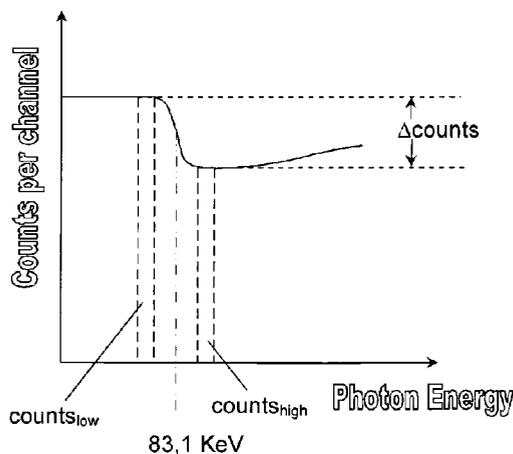




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(54) **Titre : PROCÉDE POUR DETERMINER LA CONCENTRATION D'UN ELEMENT DANS UN MATERIAU**
(54) **Title: METHOD FOR DETERMINING THE CONCENTRATION OF AN ELEMENT IN A MATERIAL**



(57) **Abrégé/Abstract:**

A method for determining the concentration of an element in a material, such as heavy elements like mercury, lead, uranium, plutonium, tungsten or gold in a light matrix material like coal or refined oils, that can be performed as an online measurement and that makes it possible to precisely determine even very low concentrations, especially below 0,1%, is provided. According to the Invention the characteristic absorption is measured using a technique such as X-ray absorption spectroscopy (e.g. XAS, XANES, EXAFS). This measurement takes advantage of the fact that the absorption of an element significantly rises when the energy of the irradiating beam goes beyond an characteristic absorption edge of the element. So, according to the invention, the difference between the intensity of the transmitted X-ray beam in an energy interval above the absorption edge and in an energy interval below the absorption edge is determined.

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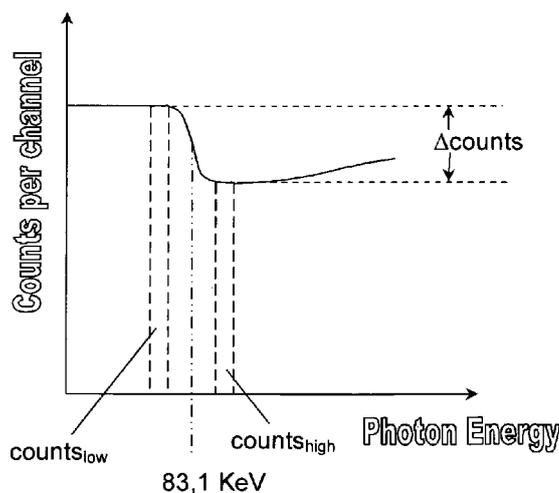


Fig. 2

(57) **Abstract:** A method for determining the concentration of an element in a material, such as heavy elements like mercury, lead, uranium, plutonium, tungsten or gold in a light matrix material like coal or refined oils, that can be performed as an online measurement and that makes it possible to precisely determine even very low concentrations, especially below 0,1%, is provided. According to the Invention the characteristic absorption is measured using a technique such as X-ray absorption spectroscopy (e.g. XAS, XANES, EXAFS). This measurement takes advantage of the fact that the absorption of an element significantly rises when the energy of the irradiating beam goes beyond a characteristic absorption edge of the element. So, according to the invention, the difference between the intensity of the transmitted X-ray beam in an energy interval above the absorption edge and in an energy interval below the absorption edge is determined.

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Method for determining the concentration of an element in a material

5 Description

Technical field of the invention and prior art

The invention relates to a method for determining the concentration of at least one
10 element in a material.

In many industrial applications it is necessary to determine the concentration of one or more elements in a bulk material. Often it is necessary or at least desired to determine said concentration in an online measurement. For example it is a
15 well known problem that the total ash content in coal needs to be measured. Here, the total ash content is typically in the range of 1 to 50 percent. There are established techniques how to perform a measurement in order to determine the total ash content. The most common method is the so-called dual energy method. The sample is irradiated with two X-ray or gamma ray beams of different energy.
20 The transmission is measured and the area weight and the average atomic number are determined. If the composition of the ash is known, one can calculate the ash content of the coal. In a more sophisticated method the surface of the coal is irradiated with an X-ray beam and the fluorescence peaks of the metals constituting the ash are measured (so called XRF measurement). With this
25 method not only the total ash content but also the composition of the ash can be determined. Because of the higher energies of the fluorescence peaks of heavier elements the measurement of heavier elements is easier than the measurement of light elements.

30 Especially because of strict environment regulations it is not only necessary to determine the total ash content of coal but also the precise content of some poisonous elements, especially mercury. Because mercury is a heavy element

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with a characteristic K_{α} -line of high energy, the measurement of mercury would basically be easy with the above mentioned XRF technique. But it turned out that because of the very low mercury concentration of real-life samples, the results of online measurements using XRF technique are not satisfactory.

5

Summary of the invention

It is an object of this invention to provide a method for determining the concentration of an element in a material that can be performed as an online
10 measurement and that makes it possible to precisely determine even very low concentrations, especially below 0,1%.

According to the invention, not the characteristic emission of the element of interest is measured, but the characteristic absorption. This measurement takes
15 advantage of the fact that the absorption of an element significantly rises when the energy of the irradiating beam goes beyond an characteristic absorption edge of the element. So, according to the invention, the difference between the intensity of the transmitted X-ray beam in an energy interval above the absorption edge and in an energy interval below the absorption edge is determined. This difference
20 is in many applications directly linear to the concentration of the searched element over a large scale. In many applications, especially if the concentration of the searched element is low and if the searched element is a high Z element and the bulk material is a low Z material, the measurement of the absorption resulting from excitation has many advantages in relation to the measurement of the
25 characteristic emission. One of the most important aspects is that the relevant signals increase to a large extend in a linear manner with the thickness of the irradiated sample. So, one can enhance the measurement by using thicker layers of bulk material as long as the absorption of the material in the used energy range is low enough such that the bulk material is penetrated by the radiation such that
30 the radiation can be detected by the detector. This is not at all the case when measuring the fluorescence, due to geometrical reasons.

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Usually it will be necessary to keep the thickness of the sample constant or to measure the thickness or the area weight of the sample in order to take the thickness into account, when determining the concentration of the searched element.

5

The invention will now be explained in detail in view of preferred embodiments with reference to the accompanying figures. The figures show:

Brief description of the drawings

10

Figure 1 a diagrammatic representation of a first embodiment of the invention,

Figure 2 a typical spectrum used for determining the concentration of mercury in coal with the photon energy on the X-axis and the counts on the Y-axis,

15 Figure 3 a diagrammatic representation of a second embodiment of the invention,

Figure 4 a typical spectrum as in Figure 2 but with the absorption-coefficient on the Y-axis,

Figure 5 a variation of the second embodiment, and

20 Figure 6 a further variation of the second embodiment.

Detailed description of preferred embodiments

The invention will now be explained in detail. In the examples described the
25 content of mercury in coal is determined. This is an important application of the invention but it needs to be mentioned that a lot other applications are possible. Especially if traces of a high-Z element, such as mercury, lead, uranium, plutonium, tungsten, gold etc. are to be measured in a bulk material with relatively low-Z average, such as coke, coal, petroleum, wood, ore, e.g. iron ore, and other
30 low-Z minerals, etc. the method is very useful.

Figure 1 shows all essential parts of a first embodiment of an apparatus for carrying out the method according to the invention. This apparatus comprises the

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following elements: An X-ray source 10, a first collimator 14 downstream of the X-ray source 10, a second collimator 42 downstream of the first collimator 14, an energy dispersive detector 40, a multi-channel analyzer 44 connected to the energy dispersive detector and a computing unit 50 connected to the multi-channel analyzer. The X-ray source, the first collimator, the second collimator and the energy dispersive detector are collinear in respect to each other. A filter 12 might be provided downstream of the X-ray source. The atomic number of the filter material should be higher by two or three elements than the searched element. For Mercury a lead filter can be used. A conveyor belt 30 for conveying the bulk material B - here coal - is provided between the two collimators. The conveyor, especially a conveyor belt, preferably extends perpendicular to the X-ray beam propagating from the X-ray source to the X-ray detector. A means for measuring the area weight and/or the thickness of the bulk material on the conveyor is provided. This means can be in form of a CS^{137} transmission line consisting of a CS^{137} source and a detector (as will be described later). If the density of the material is constant, the layer-thickness t can be measured. A means for measuring the layer-thickness can be a laser scanner 20, but other measuring means are also possible. The means for measuring the thickness of the bulk material is also connected to the computing unit 50.

20

Since Compton- and/or Rayleigh scattered radiation might influence the measurement, especially due to an imperfect collimating, an additional detector 70 can be provided. In the embodiment shown this detector is arranged in a backscatter geometry, but it would also be possible to arrange the same in a transmission geometry. The signals of the additional detector 70 are transmitted to the computing unit 50, such that the measured scattered radiation can be used for correction purposes.

25

The X-ray source generates a continuum around the K-absorption edge of mercury at 83,1 KeV, for example between 70 and 90 KeV. All other energies can be filtered, since they are of no use for this measurement.

30

- 5 -

A typical spectrum that is "seen" by the detector, analyzed by the multi-channel analyzer and used by the computing unit to compute the desired information is shown in Figure 2. One can clearly see the absorption edge at 83,1 keV . At this energy the transmission drops in a sharp step due to the change of the absorption at the absorption edge. The "height" of this step (later referred to as Δcounts) is directly proportional to the thickness of the material and to the concentration of the element of interest (here mercury), as long as a linear approximation is applicable. Such a linear approximation is in this example applicable for a thickness of the bulk material up to at least 5 Inches and for concentrations of mercury up to at least 0,1%. It goes without saying that the method described here is also applicable in the non-linear range; the non-linear formula will be given later. But in order to explain the invention by means of simple mathematics, the linear approximation is used in the following:

In a first energy interval just below the absorption edge the photons are counted in a defined time interval. The energy interval depends on the energy resolution of the detector system and is not rather critical, for example 5 keV. The time interval should be long enough to reduce the statistical error sufficiently, for example 30s. The number of photons counted is referred to as $\text{counts}_{\text{low}}$. Respectively photons are counted just above the absorption edge in a second energy interval with the same width as the first energy interval. The time interval is of course identical to the time interval of the first energy interval. The number of photons counted is referred to as $\text{counts}_{\text{high}}$. As long as the element searched (here: mercury) is not present in the bulk material, $\text{counts}_{\text{low}}$ and $\text{counts}_{\text{high}}$ are basically identical. This changes when the element searched is present. In this case $\text{counts}_{\text{high}}$ is less than $\text{counts}_{\text{low}}$. The difference is referred to as Δcounts .

In linear approximation, the following mathematics apply:

$$\Delta\text{counts} = \text{counts}_{\text{low}} - \text{counts}_{\text{high}}$$

and

$$\text{counts}_{\text{high}} = \text{counts}_{\text{low}} * c * d * \text{const}$$

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In the embodiment shown in Figure 3 two gamma-ray sources are used, namely an Am-241 source 11 at 59,5 keV for the energy below the absorption edge and a Cd-109 source 11' at 88 keV above the absorption edge. The first detector 41 is assigned to the Am-source 11 and the second detector 41' is assigned to the Cd-source 11'. Collimators 14, 14'; 42, 42' are used as in the first embodiment. An alternative for the low energy source could be Ba133 (80keV). Alternatives for the high energy source could be Co57, EU155, EU152, or EU154.

As can be seen from Figure 3, filters 12, 12a, 12b, and 12c can be used, both, on the source-side and on the detector side. On the source-side the filters 12, 12a improve the monochromatic nature of the incident beam, on the detector side the filters 12b, 12c help to eliminate scattered radiation (in this case together with the collimators 42, 42').

In this second embodiment not the thickness d of the bulk material is measured, but the area weight of the bulk material. This leads to more precise results if the density of the material is not constant. This measurement is performed as is known in the art by a high energy transmission measurement using a Cs-137 source 60 and a respective detector 62. This measurement of the the area density could also be used in the first embodiment instead of the measurement of the thickness of the bulk material via a laser scanner.

The advantage in respect to the first embodiment is that the detectors do not need to be energy dispersive and that no multi-channel analyzer is necessary.

The mathematics are as follows:

First, the well-known general formula applies:

30

$$I = I_0 * e^{-\mu * \rho * d} \quad (1)$$

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with

I_0 = incident energy

I = measured energy

5 μ = absorption coefficient

so,

$$-\mu_{high} * \rho * d = \ln\left(\frac{I}{I_0}\right)_{high}$$

and

$$-\mu_{low} * \rho * d = \ln\left(\frac{I}{I_0}\right)_{low}$$

10 and for the measurement with the Cs-source

as can be seen from Figure 4

$$\mu_{HG} = \mu_{high} - \mu_{low}$$

15

this leads to

$$\mu_{HG} = \mu_{Cs} \frac{\ln\left(\frac{I}{I_0}\right)_{high} - \ln\left(\frac{I}{I_0}\right)_{low}}{\ln\left(\frac{I}{I_0}\right)_{Cs}}$$

20 Figure 5 shows a variation of what is shown in Figure 4. Here, a common detector 45 and a discriminator 65 are used. The Am-241 source 11 and the Cd-109 source 11' point towards this common detector 45 and the energies are separated by the discriminator 65.

25 Figure 6 shows a variation of what is shown in Figure 5. Here, the beam of the Cd-109 source 11' extends through the Am-241 source 11, so that both beams

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pass through the same area of the bulk material, which is to be preferred if the surface of the bulk material has not a uniform shape.

5 In order to further refine the measurement a measurement of the moisture of the bulk material can be measured, especially via a microwave measurement. Such a measurement can be performed in all described embodiments.

Finally it should be noted that an additional detector 70 for measuring scattered radiation can be provided in all shown embodiments.

10

As has already been pointed out, the method according the invention is especially useful to determine the content of heavy trace elements in coal, coke, petroleum and petroleum derivates in an online measuring process. But it needs to be pointed out that the method can also be used for other problems such as
15 detecting uranium or plutonium in cargo or to detect tungsten or lead in gold.

Claims

1. A method for determining the concentration of an element in a material comprising the following steps:
 - 5 - transporting the material on a conveying means,
 - irradiating the material with an X-ray beam having a continuum in the area of an absorption edge of the element to be measured,
 - measuring the intensity of the transmitted X-ray beam with an energy dispersive detector (40),
 - 10 - determining the intensity of the transmitted X-ray beam in an energy interval above the absorption edge and in an energy interval below the absorption edge, and
 - computing the concentration of the element on the basis of said intensities, via a calculating means such that the concentration of the element is determined automatically in an online process.
2. The method according to claim 1, wherein the atomic number of the element is higher than the mean atomic number of the material.
- 20 3. The method according to claim 2, wherein the material is selected from the group consisting of coal, a product refined from coal, oil, and a product refined from oil.
4. The method according to any one of claims 1-3, wherein the element is selected from the group consisting of mercury, lead, uranium, plutonium, tungsten, and gold.
- 25 5. The method according to any one of claims 1-4, wherein the concentration of the element in the material is 0.1 % or less.
- 30 6. The method according to any one of claims 1-5, comprising the additional step of measuring the density or the thickness of the irradiated material.

7. The method according to any one of claims 1-6, comprising the additional step of measuring scattered radiation.
8. An apparatus for determining the concentration of an element in a material, said apparatus comprising:
- an X-ray source (10), generating an X-ray beam having a continuum in the area of the element to be measured;
 - a conveying means for transporting the material,
 - an energy dispersive X-ray detector (40) being arranged in a collinear manner to the X-ray source (10);
 - a discriminator or a multi-channel analyzer (44) being connected to the X-ray detector (40); and
 - a computing unit (50) being connected to the discriminator or multi-channel analyzer for automatically determining the concentration of the element in an online process.

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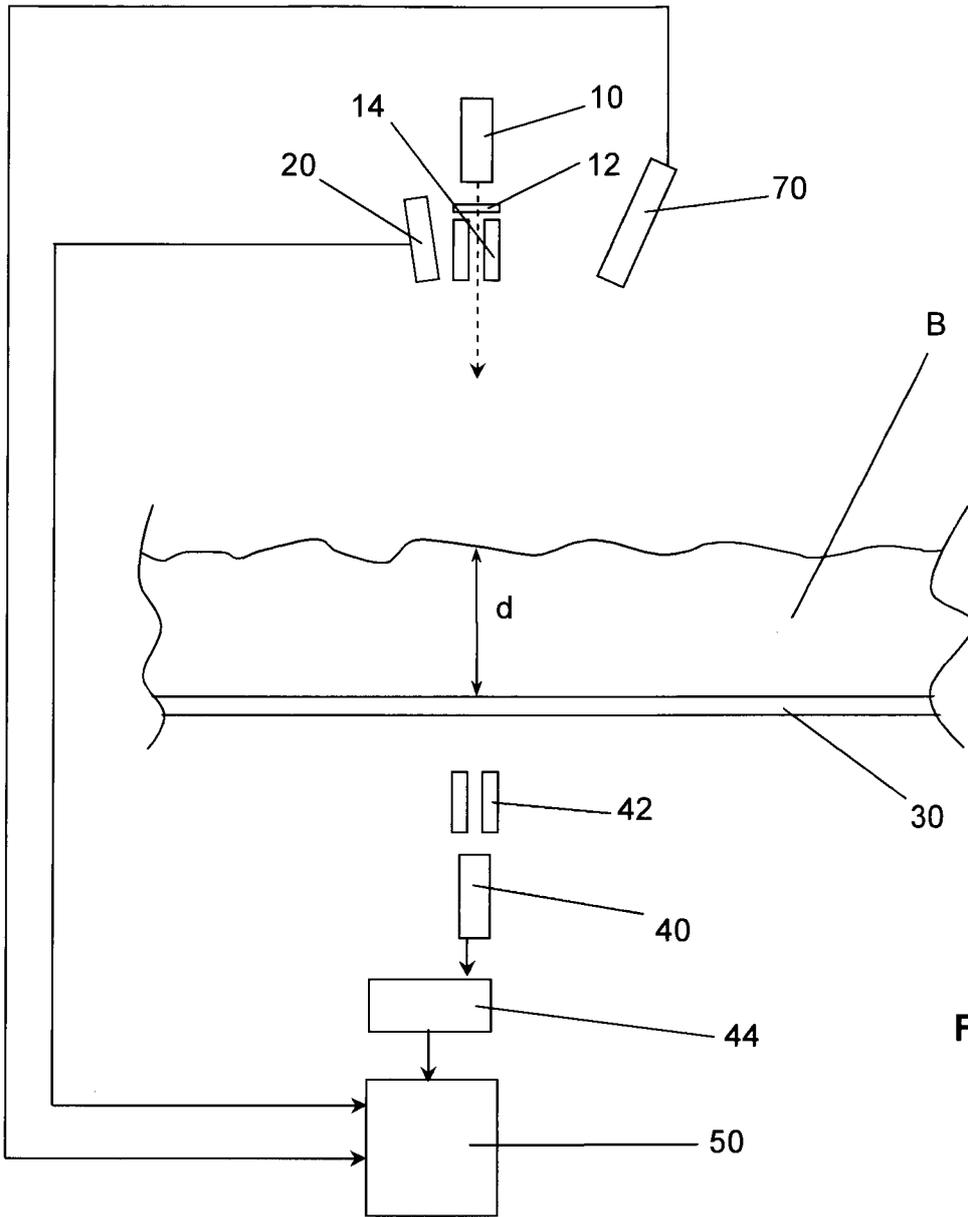


Fig. 1

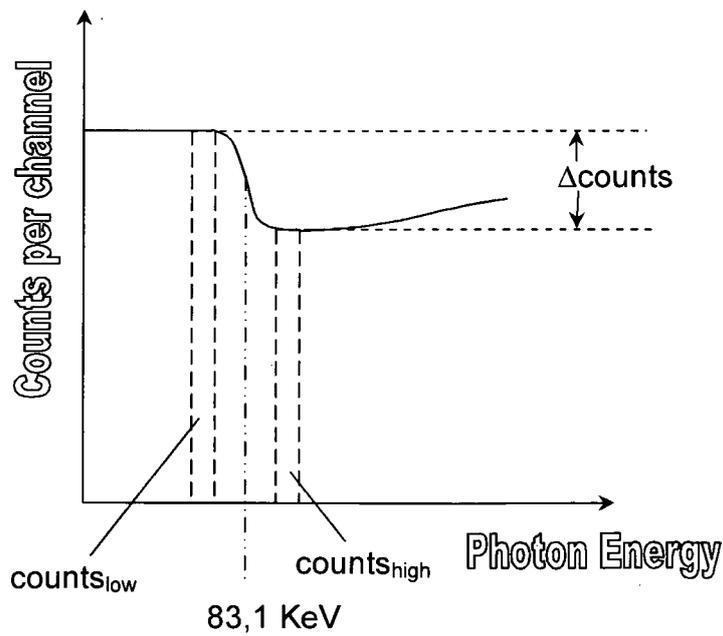


Fig. 2

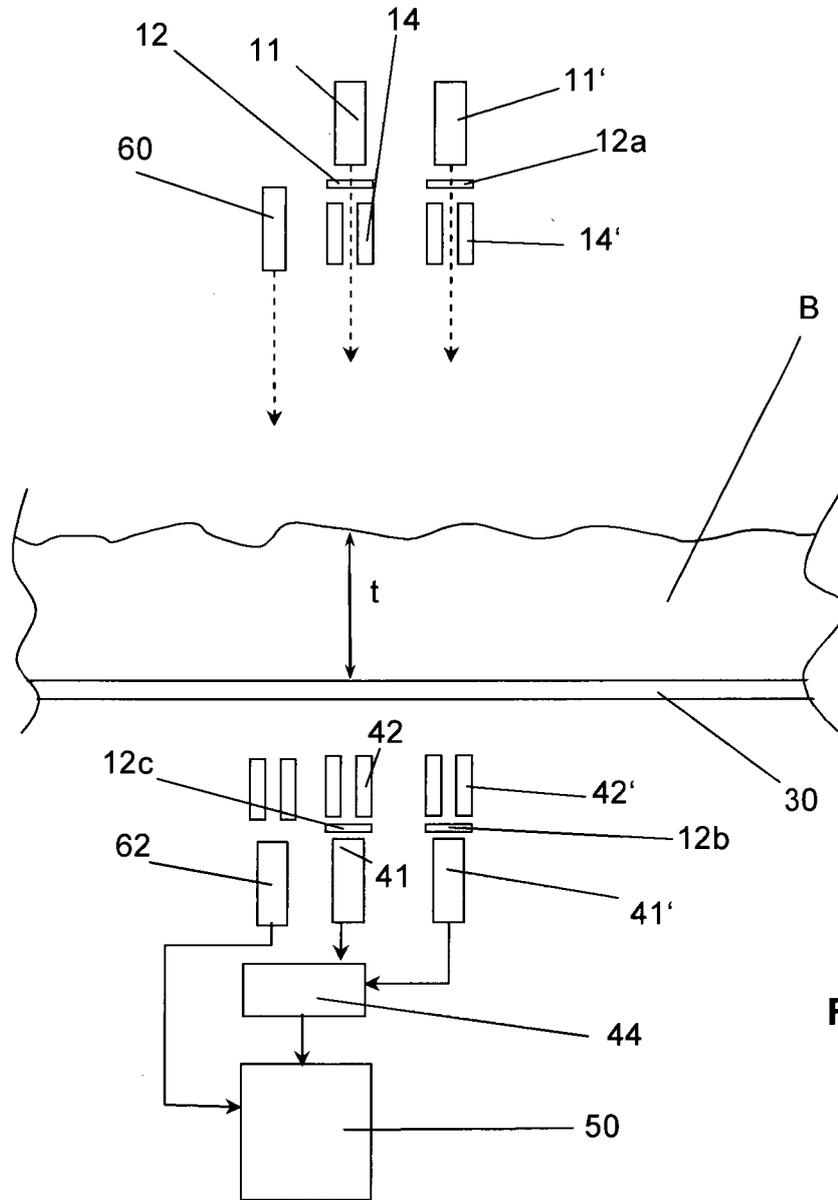


Fig. 3

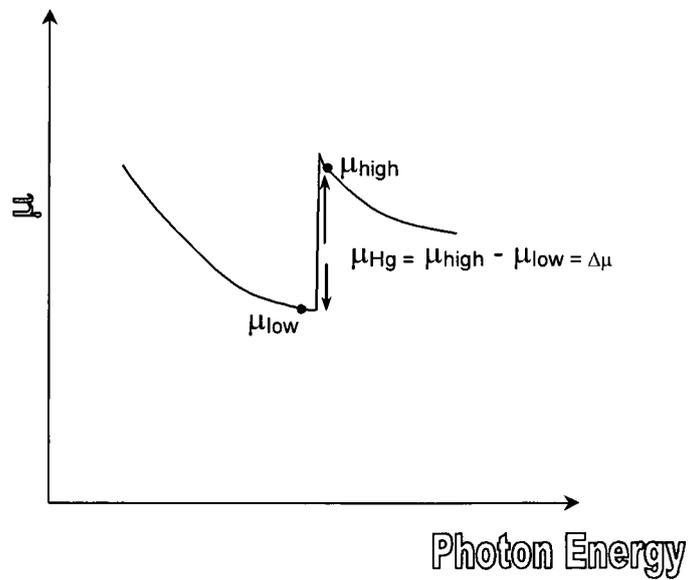


Fig. 4

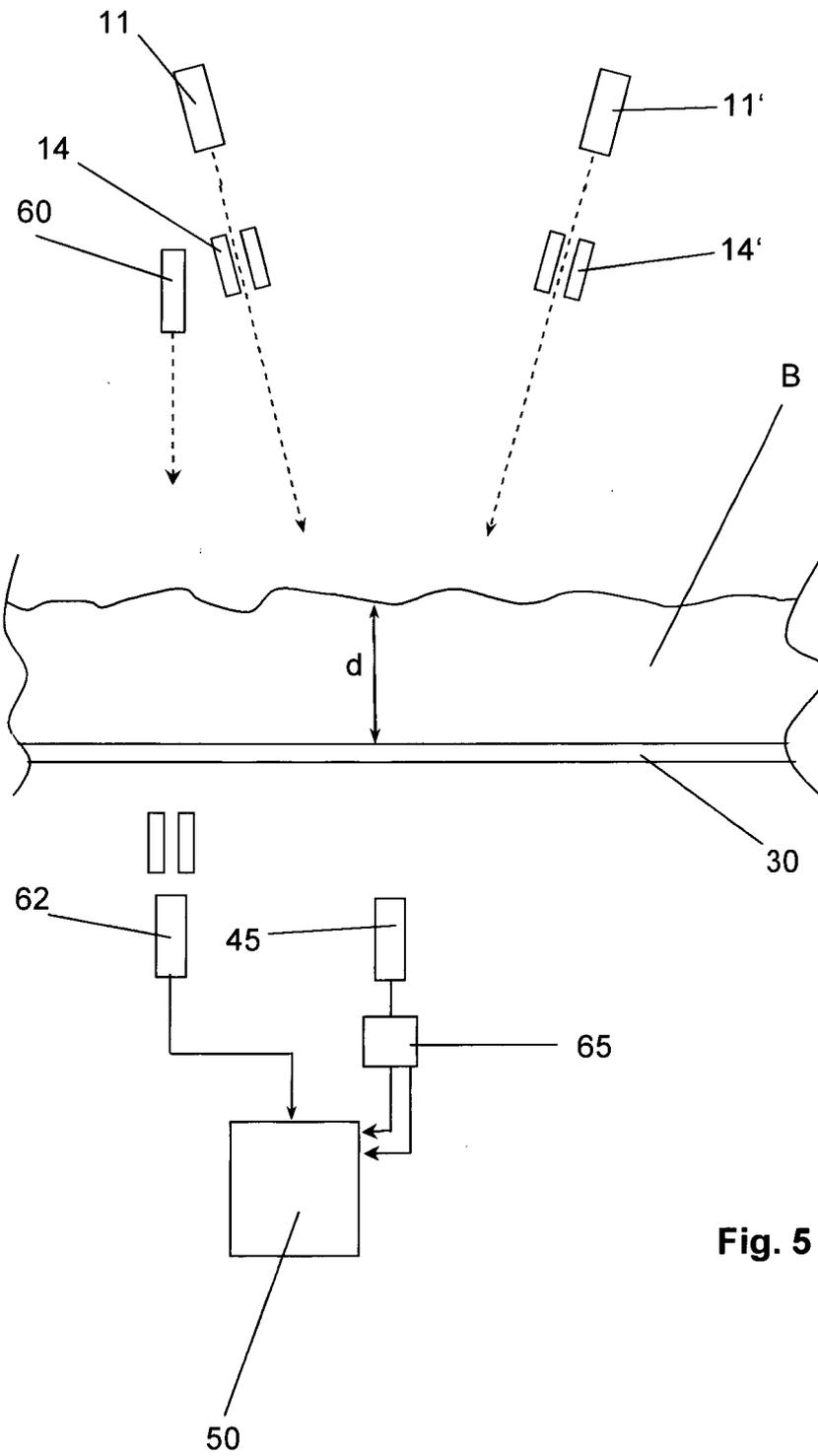


Fig. 5

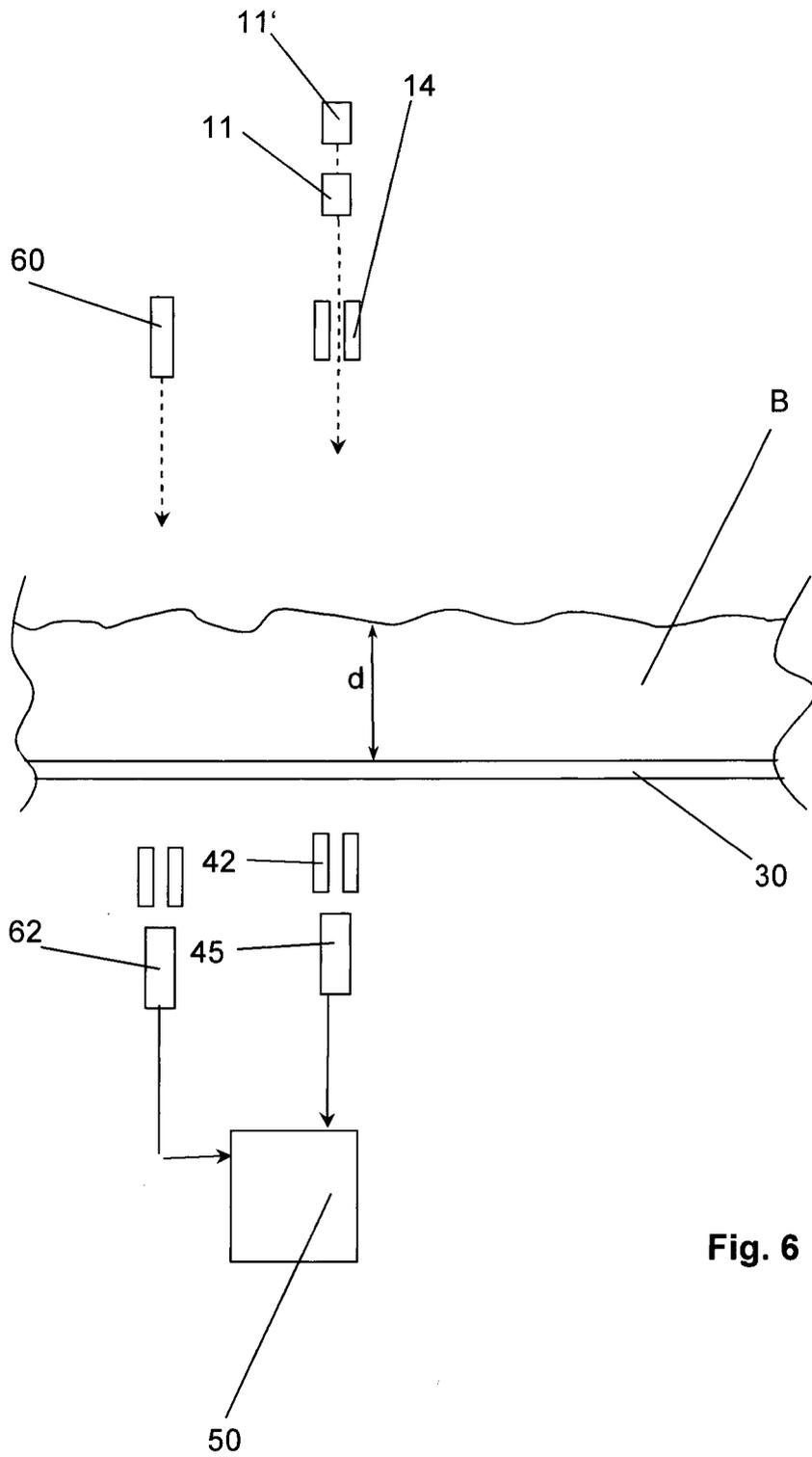


Fig. 6

