#### (12) INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

### (19) World Intellectual Property Organization

International Bureau





(10) International Publication Number WO 2012/146736 A1

# (43) International Publication Date 1 November 2012 (01.11.2012)

(21) International Application Number:

PCT/EP2012/057806

English

(22) International Filing Date:

27 April 2012 (27.04.2012)

(26) Publication Language: English

(30) Priority Data:

(25) Filing Language:

 11163839.1
 27 April 2011 (27.04.2011)
 EP

 11163841.7
 27 April 2011 (27.04.2011)
 EP

 11163840.9
 27 April 2011 (27.04.2011)
 EP

 11163872.2
 27 April 2011 (27.04.2011)
 EP

 11163871.4
 27 April 2011 (27.04.2011)
 EP

(71) Applicant (for all designated States except US): GRINDEKS, A JOINT STOCK COMPANY [LV/LV]; 53, Krustpils street, LV-1057 Riga (LV).

(72) Inventors; and

(75) Inventors/Applicants (for US only): STONANS, Ilmars [LV/LV]; 53, Krustpils street, LV-1057 Riga (LV). KALVINS, Ivars [LV/LV]; Aizkraukles street 21, LV-1006 Riga (LV). DOMBROVA, Maija [LV/LV]; 21, Aizkraukles street, LV-1006 Riga (LV). LIEPINS, Edgars [LV/LV]; 21 Aizkraukles street, LV-1006 Riga (LV).

- (81) Designated States (unless otherwise indicated, for every kind of national protection available): AE, AG, AL, AM, AO, AT, AU, AZ, BA, BB, BG, BH, BR, BW, BY, BZ, CA, CH, CL, CN, CO, CR, CU, CZ, DE, DK, DM, DO, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, GT, HN, HR, HU, ID, IL, IN, IS, JP, KE, KG, KM, KN, KP, KR, KZ, LA, LC, LK, LR, LS, LT, LU, LY, MA, MD, ME, MG, MK, MN, MW, MX, MY, MZ, NA, NG, NI, NO, NZ, OM, PE, PG, PH, PL, PT, QA, RO, RS, RU, RW, SC, SD, SE, SG, SK, SL, SM, ST, SV, SY, TH, TJ, TM, TN, TT, TZ, UA, UG, US, UZ, VC, VN, ZA, ZM, ZW.
- (84) Designated States (unless otherwise indicated, for every kind of regional protection available): ARIPO (BW, GH, GM, KE, LR, LS, MW, MZ, NA, RW, SD, SL, SZ, TZ, UG, ZM, ZW), Eurasian (AM, AZ, BY, KG, KZ, RU, TJ, TM), European (AL, AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HR, HU, IE, IS, IT, LT, LU, LV, MC, MK, MT, NL, NO, PL, PT, RO, RS, SE, SI, SK, SM, TR), OAPI (BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG).

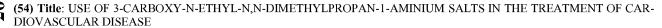
#### **Declarations under Rule 4.17:**

— of inventorship (Rule 4.17(iv))

#### Published:

- with international search report (Art. 21(3))
- before the expiration of the time limit for amending the claims and to be republished in the event of receipt of amendments (Rule 48.2(h))





WO 2012/146736 PCT/EP2012/057806

### Description

5

15

20

Use of 3-carboxy-*N*-ethyl-*N*,*N*-dimethylpropan-1-aminium salts in the treatment of cardiovascular disease

Technical Field

The present invention relates to new compound 3-carboxy-*N*-ethyl-*N*,*N*-dimethylpropan-1-aminium salts, and to a method of preparation thereof (compound of formula 4)

10
$$R_{1} = \begin{pmatrix} AcO \\ OOC \end{pmatrix}, \begin{pmatrix} COOH \\ OOC \end{pmatrix}$$

The present invention relates also to use of 3-carboxy-*N*-ethyl-*N*,*N*-dimethylpropan-1-aminium salts in the treatment of cardiovascular disease.

Background Art

Cardiovascular diseases (CVDs) are a group of disorders of the heart and blood vesels.

An estimated 16.7 million - or 29.2% of total global deaths - result from the various forms of cardiovascular disease (CVD).

Myocardial infarction (heart attack) is a serious result of coronary artery disease. Myocardial infarction (MI) is the irreversible necrosis of heart muscle secondary to prolonged ischemia. A heart attack or myocardial infarction is a medical emergency in which the supply of blood to the heart is suddenly and severely reduced or cut off, causing the muscle to die from lack of oxygen. More than 1.1 million people experience a heart attack (myocardial infarction) each year, and for many of them, the heart attack is their first symptom of coronary artery disease. A heart attack may be severe enough to cause death or it may be silent. As many as one out of every five people have only mild symptoms or none at all, and the heart attack may only be discovered by routine electrocardiography done some time later.

A heart attack (myocardial infarction) is usually caused by a blood clot that blocks an artery of the heart. The artery has often already been narrowed by fatty deposits on its walls. These deposits can tear or break open, reducing the flow of blood and WO 2012/146736 2 PCT/EP2012/057806

releasing substances that make the platelets of the blood sticky and more likely to form clots. Sometimes a clot forms inside the heart itself, then breaks away and gets stuck in an artery that feeds the heart. A spasm in one of these arteries causes the blood flow to stop.

- γ-Butyrobetaine, from which the mammalian organism synthesises carnitine, was primarily characterised as a toxic substance which accelerates respiration, causes salivation and lacrimation, pupil dilation, vasoconstriction and heart stop in diastole LINNEWEH, W. Gamma-Butyrobetain, Crotonbetain und Carnitin im tierischen Stoffwechsel. *Hoppe-Seylers Zeitschrift für physiologische Chemie.* 1929, vol.181, p.42-53. At the same time, in later papers other authors ascertained that γ-butyrobetaine is extremely low toxic (LD50>7000 mg/kg, s.c.) ROTZSCH, W. Iber die Toxizitat des Carnitins und einiger verwandter Stoffe. *Acta biol. med. germ..* 1959, vol.3, p.28-36.
- In the literature data on nonsubstututed  $\gamma$ -butyrobetaine cardiovascular effects are missed, thought it was reported **HOSEIN**, **E.A**. Pharmacological actions of  $\gamma$ -butyrobetaine. *Nature*. 1959, vol.183, p.328-329. that  $\gamma$ -butyrobetaine is a substance similar to acetyl choline with a prolonged action. However, later the same authors reported that by an error the experiments involved, instead of  $\gamma$ -butyrobetaine, its methyl esther which in fact possesses cholinergic properties.
- 20 Contrary to the former *γ*-butyrobetaine was characterised as a pharmacologically inert substance **HOSEIN**, **E.A.** Isolation and probable functions of betaine esters in brain metabolism. *Nature*. 1960, vol.187, p.321-322.

As structurally related compounds to 3-carboxy-*N*-ethyl-*N*,*N*-dimethylpropan-1-aminium salts are disclosed in:

25

30

35

- GB 1238868 A 14.07.1971 were disclosed betaines, such as 4trimethylammoniobutanoate, used for polymers. However no pharmacological propeties of these betaines weren't presented;
- US 5973026 A (XEROX CORP) 26.10.1999 were disclosed 4-trimethylammoniobutanoate and 3-[diethyl(methyl)ammonio]propionate for using for ink compositions;
- LLOYD ANDREW, et al. A comparison of glycine, sarcosine, N,N-dimethylglycine, glycinebetaine and N-modified betaines as liposome cryoprotectants. *Journal of pharmacy and pharmacology.* 1992, vol.44, no.6, p.507-511 disclosed 2-[ethyl(dimethyl)ammonio]acetate used as cryoprotectants for liposomes;

WO 2012/146736 PCT/EP2012/057806

• DAVID B., THOMAS, et al. Synthesis, Characterization, and Aqueous Solution Behavior of Electrolyte- and pH-Responsive Carboxybetaine-Containing Cyclocopolymers. *Macromolecules*. 2003, vol.36, no.26, p.9710-9715 disclose 4-[diallyl(methyl)ammonio]butanoate and its synthesis starting from *N,N*-diallyl-*N*-methylaminiom and ethyl 4-bromobutanoate. The free acis is obtained from the ester in a second step using Amberlite ion exchange resin. The product is used as intermediate to synthesise polymers;

5

10

15

20

25

- Prelog V. 1930, vol.2, p.712-722 disclosed the synthesis of 4trimethylammoniobutanoate starting from 4-dimethylammoniobutanoate and methyliodide;
- 4-Trimethylammoniobutanoate and its synthesis starting from trimethylamine and ethyl 4-bromobutanoate was described JP 2009096766 A (KONAN GAKUEN) 07.05.2009. The free acid is obtained from the ester in a second step using Amberlite ion exchange resin;
- WO 2008/055843 A (KALVINSH IVARS; CHERNOBROVIJS
   ALEKSANDRS; VARACHEVA LARISA; PUGOVICHS OSVALDS)
   15.05.2008 was described 4-trimethylammoniobutanoate and synthesis,
   which started from the correspondin ester and using KOH-solution;
- CA 2508094 A (VIVIER CANADA INC) 20.11.2006 was disclosed betaines, such as 4-trimethylammoniobutanoate, for use as medicament for accelerating collagen synthesis;
- US 5965615 A (TAIHO PHARMACEUTICAL CO LTD; VALSTS
   ZINATNISKA IESTADE BEZP) 12.10.1999 was disclosed 4 trimethylammoniobutanoate as a medicament for the treatment of
   myocardial metabolic disorder, the same compound was disclosed in
   US 2007191381 A (CONCERT PHARMACEUTICALS INC) 16.08.2007
   for treatment of myocardial infarction.
- 30 3- (2,2,2-Trimethylhydrazinium) propionate dihydrate is known as compound with cardioprotective properties (this substance being known under its International Nonproprietary Name of Meldonium). 3- (2,2,2-Trimethylhydrazinium) propionate is disclosed in US 4481218 (INST ORGANICHESKOGO SINTEZA) 06.11.1984 as well in US 4451485 A (INSTITU ORCH SINTEZA AKADEMII) 29.05.1984.

WO 2012/146736 4 PCT/EP2012/057806

It is well known that 3- (2, 2,2-trimethylhydrazinium) propionate as dihydrate is widely used for controlling carnitine and gamma-butyrobetaine concentration ratio and consequently the speed of fatty acid beta-oxidation in the body **DAMBROVA**M., LIEPINSH E., KALVINSH I. I. Mildronate: cardioprotective action through carnitine-lowering effect. *Trends in Cardiovascular Medicine*, 2002, vol.12, no.6, p.275-279.

Due to these properties, Meldonium is extensively applied in medicine as an antiischemic, stress-protective and cardioprotective drug in treating various cardiovascular diseases and other pathologies involving tissue ischemia KARPOV R.S.,

- KOSHELSKAYA O.A., VRUBLEVSKY A.V., SOKOLOV A.A., TEPLYAKOV A.T., SKARDA I., DZERVE V., KLINTSARE D., VITOLS A., KALNINS U., KALVINSH I., MATVEYA L., URBANE D. Clinical Efficacy and Safety of Mildronate in Patients With Ischemic Heart Disease and Chronic Heart Failure. *Kardiologiya*. 2000, no.6, p.69-74. In the treatment of cardiovascular diseases the mechanism of action of 3-
- 15 (2,2,2-trimethylhydrazinium)propionate based on limitation of carnitine biosynthesis rate and related long-chain fatty acid transport limitation through mitochondria membranes SIMKHOVICH B.Z., SHUTENKO Z.V., MEIRENA D.V., KHAGI K.B., MEZHAPUKE R.J., MOLODCHINA T.N., KALVINS I.J., LUKEVICS E. 3-(2,2,2,-Trimethylhydrazinium)propionate (THP) a novel gamma-butyrobetaine
- 20 hydroxylase inhibitor with cardioprotective properties. *Biochemical Pharmacology*. 1988, vol.37, p.195-202. , **KIRIMOTO T., ASAKA N., NAKANO M., TAJIMA K., MIYAKE H., MATSUURA N**. Beneficial effects of MET-88, a γ-butyrobetaine hydroxylase inhibitor in rats with heart failure following myocardial infarction. *European Journal of Pharmacology*. 2000, vol.395, no.3, p.217-224.

# 25 Summary of invention

As it was known what Meldonium dihydrate has cardioprotective effect; however there are no data that  $\gamma$ -butyrobetaine itself has pronounced cardioprotective effect. In the patent EP 0845986 B (KALVINSH IVARS, VEVERIS MARIS) 02.04.2003 is disclosed pharmaceutical composition of Meldonium dihydrate and  $\gamma$ -butyrobetaine for use in the treatment of cardiovascular diseases.

- An object of the present invention is to provide a compound, which has pronounced cardioprotective effect.
- The above-mentioned object is attained by providing new compounds 3-carboxy-*N*-ethyl-*N*,*N*-dimethylpropan-1-aminium salts (compound of formula 4), which has
- 35 similar structure to Meldonium or  $\gamma$ -butyrobetaine.

$$-N$$
 $+R_1$ 
COOH

$$\mathsf{R}_1 = \underbrace{\begin{smallmatrix} \mathsf{AcO} \\ \mathsf{OOC} \end{smallmatrix}}^{\mathsf{AcO}}, \underbrace{\begin{smallmatrix} \mathsf{COOH} \\ \mathsf{HN} \end{smallmatrix}}^{\mathsf{NH}} \underbrace{\begin{smallmatrix} \mathsf{COOH} \\ \mathsf{NH} \end{smallmatrix}}_{\mathsf{OOC}}, \underbrace{\begin{smallmatrix} \mathsf{COOH} \\ \mathsf{H}_2\mathsf{PO}_4 \end{smallmatrix}}^{\mathsf{H}_2\mathsf{PO}_4}$$

To our surprise 3-carboxy-*N*-ethyl-*N*,*N*-dimethylpropan-1-aminium salts posses pronounced cardioprotective effect and are more effective as Meldonium dihydrate *in vivo* myocardial infarction models, due to this property 3-carboxy-*N*-ethyl-*N*,*N*-dimethylpropan-1-aminium salts may be used in medicine. 3-Carboxy-*N*-ethyl-*N*,*N*-dimethylpropan-1-aminium salts can be use as a solution for injection.

10 The following object of the present invention is a method of preparation of said compound of formula **4**.

There is disclosed process, which can be used in purpose to prepare target compound 3-carboxy-*N*-ethyl-*N*,*N*-dimethylpropan-1-aminium salts of formula 4, see scheme bellow.

Br COOEt 
$$\frac{\text{EtMe}_2\text{N}}{2}$$
 COOEt  $\frac{\text{-N}_+^+\text{Br}}{3}$  COO

15

20

25

Process for preparing 3-carboxy-*N*-ethyl-*N*,*N*-dimethylpropan-1-aminium salt of formula 4 involves the following process steps:

- a) adding *N*,*N*-dimethylethylamine to ethyl 4-bromobutanoate (1) in appropriate solvent to obtain 4-ethoxy-*N*-ethyl-*N*,*N*-dimethyl-4-oxo-1-butanaminium bromide (2);
  - b) passing 4-ethoxy-*N*-ethyl-*N*,*N*-dimethyl-4-oxo-1-butanaminium bromide (2) through ion exchange resin column to obtain 4- [ethyl(dimethyl)ammonio] butanoate (3);
  - c) adding acid which is selected from 2-(acetyloxy)benzoic acid (4 a) or (*E*)-butenedioic acid (4 b) or succinic acid (4 c) or 2,6-dioxo-1,2,3,6-

PCT/EP2012/057806 WO 2012/146736

> tetrahydropyrimidine-4-carboxylic acid monohydrate (4 d) or phosphoric acid (4 e) to 4-[ethyl(dimethyl)ammonio] butanoate (3) in appropriate solvent to obtain 3-carboxy-N-ethyl-N,N-dimethylpropan-1-aminium salt (4).

5

15

20

25

30

35

# Description of embodiments

The present invention will be described in more detail by referring to the following non-limiting examples.

10 Preparation of 4-ethoxy-N-ethyl-N,N-dimethyl-4-oxo-1-butanaminium bromide (2) Procedure A

To a solution of ethyl 4-bromobutanoate (1) (20.0 g, 102.5 mmol) in acetonitrile (70 ml) N, N-dimethylethylamine (15 ml, 139 mmol) was added and stirred at ambient temperature for 3 days. The reaction mixture was evaporated, the residue was triturated with acetone (50 ml), filtered, washed with ether, and dried to afford 26.051 g (94.8%) of the 4-ethoxy-N-ethyl-N,N-dimethyl-4-oxo-1-butanaminium bromide. LCMS (ESI+, *m/z*): [M-Br-]+ 188, purity 98.9%.

[0001] <sup>1</sup>H NMR (CDCl<sub>3</sub>, HMDSO)  $\delta$ : 1.26 (t,  $\mathcal{L}$ =7.2 Hz, 3H); 1.44 (t,  $\mathcal{L}$ =7.4 Hz, 3H); 2.00-2.11 (m, 2H); 2.52 (t,  $\sqrt{=}6.6$  Hz, 2H); 3.40 (s, 6H); 3.64-3.73 (m, 2H); 3.69 (q,  $\mathcal{J}=7.4$  Hz, 2H); 4.14 (q,  $\mathcal{J}=7.2$  Hz, 2H).

### Procedure B

To a solution of ethyl 4-bromobutanoate (1) (19.5 g, 100 mmol) in acetone (70 ml) N, N-dimethylethylamine (15 ml, 139 mmol) was added and stirred at ambient temperature for 3 days. The reaction mixture was filtered; the solid material was washed with an acetone, ether, and dried to afford 24.19 g (90.2%) of the title compound 2. The filtrate was evaporated; the residue (2.147 g) was triturated with ether and dried to give an extra batch (0.962 g, 3.6%) of the product 2 of the same quality as the main portion. The evaporation of the ether washings allowed recovering 0.956 g (4.9 mmol, 4.9%) of the starting material 1. 4-ethoxy-N-ethyl-*N,N*-dimethyl-4-oxo-1-butanaminium bromide: LCMS (ESI+, *m/z*): [M-Br-]+ 188, purity 98.4%.

<sup>1</sup>H NMR (CDCl<sub>3</sub>, HMDSO)  $\delta$ : 1.26 (t, J=7.2 Hz, 3H); 1.44 (t, J=7.4 Hz, 3H); 2.00-2.11 (m, 2H); 2.52 (t,  $\mathcal{L}$ =6.6 Hz, 2H); 3.40 (s, 6H); 3.64-3.73 (m, 2H); 3.69 (g,  $\mathcal{L}$ =7.4 Hz, 2H); 4.14 (q,  $\mathcal{L}$ =7.2 Hz, 2H).

Preparation of 4-[ethyl(dimethyl)ammonio]butanoate (3)

A solution of 4-ethoxy-*N*-ethyl-*N*,*N*-dimethyl-4-oxo-1-butanaminium bromide (2) (12.00 g, 44.7 mmol) in water (10 ml) was passed through Amberlite ® IRA-410 (OH) ion exchange resin column (250 ml) eluting slowly ( $\it ca.$  10 drops/min) with ethanol (TLC control). The eluate was evaporated and the residue (12 g) was dissolved in water (50 ml). To this solution DOWEX® 50WX8 ion exchange resin (5 g) was added and stirred at ambient temperature for 0.5 h. The reaction mixture was filtered through celite (1 cm) and the eluate was evaporated. The residue was azeotropically dried with isopropanol, acetonitrile, and acetone. The obtained solid was triturated with acetone (10 ml) and the mixture was kept at 0°C for 2 h. The precipitate was filtered and dried *in vacuo* over  $P_2O_5$  to give 4.65 g (65%) of the 4-[ethyl(dimethyl)ammonio]butanoate (3).

(DMSO-d<sub>6</sub>, HMDSO) δ: 1.24 (t, *J*=7.3 Hz, 3H); 1.66-1.76 (m, 2H); 1.81 (t, *J*=6.4 Hz, 2H); 2.95 (s, 6H); 3.16-3.23 (m, 2H); 3.29 (q, *J*=7.3 Hz, 2H). LCMS (ESI+, *m/z*): 160 [M+H]+.

Anal. Calc. for C<sub>8</sub>H<sub>17</sub>NO<sub>2</sub> · 1.55 H<sub>2</sub>O: C 51.34; H 10.82; N 7.48.

Found: C 51.36, H 11.40, N 7.34.

Preparation of 3-carboxy-N-ethyl-N,N-dimethylpropan-1-aminium 2-

20 (acetyloxy)benzoate (4 a)

5

10

15

25

30

3-Carboxy-*N*-ethyl-*N*,*N*-dimethylpropan-1-aminium 2-(acetyloxy)benzoate was prepared in a form of a water mixture. Thus, *ca.* 90% 4-[ethyl-(dimethyl)ammonio]butanoate (3) (2.20 g, 12.44 mmol) and 2-(acetyloxy)-benzoic acid (2.266 g, 12.57 mmol) were placed in a volumetric flask and diluted with water up to 100 ml. The content of the mixture dissolves by heating and precipitates by lowering of the temperature. According to <sup>1</sup>H-NMR, the precipitated solid material consists of almost pure 2-(acetyloxy)-benzoic acid.

Preparation of 3-carboxy-*N*-ethyl-*N*,*N*-dimethylpropan-1-aminium (2*E*)-3-carboxyacrylate (4 b)

To a solution of 4-[ethyl(dimethyl)ammonio]butanoate (3) (2.0 g, 12.56 mmol) in anh. ethanol (10 ml) a hot (60°C) solution of (*E*)-butenedioic acid (1.46 g, 12.56 mmol) in ethanol (50 ml) was added. The reaction mixture was allowed to stand at ambient temperature for 2 h, the precipitated crystals were filtered and dried over

 $P_2O_5$  to give 2.98 g (85%) of the 3-carboxy-*N*-ethyl-*N*,*N*-dimethylpropan-1-aminium (2*E*)-3-carboxyacrylate. M.p. 122-123°C.

- <sup>1</sup>H-NMR (D<sub>2</sub>O, DSS) δ: 1.36 (tt,  $\mathcal{L}$ =1.9, 7.3 Hz, 3H); 2.06 (m, 2H); 2.49 (t,  $\mathcal{L}$ =7.1 Hz, 2H); 3.06 (s, 6H); 3.31 (m, 2H); 3.40 (q,  $\mathcal{L}$ =7.3 Hz, 2H); 6.75 (s, 1.9H, CH=CH).
- 5 LCMS ESI+ (*m/z*): 160 [M+H]+. Titration assays: water content (Fisher) 0.13%, betaine content (HClO<sub>4</sub>) 93.0%, (*E*)-butenedioic acid content 46.1%.
  Anal. Calc. for C<sub>8</sub>H<sub>17</sub>NO<sub>2</sub> · 1.2 C<sub>4</sub>H<sub>4</sub>O<sub>4</sub> (46.7%): C 51.50, H 7.36, N 4.69.
  Found: C 51.52, H 7.35, N 4.61.
- Preparation of 3-carboxy-*N*-ethyl-*N*,*N*-dimethylpropan-1-aminium 3-carboxypropanoate (4 c)
  3-Carboxy-*N*-ethyl-*N*,*N*-dimethylpropan-1-aminium 3-carboxypropanoate was prepared in a form of a water solution. Thus, *ca.* 90% 4-[ethyl-(dimethyl)ammonio]butanoate (3) (2.20 g, 12.44 mmol) and succinic acid (1.49 g, 12.62 mmol) were placed in a volumetric flask and dissolved and diluted with water up to 100 ml.
  - Preparation of 3-carboxy-*N*-ethyl-*N*,*N*-dimethylpropan-1-aminium 2,6-dioxo-1,2,3,6-tetrahydropyrimidine-4-carboxylate (4 d)
- To a solution of 4-[ethyl(dimethyl)ammonio]butanoate (3) (2.0 g, 12.56 mmol) in isopropanol (100 ml) 2,6-dioxo-1,2,3,6-tetrahydropyrimidine-4-carboxylic acid monohydrate (2.187 g, 12.56 mmol) was added and the reaction mixture was heated to reflux until all the carboxylic acid dissolved. The reaction mixture was allowed to cool to ambient temperature, the precipitated crystals were filtered,
- washed with isopropanol (5 ml) and diethyl ether (20 ml), and dried over  $P_2O_5$  to give 3.238 g (97.4%) of the 3-carboxy-*N*-ethyl-*N*,*N*-dimethylpropan-1-aminium 2,6-dioxo-1,2,3,6-tetrahydropyrimidine-4-carboxylate. M.p. 150.7°C.
  - <sup>1</sup>H-NMR (D<sub>2</sub>O, DSS) δ: 1.36 (tt, J=2.0, 7.3 Hz, 3H); 2.05 (m, 2H); 2.47 (t, J=7.0 Hz, 2H); 3.07 (s, 6H); 3.31 (m, 2H); 3.41 (q, J=7.3 Hz, 2H); 6.20 (s, 1H, C=CH).
- 30 LCMS ESI+ (*m/z*): 160 [M+H]+.

Anal. Calc. for  $C_8H_{17}NO_2 \cdot C_5H_4N_2O_4$  (49.5%): C 49.52, H 6.71, N 13.33. Found: C 49.59, H 6.69, N 13.26.

Preparation of 3-carboxy-*N*-ethyl-*N*,*N*-dimethylpropan-1-aminium dihydrogen phosphate (4 e)

To a solution of 4-[ethyl(dimethyl)ammonio]butanoate (3) (6.4 g, 40 mmol) in water (10 ml) a solution of 85% aq.  $H_3PO_4$  (4.73 g, 40 mmol) in acetone (10 ml) was added and the resulting solution was stirred at ambient temperature for 10 min. The reaction mixture was evaporated and azeotropically dried several times with acetone by rotary evaporator at 45°C. The obtained white crystalline substance was dried over  $P_2O_5$  to give 9.82 g (95%) of the 3-carboxy-N-ethyl-N, N-dimethylpropan-

<sup>1</sup>H-NMR (D<sub>2</sub>O, DSS) δ: 1.36 (tt, J=1.8, 7.3 Hz, 3H); 2.06 (m, 2H); 2.50 (t, J=7.0 Hz, 2H); 3.06 (s, 6H); 3.32 (m, 2H); 3.41 (q, J=7.3 Hz, 2H). LCMS ESI+ (m/z): 160

10 [M+H]\*. Titration assays: water content (Fisher) 0.356%, betaine content (HClO<sub>4</sub>) – 95.682%.

Anal. Calc. for  $C_8H_{17}NO_2 \cdot 0.052 H_2O (0.356\%) \cdot 1.07 H_3PO_4 (39.6\%)$ : C 36.26; H 7.73; N 5.29.

Found: C 36.20, H 7.72, N 5.11.

- The purity of the obtained 3-carboxy-*N*-ethyl-*N*, *N*-dimethylpropan-1-aminium dihydrogen phosphate was increased by crystallization from methanol. Thus, the 3-carboxy-*N*-ethyl-*N*, *N*-dimethylpropan-1-aminium dihydrogen phosphate (6.9 g) was crystallized from methanol (40 ml) to afford 5.326 g (77%) of the purified 3-carboxy-*N*-ethyl-*N*, *N*-dimethylpropan-1-aminium dihydrogen phosphate with m.p. 139°C.
- 20 Calc. for C<sub>8</sub>H<sub>17</sub>NO<sub>2</sub> · H<sub>3</sub>PO<sub>4</sub> (38.1%): C 37.36; H 7.84; N 5.45. Found: C 37.52, H 7.85, N 5.39

1-aminium dihydrogen phosphate. M.p. 110-135°C.

# Cardioprotective activity

25 Fifty male, 10 weeks old Wistar rats weighing 200-250 g were housed under standard conditions (21-23°C, 12 h light-dark cycle) with unlimited access to food (R3 diet, Lactamin AB, Sweden) and water.

Rats were adapted to local conditions for two weeks before the start of treatment.

Meldonium dihydrate at a dose of 20 mg/kg, gamma-butyrobetaine at a dose of 20 mg/kg and 3-carboxy-*N*-ethyl-*N*,*N*-dimethylpropan-1-aminium salts at dose of 20mg/kg were administered *p.o.* daily for 8 weeks. Control rats received water.

Isolated rat heart infarction study

The isolated rat heart experiment was performed essentially as described earlier

(Liepinsh et al., *J. Cardiovasc. Pharmacol.* **2006**; 48(6):314-9). Twenty-four hours

after the last drug administration hearts were excised and retrogradely perfused via

WO 2012/146736 10 PCT/EP2012/057806

the aorta at a constant pressure with oxygenated Krebs-Henseleit buffer at 37°C. The heart rate, left ventricle end-diastolic pressure and left ventricle developed pressure were continuously recorded. Coronary flow was measured using an ultrasound flow detector (HSE) and the PowerLab 8 /30 system from ADInstruments. The hearts were perfused for 20 min to stabilize the hemodynamic functions and then occlusion was performed for 60 min by constricting threads through a plastic tube. Successful occlusion was confirmed by a coronary flow decrease of about 40 percent. Reperfusion was achieved by releasing the threads. At the end of the 150-min reperfusion period, the risk zone was delineated with 10 0.1% methylene blue. The hearts were then sectioned transversely from the apex to the base in five slices 2 mm in thickness and incubated in 1% triphenyltetrazolium chloride in phosphate buffer (pH 7.4, 37°C) for 10 min to stain viable tissue red and necrotic tissue white. Computerized planemetric analysis of Sony A900 photographs was performed using Image-Pro Plus 6.3 software to determine the area at risk and area of necrosis expressed as a % of the left ventricle. The 15

Infarct Size = Area of Necrosis/Area at Risk x 100%.

obtained values were then used to calculate the infarct size (IS) as a % of risk area

20

Effects in isolated rat heart infarction model

according to the formula:

The anti-infarction effect of examined substances was investigated in an isolated rat heart infarction model. During occlusion of left coronary artery, the coronary flow in all experimental groups was decreased for 40% (from 11 ml/min to 7 ml/min).

- 25 Moreover, the drop of developed left ventricular pressure for 50% was observed. The heart rate during the occlusion period did not change significantly. In reperfusion stage, coronary flow, developed left ventricular pressure, ±dp/dt values were recovered till about 80% of control level. There were no significant differences between control and treatment groups.
- 30 Effects of Meldonium dihydrate (20 mg/kg), gamma-butyrobetaine (20 mg/kg) and 3-carboxy-*N*-ethyl-*N*,*N*-dimethylpropan-1-aminium salts (20 mg/kg) after 2 weeks of treatment on infarct size in the isolated rat heart infarction experiment are presented in *Table 1*, *Table 2*, *Table 3*, *Table 4*, *Table 5*, *Table 6*

Table 1

Effects of Meldonium dihydrate, gamma-butyrobetaine and 3-carboxy-*N*-ethyl-*N*,*N*-dimethylpropan-1-aminium 2-(acetyloxy)benzoate on infarct size

5

	Infarct size, % of control
Control	100.0 ±5.9
Meldonium dihydrate 20 mg/kg	117.9 ±7.9
Gamma-butyrobetaine 20 mg/kg	87.6 ±11.4
3-Carboxy-N-ethyl-N,N-	
dimethylpropan-1-aminium 2-	61.6 ±6.7*,*
(acetyloxy)benzoate 20 mg/kg	

Table 2

Effects of Meldonium dihydrate, gamma-butyrobetaine and 3-carboxy-*N*-ethyl-*N,N*-dimethylpropan-1-aminium (2*E*)-3-carboxyacrylate on infarct size

	Infarct size, % of control
Control	100.0 ±5.9
Meldonium dihydrate 20 mg/kg	117.9 ±7.9
Gamma-butyrobetaine 20 mg/kg	87.6 ±11.4
3-carboxy-N-ethyl-N,N-	
dimethylpropan-1-aminium (2 <i>E</i> )-3-	46.5 ±7.0*,#,\$
carboxyacrylate 20 mg/kg	

Table 3

10 Effects of Meldonium dihydrate, gamma-butyrobetaine and 3-carboxy-*N*-ethyl-*N*,*N*-dimethylpropan-1-aminium 2,6-dioxo-1,2,3,6-tetrahydropyrimidine-4-carboxylate on infarct size

	Infarct size, % of control	
Control	100.0 ±5.9	
Meldonium dihydrate 20 mg/kg	117.9 ±7.9	
Gamma-butyrobetaine 20 mg/kg	87.6 ±11.4	
3-carboxy-N-ethyl-N,N-		
dimethylpropan-1-aminium 2,6-	60.6 ±6.7* <sup>,#,\$</sup>	
dioxo-1,2,3,6-tetrahydropyrimidine-	00.0 ±0.7	
4-carboxylate 20 mg/kg		

WO 2012/146736 12 PCT/EP2012/057806

Table 4

Effects of Meldonium dihydrate, gamma-butyrobetaine and 3-carboxy-*N*-ethyl-*N*,*N*-dimethylpropan-1-aminium dihydrogen phosphate on infarct size

	Infarct size, % of control
Control	100.0 ±5.9
Meldonium dihydrate 20 mg/kg	117.9 ±7.9
Gamma-butyrobetaine 20 mg/kg	87.6 ±11.4
3-carboxy-N-ethyl-N,N-	
dimethylpropan-1-aminium	56.1 ±4.4* <sup>,#,\$</sup>
dihydrogen phosphate 20 mg/kg	

5 Table 5

Effects of Meldonium dihydrate, gamma-butyrobetaine and 3-carboxy-*N*-ethyl-*N*,*N*-dimethylpropan-1-aminium 3-carboxypropanoate on infarct size

	Infarct size, % of control
Control	100.0 ±5.9
Meldonium dihydrate 20 mg/kg	117.9 ±7.9
Gamma-butyrobetaine 20 mg/kg	87.6 ±11.4
3-carboxy-N-ethyl-N,N-	
dimethylpropan-1-aminium 3-	62.9 ±4.7* <sup>,#,\$</sup>
carboxypropanoate 20 mg/kg	

Each values in mentioned Tables from 1-5 represents the mean  $\pm$  s.e.m. of 9-10 animals.

10 \*p<0.05 compared with control group; #p<0.05 compared with Gamma-butyrobetaine group, \$p<0.05 compared with Meldonium dihydrate group

15

As it is presented in *Tables 1-5*, Meldonium dihydrate treatment at a dose of 20 mg/kg had no therapeutical effect; gamma-butyrobetaine has decreased infarct size by 12.4 %.

3-Carboxy-*N*-ethyl-*N*,*N*-dimethylpropan-1-aminium 2-(acetyloxy)benzoate at dose of 20 mg/kg decreased infarction size by 38.4 %.

Carboxy-*N*-ethyl-*N*,*N*-dimethylpropan-1-aminium (2*E*)-3-carboxyacrylate at dose of 20 mg/kg decreased infarction size by 53.5 %.

WO 2012/146736 13 PCT/EP2012/057806

- 3-Carboxy-*N*-ethyl-*N*,*N*-dimethylpropan-1-aminium 2,6-dioxo-1,2,3,6-tetrahydropyrimidine-4-carboxylate at dose of 20 mg/kg decreased infarction size by 39.4 %.
- 3-Carboxy-*N*-ethyl-*N*,*N*-dimethylpropan-1-aminium dihydrogen phosphate at dose of 20 mg/kg decreased infarction size by 43.9 %.

5

3-Carboxy-*N*-ethyl-*N*,*N*-dimethylpropan-1-aminium 3-carboxypropanoate at dose of 20 mg/kg decreased infarction size by 37.1 %.

### Claims

20

1. 3-Carboxy-N-ethyl-N,N-dimethylpropan-1-aminium 2-(acetyloxy)benzoate

5 2. 3-Carboxy-*N*-ethyl-*N*,*N*-dimethylpropan-1-aminium (2*E*)-3-carboxyacrylate

3. 3-Carboxy-*N*-ethyl-*N*,*N*-dimethylpropan-1-aminium 2,6-dioxo-1,2,3,6-tetrahydropyrimidine-4-carboxylate

10 4. 3-Carboxy-*N*-ethyl-*N*,*N*-dimethylpropan-1-aminium dihydrogen phosphate

$$- \bigvee_{1}^{H_2 PO_4} COOH$$

5. 3-Carboxy-N-ethyl-N,N-dimethylpropan-1-aminium 3-carboxypropanoate

- 6. A process for preparing 3-carboxy-*N*-ethyl-*N*,*N*-dimethylpropan-1-aminium salt comprising:
  - a. adding *N,N*-dimethylethylamine to ethyl 4-bromobutanoate in appropriate solvent to obtain 4-ethoxy-*N*-ethyl-*N,N*-dimethyl-4-oxo-1-butanaminium bromide;
  - b. passing 4-ethoxy-*N*-ethyl-*N*,*N*-dimethyl-4-oxo-1-butanaminium bromide through ion exchange resin column to obtain 4
    [ethyl(dimethyl)ammonio] butanoate;

- c. adding acid selected from the group, consisting of 2(acetyloxy)benzoic acid, fumaric acid, orotic acid, phosphoric acid
  and succinic acid in appropriate solvent to obtain the corresponding
  3-carboxy-*N*-ethyl-*N*,*N*-dimethylpropan-1-aminium salt.
- 5 7. A process according to claim 6, wherein in step a) the appropriate solvent is acetonitrile or acetone.
  - 8. 3-Carboxy-*N*-ethyl-*N*,*N*-dimethylpropan-1-aminium salt, selected from the group consisting of 3-carboxy-*N*-ethyl-*N*,*N*-dimethylpropan-1-aminium 3-carboxypropanoate, 2-(acetyloxy)benzoate, (2*E*)-3-carboxyacrylate, 2,6-dioxo-1,2,3,6-tetrahydropyrimidine-4-carboxylate and dihydrogenphosphate for use as a medicament.
  - 9. The use according to claim 8 wherein the medicament is provided for treatment of cardiovascular diseases.

10

15

- 10. The use according to claim 9, wherein the cardiovascular disease is ischemic heart disease.
- 11. The use according to claim 10, wherein wherein the ischemic heart disease is myocardial infarction.

#### INTERNATIONAL SEARCH REPORT

International application No PCT/EP2012/057806

A. CLASSIFICATION OF SUBJECT MATTER INV. C07C55/10 C07C57/15

C07D239/557

A61K31/616

C07C69/86 A61K31/205 C07C227/18 A61K31/197

C07C229/12 A61K31/513

A61P9/00

According to International Patent Classification (IPC) or to both national classification and IPC

#### B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

C07C C07D

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practicable, search terms used)

EPO-Internal, CHEM ABS Data, WPI Data

	ENTS CONSIDERED TO BE RELEVANT	
Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
Х	WO 2010/151095 A1 (TETRA SIA [LV]; KALVINS IVARS [LV]; BIRMANS ANATOLIJS [LV];	1,8-11
γ	VEVERIS MA) 29 December 2010 (2010-12-29) abstract	2-4
Å	page 2 pages 22-29; example 10 claims 1-25	6,7
X	WO 2010/149654 A1 (GRINDEKS JSC [LV]; KALVINS IVARS [LV]; BIRMANS ANATOLIJS [LV]; VEVERIS) 29 December 2010 (2010-12-29)	5,8-11
Y A	abstract pages 2-4 examples 1-5 claims 1-15	2-4 6,7
	-/	

X Further documents are listed in the continuation of Box C.	X See patent family annex.
"Special categories of cited documents:  "A" document defining the general state of the art which is not considered to be of particular relevance  "E" earlier application or patent but published on or after the international filing date  "L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)  "O" document referring to an oral disclosure, use, exhibition or other means  "P" document published prior to the international filing date but later than the priority date claimed	"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention  "X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone  "Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art  "&" document member of the same patent family
Date of the actual completion of the international search	Date of mailing of the international search report
21 August 2012	29/08/2012
Name and mailing address of the ISA/ European Patent Office, P.B. 5818 Patentlaan 2 NL - 2280 HV Rijswijk Tel. (+31-70) 340-2040, Fax: (+31-70) 340-3016	Authorized officer  Dunet, Guillaume

# INTERNATIONAL SEARCH REPORT

International application No
PCT/EP2012/057806

C(Continua	tion). DOCUMENTS CONSIDERED TO BE RELEVANT	1017 11 20127 037 000
Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
Υ	EP 2 070 529 A1 (GRINDEKS JSC [LV]) 17 June 2009 (2009-06-17)	3
Α	abstract claims 1-7	1,2,4-11
Α	WO 97/06794 A1 (KALVINSH IVARS [LV]; VEVERIS MARIS [LV]) 27 February 1997 (1997-02-27) abstract claims 1-10	1-11
Α	US 5 965 615 A (KALVINSH IVARS [LV] ET AL) 12 October 1999 (1999-10-12) cited in the application abstract examples 1-5 claims 1-4	1-11
X,P	WO 2011/048201 A1 (GRINDEKS JSC [LV]; KALVINS IVARS [LV]; DAMBROVA MAIJA [LV]; LIEPINS ED) 28 April 2011 (2011-04-28) abstract pages 5-12 claims 1-10	1-11

# INTERNATIONAL SEARCH REPORT

Information on patent family members

International application No
PCT/EP2012/057806

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
WO 2010151095 A1	29-12-2010	CA 2766048 A1 EP 2445861 A1 KR 20120050437 A US 2012088742 A1 WO 2010151095 A1	29-12-2010 02-05-2012 18-05-2012 12-04-2012 29-12-2010
WO 2010149654 A1	29-12-2010	NONE	
EP 2070529 A1	17-06-2009	AT 527996 T EP 2070529 A1 WO 2009074498 A1	15-10-2011 17-06-2009 18-06-2009
WO 9706794 A1	27-02-1997	AT 235901 T CA 2229228 A1 DE 69627160 D1 DE 69627160 T2 EP 0845986 A1 ES 2196167 T3 JP 3072858 B2 JP H10512286 A LV 11728 B US 5859056 A WO 9706794 A1	15-04-2003 27-02-1997 08-05-2003 12-02-2004 10-06-1998 16-12-2003 07-08-2000 24-11-1998 20-08-1997 12-01-1999 27-02-1997
US 5965615 A	12-10-1999	EP 0781554 A1 JP 3119430 B2 JP 10095731 A US 5965615 A WO 9704767 A1	02-07-1997 18-12-2000 14-04-1998 12-10-1999 13-02-1997
WO 2011048201 A1	28-04-2011	EP 2491005 A1 WO 2011048201 A1	29-08-2012 28-04-2011