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INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT) (51) International Patent Classification 5: WO 92/17639 (11) International Publication Number: **A1** D21C 9/14, 9/153 (43) International Publication Date: 15 October 1992 (15.10.92) PCT/US92/02533 (21) International Application Number: (81) Designated States: AT (European patent), BE (European patent), BR, CA, CH (European patent), DE (European (22) International Filing Date: 26 March 1992 (26.03.92) patent), DK (European patent), ES (European patent), FI, FR (European patent), GB (European patent), GR (European patent), IT (European patent), JP, LU (European patent), MC (European patent), NL (European pa-(30) Priority data: 8 April 1991 (08.04.91) 682,728 US tent), SE (European patent). (71) Applicant: INTERNATIONAL PAPER COMPANY [US/ Published US]; Corporate Research Center, Long Meadow Road, With international search report. Tuxedo, NY 10987 (US). (72) Inventors: BHATTACHARJEE, Shyam, S.; 22 Chatham Road, Monroe, NY 10950 (US). RENARD, Jean, J.; 1405 Oak Tree Court, Mobile, AL 36608 (US). (74) Agents: HODGES, Paul, E. et al.; Luedeka, Hodges, Neely & Graham, 1030 First American Center, Knoxville, TN 37902 (US). (54) Title: METHOD FOR REDUCING COLORED MATTER FROM BLEACH EFFLUENT USING A DZD BLEACH **SEQUENCE** (57) Abstract A process for bleaching cellulosic pulp to a G.E. brightness of at least about 75 and a viscosity of at least about 14 wherein the effluent from the the bleaching process is of reduced colored matter content and exhibits reduced COD. The process comprises DZD stages without interstage treatment of the pulp other than by an optional water wash.

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"METHOD FOR REDUCING COLORED MATTER FROM BLEACH EFFLUENT USING A DZD BLEACH SEQUENCE"

This invention relates to the bleaching of cellulosic pulps, and particularly kraft pulps, and to the effluent from such bleaching activities.

In the bleaching of cellulosic pulp, it is generally accepted that the use of chlorine, chlorine dioxide or a combination of these two compounds effects both chlorination and oxidation of the components of the pulp. Further, it is generally recognized that cellulosic pulps contain not only cellulose, but also many other components such as lignin, resins, pentosans, other miscellaneous organic matter, and mineral substances. Thus, when either chlorine and/or chlorine dioxide is used as the bleaching agent in the first stage of a given bleaching sequence, e.g. the well-known CEDED sequence, there occurs a large variety of chemical reactions that are in addition to the desired chlorination of lignin, the latter being that reaction which renders the lignins soluble in alkali so that they can be removed from the pulp in a subsequent stage of the bleaching process. Many of these chemical reactions result in the production of colored compounds, i.e. chromophores, that must be removed from the pulp if one is to obtain a non-colored paper product from the pulp. Oxidation reactions by the chlorine-containing bleaching contribute to such colored compounds, and the substitution of chlorine dioxide for the elemental chlorine has been noted to reduce the extent of such oxidation reactions. For this and other reasons, chlorine dioxide is preferred as the bleaching agent in most modern chlorine-based bleaching sequences.

Removal of the chlorinated lignins which are soluble in alkali solution is one of the primary functions of the extraction (E) stage in a cellulosic pulp bleaching sequence. Commonly, the extraction is effected at a temperature of about 60° to 70° C and employing NaOH as the alkali. Sufficient alkali is employed to develop a pH of

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the pulp of about 11. Under these reaction conditions, not only are the chlorinated lignins brought into solution, but also substantial quantities of the other compounds present in the pulp also are brought into the solution. to >80% of the total color in the pulp, >30% of the BOD (Biological Oxygen Demand), >50% of the COD (Chemical Oxygen Demand), and >14% of the total chlorides in the pulp are extracted into the solution in the extraction stage. Thus, the effluent from the extraction stage is found to contain these compounds. Still further, recent studies found that certain of the chlorine-containing compounds (or their precursors) formed in the initial chlorination stage and/or in the subsequent extraction stage are carcinogenic, toxic or otherwise harmful.

Under present and proposed environmental regulations by various governing agencies, the presence of various chlorine-containing compounds either in the pulp or the effluent from the bleaching process must be eliminated or limited to very small quantities. Especially, the dioxin class of chlorine-containing compounds has been targeted for such reduction. Further, these same agencies have developed standards for the BOD, COD and presence of colored matter in the effluent from bleaching and other processes employed in pulp and paper mills which must be met before the effluent can be discharged to environment. In an effort to meet such present and/or proposed guidelines, considerable effort has been and is being expended to develop bleaching processes which meet the guidelines. Unfortunately, it has been found that alternatives to the heretofore used bleaching sequences are economically expensive or in some instances may create problems with respect to the quantity or quality of the paper product produced from the pulp. oxygenation has been proposed as a substitute for chlorination, but oxygen tends to degrade the viscosity of the hence the paper produced from such pulp unacceptable strength values. Moreover, certain of the

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proposed modifications to conventional bleaching sequences have failed to produce the desired brightness of the pulp, which results in an unacceptable non-white paper product. Even in those instances where the brightness of the pulp appears to be at an acceptable level, it has been found that the brightness of paper produced from such pulps often reverts to lower and unacceptable brightness values upon the passage of time.

Because of the environmental demand of decreasing the amount of chlorinated organic compounds in bleached pulp and effluents, the current trend in bleaching is to reduce the chlorine charge substantially. It is predicted that chlorine consumption in the pulp and paper industry will fall rapidly, while caustic consumption will hold steady or rise. Consequently, the price of caustic soda is expected to rise rapidly. For this reason, there is a need for a caustic saving technology.

In accordance with the present invention, it has been found that pulp of good viscosity and brightness, and of lower undesirable chlorine-containing compounds (BOC1-Bound Organic Chloride) may be obtained by means of a bleaching sequence in which the initial stage comprises the use of chlorine dioxide (D) or a mixture of chlorine dioxide and chlorine and wherein this initial stage is followed by an ozonation (Z) stage and wherein the Z stage is followed by a further D stage (D_1) , all without an alkaline extraction stage between either of such stages. as the prior art teaches that the use of an alkaline extraction stage following the first chlorination stage is important in achieving a bleached pulp having acceptable viscosity and brightness values, it is surprising to find that the combination of three initial stages of DZD, without an intervening alkaline extraction stage, followed by ZD, ED, or a peroxide (P) stage, will produce an acceptable pulp. As described more fully hereinafter, selection of the operating parameters associated with such initial three stages of the sequence has been found to be

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of importance if one is to obtain the observed desirable Elimination of the alkaline extraction stage economic advantage associated with provides elimination of the NaOH and results in reduced COD and color in the effluent from the bleaching operation. Whereas it is not fully understood, it has been further noted that the effluent from the present bleaching sequence contains less colored matter. This unexpected result provides a significant advantage relating to the use and handling of such effluent, while also contributing to the ability to comply with applicable legal and/or regulatory agency guidelines relating to the disposal of effluent.

The present method is useful in processing both softwoods and hardwoods, and particularly kraft pulps of such woods. In the initial stage of the present bleaching process (sequence), the pulp at a pH of between about 2 and about 3 and at a consistency of between about 3% and about 10% is contacted with a quantity of chlorine dioxide (the preferred chlorination agent) (D stage) or chlorine dioxide with substitution of elemental chlorine for up to about 50% of the chlorine dioxide (D \rightarrow CD stage) in the course of carrying out such initial stage. In the preferred embodiment of such initial stage, the quantity of chlorination agent is calculated as:

Cl factor = $\frac{\text{% Cl}_2 + \text{% ClO}_2 \times 2.63}{\text{Kappa No.}}$

A chlorination factor of between about 0.1 and about 0.25 has been found to be effective in accomplishing the objectives of the present invention. Chlorination factors of less than about 0.1 fail to provide sufficient available chlorine to accomplish the desired degree of chlorination of the pulp and chlorination factors greater than about 0.25 tend to produce undesirable quantities of dioxin-type compounds, bound organic chlorides, and other objectionable compounds in the pulp. In the preferred process, elemental

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chlorine is used limitedly inasmuch as elemental chlorine is believed to promote the production of dioxins, among other things. For present purposes, however, when reference is made to the D stage, it is to be understood that no substantial elemental chlorine is employed. On the other hand, it is permissible in the present process to employ an initial chlorination stage in which up to about 50% elemental chlorine is substituted for chlorine dioxide, i.e. a D \rightarrow C_D stage.

The pulp, in the first step, i.e. the D or the D \rightarrow C_D stage, of the present bleaching process, is at a consistency of between about 3% and about 10%, based on OD pulp. Within such consistency range, it has been found that there is obtained optimum contact, hence reactivity, between the active chlorine and the cellulosic pulp. Further, at such consistencies and employing the chlorination factors referred to above, the reaction time during such initial stage need not exceed about 45 minutes at a pulp temperature of about 50°C .

Following the first step, the pulp, at a consistency of between about 1% and 30% based on OD pulp, a pH of between about 2 and about 5, and without any intervening treatment other than an optional water wash, is subjected to a second stage including contacting the pulp with between about 0.1% and 1% ozone (Z stage). In one embodiment, 100% oxygen is passed through a Welsbach Ozone Generator (Model No. T408) which converts approximately 1.5 to 3.0% of the oxygen to ozone. This oxygen/ozone mixture is bubbled into the bottom of a reactor containing the pulp through an inlet port at the rate of about 2 liters/min and a pressure of about 6 psig. An outlet port at the top of the reactor permits exit of the oxygen/ozone mixture after its passage through the pulp slurry in the reactor. Analysis of the ozone concentration at the inlet and outlet ports provides a measure of the ozone (based on dry weight pulp) consumed by reaction during the residence time of the gas mixture within the reactor. The pH of the pulp at the

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completion of this second step is normally less than about 3.

In the third step of the present bleaching process, the pulp, after ozonation in the second step and without an intervening treatment, other than an optional water wash, at a consistency of about 10% is contacted with chlorine dioxide at a temperature of between about 60°C and about 70°C for about 1-3 hours (D stage). Longer reaction times (3h) have not been found necessary. During this third step, sufficient alkali such as sodium hydroxide or sodium carbonate is added to the pulp such that the pH of the pulp at the end of the treatment period is between about 2.5 and 3.5. Within this pH range, it has been found that the effectiveness of this stage of the process is optimized.

The results of bleaching both softwood and hardwood kraft pulps in accordance with the present process are given in TABLE I. In such TABLE, there are also presented the results of the bleaching of the same pulp employing DED and DEDED bleaching sequences. From such TABLE, it will be noted that the present process (DZD, DZDED sequences) produced bleached pulps having a brightness essentially the same as the brightness of pulps produced by the conventional sequences DED, DEDED and with no significantly lesser viscosity that such conventionally bleached pulps. Further, the present process permitted elimination of up to 5.8 lb of NaOH for each lb of ozone employed, resulting in a substantial caustic savings by reason of the use of the present process.

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		NaOH (1b) Replaced by 1 1b of Ozone	4.6	4.6	 	2.5
		Pulp Ozone	0.25	0.25	0.75 0.5	0.75
		s on OD	1.8 0.65	2.30	3.98 1.1 1.1	2.78
	Sequences	\$ Chemicals on OD Pulp Total Active Chlorine NaOH Ozo	4.94	5.46 5.46	11.0 11.0 11.0	8.07
TABLE I	Comparison of Bleaching Sequences	perties Viscosity	20.0	18.8 18.2	20.2 18.4 0.5	20.5 19.3
	omparison of	Pulp Properties Brightness Viscos % GE	80.6	88.5	88.5 88.6 87.5	88.9
	Ö	Bleaching Sequence	DED DZD	DEDED DZDED	DEDED DZDED DZDED	DEDED
		Pulp	. Northern Hardwood Kraft	(Visc., 29.0 cP)	Southern Pine Kraft (Kappa No. 31.1) (Visc., 29.6 cP)	Southern Pine Kraft (Kappa No. 18) (Visc., 30 cp)

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Additionally, in TABLE II, there is presented the results of several examples of the bleaching of southern softwood kraft pulp of a Kappa No. of 31.4 and a viscosity of 30.1 cP, employing the present process. In such TABLE II, it is to be noted that the initial stage is identified as $D \rightarrow C_{p}$, but that at the noted chlorination factors compared, there are examples where there was 100% substitution of chlorine dioxide for the elemental chlorine, so that such examples represent a D stage wherein no elemental chlorine was used, i.e. a DZDED sequence. From TABLE II, it may be seen that below 50% substitution of chlorine dioxide for elemental chlorine results is a reduction in pulp brightness, and at all but one of the listed chlorination factors, the viscosity of the pulp was adversely affected by the greater amount of elemental chlorine. Further, at all tested chlorination factors, the bound organic chlorine (BOCl) on the pulp was greater than 200 ppm except in those examples where only chlorine dioxide was employed in the initial stage, with the sole exception of the example where the chlorination factor was at 0.1 which is believed to be borderline as regards the ability of the process to produce pulp of an acceptable brightness.

TABLE II

D+ColdED Bleaching of Southern Pine Kraft (Kappa No. 31.4; Viscosity, 30.1 cP)

Filtrate AOX (kg/ton pulp)	6.1	6.0	2.1	8	4.6	2.6	4.0	3.3	1.2	1.3	1.1	1.1
Pulp Bocl (ppm)	248	205	128	288	268	140	268	215	105	220	173	66
Viscosity CP	16.0	16.5	20.0	17.2	17.1	18.7	16.2	15.6	17.5	14.5	15.1	17.1
Brightness % GE	87.2	88.8	88.5	84.0	86.6	87.1	83.7	86.6	87.6	81.8	84.8	84.9
Z Stage % Ozone	0.25	0.25	0.25	0.5	0.5	0.5	0.75	0.75	0.75	1.0	1.0	1.0
% Total <u>Active Chlorine</u>	7.85	7.85	7.75	6.28	6.28	6.28	4.71	4.71	4.71	3.14	3.14	3.14
Stage % ClO ₂ Sub	20	50	100	20	20	100	20	20	100	20	20	100
D+C Stage C1 Factor % C1O ₂	0.25	0.25	0.25	0.20	0.20	0.20	0.15	0.15	0.15	0.1	0.1	0.1

% Total Active Chlorine = Kappa No. x Cl Factor = % Cl₂ + % Clo₂ x 2.63

D, Stage: 1.2% ClO₂, 0.8% NaOH, 70°C, 3h

E Stage: 0.5% NaOH, 70°C, 1h

D₂ Stage: 0.3% ClO₂, 70° C, 3h

^{*} Conditions for DED Stages

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Further comparison of the present process with the conventional DEDED bleaching sequence is given in TABLE III. In TABLE III the DZD sequence described hereinabove was followed by either a further ZD or a D stage. As shown in the TABLE, either of these sequences produced a pulp having a brightness substantially equal to or greater than the brightness of the pulp bleached employing the conventional DEDED sequence. The viscosity of the pulps produced employing the present process with the added ZD or D stages was not materially reduced relative to the viscosity of the pulp produced by the DEDED process, and in any event such viscosity was well above the generally accepted minimum viscosity of about 14 for producing most papers from such pulps.

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TABLE III

Non-alkaline Bleaching of Southern Pine Kraft Pulp

(Kappa No. 31)

5		DZD	DZDZD 		DEDED
10	<u>lst Stage</u> : pH % ClO ₂	2.1 2.5	2.1 2.5	2.1 2.5	2.1 2.5
	<u>2nd Stage</u> : pH % Ozone % NaOH	2.5 0.75	2.5 0.75 	2.5 0.75	10.9 1.5
15	3rd Stage: pH % ClO ₂	2.7 1.2	2.1	3.3 1.2	3.0 1.2
`	% Na₂CO₃ NaOH	1.0		1.0	 0.48
20	4th Stage: pH % Ozone % NaoH	2.5 0.2	2.5		10.9 0.5
25	5th Stage: pH % ClO ₂	2.5 0.3	2.5	2.4 0.3	3.5 0.3
30	Pulp Properties: Brightness, % GE Reverted Brightness, %G Viscosity, cP	88.1 E 83.6 18.7	85.8 79.9 18.3	84.8 80.7 19.6	87.1 84.9 23.0

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Southern pine kraft pulp was bleached in accordance with the present DZD process followed by either ED, ZD or D stages, and by the conventional DEDED bleaching sequence. As shown in TABLE IV, among other things, the reverted brightness of the pulps produced in accordance with the present process (DZDED and DZDZD) is similar to the reverted brightness of the DEDED bleached pulp with the exception of the DZDD sequence. In the DZDD sequence, the reverted brightness shown is not deemed to be detrimentally low, however. Likewise, the strength properties of paper formed from the pulps bleached in accordance with the present process were found to be essentially equal to the same strength properties of paper formed from the pulp bleached in accordance with the conventional DEDED process.

From TABLE IV it is further seen that the effluent from the pulps bleached in accordance with the present process exhibited approximately 50% less colored And in like manner, such pulps exhibited a COD that was about 2/3 of the COD of the conventionally bleached pulp. Both these benefits were unexpected and the reason for the same is not known with certainty. of the reduction in colored matter and COD of the pulps bleached using the present process, it is possible to reuse the effluent within the bleaching process for a longer time before there occurs a buildup of colored matter or COD such that the effluent must be replaced, treated and discharged. Alternatively, the effluent from the present process may be treated for discharge as a waste stream at a substantially lesser cost than the effluent from the conventional pulp bleaching process.

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TABLE IV
Southern Pine Bleached Kraft Properties

		DEDED	DZDED	DZDZD	DZDD
	Brightness, % GE	85.7	87.7	86.5	85.1
5	Reverted Br., % GE	83.9	84.8	83.9	80.6
	Viscosity, cP	23.8	18.2	15.7	19.7
	TAPPI Handsheet Strength Properties (400 CSF)			•	
	B.L. (KM)	9.96	9.46	9.91	9.63
10	Burst Factor	79.1	79.1	79.5	81.0
	Tear Factor	97.3	91.9	97.1	94.4
	BOC1 (ppm)	210	178	193	224
	Effluent Properties: (kg/t	p)			
	Color	97.2	55.4	48.9	53.9
15	COD	84.3	60.9	57.0	54.2
	AOX	1.55	1.70	1.37	1.55
	<pre>% Chemicals on OD Pulp</pre>				
	Clo ₂	4.0	4.0	4.0	4.0
	03	0	0.75	0.95	0.75
20	NaOH	2.48	0	0	0
	Na_2CO_3		1.0	1.0	1.0

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As noted, the present bleaching sequence can be made completely free of alkaline extraction stage(s), thereby providing the noted savings in alkali. Such elimination of the alkaline extraction stage(s) in the bleaching sequence does not materially affect those properties of the pulp which make it acceptable in the formation of paper of a printing quality. Importantly, such elimination of the alkaline extraction stage(s) has been found to beneficially reduce the colored matter in the effluent from the bleaching process, to reduce the COD of such effluent. Further, it is possible to reduce the AOX of such effluent using the DZDZD sequence, and to reduce the BOCl of the pulp by both DZDZD and DZDED sequences.

As noted, the bleaching process of the present invention may be, and preferably is, carried out without employing a water wash (i.e. D, Z, D) of the process. Table V presents the results of bleaching softwood kraft pulp of Kappa No. 31.4 both with and without interstage water washing of the pulp. Whereas lack of such interstage water washes resulted in lower final brightness values of the pulp, such brightness values are acceptable for many uses of such pulp.

TABLE V

Three-Stage DZD Bleaching of Softwood Kraft Pulp

(Kappa No. 31.4)

		DZD_Bleaching						
5		With interstage <u>Washing</u>	Without interstage betwn. 1st D and Z Stage	Without in washing Z and 2nd	between			
		1	2	3	4			
10	1st Stage:							
	Cl Factor	0.23	0.23	0.23	0.23			
	% Clo ₂	2.75	2.75	2.75	2.75			
	2nd Stage:							
	% Ozone	0.75	0.75	0.75	0.75			
15	3rd Stage:							
	% ClO,	1.2	1.2	1.2	1.5			
	% NA ₂ ČO ₃	1.2	1.2	1.2	1.5			
	End pH	3.7	3.5	2.8	3.0			
	Pulp Properties:							
20	Brightness, % GE	81.3	75.9	79.7	84.8			
	Viscosity	21.2	21.9	20.9	21.3			

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IN THE CLAIMS:

Claim 1. A process for the bleaching of a cellulosic pulp to a target brightness of at least about G.E. 75 and a viscosity of at least about 14 cP comprising the sequential steps of:

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wash,

a second step including contacting said
pulp at a pH of less than about 5 at
the commencement of said second step
and a consistency of between about 0.1%
and about 30% at between about 30°C and
50°C with ozone, and thereafter, without an intervening treatment other than
an optional water wash,

a first step including contacting said pulp at a consistency of between about 2% and about 15% and a pH of between about 1.5 and about 2.5 at the commencement of said first step with a chlorination agent including at least 50% chlorine dioxide and of a chlorination factor of between about 0.1 and 0.25 for a time period in excess of about 45 minutes at a temperature of between about 30°C and 60°C, and thereafter, without an intervening treatment other than an optional water

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a third step including contacting said pulp at a consistency of about 10% with chlorine dioxide for a period of between about 1 and 3 hours and at a temperature of between about 60° and about 70°C, wherein the effluent from said bleaching process contains less colored matter and exhibits a lower COD than the effluent from a pulp bleaching process wherein there is employed an alkali extraction following an initial chlorination step.

Claim 2. The bleaching process of Claim 1 and including in said third step the addition to said pulp of sufficient alkali to provide a pH of the pulp of about 3 at the completion of said third step.

Claim 3. The bleaching process of Claim 1 wherein said first step comprises a D or D- C_D stage, said second step comprises a Z stage, said third step comprises a D or D- C_D stage, and including a fourth step comprising a stage selected from the group consisting of an E stage followed by a D stage, a peroxide stage, a Z stage, a Z stage followed by a D stage, and a D stage followed by a further D stage.

Claim 4. The bleaching process of Claim 1 in which the steps are carried out without a water wash between said first and second steps.

Claim 5. The bleaching process of Claim 1 in which the steps are carried out with no water wash between said second and third steps.

Claim 6. A method for reducing the colored matter in the effluent from a pulp bleaching process which employs an initial chlorination stage comprising subjecting the pulp to a bleaching sequence in which the first three stages comprise D, Z, D, in that order and without interstage treatment of the pulp by other than a water wash.

Claim 7. The method of Claim 4 wherein the pH of the pulp during the initial three stages of the bleaching sequence is maintained between about 2 and 4.

INTERNATIONAL SEARCH REPORT

International Application No. PCT/US92/02533

1. CLASSIFICATION OF SUBJECT MATTER (if several classification symbols apply, indicate ail) 6										
According to International Patent Classification (IPC) or to both National Classification and IPC IPC (5): D21C 9/14; D21C 9/153 U.S.C1.: 162/65,88,89										
II. FIELDS SEARCHED										
		Minimum Documer	ntation Searched 7							
Classification System Classification Sympols										
U.S.Cl. 162/65,88,89										
Documentation Searched other than Minimum Documentation to the Extent that such Documents are Included in the Fields Searched *										
III. DOCL	III. DOCUMENTS CONSIDERED TO SE RELEVANT									
Category *	Citat	ion of Document, 11 with indication, where app	ropriate, of the relevant passages 12	Relevant to Claim No. 13-						
Y	US, A	A, 4,959,124 (TSAI) 25 Sep column 3, lines 3-23.	tember 1990	1-7						
Y	US, A	A, 4,080,239 (KEMIF ET AL. column 4, lines 3-13.) 21 March 1978	1-7						
Y	69th Annual Meeting Technical Section Canadian Pulp and Paper Association, 01 February 1983, Liebergott et al., "Oxidative Bleeching - A Review", pages 169-183. See pages 175 and 176.									
A	US, A, 4,372,812 (PHILLIPS ET AL.) 08 February 1983 1-7 SEe column 3, line 58 to column 4, line 2.									
* Special categories of cited documents: 10 "A" document defining the general state of the art which is not considered to be of particular relevance "E" earlier document but published on or after the international filling date "L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified) "O" document referring to an oral disclosure, use, exhibition or other means "P" document published prior to the international filling date but later than the priority date claimed "CERTIFICATION "T" later document published after the international filling date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the cannot be considered novel or cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art. "A" document published after the international filling date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the cannot be considered novel or cannot be considered to involve an inventive atep when the cannot be considered to involve an inventive step when the cannot be considered to involve an inventive step when the cannot be considered to involve an inventive step when the cannot be considered to involve an inventive step when the cannot be considered to involve an inventive step when the cannot be considered to involve an inventive step when the cannot be considered to involve an inventive step when the cannot be considered to involve an inventive step when the cannot be considered to involve an inventive step when the cannot be considered to involve an inventive step when the cannot be considered to involve an inventive step when the cannot be considered to involve an inventive step when the cannot be considered to involve an inventive step when the cannot be c										
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