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(54) LUBRICATING COMPOSITION INCLUDING ESTERIFIED COPOLYMER AND METHOD

SCHMIERMITTELZUSAMMENSETZUNG MIT VERESTERTEM COPOLYMER UND VERFAHREN

COMPOSITION LUBRIFIANTE CONTENANT UN COPOLYMÈRE ESTÉRIFIÉ ET PROCÉDÉ
AFFÉRENT

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(73) Proprietor: **The Lubrizol Corporation**

Wickliffe, OH 44092-2298 (US)

(72) Inventor: **BARTON, William R.S.**

Belper Derby GB (GB)

(74) Representative: **D Young & Co LLP**

**120 Holborn
London EC1N 2DY (GB)**

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Description

FIELD OF INVENTION

5 **[0001]** The invention relates to the use of an esterified copolymer to improve high-temperature high-shear performance of a lubricating oil for a driveline device.

BACKGROUND

10 **[0002]** Viscosity modifiers (also known as viscosity index (VI) improvers) are often added to lubricating oil compositions to improve the viscosity index of the lubricant. The viscosity modifier is combined, in the lubricating oil composition, with a base oil, together with other performance additives, such as dispersants, detergents, friction modifiers, corrosion inhibitors, pour point depressants, and so forth, depending on the particular application. Typical viscosity modifiers include polymers of alkyl methacrylates, alkyl acrylates, olefins (such as copolymers of ethylene and propylene), copolymers of styrene and conjugated dienes, and copolymers of maleic anhydride with alpha-olefins or styrene, and their esterified derivatives. Viscosity modifiers comprising ester groups tend to incorporate ester functional groups in pendent/grafted/branched groups. The ester functional groups may be derived from linear or branched alkyl alcohols with 1 to 40 carbon atoms. Examples of such polymers are disclosed in U.S. Patent Nos. 5,435,928; 6,174,843; 6,419,714; 6,544,935; and 7,254,249; and International Application Nos. WO 07/133999 and WO 2010/014655.

20 **[0003]** Lubricating oils are often subjected to high temperatures and high shear conditions. Under certain conditions, a viscosity modifier can be mechanically degraded by the moving parts and partially lose its ability to increase the viscosity of the base oil. The degradation causes lowered lubricant film thickness, which can, in turn, lead to accelerated wear. The shear stability of viscosity modifiers tends to increase as the molecular weight decreases. However, the thickening capability of the viscosity modifier also decreases as the molecular weight decreases.

25 **[0004]** Compositions have been formulated to achieve minimum high temperature/high shear rate (HTHS) viscosity to protect against excessive wear in various areas of an automobile, such as driveline devices and in the engine. High HTHS viscosity lubricating compositions, although good for wear protection, tend to suffer from poor fuel economy. By contrast, fluids with relatively low HTHS viscosities typically also exhibit low kinematic viscosity (KV), resulting in formation of a thinner lubricating film, which can improve fuel economy but negatively impact wear protection.

30 **[0005]** So for either enhanced wear protection, or for improved fuel efficiency, it is advantageous for the ratio of HTHS/KV to be as high as possible. With this relationship it would be possible to formulate, 1) lubricating oils with the same KV but higher HTHS, or in other words, to have the same fuel efficiency but improved wear protection, 2) lubricating oils with the same HTHS but reduced KV, i.e. the same wear protection but improved fuel efficiency, and 3) reduce KV but increase HTHS.

35 **[0006]** It would be desirable to provide a lubricating composition that provides good HTHS and shear stability for durability while providing a low kinematic viscosity to reduce friction and improve fuel economy.

BRIEF DESCRIPTION OF THE DRAWINGS

40 **[0007]**

Figure 1 is a graph showing the viscosity of the fluids of Example 2 under high shear rates at 50°C.

Figure 2 is a graph showing the viscosity of the fluids of Example 2 under high shear rates at 100°C.

Figure 3 is a graph showing the viscosity of the fluids of Example 2 under high shear rates at 125°C.

45 Figure 4 is a graph showing the viscosity of the fluids of Example 2 under high shear rates at 150°C.

BRIEF DESCRIPTION

50 **[0008]** The present invention concerns the use of an esterified copolymer to improve high-temperature high-shear (HTHS) performance measured according to ASTM D4683-10 of a lubricating oil, the esterified copolymer including a polymeric backbone comprising at least 60% of units (A1) derived from a vinyl monomer and units (A2) derived from a carboxylic acid monomer, wherein the ratio of (A1) to (A2) is from 1:3 to 3:1, the vinyl monomer comprising a vinyl aliphatic monomer, the carboxylic acid monomer comprising an ethylenically unsaturated carboxylic acid or derivative thereof, esterified with a primary alcohol, and no more than 30 mole% of units derived from one or more monomers capable of polymerizing with one or both of the vinyl monomer and carboxylic acid monomer, in a lubricating oil for a driveline device.

[0009] In one embodiment the use is to improve kinematic viscosity performance measured according to ASTM D445 of the lubricating oil as temperature increases. In one embodiment the use is to improve kinematic viscosity performance

measured according to ASTM D445 of the lubricating oil as shear rate increases.

DETAILED DESCRIPTION

[0010] A method for determining viscosity index (VI) of a lubricating composition is according to the procedure outlined in ASTM D2270, "Standard Practice for Calculating Viscosity Index From Kinematic Viscosity at 40 and 100°C," ASTM International, West Conshohocken, PA, DOI: 10.1520/D2270-10E01.

[0011] The kinematic viscosity (KV) is according to ASTM D445. The KV can be expressed as KV₁₀₀ or KV₄₀ (KV refers to Kinematic Viscosity, and 100 or 40 refers to the Celsius temperature at which the viscosity of the oil is measured). Unless otherwise noted, KV values are determined by this method and are reported in Centistokes (cSt). 1 cSt = 1 mm²/s.

[0012] A method for determining the High Temperature High Shear (HTHS) viscosity of a lubricating composition is according to the procedure ASTM D4683-10, "Standard Test Method for Measuring Viscosity of New and Used Engine Oils at High Shear Rate and High Temperature by Tapered Bearing Simulator Viscometer at 150 °C," ASTM International, West Conshohocken, PA. This test method determines the viscosity of an oil at 150 °C and 1.0·10⁶ s⁻¹ using a viscometer having a slightly tapered rotor and stator called the Tapered Bearing Simulator (TBS) Viscometer. Alternate methods for measuring HTHS viscosity include the Tapered Plug Viscometer (TPV, ASTM D4741) and the high pressure capillary viscometer (ASTM D5481).

[0013] For some applications, the lubricating composition can have a ratio of kinematic viscosity (KV₁₀₀) to HTHS of up to 2.6-3.6.

[0014] The viscosity at low temperatures can be measured using the Brookfield Viscosity method as described in ASTM method D2983-09, "Standard Test Method for Low-Temperature Viscosity of Lubricants Measured by Brookfield Viscometer." According to SAE J306, "Automotive Gear Lubricant Viscosity Classification," for SAE 75W automotive gear oils, the maximum viscosity measured at -26°C is also 150,000 cP.

[0015] The inventors have found that an esterified copolymer according to embodiments of the invention can provide a lubricant with at least one (or at least two, or all) of acceptable or improved HTHS, KV, shear stability, acceptable or improved viscosity index control, and acceptable or improved low temperature viscosity.

[0016] Accordingly, an exemplary lubricating composition includes (A) an esterified copolymer and an oil of lubricating viscosity. The esterified copolymer is formulated to provide the lubricating composition with a high ratio of high temperature/high shear rate (HTHS) to kinematic viscosity (KV). The resulting lubricating composition can have improved wear protection while providing low kinematic viscosity for improved fuel efficiency.

[0017] The exemplary lubricating composition finds use, for example, as an oil for a driveline device, such as a transmission. In one embodiment, the lubricating composition is used in a syncromesh transmission fluid (SSTF).

[0018] An exemplary lubricating composition can include from 0.1 to 50 wt. %, or 1.0 to 20 wt. %, or 1.5 to 10 wt. % of the exemplary viscosity modifier, such as less than 5 wt. % of the viscosity modifier.

[0019] Described herein is a use of the lubricating composition described herein in a transmission which includes providing the transmission with the lubricating composition.

[0020] Also described is a method of providing a lubricating composition with at least one (or at least two, or up to all) of acceptable or improved shear stability, acceptable or improved viscosity index control, acceptable or improved oxidation control, and acceptable or improved low temperature viscosity which includes providing the lubricating composition with an esterified copolymer.

I. The Viscosity Modifier (A)

[0021] The exemplary viscosity modifier is an esterified copolymer. The exemplary esterified copolymer (A) disclosed herein includes a polymeric backbone comprising at least 70% of units (A1) derived from a vinyl monomer and units (A2) derived from a carboxylic acid monomer, which can be an ethylenically unsaturated carboxylic acid or derivative thereof, esterified with a primary alcohol, wherein the ratio of (A1) to (A2) is from 1:3 to 3:1. The vinyl monomer comprises a vinyl aliphatic monomer. Exemplary aliphatic vinyl monomers include alpha-olefins. By way of example, the exemplary copolymer can include a polymeric backbone derived from 1-dodecene and maleic anhydride, as the carboxylic acid monomer.

[0022] In the exemplary copolymer, at least 70%, or at least 80%, or at least 90%, or at least 95%, such as 70%-95%, and up to 100% of the units in the backbone, is derived from the vinyl monomer and the carboxylic acid monomer. Pendant groups may be grafted to the backbone, such as by esterification and/or amidization/imidization of the units of the backbone that are derived from the carboxylic acid monomer.

[0023] In general, the polymeric backbone can be an alternating structure whereby each carboxylic acid unit is spaced from the next carboxylic acid unit by at least one unit derived from a vinyl monomer, such as an alpha-olefin. The exemplary esterified copolymer can have at least 20 or at least 100 units derived from these monomers in its backbone. In one embodiment, the backbone chain of monomer units derived from the selected monomers is of no more than

10,000 such monomer units, or no more than 1000 such monomer units.

[0024] A molar ratio of the vinyl monomer units (A1) to carboxylic acid monomer units (A2) in the copolymer is 1:3 to 3:1 or from 1:2 to 2:1, or 0.6:1 to 1:1. In one embodiment, the molar ratio is about 0.7:1 to 1:1.1 in the copolymer. It is to be appreciated, however, that the molar ratios used in the preparation of the copolymer may differ from those in the

[0025] The copolymer backbone optionally includes units derived from monomers other than the vinyl monomer (A1) and carboxylic acid monomer (A2) disclosed herein. For example, the backbone may include units derived from one or more monomers capable of polymerizing with one or both of the vinyl monomer and carboxylic acid monomer such as acrylates or methacrylates. The units derived from these other monomers constitute no more than 30 mole % or no more than 20 mole % or no more than 10 mole % of the monomer derived units in the copolymer. For example, 1-5% of the units in the backbone may be formed from these other monomers.

[0026] The exemplary esterified copolymer further includes ester pendent groups formed by esterification of carboxylic acid units of the copolymer with a primary alcohol, such as one or more of a linear alcohol and a branched alcohol, cyclic or acyclic alcohol, or a combination thereof.

[0027] The exemplary esterified copolymer may further include a nitrogen containing group (such as amino-, amido- and/or imido- group, or a nitrogen containing salt.

[0028] In one embodiment, the ester groups and nitrogen containing groups (such as amino-, amido- and/or imido- group) are sufficient to provide 0.01 wt. % to 1.5 wt. % (or 0.02 wt. % to 0.75 wt. %, or 0.04 wt. % to 0.25 wt. %) nitrogen to the esterified copolymer.

[0029] Weight average molecular weight (M_w) as used herein, is measured by gel permeation chromatography (GPC), also known as size-exclusion chromatography, employing a polystyrene standard. Typically the weight average molecular weight is measured on the final esterified copolymer, optionally reacted with a nitrogen-containing compound. The M_w of the exemplary polymer, before esterification, can range from 3000 to 50,000, or 5000 to 30,000, or 5000 to 25,000, or 10,000 to 17,000. The M_w of the exemplary polymer, before esterification, can also range from about 5000 to 10,000, or 12,000 to 18,000, or 9000 to 15,000, or 15,000 to 20,000, or 8000 to 21,000, and in one embodiment, may be at least 10,000. The M_w of the exemplary esterified polymer, after esterification and optional reaction with the nitrogen-containing compound, can range from 5000 to 50,000, and in one embodiment, may be 5000 to 35,000.

A. The Vinyl Monomer Units of the Viscosity Modifier

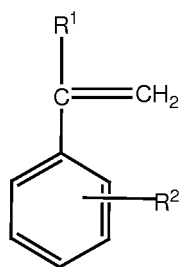
[0030] The esterified copolymer (A) includes vinyl monomer units (A1) derived from at least one of a vinyl aromatic monomer or a vinyl aliphatic monomer. In the present invention the vinyl monomer units at least include units derived from the vinyl aliphatic monomer.

[0031] Where both are present, the arrangement of vinyl monomer units in the backbone can be largely random and depend on the ratio of the two vinyl monomers.

[0032] An exemplary vinyl aliphatic monomer is a polymerizable aliphatic monomer, specifically, an aliphatic compound substituted with a vinyl group ($-\text{CH}=\text{CH}_2$). Examples of vinyl aliphatic monomers include alpha-olefins selected from C_6 - C_{30} alpha-olefins such C_8 - C_{20} alpha-olefins, or C_{10} - C_{18} alpha-olefins, or C_{10} - C_{14} alpha-olefins. The alpha-olefin can be linear or branched. Examples of suitable linear alpha-olefins include 1-decene, 1-undecene, 1-dodecene, 1-tridecene, 1-tetradecene, 1-pentadecene, 1-hexadecene, 1-heptadecene 1-octadecene, and mixtures thereof. An example of a useful vinyl aliphatic monomer is 1-dodecene. In an embodiment, the vinyl aliphatic monomer is a hydrocarbon and does not contain a heteroatom, such as O, N or S. In an embodiment, the vinyl aliphatic monomer is an alpha-olefin.

[0033] An exemplary vinyl aromatic monomer, where present, is a polymerizable aromatic monomer, specifically, an aromatic compound substituted with a vinyl group ($-\text{CH}=\text{CH}_2$).

[0034] Suitable vinyl aromatic monomers are those corresponding to Formula I:



Formula I:

wherein R¹ and R² independently represent a hydrogen atom, an alkyl group having 1 to 4 carbon atoms, or a halogen group. The vinyl aromatic monomer may be selected from styrene, alpha-alkylstyrenes, nuclear alkylstyrenes, chlorostyrenes, dichlorostyrenes, vinylnaphthalene, and mixtures of these. Specific examples include styrene, alpha-methylstyrene, alpha-ethylstyrene, alpha-isopropylstyrene, p-tert-butylstyrene, nuclear alkylstyrenes such as o-methylstyrene, m-methylstyrene, p-methylstyrene, o-methyl-alpha-methylstyrene, m-methyl-alpha-methylstyrene, p-methyl-alpha-methylstyrene, m-isopropyl-alpha-methylstyrene, p-isopropyl-alpha-methylstyrene, m-isopropylstyrene, p-isopropylstyrene, vinylnaphthalene, and mixtures thereof.

[0035] In the exemplary embodiment, a majority of the vinyl monomer units of A1 (at least 50% of the vinyl monomer units or at least 80%, or at least 90%, or 100% of the vinyl monomer units) are derived from the vinyl aliphatic monomer.

B. The Carboxylic Acid Units of the Viscosity Modifier

[0036] The exemplary ethylenically unsaturated carboxylic acid or derivative thereof, esterified with a primary alcohol, used in forming the carboxylic acid units (A2) of the exemplary esterified copolymer (A) may be a mono- or dicarboxylic acid or an anhydride or other derivative thereof that may be wholly esterified, partially esterified, or a mixture thereof. When partially esterified, other functional groups may include acids, salts or mixtures thereof. Suitable salts include alkali metals, alkaline earth metals, and mixtures thereof. The salts may include lithium, sodium, potassium, magnesium, calcium or mixtures thereof.

[0037] Exemplary unsaturated α,β -ethylenically unsaturated carboxylic acids or derivatives thereof which may be used in forming the carboxylic acid units of the copolymer include acrylic acid, methyl acrylate, methacrylic acid, maleic acid, fumaric acid, itaconic acid, alpha-methylene glutaric acid, and anhydrides and mixtures thereof, and substituted equivalents thereof. Suitable examples of monomers for forming the carboxylic acid unit include itaconic anhydride, maleic anhydride, methyl maleic anhydride, ethyl maleic anhydride, dimethyl maleic anhydride, and mixtures thereof. In one embodiment, the carboxylic acid unit includes units derived from maleic anhydride or derivatives thereof.

[0038] In the exemplary unsaturated carboxylic acids or derivatives thereof, a carbon-to-carbon double bond is typically in an alpha, beta-position relative to at least one of the carboxy functions (e.g., in the case of itaconic acid, anhydride or esters thereof) and may be in an alpha, beta-position to both of the carboxy functions of an alpha, beta-dicarboxylic acid, anhydride or the ester thereof (e.g., in the case of maleic acid or anhydride, fumaric acid, or ester thereof). In one embodiment, the carboxy functions of these compounds will be separated by up to 4 carbon atoms, such as 2 carbon atoms.

[0039] Other suitable monomers for forming the carboxylic acid monomer unit of the exemplary esterified copolymer are described in U.S. Pub. No. 20090305923.

C. Alcohols for Esterification of the Carboxylic Acid Unit

[0040] The carboxylic acid unit of the exemplary esterified copolymer (A) is wholly or partially esterified with a primary alcohol. The ester groups are usually formed by reacting the carboxy-containing copolymer with alcohols although in some embodiments, especially for lower alkyl esters, the ester group may be incorporated from one of the monomers used to prepare the copolymer.

[0041] Suitable primary alcohols for use herein may contain 1 to 150 carbon atoms, or 4 to 50, 2 to 20, or 8 to 20 (such as 4 to 20, or 4 to 16, or 8 to 12 carbon atoms). The primary alcohol may be linear or may be branched at the α -, or β -, or higher position, cyclic or acyclic, or a combination thereof. In one embodiment, a mixture of linear and branched alcohols is employed in forming the esterified copolymer described herein. In one exemplary embodiment, at least 0.1 % of the carboxylic acid units in the copolymer are esterified with an alcohol branched at the β - or higher position.

[0042] In one embodiment, 10, 20, or 30 to 100 mole %, alternatively 30 to 70 mole %, based on the total number of moles of carboxyl groups in the copolymer contain ester groups having 12 to 19 carbon atoms in the alkyl group (that is, in the alcohol-derived or alkoxy portion of the ester) and 70 or 80 to 0 mole %, alternatively 80 to 30 mole %, based on the total number of moles of carboxyl groups in the esterified copolymer, contain ester groups having 8 to 10 carbon atoms in the alcohol portion. In one embodiment, the ester contains at least 45 mole %, based on moles of carboxyl groups in the esterified copolymer, of ester groups containing from 12 to 18 carbon atoms in the alcohol portion. In an optional embodiment, the esterified copolymer has up to 20 mole % or 0 to 5% or 1 to 2%, based on the total number of moles of carboxyl groups in the copolymer, of ester groups having from 1 to 6 carbon atoms in the alcohol portion. In one embodiment, the compositions are substantially free of ester groups containing from 3 to 7 carbon atoms.

[0043] In one embodiment, 0.1 to 99.99 (or 1 to 90, or 2 to 50, or 2.5 to 20, or 5 to 15) percent of the carboxylic acid units esterified are esterified with a primary alcohol branched at the β - or higher position, 0.1 to 99.99 (or 1 to 90, or 2 to 50, or 2.5 to 20, or 5 to 15) percent of the carboxylic acid units esterified are esterified with a linear alcohol or an alpha-branched alcohol, and 0.01 to 10 % (or 0.1 % to 20 %, or 0.02 % to 7.5 %, or 0.1 to 5 %, or 0.1 to less than 2 %) of the carboxylic acid units has at least one nitrogen-containing group, such as an amino-, amido- and/or imido-group,

as described below. As an example, 5 to 15 percent of the carboxylic acid units of the copolymer are esterified with a primary alcohol branched at the β - or higher position, 0.1 to 95 percent of the carboxylic acid units are esterified with a linear alcohol or an alpha-branched alcohol, and 0.1 to less than 2% of the carboxylic acid units has at least one nitrogen-containing group.

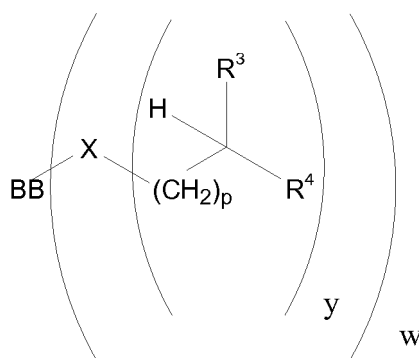
[0044] Examples of useful primary alcohols include butanol, heptanol, octanol, 2-ethylhexanol, decanol, dodecanol, tridecanol, tetradecanol, pentadecanol, hexadecanol, heptadecanol, octadecanol, and combinations thereof.

[0045] Other exemplary primary alcohols include commercially available mixtures of alcohols. These include oxoalcohols which may comprise, for example, various mixtures of alcohols having from 8-24 carbon atoms. Of the various commercial alcohols useful herein, one contains 8 to 10 carbon atoms, and another 12 to 18 aliphatic carbon atoms. The alcohols in the mixture may include one or more of, for example, octyl alcohol, decyl alcohol, dodecyl alcohol, tetradecyl alcohol, pentadecyl alcohol, and octadecyl alcohol. Several suitable sources of these alcohol mixtures are the technical grade alcohols sold under the name NEODOL® alcohols (Shell Oil Company, Houston, Tex.) and under the name ALFOL® alcohols (Sasol, Westlake, La.), and fatty alcohols derived from animal and vegetable fats and sold commercially by, for example, Henkel, Sasol, and Emery.

[0046] Tertiary alkanolamines, i.e., N,N-di-(lower alkyl)amino alkanolamines, are other alcohols that may be used to prepare the esterified copolymers. Examples include N,N-dimethylethanolamine, N,N-diethylethanolamine, 5-diethyl-amino-2-pentanol, and combinations thereof.

[0047] Exemplary primary alcohols branched at the β - or higher position may include Guerbet alcohols. Methods to prepare Guerbet alcohols are disclosed in U.S. Patent No. 4,767,815 (see column 5, line 39 to column 6, line 32).

[0048] The primary alcohol branched at the β - or higher position may be used to provide pendent groups as represented within ()_w of the Formula II:



Formula II

wherein

(BB) is a copolymer backbone derived from the carboxylic acid monomer units and vinyl monomer units, i.e. the carbon backbone not including the carboxyl group;

X is a functional group which either (i) contains a carbon and at least one oxygen or nitrogen atom or (ii) is an alkylene group with 1 to 5 carbon atoms (typically -CH₂-), connecting the copolymer backbone and a branched hydrocarbyl group contained within ()_y;

w is the number of pendent groups attached to the copolymer backbone, which may be in the range of 2 to 2000, or 2 to 500, or 5 to 250;

y is 0, 1, 2 or 3, provided that in at least 1 mol. % of the pendent groups, y is not zero; and with the proviso that when y is 0, X is bonded to a terminal group in a manner sufficient to satisfy the valence of X, wherein the terminal group is selected from hydrogen, alkyl, aryl, a metal (typically introduced during neutralization of ester reactions. Suitable metals include calcium, magnesium, barium, zinc, sodium, potassium or lithium) or ammonium cation, and mixtures thereof;

p is an integer in the range of 1 to 15 (or 1 to 8, or 1 to 4);

R³ and R⁴ are independently linear or branched hydrocarbyl groups, and the combined total number of carbon atoms present in R³ and R⁴ is at least 12 (or at least 16, or at least 18 or at least 20).

[0049] In different embodiments the copolymer with pendent groups may contain 0.10% to 100%, or 0.5% to 20%, or 0.75% to 10%, branched hydrocarbyl groups represented by a group within ()_y of the Formula II, expressed as a percentage of the total number of pendent groups. The pendent groups of Formula II may also be used to define the

ester groups as defined above by the phrase "a primary alcohol branched at the β - or higher position".

[0050] In different embodiments the functional groups defined by X in Formula II above, may comprise at least one of $-\text{CO}_2-$, $-\text{C}(\text{O})\text{N}=$ or $-(\text{CH}_2)_v-$, wherein v is an integer in the range of 1 to 20, or 1 to 10, or 1 to 2.

[0051] In one embodiment X is derived from an α,β -ethylenically unsaturated dicarboxylic acid or derivatives thereof. Examples of suitable carboxylic acids or derivatives thereof may include maleic anhydride, maleic acid, (meth)acrylic acid, itaconic anhydride, or itaconic acid. In one embodiment, the ethylenically unsaturated carboxylic acid or derivatives thereof may be at least one of maleic anhydride and maleic acid.

[0052] In one embodiment X is other than an alkylene group, connecting the copolymer backbone and the branched hydrocarbyl groups.

[0053] In different embodiments the pendent groups may be esterified, amidated or imidated functional groups.

[0054] Examples of suitable groups for R^3 and R^4 in Formula II include: alkyl groups containing C_{15-16} polymethylene groups, such as 2- C_{1-15} alkyl-hexadecyl groups (e.g., 2-octylhexadecyl) and 2-alkyl-octadecyl groups (e.g., 2-ethyl-octadecyl, 2-tetradecyl-octadecyl and 2-hexadecyloctadecyl); alkyl groups containing C_{13-14} polymethylene groups, such as 1- C_{1-15} alkyl-tetradecyl groups (e.g., 2-hexyltetradecyl, 2-decyltetradecyl and 2-undecyltridecyl) and 2- C_{1-15} alkyl-hexadecyl groups (e.g., 2-ethyl-hexadecyl and 2-dodecylhexadecyl); alkyl groups containing C_{10-12} polymethylene groups, such as 2- C_{1-15} alkyl-dodecyl groups (e.g., 2-octyldodecyl) and 2- C_{1-15} alkyl-dodecyl groups (2-hexyldodecyl and 2-octyldodecyl), 2- C_{1-15} alkyl-tetradecyl groups (e.g., 2-hexyltetradecyl and 2-decyltetradecyl); alkyl groups containing C_{6-9} polymethylene groups, such as 2- C_{1-15} alkyl-decyl groups (e.g., 2-octyldecyl) and 2,4-di- C_{1-15} alkyl-decyl groups (e.g., 2-ethyl-4-butyl-decyl); alkyl groups containing C_{1-5} polymethylene groups, such as 2-(3-methylhexyl)-7-methyl-decyl and 2-(1,4,4-trimethylbutyl)-5,7,7-trimethyl-octyl groups; and mixtures of two or more branched alkyl groups, such as alkyl residues of oxoalcohols corresponding to propylene oligomers (from hexamer to undecamer), ethylene/propylene (molar ratio 16:1-1:11) oligomers, isobutene oligomers (from pentamer to octamer), and C_{5-17} α -olefin oligomers (from dimer to hexamer).

[0055] The pendent groups in Formula II may contain a total combined number of carbon atoms on R^3 and R^4 in the range of 12 to 60, or 14 to 50, or 16 to 40, or 18 to 40, or 20 to 36.

[0056] Each of R^3 and R^4 may individually contain 5 to 25, or 8 to 32, or 10 to 18 methylene carbon atoms. In one embodiment, the number of carbon atoms on each R^3 and R^4 group may be 10 to 24.

[0057] In different embodiments, the primary alcohol branched at the β - or higher position may have at least 12 (or at least 16, or at least 18 or at least 20) carbon atoms. The number of carbon atoms may range from at least 12 to 60, or at least 16 to 30.

[0058] Examples of suitable primary alcohols branched at the β - or higher position include 2-ethylhexanol, 2-butyloctanol, 2-hexyldecanol, 2-octyldodecanol, 2-decyltetradecanol, and mixtures thereof.

D. Nitrogen-Containing Group of the Viscosity Modifier

[0059] The exemplary esterified copolymer (A) may include a nitrogen-containing group, such as an amino-, amido- and/or imido- group. The nitrogen-containing group may be derived from a nitrogen-containing compound capable of being incorporated during copolymerization (or through reaction with the carboxylic acid units to form a salt), such as an amine, to form a salt, amide, imide, or mixture thereof.

[0060] Examples of suitable nitrogen-containing compounds capable of being incorporated into the copolymer include N,N-dimethylacrylamide, N-vinyl carbonamides, such as, N-vinyl-formamide, N-vinylacetamide, N-vinyl propionamides, N-vinyl hydroxyacetamide, vinyl pyridine, N-vinyl imidazole, N-vinyl pyrrolidinone, N-vinyl caprolactam, dimethylaminoethyl acrylate, dimethylaminoethyl methacrylate, dimethylaminobutyl acrylamide, dimethylaminopropyl methacrylate, dimethylaminopropyl acrylamide, dimethylaminopropyl methacrylamide, dimethylaminoethyl acrylamide, and mixtures thereof.

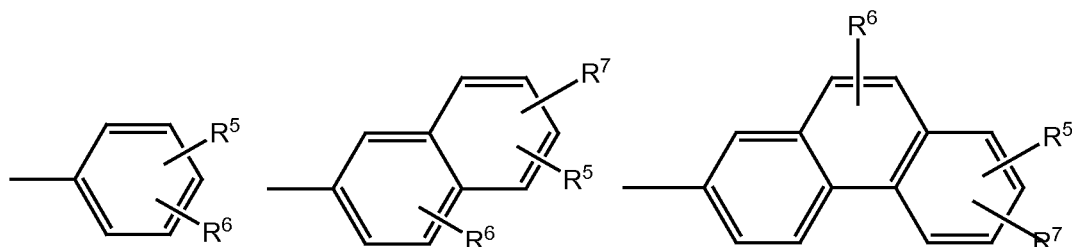
[0061] The copolymer may include a nitrogen containing group that may be capable of reacting with the esterified copolymer backbone, typically for capping the copolymer backbone. The capping may result in the copolymer having ester, amide, imide, and/or or amine groups.

[0062] The nitrogen-containing group may be derived from a primary or secondary amine, such as an aliphatic amine, aromatic amine, aliphatic polyamine, aromatic polyamine, polyaromatic polyamine, or combination thereof.

[0063] In one embodiment, the nitrogen containing group may be derived from an aliphatic amine, such as a $\text{C}1\text{-C}30$ or $\text{C}1\text{-C}24$ aliphatic amine. Examples of suitable aliphatic amines include aliphatic monoamines and diamines, which may be linear or cyclic. Examples of suitable primary amines include methylamine, ethylamine, propylamine, butylamine, pentylamine, hexylamine, heptylamine, octylamine, decylamine, dodecylamine, tetradecylamine, hexadecylamine octadecylamine, oleylamine, dimethylaminopropylamine, diethylaminopropylamine, dibutylaminopropylamine, dimethylaminoethylamine, diethylaminoethylamine, and dibutylaminoethylamine. Examples of suitable secondary amines include dimethylamine, diethylamine, dipropylamine, dibutylamine, diamylamine, dihexylamine, diheptylamine, methylethylamine, ethylbutylamine, diethylhexylamine, and ethylamylamine. The secondary amines may be cyclic amines such as

aminoethylmorpholine, aminopropylmorpholine, 1-(2-aminoethyl)pyrrolidone, piperidine, 1-(2-aminoethyl)piperidine, piperazine and morpholine. Examples of suitable aliphatic polyamines include tetraethylene pentamine, pentaethylene hexamine, diethylenetriamine, triethylenetetramine, and polyethyleneimine.

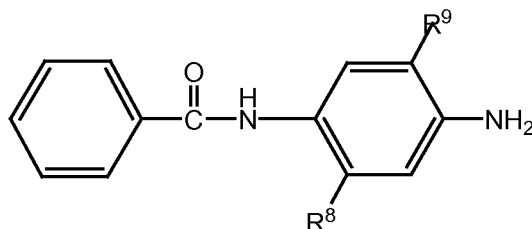
[0064] In another embodiment, the nitrogen containing group may be derived from an amine which is aromatic. Aromatic amines include those which can be represented by the general structure $\text{NH}_2\text{-Ar}$ or T-NH-Ar , where T may be alkyl or aromatic, Ar is an aromatic group, including nitrogen-containing aromatic groups and Ar groups including any of the following structures:



as well as multiple non-condensed or linked aromatic rings. In these and related structures, R^5 , R^6 , and R^7 can be independently selected from, among other groups disclosed herein, -H, -C_{1-18} alkyl groups, nitro groups, -NH-Ar , -N=N-Ar , -NH-CO-Ar , -OOC-Ar , -OOC-C_{1-18} alkyl, -COO-C_{1-18} alkyl, -OH, $\text{-O-(CH}_2\text{CH}_2\text{-O)}_n\text{C}_{1-18}$ alkyl groups, and $\text{-O-(CH}_2\text{CH}_2\text{O)}_n\text{Ar}$ (where n is 0 to 10).

[0065] Exemplary aromatic amines include those amines wherein a carbon atom of the aromatic ring structure is attached directly to the amino nitrogen. The aromatic amines may be monoamines or polyamines. The aromatic ring may be a mononuclear aromatic ring (i.e., one derived from benzene) but can include fused aromatic rings, especially those derived from naphthalene. Examples of aromatic amines include aniline, N-alkylanilines, such as N-methylaniline and N-butaniline, di-(para-methylphenyl)amine, 4-aminodiphenylamine, N,N-dimethylphenylenediamine, naphthylamine, 4-(4-nitrophenyl-azo)aniline (disperse orange 3), sulfamethazine, 4-phenoxyaniline, 3-nitroaniline, 4-aminoacetanilide (N-(4-aminophenyl)acetamide), 4-amino-2-hydroxy-benzoic acid phenyl ester (phenyl amino salicylate), N-(4-amino-phenyl)-benzamide, various benzylamines such as 2,5-dimethoxybenzylamine, 4-phenylazoaniline, and combinations and substituted versions of these. Other examples include para-ethoxyaniline, para-dodecylaniline, cyclohexyl-substituted naphthylamine, and thienyl-substituted aniline. Examples of other suitable aromatic amines include amino-substituted aromatic compounds and amines in which the amine nitrogen is a part of an aromatic ring, such as 3-aminoquinoline, 5-aminoquinoline, and 8-aminoquinoline. Also included are aromatic amines, such as 2-aminobenzimidazole, which contains one secondary amino group attached directly to the aromatic ring and a primary amino group attached to the imidazole ring. Other amines include N-(4-anilinophenyl)-3-aminobutanamide, and 3-amino propyl imidazole, and 2,5-dimethoxybenzylamine.

[0066] Additional aromatic amines and related compounds are disclosed in U.S. Patent Nos. 6,107,257 and 6,107,258. Examples of these include aminocarbazoles, benzoimidazoles, aminoindoles, aminopyrroles, amino-indazolinones, aminopyrimidines, mercaptotriazoles, aminophenothiazines, aminopyridines, aminopyrazines, aminopyrimidines, pyridines, pyrazines, pyrimidines, aminothiadiazaoles, aminothiathiadiazaoles, and aminobenzotriazoles. Other suitable amines include 3-amino-N-(4-anilinophenyl)-N-isopropyl butanamide, and N-(4-anilinophenyl)-3-((3-aminopropyl)-(cocoalkyl)amino)butanamide. Other aromatic amines which can be used include various aromatic amine dye intermediates containing multiple aromatic rings linked by, for example, amide structures. Examples include materials of the general structure:



and isomeric variations thereof, where R^8 and R^9 are independently alkyl or alkoxy groups such as methyl, methoxy, or ethoxy.

[0067] In one instance, R^8 and R^9 are both -OCH_3 and the material is known as Fast Blue RR [CAS# 6268-05-9]. In another instance, R^9 is -OCH_3 and R^8 is -CH_3 , and the material is known as Fast Violet B [99-21-8]. When both R^8 and

R⁹ are ethoxy, the material is known as Fast Blue BB [120-00-3]. U.S. Patent No. 5,744,429 discloses other aromatic amine compounds useful herein, particularly aminoalkylphenothiazines. N-aromatic substituted acid amide compounds, such as those disclosed in U.S. Pub. No. 20030030033, may also be used herein. Suitable aromatic amines include those in which the amine nitrogen is a substituent on an aromatic carboxylic compound, that is, the nitrogen is not sp²

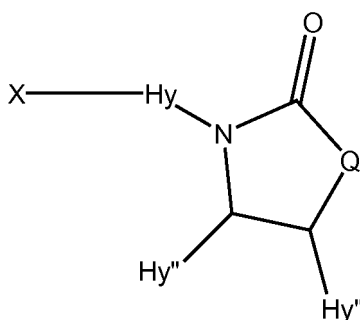
hybridized within an aromatic ring.
[0068] The aromatic amine may have an N-H group capable of condensing with the pendent carbonyl-containing group. Certain aromatic amines are commonly used as antioxidants. Examples of these are alkylated diphenylamines, such as nonyldiphenylamine and dinonyldiphenylamine. To the extent that these materials will condense with the carboxylic functionality of the polymer chain, they are also suitable for use herein. However, it is believed that the two aromatic groups attached to the amine nitrogen reduce its reactivity. Thus, suitable amines include those having a primary nitrogen atom (-NH₂) or a secondary nitrogen atom in which one of the hydrocarbyl substituents is a relatively short chain alkyl group, e.g., methyl. Among such aromatic amines are 4-phenylazoaniline, 4-aminodiphenylamine (ADPA), 2-aminobenzimidazole, and N,N-dimethylphenylenediamine. Some of these and other aromatic amines may also impart antioxidant performance to the copolymers, in addition to dispersancy and other properties.

[0069] In one embodiment, the amine component of the copolymer further includes an amine having at least two N-H groups capable of condensing with the carboxylic functionality of the copolymer. This material is referred to hereinafter as a "linking amine" as it can be employed to link together two of the copolymers containing the carboxylic acid functionality. It has been observed that higher molecular weight materials may provide improved performance, and this is one method to increase the material's molecular weight. The linking amine can be either an aliphatic amine or an aromatic amine; if it is an aromatic amine, it is considered to be in addition to and a distinct element from the aromatic amine described above, which typically will have only one condensable or reactive NH group, in order to avoid excessive crosslinking of the copolymer chains. Examples of such linking amines include aromatic amines having three or more aromatic rings, such as formaldehyde coupled ADPA or similar type aromatic amines disclosed in WO2011146692 to the Lubrizol Corporation, published November 24, 2011, for example, at paragraphs [0068 to [0080]. Other examples of linking amines can include ethylenediamine, phenylenediamine, and 2,4-diaminotoluene; others include propylenediamine, hexamethylenediamine, and other, α - ω -polymethylenediamines. The amount of reactive functionality on such a linking amine can be reduced, if desired, by reaction with less than a stoichiometric amount of a blocking material such as a hydrocarbyl-substituted succinic anhydride.

[0070] In one embodiment, the amine includes nitrogen-containing compounds capable of reacting directly with a copolymer backbone. Examples of suitable amines include N-*p*-diphenylamine, 4-anilinophenyl methacrylamide, 4-anilinophenyl maleimide, 4-anilinophenyl itaconamide, acrylate and methacrylate esters of 4-hydroxydiphenylamine, and the reaction product of *p*-aminodiphenylamine or *p*-alkylaminodiphenylamine with glycidyl methacrylate.

[0071] In one embodiment, the exemplary copolymer provides for oxidation control. Typically, the copolymer with oxidation control contains an incorporated residue of an amine-containing compound such as morpholines, pyrrolidinones, imidazolidinones, amino amides (such as acetamides), β -alanine alkyl esters, and mixtures thereof. Examples of suitable nitrogen-containing compounds include 3-morpholin-4-yl-propylamine, 3-morpholin-4-yl-ethylamine, β -alanine alkyl esters (typically alkyl esters have 1 to 30, or 6 to 20 carbon atoms), or mixtures thereof.

[0072] In one embodiment, the compounds based on imidazolidinones, cyclic carbamates or pyrrolidinones may be derived from a compound of general structure:



wherein

X = -OH or NH₂;

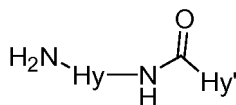
Hy'' is hydrogen, or a hydrocarbyl group (typically alkyl, or C₁₋₄-, or C₂- alkyl);

Hy is a hydrocarbylene group (typically alkylene, or C₁₋₄-, or C₂- alkylene);

Q = >NH, >NR, >CH₂, >CHR, >CR₂, or -O- (typically >NH, or >NR) and R is C₁₋₄ alkyl.

[0073] In one embodiment, the imidazolidinone includes 1-(2-amino-ethyl)-imidazolidin-2-one (may also be called aminoethylethyleneurea), 1-(3-aminopropyl)-imidazolidin-2-one, 1-(2-hydroxy-ethyl)-imidazolidin-2-one, 1-(3-amino-propyl)-pyrrolidin-2-one, 1-(3-amino-ethyl)-pyrrolidin-2-one, or mixtures thereof.

[0074] In one embodiment, the amide may be represented by the general structure:



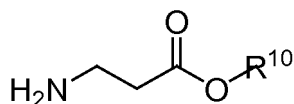
wherein:

Hy is a hydrocarbylene group (typically alkylene, or C_{1-4} -, or C_2 - alkylene); and

Hy' is a hydrocarbyl group (typically alkyl, or C_{1-4} -, or methyl).

[0075] Examples of a suitable amide include N-(2-amino-ethyl)-acetamide, or N-(2-amino-propyl)-acetamide.

[0076] In one embodiment, the β -alanine alkyl esters may be represented by the general structure:



wherein:

R^{10} is an alkyl group having 1 to 30, or 6 to 20 carbon atoms.

[0077] Examples of suitable β -alanine alkyl esters include β -alanine octyl ester, β -alanine decyl ester, β -alanine 2-ethylhexyl ester, β -alanine dodecyl ester, β -alanine tetradecyl ester, or β -alanine hexadecyl ester.

[0078] In one embodiment, the copolymer may be reacted with an amine selected from the group consisting of 1-(2-amino-ethyl)-imidazolidin-2-one, 4-(3-aminopropyl)morpholine, 3-(dimethylamino)-1-propylamine, N-phenyl-*p*-phenylenediamine, N-(3-aminopropyl)-2-pyrrolidinone, aminoethyl acetamide, β -alanine methyl ester, 1-(3-aminopropyl) imidazole, and mixtures thereof.

[0079] In one embodiment, the copolymer may be reacted with an amine-containing compound selected from morpholines, imidazolidinones, and mixtures thereof. In one embodiment, the nitrogen-containing compound is selected from 1-(2-aminoethyl)imidazolidinone, 4-(3-aminopropyl)morpholine, 3-(dimethylamino)-1-propylamine, N-phenyl-*p*-phenylenediamine, N-(3-aminopropyl)-2-pyrrolidinone, aminoethyl acetamide, β -alanine methyl ester, 1-(3-aminopropyl) imidazole, and combinations thereof.

[0080] The ester group and/or nitrogen containing group may be sufficient to provide 0.01 wt. % to 1.5 wt. % (or 0.02 wt. % to 0.75 wt. %, or 0.04 wt. % to 0.25 wt. %) nitrogen to the copolymer.

E. Formation of the Esterified Copolymer (A)

[0081] The esterified copolymer (A) can be formed by a method which includes:

(1) reacting (i) a vinyl monomer and (ii) a carboxylic acid monomer comprising an α,β -ethylenically unsaturated dicarboxylic acid or derivative thereof, to form a copolymer backbone, wherein the carboxylic acid monomer optionally has ester groups,

(2) optionally, esterifying the copolymer backbone of step (1) to form an esterified copolymer, and

(3) optionally, reacting the copolymer of step (1) or (2) with an nitrogen-containing compound in an amount to provide an esterified copolymer with at least 0.01 wt. % nitrogen; and whereby the resulting copolymer (A) is esterified in at least one of (1), (2), and (3).

1. Formation of the Copolymer Backbone

[0082] The copolymer backbone of the esterified copolymer (A) may optionally be prepared in the presence of a free radical initiator, solvent, or mixtures thereof. It will be appreciated that altering the amount of initiator can alter the number average molecular weight and other properties of the exemplary copolymer.

[0083] The copolymer backbone may be prepared by reacting the carboxylic acid monomer with the vinyl monomer.

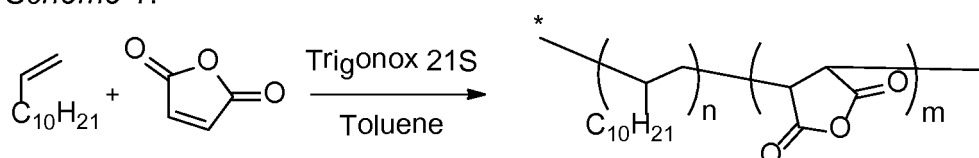
[0084] The solvent can be a liquid organic diluent. Generally, the solvent has as a boiling point that is high enough to

provide the required reaction temperature. Illustrative diluents include toluene, t-butyl benzene, benzene, xylene, chlorobenzene, various petroleum fractions boiling above 125°C, and mixtures thereof.

[0085] The free radical initiator can include one or more peroxy compounds, such as peroxides, hydroperoxides, and azo compounds which decompose thermally to provide free radicals. Other suitable examples are described in J. Brandrup and E. H. Immergut, Editor, "Polymer Handbook", 2nd edition, John Wiley and Sons, New York (1975), pages II-1 to II-40. Examples of a free radical initiator include those derived from a free radical-generating reagent, and examples include benzoyl peroxide, t-butyl perbenzoate, t-butyl metachloroperbenzoate, t-butyl peroxide, sec-butylperoxydicarbonate, azobisisobutyronitrile, t-butyl peroxide, t-butyl hydroperoxide, t-amyl peroxide, cumyl peroxide, t-butyl peroctoate, t-butyl-m-chloroperbenzoate, azobisisovaleronitrile, and mixtures thereof. In one embodiment, the free radical generating reagent is t-butyl peroxide, t-butyl hydroperoxide, t-amyl peroxide, cumyl peroxide, t-butyl peroctoate, t-butyl-m-chloroperbenzoate, azobisisovaleronitrile or mixtures thereof. Commercially available free radical initiators include classes of compound sold under the trademark Trigonox®-21 from Akzo Nobel.

[0086] An exemplary backbone polymer can be formed as follows: alpha-olefin is reacted with maleic anhydride in the presence of radical initiator and optionally in the presence of solvent. A solvent such as toluene can be used to lower backbone length by diluting the monomer concentration and through chain transfer to the benzylic protons. Scheme 1 shows an example where the alpha-olefin is 1-dodecene, the initiator is *tert*-butyl peroxy-2-ethylhexanoate (sold under the tradename Trigonox 21S by Akzo Nobel), and the solvent is toluene.

Scheme 1:



where *n* and *m* are independently at least 1, such as an integer from 1 to 10, or from 1 to 5, or from 1 to 3 in each segment of the copolymer (denoted by the two asterisks). As will be appreciated, the resulting backbone copolymer can have random variation of *n* and *m*.

[0087] Copolymerization of 1-dodecene and maleic anhydride in toluene under conventional free radical polymerization techniques leads to a low molecular weight copolymer.

2. Esterification of the Copolymer Backbone

[0088] Esterification (or transesterification, when the copolymer backbone already contains ester groups and those of a different type are desired) of the exemplary copolymer backbone can be accomplished by heating any of the copolymers described above and one or more desired alcohols and/or alkoxylates under conditions typical for effecting esterification. Such conditions include, for example, a temperature of at least 80°C, such as up to 150°C or higher, provided that the temperature is maintained below the lowest decomposition temperature of any component of the reaction mixture or products thereof. Water or lower alcohol is normally removed as the esterification proceeds. These conditions may optionally include the use of a substantially inert, normally liquid, organic solvent or diluent, such as mineral oil, toluene, benzene, xylene, or the like, and an esterification catalyst, such as one or more of toluene sulfonic acid, sulfuric acid, aluminum chloride, boron trifluoride-triethylamine, methane sulfonic acid, trifluoromethanesulfonic acid, hydrochloric acid, ammonium sulfate, and phosphoric acid. Further details of conducting the esterification can be found in U.S. Patent No. 6,544,935, at column 11.

[0089] In one embodiment, at least 2%, or at least 5%, or in certain embodiments 10% to 20%, of the carboxy functions of the copolymer remain un-converted to ester groups. Most of these will subsequently be converted to nitrogen-containing groups. An excess of alcohols and/or alkoxylates over the stoichiometric requirement for complete esterification of the carboxy functions may be used in the esterification process provided the ester content of the polymer remains in an appropriate range, e.g., within the 90 to 99% range. The excess of alcohols and alkoxylates or unreacted alcohols and alkoxylates need not be removed as such alcohols and alkoxylates can serve, for example, as diluent or solvent in the exemplary lubricating composition. Similarly, optional reaction media, e.g., toluene, need not be removed as they can similarly serve as diluent or solvent in the lubricating composition. In other embodiments, unreacted alcohols, alkoxylates and diluents are removed by well-known techniques, such as distillation.

3. Formation of Nitrogen-containing Groups on the Copolymer backbone

[0090] The nitrogen-containing compound may be directly reacted onto the copolymer backbone by grafting of the

amine, or other nitrogen-containing functional group, onto the copolymer backbone either (i) in a solution using a solvent, or (ii) under reactive extrusion conditions in the presence or absence of solvent. The amine-functional monomer may be grafted onto the copolymer backbone in multiple ways. In another embodiment, the grafting is carried out in solution or solid form through a free radical initiator. Solution grafting is a well-known method for producing grafted copolymers.

In such a process, reagents are introduced either neat or as solutions in appropriate solvents. The desired copolymer product may then be separated from the reaction solvents and/or impurities by appropriate purification steps.

[0091] In one embodiment, the nitrogen-containing compound may be directly reacted onto the copolymer backbone by free radical catalyzed grafting of the copolymer in a solvent, such as an organic solvent such as benzene, *t*-butyl benzene, toluene, xylene, hexane, or a combination thereof. The reaction may be carried out at an elevated temperature in the range of 100°C to 250°C or 120°C to 230°C, or 160°C to 200°C, e.g., above 160°C, in a solvent, such as a mineral lubricating oil solution containing, e.g., 1 to 50, or 5 to 40 wt. %, based on the initial total oil solution of the copolymer and optionally under an inert environment.

[0092] In one embodiment, the amine can have more than one nitrogen and can be selected from aliphatic amines and aromatic amines such that the R group attached to the amine that reacts with the carboxylic acid monomer derived groups contains at least one nitrogen atom, optionally substituted with hydrocarbyl groups. The hydrocarbyl groups can be selected from aliphatic, aromatic, cyclic, and acyclic hydrocarbyl groups. As the amine, one or more of the following may be used: 1-(2-amino-ethyl)-imidazolidin-2-one, 4-(3-aminopropyl)morpholine, 3-(dimethylamino)-1-propylamine, *N*-phenyl-*p*-phenylenediamine, *N*-(3-aminopropyl)-2-pyrrolidinone, aminoethyl acetamide, β -alanine methyl ester, and 1-(3-aminopropyl) imidazole.

[0093] In another embodiment, the nitrogen-containing compound may be a monomer that can polymerize with both the vinyl monomer and the carboxylic acid monomer such that the nitrogen-containing monomer is incorporated into the backbone. For example, a free radical catalyzed reaction is employed.

II. The Lubricating Composition

[0094] A lubricating oil composition described herein includes the esterified copolymer, an oil of lubricating viscosity (or "base oil"), and optionally other performance additives, such as other viscosity modifiers, dispersants, detergents, extreme pressure agents, antiwear agents, antiscuffing agents, pour point depressants, and corrosion inhibitors.

[0095] The lubricating composition may include the oil of lubricating viscosity as a minor or major component thereof, such as at least 5 wt. %, or at least 20 wt. %, or at least 30 wt. %, or at least 40 wt. %, or at least 60 wt. % of the lubricating composition.

[0096] The esterified copolymer (A) may be at least 0.1 wt. %, or at least 1 wt. %, or up to 10 wt. % or up to 5 wt. % of the lubricating composition.

[0097] The base oil (C) in the lubricating composition can be for example, from 10 to 95 wt. %, or 20-80 wt. %. Other performance additives (other than the viscosity modifier can be for example, from 0.2 to 40 wt. %, or 0.5-5 wt. % of the lubricating composition.

[0098] An exemplary lubricating composition includes at least 1 wt. % of the exemplary viscosity modifier and optionally, a pour point depressant at less than 1 wt. %, and optionally one or more performance additives.

III. Oil of Lubricating Viscosity

[0099] Suitable oils of lubricating viscosity (C) include natural and synthetic oils, oils derived from hydrocracking, hydrogenation, and hydrofinishing, unrefined, refined and re-refined oils, and mixtures thereof.

[0100] Unrefined oils are those obtained directly from a natural or synthetic source generally without (or with little) further purification treatment.

[0101] Refined oils are similar to the unrefined oils except they have been further treated in one or more purification steps to improve one or more properties. Purification techniques are known in the art and include solvent extraction, secondary distillation, acid or base extraction, filtration, percolation and the like.

[0102] Re-refined oils are also known as reclaimed or reprocessed oils, and are obtained by processes similar to those used to obtain refined oils and often are additionally processed by techniques directed to removal of spent additives and oil breakdown products.

[0103] Natural oils useful in as oils of lubricating viscosity include animal oils or vegetable oils (e.g., castor oil or lard oil), mineral lubricating oils, such as liquid petroleum oils and solvent-treated or acid-treated mineral lubricating oils of the paraffinic, naphthenic or mixed paraffinic-naphthenic types, and oils derived from coal or shale or mixtures thereof.

[0104] Synthetic lubricating oils useful as oils of lubricating viscosity include hydrocarbon oils, such as polymerized and copolymerized olefins (e.g., polybutylenes, polypropylenes, propyleneisobutylene copolymers); poly(1-hexenes), poly(1-octenes), poly(1-decenes), and mixtures thereof; alkyl-benzenes (e.g., dodecylbenzenes, tetradecylbenzenes, dinonylbenzenes, di-(2-ethylhexyl)-benzenes); polyphenyls (e.g., biphenyls, terphenyls, alkylated polyphenyls); alkylated

diphenyl ethers and alkylated diphenyl sulfides and the derivatives, analogs and homologs thereof, and mixtures thereof.

[0105] Other synthetic lubricating oils include polyol esters (such as Priolube®3970), diesters, liquid esters of phosphorus-containing acids (e.g., tricresyl phosphate, trioctyl phosphate, and the diethyl ester of decane phosphonic acid), or polymeric tetrahydrofurans. Synthetic oils may be produced by Fischer-Tropsch reactions and typically may be hydroisomerized Fischer-Tropsch hydrocarbons or waxes. In one embodiment oils may be prepared by a Fischer-Tropsch gas-to-liquid synthetic procedure as well as other gas-to-liquid (GTL) oils.

[0106] Oils of lubricating viscosity may also be defined as specified in the American Petroleum Institute (API) Base Oil Interchangeability Guidelines. The five base oil groups are as follows: Group I (sulfur content >0.03 wt. %, and/or <90 wt. % saturates, viscosity index 80-120); Group II (sulfur content ≤0.03 wt. %, and ≥90 wt. % saturates, viscosity index 80-120); Group III (sulfur content ≤0.03 wt. %, and ≥90 wt. % saturates, viscosity index ≥120); Group IV (all polyalphaolefins (PAOs)); and Group V (all others not included in Groups I, II, III, or IV). The exemplary oil of lubricating viscosity includes an API Group I, Group II, Group III, Group IV, Group V oil, or mixtures thereof. In some embodiments, the oil of lubricating viscosity is an API Group I, Group II, Group III, or Group IV oil, or mixtures thereof. In some embodiments, the oil of lubricating viscosity is an API Group I, Group II, or Group III oil, or mixtures thereof.

[0107] In some embodiments, at least 5 wt.%, or at least 10 wt.%, or at least 20 wt.%, or at least 40 wt. % of the lubricating composition is a polyalphaolefin (Group IV).

IV. Performance additives

[0108] In one embodiment, the lubricating composition or lubricant concentrate includes at least one performance additive (other than the viscosity modifier discussed above, which for convenience of the present description and claims, is not considered "performance additives"). The additional performance additive can include at least one of metal deactivators, detergents, dispersants, extreme pressure agents, antiwear agents, antioxidants, corrosion inhibitors, foam inhibitors, demulsifiers, pour point depressants, other viscosity modifiers, friction modifiers, seal swelling agents or mixtures thereof. In one embodiment, the performance additives may be used alone or in combination.

[0109] The total combined amount of the performance additives present ranges from 0 wt. % to 30 wt. %, or from 1 wt. % to 25 wt. %, or from 2 wt. % to 20 wt. %, or from 3 wt. % to 10 wt. % of the lubricating composition. Although one or more of the performance additives may be present, it is common for the performance additives to be present in different amounts relative to each other.

[0110] In the case of a lubricant concentrate (which may be combined with additional oil to form, in whole or in part, a finished lubricating composition), the ratio of the various performance additives to the oil of lubricating viscosity and/or to diluent oil include the ranges of 80:20 to 10:90 by weight.

[0111] Exemplary friction modifiers include fatty amines, esters such as borated glycerol esters, fatty phosphites, fatty acid amides, fatty epoxides, borated fatty epoxides, alkoxylated fatty amines, borated alkoxylated fatty amines, metal salts of fatty acids, fatty imidazolines, condensation products of carboxylic acids and polyalkylene-polyamines, amine salts of alkylphosphoric acids, molybdenum dithiocarbamate or mixtures thereof.

[0112] Exemplary antioxidants useful as oxidation inhibitors include sulfurized olefins, hindered phenols, diarylamines (such as, diphenylamines e.g., alkylated diphenylamines and phenyl-alpha-naphthylamines), molybdenum dithiocarbamates, and mixtures and derivatives thereof. Antioxidant compounds may be used alone or in combination.

[0113] Exemplary detergents include neutral or overbased, Newtonian or non-Newtonian, basic salts of alkali, alkaline earth and transition metals with one or more of a phenate, a sulfurized phenate, a sulfonate, a carboxylic acid, a phosphorus acid, a mono- and/or a di-thiophosphoric acid, a saligenin, an alkylsalicylate, a salixarate or mixtures thereof.

[0114] Exemplary dispersants are often known as ashless-type dispersants because, prior to mixing in a lubricating oil composition, they do not contain ash-forming metals and they do not normally contribute any ash forming metals when added to a lubricant and polymeric dispersants. Ashless type dispersants are characterized by a polar group attached to a relatively high molecular weight hydrocarbon chain. Typical ashless dispersants include succinimides, phosphonates, and combinations thereof.

[0115] Exemplary succinimide dispersants can include N-substituted long chain alkenyl succinimides as well as post-treated versions thereof. U.S. Patent Nos. 3,215,707; 3,231,587; 3,515,669; 3,579,450; 3,912,764; 4,605,808; 4,152,499; 5,071,919; 5,137,980; 5,286,823; 5,254,649 describe methods for forming such dispersants and their components. Post-treated dispersants include those further treated by reaction with materials such as urea, boron, thiourea, dimercaptothiadiazoles, carbon disulfide, aldehydes, ketones, carboxylic acids, hydrocarbon-substituted succinic anhydrides, nitriles, epoxides and phosphorus compounds.

[0116] For example such dispersants can be produced by reaction of a C3-C6 polyalkylene (e.g., polypropylene, polyisobutylene, polypentylene, polyheptylene, most often polyisobutylene) or derivative thereof (e.g., a chlorinated derivative) with a mono- or α,β unsaturated-dicarboxylic acid or anhydride thereof (such as maleic anhydride or succinic anhydride) to produce an acylated C3-C6 polyalkylene compound, which is reacted with an amine, such as a primary amine or a polyamine, such as a polyethylene amine, to produce the dispersant.

[0117] Other exemplary dispersants can be derived from polyisobutylene, an amine and zinc oxide to form a polyisobutylene succinimide complex with zinc.

[0118] Another class of ashless dispersant is acylated polyalkylene polyamines of the type described in U.S. Patent No. 5,330,667.

[0119] Another class of ashless dispersants is Mannich bases. Mannich dispersants are the reaction products of alkyl phenols with aldehydes (especially formaldehyde) and amines (especially polyalkylene polyamines). The alkyl group typically contains at least 30 carbon atoms.

[0120] Antiwear agents can include compounds such as metal thiophosphates, especially zinc dialkyldithiophosphates (ZDDP); phosphoric acid esters or salt thereof; phosphites; and phosphorus-containing carboxylic esters, ethers, and amides; antiscuffing agents including organic sulfides and polysulfides, such as benzyldisulfide, bis-(chlorobenzyl)disulfide, dibutyl tetrasulfide, di-tertiary butyl polysulfide, di-tert-butylsulfide, sulfurized Diels-Alder adducts, or alkyl sulphenyl N'N'-dialkyl dithiocarbamates.

[0121] Extreme Pressure (EP) agents that are soluble in the oil include sulfur- and chlorosulfur-containing EP agents, chlorinated hydrocarbon EP agents and phosphorus EP agents. Examples of such EP agents include chlorinated wax; sulfurized olefins (such as sulfurized isobutylene), organic sulfides and polysulfides such as dibenzyldisulfide, bis-(chlorobenzyl) disulfide, dibutyl tetrasulfide, sulfurized methyl ester of oleic acid, sulfurized alkylphenol, dimercaptothiadiazoles, sulfurized dipentene, sulfurized terpene, and sulfurized Diels-Alder adducts; phosphosulfurized hydrocarbons such as the reaction product of phosphorus sulfide with turpentine or methyl oleate; phosphorus esters such as the dihydrocarbon and trihydrocarbon phosphites, e.g., dibutyl phosphite, diheptyl phosphite, dicyclohexyl phosphite, pentylphenyl phosphite; dipentylphenyl phosphite, tridecyl phosphite, distearyl phosphite and polypropylene substituted phenol phosphite; metal thiocarbamates such as zinc dioctyldithiocarbamate and barium heptylphenol diacid; amine salts of alkyl and dialkylphosphoric acids or derivatives including, for example, the amine salt of a reaction product of a dialkyldithiophosphoric acid with propylene oxide and subsequently followed by a further reaction with P_2O_5 ; and mixtures thereof (as described, for example, in U.S. Patent No. 3,197,405).

[0122] Exemplary corrosion inhibitors can include octylamine octanoate, condensation products of dodeceny succinic acid or anhydride and a fatty acid such as oleic acid with a polyamine; metal deactivators including derivatives of benzotriazoles, thiadiazoles such as dimercaptothiadiazole and its derivatives, 1,2,4-triazoles, benzimidazoles, 2-alkyldithiobenzimidazoles, and 2-alkyldithiobenzothiazoles.

[0123] Suitable foam inhibitors include silicones, copolymers of ethyl acrylate and 2-ethylhexylacrylate which optionally further include vinyl acetate; and demulsifiers including polyethylene glycols, polyethylene oxides, polypropylene oxides and (ethylene oxide-propylene oxide) polymers.

[0124] Pour point depressants, including esters of maleic anhydride-styrene, polymethacrylates, polyacrylates or polyacrylamides; and seal swell agents including Exxon Necton-37™ (FN 1380) and Exxon Mineral Seal Oil (FN 3200); may also be used in the exemplary lubricating composition or lubricant concentrate.

[0125] In one embodiment, the exemplary lubricating composition or lubricant concentrate is free of sulfurized olefins and amine phosphates. By "free," it is meant that these ingredients, individually or in combination, amount to less than 0.001% of the lubricating composition.

V. Industrial Application

[0126] The method employing the esterified copolymer and the exemplary lubricating composition may be suitable for refrigeration lubricants, greases, gear oils, axle oils, drive shaft oils, traction oils, manual transmission oils, automatic transmission oils, metal working fluids, hydraulic oils, and internal combustion engine oils. It finds particular application as a vehicle driveline oil, such as a transmission oil. The exemplary lubricating composition may be supplied to a mechanical device, such as an automobile transmission, and used for lubrication during normal operation of the mechanical device.

[0127] In several embodiments a suitable lubricating composition includes the components present (on an actives basis) in ranges as shown Table I.

TABLE 1

	Embodiments (wt. % of lubricating composition)		
	A	B	C
Esterified Copolymer	1-25	1-15	1-10
Other Performance Additives	0-20	0.5-20	0.5-15
Oil of Lubricating Viscosity	98.89-51.0	98.45-63.0	98.4-73.2

(continued)

	Embodiments (wt. % of lubricating composition)		
	A	B	C
Total of components	100	100	100

[0128] The following examples provide an illustration of the invention. These examples are non-exhaustive and are not intended to limit the scope of the invention.

EXAMPLES

Example 1: Preparation of Esterified Copolymer

[0129] A 5L reaction flask was charged with 490.3 g of maleic anhydride (MAA), 841.6 g of 1-dodecene and 860.2 g of toluene. A molar ratio of the 1-dodecene: maleic anhydride was 1:1. The flask was fitted with a PTFE gasket and a 5-port flange lid equipped with an overhead stirrer, stirrer guide, thermocouple, nitrogen inlet (250 cm³/min nitrogen), silicon septa with peristaltic pump attached via cannula and a water condenser. The reaction flask and its contents were heated to 105 °C.

[0130] A mixture of 6.55 grams of Trigonox®C (a commercially available tert-butyl peroxybenzoate initiator from Akzo Nobel), and 124.5 g of toluene were mixed together and added to the reaction flask over 480 min. via the peristaltic pump. The reaction mixture was stirred at 105 °C for several hours.

[0131] An additional mixture of 19.65 grams of Trigonox®C (a commercially available tert-butyl peroxybenzoate initiator from Akzo Nobel), and 373.35 g of toluene were mixed together and added to the reaction flask over 480 min. via the peristaltic pump. The reaction mixture was stirred at 105°C for several hours.

[0132] The reaction flask was fitted with a Dean-Stark trap and the reaction mixture heated to 110 °C with stirring. 725 g of Alfol 810™ was added over 80 min and the resulting reaction mixture stirred for 1 hr. An additional 725 g of Alfol 810™ and 34.3 g of a 70% aqueous solution of methane sulfonic acid were added to the reaction flask over 80 min while gradually increasing the reaction temperature to 145°C. After several hours 72g of n-butanol and 17.2 g of 70% aqueous solution of methane sulfonic acid were added and stirred for 3 hr. An additional 72g of n-butanol was added and the reaction stirred for 2 hours. An additional 72g of n-butanol was added and the reaction stirred for several hours. An additional 72g of n-butanol was added and the reaction stirred for 3 to 4 hours. An additional 72g of n-butanol was added and the reaction stirred for 2 to 3 hours. Sufficient sodium hydroxide (50% aqueous sodium hydroxide solution) was added to quench the methane sulfonic acid and the mixture was stirred for 1 hour followed by the addition of 10.15 g of 3-(dimethylamino)-1-propylamine and stirred for an additional 2 hours. 309.1 g of Nexbase 3050 was added and stirred. The reaction flask was fitted for vacuum stripping and the resultant product vacuum-stripped (-26 in Hg) at 150°C and held for 3 hours. The vacuum was removed and the flask was cooled to 120°C.

[0133] The resulting reaction mixture was filtered twice using fax-5 and filter cloth. GPC run in tetrahydrofuran against polystyrene standards showed a M_w of 24,100, M_n of 16,000.

Example 2: Performance of Esterified Copolymer

[0134] Table 1 shows gear oil fluids prepared to meet SAE 75W-90 or 80W-90 grades targeting a kinematic viscosity of 17.5 cSt fluid at 100°C. Brightstock would not be capable of thickening PAO 4 sufficiently to make the particular viscosity and is formulated as a typical 80W-90 fluid with a pour point depressant (PPD) to enable improved cold flow properties.

TABLE 2

	Fluid 1	Fluid 2	Fluid 3	Fluid 4	Fluid 5
VIS Grade	SAE 75W-90	SAE 80W-90	SAE 75W-90	SAE 75W-90	SAE 75W-90
Base oil 1	43.72		62.06	45.23	68.14
Base oil 2		47			
Additive Package	10	10	10	10	10
PPD		2			

(continued)

	Fluid 1	Fluid 2	Fluid 3	Fluid 4	Fluid 5
5	VM1	46.28			
	VM2	41			
	VM3		27.94		
	VM4			44.77	
10	VM5				21.86
15	VM1=esterified copolymer prepared similar to Example 1 but having a lower molecular weight (Mw) of about 15,000-16,000 VM2=ExxonMobil™ 150BS Brightstock VM3=conventional poly(isobutylene) VM4=Uniroyal™ 100 cSt PAO VM5=ethylene propylene copolymer				

[0135] The fluids are compared in a number of key viscosity evaluations. Fluid 1 has the most favorable VI and HTHS. The viscosity reduction as temperature increases can be linked to VI but as shear rate increases the impact of the non-Newtonian behavior can manifest in differing levels of shear thinning at high shear rate. Comparison of fluid 3 and fluid 5 shows equivalent HTHS, equivalent KV100, but a substantial difference in VI Only Fluid 1 is according to the invention.

TABLE 3

	Fluid 1	Fluid 2	Fluid 3	Fluid 4	Fluid 5
25	VIS Grade	SAE 75W-90	SAE 80W-90	SAE 75W-90	SAE 75W-90
	KV40	102.6	186.3	127.2	112.8
30	KV100	17.55	17.57	17.63	16.92
	Viscosity Index (VI)	189	102	153	164
	HTHS	6.02	5.05	5.56	5.36

[0136] High shear rheology is also conducted on these fluids. The PSC Instruments USV (Ultra-Shear Viscometer) is used to determine the viscosity of the fluids at shear rates higher than the typical $1 \times 10^6 \text{ s}^{-1}$ as run in the HTHS using a tapered bearing or plug simulator (TBS or TPS). Similar to the traditional HTHS-capable instruments, USV uses a classic rotor-stator design to apply a very quick pulse to the rotor providing shear rates up to $10,000,000 \text{ s}^{-1}$. Due to the susceptibility of some polymer-containing solutions to be affected by permanent shear loss during testing, fresh sample is run at each individual shear rate.

[0137] The procedure for testing the samples is as follows: temperature and shear rate parameters are assigned in the computer. The application allows for multiple temperatures and shear rates to be assigned, but as stated individual temperatures and shear rates are used. Once the temperature and shear rate profile has been loaded, the new sample is loaded. Numerous flushing steps of the current test fluid are required before the sample (0.5 mL) is loaded into the test chamber. Once loaded and set to run, the motor and flywheel are set to spin to provide the appropriate shear rate. A clutch engages the rotor and three pulses, approximately five second apart, are applied to the sample. The viscosity is determined by calculating the average of the resistance on the rotor.

[0138] Figures 1 to 4 summarize the responses of the fluids to high shear from 50 to 150°C. At lower operational temperatures under low load the higher viscosity or thicker films can be detrimental to operational efficiency. Fluid 1 has lower viscosity at 50°C under high shear conditions than the comparative fluids which have an impact on the operational efficiency under low load and high shear rates.

[0139] At high temperatures fluid film support can be compromised and can negatively impact durability. For non-Newtonian fluids the rate of drop in viscosity will depend on the polymer type and Figure 2 shows that as temperature increased to 100°C (fluids normalized to KV at 100°C) fluid 5 shows the most substantial reduction in viscosity, fluid 2 shows the least reduction indicating the least and most Newtonian in nature respectively.

[0140] Figures 3 and 4 show the fluid response to high shear at 125°C and 150°C. At these temperatures viscosity is substantially lower than at 100°C. Thus, fluid film protection under these conditions can be compromised. Fluid 1 shows the lowest drop in viscosity at high temperature (>100°C) and high shear rate.

[0141] Unless otherwise indicated, each chemical or composition referred to herein should be interpreted as being a commercial grade material which may contain the isomers, by-products, derivatives, and other such materials which are normally understood to be present in the commercial grade. However, the amount of each chemical component is presented exclusive of any solvent or diluent oil, which may be customarily present in the commercial material, unless otherwise indicated. As used herein, the expression "consisting essentially of" permits the inclusion of substances that do not materially affect the basic and novel characteristics of the composition under consideration. As used herein any member of a genus (or list) may be excluded from the claims.

[0142] As used herein, the term "(meth) acrylic" and related terms includes both acrylic and methacrylic groups.

[0143] As used herein, the term "a primary alcohol branched at the β - or higher position" relates to an alcohol with branching at the 2- position or a higher position (e.g., 3-, or 4-, or 5-, or 6-, or 7- position, etc.).

[0144] As used herein, the term "hydrocarbyl substituent" or "hydrocarbyl group" is used in its ordinary sense, which is well-known to those skilled in the art. Specifically, it refers to a group having a carbon atom directly attached to the remainder of the molecule and having predominantly hydrocarbon character. Examples of hydrocarbyl groups include:

- a. hydrocarbon substituents, that is, aliphatic (e.g., alkyl or alkenyl), alicyclic (e.g., cycloalkyl, cycloalkenyl) substituents, and aromatic-, aliphatic-, and alicyclic-substituted aromatic substituents, as well as cyclic substituents wherein the ring is completed through another portion of the molecule (e.g., two substituents together form a ring);
 - b. substituted hydrocarbon substituents, that is, substituents containing non-hydrocarbon groups which, in the context of this invention, do not alter the predominantly hydrocarbon nature of the substituent (e.g., halo (especially chloro and fluoro), hydroxy, alkoxy, mercapto, alkylmercapto, nitro, nitroso, and sulfoxy);
 - c. hetero substituents, that is, substituents which, while having a predominantly hydrocarbon character, in the context of this invention, contain other than carbon in a ring or chain otherwise composed of carbon atoms; and
 - d. heteroatoms include sulfur, oxygen, nitrogen, and encompass substituents as pyridyl, furyl, thienyl and imidazolyl.
- In general, no more than two, in one aspect no more than one, non-hydrocarbon substituent will be present for every ten carbon atoms in the hydrocarbyl group; typically, there will be no non-hydrocarbon substituents in the hydrocarbyl group.

Claims

1. The use of an esterified copolymer to improve high-temperature high-shear (HTHS) performance measured according to ASTM D4683-10 of a lubricating oil, the esterified copolymer including a polymeric backbone comprising at least 70% of units (A1) derived from a vinyl monomer and units (A2) derived from a carboxylic acid monomer, wherein the molar ratio of (A1) to (A2) is from 1:3 to 3:1, the vinyl monomer comprising a vinyl aliphatic monomer, the carboxylic acid monomer comprising an ethylenically unsaturated carboxylic acid or derivative thereof, esterified with a primary alcohol, and no more than 30 mole% of units derived from one or more monomers capable of polymerizing with one or both of the vinyl monomer and carboxylic acid monomer, in a lubricating oil for a driveline device.
2. The use according to any previous claim to improve kinematic viscosity performance measured according to ASTM D445 of the lubricating oil as temperature increases.
3. The use according to any previous claim to improve kinematic viscosity performance measured according to ASTM D445 of the lubricating oil as shear rate increases.

Patentansprüche

1. Verwendung eines veresterten Copolymers zur Verbesserung der gemäß ASTM D4683-10 gemessenen Leistungsfähigkeit bei hoher Temperatur und hoher Scherung (HTHS) eines Schmieröls, wobei das veresterte Copolymer eine polymere Hauptkette mit mindestens 70 % Einheiten (A1), die sich von einem Vinylmonomer ableiten, und Einheiten (A2), die sich von einem Carbonsäuremonomer ableiten, wobei das Molverhältnis von (A1) zu (A2) 1:3 bis 3:1 beträgt, wobei das Vinylmonomer ein vinylaliphatisches Monomer umfasst, wobei das Carbonsäuremonomer eine ethylenisch ungesättigte Carbonsäure oder ein Derivat davon, die bzw. das mit einem primären Alkohol verestert ist, umfasst, und höchstens 30 Mol-% Einheiten, die sich von einem oder mehreren Monomeren, die mit dem Vinylmonomer und/oder dem Carbonsäuremonomer polymerisiert werden können, enthält, in einem Schmieröl für eine Antriebsstrangvorrichtung.

2. Verwendung nach einem der vorhergehenden Ansprüche zur Verbesserung der gemäß ASTM D445 gemessenen Leistungsfähigkeit hinsichtlich kinematischer Viskosität des Schmieröls mit zunehmender Temperatur.
3. Verwendung nach einem der vorhergehenden Ansprüche zur Verbesserung der gemäß ASTM D445 gemessenen Leistungsfähigkeit hinsichtlich kinematischer Viskosität des Schmieröls mit zunehmender Scherrate.

Revendications

1. Utilisation d'un copolymère estérifié pour améliorer la performance sous cisaillement élevé à haute température (HTHS) mesurée selon la méthode ASTM D4683-10 d'une huile lubrifiante, le copolymère estérifié comprenant une chaîne principale polymère comprenant au moins 70 % de motifs (A1) dérivés d'un monomère vinylique et de motifs (A2) dérivés d'un monomère acide carboxylique, le rapport molaire de (A1) à (A2) étant de 1:3 à 3:1, le monomère vinylique comprenant un monomère vinylique aliphatique, le monomère acide carboxylique comprenant un acide carboxylique à insaturation éthylénique ou un dérivé de celui-ci, estérifié avec un alcool primaire, et pas plus de 30 % en mole de motifs dérivés d'un ou plusieurs monomères capables de polymériser avec l'un ou l'autre du monomère vinylique ou du monomère acide carboxylique ou les deux, dans une huile lubrifiante pour un dispositif de chaîne cinématique.
2. Utilisation selon une quelconque revendication précédente pour améliorer la performance de viscosité cinématique mesurée selon la méthode ASTM D445 de l'huile lubrifiante lorsque la température augmente.
3. Utilisation selon une quelconque revendication précédente pour améliorer la performance de viscosité cinématique mesurée selon la méthode ASTM D445 de l'huile lubrifiante lorsque le taux de cisaillement augmente.

FIGURE 1

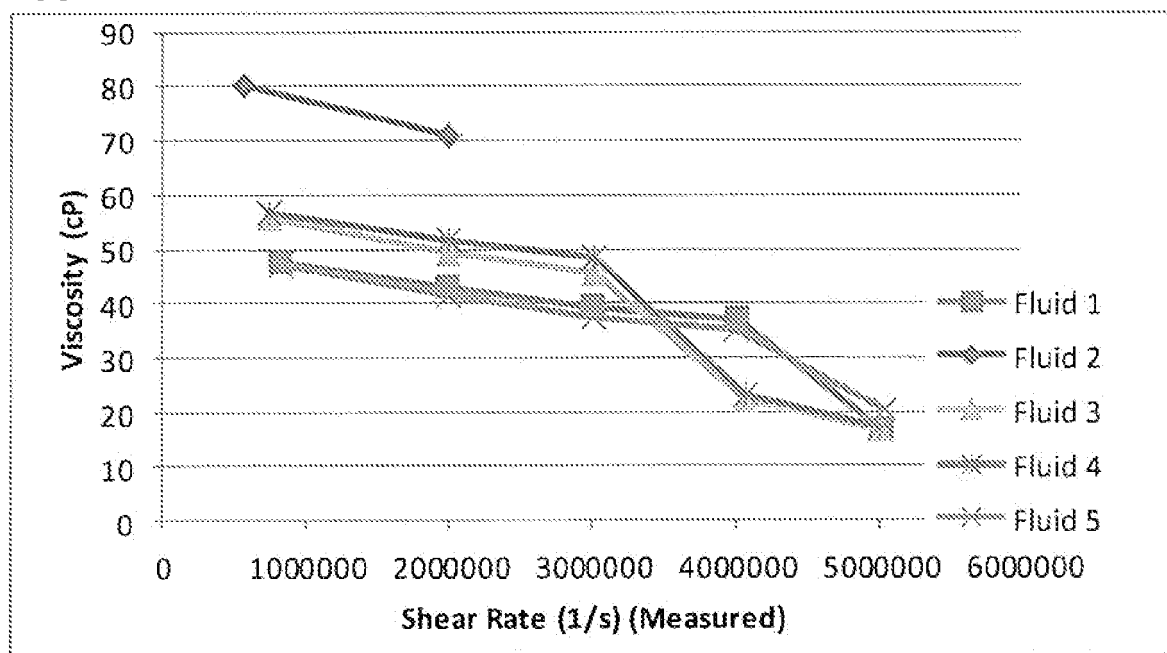


FIGURE 2

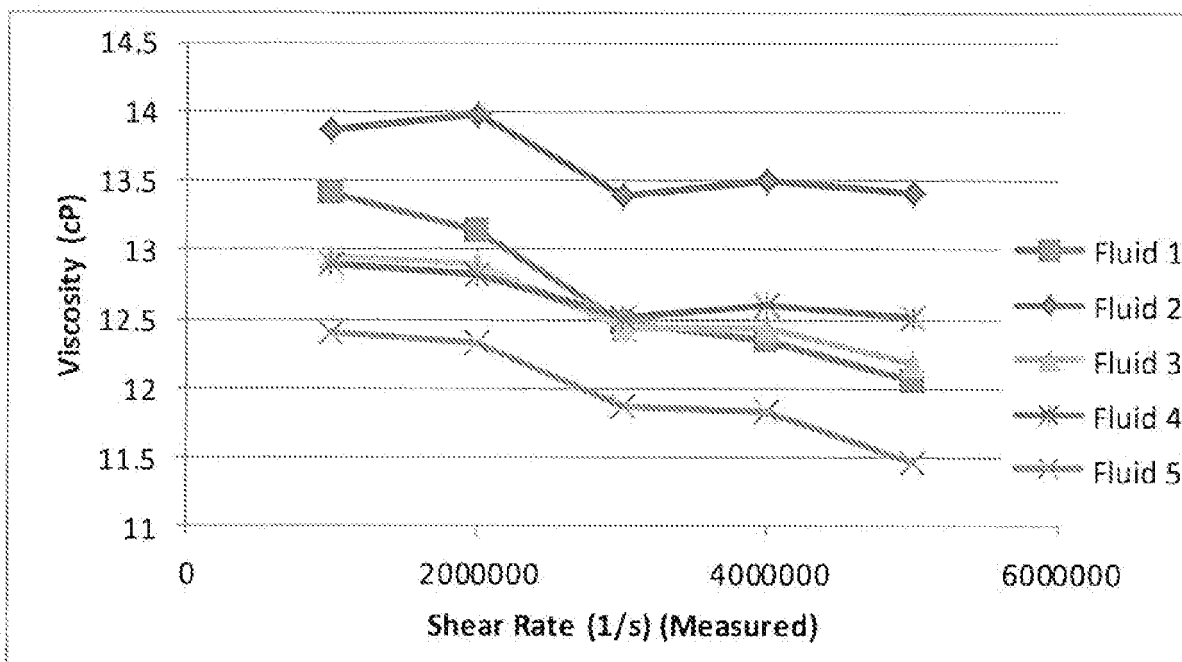


FIGURE 3

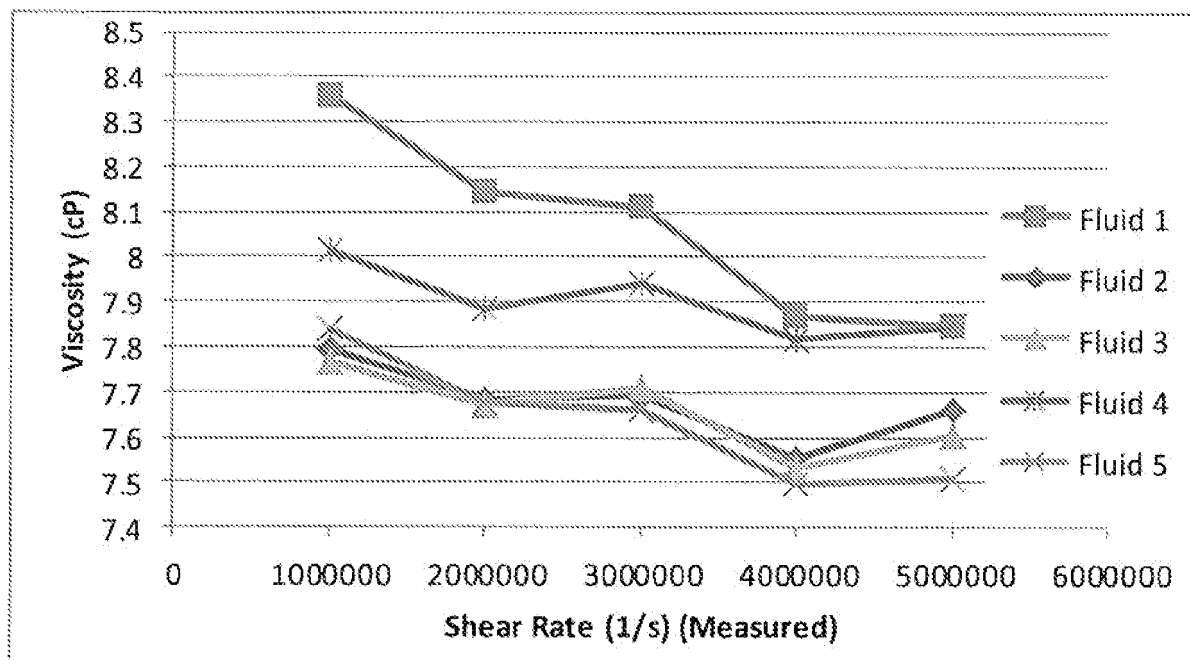
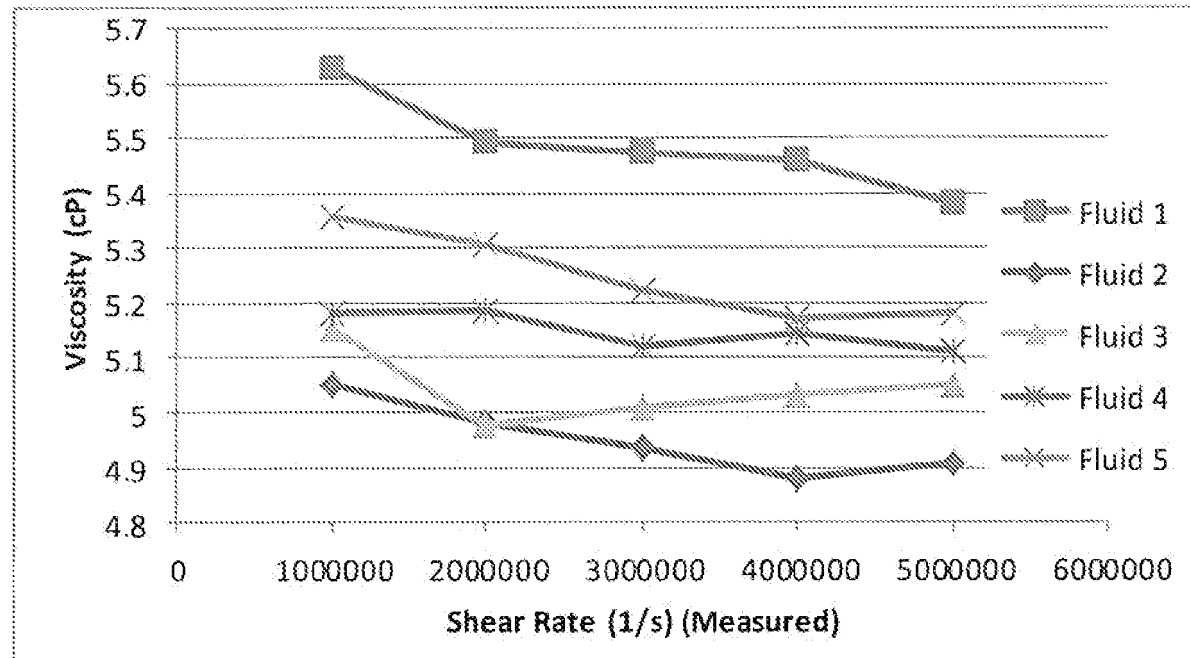


FIGURE 4



REFERENCES CITED IN THE DESCRIPTION

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