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(71) Applicant (for all designated States except US): **3V SIG-MA S.P.A.** [IT/IT]; Via Fatebenefratelli 20, I-20121 Milano MI (IT).

(72) Inventors; and

(75) Inventors/Applicants (for US only): **MAESTRI, Francesco** [IT/IT]; Via Autostrada 2, I-24126 Bergamo BG (IT). **DEL SORDO, Simone** [IT/IT]; Piazza Carlo Alberto Dalla Chiesa 22, I-24047 Treviglio BG (IT). **BERTE', Ferruccio** [IT/IT]; Via F. Nullo 3, I-24128 Bergamo BG (IT).

(74) Agents: **ADORNO, Silvano** et al.; Società Italiana Brevetti S.p.A., Via Carducci 8, I-20123 Milano MI (IT).

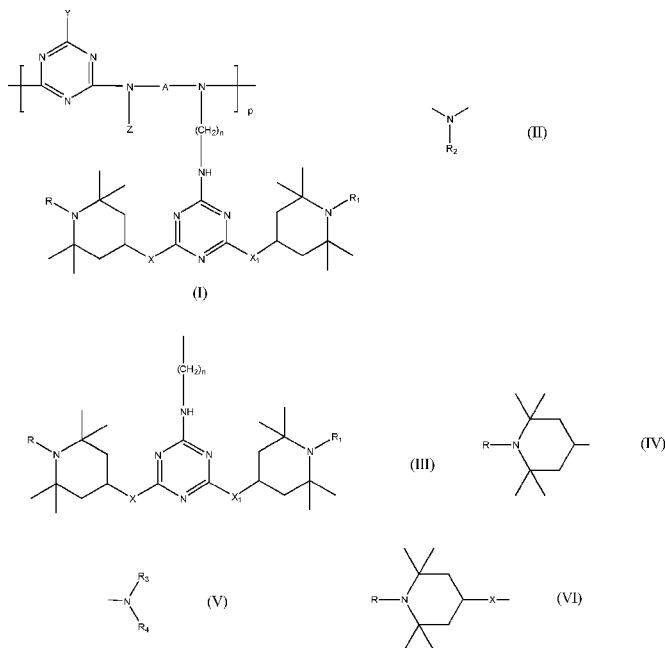
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(54) Title: NEW STERICALLY HINDERED POLYMERIC AMINES AND THEIR USE AS POLYMER STABILIZERS



(57) Abstract: The present invention relates to polypiperidine compounds of polymeric type which are capable of conferring to the polymeric materials, particularly to polyolefins, a high stability against photodegradation and oxidative action of air, belong to the HALS category and have general formula (I): wherein p is from 3 to 20; n is from 2 to 12; R and R₁, which are the same or different, are selected in the group consisting of hydrogen, linear and branched C₁-C₁₂ alkyl groups, alkenyl groups having 3 to 8 carbon atoms and aralkyl groups having 7 to 19 carbon atoms; X and X₁, which are the same or different, are selected in the group consisting of oxygen and groups of formula (II) wherein R₂ is selected in the group consisting of hydrogen, linear and branched C₁-C₁₂ alkyl groups, cycloalkyl groups having 5 to 12 carbon atoms and aralkyl groups having 7 to 12 carbon atoms; A represents a $-(CH_2)_a-$ group wherein a is from 2 to 12, with the proviso that a is different from n; Z is selected in the group consisting of C₁-C₁₈ alkyl groups, groups of formula (III) wherein n, X, X₁, R and R₁ are as above defined, and groups of formula (IV) wherein R is as above defined; Y represents a substituent selected in the group consisting of the groups of general formula (V) the groups O-R₄ and S-R₄, wherein R₃ and R₄, which may be the same or different, represent hydrogen, a linear and branched C₁-C₁₈ alkyl group, a cycloalkyl group having 5 to 12 carbon

atoms, an aralkyl group having 7 to 12 carbon atoms, an aryl group having 6 to 12 carbon atoms or may form, together with the nitrogen atom to which they are linked, a heterocycle containing 5 to 7 carbon atoms; and the piperidine group (VI) wherein R and X are as above defined. The invention further relates to the processes for preparation of the compounds according to the invention.

NEW STERICALLY HINDERED POLYMERIC AMINES AND THEIR USE AS POLYMER STABILIZERS

The present invention relates to novel polypiperidine compounds, of polymeric
5 type, which are able to impart to synthetic polymeric materials, in particular to
polyolefins, a high stability towards photodegradation and air oxidation. The invention
further relates to processes for the preparation of the compounds according to the
invention.

It is known that synthetic polymeric materials are subject to deterioration due to
10 the action of heat, sun light and air, which can cause degradation, loss of mechanical
properties, discoloring and other undesired effects.

Various classes of compounds have been proposed for the stabilization of
polymeric materials, principally against UV radiation of the solar light, some of which
had a wide range of applicability, such as for example benzophenones and
15 benzotriazoles. These compounds confer to the polymers an acceptable stability, which
is however not yet sufficient for the practical needs with reference to the fibers, films
and raffia made of olefinic polymers.

Polyalkylpiperidine derivatives, normally denominated HALS (Hindered Amine
Light Stabilizers) are very effective for the stabilization of polymeric materials. A
20 number of patents describe their preparation, use and the obtained results, such as for
example US 4,477,615, US 4,331,586, US 4,086,204.

However, not always do said known stabilizers give satisfying results and the
polymeric materials remain subject to undesired deterioration due to heat, light and air,
with the above mentioned negative consequences. One of the reasons for this
25 deterioration is diffusion and migration of the polypiperidine stabilizers inside the
polymeric material whose stability towards oxidative action and photodegradation has
to be improved. A reduced migration of the polypiperidine derivatives may therefore be
a factor for increasing stability of synthetic polymeric material towards oxidative action
and photodegradation caused by heat, light and air. A further drawback of the stabilizer
30 compounds according to the prior art is that their synthesis requires the use of
industrially expensive working conditions.

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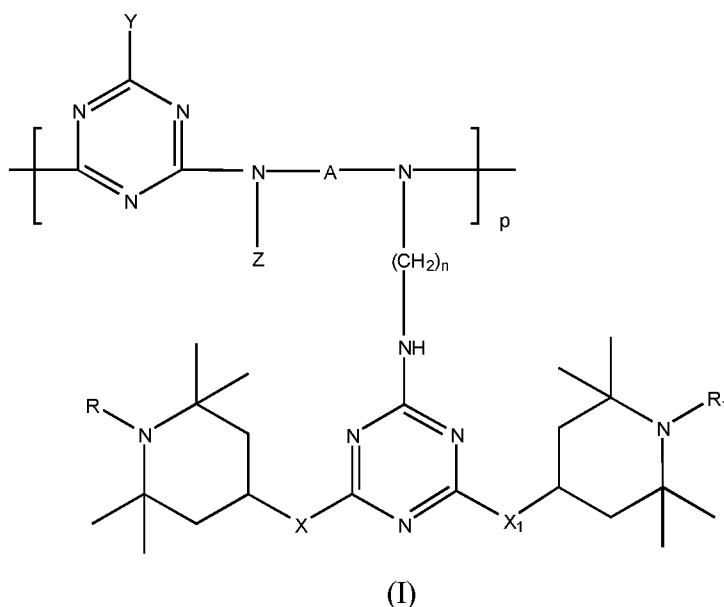
Object of the present invention is providing new high molecular weight polypiperidine compounds of polymeric type which are able to confer to the synthetic polymeric materials the above mentioned improved stability features. Said object is obtained with the polymeric polypiperidine compounds whose main features are specified in claim 1, and with a process for the fabrication thereof, whose main features are specified in claim 8. Other features of the invention are specified in the remaining claims.

Further advantages and features of the compounds according to the present invention will be clear to the skilled persons from the following detailed and non-limiting description of some embodiments thereof, with reference to the attached drawings wherein:

- figure 1 shows a chromatogram referring to a high molecular weight polymeric polypiperidine compound according to the invention, obtained as described in the following example 1;
- figure 2 shows a chromatogram referring to a high molecular weight polymeric polypiperidine compound according to the invention, obtained as described in the following example 2;
- figure 3 shows a chromatogram referring to an oligomeric piperidine compound, obtained as described in the following comparative example 3; and
- figure 4 shows a chromatogram referring to an oligomeric piperidine compound, obtained as described in the following comparative example 4.

According to the present invention, it is possible to confer to the synthetic polymeric materials a particularly high stability towards photodegradation and air oxidative action by using polymeric HALS, which are able to guarantee a reduced migration inside the polymeric materials to be stabilized, having the following general formula (I):

- 3 -

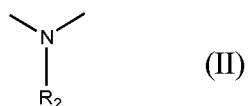


wherein **p** is from 3 to 20;

n is from 2 to 12;

R and **R**₁, which are the same or different, are selected in the group consisting of hydrogen, linear and branched C₁-C₁₂ alkyl groups, alkenyl groups having 3 to 8 carbon atoms and aralkyl groups having 7 to 19 carbon atoms;

X and **X**₁, which are the same or different, are selected in the group consisting of oxygen and groups of formula (II)



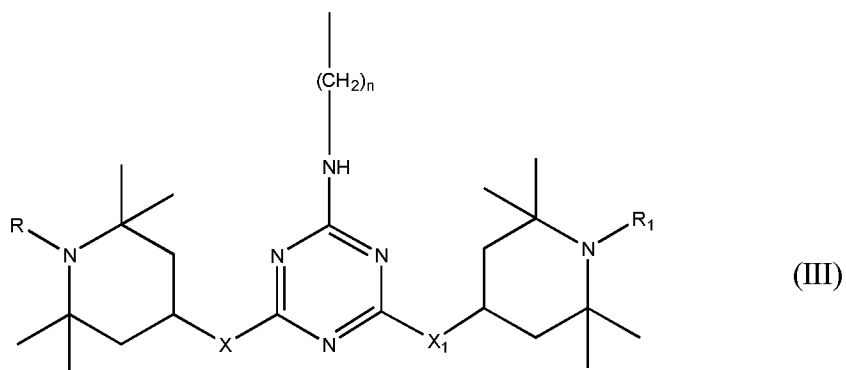
wherein **R**₂ is selected in the group consisting of hydrogen, linear and branched C₁-C₁₂ alkyl groups, cycloalkyl groups having 5 to 12 carbon atoms and aralkyl groups having 7 to 12 carbon atoms;

A represents a $-(CH_2)_a-$ group wherein **a** is from 2 to 12, with the proviso that **a** is different from **n**;

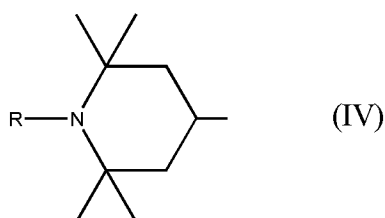
Z is selected in the group consisting of C₁-C₁₈ alkyl groups, groups of formula

(III)

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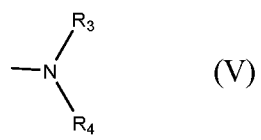


wherein **n**, **X**, **X₁**, **R** and **R₁** are as above defined,
and groups of formula (IV)



5 wherein **R** is as above defined;

Y represents a substituent selected in the group consisting of the groups of general formula (V)

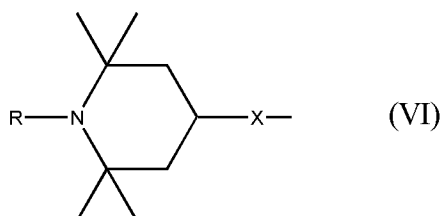


the groups **O-R₄** and **S-R₄**,

10 wherein **R₃** and **R₄**, which may be the same or different, are selected in the
group consisting of hydrogen, linear and branched C₁-C₁₈ alkyl groups,
cycloalkyl groups having 5 to 12 carbon atoms, aralkyl groups having 7 to
12 carbon atoms, aryl groups having 6 to 12 carbon atoms, or may form,
together with the nitrogen atom to which they are linked, a heterocycle
15 containing 5 to 7 carbon atoms;

and the piperidine group (VI)

- 5 -



wherein **R** and **X** are as above defined.

It has been found and it is claimed by the present inventors that the polypiperidine derivatives of formula (I), wherein p is included between 3 and 20, are capable of conferring an improved stability to the synthetic polymeric materials towards oxidative action and photodegradation. As a matter of fact, the reduced diffusion inside the synthetic polymeric material of the novel polymeric HALS of formula (I) (according to the REACH legislation) guarantees a lower degradation due to exposure to sun light and heat. Further, the process according to the present invention for the preparation thereof is simple and industrially not very expensive.

Preferably, the polypiperidine derivatives according to the present invention belong to the general formula (I) wherein p is from 3 to 8, and more preferably from 3 to 6.

It has been noted that the polypiperidine derivatives of polymeric type according to formula (I), in particular with $n=3$ and $A=-(CH_2)_a-$ with $a=2$ confer to the polymeric materials a better stability against photodegradation and the oxidative action of air.

Further, it has been found that the stability is also improved when the group Z corresponds to the group of formula (III), where R and R_1 are the same and particularly hydrogen or methyl and when X and X_1 , equal to each other, are groups of formula (II) wherein particularly R_2 is a butyl group.

Further improvements in the stability conferred to the polymeric materials are observed when Y is the group of formula (VI) or a morpholine group.

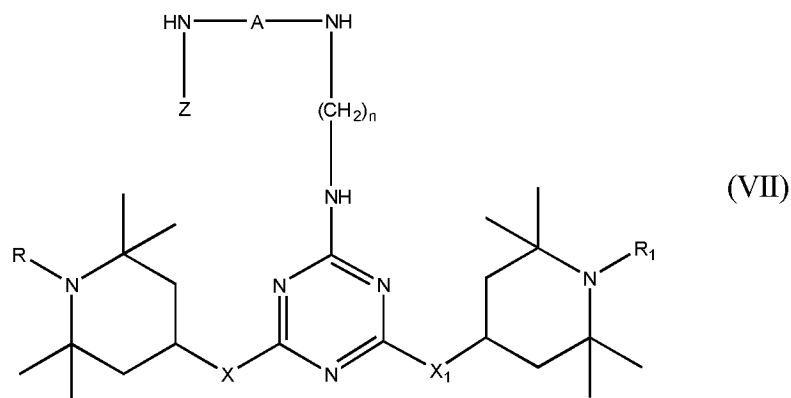
The polypiperidine compounds of formula (I) wherein Y is a morpholine group advantageously show a particularly good performance in the stabilization of polyamides (PA) and of the acrylonitrile-butadiene-styrene copolymers (ABS).

The present invention also relates to the processes for the preparation of polymeric polypiperidine compounds of formula (I).

Obtaining the polymeric polypiperidine compounds of formula (I) is possible,

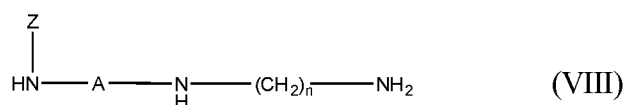
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according to the present invention, through a polycondensation reaction between an intermediate of general formula (VII)



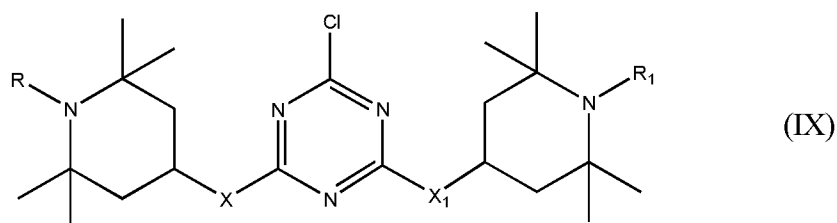
wherein Z, A, R, R₁, X, X₁ and n are as above defined,
and cyanuric chloride.

When Z is a C₁-C₁₈ alkyl group, the group of formula (VII) can be obtained through a process consisting in reacting an amine having the following general formula (VIII)



wherein Z, A and n are as above defined,

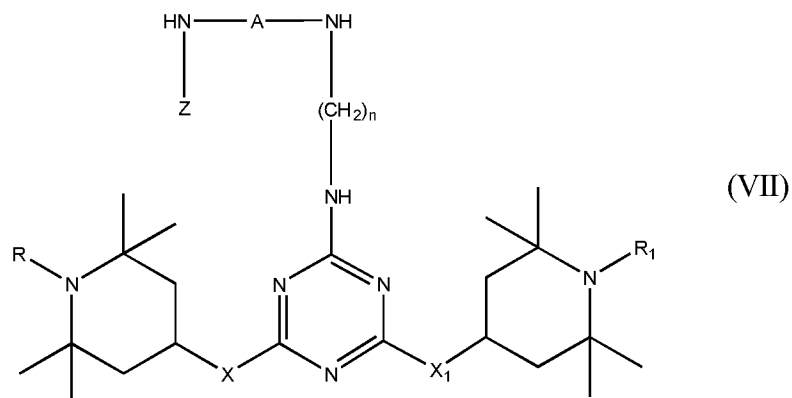
with an equivalent of the derivative having the following general formula (IX) whose preparation is well known in the literature (for example, Example 3 of WO2007/057265A2)



wherein R, R₁, X and X₁ are as above defined.

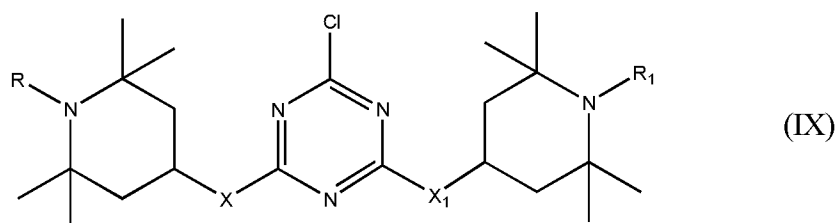
Thus, a derivative is obtained that can be represented through the following simplified formula (VII):

- 7 -



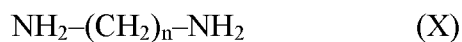
wherein $Z = C_1-C_{18}$ and A, R, R_1 , X, X_1 and n are as above defined.

Alternatively, the derivative of general formula (IX)



5 wherein R, R_1 , X and X_1 are as above defined

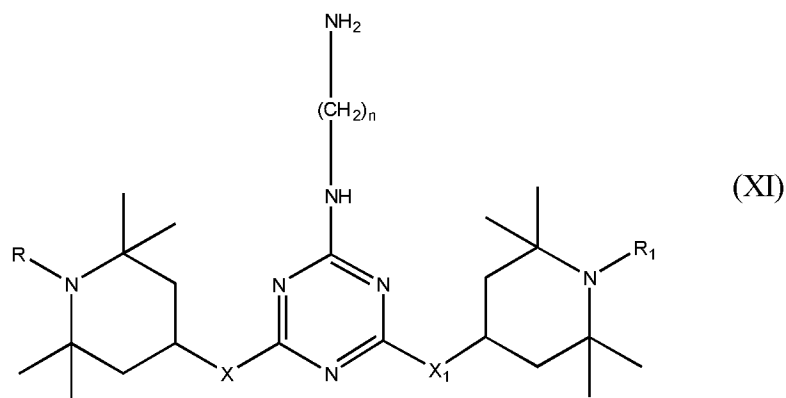
is reacted in equimolar quantities with a generic diamine of formula (X)



wherein n is as above defined,

a derivative of formula (XI) is obtained, whose preparation is also described in

10 Example 1 of US 4,322,337,



wherein R, R_1 , X, X_1 and n are as above defined.

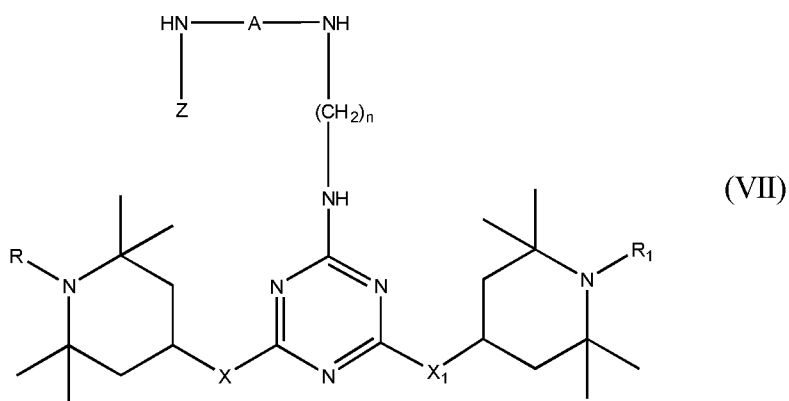
By reaction of two equivalent moles of the derivative of general formula (XI) with a group of formula (XII)

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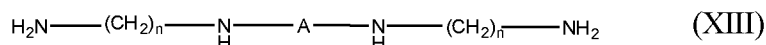
wherein A is as above defined and B represents a halogen selected between chlorine and bromine

the derivative of general formula (VII) can be obtained



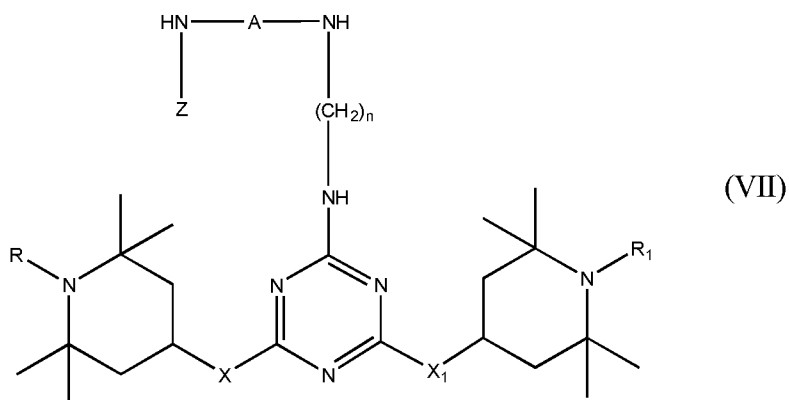
wherein Z = formula (III) group and A, R, R₁, X, X₁ and n are as above defined.

As a further alternative, by reacting an amine of general formula (XIII)



wherein A and n are as above defined

with the above described derivative having the general formula (IX) a derivative can be obtained represented by simplified formula (VII)



wherein Z = group of formula (III) and A, R, R₁, X, X₁ and n are as above defined.

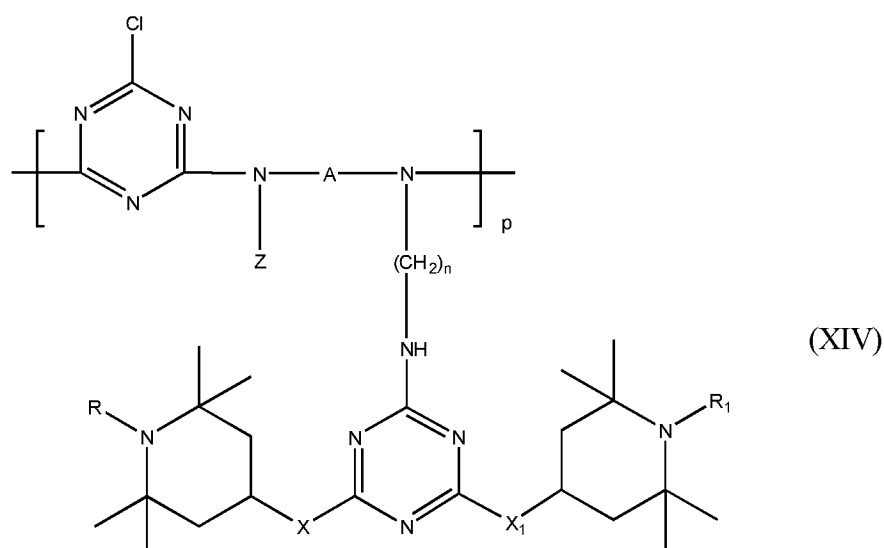
The synthetic polycondensation step, that involves as reactants the formula (VII) reactant and cyanuric chloride, takes place in an inert aprotic solvent selected among toluene, xylene, mesitylene, tetralin, decalin, acetone, dimethylformamide or other

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aprotic solvents. Preferably, the reaction is carried out in the presence of an acidity acceptor; highly water soluble inorganic bases are particularly preferable, such as for example alkaline metal hydroxides or carbonates. Sodium hydroxide or potassium hydroxide are normally used, and the reaction is carried out by using water solutions of these inorganic bases at a weight percent concentration of from 20% to 50%, preferably from 25% to 50%. The reaction takes place in a temperature range of from 5°C and 140°C, preferably from 10°C and 130°C and more preferably from 25°C and 120°C.

The reaction involves the initial mixing of a solution of the compound of formula (VII) with cyanuric chloride (in solution or solid). The mixing is carried out at a temperature of between 5°C and 60°C, preferably between 20°C and 35°C. The addition of cyanuric chloride (in solution or solid) to a solution of the compound of formula (VII) as well as the addition of a solution of the compound of formula (VII) to a solution of cyanuric chloride, brings about the formation of a reactive species capable of subsequently initiating the polycondensation process. Anyway, it has been found that the addition of a solution of cyanuric chloride to a solution of the compound of formula (VII) is preferable.

The addition of an acidity corrector, and the progressive heating of the reaction mixture at a temperature of between 80°C and 120°C, preferably of between 90°C and 110°C brings about formation of the polymeric polypiperidine derivative of formula (XIV)

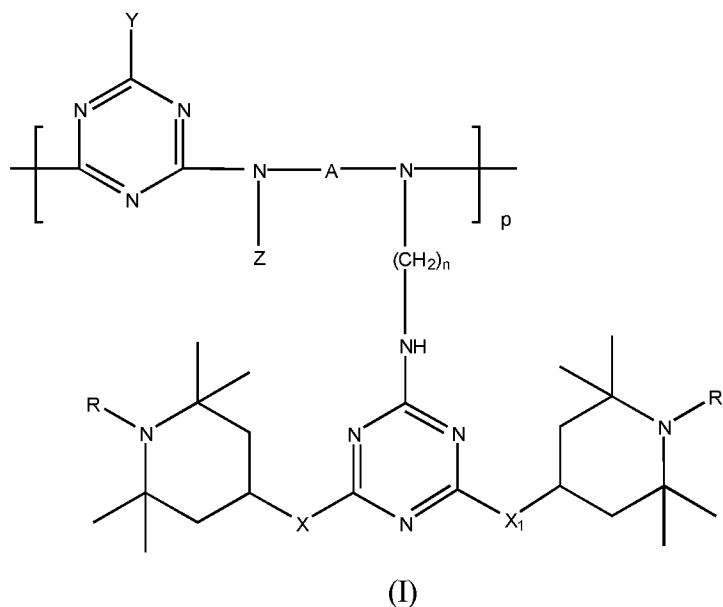


wherein $p = 3-20$ and Z, A, R, R_1, X, X_1 and n have the above defined meaning.

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The use of a light excess of the compound of formula (VII) during the polycondensation step has been found to be advantageous. An excess of between 5 and 25 moles %, and preferably of between 10 and 20 moles % has allowed a polymerization grade p included between 3 and 20, preferably between 3 and 8 and even more preferably between 3 and 6.

By reaction of the polymeric derivative of formula (XIV) with the suitable groups H-Y, wherein Y corresponds to the group of formula (V) or to groups OR_4 and SR_4 , with R_3 and R_4 being as above defined, new polymeric HALS of formula (I) have been obtained:



wherein p , Y , Z , A , X , X_1 , R , R_1 and n have the above defined meanings.

The reaction of the polymeric derivative of formula (XIV) with H-Y, wherein H-Y is morpholine, is particularly simple and economical since it proceeds under mild conditions and provides the desired polypiperidine compound of formula (I) in high yields.

The terminal groups of the HALS of formula (I) can be H, OH, OR_5 wherein R_5 =alkyl or amine group, particularly amine group derived from formula (VII).

The step sequences of the process according to the invention can be used in order to obtain high polycondensation values in other polypiperidine derivatives (HALS) in the repeating unit of which a triazine unit is present.

The quantities of HALS according to the present invention necessary for an

efficient stabilization of the polymeric materials depends on different factors, such as the type and the features of the polymeric material to be stabilized, the use for which said material is intended, the intensity of the radiations and the duration of the foreseen exposure to which the material has to be subjected.

5 In a particular embodiment thereof the present invention consists in adding the compounds of formula (I) to the polymeric material that has to be stabilized in quantities of between 0.01 and 5% by weight with respect to the polymeric material, preferably between 0.1 and 1.0%. Particularly advantageous results are obtained if the polymeric material is a polyolefin.

10 In further embodiments of the present invention the above mentioned composition comprises, as polyolefin material stabilizers, besides the HALS of formula (I), other monomeric, polymeric or macromolecular HALS of different nature.

 The polymers that can be advantageously stabilized according to the present invention are polyethylene, polypropylene, polystyrene, polybutadiene, polyisoprene,
15 and copolymers thereof such as the acrylonitrile-butadiene-styrene copolymer; polyvinylchloride, polyvinylidene chloride and copolymers thereof, polyvinylacetate and copolymers thereof, in particular with ethylene, polyesters such as polyethyleneterephthalate; polyamides, such as for example nylon 6 and nylon 6,6; polyurethans.

20 The compounds of the present invention can be incorporated into the polymeric materials by any known method for mixing additives and polymeric materials such as by:

 - mixing with the polymer, which may be in form of powder or granulated in a suitable mixer or extruder;

25 - adding in the form of a solution or suspension in a suitable solvent and subsequently removing the solvent from the polymer, which may be in form of powder, granulated or in suspension, after thorough mixing;

 - adding to the polymer during preparation thereof, for example during the last stage of preparation.

30 The compounds of formula (I) may be added to the polymeric materials to be stabilized together with HALS of different type as above mentioned, as well as

antioxidants based on phenols, amines, phosphites; UV radiation sorbers based on benzophenones, benzotriazoles; nickel stabilizers; plasticizers, lubricants, antistatic agents, fire retardants, corrosion inhibitors, metal deactivators, mineral fillers such as titanium dioxide, aluminium oxide and similar. Some examples of said additives are the following:

A. ANTIOXIDANTS

1. Alkylated phenols, such as: 2,6-di-tert-butyl-4-methylphenol; 2-(tert-butyl)-4,6-dimethylphenol; 2,6-di-tert-butyl-4-ethylphenol; 2,6-di-tert-butyl-4-butylphenol; 2,6-di-tert-butyl-4-isobutylphenol; 2,6-di-cyclopentyl-4-methylphenol; 2-(α -methylcyclohexyl)-4,6-dimethylphenol; 2,6-di-octadecyl-4-methylphenol; 2,4,6-tricyclohexylphenol; 2,6-di-tert-butyl-4-(methoxymethyl)phenol; linear or branched nonylphenols, such as 2,6-di-cyclononyl-4-methylphenol; 2,4-dimethyl-6-(1'-methylundecyl)phenol; 2,4-dimethyl-6-(1'-heptadecyl)phenol and mixtures thereof.
2. Alkyl-thiomethyl phenols, such as for example 2,4-di-octylthiomethyl-6-tert-butylphenol, 2,4-di-octylthiomethyl-6-methylphenol, 2,4-di-octylthiomethyl-6-ethylphenol, 2,6-di-dodecylthiomethyl-4-nonylphenol .
3. Hydroquinones and alkylated hydroquinones, such as for example: 2,6-di-tert-butyl-4-methoxyphenol; 2,5-di-tert-butyl-hydroquinone; 2,5-di-tert-amyl-hydroquinone; 2,6-diphenyl-4-octadecyloxyphenol; 2,6-di-tert-butyl-hydroquinone; 2,5-di-tert-butyl-4-hydroxyanisole; 3,5-di-tert-butyl-4-hydroxyanisole; 3,5-di-tert-butyl-4-hydroxyphenylstearate; bis(3,5-di-tert-butyl-4-hydroxyphenyl)adipate.
4. Tocopherols, for example α -tocopherol; γ -tocopherol; β -tocopherol; δ -tocopherol and mixtures thereof (vitamin E).
5. Hydroxylated thiodiphenyl ethers, such as 2,2'-thiobis(6-tert-butyl-4-methylphenol); 2,2'-thiobis(4-octylphenol); 4,4'-thiobis(6-tert-butyl-3-methylphenol); 4,4'-thiobis(6-tert-butyl-2-methylphenol); 4,4'-bis[2,6-dimethyl-4-hydroxyphenyl]disulfide.
6. Alkylidene bisphenols, such as 2,2'-methylene-bis(6-tert-butyl-4-methylphenol); 2,2'-methylene-bis(6-tert-butyl-4-ethylphenol); 2,2'-methylene-bis(4-methyl-6-(α -methylcyclohexyl)phenol); 2,2'-methylene-bis(4-methyl-6-cyclohexylphenol); 2,2'-methylene-bis(6-nonyl-4-methylphenol); 2,2'-methylene-bis-

(4,6-di-tert-butylphenol); 2,2'-ethylidene-bis(4,6-di-tert-butylphenol); 2,2'-ethylidene-bis(6-tert-butyl-4-isobutylphenol); 2,2'-methylene-bis(6-(α -methylbenzyl)-4-nonylphenol); 2,2'-methylenebis(6-(α - α -dimethylbenzyl)-4-nonylphenol); 4,4'-methylenebis(2,6-di-tert-butyl-phenol); 4,4'-methylenebis(6-tert-butyl-2-methylphenol); 1,1-bis(5-tert-butyl-4-hydroxy-2-methylphenyl)butane; 2,6-bis(3-tert-butyl-5-methyl-2-hydroxybenzyl)-4-methylphenol; 1,1,3-tris(5-tert-butyl-4-hydroxy-2-methylphenyl)butane; 1,1-bis(5-tert-butyl-4-hydroxy-2-methylphenyl)-3-n-dodecyl-mercaptobutane; ethylene glycol bis-(3,3-bis-(3'-tert-butyl-4'-hydroxyphenyl)-butyrate); bis(2-(3'-tert-butyl-2'-hydroxy-5'-methylbenzyl)-6-tert-butyl-4-

methylphenyl)terephthalate; bis(3-tert-butyl-4-hydroxy-5-methylphenyl)dicyclopentadiene; 1,1-bis(3,5-dimethyl-2-hydroxyphenyl)butane; 2,2-bis-(3,5-di-tert-butyl-4-hydroxyphenyl)propane; 2,2-bis-(5-tert-butyl-4-hydroxy-2-methylphenyl)-4-n-dodecyl-mercaptobutane, 1,1,5,5-tetra-(5-ter-butyl-4-hydroxy-2-methylphenyl)pentane.

7. O-, N- and S-benzyl derivates such as: 3,5,3',5'-tetra-ter-butyl-4-4'-dihydroxydibenzyl ether; octadecyl-4-hydroxy-3,5-dimethylbenzyl-mercapto acetate; tridecyl-4-hydroxy-3,5-di-ter-butyl-benzylmercapto acetate; tri(3,5-di-tert-butyl-4-hydroxybenzyl)amine; bis(4-tert-butyl-3-hydroxy-2,6-dimethylbenzyl)dithioterephthalate; bis(3,5-di-tert-butyl-4-hydroxybenzyl)disulphide; isooctyl-3,5-di-tert-butyl-4-hydroxybenzylmercaptoacetate.

8. Malonates containing the hydroxybenzyl groups such as; dioctadecyl-2,-2,2-bis(3,5-di-tert-butyl-2-hydroxybenzyl)malonate; dioctadecyl-2,- (3 -tert-butyl-4-hydroxy-5-methylbenzyl)malonate; di-dodecylmercaptoethyl-2,2'-bis-(3,5-di-tert-butyl-4-hydroxybenzyl)malonate; bis-(4-(1,1,3,3-tetramethylbutyl)-phenyl)-2,2-bis-(3,5-di-tert-butyl-4-hydroxybenzyl)malonate.

9. Hydroxybenzyl aromatic compounds, such as 1,3,5-tris-(3,5-di-tert-butyl-4-hydroxybenzyl)-2,4,6-trimethylbenzene; 1,4-bis(3,5-di-tert-butyl-4-hydroxybenzyl)-2,3,5,6-tetramethylbenzene; 2,4,6-tris-(3,5-di-tert-butyl-4-hydroxybenzyl)-phenol.

10. Triazine derivates, such as 2,4-bis(octylmercapto)-6-(3,5-di-tert-butyl-4-hydroxyanilino)-1,3,5-triazine; 2-octylmercapto-4,6-bis(3,5-di-tert-butyl-4-

hydroxyanilino)-1,3,5-triazine; 2-octylmercapto-4,6-bis(3,5-di-tert-butyl-4-hydroxyphenoxy)-1,3,5-triazine; 2,4,6-tris(3,5-di-tert-butyl-4-hydroxyphenoxy)-1,3,5-triazine; 1,3,5-tris(3,5-di-tert-butyl-4-hydroxybenzyl)-isocyanurate; 1,3,5-tris-(4-tert-butyl-3-hydroxy-2,6-dimethylbenzyl)isocyanurate; 2,4,6-tris(3,5-di-tert-butyl-4-hydroxyphenylethyl)-1,3,5-triazine; 1,3,5-tris-(3,5-di-tert-butyl-4-hydroxyphenylpropionyl)hexahydro-1,3,5-triazine; 1,3,5-tris-(3,5-dicyclohexyl-4-hydroxybenzyl)isocyanurate.

11. Benzylphosphonates, such as for example: dimethyl-2,5-di-tert-butyl-4-hydroxybenzylphosphonate; diethyl-3,5-di-tert-butyl-4-hydroxybenzylphosphonate; di-octadecyl-3,5-di-tert-butyl-4-hydroxybenzylphosphonate; di-octadecyl-5-tert-butyl-4-hydroxy-3-methylbenzylphosphonate; calcium salt of the monoethylic ester of 3,5-di-tert-butyl-4-hydroxybenzylphosphonic acid.

12. Acylamino phenols such as lauric acid 4-hydroxyanilide, stearic acid 4-hydroxyanilide, octyl N-(3,5-di-tert-butyl-4-hydroxyphenyl)carbamate.

13. β -(3,5-di-tert-butyl-4-hydroxyphenyl)propionic acid esters with mono- or polyhydric alcohols such as; methanol, ethanol, n-octanol, iso-octanol, octadecanol; 1,6-esandiol, 1,9-nonadiol, ethylenic glycol, 1,2-propandiol, neopentyl glycol, thiodiethylene glycol, diethylene glycol, triethylene glycol, pentaerithrol, tri-(hydroxyethyl)isocyanurate; N,N'-bis(hydroxyethyl)oxamide; 3-thioundecanol; 3-thiopentadecanol; trimethyl hexanediol; trimethylolpropane; 4-hydroxymethyl-1-phospho-2,6,7-trioxabicyclo(2,2,2)octane.

14. β -(5-tert-butyl-4-hydroxy-3-methylphenyl)propionic acid esters with mono- or polyhydric alcohols such as; methanol, ethanol, n-octanol, iso-octanol, octadecanol; 1,6-esandiol, 1,9-nonadiol, ethylenic glycol, 1,2-propandiol, neopentyl glycol, thiodiethylene glycol, diethylene glycol, triethylene glycol, pentaerithrol, tri-(hydroxyethyl)isocyanurate; N,N'-bis(hydroxyethyl)oxamide; 3-thioundecanol; 3-thiopentadecanol; trimethyl hexanediol; trimethylolpropane; 4-hydroxymethyl-1-phospho-2,6,7-trioxabicyclo(2,2,2)octane.

15. β -(3,5-dicyclohexyl-4-hydroxyphenyl)propionic acid esters with mono- or polyhydric alcohols such as; methanol, ethanol, n-octanol, iso-octanol, octadecanol; 1,6-esandiol, 1,9-nonadiol, ethylenic glycol, 1,2-propandiol, neopentyl glycol,

thiodiethylene glycol, diethylene glycol, triethylene glycol, pentaerithrol, tri-(hydroxyethyl)isocyanurate; N,N'-bis(hydroxyethyl)oxamide; 3-thioundecanol; 3-thiopentadecanol; trimethyl hexanediol; trimethylolpropane; 4-hydroxymethyl-1-phospho-2,6,7-trioxabicyclo(2,2,2)octane.

5 16. 3,5,-di-tert-butyl-4-hydroxyphenyl acetic acid esters with mono- or polyhydric alcohols such as; methanol, ethanol, n-octanol, iso-octanol, octadecanol; 1,6-esandiol, 1,9-nonadiol, ethylenic glycol, 1,2-propandiol, neopentyl glycol, thiodiethylene glycol, diethylene glycol, triethylene glycol, pentaerithrol, tri-(hydroxyethyl)isocyanurate; N,N'-bis(hydroxyethyl)oxamide; 3-thioundecanol; 3-
10 thiopentadecanol; trimethyl hexanediol; trimethylolpropane; 4-hydroxymethyl-1-phospho-2,6,7-trioxabicyclo(2,2,2)octane.

17. β -(3,5-di-tert-butyl-4-hydroxyphenyl)propionic acid amides such as: N,N'-bis(3,5-di-tert-butyl-4-hydroxyphenylpropionil)-hexamethylene diamide; N,N'-bis-(3,5-di-tert-butyl-4-hydroxyphenylpropionyl)-trimethylendiamide; N,N'-bis-(3,5-di-tert-
15 butyl-4-hydroxyphenylpropionyl)hydrazide; N,N'-bis(2-(3-(3,5-di-tert-butyl-4-hydroxyphenyl)propionyloxy)ethyl)-oxamide.

18. Ascorbic acid (Vitamin C).

19. Amine antioxidants such as: N,N'-diisopropyl-p-phenylenediamine; N,N'-di-sec-butyl-p-phenylenediamine; N,N'-bis(1,4-dimethyl-pentyl)-p-
20 phenylenediamine; N,N'-bis(1-ethyl-3-methylpentyl)-p-phenylenediamine; N,N'-bis(1-methylheptyl)-p-phenylenediamine; N,N'-dicyclohexyl-p-phenylenediamine; N,N'-diphenyl-p-phenylenediamine; N,N'-bis-(2-naphtyl)-p-phenylenediamine; N-isopropyl-N'-phenyl-p-phenylenediamine; N-(1,3-dimethyl-butyl)-N'-phenyl-p-phenylenediamine; N-1-methylheptyl)-N'-phenyl-p-phenylenediamine; N-cyclohexyl-N'-phenyl-p-
25 phenylenediamine; 4-(p-toluensulfamoyl)-diphenylamine; N,N'-dimethyl-N,N'-di-sec-butyl-p-phenylendimine; diphenylamine; N-allyl-diphenylamine; 4-isopropoxy-diphenylamine; N-phenyl-1-naphtylamine; N-(4-ter-octylphenyl)-1-naphtylamine; N-phenyl-2-naphtylamine; p,p'-di-ter-octyldiphenylamine; 4-n-butyl-aminophenol; 4-butyryl-aminohenol; 4-nonanoylaminophenol; 4-dodecanoyl-aminophenol; 4-
30 octadecanoyl-aminophenol; bis(4-mothoxyphenyl)amine; 2,6-di-ter-butyl-4-dimethylaminomethylphenol; 2,4'-diaminodiphenylmethane; 4,4'-diaminodiphenyl-

methane; N,N,N',N'-tetramethyl-4,4'-diaminodiphenylmethane; 1,2-bis-((2-methylphenyl)amino)ethane; 1,2-bis-(phenylamino)propane; o-tolil-biguanide; bis-(4-(1',3'-dimethylbutyl)phenyl)amine); ter-octyl-N-phenyl-1-naphtylamine; mixtures of dialkylated tert-butyl/tert-octyl-diphenylamines; mixtures of mono- and di-alkyl
 5 nonyldiphenylamines; mixtures of mono- and di-alkyl dodecyldiphenylamines; mixtures of mono- and di-alkyl isopropyl/isoheptyldiphenylamines; mixtures of mono- and di-alkyl tertbutyldiphenylamines; 2,3-dihydro-3,3-dimethyl-4H-1,4-benzothiazine; phenothiazine; mixtures of mono- and di-alkyl tert-butyl/tert-octylphenothiazine; mixtures of mono- and di-alkyl tert-octyl phenothiazine; N-allyl phenothiazine;
 10 N,N,N',N'-tetraphenyl-1,4-diamino-2-butene; N,N'-bis-(2,2,6,6-tetramethyl-piperidinyl-4-hexamethylenediamine; bis(2,2,6,6- tetramethyl-piperid -4-yl)sebacate; 2,2,6,6-tetramethyl-piperid -4-one; 2,2,6,6- tetramethyl-piperid -4-ol.

B. UV ADSORBERS AND LIGHT STABILIZERS

1. 2-(2'- hydroxyphenyl)benzotriazoles, such as: 2-(2'- hydroxy-5-methylphenyl)benzotriazole; 2-(3',5'-di-tert-butyl-2'-hydroxyphenyl)benzotriazole; 2-(5'-tert-butyl-2'- hydroxyphenyl)benzotriazole; 2-(2'-hydroxy-5'-(1,1,3,3-tetramethylbutyl)phenyl)benzotriazole; 2-(3',5'-di-tert-butyl-2'- hydroxyphenyl)-5-chloro-benzotriazole; 2-(3'-tert-butyl-2'-hydroxy-5'-methylphenyl)-5-chloro-benzotriazole; 2-(3'-sec-butyl-5'-tert-butyl-2'- hydroxyphenyl)benzotriazole; 2-(2'-hydroxy-4'-
 20 octyloxyphenyl)benzotriazole; 2-(3',5'-di-tert-butyl-2'- hydroxyphenyl)-benzotriazole; 2-(3',5'-bis-(α,α -dimethylbenzyl)-2'- hydroxyphenyl)benzotriazole; 2-(3'-tert-butyl-5'-(2-(2-ethylhexyloxy)-carbonyl-ethyl)-2'- hydroxyphenyl)-5-chloro-benzotriazole; 2-(3'-tert-butyl-2'-hydroxy-5'-(2-octyloxycarbonyl-ethyl)phenyl)-5-chloro-benzotriazole; 2-(3'-tert-butyl-2'-hydroxy-5'-(2-methoxycarbonyl-ethyl)phenyl)-5-chloro-benzotriazole; 2-(3'-
 25 (3'-tert-butyl-2'-hydroxy-5'-(2-methoxycarbonyl-ethyl)phenyl)-benzotriazole; 2-(3'-tert-butyl-2'-hydroxy-5'-(2-octyloxycarbonyl-ethyl)phenyl)-benzotriazole; 2-(3'-tert-butyl-5'-(2-(2-ethylhexyloxy)-carbonyl-ethyl)-2'- hydroxyphenyl)-benzotriazole; 2-(3'-dodecyl-2'-hydroxy-5'-methylphenyl)-benzotriazole; 2-(3'-tert-butyl-2'-hydroxy-5'-(2-isooctyloxycarbonyl-ethyl)-phenyl)-benzotriazole; 2,2'-methylene-bis-(4-(1,1,3,3-tetramethylbutyl)-6-benzotriazol-2-ylphenol); the transesterification product of 2-(3'-
 30 tert-butyl-5'-(2-methoxycarbonyl-ethyl)-2'- hydroxyphenyl)-2H-benzotriazole with

polyethyleneglycole 300; (R-CH₂-CH₂-COO-CH₂-CH₂)₂-wherein R can be: 3'-tert-butyl-4'-hydroxy-5'-2H-benzotriazole-2-ylphenyl; 2-(2'-hydroxy-3'-(α,α -dimethylbenzyl)-5'-(1,1,3,3-tetramethylbutyl)-phenyl)benzotriazole; 2-(2'-hydroxy-3'-(1,1,3,3-tetramethylbutyl)-5'-(α,α -dimethylbenzyl) phenyl)benzotriazole.

5 2. 2-hydroxybenzophenones such as for example the 4-hydroxy-; 4-methoxy-; 4-octyloxy-; 4-decyloxy-; 4-dodecyloxy-; 4-benzyloxy-; 4,2',4'-tri-hydroxy- and 2'-hydroxy-4,4'-dimethoxy derivatives.

3. Esters of substituted and non-substituted benzoic acids, such as for example: 4-tertbutyl-phenyl-salicylate; phenyl salicylate; octylphenyl salicylate; dibenzoyl
10 resorcinol; bis-(4-tert-butyl-benzoyl)-resorcinol; benzoyl resorcinol; 2,4-di-tert-butylphenyl 3,5-di-tert-butyl-4-hydroxybenzoate; hexadecyl 3,5-di-tert-butyl-4-hydroxybenzoate; octadecyl 3,5-di-tert-butyl-4-hydroxybenzoate; 2-methyl-4,6-di-tert-butylphenyl 3,5-di-tert-butyl-4-hydroxybenzoate:

4. Acrilates, such as for example: ethyl α -cyano- β,β -diphenylacrilate; isooctyl α -
15 cyano- β,β -diphenylacrilate; methyl α -carbomethoxycinnamate; methyl α -cyano- β -methyl-p-methoxy-cinnamate; butyl α -cyano- β -methyl-p-methoxy-cinnamate; methyl α -carbomethoxy-p-methoxycinnamate and N-(β -carbomethoxy- β -cyanovinyl)-2-methylindoline.

5. Nickel derivatives such as for example: nickel complexes 1:1 or 1:2 with 2,2'-
20 thio-bis-(4-(1,1,3,3-tetramethylbutyl)phenol, with or without ligands such as n-butylamine, triethanolamine or N-cyclohexyldiethanolamine; nickel dibutyldithiocarbamate; nickel salts of mono-alkyl esters (for example methyl or ethyl esters) of 4-hydroxy-3,5-di-tert-butylbenzylfosfonic acid; nickel complexes of keto-oximes, for example of 2-hydroxy-4-methylphenyl undecyl-keto-oxime; nickel
25 complexes of 1-phenyl-4-lauroyl-5-hydroxy-pyrazole, with or without additional ligands.

6. Sterically hindered amines, such as for example: bis (2,2,6,6-tetramethyl-4-piperidinyl)sebacate CAS RN=52829-07-9; bis (2,2,6,6-tetramethyl-4-piperidinyl)succinate; bis (1,2,2,6,6-pentamethyl-4-piperidinyl)sebacate CAS
30 RN=41556-26-7; (1,2,2,6,6-pentamethyl-4-piperidinyl)methyl sebacate CAS RN=82919-37-7; bis (1-octyloxy-2,2,6,6-tetramethyl-4-piperidinyl)sebacate CAS

RN=129757-67-1; bis (1,2,2,6,6-pentamethyl-4-piperidiny) n-butyl-3,5-di-ter-butyl-4-hydroxybenzylmalonate; the condensation product of 1-(2-hydroxyethyl)-2,2,6,6-tetramethyl-4-hydroxypiperidine with succinic acid CAS RN=65447-77-0; cyclic or linear condensates of N,N'-bis-(2,2,6,6-tetramethyl-4-piperidiny)hexamethylenediamine with 4-tert-octylamino-2,6-dichloro-1,3,5-triazine CAS RN=71878-19-8; tris(2,2,6,6-tetramethyl-4-piperidiny)nitrilotriacetate; tetra(2,2,6,6-tetramethyl-4-piperidiny)-1,2,3,4-butan-tetracarboxylate; 1,1'-(1,2-ethanediyl)-bis(3,3,5,5-tetramethylpiperazinone); 4-benzoyl-2,2,6,6-tetramethylpiperidine; 4-stearyloxy-2,2,6,6-tetramethyl-piperidine CAS RN=167078-06-0; bis(1,2,2,6,6-pentamethyl-4-piperidiny)-2-n-butyl-2-(hydroxy-3,5-di-tert-butylbenzyl)malonate CAS RN=63843-89-0; 3-n-octyl-7,7,9,9-tetramethyl-1,3,8-triazaspiro(4,5)decan-2,4-dione; bis(1-octiloxy-2,2,6,6-tetramethylpiperidiny)sebacate; bis(1-octiloxy-2,2,6,6-tetramethyl-piperidiny)succinate; linear or cyclic condensates of N,N'-bis-(2,2,6,6-tetramethyl-4-piperidiny)-hexamethylenediamine with 4-morpholino-2,6-dichloro-1,3,5-triazine CAS RN=82451-48-7; cyclic or linear condensates of N,N'-bis-(1,2,2,6,6-pentamethyl-4-piperidiny)-hexamethylenediamine with 4-morpholino-2,6-dicloro-1,3,5-triazine CAS RN=193098-40-7; the condensation product of 2-chloro-4,6-bis(4-n-butylamino-2,2,6,6-tetramethylpiperidiny)-1,3,5-triazine with 1,2-bis(3-amino-propylamino)ethane; the condensation product of 2-chloro-4,6-bis(4-n-butylamino-1,2,2,6,6-pentamethylpiperidiny)-1,3,5-triazine with 1,2-bis(3-aminopropylamino)ethane CAS RN=106990-43-6; 8-acetyl-3-dodecyl - 7,7,9,9-tetramethyl-1,3,8-triazaspiro(4,5)decan-2,4-dione; 3-dodecyl-1-(2,2,6,6-tetramethyl-4-piperidiny)pyrrolidin-2,5-dione; 3-dodecyl-1-(1,2,2,6,6-pentamethyl-4-piperidiny)pyrrolidin-2,5-dione; mixture of 4-hexadecyloxy- and 4-octadecyloxy-2,2,6,6-tetramethylpiperidine; the condensation product of N,N'-bis-(2,2,6,6-tetramethyl-4-piperidiny)hexamethylenediamine with 4-cyclohexylamine-2,6-dichloro-1,3,5-triazine; N-(2,2,6,6-tetramethyl-4-piperidiny)n-dodecyl succinimide; 2-undecyl-7,7,9,9-tetramethyl-1-oxo-3,8-diazo-4-oxo-spiro(4,5)decane; the condensation product of 7,7,9,9-tetramethyl-2-cycloundecyl-1-oxo-3,8-diazo-4-oxo-spiro(4,5)decane with epichlorodrine; 1,1-bis(1,2,2,6,6-pentamethyl-4-piperidinyloxycarbonyl)-2-(4-methoxyphenyl)ethene; N,N'-bis-formil-N,N'-bis(2,2,6,6-

tetramethyl-4-piperidiny)hexamethylenediamine diester of 4-methoxy-methylen-malonic acid with 1,2,2,6,6-pentamethyl-4-hydroxypiperidine; poly(methylpropyl-3-oxy-4-(2,2,6,6-tetramethyl-4-piperidiny))-siloxane; the reaction product of the copolymer maleic acid/alfa-olefins with 2,2,6,6-tetramethyl-4-aminopiperidine or
 5 1,2,2,6,6-pentamethyl-4-amino-piperidine; condensation product of 2-chloro-4,6-bis-(4-n-butylamino-1-cyclohexyloxy-2,2,6,6-tetramethylpiperidiny)-1,3,5-triaziny with 1,2-bis-(3-amino-propylamino)-ethane; condensation product of 1,6-hexandiamine-bis(2,2,6,6-tetramethyl-4-piperidiny)- with the condensation product of 2,4,6-trichloro-1,3,5-triazine with di-n-butylamine and
 10 N-butyl-2,2,6,6-tetramethyl-4-piperidinamine CAS RN=192268-64-7; derivates of 7-oxa-3,20-diaza-dispiro-(5.1.11.2)-eneicosanone identified by CAS RN 64338-16-5; 85099-51-0; 85099-50-9; 202483-55-4; reaction product of 2,2,6,6-tetramethyl-4-piperidine with the polymer obtainable by copolymerization of maleic anhydride with alkenes C20-24 CAS RN=152261-33-1; products described in EP 782994.

15 7. Oxamides, such as for example: 4,4'-dioctyloxy-oxalanilide; 2,2'-diethoxy-oxalanilide; 2,2'-dioctyloxy-5,5'-di-ter-butyl-oxalanilide; 2,2'-didodecyloxy-5,5'-di-ter-butyl-oxalanilide; 2-ethoxy-2'-ethyloxy-oxalanilide; N,N'-bis(3-dimethylaminopropyl)oxalanilide; 2-ethoxy-2'-ethyl-5,4'-di-ter-butyl-oxalanilide; mixtures of oxalanilides o- and p- methoxy disubstituted and mixtures of oxalanilides o- and p- ethoxy disubstituted.
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8. 2-(2-hydroxyphenyl)-1,3,5-triazines, such as for example: 2,4,6-tris(2-hydroxy-4-octyloxyphenyl)-1,3,5-triazine; 2-(2-hydroxy-4-octyloxyphenyl)-4,6-bis(2,4-dimethylphenyl)-1,3,5-triazine; 2,4-bis(2-hydroxy-4-propyloxyphenyl)-6-(2,4-dimethylphenyl)-1,3,5-triazine; 2-(2-hydroxy-4-octyloxyphenyl)-4,6-bis-(4-methyl-phenyl)-1,3,5-triazine; 2-(2-hydroxy-4-dodecyloxyphenyl)-4,6-bis-(2,4-dimethyl-phenyl)-1,3,5-triazine; 2-(2-hydroxy-4-tridecyloxyphenyl)-4,6-bis-(2,4-dimethyl-phenyl)-1,3,5-triazine; 2-(2-hydroxy-4-(2-hydroxy-3-butyloxy-propoxy)phenyl)-4,6-bis(2,4-dimethyl)-1,3,5-triazine; 2-(2-hydroxy-4-(2-hydroxy-3-octyloxy-propoxy)-phenyl)-4,6-bis(2,4-dimethyl)-1,3,5-triazine; 2-(4-(dodeculoxy/tridecyloxy-2-hydroxypropoxy)-2-hydroxyphenyl)-4,6-bis(2,4-dimethylphenyl)-1,3,5-triazine; 2-(2-hydroxy-4-(2-hydroxy-3-dodecyloxy-propoxy)phenyl)-4,6-bis(2,4-dimethylphenyl)-
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1,3,5-triazine; 2-(2-hydroxy-4-hexyloxy)phenyl-4,6-diphenyl-1,3,5-triazine; 2-(2-hydroxy-4-methoxyphenyl)-4,6-diphenyl-1,3,5-triazine; 2,4,6-tris(2-hydroxy-4-(3-butoxy-2-hydroxy-propoxy)phenyl)-1,3,5-triazine; 2-(2-hydroxyphenyl)-4-(4-methoxyphenyl)-6-phenyl-1,3,5-triazine; 2-(2-hydroxy-4-(3-(2-ethylhexyl-1-oxy)-2-hydroxypropyloxy)phenyl)-4,6-bis(2,4-dimethylphenyl)-1,3,5-triazine.

C. METAL DEACTIVATORS

For example: N,N'-diphenyloxamide; N-salicylal-N'-bis-salicyloyl-hydrazine; N,N'-bis(salicyloyl)hydrazine; N,N'-bis(3,5-di-tert-butyl-4-hydroxyphenylpropionyl)hydrazine; 3-salicyloylamino-1,2,4-triazole; bis(benzylidene)oxalyl dihydrazide; oxalanilide; isoftaloyl dihydrazide; sebacoyl bisphenylhydrazide; N,N'-diacetyladiopoyl dihydrazide; N,N'-bis(salicyloyl)oxalyl dihydrazide; N,N'-bis(salicyloyl)thiopropionyl dihydrazide.

D. PHOSPHITES AND PHOSPHONITES

For example: triphenyl phosphite; diphenyl alchil phosphites; phenyl dialkyl phosphites; tris(nonylphenyl)phosphite; trilauryl phosphite; trioctadecyl phosphite; distearyl pentaerythritol diphosphite; tris(2,4-di-tert-butyl-phenyl) phosphite; diisodecyl pentaerythritol diphosphite; bis(2,4-di-tert-butylphenyl) phosphite; diisodecyl pentaerythritol diphosphite; bis(2,4-di-tert-butylphenyl) pentaerythritol diphosphite; bis(2,6-di-tert-butyl-4-methylphenyl)-pentaerythritol diphosphite; diisodecyloxy-pentaerythritol diphosphite; bis-(2,4-di-tert-butyl-6-methylphenyl)pentaerythritol diphosphite; bis(2,4,6-tris(ter-butylphenyl)pentaerythritol diphosphite; tristearyl sorbitol triphosphite; bis(2,4-di-tert-butyl-6-methylphenyl) methyl phosphite; bis(2,4-di-tert-butyl-6-methylphenyl) ethyl phosphite; 2,2',2''-nitrilo(triethyltris(3,3',5,5'-tetra-tert-butyl-1,1'-biphenyl-2,2'-idyl) phosphite); 2-ethylhexyl (3,3',5,5'-tetra-tert-butyl-1,1'-biphenyl-2,2'-idyl) phosphite); tetra(2,4-di-tert-butylphenyl) 4-4'-biphenylene diphosphonite.

E. HYDROXYLAMINES

For example: N,N-dibenzylhydroxylamine; N,N-diethylhydroxylamine; N,N-

dioctylhydroxylamine; N,N-dilaurylhydroxyl-amine; N,N-ditetradecylhydroxylamine; N,N-dihexadecylhydroxylamine; N,N-dioctadecylhydroxylamine; N-hexadecyl-N-octadecylhydroxylamine; N-heptadecyl-N-octadecylhydroxylamine; N,N-dialkylhydroxylamines derived from the hydrogenated tallow amines.

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F. NITRONES

For example: N-benzyl-alfa-phenyl-nitrone; N-ethyl-alfa-methyl-nitrone; N-octyl-alfa-eptyl-nitrone; N-lauryl-alfa-undecyl-nitrone; N-tetradecyl-alfa-tridecyl-nitrone; N-hexadecyl-alfa-pentadecyl-nitrone; N-octadecyl-alfa-pentadecyl-nitrone; N-heptadecyl-
10 alfa-heptadecyl-nitrone; N-octadecyl-alfa-hexadecyl-nitrone; nitrones derived from N,N-dialkylhydroxylamines obtained from amines of hydrogenated tallow.

G. THIOSYNERGIC DERIVATES

For example dilauryl thiodipropionate or stearyl thiodipropionate.

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H. ANTIPEROXIDE AGENTS

For example esters of the thiodipropionic acid with lauryl, stearyl, miristic or tridecyl alcohols; mercaptobenzimidazole or 2-mercapto-benzimidazole zinc salt; zinc dibutyldithiocarbamate; dioctadecyl disulphide; pentaerythritol tetrakis(β -
20 dodecylmercapto)propionate.

I. POLYAMIDE STABILIZERS

For example copper salts in combination with iodides and/or phosphorated compounds and bivalent manganese salts.

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L. BASIC CO-STABILIZERS

For example: melamine; polyvinylpolypyrrolidone; dicyandiamide; triallylcyanurate; urea derivates; hydrazine derivates; amines; polyamides; polyurethans; alkaline metal and alkaline-earth metal salts of long-chain fatty acids such
30 as calcium stearate, zinc stearate, magnesium behenate, magnesium stearate, sodium ricinoleate, potassium palmitate, pyrocatechol antimonium or zinc salts.

M. NUCLEATING AGENTS

For example: inorganic substances such as talc; metal oxides such as titanium dioxide or magnesium oxide; phosphates, carbonates or sulphates of earth-alkaline metal salts; organic compounds such as mono or polycarboxylic acids and salts thereof, such as 4-ter-butylbenzoic acid, adipic acid, diphenylacetic acid, sodium succinate; sodium benzoate; polymeric compounds such as anionic copolymers.

N. BENZOFURANONES AND INDOLINONES

For example the ones described in USP 4,325,863; USP 4,338,244; USP 5,175,312; USP 5,216,052; USP 5,252,643; DE-A-4316611; DE-A-4316622; DE-A-4316876; EP-A-0589839; EP-A-0591102; 3-(4-(2-acetoethoxy)phenyl)-5,7-di-ter-butyl-benzofuran-2-one; 5,7-di-ter-butyl-3-(4-(2-stearoyloxyethoxy)phenyl)benzofuran-2-one; 3,3'-bis(5,7-di-ter-butyl-3-(4-(2-hydroxyethoxy)phenyl)benzofuran-2-one); 5,7-di-ter-butyl-3-(4-ethoxyphenyl)benzofuran-2-one; 3-(4-acetoxy-3,5-dimethylphenyl)-5,7-di-ter-butyl-benzofuran-2-one; 3-(3,5-dimethyl-4-pivaloyloxyphenyl)-5,7-di-ter-butyl-benzofuran-2-one; 3-(2,3-di-methylphenyl)-5,7-di-ter-butyl-benzofuran-2-one.

O. FILLERS AND REINFORCING AGENTS

For example: calcium carbonate; silicates; glass fibers; asbestos; talc; kaolin; mica; barium sulphate; metal oxides and hydroxides, carbon black; graphite; wood flour or fiber or other natural products; synthetic fibers.

P. OTHER ADDITIVES

For example plasticizers, lubricants, emulsifiers, pigments, rheology modifiers; catalysts; flow control agents; optical bleach; anti flame agents; antistatic agents, swelling agents.

The invention will be further described in greater detail with reference to the below given working examples, wherein working conditions are described that allow

new polymeric polypiperidine derivatives of general formula (I), with p included between 3 and 20, to be obtained according to the present invention.

Although it is clear that general formula (I) represents the primary structure of the polypiperidine derivative, being this a polymer it may happen that different kinds of oligomeric structures are present in mixture with the primary structure (I).

Characterization of the products was directed to the determination of the molecular weight distribution (MWD) by conventional size exclusion chromatography (SEC). The technique used a differential reflectometer as concentration sensor and a relative calibration built with polystyrene (PS) standards with MWD narrow.

The samples have been solubilised in the SEC mobile phase at the desired concentration. A Waters modular chromatographic HPLC/SEC system was used.

In the sample characterization the chromatographic system has used the following experimental conditions:

- Mobile phase: THF + 0,08 M diethanolamine;
- Columns bench: 2 Mesopore Polymer Laboratories;
- Temperature: 30°C;
- Flow: 0.6 l/min;
- Degassing: helium
- Concentration: » 10 mg/ml;
- Injection volume: 50 µL;
- Test time: 40 minuts;
- Calibration: polinomial relative of 3° order built with 8 polystyrene (PS) standards with MWD narrow with peak molecular weight Mp of between 95000 and 382 g/mole.

For each sample and in comparison between the sample the following parameters have been determined:

- MWD;
- Molecular weight average Mn, Mw;
- polydispersity index: Mw/Mn.

Preparation examples:**Example 1****Preparation example**

A polymeric HALS having structure (I) with $p=3-20$, $A=-(CH_2)_a-$ wherein $a=2$,
5 Z =formula (III) group, $n=3$, X and X_1 = formula (II) group with R_2 = butyl, R and R_1 =
H and Y = morpholine residue, has been prepared with the following described
procedure.

0.24 moles of N-(2,2,6,6-tetramethyl-4-piperidiny)-butylamine were added in 30
minutes, under stirring at room temperature, to a solution obtained by dissolving 0.12
10 moles of cyanuric chloride in 170 g of xylene. Then, 40 g of water and 0.27 moles of
sodium hydroxide in form of 30% water solution were added and slow heating up to
95°C was applied, this temperature was maintained for two hours, and the water phase
was then removed. The obtained xylene solution, containing 0.12 moles of a compound
of general formula (IX) wherein R and R_1 = H and X and X_1 being both the group of
15 formula (II) wherein R_2 = butyl, was additioned with 0.06 moles of N,N'-
bis(aminopropyl)-etilendiamine, corresponding to an amine of general formula (XIII)
wherein $n=3$ and $A=-(CH_2)_a-$ wherein $a=2$ and the acidity was neutralized with a light
alkali excess. The mixture was gradually heated up to the full reflux temperature, by
removing reaction water by distillation, and then maintained at the same temperature for
20 three hours. The mixture was allowed to cool at 90°C, and 100 g of water were added;
after stirring for 30' at 80-90°C the water phase was removed.

At the xylene solution containing 0.06 moles of an amine of general formula (VII)
wherein Z is the group of formula (III) and n , A , R , R_1 , X , X_1 are as above defined, a
mixture has been added that was obtained by dissolving 0.054 of cyanuric chloride in
25 100 g of xylene, under stirring and by maintaining the temperature below 30°C. Once
the addition was completed, the mixture was kept under stirring for one hour and then
slowly heated up to 60°C. At this temperature a slight excess (0.113 moles) of sodium
hydroxide in 30% water solution was added and the mixture was slightly heated up to
90°C, at which temperature stirring was maintained for 1 h. Then, 100 g of water were
30 added and stirring was maintained for 30' and then the water phase was removed.

0.060 moles of morpholine and a light excess of sodium hydroxide (0.070 moles)

- 25 -

in 30% water solution were added to the so obtained xilene solution. The reaction mixture has been slowly heated, up to 110°C, temperature that has been maintained for 1 h. 100 g of water were added to the reaction mixture, once it was cooled at 90°C; after stirring for 30 minutes at 80-90°C the water phase was removed.

5 Thus a xilene solution was obtained that, after filtration for removal of possible undissolved particles, was brought to dryness by distillation under vacuum of the solvent, thus obtaining by cooling the melt 77.6 g of solid product (HALS 1).

Mn: 4410 g/mol (corresponding to an index $p=3.30$)

MW: 8539 g/mol

10 Polydispersity (MW/Mn): 1.94

The GPC analysis shows a chromatogram as shown in Fig. 1

Example 2

Preparation example

A polymeric HALS having structure (I) with $p=3-20$, $A=-(CH_2)_a-$ wherein $a=2$,
15 Z =formula (III) group, $n=3$, X and X_1 = formula (II) group with R_2 = butyl, R and R_1 = H and Y = group of formula (VI) wherein X and R are as defined, has been prepared with the following described procedure.

0.24 moles of N-(2,2,6,6-tetramethyl-4-piperidiny)-butylamine were added in 30 minutes, under stirring at room temperature, to a solution obtained by dissolving 0.12
20 moles of cyanuric chloride in 170 g of xilene. Then, 40 g of water and 0.27 moles of sodium hydroxide in form of 30% water solution were added and slow heating up to 95°C was applied, this temperature was maintained for two hours, and the water phase was then removed. The obtained xilene solution, containing 0.12 moles of a compound of general formula (IX) wherein R and R_1 = H and X and X_1 being both the group of
25 formula (II) wherein R_2 = butyl, was additioned with 0.06 moles of N,N'-bis(aminopropyl)-etilendiamine, corresponding to an amine of general formula (XIII) wherein $n=3$ and $A=-(CH_2)_a-$ wherein $a=2$ and the acidity was neutralized with a light alkali excess. The mixture was gradually heated up to the full reflux temperature, by removing reaction water by distillation, and then maintained at the same temperature for
30 three hours. The mixture was allowed to cool at 90°C, and 100 g of water were added; after stirring for 30' at 80-90°C the water phase was removed.

At the xilene solution containing 0.06 moles of an amine of general formula (VII) wherein Z is the group of formula (III) and n, A, R, R₁, X, X₁ are as above defined, a mixture has been added that was obtained by dissolving 0.054 of cyanuric chloride in 100 g of xilene, under stirring and by maintaining the temperature below 30°C. Once the addition was completed, the mixture was kept under stirring for one hour and then slowly heated up to 60°C. At this temperature a slight excess (0.113 moles) of sodium hydroxide in 30% water solution was added and the mixture was slightly heated up to 90°C, at which temperature stirring was maintained for 1 h. Then, 100 g of water were added and stirring was maintained for 30' and then the water phase was removed.

0.054 moles of N-(2,2,6,6-tetramethyl-4-piperidiny)-butylamine and a light excess of sodium hydroxide (0.063 moles) in 30% water solution were added to the obtained xilene solution. The reaction mixture has been slowly heated, up to 110°C, temperature that has been maintained for 1 h. 100 g of water were added to the reaction mixture, once it was cooled at 90°C; after stirring for 30 minutes at 80-90°C the water phase was removed.

Thus a xilene solution was obtained that, after filtration for removal of possible undissolved particles, was brought to dryness by distillation under vacuum of the solvent, thus obtaining by cooling the melt 82.7 g of solid product (HALS 2).

Mn: 5597 g/mol (corresponding to an index p=3.83)

MW: 13159 g/mol

Polydispersity (MW/Mn): 2.35

The GPC analysis shows a chromatogram as shown in Fig. 2.

Example 3

Comparative preparation example

A oligomeric HALS of structure (I) with $p < 3$, $A = -(CH_2)_a-$ wherein $a = 2$, Z=formula (III) group, $n=3$, X and X₁ = formula (II) group with R₂ = butyl, R and R₁ = H and Y = morpholine residue (HALS 3), has been prepared by adapting the procedure given in Example 1 of US 4,477,615, using the suitable amine (XIII) with $a = 2$ and $n = 3$ instead of diethylenetriamine and 2,4-dichloro-6-morpholino-1,3,5-triazine instead of 2,4-dichloro-6-[(2',2',6',6'-tetramethyl-piperidin-4'-yl)-butylamino]-1,3,5-triazine.

Mn: 2586 g/mol (corresponding to an index p=1.94)

- 27 -

MW:3553 g/mol

Polydispersity (MW/Mn): 1.37

The GPC analysis shows a chromatogram as shown in Fig. 3.

Example 4

5

Comparative preparation example

A oligomeric HALS of structure (I) with $p < 3$, $A = -(CH_2)_a-$ wherein $a = 2$, $Z =$ formula (III) group, $n = 3$, X and $X_1 =$ formula (II) group with $R_2 =$ butyl, R and $R_1 =$ H and $Y =$ group of formula (VI) wherein X and R are as defined (HALS 4), has been prepared by adapting the procedure given in Example 1 of US 4,477,615, using the suitable amine (XIII) with $a = 2$ and $n = 3$ instead of diethylenetriamine.

10

Mn: 2553 g/mol (corresponding to an index $p = 1.94$)

MW:3342 g/mol

Polydispersity (MW/Mn): 1.31

The GPC analysis shows a chromatogram as shown in Fig. 4.

15

In the following applicative examples, the names HALS 1, HALS 2, HALS 3 and HALS 4 indicate the compounds prepared according to the preparation examples 1, 2, 3 and 4 respectively.

20

The names HALS 5 and HALS 6 indicate compounds that were prepared by repeating the procedure described in preparation example 2 of US2010/0204371A1 and in preparation example 2 of US4477615, respectively.

Applicative examples

Applicative example 1

25

1000 parts by weight of powder unstabilized polypropylene homopolymer (fluidity index: about 10-12 g/10' at 230°C-216 kPa) have been mixed in a laboratory mixer with 1.0 parts by weight of calcium stearate, 0.50 parts by weight of tris-(2,4-di-tert-butyl-phenyl)phosphite, 0.50 parts by weight of 1,3,5-tris-(3,5-di-ter-butyl-4-hydroxybenzyl)isocyanurate and 1.5 parts by weight of one of the stabilizers HALS 1, 3 and 4, respectively.

30

The so obtained dry mixture was extruded in a laboratory extruder at 230°C thus obtaining a granulated compound.

The granulate was then transformed into a film, having a thickness of 100 μm , by means of a compression molding laboratory press at 210°C.

Samples obtained from the different films were then exposed in Weather-Ometer Ci35A (T black panel 63 $\pm$ 2 °C, dry cycle). The samples were periodically taken and subjected to a Carbonyl index control, by means of FT-IR spectrophotometric technique. The growth of the carbonyl index value provides an indication about the degree of photo oxidative degradation of the material and may be correlated with the losing of mechanical properties following the material photooxidation.

The parameter used in order to compare the light resistance of the samples was the $t_{0.10}$ defined as the “time of WOM exposure, in hours, when a Carbonyl index value of 0.10 was reached”.

The obtained experimental results are summarized in Table 1.

TABLE 1. Light stability of a PP film of 100 μm

<i>Stabilization</i>	<i>t_{0.10}</i>
Without stabilizer	650
0.15% HALS 3	3200
0.15% HALS 4	3350
0.15% HALS 1	3400

Applicative example 2

The dry mixture obtained as described in applicative example 1, but using in this second example 1.5 parts by weight of one of the stabilizers HALS 1, HALS 2 and HALS 4, respectively, was extruded in a laboratory extruder at 230°C thus obtaining a granulated compound.

The granulate was then transformed into a film, having a thickness of 100 μm , by means of a compression molding laboratory press at 210°C.

Samples obtained from the different films were then exposed in Weather-Ometer Ci35A (T black panel 63 $\pm$ 2 °C, dry cycle). The samples were periodically taken and subjected to a Carbonyl index control, by means of FT-IR spectrophotometric technique. The growth of the carbonyl index value provides an indication about the degree of photo oxidative degradation of the material and may be correlated with the losing of mechanical properties following the material photooxidation.

The parameter used in order to compare the light resistance of the samples was the $t_{0.10}$ defined as the “time of WOM exposure, in hours, when a Carbonyl index value of 0.10 was reached”.

The obtained experimental results are summarized in Table 2.

5 **TABLE 2. Light stability of a PP film of 100 μ m**

<i>Stabilization</i>	<i>$t_{0.10}$</i>
Without stabilizer	650
0.15% HALS 4	3350
0.15% HALS 1	3400
0.15% HALS 2	3700

Applicative example 3

1000 parts by weight of unstabilized polyamide-6 were mixed in a laboratory
 10 mixer with 0.25 parts by weight of tris-(2,4-di-tert-butyl-phenyl)phosphate, 0.50 parts by weight of 1,3,5-tris-(3,5-di-ter-butyl-4-hydroxybenzyl)isocyanurate and 5 parts by weight of one of the stabilizers HALS 1, HALS 2, HALS 5 and HALS 6, respectively.

The dry mixture was extruded in a laboratory extruder at 260°C and granulated after cooling the extrudate.

15 The granulate was then transformed, by spinning, into a bulked filament of count 270/60 (4.5 dtex/bulk), by means of a laboratory extruder at 270°C.

The filament was exposed in Weather-Ometer Ci65 (T black panel 63±2 °C, dry cycle). Light resistance was monitored by periodically taking samples and subjecting to evaluation of tensile strength by checking the load at break.

20 The parameter used in order to compare the light resistance of the samples was the t_{50} defined as the “time of WOM exposure, in hours, when a load at break of 50% of the initial value was reached”.

The obtained experimental results are summarized in Table 3.

Table 3

<i>Stabilization</i>	<i>t_{50} WOM hours</i>
Without stabilizer	750
0.50% HALS 1	3200
0.50% HALS 2	3080
0.50% HALS 5	2950

- 30 -

0.50% HALS 6	3000
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Applicative example 4

1000 parts by weight of unstabilized ABS (acrylonitrile-butadiene-styrene copolymer) were mixed in a laboratory mixer with 3 parts by weight of one of the stabilizers HALS 1, HALS 2, HALS 5 and HALS 6, respectively.

The dry mixture was extruded in a laboratory extruder at 250°C and granulated after cooling the extrudate.

The granulate was then transformed into plates having 2mm thickness, by die casting at 260°C.

The plates were exposed in Weather-Ometer Ci35A according to SAEJ1885.

Light resistance was monitored by periodically taking the plates and subjecting them to evaluation of the color.

The parameter used in order to compare the light resistance of the plates was the Yellowness Index (E313), A lower value of Yellowness Index indicates a lower yellowing of the formulation exposed in WOM and therefore a better resistance of the formulation to degradation.

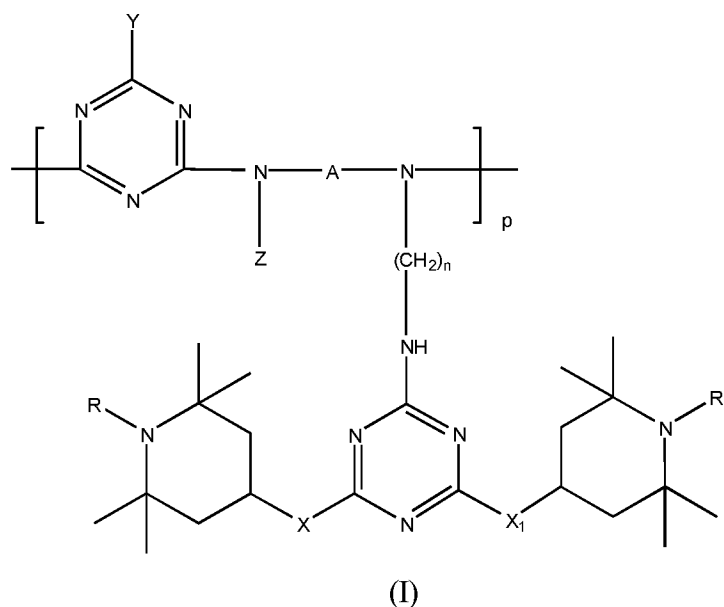
The obtained experimental results are summarized in Table 4.

Table 4

	<i>Yellowing Index (E313) vs KJ/m2 at 340 nm by WOM Ci35A exposure</i>			
<i>Stabilization</i>	<i>Initial</i>	<i>112 KJ</i>	<i>224 KJ</i>	<i>448 KJ</i>
Without stabilizer	17	19	27	40
0.30% HALS 1	14	9.5	13	26
0.30% HALS 2	15	10.5	15	30
0.30% HALS 5	15.5	11	16	32
0.30% HALS 6	15.5	11	15.7	31

CLAIMS

1. Stabilizing polymeric polypiperidine compounds of general formula (I),

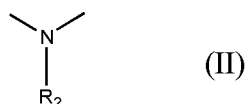


5 wherein **p** is from 3 to 20;

n is from 2 to 12;

R and **R₁**, which are the same or different, are selected in the group consisting of hydrogen, linear and branched C₁-C₁₂ alkyl groups, alkenyl groups having 3 to 8 carbon atoms and aralkyl groups having 7 to 19 carbon atoms;

10 **X** and **X₁**, which are the same or different, are selected in the group consisting of oxygen and groups of formula (II)

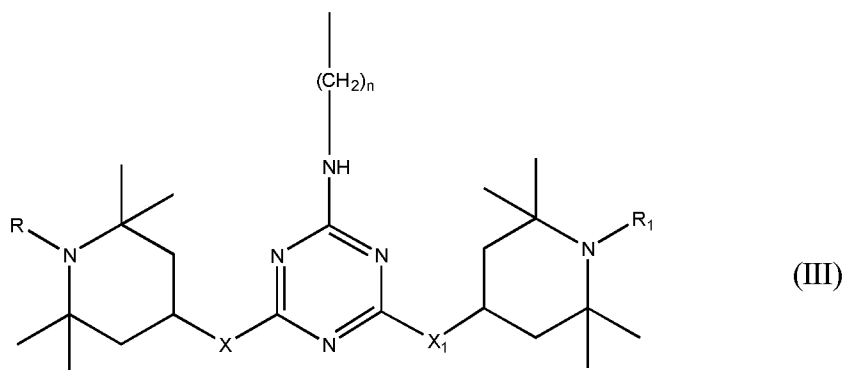


15 wherein **R₂** is selected in the group consisting of hydrogen, linear and branched C₁-C₁₂ alkyl groups, cycloalkyl groups having 5 to 12 carbon atoms and aralkyl groups having 7 to 12 carbon atoms;

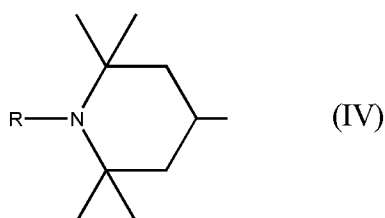
A represents a -(CH₂)_a- group wherein **a** is from 2 to 12, with the proviso that **a** is different from **n**;

Z is selected in the group consisting of C₁-C₁₈ alkyl groups, groups of formula (III)

- 32 -

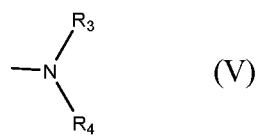


wherein **n**, **X**, **X₁**, **R** and **R₁** are as above defined,
and groups of formula (IV)



5 wherein **R** is as above defined;

Y represents a substituent selected in the group consisting of the groups of general formula (V)

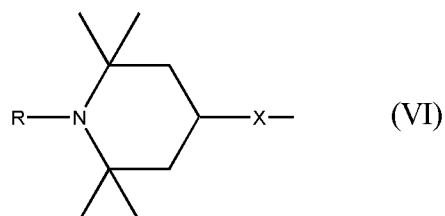


the groups **O-R₄** and **S-R₄**,

10 wherein **R₃** and **R₄**, which may be the same or different, are selected in the
group consisting of hydrogen, linear and branched C₁-C₁₈ alkyl groups,
cycloalkyl groups having 5 to 12 carbon atoms, aralkyl groups having 7 to
12 carbon atoms, aryl groups having 6 to 12 carbon atoms or may form,
together with the nitrogen atom to which they are linked, a heterocycle
15 containing 5 to 7 carbon atoms;

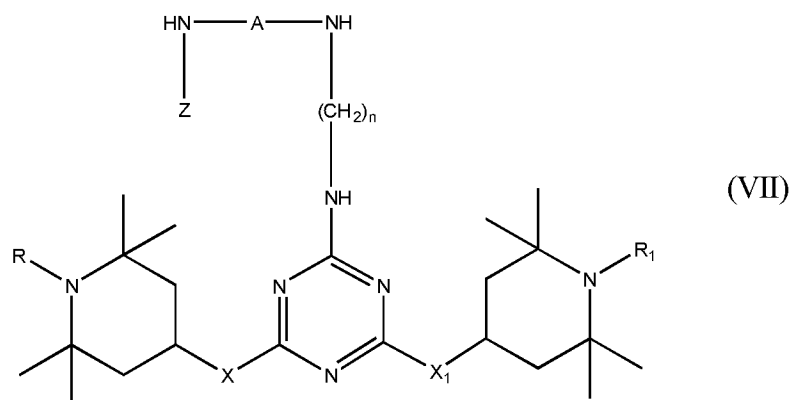
and the piperidine group (VI)

- 33 -



wherein R and X are as above defined.

2. Compounds according to claim 1, wherein p is from 3 to 8.
3. Compounds according to claim 2, wherein p is from 3 to 6.
- 5 4. Compounds according to claim 1 wherein R and R₁ are the same and represent hydrogen.
5. Compounds according to any one of the preceding claims, wherein n = 3 and a = 2.
6. Compounds according to any one of the preceding claims, wherein Z is
- 10 the group of formula (III).
7. Compound according to any one of the preceding claims, wherein X and X₁ are the same and correspond to the group of formula (II) with R₂ being n-butyl.
8. Compounds according to any one of the preceding claims, wherein Y is a morpholine group.
- 15 9. Compounds according to any one of the claims 1 to 7, wherein Y is the group of formula (VI).
10. A process for the preparation of the compounds according to claim 1, comprising:
 - a first reaction between cyanuric chloride and an intermediate of general
 - 20 formula (VII)

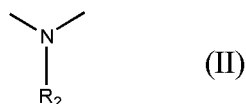


- 34 -

wherein n is between 2 and 12;

R and **R₁**, which are the same or different, are selected in the group consisting of hydrogen, linear and branched C₁-C₁₂ alkyl groups, alkenyl groups having 3 to 8 carbon atoms and aralkyl groups having 7 to 19 carbon atoms;

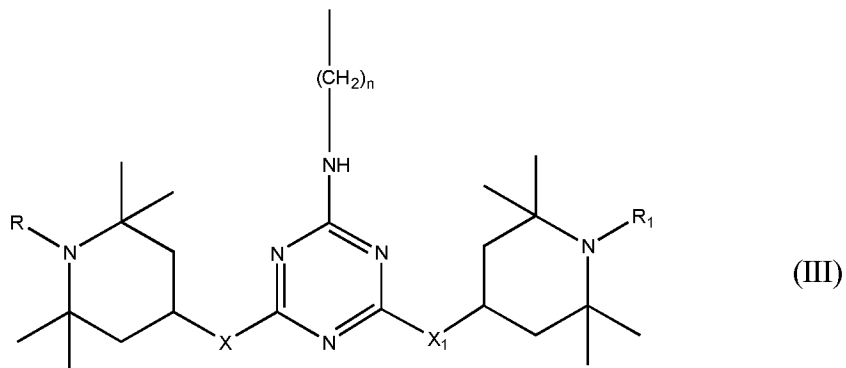
5 **X** and **X₁**, which are the same or different, are selected in the group consisting of oxygen and groups of formula (II)



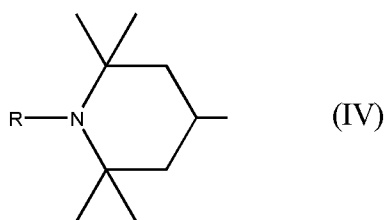
10 wherein **R₂** is selected in the group consisting of hydrogen, linear and branched C₁-C₁₂ alkyl groups, cycloalkyl groups having 5 to 12 carbon atoms and aralkyl groups having 7 to 12 carbon atoms;

A represents a $-(CH_2)_a-$ group wherein **a** is from 2 to 12, with the proviso that **a** is different from **n**;

Z is selected in the group consisting of C₁-C₁₈ alkyl groups, groups of formula (III)



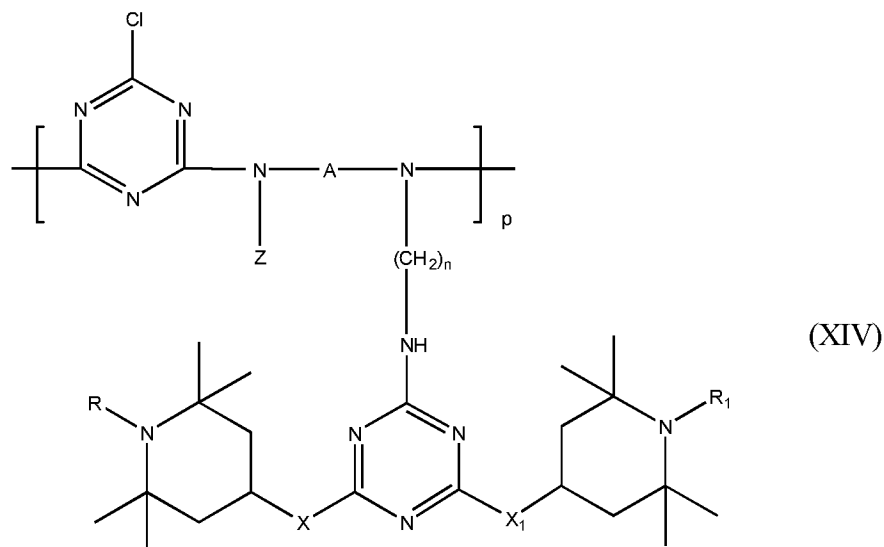
15 wherein **n**, **X**, **X₁**, **R** and **R₁** are as above defined, and groups of formula (IV)



wherein **R** is as above defined;

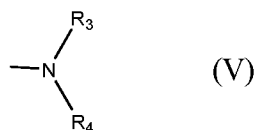
20 with formation of a compound of formula (XIV)

- 35 -



wherein p, Z, A, R, R₁, X, X₁ and n have the above defined meanings; and

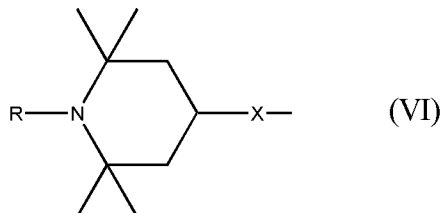
- a second reaction of the compound of formula (XIV) with the suitable group H-Y, wherein Y represents a substituent selected among the groups of general formula (V),



the groups O-R₄ and S-R₄,

wherein **R₃ and R₄**, which may be the same or different, represent hydrogen, a linear and branched C₁-C₁₈ alkyl group, a cycloalkyl group having 5 to 12 carbon atoms, an aralkyl group having 7 to 12 carbon atoms, an aryl group having 6 to 12 carbon atoms or may form, together with the nitrogen atom to which they are linked, a heterocycle containing 5 to 7 carbon atoms;

and the piperidine group (VI)



wherein R and X are as above defined.

11. A process according to the preceding claim, characterized in that said

first reaction is carried out in the presence of an acidity corrector.

12. A process according to the preceding claim, characterized in that a water solution of sodium hydroxide or potassium hydroxide at a concentration of between 20% and 50% by weight, preferably between 25% and 50% by weight is used as acidity
5 corrector.

13. A process according to claim 10, characterized in that said first reaction takes place in a range of temperature of between 5°C and 140°C, preferably between 10°C and 130°C and more preferably of between 25°C and 120°C.

14. A process according to the preceding claim, characterized in that said
10 first reaction comprises the following steps:

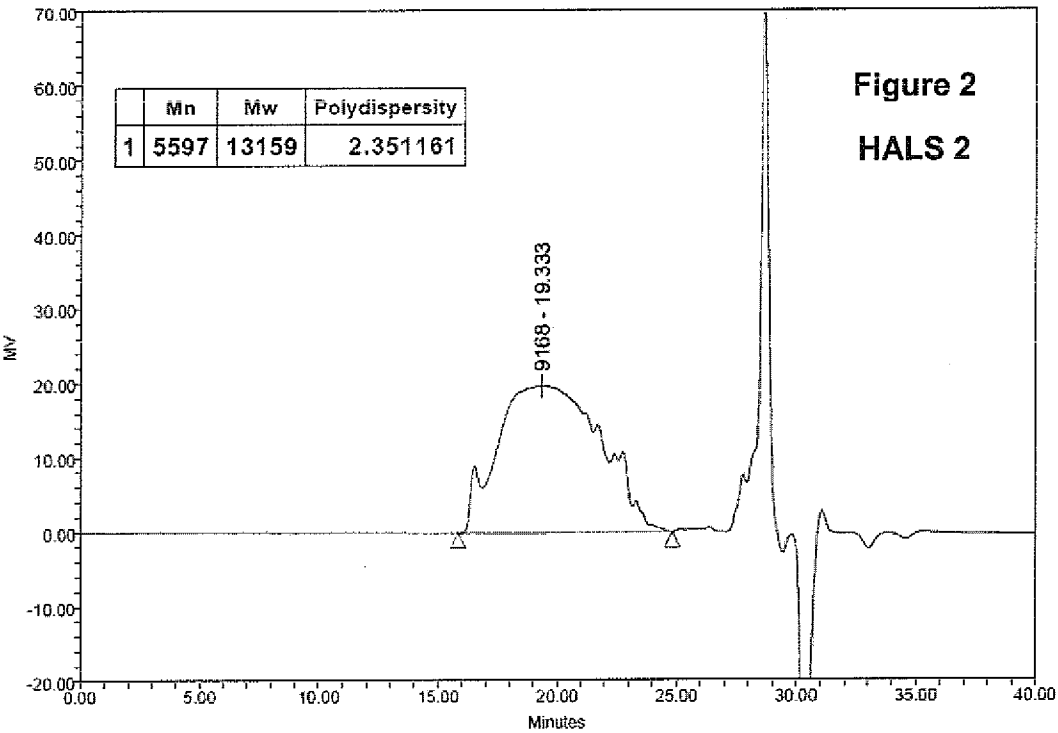
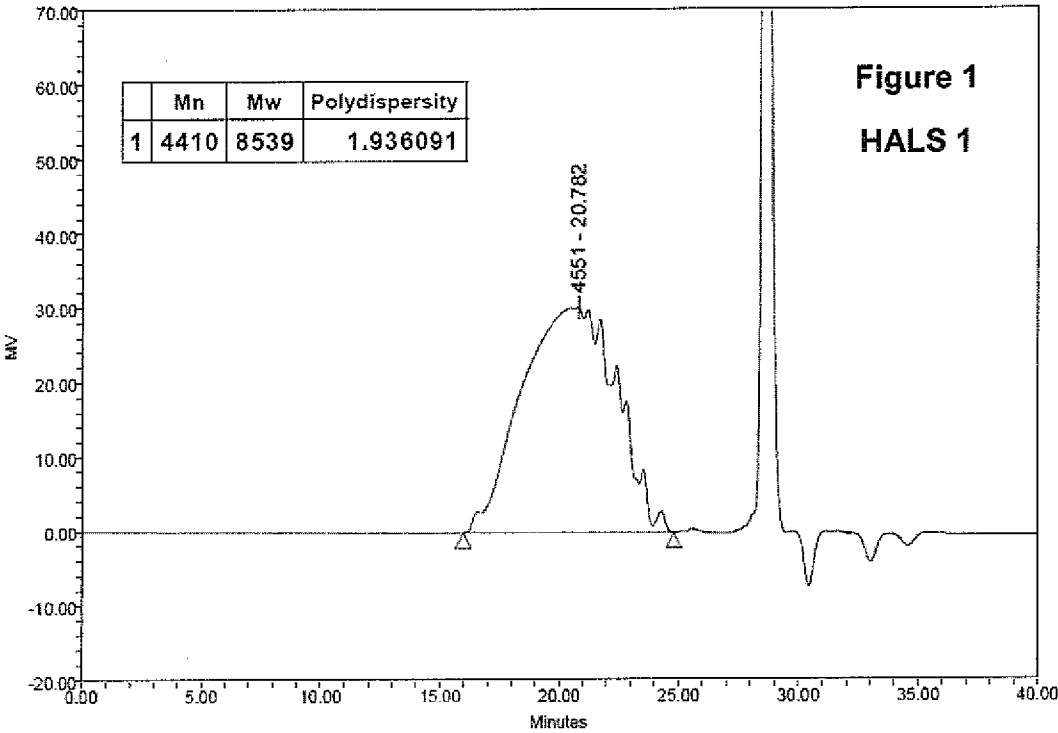
- addition of a solution of cyanuric chloride to a solution of the compound of formula (VII);
- mixing at a temperature of between 5°C and 60°C, preferably between 20°C and 35°C;
- 15 - adding a acidity corrector; and
- progressively heating at a temperature in the range of between 80°C and 120°C, preferably between 90°C and 110°C.

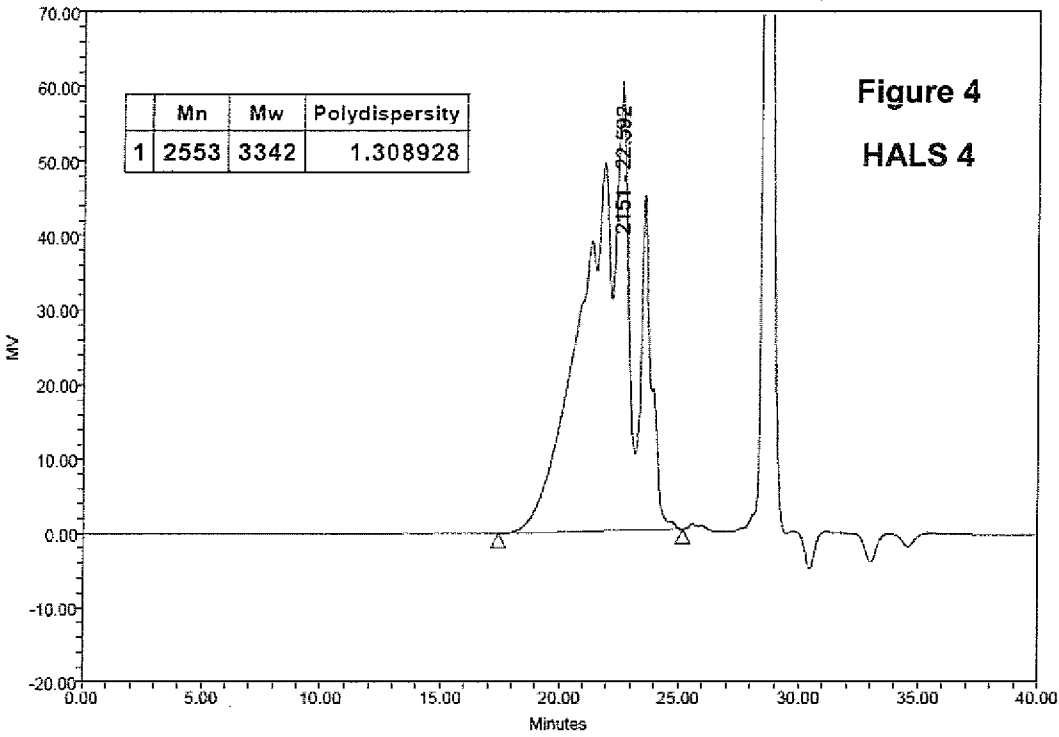
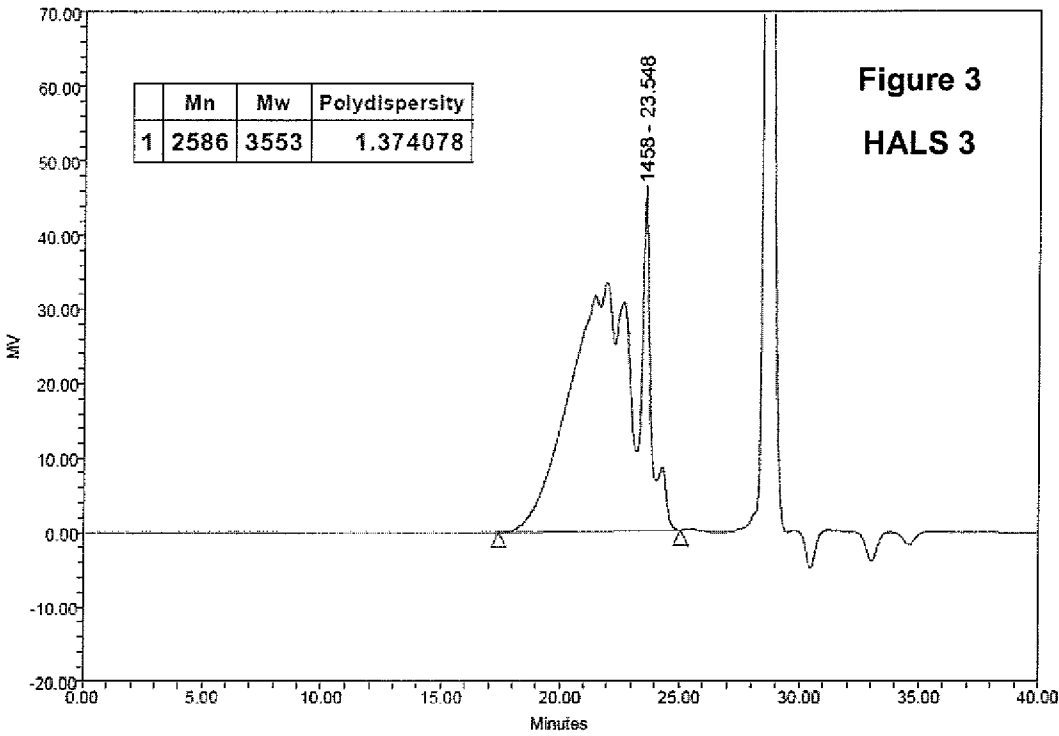
15. Composition comprising a polymeric material and compounds according to claim 1, characterized in that it comprises from 0.01% to 5% by weight of one or
20 more compounds according to claim 1.

16. Composition according to claim 15, wherein the polymeric material is selected in the group of polyethylene, polypropylene, polystyrene, polybutadiene, polyisoprene, and copolymers thereof, polyvinylchloride, polyvinylidene chloride and copolymers thereof, polyvinylacetate and copolymers thereof, in particular with
25 ethylene, polyesters such as polyethyleneterephthalate; polyamides such as nylon 6 and nylon 6,6, and copolymers thereof, acrylonitrile-butadiene-styrene copolymers (ABS), polyurethans.

17. Composition according to claim 15, wherein the polymeric material is a polyolefin.

30 18. Composition according to claim 15, also comprising other monomeric, polymeric or macromolecular HALS.





INTERNATIONAL SEARCH REPORT

International application No

PCT/EP2011/068867

A. CLASSIFICATION OF SUBJECT MATTER

INV. C07D401/14 C08K5/3492 C09K15/30
ADD.

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

C07D C08K C09K

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

EPO-Internal, WPI Data

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	US 4 477 615 A (RASPANTI GIUSEPPE [IT] ET AL) 16 October 1984 (1984-10-16) cited in the application example 6	1-18
X	----- EP 2 216 364 A1 (3V SIGMA SPA [IT]) 11 August 2010 (2010-08-11) page 3; examples	1-18
X	----- EP 1 780 237 A2 (3V SIGMA SPA [IT]) 2 May 2007 (2007-05-02) page 3	1-18
X	----- EP 2 112 198 A1 (3V SIGMA SPA [IT]) 28 October 2009 (2009-10-28) * see formula A *	1-18



Further documents are listed in the continuation of Box C.



See patent family annex.

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Date of the actual completion of the international search

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European Patent Office, P.B. 5818 Patentlaan 2
NL - 2280 HV Rijswijk
Tel. (+31-70) 340-2040,
Fax: (+31-70) 340-3016

Authorized officer

Lauro, Paola

INTERNATIONAL SEARCH REPORT

Information on patent family members

International application No

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