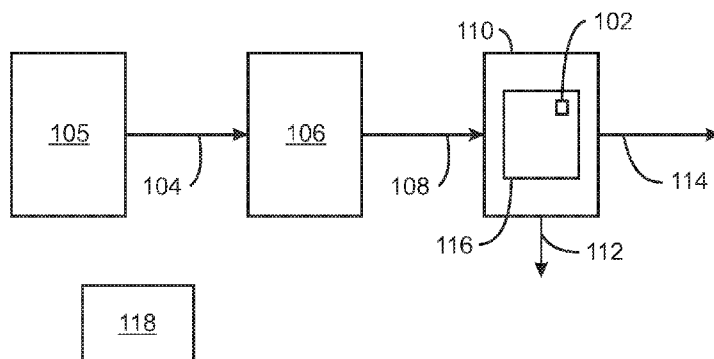




- (51) International Patent Classification:
F25J 3/02 (2006.01)
- (21) International Application Number:
PCT/US2016/032416
- (22) International Filing Date:
13 May 2016 (13.05.2016)
- (25) Filing Language: English
- (26) Publication Language: English
- (30) Priority Data:
62/183,033 22 June 2015 (22.06.2015) US
- (71) Applicant (for all designated States except US): **EXXON-MOBIL UPSTREAM RESEARCH COMPANY** [US/US]; CORP-URC-E2.4A.296, Spring, TX 77389 (US).
- (72) Inventors; and
- (71) Applicants (for US only): **VALENCIA, Jaime, A.** [US/US]; 5503 Peace Court, Houston, TX 77041 (US). **KELMAN, Scott, D.** [NZ/US]; 1207 Harvard St., Houston, TX 77008 (US). **KIMBLE, E., Lawrence** [US/US]; 4814 Rebel Ridge Drive, Sugar Land, TX 77478 (US).
- (74) Agents: **SCHMIDT, Jennifer, A.** et al.; Exxonmobil Upstream Research Company, CORP-URC-E2.4A.296, 22777 Springwoods Village Parkway, Spring, TX 77389 (US).
- (81) Designated States (unless otherwise indicated, for every kind of national protection available): AE, AG, AL, AM, AO, AT, AU, AZ, BA, BB, BG, BH, BN, BR, BW, BY, BZ, CA, CH, CL, CN, CO, CR, CU, CZ, DE, DK, DM, DO, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, GT, HN, HR, HU, ID, IL, IN, IR, IS, JP, KE, KG, KN, KP, KR, KZ, LA, LC, LK, LR, LS, LU, LY, MA, MD, ME, MG, MK, MN, MW, MX, MY, MZ, NA, NG, NI, NO, NZ, OM, PA, PE, PG, PH, PL, PT, QA, RO, RS, RU, RW, SA, SC, SD, SE, SG, SK, SL, SM, ST, SV, SY, TH, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, ZA, ZM, ZW.
- (84) Designated States (unless otherwise indicated, for every kind of regional protection available): ARIPO (BW, GH, GM, KE, LR, LS, MW, MZ, NA, RW, SD, SL, ST, SZ, TZ, UG, ZM, ZW), Eurasian (AM, AZ, BY, KG, KZ, RU, TJ, TM), European (AL, AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HR, HU, IE, IS, IT, LT, LU, LV, MC, MK, MT, NL, NO, PL, PT, RO, RS, SE, SI, SK, SM, TR), OAPI (BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, KM, ML, MR, NE, SN, TD, TG).

[Continued on next page]

(54) Title: PURGE TO INTERMEDIATE PRESSURE IN CRYOGENIC DISTILLATION



(57) Abstract: A system and method for an intermediate pressure vessel to receive a purge in a cryogenic distillation column system. The vessel operates at an intermediate pressure to reduce solidification of the purge. The cryogenic distillation column receives a natural gas having methane and acid gas. The column discharges an overhead stream rich in the methane, and a bottoms stream rich in the acid gas. An overhead system receives the overhead stream, and discharges a vapor methane product, and a cooled liquid for reflux to the cryogenic distillation column.

100
FIG. 1

**Declarations under Rule 4.17:**

- *as to applicant's entitlement to apply for and be granted a patent (Rule 4.17(ii))*
- *as to the applicant's entitlement to claim the priority of the earlier application (Rule 4.17(iii))*

Published:

- *with international search report (Art. 21(3))*

PURGE TO INTERMEDIATE PRESSURE IN CRYOGENIC DISTILLATION

CROSS REFERENCE TO RELATED APPLICATIONS

[0001] This application claims the priority and benefit of U.S. Provisional Application No. 62/183,033 filed June 22, 2015, entitled PURGE TO INTERMEDIATE PRESSURE IN CRYOGENIC DISTILLATION, the entirety of which is incorporated by reference herein.

FIELD OF THE DISCLOSURE

[0002] The present techniques relate generally to purification of natural gas via cryogenic distillation, and more particularly, to one or more purges to an intermediate pressure vessel from the cryogenic distillation system.

BACKGROUND

[0003] This section is intended to introduce various aspects of the art, which may be associated with exemplary embodiments of the present techniques. This discussion is believed to assist in providing a framework to facilitate a better understanding of particular aspects of the present techniques. Accordingly, it should be understood that this section should be read in this light, and not necessarily as admissions of prior art.

[0004] Most raw natural gas extracted from the Earth is primarily methane (CH_4) with smaller amount of other hydrocarbons and varying degrees of different “contaminants.” The methane component, as a low molecular weight hydrocarbon, is typically the desirable component within the raw natural gas. Applications for methane recovered from raw natural gas may include fuel for vehicles, power generation, and so on. Moreover, in many cities, such CH_4 is piped into homes for domestic heating and cooking purposes. In this context, methane is usually known as natural gas.

[0005] The demand to provide effective techniques to separate and remove contaminants from raw natural gas to give purified natural gas or methane has significantly increased. The contaminants and impurities in raw natural gases may include acid gases such as carbon dioxide (CO_2), hydrogen sulfide (H_2S), mercaptans, and the like. The contaminants may also include nitrogen (N_2), helium (He), water vapor, liquid water, mercury, and so forth. Such contaminants and impurities may lead to equipment malfunction, production failure, product contamination, among other detrimental production issues. For example, CO_2 contaminant when combined with water may create a corrosive form of carbonic acid. Additionally, for example, at CO_2 concentrations of more than 2%, the presence of the CO_2 may reduce the BTU

value of the natural gas and lower the economic viability of the natural gas.

[0006] To remove CO₂ and other acid gases, the natural gas may be subjected to cryogenic distillation. Distillation is a widely used way of effecting relative-volatility based separations. A cryogenic distillation system may provide for separating mixtures based on their relative volatility at cryogenic temperatures. In some cryogenic distillation systems, one or more components may solidify. Indeed, in the cryogenic distillation, the operating temperature, pressure, and component concentrations may lead to solidification of the CO₂. Therefore, certain purification systems may control such solidification. For instance, the system may provide for a freezing section in a cryogenic distillation column in which CO₂ solidification is permitted and managed. If so, solidification of CO₂ outside of the freezing section is typically undesirable. Such solidification of CO₂ outside of the freezing section may be more prone to occur during startup of the distillation system when disturbances are more likely during initial conditions and because of the transient and non-linear behavior of startup. The undesired solidification, e.g., CO₂ solidification outside of the freezing section, may occur in the distillation column or elsewhere in the distillation column system, such as in the overhead system of the distillation column.

[0007] In general, during normal operations of the cryogenic distillation column, CO₂ contaminant, among other contaminants, may be separated and removed from the raw natural gas to produce a purified methane gas product. To startup the cryogenic distillation column, prior to steady state or normal operating conditions, various startup techniques may be implemented. For instance, the startup may be “assisted” in which solidification inhibitors, and/or a clean methane flow for liquid reflux, are utilized until normal operation is reached, while avoiding solidification in undesirable locations. Unfortunately, the solidification inhibitors and the clean methane gas either impose additional costs or may not be readily available.

[0008] In the worldwide natural-gas production industry where hundreds of trillions of standard cubic meters of raw natural gas are extracted from the Earth and processed, there remains an ongoing need to continuously improve such extraction and processing techniques. In particular, there remains a need for improved cryogenic distillation systems for separating contaminants from raw gas streams.

SUMMARY

[0009] An aspect relates to cryogenic distillation column system including a cryogenic distillation column having a freezing section, wherein the cryogenic distillation column is

configured to receive a natural gas having methane and acid gas, and to discharge an overhead stream rich in the methane, and a bottoms stream rich in the acid gas. The system has an overhead system configured to receive the overhead stream, and to discharge a vapor methane product, and a cooled liquid for reflux to the cryogenic distillation column. The system
5 includes an intermediate pressure vessel configured to receive a purge and operate at an intermediate pressure to reduce solidification of the purge.

[0010] Another aspect relates to a method of operating a cryogenic distillation column system, the method including feeding natural gas having methane and acid gas to a cryogenic distillation column having a freezing section, wherein the acid gas includes carbon dioxide.
10 The method includes discharging an overhead stream rich in the methane from the cryogenic distillation column to an overhead system. The method includes discharging, via a purge valve, a purge to an intermediate pressure vessel to reduce solidification of the carbon dioxide during startup of the cryogenic distillation column system.

[0011] Yet another aspect relates to a method of operating a cryogenic distillation column system, the method including feeding natural gas having methane and acid gas to a cryogenic distillation column having a freezing section, the acid gas including carbon dioxide. The method includes discharging an overhead stream rich in the methane from the cryogenic distillation column to an overhead system. The method includes purging to an intermediate pressure vessel to reduce solidification of carbon dioxide in the cryogenic distillation column
15 system during startup of the cryogenic distillation column system. Further, the method includes maintaining an operating pressure of the intermediate pressure vessel to reduce solidification of carbon dioxide in the purging. The method includes discharging purge material from the intermediate pressure vessel to a flare system or a recovery system.
20

DESCRIPTION OF THE DRAWINGS

[0012] The advantages of the present techniques are better understood by referring to the following detailed description and the attached drawings, in which:

[0013] FIG. 1 is a block flow diagram of an exemplary system for the production of natural gas or methane;

[0014] FIG. 2 is a simplified process flow diagram of a cryogenic distillation column system of the production system of FIG. 1;

[0015] FIG. 3 is a simplified process flow diagram of another configuration of the cryogenic distillation column system of FIG. 2;

[0016] FIG. 4 is a simplified process flow diagram of an overhead system of the cryogenic

distillation column systems of FIGS. 2 and 3;

[0017] FIG. 5 is a simplified process flow diagram of a vent system for the cryogenic distillation column systems of FIGS. 2 and 3;

[0018] FIG. 6 is a block diagram of an exemplary method of operating a cryogenic distillation column system; and

[0019] FIG. 7 is a simplified process flow diagram of a cryogenic distillation column system employed in the production and purification of natural gas and methane.

DETAILED DESCRIPTION

[0020] In the following detailed description section, specific embodiments of the present techniques are described. However, to the extent that the following description is specific to a particular embodiment or a particular use of the present techniques, this is intended to be for exemplary purposes only and simply provides a description of the exemplary embodiments. Accordingly, the techniques are not limited to the specific embodiments described below, but rather, include all alternatives, modifications, and equivalents falling within the true spirit and scope of the appended claims.

[0021] In the production of natural gas to give a methane gas product, embodiments of the present techniques provide for purges from a cryogenic distillation column system to facilitate startup of the cryogenic distillation column system. The purges may discharge from the relatively high-pressure distillation to intermediate pressure prior to introduction of the purge material to a lower pressure recovery or disposal system, such as a flare system. Therefore, as discussed below, purge streams having contaminant concentrations, such as CO₂ concentrations, may be beneficially discharged without experiencing solidification, or if some solidification does occur, the solids may be more suitably managed and eventually melted, vaporized, or sublimed. Moreover, in contrast to “assisted” start-up methods in which solidification inhibitors and/or a clean methane supply is utilized, embodiments herein may advantageously provide for an “unassisted” startup. This startup may be unassisted in the sense that raw natural gas is the source of gas used to generate the liquid reflux for the cryogenic distillation column, and to transition the cryogenic distillation to normal operation. Beneficially, the use of solidification inhibitors and/or a supply of clean natural gas may be avoided.

[0022] The cryogenic distillation column may be configured with a freezing section in which CO₂ solidification is permitted and managed. Solidification of CO₂ outside of the freezing section is normally undesirable. Undesired solidification, i.e., CO₂ solidification

outside of the freezing section, may occur in the distillation column or elsewhere in the distillation column system, such as in the overhead system of the distillation column. This solidification of CO₂ outside of the freezing section may be more prone to occur during an “unassisted” startup of the cryogenic distillation system where high CO₂ content gas circulates through the rectifier and overhead system before the CO₂ is reduced to a low enough content such that liquid reflux can be generated. Therefore, the distillation column system is desirably configured with purge points that allow discharge of problematic fluids or streams that might experience solidification, or that might contribute to solidification in the distillation column, piping, valves, or associated equipment. The purge discharge may occur temporarily, intermittently, or continuously.

[0023] A slipstream or all of an overhead stream unusually high in CO₂ concentration, for example, may be purged to a flare or recovery system, for instance. Further, warm fluids, such as from vessels or pipes that are being cooled for service, and which might disrupt the startup, may also be purged to the flare. Flare systems, including flare headers, typically operate at low pressure, such as at near atmospheric or at only a few pounds per square inch (psig). Thus, a purge of a stream high in CO₂ concentration from the higher pressure cryogenic distillation system to a low pressure flare or recovery system may result in solidification of CO₂ in the purge or vent. In response, embodiments of the present techniques provide an intermediate-pressure drum or vessel disposed operationally between the distillation system and the flare system. The intermediate pressure vessel is configured to receive the purges or vents, and to discharge the purged material to a flare header and/or directly to a flare knockout drum (also known as a knockout pot). The intermediate pressure vessel is maintained at an intermediate pressure to provide for a more gradual step-down in pressure of the purges/vents from the distillation column to the flare so to avoid or reduce unwanted solidification.

[0024] For ease of reference, certain terms used in this application and their meanings as used in this context are set forth. To the extent a term used herein is not defined below, it should be given the broadest definition persons in the pertinent art have given that term as reflected in at least one printed publication or issued patent. Further, the present techniques are not limited by the usage of the terms shown below, as all equivalents, synonyms, new developments, and terms or techniques that serve the same or a similar purpose are considered to be within the scope of the present claims.

[0025] As one of ordinary skill would appreciate, different persons may refer to the same feature or component by different names. This document does not intend to distinguish between components or features that differ in name only. The figures are not necessarily to

scale. Certain features and components herein may be shown exaggerated in scale or in schematic form and some details of conventional elements may not be shown in the interest of clarity and conciseness. When referring to the figures described herein, the same reference numerals may be referenced in multiple figures for the sake of simplicity. In the following
5 description and in the claims, the terms “including” and “comprising” are used in an open-ended fashion.

[0026] As used herein, “substantially”, “predominately”, and other words of degree are relative modifiers intended to indicate permissible variation from the characteristic so modified. It is not intended to be limited to the absolute value or characteristic which it
10 modifies, but rather possessing more of the physical or functional characteristic than its opposite, and preferably, approaching or approximating such a physical or functional characteristic.

[0027] The term “rich” may refer to a compound or class of compounds that is most represented in a stream, by weight or mole percent. The term “rich” can refer to a compound
15 or class of compounds that is a majority of a stream, by weight or mole percent. Lastly, the term “rich”, when used in reference to a product stream, indicates that the product stream comprises a relatively higher amount of a certain component than the feed stream from which the product stream originated.

[0028] “Exemplary” is used exclusively herein to mean “serving as an example, instance,
20 or illustration.” Any embodiment described herein as “exemplary” is not to be construed as preferred or advantageous over other embodiments.

[0029] The term “acid gas” refers to any gas that dissolves in water to produce an acidic solution. Non-limiting examples of acid gases include hydrogen sulfide (H₂S), carbon dioxide (CO₂), sulfur dioxide (SO₂), or mixtures thereof. The term “trace sulfur compounds” includes
25 carbon disulfide (CS₂), carbonyl sulfide (COS), mercaptans, or mixtures thereof. The term “acid gas injection” (AGI) refers to the disposal of an acid gas stream by compressing it and introducing the pressurized stream into a subterranean reservoir.

[0030] The term “cooling cycle” may refer to a refrigeration system comprising compression, heat exchange, and pressure reduction means to ultimately provide cooling.

[0031] The term “Controlled Freeze Zone™ (CFZ™) process” is a “cryogenic distillation”
30 process that takes advantage of the freezing potential of carbon dioxide in cryogenic distillation, rather than avoiding solidification of carbon dioxide. In the CFZ™ process, acid gas components are separated by cryogenic distillation through the controlled freezing and melting of carbon dioxide in a column, without the use of freeze-suppression additives. The

CFZTM process uses a cryogenic distillation column with a special internal freezing section, e.g., CFZTM section or Controlled Freeze ZoneTM section, to handle the desired controlled solidification and subsequent melting of carbon dioxide. This CFZTM section or Controlled Freeze ZoneTM section generally does not contain packing or typical trays such as in conventional distillation columns. Instead, this section typically contains one or more banks of spray nozzles and a melt tray. Of course, other configurations of the CFZTM section or freezing section may be employed. In operation, solid CO₂ forms in the vapor space in the CFZTM section and falls into the liquid on the melt tray. Substantially all of the solids that form are confined to the freezing section. The portions of the distillation column above and below the freezing section of the column may be similar to conventional cryogenic demethanizer columns. For example, in the rectification section above the CFZTM section, methane is enriched to produce an overhead methane-rich stream, while the stripping section below the CFZTM section generates a liquid bottoms stream containing contaminants such as CO₂ and hydrogen sulfide with very little residual methane. The bottoms stream may be used for enhanced oil recovery or reinjected below ground for disposal by geosequestration.

[0032] The term “cryogenic” in the natural gas industry is used to refer to low temperature conditions, often those involved with propane at atmospheric pressure, such as -40 °F degrees Fahrenheit (-40 °C degrees Celsius) and colder. Cryogenic used in reference to a fluid means that the fluid is at a cryogenic temperature. Cryogenic used in reference to an item or material means that the item or material is suitable for operating at a cryogenic temperature and/or suitable for containing a cryogenic fluid. For example, a cryogenic riser is a riser that is suitable for containing a cryogenic fluid.

[0033] The term “heat exchanger” refers to a device designed to efficiently transfer or “exchange” heat from one matter to another. Exemplary heat exchanger types include co-current or counter-current heat exchangers, indirect heat exchangers (e.g. spiral wound heat exchanger, plate-frame heat exchanger, plate-fin heat exchanger, shell-and-tube heat exchanger, etc.), direct contact heat exchangers, or combinations thereof.

[0034] The terms “rectifying section”, rectifying zone”, “rectification section”, and “rectification zone” refer to the stages above a feed point of a distillation column where the concentration of the more volatile component increases in both the liquid and the vapor. In a particular example of a cryogenic distillation column, the rectifying section refers to a section of the cryogenic distillation column where an overhead methane-rich vapor stream may be purified to meet pipeline or liquid natural gas (LNG) feed quality via conventional distillation.

[0035] The term “reflux” refers to a stream introduced into a distillation column at any

location above the location at which the feed is introduced into the column, wherein the reflux is used to reduce the content of heavy components in the overhead product stream.

[0036] The terms “stripping section” and “stripping zone” refer to the stage below the feed point of a distillation column where the concentration of the more volatile component decreases in both the liquid and the vapor. For example, the stripping section refers to a section of a cryogenic distillation column in which a liquid bottoms stream rich in contaminants such as CO₂ and H₂S is produced.

[0037] A distillation process may be used to separate contaminants from a raw natural gas stream. The distillation process is the most widely used way of effecting relative volatility based separations. The separation of carbon dioxide from natural gas, in essence methane, by distillation involves cryogenic temperatures that lead to solidification of CO₂ in an intermediate portion of a distillation column. As described above, the Controlled Freeze Zone™ (CFZ™) process includes a CFZ™ column that has a specially designed freezing section, i.e., a CFZ™ section or zone, to address the freezing and melting of solids within this intermediate portion of the column. Specifically, the CFZ™ section acts to control and to confine the freezing and melting of solids within its section.

[0038] During normal operations, the cryogenic distillation operation may include feeding the natural gas into the distillation column where a stream that includes lighter vapors, e.g., methane vapors, may rise upward into a rectification zone. In certain embodiments, these light vapors may flow through a freezing section. Contaminants, for example, an acid gas stream that includes CO₂, may be removed from the upward flowing vapor stream, the vapor stream discharging as an overhead stream rich in CH₄ and with reduced CO₂. The acid gas stream flows downward in the cryogenic distillation column through a stripping section and discharges liquid rich in CO₂ as a bottoms stream for disposal or recovery. The cryogenic distillation column may receive and process a natural gas feed with a wide range of CO₂ content, for example, from about 8 mol% to about 75 mol%. The overhead stream rich in CH₄ may be partially condensed to provide a liquid reflux that may be collected in a reflux accumulator. The liquid reflux in the reflux accumulator may be refluxed, e.g., via a reflux pump and piping, into the rectification zone to promote and complete the separation and removal of the contaminants.

[0039] However, before the cryogenic distillation column can produce a purified CH₄ gas product at normal operating conditions, the column generally undergoes start-up if the column has been shut down or idled, for example. The start-up may include taking the cryogenic distillation column from initial operating conditions to normal operating conditions. Several

techniques have been devised to start cryogenic distillation columns for natural gas. For example, an assisted startup technique may make use of solidification inhibitors during startup, while another assisted startup technique employs relatively clean methane to generate clean reflux liquids during startup. However, unassisted startup techniques, depend primarily or
5 only on the components available to the unit from the feed gas and typical utility streams, and generally have no requirements for external or special fluids.

[0040] In the present embodiments, the startup technique may use the overhead stream in a cooling cycle of the cryogenic distillation column. The cooling cycle may include the compression, cooling, and expansion of the overhead stream to provide a liquid for the liquid
10 spray into the freezing section of the column. During the initial phase of startup, the overhead stream may include a high concentration of CO₂ contaminant, for example, greater than 10 mol%, or greater than 12 mol%, or greater than 14 mol%, or greater than 16 mol%, or greater than 18 mol%, or greater than 20 mol%, as compared to the less than 6 mol%, or less than 4 mol%, or less than 2 mol% concentration of CO₂ contaminant that is typically present in normal
15 operations. The presence of fluids with CO₂ concentrations well in excess of the normal range for non-solidification portions of the equipment requires special attention if CO₂ solidification in undesired locations is to be avoided.

[0041] The start-up of the cryogenic distillation column may involve establishing the appropriate distillation column temperature and composition profiles that provide the desired
20 separations and product purities while avoiding solidification, or at least limiting the solidification and melting of solids to within a freezing section zone. Unfortunately, system disturbances in the cryogenic distillation process that may occur during startup operations may cause the contaminants to solidify within the column, and may even cause contaminants to solidify outside of a freezing section of the column. The formation of solid CO₂ may
25 accumulate on system equipment and eventually lead to plugging and equipment malfunction. Therefore, embodiments described herein are directed to a system and method during start-up operations to eliminate or reduce solid component formation or to confine solid component formation to a freezing section of the cryogenic distillation column in order to prevent or reduce fouling of process equipment. Again, during start-up operations, the overhead stream
30 discharging from the cryogenic distillation column may include contaminants (e.g., CO₂) that can solidify during the start-up operations. Thus, purging from the distillation system streams having high concentrations in contaminants and/or being exposed to operating conditions that would lead to unwanted solidification may be needed to reduce unwanted solidification. It should be noted that while the discussion herein may at times focus on aspects of the CFZ™

technology, the present techniques are applicable to other technologies involving cryogenic distillation in general of raw natural gas, with or without a freezing section.

[0042] FIG. 1 is a block flow diagram of a natural-gas production system **100** for producing and processing natural gas or methane. The production system **100** has an intermediate-pressure purge vessel, which is labeled herein as an intermediate pressure vessel **102**, to facilitate startup of cryogenic distillation in the purification of the natural gas. The production system **100** processes raw natural gas **104** extracted from the Earth via one or more production wells **105**. The raw natural gas **104** may include a substantial amount of contaminants, e.g., acid gases such as CO₂ and/or H₂S, and in some embodiments, may have a low-British Thermal Unit (BTU) content, e.g., between about 500 and 950 BTUs per standard cubic foot. In the production system **100**, the raw natural gas **104** may be subjected to dehydration, refrigeration cooling, and other processing in one or more different types of treatment units **106**, prior to introduction as feed **108** into a purification system **110**. The purification unit **110** removes contaminants as liquid acid gas **112** to give a product **114**, e.g., a purified product. The product **114**, which is primarily methane, may be used for power generation and/or other applications.

[0043] The dehydration in a treatment unit **106** may involve removal of water vapor using glycol dehydration, or molecular sieve desiccants, among other processes. Generally, dehydration may remove water from the raw natural gas **104** to prepare the natural gas for cryogenic separation in the purification system **110** in which water would freeze and plug piping and equipment. The dehydrated raw natural gas may be further processed by refrigeration cooling or chilling prior to being fed into cryogenic distillation in the purification system **110**. The refrigeration may be in a treatment unit **106** or in the purification system **110**. The dehydrated raw natural gas may enter a chiller or refrigeration system where its temperature may be lowered, for example, to about -60 °F (approximately -51 °C). A dry chilled natural gas may emerge from the chiller to be introduced as feed **108** into a cryogenic distillation column system **116** in the purification system **110**. Alternatively, the dry chilled natural gas could be subjected to further cooling as by expansion through a Joule-Thompson (JT) valve, for example, prior to introduction as feed **108** into the cryogenic distillation column system **116** in the purification system **110**. Of course, other processing of the raw natural gas **104** prior to introduction to the cryogenic distillation column system **116** may be implemented. Moreover, natural gas liquids (NGL) may also be removed from the feed **108** in the purification system **110** if not removed in treatment unit **106**.

[0044] Again, the purification system **110** may remove contaminants, such as acid gas **112**, from the feed **108** via a cryogenic distillation column system **116** to give a product **114**, i.e., a

purified CH₄ product. As mentioned, the cryogenic distillation may involve the Controlled Freeze Zone™ (CFZ™) technology available from ExxonMobil, or other cryogenic processes. These cryogenic processes can generate a liquid acid gas **112** that includes CO₂ and H₂S, as well as other compounds. In addition to removing the liquid acid gas **112**, the purification system **110** may also remove higher carbon number hydrocarbons, e.g., C₂ and higher. Moreover, as also indicated, the cryogenic distillation column system **116** may include an intermediate pressure vessel **102** to facilitate startup of the cryogenic distillation column system **116**. The intermediate pressure vessel **102** will be discussed in more detail below with respect to subsequent figures.

[0045] Additionally, the production system **100** may generally include one or more control systems **118** which may include a distributed control system (DCS), programmable logic controller (PLC), and so on. The control system **118** may have a human interface (HMI) and facilitate control of the production system **100** including the intermediate pressure vessel **102** and the purges in the purification system **110** to the intermediate pressure. In general, the control system **118** may direct the operation of equipment associated with the system **100**. The control system **118** may include instrumentation, computers, computer memory, a processor, and so forth. Further, the control system **118** may include control schemes stored in memory and executable by the processor to implement or facilitate control actions associated with the operation of the equipment.

[0046] In sum, the raw natural gas **104** may be produced from a well **105** and then dehydrated and chilled in a treatment unit **106**, and fed as feed **108** to the purification system **110** and its cryogenic distillation column system **116**. The feed **108** may include methane (CH₄) and at least one acid gas component, e.g., carbon dioxide (CO₂). In examples, the feed **108** may include a CO₂ concentration ranging from about 8 mol% to about 70 mol%, along with concentrations of hydrogen sulfide (H₂S), mercaptans, heavier hydrocarbons, among other contaminants and impurities. At higher concentrations, the CO₂ contaminant acts as a diluent and significantly lowers the heat content of natural gas. In the presence of water, the CO₂ contaminant renders the natural gas as a corrosive gas. Removal of the CO₂ contaminant from the feed **108** is desirable to provide a natural gas product for delivery to a pipeline, natural gas liquids recovery, helium recovery, nitrogen recovery, or conversion to liquid natural gas, among others applications.

[0047] FIG. 2 is a simplified process flow diagram of a cryogenic distillation column system **116** of the production system **100** of FIG. 1. The system **116** includes the intermediate pressure vessel **102** to receive and collect one or more purges **200** from within the cryogenic

distillation column system **116**, such as from an overhead system or from the distillation column. Then, the purge material **202** is discharged generally as a vapor stream from the intermediate pressure vessel **102** to a flare system/recovery system (not shown). The pressure in the intermediate pressure vessel **102** may be maintained at an intermediate pressure via a pressure control valve **204** or backpressure control valve, for example. Moreover, the intermediate pressure vessel **102** may have a heater **206** to melt solids that may accumulate in the pressure vessel **102**. While the heater **206** is depicted as disposed internal to the intermediate pressure vessel **102**, the heater **206** may instead be disposed external to the intermediate pressure vessel **102**.

[0048] The cryogenic process of system **116** employs a cryogenic distillation column **208** that separates and removes the CO₂ contaminant from the feed **108**. In certain embodiments, the column **208** may be a CFZTM column. In other embodiments, the column **208** is not a CFZTM column. In general, the column **208** may be used for the condensation, separation, and removal of the CO₂ contaminant from the feed **108**, e.g., dehydrated and chilled natural gas, to produce a product **114**, i.e., a purified CH₄ gas product. In the illustrated embodiment, the column **208** has a freezing section **210**. In certain embodiments, the freezing section **210** may be a CFZTM section. In other embodiments, the column **208** and freezing section **210** is not associated with the CFZTM technology. The feed **108** may enter the column **208** at a stripping section below the freezing section **210**. A vapor stream richer in the more volatile compound, e.g., methane, may flow upward through the freezing section **210** and continue through a rectifying section of the column **208** to discharge as a vapor overhead stream **212**. A liquid stream richer in the less volatile component, e.g., CO₂, may travel down the column **208** through the stripping section and discharge as a bottoms stream rich in acid gas **112** including CO₂. A portion of the bottoms stream may be vaporized in a reboiler (not shown) and return as vapor and reintroduced to the column **208**.

[0049] The overhead stream **212** rich in methane, i.e., a purified CH₄ stream, may enter an overhead system **214** to give a methane reflux **216** and a product **114**, i.e., a methane product. To do so, the overhead system **214** may include a condensation system **218** and a reflux system **220**. As discussed below, the condensation system **218** may include a heat exchanger, compressor, cooler, a Joule-Thompson (JT) expansion valve, turbo-expander, and so forth. The reflux **216** may provide a rectifier reflux **224**. The reflux **216** may additionally provide a stripper reflux, or spray liquids **228**, to the freezing section **210**. It should be noted that where a spray assembly is not employed in the cryogenic distillation, the spray liquids **228** may be merely labeled as a reflux stream, which may enter through column through a distributor, for

example. Where a spray is employed, the spray liquids **228** may flow through a spray distributor or spray assembly in the column **208** to spray in the freezing section **210**. Further, a draw **226** may be taken from the bottom of the rectifying section, for example, to supplement the stripper reflux or spray liquids **228**. Moreover, as discussed below, a startup stream **222** from the overhead system **214** may enter across, for example, a JT valve **230**, i.e., a JT-expansion valve, to the spray assembly at the freezing section **210**. Furthermore, during startup, purges may be collected from the column **208**. For instance, a purge (not shown) may discharge from the bottom of the rectification section to the intermediate pressure vessel **102**.

[0050] As for the reflux system **220**, the reflux system **220** may include a reflux accumulator vessel, a reflux pump, and other equipment. During startup of the cryogenic distillation column system **116**, the overhead system **214** may provide the aforementioned startup stream **222** rich in methane, but with CO₂ content in excess of normal operating conditions. The startup stream **222** may be vapor and provided from the condensation system **218** to the column **208**, bypassing the reflux system **220**. Provision of this startup stream **222** may be discontinued once a reflux **216** via the reflux system **220** is established. The startup stream **222** may go across a JT valve **230** or nozzle at the distillation column **208** to condense into liquid. Moreover, the startup stream **222** may be provided for spray in the distillation column **208** during startup. Of course, various other configurations for a startup stream are contemplated.

[0051] In general, before normal operations can begin, start-up operations may take place to move the column system **116**, e.g., a cryogenic distillation system, such as a CFZ™ column system, from a shutdown and empty state into a steady state, e.g., normal operations. As will be further described, the startup operations of the system **116** may progressively reduce the CO₂ content, originating from the feed **108**, in a startup stream **222** that may serve as a spray to lower the concentration of the CO₂ contaminant in the overhead stream **212** until a non-solidifying concentration target is reached. The startup operations may employ the condensation system **218** having a cooling cycle, as will also be later described, which continually recycles the overhead stream **212** until the concentration of CO₂ in the overhead stream **212** is lowered to a normal, non-solidifying range of about 2 mol% to about 6 mol%, or from about 3 mol% to about 5 mol%. However, there are often many challenges related to the startup of a distillation column **208** due to nonlinear and wide operating ranges of the column during transient periods. For instance, in the “unassisted” startup of the distillation column **208**, high CO₂ content gas circulates through the rectifier portion of column **208** and through the overhead system **214**, before the CO₂ content is reduced to a non-solids forming

concentration and before liquid reflux is generated. Residual fluids having high CO₂ content may remain in vessels or lines not yet placed in full operation in the startup sequence. Unless these fluids having high CO₂ are purged, the fluids can lead to solidification when exposed to cold temperatures as the vessels, lines and other system components having the residual fluids are brought into full (normal) operation at cryogenic temperatures.

[0052] In embodiments, the pressure vessel **102** accepts purges **200**, so that solidification outside of the freezing section **210** is reduced or prevented. In the illustrated embodiment, one or more purges **200** are vented from the overhead system **214** to the intermediate pressure vessel **102**. Of course, purges **200** may originate and be collected from systems and equipment within the distillation column system other than the overhead system **214**. Indeed, purge(s) **200** may typically also originate from systems and components connecting to column **208**. Moreover, as previously indicated, the control system **118** may direct the operation of equipment, the amount and location of the purges/vents, the pressure control valve **204**, and so on.

[0053] Certain embodiments may provide the intermediate pressure drum or vessel **102** such that little or no solids are formed via the purges **200**. If solids are likely to form even at the intermediate pressure, the contents of the vessel **102** in particular embodiments may be heated via the heater **206** such that solids formed are melted or solidification conditions are generally avoided or not achieved. The contents of the intermediate pressure vessel **102** can then be sent as purge material **202** in a more controlled fashion to a flare system. In other words, such a configuration may be more controlled than a direct purge from the distillation to the flare system, without an intermediate vessel. Moreover, the flare system may be a conventional flare system serving the overall facility. By providing the intermediate pressure vessel **102** operating at an exemplary pressure only marginally lower (e.g., about 50 psig lower, or about 75 psig lower, or about 100 psig lower, or about 125 psig lower, or about 150 psig lower, or about 175 psig lower, or about 200 psig lower) than the operating pressure of the distillation and respective purge, the solidification by expansion cooling across the purge control valves is substantially reduced or eliminated. Advantageously, the purges **200** and the intermediate pressure vessel **102** may promote reliable cool-down of the reflux accumulator and piping that does not come into play until later in the startup procedure and in normal operation. Indeed, the fluids used for this cooling may be sent to the purge vessel or intermediate pressure vessel **102** rather than disrupting the startup procedure. In contrast, keeping these fluids, which are warmed by cooling the reflux system, in the cryogenic distillation column system **116** rather than purging these fluids could disrupt the startup

procedure by introducing relatively warm fluid streams into an otherwise colder portion of the system **116**. Furthermore, the purges **200** and the intermediate pressure vessel **102** may provide for reliable purging of high CO₂ fluids. For example, high CO₂ fluids may be purged from the bottoms of the rectifying section in the column **208** and/or from an associated spray pump or reflux pump, for instance.

[0054] FIG. 3 is a simplified process flow diagram of the cryogenic distillation column system **116A** which is another configuration of the cryogenic distillation column system **116** of FIG. 2. The system **116A** includes the intermediate pressure vessel **102**. In the illustrated embodiment of FIG. 3, the cryogenic distillation column **208** of FIG. 2 is configured as a split-tower design into two columns **300** and **302**. This split-tower design may be implemented in an effort to facilitate offshore deployment, to provide for ease of transportation and installation, to facilitate start-up of the column **208**, and so forth. In the embodiment illustrated in Figure 3, the column **208** of Figure 2, is replaced by the first column **300**, where a stripping section and a freezing section **210** are disposed, and the second column **302**, which includes a rectifying section to receive vapor **303** from the first column **300**. The arrangement of the columns **300** and **302** may vary depending on usage requirements. In the illustrated embodiment, the second column **302** may operate as typical distillation column. In other embodiments, the freezing section **210** may be included in the second column **302** with and below the rectifying section. Other column configurations may be used.

[0055] The cryogenic distillation column system **116A** includes the overhead system **214**. Moreover, as previously indicated, the control system **118** may direct the operation of equipment including operation of the overhead system, the amount and location of the purges/vents, and so forth. The control system **118** may, for example, direct the pressure control valve **204** on the intermediate pressure vessel **102** to maintain the vessel **102** at a desired pressure.

[0056] Referring to FIG. 3, during startup operations the feed **108** may be introduced into the first column **300** at the stripping section to produce a CO₂ rich bottoms stream, i.e., a bottoms stream of liquid acid gas **112**, and a freezing section vapor stream. The CO₂ rich bottoms stream **112** may exit the column **300** at a bottom outlet, as shown in FIG. 3. The freezing section vapor stream may be at a temperature and pressure with a substantially lower concentration of CO₂ than stream **112**, and may flow into the freezing section **210**. The vapor stream exiting the freezing section **210** may discharge from the lower or first column **300** as the vapor **303** stream to the second column **302**, and continue to flow upwards through the rectifying section of the column **302** and exit as the overhead stream **212**.

[0057] As discussed with respect to FIG. 2, the overhead system **214** may provide a reflux **216**. In the illustrated embodiment of FIG. 3 for normal operation, the reflux **216** may be split and sent as spray liquids **228** to the first column **300** and as rectifier reflux **224** to the second column **302**. In certain embodiments, the spray liquids **228** may be CFZTM spray liquids. In examples, the bottoms or draw **226** of the rectifying section or the second column **302** may constitute the majority of the reflux or spray liquids **228** during normal operations. The cryogenic distillation column **208** of FIG. 2 or the first column **300** of FIG. 3 may include a spray assembly **308** to spray the spray liquids **228**. Such spray may facilitate operation of the freezing section **210** and the stripping section. It should be noted that during startup prior to establishment of a consistent or reliable reflux **216**, a startup stream **222** may be provided from the overhead system **214**, e.g., from the condensation system **218**, to at least the first column **300** for the liquid spray through the spray assembly **308**.

[0058] As discussed, an unassisted startup procedure for the cryogenic distillation system may be one that generally does not involve the use of solidification inhibitors or the need for a supply of clean methane to generate clean reflux liquids. The unassisted startup may handle high CO₂ streams at non-solidifying conditions, and at solidifying conditions in the freezing section **210**. After being chilled in upstream feed chillers and introduced into the first column **300**, high CO₂ vapors, e.g., about 20 mol% CO₂, emerge from the melt tray of the freezing section **210** and discharge in the overhead stream **212**.

[0059] During startup, the overhead stream **212** is compressed by the condensation system **218** of the overhead system **214** to pressures, e.g., 1000-1500 psig, which are above the CH₄-CO₂ solidification dome, e.g., 719 psig, of the thermodynamic equilibrium phase diagram, such as the phase diagram depicted in Donnelly, H.G. and Katz, D.L., Industrial and Engineering Chemistry, Vol. 46, No. 3, pp. 513 and 515, year 1954. The compressed stream is then chilled against itself, i.e., against the lower-pressure overhead stream **212** feed to the compressor, and is sent as startup stream **222** across a JT valve **230** into the freezing section **210** through startup nozzles or spray assembly **308**. The effect of dropping from the compressed pressure to the pressure of the freezing section **210** (e.g., 550-600 psig) lowers the temperature of the stream **222**, condensing some vapor in the stream **222** into liquid. This condensed liquid in stream **222** is sprayed via the spray assembly **308** into the freezing section **210**. The spray of stream **222**, over time during the startup, reduces progressively the concentration of CO₂ in the vapor stream leaving the freezing section **210**. In embodiments, an equilibrium involving temperature and composition gradually shifts over time to reduce CO₂ content. In other words, the stream **222** over time is gradually cooler and progressively lower in CO₂ content, and the

vapor stream leaving the freezing section **210** becomes progressively lower in CO₂ content. The overhead stream **212** enters an overhead heat exchanger and compressor of the condensation system **218** to continue the cycling loop. As discussed below, the startup stream **222** may be discontinued when the normal reflux becomes available. It should be noted that
5 the spray assembly **308** may also be employed during normal operations. Moreover, the spray assembly **308** and the freezing section **210** may be in a combined freezing section **304**.

[0060] After some time during startup, as the CO₂ concentration drops into a non-solidifying range, e.g., to 2-6 mol% CO₂, or from 3-5 mol%, a portion of the high-pressure cooled vapors of the startup stream **222** is diverted towards the reflux system **220**. The diverted
10 vapors are sent across a valve, e.g., the JT valve **406** of FIG. 4, to generate reflux liquids for accumulation in the reflux accumulator. Once sufficient reflux liquids are accumulated, the reflux **216** can be directed, such as via rectifier reflux **224** stream, to the rectifying section of column **208** (FIG. 2) or to the second column **302**, e.g., rectifying column, in the split tower configuration. In turn, when sufficient liquids accumulate at the bottom of the rectifying
15 section of column **208** or at the bottom of the second column **302** of the split tower configuration, a draw **226** can provide a source of spray to the freezing section **210**. In other words, a spray pump can be started and provide sprays via draw **226**, spray liquids **228**, and spray assembly **308**, as in normal operations, to the freezing section **210**. At that point in time for certain embodiments, the flow of high-pressure compressed overhead vapors via startup
20 stream **222** to the freezing section **210** can be discontinued and startup concluded, and the cryogenic distillation system **116** or **116A** stabilized toward steady conditions and normal operation.

[0061] At various stages of these techniques, there may be a need to cool certain vessels and piping, and purge some volumes that may contain residual high CO₂ concentration vapors.
25 Indeed, keeping the fluids of these cooling operations in the distillation system would have disruptive repercussions and are instead generally better purged from the distillation system, such as on an intermittent and/or substantially continuous basis. However, the purging of fluids from the distillation system to a near-atmospheric flare system, for example, may carry the potential for plugging of the purging system. Indeed, the temperature and pressures
30 downstream of the purge valves (without the intervening vessel **102**) can frequently be in the thermodynamic solidification regime. Moreover, the solids may not be swept away efficiently by the vapor or liquid being purged. Plugging of the purge system can prevent appropriate purging of the cryogenic distillation column system **116** or **116A** and, in turn, prevent the successful startup of the system **116** or **116A**.

[0062] Therefore, as discussed, systems **116** and **116A** include the intermediate pressure vessel **102** to receive and collect one or more purges **200** from within the distillation column system **116** or **116A**, such as from an overhead system **214** or from the cryogenic distillation column **208** or from the second column **302**, i.e., rectifying column, in the split-tower configuration. Then, the purge material **202** is discharged from the intermediate pressure vessel **102** to a flare system/recovery system (not shown). As mentioned, the pressure in the vessel **102** may be maintained at an intermediate pressure, e.g., at least about 50 psig lower than the pressure upstream of the purge, via, for example, a pressure control valve **204** or backpressure control valve. Thus, solidification of the purges **200** may be reduced or avoided. In general, examples may maintain the operating of pressure of the intermediate pressure vessel **102** via a pressure control valve **204** disposed on a conduit conveying the purge material from the intermediate pressure vessel to the flare system or the recovery system. Moreover, the intermediate pressure vessel **102** may have a heater **206** to prevent solids from forming and/or to melt solids that may accumulate in the vessel **102**. The heater **206** may also vaporize liquids.

[0063] FIG. 4 is a simplified process flow diagram of an overhead system **214** of the cryogenic distillation column systems of FIGS. 2 and 3. As mentioned, the start-up operations of the cryogenic distillation systems **116** and **116A** may progressively reduce the CO₂ content in a startup stream **222**. The stream **222** may serve as a reflux or a spray to lower the concentration of the CO₂ contaminant in the vapor flowing upward through the column. The stream **222** may be employed, for example, until a non-solidifying CO₂ concentration target is reached in the overhead stream **212**. The start-up operations may include a cooling cycle, as depicted, that continually recycles the overhead stream **212** until the concentration of CO₂ is lowered to a normal, non-solidifying range of, for example, about 2 mol% to about 6 mol%. In the illustrated embodiment, the overhead system **214** has a condensation system **218** (FIGS. 2 and 3) having a heat exchanger **400** such as a plate-fin heat exchanger, a compressor **402**, a cooler **404** such as an air cooler, and at least one JT valve **406**, and so on. The overhead system **214** also includes a reflux system **220** (FIGS. 2 and 3) having a reflux accumulator or vessel **408**, a reflux pump **410**, and other equipment.

[0064] In operation, the overhead stream **212** from the column **208** (FIG. 2) or column **302** (FIG. 3) enters the heat exchanger **400** and is then sent as warm vapor **412**, e.g., at 60 °F, to the compressor **402**. The warm vapor **412** may be compressed by the compressor **402** and then cooled as compressed vapor **414** in the cooler **404** to give a cooled compressed vapor **416**. The cooled compressed vapor **416** may be analogous to the product **114**, i.e., methane product, of FIGS. 1-3. In addition, the cooled compressed vapor **416** may be returned as recycle **418** to

the heat exchanger **400** for cross-exchange with the incoming overhead stream **212**. In some embodiments, the compressor **402** may increase the pressure of the warm vapor **412** stream from about 600 psig to about 1250 psig, for example. The compressed vapor **414** and cooled compressed vapor **416** may have the same or similar composition as the overhead stream **212** that entered the heat exchanger **400**.

[0065] The cooled compressed vapor **416** as recycle **418** may be further cooled in the heat exchanger **400** to an operating temperature of about -90 °F, for example, and discharged as cold vapor **420**. The cold vapor **420** may be initially sent as the startup stream **222** for spraying in the column **208** or **300** during startup. Subsequently, as non-solidifying conditions are met, cold vapor **420** may be directed across the JT valve **406** to condense some of the vapor **420** and give a partially liquefied stream **422**. The liquid in stream **422** may be the reflux accumulated in the reflux vessel **408**, i.e., reflux accumulator vessel. Once an adequate amount of liquid is accumulated in the reflux vessel **408**, the reflux pump **410** may provide reflux **216** to the cryogenic distillation column **208** (FIG. 2) or to the second column **302** (FIG. 3), i.e., rectifying column. Further, a vapor flow **424** may be established from the reflux vessel **408** vapor space to the inlet of the heat exchanger **400** or elsewhere in the overhead system.

[0066] As discussed, the overhead system **214** may send intermittent or continuous purges **200** to the intermediate pressure vessel **102**. The purge materials **202** may be sent from the intermediate pressure vessel **102** to a flare or recovery system. Again, the pressure in the vessel **102** may be maintained at an intermediate pressure via a pressure control valve **204**, for example. Moreover, the intermediate pressure vessel **102** may have a heater (e.g., heater **206** of FIGS. 2 and 3) to melt solids that may accumulate in the vessel **102**. A control system **118** may facilitate control of the pressure and any heating in the vessel **102**. Lastly, exemplary purge points **426** are noted in FIG. 4. Of course, other locations for purge points may be implemented. These purge points **426** and other purge points may be the source location of some of the purges **200** to the intermediate pressure vessel **102**.

[0067] FIG. 5 is a simplified process flow diagram of an exemplary vent system **501** for the distillation column systems of FIGS. 2 and 3. The vent system includes aforementioned intermediate pressure vessel **102** and a flare knockout drum or vessel **500**. The intermediate pressure vessel may discharge the purge material **202** stream, i.e., a vent stream of the purge material from the distillation system, through a dedicated subheader, as depicted, or may discharge through a shared subheader or header. The flare knockout vessel **500** may receive other vent or purge streams (not shown) via flare headers and subheaders from various units and processes within the production system **100** and other systems. The flare knockout vessel

500 may discharge an overall vapor stream **502** to the flare (not shown). The knockout vessel may also have a heater **504** to vaporize any liquids generally that might accumulate in the knockout vessel **500**. Further, the heater **504** may vaporize and or melt any CO₂ that solidifies in the flare knockout vessel **500**.

5 **[0068]** In certain embodiments, the pressure control valve **204** may be disposed at or near the flare knockout vessel **500**. Indeed, the pressure control valve **204** may be placed on top of the flare knockout vessel **500** such that any solids that may form across the control valve **204** upon the pressure reduction may fall freely or substantially freely into the flare knockout vessel **500** to the bottom portion of the knockout vessel **500**. Further, the vapor discharge of the flare
10 knockout vessel **500** for overall vapor stream **502** may be located at a sufficient distance away from the entrance of the purge material **202** stream, i.e., the purge vapor stream, so to prevent or reduce any solids carry over in the overall vapor stream **502** to the flare. Additionally, the flare knockout vessel **500** may be configured with flow components, e.g., baffles, plates, etc. (not shown), so that the vapor exiting in the vapor stream **502** is forced to make a sudden change
15 in direction prior to exiting the vessel **500**. Such flow components may facilitate separation of any entrained solids back into the vessel **500** and prevent the solids from exiting the vessel **500** with the vapor stream **502**.

[0069] Further, a pressure element **506** may be disposed on the intermediate pressure vessel **102** to sense and indicate the pressure of the intermediate pressure vessel **102**. The indication
20 of the pressure via the pressure element **506** may be an input **508** to a control block of the control system **118** to adjust the valve opening position of the pressure control valve **204** to maintain the operating pressure of the vessel **102** at the pressure set point of the control block. Such a control scheme may be employed whether the control valve **204** is located at or near the knockout vessel **500** or further upstream closer to the intermediate pressure vessel **102**.

25 **[0070]** In summary, the separation of CO₂ from natural gas, in essence methane, by distillation may involve cryogenic temperatures that would lead to solidification of CO₂ in an intermediate portion of a distillation column or tower. Such solidification may plug and render conventional distillation towers inoperable. In contrast, a specially designed freezing section (e.g., freezing section **210** of FIGS. 2 and 3) in an intermediate portion of the column or tower,
30 may address the freezing and melting of solids within this intermediate portion of an otherwise conventional distillation tower. In particular embodiments, a freezing section may allow and manage freezing and solids formation, and the melting of the solids formed.

[0071] The startup of a cryogenic distillation column system may involve establishing the appropriate distillation column temperature and composition profiles that provide the desired

separations and product purities while avoiding solidification. In some embodiments, solidification and melting of solids may be allowed and substantially limited to within the freezing section. As mentioned, several techniques have been devised to start cryogenic distillation units. One technique makes use of solidification inhibitors during startup, while
5 another technique makes use of relatively clean methane to generate clean reflux liquids during start up. Yet other startup techniques, such as the unassisted or bootstrap techniques, may generally depend only on the components available to the unit from the feed and has no or little requirements for external or special fluids.

[0072] The unassisted or bootstrap approach may involve the handling of high CO₂ vapor
10 streams, e.g., having about 20% CO₂, in portions of the distillation system that normally do not see vapor streams with CO₂ concentrations above 2% to 6%. The highest CO₂ concentrations in vapors rising from the feed stage generally occur in time at initiation of startup. The rising vapors are at a cold temperature but generally not so cold to solidify CO₂. The high CO₂ concentration vapors discharge from the column overhead and are compressed to a pressure
15 above the solidification “dome.” The compressed vapors are then reduced in pressure (e.g., through valve **230**) at the inlet of special spray nozzles in the freezing section of the cryogenic distillation column. The compressed vapor may experience the pressure reduction by flowing through a pressure reducing valve (JT valve **230**) at the inlet of the spray nozzles. This expansion via reduction in pressure lowers the temperature of the compressed vapor flowing
20 across the pressure reducing valve at the inlet to the spray nozzles. This reduction in pressure and reduction in temperature of the compressed vapor during startup can result in solids formation but if so, the solids formation occurs in the one section (freezing section) capable of handling solids that may form.

[0073] As the expanded fluids discharging from the JT valve at the inlet of the spray
25 nozzles cool the freezing section, the vapors rising from the freezing section become progressively colder over time during startup. The rising vapors are sent as a column overhead stream (e.g., the overhead stream **212**) to a heat exchanger (e.g., cross-exchanger or heat exchanger **400**) for cross-exchange with the compressed vapor (e.g., high-pressure vapor **416** or recycle **418**) discharged by the compressor through a cooler (e.g., cooler **404**). The
30 compressed vapor (e.g., startup stream **222**) exiting the cross-exchanger enters the JT valve (e.g., valve **230** of FIGS. 2 and 3) at the inlet of the spray nozzle. Consequently, during startup, the system vapors having initially high CO₂ change progressively over time to colder and colder temperatures, and to lower and lower CO₂ content in response to the compressed fluid going through the JT valve to the spray nozzles. As the startup progresses, the vapors rising from the

freezing section and discharging the column as an overhead stream (e.g., **212**) fall to within the normal range of 2 to 6% CO₂ content. Upon this occurrence, normal reflux generation can be initiated and reflux directed to the rectifier section or rectifier tower (e.g., the second column **302**). Liquids that accumulate at the bottom of the rectifier or second column **302** can serve as a normal source to the spray nozzle for sprays to the freezing section. At this point in time, the distillation column system can progressively be transitioned to normal spraying operations, and the supply of high pressure fluids (e.g., the compressed fluid or startup stream **222**) through the start-up JT valve to the spray nozzles discontinued.

[0074] At various times and positional stages of this startup, there is the need to cool vessels and piping. Cool process streams may be introduced to the vessels and piping to displace volumes which may contain warm fluids and/or residual high-CO₂-concentration vapors. Discharging these displaced fluids (warm and/or having high CO₂) of the cooling operations into the distillation process can have disruptive repercussions. Therefore, these displaced warmer fluids may instead be purged from the distillation process. However, purging of fluids on a continuous or intermittent basis from process pressure, e.g., 550 to 625 psia, to atmospheric pressure carries the potential of frequent plugging of the purging system. The temperatures and compositions downstream of the purge control valve can frequently be within the thermodynamic solidification regime and the solids are not always swept efficiently by the vapor or liquid being purged. Plugging of the purge system can prevent appropriate purging of the distillation column system and, therefore, prevent or hinder startup of the column system. Accordingly, there is a need for a reliable means and method of purging residual fluids that have undesirable compositions and warm temperatures during the startup process. To satisfy this need, embodiments of the present techniques provide an intermediate purge vessel or intermediate pressure vessel (e.g., the intermediate pressure vessel **102**).

[0075] Certain embodiments provide the intermediate pressure vessel as an intermediate purge drum at an intermediate pressure such that little or no solids are formed. If any solids are formed, the solids may be directed towards the bottom of the vessel where a heater may melt the solids. The contents of this intermediate pressure vessel can then in a more controlled fashion be discharged to atmosphere via the more conventional flaring system serving the overall facility, for example.

[0076] It should be noted that the triple point pressure of CO₂ is 75.1 psia, and that purges of high CO₂ containing fluids to conditions below this pressure and certainly to atmospheric pressure would normally result in solids formation. In contrast, purges to intermediate pressures, such as with the intermediate vessel **102**, have a dual benefit of (1) not incurring

much cooling due to the much reduced pressure drop and lesser associated JT effect, and (2) operating conditions remaining significantly away from solidification conditions. Thus, while the pressure in the intermediate pressure vessel **102** could be as low as 200 psig or 300 psig in some embodiments to preclude most or all CO₂ solidification, a desirable approach may be to operate the intermediate pressure vessel **102** at higher pressure for only sufficient pressure drop to promote reliable movement of the fluids, such as a pressure drop of about 50 psi or less from the purge point across the purge valve to the intermediate pressure vessel **102**.

[0077] Certain embodiments provide reliable techniques of cooling the reflux accumulator and reflux piping that do not come into play until later in the startup procedure. Fluids used for this cooling may be sent to a purge drum or the intermediate pressure vessel rather than disrupting the startup process by introducing warm fluid streams into an otherwise colder portion of the process. Embodiments may also provide reliable techniques of purging high CO₂ warm fluids from the bottoms of the rectifier and associated spray pump piping without fear of fouling or plugging with CO₂ solids. In other words, such purging may protect the suction of the spray pump or related piping when starting the spray pump and thus very cold liquids would otherwise come in contact with pockets of residual high CO₂ fluids.

[0078] FIG. 6 is a block diagram of an exemplary method **600** of operating a cryogenic distillation column system. The cryogenic distillation may purify natural gas or methane. The method includes feeding (block **602**) natural gas having methane and acid gas to a cryogenic distillation column having a freezing section. The acid gas includes carbon dioxide and other acid gases. The method includes discharging (block **604**) an overhead stream rich in the methane from the cryogenic distillation column to an overhead system. This overhead stream may be used to facilitate startup of the cryogenic distillation column system. Also, this overhead stream rich in methane may be ultimately sent as a product methane stream. The cryogenic distillation column may also discharge a bottoms stream rich in the acid gas for disposal or additional processing. The term “rich” for various streams may refer to an amount, by mole or weight, of at least about 30%, or at least about 50%, or at least about 90%, or at least about 94%, and so on, of a compound (e.g., methane, carbon dioxide) or class of compounds (e.g., acid gas) in a stream.

[0079] Further, the method includes discharging (block **606**), via a purge valve, a purge from the cryogenic distillation column system, such as from the overhead system or from the rectifying section or rectifying column, to an intermediate pressure vessel to reduce solidification of the carbon dioxide during startup of the cryogenic distillation column system. The purge may be a plurality of purges, and the purge valve a plurality of respective purge

valves. The purge material may discharge from the intermediate pressure vessel to a flare system or recovery system. The pressure of the intermediate pressure vessel may be controlled via a pressure control valve disposed on a conduit conveying the purge material from the intermediate pressure vessel to the flare system or the recovery system. In certain
5 embodiments, the pressure control valve is disposed at or near a flare knockout drum of the flare system. Furthermore, the method includes heating (block **608**) contents, e.g., purge material, in the intermediate pressure vessel via a heater. The heater may typically be disposed in the intermediate pressure vessel.

[0080] Additionally, the method may include providing (block **610**) a cooled vapor from
10 the overhead system through a Joule-Thompson (JT) expansion valve to the cryogenic distillation column during startup of the cryogenic distillation column system. The provision of this cooled vapor may be discontinued once startup is complete or nearing completion and normal operation is implemented.

[0081] FIG. 7 is a simplified process flow diagram of a cryogenic distillation column
15 system **116B** employed in the production and purification of natural gas and/or methane. The cryogenic distillation column system is configured to receive a feed gas having carbon dioxide content. The intermediate pressure vessel **102** may be used during startup of the column system **116B**. After startup and during normal operation, the intermediate pressure vessel **102** may continue to receive purges or vents, or may be placed out-of-service for normal operations.

[0082] As shown in Fig. 7, the column system **116B** may include a cryogenic column **702**
20 with an associated reboiler system **704** and an overhead system **214**. Further, while the column **702** is depicted as a single column or tower, the column **702** may be implemented as two columns or towers.

[0083] In the illustrated example, the column **702** may include three separate sections, such
25 as an upper distillation section or “rectification section” **708**, a middle distillation section or “freezing section” **710**, a lower distillation section or “stripping section” **712**, and a freezing section **710**. The freezing section **710** may include a spray nozzle bank **714**, a vapor zone **716**, and a melt tray **718**. In particular embodiments, the vapor zone **716** may accommodate vapor generated in the freezing section **710** and with the vapor labeled as CFZ™ vapor in certain
30 examples. Such vapor may travel up the column **702** and ultimately discharge as an overhead stream **734** from the column **702**.

[0084] A feed stream, such as a chilled two-phase fluid **720**, e.g., liquid/vapor phase fluid, may be introduced into the stripping section **712** where the two-phase fluid **720** may include a CO₂ concentration and a CH₄ concentration, among other heavier hydrocarbons and

contaminants. Within the stripping section **712**, the two-phase fluid **720** may be separated into a liquid and a vapor.

[0085] Within the stripping section **712**, the liquid of the two-phase fluid **720** may collect on the series of trays **722** (or packing) and flow into a bottom portion of the stripping section **712** to form a liquid pool **724**. The liquid may primarily include liquid CO₂, and dissolved H₂S, if present in the feed. The vapor of the two-phase fluid **720** may leave the stripping section **712** and proceed upward into risers in the melt tray **718**. As the vapor continues upward, it may enter the vapor zone **716**. The risers of the melt tray **718** may act as a vapor distributor for uniform distribution through the vapor zone **716**. As the vapor continues to rise upward into the vapor zone **716**, the vapor may contact a spray of cold liquid **726** emitted by the spray nozzle bank **714**. When in contact with the spray of cold liquid **726**, a portion of the CO₂ in the vapor may solidify within the freezing section **710**. This may act to remove or “freeze out” the CO₂ contaminant within the vapor. The solidified CO₂ **728** may fall onto the melt tray **718** where it may be heated to form liquid CO₂ **730** that may be collected on the melt tray **718**. The liquid CO₂ **730** may then flow into the stripping section **712** and into the liquid pool **724**, which primarily includes liquid CO₂. The liquid pool **724** and liquid CO₂ **730** may exit the stripping section **712** as a sour liquid **732**.

[0086] With contaminants removed, the vapor may continue to flow upward and into the rectification section **708**, along with other light gases. The vapor may include a sweet gas **734**, such as methane, CH₄, that may exit overhead of the column **702**. The sweet gas **734**, analogous to overhead stream **212** and the product **114** of previous figures, may be commercialized or used as an on-site fuel gas. It should be noted that a portion of the gas stream **734** may be used as stream **222**, i.e., a spray stream, during startup or subsequently condensed in the overhead system **214** to form a condensed or partially-condensed stream for reflux **216** during normal operation. The partially-condensed stream may enter a reflux accumulator where a reflux **216**, e.g., liquid condensate, may be collected and recycled back as reflux **754** into the column **702**. Liquids **752** collected at the bottom of the rectifier section may be the main source of liquids for the spray of cold liquid **726**. A portion of the reflux **216** stream may also be a source of liquids for the spray of cold liquid **726**.

[0087] Similarly, a portion of the sour liquid **732** exiting the stripping section **712** may be heated in a reboiler **744** and returned to the bottom of the stripping section **712** as a reboiler stream, e.g., a vapor stream **746**. The vapor stream **746** may provide energy to the bottom of the column **702** to boil off methane and other light components that may be dissolved in the sour liquid **732**. A residual reboiled liquid, e.g., a bottoms product or liquid acid gas **112**, may

exit the reboiler 744.

[0088] In summary, some embodiments may be directed to a cryogenic distillation column system with a cryogenic distillation column having a freezing section. The cryogenic distillation column may be a single column, or two columns in a split-tower design. The cryogenic distillation column is configured to receive a natural gas having methane and acid gas, and to discharge an overhead stream rich in the methane, e.g., have a majority mole percent of methane, and a bottoms stream rich in the acid gas e.g., have a majority mole percent of acid gas. The acid gas discharged as a bottom stream may be generally liquid acid gas. An overhead system is configured to receive the overhead stream, and to discharge a vapor methane product and a cooled liquid for reflux to the cryogenic distillation column. An intermediate pressure vessel is configured to receive one or more purges in the cryogenic distillation column system. The intermediate pressure vessel operates at an intermediate pressure to reduce solidification of the purges.

[0089] The respective purge point for each of the purges may have a purge valve to facilitate provision of the purge to the intermediate pressure vessel. The purges may be from the overhead system, from a bottom of a rectifying section, and/or from other points in the distillation system. Furthermore, a pressure control valve may maintain an operating pressure of the intermediate pressure vessel. In particular embodiments, the pressure control valve is configured to maintain the operating pressure of the intermediate pressure vessel to within at least 200 psig of a purge pressure of the purge. The pressure control valve may be configured to maintain the operating pressure of the intermediate pressure vessel to a pressure above the triple point pressure of carbon dioxide. The intermediate pressure vessel may be configured to discharge to a flare system. The intermediate pressure vessel may be configured to discharge a vapor stream to a flare knockout drum. Further, the pressure control valve may be disposed on the vapor stream at or near the flare knockout drum. Also, the intermediate pressure vessel may have a heater. Lastly, the cryogenic distillation column overhead system may be configured to discharge a cooled vapor through a Joule-Thompson (JT) valve to the cryogenic distillation column during startup prior to significant normal reflux being established.

[0090] Other embodiments relate to a method of operating a cryogenic distillation column system, the method including feeding natural gas having methane and acid gas including carbon dioxide to a cryogenic distillation column having a freezing section. The method includes discharging an overhead stream rich in the methane from the cryogenic distillation column to an overhead system. Further, the method includes discharging, via a purge valve, a purge to an intermediate pressure vessel to reduce solidification of the carbon dioxide during

startup of the cryogenic distillation column system. The purge may be a plurality of purges, and the purge valve comprises a plurality of respective purge valves. The method may include conveying vapor from the intermediate pressure vessel to a flare system. The method may include maintaining an operating pressure of the intermediate pressure vessel via a pressure control valve. In particular examples, the pressure control valve may be disposed at or near a flare knockout drum. Moreover, the method may include heating contents in the intermediate pressure vessel via a heater. The method may involve providing a cooled vapor from the overhead system through a Joule-Thompson (JT) expansion valve to the cryogenic distillation column during startup of the cryogenic distillation column system. Additionally, a bottoms stream rich in the acid gas may be discharged from the cryogenic distillation column. Also, a methane product stream may be discharged from the overhead system.

[0091] Yet other embodiments may relate to a method of operating a cryogenic distillation column system, the method including feeding natural gas having methane and acid gas to a cryogenic distillation column having a freezing section, the acid gas including carbon dioxide. The method includes discharging an overhead stream rich in the methane from the cryogenic distillation column to an overhead system. The method includes purging from the cryogenic distillation column system during startup of the cryogenic distillation column system to a flare system or recovery system to reduce solidification of carbon dioxide in the distillation column system. In other words, purging of the relevant fluids reduces undesirable solidification of carbon dioxide in the distillation system. Further, the purge is via an intermediate pressure vessel to reduce solidification of carbon dioxide in the purge. In other words, the method includes maintaining an operating pressure of the intermediate pressure vessel to reduce solidification of carbon dioxide in the purging. The pressure of the intermediate pressure vessel may be maintained via a pressure control valve or a backpressure control valve disposed on a conduit conveying the purge material from the intermediate pressure vessel to the flare system or the recovery system. The pressure control valve or the backpressure control valve is disposed at or near a flare knockout drum of the flare system. The method may include heating the purge material in the intermediate pressure vessel via a heater. Moreover, the method may include providing a cooled vapor from the overhead system through a Joule-Thompson (JT) expansion valve to the cryogenic distillation column during startup of the cryogenic distillation column system. Lastly, the method may include discharging a bottoms stream rich in the acid gas from the cryogenic distillation column, and discharging a methane product stream from the overhead system.

[0092] While the present techniques may be susceptible to various modifications and

alternative forms, the embodiments discussed above have been shown only by way of example. However, it should again be understood that the techniques are not intended to be limited to the particular embodiments disclosed herein. Indeed, the present techniques include all alternatives, modifications, and equivalents falling within the true spirit and scope of the
5 appended claims.

CLAIMS

What is claimed is:

1. A cryogenic distillation column system comprising:
a cryogenic distillation column configured to receive a natural gas comprising methane
5 and acid gas, and to discharge:
an overhead stream rich in the methane; and
a bottoms stream rich in the acid gas;
an overhead system configured to receive the overhead stream, and to discharge:
a vapor methane product;
10 a cooled liquid for reflux to the cryogenic distillation column; and
an intermediate pressure vessel configured to receive a purge and operate at an
intermediate pressure to reduce solidification of the purge.
2. The system of claim 1, wherein the overhead system is configured to discharge a cooled
15 vapor through a Joule-Thompson (JT) valve to the cryogenic distillation column during startup,
and wherein the intermediate pressure vessel is configured to receive the purge during startup
of the cryogenic distillation column system.
3. The system of claim 1 or 2, comprising a purge valve to provide the purge to the
20 intermediate pressure vessel, and wherein the intermediate pressure vessel comprises a heater.
4. The system of any one of claims 1 to 3, wherein the purge is from the overhead system.
5. The system of any one of claims 1 to 4, wherein the cryogenic distillation column
25 comprises two columns, and wherein the purge comprises a plurality of purges.
6. The system of any one of claims 1 to 5, wherein the cryogenic distillation column has
a freezing section, and wherein the intermediate pressure vessel is configured to discharge to a
flare system.
- 30 7. The system of any one of claims 1 to 6, comprising a pressure control valve to maintain
an operating pressure of the intermediate pressure vessel.
8. The system of claim 7, wherein the pressure control valve is configured to maintain the

operating pressure of the intermediate pressure vessel to within at least 200 psig of a purge pressure of the purge.

9. The system of claim 7, wherein the pressure control valve is configured to maintain the
5 operating pressure of the intermediate pressure vessel to a pressure above the triple point pressure of carbon dioxide.

10. The system of any one of claims 7 to 9, wherein the intermediate pressure vessel is
configured to discharge a vapor stream to a flare knockout drum, and wherein the pressure
10 control valve is disposed on the vapor stream at or near the flare knockout drum.

11. A method of operating a cryogenic distillation column system, the method comprising:
feeding natural gas comprising methane and acid gas to a cryogenic distillation column,
wherein the acid gas comprises carbon dioxide;
15 discharging an overhead stream rich in the methane from the cryogenic distillation column to an overhead system; and
discharging, via a purge valve, a purge to an intermediate pressure vessel to reduce solidification of the carbon dioxide during startup of the cryogenic distillation column system.

20 12. The method of claim 11, comprising maintaining an operating pressure of the intermediate pressure vessel via a pressure control valve.

13. The method of claim 12, wherein the pressure control valve is disposed at or near a
flare knockout drum.

25 14. The method of any one of claims 11 to 13, comprising conveying vapor from the intermediate pressure vessel to a flare system.

15. The method of any one of claims 11 to 14, comprising heating contents in the
30 intermediate pressure vessel via a heater.

16. The method of any one of claims 11 to 15, wherein the purge comprises a plurality of purges, and the purge valve comprises a plurality of respective purge valves.

17. The method of any one of claims 11 to 16, comprising providing a cooled vapor from the overhead system through a Joule-Thompson (JT) expansion valve to the cryogenic distillation column during startup of the cryogenic distillation column system, wherein the cryogenic distillation column has a freezing section.

5

18. The method of any one of claims 11 to 17, comprising discharging a bottoms stream rich in the acid gas from the cryogenic distillation column.

10

19. The method of any one of claims 11 to 18, comprising discharging a methane product stream from the overhead system.

20. A method of operating a cryogenic distillation column system, the method comprising: feeding natural gas comprising methane and acid gas to a cryogenic distillation column having a freezing section, the acid gas comprising carbon dioxide;

15

discharging an overhead stream rich in the methane from the cryogenic distillation column to an overhead system;

purging to an intermediate pressure vessel to reduce solidification of carbon dioxide in the cryogenic distillation column system during startup of the cryogenic distillation column system;

20

maintaining an operating pressure of the intermediate pressure vessel to reduce solidification of carbon dioxide in the purging; and

discharging purge material from the intermediate pressure vessel to a flare system or a recovery system.

25

21. The method of claim 20, wherein maintaining the operating of pressure of the intermediate pressure vessel comprises maintaining the operating pressure of the intermediate pressure vessel via a backpressure control valve disposed on a conduit conveying the purge material from the intermediate pressure vessel to the flare system or the recovery system.

30

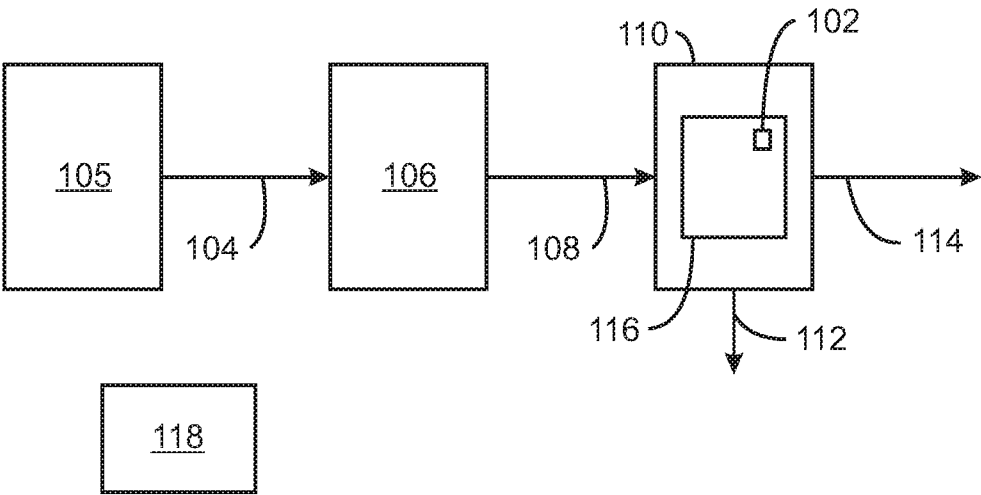
22. The method of claim 21, wherein the backpressure control valve is disposed at or near a flare knockout drum of the flare system.

23. The method of any one of claims 20 to 22, comprising heating the purge material in the intermediate pressure vessel via a heater.

24. The method of any one of claims 20 to 23, comprising providing a cooled vapor from the overhead system through a Joule-Thompson (JT) expansion valve to the cryogenic distillation column during startup of the cryogenic distillation column system.

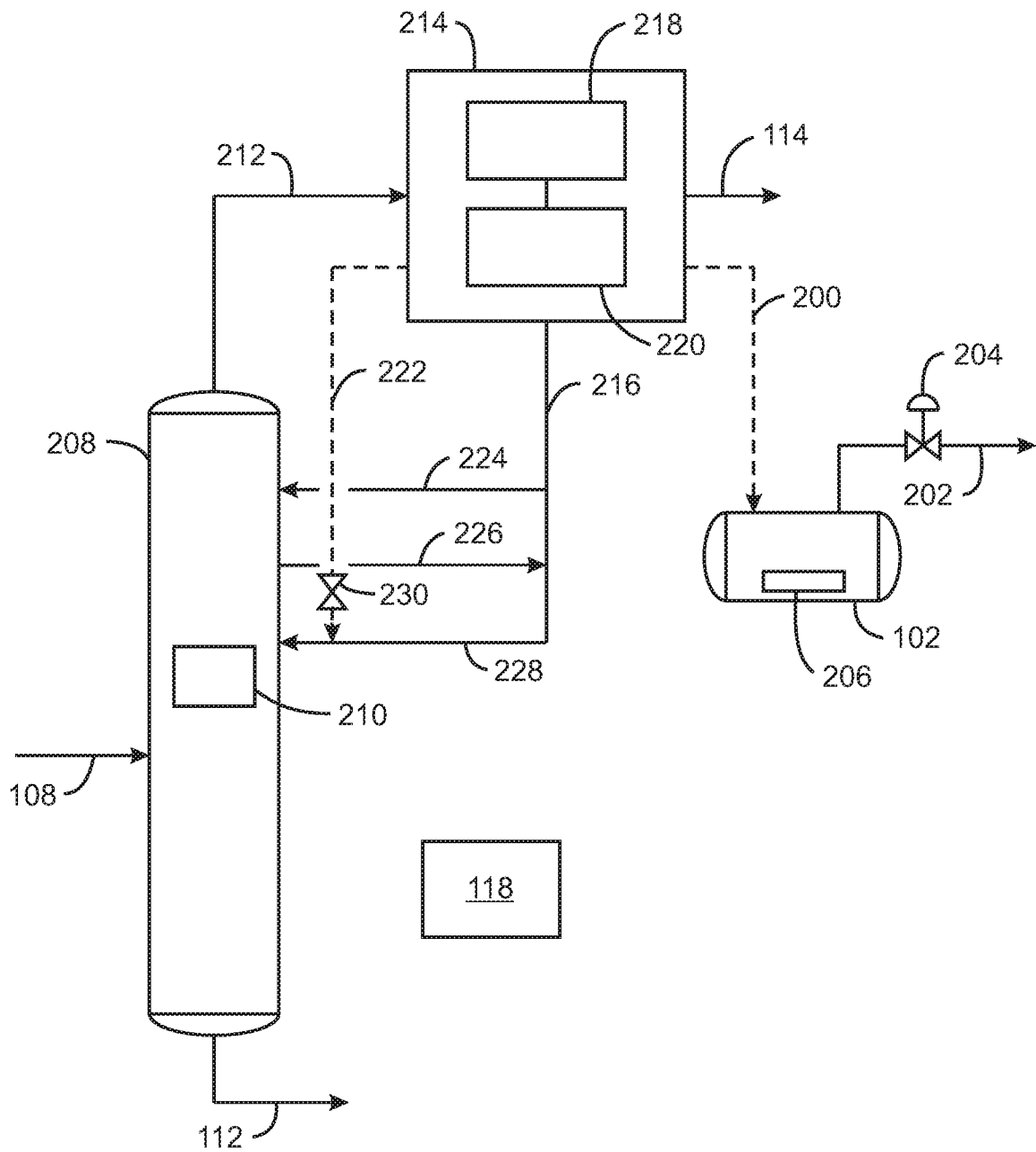
5

25. The method of any one of claims 20 to 24, comprising:
discharging a bottoms stream rich in the acid gas from the cryogenic distillation column; and
discharging a methane product stream from the overhead system.



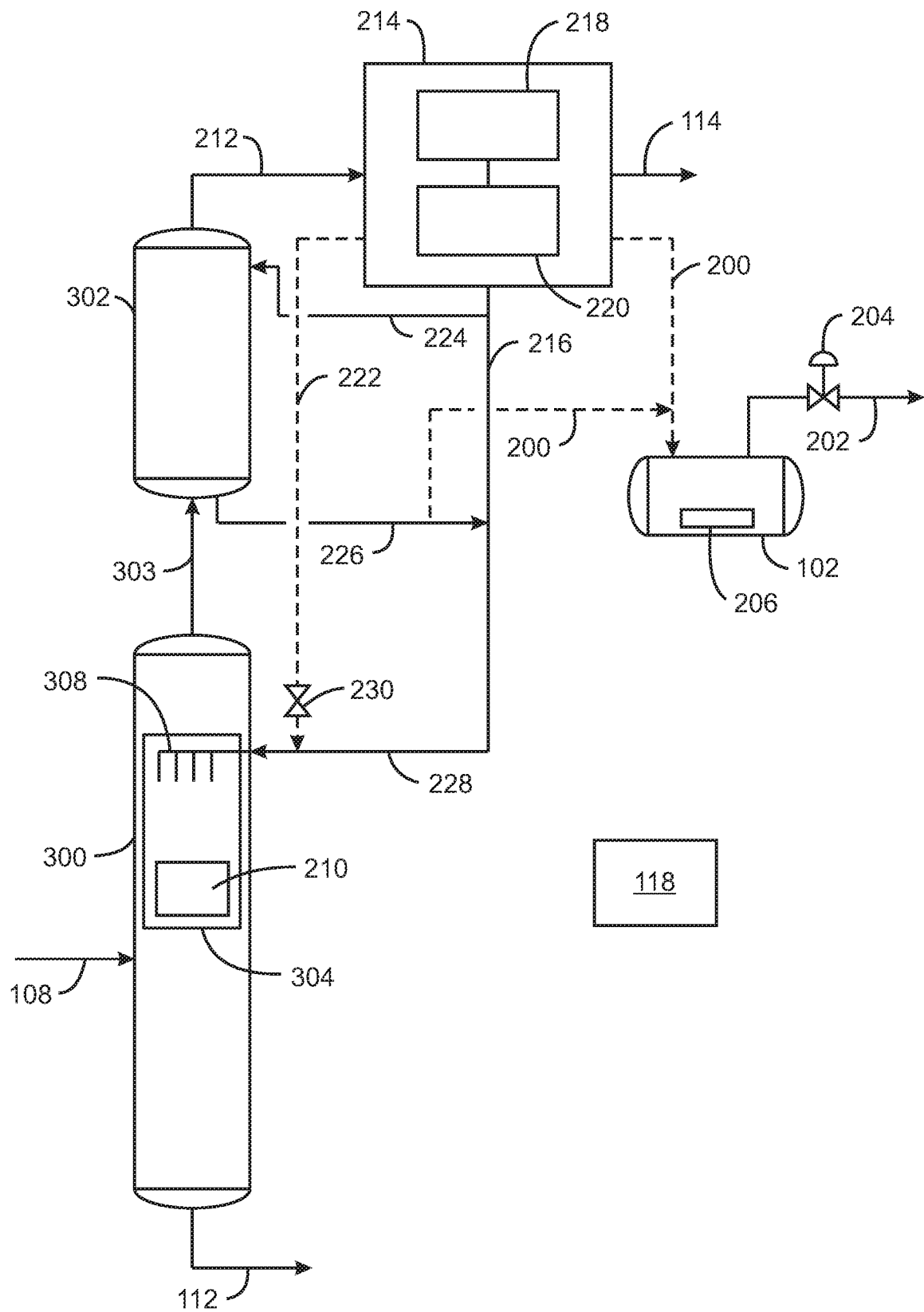
100
FIG. 1

2/7



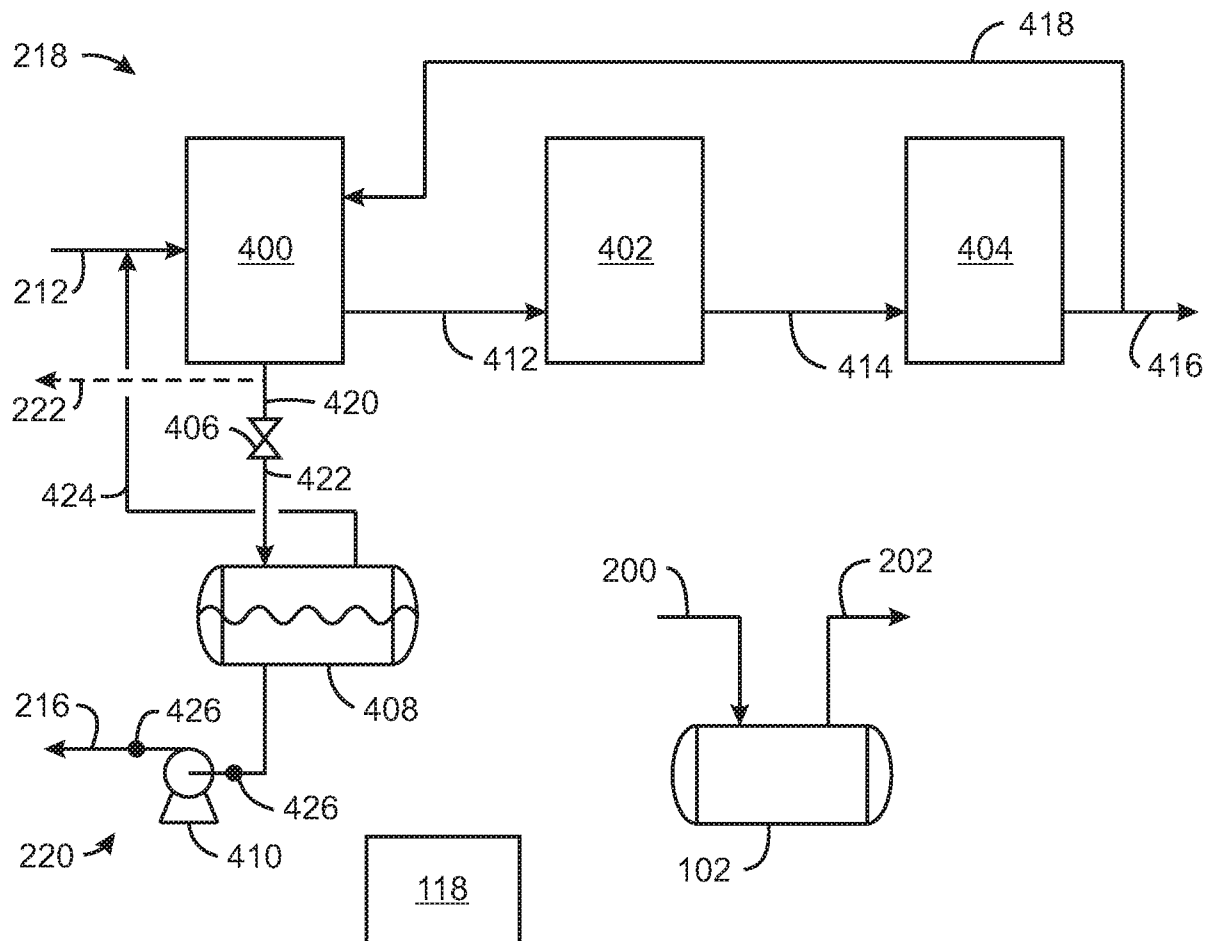
116
FIG. 2

3/7

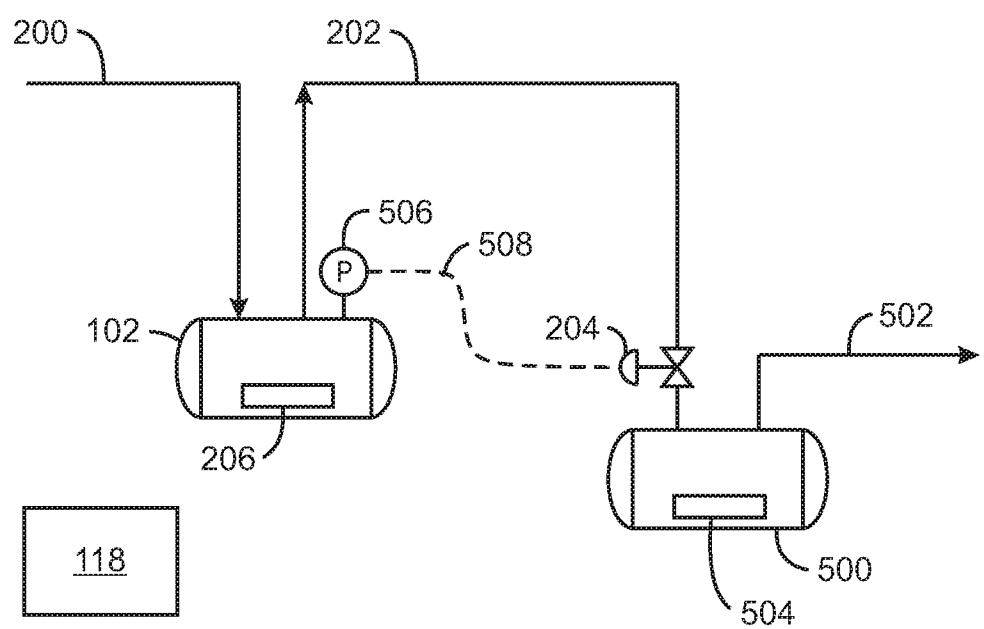


116A
FIG. 3

4/7

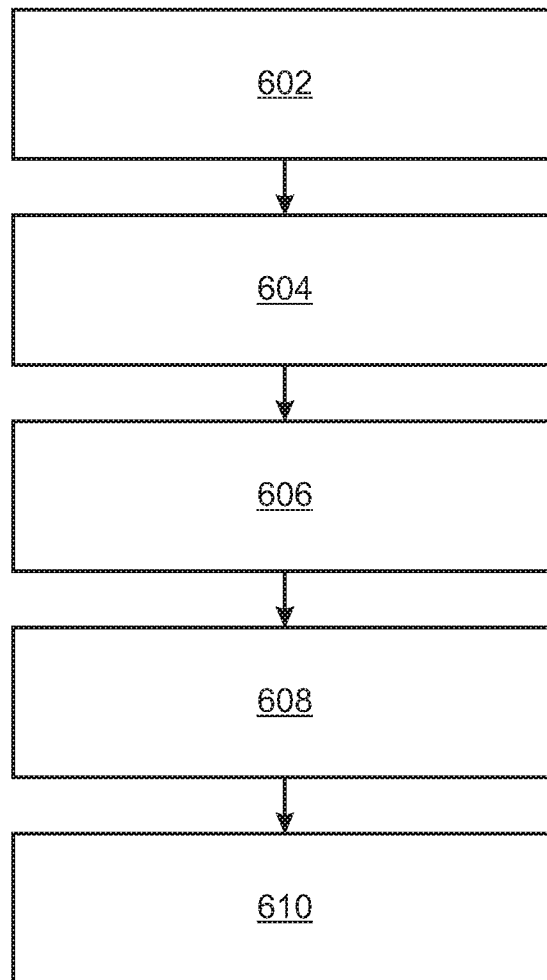


214
FIG. 4



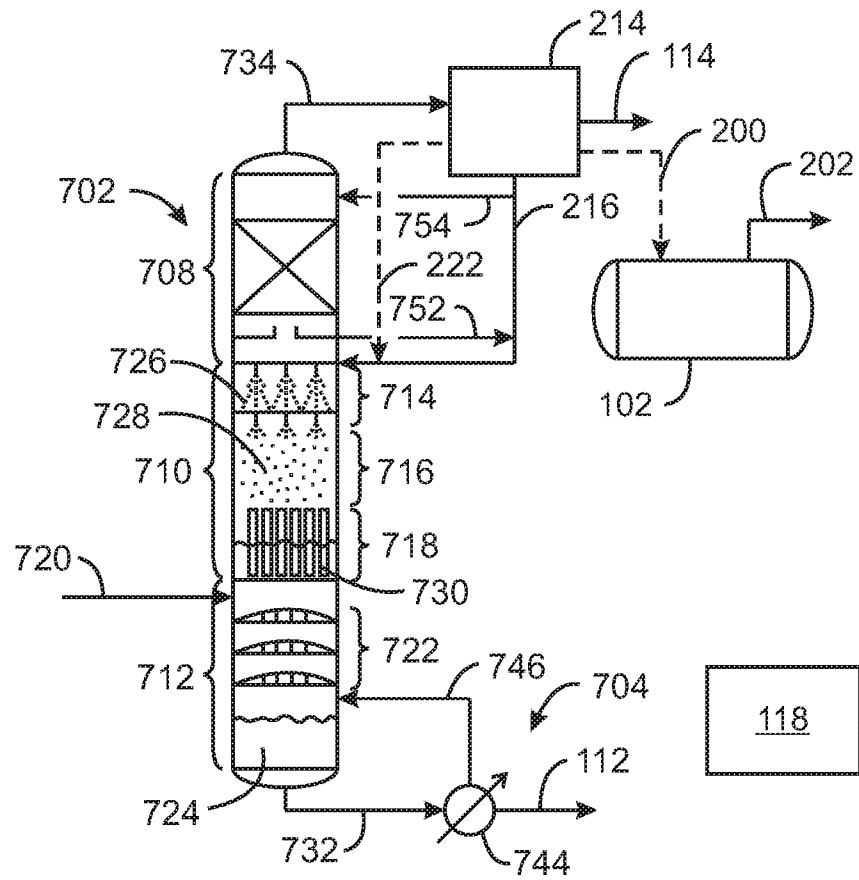
501
FIG. 5

6/7



600
FIG. 6

7/7



116B
FIG. 7

INTERNATIONAL SEARCH REPORT

International application No
PCT/US2016/032416

A. CLASSIFICATION OF SUBJECT MATTER
INV. F25J3/02
ADD.

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)
F25J

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practicable, search terms used)

EP0-Internal, WPI Data

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
Y	US 2009/266107 A1 (SINGH VIKRAM [US] ET AL) 29 October 2009 (2009-10-29) figure 1	1-25
Y	----- US 5 062 270 A (HAUT RICHARD C [NO] ET AL) 5 November 1991 (1991-11-05) figure 2	1-25
Y	----- WO 2013/171421 A2 (AIR LIQUIDE [FR]) 21 November 2013 (2013-11-21) page 1, line 18 - line 23; figure 1 page 2, line 6 - line 15 -----	1-25



Further documents are listed in the continuation of Box C.



See patent family annex.

* Special categories of cited documents :

"A" document defining the general state of the art which is not considered to be of particular relevance

"E" earlier application or patent but published on or after the international filing date

"L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)

"O" document referring to an oral disclosure, use, exhibition or other means

"P" document published prior to the international filing date but later than the priority date claimed

"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention

"X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone

"Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art

"&" document member of the same patent family

Date of the actual completion of the international search

10 August 2016

Date of mailing of the international search report

22/08/2016

Name and mailing address of the ISA/

European Patent Office, P.B. 5818 Patentlaan 2
NL - 2280 HV Rijswijk
Tel. (+31-70) 340-2040,
Fax: (+31-70) 340-3016

Authorized officer

Schopfer, Georg

INTERNATIONAL SEARCH REPORT

Information on patent family members

International application No

PCT/US2016/032416

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
US 2009266107 A1	29-10-2009	AU 2007345353 A1 CA 2674618 A1 GB 2458434 A US 2009266107 A1 WO 2008091317 A2	31-07-2008 31-07-2008 23-09-2009 29-10-2009 31-07-2008
US 5062270 A	05-11-1991	CA 2049236 A1 GB 2247518 A NO 913385 A US 5062270 A	01-03-1992 04-03-1992 02-03-1992 05-11-1991
WO 2013171421 A2	21-11-2013	FR 2990749 A1 WO 2013171421 A2	22-11-2013 21-11-2013