

(12) INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

(19) World Intellectual Property Organization  
International Bureau



(43) International Publication Date  
5 December 2002 (05.12.2002)

PCT

(10) International Publication Number  
WO 02/096967 A1

(51) International Patent Classification<sup>7</sup>: C08F 297/00, C08G 83/00

(21) International Application Number: PCT/CA02/00792

(22) International Filing Date: 29 May 2002 (29.05.2002)

(25) Filing Language: English

(26) Publication Language: English

(30) Priority Data: 60/293,508 29 May 2001 (29.05.2001) US

(71) Applicant: THE UNIVERSITY OF WESTERN ONTARIO [CA/CA]; Stevenson Lawson Building, Room 319, London, Ontario N6A 5B8 (CA).

(81) Designated States (*national*): AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, OM, PH, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TN, TR, TT, TZ, UA, UG, UZ, VN, YU, ZA, ZM, ZW.

(84) Designated States (*regional*): ARIPO patent (GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZM, ZW), Eurasian patent (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European patent (AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, TR), OAPI patent (BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG).

(71) Applicants and

(72) Inventors: PUSKAS, Judit, E. [CA/CA]; 359 Ridout St. South, London, Ontario N6C 3Z6 (CA). PAULO, Christophe [FR/FR]; 5 Grande rue de la Croix rousse, F-69004 Lyon (FR). ANTONY, Prince [IN/CA]; 380 Fox Avenue, London, Ontario N6G 1H6 (CA).

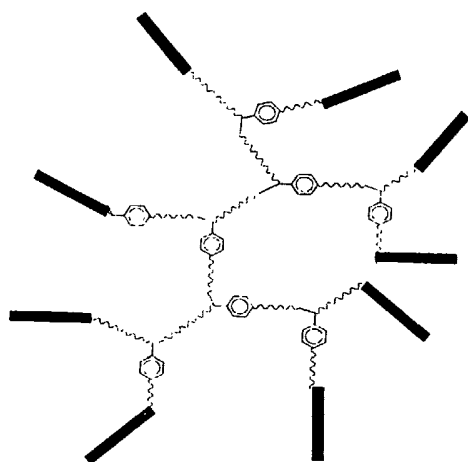
(74) Agent: HILL & SCHUMACHER; 87 Falcon Street, Toronto, Ontario M4S 2P4 (CA).

**Published:**

- with international search report
- before the expiration of the time limit for amending the claims and to be republished in the event of receipt of amendments

For two-letter codes and other abbreviations, refer to the "Guidance Notes on Codes and Abbreviations" appearing at the beginning of each regular issue of the PCT Gazette.

(54) Title: ARBORESCENT THERMOPLASTIC ELASTOMERS AND PRODUCTS THEREFROM



(57) Abstract: The present invention provides highly branched block copolymers comprising branched soft segments with a low glass-transition temperature ( $T_g$ ) and hard segments with a high  $T_g$  or crystalline melting point that exhibit thermoplastic elastomeric properties. More particularly the invention provides a highly branched block copolymer of a polyisooolefin and a polymonovinylidene arene characterized by having thermoplastic elastomeric properties wherein the aforesaid block copolymer comprises a highly branched polyisooolefin block and some of the branches of the aforesaid polyisooolefin block terminate in polymonovinylidene arene endblocks.



WO 02/096967 A1

## ARBORESCENT THERMOPLASTIC ELASTOMERS AND PRODUCTS THEREFROM

5

### FIELD OF THE INVENTION

This invention relates to arborescent or highly branched block copolymers comprising branched soft segments with a low glass-transition temperature ( $T_g$ ) and hard segments with a high  $T_g$  or crystalline melting point that exhibit thermoplastic elastomeric properties. More particularly this invention relates to highly branched block copolymers of polyisoolefins and polymonovinylidene arenes that exhibit thermoplastic elastomeric properties.

10

### BACKGROUND OF THE INVENTION

Thermoplastic elastomers (TPEs) are polymeric materials, which combine the properties of vulcanized rubbers and the processability and recyclability of thermoplastics, see for example B. M. Walker, "*Handbook of Thermoplastic Elastomers*", Van Nostrand Reinhold, New York (1979). While blends of elastomers and plastics are not compatible and show gross phase separation, block copolymers can only phase separate on a microscopic scale due to the connectivity of elastomeric and plastic blocks.

15

20

Branched polymers are of commercial interest due to their having markedly lower viscosity and less shear sensitivity than their linear counterparts. Thus branched block copolymers that have the added benefit of being thermoplastic elastomers should have a wide variety of commercial applications depending upon the elastomer and the thermoplastic used to form the block copolymer.

25

30

About 40% of TPEs are block copolymers, which contain both soft segments with a low glass-transition temperature ( $T_g$ ) and hard segments with a high  $T_g$  or crystalline melting point, see G. Holden, in "*Rubber Technology*", ed. M. Morton, Van Nostrand Reinhold, New York, Ch.16, 465 (1987). The hard segments associate, leading to physical crosslinks, which disappear when heated above a certain temperature (Order - Disorder Temperature, ODT)

and reappear immediately on cooling. The hard phase determines the mechanical strength, heat resistance, upper service temperature and strongly affects the oil and solvent resistance of a TPE. The chemical nature of the soft segments has an influence on elastic behavior, low temperature flexibility, thermal stability and aging resistance. According to present understanding in the field, in order to get good phase separation in block-type TPEs leading to good mechanical properties, the length of the elastomer chains should be as uniform as possible. This can be achieved by living polymerization, a unique process without termination and other side reactions of the growing polymer chain. Living conditions producing relatively uniform polymers can be achieved in anionic, cationic and radical systems.

An important commercial example of thermoplastic elastomeric block copolymers is styrene-elastomer-styrene, produced by living anionic polymerization. Most of the styrenic block copolymers have the general formula S-E-S, where S represents a hard amorphous polystyrene block and E represents a soft elastomeric block. Many of the polystyrene-polydiene block copolymers that are TPEs have the basic structure poly(styrene-*block*-butadiene-*block*-styrene) (S-B-S) or poly(styrene-*block*-isoprene-*block*-styrene) (S-I-S). The applications of these block copolymers are numerous. Important applications include solvent based and hot melt adhesives, sealants, coatings, hose, asphalt modifiers and sporting goods and automobiles, see G. Holden, N. R. Legge, R. Quirk, H. E. Schroeder (Eds.), *Thermoplastic Elastomers-A comprehensive Review*, Hanser Publishers, Munich (1996) and G. Holden, in *Encyclopedia of Polymer Science and Engineering*, ed. J. I. Kroschwitz, John Wiley and Sons, New York, Vol. 5, 416 (1996).

Recently, TPEs from another class of styrenic block copolymers have been developed with polyisobutylene (PIB) elastomeric segments, see United States Patent No. 4,946,899 issued to J. P. Kennedy et al. Figures 1A and 1B show a diagrammatic representation of the first generation of these PIB-based TPEs which are linear triblock (Figure 1A) and triarm—star block structures shown in Figure 1b. An important advantage of these TPEs based on

polyisobutylene-polystyrene (S-IB-S) block copolymers is that there is no need of hydrogenation of the elastomeric segments like that in the case of S-B-S or S-I-S, because of the presence of a saturated PIB elastomeric block. These novel TPEs were found to have excellent damping characteristics (similar to butyl rubber over a wide frequency range), oxidative and hydrolytic stability and good gas barrier properties, see K. Koshimura, H. Sato, *Polym. Bull*, **29**, 705 (1992) and J. P. Puskas, G. Kaszas, *Rubber Chem. Technol.*, **66**, 462 (1996).

The first generation of these TPEs were linear and triarm-star blocks, whose synthesis and basic chemical characterization have been reported, see G. Kaszas, J.E. Puskas, W.G. Hager and J.P. Kennedy, *J. Polym. Sci., Polym. Chem.*, **A29**, 427 (1991), and J. E. Puskas, G. Kaszas, J. P. Kennedy, W. G. Hager, *J. Polym. Sci., Polym. Chem.*, **A30**, 41 (1992). The living polymerization of IB by di- and tri-functional initiators gives a uniform rubbery mid-block, followed by the sequential addition of styrene (St) which results in a glassy outer block.

The architecture of copolymers can be controlled by the synthesis procedure, and TPEs with various composition and molecular weight (MW) have been prepared and characterized; for a review see J. P. Puskas, G. Kaszas, *Rubber Chem. Technol.*, **66**, 462 (1996). The most frequently used initiators are di- and tricumyl derivatives, especially di- and tricumyl-ether and -chloride. The co-initiator mainly used for making high molecular weight PIBs, suitable for block copolymer synthesis, is  $\text{TiCl}_4$ . The control of living IB polymerization is further improved by the use of electron pair donors like dimethyl sulfoxide or dimethyl acetamide and a proton trap such as di-tert-butyl pyridine (DtBP). These additives lead to better control of IB polymerization, resulting in narrow molecular weight distribution (MWD) PIB, and also improve the blocking efficiency of St monomers during TPE synthesis.

The S-IB-S triblocks and three-arm radial blocks obtained in the absence of electron pair donor and/or proton trap exhibit poor tensile properties due to inefficient blocking. The process has successfully been scaled up to a lb/batch scale (G. Kaszas, *Polym. Mater. Sci. Eng.*, **67**, 325 (1992) and an inventory of block copolymers have been prepared and characterized (P. Antony, J.E.Puskas:

Proceedings of the Polymer Processing Society Meeting, May 21-24, Montreal, Canada (2001). Kuraray Inc., Japan, recently test-marketed linear S-IB-S block copolymers. The important emerging applications of these first generation linear and tri-star block copolymer materials include medical applications such as  
5 rubber stoppers for drugs and blood, gaskets and caps for syringes, blood and drug storage bags and tubes and the like as disclosed in Japanese Patent No. 5,212,104; Japanese Patent No. 5,269,201; and Japanese Patent No. 5,295,054.

Due to the high oxidative and chemical stability of the S-IB-S macromolecules, these materials have the potential to replace silicone rubber or  
10 other soft to semirigid bio-implantable polymers; United States Patent Nos. 5,741,331 (1998); 6,102,939 (2000); and 6,197,240 (2001) issued to Pinchuk, ; L.Pinchuk, I.J.Khan, J.B.Martin and G.J.Wilson: Polyisobutylene-Based Thermoplastic Elastomers For Ultra Long Term Implant Applications; Sixth World Biomaterials Congress Transactions, 1452 (2001), and Pinchuk, L., Khan, I.J.,  
15 Martin, J.B., Bridgeman, J., Wilson, G.J., Glass, J., Si, J. and Kennedy, J.P.; A New Family of Thermoplastic Elastomers for Ultra-Long Term Implant Based Upon a Backbone of Alternating Quaternary and Secondary Carbons"; 24<sup>th</sup> Annual Meeting of the Society for Biomaterials; April 22-26, San Diego, (1998), p. 173; are directed to biologically stable, non-biodegradable implant devices and  
20 methods of producing same, using linear or star polyolefin copolymers having the structures shown in Figure 1. They also disclosed the structure shown in Figure 2A.

The second generation of PIB-based TPEs shown in Figure 2A are an even more recent development and include multi-arm-star blocks (S. Jacob, J. P. Kennedy, *Advances in Polym. Sci*, **146**, 1 (1999); J. E. Puskas, W. Pattern, P. M. Wetmore, V. Krukonis, *Rubber Chem. Technol.* **72**, 559 (1998). Multiarm-stars are expected to have improved strength and shear stability. Three-arm stars were shown to have considerably higher strength than comparable linear triblocks, but their shear sensitivity is still a concern; shearing off an arm would  
30 lead to a linear triblock. For polymers prepared by anionic polymerization beyond six arms no change in properties was found (L. K. Bi, L. J. Fetters, G.

Quack, R. N. Young "The Synthesis, Rheology and Morphology of Star Block Copolymers", Rubbercon 77, Int. Rubber Conf., Akron, Ohio, Paper No. 11, 11 (1977).

Although there are several reports on the synthesis and characterization of linear triblock, three-, six-, octa- and multi-arm (10-60) star S-IB-S block copolymers in the literature, the effect of hard and soft segment composition and molecular architecture on the mechanical properties of these TPEs has not been studied systematically. It has been reported that the minimum PS molecular weight required for good phase separation and physical properties is around 8000, and the maximum tensile strength achieved was 17-24 MPa with 400 - 600 % elongation, depending on the measurement method (for reviews see J. P. Puskas, G. Kaszas, *Rubber Chem. Technol.*, **66**, 462 (1996) and J.E.Puskas, G.Kaszas, *Progr. Polym. Sci.*, Elsevier Science Ltd., **25(3)**, 403 (2000)).

Referring again to United States Patent No. 4,946,899, this patent discloses a thermoplastic elastomeric star-shaped block copolymer comprising a polyisobutylene midblock and endblocks of polymerized styrene wherein the ratio of the the weight average molecular weight to the number average molecular weight of the midblock is from about 1.01 to about 1.5. It is further disclosed that each branch of the aforesaid star-shaped block copolymer is of essentially equal length.

United States Patent No. 5,428,111 issued to Faust et al. discloses a method of producing block copolymers having polyolefin midblocks and styrenic end blocks by the living polymerization of aromatic styrenic monomers initiated from living polyisoolefin chain end.

United States Patent No. 5,458,796 issued to Storey et al. discloses a process for the synthesis of star polymers, specifically polyisobutylene star-branched polymers by living carbocationic polymerization of isobutylene homopolymers, random copolymers or block copolymers using a monofunctional initiator and subsequently linking the preformed arms using divinylbenzene or diisopropenylbenzene.

United States Patent No. 5,721,331 issued to Shachi et al. is directed to a process of producing isobutylene-based block copolymers composed of isobutylene units and styrene blocks. This patent discloses that, if one uses the process disclosed in United States Patent No. 4,946,899 issued to J. P. Kennedy et al., and continues the polymerization after the styrene monomer is consumed, the active living chain ends can attack the styrene block of another chain, creating multiblocks shown in Figure 1C. This reaction, coupling via electrophilic substitution on the aromatic rings of the PS blocks, is well-known (Fodor Zs, Gyor M, Wang HC, Faust R. *J Macromol Sci, Pure Appl Chem A* 1993;30(5):349-63). Shachi claimed that this chain coupling, evidenced by multimodal molecular weight distribution (termed as "continuous distribution" in United States Patent No. 5,721,331 where the GPC peaks correspond to multiplets of the  $M_n$  of the starting linear triblock material), leads to improved properties, namely higher tensile strength and lower tensile set values. Similar multiblocks, consisting of polyisobutylene rubbery segments and cyclized polyisoprene hard segments and exhibiting thermoplastic elastomeric properties, are disclosed in United States Patent No. 4,910,261 issued to G.Kaszas, J.E.Puskas and J.P.Kennedy. These materials are a mixture of linear chains and various multiblocks. In the aforementioned two patents the polyisobutylene rubbery blocks are linear. The branched structure of the materials is the result of branching of the outer plastic blocks of the TPE.

Kee and Gauthier in *Macromolecules*, **32**, 6478 et seq. (1999) describe the preparation of highly branched polystyrene-polyisoprene copolymers by the successive grafting of the polymeric building blocks (graft on graft) which have a well-defined structure and the average molecular weight distributions of both the highly branched polystyrene and the polystyrene-polyisoprene graft copolymer are very narrow, being 1.1 or less. United States Patent No. 6,156,859 issued to Langstein et al. discloses a process for producing highly branched polyisoolefins by the reaction of isoolefins by polymerizing an isoolefin in the presence of a multifunctional monomer and an alkylaluminum at a temperature between 20°C and -100°C.

Puskas et al. reported the synthesis and characterization of arborescent polyisobutylenes by copolymerising an inimer with isobutylene (*Makromol. Chem, Macromol. Symp.* **132**, 117 (1998); Proceedings of the World Polymer Congress (IUPAC Macro 2000), 384 (2000)). Arborescent (hyperbranched) polymers belong to the class of dendritic polymers but are characterized by an irregular tree-like structure. These polymers are a relatively recent development and very little structure-property relationship data is available, see D. A. Tomalia: *Makromol. Chem., Macromol. Symp.* **101**, 243 (1996).

Heretofore there has not been disclosed an arborescent branched block copolymer with thermoplastic elastomeric properties comprising a highly branched polyisooolefin block wherein the branches are of irregular length in which some of the branches of the aforesaid polyisooolefin bear rigid polymer end block segments. Thus, it would be very advantageous to provide such materials that have thermoplastic elastomeric properties comparable, or superior, to the materials shown in Figure 2A but which can be produced more economically than these materials. Such materials would have many uses including use as implantable prostheses in medical applications.

## SUMMARY OF THE INVENTION

The present invention provides an arborescent branched block copolymer comprising an arborescent elastomeric polymer block with a low glass-transition temperature ( $T_g$ ) and rigid polymer block segments with a high  $T_g$  or crystalline melting point.

The present invention also provides an arborescent branched block copolymer of a polyisooolefin and a polymonovinylidene arene characterized by having thermoplastic elastomeric properties wherein the aforesaid block copolymer comprises an arborescent polyisooolefin polymer block and some of the branches of the aforesaid arborescent polyisooolefin polymer block terminate in polymonovinylidene arene plastic endblocks.



The present invention also provides a process for producing an arborescent branched block copolymer of a polyisooolefin and a polymonovinylidene arene characterized by having thermoplastic elastomeric properties wherein the aforesaid arborescent branched block copolymer comprises an arborescent polyisooolefin polymer block and some of the branches of the aforesaid arborescent polyisooolefin polymer block terminate in polymonovinylidene arene plastic endblocks, comprising:

a) polymerizing an isooolefin in the presence of at least one inimer and a Lewis acid halide coiniciator, at a temperature of between about  $-20^{\circ}\text{C}$  and  $100^{\circ}\text{C}$  to produce an arborescent elastomeric branched polymer, the inimer including at least one group for (co)polymerizing in a cationic polymerization of the isooolefin and at least one group for initiating cationic polymerization of the isooolefin; and thereafter

b) adding a compound having an effective electron pair donor for improving blocking efficiency and adding a monovinylidene arene suitable for the production of the polyvinylidene plastic blocks; and

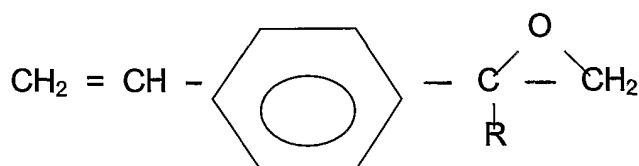
c) terminating said polymerization reaction after a selected period of time by addition of an effective nucleophile compound which terminates said polymerization reaction, said selected period of time being sufficiently long enough to ensure production of at least individual units of said arborescent branched block copolymer.

In the process the *inimer*, a compound carrying both an *initiator* and a *monomer* functionality (IM), is copolymerized with one or more olefins. Very high MW arborescent PIBs are produced using 4-(2-hydroxy-isopropyl) styrene and 4-(2-methoxy-isopropyl) styrene as IM in a "one-pot" living-type polymerization system. The reactive chain ends of arborescent PIB are blocked with a material such as styrene to form PIB-PS blocks.

In another aspect of the invention there is provided a process for producing an arborescent branched polyisooolefin block copolymer characterized by having elastomeric properties, comprising:

a) polymerizing an isoolefin in the presence of a multifunctional monomer and a Lewis acid halide coinitiator at a temperature of between about  $-20^{\circ}\text{C}$  and  $-100^{\circ}\text{C}$  to produce an elastomeric arborescent branched polyisoolefin polymer, the multifunctional monomer having a formula

5



10

where R can be H,  $\text{CH}_3$  or an alkyl or aryl group; thereafter

b) adding a compound having an effective electron pair donor for improving blocking efficiency and adding a monovinylidene arene suitable for the production of the polyvinylidene arene plastic endblocks; and

15

c) terminating said polymerization reaction after a selected period of time by addition of an effective nucleophile compound which terminates said polymerization reaction, said selected period of time being sufficiently long enough to ensure production of at least individual units of said arborescent branched block copolymer.

20

The present invention also provides a non-biodegradable article of manufacture comprising an implantable prosthesis or implant formed of an arborescent branched block copolymer of a polyisoolefin and a polymonovinylidene arene characterized by having thermoplastic elastomeric properties wherein the aforesaid arborescent branched block copolymer comprises an arborescent branched polyisoolefin polymer block and some of the branches of the aforesaid arborescent branched polyisoolefin block terminate in

25

polymonovinylidene arene plastic endblocks.

30

## BRIEF DESCRIPTION OF THE DRAWINGS

The present invention will now be described by way of example only, reference being had to the accompanying drawings in which:

5 Figure 1A shows a diagrammatic representation of PRIOR ART first generation linear triblock polyisobutylene-polystyrene (PIB-PS) block-type thermoplastic elastomers (TPEs);

Figure 1B shows a diagrammatic representation of PRIOR ART first generation triarm-star block polyisobutylene-polystyrene (PIB-PS) block-type thermoplastic elastomers (TPEs);

10 Figure 1C shows a diagram of the structure of PRIOR ART PIB-PS multiblock copolymers disclosed in United States Patent No. 5,721,331;

Figure 2A shows a diagrammatic representation of PRIOR ART second generation multiarm-star block polyisobutylene-polystyrene (PIB-PS) block-type thermoplastic elastomers (TPEs);

15 Figure 3 shows an example reaction to produce the elastomeric midblock section of the highly branched block copolymers of the present invention;

Figure 4 shows a diagrammatic representation of a highly branched block copolymer produced according to the process illustrated in Figure 3 comprising soft segments with a low glass-transition temperature ( $T_g$ ) and hard segments with a high  $T_g$  or crystalline melting point that exhibit thermoplastic elastomeric properties produced in accordance with the present invention; and

20 Figure 5 shows several examples of inimers which may be used in the process shown in Figure 3 to obtain arborescent copolymers.

## 25 DETAILED DESCRIPTION OF THE INVENTION

Referring to Figure 4, the present invention provides highly branched block copolymers comprising soft elastomeric segments with a low glass-transition temperature ( $T_g$ ) and hard thermoplastic segments with a high  $T_g$  or crystalline melting point so that the entire structure exhibits thermoplastic elastomeric properties.

30

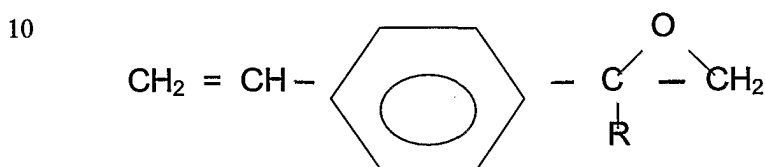
The present invention provides in a preferred embodiment of the invention a highly branched block copolymer of a polyisoolefin and a polymonovinylidene arene characterized by having thermoplastic elastomeric properties wherein the aforesaid block copolymer comprises a highly branched polyisoolefin block and some of the branches of the aforesaid polyisoolefin block terminate in polymonovinylidene arene endblocks. In addition to the term "highly branched", the terms "arborescent" and "hyperbranched" also may be used to describe the structure disclosed herein. The highly branched or arborescent polymers have an irregular tree-like structure.

The highly branched block copolymers of the present invention which have been found to exhibit thermoplastic elastomeric properties contain a highly branched polyisoolefin midblock that is synthesized using a process which involves use of an amount of *inimer*, a compound carrying both an *initiator* and a *monomer* functionality (IM), which is copolymerized with an olefin. Referring to Figure 3, the inimers used in the present invention to initiate the polymerization and to introduce the random branching points into the elastomeric midblock can be represented by the general formula A—B, where A is a polymerizable function such as a vinylic double bond shown in Figure 3 and B is an initiating group. For isobutylene polymerization B can be a tertiary ether, tertiary chloride, tertiary methoxy group or tertiary ester, well-known initiators for isobutylene polymerization. Very high MW arborescent PIBs were successfully synthesized by this method using 4-(2-hydroxy-isopropyl) styrene and 4-(2-methoxy-isopropyl) styrene as IM in a "one-pot" living-type polymerization system. The reactive chain ends of arborescent PIB have been blocked with St to form PIB-PS blocks.

Figure 5 shows the formulas for several different inimers that may be used in the synthesis process of Figure 3. These include 4-(2-hydroxyisopropyl)styrene; 4-(2-methoxyisopropyl)styrene; 4-(2-chloroisopropyl)styrene; 4-(2-acetoxisopropyl)styrene; 2,3,5,6-tertamethyl-4-(2-hydroxy isopropyl )styrene; 3-(2-methoxyisopropyl)styrene; 4-(epoxyisopropyl)styrene; 4,4,6-trimethyl-6-hydroxyl-1-heptene; 4,4,6-trimethyl-6-

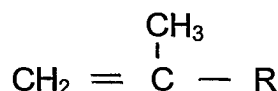
chloro-1-heptene; 4,4,6-trimethyl-6,7-epoxy-1-heptene; 4,4,6,6,8-pentamethyl-8-hydroxyl-1-nonene, 4,4,6,6,8-pentamethyl- 8-chloro-1-nonene; 4,4,6,6,8-pentamethyl-8,9-epoxy-1-nonene; 3,3,5-trimethyl-5- hydroxyl-1-hexene; 3,3,5-trimethyl-5- chloro-1-hexene; 3,3,5-trimethyl-5-6-epoxy-1-hexene; 3,3,5,5,7-pentamethyl-7-hydroxyl-1-octene, 3,3,5,5,7-pentamethyl- 7-chloro-1-octene; 3,3,5,5,7-pentamethyl-7,8-epoxy-1-octene.

If the initiator function B shown in Figure 3 is an oxirane ring, so that the initiator has the formula



where R can be H, CH<sub>3</sub> or an alkyl or aryl group, then each branching point has a hydroxy group incorporated into the polystyrene elastomeric block, as described in J. Song, J. Bodis and J.E.Puskas: Direct Functionalization of Polyisobutylene by Living Initiation with  $\alpha$ -Methylstyrene Epoxide. J.Polym.Sci., Polym. Chem. 40(8), 1005-1015 (2002). This very advantageously increases the hydrophilicity of the rubbery segments, which is very important and useful in biomedical applications.

The isoolefins which are used in the synthesis of the highly branched polyisoolefin are those with the formula



where R represents a C<sub>1</sub> to C<sub>4</sub> alkyl group such as methyl, ethyl or propyl. Isobutylene and 2-methyl-1-butene are preferred, and isobutylene is particularly preferred.

Monovinylidene arenes suitable for the production of the polyvinylidene blocks that form endblocks on some of the branches of the highly branched polyisoolefin are selected from the group consisting of C<sub>8</sub> to C<sub>12</sub> monovinylidene

arenes which may be substituted with a C<sub>1</sub> to C<sub>12</sub> alkyl or alkoxy group or a fluorine, chlorine, bromine or iodine atom on the aromatic ring. Preferred monovinylidene arenes are selected from the group consisting of styrene, p-methylstyrene, p-tertiarybutylstyrene, p-chlorostyrene, indene and mixtures thereof. Particularly preferred is styrene.

The highly branched polyisooolefin that is used as a basis for producing the highly branched block copolymers of the present invention have a branching frequency of from about 5 to about 60, preferably of from about 8 to about 35.

In order that the highly branched block copolymers of the present invention exhibit thermoplastic elastomeric properties it is preferred that the weight percent of the polymonovinylidene arene endblocks in the block copolymers be of from about 5 to about 50 weight percent.

The number average molecular weight, M<sub>n</sub>, of the highly branched polyisolefins is from about 10,000 to about 2,000,000, preferably from about 500,000 to about 1,000,000. The molecular weight distribution of the highly branched polyisooolefin is from about 1 to about 20, preferably from about 1.2 to about 2.8.

The process according to the present invention is carried out in an inert organic solvent or solvent mixture in order that the highly branched polyisooolefin and the final block copolymer remain in solution and at the same time there is some degree of polarity so that the polymerization proceeds at a reasonable rate. In order to fulfil these requirements a single solvent that may be used is n-butyl chloride or a mixture of a nonpolar solvent and a polar solvent wherein suitable nonpolar solvents include methylcyclohexane and cyclohexane and appropriate polar solvents include ethyl chloride, methyl chloride and methylene chloride. Especially preferred is a mixture of methylcyclohexane and methyl chloride. To achieve suitable solubility and polarity it has been found that the ratio of the nonpolar solvent to the polar solvent on a weight basis should be from about 80:20 to about 40:60, preferably about 60:40.

The temperature range within which the process is carried out is from about -20 to -100<sup>0</sup> C, preferably from about -50 to -80<sup>0</sup> C. The procedure of the

present invention is preferably carried out using a 1 to 30 percent polyisooolefin solution (weight/weight basis), especially between 5 and 10 weight percent.

In order to produce the highly branched block copolymers of the present invention it is necessary to employ a coinitiator, a Lewis acid halide. Suitable Lewis acid halides include boron trichloride, aluminum trichloride and titanium tetrachloride, titanium tetrachloride being preferred. The ratio of the initiator to the monovinylidene arene on a molar basis is from about 1:1 to about 1:30, preferably from about 1:10 to about 1:20.

The highly branched block copolymers of the present invention may also be produced in a one-step process wherein the isooolefin is copolymerized with the initiator monomer in conjunction with the coinitiator in a solution at a temperature of from about  $-20$  to  $-100^{\circ}$  C. An electron donor and a proton trap are then introduced, followed by the addition of a pre-chilled solution of the monovinylidene arene in the solvent and the polymerization is allowed to continue until it is terminated by the addition of a prechilled nucleophile such as methanol. The polymerization reaction is allowed to proceed for a preselected period of time prior to being terminated in order to produce the arborescent branched block copolymer structure as shown in Figure 4.

Alternatively; in order to connect two or more of the arborescent structures shown in Figure 4, the polymerization process is allowed to continue after all the styrene is consumed, as disclosed in United States Patent No. 5,721,331 issued to Shachi et al., which is incorporated herein by reference. This patent discloses that when the polymerization process is continued after the styrene monomer is consumed, the active living chain ends can attack the styrene block of another chain, creating multiblocks with each block being one of the arborescent blocks shown in Figure 4. The individual arborescent branched block copolymers are bound together wherein at least one of the polymonovinylidene arene plastic endblocks on one arborescent branched block copolymer is chemically bound to one polymonovinylidene arene plastic endblock on another arborescent branched block copolymer. Thus when multiblocks are the desired end product,

the polymerization reaction is allowed to proceed for a longer period of time prior to terminating the polymerization reaction by addition of a suitable nucleophile.

The production of the highly branched block copolymers necessitates the use of additives such as electron pair donors to improve blocking efficiency and proton traps to minimize homopolymerization. Examples of suitable electron pair donors are those nucleophiles that have an electron donor number of at least 15 and no more than 50 as tabulated by Viktor Gutmann in "The Donor Acceptor Approach to Molecular Interactions", Plenum Press(1978) and include ethyl acetate, dimethylacetamide, dimethylformamide and dimethyl sulphoxide, dimethyl acetamide being preferred. Suitable proton traps include 2,6-ditertiarybutylpyridine, 4-methyl-2,6-ditertiarybutylpyridine and diisopropylethylamine, 2,6-ditertiarybutylpyridine being preferred.

While not wishing to be bound by any theory it is believed that the degree of branching of the polyisoolefin, the molecular weight distribution of the polyisoolefin, the weight ratio of the coiniciator to the polyisoolefin, the molar ratio of the coiniciator to the monovinylidene arene, the reaction temperature and the reaction time may affect the extent to which endblocking of the polyisoolefin branches occurs. As a consequence by varying some of the aforementioned parameters it is possible to produce highly branched block copolymers with different thermoplastic elastomeric properties.

### EXAMPLES

The highly branched polyolefins that are the basis for the production of the highly branched block copolymers of the present invention were prepared using the general procedure outlined above. The polymerizations were carried out under a dry nitrogen atmosphere in a Mbraun Labmaster 130 glove box maintained at a temperature of  $-80^{\circ}$  C with an FTS Flexi Cool Immersion Cooler. A 3-necked round bottom flask of an appropriate size, equipped with an overhead stirrer was charged with a mixture of methylcyclohexane and methyl chloride 60/40 (v/v) and then in sequence was added the initiator-monomer, 4-(2-methoxy)isopropylstyrene, the proton trap 2,6-ditertiarybutylpyridine, and the



isoolefin, in this instance isobutylene. Polymerization was then begun by the rapid addition of a chilled stock solution of the coinitiator, titanium tetrachloride in methylcyclohexane. Upon completion the reaction was quenched by the addition of prechilled methanol. The polymer was isolated and purified in the usual manner and dried in a vacuum oven for one week.

The block polymers of the present invention were characterized in the following manner. The polymer Mns, Mws, and MWDs were determined by Size Exclusion Chromatography using a Waters system equipped with six Styragel columns (106, 105, 104, 103, 500 and 100 A pore sizes). A Waters 410 DRI detector, a Dawn DSP 18 angle Laser Light Scattering detector and a Waters 996 PDA detector. The elution medium, tetrahydrofuran, maintained at a temperature of 35 °C was pumped at a rate of 1ml/min. Absolute MWs were obtained using Astra software. As the highly branched polyisobutylene contained less than 1 percent of the initiator monomer, the value  $dn/dc=0.93$  from J. Wiley & Sons, Polymer Handbook 3rd Edition, was used for the calculations. In the case of the highly branched block copolymers of polyisobutylene and polystyrene the formula  $dn/dc(\text{block}) = (\text{wt}\% \text{styrene}) \times dn/dc(\text{styrene}) + (\text{wt}\% \text{PIB}) \times dn/dc(\text{PIB})$ . These methods gave good agreement with data obtained by assuming 100% mass recovery on the SEC columns. <sup>1</sup>H NMRs were conducted using a Varian Gemini 300 MHz NMR with deuterated chloroform as solvent.

The branching frequencies of the polyisobutylenes were determined by the selective destruction of the branching points using a procedure based on that reported by Kennedy et al. in Polymer Bulletin, 1981, in which the aromatic core of a three-arm star branched polymer is selectively destroyed without affecting the polymer chains. From the ratios of the number average molecular weights before and after link destruction the branching frequencies were then calculated. The following examples are intended to illustrate the embodiments of the invention and are not intended to limit the scope of the invention.

Table 1 lists the molecular weight and branching frequency data for the highly branched polyisobutylenes used in the examples.

**TABLE 1**

Sample ID	M <sub>n</sub> g/mol	M <sub>w</sub> g/mol	MWD	Branching frequency
PIB 1	561,000	681,000	1.2	33.0
PIB 2	392,000	1,039,000	2.6	11.6

<sup>1</sup>Branching frequency determined experimentally

5

**EXAMPLE 1**

The inimers used are listed in Table 1. The arborescent polyisobutylenes listed in Table 1 were produced with 4-(2-methoxy)isopropylstyrene as an inimer. In a Mbraun Labmaster 130 glove box under an atmosphere of nitrogen a 3-necked flask equipped with an overhead stirrer was charged with a mixture of methylcyclohexane and methyl chloride and cooled to a temperature of  $-80^{\circ}\text{C}$  by means of a FTS Flexi Cool Immersion Cooler. To this solution there was then added 0.196g (1.024 mmol) 2,6-ditertiarybutylpyridine, 0.08g (0.968 mmol) N,N-dimethylacetamide and 5g (0.048 mol) styrene. To this solution was added 10g PIB 1 which had been dissolved overnight in 100ml of methylcyclohexane affording a 2.5 weight percent solution of the PIB 1 in 200ml of a 60/40 (w/w) mixture of methylcyclohexane and methyl chloride. The reaction mixture was thoroughly stirred and polymerization was initiated by the addition of 2.02g (0.011 mol) of titanium tetrachloride. After approximately 50 minutes stirring the reaction was terminated by the addition of prechilled methanol. Analysis by Size Exclusion Chromatography revealed that in addition to the polyisobutylene-polystyrene block copolymer there was a significant amount of homopolystyrene present, necessitating that the total product be washed with methyl ethyl ketone three times and then dried. <sup>1</sup>H NMR spectroscopic analysis revealed that the amount of polystyrene in the block copolymer was 9.4 weight percent. The M<sub>n</sub> of the block copolymer PIB 1-PSty was 595,400 g/mol and the M<sub>w</sub>/M<sub>n</sub> was 1.3 as measured by Size Exclusion Chromatography coupled with Multiangle Laser Light Scattering.

25

## EXAMPLE 2

This was performed in an analogous manner to Example 1. To a mixture of methylcyclohexane and methyl chloride maintained at a temperature of  $-80^{\circ}\text{C}$  was added 0.051g (0.267 mmol) 2,6-ditertiarybutylpyridine, 0.12g (1.377 mmol) N,N- dimethylacetamide and 2.5g (0.024 mol) styrene. To this solution was added 5g PIB 2 which had been dissolved overnight in 50ml of methylcyclohexane affording a 2.5 weight percent solution of PIB 2 in 100ml of a 60/40 (w/w) mixture of methylcyclohexane and methyl chloride. The reaction mixture was stirred thoroughly and the polymerization was initiated by the addition of 3.8g (0.020 mol) of titanium tetrachloride. After approximately 60 minutes stirring the reaction was terminated by the addition of prechilled methanol. Analysis by Size Exclusion Chromatography revealed that other than the block copolymer there was very little polystyrene present.  $^1\text{H}$  NMR spectroscopic analysis indicated that the amount of polystyrene present in the block copolymer was 16.4 weight percent. The  $M_n$  of the block copolymer PIB 2-PSty was 543,400 g/mol and the  $M_w/M_n$  was 2.6 as measured by Size Exclusion Chromatography coupled with Multiangle Laser Light Scattering.

### Physical Characterization

Physical characterization of the highly branched block copolymers PIB 1-PSty and PIB 2-PSty1 was carried out by compression moulding the samples (thickness~0.8mm) at  $210^{\circ}\text{C}$  for 3 minutes in an electrically heated hydraulic press at a pressure of 10 MPa. Prior to the pressure being applied the samples were preheated for 5 minutes. The results are shown in Table 2.

The above results clearly indicate that with the introduction of the polystyrene endblocks on to some of the entangled branches of the polyisobutylenes, the block copolymers exhibit a self-reinforcing characteristic as a result of the fact that there is phase separation between the polyisobutylene block and the polystyrene blocks. This is particularly evident in the significant increase in the tensile strengths of the block copolymers PIB 1-PSty and PIB 2-PSty relative to the tensile strengths of PIB 1 and PIB 2 respectively.

Atomic force microscopy substantiates this conclusion in that the phase contrast picture of the block copolymer PIB 2-PSty exhibits polystyrene blocks irregularly distributed amongst the polyisobutylene block.

Three further reactions were carried out using the highly branched polyisobutylene PIB 2 but varying the conditions such as the length of reaction time, the order of addition, the amount of the initiator and the amounts of the electron donor and the proton trap.

### EXAMPLE 3

This was performed in an analogous manner to Example 1. 5g of PIB 2 was dissolved in 128ml of methylcyclohexane and the solution was cooled to  $-80^{\circ}\text{C}$ . To this solution was then added 50ml methyl chloride, 0.196g (1.02 mmol) 2,6-ditertiarybutyl pyridine, 0.084g (0.968 mmol) N,N-dimethylacetamide and 6.06g (0.032mol) titanium tetrachloride. Polymerization was then initiated by the addition of 5g (0.048mol) styrene and allowed to proceed for 16 hours, at which point it was then terminated by the addition of methanol. The resultant block copolymer PIB 2-PSty 3 was extracted with methyl ethyl ketone and dried.  $^1\text{H}$  NMR spectroscopic analysis of the block copolymer revealed that it contained 34 weight percent polystyrene. The  $M_n$  of the final product was 585,405 g/mol and the  $M_w/M_n$  was 2.7 as measured by Size Exclusion Chromatography coupled with Multiangle Laser Light Scattering.

### EXAMPLE 4

This reaction was performed in an analogous manner to Example 1. An amount of 5g of PIB 2 was dissolved in 167ml of methylcyclohexane and the solution and the solution was cooled to  $-80^{\circ}\text{C}$ . To this solution was then added 101 ml of methyl chloride, 0.15g (0.8 mmol) ditertiarybutyl pyridine, 0.07g (0.806 mmol) N,N-dimethylacetamide and 31.2g (0.300 mol) styrene. The polymerization was initiated by the addition of 9.11g (0.048 mol) titanium tetrachloride. The reaction was allowed to proceed for 8 hours and then terminated by the addition of methanol. The resultant block copolymer PIB 2-PSty 4 was then extracted with methyl ethyl ketone and dried.  $^1\text{H}$  NMR

spectroscopic analysis of the block copolymer revealed that the block copolymer contained 31 weight percent styrene. The  $M_n$  of the final product was 560,000 g/mol and the  $M_w/M_n$  was 2.3. The products from Examples 3 and 4 were not elastomeric like the starting PIB 2 but very strong, tough materials that extended like a crosslinked rubber and regained their shape when the stress was released. This is evident from their physical characteristics which are listed in Table 2.

### EXAMPLE 5

This reaction was performed in an analogous manner to Example 1. An amount of 5g of PIB 2 was dissolved in 167ml of methylcyclohexane and the solution and the solution was cooled to  $-80^{\circ}\text{C}$ . To this solution was then added 101ml methyl chloride, 0.290g (1.514 mmol) ditertiarybutyl pyridine, 0.131g (1.506 mmol) N,N-dimethylacetamide and 31.2g (0.300 mol) styrene. The polymerization was initiated by the addition of 9.11g (0.048 mol) titanium tetrachloride. The reaction was allowed to proceed for 8 hours and then terminated by the addition of methanol. The resultant block copolymer PIB 2-PSty 4 was then extracted with methyl ethyl ketone and dried.  $^1\text{H}$  NMR spectroscopic analysis of the block copolymer revealed that the block copolymer contained 27 weight percent styrene. The  $M_n$  of the final product was 530,000 g/mol and the  $M_w/M_n$  was 2.3.

### EXAMPLE 6

This example is to demonstrate that epoxy initiation introduces a hydroxy functional group into the PIB chain. The reaction vessel in a glove box under dry nitrogen was charged with 100 ml total reaction volume containing 2.1 mol/L isobutylene (IB) monomer, 0.019 mol/L 2-methyl-2-phenyl-oxirane or  $\alpha$ -Methylstyrene epoxyde (MSE) initiator, 0.007 mol/L 2,6-di-(tertiary-butyl)-pyridine proton trap and methylcyclohexane/methyl chloride 60/40 volume/volume solvent mixture, cooled to  $-60^{\circ}\text{C}$ . The reaction was started with the addition of 0.064 mol/L prechilled  $\text{TiCl}_4$ . The reaction was terminated after 10 minutes by prechilled methanol. 2.4 g polymer was recovered. The molecular weight of the

polymer was  $M_n = 4346$  g/mol with  $M_w/M_n = 1.07$ . The hydroxy functionality was determined by FTIR using 2-phenylpropanol to create a calibration curve: functionality  $F_n = 1.1$  was found. The hydroxy group was reacted with trimethylchlorosilane as described<sup>25</sup> and the functionality was determined by <sup>1</sup>H NMR.  $F_n = 1.03$  was found.

The products from Examples 3, 4 and 5 were not elastomeric like the starting PIB 2 but very strong, tough materials that extended like a crosslinked rubber and regained their shape when the stress was released. This is evident from their physical characteristics which are listed in Table 2.

**TABLE 2**

PROPERTIES	PIB 1-Psty 1	PIB 2-Psty 2	PIB 2-Psty 3	PIB 2-Psty 4	PIB 2-Psty 5
100% Modulus (MPa)	0.42	0.62	0.70	0.64	0.28
200% Modulus (MPa)	0.53	0.76	0.89	0.79	0.32
300% Modulus (MPa)	0.58	0.88	1.14	1.02	0.37
Tensile strength (Mpa)	4.0 (PIB 1 0.13)	7.3 (PIB 2 0.15)	9.0 (PIB 2 0.15)	7.0 (PIB 2 0.15)	3.6 (PIB 2 0.15)
Elongation at break (%)	1830	1150	950	980	1380
Hardness (ShoreA)	27	34	-	-	-

Surprisingly, these arborescent blocks displayed thermoplastic elastomeric properties, as evidenced by the results summarized in Table 2 which shows these arborescent TPEs display remarkable tensile strength and elongation values. For comparison, the semicommercial Kuraray sample, most

likely produced by the process disclosed in United States Patent No. 5,721,331 issued to Shachi et al., displayed 10 MPa tensile strength with 600 % elongation at  $M_n = 60,000$  g/mol and 30 wt% PS content.

In addition to the superior elastomeric properties summarized in Table 2,  
5 an advantage of the method of producing the arborescent block copolymer structures disclosed herein over the multiarm-star structures is that multifunctional initiators as used to produce the latter (see S. Jacob, J. P. Kennedy, *Advances in Polym. Sci*, **146**, 1 (1999)) are difficult to make and they are very expensive. Making the multiarm structures by coupling living chains  
10 (United States Patent Nos. 5,428,111 and 5,458,796), is a very slow process, and can take up to 100 hours due to the high viscosity, and the number of arms is undefined. In contrast, it is very easy to make high molecular weight arborescent polyisobutylene block copolymers disclosed herein. They also exhibit better tensile strength and greater elongation in spite of the irregular  
15 structure.

The arborescent branched block copolymers disclosed herein characterized by having thermoplastic elastomeric properties are very useful for many applications and of particular interest as materials for medical devices or prostheses having resistance to cracking and fissuring after implantation. Other  
20 uses and applications will be evident to those skilled in the art where materials having combined thermoplastic elastomeric properties are needed, including gaskets, seals and the like.

As used herein, the terms "comprises" and "comprising" are to be construed as being inclusive and open ended, and not exclusive. Specifically,  
25 when used in this specification including claims, the terms "comprises" and "comprising" and variations thereof mean the specified features, steps or components are included. These terms are not to be interpreted to exclude the presence of other features, steps or components.

The foregoing description of the preferred embodiments of the  
30 invention has been presented to illustrate the principles of the invention and not to limit the invention to the particular embodiment illustrated. It is

intended that the scope of the invention be defined by all of the embodiments encompassed within the following claims and their equivalents.

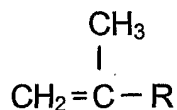
5



**WHAT IS CLAIMED IS:**

1. An arborescent branched block copolymer comprising an arborescent elastomeric polymer block with a low glass-transition temperature ( $T_g$ ) and rigid polymer block segments with a high  $T_g$  or crystalline melting point.
2. The arborescent branched block copolymer according to claim 1 wherein said arborescent elastomeric polymer block with a low glass-transition temperature ( $T_g$ ) includes an arborescent polyisoolefin block, wherein said rigid polymer block segments with a high  $T_g$  or crystalline melting point include polymonovinylidene arene plastic endblocks, and wherein some of the branches of the aforesaid arborescent polyisoolefin block terminate in said polymonovinylidene arene plastic endblocks.
3. The arborescent branched block copolymer of claim 2 wherein the total weight percent of the polymonovinylidene arene plastic endblocks is from about 5 weight percent to about 95 weight percent.
4. The arborescent branched block copolymer of claims 2 or 3 wherein said polymonovinylidene arene plastic endblocks are styrene plastic endblocks.
5. The arborescent branched block copolymer according to claim 1 wherein the arborescent branched block copolymer exhibits thermoplastic elastomeric properties.
6. An arborescent branched block copolymer of a polyisoolefin and a polymonovinylidene arene characterized by having thermoplastic elastomeric properties wherein the aforesaid block copolymer comprises an arborescent polyisoolefin polymer block and some of the branches of the aforesaid arborescent polyisoolefin polymer block terminate in polymonovinylidene arene plastic endblocks.

7. The arborescent branched block copolymer of claim 6 wherein the polyisooolefin comprises an isooolefin having the formula



wherein R is a C<sub>1</sub>-C<sub>4</sub> alkyl group.

8. The arborescent branched block copolymer of claim 7 wherein the isooolefin is selected from the group of isobutylene and 2-methyl-1-butene.

9. The arborescent branched block copolymer of claim 8 wherein the isooolefin is isobutylene.

10. The arborescent branched block copolymer of claims 6, 7, 8 or 9 wherein the polymonovinylidene arene comprises a monovinylidene arene selected from the group consisting of C<sub>8</sub> to C<sub>12</sub> monovinylidene arenes which may be substituted with a C<sub>1</sub> to C<sub>4</sub> alkyl group or alkoxy group or a chlorine or bromine atom on the aromatic ring.

11. The arborescent branched block copolymer of claim 10 wherein the monovinylidene arene is selected from the group consisting of styrene, p-methylstyrene, p-ditertiarybutylstyrene, p-methoxystyrene, p-chlorostyrene and mixtures thereof.

12. The arborescent branched block copolymer of claim 10 wherein the monovinylidene arene is styrene.

13. The arborescent branched block copolymer of claims 6, 7, 8, 9, 10, 11 or 12 wherein the arborescent polyisooolefin polymer block has a branching frequency of from about 5 to about 60.

14. The arborescent branched block copolymer of claim 12 wherein the arborescent polyisoolefin polymer block has a branching frequency of from about 8 to about 35.
15. The arborescent branched block copolymer of claims 6, 7, 8, 9, 10, 11, 12, 13 or 14 wherein the total weight percent of the polymonovinylidene arene endblocks is from about 5 to about 50 weight percent.
16. The arborescent branched block copolymer of claim 6 wherein the number average molecular weight of the arborescent polyisoolefin polymer block is from about 10,000 to about 2,000,000.
17. The arborescent branched block copolymer of claim 16 wherein the number average molecular weight of the arborescent polyisoolefin polymer block is from about 15,000 to about 2,500,000.
18. The arborescent branched block copolymer of claim 6 wherein the molecular weight distribution (MWD) of the arborescent polyisoolefin polymer block is in a range  $1 < \text{MWD} \leq 20$ .
19. The arborescent branched block copolymer of claim 18 wherein the molecular weight distribution of the highly branched polyisoolefin block is from about 1.2 to about 2.8.
20. An arborescent branched block copolymer of polyisobutylene and polystyrene characterized by having thermoplastic elastomeric properties wherein the arborescent branched block copolymer comprises an arborescent polyisobutylene polymer block and some of the branches of the aforesaid arborescent polyisobutylene polymer block terminate in polystyrene endblocks.

21. The arborescent branched block copolymer of claim 20 wherein said arborescent polyisobutylene polymer block has a branching frequency of from about 8 to about 35.

22. The arborescent branched block copolymer of claim 20 wherein the total weight percent of the polystyrene endblocks is from about 5 to about 50 weight percent.

23. The arborescent branched block copolymer claim 20 wherein the molecular weight distribution of the arborescent polyisobutylene is from about 1.1 to about 20.

24. The arborescent branched block copolymer of claim 6 wherein said arborescent branched block copolymer is a first arborescent branched block copolymer, including at least two arborescent branched block copolymers connected together wherein at least one of said polymonovinylidene arene plastic endblocks on one of said at least two arborescent branched block copolymers is chemically bound to one polymonovinylidene arene plastic endblock on the other of said at least two arborescent branched block copolymers.

25. A process for producing an arborescent branched block copolymer of a polyisooolefin and a polymonovinylidene arene characterized by having thermoplastic elastomeric properties wherein the aforesaid arborescent branched block copolymer comprises an arborescent polyisooolefin polymer block and some of the branches of the aforesaid arborescent polyisooolefin polymer block terminate in polymonovinylidene arene plastic endblocks, comprising:

a) polymerizing an isooolefin in the presence of at least one inimer and a Lewis acid halide coiniciator, at a temperature of between about  $-20^{\circ}\text{C}$  and  $-100^{\circ}\text{C}$  to produce an arborescent elastomeric branched polymer, the inimer including at least one group for (co)polymerizing in a cationic polymerization of

the isoolefin and at least one group for initiating cationic polymerization of the isoolefin; and thereafter

b) adding a compound having an effective electron pair donor for improving blocking efficiency and adding a monovinylidene arene suitable for the production of the polyvinylidene plastic blocks; and

c) terminating said polymerization reaction after a selected period of time by addition of an effective nucleophile compound which terminates said polymerization reaction, said selected period of time being sufficiently long enough to ensure production of at least individual units of said arborescent branched block copolymer.

26. The process according to claim 25 wherein said selected period of time is sufficiently long so that multiple arborescent branched block copolymers are bound together wherein at least one of said polymonovinylidene arene plastic endblocks on one arborescent branched block copolymer is chemically bound to one polymonovinylidene arene plastic endblock on another arborescent branched block copolymer.

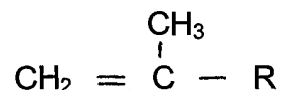
27. The process according to claims 25 or 26 wherein said nucleophile compound is selected from the group consisting of methanol, ethanol, isopropanol and water.

28. The process according to claims 25, 26 or 27 wherein said monovinylidene arenes that polymerize to form the polyvinylidene blocks that form endblocks on some of the branches of the highly branched polyisoolefin are selected from the group consisting of C<sub>8</sub> to C<sub>12</sub> monovinylidene arenes which may be substituted with a C<sub>1</sub> to C<sub>4</sub> alkyl group or alkoxy group or a fluorine, chlorine, bromine or iodine atom on the aromatic ring.

29. The process according to claims 25, 26 or 27 wherein said monovinylidene arenes that polymerize to form the polyvinylidene blocks that

form endblocks on some of the branches of the highly branched polyisooolefin are selected from the group consisting of styrene, p-methylstyrene, p-tertiarybutylstyrene, p-chlorostyrene, indene and mixtures thereof.

30. The process according to claims 25, 26, 27, 28 or 29 wherein said isoolefin used in the synthesis of the highly branched polyisooolefin has a formula



where R represents a C<sub>1</sub> to C<sub>4</sub> alkyl group selected from the group consisting of methyl, ethyl and propyl groups.

31. The process according to claim 29 wherein the isoolefin is selected from the group of isobutylene and 2-methyl-1-butene.

32. The process according to claim 31 wherein the isoolefin is isobutylene.

33. The process according to claims 25, 26, 27, 28, 29, 30, 31 or 32 wherein said at least one inimer is selected from the group consisting of 4-(2-hydroxyisopropyl) styrene and 4-(2-methoxyisopropyl) styrene.

34. The process according to claims 25, 26, 27, 28, 29, 30, 31 or 32 wherein said at least one inimer is selected from the group consisting of 4-(2-hydroxyisopropyl)styrene; 4-(2-methoxyisopropyl)styrene; 4-(2-chloroisopropyl)styrene; 4-(2-acetoxyisopropyl)styrene; 2,3,5,6-tetramethyl-4-(2-hydroxyisopropyl)styrene; 3-(2-methoxyisopropyl)styrene; 4-(epoxyisopropyl)styrene; 4,4,6-trimethyl-6-hydroxyl-1-heptene; 4,4,6-trimethyl-6-chloro-1-heptene; 4,4,6-trimethyl-6,7-epoxy-1-heptene; 4,4,6,6,8-pentamethyl-8-hydroxyl-1-nonene, 4,4,6,6,8-pentamethyl-8-chloro-1-nonene; 4,4,6,6,8-pentamethyl-8,9-epoxy-1-nonene; 3,3,5-trimethyl-5-hydroxyl-1-hexene; 3,3,5-trimethyl-5-chloro-1-hexene; 3,3,5-trimethyl-5,6-epoxy-1-hexene; 3,3,5,5,7-

pentamethyl-7-hydroxyl-1-octene, 3,3,5,5,7-pentamethyl-7-chloro-1-octene;  
3,3,5,5,7-pentamethyl-7,8-epoxy-1-octene.

35. The process according to claims 25, 26, 27, 28, 29, 30, 31, 32, 33 or 34 wherein said compound having an electron pair donor includes moieties which function as proton traps for reducing homopolymerization.

36. The process according to claims 25, 26, 27, 28, 29, 30, 31, 32, 33 or 34 wherein said compound having an electron pair donor includes nucleophiles that have an electron donor number of at least 15.

37. The process according to claim 36 wherein said compound having an electron pair donor is selected from the group consisting of ethyl acetate, dimethylacetamide, dimethylformamide, dimethyl sulphoxide and dimethyl acetamide.

38. The process according to claim 35 wherein said compound having an electron pair donor which include moieties which function as proton traps are selected from the group consisting of 2,6-ditertiarybutylpyridine, 4-methyl-2,6-ditertiarybutylpyridine and diisopropylethylamine, pyridine, and 2,6-dimethylpyridine.

39. The process according to claims 25, 26, 27, 28, 29, 30, 31, 32, 33, 34, 35, 36, 37 or 38 wherein said Lewis acid halide coiniciator is selected from the group consisting of boron trichloride, aluminum trichloride and titanium tetrachloride.

40. The process according to claim 25 wherein said Lewis acid halide coiniciator is titanium tetrachloride.

41. The process according to claim 25 wherein said at least one inimer is a compound having a formula A - R1 - B, where R1 can be an aliphatic and

aromatic moiety, A is a vinyl group that can be polymerized under cationic condition, wherein B is an oxirane ring so that each branching point of the polyisobutylene elastomeric blocks includes a hydroxy group attached thereto.

42. The process according to claim 25 wherein the isoolefin is polymerized in the presence of one or more inert, organic solvents.

43. The process according to claim 42 wherein the one or more inert, organic solvents is a mixture of a nonpolar organic solvent and a polar organic solvent.

44. The process according to claim 43 wherein said nonpolar organic solvent is selected from the group consisting of hexane, methylcyclohexane, cyclopentane and cyclohexane.

45. The process according to claim 43 wherein said polar organic solvent is selected from the group consisting of n-butylchloride, ethyl chloride, methyl chloride and methylene chloride.

46. The process according to claim 43 wherein the one or more inert, organic solvents is a mixture of methylcyclohexane and methyl chloride.

47. The process according to claim 43 wherein said mixture of a nonpolar organic solvent and a polar organic solvent has a ratio of the nonpolar solvent to the polar solvent on a weight basis from about 80:20 to about 40:60, preferably about 60:40.

48. The process according to claim 47 wherein said ratio of the nonpolar solvent to the polar solvent on a weight basis is about 60:40.

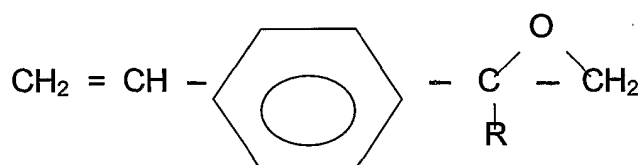


49. The process according to claim 48 wherein a ratio of the coiniciator to the monovinylidene arene on a molar basis is from about 1:1 to about 1:30, preferably from about 1:10 to about 1:20.

50. The process according to claim 49 wherein the ratio of the coiniciator to the monovinylidene arene on a molar basis is from about 1:10 to about 1:20.

51. A process for producing an arborescent branched polyisooolefin block copolymer characterized by having elastomeric properties, comprising:

a) polymerizing an isooolefin in the presence of a multifunctional monomer and a Lewis acid halide coiniciator at a temperature of between about  $-20^{\circ}\text{C}$  and  $-100^{\circ}\text{C}$  to produce an elastomeric arborescent branched polyisooolefin polymer, the multifunctional monomer having a formula



where R can be H,  $\text{CH}_3$  or an alkyl or aryl group; thereafter

b) adding a compound having an effective electron pair donor for improving blocking efficiency and adding a monovinylidene arene suitable for the production of the polyvinylidene arene plastic endblocks; and

c) terminating said polymerization reaction after a selected period of time by addition of an effective nucleophile compound which terminates said polymerization reaction, said selected period of time being sufficiently long enough to ensure production of at least individual units of said arborescent branched block copolymer.

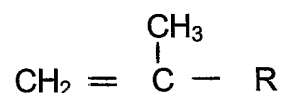
52. The process according to claim 51 wherein said selected period of time is sufficiently long so that multiple arborescent branched block copolymers are bound together wherein at least one of said polymonovinylidene arene plastic

endblocks on one arborescent branched block copolymer is chemically bound to one polymonovinylidene arene plastic endblock on another arborescent branched block copolymer.

53. The process according to claims 51 or 52 wherein said Lewis acid halide coinitiator is selected from the group consisting of boron trichloride, aluminum trichloride and titanium tetrachloride.

54. The process according to claims 51 or 52 wherein said Lewis acid halide coinitiator is titanium tetrachloride.

55. The process according to claims 51, 52, 53 or 54 wherein said isoolefin used in the synthesis of the highly branched polyisoolefin has a formula



where R represents a C<sub>1</sub> to C<sub>4</sub> alkyl group selected from the group consisting of methyl, ethyl and propyl.

56. The process according to claim 55 wherein said C<sub>1</sub> to C<sub>4</sub> alkyl group is selected from the group consisting of isobutylene and 2-methyl-1-butene.

57. The process according to claim 51 wherein the isoolefin is isobutylene.

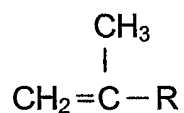
58. The process according to claim 55 wherein the isoolefin is isobutylene.

59. The process according to claim 51 wherein said at least one multifunctional monomer is selected from the group consisting of 4-(2-hydroxy-isopropyl) styrene and 4-(2-methoxy-isopropyl) styrene.

60. The process according to claims 51, 52, 53, 54, 55, 56, 57, 58 or 59 wherein the isoolefin is polymerized in the presence of one or more inert, organic solvents.
61. The process according to claim 60 wherein the one or more inert, organic solvents is a mixture of a nonpolar organic solvent and a polar organic solvent.
62. The process according to claim 61 wherein said nonpolar organic solvent is selected from the group consisting of methylcyclohexane and cyclohexane.
63. The process according to claim 61 wherein said polar organic solvent is selected from the group consisting of ethyl chloride, methyl chloride and methylene chloride.
64. The process according to claim 60 wherein the one or more inert, organic solvents is a mixture of methylcyclohexane and methyl chloride.
65. The process according to claim 61 wherein said mixture of a nonpolar organic solvent and a polar organic solvent has a ratio of the nonpolar solvent to the polar solvent on a weight basis from about 80:20 to about 40:60, preferably about 60:40.
66. The process according to claim 65 wherein said ratio of the nonpolar solvent to the polar solvent on a weight basis is about 60:40.
67. A non-biodegradable article of manufacture comprising an implantable prosthesis or implant formed of an arborescent branched block copolymer of a polyisoolefin and a polymonovinylidene arene characterized by having thermoplastic elastomeric properties wherein the aforesaid arborescent branched block copolymer comprises an arborescent branched polyisoolefin polymer block

and some of the branches of the aforesaid arborescent branched polyisooolefin block terminate in polymonovinylidene arene plastic endblocks.

68. The article of manufacture of claim 67 wherein the polyisooolefin comprises an isooolefin having the formula



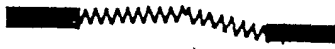
wherein R is a C<sub>1</sub>-C<sub>4</sub> alkyl group.

69. The article of manufacture of claim 68 wherein the isooolefin is selected from the group of isobutylene and 2-methyl-1-butene.

70. The article of manufacture of claim 69 wherein the isooolefin is isobutylene.

71. The article of manufacture of claims 67, 68, 69 or 70 wherein said article is constructed as an endoluminal or vascular device for in vivo implant.

First generation

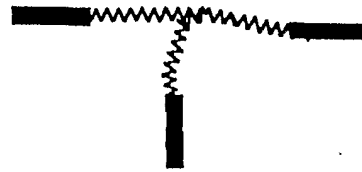


PS PIB

Linear triblock

PRIOR ART

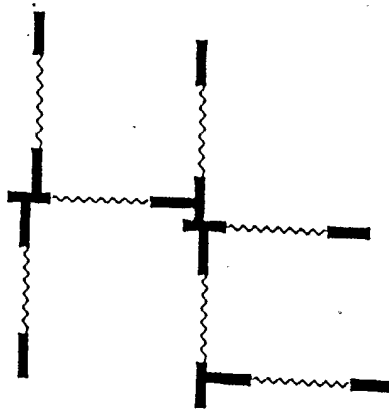
FIGURE 1A



Triarm-star block

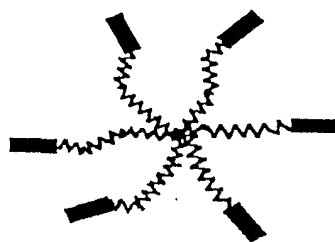
PRIOR ART

FIGURE 1B



PRIOR ART

FIGURE 1C



Multiarm-star block

PRIOR ART

FIGURE 2 A

3/7

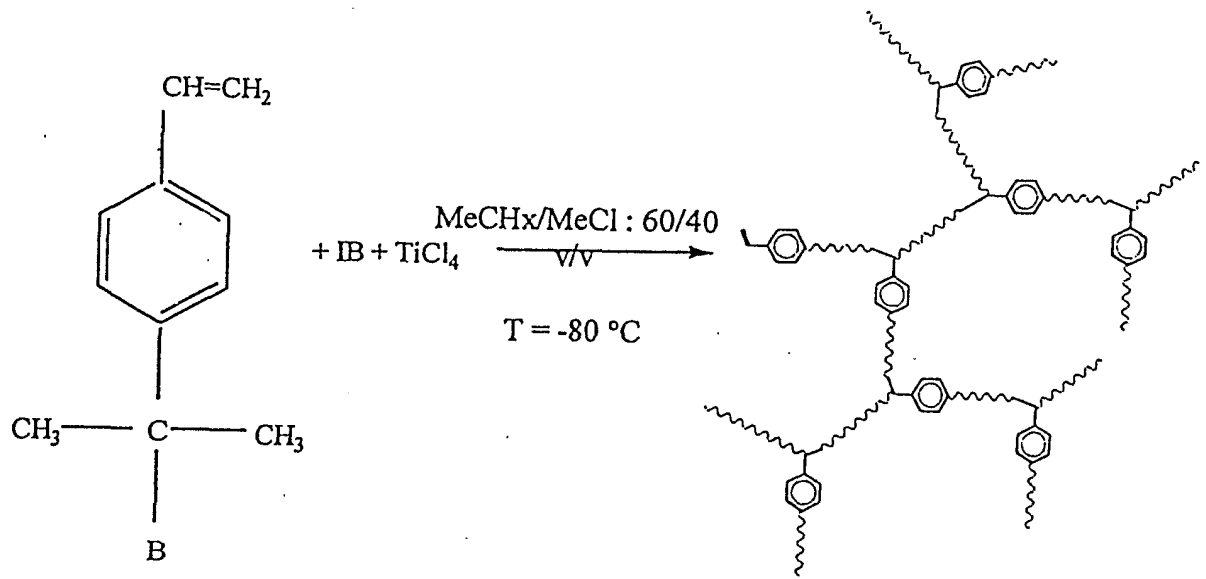


FIGURE 3

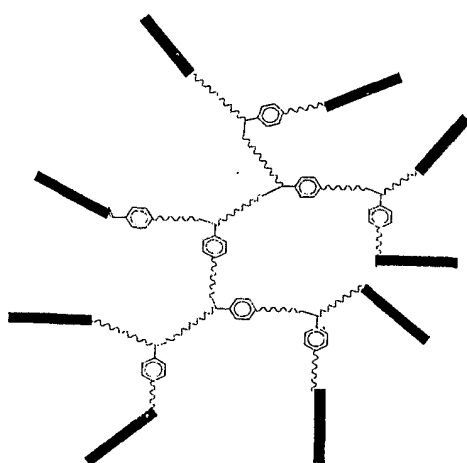
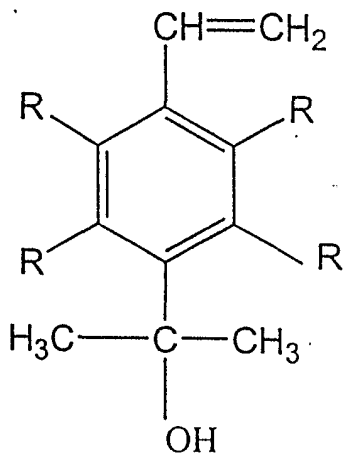


FIGURE 4

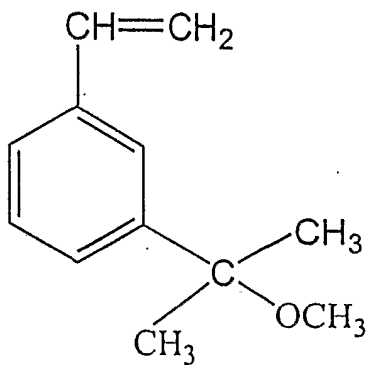




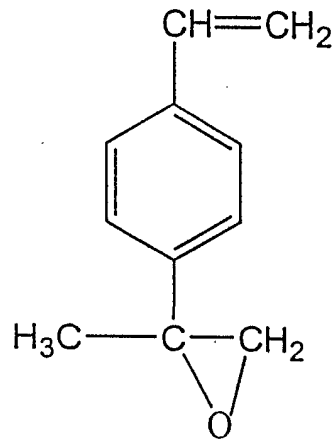
6/7



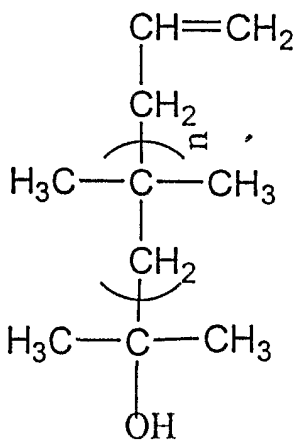
R = methyl  
2,3,5,6-tetramethyl-  
4-(2-hydroxyisopropyl)styrene



3-(2-methoxyisopropyl)  
styrene

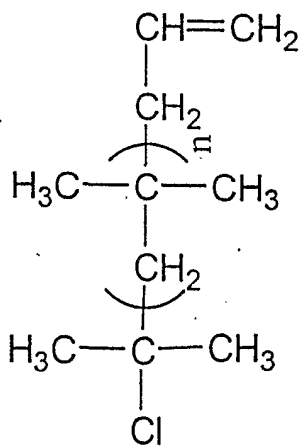


4-(epoxyisopropyl)styrene

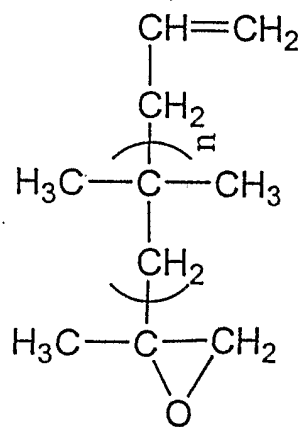


n = 1

4,4,6-trimethyl-6-  
hydroxyl-1-heptene



4,4,6-trimethyl-6-  
chloro-1-heptene



4,4,6-trimethyl-6-  
6,7-epoxy-1-heptene

n = 2

4,4,6,6,8-pentamethyl-  
8-hydroxyl-1-nonene

4,4,6,6,8-pentamethyl-  
8-chloro-1-nonene

4,4,6,6,8-pentamethyl-  
8,9-epoxy-1-nonene

FIGURE 5 (continued)

7/7

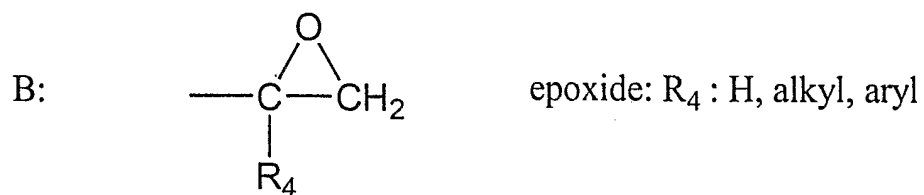
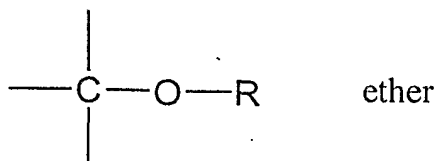
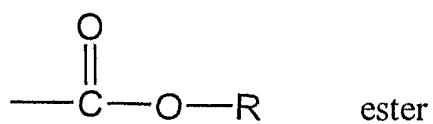
 $R_1, R_3 \longrightarrow$  H, alkyl, aryl $R_2 \longrightarrow$  X halide (Cl, Br, F, I)

FIGURE 5 (continued)

## INTERNATIONAL SEARCH REPORT

Inter Application No

PCT/CA 02/00792

## A. CLASSIFICATION OF SUBJECT MATTER

IPC 7 C08F297/00 C08G83/00

According to International Patent Classification (IPC) or to both national classification and IPC

## B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC 7 C08F C08G C08L C09J C09D

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

EPO-Internal, WPI Data, PAJ, CHEM ABS Data

## C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category °	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	US 4 946 899 A (KENNEDY JOSEPH P ET AL) 7 August 1990 (1990-08-07) whole document	1-66
Y	--- PUSKAS J E ET AL: "STAR-BRANCHED AND HYPERBRANCHED POLYISOBUTYLENES" MACROMOLECULAR SYMPOSIA, WILEY VCH, WEINHEIM, DE, vol. 132, 1 July 1998 (1998-07-01), pages 117-126, XP000790730 ISSN: 1022-1360 whole document	1-66
Y	--- US 5 458 796 A (STOREY ROBSON F ET AL) 17 October 1995 (1995-10-17) whole document --- -/--	1-66

 Further documents are listed in the continuation of box C. Patent family members are listed in annex.

° Special categories of cited documents:

- \*A\* document defining the general state of the art which is not considered to be of particular relevance
- \*E\* earlier document but published on or after the international filing date
- \*L\* document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)
- \*O\* document referring to an oral disclosure, use, exhibition or other means
- \*P\* document published prior to the international filing date but later than the priority date claimed

- \*T\* later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention
- \*X\* document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone
- \*Y\* document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.
- \*&\* document member of the same patent family

Date of the actual completion of the international search

1 October 2002

Date of mailing of the international search report

09/10/2002

Name and mailing address of the ISA

European Patent Office, P.B. 5818 Patentlaan 2  
NL - 2280 HV Rijswijk  
Tel. (+31-70) 340-2040, Tx. 31 651 epo nl,  
Fax: (+31-70) 340-3016

Authorized officer

Gold, J

## INTERNATIONAL SEARCH REPORT

Interr    Application No  
PCT/CA 02/00792

## C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT

Category °	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
P, X	WO 02 32982 A (AVECIA LTD ; STEENWINKEL PABLO (NL); TENNEBROEK RONALD (NL); AVECIA) 25 April 2002 (2002-04-25) whole document -----	1-66

# INTERNATIONAL SEARCH REPORT

Information on patent family members

Inter:      Application No
PCT/CA 02/00792

Patent document cited in search report		Publication date		Patent family member(s)	Publication date
US 4946899	A	07-08-1990	WO	9111468 A1	08-08-1991
			US	RE34640 E	14-06-1994
<hr style="border-top: 1px dashed black;"/>					
US 5458796	A	17-10-1995	AU	693156 B2	25-06-1998
			AU	1613995 A	12-10-1995
			BR	9501267 A	31-10-1995
			CA	2145811 A1	01-10-1995
			CN	1115321 A	24-01-1996
			DE	69521231 D1	19-07-2001
			DE	69521231 T2	16-05-2002
			EP	1099717 A2	16-05-2001
			EP	0675153 A1	04-10-1995
			ES	2159603 T3	16-10-2001
			JP	7268047 A	17-10-1995
			SG	24115 A1	10-02-1996
			ZA	9502553 A	01-12-1995
<hr style="border-top: 1px dashed black;"/>					
WO 0232982	A	25-04-2002	AU	9399301 A	29-04-2002
			WO	0232982 A1	25-04-2002
<hr style="border-top: 1px dashed black;"/>					