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(54) Title: F,G,H,I AND K CRYSTAL FORMS OF IMATINIB MESYLATE

(57) Abstract: The invention relates to the F-crystal form, G-crystal form, H-crystal form, I-crystal form and K-crystal form of the methanesulfonic acid addition salt of 4-(4-methylpiperazin-1-ylmethyl)- N-[4-methyl-3-(4-(pyridin-3-yl)pyrimidin-2-ylamino)phenyl]-benzamide, certain processes for their preparation, pharmaceutical compositions containing these crystal forms, their use in diagnostic methods or for the therapeutic treatment of warm-blooded animals, and their use as an intermediate or for the preparation of pharmaceutical preparations for use in diagnostic methods or for the therapeutic treatment of warm-blooded animals, especially humans.

WO 2007/059963 A1

F, G, H, I and K Crystal Forms of Imatinib Mesylate

The invention relates to particular crystal forms of the methanesulfonic acid addition salt of 4-(4-methylpiperazin-1-ylmethyl)-N-[4-methyl-3-(4-(pyridin-3-yl)pyrimidin-2-ylamino)phenyl]-benzamide (the compound of formula I, see below), certain processes for their preparation, pharmaceutical compositions containing these crystal forms, and their use in diagnostic methods or, preferably, for the therapeutic treatment of warm-blooded animals, especially humans, and their use as an intermediate or for the preparation of pharmaceutical preparations for use in diagnostic methods or, preferably, for the therapeutic treatment of warm-blooded animals, especially humans.

Background to the invention

The preparation of 4-(4-methylpiperazin-1-ylmethyl)-N-[4-methyl-3-(4-(pyridin-3-yl)pyrimidin-2-ylamino)phenyl]-benzamide, also known as Imatinib, and its use, especially as an anti-tumour agent, are described in Example 21 of EP-A-0 564 409, which was published on 6 October 1993, and in equivalent applications in numerous other countries. The compound is exemplified in these publications only in free form (not as a salt).

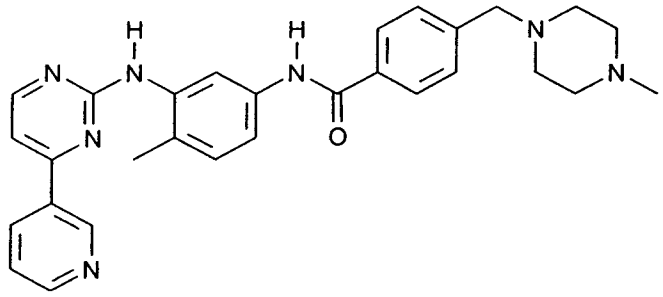
4-(4-Methylpiperazin-1-ylmethyl)-N-[4-methyl-3-(4-(pyridin-3-yl)pyrimidin-2-ylamino)phenyl]-benzamide mesylate, also known as Imatinib mesylate or STI571, the alpha and the beta crystal form thereof, as well as its pharmaceutical use are described in US 6,894,051. Imatinib mesylate is the active ingredient of the drug Gleevec® (Glivec®) which is an approved medicament for the treatment of Chronic Myeloid Leukemia (CML) and gastrointestinal stromal tumors (GIST). Another polymorph of Imatinib mesylate, the so-called H1-form, is described in WO2004/106326.

It has now been surprisingly found that under certain conditions new crystal forms of the methanesulfonate salt may be found, which are described hereinafter as F-crystal form, G-crystal form, H-crystal form, I-crystal form and K-crystal form and which forms have advantageous utilities and properties.

Detailed description of the invention

The invention is described in more detail in the following with the help of drawings and other aids.

The invention relates especially to essentially pure crystal forms, preferably those which are referred to hereinafter as the F-crystal form, G-crystal form, H-crystal form, I-crystal form and K-crystal form of the methanesulfonic acid addition salt of Imatinib of formula I,



(I).

Description of the drawings

The F-crystal form of the methanesulfonic acid addition salt of a compound of formula I is characterized by lines in the X-ray diffraction diagram observed at an angle of refraction 2theta of 8.4° and 8.6°.

Fig. 1 shows the X-ray diffraction diagram of the F-crystal form of the methanesulfonic acid addition salt of a compound of formula I. In the X-ray diagram, the angle of refraction 2theta is plotted on the horizontal axis (*x*-axis) and the relative line intensity (background-corrected peak intensity) on the vertical (*y*-axis). X-ray powder diffraction patterns are measured on a Bruker D8 GADDS Discover Diffractometer with Cu Kα radiation source (Kα1 radiation, wavelength λ = 1.54060 Angström). The optical density of the lines on the film is proportional to the light intensity. The film is scanned in using a line scanner. The strongest line in the X-ray diffraction diagram is observed at an angle of refraction 2theta of 20.9°. More broadly, the F-crystal form is characterized by refractions at angles of refraction 2theta of 8.4°, 8.6°, 13.3°, 16.2°, 16.8°, 17.1°, 19.5°, 20.9°, 23.6° and 24.5°. In essentially pure material of the F-crystal form of the methanesulfonic acid addition salt of a compound of formula I, lines can

be observed at angles of refraction 2theta 8.4°, 8.6°, 10.4°, 13.3,° 14.7°, 16.2°, 16.8°, 17.1°, 19.5°, 20.9°, 22.2°, 23.1°, 23.6°, 24.5°, 25.1°, 26.0°, 26.9°, 28.5°, 29.1° and 30.3°.

The G-crystal form of the methanesulfonic acid addition salt of a compound of formula I is characterized by a line in the X-ray diffraction diagram observed at an angle of refraction 2theta of 10.5° together with the absence of any lines between an angle of refraction 2theta of 4° and 8°.

Fig. 2 shows the X-ray diffraction diagram of the G-crystal form of the methanesulfonic acid addition salt of a compound of formula I. In the X-ray diagram, the angle of refraction 2theta is plotted on the horizontal axis (*x*-axis) and the relative line intensity (background-corrected peak intensity) on the vertical (*y*-axis). X-ray powder diffraction patterns are measured on a Bruker D8 GADDS Discover Diffractometer with Cu Kα radiation source (Kα1 radiation, wavelength λ = 1.54060 Angström). The optical density of the lines on the film is proportional to the light intensity. The film is scanned in using a line scanner. In the X-ray diffraction diagram of the G-crystal form lines are observed at an angle of refraction 2theta of 10.5°, 18.1° and 18.7°. More broadly, the G-crystal form is characterized by refractions at angles of refraction 2theta of 10.5°, 15.0°, 17.2°, 18.1°, 18.7°, 19.2°, 21.1° and 21.3°. In essentially pure material of the G-crystal form of the methanesulfonic acid addition salt of a compound of formula I, lines can be observed at angles of refraction 2theta 10.5°, 12.9°, 13.9°, 14.1°, 15.0°, 16.6°, 17.2°, 17.5°, 18.1°, 18.7°, 19.2°, 19.8°, 20.6°, 21.1°, 21.3°, 21.7°, 22.1°, 22.8°, 23.9°, 24.3°, 25.1° and 28.6°.

The H-crystal form of the methanesulfonic acid addition salt of a compound of formula I is characterized by a line in the X-ray diffraction diagram observed at an angle of refraction 2theta of 32.9°.

Fig. 3 shows the X-ray diffraction diagram of the H-crystal form of the methanesulfonic acid addition salt of a compound of formula I. In the X-ray diagram, the angle of refraction 2theta is plotted on the horizontal axis (*x*-axis) and the relative line intensity (background-corrected peak intensity) on the vertical (*y*-axis). X-ray powder diffraction patterns are measured on a Bruker D8 GADDS Discover Diffractometer with Cu Kα radiation source (Kα1 radiation, wavelength λ = 1.54060 Angström). The optical density of the lines on the film is proportional to the light intensity. The film is scanned in using a line scanner. In the X-ray diffraction

diagram the H-crystal form is dominated by a line at an angle of refraction 2theta of 25.1°. More broadly, the H-crystal form is characterized by refractions at angles of refraction 2theta of 10.5°, 22.8°, 25.1° and 32.9°. In essentially pure material of the H-crystal form of the methanesulfonic acid addition salt of a compound of formula I, lines can be observed at angles of refraction 2theta 10.5°, 13.8°, 15.7°, 18.1°, 21.0°, 22.8°, 24.3°, 25.1°, 26.3°, 29.7° and 32.9°.

The I-crystal form of the methanesulfonic acid addition salt of a compound of formula I is characterized by a line in the X-ray diffraction diagram observed at an angle of refraction 2theta of 12.9° together with the absence of any lines at an angle of refraction 2theta below 9° and above 29°.

Fig. 4 shows the X-ray diffraction diagram of the I-crystal form of the methanesulfonic acid addition salt of a compound of formula I. In the X-ray diagram, the angle of refraction 2theta is plotted on the horizontal axis (*x*-axis) and the relative line intensity (background-corrected peak intensity) on the vertical (*y*-axis). X-ray powder diffraction patterns are measured on a Bruker D8 GADDS Discover Diffractometer with Cu Kα radiation source (Kα1 radiation, wavelength λ = 1.54060 Angström). The optical density of the lines on the film is proportional to the light intensity. The film is scanned in using a line scanner. In the X-ray diffraction diagram of the I-crystal form lines are observed at an angle of refraction 2theta of 12.9°, 17.1°, 20.9°, 23.9° and 24.3°. More broadly, the I-crystal form is characterized by refractions at angles of refraction 2theta of 12.9°, 14.1°, 17.1°, 18.0°, 18.7°, 19.1°, 19.8°, 20.9°, 23.9°, 24.3° and 25.2°. In essentially pure material of the I-crystal form of the methanesulfonic acid addition salt of a compound of formula I, lines can be observed at angles of refraction 2theta 9.6°, 12.9°, 14.1°, 15.2°, 15.6°, 17.1°, 18.0°, 18.7°, 19.1°, 19.8°, 20.9°, 23.4°, 23.9°, 24.3°, 25.2° and 28.4°.

The K-crystal form of the methanesulfonic acid addition salt of a compound of formula I is characterized by a line in the X-ray diffraction diagram observed at an angle of refraction 2theta of 37.9°.

Fig. 5 shows the X-ray diffraction diagram of the K-crystal form of the methanesulfonic acid addition salt of a compound of formula I. In the X-ray diagram, the angle of refraction 2theta is plotted on the horizontal axis (*x*-axis) and the relative line intensity (background-corrected

peak intensity) on the vertical (*y*-axis). X-ray powder diffraction patterns are measured on a Bruker D8 GADDS Discover Diffractometer with Cu Kα radiation source (Kα1 radiation, wavelength λ = 1.54060 Angström). The optical density of the lines on the film is proportional to the light intensity. The film is scanned in using a line scanner. In the X-ray diffraction diagram the K-crystal form is further characterized by a line at an angle of refraction 2theta of 21.0°. More broadly, the K-crystal form is characterized by refractions at angles of refraction 2theta of 12.1°, 14.1°, 18.2°, 18.4°, 21.0°, 23.4° and 28.4°. In essentially pure material of the K-crystal form of the methanesulfonic acid addition salt of a compound of formula I, lines can be observed at angles of refraction 2theta 12.1°, 12.9°, 13.6°, 14.1°, 15.2°, 17.2°, 18.2°, 18.4°, 19.8°, 21.0°, 22.4°, 23.4°, 24.3°, 25.2°, 28.4°, 29.2° and 37.9°.

The term "essentially pure" is understood in the context of the present invention to mean especially that at least 90, preferably at least 95, and most preferably at least 99 per cent by weight of the crystals of an acid addition salt of formula I are present in the specified crystal form according to the invention.

In the context with stating that the F-crystal form of the methanesulfonic acid addition salt of a compound of formula I exhibits an X-ray diffraction diagram essentially as in Fig. 1, the term "essentially" means that at least the major lines of the diagram depicted in Fig. 1, i.e. those having a relative line intensity of more than 20%, especially more than 30 %, as compared to the most intense line in the diagram, have to be present.

Alternatively, the F-crystal form of the methanesulfonic acid addition salt of a compound of formula I is characterized by a DSC curve showing a melting event at 95°C followed by an exothermic recrystallization and a second melting at about 223°C.

In one preferred embodiment, the essentially pure methanesulfonic acid addition salt of a compound of formula I in the F-crystal form shows the X-ray diffraction diagram indicated in Fig. 1.

High preference is also given for the F-crystal form of the methanesulfonic acid addition salt of a compound of formula I which shows an X-ray diffraction diagram of the type shown in Fig. 1, in which the relative peak intensities of each peak do not deviate by more than 10%

from the relative peak intensities in the diagram shown in Fig. 1, especially an X-ray diffraction diagram identical to that shown in Fig. 1.

In the context with stating that the G-crystal form of the methanesulfonic acid addition salt of a compound of formula I exhibits an X-ray diffraction diagram essentially as in Fig. 2, the term "essentially" means that at least the major lines of the diagram depicted in Fig. 2, i.e. those having a relative line intensity of more than 20%, especially more than 30 %, as compared to the most intense line in the diagram, have to be present.

In one preferred embodiment, the essentially pure methanesulfonic acid addition salt of a compound of formula I in the G-crystal form shows the X-ray diffraction diagram indicated in Fig. 2.

High preference is also given for the G-crystal form of the methanesulfonic acid addition salt of a compound of formula I which shows an X-ray diffraction diagram of the type shown in Fig. 2, in which the relative peak intensities of each peak do not deviate by more than 10% from the relative peak intensities in the diagram shown in Fig. 2, especially an X-ray diffraction diagram identical to that shown in Fig. 2.

In the context with stating that the H-crystal form of the methanesulfonic acid addition salt of a compound of formula I exhibits an X-ray diffraction diagram essentially as in Fig. 3, the term "essentially" means that at least the major lines of the diagram depicted in Fig. 3, i.e. those having a relative line intensity of more than 20%, especially more than 30 %, as compared to the most intense line in the diagram, have to be present.

In another preferred embodiment, the essentially pure methanesulfonic acid addition salt of a compound of formula I in the H-crystal form shows the X-ray diffraction diagram indicated in Fig. 3.

High preference is also given for the H-crystal form of the methanesulfonic acid addition salt of a compound of formula I which shows an X-ray diffraction diagram of the type shown in Fig. 3, in which the relative peak intensities of each peak do not deviate by more than 10% from the relative peak intensities in the diagram shown in Fig. 3, especially an X-ray diffraction diagram identical to that shown in Fig. 3.

In the context with stating that the I-crystal form of the methanesulfonic acid addition salt of a compound of formula I exhibits an X-ray diffraction diagram essentially as in Fig. 4, the term "essentially" means that at least the major lines of the diagram depicted in Fig. 4, i.e. those having a relative line intensity of more than 20%, especially more than 30 %, as compared to the most intense line in the diagram, have to be present.

In a further preferred embodiment, the essentially pure methanesulfonic acid addition salt of a compound of formula I in the I-crystal form shows the X-ray diffraction diagram indicated in Fig. 4.

High preference is also given for the I-crystal form of the methanesulfonic acid addition salt of a compound of formula I which shows an X-ray diffraction diagram of the type shown in Fig. 4, in which the relative peak intensities of each peak do not deviate by more than 10% from the relative peak intensities in the diagram shown in Fig. 4, especially an X-ray diffraction diagram identical to that shown in Fig. 4.

In the context with stating that the K-crystal form of the methanesulfonic acid addition salt of a compound of formula I exhibits an X-ray diffraction diagram essentially as in Fig. 5, the term "essentially" means that at least the major lines of the diagram depicted in Fig. 5, i.e. those having a relative line intensity of more than 20%, especially more than 30 %, as compared to the most intense line in the diagram, have to be present.

In another preferred embodiment, the essentially pure methanesulfonic acid addition salt of a compound of formula I in the K-crystal form shows the X-ray diffraction diagram indicated in Fig. 5.

High preference is also given for the K-crystal form of the methanesulfonic acid addition salt of a compound of formula I which shows an X-ray diffraction diagram of the type shown in Fig. 5, in which the relative peak intensities of each peak do not deviate by more than 10% from the relative peak intensities in the diagram shown in Fig. 5, especially an X-ray diffraction diagram identical to that shown in Fig. 5.

The invention expressly relates also to those forms of the methanesulfonic acid addition salt of a compound of formula I in which crystals of the F-crystal form, G-crystal form, H-crystal form, I-crystal form and K-crystal form are present along with other crystal forms, in particular the α-crystal form, the β-crystal form, and/or the amorphous form of the Imatinib mesylate. Preferred, however, is the essentially pure form in the F-crystal form, G-crystal form, H-crystal form, I-crystal form or K-crystal form.

Particularly special preference is for the crystal forms of the methanesulfonic acid addition salt of a compound of formula I obtainable as described in the Examples.

One utility of the F-crystal form, G-crystal form, H-crystal form, I-crystal form and K-crystal form of the methanesulfonic acid addition salt of a compound of formula I is the use as an intermediate for the preparation of a distinct crystal form of the methanesulfonic acid addition salt of a compound of formula I, especially the β-crystal form. The (preferably essentially pure) β-crystal form is obtainable by

a) digesting the F-crystal form , G-crystal form, H-crystal form, I-crystal form or the K-crystal form of the methanesulfonic acid addition salt of a compound of formula I with a suitable polar solvent, especially an alcohol, most especially methanol, or also a ketone (especially in a mixture with water, for example water/acetone), typically acetone, a N,N-di-lower alkyl-lower alkanecarboxamide, typically N,N-dimethylformamide or -acetamide, or a hydrophilic ether, typically dioxane, preferably in the presence of some water, or mixtures thereof, in suspension at a suitable temperature, preferably a temperature between 20 and 50°C, for example at about 25°C, or

b) dissolving the F-crystal form, G-crystal form, H-crystal form, I-crystal form or the K-crystal form of the methanesulfonic acid addition salt of a compound of formula I with a suitable polar solvent, such as especially an alcohol, typically methanol or ethanol, a ketone (especially in a mixture with water, for example water/acetone) typically acetone, a N,N-di-lower alkyl-lower alkanecarboxamide, typically N,N-dimethylformamide or -acetamide, or a hydrophilic ether, typically dioxane, or mixtures thereof, preferably in the presence of some water, at a suitable temperature, especially after heating the solvent, or while warming during the dissolution process, in both cases preferably to 25°C up to the reflux temperature of the reaction mixture, and then initiating crystallisation by adding a small amount of the β-crystal form as seed crystal at a suitable temperature, for example between 0 and 70°C, preferably between 20 and 70°C.

One of the advantages of having access to different crystal forms of the compound of formula I is the fact that distinct crystal forms are prone to incorporate distinct impurities upon crystallization, i.e. an impurity incorporated in crystal form β is not necessarily also incorporated in the F-crystal form, G-crystal form, H-crystal form, I-crystal form or K-crystal form. With other words, preparing consecutively distinct crystal forms of the same material increases the purity of the finally obtained substance. Furthermore, distinct crystal forms display different physical properties such as melting points, hygroscopicities, solubilities, flow properties or thermodynamic stabilities, and, hence, distinct crystal forms allow the choice of the most suitable form for a certain use or aspect, e.g. the use as an intermediate in the process of drug manufacture or in distinct administration forms like tablets, capsules, ointments or solutions.

The F-crystal form, G-crystal form, H-crystal form, I-crystal form and K-crystal form of the methanesulfonic acid addition salt of a compound of formula I possess valuable pharmacological properties and may, for example, be used as an anti-tumour agent or as an agent to treat restenosis.

The present invention relates especially to the F-crystal form, G-crystal form, H-crystal form, I-crystal form and K-crystal form of the methanesulfonic acid addition salt of a compound of formula I in the treatment of one of the said diseases mentioned herein or in the preparation of a pharmacological agent for the treatment thereof.

The antiproliferative, especially anti-tumour, activity of the methanesulfonic acid addition salt of a compound of formula I *in vivo* is, for example, described for the treatment of abl-dependent tumours in Nature Med. 2, 561-6 (1996).

The invention relates also to a method for the treatment of warm-blooded animals suffering from said diseases, especially leukemia, wherein a quantity of the F-crystal form , G-crystal form, H-crystal form, I-crystal form or K-crystal form of the methanesulfonic acid addition salt of a compound of formula I which is effective against the disease concerned, especially a quantity with antiproliferative efficacy, is administered to warm-blooded animals in need of such treatment. The invention relates moreover to the use of the F-crystal form, G-crystal form, H-crystal form, I-crystal form and K-crystal form of the methanesulfonic acid addition

salt of a compound of formula I for the preparation of pharmaceutical compositions for use in treating the human or animal body, especially for the treatment of tumours, such as gliomas or prostate tumours.

In preferred embodiments, the present invention relates to the use in of the F-crystal form, G-crystal form, H-crystal form, I-crystal form and K-crystal form of the methanesulfonic acid addition salt of a compound of formula I in the treatment of one of the disorders listed below:

1. metastatic, inoperable GIST,
2. advanced chronic myeloid leukemia,
3. newly diagnosed chronic myeloid leukemia,
4. pediatric Philadelphia chromosome-positive chronic myeloid leukemia,
5. Philadelphia chromosome-positive acute lymphocytic leukemia (ALL),
6. glioblastoma multiforme, preferably in combination with hydroxyurea,
7. dermatofibrosarcoma protuberans (DFSP),
8. hypereosinophilic sindrome (HES), and
9. chronic myelomonocytic leucemia (CMML).

Depending on species, age, individual condition, mode of administration, and the clinical picture in question, effective doses, for example daily doses of about 50–2500 mg, preferably 100–1000 mg, especially 250–800 mg, of Imatinib having the F-crystal form G-crystal form, H-crystal form, I-crystal form or the K-crystal form, are administered to warm-blooded animals of about 70 kg bodyweight. Preferably, daily dosages of 400 mg or 600 mg are administered orally once daily, preferably together with a meal and a large glass of water (about 200 mL).
800 mg daily dosages are preferably administered in the form of 400 mg dosages twice daily together with food.

The F-crystal form, G-crystal form, H-crystal form, I-crystal form and K-crystal form described herein can be utilized to prepare stable pharmaceutical dosage forms. Hence, the invention relates also to pharmaceutical preparations which contain an amount, especially an effective amount for prevention or treatment of one of the diseases mentioned herein, of the methanesulfonic acid addition salt of a compound of formula I in the F–crystal form, G-crystal form, H-crystal form, I-crystal form or K-crystal form, together with pharmaceutically acceptable carriers which are suitable for topical, enteral, for example oral or rectal, or

parenteral administration and may be inorganic or organic and solid or liquid. Especially tablets or gelatin capsules containing the active substance together with diluents, for example lactose, dextrose, sucrose, mannitol, sorbitol, cellulose, and/or glycerin, and/or lubricants, for example silica, talc, stearic acid, or salts thereof, typically magnesium or calcium stearate, and/or polyethylene glycol, are used for oral administration. Tablets may likewise contain binders, for example magnesium aluminium silicate, starches, typically corn, wheat or rice starch, gelatin, methylcellulose, sodium carboxymethylcellulose and/or polyvinylpyrrolidone, and, if so desired, disintegrants, for example starches, agar, alginic acid, or a salt thereof, typically sodium alginate, and/or effervescent mixtures, or adsorbents, colouring agents, flavours, and sweetening agents. The pharmacologically active compounds of the present invention may further be used in the form of preparations for parenteral administration or infusion solutions. Such solutions are preferably isotonic aqueous solutions or suspensions, these possibly being prepared before use, for example in the case of lyophilised preparations containing the active substance either alone or together with a carrier, for example mannitol. The pharmaceutical substances may be sterilised and/or may contain excipients, for example preservatives, stabilisers, wetting agents and/or emulsifiers, solubilisers, salts for the regulation of osmotic pressure, and/or buffers. The present pharmaceutical preparations which, if so desired, may contain further pharmacologically active substances, are prepared in a manner known per se, for example by means of conventional mixing, granulating, coating, dissolving or lyophilising processes, and contain from about 1% to 100%, especially from about 1% to about 20%, of the active substance or substances. In a preferred embodiment, the tablet or capsule contains 50 mg 100 mg of the of the methanesulfonic acid addition salt of a compound of formula I in the F–crystal form, G-crystal form, H-crystal form, I-crystal form or K-crystal form, optionally together with pharmaceutically acceptable carriers.

In one embodiment, the capsule is a hard gelatine capsule containing a dry powder blend. The capsule shell preferably contains gelatine and titanium dioxide as well as red iron oxide. The ratio of weight of capsule fill to capsule shell is preferably between about 100:25 and 100:50, more preferably between 100:30 and 100:40.

In another embodiment, a film coated tablet is used comprising 100 mg, 400 mg or 800 mg drug substance together with inactive excepients selected from colloidal anhydrous silica, polyvinylpyrrolidone, magnesium stearate and microcrystalline cellulose.

The following Examples illustrate the invention without limiting the scope thereof. Temperatures are given in degrees Celsius (°C).

Examples

Example 1: Preparation of crystalline form F of Imatinib Mesylate using Benzyl Alcohol

About 500mg of Imatinib mesylate is first dissolved in about 100ml of water. About 50µl of the stock solution is dispensed manually into a CRISSY 96-well block, to have a total amount of drug substance of 5mg per well. The solution is flushed with nitrogen at room temperature to dry the solution. The dry precipitate is resuspended with 250µl of benzyl alcohol. The suspension or solution is agitated using High-speed vortexer at about 45-55°C for about 2 hrs. The solution is then allowed to evaporate at 45°C to 55°C under a stream of nitrogen.

Example 2: Preparation of crystalline form F of Imatinib Mesylate using a mixture of Benzyl Alcohol and Ethyl Acetate

About 500mg of Imatinib mesylate is first dissolved in about 100ml of water. About 50µl of the stock solution is dispensed manually into a CRISSY 96-well block, to have a total amount of drug substance of 5mg per well. The solution is flushed with nitrogen at room temperature to dry the solution. The dry precipitate is resuspended with a mixture of 222µl of benzyl alcohol and 28µl of ethyl acetate. The suspension or solution is agitated using High-speed vortexer at about 45-55°C for about 2 hrs. The solution is then allowed to evaporate at 45°C to 55°C under a stream of nitrogen.

Example 3: Preparation of crystalline form F of Imatinib Mesylate using a mixture of Benzyl Alcohol and 1,4-Dioxane, 3-Pentanone or Diisopropyl Ether

About 500mg of Imatinib mesylate is first dissolved in about 100ml of water. About 50µl of the stock solution is dispensed manually into a CRISSY 96-well block, to have a total amount of drug substance of 5mg per well. The solution is flushed with nitrogen at room temperature to dry the solution. The dry precipitate is resuspended with a mixture of 214µl of benzyl

alcohol and 36µl of 1,4-dioxane, 3-pentanone or diisopropyl ether. The suspension or solution is agitated using High-speed vortexer at about 45-55°C for about 2 hrs. The solution is then allowed to evaporate at 45°C to 55°C under a stream of nitrogen. The crystals obtained from the mixture consisting of benzyl alcohol/diisopropyl ether. A DSC curve (recorded using a Perkin Elmer DSC-7 instrument with a heating rate of 10K/min and a sample mass of about 0.7mg) shows a melting event at 95°C followed by an exothermic recrystallization and a second melting at about 223°C.

Example 4: Preparation of crystalline form F of Imatinib Mesylate using a mixture of Benzyl Alcohol and Acetonitrile or Dimethyl Formamide

About 500mg of Imatinib mesylate is first dissolved in about 100ml of water. About 50µl of the stock solution is dispensed manually into a CRISSY 96-well block, to have a total amount of drug substance of 5mg per well. The solution is flushed with nitrogen at room temperature to dry the solution. The dry precipitate is resuspended with a mixture of 200µl of benzyl alcohol and 50µl of acetonitrile or dimethyl formamide. The suspension or solution is agitated using High-speed vortexer at about 45-55°C for about 2 hrs. The solution is then allowed to evaporate at 45°C to 55°C under a stream of nitrogen.

Example 5: Tablets with Imatinib mesylate, F-crystal form

Tablets containing 100 mg of the active substance named in the title are usually prepared in the following composition:

Composition

Active ingredient	100 mg
Crystalline lactose	240 mg
Avicel	80 mg
PVPPXL	20 mg
Aerosil	2 mg
Magnesium stearate	5 mg
	---
	447 mg

Preparation: The active substance is mixed with carrier materials and compressed on a tableting machine (Korsch EKO, punch diameter 10 mm).
Avicel is microcrystalline cellulose (FMC, Philadelphia, USA).
PVPPXL is polyvinylpolypyrrolidone, cross-linked (BASF, Germany).
Aerosil is silicon dioxide (Degussa, Germany).

Example 6: Capsules with Imatinib mesylate, F-crystal form

Capsules containing 100 mg of the compound named in the title as active substance are usually prepared in the following composition:

Composition

Active ingredient	100 mg
Avicel	200mg
PVPPXL	15 mg
Aerosil	2 mg
Magnesium stearate	1.5 mg
	318.5 mg

The capsules are prepared by mixing the components and filling the mixture into hard gelatin capsules, size 1.

Example 7: Preparation of crystalline form G of Imatinib Mesylate using a mixture of 3-pentanone and cyclohexane

About 500mg of Imatinib mesylate is first dissolved in about 100ml of water. About 50µl of the stock solution is dispensed manually into a CRISSY 96-well block, to have a total amount of drug substance of 5mg per well. The solution is flushed with nitrogen at room temperature to dry the solution. The dry precipitate is resuspended with a mixture of 125µl of 3-pentanone and 125µl of cyclohexane. The suspension or solution is agitated using High-speed vortexer at about 45-55°C for about 2 hrs. The solution is then allowed to evaporate at 45°C to 55°C under a stream of nitrogen.

Example 8: Tablets with Imatinib mesylate, G-crystal form

Tablets containing 100 mg of the active substance named in the title are usually prepared in the following composition:

Composition	
Active ingredient	100 mg
Crystalline lactose	240 mg
Avicel	80 mg
PVPPXL	20 mg
Aerosil	2 mg
Magnesium stearate	5 mg
	447 mg

Preparation: The active substance is mixed with carrier materials and compressed on a tableting machine (Korsch EKO, punch diameter 10 mm).
Avicel is microcrystalline cellulose (FMC, Philadelphia, USA).
PVPPXL is polyvinylpolypyrrolidone, cross-linked (BASF, Germany).
Aerosil is silicon dioxide (Degussa, Germany).

Example 9: Capsules with Imatinib mesylate, G-crystal form

Capsules containing 100 mg of the compound named in the title as active substance are usually prepared in the following composition:

Composition	
Active ingredient	100 mg
Avicel	200mg
PVPPXL	15 mg
Aerosil	2 mg
Magnesium stearate	1.5 mg
	318.5 mg

The capsules are prepared by mixing the components and filling the mixture into hard gelatin capsules, size 1.

Example 10: Preparation of crystalline form H of Imatinib Mesylate using a mixture of 3-Pentanone and N,N-Dimethylformamide

About 500mg of Imatinib mesylate is first dissolved in about 100ml of water. About 50µl of the stock solution is dispensed manually into a CRISSY 96-well block, to have a total amount of drug substance of 5mg per well. The solution is flushed with nitrogen at room temperature to dry the solution. The dry precipitate is resuspended with a mixture of 125µl of 3-pentanone and 125µl of N,N-dimethylformamide. The suspension or solution is agitated using High-speed vortexer at about 45-55°C for about 2 hrs. The solution is then allowed to evaporate at 45°C to 55°C under a stream of nitrogen.

Example 11: Tablets with Imatinib mesylate, H-crystal form

Tablets containing 100 mg of the active substance named in the title are usually prepared in the following composition:

Composition

Active ingredient	100 mg
Crystalline lactose	240 mg
Avicel	80 mg
PVPPXL	20 mg
Aerosil	2 mg
Magnesium stearate	5 mg
	447 mg

Preparation: The active substance is mixed with carrier materials and compressed on a tableting machine (Korsch EKO, punch diameter 10 mm).
Avicel is microcrystalline cellulose (FMC, Philadelphia, USA).
PVPPXL is polyvinylpolypyrrolidone, cross-linked (BASF, Germany).

Aerosil is silicon dioxide (Degussa, Germany).

Example 12: Capsules with Imatinib mesylate, H-crystal form

Capsules containing 100 mg of the compound named in the title as active substance are usually prepared in the following composition:

Composition	
Active ingredient	100 mg
Avicel	200mg
PVPPXL	15 mg
Aerosil	2 mg
Magnesium stearate	1.5 mg
	318.5 mg

The capsules are prepared by mixing the components and filling the mixture into hard gelatin capsules, size 1.

Example 13: Preparation of crystalline form I of Imatinib Mesylate using a mixture of Ethyl Acetate and Diethyl Ether

About 500mg of Imatinib mesylate is first dissolved in about 100ml of water. About 50µl of the stock solution is dispensed manually into a CRISSY 96-well block, to have a total amount of drug substance of 5mg per well. The solution is flushed with nitrogen at room temperature to dry the solution. The dry precipitate is resuspended with a mixture of 125µl of ethyl acetate and 125µl of diethyl ether. The suspension or solution is agitated using High-speed vortexer at about 45-55°C for about 2 hrs. The solution is then allowed to evaporate at 45°C to 55°C under a stream of nitrogen.

Example 14: Tablets with Imatinib mesylate, I-crystal form

Tablets containing 100 mg of the active substance named in the title are usually prepared in the following composition:

Composition	
Active ingredient	100 mg
Crystalline lactose	240 mg
Avicel	80 mg
PVPPXL	20 mg
Aerosil	2 mg
Magnesium stearate	5 mg
	447 mg

Preparation: The active substance is mixed with carrier materials and compressed on a tableting machine (Korsch EKO, punch diameter 10 mm).
Avicel is microcrystalline cellulose (FMC, Philadelphia, USA).
PVPPXL is polyvinylpolypyrrolidone, cross-linked (BASF, Germany).
Aerosil is silicon dioxide (Degussa, Germany).

Example 15: Capsules with Imatinib mesylate, I-crystal form

Capsules containing 100 mg of the compound named in the title as active substance are usually prepared in the following composition:

Composition	
Active ingredient	100 mg
Avicel	200mg
PVPPXL	15 mg
Aerosil	2 mg
Magnesium stearate	1.5 mg
	318.5 mg

The capsules are prepared by mixing the components and filling the mixture into hard gelatin capsules, size 1.

Example 16: Preparation of crystalline form K of Imatinib Mesylate using a mixture of Ethyl Acetate and N,N-Dimethylformamide

About 500mg of Imatinib mesylate is first dissolved in about 100ml of water. About 50µl of the stock solution is dispensed manually into a CRISSY 96-well block, to have a total amount of drug substance of 5mg per well. The solution is flushed with nitrogen at room temperature to dry the solution. The dry precipitate is resuspended with a mixture of 125µl of 3- ethyl acetate and 125µl of N,N-dimethylformamide. The suspension or solution is agitated using High-speed vortexer at about 45-55°C for about 2 hrs. The solution is then allowed to evaporate at 45°C to 55°C under a stream of nitrogen.

Example 17: Tablets with Imatinib mesylate, K-crystal form

Tablets containing 100 mg of the active substance named in the title are usually prepared in the following composition:

Composition

Active ingredient	100 mg
Crystalline lactose	240 mg
Avicel	80 mg
PVPPXL	20 mg
Aerosil	2 mg
Magnesium stearate	5 mg
	447 mg

Preparation: The active substance is mixed with carrier materials and compressed on a tableting machine (Korsch EKO, punch diameter 10 mm).
Avicel is microcrystalline cellulose (FMC, Philadelphia, USA).
PVPPXL is polyvinylpolypyrrolidone, cross-linked (BASF, Germany).
Aerosil is silicon dioxide (Degussa, Germany).

Example 18: Capsules with Imatinib mesylate, K-crystal form

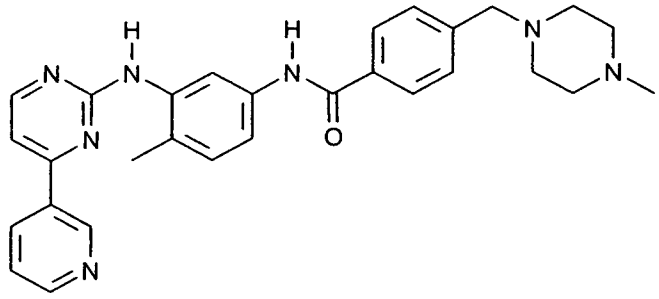
Capsules containing 100 mg of the compound named in the title as active substance are usually prepared in the following composition:

Composition	
Active ingredient	100 mg
Avicel	200mg
PVPPXL	15 mg
Aerosil	2 mg
Magnesium stearate	1.5 mg
	318.5 mg

The capsules are prepared by mixing the components and filling the mixture into hard gelatin capsules, size 1.

What is claimed is:

1. Crystalline form F of the methanesulfonic acid addition salt of a compound of formula I



(I).

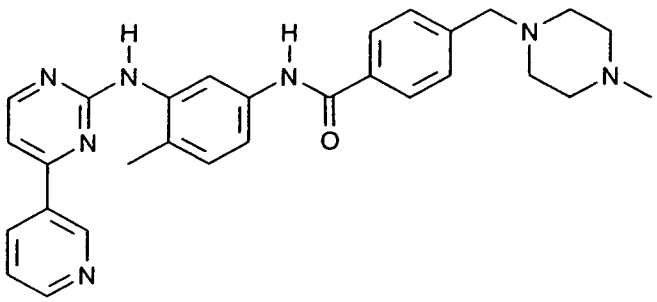
2. The crystalline form according to claim 1, which shows on X-ray diffraction peaks at an angle of refraction 2theta of 8.4° and 8.6°.

3. The crystalline form according to claim 1, which shows on X-ray diffraction peak at an angle of refraction 2theta of 20.9°.

4. The crystalline form of the methanesulfonic acid addition salt of a compound of formula I according to claim 1, which shows in an X-ray diffraction diagram lines at the following angles of refraction 2theta: 8.4°, 8.6°, 13.3°, 16.2°, 16.8°, 17.1°, 19.5°, 20.9°, 23.6° and 24.5°.

5. The crystalline form of the methanesulfonic acid addition salt of a compound of formula I according to any one of claims 1 to 4, which is present in essentially pure form.

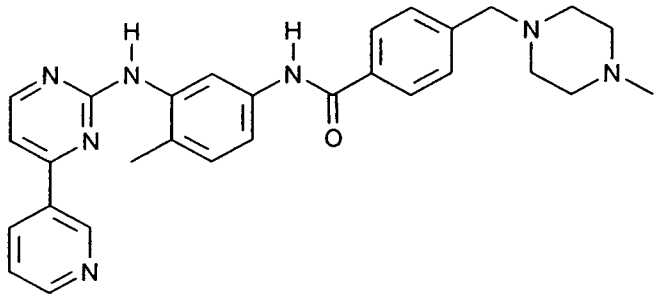
6. A crystalline form of the methanesulfonic acid addition salt of a compound of formula I



(I)

which shows an X-ray diffraction diagram of the type shown in Fig. 1, in which the relative peak intensities of each peak do not deviate by more than 10% from the relative peak intensities in the diagram shown in Fig. 1.

7. A composition containing the methanesulfonic acid addition salt of a compound of formula I



(I)

comprising a crystalline form of the methanesulfonic acid addition salt of a compound of formula I according to any one of the claims 1 to 4.

8. The composition according to claim 7 comprising additionally at least one distinct form of the methanesulfonic acid addition salt of a compound of formula I selected from the amorphous form, the α-crystal form, the β-crystal form and the H1-crystal form.

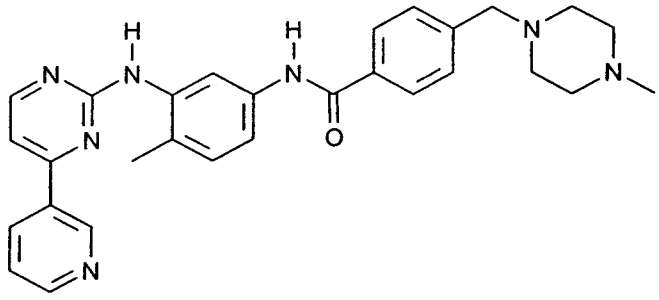
9. A pharmaceutical composition comprising a crystalline form of the methanesulfonic acid addition salt of a compound of formula I according to any one of the claims 1 to 4, optionally together with pharmaceutically acceptable carrier.

10. The pharmaceutical composition according to claim 9 comprising additionally at least one distinct form of the methanesulfonic acid addition salt of a compound of formula I selected from the amorphous form, the α-crystal form, the β-crystal form and the H1-crystal form.

11. The pharmaceutical composition according to claim 9 or 10 comprising between 50 mg and 800 mg of a crystalline form of the methanesulfonic acid addition salt of a compound of formula I according to any one of the claims 1 to 4.

12. The pharmaceutical composition according to any one of claims 9 to 11 which is a capsule containing a dry powder blend comprising between 50 mg and 200 mg of a crystalline form of the methanesulfonic acid addition salt of a compound of formula I according to any one of the claims 1 to 4.

13. A crystalline form of the methanesulfonic acid addition salt of a compound of formula I

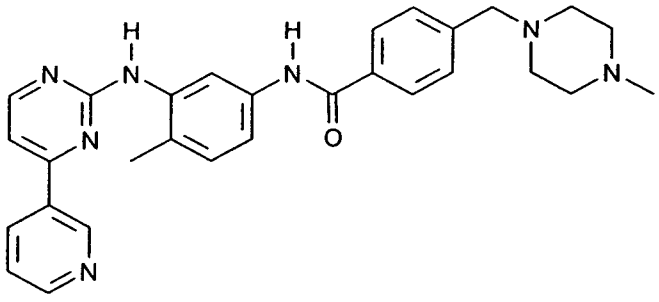


(I)

characterized by a DSC curve showing a melting event at 95°C followed by an exothermic recrystallization and a second melting at about 223°C.

14. A pharmaceutical composition comprising a crystalline form of the methanesulfonic acid addition salt of a compound of formula I according to claim 13 and optionally a pharmaceutically acceptable carrier.

15. Crystalline form G of the methanesulfonic acid addition salt of a compound of formula I



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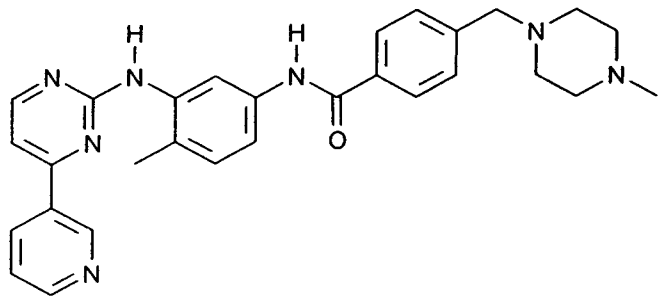
16. The crystalline form according to claim 15, which shows on X-ray diffraction peak at an angle of refraction 2theta of 10.5° in the absence of any lines between an angle of refraction 2theta of 4° and 8°.

17. The crystalline form according to claim 15, which is characterized by X-ray diffraction peaks at an angle of refraction 2theta of 10.5°, 18.1° and 18.7°.

18. The crystalline form of the methanesulfonic acid addition salt of a compound of formula I according to claim 15 or 16, which shows in an X-ray diffraction diagram lines at the following angles of refraction 2theta: 10.5°, 15.0°, 17.2°, 18.1°, 18.7°, 19.2°, 21.1° and 21.3°

19. The crystalline form of the methanesulfonic acid addition salt of a compound of formula I according to any one of claims 15 to 18, which is present in essentially pure form.

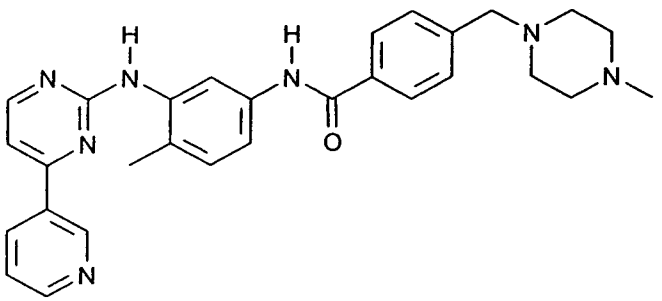
20. A crystalline form of the methanesulfonic acid addition salt of a compound of formula I



(I)

which shows an X-ray diffraction diagram of the type shown in Fig. 2, in which the relative peak intensities of each peak do not deviate by more than 10% from the relative peak intensities in the diagram shown in Fig. 2.

21. A composition containing the methanesulfonic acid addition salt of a compound of formula I



(I)

comprising a crystalline form of the methanesulfonic acid addition salt of a compound of formula I according to any one of the claims 15 to 18.

22. The composition according to claim 21 comprising additionally at least one distinct form of the methanesulfonic acid addition salt of a compound of formula I selected from the amorphous form, the α-crystal form, the β-crystal form and the H1-crystal form.

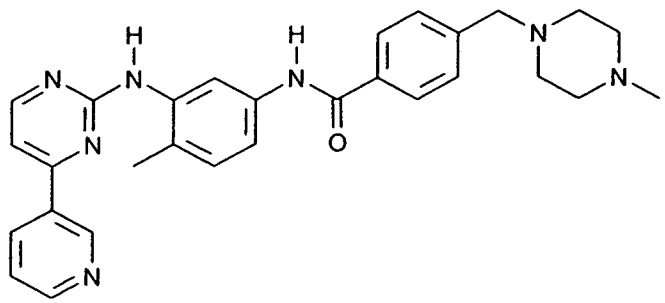
23. A pharmaceutical composition comprising a crystalline form of the methanesulfonic acid addition salt of a compound of formula I according to any one of the claims 15 to 18, optionally together with pharmaceutically acceptable carrier.

24. The pharmaceutical composition according to claim 23 comprising additionally at least one distinct form of the methanesulfonic acid addition salt of a compound of formula I selected from the amorphous form, the α-crystal form, the β-crystal form and the H1-crystal form.

25. The pharmaceutical composition according to claim 23 or 24 comprising between 50 mg and 800 mg of a crystalline form of the methanesulfonic acid addition salt of a compound of formula I according to any one of the claims 15 to 18.

26. The pharmaceutical composition according to any one of claims 23 to 25 which is a capsule containing a dry powder blend comprising between 50 mg and 200 mg of a crystalline form of the methanesulfonic acid addition salt of a compound of formula I according to any one of the claims 15 to 18.

27. Crystalline form H of the methanesulfonic acid addition salt of a compound of formula I



(I).

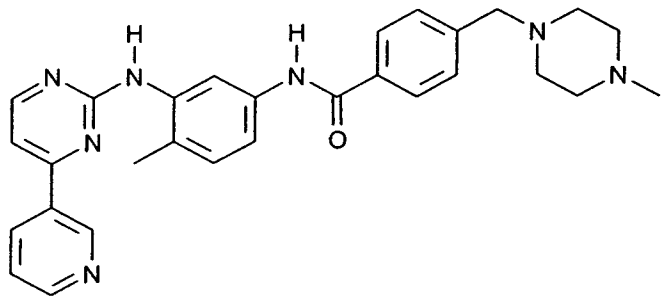
28. The crystalline form according to claim 27, which shows on X-ray diffraction a peak at an angle of refraction 2theta of 32.9°.

29. The crystalline form according to claim 27, which is characterized by an X-ray diffraction peak at an angle of refraction 2theta of 25.1°.

30. The crystalline form of the methanesulfonic acid addition salt of a compound of formula I according to any one of claims 27 to 29, which shows in an X-ray diffraction diagram lines at the following angles of refraction 2theta: 10.5°, 22.8°, 25.1° and 32.9°.

31. The crystalline form of the methanesulfonic acid addition salt of a compound of formula I according to any claim 27 to 30, which is present in essentially pure form.

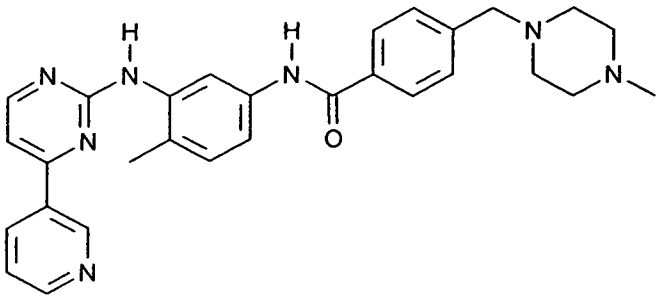
32. A crystalline form of the methanesulfonic acid addition salt of a compound of formula I



(I)

which shows an X-ray diffraction diagram of the type shown in Fig. 3, in which the relative peak intensities of each peak do not deviate by more than 10% from the relative peak intensities in the diagram shown in Fig. 3.

33. A composition containing the methanesulfonic acid addition salt of a compound of formula I



(I)

comprising a crystalline form of the methanesulfonic acid addition salt of a compound of formula I according to any one of the claims 27 to 30.

34. The composition according to claim 33 comprising additionally at least one distinct form of the methanesulfonic acid addition salt of a compound of formula I selected from the amorphous form, the α-crystal form, the β-crystal form and the H1-crystal form.

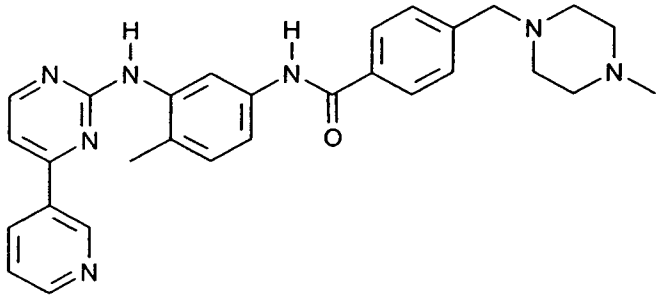
35. A pharmaceutical composition comprising a crystalline form of the methanesulfonic acid addition salt of a compound of formula I according to any one of the claims 27 to 31, optionally together with pharmaceutically acceptable carrier.

36. The pharmaceutical composition according to claim 35 comprising additionally at least one distinct form of the methanesulfonic acid addition salt of a compound of formula I selected from the amorphous form, the α-crystal form, the β-crystal form and the H1-crystal form.

37. The pharmaceutical composition according to claim 35 or 36 comprising between 50 mg and 800 mg of a crystalline form of the methanesulfonic acid addition salt of a compound of formula I according to any one of the claims 27 to 31.

38. The pharmaceutical composition according to any one of claims 35 to 37 which is a capsule containing a dry powder blend comprising between 50 mg and 200 mg of a crystalline form of the methanesulfonic acid addition salt of a compound of formula I according to any one of the claims 27 to 31.

39. Crystalline form I of the methanesulfonic acid addition salt of a compound of formula I



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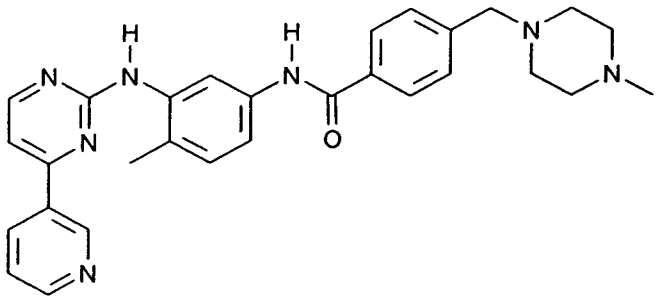
40. The crystalline form according to claim 39, which shows on X-ray diffraction a peak at an angle of refraction 2theta of 12.9° in the absence of any lines at an angle of refraction 2theta below 9° and above 29°.

41. The crystalline form according to claim 39, which is characterized by X-ray diffraction peaks at an angle of refraction 2theta of 12.9°, 17.1°, 20.9°, 23.9° and 24.3°.

42. The crystalline form of the methanesulfonic acid addition salt of a compound of formula I according to claim 39, which shows in an X-ray diffraction diagram lines at the following angles of refraction 2theta: 12.9°, 14.1°, 17.1°, 18.0°, 18.7°, 19.1°, 19.8°, 20.9°, 23.9°, 24.3° and 25.2°.

43. The crystalline form according to any claims 39 to 42, which is present in essentially pure form.

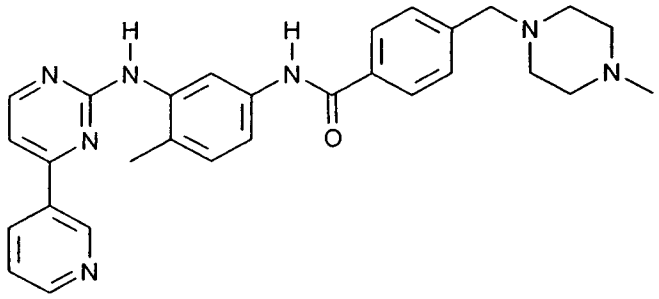
44. A crystalline form of the methanesulfonic acid addition salt of a compound of formula I



(I)

which shows an X-ray diffraction diagram of the type shown in Fig. 4, in which the relative peak intensities of each peak do not deviate by more than 10% from the relative peak intensities in the diagram shown in Fig. 4.

45. A composition containing the methanesulfonic acid addition salt of a compound of formula I



(I)

comprising a crystalline form of the methanesulfonic acid addition salt of a compound of formula I according to any one of the claims 39 to 44.

46. The composition according to claim 45 comprising additionally at least one distinct form of the methanesulfonic acid addition salt of a compound of formula I selected from the amorphous form, the α-crystal form, the β-crystal form and the H1-crystal form.

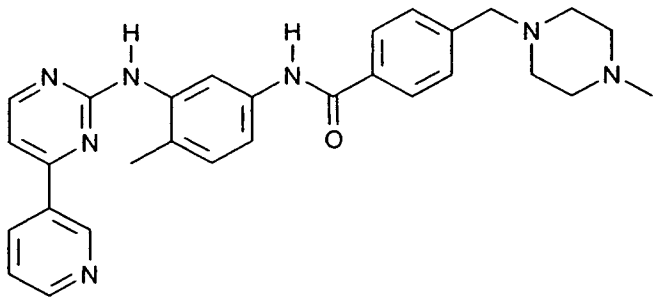
47. A pharmaceutical composition comprising a crystalline form of the methanesulfonic acid addition salt of a compound of formula I according to any one of the claims 39 to 44, optionally together with pharmaceutically acceptable carrier.

48. The pharmaceutical composition according to claim 47 comprising additionally at least one distinct form of the methanesulfonic acid addition salt of a compound of formula I selected from the amorphous form, the α-crystal form, the β-crystal form and the H1-crystal form.

49. The pharmaceutical composition according to claim 47 or 48 comprising between 50 mg and 800 mg of a crystalline form of the methanesulfonic acid addition salt of a compound of formula I according to any one of the claims 39 to 44.

50. The pharmaceutical composition according to any one of claims 47 to 49 which is a capsule containing a dry powder blend comprising between 50 mg and 200 mg of a crystalline form of the methanesulfonic acid addition salt of a compound of formula I according to any one of the claims 39 to 44.

51. Crystalline form K of the methanesulfonic acid addition salt of a compound of formula I



(I).

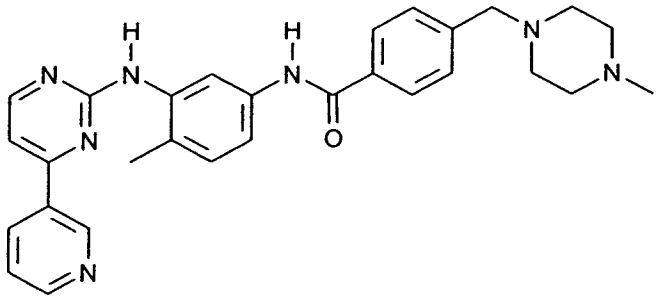
52. The crystalline form according to claim 51, which shows on X-ray diffraction a peak at an angle of refraction 2theta of 37.9°.

53. The crystalline form according to claim 51, which is characterized by an X-ray diffraction peak at an angle of refraction 2theta of 21.0°.

54. The crystalline form of the methanesulfonic acid addition salt of a compound of formula I according to claim 51 or 52, which shows in an X-ray diffraction diagram lines at the following angles of refraction 2theta: 12.1°, 14.1°, 18.2°, 18.4°, 21.0°, 23.4° and 28.4°.

55. The crystalline form according to any claim 51 to 54, which is present in essentially pure form.

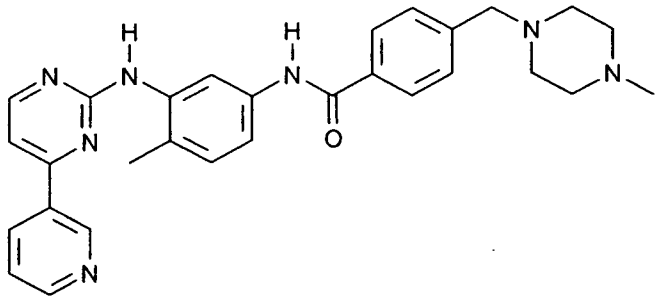
56. A crystalline form of the methanesulfonic acid addition salt of a compound of formula I



(I)

which shows an X-ray diffraction diagram of the type shown in Fig. 5, in which the relative peak intensities of each peak do not deviate by more than 10% from the relative peak intensities in the diagram shown in Fig. 5.

57. A composition containing the methanesulfonic acid addition salt of a compound of formula I



(I)

comprising a crystalline form of the methanesulfonic acid addition salt of a compound of formula I according to any one of the claims 51 to 56.

58. The composition according to claim 57 comprising additionally at least one distinct form of the methanesulfonic acid addition salt of a compound of formula I selected from the amorphous form, the α-crystal form, the β-crystal form and the H1-crystal form.

59. A pharmaceutical composition comprising a crystalline form of the methanesulfonic acid addition salt of a compound of formula I according to any one of the claims 51 to 56, optionally together with pharmaceutically acceptable carrier.

60. The pharmaceutical composition according to claim 59 comprising additionally at least one distinct form of the methanesulfonic acid addition salt of a compound of formula I selected from the amorphous form, the α-crystal form, the β-crystal form and the H1-crystal form.

61. The pharmaceutical composition according to claim 59 or 60 comprising between 50 mg and 800 mg of a crystalline form of the methanesulfonic acid addition salt of a compound of formula I according to any one of the claims 51 to 56.

62. The pharmaceutical composition according to any one of claims 59 to 61 which is a capsule containing a dry powder blend comprising between 50 mg and 200 mg of a crystalline form of the methanesulfonic acid addition salt of a compound of formula I according to any one of the claims 51 to 56.

63. The capsule according to claim 12, 26, 38, 50 or 62, wherein the shell contains gelatine.

64. The capsule according to claim 12, 26, 38, 50 or 62, wherein the shell contains titanium dioxide.

65. The capsule according to claim 12, 26, 38, 50 or 62, wherein the shell contains red iron oxide.

66. The capsule according to claim 12, 26, 38, 50 or 62, wherein the ratio of weight of capsule fill to capsule shell is between about 100:25 and 100:50.

67. The capsule according to claim 12, 26, 38, 50 or 62, wherein the ratio of weight of capsule fill to capsule shell is between 100:30 and 100:40.

68. The pharmaceutical composition according to any one of claims 9 to 11, 23 to 25, 35 to 37, 47 to 49 or 59 to 61, which is a tablet comprising 100 mg, 400 mg or 800 mg drug substance together with inactive excepients.

69. The tablet according to claim 68 wherein the inactive excepients are selected from colloidal anhydrous silica, polyvinylpyrrolidone, magnesium stearate and microcrystalline cellulose.

70. The use of a crystalline form of the methanesulfonic acid addition salt of a compound of formula I according to any one of the claims 1 to 4, 15 to 18, 27 to 30, 39 to 42 or 51 to 54 for the preparation of a medicament for the treatment of a disease selected from metastatic, inoperable GIST, advanced chronic myeloid leukemia, newly diagnosed chronic myeloid leukemia, pediatric Philadelphia chromosome-positive chronic myeloid leukemia, Philadelphia chromosome-positive acute lymphocytic leukemia (ALL), glioblastoma multiforme, dermatofibrosarcoma protuberans (DFSP), hypereosinophilic sindrome (HES), and chronic myelomonocytic leucemia (CMML).

71. Method of treating a disease selected from metastatic, inoperable GIST, advanced chronic myeloid leukemia, newly diagnosed chronic myeloid leukemia, pediatric

Philadelphia chromosome-positive chronic myeloid leukemia, Philadelphia chromosome-positive acute lymphocytic leukemia (ALL), glioblastoma multiforme, dermatofibrosarcoma protuberans (DFSP), hypereosinophilic sindrome (HES), and chronic myelomonocytic leucemia (CMML in a warm-blooded animal in need thereof comprising administering to the animal a crystalline form of the methanesulfonic acid addition salt of a compound of formula I according to any one of the claims 1 to 4, 15 to 18, 27 to 30, 39 to 42 or 51 to 54 in a quantity which is therapeutically effective against the respective disease.

72. The method according to claim 71, wherein a daily dosages of 400 mg or 600 mg is administered orally to the patient.

73. The method according to claim 72, wherein the total daily dosage is administered once daily with a meal and a large glass of water of about 200 mL.

74. The method according to claim 71, wherein a daily dosages of 800 mg is administered orally to the patient.

75. The method according to claim 74, wherein the total daily dosage is administered as 400 mg doses twice daily together with food.

FIG. 1 (F form)

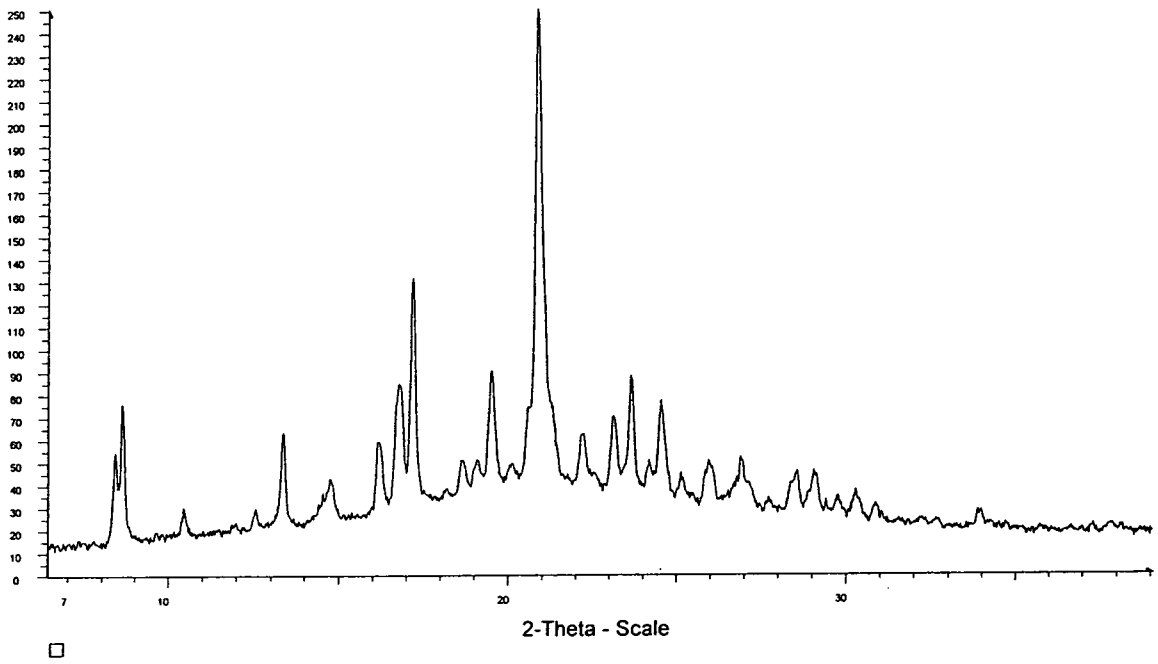

250
240
230
220
210
200
190
180
170
160
150
140
130
120
110
100
90
80
70
60
50
40
30
20
10
0
7
10
20
30
2-Theta - Scale

FIG. 2 (G form)

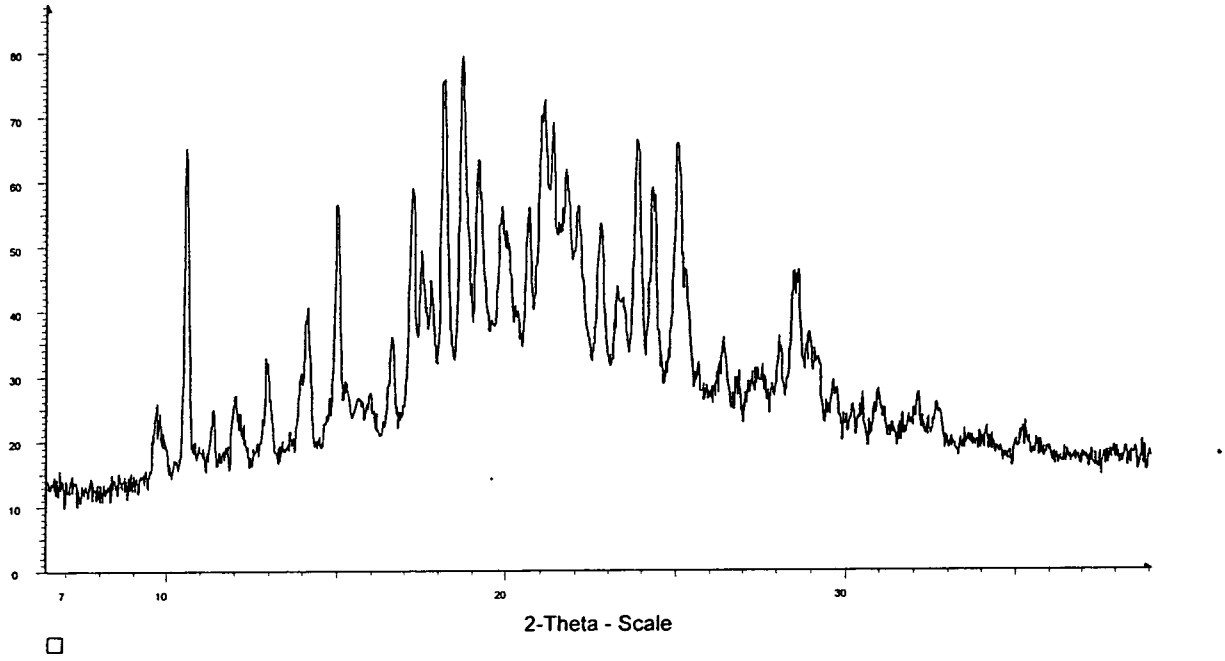


FIG. 3 (H form)

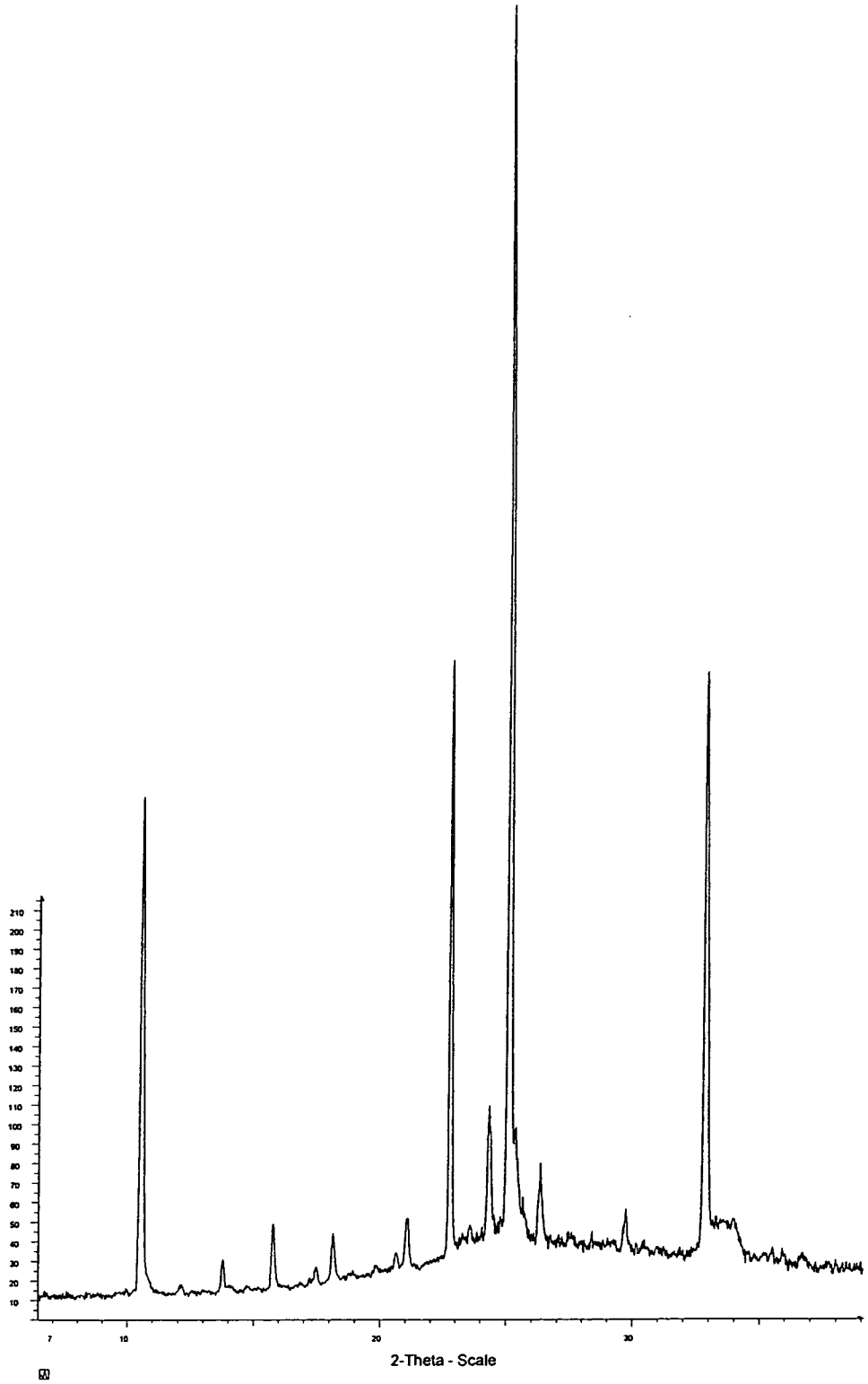


FIG. 4 (I form)

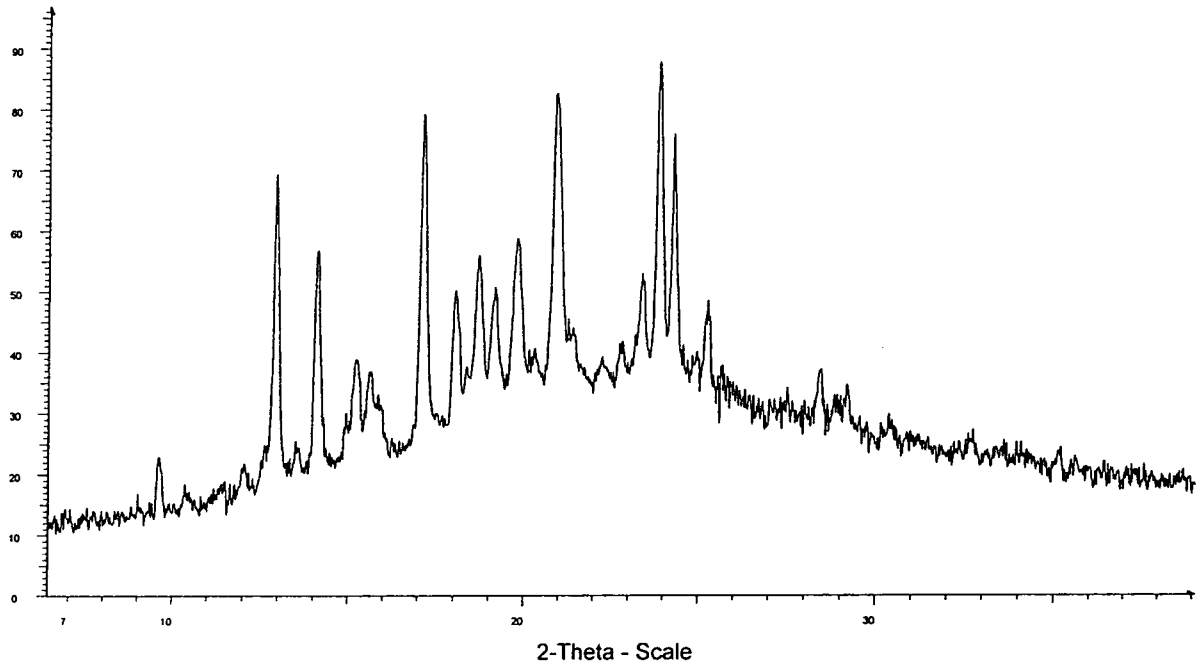
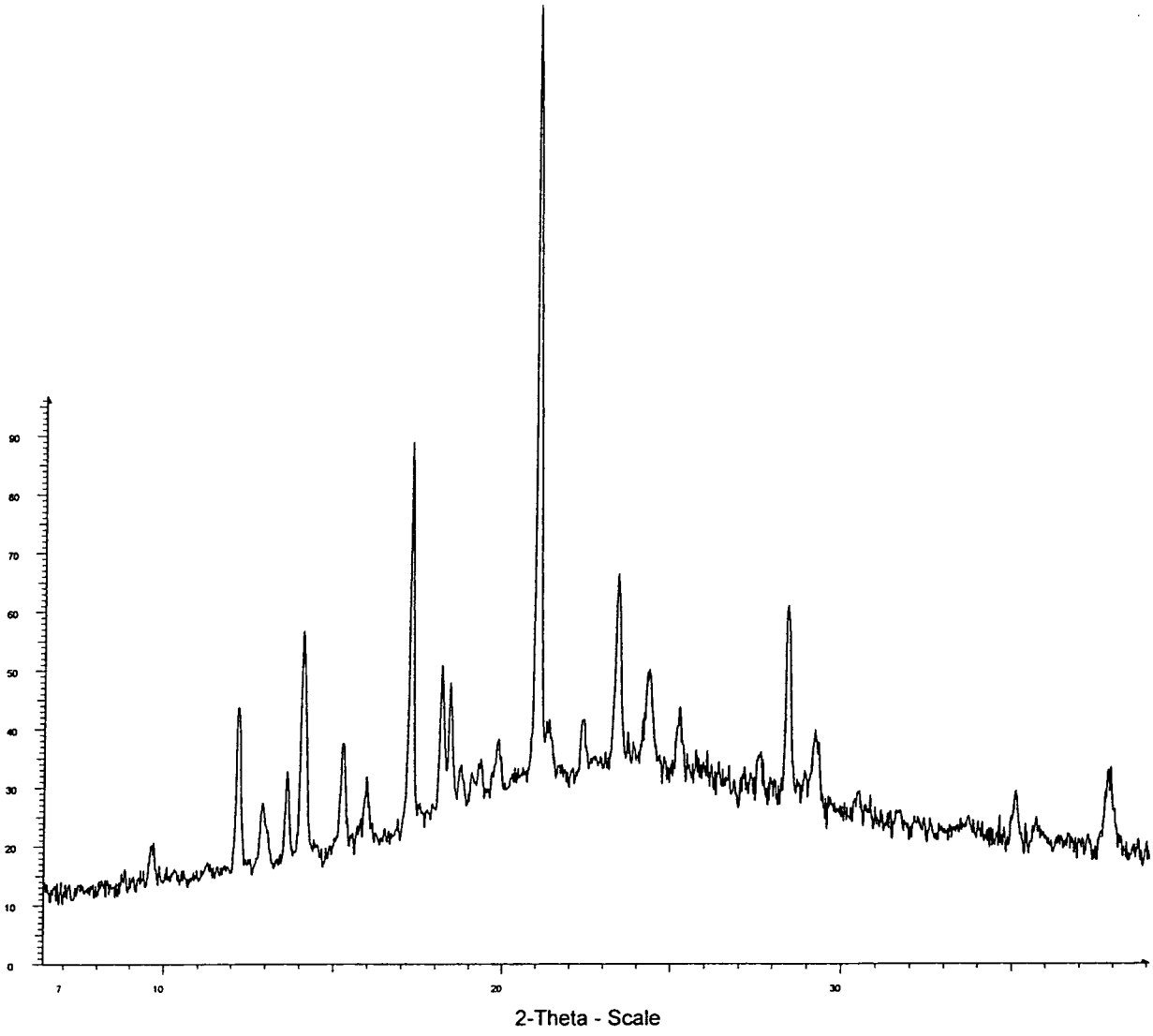


FIG. 5 (K form)



INTERNATIONAL SEARCH REPORT

International application No
PCT/EP2006/011240

A. CLASSIFICATION OF SUBJECT MATTER
INV. C07D401/04 A61K31/506 A61P35/00

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)
C07D A61P A61K

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)
EPO-Internal, CHEM ABS Data

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	WO 2005/077933 A (NATCO PHARMA LTD [IN]; AMALA KOMPELLA [IN]; SRINIVASA RAO THUNGATHURTH) 25 August 2005 (2005-08-25) the whole document	1-75
X	WO 2005/095379 A2 (INST FARMACEUTYCZNY [PL]; SZCZEPEK WOJCIECH [PL]; SAMSON-LAZINSKA DORO) 13 October 2005 (2005-10-13) the whole document	1-75
X	WO 2004/106326 A (HETERO DRUGS LTD [IN]; PARTHASARADHI REDDY BANDI [IN]; RATHNAKAR REDDY) 9 December 2004 (2004-12-09) cited in the application the whole document	1-75

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[X] Further documents are listed in the continuation of Box C.

[X] See patent family annex.

* Special categories of cited documents :

"A" document defining the general state of the art which is not considered to be of particular relevance

"E" earlier document but published on or after the international filing date

"L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)

"O" document referring to an oral disclosure, use, exhibition or other means

"P" document published prior to the international filing date but later than the priority date claimed

"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention

"X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone

"Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.

"&" document member of the same patent family

Date of the actual completion of the international search	Date of mailing of the international search report
17 April 2007	03/05/2007

Name and mailing address of the ISA/	Authorized officer
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INTERNATIONAL SEARCH REPORT

International application No
PCT/EP2006/011240

C(Continuation). DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	WO 99/03854 A (NOVARTIS AG [CH]; NOVARTIS ERFIND VERWALT GMBH [AT]; ZIMMERMANN JUERG) 28 January 1999 (1999-01-28) cited in the application the whole document	1-75
P,X	WO 2006/054314 A (NATCO PHARMA LTD [IN]; KOMPELLA AMALA KISHAN [IN]; ADIBHATLA KALI SATY) 26 May 2006 (2006-05-26) the whole document	1-75
P,X	WO 2006/048890 A (SUN PHARMACEUTICAL IND LTD [IN]; PATEL HETALKUMAR VIRENDRABHAI [IN]; J) 11 May 2006 (2006-05-11) the whole document	1-75
P,X	WO 2006/024863 A (CIPLA LTD [IN]; WAIN CHRISTOPHER PAUL [GB]; PATHI SRINIVAS LAXMINARAYA) 9 March 2006 (2006-03-09) the whole document	1-75
E	WO 2007/023182 A (NOVARTIS AG [CH]; NOVARTIS PHARMA GMBH [AT]; MUTZ MICHAEL [DE]) 1 March 2007 (2007-03-01) the whole document	1-75

INTERNATIONAL SEARCH REPORT

International application No.
PCT/EP2006/011240

Box II Observations where certain claims were found unsearchable (Continuation of item 2 of first sheet)

This International Search Report has not been established in respect of certain claims under Article 17(2)(a) for the following reasons:

1. [X] Claims Nos.:
because they relate to subject matter not required to be searched by this Authority, namely:

Although claims 71-75 are directed to a method of treatment of the human/animal body, the search has been carried out and based on the alleged effects of the compound/composition.

2. [ ] Claims Nos.:
because they relate to parts of the International Application that do not comply with the prescribed requirements to such an extent that no meaningful International Search can be carried out, specifically:

3. [ ] Claims Nos.:
because they are dependent claims and are not drafted in accordance with the second and third sentences of Rule 6.4(a).

Box III Observations where unity of invention is lacking (Continuation of item 3 of first sheet)

This International Searching Authority found multiple inventions in this International application, as follows:

1. [ ] As all required additional search fees were timely paid by the applicant, this International Search Report covers all searchable claims.

2. [ ] As all searchable claims could be searched without effort justifying an additional fee, this Authority did not invite payment of any additional fee.

3. [ ] As only some of the required additional search fees were timely paid by the applicant, this International Search Report covers only those claims for which fees were paid, specifically claims Nos.:

4. [ ] No required additional search fees were timely paid by the applicant. Consequently, this International Search Report is restricted to the invention first mentioned in the claims; it is covered by claims Nos.:

Remark on Protest

[ ] The additional search fees were accompanied by the applicant's protest.

[ ] No protest accompanied the payment of additional search fees.

INTERNATIONAL SEARCH REPORT

Information on patent family members

International application No
PCT/EP2006/011240

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