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(54) Title: METHOD FOR PRODUCING COUPLED ASYMMETRIC POLYMERS

(57) Abstract

A process for preparing a coupled asymmetric block copolymer of a vinyl aromatic hydrocarbon and a conjugated diene which comprises: (a) adding sufficient anionic polymerization initiator to the vinyl aromatic hydrocarbon monomer to start polymerization of the vinyl aromatic hydrocarbon, (b) polymerizing the vinyl aromatic hydrocarbon for a time sufficient to form a first set of vinyl aromatic hydrocarbon blocks having a desired molecular weight, (c) adding sufficient anionic polymerization initiator to start polymerization of a second set of vinyl aromatic hydrocarbon blocks, (d) completing the polymerization of the vinyl aromatic hydrocarbon for a time such that the first set of vinyl aromatic hydrocarbon blocks continues to grow to the desired molecular weight and the second set of vinyl aromatic hydrocarbon blocks grows to a desired smaller molecular weight, (e) adding a conjugated diene and polymerizing it on the ends of the vinyl aromatic hydrocarbon blocks for a time sufficient to form vinyl aromatic hydrocarbon-conjugated diene arms of the desired molecular weight, and (f) adding a coupling agent and coupling the arms to form the asymmetric block copolymer.

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METHOD FOR PRODUCING COUPLED ASYMMETRIC POLYMERS

The present invention relates to a method for producing asymmetric polymers. More particularly, this invention relates to a method for producing asymmetric polymers having vinyl aromatic hydrocarbon-conjugated diene copolymer arms wherein the vinyl aromatic hydrocarbon blocks of a controlled portion of the arms differ in molecular weight.

Heretofore, several methods have been proposed for preparing asymmetric radial polymers. As is well known in the prior art, linear coupled polymers are comprised of two arms coupled in the center by a coupling agent and radial polymers comprise three or more arms extending outwardly from a coupling agent nucleus. The known asymmetric polymers contain arms of at least two different polymers, which polymers may vary as to chemical composition, structure and/or molecular weight. A principle difference in the methods frequently used to prepare asymmetric polymers resides in the selection of a coupling agent which forms the center or nucleus of the polymer. Multifunctional coupling agents such as silicon tetrachloride have been used to form asymmetric radial polymers having three or four arms. Star-shaped radial polymers having many more arms have been formed using a poly alkenyl aromatic compound, such as divinyl benzene, as the coupling agent as described in Canadian Patent 716,645 and U.S. Patent 3,985,830. Linear coupled polymers are formed using difunctional coupling agents such as dibromoethane.

Heretofore, linear coupled polymers were made by coupling two identical block copolymer arms, A-B (where A is a vinyl aromatic hydrocarbon and B is a conjugated diene) to form a symmetrical triblock polymer, A-B-A. Generally, asymmetric radial polymers referred to in the prior art are asymmetric in the sense that there are at least two different types of arms, one of which is an A-B

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arm and the other is either a B or an A arm, i.e., homopolymer polydiene or homopolymer poly vinyl aromatic hydrocarbon. The present invention provides a method for making asymmetric polymers which have at least two sets of A-B block copolymer arms wherein 5 the A blocks of the block copolymer arms have differing molecular weights. Such polymers have the advantage of having a relatively high styrene content while at the same time maintaining a soft, tacky character. Further, the A blocks of differing molecular weight are made in a single reactor. Therefore, the process of the 10 present invention also has the advantage of requiring a reduced amount of process equipment.

A process is provided herein for preparing a coupled asymmetric block copolymer of a vinyl aromatic hydrocarbon and a conjugated diene, comprising:

15 (a) adding sufficient anionic polymerization initiator to a vinyl aromatic hydrocarbon monomer to start polymerization of that monomer;

(b) polymerizing the vinyl aromatic hydrocarbon for a time sufficient to form a first set of vinyl aromatic hydrocarbon blocks 20 having a desired molecular weight;

(c) adding sufficient anionic polymerization initiator to start polymerization of a second set of vinyl aromatic hydrocarbon blocks;

(d) completing the polymerization of the vinyl aromatic 25 hydrocarbon for a time such that the first set of block copolymer arms continues to grow to the desired molecular weight and the second set of vinyl aromatic hydrocarbon blocks grows to a desired smaller molecular weight;

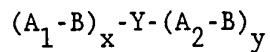
(e) adding conjugated diene and polymerizing on the ends of the 30 vinyl aromatic hydrocarbon blocks for a time sufficient to form vinyl aromatic hydrocarbon- conjugated diene block copolymer arms of the desired molecular weight; and

(f) adding a coupling agent and coupling the arms to form the asymmetric block copolymer.

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The asymmetric block copolymer will be linear if a difunctional coupling agent such as dibromoethane is used. The asymmetric block copolymer will be a four-armed radial block copolymer if a tetrafunctional coupling agent such as silicon tetrachloride is used. The asymmetric block copolymer will be a star block copolymer if divinylbenzene or some other coupling monomer is used as the coupling agent. The preferred vinyl aromatic hydrocarbon is styrene and the preferred conjugated dienes are butadiene and isoprene.

Accordingly, the block copolymers of the present invention have an idealized structure as follows:



The subscripts x and y range from 1 to 40. The A blocks are polymer blocks of a vinyl aromatic hydrocarbon. A_1 and A_2 are blocks of different molecular weights. Preferably, the vinyl aromatic hydrocarbon is styrene. Other useful vinyl aromatic hydrocarbons include alphamethyl styrene, various alkyl-substituted styrenes, alkoxy-substituted styrenes, vinyl naphthalene, vinyl toluene and the like. The B blocks are polymer blocks of conjugated dienes. Preferred dienes include butadiene and isoprene, with isoprene being much preferred. Other dienes may also be used, including piperylene, methylpentadiene, phenylbutadiene, 3,4-dimethyl-1,3-hexadiene, 4,5-diethyl-1,3-octadiene and the like, preferably those conjugated dienes containing 4 to 8 carbon atoms. Mixtures of conjugated dienes may also be employed.

The Y moiety stands for the multifunctional coupling agent or coupling monomers or initiators used herein. There are a wide variety of coupling agents that can be employed. Any polyfunctional coupling agent which contains at least two reactive sites can be employed. Examples of the types of compounds which can be used include the polyepoxides, polyisocyanates, polyimines, polyaldehydes, polyketones, polyanhydrides, polyesters, polyhalides, and the like. These compounds can contain two or more types of functional groups such as the combination of epoxy and

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aldehyde groups, isocyanate and halide groups, and the like.

Various other substituents which are inert in the treating reaction can be present such as hydrocarbon radicals as exemplified by the alkyl, cycloalkyl, aryl, aralkyl and alkaryl groups and the alkoxy,

5 aryloxy, alkythio, arylthio, and tertiary amino groups. Many suitable types of these polyfunctional compounds have been described in U.S. Patent Nos. 3,595,941; 3,468,972, 3,135,716; 3,078,254, and 3,594,452. The coupling agent of the present invention may be $X_3\text{Si}-(\text{CH}_2)_n-\text{Si}X_3$ where $n \geq 0$ and X is halogen or 10 alkoxy, including bis-trimethoxy-silylethane, hexachlorodisiloxane, bis-trichlorosilylethane, 1,6-bis(trichlorosilyl)-hexane, and hexachlorotriphosphazine to make 6 armed polymer blends, and octachlorotrisiloxane, $\text{Cl}_3\text{Si}-(\text{CH}_2)_n-\text{SiCl}_2-(\text{CH}_2)_n-\text{SiCl}_3$, where $n \geq 0$, to make 8 armed polymer blends.

15 A much preferred coupling agent is a polyalkenyl aromatic coupling agent. Polyalkenyl aromatic coupling agents capable of forming star shaped polymers are known in the art. See generally, Canadian patent number 716,645 and U.S. Patents 4,010,226 and 3,985,830. A detailed description of a variety of such coupling 20 agents is found in U.S. Patent 4,391,949. Examples of suitable polyvinyl aromatic compounds are 1,2-divinyl benzene, 1,3-divinylbenzene, 1,4-divinylbenzene, 1,2,4-trivinylbenzene, 1,3-divinylnaphthalene, 1,8-divinylnaphthalene, 1,3,5-trivinylnaphthalene, 2,4-divinylbiphenyl, 3,5,4'-trivinylbiphenyl, 25 1,2- divinyl-3,4-dimethylbenzene, 1,5,6-trivinyl-3,7-diethyl-naphthalene, 1,3-divinyl-4,5,6-tributylnaphthalene, 2,2'-divinyl-4-ethyl-4'-propylbiphenyl and the like. Divinyl aromatic hydrocarbons containing up to 26 carbon atoms per molecule are preferred for employment according to this invention. The most 30 preferred coupling agent is divinylbenzene because of its high efficiency in coupling to high arm numbers. Said coupling agent can be used in either its ortho, meta or para isomer and commercial divinylbenzene which is a mixture of said isomers is also quite satisfactory.

35 In general, the method of this invention may be used to prepare asymmetric polymers with any polymer containing a reactive

end group which will react with one or more functional groups contained in the selected coupling agent. The method is particularly suitable for the preparation of asymmetric polymers from so-called "living" polymers containing a single terminal metal 5 ion. As is well known in the prior art, "living" polymers are polymers containing at least one active group such as a metal atom bonded directly to a carbon atom. "Living" polymers are readily prepared via anionic polymerization. Since the present invention is particularly well suited to the preparation of asymmetric 10 polymers using "living" polymers to form the arms thereof, the invention will be described by reference to such polymers. It will, however, be appreciated that the invention would be equally useful with polymers having different reactive groups so long as the selected coupling agent contains functional groups which are 15 reactive with the reactive site contained in the polymer.

Living polymers containing a single terminal group are, of course, well known in the prior art. Methods for preparing such polymers are disclosed, for example, in U.S. Patents Nos. 3,150,209; 3,496,154; 3,498,960; 4,145,298 and 4,238,202. Methods 20 for preparing block copolymers such as those preferred for use in the method of the present invention are also disclosed, for example, in U.S. Patent Nos. 3,231,635; 3,265,765 and 3,322,856.

When the polymer product is a random or tapered copolymer, the monomers are, generally, added at the same time, although the 25 faster reacting monomer may be added slowly in some cases, while, when the product is a block copolymer, the monomer used to form the separate blocks are added sequentially.

In general, the polymers useful as arms in both the method of this invention and the asymmetric polymer of this invention may be 30 prepared by contacting the monomer or monomers with an organoalkali metal compound in a suitable solvent at a temperature within the range from -150°C to 300°C, preferably at a temperature within the range from 0°C to 100°C. Particularly effective polymerization initiators are organolithium compounds having the general formula:

wherein R is an aliphatic, cycloaliphatic, alkyl-substituted cycloaliphatic, aromatic or alkyl-substituted aromatic hydrocarbon radical having from 1 to 20 carbon atoms.

In general, the block copolymer is useful as arms in the 5 method of the present invention will have a GPC peak molecular weight within the range of 25,000 to 250,000. The polyvinyl aromatic hydrocarbon content of these polymers generally will range from 3 percent to 50 percent by weight and preferably from 10 to 30 percent by weight. The molecular weight of the longer 10 polyvinylaromatic hydrocarbon blocks will generally range from 5,000 to 25,000 and the molecular weights of the smaller polyvinyl aromatic hydrocarbon blocks will generally range from 1,000 to 15,000.

Molecular weights of linear polymers or unassembled linear 15 segments of polymers such as mono-, di-, triblock, and etc., arms of star polymers before coupling are conveniently measured by Gel Permeation Chromatography (GPC), where the GPC system has been appropriately calibrated. Polymers of known molecular weight are used to calibrate and these must be of the same molecular structure 20 and chemical composition as the unknown linear polymers or segments that are to be measured. For anionically polymerized linear polymers, the polymer is essentially monodisperse and it is both convenient and adequately descriptive to report the "peak" molecular weight of the narrow molecular weight distribution 25 observed. The "peak" molecular weight is nearly the same as the weight average molecular weight of the linear polymer or arm. Frequently, two or more peaks will be observed for unassembled 30 asymmetric stars. Measurement of the true molecular weight of the final coupled star polymer is not as straightforward or as easy to make using GPC. This is because the star shaped molecules do not separate and elute through the packed GPC columns in the same manner as do the linear polymers used for the calibration, and, hence, the time of arrival at a UV or refractive index detector is 35 not a good indicator of the molecular weight. A good method to use for a star polymer is to measure the weight average molecular

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weight by light scattering techniques. The sample is dissolved in a suitable solvent at a concentration less than 1.0 gram of sample per 100 milliliters of solvent and filtered using a syringe and porous membrane filters of less than 0.5 microns pore size directly 5 into the light scattering cell. The light scattering measurements are performed as a function of scattering angle and of polymer concentration using standard procedures. The differential refractive index (DRI) of the sample is measured at the same wavelength and in the same solvent used for the light scattering.

10 The light scattering method may also have to be used to characterize the molecular weight of the unassembled arms, if the arm molecular weight distribution is broad. With reference to the discussed analysis methods the following references are relevant:

1. Modern Size-Exclusion Liquid Chromatography, M. W. Yau, J. J. Kirkland, D. D. Bly, John Wiley & Sons, New York, NY, 1979;
2. Light Scattering from Polymer Solutions, M. B. Huglin, ed., Academic Press, New York, NY 1972;
3. W. Kay and A. J. Havlik, Applied Optics, 12, 541 (1973);
4. M. L. McConnell, American Laboratory, 63, May, 1978.

20 As described in U.S. Patent No. 4,096,203, usually the styrene is contacted with the initiator. Next, the living polymer in solution is contacted with isoprene. The resulting living polymer has a simplified structure A-B-Li. It is at this point that the living polymer is coupled.

25 In general, the living polymers used as arms in the asymmetric polymer will be contacted with the coupling agent at a temperature within the range from 0°C to 100°C at a pressure within the range from 0 bar to 7 bar and the contacting will be maintained until reaction between the arms and the coupling agent is complete or at 30 least substantially completed, generally for a period of time within the range from 1 to 180 minutes.

35 The method of this invention results in an improved relative arm distribution because the reactivity of the functional groups contained in the coupling agent stays relatively high as the number of functional groups contained in the coupling agent is reduced as

the result of reaction with the metal-carbon bond contained in the living polymer. This allows the coupling reaction to take place quickly.

In general, the polymers useful as arms in the asymmetric polymers of this invention will be in solution when contacted with the coupling agent. Suitable solvents include those useful in the solution polymerization of the polymer and include aliphatic, cycloaliphatic, alkyl- substituted cycloaliphatic, aromatic and alkyl-substituted aromatic hydrocarbons, ethers and mixtures thereof. Suitable solvents, then, include aliphatic hydrocarbons such as butane, pentane, hexane, heptane and the like, cycloaliphatic hydrocarbons such as cyclohexane, cycloheptane and the like, alkyl-substituted cycloaliphatic hydrocarbons such as methylcyclohexane, methylcycloheptane and the like, aromatic hydrocarbons such as benzene and the alkyl-substituted aromatic hydrocarbons such as toluene, xylene and the like and ethers such as tetrahydrofuran, diethylether, di-n-butyl ether and the like. Since the polymers useful in making the asymmetric polymers of this invention will contain a single terminal reactive group, the polymers used in preparation of the asymmetric polymers will be retained in solution after preparation without deactivating the reactive (living) site. In general, the coupling agents may be added to a solution of the polymer or a solution of the polymer may be added to the coupling agent.

Following the coupling reaction or when the desired coupling efficiency has been obtained, the product is neutralized such as by the addition of terminators, e.g., hydrogen, water, alcohol or other reagents, for the purpose of removing the lithium radical forming the nucleus for the condensed polymer product. The product is then recovered such as by coagulation utilizing hot water or steam or both.

For convenience, hereinafter the vinyl aromatic hydrocarbon monomers will be referred to as styrene and the polyvinyl aromatic hydrocarbon blocks will be referred to as polystyrene blocks.

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The first step is adding sufficient anionic polymerization initiator to the styrene monomer to start polymerization thereof. In general, from 5 percent to 95 percent of the initiator is required. The polymerization of the styrene monomer is carried out 5 for a time sufficient to form a first set of styrene blocks having a desired molecular weight. The time to achieve the desired molecular weight is calculated with knowledge of the reactant concentrations, the reaction temperature, and some estimate of the rate at which the reactants are consumed.

10 In general, the rate of consumption of monomer is determined by conducting experiments which measure the concentration of residual monomer as a function of time for various conditions of solvent type, monomer concentration, initiator concentration, and temperature. Alternately, in the case of living polymerizations, 15 the reaction rate may be determined by measuring the increase in polymer product molecular weight with time for the same conditions mentioned above. With knowledge derived by such efforts, one can estimate the molecular weight and the amount of monomer consumed for a particular time at specified, but not necessarily constant, 20 reaction conditions. Conversely, the time required to form a polymer of a specific weight and consumption of a specific fraction of the monomer can be estimated with knowledge of the reaction rate and conditions.

25 After completion of the foregoing steps, the reaction mixture contains living polystyrene blocks of the desired molecular weight and styrene monomer. At this point, additional initiator is added in an amount sufficient to start the polymerization of a second set 30 of polystyrene blocks. Meanwhile, the polymerization of the first set of polystyrene blocks continues. This step is carried out until complete consumption of the styrene monomer occurs. At 35 complete consumption of the styrene monomer, there exists living polystyrene blocks of the distinct molecular weights. The final molecular weight of the larger blocks should be equal to the sum of their molecular weight after the initial polymerization and the molecular weight of the smaller blocks.

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At this point, the reaction mixture contains two sets of living polystyrene blocks with different molecular weights. All of the styrene monomer should have been reacted. Then the diene monomer is added to the reaction mixture and polymerization of it 5 on the ends of the living polystyrene blocks is carried out until the diene blocks achieve their desired molecular weight, preferably when all of the diene has been reacted.

At this point, the reaction mixture contains two sets of living polystyrene-polydiene block copolymer arms. The two sets 10 have identical diene block molecular weights but different styrene block molecular weights. A coupling agent is added to this mixture and the arms are coupled to form the desired asymmetric block copolymer. The conditions of coupling will vary depending upon whether the desired polymer is linear, radial or star and also upon 15 the coupling agent used as heretofore described.

The polymers produced by this process are characterized in that they have polystyrene-polydiene block copolymer arms wherein 20 the polystyrene block molecular weights of some of the arms is greater than that of others. The coupling process will produce a statistical distribution of the arms. In other words, some of the polymers will have all arms of one type and others will have one of one type and x of another type, etc. The overall statistical distribution will be equivalent to a block copolymer which contains the differing arms in the ratios in which the arms were grown in 25 the early steps of the process. These polymers are useful as soft, tacky adhesives which may be cured to their final state by a variety of means including but not limited to chemical vulcanization and radiation crosslinking.

Examples

30 Ten asymmetric star block copolymers were made according to the process of the present invention. The molecular weights of the longer polystyrene blocks (MW long) created herein ranged from 7,000 to 21,000. The molecular weights of the shorter polystyrene blocks (MW short) ranged from 2,000 to 10,000. x is the number 35 fraction of long styrene arms as opposed to short styrene arms

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desired in the polymer and is also the molar fraction of the total amount of initiator used in the process which is to be used in the first polystyrene polymerization step to initiate the longer polystyrene blocks. Cyclohexane was added to a nitrogen purged 500 ml glass reactor. Styrene monomer was added and mixed into solution. This solution was then brought to the desired temperature. Once stabilized at the desired temperature, the first aliquot of initiator was added. The initiator was 1.4 molar sec-butyllithium in cyclohexane. At a time specified as t_1 , the second aliquot of initiator was added. The styrene was allowed to polymerize at least until t_{end} . The conditions of the styrene polymerizations are listed in Table I.

At the end of the styrene polymerization, the living styryl lithium was added to a purified solution of isoprene in cyclohexane in a 2000 ml glass reactor. Table II lists the amount of isoprene polymerized for each polymer. The mass of styrene and isoprene amounted to 15 percent of the total mass of the solution. The solution temperature was 60°C. The polymerization of isoprene was conducted at these conditions for 60 minutes. After polymerization of the isoprene the living arms were coupled using purified 55 percent divinyl benzene for 60 minutes at 60°C. 3 moles of divinyl benzene were added for each mole of lithium. Table III lists the molecular description of the polymers so made.

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Table I

Conditions for Styrene Polymerization

Polymer	Mass	Mass	Temp (°C)	First		Second	
	Cyclohexane (g)	Styrene (g)		S-Buli (ml)	t_1 (Min)	S-buli (ml)	t_{end} (Min)
1	120.9	13.4	50	0.53	2.97	2.13	14.7
2	96.6	10.7	40	0.22	1.86	0.90	36.9
3	85.8	9.5	50	0.23	7.19	0.91	22.2
4	279.7	31.1	50	0.48	3.46	1.93	22.3
5	279.7	31.1	50	0.48	3.46	1.93	22.3
6	279.7	31.1	50	0.48	3.46	1.93	22.3
7	85.8	9.5	50	0.57	4.55	0.57	19.5
8	248.3	27.6	40	1.23	2.93	1.23	40.6
9	348.4	38.7	50	1.15	10.47	1.15	31.4
10	188.0	20.9	50	0.53	5.34	0.53	28.3

Table II

Amount of Isoprene Monomer Polymerized

Polymer	Mass Isoprene (g)
1	186.8
2	189.3
3	190.5
4	168.9
5	168.9
6	168.9
7	190.5
8	172.4
9	161.3
10	179.1

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Table III

Molecular Description of Polymers

Polymer	Arms					
	Styrene			Coupled		
	MW	Mw	Content	MW	to Star	MW
1	7900	3290	0.217	8.72	44900	95
2	14100	7280	0.054	5.46	132500	91
3	19900	2610	0.169	4.56	115900	92
4	13900	8190	0.238	17.0	46500	96
5	15200	7770	0.221	15.6	51000	94
6	13200	8300	0.202	16.2	47900	94
7	11800	5460	0.354	4.85	151400	93
8	8190	6500	0.537	14.8	42600	91
9	20700	2630	0.433	20.3	41100	93
10	20300	9600	0.461	21.5	53100	88

C L A I M S

1. A process for preparing a coupled asymmetric block copolymer of a vinyl aromatic hydrocarbon and a conjugated diene which comprises:
 - (a) adding sufficient anionic polymerization initiator to the vinyl aromatic hydrocarbon monomer to start polymerization of the vinyl aromatic hydrocarbon,
 - (b) polymerizing the vinyl aromatic hydrocarbon for a time sufficient to form a first set of vinyl aromatic hydrocarbon blocks having a desired molecular weight,
 - 10 (c) adding sufficient anionic polymerization initiator to start polymerization of a second set of vinyl aromatic hydrocarbon blocks,
 - (d) completing the polymerization of the vinyl aromatic hydrocarbon for a time such that the first set of vinyl aromatic hydrocarbon blocks continues to grow to the desired molecular weight and the second set of vinyl aromatic hydrocarbon blocks grows to a desired smaller molecular weight,
 - 15 (e) adding a conjugated diene and polymerizing it on the ends of the vinyl aromatic hydrocarbon blocks for a time sufficient to form vinyl aromatic hydrocarbon-conjugated diene arms of the desired molecular weight, and
 - (f) adding a coupling agent and coupling the arms to form the asymmetric block copolymer.
2. The process of claim 1 wherein the vinyl aromatic hydrocarbon is styrene and the conjugated diene is selected from the group consisting of butadiene and isoprene.
3. The process of claim 1 wherein the asymmetric block copolymers are asymmetric radial block copolymers.
4. The process of claim 1 wherein the asymmetric block copolymers are asymmetric star block copolymers.

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5. The process of claim 1 wherein the asymmetric block copolymers are asymmetric linear block copolymers.

INTERNATIONAL SEARCH REPORT

International application No.

PCT/EP 94/03037

A. CLASSIFICATION OF SUBJECT MATTER

IPC6: C08F 297/04, C08G 81/02

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC6: C08F, C08G

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practicable, search terms used)

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	EP, A2, 0314256 (SHELL INTERNATIONALE RESEARCH MAATSCHAPPIJ B.V.), 3 May 1989 (03.05.89) ---	1-5
A	EP, A2, 0316671 (BASF AKTIENGESELLSCHAFT), 24 May 1989 (24.05.89) ---	1-5
A	EP, A2, 0362514 (HÜLS AKTIENGESELLSCHAFT PATENTABTEILUNG), 11 April 1990 (11.04.90) ---	1-5
A	EP, A1, 0540109 (SHELL INTERNATIONALE RESEARCH MAATSCHAPPIJ B.V.), 5 May 1993 (05.05.93) -----	1-5

 Further documents are listed in the continuation of Box C. See patent family annex.

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Date of the actual completion of the international search

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Information on patent family members

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International application No.
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Patent document cited in search report	Publication date	Patent family member(s)		Publication date
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		AU-A-	4259189	12/04/90
		DE-A-	3833760	12/04/90
		DE-D-	58907499	00/00/00
		ES-T-	2051940	01/07/94
		JP-A-	2142811	31/05/90
		US-A-	5147939	15/09/92
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		JP-A-	5214013	24/08/93
		US-A-	5212249	18/05/93