A medicament base containing from 5 to 40 percent saturated fatty alcohol having from 16 to 24 carbons, from 1 to 40 percent propylene carbonate, from 25 to 85 percent of glycol cosolvent, a stabilizing amount of a surfactant and optional amounts of compatible plasticizer, and/or other pharmaceutical adjuvants. The base is a suitable vehicle for all types of therapeutic agents for topical application and has shown particular advantages with anti-inflammatory topical corticoids.

5 Claims, No Drawings
1. **FATTY ALCOHOL-PROPYLENE CARBONATE GLYCOL SOLVENT CREAM VEHICLE**

**BACKGROUND OF THE INVENTION**

This invention relates to vehicles for topical application of medicaments and to mixtures of the vehicle and medicaments. In particular, this invention relates to new, improved medicament vehicles having advantages over previously known vehicles.

One of the oldest types of medicament vehicles is the ointment preparation containing active medicaments that can be readily applied and rubbed into the skin. It serves as a means for distributing the medication uniformly over the skin surface and maintaining it there until beneficial action can occur. The earliest ointment preparations were based on fats, waxes, greases and petrolatum. These are, by nature, greasy, or not water-washable and having a limited ability to release medication to the skin. A non-aqueous ointment of more recent origin is a mixture of polyethylene glycols having molecular weights of 1,000 to 20,000. This vehicle, although water-washable, has a greasy texture and does not provide an occlusive dressing on a treated surface. Prior to this invention, these anhydrous ointment bases were the only bases available for medicaments which deteriorated in the presence of moisture.

Emulsified creams, such as cold creams, were developed to reduce greasiness, while still maintaining the uncouthness and spreadability of the older greasy-type ointments. The emulsified creams have an aqueous base, however, and are not suitable for many drugs because their water content destroy the medicament. The medicament, in turn may destroy the emulsions, that is, break the emulsions and permit separation of the vehicle components. Furthermore, water is frequently not desirable in a medicament formulation because of its adverse effects on a condition being treated.

One system which is not subject to the above disadvantages is the non-aqueous fatty alcohol-propylene glycol vehicles described in U.S. Pat. No. 3,592,930 granted to Katz et al. The subject of this invention is an improved non-aqueous vehicle with a propylene carbonate solvent system.

It is accordingly the purpose of this invention to provide an essentially anhydrous, water-washable base which is more effective than standard anhydrous ointment bases of the grease type because it can preserve the activity of medicaments which deteriorate in the presence of moisture; provide an occlusive film for longer and better therapeutic activity; release the medicaments more quickly and effectively; bring dissolved therapeutic agents in known dilution in contact with the skin; spread evenly and adhere well even if the skin is moist; be readily removed from the skin or fabrics with water; provide media to readily absorb discharges from wounds; serve as an excellent levigating material for many prescribed ingredients that usually require separate treatment before being incorporated into one of the bases; provide a base for medicament formulations in which water is not desired; and because it does not hydrolyze, deteriorate, become rancid, support mold growth or require preservatives.

It is a further object of this invention to provide a vehicle using a unique solvent, new for topical preparations. Propylene carbonate has exceptional solubilizing properties, particularly for corticosteroids. By combining a glycol cosolvent and surfactant with the fatty alcohol and propylene carbonate, a stable cream can be prepared. Further, by varying the ratio of propylene carbonate and glycol cosolvent one can obtain a wide range of saturation concentrations for a medicament. Thus the ratio can be chosen to optimize drug delivery for any particular medicament.

**SUMMARY**

The composition of this invention is a substantially anhydrous vehicle composition consisting essentially of:

1. a. from 5 to 40 weight percent of saturated fatty alcohol having from 16 to 24 carbons;

2. b. from 1 to 40 weight percent of propylene carbonate;

3. c. from 25 to 85 weight percent of glycol cosolvent, the weight ratio of the glycol solvent to propylene carbonate being at least 1:2;

4. d. a stabilizing amount of surfactant; and

5. e. from 0 to 15 weight percent of compatible plasticizer.

The base is an improved vehicle for all types of therapeutic agents for topical application and offers particular advantages with anti-inflammatory topical steroids.

**DESCRIPTION OF THE PREFERRED EMBODIMENTS**

All concentrations are herein given as weight percents unless otherwise specified. It is also intended that the chemical compounds in each class of ingredients discussed hereinafter be limited to pharmaceutically acceptable, non-toxic compounds in the concentrations indicated.

The composition of this invention contains from 5 to 40 and preferably from 10 to 30 percent fatty alcohol. The fatty alcohol can be any fatty alcohol having from 16 to 24 carbons or mixtures thereof, and is preferably a saturated monohydric primary alcohol. Suitable fatty alcohols include cetyl alcohol, stearyl alcohol, behenyl alcohol, and the like.

The fatty alcohol component should be substantially free from any significant amount of unsaturated alcohols or fatty alcohols having fewer than 16 carbons, the term "substantially free from" as used herein, is defined as indicating the compositions of this invention containing less than irritating or otherwise medically undesirable amounts of the indicated substances. Since the commercially available fatty alcohols having from 16 to 24 carbons contain impurities including some proportion of fatty alcohols having fewer than 16 carbons, total avoidance of alcohols having fewer than 16 carbons from the mixture is not practicable. Careful selection of raw materials is preferable, however, to maintain the percentage of irritating alcohols to less than 10 percent of the total fatty alcohol concentration.

The composition of this invention also contains from 1 to 40 and preferably from 5 to 30 percent propylene carbonate.

The composition of this invention also contains from 25 to 85 preferably from 30 to 80 percent of glycol cosolvent. Suitable glycol cosolvents include 1,2-propanediol, 1,3-propanediol, polyethylene glycol having a molecular weight of from 100 to 800, dipropylene glycol, and the like or mixtures thereof. The weight ratio of glycol cosolvent to propylene carbonate must be at least 1:2 to provide a stable composition. It is preferably at least 1:1, and the optimum ratio is at least...
3:1. In the absence of the glycol cosolvent, the propylene carbonate and fatty alcohol do not form a physically stable mixture. Thus, the glycol cosolvent functions primarily as a coupling ingredient for the fatty alcohol and propylene carbonate. It also functions as an auxiliary solvent in the system.

The composition of this invention also contains a stabilizing amount of a surfactant, that is, an amount sufficient to maintain homogeneity of the other ingredients. The particular concentration will vary depending upon the choice of surfactant and the selection of the other ingredients. In general, stabilizing amounts can be as low as 0.1 percent or lower. In some instances as high as 10 percent or higher of surfactant may be desired. Generally from 2 to 5 percent is suitable. The amount of surfactant should be the minimum required for stability. The surfactant functions as a coupling agent, linking diverse phases and maintaining a dispersion of immiscible components. Suitable surfactants include pharmaceutically acceptable, non-toxic non-ionic, anionic and cationic surfactants. Examples of suitable non-ionic surfactants include glycerol fatty acid esters such as glycerol monostearate, glycerol fatty acid esters such as propylene glycol monostearate, polyhydric alcohol fatty acid esters such as sorbitan monostearate, polyethylene glycol fatty acid esters such as polyethylene glycol (400) monooctylate, polyethylene fatty acid esters such as polyoxyethylene (40) stearate, polyoxyethylene fatty alcohol ethers such as polyoxyethylene (20) stearyl ether, polyoxyethylene sorbitan fatty acid esters such as polyoxyethylene sorbitan monostearate, fatty acid etholamides and their derivatives such as the diethanolamide of stearic acid, and the like.

Examples of suitable anionic surfactants are soaps including alkali soaps, such as sodium, potassium and ammonium salts of aliphatic carboxylic acids, usually fatty acids, such as sodium stearate. Organic amine soaps, also included, include organic amine salts of aliphatic carboxylic acids, usually fatty acids, such as triethanolamine stearate. Another class of suitable soaps are the metallic soaps, salts of polyvalent metals and aliphatic carboxylic acids, usually fatty acids, such as aluminum stearate. Other classes of suitable anionic surfactants include sulfated fatty alcohols such as sodium lauryl sulfate, sulfated oils such as the sulfuric ester of ricinoleic acid disodium salt, and sulfonated compounds such as alkyl sulfonates including sodium octane sulfonate, amide sulfonates such as sodium N-methyl-N-oleyltaurate, sulfonated dibasic acid esters such as sodium dioctyl sulfosuccinate, alkyl aryl sulfonates such as sodium dodecylbenzene sulfonate, alkyl naphthalene sulfonates such as sodium isopropyl naphthalene sulfonate, petroleum sulfonates such as aryl naphthene with alkyl substituents. Examples of suitable cationic surfactants include amine salts such as octyldecyl ammonium chloride, quaternary ammonium compounds such as benzalkonium chloride. Other examples of these and other suitable surfactants can be found in "Pharmaceutical Emulsions and Emulsifying Agents" by Lawrence M. Spotton, second edition, The Chemist and Druggist, London; "Emulsions; Theory and Practice" by Paul Becher, Reinhold Publishing Corporation, New York; and "Detergents and Emulsifiers, 1969 Annual" by John M. McCutcheon, Morrison, N.J., the disclosures thereof being incorporated herein by reference.

The composition of this invention can also contain from 0 to 15 and preferably from 0.1 to 5 percent of a compatible plasticizer. Suitable compatible plasticizers include carboxylic vinyl polymers (Carbopols), polyethylene glycol having a molecular weight of from above 800 to 20,000; natural gums including acacia gum, guar gum, karaya, tragacanth, and the like; seaweed products such as agar, Irish moss and alginites; cellulose derivatives including cellulose ethers such as methyl cellulose, ethyl cellulose, sodium carboxymethyl cellulose and the like; starch, starch derivatives and dextrins; pectin and pectates; saponins; and water soluble or water dispersible vinyl polymers such as polyvinylpyrrolidone, polyvinyl alcohol, vinyl pyrrolidonevinyl alcohol copolymers, and the like. The plasticizer maintains homogeneity in the mixture at ambient temperatures, that is, temperatures at which the fatty alcohol is a solid. This component also improves the plasticity, and uniformity of the medicament mixture with the vehicle and provides to the vehicle smoothness and a more pleasing "feel;" hence the vehicle containing the plasticizer is more cosmetically acceptable. In general, the particular plasticizer concentration necessary to provide a desired consistency, degree of smoothness and plasticity will vary with the choice of the fatty alcohol component and cosolvent, and the ratio of these components in the vehicle. Preferably, the plasticizer concentration should be balanced so the vehicle has freeze-thaw stability, i.e., does not separate after repeated cycles of solidification (by cooling) and liquefaction (by heating). The term "compatible" is defined herein to indicate a component which will not cause separation (loss of homogeneity) of the other components at temperatures up to 45°C.

It should be understood that the medicament vehicles of this invention can also contain other non-essential ingredients. The vehicle can contain up to 10 weight percent of conventional pharmaceutical adjuvants. These adjuvants or additives are used to improve consistency, emolliency, homogeneity, spreadability, texture and appearance of the vehicle or its residual film or the stability of the medicament. They can be used to give a residual film, varying degrees of continuity, flexibility, adhesion, occlusion, water repellency, washability, and the like. Suitable auxiliary adjuvants include hydrocarbons ranging from liquid petrolatum to solid paraffins and waxes, beeswax, saturated fatty acids having from 16 to 24 carbons such as stearic acid, palmitic acid, behenic acid; fatty acid amides such as oleamide, palmitamide, stearamide, behenamide; and esters of fatty acids having from 14 to 24 carbons such as isopropyl myristate sorbitan monostearate, polyethylene glycol monostearate, propylene glycol monostearate and the corresponding mono- and diesters of other fatty acids such as oleic and acid and palmitic acid. It is preferable that the fatty acids be saturated and the fatty acids and amides be substantially free of irritating amounts of acids or amides having fewer than 14 carbons. Other optional adjuvants include miscellaneous natural products such as wool fat, wool alcohol, cholesterol and its derivatives, lecithin and proteins such as gelatin, casein, soyabean protein, egg albumen. Finely dispersed mineral solids useful as thickeners include colloidal clays such as bentonite and polyvalent metal hydroxides such as magnesium hydroxide. Suitable chemical stabilizers include citric acid and other agents to adjust pH, ethylenediamine tetraacetic acid and its salts and other chelating or sequestering agents, propyl gallate, butylated hydroxy anisole or toluene, and other...
antioxidants.

The medicament vehicle of this invention is essentially a non-aqueous base, that is, it is not an aqueous emulsion and consequently is not a "cream" in the usual sense. It is preferably totally anhydrous, but can contain minor amounts of water such as up to 3 percent water. The water concentration should not be sufficient to cause separation of the other vehicle components or precipitate medicaments dissolved in the vehicle.

The vehicle of this invention can be made thoroughly mixing the components at ambient or elevated temperatures. Preferably the components are thoroughly mixed while each is in a liquid state, and the mixture is cooled with good agitation to room temperature. Preferably, additional mechanical agitation and/or shock cooling steps are used as intermediate or final steps in the manufacturing process to impart more homogeneity or improved texture. Process equipment for these techniques includes heat exchangers, propeller mixers, colloid mills, homogenizers, roller mills and the like.

The base of this invention can be used as a vehicle for all types of medicaments or therapeutic agents for topical application including antibiotics such as oxytetracycline, chlortetracycline, streptomycin, bacitracin, chloramphenicol, tyrothricin and the like; steroids having anti-inflammatory or other beneficial activity; antihistamines such as prophenpyridamine maleate and diphenhydramine hydrochloride; anesthetics such as benzocaine and lidocaine; antibacterials including iodine; iodochlorhydroxyquin, nitrofurazone, sulfanilamide and derivatives, and benzalkonium chloride; fungicides such as undecylenic acid vitamins such as Vitamin A derivatives; and other therapeutic agents including coal tar, balsam Peru, ammoniated mercury, anthralin, chrysarobin, ichthammol, sulfur and the like.

The medicaments can be incorporated into this base by conventional techniques. A bulky, insoluble powder should be mixed beforehand with a small proportion of the base mixture, propylene carbonate, or propylene glycol, and then blended with the remainder of the base. The products are usually improved by passing them through an ointment or roller mill. Coal tar, ichthammol, balsam Peru and others that require special processing in greasy bases can be readily incorporated in the base of this invention. The medicaments can be incorporated into the final base or introduced into the base mixture with one of its components. Heat sensitive medicaments (in particular some antibiotics) can be dissolved or suspended in a small amount of propylene carbonate, glycol cosolvent or other liquid, and then mixed with the vehicle during or after its preparation.

The amount of medicament to be incorporated into the base will, of course, depend upon the type of medicament and its intended use; the determination of suitable medicament concentrations is a routine matter fully within the conventional skill of the art. In general, therapeutically effective amounts of the medicament are incorporated into the vehicle.

The vehicle of this invention is particularly suitable for use with anti-inflammatory topical steroids represented by Formulas I, II and III.

In the above formulas

\( R_1 \) is hydrogen, methyl, fluoro, or chloro and when \( Z_2 \) is a single bond, \( R_1 \) can be either \( \alpha \) or \( \beta \) oriented;

\( R_2 \) is hydrogen, chloro, or fluoro;

\( R_3 \) is keto or

\[ \begin{array}{c}
R_3' \\
\text{H}
\end{array} \]

wherein \( R_4' \) is hydrogen, hydroxy, chloro, or fluoro;

\( R_4 \) is hydrogen, methyl, hydroxy, or a conventional hydrolyzable ester thereof;
**Example 1**

The following ingredients are mixed at 80°C and cooled to room temperature with good agitation.

<table>
<thead>
<tr>
<th>ingredients</th>
<th>A</th>
<th>B</th>
<th>C</th>
<th>D</th>
<th>E</th>
</tr>
</thead>
<tbody>
<tr>
<td>Stearyl alcohol</td>
<td>16.0</td>
<td>20.0</td>
<td>25.0</td>
<td>20.0</td>
<td>16.0</td>
</tr>
<tr>
<td>Sorbitan monostearate</td>
<td>2.2</td>
<td>0.4</td>
<td>2.2</td>
<td>0.5</td>
<td>2.2</td>
</tr>
<tr>
<td>Polyoxyethylene sorbitan monostearate (Twee 60)</td>
<td>1.8</td>
<td>1.8</td>
<td>1.8</td>
<td>0.3</td>
<td>1.8</td>
</tr>
<tr>
<td>Propylene glycol</td>
<td>64.0</td>
<td>61.4</td>
<td>46.0</td>
<td>62.5</td>
<td>40.0</td>
</tr>
<tr>
<td>Propylene carbonate</td>
<td>16.0</td>
<td>16.0</td>
<td>25.0</td>
<td>16.0</td>
<td>40.0</td>
</tr>
<tr>
<td>Carboxy vinyl polymer (Carbopol)</td>
<td></td>
<td>0.4</td>
<td></td>
<td>0.7</td>
<td></td>
</tr>
</tbody>
</table>
EXAMPLE 2
A vehicle having the composition “B” of Example 1 and containing 0.025 g. of 6α-fluoro-11β-hydroxy-16α,17α-isopropylidenedioxy-21-acetoxypregna-1,4-diene-3,20-dione is prepared as follows.

The propylene glycol and propylene carbonate are mixed and heated to 80°-85°C, and the steroid is dissolved in the mixture. The carboxy vinyl polymer (Carbol) and stearyl alcohol are blended, and together with the sorbitan monostearate and polyoxyethylene sorbitan monostearate, are mixed with the steroid solution. The mixture is then cooled to room temperature while maintaining dispersion with suitable mixing equipment.

EXAMPLE 3
The following ingredients are mixed at 80°C and cooled to room temperature with good agitation.

<table>
<thead>
<tr>
<th>Ingredients</th>
<th>Concentration, Wt. Percent</th>
</tr>
</thead>
<tbody>
<tr>
<td>Propylene carbonate</td>
<td>15.0</td>
</tr>
<tr>
<td>Dipropylene glycol</td>
<td>53.0</td>
</tr>
<tr>
<td>Propylene glycol</td>
<td>30.0</td>
</tr>
<tr>
<td>Stearyl alcohol</td>
<td>3.0</td>
</tr>
<tr>
<td>Carboxy vinyl polymer</td>
<td>1.0</td>
</tr>
<tr>
<td>Sorbitan monostearate</td>
<td>1.0</td>
</tr>
</tbody>
</table>

EXAMPLE 4
Each of 0.25, 0.5 and 1.0 gm. quantities of the following anti-inflammatory steroids, when incorporated into 1,000 gm. the mixtures described in Example 1, are effective for topical treatment of inflammation:

- 9α-11β-dichloro-6α-fluoro-21-hydroxy-16α,17α-isopropylidenedioxy-21-acetoxypregna-1,4-diene-3,20-dione
- 6α-fluoro-11β,17α,21-trihydroxy-16α-methylpregna-1,4-diene-3,20-dione
- 6α-fluoro-11β,17α,21-dihydroxy-16β-methylvaleroyldeoxy-21-acetoxypregna-1,4-diene-3,20-dione
- 17α,21-dihydroxy-21-acetoxypregn-4-ene-3,11,20-trione
- 21-hydroxy-21-acetoxyregn-4-ene-3,11,20-trione
- 21-acetoxyregn-4-ene-3,20-dione
- 21-pivaloxyregn-4-3,20-dione
- 6α-fluoro-11β,17α,21-trihydroxy-16α-methylpregna-1,4-diene-3,20-dione
- 6α-fluoro-11β,17α,21-trihydroxy-16α-methylpregna-1,4-diene-3,20-dione
- 6α-fluoro-11β,17α,21-dihydroxy-16α-methylvaleroyldeoxy-21-acetoxypregna-1,4-diene-3,20-dione
- 6α-fluoro-11β,17α,21-dihydroxy-16α,17α-isopropylidenedioxy-21-acetoxypregna-1,4-diene-3,20-dione
- 6α-fluoro-11β,17α,21-trihydroxy-16α,17α-isopropylidenedioxy-21-acetoxypregna-1,4-diene-3,20-dione
- 6α-fluoro-11β,17α,21-trihydroxy-16α,17α-isopropylidenedioxy-21-acetoxypregna-1,4-diene-3,20-dione
- 6α-fluoro-11β,17α,21-trihydroxy-16α,17α-isopropylidenedioxy-21-acetoxypregna-1,4-diene-3,20-dione
- 6α-fluoro-11β,17α,21-trihydroxy-16α,17α-isopropylidenedioxy-21-acetoxypregna-1,4-diene-3,20-dione
- 6α-fluoro-11β,17α,21-trihydroxy-16α,17α-isopropylidenedioxy-21-acetoxypregna-1,4-diene-3,20-dione
- 6α-fluoro-11β,17α,21-trihydroxy-16α,17α-isopropylidenedioxy-21-acetoxypregna-1,4-diene-3,20-dione
- 6α-fluoro-11β,17α,21-trihydroxy-16α,17α-isopropylidenedioxy-21-acetoxypregna-1,4-diene-3,20-dione
- 6α-fluoro-11β,17α,21-trihydroxy-16α,17α-isopropylidenedioxy-21-acetoxypregna-1,4-diene-3,20-dione
- 6α-fluoro-11β,17α,21-trihydroxy-16α,17α-isopropylidenedioxy-21-acetoxypregna-1,4-diene-3,20-dione
- 6α-fluoro-11β,17α,21-trihydroxy-16α,17α-isopropylidenedioxy-21-acetoxypregna-1,4-diene-3,20-dione
- 6α-fluoro-11β,17α,21-trihydroxy-16α,17α-isopropylidenedioxy-21-acetoxypregna-1,4-diene-3,20-dione
- 6α-fluoro-11β,17α,21-trihydroxy-16α,17α-isopropylidenedioxy-21-acetoxypregna-1,4-diene-3,20-dione
- 6α-fluoro-11β,17α,21-trihydroxy-16α,17α-isopropylidenedioxy-21-acetoxypregna-1,4-diene-3,20-dione
- 6α-fluoro-11β,17α,21-trihydroxy-16α,17α-isopropylidenedioxy-21-acetoxypregna-1,4-diene-3,20-dione
- 6α-fluoro-11β,17α,21-trihydroxy-16α,17α-isopropylidenedioxy-21-acetoxypregna-1,4-diene-3,20-dione
- 6α-fluoro-11β,17α,21-trihydroxy-16α,17α-isopropylidenedioxy-21-acetoxypregna-1,4-diene-3,20-dione
- 6α-fluoro-11β,17α,21-trihydroxy-16α,17α-isopropylidenedioxy-21-acetoxypregna-1,4-diene-3,20-dione

EXAMPLE 5
Repeating the procedure of Example 1 with:

- a. from 5 to 40 percent of a fatty alcohol having from 16 to 24 carbons, e.g. cetyl alcohol, stearyl alcohol, behenyl alcohol, etc.
- b. from 1 to 40 percent of propylene carbonate
- c. from 25 to 85 percent glycol cosolvent such as 1,2-propanediol, 1,3-propanediol, polyethylene glycol (M.W. 100 to 800), dipropylene glycol, etc., the weight ratio of the glycol solvent to propylene carbonate being at least 1:2
- d. a stabilizing quantity of a surfactant such as sorbitan monooleate; and
- e. from 0 to 15 percent compatible plasticizer, e.g., carboxy vinyl polymer (Carbol) yields an improved medicament base according to this invention.

EXAMPLE 6
Repeating the procedure of Example 4 with the ingredients of Example 5 yields improved compositions for topical treatment of inflammation according to this invention.

We claim:

1. A substantially anhydrous vehicle composition consisting essentially of:
   - a. from 5 to 40 weight percent of saturated fatty alcohol having from 16 to 24 carbons;
b. from 1 to 40 weight percent of propylene carbonate;
c. from 25 to 85 weight percent of glycol cosolvent, weight ratio of the glycol solvent to propylene carbonate being at least 1:2;
d. a stabilizing amount of surfactant;
e. from 0 to 15 weight percent of compatible plasticizer; and
f. from 0 to 3 weight percent water; said vehicle composition being particularly suitable for providing an occlusive film, for releasing topically active corticosteroids which are soluble in propylene carbonate, and for distributing medication over the skin surface and maintaining it there until beneficial action occurs.

2. The composition of claim 1 comprising
a. from 10 to 30 weight percent of saturated fatty alcohol having from 16 to 24 carbons;

3. The composition of claim 2 wherein the compatible plasticizer concentration is from 0.1 to 5 weight percent.

4. The composition of claim 1 wherein the weight ratio of the glycol cosolvent to propylene carbonate is at least 1:1.

5. The composition of claim 1 wherein the weight ratio of the glycol cosolvent to propylene carbonate is at least 3:1.