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BLOCK COPOLYMERS, THEIR PREPARATION AND THEIR USE AS LUBRICANTS
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- (57) Claim

1. Three-block copolyethers having the general
formula (I)



wherein:

- R is selected from H, alkyl, alkylaryl;
- X is selected from O, S and N;
- R₁, R₂ and R₃, the same or different, are selected from H, CH₃, and C₂-C₁₀ alkyl radicals;
- R₄ is selected from H, alkyl, acyl;
- x, y, z, all different from zero, represent the number of ether units of each block, are integers which have such values that the molecular weight of the compound having general formula (I) is between 600 and 4000, and in addition (x+z)/y is between 1.5 and 5.0;

with the proviso that:

- a) either R₁ and R₃ are selected from H and CH₃, in

which case R_2 is selected from C_2 to C_{10} alkyl radicals,

b) or R_1 and R_3 are selected from C_2 - C_{10} alkyl radicals, in which case R_2 is selected from H and CH_3 .

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Invention Title:

BLOCK COPOLYMERS, THEIR PREPARATION AND THEIR USE AS LUBRICANTS

Our Ref : 457126
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The following statement is a full description of this invention, including the best method of performing it known to applicant(s):

BLOCK COPOLYMERS, THEIR PREPARATION AND THEIR USE AS
LUBRICANTS

The present invention relates to new oil-soluble
polyethers, their preparation, and their use in the
5 field of lubricants, particular for motor oil.

More specifically, the polyethers of the present
invention are characterized by a three-block structure
of the type ABA wherein A and B are blocks of alkylene-
oxides.

10 Lubricating oils for motors consist of formula-
tions of lubricant bases such as mineral oils, synthet-
ic bases of the hydrocarbon and ester type, mixed with
suitable additive systems.

The possibility of formulating lubricating oils
15 with polyglycols is described in literature, particu-
larly to improve some of the properties of the oils,
mainly in terms of high viscosity indexes, good filming
properties and anti-wear performance, high flash-
points.

The difficulty in asserting the use of polyglycols as lubricating bases in motor oil formulations is mainly due to their marked hydrophilic nature (polyethyleneoxides) which can be shown by their limited
5 miscibility both with mineral bases and with additives.

Oil-soluble polyglycols have recently been synthesized starting from α -epoxides substituted with alkyl groups (ethyl, butyl, hexyl etc.) capable of completely mixing with all components of motor oils.

10 For example US-A-4.481.123 describes the synthesis of a block polyglycol particularly suitable for transmission gears. This polyglycol consists of a block of polytetrahydrofuran obtained by cationic polymerization and a block of C_8-C_{26} α -alkylepoxy obtained by anionic
15 polymerization.

US-A-4.793.939 relates to the formulation of a lubricating oil consisting of a paraffinic and naphthenic mineral base and a two-block oil-soluble polyglycol base consisting of a C_2-C_4 block and another
20 C_6-C_{40} block. The introduction of the polyalkyleneglycol improves the viscosity index, the resistance to wear and friction, the good properties of the mineral base remaining unaltered.

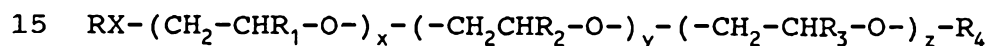
EP-A-355.977 describes the synthesis of a polyglycol to be used in formulations of two or four phase
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motor oils and in industrial lubricants.

The percentage of mineral base can vary from 0 to 40% and that of the polyglycol from 100 to 60%. It consists of a diblock structure consisting of from 35 to 80% of C_2-C_4 epoxides and from 1 to 35% of C_6-C_{30} epoxide.

The triblock polyethers which are the object of the present invention, while maintaining the excellent viscosity indexes and good viscosities at 40 and 100°C of the above polyglycols, additionally have excellent lubricating and anti-wear properties.

In accordance with this, the present invention relates to three-block copolyethers having general formula (I):



wherein:

- R is selected from H, alkyl, alkylaryl;
- X is selected from O, S and N;
- R_1, R_2 and R_3 , the same or different, are selected from H, CH_3 , and C_2-C_{10} alkyl radicals;
- R_4 is selected from H, alkyl, acyl;
- x, y, z, all different from zero, represent the number of ether units of each block, are integers which have such values that the molecular weight of the compound having general formula (I) is between 600 and 4000,

preferably between 800 and 2000, and in addition $(x+z)/y$ is between 1.5 and 5.0, preferably between 2.2 and 3.2,

with the proviso that:

- 5 a) either R_1 and R_3 are selected from H and CH_3 , in which case R_2 is selected from C_2 to C_{10} alkyl radicals, b) or R_1 and R_3 are selected from C_2 - C_{10} alkyl radicals, in which case R_2 is selected from H and CH_3 .

In one form of embodiment, the polyether having
10 general formula (I) has R and $R_2 = n\text{-C}_4\text{H}_9$, $R_3 = \text{CH}_3$, $R_1 = \text{CH}_3$, $R_4 = \text{H}$, $x = 6$, $y = 5$, $z = 6$.

In another form of embodiment, the polyether having general formula (I) has R and $R_2 = n\text{-C}_4\text{H}_9$, $R_3 = \text{CH}_3$, $R_1 = \text{CH}_3$, $R_4 = \text{H}$, $x = 6$, $y = 4$, $z = 6$.

15 In another form of embodiment, the compound having general formula (I) has $R = R_2 = R_4 = \text{H}$, $R_1 = R_3 = n\text{-C}_4\text{H}_9$, $x = 5$, $y = 4$, $z = 5$.

For the final properties of the product having general formula (I), the ratio $(x+z)/y$, which deter-
20 mines the solubility in oil of the copolymer having general formula (I), is determinant.

The compounds of the present invention are prepared by three anionic polymerization steps. The initiator molecule, which must contain at least one
25 hydrogen atom suitable for being alkoxyated, can be an

alcohol, a thiol, an amine. The basic catalysts used in the polymerization of each block comprise sodium hydroxide, potassium hydroxide, strontium carbonate, preferably potassium hydroxide.

5 When, for example, a compound having general formula (I) is to be prepared, x moles of epoxide $(\text{CH}_2-\text{CHR}_1)=\text{O}$, y moles of epoxide $(\text{CH}_2-\text{CHR}_2)=\text{O}$ and finally z moles of epoxide $(\text{CH}_2-\text{CHR}_3)=\text{O}$ are linked to the initiator molecule. Alternatively, it is possible to
10 react a polyglycol having two terminal $-\text{OH}$ with one or more alkylepoxides, still in the presence of basic catalysis. In this case the starting polyglycol acts as initiator and the final three-block polyether will contain, as internal block, the starting polyglycol.

15 At the end of the reaction with the epoxides, the reaction is inactivated by the addition of water, preferably in the form of sodium acetate trihydrate.

 The polyether of the present invention has a molecular weight of between 600 and 4000 dalton,
20 preferably between 800 and 2000 dalton.

 This new group of polyethers is characterized by a high compatibility and miscibility with mineral oils, a low capacity of water absorption, high viscosity indexes, the possibility of modifying the viscosity at
25 40 and 100°C by changing the ratio of the three poly-

alkyleneoxide blocks, excellent lubricating and anti-wear characteristics.

A further object of the present invention therefore relates to the use of the compound having general
5 formula (I) as a lubricating oil for industrial gearing, differential gears and two-cycle engine oil.

The compound having general formula (I) can also be used as a component of formulations of partially or totally synthetic lubricating oils for 4-cycle engines.
10 These latter formulations, which can also contain other additives normally used in lubricating oils (for example antioxidants, dispersers, detergents), comprise from 5 to 50% of the polyether having general formula (I), from 25 to 80% of synthetic lubricating bases and
15 from 15 to 25% of additives.

The following experimental examples provide a better illustration of the present invention.

EXAMPLE 1 - Preparation of the compound having general formula (I) wherein R and R_2 = butyl, $R_1 = R_3$ = methyl,
20 R_4 = hydrogen, $x = 6$, $y = 5$, $z = 6$.

The reaction was carried out in a 1 litre pressure-resistant reactor equipped with a magnetic drag stirrer and with one inlet for the epoxides and another for the nitrogen for the flushings, a thermometer, a
25 manometer and a safety valve. 54.9 g (0.49 moles) of

potassium butylate, previously synthesized starting from butanol and potassium hydroxide, were introduced into the reactor. The reactor flushed with nitrogen was brought to about 90°C and subsequently 165 g (2.84
5 moles) of propyleneoxide were added over a period of 60 minutes. At the end of the addition the reactor was left at 130°C for 1 hour. 240 g (2.4 moles) of hexeneoxide were then introduced into the reactor heated to 150°C over a period of 1 hour. The mixture was left to
10 react for 1 hour and finally 165 g of propyleneoxide were added over a period of an hour. At the end of the third addition the reactor was left at 150°C for a further hour.

The reactor was cooled to about 70°C and the
15 oligomer in the form of a potassium salt was subsequently discharged and treated with 38 g (0.28 moles) of sodium acetate trihydrate. The mixture was left under stirring for 12 hours at about 40°C. The oligomer in the form of an alcohol was then recovered by filtra-
20 tion on a porous septum. The reaction yield was 98.1%.

The molecular weight of the oligomer is 1272 and the polydispersity 1.17. The percentage of the two components in the final oligomer is 54.7% of propyleneoxide and 39.4% of hexeneoxide.

25 The oligomer was then characterized both from a

rheological and tribological point of view. The results are shown in table 1, which indicates, for comparative purposes, the characteristics of two commercial polyglycols, one, called NS9, consisting of about 85% by weight of polypropyleneoxide and the remaining 15% of polyethyleneoxide, the other, called SYN OA 60, consisting of a polybutyleneoxide.

TABLE 1

=====				
10	Characteristics	Ex.1	NS9	SYN OA 60

	Visc.100°C, cSt	7.66	11.21	9.3
	Visc. 40°C, cSt	43.08	57.19	57.6
	Viscosity index	147	193	143
15	-----			
	4 Wear Spheres;	0.35	0.59	0.71
	Wear diameter, mm			

	Almen- Wieland;	2050	600	600
20	Ult. tensile strength, kg			

	SRV; Friction	0.1025	0.1074	0.1095
	Coefficient			
=====				

25 From the data of table 1, it can be seen that the

product of example 1 of the present invention has better tribological characteristics than those of commercial polyethers. This proves the applicability of the product of example 1 as lubricant for industrial
5 gearing, differential gears, 2-phase motor oil.

With respect to the measurements of table 1, the kinematic viscosity at 40°C was carried out according to ASTM D445; the kinematic viscosity at 100°C according to ASTM D445; the viscosity index according to ASTM
10 D2270; the 4 wear sphere test according to ASTM D4172; the SRV according to DIN 51834.

In addition the copolymer of example 1 has complete mixability with mineral oils in all proportions.

The triblock copolymer of example 1 was also
15 tested in a "full synthetic" formulation. A formulation (SYN/P) was prepared, consisting of 65% by weight of PAO (polyalphaolefins), 15% by weight of the above product of example 1 and 20% of additives. For comparative purposes, a formulation (SYN/E) was prepared,
20 consisting of 65% by weight of PAO, 15% of esters of fatty acids with neopentyl alcohol and 20% of additives.

The 4 Wear sphere measurement of the SYN/P mixture (wear diameter in mm) gave a value of 0.49 against a
25 value of 0.51 for the comparative mixture SYN/E.

The SRV measurement (friction coefficient) of the mixture SYN/P gave a value of 0.141 against the value of 0.122 for the mixture SYN/E.

The data relating to the "full synthetic" formulation show that the composition of the present invention has tribological performances comparable to those of a commercial oil with SH performances and can therefore be used as a lubricating oil for 4-phase motors.

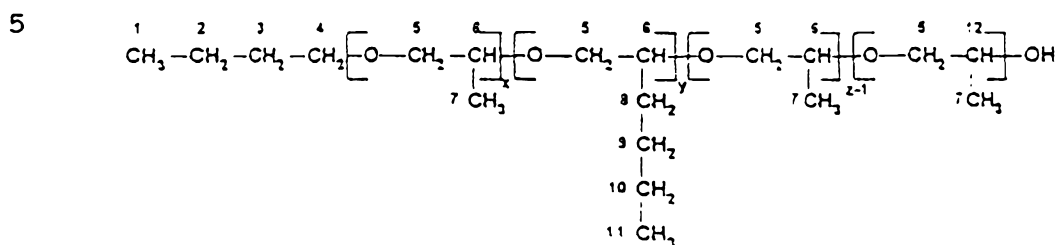
EXAMPLE 2 -Preparation of the compound having general formula (I) where R and R₂ = butyl, R₁ = R₃ = methyl, R₄ = hydrogen, x = 6, y = 4, z = 6.

The reaction was carried out in a 50 ml pressure-resistant reactor, equipped with a magnetic drag stirrer and with one inlet for the epoxides and one for the nitrogen for the flushings, a thermometer, a manometer and a safety valve. Using the same procedure described in example 1, 2.8 g (25 mmoles) of potassium butylate were reacted with 8.7 g (150 mmoles) of propyleneoxide, then with 10 g (100 mmoles) of hexeneoxide and finally with 8.7 g (150 mmoles) of propyleneoxide. The oligomer in the form of a potassium salt was treated with 1.2 g (8.8 mmoles) of sodium acetate trihydrate and finally recovered by filtration in alcoholic form. The reaction yield was 96%.

The molecular weight of the oligomer is 1172 and

the polydispersity 1.17.

The allocation of the different protons (via $^1\text{H-NMR}$ and $^{13}\text{C-NMR}$) of the three-block polyethers of examples 1 and 2, is given below.



$^1\text{H-NMR}$ (CD_2Cl_2): $\delta=0.9$ (m, C^1+C^{11}), 1.1-1.2 (m, C^7), 1.3-1.6 (m, $\text{C}^2+\text{C}^3+\text{C}^8+\text{C}^9+\text{C}^{10}$), 3.2-3.5 (m, C^5+C^6), 3.7-3.9 (m, C^{12}) ppm

$^{13}\text{C-NMR}$ (CD_2Cl_2): $\delta=14$ (C^1+C^{11}), 17.1-18.7 (C^7), 19.7 (C^2), 23.2 (C^{10}), 28 (C^9), 31.9 (C^8), 32.2 (C^3), 65.7-66.3 (C^{12}), 71.4 (C^4), 72-73.5 (C^5), 74-80 (C^6) ppm.

The oligomer was then characterized both from a rheological and from a tribological point of view with the following results:

Kinematic viscosity at 40°C (ASTM D445) = 36.75 cSt

Kinematic viscosity at 100°C (ASTM D445) = 6.29 cSt

Viscosity index (ASTM D2270) = 121

Complete miscibility with mineral oils in all proportions.

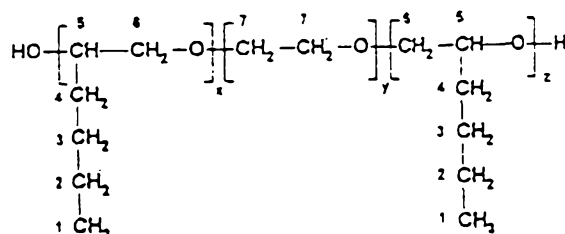
EXAMPLE 3 - Preparation of the compound having general formula (II) wherein $\text{R} = \text{R}_2 = \text{R}_4 = \text{H}$, $\text{R}_1 = \text{R}_3 = \text{butyl}$,

$x = 5, y = 4, z = 5.$

The reaction was carried out in a 50 ml pressure-resistant reactor, equipped with a magnetic drag stirrer and with one inlet for the epoxides and one for the nitrogen for the flushings, a thermometer, a manometer and a safety valve. Using the same procedure described in example 1, 1.8 g (6.7 mmoles) of potassium salt of tetraethyleneglycol were reacted with 17.5 g (174.7 mmoles) of hexeneoxide. The oligomer in the form of potassium salt was treated with 1.2 g (8.8 mmoles) of sodium acetate trihydrate and finally recovered by filtration in alcoholic form. The reaction yield was 97.5%.

The molecular weight of the oligomer is 1177 and the polydispersity 1.09.

The allocation of the different protons (via $^1\text{H-NMR}$ and $^{13}\text{C-NMR}$) of the three-block polyethers of examples 1 and 3, is given below.



$^1\text{H-NMR}$ (CD_2Cl_2): $\delta=0.8$ (t, C^1), 1.1-1.4 (m, $\text{C}^2+\text{C}^3+\text{C}^4$),

3.2-3.4 (m, C^6+C^7), 3.5-3.6 (m, C^5) ppm

$^{13}\text{C-NMR}$ (CD_2Cl_2): $\delta=13.8$ (C^1), 22.6 (C^2), 27.5 (C^3), 31 (C^4),

70-75 ($C^6 + C^7$), 78-80 (C^5) ppm

The oligomer was then characterized both from a rheological and from a tribological point of view with the following results:

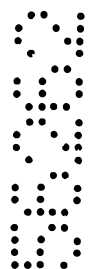
5 Kinematic viscosity at 40° C (ASTM D445) = 70.79 cSt

Kinematic viscosity at 100° C (ASTM D445) = 9.56 cSt

Viscosity Index (ASTM D2270) = 114

Complete miscibility with mineral oils in all proportions

10 Throughout the description and claims of the specification the word "comprise" and variations of the word, such as "comprising" and "comprises" is not intended to exclude other additives, components, integers or steps.



THE CLAIMS DEFINING THE INVENTION ARE AS FOLLOWS:

1. Use of three-block copolyethers having the general formula (I)



5 wherein:

-R is selected from H, alkyl, alkylaryl;

-X is selected from O, S and N;

-R₁, R₂ and R₃, the same or different, are selected from H, CH₃, and C₂-C₁₀ alkyl radicals;

10 -R₄ is selected from H, alkyl, acyl;

-x, y, z, all different from zero, represent the number of ether units of each block, are integers which have such values that the molecular weight of the compound having general formula (I) is between 600 and 4000, and in addition (x+z)/y is between 1.5 and 5.0;

15 with proviso that:

a) either R₁ and R₃ are selected from H and CH₃ in which case R₂ is selected from C₂ to C₁₀ alkyl radicals;

b) or R₁ and R₃ are selected from C₂-C₁₀ alkyl radicals, in which case R₂ is selected from H and CH₃, as anti-wear additives in lubricant oils.

2. Use of three-block copolyethers according to claim 1, wherein the ratio (x+z)/y is between 2.2 and 3.2.

3. Use of three-block copolyethers according to claim 1, wherein they have a molecular weight of between 800 and 2000 dalton.

4. Use of three-block copolyethers according to claim 1, wherein R and R₂ = n-C₄H₉, R₁ = R₃ = CH₃, R₄ = H, x = 6, y = 5, z = 6.

5. Use of three-block copolyethers according to claim 1, wherein R = R₂ = R₄ = H, x = 6, y = 5, z = 6.



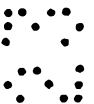
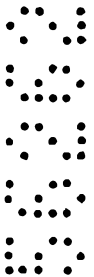
6. Use of three-block copolyethers according to claim 1, wherein $R = R_2 = R_4 = H$, $R_1 = R_3 = \text{butyl}$, $x = 5$, $y = 4$, $z = 5$.
- 5 7. Use of the compound according to claim 1 as a lubricant for industrial gearing, differential gears and 2-cycle engine oil.
8. Use according to claim 1 in lubricating oils for 4-phase motors.
- 10 9. Use of three-block copolyethers according to claim 1 substantially as hereinbefore described with reference to any of the examples.

DATED: 22 February 1999

PHILLIPS ORMONDE & FITZPATRICK

15 Attorneys for:

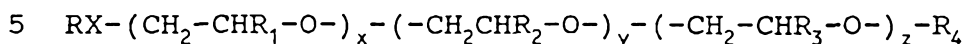
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BLOCK COPOLYMERS, THEIR PREPARATION AND THEIR USE AS
LUBRICANTS

Abstract

Block copolyethers having general formula (I):



wherein:

- R is selected from H, alkyl, alkylaryl;
- X is selected from O, S and N;
- R_1, R_2 and R_3 are selected from H, CH_3 , and C_2-C_{10} alkyl
10 radicals;
- R_4 is selected from H, alkyl, acyl;
- x, y, z represent the number of ether units and are
integers which have such values that the molecular
weight of the compound having general formula (I) is
15 between 600 and 4000, and in addition $(x+z)/y$ is
between 1.5 and 5.0; with the proviso that:
 - a) either R_1 and R_3 are selected from H and CH_3 , in
which case R_2 is selected from C_2 to C_{10} alkyl radicals,
 - b) or R_1 and R_3 are selected from C_2-C_{10} alkyl radicals,
20 in which case R_2 is selected from H and CH_3 .