

(19) World Intellectual Property Organization  
International Bureau(43) International Publication Date  
28 June 2007 (28.06.2007)

PCT

(10) International Publication Number  
WO 2007/072162 A1(51) International Patent Classification:  
*C09D 5/24* (2006.01)    *C09D 191/00* (2006.01)  
*C09D 11/00* (2006.01)

(74) Agents: SPOOR &amp; FISHER et al.; P O Box 454, 0001 Pretoria (ZA).

(21) International Application Number:  
PCT/IB2006/003666

(81) Designated States (unless otherwise indicated, for every kind of national protection available): AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, GT, HN, HR, HU, ID, IL, IN, IS, JP, KE, KG, KM, KN, KP, KR, KZ, LA, LC, LK, LR, LS, LT, LU, LV, LY, MA, MD, MG, MK, MN, MW, MX, MY, MZ, NA, NG, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RS, RU, SC, SD, SE, SG, SK, SL, SM, SV, SY, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, ZA, ZM, ZW.

(22) International Filing Date:  
18 December 2006 (18.12.2006)

(84) Designated States (unless otherwise indicated, for every kind of regional protection available): ARIPO (BW, GH, GM, KE, LS, MW, MZ, NA, SD, SL, SZ, TZ, UG, ZM, ZW), Eurasian (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European (AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IS, IT, LT, LU, LV, MC, NL, PL, PT, RO, SE, SI, SK, TR), OAPI (BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG).

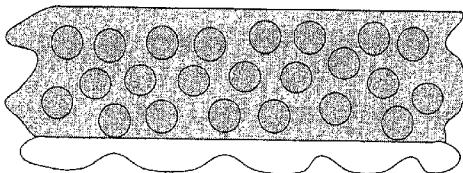
(25) Filing Language: English  
(26) Publication Language: English  
(30) Priority Data:  
2005/10436    22 December 2005 (22.12.2005) ZA

## Published:

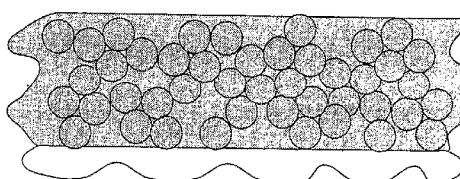
- with international search report
- before the expiration of the time limit for amending the claims and to be republished in the event of receipt of amendments

[Continued on next page]

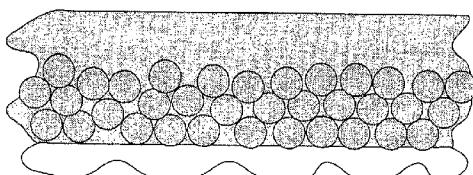
(54) Title: THICK FILM SEMICONDUCTING INKS



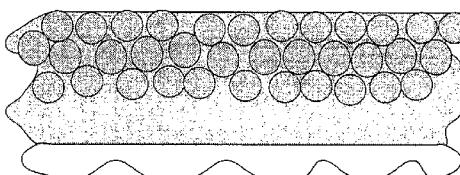
a



b



c



d

(57) Abstract: A method of producing a printable composition comprises mixing a quantity of particulate semiconductor material with a quantity of a binder. The semiconductor material is typically nanoparticulate silicon with a particle size in the range from 5 nanometres to 10 microns. The binder is a self-polymerising material comprising a natural oil, or a derivative or synthetic analogue thereof. Preferably the binder comprises a natural polymer formed by auto-polymerisation of a precursor consisting of a natural oil, or its derivatives including pure unsaturated fatty acids, mono- and di-glycerides, or methyl and ethyl esters of the corresponding fatty acids. The method may include applying the printable composition to a substrate, in single or multiple layers, and allowing the printable composition to cure to define the component or conductor on the substrate.

WO 2007/072162 A1



*For two-letter codes and other abbreviations, refer to the "Guidance Notes on Codes and Abbreviations" appearing at the beginning of each regular issue of the PCT Gazette.*

## THICK FILM SEMICONDUCTING INKS

### BACKGROUND OF THE INVENTION

THIS invention relates to printable compositions for use in electronic and electrical applications in general, and specifically in those applications where semiconducting properties are required.

Semiconductor nanoparticles, with a characteristic size of a few nanometres up to several hundred nanometres, are a widely studied type of material, in which size effects dominate over properties of the bulk material. In general, depending on the specific material and its application, three different size-related phenomena may change the electronic, optical, thermal and mechanical properties of such nanoparticles:

1. a different structure and composition compared to the known bulk phases;
2. the higher surface to volume ratio of the particles compared to the bulk phase, which causes surface states and processes to dominate; and
3. quantum confinement effects when the size of the object is similar to or smaller than the wavelength and coherence length of a fundamental excitation (electronic state, optical wavelength or phonon excitation).

Various semiconductor inks using such particles have been disclosed. For example, organic semiconducting inks comprising semiconducting particles

dispersed randomly in a matrix or binder, and inorganic semiconducting inks in which the semiconducting particles form an interconnecting structure, are known.

In such applications, the binder or matrix material is often a polymer which is soluble in an easily obtainable solvent, such as acetone, chloroform or toluene. Common examples are polystyrene and cellulose acetate butyrate (CAB), which are insulators, and polythiophenes which are conjugated hole conductors, allowing charge transport between isolated particles.

It is an object of the invention to provide alternative printable compositions comprising semiconductor particles.

#### **SUMMARY OF THE INVENTION**

According to the invention there is provided a method of producing a printable composition comprising mixing a quantity of particulate semiconductor material with a quantity of a binder, wherein the binder is a self-polymerising material comprising a natural oil, or a derivative or synthetic analogue thereof.

The binder may comprise a natural polymer formed by auto-polymerisation of a precursor consisting of a natural oil, or its derivatives including pure unsaturated fatty acids, mono- and di-glycerides, or methyl and ethyl esters of the corresponding fatty acids.

The binder may comprise a drying or semi-drying oil, or a mixture thereof.

The drying oil may contain one or more octadecanoic acids, octadecadienoic acids, and/or octadecatrienoic acids.

Preferably, the drying oil is linseed or tung oil.

In the case of a semidrying oil, the oil may be soy-bean, cottonseed or castor oil, for example.

The binder may be an octadecadienoic acid or an octadecatrienoic acid, or a mixture thereof.

For example, the binder may be linolenic acid or linoleic acid.

The method may include mixing the binder with a solvent, such as ethanol, acetone or lacquer thinners.

The composition may be applied to the substrate as a lacquer, comprising the binder and the particles, or may be applied as a varnish, comprising the binder, the particles, and a solvent.

Preferably, the volume ratio of the particulate semiconductor material to the binder is greater than 50%, and more preferably greater than 80%.

The particulate semiconductor material may have a particle size in the range from 5 nanometres to 10 microns, and preferably in the range from 50 to 500 nanometres.

In a typical example, the size range may be from 100 to 300 nanometres.

In a preferred embodiment of the method, the particulate semiconductor material comprises silicon.

Further according to the invention there is provided a method of producing an electronic component or conductor comprising preparing a printable composition as defined above, applying the printable composition to a substrate, and allowing the printable composition to cure to define the component or conductor on the substrate.

The printable composition may be applied in a single application or in multiple layers to define electronic components with desired characteristics.

Preferably, the printable composition is allowed to cure under ambient conditions.

The substrate may be rigid or flexible, and may comprise, for example, metal, glass, plastics, and paper.

The printable composition may be applied in a layer having a thickness in the range from 0.1 to 500 microns.

Typically, the layer has a thickness in the region of 100 microns.

The invention extends to a printable composition, and to an electronic component or conductor, produced by the respective method.

#### BRIEF DESCRIPTION OF THE DRAWINGS

**Figure 1(a)** is a schematic sectional diagram illustrating dispersed semiconductor particles in a binder layer deposited on a substrate, according to a first embodiment of the invention;

**Figure 1(b)** is a schematic sectional diagram similar to that of Figure 1(a) illustrating an interconnecting network of semiconductor particles in a binder layer deposited on a substrate, according to a second embodiment of the invention;

**Figure 1(c)** is a schematic sectional diagram similar to that of Figures 1(a) and 1(b) illustrating semiconductor particles concentrated in the lower part of a binder layer deposited on a substrate, according to a third embodiment of the invention;

**Figure 1(d)** is a schematic sectional diagram similar to that of Figures 1(a) to (c) illustrating semiconductor particles concentrated in the upper part of a binder layer deposited on a substrate, according to a fourth embodiment of the invention;

**Figure 2** is a schematic diagram of a natural oil suitable for use in the method of the invention;

**Figure 3** is a graph illustrating characteristic curves of Schottky diodes manufactured according to the method of the invention, using different binder materials; and

**Figure 4** is a graph illustrating the conductance of semiconducting layers, produced from silicon inks using linoleic and linolenic acid as binders, as a function of the applied gate potential in a transistor test structure.

#### DESCRIPTION OF EMBODIMENTS

The present invention concerns printable compositions for use in electronic and electrical applications in general, and specifically in those applications where semiconducting properties are required. These compositions, which can be used in the production of electronic circuits, components and as composite materials for the production of semiconducting materials and layers, comprise particles with a semiconducting functionality and a self-polymerising binder, preferably of natural origin, which cures through oxidation or hydroxation induced auto-polymerisation.

For purposes of this document, any such printable composition, generally known as an ink, lacquer, paint, varnish, suspension or the like is referred to as an "ink". Methods of applying the ink, including spraying, casting and painting, as well as conventional intaglio, flat or relief printing techniques,

are generally referred to as "printing". Dry transfer methods such as xerography and thermal printing may be included if the melting point and viscosity of the binder are such that it is solid under normal conditions. The above types of printable composition and methods of application thereof are given by way of example and other examples may occur to those skilled in the art.

As stated above, semiconductor inks are known in which the binder or matrix material is a polymer such as polystyrene or cellulose acetate butyrate (CAB), which is soluble in an easily obtainable solvent, such as acetone, chloroform or toluene.

On the other hand, binder-particle mixtures are well known in the traditional crafts as inks, lacquers and varnishes, in which inorganic particles serve as a pigment. Traditionally in a paint or lacquer, the pigment is mixed with a natural drying oil which polymerises through auto-oxidation, although more recently synthetic monomers have been applied, which may also polymerise through hydroxation. In a varnish, or ink, the binder is thinned with a suitable organic solvent, such as ethanol or methanol, or mixture of solvents, such as commercial lacquer thinners. The addition of a thinner serves the dual purpose of adjusting the rheology of the liquid and preventing premature hardening, thus prolonging storage times. The natural origin and limited processing of these binder materials leads them to be considered as ecologically friendly renewable resources.

Neither traditional drying oils, nor their derivatives, have been previously considered as binders in functional inks, such as those used for passive or active thick film electronic components, circuits, or layers. However, synthetic esters, with similar self-polymerising properties, are used commercially in both passive electronics and coating applications. Most of these are proprietary, but they generally consist of either a methyl ester or cyclic ester with aromatic side chains.

For the production of inks having photoelectronic and optoelectronic properties according to the invention, the semiconductor particles need to

be uniformly dispersed in the binder material as shown schematically in Figure 1(a). To achieve this the interfacial tension between the two materials needs to be low, to allow complete wetting of the particles. The viscosity of the ink should be kept high during deposition and curing to prevent sedimentation or flocculation of the particles.

For printing of a semiconducting layer or component, the ink should contain a high volume fraction of the semiconducting particles relative to the binder material, so that the final material contains an interconnecting network of particles, as shown schematically in Figure 1(b), which may have a regular or fractal geometry. The ink may, however, for application or storage, be diluted with any quantity of solvent or thinner, including water or a non-reactive organic liquid. The interfacial tension between the particles and the binder is therefore of little significance, although a higher value may help prevent complete wetting of the interstices between the particles, thus promoting a better conducting path between adjacent particles. In many cases, separation through settling or flocculation, as shown schematically in Figures 1(c) and 1(d), respectively, may be beneficial in the production of bilayer systems and components consisting of semiconducting and insulating layers, such as MIS-FETs and capacitors. The printed layers or components will typically have a thickness in the range from 0.1 to 500 microns.

The binder material should be a natural polymer formed by auto-polymerisation of a precursor consisting of a natural oil, or its derivatives including pure unsaturated fatty acids, mono- and di-glycerides, as well as methyl and ethyl esters of the corresponding fatty acids. Natural oils are generally a complex mixture of triglycerides, with three randomly selected fatty acids attached to a glycerol (propan-1-2-3-triol) backbone, as shown in Figure 2. Their composition, therefore, depends strongly on both the plant species and its cultivation, but through blending a consistent quality with regard to curing properties and fatty acid fractions is usually obtained.

Generally, the volume ratio of the particulate semiconductor material to the binder should be greater than 50%, and more preferably greater than 80%. In

two of the examples below, a volume ratio of the order of 90% was used. The particulate semiconductor material may have a particle size in the range from 5 nanometres to 10 microns, and is preferably nanoparticulate, with a size in the range from 50 to 500 nanometres.

The predominant components of drying oils responsible for polymerisation are the octadecanoic acids, characterised by one double bond, the octadecadienoic acids, with two double bonds, and the octadecatrienoic acids (three double bonds). Other unsaturated fatty acids also contribute to the polymerisation, with the non-polymerised saturated fatty acids acting as plasticisers.

Most common octadecatrienoic acids are all conjugated trans-fatty acids, with alternating double and single bonds, and polymerise the most effectively. The exception is linolenic acid (9cis-12cis-15cis-octadecatrienoic acid), which is a non-conjugated cis-fatty acid and is the major component of linseed oil. Besides linseed oil, the main drying oil used in coatings is tung oil, also known as china wood oil, whose main polymerising component is eleostearic acid (9cis-11trans-13trans-octadecatrienoic acid). Both of these oils have the advantage of being inexpensive and readily available compared to both synthetic binders and other natural oils.

Examples of other sources of oils rich in octadecatrienoic acids, suitable as binder precursors, are listed in Table 1 below.

Oil	Main fatty acid component
Linseed	linolenic (9cis-12cis-15cis-octadecatrienoic)
tung	eleostearic (9cis-11trans-13trans-octadecatrienoic)
calendula (true or pot marigold)	calendic (8trans-10trans-12cis-octadecatrienoic)

Jacaranda	jacaric	(8cis-10trans-12cis-octadecatrienoic)
pomegranate, gourd	punicic	(9cis-11trans-13cis-octadecatrienoic)
Catalpa	catalpic	(9trans-11trans-13cis-octadecatrienoic)

*Table 1: examples of plant oils with significant concentrations of octadecatrienoic acids suitable as self-polymerising binders for semiconductor composite materials.*

For some applications, for example where a graded interface is required, semidrying oils such as soy-bean, cottonseed and castor oil, may be used to supplement, or even replace the drying oils. The major active components of these oils are linoleic (9cis-12cis-octadecadienoic) acid and ricolenic (12-hydroxy-9-octadecanoic) acid.

Oil-based semiconducting inks are particularly well suited to offset lithography and other flat printing processes, because of their intrinsic hydrophobicity. There is already a significant body of common knowledge in the use and handling of traditional oil-based paints and inks, and they can just as easily be applied to relief, intaglio or stencil processes. In addition, the oils are soluble in most easily available non-aggressive solvents, such as ethanol or acetone, allowing simple and safe modification of their viscosity and cleaning.

For increased viscosity and a shorter curing time, methods already applied to traditional media, such as heat treatment to partially polymerise oil before mixing the ink, can be applied. A simple example of this would be to use boiled linseed oil, instead of raw linseed oil, in the ink formulation. However, if siccatives or accelerators are used, their effect on the electronic properties of the resulting material must be taken into account. Most drying agents are metallic salts of organic acids, and contain ionic

groups which can affect the doping level of semiconducting particles, and may also lead to electrical conductivity in otherwise insulating layers.

The pure fatty acids, particularly the octadecatrienoic acids, are polar molecules which are water soluble, making them suitable for water-based compositions for relief and intaglio printing processes. Without the other components found in the oil, they will form a denser, less flexible polymer, and will cure much more rapidly. In addition to the acids listed in Table 1, dienoic acids, such as linoleic (9cis-12cis-octadecadienoic) acid, which is the most common unsaturated fatty acid found in almost all plants, will polymerise to form suitable binders. Like the oils, the pure acids are non-aggressive and soluble in mild solvents like alcohols and ethers which do not attack common materials (e.g. synthetic and natural rubbers, plastics) used in printing processes.

Analogues of the oils, which can also be used as self-polymerising binders for semiconducting inks, include the mono- and di-glycerides and the methyl and ethyl esters of the unsaturated fatty acids. These are effectively intermediate forms, in which systematic removal of saturated acids, while keeping the molecules non-polar, allows the rapid polymerisation associated with pure acids and the hydrophobic properties of the oils.

Substrates on which the printable compositions can be deposited may be rigid or flexible, according to requirements. Possible rigid substrates include glass, metals and stiff or rigid plastics. Flexible substrates can be, for example, flexible plastics, thin metallic layers or paper.

### Example 1

The first example concerns the production of semiconducting inks composed of nanoparticulate silicon with raw linseed oil as a self-polymerising binder. The silicon nanoparticles were produced by mechanical attrition, for 3 hours in an orbital pulveriser, of bulk silicon. Both single crystalline wafers, boron-doped P type and antimony-doped N type,

and metallurgical grade silicon were used. Typical particle sizes were between 100 and 300 nm. Inks were produced by first diluting the oil with ethanol as a carrier solvent, and then dispersing the nanopowders at a high volume ratio, in excess of 90%, relative to the oil. Layers, typically 100 microns thick, were then printed onto paper substrates. Electrical connections for Hall effect measurements were then applied using printable silver conductor screen printing ink (Du Pont 5000). The majority carrier mobilities and concentrations, as determined by the Hall effect, are given in Table 2 below.

Nanopowder	Particle Vol. Fraction	resistivity M $\Omega$ cm	Mobility cm $^2$ V $^{-1}$ s $^{-1}$	Carrier conc $\times 10^{12}$ cm $^{-3}$
n-type Si	88%	0.26	1.6	14.9
n-type Si	90%	0.29	1.9	11.1
n-type Si	92%	0.32	1.2	17
p-type Si	88%	13	0.12	0.86
p-type Si	90%	15.5	0.15	0.28
p-type Si	92%	16.2	0.45	0.41
nm metallurgical Si	91%	2.22	0.65	4.3

*Table 2: Semiconducting characteristics of inks produced from various silicon nanoparticles, using raw linseed oil as a self-polymerising binder and ethanol as a solvent.*

### Example 2

The second example concerns the production of semiconducting inks made with different natural oils as self-polymerising binders and metallurgical grade silicon nanoparticles. The silicon nanoparticles were produced by mechanical attrition of the bulk material for 180 minutes in an orbital pulveriser. Inks were produced by mixing 20 microlitres of oil with 0.5 g silicon powder, and thinning with 50 microlitres of commercial laquer thinners, to achieve complete wetting. The mixture was then agitated for

40 minutes in an ultrasonic bath to achieve a homogeneous dispersion of the silicon powder. The following oils were compared: linseed oil; castor oil; soy bean oil; calendula oil; and a blend of calendula with non-drying oils.

For the investigation of the electrical characteristics of the inks, a 10 mm by 1mm silver strip was cast onto a glass substrate. After drying overnight, a channel of width 0.5 mm was scribed perpendicular to the length, forming the two electrical contacts. The ink was drop cast into the channel to form a symmetric Schottky diode structure. No attempt was made to control the size or spread of the ink drop. Although several compositions cured almost immediately, all diodes were left to cure for three days under ambient conditions. The positive half of the Schottky diode curve is shown in Figure 3 for exemplary structures using the different oils.

With the exception of a scaling factor, resulting from a different reverse saturation current, the curves shown in Figure 3 are qualitatively the same. They can be modelled, as shown by the solid curve, using the same contact potential difference and ideality factor. This indicates that only the semiconducting silicon particles, and not the binder materials, contribute directly to the semiconducting properties of the inks.

### **Example 3**

A third example concerns the use of the fatty acids, linolenic (9cis-12cis-15-cis-octadecatrienoic) acid and linoleic (9cis-12cis-octadecadienoic) acid, in semiconducting inks. For easy comparison of the binder properties, standardised test structures and ink compositions were used.

To construct the test structure, a silver gate contact was first printed, using Du Pont 5000 silver conductor screen printing ink, onto a 350 gsm Euro Art glossy paper substrate by tampon printing. The size of this electrode was 1 mm by 3 mm. Over this, an insulating layer was tampon printed, using Du Pont 8153 insulator. Finally, the source and drain electrodes, also Du Pont

5000 silver conductor, were overprinted using the same method. The gate length and width in the final structure were 120  $\mu\text{m}$  and 1 mm respectively.

In line with the manufacturer's recommendations for the thick film inks, the structures were dried in an oven at 120°C for 30 minutes after printing the insulator, and after printing the final silver contacts. The electrical characteristics of the device structure, without application of the semiconducting layer, were also determined.

Two different silicon powders were used in the formulation of the inks: 2503 metallurgical grade silicon, from Silicon Smelters (Pty) Ltd, Polokwane, South Africa, milled for three hours in an orbital pulveriser; and intrinsic silicon nanopowder, from MTI Crystal Corp, which had a specified particle size less than 50 nm. To fully characterise the influence of the binder, inks were produced with a volume fraction of silicon at the lower limit of 50%, comprising 0.4 g silicon powder added to a solution of 200 microlitres of the respective binder in an appropriate volume of commercial lacquer thinners. To achieve complete wetting of the powder, and similar ink viscosities, the quantity of solvent was kept the same for each type of powder, namely 1.2 ml for the metallurgical grade silicon and 4.8 ml for the smaller intrinsic nanoparticles. After addition of the powder, the mixtures were sonicated in an ultrasonic bath for two hours. Approximately five microlitres of each ink was either hand printed, or drop cast, onto the gate of a different pre-prepared structure, and allowed to dry overnight. The printed layers showed a high integrity, forming a continuous film without being powdery, whereas drop cast layers developed surface cracks during drying. All layers showed good adhesion to the substrate.

The characteristic curves of each structure were determined using a Keithley 4200 semiconductor characterisation system, for drain-source potentials up to 20V, and gate potentials of 0,  $\pm 5$  V,  $\pm 10$  V, and  $\pm 15$  V. The geometry and dimensions of the test structure were such that saturation of the drain-source current could not be attained, and only a weak field effect

was observed. All devices, therefore, exhibited a linear response, with an offset for applied gate potential indicating a finite gate-drain resistance.

The source-drain conductance for each device, after correction for the gate-drain current, is shown as a function of the applied gate potential in Figure 4. As can be seen, the conductance is higher for the printed layers compared to the bare device structure, indicating that there is electrical connectivity between the particles, even with the high binder fractions used. Also, the conductance is dependent on the applied gate voltage, confirming that the printed layers are indeed semiconducting. These effects are much less pronounced for the drop cast layer shown, suggesting that connection between the particles, or the particles and electrical contacts, is inhibited by dispersion or phase separation of the particles.

**CLAIMS**

1. A method of producing a printable composition comprising mixing a quantity of particulate semiconductor material with a quantity of a binder, wherein the binder is a self-polymerising material comprising a natural oil, or a derivative or synthetic analogue thereof.
2. A method according to claim 1 wherein the binder comprises a natural polymer formed by auto-polymerisation of a precursor consisting of a natural oil, or its derivatives including pure unsaturated fatty acids, mono- and di-glycerides, or methyl and ethyl esters of the corresponding fatty acids.
3. A method according to claim 2 wherein the binder comprises a drying or semi-drying oil, or a mixture of drying and semi-drying oils.
4. A method according to claim 3 wherein the oil is a drying oil containing one or more octadecanoic acids, octadecadienoic acids, and/or octadecatrienoic acids.
5. A method according to claim 4 wherein the drying oil is linseed or tung oil.
6. A method according to claim 3 wherein the oil is a semi-drying oil comprising soy-bean, cottonseed or castor oil.
7. A method according to claim 2 wherein the binder is an octadecadienoic acid or an octadecatrienoic acid, or a mixture thereof.
8. A method according to claim 7 wherein the binder is linolenic acid.
9. A method according to claim 7 wherein the binder is linoleic acid.

10. A method according to any one of claims 1 to 9 including mixing the binder with a solvent.
11. A method according to claim 10 wherein the solvent comprises ethanol, acetone or lacquer thinners.
12. A method according to any one of claims 1 to 11 wherein the volume ratio of the particulate semiconductor material to the binder is greater than 50%.
13. A method according to claim 12 wherein the volume ratio of the particulate semiconductor material to the binder is greater than 80%.
14. A method according to any one of claims 1 to 13 wherein the particulate semiconductor material has a particle size in the range from 5 nanometres to 10 microns.
15. A method according to claim 14 wherein the particulate semiconductor material has a particle size in the range from 50 nanometres to 500 nanometres.
16. A method according to claim 15 wherein the particulate semiconductor material has a particle size in the range from 100 nanometres to 300 nanometres.
17. A method according to any one of claims 1 to 16 wherein the particulate semiconductor material comprises silicon.
18. A method of producing an electronic component or conductor comprising preparing a printable composition according to the method of any one of claims 1 to 17, applying the printable

composition to a substrate, and allowing the printable composition to cure to define the component or conductor on the substrate.

19. A method according to claim 18 including applying the composition to the substrate as a lacquer comprising the binder and the particulate semiconductor material.
20. A method according to claim 18 including applying the composition to the substrate as a varnish comprising the binder, the particulate semiconductor material and a solvent.
21. A method according to any one of claims 18 to 20 comprising applying the printable composition in a single application.
22. A method according to any one of claims 18 to 20 comprising applying the printable composition in multiple layers to define an electronic component with desired characteristics.
23. A method according to any one of claims 18 to 22 including allowing the printable composition to cure under ambient conditions.
24. A method according to any one of claims 18 to 23 wherein the substrate is rigid and comprises metal, glass or plastics or paper.
25. A method according to any one of claims 18 to 23 wherein the substrate is flexible and comprises metal, plastics or paper.
26. A method according to any one of claims 18 to 25 including applying the printable composition to the substrate in a layer having a thickness in the range from 0.1 to 500 microns.
27. A method according to claim 26 wherein the printable composition is applied to the substrate in a layer having a thickness in the region of 100 microns.

28. A printable composition produced by the method of any one of claims 1 to 17.
29. An electronic component or conductor produced by the method of any one of claims 18 to 27.

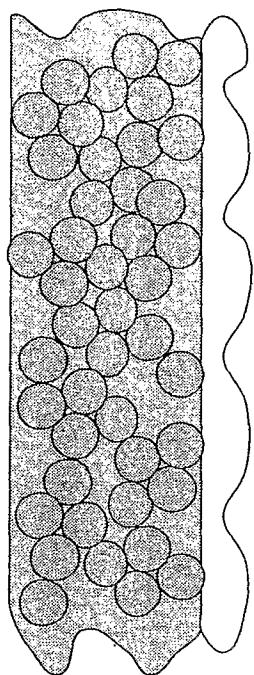


Fig. 1(b)

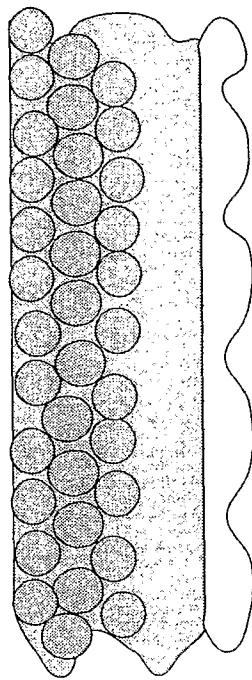


Fig. 1(d)

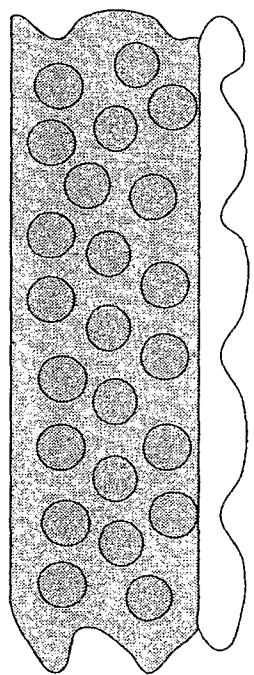


Fig. 1(a)

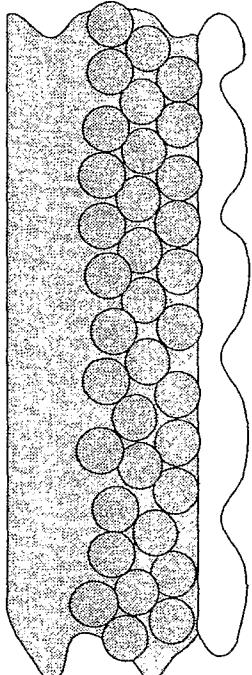


Fig. 1(c)

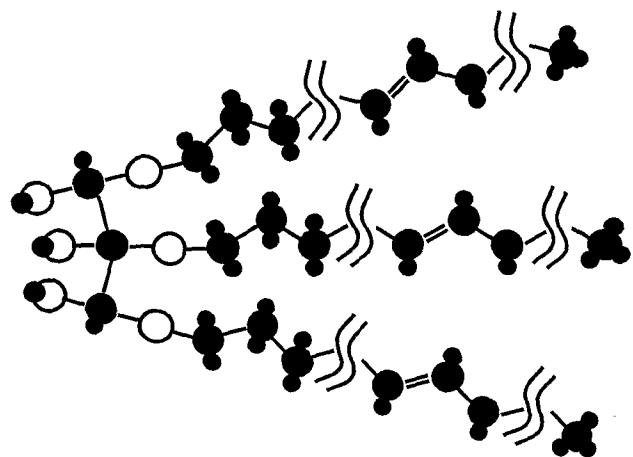


Fig.2

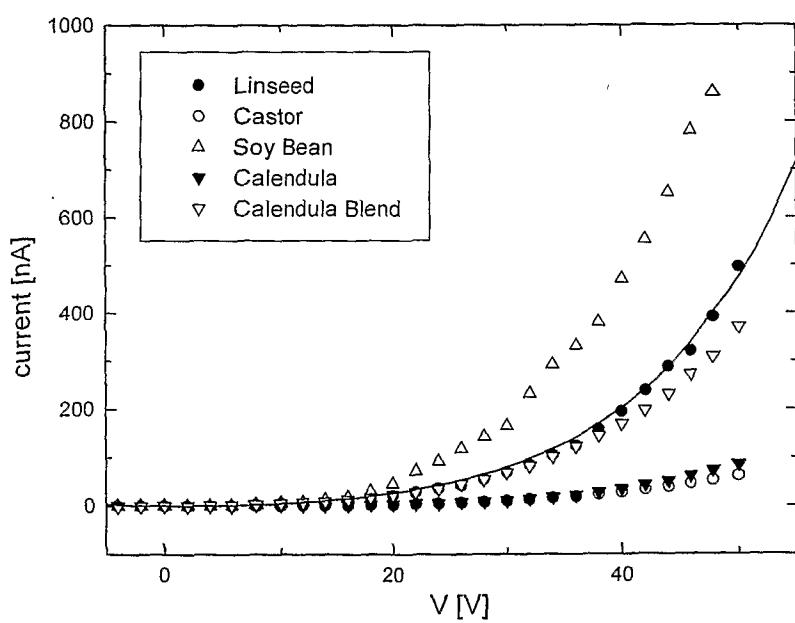


Fig.3

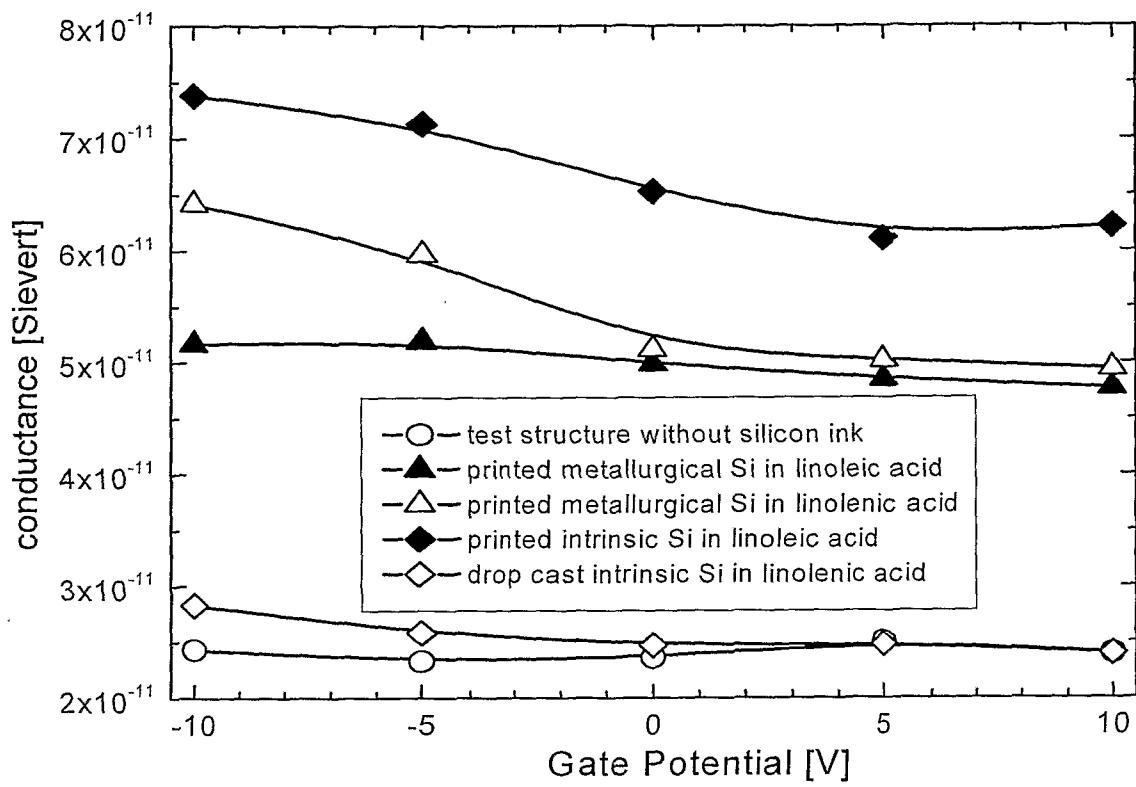


Fig. 4

## INTERNATIONAL SEARCH REPORT

International application No  
PCT/IB2006/003666A. CLASSIFICATION OF SUBJECT MATTER  
INV. C09D5/24 C09D11/00 C09D191/00

According to International Patent Classification (IPC) or to both national classification and IPC

## B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)  
C09D

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

EPO-Internal, WPI Data

## C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	US 3 947 278 A (YOUTSEY KARL J) 30 March 1976 (1976-03-30) the whole document -----	1-29
X	US 3 992 212 A (YOUTSEY KARL J ET AL) 16 November 1976 (1976-11-16) column 6, line 23 - line 48; claims; examples -----	1-29
X	EP 1 104 791 A (SICPA HOLDING SA [CH]) 6 June 2001 (2001-06-06) examples -----	1-29
X	US 3 989 644 A (BOLON DONALD A ET AL) 2 November 1976 (1976-11-02) example 2 ----- -/-	1-29

 Further documents are listed in the continuation of Box C. See patent family annex.

\* Special categories of cited documents :

- "A" document defining the general state of the art which is not considered to be of particular relevance
- "E" earlier document but published on or after the international filing date
- "L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)
- "O" document referring to an oral disclosure, use, exhibition or other means
- "P" document published prior to the international filing date but later than the priority date claimed

- "T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention
- "X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone
- "Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.
- "&" document member of the same patent family

Date of the actual completion of the international search Date of mailing of the international search report

16 April 2007

20/04/2007

Name and mailing address of the ISA/  
European Patent Office, P.B. 5818 Patentlaan 2  
NL - 2280 HV Rijswijk  
Tel. (+31-70) 340-2040, Tx. 31 651 epo nl  
Fax: (+31-70) 340-3016

Authorized officer

Schmitz, Volker

## INTERNATIONAL SEARCH REPORT

International application No  
PCT/IB2006/003666

## C(Continuation). DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	US 3 015 632 A (REMER ROBERT K) 2 January 1962 (1962-01-02) examples -----	1-29
X	US 2 014 760 A (GRAHAM DEWSBURY WILFRED ET AL) 17 September 1935 (1935-09-17) the whole document -----	1-29
X	EP 0 319 953 A (FUJI KAGAKU SHIKOGYO [JP]) 14 June 1989 (1989-06-14) examples -----	1-29
X	EP 0 285 359 A2 (TANIGUCHI INK MFG [JP]) 5 October 1988 (1988-10-05) example 2 -----	1-29
X	JP 2004 244525 A (KYOCERA CHEM CORP) 2 September 2004 (2004-09-02) abstract -----	1-29
X	JP 01 104679 A (TORAY INDUSTRIES) 21 April 1989 (1989-04-21) abstract -----	1-29

## INTERNATIONAL SEARCH REPORT

International application No  
PCT/IB2006/003666

Patent document cited in search report		Publication date	Patent family member(s)		Publication date
US 3947278	A	30-03-1976	NONE		
US 3992212	A	16-11-1976	FR	2284657 A2	09-04-1976
EP 1104791	A	06-06-2001	AT	282071 T	15-11-2004
			AU	7659700 A	04-06-2001
			CN	1425047 A	18-06-2003
			DE	60015809 D1	16-12-2004
			DE	60015809 T2	15-12-2005
			WO	0139135 A2	31-05-2001
			ES	2231270 T3	16-05-2005
			HK	1056570 A1	13-04-2006
			JP	2003515622 T	07-05-2003
			PT	1268686 T	28-02-2005
			US	7188768 B1	13-03-2007
US 3989644	A	02-11-1976	CA	1049253 A1	27-02-1979
US 3015632	A	02-01-1962	NONE		
US 2014760	A	17-09-1935	NONE		
EP 0319953	A	14-06-1989	JP	1153776 A	15-06-1989
			US	5017228 A	21-05-1991
EP 0285359	A2	05-10-1988	DE	3886246 D1	27-01-1994
			DE	3886246 T2	07-07-1994
			JP	1877322 C	07-10-1994
			JP	6004799 B	19-01-1994
			JP	63243178 A	11-10-1988
			US	4938801 A	03-07-1990
			US	5074915 A	24-12-1991
JP 2004244525	A	02-09-2004	NONE		
JP 1104679	A	21-04-1989	JP	1981489 C	25-10-1995
			JP	7013212 B	15-02-1995