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The instant disclosure describes methacrylate coated carrier resins with hydrophobic substituents for charge control and improved RH sensitivity.

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ABSTRACT

The instant disclosure describes methacrylate coated carrier resins with hydrophobic substituents for charge control and improved RH sensitivity.

NEGATIVE CHARGE CARRIER RESINS WITH RH SENSITIVITY

FIELD

[0001] The instant disclosure relates generally to carrier resins, and specifically, carrier coating resins having both an ester functionality and a hydrophobic substituent where such resins exhibit both improved RH sensitivity and high and tunable charge.

BACKGROUND

[0002] The properties of a toner are set through the selection of materials and amounts of the materials of the toner. The charging characteristics of a toner are also dependent on the carrier used in a developer composition, such as, the carrier coating. Toners typically comprise one or more external surface additives, for example, silica, titanium dioxide, zinc stearate and the like. The carrier coating can interact with a toner surface additive.

[0003] Toners having a triboelectric charging property within the range of about $-30\ \mu\text{C/g}$ and about $-45\ \mu\text{C/g}$ may be achieved when using smaller-sized silica particles as external additives, for example, silica particles having average sizes less than 20 nm, such as, for example, R805 ($\sim 12\ \text{nm}$) and/or R972 ($\sim 16\ \text{nm}$) (Evonik, NJ). However, developability at areas of low toner area coverage degrades over time. That has been attributed to the small-sized additives being impacted into the toner surface over time.

[0004] The problem with smaller-sized additives has been addressed by using larger-sized additives, i.e., additives having a size of 40 nm or larger such as, for example, RX50 silica, RX515H silica or SMT5103 titania (Evonik, NJ). However, although the above developability problem is addressed, the toners do not exhibit as high a triboelectric charging ability and also exhibit charge through.

[0005] New carrier coatings are being developed that enable higher charging developers, particularly those with larger-sized additive package components. However, when such developers are tested at low area coverage followed by high area coverage, the developers tend to exhibit low or wrong sign toner due to charge through, i.e., the incumbent toner in the device becomes less negative or even wrong sign, i.e., positive, and the new (fresh) toner

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added may charge very negative. The presence of low charge and/or wrong sign toner can result in objectionable background.

[0006] There remain problems with providing high charge with good relative humidity (RH) sensitivity of charge to changing environmental conditions for carrier coating resin designs. For example, there remains a requirement to be able to tune the charge of the carrier resin to produce higher charge for those situations where higher charge is required. To obtain both high charge and good RH sensitivity through a single design has been challenging.

[0007] Further, toners containing silica as a surface additive, which silica can be a charge driver for the toner, are known to be very RH sensitive. Thus, it is important to provide new carrier resin designs that work well with silica to improve RH sensitivity, while maintaining high charge.

SUMMARY

[0008] The instant disclosure describes methacrylate carrier polymer resins having hydrophobic substituent groups for charge control and improved RH sensitivity, including carrier compositions comprising said methacrylate carrier polymer resins and developers comprising said carrier resins.

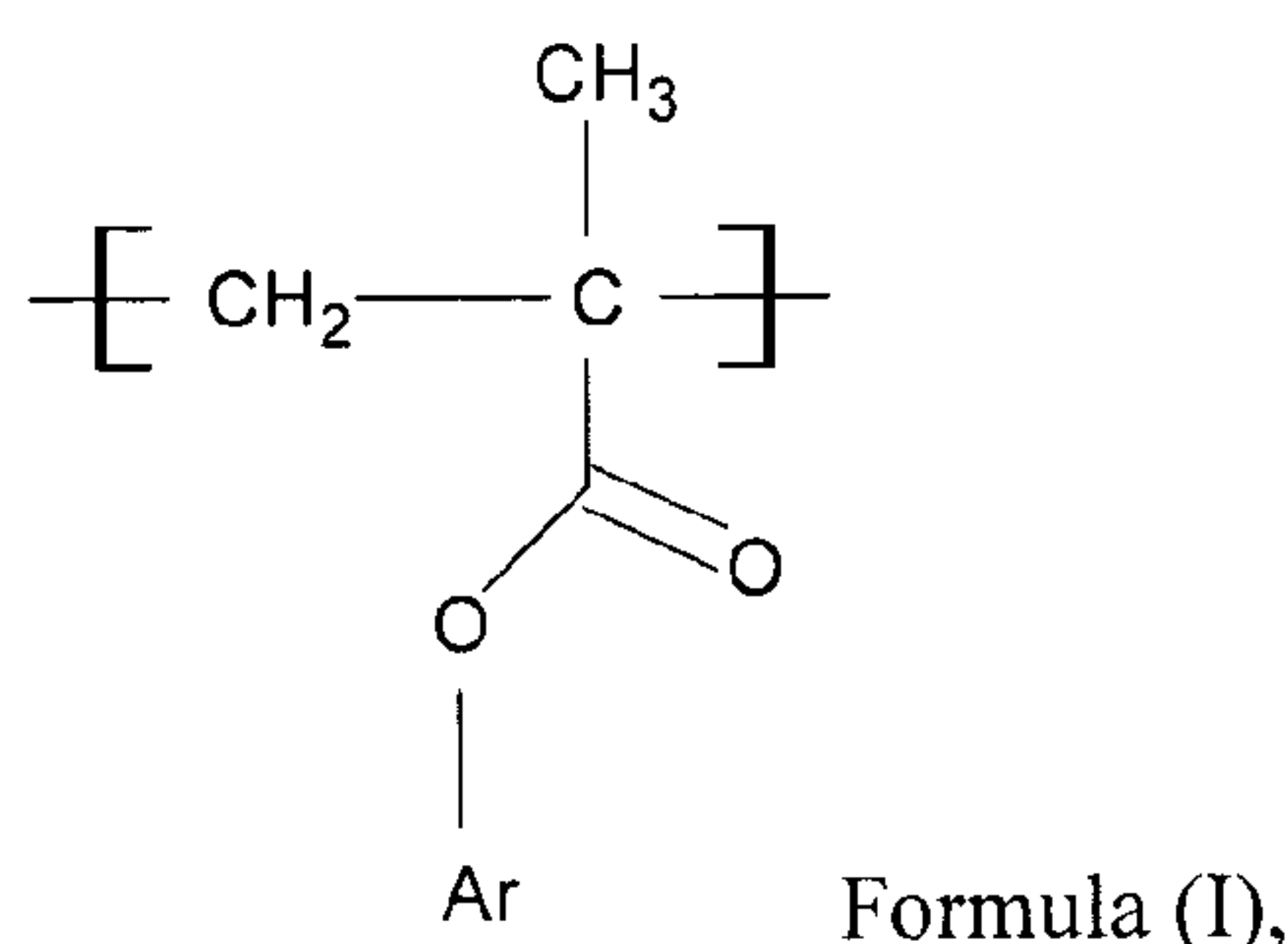
[0009] In embodiments, a carrier composition is disclosed including a polymer coating resin having at least one methacrylate monomer and a hydrophobic substituent, where the highest occupied molecular orbital (HOMO) of the methacrylate-containing polymer unit has greater than about 25% of the electron density of those molecular orbitals located on the hydrophobic substituent and less than about 75% of the electron density of those molecular orbitals on the carbonyl oxygen of the methacrylate moiety.

[0010] In embodiments, a developer is disclosed comprising a carrier composition comprising a polymer coating resin comprising at least one methacrylate monomer; a toner; and a toner surface additive, where the carrier composition further comprises a hydrophobic substituent, where the resin has the HOMO electron density substantially located on the hydrophobic substituent, and where the HOMO electron density located on the carbonyl function of the at least one methacrylate monomer is effectively reduced.

[0017] In accordance with an aspect, there is provided a toner carrier composition comprising:

a polymer coating resin consisting of a methacrylate monomer having a hydrophobic substituent, Ar, and a secondary aminoacrylate monomer, wherein the highest occupied molecular orbital (HOMO) of the methacrylate-containing polymer has greater than about 25% of the electron density of those molecular orbitals located on the hydrophobic substituent and less than about 75% of the electron density of those molecular orbitals on the carbonyl group of the methacrylate moiety, and wherein said methacrylate monomer having a hydrophobic substituent is present from about 98% to about 100% by weight of the polymer and said secondary aminoacrylate monomer is present (a) from about 0.5 to about 1%; or (b) from about 1 to about 1.5%; or (c) from about 1.5 to about 2% by weight of the polymer,

further wherein the methacrylate monomer has the formula as set forth in formula (I):



wherein Ar comprises at least one aromatic group, and

the secondary amino acrylate monomer is selected from the group consisting of dimethylaminoethyl methacrylate, diethylaminoethyl methacrylate, and diethylaminopropyl methacrylate; and

a carrier core particle, wherein the polymer coating resin is adhered to the surface of the carrier core particles.

BRIEF DESCRIPTION OF THE DRAWINGS

[0011] The foregoing summary, embodiments and other aspects of the present disclosure will be best understood with reference to a detailed description of specific embodiments, which follows, when read in conjunction with the accompanying drawings.

[0012] Figure 1 shows a schematic diagram of HOMO and LUMO energies for a carrier resin and toner additive. HOMO is highest occupied molecular orbital and LUMO is lowest unoccupied molecular orbital.

[0013] Figure 2 shows a schematic diagram of HOMO and LUMO energies for a carrier resin and toner additive.

DETAILED DESCRIPTION

[0014] The disclosure relates to methacrylate carrier resins where the resins have the highest occupied molecular orbital (HOMO) electron density substantially located on a hydrophobic group, and where the HOMO electron density located on the carbonyl function of said methacrylate is effectively reduced, thereby obtaining both high charge and improvement in RH sensitivity.

[0015] In embodiments, a carrier composition is disclosed including a polymer coating resin having at least one methacrylate monomer and a hydrophobic substituent, where the HOMO of the methacrylate-containing polymer unit has greater than about 25% of the electron density of those molecular orbitals located on the hydrophobic substituent and less than about 75% of the electron density of those molecular orbitals on the carbonyl oxygen of the methacrylate moiety.

[0016] By, "RH sensitivity," is meant a toner retains or comprises sufficient charge, particularly when exposed to higher humidity levels. The RH ratio is a useful metric for determining RH sensitivity. RH ratio generally is the ratio of charge, in $\mu\text{C/g}$, in the A zone to that in the C zone, where the A zone generally comprises 28°C and 85% RH and the C zone generally comprises 12°C and 15% RH. The Q/M charging RH ratio for the developer can be quantified as the ratio of the toner Q/M in A-zone to the toner Q/M in C-zone measured under otherwise similar conditions. The RH ratio can also be measured as the ratio

of the toner charge Q/D in A-zone to the toner Q/D in C-zone. Thus, the RH ratio defined in those ways generally takes a value from 0 if the charge in A-zone is zero, to a value of 1, if the charge in both zones is the same. Thus, a value of 0.5 RH ratio of Q/M would indicate the Q/M charge in A-zone was one-half the value in C-zone. In some instances the value of the RH ratio might be larger than 1, indicating that charge in A-zone is higher than charge in C-zone when measured under otherwise similar conditions, though this is not common. An RH ratio could in some situations be negative, indicating that the sign of the charge also changes with the change in environmental zones, which is also uncommon. It should be noted that other environmental zones may be used instead of A-zone and C-zone to define the RH ratio of the developer.

[0017] As used herein, the modifier, "about," used in connection with a quantity is inclusive of the stated value and has the meaning dictated by the context (for example, it includes at least the degree of error associated with the measurement of the particular quantity). In embodiments, "about," is interpreted to no more than 10% of a stated value. When used in the context of a range, the modifier "about" should also be considered as disclosing the range defined by the absolute values of the two endpoints. For example, the range "from about 2 to about 4" also discloses the range "from 2 to 4."

[0018] By, "negative additives that are negatively chargeable to a reference carrier," is meant that the additives are negatively charging relative to the carrier surface measured by determining the triboelectric charge of the additive by mixing same directly with the carrier. Similarly, by, "positive additives that are positively chargeable to a carrier," is meant that the additives are positively charging relative to the carrier surface measured by determining the triboelectric charge of the additive by mixing same directly with the carrier.

[0019] Negative additives that are negatively chargeable to a carrier include, for example, silica particles, alumina particles, or any small-sized particles (e.g., from about 7 to about 100 nm in volume average particle diameter as determined by any suitable technique) including, for example, polymeric microspheres, optionally treated with a composition rendering the particles negatively chargeable to a carrier on triboelectric contact therewith. The treating material may be, for example, a fluorosilane, such as exemplified in U.S. Pat. No. 4,973,540, other halogen-containing

organosilanes such as described in U.S. Pat. No. 5,376,172, incorporated herein by reference in entirety, silazanes, siloxanes and the like.

Toners

[0020] In embodiments, a developer is disclosed comprising a methacrylate-coated carrier and a toner, where the toner comprises a resin, such as, for example, an emulsion aggregation toner, comprising, but not limited to, one or more of a latex resin, a colorant, a wax, and a polymer shell.

[0021] In embodiments, the latex resin may be composed of a first and a second monomer composition. Any suitable monomer or mixture of monomers may be selected to prepare the first monomer composition and the second monomer composition. The selection of monomer or mixture of monomers for the first monomer composition is independent of that for the second monomer composition, and vice versa. Exemplary monomers for the first and/or the second monomer compositions include, but are not limited to, styrene, alkyl acrylate such as methyl acrylate, ethyl acrylate, butyl acrylate, isobutyl acrylate, dodecyl acrylate, n-octyl acrylate, 2-chloroethyl acrylate; β -carboxy ethyl acrylate (β -CEA), phenyl acrylate, methyl alphachloroacrylate, methyl methacrylate, ethyl methacrylate, butyl methacrylate, butadiene, isoprene; methacrylonitrile, acrylonitrile; vinyl ethers such as vinyl methyl ether, vinyl isobutyl ether, vinyl ethyl ether and the like; vinyl esters such as vinyl acetate, vinyl propionate, vinyl benzoate, vinyl butyrate; vinyl ketones such as vinyl methyl ketone, vinyl hexyl ketone, methyl isopropenyl ketone and the like; vinylidene halides such as vinylidene chloride, vinylidene chlorofluoride and the like; N-vinyl indole, N-vinyl pyrrolidene and the like; methacrylate, acrylic acid, methacrylic acid, acrylamide, methacrylamide, vinylpyridine, vinylpyrrolidone, vinyl-N-methylpyridinium chloride, vinyl naphthalene, p-chlorostyrene, vinyl chloride, vinyl bromide, vinyl fluoride, ethylene, propylene, butylene, isobutylene, and the like, and the mixture thereof. In case a mixture of monomers is used, typically the latex polymer will be a copolymer.

[0022] Illustrative examples of such latex copolymer include poly(styrene-n-butyl acrylate- β -CEA), poly(styrene-alkyl acrylate), poly(styrene-1,3-diene), poly(styrene-alkyl methacrylate), poly(alkyl methacrylate-alkyl acrylate), poly(alkyl methacrylate-aryl acrylate), poly(aryl methacrylate-alkyl acrylate), poly(alkyl methacrylate), poly(styrene-alkyl

acrylate-acrylonitrile), poly(styrene-1,3-diene-acrylonitrile), poly(alkyl acrylate-acrylonitrile), poly(styrene-butadiene), poly(methylstyrene-butadiene), poly(methyl methacrylate-butadiene), poly(ethyl methacrylate-butadiene), poly(propyl methacrylate-butadiene), poly(butyl methacrylate-butadiene), poly(methyl acrylate-butadiene), poly(ethyl acrylate-butadiene), poly(propyl acrylate-butadiene), poly(butyl acrylate-butadiene), poly(styrene-isoprene), poly(methylstyrene-isoprene), poly(methyl methacrylate-isoprene), poly(ethyl methacrylate-isoprene), poly(propyl methacrylate-isoprene), poly(butyl methacrylate-isoprene), poly(methyl acrylate-isoprene), poly(ethyl acrylate-isoprene), poly(propyl acrylate-isoprene), poly(butyl acrylate-isoprene); poly(styrene-propyl acrylate), poly(styrene-butyl acrylate), poly(styrene-butadiene-acrylonitrile), poly(styrene-butyl acrylate-acrylonitrile), and the like.

[0023] In embodiments, the first monomer composition and the second monomer composition may be substantially water insoluble, generally hydrophobic, or may be readily dispersed in the aqueous phase with adequate stirring when added to the reaction vessel.

[0024] The weight ratio of the first monomer composition and the second monomer composition may be generally in the range of from about 0.1:99.9 to about 50:50, from about 0.5:99.5 to about 25:75, from about 1:99 to about 10:90.

[0025] In embodiments, the first monomer composition and the second monomer composition are the same. Examples of the first/second monomer composition may be a mixture comprising styrene and alkyl acrylate such as a mixture comprising styrene, n-butyl acrylate and β -CEA. Based on total weight of the monomers, styrene may generally be present in an amount from about 1% to about 99%, from about 50% to about 95%, from about 70% to about 90%, although it may be present in greater or lesser amounts; alkyl acrylate such as n-butyl acrylate may generally be present in an amount from about 1% to about 99%, from about 5% to about 50%, from about 10% to about 30%, although may be present in greater or lesser amounts.

[0026] In embodiments, a suitable resin is a polyester. Suitable polyester resins include, for example, those which are sulfonated, non-sulfonated, crystalline, amorphous, combinations thereof and the like. The polyester resins may be linear, branched, crosslinked, combinations thereof and the like. Polyester resins may include those described, for example, in U.S. Pat.

Nos. 6,593,049; 6,830,860; 7,754,406; 7,781,138; 7,749,672; and 6,756,176.

[0027] When a mixture is used, such as, amorphous and crystalline polyester resins, the ratio of crystalline polyester resin to amorphous polyester resin may be in the range from about 1:99 to about 30:70; from about 5:95 to about 25:75; from about 5:95 to about 15:95.

[0028] A polyester resin may be obtained synthetically, for example, in an esterification reaction involving a reagent comprising a carboxylic acid group and another reagent comprising an alcohol. In embodiments, the alcohol reagent comprises two or more hydroxyl groups, in embodiments, three or more hydroxyl groups. In embodiments, the acid comprises two or more carboxylic acid groups, in embodiments, three or more carboxylic acid groups. Reagents comprising three or more functional groups enable, promote or enable and promote polymer branching and crosslinking. In embodiments, a polymer backbone or a polymer branch comprises at least one monomer unit comprising at least one pendant group or side group, that is, the monomer reactant from which the unit was obtained comprises at least three functional groups.

[0029] Examples of polyacids or polyesters that may be used for preparing an amorphous polyester resin include terephthalic acid, phthalic acid, isophthalic acid, fumaric acid, trimellitic acid, diethyl fumarate, dimethyl itaconate, cis-1,4-diacetoxy-2-butene, dimethyl fumarate, diethyl maleate, maleic acid, succinic acid, itaconic acid, succinic acid, cyclohexanoic acid, succinic anhydride, dodecylsuccinic acid, dodecylsuccinic anhydride, glutaric acid, glutaric anhydride, adipic acid, pimelic acid, suberic acid, azelaic acid, dodecanedioic acid, dimethyl naphthalenedicarboxylate, dimethyl terephthalate, diethyl terephthalate, dimethylisophthalate, diethylisophthalate, dimethylphthalate, phthalic anhydride, diethylphthalate, dimethylsuccinate, naphthalene dicarboxylic acid, dimer diacid, dimethylfumarate, dimethylmaleate, dimethylglutarate, dimethyladipate, dimethyl dodecylsuccinate, and combinations thereof. The polyacid or polyester reagent may be present, for example, in an amount from about 40 to about 60 mole% of the resin, from about 42 to about 52 mole% of the resin, from about 45 to about 50 mole% of the resin, and optionally a second polyacid can be used in an amount from about 0.1 to about 10 mole% of the resin.

[0030] Examples of polyols which may be used in generating an amorphous polyester resin include 1,2-propanediol, 1,3-propanediol, 1,2-butanediol, 1,3-butanediol, 1,4-butanediol, pentanediol, hexanediol, 2,2-dimethylpropanediol, 2,2,3-trimethylhexanediol, heptanediol, dodecanediol, bis(hydroxyethyl)-bisphenol A, bis(2-hydroxypropyl)-bisphenol A, 1,4-cyclohexanedimethanol, 1,3-cyclohexanedimethanol, xylenedimethanol, cyclohexanediol, diethylene glycol, bis(2-hydroxyethyl) oxide, dipropylene glycol, dibutylene glycol, and combinations thereof. The amount of polyol may vary, and may be present, for example, in an amount from about 40 to about 60 mole% of the resin, from about 42 to about 55 mole% of the resin, from about 45 to about 53 mole% of the resin, and a second polyol may be used in an amount from about 0.1 to about 10 mole%, from about 1 to about 4 mole% of the resin..

[0031] Polycondensation catalysts may be used in forming the amorphous (or crystalline) polyester resin, and include tetraalkyl titanates, dialkyltin oxides, such as, dibutyltin oxide, tetraalkyltins, such as, dibutyltin dilaurate, and dialkyltin oxide hydroxides, such as, butyltin oxide hydroxide, aluminum alkoxides, alkyl zinc, dialkyl zinc, zinc oxide, stannous oxide, or combinations thereof. Such catalysts may be used in amounts of, for example, from about 0.01 mole% to about 5 mole% based on the starting polyacid or polyester reagent(s) used to generate the polyester resin.

[0032] In embodiments, the resin may be a crosslinkable resin. A crosslinkable resin can be a resin including a crosslinkable group or groups such as a C=C bond or a pendant group or side group, such as, a carboxylic acid group. The resin may be crosslinked, for example, through a free radical polymerization with an initiator.

[0033] Examples of amorphous resins which may be used include alkali sulfonated-polyester resins, branched alkali sulfonated-polyester resins, alkali sulfonated-polyimide resins and branched alkali sulfonated-polyimide resins. Alkali sulfonated polyester resins may be useful in embodiments, such as, the metal or alkali salts of copoly(ethylene-terephthalate)-copoly(ethylene-5-sulfo-isophthalate), copoly(propylene-terephthalate)-copoly(propylene-5-sulfo-isophthalate), copoly(diethylene-terephthalate)-copoly(diethylene-5-sulfo-isophthalate), copoly(propylene-diethylene-terephthalate)-copoly(propylene-diethylene-5-sulfoisophthalate), copoly(propylene-butyleneterephthalate)-copoly(propylene-butyleneterephthalate-5-sulfo-isophthalate), copoly(propoxylated bisphenol-A-fumarate)-

copoly(propoxylated bisphenol A-5-sulfo-isophthalate), copoly(ethoxylated bisphenol-A-fumarate)-copoly(ethoxylated bisphenol-A-5-sulfo-isophthalate), and copoly(ethoxylated bisphenol-A-maleate)-copoly(ethoxylated bisphenol-A-5-sulfo-isophthalate), wherein the alkali metal is, for example, a sodium, a lithium or a potassium ion.

[0034] An unsaturated amorphous polyester resin may be used as a latex resin. Examples of such resins include those disclosed in U.S. Patent No. 6,063,827. Exemplary unsaturated amorphous polyester resins include, but are not limited to, poly(propoxylated bisphenol co-fumarate), poly(ethoxylated bisphenol co-fumarate), poly(butyloxylated bisphenol co-fumarate), poly(co-propoxylated bisphenol co-ethoxylated bisphenol co-fumarate), poly(1,2-propylene fumarate), poly(propoxylated bisphenol co-maleate), poly(ethoxylated bisphenol co-maleate), poly(butyloxylated bisphenol co-maleate), poly(co-propoxylated bisphenol co-ethoxylated bisphenol co-maleate), poly(1,2-propylene maleate), poly(propoxylated bisphenol co-itaconate), poly(ethoxylated bisphenol co-itaconate), poly(butyloxylated bisphenol co-itaconate), poly(co-propoxylated bisphenol co-ethoxylated bisphenol co-itaconate), poly(1,2-propylene itaconate) and combinations thereof.

[0035] For forming a crystalline polyester resin, suitable organic polyols include aliphatic polyols with from about 2 to about 36 carbon atoms, such as 1,2-ethanediol, 1,3-propanediol, 1,4-butanediol, 1,5-pentanediol, 2,2-dimethylpropane-1,3-diol, 1,6-hexanediol, 1,7-heptanediol, 1,8-octanediol, 1,9-nonanediol, 1,10-decanediol, 1,12-dodecanediol and the like; alkali sulfo-aliphatic diols such as sodio 2-sulfo-1,2-ethanediol, lithio 2-sulfo-1,2-ethanediol, potassio 2-sulfo-1,2-ethanediol, sodio 2-sulfo-1,3-propanediol, lithio 2-sulfo-1,3-propanediol, potassio 2-sulfo-1,3-propanediol, mixture thereof, and the like, including their structural isomers. The aliphatic polyol may be, for example, selected in an amount from about 40 to about 60 mole%, from about 42 to about 55 mole%, from about 45 to about 53 mole%, and a second polyol may be used in an amount from about 0.1 to about 10 mole%, from about 1 to about 4 mole% of the resin.

[0036] Examples of polyacid or polyester reagents for preparing a crystalline resin include oxalic acid, succinic acid, glutaric acid, adipic acid, suberic acid, azelaic acid, sebacic acid, fumaric acid, dimethyl fumarate, dimethyl itaconate, cis, 1,4-diacetoxy-2-butene, diethyl

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fumarate, diethyl maleate, phthalic acid, isophthalic acid, terephthalic acid, naphthalene-2,6-dicarboxylic acid, naphthalene-2,7-dicarboxylic acid, cyclohexane dicarboxylic acid (sometimes referred to herein, in embodiments, as cyclohexanedioic acid), malonic acid and mesaconic acid, a polyester or anhydride thereof; and an alkali sulfo-organic polyacid, such as, the sodio, lithio or potassio salt of dimethyl-5-sulfo-isophthalate, dialkyl-5-sulfo-isophthalate-4-sulfo-1,8-naphthalic anhydride, 4-sulfo-phthalic acid, dimethyl-4-sulfo-phthalate, dialkyl-4-sulfo-phthalate, 4-sulfophenyl-3,5-dicarbomethoxybenzene, 6-sulfo-2-naphthyl-3,5-dicarbomethoxybenzene, sulfo-terephthalic acid, dimethyl-sulfo-terephthalate, 5-sulfo-isophthalic acid, dialkyl-sulfo-terephthalate, sulfo-p-hydroxybenzoic acid, N,N-bis(2-hydroxyethyl)-2-amino ethane sulfonate, or mixtures thereof. The polyacid may be selected in an amount of, for example, in embodiments from about 40 to about 60 mole%, from about 42 to about 52 mole%, from about 45 to about 50 mole%, and optionally, a second polyacid may be selected in an amount from about 0.1 to about 10 mole% of the resin.

[0037] Specific crystalline resins include poly(ethylene-adipate), poly(propylene-adipate), poly(butylene-adipate), poly(pentylene-adipate), poly(hexylene-adipate), poly(octylene-adipate), poly(ethylene-succinate), poly(propylene-succinate), poly(butylene-succinate), poly(pentylene-succinate), poly(hexylene-succinate), poly(octylene-succinate), poly(ethylene-sebacate), poly(propylene-sebacate), poly(butylene-sebacate), poly(pentylene-sebacate), poly(hexylene-sebacate), poly(octylene-sebacate), poly(decylene-sebacate), poly(decylene-decanoate), poly(ethylene-decanoate), poly(ethylene dodecanoate), poly(nonylene-sebacate), poly(nonylene-decanoate), copoly(ethylene-fumarate)-copoly(ethylene-sebacate), copoly(ethylene-fumarate)-copoly(ethylene-decanoate), copoly(ethylene-fumarate)-copoly(ethylene-dodecanoate), copoly(2,2-dimethylpropane-1,3-diol-decanoate)-copoly(ethylene-adipate), alkali copoly(5-sulfoisophthaloyl)-copoly(propylene-adipate), alkali copoly(5-sulfoisophthaloyl)-copoly(butylene-adipate), alkali copoly(5-sulfo-isophthaloyl)-copoly(pentylene-adipate), alkali copoly(5-sulfo-isophthaloyl)-copoly(hexylene-adipate), alkali copoly(5-sulfo-isophthaloyl)-copoly(octylene-adipate), alkali copoly(5-sulfo-isophthaloyl)-copoly(ethylene-adipate), alkali copoly(5-sulfo-isophthaloyl)-copoly(propylene-adipate), alkali copoly(5-sulfo-isophthaloyl)-copoly(butylene-adipate), alkali copoly(5-sulfo-isophthaloyl)-copoly(pentylene-adipate),

alkali copoly(5-sulfo-isophthaloyl)-copoly(hexylene-adipate), alkali copoly(5-sulfo-isophthaloyl)-copoly(octylene-adipate), alkali copoly(5-sulfoisophthaloyl)-copoly(ethylene-succinate), alkali copoly(5-sulfoisophthaloyl)-copoly(propylene-succinate), alkali copoly(5-sulfoisophthaloyl)-copoly(butylenes-succinate), alkali copoly(5-sulfoisophthaloyl)-copoly(pentylene-succinate), alkali copoly(5-sulfoisophthaloyl)-copoly(hexylene-succinate), alkali copoly(5-sulfoisophthaloyl)-copoly(octylene-succinate), alkali copoly(5-sulfo-isophthaloyl)-copoly(ethylene-sebacate), alkali copoly(5-sulfo-isophthaloyl)-copoly(propylene-sebacate), alkali copoly(5-sulfo-isophthaloyl)-copoly(butylene-sebacate), alkali copoly(5-sulfo-isophthaloyl)-copoly(pentylene-sebacate), alkali copoly(5-sulfo-isophthaloyl)-copoly(hexylene-sebacate), alkali copoly(5-sulfo-isophthaloyl)-copoly(octylene-sebacate), alkali copoly(5-sulfo-isophthaloyl)-copoly(ethylene-adipate), alkali copoly(5-sulfo-isophthaloyl)-copoly(propylene-adipate), alkali copoly(5-sulfo-isophthaloyl)-copoly(butylene-adipate), alkali copoly(5-sulfo-isophthaloyl)-copoly(pentylene-adipate), alkali copoly(5-sulfo-isophthaloyl)-copoly(hexylene-adipatenonylene-decanoate), poly(octylene-adipate), and so on, wherein alkali is a metal like sodium, lithium or potassium. Examples of polyamides include poly(ethylene-adipamide), poly(propylene-adipamide), poly(butylenes-adipamide), poly(pentylene-adipamide), poly(hexylene-adipamide), poly(octylene-adipamide), poly(ethylene-succinimide), and poly(propylene-sebecamide). Examples of polyimides include poly(ethylene-adipimide), poly(propylene-adipimide), poly(butylene-adipimide), poly(pentylene-adipimide), poly(hexylene-adipimide), poly(octylene-adipimide), poly(ethylene-succinimide), poly(propylene-succinimide), and poly(butylene-succinimide).

[0038] Suitable crystalline resins which may be utilized, optionally in combination with an amorphous resin as described above, include those disclosed in U.S. Pub. No. 2006/0222991.

[0039] In embodiments, a suitable crystalline resin may include a resin formed of ethylene glycol and a mixture of dodecanedioic acid and fumaric acid.

[0040] Any suitable surfactants may be used for the preparation of latex, pigment or wax dispersions according to the present disclosure. Depending on the emulsion system, any

desired nonionic or ionic surfactant such as anionic or cationic surfactant may be contemplated.

[0041] Examples of suitable anionic surfactants include, but are not limited to, sodium dodecylsulfate, sodium dodecylbenzene sulfonate, sodium dodecylnaphthalenesulfate, dialkyl benzenealkyl sulfates and sulfonates, abitic acid, NEOGEN R[®] and NEOGEN SC[®] available from Kao, Tayca Power[®], available from Tayca Corp., DOWFAX[®], available from Dow Chemical Co., and the like, as well as mixtures thereof. Anionic surfactants may be employed in any desired or effective amount, generally at least about 0.01 % by weight of total monomers used to prepare the latex polymer, at least about 0.1 % by weight of total monomers used to prepare the latex polymer, no more than about 10 % by weight of total monomers used to prepare the latex polymer, no more than about 5 % by weight of total monomers used to prepare the latex polymer, although the amount can be outside of those ranges.

[0042] Examples of suitable cationic surfactants include, but are not limited to, dialkyl benzenealkyl ammonium chloride, lauryl trimethyl ammonium chloride, alkylbenzyl methyl ammonium chloride, alkyl benzyl dimethyl ammonium bromide, benzalkonium chloride, cetyl pyridinium bromide, C₁₂, C₁₅, and C₁₇ trimethyl ammonium bromides, halide salts of quaternized polyoxyethylalkylamines, dodecylbenzyl triethyl ammonium chloride, MIRAPOL[®] and ALKAQUAT[®] (available from Alkaril Chemical Company), SANIZOL[®] (benzalkonium chloride, available from Kao Chemicals), and the like, as well as mixtures thereof.

[0043] Examples of suitable nonionic surfactants include, but are not limited to, polyvinyl alcohol, polyacrylic acid, methalose, methyl cellulose, ethyl cellulose, propyl cellulose, hydroxy ethyl cellulose, carboxy methyl cellulose, polyoxyethylene cetyl ether, polyoxyethylene lauryl ether, polyoxyethylene octyl ether, polyoxyethylene octylphenyl ether, polyoxyethylene oleyl ether, polyoxyethylene sorbitan monolaurate, polyoxyethylene stearyl ether, polyoxyethylene nonylphenyl ether, dialkylphenoxypoly(ethyleneoxy)ethanol (available from Rhone-Poulenc as IGEPAL CA-210[®], IGEPAL CA-520[®], IGEPAL CA-720[®], IGEPAL CO-890[®], IGEPAL CO-720[®], IGEPAL CO-290[®], IGEPAL CA-210[®], ANTAROX 890[®], and ANTAROX 897[®]), and the like, as well as mixtures thereof.

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[0044] Any suitable initiator or mixture of initiators may be selected in the latex process and the toner process according to the present disclosure. In typical embodiments, the initiator is selected from various known free radical polymerization initiators. The free radical initiator can be any free radical polymerization initiator capable of initiating a free radical polymerization process, and mixtures thereof, typically free radical initiators capable of providing free radical species upon heating to above about 30° C.

[0045] Although water-soluble free radical initiators that are traditionally used in emulsion polymerization reactions are typically selected, it is also within the scope of the present disclosure that other free radical initiators are employed. Examples of suitable free radical initiators include, but are not limited to, peroxides such as ammonium persulfate, hydrogen peroxide, acetyl peroxide, cumyl peroxide, tert-butyl peroxide, propionyl peroxide, benzoyl peroxide, chlorobenzoyl peroxide, dichlorobenzoyl peroxide, bromomethylbenzoyl peroxide, lauroyl peroxide, sodium persulfate, potassium persulfate, diisopropyl peroxy carbonate, tetralin hydroperoxide, 1-phenyl-2-methylpropyl-1-hydroperoxide, tert-butylhydroperoxide, triphenylacetate, tert-butyl performate, tert-butyl peracetate, tert-butyl perbenzoate, tert-butyl perphenylacetate, tert-butyl permethoxyacetate, and tert-butyl per-N-(3-tolyl)carbamate; azo compounds such as 2,2'-azobispropane, 2,2'-dichloro-2,2'-azobispropane, 1,1'-azo(methylethyl)diacetate, 2,2'-azobis(2-amidinopropane)hydrochloride, 2,2'-azobis(2-amidinopropane)-nitrate, 2,2'-azobisisobutane, 2,2'-azobisisobutylamide, 2,2'-azobisisobutyronitrile, methyl 2,2'-azobis-2-methylpropionate, 2,2'-dichloro-2,2'-azobisbutane, 2,2'-azobis-2-methylbutyronitrile, dimethyl 2,2'-azobisisobutyrate, 1,1'-azobis(sodium 1-methylbutyronitrile-3-sulfonate), 2-(4-methylphenylazo)-2-methylmalonodinitrile, 4,4'-azobis-4-cyanovaleric acid, 3,5-dihydroxymethylphenylazo-2-methylmalonodinitrile, 2-(4-bromophenylazo)-2-allylmalonodinitrile, 2,2'-azobis-2-methylvaleronitrile, dimethyl 4,4'-azobis-4-cyanovalerate, 2,2'-azobis-2,4-dimethylvaleronitrile, 1,1'-azobiscyclohexanenitrile, 2,2'-azobis-2-propylbutyronitrile, 1,1'-azobis-1-chlorophenylethane, 1,1'-azobis-1-cyclohexanecarbonitrile, 1,1'-azobis-1-cycloheptanenitrile, 1,1'-azobis-1-phenylethane, 1,1'-azobiscumene, ethyl 4-nitrophenylazobenzylcyanoacetate, phenylazodiphenylmethane, phenylazotriphenylmethane, 4-nitrophenylazotriphenylmethane, 1'-azobis-1,2-diphenylethane, poly(bisphenol A-4,4'-

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azobis-4-cyanopentano-ate), and poly(tetraethylene glycol-2,2'-azobisisobutyrate); and 1,4-bis(pentaethylene)-2-tetrazene, and 1,4-dimethoxycarbonyl-1,4-diphenyl-2-tetrazene; and the like; and the mixture thereof.

[0046] More typical free radical initiators include, but are not limited to, ammonium persulfate, hydrogen peroxide, acetyl peroxide, cumyl peroxide, tert-butyl peroxide, propionyl peroxide, benzoyl peroxide, chlorobenzoyl peroxide, dichlorobenzoyl peroxide, bromomethylbenzoyl peroxide, lauroyl peroxide, sodium persulfate, potassium persulfate, diisopropyl peroxy carbonate and the like.

[0047] Based on total weight of the monomers to be polymerized, the initiator may generally be present in an amount from about 0.1% to about 5%, from about 0.4% to about 4%, from about 0.5% to about 3%, although it may be present in greater or lesser amounts.

[0048] A chain transfer agent may optionally be used to control the polymerization degree of the latex, and thereby control the molecular weight and molecular weight distribution of the product latexes of the latex process and/or the toner process according to the present disclosure. As a skilled artisan can appreciate, typically, the chain transfer agent becomes part of the latex polymer.

[0049] In embodiments, the chain transfer agent has a carbon-sulfur covalent bond. The carbon-sulfur covalent bond has usually absorption peak in a wave number region ranging from 500 to 800 cm^{-1} in an infrared absorption spectrum. When the chain transfer agent is incorporated into the latex and the toner made from the latex, the absorption peak may be changed, for example, to a wave number region of 400 to 4,000 cm^{-1} .

[0050] Exemplary chain transfer agents include, but are not limited to, n-C₃₋₁₅ alkylmercaptans such as n-propylmercaptan, n-butylmercaptan, n-amylmercaptan, n-hexylmercaptan, n-heptylmercaptan, n-octylmercaptan, n-nonylmercaptan, n-decylmercaptan, and n-dodecylmercaptan; branched alkylmercaptans such as isopropylmercaptan, isobutylmercaptan, s-butylmercaptan, tert-butylmercaptan, cyclohexylmercaptan, tert-hexadecylmercaptan, tert-laurylmercaptan, tert-nonylmercaptan, tert-octylmercaptan, and tert-tetradecylmercaptan; aromatic ring-containing mercaptans such as allylmercaptan, 3-phenylpropylmercaptan, phenylmercaptan, and

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mercaptotriphenylmethane. As a skilled artisan understands, the term, “mercaptan,” and, “thiol,” may be used interchangeably to mean C-SH group.

[0051] Typical examples of such chain transfer agents also include, but are not limited to, dodecanethiol, butanethiol, isooctyl-3-mercaptopropionate, 2-methyl-5-t-butyl-thiophenol, carbon tetrachloride, carbon tetrabromide and the like.

[0052] Based on total weight of the monomers to be polymerized, the chain transfer agent may generally be present in an amount from about 0.1% to about 7%, from about 0.5% to about 6%, from about 1.0% to about 5%, although it may be present in greater or lesser amounts.

[0053] In various embodiments, a branching agent may optionally be included in the first/second monomer composition to control the branching structure of the target latex. Exemplary branching agents include, but are not limited to, decanediol diacrylate (ADOD), trimethylolpropane, pentaerythritol, trimellitic acid, pyromellitic acid, and mixtures thereof. In a specific embodiment, the branching agent is ADOD, which may be commercially available from Shin-Najamura Co., Japan.

[0054] Based on total weight of the monomers to be polymerized, the branching agent may generally be present in an amount from about 0% to about 2%, from about 0.05% to about 1.0%, from about 0.1% to about 0.8%, although it may be present in greater or lesser amounts.

[0055] In the latex process and toner process of the disclosure, emulsification may be done by any suitable process such as mixing at elevated temperature. For example, the emulsion mixture may be mixed in a homogenizer set at about 200 to about 400 rpm and at a temperature of from about 40° C to about 80° C for a period of from about 1 minute to about 20 minutes.

[0056] Any type of reactor may be suitably used without restriction. The reactor should include means for stirring the compositions therein. Typically, the reactor includes at least one impeller. For forming the latex and/or toner, the reactor is operated throughout the process where impellers operate at a rate of about 10 to about 1,000 rpm.

[0057] Following completion of the monomer addition, the latex may be permitted to stabilize by maintaining the conditions for a period of time, for example for about 10 to about

300 minutes, before cooling. Optionally, the latex formed by the above process may be isolated by standard methods known in the art, for example, coagulation, dissolution and precipitation, filtration, washing, drying, or the like.

[0058] The latex of the present disclosure may be selected for emulsion/aggregation/coalescence processes for forming toners, inks and developers by known methods.

[0059] The latex of the present disclosure may be melt blended or otherwise mixed with various toner ingredients such as a wax dispersion, a coagulant, an optional silica, an optional charge enhancing additive or charge control additive, an optional surfactant, an optional emulsifier, an optional flow additive, and the like. Optionally, the latex (e.g. around 40 % solids) may be diluted to the desired solids loading (e.g. around 12 to 15 % by weight solids), before formulated into the toner composition.

[0060] Based on the total toner weight, the latex may generally be present in an amount from about 50% to about 99%, from about 60% to about 98%, from about 70% to about 95%, although it may be present in greater or lesser amounts. Methods of producing such latex resins may be carried out as described in the disclosure of U.S. Pat. No. 7,524,602.

Colorants

[0061] Various known suitable colorants, such as dyes, pigments, mixtures of dyes, mixtures of pigments, mixtures of dyes and pigments, and the like, may be included in the toner. The colorant may be included in the toner in an amount of, for example, 0 to about 35 % by weight of the toner, from about 1 to about 15 weight % of the toner, from about 3 to about 10 % by weight of the toner, although amounts outside those ranges may be utilized.

[0062] As examples of suitable colorants, mention may be made of carbon black like REGAL 330[®]; magnetites, such as Mobay magnetites MO8029[™], MO8060[™]; Columbian magnetites; MAPICO BLACKS[™] and surface treated magnetites; Pfizer magnetites CB4799[™], CB5300[™], CB5600[™], MCX6369[™]; Bayer magnetites, BAYFERROX 8600[™], 8610[™]; Northern Pigments magnetites, NP-604[™], NP-608[™]; Magnox magnetites TMB-100[™], or TMB-104[™]; and the like. As colored pigments, there can be selected cyan, magenta, yellow, red, green, brown, blue or mixtures thereof. Generally, cyan, magenta, or

yellow pigments or dyes, or mixtures thereof, are used. The pigment or pigments are generally used as water based pigment dispersions.

[0063] Specific examples of pigments include SUNSPERSE 6000, FLEXIVERSE and AQUATONE water based pigment dispersions from SUN Chemicals, HELIOGEN BLUE L6900™, D6840™, D7080™, D7020™, PYLAM OIL BLUE™, PYLAM OIL YELLOW™, PIGMENT BLUE 1™ available from Paul Uhlich & Co., Inc., PIGMENT VIOLET 1™, PIGMENT RED 48™, LEMON CHROME YELLOW DCC 1026™, E.D. TOLUIDINE RED™ and BON RED C™ available from Dominion Color Corp., Ltd., Toronto, CA, NOVAPERM YELLOW FGL™, HOSTAPERM PINK E™ from sanofi, and CINQUASIA MAGENTA™ available from E.I. DuPont de Nemours & Co., and the like. Generally, colorants that can be selected are black, cyan, magenta, or yellow, and mixtures thereof. Examples of magentas are 2,9-dimethyl-substituted quinacridone and anthraquinone dye identified in the Color Index (CI) as CI 60710, CI Dispersed Red 15, diazo dye identified in the Color Index as CI 26050, CI Solvent Red 19, and the like. Illustrative examples of cyans include copper tetra(octadecyl sulfonamido) phthalocyanine, x-copper phthalocyanine pigment listed in the Color Index as CI 74160, CI Pigment Blue, Pigment Blue 15:3, and Anthrathrene Blue, identified in the Color Index as CI 69810, Special Blue X-2137, and the like. Illustrative examples of yellows are diarylide yellow 3,3-dichlorobenzidine acetoacetanilides, a monoazo pigment identified in the Color Index as CI 12700, CI Solvent Yellow 16, a nitrophenyl amine sulfonamide identified in the Color Index as Foron Yellow SE/GLN, CI Dispersed Yellow 33 2,5-dimethoxy-4-sulfonanilide phenylazo-4'-chloro-2,5-dimethoxy acetoacetanilide, and Permanent Yellow FGL. Colored magnetites, such as mixtures of MAPICO BLACK™, and cyan components may also be selected as colorants. Other known colorants can be selected, such as Levanyl Black A-SF (Miles, Bayer) and Sunspere Carbon Black LHD 9303 (Sun Chemicals), and colored dyes such as Neopen Blue (BASF), Sudan Blue OS (BASF), PV Fast Blue B2G01 (sanofi), Sunspere Blue BHD 6000 (Sun Chemicals), Irgalite Blue BCA (Ciba-Geigy), Paliogen Blue 6470 (BASF), Sudan III (Matheson, Coleman, Bell), Sudan II (Matheson, Coleman, Bell), Sudan IV (Matheson, Coleman, Bell), Sudan Orange G (Aldrich), Sudan Orange 220 (BASF), Paliogen Orange 3040 (BASF), Ortho Orange OR 2673 (Paul Uhlich), Paliogen Yellow 152, 1560 (BASF),

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Lithol Fast Yellow 0991K (BASF), Paliotol Yellow 1840 (BASF), Neopen Yellow (BASF), Novoperm Yellow FG 1 (sanofi), Permanent Yellow YE 0305 (Paul Uhlich), Lumogen Yellow D0790 (BASF), Sunsperse Yellow YHD 6001 (Sun Chemicals), Suco-Gelb L1250 (BASF), Suco-Yellow D1355 (BASF), Hostaperm Pink E (sanofi), Fanal Pink D4830 (BASF), Cinquasia Magenta (DuPont), Lithol Scarlet D3700 (BASF), Toluidine Red (Aldrich), Scarlet for Thermoplast NSD PS PA (Ugine Kuhlmann of Canada), E.D. Toluidine Red (Aldrich), Lithol Rubine Toner (Paul Uhlich), Lithol Scarlet 4440 (BASF), Bon Red C (Dominion Color Company), Royal Brilliant Red RD-8192 (Paul Uhlich), Oracet Pink RF (Ciba-Geigy), Paliogen Red 3871K (BASF), Paliogen Red 3340 (BASF), Lithol Fast Scarlet L4300 (BASF), combinations of the foregoing, and the like.

Wax

[0064] In addition to the polymer resin, the toners of the present disclosure may contain a wax, which can be either a single type of wax or a mixture of two or more different waxes. A single wax can be added to toner formulations, for example, to improve particular toner properties, such as toner particle shape, presence and amount of wax on the toner particle surface, charging and/or fusing characteristics, gloss, stripping, offset properties, and the like. Alternatively, a combination of waxes can be added to provide multiple properties to the toner composition.

[0065] When included, the wax may be present in an amount of, for example, from about 1 wt% to about 25 wt% of the toner particles, from about 5 wt% to about 20 wt% of the toner particles. A wax can have a melting point of at least about 30°C., at least about 40°C., at least about 50°C.

[0066] Waxes that may be selected include waxes having, for example, a weight average molecular weight of from about 500 to about 20,000, from about 1,000 to about 10,000. Waxes that may be used include, for example, polyolefins such as polyethylene, polypropylene, and polybutene waxes such as commercially available from Allied Chemical and Petrolite Corporation, for example POLYWAXTM polyethylene waxes from Baker Petrolite, wax emulsions available from Michaelman, Inc. and the Daniels Products Company, EPOLENE N-15TM commercially available from Eastman Chemical Products, Inc., and VISCOL 550-PTM, a low weight average molecular weight polypropylene available

from Sanyo Kasei K. K.; plant-based waxes, such as carnauba wax, rice wax, candelilla wax, sumacs wax, and jojoba oil; animal-based waxes, such as beeswax; mineral-based waxes and petroleum-based waxes, such as montan wax, ozokerite, ceresin, paraffin wax, microcrystalline wax, and Fischer-Tropsch wax; ester waxes obtained from higher fatty acid and higher alcohol, such as stearyl stearate and behenyl behenate; ester waxes obtained from higher fatty acid and monovalent or multivalent lower alcohol, such as butyl stearate, propyl oleate, glyceride monostearate, glyceride distearate, and pentaerythritol tetra behenate; ester waxes obtained from higher fatty acid and multivalent alcohol multimers, such as diethyleneglycol monostearate, dipropyleneglycol distearate, diglyceryl distearate, and triglyceryl tetrastearate; sorbitan higher fatty acid ester waxes, such as sorbitan monostearate, and cholesterol higher fatty acid ester waxes, such as cholesteryl stearate. Examples of functionalized waxes that may be used include, for example, amines, amides, for example AQUA SUPERSLIP 6550TM, SUPERSLIP 6530TM available from Micro Powder Inc., fluorinated waxes, for example POLYFLUO 190TM, POLYFLUO 200TM, POLYSILK 19TM, POLYSILK 14TM available from Micro Powder Inc., mixed fluorinated, amide waxes, for example MICROSPERSION 19TM also available from Micro Powder Inc., imides, esters, quaternary amines, carboxylic acids or acrylic polymer emulsion, for example JONCRYL 74TM, 89TM, 130TM, 537TM and 538TM, all available from SC Johnson Wax, and chlorinated polypropylenes and polyethylenes available from Allied Chemical and Petrolite Corporation and SC Johnson wax. Mixtures and combinations of the foregoing waxes may also be used in embodiments. Waxes may be included as, for example, fuser roll release agents.

Toner Preparation

[0067] The toner particles may be prepared by any method within the purview of one skilled in the art. Although embodiments relating to toner particle production are described below with respect to emulsion-aggregation processes, any suitable method of preparing toner particles may be used, including chemical processes, such as suspension and encapsulation processes disclosed in U.S. Pat. Nos. 5,290,654 and 5,302,486. In embodiments, toner compositions and toner particles may be prepared by aggregation and coalescence processes

in which small-size resin particles are aggregated to the appropriate toner particle size and then coalesced to achieve the final toner particle shape and morphology.

[0068] In embodiments, toner compositions may be prepared by emulsion-aggregation processes, such as a process that includes aggregating a mixture of an optional wax and any other desired or required additives, and emulsions including the resins described above, optionally with surfactants as described above, and then coalescing the aggregate mixture. A mixture may be prepared by adding an optional wax or other materials, which may also be optionally in a dispersion(s) including a surfactant, to the emulsion, which may be a mixture of two or more emulsions containing the resin. The pH of the resulting mixture may be adjusted by an acid (i.e., a pH adjustor) such as, for example, acetic acid, nitric acid or the like. In embodiments, the pH of the mixture may be adjusted to from about 2 to about 4.5. Additionally, in embodiments, the mixture may be homogenized. If the mixture is homogenized, homogenization may be accomplished by mixing at about 600 to about 4,000 rpm. Homogenization may be accomplished by any suitable means, including, for example, an IKA ULTRA TURRAX T50 probe homogenizer.

[0069] Following the preparation of the above mixture, an aggregating agent may be added to the mixture. Any suitable aggregating agent may be utilized to form a toner. Suitable aggregating agents include, for example, aqueous solutions of a divalent cation or a multivalent cation material. The aggregating agent may be, for example, polyaluminum halides such as polyaluminum chloride (PAC), or the corresponding bromide, fluoride, or iodide, polyaluminum silicates such as polyaluminum sulfosilicate (PASS), and water soluble metal salts including aluminum chloride, aluminum nitrite, aluminum sulfate, potassium aluminum sulfate, calcium acetate, calcium chloride, calcium nitrite, calcium oxylate, calcium sulfate, magnesium acetate, magnesium nitrate, magnesium sulfate, zinc acetate, zinc nitrate, zinc sulfate, zinc chloride, zinc bromide, magnesium bromide, copper chloride, copper sulfate, and combinations thereof. In embodiments, the aggregating agent may be added to the mixture at a temperature that is below the glass transition temperature (T_g) of the resin.

[0070] The aggregating agent may be added to the mixture utilized to form a toner in an amount of, for example, from about 0.1 parts per hundred (pph) to about 1 pph, from about 0.25 pph to about 0.75 pph.

[0071] The gloss of a toner may be influenced by the amount of retained metal ion, such as Al^{3+} , in the particle. The amount of retained metal ion may be adjusted by the addition of a chelator, such as, EDTA. In embodiments, the amount of retained metal ion, for example Al^{3+} , in toner particles of the present disclosure may be from about 0.1 pph to about 1 pph, from about 0.25 pph to about 0.8 pph.

[0072] The present disclosure also provides a melt mixing process to produce low cost and safe cross-linked thermoplastic binder resins for toner compositions which have low fix temperatures and high offset temperatures, and which show minimized or substantially no vinyl offset. In this process, unsaturated base polyester resins or polymers are melt blended, that is, in the molten state under high shear conditions producing substantially uniformly dispersed toner constituents, and which process provides a resin blend and toner product with optimized gloss properties (see, e.g., U.S. Pat. No. 5,556,732).

[0073] To control aggregation and coalescence of the particles, in embodiments, the aggregating agent may be metered into the mixture over time. For example, the agent may be metered into the mixture over a period of from about 5 to about 240 minutes, from about 30 to about 200 minutes. The addition of the agent may also be done while the mixture is maintained under stirred conditions, in embodiments from about 50 rpm to about 1,000 rpm, from about 100 rpm to about 500 rpm, and at a temperature that is below the T_g of the resin as discussed above.

[0074] The particles may be permitted to aggregate until a predetermined desired particle size is obtained. A predetermined desired size refers to the desired particle size to be obtained as determined prior to formation, and the particle size being monitored during the growth process until such particle size is reached. Samples may be taken during the growth process and analyzed, for example with a COULTER COUNTER, for average particle size. The aggregation thus may proceed by maintaining the elevated temperature, or slowly raising the temperature to, for example, from about 40° C to about 100° C, and holding the mixture at

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this temperature for a time from about 0.5 hours to about 6 hours, from about hour 1 to about 5 hours, while maintaining stirring, to provide the aggregated particles. Once the predetermined desired particle size is reached, then the growth process is halted. The particle size may be from about 3 to 8 μm , from about 4 to about 7 μm , from about 5 to about 6 μm .

Shell Resin

[0075] In embodiments, a shell may be applied to the formed aggregated toner particles. Any resin described above as suitable for the core resin may be utilized as the shell resin. The shell resin may be applied to the aggregated particles by any method within the purview of those skilled in the art. In embodiments, the shell resin may be in an emulsion including any surfactant described above. The aggregated particles described above may be combined with said emulsion so that the resin forms a shell over the formed aggregates. In embodiments, an amorphous polyester may be utilized to form a shell over the aggregates to form toner particles having a core-shell configuration.

[0076] The shell component may be from about 5 to about 40% by weight of the toner particles, from about 10% to about 30% by weight.

[0077] Alternatively, a thicker shell may be desirable to provide improved charging characteristics due to the higher surface area of the toner particle. Thus, the shell resin may be present in an amount of at least about 30% by weight of the toner, at least about 35% by weight of the toner, at least about 40% by weight of the toner.

[0078] In embodiments, a photoinitiator may be included in the resin solution for forming the shell. Thus, the photoinitiator may be in the core, the shell or both. The photoinitiator may be present in an amount of from about 1% to about 5% by weight of the toner particles, from about 2% to about 4% by weight of the toner particles.

[0079] Emulsions including these resins may have a solids loading of from about 5% solids by weight to about 20% solids by weight, from about 12% solids by weight to about 17% solids by weight.

[0080] Once the desired final size of the toner particles is achieved, the pH of the mixture may be adjusted with a base to a value of from about 6 to about 10, from about 6.2 to about 8. The adjustment of the pH may be utilized to freeze, that is, to stop, toner growth. The base utilized to stop toner growth may include any suitable base such as, for example, alkali

metal hydroxides such as, for example, sodium hydroxide, potassium hydroxide, ammonium hydroxide, combinations thereof, and the like. In embodiments, a chelator, such as, ethylene diamine tetraacetic acid (EDTA) may be added to help adjust the pH to the desired values noted above. The base may be added in amounts from about 2 to about 25% by weight of the mixture, from about 4 to about 10% by weight of the mixture. In embodiments, the shell has a higher T_g than the aggregated toner particles.

Coalescence

[0081] Following aggregation to the desired particle size, with the formation of an optional shell as described above, the particles may then be coalesced to the desired final shape, the coalescence being achieved by, for example, heating the mixture to a temperature of from about 55° C to about 100° C, from about 65° C to about 75° C, which may be below the melting point of a crystalline resin to prevent plasticization. Higher or lower temperatures may be used, it being understood that the temperature is a function of the resins used for the binder.

[0082] Coalescence may proceed and be accomplished over a period of from about 0.1 to about 9 hours, from about 0.5 to about 4 hours.

[0083] After coalescence, the mixture may be cooled to room temperature (RT), such as from about 20° C to about 25° C. The cooling may be rapid or slow, as desired. A suitable cooling method may include introducing cold water to a jacket around the reactor. After cooling, the toner particles may be optionally washed with water, and then dried. Drying may be accomplished by any suitable method for drying including, for example, freeze drying.

Additives

[0084] In embodiments, the toner particles may also contain other optional additives, as desired or required. For example, the toner may include any known charge additives in amounts of from about 0.1 to about 10 wt%, from about 0.5 to about 7 wt% of the toner. Examples of such charge additives include alkyl pyridinium halides, bisulfates, the charge control additives of U.S. Pat. Nos. 3,944,493, 4,007,293, 4,079,014, 4,394,430 and 4,560,635, negative charge enhancing additives, like aluminum complexes, and the like.

[0085] Surface additives can be added to the toner compositions of the present disclosure after washing or drying. Examples of such surface additives include, for example, metal salts, metal salts of fatty acids, colloidal silicas, metal oxides, strontium titanates, mixtures thereof and the like. Hence, surface additives can include a silica, a titania, an alumina and so on. Surface additives may be present in an amount of from about 0.1 to about 10 wt%, from about 0.5 to about 7 wt% of the toner. Examples of such additives include those disclosed in U.S. Pat. Nos. 3,590,000, 3,720,617, 3,655,374 and 3,983,045. Other additives include zinc stearate and AEROSIL R972[®] available from Degussa. The coated silicas of U.S. Pat. Nos. 6,190,815 and 6,004,714 can also be present in an amount of from about 0.05 to about 5%, from about 0.1 to about 2% of the toner, which additives can be added during the aggregation or blended into the formed toner product.

[0086] The characteristics of the toner particles may be determined by any suitable technique and apparatus. Volume average particle diameter D_{50v} , GSD_v , and GSD_n may be measured by means of a measuring instrument such as a Beckman Coulter MULTISIZER 3, operated in accordance with manufacturer instructions. Developers produced in accordance with the present disclosure may possess excellent charging characteristics when exposed to extreme relative humidity (RH) conditions.

[0087] A toner of the present disclosure may have a gloss as measured by, for example, a Gardner device of from about 20 gloss units (gu) to about 100 gu, from about 50 gu to about 95 gu, from about 60 gu to about 90 gu.

[0088] In embodiments, toners of the present disclosure may be utilized as ultra low melt (ULM) toners. In embodiments, the dry toner particles, exclusive of external surface additives, may have the following characteristics:

(1) circularity of from about 0.9 to about 1 (measured with a Sysmex 3000 analyzer), from about 0.95 to about 0.99, from about 0.96 to about 0.98;

(2) T_g of from about 45° C to about 60° C, from about 48° C to about 55° C.; and

(3) melt flow index (MFI) g/10 min (5 kg/130° C) of from about 79.0 to about 172.5.

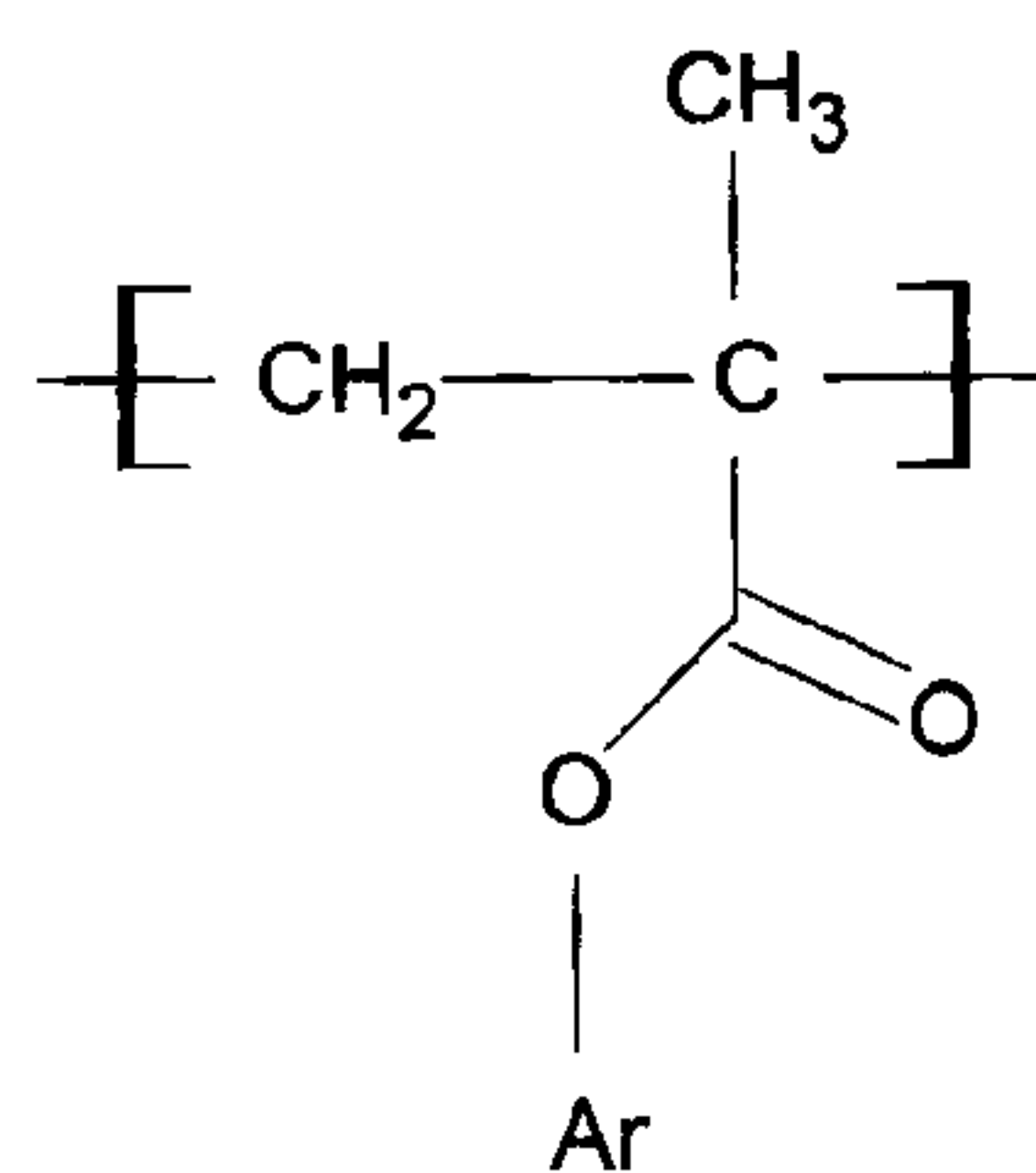
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[0089] In embodiments, a carrier composition is disclosed comprising a polymer coating resin comprised of at least one methacrylate monomer comprising a hydrophobic substituent, where optionally, the monomer can have a carbon:oxygen (C/O) ratio of at least about 4.75, at least about 4.8, at least about 4.9, at least about 5, at least about 5.1 or more.

[0090] The carrier resin may be a methacrylate monomer, having a T_g of between about 80° C to 140° C and may further include a toner surface additive. In embodiments, the toner surface additive is a silica. Such resins as described possess an energy gap for forward charge transfer of an electron from the monomer to the toner additive that is lower than the energy gap for reverse charge transfer of an electron from the toner surface additive to the monomer.

[0091] The at least one methacrylate monomer can have the formula as set forth in formula (I):



Formula (I),

wherein Ar comprises at least one aromatic group.

[0092] The aryl group can be, for example, a benzyl group, a naphthyl group, a phenyl group and so on.

[0093] In embodiments, the gap for forward charge transfer of electrons for said monomer is less than about 5.1 eV, less than about 5 eV, less than about 4.9 eV or less, including that the gap for reverse charge transfer of electrons from said toner additive is greater than the forward gap.

[0094] In embodiments, the resin has the HOMO electron density substantially located on a hydrophobic group, and whereby the HOMO electron density located on the carbonyl function or group of the methacrylate is effectively reduced. In embodiments, a monomer

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comprising the resin comprises, but is not limited to, hydrophobic groups, such as, e.g., phenyl and benzyl substituents.

[0095] In embodiments, the coating resin may further include a secondary amino acrylate monomer. Such secondary amino acrylate monomers include, but are not limited to, dimethylaminoethyl methacrylate (DMAEMA), diethylaminoethyl methacrylate (DEAEMA) and diethylaminopropyl methacrylate (DEAPMA). The secondary amino acrylate monomers may be present at from about 0.5 to about 1%, from about 1 to about 1.5%, from about 1.5 to about 2% by weight of the resin coating.

[0096] In embodiments, the carrier resin can comprise more than one species of resin. Hence, when two resins are present, those may be mixed in proportions of from about 30 to about 70 wt% to about 70 to about 30 wt%, from about 40 to about 60 wt% to about 60 to about 40 wt%. The coating may have a coating weight of, for example, from about 0.1 to about 5% by weight of the carrier, from about 0.5 to about 2% by weight of the carrier.

[0097] In embodiments, the carrier resin polymer optionally may be copolymerized with any desired comonomer, so long as the resulting copolymer retains suitable charge transfer and RH sensitivity properties. Suitable comonomers can include monoalkyl or dialkyl amines, such as a DMAEMA, cyclohexyl methacrylate (CHMA), diethylaminoethyl methacrylate, diisopropylaminoethyl methacrylate, t-butylaminoethyl methacrylate, and the like.

[0098] The carrier particles may be prepared by mixing the carrier core with polymer in an amount from about 0.05 to about 10 % by weight, from about 0.01 % to about 3 % by weight, based on the weight of the coated carrier particles, until adherence thereof to the carrier core by mechanical impaction and/or electrostatic attraction. In embodiments, the T_g of the resin monomers is from about 53° C to about 80° C, from about 80° C to about 100° C, from about 100° C to about 140° C. In embodiments, the T_g is from about 80° C to about 140° C.

[0099] Various effective suitable means can be used to apply the polymer to the surface of the carrier core particles, for example, cascade roll mixing, tumbling, milling, shaking, electrostatic powder cloud spraying, fluidized bed, electrostatic disc processing, electrostatic curtain, combinations thereof, and the like. The mixture of carrier core particles and polymer

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may then be heated to enable the polymer to melt and to fuse to the carrier core particles using a known method and device, such as, a kiln or an extruder. The coated carrier particles may then be cooled and thereafter classified to a desired particle size.

[00100] The carrier particles can be mixed with the toner particles in various suitable combinations. The concentrations may be from about 1% to about 20% by weight of the toner composition. However, different toner and carrier percentages may be used to achieve a developer composition with desired characteristics.

Developers

[00101] The toner particles are mixed with carrier particles to achieve a developer composition. The toner concentration in the developer may be from about 1% to about 25% by weight of the total weight of the developer, from about 2% to about 15% by weight of the total weight of the developer.

[00102] The developers of interest, comprising a carrier resin of interest, comprise good charging under a range of environmental conditions. While it is desirable for the RH ratio to be as close as possible to 1, in general a desirable RH ratio is least about 0.5, at least about 0.6, at least about 0.7 and so on. It is desirable that image density in printing does not change too much, a lower Q/M charge can tend to increase image density and a higher Q/M charge can lower image density. Also, the risk of background on the print increases as the Q/D charge drops, so a high RH ratio may result in increased background. The printer may also have software and hardware control systems to compensate for the change in charge level, however, a high RH ratio can result in insufficient control latitude over different environments, resulting in image density changes, background or both.

Imaging

[00103] The toners can be utilized for electrophotographic processes, including those disclosed in U.S. Pat. No. 4,295,990, the disclosure of which is hereby incorporated by reference in entirety. In embodiments, any known type of image development system may be used in an image developing device, including, for example, magnetic brush development, jumping single-component development, hybrid scavengeless development (HSD), and the like. These and similar development systems are within the purview of those skilled in the art.

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[00104] It is envisioned that the toners of the present disclosure may be used in any suitable procedure for forming an image with a toner, including in applications other than xerographic applications.

[00105] Utilizing the toners of the present disclosure, images may be formed on substrates, including flexible substrates, having a toner pile height of from about 1 μm to about 6 μm , from about 2 μm to about 4.5 μm , from about 2.5 to about 4.2 μm .

[00106] In embodiments, the toner of the present disclosure may be used for a xerographic print protective composition that provides overprint coating properties including, but not limited to, thermal and light stability and smear resistance, particularly in commercial print applications. More specifically, such overprint coating as envisioned has the ability to permit overwriting, reduce or prevent thermal cracking, improve fusing, reduce or prevent document offset, improve print performance, and protect an image from sun, heat, and the like. In other embodiments, the overprint compositions may be used to improve the overall appearance of xerographic prints due to the ability of the compositions to fill in the roughness of xerographic substrates and toners, thereby forming a level film and enhancing gloss.

Rational Design

[00107] In embodiments, a controller/central processing unit (CPU) may be used to design carrier resins and/or to predict key attributes for resins.

[00108] In embodiments, a carrier resin is selected by identifying a test polymer and modeling the surface of the polymer; identifying a test toner additive and modeling the surface of the toner additive; determining surface electron properties of the polymer and the toner additive using a density functional method, where the method determines structure calculations for local and gradient-dependent function; determining the initial structure, optimized structure and electronic properties of adsorbed test polymer complexes at the toner additive; determining geometry optimization convergence for the adsorbed polymer complexes on the toner additive, where the optimization is achieved when the energy, gradient and displacement are lower than about 2×10^{-5} Ha, about 4×10^{-3} Ha/Å, and about 5×10^{-3} Å, respectively; determining the likely direction of charge transfer between the polymer and additive by calculating the HOMO and LUMO for the polymer-toner additive complexes; determining Fukui functions for surface electron densities in HOMO and LUMO

to describe the active sites of the donor-acceptor complexes in charge transfer; and determining the lowest energy gap for both forward electron transfer and reverse electron transfer, where when the reverse gap is higher than the forward gap for the adsorbed polymer complexes on the toner additive, the negative gap difference is predictive of high negative toner charge in charging of toner resins comprising the polymer-toner additive complexes.

[00109] Those of skill in the art will recognize that many of the functions and aspects of such a method may be implemented on a computer or computers. The hardware of such computer platforms typically is general purpose in nature, albeit with an appropriate network connection for communication via the intranet, the internet and/or other data networks.

[00110] As known in the data processing and communications arts, each such general purpose computer typically comprises a central processor, an internal communication bus, various types of memory (RAM, ROM, EEPROM, cache memory etc.), disk drives or other code and data storage systems, and one or more network interface cards or ports for communication purposes. The computer system also may be coupled to a display and one or more user input devices, such as, alphanumeric and other keys of a keyboard, a mouse, a trackball and the like. A service-related user interface, for interactive control, enables operation of the computer system. The user interface elements may be locally coupled to the computer system, for example, in a workstation configuration, or the user interface elements may be remote from the computer and communicate therewith via a network. The elements of such a general purpose computer system also may be combined with or built into routing elements or nodes of the network.

[00111] The software functionalities (e.g., many of the operations described above) involve programming of controllers, including executable code as well as associated stored data. The software code is executable by the general purpose computer that functions as the particular computer. In operation, the executable program code and possibly the associated data are stored within the general purpose computer platform. At other times, however, the software may be stored at other locations and/or transported for loading into the appropriate general purpose computer system. Hence, the embodiments involve one or more software products in the form of one or more modules of code carried by at least one machine-readable medium. Execution of such code by a processor of the computer platform enables

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the platform to implement the system or platform functions, in essentially the manner performed in the embodiments discussed and illustrated herein.

[00112] As used herein, terms, such as, controller or CPU or computer or machine readable medium refer to any medium that participates in providing instructions to a processor for execution. Such a medium may take many forms, including but not limited to, non-volatile media, volatile media and transmission media. Non-volatile media include, for example, optical or magnetic disks, such as any of the storage devices in any computer(s). Volatile media include dynamic memory, such as main memory of such a computer platform. Physical transmission media include coaxial cables; copper wire and fiber optics, including the wires that comprise a bus within a computer system. Carrier wave transmission media may take the form of electric or electromagnetic signals, or acoustic or light waves such as those generated during radio frequency (RF) and infrared (IR) data communications. Common forms of computer-readable media therefore include, for example: a floppy disk, a flexible disk, hard disk, magnetic tape, any other magnetic medium, a CD-ROM, DVD, any other optical medium, punch cards, paper tape, any other physical medium with patterns of holes, a RAM, a PROM, and EPROM, a FLASH-EPROM, any other memory chip or cartridge, a carrier wave transporting data or instructions, cables or links transporting such a carrier wave, or any other medium from which a computer may read programming code and/or data. Many of those forms of computer readable media may be involved in carrying one or more sequences of one or more instructions to a processor for execution.

[00113] The following Examples are submitted to illustrate embodiments of the present disclosure. The Examples are intended to be illustrative only and are not intended to limit the scope of the present disclosure. Also, parts and percentages are by weight unless otherwise indicated.

EXAMPLES

Silica Charge with Carrier Resins

[00114] There are problems in providing high charge with good RH sensitivity of charge to changing environmental conditions for carrier coating resins. For example, polymer resins, such as, poly(cyclohexyl methacrylate) (PCHMA) could be used as carrier coating

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resins to provide similar charge as such resins including poly(methyl methacrylate) (PMMA) coated resin, but with improved RH sensitivity. There remains a need to tune the charge of the carrier resin to produce higher charge for those situations where higher charge is required.

[00115] In general, it seems that a higher carbon-to-oxygen (C/O) ratio of the monomer in the carrier resin improves RH sensitivity, while still providing good charge. For example, increasing the amount of CHMA in copolymers of CHMA and PMMA, where the DMAEMA is kept constant at 1% in the copolymer, demonstrates such behavior. Thus, it was desirable to increase the carbon content of a resin, for example, by replacing a methyl group in PMMA with a cyclohexyl group in CHMA.

[00116] Thus, the data demonstrate that resins, such as, PMMA, PCHMA and copolymers containing DMAEMA provide acceptable charge levels with silica, at least under dry conditions. Under high RH conditions, a higher C/O ratio is desired, as was demonstrated for methacrylates that contain cyclic hydrocarbon groups, such as, cyclohexyl in PCHMA.

[00117] Based on modeling studies, for high negative toner charge in charging of toners with silica and carriers with a polymeric resin coating the following attributes are desirable:

- 1) the gap for forward charge transfer from the HOMO in the carrier resin to the LUMO in the toner silica additive should be low;
- 2) the reverse gap should be higher than the forward gap (a negative gap difference; subtracting 1) from 2));
- 3) the resin is comprised of a high C/O ratio monomer, which limits bulk water adsorption; and
- 4) the resin has the HOMO electron density located substantially on the hydrophobic group.

[00118] The first two conditions provide the requisite charge transfer and charge level, while the last two conditions improve the RH sensitivity of charge to water adsorption. To provide functional specification for the carrier coating resin materials that will provide high negative charge to silica on a toner surface and to provide a description of chemical functionality that will in addition provide an improved RH sensitivity, computer modeling

was used to understand and to define properties that are important, inter alia, to design (and to predict properties of) new materials that have both high charge and good RH sensitivity.

Materials

[00119] Methacrylates with methyl, cyclohexyl, dimethylamino, phenyl and benzyl substituents were studied.

Modeling

[00120] For all substituted methacrylates, a trimer was used to represent the polymer to be examined. To distinguish possible effects of C (carbon) rich and O (oxygen) rich functional groups (alkyl/aromatic and acyl) in the polymer, all three acyl groups were designed to coordinate to the same side.

[00121] To mimic the surface hydroxyl group of the silica model, a one layer cylinder-like silica model was used to design the surface-treated silicas with the formula, $\text{Si}_{12}\text{O}_{32}\text{H}_{16}$. In the model, all silicas were in tetrahedral geometry and connected by oxygen. The edge of the cylinder was terminated by two hydroxyl groups to represent the geminal silanols $[\text{Si}(\text{OH})_2]$ which typically are on the surface of β -cristobalite, identified experimentally on the amorphous silica surface as one of the two types of surface hydroxyl group of untreated silica (see, e.g., Leonardelli, et al., J Am Chem Soc (1992) 114:6412; Vigné-Maeder & Sautet, Phys Chem (1997) 101(41):8197). All calculations were performed with the DMol3 module from the Accelrys Materials Studio 4.2 commercial software package. Density functional theory (DFT) was used for the study of surface electronic properties of all models and the coupled toner/carrier complexes. Due to the advantages of accuracy at reasonable computational efficiency, the DFT method has been successfully applied to the electronic structure modeling of materials.

[00122] Recent extensions of the DMol3 density functional method are designed to do electronic structure calculations for local and gradient-dependent functional, depending on the accuracy needed. In the present modeling example, Perdew's 91 generalized gradient approximations (PW91PW91) were employed as the density functional method. For basis sets, a double numerical basis set with d-polarization functions (DND) was used for all

calculations. For different basis set types, it has been reported that DND performs better than a Gaussian-type basis set of the same size, which is 6-31G*. The DND numerical solutions can give highly accurate DFT solutions for the separated atoms limit for molecular and solids calculations.

[00123] The initial structure, optimized structure and electronic properties of adsorbed polymer complexes on the silica were studied. The geometry optimization convergence was achieved when the energy, gradient and displacement were lower than 2×10^{-5} Ha, 4×10^{-3} Ha/Å, and 5×10^{-3} Å, respectively. Ha is the Hartree atomic units (au), where $1 \text{ au} = 4.359 \times 10^{-18}$ Joules. The calculations of HOMO-LUMO orbitals were performed to understand the direction of charge transfer of the above models and to identify the most essential factor that could affect electron transfer in the complex models.

[00124] Electron transfer active sites on different materials are crucial to triboelectric charge since the sites are the root and destination of electron transfer, and the relative ability of donating and accepting electrons will directly determine the triboelectric charge properties of certain toner/charger pairs. Generated surface electron densities in HOMO and LUMO are used to describe the active sites of the donor-acceptor complex in charge transfer. Fukui functions for $+0.1 |e|$ and $-0.1 |e|$ charges also were calculated.

[00125] To explore further the reverse gap and forward gap of charge transfer, the excited orbitals for the above systems were investigated. Generally, 10 levels of orbital above (M+9) and below (n-9) the Fermi levels were calculated. The lowest energy gap for both forward electron transfer and reverse electron transfer were collected from the set of twenty orbitals. The calculation errors of the energy gaps were evaluated by comparing the forward and reverse electron transfer barriers of DMEAMA dimer and trimer/silica complexes.

Results

[00126] The key attributes for high negative toner charge are:

- 1) the gap for the forward charge transfer should be low;
- 2) the reverse gap should be higher than the forward gap (a negative gap difference, subtracting 1) from 2));
- 3) the resin has a high C/O ratio monomer; and

4) the water adsorption at the charging site must be low.

Attributes (1) and (2): The gap for the forward charge transfer should be low and the reverse gap should be higher than the forward gap (negative gap difference).

[00127] It is art recognized that in the usual intramolecular electron transfer, within a single material, the adsorption of sufficient energy from a photon or collision or thermal energy can result in transfer of an electron from the HOMO to the LUMO. Since the electron and hole (left when the electron leaves the HOMO) are both on the same molecule, there is no net charge on the molecule. The size of the energy gap determines the amount of energy required to transfer the electron between the orbitals. So, as shown in Figure 1, both the carrier resin, for example, and toner additive, before in contact, have a HOMO and a LUMO and an associated gap. It should be noted that there are also potentially other energy levels above the LUMO (known as LUMO+1, LUMO+2 etc. of increasing energy) and below HOMO (known as HOMO-1, HOMO-2 etc. of decreasing energy). So, in general, it is possible to transfer an electron from a HOMO- n to a LUMO + m , where $n, m \geq 0$, within a material. (Note HOMO $n=0$ is usually written as HOMO and LUMO $m=0$ as LUMO for simplicity.)

[00128] On contact of two materials, such as, between the toner additive and the carrier, a number of different possibilities arise for the location of HOMO- n and the LUMO+ m , and thus, the result of charge transfer has a number of different possibilities. The contact of the two materials may result in the HOMO- n being located on the carrier resin and LUMO+ m on the toner additive. The electron transfer will charge the carrier resin positive and the toner additive negative (desired transfer for negative charging toner) (Figure 2). On the other hand, if the LUMO+ m is located on the carrier resin and the HOMO- n is on the toner additive, the electron transfer will charge the toner additive positive and the carrier resin negative (undesired transfer for negative charging toner). Of course, the HOMO and LUMO may be located on just one molecule or could be partially on both molecules. The disposition of the frontier molecular orbitals that results is a consequence of the properties of the two materials and interaction therebetween, that interaction also depending on the orientation of the two molecules in contact. In a bulk sample of material, different orientations of the

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molecules in contact will be obtained randomly. So, the overall charge transferred is the sum of the different processes. Fortunately, the important processes for charge transfer will be that of the lowest energy, so the lowest energy gap for the forward charge transfer desired (e.g., negative toner charge) and the lowest energy gap for reverse charge transfer (i.e., positive toner charge) are determined.

[00129] Thus, for high negative toner charge in the charging of toners with silica and carriers with a polymeric resin coating:

- 1) the gap for the forward charge transfer should be low and
- 2) the reverse gap should be higher than the forward gap (a negative gap difference, subtracting 1) from 2)).

[00130] Table 1 below shows electron charge transfer to silica (desirable) and electron charge transfer to a polymer (not desirable) for a number of different coating materials.

Table 1. Data for Charge Transfer

	Charge Transfer Polymer to Silica (eV)	Charge Transfer Silica to Polymer (eV)	Positive Carrier Charge Prediction	Measured Charging	Calculated C/O Ratio	Delocalization on Hydrophobic Group		Measured RH Ratio
						HOMO	Fukui (-) Function	
MMA	4.793	6.236	Good	Good	2.5	4%	12%	Worse than CHMA
CHMA	4.667	6.513	Good	Good	5.0	18%	15%	Good
CHMA/DMAEMA	4.275	6.255	Good	Good	≈5	No data	No data	Worse than CHMA as more DMAEMA added
DMAEMA	3.734	5.233	Good	No data	4	No data	No data	No data

PHMA	4.720	5.659	Good	No data	5	90%	90%	No data
BMA	4.960	5.158	Good	No data	5.5	50%	50%	No data
HMA	5.263	>6.005	Lower	No data	5	6%	6%	No data

[00131] Thus, as expected, data show that polymers based on PMMA, PCHMA and DMAEMA/PCHMA have good high charging, as all have a charge transfer gap for forward transfer that is about equal to or lower than that of PMMA. In all cases, the energy for reverse transfer is higher and thus not favored.

[00132] The last four entries of Table 1 show that DMAEMA and PHMA (phenyl methacrylate) present with lower and approximately equal forward gap energies to MMA and CHMA, with calculation error of ≈ 0.045 eV (the error mostly arising from the size difference among polymer trimer models), BMA (benzyl methacrylate) is close to the gap of MMA, however, HMA (hexyl methacrylate) has a larger gap than PMMA and thus will not charge as high. Hence, carrier polymers made from DMAEMA or PHMA will be suitable to charge silica. Also, DMAEMA can be used to reduce the gap for forward electron transfer to the silica toner additive. The combination of BMA or PHMA with some DMAEMA would also provide a suitable gap, with equal or lower energy to CHMA or PMMA, and lower forward energy gap than reverse gap, as required for net negative charge transfer to silica. DMAEMA does however increase the relative humidity sensitivity of the charge because of the polar nitrogen atom which increases water adsorption at the charge site (which is the N atom).

Attribute (3): Carbon to Oxygen Ratio.

[00133] The MMA C/O ratio is low at 2.5 and has a poor RH sensitivity. CHMA has a C/O ratio of 5, and thus much improved RH ratio. DMAEMA has a C/O ratio of 4 and thus can degrade RH sensitivity, so should be used in small amounts to maintain the best RH sensitivity. Hexyl methacrylate and phenyl methacrylate maintain the high C/O ratio of 5, similar to CHMA, and should also provide a good RH ratio. Phenyl acrylate has been used as a potential carrier (see, e.g., US Pat. No. 6,511,780) however, phenyl acrylate has a lower C/O ratio (4.5) compared to CHMA, and

therefore is not expected to provide as good RH sensitivity as phenyl methacrylate. Indeed, all things being equal, the C/O ratio, and thus RH sensitivity, will show greater improvement for any methacrylate as compared to the acrylate analogue, and as such, methacrylates represent the preferred ester. In general, methacrylates have higher T_g than acrylates, which also is preferred for robust coating that will not soften on the carrier surface in the developer housing. Benzyl methacrylate has the highest C/O ratio (i.e., 5.5), and therefore is expected to provide even greater RH sensitivity than CHMA.

Attribute (4): Water Adsorption at the Charging Site

[00134] As stated above, another key attribute is that the resin has the HOMO electron density located substantially on the hydrophobic group and a reduced electron density on the carbonyl (C=O) oxygen atom. Calculated electron density distributions for the HOMO of an isolated trimer (three repeat units for each monomer candidate was carried out) for PMMA (with a methyl hydrophobic substituent), PHMA (with a hexyl hydrophobic substituent), PCHMA (with a cyclohexyl hydrophobic substituent), PPHMA (with a phenyl hydrophobic substituent), and PBMA (with a benzyl hydrophobic substituent).

[00135] In PMMA, the carbonyl oxygen has the HOMO frontier molecular orbitals located for the most part on the carbonyl oxygen atom, which is very polar and attracts water in higher relative humidity, and interacting with that critical charging site the HOMO is the electron transfer site to provide negative charge to the toner silica additive acceptor site. Thus, water adsorption at high RH will disturb the charge transfer (either by physically blocking the HOMO charge site or by actually disturbing the HOMO itself). For PCHMA, with the higher C/O ratio there is some delocalization of the HOMO from the carbonyl oxygen onto the hydrophobic substituent, the cyclohexyl group. That trend is extended for PBMA, where roughly half of HOMO electron density stays located on the carbonyl oxygen atom and half are now on the hydrophobic substituent, the benzyl group. Since that part of the molecule is more hydrophobic than the carbonyl oxygen, it is less sensitive to water adsorption at high RH, and thus will be less disturbed by water. The ideal material will have the majority of the HOMO molecular orbitals on the hydrophobic substituent where it will not be disturbed as much by water adsorption, and little of the HOMO molecular orbitals on

the carbonyl oxygen, as is the case of PPHMA where the HOMO is mostly located on the phenyl ring. The approximate electron density for the HOMO for the polymers studies is summarized in Table 1.

[00136] An alternative way to study the electron density distribution is to assess the, “softness,” of the HOMO as given by the Fukui (-) function with an overall positive charge of +0.1. The Fukui function can be viewed as the location of the active polarizable electron densities for charge transfer to the silica, the calculation of the Fukui function is described, for example, in, Ayers et al., in, “Chemical Reactivity Theory: A Density Functional View,” Chatteraj, ed., Chap. 18, pp. 255-267, CRC Press, 2009. As with the HOMO electron densities, there is a trend of increased delocalization of the Fukui (-) function electron densities from the carbonyl oxygen to the hydrophobic group, as the hydrophobic group is changed from methyl, to hexyl, to cyclohexyl, and finally to benzyl and phenyl, both of which have a majority of the electron density on the hydrophobic group. Again, as the electron density is on the hydrophobic group and on the polar carbonyl group, it is expected to result in improved RH sensitivity (again, see Table 1).

[00137] It will be appreciated that several of the above-disclosed and other features and functions, or alternatives thereof, may be desirably combined into many other different systems or applications. Also various presently unforeseen or unanticipated alternatives, modifications, variations or improvements therein may be subsequently made by those skilled in the art, which are also intended to be encompassed by the following claims.

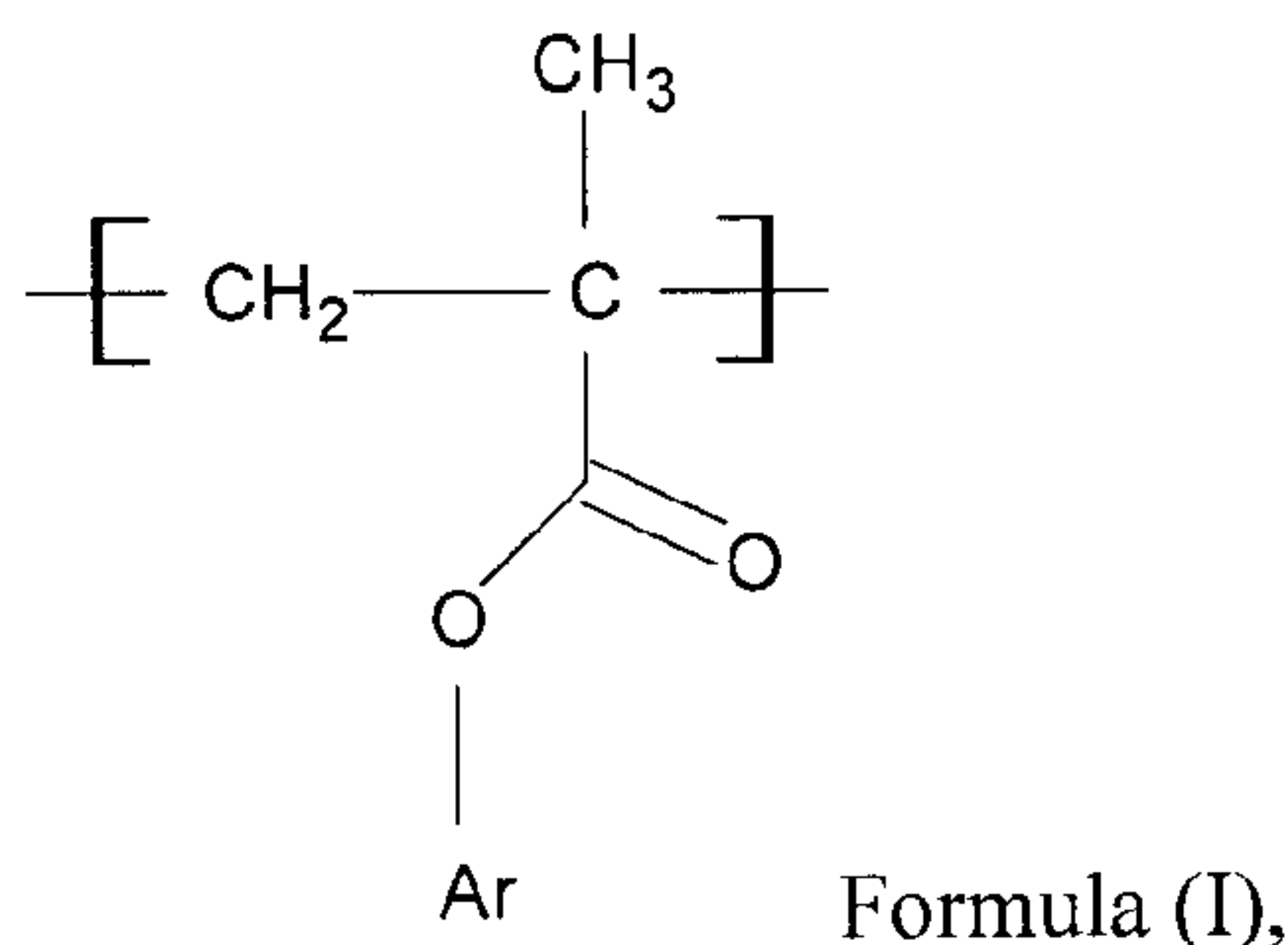
[00138] Unless specifically recited in a claim, steps or components of claims should not be implied or imported from the specification or any other claims as to any particular order, number, position, size, shape, angle, color or material.

We claim:

1. A toner carrier composition comprising:

a polymer coating resin consisting of a methacrylate monomer having a hydrophobic substituent, Ar, and a secondary aminoacrylate monomer, wherein the highest occupied molecular orbital (HOMO) of the methacrylate-containing polymer has greater than about 25% of the electron density of those molecular orbitals located on the hydrophobic substituent and less than about 75% of the electron density of those molecular orbitals on the carbonyl group of the methacrylate moiety, and wherein said methacrylate monomer having a hydrophobic substituent is present from about 98% to about 100% by weight of the polymer and said secondary aminoacrylate monomer is present (a) from about 0.5 to about 1%; or (b) from about 1 to about 1.5%; or (c) from about 1.5 to about 2% by weight of the polymer,

further wherein the methacrylate monomer has the formula as set forth in formula (I):



wherein Ar comprises at least one aromatic group, and

the secondary amino acrylate monomer is selected from the group consisting of dimethylaminoethyl methacrylate, diethylaminoethyl methacrylate, and diethylaminopropyl methacrylate; and

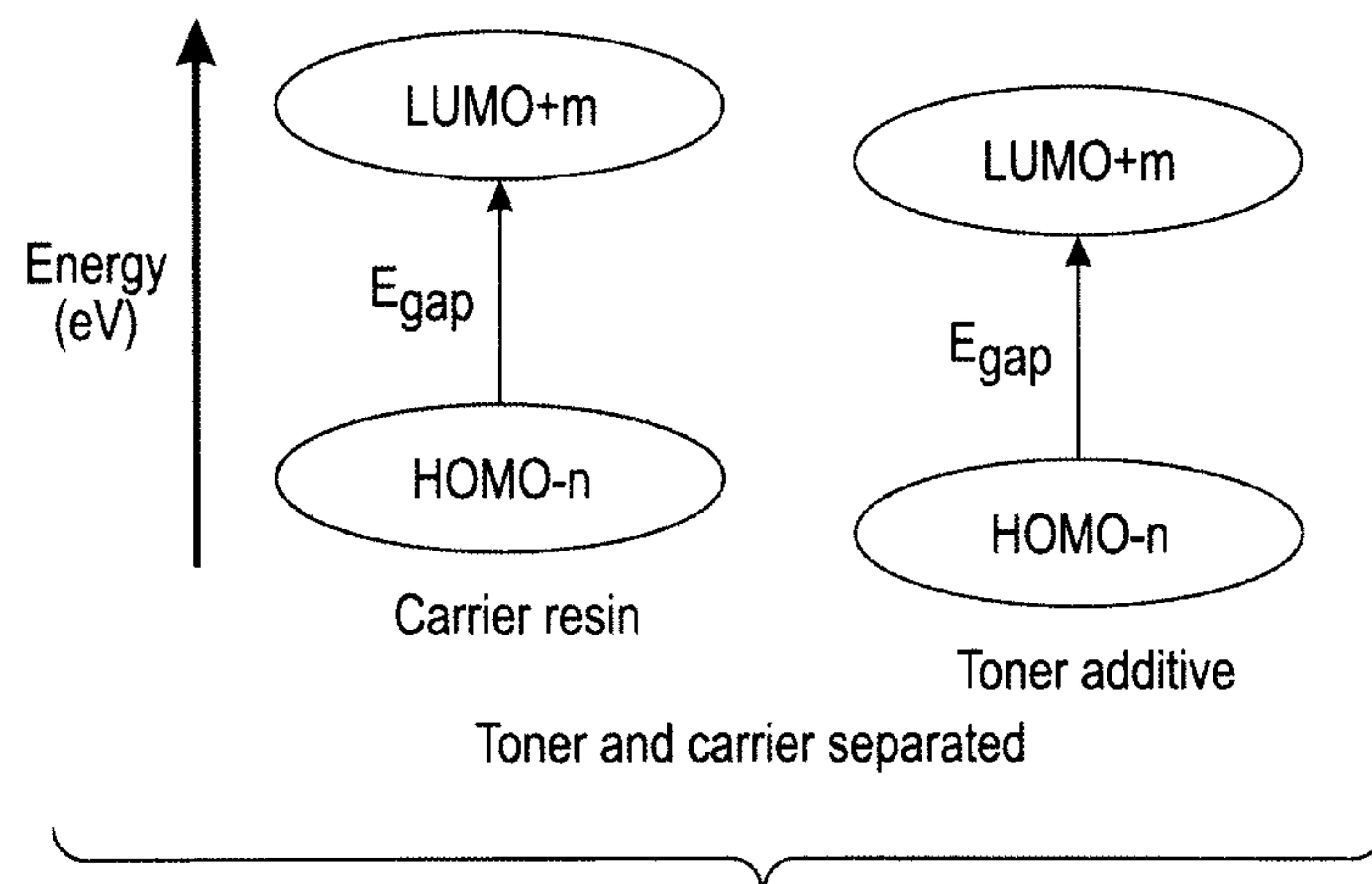
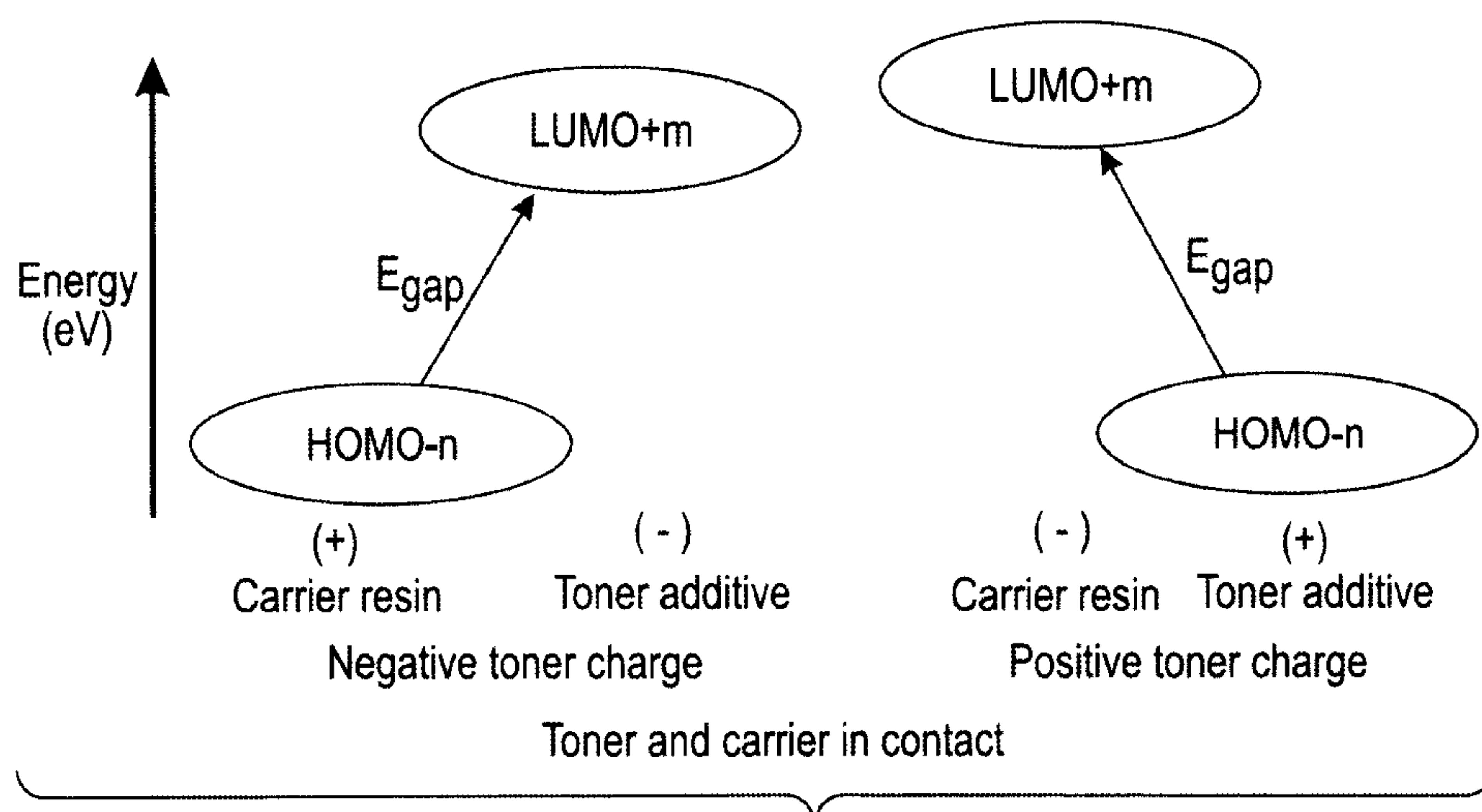
a carrier core particle, wherein the polymer coating resin is adhered to the surface of the carrier core particles.

2. The carrier composition of claim 1, wherein the monomer exhibits increased delocalization of Fukui (-) function electron densities from the carbonyl group to the hydrophobic substituent as compared to methyl-containing, hexyl-containing or cyclohexyl-containing methacrylate polymers.

3. The carrier composition of claim 1 or 2, wherein Ar is a benzyl group, a naphthyl group or a phenyl group.
4. The carrier composition of any one of claims 1 to 3, wherein the methacrylate monomer has a carbon:oxygen (C/O) ratio of 4.75 or more.
5. The carrier composition of any one of claims 1 to 4, wherein said secondary aminoacrylate monomer is present at about 1% by weight of the polymer.
6. The carrier composition of any one of claims 1 to 5, wherein said at least one methacrylate monomer has an improved charging RH ratio as compared to that of methylmethacrylate (MMA).
7. A developer comprising:
 - the carrier composition of any one of claims 1 to 6;
 - a toner;
 - and a toner surface additive,wherein the HOMO electron density located on the carbonyl group of said at least one methacrylate monomer is reduced.
8. The developer of claim 7, wherein said toner surface additive comprises a silica, a titania, an alumina or mixtures thereof.
9. The developer of claim 8, wherein said silica comprises at least one silica.
10. The developer of claim 7 to 9, wherein an energy gap for forward charge transfer of electrons for said at least one methacrylate monomer and a toner surface additive is at less than about 5.1 eV.

11. The developer of any one of claims 7 to 10, wherein said toner comprises an emulsion aggregation toner.

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**FIG. 1****FIG. 2**