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12/974,495 21 December 2010 (21.12.2010) US(71) Applicant (for all designated States except US): **APPLIED MATERIALS, INC.** [US/US]; 3050 Bowers Avenue, Santa Clara, California 95054 (US).

(72) Inventors; and

(75) Inventors/Applicants (for US only): **LIANG, Jingmei** [CN/US]; 1256 Mountain Quail Circle, San Jose, California 95120 (US). **INGLE, Nitin, K.** [US/US]; 1435 Graywood Drive, San Jose, California 95129 (US).(74) Agents: **BERNARD, Eugene, J.** et al.; TOWNSEND AND TOWNSEND AND CREW LLP, 1400 Wewatta Street Suite 600, Denver, CO 80202 (US).

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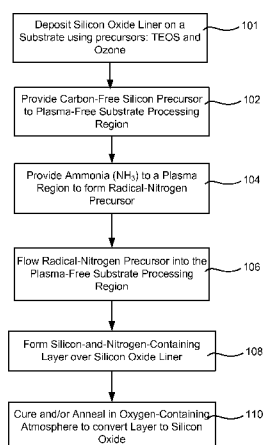


FIG 1

(57) Abstract: Methods of forming silicon oxide layers are described. The methods include mixing a carbon-free silicon-containing precursor with a radical-nitrogen precursor, and depositing a silicon-and-nitrogen-containing layer on a substrate. The radical-nitrogen precursor is formed in a plasma by flowing a hydrogen-and-nitrogen-containing precursor into the plasma. Prior to depositing the silicon-and-nitrogen-containing layer, a silicon oxide liner layer is formed to improve adhesion, smoothness and flowability of the silicon-and-nitrogen-containing layer. The silicon-and-nitrogen-containing layer may be converted to a silicon-and-oxygen-containing layer by curing and annealing the film. Methods also include forming a silicon oxide liner layer before applying a spin-on silicon-containing material.

## FLOWABLE DIELECTRIC USING OXIDE LINER

### CROSS-REFERENCES TO RELATED APPLICATIONS

This application is a PCT application of U.S. Patent Application No. 12/974,495 filed  
5 December 21, 2010, entitled "FLOWABLE DIELECTRIC USING OXIDE LINER," and  
claims the benefit of U.S. Provisional Patent Application No. 61/292,520, filed January 6,  
2010, entitled "RADICAL COMPONENT DIELECTRIC USING OXIDE LINER," the  
entire disclosure of which is incorporated herein by reference for all purposes.

### BACKGROUND OF THE INVENTION

10 Semiconductor device geometries have dramatically decreased in size since their introduction  
several decades ago. Modern semiconductor fabrication equipment routinely produces  
devices with 45 nm, 32 nm, and 28 nm feature sizes, and new equipment is being developed  
and implemented to make devices with even smaller geometries. The decreasing feature  
sizes result in structural features on the device having decreased spatial dimensions. The  
15 widths of gaps and trenches on the device narrow to a point where the aspect ratio of gap  
depth to its width becomes high enough to make it challenging to fill the gap with dielectric  
material. The depositing dielectric material is prone to clog at the top before the gap  
completely fills, producing a void or seam in the middle of the gap.

Over the years, many techniques have been developed to avoid having dielectric material  
20 clog the top of a gap, or to "heal" the void or seam that has been formed. One approach has  
been to start with highly flowable precursor materials that may be applied in a liquid phase to  
a spinning substrate surface (e.g., SOG deposition techniques). These flowable precursors  
can flow into and fill very small substrate gaps without forming voids or weak seams.  
However, once these highly flowable materials are deposited, they have to be hardened into a  
25 solid dielectric material.

In many instances, the hardening process includes a heat treatment to remove carbon and  
hydroxyl groups from the deposited material to leave behind a solid dielectric such as silicon  
oxide. Unfortunately, the departing carbon and hydroxyl species often leave behind pores in  
the hardened dielectric that reduce the quality of the final material. In addition, the hardening  
30 dielectric also tends to shrink in volume, which can leave cracks and spaces at the interface of

the dielectric and the surrounding substrate. In some instances, the volume of the hardened dielectric can decrease by 40% or more.

Thus, there is a need for new deposition processes and materials to form dielectric materials on structured substrates without generating voids, seams, or both, in substrate gaps and trenches. There is also a need for materials and methods of hardening flowable dielectric materials with fewer pores and a lower decrease in volume. This and other needs are addressed in the present application.

### BRIEF SUMMARY OF THE INVENTION

Methods of forming silicon oxide layers are described. The methods include subsequently mixing a carbon-free silicon-containing precursor with a radical-nitrogen precursor, and depositing a silicon-and-nitrogen-containing layer on a substrate. The radical-nitrogen precursor is formed in a plasma by flowing a hydrogen-and-nitrogen-containing precursor into the plasma. Prior to depositing the silicon-and-nitrogen-containing layer, a silicon oxide liner layer is formed to improve adhesion, smoothness and flowability of the silicon-and-nitrogen-containing layer. The silicon-and-nitrogen-containing layer may be converted to a silicon-and-oxygen-containing layer by curing and annealing the film. Methods also include forming a silicon oxide liner layer before applying a spin-on silicon-containing material.

Embodiments of the invention include methods of forming a flowable dielectric layer on a substrate. The methods include the sequential steps of (1) forming a generally conformal silicon oxide liner layer on the substrate by exposing the substrate to a silicon-containing liner precursor and an oxygen-containing liner precursor while the substrate is maintained at a liner deposition temperature, and (2) forming a carbon-free flowable silicon-nitrogen-and-hydrogen-containing layer on the substrate while the substrate is maintained at a bulk deposition temperature.

Additional embodiments and features are set forth in part in the description that follows, and in part will become apparent to those skilled in the art upon examination of the specification or may be learned by the practice of the invention. The features and advantages of the invention may be realized and attained by means of the instrumentalities, combinations, and methods described in the specification.

### BRIEF DESCRIPTION OF THE DRAWINGS

A further understanding of the nature and advantages of the present invention may be realized by reference to the remaining portions of the specification and the drawings wherein like

reference numerals are used throughout the several drawings to refer to similar components. In some instances, a sublabel is associated with a reference numeral and follows a hyphen to denote one of multiple similar components. When reference is made to a reference numeral without specification to an existing sublabel, it is intended to refer to all such multiple similar components.

Fig. 1 is a flowchart illustrating selected steps for making a silicon oxide film according to embodiments of the invention.

Fig. 2 is another flowchart illustrating selected steps for forming a silicon oxide film in a substrate gap according to embodiments of the invention.

Fig. 3 shows a substrate processing system according to embodiments of the invention.

Fig. 4A shows a substrate processing chamber according to embodiments of the invention.

Fig. 4B shows a showerhead of a substrate processing chamber according to embodiments of the invention.

#### DETAILED DESCRIPTION OF THE INVENTION

Methods of forming silicon oxide layers are described. The methods include mixing a carbon-free silicon-containing precursor with a radical-nitrogen precursor, and depositing a silicon-and-nitrogen-containing layer on a substrate. The radical-nitrogen precursor is formed in a plasma by flowing a hydrogen-and-nitrogen-containing precursor into the plasma. Prior to depositing the silicon-and-nitrogen-containing layer, a silicon oxide liner layer is formed to improve adhesion, smoothness and flowability of the silicon-and-nitrogen-containing layer. The silicon-and-nitrogen-containing layer may be converted to a silicon-and-oxygen-containing layer by curing and annealing the film. Methods also include forming a silicon oxide liner layer before applying a spin-on silicon-containing material.

Introducing an oxide liner layer between a substrate and a silicon-and-nitrogen-containing layer appears to improve adhesion and reduce incidences of delamination and cracking during and after subsequent processing. Silicon oxide films formed according to disclosed embodiments using a silicon oxide liner have also exhibited a smoother outer surface indicating a modification in the deposition dynamics. Without binding the coverage of the claims to theoretical mechanisms which may or may not be entirely correct, silanol groups may be present on the exposed surface of the silicon oxide liner layer and may serve to increase mobility thereby increasing nascent flowability of the silicon-and-nitrogen-

containing layer. Other benefits of including a silicon oxide layer include a more rapid initial growth rate on silicon nitride, a common underlying material in some applications. Other benefits also include accommodating contraction or expansion of the overlying layer relative to the underlying substrate. Following the deposition of the multilayer, the silicon-and-nitrogen-containing layer may be cured and/or annealed in oxygen-containing environments to convert the layer to silicon oxide.

Additional details about the methods and systems of forming the silicon oxide layer will now be described.

#### Exemplary Silicon Oxide Formation Process

Fig. 1 is a flowchart showing selected steps in methods 100 of making silicon oxide films according to embodiments of the invention. The method 100 includes depositing a silicon oxide liner on a substrate with concurrent flows of TEOS and ozone ( $O_3$ ) in operation 101. As with other liner depositions described herein, the substrate temperature during deposition is greater than  $400^\circ\text{C}$ , greater than  $500^\circ\text{C}$  and greater than  $600^\circ\text{C}$  in embodiments of the invention. Additives such as water ( $H_2O$ , steam), HMDS and TMDSO may be added to TEOS and ozone ( $O_3$ ) in order to ensure a more flowable or smooth deposition. The relatively high temperature of the substrate relative to the radical-component deposition which follows facilitates deposition on inert surfaces such as silicon nitride. The liner then presents a silicon oxide surface which is less inert and more conducive to relatively low-temperature deposition. Such deposition processes are known in the art as sub-atmospheric CVD (SACVD) but may also be conducted at pressures in excess of 1 atm. Exemplary flow-rates of TEOS may be greater than 0.1 gm/min (grams per minute), greater than 0.5 gm/min, greater than 1 gm/min and greater than 3 gm/min in different embodiments. Ozone may be flowed at greater than 1,000 sccm, greater than 3,000 sccm, greater than 10,000 sccm or greater than 30,000 sccm in different embodiments. Relatively inert carrier gases may be used to deliver the TEOS and optional additives to the substrate and the masses of the carrier gas are not included in the gm/min delivery rates given above.

The method continues and includes providing a carbon-free silicon-containing precursor to a substrate processing region (operation 102). The carbon-free silicon-containing precursor may be, for example, a silicon-and-nitrogen precursor, a silicon-and-hydrogen precursor, or a silicon-nitrogen-and-hydrogen-containing precursor, among other classes of silicon precursors. The silicon-precursor may be oxygen-free in addition to carbon-free. The lack of oxygen results in a lower concentration of silanol ( $Si-OH$ ) groups in the silicon-and-nitrogen

layer formed from the precursors. Excess silanol moieties in the deposited film can cause increased porosity and shrinkage during post deposition steps that remove the hydroxyl (-OH) moieties from the deposited layer.

Specific examples of carbon-free silicon-containing precursors may include silyl-amines such as  $\text{H}_2\text{N}(\text{SiH}_3)$ ,  $\text{HN}(\text{SiH}_3)_2$ , and  $\text{N}(\text{SiH}_3)_3$ , among other silyl-amines. The flow rates of a silyl-amine may be greater than or about 200 sccm, greater than or about 300 sccm or greater than or about 500 sccm in different embodiments. All flow rates given herein refer to a dual chamber substrate processing system. Single wafer systems would require half these flow rates and other wafer sizes would require flow rates scaled by the processed area. These silyl-amines may be mixed with additional gases that may act as carrier gases, reactive gases, or both. Examples of these additional gases may include  $\text{H}_2$ ,  $\text{N}_2$ ,  $\text{NH}_3$ , He, and Ar, among other gases. Examples of carbon-free silicon-containing precursors may also include silane ( $\text{SiH}_4$ ) either alone or mixed with other silicon (e.g.,  $\text{N}(\text{SiH}_3)_3$ ), hydrogen (e.g.,  $\text{H}_2$ ), and/or nitrogen (e.g.,  $\text{N}_2$ ,  $\text{NH}_3$ ) containing gases. Carbon-free silicon-containing precursors may also include disilane, trisilane, even higher-order silanes, and chlorinated silanes, alone or in combination with one another or the previously mentioned carbon-free silicon-containing precursors. The carbon-free silicon-containing precursor is not excited in a plasma region (e.g. a remote plasma region) before entering the plasma-free substrate processing region.

Ammonia ( $\text{NH}_3$ ) is delivered to a plasma region to form a radical-nitrogen precursor (operation 104). The radical-nitrogen precursor is a nitrogen-radical-containing precursor generated in the plasma region outside the substrate processing region from the ammonia. For example, the stable nitrogen precursor compound containing  $\text{NH}_3$  may be activated in a chamber plasma region or a remote plasma system (RPS) outside the processing chamber to form the radical-nitrogen precursor, which is then transported into the substrate processing region (operation 106). The flow rate of the ammonia may be greater than or about 300 sccm, greater than or about 500 sccm or greater than or about 700 sccm in different embodiments while additional precursors such as nitrogen ( $\text{N}_2$ ) and hydrogen ( $\text{H}_2$ ) may be included to adjust the nitrogen:hydrogen atomic flow ratio. The radical-nitrogen precursor may also be produced without using  $\text{NH}_3$ . Stable nitrogen precursors flowed into the remote plasma region may include one or more of  $\text{H}_2$ ,  $\text{N}_2$  and  $\text{N}_2\text{H}_4$ , in embodiments of the invention. The radical-nitrogen precursor produced in the chamber plasma region may be one or more of  $\cdot\text{N}$ ,  $\cdot\text{NH}$ ,  $\cdot\text{NH}_2$ , etc., and may also be accompanied by ionized species formed in the plasma.

In embodiments employing a chamber plasma region, the radical-nitrogen precursor is generated in a section of the substrate processing system partitioned from a substrate processing region where the precursors mix and react to deposit the silicon-and-nitrogen layer on a deposition substrate (e.g., a semiconductor wafer). The radical-nitrogen precursor may also be accompanied by a carrier gas such as helium, argon etc. The substrate processing region may be described herein as "plasma-free" during the growth of the silicon-and-nitrogen-containing layer and during the low temperature ozone cure. "Plasma-free" does not necessarily mean the region is devoid of plasma. Ionized species created within the plasma region do travel through pores (apertures) in the partition (showerhead) but the carbon-free silicon-containing precursor is not substantially excited by the plasma power applied to the plasma region. The borders of the plasma in the chamber plasma region are hard to define and may encroach upon the substrate processing region through the apertures in the showerhead. In the case of an inductively-coupled plasma, a small amount of ionization may be effected within the substrate processing region directly. Furthermore, a low intensity plasma may be created in the substrate processing region without eliminating the flowable nature of the forming film. Plasmas in the substrate processing region having much lower ion density than the chamber plasma region during the creation of the radical nitrogen precursor do not deviate from the scope of "plasma-free" as used herein.

In the substrate processing region, the carbon-free silicon-containing precursor and the radical-nitrogen precursor mix and react to form a silicon-and-nitrogen-containing film on the deposition substrate (operation 108). The deposited silicon-and-nitrogen-containing film may deposit conformally with recipe combinations which result in low deposition rates or high radical nitrogen fluxes at the deposition surface. In other embodiments, the deposited silicon-and-nitrogen-containing film has flowable characteristics unlike conventional silicon nitride ( $\text{Si}_3\text{N}_4$ ) film deposition techniques. The flowable nature of the formation allows the film to flow into narrow gaps trenches and other structures on the deposition surface of the substrate. The temperature of the substrate during deposition (operation 108) is less than  $120^\circ\text{C}$ , less than  $100^\circ\text{C}$ , less than  $80^\circ\text{C}$  and less than  $60^\circ\text{C}$  in different embodiments.

The flowability may be due to a variety of properties which result from mixing a radical-nitrogen precursors with the unexcited carbon-free silicon-containing precursor. These properties may include a significant hydrogen component in the deposited film and/or the presence of short chained polysilazane polymers. These short chains grow and network to form more dense dielectric material during and after the formation of the film. For example the deposited film may have a silazane-type, Si-NH-Si backbone (i.e., a Si-N-H film). When

both the silicon-containing precursor and the radical-nitrogen precursor are carbon-free, the deposited silicon-and-nitrogen-containing film is also substantially carbon-free. Of course, "carbon-free" does not necessarily mean the film lacks even trace amounts of carbon.

Carbon contaminants may be present in the precursor materials that find their way into the deposited silicon-and-nitrogen precursor. The amount of these carbon impurities however are much less than would be found in a silicon-containing precursor having a carbon moiety (e.g., TEOS, TMDSO, etc.), for example, in the liner layer grown in operation 101.

Following the deposition of the silicon-and-nitrogen-containing layer, the deposition substrate may be cured and/or annealed in oxygen-containing atmosphere(s) (operation 110).

The curing may occur in an ozone-containing atmosphere at a substrate temperature below or about 400°C. Under some conditions (e.g. between substrate temperatures from about 100°C to about 200°C) the conversion has been found to be substantially complete so a relatively high temperature anneal in an oxygen-containing environment may be unnecessary in embodiments of the invention. Following curing of the silicon-and-nitrogen containing layer, it may be desirable to anneal the substrate in an oxygen-containing atmosphere to further convert the film to silicon oxide. The oxygen-containing atmosphere may include one or more oxygen-containing gases such as molecular oxygen (O<sub>2</sub>), ozone (O<sub>3</sub>), water vapor (H<sub>2</sub>O), hydrogen peroxide (H<sub>2</sub>O<sub>2</sub>) and nitrogen-oxides (NO, NO<sub>2</sub>, etc.), among other oxygen-containing gases. The oxygen-containing atmosphere may also include radical oxygen and hydroxyl species such as atomic oxygen (O), hydroxides (OH), etc., that may be generated remotely and transported into the substrate chamber. Ions of oxygen-containing species may also be present. The oxygen anneal temperature of the substrate may be between about 500°C and about 1100°C. When a plasma is used, it may be in the substrate processing region, in a separate region separated by a showerhead or in a remote plasma system (RPS).

The oxygen-containing atmospheres of both the curing and oxygen anneal provide oxygen to convert the silicon-and-nitrogen-containing film into the silicon oxide (SiO<sub>2</sub>) film. As noted previously, the lack of carbon in the silicon-and-nitrogen-containing film results in significantly fewer pores formed in the final silicon oxide film. It also results in less volume reduction (i.e., shrinkage) of the film during the conversion to the silicon oxide. For example, where a silicon-nitrogen-carbon layer formed from carbon-containing silicon precursors and radical-nitrogen may shrink by 40 vol.% or more when converted to silicon oxide, the substantially carbon-free silicon-and-nitrogen films may shrink by about 17 vol.% or less.



Referring now to Fig. 2, another flowchart is shown illustrating selected steps in methods 200 for forming a silicon oxide film in a substrate gap according to embodiments of the invention. The method 200 includes depositing a silicon oxide liner on a patterned substrate having gaps in the spacing and structure of device components (e.g., transistors) formed on the substrate.

- 5 The gaps may have a height and width that define an aspect ratio (AR) of the height to the width (i.e., H/W) that is significantly greater than 1:1 (e.g., 5:1 or more, 6:1 or more, 7:1 or more, 8:1 or more, 9:1 or more, 10:1 or more, 11:1 or more, 12:1 or more, etc.). In many instances the high AR is due to small gap widths of that range from about 90 nm to about 22 nm or less (e.g., about 90 nm, 65 nm, 45 nm, 32 nm, 22 nm, 16 nm, etc.).
- 10 The silicon oxide liner is deposited with concurrent flows of TEOS and oxygen (O<sub>2</sub>) (operation 201). Substrate temperatures may be the same as the embodiments described with reference to Fig. 1 and the same additives may be added for the same purposes. The relatively high deposition temperature allows the deposition to proceed more rapidly than if the radical-component deposition were attempted without the liner layer. The liner presents a
- 15 silicon oxide surface which is less inert and more conducive to relatively low-temperature deposition. Such deposition processes are known in the art as sub-atmospheric CVD (SACVD) but may also be conducted at pressures in excess of 1 atm. Exemplary flow-rates of TEOS may be greater than 0.1 gm/min (grams per minute), greater than 0.5 gm/min, greater than 1 gm/min and greater than 3 gm/min in different embodiments. Oxygen may be
- 20 flowed at greater than 3,000 sccm, greater than 10,000 sccm, greater than 30,000 sccm or greater than 60,000 sccm in different embodiments. Relatively inert carrier gases are used to deliver the TEOS and optional additives to the substrate and the masses of the carrier gas are not included in the gm/min delivery rates given above.

- The substrate is then transferred to a substrate processing region (operation 202) and
- 25 ammonia (NH<sub>3</sub>) is excited in a separate chamber plasma region to form a radical-nitrogen precursor 204. A plasma in the chamber plasma region creates the radical-nitrogen precursor which flows through apertures in a showerhead separating the chamber plasma region from the substrate processing region. A carbon-free silicon-containing precursor is mixed with the radical nitrogen precursor in the substrate processing region (operation 206). A flowable
- 30 silicon-and-nitrogen-containing layer is deposited on the substrate (operation 208). Because the layer is flowable, it can fill the gaps having the high aspect ratios without creating voids or weak seams around the center of the filling material. For example, a depositing flowable material is less likely to prematurely clog the top of a gap before it is completely filled to

leave a void in the middle of the gap. The substrate temperature is below the temperatures discussed with reference to FIG. 1 in embodiments.

The as-deposited silicon-and-nitrogen-containing layer may then be cured in an ozone-containing atmosphere and/or annealed in an oxygen-containing atmosphere (operation 210) to transition the silicon-and-nitrogen-containing layer to silicon oxide. A further anneal (not shown) may be carried out in an inert environment at a higher substrate temperature in order to densify the silicon oxide layer. Curing and annealing the as-deposited silicon-and-nitrogen-containing layer in an oxygen-containing atmosphere forms a silicon oxide layer on the substrate, including the substrate gap. In embodiments, the processing parameters of operation 210 possess the same ranges described with reference to FIG. 1. As noted above, the silicon oxide layer has fewer pores and less volume reduction than similar layers formed with carbon-containing precursors that have significant quantities of carbon present in the layer before the heat treatment step. In many cases, the volume reduction is slight enough (e.g., about 15 vol.% or less) to avoid post heat treatment steps to fill, heal, or otherwise eliminate spaces that form in the gap as a result of the shrinking silicon oxide.

#### Exemplary Silicon Oxide Deposition System

Deposition chambers that may implement embodiments of the present invention may include high-density plasma chemical vapor deposition (HDP-CVD) chambers, plasma enhanced chemical vapor deposition (PECVD) chambers, sub-atmospheric chemical vapor deposition (SACVD) chambers, and thermal chemical vapor deposition chambers, among other types of chambers. Specific examples of CVD systems that may implement embodiments of the invention include the CENTURA ULTIMA® HDP-CVD chambers/systems, and PRODUCER® PECVD chambers/systems, available from Applied Materials, Inc. of Santa Clara, Calif.

Examples of substrate processing chambers that can be used with exemplary methods of the invention may include those shown and described in co-assigned U.S. Provisional Patent App. No. 60/803,499 to Lubomirsky et al, filed May 30, 2006, and titled "PROCESS CHAMBER FOR DIELECTRIC GAPFILL," the entire contents of which is herein incorporated by reference for all purposes. Additional exemplary systems may include those shown and described in U.S. Pat. Nos. 6,387,207 and 6,830,624, which are also incorporated herein by reference for all purposes.

Embodiments of the deposition systems may be incorporated into larger fabrication systems for producing integrated circuit chips. FIG. 3 shows one such system 300 of deposition,

baking and curing chambers according to disclosed embodiments. In the figure, a pair of FOUPs (front opening unified pods) 302 supply substrate substrates (e.g., 300 mm diameter wafers) that are received by robotic arms 304 and placed into a low pressure holding area 306 before being placed into one of the wafer processing chambers 308a-f. A second robotic arm 310 may be used to transport the substrate wafers from the holding area 306 to the processing chambers 308a-f and back.

The processing chambers 308a-f may include one or more system components for depositing, annealing, curing and/or etching a flowable dielectric film on the substrate wafer. In one configuration, two pairs of the processing chamber (e.g., 308c-d and 308e-f) may be used to deposit the flowable dielectric material on the substrate, and the third pair of processing chambers (e.g., 308a-b) may be used to anneal the deposited dielectric. In another configuration, the same two pairs of processing chambers (e.g., 308c-d and 308e-f) may be configured to both deposit and anneal a flowable dielectric film on the substrate, while the third pair of chambers (e.g., 308a-b) may be used for UV or E-beam curing of the deposited film. In still another configuration, all three pairs of chambers (e.g., 308a-f) may be configured to deposit and cure a flowable dielectric film on the substrate. In yet another configuration, two pairs of processing chambers (e.g., 308c-d and 308e-f) may be used for both deposition and UV or E-beam curing of the flowable dielectric, while a third pair of processing chambers (e.g. 308a-b) may be used for annealing the dielectric film. Any one or more of the processes described may be carried out on chamber(s) separated from the fabrication system shown in different embodiments.

In addition, one or more of the process chambers 308a-f may be configured as a wet treatment chamber. These process chambers include heating the flowable dielectric film in an atmosphere that include moisture. Thus, embodiments of system 300 may include wet treatment chambers 308a-b and anneal processing chambers 308c-d to perform both wet and dry anneals on the deposited dielectric film.

A silicon oxide liner layer may deposited in one of the chambers which is configured to deposit a liner layer via sub-atmospheric chemical vapor deposition (SACVD). Other names may be used to describe relatively high pressure processes involving exposure of the substrate to a combination of TEOS and ozone or TEOS and oxygen (O<sub>2</sub>). Such systems are also available from Applied Materials, Inc. of Santa Clara, California. The silicon oxide liner layer thickness may be less than 100Å, less than 75Å, less than 50Å and less than 25Å in different embodiments. Following deposition of a silicon oxide liner layer, the substrate may be transferred to a radical-component CVD chamber as described in FIG. 4. Alternatively, a

spin-on dielectric (SOD), spin-on glass (SOG) or other spin-on silicon-containing film may be applied. Spin-on flowable materials will offer similar benefits to radical-component flowable deposition in that the film will exhibit less delamination and cracking. SOD films containing silicon and nitrogen will exhibit these benefits in embodiments of the invention.

5 FIG. 4A is a substrate processing chamber 400 according to disclosed embodiments. A remote plasma system (RPS) 410 may process a gas which then travels through a gas inlet assembly 411. Two distinct gas supply channels are visible within the gas inlet assembly 411. A first channel 412 carries a gas that passes through the remote plasma system RPS 410, while a second channel 413 bypasses the RPS 400. The first channel 402 may be used  
10 for the process gas and the second channel 413 may be used for a treatment gas in disclosed embodiments. The lid (or conductive top portion) 421 and a perforated partition 453 are shown with an insulating ring 424 in between, which allows an AC potential to be applied to the lid 421 relative to perforated partition 453. The process gas travels through first channel 412 into chamber plasma region 420 and may be excited by a plasma in chamber plasma  
15 region 420 alone or in combination with RPS 410. The combination of chamber plasma region 420 and/or RPS 410 may be referred to as a remote plasma system herein. The perforated partition (also referred to as a showerhead) 453 separates chamber plasma region 420 from a substrate processing region 470 beneath showerhead 453. Showerhead 453 allows a plasma present in chamber plasma region 420 to avoid directly exciting gases in  
20 substrate processing region 470, while still allowing excited species to travel from chamber plasma region 420 into substrate processing region 470.

Showerhead 453 is positioned between chamber plasma region 420 and substrate processing region 470 and allows plasma effluents (excited derivatives of precursors or other gases) created within chamber plasma region 420 to pass through a plurality of through holes 456  
25 that traverse the thickness of the plate. The showerhead 453 also has one or more hollow volumes 451 which can be filled with a precursor in the form of a vapor or gas (such as a silicon-containing precursor) and pass through small holes 455 into substrate processing region 470 but not directly into chamber plasma region 420. Showerhead 453 is thicker than the length of the smallest diameter 450 of the through-holes 456 in this disclosed  
30 embodiment. In order to maintain a significant concentration of excited species penetrating from chamber plasma region 420 to substrate processing region 470, the length 426 of the smallest diameter 450 of the through-holes may be restricted by forming larger diameter portions of through-holes 456 part way through the showerhead 453. The length of the

smallest diameter 450 of the through-holes 456 may be the same order of magnitude as the smallest diameter of the through-holes 456 or less in disclosed embodiments.

In the embodiment shown, showerhead 453 may distribute (via through holes 456) process gases which contain oxygen, hydrogen and/or nitrogen and/or plasma effluents of such process gases upon excitation by a plasma in chamber plasma region 420. In embodiments, process gases excited in RPS 410 and/or chamber plasma region 420 include ammonia ( $\text{NH}_3$ ) and nitrogen ( $\text{N}_2$ ) and/or hydrogen ( $\text{H}_2$ ). Generally speaking, the process gas introduced into the RPS 410 and/or chamber plasma region 420 through first channel 412 may contain one or more of oxygen ( $\text{O}_2$ ), ozone ( $\text{O}_3$ ),  $\text{N}_2\text{O}$ ,  $\text{NO}$ ,  $\text{NO}_2$ ,  $\text{NH}_3$ ,  $\text{N}_x\text{H}_y$  including  $\text{N}_2\text{H}_4$ , silane, disilane, TSA and DSA. The process gas may also include a carrier gas such as helium, argon, nitrogen ( $\text{N}_2$ ), etc. The second channel 413 may also deliver a process gas and/or a carrier gas, and/or a film-curing gas used to remove an unwanted component from the growing or as-deposited film. Plasma effluents may include ionized or neutral derivatives of the process gas and may also be referred to herein as a radical-oxygen precursor and/or a radical-nitrogen precursor referring to the atomic constituents of the process gas introduced.

In embodiments, the number of through-holes 456 may be between about 60 and about 2000. Through-holes 456 may have a variety of shapes but are most easily made round. The smallest diameter 450 of through holes 456 may be between about 0.5 mm and about 20 mm or between about 1mm and about 6mm in disclosed embodiments. There is also latitude in choosing the cross-sectional shape of through-holes, which may be made conical, cylindrical or a combination of the two shapes. The number of small holes 455 used to introduce a gas into substrate processing region 470 may be between about 100 and about 5000 or between about 500 and about 2000 in different embodiments. The diameter of the small holes 455 may be between about 0.1 mm and about 2 mm.

FIG. 4B is a bottom view of a showerhead 453 for use with a processing chamber according to disclosed embodiments. Showerhead 453 corresponds with the showerhead shown in FIG. 4A. Through-holes 456 are depicted with a larger inner-diameter (ID) on the bottom of showerhead 453 and a smaller ID at the top. Small holes 455 are distributed substantially evenly over the surface of the showerhead, even amongst the through-holes 456 which helps to provide more even mixing than other embodiments described herein.

An exemplary film is created on a substrate supported by a pedestal (not shown) within substrate processing region 470 when plasma effluents arriving through through-holes 456 in showerhead 453 combine with a silicon-containing precursor arriving through the small holes

455 originating from hollow volumes 451. Though substrate processing region 470 may be equipped to support a plasma for other processes such as curing, no plasma is present during the growth of the exemplary film.

A plasma may be ignited either in chamber plasma region 420 above showerhead 453 or substrate processing region 470 below showerhead 453. A plasma is present in chamber plasma region 420 to produce the radical nitrogen precursor from an inflow of a nitrogen-and-hydrogen-containing gas. An AC voltage typically in the radio frequency (RF) range is applied between the conductive top portion 421 of the processing chamber and showerhead 453 to ignite a plasma in chamber plasma region 420 during deposition. An RF power supply generates a high RF frequency of 13.56 MHz but may also generate other frequencies alone or in combination with the 13.56 MHz frequency.

The top plasma may be left at low or no power when the bottom plasma in the substrate processing region 470 is turned on to either cure a film or clean the interior surfaces bordering substrate processing region 470. A plasma in substrate processing region 470 is ignited by applying an AC voltage between showerhead 453 and the pedestal or bottom of the chamber. A cleaning gas may be introduced into substrate processing region 470 while the plasma is present.

The pedestal may have a heat exchange channel through which a heat exchange fluid flows to control the temperature of the substrate. This configuration allows the substrate temperature to be cooled or heated to maintain relatively low temperatures (from room temperature through about 120°C). The heat exchange fluid may comprise ethylene glycol and water. The wafer support platter of the pedestal (preferably aluminum, ceramic, or a combination thereof) may also be resistively heated in order to achieve relatively high temperatures (from about 120°C through about 1100°C) using an embedded single-loop embedded heater element configured to make two full turns in the form of parallel concentric circles. An outer portion of the heater element may run adjacent to a perimeter of the support platter, while an inner portion runs on the path of a concentric circle having a smaller radius. The wiring to the heater element passes through the stem of the pedestal.

The substrate processing system is controlled by a system controller. In an exemplary embodiment, the system controller includes a hard disk drive, a floppy disk drive and a processor. The processor contains a single-board computer (SBC), analog and digital input/output boards, interface boards and stepper motor controller boards. Various parts of CVD system conform to the Versa Modular European (VME) standard which defines board,

card cage, and connector dimensions and types. The VME standard also defines the bus structure as having a 16-bit data bus and a 24-bit address bus.

The system controller controls all of the activities of the CVD machine. The system controller executes system control software, which is a computer program stored in a computer-readable medium. Preferably, the medium is a hard disk drive, but the medium may also be other kinds of memory. The computer program includes sets of instructions that dictate the timing, mixture of gases, chamber pressure, chamber temperature, RF power levels, susceptor position, and other parameters of a particular process. Other computer programs stored on other memory devices including, for example, a floppy disk or other another appropriate drive, may also be used to instruct the system controller.

A process for depositing a film stack on a substrate or a process for cleaning a chamber can be implemented using a computer program product that is executed by the system controller. The computer program code can be written in any conventional computer readable programming language: for example, 68000 assembly language, C, C++, Pascal, Fortran or others. Suitable program code is entered into a single file, or multiple files, using a conventional text editor, and stored or embodied in a computer usable medium, such as a memory system of the computer. If the entered code text is in a high level language, the code is compiled, and the resultant compiler code is then linked with an object code of precompiled Microsoft Windows® library routines. To execute the linked, compiled object code the system user invokes the object code, causing the computer system to load the code in memory. The CPU then reads and executes the code to perform the tasks identified in the program.

The interface between a user and the controller is via a flat-panel touch-sensitive monitor. In the preferred embodiment two monitors are used, one mounted in the clean room wall for the operators and the other behind the wall for the service technicians. The two monitors may simultaneously display the same information, in which case only one accepts input at a time. To select a particular screen or function, the operator touches a designated area of the touch-sensitive monitor. The touched area changes its highlighted color, or a new menu or screen is displayed, confirming communication between the operator and the touch-sensitive monitor. Other devices, such as a keyboard, mouse, or other pointing or communication device, may be used instead of or in addition to the touch-sensitive monitor to allow the user to communicate with the system controller.

As used herein "substrate" may be a support substrate with or without layers formed thereon. The support substrate may be an insulator or a semiconductor of a variety of doping concentrations and profiles and may, for example, be a semiconductor substrate of the type used in the manufacture of integrated circuits. A layer of "silicon oxide" may include  
5 minority concentrations of other elemental constituents such as nitrogen, hydrogen, carbon and the like. A gas in an "excited state" describes a gas wherein at least some of the gas molecules are in vibrationally-excited, dissociated and/or ionized states. A gas may be a combination of two or more gases. The term "trench" is used throughout with no implication that the etched geometry has a large horizontal aspect ratio. Viewed from above the surface,  
10 trenches may appear circular, oval, polygonal, rectangular, or a variety of other shapes. The term "via" is used to refer to a low aspect ratio trench which may or may not be filled with metal to form a vertical electrical connection. The term "precursor" is used to refer to any process gas which takes part in a reaction to either remove or deposit material from a surface.

Having described several embodiments, it will be recognized by those of skill in the art that  
15 various modifications, alternative constructions, and equivalents may be used without departing from the spirit of the invention. Additionally, a number of well-known processes and elements have not been described in order to avoid unnecessarily obscuring the present invention. Accordingly, the above description should not be taken as limiting the scope of the invention.

Where a range of values is provided, it is understood that each intervening value, to the tenth  
20 of the unit of the lower limit unless the context clearly dictates otherwise, between the upper and lower limits of that range is also specifically disclosed. Each smaller range between any stated value or intervening value in a stated range and any other stated or intervening value in that stated range is encompassed. The upper and lower limits of these smaller ranges may  
25 independently be included or excluded in the range, and each range where either, neither or both limits are included in the smaller ranges is also encompassed within the invention, subject to any specifically excluded limit in the stated range. Where the stated range includes one or both of the limits, ranges excluding either or both of those included limits are also included.

As used herein and in the appended claims, the singular forms "a", "an", and "the" include  
30 plural referents unless the context clearly dictates otherwise. Thus, for example, reference to "a process" includes a plurality of such processes and reference to "the precursor" includes reference to one or more precursor and equivalents thereof known to those skilled in the art, and so forth.



Also, the words "comprise," "comprising," "include," "including," and "includes" when used in this specification and in the following claims are intended to specify the presence of stated features, integers, components, or steps, but they do not preclude the presence or addition of one or more other features, integers, components, steps, acts, or groups.

WHAT IS CLAIMED IS:

1. A method of forming a flowable dielectric layer on a substrate, the method comprising the sequential steps of:

forming a generally conformal silicon oxide liner layer on the substrate by  
5 exposing the substrate to a silicon-containing liner precursor and an oxygen-containing liner precursor, wherein the substrate is maintained at a liner deposition temperature; and  
forming a carbon-free flowable silicon-nitrogen-and-hydrogen-containing layer on the substrate, wherein the substrate is maintained at a bulk deposition temperature.

2. The method of claim 1 wherein forming the carbon-free flowable  
10 silicon-nitrogen-and-hydrogen-containing layer comprises:

flowing a nitrogen-and-hydrogen-containing gas into a plasma region to produce a radical-nitrogen precursor;

combining a carbon-free silicon-containing precursor with the radical-nitrogen precursor in a plasma-free substrate processing region; and

15 depositing a silicon-and-nitrogen-containing layer on the substrate.

3. The method of claim 2 wherein the nitrogen-and-hydrogen-containing gas comprises ammonia (NH<sub>3</sub>).

4. The method of claim 2 wherein the nitrogen-and-hydrogen-containing gas comprises at least one of nitrogen (N<sub>2</sub>), hydrogen (H<sub>2</sub>), hydrazine (N<sub>2</sub>H<sub>4</sub>) and ammonia  
20 (NH<sub>3</sub>).

5. The method of claim 1 wherein a thickness of the generally conformal silicon oxide liner is less than or about 100Å.

6. The method of claim 2 wherein the carbon-free silicon-containing precursor comprises N(SiH<sub>3</sub>)<sub>3</sub>.

25 7. The method of claim 1 wherein the carbon-free flowable silicon-nitrogen-and-hydrogen-containing layer comprises a carbon-free Si-N-H layer.

8. The method of claim 1 wherein the bulk-deposition temperature is lower than the liner deposition temperature.

9. The method of claim 1 wherein the bulk-deposition temperature is less  
30 than 120°C.

10. The method of claim 1 wherein the liner deposition temperature is greater than 400°C.

11. The method of claim 1 further comprising a cure step maintaining a temperature of the substrate below or about 400°C in an ozone-containing atmosphere.

5 12. The method of claim 11 further comprising a post-cure step of raising a temperature of the substrate to an oxygen anneal temperature above or about 600°C in an oxygen-containing atmosphere comprising one or more gases selected from the group consisting of atomic oxygen, ozone, and steam (H<sub>2</sub>O).

10 13. The method of claim 2 wherein the plasma region is in a remote plasma system.

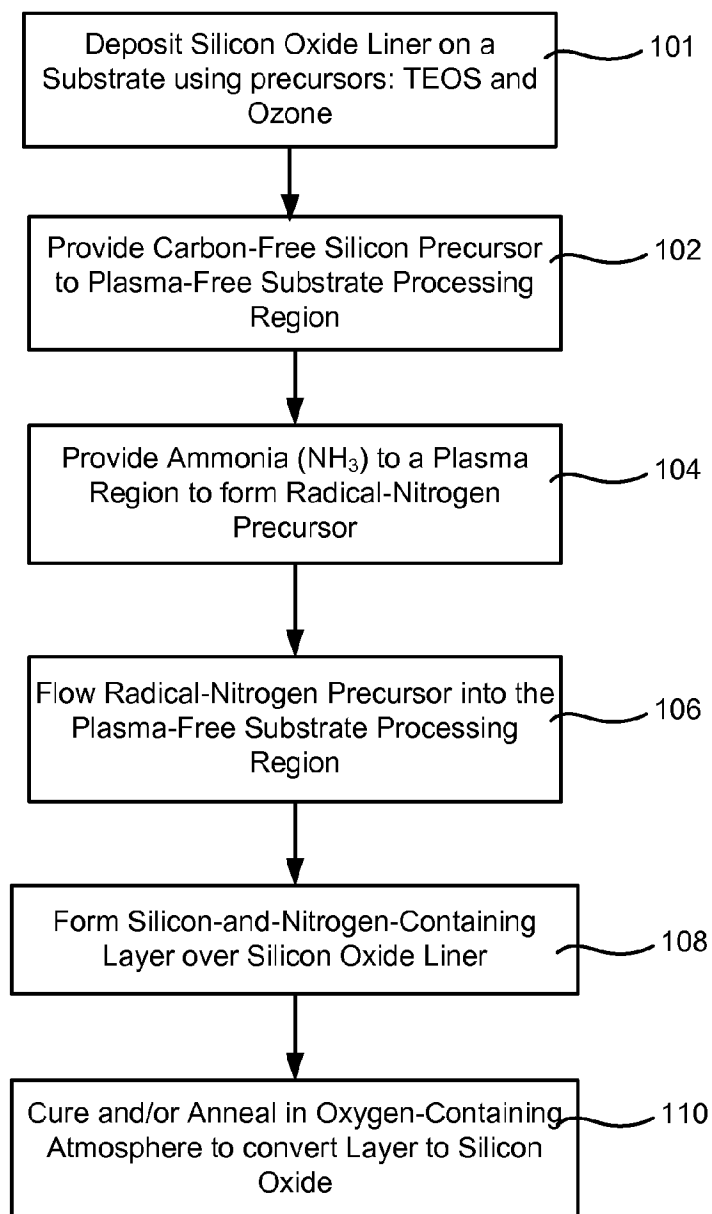
14. The method of claim 2 wherein the plasma region is a partitioned portion of the substrate processing chamber separated from the plasma-free substrate processing region by a showerhead.

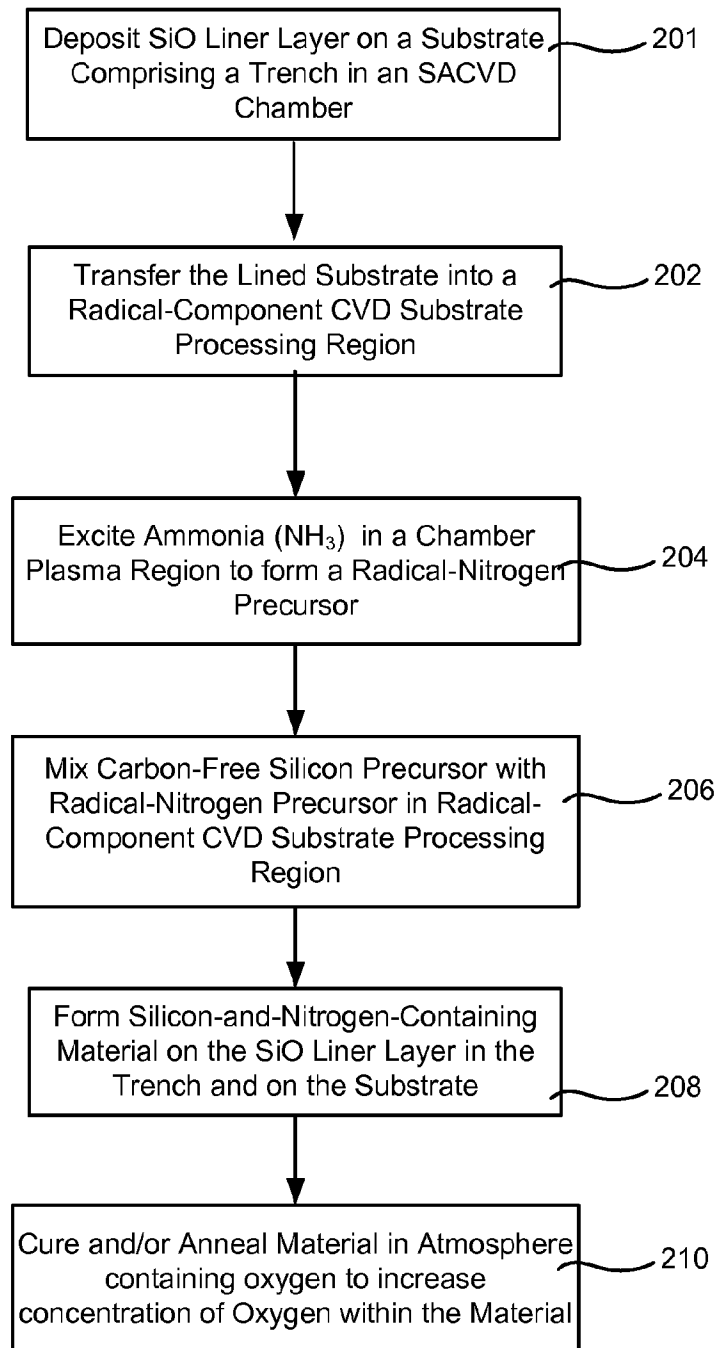
15 15. The method of claim 1 wherein the oxygen-containing liner precursor comprises ozone.

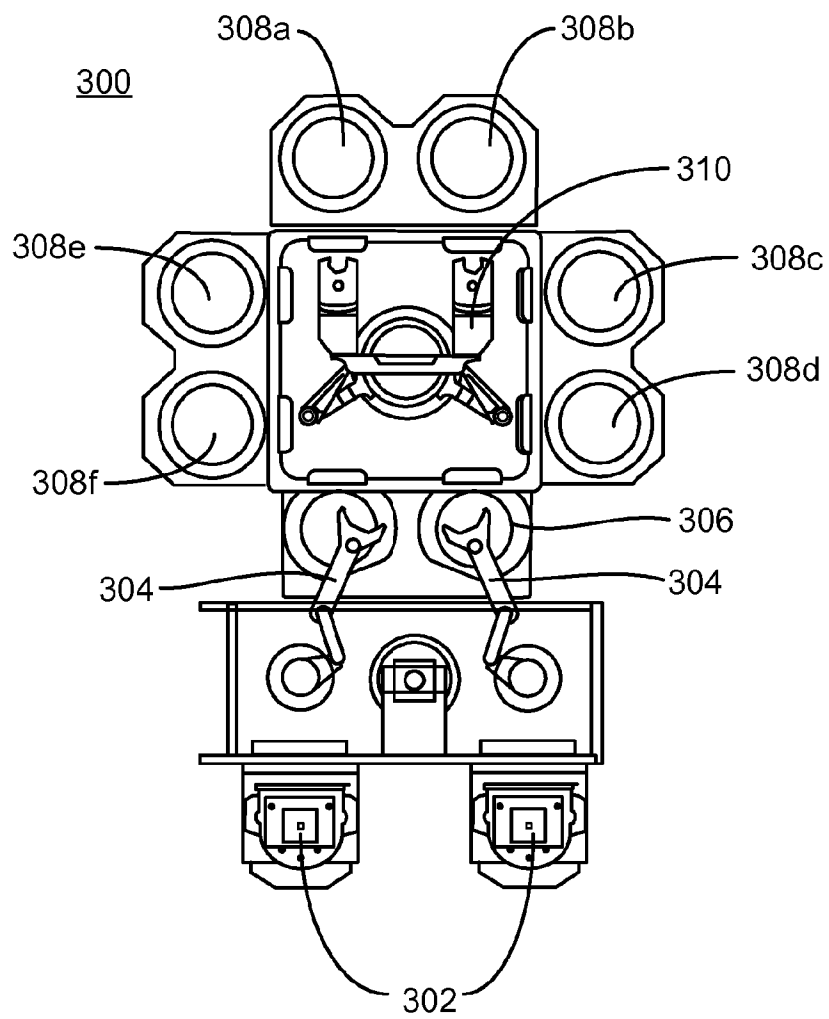
16. The method of claim 1 wherein the oxygen-containing liner precursor comprises oxygen (O<sub>2</sub>).

17. The method of claim 1 wherein the silicon-containing liner precursor comprises TEOS.

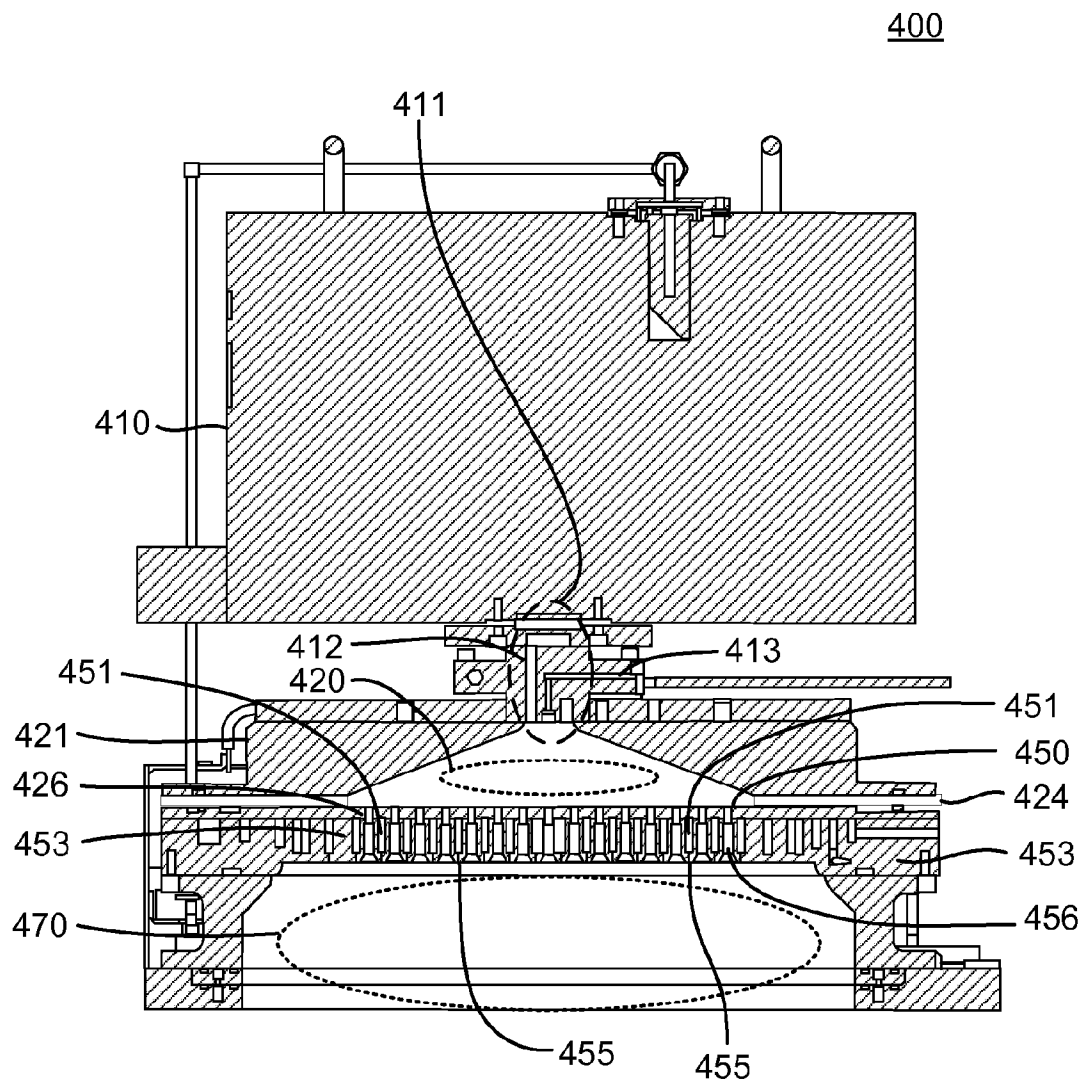
20 18. The method of claim 1 wherein the operation of depositing a silicon-and-nitrogen-containing layer comprises depositing a spin-on dielectric (SOD) layer.

**100****FIG 1**

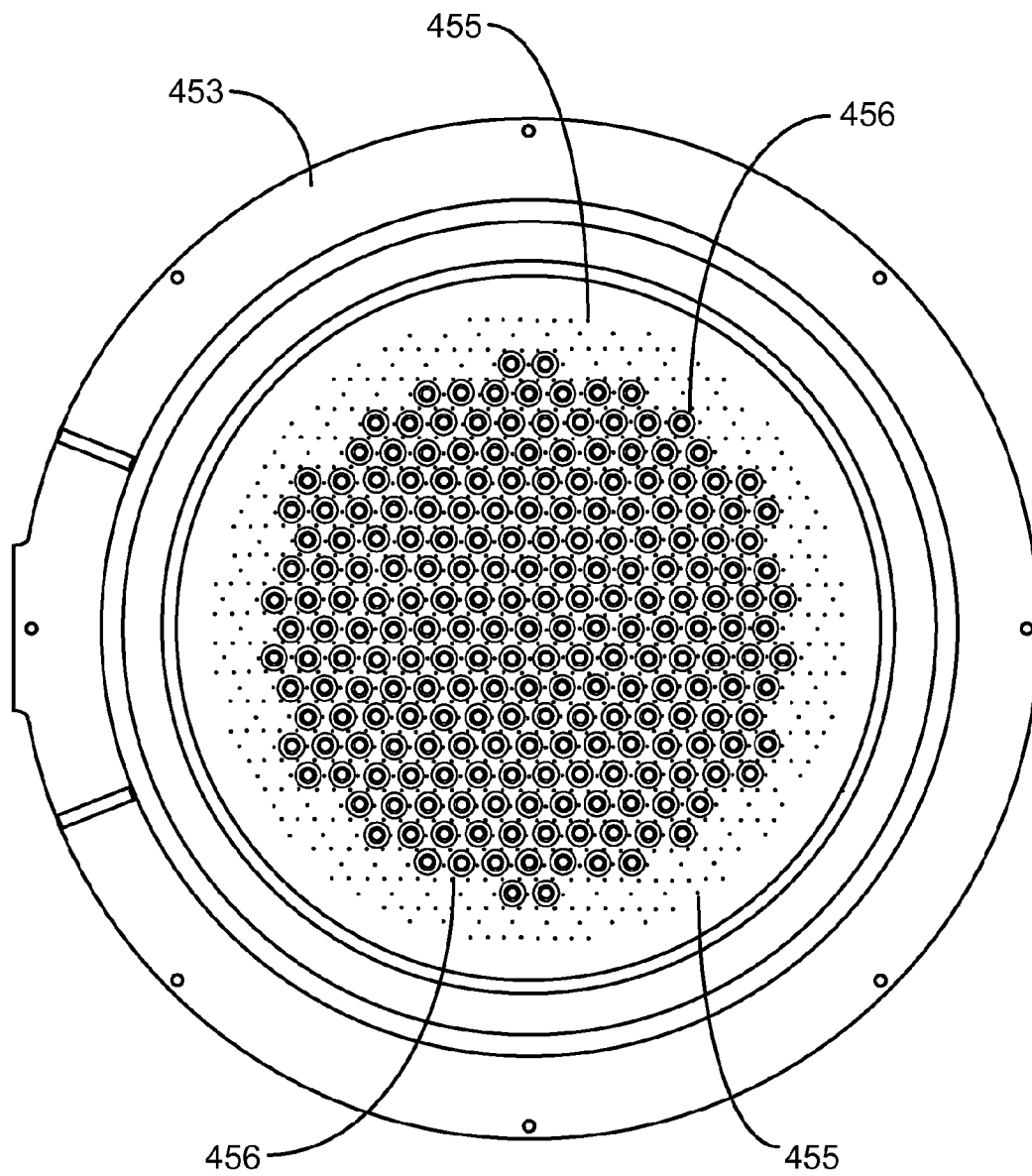
**200****FIG 2**



**FIG. 3**



**FIG. 4A**

**FIG. 4B**