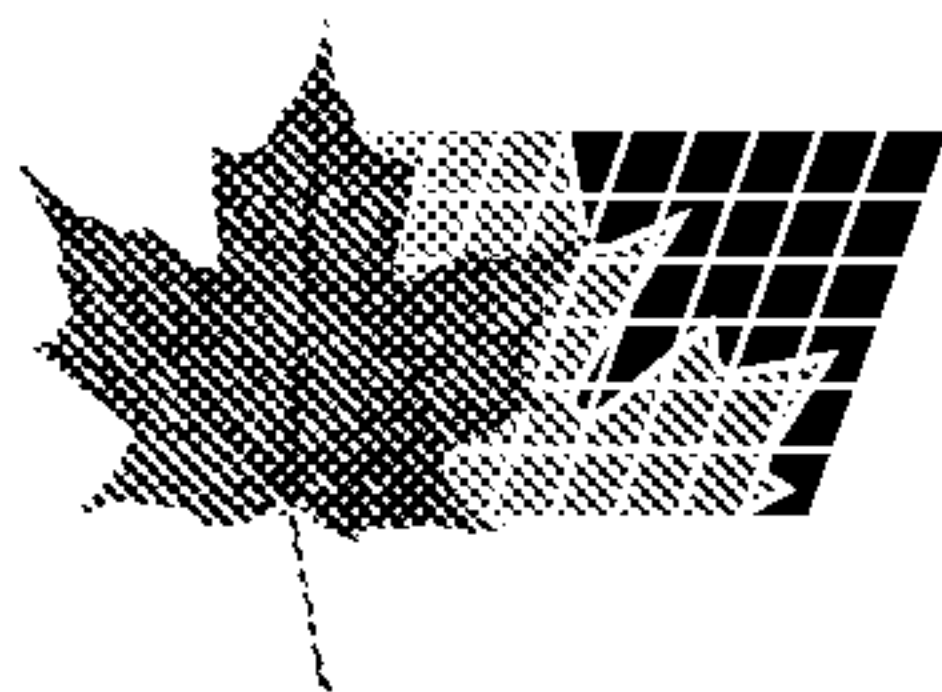




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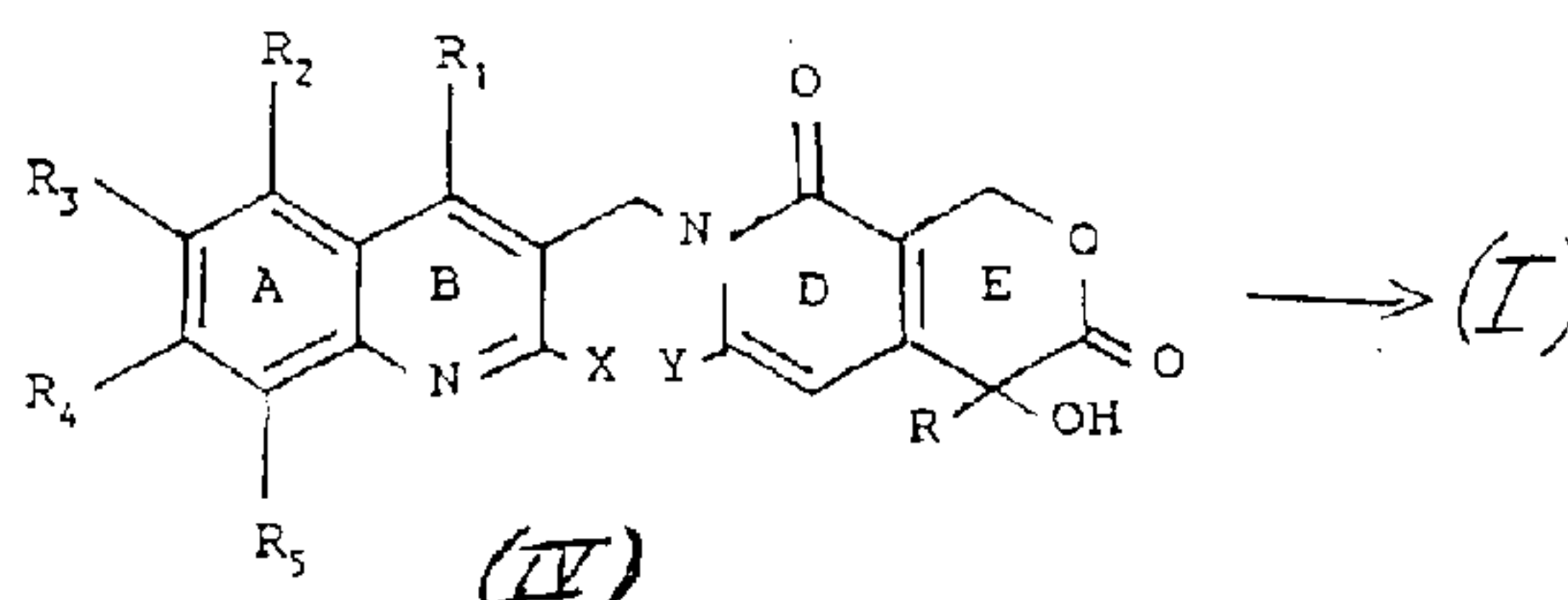
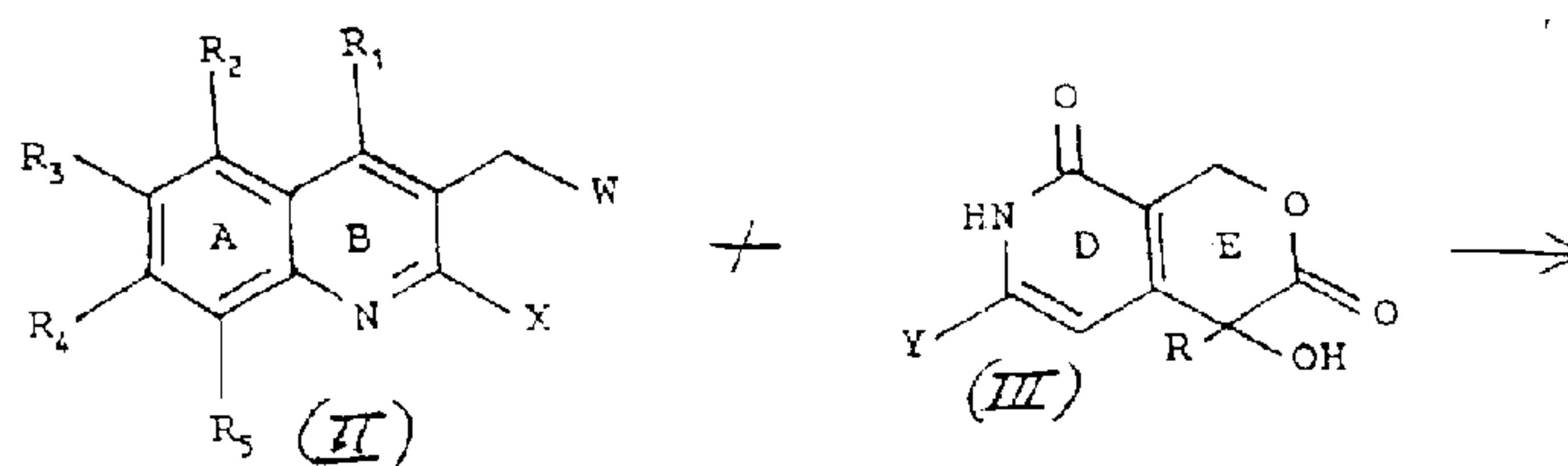
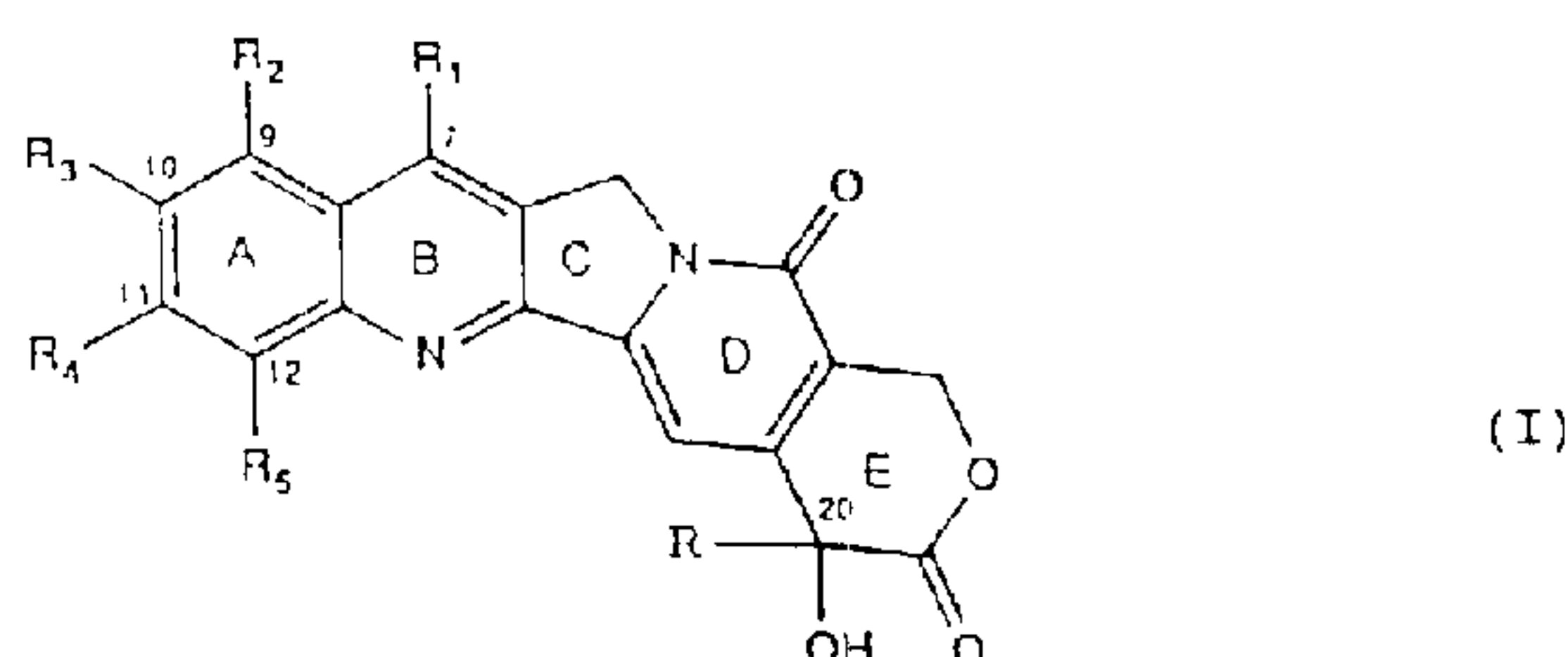
(72) COMINS, DANIEL L., US
(72) BAEVSKY, MATTHEW F., US
(71) NORTH CAROLINA STATE UNIVERSITY, US

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(30) 1990/12/20 (632,970) US

(54) **METHODE DE PREPARATION DE CAMPTOTHECINE ET
D'ANALOGUES DE CELLE-CI**

(54) **METHOD OF MAKING CAMPTOTHECIN AND
CAMPTOTHECIN ANALOGS**



(57) Compounds of Formula I (See formula I) are made in accordance with the following scheme: (See formula II, III, IV) wherein R may be loweralkyl; R₁ may be H, loweralkyl, loweralkoxy, or halo; R₂, R₃, R₄, and R₅ may each 26 independently be H, amino, hydroxy, loweralkyl, loweralkoxy, loweralkylthio, di(loweralkyl)amino, cyano, methylenedioxy, formyl, nitro, halo, trifluoromethyl, aminomethyl, azido, amide, hydrazino, or any of the twenty standard amino acids bonded to the A ring via the amino-nitrogen atom; Y is H and W and X are halogen. Also disclosed are novel processes for making starting materials for the scheme given above, and novel intermediates employed in these processes.

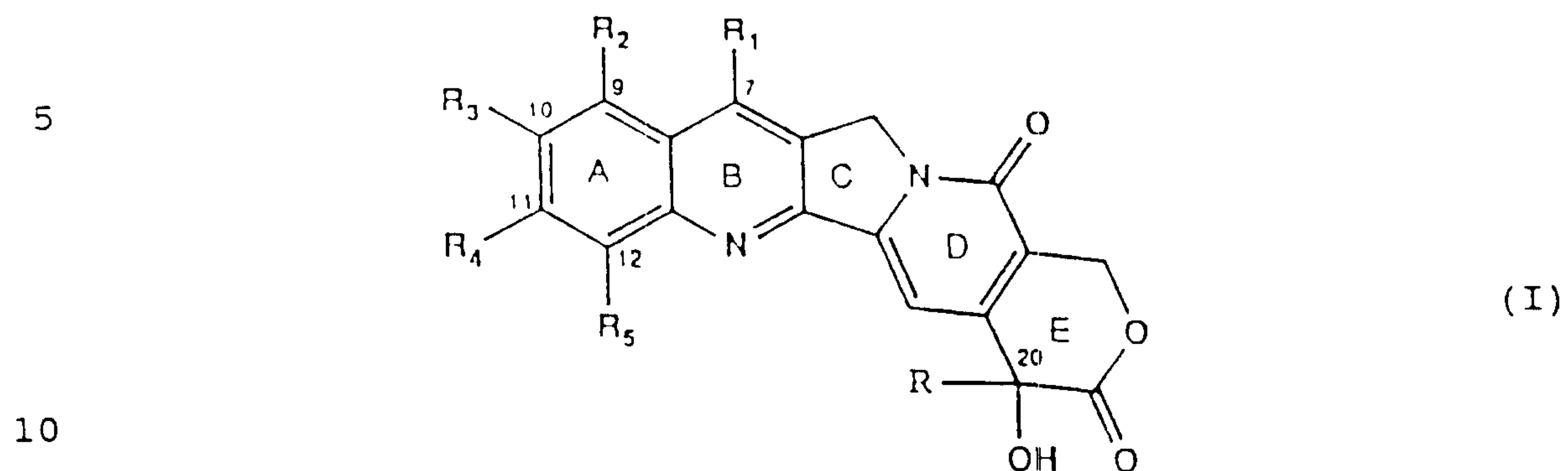


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METHOD OF MAKING CAMPTOTHECIN AND CAMPTOTHECIN ANALOGS

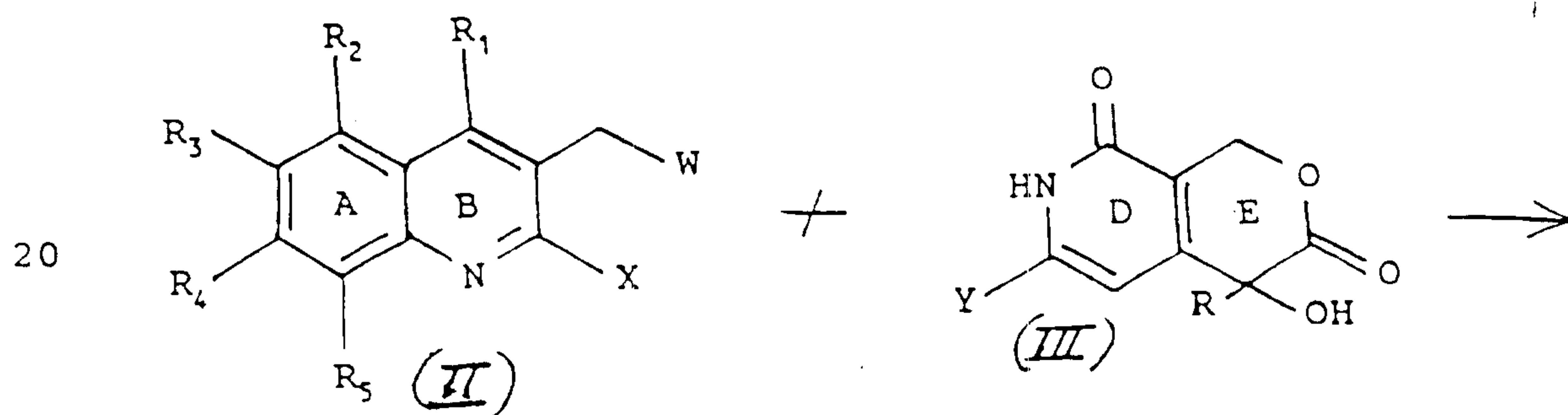
Abstract

Compounds of Formula I

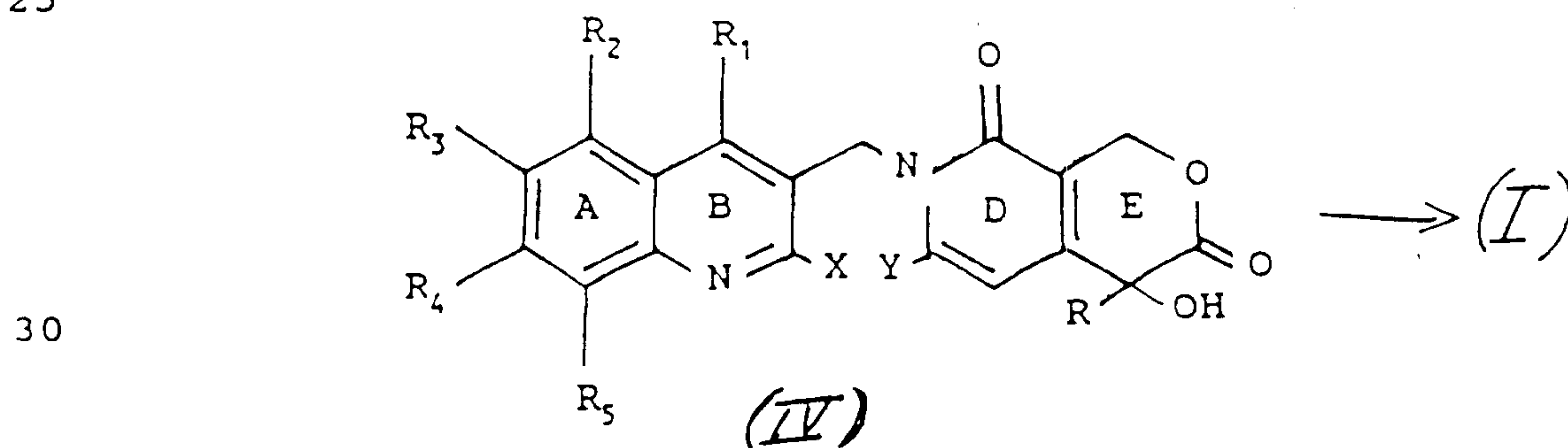


are made in accordance with the following scheme:

15



25



wherein R may be loweralkyl; R₁ may be H, loweralkyl,
loweralkoxy, or halo; R₂, R₃, R₄, and R₅ may each

independently be H, amino, hydroxy, loweralkyl, loweralkoxy, loweralkylthio, di(loweralkyl)amino, cyano, methylenedioxy, formyl, nitro, halo, trifluoromethyl, aminomethyl, azido, amido, hydrazino, or any of the twenty standard amino acids bonded to the A ring via the amino-nitrogen atom; Y is H and W and X are halogen. Also disclosed are novel processes for making starting materials for the scheme given above, and novel intermediates employed in these processes.

METHOD OF MAKING CAMPTOTHECIN AND CAMPTOTHECIN ANALOGS

Field of the Invention

The present invention provides a parallel synthesis of camptothecin and camptothecin analogs via novel intermediates at high yields.

5

Background of the Invention

Camptothecin (Chem. Abstracts Registry No. 7689-03-4) is a naturally occurring compound found in *Camptotheca acuminata* (Nyssaceae) which has anti-leukemic and antitumor properties. Numerous
10 camptothecin analogs having like properties are known, examples being those described in U.S. Patent No. 4,894,456 to Wall et al. and European Patent Application No. 0 325 247 of Yaegashi et al.

A number of syntheses for camptothecin are
15 known. Several routes are reviewed in Natural Products Chemistry, Vol. 2, 358-361 (K. Nakanishi, T. Goto, S. Itô, S. Natori and S. Nozoe eds.) and in J. Cai and C. Hutchinson, Camptothecin, in The Alkaloids, Vol. XXI, 101-137 (Academic Press 1983). The biosynthesis of
20 camptothecin is described in Natural Products Chemistry, Vol. 3, 573-574 (K. Nakanishi et al. eds.). A recent synthetic route is described in U.S. Patent

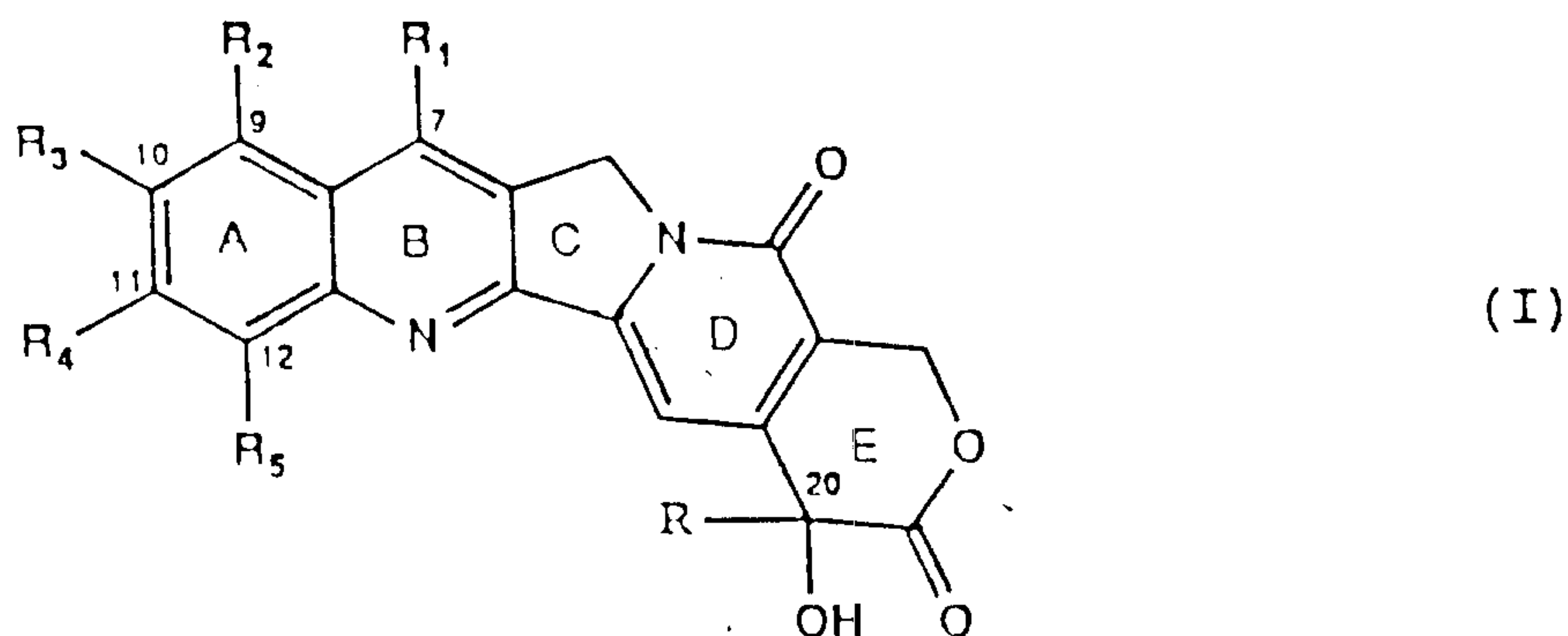
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No. 4,894,456 to Wall et al. (see also references cited therein).

A problem with prior methods of synthesizing camptothecin is that they are largely linear syntheses. Such syntheses provide low yields of the final product because of the sequential loss in product during each step of the total synthesis. Parallel syntheses (i.e., a strategy in which two synthetic paths are followed separately and the products thereof combined to form the final product) provide higher yields, but few such syntheses have been available for camptothecin. Accordingly, an object of the present invention is to provide a parallel synthetic method for making camptothecin and analogs thereof.

Summary of the Invention

The present invention provides a method of making compounds of Formula I below:



wherein:

R may be loweralkyl, preferably ethyl.

R₁ may be H, loweralkyl, loweralkoxy, or halo (e.g., chloro). Preferably R₁ is H.

R₂, R₃, R₄, and R₅ may each independently be H, amino, hydroxy, loweralkyl, loweralkoxy, loweralkylthio, di(loweralkyl)amino, cyano, methylenedioxy, formyl, nitro, halo, trifluoromethyl,

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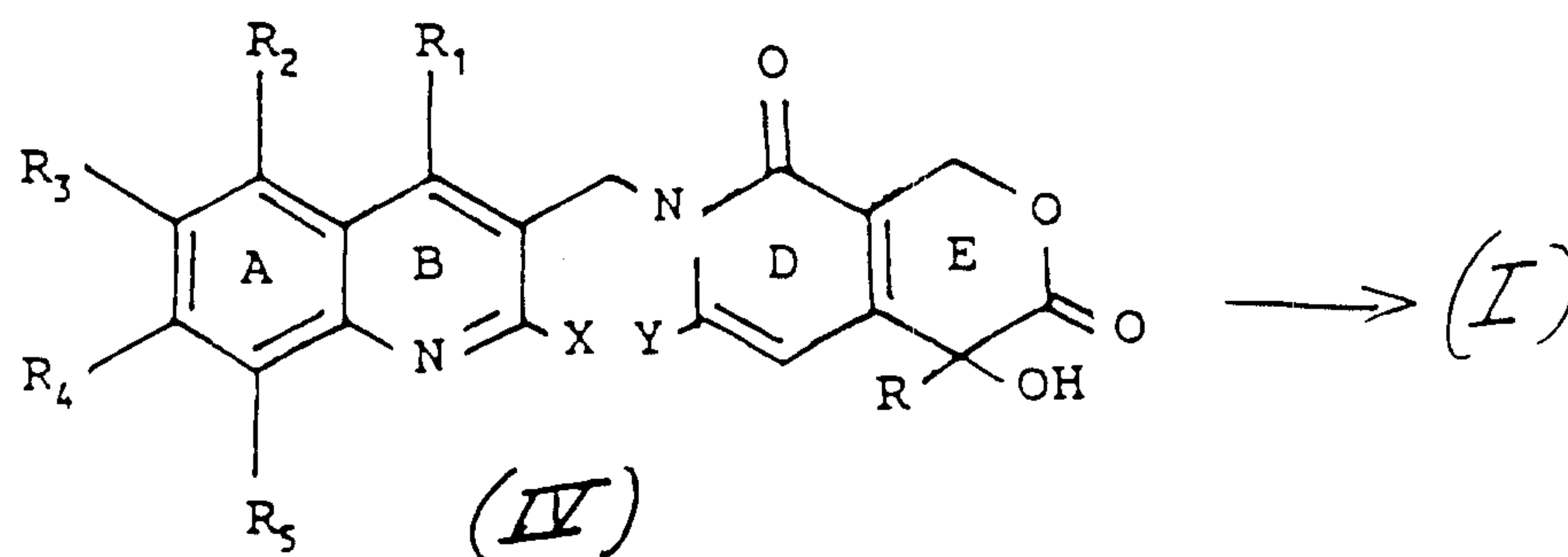
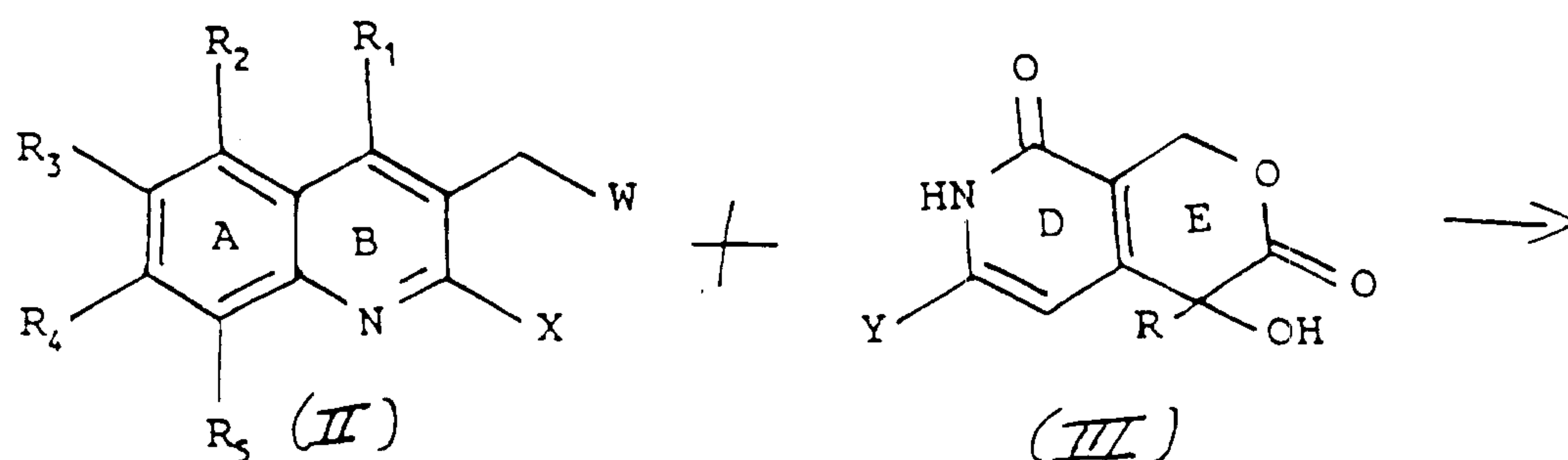
aminomethyl, azido, amido, hydrazino, or any of the twenty standard amino acids bonded to the A ring via the amino-nitrogen atom (numbering in Formula I is by the Le Men-Taylor numbering system and rings are lettered in the conventional manner. See J. Cai and C. Hutchinson, supra at 102).

At least two of R_2 , R_3 , R_4 , and R_5 may be H, and in a preferred embodiment R_2 , R_4 , and R_5 are H.

Preferably: R_2 is H or amino; R_3 is H or hydroxy; R_4 is H; and R_5 is H.

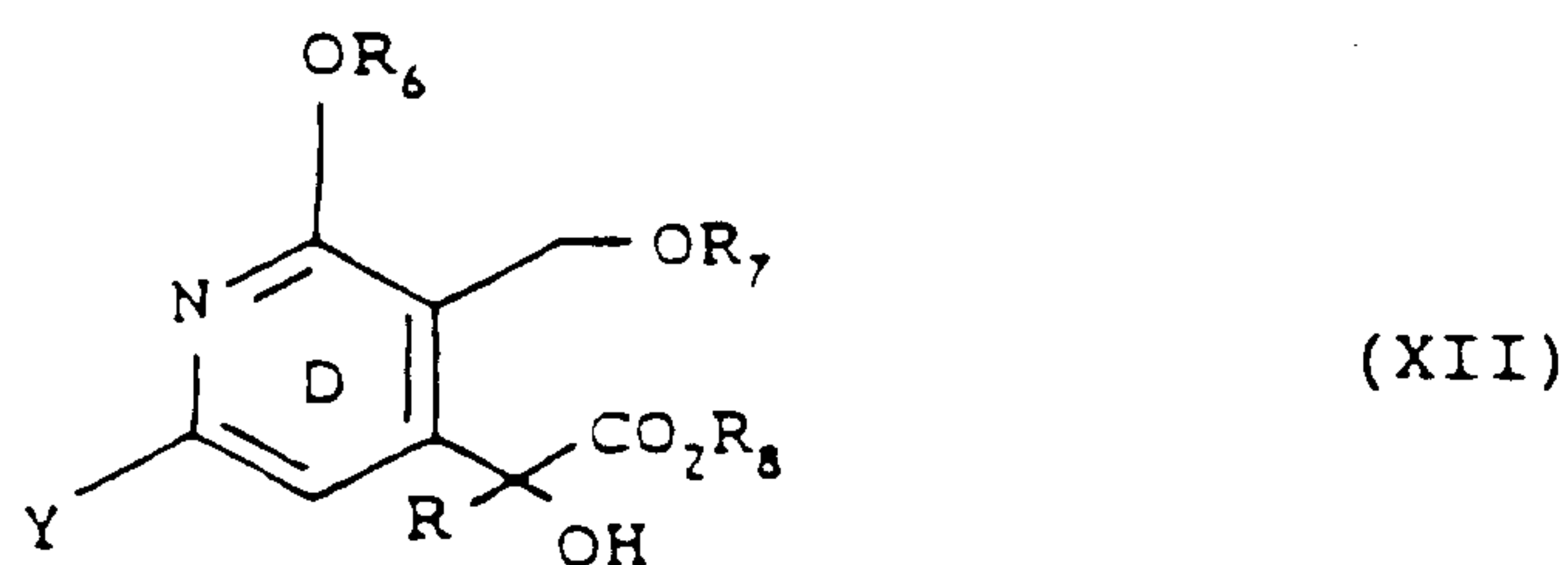
In the invention, a compound of Formula I is produced according to scheme A below, where Y is H, R_1 through R_5 are as given in connection with Formula I above, X is halogen, preferably bromo or iodo; and W is halogen, preferably chloro.

Scheme A

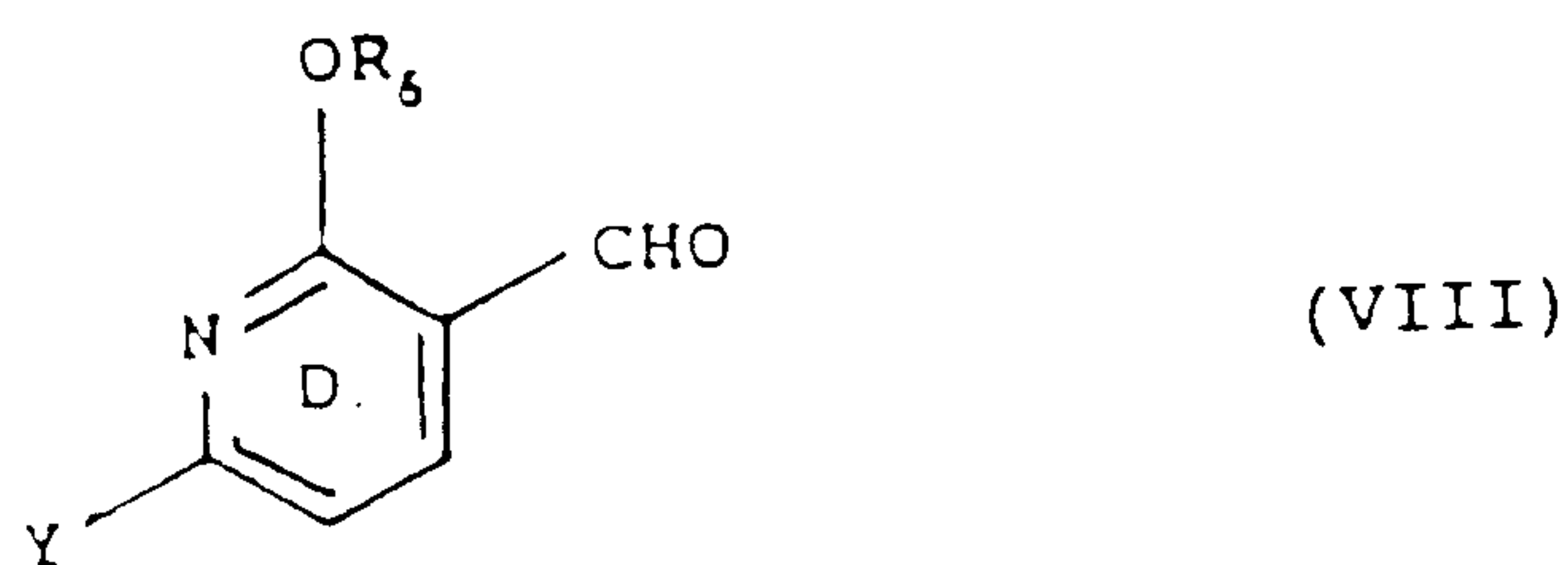


3a

In accordance with one embodiment of the present invention, a method of making a compound of Formula XII:

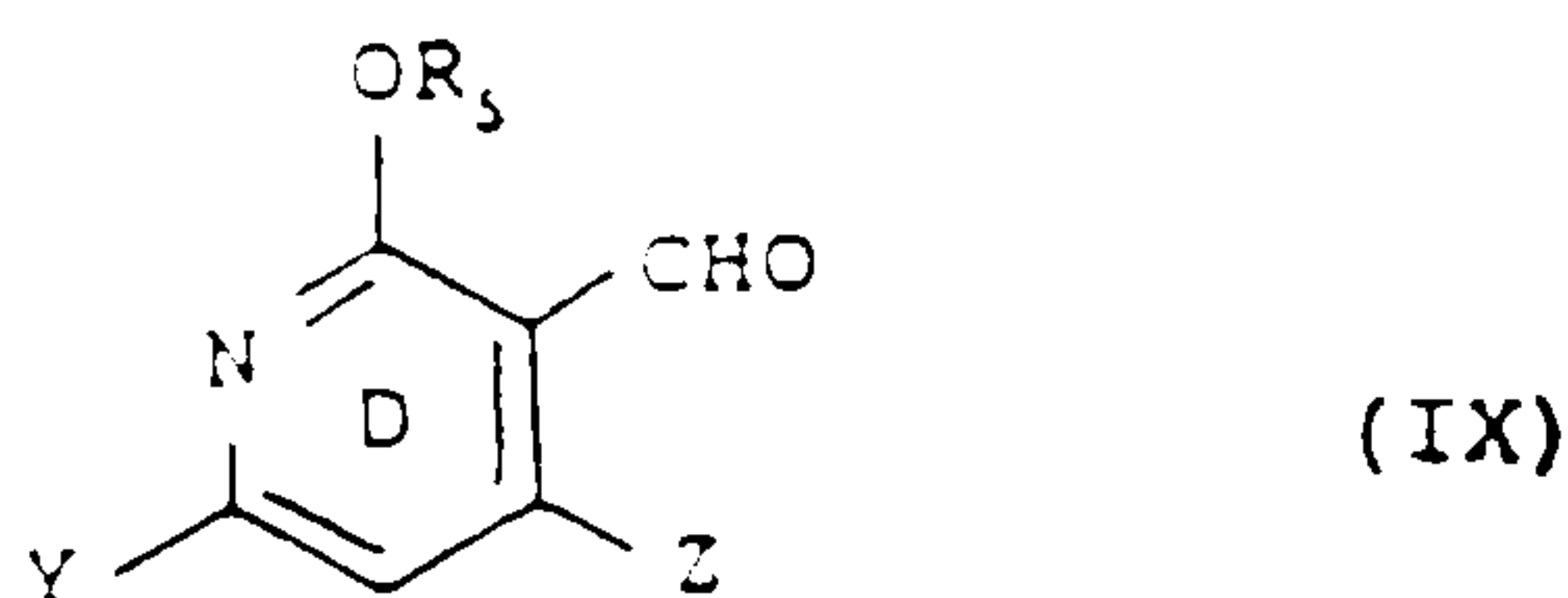


10 wherein Y is H or Cl, and R, R₆, R₇, and R₈ are loweralkyl, from a carboxaldehyde of Formula VIII:



20 comprises

halogenating the carboxaldehyde of Formula VIII to produce a 4-halo-pyridinecarboxaldehyde of Formula IX:

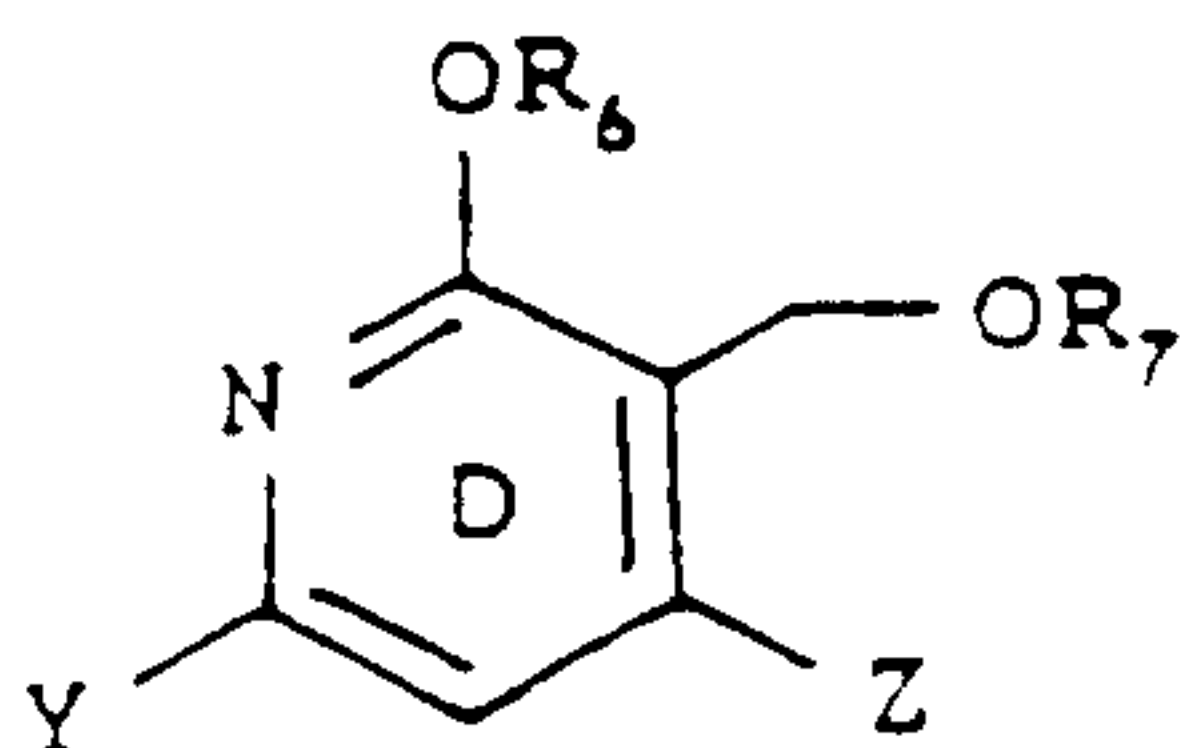


30 wherein Z is selected from the group consisting of bromo and iodo; and then

3 b

reducing the 4-halo-pyridinecarboxaldehyde of Formula IX to give an alkoxymethylpyridine of Formula X:

5



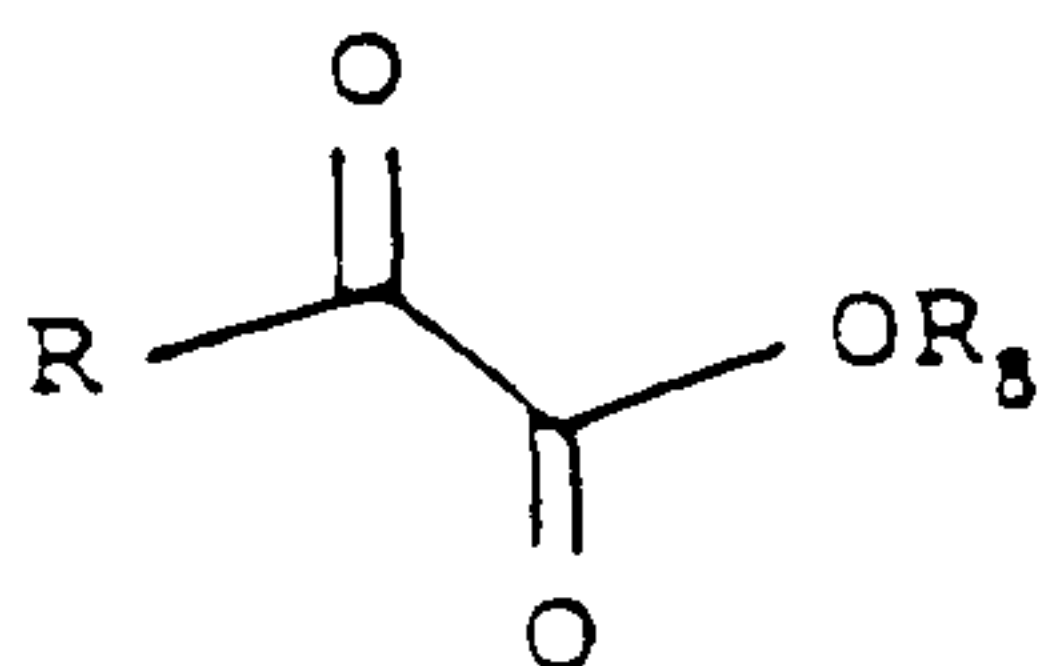
(X)

10

wherein R_7 is loweralkyl; and then

alkylating the alkoxymethylpyridine of Formula X with a compound of Formula XI:

15

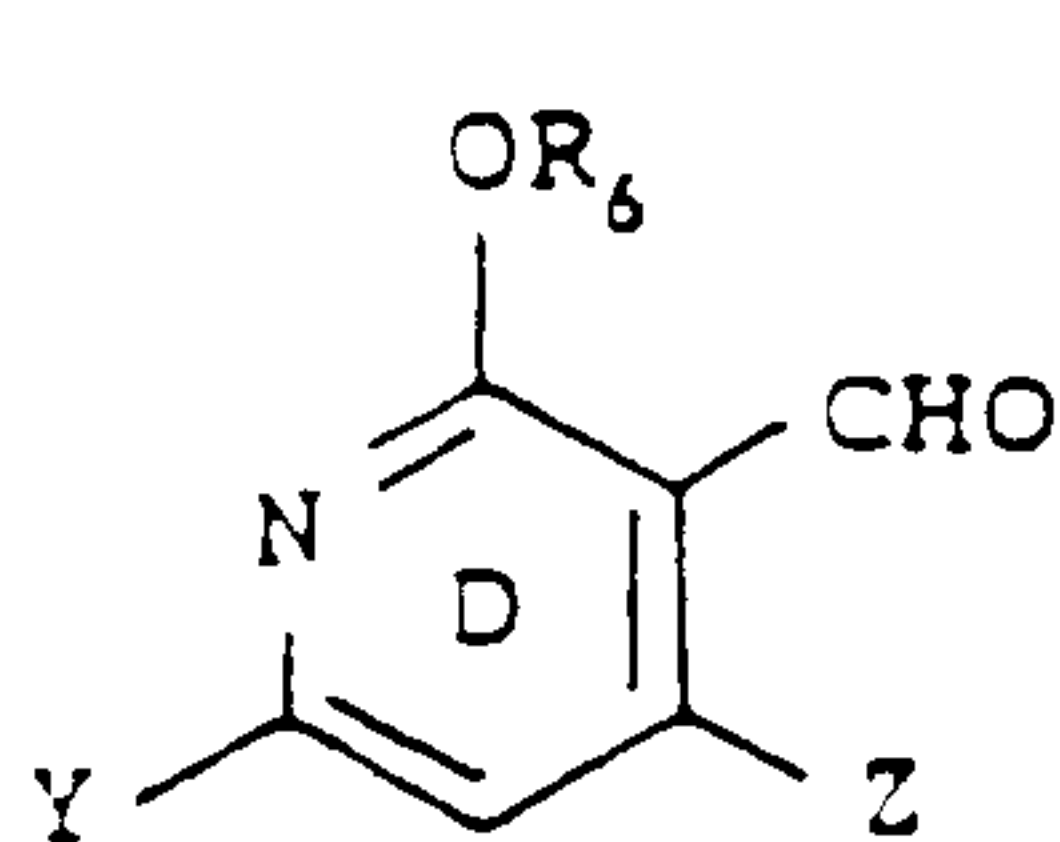


(XI)

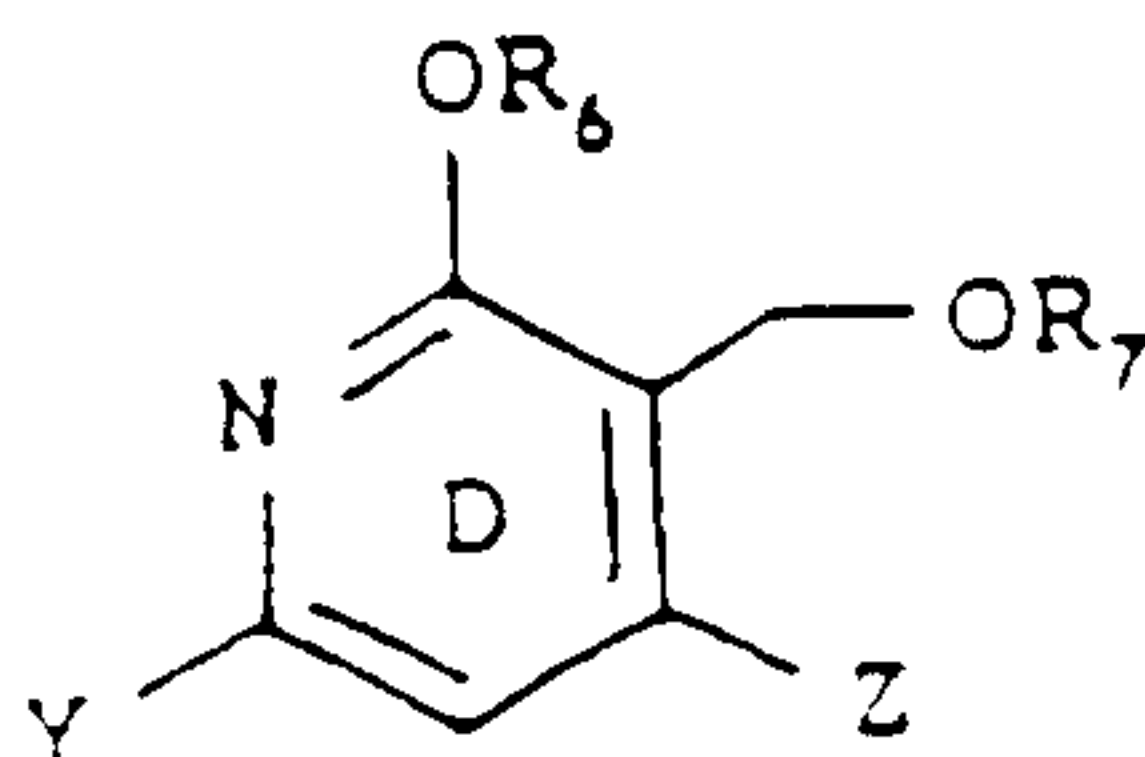
20 to yield the compound of Formula XII.

In accordance with another embodiment of the present invention, is a compound selected from the group consisting of compounds of Formula IX, Formula X, and Formula XII:

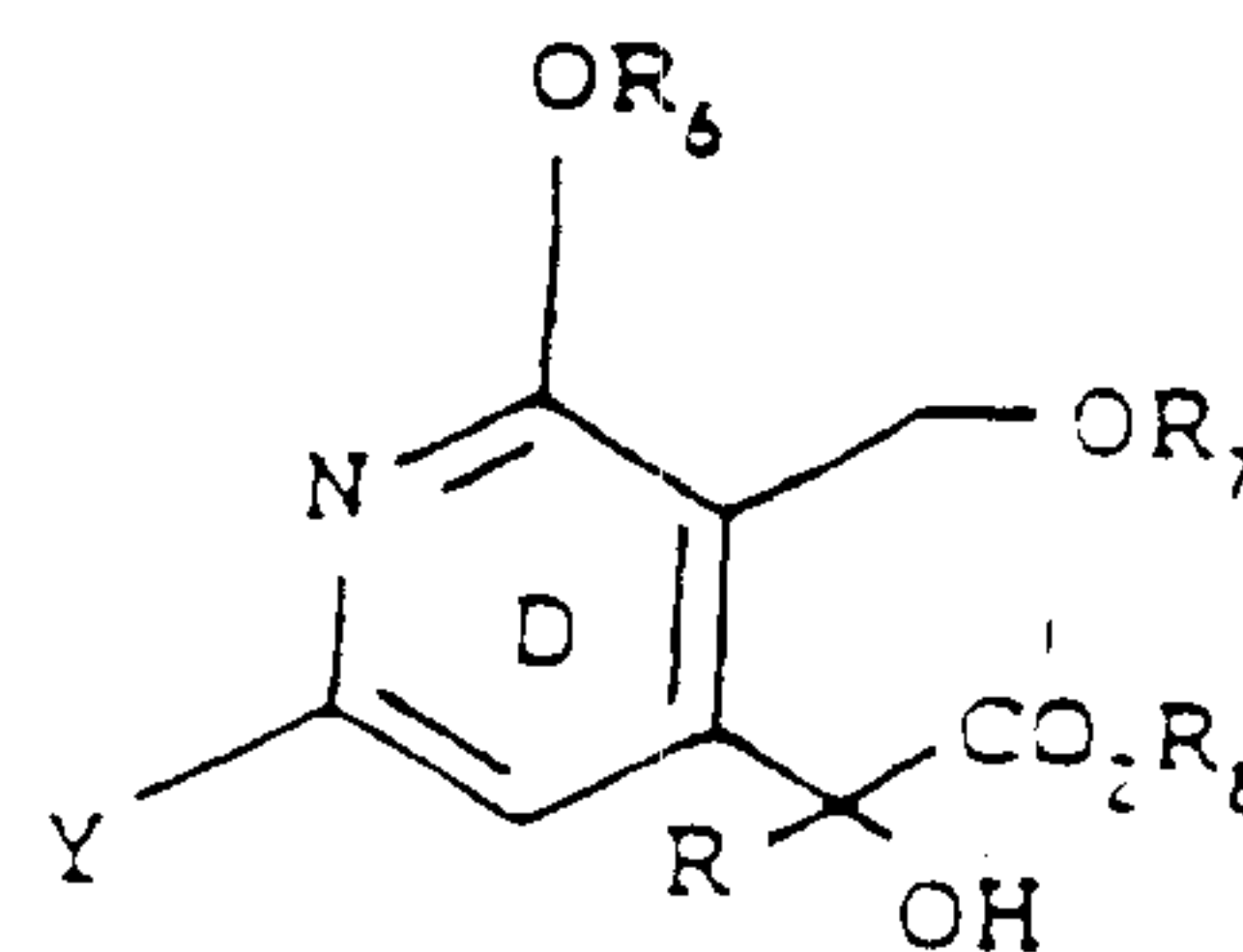
25



(IX)



(X)



(XII)

wherein in each of the above formulae:

30 R , R_6 , R_7 , and R_8 are loweralkyl;

Y is Cl or H; and

Z is halogen.

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Detailed Description of the Invention

As used herein, the term "loweralkyl" means a linear or branched alkyl group with 1-8, preferably 1-4, carbon atoms, such as methyl, ethyl, propyl, isopropyl, n-butyl, tert-butyl, hexyl, and octyl. This definition also applies to a loweralkyl moiety in the loweralkoxy, loweralkylthio, and di(loweralkyl)amino groups. Thus, examples of loweralkoxy groups are methoxy, ethoxy, propoxy, sec-butoxy, and isohexoxy; examples of loweralkylthio groups are methylthio, ethylthio, tert-butylthio, and hexylthio; and examples of di(loweralkyl)amino groups are dimethylamino, diethylamino, diisopropylamino, di(n-butyl)amino, and dipentylamino.

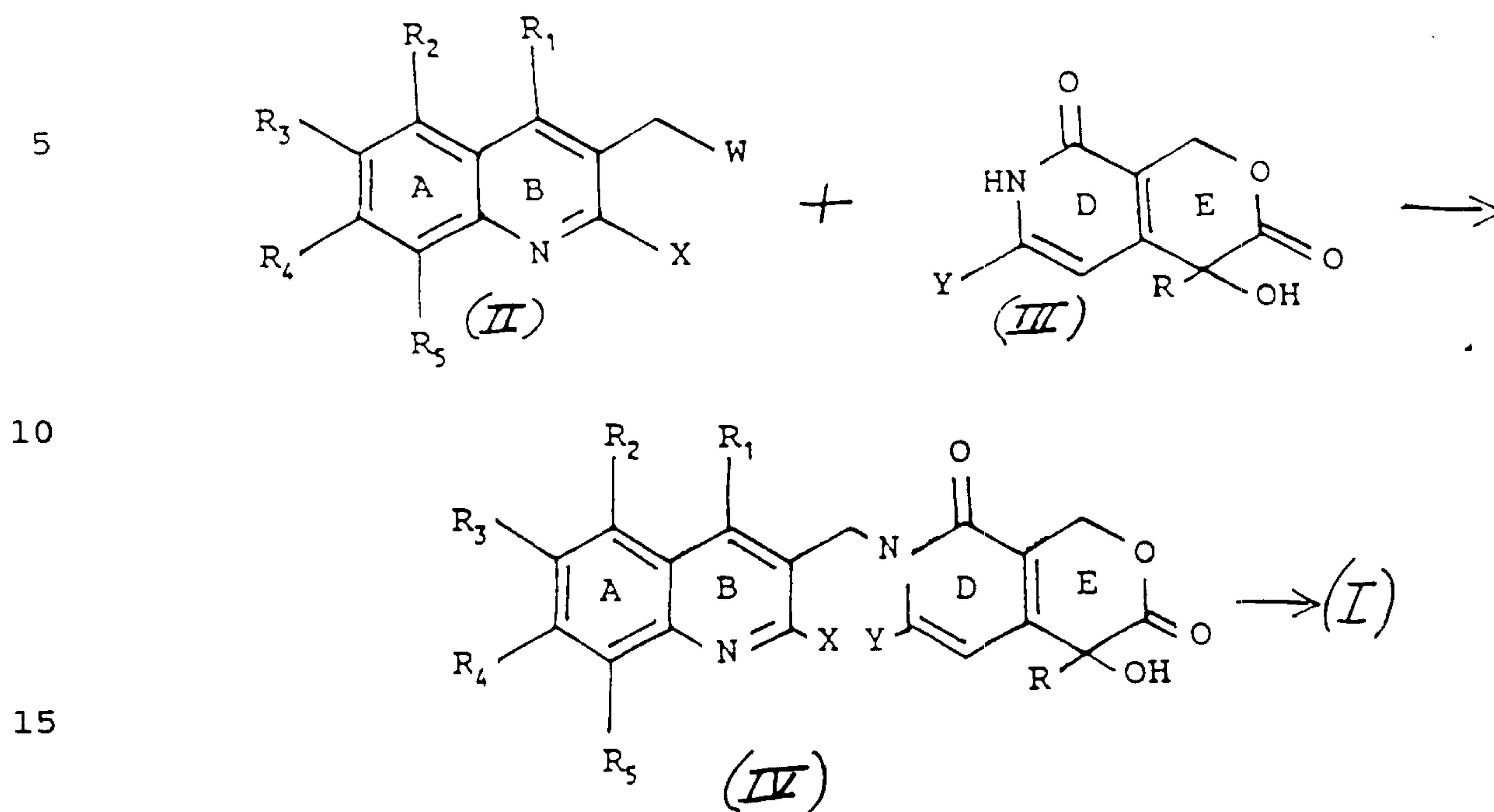
The terms "halo" and "halogen" as used herein refers to a substituent which may be fluoro, chloro, bromo, or iodo.

Substituents on the "A" ring of the compounds disclosed herein may be joined together to form a bifunctional substituent such as the methylenedioxy group. Methyleneedioxy substituents may be bonded to any two consecutive positions in the A ring, for example, the 9,10, the 10,11, or the 11,12 positions.

Substituents which are standard amino acids may be any of the twenty amino acids commonly found in naturally occurring proteins, and are well known in the art. These provide a substituent of the formula -NHCH(R)COOH , with R being the side chain of any of the twenty standard amino acids. The amino acids may be of any configuration, but preferably have an (L) configuration.

A compound of Formula I is produced in accordance with Scheme A below by alkylating a pyridone of Formula III with a chloromethylquinoline of Formula II to produce a compound of Formula IV, and then cyclizing the compound of Formula IV to yield the compound of Formula I.

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Scheme A

In Scheme A: Y is H; R and R₁ through R₅ are as given in connection with Formula I above; X is halogen, preferably bromo or iodo; and W is halogen, preferably chloro.

The starting materials of Scheme A, the compounds of Formula II and III, are prepared in accordance with Schemes B and C below.

25 The pyridone of Formula III may be alkylated with a halomethylquinoline of Formula II in a suitable solvent, such as a polar protic solvent (e.g., isopropyl alcohol, ethanol, methanol), an aprotic solvent (e.g., 1,2-dimethoxyethan, tetrahydrofuran, toluene, acetonitrile, or dimethylformamide) or alternatively in an aqueous solution in the presence of a phase transfer catalyst. The reaction is preferably carried out under mildly basic conditions, to minimize attack on the pyridone ring oxygen. The reaction may be carried out as a single step, or may conveniently be carried out in two stages by, first, forming the anion of of the pyridone by addition of an alkali earth salt

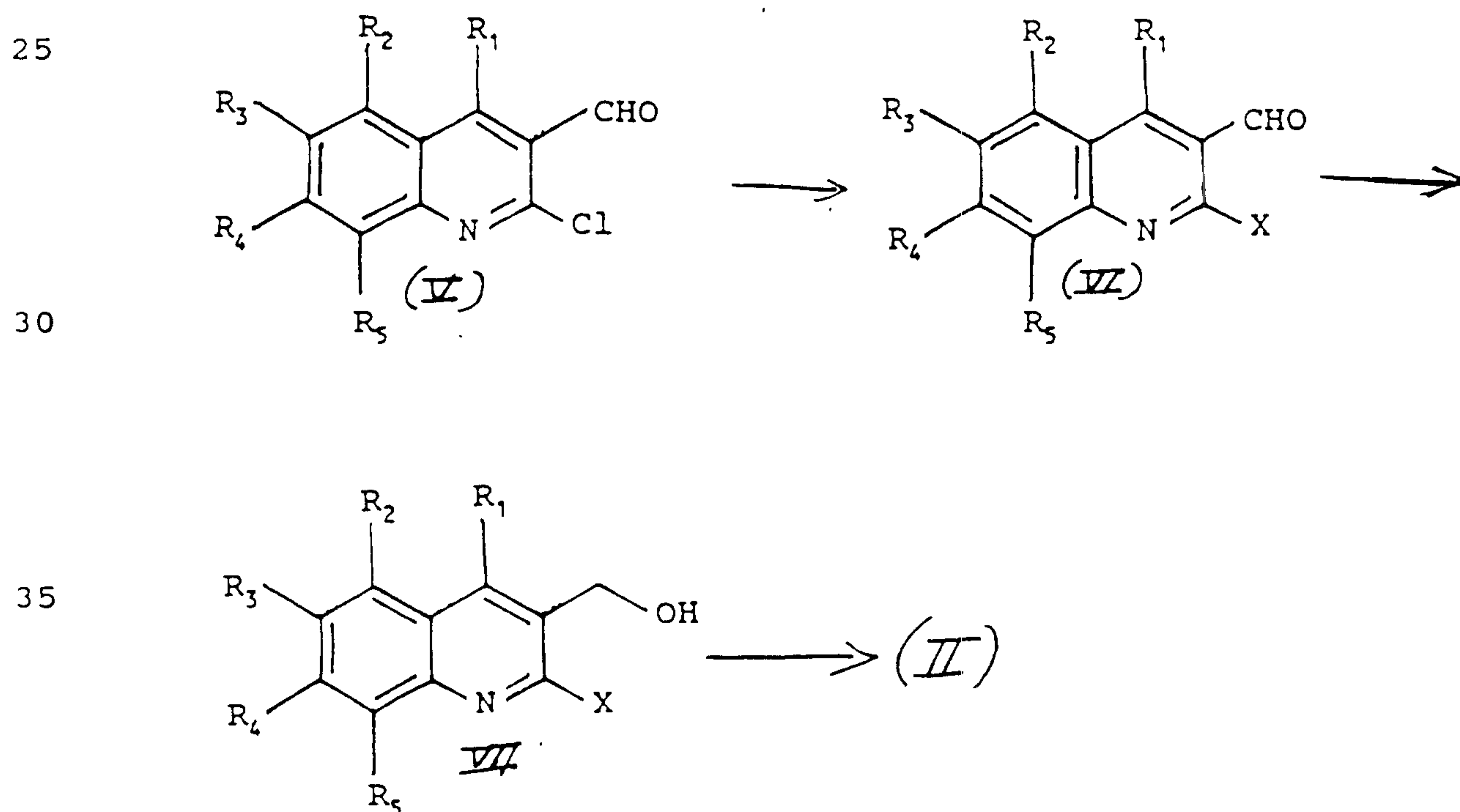
-6-

(e.g., potassium tert-butoxide) at about room temperature, and then adding the halomethylquinoline to the reaction solution and heating the solution between about 60° to about 100° Centigrade for 4-24 hours.

5 The compound of Formula IV may be cyclized to yield the compound of Formula I by an intramolecular Heck reaction. The reaction is carried out in the presence of a palladium catalyst (e.g., palladium acetate) under basic conditions in a polar aprotic solvent such as acetonitrile or dimethylformamide. A phase transfer catalyst such as a tetraalkylammonium halide salt is preferably included. The reaction should be carried out in an inert atmosphere, such as under argon. The reaction mixture may be heated to a temperature between about 50° to about 100° C for about 1 to 24 hours. Variations on these conditions will be apparent from the literature on the Heck reaction. See, e.g., R. Grigg et al. *Tetrahedron* 46, 4003-4008 (1990).

20 The compounds of Formula II may be prepared in accordance with Scheme B below, where R₁ through R₅ are as given in connection with Formula I above, and X is Bromo or Iodo, preferably Iodo.

Scheme B



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The starting materials in Scheme B, the compounds of Formula V, are made by known techniques, such as by chlorination of a quinoline. See, e.g., *Progress in Heterocyclic Chemistry* 2, 180 (H. Suschitzky and E. Scriven eds. 1990). In the alternative, compounds of Formula V may be made from the substituted acetanilide as described by O. Meth-Cohn et al., *J. Chem. Soc. Perkin Trans. I* 1981, 1520.

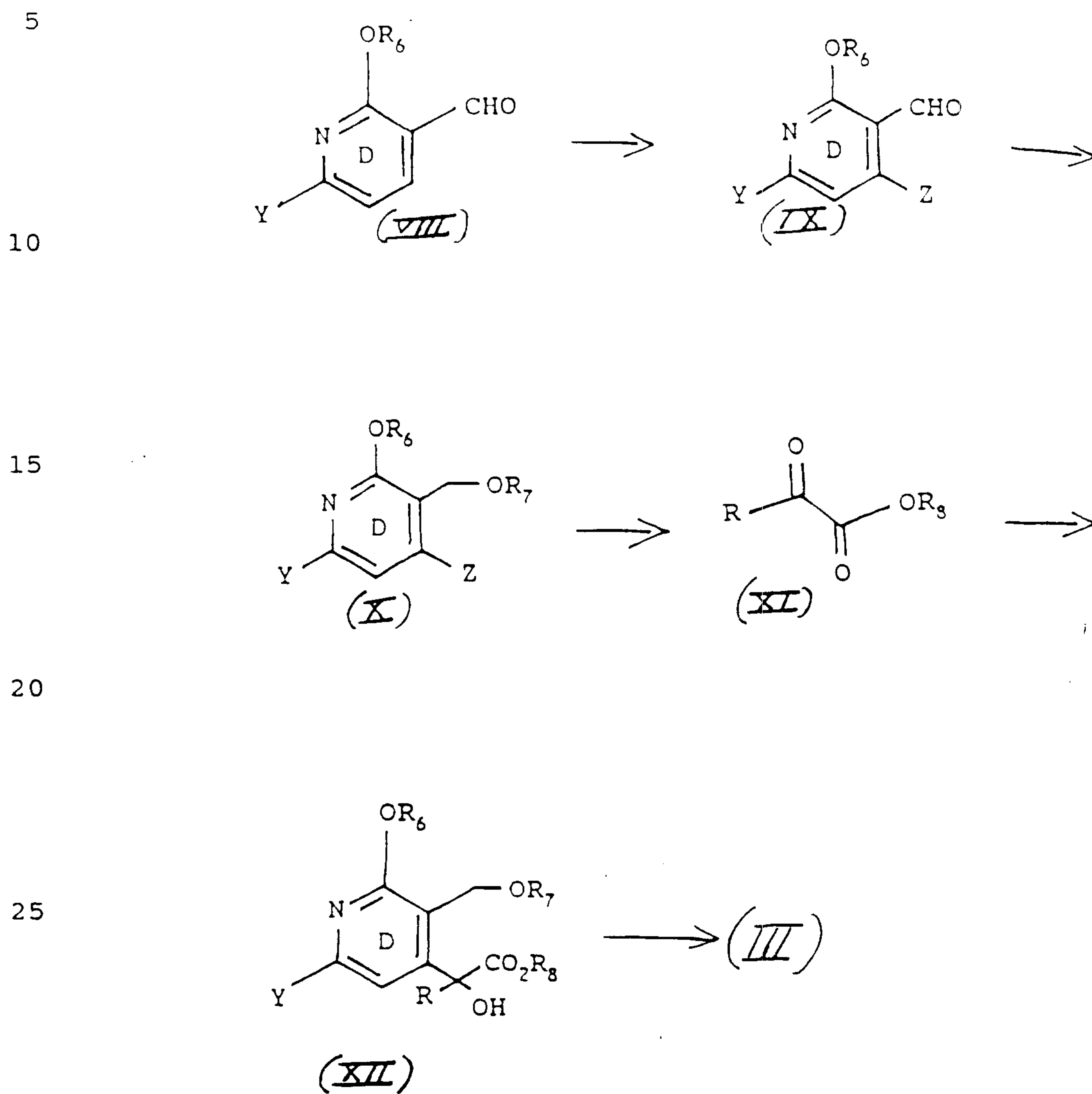
The halo group on the carboxaldehyde of Formula V is exchanged with an Iodo or Bromo (preferably Iodo) to produce the carboxaldehyde of Formula VI. The exchange reaction may be carried out in acetonitrile in the presence of a catalytic amount of a strong acid, such as HCl, by heating the reaction mixture to between about 70° to about 90° C for at least about 4 hours.

The carboxaldehyde of Formula VI is then reduced to produce the hydroxymethylquinoline of Formula VII. The reaction is carried out with a mild reducing agent to avoid reducing the quinoline ring, at a temperature of from about 0° to about 25° C, in an alcohol solvent. An alternative route for producing a compound of Formula VII is disclosed in N. Narasimham et al., *J. Chem. Soc., Chem. Commun.*, 1985, 1368-1369.

A compound of Formula II is produced from the hydroxymethylquinoline of Formula VII in accordance with conventional procedures in a solvent in which the reactants are soluble, such as dimethylformamide. The reaction is preferably carried out at lower temperatures to provide a higher yield.

The compounds of Formula III above are preferably prepared in accordance with Scheme C below, wherein R is as given in connection with Formula I above, R₆ and R₇ are loweralkyl, preferably methyl, R₈ is loweralkyl, preferably ethyl, Y is Cl or H, and Z is halo, preferably bromo or iodo.

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Scheme C

The starting materials for Scheme C, the compounds of Formula VIII, may be prepared in accordance with known techniques. For example, the synthesis of 2-methoxy-3-pyridinecarboxaldehyde is disclosed in D. Comins and M. Killpack, *J. Org. Chem.* **55**, 69-73 (1990).

In Scheme C, the carboxaldehyde of Formula

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VIII is halogenated to produce the 4-halo-3-pyridinecarboxaldehyde of Formula IX. Halogenation at the 4- position may be carried out by reacting the carboxaldehyde of Formula VIII with a lithiated
5 diamine, such as lithiated N,N,N'-trimethylethylenediamine, in dimethoxyethane or tetrahydrofuran to protect the aldehyde and direct subsequent C-4 lithiation, and by then lithiating the C-4 position of the pyridine with a suitable lithiating
10 reagent, such as n-butyllithium. See D. Comins and M. Killpack, *supra*. The C-4 lithiated pyridine intermediate is preferably halogenated by adding the intermediate to a solution of iodine or bromine in a polar or nonpolar organic solvent, preferably at a
15 temperature of at least as low as about -70°C.

The compound of Formula IX is reduced in an alcoholic acidic media in the presence of a trialkylsilane to yield the alkoxymethylpyridine of Formula X. The acid should be a strong acid, such as
20 sulfuric or trifluoroacetic acid. At least about 2 molar equivalents of a suitable alcohol (e.g., methanol, ethanol, *tert*-butanol) should be included to convert the aldehyde to the ether. Reference may be made to the literature on the silane reduction of
25 aldehydes for conditions and variations on this reaction. See, e.g., M. Doyle et al., *J. Am. Chem. Soc.* **94:10**, 3659-3661 (1972).

The compound of Formula X is lithiated at the C-4 position with a lithiating agent such as n-
30 butyllithium, and then reacted with a compound of Formula XI such as an alkyl α -ketobutyrate (e.g., methyl α -ketobutyrate, ethyl α -ketobutyrate, *tert*-butyl α -ketobutyrate) to produce the compound of Formula XII in essentially the manner described by R. Lyle et al.,
35 *J. Org. Chem.* **38**, 3268-3271 (1973). The reaction may be carried out in a tetrahydrofuran or ether solvent at a temperature of at least as low as about -50°C, with

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the alkyl α -ketobutyrate being added to the reaction solution as a single aliquot.

5 The compound of Formula XII is then cyclized to yield the compound of Formula III. Cyclization may be carried out by reacting the compound of Formula XII with bromo- or iodotrimethylsilane (preferably
10 iodotrimethylsilane) in a neutral or polar aprotic solvent such as acetonitrile, followed by reaction with a strong acid solution to cleave the ethers and yield the compound of Formula III (the ring forming spontaneously upon cleavage of the ethers). The bromo- or iodotrimethylsilane is preferably generated *in situ* in accordance with known techniques, such as by the reaction of chlorotrimethylsilane with a halogen salt
15 or elemental halogen. See A. Schmidt, *Aldrichimica Acta* 14, 31-38 (1981).

When Y is halo in the compound of Formula III, the compound may be hydrogenated by any suitable technique, preferably by catalytic hydrogenation in the
20 presence of a palladium catalyst in a hydrogen atmosphere under pressure (e.g., at least three atmospheres). See generally J. March, *Advanced Organic Chemistry*, 510-511 (3d. Ed. 1985).

As alternatives to Scheme C, a compound of
25 Formula III, where Y is H, may be prepared in the manner described in D. Comins, Ph.D. Thesis, University of New Hampshire, Durham, NH, at 25-29 (1977), and as described in Lyle et al., *J. Org. Chem.* 38, 3268-3271 (1973).

30 The discussion herein is, for simplicity, given without reference to stereoisomerism. However, the compounds of Formula I have an asymmetric carbon atom at the C-20 position. Thus, the present invention is concerned with the synthesis of both (i) racemic
35 mixtures of the compound of Formula I and (ii) enantiomeric forms of the compound of Formula I, particularly the 20-(S) form. The resolution of

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racemates into enantiomeric forms can be done in connection with the last step of the process, or in preceeding steps involving the synthesis of an intermediate having an asymmetric carbon atom, by known
5 procedures. For example, the racemate may be converted with an optically active reagent into a diasteriomeric pair, and the diasteriomeric pair subsequently separated into the enantiomeric forms.

Specific examples of compounds which may be
10 prepared by the method of the present invention include 9-methoxy-camptothecin, 9-hydroxy-camptothecin, 9-nitro-camptothecin, 9-amino-camptothecin, 10-hydroxy-camptothecin, 10-nitro-camptothecin, 10-amino-camptothecin, 10-chloro-camptothecin, 10-methyl-
15 camptothecin, 11-methoxy-camptothecin, 11-hydroxy-camptothecin, 11-nitro-camptothecin, 11-amino-camptothecin, 11-formyl-camptothecin, 11-cyano-camptothecin, 12-methoxy-camptothecin, 12-hydroxy-camptothecin, 12-nitro-camptothecin, 10,11-dihydroxy-camptothecin, 10,11-dimethoxy-camptothecin, 7-methyl-
20 10-fluoro-camptothecin, 7-methyl-10-chloro-camptothecin, 7-methyl-9,12-dimethoxy-camptothecin, 9,10,11-trimethoxy-camptothecin, 10,11-methylenedioxy-camptothecin and 9,10,11,12-tetramethyl-camptothecin.

25 Compounds of Formula I have antitumor and antileukemic activity. Additionally, compounds of Formula I wherein R₁ is halo are useful as intermediates for, among other things, making compounds of Formula I wherein R₁ is loweralkyl.

30 Those skilled in the art will appreciate that additional changes can be made in the compounds of Formula I (see, for examples, J. Cai and C. Hutchinson, supra), which changes will not adversely affect the new processes disclosed herein and do not depart from the
35 concept of the present invention.

In the Examples which follow, "mg" means milligrams, "M" means Molar, mL means milliliter(s),

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"mmol" means millimole(s), "Bu" means butyl, "THF" means tetrahydrofuran, "h" means hours, "min" means minutes, "C" means Centigrade, "p.s.i." means pounds per square inch, "DMF" means dimethylformamide, "TLC" means thin layer chromatography, and "PLC" means preparative thin layer chromatography.

EXAMPLE 1

6-Chloro-2-methoxy-3-pyridinecarboxaldehyde

To a solution of tert-butyllithium (1.7 M in pentane, 48.5 mL, 83.0 mmol) in 150 mL of THF at -78°C was added 6-chloro-2-methoxypyridine (8.94 mL, 75.0 mmol) over 5 min. The reaction mixture was stirred at -78°C for 1 h, then dimethylformamide (7.55 mL, 97 mmol) was added and the mixture was stirred at this temperature for 1.5 h. After the addition of glacial acetic acid (8.6 mL, 150 mmol), the reaction mixture was allowed to warm to room temperature over a 30-min period, then diluted with ether (200 mL). The organic phase was washed with saturated aqueous NaHCO₃ (100 mL) and brine (100 mL), and was dried over MgSO₄. Concentration afforded the crude product as a light yellow solid which was recrystallized from hexanes to give 9.6 g (75%) of 6-chloro-2-methoxy-3-pyridinecarboxaldehyde as a white solid: mp 80-81°C (mp 62-64°C) (See Daintier, R.S.; Suschitzky, H.; Wakefield, B.J. Tetrahedron Lett. 1984, 25, 5693.). ¹H NMR (300 MHz, CDCl₃) δ 10.31 (s, 1H), 8.07 (d, 1H, J = 9 Hz), 7.03 (d, 1H, J = 9 Hz), 4.09 (s, 3H); IR (nujol) 1685, 1580, 1565, 1270, 1140, 1090, 1005, 905, 820, 755 cm⁻¹.

30

EXAMPLE 2

6-Chloro-4-iodo-2-methoxy-3-pyridinecarboxaldehyde

To a solution of N,N,N'-trimethylethylenediamine (2.46 mL, 19.23 mmol) in 15 mL of 1,2-dimethoxyethane at -23°C was added n-BuLi (9.22 mL, 19.23 mmol), and the solution was stirred at -23°C for 20 min. The mixture

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was transferred using a double-tipped needle to a solution of 6-chloro-2-methoxy-3-pyridinecarboxaldehyde (3.0 g, 17.5 mmol) in 40 mL of 1,2-dimethoxyethane at -23°C. After stirring for 15 min, *n*-BuLi (12.6 mL, 26.2 mmol) was added and the dark mixture was stirred an additional 2 h at -23°C. The solution was transferred using a double-tipped needle to a solution of iodine (8.04 g, 31.7 mmol) in 40 mL of 1,2-dimethoxyethane at -78°C. After stirring at -78°C for 30 min, the cooling bath was removed and the reaction mixture was allowed to warm for 20 min, then quenched with water. The mixture was extracted with ether (2 x 30 mL) and the combined organic layers were washed successively with 30-mL portions of 10% aqueous Na₂S₂O₃, water and brine, and dried over MgSO₄. Concentration afforded 4.62 g (89%) of crude product to which was added 50 mL of hexanes. The mixture was stirred and allowed to stand at 5°C overnight. Filtration gave 2.67 g of 6-Chloro-4-iodo-2-methoxy-3-pyridinecarboxaldehyde as a yellow solid: mp 120-124°C. Concentration of the hexane washings and purification of the residue by radial preparative thin-layer chromatography (silica gel, 5% ethyl acetate/hexanes) gave an additional 1.41 g of product (mp 120-124°C), raising the total yield of the compound to 78%. Recrystallization from hexanes gave an analytical sample as a bright yellow solid: mp 129-130°C. ¹H NMR (300 MHz, CDCl₃) δ 10.16 (s, 1H), 7.59 (s, 1H), 4.07 (s, 1H); IR (nujol) 1690, 1350, 1260, 1095, 1010, 900, 840 cm⁻¹.

30

EXAMPLE 3**2-Chloro-4-iodo-6-methoxy-5-(methoxymethyl)pyridine**

To a mixture of 6-chloro-4-iodo-2-methoxy-3-pyridinecarboxaldehyde (1.07 g, 3.60 mmol), triethylsilane (0.86 mL, 5.40 mmol) and methanol (0.43 mL, 10.6 mmol) at 0°C was added trifluoroacetic acid (2.2 mL, 28.6 mmol), and the resulting solution was

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-14-

stirred at 25°C for 14 h. After dilution with ether (30 mL), saturated NaHCO₃ was added until the aqueous phase was rendered basic. The aqueous layer was extracted with ether (10 mL), and the combined ether layers were washed with water (10 mL) and brine (10 mL), and dried (Na₂SO₄). Concentration gave the crude product which was purified by radial PLC (silica gel, 5% ethyl acetate/hexanes) to afford 2-chloro-4-iodo-6-methoxy-5-(methoxymethyl)pyridine as a white solid (1.05 g, 93%): mp 69-72°C. Recrystallization from hexanes provided an analytical sample: mp 74-75°C. ¹H NMR (300 MHz, CDCl₃) δ 7.40 (s, 1H), 4.53 (s, 2H), 3.96 (s, 3H), 3.42 (s, 3H); IR (nujol) 1550, 1300, 1115, 1090, 1020, 940, 905, 830, 720 cm⁻¹.

15

EXAMPLE 4

Ethyl 2-Hydroxy-2-(6'-chloro-2'-methoxy-3'-methoxymethyl-4'-pyridyl)butyrate

To a solution of 2-chloro-4-iodo-6-methoxy-5-(methoxymethyl)pyridine (2.28 g, 7.30 mmol) in 50 mL of THF at -90°C was added n-BuLi (3.46 mL, 8.03 mmol) over 5 min and the resulting solution was stirred at -90°C for 30 min. Ethyl α-ketobutyrate (1.25 mL, 9.45 mmol) was added, the reaction mixture was stirred at -90°C for 30 min, then allowed to warm at ambient for 20 min, and quenched with saturated NH₄Cl. After removal of most of the solvent under reduced pressure, the residue was taken up in 40 mL of ether, washed with dilute NaHCO₃ (15 mL) and brine (15 mL), and was dried over MgSO₄. Evaporation of the solvent in vacuo and purification of the residue by radial PLC (10% acetone/hexanes) afforded ethyl 2-hydroxy-2-(6'-chloro-2'-methoxy-3'-methoxymethyl-4'-pyridyl)butyrate (1.53 g, 66%) as a light yellow, viscous oil. ¹H NMR (300 MHz, CDCl₃) δ 7.07 (s, 1H), 4.75 (d, 1H, J = 12 Hz), 4.47 (d, 1H, J = 12 Hz), 4.24 (q, 1H, J = 6 Hz), 4.17 (q, 1H, J = 6 Hz), 3.96 (s, 3H), 3.37 (s, 3H), 2.16 (m,

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2H), 1.24 (t, 3H, J = 6 Hz); IR (film) 3400, 1735, 1580, 1555, 1305, 1235, 1130, 1090, 1020, 905, 830, 730 cm^{-1} .

EXAMPLE 5

5 9-Chloro-7-oxopyrido[5,4-c]-2-oxo-3-ethyl-3-hydroxy-3,6-dihydropyran

To a stirred mixture of the hydroxy ester prepared in Example 4 above (1.53 g, 4.82 mmol) and sodium iodide (2.89 g, 19.3 mmol) in dry CH_3CN (35 mL) at 25°C was added dropwise chlorotrimethylsilane (2.45 mL, 19.3 mmol). The resulting solution was heated at reflux for 4 h, the solvent was removed under reduced pressure, and 100 mL of 6N HCl was added to the residue. After heating at a gentle reflux for 4 h, the mixture was stirred at 25°C overnight, then extracted with six 30-mL portions of CHCl_3 containing 5% CH_3OH . The combined organic extracts were washed with 40 mL of half-saturated NaCl containing $\text{Na}_2\text{S}_2\text{O}_3$, followed by 40 mL of saturated NaCl. After drying over Na_2SO_4 , the solvent was removed under reduced pressure and the residue was purified by radial PLC (silica gel, 5% $\text{CH}_3\text{OH}/\text{CHCl}_3$) to give 9-chloro-7-oxopyrido[5,4-c]-2-oxo-3-ethyl-3-hydroxy-3,6-dihydropyran (743 mg, 63%) as an off-white solid: mp 205-207°C. Recrystallization from $\text{CHCl}_3/\text{CH}_3\text{OH}$ gave an analytically pure sample as a white solid: mp 207-208°C. ^1H NMR (300 MHz, CDCl_3 DMSO- d_6) δ 6.79 (s, 1H), 5.49 (d, 1H, J = 15 Hz), 5.13 (d, 1H, J = 15 Hz), 1.78 (q, 2H, J = 6 Hz), 0.93 (t, 3H, J = 9 Hz), IR (nujol) 3450, 1740, 1640, 1600, 1560, 1320, 1225, 1140, 1035, 995, 940 cm^{-1} .

EXAMPLE 6

7-Oxopyrido[5,4-c]-2-oxo-3-ethyl-3-hydroxy-3,6-dihydropyran

35 A mixture of the chloropyridone prepared in Example 5 above (400 mg, 1.64 mmol) and sodium acetate

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(400 mg, 4.86 mmol) in 25 mL of ethanol was hydrogenated over 10% Pd/C (100 mg) at 42 psi for 4 h. The mixture was filtered through celite and the solids were washed with CH₃OH. The filtrate was concentrated and the residue was purified by radial PLC (silica gel, 5% CH₃OH/CHCl₃) to give the pure product (256 mg, 75%) as a white solid: mp 230-232°C (dec.). Recrystallization from CHCl₃/CH₃OH afforded an analytical sample: mp 232°C (dec.). ¹H NMR (300 MHz, CHCl₃/DMSO-d₆) δ 7.30 (d, 1H, J = 6 Hz), 6.49 (d, 1H, J = 6 Hz), 5.42 (d, 1H, J = 18 Hz), 5.12 (d, 1H, J = 18 Hz), 1.79 (m, 2H), 0.91 (t, 3H, J = 6 Hz); IR (nujol) 3300, 1750, 1640, 1620, 1555, 1065, 1030, 995, 805, cm⁻¹.

15

EXAMPLE 72-Chloro-3-quinolinecarboxaldehyde

To a solution of 0.46 mL (3.30 mmol) of diisopropylamine in 8 mL of THF at 0°C was added 1.53 mL (3.30 mmol) of n-BuLi dropwise. After 20 min the solution was cooled to -78°C and 2-chloroquinoline (491 mg, 3.0 mmol) was added neat. The mixture was stirred at -78°C for 30 min, then dimethylformamide (0.39 mL, 5.04 mmol) was added dropwise and the reaction mixture was stirred an additional 30 min at this temperature. After quenching at -78°C with glacial acetic acid (1 mL), the mixture was warmed to room temperature and diluted with ether (30 mL). The organic phase was washed with saturated NaHCO₃ solution (10 mL) and brine (10 mL), and was dried over MgSO₄. Concentration afforded 2-chloro-3-quinolinecarboxaldehyde (530 mg, 92%) as a light yellow solid (mp 145-149°C), which was used directly in the next step without further purification. Recrystallization from ethyl acetate afforded the pure compound as light yellow needles: mp 149-150°C (mp 148-149°C reported in Meth-Cohn, O.; Narhe, B.; Tarnowski, B. J. Chem. Soc. Perkin Trans. I

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1981, 1520.). ^1H NMR (300 MHz, CDCl_3) δ 10.57 (s, 1H), 8.77 (s, 1H), 8.08 (d, 1H, $J = 9$ Hz), 8.0 (d, 1H, $J = 9$ Hz), 7.90 (t, 1H, $J = 9$ Hz), 7.67 (t, 1H, $J = 9$ Hz); IR (nujol) 1685, 1575, 1045, 760, 745 cm^{-1} .

5

EXAMPLE 8

Preparation of 2-Chloro-3-quinoline-
carboxaldehyde from acetanilide

Following a literature procedure (see Meth-Cohn, O.; Narhe, B.; Tarnowski, B. J. Chem. Soc. Perkin Trans. I 1981, 1520), phosphorus oxychloride (24.0 mL, 260 mmol) was added dropwise to an ice-cold solution of dimethylformamide (7.20 mL, 93.0 mmol) and the deep-red solution was stirred at 0°C for 30 min. Acetanilide (5.0 g, 37.0 mmol) was added neat and the mixture was stirred at 0°C for 30 min, then heated at 75°C for 16 h. The cooled mixture was poured into 250 mL of ice-water and stirred at 0-5°C for 30 min. The product was filtered, washed with water, and recrystallized from ethyl acetate to give 5.2 g (74%) of 2-Chloro-3-quinoline-carboxaldehyde as a light yellow solid: mp 147-149°C.

EXAMPLE 9

2-Iodo-3-quinolinecarboxaldehyde

A mixture of the aldehyde prepared in accordance with Example 7 or 8 above (5.0 g, 26.2 mmol), sodium iodide (10.0 g, 66.7 mmol) and concentrated HCl (1 mL) in 100 mL of CH_3CN was heated at reflux for 4.5 h. After removal of most of the solvent in vacuo, aqueous Na_2CO_3 was added until the mixture was basic, and the product was filtered and washed with water. The crude product was recrystallized from 95% ethanol to give 6.51 g (88%) of 2-iodo-3-quinolinecarboxaldehyde as off-white fluffy needles: mp 156-157°C (mp 150-152°C reported in Meth-Cohn, O.; Narhe, B.; Tarnowski, B.; Hayes, R.; Keyzad, A.;

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Rhavati, S.; Robinson, A. J. Chem. Soc. Perkin Trans. I 1981, 2509). ¹H NMR (300 MHz, CDCl₃) δ 10.29 (s, 1H), 8.57 (s, 1H), 8.12 (d, 1H, J = 9 Hz), 7.98 (d, 1H, J = 9 Hz) 7.88 (t, 1H, J = 9 Hz), 7.68 (t, 1H, J = 9 Hz); IR (nujol) 1680, 1610, 1570, 1555, 1315, 1020, 1005, 750, 740 cm⁻¹.

EXAMPLE 10

3-Hydroxymethyl-2-iodoquinoline

To a stirred solution of 2-iodo-3-quinolinecarboxaldehyde (595 mg, 2.10 mmol) in 40 mL of CH₃OH at 0°C was added NaBH₄ (86 mg, 2.31 mmol), and the mixture was stirred at 0°C for 30 min. After concentrating the mixture to approximately one-half of its original volume, water (30 mL) was added and the mixture was allowed to stand at 5°C overnight. The solids were filtered and the crude product (570 mg, 95%) was recrystallized from methanol to give 3-hydroxymethyl-2-iodoquinoline (505 mg, 84%) as colorless needles: mp 189-190°C. ¹H NMR (300 MHz, CDCl₃) δ 8.19 (s, 1H), 7.99 (d, 1H, J = 9 Hz), 7.87 (d, 1H, J = 9 Hz), 7.68 (m, 1H), 7.58 (t, 1H, J = 9 Hz), 5.45 (t, 1H, J = 6 Hz), 4.66 (d, 2H, J = 6 Hz); IR (nujol) 3350, 1580, 1320, 1125, 1060, 995, 755, 720, cm⁻¹.

25

EXAMPLE 11

3-Chloromethyl-2-iodoquinoline

To a stirred mixture of 3-hydroxymethyl-2-iodoquinoline prepared in accordance with Example 10 above (350 mg, 1.23 mmol) and triphenylphosphine (483 mg, 1.84 mmol) in 10 mL of dry DMF at -23°C was added N-chlorosuccinimide (246 mg, 1.84 mmol), and the mixture was stirred for 1 h at -23°C. After the addition of 40 mL of dilute aqueous NaHCO₃, the mixture was extracted with ethyl acetate (20 mL) and then ether (2 x 15 mL). The combined organic extracts were washed

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successively with 20-mL portions of dilute NaHCO_3 , water and brine, and were dried over MgSO_4 . Concentration and purification of the residue by radial PLC (silica gel, 10% ethyl acetate/hexanes) afforded 312 mg (84%) of 3-chloromethyl-2-iodoquinoline as a white crystalline solid: mp 138-140°C. Recrystallization from hexanes afforded an analytical sample as colorless needles: mp 139-140°C. ^1H NMR (300 MHz, CDCl_3) δ 8.17 (s, 1H), 8.07 (d, 1H, $J = 9$ Hz), 7.84 (d, 1H, $J = 9$ Hz), 7.75 (t, 1H, $J = 9$ Hz), 7.62 (t, 1H, $J = 9$ Hz), 4.80 (s, 1H); IR (nujol) 1585, 1555, 1260, 1010, 780, 755, 710 cm^{-1} .

EXAMPLE 12

8-(2'-Iodo-3'-quinolylmethyl)-7-oxopyrido[5,4-c]-2-oxo-3-ethyl-3-hydroxy-3,6-dihydropyran

To a solution containing 45 mg (0.40 mmol) of potassium tert-butoxide in 4 mL of dry isopropyl alcohol at 25°C was added 55 mg (0.26 mmol) of 7-oxopyrido[5,4-c]-2-oxo-3-ethyl-3-hydroxy-3,6-dihydropyran prepared in accordance with Example 6 above and the mixture was stirred at 25°C for 30 min. A solution of 3-chloromethyl-2-iodoquinoline prepared in accordance with Example 11 above (104 mg, 0.35 mmol) in 1 mL of CH_3OH was added dropwise to the white suspension, and the resulting solution was heated at 75°C for 24 h. After quenching the reaction mixture with saturated NH_4Cl , the solvents were removed under reduced pressure, and the residue was taken up in CH_2Cl_2 (20 mL) and washed with brine (2 x 10 mL). Concentration and purification of the residue by radial PLC (2% $\text{CH}_3\text{OH}/\text{CHCl}_3$) gave the product (99 mg, 80%) as a white solid: mp 171-174°C (dec.). Recrystallization from ethyl acetate/hexanes afforded an analytical sample: mp 174°C (dec.). ^1H NMR (300 MHz, CDCl_3) δ 8.05 (d, 1H, $J = 9$ Hz), 7.70-7.80 (m, 3H), 7.55-7.61 (m, 2H), 6.61 (d, 1H, $J = 9$ Hz), 5.63 (d, 1H, $J = 15$ Hz), 5.43 (d, 1H, $J = 15$ Hz), 5.27 (d, 1H, $J = 9$ Hz),

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5.22 (d, 1H, J = 9 Hz); IR (nujol) 3350, 1750, 1650, 1590, 1565, 1160, 1140, 1000, 750 cm⁻¹.

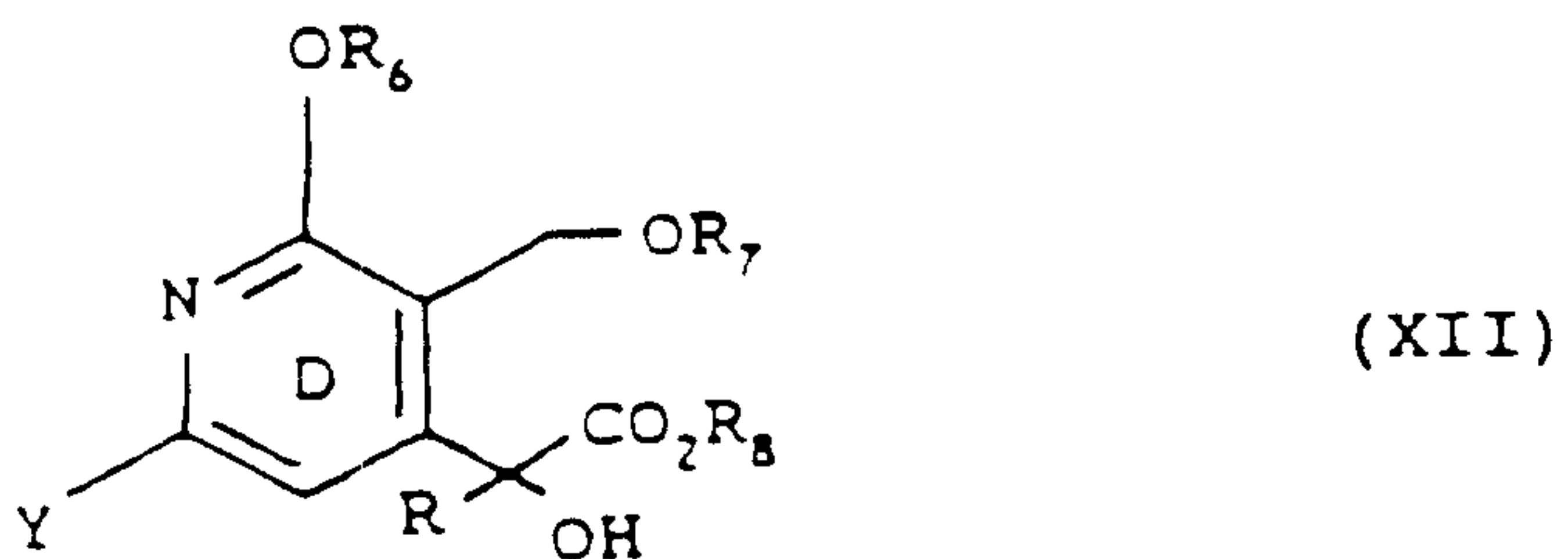
EXAMPLE 13(±)-Camptothecin

5 A mixture of 8-(2'-iodo-3'-quinolylmethyl)-7-oxopyrido[5,4-c]-2-oxo-3-ethyl-3-hydroxy-3,6-dihydropyran prepared in accordance with Example 12 above (76 mg, 0.16 mmol), K₂CO₃ (44 mg, 0.32 mmol), tetrabutylammonium bromide (52 mg, 0.16 mmol) and
10 Pd(OAc)₂ (3.6 mg, 0.016 mmol) in 15 mL of dry acetonitrile under argon was heated at 90°C for 5 h. TLC analysis of the reaction mixture showed a single spot which was highly U.V. active. The mixture was cooled, concentrated, and the residue was taken up in
15 30 mL of CHCl₃ containing 10% CH₃OH. This was washed with two 10-mL portions of saturated aqueous NH₄Cl. The organic layer was dried over Na₂SO₄ and concentrated. The dark residue was subjected to radial PLC (silica gel, 4% CH₃OH/CHCl₃), to give 17 mg of an orange solid
20 which was shown by NMR analysis to be a mixture of impure (±)-camptothecin and tetrabutylammonium bromide. The aqueous washings were filtered to give a yellow solid which was purified by radial PLC (silica gel, 4% CH₃OH/CHCl₃) to afford (±)-camptothecin (26 mg, 47%) as
25 a yellow solid: mp 275-277°C (mp 275-277°C reported in Volman, R.; Danishefsky, S.; Eggler, J.; Soloman, D.M. J. Am. Chem. Soc. 1971, 93, 4074.).

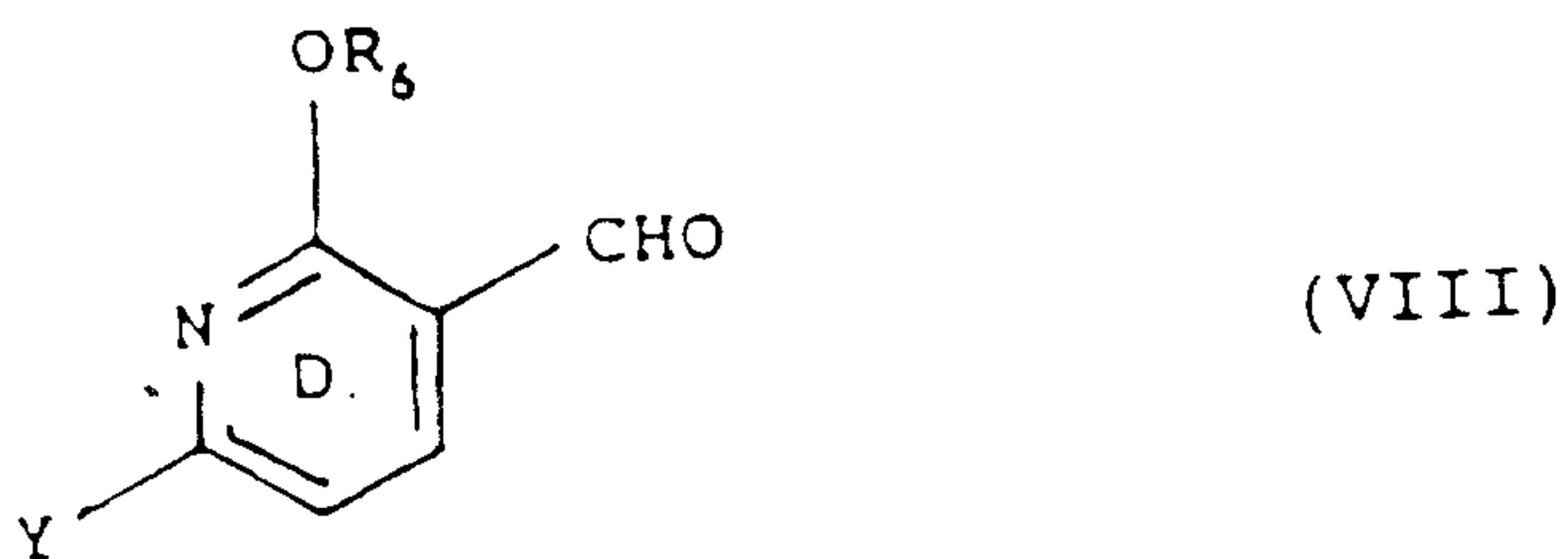
30 The foregoing examples are illustrative of the present invention, and are not to be construed as limiting thereof. The invention is defined by the following claims, with equivalents of the claims to be included therein.

THE EMBODIMENTS OF THE INVENTION IN WHICH AN EXCLUSIVE PROPERTY OR PRIVILEGE IS CLAIMED ARE DEFINED AS FOLLOWS:

1. A method of making a compound of Formula XII:

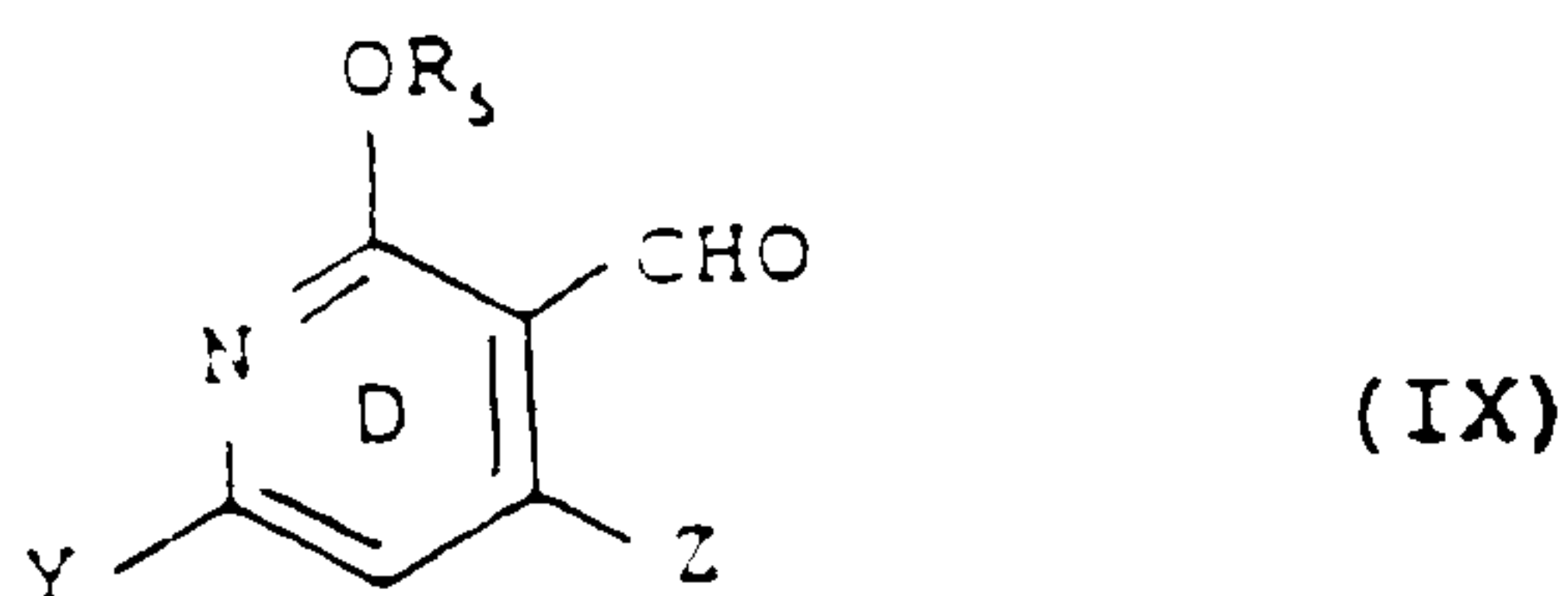


wherein Y is H or Cl, and R, R₆, R₇, and R₈ are loweralkyl, from a carboxaldehyde of Formula VIII:

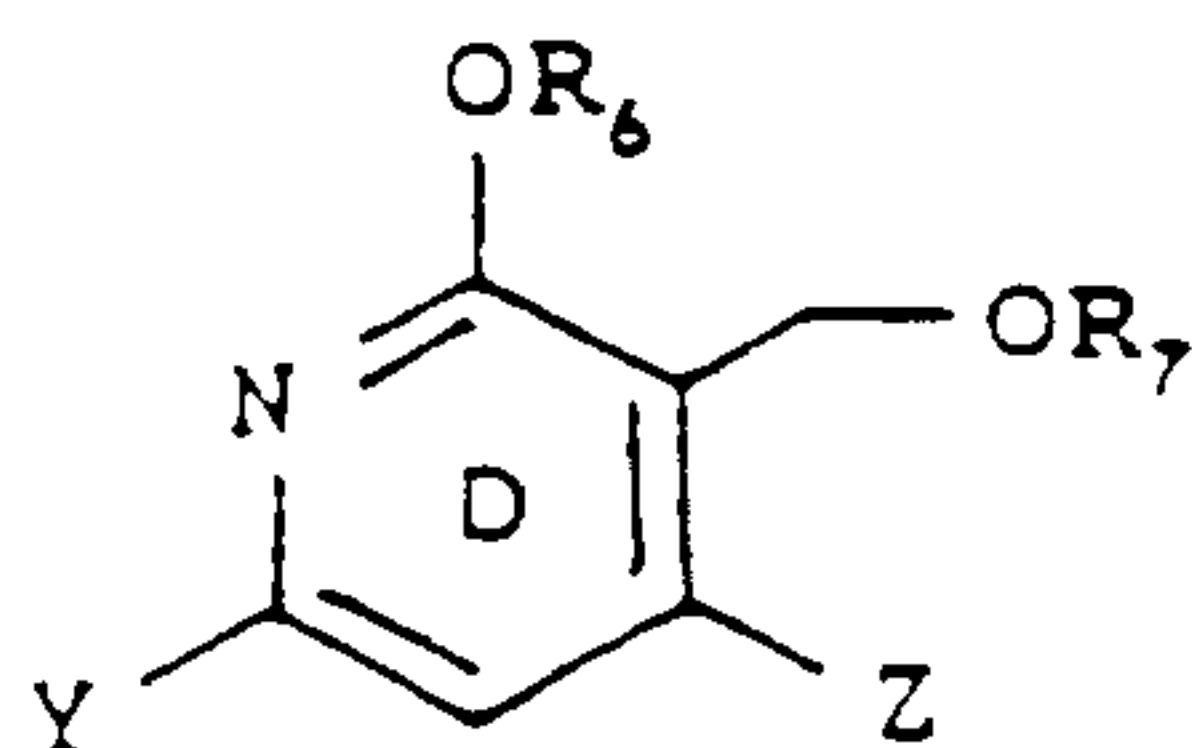


comprising

halogenating the carboxaldehyde of Formula VIII to produce a 4-halo-pyridinecarboxaldehyde of Formula IX:

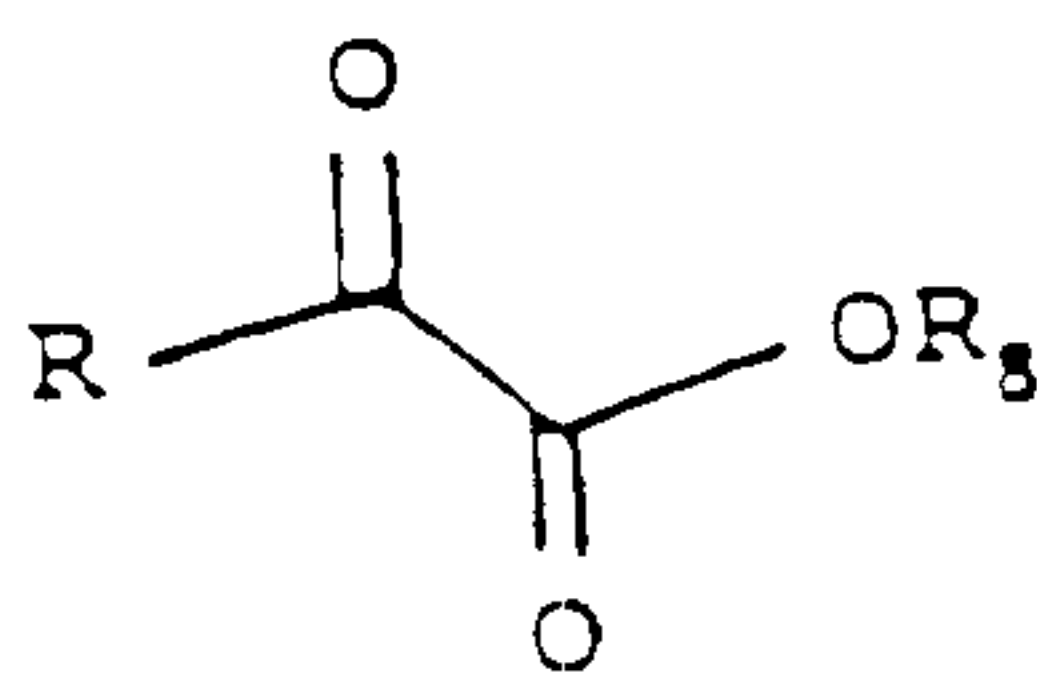


wherein Z is selected from the group consisting of bromo and iodo; and then
 reducing the 4-halo-pyridinecarboxaldehyde of Formula IX to give an
 alkoxymethylpyridine of Formula X:



(X)

wherein R₇ is loweralkyl; and then
 alkylating the alkoxymethylpyridine of Formula X with a compound of
 Formula XI:



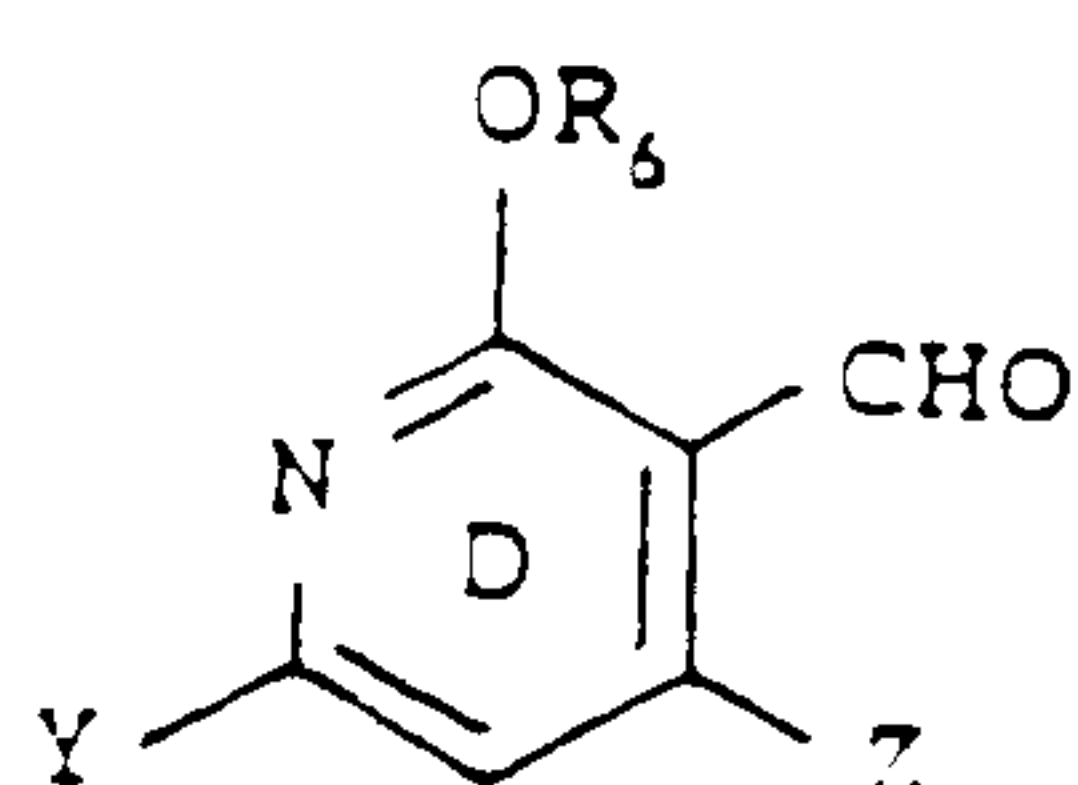
(XI)

to yield the compound of Formula XII.

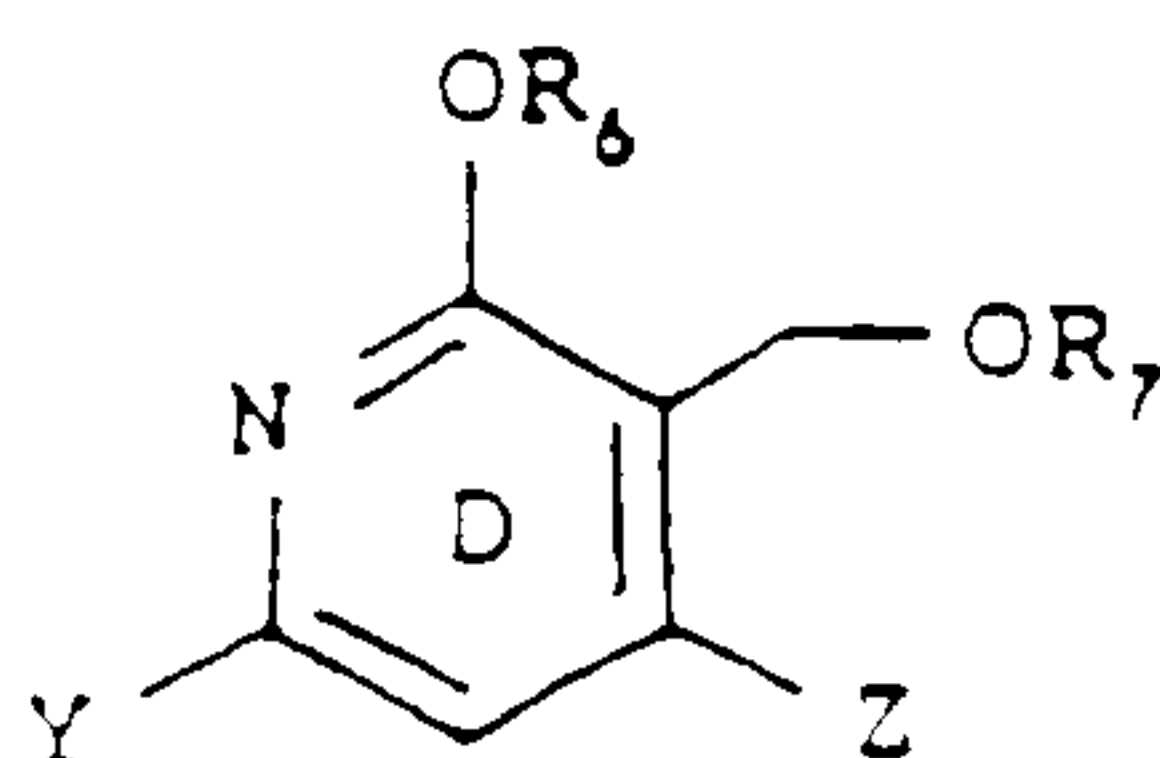
2. A method according to claim 1, wherein said step of halogenating the
 carboxyaldehyde of Formula VIII to produce a 4-halo-pyridinecarboxaldehyde of
 Formula IX is carried out by

- (a) reacting the carboxaldehyde of Formula VIII with a lithiated diamine
 to direct subsequent C-4 lithiation of the compound of Formula VIII, then
- (b) lithiating the C-4 position of the compound of Formula VIII, and then
- (c) halogenating the C-4 position of the compound of Formula VIII to
 yield the compound of Formula IX.

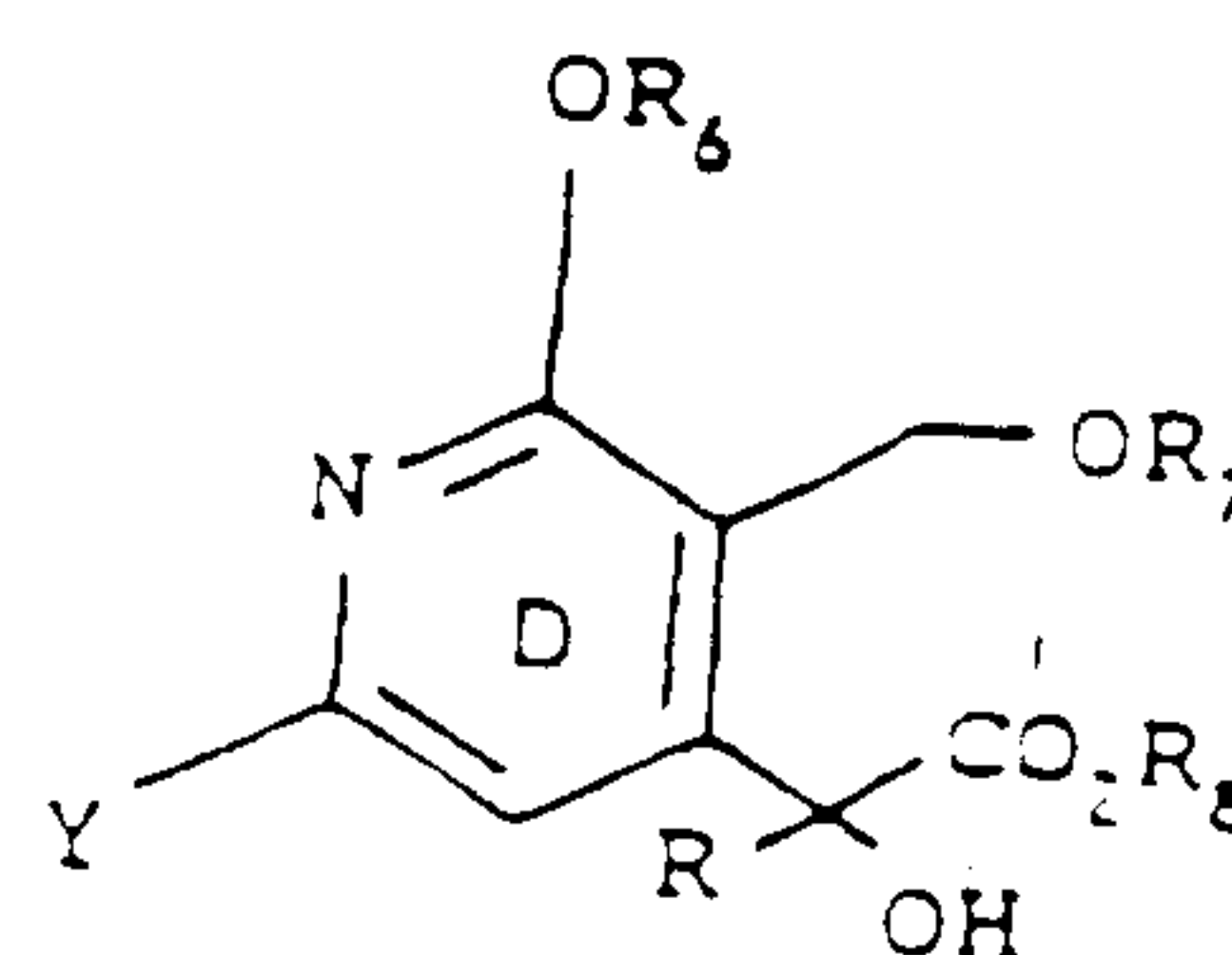
3. A method according to claim 2, wherein said step (c) is carried out in a polar or nonpolar organic solvent at a temperature at least as low as about -70°C.
4. A method according to claim 1, wherein said step of reducing the 4-halopyridinecarboxaldehyde of Formula IX is carried out in an alcoholic acidic medium in the presence of a trialkylsilane to give the alkoxymethylpyridine of Formula X.
5. A method according to claim 4, wherein said alcoholic acidic medium comprises an acid selected from the group consisting of sulfuric acid and trifluoroacetic acid.
6. A method according to claim 1, wherein said step of alkylating the alkoxymethylpyridine of Formula X with a compound of Formula XI is carried out by:
 - (a) lithiating the compound of Formula X at the C-4 position; and then
 - (b) reacting the compound of Formula X with the compound of Formula XI to yield the compound of Formula XII.
7. A compound selected from the group consisting of compounds of Formula IX, Formula X, and Formula XII:



(IX)



(X)



(XII)

wherein in each of the above formulae:

R, R₆, R₇, and R₈ are loweralkyl;

Y is Cl or H; and

Z is halogen.

8. A compound according to claim 7, wherein Z is bromo or iodo.
9. A compound according to claim 7, wherein the compound has Formula X, wherein R_6 and R_7 are each independently selected from the group consisting of methyl, ethyl, propyl and butyl.
10. A compound according to claim 9 wherein Z is iodo.
11. A compound according to claim 9 or 10 wherein R_6 and R_7 are methyl.
12. A compound according to claim 7, wherein the compound has Formula IX, wherein R_6 is selected from the group consisting of methyl, ethyl, propyl and butyl.
13. A compound according to claim 12 wherein R_6 is methyl.
14. A compound according to claim 12 or 13 wherein Z is iodo.
15. A compound according to any one of claims 7, 9 or 14 wherein Y is Cl.

