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(54) **Title:** METHODS FOR GRAPHENE-ASSISTED FABRICATION OF MICRO- AND NANOSCALE STRUCTURES AND DEVICES FEATURING THE SAME

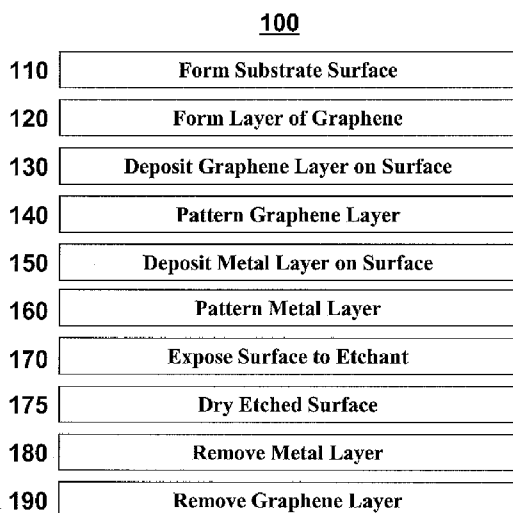


Figure 1

(57) **Abstract:** Methods for graphene-assisted fabrication of a surface on a substrate are disclosed herein. In an exemplary method, fabricating an etched surface on a substrate includes, depositing at least one layer of graphene on the surface on the substrate, patterning the deposited layer of graphene, and exposing the surface on a substrate to an acid to etch the surface on the substrate. The method can further include forming the layer of graphene from graphite. In some embodiments, the layer of graphene is formed by mechanically exfoliating the layer of graphene from the graphite. Alternatively, the layer of graphene can be formed by chemically exfoliating the graphene from the graphite, or other carbon materials, and/or utilizing vapor deposition to form the layer of graphene from the graphite, or other carbon materials.

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METHODS FOR GRAPHENE-ASSISTED FABRICATION OF MICRO- AND
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CROSS REFERENCE TO RELATED APPLICATIONS

5 This application claims priority to U.S. Provisional Application No.
61/186,577, entitled "Graphene-Assisted Fabrication Of Nanoscale Structures", filed
on June 12, 2009 and U.S. Provisional Application No. 61/118,919, entitled "Three-
Terminal Device Using Mechanically-Vibrating, High Transconductance Material",
filed on December 1, 2008, each of which is incorporated by reference in its entirety
10 herein.

STATEMENT REGARDING FEDERALLY-SPONSORED RESEARCH

 This invention was made with government support under CHE-
0641523 awarded by the National Science Foundation. The government has certain
15 rights in the invention.

BACKGROUND

FIELD

 The present application relates to methods for fabricating micro- and
20 nanoscale structures and devices featuring such fabricated structures.

BACKGROUND ART

 Silicon dioxide (SiO₂) thermally grown on top of silicon is an
important insulating material in the electronic industry. Many applications call for
25 the patterned removal (e.g., etching) of silicon dioxide to form a variety of devices.
The removal of silicon dioxide can be accomplished using a hydrofluoric acid (HF)
etch that quickly dissolves the silicon dioxide, while leaving the silicon intact.

 Further, many applications in the micro- and nano-electronic fields call
for fabrication of devices with parts that are unsupported by the silicon substrate, e.g.,
30 suspended above the substrate. Examples of such devices would include nano-
electromechanical switches, resonators and mass sensors. However, such devices are

often difficult to fabricate, as material needs to be from under the device surface (e.g., the electrodes, etc.), while maintaining the structural integrity of the device.

Another challenge in small scale fabrication is the production of closed or covered cavities in a substrate surface layer such as silicon dioxide. Such cavities can serve as, for example, nanoscale reservoirs in nanofluidic devices. Accordingly, there is a need in the art for techniques for to fabricate micro- and nanoscale structures on the surfaces of substrates.

SUMMARY

10 Methods for graphene-assisted fabrication of a surface on a substrate are disclosed herein. In an exemplary method, fabricating an etched surface on a substrate includes, depositing at least one layer of graphene on the surface on the substrate, patterning the deposited layer of graphene, and exposing the surface on a substrate to an acid to etch the surface on the substrate. The method can further include forming the layer of graphene from graphite. In some embodiments, the layer of graphene is formed by mechanically exfoliating the layer of graphene from the graphite. Alternatively, the layer of graphene can be formed by chemically exfoliating the graphene from the graphite, or other carbon materials, and/or utilizing vapor deposition to form the layer of graphene from the graphite, or other carbon materials.

The method can also include depositing at least one layer of metal on top of the deposited at least one layer of graphene, where the layer of metal leaves at least one portion of an edge of the layer of graphene exposed. In some embodiments, the surface on the substrate can be silicon dioxide and the acid can be hydrofluoric acid and the layer of metal can be gold. In the same or another embodiment, patterning the deposited layer of graphene includes utilizing lithography to pattern the deposited layer of graphene. Patterning the deposited layer of graphene can also include oxygen plasma etching to pattern the deposited layer of graphene.

In some exemplary embodiments exposing the surface of substrate to an acid includes acid vapor phase etching the surface and, in the same or yet other embodiments, exposing the surface to an acid includes exposing the surface to a buffered oxide etchant, which can be, e.g., diluted with water.

The disclosed subject matter further includes a graphene fabricated device including at least one layer of graphene partially suspended above a surface on a substrate. This graphene fabricated device can also include at least one layer of metal deposited on the layer of graphene. in some embodiments the layer of metal is
5 gold.

A graphene fabricated device including at least one channel etched into a silicon substrate beneath at least one flake of graphene is also disclosed herein. In some embodiments, this graphene fabricated device can also include at least one layer of metal deposited on the at least one layer of graphene, such that the at least
10 one layer of metal substantially covers the layer of graphene leaving at least one portion of an edge of the at least one layer of graphene exposed. The channel can also be a nanoscale channel.

A graphene fabricated device including at least one buried cavity etched into a surface on a substrate beneath at least one flake of graphene is further
15 described herein. In some embodiments, the graphene fabricated device further includes at least one layer of metal deposited on the layer of graphene, such that the layer of metal substantially covers the layer of graphene leaving at least one portion of an edge of the layer of graphene exposed and covering the remainder of the layer of graphene, and where the exposed portion of the layer of graphene has a first width
20 and the covered portion of the layer of graphene has a second width, the first width being less than the second width.

BRIEF DESCRIPTION OF THE DRAWINGS

The accompanying drawings, which are incorporated and constitute
25 part of this disclosure, illustrate some embodiments of the disclosed subject matter.

FIG. 1 illustrates a method for graphene-assisted fabrication of a substrate surface in accordance with an exemplary embodiment of the disclosed subject matter.

FIGS. 2(a)-(b) illustrate a substrate surface fabricated to illustrate the
30 etching of the substrate surface beneath a layer of graphene in accordance with an exemplary embodiment of the disclosed subject matter.

FIG. 2(c) is a graph showing the depths of etching that occurs beneath certain layers of graphene in accordance with an exemplary embodiment of the disclosed subject matter.

5 FIG. 2(d) illustrates a vapor phase etching chamber used to fabricate a substrate surface in accordance with an exemplary embodiment of the disclosed subject matter.

FIGS. 3(a)-(c) illustrate a substrate surface fabricated to illustrate the impermeability of graphene in accordance with an exemplary embodiment of the disclosed subject matter.

10 FIGS. 3(d)-(f) illustrate a substrate surface fabricated to create a channel in the substrate surface in accordance with an exemplary embodiment of the disclosed subject matter.

FIGS. 4(a)-(b) illustrate a substrate surface fabricated to create a device suspended above the substrate in accordance with an exemplary embodiment of the disclosed subject matter.

15 FIGS. 5(a)-(d) illustrate a substrate surface fabricated to create a cavity in the substrate surface in accordance with an exemplary embodiment of the disclosed subject matter.

20 Throughout the figures and specification the same reference numerals are used to indicate similar features and/or structures.

DETAILED DESCRIPTION

The techniques described herein are useful for etching a substrate surface with the assistance of graphene. Although the description is focused on 25 examples utilizing a silicon dioxide substrate surface, the techniques herein can also be useful for etching other substrate surfaces, such as, e.g., quartz, polysilicon, silicon and silicon nitride

The subjected matter disclosed herein provides methods for graphene-assisted etching of substrate surfaces and devices formed using such methods. The 30 techniques described herein make use of the recent discovery of graphene, a carbon allotrope comprising of a hexagonal lattice of sp^2 -hybridized carbon atoms. One property of graphene is that it allows for the etching of silicon dioxide from underneath a graphene layer placed on the silicon. Further, the etching process is

accelerated along the SiO₂-graphene interface and it has also been found, as detailed below, that graphene is not permeable to most etchants, such as hydrofluoric acid, and may not be permeable to anything, even helium. Accordingly, a layer of graphene is deposited on a substrate surface to facilitate etching of that surface in areas that would otherwise be unreachable. A layer of gold can be deposited on the graphene and further on portions of the surrounding substrate surface. When the substrate surface is exposed to an etchant it will be etched where it is exposed directly to the etchant and where the etchant can enter underneath the graphene by means of an edge of graphene being exposed to the etchant, but the surface will not be substantially etched beneath the edge of the gold layer exposed to the etchant.

Figure 1 illustrates a method 100 for graphene-assisted fabrication of a substrate surface. Method 100 can include forming 110 the substrate surface, or in some embodiments, the substrate surface can be provided fully formed 110. The substrate surface can be formed 110 by any suitable method known in the art, depending on the composition of the substrate surface and the substrate base upon which it is formed 110. In one exemplary embodiment, the substrate surface is composed of silicon dioxide (SiO₂) and is formed 110 on a silicon substrate base. For the example of silicon dioxide, the substrate surface can be formed 110 thermally growing the silicon dioxide on the silicon substrate base. Other techniques such as vapor deposition are also within the scope of the disclosed subject matter.

As further illustrated in Figure 1, method 100 can further include forming 120 one or more graphene layers. The one or more graphene layers, or "flakes", can be formed 120 by any process known in the art for forming graphene flakes, e.g., by mechanical exfoliation. The mechanical exfoliation process includes using adhesive tape, e.g., Scotch® tape, to remove graphene flakes from a stock of graphite. Other techniques for forming graphene layers, such as chemical exfoliation of graphite or vapor deposition, can also be used to form 120 the one or more graphene flakes. For example, in one embodiment utilizing a chemical exfoliation technique, the graphene can be formed 120 by chemically splitting individual sheets of graphene off of a graphite stock. Such a process can be performed in solution and then the solution can be sprayed onto the surface on the substrate, thus depositing 130 the graphene layers. The solution can then be dried leaving one or more layers of graphene deposited 130 on the substrate surface.

The graphene layers are then deposited **130** onto the substrate surface, which can be performed, in some embodiments, by laying the graphene layers on the substrate surface. In the embodiment where the graphene is formed **120** using adhesive tape, the graphene can be deposited **130** on the substrate surface by
5 transferring the flake directly from the adhesive side of the tape to the surface, e.g., placing the adhesive side of the tape, with the flake adhered to it, directly on the substrate surface and then removing the tape from the substrate surface.

In other embodiments, the graphene can be deposited **130** on the surface on the substrate during its formation **120**. For example, where the graphene
10 layers are formed **120** by vapor deposition, such layers can be formed **120** and deposited **130** onto the substrate surface in one step. In one such embodiment utilizing a chemical vapor deposition technique, the graphene layers can formed **120** by growing the graphene on the surface of a metal, such as nickel or copper, then chemically dissolving the metal transferring **130** the graphene to the substrate surface.
15 Transferring **130** the graphene to the substrate surface can be accomplished by adhering the grown **120** graphene to an adhesive transfer material, chemically dissolving the metal, and then depositing **130** the graphene onto the substrate surface, e.g., essentially stamping the graphene onto the surface.

The method **100** further includes patterning **140** the graphene layers
20 into desired shapes, useful for particular applications, e.g., micro- or nano-electromechanical devices, such as micro-mirrors, accelerometers, switches, Fabir-Pero cavities, resonators, mass sensors, force sensors, etc. Patterning **140** the graphene layers can be performed either before or after the deposition **130** of the graphene layers onto the substrate surface. Further, the graphene layers can be
25 patterned **140** utilizing any appropriate technique known in the art. For example, the graphene flakes can be patterned **140** by oxygen plasma etching, e.g., for 6 seconds at 50 W and 200 milliTorr (mT) through a mask in an electron beam resist (PMMA 950k), to create the required shapes for a given application. The graphene layers can also be patterned **140** utilizing lithographic techniques, such as photolithography or
30 other kinds of lithography.

In some embodiments, the method **100** further includes depositing **150** one or more layers of metal onto the substrate surface. The metal layer can be deposited **150** utilizing any appropriate technique known in the art. For example, the

metal layers can be deposited **150** using metal evaporation, e.g., in a vacuum using either electron beam assisted or thermally assisted, sputtering, or electrodeposition. The metal used for the metal layer will depend on the desired application and in some embodiments is gold, aluminum, copper, titanium or other metals used in semiconductor processing, as is known in the art. In other embodiments, the gold layer is deposited **150** onto a layer of another metal, such as chromium, for adhesive purposes. The gold layer can be deposited **150** at an appropriate thickness for the particular application, e.g., 30-150 nm thick gold layer deposited **150** onto a 1-10 nm thick layer of chromium. In one embodiment a 100 nm thick gold layer was deposited **150** onto a 1 nm thick layer of chromium. In other embodiments, a 30-150 nm thick gold layer can be deposited **150** directly onto the substrate surface. Either before, during or after being deposited **150**, the one or more metal layers can be patterned **160** utilizing any appropriate technique known in the art, such as electron beam or optical lithography or other lithographic techniques, to create the desired shapes for a particular application.

The method **100** further includes exposing **170** the substrate surface to an etchant capable of removing at least portions of the substrate surface. Depending on the composition of the surface, the etchant can be an acid. In an exemplary embodiment where the substrate surface is composed of silicon dioxide the etchant can be hydrofluoric acid. Other etchants can also be effective at etching **170** the substrate surface depending on the composition of the substrate surface. For example, where the substrate surface is composed of silicon, it can be etched **170** using potassium hydroxide. Exposing **170** the substrate surface to an etchant can be performed using a vapor phase technique which, in one exemplary embodiment, employs a chamber containing a heating element for receiving and controlling the temperature of the substrate surface, and a container for holding the etchant, e.g., a beaker holding hydrofluoric acid. In one embodiment, the heating element heats a silicon dioxide substrate surface to 60° C which can produce an etching rate of about 1 nm/min.

In another exemplary embodiment, the silicon substrate surface can be exposed **170** to an etchant in liquid form to etch the silicon dioxide. The liquid can be, for example, a buffered oxide etchant such as hydrofluoric acid diluted with water at a concentration of, e.g., 50:1. The substrate surface can be exposed **170** to the

liquid etchant for a period of time sufficient to remove the desired amount of the surface. In one example, a silicon dioxide substrate surface was exposed **170** to hydrofluoric acid diluted to 50:1 for 15 minutes to remove 150 nm of silicon dioxide.

After etching **170** the metal layer can be removed **180** utilizing any appropriate technique known in the art, such as etching **180** with aqua regia or a specifically formulated gold etchant (e.g., Transene TFA). Similarly, the graphene layer can also be removed **190** utilizing any appropriate technique known in the art, such as oxygen plasma etching. Further in one exemplary embodiment, after etching **170** the fabricated device can be dried **175** utilizing any technique known in the art that prevents the collapse of the suspended graphene or graphene and metal structures. For example, in an embodiment utilizing a buffered oxide etch **170** a critical point drying **175** technique utilizing ethanol can be used to reduce the probability of collapse of the suspended structure due to surface tension of the drying and/or etching liquid.

Figures 2(a)-(c) illustrate a silicon dioxide substrate surface fabricated in accordance with an exemplary method **100**. Figure 2(a) is an image of a multilayer graphene flake **201**, deposited **130** on to a SiO₂ substrate surface **202**, from which graphene layers, **1L**, **2L** and **3L**, were patterned **140** into circular shapes using oxygen plasma etching (e.g., for 6 seconds at 50 W and 200 milliTorr (mT)). Graphene layer **1L**, as illustrated in Figure 2(a), is a single layer (a monolayer) of graphene, graphene layer **2L** is a double layer, and graphene layer **3L** is a triple layer. Figure 2(a) further illustrates, in the inset **203**, a magnified image of layer **1L**. Figure 2(b) is an atomic force microscope (AFM) image of the substrate **202** in Figure 2(a) after exposure **170** to hydrofluoric acid, showing the etched cavity **209** centered at point **A**, where the graphene layer was located. As illustrated in Figure 2(b), the area under the graphene layer centered on point **A** was etched significantly more than the surrounding areas, such as at point **B**, which were not covered with a layer of graphene. Figure 2(c) illustrates the cross-sectional depth of the removed silicon dioxide **202** at each of the sites **1L**, **2L** and **3L**. As illustrated in Figures 2(b) and 2(c), exposure **170** to the hydrofluoric acid remove more of the silicon dioxide substrate surface **202** underneath the graphene than in the surrounding areas of the surface. Further illustrated in Figure 2(c), the depth of etching under single layer graphene **1L** was larger than under double layer **2L**, which itself was deeper than the

etch under triple layer 3L. Such exemplary results are consistent with the theory that hydrofluoric acid diffuses very fast across the SiO₂-graphene interface and then etches down at a normal rate, e.g., the rate it would etch SiO₂ without the presence of graphene. As a result, etching 170 proceeds as if it were catalyzed by graphene.

5 Figure 2(d) illustrates an exemplary vapor etching chamber 204 containing a heating element 205 for receiving a substrate 206 having a surface 202 and a container 207 for holding an etchant 208, e.g., hydrofluoric acid. In one exemplary embodiment, heating element 205 heats the substrate surface to 50-60° C and the substrate surface 202 is composed of silicon dioxide having at least one
10 graphene layer deposited thereon.

Figure 3(a) illustrates a silicon dioxide substrate surface 202 that has a layer of graphene 301 deposited 130 onto it followed by a layer of gold 302 deposited 150 on the graphene layer such that the gold layer 302 covers the entire edge of the graphene layer 301 but leaves the central portion of the graphene 301 exposed.

15 Figure 3(b) is an AFM image of the silicon dioxide surface 202 of Figure 3(a) after exposure 170 to an etchant, e.g., hydrofluoric acid. As shown in Figure 3(b), the hydrofluoric acid etched away the silicon dioxide surrounding the gold layer 302 but did not etch the silicon dioxide underneath the gold layer 302 or the exposed graphene 301. Thus, the hydrofluoric acid does not permeate the graphene layer 301.

20 Figure 3(c) is an illustration of the etched substrate surface 202 of Figure 3(b) showing the graphene layer 301 with the overlaying gold masking layer 302, all located on the substrate base 206.

Figure 3(d) illustrates a silicon dioxide substrate surface 202 that has a layer of graphene 301 deposited 130 onto it followed by a layer of gold 302 deposited
25 150 on the graphene layer 301 such that the gold layer 302 cover almost all of the graphene layer 301, except a portion 303, having length ΔL_{Gr} , which can be, e.g., 1-50 μm and in some embodiments 1-30 μm , and further is less than length L_{Gr} . The graphene layer 301 has a length L_{Gr} , which can be, e.g., 1-50 μm and in some embodiments 1-30 μm , and a width W_{Gr} , which can be, e.g., 1-50 μm and in some
30 embodiments 1-30 μm , and in one embodiment is less than length L_{Gr} . The gold layer 302 has a length L_{Au} , which can be, e.g., 1-50 μm and in some embodiments 1-30 μm and is either shorter than length L_{Gr} , by at least ΔL_{Gr} , or is deposited 150 offset by length of at least ΔL_{Gr} , and has a width W_{Au} , which can be, e.g., 1-50 μm

and in some embodiments 1-30 μm and in the same or another embodiment is wider than graphene width W_{Gr} such that both edges along length of the graphene layer **301** are covered by the gold layer **302**.

Figure 3(e) is an AFM image of the silicon dioxide surface **202** of
5 Figure 3(d) after exposure **170** to an etchant, e.g., hydrofluoric acid. As shown in Figure 3(e), the hydrofluoric acid etched away the silicon dioxide **202** surrounding the gold layer **302** as well as the silicon dioxide **202** underneath the graphene layer **301**. Further, Figure 3(e) illustrates that the hydrofluoric acid etches **170** for approximately the length of the graphene layer that was covered by the gold layer,
10 forming a sloped channel **304** having depth H_{C} , e.g., 200 nm deep at the entry point **A** and sloping up to the original surface of the silicon dioxide **202**, length L_{C} , e.g., approximately 30 μm long and width W_{C} , e.g., 2 μm wide. Figure 3(f) is an illustration of the configuration of Figure 3(e), showing the gold capping layer **302** on top of the graphene layer **301** and the channel **304** that was etched **170** out from
15 underneath the graphene **301**. Further, Figure 3(f) is also an illustration of a graphene fabricated device **305** having a channel **304** etched **170** into the silicon dioxide substrate surface **202** beneath a layer of graphene **301**. The graphene fabricated device **305** in Figure 3(f) further includes a layer of metal **302**, e.g., gold, covering substantially all of the layer of graphene **301**, but leaving a portion **303** (as shown in
20 Figure 3(d)) exposed.

Figure 4(a) illustrates a silicon dioxide substrate surface **202** having a graphene layer and gold layer deposited **130**, **150** before and after etching **170**. Figure 4(b) is a scanning electron microscope (SEM) image of a silicon substrate **206** prepared in the manner illustrated in Figure 4(a). As shown in Figure 4(b), etching
25 **170** of the silicon dioxide substrate surface removes a portion of the SiO_2 beneath the graphene **301** and gold layers **302**, forming a suspended device **401** on the base silicon substrate **206**. Figures 4(a)-(b) further illustrate a graphene fabricated device **401** which includes at least one layer of graphene **301** suspended above a silicon substrate **206**. As illustrated in Figure 4(a), the suspended graphene fabricated device
30 **401** can further include at least one layer of metal **302**, e.g., gold, deposited on the graphene **301**.

Figure 5(a) illustrates a silicon dioxide substrate surface **202** having a gold layer **302** deposited **150** on top of a layer of graphene **301** deposited **130** on the

SiO₂ surface 202. In this example the graphene layer 301 has been patterned 140 into a circular shape having a tail ending with an exposed portion 303 ending at point A, which is not covered by the gold layer 302, as illustrated in Figure 5(a). As illustrated in Figure 5(a), the layer of graphene can be patterned 140 such that it has a first

5 portion 502, having a first width W_1 , e.g., 1-50 μm and in some embodiments 1-30 μm , and length L_{Gr1} , e.g., 1-50 μm and in some embodiments 1-30 μm , and it has a second portion 503, having a second width W_2 , e.g., 1-50 μm and in some

embodiments 1-30 μm . The first portion 502 extends from underneath the metal layer 302 to include exposed portion 303, which extends from under the metal layer

10 302 by length ΔL_{Gr} , which can be, e.g., 1-50 μm and in some embodiments 1-30 μm . In one exemplary embodiment, the first width W_1 of the first portion 502, and correspondingly the width W_1 of the exposed portion 303, is smaller than the width W_2 of the second portion (e.g., the covered portion) 503.

Exposed portion 303 has a length ΔL_{Gr} , which can be, e.g., 1-50 μm

15 and in some embodiments 1-30 μm . After etching 170 the SiO₂ surface 202, the resulting silicon dioxide substrate surface 202 appears as illustrated in Figure 5(b), which is a AFM image of the substrate 202 after etching 170 and removal 180, 190 of the graphene and gold layers. Figures 5(c)-(d) are 3D renderings of the processes illustrated in Figures 5(a)-(b), with Figure 5(c) illustrating the layers of graphene 301 and gold 302 deposited 130, 150 on the SiO₂ surface 202 and Figure 5(d) illustrates

20 the same configuration after etching 170, showing the removal of a cavity 501 of SiO₂ beneath the graphene and gold layers 301, 302.

Figures 5(b) and 5(d) further illustrates a graphene fabricated device 500 which includes a cavity 501 etched into the substrate surface 202 beneath at least

25 one flake of graphene. As illustrated in Figure 5(d), the cavity 501 can include a first portion 504, having a first width W_1 , e.g., 1-50 μm and in some embodiments 1-30 μm , and length L_{Gr2} , e.g., 1-50 μm and in some embodiments 1-30 μm , and it has a second portion 505, having a second width W_2 , e.g., 1-50 μm and in some

embodiments 1-30 μm . In one embodiment the length L_{Gr2} of the first portion 504

30 can be $(L_{Gr1} - \Delta L_{Gr})$, e.g., the length of the first portion 502 of the graphene layer 301 minus the length ΔL_{Gr} of the exposed portion 303. As illustrated in Figure 5(d), the graphene fabricated device 500 can further include at least one layer of metal 302, e.g., gold, deposited 150 on the layer of graphene 301, such that substantially all of

the graphene layer 301 is covered except for at least one edge of the first portion that is left exposed. In addition, though not illustrated in Figure 5(d), the graphene layer 301 can remain in place, e.g., suspended, from the layer of metal 302 after exposure 170 to an etchant.

5 It will be understood that the foregoing is only illustrative of the principles described herein, and that various modifications can be made by those skilled in the art without departing from the scope and spirit of the disclosed subject matter. For example, the methods described herein are used for etching silicon
10 dioxide. It is understood that that techniques described herein are useful for other materials that can be etched utilizing graphene. Moreover, features of embodiments described herein can be combined and/or rearranged to create new embodiments.

CLAIMS

We claim:

1. A method for fabricating an etched surface on a substrate comprising:
depositing at least one layer of graphene on the surface on the substrate;
5 patterning the deposited at least one layer of graphene; and
exposing the surface on the substrate to an acid to etch the surface on the substrate.
2. The method of claim 1, further comprising forming the at least one layer of graphene from graphite.
- 10 3. The method of claim 2, wherein forming the at least one layer of graphene comprises mechanically exfoliating the layer of graphene from the graphite.
4. The method of claim 2, wherein forming the at least one layer of graphene comprises chemically exfoliating the graphene from the graphite.
- 15 5. The method of claim 2, wherein forming the at least one layer of graphene comprises utilizing vapor deposition to form the layer of graphene from the graphite.
6. The method of claim 1, further comprising depositing at least one layer of metal on top of the deposited at least one layer of graphene, wherein the at least one layer of metal leaves at least one portion of an edge of the at least
20 one layer of graphene exposed.
7. The method of claim 1, wherein the surface on the substrate comprises silicon dioxide.
8. The method of claim 1, wherein the acid comprises hydrofluoric acid.
9. The method of claim 6, wherein the layer of metal comprises a layer of gold.

10. The method of claim 1, wherein patterning the deposited at least one layer of graphene comprises utilizing lithography to pattern the deposited at least one layer of graphene.
- 5 11. The method of claim 1, wherein patterning the deposited at least one layer of graphene comprises oxygen plasma etching to pattern the deposited at least one layer of graphene.
12. The method of claim 1, wherein exposing the surface on the substrate to an acid comprises acid vapor phase etching the surface on the substrate.
- 10 13. The method of claim 1, wherein exposing the surface on the substrate to an acid comprises exposing the surface on the substrate to a buffered oxide etchant.
14. A graphene fabricated device comprising at least one layer of graphene partially suspended above a substrate.
- 15 15. The graphene fabricated device of claim 14, further comprising at least one layer of metal deposited on the at least one layer of graphene
16. The graphene fabricated device of claim 15, wherein the at least one layer of metal comprises gold.
17. A graphene fabricated device comprising at least one channel etched into a surface on a substrate beneath at least one flake of graphene.
- 20 18. The graphene fabricated device of claim 17, further comprising at least one layer of metal deposited on the at least one layer of graphene, wherein the at least one layer of metal substantially covers the at least one layer of graphene leaving at least one portion of an edge of the at least one layer of graphene exposed.
- 25 19. The graphene fabricated device of claim 17, where the at least one channel comprises a nanoscale channel.

20. A graphene fabricated device comprising at least one cavity etched into a surface on a substrate beneath at least one flake of graphene.

21. The graphene fabricated device of claim 20, further comprising at least one layer of metal deposited on the at least one layer of graphene,

5 wherein the at least one layer of metal substantially covers the at least one layer of graphene leaving at least one portion of an edge of the at least one layer of graphene exposed and covering the remainder of the at least one layer of graphene, and

10 wherein the exposed portion of the at least one layer of graphene has a first width and the covered portion of the at least one layer of graphene has a second width, the first width being less than the second width.

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110	Form Substrate Surface
120	Form Layer of Graphene
130	Deposit Graphene Layer on Surface
140	Pattern Graphene Layer
150	Deposit Metal Layer on Surface
160	Pattern Metal Layer
170	Expose Surface to Etchant
175	Dry Etched Surface
180	Remove Metal Layer
190	Remove Graphene Layer

Figure 1

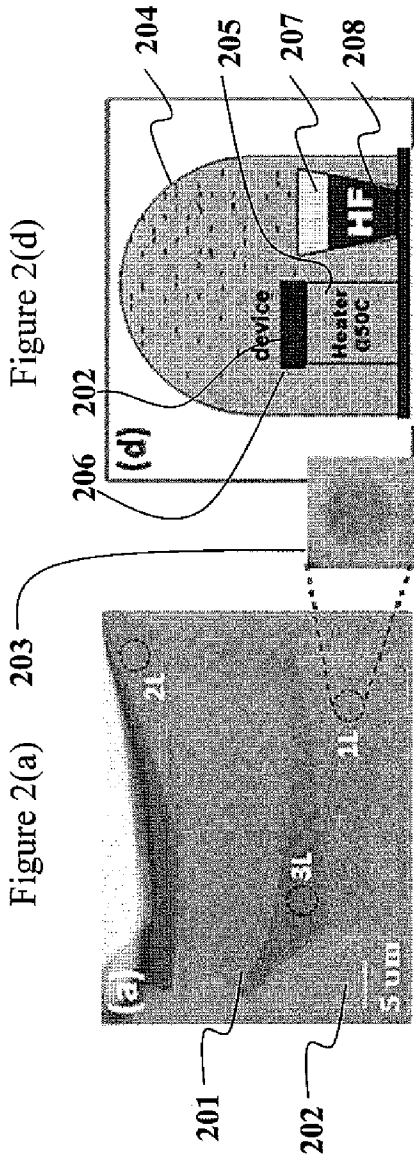


Figure 2(d)

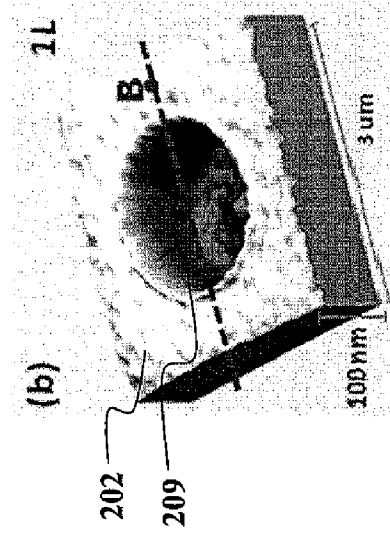
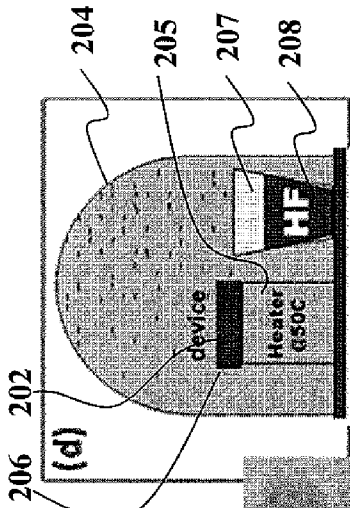


Figure 2(b)

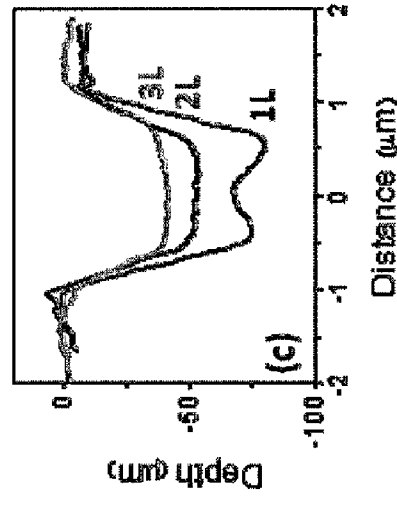


Figure 2(c)

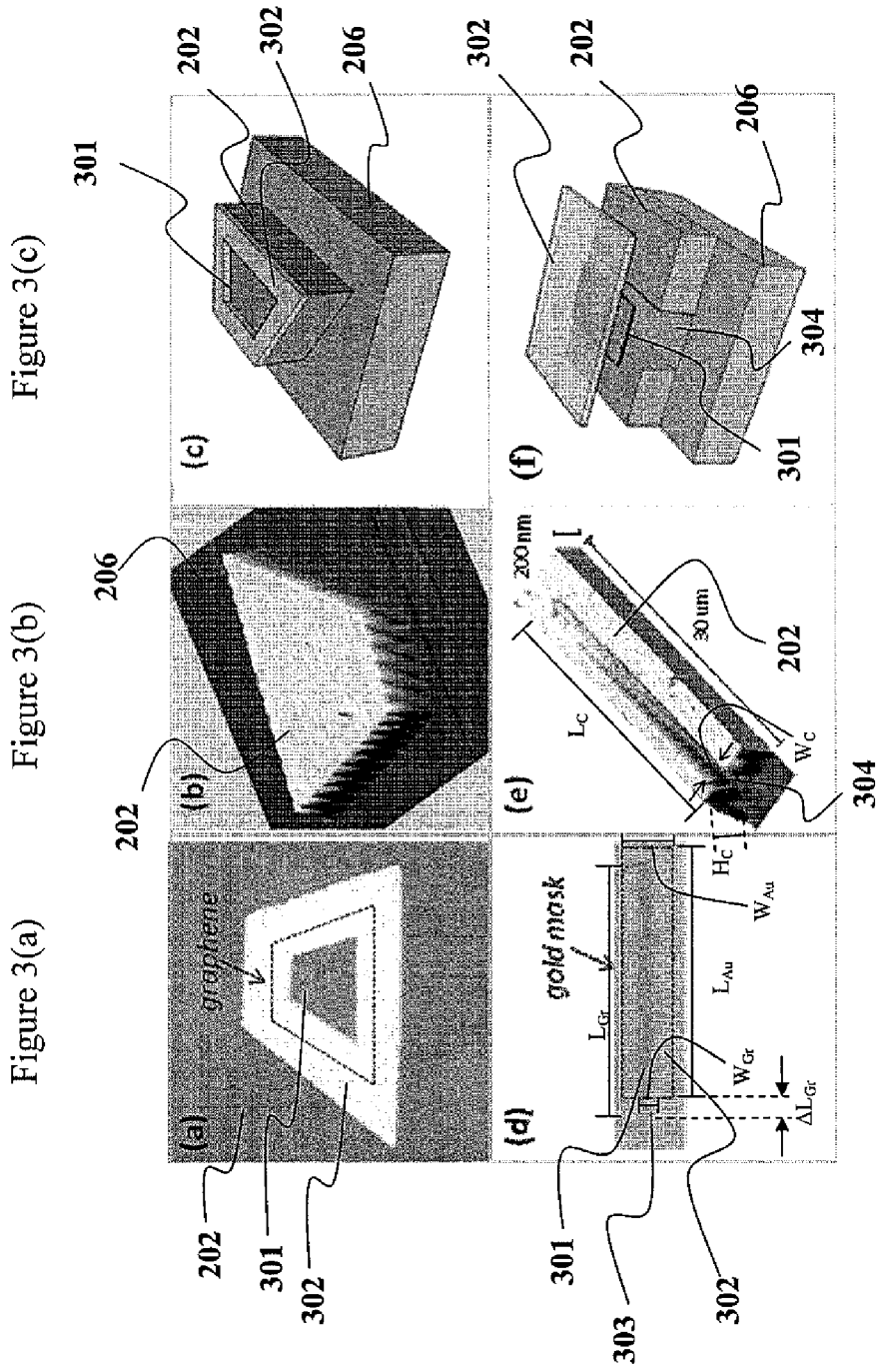


Figure 3(c)

Figure 3(b)

Figure 3(a)

Figure 3(f)

Figure 3(e)

Figure 3(d)

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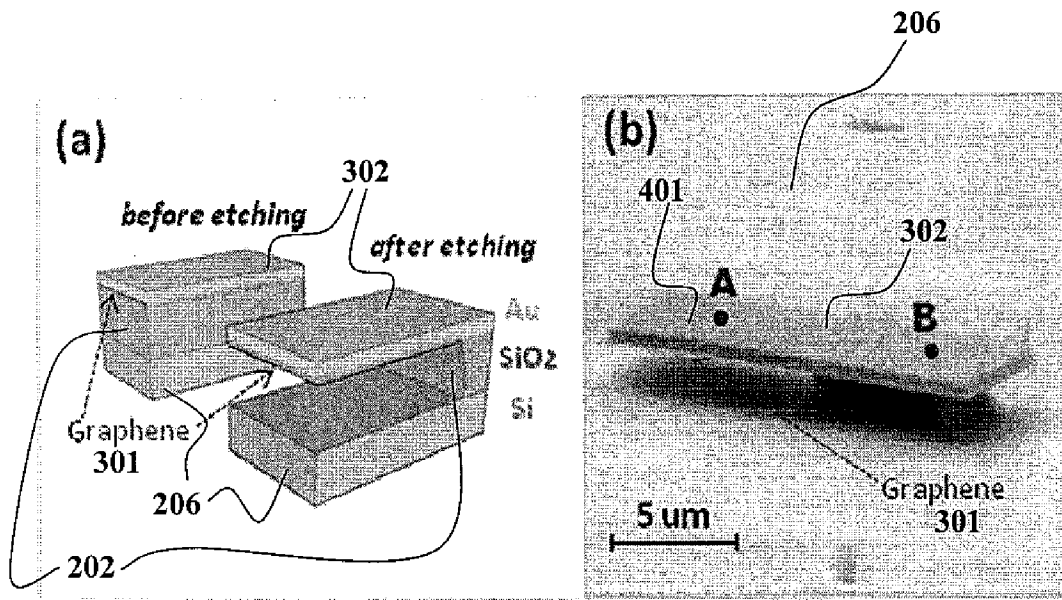


Figure 4(a)

Figure 4(b)

Figure 5(a)

Figure 5(b)

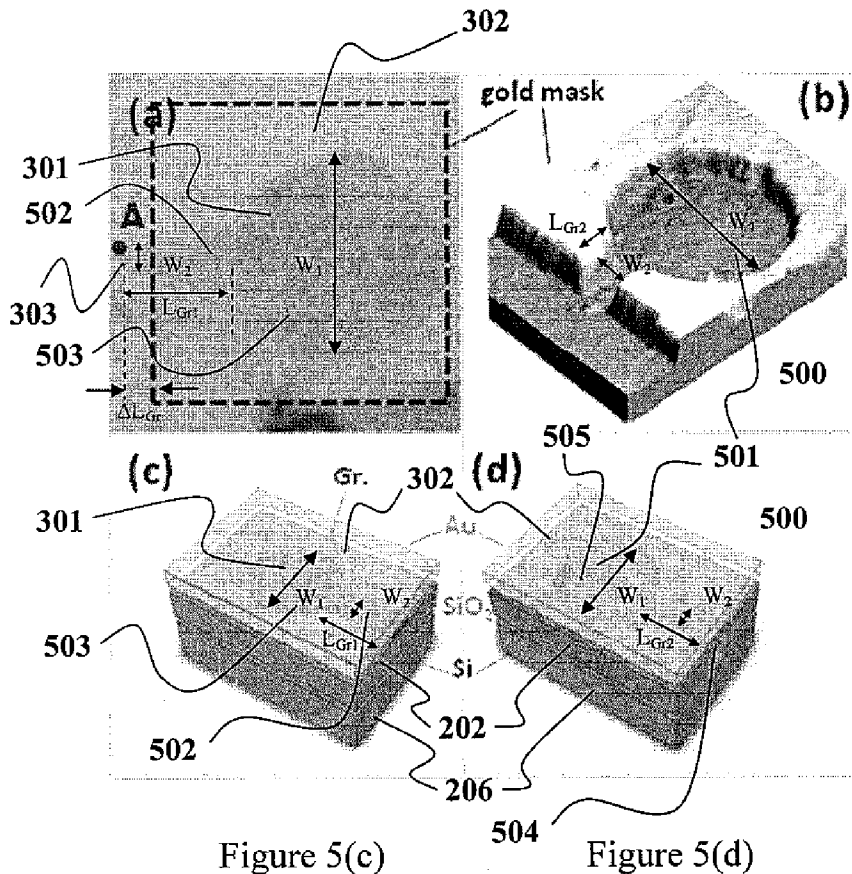


Figure 5(c)

Figure 5(d)

INTERNATIONAL SEARCH REPORT

International application No.

PCT/US 09/66220

A. CLASSIFICATION OF SUBJECT MATTER IPC(8) - G11C 11/18; H01L 43/06 (2010.01) USPC - 365/170; 257/421; 257/9 According to International Patent Classification (IPC) or to both national classification and IPC																
B. FIELDS SEARCHED Minimum documentation searched (classification system followed by classification symbols) IPC(8) - G11C 11/18; H01L 43/06 (2010.01) USPC - 365/170; 257/421; 257/9 Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched USPC - 365/170; 257/421; 257/9 (keyword search) Electronic data base consulted during the international search (name of data base and, where practicable, search terms used) PubWEST, WIPO, Web: Google, Google Scholar Search Terms Used: graphene, graphite, patterning, etching, exfoliate, peeling, mechanical, chemical, SiO ₂ , HF, gold, suspended, floating, channel, nanoscale, oxide, etchant, lithography, oxygen plasma																
C. DOCUMENTS CONSIDERED TO BE RELEVANT																
<table border="1"> <thead> <tr> <th>Category*</th> <th>Citation of document, with indication, where appropriate, of the relevant passages</th> <th>Relevant to claim No.</th> </tr> </thead> <tbody> <tr> <td>X</td> <td>Bolotin et al., "Ultrahigh electron mobility in suspended graphene," Solid State Communications Vol 146, pg 351-355, published on-line on 06 March 2008 (06.03.2008), Entire document, especially: pg 351, col 2; pg 352; Fig 1 [online]. Retrieved from the Internet on [11 January 2010] Retrieved from: <URL: http://pico.phys.columbia.edu/pdf_papers/SSC_146_2008_KB.pdf</td> <td>14-17, 19 ----- 1-13, 18, 20, 21</td> </tr> <tr> <td>Y</td> <td>Bunch, "Mechanical and electrical properties of graphene sheets," PhD Dissertation, Cornell University, May 2008 (05.2008), Entire document, especially: pg 40, para [03]; pg 41, para [01]; pg 42, para [01]-[03]; pg 44, para [01], [03]; pg 65, para [01]; Fig 5.1A, pg 66; pg 86, para [01]; Fig 6.5, pg 95 [online]. Retrieved from the Internet on [11 January 2010] <URL: www.lassp.cornell.edu/lassp_data/mceuen/homepage/.../bunch_thesis.pdf</td> <td>1-13, 20, 21</td> </tr> <tr> <td>Y</td> <td>Stolyarova et al., "High-resolution scanning tunneling microscopy imaging of mesoscopic graphene sheets on an insulating surface," Proceedings of the National Academy of Sciences Vol 104, No 22, pg 9209-9212, published on-line on 29 May 2007 (29.05.2007), Entire document, especially: Fig 1b, pg 9210 [online]. Retrieved from the Internet on [11 January 2010] Retrieved from: <URL: http://www.pnas.org/content/104/22/9209.abstract</td> <td>6, 9, 18, 21</td> </tr> <tr> <td>Y</td> <td>Schultz et al., "Synthesis of linked carbon monolayers: Films, balloons, tubes, and pleated sheets," Proceedings of the National Academy of Sciences Vol 105, No 21, pg 7353-7358, published on-line on 27 May 2008 (27.05.2008), Entire document, especially: pg 7354, col 2; pg 7355, col 1; pg 7536, col 2 [online]. Retrieved from the Internet on [14 January 2010]</td> <td>8, 12</td> </tr> </tbody> </table>	Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.	X	Bolotin et al., "Ultrahigh electron mobility in suspended graphene," Solid State Communications Vol 146, pg 351-355, published on-line on 06 March 2008 (06.03.2008), Entire document, especially: pg 351, col 2; pg 352; Fig 1 [online]. Retrieved from the Internet on [11 January 2010] Retrieved from: <URL: http://pico.phys.columbia.edu/pdf_papers/SSC_146_2008_KB.pdf	14-17, 19 ----- 1-13, 18, 20, 21	Y	Bunch, "Mechanical and electrical properties of graphene sheets," PhD Dissertation, Cornell University, May 2008 (05.2008), Entire document, especially: pg 40, para [03]; pg 41, para [01]; pg 42, para [01]-[03]; pg 44, para [01], [03]; pg 65, para [01]; Fig 5.1A, pg 66; pg 86, para [01]; Fig 6.5, pg 95 [online]. Retrieved from the Internet on [11 January 2010] <URL: www.lassp.cornell.edu/lassp_data/mceuen/homepage/.../bunch_thesis.pdf	1-13, 20, 21	Y	Stolyarova et al., "High-resolution scanning tunneling microscopy imaging of mesoscopic graphene sheets on an insulating surface," Proceedings of the National Academy of Sciences Vol 104, No 22, pg 9209-9212, published on-line on 29 May 2007 (29.05.2007), Entire document, especially: Fig 1b, pg 9210 [online]. Retrieved from the Internet on [11 January 2010] Retrieved from: <URL: http://www.pnas.org/content/104/22/9209.abstract	6, 9, 18, 21	Y	Schultz et al., "Synthesis of linked carbon monolayers: Films, balloons, tubes, and pleated sheets," Proceedings of the National Academy of Sciences Vol 105, No 21, pg 7353-7358, published on-line on 27 May 2008 (27.05.2008), Entire document, especially: pg 7354, col 2; pg 7355, col 1; pg 7536, col 2 [online]. Retrieved from the Internet on [14 January 2010]	8, 12	<input type="checkbox"/> Further documents are listed in the continuation of Box C.
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* Special categories of cited documents: "A" document defining the general state of the art which is not considered to be of particular relevance "E" earlier application or patent but published on or after the international filing date "L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified) "O" document referring to an oral disclosure, use, exhibition or other means "P" document published prior to the international filing date but later than the priority date claimed	"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention "X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone "Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art "&" document member of the same patent family															
Date of the actual completion of the international search 16 January 2010 (16.01.2010)	Date of mailing of the international search report 03 FEB 2010															
Name and mailing address of the ISA/US Mail Stop PCT, Attn: ISA/US, Commissioner for Patents P.O. Box 1450, Alexandria, Virginia 22313-1450 Facsimile No. 571-273-3201	Authorized officer: Lee W. Young PCT Helpdesk: 571-272-4300 PCT OSP: 571-272-7774															