(51) International Patent Classification 6:
C07C 227/18

(11) International Publication Number: WO 96/16927
(43) International Publication Date: 6 June 1996 (06.06.96)

(21) International Application Number: PCT/EP95/04635
(22) International Filing Date: 24 November 1995 (24.11.95)
(30) Priority Data:
MI94A002412 29 November 1994 (29.11.94) IT

(71) Applicant (for all designated States except US): ZAMBON GROUP S.P.A. [IT/IT]; Via della Chimica, 9, I-36100 Vicenza (IT).

(72) Inventors; and
(75) Inventors/Applicants (for US only): CANNATA, Vincenzo [IT/IT]; Via Annibale Cl6, 12, I-40037 Sasso Marconi (IT). VELGI, Corrado [IT/IT]; Via Venezia, 23, I-36040 Sarego (IT). BARRECA, Giuseppe [IT/IT]; Via Vincenzo da Seregno, 46/6, I-20161 Milano (IT).


Published
With international search report.
Before the expiration of the time limit for amending the claims and to be republished in the event of the receipt of amendments.

(54) Title: PROCESS FOR PREPARING 5-AMINO-2,4,6-TRIODISOPTHALIC ACID DICHLORIDE BY CHLORINATION OF THE CORRESPONDING ACID IN THE PRESENCE OF A TERTIARY AMINE SALT OR QUATERNARY AMMONIUM SALT.

(57) Abstract

A process for the preparation of 5-amino-2,4,6-triodoisophthalic acid dichloride by chlorinating 5-amino-2,4,6-triodoisophthalic acid with thionyl chloride in the presence of a suitable solvent and of a tertiary amine salt or quaternary ammonium salt in a molar ratio from 1:1 to 1:2 with respect to 5-amino-2,4,6-triodoisophthalic acid is described. 5-amino-2,4,6-triodoisophthalic acid dichloride is an intermediate useful for the preparation of iodinated contrast agents.
FOR THE PURPOSES OF INFORMATION ONLY

Codes used to identify States party to the PCT on the front pages of pamphlets publishing international applications under the PCT.

<table>
<thead>
<tr>
<th></th>
<th></th>
<th></th>
<th></th>
<th></th>
<th></th>
</tr>
</thead>
<tbody>
<tr>
<td>AT</td>
<td>Austria</td>
<td>GB</td>
<td>United Kingdom</td>
<td>MR</td>
<td>Mauritania</td>
</tr>
<tr>
<td>AU</td>
<td>Australia</td>
<td>GE</td>
<td>Georgia</td>
<td>MW</td>
<td>Malawi</td>
</tr>
<tr>
<td>BB</td>
<td>Barbados</td>
<td>GN</td>
<td>Guinea</td>
<td>NE</td>
<td>Niger</td>
</tr>
<tr>
<td>BE</td>
<td>Belgium</td>
<td>GR</td>
<td>Greece</td>
<td>NL</td>
<td>Netherlands</td>
</tr>
<tr>
<td>BF</td>
<td>Burkina Faso</td>
<td>HU</td>
<td>Hungary</td>
<td>NO</td>
<td>Norway</td>
</tr>
<tr>
<td>BG</td>
<td>Bulgaria</td>
<td>IE</td>
<td>Ireland</td>
<td>NZ</td>
<td>New Zealand</td>
</tr>
<tr>
<td>BJ</td>
<td>Benin</td>
<td>IT</td>
<td>Italy</td>
<td>PL</td>
<td>Poland</td>
</tr>
<tr>
<td>BR</td>
<td>Brazil</td>
<td>JP</td>
<td>Japan</td>
<td>PT</td>
<td>Portugal</td>
</tr>
<tr>
<td>BY</td>
<td>Belarus</td>
<td>KE</td>
<td>Kenya</td>
<td>RO</td>
<td>Romania</td>
</tr>
<tr>
<td>CA</td>
<td>Canada</td>
<td>KG</td>
<td>Kyrgyzstan</td>
<td>RU</td>
<td>Russian Federation</td>
</tr>
<tr>
<td>CF</td>
<td>Central African Republic</td>
<td>KP</td>
<td>Democratic People’s Republic of Korea</td>
<td>SD</td>
<td>Sudan</td>
</tr>
<tr>
<td>CG</td>
<td>Congo</td>
<td>KL</td>
<td>Australia</td>
<td>SE</td>
<td>Sweden</td>
</tr>
<tr>
<td>CH</td>
<td>Switzerland</td>
<td>KZ</td>
<td>Kazakhstan</td>
<td>SI</td>
<td>Slovenia</td>
</tr>
<tr>
<td>CI</td>
<td>Côte d’Ivoire</td>
<td>LI</td>
<td>Liechtenstein</td>
<td>SK</td>
<td>Slovakia</td>
</tr>
<tr>
<td>CM</td>
<td>Cameroon</td>
<td>LK</td>
<td>Sri Lanka</td>
<td>SN</td>
<td>Senegal</td>
</tr>
<tr>
<td>CN</td>
<td>China</td>
<td>LU</td>
<td>Luxembourg</td>
<td>TD</td>
<td>Chad</td>
</tr>
<tr>
<td>CS</td>
<td>Czechoslovakia</td>
<td>LV</td>
<td>Latvia</td>
<td>TG</td>
<td>Togo</td>
</tr>
<tr>
<td>CZ</td>
<td>Czech Republic</td>
<td>MC</td>
<td>Monaco</td>
<td>TJ</td>
<td>Tajikistan</td>
</tr>
<tr>
<td>DE</td>
<td>Germany</td>
<td>MD</td>
<td>Republic of Moldova</td>
<td>TT</td>
<td>Trinidad and Tobago</td>
</tr>
<tr>
<td>DK</td>
<td>Denmark</td>
<td>MG</td>
<td>Madagascar</td>
<td>UA</td>
<td>Ukraine</td>
</tr>
<tr>
<td>ES</td>
<td>Spain</td>
<td>ML</td>
<td>Mali</td>
<td>US</td>
<td>United States of America</td>
</tr>
<tr>
<td>FI</td>
<td>Finland</td>
<td>MN</td>
<td>Mongolia</td>
<td>UZ</td>
<td>Uzbekistan</td>
</tr>
<tr>
<td>FR</td>
<td>France</td>
<td></td>
<td></td>
<td>VN</td>
<td>Viet Nam</td>
</tr>
</tbody>
</table>
PROCESS FOR PREPARING 5-AMINO-2,4,6-TRIIOIDOISOPHTHALIC ACID DICHLORIDE
BY CHLORINATION OF THE CORRESPONDING ACID IN THE PRESENCE OF A TERTIARY
AMINE SALT OR QUARTERNARY AMMONIUM SALT

The present invention relates to a process for the preparation of an intermediate useful in the
synthesis of organic compounds and, more particularly, it relates to a process for the
preparation of 5-amino-2,4,6-triiodoisophthalic acid dichloride.
5-Amino-2,4,6-triiodoisophthalic acid dichloride is a known compound useful for the
preparation of iodinated contrast agents among which Iopamidol (British patent No.
1,472,050 - Savac AG) and Ioversol (European patent application No. 0083964 -
Mallinckrodt Inc.) can be cited.

Several examples of synthesis of 5-amino-2,4,6-triiodoisophthalic acid dichloride are reported
in the literature and all of them foresee the chlorination of 5-amino-2,4,6-triiodoisophthalic
acid with thionyl chloride.

In particular, we can cite the syntheses described in the already mentioned British patent No.
1,472,050, in the Belgian patent No. 852,418 (Mallinckrodt Inc.) and in the U.S.A. patent No.
3,655,752 (Sterling Drug Inc.) which use a strong excess of thionyl chloride and which
require a long and cumbersome work up, hardly suitable under an industrial viewpoint, even if
in some cases they allow to afford the desired dichloride with high yields.

The synthesis described in the European patent application No. 0118347 (Guerbet S.A.)
foresees the use of excess thionyl chloride too, but in the presence of catalytic amounts of
N,N-dimethylformamide. The yields are high but the work up requires the removal of excess
thionyl chloride by evaporation also in this case.

The use of solvents such as ethyl acetate, as described in the International patent applications
No. WO 91/09007 (Mallinckrodt Inc.) and No. WO 93/10825 (Mallinckrodt Inc.) or in the
already mentioned European patent application No. 0083964, does not allow to obtain the
desired dichloride with satisfactory yields.

We have now found that, by carrying out the chlorination reaction of 5-amino-2,4,6-
triiodoisophthalic acid with thionyl chloride in a suitable solvent in the presence of a tertiary
amine salt or quaternary ammonium salt, the corresponding dichloride is obtained not only in
high yields but also substantially free from impurities and in crystalline form by simple
dilution of the reaction mixture with water.
Therefore, object of the present invention is a process for the preparation of 5-amino-2,4,6-triiodoisophthalic acid dichloride by chlorination of 5-amino-2,4,6-triiodoisophthalic acid with thionyl chloride in the presence of a solvent characterized in that the reaction is carried out in the presence of a tertiary amine salt or quaternary ammonium salt in a molar ratio from 1:1 to 1:2 with respect to 5-amino-2,4,6-triiodoisophthalic acid.

The 5-amino-2,4,6-triiodoisophthalic acid dichloride obtained according to the process of the present invention is useful as intermediate in the synthesis of iodinated contrast agents.

The amount of thionyl chloride used in the process object of the present invention is generally from 2 to 8 moles with respect to 5-amino-2,4,6-triiodoisophthalic acid.

Preferably, from 4 to 6 moles of thionyl chloride by mole of 5-amino-2,4,6-triiodoisophthalic acid are used.

Tertiary amine salts which can be used in the process object of the present invention are generally hydrohalides, preferably hydrochlorides or hydrobromides.

Quaternary ammonium salts which can be used in the process object of the present invention are generally halides, preferably chlorides or bromides.

The tertiary amines are generally trialkylamines, preferably triethylamine.

Quaternary ammonium salts are generally tetraalkylammonium salts, preferably tetraethylammonium or tetrabutylammonium salts.

Examples of tertiary amine salts and of quaternary ammonium salts used in the process object of the present invention are triethylamine hydrochloride, triethylamine hydrobromide, tetrabutylammonium chloride, tetrabutylammonium bromide, tetraethylammonium bromide and tetrabutylammonium chloride.

Preferably triethylamine hydrochloride, which can be optionally prepared in situ, is used.

The amount of tertiary amine salt or quaternary ammonium salt is preferably equimolar (molar ratio 1:1) with respect to 5-amino-2,4,6-triiodoisophthalic acid.

Suitable solvents are ethyl acetate, methylene chloride, chloroform and 1,2-dichloroethane.

Preferably methylene chloride is used.

A preferred practical embodiment of the present invention is the following.

Thionyl chloride is added to a suspension of 5-amino-2,4,6-triiodoisophthalic acid, triethylamine and methylene chloride and the reaction mixture is heated under reflux for some
hours.
At the end of the addition, water is added and the precipitation of a crystalline product is observed.

By simple filtration and washing, 5-amino-2,4,6-triiodoisophthalic acid dichloride is obtained in pure form.
The characterizing feature of the process object of the invention is the presence of amounts at least equimolar of a tertiary amine salt or of a quaternary amine salt.
The use of said salt in the above indicated amounts allows to obtain the desired product with extremely high yields and, above all, free from impurities which should make difficult its isolation with a purity degree suitable for its use as intermediate in the synthesis of iodinated contrast agents.

It is worth underlining that the substantial absence of by-products in the reaction of preparation of 5-amino-2,4,6-triiodoisophthalic acid dichloride according to the process object of the present invention does not result exclusively in an improvement of the overall yield of the process with respect to the known methods, but allows also the isolation of the desired product with simple operations of dilution in water of the reaction mixture.

It is evident to the man skilled in the art the advantage deriving from the possibility of carrying out a process for the synthesis of an intermediate with high yields, high purity, through extremely simple operations, without requiring either the removal of thionyl chloride by evaporation or repeated purification operations for isolating the desired product in pure form.
As already underlined, contrary to what described in the known processes for the preparation of 5-amino-2,4,6-triiodoisophthalic acid dichloride, the isolation of the pure product is carried out by simple dilution with water of the reaction mixture.

As far as we know, a mechanism able to explain the unexpected advantages deriving from the use of a tertiary amine salt or of a quaternary ammonium salt according to the process object of the present invention cannot be assumed.

Probably, the salt forms an adduct with thionyl chloride.
In this connection, it is worth noting that the literature (European patent application No. 0026281 - Bracco Industria Chimica S.p.A.) describes the preparation of 5-methylamino-2,4,6-triiodoisophthalic acid dichloride by reaction with thionyl chloride in the presence of
small amounts of quinoline, without reporting the yields yet.
However, the use of quinoline or of a tertiary amine salt or quaternary ammonium salt in molar amounts significantly lower than 1:1 with respect to 5-aminoo-2,4,6-triiodoisophthalic acid does not allow to obtain the desired product with high yields and, above all, with a suitable purity degree. In particular, by using quinoline in catalytic amounts as described in the European patent application No. 0026281, the desired dichloride is obtained in admixture with not negligible amounts of by-products and then with a purity degree not suitable for the use as intermediate in the subsequent steps of the process for the preparation of iodinated contrast agents.

In order to illustrate the present invention the following examples are now given.

Example 1

Methylene chloride (280 g), triethylamine (25.5 g; 0.25 moles) and water (2.25 g; 0.125 moles) were charged in this order into a reactor.

After addition of thionyl chloride (14.9 g; 0.125 moles) dropwise while keeping the temperature below 30°C, 5-aminoo-2,4,6-triiodoisophthalic acid (140 g; 0.25 moles) and, in about 1.5 hours, thionyl chloride (143 g; 1.2 moles) were added while keeping the internal temperature below 36°C and adjusting the addition rate with the gas emission.

At the end of the addition, the reaction mixture was kept under reflux (43°C) for 28 hours.

After cooling to about 30°C, methylene chloride (65 g) and, in small portions, water (100 g) were added.

At about half of the addition, the formation of a crystalline precipitate was observed.

After filtration, the resultant product was reduced to pulp with water (200 g) and then dried in oven under vacuum at 60°C for about 21 hours obtaining 5-aminoo-2,4,6-triiodoisophthalic acid dichloride (139 g; 93.3% yield) practically pure by HPLC analysis and by thin layer chromatography.

Example 2

Methylene chloride (280 g), triethylamine (51 g; 0.5 moles) and water (4.5 g; 0.25 moles) were charged in this order into a reactor.

After addition of thionyl chloride (30 g; 0.25 moles) dropwise while keeping the temperature below 30°C, 5-aminoo-2,4,6-triiodoisophthalic acid (140 g; 0.25 moles) and, in about 2 hours,
thionyl chloride (143 g, 1.2 moles) were added while keeping the internal temperature below 36°C and adjusting the addition rate with the gas emission.

At the end of the addition, the reaction mixture was kept under reflux (about 43°C) for 22 hours.

After cooling to about 30°C, water (100 g) was added in small portions.

At about half of the addition, the formation of a crystalline precipitate was observed.

Subsequently, the addition of water (250 g total) was completed and the precipitate was filtered.

The resultant product was reduced to pulp twice with water (2x300 g) and then dried in oven under vacuum at 60°C for about 21 hours obtaining 5-amino-2,4,6-triiodoisophthalic acid dichloride (140.7 g, 94.4% yield) practically pure by HPLC analysis and by thin layer chromatography.

Example 3

Methylene chloride (112 g) and tetrabutylammonium bromide (3.23 g, 0.1 moles) were charged in this order into a reactor.

5-Amino-2,4,6-triiodoisophthalic acid (55.9 g, 0.1 moles) and, in about 2 hours, thionyl chloride (57.1 g, 0.479 moles) were added to the mixture while keeping the internal temperature below 36°C and adjusting the addition rate with the gas emission.

At the end of the addition, the reaction mixture was kept under reflux (43°C) for 27 hours.

After cooling to about 20°C, water (60 g) was added in small portions.

At about half of the addition, the formation of a crystalline precipitate was observed.

After filtration, the resultant product was washed with water (5x50 g) and then dried in oven under vacuum at 50°C for about 21 hours obtaining 5-amino-2,4,6-triiodoisophthalic acid dichloride (49.99 g, 83.8% yield) practically pure by HPLC analysis and by thin layer chromatography.
- 6 -

Claims

1) A process for the preparation of 5-amino-2,4,6-triodoisophthalic acid dichloride by chlorination of 5-amino-2,4,6-triodoisophthalic acid with thionyl chloride in the presence of a solvent characterized in that the reaction is carried out in the presence of a tertiary amine salt or quaternary ammonium salt in a molar ratio from 1:1 to 1:2 with respect to 5-amino-2,4,6-triodoisophthalic acid.

2) A process according to claim 1 wherein the amount of thionyl chloride is from 2 to 8 moles by mole of 5-amino-2,4,6-triodoisophthalic acid.

3) A process according to claim 2 wherein the amount of thionyl chloride is from 4 to 6 moles by mole of 5-amino-2,4,6-triodoisophthalic acid.

4) A process according to claim 1 wherein the tertiary amine salts or quaternary amine salts are tertiary amine hydrohalides or quaternary ammonium halides.

5) A process according to claim 4 wherein the salts are tertiary amine hydrochlorides or hydrobromides or quaternary ammonium chlorides or bromides.

6) A process according to claim 1 wherein the tertiary amines are trialkylamines.

7) A process according to claim 6 wherein the tertiary amine is triethylamine.

8) A process according to claim 1 wherein the quaternary ammonium salts are tetraalkylammonium salts.

9) A process according to claim 8 wherein the quaternary ammonium salt is selected among tetraethylammonium and tetrabutylammonium salts.

10) A process according to claim 1 wherein the tertiary amine salts or quaternary ammonium salts are selected among triethylamine hydrochloride, triethylamine hydrobromide, tetrabutylammonium chloride, tetrabutylammonium bromide, tetraethylammonium bromide and tetraethylammonium chloride.

11) A process according to claim 10 wherein triethylamine hydrochloride is used.

12) A process according to claim 1 wherein the amount of tertiary amine salt or quaternary ammonium salt is equimolar with respect to 5-amino-2,4,6-triodoisophthalic acid.

13) A process according to claim 1 wherein the solvent is selected among ethyl acetate, methylene chloride, chloroform and 1,2-dichloroethane.

14) A process according to claim 13 wherein the solvent is methylene chloride.
**INTERNATIONAL SEARCH REPORT**

**A. CLASSIFICATION OF SUBJECT MATTER**

IPC 6 C07C227/18

According to International Patent Classification (IPC) or to both national classification and IPC.

**B. FIELDS SEARCHED**

Minimum documentation searched (classification system followed by classification symbols)

IPC 6 C07C

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched.

Electronic data base consulted during the international search (name of data base and, where practical, search terms used).

**C. DOCUMENTS CONSIDERED TO BE RELEVANT**

<table>
<thead>
<tr>
<th>Category</th>
<th>Citation of document, with indication, where appropriate, of the relevant passages</th>
<th>Relevant to claim No.</th>
</tr>
</thead>
<tbody>
<tr>
<td>A</td>
<td>EP A 0 026 281 (BRACCO INDUSTRIA CHIMICA) 8 April 1981 cited in the application see page 18, line 11 - line 23</td>
<td>1</td>
</tr>
<tr>
<td>A</td>
<td>EP A 0 118 347 (GUEBERT S.A.) 12 September 1984 cited in the application see page 9, line 19 - line 27</td>
<td>1</td>
</tr>
<tr>
<td>A</td>
<td>EP A 0 083 964 (MALLINCKRODT INC.) 20 July 1983 see page 5, line 5 - line 24</td>
<td>1</td>
</tr>
</tbody>
</table>

Further documents are listed in the continuation of box C.

Patent family members are listed in annex.

"A" document defining the general state of the art which is not considered to be of particular relevance

"E" earlier document but published on or after the international filing date

"I" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)

"O" document referring to an oral disclosure, use, exhibition or other means

"P" document published prior to the international filing date but later than the priority date claimed

"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention

"X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone or in combination with one or more other such documents, such combination being obvious to a person skilled in the art.

"Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.

"&" document member of the same patent family

Date of the actual completion of the international search: 19 March 1996

Date of mailing of the international search report: 2.04.96

Name and mailing address of the ISA

European Patent Office, P.B. 5818 Patentfaan 2 NL - 2280 HV Rijswijk
Tel. (+31-70) 340-2040, Tx. 31 651 epo nl, Fax (+31-70) 340-3016

Authorized officer

Goetz, G

Form PCT/ISA/218 (second sheet) (July 1993)
<table>
<thead>
<tr>
<th>Patent document cited in search report</th>
<th>Publication date</th>
<th>Patent family member(s)</th>
<th>Publication date</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td></td>
<td>AR-A- 225836</td>
<td>30-04-82</td>
</tr>
<tr>
<td></td>
<td></td>
<td>AT-T- 1995</td>
<td>15-12-82</td>
</tr>
<tr>
<td></td>
<td></td>
<td>AU-B- 541428</td>
<td>10-01-85</td>
</tr>
<tr>
<td></td>
<td></td>
<td>AU-B- 6061680</td>
<td>12-02-81</td>
</tr>
<tr>
<td></td>
<td></td>
<td>CA-A- 1128065</td>
<td>20-07-82</td>
</tr>
<tr>
<td></td>
<td></td>
<td>JP-C- 1194629</td>
<td>12-03-84</td>
</tr>
<tr>
<td></td>
<td></td>
<td>JP-A- 56029553</td>
<td>24-03-81</td>
</tr>
<tr>
<td></td>
<td></td>
<td>JP-B- 58027264</td>
<td>08-06-83</td>
</tr>
<tr>
<td></td>
<td></td>
<td>SI-A- 8012011</td>
<td>31-12-94</td>
</tr>
<tr>
<td></td>
<td></td>
<td>US-A- 4352788</td>
<td>05-10-82</td>
</tr>
<tr>
<td></td>
<td></td>
<td>JP-A- 59193857</td>
<td>02-11-84</td>
</tr>
<tr>
<td></td>
<td></td>
<td>AU-B- 552188</td>
<td>22-05-86</td>
</tr>
<tr>
<td></td>
<td></td>
<td>AU-B- 9151182</td>
<td>21-07-83</td>
</tr>
<tr>
<td></td>
<td></td>
<td>CA-A- 1198739</td>
<td>31-12-85</td>
</tr>
<tr>
<td></td>
<td></td>
<td>JP-C- 1599742</td>
<td>31-01-91</td>
</tr>
<tr>
<td></td>
<td></td>
<td>JP-B- 2024252</td>
<td>29-05-90</td>
</tr>
<tr>
<td></td>
<td></td>
<td>JP-A- 58131970</td>
<td>06-08-83</td>
</tr>
</tbody>
</table>