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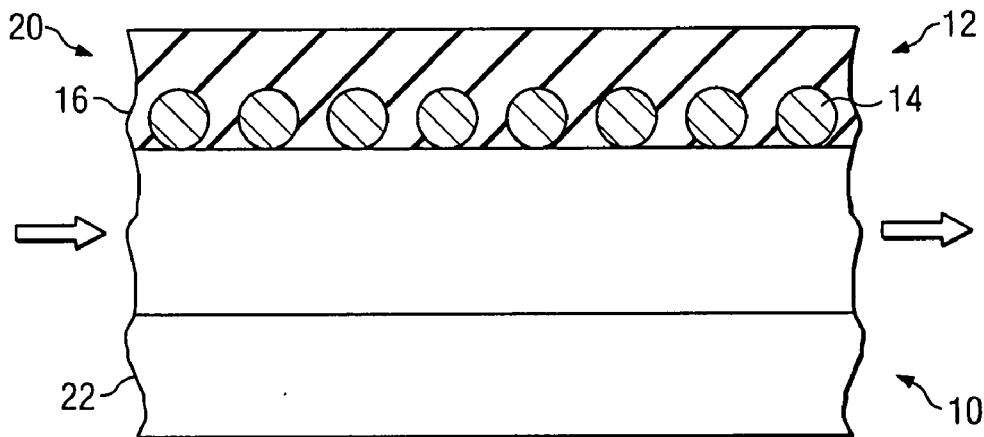
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(57) ABSTRACT

The present invention provides a sensor that includes an optical device as a support for a thin film formed by a matrix containing resonant nanoparticles. The nanoparticles may be optically coupled to the optical device by virtue of the geometry of placement of the thin film. Further, the nanoparticles are adapted to resonantly enhance the spectral signature of analytes located near the surfaces of the nanoparticles. Thus, via the nanoparticles, the optical device is addressable so as to detect a measurable property of a sample in contact with the sensor. The sensors include chemical sensors and thermal sensors. The optical devices include reflective devices and waveguide devices. Still further, the nanoparticles include solid metal particles and metal nanoshells. Yet further, the nanoparticles may be part of a nano-structure that further includes nanotubes.



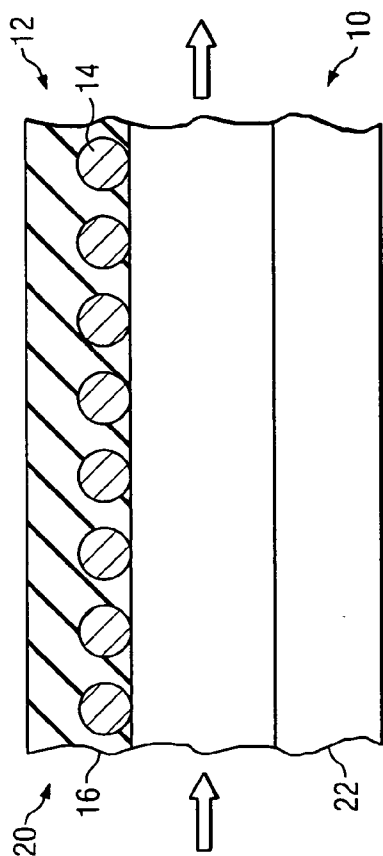


FIG. 1a

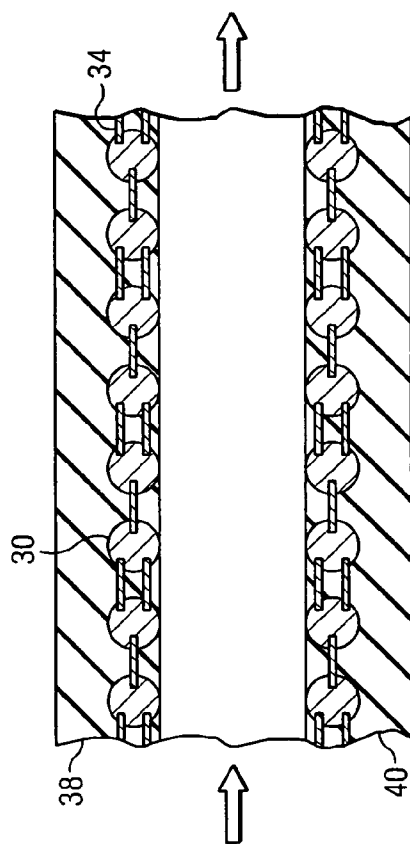


FIG. 1b

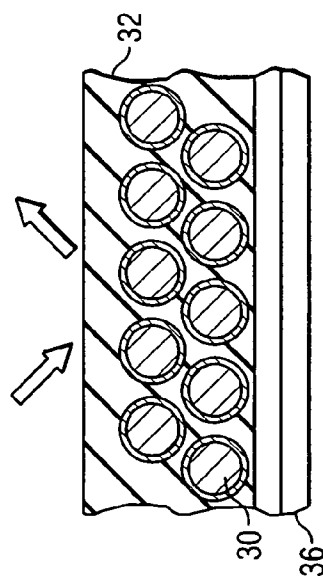


FIG. 2a

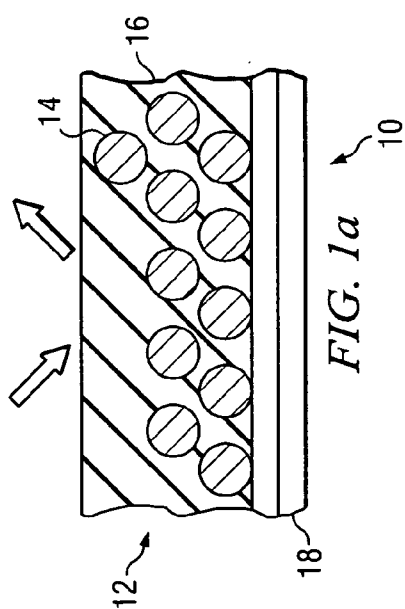


FIG. 2b

NANOPARTICLE-BASED ALL-OPTICAL SENSORS

CROSS-REFERENCE TO RELATED APPLICATIONS

[0001] This application is a divisional application of U.S. patent application Ser. No. 10/280,481, filed Oct. 24, 2002, entitled "Nanoparticle-Based All-Optical Sensors" which claims benefit of U.S. provisional patent application Ser. No. 60/335,670, filed Oct. 24, 2001, entitled "Nanoshell-Based All-Optical Sensors", and U.S. provisional patent application Ser. No. 60/339,415, filed Oct. 26, 2001, entitled "Light Interaction Between Gold Nanoshells Plasmon Resonance and Planar Optical Waveguides," each hereby incorporated herein by reference.

STATEMENT REGARDING FEDERALLY SPONSORED RESEARCH OR DEVELOPMENT

[0002] This invention was made with government support under Grant No. ECS-9801707 awarded by the National Science Foundation and Grant No. DAAD19-99-1-0315 awarded by the Army. The United States government has certain rights in the invention.

BACKGROUND OF THE INVENTION

[0003] 1. Field of the Invention

[0004] The present invention relates generally to chemical sensors that detect changes in analyte concentration or changes in temperature. More particular, the present invention relates to such sensors that incorporate resonant nanoparticles embedded in a matrix that is supported by an optically addressable device.

[0005] 2. Background of the Invention

[0006] Sensors are used to detect the concentration of an analyte and to detect the ambient temperature, among other applications. Chemical sensors based on surface plasmon resonance (SPR) are known. These sensors utilize the enhancement of Raman scattering from a Raman-active analyte that occurs when the analyte is brought near to a plasmon resonant surface. This enhancement is commonly termed surface enhanced Raman scattering (SERS).

[0007] The SERS effect is related to the field strength near the surface of the substrate upon illumination and how the analyte is bound to the surface, whether the substrate is a roughened metal surface or an aggregate of metallic nanoparticles. The strongest field enhancement is obtainable at the plasmon resonance of the metal substrate or particle. It is for this reason that gold colloid (plasmon resonance=520 nm) is such an efficient SERS enhancer under visible Raman excitation (typically with an argon ion laser at 514 nm).

[0008] Chemical sensors that incorporate metal nanoshells embedded in a matrix are described in commonly assigned co-pending patent application Ser. No. 09/616,154, filed Jul. 14, 2000, hereby incorporated herein by reference. Metal nanoshells include composite, layered nanoparticles that may include a dielectric core and a metal shell.^{1,2} Metal nanoshells are described in commonly assigned U.S. Pat. No. 6,344,272 and U.S. patent application Ser. No. 10/013,259, filed Nov. 5, 2001, each hereby incorporated herein by reference. By varying the relative dimensions of the core and shell layers, the optical absorption resonance of metal

nanoshells can be controlled and tuned across a broad region of the optical spectrum from the visible to the mid infrared.³ These frequency-agile properties are unique to nanoshells, and promise broad applicability across a range of technological applications.

[0009] Enormous increases in the detection sensitivity of molecules via the Surface-Enhanced Raman Effect can be achieved when the molecules of interest are on or near the surface of metal nanoshells and the metal nanoshell resonance is tuned to the wavelength of the excitation laser.⁴ Enhancements of $>10^6$ with infrared excitation have been observed in highly absorptive solutions, which are equivalent to enhancements of 10^{12} - 10^{14} in thin film geometries where the Raman signal is not reabsorbed. Furthermore, the metal nanoshell resonance can be tuned to the infrared region of the spectrum so that this detection capability can be realized in a region of the spectrum where compact and inexpensive semiconductor laser sources are available.

[0010] Nevertheless, despite continuing progress in SPR chemical sensors, there remains a need for a surface plasmon resonant chemical sensor having a controlled optical geometry.

SUMMARY OF THE INVENTION

[0011] The present invention is based on the discovery that an optical device may be used as a support for a thin film formed by resonant nanoparticles embedded in a matrix. The nanoparticles may be optically coupled to the optical device by virtue of the geometry of the thin film. Further, the nanoparticles are adapted to resonantly enhance the spectral signature of analytes located near the surfaces of the nanoparticles. Thus, via the nanoparticles, the optical device is addressable so as to detect a measurable property of a sample in contact with the sensor.

[0012] The optical device may be a reflective device. A reflective device preferably incorporates a reflective surface. For the purposes of the present specification a reflective surface denotes a surface having a ratio of reflectance to transmittance of at least 1.

[0013] Alternatively, the optical device may be a waveguide device. A thin film formed by the and the embedded nanoparticles may serve as a cladding layer for the waveguide device.

[0014] The measurable property may be the concentration of the analyte in a sample in chemical with the matrix, where the chemical contact allows an exchange of analyte between the and the matrix.

[0015] Alternative, the measurable property may be the temperature of a sample or environment in contact with the matrix, where the thermal contact allows an exchange of heat between the and the matrix.

[0016] It will be understood that according to some embodiments, the nanoparticles are dispersed. Thus, in these embodiments the sensor does not include a matrix.

[0017] The resonant nanoparticles are adapted to impart ultrahigh sensitivity to the sensor. Further, the sensor can be addressed and read out optically, providing remote sensing capabilities.

[0018] Thus, the present invention comprises a combination of features and advantages which enable it to overcome

various problems of prior devices. The various characteristics described above, as well as other features, will be readily apparent to those skilled in the art upon reading the following detailed description of the preferred embodiments of the invention, and by referring to the accompanying drawings, wherein like numbers refer to like elements.

BRIEF DESCRIPTION OF THE DRAWINGS

[0019] For a more detailed description of the preferred embodiment of the present invention, reference will now be made to the accompanying drawings, wherein:

[0020] **FIG. 1** is: (a) a cross-sectional view of a reflective chemical sensor according to an embodiment of the present invention; (b) a cross-sectional view of a waveguide chemical sensor according to another embodiment of the present invention; and

[0021] **FIG. 2** is: (a) a cross-sectional view of a reflective thermal sensor according to an still another embodiment of the present invention; (b) a cross-sectional view of a waveguide thermal sensor according to yet another embodiment of the present invention.

DETAILED DESCRIPTION OF THE PREFERRED EMBODIMENT

[0022] All-Optical Chemical Sensor

[0023] Referring initially to **FIG. 1**, according to an embodiment of the present invention, a chemical sensor **10** includes of a thin film **12** of resonant nanoparticles **14** embedded in a semipermeable matrix **16**. Matrix **16** is preferably semipermeable, more preferably permeable to an analyte of interest.

[0024] Matrix **16** is preferably transparent to the optical sampling wavelength and not Raman active at the Stokes shifts of interest. The optical sample wavelength may be 820 nm. Alternatively, the optical sampling wavelength may be any suitable laser wavelength. Matrix **16** may be any suitable inorganic or polymeric material. One excellent candidate inorganic material for such a matrix material is mesoporous silica.

[0025] The basic device geometry is shown in **FIG. 1**. The optical sampling geometry can be as a layer deposited onto a reflective substrate **18** (**FIG. 1(a)**) exposed to incident light. Alternatively, the optical sampling geometry can be as a cladding layer **20** in a waveguide structure **22** (**FIG. 1(b)**), where the Raman excitation is a result of the evanescent wave of the guided optical mode propagating in that structure.

[0026] In either geometry, the analytes of interest are exposed to the semipermeable layer, diffuse through this layer and are adsorbed onto the surfaces of the embedded nanoparticles. The scattered light is modulated by the Stokes modes of the analyte molecules, and detection consists of spectral analysis of the scattered light using a standard dispersive geometry and lock-in based photodetection.

[0027] One direct advantage of Raman-based chemical sensing is its insensitivity to an H₂O solvent. This approach can be used in analytical scenarios such as VOCs (volatile organic compounds) in groundwater samples or hydrocarbon mixtures in petroleum refinery or recovery. This geometry should also be amenable to vapor phase sampling of analytes.

[0028] The analyte may be any suitable analyte such disclosed in commonly assigned co-pending patent application Ser. No. 09/616,154, filed Jul. 14, 2000. The analyte may be a Raman-active chemical to be detected. Alternatively, the analyte may be a complex of a non-Raman active chemical to be detected with a Raman-active moiety.

[0029] All-Optical Temperature Sensor

[0030] The active medium of this sensor consists of nanoparticles **30** whose resonances are tuned to match the pump laser wavelength.

[0031] The nanoparticles can be functionalized with molecules that exhibit a strong Raman response. A variety of candidate molecules may be used, such as para-mercaptoaniline, which can be bound to the surface of the nanoparticles and which yields three strong Stokes modes.

[0032] Alternatively the nanoparticles can be embedded in a medium **32** exhibiting a strong Raman response.

[0033] For high temperature operation, a composite of semiconducting carbon nanotubes **34** and nanoparticles **30** can be used. Since the peak amplitudes of the corresponding Stokes and anti-Stokes modes of the Raman-active molecules are related by the Boltzmann distribution, their ratio provides an optical readout of the ambient temperature of the sensor.

[0034] A schematic of this sensor is shown in **FIG. 2**. As for the chemical sensor described above, the optical sampling geometry can be as a layer deposited onto a reflective substrate **36** (**FIG. 1(a)**) exposed to incident light. Alternatively, the optical sampling geometry can be as a layer **38** in a waveguide structure **40** (**FIG. 1(b)**), where the Raman excitation is a result of the evanescent wave of the guided optical mode propagating in that structure.

[0035] This sensor can be designed for operation with a predetermined wavelength of light. According to some embodiments, the wavelength is 820 nm. Alternatively longer wavelengths, such as 1.06 μ m may be selected, to eliminate the resonant Raman response when semiconducting carbon nanotubes are used.⁶

[0036] Nanoparticles

[0037] According to some embodiments, the resonant nanoparticles are solid metal nanoparticles. The shape of the metal nanoparticles may be selected so as to adjust the wavelength of the resonance. Thus, contemplated shapes include spheroids, ellipsoids, needles, and the like. Further the metal nanoparticles may be aggregated into multi-nanoparticle aggregates so as to adjust the wavelength of the resonance. Still further, the metal nanoparticles may be embedded in a matrix material that is capable of adjusting the wavelength of the resonance. For example, the matrix may be any dielectric material suitable to form the core of a metal nanoshell.

[0038] According to other embodiments, the resonant nanoparticles are metal nanoshells. The metal nanoshells may be metal nanoshells having multiple shell layers, termed herein multilayer nanoshells. Further the metal nanoshells may be aggregated into multi-nanoshell aggregates so as to adjust the wavelength of the resonance.

[0039] The wavelength of the resonance is preferably selected so as to provide surface enhanced Raman scattering.

The wavelength may be controlled by controlling the geometry of the metal nanoparticle¹¹.

[0040] According to some embodiments of the present invention, the nanoparticles are islands, such as may be formed as a stamped surface¹⁰.

[0041] According to some embodiments of the present invention, the nanoparticles are arranged in a random array. Random as used herein denotes lacking X-ray scattering peaks with the range of length scales up to mesoscopic. According to some embodiments of the present invention, the nanoparticles are arranged in a regular array. Regular as used herein denotes possessing at least one X-ray scattering peak with the range of length scales up to mesoscopic.

[0042] According to some embodiments of the present invention, the nanoparticles are arranged in a two dimensional array. Alternatively, according to some embodiments of the present invention, the nanoparticles are arranged in a three dimensional array. Yet alternatively, the thin film may contain an arrangement of nanoparticles having a fractional dimension between two and three.

[0043] Optical Device

[0044] It will be understood that the present optical device, such as a reflective device or a waveguide device, may be a component in an optical apparatus. Optical apparatuses that are contemplated include optical computing elements, holographic devices, optical correlators, optical phase conjugators, bistable memory devices, optical limiters, polarization filters, and infrared and visible light detectors.

[0045] When the optical device includes a reflective surface, the reflective surface may be a mirror. Alternatively, a reflective surface may be a stack of dielectric thin films of alternating high and low refractive index. Such stacks are known that approach upwards of at least 90% reflectance. A spacer layer may be disposed between the reflective surface and the thin film containing the nanoparticles. The spacer layer may be formed of a dielectric material.

[0046] When the optical device includes a waveguide, the waveguide may include a dielectric layer supported on a metal layer. The thickness of the dielectric layer is preferably selected so as to support optical waves propagating parallel to the interface between the dielectric layer and the metal layer. The thin film layer containing the resonance nanoshells may form a cladding layer of the waveguide.

[0047] Methods of making the present optical devices include conventional microfabrication techniques such as known to one of ordinary skill in the art.

[0048] Optical Coupling

[0049] The thin film is preferably optically coupled to the optical device. The optical coupling preferably occurs as a result of the geometry of the thin film with respect to the optical device. It will be understood that the preferred average distance between a nanoparticle and a surface of the optical device may vary according to the wavelength of the maximum resonance of the nanoparticle, also termed herein resonant wavelength.

[0050] The average nanoparticle distance to the nearest surface of the optical device is preferably up to a value on the order of the resonant wavelength. The average distance

to the nearest surface is preferably determined as the average length of a vector oriented perpendicular to the outer surface of the optical device and extending from that outer surface to the center of mass of a nanoparticle.

[0051] The average nanoparticle distance to a light directing surface as disclosed herein is likewise preferably up to a value on the order of the resonant wavelength. The average distance to the light directing surface is preferably determined as the average length of a vector oriented perpendicular to the light directing surface and extending from that light directing surface to the center of mass of a nanoparticle.

[0052] The light directing surface may be a metal surface in a waveguide. Alternatively, the light directing surface may be a reflective surface.

[0053] Exemplary light scattering experiments described in U.S. Provisional Application No. 60/339,415 that were performed on gold nanoshells randomly deposited on a dielectric layer supported on a gold layer show a change in the scattering spectrum of the nanoshells due to coupling of light with the waveguide modes. Thus, these experiments demonstrated optically coupling of metal nanoshells deposited on a waveguide structure with the waveguide. It is believed that these results extend to the present nanoparticles embedded in the present matrix supported on the present optical device.

[0054] Thin Film Formation

[0055] Forming the thin film preferably includes depositing a matrix material onto the optical device. The exposed surface of the optical device may be a metallic material. Alternatively, the exposed surface of the optical device may be a non-metallic material such as a dielectric material. The deposition may include spin-coating the matrix material. The matrix material may be in the form of a fluid precursor during the deposition. The formation of the thin film then includes drying the fluid precursor so as to form the matrix as a solid that is preferably still gas or liquid permeable. Suitable inorganic materials include silica or other oxides that may be formed by a sol-gel process. Suitable polymeric materials include polyvinyl acetate (PVA).

[0056] The nanoparticles may be mixed into the fluid precursor prior to deposition. Metal nanoshells have been successfully mixed by the present inventors into various polymers including PVA, polyvinylpropylene (PVP), polymethylmethacrylate (PMMA), and polydimethylsiloxane (PDMS). Further, methods for incorporating gold nanoparticles in a silica sol-gel matrix are known to one of ordinary skill in the art. These methods are contemplated for incorporating the present nanoparticles into inorganic oxide matrices.

[0057] Alternatively, nanoparticles or other nanostructure may be formed on the optical device so as to form a composite structure, followed by depositing the fluid precursor to the composite structure.

[0058] According to some embodiments, forming the composite structure includes evaporating a solution a concentrated solution of the nanoparticle. A suitable exemplary method in which the optical device is a waveguide and the nanoparticles are gold nanoshells is described in the paper entitled "Light Interaction Between Gold Nanoshell Plas-

mon Resonance and Planar Optical Waveguides" contained in Provisional Application No. 60/339,415, which is incorporated herein by reference.

[0059] In an exemplary method, an approximately 200 nm thick layer of gold was sputter coated onto an indium tin oxide (ITO) coated glass slide. Self-Assembled Monolayers (SAM's) of a cationic polyelectrolyte PDDA (poly(diallyldimethylammonium chloride) and anionic sheets of an exfoliated synthetic clay (Laponite RD, a synthetic form of hectorite) were deposited on the gold surface to control the spacing s to nominally nm precision between the gold surface and the gold nanoshells. A sub monolayer of gold nanoshells, with an average spacing of 200 nm and approximately 27% coverage (as determined by scanning electron microscopy) was deposited on the SAM's by evaporating 10-20 μ l of concentrated aqueous solution containing gold nanoshells.

[0060] According to other embodiments, forming the composite structure includes mask-free lithographic formation of metal structures, such as metallic arrays.

[0061] In an exemplary method¹⁰, PDMS stamps were prepared in a standard way using an elastomer kit (Sylgard 184, Dow Corning). Diffraction gratings were purchased from Edmund Optics. Glass microscope slides were cleaned in piranha etch (7:3 v/v 98% H₂SO₄:30% H₂O₂) for 1 hour, rinsed in ultrapure water (Milli-Q system, Millipore) and dried with a stream of filtered N₂. n-Propyltrimethoxysilane (PTMS), HAuCl₄, and K₂CO₃ were purchased from Sigma-Aldrich Corp. and used as received. Silver plating was accomplished using a commercially available silver plating kit (HE-300, Peacock Laboratories Inc.) Scanning electron microscopy (SEM) was performed on a Phillips XL-30 ESEM. Atomic force microscopy (AFM) was performed on a Digital Instruments Nanoscope III.

[0062] Glass microscope slides were patterned with PTMS using stamps made from diffraction gratings and standard microcontact printing procedures. After the siloxane molecules had condensed on the surface (12 hours) the slides were exposed to a solution of SnCl₂ (Peacock Laboratories Inc.) for 5-10 seconds which activates the unstamped regions for metal reduction. Once activated the slides were washed with Milli-Q water and immediately exposed to silver or gold electroless plating solutions for a period of seconds or minutes until the metal had reduced onto the activated regions of the slides. Typical plating times ranged from 15 seconds to 1 minute. The silvering solution was used according to the provided instructions, while the gold solution was prepared by diluting 1 mL of a 1% HAuCl₄ solution in 100 mL H₂O and adding 25 mg K₂CO₃. After plating samples were rinsed well with water and dried with filtered nitrogen.

REFERENCES

[0063] The following references, referred to herein by accompanying number are hereby incorporated herein by reference in their entirety for all purposes:

[0064] 1. R. D. Averitt, D. Sarkar & N. J. Halas. Plasmon Resonance Shifts of Au Coated Au₂S Nanoshells: Insight into Multicomponent Nanoparticle Growth. *Physical Review Letters* 78, 4217-4220 (1997).

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[0069] 6. S. D. M. Brown, P. Corio, A. Marucci, M. S. Dresselhaus, M. A. Pimenta & K. Kneipp. Anti-Stokes Raman Spectra of single-walled carbon nanotubes. *Physical Review B: Rapid Communications* 61, 5137-5140 (2000).

[0070] 7. Y. Xia & G. M. Whitesides. Soft Lithography. *Angewandte Chemie International Edition* 37, 550-575 (1998).

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[0072] 9. K. F. Kelly, G. D. Hale, D. Sarkar & N. J. Halas. Threefold Electron Scattering on Graphite Observed with C60-Modified STM Tips. *Science* 273, 1371-1373 (1996).

[0073] 10. C. E. Moran, C. Radloff, and N. J. Halas. Mask-Free Passivation Stamp (MAPS) Lithography: Larger Area Fabrication and Geometric Variation of Submicron Metal Line and Island Arrays. _____. Preprint provided concurrently herewith.

[0074] 11. J. B. Jackson, S. L. Westcott, L. R. Hirsch, J. L. West, and N. J. Halas, Controlling the Surface Enhanced Raman Effect Via the Nanoshell Geometry. *Applied Physics Letters* _____, _____ (____). Preprint provided concurrently herewith.

[0075] Should the disclosure of any of the patents, patent applications, preprints, and publications that are incorporated herein conflict with the present specification to the extent that it might render a term unclear, the present specification shall take precedence.

[0076] While preferred embodiments of this invention have been shown and described, modifications thereof can be made by one skilled in the art without departing from the spirit or teaching of this invention. The embodiments described herein are exemplary only and are not limiting. Many variations and modifications of the system and apparatus are possible and are within the scope of the invention. Accordingly, the scope of protection is not limited to the embodiments described herein, but is only limited by the claims that follow, the scope of which shall include all equivalents of the subject matter of the claims.

What is claimed is:

1. A sensor comprising:

an optical device; and

a thin film supported by said device, said thin film comprising

a matrix; and

a plurality of plasmon resonant particles embedded in said matrix;

wherein said matrix is selected from the group consisting of a transparent matrix, a Raman-active matrix, a

semipermeable matrix, a mesoporous matrix, an inorganic oxide, a polymer, a material selected so as to control the maximum wavelength of the resonance of said resonant particles, and combinations thereof.

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