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(54) Title: CATALYSTS SYSTEMS BASED ON CARBONYLAMINO FULVENES

(57) Abstract: The present invention discloses metallic complexes based on carbonylamino fulvene ligands, their method of preparation and their use in the oligomerisation or polymerisation of ethylene and alpha-olefins.



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CATALYSTS SYSTEMS BASED ON CARBONYLAMINO FULVENES.

The present invention discloses catalyst components based on carbonylamino fulvene ligands their method of preparation and their use in the polymerisation of ethylene and alpha-olefins.

Several ligands have been described in literature, some of which were tested in complexation with metals but none of them have been used as catalysts for the polymerisation of ethylene or alpha-olefins. Some ligands are described for example in Lloyd and Preston (D. Lloyd, N.W. Preston, J. Chem. Soc. C, **1969**, 2464-2469.) or by Linn and Sharkey (W.J. Linn, W.G. Sharkey J. Am. Chem. Soc. **1957**, 79, 4970-2.) or in Snyder et al. (C.A. Snyder, J.P. Selegue, N.C. Tice, C.E. Wallace. M.T. Blankenbuehler, S. Parkin, K.D.E. Allen, R.T. Beck, J. Am. Chem. Soc. **2005**, 127, 15010-11.) or in Dong et al. (Y.B. Dong, Y. Geng, J.P. Ma and R.Q. Huang, Inorg. Chem. **2005**, 44, 1693-1703.)

There is a need to develop new catalyst system having good activity and able to produce polymers tailored to specific needs.

It is an aim of the present invention to prepare new catalyst components that can be used in the polymerisation of ethylene and alpha-olefins.

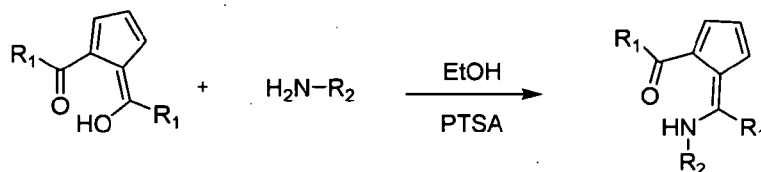
It is also an aim of the present invention to provide very active catalyst components.

It is another aim of the present invention to provide a method for polymerising or copolymerising ethylene and alpha-olefins.

The present invention reaches, at least partially, any one of those aims.

Accordingly, the present invention discloses a method for preparing a metallic complex that comprises the steps of:

- a) preparing a carbonylamino fulvene ligand by condensation reaction of an hydroxycarbonyl fulvene ligand with a primary amine, in a polar solvent and with an acid catalyst



wherein R₁ and R₂ are the same or different and are selected from alkyl, aryl, alkylaryl, arylalkyl having at most 20 carbon atoms or heteroatoms-containing groups;

- b) providing a metallic precursor MZ_n wherein M is a metal Group 6 to 11 of the Periodic Table, Z is a negative counter-anion and n is the valence of M;
 c) complexing the metallic precursor of step b) with the carbonylamino fulvene of step a);
 d) retrieving a metallic complex.

Preferably, both R₁ are the same and are selected from alkyl, unsubstituted or substituted phenyl (Ph), CHPh₂ wherein Ph may be substituted or not, or the R₁ groups include heteroatom(s)-containing units. More preferably R is CHPh₂ or cyclohexyl or paramethoxyphenyl. Most preferably R₁ is CHPh₂.

Preferably R₂ is CH₂pyridine or includes heteroatoms-containing units. More preferably R₂ is CH₂pyridine or furan.

Preferred embodiments according to the present invention are characterised by the following pairs:

Both R₁ are CHPh₂ and R₂ is CH₂pyridine,

Both R₁ are paramethoxyphenyl and R₂ is CH₂pyridine,

Both R₁ are cyclohexyl and R₂ is furan.

Most preferably, both R₁ are CHPh₂ and R₂ is CH₂pyridine.

Preferably, M is CrII, CrIII or Ni, more preferably, it is CrII.

Preferably Z is halogen or acetate, more preferably, it is Cl.

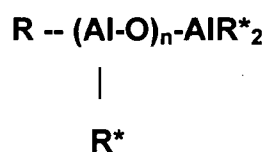
The preferred polar solvent is ethanol and the preferred acid catalyst is paratoluene sulfonic acid.

Several types of metallic complexes can be formed, one where the metal is coordinated to one ligand and one where the metal is coordinated to two ligands. The relative amounts of each ligand and metal unit depend upon the nature of ligand and of the metal. The amount of ligand must therefore be of at least one equivalent of ligand per metallic equivalent. In a preferred embodiment according to the present invention, the metal is CrII and it is coordinated to one ligand or two ligands.

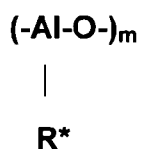
The present invention further discloses an active catalyst system comprising the metallic complex and an activating agent having an ionising action.

Suitable activating agents are well known in the art. The activating agent can be an aluminium alkyl represented by formula $\text{AlR}^+_n\text{X}_{3-n}$ wherein R^+ is an alkyl having from 1 to 20 carbon atoms and X is a halogen. The preferred alkylating agents are triisobutyl aluminium (TIBAL) or triethyl aluminium (TEAL).

Alternatively and preferably, it is an aluminoxane and comprise oligomeric linear and/or cyclic alkyl aluminoxanes represented by formula



for oligomeric, linear aluminoxanes and by formula



for oligomeric, cyclic aluminoxane,

wherein n is 1-40, preferably 10-20, m is 3-40, preferably 3-20 and R^* is a C_1 - C_8 alkyl group and preferably methyl.

The amount of activating is selected to give an Al/M ratio of from 100 to 3000, preferably of about 1000.

Suitable boron-containing activating agents may comprise a triphenylcarbenium boronate such as tetrakis-pentafluorophenyl-borato-triphenylcarbenium as described in EP-A-0427696, or those of the general formula $[L^+H] + [B Ar_1 Ar_2 X_3 X_4]^-$ as described in EP-A-0277004 (page 6, line 30 to page 7, line 7). The amount of boron-containing activating agent is selected to give B/M ratio of from 0.5 to 5, preferably of about 1.

The preferred activating agent is methylaluminoxane (MAO).

In another embodiment, according to the present invention, the metallic complex may be deposited on a conventional support impregnated with an activating agent.

Preferably, the conventional support is silica impregnated with methylaluminoxane (MAO). Alternatively, it can be an activating support such as fluorinated alumina silica.

The present invention further discloses a method for preparing an active catalyst system that comprises the steps of:

- a) providing a carbonylamino fulvene ligand;
- b) complexing the ligand of step a) with a metallic salt MZ_n in a solvent with an acid catalyst;
- c) retrieving a catalyst component;
- d) optionally depositing the catalyst component of step c) on a support;
- e) activating the catalyst component of step c) or step d) with an activating agent having an ionising action;
- f) optionally adding a scavenger;
- g) retrieving an active oligomerisation or polymerisation catalyst system.

Alternatively, in step d), the catalyst component is deposited on a support impregnated with an activating agent or on an activating support. In that case, the activating step e) is not necessary.

The scavenger may be selected from triethylaluminium, triisobutylaluminum, tris-n-octylaluminium, tetraisobutyldialuminoxane or diethyl zinc.

The active catalyst system is used in the oligomerisation and in the polymerisation of ethylene and alpha-olefins.

The present invention discloses a method for the oligomerisation or the homo- or co-polymerisation of ethylene and alpha-olefins that comprises the steps of:

- a) injecting the active catalyst system into the reactor;
- b) injecting the monomer and optional comonomer;
- c) maintaining under polymerisation conditions;
- d) retrieving the oligomers and/or polymer.

The pressure in the reactor can vary from 0.5 to 60 bars, preferably from 15 to 45 bars. The productivity of the catalyst system increases with increasing pressure.

The polymerisation temperature can range from 10 to 100°C, preferably from 25 to 55°C. The productivity of the catalyst system decreases with increasing temperature.

Preferably the monomer and optional comonomer are selected from ethylene, propylene or 1-hexene.

The present invention also discloses the polymers obtained with the new catalyst systems.

List of figures.

Figure 1 represents examples of ligands prepared according to the present invention.

Figure 2 represents the molecular structure of ligand D obtained by X-Ray.

Figure 3 represents the molecular structure of the metallic compound resulting from the complexation of CrCl_2 with ligand L.

Examples.

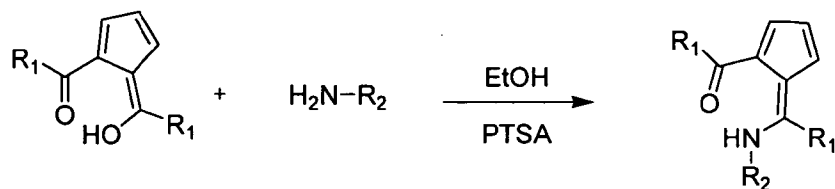
Synthesis of ligands.

The ligands were prepared following methods similar to those described for example in Lloyd and Preston (D. Lloyd, N.W. Preston, J. Chem. Soc. C, 2464-2469, 1969.)

All reactives were purchased from commercially available sources and used without purification and the solvents were purified following standard procedures. The NMR spectra were recorded either on a Brücker ARX 200 spectrometer, at 200 MHz for ^1H spectra and at 50 MHz for ^{13}C spectra, or on a Brücker AC 300P at 300 MHz for ^1H spectra and at 75 MHz for ^{13}C spectra. Mass spectras were obtained with a high resolution mass spectrometer Varian MAT 311 and microanalyses were carried out on a Flash EA1112 CHNS/O Thermo Electron (Centre Régional de Mesures des Physiques de l'Ouest, Rennes, France). Crystallographic data collection, unit cell constant and space group determination were carried out on an automatic 'Enraf Nonius FR590' NONIUS Kappa CCD diffractometer with graphite monochromatised $\text{Mo-K}\alpha$ radiation at 120 K. The cell parameters are obtained with Denzo and Scalepack with 10 frames (psi rotation: 1° per frame). The structure was solved with SIR-97. The whole structure was refined with SHELXL97 by the full-matrix least-square techniques.

Parallel synthesis of carbonylamino fulvene ligands.

Carbonylamino fulvene ligands A, B, C, D, ... U were synthesised in parallel with a Büchi Syncore. In each tube, 0.2 to 0.6 mmol of hydroxycarbonyl fulvene were introduced with 1.1 equivalents of amine, 0.5 mg of paratoluene-sulfonic acid (PTSA) and 20 ml of ethanol.



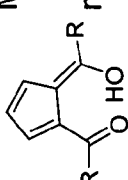
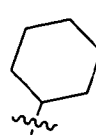
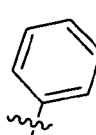
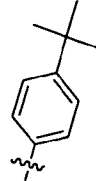
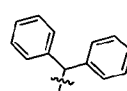
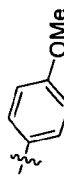
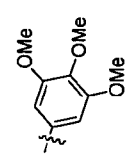
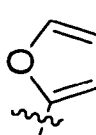
The mixtures were heated for a period of time of 12 h at a temperature of 80 °C. Ligands A, B, C, J, K and L were heated for 12 additional hours at a temperature of 110 °C. Ligands D and H were crystallised from ethanol. Otherwise, the solvent was evaporated under vacuum for 1 night and crude products were purified by column chromatography on silica gel and dried over MgSO₄ to afford new carbonylamino fulvene ligands. Several ligands synthesised according to the present invention are displayed in Figure 1. The results for different amines are displayed in TABLE I for benzylamine, in TABLE II for furfurylamine and in TABLE III for picolylamine.

EtOAc= ethyl acetate, DCM=dichloromethane, C7=heptane, EtOH=ethanol

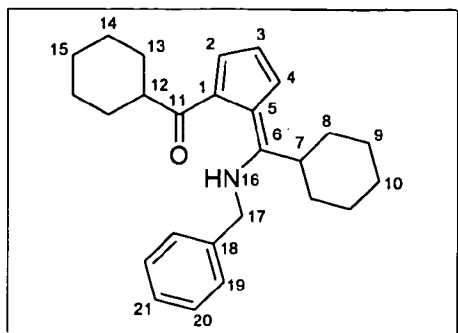
O=C1C(=C(R)C=C1)C(=O)NCC2=CC=CC(=O)O2

EtOAc= ethyl acetate, **DCM**=dichloromethane, **C7**=heptane, **EtOH**=ethanol

TABLE III Synthesis with 2-picolyamine

 R=	 M hydroxycarbonylfulvene (g/mol)	 m hydroxycarbonylfulvene (mg)	 n hydroxycarbonylfulvene (mmol)	 V amine (μ L)	 n amine (mmol)	 C	 E	 I	 L	 O	 R	 U
	286,41	274,31	386,53	454,56	334,37	454,16	254,24					
	174,7	158,1	223,7	205,6	170,3	161,8	114,3					
	0,610	0,576	0,579	0,452	0,509	0,356	0,450					
	73	69	70	54	61	43	54					
	0,671	0,634	0,637	0,498	0,560	0,392	0,495					
	EtOAc/C7 1/1	EtOAc/C7 1/1	EtOAc/C7 1/1	Ether/C7 1/1	EtOAc/C7 2/1	EtOAc/C7 2/1	EtOAc/C7 1/1					
	376,55	364,45	476,67	544,70	424,51	544,30	344,38					
	187	193	187	174	129	109	93					
	0,497	0,530	0,392	0,319	0,304	0,200	0,270					
	81	92	68	71	60	56	60					

EtOAc= ethyl acetate, DCM=dichloromethane, C7=heptane, EtOH=ethanol

Ligand A: 1-(cyclohexanoyl)-6-benzylamino-6-cyclohexyl fulvene

$C_{26}H_{33}NO$
Mol. Wt.: 375,54632

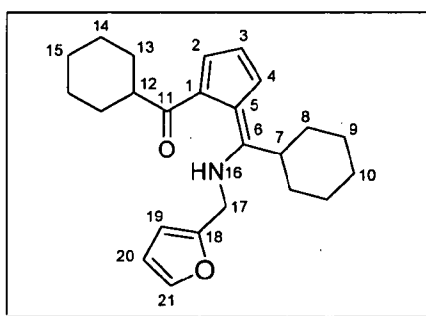
Pale green solid
Yield : 66 %

1H NMR ($CDCl_3$, 300 MHz, ppm) δ : 14.87 (1H, s, $N^{16}H$), 7.31 (2H, d, $J=0.03$ Hz, C^2H and C^4H), 7.21 (5H, m, $C^{19}H$, $C^{20}H$ and $C^{21}H$), 6.19 (1H, s, C^3H), 4.63 (2H, d, $J=0.23$ Hz, $C^{17}H$), 3.44-2.75 (2H, m, $C^{12}H$ and C^7H), 1.83-0.97 (20H, m, C^8H , C^9H , $C^{10}H$, $C^{13}H$, $C^{14}H$, $C^{15}H$)

^{13}C NMR ($CDCl_3$, 50 MHz, ppm) δ : 199.23 (C^{11}), 173.58 (C^6), 137.93 (C^{18}), 131.82 (C^2), 130.60 (C^4), 128.84 (C^{20}), 127.67 (C^{21}), 127.14 (C^{19}), 123.97 (C^1), 117.01 (C^5), 116.03 (C^3), 47.87 (C^{17}), 47.04 (C^{12}), 42.03 (C^{15}), 32.26 (C^8), 31.06 (C^{10}), 26.99 (C^{13}), 26.32 (C^{14}), 26.21 (C^9), 25.85 (C^7).

HRMS: Calcd. for $M^+.(C_{26}H_{33}NO)$ m/z = 375.25621, found 375.2571 (2 ppm).

Anal. Cald for $C_{26}H_{33}NO$: C: 83.15, H: 8.86, N: 3.73, O: 4.26 found C: 82.71, H: 8.91, N: 4.00.

Ligand B: 1-(cyclohexanoyl)-6-(furan-2-yl methyl)amino-6-cyclohexyl fulvene

$C_{24}H_{31}NO_2$
Mol. Wt.: 365,51

Pale brown solid
Yield : 85 %

1H NMR ($CDCl_3$, 300 MHz, ppm) δ : 14.92 (1H, s, $N^{16}H$), 7.56 (1H, d, $J=0.02$ Hz, $C^{21}H$), 7.43 (2H, d, $J=0.025$ Hz, C^2H and C^4H), 7.28 (1H, m, $C^{20}H$), 6.34-6.40 (2H, m, C^3H and $C^{19}H$), 4.78 (2H, d, $J=0.24$ Hz, $C^{17}H$), 3.52-3.09 (2H, m, $C^{12}H$ and C^7H), 2.19-1.31 (20H, m, C^8H , C^9H , $C^{10}H$, $C^{13}H$, $C^{14}H$, $C^{15}H$)

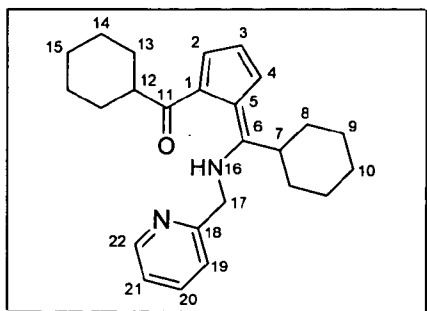
^{13}C NMR ($CDCl_3$, 50 MHz, ppm) δ : 199.11 (C^{11}), 173.22 (C^6), 150.40 (C^{18}), 142.43 (C^{21}), 132.02 (C^2), 130.83 (C^4), 124.01 (C^1), 117.06 (C^5), 116.12 (C^3), 110.68 (C^{20}),

107.90 (C¹⁹), , 46.93 (C⁷), 41.74 (C¹⁰), 40.95 (C¹⁷), 32.44 (C¹³), 30.98 (C¹⁵), 27.00 (C⁸), 26.28 (C⁹), 26.17 (C¹⁴), 25.87 (C¹²).

HRMS: Calcd. for M⁺.(C₂₄H₃₁NO₂) m/z= 365.23548, found 365.2344 (2 ppm).

Anal. Cald for C₂₄H₃₁NO₂: C: 78.86, H: 8.55, N: 3.83, O: 8.75 found C: 78.54, H: 8.60, N: 3.89.

Ligand C: 1-(cyclohexanoyl)-6-(pyridine-2-methyl)amino-6-cyclohexyl fulvene



C₂₅H₃₂N₂O
Mol. Wt.: 376,53438

Pale brown solid
Yield : 81 %

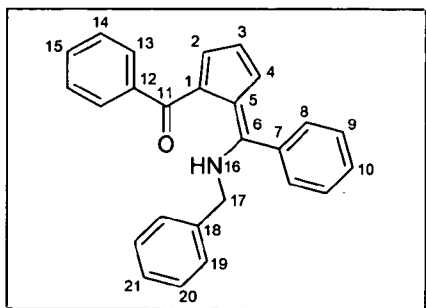
¹H NMR (CDCl₃, 300 MHz, ppm) δ: 15.01 (1H, s, N¹⁶H), 8.59 (1H, d, J=0.01 Hz, C²²H), 7.68 (1H, t, J=0.025 Hz, C²⁰H), 7.45-7.51 (2H, m, C²H and C⁴H), 7.36 (1H, s, C¹⁹H), 7.22 (1H, t, J=0.02 Hz, C²¹H), 6.34 (1H, t, J=0.02 Hz, C³H), 4.92 (2H, d, J=0.26 Hz, C¹⁷H), 3.60-2.95 (2H, m, C¹²H and C⁷H), 1.96-1.31 (20H, m, C⁸H, C⁹H, C¹⁰H, C¹³H, C¹⁴H, C¹⁵H)

¹³C NMR (CDCl₃, 50 MHz, ppm) δ: 199.72 (C¹¹), 173.84 (C⁶), 157.76 (C¹⁸), 149.35 (C²²), 137.09 (C¹⁹), 135.48 (C²⁰), 132.15 (C²), 130.88 (C⁴), 124.03 (C¹), 122.59 (C²¹), 117.06 (C⁵), 116.28 (C³), 49.88 (C¹⁷), 46.95 (C⁷), 41.86 (C¹⁰), 32.38 (C¹³), 31.02 (C¹⁵), 26.82 (C⁸), 26.26 (C⁹), 26.17 (C¹⁴), 25.76 (C¹²).

HRMS: Calcd. for M⁺.(C₂₅H₃₂N₂O) m/z= 376.25146, found 376.2503 (3 ppm).

Anal. Cald for C₂₅H₃₂N₂O : C: 79.75, H: 8.57, N: 7.44, O: 4.25 found C: 79.34, H: 8.67, N: 7.27.

Ligand D: 1-benzoyl-6-benzylamino-6-phenyl fulvene



C₂₆H₂₁NO
Mol. Wt.: 363,45104

Brown solid
Yield : 90 %

¹H NMR (CDCl₃, 300 MHz, ppm) δ: 15.28 (1H, s, N¹⁶H), 7.51 (2H, m, C¹³H), 7.20-6.88 (13H, m, C⁸H, C⁹H, C¹⁰H, C¹⁴H, C¹⁵H, C¹⁹H, C²⁰H, C²¹H), 6.74 (1H, m, C²H), 5.96 (1H, m, C⁴H), 5.75 (1H, t, J=0.02 Hz, C³H), 3.68 (2H, d, J=0.03 Hz, C¹⁷H)

¹³C NMR (CDCl₃, 50 MHz, ppm) δ: 191.13 (C¹¹), 166.39 (C⁶), 142.42 (C¹²), 139.71 (C⁷), 137.05 (C¹⁸), 135.80 (C¹⁵), 134.04 (C¹⁰), 129.94 (C²), 129.72 (C⁴), 128.94 (C²⁰), 128.92 (C¹³), 128.40 (C¹⁴), 128.32 (C⁹), 127.87 (C⁸), 127.82 (C²¹), 127.47 (C¹⁹), 125.35 (C¹), 119.98 (C⁵), 118.20 (C³), 49.76 (C¹⁷).

HRMS: Calcd. for M⁺.(C₂₆H₂₁NO) m/z= 363.16231, found 363.1594 (1 ppm).

Anal. Cald for C₂₆H₂₁NO: C: 85.92, H: 5.82, N: 3.85, O: 4.40 found C: 85.46, H: 6.06, N: 3.99.

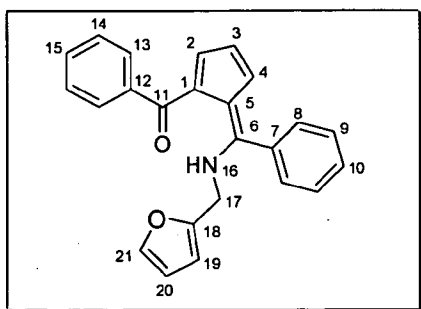
Crystallography. Single crystals of compound **D** suitable for a single crystal X-ray determination were obtained by evaporation of a saturated solution in THF.

Empirical formula	C ₂₆ H ₂₁ N O
Formula weight	363.44
Temperature	120(2) K
Wavelength	0.71073 Å
Crystal system, space group	Monoclinic, P 21/c
Unit cell dimensions	a = 6.22050(10) Å alpha = 90 deg. b = 17.5271(3) Å beta = 91.6710(10) deg. c = 18.2524(3) Å gamma = 90 deg.
Volume	1989.16(6) Å ³
Z, Calculated density	4, 1.214 Mg/m ³
Absorption coefficient	0.073 mm ⁻¹
F(000)	768
Crystal size	0.2 x 0.15 x 0.1 mm
Theta range for data collection	3.22 to 27.47 deg.
Limiting indices	-8<=h<=8, -22<=k<=22, -23<=l<=23
Reflections collected / unique	8885 / 4522 [R(int) = 0.0472]
Completeness to theta =	27.47 99.3 %
Absorption correction	none
Refinement method	Full-matrix least-squares on F ²
Data / restraints / parameters	4522 / 0 / 257
Goodness-of-fit on F ²	1.078
Final R indices [I>2sigma(I)]	R1 = 0.0607, wR2 = 0.1598

R indices (all data)	R1 = 0.0736, wR2 = 0.1702
Extinction coefficient	0.075(9)
Largest diff. peak and hole	0.337 and -0.271 e.Å ⁻³

The molecular structure of ligand D obtained by X-Ray is displayed in figure 2.

Ligand E: 1-benzoyl-6-(1-(furan-2-methyl)amino)-6-phenyl fulvene



$C_{24}H_{19}NO_2$
Mol. Wt.: 353,41316

Red oil
Yield : 94 %

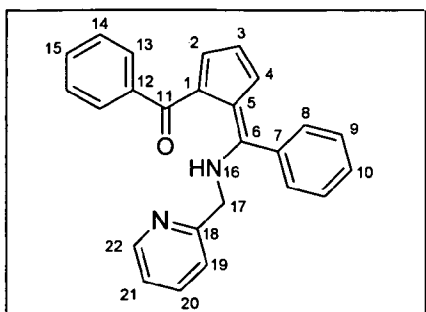
¹H NMR (CDCl₃, 300 MHz, ppm) δ: 14.36 (1H, s, N¹⁶H), 7.83 (2H, m, C¹³H), 7.59-7.50 (8H, m, C⁸H, C⁹H, C¹⁰H, C¹⁴H, C¹⁵H), 7.43 (1H, d, J=0.01 Hz, C²H), 7.20 (1H, m, C²¹H), 6.50 (1H, t, J=0.01 Hz, C²⁰H), 6.37 (1H, m, C⁴H), 6.33-6.30 (2H, m, C³H and C¹⁹H), 4.49 (2H, d, J=0.02 Hz, C¹⁷H)

¹³C NMR (CDCl₃, 50 MHz, ppm) δ: 191.13 (C¹¹), 166.07 (C⁶), 149.82 (C¹⁸), 142.82 (C²¹), 142.30 (C¹²), 139.99 (C⁷), 135.95 (C¹⁵), 133.85 (C¹⁰), 129.95 (C²), 129.77 (C⁴), 128.92 (C¹³), 128.53 (C¹⁴), 128.35 (C⁹), 127.84 (C⁸), 125.48 (C¹), 119.92 (C⁵), 118.35 (C³), 110.64 (C²⁰), 108.22 (C¹⁹), 42.93 (C¹⁷).

HRMS: Calcd. for M⁺.(C₂₄H₁₉NO₂) m/z= 353.14158, found 353.1417 (0 ppm).

Anal. Cald for C₂₄H₁₉NO₂: C: 81.56, H: 5.42, N: 3.96, O: 9.05 found C: 80.35, H: 5.64, N: 4.13.

Ligand F: 1-benzoyl-6-(1-(pyridine-2-methyl)amino)-6-phenyl fulvene



$C_{25}H_{20}N_2O$
Mol. Wt.: 364,4391

Brown solid
Yield : 92 %

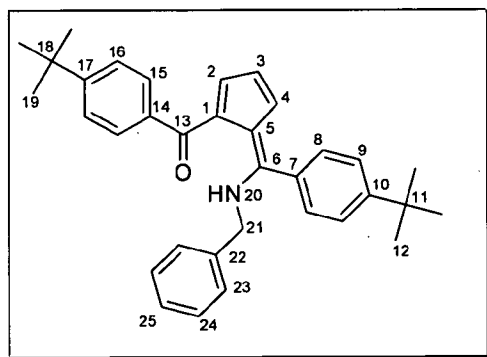
^1H NMR (CDCl_3 , 300 MHz, ppm) δ : 14.49 (1H, s, N^{16}H), 8.56 (1H, d, $J=0.01$ Hz, C^{22}H), 7.80 (2H, m, C^{13}H), 7.70 (1H, t, $J=0.025$ Hz, C^{20}H), 7.54-7.42 (9H, m, C^8H , C^9H , C^{10}H , C^{14}H , C^{15}H , and C^{19}H), 7.22-7.17 (2H, m, C^2H , C^{21}H), 6.48 (1H, m, C^4H), 6.28 (1H, t, $J=0.02$ Hz, C^3H), 4.67 (2H, d, $J=0.02$ Hz, C^{17}H)

^{13}C NMR (CDCl_3 , 50 MHz, ppm) δ : 191.14 (C^{11}), 166.69 (C^6), 156.79 (C^{18}), 149.59 (C^{22}), 142.27 (C^7), 139.99 (C^{12}), 137.13 (C^{20}), 136.02 (C^{10}), 133.89 (C^{15}), 129.95 (C^2), 129.72 (C^4), 128.92 (C^{13}), 128.31 (C^{14}), 128.30 (C^9), 127.81 (C^8), 125.50 (C^5), 122.60 (C^{19}), 121.31 (C^{21}), 119.97 (C^1), 118.33 (C^3), 51.41 (C^{17}).

HRMS: Calcd. for $\text{M}^+(\text{C}_{25}\text{H}_{20}\text{N}_2\text{O})$ m/z = 364.15756, found 364.1557 (5 ppm).

Anal. Calcd for : C: 82.39, H: 5.53, N: 7.69, O: 4.39 found C: 82.53, H: 5.56, N: 7.61.

Ligand G: 1-(4-tertbutylbenzoyl)-6-benzylamino-6-(4-tertbutylphenyl) fulvene



$\text{C}_{34}\text{H}_{37}\text{NO}$
Mol. Wt.: 475,66368

Yellow solid
Yield : 85 %

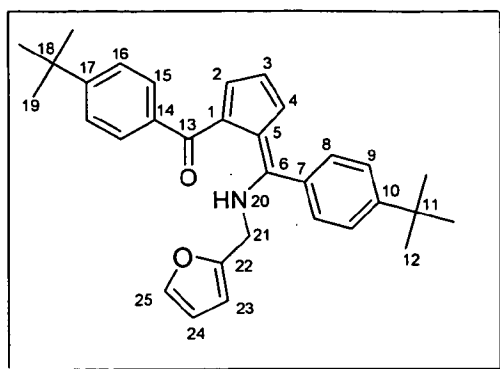
^1H NMR (CDCl_3 , 300 MHz, ppm) δ : 14.32 (1H, s, N^{20}H), 7.57 (2H, m, C^{15}H), 7.40-7.13 (11H, m, C^8H , C^9H , C^{16}H , C^{23}H , C^{24}H , C^{25}H), 6.34 (1H, m, C^2H), 6.11 (1H, m, C^4H), 5.17 (1H, t, $J=0.02$ Hz, C^3H), 4.38 (2H, d, $J=0.02$ Hz, C^{21}H), 1.30 (18H, d, $J=0.03$ Hz, C^{12}H_3 and C^{19}H_3).

^{13}C NMR (CDCl_3 , 50 MHz, ppm) δ : 191.05 (C^{13}), 166.71 (C^6), 153.13 (C^{17}), 152.84 (C^{10}), 139.50 (C^{14}), 139.28 (C^7), 137.13 (C^{22}), 135.51 (C^2), 131.01 (C^4), 128.87 (C^{24}), 128.79 (C^8), 128.20 (C^9), 127.63 (C^{16}), 127.44 (C^{15}), 125.19 (C^{25}), 125.07 (C^{23}), 124.70 (C^1), 119.92 (C^5), 117.70 (C^3), 49.68 (C^{21}), 34.92 (C^{11} and C^{18}), 31.35 (C^{12} and C^{19}).

HRMS: Calcd. for $\text{M}^+(\text{C}_{34}\text{H}_{37}\text{NO})$ m/z = 475.28752, found. 475.2899 (5 ppm)

Anal. Calcd for $\text{C}_{34}\text{H}_{37}\text{NO}$: C: 85.85, H: 7.84, N: 2.94, O: 3.36 found C: 86.13, H: 7.99, N: 2.99.

Ligand H: 1-(4-tertbutylbenzoyl)-6-(1-(furan-2-methyl)amino)-6-(4-tertbutylphenyl) fulvene



$C_{32}H_{35}NO_2$
Mol. Wt.: 465,6258

Yellow solid
Yield : 77 %

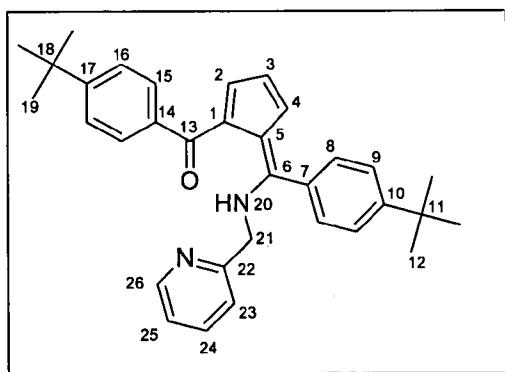
1H NMR ($CDCl_3$, 300 MHz, ppm) δ : 14.14 (1H, s, $N^{20}H$), 7.56 (2H, d, $J=0.03$ Hz, $C^{15}H$), 7.42-7.27 (6H, m, C^8H , C^9H , $C^{16}H$), 7.23 (1H, s, C^2H), 7.03 (1H, d, $J=0.01$ Hz, $C^{25}H$), 6.32 (1H, t, $J=0.01$ Hz, $C^{24}H$), 6.18-6.09 (1H, m, C^3H , C^4H and $C^{23}H$), 4.32 (2H, d, $J=0.02$ Hz, $C^{21}H$), 1.26 (18H, d, $J=0.03$ Hz, $C^{12}H_3$ and $C^{19}H_3$)

^{13}C NMR ($CDCl_3$, 50 MHz, ppm) δ : 191.09 (C^{13}), 166.41 (C^6), 153.17 (C^{17}), 152.91 (C^{10}), 149.98 (C^{22}), 142.66 (C^{25}), 139.62 (C^{14}), 139.46 (C^7), 135.72 (C^2), 130.88 (C^4), 128.93 (C^8), 128.38 (C^9), 125.39 (C^{16}), 125.15 (C^{15}), 124.71 (C^1), 119.93 (C^5), 117.92 (C^3), 110.58 (C^{24}), 108.08 (C^{23}), 42.99 (C^{21}), 34.88 (C^{11} and C^{18}), 31.39 (C^{12} and C^{19}).

HRMS: Calcd. for M^+ ($C_{32}H_{35}NO_2$) m/z = 465.26678, found 465.2669 (0 ppm).

Anal. Cald for $C_{32}H_{35}NO_2$: C: 82.54, H: 7.58, N: 3.01, O: 3.01 found C: 81.98, H: 7.57, N: 3.14.

Ligand I: 1-(4-tertbutylbenzoyl)-6-(1-(pyridine-2-methyl)amino)-6-(4-tertbutylphenyl) fulvene



$C_{33}H_{36}N_2O$
Mol. Wt.: 476,65174

Yellow solid
Yield : 68 %

1H NMR ($CDCl_3$, 300 MHz, ppm) δ : 14.35 (1H, s, $N^{20}H$), 8.37 (1H, m, $C^{26}H$), 7.54 (2H, m, $C^{15}H$), 7.48 (1H, d, $C^{24}H$), 7.35-7.19 (7H, m, C^8H , C^9H , $C^{16}H$, C^2H), 7.05 (1H, m, $C^{23}H$), 7.01 (1H, t, $J=0.02$ Hz, $C^{25}H$), 6.34 (1H, m, C^4H), 6.11 (1H, t, $J=0.02$ Hz, C^3H), 4.52 (2H, d, $J=0.02$ Hz, $C^{21}H$), 1.24 (18H, s, $C^{12}H_3$ and $C^{19}H_3$)

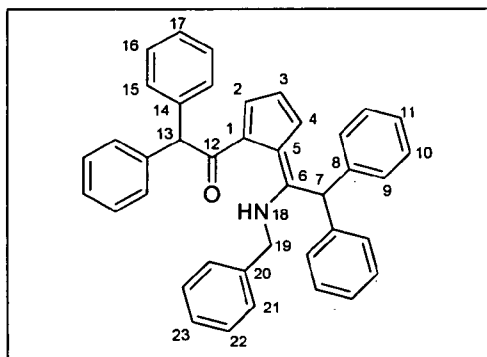
^{13}C NMR ($CDCl_3$, 50 MHz, ppm) δ : 191.07 (C^{13}), 167.04 (C^6), 157.02 (C^{26}), 153.21 (C^{17}), 152.85 (C^{10}), 149.53 (C^{24}), 139.70 (C^7), 139.46 (C^{14}), 137.12 (C^{23}), 135.90 (C^2), 130.93 (C^4), 128.99 (C^8), 128.21 (C^9), 125.46 (C^{16}), 125.16 (C^{15}), 124.74 (C^1),

122.50 (C²²), 121.24 (C²⁵), 120.06 (C⁵), 118.02 (C³), 51.47 (C²¹), 34.92 (C¹¹ and C¹⁸), 31.39 (C¹² and C¹⁹).

HRMS: Calcd. for M⁺.(C₃₃H₃₆N₂O) m/z= 476.28276, found 476.2830 (0 ppm).

Anal. Cald for C₃₃H₃₆N₂O : C: 83.15, H: 7.61, N: 5.88, O: 3.36 found C: 82.65, H: 7.77, N: 5.81.

Ligand J: 1-(diphenylacetyl)-6-benzylamino-6-(diphenylmethyl) fulvene



C₄₀H₃₃NO
Mol. Wt.: 543,69612

Yellow solid
Yield : 52 %

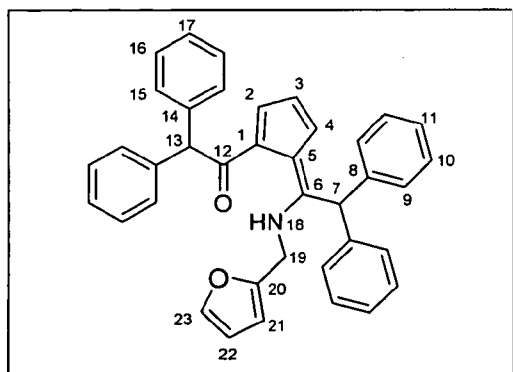
¹H NMR (CDCl₃, 300 MHz, ppm) δ: 14.95 (1H, s, N¹⁸H), 7.64 (1H, m, C²H), 7.38-7.10 (25H, m, C⁹H, C¹⁰H, C¹¹H, C¹⁵H, C¹⁶H, C¹⁷H, C²¹H, C²²H and C²³H), 7.04 (1H, m, C⁴H), 6.31 (1H, s, C¹³H), 6.25-6.22 (2H, m, C³H and C⁷H), 4.54 (2H, d, J=0.02 Hz, C¹⁹H).

¹³C NMR (CDCl₃, 50 MHz, ppm) δ: 193.03 (C¹²), 167.51 (C⁶), 141.30 (C¹⁴), 138.85 (C⁸), 136.42 (C²⁰), 134.93 (C²), 133.10 (C⁴), 129.27 (C¹⁶), 128.95 (C¹⁵), 128.92 (C¹⁰), 128.58 (C²²), 128.39 (C⁹), 127.31 (C¹⁷), 126.96 (C²¹), 126.65 (C¹¹), 125.92 (C¹), 120.39 (C⁵), 117.60 (C³), 58.95 (C¹³), 52.10 (C¹⁹), 49.46 (C⁷).

HRMS: Calcd. for M⁺.(C₄₀H₃₃NO) m/z= 543.25621, found 543.2539 (4 ppm).

Anal. Cald for C₄₀H₃₃NO: C: 88.36, H: 6.12, N: 2.58, O: 2.94 found C: 87.74, H: 6.07, N: 2.52.

Ligand K: 1-(diphenylacetyl)-6-(1-(furan-2-methyl)amino)-6-(diphenylmethyl) fulvene



C₃₈H₃₁NO₂
Mol. Wt.: 533,65824

Yellow solid
Yield : 73 %

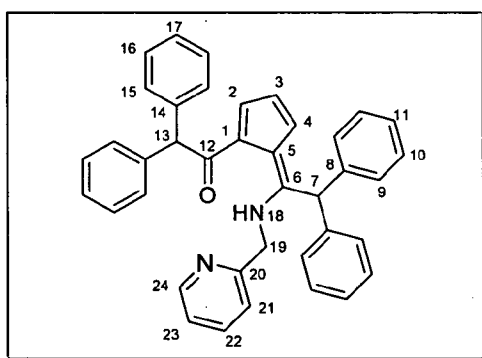
¹H NMR (CDCl₃, 300 MHz, ppm) δ: 14.80 (1H, s, N¹⁸H), 7.64 (1H, m, C²H), 7.45-7.27 (21H, m, C⁹H, C¹⁰H, C¹¹H, C¹⁵H, C¹⁶H, C¹⁷H and C²³H), 7.07 (1H, m, C⁴H), 6.44 (1H, s, C¹³H), 6.32 (1H, m, C²²H), 6.25 (1H, t, J=0.02Hz, C³H), 6.19 (1H, s, C²¹H), 4.42 (2H, d, J=0.02 Hz, C¹⁹H).

¹³C NMR (CDCl₃, 50 MHz, ppm) δ: 192.99 (C¹²), 167.02 (C⁶), 149.35 (C²⁰), 142.34 (C²³), 141.27 (C¹⁴), 138.66 (C⁸), 135.22 (C²), 133.07 (C⁴), 129.27 (C¹⁶), 128.99 (C¹⁵), 128.93 (C¹⁰), 128.36 (C⁹), 127.36 (C¹⁷), 126.62 (C¹¹), 126.01 (C¹), 120.29 (C⁵), 117.74 (C³), 110.52 (C²²), 107.87 (C²¹), 58.94 (C¹³), 52.10 (C¹⁹), 43.26 (C⁷).

HRMS: Calcd. for M⁺.(C₃₈H₃₁NO₂) m/z= 533.23548, found 533.2349 (1 ppm).

Anal. Cald for C₃₈H₃₁NO₂: C: 85.52, H: 5.86, N: 2.62, O: 6.00 found C: 85.27, H: 5.95, N: 2.59.

Ligand L: 1-(diphenylacetyl)-6-(1-(pyridine-2-methyl)amino)-6-(diphenylmethyl) fulvene



C₃₉H₃₂N₂O
Mol. Wt.: 544,68418

Brown solid
Yield : 70 %

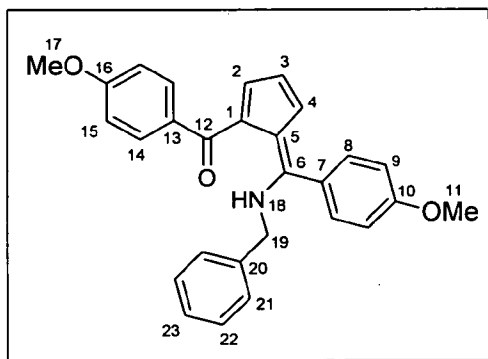
¹H NMR (CDCl₃, 300 MHz, ppm) δ: 14.94 (1H, t, J=0.02Hz, N¹⁸H,), 8.45 (1H, d, J=0.02Hz C²⁴H), 7.68 (1H, m, C²H), 7.50 (1H, t, J=0.03Hz C²²H), 7.46-7.27 (21H, m, C⁹H, C¹⁰H, C¹¹H, C¹⁵H, C¹⁶H, C¹⁷H, C²¹H and C²³H), 7.09 (1H, m, C⁴H), 6.39 (1H, s, C¹³H), 6.27 (1H, t, J=0.02Hz, C³H), 6.25 (1H, s, C⁷H), 4.76 (2H, d, J=0.02 Hz, C¹⁹H).

¹³C NMR (CDCl₃, 50 MHz, ppm) δ: 193.22 (C¹²), 167.99 (C⁶), 156.45 (C²⁴), 149.16 (C²²), 141.25 (C¹⁴), 138.54 (C⁸), 136.60 (C²¹), 135.50 (C²), 133.42 (C⁴), 129.27 (C¹⁶), 129.00 (C¹⁵), 128.85 (C¹⁰), 128.42 (C⁹), 127.22 (C¹⁷), 126.68 (C¹¹), 126.06 (C¹), 122.20 (C²²), 120.8 (C²⁵), 120.38 (C⁵), 117.90 (C³), (C²²), (C²¹), 59.03 (C¹³), 52.06 (C¹⁹), 51.17 (C⁷).

HRMS: Calcd. For M⁺.(C₃₉H₃₂N₂O) m/z= 544.25146, found 544.2533 (3 ppm).

Anal. Cald for C₃₉H₃₂N₂O: C: 86.00, H: 5.92, N: 5.14, O: 2.94 found C: 86.08, H: 6.14, N: 4.80.

Ligand M: 1-(4-methoxybenzoyl)-6-(benzylamino)-6-(4-methoxyphenyl) fulvene



$C_{28}H_{25}NO_3$
Mol. Wt.: 423,503

Yellow solid
Yield : 82 %

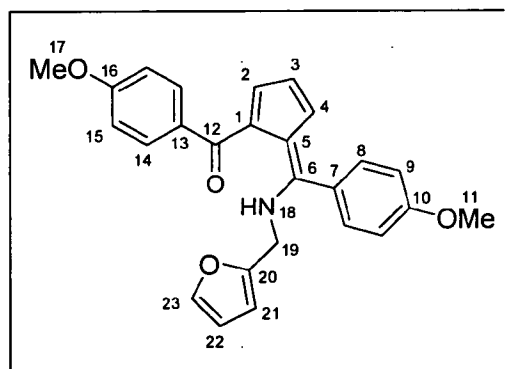
1H NMR ($CDCl_3$, 300 MHz, ppm) δ : 14.26 (1H, s, $N^{18}H$), 7.64 (2H, m, $C^{14}H$), 7.22-7.12 (7H, m, C^8H , $C^{21}H$, $C^{22}H$ and $C^{23}H$), 7.03 (1H, m, C^2H), 6.90 (4H, m, C^9H and $C^{15}H$), 6.12 (1H, m, C^4H), 3.77 (1H, t, $J=0.02Hz$, C^3H), 4.39 (2H, d, $J=0.02 Hz$, $C^{19}H$), 3.78 (3H, s, $C^{17}H$), 3.77 (3H, s, $C^{11}H$).

^{13}C NMR ($CDCl_3$, 50 MHz, ppm) δ : 190.41 (C^{12}), 166.52 (C^6), 161.26 (C^{16}), 160.56 (C^{10}), 138.87 (C^{20}), 137.27 (C^{13}), 135.20 (C^2), 134.82 (C^4), 131.06 (C^{15}), 130.01 (C^9), 128.81 (C^{22}), 127.61 (C^{23}), 127.28 (C^{21}), 126.14 (C^7), 125.19 (C^1), 120.09 (C^5), 117.59 (C^3), 113.56 (C^{14}), 113.06 (C^8), 55.42 (C^{17}), 55.40 (C^{11}), 49.57 (C^{19}).

HRMS: Calcd. for $M^+.(C_{28}H_{25}NO_3)$ $m/z = 423.18344$, found 423.1821 (3 ppm).

Anal. Cald for $C_{28}H_{25}NO_3$: C: 79.41, H: 5.95, N: 3.31, O: 11.33 found C: 79.62, H: 5.95, N: 3.36.

Ligand N: 1-(4-methoxybenzoyl)-6-(1-(furan-2-methyl)amino)-6-(4methoxyphenyl) fulvene



$C_{26}H_{23}NO_4$
Mol. Wt.: 413,46512

Yellow solid
Yield : 36 %

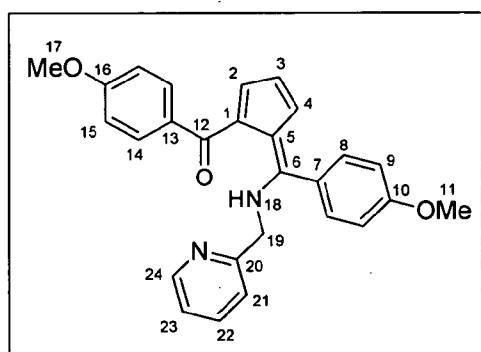
1H NMR ($CDCl_3$, 300 MHz, ppm) δ : 14.07 (1H, s, $N^{18}H$), 7.63 (2H, d, $J=0.04Hz$, $C^{14}H$), 7.29 (2H, d, $J=0.04Hz$, C^8H), 7.27 (1H, m, $C^{23}H$), 7.03 (1H, m, C^2H), 6.92 (2H, d, $J=0.04Hz$, $C^{15}H$), 6.83 (2H, d, $J=0.04Hz$, C^9H), 6.33 (1H, m, C^4H), 6.22 (1H, m, $C^{22}H$), 6.15 (1H, d, $J=0.01Hz$, $C^{21}H$), 6.12 (1H, t, $J=0.02Hz$, C^3H), 4.35 (2H, D, $J=0.02Hz$, $C^{19}H$), 3.80 (3H, s, $C^{17}H$), 3.78 (3H, s, $C^{11}H$).

^{13}C NMR (CDCl_3 , 50 MHz, ppm) δ : 190.41 (C^{12}), 166.52 (C^6), 161.26 (C^{16}), 160.56 (C^{10}), 142.64 (C^{23}), 139.15 (C^{20}), 137.27 (C^{13}), 135.35 (C^2), 134.82 (C^4), 131.07 (C^{15}), 130.17 (C^9), 126.14 (C^7), 125.19 (C^1), 120.09 (C^5), 117.74 (C^3), 113.60 (C^{14}), 113.02 (C^8), 110.51 (C^{22}), 107.96 (C^{21}), 55.40 (C^{17} and C^{11}), 42.84 (C^{19}).

HRMS: Calcd. for $\text{M}^+(\text{C}_{26}\text{H}_{23}\text{NO}_4)$ m/z = 413.16271, found 413.1614 (3 ppm)

Anal. Calcd for : C: 75.53, H: 5.61, N: 3.39, O: 15.48 found C: 74.96, H: 5.59, N: 3.36.

Ligand O: 1-(4-methoxybenzoyl)-6-(1-(pyridine-2-methyl)amino)-6-(4-methoxyphenyl) fulvene



$\text{C}_{27}\text{H}_{24}\text{N}_2\text{O}_3$
Mol. Wt.: 424.49106

Yellow solid
Yield : 60 %

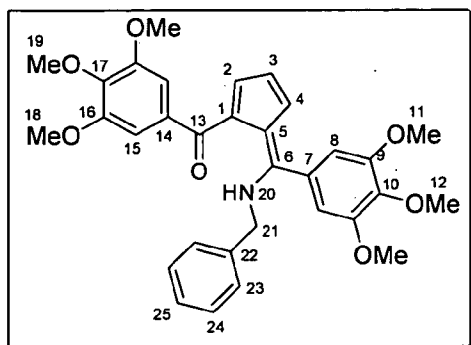
^1H NMR (CDCl_3 , 300 MHz, ppm) δ : 14.38 (1H, s, N^{18}H), 8.53 (1H, d, $J=0.01\text{Hz}$, C^{24}H), 7.80 (2H, d, $J=0.03\text{Hz}$, C^{14}H), 7.70 (1H, t, $J=0.02\text{Hz}$, C^{22}H), 7.42 (1H, d, $J=0.02\text{Hz}$, C^{21}H), 7.33 (2H, d, $J=0.02\text{Hz}$, C^8H), 7.22-7.17 (2H, m, C^2H and C^{23}H), 6.96 (4H, d, $J=0.03\text{Hz}$, C^{15}H and C^9H), 6.47 (1H, m, C^4H), 6.26 (1H, t, $J=0.02\text{Hz}$, C^3H), 4.67 (2H, d, $J=0.03\text{Hz}$, C^{19}H), 3.88 (3H, s, C^{17}H), 3.87 (3H, s, C^{11}H).

^{13}C NMR (CDCl_3 , 50 MHz, ppm) δ : 190.46 (C^{12}), 166.80 (C^6), 161.35 (C^{16}), 160.55 (C^{10}), 156.90 (C^{24}), 149.01 (C^{22}), 139.34 (C^{20}), 137.61 (C^{13}), 135.57 (C^2), 134.68 (C^4), 131.14 (C^{15}), 130.01 (C^9), 125.90 (C^7), 125.44 (C^1), 122.65 (C^{21}), 121.45 (C^{23}), 120.19 (C^5), 117.92 (C^3), 113.67 (C^{14}), 113.08 (C^8), 55.39 (C^{17} and C^{11}), 50.99 (C^{19}).

HRMS: Calcd. for $\text{M}^+(\text{C}_{27}\text{H}_{24}\text{N}_2\text{O}_3)$ m/z = 424.17869, found 424.1775 (2 ppm).

Anal. Calcd for $\text{C}_{27}\text{H}_{24}\text{N}_2\text{O}_3$: C: 76.39, H: 5.70, N: 6.60, O: 11.31 found C: 76.20, H: 5.73, N: 6.77.

Ligand P: 1-(3,4,5-trimethoxybenzoyl)-6-(benzylamino)-6-(3,4,5-trimethoxyphenyl) fulvene



$C_{32}H_{33}NO_7$
Mol. Wt.: 543,60692

Yellow solid
Yield : 54 %

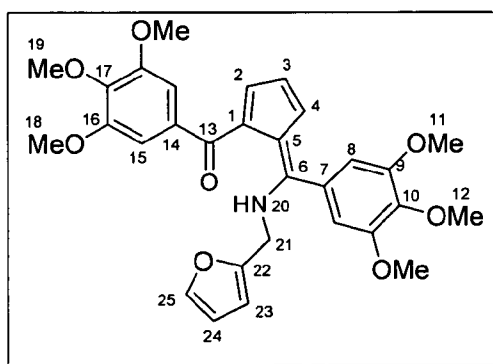
1H NMR ($CDCl_3$, 300 MHz, ppm) δ : 14.36 (1H, t, $J=0.02$ Hz, $N^{20}H$), 7.33-7.22 (6H, m, C^2H , $C^{23}H$, $C^{24}H$ and $C^{25}H$), 7.03 (2H, s, $C^{15}H$), 6.57 (1H, m, C^4H), 6.51 (2H, s, C^8H), 6.27 (1H, t, $J=0.01$ Hz, C^3H), 4.50 (2H, d, $J=0.02$ Hz, $C^{21}H$), 3.92 (12H, s, $C^{11}H$ and $C^{18}H$), 3.74 (6H, s, $C^{12}H$ and $C^{19}H$).

^{13}C NMR ($CDCl_3$, 50 MHz, ppm) δ : 190.40 (C^{13}), 166.52 (C^6), 152.95 (C^{16}), 152.60 (C^9), 139.70 (C^{17}), 139.39 (C^{10}), 138.70 (C^{14}), 137.58 (C^{22}), 137.53 (C^2), 135.64 (C^4), 128.96 (C^7), 128.80 (C^{24}), 127.64 (C^{23}), 124.78 (C^1), 119.46 (C^5), 118.17 (C^3), 106.42 (C^{15}), 105.63 (C^8), 61.05 (C^{19}), 60.94 (C^{12}), 56.26 (C^{18}), 56.12 (C^{11}), 49.56 (C^{21}).

HRMS: Calcd. for $M^+.(C_{32}H_{33}NO_7)$ m/z = 543.22570, found 543.2273 (2 ppm).

Anal. Cald for $C_{32}H_{33}NO_7$: C: 70.70, H: 6.12, N: 2.58, O: 20.60 found C: 70.93, H: 6.21, N: 2.61.

Ligand Q: 1-(3,4,5-trimethoxybenzoyl)-6-(1-(furan-2-methyl)amino)-6-(3,4,5-trimethoxyphenyl) fulvene



$C_{30}H_{31}NO_8$
Mol. Wt.: 533,56904

Yellow solid
Yield : 55 %

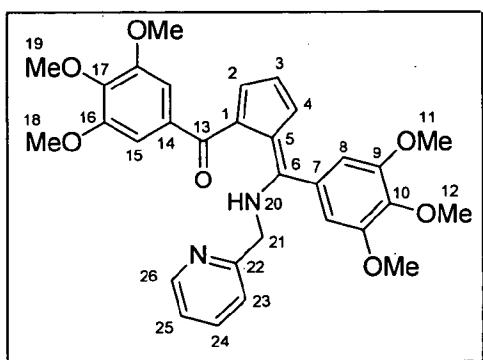
1H NMR ($CDCl_3$, 300 MHz, ppm) δ : 14.13 (1H, t, $J=0.02$ Hz, $N^{20}H$), 7.38 (1H, m, $C^{25}H$), 7.21 (1H, q, $J=0.01$ Hz, C^2H), 7.00 (1H, s, $C^{15}H$), 6.69 (1H, s, C^8H), 6.58 (1H, q, $J=0.01$ Hz, C^4H), 6.34 (1H, q, $J=0.01$ Hz, $C^{24}H$), 6.29-6.25 (2H, m, C^3H and $C^{23}H$), 4.47 (2H, d, $J=0.02$ Hz, $C^{21}H$), 3.96 (3H, s, $C^{19}H$), 3.92 (3H, s, $C^{12}H$), 3.91 (6H, s, $C^{18}H$), 3.87 (6H, s, $C^{11}H$).

^{13}C NMR (CDCl_3 , 50 MHz, ppm) δ : 190.37 (C^{13}), 166.03 (C^6), 153.08 (C^{16}), 152.57 (C^9), 149.95 (C^{25}), 142.65 (C^{22}), 139.70 (C^{17}), 139.63 (C^{10}), 138.82 (C^{14}), 137.44 (C^2), 135.86 (C^4), 128.82 (C^7), 124.92 (C^1), 119.39 (C^5), 118.31 (C^3), 110.61 (C^{22}), 108.12 (C^{15}), 105.85 (C^8), 61.06 (C^{19}), 60.93 (C^{12}), 56.27 (C^{18}), 56.25 (C^{11}), 42.92 (C^{21}).

HRMS: Calcd. for $\text{M}^+(\text{C}_{30}\text{H}_{31}\text{NO}_8)$ m/z = 533.20497, found 533.2035 (2 ppm).

Anal. Cald for $\text{C}_{30}\text{H}_{31}\text{NO}_8$: C: 67.53, H: 5.86, N: 2.63, O: 23.99 found C: 67.82, H: 5.99, N: 4.94.

Ligand R: 1-(3,4,5-trimethoxybenzoyl)-6-(1-(pyridine-2-methyl)amino)-6-(3,4,5-trimethoxyphenyl) fulvene



$\text{C}_{31}\text{H}_{32}\text{N}_2\text{O}_7$
Mol. Wt.: 544,59498

Yellow solid
Yield : 56 %

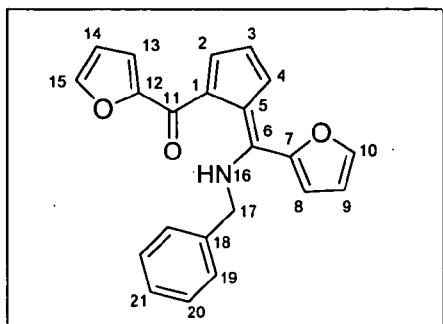
^1H NMR (CDCl_3 , 300 MHz, ppm) δ : 14.30 (1H, t, $J=0.02\text{Hz}$, N^{20}H), 8.55 (1H, d, $J=0.01\text{Hz}$, C^{26}H), 7.70 (1H, t, $J=0.02\text{Hz}$, C^{24}H), 7.38 (1H, d, $J=0.03\text{Hz}$, C^{23}H), 7.24-7.18 (2H, m, C^2H and C^{25}H), 7.02 (1H, s, C^{15}H), 6.62 (1H, s, C^8H), 6.58 (1H, q, $J=0.02\text{Hz}$, C^4H), 6.27 (1H, t, $J=0.01\text{Hz}$, C^3H), 4.65 (2H, d, $J=0.02\text{ Hz}$, C^{21}H), 3.92 (12H, s, C^{11}H and C^{18}H), 3.75 (6H, s, C^{12}H and C^{19}H).

^{13}C NMR (CDCl_3 , 50 MHz, ppm) δ : 190.38 (C^{13}), 166.68 (C^6), 157.03 (C^{26}), 152.98 (C^{16}), 152.56 (C^9), 149.60 (C^{25}), 139.62 (C^{17} and C^{10}), 138.82 (C^{14}), 137.46 (C^2), 137.01 (C^{22}), 135.94 (C^4), 128.89 (C^7), 124.96 (C^1), 122.58 (C^{25}), 121.36 (C^{23}), 119.50 (C^5), 118.31 (C^3), 106.48 (C^{15}), 105.80 (C^8), 61.01 (C^{19}), 60.93 (C^{12}), 56.24 (C^{18}), 56.15 (C^{11}), 51.27 (C^{21}).

HRMS: Calcd. for $\text{M}^+(\text{C}_{31}\text{H}_{32}\text{N}_2\text{O}_7)$ m/z = 544.22095, found 544.2181 (5 ppm).

Anal. Cald for $\text{C}_{31}\text{H}_{32}\text{N}_2\text{O}_7$: C: 68.37, H: 5.92, N: 5.14, O: 20.56 found C: 68.16, H: 6.02, N: 4.96.

Ligand S: 1-furanoyl-6-benzylamino-6-furanyl fulvene



$C_{22}H_{17}NO_3$
Mol. Wt.: 343,38

Red solid
Yield : 96 %

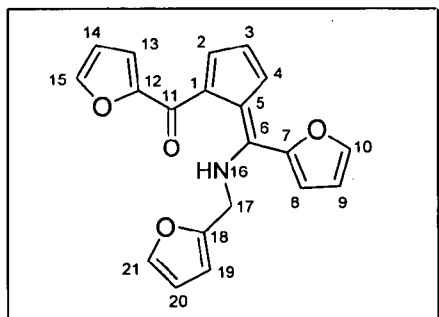
1H NMR ($CDCl_3$, 300 MHz, ppm) δ : 13.90 (1H, s, $N^{16}H$), 7.80 (1H, q, $J=0.01$ Hz, $C^{15}H$), 7.68-7.67 (2H, m, $C^{10}H$ and $C^{13}H$), 7.35-7.21 (5H, m, $C^{19}H$, $C^{20}H$ and $C^{21}H$), 7.20 (1H, m, C^2H), 6.92 (1H, q, $J=0.01$ Hz, $C^{14}H$), 6.76 (1H, d, $J=0.01$ Hz, C^8H), 6.59-6.57 (2H, m, C^4H and C^9H), 6.39 (1H, t, $J=0.01$ Hz, C^3H), 4.65 (2H, d, $J=0.02$ Hz, $C^{17}H$)

^{13}C NMR ($CDCl_3$, 50 MHz, ppm) δ : 176.37 (C^{11}), 154.33 (C^7), 153.95 (C^{12}), 146.13 (C^6), 145.32 (C^{15}), 144.65 (C^{10}), 137.64 (C^2), 137.07 (C^{18}), 134.91 (C^4), 128.73 (C^{20}), 127.69 (C^{21}), 127.44 (C^{19}), 124.71 (C^1), 119.88 (C^5), 118.83 (C^3), 117.55 (C^{13}), 117.34 (C^{14}), 11.60 (C^8), 111.45 (C^9), 50.03 (C^{17}).

HRMS: Calcd. for $M^+.(C_{22}H_{17}NO_3)$ m/z = 343.12084, found 343.1210 (0 ppm).

Anal. Cald for $C_{22}H_{17}NO_3$: C: 76.95, H: 4.99, N: 4.08, O:13.98 found C: 76.75, H: 5.15, N: 3.95.

Ligand T: 1-(furanoyl)-6-(1-(furan-2-methyl)amino)-6-(furanyl) fulvene



$C_{20}H_{15}NO_4$
Mol. Wt.: 333,34

Red oil
Yield : 97 %

1H NMR ($CDCl_3$, 300 MHz, ppm) δ : 13.76 (1H, s, $N^{16}H$), 7.79 (1H, q, $J=0.01$ Hz, $C^{15}H$), 7.70 (1H, m, $C^{10}H$), 7.66 (1H, m, $C^{13}H$), 7.36 (1H, m, $C^{19}H$), 7.19 (1H, d, $J=0.01$ Hz, C^2H), 6.91 (1H, q, $J=0.01$ Hz, $C^{14}H$), 6.84 (1H, d, $J=0.01$ Hz, $C^{21}H$), 6.62 (1H, q, $J=0.01$ Hz, $C^{20}H$), 6.56 (1H, q, $J=0.01$ Hz, C^4H), 6.38 (1H, t, $J=0.01$ Hz, C^3H), 6.31 (1H, m, C^8H), 6.27 (1H, m, C^9H), 4.62 (2H, d, $J=0.02$ Hz, $C^{17}H$)

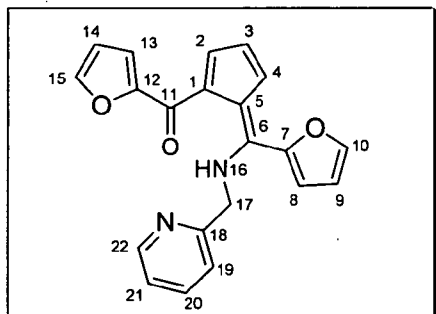
^{13}C NMR ($CDCl_3$, 50 MHz, ppm) δ : 176.36 (C^{11}), 153.93 (C^7), 153.85 (C^{12}), 149.92 (C^{21}), 146.01 (C^6), 145.32 (C^{15}), 144.80 (C^{10}), 142.61 (C^{18}), 137.94 (C^2), 134.98 (C^4),

124.84 (C¹), 119.83 (C⁵), 119.82 (C³), 117.59 (C¹³ and C¹⁴), 111.60 (C⁸), 111.52 (C⁹), 110.52 (C²⁰), 108.02 (C¹⁹), 43.07 (C¹⁷).

HRMS: Calcd. for M⁺.(C₂₀H₁₅NO₄) m/z= 333.10011, found 333.1008 (2 ppm).

Anal. Cald for C₂₀H₁₅NO₄: C: 72.06, H: 4.54, N: 4.20, O: 19.20, found C: 72.11, H: 4.55, N: 4.18.

Ligand U: 1-(furanoyl)-6-(1-(pyridine-2-methyl)amino)-6-(furanyl) fulvene



C₂₁H₁₆N₂O₃
Mol. Wt.: 344,36

Red solid
Yield : 60 %

¹H NMR (CDCl₃, 300 MHz, ppm) δ: 13.90 (1H, s, N¹⁶H), 8.54 (1H, d, J=0.02 Hz, C²²H), 7.80 (1H, q, J=0.01 Hz, C¹⁵H), 7.70-7.60 (3H, m, C¹⁰H, C¹³H and C²⁴H), 7.42 (1H, d, J=0.03 Hz, C¹⁹H), 7.21-7.17 (2H, m, C²¹H and C²H), 6.92 (1H, q, J=0.01 Hz, C¹⁴H), 6.79 (1H, d, J=0.01 Hz, C⁸H), 6.57-6.54 (2H, m, C⁴H and C⁹H), 6.38 (1H, t, J=0.01 Hz, C³H), 4.78 (2H, d, J=0.02 Hz, C¹⁷H).

¹³C NMR (CDCl₃, 50 MHz, ppm) δ: 176.40 (C¹¹), 157.15 (C²²), 154.59 (C⁷), 153.92 (C¹²), 149.35 (C²⁰), 145.94 (C⁶), 145.35 (C¹⁵), 144.92 (C¹⁰), 137.98 (C²), 137.01 (C¹⁸), 135.15 (C⁴), 124.88 (C¹), 122.45 (C¹⁹), 121.25 (C²¹), 119.87 (C⁵), 119.07 (C³), 117.63 (C¹³), 117.59 (C¹⁴), 111.61 (C⁸), 111.37 (C⁹), 51.77 (C¹⁷).

HRMS: Calcd. for M⁺.(C₂₁H₁₆N₂O₃) m/z= 344.11609, found 344.1155 (1 ppm).

Anal. Cald for C₂₁H₁₆N₂O₃: C: 73.24, H: 4.68, N: 8.13, O: 13.94, found C: 72.77, H: 4.73, N: 7.76.

Preparation of metallic complexes from carbonylaminofulvenes.

CrCl₃ / ligand A complex.

7.51 mg (20 μmol) of ligand A and 3.75 mg (10 μmol) of CrCl₃.3THF were introduced in a Schlenk with 100 μL of tetrahydrofuran (THF). The mixture was placed under

stirring for 2 h at room temperature. The solvent was evaporated under vacuum overnight to yield a yellow brown solid.

CrCl₂ / ligand A complex.

7.51 mg (20 μmol) of ligand A and 1.23 (10 μmol) of CrCl₂ were introduced in a Schlenk with 100 μL of THF. The mixture was placed under stirring for 2 h at room temperature. The solvent was evaporated under vacuum overnight to yield a yellow brown solid.

CrCl₃ / ligand D complex.

7.27 mg (20 μmol) of ligand D and 3.75 mg (10 μmol) of CrCl₃.3THF were introduced in a Schlenk with 100 μL of tetrahydrofuran (THF). The mixture was placed under stirring for 2 h at room temperature. The solvent was evaporated under vacuum overnight to yield a yellow brown solid.

CrCl₃ / ligand H complex.

9.31 mg (20 μmol) of ligand H and 3.75 mg (10 μmol) of CrCl₃.3THF were introduced in a Schlenk with 100 μL of tetrahydrofuran (THF). The mixture was placed under stirring for 2 h at room temperature. The solvent was evaporated under vacuum overnight to yield a yellow brown solid.

CrCl₃ / ligand E complex.

7.07 mg (20 μmol) of ligand E and 3.75 mg (10 μmol) of CrCl₃.3THF were introduced in a Schlenk with 100 μL of tetrahydrofuran (THF). The mixture was placed under stirring for 2 h at room temperature. The solvent was evaporated under vacuum overnight to yield a red brown solid.

CrCl₃ / ligand C complex.

7.53 mg (20 μmol) of ligand C and 3.75 mg (10 μmol) of CrCl₃.3THF were introduced in a Schlenk with 100 μL of tetrahydrofuran (THF). The mixture was placed under stirring for 2 h at room temperature. The solvent was evaporated under vacuum overnight to yield a yellow brown solid.

CrCl₃ / ligand B complex.

7.31 mg (20 μmol) of ligand B and 3.75 mg (10 μmol) of $\text{CrCl}_3 \cdot 3\text{THF}$ were introduced in a Schlenk with 100 μL of tetrahydrofuran (THF). The mixture was placed under stirring for 2 h at room temperature. The solvent was evaporated under vacuum overnight to yield a yellow brown solid.

CrCl_3 / ligand O complex.

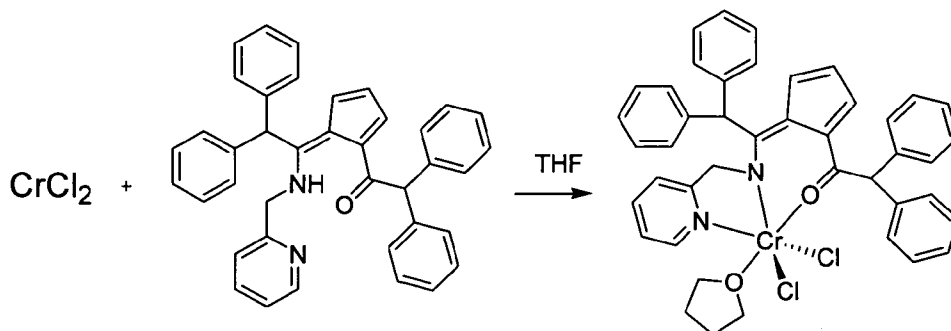
8.49 mg (20 μmol) of ligand O and 3.75 mg (10 μmol) of $\text{CrCl}_3 \cdot 3\text{THF}$ were introduced in a Schlenk with 100 μL of tetrahydrofuran (THF). The mixture was placed under stirring for 2 h at room temperature. The solvent was evaporated under vacuum overnight to yield a yellow brown solid.

CrCl_3 / ligand L complex.

10.89 mg (20 μmol) of ligand L and 3.75 mg (10 μmol) of $\text{CrCl}_3 \cdot 3\text{THF}$ were introduced in a Schlenk with 100 μL of tetrahydrofuran (THF). The mixture was placed under stirring for 2 h at room temperature. The solvent was evaporated under vacuum overnight to yield a yellow brown solid.

CrCl_2 / ligand L complex.

10.89 mg (20 μmol) of ligand L and 1.23 (10 μmol) of CrCl_2 were introduced in a Schlenk with 100 μL of THF. The mixture was placed under stirring for 2 h at room temperature. The solvent was evaporated under vacuum overnight to yield a yellow brown solid.



The complex was recrystallised by slow diffusion of pentane in a saturated solution of the complex in THF. The crystals obtained were suitable for X-Ray analysis.

The complex crystallises in a monoclinic environment with space group P 21/c. The chromium atom is coordinated by one molecule of tridentate fulvene by its oxygen atom and its two nitrogen atoms. The chromium atom is further coordinated by two chlorine atom and a THF molecule. This can be seen in Figure 3. The complex is characterised as follows.

Empirical formula	C ₄₃ H ₃₉ Cl ₂ Cr N ₂ O ₂
Formula weight	738.66
Temperature	100(2) K
Wavelength	0.71073 Å
Crystal system, space group	Monoclinic, P 21/c
Unit cell dimensions	a = 14.987(2) Å alpha = 90 deg. b = 14.175(2) Å beta = 99.485(9) deg. c = 17.412(3) Å gamma = 90 deg.
Volume	3648.4(9) Å ³
Z, Calculated density	4, 1.345 Mg/m ³
Absorption coefficient	0.500 mm ⁻¹
F(000)	1540
Crystal size	0.2 x 0.2 x 0.03 mm
Theta range for data collection	2.92 to 27.48 deg.
Limiting indices	-19 ≤ h ≤ 19, -18 ≤ k ≤ 11, -22 ≤ l ≤ 22
Reflections collected / unique	38703 / 8296 [R(int) = 0.0659]
Completeness to theta =	27.48 99.2 %
Absorption correction	Semi-empirical from equivalents
Max. and min. transmission	0.985 and 0.885
Refinement method	Full-matrix least-squares on F ²
Data / restraints / parameters	8296 / 0 / 451
Goodness-of-fit on F ²	1.030
Final R indices [I > 2σ(I)]	R1 = 0.0423, wR2 = 0.0840
R indices (all data)	R1 = 0.0728, wR2 = 0.0941
Largest diff. peak and hole	0.331 and -0.492 e.Å ⁻³

Homogeneous polymerisation of ethylene.

The metallic catalyst component were activated with 1.625 mL of methylaluminoxane (MAO). The solution was stirred for 5 minutes and then diluted with 3.375 mL of toluene. The reactor was dried under nitrogen at a temperature of 90 °C for a period of time of 30 minutes. The reactor was brought to a polymerisation temperature of 35 °C and 50 mL of toluene were added to the reactor under nitrogen. A scavenger solution consisting of 0.5 mL of MAO (30%) and 4.5 mL of toluene was added to the reactor and the solution was stirred for a few minutes. The solution of activated catalyst was added to the reactor under nitrogen. The flux of nitrogen was interrupted, the reactor was purged and placed under an ethylene pressure of 15 bars. It was placed under stirring for a period of time of 1 h. The reactor was purged and the polymerisation was stopped by adding a 10% solution of MeOH/HCl. The polymer was washed 3 times with 30 mL of MeOH and 3 times with 30 mL of acetone. The polymer was dried under vacuum overnight at room temperature. The results are summarised in Table IV for the chromium-based catalyst systems.

TABLE IV.

<i>M/L</i>	<i>m PE (g)</i>	<i>Activity (kg_{PE}/(mol.h))</i>	<i>Mn</i>	<i>Tm (°C)</i>
CrCl ₃ /A	0,941	94,1	-	137
CrCl ₂ /A	0,976	97,6	insoluble	137
CrCl ₃ /D	0,995	99,5	unfilterable	119
CrCl ₃ /H	1,045	104,5	unfilterable	135
CrCl ₃ /E	1,244	124,4	insoluble	129
CrCl ₃ /C	1,283	128,3	unfilterable	135
CrCl ₃ /B	2,895	289,5	unfilterable	133
CrCl ₃ /O	3,648	364,8	insoluble	136
CrCl ₃ /L	5,415	541,5	insoluble	135
CrCl ₂ /L	6,915	691,5	insoluble	136

For all polymerisations, the conditions were as follows: Cr 10 μmol , ligand 20 μmol , polymerisation temperature 35 °C, ethylene pressure 15 bars, 1000 eq. MAO, solvent: toluene, polymerisation time 1h.

The highest activities were obtained with the catalyst systems based on CrCl_3 / ligand L and CrCl_2 / ligand L. As depicted in Table V, CrCl_2 based system is a selective catalyst whereas CrCl_3 based system is a mixed polymer/oligomer catalyst.

TABLE V

<i>M/L</i>	<i>m PE (g)</i>	<i>Activity (kg_{PE}/(mol.h))</i>	<i>Consumption (kg_{C₂H₄}/(mol.h))</i>
CrCl_2/L	4.11	411	385
CrCl_3/L	3.38	338	590

For all polymerisations, the conditions were as follows: Cr 10 μmol , ligand 20 μmol , polymerisation temperature 35 °C, ethylene pressure 15 bars, 1000 eq. MAO, solvent: toluene, polymerisation time 1h.

The activity of the system CrCl_2 / ligand L has been studied as a function of temperature and of ethylene pressure. The results are displayed in Table VI. It can be concluded that the activity of the catalyst system increases with increasing pressure and decreases when the temperature is raised above 35 °C. It can also be concluded that the activity increases with a lower amount of catalyst.

TABLE VI

<i>kg_{PE}/(mol.h)</i>	<i>5 $\mu\text{mol Cr}$</i>		<i>2.5 $\mu\text{mol Cr}$</i>
	<i>15 bars</i>	<i>45 bars</i>	<i>45 bars</i>
25 °C	773	1 722	3 606
35 °C	1 016	1 341	2 790
55 °C	619	806	1 196

For all polymerisations, the conditions were as follows: 1000 eq. MAO, solvent: toluene, polymerisation time 1h.

The consumptions were also measured and the results are displayed in Table VII.

TABLE VII

<i>Temp.</i> (°C)	<i>m PE</i> (g)	<i>Activity</i> (kg _{PE})/(mol.h)	<i>Consumption</i> (kg _{C₂H₄})/(mol.h)
35	7.49	2 996	2 729
25	10.77	4 308	5 738

For all polymerisations, the conditions were as follows: Cr 2.5 µmol, ligand L 5 µmol, ethylene pressure 45 bars, 1000 eq. MAO, solvent: toluène, polymerisation time 1h.

Polymerisation of ethylene with supported catalyst systems.

The activity of the unsupported CrCl₂ / ligand L catalyst system was evaluated in heptane. The catalyst system was not selective in polyethylene as the consumption of ethylene was larger than the amount of ethylene present in the polyethylene. The polymerisation conditions were as follows:

complexation time: 2 hours,

5 µmol of ligand with 2.5 µmol of Cr,

polymerisation temperature: 25 °C,

polymerisation pressure: 45 bars,

1000 equ. of MAO,

solvent: heptane,

polymerisation time: 1 hour.

The results are displayed in Table VIII.

TABLE VIII

<i>m PE (g)</i>	<i>Activity</i> (kg _{PE})/(mol.h)	<i>Consumption</i> (kg _{C₂H₄})/(mol.h)	<i>Tm (°C)</i>
6,77	2 708	6 768	137

Impregnation of the catalyst on silica/MAO.

5 μmol of complex CrCl_2/L were dissolved in 600 μl of toluene and then introduced in a schlenk with 100 mg of silica/MAO (50 $\mu\text{mol}_{\text{Cr}}/\text{g}_{\text{Si}}$) under stirring for a period of time of 30 minutes. The impregnated silica was filtered and washed either once with 600 μl of toluene and three times with 600 μl of heptane (condition 1) or three times with 600 μl of heptane (condition 2).

Polymerisation of ethylene with impregnated silica/MAO.

The reactor was dried under nitrogen for a period of time of 30 minutes and at a temperature of 90 °C. 50 mL of heptane were then introduced into the reactor with 100 mL of scavenger, MAO (30 %) diluted in 5 mL of heptane, at a temperature of 25 °C. 50 mg of silica, containing about 2.5 μmol of activated catalyst (50 $\mu\text{mol}_{\text{Cr}}/\text{g}_{\text{SiO}_2}$) were introduced into the reactor with 5 mL of heptane. The polymerisation reaction was carried out at a temperature of 25 °C under an ethylene pressure of 45 bars and for a period of time of 1 hour for conditions 1 and 2. The results are displayed in Table IX.

TABLE IX

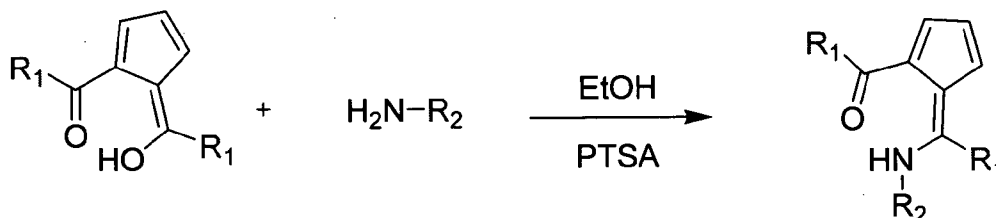
	<i>m PE</i> (g)	<i>Activity</i> ($\text{kg}_{\text{PE}}/(\text{mol.h})$)	<i>Consumption</i> ($\text{kg}_{\text{C}_2\text{H}_4}/(\text{mol.h})$)	<i>Activity</i> ($\text{g}_{\text{PE}}/(\text{g}_{\text{Si}}.\text{h})$)	<i>Consumption</i> ($\text{g}_{\text{C}_2\text{H}_4}/(\text{g}_{\text{Si}}.\text{h})$)	<i>Tm</i> (°C)
Cond. 1	~ 0	0	4 836	0	242	-
Cond. 2	0,77	308	5 061	15,4	253	138

Polymerisation of alpha-olefins.

The unsupported catalyst system CrCl_2/L was used for the polymerisation of hexene with the following conditions: $\text{CrCl}_2/\text{L}/\text{MAO}/\text{hexene} = 1/100/2000$. After a period of time of 24 hours and a polymerisation temperature of 30 °C the yield was of about 4.5 %.

CLAIMS.

1. A method for preparing a metallic complex that comprises the steps of:
 - a) preparing a carbonylamino fulvene ligand by condensation reaction of an hydroxycarbonyl fulvene ligand with a primary amine;



wherein R₁ and R₂ are the same or different and are selected from alkyl, aryl, alkylaryl, arylalkyl having at most 20 carbon atoms or heteroatoms-containing groups;

- b) providing a metallic precursor MZ_n wherein M is a metal Group 6 to 11 of the Periodic Table, Z is a negative counter-anion and n is the valence of M;
 - c) complexing the metallic precursor of step b) with the carbonylamino fulvene of step a);
 - d) retrieving a metallic complex.
2. The method of claim 1 wherein R₁ is alkyl, substituted or unsubstituted phenyl group, CHPh₂ wherein phenyl group Ph is substituted or unsubstituted or heteroatom-containing group.
3. The method of claim 2 wherein R₁ is CHPh₂ or paramethoxyphenyl or cyclohexyl.
4. The method of any one of claims 1 to 3 wherein R₂ is CH₂pyridine or furan.
5. The method of any one of the preceding claims wherein R₁ is CHPh₂ and R₂ is CH₂pyridine, or wherein R₁ is paramethoxyphenyl and R₂ is CH₂pyridine, or wherein R₁ is cyclohexyl and R₂ is furan.

6. The method of any one of the preceding claims wherein M is CrII, CrIII or Ni.
7. The method of any one of the preceding claims wherein Z is halogen or acetate.
8. A metallic complex obtainable by the method of any one of claims 1 to 7.
9. An active catalyst comprising the metallic complex of claim 8, an activating agent having an ionising action and optionally a support.
10. The active catalyst system of claim 9 wherein the activating agent is methylaluminumoxane.
11. A method for preparing an active catalyst system that comprises the steps of:
 - a) providing a carbonylamino fulvene ligand;
 - b) complexing the ligand of step a) with a metallic salt MZ_n in a solvent;
 - c) retrieving a catalyst component;
 - d) optionally depositing the catalyst component of step c) on a support;
 - e) activating the catalyst component of step c) or of step d) with an activating agent having an ionising action;
 - f) optionally adding a scavenger;
 - g) retrieving an active oligomerisation or polymerisation catalyst system.
12. A method for oligomerising or for homo- or co-polymerising ethylene and alpha-olefins that comprises the steps of:
 - a) injecting the active catalyst system of claim 9 or claim 10 into the reactor;
 - b) injecting the monomer and optional comonomer;
 - c) maintaining under polymerisation conditions;
 - d) retrieving the oligomers and/or polymer.

13. The method of claim 12 wherein the monomer and comonomer are selected from ethylene, propylene or 1-hexene.

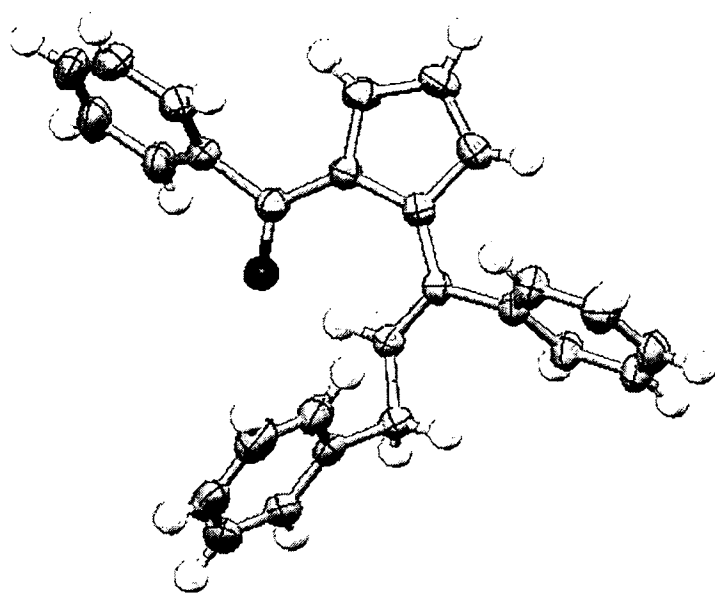


FIGURE 1

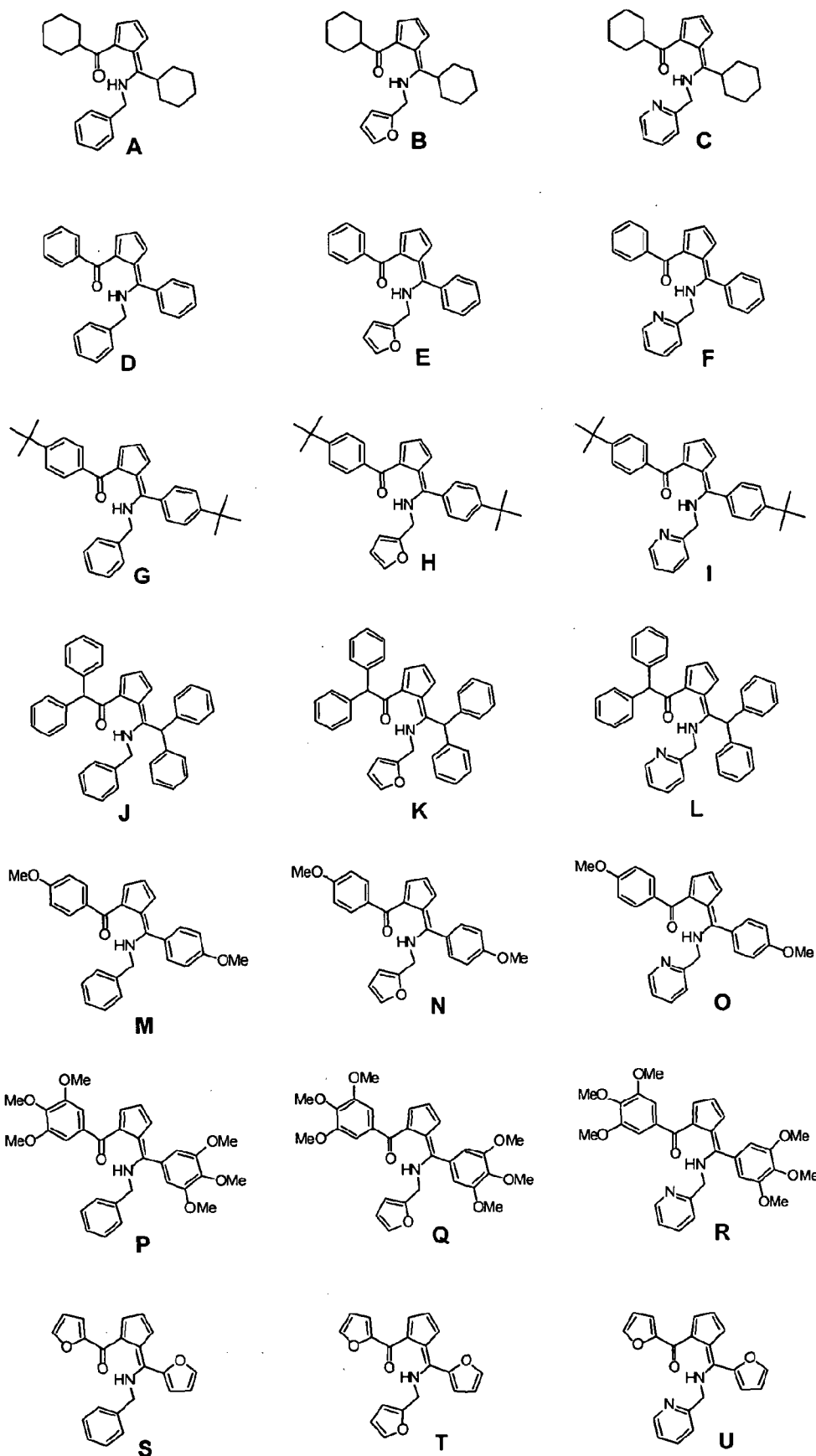


FIGURE 2

SUBSTITUTE SHEET (RULE 26)

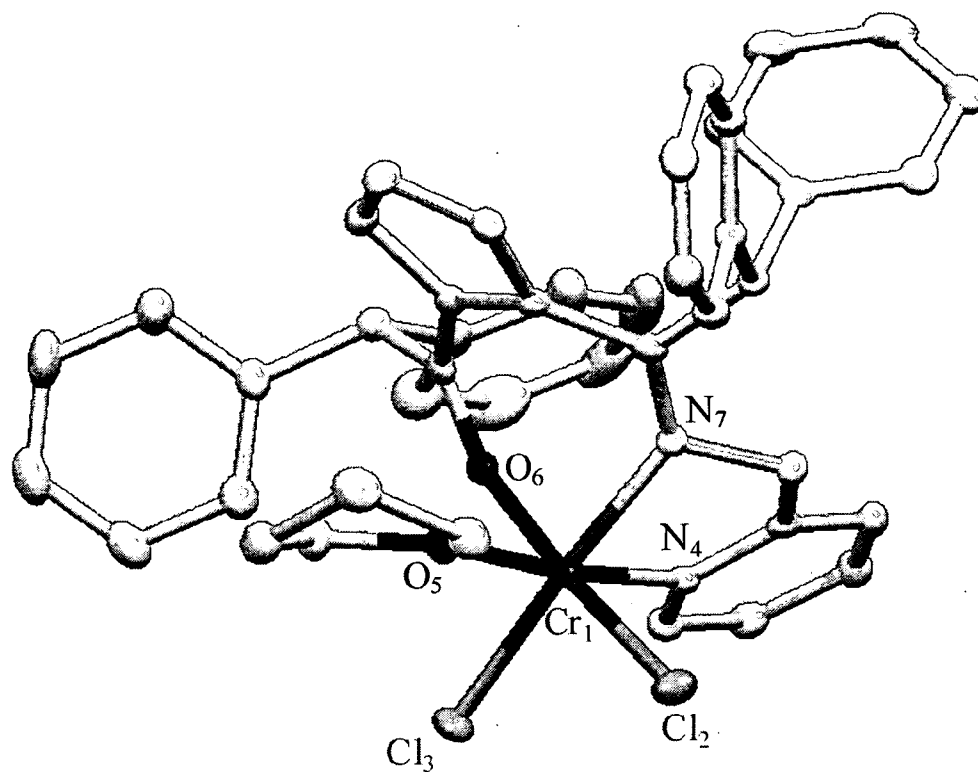


FIGURE 3