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(54) Title: PERFLUOROELASTOMER COMPOSITIONS INCLUDING BARIUM TITANATE FILLERS

(57) Abstract: Provided herein is a fluorine-containing elastomer composition having a first curable perfluoropolymer comprising tetrafluoroethylene, at least one perfluoroalkylvinyl ether and at least one cure site monomer having a functional group to permit crosslinking of the perfluoropolymer; and barium titanate.



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TITLE OF THE INVENTION

[0001] Perfluoroelastomer Compositions Including Barium Titanate Fillers

CROSS-REFERENCE TO RELATED APPLICATIONS

[0002] This application claims the benefit under 35 U.S.C. §119(e) of U.S. Provisional
5 Patent Application No. 61/138,333, filed December 17, 2008, the entire disclosure of which is
incorporated herein by reference.

BACKGROUND OF THE INVENTION

FIELD OF THE INVENTION

[0003] The present invention relates to a fluorine-containing elastomer composition
10 prepared using perfluoroelastomers and having a barium titanate filler system. The present
invention also relates to curable compositions and molded articles formed from such
perfluoroelastomer compositions.

DESCRIPTION OF RELATED ART

[0004] Fluorine-containing elastomers, particularly perfluoroelastomers (FFKM)s mainly
15 comprising a tetrafluoroethylene (TFE) unit exhibit excellent chemical resistance, solvent
resistance and heat resistance, and therefore are widely used as a basis for sealing materials and
other products exposed to harsh environmental conditions.

[0005] Perfluoroelastomeric materials are known for their chemical resistance, plasma
resistance, and when used in compositions having typical filler or reinforcing systems for
20 acceptable compression set resistance levels and mechanical properties. As such, they have
been applied for many uses, including for use as elastomeric sealing materials in applications
where a seal or gasket will be subject to highly corrosive chemicals and/or extreme operating
conditions, and for use in forming molded parts that are capable of withstanding deformation.
FFKM)s are also well known for use in the semiconductor manufacturing industry as sealing
25 materials due to their chemical and plasma resistance. Such materials are typically prepared
from perfluorinated monomers, including at least one perfluorinated cure site monomer. The
monomers are polymerized to form a perfluorinated polymer having the cure sites from the cure
site monomer(s) and then cured (cross-linked) to form an elastomer. Typical FFKM
compositions include a polymerized perfluoropolymer as noted above, a curing agent that reacts

with the reactive cure site group on the cure site monomer, and any desired fillers. The cured perfluoroelastomer exhibits typical elastomeric characteristics.

[0006] FFKMs are generally known for use as O-rings and related sealing parts for high-end sealing applications due to their high purity, excellent resistance to heat, plasma, chemicals and other harsh environments. Industries that require their use in such environments include semiconductor, aerospace, chemical and pharmaceutical. The development of new perfluoroelastomer compositions using these materials faces ever-increasing demands and challenges for FFKMs and compositions based on FFKMs that have the ability to provide greater thermal, chemical and plasma resistance. Industry demands, particularly in the semiconductor area, continue to require enhanced performance of such seals to meet new end-use applications that have increasingly aggressive environments as well as lower and lower contamination and particulation requirements.

[0007] As is recognized in the art, different FFKM compositions may include different curatives (curing agents) depending on the type of cure site monomer (CSM) structure and corresponding curing chemistry. Such compositions may also include a variety of fillers and combinations of fillers to achieve target mechanical properties, compression set or improved chemical and plasma resistance. For semiconductor sealing applications, both inorganic and organic fillers can be used to improve plasma resistance depending on the type of plasma chemistry. Typical fillers include carbon black, silica, alumina, fluoroplastics, barium sulfate and other polymers and plastics. Fillers used in some FFKM compositions for semiconductor applications include fluoroplastic filler particles formed of polytetrafluoroethylene (PTFE) or melt-processible perfluorinated copolymers such as copolymers of tetrafluoroethylene (TFE) and hexafluoropropylene (HFP) (also referred to as FEP-type copolymers) or of TFE and perfluoroalkylvinyl ethers (PAVEs) (known as PFA-type copolymers), particularly in micro- or nanometer-sized particles.

[0008] U.S. Patent No. 6,710,132 discloses a blend of an FFKM with semi-crystalline fluoroplastic particles (such as PTFE), wherein the particles have a core-shell structure and are formed by latex blending of these materials.

[0009] U.S. Patent No. 4,713,418 discloses a composition formed by melt blending an FFKM and a melt-processible thermoplastic fluoropolymer. The patent asserts that particles of about 10 microns are reformed from some of the melted thermoplastic upon recrystallization. U.S. Patent Publication No. 2005/0261431 A1 discloses melt blending an FFKM and a

semicrystalline polymer such as PTFE and/or a copolymer, such as the PFA-type copolymer, of greater than an average size of 100 nm wherein blending temperature or curing temperature exceeds the melting temperature of the fluoroplastic fillers.

[0010] U.S. Patent No. 7,019,083 and International Publication WO2006/120882 A1 disclose crosslinkable fluoroplastics.

[0011] When an FFKM composition includes a semicrystalline fluoroplastic particle filler, such as microparticles or nanoparticles of PTFE or copolymers such as those of the PFA-type, good physical properties, good plasma resistance and excellent purity are achieved. For semiconductor applications, such systems also help to avoid metallic particulation and contamination at a level improved over FFKMs, which have inorganic fillers such as metal oxides. However, there is a need in the art to develop more simplified processing methods to form fluoropolymer-filled FFKMs. Latex blending can be expensive for large-scale, commercial batches and melt blending generally requires temperatures of up to 350°C. Filler loading in many commercial products is generally limited to up to about 30 weight percent of the base polymer. Due to the use of the fluoropolymeric fillers, such compositions can also sometimes have a relatively high compression set especially at high temperatures (e.g., >300 °C). Moldability and bondability can also be limited due to use of such fluoropolymeric fillers.

[0012] In certain semiconductor processes such as atomic layer deposition (ALD), corrosive and reactive gases such as chlorinated fluorine gas (ClF_3) is used for etching at high temperatures (such as about 280 to about 300°C). Perfluoroelastomer seals used in such applications face the challenges of requiring high temperature resistance as well as ClF_3 resistance. Such perfluoroelastomers should also be able to exhibit generally acceptable strong chemical resistance to ClF_3 as well as other plasma-type gases, such as NF_3 and maintain good properties in various chemical vapor deposition processes including ALD.

[0013] Products are known in the art for use in ALD-type applications and other uses, such as Kalrez® 8575 sold by DuPont Elastomers. These ALD-type products are believed to include FFKM including barium sulfate (BaSO_4) and titanium dioxide (TiO_2) fillers. However there is a need in the art for additional mechanical stability under a compression load at temperatures of about 300°C.

[0014] Chemraz® XRZ seals from Greene, Tweed & Co., Inc., Kulpsville, PA are very useful in ALD-type applications and represents a strong plasma-resistant product in this area.

[0015] There is a desire, however, to continually improve in the areas of high temperature properties and ClF₃ exposure.

5 [0016] U.S. Published Patent Application No. 2005-0107544 A1 discloses a fluoroelastomer composition that has a fluoroelastomer, curative and barium sulfate filler, wherein the barium sulfate filler has an average particle size less than 200 nm. The composition is described as losing less weight and producing fewer particles when exposed to reactive plasmas than similar compositions having barium sulfate particles, which are of a
10 larger size.

[0017] U.S. Published Patent Application No. 2005-0038165 A1 discloses a non-black filled perfluoroelastomer composition that has a perfluoroelastomer having nitrile groups, which groups provide the perfluoroelastomer cure sites; curatives, which may be bis(aminophenol) and compounds that decompose at curing temperatures to generate ammonia;
15 and 1 to 25 parts per hundred of hydrophobic silica filler. The cured compositions are described as having surprisingly better compression set than similar compounds containing hydrophilic silica.

[0018] U.S. Patent No. 6,642,300 discloses a filler for crosslinkable elastomers such as perfluoroelastomers or silicone compositions wherein the filler is, for example, an imide such
20 as polyimide, polyamideimide and polyetherimide or an organic filler prepared from engineering plastics having a heat resistance of not less than 150° C, such as polyarylates, polysulfones, polyether sulfones, polyphenylene sulfides, polyether ether ketones and polyoxy benzoates.

[0019] European Patent Application No. 1 464 671 A1 discloses a fluorine-containing
25 elastomer molded article obtained by crosslinking with a heat resistant crosslinking agent a crosslinkable fluorine-containing elastomer composition with an inorganic filler that has an average primary particle size of 0.5 μm or less. Examples of the fillers are α-type aluminum oxide and aluminum nitride, wherein the fluorine-containing elastomer may be a perfluoroelastomer, which has a cyano or a carboxylic acid group.

30 [0020] U.S. Patent No. 6,803,402 discloses elastomeric molded articles and cross-linkable fluorine-containing elastomers for use in semiconductor production apparatuses having a filler system which also includes particle of aluminum oxide having an average particle size of 0.5

μm or less. The elastomer molded articles are described as having excellent plasma resistance and low formation of microparticles generated after irradiation of plasma, so as to provide clean molded articles, particularly sealing material.

[0021] U.S. Patent No. 6,515,064 discloses elastomeric molded articles obtained by crosslinked elastomers having a carbon black filler which has an average particle size of 700 μm or less, an impurity metal content measured by an ashing analysis method of not more than 300 ppm or not more than 100 ppm, with low particle generation in plasma environments.

[0022] While such compositions as are noted above are present in the art, there is a continuing need in the art to further improve perfluoroelastomer compositions so that, upon cure, they are able to provide good mechanical properties and compression set while being able to meet the increasingly demanding requirements for use in high-end sealing applications like those of semiconductor processing and to withstand high temperatures as well as exposure to harsh plasma such as ClF₃ and NF₃.

BRIEF SUMMARY OF THE INVENTION

[0023] Problem to be solved by the invention: It is an object of the present invention to provide a fluorine-containing elastomer composition which has a short crosslinking time and good mechanical properties but which also demonstrates enhanced thermal resistance properties to about 300°C as well as the ability to continue to maintain good plasma resistance in harsh environments such as NF₃ plasma gases, but also to provide enhanced resistance to including

ClF₃ plasma. The method of solving the problems of the prior art is provided by the present invention and its unique filler system including barium titanate.

[0024] The invention includes a fluorine-containing elastomer composition comprising a first curable perfluoropolymer comprising tetrafluoroethylene, at least one perfluoroalkylvinyl ether and at least one cure site monomer having a functional group to permit crosslinking of the perfluoropolymer; and barium titanate.

[0025] The fluorine-containing elastomer composition preferably has barium titanate with an average particle size of at least 200 nm, and in a further embodiment, with an average particle size of about 300 nm to about 1200 nm, and in yet a further embodiment, an average particle size of about 500 nm to about 1000 nm. The barium titanate may be present in a mixture of at least two different average particle sizes.

[0026] In one embodiment, the barium titanate is present in the fluorine-containing elastomer composition in an amount of about 1 to about 200 parts by weight of the barium titanate per 100 parts by weight of the curable perfluoropolymer, and in a further embodiment, about 1 to about 100 parts by weight barium titanate per 100 parts by weight of the

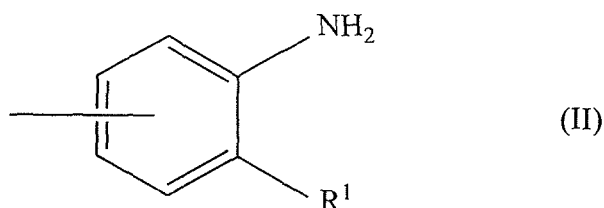
perfluoropolymer, and preferably about 1 to about 50 parts by weight of the perfluoropolymer.

In a further embodiment, the barium titanate is present in the fluorine-containing elastomer composition in an amount of about 5 to about 100 parts by weight of barium titanate per 100 parts by weight of curable perfluoropolymer, more preferably about 5 to about 50 parts by weight of the barium titanate per 100 parts by weight of the curable perfluoropolymer, even more preferably about 5 to about 30 parts by weight of barium titanate per 100 parts curable perfluoropolymer, and in a further embodiment about 10 to about 20 parts by weight of the barium titanate per 100 parts by weight of the curable perfluoropolymer.

[0027] In one embodiment, the functional group of the at least one cure site monomer is selected from the group consisting of nitrile, carboxyl and alkoxycarbonyl.

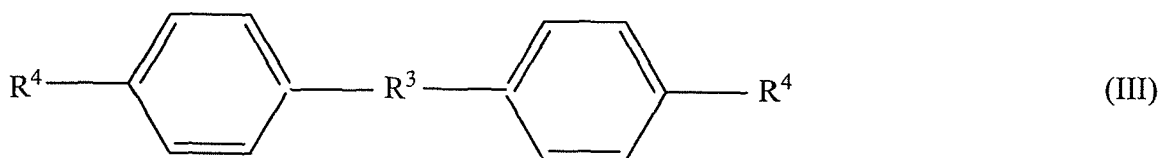
In yet a further embodiment, the composition may comprise a crosslinking agent which is capable of reacting with the functional group of the at least one cure site monomer, wherein the functional group of the at least one cure site monomer may optionally be selected from the group consisting of nitrile, carboxyl and alkoxycarbonyl and the crosslinking agent may be selected from the following compounds:

(i) a compound containing at least two crosslinkable reaction groups represented by the formula (II):

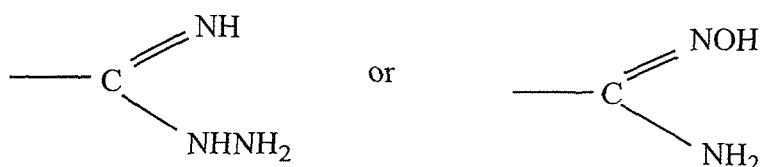


wherein R^1 groups are the same or different and each is $-NH_2$, $-NHR^2$, $-OH$ or $-SH$; R^2 is a monovalent organic group,

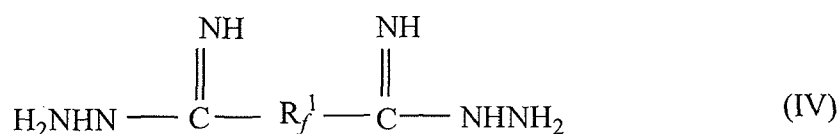
(ii) a compound represented by the formula (III):



wherein R^3 is $-\text{SO}_2-$, $-\text{O}-$, $-\text{CO}-$, an alkylene group having 1 to 6 carbon atoms, a perfluoroalkylene group having 1 to 10 carbon atoms or a single bond; R^4 is

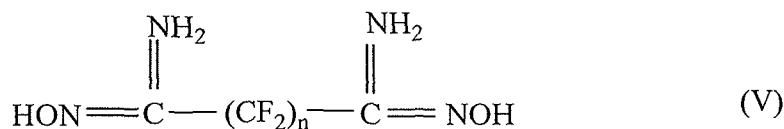


(iii) a compound represented by the formula (IV):



wherein R_f^1 is a perfluoroalkylene group having 1 to 10 carbon atoms, and

(iv) a compound represented by the formula (V):



wherein n is an integer of 1 to 10.

[0028] In one embodiment of the invention the composition may optionally also include a second curable perfluoropolymer which may be the same or different from the first curable perfluoropolymer. Such a second curable perfluoropolymer preferably comprises tetrafluoroethylene, a second perfluoroalkylvinyl ether (which may be the same or different from the first perfluoroalkylvinyl ether in the first curable perfluoropolymer) and a second cure site monomer (which may be the same or different from the first cure site monomer in the first perfluoropolymer). In such an embodiment, the content of the first perfluoroalkylvinyl ether in the first curable perfluoropolymer may differ from the content of the second perfluoroalkylvinyl ether in the second curable perfluoropolymer, with a preferred difference in content of the first perfluoroalkylvinyl ether in the first curable perfluoropolymer and the second perfluoroalkylvinyl ether in the second curable perfluoropolymer of about 5 to about 25 percent by mole.

[0029] In yet a further embodiment, the fluorine-containing elastomer composition has a perfluoroalkylvinyl ether, which is a perfluoromethylvinyl ether.

[0030] In a further, preferred embodiment, the fluorine-containing elastomer composition has barium titanate in a stoichiometric ratio and chemical structure of BaTiO₃.

[0031] In another embodiment, the fluorine-containing elastomer composition may also include a cross-linking agent capable of reacting with the functional group of the at least one
5 cure site monomer, wherein the functional group is selected from the group consisting of nitrile, carboxyl and alkoxycarbonyl, and the cross-linking agent is present in the composition in an amount of about 0.6 to about 0.9 weight percentage.

[0032] The invention also includes sealing materials for semiconductor manufacturing equipment made from the fluorine-containing elastomer compositions described herein.

10 [0033] The invention further includes a cured perfluoroelastomeric composition, comprising (a) a cured perfluoroelastomer formed by the curing reaction of (i) a first curable perfluoropolymer comprising tetrafluoroethylene, at least one perfluoroalkylvinyl ether and at least one first cure site monomer having a functional group to permit crosslinking of the perfluoropolymer and (ii) a first curative; and (b) barium titanate. In one embodiment, the
15 barium titanate has particle sizes and may be present in amounts as noted above.

[0034] In one embodiment, the cured perfluoroelastomeric composition also comprises a second cured perfluoroelastomer formed from the curing reaction of (i) a second curable perfluoropolymer having a second cure site monomer, wherein the second curable perfluoropolymer may be the same or different from the first curable perfluoropolymer and (ii)
20 a curative which may be the same or different from the first curative, wherein the cure site of the at least one first cure site monomer and/or the cure site of the second cure site monomer is a functional group selected from the group consisting of a nitrile group, a carboxyl group and an alkoxycarbonyl group. In addition, at least one of the first perfluoroelastomer and the second perfluoroelastomer preferably has a benzoimidazole cross-linking structure.

25 [0035] In yet a further embodiment of the cured perfluoroelastomeric composition the barium titanate has a stoichiometric ratio and a chemical structure of BaTiO₃.

[0036] The invention also includes molded articles comprising the cured perfluoroelastomeric compositions described herein. Such molded articles may be O-rings, seals or gaskets. The molded articles may be bonded to a surface comprising a metal or a metal
30 alloy, which surfaces may be the surfaces of doors for the sealing of a semiconductor-processing chamber.

[0037] The invention further includes a method for making a cured perfluoroelastomeric composition which method comprises (a) preparing a curable perfluoroelastomeric composition by combining: (i) a first curable perfluoropolymer comprising tetrafluoroethylene, a perfluoroalkyl vinyl ether and at least one first cure site monomer having a cure site; (ii) at least one curative capable of curing the cure site of the at least one first cure site monomer; and (iii) barium titanate; and (b) curing the curable perfluoropolymer in the perfluoroelastomeric composition to form a cured perfluoroelastomeric composition. In one embodiment, the barium titanate has particle sizes and may be present in amounts as noted above.

[0038] In the method noted herein, the barium titanate may have a stoichiometric ratio and a chemical structure of BaTiO_3 .

[0039] Preferably, the cured perfluoroelastomeric composition formed by the method herein comprises a benzoimidazole cross-linking structure.

[0040] The method may further comprise forming the curable perfluoroelastomeric composition into a molded article while curing the curable perfluoroelastomeric composition.

[0041] The invention further includes an improvement to a method of processing in a processing apparatus having a sealing material therein, wherein the process includes use of high temperatures and/or use of a ClF_3 and/or NF_3 gas or plasma, wherein the improvement comprises the sealing material including a cured perfluoroelastomeric composition comprising barium titanate.

DETAILED DESCRIPTION OF THE INVENTION

[0042] New perfluoroelastomer compositions, and/or molded articles made therefrom such as sealing members including O-rings, seals, gaskets and the like as described herein provide required chemical and plasma resistance, and more particularly provide excellent levels of high-temperature resistance and enhanced plasma resistance for such molded articles when exposed to remote NF_3 plasma as well as when exposed to ClF_3 plasma. Molded articles made in accordance with the curable and cured compositions herein have the characteristics suitable for use in semiconductor plasma and gas chemical vapor deposition (CVD) applications including high density plasma CVD (HDPCVD), plasma-enhanced CVD (PECVD) and atomic layer deposition (ALD) and plasma-enhanced atomic layer deposition (PEALD). From processing and performance perspectives, the curable and cured compositions described herein provide

articles that perform equivalent to or better than various prior art filled FFKM compositions which incorporate semicrystalline fluoroplastics as particulate fillers as described in the Background hereof and provide additional chemical resistance in ClF_3 plasma in comparison to such perfluoroelastomer compositions with other prior art fillers, or as blended using two different perfluoroelastomers having varied PAVE content without a filler system.

[0043] The cured perfluoroelastomer compositions may be formed from a curable composition having one or more types of curable perfluoropolymer in the composition. In one embodiment, there are at least two such perfluoropolymers. If more than one perfluoropolymer is used, one embodiment provides that certain benefits may be achieved by having varying perfluoroalkylvinyl ether (PAVE) monomer content in at least two of the perfluoropolymers. The difference in content, measured in percentage by mole, between any two such differing PAVE-content perfluoropolymers is preferably, in such an embodiment, from about 5% to about 25%.

[0044] In such compositions, the preferred filler system herein includes barium titanate.

The barium titanate preferably is present in its stoichiometric form BaTiO_3 , however, it should be understood based on this disclosure that modifications or variants of the stoichiometric form, wherein the stoichiometric ratios of barium to titanium (1:1), barium to oxygen (1:3) and/or titanium to oxygen (1:3) in the molecule vary, may also fall within the scope of the invention provided such compounds having variations in stoichiometric ratio are still able to provide at least some of the beneficial effects of the invention as described herein.

[0045] In addition, in some embodiments herein, the compositions are directed to achieving and/or maintaining shorter perfluoroelastomer crosslinking times, which is achieved when using the unique fillers herein with a perfluoroelastomer in the composition that has a generally lower content of PAVE than is normally used in such compositions.

[0046] As used in this application, "perfluoroelastomer" or "cured perfluoroelastomer" unless otherwise indicated, includes any cured elastomeric material or composition that is formed by curing a curable perfluoropolymer(s) such as the curable perfluoropolymers in the curable perfluoroelastomeric compositions described herein. A "curable perfluoropolymer" (sometimes referred to in the art as a "perfluoroelastomer" or more appropriately a "perfluoroelastomer gum") that may be used to form a cured perfluoroelastomer is a polymer that is substantially completely fluorinated, which is preferably completely perfluorinated on its

polymeric backbone. It will be understood, based on this disclosure, that some residual hydrogen may be present in some perfluoroelastomers within the crosslinks of those materials due to use of hydrogen as part of certain functional crosslinking groups. Cured materials, such as perfluoroelastomers are generally cross-linked polymeric structures.

5 [0047] The curable perfluoropolymers that are used in perfluoroelastomeric compositions to form cured perfluoroelastomers upon cure are formed by polymerizing one or more perfluorinated monomers, one of which is preferably a perfluorinated cure site monomer having a functional group to permit curing, wherein the functional group includes a reactive group that may not be perfluorinated. One or more perfluoropolymers, and preferably at least one curing
10 agent, are combined in a perfluoroelastomeric composition (which may be a blended composition including more than one curable perfluoropolymeric compound) that is then cured forming the resulting crosslinked, cured perfluoroelastomeric compositions as described herein.

[0048] As used herein, a "perfluoroelastomeric composition" is a polymeric composition including one or more, and preferably more than one curable perfluoropolymers, each of which
15 is formed by polymerizing two or more perfluorinated monomers, including at least one perfluorinated monomer which has at least one functional group to permit curing, i.e. at least one cure site monomer. Such materials are also referred to generally as FFKMs in accordance with the American Standardized Testing Methods (ASTM) standardized rubber definitions and as described further herein.

20 [0049] As used herein, "compression set" refers to the propensity of an elastomeric material to remain distorted and not return to its original shape after a deforming compressive load has been removed. The compression set value is expressed as a percentage of the original deflection that the material fails to recover. For example, a compression set value of 0% indicates that a material completely returns to its original shape after removal of a deforming
25 compressive load. Conversely, a compression set value of 100% indicates that a material does not recover at all from an applied deforming compressive load. A compression set value of 30% signifies that 70% of the original deflection has been recovered. Higher compression set values generally indicate a potential for seal leakage and so compression set values of 30% or less are preferred in the sealing arts.

30 [0050] As described herein, the invention includes a preferred curable perfluoroelastomeric composition, cured perfluoroelastomer compositions and molded articles formed from the same as well as methods for making such compositions and improvements to methods of processing

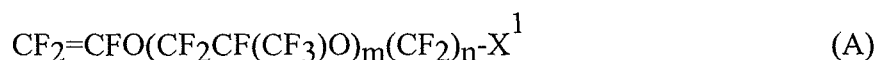
wherein the process involves high temperatures and/or harsh chemical gases or plasmas such as NF₃ or ClF₃.

[0051] Such perfluoroelastomeric compositions preferably include one or more perfluorinated copolymers. If two or more such copolymers are used, it is preferred that at least one of such perfluorinated copolymers has a higher content of tetrafluoroethylene (TFE) than a second polymer in the composition. Other suitable co-monomers may include other ethylenically unsaturated fluoromonomers. Each such polymer also has one or more perfluoroalkylvinyl ethers (PAVEs), which include alkyl or alkoxy groups that may be straight or branched and which may also include ether linkages, wherein preferred PAVEs for use herein include, for example, perfluoromethylvinyl ether (PMVE), perfluoroethylvinyl ether (PEVE), perfluoropropylvinyl ether (PPVE), perfluoromethoxyvinyl ether and other similar compounds, with especially preferred PAVEs being PMVE, PEVE and PPVE, and most preferred being PMVE which provides excellent mechanical strength to resulting articles formed from curing the curable compositions herein. The PAVEs may be used alone or in combinations of the above-noted PAVE types within the curable perfluoropolymers and in the ultimate curable compositions.

[0052] Preferred perfluoropolymers are co-polymers of TFE, at least one PAVE, and at least one perfluorinated cure site monomer that incorporates a functional group to permit crosslinking of the curable polymer when the function group of the cure site monomer reacts with a curative or curing agent to form a cross-link in the cured elastomeric structure. The cure site monomers may be of a variety of types with preferred cure sites noted herein. While preferred cure sites include those having a nitrogen-containing group, a carboxyl group or an alkylcarbonyl group, other cure sites such as iodine, bromine and other halogenated cures as well as other cure sites such as cyano-containing cure sites known in the art may also be used, particularly if additional curable perfluoropolymers are provided to the composition. Consequently, while the disclosure herein provides a variety of preferred curatives (also referred to herein as crosslinking agents, curing agents), if other cure sites known in the art are used, other curatives that are capable of curing such alternative cure sites may accordingly be used. For example, it is within the scope of the invention to use a perfluoropolymer that has a curesite monomer with a functional cure site group that is halogenated, e.g., -CH₂I, -CH₂Br,

and other known halogenated cure site functional groups, in which case suitable curatives for such groups, including organic peroxide-based curatives and co-curatives may be used.

[0053] Exemplary, but not limiting, cure site monomers are listed below, most of which are PAVE-based in structure and have a reactive site, although they may vary, are those having the following structure (A):



wherein m is 0 or an integer from 1 to 5, n is an integer from 1 to 3 and X^1 is a nitrogen-containing group, such as nitrile or cyano, a carboxyl group and/or an alkoxycarbonyl group.

The functional groups noted herein, such as the nitrogen-containing groups, are the sites for crosslinking. Compounds according to formula (A) may be used alone or in various, optional, combinations thereof. From a crosslinking perspective, it is preferred that the crosslinking functional group is a nitrogen-containing group, preferably a nitrile group. However, other groups may also be used.

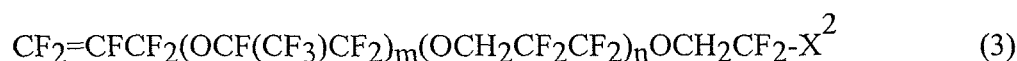
[0054] Further examples of cure site monomers according to formula (A) include formulas (1) through (17) below:



wherein Y is H or F, n is an integer from 1 to about 8



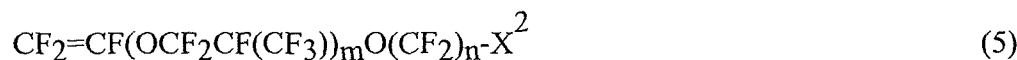
wherein R_f^2 is $(-\text{CF}_2)_n$, $(-\text{OCF}_2)_n$ and n is 0 or an integer from 1 to about 5



wherein m is 0 or an integer from 1 to about 5 and n is 0 or an integer of from 1 to about 5



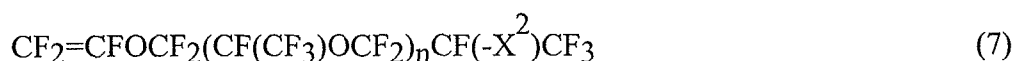
wherein m is 0 or an integer from 1 to about 5, and n is 0 or an integer of from 1 to about 5



wherein m is 0 or an integer from 1 to about 5, and n is an integer of from 1 to about 8



wherein m is an integer from 1 to about 5



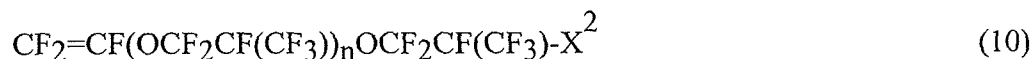
wherein n is an integer from 1 to about 4



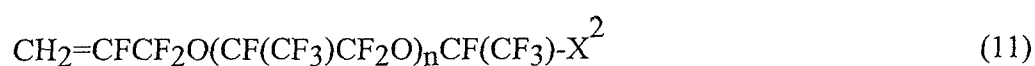
wherein n is an integer of from 2 to about 5



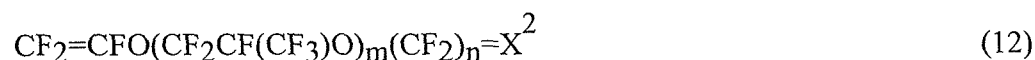
wherein n is an integer from 1 to about 6



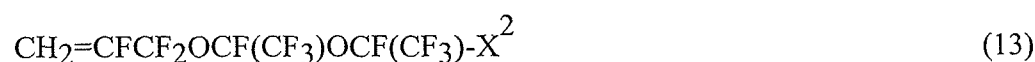
wherein n is an integer from 1 to about 2



10 wherein n is 0 or an integer from 1 to about 5



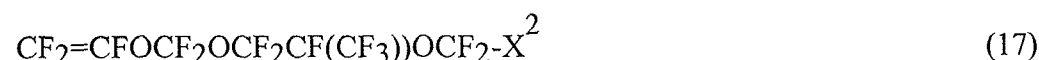
wherein m is 0 or an integer from 1 to about 4 and n is an integer of 1 to about 3



wherein m is an integer greater than 0



wherein n is an integer that is at least 1



20 wherein X^2 can be a monomer reactive site subunit such as a nitrile (-CN), carboxyl (-COOH),

or an alkoxycarbonyl group ($-\text{COOR}^5$, wherein R^5 is an alkyl group of 1 to about 10 carbon

atoms which may be fluorinated or perfluorinated), and the like. It is preferred, in some embodiments, that perfluorinated compounds having no hydrogen atoms are used if excellent heat resistance is desired for the perfluoroelastomer resulting from curing the

25 perfluoropolymers as well as for preventing decrease in molecular weight due to chain transfer when synthesizing the perfluoroelastomer by polymerization reaction. Further, compounds

having a $\text{CF}_2=\text{CFO}-$ structure are preferred from the viewpoint of providing excellent polymerization reactivity with TFE.

[0055] Suitable cure site monomers may also include those having nitrogen-containing cure sites such as nitrile or cyano cure sites, for preferred crosslinking reactivity. However, cure sites (having multiple and varied backbones in addition to those noted above) and having carboxyl, COOH and other similar cure sites known in the art and to be developed may also be used. The cure site monomers may be used alone or in varied combinations.

[0056] Examples of perfluoropolymers and resulting elastomers formed therefrom using cure site monomers such as those noted above may be found in WO 00/29479 A1, incorporated herein in relevant part with respect to such perfluoroelastomers, their content and methods of making the same. Reference is also made to Japanese Kokai Patents No. H09-512569 A and H11-092529 A and to Published U.S. Patent Application No. US-2008-0287627-A1.

[0057] Perfluoropolymers for use in the compositions claimed herein may be synthesized using any known or to be developed polymerization technique for forming fluorine-containing elastomers using polymerization, including, for example, emulsion polymerization, latex polymerization, chain initiated polymerization, batch polymerization and others. Preferably, the polymerization is undertaken so that reactive cure sites are located either on either or both terminal ends of the polymer backbone and/or are depending from the main polymer backbone.

[0058] One possible method of making the polymers includes radical polymerization using an initiator such as those known in the art for polymerization of fluorine-containing elastomers (organic or inorganic peroxide and azo compounds). Typical initiators are persulfates, percarbonates, peresters and the like, with preferred initiators being include salts of persulfuric acid, oxidizing carbonates and esters, and ammonium persulfate, with the most preferred being ammonium persulfate (APS). These initiators may be used alone or with reducing agents, such as sulfites and sulfite salts.

[0059] A wide variety of emulsifiers for emulsion polymerization can be used, but preferred are salts of carboxylic acid having a fluorocarbon chain or a fluoropolyether chain, to suppress chain transfer reactions to the emulsifier molecules that occur during polymerization. The amount of emulsifier is generally used in amounts of about 0.05 to 2 weight percent, and preferably 0.2 to 1.5 weight percent, based on the added water. It is noted that a special arrangements should be used to avoid an ignition source, such as sparks, near the polymerization equipment. See, G.H. Kalb, Advanced Chemistry Series, 129, 12 (1973).

[0060] Polymerization pressure may vary, and can generally be in the range 0.5 to 7 MPa. The higher the polymerization pressure is, the higher the polymerization rate will be. Accordingly if productivity enhancement is desired, the polymerization pressure is preferably at least 0.7 MPa.

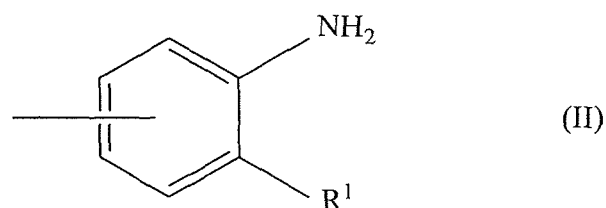
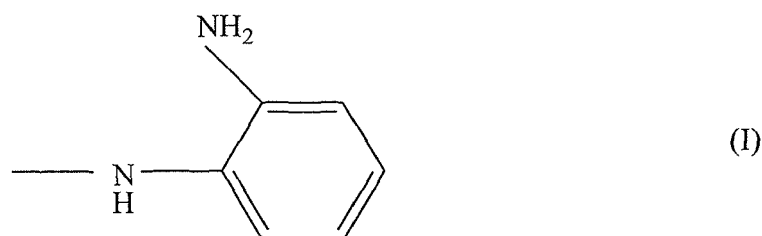
5 [0061] Standard polymerization procedures known in the art may be used. If a nitrogen-containing group, such as nitrile or cyano, a carboxyl group, or an alkoxycarbonyl group is to used in the curable perfluoropolymers herein, it may be included in the polymer by copolymerizing an additional monomer having the crosslinking site containing that group. The cure-site monomer may be added and copolymerized when preparing the fluorine-containing
10 elastomer. A further method for providing such a group to the polymer is by subjecting a polymerization product to an acid treatment to convert a group such as a metallic salt or ammonium salt of a carboxylic acid contained in the polymerization product to a carboxyl group. Examples of a suitable acid treatment method are washing with hydrochloric acid, sulfuric acid, nitric acid or fuming sulfuric acid or by decreasing a pH value of a mixture
15 system after the polymerization reaction to 3 or less by using the above-mentioned acids. Another method for introducing a carboxyl group is by oxidizing a crosslinkable polymer having iodine and bromine, with fuming nitric acid.

[0062] Uncured perfluoropolymers are commercially available, including perfluoropolymers sold by Dyneon, Daiel-Perfluor® and other similar polymers, available from
20 Daikin Industries, Ltd. of Osaka, Japan. Other suitable materials are available also from Solvay Solexis in Italy, Asahi Glass, Japan, and W.L. Gore.

[0063] Curing agents (also referred to herein as crosslinking agents) for use with various perfluoroelastomer compositions and elastomer-containing compositions of the present invention are for use with various cure sites described herein and should be capable of curing
25 (*i.e.*, capable of crosslinking) or otherwise undergoing a curing reaction with the functional group of the cure site monomer or cure site in the various uncured perfluoropolymers in the compositions to form crosslinks. Preferred crosslinking or curing agents are oxazole, imidazole, thiazole, triazine, amidoxime, and amidrazone crosslinking agents. Of these, imidazole is preferred in that crosslinked article providing excellent mechanical strength, heat
30 resistance, chemical resistance, cold resistance is achievable, particularly a cured article which is balanced and excellent with respect to heat resistance and cold resistance.

[0064] For nitrogen-containing cure sites, other curatives such as bisphenyl-based curatives and derivatives thereof, including bisaminophenol and its salts, and tetraphenyltin may be used. Examples of suitable curatives may be found, for example, in U.S. Patents Nos. 7,521,510 B2, 7,247,749 B2 and 7,514,506 B2, each of which is incorporated herein in relevant part with respect to the listing of various curatives for cyano-group containing perfluoropolymers. In addition, the perfluoropolymers may be cured using radiation-curing technology.

[0065] Most preferred are cyano-group containing cure sites cured with curatives that are aromatic amines having at least two crosslinkable groups as in formulas (I) and (II) below, or a combination thereof, which form benzoimidazole cross-linking structures upon cure. These curatives are known in the art and discussed in relevant part and with specific examples in U.S. Patents Nos. 6,878,778 and US 6,855,774, which are incorporated herein in their entirety.

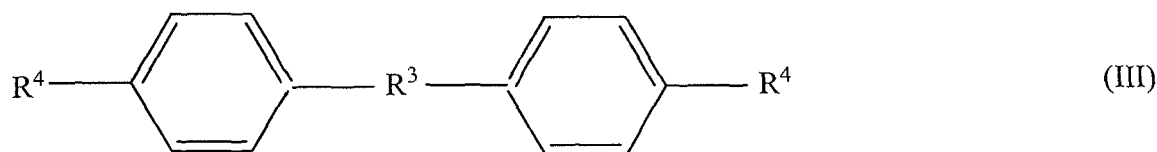


wherein R^1 is the same or different in each group according to formula (II) and may be NH_2 , NHR^2 , OH , SH or a monovalent organic group or other organic group such as alkyl, alkoxy, aryl, aryloxy, aralkyl and aralkyloxy of from about 1 to about 10 carbon atoms, wherein the non-aryl type groups may be branched or straight chain and substituted or unsubstituted and R^2 may be $-NH_2$, $-OH$, $-SH$ or a monovalent or other organic group such as an aliphatic hydrocarbon group, a phenyl group and a benzyl group, or alkyl, alkoxy, aryl, aryloxy, aralkyl and aralkyloxy groups, wherein each group is from about 1 to about 10 carbon atoms, wherein the non-aryl type groups may be branched or straight chain and substituted or unsubstituted. Preferred monovalent or other organic groups, such as alkyl and alkoxy (or perfluorinated

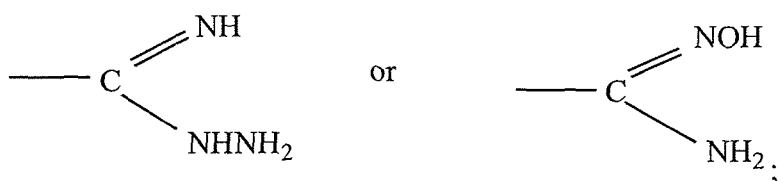
versions thereof) are from 1 to 6 carbon atoms, and preferred aryl type groups are phenyl and benzyl groups. Examples thereof include $-\text{CF}_3$, $-\text{C}_2\text{F}_5$, $-\text{CH}_2\text{F}$, $-\text{CH}_2\text{CF}_3$ or $-\text{CH}_2\text{C}_2\text{F}_5$, a phenyl group, a benzyl group; or a phenyl or benzyl group wherein 1 to about 5 of the hydrogen atoms are substituted by fluorine atoms such as $-\text{C}_6\text{F}_5$, $-\text{CH}_2\text{C}_6\text{CF}_5$, wherein groups may be further substituted, including with $-\text{CF}_3$ or other lower perfluoroalkyl groups, or, phenyl or benzyl groups in which 1 to 5 hydrogen atoms are substituted by CF_3 such as for example $\text{C}_6\text{H}_{5-n}(\text{CF}_3)_n$, $-\text{CH}_2\text{C}_6\text{H}_{5-n}(\text{CF}_3)_n$ (wherein n is from 1 to about 5). Hydrogen atoms may be further substituted with phenyl or benzyl groups. However, a phenyl group and CH_3 are preferred as providing superior heat resistance, good cross-linking reactivity and relatively easy synthesis.

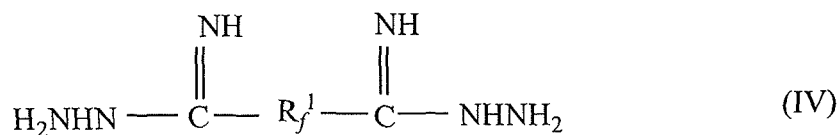
[0066] A structure having formula (I) or (II) incorporated in an organic amine should include at least two such groups of formula (I) or (II) such that at least two cross-linking reactive groups are provided.

[0067] Also useful herein are curatives having formulas (III), (IV) and (V) shown below.

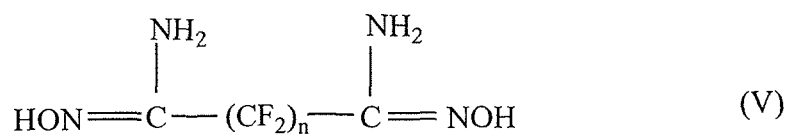


wherein R^3 is preferably SO , O or CO or an organic or alkylene type group, such as an alkyl, alkoxy, aryl, aralkyl or aralkoxy group of from one to six carbon atoms or perfluorinated versions of such groups, having from about one to about 10 carbon atoms, and being branched or straight chain, saturated or unsaturated, and branched or straight chain (with respect to the non-aryl type groups) or a single bond. R^4 is preferably a reactive side group such as those set forth below:





wherein R_f^1 is a perfluoroalkyl or perfluoroalkoxy group of from about 1 to about 10 carbon atoms that may be a straight or branched chain group and/or saturated or unsaturated and/or substituted or unsubstituted; and



wherein n is an integer of about 1 to about 10.

[0068] With respect to heat resistance, oxazole, imidazole, thiazole and triazine

crosslinking agents are preferred and can include the formula compounds listed below and discussed further below with respect to Formulae (I), (II), (III), (IV) and (V), specifically,

formula (II) wherein R^1 is the same or different and each is $-\text{NH}_2$, $-\text{NHR}^2$, $-\text{OH}$ or $-\text{SH}$,

wherein R^2 is a monovalent organic group, preferably not hydrogen; formula (III) wherein R^3

is $-\text{SO}_2-$, $-\text{O}-$, $-\text{CO}-$, and alkylene group of 1 to about 6 carbon atoms, a perfluoroalkylene

group of 1 to about 10 carbon atoms or a single bond and R^4 is as noted below; formula (IV)

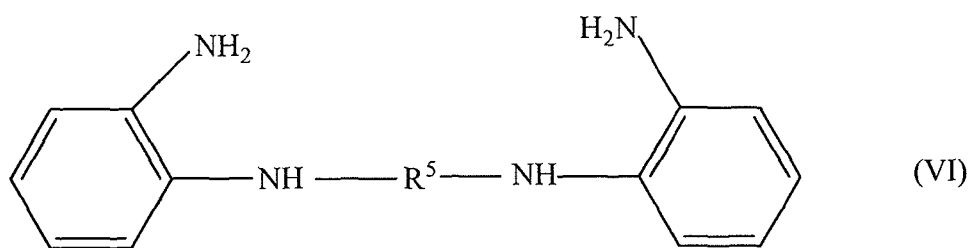
wherein R_f^1 is a perfluoroalkylene group of 1 to about 10 carbon atoms, and formula (V)

wherein n is an integer of 1 to about 10. Of such compounds, those of formula (II) as noted herein are preferred for heat resistance, which is enhanced due to stabilization of the aromatic rings after crosslinking. With respect to R^1 in the formula (II), it is preferred also to use -

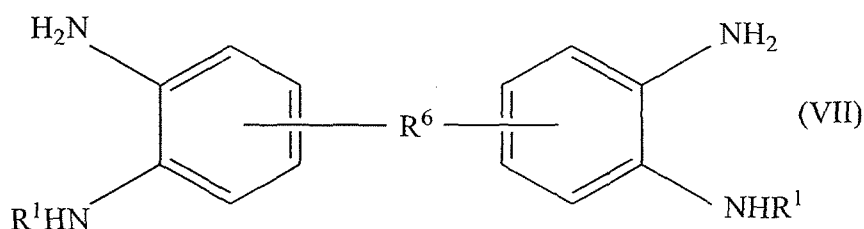
NHR^2 as R^1 , since an $\text{N}-\text{R}^2$ bond (wherein R^2 is a monovalent organic group and not hydrogen) is higher in oxidation resistance than an $\text{N}-\text{H}$ bond,

[0069] Compounds having at least two groups as in formula (II) are preferred and having 2 to 3 crosslinkable reactive groups thereon, more preferably having 2 crosslinkable groups.

[0070] Exemplary curatives based on the above preferred formulae include at least two functional groups, such as the following structures formula (VI), (VII) or (VIII):

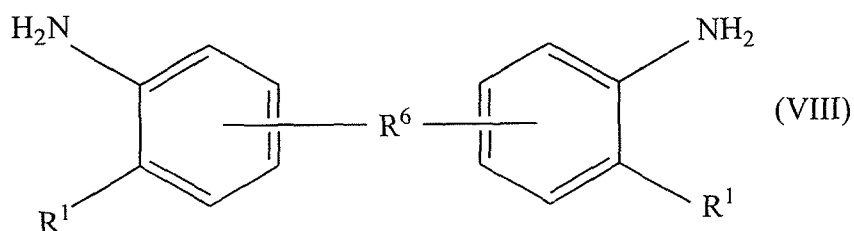


wherein R^5 represents a saturated or unsaturated, branched or straight chain, substituted or unsubstituted group such as alkyl, alkoxy, aryl, SO, O, CO, or similar groups which are perfluorinated with respect to the carbon atoms and which is preferably about 1 to about 10 carbon atoms;

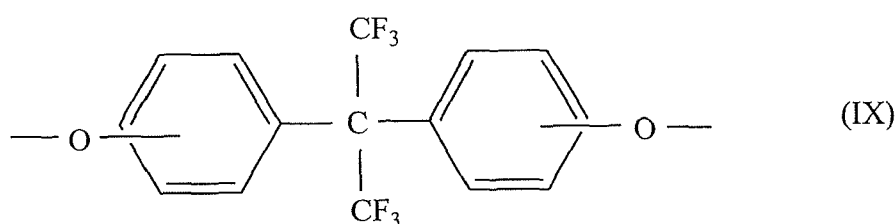


wherein R^1 is as defined elsewhere herein and R^6 may be O, SO₂, CO or an organic group which may be perfluorinated, such as alkyl, alkoxy, aryl, aryloxy, aralkyl and aralkyloxy of from about 1 to about 10 carbon atoms, wherein the non-aryl type groups may be branched or straight chain and substituted or unsubstituted, or a single or alkylene bond.

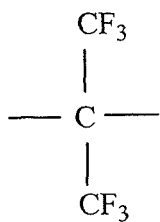
[0071] From the view of easy synthesis, preferred crosslinking agents as represented by formula (II) are shown below in formula (VIII).



wherein R^1 is as above and R^6 is $-\text{SO}_2$, $-\text{O}-$, $-\text{CO}-$, an alkylene group of 1 to about 6 carbon atoms, a perfluoroalkylene group of 1 to about 10 carbon atoms, a single bond or a group as shown in Formula (IX):

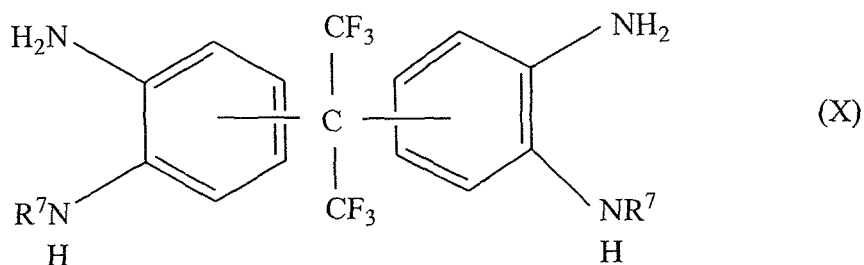


wherein this formula provides an easier the synthesis. Preferred examples of alkylene groups of from 1 to about 6 carbon atoms are methylene, ethylene, propylene, butylene, pentylene, hexylene and the like. Examples of perfluoroalkylene groups of 1 to about 10 carbon atoms are



and the like. These compounds are known as examples of bisaminophenyl

compounds. See, as a reference for example, the compounds in Japanese Patent No. 2-591177 B and Japanese Kokai Application No. 8-120146 A and similar patents. Preferred compounds according to this structure include those of formula (X):



wherein R^7 is the same or different in each instance and each R^7 is hydrogen, an alkyl group of 1 to about 10 carbon atoms; a partially fluorinated or perfluorinated alkyl group of 1 to 10 carbon atoms; a phenyl group; a benzyl group; or a phenyl or benzyl group in which 1 to about

5 hydrogen atoms have been replaced by fluorine or a lower alkyl or perfluoroalkyl group such as CF₃.

[0072] Non-limited examples of curatives include 2,2-bis(2,4-diaminophenyl)hexafluoropropane, 2,2-bis[3-amino-4-(N-methylamino)phenyl]hexafluoropropane, 2,2-bis[3-amino-4-(N-ethylamino)phenyl] hexafluoropropane, 2,2-bis[3-amino-4-(N-propylamino)phenyl] hexafluoropropane, 2,2-bis[3-amino-4-(N-phenylamino)phenyl]hexafluoropropane, 2,2-bis[3-amino-4-(N-perfluorophenylamino)phenyl]hexafluoropropane, 2,2-bis[3-amino-4(N-benzylamino)phenyl]hexafluoropropane, and similar compounds. Of these, for preferred excellent heat resistance properties, 2,2-bis[3-amino-4(N-methylamino)phenyl]hexafluoropropane, 2,2-bis[3-amino-4-(N-ethylamino)phenyl]hexafluoropropane, 2,2-bis[3-amino-4-(N-propylamino)phenyl]hexafluoropropane and 2,2-bis[3-amino-4-(N-phenylamino)phenyl]hexafluoropropane are preferred. For most preferred excellent heat resistant properties, 2,2-bis[3-amino-4-(N-phenylaminophenyl)]hexafluoropropane is preferred.

[0073] Other suitable curatives include oxazole, imidazole, thiazole, triazine, amidoxime and amidrazone crosslinking agents, and particularly bisaminophenols, bisamidines, bisamidoximes, bisamidrazones, monoamidines, monoamidoximes and monoamidrazones as known in the art or to be developed, examples of which are set forth, for example in U.S. Patent Publication No. 2004/0214956 A1, incorporated herein in relevant part by reference, including the curatives and co-curatives and accelerators therein. Imidazoles are useful in that they can provide good mechanical strength, heat resistance, chemical resistance, and low temperature capacity, as well as a good balance of crosslinking properties and high and low temperature properties. The bisamidoxime, bisamidrazone, bisaminophenol, bisaminothiophenol or bisdiaminophenyl curatives can react with nitrile or cyano groups, carboxyl groups, and/or alkoxycarbonyl groups in the perfluoropolymer to form a perfluoroelastomer preferred in some embodiments herein having an oxazole ring, a thiazole ring, an imidazole ring, or a triazine ring as crosslinks in the resulting cured articles formed from the compositions herein.

[0074] In one embodiment herein, a compound can be used including at least two chemical groups with cross-linking reactive groups as in Formula (I) or (II) in order to increase heat resistance and to stabilize an aromatic ring system. For groups such as in (I) or (II), having two

to three such groups, it is preferred to have at least two in each group (I) or (II), as having a lesser number of groups may not provide adequate cross-linking.

[0075] In one further embodiment, the curable perfluoroelastomeric composition includes at least two curable perfluoropolymers, a first perfluoropolymer and a second

perfluoropolymer, which have differing PAVE content, however, it should be understood that the invention can be practiced with only one curable perfluoropolymer and/or that additional such perfluoropolymers may be combined with a first and second perfluoropolymer, which may also have varying amounts of PAVE monomer in the polymer chain, provided that if two or more such polymers are used, it is preferred that at least the first and the second

perfluoropolymers (and in cured compositions the first and second perfluoroelastomers) have varying amounts of PAVE monomer in their polymeric chain as discussed herein. When using varying PAVE content, it is preferred to use a high-PAVE content curable perfluoropolymer and a low-PAVE content curable perfluoropolymer, alone or in a combination with one or more additional curable perfluoropolymers as use of two PAVES, which are both low-PAVE content polymers, while beneficial in some ways, contributes to increasing crosslinking time.

However, the better the PAVE differential approaching the optimum ranges described herein, the better and the shorter the crosslinking time will be. Such preferred differential PAVE content also contributes to the easy adjustment of hardness in end products, such as molded articles, formed from the fluorine-containing elastomer compositions.

[0076] When using one or more perfluoropolymers, it is preferred that at least one of or the first in a blend of curable perfluoropolymer preferably includes tetrafluoroethylene in an amount of about 0 to about 58.5 mole percent, and preferably about 49.8 to about 63.1 mole percent. A first perfluoroalkylvinyl ether (which may include at least one perfluoroalkylvinyl ether which can be used alone or in combination with other perfluoroalkylvinyl ethers), and at least one first cure site monomer having a cure site. The perfluoroalkylvinyl ether is preferably present in the first curable perfluoropolymer in an amount of about 31.5 to about 99.99, and preferably about 34 to about 49.75 mole percentage of the perfluoropolymer or about 38 to about 50 mole percent of the polymer. In one embodiment, in a fluorine-containing elastomer composition, the first perfluoropolymer is the high PAVE-content curable perfluoropolymer and the content of PAVE therein is preferably at least about 38% by mole, and more preferably at least about 40% by mole to enhance crosslinking speed of the composition. Corresponding in this embodiment, the PAVE content is preferably no greater than about 50% by mole, more

preferably no greater than about 45% by mole and most preferably no greater than about 42% by mole so as to increase the polymerization rate in synthesizing the polymer. Other perfluoropolymers known in the art and having the same or varied amounts of TFE/PAVE may also be used.

- 5 [0077] Most preferably, the first curable perfluoropolymer has a molar percentage ratio of tetrafluoroethylene to the first perfluoroalkylvinyl ether(s) in the polymer chain of about 0:100 to about 65:35, and more preferably from about 50:50 to about 65:35.

[0078] The at least one first cure site monomer having a cure site is preferably a single cure site monomer, but combinations of cure site monomers having the same functional active cure
10 group or varying types of cure site monomers having differing cure site groups (as in a dual cure composition) may also be used herein.

[0079] If a blend of perfluoropolymers having varying PAVE content are used, then in such embodiment, the second curable perfluoropolymer preferably includes a higher content of tetrafluoroethylene than the first curable perfluoropolymer, and more preferably about 65 to
15 about 85.5 mole percentage of tetrafluoroethylene, and most preferably about 64.7 to about 82.5 mole percent in the second curable polymer. The second perfluoroalkylvinyl ether in the second curable perfluoropolymer, which is the low-PAVE content perfluoropolymer may also be one or more perfluoroalkylvinyl ethers used alone or in combination, and the second perfluoroalkylvinyl ether(s) may be the same or different from the first perfluoroalkylvinyl
20 ether(s) in the first curable perfluoropolymer. The second perfluoroalkylvinyl ether is preferably present at about 4.5 to about 35 mole percent, and more preferably about 14.6 to about 34.83 mole percent of the second perfluoropolymer.

[0080] In one embodiment, in a fluorine-containing elastomer composition, the second curable PAVE is present in the polymer preferably in an amount of at least about 18% by mole,
25 more preferably at least about 21% by mole and most preferably at least about 25% by mole, so as to contribute to a lower glass transition temperature and satisfactory low temperature properties. In that embodiment, the PAVE content is also preferably no greater than about 35% by mole, and more preferably no greater than about 32% by mole and most preferably no greater than about 30% by mole so as to contribute to an increase in hardness of resulting
30 crosslinked articles formed from curing the fluorine-containing elastomer composition and for enhancing sealing properties of such articles.

[0081] Most preferably, the second curable perfluoropolymer has a molar percentage ratio in the polymer chain of tetrafluoroethylene to the second perfluoroalkylvinyl ether(s) of about 65:35 to about 95:5, and more preferably about 65:35 to about 85:15.

[0082] It is most preferred, when using a varying PAVE blend of curable

5 perfluoropolymers, that the difference in PAVE content in the first curable perfluoropolymer and the second curable perfluoropolymer is at least 5 molar percent to provide adjustment of hardening of the crosslink. It is preferred, however, that the difference in the PAVE content between the first and second curable perfluoropolymers be at least about 8 molar percent, and more preferred that it is at least about 10 molar percent. It is further preferred that the difference
10 in content in PAVE content is less than about 25 molar percent, more preferably less than about 15 molar percent and most preferably less than about 10 molar percent to avoid increases in the glass transition point.

[0083] The second curable perfluoropolymer preferably also comprises at least one second cure site monomer having a cure site. The second cure site monomer(s) may be the same or

15 different from the at least one first cure site monomer(s) used in the first curable perfluoropolymer, although it is preferred that the first and second cure site monomer(s) are either of the same type (meaning that they are the same or have the same cure site functional group(s)) or are capable of being cured by the same curative for convenience and compatibility. Although it should be understood by those skilled in the art, based on this disclosure that dual
20 cure materials may be used, or varying cures between the first and second curable polymers within the scope of the invention, provided that preferably adequate curing is obtained through use of the appropriate curatives.

[0084] In the curable perfluoropolymers herein, the polymers preferably include the cure site monomer(s) in amounts of about 0.01 to about 10 molar percentage of the polymer chains,

25 and more preferably about 0.05 to about 3 molar percent. While various cure site functional groups may be used within the scope of the invention, it is preferred that the cure sites of the at least one cure site monomer each is a functional group that is a nitrogen-containing group such as nitrile, a carboxyl group or an alkoxycarbonyl group. The monomers may be configured so that the first and/or the second cure site monomer(s) provide a functional group such as a
30 nitrogen-containing group on one or two of the terminal ends of the first or second curable perfluoropolymer, respectively. Alternatively, or in addition to terminal end placement of groups, such cure site groups having nitrogen may also be situated so as to depend from the

polymer backbone of either the first and/or the second curable perfluoropolymer. However, as noted elsewhere herein, it is within the scope of the invention to use a variety of cure site functional groups as are already known or to be developed in the art.

[0085] In one preferred embodiment, in a fluorine-containing elastomer composition, the content of the at least one cure site monomer is at least about 0.1 % by mole, more preferably at least about 0.2% by mole and most preferably at least about 0.3% by mole in order to provide enhanced crosslinkability. Further, in such embodiment, the at least one cure site monomer is no greater than about 2.0% by mole, more preferably no greater than about 1.0% by mole and most preferably no greater than about 0.5% by mole to avoid use of excessive cure site monomer due to expense associated therewith.

[0086] The curable perfluoropolymer compositions described herein are preferably a combination of two curable perfluoropolymers, but the invention may also include within its scope the addition of further such curable perfluoropolymers without departing from the spirit of the invention.

[0087] In addition to the curable perfluoropolymers described above, the curable perfluoroelastomer composition preferably also includes at least one curative(s) which is/are capable of curing the at least one first cure site monomer and the at least one second cure site monomer. Preferably, if a functional-group containing cure site is used in the curable perfluoropolymer, the at least one curative is selected so as to react with the functional group(s) of the cure site monomer(s) in order to form cross-linking structures such as benzoimidazole cross-linking structures. Suitable curatives are as noted elsewhere herein and are included in the curable perfluoroelastomeric composition in amounts of about 0.3 to about 10 parts by weight per hundred parts of the curable composition based on the total weight of the curable perfluoropolymer(s), and more preferably about 0.6 to about 0.9 parts by weight per hundred parts of the curable perfluoroelastomeric composition thereof. The preferred range is most beneficial to provide good strength characteristics and to avoid cracking or structural defects when under high-temperature compression forces. It also provides preferred compression set characteristics.

[0088] In one embodiment of the fluorine-containing elastomers herein, the curative is one of those listed as preferred above, and is present in an amount by weight of at least about 0.3 parts by weight, and more preferably at least about 0.5 parts by weight or at least about 0.7 parts by weight, and most preferably at least about 0.6 parts by weight based on 100 parts by

weight of elastomer (in this case, uncured polymer) in the composition, with greater amounts enhancing crosslinking. The curative or crosslinking agent is preferably no greater than 10.0 parts by weight, and more preferably no greater than 2.0 parts by weight and most preferably no greater than 0.9 parts by weight based on 100 parts by weight of the elastomers in the composition.

[0089] In addition to the preferred curatives noted herein for use with fluorine-containing curable perfluoropolymers having nitrile groups and the like, it is within the scope of the invention to cure the nitrile groups using curatives known in the art for the perfluoropolymers added to the compositions herein. Examples of other curatives known in the art include organotins such as tetraphenyltin, triphenyltin and the like (as these compounds form preferred triazine rings). If an organotin compound is used, it is preferred to be present in an amount of about 0.05 to about 10 parts by weight, more preferably about 1 to about 5 parts by weight, based on 100 parts by weight of the curable perfluoropolymers in the composition. If the organotin is present in an amount of less than about 0.05 parts, there is a tendency for the polymer to not sufficiently crosslink upon curing and if the amount is more than about 10 parts, physical properties of the formed articles tend to deteriorate.

[0090] In addition to the above-described curable perfluoroelastomeric composition, described herein are cured perfluoroelastomeric composition includes at least a first cured perfluoroelastomer and a second cured perfluoroelastomers, which cured elastomers are formed from the at least one, and preferably at least two of the above-described curable perfluoropolymers in the composition after the cure is complete such that the curative(s) in the curable perfluoroelastomeric composition has/have been substantially reacted and incorporated into the cross-linked cured perfluoroelastomeric composition.

[0091] It is within the scope of the invention to combine these curable materials in varying amounts so that the weight percentage ratio of the first perfluoroelastomer to the second perfluoroelastomer in the composition is about 1:99 to about 99:1, more preferably about 20:80 to about 80:20, and most preferably about 75:25 to about 45:55.

[0092] Such cured perfluoroelastomer compositions formed from curable perfluoroelastomeric compositions as noted herein may be cured and shaped so as to form a molded article(s). Generally, the molded articles will be formed as sealing members such as O-rings, seals, gaskets, inserts and the like, but other shapes and uses known or to be developed in the art are contemplated herein.

[0093] The molded article may be bonded to a surface for forming, for example, bonded seals. Such bonded seals may be used, for example for forming pre-bonded doors, gates, and slit valve doors for use, e.g., in semiconductor processing. The surfaces to which such molded articles, such as seals may be bonded include polymeric surfaces as well as metal and metal alloy surfaces. In one embodiment, the invention includes a gate or slit valve door formed of, e.g., stainless steel or aluminum, to which an O-ring seal conforming to a recess in the door configured for receiving the seal. The bonding may occur through use of a bonding composition or through an adhesive. Further, a bonding agent may be prepared which is formed of a fluorosolvent, such as one of several Fluorinert® solvents from 3M capable of dissolving a perfluoropolymer, at least one curable perfluoropolymer and at least one suitable curative capable of crosslinking the cure site on the curable perfluoropolymer.

[0094] The bonding agent may be applied to the O-ring or the recess of the door either after initial curing of an extruded polymer in a mold for making an O-ring and prior to bonding the seal to a surface such as a door, or the bonding agent can be applied to an extruded polymer which can be molded and cured in situ in the surface (door) to which it is to be bonded so that upon heat curing, the perfluoropolymers are cured in the O-ring and also within the bonding agent at the same time. Preferably, although not necessarily, the perfluoropolymer used in the bonding agent is the same as at least one of the perfluoropolymers in the perfluoroelastomer compositions described herein. The bonding agent may also preferably include both perfluoropolymers used in the curable perfluoroelastomeric compositions described herein and/or can be useful using any suitable curable perfluoropolymer capable of curing to bond the compositions to the intended surface.

[0095] In preparing the fluorine-containing compositions herein, the compositions preferably include barium titanate with an average particle size of at least 200 nm. In preferred embodiments, the average particle size is about 300 nm to about 1200 nm, or about 500 nm to about 1000 nm. However, the barium titanate may also be present in a mixture of at least two different average particle sizes.

[0096] In an embodiment herein, the barium titanate is preferably present in the fluorine-containing elastomer composition in an amount of about 1 to about 200 parts by weight of barium titanate per 100 parts by weight of the curable perfluoropolymer, more preferably about 1 to about 100 parts by weight of barium titanate, and most preferably 1 to about 50 parts by weight of the barium titanate per 100 parts by weight of the curable perfluoropolymer. In

another embodiment, the barium titanate is preferably present in an amount of about 5 to about 200 parts by weight of barium titanate per 100 parts by weight of the perfluoropolymer, more preferably about 5 to about 100 parts by weight barium titanate, yet more preferably about 5 to about 50 parts by weight barium titanate, even more preferably about 5 to about 30 parts by weight of the barium titanate, and most preferably about 10 to about 20 parts by weight of the barium titanate, each range being based on 100 parts by weight of the curable perfluoropolymer.

[0097] Also described herein is a method for making a cured perfluoroelastomeric composition as described hereinabove. In the method, a curable perfluoroelastomeric composition is prepared by combining at least one curable perfluoropolymer as described elsewhere herein and at least one curative capable of curing the cure site of the at least one first and second cure site monomer(s) along with barium titanate in amounts noted elsewhere herein.

[0098] The polymers and barium titanate may be combined using typical rubber processing equipment such as an open roll, Banbury mixer, kneader or the like. The compositions may also be prepared using a method of a closed mixer and a method of coagulation through emulsion mixing. Preferably a typical mixer, such as a two-roll mixer as is typically used for combining perfluoropolymers (also referred to as perfluoroelastomer gum). Preferably, in this method, the polymers are mixed at room temperatures, or at elevated temperatures of about 30°C to about 100°C, and preferably about 50°C.

[0099] If desired, additives may also be admixed into the composition at this point. Additives are optional and not required due to the unique nature of the interaction of the first and second curable perfluoropolymers and/or due to the unique filler system having barium titanate. However, if desired to alter certain properties, cure accelerators, co-curatives, co-agents, processing aids, plasticizers, fillers and modifiers such as silica, fluoropolymers (TFE and its melt-processible copolymers as well as core-shell modified fluoropolymers as are known in the art in micropowder, pellet, fiber and nanopowder forms), fluorographite, silica, barium sulfate, carbon, carbon black, carbon fluoride, clay, talc, metallic fillers (titanium oxide, aluminum oxide, yttrium oxide, silicon oxide), metal carbides (silicon carbide, aluminum carbide), metallic nitrides (silicon nitride, aluminum nitride), other inorganic fillers (aluminum fluoride, carbon fluoride), colorants, organic dyes and/or pigments, such as azo, isoindolenone, quinacridone, diketopyrrolopyrrole, anthraquinone, and the like, imide fillers (such as polyimide, polyamide-imide and polyetherimide), ketone plastics (such as polyarylene ketones

(PAEK polymers) including PEEK, PEK and PEKK), polyarylates, polysulfones, polyethersulfones, polyphenylene sulfides, polyoxybenzoate, and the like may be used in amounts known in the art and/or which may be varied for different properties. All of the fillers herein may be used alone or in combinations of two or more such fillers and additives.

5 [0100] Preferably, any optional fillers used total less than about 50 parts, and most preferably less than about 30 parts per hundred parts of the combined curable perfluoropolymers in the composition. Organic fillers, providing heat resistance, and plasma resistance (reduced numbers of particles and low weight reduction rates at emission of plasma), include of those mentioned above, organic pigments, imide fillers with imide structures such as
10 polyimide, polyamide imide and polyetherimide, and ketone-based engineering plastics like PEEK, PEKK and PEK, with organic pigments being preferred.

[0101] Pigmented fillers which are preferred for heat resistance and chemical resistance and having less effect on end characteristics of the molded articles formed from the compositions described herein include quinacridone, diketopyrrolopyrrole and anthraquinone pigments and
15 dyes, with quinacridone being preferred.

[0102] Of the additional inorganic fillers, preferred fillers for shielding plasma effects include aluminum oxide, yttrium oxide, silicon oxide, polyimide and carbon fluoride.

[0103] Preferably after the polymers are combined, the first and second curable perfluoropolymers in the perfluoroelastomeric composition are cured at least partially, and
20 preferably substantially and to as complete an extent as is possible as described further herein with or without optional postcure, to form a cured perfluoroelastomeric composition as described herein.

[0104] Depending on the cure sites and curative, various cross-link structures can be formed upon curing. Preferably, functional cure groups are used on the cure site monomers, so
25 that the cured perfluoroelastomeric composition includes a curative(s) which form benzoimidazole cross-linking structures.

[0105] The curable perfluoroelastomeric composition is preferably cured at temperatures and for times sufficient to allow the curing reaction to proceed until the curable perfluoropolymers in the composition are substantially cured, preferably at least 70% cured.

30 Preferred curing temperatures and times are about 150°C to about 250°C, for about 5 to about 40 minutes. Following curing, an optional postcure may be used. Acceptable postcure temperatures and times are about 250°C to about 320°C for about 5 to about 48 hours.

[0106] While curing, the curable perfluoroelastomeric compositions described herein can be formed into a molded article while simultaneously curing using heat and pressure applied by to a mold. Preferably, the combined curable perfluoropolymers are formed into a preform, such as an extruded rope or other shape useful for including the preform in a mold having a recess
5 shaped to receive the preform and for forming a molded article while curing.

[0107] In addition to fillers, it is within the scope of the invention to include additional curable and noncurable perfluoropolymers having varied types, including the same or different cure site monomers to those preferred herein. Additional curatives and cure accelerators, either to work with or accelerate the cure of the first perfluoropolymer and/or the second
10 perfluoropolymer or to cure and/or accelerate cure of any additional optional curable perfluoropolymers may also be included herein. Noncurable perfluoropolymers include those that lack a reactive cure site and are formed from one or more ethylenically unsaturated monomers (such as TFE, HFP and PAVE). Additional curable perfluoropolymers may be any of the curable perfluoropolymers noted herein as well as those having cure sites suitable for
15 crosslinking with curing systems such as organic peroxide and the like as described above and as are known in the art, *e.g.*, tetraphenyl tin cures, bisaminophenyl-based cures and the like. Such polymers may be added to develop alternative blends and to modify the properties of the compositions noted herein.

[0108] The perfluoroelastomers of the invention are alternatives to and generally show
20 improved properties in comparison to semicrystalline fluoroplastic-filled FFKM compositions as used in the prior art and are improvements over plasma-resistant compositions of the present applicants based on blends of perfluoroelastomers having varying amounts of PAVE. The compositions can be made without additional use of such fluoroplastic particle fillers and without the need for high temperature mixing. When blends are used, the higher TFE content
25 of the second polymer while allowing the cured perfluoropolymer in the composition to remain amorphous, changes the nature of the other perfluoropolymer such that the higher TFE content perfluoropolymer acts in the role of a "filler" in the other curable perfluoropolymer. Thus, the molded articles produced by the elastomeric compositions of the present invention are more resilient to cracking under harsh chemical, thermal and plasma conditions.

[0109] In contrast to the prior art, the blended perfluoropolymer compositions having
30 varying PAVE content as discussed herein produce an FFKM composition at a molecular level that results in desired properties for intended uses without the need for additional fillers.

Further, in contrast to the prior art, the compositions of the present invention, because they do not require a semicrystalline polymer component and remain in an amorphous state if unfilled, can be easily processed. Such compositions are further improved using the barium titanate filler herein.

5 [0110] As discussed, when using a blend of perfluoropolymers of varying PAVE content, the amorphous, high-TFE curable perfluoropolymer in the composition is believed on theory to be instrumental to achieving the basic desired properties of the resultant perfluoroelastomeric composition. The mole percentage of TFE in the high TFE-content curable perfluoropolymer in the composition should not exceed about 95%, and particularly should avoid approaching the
10 crystalline point where a melting point would be discernible. The crosslinked elastomeric compositions and molded articles prepared therefrom display excellent thermal resistance with very low compression set. In addition, due to their high purity and excellent plasma resistance, they can be used for semiconductor sealing applications.

[0111] The hardness of preferred cross-linked perfluoroelastomer compositions formed
15 from blended perfluoropolymers as described herein may be from about 40 to about 95 Shore A hardness, but is preferably at least about 50 Shore A, and more preferably at least about 55 Shore A, and most preferably at least about 60 Shore A. It is further preferred that hardness is no greater than about 95 Shore A, more preferred that it is no greater than about 90 Shore A and most preferred that it is no greater than about 85 Shore A. Such preferred hardness values
20 provide more and increasingly superior sealing characteristics.

[0112] The resulting cured perfluoroelastomer compositions described herein also have superior chemical resistance, plasma resistance, good mechanical strength and heat resistance. It is also possible to adjust the hardness levels of the resulting perfluoroelastomer composition, with or without use of fillers, by using varying combinations of the perfluoropolymers noted
25 herein. The outgassing component released from the resulting perfluoroelastomer compositions may also be reduced thereby assisting in avoidance of environmental pollution. It is also, therefore, useful for sealing semiconductor equipment, as an O-ring, square ring, corner ring, gasket, packing, oil seal, pairing seal, lip seal, door seal and the like. Such seals and related gasketing products can be used in various types of semiconductor processing equipment for
30 providing semiconductor products having higher demands in manufacture and clarity, such as liquid crystal or plasma panel displays.

[0113] Exemplary equipment in which sealing products formed from the perfluoropolymer compositions herein may be used include etching equipment, such as dry-etching, plasma-etching, reactive ion-etching, reactive ion beam-etching, sputter-etching, ion beam etching, wet etching and ashing equipment; cleaning apparatuses such as dry-etching cleaning, UV/O₃ cleaning, ion beam cleaning, laser beam cleaning, plasma cleaning and gas etching cleaning apparatuses; extractor cleaning apparatuses, such as Soxhlet extraction cleaning, high-temperature, high-pressure extractor cleaning, microwave extractor cleaning, and supercritical extractor cleaning apparatuses; exposure devices such as steppers and coater developers; polishing apparatuses such as CMP equipment; coating equipment such as CVD and sputtering equipment; and diffusion - ion implantation equipment such as oxidation diffusion equipment and ion implantation equipment.

[0114] In one preferred embodiment, the present invention relates to a fluorine-containing elastomer composition comprising two or more kinds of perfluoroelastomers, referred to herein as perfluoroelastomers (A), having different contents of a perfluoroalkylvinyl ether (PAVE) unit (a) and including barium titanate.

[0115] In the two or more kinds of perfluoroelastomers (A), there is a difference in the content of PAVE unit (a) between any two kinds of perfluoroelastomers (A) that is preferably not less than 5 % by mole, more preferably not less than 8 % by mole, further preferably not less than 10 % by mole, from the viewpoint of easy adjustment of hardness of a crosslinked article. In addition, the difference in a content of the PAVE unit (a) in each kind of perfluoroelastomer (A) is preferably not more than 25 % by mole, more preferably not more than 20 % by mole, further preferably not more than 15 % by mole, in that a glass transition temperature of the perfluoroelastomer having a smaller content of PAVE unit (a) is not elevated. In such compositions it is further preferred that a blend of more than one particle size of barium titanate is used.

[0116] In addition, in such embodiment, in any two kinds of perfluoroelastomers (A) among two or more kinds of perfluoroelastomers (A), assuming that the perfluoroelastomer having the larger content of PAVE unit (a) is referred to as a perfluoroelastomer (A1) and the perfluoroelastomer having the smaller content of PAVE unit (a) is referred to as a perfluoroelastomer (A2), the content of PAVE unit (a) in the perfluoroelastomer (A1) is preferably not less than 38 % by mole, more preferably not less than 40 % by mole, in that a crosslinking speed of the composition is faster. Also the content of PAVE in the

perfluoroelastomer (A1) is preferably not more than 50 % by mole, more preferably not more than 45 % by mole, further preferably not more than 42 % by mole, in that a polymerization rate in synthesizing a polymer is higher.

[0117] In this embodiment, the content of PAVE in the perfluoroelastomer (A2) is preferably not less than 18 % by mole, more preferably not less than 21 % by mole, further preferably not less than 25 % by mole, from the viewpoint of a low glass transition temperature and satisfactory low temperature properties. Also the content of PAVE in the perfluoroelastomer (A2) is preferably not more than 35 % by mole, more preferably not more than 32 % by mole, further preferably not more than 30 % by mole, from the viewpoint of increase in hardness of a crosslinked article, thus enhancing sealing property of the sealing material.

[0118] Further, in this embodiment, a third perfluoroelastomer other than the perfluoroelastomers (A1) and (A2) may be mixed into the composition.

[0119] If only perfluoroelastomers having a smaller content of PAVE unit are used, a crosslinking time becomes longer. On the contrary, according to the present invention, as mentioned above, shortening of a crosslinking time can be achieved by combining at least two kinds of perfluoroelastomers, that is, one having a larger content of PAVE unit and another one having a smaller content of PAVE unit. Further hardness of an obtained molded article can be easily adjusted by combining such two kinds of perfluoroelastomers.

[0120] In this case, examples of PAVE are, as set forth in the above-noted embodiments, for instance, perfluoromethylvinyl ether (PMVE), perfluoropropylvinyl ether (PPVE) and the like. These can be used alone or can be used in optional combination thereof to such an extent not to impair the effect of the present invention.

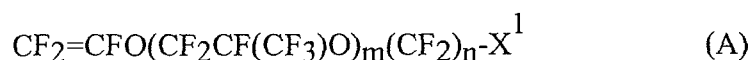
[0121] Among these, PMVE is preferable from the viewpoint of an excellent mechanical strength of a cured article.

[0122] It is also preferable in this embodiment, that the perfluoroelastomer (A) further contains a monomer unit (b) having at least one kind selected from the group consisting of a nitrile group, a carboxyl group and an alkoxycarbonyl group.

[0123] From the viewpoint of enhancing crosslinkability of the crosslinkable elastomer, the content of monomer unit (b) in perfluoroelastomer (A) is not less than 0.1 % by mole, preferably not less than 0.2 % by mole, more preferably not less than 0.3 % by mole. In addition, the content of monomer unit (b) in the perfluoroelastomer (A) is not more than 2.0 %

by mole, preferably not more than 1.0 % by mole, more preferably not more than 0.5 % by mole in that the amount of expensive monomer unit (b) can be reduced.

[0124] Examples of the monomer unit (b) are, for instance, monomers represented by the formula (A) as noted hereinabove:



wherein m is 0 or an integer of 1 to 5, n is an integer of 1 to 3, X^1 is a nitrile group, a carboxyl group or an alkoxycarbonyl group. These can be used alone or can be used in optional combination thereof.

[0125] The nitrile group, carboxyl group or alkoxycarbonyl group can function as a cure site. In addition, from the viewpoint of excellent crosslinkability, the monomer unit (b) is preferably a nitrile group-containing monomer in which a cure site is a nitrile group.

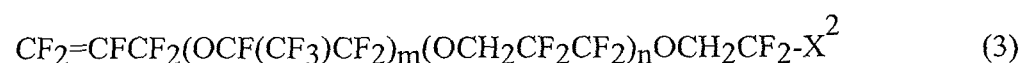
[0126] Examples of the monomer unit (b) are monomers represented by the formulae (1) to (17) in a manner as noted above:



wherein Y is hydrogen atom or fluorine atom, n is an integer of 1 to 8,



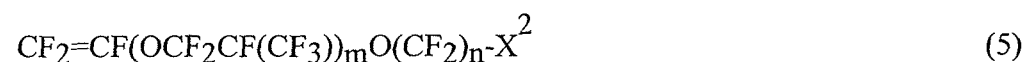
where R_f^2 is $-(\text{OCF}_2)_n-$ or $-(\text{OCF}_2)_n-$, n is 0 or an integer of 1 to 5,



wherein m is 0 or an integer of 1 to 5, n is 0 or an integer of 1 to 5,



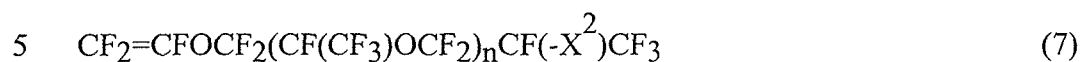
wherein m is 0 or an integer of 1 to 5, n is 0 or an integer of 1 to 5,



wherein m is 0 or an integer of 1 to 5, n is an integer of 1 to 8,



wherein m is an integer of 1 to 5,



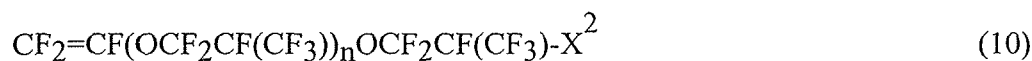
wherein n is an integer of 1 to 4,



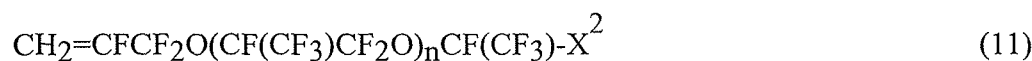
wherein n is an integer of 2 to 5,



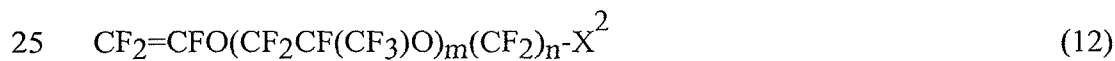
wherein n is an integer of 1 to 6,



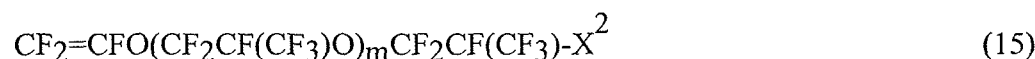
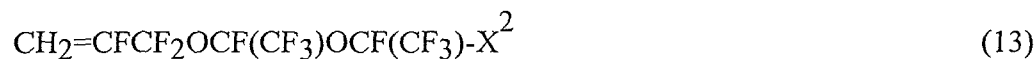
wherein n is an integer of 1 to 2,



wherein n is 0 or an integer of 1 to 5,



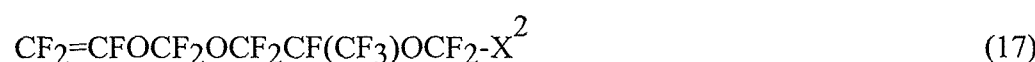
wherein m is 0 or an integer of 1 to 5, n is an integer of 1 to 3,



wherein m is an integer of not less than 0,



wherein n is an integer of not less than 1,



in which in the formulae (1) to (17), X^2 is a nitrile group (-CN group), a carboxyl group (-COOH group) or an alkoxycarbonyl group (-COOR⁵, wherein R⁵ is an alkyl group having 1 to 10 carbon atoms which may have fluorine atom). Among these, perfluorinated compounds containing no hydrogen atom are preferable from the viewpoint of excellent heat resistance of the perfluoroelastomer (A) and for preventing decrease of a molecular weight due to chain transfer when synthesizing the perfluoroelastomer by polymerization reaction. In addition, a compound having a CF₂=CFO- structure is preferable from the viewpoint of excellent polymerization reactivity with tetrafluoroethylene.

[0127] Examples of such perfluoroelastomers (A) are those disclosed in Japanese Kokai No. 9-512569A, International Application WO 00/29479, and Japanese Kokai No. 11-92529A, etc.

[0128] Those perfluoroelastomers (A) in this embodiment can be prepared by known methods.

[0129] The radical polymerization initiator used in this embodiment of the present invention may be one that is used for polymerization of fluorine-containing rubbers, and examples thereof are organic and inorganic peroxides and azo compounds. Represented initiators are persulfates, percarbonates, peresters and the like, and a preferable initiator is APS. APS may be used alone or can be used in combination with reducing agents such as sulfites and sulfates.

[0130] As noted elsewhere herein, the emulsifier used for emulsion polymerization can be selected from a wide range, and from the viewpoint of inhibiting a chain transfer reaction to the emulsifier molecules which occurs during the polymerization, salts of carboxylic acid having a fluorocarbon chain or a fluoropolyether chain are preferable. The amount of the emulsifier is preferably about 0.05 to 2 % by weight, particularly preferably 0.2 to 1.5 % by weight based on the added water.

[0131] The monomer mixture gas used in the present invention is explosive as described in Advances in Chemistry Series, G.H. Kalb, et al., 129, 13 (1973), and therefore it is necessary to take any measures for polymerization equipment not to cause sparking which becomes an ignition source.

[0132] The polymerization pressure can be changed in a wide range, and generally is within a range from 0.5 to 7 MPa. The higher the polymerization pressure is, the higher a

polymerization rate is. Accordingly from the viewpoint of enhancement of productivity, the polymerization pressure is preferably not less than 0.7 MPa.

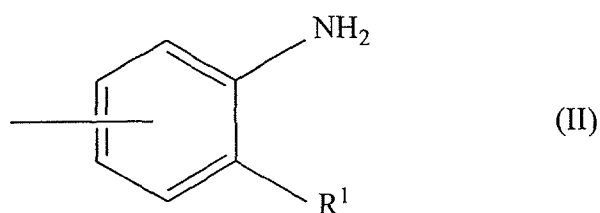
[0133] For introducing at least one selected from the group consisting of a nitrile group, a carboxyl group and an alkoxycarbonyl group to the fluorine-containing elastomer used in the present invention, as mentioned above, there is a method of copolymerizing by adding a monomer having a cure site when preparing the fluorine-containing elastomer in this embodiment. An example of another method is a method of subjecting a polymerization product to acid treatment to convert a group such as a metallic salt or ammonium salt of a carboxylic acid contained in the polymerization product to carboxyl group. An example of a suitable acid treatment method is a method of washing with hydrochloric acid, sulfuric acid or nitric acid or a method of decreasing a pH value of a mixture system after the polymerization reaction to 3 or less by using the mentioned acid.

[0134] In addition, it is possible to introduce a carboxyl group by oxidizing a crosslinkable elastomer containing iodine or bromine with a fuming nitric acid.

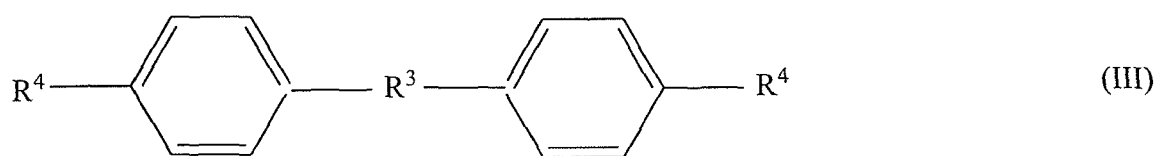
[0135] It is preferable that the fluorine-containing elastomer composition of this embodiment of the present invention comprises the crosslinking agent (B) crosslinkable with the group of the above-mentioned fluorine-containing elastomer being capable of acting as a cure site.

[0136] The crosslinking agent(s) (B) used in the present invention is at least one crosslinking agent selected from the group consisting of an oxazole crosslinking agent, an imidazole crosslinking agent, a thiazole crosslinking agent, a triazine crosslinking agent, an amidoxime crosslinking agent and an amidrazone crosslinking agent. Of these, an imidazole crosslinking agent is preferable in that a crosslinked article having excellent mechanical strength, heat resistance, chemical resistance and cold resistance, particularly a crosslinked article being excellent in heat resistance and cold resistance in good balance can be provided.

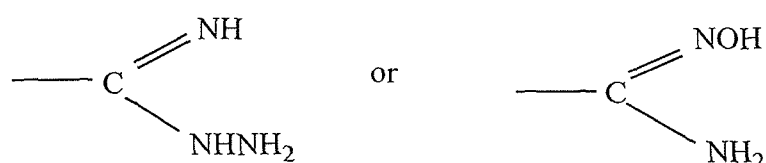
[0137] From the viewpoint of heat resistance, the preferred examples of an oxazole crosslinking agent, an imidazole crosslinking agent, a thiazole crosslinking agent and a triazine crosslinking agent is at least one compound selected from the group consisting of a compound containing at least two crosslinkable reaction groups represented by the formula (II) as noted above:



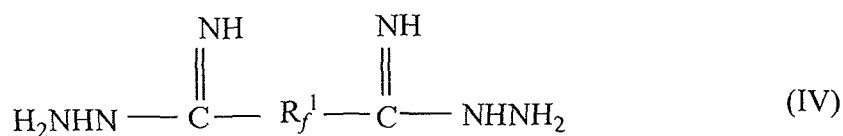
wherein R^1 groups are the same or different and each is $-NH_2$, $-NHR^2$, $-OH$ or $-SH$; R^2 is a
 5 monovalent organic group,
 a compound represented by the formula (III):



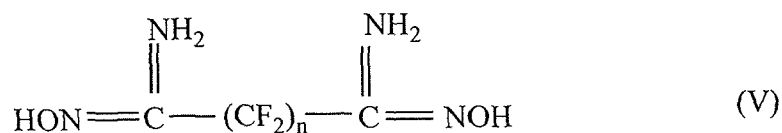
wherein R^3 is $-SO_2-$, $-O-$, $-CO-$, an alkylene group having 1 to 6 carbon atoms, a
 perfluoroalkylene group having 1 to 10 carbon atoms or a single bond; R^4 is



a compound represented by the formula (IV):



in which R_f^1 is a perfluoroalkylene group having 1 to 10 carbon atoms, and a compound
 represented by the formula (V):



in which n is an integer of 1 to 10.

[0138] Of these compounds, as with other embodiments noted herein, the compound containing at least two crosslinkable reaction groups represented by the formula (II) is preferable in that heat resistance is enhanced due to stabilization by aromatic rings after the crosslinking.

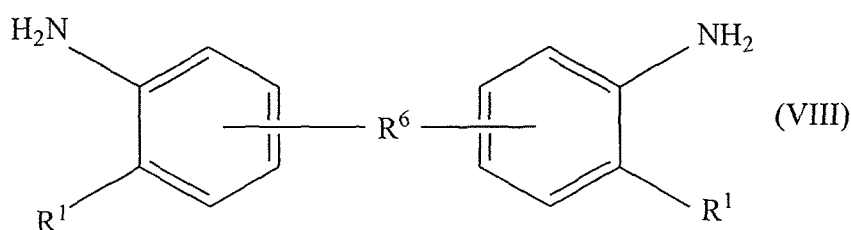
[0139] The compound containing at least two crosslinkable reaction groups represented by the formula (II) is preferably one having 2 or 3 crosslinkable reaction groups, more preferably one having 2 crosslinkable reaction groups. When the number of crosslinkable reaction groups represented by the formula (II) is less than 2, crosslinking cannot be achieved.

[0140] R^2 contained in the substituent R^1 of the crosslinkable reaction group represented by the formula (II) is a monovalent organic group other than hydrogen atom. Since an N- R^2 bond is higher in oxidation resistance than a N-H bond, it is preferable to use -NHR^2 as the substituent R^1 as noted above.

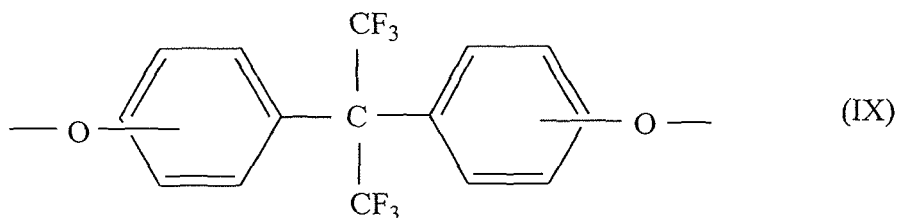
[0141] The monovalent organic group is not limited particularly, and examples thereof are an aliphatic hydrocarbon group, a phenyl group and a benzyl group. Specifically, for example, at least one of R^2 is a lower alkyl group having 1 to 10, particularly 1 to 6 carbon atoms such as -CH_3 , $\text{-C}_2\text{H}_5$ or $\text{-C}_3\text{H}_7$; a fluorine atom-containing lower alkyl group having 1 to 10, particularly 1 to 6 carbon groups such as -CF_3 , $\text{-C}_2\text{F}_5$, $\text{-CH}_2\text{F}$, $\text{-CH}_2\text{CF}_3$ or $\text{-CH}_2\text{C}_2\text{F}_5$; a phenyl group; a benzyl group; a phenyl group or a benzyl group, in which 1 to 5 hydrogen atoms are substituted by fluorine atoms such as $\text{-C}_6\text{F}_5$ or $\text{-CH}_2\text{C}_6\text{F}_5$; or a phenyl group or a benzyl group, in which 1 to 5 hydrogen atoms are substituted by -CF_3 such as $\text{-C}_6\text{H}_5\text{-}_n(\text{CF}_3)_n$ or $\text{-CH}_2\text{C}_6\text{H}_5\text{-}_n(\text{CF}_3)_n$ wherein n is an integer of 1 to 5.

[0142] Among these, a phenyl group and -CH₃ are preferable from the viewpoint of especially excellent heat resistance, satisfactory crosslinkability and relatively easy synthesis.

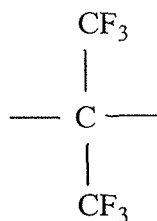
[0143] From the viewpoint of easy synthesis, preferable as the crosslinking agent (B) are compounds which have two crosslinkable reaction groups represented by the formula (II) and are represented by the formula (VIII) as noted above:



10 wherein R¹ is as defined above, R⁶ is -SO₂-, -O-, -CO-, an alkylene group having 1 to 6 carbon atoms, a perfluoroalkylene group having 1 to 10 carbon atoms, a single bond or a group represented by:

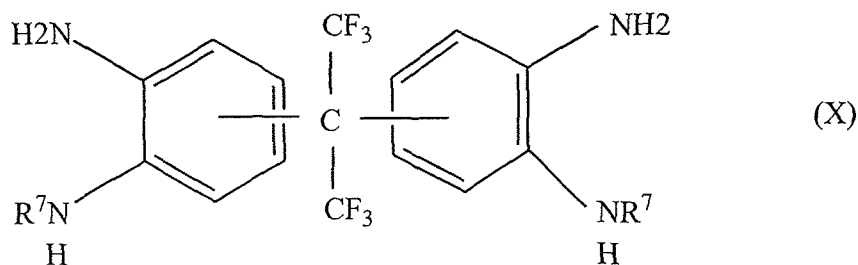


15 [0144] Preferable examples of alkylene groups having 1 to 6 carbon atoms, as noted above, are methylene, ethylene, propylene, butylene, pentylene, hexylene and the like. Examples of perfluoroalkylene groups having 1 to 10 carbon atoms are



20 and the like. These compounds are known as examples of bisdiaminophenyl compound in Japanese Patent No. 2-59177B, and Japanese Kokai No. 8-120146A, etc.

[0145] Among these compounds, more preferable compounds as the crosslinking agent (B) are compounds represented by the formula (X):



5 wherein R⁷ groups are the same or different and each is hydrogen atom, an alkyl group having 1 to 10 carbon atoms; an alkyl group having 1 to 10 carbon atoms and containing fluorine atom; a phenyl group; a benzyl group; or a phenyl group or benzyl group, in which 1 to 5 hydrogen atoms are replaced by fluorine atoms and/or -CF₃.

[0146] Non-limiting examples thereof are as with other embodiments herein, for instance,
 10 2,2-bis(3,4-diaminophenyl)hexafluoropropane, 2,2-bis[3-amino-4-(N-methylamino)phenyl]hexafluoropropane, 2,2-bis[3-amino-4-(N-ethylamino)phenyl]hexafluoropropane, 2,2-bis[3-amino-4-(N-propylamino)phenyl]hexafluoropropane, 2,2-bis[3-amino-4-(N-phenylamino)phenyl]hexafluoropropane, 2,2-bis[3-amino-4-(N-perfluorophenylamino)phenyl]hexafluoropropane, 2,2-bis[3-amino-4-(N-benzylamino)phenyl]hexafluoropropane, and the like. Of these, from the viewpoint of excellent heat resistance, 2,2-bis[3-amino-4-(N-methylamino)phenyl]hexafluoropropane, 2,2-bis[3-amino-4-(N-ethylamino)phenyl]hexafluoropropane, 2,2-bis[3-amino-4-(N-propylamino)phenyl]hexafluoropropane and 2,2-bis[3-amino-4-(N-phenylamino)phenyl]hexafluoropropane are preferable, and from the viewpoint of particularly excellent heat resistance, 2,2-bis[3-amino-4-(N-phenylamino)phenyl]hexafluoropropane is preferable.

[0147] The bisamidoxime crosslinking agent, bisamidrazone crosslinking agent, bisaminophenol crosslinking agent, bisaminothiophenol crosslinking agent or bisdiaminophenyl
 25 crosslinking agent react with a nitrile group, a carboxyl group or an alkoxycarbonyl group of the fluorine-containing elastomer and form an oxazole ring, a thiazole ring, an imidazole ring or a triazine ring to provide a crosslinked article.

[0148] An amount of the crosslinking agent (B) is preferably not less than 0.3 part by weight, more preferably not less than 0.5 part by weight, further preferably not less than 0.7 part by weight based on 100 parts by weight of the elastomer, from the viewpoint of enhancing crosslinkability of the composition. In addition, the amount of the crosslinking agent (B) is preferably not more than 10.0 parts by weight, more preferably not more than 2.0 parts by weight based on 100 parts by weight of the elastomer.

[0149] In the present invention, in addition to the above-mentioned crosslinking agents, other crosslinking agents can be used together.

[0150] When the fluorine-containing elastomer contains a nitrile group, the fluorine-containing elastomer composition in this embodiment of the present invention may comprise an organotin compound such as tetraphenyltin, triphenyltin or the like because the nitrile group forms a triazine ring, thus making it possible to achieve triazine crosslinking.

[0151] In this embodiment of the present invention, an amount of such an organotin compound is preferably 0.05 to 10 parts by weight, more preferably 1 to 5 parts by weight based on 100 parts by weight of the fluorine-containing elastomer. When the amount of organotin compound is less than 0.05 part by weight, there is a tendency that the fluorine-containing elastomer is not sufficiently crosslinked, and when the amount of organotin compound is more than 10 parts by weight, physical properties of a crosslinked article tends to be deteriorated.

[0152] In the fluorine-containing elastomer composition of this embodiment of the present invention, usual additives may be added, as the case demands, to crosslinkable elastomer compositions, for example, a filler, a processing aid, a plasticizer and a colorant may be blended thereto. In addition, one or more usual crosslinking agents or crosslinking accelerators different from the above-mentioned ones may be blended to the composition. Also, a different kind of elastomer may be mixed to an extent not to impair the effects of the present invention.

[0153] Examples of a filler for this embodiment are organic fillers, and from the viewpoint of heat resistance and plasma resistance (reduced number of particles and low weight reduction rate at emission of plasma), there are preferably exemplified organic pigments; imide fillers having an imide structure such as polyimide, polyamide imide and polyetherimide; ketone engineering plastics such as polyether ether ketone (PEEK) and polyether ketone (PEK), and organic pigments are particularly preferable.

[0154] Examples of organic pigments for use in this embodiment are condensed azo pigments, isoindolenone pigments, quinacridone pigments, diketopyrrolopyrrole pigments, anthraquinone pigments, and the like. Among these pigments, from the viewpoint of excellent heat resistance and chemical resistance and less effect on characteristics of a molded article, quinacridone pigments, diketopyrrolopyrrole pigments and anthraquinone pigments are preferable, and quinacridone pigments are more preferable.

[0155] Further the fluorine-containing crosslinkable composition of this embodiment of the present invention may contain a general filler.

[0156] Examples of such general fillers are organic fillers made of engineering plastics such as polyarylate, polysulfone, polyether sulfone, polyphenylene sulfide, polyoxybenzoate and polytetrafluoroethylene powder; metallic oxide fillers such as aluminum oxide, silicon oxide, yttrium oxide and titanium oxide; metallic carbides such as silicon carbide and aluminum carbide, metallic nitride fillers such as silicon nitride and aluminum nitride; and inorganic fillers such as aluminum fluoride, carbon fluoride, barium sulfate, carbon black, silica, clay and talc.

[0157] Among these fillers, from the viewpoint of an effect of shielding various plasmas, aluminum oxide, yttrium oxide, silicon oxide, polyimide and carbon fluoride are preferable.

[0158] Also, the above-mentioned inorganic fillers and organic fillers may be used alone or may be blended in combination of two or more thereof.

[0159] The fluorine-containing elastomer composition of this embodiment of the present invention can be prepared by mixing each of the above-mentioned components by using usual processing equipment for rubber, for example, an open roll, Banbury mixer, kneader, or the like. In addition, the composition can be prepared also by a method of using a closed mixer and a method of co-coagulation through emulsion mixing.

[0160] A hardness in Shore A of a crosslinked article obtained by crosslinking the fluorine-containing elastomer of this embodiment of the present invention is preferably not less than 50, more preferably not less than 55, further preferably not less than 60 in that sealing property of the sealing material made of the elastomer composition of the present invention is satisfactory. In addition, the hardness of a crosslinked article is preferably not more than 95, more preferably not more than 90, further preferably not more than 85 in that sealing property of the sealing material made of the elastomer composition of the present invention is satisfactory.

[0161] A crosslinked article obtained by crosslinking and molding the fluorine-containing elastomer composition of this embodiment of the present invention is excellent in chemical resistance, mechanical strength and heat resistance. Also, according to this embodiment of the present invention, since adjustment of the hardness can be carried out by combining two kinds of perfluoroelastomers, the hardness can be adjusted to a desired one even without adding a filler. In this case, the cured article is suitable as a sealing material for sealing of, for example, semiconductor equipment from the viewpoint of improvement in reduction of pollution of working environment because outgas component generated from the cured article is reduced. Examples of the sealing material are O-ring, square ring, gasket, packing, oil seal, bearing seal, lip seal, etc.

[0162] In the present invention, the semiconductor manufacturing equipment is not limited particularly to equipment for producing semiconductors and encompasses whole manufacturing equipment used in the field of semiconductors where a high degree of cleanliness is required, such as equipment for manufacturing a liquid crystal panel and a plasma panel. Examples of the semiconductor manufacturing equipment are as follows.

(1) Etching system

Dry etching equipment
Plasma etching machine
Reactive ion etching machine
Reactive ion beam etching machine
Sputter etching machine
Ion beam etching machine
Wet etching equipment
Ashing equipment

(2) Cleaning system

Dry etching cleaning equipment
UV/O₃ cleaning machine
Ion beam cleaning machine
Laser beam cleaning machine
Plasma cleaning machine
Gas etching cleaning machine
Extractive cleaning equipment

Soxhlet extractive cleaning machine
 High temperature high pressure extractive
 cleaning machine

Microwave extractive cleaning machine

5 Supercritical extractive cleaning machine

(3) Exposing system

Stepper

Coater and developer

(4) Polishing system

10 CMP equipment

(5) Film forming system

CVD equipment

Sputtering equipment

(6) Diffusion and ion implantation system

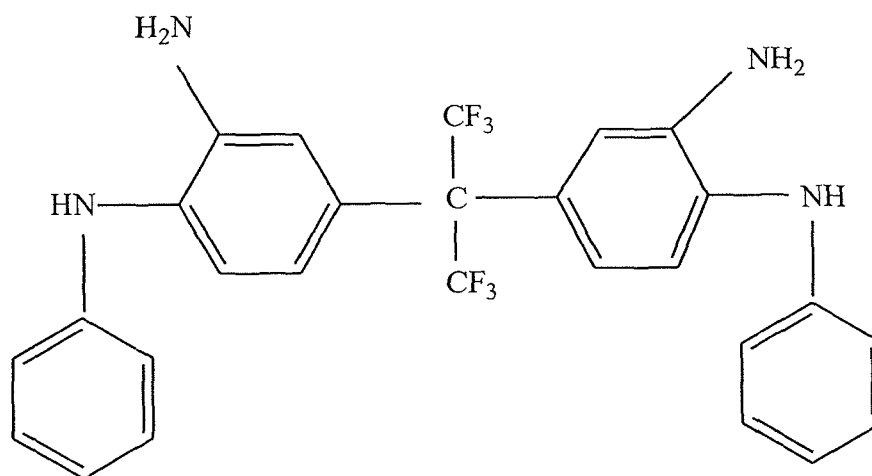
15 Oxidation and diffusion equipment

Ion implantation equipment

[0163] The invention and various embodiments will now be explained with respect to the following non-limiting Examples.

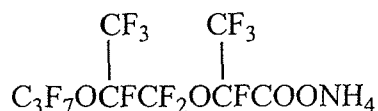
EXAMPLE 1

20 [0164] A compound used in this Example is shown below. The compound (NPh-AF) represented below is used as a curing or crosslinking agent.



[0165] Preparation Example 1: Synthesis of Perfluoroelastomer (1):

[0166] Into a 6-liter stainless steel autoclave having no ignition source were poured 2.34 liters of pure water, 23.4 g of



as an emulsifying agent and 0.21 g of $(\text{NH}_4)_2\text{CO}_3$, and the inside of the system was sufficiently

5 replaced with nitrogen gas and subjected to deaeration. Then, the autoclave was heated up to 52°C with stirring at 600 rpm, and a gas mixture of tetrafluoroethylene (TFE) and perfluoro(methyl vinyl ether) (PMVE) (molar ratio of TFE/PMVE = 22/78) was introduced so that the inside pressure became 0.78 MPa·G. Then, after introducing 0.82 g of

10 $\text{CF}_2=\text{CFO}(\text{CF}_2)_5\text{CN}$ with pressurized nitrogen gas, a solution prepared by dissolving 12.3 g of ammonium persulfate (APS) in 30 g of water, was introduced with pressurized nitrogen gas to initiate a reaction.

[0167] As the polymerization proceeded, the inside pressure of the reactor decreased, and pressurized TFE and PMVE were introduced so that the inside pressure became 0.78 MPa·G. Until completion of the polymerization, 323 g of TFE and 356 g of PMVE were introduced in a
15 specific ratio. During the reaction, pressurized $\text{CF}_2=\text{CFO}(\text{CF}_2)_5\text{CN}$ was introduced 17 times, totaling 14.67 g to obtain 2,989 g of an aqueous dispersion having a solids content of 21.2% by weight.

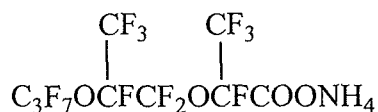
[0168] Out of the obtained aqueous dispersion, 500 g was distilled with 500 g water, and the distilled solution was slowly added to 2,800 g of 3.5% by weight aqueous solution of
20 hydrochloric acid with stirring. After completion of the addition, the solution was stirred for five minutes, and then a coagulated product was filtered off. The obtained polymer was poured into 2 kg of HCFC-141b, followed by 5-minute stirring and filtering off again. Thereafter washing with HCFC-141b and filtering off were repeated four more times, followed by vacuum drying at 60°C for 72 hours to obtain 110 g of a polymer (Perfluoroelastomer (1)).

25 [0169] According to F-NMR analysis, contents of each monomer of the obtained Perfluoroelastomer (1) are as shown in Table 1.

EXAMPLE 2

[0170] Preparation Example 2: Synthesis of Perfluoroelastomer (2)

[0171] Into a 6-liter stainless steel autoclave having no ignition source were poured 2.34 liters of pure water, 23.4 g of



5 as an emulsifying agent and 0.21 g of $(\text{NH}_4)_2\text{CO}_3$, and the inside of the system was sufficiently replaced with nitrogen gas and subjected to deaeration. Then the autoclave was heated up to 52°C with stirring at 600 rpm, and a gas mixture of TFE and PMVE (molar ratio of TFE/PMVE = 41/59) was introduced so that the inside pressure became 0.78 MPa·G. Then, after introducing 0.87 g of $\text{CF}_2=\text{CFO}(\text{CF}_2)_5\text{CN}$ with pressurized nitrogen gas, a solution prepared
10 by dissolving 12.3 g of APS in 30 g of water, was introduced with pressurized nitrogen gas to initiate a reaction.

[0172] As the polymerization proceeded, the inside pressure of the reactor decreased, and pressurized TFE and PMVE were introduced so that the inside pressure became 0.78 MPa·G. Until completion of the polymerization, 400 g of TFE and 284 g of PMVE were introduced in a
15 specific ratio. During the reaction, pressurized $\text{CF}_2=\text{CFO}(\text{CF}_2)_5\text{CN}$ was introduced 17 times totaling 14.72 g to obtain 3,087 g of an aqueous dispersion having a solids content of 22.5% by weight.

[0173] Out of the obtained aqueous dispersion, 500 g was distilled with 500 g water, and the distilled solution was slowly added to 2,800 g of 3.5% by weight aqueous solution of
20 hydrochloric acid with stirring. After completion of the addition, the solution was stirred for five minutes, and then a coagulated product was filtered off. The obtained polymer was poured into 2 kg of HCFC-141b, followed by 5 minutes of stirring and filtering off again. Thereafter washing with HCFC-141b and filtering off were repeated four more times, followed by vacuum drying at 60°C for 72 hours to obtain 110 g of a polymer (Perfluoroelastomer 2).

25 [0174] According to F-NMR analysis, contents of each monomer of the obtained Perfluoroelastomer (2) is shown in Table 1.

TABLE 1

Content (% by mole)	Perfluoroelastomer (1)	Perfluoroelastomer (2)
---------------------	------------------------	------------------------

PMVE	41.7	30.2
TFE	57.9	69.4
CF ₂ =CFO(CF ₂) ₅ CN	0.43	0.43

EXAMPLE 3

[0175] Perfluoroelastomer (1) from Example 1, Perfluoroelastomer (2) from Example 2 and NPh-AF (shown above) as a cross-linking agent were mixed in amounts as shown in Table 2, and kneaded with an open roll to prepare a crosslinkable fluorine-containing elastomer composition.

[0176] This fluorine-containing elastomer composition was subjected to crosslinking by pressing at 180°C for 20 minutes and further crosslinking in an oven at 290°C for 18 hours to make a test sample O-ring (P-24). With respect to this test sample, crosslinkability at crosslinking and physical properties in the normal state were measured by the following methods. The results are shown in Table 2.

[0177] Crosslinkability: With respect to each crosslinkable composition, a vulcanization curve was obtained at 180°C by using JSR type Curastometer Model II, and a minimum torque (M_L), a maximum torque (M_H), and induction time (T₁₀) and an optimum vulcanization time (T₉₀) were determined.

[0178] Physical Properties in the Normal State: According to JIS K 6301, a 100% modulus (M₁₀₀), a tensile strength (T_B), an elongation (E_B) and hardness (H_S) of a 2 mm thick crosslinked article in a normal state (25°C) were measured.

COMPARATIVE EXAMPLE 1

[0179] A composition was prepared in the same manner as in Example 3 with the exception that only Perfluoroelastomer (1) was used as a perfluoroelastomer instead of a combination use of Perfluoroelastomers (1) and (2). Then crosslinkability at crosslinking and physical properties in a normal state were measured in the same manner as Example 3. The results are shown also in Table 2.

COMPARATIVE EXAMPLE 2

[0180] A composition was prepared in the same manner as in Example 3 with the exception that only Perfluoroelastomer (2) was used as a perfluoroelastomer instead of a combination use
5 of Perfluoroelastomers (1) and (2). Then crosslinkability at crosslinking and physical properties in a normal state were measured in the same manner as Example 3. The results are shown also in Table 2.

TABLE 2

Amount (parts per 100)	Example 3	Comparative Example 1	Comparative Example 2
Perfluoroelastomer (1)	50	100	--
Perfluoroelastomer (2)	50	--	100
NPh-AF	1.0	1.0	1.0
Results of evaluation			
Crosslinkability			
M _L (kgf)	0.72	0.42	1.12
M _H (kgf)	3.03	2.50	3.33
T ₁₀ (min)	5.6	4.3	7.3
T ₉₀ (min)	16.3	9.5	44.5
Physical Properties in Normal State			
M ₁₀₀ (MPa)	2.1	1.3	3.2
T _B (MPa)	16.0	6.7	20.8
E _B (%)	250	275	225
H _S (Shore A)	68	59	78

EXAMPLES 4-14

[0181] The elastomeric compositions in the Examples 4-14 were produced by mixing or blending the perfluoropolymers with any desired additives. The polymer(s) and any additives may be blended using an internal mixer such as those commercially available from C. W. Brabender Instruments, Inc. of S. Hackensack, N.J. or other internal mixers such as are commercially available from Morijama of Farmingdale, N.Y. Examples 4-12 were prepared by blending in a two-roll mixer (also called "open mill mixer"), with 6-inch diameter, using a regular rubber compounding mixing method known in the art. The roll temperature was about 50° C with a roll speed of about 25 to 35 rpm for a batch weight of 500 grams.

[0182] The method of preparation included adding the first polymer onto the mill at 50° C and mixing for approximately 8-9 minutes. After the first curable perfluoropolymer was sheeted out on the mill, the second curable perfluoropolymer was added. After mixing both polymers for another 3-4 minutes, the curative was added and mixed thoroughly. After adding the curative, the blend was cut and mixed on the rolls 3 times, and then cut/mixed 30 times.

The blend was taken off the mill and cooled down to about room temperature. Once cooled, the blend was placed back into the mill and cut/mixed multiple times.

[0183] A standard regular rubber compounding procedure was used except for the mixing temperature 50° C and the cutting and remixing steps (3+30+30 cut /remixing).

5 [0184] For Examples 13 and 14, the internal mixer first used was at 50° C to mix a master batch concentrate by adding the first perfluoropolymer and the curative based on the formulation of the master batch concentrate. The master batch included Polymer A in 100 parts by weight and NPh-AF (in an amount of 6.8 parts per hundred parts by weight). The master batch was then diluted to prepare the formulations noted for Examples 13 and 14. The master
10 batch concentrate was unloaded from the internal mixer and moved to an open mill. The same procedures as described above were followed for Examples 4-12 except for adding the master batch concentrate having curatives in the last step (instead of the curative only).

[0185] It will be recognized that these procedures were for these examples only and that any other standard rubber compounding mixer or regular internal mixer could be used within
15 the invention.

[0186] Examples of elastomer compositions and molded articles prepared according to the methods of the present invention are set forth below in Table 3. Included in Table 3 are the physical characteristics of elastomers formed by the present invention. Examples 4-14 were tested for properties using ASTM procedures identified in Table 4, and the physical property
20 parameters were recorded: T_b (tensile at break in psi (MPa)); E_B (elongation at break in %); M_{100} (100% modulus in psi (MPa)), Hardness (Durometer M) and Compression set of the O-rings prepared from the elastomers.

[0187] The example compounds were prepared from a commercially available perfluoropolymer composition and mixed as discussed herein. In the examples, Polymer A
25 from Daikin Industries is the first perfluoropolymer as defined herein and is a perfluoropolymer of TFE and perfluoromethylvinyl ether (PMVE) in a molar ratio of 60/40. Polymer B from Daikin Industries is the second polymer used and is a perfluoropolymer of TFE and PMVE in a molar ratio of 70/30. In both of Polymers A and B, there is 0.6 mole percent of a cure site monomer including a cyano-functional group for curing. The curative, NPh-AF is 4,4'-[2,2,2-
30 Trifluoro-1-(trifluoromethyl) ethylidene]bis[N1-phenyl-1,2-benzenediamine]. Each of the components, Polymer A and B, and NPh-AF are commercially available. The weight

percentage ratios of the resultant example elastomers (Examples 4-14) vary as described in Table 3. The compounds were cured at 182°C for 30 minutes followed by postcuring at 290°C for 18 hours.

[0188] The molded articles, most commonly an O-ring, seal or gasket, formed from an elastomer of the present invention, are described herein. Seals can be formed from the perfluoropolymers by a variety of processing methods, such as compression molding, injection molding, extrusion, etc. Molding was used in the Examples herein.

[0189] As illustrated in Table 3, after testing in remote NF₃ plasma at high temperature, the samples showed insignificant weight loss. The sample integrity and surface appearance showed no significant change. Compared to the commercial samples as controls, the working examples display better resistance to remote NF₃ plasma. In direct plasma environments (O₂, O₂+CF₄), the results are comparable to the commercial products.

[0190] O-rings were formed from the resultant elastomers of Examples 4-14 for evaluation of compression set. Examples 4-14 each showed less than 25 percent compression set after 300°C for 70 hours under 25% deflection and showed less than 40 percent compression set after 300°C for 168 hours under 25% deflection. This is a significant improvement in the art.

[0191] The O-rings formed from the elastomers of Examples 4-14 were subjected to a plasma gas environment and evaluated. The percent loss is defined for each in Table 3 based on the internal screening method listed in Table 4.

Table 3

Formulation	Ex. 4	Ex. 5	Ex. 6	Ex. 7	Ex. 8	Ex. 9	Ex. 10	Ex. 11	Ex. 12	Ex. 13	Ex. 14
Polymer B (weight ratio)	50	50	50	65	65	65	35	35	35	50	50
Polymer A (weight ratio)	50	50	50	35	35	35	65	65	65	50	50
NPh-AF	0.7	0.8	0.9	0.7	0.8	0.9	0.7	0.8	0.9	0.9	0.9
T _B : psi (MPa)	944 (6.51)	1049 (7.23)	978 (6.74)	1388 (9.57)	1130 (7.79)	1370 (9.45)	847 (5.84)	933 (6.43)	894 (6.16)	1103 (7.61)	1052 (7.26)
E _B (%)	271	262	261	263	258	260	270	262	249	256	261
M ₁₀₀ : psi (MPa)	193 (1.33)	216 (1.49)	200 (1.38)	242 (1.67)	236 (1.63)	250 (1.72)	184 (1.27)	198 (1.37)	217 (1.50)	231 (1.59)	217 (1.50)
Density	2.055	2.0575	2.056	2.061	2.061	2.063	2.050	2.056	2.055	2.054	2.057
Shore M Hardness	71	71.5	71.5	72.5	72.5	73	70	70	70	71.5	
Shore A Hardness	63.5	64	64.5	65	65.5	66.5	61.5	62	62.5	65.5	63.5
Compression set (%) (300°C/70hrs)	21.2	19.7	17.3	20.3	20.3	17.4	17.7	18.5	17.1	17.1	17.7
Compression set (%) (300°C/168hrs)	33.5	30.9	29.4	27.6	33.5	30.0	35.0	25.2	31.4	31.6	29.7
Stiction: lbs (N)	41.0 (182.4)	40.6 (180.6)	41.5 (184.6)	45.9 (204.2)	46.0 (204.6)	46.9 (208.6)	38.9 (173.0)	40.2 (178.8)	40.5 (180.1)	40.1 (178.4)	40.4 (179.7)
NF ₃ remote, 6hrs/220°C (% loss)	0.023	0.022	-0.024	0.108	0.116	0.069	0.070	-0.022	0.046	0.000	-0.023
O ₂ direct ICP, 30 min. (% loss)	2.82	2.94	2.87	2.72	2.84	2.95	2.97	3.07	2.84	2.8	2.77
O ₂ +CF ₄ direct ICP (% loss)	2.98	2.89	2.93	2.92	2.98	2.91	2.95	3.29	2.91	2.95	3.02
NF ₃ remote, 12hrs/250°C loss	0.084	0.006	-0.041	-0.001	0.065	0.034	0.039	0.073	0.023	0.035	0.023
O ₂ Direct, 80°C/60min ICP (% loss)	4.243	4.395	4.690	4.718	5.081	5.134	5.482	5.546	5.522	5.493	5.255
O ₂ +CF ₄ Direct/80°C/ 60 min, ICP (% loss)	4.469	4.687	4.875	4.907	5.275	5.526	5.818	6.083	5.785	5.569	5.550
RIE O ₂ plasma, 90 min. (% loss)	3.453	3.288	3.006	3.076	2.609	3.234	3.235	3.331	3.413	3.377	3.586

Table 4

Properties	Testing Method
M _H : lb inch (N·m or kg·f as noted)	ASTM D 5289, average of 2 min sets; 360°F (182.2°C)/60 min.
M _L : lb inch (N·m or kg·f as noted)	
T ₁₀ (min)	
T ₅₀ (min)	
T ₉₀ (min)	
T _{s2} (min)	
T _b : psi (MPa)	Average of 10 O-rings, 20"/min, ASTM D 1414, ASTM D 412
E _b (%)	
M ₁₀₀ : psi (MPa)	
Density	ASTM D 792
Shore M Hardness	ASTM D 2240
Shore A Hardness	ASTM D 2240
Compression set (%) (300°C/70 h)	ASTM D 1414/ASTM D 395 average of 10
Compression set (%) (300°C/168 h)	ASTM D1414 /ASTM D395 average of 10
Stiction: lbs. (N)	Compression 25% between two Aluminum substrates, conditioning 392°F (200°C) for 24 hrs and cool for 1 h and then push O-ring of Al at 0.5"/min (0.2 mm/second).
NF ₃ remote, 6h/220°C (% loss)	NF ₃ /Ar 1:1, 3 Torr, 220°C (set up 300°C)
NF ₃ remote, 12h/220°C (% loss)	NF ₃ /Ar 1:1, 3 Torr, 220°C (set up 300°C)
O ₂ direct ICP, 30 min. (% loss)	Power 400W, flow 16 standard cm ³ per minute, pressure: 10 Pa, time: 30 min.
O ₂ +CF ₄ direct ICP (% loss)	Power 400W, O ₂ /CF ₄ 16/16standard cm ³ per
NF ₃ remote, 12 h/250°C (% loss)	Power 2500W, pressure 990 mTorr, NF ₃ 400 ₃
O ₂ Direct, 80°C/60min ICP (% loss)	Power 590W, pressure 100 mTorr, O ₂ : 32
O ₂ +CF ₄ Direct/80°	Power 590W, pressure 100 mTorr, O ₂ /CF ₄
RIE O ₂ plasma AEC, 4400 seconds (% loss)	Power 300W, pressure 300 mTorr. O ₂ : 60
RIE O ₂ +CF ₄ plasma, AEC, 4400 seconds (% loss)	Power 300W, pressure 300 mTorr, O ₂ /CF ₄ : 30/30 standard cm ³ per minute

EXAMPLE 15

[0192] As noted elsewhere herein, the various fluorine-containing elastomers and perfluoroelastomer blends described herein can be bonded and molded onto metals and other substrates to form bonded products. Bonding samples were prepared and evaluated using the current ASTM-D-429 A method involving pulling two metal plates apart at a rate of 0.4 mm/second (1 inch/minute) at room temperature and recording bonding force. A blend of two perfluoropolymers according to the invention was molded in between two circular metal plate surfaces, each of which surface was treated with a bonding agent prior to the bonding process. The metal plates, which were 2 in² (12.9 cm²) in surface area, were sandblasted with 36-sized grit before use.

[0193] Perfluoroelastomer compound C was dissolved in Fluorinert® FC-77 to bond the blended composition of Example 9 in Table 3 onto aluminum and steel surfaces. Example 9's compositions included Polymer A in 35 parts per hundred, Polymer B in 65 parts per hundred and NPh-AF (in 0.9 parts per hundred). The bonding agent (Perfluoroelastomer compound C) included a perfluoroelastomer polymer having a nitrile-functionalized cure site monomer (FFKM, 100 parts), Aerosil® R972 (12 parts per hundred parts FFKM), 2,2-bis[3-amino-4-hydroxyphenyl]hexafluoropropane (1.5 parts per hundred curative), Cromophthal Blue A3R (0.5 parts per hundred as a colorant), Cromophthal Yellow 2RF (0.5 parts per hundred as a colorant), and Varox® DBPH-50 (1 part per hundred). In the polymer FFKM D, the molar ratio of TFE:PMVE: CSM of 53:44:3.

[0194] After dissolving Compound C in FC-77, the mixture was applied as a thin layer to a metal substrate. The solution was allowed to dry for 30 minutes and a pressed portion of the composition of Example 9 was molded onto the substrates at 360°F (182.2°C) for 30 minutes, and then postcured at 550°F (287.8°C) for 22 hours.

[0195] Table 5 shows the results from the bonding force tests.

TABLE 5

Bonding Agent	Substrate	Before Postcure	After Postcure	Bonding Force lbs (N)
Compound C in FC-77, 1:15	Al	2-bonded*	2-bonded	1337 (5947), 895 (3981)
Compound C in FC-77, 1:6	Al	2-bonded	2-bonded	1390 (6183), 1449 (6445)
Compound C in FC-77, 1:15	Steel	2-bonded	2-bonded	1436 (6387), 1603 (7130)
Compound C in FC-77, 1:6	Steel	bonded*	bonded	1328 (5907), --

*“2-bonded” means that both samples made were bonded and “bonded” means that only one sample was made and tested.

EXAMPLES 16-41

- 5 [0196] Table 6 shows various compositions having at least one perfluoroelastomer polymer. Perfluoropolymers (1) and (2) (FFKM (1) and (2)) used in earlier examples herein are adapted for sole use and combined use in various of the examples 16-33 as are other perfluoropolymer compounds, including FFKM (3) (PFE131TX of Dyneon having 1.2 mol% of a cyano functional cure site monomer), FFKM (4) (PFE133TBX of Dyneon also having 1.2
- 10 mol% of a cyano functional cure site monomer and having 20 parts of a PFA copolymer), FFKM (5) (having 53.2% tetrafluoroethylene monomer, 44.1% perfluoromethyl vinyl ether monomer and 2.7 mol% of cure site monomers with cyano functional curing groups) and FFKM (6) (PFR95HT available from Solvay and having a peroxide curable functional group). Four different curatives were used as well, bisaminophenol (BOAP), Nph-AF, and two
- 15 fluorochemical acid onium compounds (1) and (2). In these Examples, there was examination of different polymers and curatives as well as various types of fillers and varying filler sizes. The effects are shown in Table 7. Fillers used include barium sulfate (Barifine® BF 20, from Sakai Chemical, Japan) having a mean particle size of about 0.03 microns, titanium dioxide (Ti-Pure® from DuPont); barium sulfate of average particle size 1.7 microns (Blank Fixe N® from
- 20 Solvay), barium titanate powder having particle size 50 nm (TPL, Inc., New Mexico), barium titanate of particle size less than 100 nm (NanOxide™ from TPL, Inc.), barium titanate nanopowder of particle size of about 100 nm (Inframat® from Advanced Materials, Connecticut), alumina powder (Alumina AKP G-0625), alumina titania powder (Inframat Nanox™ S2613P), nickel/nickel oxide powder (Quantum Sphere, QSI-Nano®), nickel oxide
- 25 powder having 20 nm particle size (Inframat), and silica carbide (45-55nm).

[0197] Based on the same polymer system, there was also a comparison of different fillers with respect to the ability of the compositions to provide resistance to ClF₃. ClF₃ was introduced at 280°C in a 50% mixture of ClF₃ and argon, each at 200 standard cubic centimeters per minute with a 30 minute exposure. Tests for O₂ and NF₃ plasma were direct plasma tests (RIE 90 minutes). The results show that certain of the fillers tried, including Ni/NiO, alumina and SiC fillers were not as effective as other fillers. The best results were achieved with BaSO₄ and BaTiO₃. It appeared that the BaSO₄ had less ClF₃ resistance but yielded better compression set based on the data. However, the BaSO₄ containing formulation also gave a long T₉₀ time (indicating slower curing). The BaTiO₃ showed the best ClF₃ resistance overall. Despite such benefit, the compression set data for barium titanate formulations using 50-100 nm particles was higher than preferred (100-120%). Such testing based on compression set would have otherwise screened out this filler as a potential candidate for overall use, however, surprisingly as seen in the next group of examples beginning with Example 34 in Table 8, use of the same filler but with a larger nanoparticle size (when larger sizes are typically thought to contribute further to impurities in processing) provided improved compression set, resulting in preferred embodiments in which both strong ClF₃ plasma resistance as well as reasonable compression set were achieved.

[0198] Following these results, further testing was done demonstrating the effects of barium titanate in different sizes (including 700 nm size - available from Advanced Materials, Connecticut) and using the preferred blend of 50/50 Perfluoroelastomer (1) and Perfluoroelastomer (2) as these appeared to yield the best base formulations in this context. Compositions tested are set forth in Table 8 with data and results in Table 9, showing that tests were done in both 50/50 ClF₃/Ar mixtures using conditions as noted in Table 8, as well as using 100% ClF₃ plasma at 350°C, introduced at 30 standard cubic centimeters per minute for 60 minutes. Both remote NF₃ plasma in argon (fed at 50 standard cubic centimeters per minute, in 800 torr, for 6 hours) and direct NF₃ plasma (RIE, 90 minutes) were used for testing

along with direct O₂ plasma. As can be seen from Table 9, the 700nm BaTiO₃ filler gives unexpected and surprising results in a formulation having very good compression set (30%) while also keeping the best ClF₃ resistance. The other results are also surprisingly good. Other properties such as stiction, curing speed, and remote NF₃ resistance are good too.

TABLE 6

Example No.:	16	17	18	19	20	21	22	23	24	25	26	27	28	29	30	31	32	33
FFKM (1)	100	35				100	35				35	35	35	35	35	35	35	35
FFKM (2)		65					65				65	65	65	65	65	65	65	65
FFKM (3)			100					100										
FFKM (4)										117.6								
FFKM (5)				100														
FFKM (6)					100				100									
Barifine BF-20	14	14	14	14	14					14	14							
Ti-Pure R102	1	1	1	1	1	1	1	1	1	1	1							
Blank Fixe N						14	14	14	14									
Nanoxide HFB1000												15						
Barium Titanate (100 nm)													15		10			
Alumina AKP G-0625														5	5			
Alumina Titania														10				
Ni/NiO																10		
Nickel Oxide (20 nm)																5	15	
SiC (45-55 nm)																		15
Nph-AF	0.9	0.9				0.9	0.9					0.9	0.9	0.9	0.9	0.9	0.9	0.9
fluorochemical acid onium (1)			3					3		3								
fluorochemical acid onium (2)			1.25					1.25		1.25								
BOAP				1							0.64							
Luperox 101					0.9				0.9									
Total	115.9	115.9	119.25	116	115.9	115.9	115.9	119.25	115.9	136.85	115.64	115.9	115.9	115.9	115.9	115.9	115.9	115.9

TABLE 7

Example No.:	16	17	18	19	20	21	22	23	24	25	26	27	28	29	30	31	32	33
Curing Properties																		
MDR T (°F)	360	360	370	350	320	360	360	370	320	370	360	360	360	360	360	340	360	360
MDR Time (min)	60	60	60	60	60	60	60	60	60	60	60	60	60	60	60	60	60	60
M _H (lbs. in.)	5.73	6.58	7.26	5.13	9.12	11.06		9.47	14.52	9.29	7.23	6.54	6.67	5.4	10.75	14.24	12.56	5.65
M _L (lbs. in.)	2.95	4.07	0.4	0.05	1.13	2.59		0.56	1.61	0.71	4.17	4.06	3.98	4.27	4.17	2.96	6.78	3.27
T ₉₀ (mins.)	52.13	51.3	6.79	51.51	1.8	38.48		7.46	9.16	6.1	43.33	21.02	20.36		17.48	1.94	3.81	48.51
Press Temp. (°F)	340	350	370	350	320	360	360	370	320	350	350	350	350		360	340	340	350
Time (min.)	30	30	20	30	10	30	30	20	10	15	30	30	30		30	30	30	30
Post Cure Temp. (°F)	550	550	550	550	450	550	550	550	450	484	550	550	550		550	550	550	550
Time (hr)*	18	18	24	45	9	22	22	24	9	24	18	18	18		22	22	22	18
Physical Properties																		
Tensile Strength (psi)	blistered after Postcure	1695	1613	832	2187	1177	1425	1141	1660	1672	924	1467	2165		1570	1192	1094	1281
Elongation (%)		257	235	169	264	310	264	255	259	206	271	324	301		311	230	281	229
Hardness (M)		81	80	72	80	73	78	77	78	85	76	78	76		82	78	75	78
Comp. Set. % (300C/70hr)		100	39.7 (1/2 split)	all split	66.67	31.68	30.88	23.53	62.25	all split	79.6	120.5	85.4		100	122.6	131.5	125.5
Stiction (lbs)		94.8	40.1	73.2	90	44.2	58.6	47.2	114.9	101	89.3	76	77.9		110	82.5	80.3	87.2
Chemical/Plasma Properties																		
ClF ₃ (%wt. loss)		0.39	0.51	1.00	0.75	0.44	0.36	0.70	0.77	0.41	0.36	0.14	0.20		4.40	4.07	1.12	1.07
NF ₃ (%wt. loss)		2.13	1.778	2.879	3.98	4.169	4.088	4.693	6.673	1.571	2.309	1.832	1.578		4.18	15.10 ₉	9.312	15.60 ₁
O ₂ (%wt. loss)		0.821	0.522	1.172	0.89	1.745	1.598	1.44	1.823	0.697	1.165	1.221	1.204		0.86	0.917	1.114	1.16

*not molded well

*post cures in some samples done in cycles (Exs. 18, 23, 30 - 2/20/2), (Ex. 19 - step), (Exs. 21,22, 31,32 - 2/18/2), (Ex. 25 - 6/16/2)

TABLE 8

Example	34	35	36	37	38	39	40	41
Perfluoropolymer (1)	35	35	35	35	35	35	35	35
Perfluoropolymer (2)	65	65	65	65	65	65	65	65
BaTiO ₃ (600-700 nm)	15	15	15	10	10		15	15
BaTiO ₃ (100 nm)				5		5		
BaSO ₄ (1.7 micron)					5	10	5	5
TiO ₂	1		1	1	1	1	1	1
Nph-AF	0.9	0.9		0.9	0.9	0.9	0.9	
BOAP			0.64					0.64
Total	116.9	115.9	116.64	116.9	116.9	116.9	121.9	121.64

TABLE 9

Example	34	35	36	37	38	39	40	41
Curing Properties								
M _H (lbs. in.)	12.1	12.17	11.98	12.46	12.41	12.77	12.82	12.95
M _L (lbs. in.)	4.64	4.84	4.86	4.87	1.87	2.93	4.82	5.09
T ₉₀ (mins.)	39.88	25.91	14.83	28.19	39.61	33.21	42.59	16.22
Density (g/ml)	2.245	2.236	2.249	2.241	2.227	2.213	2.29	2.293
Physical Properties								
Tensile Strength (psi)	1930	1372	1493	1423	1584	1508	1506	1424
Elongation (%)	263	242	265	257	267	262	264	271
Hardness (M)	76	76	76	77	77	77	78	77
Comp. Set. % (300C/70hr/25% deflection)	32.6	30.6	36.5	36.5	26.5	40.4	40.2	52.0
Stiction (lbs)	44.2	46.7	50.4	54.5	65.1	59	65	67.3
Chemical/Plasma Properties								
ClF ₃ (%wt. loss) (50%Ar)	0.34	-0.18	0.93	0.38	0.41	0.33	0.28	0.56
ClF ₃ (%hardness change) (50%Ar)	0.4	-0.4	0.0	0.0	-0.5	-0.4	-0.4	-0.5
ClF ₃ (%wt. loss) (100%)	0.16	0.10	0.23	0.21	0.13	0.18	0.15	0.19
ClF ₃ (%hardness change)(100%)	-0.2	-0.5	-0.5	-0.3	-1.1	-0.4	-0.6	-1.5
NF ₃ (%wt. loss (remote)	0.07	0.05	0.11	0.11	0.04	0.05	0.04	0.03
NF ₃ (%wt. loss)(direct)	3.29	3.258	3.464	2.344	3.614	2.533	2.354	3.006
O ₂ (%wt. loss) (direct)	1.737	1.804	1.86		1.822	1.438	1.854	1.642

[0199] It will be appreciated by those skilled in the art that changes could be made to the embodiments described above without departing from the broad inventive concept thereof. It is understood, therefore, that this invention is not limited to the particular embodiments disclosed, but it is intended to cover modifications within the spirit and scope of the present invention as

5 defined by the appended claims.

CLAIMS

We claim:

1. A fluorine-containing elastomer composition comprising
a first curable perfluoropolymer comprising tetrafluoroethylene, at least one
5 perfluoroalkylvinyl ether and at least one cure site monomer having a functional group to
permit crosslinking of the perfluoropolymer; and
barium titanate.
2. The fluorine-containing elastomer composition according to claim 1 wherein the
barium titanate has an average particle size of at least 200 nm.
- 10 3. The fluorine-containing elastomer composition according to claim 2, wherein the
barium titanate has an average particle size of about 300 nm to about 1200 nm.
4. The fluorine-containing elastomer composition according to claim 2, wherein the
barium titanate has an average particle size of about 500 nm to about 1000 nm.
5. The fluorine-containing elastomer composition according to claim 2, wherein the
15 barium titanate is present in a mixture of at least two different average particle sizes.
6. The fluorine-containing elastomer composition according to claim 1, comprising
from about 1 to about 200 parts by weight of the barium titanate per 100 parts by weight of the
curable perfluoropolymer.
7. The fluorine-containing elastomer composition according to claim 6, comprising
20 from about 1 to about 100 parts by weight of the barium titanate per 100 parts by weight of the
curable perfluoropolymer.
8. The fluorine-containing elastomer composition according to claim 7, comprising
from about 1 to about 50 parts by weight of the barium titanate per 100 parts by weight of the
curable perfluoropolymer.
- 25 9. The fluorine-containing elastomer composition according to claim 8, comprising
from about 5 to about 50 parts by weight of the barium titanate per 100 parts by weight of the
curable perfluoropolymer
10. The fluorine-containing elastomer composition according to claim 9, comprising
about 5 to about 30 parts by weight of the barium titanate per 100 parts by weight of the curable
30 perfluoropolymer.

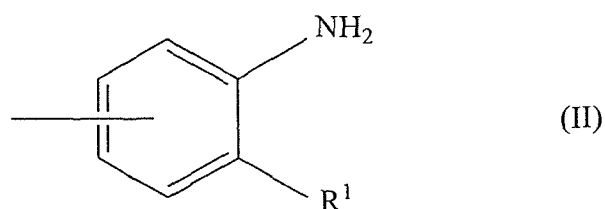
11. The fluorine-containing elastomer composition according to claim 10, comprising about 10 to about 20 parts by weight of the barium titanate per 100 parts by weight of the curable perfluoropolymer.

12. The fluorine-containing elastomer composition according to claim 1, wherein the functional group of the at least one cure site monomer is selected from the group consisting of nitrile, carboxyl and alkoxy carbonyl.

13. The fluorine-containing elastomer composition according to claim 1, further comprising a crosslinking agent which is capable of reacting with the functional group of the at least one cure site monomer.

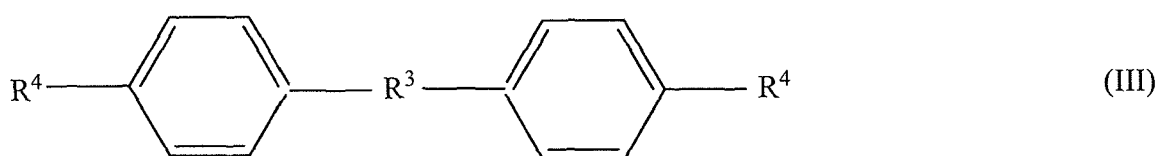
14. The fluorine-containing elastomer composition of claim 13, wherein the functional group of the at least one cure site monomer is selected from the group consisting of nitrile, carboxyl and alkoxy carbonyl and the crosslinking agent is a

(i) a compound containing at least two crosslinkable reaction groups represented by the formula (II):

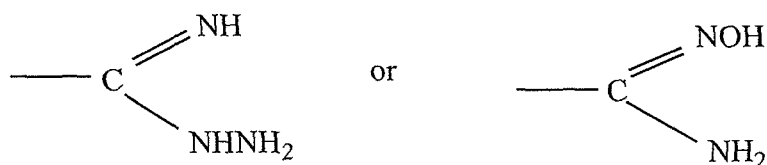


wherein R^1 groups are the same or different and each is $-NH_2$, $-NHR^2$, $-OH$ or $-SH$; R^2 is a monovalent organic group,

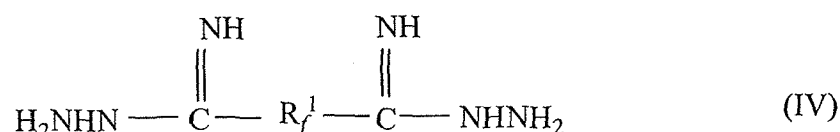
(ii) a compound represented by the formula (III):



wherein R^3 is $-SO_2-$, $-O-$, $-CO-$, an alkylene group having 1 to 6 carbon atoms, a perfluoroalkylene group having 1 to 10 carbon atoms or a single bond; R^4 is

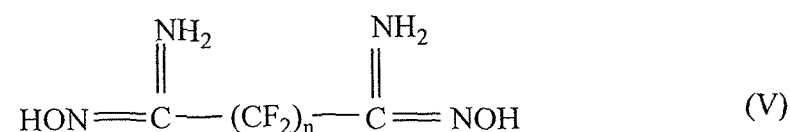


(iv) a compound represented by the formula (IV):



wherein R_f^1 is a perfluoroalkylene group having 1 to 10 carbon atoms, and

(v) a compound represented by the formula (V):



wherein n is an integer of 1 to 10.

15. The fluorine-containing elastomer composition according to claim 1, further comprising a second curable perfluoropolymer which may be the same or different from the first curable perfluoropolymer.

16. The fluorine-containing elastomer composition according to claim 15, wherein the second curable perfluoropolymer comprises tetrafluoroethylene, at least one second perfluoroalkylvinyl ether and at least one second cure site monomer having a functional group to permit crosslinking of the perfluoropolymer and wherein content of the first perfluoroalkylvinyl ether in the first perfluoropolymer is different from content of the second perfluoroalkylvinyl ether in the second curable perfluoropolymer.

17. The fluorine-containing elastomer composition according to claim 16, wherein the content of the first perfluoroalkylvinyl ether in the first curable perfluoropolymer differs from the content of the second perfluoroalkylvinyl ether in the second curable perfluoropolymer by about 5 to about 25 % by mole.

18. The fluorine-containing elastomer composition according claim 1, wherein the perfluoroalkylvinyl ether is perfluoromethylvinyl ether.

19. The fluorine-containing elastomer composition according to claim 1, wherein the barium titanate has a stoichiometric ratio and a chemical structure of BaTiO_3 .

20. The fluorine-containing elastomer composition according to claim 1, wherein the composition comprises a cross-linking agent capable of reacting with the functional group of the at least one cure site monomer, the functional group is selected from the group consisting of

nitrile, carboxyl and alkoxycarbonyl, and the cross-linking agent is present in the composition in an amount of about 0.6 to about 0.9 weight percentage.

21. A sealing material for semiconductor manufacturing equipment made from the fluorine-containing elastomer composition of claim 1.

22. A cured perfluoroelastomeric composition, comprising

(a) a cured perfluoroelastomer formed by the curing reaction of (i) a first curable perfluoropolymer comprising tetrafluoroethylene, at least one perfluoroalkylvinyl ether and at least one first cure site monomer having a functional group to permit crosslinking of the first curable perfluoropolymer and (ii) a first curative; and

(b) barium titanate.

23. The cured perfluoroelastomeric composition according to claim 22, further comprising a second cured perfluoroelastomer formed from the curing reaction of (i) a second curable perfluoropolymer having a second cure site monomer, wherein the second curable perfluoropolymer may be the same or different from the first curable perfluoropolymer and (ii) a curative which may be the same or different from the first curative, wherein the cure site of the at least one first cure site monomer and/or the cure site of the second cure site monomer is a functional group selected from the group consisting of a nitrile group, a carboxyl group and an alkoxycarbonyl group.

24. The cured perfluoroelastomeric composition according to claim 23, wherein at least one of the first perfluoroelastomer and the second perfluoroelastomer has a benzoimidazole cross-linking structure.

25. The cured perfluoroelastomeric composition according to claim 22, wherein the barium titanate has a stoichiometric ratio and a chemical structure of BaTiO_3 .

26. The cured perfluoroelastomeric composition according to claim 22, wherein the barium titanate has an average particle size of at least 200 nm.

27. The cured perfluoroelastomeric composition according to claim 26, wherein the barium titanate has an average particle size of about 300 nm to about 1200 nm.

28. The cured perfluoroelastomeric composition according to claim 27, wherein the barium titanate has an average particle size of about 500 nm to about 1000 nm.

29. The cured perfluoroelastomeric composition according to claim 22, wherein the barium titanate is present in a mixture of at least two different average particle sizes.

30. The cured perfluoroelastomeric composition according to claim 22, comprising from about 1 to about 50 parts by weight of the barium titanate per 100 parts by weight of the first curable perfluoropolymer.

5 31. The cured perfluoroelastomeric composition according to claim 30, comprising about 5 to about 30 parts by weight of the barium titanate per 100 parts by weight of the first curable perfluoropolymer.

32. The cured perfluoroelastomeric composition according to claim 31, comprising about 10 to about 20 parts by weight of the barium titanate per 100 parts by weight of the first curable perfluoropolymer.

10 33. A molded article comprising the cured perfluoroelastomeric composition according to claim 22.

34. The molded article according to claim 33, wherein the molded article is an O-ring, a seal or a gasket.

15 35. The molded article according to claim 33, wherein the molded article is bonded to a surface comprising a metal or a metal alloy.

36. The molded article according to claim 35, wherein the molded article is bonded to surface of a door for sealing a semiconductor-processing chamber.

37. A method for making a cured perfluoroelastomeric composition comprising:

(a) preparing a curable perfluoroelastomeric composition by combining:

20 (i) a first curable perfluoropolymer comprising tetrafluoroethylene, a perfluoroalkyl vinyl ether and at least one first cure site monomer having a cure site;

(ii) at least one curative capable of curing the cure site of the at least one first cure site monomer; and

(iii) barium titanate;

25 (b) curing the curable perfluoropolymer in the perfluoroelastomeric composition to form a cured perfluoroelastomeric composition.

38. The method for making a cured perfluoroelastomeric composition according to claim 37, wherein the barium titanate has a stoichiometric ratio and a chemical structure of BaTiO_3 .

30 39. The method for making a cured perfluoroelastomeric composition according to claim 37, wherein the cured perfluoroelastomeric composition comprises a benzoimidazole cross-linking structure.

40. The method for making a cured perfluoroelastomeric composition according to claim 37, further comprising forming the curable perfluoroelastomeric composition into a molded article while curing the curable perfluoroelastomeric composition.

5 41. The method for making a cured perfluoroelastomeric composition according to claim 37, wherein the barium titanate has an average particle size of at least 200 nm.

42. The method for making a cured perfluoroelastomeric composition according to claim 41, wherein the barium titanate has an average particle size of about 300 nm to about 1200 nm.

10 43. The method for making a cured perfluoroelastomeric composition according to claim 42, wherein the barium titanate has an average particle size of about 500 nm to about 1000 nm.

44. The method for making a cured perfluoroelastomeric composition according to claim 37, wherein the barium titanate is present in a mixture of at least two different average particle sizes.

15 45. The method for making a cured perfluoroelastomeric composition according to claim 37, comprising from about 1 to about 200 parts by weight of the barium titanate per 100 parts by weight of the first curable perfluoropolymer.

20 46. The method for making a cured perfluoroelastomeric composition according to claim 45, comprising about 5 to about 50 parts by weight of the barium titanate per 100 parts by weight of the first curable perfluoropolymer.

47. The method for making a cured perfluoroelastomeric composition according to claim 46, comprising about 10 to about 20 parts by weight of the barium titanate per 100 parts by weight of the first curable perfluoropolymer.

25 48. In a method of processing in a processing apparatus having a sealing material therein, wherein the process includes use of high temperatures and/or use of a ClF_3 and/or NF_3 gas or plasma, the improvement comprising,

the sealing material including a cured perfluoroelastomeric composition comprising barium titanate.

INTERNATIONAL SEARCH REPORT

International application No.

PCT/US 09/67592

A. CLASSIFICATION OF SUBJECT MATTER

IPC(8) - C08L 27/18, 27/22 (2010.01)

USPC - 524/546; 525/326.3

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC(8) -- C08L 27/18, 27/22 (2010.01) and USPC -- 524/546; 525/326.3

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched
USPC -- 524/544, 545, 546; 525/199, 326.3

Electronic data base consulted during the international search (name of data base and, where practicable, search terms used)

PubWest (PGPB,USPT,USOC,EPAB,JPAB); USPTO; Espacenet; Google Patents; Google Scholar; Google -- please see extra sheet for Search Terms Used

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
Y	US 2008/0287627 A1 (Noguchi et al.) 20 November 2008 (20.11.2008) para [0009]; [0027]; [0031]; [0034]; [0035]; [0036]; [0038]; [0043]; [0044]; [0045]; [0052]; [0064]; [0088];	1-48
Y	US 2006/0293432 A1 (Hirano et al.) 28 December 2006 (28.12.2006) para [0003]; [0021]; [0022]; [0027]; [0032]; [0051]; [0071]; [0102]; [0103]; [0110]; [0124]; [0126]	1-48
Y	US 2007/0252771 A1 (Maezowa et al.) 01 November 2007 (01.11.2007) para [0019]; [0082]; [0084]; [0086]; [0110]; [0131]; [0132]; [0133]	5, 29, 44

☐ Further documents are listed in the continuation of Box C.

* Special categories of cited documents:

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"O" document referring to an oral disclosure, use, exhibition or other means

"P" document published prior to the international filing date but later than the priority date claimed

"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention

"X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone

"Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art

"&" document member of the same patent family

Date of the actual completion of the international search

20 JANUARY 2010 (20.01.2010)

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INTERNATIONAL SEARCH REPORT

International application No.

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Search Terms Used:

AVERAGE PARTICLE SIZE BARIUM TITANATE COMBINATION DIFFERENT FILLERS\$ FLUOROPOLYMER FLUORORESIN
FLUROELASTOMER FLUORINATED POLYMER FLUORINATED ELASTOMER FLUORINATED RESIN MULTIPLE
PERFLUOROALKYL VINYL ETHER PERFLUORO\$ PLASMA PLASMA RESISTAN\$ PLASMA-RESISTAN\$ POLYMER POLYMERIC
TWO VARIED VARIOUS