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- (71) **Applicant:** L'AIR LIQUIDE, SOCIETE ANONYME POUR L'ETUDE ET EXPLOITATION DES PROCEDES GEORGES CLAUDE [FR/FR]; 75 quai d'Orsay, F-75007 Paris (FR).
- (72) **Inventors; and**
- (71) **Applicants (for US only):** SANDERS, Edgar, S., Jr. [US/US]; 3 Penn State Drive, Newark, Delaware 19713 (US). KULKARNI, Sudhir S. [US/US]; 1801 Breen Lane, Wilmington, Delaware 19810 (US).
- (74) **Agents:** CRONIN, Christopher J. et al.; Attn: Intellectual Property Department, 2700 Post Oak Blvd., Suite 1800, Houston, Texas 77056 (US).
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(54) **Title:** MEMBRANES MADE FROM 6FDA, DETDA, AND DABA-BASED POLYMERS

(57) **Abstract:** A polymeric membrane made of a polyimide polymer or copolymer essentially consisting of repeating units of dianhydride-derived units and diamine-derived units. At least some of the dianhydride-derived units are derived from 2,2-bis (3,4-dicarboxyphenyl) hexafluoropropane (6FDA). At least some of the diamine-derived units are derived from 2,5-diethyl-6-methyl-1,3-diamino benzene (DETD). The balance of the diamine-derived units are derived from a diamino benzene ring with a carboxylic acid substituent.



MEMBRANES MADE FROM 6FDA, DETDA, AND DABA-BASED POLYMERS

Background

Field of the Invention

5 The present invention relates to carbon molecular sieve membranes and gas separations utilizing the same.

Related Art

10 Membranes are viewed as selective barriers between two phases. Due to the random thermal fluctuations within the polymer matrix, gas molecules from the high partial pressure side sorb into the membrane and diffuse through under the influence of a chemical potential gradient, and finally desorb to the low partial pressure side. Two terms, "permeability" and "selectivity", are used to describe the most important properties of membranes-productivity and separation efficiency
15 respectively. Permeability (P) equals the pressure and thickness normalized flux, as shown in the following equation:

$$P_i = \frac{n_i \cdot l}{\Delta p_i} \quad (1)$$

20 where n_i is the penetrant flux through the membrane of thickness (l) under a partial pressure (Δp_i). The most frequently used unit for permeability, Barrer, is defined as below:

$$\text{Barrer} = 10^{-10} \frac{\text{cc(STP)} \cdot \text{cm}}{\text{cm}^2 \cdot \text{s} \cdot \text{cmHg}} \quad (2)$$

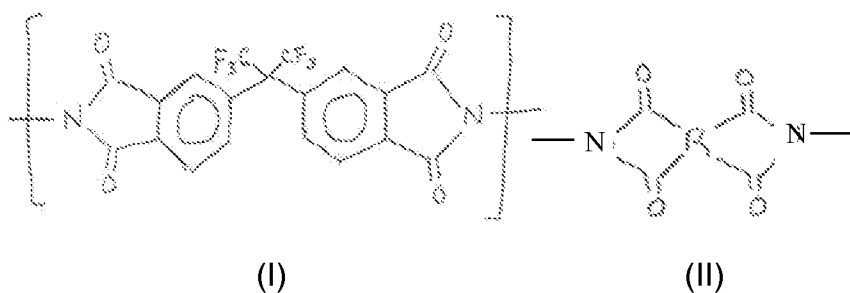
25 Selectivity is a measure of the ability of one gas to flow through the membrane over that of another gas. When the downstream pressure is negligible, the ideal selectivity (based upon the permeabilities of pure gases) of the membrane, can be used to approximate the real selectivity (based upon the permeabilities of the gases in a gas mixture). In this case, the selectivity ($\alpha_{A/B}$) is the permeability of a first gas A divided by the permeability of a second gas B.

 Currently, polymeric membranes are well studied and widely available for gaseous separations due to easy processability and low cost. In particular,

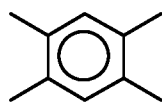
polyimides have high glass transition temperatures, are easy to process, and have one of the highest separation performance properties among other polymeric membranes. The patent literature (including US 2011/138852; US 5,618,334; US 5,928,410; and US 4,981,497) discloses one particular class of polyimides for use in polymeric gas separation membranes that is based upon the reaction of a diamine(s) with 2,2-bis (3,4-dicarboxyphenyl) hexafluoropropane dianhydride (6FDA).

Summary

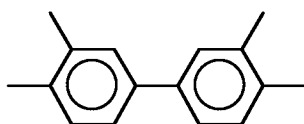
There is disclosed a membrane comprising a polyimide polymer or copolymer having repeating units of dianhydride-derived units and diamine-derived units, at least some of, and as much as 100% of, the dianhydride-derived units being the dianhydride-derived moiety of formula (I) with a balance of the dianhydride-derived units, if any, being the dianhydride-derived moiety of formula (II)



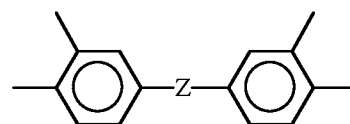
Each R is a molecular segment independently selected from the group consisting of formula (1), formula (2), formula (3), and formula (4)



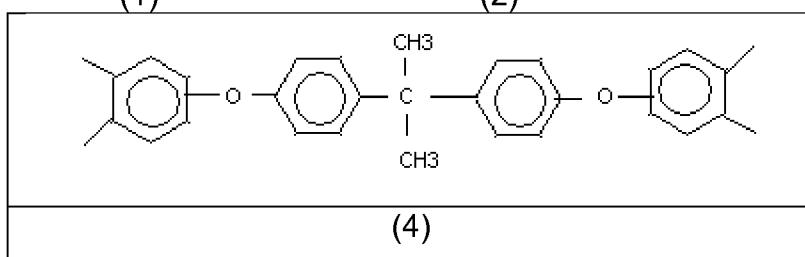
(1)



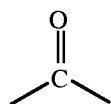
(2)



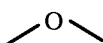
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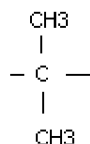
Each Z is a molecular segment independently selected from the group consisting of formula (5), formula (6), formula (7), and formula (8)



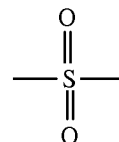
(5)



(6)

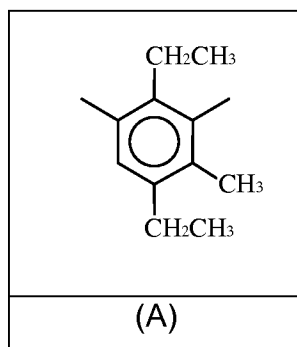


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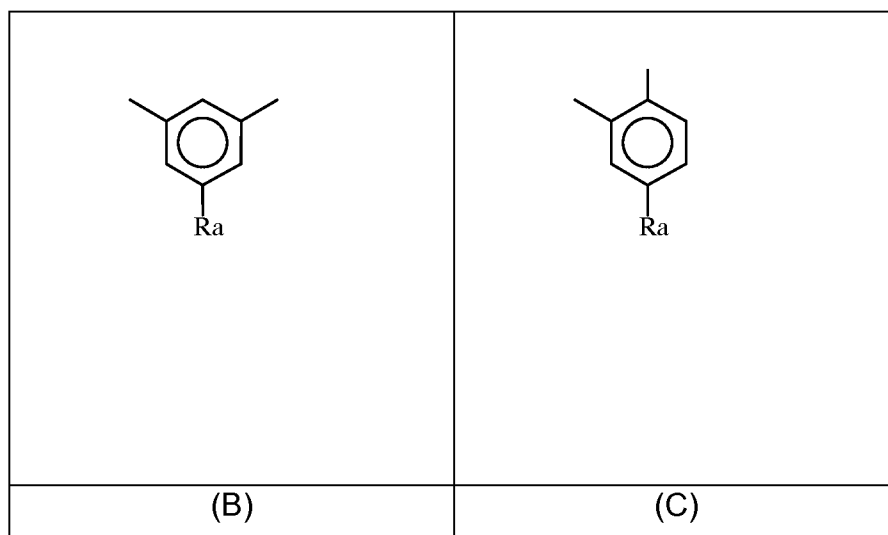


(8)

At least some of the diamine-derived units are the diamine-derived moiety of formula (A)



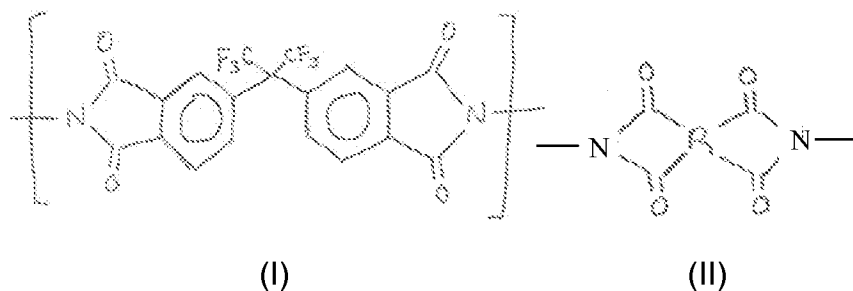
At least some of the diamine-derived units are the diamine-derived moiety of formula (B) or formula (C):



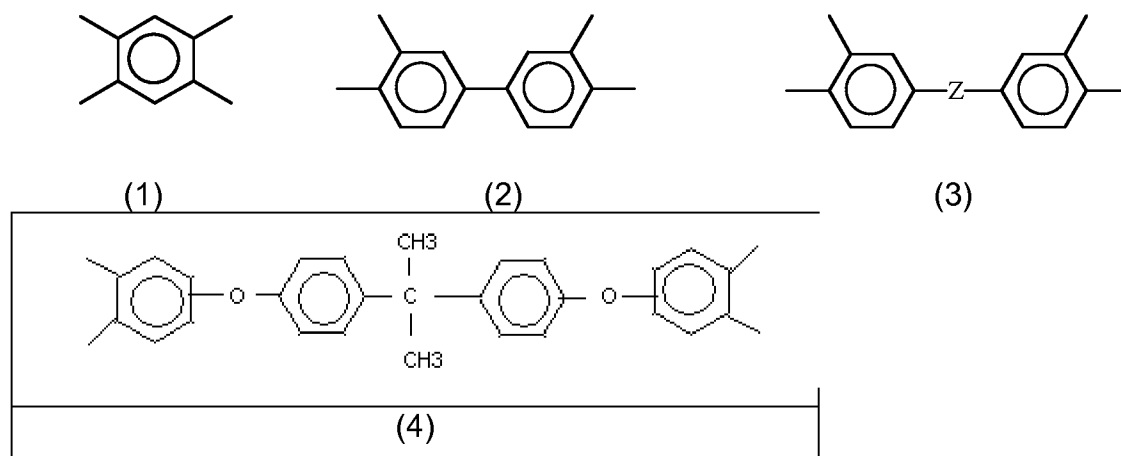
Each R_a is a straight or branched C_1 to C_6 alkyl group having a terminal carboxylic acid group.

There is also disclosed a method of producing a hollow fiber membrane that includes the following steps. A solution or suspension of a polyimide polymer

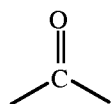
or copolymer binder and a solvent for the binder is prepared. A spinneret adapted and configured to continuously extrude one or more nascent hollow fibers is provided. The spinneret has an inner annular channel disposed concentrically within an outer annular channel. A bore fluid is fed through the inner annular channel to form a cylindrical fluid stream positioned concentrically within the fiber. The solution or suspension is fed through the outer annular channel so that it surrounds the cylindrical fluid stream to form a nascent hollow fiber. The nascent hollow fiber is passed from the spinneret through an air gap. The nascent hollow fiber is immersed in a liquid coagulant to facilitate phase inversion. The fiber is removed from the coagulant. The polyimide polymer or copolymer has repeating units of dianhydride-derived units and diamine-derived units, at least some of, and as much as 100% of, the dianhydride-derived units being the dianhydride-derived moiety of formula (I) with a balance of the dianhydride-derived units, if any, being the dianhydride-derived moiety of formula (II)



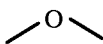
Each R is a molecular segment independently selected from the group consisting of formula (1), formula (2), formula (3), and formula (4)



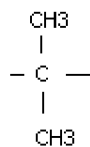
Each Z is a molecular segment independently selected from the group consisting of formula (5), formula (6), formula (7), and formula (8)



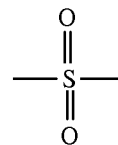
(5)



(6)

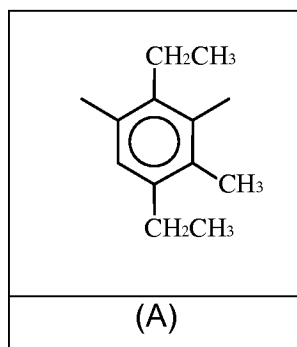


(7)

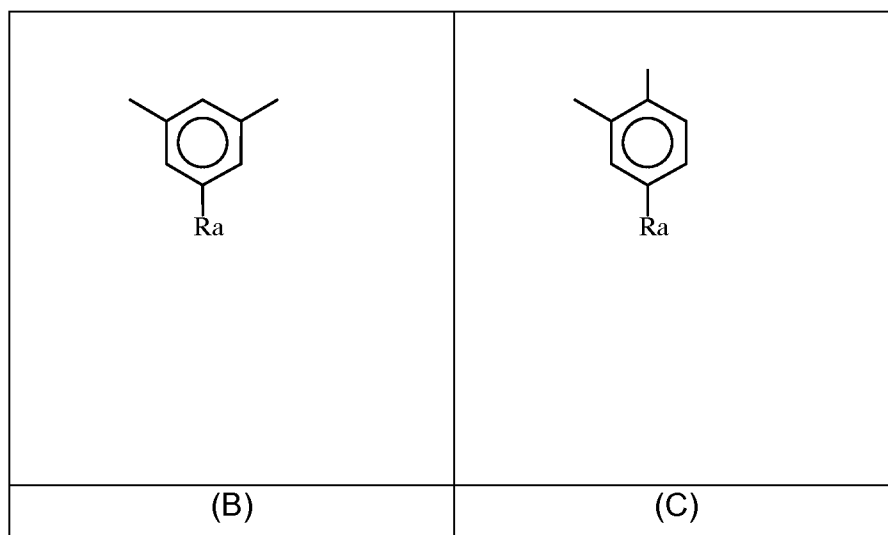


(8)

At least some of the diamine-derived units are the diamine-derived moiety of formula (A)



At least some of the diamine-derived units are the diamine-derived moiety of formula (B) or formula (C):



Each R_a is a straight or branched C_1 to C_6 alkyl group having a terminal carboxylic acid group.

The membrane and/or method may include one or more of the following aspects:

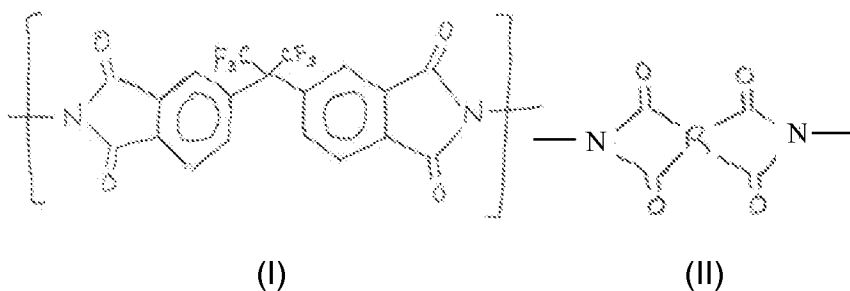
- 100% of the dianhydride-derived units are the dianhydride-derived moiety of formula (I).
- R_a is a $-\text{COOH}$ group.
- less than 100% of the dianhydride-derived units are the dianhydride-derived moiety of formula (I) and the balance of the dianhydride-derived units are the dianhydride-derived moiety of formula (II).
- R_a is a $-\text{COOH}$ group.
- R is the molecular segment of formula (2).
- Z is the molecular segment of formula (5).
- R_a is a $-\text{COOH}$ group.
- R is the molecular segment of formula (3).
- Z is the molecular segment of formula (5).
- R_a is a $-\text{COOH}$ group.
- a ratio of the diamine-derived moieties of formula (A) to that of either formula (B) or formula (C) ranges from about 3:2 to about 2:3.

Description of Preferred Embodiments

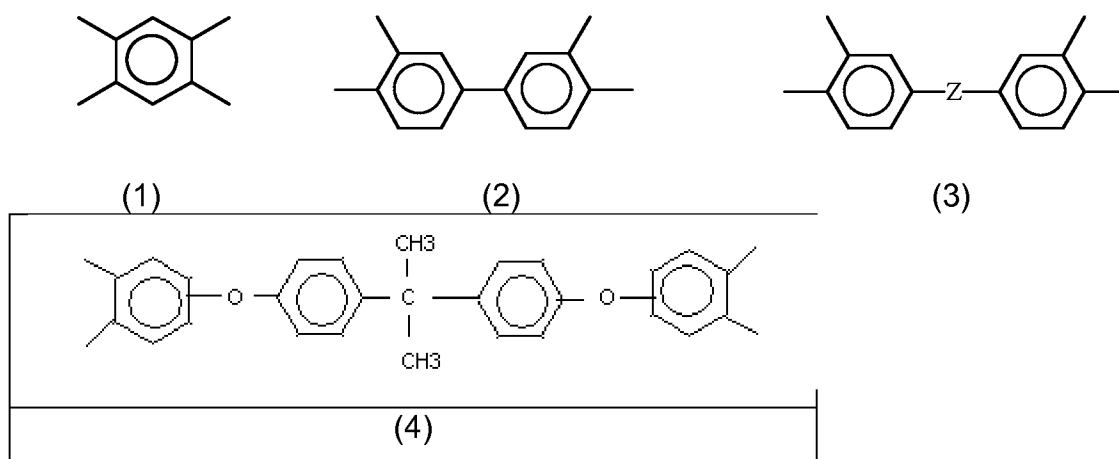
The membranes of the invention are expected to exhibit relatively high permeabilities and selectivities in various gas separations, including CO_2/CH_4 , O_2/N_2 , and $\text{C}_3\text{H}_6/\text{C}_3\text{H}_8$.

The membrane is made from a polyimide polymer or copolymer having repeating units of dianhydride-derived units and diamine-derived units.

At least some (and as much as 100%) of the dianhydride-derived units are the dianhydride-derived moiety of formula (I) with the balance (if any) being the dianhydride-derived moiety of formula (II):

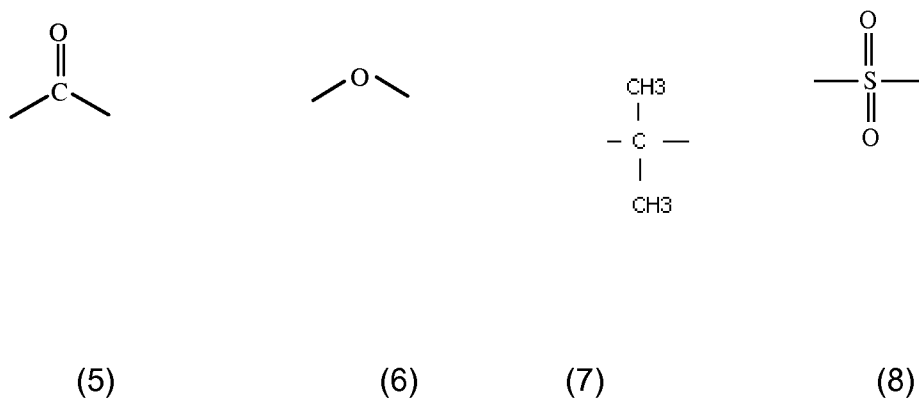


Each R is a molecular segment independently selected from the group consisting of formula (1), formula (2), formula (3), and formula (4):



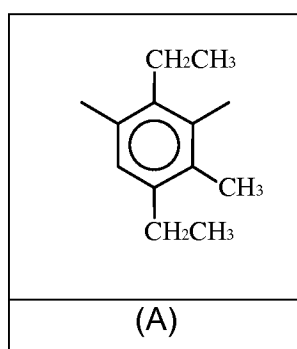
By independently selected, we mean that each R need not be the same, however, typically it is.

Each Z is a molecular segment independently selected from the group consisting of formula (5), formula (6), formula (7), and formula (8).

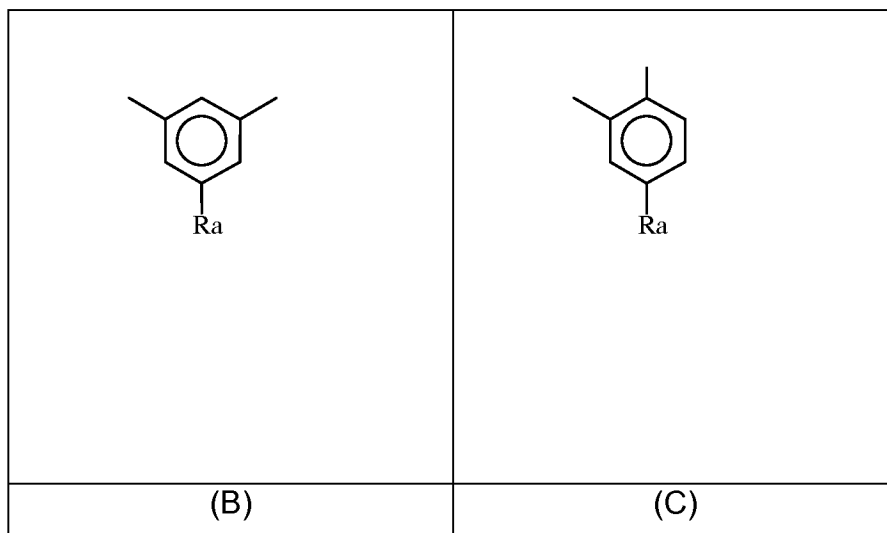


- 5 By independently selected, we mean that each Z need not be the same, however, typically it is.

At least some of the diamine-derived units are the diamine-derived moiety of formula (A):



At least some of the diamine-derived units are the diamine-derived moiety of formula (B) or formula (C):



Each R_a is a straight or branched C_1 to C_6 alkyl group having a terminal carboxylic acid group. Typically, R_a is a $-COOH$ group. The diamine-derived units of formula (B) where R_a is a $-COOH$ group is conventionally termed DABA. Typically, the ratio of the diamine-derived moieties of formula (A) to that of either formula (B) or formula (C) ranges from about 3:2 to about 2:3.

The dianhydride-derived moiety of formula (I) is conventionally termed 6FDA and is derived from 2,2'-bis(3,4-dicarboxyphenyl hexafluoropropane). We believe that the 6FDA molecular segment exhibits many attractive properties. Polyimides incorporating 6FDA have limited rotational mobility of the polymer chain. This results in polyimides with both hindered packing density and high glass transition. These molecular properties result in membranes with relatively high permeance for a given selectivity as well as high temperature capability.

The diamine-derived moiety of formula (A) is conventionally termed DETDA. The DETDA molecular segment includes an ethyl group that is ortho to the phenyl to nitrogen bond of the imide linkage of the polyimide. We believe that this relatively bulky group sterically inhibits rotation of the polymer around that bond. Without being bound by any particular theory, we believe that this sterically inhibited rotation renders the polyimide more rigid. The relatively bulky group also inhibits close chain packing. These molecular properties can further enhance the desired membrane properties of high permeance and temperature capability.

The R_a substituent of the diamine-derived moiety of formula (B) or formula (C) allows cross-linking between the R_a substituents on adjacent polymer chains

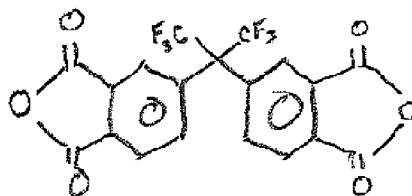
via the esterification route. When R_a is a $-\text{COOH}$ group, the diamine-derived moiety of formula (B) is conventionally termed DABA. Cross-linking can be used to achieve higher selectivity for higher separation efficiency and to maintain this high selectivity when exposed to plasticizing species in the gas mixture to be separated as well as conditions such as high temperature and high pressure.

In one embodiment, 100% of the dianhydride-derived units are the dianhydride-derived moiety of formula (I). Typically, R_a is a $-\text{COOH}$ group.

In another embodiment, less than 100% of the dianhydride-derived units are the dianhydride-derived moiety of formula (I) with the balance being the dianhydride-derived units are the dianhydride-derived moiety of formula (II). Typically, R_a is a $-\text{COOH}$ group. Also typically, R is the molecular segment of either formula (2) or (3) where Z is the molecular segment of formula (5).

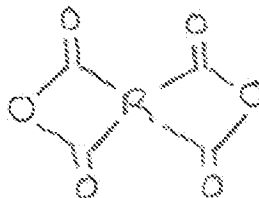
The polyimide polymer or copolymer may be synthesized by reacting, in any one of a wide variety of known polyimide synthesis methods, stoichiometric amounts of one or more dianhydrides and one or more diamines to form the intermediate poly(amic acid) followed by removal of water to form the polyimide by ring-closing. The skilled artisan will understand that a stoichiometric amount of a dianhydride reacted with a stoichiometric amount of a mixture of diamines will result in a random copolymer. Alternatively, a block copolymer of the dianhydride and one or more diamines may be synthesized according to known methods in which case the diamines are not initially in admixture. The skilled artisan will similarly understand that a stoichiometric amount of a mixture of dianhydrides reacted with a stoichiometric amount of a diamine will also form a random copolymer and that a block copolymer may alternatively be synthesized according to known methods in which case the dianhydrides are not initially in admixture. Finally, the skilled artisan will further understand that a stoichiometric amount of a mixture of dianhydrides reacted with a stoichiometric amount of a mixture of diamines will result in a random polymer and that a block copolymer may alternatively be synthesized according to known methods in which case the dianhydrides are not initially in admixture and the diamines are not initially in admixture.

At least some of the dianhydride is 2,2'-bis(3,4-dicarboxyphenyl hexafluoropropane) which conventionally termed 6FDA and whose molecular structure is shown by formula I':



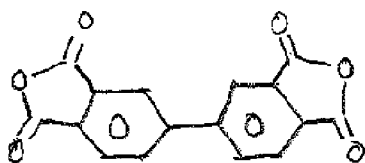
(I').

When less than 100% of the dianhydride is 6FDA, the balance of the dianhydride is shown by formula II':

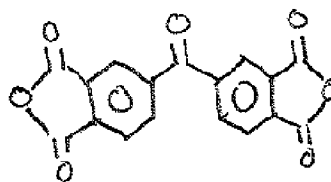


(II').

Each R is as defined above. Typically, the balance of the dianhydride is 4,4'-bipthalic dianhydride (BPDA) or benzophenone-3,3',4,4'-tetracarboxylic dianhydride (BTDA):

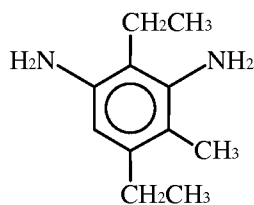


BPDA



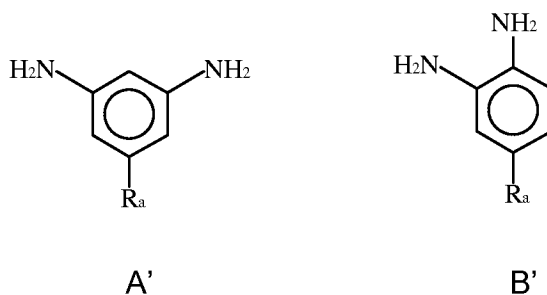
BTDA

Some of the diamines are 2,5-diethyl-6-methyl-1,3-diamino benzene (DETDA):

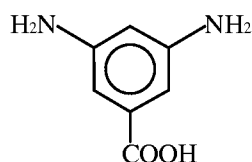


DETDA

The balance of the diamines are the diamine of formula A' or formula B', where R_a is as defined above.



When the balance of the diamines are the diamine of formula (A) and R_a is a –
COOH, the balance of the diamines are conventionally termed 3,5-
5 diaminobenzoic acid (DABA).



DABA

When the balance of the diamines are the diamine of formula (B) and R_a is a –
COOH, the balance of the diamines are conventionally termed 3,4-
10 diaminobenzoic acid.

More typical polyimides include: 6FDA/DETDA:DABA (polymerized using
6FDA and a mixture of DETDA and DABA); 6FDA:BPDA/DETDA:DABA
(polymerized using a mixture of 6FDA and BPDA and a mixture of DETDA and
DABA), and 6FDA:BTDA/DETDA:DABA (polymerized with a mixture of 6FDA and
15 BTDA and a mixture of DETDA and DABA).

While the membrane may have any configuration known in the field of gas
separation, typically it is formed as a plurality of hollow fibers.

The polyimide is optionally dried and later dissolved in a suitable solvent to
provide a precursor solution (known as a spin dope in the case of hollow fiber
20 spinning). The drying may be carried out in, for example, a drying vacuum oven,
typically at a temperature ranging from 110-150 °C for at least 6 hours (and as
much as 6-12 hours). Drying is considered to be completed once a steady weight
is achieved. Other known methods of drying such as heating in an inert gas purge
may additionally or alternatively be employed.

Dissolution in, and homogenous distribution of, the polyimide in the solvent may be enhanced by mixing with any known mixing device, including rollers, stirrer bars, and impellers. In the case of a hollow fiber membrane, the precursor solution may be mixed for 6 hours to 30 days (optionally 3-10 days or even 3-7 days).

The concentration of the polyimide in the precursor solution is typically driven by the configuration of the polymeric membrane. For example, a concentration ranging from 15-35 wt % (or optionally 18-30 wt % or even 22-28 wt %) is suitable for spinning hollow fibers.

Suitable solvents may include, for example, dichloromethane, tetrahydrofuran (THF), N-methyl-2-pyrrolidone (NMP), and others in which the resin is substantially soluble, and combinations thereof. For purposes herein, "substantially soluble" means that at least 98 wt % of the polymer in the solution is solubilized in the solvent. Typical solvents include N-methylpyrrolidone (NMP), N,N-dimethylacetamide (DMAC), N,N-dimethylformamide (DMF), dimethyl sulfoxide (DMSO), *gamma*-butyrolactone (BLO), dichloromethane, THF, glycol ethers or esters, and mixtures thereof.

In a membrane configured as hollow fibers, the hollow fibers may be spun by any conventional method. A typical procedure for producing hollow fibers of this invention can be broadly outlined as follows. A bore fluid is fed through an inner annular channel of spinneret designed to form a cylindrical fluid stream positioned concentrically within the fibers during extrusion of the fibers. A number of different designs for hollow fiber extrusion spinnerets known in the art may be used. Suitable embodiments of hollow-fiber spinneret designs are disclosed in US 4,127,625 and US 5,799,960, the entire disclosures of which are hereby incorporated by reference. The bore fluid is preferably water, but a mixture of water and an organic solvent (for example NMP) may be used as well. The precursor solution is fed through an outer annular channel of the spinneret so that it surrounds the bore fluid to form a nascent polymeric hollow fiber.

The diameter of the eventual solid polymeric fiber is partly a function of the size of the hollow fiber spinnerets. The outside diameter of the spinneret can be from about 400 μm to about 2000 μm , with bore solution capillary-pin outside diameter from 200 μm to 1000 μm . The inside diameter of the bore solution capillary is determined by the manufacturing limits for the specific outside

diameter of the pin.

The temperature of the solution during delivery to the spinneret and during spinning of the hollow fiber depends on various factors including the desired viscosity of the dispersion within the spinneret and the desired fiber properties. At higher temperature, viscosity of the dispersion will be lower, which may facilitate extrusion. At higher spinneret temperature, solvent evaporation from the surface of the nascent fiber will be higher, which will impact the degree of asymmetry or anisotropy of the fiber wall. In general, the temperature is adjusted to maintain the desired viscosity of the dispersion and the fiber wall asymmetry. Typically, the temperature is from about 20 °C to about 100 °C, preferably from about 20 °C to about 60 °C.

Upon extrusion from the spinneret, the nascent polymeric hollow fiber is passed through an air gap and immersed in a suitable liquid coagulant bath to facilitate phase inversion of the dissolved polyimide and solidification of the fiber structure. The coagulant constitutes a non-solvent or a poor solvent for the polymer while at the same time a good solvent for the solvent within the dispersion. As a result, the solvent for the polymer is extracted from the nascent fiber causing the polymer to solidify as it is drawn through the quench bath. Suitable liquid coagulants include water (with or without a water-soluble salt) and/or alcohol with or without other organic solvents. Typically, the liquid coagulant is water.

The solidified fiber is then withdrawn from the coagulant and wound onto a rotating take-up roll, drum, spool, bobbin or other suitable conventional collection device. Before or after collection, the fiber may optionally be washed to remove any residual solvent. After collection, the fiber may optionally be dried to remove any remaining volatile material.

Other exemplary conventional processes for producing polymeric hollow fibers are disclosed in US 5,015,270, US 5,102,600, and Clausi, et al., (Formation of Defect-free Polyimide, Hollow Fiber Membranes for Gas Separations, Journal of Membrane Science, 167 (2000) 79-89), the entire disclosures of which are hereby incorporated by reference herein.

The completed fibers have an outer diameter that typically ranges from about 150-550 μm (optionally 200-300 μm) and an inner diameter that typically ranges from 75-275 μm (optionally 100-150 μm). In some cases unusually thin

walls (for example, thicknesses less than 30 μm) may be desirable to maximize productivity while maintaining desirable durability.

The hollow fibers may be used to make a gas separation membrane according to any number of well-known methods of manufacturing gas separation membranes based upon polymeric hollow fibers. Alternatively, the hollow fibers may be used to make a carbon molecular sieve (CMS) membrane after pyrolysis to substantially carbonize the hollow fibers according to any number of well-known methods of manufacturing CMS membranes.

While the invention has been described in conjunction with specific embodiments thereof, it is evident that many alternatives, modifications, and variations will be apparent to those skilled in the art in light of the foregoing description. Accordingly, it is intended to embrace all such alternatives, modifications, and variations as fall within the spirit and broad scope of the appended claims. The present invention may suitably comprise, consist or consist essentially of the elements disclosed and may be practiced in the absence of an element not disclosed. Furthermore, if there is language referring to order, such as first and second, it should be understood in an exemplary sense and not in a limiting sense. For example, it can be recognized by those skilled in the art that certain steps can be combined into a single step.

The singular forms "a", "an" and "the" include plural referents, unless the context clearly dictates otherwise.

"Comprising" in a claim is an open transitional term which means the subsequently identified claim elements are a nonexclusive listing i.e. anything else may be additionally included and remain within the scope of "comprising."

"Comprising" is defined herein as necessarily encompassing the more limited transitional terms "consisting essentially of" and "consisting of"; "comprising" may therefore be replaced by "consisting essentially of" or "consisting of" and remain within the expressly defined scope of "comprising".

"Providing" in a claim is defined to mean furnishing, supplying, making available, or preparing something. The step may be performed by any actor in the absence of express language in the claim to the contrary.

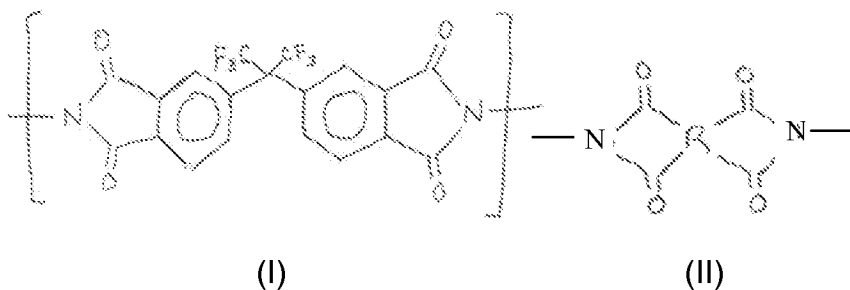
Optional or optionally means that the subsequently described event or circumstances may or may not occur. The description includes instances where the event or circumstance occurs and instances where it does not occur.

Ranges may be expressed herein as from about one particular value, and/or to about another particular value. When such a range is expressed, it is to be understood that another embodiment is from the one particular value and/or to the other particular value, along with all combinations within said range.

- 5 All references identified herein are each hereby incorporated by reference into this application in their entireties, as well as for the specific information for which each is cited.

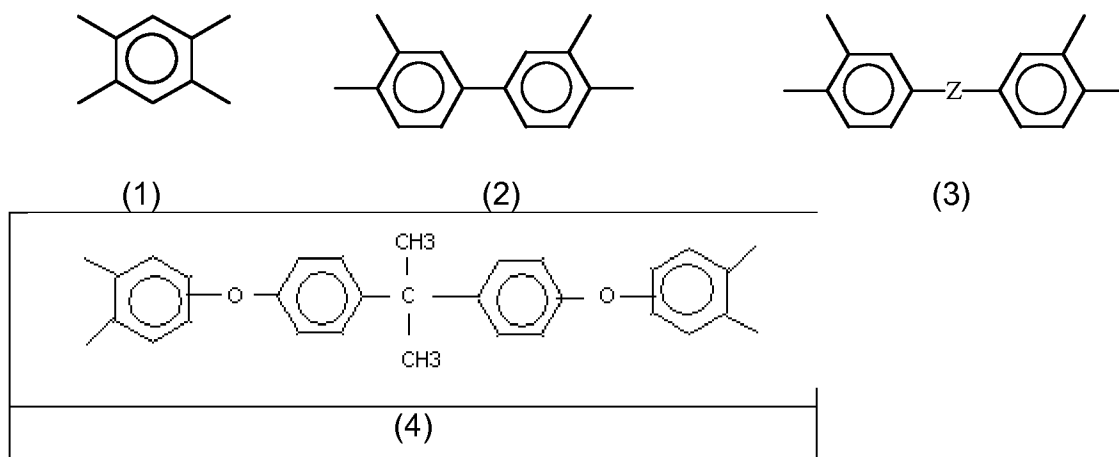
What is claimed is:

1. A membrane comprising a polyimide polymer or copolymer having repeating units of dianhydride-derived units and diamine-derived units, at least
 5 some of, and as much as 100% of, the dianhydride-derived units being the dianhydride-derived moiety of formula (I) with a balance of the dianhydride-derived units, if any, being the dianhydride-derived moiety of formula (II)



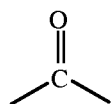
10 wherein:

each R is a molecular segment independently selected from the group consisting of formula (1), formula (2), formula (3), and formula (4)

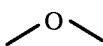


;

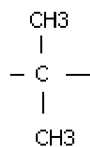
15 each Z is a molecular segment independently selected from the group consisting of formula (5), formula (6), formula (7), and formula (8)



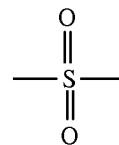
(5)



(6)



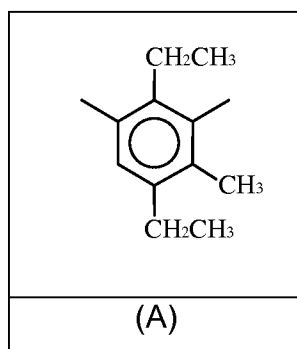
(7)



(8)

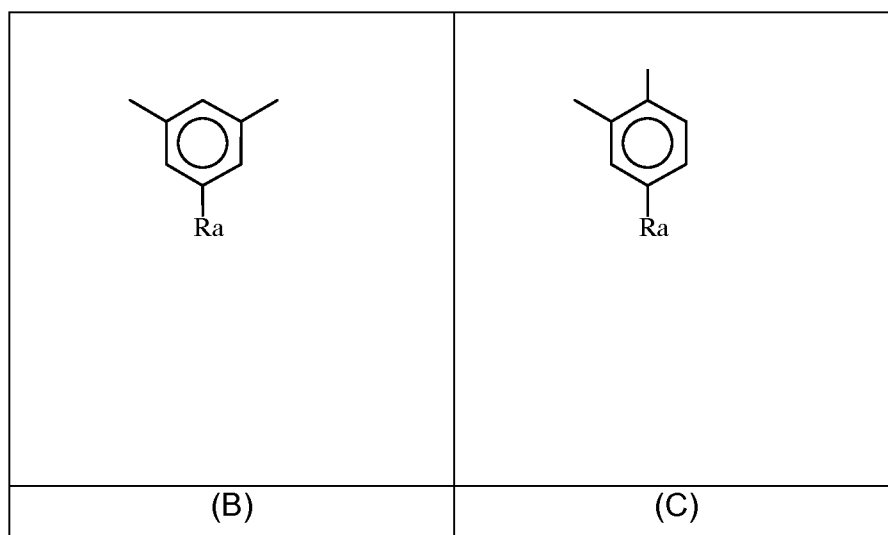
;

at least some of the diamine-derived units are the diamine-derived moiety of formula (A)



5

at least some of the diamine-derived units are the diamine-derived moiety of formula (B) or formula (C):



; and

each R_a is a straight or branched C₁ to C₆ alkyl group having a terminal carboxylic acid group.

2. The membrane of claim 1, wherein 100% of the dianhydride-derived
5 units are the dianhydride-derived moiety of formula (I).

3. The membrane of claim 2, wherein R_a is a –COOH group.

4. The membrane of claim 1, wherein less than 100% of the
10 dianhydride-derived units are the dianhydride-derived moiety of formula (I) and the balance of the dianhydride-derived units are the dianhydride-derived moiety of formula (II).

5. The membrane of claim 4, wherein R_a is a –COOH group.
15

6. The membrane of claim 4, wherein R is the molecular segment of
formula (2).

7. The membrane of claim 6, wherein Z is the molecular segment of
20 formula (5).

8. The membrane of claim 7, wherein R_a is a –COOH group.

9. The membrane of claim 4, wherein R is the molecular segment of
25 formula (3).

10. The membrane of claim 9, wherein Z is the molecular segment of
formula (5).

11. The membrane of claim 10, wherein R_a is a $-\text{COOH}$ group.

12. The membrane of claim 1, wherein a ratio of the diamine-derived
5 moieties of formula (A) to that of either formula (B) or formula (C) ranges from about 3:2 to about 2:3.

13. A method of producing a hollow fiber membrane, comprising the steps of:

10 preparing a solution or suspension of a polyimide polymer or copolymer binder and a solvent for the binder;

providing a spinneret adapted and configured to continuously extrude one or more nascent hollow fibers, the spinneret having an inner annular channel disposed concentrically within an outer annular channel;

15 feeding a bore fluid through the inner annular channel to form a cylindrical fluid stream positioned concentrically within the fiber;

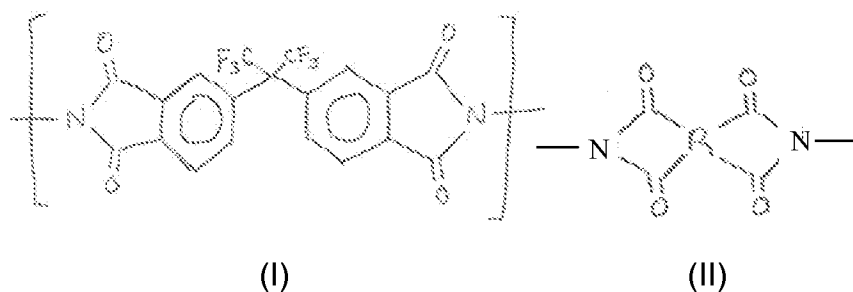
feeding the solution or suspension through the outer annular channel so that it surrounds the cylindrical fluid stream to form a nascent hollow fiber;

passing the nascent hollow fiber from the spinneret through an air gap;

20 immersing the nascent hollow fiber in a liquid coagulant to facilitate phase inversion; and

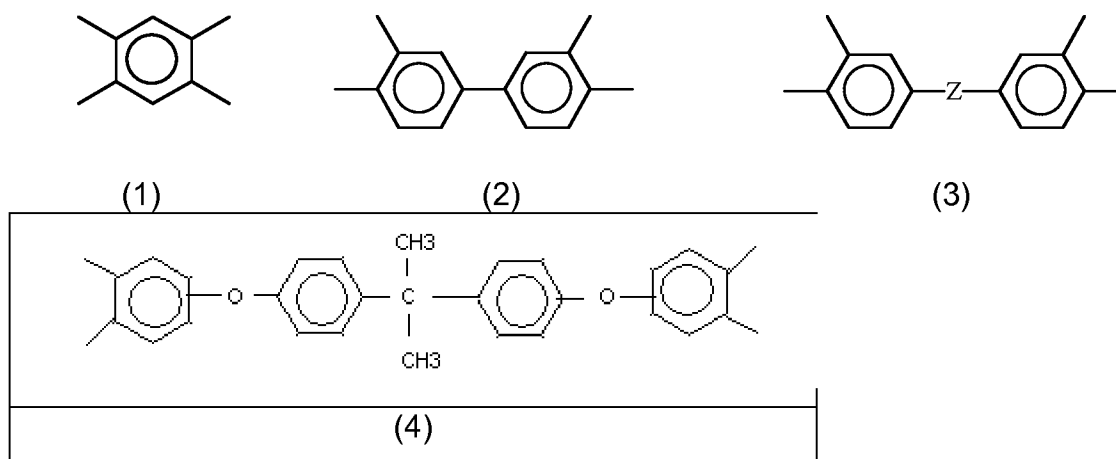
removing the fiber from the coagulant, wherein:

the polyimide polymer or copolymer has repeating units of dianhydride-derived units and diamine-derived units, at least some of, and as much as 100%
25 of, the dianhydride-derived units being the dianhydride-derived moiety of formula (I) with a balance of the dianhydride-derived units, if any, being the dianhydride-derived moiety of formula (II)



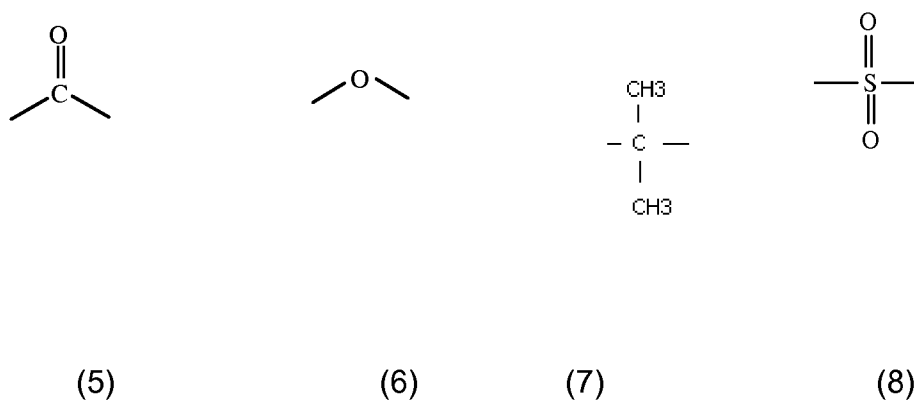
wherein:

each R is a molecular segment independently selected from the group
 5 consisting of formula (1), formula (2), formula (3), and formula (4)



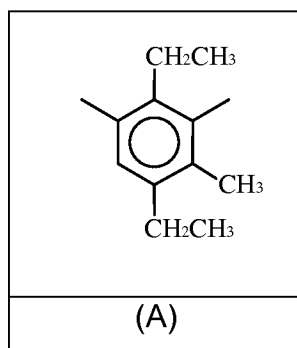
;

each Z is a molecular segment independently selected from the group
 consisting of formula (5), formula (6), formula (7), and formula (8)

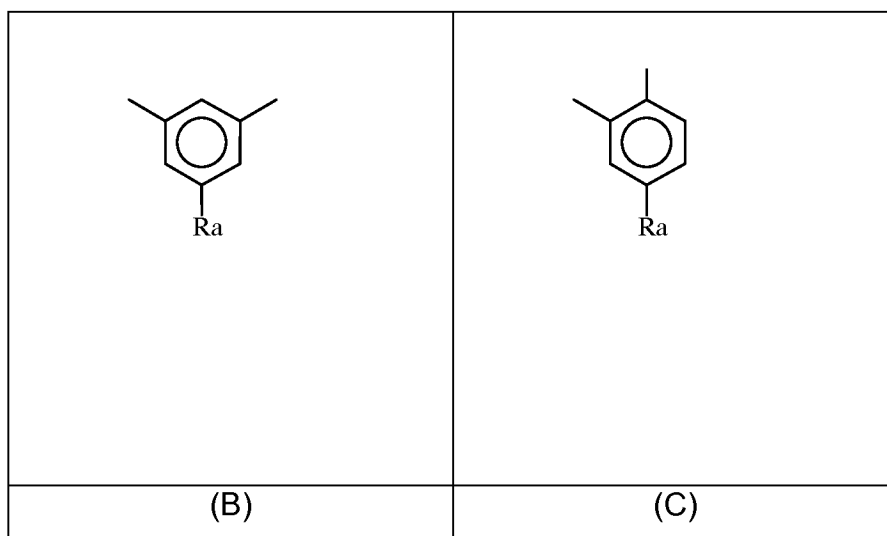


;

10 at least some of the diamine-derived units are the diamine-derived moiety
 of formula (A)



at least some of the diamine-derived units are the diamine-derived moiety of formula (B) or formula (C):



; and

5 each R_a is a straight or branched C_1 to C_6 alkyl group having a terminal carboxylic acid group.

INTERNATIONAL SEARCH REPORT

International application No
PCT/US2014/071492

A. CLASSIFICATION OF SUBJECT MATTER
INV. B01D69/08 B01D71/64
ADD.

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)
B01D

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practicable, search terms used)

EPO-Internal, WPI Data

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
Y	<p>CANGHAI MA ET AL: "High-performance ester-crosslinked hollow fiber membranes for natural gas separations", JOURNAL OF MEMBRANE SCIENCE, vol. 428, 1 February 2013 (2013-02-01), pages 251-259, XP055181661, ISSN: 0376-7388, DOI: 10.1016/j.memsci.2012.10.024 the whole document</p> <p>-----</p> <p>-/--</p>	1-13



Further documents are listed in the continuation of Box C.



See patent family annex.

* Special categories of cited documents :

"A" document defining the general state of the art which is not considered to be of particular relevance

"E" earlier application or patent but published on or after the international filing date

"L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)

"O" document referring to an oral disclosure, use, exhibition or other means

"P" document published prior to the international filing date but later than the priority date claimed

"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention

"X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone

"Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art

"&" document member of the same patent family

Date of the actual completion of the international search

8 April 2015

Date of mailing of the international search report

23/04/2015

Name and mailing address of the ISA/

European Patent Office, P.B. 5818 Patentlaan 2
NL - 2280 HV Rijswijk
Tel. (+31-70) 340-2040,
Fax: (+31-70) 340-3016

Authorized officer

Kukolka, Florian

INTERNATIONAL SEARCH REPORT

International application No
PCT/US2014/071492

C(Continuation). DOCUMENTS CONSIDERED TO BE RELEVANT		
Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
Y	GRUBB T L ET AL: "Highly soluble polyimides from sterically hindered diamines", POLYMER, ELSEVIER SCIENCE PUBLISHERS B.V, GB, vol. 40, no. 15, 1 July 1999 (1999-07-01), pages 4279-4288, XP027262490, ISSN: 0032-3861 [retrieved on 1999-07-01] figure 1	1-13
A	----- YUAN ZHANG ET AL: "Current status and development of membranes for CO ₂ /CH ₄ separation: A review", INTERNATIONAL JOURNAL OF GREENHOUSE GAS CONTROL, vol. 12, 20 December 2012 (2012-12-20), pages 84-107, XP055181652, ISSN: 1750-5836, DOI: 10.1016/j.ijggc.2012.10.009 the whole document	1-13
A	----- QIU WULIN ET AL: "Gas separation performance of 6FDA-based polyimides with different chemical structures", POLYMER, vol. 54, no. 22, 19 September 2013 (2013-09-19), pages 6226-6235, XP028734520, ISSN: 0032-3861, DOI: 10.1016/J.POLYMER.2013.09.007 the whole document	1-13
A	----- US 2003/140789 A1 (KOROS WILLIAM J [US] ET AL) 31 July 2003 (2003-07-31) claims 1-25 -----	1-13

INTERNATIONAL SEARCH REPORT

Information on patent family members

International application No

PCT/US2014/071492

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
US 2003140789 A1	31-07-2003	AU 2002360746 A1	09-07-2003
		US 2003131731 A1	17-07-2003
		US 2003140789 A1	31-07-2003
		WO 03053548 A1	03-07-2003
