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<p>(21) International Application Number: PCT/NL99/00115</p> <p>(22) International Filing Date: 4 March 1999 (04.03.99)</p> <p>(30) Priority Data: 1008477 4 March 1998 (04.03.98) NL</p> <p>(71) Applicant (for all designated States except US): INSTI- TUUT VOOR AGROTECHNOLOGISCH ONDERZOEK (ATO-DLO) [NL/NL]; P.O. Box 17, NL-6700 AA Wageningen (NL).</p> <p>(72) Inventors; and (75) Inventors/Applicants (for US only): LUITJES, Hendrikus [NL/NL]; Rijnhof 19, NL-6715 LR Ede (NL). JANSEN, Johannes, Carolus [NL/NL]; Troelstraweg 59, NL-6702 AG Wageningen (NL).</p> <p>(74) Agent: DE BRUIJN, Leendert, C.; Nederlandsch Octrooi- reau, Scheveningseweg 82, P.O. Box 29720, NL-2502 LS The Hague (NL).</p>	<p>(81) Designated States: AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, CA, CH, CN, CU, CZ, DE, DK, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MD, MG, MK, MN, MW, MX, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT, UA, UG, US, UZ, VN, YU, ZW, ARIPO patent (GH, GM, KE, LS, MW, SD, SL, SZ, UG, ZW), Eurasian patent (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European patent (AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE), OAPI patent (BF, BJ, CF, CG, CI, CM, GA, GN, GW, ML, MR, NE, SN, TD, TG).</p> <p>Published <i>With international search report.</i> <i>With amended claims.</i></p>	
<p>(54) Title: BICYCLOOCTANE DERIVATIVES AS PLASTICISERS</p>		
<p>(57) Abstract</p> <p>The use of isosorbide esters, isosorbide polyesters, isosorbide ethers, isosorbide carbonates, isosorbide thioethers, isosorbide thioesters, isosorbide amides, isosorbide (thio)urethanes, isosorbide urea, isosorbide phosphates and isosorbide phosphonates as a plasticiser is described. Instead of isosorbide it is also possible for corresponding diols such as isomannide and similar compounds, in which the ring oxygen atoms of the isosorbide have been replaced by carbon or by other heteroatoms, to serve as a base. In particular, this relates to C³-C₁₁ alkanooates of isosorbide, isomannide and the like. In particular, the compounds are used as a plasticiser in poly(vinyl chloride), optionally in conjunction with usual plasticisers such as dioctyl phthalates.</p>		

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Bicyclooctane derivatives as plasticisers

The invention relates to the use of bicyclo[3.3.0]octane derivatives as plasticisers or solvents for polymers, including thermoplastics.

Plasticisers are indispensable components for thermoplastic materials such as poly(vinyl chloride), and other polymer systems. The most commonly used plasticisers are diesters of dicarboxylic acids, among which the phthalic acid diesters are frontrunners in the plasticiser market with 89%, about half of which is made up of the isomeric diisooctyl esters. Whilst the existing plasticisers are generally satisfactory, there is a need for the range of plasticisers to be extended. Moreover, the existing plasticisers are virtually without exception based on petroleum products, whereas it is desirable for these to be replaced, as far as possible, by renewable raw materials. From an environmental and health point of view it is likewise desirable for dialkyl phthalates to be replaced.

Polyesters based on isosorbide and dicarboxylic acids (adipic acid, suberic acid, sebacic acid and dodecanedioic acid) having molecular weights above 10,000 have been suggested by Braun and Bergmann (*Angew. Makromol. Chem.* **199** (1992), 191-205) as plasticisers for PVC. JP-A-8-173788 describes the use of fatty acid diesters of sorbitans and isosorbide as an emulsifier. JP-A-59-175408 discloses the cosmetic use of esters of isosorbide with fatty acids having more than 8 carbon atoms. WO 96/33689 describes the use of a plasticising solvent, including dimethyl isosorbide, in film-forming water-in-oil emulsions for use in cosmetics.

We have now found that esters and ethers based on isosorbide and analogous derivatives having a low molecular weight have excellent characteristics as plasticisers and solvents for polymer materials. The derivatives are defined in the appended claims. The central unit of these derivatives is a bicyclo[3.3.0]octane-4,8-diyl system, whose carbon atoms in the 2- and 6- position may be replaced by a heteroatom, such as sulphur and in particular oxygen. The attachment of side chains in the 4- and 8- positions can take the form of an ether, ester, (thio)carbonate, thioether, thioester, amide, (thio)-urethane, urea, phosphate or phosphonate. Esters are preferred, but carbonates, urethanes and especially ethers can also advantageously be used. Phosphates and phosphonates are preferred when in addition to the plasticising effect a flame retardant effect is desired.

The side chains thus attached are preferably medium-length alkyl groups. Medium length means at least 3 carbon atoms, preferably at least 4 carbon atoms, especially (for ethers) at least 6 carbon atoms, up to 10 or even 12 carbon atoms. Where reference is made to alkyl, alkenyl etc., these terms include both linear and branched groups; depending on the particular use, branched groups, e.g. isobutyl, isooctyl, isononyl, 2-ethylhexyl and the corresponding acyl groups, or unbranched groups may be preferred. Examples of suitable alkanoyl groups include butanoyl, hexanoyl, 2-ethylhexanoyl, octanoyl, decanoyl, and unsaturated groups such as benz(o)yl and undecenoyl. Mixtures of alkyl groups are suitable as well, their average chain length is preferably C_4-C_{12} .

It is also possible for oligoester side groups of a diacid and a diol, for example succinic acid, which may or may not be substituted, and isosorbide, or of a hydroxy acid such as hydroxystearic acid or caprolactone, to be attached, with an alkyl or alkanoyl group as the terminal group. The chain length of such oligomeric side groups is 1-20, preferably 1-10, most preferably 1-5 repeating units on either side of the central unit, with molecular weights preferably between about 600 and 2000.

The derivatives according to the invention can be prepared in a manner known per se. Derivatives where X represents an oxygen atom in formula 1 can be prepared starting from isosorbide or the isomeric dianhydrohexitols such as isomannide and isoidide. The dianhydrohexitols can in turn be obtained from the corresponding monosaccharides (glucose, mannose) and di- and polysaccharides (sucrose, maltose, lactose, starch, cellulose, galactomannans and the like). Derivatives where one of the atoms X is a nitrogen or a sulphur atom can be prepared in a similar manner from a suitable amino or thio sugar. Derivatives where both atoms X are sulphur atoms can be prepared starting from 1,4-dithiapentalen-3-one which, by reduction and addition, can be converted into a 4,8-disubstituted 2,6-dithiabicyclooctane. Derivatives where one of the groups Y contains a nitrogen or sulphur atom can be obtained from the corresponding 2-amino- or 2-thio sugars, such as a hydrolysis product of chitin. The side chains can be introduced by esterification (eg. using a reactive carboxylic acid derivative), etherification, isocyanate addition, and the like. Polyester side groups can be introduced in a manner known per se, by polycondensation of the suitable dicarboxylic acids and diols or hydroxy acids.

The derivatives according to the invention can be worked into polymer materials in a manner known per se. In general, the polymer and the plasticiser can be mixed in a ratio of between 100:1 and 1:9. The ratio is preferably from 10:1 to 10:8. In addition to the plasticisers or plasticising solvents according to the invention, other customary components are used such as stabilisers, flow improvers, pigments, antioxidants, UV absorbers, flame retardants, fillers, oligomeric or polymeric resins or varnishes, reactive monomers, activators, starters, desiccants, lubricants, waxes, solvents, biocides and the like. This may involve, for example, organic compounds of calcium, magnesium, zinc or barium, β -diketo compounds, β -ketoesters, β -aminocrotonates, uracil derivatives, dihydropyridine derivatives, sterically hindered phenols, sterically hindered amines, phosphites, polyols and hydrotalcites. It is also possible, advantageously, for the plasticisers according to the invention to be combined with known plasticisers such as dialkyl phthalates, dialkyl adipates, dialkyl azelates and dialkyl sebacates, alkylbenzyl phthalates, trialkyl trimellitates, triaryl phosphates, citric acid esters, alkyl benzoates and polyesters based on adipic acid or azelaic acid, and thus for an optimal combination to be achieved in terms of compatibility, renewability, degradability, and plasticising properties.

The derivatives according to the invention can be used for plasticising and/or solubilising any polymer systems, including thermoplasts (PVC etc.), rubbers, inks, coatings, adhesives, sealants, foams and thermosetting resins.

Examples of the use as plasticiser are given below. The plasticisers are used in a manner known for plasticisers and plasticising solvents, as described e.g. in: I. Skeist (ed.), Handbook of Adhesives, 3rd ed., Van Nostrand Reinhold, NY (1990), and H.F. Mark, Encyclopedia of polymer Science and Engineering, 2nd ed., NY (1985).

Sealants: The derivatives of the invention can be used as a plasticiser in sealants and caulks, for example in a one or two-component polysulphide sealant or in acrylates or polysiloxanes or natural or synthetic rubbers. In addition to the plasticisers, other components may be used in a sealant: epoxidised unsaturated oil (soy, fish, linseed oil), inorganic pigments, desiccants, fillers, and activators such as silica, calcium carbonate, titanium dioxide, lime, zeolites, and adhesion aids, such as organosilicon compounds, and aqueous or organic solvents).

	Example:	general
	50 % liquid polysulphide polymer	30-70%
	4 % epoxidised soya oil	0-10%
	2 % silica	0-5%
5	22 % titanium dioxide	general: inorganic pigment 1-25%
	5 % calcium carbonate	in general:
	2 % lime	fillers, activators etc. 1-15%
	2 % zeolite	
	4 % calcium peroxide (curing agent)	1-10%
10	2 % plasticiser according to the invention	1-20%
	1 % 3-aminopropyltriethoxysilane	0.1-2.5%
	Remainder: toluene	

Printing inks: The derivatives of the invention can be used as plasticiser or solvent in printing ink formulations for application in letterpress (relief), gravure (intaglio), offset lithography (planographic), screen (stencil), electrostatic (reprography) and jet (ink spray) printing. Suitable polymer systems for printing inks include acrylates, rosins, polyamides, polyesters, hydrocarbon resins, alkyd resins, nitrocellulose, cellulose acetates, etc.

Example of flexographic ink formulation (phr = parts per hundred):

	35 phr titanium dioxide	5-45%
20	12 phr polyamide resin	5-25%
	2 phr nitrocellulose	1-10%
	3 phr plasticiser of the invention	2-20%
	5 phr isopropyl acetate	general:
	43 phr isopropanol	20-60% solvent

25 Example of gravure ink formulation:

	10 phr organic pigment or dye	1-15%
	5 phr inorganic extender	1-20%
	5 phr plasticiser of the invention	2-20%
	12 phr acrylic varnish	general:
30	38 phr nitrocellulose varnish	varnish 20-80%
	5 phr wax compound	2-8%
	15 phr ethanol	general:
	10 phr isopropyl acetate	solvent 5-35%

Example of gravure ink formulation:

	12 phr organic pigment	1-15%
	10 phr titanium dioxide	general: inorganic pigment 1-20%
	4 phr plasticiser of the invention	2-20%
5	15 phr maleic varnish	general:
	40 phr nitrocellulose varnish	varnish 20-80%
	4 phr polyethylene wax	2-8%
	10 phr ethanol	general:
	5 phr isopropyl acetate	solvent 5-35%

10 Example of screen-printing ink formulation

	3 phr organic pigment	2-20%
	25 phr talc	5-40%
	36 phr acrylated monomer	general:
	26 phr 1,6-hexanediol diacrylate	reactive monomer 40-75%
15	10 benzophenone	general: initiator 1-15%
	3 phr plasticiser of the invention	1-20%

Adhesives and coatings: The derivatives of the invention can be used as plasticisers in adhesives. Their function is to improve the flexibility, wetting properties and water resistance. The derivatives of the invention are applied as plasticiser for adhesives and sealants of the following classes: natural rubber; synthetic rubber, e.g. butyl, nitrile, neoprene, isoprene, styrene-butadiene rubber and copolymers thereof; carboxylated rubber and carboxyl functional polymers, e.g. and acrylic acid polymers and copolymers; phenolic and amino resin (e.g. urea, melamine); polysulphide resins and adhesives; epoxy resins and adhesives; polyurethanes and isocyanate-functional adhesives; polyvinyl alcohol and polyvinyl acetate and acetal adhesives; acrylate, cyanoacrylate and acrylic acid adhesives and their copolymers; polyester and polyamide; silicone adhesives

The types of adhesives include: reactive one-and two component adhesives; hot-melt adhesives; delayed-tack adhesives; solution adhesives; in particular pressure-sensitive adhesives.

30 Example: two-component vinyl flooring adhesive

Part A:	general:
212.5 phr hydrocarbon resin	resin
37.5 phr mineral spirit	solvent

	12.5 phr plasticiser of the invention	plasticiser
	12.5 phr methanol	solvent
	7.5 phr nonionic surfactant	surfactant
	50 phr clay	filler
5	Part B:	
	100 phr high-solids SBR latex	latex/polymer
	0.5 phr phosphate stabiliser	
	0.5 phr potassium hydroxide	
	0.1 phr defoamer	
10	Water to total solids content 60%	water

Rubbers and thermoplastic elastomers: The plasticisers of the invention can be used as plasticisers in cured or non-cured natural and synthetic rubbers, and in thermoplastic elastomers. These include all common rubbers, in particular acrylonitrile butadiene rubber (NBR), chloroprene rubber (CR), styrene butadiene rubber (SBR), polybutadiene (BR),
 15 1 to 100 parts, in particular, 5-30 parts. The derivatives are used for decreasing the glass temperature of the rubber and for increasing tensile strength and strength on rupture.

Thermosets: The plasticisers of the invention can be used as plasticisers in thermosetting resins (e.g. polyesters, amino resins) in order to improve the toughness and elongation.

Foams: The derivatives of the invention can also be used as plasticisers in polymeric
 20 foams, in particular polyurethane, polyether and latex (natural or synthetic rubber) foams. They can be used to increase compression strength of the foam, refine the cell structure, improve its insulation resistance and increase its tenacity or other chemical, physical or mechanical properties.

Example 1: Preparation of isosorbide dioctanoate (ISDO)

25 In a 500 ml flask, provided with a Dean-Stark apparatus, a solution of 25 g (0.171 mol) of isosorbide, 54 g of n-octanoic acid (0.375 mol) and 1.0 g of p-toluenesulphonic acid was boiled for 2 hours in 130 ml of xylene. After cooling, 250 ml of diethyl ether were added and the solution was washed with 1225 ml of 0.1 M NaOH and then with water until the water phase remained neutral. The organic phase was
 30 evaporated, whereupon 65 g (95%) of product were isolated.

¹³C-NMR (δ , ppm, CDCl₃): 13.9 (-CH₃), 22.5 + 24.7 + 28.8 + 31.1 + 31.5 + 33.8 + 70.2 + 73.3 (-CH₂-), 73.6 + 77.7 (CHO), 80.6 + 85.8 (CHOR), 172.7 (OCO).

Examples 2-5: *Preparation of isosorbide dibutanoate (ISDB), isosorbide dihexanoate (ISDH) and isosorbide bis(2-ethylhexanoate) (ISDEH) and isomannide dioctanoate (IMDO)*

Using the method of Example 1, starting from the appropriate alkanolic acids and dianhydrohexitols, the title compounds were obtained:

ISB: Yield 87%, $^{13}\text{C-NMR}$: 13.3(-CH₃), 18.1 + 35.5 + 70.2 + 73.0 (-CH₂-), 73.6 + 77.6 (CHO), 80.5 + 85.7 (CHOR), 172.1 (OCO).

ISH: Yield 86%, $^{13}\text{C-NMR}$: 13.5 (-CH₃), 22.0 + 24.2 + 31.0 + 33.7 + 70.1 + 73.0 (-CH₂-), 73.6 + 77.6 (CHO), 80.5 + 85.7 (CHOR), 172,1 (OCO).

ISEH: Yield 89%, $^{13}\text{C-NMR}$: 11.5 + 13.6 (-CH₃), 22.4 + 25.1 + 29.1 + 31.4 + 70.4 + 73.0 (-CH₂-), 46.7 (-CH-), 73.4 + 77.5 (CHO), 80.5 + 85.9 (CHOR), 174.8 (OCO).

IMO: Yield 91%, $^{13}\text{C-NMR}$: 14.0 (-CH₃), 22.5 + 24.8 + 28.8 + 28.9 + 31.6 + 33.9 + 70.4 (-CH₂-), 73.5 (CHO), 80.3 (CHOR), 173.2 (OCO).

Example 6: *Preparation of plasticised PVC*

A two-roll calender was used to make a film from the PVC compound at 142°C, followed by determination of the glass transition temperature (T_g) by means of DMTA (Dynamic-mechanical thermal analysis).

100 parts of PVC

50 parts of plasticiser according to Example 1

1.0 part of hydrotalcite

0.3 part of zinc laurate

0.3 part of calcium stearate

The T_g was 0°C; that of non-plasticised PVC that otherwise had the same composition was 80°C, and that of PVC plasticised with the same amount of dioctyl phthalate (DOP) was -13°C. The modulus of elasticity at -20°C was 10⁹ Pa, at 40°C 10⁷ Pa and at 100°C 3.10⁶ Pa, values comparable to those of PVC plasticised with DOP.

Example 7: *Preparation of plasticised PVC*

A film obtained according to Example 6 on the basis of the following compounds:

100 parts of PVC

30-50 parts of plasticiser according to Examples 1-5

1.0 part of hydrotalcite

0.3 part of zinc laurate

0.3 part of calcium stearate

was made into small rods whose glass transition temperature was determined by means of DMTA and whose modulus of elasticity and breaking behaviour were determined on a tensile testing machine. The results - as yet not optimised - were compared with those of dibutyl phthalate (DBP) and dioctyl phthalate (DOP) and are shown in Table 1:

Table 1

plasticiser	amount (phr)	Tg(E"max) (°C)	E(Mpa) at T=17,6°C	breaking tension (MPa)	breaking elongation (%)
ISDB	30	25	670	28	165
ISDH	30	21	310	26	142
ISDEH	30	26	n.d.	n.d.	n.d.
ISDO	30	18	600	24	135
IDMO	30	27	390	29	207
DBP	30	14	110	18	159
DOP	30	23	240	17	130
ISDB	50	2	12	18	193
ISDH	50	-7	14	10	149
ISDEH	50	-5	11	11	234
ISDO	50	-9	15	18	228
IMDO	50	-20	26	15	152
DBP	50	-16	7	13	237
DOP	50	-13	21	14	205
none	-	85	n.d.	n.d.	n.d.

n.d. = not determined

Example 8: Preparation of plasticised NBR

To a commercial acrylonitrile-butadiene rubber (NBR) compound containing carbon black, stabilisers, crosslinkers and further conventional additives was added 20

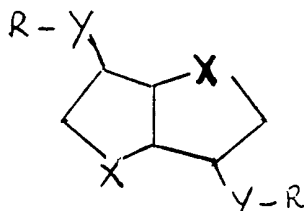
phr of plasticiser according to the invention (isosorbide bis(2-ethylhexanoate ISDEH) 40°C. After extensive mixing, the mixture was vulcanised at 170°C in a die having dimensions of 180 x 120 x 1 mm. The vulcanisation time was 7 minutes. The vulcanised material was conditioned at 50% relative humidity for two days and punched in accordance with ISO 37. The mechanical properties were determined using a tensile strength tester (Zwick Z010). The E-modulus was determined at 1 mm/min, the other properties at 500 mm/min, in accordance with ISO 1184. Table 2 shows the results, with standard deviations in parentheses.

Table 2

<i>Property</i>	<i>without plasticiser</i>	<i>20 phr DOP</i>	<i>20 phr ISDEH</i>	<i>unit</i>
E-modulus	7 (1)	2 (0)	3 (0)	N/mm ²
Stress at 100% strain	4.2 (0.1)	1.5 (0.1)	1.6 (0.2)	N/mm ²
Stress at 300% strain	15.1 (0.5)	7.1 (0.3)	7.7 (0.4)	N/mm ²
Stress at break	15.4 (1.8)	13.0 (1.4)	13.0 (0.8)	N/mm ²
Strain at break	354 (49)	499 (38)	469 (26)	%

CLAIMS

1. Use of a bicyclo[3.3.0]octane derivative having formula 1:



where

- X represents $-\text{CH}_2-$, $-\text{O}-$, $-\text{S}-$, $-\text{NR}^1-$ or $-\text{POOR}^2-$,
- Y represents $-\text{O}-$, $-\text{O}-\text{CO}-$, $-\text{O}-\text{CO}-\text{O}-$, $-\text{O}-\text{CS}-\text{S}-$, $-\text{S}-$, $-\text{S}-\text{CO}-$, $-\text{O}-\text{CO}-\text{NH}-$, $-\text{S}-\text{CO}-\text{NH}-$, $-\text{NH}-\text{CO}-$, $-\text{NH}-\text{CO}-\text{NH}-$, $-\text{O}-\text{POOR}^2-$ or $-\text{O}-\text{POOR}^2-\text{O}-$,
- R represents R^3 or CH_3CO or, if $\text{Y} = \text{O}$, represents $-\text{SiR}^2_3$, $-\text{SiR}^2_2-\text{OR}^2$, $-\{\text{CO}-\text{D}-\text{CO}-\text{O}-\text{E}-\text{O}\}_n-(\text{CO})_p\text{R}^3$, or $-\{\text{CO}-\text{E}-\text{O}\}_n-(\text{CO})_p\text{R}^3$,
- R^1 represents C_1-C_4 alkyl or C_1-C_4 alkanoyl,
- R^2 represents C_1-C_4 alkyl,
- R^3 represents C_2-C_{18} alkyl, C_2-C_{18} alkenyl, C_3-C_{10} cycloalkyl or C_7-C_{18} aryl-alkyl, where one or more of the methylene groups can be replaced by an oxygen atom,
- D represents C_2-C_{10} alkylene or C_2-C_{10} alkenylene,
- E represents C_2-C_{10} alkylene, cycloalkylene or bicycloalkylene, where one or more of the methylene groups can be replaced by an oxygen atom,
- n is a number from 1 to 20, in particular 1 to 10, and
- p is 0 or 1,

as a plasticiser or solvent for a polymer.

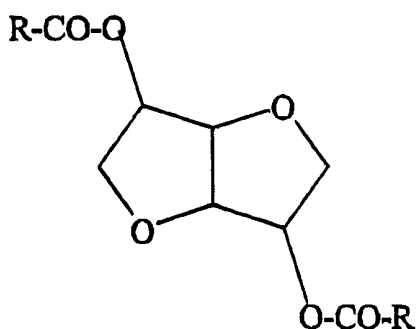
2. Use according to Claim 1, wherein X represents $-\text{O}-$.
3. Use according to Claim 1 or 2, wherein Y represents $-\text{O}-\text{CO}-$.
4. Use according to any one of Claims 1-3, wherein R represents C_3-C_{10} alkyl or C_3-C_{10} alkenyl.

5. Use according to any one of Claims 1-4, wherein 1-900 parts by weight, in particular 10-80 parts by weight, of the bicyclooctane derivative are used per 100 parts of synthetic polymer to be plasticised.
6. Use according to any one of Claims 1-5, wherein a dialkyl phthalate, adipate or sebacate, or trialkyl trimellitate, phosphate, citrate and/or a polyester are also used.
7. Use according to any one of Claims 1-6, wherein said polymer is PVC.
8. Use according to any one of Claims 1-6, wherein said polymer is a rubber.
9. Use according to any one of Claims 1-6, for plasticising coatings or adhesives.
10. Use according to any one of Claims 1-6, for plasticising an ink.
11. Use according to any one of Claims 1-6, for plasticising a sealant.
12. Use according to any one of Claims 1-6, for plasticising a polymer foam.
13. Bicyclo[3.3.0]octane derivative of formula 1, in which:
X represents -CH₂-, -O-, -S-, -NR¹- or -POOR²-,
Y represents -O-CO- or -O-CO-O-, -O-POOR²- or -O-POOR²-O-, and
R represents C₃-C₁₀ alkyl or C₃-C₁₀ alkenyl.
14. Bicyclo[3.3.0]octane derivative according to Claim 7, which is a dianhydrohexitol diester having formula 1 in which:
X represents -O-, Y represents -O-CO- and R represents C₄-C₈ alkyl.
15. Mixture of plasticisers which comprises 20-90 wt% of a bicyclo[3.3.0]octane derivative as defined in any one of Claims 1-6 and 10-80 wt% of a dialkyl phthalate, adipate, sebacate, or a trialkyl trimellitate, phosphate, citrate, an alkyl benzoate and/or a polyester.

AMENDED CLAIMS

[received by the International Bureau on 23 August 1999; (23.08.99);
original claims 1-15 replaced by new claims 1-11 (2 pages)]

1. Use of a bicyclo[3.3.0]octane derivative having formula 1:



- wherein R represents C₅-C₁₁ alkyl or alkenyl,
as a plasticiser or solvent for a polymer.
2. Use according to Claim 1, wherein 1-900 parts by weight, in particular 10-80 parts by weight, of the bicyclooctane derivative are used per 100 parts of synthetic polymer to be plasticised.
3. Use according to Claim 1 or 2, wherein a dialkyl phthalate, adipate or sebacate, or trialkyl trimellitate, phosphate, citrate and/or a polyester are also used.
4. Use according to any one of Claims 1-3, wherein said polymer is PVC.
5. Use according to any one of Claims 1-3, wherein said polymer is a rubber.
6. Use according to any one of Claims 1-3, for plasticising coatings or adhesives.
7. Use according to any one of Claims 1-3, for plasticising an ink.
8. Use according to any one of Claims 1-3, for plasticising a sealant.
9. Use according to any one of Claims 1-3, for plasticising a polymer foam.

10. Bicyclo[3.3.0]octane derivative of formula 1, in which:
R represents C₅-C₁₁ alkyl or C₅-C₁₁ alkenyl.

11. Mixture of plasticisers which comprises 20-90 wt% of a bicyclo[3.3.0]octane derivative as defined in Claim 10 and 10-80 wt% of a dialkyl phthalate, adipate, sebacate, or a trialkyl trimellitate, phosphate, citrate, an alkyl benzoate and/or a polyester.

INTERNATIONAL SEARCH REPORT

International Application No

PCT/NL 99/00115

A. CLASSIFICATION OF SUBJECT MATTER
 IPC 6 C08K5/15 C08K5/3417 C08K5/34 C08K5/45 C08K5/00
 C08K5/5377 C07D493/04 //(C07D493/04, 307:00, 307:00)

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED
 Minimum documentation searched (classification system followed by classification symbols)
 IPC 6 C08K

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category °	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	WO 96 33689 A (PROCTER & GAMBLE) 31 October 1996 zie conclusies; blz. 10, regel 6 ---	1, 2, 4, 13
A	DE 33 47 075 A (BAYER AG) 4 July 1985 zie conclusies; blz. 6, formule II ---	1-15
A	BRAUN D ET AL: "POLYESTERS WITH 1.4:3.6-DIANHYDROSORBITOL AS POLYMERIC PLASTICIZERS FOR PVC" ANGEWANDTE MAKROMOLEKULARE CHEMIE. APPLIED MACROMOLECULAR CHEMISTRY AND PHYSICS, vol. 199, 1 August 1992, pages 191-205, XP000293988 cited in the application zie blz. 191, samenvatting ---	1-15
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Further documents are listed in the continuation of box C. Patent family members are listed in annex.

° Special categories of cited documents :

"A" document defining the general state of the art which is not considered to be of particular relevance	"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention
"E" earlier document but published on or after the international filing date	"X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone
"L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)	"Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.
"O" document referring to an oral disclosure, use, exhibition or other means	"&" document member of the same patent family
"P" document published prior to the international filing date but later than the priority date claimed	

Date of the actual completion of the international search 10 June 1999	Date of mailing of the international search report 22/06/1999
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Name and mailing address of the ISA European Patent Office, P.B. 5818 Patentlaan 2 NL - 2280 HV Rijswijk Tel. (+31-70) 340-2040, Tx. 31 651 epo nl, Fax: (+31-70) 340-3016	Authorized officer Engel, S
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INTERNATIONAL SEARCH REPORT

International Application No

PCT/NL 99/00115

C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	PATENT ABSTRACTS OF JAPAN vol. 009, no. 028 (C-264), 6 February 1985 & JP 59 175408 A (NIHON SAAFUAKUTANTO KOGYO KK), 4 October 1984 cited in the application see abstract ---	1-15
X	ABENHAIM D ET AL: "SELECTIVE ALKYLATIONS OF 1,4:3,6-DIANHYDRO-D-GLUCITOL (ISOSORBIDE)" CARBOHYDRATE RESEARCH, vol. 261, no. 2, 17 August 1994, pages 255-266, XP002020671 zie blz. 257, schema 1, verbindingen 4c, 4d, 4e en 4f ---	13,14
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