



## INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

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<p>(21) International Application Number:      PCT/US91/09501 (22) International Filing Date:        17 December 1991 (17.12.91) (30) Priority data: 632,780                            24 December 1990 (24.12.90)    US (71) Applicant: EASTMAN KODAK COMPANY [US/US]; 343 State Street, Rochester, NY 14650 (US). (72) Inventor: COOK, Gregory, Allen ; 14237 Shenandoah Avenue, Baton Rouge, LA 70817 (US). (74) Agent: MONTGOMERY, Mark, A.; 343 State Street, Rochester, NY 14650-2201 (US).</p>	<p>(81) Designated States: AT (European patent), BE (European patent), CA, CH (European patent), DE (European patent), DK (European patent), ES (European patent), FR (European patent), GB (European patent), GR (European patent), IT (European patent), JP, LU (European patent), MC (European patent), NL (European patent), SE (European patent).</p> <p><b>Published</b> <i>With international search report.</i></p>	
<p>(54) Title: EMULSIFIABLE LINEAR POLYETHYLENE COMPOSITIONS AND PROCESS FOR PREPARATION</p>		
<p>(57) Abstract</p> <p>Emulsifiable linear polyethylene compositions which exhibit good hardness and color characteristics can be obtained by oxidizing a blend of 90 to 60 % by weight of either linear low density polyethylene or high density polyethylene and 10 to 40 % by weight paraffin wax to an acid number of 12 to 30 mg KOH/g. The water emulsions obtained from the emulsifiable polyethylene compositions are useful as textile lubricants and fruit coatings.</p>		

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EMULSIFIABLE LINEAR POLYETHYLENE COMPOSITIONS  
AND PROCESS FOR PREPARATION

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Background of the Invention

Field of the Invention:

The present invention relates to novel emulsifiable linear polyethylene containing compositions which possess good color characteristics, such as high light transmittance, high density, and low penetration hardness values. The present invention also relates to a process for the production of such emulsifiable linear polyethylene compositions and the emulsions prepared from such emulsifiable linear polyethylene compositions.

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Discussion of the Background:

Water emulsifiable polyethylene waxes are widely used as self-polishing waxes in commercial applications. These waxes are relatively easy to use, are inexpensive, and provide good results. The waxes, in addition to being used in self-polishing wax formulations for floor polish applications, are also useful in formulating textile finishes, paper coatings, and the like. However, for applications such as textile lubricants and fruit coatings, waxes characterized by having high densities and low penetration hardness values as well as good color characteristics are desired.

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One method for producing emulsifiable polyethylene compositions involves a two-step procedure. The polyethylene is first thermally degraded and then oxidized. However, the presence of catalyst residues, such as Ti, Ca, Al, and Cl, results in a discoloration of the polyethylene during the thermal degradation step. Thus, emulsifiable

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polyethylenes produced by this process do not possess good color characteristics and are not suitable for use as textile lubricants and fruit coatings.

5 A process for oxidizing blends of specific polyethylene components and a specific microcrystalline wax is disclosed in U. S. Patent No. 2,879,239. The products produced by the process are said to be especially valuable for the preparation of emulsions which, when spread on a surface, leave  
10 hard, tough films having a high gloss.

U. S. Patent 4,632,955 discloses emulsifiable polyethylene paraffin blend compositions prepared by oxidizing a mixture of: (1) low density polyethylene (LDPE); (2) low density polyethylene wax; and (3)  
15 paraffin wax. By first blending the LDPE with paraffin wax, the initial viscosity of the polyethylene is lowered, and the thermal degradation step may be omitted. This method provides a savings in energy and a product with better color  
20 characteristics. However, the emulsifiable LDPE compositions possess low densities and high penetration hardness values and are, thus, unsuitable for use as textile lubricants or fruit coatings.

Thus, there is a need for emulsifiable  
25 polyethylene compositions which possess good color characteristics, high densities, and low penetration hardness values.

#### Summary of the Invention

30 Accordingly, one object of this invention is to provide novel emulsifiable polyethylene compositions which possess good color characteristics, high densities, and low penetration hardness values.

It is another object of the present invention to  
35 provide a process for the preparation of emulsifiable polyethylene compositions which possess good color characteristics, high densities, and low penetration

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hardness values.

It is a further object of the present invention to provide water-based emulsions of emulsifiable polyethylene compositions which possess high light  
5 transmittance, good color characteristics, high densities, and low penetration hardness values.

These and other objects which will become apparent during the following description of the present invention have been achieved by the oxidation  
10 of a blend of (a) about 90 to 60% by weight of either linear low density polyethylene (LLDPE) or high density polyethylene (HDPE) and (b) about 10 to 40% by weight of paraffin wax, to an acid number of about 12 to 30 milligrams KOH/gm.

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#### Detailed Description of the Preferred Embodiments

The emulsifiable polyethylene compositions produced by the present invention have high densities and low penetration hardness values. These  
20 emulsifiable polyethylene compositions have hardness characteristics of linear polyethylenes despite the presence of the soft paraffin wax. These oxidized polyethylene compositions are prepared without the need of thermal degradation step and thus have good  
25 color characteristics.

The LLDPE useful in this invention can have densities between about 0.931 g/cc and 0.940 g/cc and melt indices of no lower than about 55 dg/min preferably between about 75 and 125 dg/min. The  
30 LLDPE can be prepared by any conventional process, such as the polymerization of high purity polymerization-grade ethylene. The polymerization can be carried out in the gas-phase utilizing either a fluid bed or stirred bed. Alternatively, the  
35 polymerization can be carried out in the liquid-phase using either a slurry or solution. Further details for the preparation of LLDPE can be found, e.g., in

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Kirk-Othmer, "Encyclopedia of Chemical Technology", Volume 16, John Wiley, pp. 388-394 (1981).

Suitable HDPE can have densities between about 0.960 g/cc and 0.975 g/cc and melt indices of no  
5 lower than about 55 dg/min and preferably between about 75 and 125 dg/min. The HDPE used in the present invention can be produced by any conventional process, such as the Solvay process. Again, the polymerization can be carried out in either the gas-  
10 phase or liquid-phase, and further details can be found in Kirk-Othmer, "Encyclopedia of Chemical Technology", Volume 16, John Wiley, pp. 428-430 (1981).

Suitable LLDPE and HDPE contain less than about  
15 0.03% by weight of color imparting impurities such as ash preferably less than about 0.02% by weight. If the impurity content is above 0.03% by weight, the composition will have poor color characteristics.

The paraffin waxes useful in this invention are  
20 those having a melting point between about 40°C and 75°C, preferably between about 52°C and 64°C, which are obtained during petroleum refining processes. Such waxes are well known in the art and are readily commercially available.

25 The emulsifiable linear polyethylene compositions of the present invention preferably contain: (a) about 85 to 70% by weight LLDPE or HDPE and (b) about 15 to 30% by weight paraffin wax, based on the total weight of the composition. The  
30 emulsifiable linear polyethylene compositions of the present invention more preferably contain less than about 25% by weight paraffin wax.

The polyethylene and paraffin wax are blended by  
35 any suitable means known in the art such as for example, heated tumblers, blenders, extruders and the like. The molten blend can then be oxidized in any suitable apparatus by passing an oxygen containing

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gas such as air, oxygen, and inert gas and oxygen mixtures, into the blend at a temperature of less than about 135°C to 190°C, preferably from 150°C to 175°C, to obtain an oxidized blend having the desired acid number. The emulsifiable polyethylene compositions of the present invention are preferably oxidized to have an acid number from 15 milligrams KOH/gm to 24 milligrams KOH/gm. The oxidized blend can, if desired, be stabilized with conventional commercially available stabilizers.

The oxidized blends of the present invention are readily emulsifiable by processes well known in the art, such as water-to-wax and wax-to-water techniques. These emulsions are generally prepared using well known emulsifiers or surfactants and a base, water and the oxidized blend. These emulsions preferably contain between about 50 and 95% by weight water and between about 5 and 50% by weight of solids. The emulsion also preferably contains between about 4 and 40% by weight of the polyethylene composition. These emulsions more preferably contain between about 60 and 80% by weight water.

The emulsifiable polyethylene compositions prepared according to the present invention have a Gardner color of no higher than from about 2 preferably no higher than about 1 and a penetration hardness of less than or equal to 1 dmm. The compositions of the present invention provide excellent nonionic emulsions having light transmittance values of at least 60% and are useful in textile coatings and fruit coatings.

Other features of the present invention will become apparent in the course of the following descriptions of exemplary embodiments which are given for illustration of the invention and are not intended to be limiting thereof.

ExamplesOxidation

Oxidation of the blends was generally conducted as described below:

5           Atmospheric Pressure Cooxidation was conducted as follows. A 500-ml, four-neck flask was equipped with a Teflon blade stirrer, thermocouple, and spargetube and charged with 200 grams of the appropriate blend. The contents were heated to 150°C and oxygen was passed through the vigorously stirred molten wax at a rate of 1.7 liters per minute. The acid number was monitored throughout the oxidation, and the reaction was terminated near a target acid number of about 16 to 17.           Pressure Cooxidation was conducted as follows. A 2-liter Parr autoclave was charged with 700 grams of the appropriate blend, and the contents were heated to 150°C under 30 psig (207 kilopascal) nitrogen. The molten wax was stirred at 800 rpm, and the inlet gas was switched from nitrogen to compressed air. Oxidation was carried out at 150°C to 175°C, 100 to 175 psig (689 to 1207 kilopascal), and 2.5 to 3.0 liters per minute air flow. The acid number was measured periodically during the course of the cooxidation by collecting a small portion of the wax through a bottom drain valve. Upon reaching a target acid number near about 16 to 17, the product was dumped into a silicone lined box via the bottom drain valve.

30           Emulsification

The emulsifiable polyethylene compositions prepared in the Examples given below can be converted to water-emulsions by conventional wax-to-water methods. A typical nonionic emulsion is as follows. A 300-ml Parr autoclave is charged with 40 grams wax, 12 grams nonionic surfactant (Igepal CO-630), 1 gram potassium hydroxide, and sufficient demineralized

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water to make a 30 percent solids emulsion. The amount of KOH varies with the wax acid number and is calculated using the equation below:

$$\text{KOH, Grams} = \text{Acid Number} \times 0.0465.$$

5           The contents of the autoclave are stirred and heated to from 140°C to 145°C resulting in a pressure rise to approximately 120 psi (827 kilopascal). Stirring is continued for 45 minutes at 140°C. The heating mantle is removed, and the autoclave is  
10 immersed in an ice-water bath with continued agitation. Cooling of the emulsion to room temperature is achieved within 10 minutes.

Examples 1-3 and Comparative Examples 1-4

15           Emulsifiable LLDPE compositions were prepared using the ingredients, amounts, and oxidation conditions given in Table 1. The physical properties of the resulting emulsifiable LLDPE compositions for Examples 1-3 and Comparative Examples 1-4 are presented in Table 2.

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Table 1

Example	Composition				Oxidation Conditions				Residence Time(min)
	LLDPE	Melt Index of LLDPE	Density of LLDPE	Paraffin Wax*	LLDPE:Wax	Initial Temp(°C)	Temp. after 1 Hr (°C)	Pressure	
Example 1	XD-61501.02 <sup>(a)</sup>	85	0.935	SW100	80:20	175	150	1 atm	200
Example 2	XD-61501.02 <sup>(a)</sup>	85	0.935	SW100	80:20	150	150	100 psi <sup>(a)</sup>	362
Example 3	XD-61501.02 <sup>(a)</sup>	85	0.935	SW100	85:15	165	150	100 psi <sup>(a)</sup>	221
Comparative Example 1	Sclair 2316 <sup>(a)</sup>	59	0.935	SW100	80:20	175	150	1 atm	270
Comparative Example 2	Sclair 2316 <sup>(a)</sup>	59	0.935	SW100	80:20	150	150	100 psi <sup>(a)</sup>	202
Comparative Example 3	GRSN-7147 <sup>(a)</sup>	42	0.929	SW100	75:25	175	150	100 psi <sup>(a)</sup>	210
Comparative Example 4	Lotrex <sup>(a)</sup> MW 1210	25	0.940	SW100	75:25	175	175	1 atm	365

(a) Produced by Dow Chemical having octene comonomer.  
 (b) Produced by DuPont having butene comonomer and contains an unusually large amount of ash.  
 (c) Produced by Union Carbide having butene comonomer.  
 (d) Produced by Cdf Chimie having butene comonomer.  
 (e) 689 kilopascal

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Table 2

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<u>Properties</u>						
<u>Example</u>	<u>Acid Number</u> <u>(mg KOH/g)</u>	<u>Viscosity</u> <u>(125°C, cp)</u>	<u>Penetration</u> <u>Hardness</u> <u>(dmm)</u>	<u>Density</u> <u>(g/cc)</u>	<u>Gardner</u> <u>Color</u>	
15	Example 1	18.1	416	1.0	0.954	1
20	Example 2	21.2	290	1.0	0.956	1
	Example 3	16.1	725	<0.1	0.954	1
25	Comparative Example 1	17.6	1200	0.5	0.955	6
	Comparative Example 2	16.0	761	0.6	0.952	2
30	Comparative Example 3	16.5	257	2.7	0.946	1
35	Comparative Example 4	15.9	352	1.4	0.954	1

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5           As can be seen from the results presented in  
Table 2, the emulsifiable LLDPE compositions of the  
present invention all display good color and  
penetration hardness, while the emulsifiable LLDPE  
10           compositions prepared not in accordance with the  
present method exhibit either poor hardness  
properties or color.

          Thus, the compositions of Examples 1-3 all  
possess a high density, a low penetration hardness  
value, and a Gardner color of 1. In contrast, the  
15           compositions of Comparative Examples 1 and 2, which  
are prepared from LLDPE containing a large amount of  
ash, display Gardner colors of 6 and 2, respectively,  
and those of Comparative Examples 3 and 4, which are  
prepared from LLDPE with low melt indices, exhibit  
20           penetration hardness values greater than 1 dmm.

Examples 4 and 5 and Comparative Examples 5 and 6

          Emulsifiable HDPE compositions were prepared as  
described in Table 3, and the properties of the  
25           resulting compositions are given in Table 4.

Table 3

Example	Composition			Oxidation Conditions					
	HDPE (Comonomer)	Melt Index of HDPE	Density of HDPE	Paraffin Wax	HDPE:Wax	Initial Temp.(°C)	Temp. after 1 hr.(°C)	Pressure	Residence Time(min)
Example 4	XD-5320.15 <sup>(a)</sup>	80	0.96	SW100	80:20	175	150	1 atm	313
Example 5	XD-5320.15 <sup>(a)</sup>	80	0.96	SW100	80:20	150	150	100 psi <sup>(a)</sup>	399
Comparative Example 5	Norsoplast <sup>(b)</sup> MY 1610	16	0.957	M150	50:50	175	150	100 psi <sup>(a)</sup>	198
Comparative Example 6	Norsoplast <sup>(b)</sup> MY 1610	16	0.957	SW100	60:40	175	150	100 psi <sup>(a)</sup>	206

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(a) Produced by Dow Chemical.  
 (b) Produced by Cdf Chimie.  
 (c) 689 kilopascal.

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Table 4

Properties

10	<u>Example</u>	<u>Acid number</u> <u>(mg KOH/g)</u>	<u>Viscosity</u> <u>(125°C, cp)</u>	<u>Penetration</u> <u>Hardness</u> <u>(dmm)</u>	<u>Density</u> <u>(g/cc)</u>	<u>Gardner</u> <u>Color</u>
15	Example 4	20.1	407	<0.1	0.972	1
	Example 5	21.5	255	<0.1	0.973	1
20	Comparative Example 5	16.7	82.5	3.5	0.953	1
25	Comparative Example 6	17.5	126	2.0	0.951	1

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5           Again, as can be seen from the results presented  
in Table 4, emulsifiable HDPE compositions prepared  
according to the present invention display superior  
penetration hardness values as compared to those of  
the Comparative Examples.

10           Obviously, numerous modifications and variations  
of the present invention are possible in light of the  
above teachings. It is therefore to be understood  
that within the scope of the appended claims, the  
invention may be practiced otherwise than as  
15           specifically described herein.

Claims

I claim:

1. An emulsifiable polyethylene composition  
5 comprising:
  - (a) 90 to 60% by weight, based on the total  
weight of the composition, of at least one  
polyethylene selected from the group  
10 consisting of (1) linear low density  
polyethylene having a melt index no lower  
than 55 dg/min and a density between 0.931  
and 0.940 g/cc and (2) high density  
polyethylene having a melt index no lower  
15 than 55 dg/min and a density between 0.960  
and 0.975 g/cc, and
  - (b) 10 to 40% by weight, based on the total  
weight of said composition of paraffin wax;  
wherein said emulsifiable polyethylene  
composition has been oxidized to an acid number of 12  
20 to 30 mg KOH/g.
2. The emulsifiable polyethylene composition  
according to Claim 1, wherein said polyethylene has a  
melt index between 75 and 125 dg/min.
3. The emulsifiable polyethylene composition  
25 according to Claim 1, wherein said composition has a  
penetration hardness value no higher than 1 dmm and a  
gardner color no higher than 2.
4. The emulsifiable polyethylene composition  
according to Claim 3, wherein said composition  
30 contains less than 0.03% by weight ash and has a  
gardner color no higher than 1.
5. The emulsifiable polyethylene composition  
according to Claim 1, wherein said paraffin wax is  
present in the composition in an amount between 15  
35 and 30% by weight and has a melting point between  
40°C and 75°C.
6. The emulsifiable polyethylene composition

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according to Claim 5 wherein said paraffin wax has a melting point of 52°C to 64°C.

5 7. A water-polyethylene emulsion comprising 50 to 95% by weight water based on the total weight of said emulsion; and an emulsifiable polyethylene composition, comprising:

10 (a) 90 to 60% by weight, based on the total weight of the composition of at least one polyethylene selected from the group consisting of (1) linear low density polyethylene having a melt index between 55 and 125 dg/min and a density between 0.931 and 0.940 g/cc and (2) high density polyethylene having a melt index between 55 and 125 dg/min and a density between 0.960 and 0.975 g/cc and

15 (b) 10 to 40% by weight, based on the total weight of said composition, of paraffin wax;

20 wherein said emulsifiable polyethylene composition has been oxidized to an acid number of 12 to 30 mg KOH/g.

25 8. The emulsion according to Claim 7 wherein the emulsion contains between 60 and 80% by weight water and between 20 to 40% by weight solids.

9. The emulsion according to Claim 7 wherein said paraffin wax has a melting point between 40°C and 75°C.

30 10. The emulsion according to Claim 7 wherein the light transmittance of the emulsion is at least 60%.

11. A process for preparing an emulsifiable polyethylene composition comprising:

35 (i) blending (a) 90 to 60% by weight, based on the total weight of the composition, of at least one polyethylene selected from the group consisting of (1) linear low density

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polyethylene having a melt index no lower than 55 dg/min and a density between 0.931 and 0.940 g/cc and (2) high density polyethylene having a melt index no lower than 55 dg/min and a density between 0.960 and 0.975 g/cc, and (b) 10 to 40% by weight, based on the total weight of said composition of paraffin wax, to give a polyethylene-paraffin wax blend; and

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(ii) oxidizing said polyethylene-paraffin wax blend to give an emulsifiable polyethylene composition having an acid number between 12 and 30 mg KOH/g.

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12. The process according to Claim 11 wherein oxidizing Step (ii) utilizes an oxygen containing gas as an oxidant.

13. The process according to Claim 12 wherein said oxidizing Step (ii) is conducted at a temperature from 135°C to 190°C.

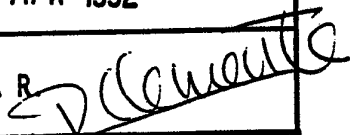
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14. The process according to Claim 13 wherein said oxidizing step is conducted at a temperature from 150°C to 175°C.

**INTERNATIONAL SEARCH REPORT**

PCT/US 91/09501

International Application No

<b>I. CLASSIFICATION OF SUBJECT MATTER</b> (if several classification symbols apply, indicate all) <sup>6</sup>		
According to International Patent Classification (IPC) or to both National Classification and IPC		
Int.Cl. 5 C09G1/10;	C08F8/06;	C08L23/30; C10G73/38
<b>II. FIELDS SEARCHED</b>		
Minimum Documentation Searched <sup>7</sup>		
Classification System	Classification Symbols	
Int.Cl. 5	C09G ; C08F ; C08L ; C10G	
Documentation Searched other than Minimum Documentation to the Extent that such Documents are Included in the Fields Searched <sup>8</sup>		
<b>III. DOCUMENTS CONSIDERED TO BE RELEVANT<sup>9</sup></b>		
Category *	Citation of Document, <sup>11</sup> with indication, where appropriate, of the relevant passages <sup>12</sup>	Relevant to Claim No. <sup>13</sup>
Y	DE,A,1 495 938 (ALLIED CHEMICAL CORPORATION) 14 May 1969 see claims	1-4, 6-8, 10-14
Y	FR,A,1 323 476 (FARBWERKE HOECHST AKTIENGESELLSCHAFT) 25 February 1963 see the whole document	1-4, 6-8, 10-14
A	EP,A,0 296 490 (BASF AKTIENGESELLSCHAFT) 28 December 1988 see the whole document	1-14
A	GB,A,1 050 548 (EASTMAN KODAK COMPANY) 7 December 1966 see the whole document	1-14
<p>* Special categories of cited documents : <sup>10</sup></p> <p>"A" document defining the general state of the art which is not considered to be of particular relevance</p> <p>"E" earlier document but published on or after the international filing date</p> <p>"L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)</p> <p>"O" document referring to an oral disclosure, use, exhibition or other means</p> <p>"P" document published prior to the international filing date but later than the priority date claimed</p> <p>"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention</p> <p>"X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step</p> <p>"Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.</p> <p>"A" document member of the same patent family</p>		
<b>IV. CERTIFICATION</b>		
Date of the Actual Completion of the International Search	Date of Mailing of this International Search Report	
09 APRIL 1992	15 APR 1992	
International Searching Authority	Signature of Authorized Officer	
EUROPEAN PATENT OFFICE	CLEMENTE GARCIA R 	

**ANNEX TO THE INTERNATIONAL SEARCH REPORT  
ON INTERNATIONAL PATENT APPLICATION NO. US 9109501  
SA 55524**

This annex lists the patent family members relating to the patent documents cited in the above-mentioned international search report. The members are as contained in the European Patent Office EDP file on The European Patent Office is in no way liable for these particulars which are merely given for the purpose of information. 09/04/92

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
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