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<b>(54) Title:</b> PERSONAL CARE COMPOSITIONS		
<b>(57) Abstract</b>  Disclosed are personal care compositions, especially hair styling compositions, containing silicone microemulsions, and corresponding applications of such compositions. The disclosed invention further relates to stable, hair styling compositions containing a silicone microemulsion and lower monohydric alcohol. A preferred personal care composition comprises: a) from about 0.01 % to about 20 % of a non-silicone containing polymer suitable for hair styling; b) an organopolysiloxane microemulsion comprising: (i) an organopolysiloxane dispersed as particles in the microemulsion, wherein the organopolysiloxane is substantially free of amino groups in combination with hydroxyl groups and has an average particle size of less than about 80 nanometers, and (ii) a surfactant system for dispersing the organopolysiloxane in the microemulsion, which is compatible with the hair styling polymer; wherein the amount of microemulsion is such that the personal care composition comprises from about 0.01 % to about 10 % of the organopolysiloxane; and c) a carrier comprising: (i) from about 3 % to about 99 %, by weight of the composition, of a first solvent selected from the group consisting of water; water soluble organic solvents; organic solvents which are strongly to moderately strong in hydrogen-bonding parameter; and mixtures thereof; wherein the first solvent is other than C <sub>1</sub> -C <sub>3</sub> monohydric alcohol, C <sub>1</sub> -C <sub>3</sub> ketone and C <sub>1</sub> -C <sub>3</sub> ether, and (ii) optionally, from about 0 % to about 55 % of a second solvent selected from the group consisting of C <sub>1</sub> -C <sub>3</sub> monohydric alcohols, C <sub>1</sub> -C <sub>3</sub> ketones, C <sub>1</sub> -C <sub>3</sub> ethers, and mixtures thereof.		

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## PERSONAL CARE COMPOSITIONS

### TECHNICAL FIELD

The present invention relates to personal care compositions, especially hair styling compositions, containing silicone microemulsions, and corresponding applications of such compositions. The invention further relates to stable, hair styling compositions containing a silicone microemulsion and lower monohydric alcohol.

### BACKGROUND OF THE INVENTION

The desire to have the hair retain a particular style or shape is widely held. The most common methodology for accomplishing this is by spraying a composition, typically from a mechanical pump spray device or from a pressurized aerosol canister, to the hair. Other means of providing style or shaping to the hair are mousses, gels, lotions and the like. Such compositions provide temporary setting benefits and can usually be removed by water or by shampooing. The materials used in these types of hair styling compositions are generally resins, gums, and adhesive polymers which are capable of imparting style or shape to the hair. Many of these products also contain lower alcohols in order to obtain good films of the polymer in a short period of time.

Many people desire a high level of style retention, or hold, from a styling product. Unfortunately, most current hair styling products having good hold characteristics suffer from the disadvantages of being either too stiff, not smooth or too sticky upon drying. Stiff compositions tend to be brittle and break down under common stresses such as wind, brushing, combing. Stiff compositions also tend to feel and look unnatural. Sticky compositions overcome many of the foregoing disadvantages of stiff compositions, because sticky compositions tend to be more forgiving, i.e., flexible, under stress and allow for restyling of the hair. However, sticky compositions have the disadvantage of leaving the hair with a heavy, coated feel and with a limp and unattractive appearance. Also, sticky compositions cause the hair to quickly become soiled from common contaminant sources such as dust, dirt, lint, sebum, etc.

One approach to minimizing stiffness and roughness of a hair styling composition is the incorporation of silicones, including silicone emulsions and

microemulsions, in such compositions. Silicones tend to provide a desirably smooth or soft hair feel. Unfortunately, silicone emulsions tend to be difficult to formulate in hair styling compositions. Hair styling compositions tend to be complex, requiring a number of ingredients for different purposes, with potential for incompatibilities. For example, silicone emulsions tend to be unstable in compositions containing lower alcohols, resulting in phase separation of the composition. Other incompatibilities in the system, e.g., polymer-polymer or polymer-surfactant interactions, can also result in phase separation. This phase separation is not only undesirable for visual esthetic reasons, but for performance reasons as well. When the product phase separates, hold and/or hair feel properties of the product tend to be negatively impacted.

Therefore, a need exists for hair styling compositions providing good style retention without the disadvantages of stiff or sticky compositions. There is a particular need for hair styling compositions containing lower alcohols providing good style retention without the disadvantages of stiff or sticky compositions.

It has surprisingly been found that stable hair styling compositions providing these benefits and containing up to about 55% monohydric alcohol can be provided by employing a combination of a (i) non-silicone containing hair styling polymer, (ii) a silicone microemulsion substantially free of a combination of amino groups and hydroxyl groups and having a specific particle size, namely less than about 80nm, and (iii) certain solvent systems for the hair styling polymer.

It is therefore an object of this invention to provide stable hair styling compositions containing silicone microemulsions, especially compositions which also contain lower alcohols.

It is another object of this invention to provide hair styling compositions that provide good style retention without unacceptable stiffness or stickiness. Another object of this invention is to provide hair styling compositions that both look and feel natural.

It is another object of this invention to provide methods for styling and holding hair.

These and other objects will become readily apparent from the detailed description which follows.

#### SUMMARY OF THE INVENTION

The present invention relates to hair styling compositions comprising:

- a) from about 0.01% to about 20% of a non-silicone containing polymer suitable for hair styling;
- b) a organopolysiloxane microemulsion comprising:

- (i) a organopolysiloxane dispersed as particles in the microemulsion, the organopolysiloxane being substantially free of amino and hydroxyl groups in combination and having an average particle size of less than about 80 nanometers, and
- (ii) a surfactant system for dispersing the organopolysiloxane in the microemulsion;

wherein the amount of microemulsion is such that the personal care composition comprises from about 0.01% to about 10% of the organopolysiloxane; and

- c) a carrier comprising:
  - (i) from about 3% to about 99.9%, by weight of the composition of a first solvent selected from the group consisting of water; water soluble organic solvents; organic solvents which are strongly to moderately strong in hydrogen-bonding parameter; and mixtures thereof; wherein the first solvent is other than C<sub>1</sub>-C<sub>3</sub> monohydric alcohol, C<sub>1</sub>-C<sub>3</sub> ketone and C<sub>1</sub>-C<sub>3</sub> ether; and
  - (ii) optionally, from about 0% to about 55% of a second solvent selected from the group consisting of C<sub>1</sub>-C<sub>3</sub> monohydric alcohols, C<sub>1</sub>-C<sub>3</sub> ketones, C<sub>1</sub>-C<sub>3</sub> ethers, and mixtures thereof.

In a preferred embodiment, the composition comprises the second solvent which is preferably a C<sub>1</sub>-C<sub>3</sub> monohydric alcohol. Compositions of this type preferably comprise water as the first solvent.

#### DETAILED DESCRIPTION OF THE INVENTION

The essential components of the present invention are described below. Also included is a nonexclusive description of various optional and preferred components useful in embodiments of the present invention.

The present invention can comprise, consist of, or consist essentially of any of the required or optional ingredients and/or limitations described herein.

All percentages and ratios are calculated on a weight basis unless otherwise indicated. All percentages are calculated based upon the total composition unless otherwise indicated.

All molecular weights are weight average molecular weights and are given in units of grams per mole.

All ingredient levels are in reference to the active level of that ingredient, and are exclusive of solvents, by-products, or other impurities that may be present in commercially available sources, unless otherwise indicated.

All measurements made are at ambient room temperature, which is approximately 73°F, unless otherwise designated.

All documents referred to herein, including all patents, patent applications, and printed publications, are hereby incorporated by reference in their entirety in this disclosure.

The term "suitable for application to human hair" and the like, as used herein, means that the compositions or components thereof so described are suitable for use in contact with human hair and the scalp and skin without undue toxicity, incompatibility, instability, allergic response, and the like.

The components of the compositions hereof are selected such that the total composition will be compatible. As used herein, compatible means there is no marked phase separation, e.g., excessive cloudiness, layering or precipitation of the composition which negatively impacts the esthetic or functional properties of the composition in a significant manner.

Preferred compositions are those wherein the mixture of essential components, namely the hair styling polymer, silicone microemulsion, and carrier is a substantially homogeneous solution or dispersion (preferably microdispersion), more preferably substantially clear to translucent in appearance. Preferred final compositions are those also characterized by these properties.

Preferred compositions are those wherein the mixture of essential components provide a % transmittance of at least about 50% at a wavelength of 460 nm as determined by standard spectroscopy methods. Preferred final compositions are those also characterized by these properties.

#### **Hair Styling Polymers**

The composition of the present invention comprises a non-silicone containing hair styling polymer for providing stylability to the hair. Hair styling polymers possess adhesive properties such that they are capable of shaping or styling the hair, and should be removable by shampooing or rinsing the hair. One or more hair styling polymers may be used. The total amount of hair styling polymer is generally from about 0.01% to about 20%, preferably from about 0.1% to about 15%, more preferably from about 0.5% to about 10%. A variety of hair styling polymers are suitable in the present invention. Particular polymers will be selected by the skilled artisan considering the solubility of the polymer in the composition and the ionicity of the composition.

Suitable hair styling polymers are those which are soluble or dispersible in the carrier described herein in the ratios employed in the composition such that the polymer forms a substantially homogeneous solution or dispersion (preferably a microdispersion) with the carrier. Solubility/dispersibility is determined at ambient conditions of temperature and pressure (25°C and 101.3 kPa (1 Atm)).

Solubility/dispersibility of the polymer should be determined after neutralization, if any.

Suitable non-silicone-containing hair styling polymers include nonionic, anionic, cationic, and amphoteric polymers, and mixtures thereof. The non-silicone-containing hair styling polymers are preferably present in a combined amount of from about 0.01% to about 20%, more preferably from about 0.1% to about 15%, and most preferably from about 0.5% to about 10% by weight of composition.

Suitable cationic polymers include Polyquaternium-4 (Celquat H-100; L200 - supplier National Starch); Polyquaternium-10 (Celquat SC-240C; SC-230 M - supplier National Starch); (UCARE polymer series - JR-125, JR-400, LR-400, LR-30M, LK, supplier Amerchol); Polyquaternium-11 (Gafquat 734; 755N - supplier ISP); Polyquaternium-16 (Luviquat FC 370; FC550; FC905; HM-552 supplier by BASF); PVP/Dimethylaminoethylmethacrylate (Copolymer 845; 937; 958- ISP supplier); Vinyl Caprolactam/PVP/Dimethylaminoethyl Methacrylate copolymer (Gaffix VC-713; H2OLD EP-1 - supplier ISP); Chitosan (Kytamer L; Kytamer PC - supplier Amerchol); Polyquaternium-7 (Merquat 550 - supplier Calgon); Polyquaternium-18 (Mirapol AZ-1 supplied by Rhone-Poulenc); Polyquaternium-24 (Quatrisoft Polymer LM-200 - supplier Amerchol); Polyquaternium-28 (Gafquat HS-100 - supplier ISP); Polyquaternium-46 (Luviquat Hold - supplier BASF); and Chitosan Glycolate (Hydagen CMF; CMFP - supplier Henkel); Hydroxyethyl Cetyldimonium Phosphate (Luviquat Mono CP - supplier BASF); and Guar Hydroxylpropyl Trimonium Chloride (Jaguar C series -13S, -14S, -17, 162, -2000, Hi-CARE 1000 - supplier Rhône-Poulenc).

Preferred cationic polymers are Polyquaternium-4; Polyquaternium-10; Polyquaternium-11; Polyquaternium-16; PVP/Dimethylaminoethylmethacrylate; Vinyl Caprolactam/PVP/Dimethylaminoethyl Methacrylate copolymer; and Chitosan.

Suitable amphoteric polymers include Octylacrylamide/Acrylates/Butylaminoethyl Methacrylate Copolymer (Amphomer 28-4910, Amphomer LV-71 28-4971, Lovocryl-47 28-4947 - National Starch supplier), and Methacryloyl ethyl betaine/methacrylates copolymer (Diaformer series supplier Mitsubishi). Preferred are Octylacrylamide/Acrylates/Butylaminoethyl Methacrylate Copolymer.

Especially preferred polymers for relatively low alcohol systems (e.g., less than about 55% alcohol) are those which are partially zwitterionic in that they always possess a positive charge over a broad range of pH but contain acidic groups which are only negatively charged at basic pH. Therefore the polymer is positively

charged at lower pH and neutral (have both negative and positive charge) at higher pHs. The zwitterionic polymer may be selected from cellulose derivatives, wheat derivatives and chitin derivatives such as are known in the art. Nonlimiting examples of zwitterionic polymers useful herein include Polyquaternium-47 (Merquat 2001 - supplier Calgon (a zwitterionic copolymer of acrylic acid, methacryl amido propyl trimethyl ammonium chloride, and methyl acrylate)); Carboxyl Butyl Chitosan (Chitolam NB/101 - marketed by Pilot Chemical Company, developed by Lamberti); and Dicarboxyethyl Chitosan (N-[(3'-hydroxy-2',3'-dicarboxy)ethyl]-beta-D-(1,4)-glucosamine) (available from Amerchol as, e.g., CHITOLAM NB/101).

Useful nonionic polymers include PVP or Polyvinylpyrrolidone (PVP K-15, K-30, K-60, K-90, K-120 - supplier ISP) (Luviskol K series 12, 17, 30, 60, 80, & 90 - supplier BASF); PVP/VA (PVP/VA series S-630; 735, 635, 535, 335, 235 - supplier ISP) (Luviskol VA); PVP/DMAPA acrylates copolymer (Styleze CC-10 - supplier ISP); PVP/VA/Vinyl Propionate copolymer (Luviskol VAP 343 E, VAP 343 I, VAP 343 PM - supplier BASF); Hydroxyethyl Cellulose (Cellosize HEC - supplier Amerchol); and Hydroxylpropyl Guar Gum (Jaguar HP series -8, -60, -105, -120 - supplier Rhône-Poulenc).

Preferred nonionic polymers are PVP or Polyvinylpyrrolidone; PVP/VA; PVP/DMAPA acrylates copolymer; and Hydroxylpropyl Guar Gum.

Anionic polymers suitable for use herein include VA/Crotonates/Vinyl Neodecanonate Copolymer (Resyn 28-2930 - National Starch supplier); Butyl Ester of PVM/MA (Gantrez A-425; ES-425; ES-435 - supplier ISP); Ethyl Ester of PVM/MA (Gantrez ES-225; SP-215 - supplier ISP); Acrylates/acrylamide copolymer (Luvimer 100P; Lumiver Low VOC, supplier BASF); Methacrylate Copolymer (Balance 0/55 - National Starch supplier); Vinyl Acetate/Crotonic Acid copolymer (Luviset CA 66 - supplier BASF); Isopropyl Ester of PVM/MA Copolymer (Gantrez ES-335 - supplier ISP); Acrylates Copolymer; Methacrylates/acrylates copolymer/amine salt (Diahold polymers - supplier Mitsubishi); 2-Butenedioic Acid (Z)-, Monoethyl Ester, Polymer with Methoxyethene (Omnirez 2000); VA/Butyl maleate/Isobornyl Acrylate (Advantage Plus terpolymer - supplier ISP); Acrylates Copolymer (Amerhold DR-25 - supplier Amerchol); Acrylates/Hydroxyesteracrylates Copolymer (Acudyne 255 supplier Rohm & Haas); vinyl Acetate/Crotonic Acid/Vinyl Propionate copolymer (Luviset CAP - supplier BASF); PVP/Acrylates copolymer (Luviflex VBM 35 - supplier BASF); Diglycol/CHDM/Isophthalates/SIP Copolymer (Eastman AQ 48, AQ 55 - supplier Eastman Chemicals); Acrylates/Octacrylamide Copolymer (Versatyl-42 or



Amphomer HC- National Starch supplier); TBA/AA copolymer (75/25 - Mitsubishi Chemical Corp.); Acrylates Copolymer (Aculyn 33 - supplier Rohm & Haas); Acrylates/Stearth-20 Methacrylate Copolymer (Aculyn 22 - supplier Rohm & Haas); and Carbomer (supplier B.F. Goodrich).

Preferred anionic polymers are VA/Crotonates/Vinyl Neodecanonate Copolymer; Butyl Ester of PVM/MA; Ethyl Ester of PVM/MA; Acrylates/acrylamide copolymer; Methacrylate Copolymer; and Vinyl Acetate/Crotonic Acid copolymer.

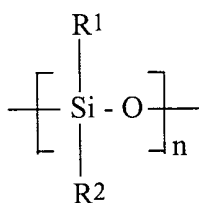
### **Organopolysiloxane Microemulsions**

The compositions of the present invention contain an organopolysiloxane microemulsion comprising polysiloxane particles dispersed in a suitable carrier (typically aqueous) with the aid of a surfactant. The organopolysiloxane is substantially free of, preferably essentially free of, more preferably contains no, organopolysiloxane having both amino groups and hydroxyl groups. The microemulsions are preferably included in an amount such that the composition contains from about 0.01 to about 10% of the dispersed polysiloxane, more preferably about 0.05% to about 6%, most preferably about 0.1% to about 4%.

Organopolysiloxane microemulsions can be produced by the emulsion polymerization of organosiloxane having a low degree of polymerization in a solvent comprising water. The organopolysiloxane is stabilized in the microemulsion by a surfactant, e.g., a nonionic surfactant and an ionic surfactant. The average particle size of the emulsion after emulsion polymerization (corresponding to the organopolysiloxane in the emulsion) is less than about 80 nanometers (nm), preferably less than about 60 nm, more preferably less than about 40 nm. Particle size of a microemulsion can be determined by conventional methods, e.g., using a Leeds & Northrup Microtrac UPA particle sizer. Microemulsions having these particle sizes are more stable and have better external appearance than those having larger particle sizes. Furthermore, the degree of polymerization (DP) of the polysiloxane after emulsion polymerization is preferably in the range of from 3 to 5,000, more preferably in the range of from 10 to 3,000.

The organopolysiloxane in the microemulsion can be a linear or branched chain siloxane fluid having a viscosity of about 20-3,000,000 mm<sup>2</sup>/s (cs), preferably 300-300,000 cs, more preferably 350-200,000 cs, at 25°C.

Suitable organopolysiloxanes may contain the difunctional repeating "D" unit:



wherein  $n$  is greater than 1 and  $\text{R}^1$  and  $\text{R}^2$  are each independently  $\text{C}_1$ - $\text{C}_7$  alkyl or phenyl. A mixture of siloxanes may be used. Exemplary siloxanes include polydimethylsiloxane, polydiethylsiloxane, polymethylethylsiloxane, polymethylphenylsiloxane, and polydiphenylsiloxane. Siloxane polymers with dimethylsiloxane "D" units are preferred from an economic standpoint. However,  $\text{R}^1$  and  $\text{R}^2$  may independently be a functional group other than methyl, e.g., carboxyalkyl, haloalkyl, acrylate, acryloxy, acrylamide, vinyl or mercaptoalkyl.

The siloxane may be terminated with hydroxy groups, alkoxy groups such as methoxy, ethoxy, and propoxy, or trimethylsiloxy groups, preferably hydroxy or trimethylsiloxy.

The emulsion can be prepared by the emulsion polymerization process described in EP 459500 (published December 4, 1992), incorporated herein by reference. In that process, stable, oil free polysiloxane emulsions and microemulsions are prepared by mixing a cyclic siloxane, a nonionic surfactant, an ionic surfactant, water, and a condensation polymerization catalyst. The mixture is heated and agitated at polymerization reaction temperature until essentially all of the cyclic siloxane is reacted, and a stable, oil free emulsion or microemulsion is formed. The reaction mix, especially surfactant levels, and conditions are controlled in order to provide the desired organopolysiloxane particle size. The emulsions and microemulsions typically have a pH of about 3 to about 10 (e.g., 6-7.5), and contain about 10 to about 70% by weight siloxane polymer, preferably about 25 to about 60%, about 0% to about 30% by weight nonionic surfactant, about 0% to about 30% by weight ionic surfactant, preferably about 0% to about 20%, the balance being water. Preferred emulsions and methods of making them are further described in U.S. Patent Application Serial No. 08/929,721, filed on September 15, 1997 in the names of Ronald P. Gee and Judith M. Vincent, incorporated herein by reference in its entirety.

Microemulsions can also be produced by the emulsion polymerization process described in EPA 0268982, published June 6, 1988, assigned to Toray, incorporated herein by reference in its entirety. In this process, the microemulsion is prepared by a process in which a crude emulsion, consisting of polysiloxane having a low degree of polymerization, a first surfactant (anionic, cationic, and nonionic

surfactants), and water, is slowly dripped into an aqueous solution containing a catalytic quantity of a polymerization catalyst and a second surfactant which acts as an emulsifying agent (which may be the same as the first surfactant, however, the surfactants should be compatible in the reaction mixture considering the ionicity of the reaction mixture). The reaction mix and conditions are controlled to provide the desired organopolysiloxane particle size. Therefore, a dropwise addition of the crude emulsion into the aqueous solution of catalyst and surfactant of 30 minutes or longer is preferred in order to produce microemulsions having smaller particle sizes. In addition, the quantity of surfactant used in the catalyst plus the surfactant aqueous solution is from about 5 to about 70 weight %, more preferably from about 25 to about 60 per 100 weight parts polysiloxane in the crude emulsion.

Any conventional nonionic surfactant can be used to prepare the microemulsion. Exemplary types of nonionic surfactants include silicone polyethers, both grafted and linear block, ethoxylated fatty alcohols, ethoxylated alcohols, ethoxylated alkyl phenols, Isolaureth-6 (polyethylene glycol ether of branched chain aliphatic C<sub>12</sub> containing alcohols having the formula C<sub>12</sub>H<sub>25</sub>(OCH<sub>2</sub>CH<sub>2</sub>)<sub>6</sub>OH), fatty acid alkanolamides, amine oxides, sorbitan derivatives (e.g., commercially available from ICI Americas, Inc., Wilmington, DE, under the tradenames SPAN and TWEEN), and propylene oxide-ethylene oxide block polymers (e.g., commercially available from BASF Corp., Parsippany, NJ under the trademark PLURONIC). Ionic surfactants useful in preparing the microemulsion include any conventional anionic surfactant such as sulfonic acids and their salt derivatives. Ionic surfactants also include any conventional cationic surfactant used in emulsion polymerization. Surfactants of these types are well known in the art and are commercially available from a number of sources. Specific examples of these surfactant types are also disclosed in the above referenced patent application Serial No. 08/929,721.

The surfactant can be used in the form of a single type of surfactant (e.g., anionic, cationic or nonionic), or the surfactant can be used as a combination of two or more types provided that they are compatible with each other and the other components of the composition. Preferred combinations of surfactant types include the combination of two or more types of anionic surfactants, the combination of two or more types of nonionic surfactants, the combination of two or more types of cationic surfactants, the combination of two or more types of surfactants selected from both the anionic and nonionic surfactants; and the combination of two or more types of surfactants selected from both the cationic and nonionic surfactants.

The catalyst employed in the emulsion polymerization may be any catalyst capable of polymerizing cyclic siloxanes in the presence of water, including condensation polymerization catalysts capable of cleaving siloxane bonds. Exemplary catalysts include strong acids and strong bases, ionic surfactants such as dodecylbenzenesulfonic acid, phase transfer catalysts, and ion exchange resins where a catalyst is formed *in situ*. As will be understood by those skilled in the art, a given surfactant may also serve as the polymerization catalyst (e.g., alkylbenzenesulfonic acids, or quaternary ammonium hydroxides or salt thereof may function as both a surfactant and the polymerization catalyst).

A surfactant system, catalyst and resulting microemulsion suitable for use in the compositions of the present invention can be selected by the skilled artisan considering the ionicity of the composition. In general, these materials are selected such that the total composition will be compatible.

The silicone micremulsion may contain a silicone polyether copolyol, such as described herein. Alternatively or additionally, the compositions hereof may contain a silicone polyether. If the microemulsion is supplied as a material not containing a silicone polyether already present in the emulsion, a silicone polyether may be added prior to making the batch composition hereof. Where the polyether is not dispersible in the silicone microemulsion, it is preferably mixed in about an equal portion of water containing from 10-50% C<sub>1</sub>-C<sub>3</sub> monohydric alcohol, preferably ethanol, prior to combination with the silicone microemulsion. This pre-mix is then added to the other ingredients of the composition which have preferably been pre-mixed.

Organopolysiloxane microemulsions are available from a number of commercial sources.

The following organopolysiloxane microemulsions are manufactured by Dow Corning of Midland, MI:

Microemulsions containing dimethicone copolyol:

Microemulsion Trade Name	Si Type	Internal phase viscosity (cps)	Surfactant	Si particle size, nm
DC 2-5791	Dimethylsiloxanol, Dimethyl cyclosiloxane	70-90M	Triethanolamine dodecylbenzene sulfonate, Polyethylene oxide lauryl ether	<50

DC 2-5791 - sp	Dimethylsiloxanol, Dimethyl cyclosiloxane	70-90M	Triethanolamine dodecylbenzene sulfonate, Polyethylene oxide lauryl ether	<40
DC 2-5932	Dimethylsiloxanol, Dimethyl cyclosiloxane	1-2M	Cetrimonium Chloride, Trideceth-12	<30

Microemulsions not containing dimethicone copolyol:

Microemulsion Trade Name	Si Type	Internal phase viscosity (cps)	Surfactant	Si particle. size, nm
DC 2-1470	Dimethylsiloxanol, Dimethyl cyclosiloxane	15-20M	Triethanolamine dodecylbenzene sulfonate, Polyethylene oxide lauryl ether	<50
DC 2-1845	Dimethylsiloxanol, Dimethyl cyclosiloxane	4-8M	Triethanolamine dodecylbenzene sulfonate, Polyethylene oxide lauryl ether	<40
DC 2-1845 - HV	Dimethylsiloxanol, Dimethyl cyclosiloxane	60-70M	Triethanolamine dodecylbenzene sulfonate, Polyethylene oxide lauryl ether	<35
DC 2-1550	Dimethylsiloxanol, Dimethyl cyclosiloxane	100-600M	Triethanolamine dodecylbenzene sulfonate, Polyethylene oxide lauryl ether	≤ 50
DC 2-1281	Dimethylsiloxanol, Dimethyl cyclosiloxane	1-2M	Cetrimonium Chloride, Trideceth-12	<30
DC 2-8194	Dimethyl, aminomethyl propyl siloxane	4-6M	Cetrimonium Chloride, Trideceth-12	≤ 30
DC 2-1716 MEM	Dimethylsiloxanol with methyl silsequioxane, Octamethyl cyclotretrasiloxane	10-30M	Cetrimonium Chloride, Trideceth-12	<u>50-80</u>

Where the composition contains a hair styling polymer which is an anionic acrylate polymer, DC 2-1845 and DC 2-5791 are preferred microemulsions. When the hair styling polymer is a cationic polymer comprising nitrogen, the DC 2-8194, DC2-1281, and/or DC 2-5932 microemulsions are preferred.

### **Carrier**

The compositions of the invention also comprise a carrier for the hair styling polymer and the silicone microemulsion. Suitable carriers are those in which the hair styling polymer is soluble or dispersible, preferably soluble or microdispersible, and wherein the organopolysiloxane of the microemulsion is dispersible. Choice of an appropriate carrier will also depend on the particular end use and product form contemplated (e.g., the hair styling polymer to be used, and the product form, e.g., for hair styling compositions such as hair spray, mousse, tonic, lotion or gel). The carrier is preferably suitable for application to the hair.

The carrier is present at from about 0.5% to about 99.5%, preferably from about 5% to about 99.5%, most preferably from about 50% to about 95%, of the composition.

The compositions of the present invention comprise one or more suitable solvents for the hair styling polymer. Preferred solvent systems are those which form a homogeneous solution or dispersion (preferably microdispersion) with the hair styling polymer in the weight ratios used in the composition. Preferred solvent systems are those which form a substantially clear to translucent solution or dispersion (preferably microdispersion) with the hair styling polymers in the weight ratios used in the composition.

Preferred solvents include those selected from the group consisting of water; water soluble organic solvents; organic solvents which are strongly to moderately strong in hydrogen-bonding parameter; and mixtures thereof; wherein the solvent is other than C<sub>1</sub>-C<sub>3</sub> monohydric alcohol, C<sub>1</sub>-C<sub>3</sub> ketone and C<sub>1</sub>-C<sub>3</sub> ether. Water is a preferred solvent. At least about 3% to about 99%, preferably at least about 5% to about 98%, of this type of solvent is used in the composition.

Exemplary water soluble organic solvents other than C<sub>1</sub>-C<sub>3</sub> monohydric alcohols, ketones and ethers include propylene glycol, glycerine, phenoxyethanol, dipropylene glycol, sugars, and mixtures thereof.

Solvents which are moderately strong to strong in hydrogen-bonding parameter other than C<sub>1</sub>-C<sub>3</sub> monohydric alcohols, ketones and ethers include esters, ethers, ketones, glycol monoethers (moderately H-bonded) and alcohols, amines, acids, amides and aldehydes (strongly H-bonded). A description and examples of

solvents of this type are disclosed in Polymer Handbook, 2d. Ed., J. Brandrup and E.H. Immergut, Editors, John Wiley & Sons, N.Y., 1975, Section IV, page 337 - 348 (Table 2). Preferred solvents of this type are dibutyl phthalate, propylene carbonate, propylene glycol monomethyl ether, methyl acetate, methyl propionate and mixtures thereof. Propylene glycol monomethyl ether, methyl acetate, methyl propionate and mixtures thereof are preferred; methyl acetate is most preferred.

Other solvents suitable for use herein are water soluble, organic volatile solvents selected from C<sub>1</sub>-C<sub>3</sub> monohydric alcohols, C<sub>1</sub>-C<sub>3</sub> ketones, C<sub>1</sub>-C<sub>3</sub> ethers, and mixtures thereof, monohydric alcohols being preferred. Preferred solvents of this type are methylal, ethanol, *n*-propanol, isopropanol, acetone and mixtures thereof. More preferred are ethanol, *n*-propanol, isopropanol, and mixtures thereof. Where the composition comprises about 40% or more of such solvents, at least about 5% of water, a water soluble organic solvent, and/or an organic solvent which is strongly to moderately strong in hydrogen-bonding parameter is used.

In a preferred embodiment, the carrier comprises (i) a solvent selected from the group consisting of water; water soluble organic solvents; organic solvents which are strongly to moderately strong in hydrogen-bonding parameter; and mixtures thereof; wherein the solvent is other than C<sub>1</sub>-C<sub>3</sub> monohydric alcohol, C<sub>1</sub>-C<sub>3</sub> ketone and C<sub>1</sub>-C<sub>3</sub> ether; (ii) a solvent selected from the groups consisting of C<sub>1</sub>-C<sub>3</sub> monohydric alcohols, C<sub>1</sub>-C<sub>3</sub> ketones, C<sub>1</sub>-C<sub>3</sub> ethers, and mixtures thereof; and (iii) mixtures thereof; preferably a mixture thereof. Especially preferred are a mixture of water and C<sub>1</sub>-C<sub>3</sub> monohydric alcohol, e.g., water-ethanol or water-isopropanol-ethanol. Another particularly preferred solvent system comprises one or more of propylene glycol monomethyl ether, methyl acetate, and methyl propionate, preferably methyl acetate, optionally with one or more of water or a C<sub>1</sub>-C<sub>3</sub> monohydric alcohol.

The carrier may include other solvents, e.g., hydrocarbons (such as isobutane, hexane, decene, acetone), halogenated hydrocarbons (such as Freon), linalool, volatile silicon derivatives, especially siloxanes (such as phenyl pentamethyl disiloxane, methoxypropyl heptamethyl cyclotetrasiloxane, chloropropyl pentamethyl disiloxane, hydroxypropyl pentamethyl disiloxane, octamethyl cyclotetrasiloxane, decamethyl cyclopentasiloxane), and mixtures thereof.

Solvents used in admixture may be miscible or immiscible with each other. However, in the final composition such solvents should be compatible with each other and other components in the composition such that solids do not precipitate.

Reduced "volatile organic compound" or "VOC" compositions may be desirable. In this regard, "VOC" refers to those organic compounds that contain less than 12 carbon atoms or have a vapor pressure greater than about 0.1 mm of mercury. For example, the composition may have, as initially applied, a total VOC content of about 55% or less (e.g., in preferred hairsprays), about 16% or less (e.g., in preferred mousses), or about 6% or less (e.g., in preferred gels). The VOC may be based on actual VOC content, or the VOC which is delivered upon initial dispensing from a package.

The carrier may also comprise conventional components such as are known in the art suitable for a given product form.

#### **Optional Components**

The present compositions can contain a wide variety of other optional ingredients that are suitable for application to human hair, including among them any of the types of ingredients known in the art for use in hair care compositions, especially hair setting compositions like hair spray compositions, mousses, gels and tonics. Generally, such other adjuvants collectively comprise from about 0.05% to about 20% by weight, preferably from about 0.1% to about 10%, more preferably 0.05% to about 5% by weight of the compositions. Such conventional optional adjuvants are well known to those skilled in the art and include, but are not limited to, plasticizers, surfactants (which may be anionic, cationic, amphoteric or nonionic), neutralizing agents, propellants, hair conditioning agents (e.g., silicone fluids, fatty esters, fatty alcohols, long chain hydrocarbons, isobutene, cationic surfactants, etc.), emollients, lubricants and penetrants such as various lanolin compounds, vitamins, proteins, preservatives, dyes, tints, bleaches, reducing agents and other colorants, sunscreens, thickening agents (e.g., polymeric thickeners, such as xanthan gum), physiologically active compounds for treating the hair or skin (e.g., anti-dandruff actives, hair growth actives) and perfume.

Non-exclusive examples of certain types of optional components are provided below.

##### **a) Plasticizers**

The compositions hereof may contain a plasticizer for the hair styling polymer. Any plasticizer suitable for use in hair care products or for topical application to the hair or skin can be used. A wide variety of plasticizers are known in the art. These include glycerine, diisobutyl adipate, butyl stearate, propylene glycol, diethylene glycol, other glycols, tri-C<sub>2</sub>-C<sub>8</sub> alkyl citrates, including triethyl citrate and analogs of triethyl citrate.



Plasticizers are typically used at levels of from about 0.01% to about 200%, preferably from about 0.05% to about 100%, more preferably from about 0.1% to about 50%, by weight of the polymer.

**b) Surfactants**

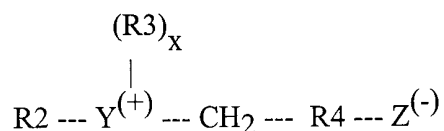
The hair styling compositions can contain one or more surfactants, e.g., for emulsifying hydrophobic components which may be present in the composition. Surfactants are preferred for use in mousse products. Generally, if used, such surfactants will be used at a total level of from about 0.01% to about 10%, preferably from about 0.01% to about 5% and more preferably from about 0.01% to about 3%, by weight of the composition. A wide variety of surfactants can be used, including anionic, cationic, amphoteric, and zwitterionic surfactants.

Anionic surfactants include, for example: alkyl and alkenyl sulfates; alkyl and alkenyl ethoxylated sulfates; (preferably having an average degree of ethoxylation of 1 to 10), succinamate surfactants, such as alkylsulfosuccinamates and dialkyl esters of sulfosuccinic acid; neutralized fatty acid esters of isethionic acid; and alkyl and alkenyl sulfonates, including, for example, olefin sulfonates and beta-alkoxy alkane sulfonates. Preferred are alkyl and alkenyl sulfates and alkyl and alkenyl ethoxylated sulfates such as the sodium and ammonium salts of C<sub>12</sub>-C<sub>18</sub> sulfates and ethoxylated sulfates with a degree of ethoxylation of from 1 to about 6, preferably from 1 to about 4, e.g., lauryl sulfate and laureth (3.0) sulfate.

Amphoteric surfactants include those which can be broadly described as derivatives of aliphatic secondary and tertiary amines in which the aliphatic radical can be straight chain or branched and wherein one of the aliphatic substituents contains from about 8 to about 18 carbon atoms and one contains an anionic water solubilizing group, e.g., carboxy, sulfonate, sulfate, phosphate, or phosphonate. Examples of compounds falling within this definition are sodium 3-dodecylaminopropionate, N-alkyltaurines such as the one prepared by reacting dodecylamine with sodium isethionate according to the teaching of U.S. Patent 2,658,072, N-higher alkyl aspartic acids such as those produced according to the teaching of U.S. Patent 2,438,091, and the products sold under the trade name "Miranol" and described in U.S. Patent 2,528,378. Others include alkyl, preferably C<sub>6</sub>-C<sub>22</sub> and most preferably C<sub>8</sub>-C<sub>12</sub>, amphoglycinates; alkyl, preferably C<sub>6</sub>-C<sub>22</sub> and most preferably C<sub>8</sub>-C<sub>12</sub>, amphopropionates; and mixtures thereof.

Suitable zwitterionic surfactants for use in the present compositions can be exemplified by those which can be broadly described as derivatives of aliphatic quaternary ammonium, phosphonium, and sulfonium compounds, in which the aliphatic radicals can be straight chain or branched, and wherein one of the aliphatic

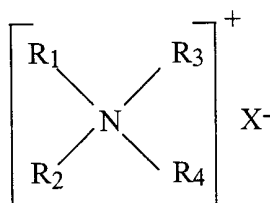
substituents contains from about 8 to 18 carbon atoms and one contains an anionic water-solubilizing group, e.g., carboxy, sulfonate, sulfate, phosphate, or phosphonate. A general formula for these compounds is:



wherein R2 contains an alkyl, alkenyl, or hydroxy alkyl radical of from about 8 to about 18 carbon atoms, from 0 to about 10 ethylene oxide moieties and from 0 to 1 glyceryl moiety; Y is selected from the group consisting of nitrogen, phosphorus, and sulfur atoms; R3 is an alkyl or monohydroxyalkyl group containing 1 to about 3 carbon atoms; x is 1 when Y is sulfur or phosphorus, 1 or 2 when Y is nitrogen; R4 is an alkylene or hydroxyalkylene of from 1 to about 4 carbon atoms and Z is a radical selected from the group consisting of carboxylate, sulfonate, sulfate, phosphonate, and phosphate groups. Classes of zwitterionics include alkyl amino sulfonates, alkyl betaines, and alkyl amido betaines.

Cationic surfactants useful in compositions of the present invention contain amino or quaternary ammonium hydrophilic moieties which are positively charged when dissolved in the aqueous composition of the present invention. Cationic surfactants among those useful herein are disclosed in the following documents, all incorporated by reference herein: M. C. Publishing Co., McCutcheon's, Detergents & Emulsifiers, (North American edition 1979); Schwartz, et al., Surface Active Agents, Their Chemistry and Technology, New York: Interscience Publishers, 1949; U.S. Pat. No. 3,155,591, Hilfer, issued Nov. 3, 1964; U.S. Pat. No. 3,929,678, Laughlin, et al., issued Dec. 30, 1975; U.S. Pat. No. 3,959,461, Bailey, et al., issued May 25, 1976; and U.S. Pat. No. 4,387,090, Bolich, Jr., issued June 7, 1983.

Among the quaternary ammonium-containing cationic surfactant materials useful herein are those of the general formula:



wherein R<sub>1</sub> is an aliphatic group of from 1 to 22 carbon atoms, or an aromatic, aryl or alkylaryl group having from 12 to 22 carbon atoms; R<sub>2</sub> is an aliphatic group having from 1 to 22 carbon atoms; R<sub>3</sub> and R<sub>4</sub> are each alkyl groups having from 1 to 3 carbon atoms, and X is an anion selected from halogen, acetate, phosphate,

nitrate and alkylsulfate radicals. The aliphatic groups may contain, in addition to carbon and hydrogen atoms, ether linkages, and other groups such as amido groups. Other quaternary ammonium salts useful herein are diquaternary ammonium salts.

Salts of primary, secondary and tertiary fatty amines are also suitable cationic surfactants for use herein. The alkyl groups of such amines preferably have from 12 to 22 carbon atoms, and may be substituted or unsubstituted. Secondary and tertiary amines are preferred, tertiary amines are particularly preferred. Such amines, useful herein, include stearamido propyl dimethyl amine, diethyl amino ethyl stearamide, dimethyl stearamine, dimethyl soyamine, soyamine, myristyl amine, tridecyl amine, ethyl stearylamine, N-tallowpropane diamine, ethoxylated (5 moles E.O.) stearylamine, dihydroxy ethyl stearylamine, and arachidyl-behenylamine. Cationic amine surfactants included among those useful in the present invention are disclosed in U.S. Pat. No. 4,275,055, Nachtigal, et al., issued June 23, 1981 (incorporated by reference herein).

Suitable cationic surfactant salts include the halogen, acetate, phosphate, nitrate, citrate, lactate and alkyl sulfate salts.

Nonionic surfactants include polyethylene oxide condensates of alkyl phenols (preferably C<sub>6</sub>-C<sub>12</sub> alkyl, with a degree of ethoxylation of about 1 to about 6), condensation products of ethylene oxide with the reaction product of propylene oxide and ethylene diamine, condensation products of aliphatic alcohols with ethylene oxide, long chain (i.e., typically C<sub>12</sub>-C<sub>22</sub>) tertiary amine oxides, long chain tertiary phosphine oxides, dialkyl sulfoxides containing one long chain alkyl or hydroxy alkyl radical and one short chain (preferably C<sub>1</sub>-C<sub>3</sub>) radical, silicone copolyols, and C<sub>1</sub>-C<sub>4</sub> alkanol amides of acids having a C<sub>8</sub>-C<sub>22</sub> acyl moiety. Preferred nonionic surfactants are C<sub>1</sub>-C<sub>4</sub> alkanol amides of acids having a C<sub>8</sub>-C<sub>22</sub> acyl moiety, polyoxyethylene glycol stearyl ethers, and mixtures thereof. Specific examples which are preferred are Lauramide DEA, Steareth-21, Steareth-2, and Na Cocoyl Isethionate..

Additional surfactants suitable for use herein include those described in reference to the microemulsion.

**c) Neutralizing Agents**

Hair styling polymers which have acidic functionalities, such as carboxyl groups, are preferably used in at least partially neutralized form to promote solubility/dispersibility of the polymer. In addition, use of the neutralized form aids in the ability of the dried hair styling compositions to be removed from the hair by shampooing. The degree of neutralization must balance shampoo removability versus humidity resistance. Neutralization levels in excess of what is required for

shampoo removability will result in excessively sticky products that will not hold as well in high humidity. When available acidic monomers are neutralized, it is preferred that from about 5% to 60%, more preferably from about 10% to about 40%, and even more preferably from about 12% to about 30% of the polymer (on a total polymer weight basis) be neutralized. The optimal level of neutralization for a specific polymer will depend on the polarity of the monomers selected, the specific ratios of the monomers to each other, and the percentage of acidic monomers. The level of base needed to neutralize the acid groups in a polymer for a specific % neutralization of the polymer may be calculated from the following equation:

$$\begin{array}{l} \text{\% Base in} \\ \text{composition} \end{array} = A \times (B/100) \times (C/D)$$

A = % Polymer in composition

B = % of polymer to be neutralized (assuming acid groups are available)

C = MW of Base

D = MW of Acid monomer

Any conventionally used base, including organic or inorganic (metallic or other) bases, can be used for neutralization of the polymers. Metallic bases are particularly useful in the present compositions. Hydroxides, where the cation is ammonium, an alkali metal or an alkaline earth metal, are suitable neutralizers for use in the present compositions. Preferred inorganic neutralizing agents for use in the compositions of the present invention are potassium hydroxide and sodium hydroxide. Examples of other suitable neutralizing agents which may be included in the hair styling compositions of the present invention include amines, especially amino alcohols such as 2-amino-2-methyl-1,3-propanediol (AMPD), 2-amino-2-ethyl-1,3-propanediol (AEPD), 2-amino-2-methyl-1-propanol (AMP), 2-amino-1-butanol (AB), monoethanolamine (MEA), diethanolamine (DEA), triethanolamine (TEA), monoisopropanolamine (MIPA), diisopropanolamine (DIPA), triisopropanolamine (TIPA), dimethyl laurylamine (DML), dimethyl myristamine (DMM) and dimethyl stearamine (DMS).

Polymers having basic functionalities, e.g., amino groups, are preferably at least partially neutralized with an acid, e.g., hydrochloric acid.

Neutralization can be accomplished by techniques well known in the art, and before or after polymerization of the monomers comprising the hair styling polymer.

**d) Hair Conditioning Polymers**

The compositions of the invention may include a hair conditioning polymer for purposes of improved wet combing, dry combing and/or improved manageability (e.g., frizz or static control). Hair conditioning polymers are typically used at a level of from about 0.001% to about 6%, more preferably from about 0.01% to about 5% of the composition.

Cationic and zwitterionic hair conditioning polymers are preferred. Suitable hair conditioning polymers include cationic polymers having a weight average molecular weight of from about 5,000 to about 10 million, and will generally have cationic nitrogen-containing moieties such as quaternary ammonium or cationic amino moieties, and mixtures thereof. Cationic charge density should be at least about 0.1 meq/gram, preferably less than about 3.0 meq/gram, which can be determined according to the well known Kjeldahl Method. Those skilled in the art will recognize that the charge density of amino-containing polymers can vary depending upon pH and the isoelectric point of the amino groups. The charge density should be within the above limits at the pH of intended use. Any anionic counterions can be utilized for the cationic polymers so long as they are compatible.

The cationic nitrogen-containing moiety will be present generally as a substituent, on a fraction of the total monomer units of the cationic hair conditioning polymers. Thus, the cationic polymer can comprise copolymers, terpolymers, etc. of quaternary ammonium or cationic amine-substituted monomer units and other non-cationic units referred to herein as spacer monomer units. Such polymers are known in the art, and a variety can be found in *International Cosmetic Ingredient Dictionary*, Sixth Edition, 1995, which is incorporated by reference herein in its entirety.

Suitable cationic polymers include, for example, copolymers of vinyl monomers having cationic amine or quaternary ammonium functionalities with water soluble spacer monomers such as acrylamide, methacrylamide, alkyl and dialkyl acrylamides, alkyl and dialkyl methacrylamides, alkyl acrylate, alkyl methacrylate, vinyl caprolactone, and vinyl pyrrolidone. The alkyl and dialkyl substituted monomers preferably have C<sub>1</sub>-C<sub>7</sub> alkyl groups, more preferably C<sub>1</sub>-C<sub>3</sub> alkyl groups. Other suitable spacer monomers include vinyl esters, vinyl alcohol (made by hydrolysis of poly-vinyl acetate), maleic anhydride, propylene glycol, and ethylene glycol. The cationic polymers are described in detail in U.S. Patent 4,733,677 which is hereby incorporated by reference to further describe the cationic polymers used for conditioning purposes.

The cationic amines can be primary, secondary, or tertiary amines, depending upon the particular species and the pH of the composition. In general, secondary and tertiary amines, especially tertiary amines, are preferred. The cationic polymers can comprise mixtures of monomer units derived from amine- and/or quaternary ammonium-substituted monomer and/or compatible spacer monomers.

Other cationic polymers that can be used include polysaccharide polymers, such as cationic cellulose derivatives, cationic starch derivative, and cationic guar gum derivatives. Other materials include quaternary nitrogen-containing cellulose ethers as described in U.S. Patent 3,962,418, and copolymers of etherified cellulose and starch as described in U.S. Patent 3,958,581, which descriptions are incorporated herein by reference.

Where the composition comprises a neutralized, anionic hair styling polymer and a zwitterionic hair conditioning polymer, the pH of the zwitterion is preferably adjusted to that of the neutralized hair styling polymer prior to combination therewith. Neutralization may be achieved by conventional methods using pH-adjusting agents such as are known in the art.

**e) Propellants**

When the hair styling compositions are to be dispensed from a pressurized aerosol container (e.g., certain hair sprays and mousses), a propellant which consists of one or more of the conventionally-known aerosol propellants can be used to propel the compositions. A suitable propellant for use can be any gas conventionally used for aerosol containers, preferably a liquifiable gas. Suitable propellants for use are volatile hydrocarbon propellants which can include liquified lower hydrocarbons of 3 to 4 carbon atoms such as propane, butane and isobutane. Other suitable propellants are hydrofluorocarbons such as 1,2-difluoroethane (Hydrofluorocarbon 152A) supplied as Dymel 152A by DuPont. Other examples of propellants are dimethylether, nitrogen, carbon dioxide, nitrous oxide, and atmospheric gas. For hair sprays and mousses, the selection of appropriate hydrocarbons is made to provide a stable system giving the desired spray/foam quality.

The aerosol propellant may be mixed with the present hair styling compositions and the amount of propellant to be mixed is governed by normal factors well known in the aerosol art. Generally, for liquifiable propellants, the level of propellant is from about 1% to about 60% by weight of the total composition. For hair sprays, the propellant level is from about 10% to about 60% by weight of the total composition, preferably from about 15% to about 50% by weight of the total composition. For mousses, the level of propellant is generally from about 1%

to about 30% and more preferably from about 4% to about 15% by weight of the total composition.

Alternatively, pressurized aerosol dispensers can be used where the propellant is separated from contact with the hair styling composition such as a two compartment can of the type sold under the tradename SEPRO from American National Can Corp.

Other suitable aerosol dispensers are those characterized by the propellant being compressed air which can be filled into the dispenser by means of a pump or equivalent device prior to use. Such dispensers are described in U.S. Patents 4,077,441, March 7, 1978, Olofsson and 4,850,577, July 25, 1989, TerStege, both incorporated by reference herein, and in U.S. Serial No. 07/839,648, Gosselin et al., filed February 21, 1992, also incorporated by reference herein. Compressed air aerosol containers suitable for use herein are also those previously marketed by The Procter & Gamble Company under their tradename VIDAL SASSOON AIRSPRAY<sup>®</sup> hair sprays.

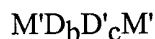
Furthermore, non-aerosol foams may also be mixed with the present hair styling composition such that the final composition is dispensable as a stable foam. A composition is "dispensable as a stable foam" when it produces a foam when dispensed from a package or container which is either pressurized or equipped with an air or gas mixing device like the F2 non-aerosol foamer described in U.S. Patents 5,271,530; 5,337,929; and 5,443,569; all of which are herein incorporated by reference.

**f) Silicone Polyether Surfactant**

Compositions of the present invention may contain a silicone polyether suitable for stabilizing the organopolysiloxane microemulsion. The silicone polyether is especially preferred for relatively high alcohol compositions (e.g., at least about 40% monohydric alcohol). The silicone polyether comprises a polymeric portion comprising repeating organosiloxane units, and a polymeric portion comprising repeating alkylene oxide units (i.e., a silicone-polyoxyalkylene copolymer). Suitable silicone polyethers are those which are surface active in the carrier employed in the compositions of the invention. As will be understood in the art, the surface activity of the silicone polyether will depend on the molecular weight of the polymeric portion comprising repeating organosiloxane units. This portion should be of sufficiently large molecular weight such that it is insoluble in the carrier, yet not so large that it renders the whole molecule insoluble in the carrier. When used, the silicone polyether is used in an amount effective to stabilize the microemulsion, preferably from about 0.02% to about 7%, more preferably about 0.05% to about 5%, of the total composition.

The silicone polyether may be a silicone - linear polyoxyalkylene block copolymer (wherein the polymeric backbone comprises silicone blocks and polyoxyalkylene blocks, optionally having grafts), a silicone - graft polyoxyalkylene copolymer (wherein the polymeric backbone comprises silicone blocks and the polyoxyalkylene blocks are present as grafts rather than in the backbone), or a mixture thereof. Linear copolymers are preferred.

Silicone linear block polyethers, particularly preferred for relatively high alcohol compositions (e.g., at least about 40% monohydric alcohol) include those having the formula (I):



wherein

**M'** is a monofunctional unit  $R_2R'SiO_{1/2}$ ;

**D** is a difunctional unit  $R_2SiO_{2/2}$ ;

**D'** is a difunctional unit  $RR'SiO_{2/2}$ ;

**R** is independently H, C<sub>1</sub>- C<sub>6</sub> alkyl, or aryl, preferably H or C<sub>1</sub>- C<sub>4</sub> alkyl, more preferably CH<sub>3</sub>;

**R'** is independently, an oxyalkylene containing moiety, H, or CH<sub>3</sub>;

**b** is an integer of from about 10 to about 1000, preferably about 10 to about 500, more preferably about 20 to about 200; and



**c** is an integer of from 0 to about 100, preferably 0 to about 50, more preferably **c** is 0, provided that when **c** is 0, at least one **M'** contains an oxyalkylene moiety.

Preferred **R'** in structure (I) are those having the formula:



wherein

**R''** is a divalent radical for connecting the oxyalkylene portion of moiety **R'** to the siloxane backbone, preferably  $-(C_mH_{2m})-$ , wherein **m** is an integer of from 2 to 8, preferably from 2-6, more preferably from 3-6;

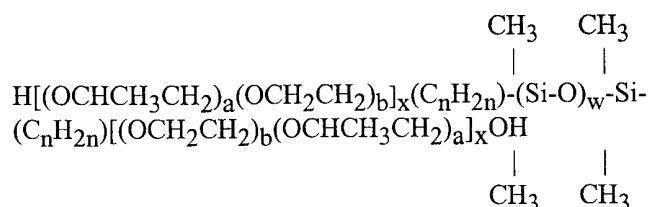
**R'''** is a terminating radical for the oxyalkylene portion of the moiety **R'**, e.g., H, hydroxyl,  $C_1 - C_6$  alkyl, aryl, alkoxy (e.g.,  $C_1 - C_6$ ) or acyloxy (e.g.,  $C_1 - C_6$ ), preferably hydroxyl;

**n** is an integer of from 2 to 4, preferably 2 to 3 (i.e., the oxyalkylene group may contain ethylene oxide, propylene oxide and/or butylene oxide units); and

**y** is 1 or greater, wherein the total **y** from all the oxyalkylene units in the copolymer is 10 or greater.

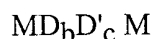
The oxyalkylene moiety of **R'** may be a random copolymer, a block copolymer or a mixture thereof. Preferred **R'** groups in structure (I) are those wherein the oxyalkylene units are selected from ethylene oxide units (EO), propylene oxide units (PO), and mixtures thereof. More preferred are those wherein the oxyalkylene units have an ethylene oxide unit (EO) to propylene oxide unit (PO) ratio of  $EO_{10-100}PO_{0-100}$ , more preferably  $EO_{20-70}PO_{20-70}$ , most preferably  $EO_{30-70}PO_{30-70}$ , based on the total oxyalkylene in the silicone polyether.

Particularly preferred silicone polyethers for relatively high alcohol compositions (e.g., at least about 40% monohydric alcohol) are those having the formula:



wherein **n** is as defined above, **x** is independently an integer of 1 or greater, **a** and **b** independently are an integer of from about 15 to about 30, and **w** is an integer of from about 20 to about 200, preferably from about 30 to about 200. Such silicone polyethers are commercially available from Goldschmidt Chemical Company under the tradename TEGOPREN 5830.

Preferred silicone graft polyethers for use herein are those having the formula (II):



wherein:

**M** is a monofunctional unit  $R_3SiO_{1/2}$ ,

**D** is a difunctional unit  $R_2SiO_{2/2}$ ,

**D'** is a difunctional unit  $RR'SiO_{2/2}$ ,

**R** is independently H, C<sub>1</sub>-C<sub>6</sub> alkyl, or aryl, preferably H or C<sub>1</sub>-C<sub>4</sub> alkyl, more preferably CH<sub>3</sub>,

**R'** is an oxyalkylene containing moiety,

**b** is an integer of from about 10 to about 1000, preferably about 100 to about 500, and

**c** is an integer of from 1 to about 100, preferably 1 to about 50.

Preferred **R'** are those having the formula



wherein

**R''** is a divalent radical for connecting the oxyalkylene portion of moiety **R'** to the siloxane backbone, preferably  $-(C_mH_{2m})-$ , wherein **m** is an integer of from 2 to 8, preferably 2-6, more preferably 3-6;

**R'''** is a terminating radical for the oxyalkylene portion of moiety **R'**, e.g., H, hydroxyl, C<sub>1</sub>-C<sub>6</sub> alkyl, aryl, alkoxy (e.g., C<sub>1</sub>-C<sub>6</sub>) or acyloxy (e.g., C<sub>1</sub>-C<sub>6</sub>), preferably hydroxy or acyloxy, more preferably hydroxyl;

**n** is an integer of from 2 to 4, preferably 2 to 3 (i.e., the oxyalkylene group may contain ethylene oxide, propylene oxide and/or butylene oxide units); and

**y** is 1 or greater.

The oxyalkylene moiety of **R'** may be a random copolymer, a block copolymer or a mixture thereof. Preferred **R'** groups are those wherein the oxyalkylene units are selected from ethylene oxide units (EO), propylene oxide units

(PO), and mixtures thereof. More preferred are those wherein the oxyalkylene units have an ethylene oxide unit (EO) to propylene oxide unit (PO) ratio of EO<sub>10-100</sub>PO<sub>0-100</sub>, more preferably EO<sub>10-30</sub>PO<sub>1-30</sub>, based on the total oxyalkylene in the silicone polyether.

In this regard, Table 1 shows some representative silicone graft polyethers:

Table 1

Silicone Polyether*	Structure	Weight % EO	Weight % PO	HLB**
A	EO	19	0	6.8
B	EO	40	0	8.0
C	EO/PO	34	0.4	6.8
D	EO/PO	41	12	8.2
E	EO/PO	34	39	5.0
F	EO/PO	32	42	6.4
G	EO/PO	30	40	5.7

\* Silicone Polyethers A and B contain less than 20 D units and less than 5 D' units. Silicone Polyethers C-G contain from 100-200 D units and 10-30 D' units.

\*\* Hydrophilic lipophilic balance (HLB) is determined by calculating the weight percent of EO and dividing this value by five.

Silicone polyethers of this type are further described in the above referenced patent application Serial No. 08/929,721.

Siloxane-oxyalkylene copolymers, i.e., silicone polyethers, can be prepared according to methods generally described in the standard text on silicone chemistry entitled "Chemistry and Technology of Silicones," by Walter Noll, Academic Press Inc., Orlando, Florida, (1968), on pages 373-376. Silicone polyethers are also available from a number of commercial sources such as:

Trade Name	Supplier	Silicone Content %	EO and/or PO	Molecular Weight
D.C.Q2-5220	Dow Corning <sup>1</sup>	14	EO & PO	3102
D.C.193	"	33	EO	-
D.C.190	"	24	EO & PO	2570
D.C. Q4-3667	"	37	EO	2400
Silwet L-7200	OSI <sup>2</sup>	31	EO & PO	19,000
Tegopren 5830/ Abil B8830	Goldschmidt A.G. <sup>3</sup>	55	40% EO/60% PO	7800

Tegopren 5830 - A	"	50	30% EO/70% PO	9000
Tegopren 5830 - B	"	50	60% EO/40% PO	9000
Abil B8851	"	-	EO & PO	>1400
Abil B8863	"	-	EO & PO	>3000
Abil EM 97 neat	"	75	60% EO/40% PO	14,000 - 15,000

<sup>1</sup> Dow Corning of Midland, Michigan

<sup>2</sup> OSI Specialties of Lisle, Il.

<sup>3</sup> Hopewell, VA

Other silicone polyethers are available as SF-1188 offered by General Electric of Waterford, N.Y., and KF353A offered by Shin Etsu Silicones of America of Torrance, Ca.. Additional silicone polyethers are described in U.S. Patent 4,871,529, incorporated herein by reference.

**g) Other optional components**

Compositions of the invention may contain a variety of other ingredients such as are conventionally used in a given product form. The compositions hereof may, for example, be in the form of a hairspray, mousse, gel, lotion, cream, pomade, spray-on product such as spray-on gel, heat protectant spray, volumizing spray, spritz, hair tonic, and the like. The compositions may be aerosol or non-aerosol. Such compositions are described, for example in California Code of Regulations, Regulation for Reducing Volatile Organic Compound Emissions from Consumer Products, Amendment 2, Consumer Products, Sections 94507-94717, Title 17, filed September 19, 1991 and effective October 21, 1991; and in Formulation and Function of Cosmetics, J. S. Jellinek, Wiley Interscience (1970), each incorporated herein by reference.

**Method of Making**

The hair styling compositions of the present invention can be made using conventional formulation and mixing techniques. The hair styling polymer and the solvent are mixed to provide a homogeneous mixture. Any other ingredients are then added and mixed to yield the final composition. If the polymer is neutralized, the neutralizer is preferably added prior to addition of other ingredients. For hair spray products, the composition is packaged in conventional mechanical pump spray devices, or alternatively, in the case of aerosol sprays products, the composition is packaged in conventional aerosol canisters along with an appropriate propellant system (also applicable for mousses). Other hair styling compositions including tonics, lotions, and gels, are typically packaged in a conventional bottle or tube.

When the compositions of the present invention comprise a silicone microemulsion, it is important to add the silicone microemulsions to the system when some water (or other polar material) is present. The water helps to maintain the stability of the silicone microemulsions in the composition. The amount of water which is preferably present depends on the type of hair styling polymer and the presence and type of silicone copolyol. If the composition does not contain a silicone copolyol, at least about 30% water is preferably present. Where the composition contains a silicone copolyol having structure (I) (e.g., Tegopren), at least about 3% water is preferably present. When the composition contains other dimethicone copolyols at least about 25% water is preferably present.

Additionally, it is desirable to not impart high shear rates to the composition once the silicone microemulsion has been added, since shear might break the emulsion. Also, the silicone microemulsion is typically added after any thickeners or surfactants.

#### **Method of Use**

The compositions of the present invention are used in conventional ways to provide the hair care benefits of the present invention. Such methods generally involve application of an effective amount of the product to dry, slightly damp, or wet hair before and/or after the hair is dried and arranged to a desired style. Application of the product is normally effected by spraying or atomizing the product using an appropriate device, e.g. a mechanical pump spray, a pressurized aerosol container, or other appropriate means. Other hair styling compositions including tonics, lotions, and gels, are typically dispensed from a conventional bottle or tube, and applied directly to the hair or first dispensed to the hand and then to the hair. The composition is then dried or allowed to dry. By "effective amount" is meant an amount sufficient to provide the hair hold and style benefits desired. In general, from about 0.5g to about 30g of product is applied to the hair, depending upon the particular product formulation, dispenser type, length of hair, and type of hair style.

The present compositions are also useful in other applications where the benefits of the composition hereof may be realized. For example, other applications in the personal care area or household care area may benefit from a composition containing a polymer having hair styling properties, along with the silicone microemulsion.

#### **Examples**

The following examples further describe and demonstrate embodiments within the scope of the present invention. The examples are given solely for the purpose of illustration and are not to be construed as limitations of the present invention, as many variations thereof are possible without departing from the spirit and scope of the invention. Ingredients are identified by chemical or CTFA name.

Example I

The following is a gel composition representative of the present invention:

<u>Component</u>	<u>Weight %</u>
Ethanol	5.00
PVP/VA Copolymer (50% active) <sup>1</sup>	6.00
Isosteareth 20 <sup>2</sup>	0.30
Perfume	0.20
Diethylene Glycol <sup>3</sup>	0.30
Triethanolamine <sup>4</sup>	0.47
Water	84.03
Glycerin	0.25
Carbomer <sup>5</sup>	0.50
Silicone Microemulsion (25% active) <sup>6</sup>	2.00
Preservatives	0.95

- 1 PVP/VA 735W, ISP
- 2 Arosurf 66 E-20, Witco
- 3 Diethylene glycol, Ashland
- 4 Triethanol amine, Dow Chemical
- 5 Carbpol 940, BF Goodrich
- 6 DC-1845 - Silicone microemulsion from Dow Corning with a particle size of 33 nm, an anionic/nonionic surfactant system, and a silicone with an internal phase viscosity = 70,000 cps.

A premix is prepared by completely dissolving the PVP/VA in 30% of the batch water (100° F) with vigorous agitation. In the following order, under continued agitation, add the Diethylene glycol, the Isosteareth-20, the perfume, the Glycerin, the ethanol and the preservatives.

The final mix is prepared under vigorous agitation by dissolving the Carbomer in 50% of the batch water (120° F) until the carbomer is fully hydrated. With mild agitation, add Triethanol amine to carbomer/water, allowing gelation to occur. With continued mild agitation, add the premix and the silicone microemulsion (premixed with the remaining water) to the mix.

Example II

The following is a gel composition representative of the present invention:

<u>Component</u>	<u>Weight %</u>
Water	76.75
Ethanol	16.00
Polyquaterium - 16 <sup>1</sup>	2.00
Benzophenone - 4 <sup>2</sup>	0.10
Perfume	0.20
Hydroxypropyl Guar <sup>3</sup>	1.00
Silicone Microemulsion (37 % active) <sup>4</sup>	3.00
Preservatives	0.95

- 1 Luviquat FC 550, BASF
- 2 Uvinul MS-40, BASF
- 3 Jaguar HP-105 , Rhone-Poulenc
- 4 DC-2-5932 silicone microemulsion from Dow Corning with a particle size of 24 nm, a cationic surfactant system, and a silicone with an internal phase viscosity = 1,200 cps.

A premix is prepared by adding the Polyquaternium-16 to 60% of the water under moderate agitation. Next with vigorous agitation, add the Benzophenone, the Ethanol, the Perfume, and the preservatives.

The final mix is prepared by adding the hydroxypropyl guar to 30% of the water, mixing vigorously until the guar hydrates completely. Next, add the premix with gentle agitation. Then add the silicone microemulsion (premixed with the remaining water) to the mix with mild agitation.

Example III

The following is a gel composition representative of the present invention:

<u>Component</u>	<u>Weight %</u>
PVP <sup>1</sup>	2.00
PVP/VA Copolymer (50% active) <sup>2</sup>	3.00
Isosteareth 20 <sup>3</sup>	0.30
Perfume	0.20
Polyquaterium-47 <sup>4</sup>	0.50
Water	88.60
Propylene Glycol	0.20
Glycerin	0.25
Hydroxyethyl Cellulose <sup>5</sup>	1.00
Silicone Microemulsion (25% active) <sup>6</sup>	3.00
Preservatives	0.95

- 1 PVP K-30, ISP
- 2 Luviskol VA 73W, BASF
- 3 Arosurf 66 E-20, Witco
- 4 Merquat - 2001, Calgon.
- 5 Cellosize PCG-10, Amerchol
- 6 DC-1550 silicone microemulsion from Dow Corning with a particle size of 50 nm, an anionic/nonionic surfactant system, and a silicone with an internal phase viscosity = 100,000 cps.

The premix is prepared by completely dissolving the PVP and PVP/VA in 30% of the batch water (120° F) with vigorous agitation. Under continued agitation, add the Isosteareth-20 and the perfume, and then remove the mix from the heat. While still vigorously agitating, add the Glycerin, the Merquat 2001, and the preservatives.

The final mix is prepared under vigorous agitation by dissolving the HEC in 50% of the batch water (155° F) until the HEC is fully hydrated. With mild agitation, add premix and then the silicone microemulsion (premixed with the remaining water).

#### Example IV

The following is a mousse composition representative of the present invention:

##### Concentrate composition:

<u>Component</u>	<u>Weight %</u>
Ethanol (200 proof)	10.00
Polyquaternium 11 (20% active) <sup>1</sup>	12.50
Polyquaternium 4 <sup>2</sup>	0.50
Perfume	0.10
Water	72.39
Glycerin	0.30
Lauramide DEA <sup>3</sup>	0.20
Stareth 21 <sup>4</sup>	0.10
Silicone Microemulsion (25% active) <sup>5</sup>	3.00
Preservatives	0.91

- 1 Gafquat - 755N, ISP
- 2 Celquat H-100, National Starch
- 3 Monamide 716, Mona
- 4 BRIJ 721, ICI
- 5 DC-2-5932 silicone microemulsion from Dow Corning with a particle size of 24 nm, a cationic surfactant system, and a silicone with an internal phase viscosity = 1,200 cps.

The first premix is prepared by adding the ethanol to ½ of the water (save remaining ½ for premixing Silicone Microemulsion, preservatives and surfactants). Next, add the Celquat H-100 and mix until uniformly dissolved in ethanol/water. Add the Gafquat-755N and mix until uniformly dissolved. With vigorous agitation, add the perfume and the glycerin.

The second premix is prepared by dissolving the surfactants (Lauramide DEA, Stareth 21) in the remaining water with heat (120° F) and agitation. Remove the mix from heat when the surfactants are completely dissolved. Next, add the preservatives and the silicone microemulsion to the mix with vigorous agitation.

The final concentrate step involves combining the first and second premixes under vigorous agitation and placing the concentrate in a suitable aluminum can at 7% of total fill with an appropriate mousse valve. The valve is crimped to the can and a vacuum is applied to deaerate. The can is then pressurized with propellant.

<u>Can Composition:</u>	<u>Weight %</u>
Concentrate Composition (above)	93.00
Hydrofluorocarbon 152A (propellant)	4.76
Isobutane (A31 propellant)	2.24



Example V

The following is a mousse composition representative of the present invention:

Concentrate composition:

<u>Component</u>	<u>Weight %</u>
Ethanol (200 proof)	6.00
Butyl Ester of PVM/MA copolymer (50% active) <sup>1</sup>	4.00
PVP/VA (50% active) <sup>2</sup>	2.00
Perfume	0.10
Water	83.19
Aminomethyl propanol <sup>3</sup>	0.10
Polyquaterium - 47 <sup>4</sup>	0.30
Lauramide DEA <sup>5</sup>	0.30
Steareth 21 <sup>6</sup>	0.10
Silicone Microemulsion (25% active) <sup>7</sup>	3.00
Preservatives	0.91

- 1 Gantrez ES-425, ISP
- 2 PVP/VA 735W, ISP
- 3 AMP - regular, Angus
- 4 Merquat - 2001, Calgon
- 5 Monamide 716, Mona
- 6 BRIJ 721, ICI
- 7 DC 2-5791 - Silicone microemulsion from Dow Corning with a particle size of 49 nm, an anionic/nonionic surfactant system, and a silicone with an internal phase viscosity = 80,000 cps.

The first premix is prepared by adding the Gantrez and PVP/VA to ethanol. The premix is then neutralized with AMP under vigorous agitation. Under continued agitation, add ½ of the water (save remaining ½ for premixing the Silicone Microemulsion, preservatives and surfactants) and the perfume.

The second premix is prepared by dissolving the surfactants (Lauramide DEA, Steareth 21) in the remaining water with heat (120° F) and agitation. Remove the mix from heat when the surfactants are completely dissolved. Adjust the pH of the Polyquaterium-47 with AMP to be equal to that of the first neutralized Gantrez premix above. Add the pH adjusted Polyquaterium-47 to the mix with vigorous agitation. Next, add the preservatives and the silicone microemulsion to the mix with vigorous agitation.

The final concentrate step involves combining the first and second premixes under vigorous agitation and placing the concentrate in a suitable aluminum can at 7% of total fill with an appropriate mousse valve. The valve is crimped to the can and a vacuum is applied to deaerate. The can is then pressurized with propellant.

<u>Can Composition:</u>	<u>Weight %</u>
Concentrate Composition (above)	93.00
Hydrofluorocarbon 152A (propellant)	4.76
Isobutane (A31 propellant)	2.24

Example VI

The following is a mousse composition representative of the present invention:

<u>Component</u>	<u>Weight %</u>
Ethanol	10.00
PVP/DMAPA Acrylates Copolymer (10% active) <sup>1</sup>	15.00
Perfume	0.25
Water	63.70
Oleth-20 <sup>2</sup>	0.30
Silicone Microemulsion (25% active) <sup>3</sup>	2.00
Preservatives	0.75
Propellants	8.00

1 Styleze CC-10, ISP - (note that this polymer is cationic)

2 Monamide 716, Mona

3 DC 2-1281 - Dow Corning silicone microemulsion with a cationic surfactant system and with a particle size 25nm and an internal phase viscosity of 1,200 cps. (no dimethicone copolyol)

The first premix involves adding the Styleze CC-10 to ½ the water and the ethanol.

The second premix involves dissolving the surfactant (Oleth-20) in the remaining water with heat (120° F) under vigorous agitation. Remove the mix from heat when surfactant is completely dissolved. Next, add the preservatives, the perfume, and the silicone microemulsion to the mix with vigorous agitation.

The final concentrate step involves combining the first and second premixes under vigorous agitation and placing the concentrate in a suitable aluminum can at 7% of total fill with an appropriate mousse valve. The valve is crimped to the can and a vacuum is applied to deaerate. The can is then pressurized with propellant.

<u>Can Composition:</u>	<u>Weight %</u>
Concentrate Composition (above)	92.00
Hydrofluorocarbon 152A (propellant)	4.00
Isobutane (A31 propellant)	4.00

Example VII

The following is a non-aerosol mousse composition representative of the present invention:

<u>Component</u>	<u>Weight %</u>
Water	90.64
PVP/VA Copolymer (50% active) <sup>1</sup>	3.00
PVP/Dimethylaminoethylmethacrylate Copolymer (20% active) <sup>2</sup>	2.50
Sodium Cocoyl Isethionate <sup>3</sup>	0.75
Isosteareth-10 <sup>4</sup>	0.10
Perfume	0.10
Preservatives	0.91
Silicone Microemulsion (25% active) <sup>5</sup>	2.00

1 PVP/VA 735W, ISP

2 Copolymer 845, ISP

3 Aerosurf 66E10, Witco

- 4 Tauranol I-78, Finetex  
 5 DC 2-1281 - Dow Corning silicone microemulsion with a cationic surfactant system and with a particle size 25nm and an internal phase viscosity Of 1,200 cps. (no dimethicone copolyol)

The first premix is prepared by adding both polymers PVP/VA and Copolymer 845 to half of the water with agitation and then adding the preservatives.

The second premix is prepared by adding the Sodium Cocoyl Isethionate to 40% of the water (heat and vigorous agitation may be required). Combine the Isostearth-10 and perfume and add to the mix with agitation. Next, add the silicone microemulsion, premixed with the remaining (10%) of the water with agitation.

The final concentrate step involves combining the first and second premixes.

#### Example VIII

The following is a Sculpting Spray representative of the present invention:

##### Concentrate composition:

<u>Component</u>	<u>Weight %</u>
Ethanol (200 proof)	50.00
Butyl Ester of PVM/MA copolymer (50% active in ethanol) <sup>1</sup>	12.00
Perfume	0.10
Water	34.64
Aminomethyl propanol <sup>2</sup>	0.26
Silicone Microemulsion (25% active) <sup>3</sup>	3.00

1 Omnirez-2000, ISP

2 AMP-95, Angus

3 DC-1845 - Silicone microemulsion from Dow Corning with a particle size of 33 nm, an anionic/nonionic surfactant system, and a silicone with an internal phase viscosity = 77,000 cps.

This product is prepared by combining the alcohol and water (95% of the water). Begin agitation and add the neutralizer (Aminomethyl propanol). Next, add the Omnirez-2000 and mix until completely dissolved. Then add the fragrance with agitation and finally add the silicone microemulsion premixed with the remaining water.

#### Example IX

The following is a Volumizing Spray representative of the present invention:

##### Concentrate composition:

<u>Component</u>	<u>Weight %</u>
Ethanol (200 proof)	35.00
Octylacrylamide/Acrylates/Butylaminoethyl Methacrylate copolymer <sup>1</sup>	2.50
KOH, (45% active)	0.62
Propylene glycol	0.10
Glycerin	0.10
Perfume	0.10
Water	58.58
Silicone Microemulsion (25% active) <sup>2</sup>	3.00

1 Amphomer LV-71, National Starch

- 2 DC-1845 - Silicone microemulsion from Dow Corning with a particle size of 33 nm, an anionic/nonionic surfactant system, and a silicone with an internal phase viscosity = 77,000 cps.

This product is prepared by combining the alcohol and water (95% of the water). Begin agitating and add the neutralizer (KOH). Next, add the Amphomer LV-71 and mix until completely dissolved. Add the propylene glycol and glycerin. Finally, under agitation, add the fragrance and the silicone microemulsion (premixed with the remaining water).

Example X

The following is a spray gel composition representative of the present invention:

<u>Component</u>	<u>Weight %</u>
Ethanol	40.00
PVP/VA Copolymer (50% active) <sup>1</sup>	6.00
Polyquaterium -11 <sup>2</sup>	0.50
Isosteareth 20 <sup>3</sup>	0.30
Perfume	0.20
Diethylene Glycol <sup>4</sup>	0.30
Water	49.50
Glycerin	0.25
Silicone Microemulsion (25% active) <sup>5</sup>	2.00
Preservatives	0.95

- 1 PVP/VA 735W, ISP
- 2 Gafquat - 755N, ISP
- 3 Arosurf 66 E-20, Witco
- 4 Diethylene glycol, Ashland
- 5 DC 2-1281 - Dow Corning Silicone microemulsion with a particle size of 30 nm, and an internal phase viscosity of 1,500.

This concentrate is prepared by completely dissolving the Polyquaterium-11 and PVP/VA in the ethanol and the batch water with vigorous agitation. Under continued vigorous agitation, add the Diethylene glycol, the Isosteareth-20, the perfume, the Glycerin, and preservatives. Next, add the silicone microemulsion to the mix with mild agitation.

Example XI

The following is a Non-Aerosol Hairspray composition representative of the present invention:

<u>Component</u>	<u>Weight %</u>
Ethanol (200 proof)	55.00
Acrylate Copolymer <sup>1</sup>	4.00
Dimethyl myristalamine	0.10
Isododecane <sup>2</sup>	0.50
Perfume	0.10
Diisobutyl Adipate <sup>3</sup>	0.10
Sodium Hydroxide Solution (30% active)	1.29
Water	34.91
Silicone Microemulsion (25% active) <sup>4</sup>	4.00

- 1 75/25 Tertbutylacrylate/Acrylic Acid/, polymer molecular weight = 80,000
- 2 Permethyl 99A, Presperse Inc.
- 3 Plasthall DIBA, C. P. Hall
- 4 DC-2-5791 - Silicone microemulsion from Dow Corning with a particle size of 49 nm, an anionic/nonionic surfactant system, and a silicone with an internal phase viscosity = 77,000.

This product is prepared by completely dissolving the Acrylate Copolymer in ethanol with vigorous agitation (there should be no visible white flakes). Under continued vigorous agitation, add the dimethyl myristal amine, the Isododecane, and the DIBA. While still agitating, slowly add the NaOH. It is important to achieve a complete neutralization reaction. Next, add the perfume under

agitation. Then add the water and mix until a uniform mixture results. Finally, add the silicone microemulsion to the mix with agitation.

Example XII

The following is an Aerosol Hairspray composition representative of the present invention:

<u>Component</u>	<u>Weight %</u>
Ethanol (200 proof)	40.0
Methacrylate Copolymer (50% active) <sup>1</sup>	5.0
Perfume	0.10
Cyclomethicone	0.30
Aminomethyl propanol (AMP)	0.70
Water	24.72
Silicone Microemulsion (25% active) <sup>2</sup>	1.00
Corrosion Inhibitors	0.18
Dimethyl Ether (propellant)	14.00
Isobutane (propellant)	14.00

<sup>1</sup> Balance CR, National Starch

<sup>2</sup> DC-1845 - Silicone microemulsion from Dow Corning with a particle size of 33 nm, an anionic/nonionic surfactant system, and a silicone with an internal phase viscosity = 70,000 cps.

This concentrate is prepared by adding the AMP to the ethanol with agitation. Adding the methacrylate copolymer to the mix with agitation. Next, add the cyclomethicone and perfume to the mix with agitation. Then, add the corrosion inhibitors to the water and add the water to the batch. The batch concentrate is put into a suitable container, with a suitable valve and is pressurized with the propellants.

Example XIII

The following is an anti-bacterial lotion for hand wipes composition representative of the present invention:

<u>Component</u>	<u>Weight %</u>
Water	75.69
Ethanol	20.00
Ammonium lauryl sulfate	0.60
Triclosan <sup>1</sup>	0.15
Propylene Glycol	0.20
Polyacrylate & Polyacrylate copolymer blend <sup>2</sup>	1.00
Perfume	0.06
Polyquaterium-47 (20% active) <sup>3</sup>	0.50
Hydroxylpropyl Guar <sup>4</sup>	0.50
Silicone Microemulsion (25 % active) <sup>5</sup>	1.00
Preservatives	0.30

<sup>1</sup> Irgasan DP300, CIBA-Geigy

<sup>2</sup> Acusol 445, Rhom & Haas

<sup>3</sup> Merquat-2001, Calgon

<sup>4</sup> Jaguar HP-105, Rhone-Poulenc

<sup>5</sup> DC-1845 - Silicone microemulsion from Dow Corning with a particle size of 33 nm, an anionic/nonionic surfactant system, and a silicone with an internal phase viscosity = 77,000 cps.

A premix is prepared by adding the Polyquaternium-47 to 1/2 of the water under moderate agitation. Next the ammonium lauryl sulfate and Polyacrylate/Polyacrylate copolymer blend are added. Preservatives and silicone microemulsion are then added to the mix.

A second premix is prepared by adding the hydroxypropyl guar to 1/2 of the water, mixing vigorously until the guar hydrates completely. The first and second premixes are then combined.

A third premix is comprised of propylene glycol, Triclosan, ethanol, and perfume. This last ethanol premix is then added to the main mix.

#### Example XIV

The following is a composition for cleansing towellettes representative of the present invention:

<u>Component</u>	<u>Weight %</u>
Water	53.30
Ethanol	45.00
Chitosonium 2-hydroxypropionic acid salt <sup>1</sup>	0.50
Perfume	0.20
Silicone Microemulsion (25% active) <sup>2</sup>	1.00

1 Kytamer L, Amerchol

2 DC-2-8194 - Aminosilicone microemulsion from Dow Corning with a particle size of 30 nm, a cationic surfactant system, and a silicone with an internal phase viscosity = 4000 cps.

The premix is prepared by adding the Chitosonium 2-hydroxypropionic acid salt to 60% of the water (180° F) with vigorous agitation. Then add the silicone microemulsion (premixed with the remaining water) to the mix with mild agitation. Next under continued agitation, add the Ethanol, and the Perfume.

#### Example XV

The following is an Instant Hand Sanitizer gel composition representative of the present invention:

<u>Component</u>	<u>Weight %</u>
Ethanol	50.00
Isopropanol	5.00
Perfume	0.05
Triethanolamine <sup>1</sup>	0.94
Water	41.01
Carbomer <sup>2</sup>	1.00
Silicone Microemulsion (25% active) <sup>3</sup>	2.00

1 Triethanol amine, Dow Chemical

2 Carbopol 940, BF Goodrich

3 DC 2-5791 sp - Silicone microemulsion from Dow Corning with a particle size of 37 nm, an anionic/nonionic surfactant system, and a silicone with an internal phase viscosity = 90,000 cps.

Under vigorous agitation dissolve the Carbomer in the water (120° F) until the carbomer is fully hydrated. With mild agitation, add Triethanol amine to carbomer/water, allowing gelation to occur. With continued mild agitation, add the silicone microemulsion to the mix. Last add the ethanol premixed with the perfume.



## Example XVI

The following is a Makeup Removal gel composition representative of the present invention:

<u>Component</u>	<u>Weight %</u>
Ethanol	30.00
Acrylates Copolymer <sup>1</sup>	5.0
Sodium Hydroxide (1 normal solution)	to pH 7.5
Water	q.s. to 100
Glycerin	0.50
Silicone Microemulsion (25% active) <sup>2</sup>	2.00
Dimethicone Copolyol <sup>3</sup>	0.50

1 Aculyn 33, Rohm and Haas

2 DC-1845 - Silicone microemulsion from Dow Corning with a particle size of 33 nm, an anionic/nonionic surfactant system, and a silicone with an internal phase viscosity = 70,000 cps.

3 DC-190, Dow Corning

Add Acculyn 33 to water. Partially neutralize to pH 6.5 with NaOH. Add glycerin and ethanol. Add remaining NaOH to pH 7.5. Add silicone microemulsion premixed with dimethicone copolyol.

## Example XVII

The following is a clear acne medication composition representative of the present invention:

<u>Component</u>	<u>Weight %</u>
Water	52.50
Ethanol	39.00
Polyquaterium - 16 <sup>1</sup> (20% active)	0.50
Guar Hydroxylpropyl Trimonium Chloride <sup>2</sup>	2.00
Silicone Microemulsion (37 % active) <sup>3</sup>	4.00
Salicylic Acid	2.00

1 Luviquat FC 550, BASF

3 Jaguar C-2000, Rhone-Poulenc

4 DC-2-5932 silicone microemulsion from Dow Corning with a particle size of 24 nm, a cationic surfactant system, and a silicone with an internal phase viscosity = 1,200 cps.

Add the guar to the water, mixing vigorously until the guar hydrates completely. Then add the silicone microemulsion to the mix with mild agitation. Premix the Polyquaterium-16, Salicylic Acid, and ethanol together with agitation. Then add the ethanol premix to the guar mix.

## Example XVIII

The following is a Spray-on Sunscreen composition representative of the present invention:

<u>Component</u>	<u>Weight %</u>
Water	38.45
Acrylates Copolymer <sup>1</sup>	2.00
Acrylates/Stearth-20 Methacrylate Copolymer <sup>2</sup>	2.00
Silicone Microemulsion (25% active) <sup>3</sup>	2.00
Ethanol	30.00

Propylene glycol	1.00
Isopropyl Myristate	5.00
Oleyl Alcohol	1.00
DEA-Cetyl Phosphate <sup>4</sup>	4.00
Benzophenone-3 <sup>5</sup>	6.00
Octyl methoxycinnamate <sup>6</sup>	7.50
Vitamin E Acetate	0.05
Coconut oil	1.00

- 1 Aculyn 33, Rohm and Haas
- 2 Aculyn 22, Rohm and Haas
- 3 DC-1845 - Silicone microemulsion from Dow Corning with a particle size of 33 nm, an anionic/nonionic surfactant system, and a silicone with an internal phase viscosity = 70,000 cps.
- 4 Amphisol, Givaudan-Roure
- 5 supplied by TRI-K
- 6 Neoheliopan AV, Haarmann & Reimer

For premix 1, add Acculyn 33 and Acryln 22 to water and heat to 75° C with agitation. Cool to room temperature, then add silicone microemulsion.

For premix 2, combine remaining ingredients; heat as necessary to solubilize components (this may require heated pressure vessel to prevent evaporation of ethanol). Cool the premix to room temperature and combine with premix A.

#### Example XIX

The following is an Aerosol Shaving Cream composition representative of the present invention:

##### Concentrate composition:

<u>Component</u>	<u>Weight %</u>
Ethanol (200 proof)	4.00
Water	88.59
Polyquaternium 47 (20% active) <sup>1</sup>	2.50
Perfume	0.10
Oleyl Alcohol <sup>2</sup>	0.30
Glycerol Polyacrylate mixture <sup>3</sup>	0.30
Lauramide DEA <sup>4</sup>	0.20
Steareth 21 <sup>5</sup>	0.10
Silicone Microemulsion (25% active) <sup>5</sup>	3.00
Preservatives	0.91

- 1 Merquat 2001, Calgon
- 2 NOVOL, Shinnihon RIKA
- 3 Lubrajel Oil, ISP
- 4 Monamide 716, Mona
- 5 BRIJ 721, ICI
- 6 DC-2-8194 - Aminosilicone microemulsion from Dow Corning with a particle size of 25 nm, a cationic surfactant system, and a silicone with an internal phase viscosity = 4,000 cps.

The first premix is prepared by adding the ethanol to ½ of the water (save remaining ½ for premixing Silicone Microemulsion, preservatives and surfactants). Next, add the Polyquaternium 47 and mix until the Polyquaternium 47 is uniformly dissolved in ethanol/water. Then add the Lubrajel Oil. With vigorous agitation, add the perfume and oleyl alcohol.

The second premix is prepared by dissolving the surfactants (Lauramide DEA, Steareth 21) in the remaining water with heat (120° F) and agitation. Remove the mix from heat when the surfactants are completely dissolved. Next, add the preservatives and the silicone microemulsion to the mix with vigorous agitation.

The final concentrate step involves combining the first and second premixes under vigorous agitation and placing the concentrate in a suitable aluminum can at 7% of total fill with an appropriate mousse valve. The valve is crimped to the can and a vacuum is applied to deaerate. The can is then pressurized with propellant.

<u>Can Composition:</u>	<u>Weight %</u>
Concentrate Composition (above)	93.00
Hydrofluorocarbon 152A (propellant)	4.76
Isobutane (A31 propellant)	2.24

#### Example XX

The following is a clear 2-in-1 Shampoo composition representative of the present invention:

#### Concentrate composition:

<u>Component</u>	<u>Weight %</u>
Ethanol (200 proof)	8.00
Polyquaternium 47 (20% active) <sup>1</sup>	5.00
Dimethyl laurylamine <sup>2</sup>	1.20
Isododecane <sup>3</sup>	0.50
Perfume	0.10
Water	64.04
Ammonium lauryl sulfate	15.00
Lauramide DEA <sup>5</sup>	2.00
Steareth 21 <sup>6</sup>	0.25
Silicone Microemulsion (25% active) <sup>7</sup>	3.00
Preservatives	0.91

- 1 Merquat 2001, Calgon
- 2 C12 Alkyl Dimethyl Amine, AT-1295 LT
- 3 Permethyl 99A, Presperse Inc.
- 4 Monamide 716, Mona
- 5 BRIJ 721, ICI
- 6 DC 2-5791 - Silicone microemulsion from Dow Corning with a particle size of 33 nm, an anionic/nonionic surfactant system, and a silicone with an internal phase viscosity = 80,000 cps.

The first premix is prepared by adding the ethanol to the Polyquaternium 47 and mix until the Polyquaternium 47 is uniformly dissolved in ethanol/water. With vigorous agitation, add Dimethyl lauryl amine, the perfume and the Isododecane. The second premix is prepared by dissolving the surfactants (Lauramide DEA, Steareth 21) in the remaining water with heat (120° F) and agitation. Then add the ammonium lauryl sulfate. Remove the mix from heat when the surfactants are completely dissolved. Next, add the preservatives and the silicone microemulsion to the mix with vigorous agitation. The final concentrate step involves combining the first and second premixes with agitation

#### Example XXI

The following is a shampoo composition for a non-aerosol foamer package representative of the present invention:

Component	Weight %
Water	83.39
Ethanol	6.00
Polyquaterium-10 <sup>1</sup> (20% ACTIVE)	2.50
Lauramide DEA <sup>2</sup>	3.00
Sodium Cocoyl Isethionate <sup>3</sup>	3.00
Isosteareth-10 <sup>4</sup>	0.10
Perfume	0.10
Preservatives	0.91
Silicone Microemulsion (25% active) <sup>5</sup>	1.00

<sup>1</sup> Ucare JR-400, Amerchol

<sup>2</sup> Monamide 716, Mona

<sup>3</sup> Aerosurf 66E10, Witco

<sup>4</sup> Tauranol I-78, Finetex

<sup>5</sup> DC 2-1281 - Dow Corning silicone microemulsion with a cationic surfactant system and with a particle size 25nm and an internal phase viscosity of 1,200 cps. (no dimethicone copolyol)

The first premix is prepared by adding the Polyquaternium-11 to half of the water with agitation and then adding the preservatives. The second premix is prepared by adding the Sodium Cocoyl Isethionate and Lauryamide DEA to the remaining water (heat and vigorous agitation may be required). Combine the Isosteareth-10 and perfume and add to the mix with agitation. Next, add the silicone microemulsion with mild agitation. The final concentrate step involves combining the first and second premixes.

## WHAT IS CLAIMED IS:

1. A personal care composition suitable for styling hair, characterized in that it comprises:
  - a) from 0.01% to 20% of a non-silicone containing polymer suitable for hair styling;
  - b) an organopolysiloxane microemulsion comprising:
    - (i) a organopolysiloxane dispersed as particles in the microemulsion, wherein the organopolysiloxane is substantially free of amino groups in combination with hydroxyl groups and has an average particle size of less than 80 nanometers, and
    - (ii) a surfactant system for dispersing the organopolysiloxane in the microemulsion, which is compatible with the hair styling polymer;wherein the amount of microemulsion is such that the personal care composition comprises from 0.01% to 10% of the organopolysiloxane; and
  - c) a carrier comprising:
    - (i) from 3% to 99%, by weight of the composition, of a first solvent selected from water; water soluble organic solvents; organic solvents which are strongly to moderately strong in hydrogen-bonding parameter; and mixtures thereof; wherein the first solvent is other than C<sub>1</sub>-C<sub>3</sub> monohydric alcohol, C<sub>1</sub>-C<sub>3</sub> ketone and C<sub>1</sub>-C<sub>3</sub> ether, and
    - (ii) optionally, from 0% to 55% of a second solvent selected from C<sub>1</sub>-C<sub>3</sub> monohydric alcohols, C<sub>1</sub>-C<sub>3</sub> ketones, C<sub>1</sub>-C<sub>3</sub> ethers, and mixtures thereof.
2. The composition of Claim 1 wherein the hair styling polymer is a nonionic polymer selected from polyvinylpyrrolidone, copolymers of polyvinylpyrrolidone and vinyl acetate, and mixtures thereof.
3. The composition of any of the preceding claims wherein the hair styling polymer is an anionic acrylate polymer.
4. The composition of any of the preceding claims wherein the hair styling polymer is a cationic polymer comprising a nitrogen atom.

5. The composition of any of the preceding claims wherein the organopolysiloxane microemulsion is selected from the group comprising DC-2-5791, DC-2-1845, DC-2-5932, DC-2-8194, DC-2-1281, and mixtures thereof.
6. The composition of any of the preceding claims wherein the organopolysiloxane microemulsion is selected from the group comprising DC-2-5791, DC-2-5932 and mixtures thereof.
7. The composition of any of the preceding claims wherein the first solvent is water and the composition comprises comprises 40% or less of C<sub>1</sub>-C<sub>3</sub> monohydric alcohols, preferably ethanol, n-propanol, isopropanol, and mixtures thereof.
8. The composition of any of the preceding claims wherein the composition comprises from 40% to 55% of said second solvent, and further comprises a silicone polyether surfactant.
9. The composition of any of the preceding claims wherein the organopolysiloxane has an average particle size of less than 60 nanometers.
10. The composition of any of the preceding claims wherein the organopolysiloxane has an average particle size of less than 40 nanometers.

# INTERNATIONAL SEARCH REPORT

International Application No

PCT/IB 98/01749

**A. CLASSIFICATION OF SUBJECT MATTER**  
 IPC 6 A61K7/06 A61K7/11

According to International Patent Classification (IPC) or to both national classification and IPC

**B. FIELDS SEARCHED**

Minimum documentation searched (classification system followed by classification symbols)

IPC 6 A61K

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

**C. DOCUMENTS CONSIDERED TO BE RELEVANT**

Category	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	EP 0 468 721 A (UNILEVER) 29 January 1992 see claims 1-5 see page 4, line 30-42 ---	1, 4, 9, 10
P, X	WO 98 19654 A (PROCTER & GAMBLE) 14 May 1998 see claims 1, 3, 4 see page 11, line 25 - page 12, line 29 see page 24, line 1-13 -----	1-4, 7

Further documents are listed in the continuation of box C.

Patent family members are listed in annex.

\* Special categories of cited documents :

- "A" document defining the general state of the art which is not considered to be of particular relevance
- "E" earlier document but published on or after the international filing date
- "L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)
- "O" document referring to an oral disclosure, use, exhibition or other means
- "P" document published prior to the international filing date but later than the priority date claimed

- "T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention
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Information on patent family members

Inte. onal Application No

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