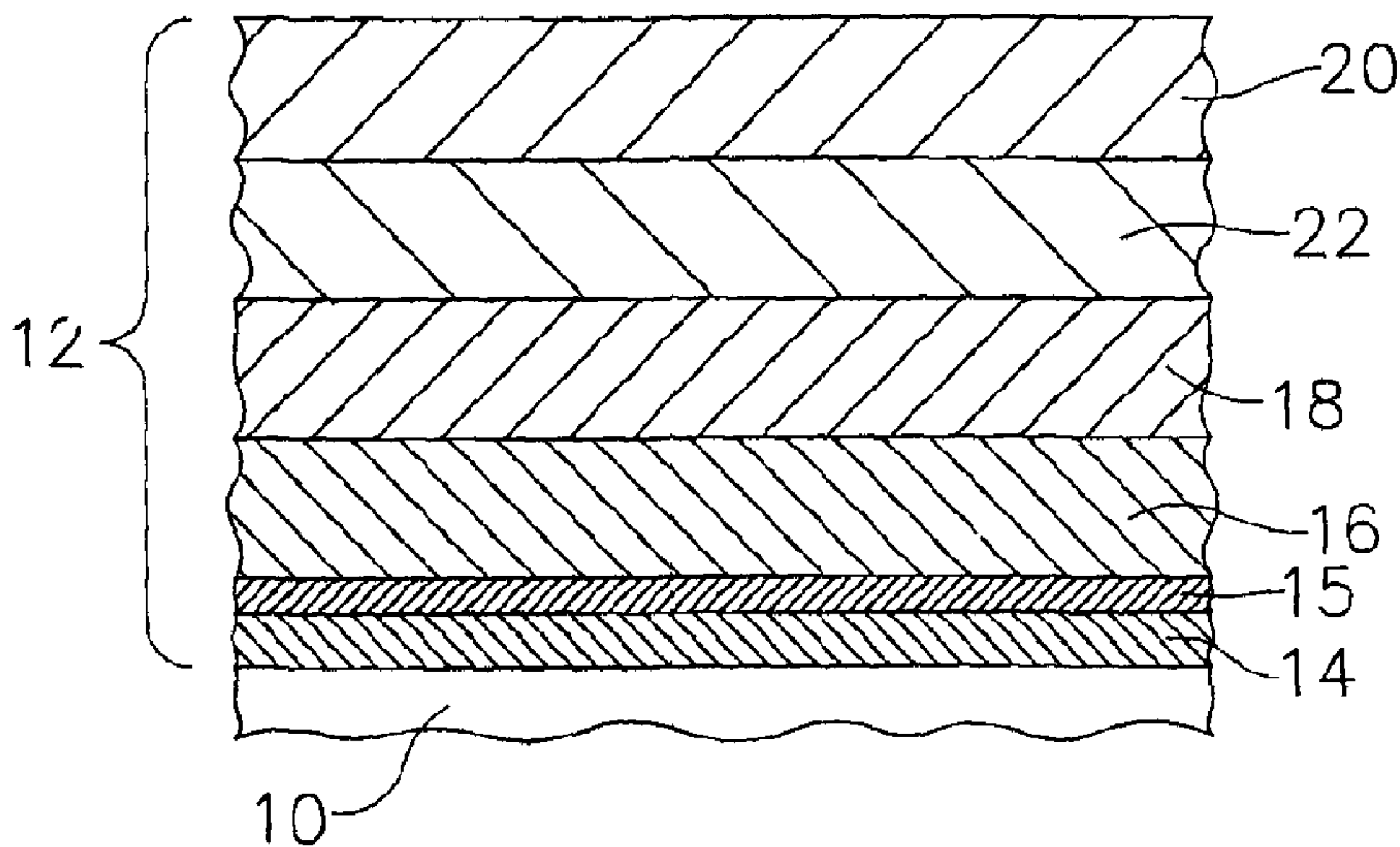




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 (54) Title: METHOD FOR MAKING SOLVENT BASED ENVIRONMENTAL BARRIER COATINGS USING SINTERING  
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(57) **Abrégé/Abstract:**

A method for making an environmental barrier coating (12) using a sintering aid comprising: combining at least an organic solvent, and a primary transition material comprising a rare earth disilicate, or a doped rare earth disilicate to produce a transition layer

**(57) Abrégé(suite)/Abstract(continued):**

slurry; combining at least an organic solvent, and a primary outer material comprising a rare earth monosilicate or a doped rare earth monosilicate to produce an outer layer slurry; combining at least an organic solvent, and a primary compliant material comprising BSAS or a rare earth doped BSAS to produce a compliant layer slurry; applying at least the transition layer slurry, and any one or more of the outer layer slurry or the compliant layer slurry, to a component (10); infiltrating a sol-gel solution comprising an aqueous solution of a water soluble salt sintering aid, or an organic solvent solution of a solvent soluble salt sintering aid into the applied slurry layers; and sintering the component (10) to produce the environmental barrier coating having at least a transition layer (16) and any one or more of an outer layer (20) or a (18) compliant layer wherein during sintering a reaction between the primary materials and the sintering aid results in the transition layer comprising a porosity of from 0% to about 15% by volume of the transition layer, the outer layer comprising a porosity of from 0% to about 30% by volume of the outer layer, and the compliant layer comprising a porosity of from about 0% to about 30% by volume of the compliant layer.

METHOD FOR MAKING SOLVENT BASED ENVIRONMENTAL BARRIER  
COATINGS USING SINTERING AIDS

ABSTRACT

A method for making an environmental barrier coating (12) using a sintering aid comprising: combining at least an organic solvent, and a primary transition material comprising a rare earth disilicate, or a doped rare earth disilicate to produce a transition layer slurry; combining at least an organic solvent, and a primary outer material comprising a rare earth monosilicate or a doped rare earth monosilicate to produce an outer layer slurry; combining at least an organic solvent, and a primary compliant material comprising BSAS or a rare earth doped BSAS to produce a compliant layer slurry; applying at least the transition layer slurry, and any one or more of the outer layer slurry or the compliant layer slurry, to a component (10); infiltrating a sol-gel solution comprising an aqueous solution of a water soluble salt sintering aid, or an organic solvent solution of a solvent soluble salt sintering aid into the applied slurry layers; and sintering the component (10) to produce the environmental barrier coating having at least a transition layer (16) and any one or more of an outer layer (20) or a (18) compliant layer wherein during sintering a reaction between the primary materials and the sintering aid results in the transition layer comprising a porosity of from 0% to about 15% by volume of the transition layer, the outer layer comprising a porosity of from 0% to about 30% by volume of the outer layer, and the compliant layer comprising a porosity of from about 0% to about 30% by volume of the compliant layer.

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METHOD FOR MAKING SOLVENT BASED ENVIRONMENTAL BARRIER  
COATINGS USING SINTERING AIDS

TECHNICAL FIELD

Embodiments described herein generally relate to methods for making solvent based environmental barrier coatings using sintering aids. More particularly, embodiments herein generally describe methods for making solvent based environmental barrier coatings for use on high temperature ceramic components.

BACKGROUND OF THE INVENTION

Higher operating temperatures for gas turbine engines are continuously being sought in order to improve their efficiency. However, as operating temperatures increase, the high temperature durability of the components of the engine must correspondingly increase. Significant advances in high temperature capabilities have been achieved through the formulation of iron, nickel, and cobalt-based superalloys. While superalloys have found wide use for components used throughout gas turbine engines, and especially in the higher temperature sections, alternative lighter-weight component materials have been proposed.

Ceramic matrix composites (CMCs) are a class of materials that consist of a reinforcing material surrounded by a ceramic matrix phase. Such materials, along with certain monolithic ceramics (i.e. ceramic materials without a reinforcing material), are currently being used for higher temperature applications. These ceramic materials are lightweight compared to superalloys yet can still provide strength and durability to the component made therefrom. Therefore, such materials are currently being considered for many gas turbine components used in higher temperature

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sections of gas turbine engines, such as airfoils (e.g. turbines, and vanes), combustors, shrouds and other like components that would benefit from the lighter-weight and higher temperature capability these materials can offer.

CMC and monolithic ceramic components can be coated with environmental barrier coatings (EBCs) to protect them from the harsh environment of high temperature engine sections. EBCs can provide a dense, hermetic seal against the corrosive gases in the hot combustion environment, which can rapidly oxidize silicon-containing CMCs and monolithic ceramics. Additionally, silicon oxide is not stable in high temperature steam, but rather, can be converted to volatile (gaseous) silicon hydroxide species. Thus, EBCs can help prevent dimensional changes in the ceramic component due to such oxidation and volatilization processes. Unfortunately, there can be some undesirable issues associated with standard, industrial coating processes such as plasma spray and vapor deposition (i.e. chemical vapor deposition, CVD, and electron beam physical vapor deposition, EBPVD) currently used to apply EBCs.

Accordingly, there remains a need for methods for making environmental barrier coatings for ceramic component that are suitable for use in the high temperature steam environments present in gas turbine engines.

#### BRIEF DESCRIPTION OF THE INVENTION

Embodiments herein generally relate to methods for making an environmental barrier coating using a sintering aid comprising: combining at least an organic solvent, and a primary transition material comprising a rare earth disilicate, or a doped rare earth disilicate to produce a transition layer slurry; applying the transition layer slurry to the component; drying the component having the applied transition layer slurry; infiltrating a sol-gel solution comprising an aqueous solution of a water soluble salt sintering aid, or an organic solvent solution of a solvent soluble salt sintering aid into the applied slurry; and sintering the component to produce the environmental barrier coating having at least the bond coat layer, and a transition layer wherein during sintering a reaction between the primary transition material and the sintering aid results in the transition layer comprising a porosity of from 0% to about 15% by volume of the transition layer.

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Embodiments herein also generally relate to methods for making an environmental barrier coating using a sintering aid comprising: combining at least an organic solvent, and a primary transition material comprising a rare earth disilicate, or a doped rare earth disilicate to produce a transition layer slurry; combining at least an organic solvent, and a primary outer material comprising a rare earth monosilicate or a doped rare earth monosilicate to produce an outer layer slurry; combining at least an organic solvent, and a primary compliant material comprising BSAS or a rare earth doped BSAS to produce a compliant layer slurry; applying at least the transition layer slurry, and any one or more of the outer layer slurry or the compliant layer slurry, to the component; drying the component; infiltrating a sol-gel solution comprising an aqueous solution of a water soluble salt sintering aid, or an organic solvent solution of a solvent soluble salt sintering aid into the applied slurry layers; and sintering the component to produce the environmental barrier coating having at least the bond coat layer, a transition layer and any one or more of an outer layer or a compliant layer wherein during sintering a reaction between the primary materials and the sintering aid results in the transition layer comprising a porosity of from 0% to about 15% by volume of the transition layer, the outer layer comprising a porosity of from 0% to about 30% by volume of the outer layer, and the compliant layer comprising a porosity of from about 0% to about 30% by volume of the compliant layer.

Embodiments herein also generally relate to methods for making an environmental barrier coating using a sintering aid comprising: combining at least an organic solvent, and a primary transition material comprising a rare earth disilicate, or a doped rare earth disilicate to produce a transition layer slurry; combining at least an organic solvent, and a primary outer material comprising a rare earth monosilicate or a doped rare earth monosilicate to produce an outer layer slurry; combining at least an organic solvent, and a primary compliant material comprising BSAS or a rare earth doped BSAS to produce a compliant layer slurry; applying the transition layer slurry to the component; drying the component; infiltrating a sol-gel solution comprising an aqueous solution of a water soluble salt sintering aid, or an organic solvent solution of a solvent soluble salt sintering aid into the applied transition layer slurry; applying any one or more of the outer layer slurry, the compliant layer slurry, or an additional

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transition layer slurry; drying the component; and sintering the component to produce the environmental barrier coating having a bond coat layer, at least one transition layer and any one or more of an outer layer or a compliant layer wherein during sintering a reaction between the primary materials and the sintering aid results in the at least one transition layer comprising a porosity of from 0% to about 15% by volume of the transition layer, the outer layer comprising a porosity of from 0% to about 30% by volume of the outer layer, and the compliant layer comprising a porosity of from about 0% to about 30% by volume of the compliant layer.

These and other features, aspects and advantages will become evident to those skilled in the art from the following disclosure.

#### BRIEF DESCRIPTION OF THE DRAWINGS

While the specification concludes with claims particularly pointing out and distinctly claiming the invention, it is believed that the embodiments set forth herein will be better understood from the following description in conjunction with the accompanying figures, in which like reference numerals identify like elements.

FIG. 1 is a schematic cross sectional view of one embodiment of a component having and environmental barrier coating in accordance with the description herein;

FIG. 2 is a SEM cross-section of an EBC coating on a SiC-SiC CMC in accordance with Example 1 herein;

FIG. 3 is a SEM cross-section of an EBC coating on SiC-SiC CMC in accordance with Example 2 herein; and

FIG. 4 is a SEM cross-section of an EBC coating on SiC-SiC CMC in accordance with Example 3 herein.

#### DETAILED DESCRIPTION OF THE INVENTION

Embodiments described herein generally relate to methods for making solvent based environmental barrier coatings using sintering aids. More particularly, embodiments herein generally describe methods for making solvent based environmental barrier coatings for use on high temperature ceramic components.

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More specifically, the EBCs described herein comprise solvent based slurries having sintering aids, which can lower the sintering temperature, thereby promoting the formation of dense EBC layers that can act as a hermetic seal to protect the underlying component from corrosion from the gases generated during high temperature combustion without damaging the component through exposure to high sintering temperatures, as explained herein below.

The EBCs described herein may be suitable for use in conjunction with CMCs or monolithic ceramics. As used herein, “CMCs” refers to silicon-containing matrix and reinforcing materials. Some examples of CMCs acceptable for use herein can include, but should not be limited to, materials having a matrix and reinforcing fibers comprising silicon carbide, silicon nitride, and mixtures thereof. As used herein, “monolithic ceramics” refers to materials comprising silicon carbide, silicon nitride, and mixtures thereof. Herein, CMCs and monolithic ceramics are collectively referred to as “ceramics.”

As used herein, the term “barrier coating(s)” can refer to environmental barrier coatings (EBCs). The barrier coatings herein may be suitable for use on “ceramic component,” or simply “component” 10 found in high temperature environments (e.g. operating temperatures of above 2100°F (1149°C)), such as those present in gas turbine engines. Examples of such ceramic components can include, for example, combustor components, turbine blades, shrouds, nozzles, heat shields, and vanes.

More specifically, EBC 12 may comprise a coating system including various combinations of the following: a bond coat layer 14, an optional silica layer 15, at least one transition layer 16, an optional compliant layer 18, an optional intermediate layer 22, and an optional outer layer 20, as shown generally in FIG. 1 and as set forth herein below.

Bond coat layer 14 may comprise silicon metal, silicide, or a combination thereof, and may generally have a thickness of from about 0.1 mils to about 6 mils (about 2.5 to about 150 micrometers). Due to the application method as described herein below, there may be some local regions where the silicon bond coat is missing, which can be acceptable. For example, in one embodiment, bond coat layer can cover about 100% of the surface

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of the component, and in another embodiment, about 90% or more of the surface area of the component.

As used herein “silicide” may include rare earth (Ln) silicides, chromium silicide (e.g. CrSi<sub>3</sub>), niobium silicide (e.g. NbSi<sub>2</sub>, NbSi<sub>3</sub>), molybdenum silicide (e.g. MoSi<sub>2</sub>, MoSi<sub>3</sub>), tantalum silicide (e.g. TaSi<sub>2</sub>, TaSi<sub>3</sub>), titanium silicide (e.g. TiSi<sub>2</sub>, TiSi<sub>3</sub>), tungsten silicide (e.g. WSi<sub>2</sub>, W<sub>5</sub>Si<sub>3</sub>), zirconium silicide (e.g. ZrSi<sub>2</sub>), hafnium silicide (e.g. HfSi<sub>2</sub>),

As used herein, “rare earth” represented “(Ln)” refers to the rare earth elements of scandium (Sc), yttrium (Y), lanthanum (La), cerium (Ce), praseodymium (Pr), neodymium (Nd), promethium (Pm), samarium (Sm), europium (Eu), gadolinium (Gd), terbium (Tb), dysprosium (Dy), holmium (Ho), erbium (Er), thulium (Tm), ytterbium (Yb), lutetium (Lu), and mixtures thereof.

Silica layer 15 can be amorphous or crystalline, and have an initial thickness of from about 0.0 mils to about 0.2 mils (about 0.0 to about 5 micrometers). However, the thickness of silica layer 15 can increase over time. Specifically, the silicon in bond coat layer 14 can oxidize slowly during the service life of the EBC to gradually increase the thickness of silica layer 15. This oxidation of bond coat 14 can protect the underlying ceramic component from oxidation since the bond coat is oxidized rather than the ceramic component. Silica layer 15 can, in some embodiments, also be doped with a doping composition, as defined herein below, due to diffusion of the sintering aid into the silica layer.

Transition layer 16 may comprise a rare earth disilicate, a doped rare earth disilicate, or a doped rare earth disilicate containing secondary materials, as defined below. More specifically, transition layer 16 may include from about 85% to about 100% by volume of the transition layer of a primary transition material and up to about 15% by volume of the transition layer of a secondary material, and in one embodiment from about 85% to about 99% by volume of the transition layer of the primary transition material and from about 1% to about 15% by volume of the transition layer of the secondary material. In another embodiment, transition layer 16 may comprise 100% primary transition material wherein the primary transition material can be doped, as described below.

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As used herein, “primary transition material” refers to a rare earth disilicate ( $\text{Ln}_2\text{Si}_2\text{O}_7$ ), or a doped rare earth disilicate. As used herein, “doped rare earth disilicate” refers to  $\text{Ln}_2\text{Si}_2\text{O}_7$  doped with a “doping composition” selected from the group consisting of iron (Fe), aluminum (Al), titanium (Ti), gallium (Ga), nickel (Ni), boron (B), an alkali, an alkali-earth, and Lnb rare earths, as defined herein below. As used herein throughout, “secondary material” refers to a material comprising a doping composition (as defined previously), and specifically, can be selected from the group consisting of  $\text{Fe}_2\text{O}_3$ , iron silicates, rare earth iron oxides,  $\text{Al}_2\text{O}_3$ , mullite, rare earth aluminates, rare earth aluminosilicates,  $\text{TiO}_2$ , rare earth titanates,  $\text{Ga}_2\text{O}_3$ , rare earth gallates, NiO, nickel silicates, rare earth nickel oxides, Lnb metals,  $\text{Lnb}_2\text{O}_3$ ,  $\text{Lnb}_2\text{Si}_2\text{O}_7$ ,  $\text{Lnb}_2\text{SiO}_5$ , borosilicate glass, alkaline earth silicates, alkaline earth rare earth oxides, alkaline earth rare earth silicates, and mixtures thereof. Any doping composition present in the primary material should correspond to the doping composition contained in any secondary material present (e.g. Fe-doped  $\text{Ln}_2\text{Si}_2\text{O}_7$  with  $\text{Fe}_2\text{O}_3$  secondary material; Ti-doped  $\text{Ln}_2\text{Si}_2\text{O}_7$  with  $\text{TiO}_2$  secondary material; or Ni-doped  $\text{Ln}_2\text{Si}_2\text{O}_7$  with rare earth nickel oxide secondary material, for example).

Each transition layer 16 may have a thickness of from about 0.1 mils to about 40 mils (about 2.5 to about 1000 micrometers), and may be made and applied to the underlying layer as set forth below. In one embodiment, there may be more than one transition layer present. In such instances, each transition layer may comprise the same or different combination of primary transition materials and secondary materials. Transition layer 16 may have a porosity level of from 0% to about 15% by volume of the transition layer, and in another embodiment, from about 0.01% to about 15% by volume of the transition layer.

Similarly, outer layer 20 may comprise a rare earth monosilicate, a doped rare earth monosilicate, or a doped rare earth monosilicate containing secondary material. More specifically, outer layer 20 can include from about 85% to about 100% by volume of the outer layer of a primary outer material and up to about 15% by volume of the outer layer of the previously defined secondary material, and in one embodiment from about 85% to about 99% by volume of the outer layer of a primary outer material and from about 1% to about 15% by volume of the outer layer of the secondary material.

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In another embodiment, outer layer 20 may comprise 100% primary outer material wherein the primary outer material can be doped as described below.

As used herein, “primary outer material” refers to a rare earth monosilicate, or a doped rare earth monosilicate. As used herein, “doped rare earth monosilicate” refers to  $\text{Ln}_2\text{SiO}_5$  doped with a doping composition, as defined previously. Outer layer 20 may have a thickness of from about 0.1 mils to about 3 mils (about 2.5 to about 75 micrometers), and may be made and applied to the underlying layer as set forth below. In one embodiment, outer layer 20 may have a porosity level of from 0% to about 30% by volume of the outer layer, and in another embodiment, from about 0.01% to about 30% by volume of the outer layer, and in another embodiment, from about 0.01% to about 15% by volume of the outer layer. In some embodiments, outer layer 20 can comprise cracks therein at a density of up to about 10 cracks/mm that can form during operation due to thermal expansion anisotropy.

In reference to the embodiments herein, “Ln<sub>b</sub> rare earth metal”, or simply “Ln<sub>b</sub>” refers to a sub-set of rare-earth metals having a melting point below at least about 1450°C including lanthanum, cerium, praseodymium, neodymium, promethium, samarium, europium, gadolinium, terbium, dysprosium, and ytterbium. In one embodiment, the sub-set can include only those rare earth elements having a melting point below about 1350°C including lanthanum, cerium, praseodymium, neodymium, promethium, samarium, europium, gadolinium, and ytterbium. The Ln<sub>b</sub> rare earth metal can be utilized with SiC-SiC CMCs having an operation limit of about 1357°C.

As used herein throughout, “alkaline earth” can refer to magnesium (Mg), calcium (Ca), strontium (Sr), and barium (Ba). As used herein, “alkali” refers to lithium (Li), potassium (K), and sodium (Na). “Iron silicates” can include compounds such as  $\text{Fe}_2\text{SiO}_4$ , and glasses of rare earth iron silicates. “Rare earth iron oxides” can include compounds such as garnets ( $\text{Ln}_3\text{Fe}_5\text{O}_{12}$ ), monoclinic ferrites ( $\text{Ln}_4\text{Fe}_2\text{O}_9$ ), and perovskites ( $\text{LnFeO}_3$ ). “Rare-earth aluminates” can include compounds such as garnets ( $\text{Ln}_3\text{Al}_5\text{O}_{12}$ ), monoclinic aluminates ( $\text{Ln}_4\text{Al}_2\text{O}_9$ ), and perovskites ( $\text{LnAlO}_3$ ). “Rare earth aluminates” can include glassy materials comprised of about 35-50wt%  $\text{Ln}_2\text{O}_3$ , about 15-25wt%  $\text{Al}_2\text{O}_3$ , and about 25-50wt%  $\text{SiO}_2$ . “Rare-earth titanates” can

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include compounds such as  $\text{Ln}_2\text{Ti}_2\text{O}_7$  (pyrochlore) and  $\text{Ln}_2\text{TiO}_5$ . “Rare-earth gallates” can include compounds such as garnets ( $\text{Ln}_3\text{Ga}_5\text{O}_{12}$ ), monoclinic gallates ( $\text{Ln}_4\text{Ga}_2\text{O}_9$ ), perovskites ( $\text{LnGaO}_3$ ), and  $\text{Ln}_3\text{GaO}_6$ . “Nickel silicates” can include compounds such as  $\text{Ni}_2\text{SiO}_4$ . “Borosilicate glass” can refer to any amorphous material containing up to about 15% by weight boron oxide ( $\text{B}_2\text{O}_3$ ), up to about 10% alkali oxide selected from the group consisting of sodium ( $\text{Na}_2\text{O}$ ), potassium ( $\text{K}_2\text{O}$ ), lithium ( $\text{Li}_2\text{O}$ ), or any combinations of thereof, up to about 10% alumina ( $\text{Al}_2\text{O}_3$ ), and a balance of silica ( $\text{SiO}_2$ ). “Alkaline earth silicates” can include compounds such as  $\text{Mg}_2\text{SiO}_4$ ,  $\text{MgSiO}_3$ ,  $\text{Ca}_2\text{SiO}_4$ ,  $\text{Ca}_3\text{SiO}_5$ ,  $\text{Ca}_3\text{Si}_2\text{O}_7$ ,  $\text{CaSiO}_3$ ,  $\text{Ba}_2\text{SiO}_4$ ,  $\text{BaSiO}_3$ ,  $\text{Ba}_2\text{Si}_3\text{O}_8$ ,  $\text{BaSi}_2\text{O}_5$ ,  $\text{Sr}_2\text{SiO}_4$ , and  $\text{SrSiO}_3$ . “Alkali earth rare earth oxides” can include compounds such as  $\text{BaLn}_2\text{O}_4$ ,  $\text{Mg}_3\text{Ln}_2\text{O}_6$ ,  $\text{SrLn}_2\text{O}_4$ , and  $\text{Sr}_2\text{Ln}_2\text{O}_5$ . “Alkaline earth rare earth silicates” can include oxyapatite materials (i.e.  $\text{Ae}_2\text{Ln}_8\text{Si}_6\text{O}_{26}$ ).

If present, compliant layer 18 may include from about 85% to about 100% by volume of the compliant layer of a primary compliant material and up to about 15% by volume of the compliant layer of a secondary compliant material, and in one embodiment from about 85% to about 99% by volume of the compliant layer of a primary compliant material and from about 1% to about 15% by volume of the compliant layer of the secondary compliant material. In another embodiment, compliant layer 18 may comprise 100% by volume of the compliant layer of a primary compliant material wherein the primary outer material may be doped with a rare earth element.

As used herein, “primary compliant material” refers to BSAS, or a rare earth doped BSAS, while “secondary compliant material” refers to  $\text{Ln}_2\text{O}_3$ ,  $\text{Ln}_2\text{Si}_2\text{O}_7$ ,  $\text{Ln}_2\text{SiO}_5$ ,  $\text{Ln}_3\text{Al}_5\text{O}_{12}$ ,  $\text{Al}_2\text{O}_3$ , mullite, and combinations thereof. Compliant layer 20 may have a thickness of from about 0.1 mils to about 40 mils (about 2.5 to about 1000 micrometers), and may be made and applied as set forth below. In one embodiment, compliant layer 18 may have a porosity level of from 0% to about 30% by volume of the compliant layer, and in another embodiment, from about 0.01% to about 30% by volume of the compliant layer, and in another embodiment, from about 0.01% to about 15% by volume of the compliant layer.

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Intermediate layer 22, if present, can comprise the previously defined primary outer materials of rare earth monosilicate or doped rare earth monosilicate. Similar to the silica layer, intermediate layer 22 can form during the service life of the EBC. More specifically, high temperature steam penetrates the outer layer 20, and as the steam reacts with the primary transition material of the transition layer to volatilize  $\text{SiO}_2$ , intermediate layer 22 can form.

By way of example, and not limitation, the EBC systems described herein may include in one embodiment, component 10, bond coat layer 14, and transition layer 16; in another embodiment, component 10, bond coat layer 14, transition layer 16, and outer layer 20; in another embodiment, component 10, bond coat layer 14, transition layer 16, compliant layer 18, and outer layer 20; in another embodiment, component 10, bond coat layer 14, transition layer 16, compliant layer 18, transition layer 16, and outer layer 20; in another embodiment, component 10, bond coat layer 14, silica layer 15, and transition layer 16; in another embodiment, component 10, bond coat layer 14, silica layer 15, transition layer 16, and outer layer 20; in another embodiment, component 10, bond coat layer 14, silica layer 15, transition layer 16, compliant layer 18, and outer layer 20; in another embodiment, component 10, bond coat layer 14, silica layer 15, transition layer 16, compliant layer 18, transition layer 16, and outer layer 20; in another embodiment, component 10, bond coat layer 14, transition layer 16, intermediate layer 22, and outer layer 20; in another embodiment, component 10, bond coat layer 14, silica layer 15, transition layer 16, intermediate layer 22, and outer layer 20; in another embodiment, component 10, bond coat layer 14, silica layer 15, transition layer 16, intermediate layer 22 (which can form during operation), and outer layer 20; and in another embodiment, component 10, bond coat layer 14, silica layer 15, transition layer 16, compliant layer 18, transition layer 16, intermediate layer 22 (which can form during operation), and outer layer 20. Such embodiments can be suitable for use in environments having a temperature up to about  $31704^\circ\text{C}$  ( $3100^\circ\text{F}$ ).

Alternately, the EBC system may comprise component 10, bond coat layer 14, transition layer 16, and compliant layer 18; in another embodiment, component 10, bond coat layer 14, silica layer 15, transition layer 16, and compliant layer 18. Such

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embodiments can be suitable for use in environments having a temperature of up to about 1538°C (2800°F).

Those skilled in the art will understand that embodiments in addition to those set forth previously are also acceptable, and that not all of the layers need to be present initially, but rather, may form during engine operation.

The EBC can be made and applied in accordance with the description below.

Bond coat layer 14 may be applied by plasma spray processes, chemical vapor deposition processes, electron beam physical vapor deposition processes, dipping in molten silicon, sputtering processes, and other conventional application processes known to those skilled in the art.

As previously described, silica layer 15 can form during the service life of the EBC. Specifically, oxygen in the surrounding atmosphere can diffuse through any of the outer layer, compliant, and transition layer(s) present in the EBC and react with the silicon of bond coat layer 14 to form silica layer 15. Alternately, silica layer 15 may be intentionally deposited by chemical vapor deposition, plasma spray, slurry deposition, or other conventional method.

Similar to silica layer 15, intermediate layer 22 can also form during the service life of the EBC when high temperature steam reacts with transition layer 16, as previously described.

The manufacturing and application processes for transition layer 16, compliant layer 18 and outer layer 20 can consist of a solvent based slurry deposition cycle including sintering aids to lower the temperature needed to densify the layers. The slurry deposition cycle can generally include slurry formation, slurry application, drying, and sintering, with optional masking, leveling, sintering aid infiltration, mask removal, and binder burnout steps, as set forth below. Those skilled in the art will understand that slurries of varying compositions can be used to make EBC layers of varying composition and that multiple slurry deposition cycles can be used to build up the total thickness of a particular layer. Each layer can have the thickness set forth previously with the average thickness per slurry deposition cycle depending primarily

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on the slurry solids loading, sintering aid concentration, and number of dip, spray, or paint passes.

The slurries described in the embodiments herein can comprise various slurry components, but generally include an organic solvent, ceramic particles, sintering aid, and organic processing aids. Particularly, the slurry may comprise from about 6.8 wt% to about 96.1 wt% solvent, from about 0 wt% to about 8.9 wt% of a dispersant, from about 0 wt% to about 15.4 wt % of a plasticizer, from about 3.9 wt% to about 93.2 wt % primary material, from about 0 wt% to about 15.4 wt% thickener, from about 0wt% to about 1wt% surfactant, and from about 0 wt% to about 20 wt % slurry sintering aid if there is one sintering aid, or alternately, from about 0 wt % to about 79.9 wt % slurry sintering aid if there are two sintering aids present; and in another embodiment, from about 0.01 wt% to about 20 wt % slurry sintering aid if there is one sintering aid, or alternately, from about 0.01 wt % to about 79.9 wt % slurry sintering aid if there are two sintering aids present, as described herein below.

More specifically, in such solvent based slurries, “organic solvent” refers to methanol, ethanol, propanol, butanol, pentanol, hexanol, heptanol, octanol, nonanol, decanol, dodecanol, acetone, methyl isobutyl ketone (MIBK), methyl ethyl ketone (MEK), toluene, ethylbenzene, propyl benzene, methoxybenzene, heptane, octane, nonane, decane, xylene, mineral spirits, naphtha (such as VM&P naphtha), tetrahydrofuran, ethers, and combinations thereof.

“Primary materials” may refer to  $\text{Ln}_2\text{Si}_2\text{O}_7$ ,  $\text{Ln}_2\text{SiO}_5$ , or BSAS depending on which layer is being made.

“Dispersant” can refer to polyacrylic acid, polyacrylic acid-polyethylene oxide copolymers, polymethacrylic acid, polyethylenimine, phosphate esters, menhaden fish oil, polyethylene oxide, polysilazane, and combinations thereof.

“Plasticizer” can refer to ethylene glycol, diethylene glycol, triethylene glycol, tetraethylene glycol glycerol, glycerin, polyethylene glycol, dibutyl phthalate, Bis(2-ethylhexyl) phthalate, Bis(n-butyl)phthalate, Butyl benzyl phthalate, Diisodecyl phthalate, Di-n-octyl phthalate, Diisooctyl phthalate, Diethyl phthalate, Diisobutyl

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phthalate, Di-n-hexyl phthalate, Di(propylene glycol) dibenzoate, Di(ethylene glycol) dibenzoate, tri(ethylene glycol) dibenzoate, and combinations thereof.

As used herein, “slurry sintering aid” can refer to sintering aid compositions suitable for inclusion in the slurry. In some embodiments, there can be from about 0 wt% to about 20 wt %, and in some embodiments from about 0.01 wt% to about 20 wt %, of a slurry sintering aid selected from iron oxide, gallium oxide, aluminum oxide, nickel oxide, titanium oxide, boron oxide, and alkaline earth oxides; carbonyl iron; iron metal, aluminum metal, boron, nickel metal, hydroxides including iron hydroxide, gallium hydroxide, aluminum hydroxide, nickel hydroxide, titanium hydroxide, alkaline earth hydroxides; carbonates including iron carbonate, gallium carbonate, aluminum carbonate, nickel carbonate, boron carbonate, and alkaline earth carbonates; oxalates including iron oxalate, gallium oxalate, aluminum oxalate, nickel oxalate, titanium oxalate; and “solvent soluble salts” including solvent soluble iron salts, solvent soluble gallium salts, solvent soluble aluminum salts, solvent soluble nickel salts, solvent soluble titanium salts, solvent soluble boron salts, and solvent soluble alkaline earth salts. In the case of the compliant layer slurry, the “slurry sintering aid” may include rare earth nitrate, rare earth acetate, rare earth chloride, rare earth oxide, ammonium phosphate, phosphoric acid, polyvinyl phosphoric acid, and combinations thereof.

In an alternate embodiment, the slurry can comprise from about 0 wt% to about 59.3 wt%, and in one embodiment from about 0.01 wt% to about 59.3 wt%, of an Lnb metal slurry sintering aid as defined previously herein, and from about 0 wt% to about 20.6 wt%, and in one embodiment from about 0.01 wt% to about 20.6 wt%, of a SiO<sub>2</sub> slurry sintering aid. In this embodiment, the Lnb and SiO<sub>2</sub> content can be held such that the mole ratio of Lnb to SiO<sub>2</sub> is about 1 to 1 for slurries containing rare earth disilicate primary transition material, and about 2 to 1 for slurries containing rare earth monosilicate primary outer material.

As used herein, “solvent-soluble iron salts” can include ethoxide, iron 2,4-pentanedionate, and iron tetramethylheptanedionate; “solvent-soluble gallium salts” can include gallium 8-hydroxyquinolate, gallium 2,4-pentanedionate, gallium

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ethoxide, gallium isopropoxide, and gallium 2,2,6,6-tetramethylheptanedionate; “solvent-soluble aluminum salts” can include butoxide, aluminum di-s-butoxide ethylacetoacetate, aluminum diisopropoxide ethylacetoacetate, aluminum ethoxide, aluminum ethoxyethoxyethoxide, aluminum 3,5-heptanedionate, aluminum isopropoxide, aluminum 9-octadecenylacetoacetate diisopropoxide, aluminum 2,4-pentanedionate, aluminum pentanedionate bis(ethylacetoacetate), aluminum 2,2,6,6-tetramethyl-3,5-heptanedionate, and aluminum phenoxide; “solvent-soluble nickel salts” can include nickel 2,4-pentanedionate, nickel 2,2,6,6-tetramethyl-3,5-heptanedionate; “solvent-soluble titanium salts” can include titanium allylacetoacetatetriisopropoxide, titanium bis(triethanolamine)diisopropoxide, titanium butoxide, titanium di-n-butoxide bis(2-ethylhexanoate), titanium diisopropoxide(bis-2,4-pentanedionate), titanium diisopropoxide bis(tetramethylheptanedionate), titanium ethoxide, titanium diisopropoxide bis(ethylacetoacetate), titanium 2-ethylhexoxide, titanium iodide triisopropoxide, titanium isobutoxide, titanium isopropoxide, titanium methacrylate triisopropoxide, titanium methacryloxyethylacetoacetate triisopropoxide, titanium methoxide, titanium methoxypropoxide, titanium methylphenoxide, titanium n-nonyloxide, titanium oxide bis(pentanedionate), titanium oxide bis(tetramethylheptanedionate), and titanium n-propoxide; “solvent-soluble boron salts” can include boron ethoxide, boron butoxide, boron isopropoxide, boron methoxide, boron methoxyethoxide, boron n-propoxide; and “solvent-soluble alkaline earth salts” can include calcium isopropoxide, calcium methoxyethoxide, calcium methoxide, calcium ethoxide, strontium isopropoxide, strontium methoxypropoxide, strontium 2,4-pentanedionate, strontium 2,2,6,6-tetramethyl-3,5-heptanedionate, magnesium ethoxide, magnesium methoxide, magnesium methoxyethoxide, magnesium 2,4-pentanedionate, magnesium n-propoxide, barium isopropoxide, barium methoxypropoxide, barium 2,4-pentanedionate, barium 2,2,6,6-tetramethyl-3,5-heptanedionate.

“Thickener” refers to polyvinyl butyral, polyvinyl acetate, poly(isobutyl methacrylate), poly[(n-butyl methacrylate-co-isobutyl methacrylate)], methyl methacrylate copolymers, ethyl methacrylate copolymers, poly methyl methacrylate, polyethyl methacrylate, polyvinylpyrrolone, ethyl cellulose, nitrocellulose, and other solvent soluble cellulose derivatives, and combinations thereof.

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“Surfactant” refers to compositions selected from the group consisting of fluorocarbons, dimethylsilicones, and ethoxylated acetylenic diol chemistries (e.g. commercial surfactants in the Surfynol® series such as Surfynol® 420 and 502 (Air Products and Chemicals, Inc.)), and combinations thereof.

Also, as used herein, “organic processing aids” refers to any dispersants, plasticizers, thickeners, and surfactants present in the slurry. These organic processing aids are comprised primarily of carbon and other elements that volatilize during processing such that they are not present in the post-sintered coating.

The slurry can be formed by combining any or all of the previously described slurry components with mixing media in a container. The mixture can be mixed using conventional techniques known to those skilled in the art such as shaking with up to about a 1 inch (about 25.4 mm) diameter alumina or zirconia mixing media, ball milling using about a 0.25 inch to about a 1 inch (about 0.64cm to about 2.54cm) diameter alumina or zirconia mixing media, attritor milling using about a 1mm to about a 5mm diameter zirconia-based mixing media, planetary ball milling using from about a 1mm to about a 5mm diameter zirconia-based media, or mechanical mixing or stirring with simultaneous application of ultrasonic energy. The mixing media or ultrasonic energy can break apart any agglomerated ceramic particles in the slurry. Any mixing media present may then be removed by straining, for example.

If not added previously, thickener may be added to the slurry if desired and the resulting mixture may be agitated by such methods as mechanical stirring, rolling, blending, shaking, and other like methods until the thickener is fully dissolved, generally after about 5 to about 60 minutes.

Similarly, if not added previously, the addition of the sintering aids may follow along with mixing using the previously described methods until the sintering aids dissolve, which is about 5 to about 60 minutes.

Once all slurry components have been mixed, the slurry can be filtered through screens of varying mesh sizes to remove any impurities that may be present, such as after the initial mixing of the slurry or after use of the slurry to deposit coating layers.

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A 325 mesh screen, for example, can be used to filter out impurities having an average size of about 44 microns or greater.

After mixing and optional filtering, the slurry can be agitated indefinitely by slow rolling, slow mechanical mixing, or other like methods to avoid trapping air bubbles in the slurry. In one embodiment, the slurry may be refreshed by adding additional solvent to account for that which has evaporated during processing. Alternately, once mixed, the slurry can be set aside until needed for application. Those skilled in the art will understand that the previous embodiment sets forth one method for making the slurry compositions described herein, and that other methods are also acceptable, as set forth in the Examples below.

Optionally, masking can be applied to the ceramic component before the slurry is applied to prevent coating specific areas of the component. Masking may be carried out using conventional techniques known to those skilled in the art including, but not limited to, tapes, tooling, and paint-on adhesives.

Once all desired masking of the ceramic component is complete, the slurry can be applied to the component to produce a coated component. The slurry can be applied to the component (or on top of a previously applied layer) using any conventional slurry deposition method known to those skilled in the art, including but not limited to, dipping the component into a slurry bath, or painting, rolling, stamping, spraying, or pouring the slurry onto the component. Slurry application can be carried out manually or it may be automated.

Once the slurry has been applied to the component, and while the slurry is still wet, it may be leveled to remove excess slurry material. Leveling may be carried out using conventional techniques such as, but not limited to, spinning, rotating, slinging the component, dripping with or without applied vibration, or using a doctor blade, to remove excess slurry material. Similar to the slurry application, leveling can be conducted manually or it may be automated.

Next, the coated component can be dried. Drying may be carried out in an enclosed area having additional organic solvent present in secondary containers. This can help

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slow the drying process because the atmosphere can be saturated with organic solvent. Since this process utilizes an organic solvent, it is not strongly sensitive to humidity. Temperature variation can be used to control the drying rate, however, those skilled in the art will understand that that temperatures can be kept below the flash point of the organic solvent. Placing a coated component in a vacuum chamber and pulling a vacuum can also be used to accelerate drying.

After drying, any masking present may then be removed by peeling off tapes and adhesives, pyrolysis of tapes and adhesives, or removing multi-use tooling. Any rough edges remaining after masking removal may be scraped or cut away using a sharp or abrasive tool.

Next, burnout of the organic processing aids may be carried out by placing the dried component in an elevated temperature environment so that any residual solvent can be evaporated and the organic processing aids can be pyrolyzed. In one embodiment, burnout of the organic processing aids may be accomplished by heating the dried component at a rate of from about 1°C/min to about 15°C/min to a temperature of from about 400°C to about 1000°C and holding the component at this temperature for from about 0 to about 10 hours. In another embodiment, the coated component may be heated at a rate of from about 2°C/min to about 6°C/min to a temperature of from about 600°C to about 800°C and holding the component at this temperature for from about 0 to about 10 hours. In another embodiment, the hold time can be eliminated by slowly ramping up to the target temperature without holding, followed by ramping up or down to another temperature at a different rate. In another embodiment, binder burnout can occur rapidly by placing the coated component into a furnace heated to a temperature of from about 400°C to about 1400°C.

Binder burnout and sintering heat treatments may be carried out in an ambient air atmosphere, or in an inert gas atmosphere where the inert gas is selected from hydrogen, a noble gas such as helium, neon, argon, krypton, xenon, or mixtures thereof. In one embodiment, the inert gas atmosphere can be used in conjunction with Ln<sub>2</sub>O<sub>3</sub> and SiO<sub>2</sub> sintering aids so as not to convert the rare earth metal to an oxide before

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it melts. Maintaining the Lnb metal in a metal state can promote liquid phase sintering and subsequent reaction with the  $\text{SiO}_2$ .

The dried component may then be sintered to produce a component comprising an environmental barrier coating. Sintering can serve to simultaneously densify and impart strength to the coating. Additionally, in the case of the transition and compliant layers of the EBC, sintering can impart a hermetic seal against high temperature steam present in the engine environment. In the case of the outer layer, sintering can provide a dense barrier against the infiltration of molten silicates, such as calcium-magnesium aluminosilicate (CMAS) that may be encountered as a result of particulate contamination (i.e. dirt and sand) in the engine environment. Sintering can be carried out using a conventional furnace, or by using such methods as microwave sintering, laser sintering, infrared sintering, and the like.

Sintering can be accomplished by heating the dried component at a rate of from about  $1^\circ\text{C}/\text{min}$  to about  $15^\circ\text{C}/\text{min}$  to a temperature of from about  $1100^\circ\text{C}$  to about  $1700^\circ\text{C}$  and holding the component at that temperature for from about 0 to about 24 hours. In another embodiment, sintering can be accomplished by heating the coated component at a rate of from about  $5^\circ\text{C}/\text{min}$  to about  $15^\circ\text{C}/\text{min}$  to a temperature of from about  $1300^\circ\text{C}$  to about  $1375^\circ\text{C}$  and holding the component at that temperature for from about 0 to about 24 hours. In another embodiment, sintering can occur rapidly by placing the coated component into a furnace heated to a temperature of from about  $1000^\circ\text{C}$  to about  $1400^\circ\text{C}$ .

In an alternate embodiment, all layers of the EBC can be applied, one on top of the other, before masking removal, organic processing aid burnout, and sintering are carried out. Those skilled in the art will understand that after application of each layer, the layer should be dried, or partially dried, before the application of the subsequent layer.

In another embodiment, the sintering aid does not need to be added directly to the transition or outer layer of the slurry to achieve the desired result. The sintering aid can be added to one layer of the EBC slurry, and during sintering, the sintering aid

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can diffuse throughout the EBC to the remaining layers. In another embodiment, a primary material slurry with no sintering aid can be densified by applying the layer, allowing it to dry, and then back infiltrating a sol-gel solution comprising a sintering aid prior to heat treatment as explained below.

Infiltration may allow for the densification of a thicker layer of EBC material at one time. Moreover, infiltration is a way to add more sintering aid after sintering if the coating is not as dense as desired. The sol-gel solution used for infiltration may be a solution of an organic solvent and a solvent soluble salt sintering aid, as defined previously, or a solution of water and a water soluble salt sintering aid.

As used herein “water soluble salt sintering aid” can refer to “water-soluble iron salts” such as iron nitrate and iron acetate; “water-soluble gallium salts” such as gallium nitrate and gallium acetate; “water-soluble aluminum salts” such as aluminum nitrate and aluminum acetate; “water-soluble nickel salts” such as nickel nitrate and nickel acetate; “water-soluble titanium salts” such as titanium chloride; “water-soluble boron salts” such as boric acid and ammonium borate; and “water-soluble alkaline earth salts” such as include  $Mg(NO_3)_2$ ,  $Ca(NO_3)_2$ ,  $Sr(NO_3)_2$ ,  $Ba(NO_3)_2$ ,  $MgC_2H_3O_2$ ,  $CaC_2H_3O_2$ ,  $SrC_2H_3O_2$ , and  $BaC_2H_3O_2$ .

As used herein, “sintering aid(s)” can refer to any of a “slurry sintering aid,” a “solvent soluble sintering aid,” or a “water soluble salt sintering aid,” as defined previously. Without intending to be limited by theory, the inclusion of sintering aids to the EBC embodiments herein can increase the rate of diffusion of primary material such that surface area reduction (i.e. high surface area particles consolidating to form a dense coating) can occur at lower temperatures than it would have absent the sintering aid. As previously described, sintering at lower temperatures (i.e. about 1357°C or below) can not only result in a highly dense (i.e. greater than about 85% for the transition layer, greater than about 70% for the compliant layer, and greater than about 70% for the outer layer) coating that can be less susceptible to the penetration of hot steam from the engine environment, but can also help prevent the degradation of the mechanical properties of the underlying component that could result from prolonged exposure to higher temperatures.

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Sintering aids can act in a variety of ways depending on the amount of sintering aid included in the EBC and the time at which the coating is exposed to sintering temperatures. For example, in one embodiment, the sintering aid can dissolve completely into the primary material (i.e. primary transition, outer, or compliant, material) to “dope” the material. In another embodiment, if the amount of sintering aid that is soluble in the primary material is exceeded, the remaining insoluble portion of sintering aid can react with the primary material to form the secondary material (i.e. secondary transition, outer, or compliant material). In another embodiment, primary material and secondary material can be present as described previously, along with residual sintering aid.

In these latter two embodiments, when the secondary material is highly volatile in high temperature steam, such as but not limited to, alkali silicates, alkaline earth silicates, mullite, iron silicate, borosilicate glass, nickel silicate, and residual sintering aids of iron, aluminum, titanium, gallium, nickel, boron, alkali, and alkali-earth compounds, as long as the total volume of secondary material, plus porosity (plus residual sintering aid when present) in either of the intermediate layer or compliant layer (when present) of the EBC remains about 15% by volume or less, the hermetic seal can be maintained. Alternately, in these latter two embodiments, when the secondary material is highly resistant to volatilization in high temperature steam, such as when the secondary material comprises a rare earth containing compound, such as but not limited to rare earth oxide, rare earth titanate, rare earth iron compound, rare earth gallate, rare earth aluminate, and rare earth aluminosilicate, the porosity in either of the intermediate or compliant layer (when present) of the EBC need remain about 15% by volume or less to maintain the hermetic seal.

It should be noted that at low levels of sintering aid, the densified coating layer might not initially include any detectable secondary materials. In some embodiments, the secondary materials may never become detectable. In other embodiments, however, after hours of exposure to high temperature steam in the engine environment, the secondary materials can become detectable using techniques such as x-ray diffraction, electron microscopy, electron dispersive spectroscopy (EDS), and the like.

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EBC embodiments described herein can offer a variety of benefits over current EBCs and manufacturing processes thereof. Specifically, as previously described, the inclusion of a sintering aid in the EBC embodiments herein can permit sintering at lower temperatures (i.e. about 1357°C or below). This can result in a highly dense (i.e. greater than about 85% for the transition layer, and greater than about 70% for each of the outer, and compliant, layers) coating that can be less susceptible to the penetration of hot steam from the engine environment, and can also help prevent the degradation of the mechanical properties of the underlying component that could result from prolonged exposure to higher temperatures. Also, the embodiments set forth herein can be made at less expense than current EBCs due to the use of the slurry deposition process, which is made possible by the incorporation of sintering aids into the various layers. Moreover, the present embodiments can provide for EBCs having a more uniform thickness than conventional techniques, such as plasma spraying, even when applying thin layers (<2 mils or less than about 50 micrometers). Additionally, the slurry deposition process can allow for the application of the EBCs to internal component passages as well as the ability to produce smooth surface finishes without an additional polishing step.

There can be occasions when the EBC develops small and/or narrow defects (c.g. about 10 microns to about 5mm in diameter; or about 10 microns to about 1mm in width) that need to be repaired. The following repair processes are applicable to the EBCs described herein and may be carried out after sintering of an individual EBC layer, or after sintering the entire applied EBC, as explained herein below.

In one embodiment, repairs may include remedying defects in one or more individual layers as the EBC is being applied using the methods described herein. In this embodiment, the repair can be carried out after sintering a given layer by applying a repair slurry comprising the same slurry materials used to make the layer having the defects. For example, if the transition layer develops a defect after sintering, the defect could be repaired using a “transition layer repair slurry” that comprises the same transition layer slurry materials used in the original application of the transition layer. In one embodiment, the repair slurry can comprise a higher solids loading of primary material ceramic particles than the original slurry layer as this can reduce

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shrinkage on drying and sintering of the repaired portion of the coating. In particular, the solids loading of primary material ceramic particles in the repair slurry can be greater than about 30% to about 55% by volume (as opposed to greater than about 10% by volume in one embodiment of the original slurry, and from about 10% to about 55% by volume in another embodiment of the original slurry used to make the layer). The repair slurry may be applied using any conventional method including those described previously, and the resulting “repair(ed) coating” may then be processed as described previously herein before application of any subsequent layer of the EBC.

In an alternate embodiment, repairs may include fixing defects after application and sintering of the entire EBC. In this embodiment, the repair may be carried out on the EBC having defects using a transition layer repair slurry comprising the same materials present in the previously defined transition layer slurry (i.e. primary transition material, a sintering aid, and optionally secondary material). This particular repair slurry can seep into any defects present in the EBC and provide a hermetic seal to the repaired EBC coating after sintering. Again, the solids loading of the transition layer repair slurry may comprise upwards of about 30% to 55% by volume.

Additionally, repair processes may be used to reduce surface roughness of a plasma sprayed EBC having any composition. Specifically, if the surface roughness of a plasma sprayed EBC is unacceptable the coating can be smoothed over by applying either of the previously described transition layer slurry or outer layer slurry. When applied over the plasma sprayed EBC, the transition layer slurry or outer layer slurry can fill in any gaps, grooves, or uneven portions of the plasma sprayed coating and reduce the surface roughness to an acceptable degree. More specifically, depending on the thickness of the transition layer slurry or outer layer slurry, surface roughness of the plasma sprayed EBC can be reduced from greater than 200 micro inch (about 5 micrometers) Ra, to between 40 micro inch Ra and 200 micro inch (about 1 to about 5 micrometers) Ra in one embodiment, and from between 40 micro inch Ra to 150 micro inch (about 1 to about 3.8 micrometers) Ra in another embodiment. In one embodiment, the transition layer slurry or outer layer slurry can comprise a thickness of at least about 0.5 mils (about 12.5 micrometers), and in another embodiment from about 0.5 mils to about 3 mils (about 12.5 to about 75 micrometers). The applied transition layer slurry can then be processed as described

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previously to produce a repaired EBC having an acceptable surface roughness. Additional slurry layers may be applied to the EBC if desired.

Such repair processes can provide the ability to repair localized defects, at varying points during the application or life of the coating, as opposed to stripping off and reapplying the entire coating. This, in turn, can result in a savings of time, labor, and materials.

## EXAMPLES

Example 1: A silicon bond coat was applied to a SiC-SiC CMC using a conventional air plasma spray process. Next, a primary transition material slurry was made by first mixing ytterbium disilicate (primary transition material), iron oxide nanoparticles (sintering aid), ethanol (solvent), and polyethylenimine (dispersant) in a plastic container, along with enough 0.25 inch (6.35mm) diameter, spherical zirconia media to line the bottom of container. This mixture was placed on a roller mill for 15 hours. After taking the container off of the roller mill, the zirconia media was removed and the slurry was filtered through a 325 mesh screen to remove any large particle agglomerates.

The resulting primary transition material slurry (Slurry A) consisted of 56.11% ytterbium disilicate, 0.54% iron oxide, 0.57% polyethylenimine, and the balance ethanol (all percents by weight). The silicon-coated ceramic component was dipped into Slurry A, dried in ambient conditions, re-dipped into Slurry A, dried in ambient conditions, and heat-treated at 10°C/minute to 1000°C to burn out any residual organic materials. Then, the temperature was increased at 10°C/minute from 1000°C to 1344°C and held for 10 hours to form a transition layer comprising iron-doped ytterbium disilicate. The entire heat treatment was carried out in air. The heating environment resulted in the transition layer having a porosity of less than 15% by volume. These dipping, drying, and heat treating processes were repeated 8 times to build thickness of the transition layer.

Next, a primary outer material slurry was made by first mixing ytterbium monosilicate (primary outer material), iron oxide nanoparticles (sintering aid), ethanol (solvent),

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and polyethylenimine (dispersant) in a plastic container, along with enough 0.25 inch (6.35mm) diameter, spherical zirconia media to line the bottom of container. This mixture was placed on a roller mill for 15 hours. After taking the container off of the roller mill, the zirconia media was removed and the slurry was filtered through a 325 mesh screen to remove any large particle agglomerates.

The resulting primary outer material slurry (Slurry B) consisted of 61.19% ytterbium monosilicate, 0.29% iron oxide, 0.61% polyethylenimine, and the balance ethanol (all percents by weight). The silicon and transition layer-coated ceramic component was dipped into Slurry B, dried in ambient conditions, re-dipped into Slurry B, dried in ambient conditions, and heat-treated at 10°C/minute to 1000°C to burn out any residual organic materials. Then, the temperature was increased at 10°C/minute from 1000°C to 1344°C and held for 10 hours to densify the transition layer. After heat treatment, the layer comprised iron-doped yttrium monosilicate. The entire heat treatment was carried out in air. The heating environment resulted in the transition layer having a porosity of less than 15% by volume. The dipping, drying, and heat treatments were repeated 2 times to build thickness of the outer layer.

FIG. 2. shows a SEM micrograph of this coating microstructure with the air plasma spray silicon bond coat (100), a thin silica layer (3.9 micrometers) (102), transition layer (104), and outer layer (106). The outer layer and transition layer appeared to consist only of iron-doped primary materials (iron-doped ytterbium monosilicate and iron-doped ytterbium disilicate, respectively), x-ray diffraction was not able to detect any secondary material present in the EBC. EDS analysis of the layers suggested that the iron oxide had dissolved into the primary materials.

**Example 2:** A CMC (101) coated with the EBC of Example 1 was exposed to 2400°F (1316°C) steam for 1000 hours. FIG. 3 shows a SEM micrograph of this coating after high temperature steam exposure with silicon bond coat (100), a silica layer (102) that has grown to approximately 7.0 micrometer thickness, an iron-doped ytterbium disilicate transition layer (104), and a dense ytterbium monosilicate outer layer (106) with cracks. Some ytterbium disilicate has formed around the cracks in the

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monosilicate layer as an artifact of high gaseous silicon content in the static atmosphere of the steam test.

Example 3: To demonstrate proof of principle, a primary transition layer was deposited on a silicon metal wafer using a slurry deposition process. A primary transition material slurry was made by first mixing ytterbium disilicate powder (primary transition material), gallium oxide powder (sintering aid), ethanol (solvent), and polyethylenimine (dispersant) in a plastic container, along with enough 0.25 inch (6.35mm) diameter, spherical zirconia media to line the bottom of container. This mixture was placed on a roller mill for 15 hours. After taking the container off of the roller mill, the zirconia media was removed and the slurry was filtered through a 325 mesh screen to remove any large particle agglomerates.

The resulting primary transition material slurry (Slurry C) consisted of 56.35% ytterbium disilicate, 0.64% gallium oxide, 0.57% polyethylenimine, and the balance ethanol (all percents by weight). The silicon-coated ceramic component was dipped into Slurry C, dried in ambient conditions, re-dipped into Slurry C, dried in ambient conditions, and heat-treated at 10°C/minute to 1000°C to burn out any residual organic materials. Then, the temperature was increased at 10°C/minute from 1000°C to 1344°C and held for 10 hours to form a transition layer comprising iron-doped ytterbium disilicate. The entire heat treatment was carried out in air. The heating environment resulted in the transition layer (108) having a porosity of less than 15% by volume as shown in FIG. 4. No secondary phases were observed in this coating via electron microscope examination. EDS analysis suggested that the gallium had dissolved into the ytterbium disilicate primary material.

This written description uses examples to disclose the invention, including the best mode, and also to enable any person skilled in the art to make and use the invention. The patentable scope of the invention may include other examples that occur to those skilled in the art in view of the description. Such other examples are intended to be within the scope of the invention.

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## WHAT IS CLAIMED IS:

1. A method for making an environmental barrier coating on a component using a sintering aid, the method comprising:

combining at least an organic solvent and a primary transition material comprising a rare earth disilicate to produce a transition layer slurry;

applying a bond coat layer to the component;

applying the transition layer slurry to the bond coat layer;

drying the applied transition layer slurry to form a dried transition layer;

infiltrating a sol-gel solution into the dried transition layer to form an infiltrated transition layer, the sol-gel solution comprising a sintering aid chosen from the group consisting of aqueous solutions of a water-soluble salt sintering aid and organic solvent solutions of a solvent-soluble salt sintering aid; and

sintering the infiltrated transition layer to produce the environmental barrier coating having at least the bond coat layer and a transition layer formed by the infiltrated transition layer;

wherein the sintering aid dissolves into the primary transition material during sintering to form a doped rare earth disilicate, optionally any of the sintering aid that does not dissolve into the primary transition material forms a secondary material, and the transition layer contains up to about 15% by volume of the secondary material;

wherein the water-soluble salt sintering aid is selected from the group consisting of water soluble iron salts, water soluble gallium salts, water soluble aluminum salts, water soluble nickel salts, water titanium salts, water soluble boron salts, water soluble alkaline earth salts; and

wherein the solvent-soluble salt sintering aid is selected from the group consisting of solvent soluble iron salts, solvent soluble gallium salts, solvent soluble aluminum salts, solvent soluble nickel salts, solvent soluble titanium salts, solvent soluble boron salts, and solvent soluble alkaline earth salts.

2. The method of claim 1 wherein the organic solvent is selected from the group consisting of methanol, ethanol, propanol, butanol, pentanol, hexanol, heptanol, octanol, nonanol, decanol, dodecanol, acetone, methyl isobutyl ketone,

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methyl ethyl ketone, toluene, ethylbenzene, propyl benzene, methoxybenzene, heptane, octane, nonane, decane, xylene, mineral spirits, naphtha, tetrahydrofuran, ethers, and mixtures thereof.

3. The method of claim 1, wherein the combining step comprises combining from about 6.8 wt % to about 96.1 wt % of the organic solvent; from about 3.9 wt % to about 93.2 wt % of the primary transition material; and optionally one or more organic processing aids to produce the transition layer slurry.

4. The method of claim 1, the method further comprising:

combining from about 6.8 wt % to about 96.1 wt % of a solvent, from about 3.9 wt % to about 93.2 wt % of a primary outer material comprising a rare earth monosilicate or a doped rare earth monosilicate, and optionally one or more organic processing aids to produce an outer layer slurry;

combining from about 6.8 wt % to about 96.1 wt % of a solvent, from about 3.9 wt % to about 93.2 wt % of a primary compliant material comprising Barium Strontium Aluminosilicate (BSAS) or rare earth doped BSAS, and optionally any of the one or more organic processing aids to produce a compliant layer slurry; and

applying any one or more of the outer layer slurry or the compliant layer slurry to the component prior to the sintering step to form any one or more of an outer slurry layer and a compliant slurry layer.

5. The method of claim 4 wherein the organic processing aids comprise:

from about 0 wt % to about 8.9 wt % dispersant;

from about 0 wt % to about 15.4 wt % plasticizer;

from about 0 wt % to about 1 wt % surfactant; and

from about 0 wt % to about 15.4 wt % thickener.

6. The method of claim 4 further comprising burning out the organic processing aids after the drying step but prior to the sintering step by heating the component at a rate of from about 1° C/minute to about 15° C/minute to a temperature

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of from about 400° C to about 1000° C, and holding the component at that temperature for up to about 10 hours.

7. The method of claim 6, wherein the sintering step is performed after burning out the organic processing aids and comprises heating the component at a rate of about 1° C/minute to about 15° C/minute to a temperature of about 1100° C to about 1700° C and holding the component at that temperature for up to about 24 hours.

8. The method of claim 4 wherein the sintering step forms an outer layer from the outer slurry layer and a compliant layer from the compliant slurry layer, the transition layer comprises a thickness of about 2.5 to about 1000 micrometers, the outer layer comprises a thickness of about 2.5 to about 75 micrometers, and the compliant layer comprises a thickness of about 2.5 to about 1000 micrometers.

9. The method of claim 1, wherein the component comprises a ceramic matrix composite or a monolithic ceramic turbine engine component selected from the group consisting of combustor components, turbine blades, shrouds, nozzles, heat shields, and vanes.

10. The method of claim 1, wherein the transition layer comprises at least one defect after the sintering step, the method further comprising repairing the defect by:

applying a transition layer repair slurry to the transition layer defect; and  
sintering the transition layer repair slurry to produce a repaired transition layer wherein the transition layer repair slurry comprises a primary transition material solids loading of from about 30% to about 55% by volume.

11. The method of claim 1 wherein the sintering aid completely dissolves into the primary transition material during sintering.

12. The method of claim 1 wherein the transition layer consists of a single phase and does not contain any of the secondary material.

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13. The method of claim 1, wherein the sol-gel solution further comprises a second sintering aid chosen from the group consisting of iron oxides and aluminum oxide.

14. A method for making an environmental barrier coating on a component using a sintering aid, the method comprising:

combining at least an organic solvent and a primary transition material comprising a rare earth disilicate to produce a transition layer slurry;

combining at least an organic solvent and a primary outer material comprising a rare earth monosilicate to produce an outer layer slurry;

combining at least an organic solvent and a primary compliant material comprising Barium Strontium Aluminosilicate (BSAS) or a rare earth doped BSAS to produce a compliant layer slurry;

applying the transition layer slurry, the outer layer slurry, and the compliant layer slurry to the component;

drying the transition layer slurry, the outer layer slurry, and, the compliant layer slurry on the component to form a dried transition layer, a dried outer layer, and a dried compliant layer;

infiltrating a sol-gel solution into the dried transition layer, the dried outer layer, and the dried compliant layer, the sol-gel solution comprising a sintering aid chosen from the group consisting of aqueous solutions of a water-soluble salt sintering aid and organic solvent solutions of a solvent-soluble salt sintering aid; and then

sintering the dried transition layer, the dried outer layer, and the dried compliant layer to produce the environmental barrier coating having a transition layer formed by the dried transition layer, an outer layer formed by the dried outer layer, and a compliant layer formed by the dried compliant layer;

wherein the sintering aid dissolves into the primary transition material and the primary outer material during sintering to form a doped rare earth disilicate in the transition layer and form a doped rare earth monosilicate in the outer layer;

wherein the water-soluble salt sintering aid is selected from the group consisting of water soluble iron salts, water soluble gallium salts, water soluble

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aluminum salts, water soluble nickel salts, water titanium salts, water soluble boron salts, and water soluble alkaline earth salts; and

wherein the solvent-soluble sintering aid is selected from the group consisting of solvent soluble iron salts, solvent soluble gallium salts, solvent soluble aluminum salts, solvent soluble nickel salts, solvent soluble titanium salts, solvent soluble boron salts, and solvent soluble alkaline earth salts.

15. The method of claim 14 wherein the organic solvents are selected from the group consisting of methanol, ethanol, propanol, butanol, pentanol, hexanol, heptanol, octanol, nonanol, decanol, dodecanol, acetone, methyl isobutyl ketone, methyl ethyl ketone, toluene, ethylbenzene, propyl benzene, methoxybenzene, heptane, octane, nonane, decane, xylene, mineral spirits, naphtha, tetrahydrofuran, ethers, and mixtures thereof.

16. The method of claim 14, wherein the combining steps comprise:  
 combining from about 6.8 wt % to about 96.1 wt % of the organic solvent;  
 from about 3.9 wt % to about 93.2 wt % of the primary transition material; and  
 optionally one or more organic processing aids to produce the transition layer slurry;  
 combining from about 6.8 wt % to about 96.1 wt % of the organic solvent,  
 from about 3.9 wt % to about 93.2 wt % of the primary outer material, and optionally  
 one or more organic processing aids to produce the outer layer slurry; and  
 combining from about 6.8 wt % to about 96.1 wt % of the organic solvent,  
 from about 3.9 wt % to about 93.2 wt % of the primary compliant material, and  
 optionally one or more organic processing aids to produce the compliant layer slurry.

17. The method of claim 16 wherein the organic processing aids comprise:  
 from about 0 wt % to about 8.9 wt % dispersant;  
 from about 0 wt % to about 15.4 wt % plasticizer;  
 from about 0 wt % to about 1 wt % surfactant; and  
 from about 0 wt % to about 15.4 wt % thickener.

18. The method of claim 16 further comprising burning out the organic processing aids after the drying step but prior to the sintering step by heating the

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component at a rate of from about 1° C/minute to about 15° C/minute to a temperature of from about 400° C to about 1000° C, and holding the component at that temperature for up to about 10 hours.

19. The method of claim 18, wherein the sintering step is performed after burning out the organic processing aids and comprises heating the component at a rate of about 1° C/minute to about 15° C/minute to a temperature of about 1100° C to about 1700° C and holding the component at that temperature for up to about 24 hours.

20. The method of claim 14 wherein the transition layer comprises a thickness of about 2.5 to about 1000 micrometers, the outer layer comprises a thickness of about 2.5 to about 75 micrometers, and the compliant layer comprises a thickness of about 2.5 to about 1000 micrometers.

21. The method of claim 14, wherein the component comprises a ceramic matrix composite or a monolithic ceramic turbine engine component selected from the group consisting of combustor components, turbine blades, shrouds, nozzles, heat shields, and vanes.

22. The method of claim 14, wherein the transition layer comprises at least one defect after the sintering step, the method further comprising repairing the defect by:

applying a transition layer repair slurry to the transition layer defect; and  
sintering the transition layer repair slurry to produce a repaired transition layer,

wherein the transition layer repair slurry comprises a primary transition material solids loading of from about 30% to about 55% by volume.

23. The method of claim 14, wherein the compliant layer comprises at least one defect after the sintering step, the method further comprising repairing the defect by:

applying a compliant layer repair slurry to the compliant layer defect; and

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sintering the compliant layer repair slurry to produce a repaired compliant layer,

wherein the compliant layer repair slurry comprises a primary compliant material solids loading of from about 30% to about 55% by volume.

24. The method of claim 14, wherein the environmental barrier coating comprises at least one defect in the outer layer after the sintering step, the method further comprising repairing the defect by:

applying a transition layer repair slurry to the outer layer defect of the environmental barrier coating; and

sintering the transition layer repair slurry to produce a repaired environmental barrier coating,

wherein the transition layer repair slurry comprises a primary transition material solids loading of from about 30% to about 55% by volume.

25. The method of claim 14 wherein the sintering aid completely dissolves into the primary transition material and the primary outer material during sintering.

26. The method of claim 14 wherein the transition layer consists of a single phase and does not contain a secondary material.

27. The method of claim 14, wherein the sol-gel solution further comprises a second sintering aid is chosen from the group consisting of iron oxides and aluminum oxide.

28. A method for making an environmental barrier coating on a component using a sintering aid, the method comprising:

combining at least an organic solvent and a primary transition material comprising a rare earth disilicate to produce a transition layer slurry;

combining at least an organic solvent and a primary outer material comprising a rare earth monosilicate to produce an outer layer slurry;

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combining at least an organic solvent and a primary compliant material comprising Barium Strontium Aluminosilicate (BSAS) or a rare earth doped BSAS to produce a compliant layer slurry;

applying the transition layer slurry to the component;

drying the transition layer slurry on the component to form a dried transition layer;

infiltrating a sol-gel solution into the dried transition layer to form an infiltrated transition layer, the sol-gel solution comprising a sintering aid chosen from the group consisting of aqueous solutions of a water-soluble salt sintering aid and organic solvent solutions of a solvent-soluble salt sintering aid;

applying the outer layer slurry and any more of the compliant layer slurry or an additional transition layer slurry over the infiltrated transition layer;

drying the outer layer slurry and the any one or more of the compliant and additional transition layer slurries on the component to form a dried outer layer and any one or more of a dried compliant layer and a dried additional transition layer, respectively; and

sintering the infiltrated transition layer, the dried outer layer, and any one or more of the dried compliant layer and the dried additional transition layer to produce the environmental barrier coating having at least one transition layer formed by the infiltrated transition layer, an outer layer formed by the dried outer layer, and any one or more of a compliant layer or an additional transition layer formed by the dried compliant layer and the dried additional transition layer, respectively;

wherein the sintering aid dissolves into the primary transition material and the primary outer material during sintering to form a doped rare earth disilicate in the transition layer and form a doped rare earth monosilicate in the outer layer;

wherein the water-soluble salt sintering aid is selected from the group consisting of water soluble iron salts, water soluble gallium salts, water soluble aluminum salts, water soluble nickel salts, water titanium salts, water soluble boron salts, and water soluble alkaline earth salts; and

wherein the solvent-soluble salt sintering aid is selected from the group consisting of solvent soluble iron salts, solvent soluble gallium salts, solvent soluble

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aluminum salts, solvent soluble nickel salts, solvent soluble titanium salts, solvent soluble boron salts, and solvent soluble alkaline earth salts.

29. The method of claim 28 wherein the transition layer comprises a thickness of about 2.5 to about 1000 micrometers, the outer layer comprises a thickness of about 2.5 to about 75 micrometers, and the compliant layer comprises a thickness of about 2.5 to about 1000 micrometers.

30. The method of claim 28, wherein the step of applying the outer layer slurry and any one or more of the compliant layer slurry or an additional transition layer slurry comprises applying each of the outer layer slurry, the compliant layer slurry, and the additional transition layer slurry to the component.

31. The method of claim 28 wherein the organic solvents are selected from the group consisting of methanol, ethanol, propanol, butanol, pentanol, hexanol, heptanol, octanol, nonanol, decanol, dodecanol, acetone, methyl isobutyl ketone, methyl ethyl ketone, toluene, ethylbenzene, propyl benzene, methoxybenzene, heptane, octane, nonane, decane, xylene, mineral spirits, naphtha, tetrahydrofuran, ethers, and mixtures thereof.

32. The method of claim 28, wherein the combining steps comprise:  
combining from about 6.8 wt % to about 96.1 wt % of the organic solvent;  
from about 3.9 wt % to about 93.2 wt % of the primary transition material; and  
optionally one or more organic processing aids to produce the transition layer slurry;  
combining from about 6.8 wt % to about 96.1 wt % of the organic solvent,  
from about 3.9 wt % to about 93.2 wt % of the primary outer material, and optionally  
one or more organic processing aids to produce the outer layer slurry; and  
combining from about 6.8 wt % to about 96.1 wt % of the organic solvent,  
from about 3.9 wt % to about 93.2 wt % of the primary compliant material, and  
optionally one or more organic processing aids to produce the compliant layer slurry.

33. The method of claim 32 wherein the organic processing aids comprise:  
from about 0 wt % to about 8.9 dispersant;

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from about 0 wt % to about 15.4 wt % plasticizer;

from about 0 wt % to about 1 wt % surfactant; and

from about 0 wt % to about 15.4 wt % thickener.

34. The method of claim 32 further comprising burning out the organic processing aids after the drying steps but prior to the sintering steps by heating the component at a rate of from about 1° C/minute to about 15° C/minute to a temperature of from about 400° C to about 1000° C, and holding the component at that temperature for up to about 10 hours.

35. The method of claim 34, wherein the sintering step is performed after burning out the organic processing aids and comprises heating the component at a rate of about 1° C/minute to about 15° C/minute to a temperature of about 1100° C to about 1700° C and holding the component at that temperature for up to about 24 hours.

36. The method of claim 28, wherein the component comprises a ceramic matrix composite or a monolithic ceramic turbine engine component selected from the group consisting of combustor components, turbine blades, shrouds, nozzles, heat shields, and vanes.

37. The method of claim 28, wherein the transition layer comprises at least one defect after the sintering step, the method further comprising repairing the defect by:

applying a transition layer repair slurry to the transition layer defect; and sintering the transition layer repair slurry to produce a repaired transition layer,

wherein the transition layer repair slurry comprises a primary transition material solids loading of from about 30% to about 55% by volume.

38. The method of claim 28, wherein the compliant layer comprises at least one defect after the sintering step, the method further comprising repairing the defect by:

applying a compliant layer repair slurry to the compliant layer defect; and

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sintering the compliant layer repair slurry to produce a repaired compliant layer,

wherein the compliant layer repair slurry comprises a primary compliant material solids loading of from about 30% to about 55% by volume.

39. The method of claim 28, wherein the environmental barrier coating comprises at least one defect in the outer layer after the sintering step, the method further comprising repairing the defect by:

applying a transition layer repair slurry to the outer layer defect of the environmental barrier coating; and

sintering the transition layer repair slurry to produce a repaired environmental barrier coating,

wherein the transition layer repair slurry comprises a primary transition material solids loading of from about 30% to about 55% by volume.

40. The method of claim 28 wherein the sintering aid completely dissolves into the primary transition material and the primary outer material during sintering.

41. The method of claim 28 wherein the transition layer consists of a single phase and does not contain a secondary material.

42. The method of claim 28, wherein the sol-gel solution further comprises a second sintering aid is chosen from the group consisting of iron oxides and aluminum oxide.

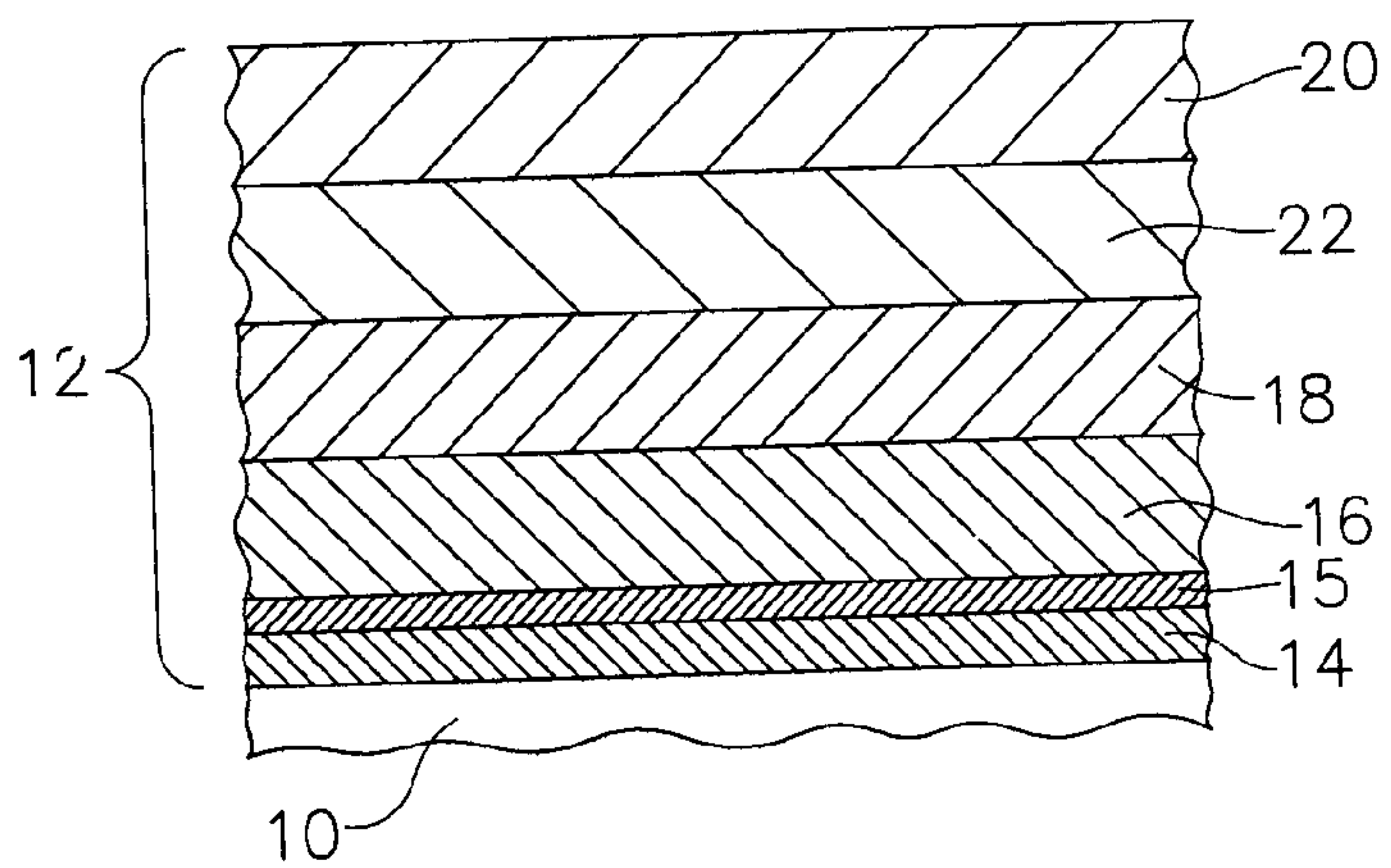


FIG. 1

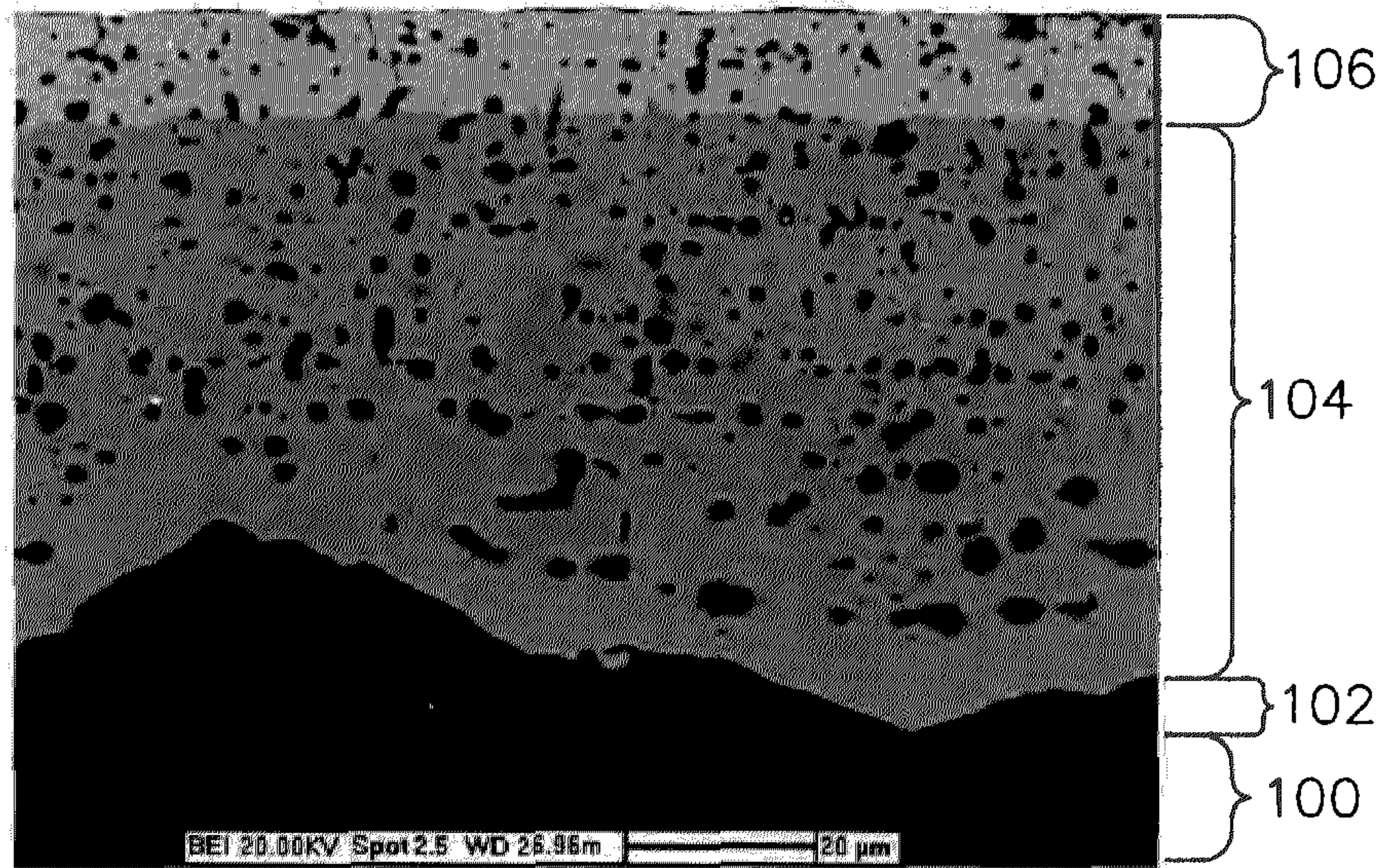


FIG. 2

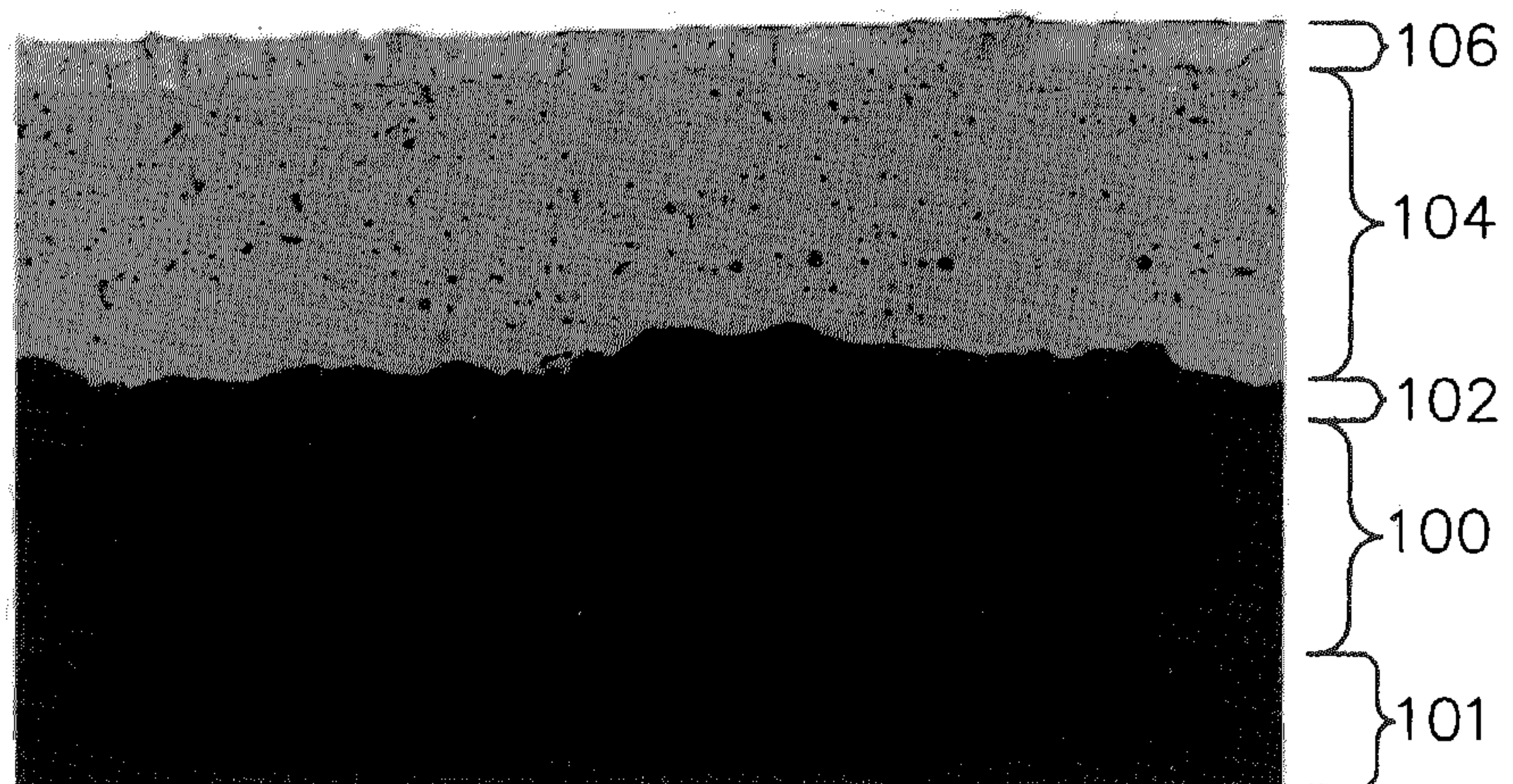


FIG. 3

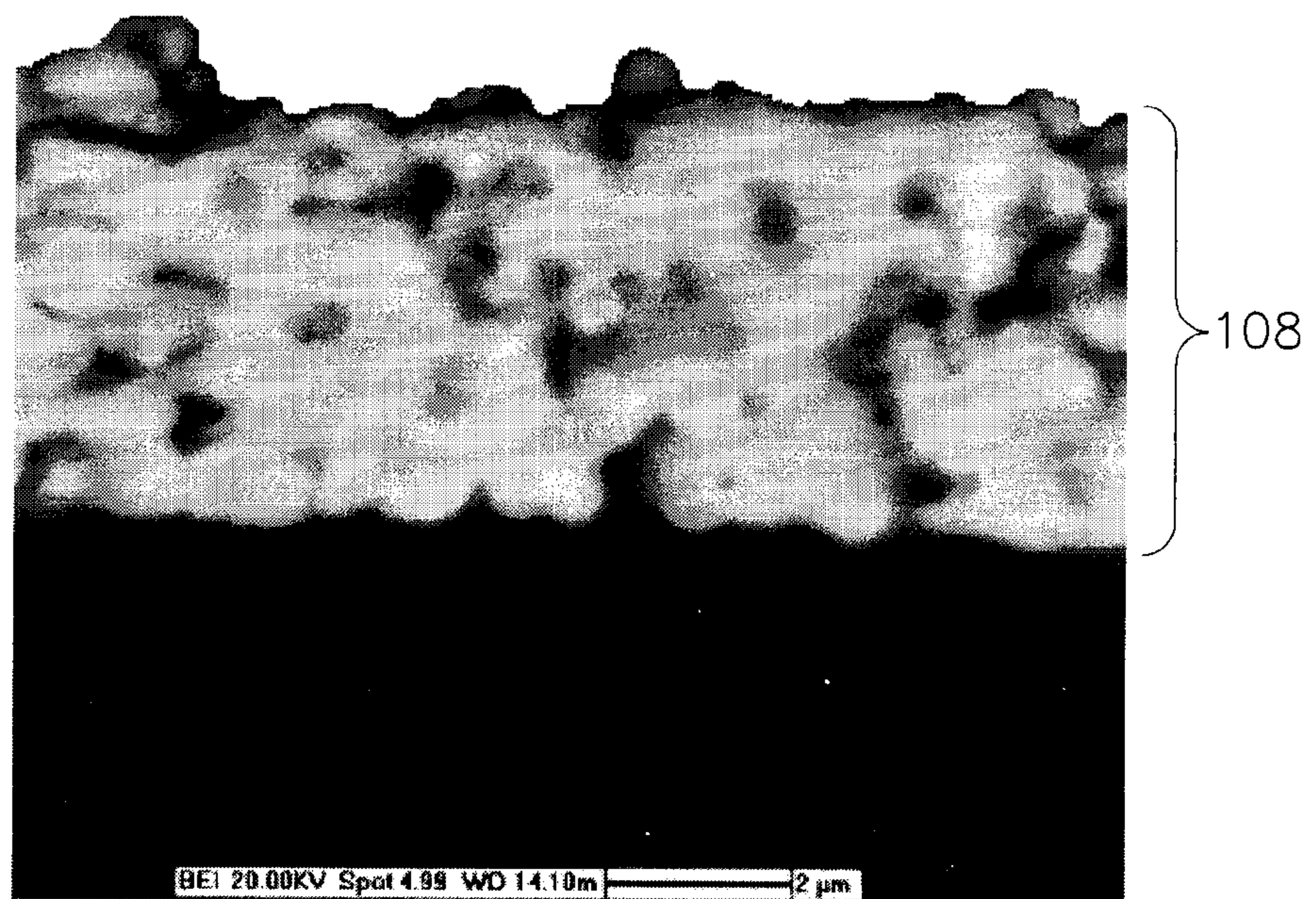


FIG. 4

