

(12) STANDARD PATENT
(19) AUSTRALIAN PATENT OFFICE

(11) Application No. **AU 2009207014 B2**

(54) Title
Intermediate film for laminated glass and laminated glass

(51) International Patent Classification(s)
C03C 27/12 (2006.01) **C07D 209/24** (2006.01)

(21) Application No: **2009207014** (22) Date of Filing: **2009.01.22**

(87) WIPO No: **WO09/093655**

(30) Priority Data

(31) Number	(32) Date	(33) Country
2008-012819	2008.01.23	JP
2008-228027	2008.09.05	JP

(43) Publication Date: **2009.07.30**

(44) Accepted Journal Date: **2013.10.31**

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(56) Related Art
JP 2004/227843
EP 477844
US 6903152
US 2006/0110593

(19) 世界知的所有権機関
国際事務局



(43) 国際公開日
2009年7月30日 (30.07.2009)

PCT

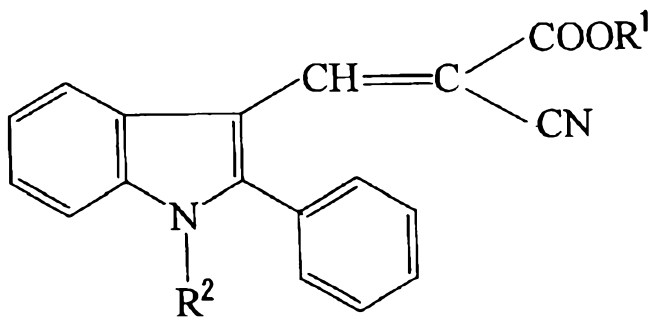
(10) 国際公開番号
WO 2009/093655 A1

- (51) 国際特許分類:
C03C 27/12 (2006.01) C07D 209/24 (2006.01)
- (21) 国際出願番号: PCT/JP2009/050975
- (22) 国際出願日: 2009年1月22日 (22.01.2009)
- (25) 国際出願の言語: 日本語
- (26) 国際公開の言語: 日本語
- (30) 優先権データ:
特願2008-012819 2008年1月23日 (23.01.2008) JP
特願2008-228027 2008年9月5日 (05.09.2008) JP
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- (81) 指定国 (表示のない限り、全ての種類の国内保護が可能): AE, AG, AL, AM, AO, AT, AU, AZ, BA, BB, BG, BH, BR, BW, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DO, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, GT, HN, HR, HU, ID, IL, IN, IS, JP, KE, KG, KM, KN, KP, KR, KZ, LA, LC, LK, LR, LS, LT, LU, LY, MA, MD, ME, MG, MK, MN, MW, MX, MY, MZ, NA, NG, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RS, RU, SC, SD, SE,

[続葉有]

(54) Title: INTERMEDIATE FILM FOR LAMINATED GLASS AND LAMINATED GLASS

(54) 発明の名称: 合わせガラス用中間膜、及び、合わせガラス



(1)

(57) Abstract: Disclosed is an intermediate film for laminated glasses, which is reduced in transmittance for ultraviolet light within a wavelength range of 380-400 nm, while maintaining high visible light transmittance. The intermediate film for laminated glasses has excellent light resistance. Specifically disclosed is an intermediate film for laminated glasses, which is characterized by containing a thermoplastic resin and an indole compound having a structure represented by general formula (1). [In the general formula (1), R¹ represents an alkyl group having 1-3 carbon atoms; and R² represents a hydrogen, an alkyl group having 1-10 carbon atoms or an aralkyl group having 7-10 carbon atoms.]

(57) 要約: 本発明は、高い可視光線透過率を保持しながら、380~400nmの波長域の紫外線透過率を低下させることができ、耐光性に優れる合わせガラス用中間膜を提供することを目的とする。本発明は、熱可塑性樹脂と、下記一般式(1)で表される構造を有するインドール化合物とを含有することを特徴とする合わせガラス用中間膜。一般式(1)中、R¹は、炭素数が1~3のアルキル基を表し、R²は、水素、炭素数が1~10のアルキル基、又は、炭素数が7~10のアラルキル基を表す。

WO 2009/093655 A1



SG, SK, SL, SM, ST, SV, SY, TJ, TM, TN, TR, TT, TZ,
UA, UG, US, UZ, VC, VN, ZA, ZM, ZW.

CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HR, HU,
IE, IS, IT, LT, LU, LV, MC, MK, MT, NL, NO, PL, PT, RO,
SE, SI, SK, TR), OAPI (BF, BJ, CF, CG, CI, CM, GA, GN,
GQ, GW, ML, MR, NE, SN, TD, TG).

(84) 指定国 (表示のない限り、全ての種類の広域保護が可
能): ARIPO (BW, GH, GM, KE, LS, MW, MZ, NA, SD,
SL, SZ, TZ, UG, ZM, ZW), ユーラシア (AM, AZ, BY,
KG, KZ, MD, RU, TJ, TM), ヨーロッパ (AT, BE, BG,

添付公開書類:
— 国際調査報告書

DESCRIPTION

INTERLAYER FILM FOR LAMINATED GLASS AND LAMINATED GLASS

5 TECHNICAL FIELD

[0001]

The present invention relates to an interlayer film for a laminated glass which can reduce transmittance of ultraviolet rays having wavelength of 380 to 400 nm and
10 has excellent durability to light exposure while maintaining high visible light transmittance.

BACKGROUND ART

[0002]

15 A laminated glass is a safety glass because few glass fragments are scattered even if it is broken by impact from the outside. Therefore, laminated glasses have been used widely for windowpanes of motor vehicles such as automobiles, aircrafts, buildings, and the like.
20 Examples of laminated glasses include glasses obtained by inserting, between at least one pair of glasses, an interlayer film for a laminated glass which contains polyvinyl acetal plasticized by a plasticizer, and then by uniting them, and then by laminating them.

25 [0003]

When laminated glasses are used as windowpanes of motor vehicles such as automobiles, aircrafts, and buildings, they are used under a condition exposed to irradiation of ultraviolet rays. A conventional
30 interlayer film for a laminated glass contains ultraviolet absorbers so as to block ultraviolet rays. Most ultraviolet absorbers contained in interlayer films for laminated glasses block only ultraviolet rays having wavelength of 380 nm or shorter. Therefore, an interlayer
35 film for a laminated glass containing the ultraviolet

absorbers cannot sufficiently block ultraviolet rays having wavelength of 380 to 400 nm.

[0004]

As an example of an interlayer film for a laminated glass to overcome the problem mentioned above, Patent Document 1 discloses an interlayer film for a laminated glass containing a synthetic resin, an ultraviolet absorber, and a yellow dye which absorbs light having wavelength of 380 to 450 nm. It is said that the interlayer film for a laminated glass disclosed in Patent Document 1 can block light having wavelength of 450 nm or shorter while maintaining lighting-transmitting properties. However, Patent Document 1 does not consider any method for homogeneously dispersing the yellow dye in the interlayer film for a laminated glass, and thus it is not possible to obtain an interlayer film for a laminated glass having high visible light transmittance. And also, the interlayer film for a laminated glass disclosed in Patent Document 1 has a problem that practically it cannot sufficiently block ultraviolet rays having wavelength of 380 to 400 nm.

[0005]

And also, Patent Document 2 discloses an interlayer film containing a synthetic resin material with an organic light absorber added thereto. As examples of the organic light absorbers, ultraviolet absorbers, blue light absorbers, infrared absorbers, and red light absorbers are described. However, the organic light absorbers described in Patent Document 2 cannot sufficiently block ultraviolet rays having wavelength of 380 to 400 nm.

Patent Document 1: Japanese Kokai Publication 2000-300149 (JP-A-2000-300149)

Patent Document 2: Japanese Kokai Publication 2007-290923 (JP-A-2007-290923)

DISCLOSURE OF THE INVENTION

PROBLEMS TO BE SOLVED BY THE INVENTION

[0006]

The present invention aims to provide an interlayer
 5 film for a laminated glass which is capable of reducing
 transmittance of ultraviolet rays having wavelength of 380
 to 400 nm, and which has excellent durability to light
 exposure while maintaining high visible light
 transmittance.

10

MEANS FOR SOLVING THE PROBLEMS

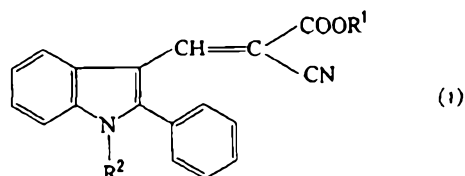
[0007]

The present invention provides an interlayer film
 for a laminated glass, which contains a thermoplastic
 15 resin and an indole compound having a structure
 represented by the following general Chemical Formula (1).

[0008]

[Chemical Formula 1]

20



25 [0009]

In the general Chemical Formula (1), R¹ represents
 an alkyl group having 1 to 3 of carbon atoms, and R²
 represents hydrogen, an alkyl group having 1 to 10 of
 carbon atoms, or an aralkyl group having 7 to 10 of carbon
 30 atoms.

The following description will discuss details of
 the present invention.

[0010]

The present inventors have found that, when an
 35 interlayer film for a laminated glass contains an indole

compound having a specific structure, the interlayer film can reduce transmittance of ultraviolet rays having wavelength of 380 to 400 nm while maintaining high visible light transmittance.

5 Laminated glasses obtained by using interlayer films containing indole compounds have a problem that, when the laminated glasses are used under exposure to sunlight, the transmittance of ultraviolet rays will increase with the passing of time, and the hue of the laminated glasses will
10 change. The present inventors have found that the number of carbon atoms in the R^1 in the general Chemical Formula (1) is related to durability of the laminated glass to light exposure and that high durability to light exposure can be achieved by setting the number of carbon atoms in
15 the R^1 within a certain range. In this way, the present inventors completed the present invention.

[0011]

The interlayer film for a laminated glass of the present invention contains an indole compound having a
20 structure represented by the general Chemical Formula (1). The interlayer film for a laminated glass of the present invention has high visible light transmittance and low transmittance of ultraviolet rays having wavelength of 380 to 400 nm because the interlayer film contains an indole
25 compound having a structure represented by the general Chemical Formula (1).

[0012]

In the general Chemical Formula (1), R^1 represents an alkyl group having 1 to 3 of carbon atoms. Examples of
30 the R^1 include methyl group, ethyl group, isopropyl group, n-propyl group and the like. Among them, methyl group, ethyl group or isopropyl group are preferable, and methyl group or ethyl group are more preferable.

In the general Chemical Formula (1), R^2 represents
35 hydrogen, an alkyl group having 1 to 10 of carbon atoms,

or an aralkyl group having 7 to 10 of carbon atoms. Among them, the alkyl group having 1 to 10 of carbon atoms is preferable, and an alkyl group having 1 to 8 of carbon atoms is more preferable. Examples of the alkyl group
5 having 1 to 10 of carbon atoms include methyl group, ethyl group, isopropyl group, n-propyl group, isobutyl group, n-butyl group, pentyl group, hexyl group, 2-ethylhexyl group, n-octyl group, and the like. Examples of the aralkyl
10 group having 7 to 10 of carbon atoms include benzyl group, phenylethyl group, phenylpropyl group, phenylbutyl group, and the like. The alkyl group may be an alkyl group having a linear main chain or a branched main chain.
[0013]

The number of carbon atoms in R^1 in the general
15 Chemical Formula (1) has a significant influence on durability of a laminated glass to light exposure. The fewer is the number of carbon atoms in R^1 , the higher is durability of the laminated glass to light exposure. And when the number of carbon atoms in R^1 is 1, durability to
20 light exposure is the highest. R^1 includes at most three carbon atoms. If the number of carbon atoms in R^1 is greater or equal to four, ultraviolet transmittance will increase with the passing of time, or the hue of the laminated glass will change when the laminated glass is
25 used under exposure to sunlight.
[0014]

In the interlayer film for a laminated glass of the present invention, the preferable content of the indole compound having a structure represented by the general
30 Chemical Formula (1) depends on the thickness of the interlayer film for a laminated glass. The preferable lower limit of the content of the indole compound having a structure represented by the general Chemical Formula (1) is 0.030 weight%, and the preferable upper limit is 0.145
35 weight% in the interlayer film. If the content of the

indole compound having a structure represented by the general Chemical Formula (1) is 0.030 to 0.145 weight% in the interlayer film, the interlayer film for a laminated glass, which can reduce transmittance of ultraviolet rays having wavelength of 380 to 400 nm while maintaining high visible light transmittance, can be obtained. If the content of the indole compound having a structure represented by the general Chemical Formula (1) is 0.030 to 0.145 weight% in the interlayer film having a thickness of 760 μm , the interlayer film for a laminated glass, which can show particularly excellent effects, can be obtained.

[0015]

The interlayer film for a laminated glass of the present invention contains a thermoplastic resin.

The thermoplastic resin is not limited to any particular resins. For examples, a polyvinyl acetal resin, an ethylene-vinyl acetate copolymer resin, an ethylene-acrylic copolymer resin, a polyurethane resin, a sulfur-containing polyurethane resin, a polyvinyl alcohol resin, and the like can be used. Among them, a polyvinyl acetal resin is preferable, because when mixed and formed together with a plasticizer, an interlayer film for a laminated glass capable of exerting excellent adhesion to the glass can be obtained.

[0016]

The polyvinyl acetal resin is not particularly limited as long as it is a polyvinyl acetal resin obtainable by acetalizing polyvinyl alcohol with aldehyde, and a polyvinyl butyral resin is suitable. And if necessary, two or more kinds of polyvinyl acetal resins may be used together.

The preferable lower limit of an acetalization degree of the polyvinyl acetal resin is 40 mol%, and the preferable upper limit is 85 mol%. And the more preferable

lower limit is 60 mol%. And the more preferable upper limit is 75 mol%.

[0017]

In the case where a polyvinyl butyral resin is used
5 as the polyvinyl acetal resin, the preferable minimum
amount of the hydroxyl group is 15 mol%. And the
preferable maximum amount of the hydroxyl group is 35 mol%.
If the amount of the hydroxyl group is less than 15 mol%,
adhesion between the interlayer film for a laminated glass
10 and the glass may be deteriorated or the penetration
resistance of the laminated glass to be obtained may be
reduced. If the amount of the hydroxyl group exceeds 35
mol%, the interlayer film for a laminated glass to be
obtained may become too hard.

15 [0018]

The polyvinyl acetal resin can be prepared by
acetalizing polyvinyl alcohol with aldehyde.

The polyvinyl alcohol can be normally obtained by
saponifying polyvinyl acetate. Generally, polyvinyl
20 alcohol having a saponification degree of 80 to 99.8 mol%
is used.

The polymerization degree of the polyvinyl alcohol
is preferably 500 in the lower limit and 4000 in the upper
limit. If the polymerization degree of the polyvinyl
25 alcohol is less than 500, the penetration resistance of
the laminated glass to be obtained may be reduced. If the
polymerization degree of the polyvinyl alcohol is more
than 4000, the moldability of the interlayer film for a
laminated glass may be deteriorated. A polymerization
30 degree of the polyvinyl alcohol is more preferably 1000 in
the lower limit and more preferably 3600 in the upper
limit.

[0019]

The aldehyde is not particularly limited, and
35 generally, aldehyde having 1 to 10 of carbon atoms is

suitably used. The aldehyde having 1 to 10 of carbon atoms is not particularly limited and examples thereof include n-butyl aldehyde, iso-butyl aldehyde, n-valeraldehyde, 2-ethylbutyl aldehyde, n-hexyl aldehyde, n-octyl aldehyde, n-nonyl aldehyde, n-decyl aldehyde, formaldehyde, acetaldehyde, benzaldehyde and the like. Out of those examples, n-butyl aldehyde, n-hexyl aldehyde and n-valeraldehyde are preferable, and n-butyl aldehyde is more preferable. These aldehydes may be used alone or in combination of two or more kinds.

[0020]

The interlayer film for a laminated glass of the present invention may contain a plasticizer.

The plasticizer is not particularly limited, and examples thereof include organic ester plasticizers such as monobasic organic acid ester and polybasic organic acid ester, phosphoric acid plasticizers such as organic phosphate plasticizer and organic phosphorous acid type plasticizer, and the like. It is preferable that the plasticizer is a liquid plasticizer.

[0021]

The monobasic organic acid ester is not particularly limited, and examples thereof include glycol type esters obtained by a reaction of glycol such as triethylene glycol, tetraethylene glycol or tripropylene glycol with monobasic organic acid such as butyric acid, isobutyric acid, caproic acid, 2-ethylbutyric acid, heptylic acid, n-octylic acid, 2-ethylhexylic acid, pelargonic acid (n-nonylic acid) or decylic acid. Out of those examples, triethylene glycol dicaproate, triethylene glycol di-2-ethylbutyrate, triethylene glycol di-n-octylate and triethylene glycol di-2-ethylhexylate, and the like are preferable.

[0022]

The polybasic organic acid ester is not particularly

limited, and examples thereof include ester compounds of polybasic organic acids such as adipic acid, sebacic acid and azelaic acid and straight-chain or branched alcohols having 4 to 8 of carbon atoms. Out of those examples,
 5 dibutyl sebacate, dioctyl azelate, dibutyl carbitol adipate and the like are preferable.

[0023]

The organic ester plasticizer is not particularly limited, and examples thereof include triethylene glycol
 10 di-2-ethylbutylate, triethylene glycol di-2-ethylhexanoate, triethylene glycol dicaprylate, triethylene glycol di-n-octanoate, triethylene glycol di-n-heptanoate, tetraethylene glycol di-n-heptanoate, tetraethylene glycol di-2-ethyl hexanoate, dibutyl sebacate, dioctyl azelate,
 15 dibutyl carbitol adipate, ethylene glycol di-2-ethylbutylate, 1,3-propylene glycol di-2-ethylbutylate, 1,4-butylene glycol di-2-ethylbutylate, 1,2-butylene glycol di-2-ethylenebutylate, diethylene glycol di-2-ethylbutylate, diethylene glycol di-2-ethyl hexanoate,
 20 dipropylene glycol di-2-ethylbutylate, triethylene glycol di-2-ethylpentanoate, tetraethylene glycol di-2-ethylbutylate, diethylene glycol dicaprylate, triethylene glycol di-n-heptanoate, tetraethylene glycol di-n-heptanoate, triethylene glycol di-2-ethylbutylate,
 25 triethylene glycol bis(2-ethylbutylate), triethylene glycol di-(2-ethyl hexanoate), triethylene glycol di-heptanoate, tetraethylene glycol di-heptanoate, dihexyl adipate, dioctyl adipate, hexylcyclohexyl adipate, diisononyl adipate, heptylnonyl adipate, dibutyl sebacate,
 30 oil-modified sebacic acid alkyd, a mixture of phosphate ester and adipate ester, a mixed adipate ester prepared by using adipate ester, alkyl alcohol having 4 to 9 of carbon atoms and cyclic alcohol having 4 to 9 of carbon atoms, and adipate ester having 6 to 8 of carbon atoms such as
 35 hexyl adipate.

[0024]

The organic phosphate plasticizer is not particularly limited, and examples thereof include tributoxyethyl phosphate, isodecylphenyl phosphate, 5 triisopropyl phosphate and the like.

[0025]

At least one kind selected from the group consisting of dihexyl adipate (DHA), triethylene glycol di-2-ethylhexanoate (3GO), tetraethylene glycol di-2-10 ethylhexanoate (4GO), triethylene glycol di-2-ethylbutylate (3GH), tetraethylene glycol di-2-ethylbutylate (4GH), tetraethylene glycol di-heptanoate (4G7), and triethylene glycol di-heptanoate (3G7) among the aforementioned plasticizer can prevent the time course 15 change in adhesive force between the interlayer film for a laminated glass and the glass when a metal salt of carboxylic acid having 5 or 6 of carbon atoms is included as an adhesion force controlling agent.

[0026]

20 The plasticizer is preferably selected from triethylene glycol di-2-ethylhexanoate (3GO), triethylene glycol di-2-ethylbutylate (3GH), tetraethylene glycol di-2-ethylhexanoate (4GO), and dihexyl adipate (DHA), because they are hardly hydrolyzed. The plasticizer is more 25 preferably tetraethylene glycol di-2-ethylhexanoate (4GO) or triethylene glycol di-2-ethylhexanoate (3GO) and particularly preferably triethylene glycol di-2-ethylhexanoate.

[0027]

30 The content of the plasticizer in the interlayer film for a laminated glass is not particularly limited. The preferable lower limit of the content of the plasticizer in the interlayer film for a laminated glass is 30 parts by weight relative to 100 parts by weight of 35 the thermoplastic resin. And the preferable upper limit is

70 parts by weight. If the content of the plasticizer is less than 30 parts by weight, the viscosity of a melted interlayer film for a laminated glass to be obtained is so high that deaeration properties upon producing the
5 laminated glass may be deteriorated. If the content of the plasticizer exceeds 70 parts by weight, some separation of the plasticizer from the interlayer film for a laminated glass may occur. The more preferable lower limit of the content of the plasticizer is 35 parts by
10 weight. And the more preferable upper limit of the content of the plasticizer is 63 parts by weight.

[0028]

In the interlayer film for a laminated glass of the present invention, additives such as ultraviolet absorbers,
15 antioxidants, light stabilizers, flame retardants, antistatic agents, adhesive strength controlling agents, moisture resistant agents, blue pigments, blue dyes, green pigments, green dyes, fluorescent whitening agents, and infrared absorbers may be contained.

20 Examples of the ultraviolet absorbers include a compound having benzotriazol structure and the like.

[0029]

The infrared absorbers are not particularly limited as long as they have property to block infrared rays. The
25 infrared absorbers are preferably at least one kind selected from the group consisting of tin-doped indium oxide particles, antimon-doped tin oxide particles, zinc oxide particles doped with elements other than zinc, hexaboride lanthanum particles, zinc antimonate particles,
30 and an infrared absorber having a phthalocyanine structure.

[0030]

The content of the infrared absorber is not particularly limited. The preferable lower limit of the content of the infrared absorber is 0.001 parts by weight
35 relative to 100 parts by weight of the thermoplastic resin.

And the preferable upper limit is 5 parts by weight. If the content of the infrared absorber is less than 0.001 parts by weight, the interlayer film for a laminated glass may not be able to block infrared rays. If the content of the infrared rays exceeds 5 parts by weight, transparency of the laminated glass may be deteriorated.

[0031]

The interlayer film for a laminated glass of the present invention preferably has a thickness of 0.1 mm in the lower limit and a thickness of 3 mm in the upper limit. If the thickness of the interlayer film for a laminated glass is less than 0.1 mm, the penetration resistance of the laminated glass to be obtained may be reduced. If the thickness of the interlayer film for a laminated glass exceeds 3 mm, the transparency of the interlayer film for a laminated glass to be obtained may be deteriorated. The thickness of the interlayer film for a laminated glass is more preferably 0.25 mm in the lower limit and 1.5 mm in the upper limit.

[0032]

The interlayer film for a laminated glass of the present invention may be an interlayer film having a single layer structure of only one resin layer. And also, the interlayer film for a laminated glass of the present invention may be an interlayer film having a multilayer laminated structure including at least two resin layers, provided that at least one of the resin layers is an indole compound-containing resin layer which contains a thermoplastic resin and an indole compound having a structure represented by the general Chemical Formula (1).

[0033]

When the interlayer film for a laminated glass of the present invention has the multilayer laminated structure, it is preferable that at least the outermost layer contains the aforementioned thermoplastic resin, the

aforementioned indole compound having a structure represented by the general Chemical Formula (1), and the aforementioned plasticizer. The thermoplastic resin contained in the outermost layer is preferably a polyvinyl acetal resin and more preferably a polyvinyl butyral resin.
5 [0034]

When the interlayer film for a laminated glass of the present invention has the multilayer laminated structure in which at least the outmost layer contains the
10 aforementioned thermoplastic resin, the aforementioned indole compound having a structure represented by the general Chemical Formula (1), and the aforementioned plasticizer, it is preferable that an intermediate layer preferably contains the thermoplastic resin and the
15 plasticizer.

Moreover, the intermediate layer preferably contains the indole compound having a structure represented by the general Chemical Formula (1). The thermoplastic resin contained in the outermost layer and the intermediate
20 layer is preferably a polyvinyl acetal resin and more preferably a polyvinyl butyral resin. When the thermoplastic resin is a polyvinyl acetal resin, the polyvinyl acetal resin contained in the intermediate layer preferably has a lower amount of the hydroxyl group than
25 that in the polyvinyl acetal resin contained in the outermost layer. An example of the polyvinyl acetal resin contained in the intermediate layer includes a polyvinyl butyral resin containing 15 to 25 mol% of the amount of the hydroxyl group with an acetylation degree of 8 to 15
30 mol% and a butyralation degree of 60 to 71 mol%.
[0035]

The interlayer film for a laminated glass of the present invention having a multilayer laminated structure can reduce transmittance of ultraviolet rays having
35 wavelength of 380 to 400 nm and can show excellent

durability to light exposure and excellent sound insulation properties, while maintaining high visible light transmittance.

[0036]

5 The interlayer film for a laminated glass of the present invention preferably has at least 60% of visible light transmittance (Tv) when measured by a method according to JIS R 3106 in which the interlayer film having a thickness of 760 μm is sandwiched between two
10 sheets of clear glasses having a thickness of 2.5 mm. When visible light transmittance (Tv) is less than 60%, the transparency of the laminated glass obtained by using the interlayer film for a laminated glass of the present invention may be deteriorated. The visible light
15 transmittance (Tv) is preferably 70% or more, more preferably 75% or more, and particularly preferably 80% or more.

 The apparatus for measuring the visible light transmittance (Tv) is not particularly limited, and a
20 spectrophotometer (U-4000, manufactured by Hitachi, Ltd.) and the like may be exemplified.

[0037]

 The interlayer film for a laminated glass of the present invention can reduce the transmittance of
25 ultraviolet rays having wavelength of 380 to 400 nm. The transmittance of ultraviolet rays having wavelength of 380 to 400 nm is preferably 2% or less, more preferably 1% or less, and particularly preferably 0.5% or less. The transmittance of ultraviolet rays having wavelength of 380
30 to 400 nm can be obtained by measuring the transmittances of ultraviolet rays at wavelength of 380 nm, 390 nm and 400 nm, and then by summing up the obtained values, and then by calculating the average value.

 The apparatus for measuring the transmittance of
35 ultraviolet rays having wavelength of 380 to 400 nm is not

particularly limited, and a spectrophotometer (U-4000, manufactured by Hitachi, Ltd.) and the like may be exemplified.

[0038]

5 As a method of producing the interlayer film for a laminated glass of the present invention, an example of the method is a method as follows: first, the plasticizer is mixed with a indole compound having a structure represented by the general Chemical Formula (1) to prepare
10 a composition; the composition to which additives are added as needed is sufficiently kneaded with a thermoplastic resin such as a polyvinyl acetal resin; and the kneaded mixture is molded into an interlayer film for a laminated glass.

15 It is preferable that the production method includes a process to prepare a composition including the plasticizer and the indole compound having a structure represented by the general Chemical Formula (1) dissolved in the plasticizer, and a process to knead the composition
20 with a thermosetting resin such as a polyvinyl acetal resin.

 In the process for preparing the composition in which the indole compound having a structure represented by the general Chemical Formula (1) is dissolved in the
25 plasticizer, it is preferable that the composition is heated to dissolve the indole compound.

[0039]

 The method for kneading the composition and the thermoplastic resin is not particularly limited, and an
30 example of the method includes a method using an extruder, a kneader, a Banbury mixer, calendaring roll and the like. In the example, a method using an extruder is preferable, and a method using a two-axis extruder is more preferable because they are suitable for continuous production.

35 [0040]

The interlayer film for a laminated glass of the present invention can reduce the transmittance of ultraviolet rays having wavelength of 380 to 400 nm while maintaining high visible light transmittance when the indole compound is contained therein. Further, the interlayer film for a laminated glass of the present invention has excellent durability to light exposure.

[0041]

Moreover, the interlayer film for a laminated glass of the present invention can reduce the number of bubble under high temperature environment.

With regard to a laminated glass in which a interlayer film containing a relatively large amount of a plasticizer is used, there is a problem that generation of bubble tends to easily occur when the laminated glass is put under high temperature environment of 80 to 150°C. In this description, containing a relatively large amount of a plasticizer refers to that the content of the plasticizer is in the range from 50 parts by weight to 70 parts by weight relative to 100 parts by weight of the thermoplastic resin.

Moreover, in the case where a polyvinyl acetal resin having a low amount of the hydroxyl group is used as thermoplastic resin, generation of bubble tends to easily occur when the laminated glass is put under high temperature environment of 80 to 150°C even though the content of the plasticizer is less than 50 parts by weight relative to 100 parts by weight of the thermoplastic resin. Especially, generation of bubble tends to easily occur when water content in a plasticizer is high.

Also, when the interlayer film for a laminated glass of the present invention has a single layer structure, it is possible to effectively prevent generation of bubble under high temperature environment even in the case where the content of the plasticizer is 50 to 70 parts by weight

relative to 100 parts by weight of the thermoplastic resin. Similarly, when the interlayer film for a laminated glass of the present invention has a single layer structure, it is possible to effectively prevent generation of bubble
5 under high temperature environment even in the case where a polyvinyl acetal resin having a low amount of the hydroxyl group is used as thermoplastic resin. Similarly, when the interlayer film for a laminated glass of the present invention has a single layer structure, it is
10 possible to effectively prevent generation of bubble under high temperature environment even in the case where water content in a plasticizer is high. One possible reason of this effect is that the interlayer film for a laminated glass containing the indole compound contacts with the
15 glass.

It is to be noted that the polyvinyl acetal resin having a low amount of the hydroxyl group refers to a polyvinyl acetal resin containing 25 mol% or less of the hydroxyl group. An example of the polyvinyl acetal resin
20 having a low amount of the hydroxyl group includes a polyvinyl butyral resin containing 15 to 25 mol% of the hydroxyl group with an acetylation degree of 8 to 15 mol% and a butyralation degree of 60 to 71 mol%.

[0042]

25 And also, in the case where the interlayer film for a laminated glass of the present invention has a multilayer structure, the number of bubble under high temperature environment can be reduced when the outermost layer of the interlayer film is the resin layer containing
30 the indole compound.

For example, with regard to the interlayer film having a structure in which an interlayer containing the plasticizer in an amount of not less than 50 parts by weight relative to 100 parts by weight of the
35 thermoplastic resin is sandwiched between the two

outermost layers containing the plasticizer in an amount of 35 to 49 parts by weight, in the case where the outermost layers contain the indole compound having a structure represented by the general Chemical Formula (1),
5 generation of bubble can be prevented even when the laminated glass is put under high temperature environment of 80 to 150°C.

[0043]

When the interlayer film for a laminated glass of
10 the present invention is sandwiched between two sheets of glasses and then laminated, the resulting product can be used as a laminated glass. The glass to be used for the laminated glass is not particularly limited and may be a generally used clear glass plate. Examples of the glass
15 include inorganic glass such as float plate glass, polished plate glass, figured glass, wired glass, colored plate glass, heat-absorbing glass, heat-reflecting glass, and green glass. Organic plastic plates such as polycarbonate and polyacrylate may also be exemplified.

20 [0044]

Two kinds or more of plate glasses may be used as the glass plate. For example, a laminated glass in which the interlayer film for a laminated glass of the present invention is sandwiched between a transparent float glass
25 plate and a colored glass plate such as a green glass may be exemplified.

When the laminated glass is used as glass for vehicles, it may be used as a windshield, a side glass, a rear glass, a roof glass, or a panorama glass.

30 Furthermore, a production method of the laminated glass is not particularly limited, and a conventionally method can be used.

[0045]

When the interlayer film for a laminated glass of
35 the present invention is used in the laminated glass, the

laminated glass shows a sufficiently low transmittance of ultraviolet rays having wavelength of 380 to 400 nm while maintaining high visible light transmittance. Moreover, the laminated glass can maintain ultraviolet blocking properties for a long time, and the hue thereof hardly changes even under exposure to sunlight.

EFFECTS OF THE INVENTION

[0046]

10 The present invention can provide an interlayer film for a laminated glass which can reduce transmittance of ultraviolet rays having wavelength of 380 to 400 nm and has excellent durability to light exposure while maintaining high visible light transmittance.

15

BEST MODE FOR CARRYING OUT THE INVENTION

[0047]

The following description will discuss details of embodiments of the present invention by showing Examples, but the present invention is not limited to those Examples.

20

[0048]

(Preparation of indole compound)

(1) Preparation of indole compound A

To 120 ml of methanol were added 23.5 g (0.10 mol) of 1-methyl-2-phenyl-1H-indole-3-carbaldehyde and 11.9 g (0.12 mol) of methyl cyanoacetate to produce a mixture. Subsequently, 2.5 g (0.03 mol) of piperidine was added to the mixture, reacted under reflux for 6 hours, and cooled to room temperature to precipitate a crystal. The obtained crystal was washed with a small amount of methanol, and then dried to give 30.9 g of a light-yellow crystalline indole compound A having a structure of the general Chemical Formula (1) in which R¹ was methyl group and R² was methyl group. The melting point of the obtained indole compound A was 193.7°C.

35

[0049]

(2) Preparation of indole compound B

An amount of 28.9 g of a light-yellow crystalline indole compound B having a structure of the general
5 Chemical Formula (1) in which R^1 was ethyl group and R^2 was methyl group was obtained by the same method as the preparation method of the indole compound A, except that ethanol was used instead of methanol and ethyl
10 cyanoacetate (0.12 mol) was used instead of methyl cyanoacetate. The melting point of the obtained indole compound B was 145°C.

[0050]

(3) Preparation of indole compound C

An amount of 32.7 g of a light-yellow crystalline
15 indole compound C having a structure of the general Chemical Formula (1) in which R^1 was isopropyl group and R^2 was methyl group was obtained by the same method as the preparation method of the indole compound A, except that isopropyl alcohol was used instead of methanol and
20 isopropyl cyanoacetate (0.12 mol) was used instead of methyl cyanoacetate. The melting point of the obtained indole compound C was 170.1°C.

[0051]

(4) Preparation of indole compound D

25 An amount of 33.7 g of a light-yellow crystalline indole compound D having a structure of the general Chemical Formula (1) in which R^1 was butyl group and R^2 was methyl group was obtained by the same method as the preparation method of the indole compound A, except that
30 butanol was used instead of methanol and butyl cyanoacetate (0.12 mol) was used instead of methyl cyanoacetate. The melting point of the obtained indole compound D was 126°C.

[0052]

35 (5) Preparation of indole compound E

An amount of 35.0 g of a light-yellow crystalline indole compound E having a structure of the general Chemical Formula (1) in which R¹ was pentyl group and R² was methyl group was obtained by the same method as the preparation method of the indole compound A, except that pentanol was used instead of methanol and pentyl cyanoacetate (0.12 mol) was used instead of methyl cyanoacetate.

[0053]

10 (6) Preparation of indole compound F

An amount of 28.4 g of a light-yellow crystalline indole compound F having a structure of the general Chemical Formula (1) in which R¹ was methyl group and R² was hydrogen was obtained by the same method as the preparation method of the indole compound A, except that 2-phenyl-1H-indole-3-carbaldehyde (0.10 mol) was used instead of 1-methyl-2-phenyl-1H-indole-3-carbaldehyde.

[0054]

(7) Preparation of indole compound G

20 An amount of 31.1 g of a light-yellow crystalline indole compound G having a structure of the general Chemical Formula (1) in which R¹ was methyl group and R² was ethyl group was obtained by the same method as the preparation method of the indole compound A, except that 1-ethyl-2-phenyl-1H-indole-3-carbaldehyde (0.10 mol) was used instead of 1-methyl-2-phenyl-1H-indole-3-carbaldehyde.

[0055]

(8) Preparation of indole compound H

30 An amount of 33.7 g of a light-yellow crystalline indole compound H having a structure of the general Chemical Formula (1) in which R¹ was methyl group and R² was butyl group was obtained by the same method as the preparation method of the indole compound A, except that 1-butyl-2-phenyl-1H-indole-3-carbaldehyde (0.10 mol) was used instead of 1-methyl-2-phenyl-1H-indole-3-carbaldehyde.

35

[0056]

(9) Preparation of indole compound I

An amount of 36.3 g of a light-yellow crystalline indole compound I having a structure of the general
5 Chemical Formula (1) in which R¹ was methyl group and R²
was -CH₂-CH(C₂H₅)-C₄H₉ group was obtained by the same method
as the preparation method of the indole compound A, except
that 1-(2-ethyl-hexyl)-2-phenyl-1H-indole-3-carbaldehyde
(0.10 mol) was used instead of 1-methyl-2-phenyl-1H-
10 indole-3-carbaldehyde.

[0057]

(10) Preparation of indole compound J

An amount of 36.9 g of a light-yellow crystalline indole compound J having a structure of the general
15 Chemical Formula (1) in which R¹ was methyl group and R²
was -CH₂Ph group was obtained by the same method as the
preparation method of the indole compound A, except that
1-benzyl-2-phenyl-1H-indole-3-carbaldehyde (0.10 mol) was
used instead of 1-methyl-2-phenyl-1H-indole-3-carbaldehyde.

20 [0058]

(11) Preparation of indole compound K

An amount of 32.4 g of a light-yellow crystalline indole compound K having a structure of the general
Chemical Formula (1) in which R¹ was butyl group and R² was
25 hydrogen was obtained by the same method as the
preparation method of the indole compound D, except that
2-phenyl-1H-indole-3-carbaldehyde (0.10 mol) was used
instead of 1-methyl-2-phenyl-1H-indole-3-carbaldehyde.

[0059]

30 (12) Preparation of indole compound L

An amount of 35.0 g of a light-yellow crystalline indole compound L having a structure of the general
Chemical Formula (1) in which R¹ was butyl group and R² was
ethyl group was obtained by the same method as the
35 preparation method of the indole compound D, except that

1-ethyl-2-phenyl-1H-indole-3-carbaldehyde (0.10 mol) was used instead of 1-methyl-2-phenyl-1H-indole-3-carbaldehyde. [0060]

(13) Preparation of indole compound M

5 An amount of 37.6 g of a light-yellow crystalline indole compound M having a structure of the general Chemical Formula (1) in which R¹ was butyl group and R² was butyl group was obtained by the same method as the preparation method of the indole compound D, except that
10 1-butyl-2-phenyl-1H-indole-3-carbaldehyde (0.10 mol) was used instead of 1-methyl-2-phenyl-1H-indole-3-carbaldehyde. [0061]

(14) Preparation of indole compound N

15 An amount of 40.2 g of a light-yellow crystalline indole compound N having a structure of the general Chemical Formula (1) in which R¹ was butyl group and R² was -CH₂-CH(C₂H₅)-C₄H₉ group was obtained by the same method as the preparation method of the indole compound D, except that 1-(2-ethyl-hexyl)-2-phenyl-1H-indole-3-carbaldehyde
20 (0.10 mol) was used instead of 1-methyl-2-phenyl-1H-indole-3-carbaldehyde. [0062]

(15) Preparation of indole compound O

25 An amount of 40.8 g of a light-yellow crystalline indole compound O having a structure of the general Chemical Formula (1) in which R¹ was butyl group and R² was -CH₂Ph group was obtained by the same method as the preparation method of the indole compound D, except that 1-benzyl-2-phenyl-1H-indole-3-carbaldehyde (0.10 mol) was
30 used instead of 1-methyl-2-phenyl-1H-indole-3-carbaldehyde. [0063]

(16) Preparation of indole compound P

35 An amount of 29.1 g of a light-yellow crystalline indole compound P having a structure of the general Chemical Formula (1) in which R¹ was hydrogen and R² was

methyl group was obtained by the same method as the preparation method of the indole compound A, except that toluene was used instead of methanol and cyanoacetic acid (0.12 mol) was used instead of methyl cyanoacetate. The
5 melting point of the obtained indole compound P was 203.5°C.

[0064]

(17) Preparation of indole compound Q

To 200 ml of N,N-dimethylformamide (DMF) were added
10 30.0g (0.10 mol) of the indole compound P and 14.4 g of anhydrous potassium carbonate to produce a mixture. Subsequently, 25.6 g (0.20 mol) of 2-bromoethanol was added to the mixture to give a liquid mixture. The obtained liquid mixture was reacted at 70°C for 6 hours,
15 and then cooled to room temperature. Thereafter, 1000 ml of water and 500 ml of ethyl acetate were added to the reaction mixture, and the organic layer was washed with 1000 ml of water three times. Hexane was added to the organic layer to precipitate a crystal. The obtained
20 crystal was dried to give 19.1 g of a light-yellow crystalline indole compound Q having a structure of the general Chemical Formula (1) in which R¹ was -C₂H₄OH group and R² was methyl group. The melting point of the obtained indole compound Q was 145.6°C.

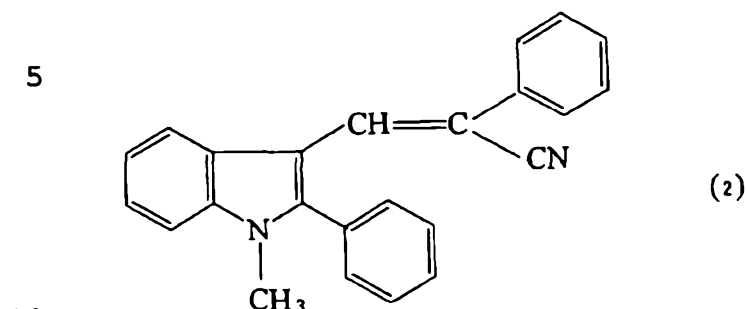
25 [0065]

(18) Preparation of indole compound R

An amount of 29.9 g of a light-yellow crystalline indole compound R having a structure represented by the following Chemical Formula (2) was obtained by the same
30 method as the preparation method of the indole compound A, except that ethanol was used instead of methanol, phenylacetonitrile (0.12 mol) was used instead of methyl cyanoacetate, and 48% solution of potassium hydroxide
5 (0.30 mol) was used instead of piperidine. The melting
35 point of the obtained indole compound R was 183.7°C.

[0066]

[Chemical Formula 2]



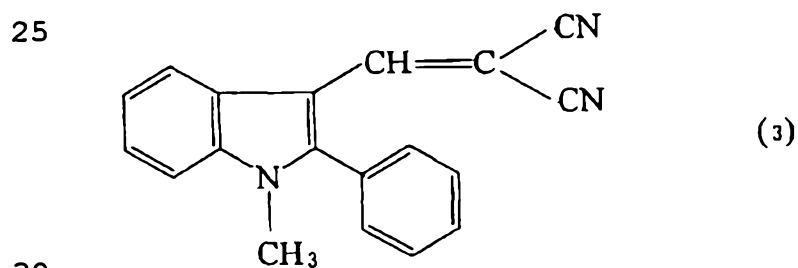
[0067]

(19) Preparation of indole compound S

An amount of 27.5 g of a light-yellow crystalline
 15 indole compound S having a structure represented by the
 following Chemical Formula (3) was obtained by the same
 method as the preparation method of the indole compound A,
 except that ethanol was used instead of methanol,
 malononitrile (0.12 mol) was used instead of methyl
 20 cyanoacetate, and triethylamine (0.03 mol) was used
 instead of piperidine. The melting point of the obtained
 indole compound S was 203.2°C.

[0068]

[Chemical Formula 3]



[0069]

(20) Preparation of indole compound T

An amount of 31.75 g of a light-yellow crystalline
 indole compound T having a structure of the general
 35 Chemical Formula (1) in which R¹ was -C₂H₄OC₂H₅ group and R²

was methyl group was obtained by the same method as the preparation method of the indole compound A, except that 2-ethoxyethanol was used instead of methanol and 2-ethoxyethyl cyanoacetate (0.12 mol) was used instead of methyl cyanoacetate. The melting point of the obtained indole compound T was 127.3°C.

[0070]

Table 1 shows the structure of the prepared indole compounds. It is to be noted that BONASORB UA3901 (produced by Orient Chemical industries, Ltd.) is an indole compound.

[0071]

[Table 1]

Indole compound	R ¹	R ²	Relationship with Chemical Formula (1)
A	-CH ₃	-CH ₃	Compatible
B	-C ₂ H ₅	-CH ₃	Compatible
C	-C ₃ H ₇	-CH ₃	Compatible
D	-C ₄ H ₉	-CH ₃	Incompatible
E	-C ₅ H ₁₁	-CH ₃	Incompatible
F	-CH ₃	-H	Compatible
G	-CH ₃	-C ₂ H ₅	Compatible
H	-CH ₃	-C ₄ H ₉	Compatible
I	-CH ₃	-CH ₂ -CH(C ₂ H ₅)-C ₄ H ₉	Compatible
J	-CH ₃	-CH ₂ -Ph	Compatible
K	-C ₄ H ₉	-H	Incompatible
L	-C ₄ H ₉	-C ₂ H ₅	Incompatible
M	-C ₄ H ₉	-C ₄ H ₉	Incompatible
N	-C ₄ H ₉	-CH ₂ -CH(C ₂ H ₅)-C ₄ H ₉	Incompatible
O	-C ₄ H ₉	-CH ₂ -Ph	Incompatible
P	-H	-CH ₃	Incompatible
Q	-C ₂ H ₄ OH	-CH ₃	Incompatible
R	Chemical Formula (2)		Incompatible
S	Chemical Formula (3)		Incompatible
T	-C ₂ H ₄ OC ₂ H ₅	-CH ₃	Incompatible
BONASORB UA3901	-		Incompatible

15 [0072]

(Example 1)

(1) Production of interlayer film for laminated glass

To 40 parts by weight of triethylene glycol di-2-ethylhexanoate (3GO) as a plasticizer were added 0.4 parts by weight of 2,6-di-t-butyl-p-cresol (BHT) as an antioxidant, 0.4 parts by weight of an ultraviolet absorber ("TINUVIN 326" produced by Ciba Specialty Chemicals Inc.) having a benzotriazol structure, and 0.048 parts by weight of the obtained indole compound A. The resultant mixture was stirred by using a stirrer at 80°C for 30 minutes to give a plasticizer solution.

The obtained plasticizer solution was sufficiently mixed with 100 parts by weight of polyvinyl butyral resin (PVB) (average polymerization degree: 1700, butyralation degree: 68.5 mol%, amount of the hydroxyl group: 30.6 mol%, amount of the acetyl group: 0.9 mol%). Then, a twin-screw aeolotropic extruder was used to produce an interlayer film for a laminated glass having a film thickness of 760 μm . In this process, for the purpose of controlling the adhesion of the interlayer film for a laminated glass, a magnesium acetate solution was added so that concentration of Mg in the interlayer film for a laminated glass was 65 ppm.

[0073]

(2) Production of laminated glass

The obtained interlayer film for a laminated glass was put for 24 hours under the constant temperature and humidity conditions, at a temperature of 23°C and a relative humidity of 28%, and thereafter sandwiched between two sheets of transparent float glass (clear glass: 300 mm in length \times 300 mm in width \times 2.5 mm in thickness) to form a laminated body. The obtained laminated body was temporarily-press-bonded by using a roller heated at 230°C. The temporarily-press-bonded laminated glass was press-bonded using an autoclave, for 20 minutes at a temperature of 135°C and a pressure of 1.2

MPa, to produce a laminated glass. Similarly, another laminated glass was produced by using float glass (clear glass) with 500 mm in length × 500 mm in width × 2.5 mm in thickness.

5 [0074]

(Examples 2 to 27, and Comparative Examples 1 to 15)

Interlayer films for a laminated glass and laminated glasses were produced in the same manner as in Example 1, except that the compositions thereof were changed as shown
10 in Tables 2 to 6. Meanwhile, interlayer films for a laminated glass and laminated glasses were produced in Examples 4, 8, and 12 in the same manner as in Example 1, except that 0.28 parts by weight of tin-doped indium oxide particles (ITO) (volume average particle diameter: 35 nm)
15 were added to the plasticizer solution as an infrared absorber.

[0075]

(Evaluations)

The following evaluations were made for each of the
20 laminated glasses obtained in Examples and Comparative Examples.

Tables 2 to 6 show the results.

[0076]

(1) Evaluation of transmittance

25 The visible light transmittance (T_v) of the obtained laminated glass (300 mm in length × 300 mm in width) was calculated according to JIS R 3106 (1998) by using a spectrophotometer ("U-4000" produced by Hitachi, Ltd.). The solar transmittance (T_s) of the obtained laminated
30 glass was also determined at wavelength of 300 to 2500 nm.

Using the spectrophotometer ("U-4000" produced by a Hitachi, Ltd.), the transmittances of the laminated glass at wavelength of 380 nm, 390 nm, and 400 nm were measured respectively to find an average value T of the
35 transmittance (380 to 400 nm). The average value T of the

transmittance (380 to 400 nm) was determined for the laminated glass before and after the light exposure test (2) to be mentioned later.

[0077]

5 (2) Light exposure test

The obtained laminated glass was exposed with ultraviolet rays for 2000 hours using an ultraviolet irradiation apparatus according to JIS R 3205 (1998), and the change in hue was evaluated with the color difference ΔE before and after ultraviolet irradiation.

[0078]

(3) Evaluation of preventive effect on pest insect attraction

A transparent adhesive was applied to the glass surface of one side of the obtained laminated glass (500 mm in length \times 500 mm in width). A halogen lamp is provided on the glass side to which the transparent adhesive is not applied, and the laminated glass was left to stand outdoors (Sekisui Chemical's Shiga Minakuchi Plant, in August, 2008) for 1 hour (from 20:00 to 21:00) with the white light of the halogen lamp on. The number of winged insects that had been captured to the transparent adhesive after being left standing was calculated.

25 [0079]

(4) Evaluation of penetration resistance

The obtained laminated glass (300 mm in length \times 300 mm in width) was adjusted so as to have a surface temperature of 23°C. Subsequently, according to JIS R 3212, a rigid sphere having a mass of 2260 g and a diameter of 82 mm was dropped from a height of 4 m to the central point of the laminated glass. The same test was made for in total six sheets of laminated glasses. The test result was expressed as "pass" in the case where a rigid sphere did not penetrate through the laminated glass within 5

seconds after the impact of the rigid sphere on the laminated glass for all the six sheets of the laminated glass. Also, the test result was expressed as "fail" in the case where the number of sheets of the laminated glass through which a rigid sphere did not penetrate within 5 seconds after the impact of the rigid sphere was less than or equal to three. In the case where the number of sheets of the laminated glass through which a rigid sphere did not penetrate within 5 seconds after the impact of the rigid sphere was four, a retest was made for another six sheets of the laminated glass. In the case where the number of sheets of the laminated glass through which a rigid sphere did not penetrate within 5 seconds after the impact of the rigid sphere was five, one more sheet of the laminated glass was additionally tested; and the test result was evaluated as "pass" in the case where a rigid sphere did not penetrate through the laminated glass within 5 seconds after impact of the rigid sphere.

The same evaluations of penetration resistance were performed for the fall heights of 5 m and 6 m, respectively, in addition to the test for the fall heights of 4 m.

[0080]

(5) Durability test to heat

The obtained laminated glass (300 mm in length × 300 mm in width) was vertically set under a constant temperature at 100°C for one month. The change in hue was evaluated based on the color difference ΔE before and after the test.

The number of bubble generated after the test was counted. With regard to bubble's size, bubbles were classified into three:

1. The length of its longest side is less than 5 mm
2. The length of its longest side is greater than or equal to 5 mm, and less than 10 mm

3. The length of its longest side is greater than or equal to 10 mm.

Then, the numbers of bubble for each classification were counted.

5 [0081]

(6) Observation of solid components

The interlayer film specimen having a size of 300 mm in length × 300 mm in width was cut from the produced interlayer film and wrapped with an aluminum bag. The wrapped sample was kept for one month at a constant temperature of 5°C. Then, the interlayer film was observed with the naked eye, and it was checked whether there are any solid components or not. The solid components are presumably either an indole compound or an ultraviolet absorber having a benzotriazol structure.

15

[0082]

[Table 2]

		Example										
		1	2	3	4	5	6	7	8	9	10	
Composition (parts by weight)	PVB	100	100	100	100	100	100	100	100	100	100	
	Plasticizer (3GO)	A	40	40	40	40	40	40	40	40	40	40
		B	0.048	0.14	0.2	0.14	-	-	-	-	-	-
		C	-	-	-	-	0.048	0.14	0.2	0.14	-	-
		D	-	-	-	-	-	-	-	-	0.048	0.14
		E	-	-	-	-	-	-	-	-	-	-
	Indole compound	Antioxidant	0.4	0.4	0.4	0.4	0.4	0.4	0.4	0.4	0.4	0.4
		Ultraviolet absorber	0.4	0.4	0.4	0.4	0.4	0.4	0.4	0.4	0.4	0.4
	ITO		0	0	0	0.28	0	0	0	0.28	0	0
	Mg/ppm		65	65	65	65	65	65	65	65	65	65
Concentration of light absorber/weight %		0.034	0.099	0.142	0.099	0.034	0.099	0.142	0.099	0.034	0.099	
Evaluation of transmittance	Transmittance/%	79.0	78.8	78.7	65.6	79.1	78.8	78.2	65.9	78.5	78.9	
	Tv	88.3	88.6	88.4	87.4	88.2	88.5	88.4	87.1	88.4	88.3	
	Before ultraviolet irradiation	0.35	0.04	0.02	0.04	0.46	0.05	0.02	0.03	0.51	0.08	
	After ultraviolet irradiation	0.40	0.16	0.10	0.18	0.50	0.26	0.13	0.18	0.57	0.30	
Light exposure test	ΔE	1.07	1.14	1.30	1.21	1.06	1.22	1.41	1.50	1.19	1.44	
Evaluation of preventive effect on pest insect attraction	The number of attracted winged insects	7	3	5	5	8	2	6	4	8	3	
	4 m	Pass	Pass	Pass	Pass	Pass	Pass	Pass	Pass	Pass	Pass	
Evaluation of penetration resistance	5 m	Pass	Pass	Pass	Pass	Pass	Pass	Pass	Pass	Pass	Pass	
	6 m	Pass	Pass	Pass	Pass	Pass	Pass	Pass	Pass	Pass	Pass	
	ΔE	0.26	0.38	0.55	0.46	0.30	0.39	0.54	0.50	0.31	0.42	
Durability test to heat	The number of bubble less than 5 mm	0	0	0	0	0	0	0	0	0	0	
	The number of bubble greater than or equal to 5 mm, and less than 10 mm	0	0	0	0	0	0	0	0	0	0	
	The number of bubble greater than or equal to 10mm	0	0	0	0	0	0	0	0	0	0	
Presence of solid components		Absent	Absent	Absent	Absent	Absent	Absent	Absent	Absent	Absent	Absent	

[0083]

[Table 3]

		Example		Comparative Example				
		11	12	1	2	3	4	
Composition (parts by weight)	PVB		100	100	100	100	100	100
	Plasticizer (3GO)		40	40	40	40	40	40
	Indole compound	A	-	-	-	-	-	-
		B	-	-	-	-	-	-
		C	0.2	0.14	-	-	-	-
		D	-	-	0.14	-	-	-
		E	-	-	-	0.14	-	-
	Antioxidant		0.4	0.4	0.4	0.4	0.4	0.4
	Ultraviolet absorber		0.4	0.4	0.4	0.4	0.4	0.8
ITO		0	0.28	0	0	0	0	
Mg/ppm		65	65	65	65	65	65	
Concentration of light absorber/ weight %		0.142	0.099	0.099	0.099	0.000	0.000	
Evaluation of transmittance	Transmittance/%	Ts	78.2	67.3	78.6	78.5	79.1	78.9
		Tv	88.3	87.2	88.5	88.3	88.6	88.1
	T(380-400nm)/%	Before ultraviolet irradiation	0.03	0.08	0.06	0.05	10.25	2.49
		After ultraviolet irradiation	0.14	0.30	0.66	0.84	10.17	2.51
Light exposure test	ΔE		1.65	1.48	3.01	3.96	0.30	0.25
Evaluation of preventive effect on pest insect attraction	The number of attracted winged insects		6	4	3	5	27	19
Evaluation of penetration resistance	4 m		Pass	Pass	Pass	Pass	Pass	Pass
	5 m		Pass	Pass	Pass	Pass	Pass	Pass
	6 m		Pass	Pass	Pass	Pass	Pass	Pass
Durability test to heat	ΔE		0.63	0.59	2.17	2.33	0.15	0.13
	The number of bubble less than 5 mm		0	0	0	0	0	0
	The number of bubble greater than or equal to 5 mm, and less than 10 mm		0	0	0	0	0	0
	The number of bubble greater than or equal to 10mm		0	0	0	0	0	0
Presence of solid components			Absent	Absent	Absent	Absent	Absent	Absent

[0084]

[Table 4]

[0085]

[Table 5]

		Comparative Example					
		5	6	7	8	9	
Composition (parts by weight)	PVB		100	100	100	100	100
	Plasticizer (3GO)		40	40	40	40	40
	Indole compound	K	0.14	-	-	-	-
		L	-	0.14	-	-	-
		M	-	-	0.14	-	-
		N	-	-	-	0.14	-
		O	-	-	-	-	0.14
	Antioxidant		0.4	0.4	0.4	0.4	0.4
	Ultraviolet absorber		0.4	0.4	0.4	0.4	0.4
ITO		0	0	0	0	0	
Mg /ppm		65	65	65	65	65	
Concentration of light absorber/weight %		0.099	0.099	0.099	0.099	0.099	
Evaluation of transmittance	Transmittance (%)	Ts	78.8	78.4	78.6	78.8	78.4
		Tv	88.4	88.4	88.5	88.3	88.2
	T(380-400nm)	Before ultraviolet irradiation	0.07	0.06	0.08	0.05	0.05
		After ultraviolet irradiation	0.82	0.75	0.88	0.76	0.92
Light exposure test	ΔE		3.33	3.26	3.10	3.81	3.45
Evaluation of preventive effect on pest insect attraction	The number of attracted winged insects		5	4	6	8	2
Evaluation of penetration resistance	4 m		Pass	Pass	Pass	Pass	Pass
	5 m		Pass	Pass	Pass	Pass	Pass
	6 m		Pass	Pass	Pass	Pass	Pass
Durability test to heat	ΔE		1.96	2.08	1.85	2.17	1.58
	The number of bubble less than 5 mm		0	0	0	0	0
	The number of bubble greater than or equal to 5 mm, and less than 10 mm		0	0	0	0	0
	The number of bubble greater than or equal to 10mm		0	0	0	0	0
Presence of solid components		Absent	Absent	Absent	Absent	Absent	

[0086]

5 [Table 6]

		Comparative Example						
		10	11	12	13	14	15	
Composition (parts by weight)	PVB	100	100	100	100	100	100	
	Plasticizer (3GO)	40	40	40	40	40	40	
	Indole compound	P	0.14	-	-	-	-	-
		Q	-	0.14	-	-	-	-
		R	-	-	0.14	-	-	-
		S	-	-	-	0.14	-	-
		T	-	-	-	-	-	0.14
	BONASORB UA3901	-	-	-	-	0.14	-	
	Antioxidant	0.4	0.4	0.4	0.4	0.4	0.4	
	Ultraviolet absorber	0.4	0.4	0.4	0.4	0.4	0.4	
ITO	0	0	0	0	0	0		
Mg/ppm	65	65	65	65	65	65		
Concentration of light absorber/ weight %		0.099	0.099	0.099	0.099	0.099	0.099	
Evaluation of transmittance	Transmittance/%	Ts	79.0	79.1	78.4	78.1	78.4	78.7
		Tv	88.1	88.6	88.2	87.6	88.3	88.5
	T(380-400nm)/%	Before ultraviolet irradiation	0.30	0.07	0.34	0.03	0.05	0.09
		After ultraviolet irradiation	1.06	0.58	1.09	0.67	0.64	0.52
Light exposure test	ΔE	2.73	3.06	4.13	3.65	3.68	2.87	
Evaluation of preventive effect on pest insect attraction	The number of attracted winged insects	3	2	4	6	3	4	
Evaluation of penetration resistance	4 m	Pass	Pass	Pass	Pass	Pass	Pass	
	5 m	Fail	Fail	Pass	Pass	Pass	Pass	
	6 m	Fail	Fail	Pass	Pass	Pass	Pass	
Durability test to heat	ΔE	1.90	2.34	3.71	3.07	3.22	1.99	
	The number of bubble less than 5 mm	0	0	0	0	0	0	
	The number of bubble greater than or equal to 5 mm, and less than 10 mm	0	0	0	0	0	0	
	The number of bubble greater than or equal to 10 mm	0	0	0	0	0	0	
Presence of solid components		Present	Present	Absent	Absent	Absent	Absent	

[0087]

(Example 28)

(1) Production of interlayer film for laminated glass

5 To 63 parts by weight of tetraethylene glycol-di-2-ethylhexanoate (4GO) as a plasticizer were added 0.4 parts by weight of 2,6-di-t-butyl-p-cresol as an antioxidant, 0.4 parts by weight of an ultraviolet absorber ("TINUVIN 326" produced by Ciba Specialty Chemicals Inc.) having a

10 benzotriazol structure, and 0.14 parts by weight of the obtained indole compound A. The resultant mixture was stirred by using a stirrer at 80°C for 30 minutes to give a plasticizer solution.

The obtained plasticizer solution was sufficiently

mixed with 100 parts by weight of polyvinyl butyral resin (PVB) (average polymerization degree: 1700, butyralation degree: 68.5 mol%, amount of the hydroxyl group: 30.6 mol%, amount of the acetyl group: 0.9 mol%). Then, a twin-screw
5 aeolotropic extruder was used to produce an interlayer film for a laminated glass having a film thickness of 760 μm . In this process, for the purpose of controlling the adhesion of the interlayer film for a laminated glass, a magnesium acetate solution was added so that concentration
10 of Mg in the interlayer film for a laminated glass was 65 ppm.

The laminated glass was produced in the same manner as in Example 1, except for the use of the obtained interlayer film for a laminated glass, and the same
15 evaluations were made.

Table 7 shows the results.

[0088]

(Examples 29 to 51, and Comparative Examples 16 to 18)

Each of the interlayer films for a laminated glass
20 was produced in the same manner as in Example 1, except that each of the compositions thereof was changed as shown in Table 7 and Table 8. The laminated glass was produced in the same manner as in Example 1, except for the use of the obtained interlayer film for a laminated glass, and
25 the same evaluations were made.

[0089]

[Table 7]

[0091]

(Example 52)

(1) Production of resin composition A

To 40 parts by weight of triethylene glycol di-2-ethylhexanoate (3GO) as a plasticizer were added 0.4 parts
5 by weight of 2,6-di-t-butyl-p-cresol as an antioxidant, 0.4 parts by weight of an ultraviolet absorber ("TINUVIN 326" produced by Ciba Specialty Chemicals Inc.) having a benzotriazol structure, 0.14 parts by weight of the
10 obtained indole compound A, and 0.28 parts by weight of tin-doped indium oxide particles (ITO) (volume average particle diameter: 35 nm). The resultant mixture was stirred by using a stirrer at 80°C for 30 minutes to give a plasticizer solution.

15 The obtained plasticizer solution was sufficiently mixed with 100 parts by weight of polyvinyl butyral resin (PVB) (average polymerization degree: 1700, butyralation degree: 68.5 mol%, amount of the hydroxyl group: 30.6 mol%, amount of the acetyl group: 0.9 mol%) to produce a resin
20 composition A. In this process, a magnesium acetate solution was added so that concentration of Mg in the resin composition A was 65 ppm.

[0092]

(2) Production of resin composition B

25 (Resin layer B)

To 60 parts by weight of triethylene glycol di-2-ethylhexanoate (3GO) as a plasticizer were added 0.4 parts
by weight of 2,6-di-t-butyl-p-cresol as an antioxidant, 0.4 parts by weight of an ultraviolet absorber ("TINUVIN
30 326" produced by Ciba Specialty Chemicals Inc.) having a benzotriazol structure, 0.14 parts by weight of the obtained indole compound A, and 0.28 parts by weight of tin-doped indium oxide particles (ITO) (volume average particle diameter: 35 nm). The resultant mixture was
35 stirred by using a stirrer at 80°C for 30 minutes to give

a plasticizer solution.

The obtained plasticizer solution was sufficiently mixed with 100 parts by weight of polyvinyl butyral resin (PVB) (average polymerization degree: 2450, butyralation degree: 65.5 mol%, amount of the hydroxyl group: 20.1 mol%, amount of the acetyl group: 13.4 mol%) to produce a resin composition B.

[0093]

(3) Production of interlayer film for laminated glass

10 The resin composition A and the resin composition B were co-extruded to produce an interlayer film for a laminated glass having a three-layer structure where a resin layer A (thickness: 330 μm), a resin layer B (thickness: 100 μm), and a resin layer A (thickness: 330 μm) were sequentially laminated.

15 The laminated glass was produced in the same manner as in Example 1, except for the use of the obtained interlayer film for a laminated glass, and the same evaluations were made.

20 Table 9 shows the results.

[0094]

(Examples 53 to 67, and Comparative Example 19)

25 Each of the interlayer films for a laminated glass was produced in the same manner as in Example 1, except that each of the compositions thereof was changed as shown in Table 9 and Table 10. The laminated glass was produced in the same manner as in Example 1, except for the use of the obtained interlayer film for a laminated glass, and the same evaluations were made.

30 Table 9 and Table 10 show the results.

[0095]

[Table 9]

			Example						Comparative Example
			52	53	54	55	58	57	19
Resin layer A	Composition (parts by weight)	PVB	100	100	100	100	100	10	100
		Plasticizer (3GO)	40	40	40	40	40	40	40
		Indole A	0.14	-	-	-	-	-	-
		Indole B	-	-	0.14	-	-	-	-
		Indole C	-	-	-	-	0.14	-	-
		Antioxidant	0.4	0.4	0.4	0.4	0.4	0.4	0.4
		Ultraviolet absorber	0.4	0.4	0.4	0.4	0.4	0.4	0.4
		ITO	0.28	0.28	0.28	0.28	0.28	0.28	0.28
Mg/ppm		85	85	85	85	85	85	85	
Resin layer B	Composition (parts by weight)	PVB	100	100	100	100	100	100	100
		Plasticizer (3GO)	60	80	80	60	80	80	80
		Indole A	0.14	1.08	-	-	-	-	-
		Indole B	-	-	0.14	1.08	-	-	-
		Indole C	-	-	-	-	0.14	1.08	-
		Antioxidant	0.4	0.4	0.4	0.4	0.4	0.4	0.4
		Ultraviolet absorber	0.4	0.4	0.4	0.4	0.4	0.4	0.4
		ITO	0.28	0.28	0.28	0.28	0.28	0.28	0.28
Evaluation of transmittance	Transmittance/%	Ts	88.3	88.8	85.8	86.0	88.5	88.2	87.7
		Tv	87.2	87.8	87.4	87.4	87.7	87.9	87.3
	T(380-400nm)/%	Before ultraviolet irradiation	0.05	0.06	0.05	0.09	0.08	0.05	10.16
		After ultraviolet irradiation	0.20	0.37	0.26	0.41	0.24	0.35	10.08
Light exposure test	ΔE	1.19	1.31	1.54	1.68	1.29	1.33	0.79	
Evaluation of preventive effect on pest insect attraction	The number of attracted winged insects	6	2	6	8	5	7	31	
Evaluation of penetration resistance	4 m	Pass	Pass	Pass	Pass	Pass	Pass	Pass	
	5 m	Pass	Pass	Pass	Pass	Pass	Pass	Pass	
	6 m	Pass	Pass	Pass	Pass	Pass	Pass	Pass	
Durability test to heat	ΔE	0.42	0.55	0.44	0.41	0.53	0.61	0.21	
	The number of bubble less than 5 mm	2	0	2	0	3	0	0	
	The number of bubble greater than or equal to 5 mm, and less than 10 mm	0	7	0	12	0	5	8	
	The number of bubble greater than or equal to 10mm	0	1	0	3	0	1	1	
Presence of solid components		Absent	Absent	Absent	Absent	Absent	Absent	Absent	

[0096]

[Table 10]

		Example									
		58	59	60	61	62	63	64	65	66	67
Resin layer A	PVB	100	100	100	100	100	100	100	100	100	100
	Plasticizer (3GO)	40	40	40	40	40	40	40	40	40	40
	Indole F	0.14	-	-	-	-	-	-	-	-	-
	Indole G	-	-	0.14	-	-	-	-	-	-	-
	Indole H	-	-	-	-	0.14	-	-	-	-	-
	Indole I	-	-	-	-	-	-	0.14	-	-	-
	Indole J	-	-	-	-	-	-	-	-	0.14	-
	Antioxidant	0.4	0.4	0.4	0.4	0.4	0.4	0.4	0.4	0.4	0.4
	Ultraviolet absorber	0.4	0.4	0.4	0.4	0.4	0.4	0.4	0.4	0.4	0.4
	ITO	0.28	0.28	0.28	0.28	0.28	0.28	0.28	0.28	0.28	0.28
Mg/ppm	65	65	65	65	65	65	65	65	65	65	
Resin layer B	PVB	100	100	100	100	100	100	100	100	100	100
	Plasticizer (3GO)	60	60	60	60	60	60	60	60	60	60
	Indole F	0.14	1.06	-	-	-	-	-	-	-	-
	Indole G	-	-	0.14	1.06	-	-	-	-	-	-
	Indole H	-	-	-	-	0.14	1.06	-	-	-	-
	Indole I	-	-	-	-	-	-	0.14	1.06	-	-
	Indole J	-	-	-	-	-	-	-	-	0.14	1.06
	Antioxidant	0.4	0.4	0.4	0.4	0.4	0.4	0.4	0.4	0.4	0.4
	Ultraviolet absorber	0.4	0.4	0.4	0.4	0.4	0.4	0.4	0.4	0.4	0.4
	ITO	0.28	0.28	0.28	0.28	0.28	0.28	0.28	0.28	0.28	0.28
Evaluation of transmittance	Transmittance/%	66.9	66.3	66.2	66.7	66.6	66.5	66.4	66.8	66.1	66.4
	Tv	87.5	87.4	87.8	87.2	87.8	87.6	87.9	87.7	87.8	87.5
	Before ultraviolet irradiation	0.08	0.05	0.04	0.03	0.07	0.03	0.06	0.04	0.06	0.06
	After ultraviolet irradiation	0.25	0.36	0.27	0.30	0.25	0.36	0.28	0.33	0.29	0.31
Light exposure test	1.28	1.30	1.45	1.49	1.38	1.41	1.32	1.36	1.41	1.51	
Evaluation of preventive effect on pest insect attraction	The number of attracted winged insects	6	5	2	7	8	2	4	6	8	3
		Pass	Pass	Pass	Pass	Pass	Pass	Pass	Pass	Pass	Pass
Evaluation of penetration resistance	4 m	Pass	Pass	Pass	Pass	Pass	Pass	Pass	Pass	Pass	Pass
	5 m	Pass	Pass	Pass	Pass	Pass	Pass	Pass	Pass	Pass	Pass
	6 m	Pass	Pass	Pass	Pass	Pass	Pass	Pass	Pass	Pass	Pass
Durability test to heat	ΔE	0.87	0.96	0.95	0.91	0.87	0.83	1.14	1.04	0.90	0.93
	The number of bubble less than 5 mm	1	0	1	0	3	0	1	0	2	0
	The number of bubble greater than or equal to 5 mm, and less than 10 mm	0	15	0	7	0	5	0	4	0	6
	The number of bubble greater than or equal to 10 mm	0	2	0	1	0	1	0	1	0	3
Presence of solid components	Absent	Absent	Absent	Absent	Absent	Absent	Absent	Absent	Absent	Absent	

[0097]

(Example 68)

To 49 parts by weight of triethylene glycol-di-2-ethylhexanoate (3GO) as a plasticizer were added 0.4 parts

by weight of 2,6-di-t-butyl-p-cresol (BHT) as an antioxidant, 0.4 parts by weight of an ultraviolet absorber ("TINUVIN 326" produced by Ciba Specialty Chemicals Inc.) having a benzotriazol structure, and 0.14
5 parts by weight of the obtained indole compound A. The resultant mixture was stirred by using a stirrer at 80°C for 30 minutes to give a plasticizer solution.

The obtained plasticizer solution was sufficiently mixed with 100 parts by weight of polyvinyl butyral resin
10 (PVB) (average polymerization degree: 2450, butyralation degree: 65.5 mol%, amount of the hydroxyl group: 20.1 mol%, amount of the acetyl group: 13.4 mol%). Then, a twin-screw aeolotropic extruder was used to produce an interlayer film for a laminated glass having a film
15 thickness of 760 μm . In this process, a magnesium acetate solution was added so that concentration of Mg in the interlayer film for a laminated glass was 65 ppm.

The laminated glass was produced in the same manner as in Example 1, except for the use of the obtained
20 interlayer film for a laminated glass, and the same evaluations were made.

Table 11 shows the results.

[0098]

(Examples 69 to 91, and Comparative Examples 20 to 22)

25 Each of the interlayer films for a laminated glass was produced in the same manner as in Example 1, except that each of the compositions thereof was changed as shown in Table 11 and Table 12. The laminated glass was produced in the same manner as in Example 1, except for
30 the use of the obtained interlayer film for a laminated glass, and the same evaluations were made.

Table 11 and Table 12 show the results.

[0099]

[Table 11]

INDUSTRIAL APPLICABILITY

[0101]

According to the present invention, it is possible to provide an interlayer film for a laminated glass which can
5 reduce transmittance of ultraviolet rays having wavelength of 380 to 400 nm and has excellent durability to light exposure while maintaining high visible light transmittance.

[0102]

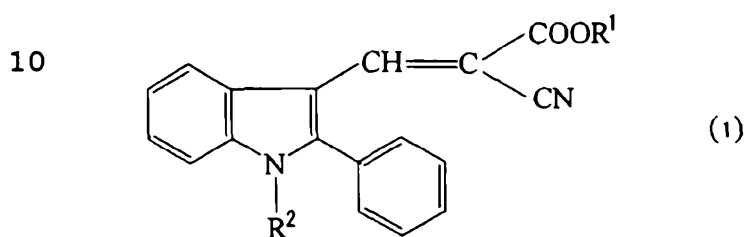
10 It is to be understood that, if any prior art publication is referred to herein, such reference does not constitute an admission that the publication forms a part of the common general knowledge in the art, in Australia or any other country.

15 [0103]

In the claims which follow and in the preceding description of the invention, except where the context requires otherwise due to express language or necessary implication, the word "comprise" or variations such as
20 "comprises" or "comprising" is used in an inclusive sense, i.e. to specify the presence of the stated features but not to preclude the presence or addition of further features in various embodiments of the invention.

CLAIMS

1. An interlayer film for a laminated glass,
which comprises a thermoplastic resin and an indole
5 compound having a structure represented by the following
general Chemical Formula (1):
[Chemical Formula 1]



15

wherein R¹ represents an alkyl group having 1 to 3 of
carbon atoms, and R² represents hydrogen, an alkyl group
having 1 to 10 of carbon atoms, or an aralkyl group having
7 to 10 of carbon atoms.

20

2. The interlayer film for a laminated glass
according to claim 1,
which further comprises a plasticizer.

25

3. The interlayer film for a laminated glass
according to claim 1 or 2,
wherein R¹ represents methyl group in the indole
compound having a structure represented by the general
Chemical Formula (1).

30

4. The interlayer film for a laminated glass
according to claim 1 or 2,
wherein the content of the indole compound is 0.030
to 0.145 weight% in the interlayer film.

35

5. The interlayer film for a laminated glass according to claim 2,
wherein the plasticizer is triethylene glycol di-2-ethylhexanoate.

5

6. The interlayer film for a laminated glass according to claim 1 or 2,
wherein the thermoplastic resin is a polyvinyl acetal resin.

10

7. The interlayer film for a laminated glass according to claim 1 or 2,
which further comprises an infrared absorber.

15

8. The interlayer film for a laminated glass according to claim 7,
wherein the infrared absorber is a tin-doped indium oxide particle.

20

9. The interlayer film for a laminated glass according to claim 1, 2, 3, 4, 5, 6, 7, or 8,
which comprises a multilayer laminated structure comprising at least two resin layers,
said at least two resin layers including an indole
25 compound-containing resin layer,
said indole compound-containing resin layer comprising the thermoplastic resin and the indole compound having a structure represented by the general Chemical Formula (1).

30

10. The interlayer film for a laminated glass according to claim 9,
wherein at least an outermost layer is the indole compound-containing resin layer.

49a

11. A laminated glass,
wherein the interlayer film for a laminated glass
according to claim 1, 2, 3, 4, 5, 6, 7, 8, 9 or 10 is
sandwiched between two sheets of glasses.

5

12. An interlayer film for a laminated glass
according to claim 1, or a laminated glass comprising the
interlayer film, substantially as herein described with
reference to the Examples, excluding Comparative Examples.