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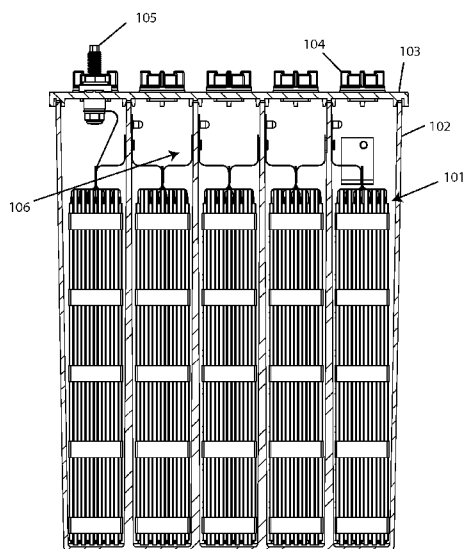


FIG. 1

(57) Abstract: A Nickel iron battery using a mono-block housing that has cells with leak-proof intercell connections. The intercell connections use compressed grommets to prevent fluid transfer between the cells. Each cell generates a charge that can be connected in series or parallel as required. The cells use an electrode plate that enables a manufacturing process which yields higher efficiency, higher throughput, and significantly lower battery cost. The anode composition of the cells has iron powder to increase cathode utilization, therefore lowering cathode material usage and lowering production cost.



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NICKEL IRON BATTERY

CROSS-REFERENCE TO RELATED APPLICATIONS

[0001] This application claims the priority benefit of U.S. Provisional Utility Application 61/837209 filed June 20, 2013, and U.S. Non-Provisional Utility Application 14/308441 filed June 18, 2014, both of which are herein incorporated by reference in their entirety.

TECHNICAL FIELD

[0002] This disclosure is related to batteries. More particularly, the disclosure discusses a Nickel Iron mono-block battery.

BACKGROUND ART

[0003] The statements in this section merely provide background information related to the present disclosure. Accordingly, such statements are not intended to constitute an admission of prior art.

[0004] Nickel Iron batteries are robust, tolerant of abuse, and safe for large scale applications. They are known for having very long calendar and cycle life in comparison to other batteries. Nickel Iron batteries perform well in cyclic applications as the main backup power source for renewable energy and traction power for forklift and locomotive propulsion because of the exceptional tolerance to electrical abuse in higher voltage multiple cell usage against deep discharge and overcharge in comparison to any other battery chemistries.

BRIEF SUMMARY OF THE INVENTION

[0005] In one embodiment, the nickel iron battery of the present disclosure comprises: a housing, wherein the housing has partitions that divide the housing into one or more cells, wherein the cell(s) are connected using intercell connections, further wherein each cell

comprises: a plurality of positive electrode plates, wherein each positive electrode plate is comprised of a multi-cavity current collector which encases multiple positive electrode pellets and a positive current collector tab; a plurality of negative electrode plates, wherein each negative electrode plate is comprised of a multi-cavity current collector which encases multiple negative electrode pellets and a negative current collector tab; a separator disposed between the positive electrode plates and the negative electrode plates; and a plurality of bands configured to contain the cell in a unified assembly; and an electrolyte disposed in said housing.

[0006] The housing can hold the other battery elements as cited above. The housing material of construction can be plastic, plastic coated metal, or any other material compatible with an alkaline electrolyte.

[0007] The intercell connections can have a material of construction that is electrically conductive such as steel, nickel, nickel plated steel or the like. The intercell connections can have a fastening method that uses metal crimping, a rivet, nut and bolt, weld, or the like.

[0008] The positive electrode plates can have a material of construction that is nickel plated steel or the like. They can contain the one or more pellets.

[0009] The multi-cavity current collectors can have a material of construction that is electrically conductive such as steel, nickel, nickel plated steel or the like.

[0010] The positive electrode pellets can have a material of construction that is nickel hydroxide, manganese oxides, or the like.

[0011] The positive current collector can have a material of construction that is nickel, nickel plated steel, or the like.

[0012] The negative electrode pellets can have a material of construction that is Iron Oxide or the like.

[0013] The negative current collector can have a material of construction that is nickel, nickel plated steel, or the like.

[0014] The separator can have a material of construction that is woven or non-woven nylon, polypropylene, or any other material compatible with an alkaline electrolyte.

[0015] The bands can have a material of construction such as polypropylene plastic or the like.

[0016] The electrolyte can be a solution containing potassium hydroxide, sodium hydroxide, lithium hydroxide, or the like.

[0017] The scope of the invention is defined by the claims, which are incorporated into this section by reference. A more complete understanding of embodiments on the present disclosure will be afforded to those skilled in the art, as well as the realization of additional advantages thereof, by consideration of the following detailed description of one or more embodiments. Reference will be made to the appended sheets of drawings that will first be described briefly.

BRIEF DESCRIPTION OF THE DRAWINGS

[0018] A clear understanding of the key features of the invention summarized above may be had by reference to the appended drawings, which illustrate the method and system of the invention, although it will be understood that such drawings depict preferred embodiments of the invention and, therefore, are not to be considered as limiting its scope with regard to other embodiments which the invention is capable of contemplating. Accordingly:

[0019] FIG. 1 is a cross-section view of a mono-block Nickel Iron battery housing design with internal inter-cell connections.

[0020] FIG. 2A shows a cross-section view of a crimped intercell connection.

[0021] FIG. 2B shows a magnified cross-section view of a crimped intercell connection.

- [0022] FIG. 2C shows an exploded cross-section view of a crimped intercell connection.
- [0023] FIG. 3 shows an electrode plate with perforated and unperforated areas.
- [0024] FIG. 4A shows an exploded perspective view of an electrode plate.
- [0025] FIG. 4B shows a side cross-section view of an electrode plate.
- [0026] FIG. 4C shows a front view of an electrode plate.
- [0027] FIG. 5A shows a cross-section view of a perforated steel sheet single-sided louver embodiment.
- [0028] FIG. 5B shows a perspective view of a perforated steel sheet single-sided louver embodiment.
- [0029] FIG. 5C shows a cross-section view of a perforated steel sheet double-sided louver embodiment.
- [0030] FIG. 5D shows a perspective view of a perforated steel sheet double-sided louver embodiment.
- [0031] FIG. 6 shows a perspective view of a cell stack.
- [0032] FIG. 7A shows a cross-section view of a cell stack.
- [0033] FIG. 7B shows a cross-section-view of a z-fold separator.
- [0034] FIG. 8 shows an alternate cell stack embodiment.
- [0035] FIG. 9A shows a cross-section view of an alternate cell stack embodiment.
- [0036] FIG. 9B shows a cross-section-view of an alternate z-fold separator embodiment.
- [0037] FIG. 10A shows anode performance without an iron powder precharge.
- [0038] FIG. 10B shows anode performance with an iron powder precharge.
- [0039] FIG. 11 shows a comparison of nickel iron cell performance both with and without iron powder added.

DETAILED DESCRIPTION

[0040] The following detailed description of the invention is merely exemplary in nature and is not intended to limit the invention or the application and uses of the invention.

Furthermore, there is no intention to be bound by any theory presented in the preceding background of the invention or the following detailed description of the invention.

[0041] Despite numerous performance advantages over other battery technologies, Nickel Iron batteries have very complex plate manufacturing processes and therefore very high cost to produce.

[0042] The present disclosure describes a novel and non-obvious nickel iron battery which has overcome the manufacturing limitations described above, while utilizing the cited advantages. The disclosed invention for Nickel Iron battery has improvements over the current state of the art in three distinct areas. These are: 1) a high voltage Nickel Iron mono-block design comprising a leak-proof inter-cell connection; 2) an electrode plate design approach and manufacturing process to yield higher efficiency, higher throughput, and significantly lower battery cost; and 3) an anode composition which has iron powder to increase cathode utilization and therefore lower cathode material usage and lower production costs.

[0043] A current state of the art nickel iron battery comprises: a cathode; an anode; a separator; an electrolyte; and a housing.

[0044] The present disclosure provides a mono-block housing design with internal inter-cell connections for the Nickel Iron battery as shown in FIG. 1. The mono-block design comprises leak-proof inter-cell connections within the battery.

[0045] In the mono-block design battery voltage can be scaled from 1V to 6V to 12V to 48V or even higher as needed for the power requirement. The advantages of mono-block Nickel Iron include higher battery energy density, lower material usage, and therefore lower manufacturing cost as shown in TABLE 1 below.

TABLE 1

	State of the Art 12V 150 Ah Nickel Iron Battery	The present invention 12V 150 Ah Nickel Iron battery
Battery Voltage V	12V	12V
Battery Capacity Ah	150 Ah	150 Ah
Battery Length cm	53 cm	19.2 cm
Battery Width cm	83 cm	23 cm
Battery Height cm	34.5 cm	36 cm
Battery Weight kg	90 kg	34 kg
Battery Volume	150 Liter	16 Liter
Specific Energy Wh/kg	20 Wh/kg	53 Wh/kg
Energy density Wh/L	12 Wh/L	112 Wh/L
Projected cost	>400\$/kWh	<100\$/kWh

[0046] A reliable mono-block Nickel Iron battery requires leak-proof inter-cell connections. Alkaline electrolyte has tendency to creep along conductive metal surfaces. The present mono-block design invention discloses a leak-proof inter-cell connection as shown in FIG. 2A, 2B, and 2C.

[0047] This present disclosure provides a low cost, permanently leak proof connection between cells in a mono-block battery. This is achieved through the compression of a plastic grommet between the positive tab and the negative tab, causing the grommet to conform to the volume of the hole in the intercell partition. The capturing and compression of the grommet material within the intercell partition hole provides a hydrostatic pressure in the grommet which permanently seals the pass through of liquid electrolyte (i.e. does not allow

liquid through the partition hole). This design also provides for a long term leak proof connection due to the full capture of the grommet not allowing for long term plastic creep that would reduce the hydrostatic pressure on the grommet and the loss of sealing between the grommet and the positive tab protrusion and/or the intercell partition.

[0048] The positive tab has a formed protrusion that has a cylindrical portion enclosed by a spherically radial portion. Upon crimping, the spherical portion is flattened to provide a “head” similar to a rivet head. The “head” provides support on the anode tab holding the assembly in the compressed state. The material of the positive tab in this embodiment is cold rolled steel with nickel plating. Alternate materials which could be used include stainless steel and nickel or any other materials compatible with the chemistry of interest.

[0049] The grommet in this embodiment has a flange but, could be made without the flange and still provide the same sealing characteristics. The material in this embodiment is PTFE. Alternate materials would include Polypropylene, Polyethylene, Polyamide or other polymers that are suitable for the application.

[0050] The negative tab has a hole sized to accept the positive tab formed protrusion and retain the “head” of the formed positive tab protrusion. In this embodiment the negative tab has a cylindrical portion enclosed by a spherically radial portion. This feature is used in the assembly process to aid in the orientation of the negative tab. Alternate embodiments may eliminate this feature or change the type of feature.

[0051] This present disclosure provides a leak-proof permanently sealed intercell connection by using the positive and negative tabs themselves. No additional sealing post is required for simpler assembly at the lowest possible cost.

[0052] Prior art designs for nickel iron pocket plate battery carry very high manufacturing costs due to a complex manufacturing process for the making of electrode plates. In the prior art of the pocket plate making process individual strips of steel are formed, punched, or

perforated and then linked together into an electrode plate. This is a complex and costly process to manufacture the plates.

[0053] In contrast, the Nickel Iron battery of the present disclosure provides a high efficiency and high throughput plate manufacturing process by forming the complete electrode plate with specified shapes and dimensions and electrode pockets in a single component. As shown in FIG. 3, this battery makes use of a steel sheet conforming to the size of the electrode plate. Perforation and pockets can be done by hydraulic or mechanical presses on multiple sheets. Both sheet forming and the perforation operations with specified shapes and dimensions can be done at the same time. An alternative approach can make use of already perforated sheet(s) with equivalent results. Already perforated steel sheet(s) may include expanded metal made by expanded metal sheer press or rotary forming die. Although FIG. 3 depicts unperforated sections **302** between perforated sections **301** comprised of electrode pockets **402**, an alternate embodiment having a continuous perforation pattern is also acceptable.

[0054] The electrode plate is comprised of a plurality of electrode pellets and perforated steel sheets as current collectors. Two (2) individual perforated steel plate components with a plurality of electrode pockets **402** and electrode pellets **404** shown in FIG. 4A and FIG. 4B can provide a robust plate making process with high efficiency, high through-put, and very cost effective assembly. The structure can be joined together by several manufacturing methods including clinching, resistance welding, or the like. Current collector side rails **405** and a current collector tab **403** are added as the final step. The current collectors **401**, current collector side rails **405**, and current collector tab **403** are joined together in this embodiment by resistance welding. In another embodiment, the joining processes can also be achieved by a mechanical clinching process.

[0055] The assembly of both positive and negative electrode plates is similar with the exception of electrode pellet chemical makeup and perforations in the current collector material. In the embodiment shown in FIG. 4A, FIG. 4B, and FIG. 4C, the plurality of electrode pellets **404** are contained in the electrode pockets **402** formed in the current collectors **401**.

[0056] In one embodiment, the perforation opening dimensions for the cathode plate current collector are 40 to 60 microns tall by 0.30 to 0.50 mm wide, but the acceptable range for these dimensions is 20 to 200 microns tall by 0.10 to 4.00 mm wide.

[0057] In one embodiment, the open area percentage, defined as the total included area of the openings per cavity face divided by the surface area of the major face of the cathode pellet, is to be between 5 to 7%, but the acceptable range is 2 to 20%.

[0058] In one embodiment, the perforation opening dimensions for the anode plate current collector are 120 to 150 microns tall by 1.00 to 2.00 mms wide, but the acceptable range for these dimensions is 20 to 200 microns tall by 0.10 to 4.00 mm wide.

[0059] In one embodiment, the open area percentage, defined as the total included area of the openings per cavity face divided by the surface area of the major face of the anode pellet, is to be between 5 to 7%, but the acceptable range is 2 to 20%.

[0060] The cell stack **600** is comprised of a plurality of electrode plates **602**, separator **601**, conductive tabs **604**, and compressive bands **603**. In this embodiment, it is arranged with separator **601** material interwoven between the positive electrode plates **606** and negative electrode plates **605**. For this embodiment, FIG. 6, 7A, and 7B show the separator material to be made of non-woven spun-bonded nylon material, but alternate materials such as polypropylene, polyethylene, and any other materials compatible with the chemistry of interest could be used.

[0061] Alternate cell stack embodiment **800** FIG. 8, 9A, and 9B show the use of a polymer frame **803** without the use of the separator material to provide the needed insulation between the electrode plates **602** while allowing the needed electrolyte into the void space between the electrode plates **602**. The material in the alternate embodiment is polypropylene, but alternate materials such as polyamide, polyethylene, and any other materials compatible with the chemistry of interest could be used. In addition, a plurality of bands **603** is used to constrain the alternate cell stack embodiment **800** by providing compression to the alternate cell stack embodiment **800** that is necessary for effective electrical and ionic conductivity. In this embodiment **800**, the bands **603** are made from polypropylene, but alternate materials such as rubber, polyethylene, and any other materials compatible with the chemistry of interest could be used.

[0062] In one embodiment, the cell stack is comprised of 13 electrode plates with 7 negative electrode plates and 6 positive electrode plates. The total number of electrode plates is dependent upon the intended amp hour capacity of the cell. Alternate embodiments may have any number of electrode plates, depending upon the required capacity of the cell.

[0063] The discharge efficiency of the iron anode is not as good as the discharge efficiency of the nickel cathode. The faradic utilization of iron oxide anode is about 30% of theoretical number where as the faradic utilization of nickel hydroxide cathode is about 100%. During first formation cycle a significant amount of expensive nickel hydroxide is wasted because the battery discharge capacity is limited by the iron oxide anode. FIG. 10 depicts the cathode discharge depth without and with the added iron powder. Cell reversible capacity is higher with added iron powder.

[0064] The preferred amount of iron powder used in the iron oxide anode is between 5% and 35%. But more preferably it is between 10% and 20%. The preferred embodiment of the iron oxide anode with iron powder may include:

Magnetite Iron oxide powder 50%-87.5%

Sulfur 0.5%-1%

Graphite conductor 5%-10%

Binder 2%-4%

Iron powder 5%-35%

[0065] The iron powder is electrochemically active and therefore can provide the discharge capacity to make up the efficiency loss in the iron oxide anode, so that the cathode capacity can be fully utilized. The preferred particle size of iron powder for good discharge efficiency is 0.5-5 microns, but more preferably between 1-2 microns. With this approach, the battery energy cost is lower with no under-utilized nickel hydroxide. FIG. 11 shows the cell performance with and without the added iron powder. The iron powders in the present disclosure can include atomized iron powder, carbonyl iron powder, iron powder-carbon composite synthesized by reducing iron compound from solution phase onto dispersed carbon powder, or the like. These types of iron powders can be produced with very high purity which can result in lower gassing during charge and therefore higher cycle efficiency.

[0066] FIG. 1 is a cross-section view of a mono-block Nickel Iron battery housing design with internal inter-cell connections. Shown are cell stacks **101**, battery container **102**, battery lid **103**, vent caps **104**, battery terminal **105**, and inter-cell connections **106**. The intercell connections **106** are configured to prevent fluid flow between the cell stacks **101**.

[0067] FIG. 2A shows a cross-section view of a crimped intercell connection **200**. Shown are an intercell partition **201**, negative tab **202**, grommet **203**, positive tab **204**, and crimped protrusion **205**. The grommet **203**, is compressed and configured to prevent fluid flow through the crimped intercell connection **200**.

[0068] FIG. 2B shows a magnified cross-section view of a crimped intercell connection **200**. Shown are an intercell partition **201**, negative tab **202**, grommet **203**, positive tab **204**,

and crimped protrusion **205**. The grommet **203**, is compressed and configured to prevent fluid flow through the crimped intercell connection **200**.

[0069] FIG. 2C shows an exploded cross-section view of a crimped intercell connection **200**. Shown are an intercell partition **201**, negative tab **202**, grommet **203**, and positive tab **204**.

[0070] FIG. 3 shows an electrode plate with perforated areas **301** and unperforated areas **302**. Also shown are electrode pockets **402** in the perforated areas **301**.

[0071] FIG. 4A shows an exploded perspective view of an electrode plate. Shown are current collectors **401**, electrode pockets **402**, current collector tab **403**, electrode pellets **404**, and current collector side rails **405**.

[0072] FIG. 4B shows a side cross-section view of an electrode plate. Shown are current collectors **401**, current collector tab **403**, electrode pellets **404**, and current collector side rail **405**.

[0073] FIG. 4C shows a front view of an electrode plate. Shown are joining locations **406** for the electrode plates. The electrode plates are paired and mirrored so that pockets **402** are formed in-between the joining locations **406**. The pockets hold the electrode pellets **404**.

[0074] FIG. 5A shows a cross-section view of a perforated steel sheet single-sided louver embodiment. Shown are single-side louvers **501**. The single-sided louvers **501** can have a half-conical shape with a single opening.

[0075] FIG. 5B shows a perspective view of a perforated steel sheet single-sided louver embodiment. Shown are single-side louvers **501**. The single-sided louvers **501** can have a half-conical shape with a single opening.

[0076] FIG. 5C shows a cross-section view of a perforated steel sheet double-sided louver embodiment. Shown are double-sided louvers **502**. The double-sided louvers **502** can have a trapezoidal shape with two openings.

[0077] FIG. 5D shows a perspective view of a perforated steel sheet double-sided louver embodiment. Shown are double-sided louvers **502**. The double-sided louvers **502** can have a trapezoidal shape with two openings.

[0078] FIG. 6 shows a perspective view of a cell stack **600**. Shown are separators **601**, electrode plates **602**, bands **603**, and conductive tabs **604**.

[0079] FIG. 7A shows a cross-section view of a cell stack. Shown are separators **601**, negative electrode plates **605**, positive electrode plates **606**, and bands **603**.

[0080] FIG. 7B shows a cross-section-view of a z-fold separator. Shown are negative electrode plates **605**, positive electrode plates **606**, and z-fold separator **701**.

[0081] FIG. 8 shows an alternate cell stack embodiment **800**. Shown are polymer frame **803**, electrode plates **602**, bands **603**, and conductive tabs **604**.

[0082] FIG. 9A shows a cross-section view of an alternate cell stack embodiment. Shown are polymer frame **803**, negative electrode plates **605**, positive electrode plates **606**, and bands **603**.

[0083] FIG. 9B shows a cross-section-view of an alternate z-fold separator embodiment. Shown are negative electrode plates **605**, positive electrode plates **606**, and polymer frame **803**.

[0084] FIG. 10A shows anode performance without an iron powder precharge.

[0085] FIG. 10B shows anode performance with an iron powder precharge.

[0086] FIG. 11 shows a comparison of nickel iron cell performance both with and without iron powder added.

[0087] All patents and publications mentioned in the prior art are indicative of the levels of those skilled in the art to which the invention pertains. All patents and publications are herein incorporated by reference to the same extent as if each individual publication was

specifically and individually indicated to be incorporated by reference, to the extent that they do not conflict with this disclosure.

[0088] While the present invention has been described with reference to exemplary embodiments, it will be readily apparent to those skilled in the art that the invention is not limited to the disclosed or illustrated embodiments but, on the contrary, is intended to cover numerous other modifications, substitutions, variations, and broad equivalent arrangements.

CLAIMS

We claim,

1. A nickel-iron battery, the battery comprising:
 - a housing, wherein the housing has partitions that divide the housing into one or more cells, wherein the cell(s) are connected using intercell connections, further wherein each cell comprises:
 - a plurality of positive electrode plates, wherein each positive electrode plate is comprised of a multi-cavity current collector which encases one or more positive electrode pellets and a positive current collector tab;
 - a plurality of negative electrode plates, wherein each negative electrode plate is comprised of a multi-cavity current collector which encases one or more negative electrode pellets and a negative current collector tab;
 - a separator disposed between the positive electrode plates and the negative electrode plates; and
 - a plurality of bands configured to contain the cell in a unified assembly; and
 - an electrolyte disposed in said housing.
2. The battery of claim 1, wherein the negative electrode pellets comprise an iron oxide anode consisting of iron powder between 5 and 50 percent by weight.
3. The battery of claim 1, wherein the negative electrode pellets comprise an iron oxide anode consisting of iron powder between 10 and 20 percent by weight.
4. The battery of claim 1, further comprising a positive extension tab and a negative extension tab, wherein the positive extension tab is connected to the positive current collector tabs and the negative extension tab is connected to the negative current collector tabs.

LmarkNiFePCT

5. The battery of claim **4**, wherein the intercell connections are achieved using a compressed plastic grommet to seal between extension tabs and an intercell partition that exists between adjacent cells.
6. The battery of claim **4**, wherein the positive extension tab has a prescribed protrusion that has a cylindrical portion enclosed by a spherically radial portion, wherein an outside diameter of the cylindrical portion is between 2 and 30 mm and the length of said cylindrical portion is between 100 and 300 % of a thickness of the intercell partition.
7. The battery of claim **5**, wherein the compressed grommet seal is achieved by forming the positive extension tab protrusion into a formed head, wherein the diameter of the head is between 105 and 200 % of the cylindrical portion outside diameter.
8. The battery of claim **1**, wherein each positive electrode plate current collector has perforations in the shape of a single-sided louver and the perforations are between 20 to 200 microns tall and 0.1 to 4.0 mm wide.
9. The battery of claim **1**, wherein each positive electrode plate current collector has perforations in the shape of a double-sided louver and the perforations are between 20 to 200 microns tall and 0.1 to 4.0 mm wide.
10. The battery of claim **1**, wherein each positive electrode plate current collector has an open area between 2 and 20 %.
11. The battery of claim **1**, wherein each positive electrode current collector has unperforated areas.
12. The battery of claim **1**, wherein each positive electrode current collector has continuous perforations.

LmarkNiFePCT

13. The battery of claim **1**, wherein each negative electrode plate current collector has perforations in the shape of a single-sided louver and the perforations are between 20 to 200 microns tall and 0.10 to 4.0 mm wide.
14. The battery of claim **1**, wherein each negative electrode plate current collector has perforations in the shape of a double-sided louver and the perforations are between 20 to 200 microns tall and 0.10 to 4.0 mm wide.
15. The battery of claim **1**, wherein each negative electrode plate current collector has an open area between 2 and 20 %.
16. The battery of claim **1**, wherein each negative electrode current collector has unperforated areas.
17. The battery of claim **1**, wherein each negative electrode current collector has continuous perforations.

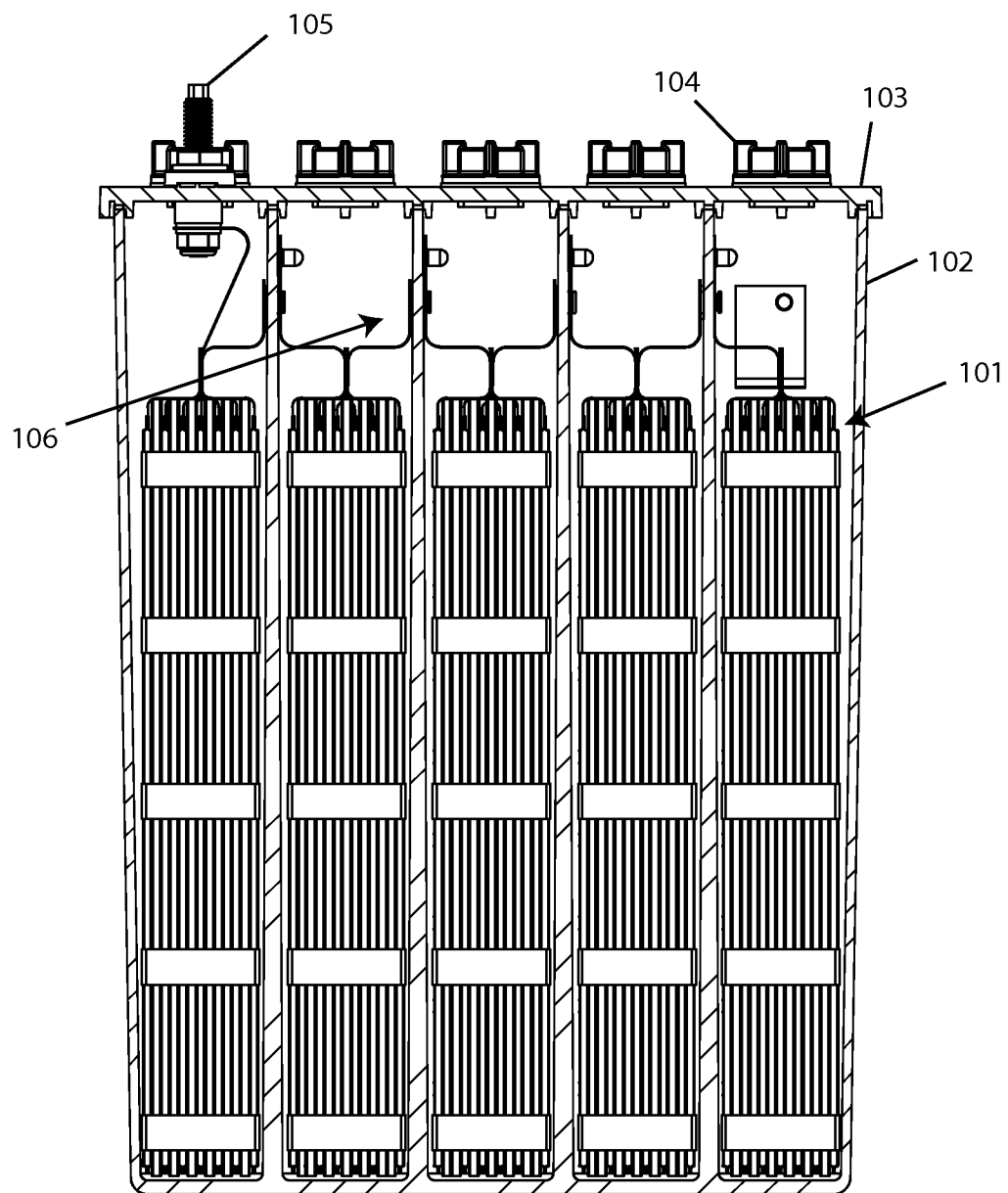


FIG. 1

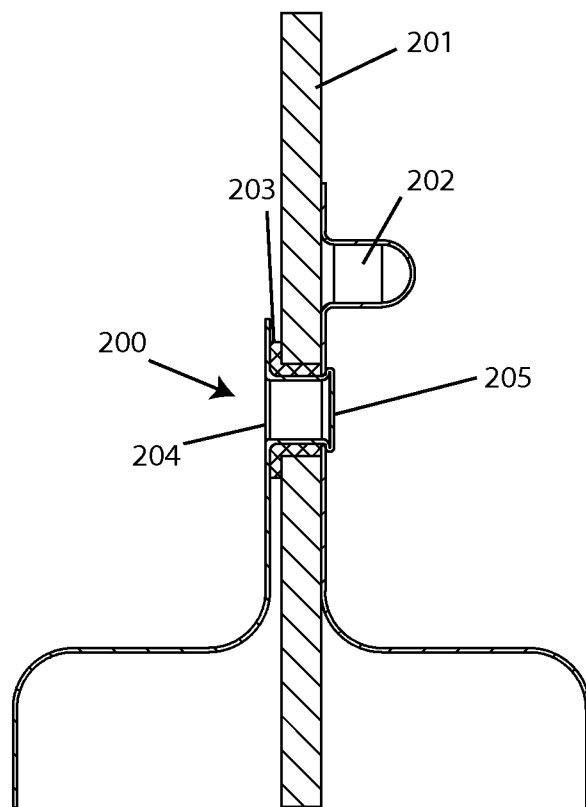


FIG. 2A

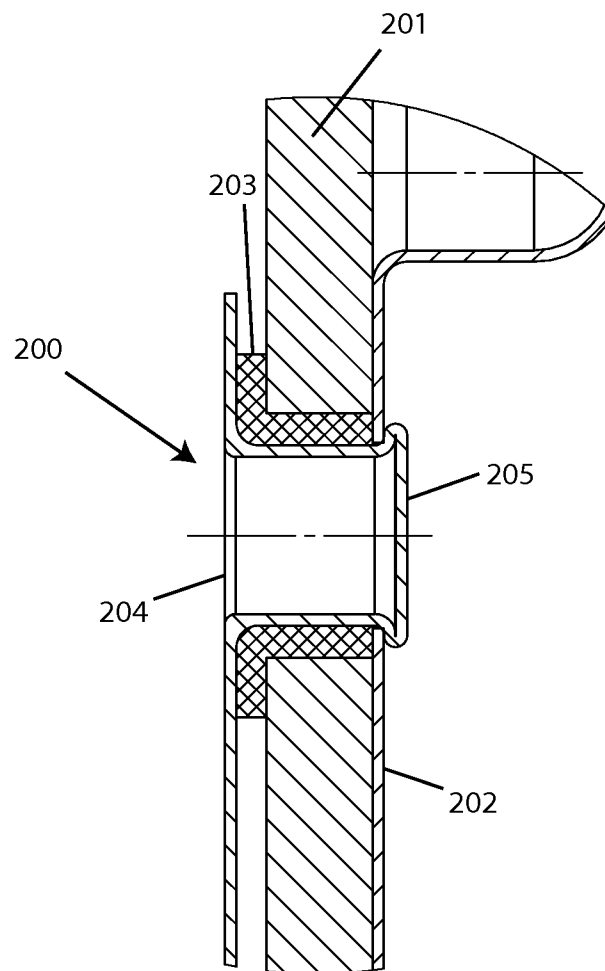


FIG. 2B

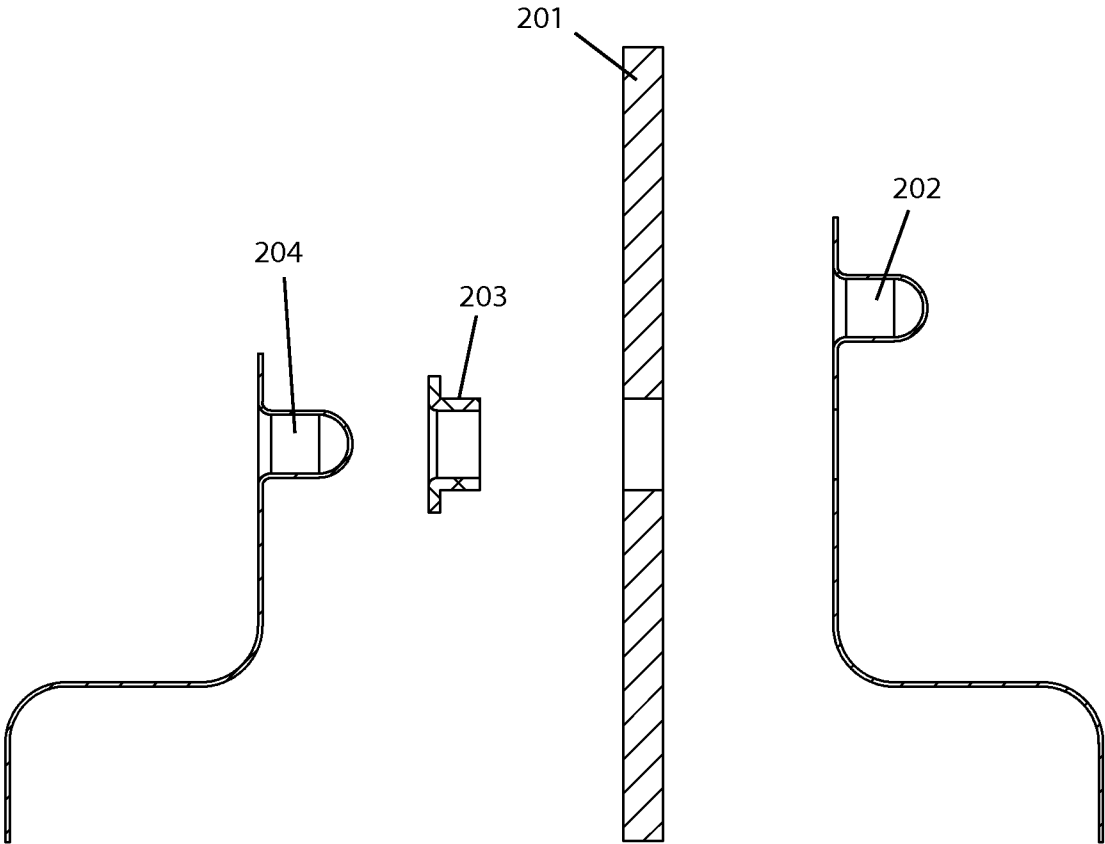


FIG. 2C

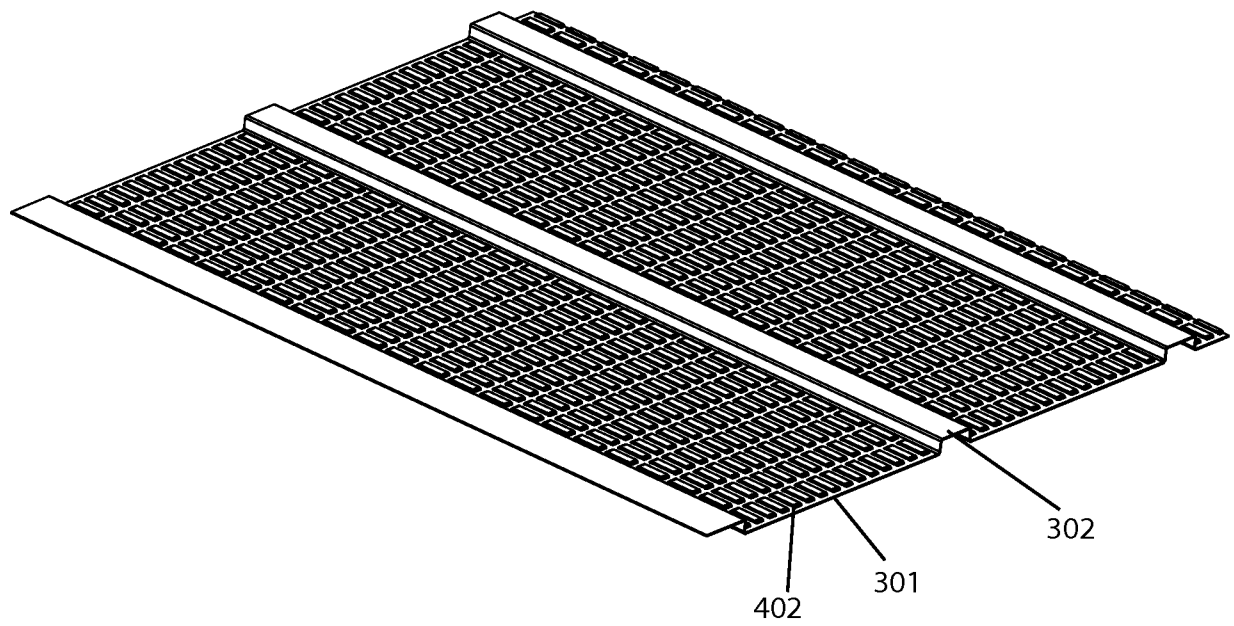


FIG. 3

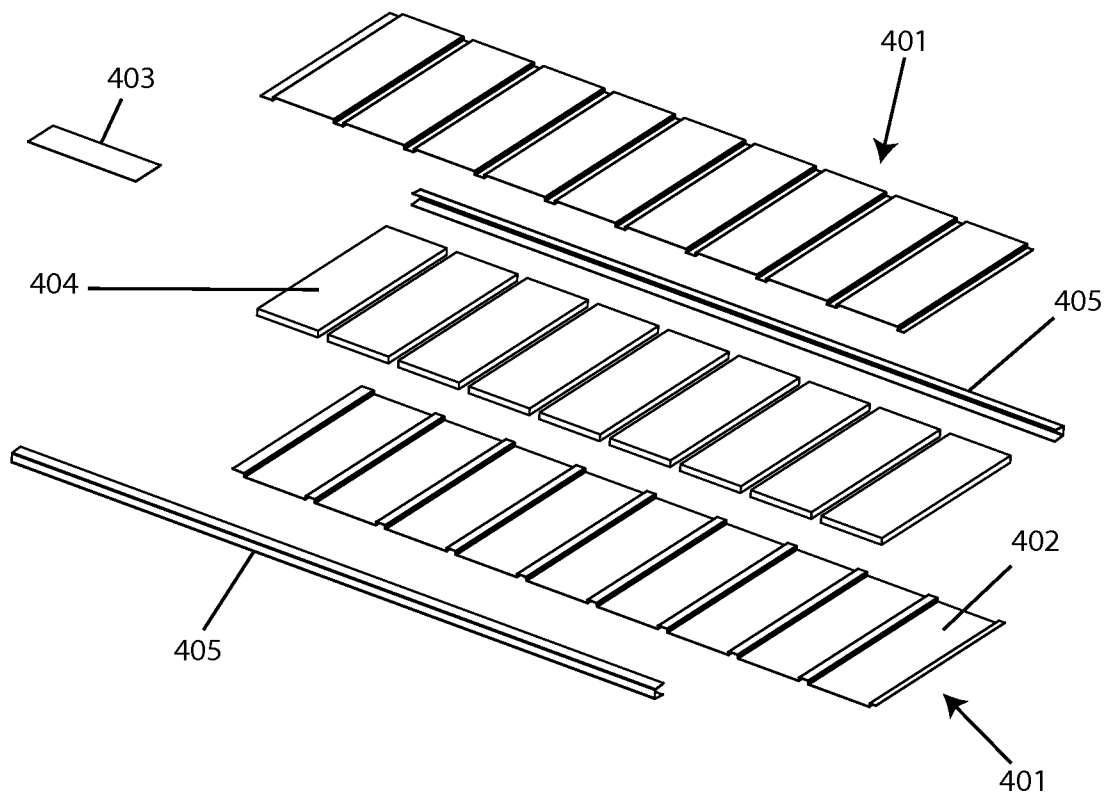


FIG. 4A

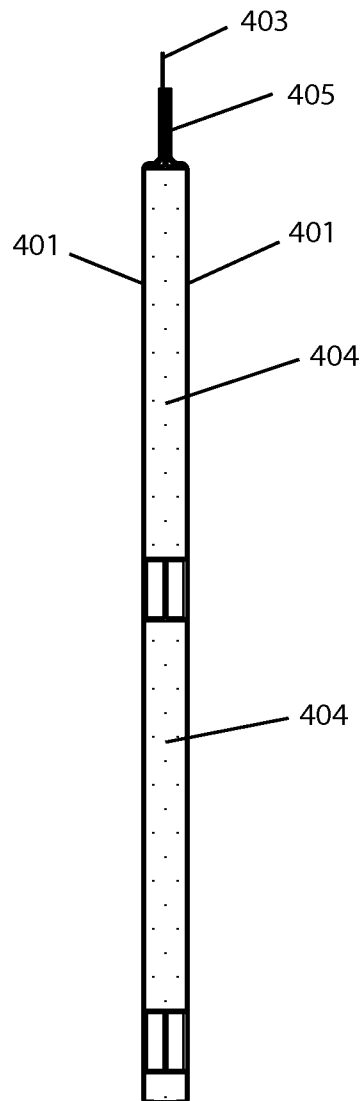


FIG. 4B

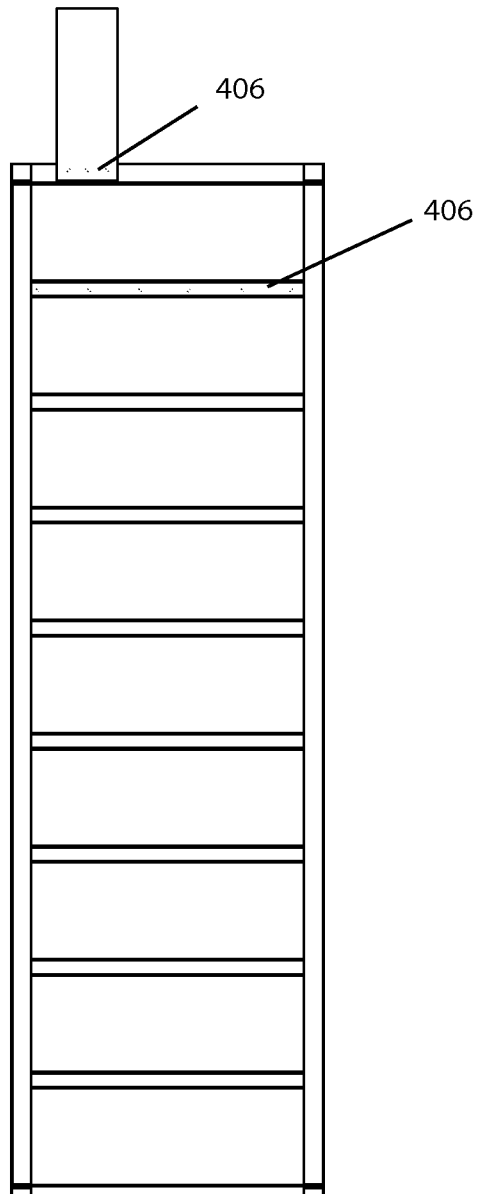


FIG. 4C

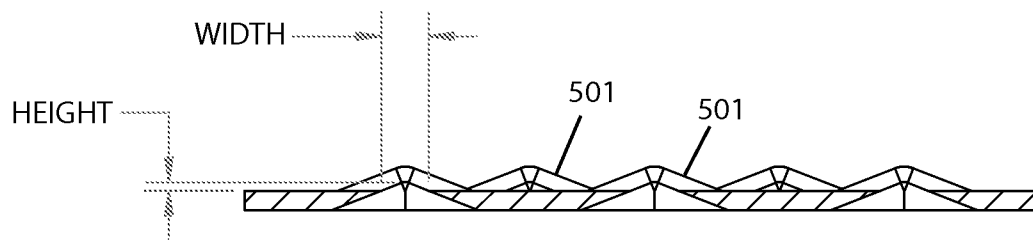


FIG. 5A

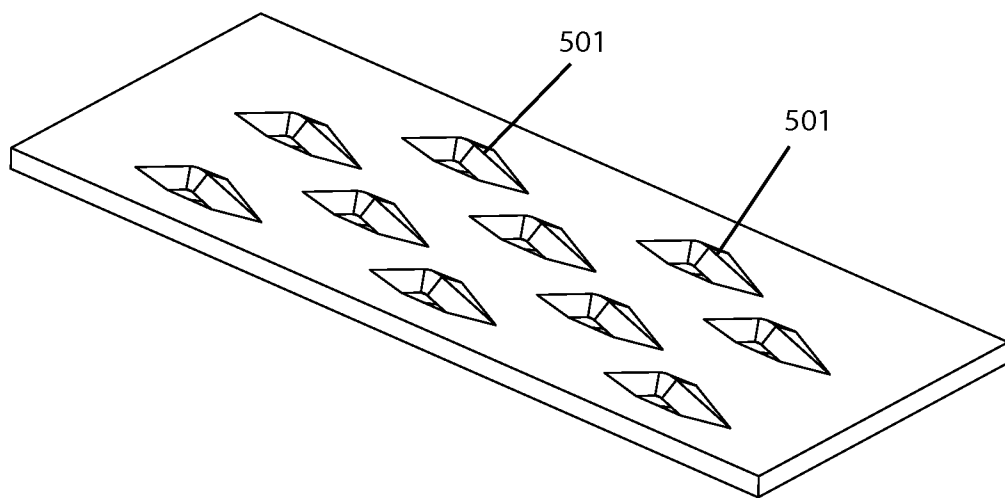


FIG. 5B

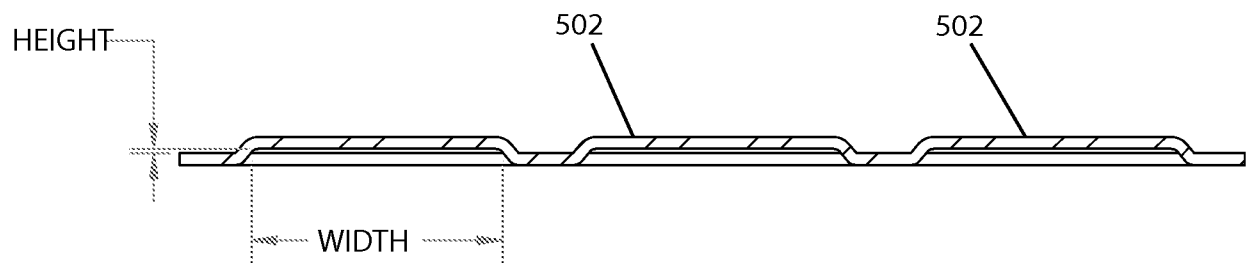


FIG. 5C

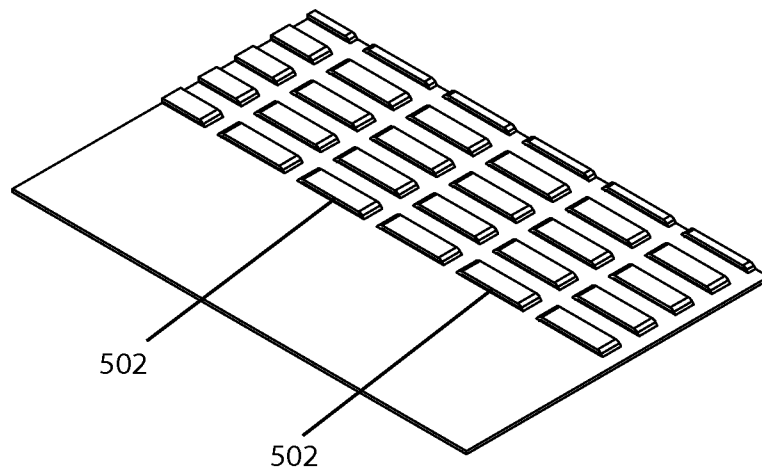


FIG. 5D

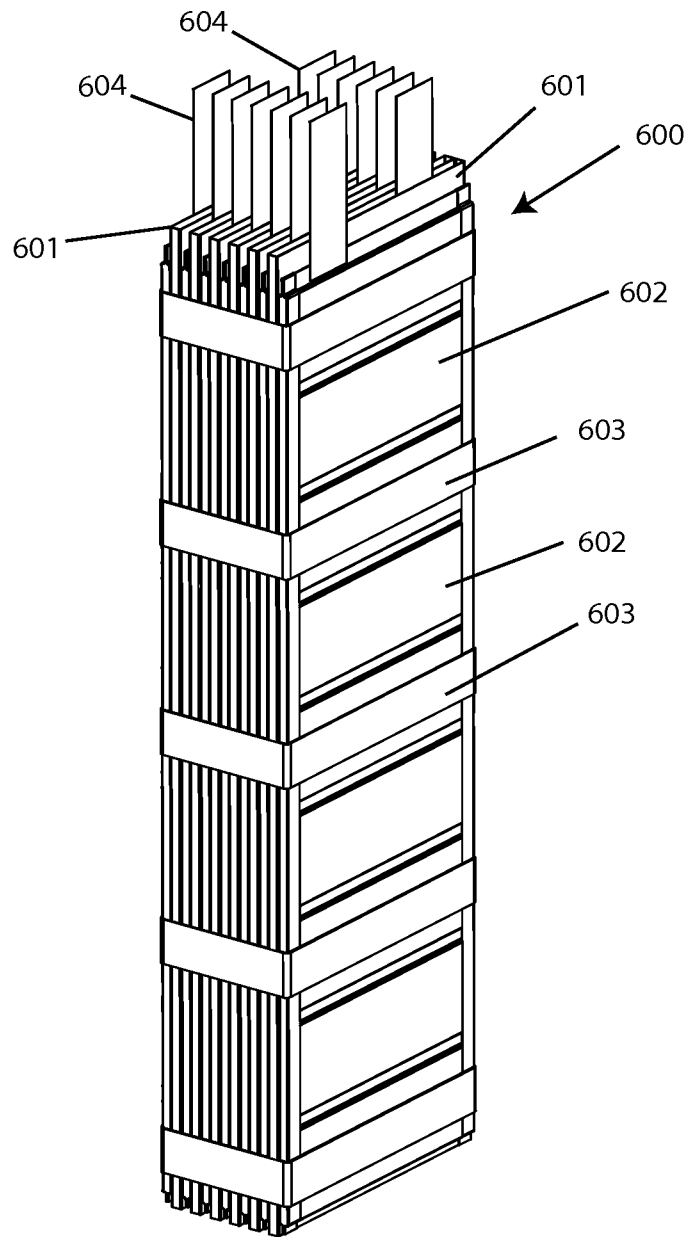


FIG. 6

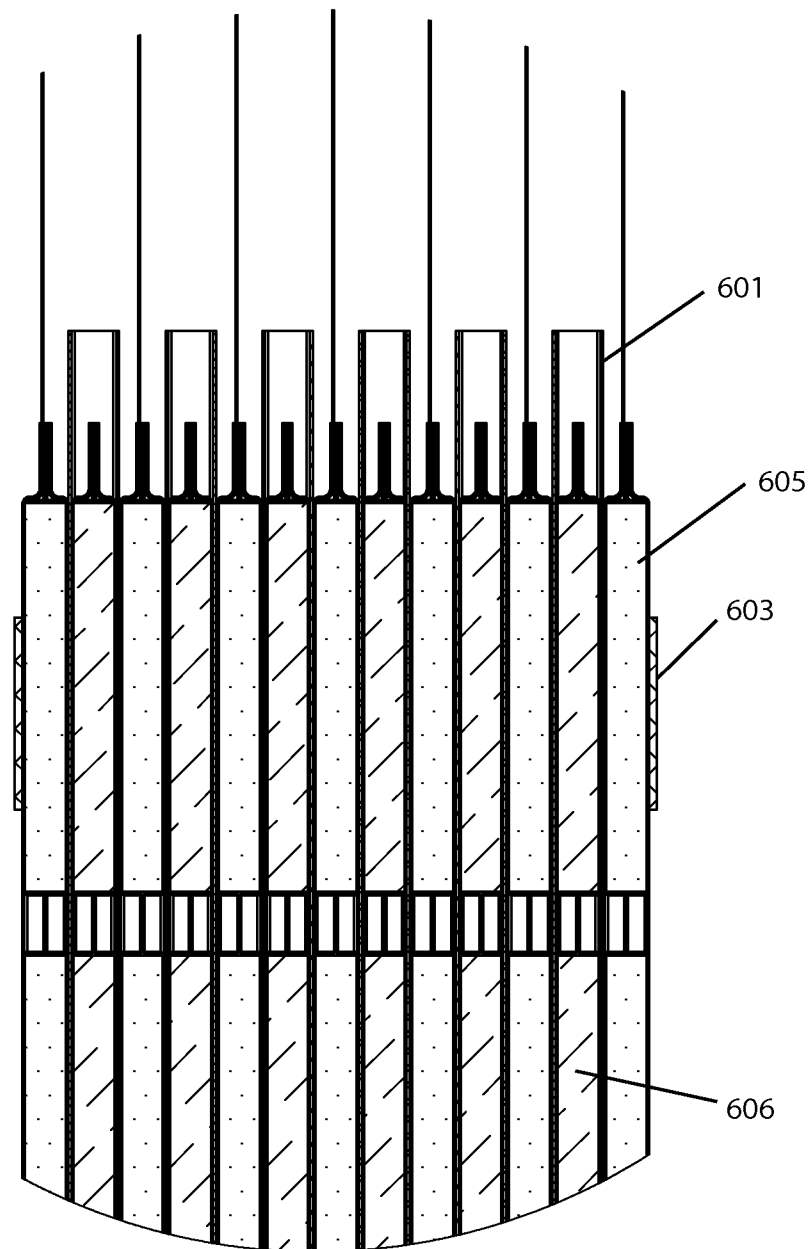


FIG. 7A

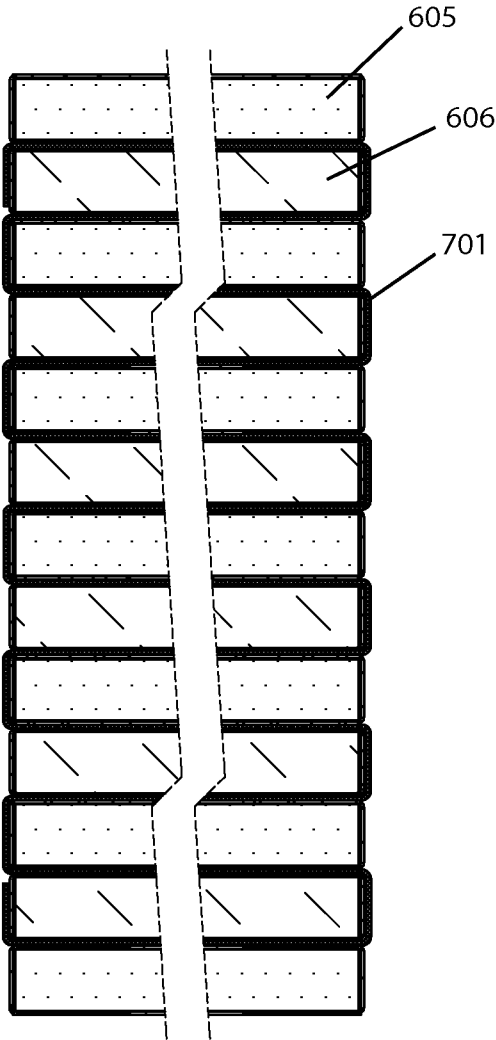


FIG. 7B

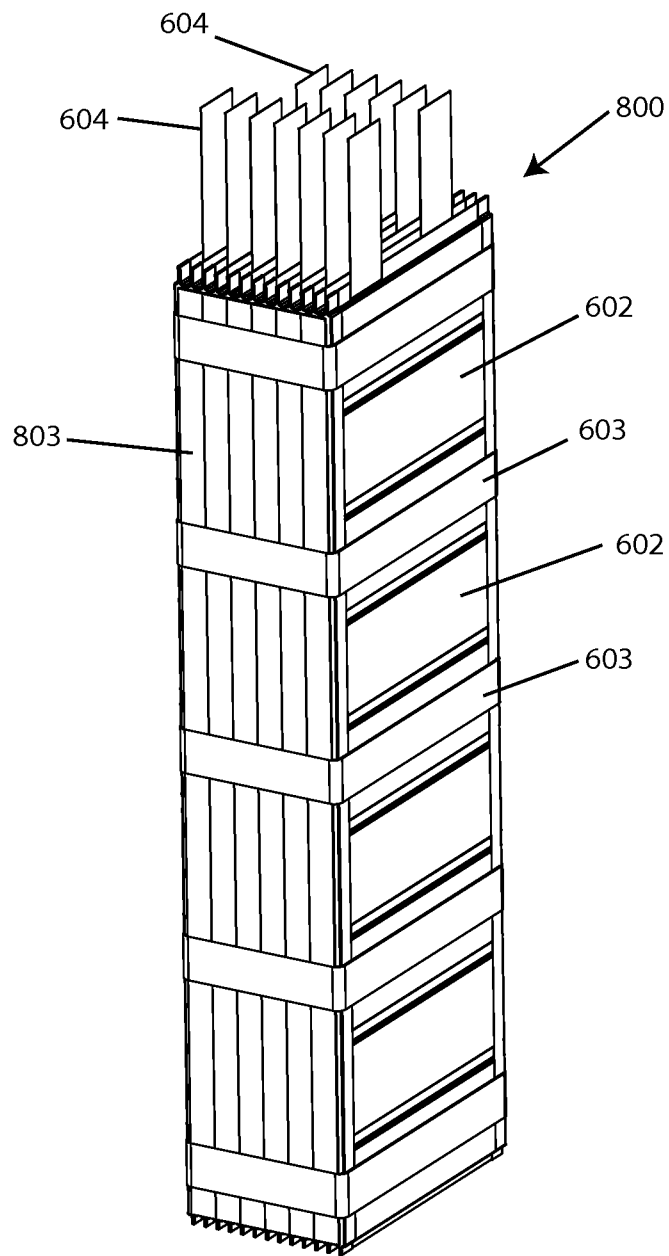


FIG. 8

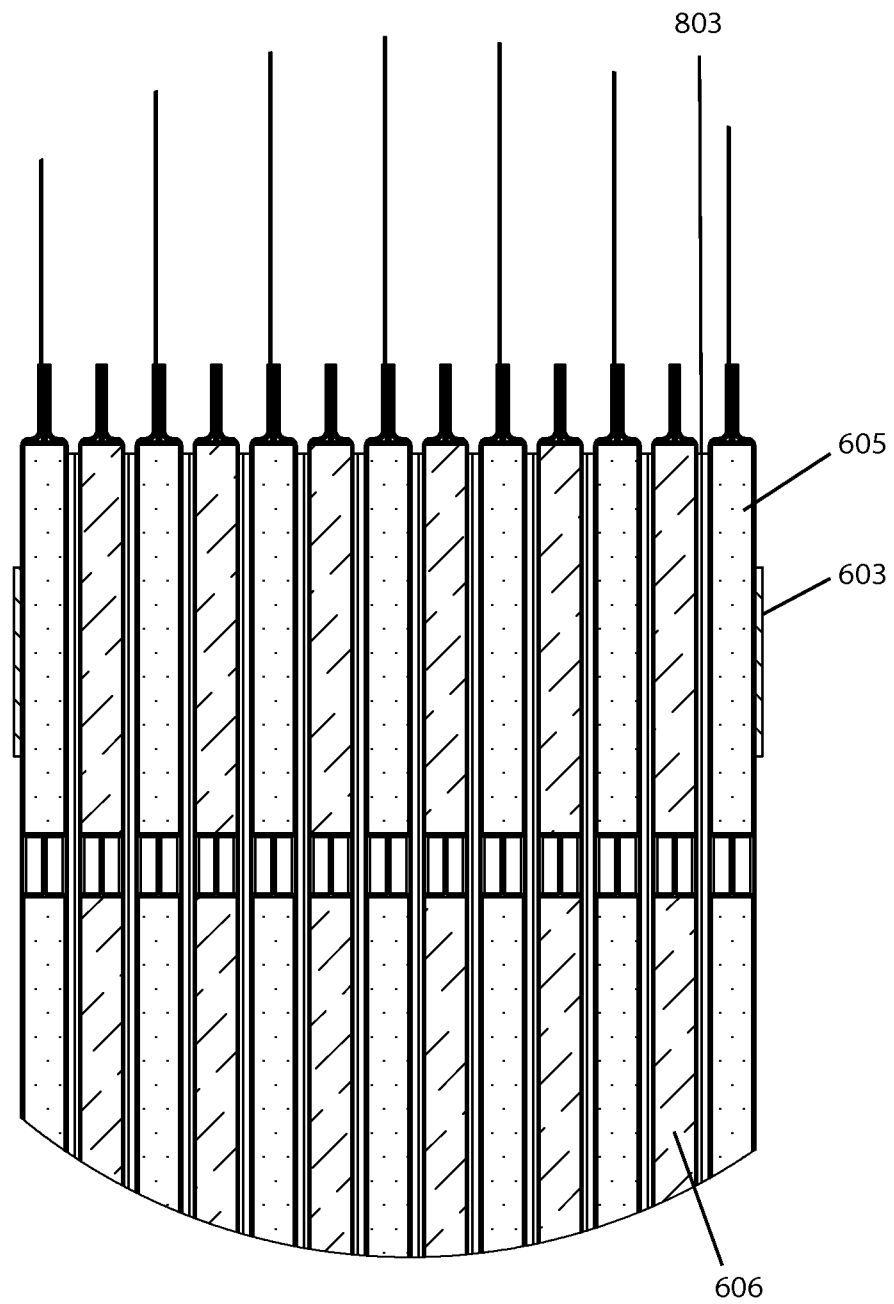


FIG. 9A

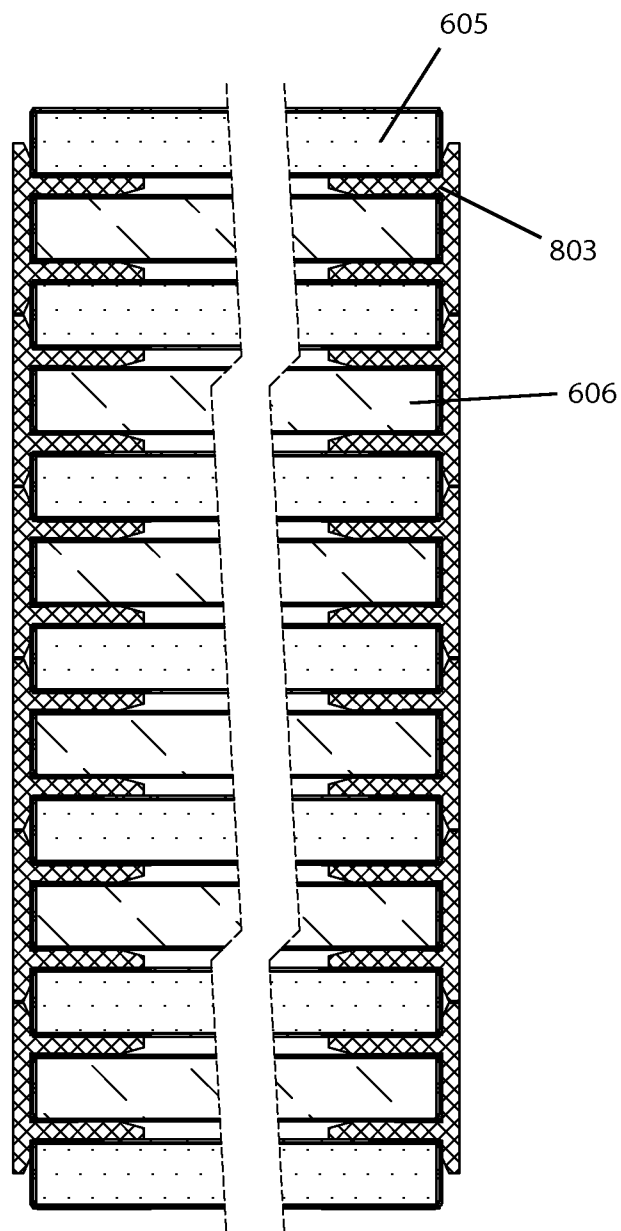


FIG. 9B

Anode Without Iron Powder

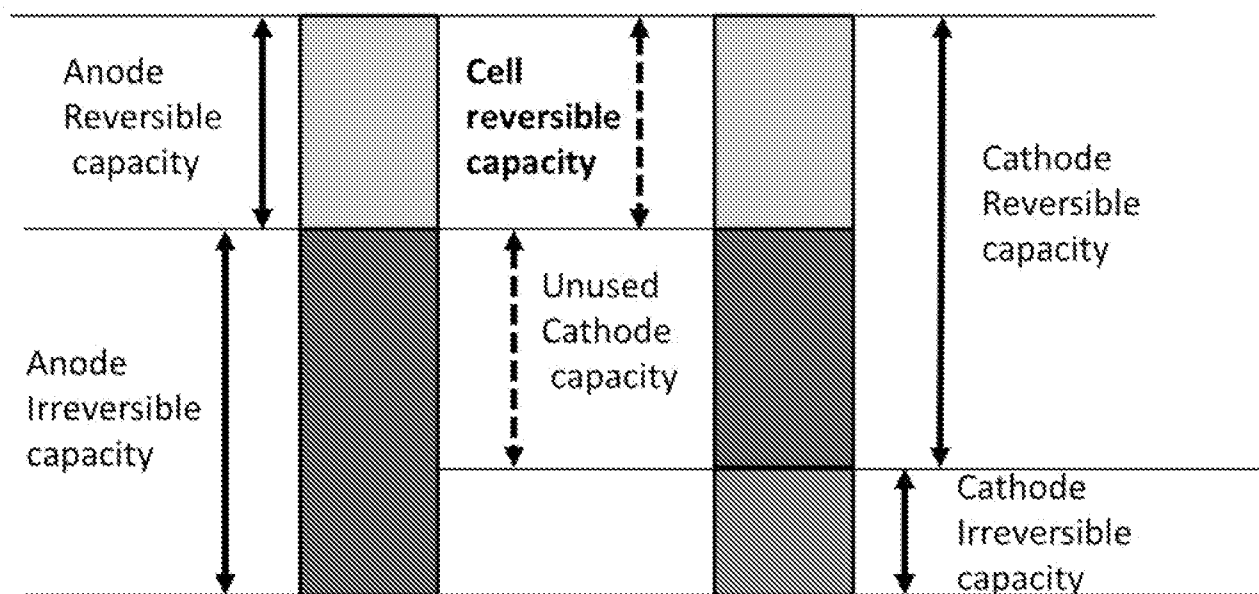


FIG. 10A

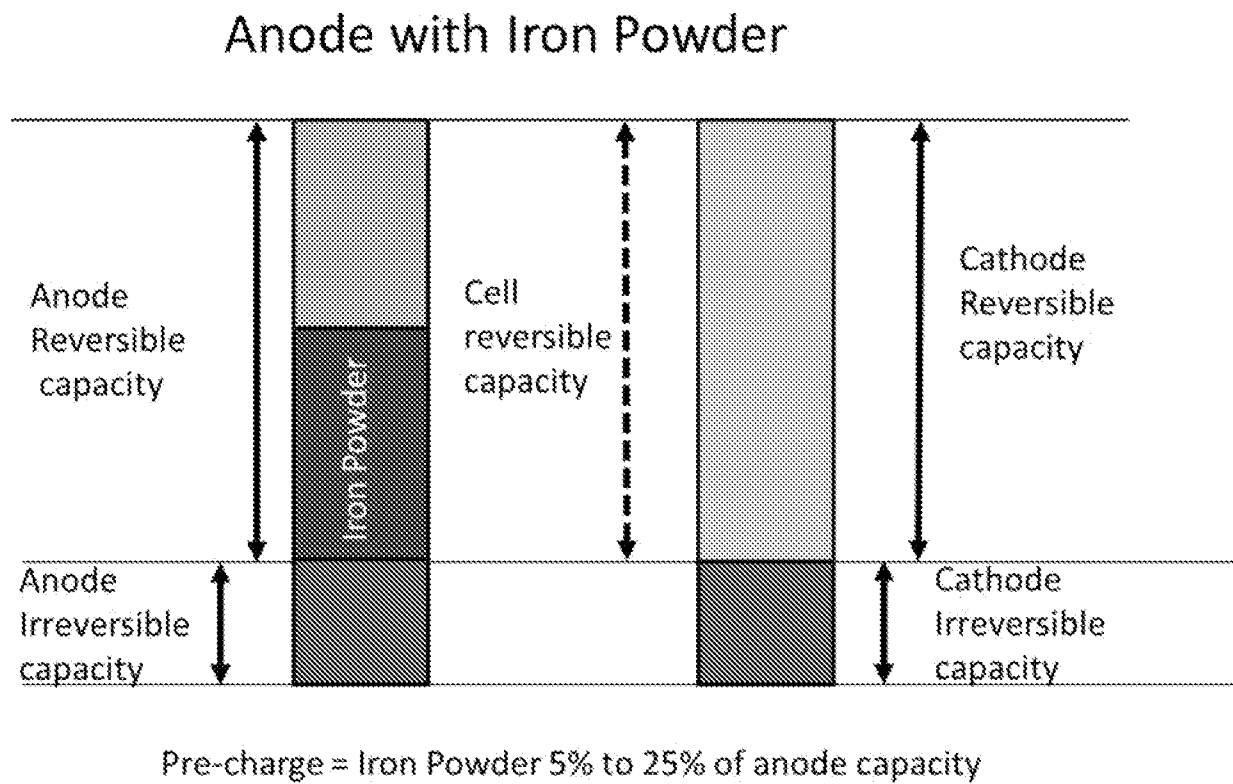


FIG. 10B

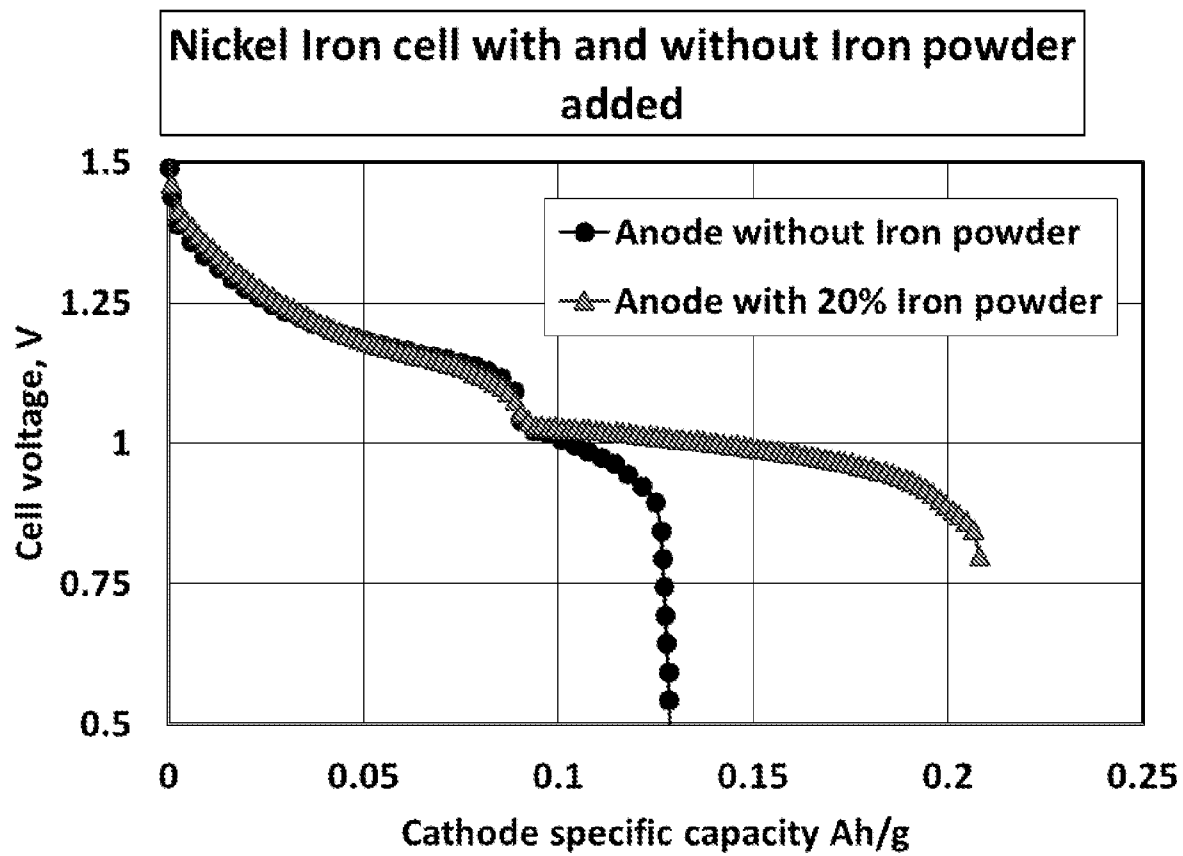


FIG.11

INTERNATIONAL SEARCH REPORT

International application No.

PCT/US 2014/043211

A. CLASSIFICATION OF SUBJECT MATTER <i>H01M 10/28 (2006.01)</i> <i>H01M 10/30 (2006.01)</i> <i>H01M 10/52 (2006.01)</i> <i>H01M 4/32 (2006.01)</i> According to International Patent Classification (IPC) or to both national classification and IPC				
B. FIELDS SEARCHED Minimum documentation searched (classification system followed by classification symbols) H01M 10/28, 10/30, 10/34, 4/32, 10/38, 10/04 Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched Electronic data base consulted during the international search (name of data base and, where practicable, search terms used) PatSearch (RUPTO internal), Espacenet, PAJ, USPTO, Information Retrieval System of FIPS				
C. DOCUMENTS CONSIDERED TO BE RELEVANT				
Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.		
Y	CN 201345381 Y (ZHENMING LIU) 11.11.2009, abstract, fig. 1-3, paragraphs [0008] - [0014], [0019] - [0021]	1-17		
Y	US 2012/0058387 A1 (SAMSUNG SDI CO., LTD) 08.03.2012, abstract, claims, paragraphs [0010] - [0018], [0031] - [0032], [0039] - [0041], [0049], fig. 1-3	1-17		
Y	RU 2064208 C1 (AKTSIONERNOE OBSHESTVO ZAKRYTOGO TIP "UNIAK") 20.07.1996, abstract, fig. 1, p. 4, col. 1, lines 18-24	2, 3		
Y	EP 1742279 A2 (GREATBATCH W TECHNOLOGIES) 10.01.2007, abstract, claims, fig. 1-14	9-17		
☐ Further documents are listed in the continuation of Box C. ☐ See patent family annex.				
* Special categories of cited documents: <table border="0"> <tr> <td style="vertical-align: top;"> "A" document defining the general state of the art which is not considered to be of particular relevance "E" earlier document but published on or after the international filing date "L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified) "O" document referring to an oral disclosure, use, exhibition or other means "P" document published prior to the international filing date but later than the priority date claimed </td> <td style="vertical-align: top;"> "T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention "X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone "Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art "&" document member of the same patent family </td> </tr> </table>			"A" document defining the general state of the art which is not considered to be of particular relevance "E" earlier document but published on or after the international filing date "L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified) "O" document referring to an oral disclosure, use, exhibition or other means "P" document published prior to the international filing date but later than the priority date claimed	"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention "X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone "Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art "&" document member of the same patent family
"A" document defining the general state of the art which is not considered to be of particular relevance "E" earlier document but published on or after the international filing date "L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified) "O" document referring to an oral disclosure, use, exhibition or other means "P" document published prior to the international filing date but later than the priority date claimed	"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention "X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone "Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art "&" document member of the same patent family			
Date of the actual completion of the international search 15 October 2014 (15.10.2014)		Date of mailing of the international search report 06 November 2014 (06.11.2014)		
Name and mailing address of the ISA/RU: FIPS, Russia, 123995, Moscow, G-59, GSP-5, Berezhkovskaya nab., 30-1 Facsimile No. +7 (499) 243-33-37		Authorized officer N. Rygalina Telephone No. (495)531-64-81		