

26425

PHILIPPINE PATENT [19]

[11] No.: 26425
[45] Issued: JUL 15 1982

- [54] Title: AZANEOPHYL AND SELAZANEOPHYL SULFIDES, AGENTS CONTAINING THEM, AND THEIR USE AS PESTICIDES
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- [22] Filed: March 22, 1989
- [21] Application Serial No: 28376

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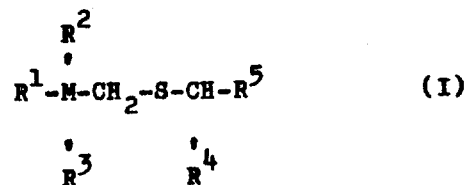
U.S. Pat. No. 4,775,664

H.H. Schubert

[57]

ABSTRACT

Azaneophyl and silazaneophyl sulfides, agents containing them, and their use as pesticides
Compounds of the formula I, their optical isomers and mixtures thereof



where

M_1 denotes C or Si,
 R^1 denotes unsubstituted or substituted pyridyl, unsubstituted or substituted pyrimidyl, and - when $M=C$ - denotes unsubstituted or substituted pyridazinyl, unsubstituted or substituted pyrazinyl, unsubstituted or substituted 1,2,4-triazinyl, unsubstituted or substituted

8 Claims. Specification: 149 page (s): Drawings: None see attached sheet sheet (s)

Examiner: EFIPANIO EVASCO

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Attorney / Agent:

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Cont. ABSTRACT:

1,2,4,5-tetrazinyl,

R^2 , R^3 independently of one another denote alkyl, alkenyl, haloalkyl, phenyl, or R^2 and R^3 denote an alkylene chain which - together with the quaternary central atom (M) - form an unsubstituted or fluorine-substituted ring having four to six ring members (when $M=Si$) or having three to six ring members (when $M=C$),

R^4 denotes H, F, $-CN$, $-CCl_3$, $-C \equiv CH$, (C_1-C_4) alkyl

$-C-NH_2$

"

S

R^5 denotes pyridyl, furyl, thienyl, all of which can be substituted, phthalimidyl, dialkylmaleiminidyl, thiophthalimidyl, dihydropthalimidyl, tetrahydropthalimidyl substituted phenyl

or R^4 and R^5 - together with the carbon atom bridging them - denote an optionally substituted indanyl, cyclopentenyl or cyclopentenyl radical,

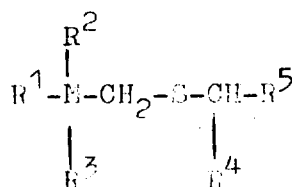
and to their salt and quaternization products for advantageous insecticidal, acaricidal or nematocidal properties. xxxxxx

CH. 39246

Abstract of the disclosure:

Azaneophyl and silazaneophyl sulfides, agents containing them, and their use as pesticides

5 Compounds of the formula I, their optical isomers and mixtures thereof



(I)

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where

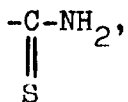
N denotes C or Si,

R¹ denotes unsubstituted or substituted
pyridyl, unsubstituted or substituted pyrimidyl,
10 and - when N = C - denotes unsubstituted
or substituted pyridazinyl, unsubstituted
or substituted pyrazinyl, unsubstituted
or substituted 1,2,4-triazinyl, unsub-
stituted or substituted 1,2,4,5-tetrazinyl,

15 R², R³ independtly of one another denote
alkyl, alkenyl, haloalkyl, phenyl, or R² and R³

denote an alkylene chain which - together with
the quaternary central atom (M) - form an
unsubstituted or fluorine-substituted ring
having four to six ring members (when M=Si)

5 or having three to six ring members (when M=C),
R⁴ denotes H, F, -CN, -CCl₃, -C=CH, (C₁-C₄)
alkyl,



10 R⁵ denotes pyridyl, furyl, thienyl, all of
which can be substituted, phthalimidyl,
dialkylmaleinimidyl, thiophthalimidyl,
dihydrophtalimidyl, tetrahydrophtalimidyl,
substituted phenyl

15 or R⁴ and R⁵ - together with the carbon atom
bridging them - denote an optionally
substituted indanyl, cyclopentenoyl or
cyclopentenyl radical,

20 and to their salts and quaternization products
for advantageous insecticidal, acaricidal or
nematocidal properties.

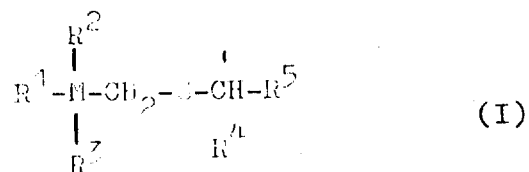
Description

Azaneophyl and silazaneophyl sulfides, agents containing them, and their use as pesticides

5 HI-A 0,248,015 describes the insecticidal and acaricidal action of certain silazaneophyl ethers and their hydrocarbon analogs. Furthermore, sulfur-containing silane derivatives are disclosed in HI-A 0,224,024, which also exhibit insecticidal and acaricidal activities.

10 Novel azaneophyl and silazaneophyl sulfides exhibiting advantageous insecticidal, acaricidal, or nematocidal properties have now been found,

15 The present invention thus relates to the compounds of the formula (I), their optical isomers and possible mixtures thereof



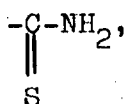
where

M denotes C or Si,

R¹ denotes unsubstituted or substituted pyridyl,
unsubstituted or substituted pyrimidyl,
and - when M = C- denotes unsubstituted
or substituted pyridazinyl, unsubstituted
or substituted pyrazinyl, unsubstituted
or substituted 1,2,4-triazinyl, unsub-
stituted or substituted 1,2,4,5-tetrazinyl,

R², R³ independently of one another denote
(C₁-C₃)alkyl, (C₂-C₈)alkenyl, (C₁-C₂)
haloalkyl, phenyl, or R² and R³ denote
an alkylene chain which - together with
the quaternary central atom (M) - form an
unsubstituted or fluorine-substituted ring
having four to six ring members (when M=Si)
or having three to six ring members
(when M=C),

R⁴ denotes H, F, -CN, -CCl₃, -C=CH, (C₁-C₄)
alkyl,



R⁵ denotes pyridyl, furyl, thienyl, all of

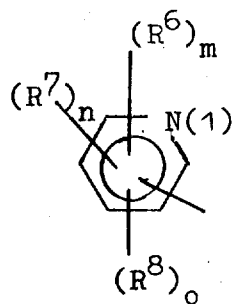
which can be substituted, phthalimidyl,
di(C₁-C₄)alkylmaleinimidyl, thiophtha-
limidyl, dihydrophthalimidyl, tetrahydro-
phthalimidyl, substituted phenyl
5 or R⁴ and R⁵ - together with the carbon atom
bridging them - denote an optionally
substituted indenyl, cyclopentenoyl or
cyclopentenyl radical,
and to their salts and quaternization products
10 which can be employed for agricultural purposes.
In this case, the salt formation or quaternization
is carried out by customary methods by an addition
reaction of suitable compounds with the basic
nitrogen or the basic nitrogen atoms of the
15 heterocyclic radicals R¹ and/or R⁵. Multiple
salt formation or quaternization is thus possible.

Suitable acids for salt formation on the nitrogen
are all inorganic or organic acids which, by
virtue of their pK_s value, are capable of
20 salt formation, for example hydrohalic acids,
nitric acid, sulfuric acid, phosphoric acid,
phosphonic acid, sulfonic acid, haloacetic acids
or oxalic acid.

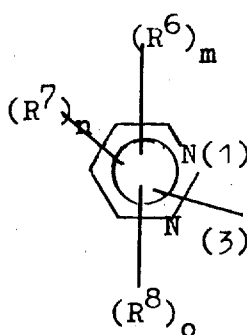
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A preferred optionally substituted pyridyl R^1 ,
 pyrimidyl R^1 , pyridazinyl R^1 , pyrazinyl R^1 ,
 1,2,4-triazinyl R^1 or 1,2,4,5-tetrazinyl R^1
 denotes a radical of the general formulae (A),
 (B), (C), (D), (E) or (F),

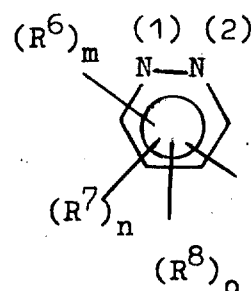
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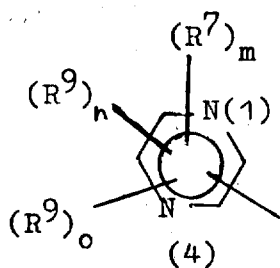
(A)



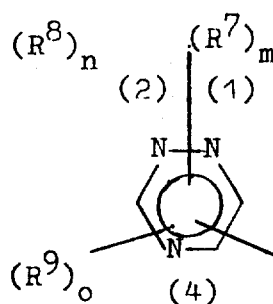
(B)



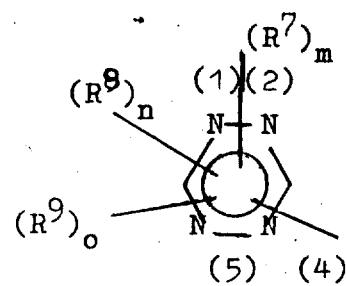
(C)



(D)



(E)



(F)

where $0 \leq m+n+o \leq 3$ and m, n, o may have the values 0 to 2. R^6 , R^7 , R^8 and R^9 independently of one another stand for (C_1-C_5) alkyl, halogen, (C_2-C_6) alkenyl, (C_2-C_6) alkynyl, (C_3-C_7) cycloalkyl, (C_1-C_3) alkoxy, (C_2-C_4) alkenyloxy, (C_2-C_4) alkynyloxy, (C_1-C_4) alkylthio, (C_3-C_6) cycloalkyloxy, (C_1-C_4) haloalkyl, (C_1-C_3) haloalkoxy, (C_1-C_3) haloalkylthio, (C_2-C_5) haloalkenyl, (C_2-C_5) haloalkenyloxy, (C_2-C_5) haloalkenylthio, or two of the radicals R^6 , R^7 , R^8 , R^9 - when they are in the ortho position - form a methylenedioxy, ethylenedioxy or (C_3-C_5) alkylene radical.

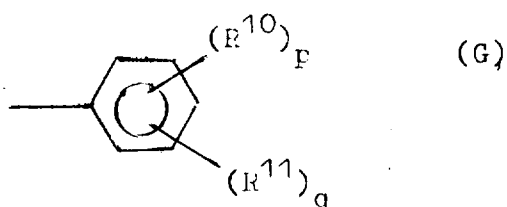
The linkage point (free valency) of the pyridyl radical (for example in formula (A)) on the quaternary central atom M in formula (I) is preferably in the 2- or 3-position of the pyridyl radical (N=position 1), particularly preferably in the 3-position of the pyridyl radical. The pyrimidyl radical (for example formula (B)) is preferably bonded to the quaternary central atom M in position 2 or 5. The preferred linkage

point of the pyridazinyl radical (for example formula (C)) is position 3, that of the pyrazinyl radical (for example formula (D)) position 2. Triazinyl radicals (for example formula (E)) and tetrazinyl radicals (for example formula (F)) are preferably linked to the quaternary central atom in the 3- or 6-position, respectively. R^1 particularly preferably denotes mono- or disubstituted pyridyl, pyrimidyl, pyridazinyl or pyrazinyl radicals of the formulae (A), (B), (C) or (D) where $m+n+o = 1$ or 2, the substituents (R^6-R^9) being oriented in particular in the para or meta position to the linkage point (quaternary central atom M). $R^1 = A$ is of particular importance.

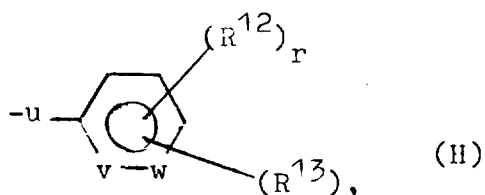
R^2 and R^3 preferably represent a (C_1-C_3)alkyl radical such as methyl, ethyl, i-propyl and n-propyl, a fluorinated methyl radical such as fluoromethyl, difluoromethyl and trifluoromethyl, or - when $M=C$ - together with the carbon atom linking them preferably denote an unsubstituted or mono- or difluorinated cyclopropyl ring.

R^4 preferably represents hydrogen, fluorine, cyano or (C_1-C_4) alkyl, particularly preferably hydrogen.

5 Substituted phenyl R^5 preferably represents a phenyl radical of the general formula (G)

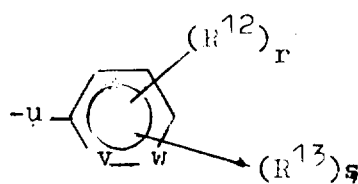


where R^{10} and R^{11} - independently of one another - can denote halogen, (C_1-C_4) alkyl, (C_1-C_4) alkoxy, (C_1-C_4) haloalkyl, phenyl, N-pyrrolyl
 10 or a group of the general formula (H)



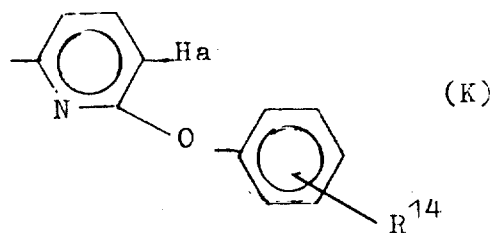
where R^{12} and R^{13} independently of one another
 denote H, halogen, (C_1-C_4) alkyl, (C_1-C_4) alkoxy
 and (C_1-C_4) haloalkyl; U denotes $-CH_2-$, $C=O$,
 -O- or -S-, preferably -O-; V, W denote CH
 5 or N, it being possible that both simultane-
 ously denote CH but not N,
 and where in formulae (G) and (H)
 p, q = an integer of 0 to 5 with the condition
 that the total $p + q$ must denote a number
 10 from 1 to 5,
 r, s = 0, 1 or 2, with the condition that the
 total of $r + s$ must be 0, 1 or 2, and
 the condition that, in the event that R^{10}
 or R^{11} corresponds to the
 15 group (H),
 p, q must denote 0 or 1, and $p + q$ must denote
 1 or 2.

Of these R^5 radicals, radicals of the formula
 (G), where (R^{10}) is H or 4-fluoro and (R^{11})
 20 is orientated in the β -position of the phenyl
 radical and denotes the radical



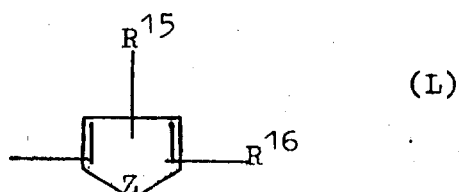
where $r + s$ preferably stands for 0, are particularly important.

As an optionally substituted pyridyl, R^5 represents a mono- or disubstituted pyridyl group of the general formula (K)



where R^{14} denotes halogen with the exception of I, (C_1-C_4) alkyl, (C_1-C_4) alkoxy or (C_1-C_4) haloalkyl and Hal denotes halogen, in particular fluorine or H.

- 5 As an optionally substituted thienyl or furyl, R^5 represents a heterocycle of the general formula (L)



where Z denotes O, S,

- 10 R^{15} denotes H, halogen, (C_1-C_4) alkyl, (C_1-C_4) alkoxy, (C_1-C_4) haloalkyl, CN or NO_2 and R^{16} denotes a generally substituted benzyl, propargyl, allyl or phenoxy.

- 15 Substituted phenyl radicals as R^5 are of particular importance for the invention.

Typical examples which are indicated for the group R^5 are the following radicals:

1entafluorophenyl, 5-benzyl-2-furyl, 4-
 phenoxyphenyl, 3-phenoxyphenyl, 3-(4-fluorophenoxy)
 phenyl, 3-(4-chlorophenoxy) phenyl, 3-(4-bromophenoxy) phenyl,
 3-(3-fluorophenoxy) phenyl, 3-(3-chlorophenoxy)
 5 phenyl, 3-(3-bromophenoxy) phenyl, 3-(2-fluoro-
 phenoxy) phenyl, 3-(2-chlorophenoxy) phenyl,
 3-(2-bromophenoxy) phenyl, 3-(4-methylphenoxy)
 phenyl, 3-(3-methylphenoxy) phenyl, 3-(2-methyl-
 phenoxy) phenyl, 3-(4-methoxyphenoxy) phenyl,
 10 3-(3-methoxyphenoxy)phenyl, 3-(2-methoxyphenoxy)
 phenyl, 3-(4-ethoxyphenoxy)phenyl, 3-(phenylthio)
 phenyl, 3-(4-fluorophenylthio)phenyl, 3-(3-
 fluorophenylthio)phenyl, 2-benzoylphenyl, 3-
 benzylphenyl, 3-(4-florobenzyl)phenyl, 3-(4-
 15 chlorobenzyl)phenyl, 3-(3,5-dichlorophenoxy)
 phenyl, 3-(3,4-dichlorophenoxy)phenyl, 3-(4-
 chloro-2-methylphenoxy)phenyl, 3-(2-chloro-
 5-methylphenoxy)phenyl, 3-(2-chloro-5-methyl-
 phenoxy)phenyl, 3-(4-ethylphenoxy)-phenyl,
 20 3-(3-chloro-4-methoxyphenoxy)phenyl, 3-(2,5-
 dichlorophenoxy)phenyl, 3-(3,5-dichlorobenzoyl)
 phenyl, 3-(3,4-dichlorobenzoyl)phenyl, 3-(4-
 methylbenzyl)phenyl, 3-(4-isopropoxyphenoxy)phenyl,
 4-fluoro-2-phenoxyphenyl, 4-chloro-2-phenoxyphenyl,

4-bromo-3-phenoxyphenyl, 4-fluoro-3-(4-fluoro-
 phenoxy)phenyl, 4-fluoro-3-(4-chlorophenoxy)
 phenyl, 4-fluoro-3-(4-bromophenoxy)phenyl, 4-
 fluoro-3-(4-methylphenoxy)phenyl, 4-fluoro-3-
 (4-methoxyphenoxy)phenyl, 4-fluoro-3-(3-fluoro-
 phenoxy)phenyl, 4-fluoro-3-(3-chlorophenoxy)
 phenyl, 4-fluoro-3-(3-bromophenoxy)phenyl, 4-
 fluoro-3-(3-methoxyphenoxy)phenyl, 4-fluoro-3-
 (4-ethoxyphenoxy)phenyl, 4-fluoro-3-(2-fluoro-
 phenoxy)phenyl, 3-methoxy-5-phenoxyphenyl, 2-
 fluoro-3-phenoxyphenyl, 2-fluoro-3-(4-fluoro-
 phenoxy)phenyl, 2-fluoro-3-(3-fluorophenoxy)
 phenyl, 2-fluoro-3-(2-fluorophenoxy)phenyl,
 3-fluoro-5-(4-fluorophenoxy)phenyl, 3-fluoro-5-
 (3-fluorophenoxy)phenyl, 3-fluoro-5-(2-fluoro-
 phenoxy)phenyl, 4-methyl-3-phenoxyphenyl,
 3-fluoro-5-(4-methylphenoxy)phenyl, 3-fluoro-5-
 (3-methoxyphenoxy)-phenyl, 2-fluoro-5-(4-
 fluorophenoxy)phenyl, 2-fluoro-5-(3-fluoro-
 phenoxy)phenyl, 2-fluoro-5-(2-fluorophenoxy)-
 phenyl, 2-chloro-3-phenoxyphenyl, 3-fluoro-5-
 phenoxyphenyl, 2-fluoro-5-phenoxyphenyl, 2-chloro-
 5-phenoxyphenyl, 2-bromo-5-phenoxyphenyl, 4-
 chloro-3-(3-methylphenoxy)phenyl, 4-chloro-3-
 (4-fluorophenoxy)phenyl, 3-chloro-5-phenoxy-
 phenyl, 3-bromo-5-phenoxyphenyl, 4-bromo-3-
 phenoxyphenyl, 4-trifluoromethyl-3-phenoxyphenyl,
 4-fluoro-3-phenylthiophenyl, 4-fluoro-3-benzyl-
 phenyl, 3-(2-pyridyloxy)phenyl, 3-(3-pyridyloxy)
 phenyl, 4-fluoro-3-(2-pyridyloxy)phenyl, 4-chloro-

3-(2-pyridyloxy)phenyl, 4-bromo-3-(2-pyridyloxy)
 phenyl, 4-methyl-3-(2-pyridyloxy)-phenyl, 4-fluoro-3-(3-
 pyridyloxy)phenyl, 4-chloro-3-(3-pyridyloxy)
 -phenyl, 4-bromo-3 (3-pyridyloxy)phenyl, 4-
 5 methyl-3-(3-pyridyloxyphenyl), 2-methyl-3-
 phenylphenyl, 2-methyl-3-(N-pyrrolyl)phenyl,
 6-phenoxy-2-pyridyl, 6-(4-fluorophenoxy)-2-
 pyridyl, 6-(4-chlorophenoxy)-2-pyridyl, 6-(4-
 bromophenoxy)-2-pyridyl, 6-(4-methylphenoxy
 10 (-2-pyridyl, 6-(4-methoxyphenoxy)-2-pyridyl,
 6-(4-ethoxyphenoxy)-2-pyridyl, 6-(3-fluorophenoxy)
 -2-pyridyl, 6-(3-chlorophenoxy)-2-pyridyl, 6-(3-
 bromophenoxy)-2-pyridyl, 6-(3-methoxyphen xy)-
 2-pyridyl, 6-(2-fluorophe oxy)-2-pyridyl, 6-
 15 (2-chlorophenoxy)-2-pyridyl, 6-(2-bromophen-
 oxy)-2-pyridyl, 5-propargyl-3-furyl, N-phthalimidyl,
 N-3,4,5,6,-phthalimidyl, 2-methyl-5-propargyl-3-
 furyl, 4-t-butylphenyl, 4-methylphenyl, 4-isopropyl-
 phenyl, 4-(2-chloro-4-trifluoromethyl-2-pyridyloxy)
 20 phenyl, 4-cyclohexylphenyl, 4-difluoromethoxyphenyl,
 4-biphenyl, 4-trimethylsilylphenyl and 4-phenoxy-
 2-thienyl.

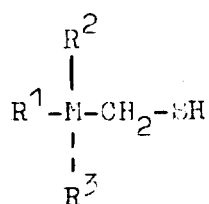
Other typical examples of the group -CH-R^5 are:

$$\begin{array}{c} | \\ \text{R}^4 \end{array}$$

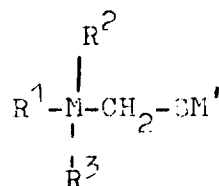
2-allyl-3-methylcyclopent-2-en-1-on-4-yl,
4-phenylindan-2-yl. R^6 - R^9 independently of
one another in particular represent methyl,
ethyl, propyl, iso-propyl, butyl, t-butyl, F,
5 Cl, Br, I, vinyl, allyl, ethynyl, propargyl,
methoxy, ethoxy, propoxy, iso-propoxy, difluoro-
methoxy, bromodifluoromethoxy, chlorodifluoro-
methoxy, trifluoromethoxy, 1,1,2,2-tetrafluoroethoxy,
2-chloro-1,1,2-trifluoroethoxy, 2,2,2-trifluo-
10 roethoxy, 2-fluoroethoxy, 1,1,2,3,3,3-hexa-
fluoropropoxy, vinyloxy, allyloxy, propargyloxy,
ethylthio, methylthio, propylthio, isobutylthio,
trifluoromethyl, 1,1,2,2-tetrafluoroethyl,
1,1,2,2-tetrafluoroethylthio, 2,2,2-trifluoro-
15 ethylthio, difluoromethylthio, trifluoromethyl-
thio, chlorodifluoromethylthio, bromodifluoro-
methylthio, heptafluoropropyl, trichlorovinyl,
trichlorovinyloxy, methylenedioxy or ethylenedioxy.

The present invention also relates to processes
20 for the preparation of the compounds of the
general formula (I), which comprise
reacting

- a) a mercaptan of the general formula (II)
or its salt of the general formula III

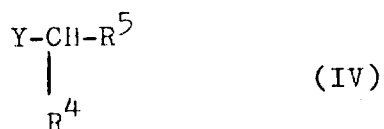


(II)



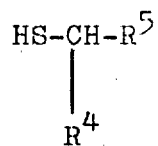
(III)

- where M' corresponds to an alkali metal
equivalent or alkaline earth metal equivalent,
5 in particular Li, Na, K, Hg, Ca, with an
alkylating agent of the general formula (IV)

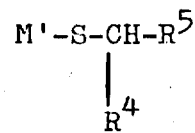


- where Y denotes a nucleofugic leaving group, such as,
for example, halogen or sulfonate, if appropriate
in the presence of a base,
10 or

- b) a mercaptan of the general formula (IV) or
its salt of the general formula (VI)

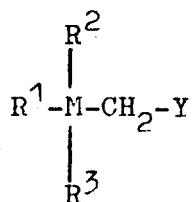


(V)



(VI)

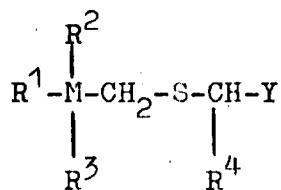
with an alkylating agent of the general formula (VII)



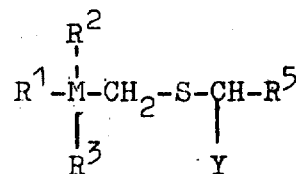
(VII)

if appropriate in the presence of a base,
or

- 5 c) a sulfide of the general formula (VIII)
or (IX)

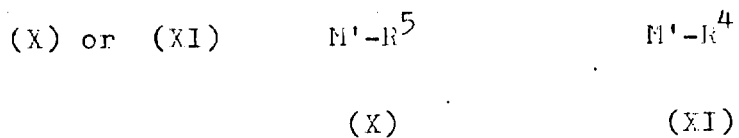


(VIII)



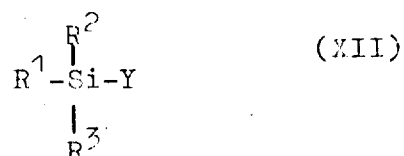
(IX)

with a metal compound of the general formula

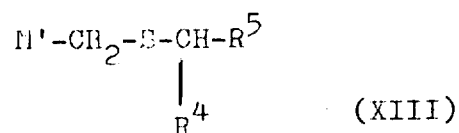


or -where M = Si -

d) a silane of the general formula (XII)

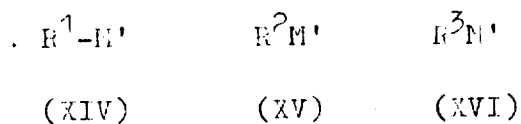


5 with an organometal compound of the general formula (XIII)

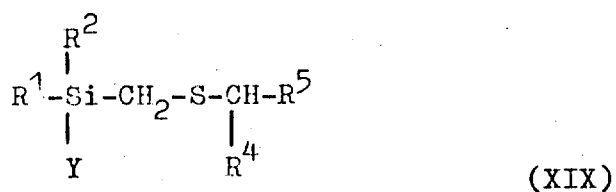
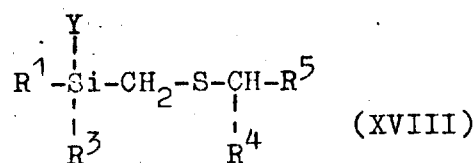
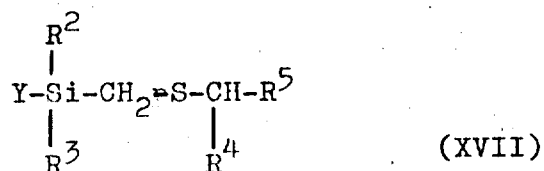


or - where M = Si-

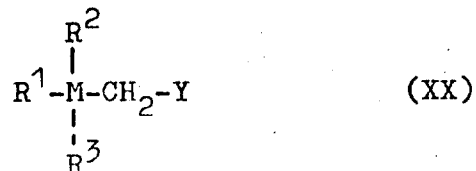
e) an organometal compound of the general formulae (XIV), (XV) or (XVI)



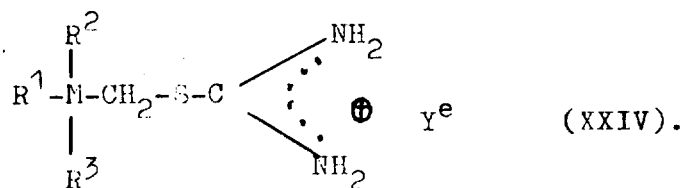
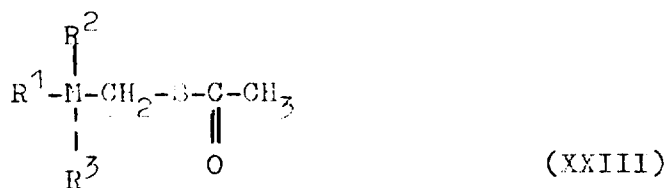
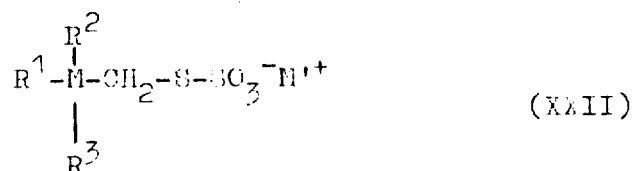
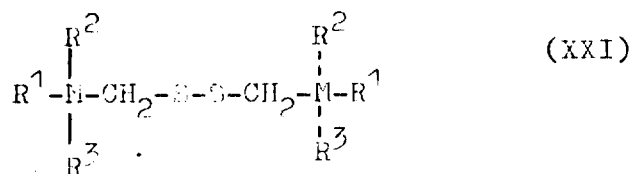
10 with a silane of the general formulae (XVII), (XVIII) or (XIX)



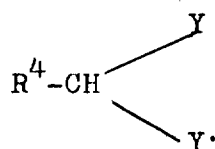
Some of the mercaptans of the general formula (II) to be used as the starting compound in preparation process a) are novel and can be prepared by a process known per se from the literature, for example by initially reacting suitable alkylating agents of the general formula (XX) (see EP-A 0,249,015 and German Patent No. 3,712,752.7) .



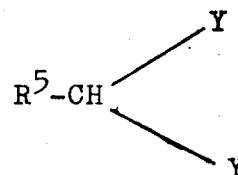
with a sulfur nucleophile, such as disulfide, thiosulfate, thioacetate or thiourea, and hydrolyzing or reducing the resulting intermediates of the general formulae (XXI), (XXII), (XXIII) or (XXIV) by customary standard methods (see also Methoden der org. Chemie /Methods of Organic Chemistry_7(Houben-Weyl), Vol. E11/1, Georg Thieme Verlag, Stuttgart 1985; p. 32 et seq)



Some of the sulfides (VIII) and (IX) to be used as starting substances in preparation process c) are novel and can be prepared by a process which is known per se from the literature by reacting an alkylating agent of the general formulae (XXV) or (XXVI)

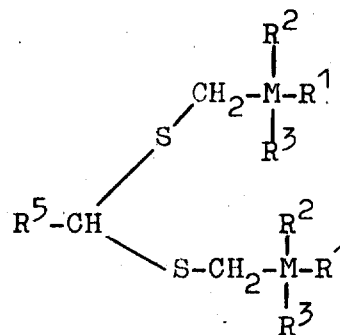
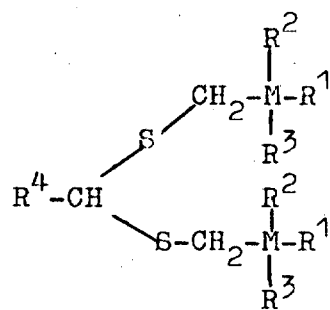


(XXV)



(XXVI)

with two equivalents of a mercaptan of the formula (II) or its salt of the formula (III), if appropriate in the presence of a base, and treating the intermediates obtained, of the general formula (XXVII) or (XXVIII),



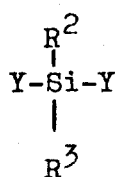
with chlorine (see Methoden der org. Chemie
 /Methods of Organic Chemistry_7(Houben-Weyl),
 Vol. 5/3, Georg Thieme Verlag, Stuttgart
 1962, p. 756), in which process compounds
 5 of the general formula (VIII) or (IX) where
 Y = Cl are formed.

In a comparable manner, chloromethyl sulfides
 of the general formula (XXIX) can be obtained.

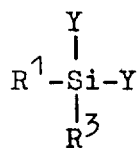


These can be used as educts for the intermediates
 10 of the general formula (XIII) which are
 required as a starting compound in preparation
 process d) and which can be prepared from them
 by processes known per se from the literature
 (see, for example, Methoden der org. Chemie
 15 /Methods of Organic Chemistry_7(Houben-Weyl),
 Vol. 13/2a, Georg Thieme Verlag, Stuttgart
 1973, p. 115).

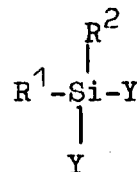
The silanes of the general formulae (XVII),
 (XVIII) and (XIX) which are to be used as
 starting compounds in preparation process
 e) can be prepared by processes known per se
 5 from the literature (see Methoden der org.
 Chemie /Methods of Organic Chemistry_7(Houben-
 Weyl), Vol. 13/5, Georg Thieme Verlag, Stuttgart
 1980) by reacting an organometal intermediate
 of the general formula (XIII) with silanes of
 10 the formulae (XXX), (XXXI) or (XXXII)



(XXX)



(XXXI)



(XXXII)

The intermediates (IV), (V), (VI), (VII), (X),
 (XI), (XII), (XIV), (XV) and (XVI) are either
 obtained as described above (EP-A 0,249,015
 and German Patent No.3,712,752.7), or alternatively
 15 they can be prepared by methods quoted in the
 current literature (see, inter alia, the
 literature quoted so far):

The mentioned process variants c), d) and e) are preferably carried out in a diluent the nature of which depends on the type of the organometal compound employed. Suitable diluents are, in particular, aliphatic and aromatic hydrocarbons, such as, for example, pentane, hexane, heptane, cyclohexane, petroleum ether, benzine, ligroin, benzene, toluene and xylene, ethers, such as, for example, diethyl ether, dibutyl ether, glycol dimethyl ether, diglycol dimethyl ether, tetrahydrofuran or dioxane, and, finally, all possible mixtures of the previously mentioned solvents.

In the abovementioned process variants, the reaction temperature is between -110°C and $+150^{\circ}\text{C}$, preferably between -75°C and $+105^{\circ}\text{C}$. The starting materials are usually employed in equimolar amounts. However, it is possible to use an excess of one or the other reactant.

What has been said for variants c)-e) is essentially also true for process variants

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a) and b) which have been mentioned further above. Alternatively, other diluents can be used. Thus, in these cases, alcohols, such as, for example, methanol, ethanol, propanol, 5 i-propanol or butanol, amides, such as, for example, dimethylformamide, dimethylacetamide or N-methylpyrrolidone, nitriles, such as, for example, acetonitrile or butyronitrile, ketones, such as, for example, acetone or MIBK, 10 and also dimethyl sulfoxide, tetramethylene sulfone or hexamethylphosphoric triamide are also suitable as diluents. However, process variants a) and b) can also be carried out without diluents.

15 Bases which can be used are inorganic bases, such as, for example, the hydroxides, hydrides, carbonates, acetates or hydrogen carbonates of alkali metals and alkaline earth metals, but also organic bases, such as, for example, 20 pyridine, triethylamine, N,N-diisopropylethylamine or diazabicyclooctane. The mercaptans (II) or (V) can alternatively be converted to their salts (III) or (VI), respectively, by adding

organometal reagents, such as, for example, methyl lithium, butyllithium, methylmagnesium halide, phenylmagnesium halide etc.

5 The compounds of the formula (I) are isolated, and, if appropriate, purified by generally customary methods, for example by distilling of the solvent (if appropriate under reduced pressure), followed by distillation chroma-
10 tography or distribution of the crude product between two phases and its subsequent customary work-up.

The compounds of the general formula (I) are readily soluble in most organic solvents.

The active compounds are well tolerated by
15 plants, have a favourable toxicity to warm-blooded animals and are suitable for combating animal pests, in particular insects, arachnids and nematodes, which are encountered in agriculture, in forestry, in the protection of stored products
20 and of materials, and in the hygiene field. They are active against normally sensitive and



resistant species and against all or individual stages of development. The abovementioned pests include :

- 5 From the order of the Isopoda, for example, *Cniscus asellus*, *Armadillidium vulgare* and *Porcellio scaber*. From the order of the Diplopoda, for example, *Blaniulus guttulatus*. From the order of the Chilopoda, for example, *Geophilus carpophagus* and *Scutigera spec.*
- 10 From the order of the Symphyla, for example, *Scutigera immaculata*. From the order of the Thysanura, for example, *Lepisma saccharina*. From the order of the Collembola, for example, *Onychiurus armatus*. From the order of the Orthoptera,
- 15 for example, *Blatta orientalis*, *Periplaneta americana*, *Leucophaea maderae*, *Blattella germanica*, *Acheta domesticus*, *Gryllotalpa spp.*, *Locusta migratoria migratorioides*, *Melanoplus differentialis* and *Schistocerca gregaria*. From
- 20 the order of the Dermaptera, for example, *Forficula auricularia*. From the order of the Isoptera, for example, *Reticulitermes spp.*. From the order of the Anoplura, for example, *Phylloxera vastatrix*, *Pemphigus spp.*, *Pediculus humanus corporis*, *Haematopinus spp.* and
- 25 *Linognathus spp.* From the order of the Mallophaga,

for example, *Trichodectes* spp. and *Damalinea*
 spp. From the order of the Thysanoptera, for
 example, *Mercinothrips femoralis* and *Thrips*
tabaci. From the order of the Heteroptera,
 5 for example, *Eurycastor* spp., *Dysdercus*
intermedius, *Liesma quadrata*, *cimex lectularius*,
Rhodnius prolixus and *Triatoma* spp. From
 the order of the Homoptera, for example,
Aleurodes brassicae, *Bemisia tabaci*, *Trialeurodes*
 10 *vaporariorum*, *Aphis gossypii*, *Brevicoryne*
brassicae, *Cryptomyzus ribis*, *Doralis fabae*,
Doralis pomi, *Eriosoma lanigerum*, *Hyalopterus*
arundinis, *Macrosiphum avenae*, *Myzus* spp.,
Ichorodon humuli, *Rhopalosiphum padi*, *Empoasca*
 15 spp., *Muscelis bilobatus*, *Aephotettix cincticeps*,
Lecanium corni, *Gaioisettia oleae*, *Maodelphax*
striatellus, *Nilaparvata lugens*, *Aonidiella*
aurantii, *Aspidiotus hederæ*, *Pseudococcus* spp.
 and *Isylla* spp. From the order of the Lepidop-
 20 tera, for example, *Pectinophora gossypiella*,
Eupalus pinarius, *Cheimatobia brumata*, *Lithoco-*
lletis blancardella, *Hyponomeuta padella*,
Plutella maculipennis, *Malcosoma neustria*,
Euproctis chrysorrhoea, *Lymantria* spp. *Bucculatrix*



thurberiella, Phyllocnistis citrella, Agrotis
 spp., Euxoa spp., Feltia spp., Earias insulana,
 Heliothis spp., Laphygma exigua, Mamestra
 brassicae, Panolis flammea, Prodenia litura,
 5 Spodoptera spp., Trichoplusia ni, Caprocapsa
 pomonella, Pieris spp., Chilo spp., Pyrausta
 nubilalis, Ephestia kuehniella, Galleria
 mellonella, Tineola bisselliella, Tinea
 pellionella, Hofmannophila pseudospretella,
 10 Cacoecia podana, Capua reticulana, Choristoneura
 fumiferana, Clysia ambiguella, Homona magnanima
 and Tortrix viridana. From the order of the
 Coleoptera, for example, Anobium punctatum Rhizopertha
 dominica, Bruchidius obtectus, Acanthoscelides
 15 obtectus, Hylotrupes bajulus, Agelastica alni,
 Leptinotarsa decemlineata, Phaedon cochleariae,
 Diabrotica spp., Psylliodes chrysocephala,
 Epilachna varivestis, Atomaria spp., Oryzaephilus
 surinamensis, Anthonomus spp., Sitophilus spp.,
 20 Otiorrhynchus sulcatus, Cosmopolites sordidus,
 Ceuthorrhynchus assimilis Hypera postica,
 Dermestes spp., Trogoderma spp., Anthrenus spp.,
 Attagenus spp., Lyctus spp., Meligethes aeneus,
 Ptinus spp., Niptus hololeucus, Gibbium psyllodes,

Tribolium spp., Tenebrio molitor, Agriotes
 spp., Conoderus spp., Melolontha melolontha,
 Amphimallon solstitialis and Costelytra
 zealandica. From the order of the Hymenoptera,
 5 for example spp., Hoplocampa spp., Lasius spp.,
 Monomorium pharaonis and Vespa spp. From the
 order of the Diptera, for example, Aedes spp.,
 Anopheles spp., Culex spp., Drosophila melanogaster,
 Musca spp., Fannia spp., Calliphora erythrocephala,
 10 Lucilia spp., Chrysomya spp., Cuterebra spp.,
 Gastrophilus spp., Hyppobosca spp., Stomoxys
 spp., Cestrus spp., Hypoderma spp., Tabanus spp., Tannia
 spp., Bibio hortulanus, Oscinella frit, Ichorbia
 spp., Pegomya hyoscyami, Ceratitis capitata, Dacus
 15 oleae and Tipula paludosa. From the order of
 the Siphonaptera, for example, Xenopsylla
 cheopis and Ceratophyllus spp. From the order of
 the Arachnida, for example, Scorpio maurus and
 Latrodectus mactans. From the order of the
 20 Acarina, for example, Acarus siro, Argas spp.,
 Ornithodoros spp., Dermanyssus gallinae, Eriophyes
 ribis, Ixyllocoptura oleivora, Boophilus spp.,
 Rhipicephalus spp., Amblyomma spp., Hyalomma spp.,
 Ixodes spp., Psoroptes spp., Chorioptes spp.,

Sarcoptes spp., Tarsonemus spp., Bryobia
praetiosa, Panonychus spp. and Tetranychus spp.

5 Furthermore the compounds have an excellent
action against Nematodes which are harmful to
plant, for example those of the species
Meloidogyne, Heterodera, Ditylenchus, Aphelen-
choides, Radopholus, Globodera, Pratylenchus,
Longidorus and Xiphinema.

10 The invention also relates to agents which
contain the compounds of the formula (I) besides
suitable formulation auxiliaries. In general,
the agents according to the invention contain 1-
95% by weight of the active substances of the
formula (I). They can be formulated in different
15 ways, depending on the biological and/or chemical-
physical parameters. Thus, the following for-
mulation possibilities are suitable: wettable
powders (WF), emulsifiable concentrates (EC),
aqueous solutions (SC), emulsions, spraying
20 solutions, dispersions on oil or water bases (SC),
suspo emulsions (SC), dusting agents (DA), dressing
agents, granules in the form of microgranules,

sprayable granules, coated granules and adsorption granules, water-dispersible granules (WG), ULV formulations, microcapsules, waxes or baits.

5 These individual formulation types are known
in principle and are described, for example,
in: Winnacker-Kuchler, "Chemische Technologie
/Chemical Technology_7", Volume 7, C. Hauser
Verlag Munich, 4th edition 1986; van Falkenberg,
"Pesticides Formulations", Marcel Dekker N.W.,
10 2nd Ed. 1972-73; E. Martens, "Spray Drying
Handbook", 3rd Ed. 1979, G. Goodwin Ltd., London.

The formulation auxiliaries required, such as
inert materials, surfactants, solvents and other
additives, are likewise known and are described,
15 for example, in: Watkins, "Handbook of Insecticide
Dust Diluents and Carriers", 2nd Ed., Barland
Books, Caldwell, N.J.; H.v. Clphen, "Introduction
to Clay Colloid Chemistry", 2nd Ed., J. Wiley
& Sons, N.Y.; Marschen, "Solvents Guide", 2nd
20 Ed., Interscience, N.Y. 1950; McCutcheon's,
"Detergents and Emulsifiers Annual", MC Publ.
Corp., Ridgewood, N.J.; Sisley and Wood,

"Encyclopedia of Surface Active Agents", Chem. Publ. Co. Inc., N.Y. 1964; Schönfeldt, "Grenzflächenaktive Athylenoxidaddukte /Surface-active ethylene oxide adducts_7", Wiss. Verlagsgesell., Stuttgart 1976; Winnacker-Küchler, "Chemische Technologie /Chemical Technology_7", Volume 7, C. Hauser Verlag Munich, 4th edition 1986.

On the basis of these formulations, it is also possible to prepare combinations with other pesticidally active substances, fertilizers and/or growth regulators, for example in the form of a readymix or as a tank mix. Wetttable powders are preparations, uniformly dispersable in water, which contain, besides the active substance and in addition to a diluent or inert substance, wetting agents, for example polyoxyethylated alkylphenols, polyoxyethylated fatty alcohols, alkyl- or alkylphenol sulfonates, and dispersing agents, for example sodium lignin-sulfonate, sodium 2,2'-dinaphthylmethane-6,6'-disulfonate, sodium dibutyl-naphthalenesulfonate, or sodium oleoylmethyltaurinate. Emulsifiable

concentrates are prepared by dissolving the active substance in an organic solvent, for example butanol, cyclohexanone, dimethylformamide, xylene or higher-boiling aromatics or hydrocarbons, with the addition of one or more emulsifiers, examples of emulsifiers which can be used are: calcium salts of alkylaryl-sulfonate, such as Ca dodecylbenzenesulfonate, or non-ionic emulsifiers, such as fatty acid polyglycol esters, alkylaryl polyglycol ethers, fatty alcohol polyglycol ethers, propylene oxide/ethylene oxide condensation products, alkyl polyether, sorbitan fatty acid esters, polyoxyethylene sorbitan fatty acid esters or polyoxyethylene sorbito esters.

Dusting agents are obtained by grinding the active compound with finely divided solid substances, for example talc, natural clays such as kaolin bentonite, pyrophyllite or diatomaceous earth. Granules can be prepared either by atomizing the active substance onto adsorptive, granulated inert material or by applying active compound concentrates onto the surface of carrier materials, such as sand, kaolinites or granulated inert material, by means of adhesives, for example polyvinyl alcohol or sodium polyacrylate, or alternatively mineral oils. Suitable active

substances can also be granulated in the manner customary for the preparation of fertilizer granules - if desired as a mixture with fertilizers.

- 5 In wettable powders, the active substance concentration is, for example approximately 10 to 90% by weight, the remainder to 100% by weight consists of customary formulation components. In the case of emulsifiable concentrates, the active substance concentration
- 10 may be approximately 5 to 80% by weight. Formulations in the form of dusts usually contain 5 to 20% by weight of active substance, sprayable solutions approximately 2 to 20% by weight. In
- 15 the case of granules, the active substance content partly depends on whether the active substance is present as a liquid or as a solid, and which granulation auxiliaries, fillers etc. are used.
- 20 Besides, the active substance formulations mentioned may contain the adhesives, wetting agents, dispersing agents, emulsifiers, penetrants,

solvents, fillers or carriers which are customary in each case.

5 For application the concentrates present in commercially available form may be diluted in a customary manner, for example in the case of wettable powders, emulsifiable concentrates, dispersions and, in some cases, also in the case of microgranules, by means of water. Preparations in the form of dusts and of granules
10 and also sprayable solutions are usually not diluted any further with other inert substances prior to application.

The dosage rate required varies with the external conditions, such as temperature, humidity
15 and the like. It can vary within wide limits, for example between 0.005 and 10.0 kg/ha or more of active substance, however, preferably it is between 0.01 and 5 kg/ha.

The active substances according to the invention
20 can be present in their commercially available formulations and in the application forms

prepared from these formulations, in mixtures
with other active substances, such as insecticides,
pheromones, sterilants, acaricides, nematocides,
fungicides, growth-regulating substances or
5. herbicides.

The insecticides include, for example, phosphoric
acid esters, carbamates, carboxylic acid esters,
formamidines, tin-compounds, substances prepared
by microorganisms and the like.

10 Preferred mixture components are

1. from the group comprising the phosphorus
compounds acephate, azamethiphos, azinphos-ethyl,
azinphos-methyl, bromophos, bromophos-ethyl,
chlorfenvinphos, chlormephos, chlorpyrifos,
15 chlorpyrifos-methyl, demeton, demeton-S-
methyl, demeton-S-methyl sulfphone, dialifos,
diazinon, dichlorvos, dicrotophos, O,O-1,2,2,2-
tetrachloroethylphosphorthioate (SD 208 304),
dimethoate, disulfoton, EPN, ethion, ethoprophos,
20 etrimfos, famphur, fenamiphos, fenitrothion, fensul-
fothion, fenthion, fonofos, formothion, hepte-
nophos, isazaphos, isothioate, isoxathion,

malathion, methacrifos, methamidophos,
 methidathion, salithion, mevinphos, monocro-
 tophos, naled, omethoate, oxydemeton-methyl,
 parathion, parathion-methyl, phenthoate,
 5 phorate, phosalone, phosfolan, phosmet, phospham-
 idon, phoxim, pirinaphos-ethyl, pirimiphos-
 methyl, profenofos, propaphos, proetamphos, pro-
 thiofos, pyraclofos, pyridapenthion, quinalphos, sul-
 profos, terephos, terbufos, tetrachlorvinphos,
 10 thiometon, triazophos, trichlorphon, vamidothion;

2. from the group comprising the carbamates
 aldicarb, 2-sec.-butylphenyl methylcarbamate
 (BIFC), carbaryl, carbofuran, carbosulfan,
 cloethocarb, benfuracarb, ethiofencarb, furathiocarb,
 15 isoprocab, methomyl, 5-methyl-m-cumenylbutyryl
 (methyl)carbamate, oxamyl, pirimicarb, propoxur,
 thiodicarb, thiofanox, ethyl 4,6,9-triaza-4-
 benzyl-6, 10-dimethyl-8-oxa-7-oxa-5,11-dithia-
 9-dodecenoate (OF 135), 1-methylthio(ethylide-
 20 neamino) N-methyl-N-(morpholiniothio)carbamate
 (UC 51717);

3. from the group comprising the carboxylic
 acid esters allethrin, alphametrin, 5-benzyl-

3-furylmethyl (E)-(1R)-cis-2,2-dimethyl-3-(2-oxothiolan-3-ylidenemethyl)cyclopropanecarboxylate, bipallethrin, bioallethrin((S)-cyclopentylisomer), bioresmethrin, biphenate, 5 (RA)-1-cyano-1-(6-phenoxy-2-pyridyl)methyl(1RS)-trans-3-(4-tert.butylphenyl)-2,2-dimethylcyclopropanecarboxylate (NCI 85193), cycloprothrin, cyfluthrin, cyhalothrin, cypermethrin, cyphenothrin, deltamethrin, empenthrin, 10 esfenvalerate, fenfluthrin, fenpropathrin, fenvalerate, flucythrinate, flumethrin, fluvalinate (D-isomer), permethrin, phenothrin ((R)-isomer), d-pralethrin, pyrethrins (natural products), resmethrin, tefluthrin, tetramethrin, tralome- 15 thrin, hydroprene, methoprene, knoprene;

4. from the group comprising the amidines
amitraz, chlordimeform;

5. from the group comprising the tin compounds
cyhexatin, fenbutatin oxides, azocyclotin;

20 6. others

abamectin, Bacillus thuringiensis, bensultrap,
 binapacryl, bromopropylate, buprofezin,
 camphechlor, cartap, chlorobenzilate, chlor-
 fluazuron, 2-(4-chlorophenyl)-4,5-diphenyl-
 5 thiophene (UBI-T 930), chlrofentezine, 2-
 naphthylmethyl cyclopropanecarboxylate (Ro
 42-0470), cyromazin, N-(3,5-dichloro-4-(1,1,
 2,3,3,3-hexafluoro-1-propyloxy)phenyl)carbamoyl
 -2-chlorobenzocarboximidate, DDT, dicofol,
 10 N-(N-(3,5-dichloro-4-(1,1,2,2-tetrafluoro-
 ethoxy)phenylamino)carbonyl)-2,6-difluorobenzamide
 (KRD 473), diflubenzuran, N-(2,3-dihydro-3-
 methyl-1,3-thiazol-2-ylidene)-2,4-xylidine,
 dinobuton, dinocap, endosulfan, ethofenprox,
 15 (4-ethoxyphenyl)(dimethyl)(3-(3-phenoxyphenyl)
 propyl)silane, (4-ethoxyphenyl) (3-(4-fluoro-
 3-phenoxyphenyl)propyl)dimethylsilane, fenoxycarb,
 2-fluoro-5-(4-(4-ethoxyphenyl)-4-methyl-1-pentyl)
 diphenyl ether (MTI 800), granulosis and nuclear
 20 polyhedrosis viruses, fen thiocarb, flubenzimine,
 flucyclexuron, flufenoxuron, gamma-HCH, hexythiazox,
 hydramethylnon (AG 217300), ivermectin, 2-nitromethyl-
 4,5-dihydro-6H-thiazine (SD 52618), 2-nitromethyl-3,4
 dihydrothiazole (D 35651), 2-nitromethylene-

1,2-thiazinan-3-ylcarbamaldehyde (WL 108477),
propargite, teflubenzuron, tetradifon, tetrasul,
thiocyclam, triflumuron, O-ethyl-N- /2-(4-
phenoxy)phenoxyethyl_7carbamate, O-/2-(4-
5 phenoxy)phenoxyethyl_7propionaldoxim, 1-(4-
phenoxy-phenoxy)-2-(2-pyridinoxy)propane.

The active substance content of the use forms
prepared from the commercially available
formulations can vary within broad ranges. The
10 active substance concentration of the use forms
can be from 0.0000001 to 100% by weight of
active substance, preferably between 0.00001
and 1% by weight. Application is effected in
a conventional fashion, matched to the use forms.

15 The active substances according to the invention
are also suitable for combating ecto- and
endoparasites preferably ectoparasitizing
insects, in the veterinary medicine area or in
the area of animal husbandry.

20 The active substances according to the invention
are applied here in a known fashion, such as

by oral application in the form of, for example,
tablets, capsules, potions or granules, by
dermal application in the form of, for example,
dipping, spraying, pouring-on and spotting-
on and powdering.

5

The dosages and formulations suitable in each
case are particularly dependent on the type and
stage of development of the productive animals
and also on the degree of infestation of the
insects, and can easily be determined and fixed
by conventional methods.

10

The following examples serve to illustrate
the invention.

A. Formulation Examples

15

a) A dusting agent is obtained by mixing 10
parts by weight of active substance and 90
parts by weight of talc as inert material
and comminuting in a hammer mill.

b) A wettable powder which is easily dispersible in

water is obtained by mixing 25 parts by weight of active substance, 65 parts by weight of kaolin-containing quartz as inert material, 10 parts by weight of potassium
5 ligninsulfonate and 1 part by weight of sodium oleoylmethyltaurinate as wetting and dispersing agent, and grinding in a pin disk mill.

c) A dispersion concentrate which is easily dispersible in water is prepared by mixing
10 40 parts by weight of active compound with 7 parts by weight of a sulfo succinic acid monoester, 2 parts by weight of the sodium salt of a ligninsulfonate and 51 parts
15 by weight of water and grinding in a ball mill, to a fineness of below 5 microns.

d) An emulsifiable concentrate can be prepared from 15 parts by weight of active substance, 75 parts by weight of cyclohexanone as solvent and 10 parts by weight of oxyethylated nonyl-
20 phenol (10 EO) as emulsifier.

5 e) Granules can be prepared from 2 to 15 parts
by weight of active substance and inert
granule carrier material, such as attapulgite,
pumice granules and/or quartz sand. A
suspension of the wettable powder from
example b)), having a solids content of 30%
is expediently used, and this is sprayed onto
the surface of attapulgite granules, which
are dried and mixed intimately. Here, the
10 proportion by weight of the wettable powder
is approx. 5% and that of the inert carrier
material approx. 95% of the finished granules.

B. Chemical Examples

Preparation instructions

15 1) 56 ml (= 140 mmol) of 2.5 M-n-butyllithium
solution in hexane are added dropwise at
-70°C to a mixture of 25.6 g (100 mmol) of
5-bromo-2-(2,2,2-trifluoro ethoxy)pyridine
(from 2,5-dibromopyridine and 2,2,2-trifluoro-
20 ethanol/sodium hydride in DME), 20 g (140
mmol) of chloro-chloromethyl-dimethylsilane
and 120 ml of anhydrous tetrahydrofuran.

After the addition is complete, stirring is continued for a further 30 minutes at -70°C , the reaction mixture is allowed to slowly come to room temperature, and 5 ml of water are finally added. After the reaction solution has been evaporated, the residue is taken up in 200 ml of hexane, and the mixture is washed twice using water. The hexane phase, which has been dried over Na_2CO_3 , contains butyl-chloromethyldimethylsilane as a by-product, besides the desired intermediate $\leq 2-(2,2,2\text{-trifluoroethoxy})\text{pyrid-5-yl}-(\text{dimethyl})-(\text{chloromethyl})\text{-silane}$.

Most of the by-product is removed by incipient distillation of the crude product at up to 50°C (internal temperature) at 0.2 mbar. The silylpyridine, which remains in the residue, is added dropwise without further purification at room temperature to a solution of 15 g (130 mmol) of potassium thio acetate in 100 ml of anhydrous

DMF. The reaction mixture is allowed to stand at 60-70°C for 1.5 h and is then poured into 1 l of ice water, and the product is extracted using several portions of ether. The extracts
5 are washed three times using a 20% strength sodium chloride solution, dried over Na_2CO_3 and evaporated. Subsequent distillation in vacuo yields 22.3 g (69%) of < 2-(2,2,2-trifluoroethoxy)-pyrid-5yl >-(dimethyl)-(acetyl-
10 thiomethyl)silane as a pale yellow oil of boiling point 95-100°C/0.1 mbar.

2) The solution of 22.3 g (69 mmol) of < 2-(2,2,2-trifluoroethoxy)-pyrid-5yl >-(dimethyl)-(acetyl-thiomethyl)-silane in 200 ml of anhydrous
15 ether is added dropwise at room temperature to a stirred suspension of 5.3 g (150 mmol) of lithiumaluminum hydride in 50 ml of anhydrous ether. After a further 3 h, the reaction mixture is decomposed by the dropwise addition of a
20 little water. The solid precipitate which forms during this process is filtered off with suction and washed thoroughly using ether. After evaporation, the filtrates yield 8.6 g (44%) of



(mercaptomethyl)-(dimethyl)-<2-(2,2,2-trifluoroethoxy)-pyrid-5-yl>-silane as a pale yellow oil.

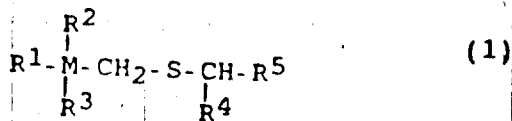
- 3) At -60°C, initially the solution of 4.3 g (15.3 mmol) of (mercaptomethyl)-(dimethyl)-<2-(2,2,2-trifluoroethoxy)-pyrid-5-yl>-silane in 10 ml of THF and then 10 ml of 1.6 M-methylolithium solution in ether are added dropwise to a solution of 5.3 g (18.8 mmol) of 4-fluoro-3-phenoxybenzyl bromide in 10 ml of anhydrous THF. During a subsequent period of approximately 2 hours, the reaction solution is allowed to come to room temperature, and stirring is continued for 2 hours at 20-25°C. After the addition of 200 ml of water, the product is extracted using two 100 ml portions of hexane. After the extracts have been washed twice using water, dried (Na₂SO₄) and evaporated 8.1 g of crude product are obtained. Through subsequent chromatography of silica gel using methylene chloride/hexane (1:1), 3.1 g (42%) of (dimethyl)<2-(2,2,2-trifluoroethoxy)-pyrid-5-yl>-<(4-fluoro-3-phenoxy-benzyl)-thiomethyl>-silane (Example 26/

0.05

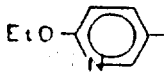
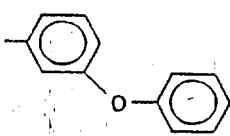
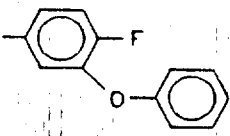
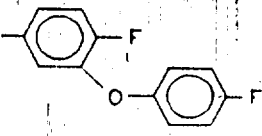
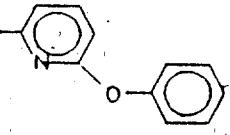
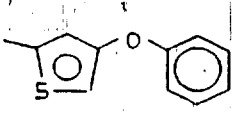
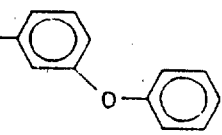
- 270-240°C (bulb tube) are obtained. In analogy to the e constructions, the compounds listed below, where M = Si, can be prepared. The syntheses of the compounds below, where M = C, proceed entirely analogously with respect to their steps 2) and 3). The preparation of the carbon compounds which are analogous to the product of step 1) is described in German Patent 3,712,752.7.

5

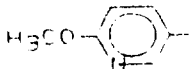
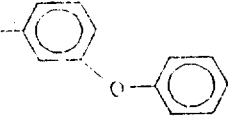
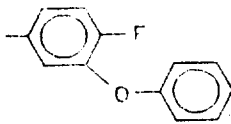
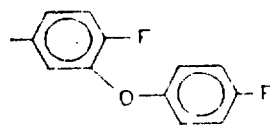
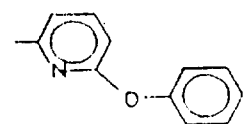
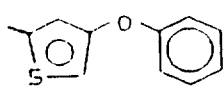
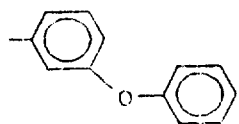
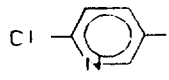
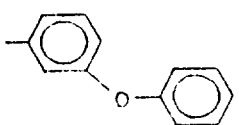
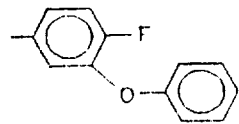
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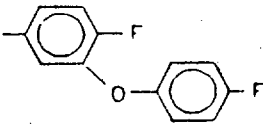
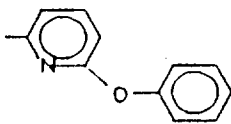
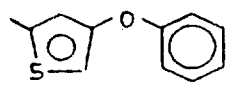
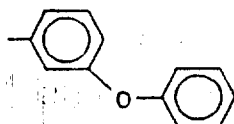
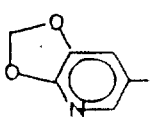
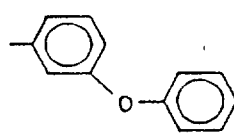
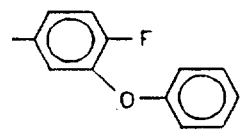
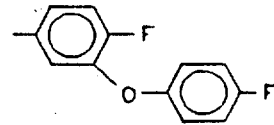
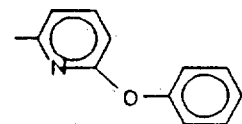
M = Si: R² = R³ = CH₃

N ^o .	R ¹	R ⁴	R ⁵	Physical data
1		H		b.p. _{0.1} = 240°C
2	"	H		b.p. _{0.03} = 235-240°C
3	"	H		
4	"	H		b.p. _{0.2} = 245-250°C
5	"	H		b.p. _{0.1} = 235-245°C
6		CN		highly viscous oil

Continuation of Table 1:

No.	R ¹	R ⁴	R ⁵	Physical data
7		H		
8		H		b.p. _{0.2} = 240-245°C
9	"	H		
10	"	H		
11	"	H		b.p. _{0.05} = 235-245°C
12	"	CN		
13		H		
14	"	H		b.p. _{0.01} = 220-225°C

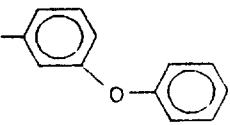
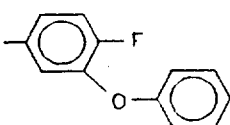
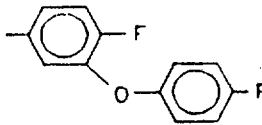
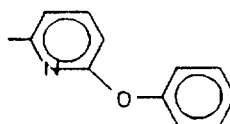
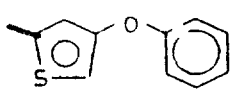
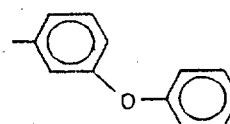
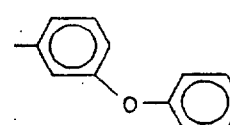
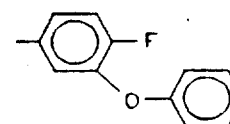
Continuation of Table 1:

No.	R ¹	R ⁴	R ⁵	Physical data
15	"	H		
16	"	H		
17	"	H		b.p. _{0.07} = 230°C
18	"	CN		
19		H		b.p. _{0.1} = 230-235°C
20	"	H		b.p. _{0.05} = 230-240°C
21	"	H		
22	"	H		b.p. _{0.02} = 240-250°C

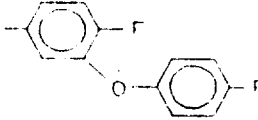
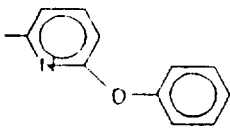
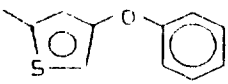
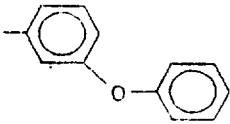
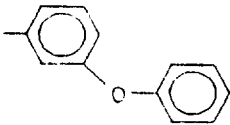
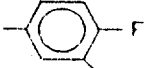
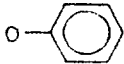
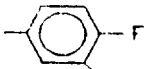
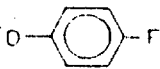
Continuation of Table I:

No.	R ¹	R ⁴	R ⁵	Physical data
23	"	H		
24	"	CN		
25	F ₃ C-CH ₂ -O-	H		b.p. _{0.2} = 245°C
26	"	H		b.p. _{0.05} = 230- 240°C
27	"	H		
28	"	H		
29	"	H		b.p. _{0.05} = 230- 240°C
30	"	CN		

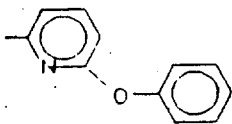
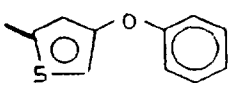
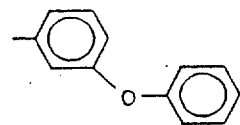
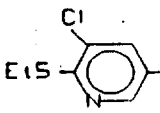
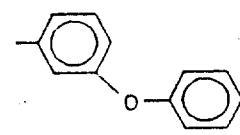
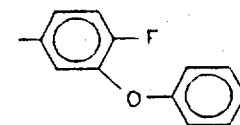
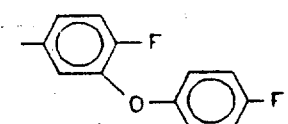
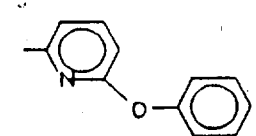
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31	$F-CH_2-CH_2-O-$ 	H		
32	"	H		b.p. _{0.07} = 235°C
33	"	H		
34	"	H		
35	"	H		
36	"	CN		
37	EtS- 	H		b.p. _{0.05} = 240-245°C
38	"	H		b.p. _{0.01} = 235-240°C

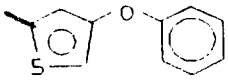
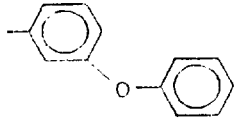
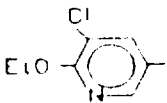
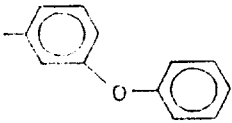
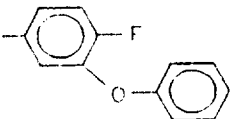
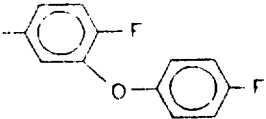
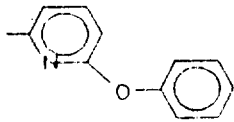
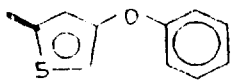
Continuation of Table 1:

No.	R ¹	R ⁴	R ⁵	Physical data
39	"	H		
40	"	H		
41	"	H		
42	"	CN		
43	F ₃ C-CH ₂ -S-		H	
44	"	H		
45	"	H		

Continuation of Table 1:

No.	R ¹	R ⁴	R ⁵	Physical data
46	"	H		
47	"	H		
48	"	CN		
49		H		
50	"	H		b.p. _{0.05} >250°C
51	"	H		
52	"	H		

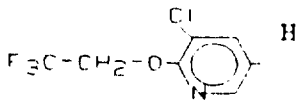
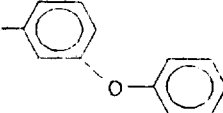
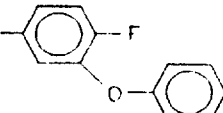
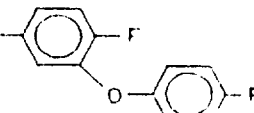
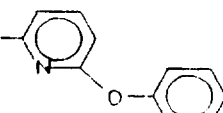
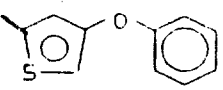
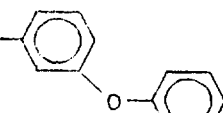
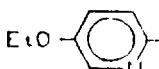
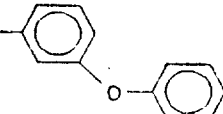
Continuation of Table 1:

No.	R ¹	R ⁴	R ⁵	Physical data
53	"	H		
54	"	CN		
55		H		b.p. _{0.01} = 250°C
56	"	H		b.p. _{0.01} = 250°C
57	"	H		
58	"	H		
59	"	H		

Continuation of Table 1:

No.	R ¹	R ⁴	R ⁵	Physical data
60	"	CN		
61		H		b.p. _{0.05} = 240-245°C
62	"	H		b.p. _{0.03} = 240-245°C
63	"	H		b.p. _{0.05} = 240-245°C
64	"	H		
65	"	H		b.p. _{0.05} = 240°C
66	"	CN		

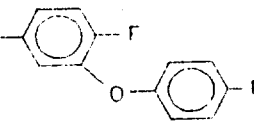
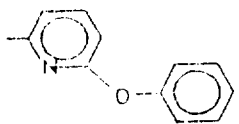
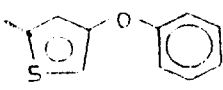
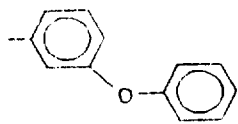
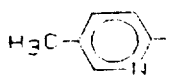
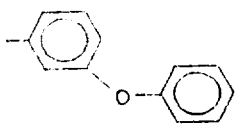
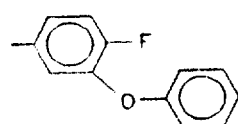
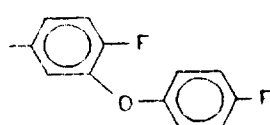
Continuation of Table 1:

No.	R ¹	R ⁴	R ⁵	Physical data
67		H		b.p. _{0.05} = 250°C
68	"	H		b.p. _{0.02} = 250°C
69	"	H		
70	"	H		
71	"	H		
72	"	CN		
73		H		

Continuation of Table 1:

No.	R ¹	R ⁴	R ⁵	Physical data
74	"	H		b.p. _{0.01} = 240-245°C
75	"	H		
76	"	H		
77	"	H		
78	"	CN		
79		H		
80	"	H		b.p. _{0.05} = 230-235°C

Continuation of Table 1:

No.	R ¹	R ⁴	R ⁵	Physical data
81	"	H		
82	"	H		
83	"	H		
84	"	CN		
85		H		b.p. _{0.02} = 240°C
86	"	H		
87	"	H		

26425

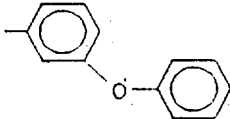
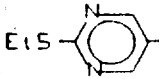
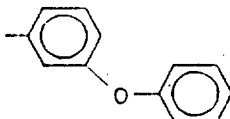
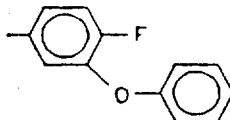
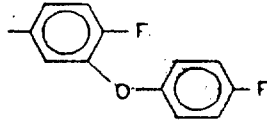
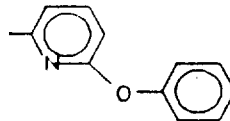
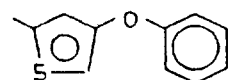
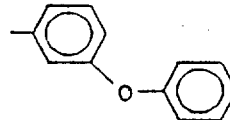
Continuation of Table 1:

No.	R ¹	R ⁴	R ⁵	Physical data
88	"	H		
89	"	H		
90	"	CN		
91		H		b.p. _{0.02} = 235- 240°C
92	"	H		
93	"	H		
94	"	H		

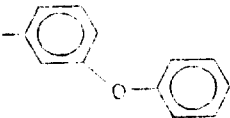
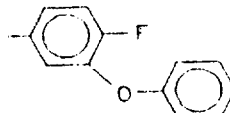
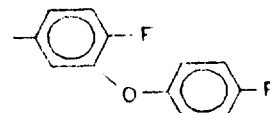
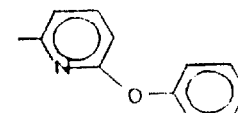
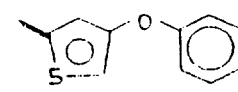
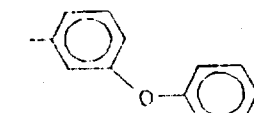
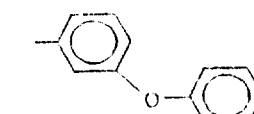
Continuation of Table 1:

No.	R ¹	R ⁴	R ⁵	Physical data
95	"	H		
96	"	CN		
97		H		pale yellow oil R _f (CH ₂ Cl ₂ / heptane 4:1) = 0.25
98	"	H		pale yellow oil R _f (CH ₂ Cl ₂ / heptane 4:1) = 0.25
99	"	H		
100	"	H		
101	"	H		

Continuation of Table 1:

No.	R ¹	R ⁴	R ⁵	Physical data
102	"	CN		
103	EtS- 	H		
104	"	H		
105	"	H		
106	"	H		
107	"	H		
108	"	CN		

Continuation of Table 1:

No.	R ¹	R ⁴	R ⁵	Physical data
109	$\text{F}_3\text{C}-\text{CH}_2-\text{O}-$		H	 pale yellow oil $R_f(\text{CH}_2\text{Cl}_2)=0.3$
110	"	H	 pale yellow oil $R_f(\text{CH}_2\text{Cl}_2)=0.3$	
111	"	H		
112	"	H		
113	"	H		
114	"	CN		
115	$\text{F}-\text{CH}_2-\text{CH}_2-\text{O}-$		H	

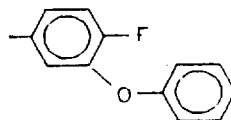
Continuation of Table 1:

No.	R ¹	R ⁴	R ⁵	Physical data
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116

"

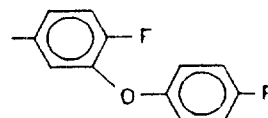
H



117

"

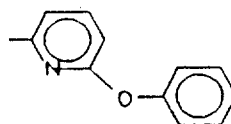
H



118

"

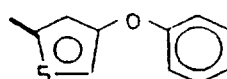
H



119

"

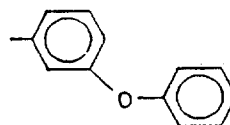
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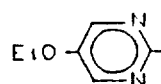
120

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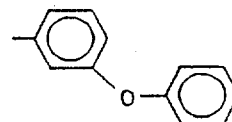
CN



121



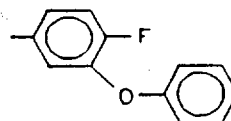
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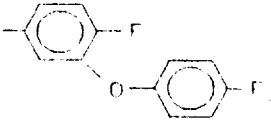
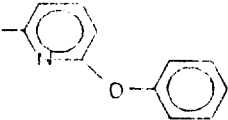
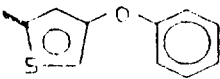
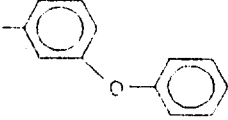
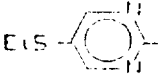
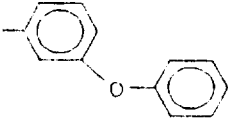
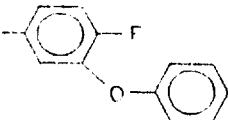
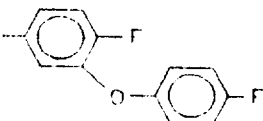
122

"

H



Continuation of Table 1:

No.	R ¹	R ⁴	R ⁵	Physical data
123	"	H		
124	"	H		
125	"	H		
126	"	CN		
127		H		
128	"	H		
129	"	H		

Continuation of Table 1:

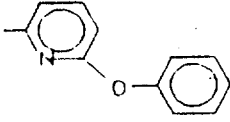
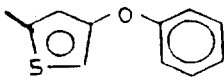
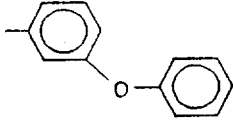
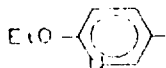
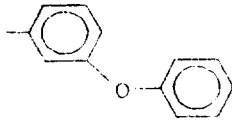
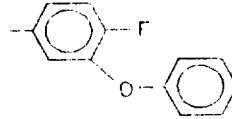
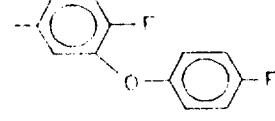
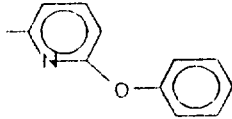
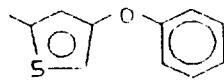
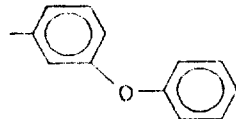
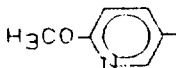
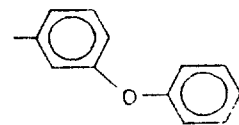
No.	R ¹	R ⁴	R ⁵	Physical data
130	"	H		
131	"	H		
132	"	CN		

Table 2:

 $M = C; R^2 = R^3 = CH_3$

No.	R^1	R^4	R^5	Physical data
133		H		b.p. _{0.001} = 224-230°C
134	"	H		b.p. _{0.05} = 230-235°C
135	"	H		
136	"	H		
137	"	H		b.p. _{0.0005} = 210°C b.p. _{0.01} = 220-230°C
138	"	CN		
139		H		

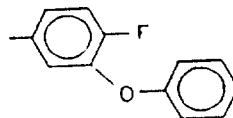
Continuation of Table 2:

No.	R ¹	R ⁴	R ⁵	Physical data
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140

"

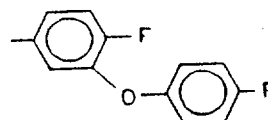
H



141

"

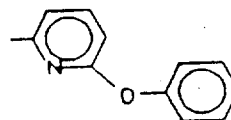
H



142

"

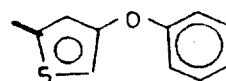
H



143

"

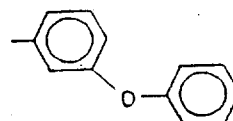
H



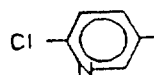
144

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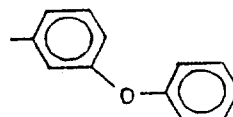
CN



145



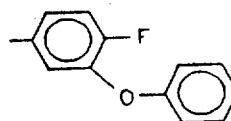
H



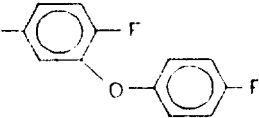
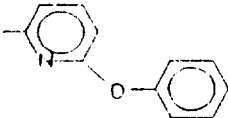
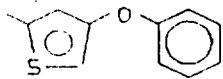
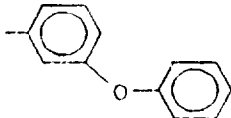
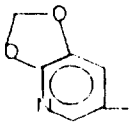
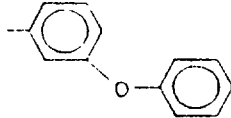
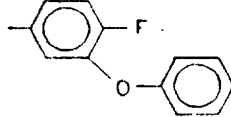
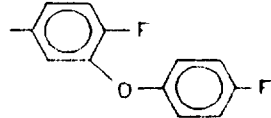
146

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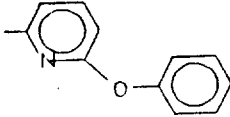
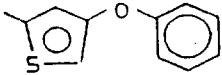
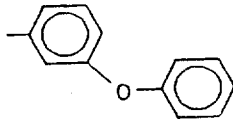
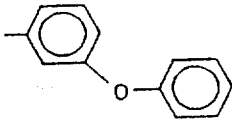
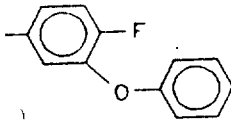
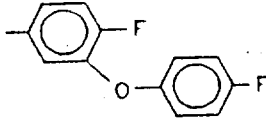
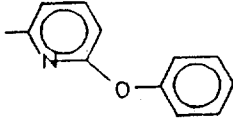
H



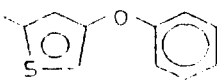
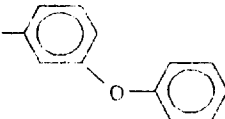
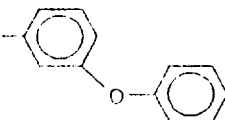
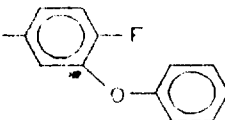
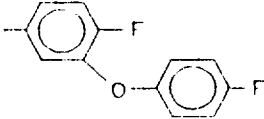
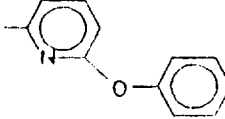
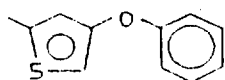
Continuation of Table 2:

No.	R ¹	R ⁴	R ⁵	Physical data
147	"	H		
148	"	H		
149	"	H		
150	"	CN		
151		H		
152	"	H		
153	"	H		

Continuation of Table 2:

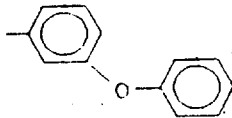
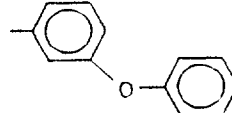
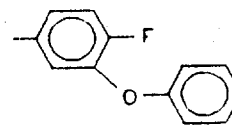
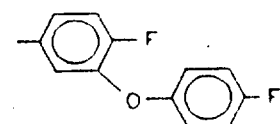
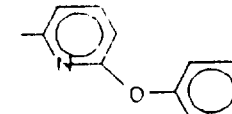
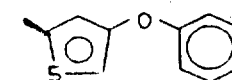
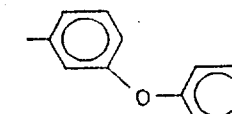
No.	R ¹	R ⁴	R ⁵	Physical data
154	"	H		
155	"	H		
156	"	CN		
157	F ₃ C-CH ₂ -O-	H		b.p. _{0.001} = 225-235°C
158	"	H		bp. _{0.01} = 220-230°C
159	"	H		b.p. _{0.01} = 230-235°C
160	"	H		

Continuation of Table 2:

No.	R ¹	R ⁴	R ⁵	Physical data	
161	"	H			
162	"	CN			
163	F-CH ₂ -CH ₂ -O-		H		
164	"	H		b.p. _{0.001} = 235°C	
165	"	H			
166	"	H			
167	"	H			

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Continuation of Table 2:

No.	R ¹	R ⁴	R ⁵	Physical data
168	"	CN		
169	EtS- 	H		
170	"	H		
171	"	H		
172	"	H		
173	"	H		
174	"	CN		

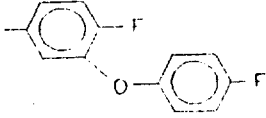
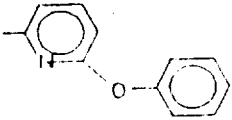
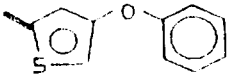
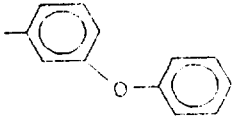
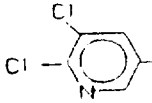
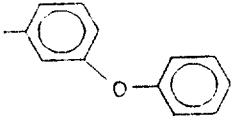
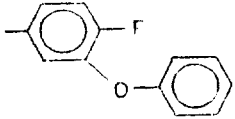
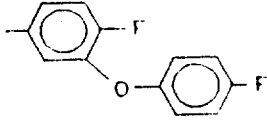
Continuation of Table 2:

No.	R ¹	R ⁴	R ⁵	Physical data
175	<chem>F3C-CH2-S-c1ccc[nH]1</chem>	H	<chem>c1ccc(Oc2ccccc2)cc1</chem>	
176	"	H	<chem>Fc1ccc(Oc2ccccc2)cc1</chem>	
177	"	H	<chem>Fc1ccc(Oc2ccc(F)cc2)cc1</chem>	
178	"	H	<chem>c1ccc(Oc2c[nH]cc3ccccc23)cc1</chem>	
179	"	H	<chem>c1ccc(Oc2cc3ccccc3s2)cc1</chem>	
180	"	CN	<chem>c1ccc(Oc2ccccc2)cc1</chem>	
181	<chem>Clc1cc(Cl)ccc1S(=O)(=O)C</chem>	H	<chem>c1ccc(Oc2ccccc2)cc1</chem>	b.p. _{0.001} = 230-240°C

Continuation of Table 2:

No.	R ¹	R ⁴	R ⁵	Physical data
182	"	H		b.p. _{0.001} = 240°C
183	"	H		
184	"	H		
185	"	H		
186	"	CN		
187		H		b.p. _{0.001} = 235- 240°C
188	"	H		b.p. _{0.003} = 240°C

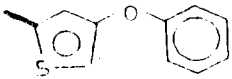
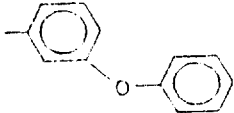
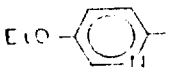
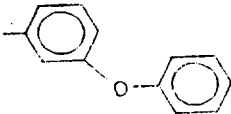
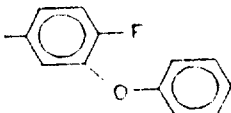
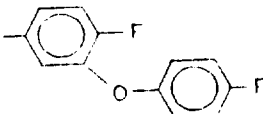
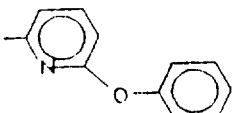
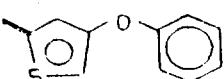
Continuation of Table 2:

No.	R ¹	R ⁴	R ⁵	Physical data
189	"	H		
190	"	H		
191	"	H		
192	"	CN		
193		H		
194	"	H		
195	"	H		

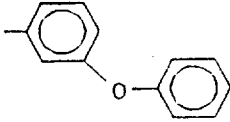
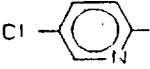
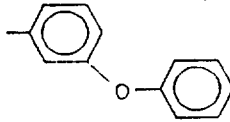
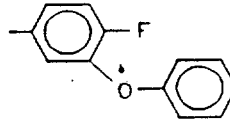
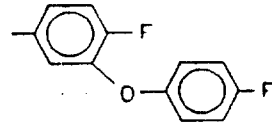
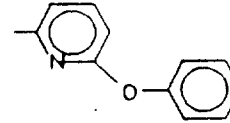
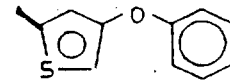
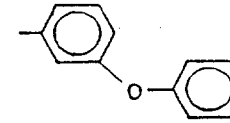
Continuation of Table 2:

No.	R ¹	R ⁴	R ⁵	Physical data
196	"	H		
197	"	H		
198	"	CN		
199		H		b.p. _{0.001} = 220-225°C
200	"	H		b.p. _{0.003} = 230°C
201	"	H		
202	"	H		

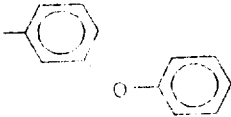
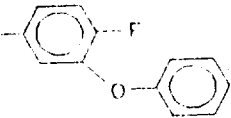
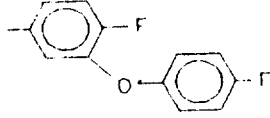
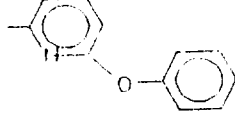
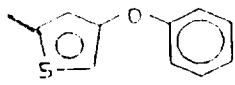
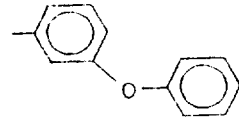
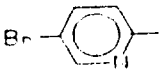
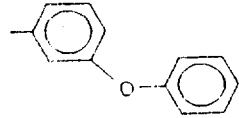
Continuation of Table 2:

No.	R ¹	R ⁴	R ⁵	Physical data
203	"	H		
204	"	CN		
205				
206	"	H		
207	"	H		
208	"	H		
209	"	H		

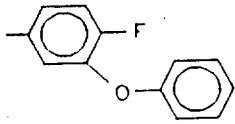
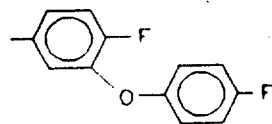
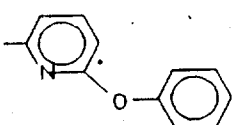
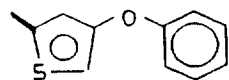
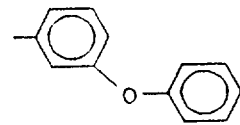
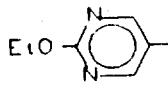
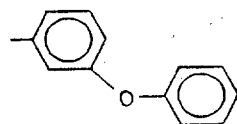
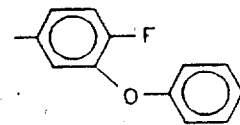
Continuation of Table 2:

No.	R ¹	R ⁴	R ⁵	Physical data
210	"	CN		
211		H		
212	"	H		
213	"	H		
214	"	H		
215	"	H		
216	"	CN		

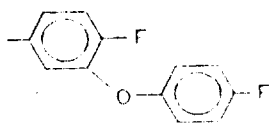
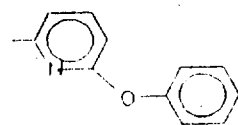
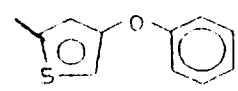
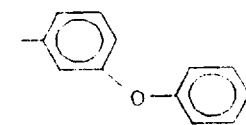
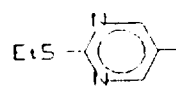
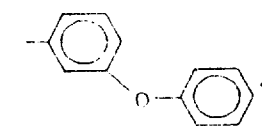
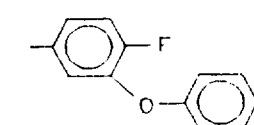
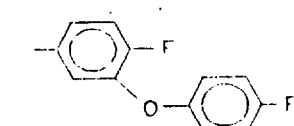
Continuation of Table 2:

No.	R ¹	R ⁴	R ⁵	Physical data
217		H		
218	"	H		
219	"	H		
220	"	H		
221	"	H		
222	"	CN		
223		H		

Continuation of Table 2:

No.	R ¹	R ⁴	R ⁵	Physical data
224	"	H		
225	"	H		
226	"	H		
227	"	H		
228	"	CN		
229				
230	"	H		

Continuation of Table 2:

No.	R ¹	R ⁴	R ⁵	Physical data
231	"	H		
232	"	H		
233	"	H		
234	"	CN		
235	EtS- 	H		
236	"	H		
237	"	H		

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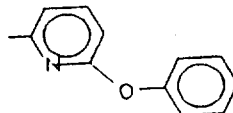
Continuation of Table 2:

No.	R ¹	R ⁴	R ⁵	Physical data
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"

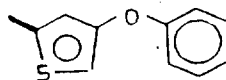
H



239

"

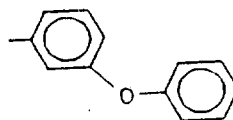
H



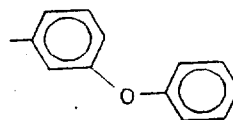
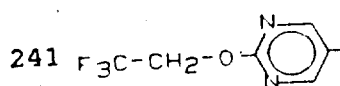
240

"

CN



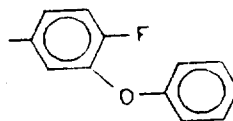
241



242

"

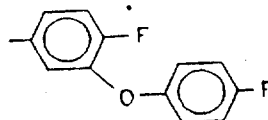
H



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"

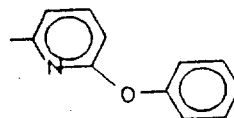
H



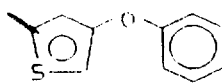
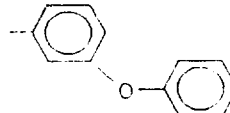
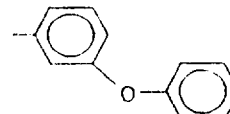
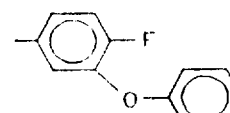
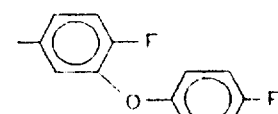
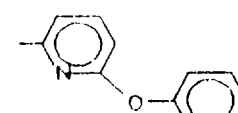
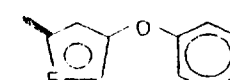
244

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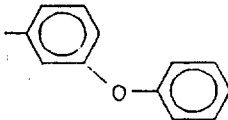
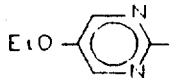
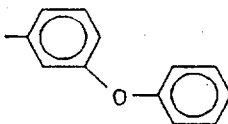
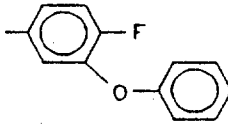
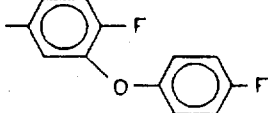
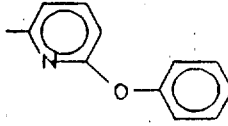
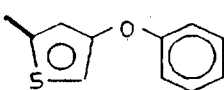
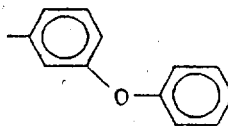
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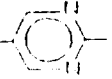
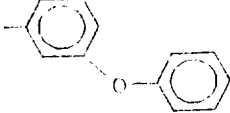
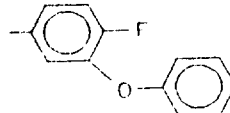
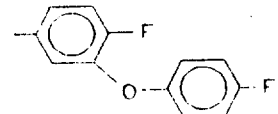
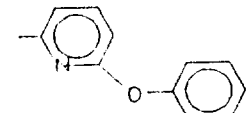
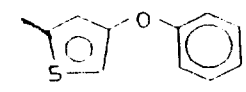
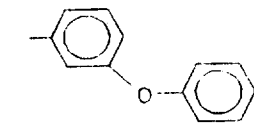
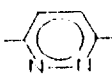
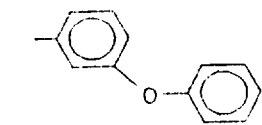
Continuation of Table 2:

No.	R ¹	R ⁴	R ⁵	Physical data	
245	"	H			
246	"	CN			
247	$\text{F}-\text{CH}_2-\text{CH}_2-\text{O}-$		H		
248	"	H			
249	"	H			
250	"	H			
251	"	H			

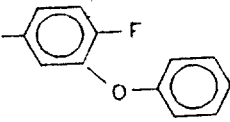
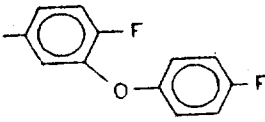
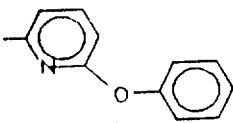
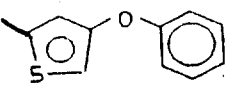
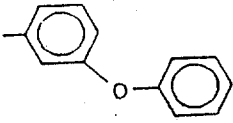
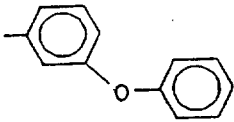
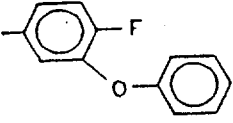
Continuation of Table 2:

No.	R ¹	R ⁴	R ⁵	Physical data
252	"	CN		
253		H		
254	"	H		
255	"	H		
256	"	H		
257	"	H		
258	"	CN		

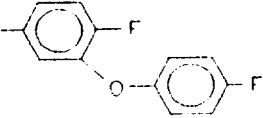
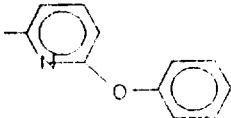
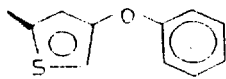
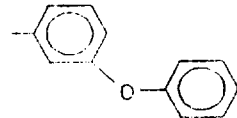
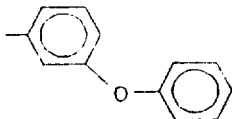
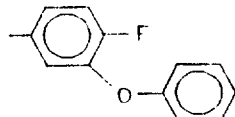
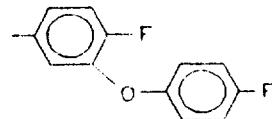
Continuation of Table 2:

No.	R ¹	R ⁴	R ⁵	Physical data
259	EtS- 	H		
260	"	H		
261	"	H		
262	"	H		
263	"	H		
264	"	CN		
265	EtO- 	H		

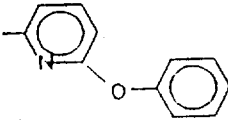
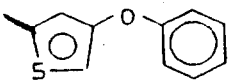
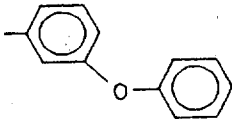
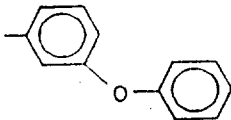
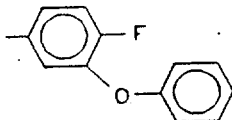
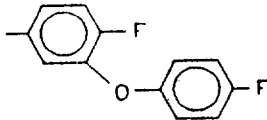
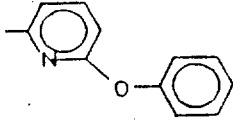
Continuation of Table 2:

No.	R ¹	R ⁴	R ⁵	Physical data
266	"	H		
267	"	H		
268	"	H		
269	"	H		
270	"	CN		
271		H		
272	"	H		

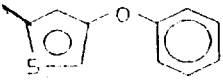
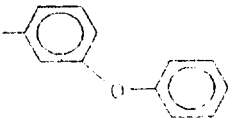
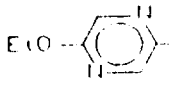
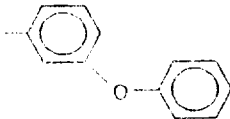
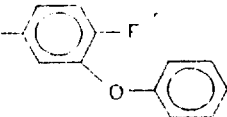
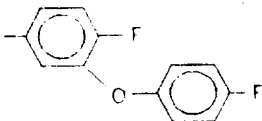
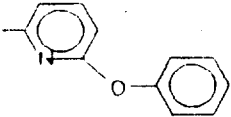
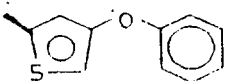
Continuation of Table 2:

No.	R ¹	R ⁴	R ⁵	Physical data
273	"	H		
274	"	H		
275	"	H		
276	"	CN		
277	EtS- 	H		
278	"	H		
279	"	H		

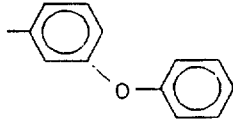
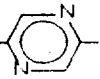
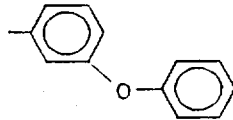
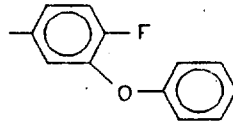
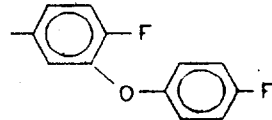
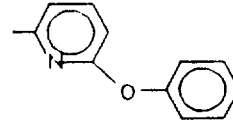
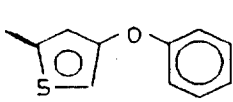
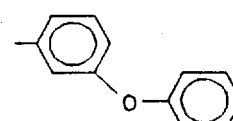
Continuation of Table 2:

No.	R ¹	R ⁴	R ⁵	Physical data
280	"	H		
281	"	H		
282	"	CN		
283	$\text{F}_3\text{C}-\text{CH}_2-\text{O}-$	H		
284	"	H		
285	"	H		
286	"	H		

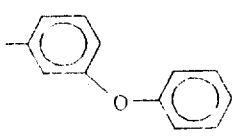
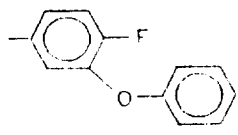
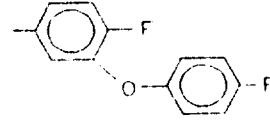
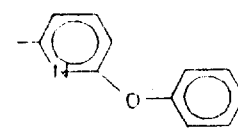
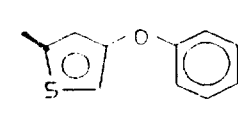
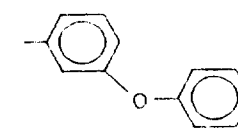
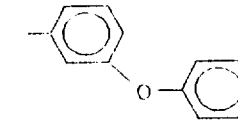
Continuation of Table 2:

No.	R ¹	R ⁴	R ⁵	Physical data
287	"	H		
288	"	CN		
289		H		
290	"	H		
291	"	H		
292	"	H		
293	"	H		

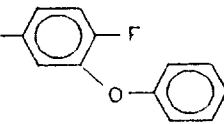
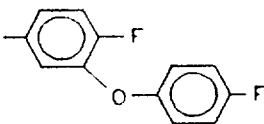
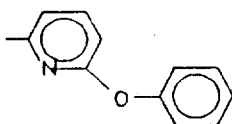
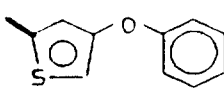
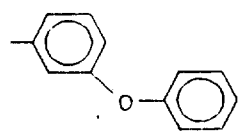
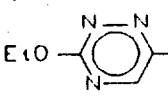
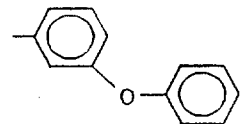
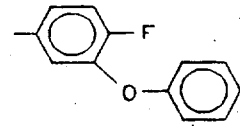
Continuation of Table 2:

No.	R ¹	R ⁴	R ⁵	Physical data
294	"	CN		
295	EtS- 	H		
296	"	H		
297	"	H		
298	"	H		
299	"	H		
300	"	CN		

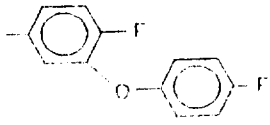
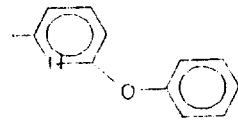
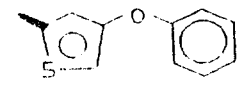
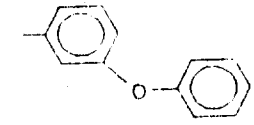
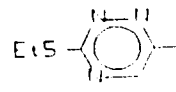
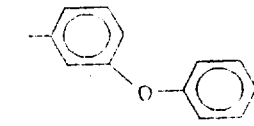
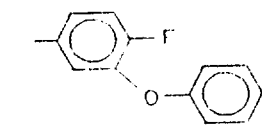
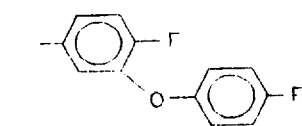
Continuation of Table 2:

No.	R ¹	R ⁴	R ⁵	Physical data
301	$\text{F}_3\text{C}-\text{CH}_2-\text{O}-$ 	H		
302	"	H		
303	"	H		
304	"	H		
305	"	H		
306	"	CN		
307	$\text{Br}-$ 	H		

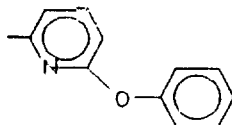
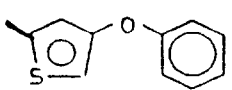
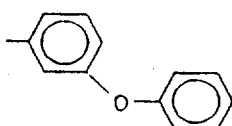
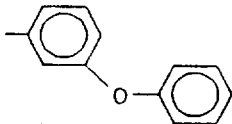
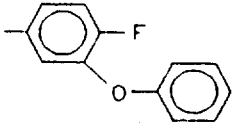
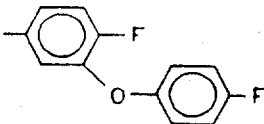
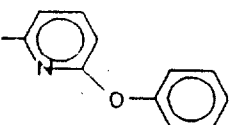
Continuation of Table 2:

No.	R ¹	R ⁴	R ⁵	Physical data
308	"	H		
309	"	H		
310	"	H		
311	"	H		
312	"	CN		
313	EtO 	H		
314	"	H		

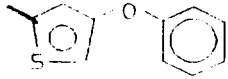
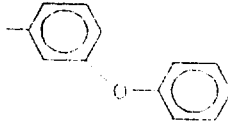
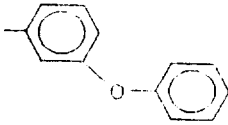
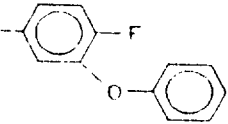
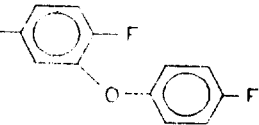
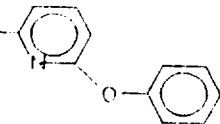
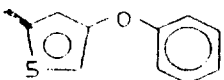
Continuation of Table 2:

No.	R ¹	R ⁴	R ⁵	Physical data
315	"	H		
316	"	H		
317	"	H		
318	"	CN		
319	EtS 	H		
320	"	H		
321	"	H		

Continuation of Table 2:

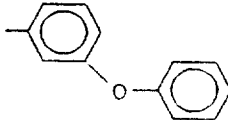
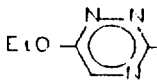
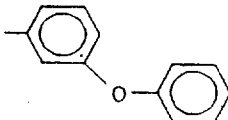
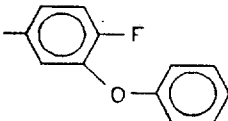
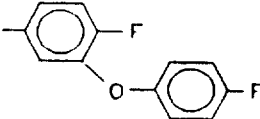
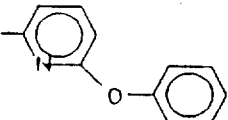
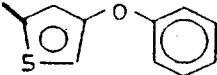
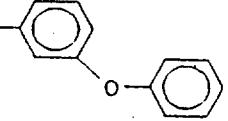
No.	R ¹	R ⁴	R ⁵	Physical data
322	"	H		
323	"	H		
324	"	CN		
325	$\text{F}_3\text{C}-\text{CH}_2-\text{O}-$ 	H		
326	"	H		
327	"	H		
328	"	H		

Continuation of Table 2:

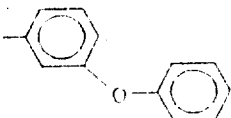
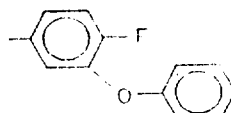
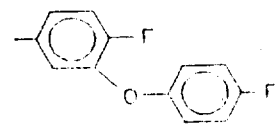
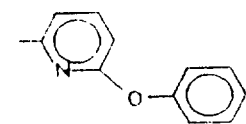
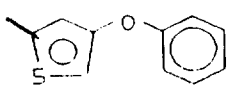
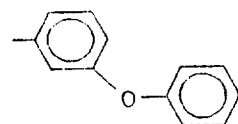
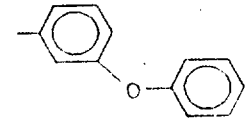
No.	R ¹	R ⁴	R ⁵	Physical data
329	"	H		
330	"	CN		
331		H		
332	"	H		
333	"	H		
334	"	H		
335	"	H		

26425

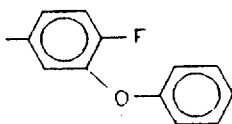
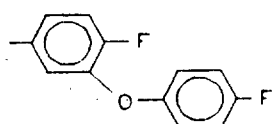
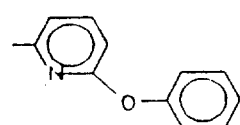
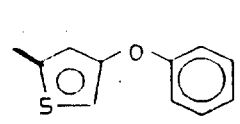
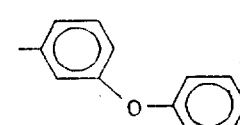
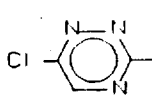
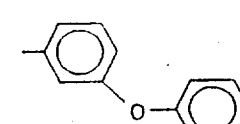
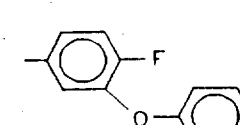
Continuation of Table 2:

No.	R ¹	R ⁴	R ⁵	Physical data
336	"	CN		
337	EtO- 	H		
338	"	H		
339	"	H		
340	"	H		
341	"	H		
342	"	CN		

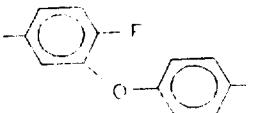
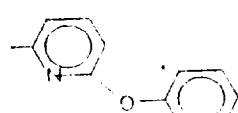
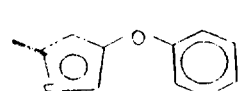
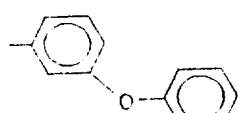
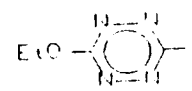
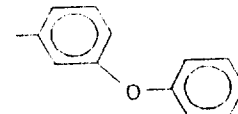
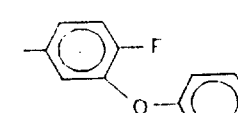
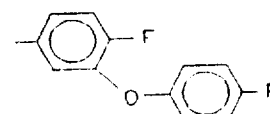
Continuation of Table 2:

No.	R ¹	R ⁴	R ⁵	Physical data
343	EtS- 	H		
344	"	H		
345	"	H		
346	"	H		
347	"	H		
348	"	CN		
349	3C-CH ₂ -O- 	H		

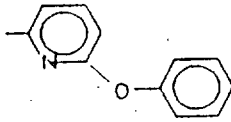
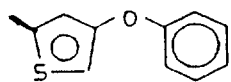
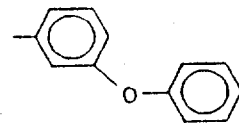
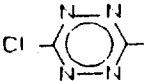
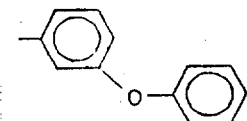
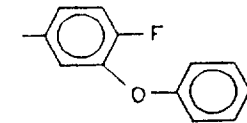
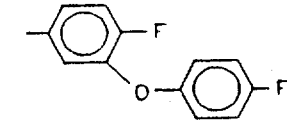
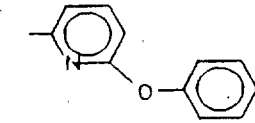
Continuation of Table 2:

No.	R ¹	R ⁴	R ⁵	Physical data
350	"	H		
351	"	H		
352	"	H		
353	"	H		
354	"	CN		
355		H		
356	"	H		

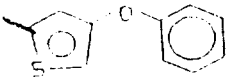
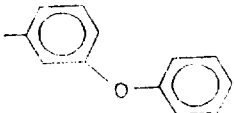
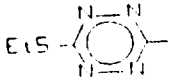
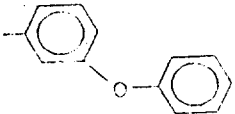
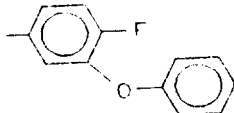
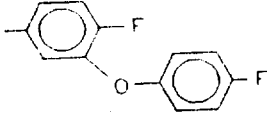
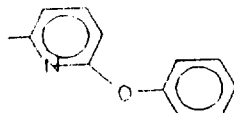
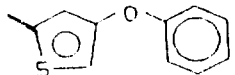
Continuation of Table 2:

No.	R ¹	R ⁴	R ⁵	Physical data
357	"	H		
358	"	H		
359	"	H		
360	"	CN		
361		H		
362	"	H		
363	"	H		

Continuation of Table 2:

No.	R ¹	R ⁴	R ⁵	Physical data
364	"	H		
365	"	H		
366	"	CN		
367		H		
368	"	H		
369	"	H		
370	"	H		

Continuation of Table 2:

No.	R ¹	R ⁴	R ⁵	Physical data
371	"	H		
372	"	CN		
373		H		
374	"	H		
375	"	H		
376	"	H		
377	"	H		

Continuation of Table 2:

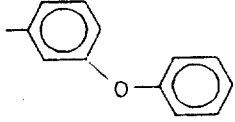
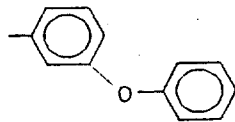
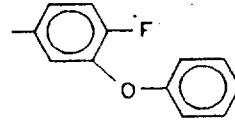
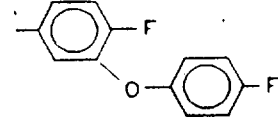
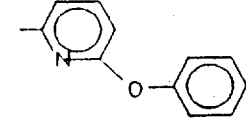
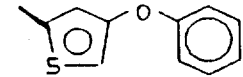
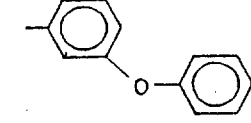
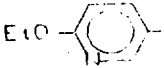
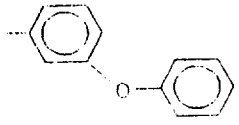
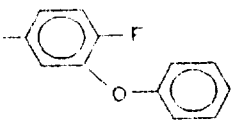
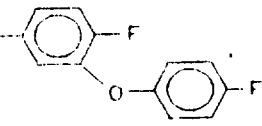
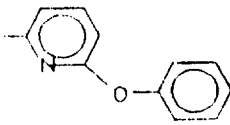
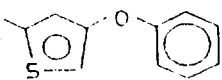
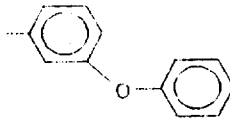
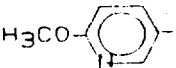
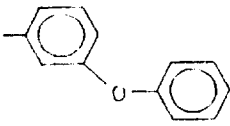
No.	R ¹	R ⁴	R ⁵	Physical data
378	"	CN		
379	$\text{F}_3\text{C}-\text{CH}_2-\text{O}-\text{C}_6\text{H}_3\text{N}_2$	H		
380	"	H		
381	"	H		
382	"	H		
383	"	H		
384	"	CN		

Table 3: $M = C$; $R^2-R^3 = -CH_2-CH_2-$

No.	R^1	R^4	R^5	Physical data
385		H		
386	"	H		
387	"	H		
388	"	H		
389	"	H		
390	"	CN		
391		H		

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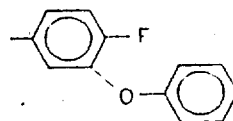
Continuation of Table 3:

No.	R ¹	R ⁴	R ⁵	Physical data
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"

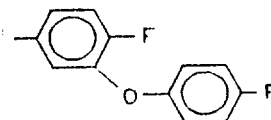
H



393

"

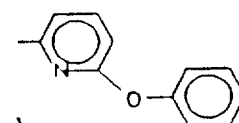
H



394

"

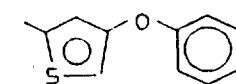
H



395

"

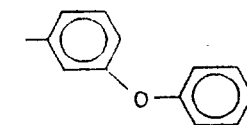
H



396

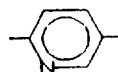
"

CN

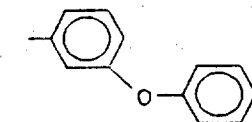


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Cl



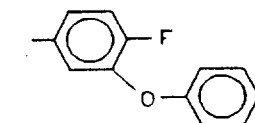
H



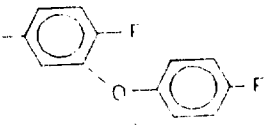
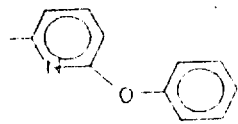
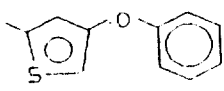
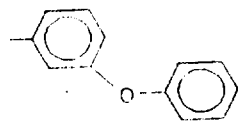
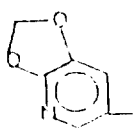
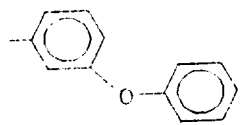
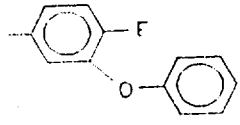
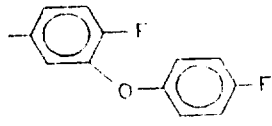
398

"

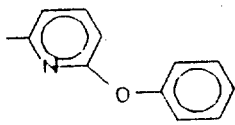
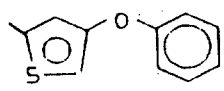
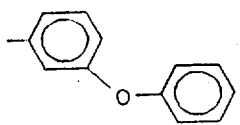
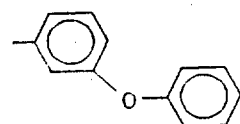
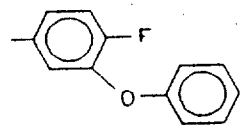
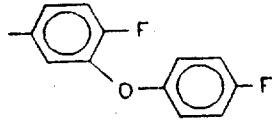
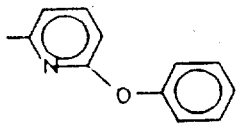
H



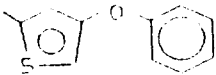
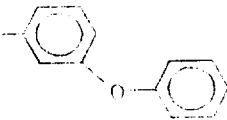
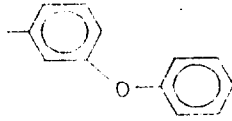
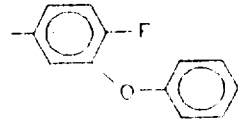
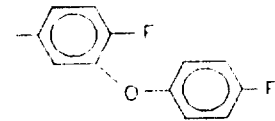
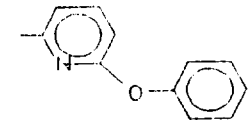
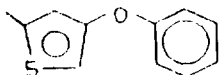
Continuation of Table 3:

No.	R ¹	R ⁴	R ⁵	Physical data
399	"	H		
400	"	H		
401	"	H		
402	"	CN		
403		H		
404	"	H		
405	"	H		

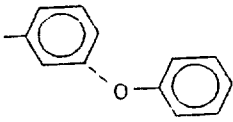
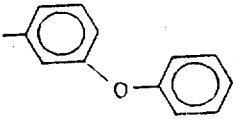
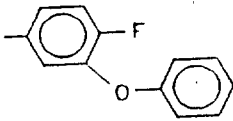
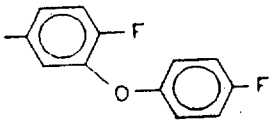
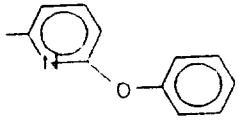
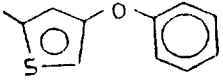
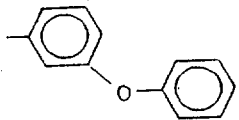
Continuation of Table 3:

No.	R ¹	R ⁴	R ⁵	Physical data
406	"	H		
407	"	H		
408	"	CN		
409	$\text{F}_3\text{C}-\text{CH}_2-\text{O}-$	H		
410	"	H		
411	"	H		
412	"	H		

Continuation of Table 3:

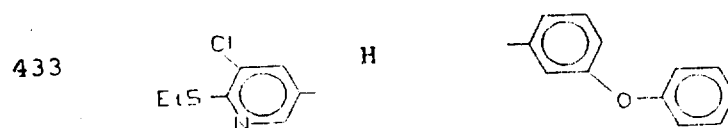
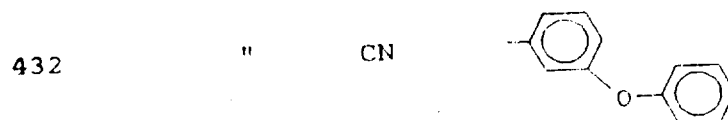
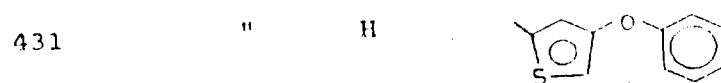
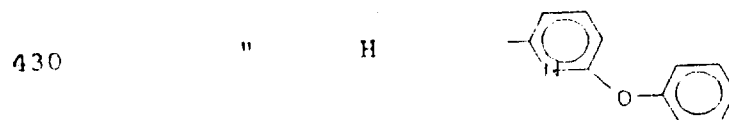
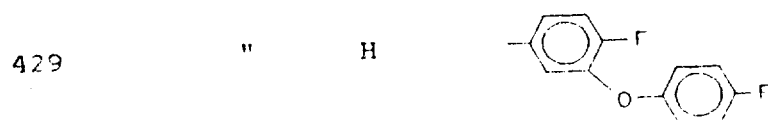
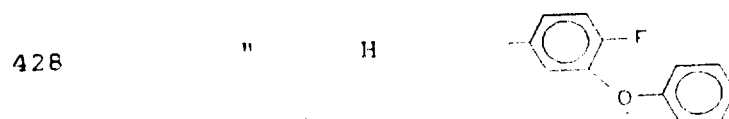
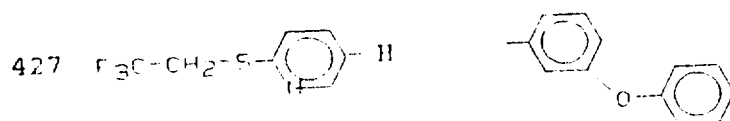
No.	R ¹	R ⁴	R ⁵	Physical data
413	"	H		
414	"	CN		
415	F-CH ₂ -CH ₂ -O-	H		
416	"	H		
417	"	H		
418	"	H		
419	"	H		

Continuation of Table 3:

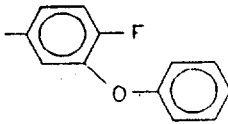
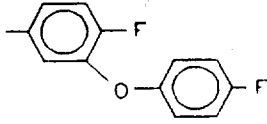
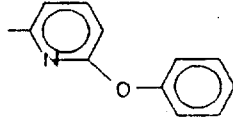
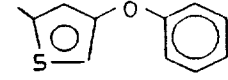
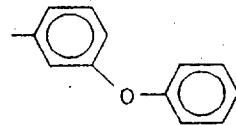
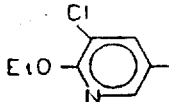
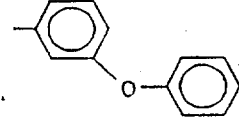
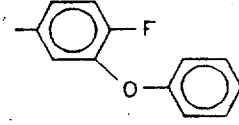
No.	R ¹	R ⁴	R ⁵	Physical data
420	"	CN		
421	EtS 	H		
422	"	H		
423	"	H		
424	"	H		
425	"	H		
426	"	CN		

Continuation of Table 3:

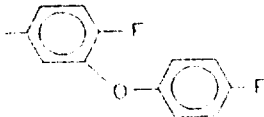
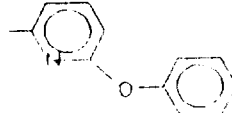
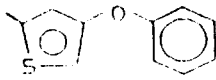
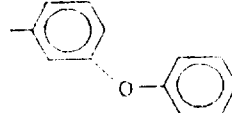
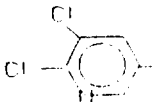
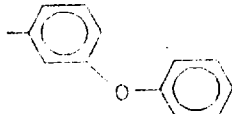
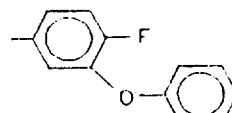
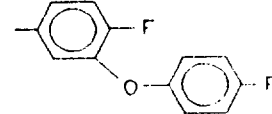
No.	R ¹	R ⁴	R ⁵	Physical data
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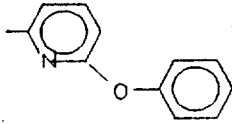
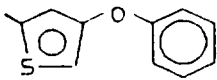
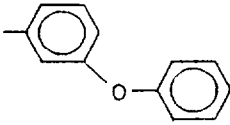
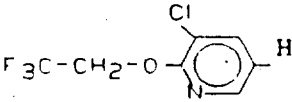
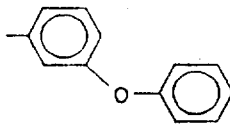
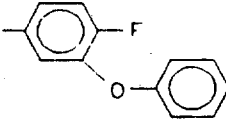
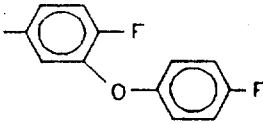
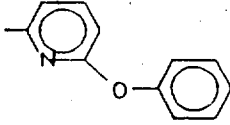
Continuation of Table 3:

No.	R ¹	R ⁴	R ⁵	Physical data
434	"	H		
435	"	H		
436	"	H		
437	"	H		
438	"	CN		
439		H		
440	"	H		

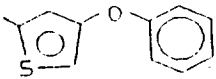
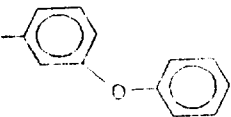
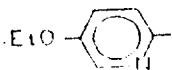
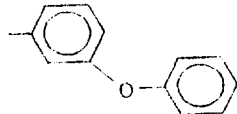
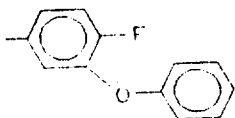
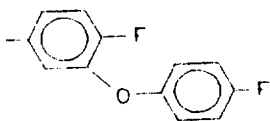
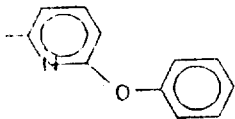
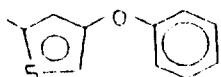
Continuation of Table 3:

No.	R ¹	R ⁴	R ⁵	Physical data
441	"	H		
442	"	H		
443	"	H		
444	"	CN		
445		H		
446	"	H		
447	"	H		

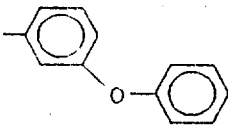
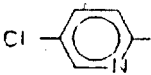
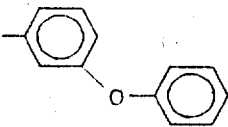
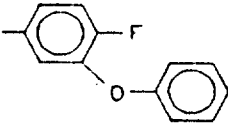
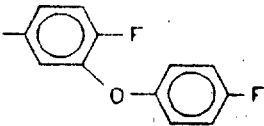
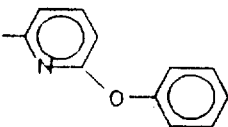
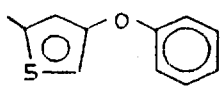
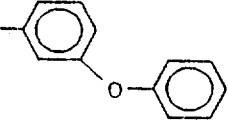
Continuation of Table 3:

No.	R ¹	R ⁴	R ⁵	Physical data
448	"	H		
449	"	H		
450	"	CN		
451		H		
452	"	H		
453	"	H		
454	"	H		

Continuation of Table 3:

No.	R ¹	R ⁴	R ⁵	Physical data
455	"	H		
456	"	CN		
457				
458	"	H		
459	"	H		
460	"	H		
461	"	H		

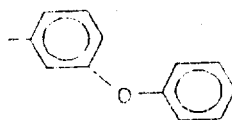
Continuation of Table 3:

No.	R ¹	R ⁴	R ⁵	Physical data
462	"	CN		
463	Cl 	H		
464	"	H		
465	"	H		
466	"	H		
467	"	H		
468	"	CN		

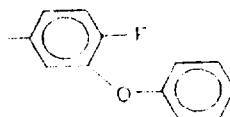
Continuation of Table 3:

No.	R ¹	R ⁴	R ⁵	Physical data
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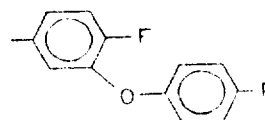
469



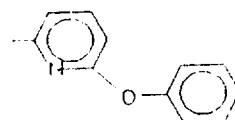
470



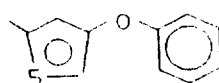
471



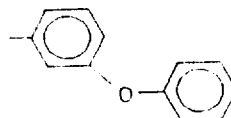
472



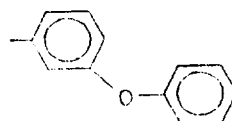
473



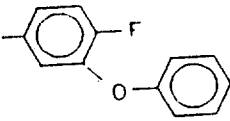
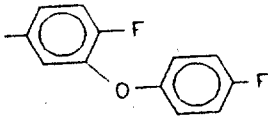
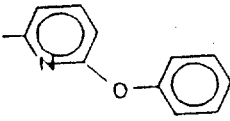
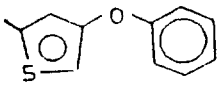
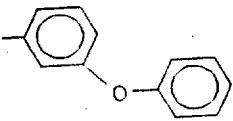
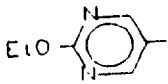
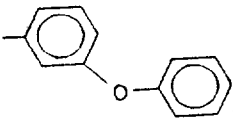
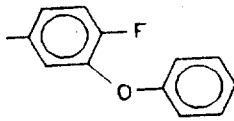
474



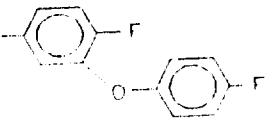
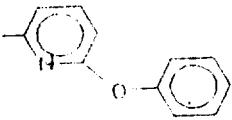
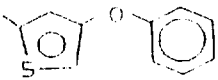
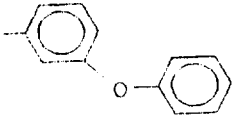
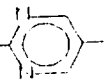
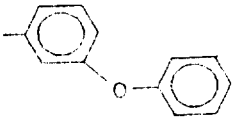
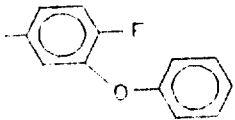
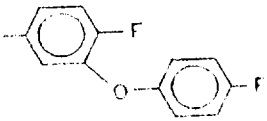
475



Continuation of Table 3:

No.	R ¹	R ⁴	R ⁵	Physical data
476	"	H		
477	"	H		
478	"	H		
479	"	H		
480	"	CN		
481		"		
482	"	H		

Continuation of Table 3:

No.	R ¹	R ⁴	R ⁵	Physical data
483	"	H		
484	"	H		
485	"	H		
486	"	CN		
487	EtS- 	H		
488	"	H		
489	"	H		

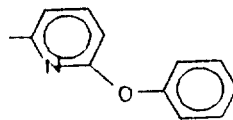
Continuation of Table 3:

No.	R ¹	R ⁴	R ⁵	Physical data
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490

"

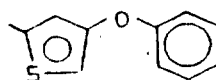
H



491

"

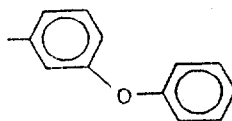
H



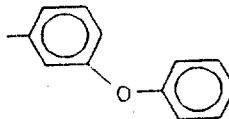
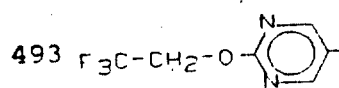
492

"

CN



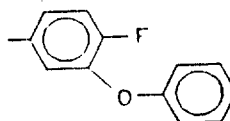
493



494

"

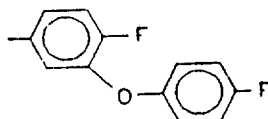
H



495

"

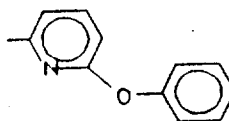
H



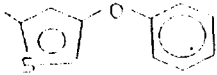
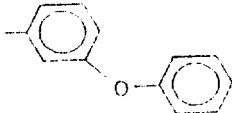
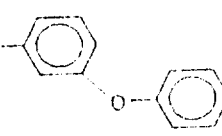
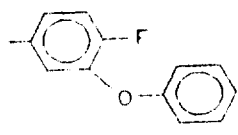
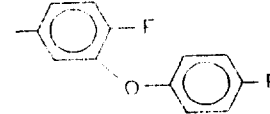
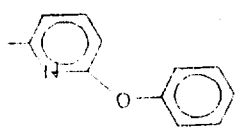
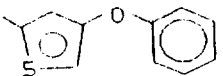
496

"

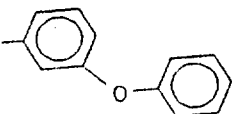
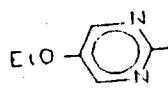
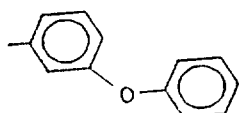
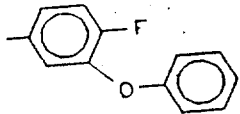
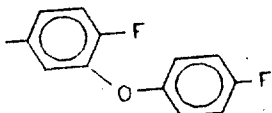
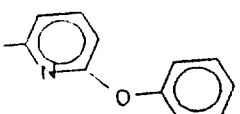
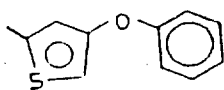
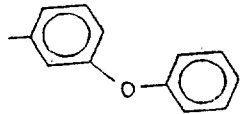
H



continuation of Table 3:

No.	R ¹	R ⁴	R ⁵	Physical data
497	"	H		
498	"	CN		
499	$\text{CH}_2\text{CH}_2\text{O}$		H	
500	"	H		
501	"	H		
502	"	H		
503	"	H		

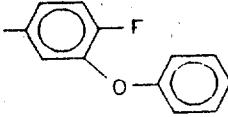
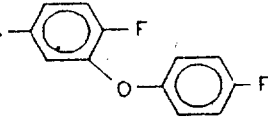
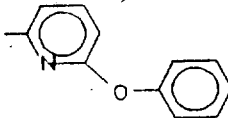
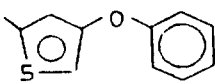
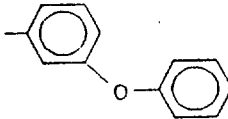
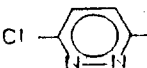
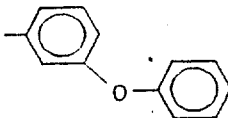
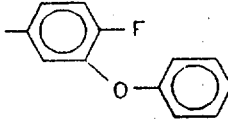
Continuation of Table 3:

No.	R ¹	R ⁴	R ⁵	Physical data
504	"	CN		
505		H		
506	"	H		
507	"	H		
508	"	H		
509	"	H		
510	"	CN		

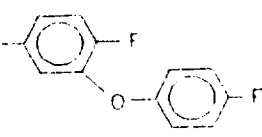
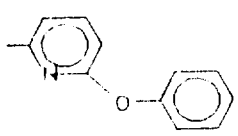
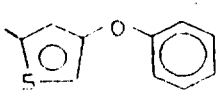
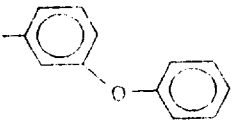
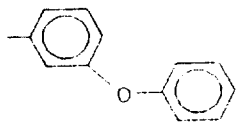
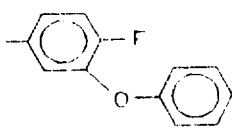
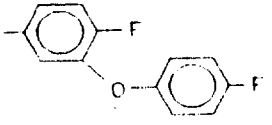
Continuation of Table 3:

No.	R ¹	R ⁴	R ⁵	Physical data
511		H		
512	"	H		
513	"	H		
514	"	H		
515	"	H		
516	"	CN		
517		H		

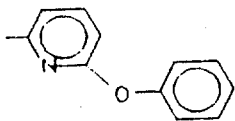
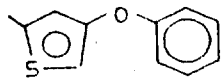
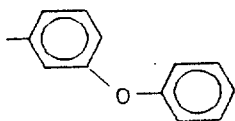
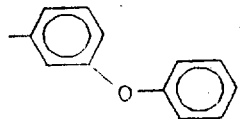
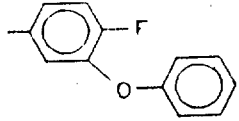
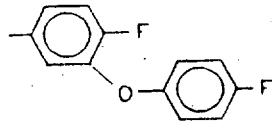
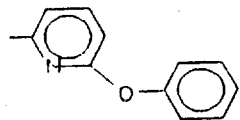
Continuation of Table 3:

No.	R ¹	R ⁴	R ⁵	Physical data
518	"	H		
519	"	H		
520	"	H		
521	"	H		
522	"	CN		
523		H		
524	"	H		

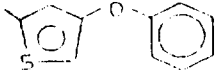
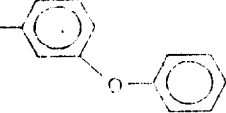
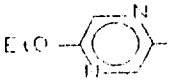
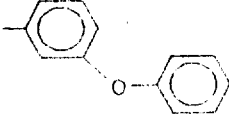
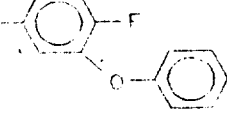
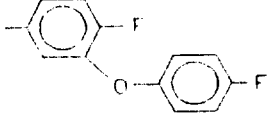
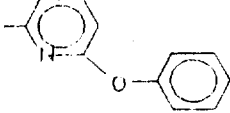
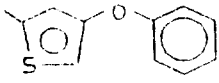
Continuation of Table 3:

No.	R ¹	R ⁴	R ⁵	Physical data
525	"	H		
526	"	H		
527	"	H		
528	"	CN		
529	EtS- 	H		
530	"	H		
531	"	H		

Continuation of Table 3:

No.	R ¹	R ⁴	R ⁵	Physical data
532	"	H		
533	"	H		
534	"	CN		
535	$\text{F}_3\text{C}-\text{CH}_2-\text{O}-$ 	H		
536	"	H		
537	"	H		
538	"	H		

Continuation of Table 3:

No.	R ¹	R ⁴	R ⁵	Physical data
539	"	H		
540	"	CN		
541		H		
542	"	H		
543	"	H		
544	"	H		
545	"	H		

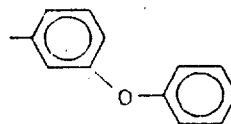
Continuation of Table 3:

No.	R ¹	R ⁴	R ⁵	Physical data
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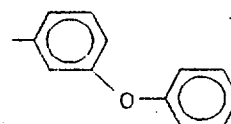
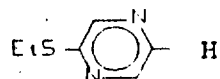
546

"

CN



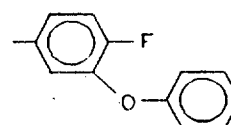
547



548

"

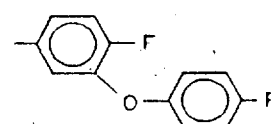
H



549

"

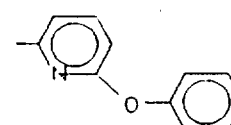
H



550

"

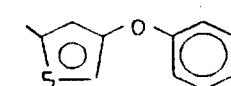
H



551

"

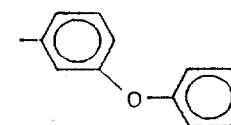
H



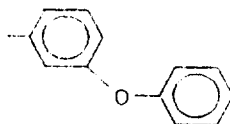
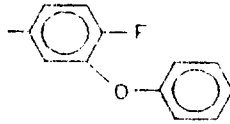
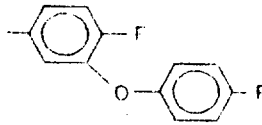
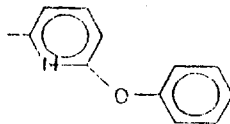
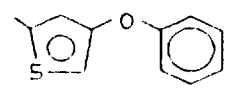
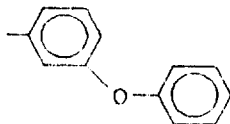
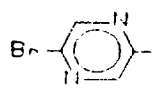
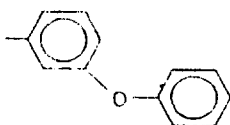
552

"

CN



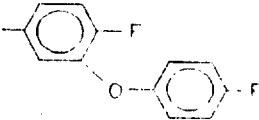
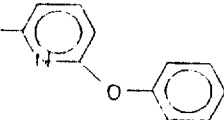
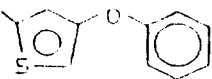
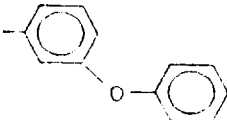
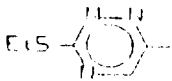
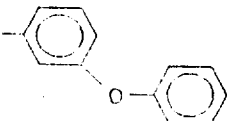
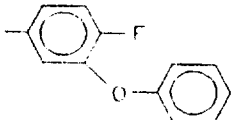
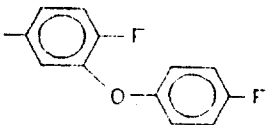
Continuation of Table 3:

No.	R ¹	R ⁴	R ⁵	Physical data
553	$\text{F}_3\text{C}-\text{CH}_2-\text{O}-$ 	H		
554	"	H		
555	"	H		
556	"	H		
557	"	H		
558	"	CN		
559	$\text{Br}-$ 	H		

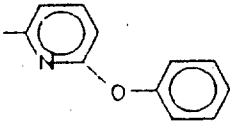
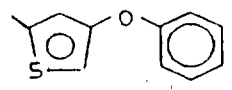
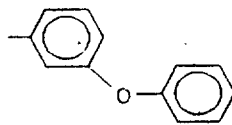
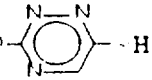
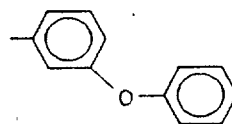
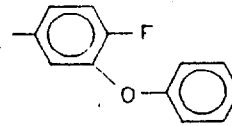
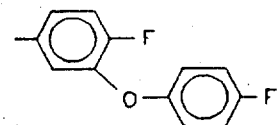
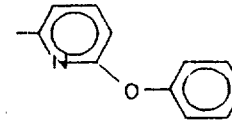
Continuation of Table 3:

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561	"	H		
562	"	H		
563	"	H		
564	"	CN		
565		H		
566	"	H		

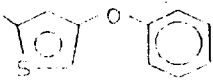
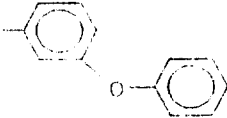
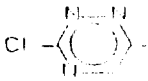
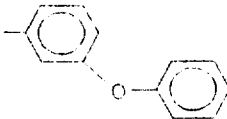
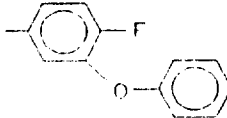
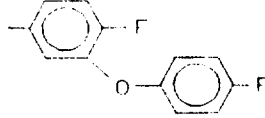
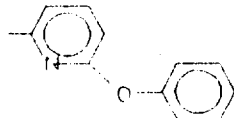
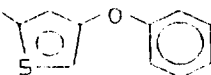
Continuation of Table 3:

No.	R ¹	R ⁴	R ⁵	Physical data
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568	"	H		
569	"	H		
570	"	CN		
571		H		
572	"	H		
573	"	H		

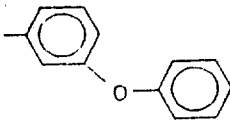
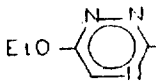
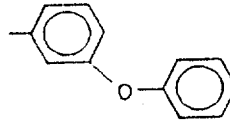
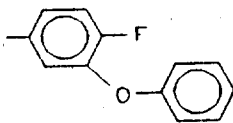
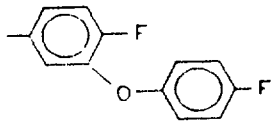
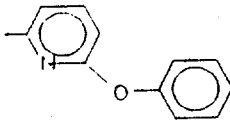
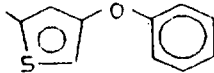
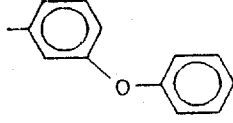
Continuation of Table 3:

No.	R ¹	R ⁴	R ⁵	Physical data
574	"	H		
575	"	H		
576	"	CN		
577	$\text{F}_3\text{C}-\text{CH}_2-\text{O}-$ 	H		
578	"	H		
579	"	H		
580	"	H		

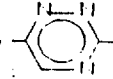
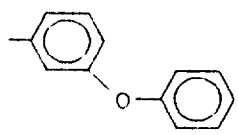
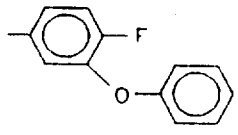
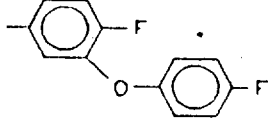
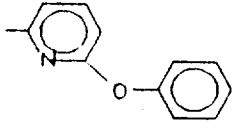
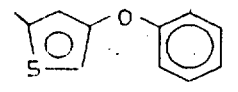
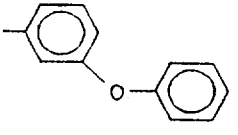
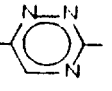
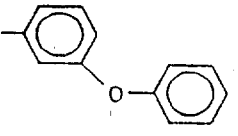
Continuation of Table 3:

No.	R ¹	R ⁴	R ⁵	Physical data
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584	"	H		
585	"	H		
586	"	H		
587	"	H		

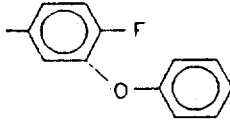
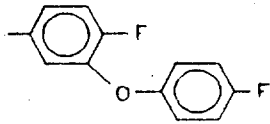
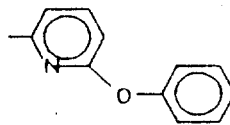
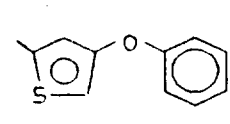
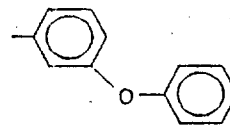
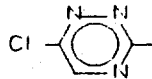
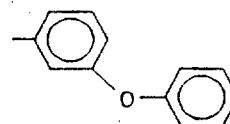
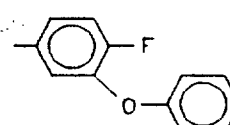
Continuation of Table 3:

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589		H		
590	"	H		
591	"	H		
592	"	H		
593	"	H		
594	"	CN		

Continuation of Table 3:

No.	R ¹	R ⁴	R ⁵	Physical data
595	EtS- 	H		
596	"	H		
597	"	H		
598	"	H		
599	"	H		
600	"	CN		
601	F ₃ C-CH ₂ -O- 	H		

Continuation of Table 3:

No.	R ¹	R ⁴	R ⁵	Physical data
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603	"	H		
604	"	H		
605	"	H		
606	"	CN		
607	Cl 	H		
608	"	H		

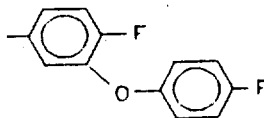
Continuation of Table 3:

No.	R ¹	R ⁴	R ⁵	Physical data
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609

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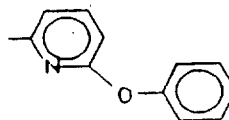
H



610

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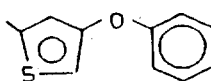
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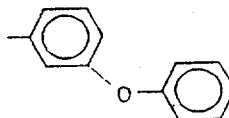
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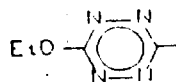
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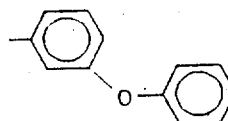
CN



613



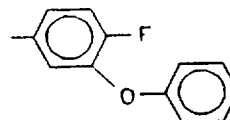
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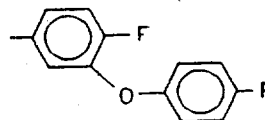
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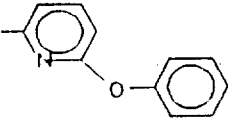
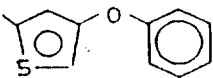
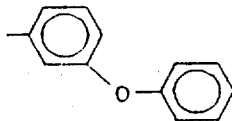
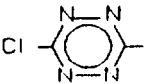
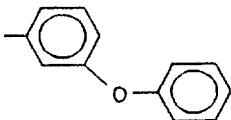
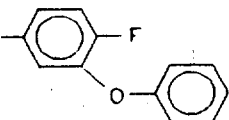
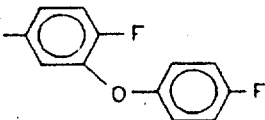
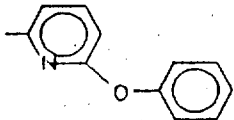
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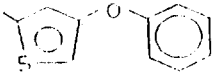
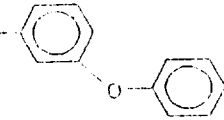
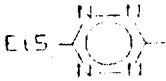
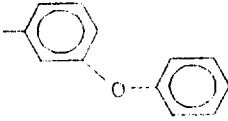
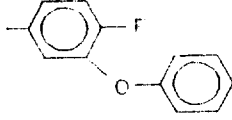
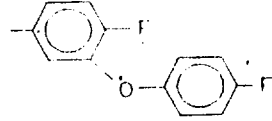
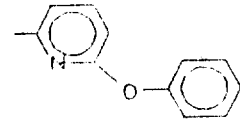
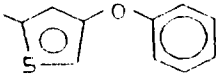
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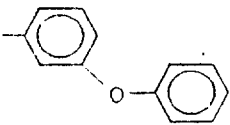
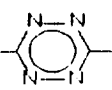
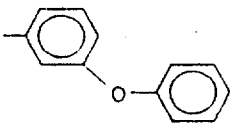
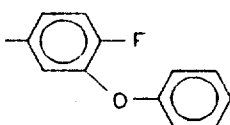
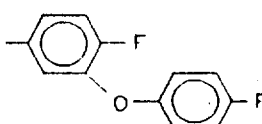
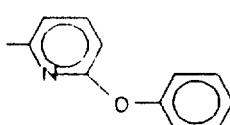
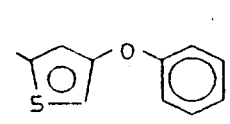
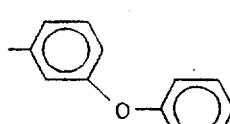
Continuation of Table 3:

No.	R ¹	R ⁴	R ⁵	Physical data
616	"	H		"
617	"	H		"
618	"	CN		"
619		H		"
620	"	H		"
621	"	H		"
622	"	H		"

Continuation of Table 3:

No.	R ¹	R ⁴	R ⁵	Physical data
623	"	H		
624	"	CN		
625		H		
626	"	H		
627	"	H		
628	"	H		
629	"	H		

Continuation of Table 3:

No.	R ¹	R ⁴	R ⁵	Physical data
630	"	CN		
631	$\text{F}_3\text{C}-\text{CH}_2-\text{O}-$ 	H		
632	"	H		
633	"	H		
634	"	H		
635	"	H		
636	"	CN		

Biological Examples

Example 1

Bean plants (*Phaseolus v.*) which were heavily infested with greenhouse red spider mites (5 *Tetranychus urticae*, entire population) were sprayed with the aqueous dilution of an emulsion concentrate which contained 1,000 ppm of the respective active substance.

10 The mortality of the mites was checked after 7 days. 100% destruction was achieved using the compounds of Example 25, 26, 157 and 158.

Example 2

Field beans (*Vicia faba*) showing a large number of cowpea aphid (15 *Aphis craccivora*) were sprayed with aqueous dilutions of wettable powder concentrates of active substance content 100 ppm until just dripping wet.

The mortality of the aphids was determined after 3 days. 100% destruction could be 20 achieved using the compounds of Example 1, 25,

26, 133, 157 and 158.

Example 3

5 The bottom insides of Petri dishes were coated with artificial nutrient medium and then sprayed with 3 ml portions of an emulsion containing 1,000 ppm of active substance after the nutrient broth had solidified. After the spray-coating had dried on, 10 cotton worm larvae (*Prodenia litura*) were placed inside and the dishes were kept for 7 days at 21°C, and the degree of activity of the respective compounds (expressed in % mortality) was then determined. In this test, the compounds 1, 2, 25, 26, 133, 134, 157 and 158 each showed an activity of 100%.

Example 4

20 Bean leaves (*Phaseolus vulgaris*) were sprayed uniformly with an aqueous emulsion of the compounds at an active substance concentration of 250 ppm and placed in observation cages together with Mexican bean beetle larvae

(*Epilachna varivestis*) which had been treated in the same way. After 48 hours, the compounds 1, 2, 25, 26, 133, 134, 157 and 158 showed 100% destruction of the test insects.

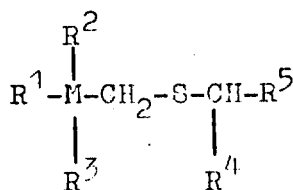
Example 5

- 5 The bottom and lid insides of glass Petri dishes were lined with filter paper disks, and the latter were sprayed with 2 ml portions of an aqueous emulsion of the compounds to be tested, containing 500 ppm of active substance. 20
- 10 winged specimens of the brown plant hopper (*Nilaparvata lugens*) were subsequently inserted into each dish, the lids were put on the bottoms, and the Petri dishes were kept for 48 hours at 22°C.
- 15 After this period, the tested compounds 1, 2, 25, 26, 133, 134, 157 and 158 each showed 100% mortality of the experimental animals.

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Patent Claims

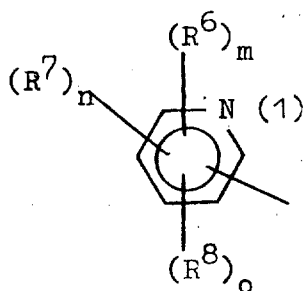
1. A compound of the formula I, its optical isomers and mixtures thereof,



where

M denotes C or Si,

R¹ denotes a radical of the formula A.

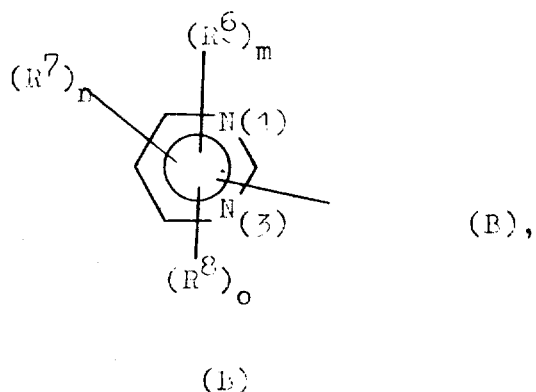


(A)

(A)

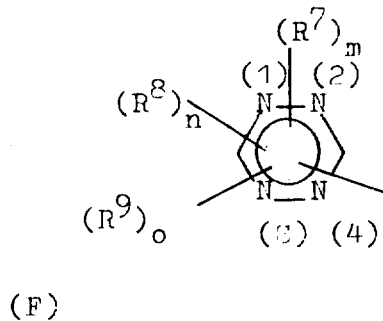
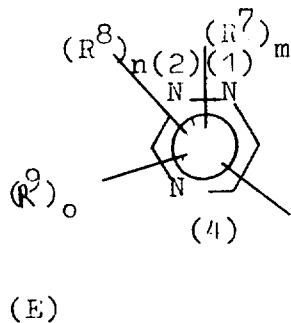
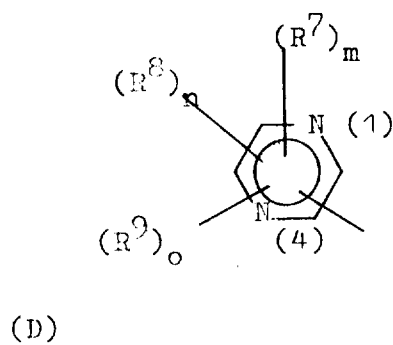
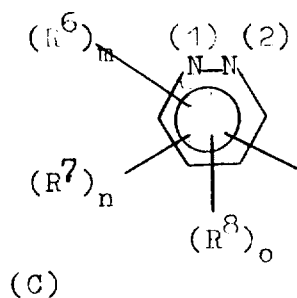
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or E



where $0 \leq m + n + o \leq 3$ and m, n, o may have the values 0 to 2, R^6, R^7, R^8 and R^9 independently of one another stand for (C_1-C_5) alkyl, halogen, (C_1-C_3) alkoxy, (C_1-C_4) alkylthio, (C_1-C_3) haloalkoxy, (C_1-C_3) haloalkylthio or two of the radicals R^6, R^7, R^8, R^9 - when they are in the ortho position - form a methylenedioxy, ethylene dioxy or (C_3-C_5) alkylene radical, or - when H denotes C - also a radical of the formula C, D, E or F.

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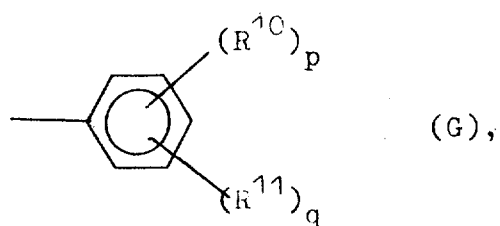
where R^6 , R^7 , R^8 , R^9 , m , n and O have the above-mentioned meaning,

R^2 , R^3 independently of one another denote

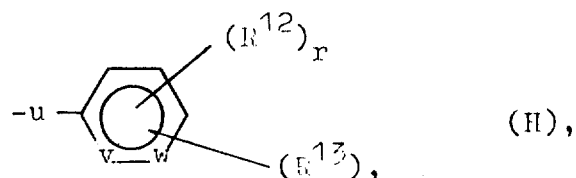
(C_1 - C_3) alkyl,

R^4 denotes H, -CN,

R^5 denotes pyridyl, thienyl or substituted phenyl of the formula (G)



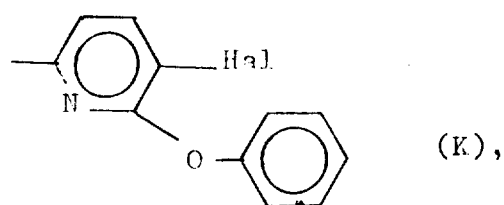
where $p, q = 1$, $R^{10} = H$ or 4-fluorine, and R^{11} in the β -position of the phenyl radical represents a radical of the formula (H)



where R^{12} and R^{13} independently of one another denote H or halogen, U denotes O and V, W denotes CH $r, s = 0, 1$ or 2 with the condition that the total on $r + s$ must 0, 1 or 2, or

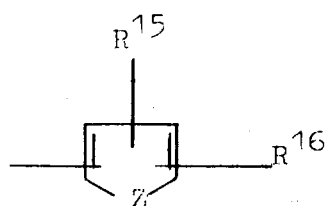
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R^5 denotes a radical of the formula (K)



where Hal denotes fluorine or H,

or R^5 denotes a radical of the formula (L)



where Z denotes H,

R^{15} denotes H and

R^{16} denotes phenoxy,

2. A compound of the formula 1 as claimed in claim 1, where R^1 denotes a radical of the general formulae (A), (B), (C), (D), (E), or (F).

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3. A compound of the formula I as claimed in claim 1 where R¹ denotes a radical of the formula A.
4. A pesticide, containing a compound of the formula I or claim 1.
5. An insecticidal agent, containing a compound of the formula I of claim 1.
6. An acaricidal agent, containing a compound of the formula I of claim 1,
7. A nematocidal agent, containing a compound of the formula I of claim 1.
8. A method of controlling noxious insects, acarids or nematodes, which comprises applying an active amount of a compound of the formula I of claim 1 to these animals, on the areas, plants or substrates infested with them.