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# Shiba et al.


[54]		HALIDE SUPERSE GRAPHIC EMULSI	
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[22]	Filed:	June 25, 1970	$(x_{i,k}, x_{i,k}) = (x_{i,k}, x_{i,k})$
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[56]		References Cited	
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2,533		50 Carroll	96/124
2,541			96/120
2,701			96/124
3,038	,800 6/19		96/139

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# [57]

# **ABSTRACT**

A silver halide photographic emulsion is super-sensitized in the green wave length region by adding thereto at least one sensitizing dye of the following formula I and at least one sensitizing dye of the following formula II; and, optionally a compound of the following formula III:

# FORMULA I

$$\begin{array}{c|c} R_1 & R_2 \\ N & N \\ Z_1 & N \\ R_2 & N \\ \end{array}$$

#### $(X_1^-)_{p-1}$ FORMULA II

# FORMULA III

The substituents are described in detail hereinbelow.

22 Claims, No Drawings

# SILVER HALIDE SUPERSENSITIZED PHOTOGRAPHIC EMULSION

#### **BACKGROUND OF THE INVENTION**

#### 1. Field of the Invention

This invention relates to a super-sensitized silver halide photographic emulsion and more particularly it relates to a supersensitized silver halide photographic emulsion for increasing particularly the spectral sensitivity in the green region. Furthermore, the invention relates to a super-sensitized silver halide photographic emulsion containing couplers for making color photographic light-sensitive materials.

# 2. Description of the Prior Art

It is well known that a spectral sensitizing method is one of the techniques employed in the manufacture of silver halide photographic emulsions, and that the sensitizing method is an indispensible technique for the production of color photographic light-sensitive materials. The spectral sensitivity obtained by the method is influenced not only by the chemical structure of the sensitizing dye to be used and the properties of the silver halide emulsion to be used, such as the composition of the halogen and the crystal system of the silver halide contained in the emulsion, the manner of chemical ripening, the silver ion concentration and the hydrogen ion concentration of the emulsion, but also by the various kinds of additives added to the silver halide emulsion.

In many cases, the spectral sensitivity is reduced. However, when a certain kind of sensitizing dye is used together with a dye or a colorless aromatic compound having a specific chemical structure according to the chemical structure and the properties of the sensitizing 35 dye, a higher spectral sensitivity is obtained than when using each of the sensitizing dyes alone. This is known as a super-sensitizing action.

In particular, silver halide emulsions used for producing color photographic light-sensitive materials 40 contain a large amount of couplers, and when a coupler is incorporated in a silver halide photographic emulsion containing a sensitizing dye, the spectral sensitivity becomes lower than the spectral sensitivity obtained by using the sensitizing dye alone. In addition, the 45 presence of the coupler reduces the super-sensitizing action.

Also, the spectral sensitivity distribution in the green region, in which man's visual sensitivity is high, is one of the most important factors for the color reproduction of color photographic light-sensitive materials and for the quality of the image obtained.

Therefore, various studies have been made concerning the sensitization of silver halide emulsions in the green region. For example, a method of using a combination of a benzimidazolocarbocyanine dye and a pseudocyanine dye is described in the specification of U. S. Pat. No. 2,701,198. However, when the method described in the aforementioned U. S. patent is applied to a silver halide emulsion containing a coupler, only a low sensitivity is obtained due to the presence of the coupler in the emulsion.

Also, a method of using a combination of a benzimidazolocarbocyanine dye and a benzoxazolocarbocyanine dye is described in the specification of Japanese Pat. Publication No. 4936/68. However, this method has the disadvantages that the sensitivity in the

shorter wave length side of the green region, e.g., in a wave length region of  $500-530~\text{m}\mu$  is large, and also the contamination effect by the dyes after development is large.

Therefore, a primary object of the present invention is to provide a silver halide photographic emulsion having both a high spectral sensitivity in a wave length region of  $500-530 \text{ m}\mu$  and less of a contamination by the dyes after development.

Another object of the present invention is to provide a silver halide photographic emulsion used for producing a color photographic light-sensitive material having a high green sensitivity in the presence of a coupler.

#### **SUMMARY OF THE INVENTION**

The above objects of the present invention can be attained by super-sensitizing a silver halide photographic emulsion with at least one sensitizing dye represented by the following formula I:

FORMULA I

$$Z_1 + C - CH = CH - CH = C$$

$$R_2$$

$$N$$

$$N$$

$$R_3$$

$$(X_1^*)_{p-1}$$

$$R_4$$

 $_{30}$  wherein  $R_1$  and  $R_2$  (which may be same or different) each represent an alkyl group, an aryl group or a substituted alkyl group; R<sub>3</sub> and R<sub>4</sub> (which may be same or different) each represent an alkyl group, an aryl group or a substituted alkyl group (including a substituted alkyl group having a sulfo group); at least one of said R<sub>3</sub> and R4 being a substituted alkyl group having a sulfo group; Z<sub>1</sub> and Z<sub>2</sub> (which may be same or different) each represents a non-metallic atomic group necessary complete a heterocyclic nucleus of the benzimidazole series; X<sub>1</sub>- represents an acid anion group usually employed in cyanine dyes; and p is an integer of 1 or 2, p being 1 when the cyanine dye of formula I forms an intermolecular salt; and at least one of the sensitizing dyes represented by the following formula II:

FORMULA II

$$Z_3$$
 + -CH=C +  $Z_4$   $R_5(X_2^*)_{q-1}$   $R_6$ 

wherein  $Z_3$  represents a non-metallic atomic group necessary to complete a heterocyclic nucleus of the 2-quinoline series;  $Z_4$  represents a non-metallic atomic group necessary to complete a benzoxazole nucleus, a naphthoxazole nucleus, a benzthiazole nucleus, a benz-selenazole nucleus, a naphthothiazole nucleus, or a naphthoselenazole nucleus, each being unsubstituted or substituted by an alkyl group, a phenyl group, a halogen atom, an alkoxyl group or an alkoxyl group having a sulfo group;  $R_5$  and  $R_6$  (which may be same or different) each represents an alkyl group or a substituted alkyl group (including a substituted alkyl group having a sulfo group); at least one of said  $R_5$  and  $R_6$  being a substituted alkyl group having a sulfo group ex-

cept where the heterocyclic nucleus containing Z<sub>4</sub> has been substituted by a substituted alkoxyl group having a sulfo group; X2- represents an acid anion group usually employed in cyanine dyes; and q is an integer of 1 or 2, q being 1 when the cyanine dye of formula II 5 forms an intermolecular salt.

Furthermore, it has been discovered according to the present invention that a particularly better result is obtained by incorporating a compound represented by the following formula III in the above-mentioned silver  $^{10}$ halide photographic emulsion:

wherein R<sub>7</sub> represents an alkyl group or a substituted alkyl group; Z<sub>5</sub> represents a non-metallic atomic group 20 necessary to complete a heterocyclic nucleus of the benzimidazole series; and  $Z_{\mbox{\tiny 6}}$  represents a phenyl group or a substituted phenyl group, such as a phenyl group substituted by alkyl groups, alkoxy groups, and the like.

## DETAILED DESCRIPTION OF THE PREFERRED **EMBODIMENTS**

In the above-mentioned formulas I and II, the alkyl groups of R<sub>1</sub> to R<sub>6</sub> include: methyl, ethyl, n-propyl, isopropyl, n-butyl and iso-butyl groups; and as the substituted alkyl groups of R<sub>1</sub> to R<sub>6</sub> there may be illustrated: an allyl group, an aralkyl group, a  $\beta$ -acetoxyethyl group, a γ-acetoxypropyl group, a γ-carboxypropyl group, a  $\beta$ -carboxyethyl group, a  $\gamma$ -sulfatopropyl group, a  $\delta$ -Nacetylsulfamyl butyl group and a  $\beta$ -hydroxyethyl group. In particular, as the substituted alkyl group having a sulfo group, there may be illustrated a  $\beta$ -sulfopropyl (3-sulfopropoxy)ethyl group, a 2-[2-(3-sulfopropoxy)ethoxy]ethyl group, and a 2-hydroxy-1-sulfopropyl group.

As the benzimidazole series nucleus completed by  $Z_1$ ,  $Z_2$  or  $Z_5$ , there may be illustrated: benzimidazole 45 and benzimidazole containing, as a component of the skeletal-structure thereof, a benzene ring substituted by at least one of the following: a halogen atom, an Nunsubstituted or alkyl substituted sulfamyl or carbamyl group, an N-disubstituted sulfamyl group such as a 50 morpholino sulfonyl group, an alkyl sulfonyl group, a trifluoromethyl group, an unsubstituted or substituted alkyl group, a cyano group, a phenyl group, a carboalkoxyl group, a hydroxyl group, and an alkoxyl group.

Also, the 2-quinoline series nucleus completed by Z<sub>3</sub>

in the above-mentioned formula II may be, for example, 2-quinoline and 2-quinoline containing, as a component of the skeletal-structure thereof, a benzene ring substituted by at least one of the following: an alkyl group and a halogen atom.

The substituted alkyl group of R<sub>7</sub> in the above-mentioned formula III is, for example, a sulfoalkyl group neutralized by a cation, such as alkali metal ions, organic ammonium ions, a pyridinium ion, a benzylthiouronium ion, and the like, usually employed in cyanine dyes, a cyanoalkyl group and an allyl group.

As X<sub>1</sub> and X<sub>2</sub> in the formulas I and II, there may be illustrated, e.g., a halogen, perchlorate, thiocyanate, ptoluene sulfonate, benzene sulfonate, methyl sulfate and ethyl sulfate ions.

The sensitizing dye represented by formula II has at least one sulfo group in the heterocyclic nucleus containing Z<sub>4</sub> or at R<sub>5</sub> or R<sub>6</sub>. The spectral sensitization obtained by using the sensitizing dye represented by the formula II is ordinarily reduced by the introduction of a sulfo group, but when the sensitizing dye represented by the formula I is used together therewith, the spectral sensitization is increased.

The sensitizing dye represented by the formula II may be substituted therein by an alkyl group, a phenyl group, a halogen atom, an alkoxyl group, or an alkoxyl group substituted by at least a sulfo group; in other words, the substituent is so selected that the maximum sensitivity obtained by using at least one sensitizing dye of formula II and at least one compound of formula I in combination does not shift to a longer wave length side than that of the sensitizing dye of formula II.

The compound represented by formula III shows a 35 super-sensitizing action to the sensitizing dye represented by formula I and to the sensitizing dye represented by formula II, and by using the compound represented by formula III together with the combination of the other two sensitizing dyes, the green sengroup, a β-sulfobutyl group, a δ-sulfobutyl group, a 2- 40 sitivity is further increased and at the same time the sensitivity in the wave length region of 500-530 MA, which is present at a shorter wave length side of the green region, is further increased.

> The compounds which may be used in the present invention will be further illustrated by reference to the following illustrative formulas, which are merely illustrative, and not limiting, in nature.

The above-mentioned sensitizing dyes or the other compounds which may be used in this invention may be synthesized by known methods.

(CH2)3SO3Na

(IIIC)

For instance, they can be synthesized by the methods described in the specifications of U. S. Pat. Nos. 50 2,503,776 and 2,778,823; Belgian Pat. Nos. 590,607; 704,296; 668,014; 669,133; 669,934; 673,554; and 697,006; British Pat. Nos. 742,112; 955,964; 975,504; and 980,234; Japanese Pat. Publication Nos. 14112/65; 23467/65; and 13823/68; and Japanese Pat. Application Nos. 55439/67 and 76071/67.

Also, other sensitizing dyes and the compounds used in this invention, which were not described above, may also be easily synthesized by referring to the specifications of the above patents.

The compounds used in this invention are incorporated in a sliver halide emulsion as solutions in water or a water-soluble organic solvent such as methanol, ethanol, and pyridine. The sensitizing dyes represented by formula I and formula II, respectively, may be added to a silver halide emulsion as a mixture of the solutions

thereof or as separate solutions. The amounts of these dyes are greatly influenced by the nature of the silver halide emulsion to be used. Usually, the total amount of

the above-mentioned sensitizing dyes (I and II) and compound III are preferably from  $1\times 10^{-6}$  mol to  $1\times 10^{-3}$  mol, per mol of silver halide. The weight ratio of the amount of the sensitizing dye represented by formula II to the amount of the sensitizing dye represented 5 by formula I is preferably from 10:1 to 1:2, but this ratio is varied by the amount of the compound represented by formula III to be added.

As the silver halide emulsion which may be used in this invention, there may be illustrated: a silver 10 iodobromide emulsion, a silver bromide emulsion, a silver chlorobromide emulsion, etc. The silver halide emulsion used in this invention is mainly a gelatino silver halide emulsion but the emulsion may contain, besides gelatin, polyvinyl alcohol, an alginic acid 15 polymer, polyvinyl imidazole, polyvinyl pyrrolidone, or copolymers thereof, or may be an emulsion thereof.

The silver halide emulsion of this invention is applied to a suitable support according to the desired use, such as a paper, a glass plate, a cellulose triacetate film, a polyethylene terephthalate film, other plastic films than above, a baryta-coated paper, a resin-coated paper, or a synthetic paper.

The present invention will be illustrated further by 25 the following examples which are merely illustrative and not limited in nature.

#### **EXAMPLE 1**

50 g of 1-phenyl-3-[3-(2,4-di-tertiaryamylphenox-yactamido)-benzamido]5-pyrazolone was dissolved in 100 ml of dibutyl phthalate by heating and the resultant solution was added to 1 liter of a 10 percent aqueous gelatin solution. Then, 50 ml of a 5 percent aqueous solution of sodium alkylbenzene sulfonate was added thereto and they were dispersed by emulsification by means of a high speed rotary mixer (the product is called the dispersion of the magenta coupler).

1 kg. of a silver iodobromide emulsion (containing 40 0.25 mol of silver and 4.0 mol percent of iodine) prepared by a conventional method was placed in a beaker and melted at 40°C. After adding to the emulsion the sensitizing dye in the amounts shown in Table 1, the emulsion was stirred for 15 minutes. Further, 400 45 g. of the dispersion of the magenta coupler as prepared above was added to the emulsion with stirring and after adding further suitable amounts of a hardening agent and an ampholite surface active agent as a wetting agent to the mixture with stirring, the resultant mixture was applied to a cellulose triacetate film in a thickness of 7 microns to obtain a sample of a green-sensitive light-sensitive material.

The sample was cut into strips and subjected to an 55 optical wedge exposure through Yellow Filter No. K-12 made by Fuji Photo Film Co. by using a sensitometer of 5400°K in color temperature of the light source. The strip thus exposed was developed for 12 minutes at 20°C in a color developer having the following composition:

N,N-diethyl-p-aminoaniline sulfate Sodium sulfite	2.0 g	
	2.0 g	
Sodium carbonate monohydrate	50.0 g	65
Hydroxylamine hydrochloride	1.5 g	
Potassium bromide	1.0 g	
Water to make	1000 ml	
$(pH, 10.8 \pm 0.1)$		

Thereafter, the strip was further processed in a first fixing solution, a bleaching solution and a second fixing solution and washed with water to obtain a magenta image. After drying the sample strips thus processed, the green filter density was measured by using an Stype Densitometer made by Fuji Photo Film Co., from which characteristic curves were obtained. From these results, the densities at the point of fog plus 0.20 were calculated and the results are shown in Table 1.

The compositions of the bleaching solution and the fixing solution used in the above processing were as follows:

#### Composition of the Bleaching Solution

Ferricyanide	100 g
Potassium bromide	20 g
Water to make	1000 ml
$(pH, 6.9 \pm 0.3)$	

#### Composition of the Fixing Solution

Нуро	200 g
Sodium sulfite	20 g
Acetic acid (28%)	45 ml
Boric acid	7.5 g
Potassium alum	20 g
Water to make	1000 ml
$(pH, 4.5 \pm 0.2)$	

#### TABLE 1

0	Dye	Amount (ml) (mol concent- ration)	Dye Amount (ml) (mol concent- ration)	(A)	(B)
	(IA)	40(5×10 <sup>-4</sup> )	_	115	0.17
		80	_	120	0.23
5		40	(IIC) 20(1×10 <sup>-3</sup> )	142	0.18
-		**	40	165	0.20
		44	120	158	0.23
			40	43	0.11
			120	50	0.12
1	(IB)	40(5×10 <sup>-1</sup> )	_	100	0.13
		80	<del>-</del>	100	0.15
0		40 "	(IIA) $40(1\times10^{-3})$	200	0.25
		••	80	186	0.25
		_	40	35	0.13
		· <del>-</del>	80	35	0.13
		40	(IIB) 40(1×10 <sup>-3</sup> )	220	0.21
		••	80	282	0.25
5		-	40	43	0.13
,		40	80	46	0.13
		40	(IID) 40(1×10 <sup>-3</sup> )	190	0.23
			80 40	220	0.25
		<del>-</del>	40 80	33	0.11
	(IC)	40(5×10 <sup>-1</sup> )	80	35	0.12
٠,	(IC)	80	_	105	
0		40	(IIE) 40(5×10 <sup>-4</sup> )	105 200	0.15 0.19
		"	80	220	0.19
		"	160	220	0.20
			40	45	0.18
		_	80	50	0.17
		_	160	58	0.17
-		40	(IIF)20(1×10 <sup>-3</sup> )	252	0.13
,			40	282	0.24
		**	80	218	0.26
			20	50	0.22
		_ '	40	60	0.22
			- 80	60	0.22
(	(IE)	40(5×10 <sup>-1</sup> )		110	0.18
)		80	_	115	0.24
		40	40(1×10 <sup>-3</sup> )	200	0.20
		**	80 `	234	0.21
			40	25	0.17
		_	80	25	0.19
(	(ID)	40(1×10 <sup>-1</sup> )		118	0.18
		80	_	118	0.14
•		40	(IIH) 40(5×10 <sup>-4</sup> )	250	0.17
		**	80	316	0.20
		_	40	33	0.15
			80	40	0.15
•	IG)	40(5×10 <sup>-1</sup> )		100	0.15

30

	80	·	115	0.22
	40	(IIB) 40(1×10 <sup>-3</sup> )	200	0.20
	44	80	220	0.25
(IF)	40		- 79	0.25
	80		85	0.27
	40	(IIC) 20(1×10 <sup>-3</sup> )	118	0.27
	**	40	150	0.30

(A): Relative yellow filter sensitivity (SY)

(B): Fog.

## **EXAMPLE 2**

As in Example 1, 1 kg of a silver iodobromide emulsion (containing 0.30 mol of silver and 6.0 mol% of iodine) was melted in a beaker at 40°C and the amounts of the sensitizing dyes I and II and the compound of formula III shown in Table 2 were added to the emulsion with stirring. 7 ml of the silver halide emulsion thus obtained was applied to a glass plate of cabinet size followed by drying to obtain a sample.

The sample was cut into strips and the strips were subjected to an optical wedge exposure by yellow light as in Example 1 and then developed in a developer having the following composition for 10 minutes at 20°C.

# Composition of the Developer

Water	500 ml
Metol	0.3 g
Sodium pyrosulfite	38 g
Hydroquinone	6 g
Sodium carbonate monohydrate	22.5 g
Citric acid	0.7 g
Potassium bromide	0.9 g
Water to make	1000 ml

The strips were then fixed and the optical densities thereof were measured, the results of which are shown in Table 2.

# TABLE 2

	(m tra	ol cone tion)		(me	ol (	unt(n conce on)	nl) Compd en-	Amt (ml) (mol concen- tration)	A	В	
IB	40(5	×10⁴)				-1:			100	0.15	
"	80	44			-		· · ·	_	97	0.16	
44	40	44		IB4	0(1	1×10	3)			0.15	4
**	**	**				44	·   —	_		0.16	
66	64	44	•	6	Ō.	66	IIIA	20(1×10-3)		0.15	
**	**	44	•			**	**	40 "		0.15	
64		44	4		4	**	.61	80 "		0.16	
44	"	44 .			4	**	IIIB	40(1×10-3)			
"	. 66	44			4 -	**	44	80 "		0.16	
**	**	44			•	44	IIIC	40(1×10 <sup>-3</sup> )		0.16	٦
"	"		<b>.</b>		4	**	"	80 "		0.17	

(A): Relative yellow filter sensitivity (Sy)

(B): Fog

What is claimed is:

1. A super-sensitized silver halide photographic emulsion containing at least one sensitizing dye represented by the following first formula:

$$\mathbf{z}_{1}$$
 $\mathbf{z}_{1}$ 
 $\mathbf{z}_{1}$ 
 $\mathbf{z}_{2}$ 
 $\mathbf{z}_{2}$ 
 $\mathbf{z}_{3}$ 
 $\mathbf{z}_{4}$ 
 $\mathbf{z}_{1}$ 
 $\mathbf{z}_{2}$ 

wherein R<sub>1</sub> and R<sub>2</sub> each represents alkyl, acetoxyalkyl or vinylalkyl; R<sub>3</sub> and R<sub>4</sub> are each alkyl, sulfoalkyl, sulfoalkoxyalkoxyalkyl or sulfoalkoxyalkyl with at least one of R<sub>3</sub> and R<sub>4</sub> being sulfoalkyl, sulfoalkoxyalkoxyalkyl or sulfoalkoxyalkyl, Z<sub>1</sub> and Z<sub>2</sub>, which may be the 65 same or different, each represents a non-metallic atomic group necessary to complete a substituted heterocyclic nucleus of the benzimidazole series, said

substituent being alkyl, halogen, trifluoroalkyl, cyano, alkyl-sulfonyl, alkylsulfomyl or alkylcarbamyl; and at least one sensitizing dye represented by the following second formula:

$$Z_3$$
 $\bigoplus_{N}$ 
 $CH=C$ 
 $Z_4$ 
 $N$ 
 $R_5$ 
 $R_6$ 

wherein  $R_5$  is alkyl or sulfoalkyl,  $R_6$  is alkyl, sulfoalkyl, sulfoalkoxyalkoxy or sulfoalkylcarbamylalkyl;  $Z_3$  represents a non-metallic atomic group necessary to complete a 2-quinoline substituted or unsubstituted heterocyclic nucleus, the substituent being alkyl or halogen; and  $Z_4$  represents a non-metallic group necessary to complete a substituted or unsubstituted heterocyclic nucleus selected from the group consisting of benzoxazole, naphthoxazole, benzthiazole, benzselenazole, naphthothiazole or naphthoselenazole, the substituent being halogen or sulfoalkoxy; with the proviso that when  $Z_4$  contains a sulfoalkoxy group,  $R_5$  and  $R_6$  is alkyl.

2. A super-sensitized silver halide emulsion according to claim 1 wherein said silver halide emulsion further contains a compound represented by the following formula III:

35 wherein R<sub>7</sub> represents an alkyl group or a substituted alkyl group wherein the substituent is a sulfo group neutralized by a cation, a cyano group or a vinyl group;
 Z<sub>5</sub> represents an atomic group necessary to complete a heterocyclic nucleus of the benzimidazole series; and
 40 Z<sub>6</sub> represents a phenyl group or a substituted phenyl group wherein the substituent is alkyl or alkoxy.

3. A silver halide emulsion according to claim 1 wherein the alkyl group of  $R_1$  to  $R_6$  is methyl, ethyl, n-propyl, iso-propyl, n-butyl or iso-butyl.

4. A silver halide emulsion according to claim 2 wherein the alkyl group of  $R_1$  to  $R_7$  is methyl, ethyl, n-propyl, iso-propyl, n-butyl or iso-butyl.

5. A silver halide emulsion according to claim 1 wherein said substituted alkyl group is a vinylmethyl group, a benzyl group, a β-acetoxyethyl group, a γ-acetoxypropyl group, a β-carboxyethyl group, a γ-sulfatopropyl group, a δ-sulfatobutyl group, a δ-N-acetylsulfamylbutyl group, a β-hydroxyethyl group, a γ-sulfopropyl group, a γ-sulfopropyl group, a δ-sulfobutyl group, a δ-sulfopropyl group, a 2-(3-sulfopropoxy)ethyl group, a 2[2-(3-sulfopropyl group.

6. A silver halide emulsion according to claim 1 wherein the heterocyclic nucleus of the benzimidazole series completed by Z<sub>1</sub>, Z<sub>2</sub> or Z<sub>5</sub> is a benzimidazole nucleus, or a benzimidazole nucleus containing, as a component of the skeletal-structure thereof, a benzene ring substituted by at least one of the groups consisting of a halogen atom, an N-unsubstituted or alkyl substituted sulfamyl or carbamyl group, an N-disubstituted sulfamyl group, an alkyl sulfonyl group, a trifluoromethyl group, an alkyl group, a cyano group, a

20 (IIB)

25

45

CH2CH2-OCH2CH2-OCH2CH2CH2-SO3HN(C2H5)3

phenyl group, a carboalkoxyl group, a hydroxyl group and an alkoxyl group.

- 7. A silver halide emulsion according to claim 1 wherein the heterocyclic nucleus of the 2-quinoline series is a 2-quinoline nucleus or a 2-quinoline nucleus 5 containing, as a component of the skeletal-structure thereof, a benzene ring substituted by at least one of the groups consisting of an alkyl group and a halogen
- 8. A silver halide emulsion as in claim 1 wherein the 10 light-sensitivity in the wave length region of 500-530  $m\mu$  is increased.
- 9. A silver halide emulsion as in claim 2 wherein the total amount of said sensitizing dyes I and II and said compound of formula III is within the range of from  $1 \times 15$  $10^{-6}$  mol to  $1 \times 10^{-3}$  mol per mol of silver halide.
- 10. A silver halide emulsion as in claim 1 wherein said emulsion contains the following sensitizing dyes:

(IG) 
$$C_2II_5$$
 $C_2II_5$ 
 $C_2I$ 

14. A silver halide emulsion as in claim 1 which contains the sensitizing dye of the following formula:

Ċ2H4NHCOC3H6SO3−

CH2CH2OCH2CH2OCH2CH2SO3 CH2CH2OCH2CH2OCH2CH2CH2CH2SO3HN(C2H5)3

11. A silver halide emulsion as in claim 1 wherein said emulsion contains one sensitizing dye of formula I and a sensitizing dye of the following formula: (IIB)

H<sub>2</sub>C C<sub>2</sub>H<sub>4</sub>NHCOC<sub>3</sub>H<sub>6</sub>SO<sub>3</sub>-

12. A silver halide emulsion according to claim 11

at least one sensitizing dye of formula II and at least one 35 compound of formula III.

Ċ∘Hs

15. A silver halide emulsion according to claim 14 wherein said emulsion contains the following sensitizing dyes:

CH2CH2-OCH2CH2-OCH2CH2-SO3

(IB)

(IIB) H<sub>3</sub>C Ċ₂H₅ Ċ₂H4NHCOC₃H6SO₃~

and a compound of the following formula:

IIIA) C<sub>2</sub>H<sub>5</sub> 65 C==CH--CH==N Ċ₂II₃

13. A silver halide emulsion according to claim 11 wherein said sensitizing dye of formula I is

5

15

16. A silver halide emulsion as in claim 1 wherein said emulsion contains the following sensitizing dyes:

(IIC) 
$$_{\rm H_3C}$$
  $_{\rm C_2H_5}$   $_{\rm C_2H_5}$   $_{\rm C_1}$   $_{\rm C_1}$   $_{\rm C_2H_2O)_2\,S\,O_3-}$ 

said emulsion contains the following sensitizing dyes:

(IID) 
$$\begin{array}{c} Se \\ C_{2H_{5}} \end{array}$$
 
$$\begin{array}{c} CH=C \\ O \\ (CH_{2})_{4} S O_{5} \end{array}$$

17. A silver halide emulsion as in claim 1 wherein

(IC)

20. A silver halide emulsion as in claim 1 wherein said emulsion contains the following sensitizing dyes:

$$_{\rm H_3C}$$
 $_{\rm CH=C}$ 
 $_{\rm CH_{2})_3SO_{3^-}}$ 
 $_{\rm C_2H_5}$ 

$$20 \xrightarrow[N]{C_2H_5} \xrightarrow[N]{C_2H_5} \xrightarrow[N]{C_2H_5} \xrightarrow[N]{C_2H_5} \xrightarrow[N]{Br} -Br$$

$$Br - \xrightarrow[N]{Br} -Br$$

$$-Br - Br$$

$$(CH_2)_3SO_3^- (CH_2)_3SO_3HN(C_2H_5)_3$$

18. A silver halide emulsion as in claim 1 wherein said emulsion contains the following sensitizing dyes:

(IIE) 
$$\begin{array}{c} S \\ C-CH = N \\ N \\ C_2H_5 \end{array}$$

(IC) 
$$C_{2}H_{5}$$
  $C_{1}H_{2}-CH=OH_{2}$   $C_{1}H_{3}C-S$   $C_{2}H_{5}$   $C_{2}H_{5}$   $C_{2}H_{5}$   $C_{2}H_{5}$ 

19. A silver halide emulsion as in claim 1 wherein said emulsion contains the following sensitizing dyes:

21. A silver halide emulsion as in claim 1 wherein said emulsion contains the following sensitizing dyes:

22. A photographic light-sensitive element compris-60 ing a support having thereon at least one layer containing the photographic silver halide emulsion as in claim 1.