



TR), OAPI (BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG). **Published:**

Declarations under Rule 4.17:

- *as to applicant's entitlement to apply for and be granted a patent (Rule 4.17(ii))*
- *as to the applicant's entitlement to claim the priority of the earlier application (Rule 4.17(iii))*

- *with international search report (Art. 21(3))*
- *before the expiration of the time limit for amending the claims and to be republished in the event of receipt of amendments (Rule 48.2(h))*

METHOD AND APPARATUS FOR ESTIMATING A FLUID PROPERTY**Inventor(s): DIFOGGIO, Rocco & BLUE, D. Duncan III****BACKGROUND****1. Technical Field**

5 [0001] The present disclosure generally relates to well bore tools and in particular to apparatus and methods for detecting a substance in a downhole fluid.

2. Background Information

[0002] Drilling tools and wireline tools are used to reach and evaluate subterranean formations that produce oil and gas. These tools often incorporate various sensors,
10 instruments and control devices in order to carry out any number of downhole operations. The operations may include formation testing, fluid analysis, and tool monitoring and control.

[0003] Information about the subterranean formations traversed by the borehole may be obtained by any number of techniques. Techniques used to obtain formation
15 information include obtaining one or more downhole fluid samples produced from the subterranean formations. Downhole fluids, as used herein include any one or any combination of drilling fluids, return fluids, connate formation fluids, and formation fluids that may be contaminated by materials and fluids such as mud filtrates, drilling fluids and return fluids. Downhole fluid samples are often retrieved from the
20 borehole and tested in a rig-site or remote laboratory to determine properties of the fluid samples, which properties are used to estimate formation properties. Modern fluid sampling also includes various downhole tests to estimate fluid properties while the fluid sample is downhole.

[0004] Some formations produce hazardous fluids, such as hydrogen sulfide and other
25 gases that may damage tools, present safety hazards to surface personnel, and that may reduce the viability of the formation for producing useful hydrocarbons. Surface testing for these fluids requires bringing the fluid to the surface. For H₂S gas measurement, unless one is careful in the selection of sample tank material, retrieval of a sample to the surface runs the risk of under-reporting the actual H₂S levels. The
30 reason is that H₂S chemically reacts with many materials. The unreacted, remaining amount of H₂S in the sample that is finally measured at the surface may be

significantly less than the total amount of H₂S that had been in the original sample so that the H₂S concentration gets under reported. That is another reason that an in-situ measurement is preferable. However, an in-situ measurement of the chemical composition of these fluids presents unique problems due to the rigorous downhole environment.

[0005] The environment in a well presents many challenges to maintain the tools used at depth due to vibration, harsh chemicals and temperature. Temperature in downhole tool applications presents a unique problem to these tools. High downhole temperatures may reach as high as 392° F (200° C) or more making it difficult to operate sensitive electronic components in the environment. Space in a downhole carrier is usually limited to a few inches in diameter. Cooling systems typically utilize large amounts of power and take up valuable space in the tool carrier and add an additional failure point in the system.

SUMMARY

[0006] The following presents a general summary of several aspects of the disclosure in order to provide a basic understanding of at least some aspects of the disclosure. This summary is not an extensive overview of the disclosure. It is not intended to identify key or critical elements of the disclosure or to delineate the scope of the claims. The following summary merely presents some concepts of the disclosure in a general form as a prelude to the more detailed description that follows.

[0007] Disclosed is an apparatus for estimating a fluid property. The apparatus may include an optic member having a material within the optic member that is responsive to the fluid property. A detector may be operably associated with the optic member that detects a change in the material, the change being indicative of the fluid property.

[0008] In another aspect, a method for estimating a fluid property includes introducing an optic member to a fluid. The optic member includes a material within the optic member that is responsive to the fluid property. The method may further include detecting a change in the material, the change being indicative of the fluid property.

BRIEF DESCRIPTION OF THE DRAWINGS

[0009] For a detailed understanding of the present disclosure, reference should be made to the following detailed description of the several non-limiting embodiments, taken in conjunction with the accompanying drawings, in which like elements have
5 been given like numerals and wherein:

FIG. 1 schematically illustrates a non-limiting example of a drilling system in a measurement-while-drilling (“MWD”) arrangement according to several non-limiting embodiments of the disclosure;

FIG. 2 illustrates non-limiting example of a gas detection tool according to
10 one or more embodiments of the disclosure;

FIG. 3 illustrates a filled fiber detector according to one or more embodiments of the disclosure; and

FIG. 4 illustrates a non-limiting method according to one or more embodiments of the disclosure.

15

DESCRIPTION OF EXEMPLARY EMBODIMENTS

[0010] The present disclosure uses terms, the meaning of which terms will aid in providing an understanding of the discussion herein. As used herein, high temperature refers to a range of temperatures typically experienced in oil production well boreholes. For the purposes of the present disclosure, high temperature and
20 downhole temperature include a range of temperatures from about 100° C to about 200° C (about 212° F to about 392° F). In recent years, as wells have gotten deeper, a few wells now exceed 200° C. One or more embodiments disclosed herein may use the term carrier. The term “carrier” as used herein means any device, device component, combination of devices, media and/or member that may be used to
25 convey, house, support or otherwise facilitate the use of another device, device component, combination of devices, media and/or member. Exemplary non-limiting carriers include drill strings of the coiled tube type, of the jointed pipe type and any combination or portion thereof. Other carrier examples include casing pipes, wirelines, wireline sondes, slickline sondes, drop shots, downhole subs, bottom hole

assemblies (BHA's), drill string inserts, modules, internal housings and substrate portions thereof.

[0011] **FIG. 1** schematically illustrates a non-limiting example of a drilling system **100** in a measurement-while-drilling ("MWD") arrangement according to several non-limiting embodiments of the disclosure. A derrick **102** supports a drill string **104** operating in this example as a carrier, which may be a coiled tube or drill pipe. The drill string **104** may carry a bottom hole assembly ("BHA") referred to as a downhole sub **106** and a drill bit **108** at a distal end of the drill string **104** for drilling a borehole **110** through earth formations.

[0012] Drilling operations according to several embodiments may include pumping drilling fluid or "mud" from a mud pit **112**, and using a circulation system **114**, circulating the mud through an inner bore of the drill string **104**. The mud exits the drill string **104** at the drill bit **108** and returns to the surface through an annular space between the drill string **104** and inner wall of the borehole **110**. The drilling fluid is designed to provide a hydrostatic pressure that is greater than the formation pressure to avoid blowouts. The pressurized drilling fluid may further be used to drive a drilling motor **116** and may be used to provide lubrication to various elements of the drill string **104**.

[0013] In the non-limiting embodiment of **FIG. 1**, the downhole sub **106** includes a formation evaluation tool **118**. The formation evaluation tool **118** may include an assembly of several tool segments that are joined end-to-end by threaded sleeves or mutual compression unions **120**. An assembly of tool segments suitable for the present disclosure may include a power unit **122** that may include one or more of a hydraulic power unit, an electrical power unit and an electro-mechanical power unit. In the example shown, a formation sample tool **124** may be coupled to the formation evaluation tool **118** below the power unit **122**.

[0014] The exemplary formation sample tool **124** shown comprises an extendable probe **126** that may be opposed by bore wall feet **128**. The extendable probe **126**, the opposing feet **128**, or both may be hydraulically and/or electro-mechanically extendable to firmly engage the well borehole wall. The formation sample tool **124** may be configured for extracting a formation core sample, a formation fluid sample,

formation images, nuclear information, electromagnetic information, and/or downhole information, such as pressure, temperature, location, movement, and other information. In several non-limiting embodiments, other formation sample tools not shown may be included in addition to the formation sample tool **124** without
5 departing from the scope of the disclosure.

[0015] Continuing now with **FIG. 1**, several non-limiting embodiments may be configured with the formation sample tool **124** operable as a downhole fluid sampling tool. In these embodiments, a large displacement volume motor/pump unit **130** may be provided below the formation sample tool **124** for line purging. A similar
10 motor/pump unit **132** having a smaller displacement volume may be included in the tool in a suitable location, such as below the large volume pump, for quantitatively monitoring fluid received by the downhole evaluation tool **118** via the formation sample tool **124**. As noted above, the formation sample tool **124** may be configured for any number of formation sampling operations. Construction and operational
15 details of a suitable non-limiting fluid sample tool **124** for extracting fluids are more described by U.S. Patent No. 5,303,775, the specification of which is incorporated herein by reference.

[0016] The downhole evaluation tool **118** may include a downhole evaluation system **134** for evaluating several aspects of the downhole sub **106**, of the drilling system
20 **100**, of the downhole fluid in and/or around the downhole sub **106**, formation samples received by the downhole sub **106**, and of the surrounding formation.

[0017] One or more formation sample containers **136** may be included for retaining formation samples received by the downhole sub **106**. In several examples, the formation sample containers **136** may be individually or collectively detachable from
25 the downhole evaluation tool **118**.

[0018] A downhole transceiver **146** may be coupled to the downhole sub **106** for bidirectional communication with a surface transceiver **140**. The surface transceiver **140** communicates received information to a controller **138** that includes a memory **142** for storing information and a processor **144** for processing the information. The
30 memory **142** may also have stored thereon programmed instructions that when executed by the processor **144** carry out one or more operations and methods that will

become apparent in view of the discussion to follow. The memory **142** and processor **144** may be located downhole on the downhole sub **106** in several non-limiting embodiments.

[0019] The system **100** shown in **FIG. 1** is only an example of how various tools may be carried into a well borehole using the downhole sub **106**. Tools according to the present disclosure may further include direct measurement tools for evaluating fluid characteristics such as content and concentrations. In one or more embodiments, the downhole sub **106** may be used to carry a downhole gas detector for evaluating fluids in-situ. The following discussion and associated figures will present several exemplary downhole gas detectors according to the disclosure.

[0020] **FIG. 2** illustrates a non-limiting example of a downhole fluid test tool **200** that may be incorporated into the downhole sub **106** as part of the downhole evaluation system **134** described above and shown in **FIG. 1**. As used herein, downhole fluid means any fluid that is carried to, carried in, encountered in, or carried from a downhole environment. A downhole fluid may include drilling fluid, return fluid, formation fluid or any combination thereof. The downhole fluid test tool **200** includes a fluid cell **202**. In the non-limiting embodiment of **FIG. 2**, an optic member **204** is shown immersed in the fluid **206**. The fluid cell **202** may include a septum **208** as an interface with the fluid cell **202** and the optic member **204**. The optic member may be introduced to the fluid via a feeder **210**, and a receiver **212** may be disposed to receive the optic member **204**.

[0021] One or more embodiments may further include a controller **214** coupled to an electromagnetic energy light source **216** and the electromagnetic energy source **216** may be coupled to the feeder **210** as shown, to the optic member **204**, or to both. In one or more embodiments, the receiver **212** may be coupled to a photo detector **218**. The photodetector **218**, depending on the detector output signal characteristics, may be connected to an analog-to-digital converter (ADC) **220** and the ADC **220** may be connected to a controller **222**. In one or more embodiments, the light source controller **214** and the controller **222** connected to the ADC **220** or photodetector **218** may be implemented as a single controller without departing from the scope of the disclosure.

[0022] The controller or controllers **214**, **222** may be implemented as downhole controllers or as surface controllers substantially as described above and shown in **FIG. 1** as surface controller **138**. The controller **214**, **222** may include a memory **142** as shown in **FIG. 1** containing programs for carrying out instructions for methods to
5 be described in more detail below.

[0023] The electromagnetic energy source **216** may be any suitable source for generating one or more wavelengths of electromagnetic energy. The generated energy may include broadband, narrow band, a single wavelength, or combinations thereof. In one or more embodiments, the electromagnetic energy source may include
10 a broadband source such as a tungsten bulb that produces substantially white light. The filament temperature of a tungsten bulb is approximately 2700 C so its operation is substantially unaffected at wellbore temperatures of 200 C. In one or more embodiments, the electromagnetic energy source may include an LED, a laser diode, or a combination thereof. The laser diodes and LEDs, when used may be provided
15 cooling to avoid dimming at elevated temperatures. In one or more embodiments, the electromagnetic energy source may include a plurality of sources arranged in an array.

[0024] The feeder **210** may include any suitable mechanism for introducing the optical member **204** to the fluid **206**. In one or more embodiments, the feeder **210** may include a probe with the optical member **204** being disposed on the probe. The
20 optical member, such as a polymer fiber optic, being insertable into the fluid cell **202** through a hole in an elastomeric, self-sealing septum **208**. Those skilled in the art with the benefit of the present disclosure will appreciate that the fluid need not be in a tool-carried fluid cell. In one or more embodiments, the fluid may be in the wellbore annulus and the optical member may be extended from the tool and into the annulus
25 fluid. In one or more embodiments, an optical member **204** may be inserted into a formation. It should also be recognized that the feeder **210** and optical member **204** need not penetrate through a fluid cell **202** as depicted in the non-limiting example of **FIG. 2**. The optical member **204** may be arranged as a loop with the photodetector **218**, ADC **220** and controller **222** being arranged co-located with the feeder **210**. The
30 feeder **210**, the receiver **212**, or each of the feeder and receiver may include a mechanism for disposing of used optical members and for dispensing new optical members for multiple test capability.

[0025] Referring now to **FIGS. 2** and **3**, the optical member **204** in several embodiments may include a fiber **300** having a colorimetric filler **302** disposed within the fiber. In one or more embodiments, the fiber **300** may be manufactured using a high-refractive-index, high-temperature polymer such as the transparent sulfone polymers, such as a polysulfone, a polyethersulfone, or a combination thereof. Each of these materials may be used in high-temperatures as may be experienced in the downhole environment. For example, polysulfones are capable of use in temperatures at least up to 174 C and polyphenylsulfones are usable to about 204 C. In one or more embodiments, the fiber 300 may be a clear

10 [0026] The colorimetric filler **302** may be any filler responsive to interaction with a gas **304** penetrating the fiber **300**. In one or more embodiments, the filler may include a small amount (1%- 2% by weight) of a compound that contains lead, iron, or another substance that changes color upon exposure to a selected gas **304** such as H₂S. In one or more embodiments, lead sulfate, which melts at the very high temperature of 1170 C, may be used as the filler **302**. In other embodiments, an iron compound may be used as the colorimetric filler **302**. For example, hydrogen sulfide reacts with ferric oxide, Fe₂O₃, which is rust colored, to form ferrous sulfide, FeS₂, which is black. Elemental iron particles might also be used as the colorimetric filler. In other non-limiting examples, high temperature organic compounds that change color upon exposure to a selected gas such as H₂S may be used as the filler **302**. Similarly, for SO₂ detection, the color-changing filler could be an alkali vanadate such as sodium vanadate, which doesn't melt until 858 C or potassium dichromate, which doesn't melt until 398 C. Similarly, other gases could be detected using other color-changing fillers.

25 [0027] Operational examples may now be described in view of the several embodiments described above and shown in **FIGS. 1-3**. Referring to **FIG. 4**, a method **400** for estimating a fluid property may include introducing an optic member to a fluid **402**. A material change may then be detected **204**, and the fluid property may be estimated **406** using the detected material change. In one or more
30 embodiments, the estimated property may be quantitative, qualitative, or a combination thereof.

[0028] In one or more embodiments, the optical member **204** may be implemented as a simple H₂S sensor that can be directly immersed into a high-temperature, high-pressure fluid downhole and produce to produce real-time H₂S concentration measurements. An advantage is that the H₂S diffuses out of the downhole fluid into a filled polymer thus accomplishing both a separation step through diffusion into the bulk polymer and the detection step through attenuation of light **306** transmitted because of discoloration of the filler as shown in **FIG. 3**. Dissolved gases **304**, such as H₂S, diffuse into the polymer, react with the filler, and discolor or darken from the outer perimeter of the member **204** toward the interior as shown in **FIG. 3**. The rate of discoloring or darkening of the polymer would be indicative of the H₂S concentration in the downhole fluid.

[0029] The polymer may be selected such that the polymer can withstand borehole temperatures up to about 200 C without melting and should be optically transparent. Selection should also take into consideration that the polymer should have a refractive index that is greater than the refractive index of crude oil (typical range of 1.40 – 1.55) and the refractive index of brine (typical range of 1.30 – 1.38). A high-refractive-index, high-temperature, transparent polymer such as the sulfone polymers could be used. For example, Udel[®] is a polysulfone usable to 174 C, Radel[®] A and Radel[®] R are polyethersulfones usable to 204 C. These exemplary polysulfones are available from Solvay Advanced Polymers 4500 McGinnis Ferry Road, Alpharetta, GA 30005-3914 USA. In one or more embodiments, the sulfone polymer may be loaded with a small amount of H₂S-indicating filler and drawn into a fiber. A light source **216** may be attached to one end of the fiber and a light detector **218** may be attached to the other end of the fiber. The middle section of fiber **204** may be immersed in the downhole fluid with appropriate fluid-tight connectors **208** at both ends. As noted above, one example of a suitable connector may include a self-sealing septum device. Light will propagate along the optic member **204** when a higher index of refraction (RI) exists within the fiber than in the medium surrounding the fiber. This condition is met by the exemplary sulfones mentioned above where Udel[®] RI = 1.650 at 486 nm, Radel[®] A RI = 1.671 at 486 nm, or Radel[®] R RI = 1.696 at 486 nm is immersed in crude oil, brine, or natural gas because the RI of crude oil is generally between 1.40 and 1.55, the RI of brine is generally between 1.30 and 1.38, and the RI of natural gas is generally between 1.1 and 1.3 depending on pressure and

temperature. By observing rate of change of the attenuation of light transmitted at one or more wavelengths through the fiber and accounting for the temperature and/or pressure dependence of the H₂S diffusion rate into the polymer, we can estimate the H₂S concentration of the downhole fluid.

- 5 **[0030]** The color-changing filler selected should be stable and it should not decompose at borehole temperatures. In one or more embodiments, the selected filler provides a significant color change upon exposure to expected concentrations of the gas in the downhole fluids. Using a lead sulfate filler provides for a very high melting temperature of 1170 C. Alternatively, an iron compound might be used as the
- 10 colorimetric filler. For example, hydrogen sulfide reacts with ferric oxide, Fe₂O₃, which is rust colored, to form ferrous sulfide, FeS₂, which is black. Elemental iron particles might also be used as the colorimetric filler. In other embodiments a high temperature organic compounds that change color upon exposure to H₂S may be used.
- 15 **[0031]** The present disclosure is to be taken as illustrative rather than as limiting the scope or nature of the claims below. Numerous modifications and variations will become apparent to those skilled in the art after studying the disclosure, including use of equivalent functional and/or structural substitutes for elements described herein, use of equivalent functional couplings for couplings described herein, and/or use of
- 20 equivalent functional actions for actions described herein. Such insubstantial variations are to be considered within the scope of the claims below.

- [0032]** Given the above disclosure of general concepts and specific embodiments, the scope of protection is defined by the claims appended hereto. The issued claims are not to be taken as limiting Applicant's right to claim disclosed, but not yet literally
- 25 claimed subject matter by way of one or more further applications including those filed pursuant to the laws of the United States and/or international treaty.

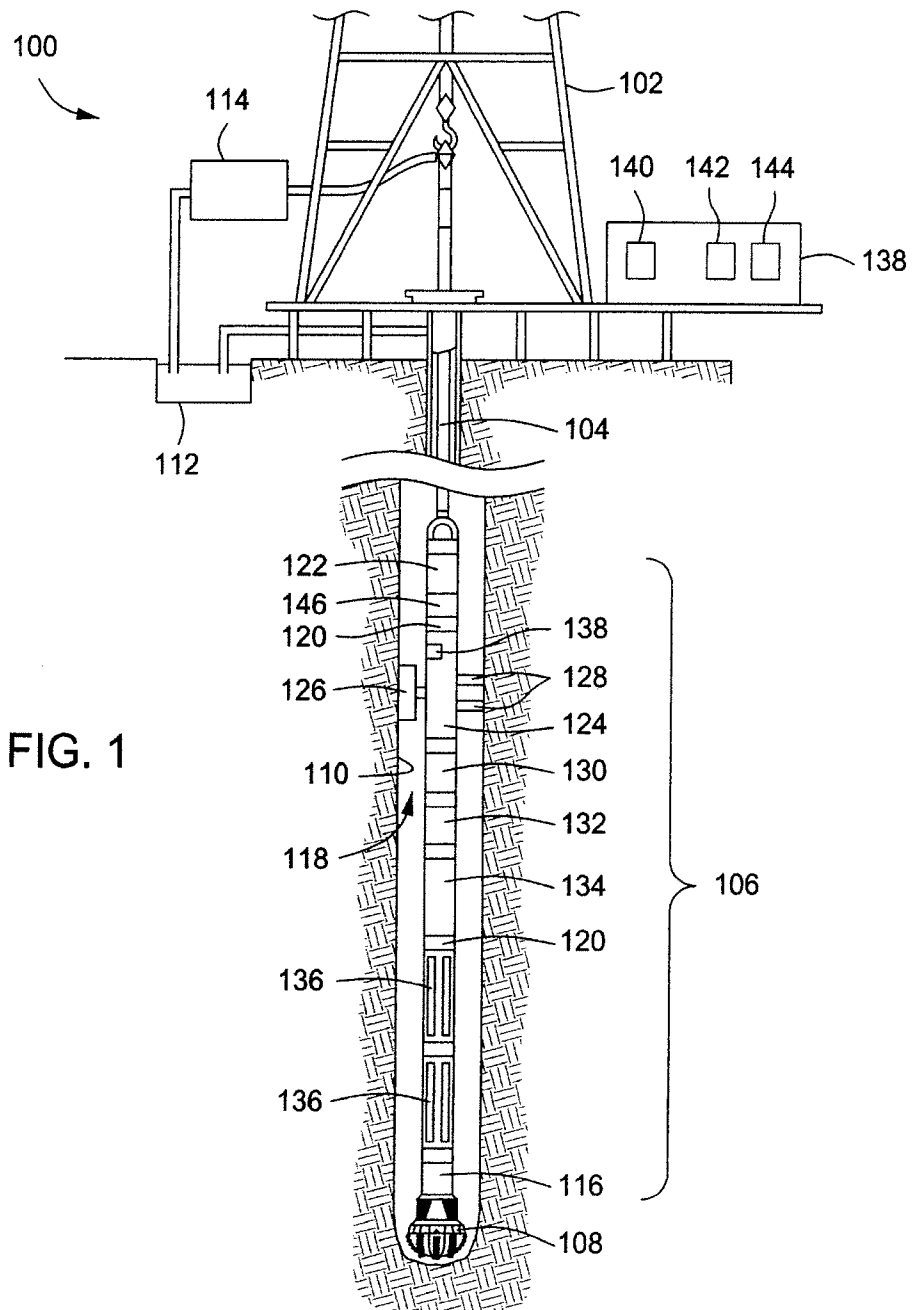
CLAIMS

What is claimed is:

- 1 1. An apparatus for estimating a fluid property comprising:
 - 2 (a) an optic member;
 - 3 (b) a material within the optic member responsive to the fluid property;
 - 4 and
 - 5 (c) a detector operably associated with the optic member that detects a
 - 6 change in the material, the change being indicative of the fluid
 - 7 property.
- 1 2. An apparatus according to claim 1, wherein the optic member comprises a
- 2 polymer fiber.
- 1 3. An apparatus according to claim 2, wherein the polymer fiber comprises one
- 2 or more sulfone polymers.
- 1 4. An apparatus according to claim 2, wherein the polymer fiber comprises one
- 2 or more of a polysulfone, a polyethersulfone, or a combination thereof.
- 1 5. An apparatus according to claim 1, wherein the material includes one or more
- 2 of lead, iron, lead sulfate, an iron compound, ferric oxide, elemental iron
- 3 particles, an organic compound, or combinations thereof.
- 1 6. An apparatus according to claim 1, wherein the fluid property includes the
- 2 presence of a gas.
- 1 7. An apparatus according to claim 6, wherein the gas includes a sulfide.
- 1 8. An apparatus according to claim 6, wherein the gas includes hydrogen sulfide.
- 1 9. An apparatus according to claim 1, further comprising a feeder that moves the
- 2 optical member toward the fluid.
- 1 10. An apparatus according to claim 1, further comprising a device for replacing
- 2 the optical member after use.

- 1 11. A method for estimating a fluid property comprising:
2 (a) introducing an optic member to a fluid, the optic member including a
3 material within the optic member responsive to the fluid property; and
4 (b) detecting a change in the material, the change being indicative of the
5 fluid property.
- 1 12. A method according to claim 11, wherein the optic member comprises a
2 polymer fiber.
- 1 13. A method according to claim 11, wherein the polymer fiber comprises one or
2 more sulfone polymers.
- 1 14. A method according to claim 12, wherein the polymer fiber comprises one or
2 more of a polysulfone, a polyethersulfone, or a combination thereof.
- 1 15. A method according to claim 11, wherein the material includes one or more of
2 lead, iron, lead sulfate, an iron compound, ferric oxide, elemental iron
3 particles, an organic compound, or combinations thereof.
- 1 16. A method according to claim 11, further comprising introducing
2 electromagnetic energy to the optic member.
- 1 17. A method according to claim 16, wherein detecting a change in the material
2 includes detecting an attenuation of the electromagnetic energy.
- 1 18. A method according to claim 11, further comprising replacing the optical
2 member after use.
- 1 19. A method according to claim 11, wherein the fluid property includes the
2 presence of a sulfide.
- 1 20. A method according to claim 11, wherein the fluid property includes the
2 presence of hydrogen sulfide.

1/3



2/3

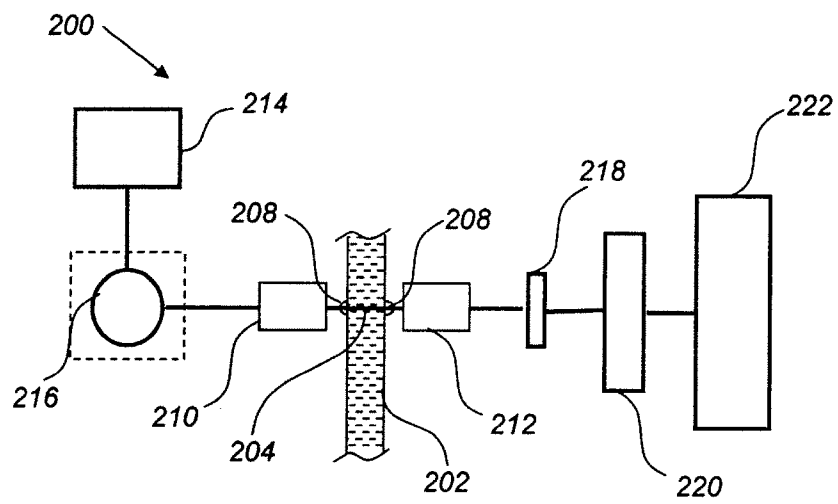


FIG 2

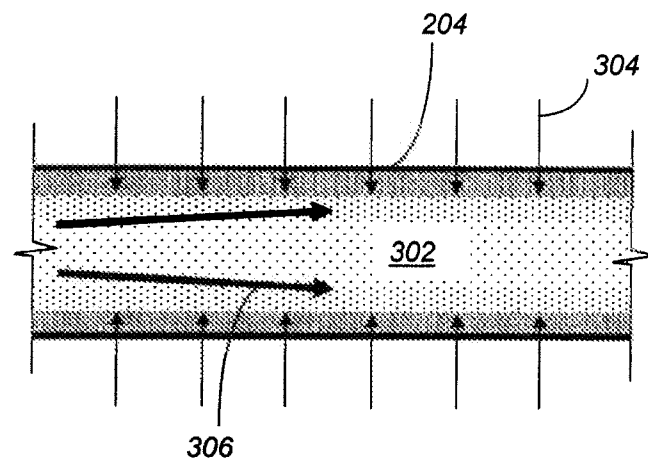


FIG 3

INTERNATIONAL SEARCH REPORT

International application No.
PCT/US2009/060753**A. CLASSIFICATION OF SUBJECT MATTER*****E21B 49/08(2006.01)i, G01N 21/17(2006.01)i***

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

E21B 49/08; E21B 47/01; E21B 49/10; G01N 21/17; G01N 21/61

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Korean utility models and applications for utility models

Japanese utility models and applications for utility models

(Chinese Patents and application for patent)

Electronic data base consulted during the international search (name of data base and, where practicable, search terms used)

eKOMPASS(KIPO internal) & Keywords: optic member, material, detector

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X Y	US 6328932 B1 (CARTER et al.) 11 December 2001 See Abstract, Figs. 1,2, Column 5, Claims 1,27	1,5-8,10-11,15-20 2-4,9,12-14
Y	US 5926269 A (VON DER ELTZ et al.) 20 July 1999 See Abstract, Fig. 1, Column 3, Claim 1	2-4,9,12-14
A	US 7025138 B2 (KURKJIAN et al.) 11 April 2006 See Abstract, Fig. 12, Claim 23	1-20
A	US 6947138 B2 (ARNO) 20 September 2005 See Abstract, Fig. 1	1-20

☐ Further documents are listed in the continuation of Box C.☒ See patent family annex.

* Special categories of cited documents:

"A" document defining the general state of the art which is not considered to be of particular relevance

"E" earlier application or patent but published on or after the international filing date

"L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of citation or other special reason (as specified)

"O" document referring to an oral disclosure, use, exhibition or other means

"P" document published prior to the international filing date but later than the priority date claimed

"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention

"X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone

"Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art

"&" document member of the same patent family

Date of the actual completion of the international search

26 MAY 2010 (26.05.2010)

Date of mailing of the international search report

28 MAY 2010 (28.05.2010)

Name and mailing address of the ISA/KR

Korean Intellectual Property Office
Government Complex-Daejeon, 139 Seonsa-ro, Seo-
gu, Daejeon 302-701, Republic of Korea

Facsimile No. 82-42-472-7140

Authorized officer

KOO, BON CHEOL

Telephone No. 82-42-481-5803



INTERNATIONAL SEARCH REPORT

Information on patent family members

International application No.

PCT/US2009/060753

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
US 6328932 B1	11.12.2001	None	
US 5926269 A	20.07.1999	EP 0832426 A1 EP 0832426 A1 JP 11-508038 A WO 97-00437 A1	16.07.2003 01.04.1998 13.07.1999 03.01.1997
US 7025138 B2	11.04.2006	GB 0129405 D0 GB 2371621 A GB 2371621 B US 2002-121370 A1	30.01.2002 31.07.2002 25.06.2003 05.09.2002
US 6947138 B2	20.09.2005	US 2004-0252307 A1 US 2006-0012794 A1 WO 2004-113881 A1	16.12.2004 19.01.2006 29.12.2004