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(54) Title: PHOTOCROMIC MATERIAL

(57) Abstract: Photochromic dyes are disclosed. A photochromic dye can include a first photo-reactive group and a second photo-reactive group. A first photochromic reaction can be induced in the first photo-reactive group of the photochromic dye by radiation having a first wavelength, and a second photochromic reaction can be induced in the second photo-reactive group of the photochromic dye by radiation having a second wavelength. The second wavelength may be greater than the first wavelength.



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Description

Title of Invention: PHOTOCROMIC MATERIAL

Technical Field

- [1] This disclosure relates generally to photochromic materials.

Background Art

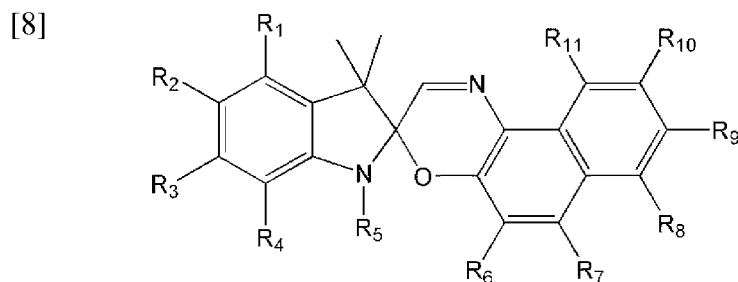
- [2] A number of photochromic materials are used in many industrial fields. The absorption spectra and thus the color of photochromic materials can change as a result of the absorption of electromagnetic radiation (e.g., visible or UV light).
- [3] Many photochromic materials possess color change properties due to the phenomenon of photochromism. Photochromism refers to a reversible, light-induced transformation of a chemical compound between at least two forms. The at least two forms inherently have different absorption spectra, so they exhibit different colors. A photochromic dye is a photochromic material exhibiting light-dependent color change properties. Absorption of light radiation can reversibly transform a photochromic dye between two forms having different absorption spectra, so as to exhibit different colors.

Disclosure of Invention

- [4] Summary of Invention

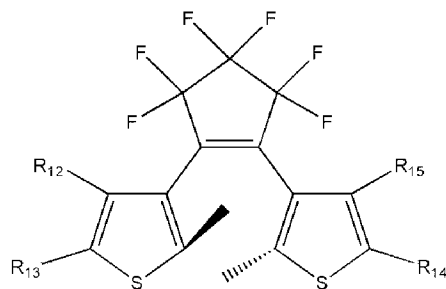
- [5] One embodiment of the disclosure describes a photochromic dye. The photochromic dye can include at least two photo-reactive groups. In one aspect, a first photo-reactive group is configured to undergo a first photochromic reaction in response to radiation having first wavelength, and a second photo-reactive group is configured to undergo a second photochromic reaction in response to radiation having second wavelength.
- [6] According to another embodiment, a photochromic dye is described. The photochromic dye can include at least two photo-reactive groups that are coupled to one another, wherein a first photo-reactive group and a second photo-reactive group are independently at least one structure selected from the group consisting of Formula I and II:

- [7] [Formula I]



- [9] [Formula II]

[10]



[11] wherein, R_1 to R_{15} independently represent hydrogen, a halogen, a substituted or unsubstituted C_{1-12} alkyl, a substituted or unsubstituted C_{1-12} alkoxy, a substituted or unsubstituted C_{1-12} alkenyl, a substituted or unsubstituted C_{1-12} conjugated alkyl, a substituted or unsubstituted aryl group, or a substituted or unsubstituted heteroaryl group, or the substituents together form an alkylene or alkenylene chain completing an aryl group.

[12] According to yet another embodiment, photochromic composition is described. In one aspect, the photochromic composition can include at least one material selected from the group consisting of a polymer, an oligomer, a monomer, or a mixture thereof, and at least one photochromic dye incorporated into at least a portion of the material. In one aspect, the at least one photochromic dye can include at least two photo-reactive groups that are coupled to one another, wherein a first photo-reactive group is configured to undergo a first photochromic reaction in response to radiation having first wavelength, and a second photo-reactive group is configured to undergo a second photochromic reaction in response to radiation having second wavelength.

[13] According to still yet another embodiment, an optical article is described. In one aspect, the optical article can include at least one optical article selected from the group consisting of ophthalmic elements, display elements, windows, mirrors, liquid crystal cell elements, and combinations thereof, and at least one photochromic dye incorporated into at least a portion of the optical article. In one aspect, the at least one photochromic dye can include at least two photo-reactive groups that are coupled to one another, wherein a first photo-reactive group is configured to undergo a first photochromic reaction in response to radiation having first wavelength, and a second photo-reactive group is configured to undergo a second photochromic reaction in response to radiation having second wavelength.

[14] The foregoing summary is illustrative only and is not intended to be in any way limiting. In addition to the illustrative aspects, embodiments, and features described above, further aspects, embodiments, and features will become apparent by reference to the following detailed description.

Mode for the Invention

[15] I. INTRODUCTION AND DEFINITIONS

- [16] In the following detailed description, reference is made to the accompanying structural formulas, which form a part hereof. In the structural formulas, similar symbols typically identify similar components, unless context dictates otherwise. The illustrative embodiments described in the detailed description, structural formulas, and claims are not meant to be limiting. Other embodiments may be utilized, and other changes may be made, without departing from the spirit or scope of the subject matter presented here.
- [17] A photochromic dye according to the present disclosure refers to a photochromic material (e.g., a chemical dye) that can exhibit two or more color changes as the wavelength of irradiation changes. A photochromic dye as described herein can be formed by linking two or more photo-reactive groups having different reactivities together in a single molecule. In one aspect, the two or more photo-reactive groups can be linked by a non-conjugated linkage.
- [18] The term “non-conjugated linkage” as used herein refers to a chemical linkage connecting two photochromic groups without a conjugation system. Since two photochromic groups are linked by the non-conjugated linkage, each photochromic group undergoes respective photochromic reactions in response to the specific wavelength of light and exhibit a specific color change.
- [19] A photochromic dye as described herein can be transformed from a colorless isomer to at least two colored isomers by the absorbance of light radiation. As such, the term “light radiation” as used herein refers to electromagnetic radiation such as, but not limited to, ultraviolet and visible radiation that is capable of causing a photo-reactive chromic group to transform from one form to another.
- [20] As used herein, the term “isomer” refers to isomers chemical compounds having the same molecular formula but different structural formulas. For example, a ring-opening or a ring-closing isomerization reaction alters the bonding structure and will generally alter the electronic properties of a molecule without altering the molecular formula. While the stability of photo-induced isomers varies from compound to compound, photo-induced isomers will generally back isomerize over time and/or in response to heat or absorbance of light radiation.
- [21] The term “photo-reactive group” as used herein means any substance having photo-reactive properties. Transformation of a photo-reactive group from one form to another can be induced by absorption of electromagnetic radiation (e.g., UV or visible light radiation). That is, a photo-reactive group can transform from one form to another form, each form having different absorption spectra so as to exhibit different colors, by absorbing light radiation.
- [22] In one aspect, a photo-reactive group can include a photochromic group. In one aspect, a photochromic reaction can be a color-change reaction that occurs in a photo-

tochromic group. As such, the transformation from one form to the other can include a photochromic reaction that causes or induces a change in the absorption spectrum of the dye such that the two forms exhibit different colors.

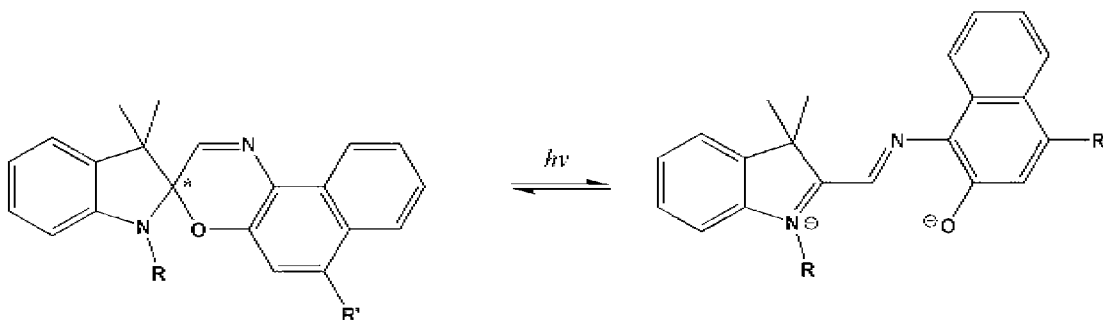
- [23] The term “photochromic moiety” as used herein refers to a part or portion of a photo-reactive group, in which a photochromic reaction occurs in order to undergo a reversible photochromic transformation from one form to another.
- [24] The term “ring-opening or closing reaction,” as used herein, refers to a ring-opening reaction or ring-closing reaction that happens when a photochromic group absorbs light radiation. According to the species of the photochromic group, a ring-opening reaction or a ring-closing reaction takes place. For example, spiropyrans and spirooxazines undergo a ring-opening reaction in response to light radiation, whereas diarylethenes and fulgides undergo a ring-closing reaction in response to light radiation.
- [25] The term “conjugation system” as used herein refers to a system where atoms covalently bond with alternating single and double bonds. In one aspect, a first photochromic reaction, as disclosed herein, can include a modification of a first conjugation system in a first photo-reactive group, and a second photochromic reaction can include modification of a second conjugation system in a second photo-reactive group. For example, absorbance of a photon of light having a given wavelength can induce the transformation of a conjugation system in a first photo-reactive group from a conjugation system having a first conjugation bond length to a conjugation system having a second and generally longer conjugation bond length. It is generally the case that the absorption spectra of molecules and thus their color changes as the conjugation bond length changes. In general, conjugation systems having less than eight conjugated double bonds only absorb light in the ultraviolet region and are colorless to the human eye. With the addition of additional conjugated double bonds (i.e., an increase in the conjugation bond length), the conjugation system absorbs photons of longer wavelength (and lower energy), eventually resulting in the absorbance of photons in the visual range of the human eye.
- [26] Thus, the photochromic reactions discussed herein can induce the transformation of a photochromic dye having at least two photo-reactive groups from a colorless isomer having a first conjugation bond length to at least two separate isomers having different and longer conjugation bond length systems that are capable of absorbing photons in the visual range of the human eye. As such, a colorless isomer of a photochromic dye can be transformed to an isomer having a first color (e.g., colorless to blue), or a colorless isomer of a photochromic dye can be transformed to an isomer having a second color (e.g., colorless to green). It is also possible for a colorless isomer of a photochromic dye to be transformed to an isomer having a third color (e.g., colorless to

cyan), wherein the colors produced by the first and second photo-reactive groups are blended in response to the first and second photo-reactive groups being isomerized more-or-less simultaneously.

- [27] For example, Scheme I below shows a ring-opening reaction of a spirooxazine in response to UV light irradiation. The spirooxazine shown in Scheme I includes a photo-reactive group and the compound can be termed “photochromic” because the spirooxazine changes its absorption properties, and thus its color, in response to UV irradiation. The spiro form of an oxazine is a colorless leuco dye having a conjugated oxazine and another conjugated aromatic portion separated by an sp³ hybridized spiro carbon (represented by *). After irradiation with UV light, the bond between the spiro carbon and the oxazine breaks, opening the ring. As a result, the spiro carbon switches to an sp² hybridization state and becomes planar, the aromatic group rotates, and an extended conjugation system is formed. The formation of the extended conjugation system allows the molecule to absorb photons of visible light, and therefore appear colorful. When the UV source is removed, the molecule will gradually relax to its ground state, the carbon-oxygen bond reforms, the spiro-carbon becomes sp³ hybridized again, and the molecule returns to its colorless state.

[28] [Scheme I]

[29]



- [30] Each photochromic group has its own absorption maximum which can induce a photochromic reaction. As used herein, the term “absorption maximum” refers to the specific wavelength at which an absorbance of a material is at its maximum. Extending conjugation length generally results in a shift of the absorption maximum to longer wavelengths. The absorbance of the material (or the amount of radiation absorbed by the material) can be determined using a spectrophotometer by exposing the material to incident radiation having a particular wavelength and intensity and comparing the intensity of radiation transmitted by the material to that of the incident radiation. For each wavelength tested, the absorbance (“A”) of the material is given by the following equation:

[31]

[32] $A = \log I_0/I$

[33]

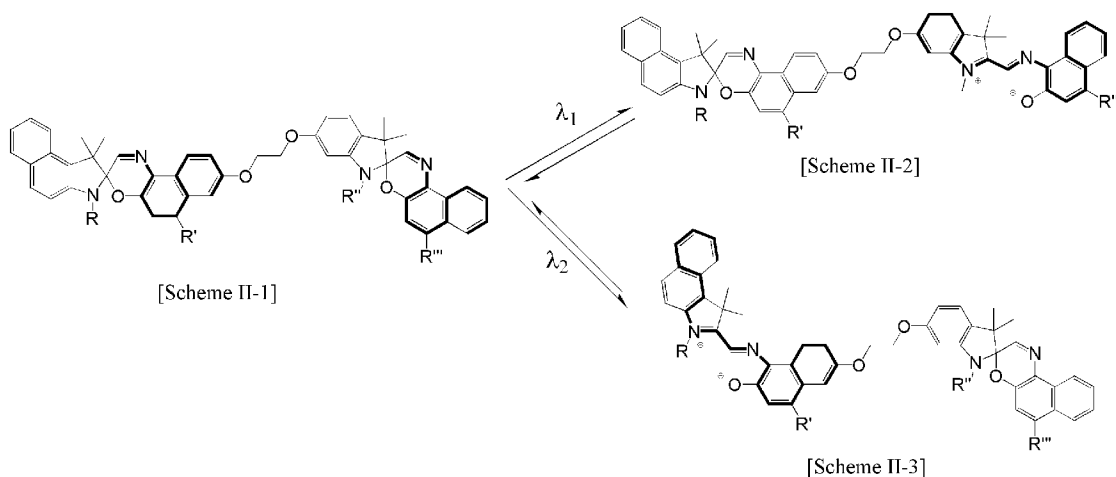
[34] wherein “ I_0 ” is the intensity of the incident radiation and “ I ” is the intensity of the transmitted radiation. An absorption spectrum for a material can be obtained by plotting the absorbance of the material vs. wavelength. By comparing the absorption spectrum of photochromic materials that were tested under the same conditions, that is using the same concentration and path length for electromagnetic radiation passing through the sample (e.g., the same cell length or sample thickness), an increase in the absorbance of one of the materials at a given wavelength can be seen as an increase in the intensity of the spectral peak for that material at that wavelength.

[35] As can be appreciated from the preceding discussion, when the photochromic dyes disclosed herein are exposed to radiation having the appropriate wavelength, the first or the second photo-reactive group, can undergo a structural rearrangement, e.g., a ring-opening or closing reaction, and/or a cis-trans isomerization, to alter the conjugation system in the first or the second photo-reactive group. Extending the conjugation system in response to the first or the second photochromic allows the dye to absorb photons of longer wavelength, which gives the dye its color.

[36] Scheme II shown below represents an example of two photochromic reactions according to one embodiment of the disclosure. The photochromic dye shown in Scheme II includes two photo-reactive groups connected by a non-conjugated linkage. The photochromic reactions shown in Scheme II are ring-opening reactions that occur in response to absorbance of different wavelengths of light (λ_1 and λ_2).

[37] [Scheme II]

[38]



[39] As shown in Scheme II-1, the two separate conjugation systems of the dye indicated by the bold lines each include six conjugated double bonds and are thus too short to for the isomer shown in Scheme II-1 to absorb visible light. As mentioned above, it is generally the case that conjugation systems having less than eight conjugated double bonds only absorb light in the ultraviolet region and are colorless to the human eye. However, the absorption spectra of molecules and thus their color changes as the con-

jugation bond length changes. That is, additional double bonds in the conjugation system allow the molecule to absorb photons of longer wavelength (and lower energy), which eventually results in the compound having a color that can be seen by the human eye.

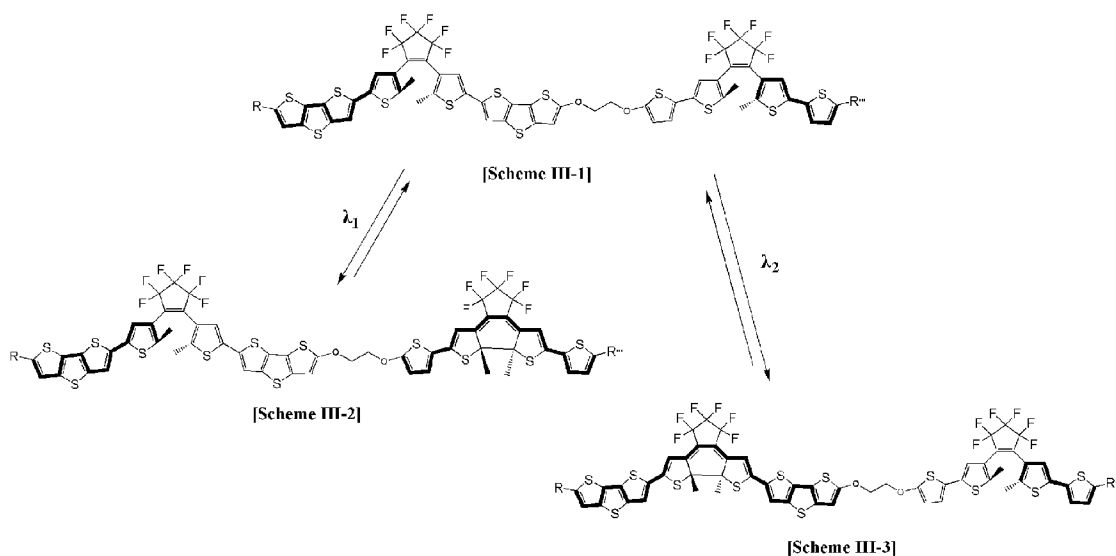
[40] Referring now to Schemes II-1 and II-2, when light having a first wavelength (λ_1) is absorbed by the first compound (Scheme II-1), the first photochromic group having an absorption maximum corresponding to the first wavelength undergoes the first photochromic reaction. Since the conjugation system is extended to ten conjugated double bonds (scheme II-2, bold line), the dye can absorb visible light to exhibit color.

[41] Referring now to Schemes II-1 and II-3, when light having a second wavelength (λ_2) is absorbed by the second compound (Scheme II-1), the second photochromic group having an absorption maximum corresponding to the second wavelength undergoes the second photochromic reaction. Since the conjugation system is extended to eleven conjugated double bonds (scheme II-3, bold line), the dye can absorb visible light to exhibit color. Because the conjugation bond length of the second photochromic group is different from the first photochromic group (i.e., eleven conjugated double bonds vs. ten conjugated double bonds), the isomer shown in Scheme II-3 will exhibit a different color that is different that the color exhibited by the isomer shown in Scheme II-2.

[42] Scheme III shown below represents another example of two photochromic reactions according to one embodiment of the disclosure. The photochromic dye shown in Scheme III includes two photo-reactive groups connected by a non-conjugated linkage. The photochromic reactions shown in Scheme III are ring-closing reactions that occur in response to absorbance of different wavelengths of light (λ_1 and λ_2).

[43] [Scheme III]

[44]



[45]

[46] As shown in scheme III, two diarylethene photo-reactive groups are coupled to one

another by a non-conjugated linkage. As in scheme II, the two separate conjugation systems of the dye indicated by the bold lines are too short for the isomer shown in Scheme III-1 to absorb visible light. However, when light having a different wavelength (λ_1 or λ_2) is absorbed by the dye (Scheme III-1), each photochromic group having an absorption maximum corresponding to each wavelength undergoes a photochromic reaction. Since the conjugation system is extended to eight or more conjugated double bonds (Scheme III-2 and III-3, bold line), the dye can absorb visible light to exhibit color. Because the conjugation bond lengths of the two photochromic groups differ from each other (i.e., eight conjugated double bonds in Scheme III-2 vs. twelve conjugated double bonds in Scheme III-3), the isomer shown in Scheme III-3 will exhibit a different color that is different than the color exhibited by the isomer shown in Scheme III-2.

[47]

[48] II. PHOTOCROMIC DYES

[49] Embodiments of the present disclosure relate to photochromic dyes. Photochromic dyes are dyes that exhibit the ability to reversibly change color in response to absorption of electromagnetic radiation. The photochromic dyes disclosed herein possess at least two photo-reactive groups, allowing the photochromic dyes to reversibly transform between at least two forms having at least two different colors.

[50] In one embodiment, the photochromic dyes disclosed herein can include at least two photo-reactive groups. In one aspect, a first photo-reactive group is configured to undergo a first photochromic reaction in response to radiation having first wavelength, and a second photo-reactive group is configured to undergo a second photochromic reaction in response to radiation having second wavelength.

[51] In one aspect, the first and the second photo-reactive groups are coupled to one another by a non-conjugated linkage. Non-limiting examples of a non-conjugated linkage that can be used to couple the first and the second photo-reactive groups include, but are not limited to, C_{1-12} alkylenes, C_{1-12} heteroalkylenes, C_{1-6} alkylenoxy compounds, and C_{1-6} alkylenedioxy compounds.

[52] In one aspect, of the photochromic dyes disclosed herein, the first photochromic reaction can include modification of a first conjugation system in the first photo-reactive group, and the second photochromic reaction can include modification of a second conjugation system in the second photo-reactive group. That is, for example, the first photochromic reaction can include lengthening the conjugation bond length of the conjugation system in the first photo-reactive group to the extent that the first photo-reactive group can absorb photons of visible light. Likewise, the second photochromic reaction can include lengthening the conjugation bond length of the conjugation system in the second photo-reactive group to the extent that the second photo-

reactive group can absorb photons of visible light. In some aspects, a modification of a conjugation system can include a shortening of a conjugation bond length such that, for example, a colored compound becomes colorless.

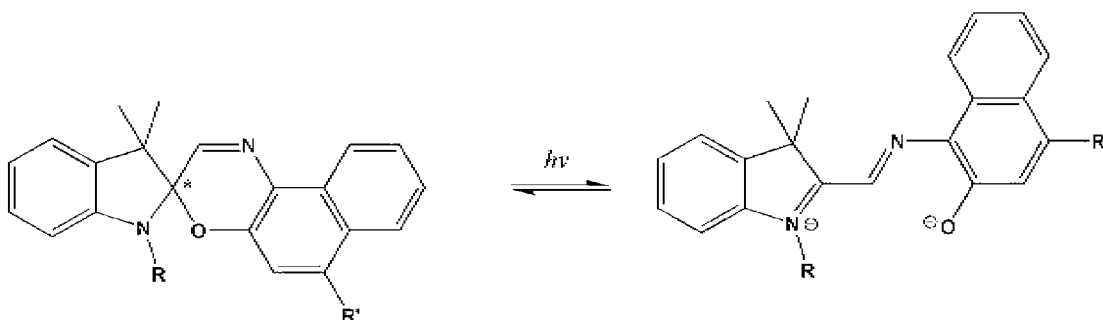
[53] As such, in one aspect, the first conjugation system can have a first conjugation bond length and the second conjugation system can have a second conjugation bond length that is different than the first conjugation bond length.

[54] Several non-limiting examples of isomerization reactions that can modify the first conjugation system in the first photo-reactive group and/or modify the second conjugation system in the second photo-reactive group independently include, but are not limited to at least one of a pericyclic reaction, a cis-trans isomerization, an intramolecular hydrogen transfer, an intramolecular group transfer, a dissociation process, an electron transfer reaction, a ring-opening reaction, or a ring closing reaction.

[55] To illustrate the formation of a conjugation system that allows a molecule to appear colorful, Scheme I will be referred to below.

[56] [Scheme I]

[57]



[58] In the absence of irradiation, the spirooxazine shown on the left-hand portion of Scheme I includes two separated portions having conjugated double bonds. That is, the conjugated systems of the oxazine on one end of the molecule and the benzene ring on the other end of the molecule are separated by the “spiro” carbon (indicated by *). The presence of the spiro linkage in between the conjugated systems prevents extended overlap of the pi-orbitals from one end of the molecule to the other with the longest conjugation system including six conjugated double bonds. As a result, the non-irradiated form of the spirooxazine is colorless.

[59] In response to irradiation ($h\nu$), however, the bond between the spiro-carbon and the oxazine breaks, and a conjugation system forms extending from one end of the molecule to the other. The photo-activated version of the spirooxazine has a conjugation system that includes ten conjugated double bonds, which allows the oxazine to absorb visible light and appear colorful.

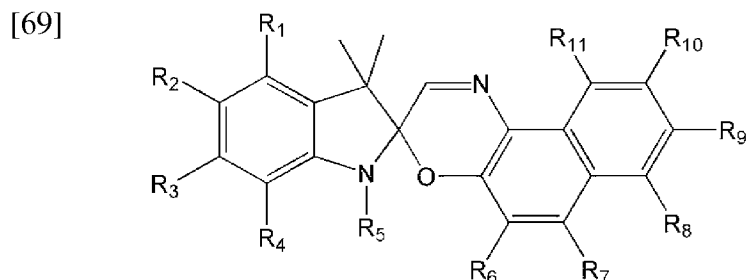
[60] In contrast to the spirooxazine discussed above, the photochromic dyes disclosed herein can include a first photo-reactive group having a first conjugation system and a

second photo-reactive group having a second conjugation system. In one aspect, the first conjugation system can be formed when the first photochromic reaction is induced, and the second conjugation system can be formed when the second photochromic reaction is induced. In one aspect, the first conjugation system has a first conjugation bond length and the second conjugation system has a second conjugation bond length that is longer than the first conjugation bond length.

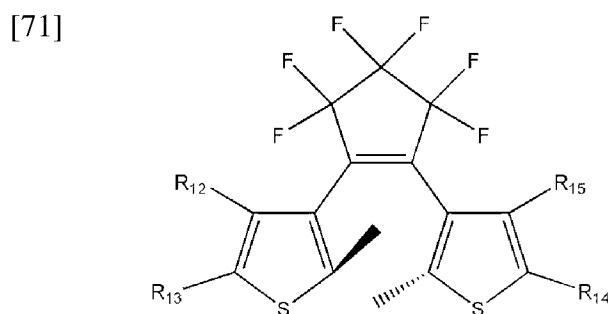
- [61] In one aspect, the first photo-reactive group and the second photo-reactive group have different absorption maxima. For example, according to a non-limiting embodiment, the difference in absorption maximum between the first photo-reactive group and the second photo-reactive group may be greater than or equal to 50 nm, greater than or equal to 100 nm, or greater than or equal to 150 nm. If the difference in absorption maximum between the photochromic groups is more than 50 nm, the first and the second photochromic reactions can be clearly distinguished according to the change of wavelength. For example, the second photochromic group may have an absorption maximum of 400 to 500 nm, 410 to 500 nm, 420 to 500 nm, 430 to 500 nm, 440 to 500 nm, or 450 to 500 nm when the first photochromic group has an absorption maximum of 250 to 350 nm, 250 to 340 nm, 250 to 330 nm, 250 to 320 nm, 250 to 310 nm, or 250 to 300 nm.
- [62] According to a non-limiting embodiment, the intensity of irradiation may be 10 to 30 mW/cm² when the photochromic dye sample according to the present disclosure is a liquid phase. However, when the photochromic dye sample is a film and the thickness of the film is about 1 μm, the intensity of irradiation may be 100 to 200 mW/cm². As the thickness of the film is increased, a stronger intensity of irradiation should be used.
- [63] In one aspect, the first photochromic reaction produces a first color and the second photochromic reaction produces a second color. For example, the first photochromic reaction can induce a transformation in the photochromic dye from colorless to blue and the second photochromic reaction can induce a transformation in the photochromic dye from colorless to green.
- [64] According to a non-limiting embodiment, the photochromic groups may be independently at least one selected from the group consisting of a spiropyran compound, a spirooxazine compound, a diarylethene compound and a fulgide compound.
- [65] Non-limiting examples of photochromic diarylethenes from which the photochromic group can be chosen include thiophene perfluoropentenes; benzothiophene perfluoropentenes; benzothiophene maleicanhydrides; benzothiophene cyanoethenes; and benzothiophene sulfone perfluoropentenes.
- [66] Non-limiting examples of photochromic fulgides from which the photochromic group can be chosen include the 3-furyl and 3-thienyl fulgides and fulgimides.
- [67] In one aspect, a photochromic dye, as disclosed herein can have at least two photo-

reactive groups coupled to one another, wherein a first photo-reactive group and a second photo-reactive group are independently at least one structure selected from the group consisting of Formula I and II:

[68] [Formula I]



[70] [Formula II]



[72] wherein, R_1 to R_{15} independently represent hydrogen, a halogen, a substituted or unsubstituted C_{1-12} alkyl, a substituted or unsubstituted C_{1-12} alkoxy, a substituted or unsubstituted C_{1-12} alkenyl, a substituted or unsubstituted C_{1-12} conjugated alkyl, a substituted or unsubstituted aryl group, or a substituted or unsubstituted heteroaryl group; or the substituents together form an alkylene or alkenylene chain completing an aryl group.

[73] In one aspect, the structures selected from the group of Formula I and Formula II are coupled to form the photochromic dye by a non-conjugated linkage. Several non-limiting examples of a non-conjugated linkage include, but are not limited to, C_{1-12} alkylenes, C_{1-12} heteroalkylenes, C_{1-12} alkoxys, C_{1-6} alkyleneoxy compounds, and C_{1-6} alkylenedioxy compounds, C_{1-6} phenyleneoxy compounds, and C_{1-6} phenylenedioxy compounds. It can be desirable to limit the length of the non-conjugated linkage because the photochromic reactions can be inhibited if the photochromic groups are able to self-associate intramolecularly can be self-associated so that the photochromic reaction can be inhibited.

[74] “Alkylene” refers to a linear or branched saturated divalent hydrocarbon radical. Examples of the alkylene group include, without limitation, methylene, ethylene, propylene, butylenes, and the like.

[75] “Heteroalkylene” refers to an alkylene chain as described above, in which one or more C-atoms have in each case been replaced by a heteroatom mutually inde-

pendently selected from the group comprising oxygen, sulfur and nitrogen (NH). Heteroalkylene groups can have 1, 2 or 3 heteroatom(s), particularly one heteroatom, selected from the group comprising oxygen, sulfur and nitrogen (NH) as the chain member(s). Heteroalkylene groups can be 2- to 12-membered, particularly 2- to 6-membered, and more particularly 2- or 3-membered.

- [76] “Alkyleneoxy” refers to a divalent group represented by the formula $-(\text{alkylene})-\text{O}-$ and includes, for example, a methyleneoxy, an ethyleneoxy, a propyleneoxy, a dimethylenedioxy, and the like.
- [77] “Alkylenedioxy” refers to a divalent group represented by the formula $-\text{O}-(\text{alkylene})-\text{O}-$ and includes, for example, a methylenedioxy, an ethylenedioxy, a propylenedioxy, a dimethylenedioxy, and the like.
- [78] Unless otherwise indicated, this disclosure uses definitions provided below.
- [79] “Substituted” groups are those in which one or more hydrogen atoms have been replaced with one or more non-hydrogen groups, provided that valence requirements are met and that a chemically stable compound results from the substitution.
- [80] “Alkyl” refers to straight chain and branched saturated hydrocarbon groups, generally having a specified number of carbon atoms (i.e., C_{1-6} alkyl refers to an alkyl group having 1, 2, 3, 4, 5, or 6 carbon atoms and C_{1-12} alkyl refers to an alkyl group having 1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11, or 12 carbon atoms). Examples of alkyl groups include, without limitation, methyl, ethyl, n-propyl, i-propyl, n-butyl, s-butyl, i-butyl, t-butyl, pent-1-yl, pent-2-yl, pent-3-yl, 3-methylbut-1-yl, 3-methylbut-2-yl, 2-methylbut-2-yl, 2,2,2-trimethyleth-1-yl, n-hexyl, and the like.
- [81] “Alkoxy” refers to alkyl-O-, alkenyl-O-, and alkynyl-O-. Examples of alkoxy groups include, without limitation, methoxy, ethoxy, n-propoxy, i-propoxy, n-butoxy, s-butoxy, t-butoxy, n-pentoxy, s-pentoxy, and the like.
- [82] “Alkenyl” refers to straight chain and branched hydrocarbon groups having one or more unsaturated carbon-carbon bonds, and generally having a specified number of carbon atoms. Examples of alkenyl groups include, without limitation, ethenyl, 1-propen-1-yl, 1-propen-2-yl, 2-propen-1-yl, 1-buten-1-yl, 1-buten-2-yl, 3-buten-1-yl, 3-buten-2-yl, 2-buten-1-yl, 2-buten-2-yl, 2-methyl-1-propen-1-yl, 2-methyl-2-propen-1-yl, 1,3-butadien-1-yl, 1,3-butadien-2-yl, and the like.
- [83] “Aryl” refers to monovalent and divalent aromatic groups, respectively, including 5- and 6-membered monocyclic aromatic groups that contain 0 to 4 heteroatoms independently selected from nitrogen, oxygen, and sulfur. Examples of monocyclic aryl groups include, without limitation, phenyl, pyrrolyl, furanyl, thiophenyl, thiazolyl, isothiazolyl, imidazolyl, triazolyl, tetrazolyl, pyrazolyl, oxazolyl, isooxazolyl, pyridinyl, pyrazinyl, pyridazinyl, pyrimidinyl, and the like. The aryl group also includes bicyclic groups, tricyclic groups, etc., including fused 5- and 6-membered

rings described above. Examples of multicyclic aryl groups include, without limitation, naphthyl, biphenyl, anthracenyl, pyrenyl, carbazolyl, benzoxazolyl, benzodioxazolyl, benzothiazolyl, benzoimidazolyl, benzothiophenyl, quinolinyl, isoquinolinyl, indolyl, benzofuranyl, purinyl, indoliziny, and the like. The aryl and arylene groups may be attached to a parent group or to a substrate at any ring atom, unless such attachment would violate valence requirements. Likewise, the aryl group may include one or more non-hydrogen substituents unless such substitution would violate valence requirements. Useful substituents include, without limitation, alkyl, alkenyl, alkynyl, haloalkyl, haloalkenyl, haloalkynyl, cycloalkyl, cycloalkenyl, alkoxy, cycloalkoxy, alkanoyl, cycloalkanoyl, cycloalkenoyl, alkoxycarbonyl, cycloalkoxycarbonyl, and halo, as defined above, and hydroxy, mercapto, nitro, amino, and alkylamino.

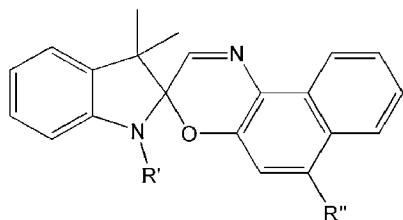
[84] In one non-limiting embodiment, R_2 and R_9 of Formula I independently represent hydrogen, a substituted or unsubstituted aryl group, or a substituted or unsubstituted heteroaryl group; or R_2 and R_3 , or R_9 and R_{10} together form an alkylene or alkenylene chain completing an aryl group.

[85] In one non-limiting embodiment, said R_2 and R_9 independently represent hydrogen or a phenyl; or R_2 and R_3 , or R_9 and R_{10} together form a benzene or naphthyl.

[86] In one non-limiting embodiment, the photochromic group of Formula I may be selected from the group consisting of the following Formula I-1 to I-9:

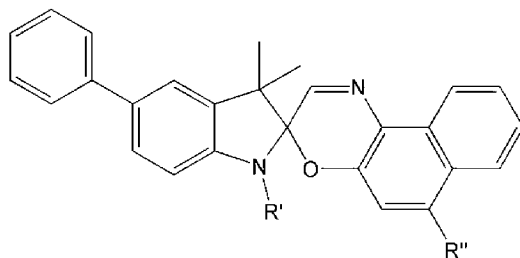
[87] [Formula I-1]

[88]



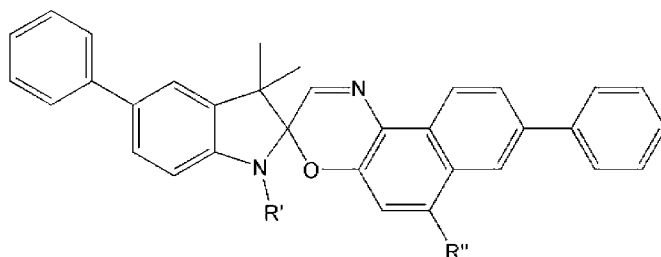
[89] [Formula I-2]

[90]



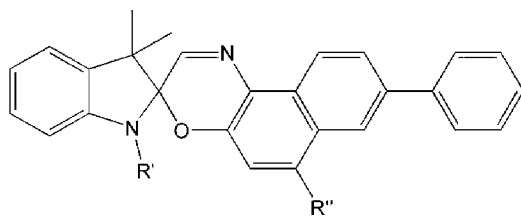
[91] [Formula I-3]

[92]



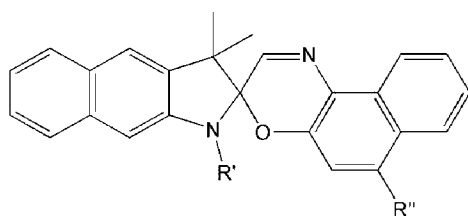
[93] [Formula I-4]

[94]



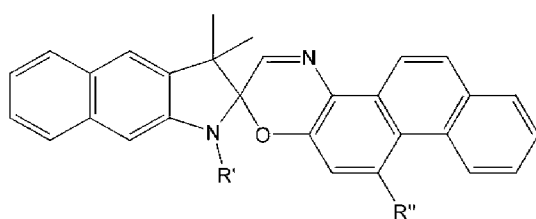
[95] [Formula I-5]

[96]



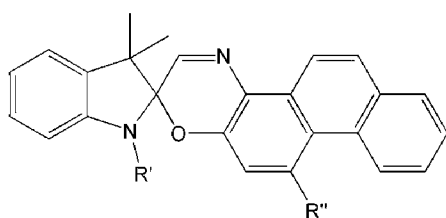
[97] [Formula I-6]

[98]



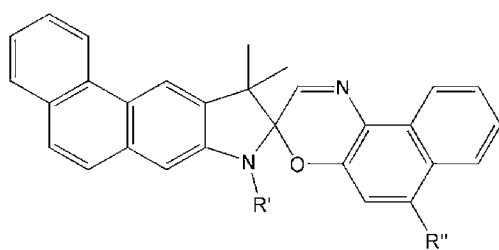
[99] [Formula I-7]

[100]



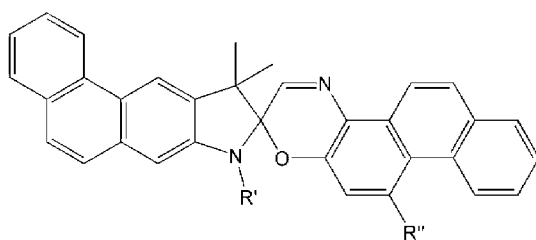
[101] [Formula I-8]

[102]



[103] [Formula I-9]

[104]



[105] wherein R' and R'' independently represent hydrogen, a halogen, a substituted or unsubstituted C₁₋₆ alkyl, or a substituted or unsubstituted C₁₋₆ alkoxy.

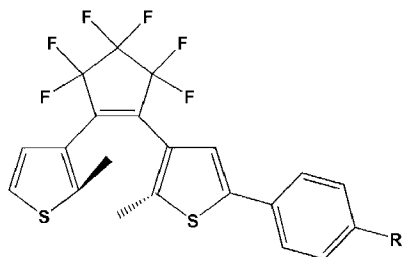
[106] In one non-limiting embodiment, R_{13} and R_{14} of Formula II independently represent hydrogen; and R_{12} and R_{15} independently represent hydrogen, a substituted or unsubstituted C_{6-12} conjugated alkyl, a substituted or unsubstituted aryl group, or a substituted or unsubstituted heteroaryl group.

[107] In one non-limiting embodiment, said R_{13} and R_{14} independently represent hydrogen; and R_{12} and R_{15} independently represent hydrogen, a phenyl or a substituted or unsubstituted dithienothiophenyl.

[108] In one non-limiting embodiment, the photochromic group of Formula II is selected from the group consisting of Formula II-1 to II-7:

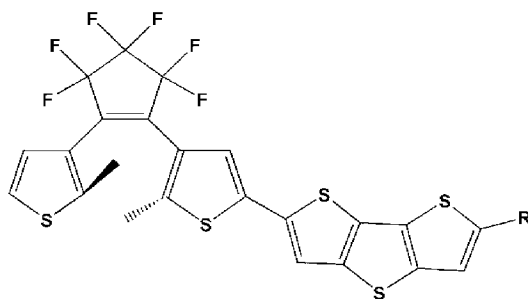
[109] [Formula II-1]

[110]



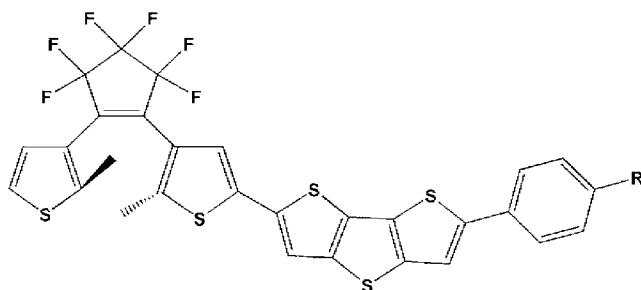
[111] [Formula II-2]

[112]



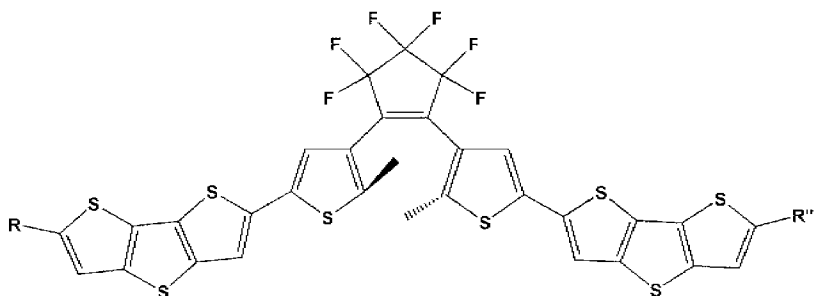
[113] [Formula II-3]

[114]



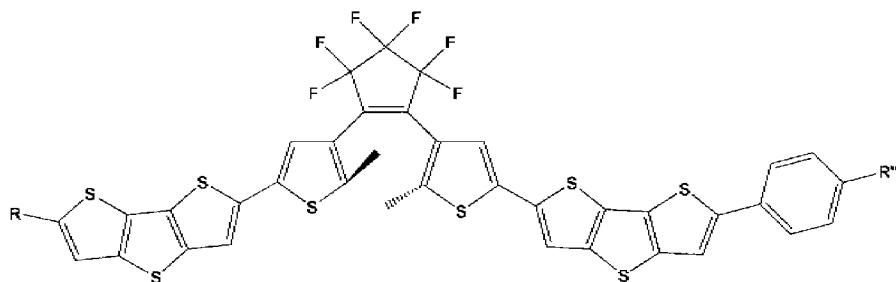
[115] [Formula II-4]

[116]



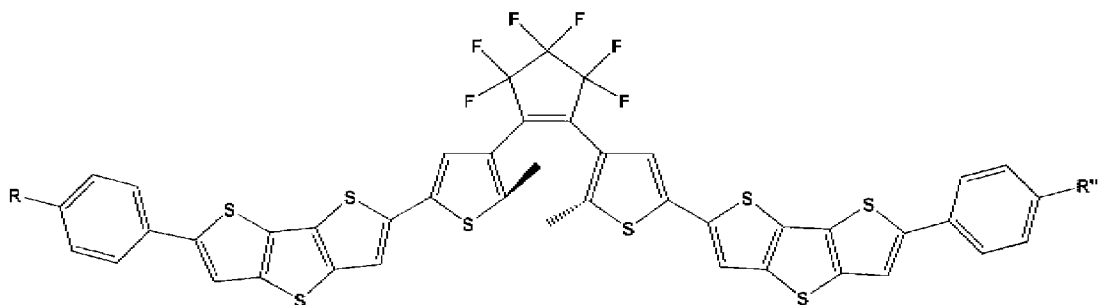
[117] [Formula II-5]

[118]



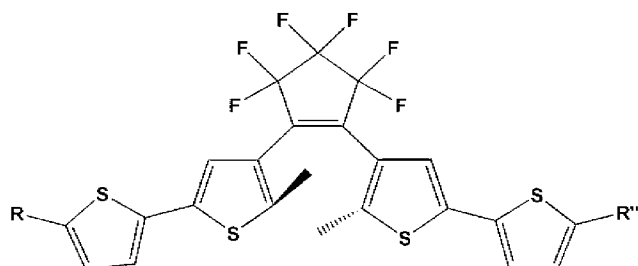
[119] [Formula II-6]

[120]



[121] [Formula II-7]

[122]



[123] Wherein, R and R''' independently represent hydrogen, or a substituted or unsubstituted C₁₋₁₂ alkyl.

[124]

[125] III. ARTICLES INCORPORATING PHOTOCHROMIC DYES

[126] This disclosure also provides a photochromic composition. The photochromic composition can include one or more of the photochromic dyes described previously herein, with the dye (or dyes) being incorporated into a portion of a first material. In illustrative embodiments, the first material may include, but is not limited to, a polymer, a biopolymer such as DNA/RNA or protein, an oligomer, a monomer or a mixture or combination thereof.

[127] The photochromic dye may be incorporated into a portion of a material, such as a polymer, oligomer or monomer to form a photochromic composition, which may be used, for example and without limitation, to form photochromic articles. As used herein the term "polymer" refers to homopolymers and copolymers as well as blends and other combinations thereof. As used herein the term "oligomer" refers to homooligomers and co-oligomers as well as blends and other combinations of two or more

monomer units that are capable of reacting with additional monomer units. As used herein the term “incorporated into” means physically and/or chemically combined with. For example, the photochromic dyes according to various non-limiting examples disclosed herein may be physically associated with a portion of a material, for example and without limitation, by mixing, combining, impregnating, inserting, or imbibing the photochromic dye into the material; and/or chemically combined with a portion of a material, for example and without limitation, by copolymerization or otherwise covalently or non-covalently bonding the photochromic dye to the material.

[128] According to various non-limiting embodiments disclosed herein, the photochromic dye may be incorporated into a portion of the material by at least one of blending and bonding the photochromic dye with the material. As used herein with reference to the incorporation of photochromic dyes into a material, the terms “blending” and “blended” mean that the photochromic dye is intermixed or intermingled with a portion of the material, but not bonded to the material. Further, as used herein with reference to the incorporation of photochromic dyes into a material, the terms “bonding” or “bonded” mean that the photochromic dye is linked to a portion of the material. For example, although not limiting herein, the photochromic dye may be linked to the material through a reactive substituent.

[129] According to one embodiment, the photochromic dye may be incorporated into a portion of a material at a rate from about 0.01 wt% to about 50 wt%, or about 0.1 wt% to about 45 wt%, or about 1 wt% to about 40 wt%, or about 2 wt% to about 35 wt%, or about 5 wt% to about 30 wt%, or about 10 wt% to about 20 wt%, or about 12 wt% to about 17 wt%, or about 15 wt%.

[130] According to one specific non-limiting embodiment, the material may be a polymeric material at least one selected from the group consisting of polyacrylates, poly-methacrylates, poly(C₁-C₁₂) alkyl methacrylates, polyoxy(alkylene methacrylates), poly(alkoxylated phenol methacrylates), cellulose acetate, cellulose triacetate, cellulose acetate propionate, cellulose acetate butyrate, poly(vinyl acetate), poly(vinyl alcohol), poly(vinyl chloride), poly(vinylidene chloride), poly(vinylpyrrolidone), poly((meth)acrylamide), poly(dimethyl acrylamide), poly(hydroxyethyl methacrylate), poly((meth)acrylic acid), thermoplastic polycarbonates, polyesters, polyurethanes, polythiourethanes, poly(ethylene terephthalate), polystyrene, poly(alpha methylstyrene), copoly(styrene-methylmethacrylate), copoly(styrene- acrylonitrile), polyvinylbutyral and polymers of members of the group consisting of polyol(allyl carbonate)monomers, mono-functional acrylate monomers, mono-functional methacrylate monomers, polyfunctional acrylate monomers, polyfunctional methacrylate monomers, diethylene glycol dimethacrylate monomers, diisopropenyl benzene monomers, alkoxylated polyhydric alcohol monomers and diallylidene pen-

taerythritol monomers.

- [131] In general, glass transition temperature (T_g) refers to the temperature at which a polymer or another material transitions from a rigid, solid state to a pliable state. For example, polymers having glass transition temperatures well below room temperature can be characterized as elastomers and/or viscous liquids, whereas polymers having glass transition temperatures well above room temperature can be characterized as rigid, structural polymers.
- [132] Because the transition of photochromic dyes from one state to another generally involves one or more structural rearrangements, the switching speed of photochromic dyes can be sensitive to the rigidity of the environment around the dye. As result, photochromic dyes switch most rapidly in solution and slowest in the rigid environment like in rigid polymer. One way that switching speeds can be increased is to incorporate the dyes into a low T_g polymer (e.g., a polymer having a glass transition temperature well below room temperature). The low T_g polymer can then be incorporated into a rigid material for structure while preserving the reduced switching speed afforded by the non-rigid material. Attaching flexible, low T_g polymers to photochromic dyes can allow them to switch much more rapidly in a rigid matrix. For example, some spirooxazines with siloxane polymers attached can switch at near solution-like speeds even though they are incorporated into a rigid matrix.
- [133] As such, in one embodiment, the dye can be incorporated into at least one material having a glass transition temperature less than about 0 °C, or less than about -5 °C, or less than about -10 °C, or less than about -15 °C, or less than about -20 °C, or less than about -25 °C, or less than about -30 °C, or less than about -35 °C, or less than about -40 °C, or less than about -45 °C, or less than about -50 °C, or less than about -55 °C, or less than about -60 °C.
- [134] In another embodiment, the dye can be incorporated into a second material, which can then be incorporated into the first material discussed above, where the second material has a glass transition temperature less than about 0 °C, or less than about -5 °C, or less than about -10 °C, or less than about -15 °C, or less than about -20 °C, or less than about -25 °C, or less than about -30 °C, or less than about -35 °C, or less than about -40 °C, or less than about -45 °C, or less than about -50 °C, or less than about -55 °C, or less than about -60 °C.
- [135] In yet another embodiment, photochromic dye can be incorporated into a second material selected from the group consisting of a polysiloxane and/or a polyacrylate having a glass transition temperature less than about 0 °C, or less than about -5 °C, or less than about -10 °C, or less than about -15 °C, or less than about -20 °C, or less than about -25 °C, or less than about -30 °C, or less than about -35 °C, or less than about -40 °C, or less than about -45 °C, or less than about -50 °C, or less than about -55 °C, or less

than about -60 °C. As discussed above, the second material including the dye can be incorporated into the first material.

- [136] A non-limiting example of a polyacrylate having a glass transition temperature less than about 0 °C is poly(butyl acrylate), which has a glass transition temperature of -49 °C. Polymerized siloxanes with organic side chains ($R \neq H$) are commonly known as silicones or as polysiloxanes. Representative examples include, but are not limited to, $[\text{SiO}(\text{CH}_3)_2]_n$ (polydimethylsiloxane) (“PDMS”) and $[\text{SiO}(\text{C}_6\text{H}_5)_2]_n$ (polydiphenylsiloxane). These compounds can be viewed as a hybrid of both organic and inorganic compounds. The organic side chains confer hydrophobic properties while the -Si-O-Si-O- backbone is purely inorganic.
- [137] PDMS is a widely used silicon-based organic polymer. Its applications range from contact lenses and medical devices to elastomers. PDMS is viscoelastic, meaning that at long flow times (or high temperatures), it acts like a viscous liquid, similar to honey. However at short flow times (or low temperatures) it acts like an elastic solid, similar to rubber. Due to its unique mechanical, chemical, and optical properties, PDMS is integrated into many optical devices. PDMS is optically clear at a wide range of wavelengths. In addition, the curing time and temperature used during cross-linking (generally with methyltrichlorosilane) can determine the refractive index (RI) of the bulk. Since the polymer can be easily molded, it has been used to form lenses and waveguides. Also, the effective RI and absorption spectrum of PDMS are changed when organic compounds are physically absorbed into the polymer.
- [138] In another embodiment, the present disclosure provides an optical article that is made of and/or incorporates one or more of the photochromic dyes or compositions disclosed herein. According to one aspect, an optical article can include at least one optical article, and at least one photochromic dye incorporated into at least a portion of the optical article.
- [139] As used herein the term “optical” means pertaining to or associated with light and/or vision. The optical elements according to various non-limiting embodiments disclosed herein may include, without limitation, ophthalmic elements, display elements, windows, mirrors, and liquid crystal cell elements. As used herein the term “ophthalmic” means pertaining to or associated with the eye and vision. Non-limiting examples of ophthalmic elements include corrective and non-corrective lenses, including single vision or multi-vision lenses, which may be either segmented or non-segmented multi-vision lenses (such as, but not limited to, bifocal lenses, trifocal lenses and progressive lenses), contact lenses, as well as other elements used to correct, protect, or enhance (cosmetically or otherwise) vision, including without limitation, magnifying lenses, protective lenses, visors, goggles, as well as, lenses for optical instruments (for example, cameras and telescopes). As used herein the term “display”

means the visible or machine-readable representation of information in words, numbers, symbols, designs or drawings. Non-limiting examples of display elements include screens, monitors, and security elements, such as security marks. As used herein the term “window” means an aperture adapted to permit the transmission of radiation therethrough. Non-limiting examples of windows include automotive and aircraft transparencies, windshields, filters, shutters, and optical switches. As used herein the term “mirror” means a surface that specularly reflects a large fraction of incident light. As used herein the term “liquid crystal cell” refers to a structure containing a liquid crystal material that is capable of being ordered. One non-limiting example of a liquid crystal cell element is a liquid crystal display.

[140] In one embodiment, the photochromic dye incorporated into the optical article can include a first photo-reactive group and a second photo-reactive group. In one aspect, a first photochromic reaction can be induced in the first photo-reactive group by radiation having a first intensity, and a second photochromic reaction can be induced in the second photo-reactive group by radiation having a second intensity. Any of the other photochromic dyes described herein can be incorporated in the optical article without limitation as to composition.

[141] In one aspect, the optical article can be fabricated from or further include at least one polymer selected from the group consisting of polyacrylates, polymethacrylates, polyalkylmethacrylates, polyoxy(alkylene methacrylates), poly(alkoxylated phenol methacrylates), cellulose acetate, cellulose triacetate, cellulose acetate propionate, cellulose acetate butyrate, poly(vinyl acetate), poly(vinyl alcohol), poly(vinyl chloride), poly(vinylidene chloride), poly(vinylpyrrolidone), poly((meth)acrylamide), poly(dimethyl acrylamide), poly(hydroxyethyl methacrylate), poly((meth)acrylic acid), thermoplastic polycarbonates, polyesters, polyurethanes, polythiourethanes, poly(ethylene terephthalate), polystyrene, poly(alpha methylstyrene), copoly(styrene-methylmethacrylate), copoly(styrene- acrylonitrile), polyvinylbutyral, and polymers of members of the group consisting of polyol(allyl carbonate)monomers, mono-functional acrylate monomers, mono-functional methacrylate monomers, poly-functional acrylate monomers, polyfunctional methacrylate monomers, diethylene glycol dimethacrylate monomers, diisopropenyl benzene monomers, alkoxylated polyhydric alcohol monomers and diallylidene pentaerythritol monomers.

[142] As described above, the dye incorporated in the optical article can be incorporated into at least one material having a glass transition temperature less than about 0 °C. In another aspect, the dye can be incorporated into a second material, which can then be incorporated into the first material discussed above, where the second material has a glass transition temperature less than about 0 °C. In yet another embodiment, photochromic dye can be incorporated into a second material selected from the group

consisting of a polysiloxane and/or a polyacrylate having a glass transition temperature less than about 0 °C.

- [143] Various non-limiting embodiments disclosed herein provide photochromic articles comprising a substrate and a photochromic dye according to any of the non-limiting embodiments discussed above connected to a portion of the substrate. As used herein, the term “connected to” means associated with, either directly or indirectly through another material or structure.
- [144] Non-limiting embodiments disclosed herein provide a method of preparing an optical element, comprising connecting a photochromic composition to at least a portion of a substrate by at least one of in-mold casting, coating and lamination.
- [145] For example, according to one non-limiting embodiment, the photochromic composition may be connected to at least a portion of a substrate by in-mold casting. According to this non-limiting embodiment, a coating composition comprising the photochromic composition, which may be a liquid coating composition or a powder coating composition, is applied to the surface of a mold. After that, the coated substrate is removed from the mold. Non-limiting examples of powder coatings in which the photochromic composition according to various non-limiting embodiments disclosed herein may be employed are set forth in U.S. Pat. No. 6,068,797, the disclosure of which is incorporated by reference herein in its entirety.
- [146] According to another non-limiting embodiment, the photochromic composition may be connected to a portion of a substrate by coating. Non-limiting examples of suitable coating methods include spin coating, spray coating (e.g., using a liquid or powder coating), curtain coating, roll coating, spin and spray coating, over-molding, and combinations thereof. For example, according to one non-limiting embodiment, the photochromic composition may be connected to the substrate by over-molding. According to this non-limiting embodiment, a coating composition comprising the photochromic composition (which may be a liquid coating composition or a powder coating composition as previously discussed) may be applied to a mold and then the substrate may be placed into the mold such that the substrate contacts the coating causing it to spread over the surface of the substrate. Thereafter, the coating composition may be at least partially set and the coated substrate may be removed from the mold. As used herein, the term “set” include, without limitation, curing, polymerizing, cross-linking, cooling, and drying. Alternatively, over-molding may be done by placing the substrate into a mold such that an open region is defined between the substrate and the mold, and thereafter injecting a coating composition comprising the photochromic composition into the open region. Thereafter, the coating composition may be at least partially set and the coated substrate may be removed from the mold.
- [147] Additionally or alternatively, a coating composition (with or without a photochromic

composition) may be applied to a substrate (for example, by any of the foregoing methods), the coating composition may be at least partially set, and thereafter, a photochromic dye may be imbibed (as previously discussed) into the coating composition.

[148] According to still another non-limiting embodiment, wherein the substrate comprises a polymeric material or an inorganic material such as glass, the photochromic composition may be connected to at least a portion of a substrate by lamination. According to this non-limiting embodiment, a film comprising the photochromic composition may be adhered or otherwise connect to at least a portion of the substrate, with or without an adhesive and/or the application of heat and pressure. Thereafter, if desired, a second substrate may be applied over the first substrate and the two substrates may be laminated together (i.e., by the application of heat and pressure) to form an element wherein the film comprising the photochromic composition is interposed between the two substrates. Methods of forming films comprising a photochromic composition may include, for example and without limitation, combining a photochromic composition with a polymeric solution or oligomeric solution or mixture, casting or extruding a film therefrom, and, if required, at least partially setting the film. Additionally or alternatively, a film may be formed (with or without a photochromic composition) and imbibed with the photochromic composition (as discussed above).

[149] Further, it will be appreciated by those skilled in the art that the photochromic compositions and articles according to various non-limiting embodiments disclosed herein may further comprise other additives that aid in the processing and/or performance of the composition or article. Non-limiting examples of such additives include from photoinitiators, thermal initiators, polymerization inhibitors, solvents, light stabilizers (such as, but not limited to, ultraviolet light absorbers and light stabilizers, such as hindered amine light stabilizers (HALS)), heat stabilizers, mold release agents, rheology control agents, leveling agents (such as, but not limited to, surfactants), free radical scavengers, adhesion promoters (such as hexanediol diacrylate and coupling agents), and combinations and mixtures thereof.

[150] This disclosure also provides a photochromic composition comprising the photochromic dye described previously herein incorporated into at least a portion of an organic material, said organic material being a polymer, an oligomer, a monomer or a mixture thereof.

[151] The photochromic dye may be incorporated into at least a portion of an organic material, such as a polymer, oligomer or monomer to form a photochromic composition, which may be used, for example and without limitation, to form photochromic articles. As used herein the term "polymer" refers to homopolymer and copolymer as well as blends and other combinations thereof. As used herein the terms "oligomer" refer to a combination of two or more monomer units that is capable of

reacting with additional monomer units. As used herein the term “incorporated into” means physically and/or chemically combined with. For example, the photochromic dyes according to various non-limiting examples disclosed herein may be physically combined with at least a portion of an organic material, for example and without limitation, by mixing or blending the photochromic dye into the organic material; and/or chemically combined with a portion of an organic material, for example and without limitation, by copolymerization or otherwise bonding the photochromic dye to the organic material.

[152] According to various non-limiting embodiment disclosed herein, the photochromic dye may be incorporated into at least a portion of the organic material by at least one of blending and bonding the photochromic dye with the organic material. As used herein with reference to the incorporation of photochromic dyes into an organic material, the terms “blending” and “blended” mean that the photochromic dye is intermixed or intermingled with a portion of the organic material, but not bonded to the organic material. Further, as used herein with reference to the incorporation of photochromic dyes into an organic material, the terms “bonding” or “bonded” mean that the photochromic dye is linked to a portion of the organic material. For example, although not limiting herein, the photochromic dye may be linked to the organic material through a reactive substituent.

[153] According to one specific non-limiting embodiment, the organic material may be a polymer at least one selected from the group consisting of polyacrylates, polymethacrylates, polyalkylmethacrylates, polyoxy(alkylene methacrylates), poly(alkoxylated phenol methacrylates), cellulose acetate, cellulose triacetate, cellulose acetate propionate, cellulose acetate butyrate, poly(vinyl acetate), poly(vinyl alcohol), poly(vinyl chloride), poly(vinylidene chloride), poly(vinylpyrrolidone), poly((meth)acrylamide), poly(dimethyl acrylamide), poly(hydroxyethyl methacrylate), poly((meth)acrylic acid), thermoplastic polycarbonates, polyesters, polyurethanes, polythiourethanes, poly(ethylene terephthalate), polystyrene, poly(alpha methylstyrene), copoly(styrene-methylmethacrylate), copoly(styrene-acrylonitrile), polyvinylbutyral and polymers of members of the group consisting of polyol(allyl carbonate)monomers, mono-functional acrylate monomers, mono-functional methacrylate monomers, polyfunctional acrylate monomers, polyfunctional methacrylate monomers, diethylene glycol dimethacrylate monomers, diisopropenyl benzene monomers, alkoxylated polyhydric alcohol monomers and diallylidene pentaerythritol monomers.

[154] In another embodiment, the present disclosure provides an optical article applying the photochromic composition according to this disclosure. As used herein the term “optical” means pertaining to or associated with light and/or vision. The optical

elements according to various non-limiting embodiments disclosed herein may include, without limitation, ophthalmic elements, display elements, windows, mirrors, and liquid crystal cell elements. As used herein the term “ophthalmic” means pertaining to or associated with the eye and vision. Non-limiting examples of ophthalmic elements include corrective and non-corrective lenses, including single vision or multi-vision lenses, which may be either segmented or non-segmented multi-vision lenses (such as, but not limited to, bifocal lenses, trifocal lenses and progressive lenses), as well as other elements used to correct, protect, or enhance (cosmetically or otherwise) vision, including without limitation, magnifying lenses, protective lenses, visors, goggles, as well as, lenses for optical instruments (for example, cameras and telescopes). As used herein the term “display” means the visible or machine-readable representation of information in words, numbers, symbols, designs or drawings. Non-limiting examples of display elements include screens, monitors, and security elements, such as security marks. As used herein the term “window” means an aperture adapted to permit the transmission of radiation therethrough. Non-limiting examples of windows include automotive and aircraft transparencies, windshields, filters, shutters, and optical switches. As used herein the term “mirror” means a surface that specularly reflects a large fraction of incident light. As used herein the term “liquid crystal cell” refers to a structure containing a liquid crystal material that is capable of being ordered. One non-limiting example of a liquid crystal cell element is a liquid crystal display.

- [155] Various non-limiting embodiments disclosed herein provide photochromic articles comprising a substrate and a photochromic composition according to any of the non-limiting embodiments discussed above connected to at least a portion of the substrate. As used herein, the term “connected to” means associated with, either directly or indirectly through another material or structure.
- [156] Non-limiting embodiments disclosed herein provide a method of preparing an optical element, comprising connecting a photochromic composition to at least a portion of a substrate by at least one of in-mold casting, coating and lamination.
- [157] For example, according to one non-limiting embodiment, the photochromic composition may be connected to at least a portion of a substrate by in-mold casting. According to this non-limiting embodiment, a coating composition comprising the photochromic composition, which may be a liquid coating composition or a powder coating composition, is applied to the surface of a mold. After that, the coated substrate is removed from the mold. Non-limiting examples of powder coatings in which the photochromic composition according to various non-limiting embodiments disclosed herein may be employed are set forth in U.S. Pat. No. 6,068,797 at col. 7, line 50 to col. 19, line 42, which disclosure is hereby specifically incorporated by reference herein.

- [158] According to another non-limiting embodiment, the photochromic composition may be connected to a portion of a substrate by coating. Non-limiting examples of suitable coating methods include spin coating, spray coating (e.g., using a liquid or powder coating), curtain coating, roll coating, spin and spray coating, over-molding, and combinations thereof. For example, according to one non-limiting embodiment, the photochromic composition may be connected to the substrate by over-molding. According to this non-limiting embodiment, a coating composition comprising the photochromic composition (which may be a liquid coating composition or a powder coating composition as previously discussed) may be applied to a mold and then the substrate may be placed into the mold such that the substrate contacts the coating causing it to spread over the surface of the substrate. Thereafter, the coating composition may be at least partially set and the coated substrate may be removed from the mold. As used herein, the term “set” include, without limitation, curing, polymerizing, cross-linking, cooling, and drying. Alternatively, over-molding may be done by placing the substrate into a mold such that an open region is defined between the substrate and the mold, and thereafter injecting a coating composition comprising the photochromic composition into the open region. Thereafter, the coating composition may be at least partially set and the coated substrate may be removed from the mold.
- [159] Additionally or alternatively, a coating composition (with or without a photochromic composition) may be applied to a substrate (for example, by any of the foregoing methods), the coating composition may be at least partially set, and thereafter, a photochromic dye may be imbibed (as previously discussed) into the coating composition.
- [160] According to still another non-limiting embodiment, wherein the substrate comprises a polymeric material or an inorganic material such as glass, the photochromic composition may be connected to at least a portion of a substrate by lamination. According to this non-limiting embodiment, a film comprising the photochromic composition may be adhered or otherwise connect to at least a portion of the substrate, with or without an adhesive and/or the application of heat and pressure. Thereafter, if desired, a second substrate may be applied over the first substrate and the two substrates may be laminated together (i.e., by the application of heat and pressure) to form an element wherein the film comprising the photochromic composition is interposed between the two substrates. Methods of forming films comprising a photochromic composition may include, for example and without limitation, combining a photochromic composition with a polymeric solution or oligomeric solution or mixture, casting or extruding a film therefrom, and, if required, at least partially setting the film. Additionally or alternatively, a film may be formed (with or without a photochromic composition) and imbibed with the photochromic composition (as discussed above).
- [161] Further, it will be appreciated by those skilled in the art that the photochromic com-

positions and articles according to various non-limiting embodiments disclosed herein may further comprise other additives that aid in the processing and/or performance of the composition or article. Non-limiting examples of such additives include from photoinitiators, thermal initiators, polymerization inhibitors, solvents, light stabilizers (such as, but not limited to, ultraviolet light absorbers and light stabilizers, such as hindered amine light stabilizers (HALS)), heat stabilizers, mold release agents, rheology control agents, leveling agents (such as, but not limited to, surfactants), free radical scavengers, adhesion promoters (such as hexanediol diacrylate and coupling agents), and combinations and mixtures thereof.

[162] The present disclosure is not to be limited in terms of the particular examples described in this application. Many modifications and variations can be made without departing from its spirit and scope, as will be apparent to those skilled in the art. Functionally equivalent methods and apparatuses within the scope of the disclosure, in addition to those enumerated herein, will be apparent to those skilled in the art from the foregoing descriptions. Such modifications and variations are intended to fall within the scope of the appended claims. The present disclosure is to be limited only by the terms of the appended claims, along with the full scope of equivalents to which such claims are entitled. It is to be understood that this disclosure is not limited to particular methods, reagents, compounds compositions or biological systems, which can, of course, vary. It is also to be understood that the terminology used herein is for the purpose of describing particular examples only, and is not intended to be limiting.

[163] With respect to the use of substantially any plural and/or singular terms herein, those having skill in the art can translate from the plural to the singular and/or from the singular to the plural as is appropriate to the context and/or application. The various singular/plural permutations may be expressly set forth herein for sake of clarity.

[164] It will be understood by those within the art that, in general, terms used herein, and especially in the appended claims (e.g., bodies of the appended claims) are generally intended as “open” terms (e.g., the term “including” should be interpreted as “including but not limited to,” the term “having” should be interpreted as “having at least,” the term “includes” should be interpreted as “includes but is not limited to,” etc.). It will be further understood by those within the art that if a specific number of an introduced claim recitation is intended, such an intent will be explicitly recited in the claim, and in the absence of such recitation no such intent is present. For example, as an aid to understanding, the following appended claims may contain usage of the introductory phrases “at least one” and “one or more” to introduce claim recitations. However, the use of such phrases should not be construed to imply that the introduction of a claim recitation by the indefinite articles “a” or “an” limits any particular claim containing such introduced claim recitation to embodiments

containing only one such recitation, even when the same claim includes the introductory phrases “one or more” or “at least one” and indefinite articles such as “a” or “an” (e.g., “a” and/or “an” should be interpreted to mean “at least one” or “one or more”); the same holds true for the use of definite articles used to introduce claim recitations. In addition, even if a specific number of an introduced claim recitation is explicitly recited, those skilled in the art will recognize that such recitation should be interpreted to mean at least the recited number (e.g., the bare recitation of “two recitations,” without other modifiers, means at least two recitations, or two or more recitations). Furthermore, in those instances where a convention analogous to “at least one of A, B, and C, etc.” is used, in general such a construction is intended in the sense one having skill in the art would understand the convention (e.g., “a system having at least one of A, B, and C” would include but not be limited to systems that have A alone, B alone, C alone, A and B together, A and C together, B and C together, and/or A, B, and C together, etc.). It will be further understood by those within the art that virtually any disjunctive word and/or phrase presenting two or more alternative terms, whether in the description, claims, or drawings, should be understood to contemplate the possibilities of including one of the terms, either of the terms, or both terms. For example, the phrase “A or B” will be understood to include the possibilities of “A” or “B” or “A and B.”

[165] In addition, where features or aspects of the disclosure are described in terms of Markush groups, those skilled in the art will recognize that the disclosure is also thereby described in terms of any individual member or subgroup of members of the Markush group.

[166] As will be understood by one skilled in the art, for any and all purposes, such as in terms of providing a written description, all ranges disclosed herein also encompass any and all possible subranges and combinations of subranges thereof. Any listed range can be easily recognized as sufficiently describing and enabling the same range being broken down into at least equal halves, thirds, quarters, fifths, tenths, etc. As a non-limiting example, each range discussed herein can be readily broken down into a lower third, middle third and upper third, etc. As will also be understood by one skilled in the art all language such as “up to,” “at least,” “greater than,” “less than,” and the like include the number recited and refer to ranges which can be subsequently broken down into subranges as discussed above. Finally, as will be understood by one skilled in the art, a range includes each individual member. Thus, for example, a group having 1-3 cells refers to groups having 1, 2, or 3 cells. Similarly, a group having 1-5 cells refers to groups having 1, 2, 3, 4, or 5 cells, and so forth.”

[167] While various aspects and embodiments have been disclosed herein, other aspects and embodiments will be apparent to those skilled in the art. The various aspects and

embodiments disclosed herein are for purposes of illustration and are not intended to be limiting, with the true scope and spirit being indicated by the following claims.

[168]

[169] **EXAMPLES**

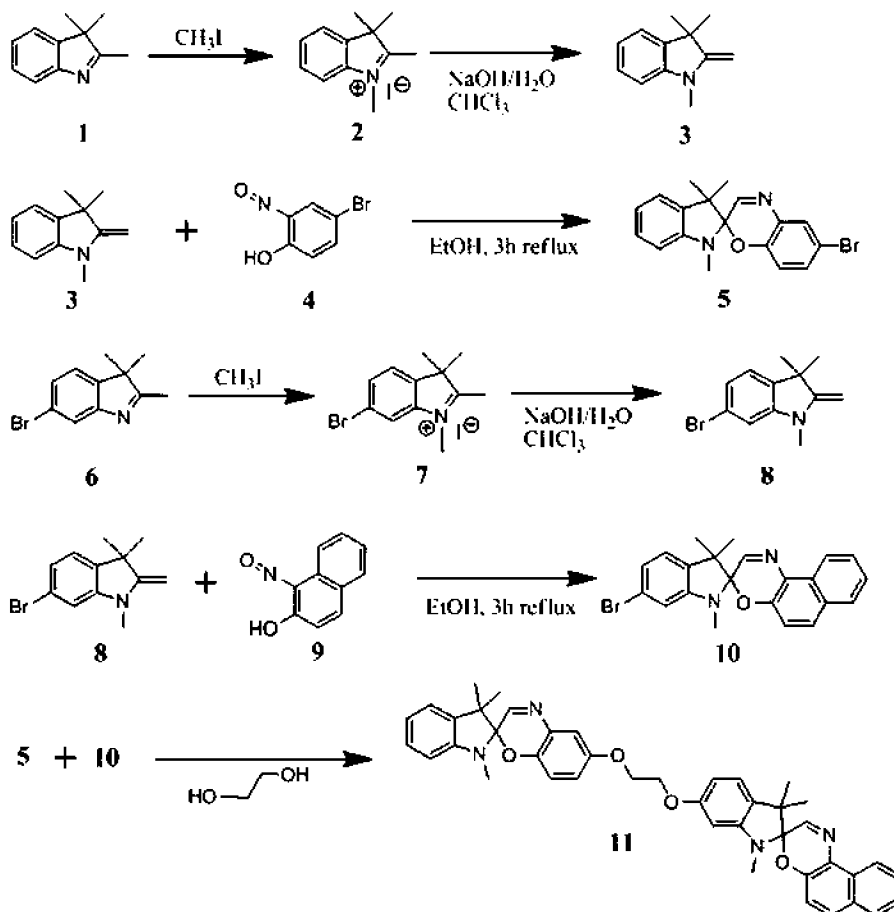
[170] The synthesis procedures used to make photochromic dyes according to various non-limiting embodiments disclosed herein are set forth in Examples 1 to 3.

[171] **EXAMPLE 1**

[172] The synthesis of

1,3,3-trimethyl-6-(2-(1',3',3'-trimethylspiro[benzo[b][1,4]oxazine-2,2'-indoline]-6-yloxy)ethoxy)spiro[indoline-2,3'-naphtho[2,1-b][1,4]oxazine]

[173]



[174] **Preparation of compound 2 :** A solution of 2,3,3-trimethyl-3H-indole and iodomethane in dichloroethane is refluxed by stirring for 4 h. Then the reaction mixture is stirred for one hour at room temperature and the precipitated solid is collected by filtration.

[175] **Preparation of compound 3:** After the crude solid of compound 2 is washed with acetone, a portion of the solid is dissolved in NaOH aqueous solution (1 M). The resultant 1,3,3-trimethyl-2-methyleneindoline is extracted with chloroform. The organic layer is dried with anhydrous Na_2SO_4 and the solvent was evaporated.

- [176] **Preparation of compound 5:** Compound 3 and 4-bromo-2-nitrosophenol are dissolved in ethanol and the mixture is stirred at 90 °C for 3 h. After cooling the mixture, the solid product, 5 is obtained by filtration.
- [177] **Preparation of compound 7:** A solution of 6-bromo-2,3,3-trimethyl-3H-indole and iodomethane in dichloroethane is refluxed by stirring for 6 h. Then the reaction mixture is stirred for one hour at room temperature and the precipitated solid, 7 is collected by filtration.
- [178] **Preparation of compound 8:** After the solid of compound 7 is washed with acetone, a portion of the solid is dissolved in NaOH aqueous solution (1 M). The resultant 6-bromo-1,3,3-trimethyl-2-methyleneindoline, 8 is extracted with chloroform. The organic layer is dried with anhydrous Na₂SO₄ and the solvent is evaporated. The product is obtained in a fairly high yield.
- [179] **Preparation of compound 10:** compound 8 and 1-nitrosonaphthalen-2-ol are dissolved in ethanol and the mixture is stirred at 90 °C for 3 h. After cooling the mixture, the solid product, 6-bromo-1,3,3-trimethylspiro[indoline-2,3'-naphtho[2,1-b][1,4]oxazine] is obtained by filtration.
- [180] **Preparation of compound 11:** Compound 5 and 10 are dissolved in toluene with the presence of 1,2-ethanediol and HCl(0.1M solution) and the mixture is refluxed for 12 hrs. The final compound 1,3,3-trimethyl-6-(2-(1',3',3'-trimethylspiro[benzo[b][1,4]oxazine-2,2'-indoline]-6-yloxy)ethoxy)spiro[indoline-2,3'-naphtho[2,1-b][1,4]oxazine] is obtained by silical gel chromatography (EA:hexane=1:4). The resultant compound is purified by reprecipitation in THF/ether.

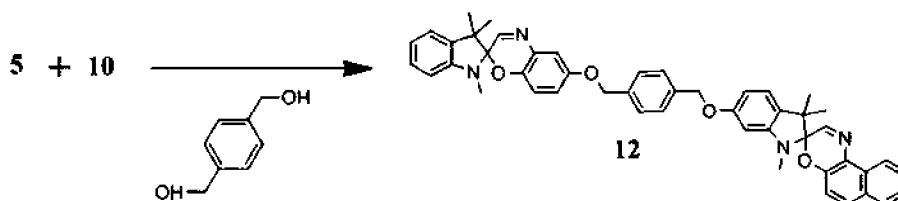
[181]

[182] Example 2

[183] The synthesis of

1,3,3-trimethyl-6-(4-((1',3',3'-trimethylspiro[benzo[b][1,4]oxazine-2,2'-indoline]-6-yloxy)methyl)benzyloxy)spiro[indoline-2,3'-naphtho[2,1-b][1,4]oxazine]

[184]



[185] The synthesis of compound 12 is quite similar to that of compound 11.

[186] Compound 5 and 10 are dissolved in dimethylformamide with the presence of 1,4-phenylenedimethanol. The reaction mixture is refluxed for overnight. The final compound,

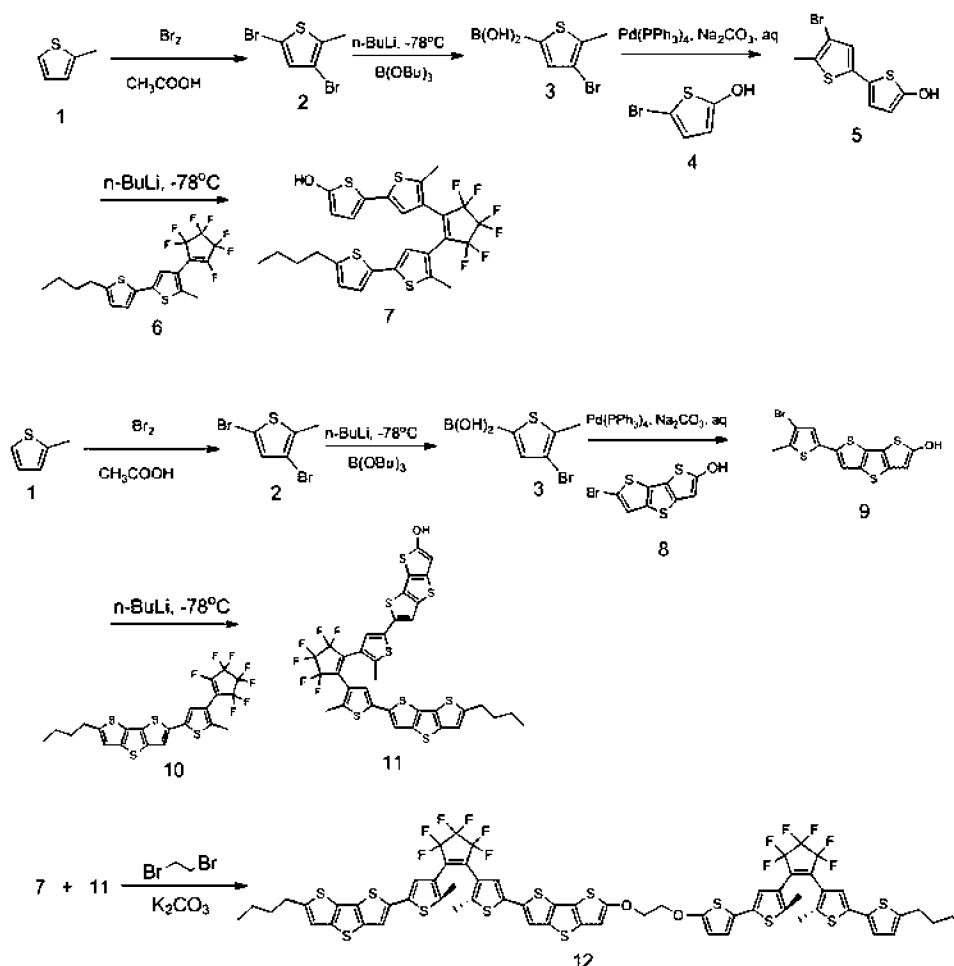
1,3,3-trimethyl-6-(4-((1',3',3'-trimethylspiro[benzo[b][1,4]oxazine-2,2'-indoline]-6-ylomethyl) benzyloxy)spiro[indoline-2,3'-naphtho[2,1-b][1,4]oxazine] is obtained by silical gel chromatography (EA:hexane=1:4). The resultant compound is purified by reprecipitation in THF/hexane.

[187]

[188] Example 3

[189] The synthesis of a photochromic dye according to the scheme III-1.

[190]



[191] **Synthesis of 3,5-dibromo-2-methylthiophene 2:** Bromine in acetic acid is slowly added to a stirred solution of 2-methylthiophene in acetic acid (100 mL) at 0 °C. The reaction mixture is stirred overnight at this temperature. The reaction is quenched by the addition of water 30 mL. The mixture is neutralized to pH 9.0 with Na₂CO₃ solid and extracted with ether. The ether extract is dried, filtrated, and concentrated. The residue is purified by distillation in vacuo. Compound 5 is obtained as a colorless oil B.p. 100 °C.

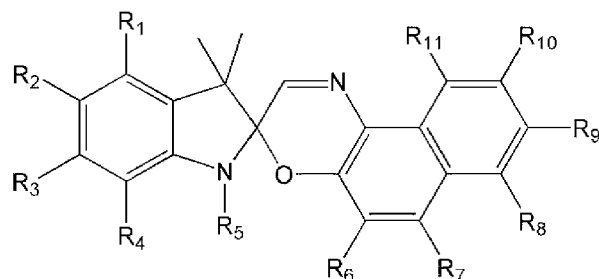
[192] **Synthesis of 3-bromo-2-methyl-5-thienylboronic acid 3:** n-Butyl lithium in hexane is added to a stirred solution of compound 2 in dry ether (150mL) at -78 °C under argon atmosphere. After stirring for 30 min, boric acid tri-butyl ester is quickly added

to the reaction mixture. The mixture is extracted with 4% aq. NaOH (100mL), the extract is collected and neutralized by with 10% HCl. The solid residue is washed, filtrated, and dried. Compound 3 is obtained as a yellowish solid. M.p. 229 °C.

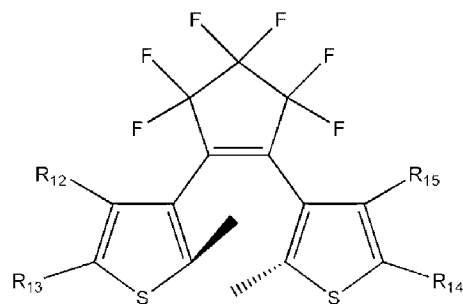
- [193] **Synthesis of 3-bromo-2-methyl-5-(4-hydroxythienyl)thiophene 5:** Compound 3 is mixed with 5-bromothiophen-2-ol in the presence of tetrakis(triphenylphosphine)palladium(0) [Pd(PPh₃)₄] and Na₂CO₃ in tetrahydrofuran (THF) for 24 h at 80 °C. The crude product is distillated in vacuo and purified by silica-gel column chromatography (eluent:hexane). Compound 5 (5.01 g) is obtained as a yellowish solid.
- [194] **Synthesis of compound 9:** The synthesis is quite similar to that for compound 5. Instead of compound 4, compound 8 is employed in the coupling reaction.
- [195] **Synthesis of 4'-(2-(5'-butyl-5-methyl-2,2'-bithiophen-4-yl)-3,3,4,4,5,5-hexafluorocyclopent-1-enyl)-5'-methyl-2,2'-bithiophen-5-ol 7:** Compound 5 in anhydrous THF is added dropwise to a 2.4 mol/L n-BuLi hexane solution at -78 °C under an argon atmosphere. The mixture is kept stirring for 1 hr and 5'-butyl-5-methyl-4-(perfluorocyclopent-1-enyl)-2,2'-bithiophene, 6 is slowly added to the reaction mixture at -78 °C and stirred for 4 h. The reaction is quenched with 50 mL water. The mixture is warmed to room temperature and extracted with ether. The organic layer is dried over MgSO₄, filtrated and evaporated. The crude product is purified by column chromatography on silica gel using hexane as the eluent.
- [196] **Synthesis of compound 11:** The synthesis is quite similar to that for compound 5. Instead of compound 6, compound 11 is employed in the coupling reaction.
- [197] **Synthesis of compound 12:** Compound 7 and 11 are mixed in THF at room temperature in the presence of dibromoethane and potassium carbonate. The reaction mixture is allowed to be refluxed overnight. The crude product is neutralized and filtered. The filtrate is purified by silical gel column chromatography.

Claims

- [Claim 1] A photochromic dye, comprising:
at least two photo-reactive groups, wherein
a first photo-reactive group is configured to undergo a first photochromic reaction in response to radiation having first wavelength, and
a second photo-reactive group is configured to undergo a second photochromic reaction in response to radiation having second wavelength.
- [Claim 2] The photochromic dye of claim 1, wherein the at least two photo-reactive groups are coupled to one another by a non-conjugated linkage selected from the group consisting of C₁₋₁₂ alkylene, C₁₋₁₂ heteroalkylene, C₁₋₁₂ alkoxy, C₁₋₆ alkyleneoxy, C₁₋₆ alkylenedioxy, C₁₋₆ phenyleneoxy, and C₁₋₆ phenylenedioxy.
- [Claim 3] The photochromic dye of claim 1, wherein
the first photochromic reaction includes modification of a first conjugation system in the first photo-reactive group, and
the second photochromic reaction includes modification of a second conjugation system in the second photo-reactive group.
- [Claim 4] The photochromic dye of claim 1, wherein the first photo-reactive group and the second photo-reactive group have different absorption maxima.
- [Claim 5] The photochromic dye of claim 4, wherein the difference of absorption maxima between the first photo-reactive group and the second photo-reactive group photochromic is greater than or equal to about 50 nm.
- [Claim 6] The photochromic dye of claim 1, wherein the at least two photo-reactive groups are independently at least one selected from the group consisting of a spiropyran compound, a spirooxazine compound, a diarylethene compound and a fulgide compound.
- [Claim 7] A photochromic dye having at least two photo-reactive groups coupled to one another, wherein
a first photo-reactive group and a second photo-reactive group are independently at least one structure selected from the group consisting of Formula I and II:
[Formula I]



[Formula II]



wherein,

R_1 to R_{15} independently represent hydrogen, a halogen, a substituted or unsubstituted C_{1-12} alkyl, a substituted or unsubstituted C_{1-12} alkoxy, a substituted or unsubstituted C_{1-12} alkenyl, a substituted or unsubstituted C_{1-12} conjugated alkyl, a substituted or unsubstituted aryl group, or a substituted or unsubstituted heteroaryl group; or the substituents together form an alkylene or alkenylene chain completing an aryl group.

[Claim 8]

The photochromic dye of claim 7, wherein,

R_2 and R_9 independently represent hydrogen, a substituted or unsubstituted aryl group, or a substituted or unsubstituted heteroaryl group; or

R_2 and R_3 , or R_9 and R_{10} together form an alkylene or alkenylene chain completing an aryl group

R_{13} and R_{14} independently represent hydrogen; and

R_{12} and R_{15} independently represent hydrogen, a substituted or unsubstituted C_{6-12} conjugated alkyl, a substituted or unsubstituted aryl group, or a substituted or unsubstituted heteroaryl group.

[Claim 9]

The photochromic dye of claim 7, wherein,

R_2 and R_9 independently represent hydrogen or a phenyl; or

R_2 and R_3 , or R_9 and R_{10} together form a benzene or naphthyl;

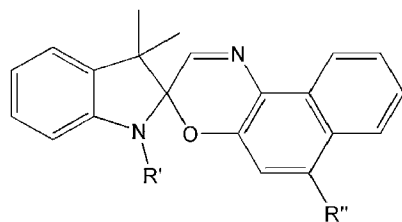
R_{13} and R_{14} independently represent hydrogen; and

R_{12} and R_{15} independently represent hydrogen, a phenyl or a substituted or unsubstituted dithienothiophenyl.

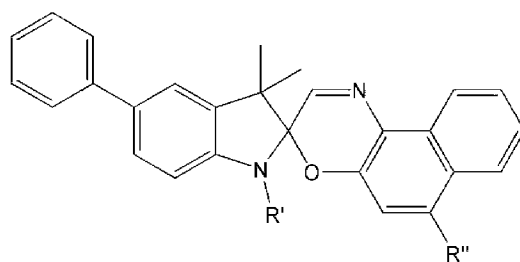
[Claim 10]

The photochromic dye of claim 7, wherein,
the photochromic group of Formula I is selected from the group
consisting of Formula I-1 to I-9:

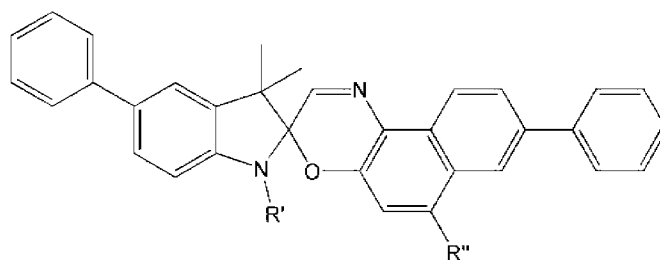
[Formula I-1]



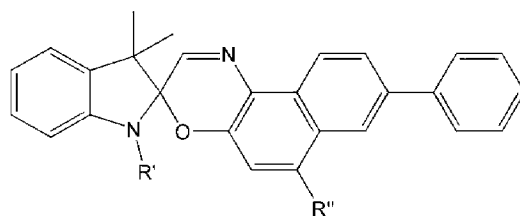
[Formula I-2]



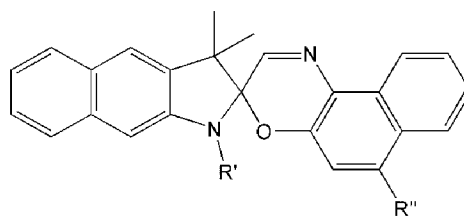
[Formula I-3]



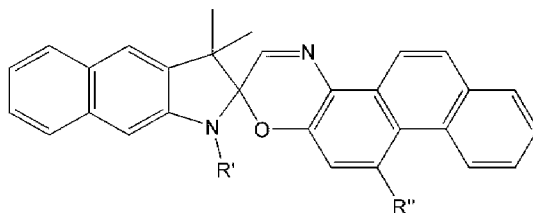
[Formula I-4]



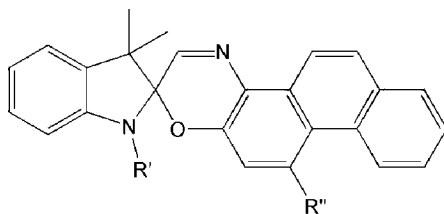
[Formula I-5]



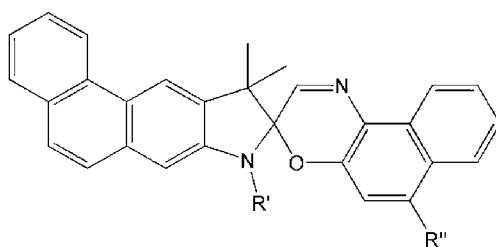
[Formula I-6]



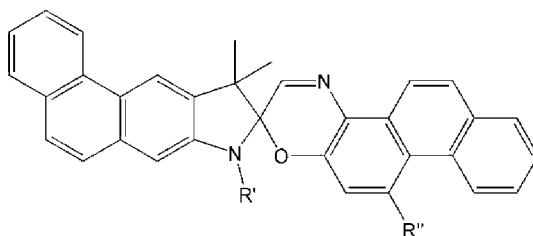
[Formula I-7]



[Formula I-8]



[Formula I-9]



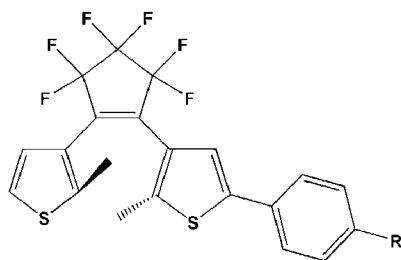
wherein,

R' and R'' independently represent hydrogen, a halogen, a substituted or unsubstituted C₁₋₆ alkyl, or a substituted or unsubstituted C₁₋₆ alkoxy.

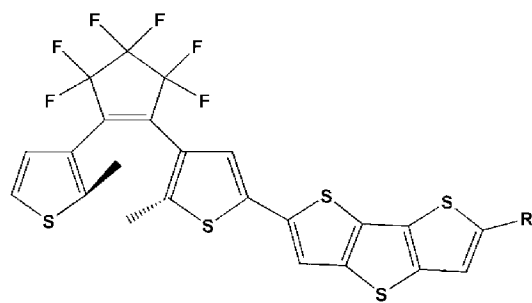
[Claim 11]

The photochromic dye of claim 7, wherein,
the photochromic group of Formula II is selected from the group consisting of Formula II-1 to II-7:

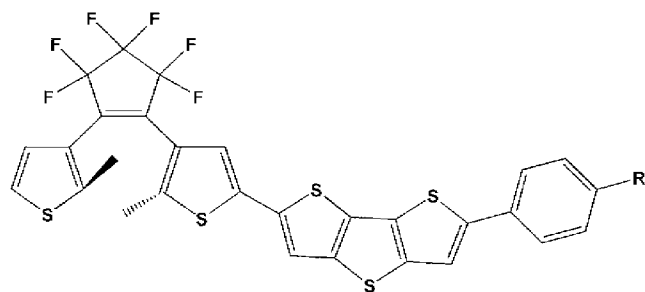
[Formula II-1]



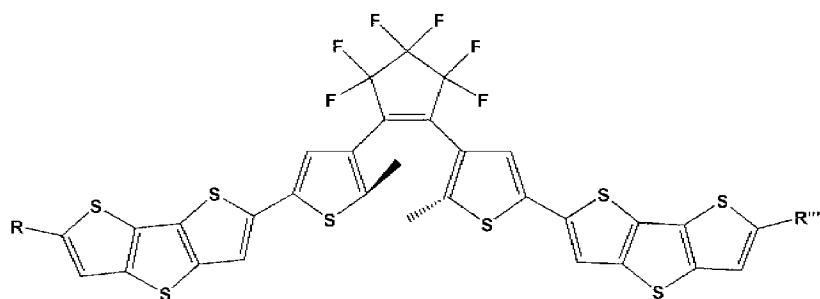
[Formula II-2]



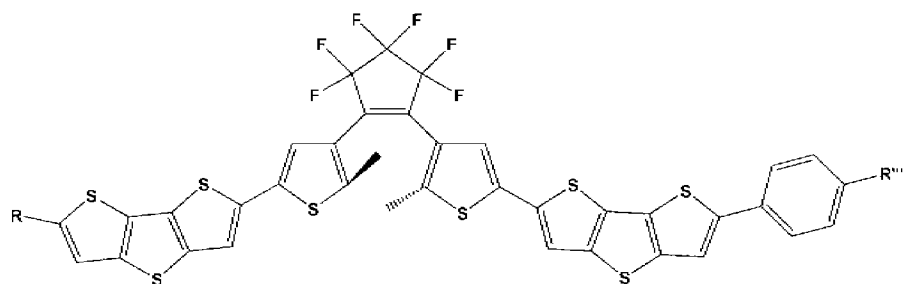
[Formula II-3]



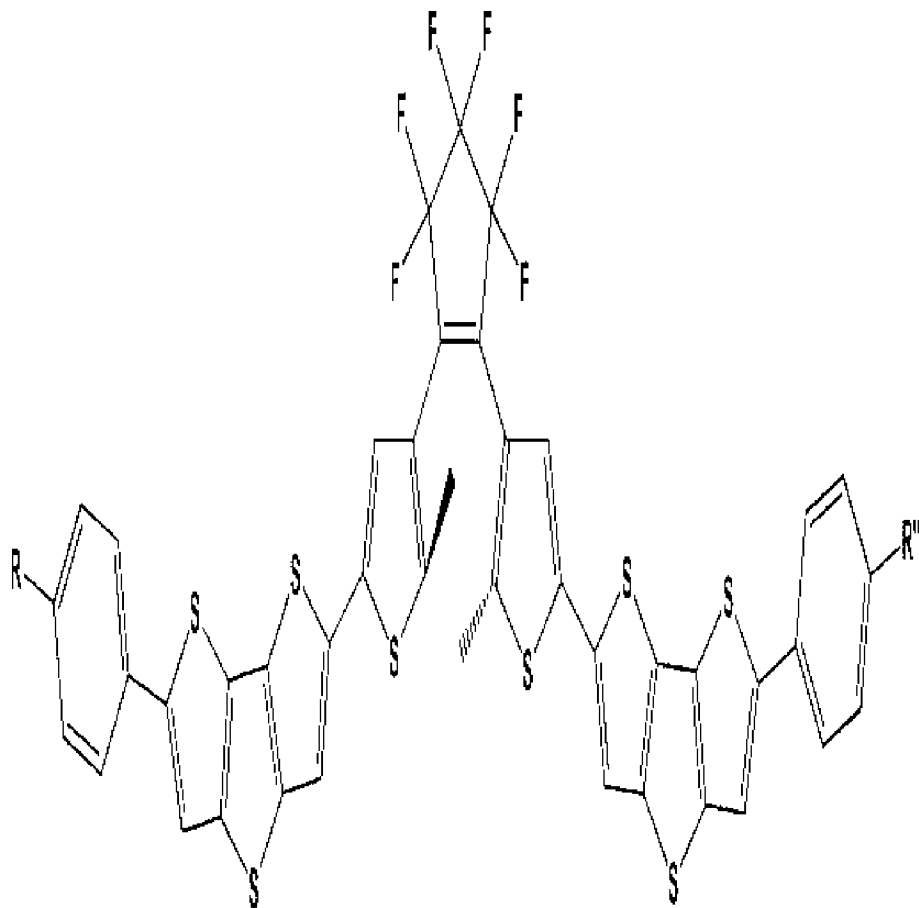
[Formula II-4]



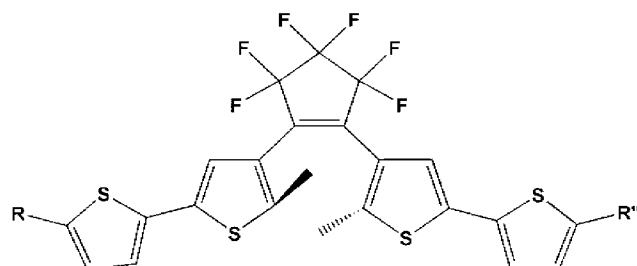
[Formula II-5]



[Formula II-6]



[Formula II-7]



Wherein,

R and R''' independently represent hydrogen, or a substituted or unsubstituted C₁₋₁₂ alkyl.

[Claim 12]

The photochromic dye of claim 7, wherein the first photo-reactive group and the second photo-reactive group are coupled to one another by a non-conjugated linkage selected from the group consisting of C₁₋₁₂ alkylene, C₁₋₁₂ heteroalkylene, C₁₋₆ alkyleneoxy, and C₁₋₆ alkylenedioxy.

[Claim 13]

The photochromic dye of claim 7, wherein the first photo-reactive group is configured to undergo a first photochromic reaction in response to radiation having first wavelength, and the second photo-reactive group is configured to undergo a second photochromic reaction

in response to radiation having second wavelength.

[Claim 14]

A photochromic composition, comprising:

at least one material selected from the group consisting of a polymer, an oligomer, a monomer, and a mixture thereof; and

at least one photochromic dye incorporated into at least a portion of the material, the at least one photochromic dye having at least two photo-reactive groups that are coupled to one another, wherein

a first photo-reactive group is configured to undergo a first photochromic reaction in response to radiation having first wavelength, and a second photo-reactive group is configured to undergo a second photochromic reaction in response to radiation having second wavelength.

[Claim 15]

The photochromic composition of claim 14, wherein the material is at least polymer one selected from the group consisting of polyacrylates, polymethacrylates, polyalkylmethacrylates, polyoxy(alkylene methacrylates), poly(alkoxylated phenol methacrylates), cellulose acetate, cellulose triacetate, cellulose acetate propionate, cellulose acetate butyrate, poly(vinyl acetate), poly(vinyl alcohol), poly(vinyl chloride), poly(vinylidene chloride), poly(vinylpyrrolidone), poly((meth)acrylamide), poly(dimethyl acrylamide), poly(hydroxyethyl methacrylate), poly((meth)acrylic acid), thermoplastic polycarbonates, polyesters, polyurethanes, polythiourethanes, poly(ethylene terephthalate), polystyrene, poly(alpha methylstyrene), copoly(styrene-methylmethacrylate), copoly(styrene- acrylonitrile), polyvinylbutyral, and polymers of members of the group consisting of polyol(allyl carbonate)monomers, mono-functional acrylate monomers, mono-functional methacrylate monomers, polyfunctional acrylate monomers, polyfunctional methacrylate monomers, diethylene glycol dimethacrylate monomers, diisopropenyl benzene monomers, alkoxylated polyhydric alcohol monomers and diallylidene pentaerythritol monomers.

[Claim 16]

The photochromic composition of claim 14, the at least one material further including at least a second material having a glass transition temperature less than about 0 °C.

[Claim 17]

The photochromic composition of claim 16, the dye being coupled to the second material.

[Claim 18]

The photochromic composition of claim 14, the at least one material further including at least one polysiloxane and/or at least one polyacrylate having a glass transition temperature less than about 0 °C.

- [Claim 19] The photochromic composition of claim 14, wherein the first photo-reactive group and the second photo-reactive group are coupled to one another by a non-conjugated linkage selected from the group consisting of C₁₋₁₂ alkylene, C₁₋₁₂ heteroalkylene, C₁₋₆ alkyleneoxy, and C₁₋₆ alkylenedioxy.
- [Claim 20] The photochromic composition of claim 14, wherein the at least two photo-reactive groups are independently at least one structure selected from the group consisting of a spiropyran compound, a spirooxazine compound, a diarylethene compound and a fulgide compound.

INTERNATIONAL SEARCH REPORT

International application No.

PCT/KR2010/009128

A. CLASSIFICATION OF SUBJECT MATTER

Int. Cl.

G03C 1/73 (2006.01)

C09B 57/00 (2006.01)

G03C 1/685 (2006.01)

C09B 19/00 (2006.01)

C09K 9/02 (2006.01)

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practicable, search terms used)

CAPLUS, EPODOC, WPI: spirooxazine, spiropyran, fulgide, photochromic, dye, ink, wavelength, intensity, polymer, biphotochromic

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	US 2003/0045714 A1 (MELZIG ET AL.) 6 March 2003 See abstract, [0009], [0010] and examples	1, 3-5, 14, 15, 17
X	JP 05-065477 A (TORAY IND INC) 19 March 1993 – Patent Abstracts of Japan machine based English translation <URL: http://www19.ipdl.inpit.go.jp/PA1/cgi-bin/PA1DETAIL > See abstract, [0044], chemical formula 1	1, 3-5, 14-18
X	JP 05-043870 A (TORAY IND INC) 23 February 1993 – Patent Abstracts of Japan machine based English translation <URL: http://www19.ipdl.inpit.go.jp/PA1/cgi-bin/PA1DETAIL > See abstract and claim 2	1, 3-5, 14-18
X	EP 1 433 620 A2 (EASTMAN KODAK COMPANY) 30 June 2004 See abstract, [0017], p. 3 line 51, [0044], [0045]	1-5, 14, 15, 19

☒ Further documents are listed in the continuation of Box C☒ See patent family annex

* Special categories of cited documents:	"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention
"A" document defining the general state of the art which is not considered to be of particular relevance	"X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone
"E" earlier application or patent but published on or after the international filing date	"Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art
"L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)	"&" document member of the same patent family
"O" document referring to an oral disclosure, use, exhibition or other means	
"P" document published prior to the international filing date but later than the priority date claimed	

Date of the actual completion of the international search

21 March 2011

Date of mailing of the international search report

- 2 APR 2011

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INTERNATIONAL SEARCH REPORT

International application No.

PCT/KR2010/009128

Box No. II Observations where certain claims were found unsearchable (Continuation of item 2 of first sheet)

This international search report has not been established in respect of certain claims under Article 17(2)(a) for the following reasons:

1. ☐ Claims Nos.:
because they relate to subject matter not required to be searched by this Authority, namely:
2. ☐ Claims Nos.:
because they relate to parts of the international application that do not comply with the prescribed requirements to such an extent that no meaningful international search can be carried out, specifically:
3. ☐ Claims Nos.:
because they are dependent claims and are not drafted in accordance with the second and third sentences of Rule 6.4(a)

Box No. III Observations where unity of invention is lacking (Continuation of item 3 of first sheet)

This International Searching Authority found multiple inventions in this international application, as follows:

As reasoned in the Supplemental Boxes

1. ☐ As all required additional search fees were timely paid by the applicant, this international search report covers all searchable claims.
2. ☐ As all searchable claims could be searched without effort justifying additional fees, this Authority did not invite payment of additional fees.
3. ☐ As only some of the required additional search fees were timely paid by the applicant, this international search report covers only those claims for which fees were paid, specifically claims Nos.:
4. ☒ No required additional search fees were timely paid by the applicant. Consequently, this international search report is restricted to the invention first mentioned in the claims; it is covered by claims Nos.: 1-5 and 14-19

Remark on Protest

- ☐ The additional search fees were accompanied by the applicant's protest and, where applicable, the payment of a protest fee.
- ☐ The additional search fees were accompanied by the applicant's protest but the applicable protest fee was not paid within the time limit specified in the invitation.
- ☐ No protest accompanied the payment of additional search fees.

INTERNATIONAL SEARCH REPORT

International application No.

PCT/KR2010/009128

C (Continuation). DOCUMENTS CONSIDERED TO BE RELEVANT		
Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	GUGLIELMETTI, R., 'New polyphotochromic systems involving spirooxazine and/or chromene joined by a (Z) ethylenic bridge', <i>Advances in Colour Science and Technology</i> , 2002, Vol. 5 (4), pages 113-116 See introduction paragraph, compound E, p. 115 para. 7	1-5
X	ORTICA, F. <i>et al.</i> , 'Photokinetic behaviour of biphotochromic supramolecular systems Part 2. A bis-benzo-[2H]-chromene and a spirooxazine-chromene with a (Z)-ethenic bridge between each moiety', <i>J. Photochem. Photobiol. A: Chemistry</i> , 2001, Vol. 139, pages 133-141 See compound SPO-CHR(Z) and Scheme 4	1-5
X	YIP, W. T. <i>et al.</i> , 'Energy Transfer in Bichromophoric Molecules: The Effect of Symmetry and Donor/Acceptor Energy Gap', <i>J. Phys. Chem. A</i> , 1999, Vol. 103, pages 10-20 See abstract, figure 1 and tables 1-9	1-5
X	US 2008/0055686 A1 (ERBEN ET AL.) 6 March 2008 See [0007], [0035], [0059], [0062]	1, 3, 4, 14-16, 18
P, X	WO 2010/020770 A1 (JAMES ROBINSON LIMITED) 25 February 2010 See abstract, example 10 and p. 3 para. 6	1, 3-5, 14, 15, 17

Supplemental Box I

(To be used when the space in any of Boxes I to IV is not sufficient)

Continuation of Box No: III

This International Application does not comply with the requirements of unity of invention because it does not relate to one invention or to a group of inventions so linked as to form a single general inventive concept.

In assessing whether there is more than one invention claimed, I have given consideration to those features which can be considered to potentially distinguish the claimed combination of features from the prior art.

This International Searching Authority has found that there are five different inventions as follows:

- **Invention 1 (claims 1-5 and 14-19):** constitutes a photochromic dye comprising at least two photo-reactive groups each configured to undergo a photochromic reaction at different wavelengths. The dye can be incorporated into a composition further consisting of a polymer, an oligomer, a monomer or a mixture thereof.
- **Invention 2 (claims 6, 7-10, 12, 13 and 20 in part):** constitutes a photochromic dye comprising at least two photo-reactive groups each configured to undergo a photochromic reaction at different wavelengths whereby the first and second photo-reactive groups are **spirooxazines** including those of formula I respectively. The dye can be incorporated into a composition further consisting of a polymer, an oligomer, a monomer or a mixture thereof.
- **Invention 3 (claims 6, 7-9, 11-13 and 20 in part):** constitutes a photochromic dye comprising at least two photo-reactive groups each configured to undergo a photochromic reaction at different wavelengths whereby the first and second photo-reactive groups are **diarylethenes** including those of formula II respectively. The dye can be incorporated into a composition further consisting of a polymer, an oligomer, a monomer or a mixture thereof.
- **Invention 4 (claims 6 and 20 in part):** constitutes a photochromic dye comprising at least two photo-reactive groups each configured to undergo a photochromic reaction at different wavelengths whereby the first and second photo-reactive groups consist of at least a **spiropyran**. The dye can be incorporated into a composition further consisting of a polymer, an oligomer, a monomer or a mixture thereof.
- **Invention 5 (claims 6 and 20 in part):** constitutes a photochromic dye comprising at least two photo-reactive groups each configured to undergo a photochromic reaction at different wavelengths whereby the first and second photo-reactive groups consist of at least a **fulgide**. The dye can be incorporated into a composition further consisting of a polymer, an oligomer, a monomer or a mixture thereof.

PCT Rule 13.2, first sentence, states that unity of invention is only fulfilled when there is a technical relationship among the claimed inventions involving one or more of the same or corresponding special technical features. PCT Rule 13.2, second sentence, defines a special technical feature as a feature which makes a contribution over the prior art.

[Continued in Supplemental Box II]

Supplemental Box

(To be used when the space in any of Boxes I to VIII is not sufficient)

Continuation of Box No: Supplemental Box I

The only feature common to all of the claims is two photo-reactive groups each configured to undergo a photochromic reaction at different wavelengths. However photochromic dyes with such a feature are well known in the art. See for example:

US 2003/0045714 A1

JP 05-065477 A

JP 05-043870 A

EP 1 433 620 A2

This means that the common feature can not constitute a special technical feature within the meaning of PCT Rule 13.2, second sentence, since it makes no contribution over the prior art.

Because the common feature does not satisfy the requirement for being a special technical feature it follows that it cannot provide the necessary technical relationship between the identified inventions. Therefore the claims do not satisfy the requirement of unity of invention *a posteriori*.

INTERNATIONAL SEARCH REPORT

Information on patent family members

International application No.

PCT/KR2010/009128

This Annex lists the known "A" publication level patent family members relating to the patent documents cited in the above-mentioned international search report. The Australian Patent Office is in no way liable for these particulars which are merely given for the purpose of information.

Patent Document Cited in Search Report			Patent Family Member		
US	2008055686	NONE			
US	2003045714	EP 0706526	JP 11252842	US	6936725
		US 2003127933	WO 9500519		
JP	5065477	JP 62288191			
JP	5043870	JP 59079020			
EP	1433620	JP 2004209984	US 2004127359	US	6841514
WO	2010020770	NONE			
Due to data integration issues this family listing may not include 10 digit Australian applications filed since May 2001.					
END OF ANNEX					