



(43) International Publication Date  
30 January 2014 (30.01.2014)

(10) International Publication Number  
**WO 2014/016852 A1**

- (51) **International Patent Classification:**  
*C07C 213/00* (2006.01)
- (21) **International Application Number:**  
PCT/IN2013/000464
- (22) **International Filing Date:**  
25 July 2013 (25.07.2013)
- (25) **Filing Language:** English
- (26) **Publication Language:** English
- (30) **Priority Data:**  
2300/DEL/2012 25 July 2012 (25.07.2012) IN
- (71) **Applicant: COUNCIL OF SCIENTIFIC & INDUSTRIAL RESEARCH** [IN/IN]; Anusandhan Bhawan, Rafi Marg, New Delhi 110001 (IN).
- (72) **Inventors: CHAVAN, Subhash, Prataprao;** National Chemical Laboratory, Dr. Homi Bhabha Road, Pune 411008, Maharashtra (IN). **GARAI, Sumanta;** National Chemical Laboratory, Dr. Homi Bhabha Road, Pune 411008, Maharashtra (IN). **PAWAR, Kailash, Pralhad;** National Chemical Laboratory, Dr. Homi Bhabha Road, Pune 411008, Maharashtra (IN).
- (74) **Agent: DHAWAN, Ramesh, Chander;** Lall Lahiri & Salhotra, Plot No. B-28, Sector 32, Institutional Area, Gurgaon 122001, Haryana (IN).
- (81) **Designated States** (*unless otherwise indicated, for every kind of national protection available*): AE, AG, AL, AM,

AO, AT, AU, AZ, BA, BB, BG, BH, BN, BR, BW, BY, BZ, CA, CH, CL, CN, CO, CR, CU, CZ, DE, DK, DM, DO, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, GT, HN, HR, HU, ID, IL, IN, IS, JP, KE, KG, KN, KP, KR, KZ, LA, LC, LK, LR, LS, LT, LU, LY, MA, MD, ME, MG, MK, MN, MW, MX, MY, MZ, NA, NG, NI, NO, NZ, OM, PA, PE, PG, PH, PL, PT, QA, RO, RS, RU, RW, SC, SD, SE, SG, SK, SL, SM, ST, SV, SY, TH, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, ZA, ZM, ZW.

- (84) **Designated States** (*unless otherwise indicated, for every kind of regional protection available*): ARIPO (BW, GH, GM, KE, LR, LS, MW, MZ, NA, RW, SD, SL, SZ, TZ, UG, ZM, ZW), Eurasian (AM, AZ, BY, KG, KZ, RU, TJ, TM), European (AL, AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HR, HU, IE, IS, IT, LT, LU, LV, MC, MK, MT, NL, NO, PL, PT, RO, RS, SE, SI, SK, SM, TR), OAPI (BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, KM, ML, MR, NE, SN, TD, TG).

**Declarations under Rule 4.17:**

- *as to applicant's entitlement to apply for and be granted a patent (Rule 4.17(ii))*
- *of inventorship (Rule 4.17(iv))*

**Published:**

- *with international search report (Art. 21(3))*

(54) **Title:** ASYMMETRIC SYNTHESIS OF (-)-VENLAFAXINE USING ORGANOCATALYST

(57) **Abstract:** The patent discloses an asymmetric synthesis of (-)-venlafaxine using an organocatalyst via a unified strategy employing organocatalytic Michael addition, regio-selective dehydration and selective epoxide ring opening.



WO 2014/016852 A1

## ASYMMETRIC SYNTHESIS OF (-)-VENLAFAXINE USING ORGANOCATALYST

### Technical field of invention:

The invention relates to the asymmetric synthesis of (-)-venlafaxine using an organo catalyst. Particularly, the invention relates to the selective synthesis of one enantiomer of venlafaxine using the organocatalyst.

### Background and prior art:

Venlafaxine is a new generation antidepressant drug, first introduced in 1993. It is used for the treatment of major depressive disorder (MDD), as a treatment for generalized anxiety disorder, and co-morbid indications in certain anxiety disorders with depression. In 2007, venlafaxine was the sixth most commonly prescribed antidepressant on the U.S. retail market, with 17.2 million prescriptions. Although venlafaxine is sold as a racemate, (-)-venlafaxine is a more potent inhibitor of norepinephrine synaptosomal uptake while (+)-venlafaxine is more selective in serotonin uptake. It is different from other antidepressants in that it has no or little activity on a variety of neuroreceptors. (e.g.  $\alpha$  or  $\beta$ -adrenergic receptors, muscarinic receptors, cholinergic receptors, histaminic receptors etc.).

There are number of racemic syntheses reported for venlafaxine, including those by the inventors. These synthetic routes for racemic venlafaxine mainly involve the condensation of cyclohexanones with 4-methoxyphenyl acetic acids or 4-methoxyphenyl acetonitriles followed by functional group manipulation.

As both enantiomers possess different biological activities, therefore asymmetric synthesis of Venlafaxine is a subject matter of interest.

Nanda et al in Tetrahedron Letters 53 (2012) 1990-1992 reported an enzyme based resolution for asymmetric synthesis of venlafaxine. Their strategy included (S)-HNL

catalyzed synthesis of cyanohydrins from cyclic ketones and lipase-PS catalyzed kinetic resolution for creation of the stereocenter.

Chem. Commun., 2006, 3110-3112 disclose  $\beta$ -Amino esters which are readily formed from rhodium(II) proline-catalyzed intermolecular C-H insertion between methyl aryldiazoacetates and a bis-silyl protected methylamine. This was applied for effective synthesis of venlafaxine with enantiomers obtained with moderate yields moderate % ee.

But prior art methods suffer from the main drawback of having to resolve the enantiomers in a separate dedicated step, and yet result in only moderate yield. Also, these processes employ hazardous and potentially explosive reagents. They need dry, inert conditions during use of Grignard's reagent and many processes need cryogenic conditions. Also, these prior art processes use metal based catalyst which are not environmentally friendly.

#### **Objects of invention:**

The main object of the invention is to provide a process for asymmetric synthesis of (-)-venlafaxine, wherein one enantiomer is obtained in high enantiomeric purity.

Another object of the invention is to provide a process to those results selectively in one enantiomer of venlafaxine, without the need for a step of resolution.

#### **Summary of Invention:**

Accordingly, the present invention provides a process for asymmetric synthesis of enantiomerically pure venlafaxine with ee  $\geq$  99% comprising the steps of:

- a. reacting anisaldehyde with nitromethane in mole ratio 1: 11.8 in presence of ammonium acetate in acetic acid under sonication condition

at room temperature ranging between 25 – 35° C for a period ranging between 2-4 hrs to obtain nitro styrene;

- b. michael addition of nitrostyrene **as** obtained in step (a) with cyclohexanone in mole ratio 1:5 in presence of proline based organocatalyst under stirring at room temperature ranging between 25 – 35° C for a period ranging between 23-25 hrs in the presence of p-toluene sulphonic acid to obtain nitro ketone;
- c. reducing nitro ketone of step (b) using NaBH<sub>4</sub> in THF:H<sub>2</sub>O (9:1) to obtain crude alcohol (2*S*)-2-((*R*)-1-(4-methoxyphenyl)-2-nitroethyl)cyclohexan-1-ol which on subjecting to nitro reduction by NiCl<sub>2</sub>.6H<sub>2</sub>O and sodium borohydride in MeOH as a solvent, afforded the resultant amine (2*S*)-2-((*R*)-2-amino-1-(4-methoxyphenyl)ethyl)cyclohexan-1-ol which on *in situ* protection by benzylchloroformate in presence of Et<sub>3</sub>N as a base furnished Cbz protected amino alcohol benzyl ((2*R*)-2-((1*S*)-2-hydroxycyclohexyl)-2-(4-methoxyphenyl)ethyl)carbamate;
- d. treating amino alcohol of step (c) with mesyl chloride in presence of Et<sub>3</sub>N as a base in DCM solvent under reflux condition at temperature ranging between 40-45 ° C for a period ranging 14-25 hrs to give the crude mesylated reaction mixture which further on treatment with DBU in acetonitrile solvent furnished selectively more substituted double bond product benzyl (*R*)-(2-(cyclohex-1-en-1-yl)-2-(4-methoxyphenyl)ethyl)carbamate;
- e. subjecting compound (*R*)-(2-(cyclohex-1-en-1-yl)-2-(4-methoxyphenyl)ethyl)carbamate of step(d) with NaH and MeI in dry THF to obtain benzyl (*R*)-(2-(cyclohex-1-en-1-yl)-2-(4-methoxyphenyl)ethyl)(methyl)carbamate ;

- f. epoxidation of benzyl *(R)*-(2-(cyclohex-1-en-1-yl)-2-(4-methoxyphenyl)ethyl)(methyl)carbamate of step (e) by treating with *m*-CPBA in presence of NaHCO<sub>3</sub> in DCM under stirring at temperature ranging between 25–35 °C for a period ranging between 1–3 hrs to afford crude epoxide benzyl ((2*R*)-2-(7-oxabicyclo[4.1.0]heptan-1-yl)-2-(4-methoxyphenyl)ethyl)(methyl)carbamate;
- g. subjecting the crude epoxide of step (f) to selective epoxide opening as well as carbamate reduction in one pot using lithium aluminum hydride at reflux condition at temperature ranging between 65–70 ° C for a period ranging 4–5 hrs in THF to afford (–)-venlafaxine.

In one embodiment of the present invention the overall yield of enantiomerically pure (–)-venlafaxine is in the range of 21–22%.

In an embodiment of the present invention the enantioselectivity of (–)-venlafaxine is in the range of 99– 99.9%.

In another embodiment of the present invention proline based organocatalyst used in step (b) is (*S*)-*N*1, *N*1-dimethyl-*N*2-(pyrrolidin-2-ylmethyl)ethane-1,2-diamine

#### **Brief Description of figures:**

Figure 1: Chromatogram for racemic venlafaxine

Figure 2: Chromatogram for optically pure venlafaxine

Figure 3 Scheme I indicates Retrosynthetic analysis of (–)-venlafaxine.

Figure 4: Scheme 2 indicates synthesis of venlafaxine

#### **Detailed description of invention:**

Abbreviations used:

PTSA: *para*-Toluene sulphonic acid.

THF: Tetrahydrofuran.

Cbz: Carbobenzyloxy.

Ms: Methanesulphonyl

5 DBU: 1, 8-Diazabicyclo[5.4.0]undec-7-ene.

m-CPBA: *meta*-Chloroperoxybenzoic acid.

DCM: Dichloromethane

LAH: Lithium aluminium hydride.

The process of the invention is outlined in Scheme 1.

10 According to retrosynthetic analysis, synthesis of (-)-venlafaxine began with Henry reaction of commercially cheap, easily available starting material anisaldehyde **6** with nitromethane in presence of ammonium acetate in acetic acid under sonication condition at room temperature to furnish nitro styrene **5** in 95% yield. Michael addition  
15 of nitro styrene **5** with cyclohexanone in presence of proline based organocatalyst **11** gives nitro keto compound **4** in 79% with  $\geq 99\%$  *ee* after stirring 24 hours at room temperature in presence of *p*-toluene sulphonic acid (PTSA) as an additive in DMF solvent. Selective reduction of keto **4** using NaBH<sub>4</sub> in THF:H<sub>2</sub>O (9:1) as solvent system afforded alcohol. The crude alcohol was subjected to nitro reduction by NiCl<sub>2</sub>·6H<sub>2</sub>O and sodium borohydride in MeOH as a solvent, then the resultant amine was *in situ*  
20 protected by benzylchloroformate in presence Et<sub>3</sub>N as a base to furnish

Cbz protected amino alcohol **7** in 75% yield.

Scheme 2. Reagents and conditions: a) Nitromethane, NH<sub>4</sub>OAc, glacial acetic acid, ), 3 hrs, 95%; b) Cyclohexanone, **11**, PTSA, DMF, 24 hrs, 79%,  $\geq 99\%$  *ee*; c) i) NaBH<sub>4</sub>,

THF:H<sub>2</sub>O (9:1), 2 hrs.; ii) NiCl<sub>2</sub>·6H<sub>2</sub>O, NaBH<sub>4</sub>, MeOH, 1.5 hrs., 0 ° C then CbzCl, Et<sub>3</sub>N, rt, overnight, 75% (over two steps); d) i) MsCl, Et<sub>3</sub>N, reflux, 14 hrs; ii) DBU, CH<sub>3</sub>CN, 24 hrs, reflux, 68% (over two steps); e) MeI, NaH, THF, overnight, rt, 92%; f) i) *m*-CPBA, NaHCO<sub>3</sub>, DCM, 2 hrs., rt. ii) LiAlH<sub>4</sub>, THF, 5 hrs, reflux, 60% , ≥99% *ee*.

5 The hydroxyl group of compound **7** was converted into corresponding mesyl derivative by using mesyl chloride in presence of Et<sub>3</sub>N as a base in DCM solvent under reflux condition. The crude mesylated reaction mixture on treatment with DBU in acetonitrile solvent furnished selectively more substituted double bond product **8** in 68% yield. After introduction of double bond dihydroxylation reaction condition was tried for installation  
10 of tertiary hydroxyl group. After successful installation of diol through dihydroxylation (OsO<sub>4</sub>, NMO), selective removal of secondary hydroxyl group failed. So it was decided to install tertiary hydroxyl group through epoxidation and followed by epoxide opening. Thus the compound **8** was subjected with NaH and MeI in dry THF to afford compound **9** in 92% yield. For epoxidation compound **9** was treated with *m*-CPBA in presence of  
15 NaHCO<sub>3</sub> in DCM to afford epoxide. The crude epoxide **10** was subjected to selective epoxide opening as well as carbamate reduction in one pot using lithium aluminum hydride at reflux condition in THF to afford (–)-venlafaxine **1** in 60% yield with ≥99% *ee*. Spectral data and optical rotation for (–)-Venlafaxine **1** is provided herein in the form of examples.

20 This strategy of asymmetric synthesis of venlafaxine **1** by using organocatalyst can be extended to the synthesis of both enantiomers by switching the stereocentre of the catalyst with no loss in the optical activity of desired product. Derivatives of venlafaxine can be prepared in the same manner.

The invention is now explained with reference to embodiments and preferred  
25 embodiments, which in no way should be construed to be restrictive.

**Examples:****Example 1:** Synthesis of (-)-Venlafaxine.

Reacting anisaldehyde (20 gm, 0.147 mol) with nitromethane (94 mL, 1.741 mol) in presence of ammonium acetate in acetic acid (24 mL, 0.419 mol) under sonication condition at room temperature (25 ° C) for a period of 3 hrs to furnish 24.7 gm nitro styrene **5** in 95% yield. Michael addition of nitro styrene **5** (3 gm, 16.8 mmol) with cyclohexanone (8.2 gm, 84 mmol) in presence of proline based organocatalyst (*S*)-*N*1,*N*1-dimethyl-*N*2-(pyrrolidin-2-ylmethyl)ethane-1,2-diamine (115 mg, 0.67 mmol) gives 6.1 gm of (*S*)-2-((*R*)-1-(4-methoxyphenyl)-2-nitroethyl)cyclohexan-1-one **4** in 79% with  $\geq 99\%$  *ee* after stirring 24 hours at room temperature (25 ° C) in presence of *p*-toluene sulphonic acid (PTSA) (127 mg, 0.67 mmol) as an additive in DMF solvent. Selective reduction of keto **4** (2 gm, 7.2 mmol) using NaBH<sub>4</sub> (0.816 gm, 21.6 mmol) in THF:H<sub>2</sub>O (9:1) (20 ml), as solvent system afforded (*2S*)-2-((*R*)-1-(4-methoxyphenyl)-2-nitroethyl)cyclohexan-1-ol. The crude alcohol (2.06 gm, 7.4 mmol) was subjected to nitro reduction by NiCl<sub>2</sub>·6H<sub>2</sub>O (4.4 gm, 18.5 mmol) and sodium borohydride (7.03 gm, 0.185 mol) in MeOH (20 mL) as a solvent, then the resultant amine was *in situ* protected by benzylchloroformate (3.7 ml, 22.2 mmol) in presence Et<sub>3</sub>N (4 mL, 29.6 mmol) as a base to furnish 2.07 gm Cbz protected amino alcohol benzyl ((*2R*)-2-((*1S*)-2-hydroxycyclohexyl)-2-(4-methoxyphenyl)ethyl)carbamate in 75% yield. The hydroxyl group of Cbz protected amino alcohol (100 mg, 0.26 mmol) was converted into corresponding mesyl derivative by using mesyl chloride (0.06 mL, 0.78 mmol) in presence of Et<sub>3</sub>N (0.22 mL, 1.56 mmol) as a base in DCM solvent under reflux condition (40 ° C) for 14 hrs. The crude mesylated reaction mixture (120 mg) on treatment with DBU (1 mL) in acetonitrile solvent (3 mL) furnished 64.6 mg of selectively more substituted double bond product **8** benzyl (*R*)-2-(cyclohex-1-en-1-yl)-2-(4-methoxyphenyl)ethyl)carbamate in 68% yield. After introduction of double bond dihydroxylation reaction condition was tried for installation of tertiary hydroxyl group.



- After successful installation of diol through dihydroxylation ( $\text{OsO}_4$ , NMO), selective removal of secondary hydroxyl group failed. So it was decided to install tertiary hydroxyl group through epoxidation and followed by epoxide opening. Thus the compound **8** (100 mg, 0.274 mmol) was subjected with NaH (22 mg, 0.55 mmol, 60%) and MeI (0.034 mL, 0.55 mmol) in dry THF (5 mL) to afford 95 mg compound **9** benzyl (*R*)-(2-(cyclohex-1-en-1-yl)-2-(4-methoxyphenyl)ethyl)(methyl)carbamate in 92% yield. To a cold (0 °C), magnetically stirred solution of *N*-methylCbz compound **9** (235 mg, 0.6 mmol) in distilled DCM (5 mL),  $\text{NaHCO}_3$  (126 mg, 1.5 mmol) was added followed by 60% *m*-CPBA (348 mg, 1.2 mmol) was added portion wise and stirred for 2 hrs at rt (25 °C). The reaction was quenched with solid  $\text{NaHCO}_3$  (300 mg) and stirred for further 15 min. The reaction mixture was extracted with DCM (3 × 5 mL) and the combined organic layer was washed with brine (7 mL) and dried over anhydrous  $\text{Na}_2\text{SO}_4$ , filtered and concentrated under reduced pressure. The crude reaction mixture was used as such in the next reaction without further purification.
- To a cold (0 °C), magnetically stirred solution of lithium aluminum hydride (100 mg, 2.5 mmol) in dry THF (5 mL), crude epoxide **10** (100 mg, 0.25 mmol) was added dropwise and refluxed (66 °C) for 5 hrs. The reaction mixture was cooled to 0 °C and excess LAH was quenched with ethyl acetate and then by addition of water, stirred for 2 hrs. Evaporation of the solvent furnished a residue which was extracted with ethyl acetate (3 × 20 mL). The combined organic layer was washed with brine (20 mL) and dried over anhydrous  $\text{Na}_2\text{SO}_4$ , filtered and concentrated under reduced pressure. Purification of the residue on a silica gel column using ethyl acetate as eluent furnished the (-)-venlafaxine **1** (103 mg, 60%) as a white solid.

**Example 2:** Characterization data of (-)-Venlafaxine.

- The product of the process enlisted in example 1 was characterized by IR and  $^1\text{H}$  and  $^{13}\text{C}$  NMR and results are as follows:

$R_f$  (100% EtOAc) 0.2 (long tail); IR ( $\text{CHCl}_3$ ): 3164, 2982, 2938, 2860, 2782, 1610, 1512  $\text{cm}^{-1}$ ;

$^1\text{H}$  NMR (200 MHz,  $\text{CDCl}_3 + \text{CCl}_4$ ):  $^1\text{H}$  NMR (200 MHz,  $\text{CDCl}_3 + \text{CCl}_4$ ): 0.83–1.00 (m, 2H), 1.23–1.76 (m, 8H), 2.28 (dd,  $J=12.2$ , 2.9 Hz, 1H), 2.33 (s, 6H), 2.93 (dd,  $J=12.2$ , 2.9 Hz, 1H), 3.28 (t,  $J=12.2$  Hz, 1H), 3.79 (s, 3H), 6.79 (d,  $J=8.8$  Hz, 2H), 7.03 (d,  $J=8.79$  Hz, 2H).  $^{13}\text{C}$  NMR (50 MHz,  $\text{CDCl}_3 + \text{CCl}_4$ ): 20.70, 21.05, 25.55, 30.72, 37.53, 44.89, 51.20, 54.36, 60.74, 73.48, 112.75, 129.43, 132.00, 157.72.

Example 3: Optical purity of (–)-Venlafaxine.

(*R*)-venlafaxine  $[\alpha] = -24.285$  ( $c = 1.04$ , EtOH).

10 Column: Kromasil 5–Amy Coat ( $250 \times 4.6$  mm)

Mobile Phase: EtOH: Pet ether: Diethylamine (05: 95: 0.5)

Wave length: 254 nm

<u>Racemic:</u>		<u>Chiral</u>	
Retention time	Area%	Retention time	Area%
15 12.075		47.587	
15.158		100.000	
15.283		52.413	

#### Advantages of invention:

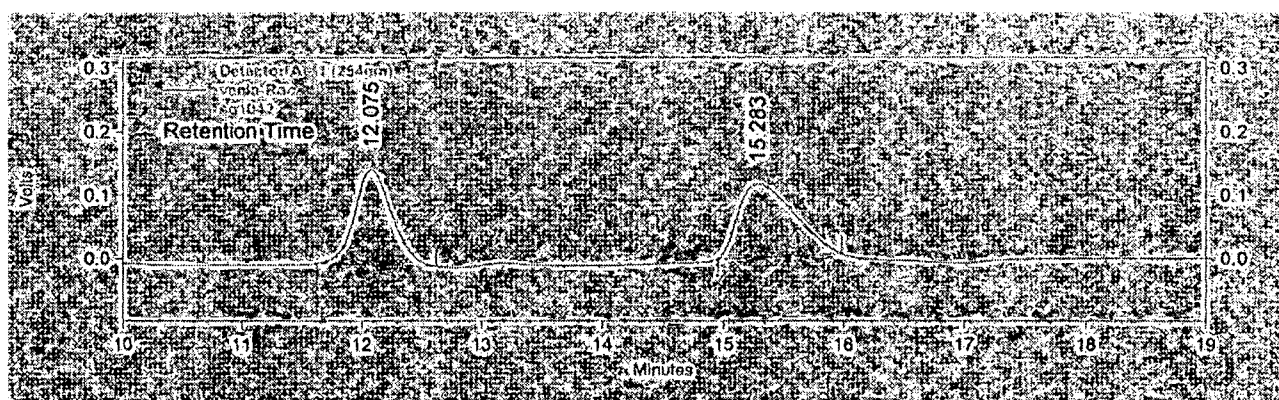
1. Use of cheap and easily available raw materials
- 20 2. Use of cheap and environmentally friendly catalyst
3. Avoidance of expensive and metal based catalyst
4. Avoidance of additional steps involving resolution of enantiomers

5. High % ee purity of product obtained

**We claim:**

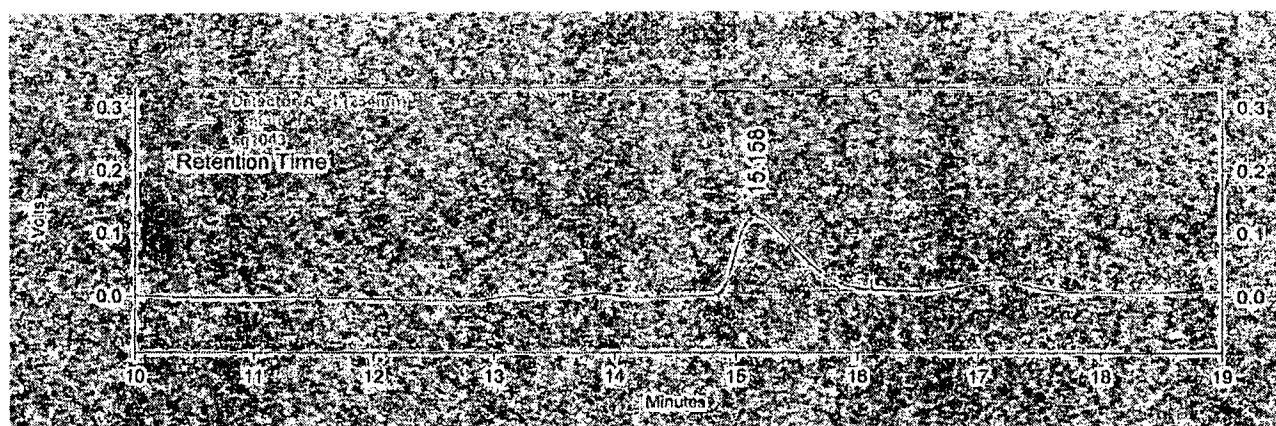
1. A process for asymmetric synthesis of enantiomerically pure venlafaxine with ee  $\geq$  99% comprising the steps of:
  - a. reacting anisaldehyde with nitromethane in mole ratio 1: 11.8 in presence of ammonium acetate in acetic acid under sonication condition at room temperature ranging between 25 – 35° C for a period ranging between 2-4 hrs to obtain nitro styrene;
  - b. michael addition of nitrostyrene as obtained in step (a) with cyclohexanone in mole ratio 1:5 in presence of proline based organocatalyst under stirring at room temperature ranging between 25 – 35° C for a period ranging between 23-25 hrs in the presence of p- toluene sulphonic acid to obtain nitro ketone;
  - c. reducing nitro ketone of step (b) using NaBH<sub>4</sub> in THF:H<sub>2</sub>O (9:1) to obtain crude alcohol (2*S*)-2-((*R*)-1-(4-methoxyphenyl)-2-nitroethyl)cyclohexan-1-ol which on subjecting to nitro reduction by NiCl<sub>2</sub>.6H<sub>2</sub>O and sodium borohydride in MeOH as a solvent, afforded the resultant amine (2*S*)-2-((*R*)-2-amino-1-(4-methoxyphenyl)ethyl)cyclohexan-1-ol which on *in situ* protection by benzylchloroformate in presence of Et<sub>3</sub>N as a base furnished Cbz protected amino alcohol benzyl ((2*R*)-2-((1*S*)-2-hydroxycyclohexyl)-2-(4-methoxyphenyl)ethyl)carbamate;
  - d. treating amino alcohol of step (c) with mesyl chloride in presence of Et<sub>3</sub>N as a base in DCM solvent under reflux condition at temperature ranging between 40-45 ° C for a period ranging 14-25 hrs to give the crude mesylated reaction mixture which further on treatment with DBU in acetonitrile solvent furnished selectively more substituted double bond product benzyl (*R*)-2-(cyclohex-1-en-1-yl)-2-(4-methoxyphenyl)ethyl)carbamate;

- e. subjecting compound  $(R)$ -(2-(cyclohex-1-en-1-yl)-2-(4-methoxyphenyl)ethyl)carbamate of step(d) with NaH and MeI in dry THF to obtain benzyl  $(R)$ -(2-(cyclohex-1-en-1-yl)-2-(4-methoxyphenyl)ethyl)(methyl)carbamate ;
- 5 f. epoxidation of benzyl  $(R)$ -(2-(cyclohex-1-en-1-yl)-2-(4-methoxyphenyl)ethyl)(methyl)carbamate of step (e) by treating with *m*-CPBA in presence of NaHCO<sub>3</sub> in DCM under stirring at temperature ranging between 25-35 °C for a period ranging between 1-3 hrs to afford crude epoxide benzyl ((2*R*)-2-(7-oxabicyclo[4.1.0]heptan-1-yl)-2-(4
- 10 methoxyphenyl)ethyl)(methyl)carbamate;
- g. subjecting the crude epoxide of step (f) to selective epoxide opening as well as carbamate reduction in one pot using lithium aluminum hydride at reflux condition at temperature ranging between 65-70 ° C for a period ranging 4-5 hrs in THF to afford (-)-venlafaxine.
- 15 2. The process according to claim 1, wherein the yield of enantiomerically pure (-)-venlafaxine is in the range of 21-58%.
3. The process according to claim 1, wherein the enantioselectivity of (-)-venlafaxine is in the range of 97- 99%.
4. The process according to claim 1, wherein proline based organocatalyst used in
- 20 step (b) is (*S*)-*N*1,*N*1-dimethyl-*N*2-(pyrrolidin-2-ylmethyl)ethane-1,2-diamine.



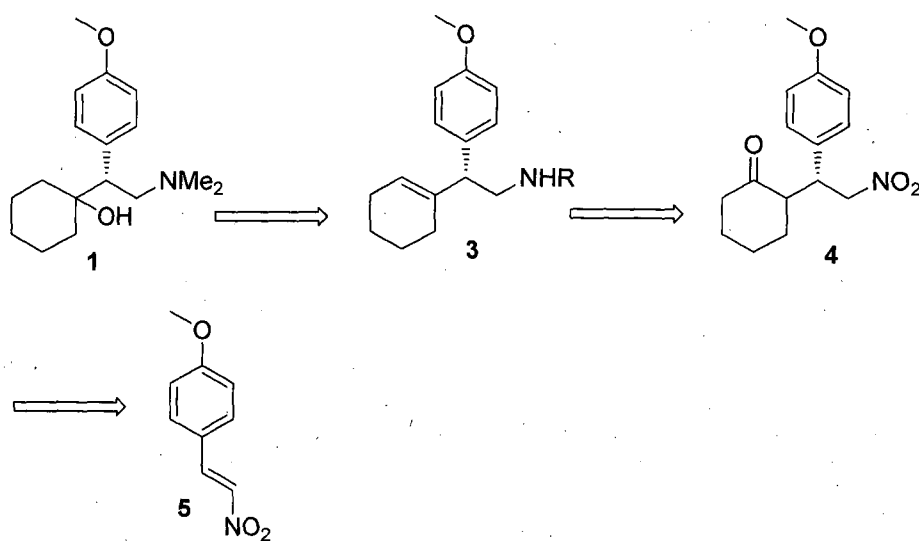
Chromatogram for racemic venlafaxine

Figure 1



Chromatogram for optically pure venlafaxine

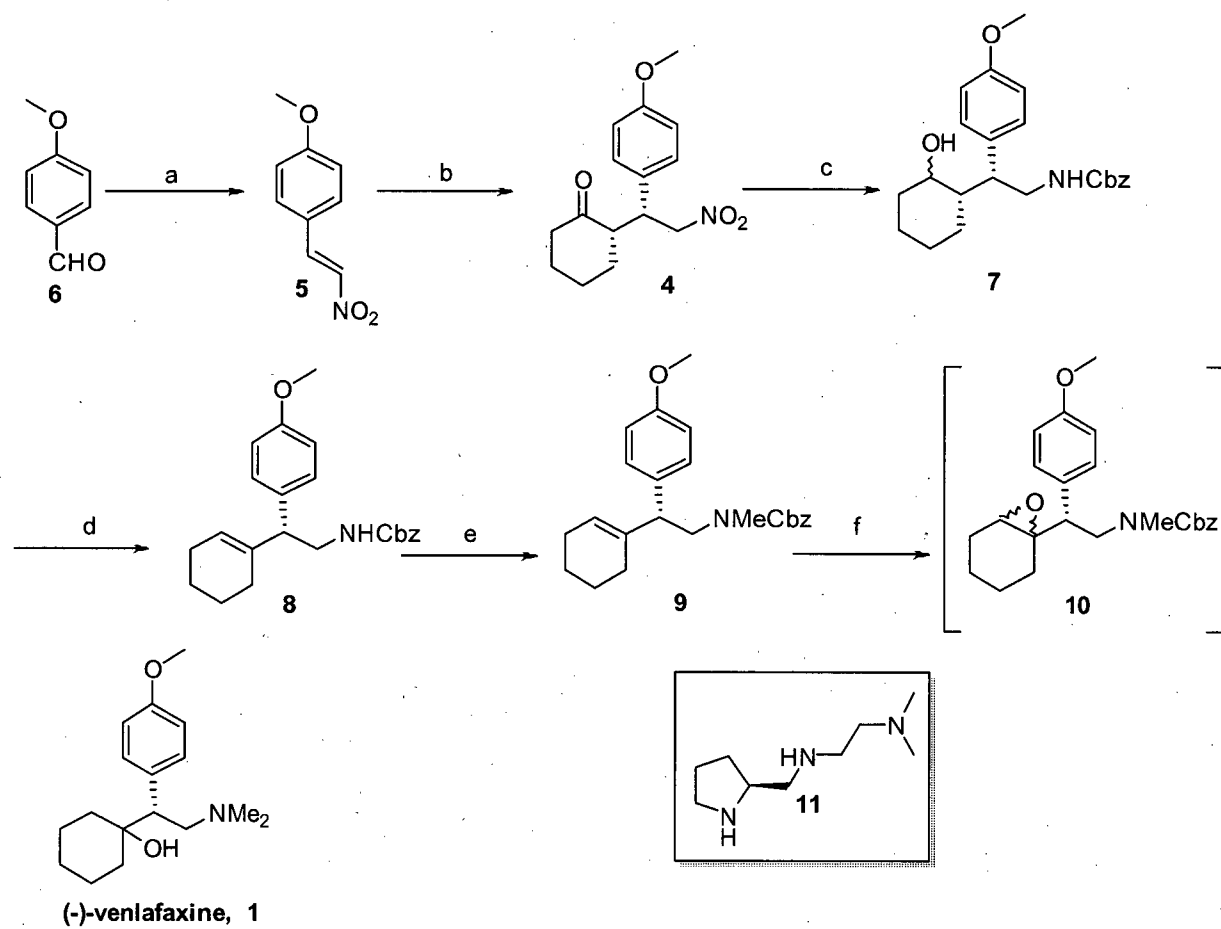
Figure 2



**Scheme 1: Retrosynthetic analysis of (-)-venlafaxine**

**Figure 3**





Scheme 2: Synthesis of venlafaxine

Figure 4

## INTERNATIONAL SEARCH REPORT

International application No

PCT/IN2013/000464

A. CLASSIFICATION OF SUBJECT MATTER  
 INV. C07C213/00  
 ADD.

According to International Patent Classification (IPC) or to both national classification and IPC

## B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)  
 C07C

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practicable, search terms used)

EPO-Internal

## C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X,P	SUBHASH P. CHAVAN ET AL: "Asymmetric total synthesis of (-)-venlafaxine using an organocatalyst", TETRAHEDRON LETTERS, vol. 54, no. 17, 1 April 2013 (2013-04-01) , pages 2137-2139, XP055085047, ISSN: 0040-4039, DOI: 10.1016/j.tetlet.2013.02.029 the whole document	1-4
A	NANDA ET AL.: "Asymmetric Synthesis of both the enantiomers of antidepressant venlafaxine and its analogues", TETRAHEDRON LETTERS, vol. 53, 2012, pages 1990-1992, XP002715302, cited in the application the whole document	1-4



Further documents are listed in the continuation of Box C.



See patent family annex.

## \* Special categories of cited documents :

"A" document defining the general state of the art which is not considered to be of particular relevance

"E" earlier application or patent but published on or after the international filing date

"L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)

"O" document referring to an oral disclosure, use, exhibition or other means

"P" document published prior to the international filing date but later than the priority date claimed

"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention

"X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone

"Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art

"&" document member of the same patent family

Date of the actual completion of the international search

23 October 2013

Date of mailing of the international search report

07/11/2013

Name and mailing address of the ISA/

European Patent Office, P.B. 5818 Patentlaan 2  
 NL - 2280 HV Rijswijk  
 Tel. (+31-70) 340-2040,  
 Fax: (+31-70) 340-3016

Authorized officer

Tabanella, Stefania