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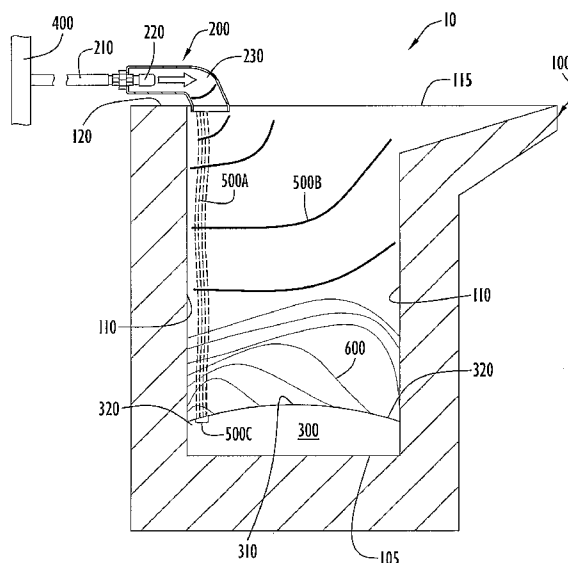
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(54) **Title:** VAPOR-REINFORCED EXPANDING VOLUME OF GAS TO MINIMIZE THE CONTAMINATION OF PRODUCTS TREATED IN A MELTING FURNACE



(57) **Abstract:** Systems and corresponding methods are described herein that provide an effective inert blanket over a metal surface (hot solid (charge) metal or molten metal) in a container such as an induction furnace. The system includes a container of metal and a system configured to delivery biphasic inert cryogen toward the metal. The delivery system may include a lance disposed at the top of the container. The lance has a hood that directs both a flow of liquid cryogen and a flow of vaporous gas toward the metal surface. The liquid cryogen contacts the metal surface, generating a volume of expanding gas over the metal surface. The vaporous cryogen creates a reinforcing vapor that slows the expansion rate of the expanding gas, localizing the expanding gas over the metal surface.



WO 2008/023229 A1

VAPOR-REINFORCED EXPANDING VOLUME OF GAS TO MINIMIZE THE CONTAMINATION OF PRODUCTS TREATED IN A MELTING FURNACE

Background

Field

This invention relates to the minimizing of contamination of molten metal during processing.

Related Art

In the metal casting industry, metals (ferrous or non-ferrous) are melted in a furnace, and then poured into molds to solidify into castings. In the foundry melting operations, metals are commonly melted in electric induction furnaces. It is often advantageous to melt and transport the metals without exposure to atmospheric air to minimize oxidation of the metal (including its alloying components), which not only increases yield and alloy recovery efficiency, but also reduces formation of metallic oxides, which can cause casting defects (inclusions), reducing the quality of the finished product. Molten metal, moreover, has a tendency to absorb gases (chiefly oxygen and hydrogen) from the atmosphere (ambient air), which cause gas-related casting defects such as porosity.

Various processes are utilized to prevent exposure of the metal to the atmospheric air, including vacuum treatment and inerting with a gas or a liquid. In vacuum treatment, a fluid-tight furnace chamber is vacuum evacuated of substantially all ambient oxygen prior to heating the metal. This process, however, requires a special vacuum furnace and is generally only suitable for small batch processes. In addition, the use of a vacuum furnace also results in the need for a substantially long cooling period, which lowers plant productivity.

With gas inerting, a continuous flow of inert gas is injected into the furnace chamber. This creates a blanket of inert gas that purges ambient oxygen from the chamber, as well as prevents the ambient air from entering the

chamber. This process, however, requires an extraordinarily large volume of gas to be used during the process, even with a substantially fluid-tight chamber. The process, moreover, fails to keep the concentration of residual oxygen low enough to prevent the formation of an oxide layer on most metal products. Hot thermal updrafts from within the hot furnace are continually pushing the incoming cold inert gas up and away from the metal surface. Thus, as the hot air and gases rise, the induced draft continually pulls fresh cold air toward the furnace. The injected inert gas will also entrain ambient air along with it as it is injected into the furnace. Because of these effects, it is difficult, if not impossible, for gas inerting techniques to provide a true inert (0% O₂) atmosphere directly at the surface of the metal.

With liquid inerting, a liquid cryogen (typically N₂ or Ar) covers the entire exposed surface of the metal (i.e., hot solid metal or molten metal). Since the liquid cryogen has higher density than its gas phase and air, it is much less likely to be pushed up and away from the melt surface by the thermal updrafts. After contacting the metal surface, within a short time, the liquid vaporizes into a gas. As the cryogen boils from liquid to gas, it expands volumetrically by a factor of about 600 – 900 times as it rises. As a result, the expansion pushes ambient air away from the surface of the metal, inhibiting oxidation. One drawback of liquid inerting is the difficulty of efficiently delivering the liquid cryogen to the furnace interior in a liquid state. The liquefied gas is extremely cold. In the storage tank and distribution piping, the liquid inert gas is continually absorbing heat from the surroundings, boiling some of the liquid to vapor inside the storage tank and distribution piping. This vapor must be vented before the liquid is injected into the chamber, otherwise flow sputtering and surging results (caused by the tendency of the gas to choke the flow of liquid in the delivery pipes). As a result, a significant portion of the cryogen supply is lost due to boiling.

Thus, there still remains a need in the art to achieve low residual oxygen concentrations through a purging process without losing substantial volumes of inert gases.

Summary

Systems and corresponding methods are described herein that provide an effective inert blanket over a metal surface in a container such as an induction furnace, tundish, etc. The system includes a container of metal (e.g., hot solid (charge) metal or molten metal) and a system configured to deliver biphasic inert cryogen toward the metal. The delivery system may include a lance disposed proximate the top of the container. The lance includes a hood that directs both a flow of liquid cryogen and a flow of vaporous cryogen toward the metal surface. The liquid cryogen travels to the metal surface, where it vaporizes to generate a volume of expanding gas. The vaporous cryogen, moreover, is directed downward, toward the expanding gas. The vaporous cryogen reinforces expanding gas, slowing its expansion rate to maintain the expanding gas over the metal surface. Thus, the liquid and vaporous gas work in tandem to inhibit the oxidation of the metal.

The system can include a number of different features, including any one or combination of the following features:

- an open vessel for containing molten metal, the vessel including a bottom wall, a side wall, and an opening;
- an inert cryogen source, the inert cryogen including a liquid flow component and a vaporous flow component;
- a delivery system disposed proximate the opening, the delivery system comprising (1) a lance including an inlet and a outlet, the inlet connected to the inert cryogen source and/or (2) a hood coupled to the outlet end of the lance, wherein the hood directs the components of the inert cryogen toward the molten metal;
- a hood configured to direct the liquid component of the inert cryogen toward the bottom wall of the vessel such that the liquid component contacts the molten metal to form an expanding volume of gas having a rate of expansion;
- a hood further configured to direct the vaporous component toward the molten metal to inhibit the rate of expansion of the expanding volume of gas;

a hood having a curved housing with an inlet and an outlet located downstream from the outlet;

a hood positioned such that the outlet of the hood is generally coplanar with or below the opening of the vessel;

a delivery system operable to generate a flow rate of inert cryogen in the range of about 0.002 lb/in² to about 0.005 lb/in², based upon the surface area of the molten metal;

diffuser operable to separate the liquid flow component from the vaporous flow component; and

a hood having a degree of curvature of about 0° to about 90°.

A method of providing a vapor blanket over a material processed within a container is also described herein. The method can include a number of different features, including any one or combination of the following features:

forming molten metal within a container, the molten metal having an exposed surface defining a surface area;

generating a biphasic inert cryogen, wherein the inert cryogen comprises a liquid flow component and a vaporous flow component;

directing the liquid flow component into contact with the molten metal to generate an expanding gaseous volume having a rate of expansion; and

directing the vaporous flow component into the container to inhibit the rate of expansion of the gaseous volume;

directing a flow of biphasic inert cryogen at a flow rate effective to generate the expanding gaseous volume that is substantially coextensive with the exposed surface of the molten metal;

determining flow rate based upon the surface area of the molten metal;

providing a flow rate in the range of about 0.002 lb/in² to about 0.005 lb/in², based upon the surface area of the molten metal;

providing a molten metal possessing a generally meniscoid shape with a raised center meniscus portion and a lower edge meniscus portion, and directing the liquid flow component into contact with the lower meniscoid portion;

maintaining the flow rate to localize the liquid flow component within a portion of the molten metal exposed surface;

providing a container including a bottom wall, a side wall, and an opening, and directing the liquid flow component proximate the side wall such that the liquid flow component contacts the molten metal at a point proximate the side wall;

directing a liquid inert cryogen from a source through a diffuser to separate the liquid flow component from the vaporous flow component; and maintaining a flow rate of the inert cryogen such that liquid flow is localized within an area smaller than the molten metal exposed surface.

The above and still further objects, features and advantages of the systems and methods described herein will become apparent upon consideration of the following detailed description of specific embodiments thereof, particularly when taken in conjunction with the accompanying drawings, wherein like reference numerals designate like components.

Brief Description of the Drawings

FIG. 1 depicts cross-sectional view of an exemplary embodiment of a container with a heated load of metal and a delivery system for a biphasic inert cryogen in accordance with an embodiment of the invention.

FIG. 2 is a close-up view of the delivery system shown in FIG. 1.

Description of Preferred Embodiments

The present invention provides a system and process wherein a vapor reinforced expanding volume of inert gas (e.g., argon, nitrogen, or carbon dioxide) is developed and maintained over the surface of metal (e.g., molten metal and/or heated metal charge) in a container such as a melting furnace or a transfer system (a ladle, a launder, etc.). The reinforced expanding volume of inert gas may be generated and maintained from a vaporizing volume of liquid cryogen situated against one or more sides of the inside surface of the container. The volumes of expanding gas may be maintained by a continuous stream of liquid cryogen replenishing the vaporizing volume of liquid cryogen from a lance system at the top of the furnace.

FIG. 1 shows a system **10** in accordance with an embodiment of the invention. As illustrated, the system **10** includes a container **100** and a biphasic cryogen delivery system **200**. The container **100** includes a bottom wall **105**, a side wall **110**, and an opening **115** defined by a rim **120**. The container **100** houses metal **300** (e.g., molten metal and/or heated charge material). By way of example, the container **100** may be a molten metal bath, an induction furnace, or a metal containment and/or transfer system such as a ladle, launder, etc. Convection movements and/or surface tension present in the molten metal form a converging meniscus with a raised central portion **310** and lower edge portion **320** disposed along the side wall **110** of the container **100**.

The biphasic cryogen delivery system **200** distributes liquid and vaporous inert cryogen into the container **100**. The system **200** may include a lance **210** disposed at the top of the container **100**. The lance **210** may communicate with an inert liquid cryogen source **400** (e.g., a storage vessel). The inert liquid cryogen may include, but is not limited to, argon, nitrogen, or carbon dioxide.

As discussed above, in traveling from the source **400** to the container **100**, the inert liquid cryogen absorbs heat, forming a vaporous/gaseous component. Consequently, a diffuser **220** may be coupled to the lance **210** to separate the vaporous component from the liquid component (i.e., the vaporous cryogen from the liquid cryogen). The diffuser **220** may include, for example, a sintered 10 – 80 μ level plug disposed at the discharge end of the lance **210**. The diffuser **220** is housed within a shroud or hood **230** configured to channel the liquid and gas components exiting the diffuser, directing them into the container **100**. Specifically, the hood **230** is shaped to direct the biphasic flow or cryogen (i.e., the flow of liquid cryogen **500A** and the flow of vaporous cryogen **500B**) toward the surface of the metal **300**.

FIG. 2 illustrates a close-up view of the hood **230** illustrated in FIG. 1. In the embodiment illustrated, the hood **230** includes an inlet end **235**, a first portion

237, a second portion **239**, and an outlet end **240**. The hood **230** curves downward, away from the longitudinal axis of the hood (indicated by **X**), creating a first or outer bend **245** and a second or inner bend **250**. The degree of curvature may include, but is not limited to, downward curvatures in the range of about 0° (where the outlet **240** is generally perpendicular to the axis **X**) to about 90° (wherein the outlet **240** is generally parallel to the axis **X**). The dimensions of the hood may be any suitable for its described purpose. By way of example, the hood **230** may have an overall length of approximately 4 – 6 inches (10.16 cm – 15.24 cm). By way of specific example, the first portion **237** (extending from the inlet **235** to the bend **245/250**) may be about 3 – 5 inches (7.62 cm – 12.7 cm) (e.g., 4 inches (10.16 cm)), while the second portion (extending from the bend **245/250** to the outlet **240**) may be about 0.5 – 3 inches (1.27 cm – 7.62 cm) (e.g., about 1.5 inches (3.81 cm)). The diameter of the hood channel (indicated as **D**) may be about 0.5 inches to 2 inches (1.27 cm – 5.08 cm) (e.g., 1 inch (3.54 cm)). Preferably, the diameter **D** of the channel is substantially continuous from the inlet **235** to the outlet **240**. The material forming the hood includes, but is not limited to, stainless steel tubing.

The hood **230** is disposed oriented to introduce the liquid cryogen **500A** and vaporous cryogen **500B** into the container. For example, the hood **230** may be disposed at a point proximate the opening **115** of the container **100**. By way of specific example, the outlet end **240** may be generally coplanar with the opening **115** of the container **100**, or may be positioned slightly below the opening **115** such that it protrudes into the container interior. The hood **230**, moreover, may be oriented on the container such that the inner bend **250** of the hood is positioned adjacent the sidewall **110**.

With this configuration, the liquid cryogen **500A** is directed along/adjacent the side wall **110** of the container **100**, permitting the liquid cryogen to reach the metal **300** and create a localized pool or volume **500C** of liquid cryogen along the lower meniscus portion **320**. This is contrary to conventional liquid cryogen delivery systems, which direct a blanket of liquid over the entire metal

surface. Instead, the delivery system **200** of the present invention controls parameters to cause the liquid cryogen **500A** to become localized on the metal **300**. That is, the liquid cryogen **500A** covers only a portion of the metal surface, localizing the liquid cryogen within an area generally adjacent the side wall **110** of the container **100**.

As noted above, the pool **500C** of liquid cryogen is formed proximate the side wall **110** of the container. It is more effective to deliver the liquid cryogen **500A** down the side wall **110** of the container (to the lower portion **320** of the meniscus) to maximize the cryogen delivered to the meniscus site, as well as to create a pool **500C** of liquid cryogen at the lowest elevation within the metal environment (e.g., the lowest level of a furnace). In contrast, delivering the liquid cryogen **500A** to the upper portion **310** of the meniscus would inhibit the amount of cryogen actually delivered to the lower portion **320** of the meniscus (along the side wall **110**) because the cryogen **500C** would become trapped within or above the charge material (solid charge that will melt during the heat cycle). Also, placing the delivery system **200** along the side wall **110** of the container **100** (e.g., perpendicular to and adjacent the pouring spout of a furnace) provides an additional benefit of automatically facilitating inert protection of the pour of the metal into the transfer ladle, launder, tundish mold, etc.

Thus, with the above hood configuration, the flow of liquid cryogen **500A** forms a small volume **500C** of liquid cryogen on the surface of the metal **300**, adjacent the side wall **110**. Due to the heat generated by the surface of the molten metal **300**, as well as the heat radiated by the furnace walls **110**, the pool of liquid cryogen **500C** vaporizes, generating an expanding volume of inert gas **600** that expands across the entire exposed surface of the metal **300**. This expansion pushes ambient air away from the surface of the metal **300**, and infiltrates any charge material melting at the molten surface. This, in turn, provides a true inert atmosphere directly at the metal surface. The expansion rate of the gas **600** is generally dependant upon the type of inert gas utilized in forming the inert blanket (e.g., argon, nitrogen, or carbon

dioxide). By way of example, as the pool **500C** of liquid cryogen boils from liquid to gas, it may expand volumetrically by a factor of about 600 – 900 times as it rises. By way of specific example, argon expands up to 840 times the liquid volume while heating up from -302°F (-185°C) to room temperature.

The faster the expanding gas **600** expands, the quicker it escapes the container **100**, becoming lost into the surrounding environment. Such a loss not only reduces the effectiveness of the inert blanket, but also alters the surrounding atmosphere (e.g., exposing users to inert gas). To minimize and/or eliminate the rate of loss of the expanding volume of gas **600** from the container **100**, the delivery system **200** further directs a shroud of vaporous cryogen **500B** into the container, where it reinforces the expanding volume of inert gas **600** generated from the pool **500C** of cryogenic liquid, maintaining the expanding volume **600** proximate the exposed metal surface. Specifically, the hood **230** directs the vaporous cryogen **500B** toward the expanding gas **600**, reinforcing the expanding gas and inhibiting its rate of expansion and diffusion into the atmosphere above the container **100**. This alleviates a major drawback of conventional liquid inerting (discussed above), where a large portion of the inert cryogen is lost (e.g., when vented off to avoid lance sputtering).

The flow rate of the biphasic cryogen **500A**, **500B** from the source **400** should be effective to provide a continuous volume of expanding inert gas **600**, to maintain a localized pool **500C** of liquid cryogen on the surface of the metal **300** (i.e., to prevent the liquid cryogen **500A** from creating a pool **500C** that covers the entire surface of the metal **300**), and to maintain the flow reinforcing vaporous cryogen **500B** toward the metal surface. Preferably, the flow rate is determined as a function of the surface area of the metal **300**. This is contrary to the prior art processes, which calculate the flow rate utilizing the volume of the metal. Preferably, the continuous stream of cryogen is maintained at a flow rate of about 0.002 lb/in² to about 0.005 lb/in² (about 0.14 g/cm² to about 0.35 g/cm²) of the exposed metal surface area in the container **100**. This maintains a flow of cryogen at a rate effective to

generate a beneficial amount vaporous cryogen **500B** capable of reinforcing the expanding gas **600**. For example, the ratio of liquid cryogen **500A** to vaporous cryogen **500B** exiting the lance **210** may be about 99/1 to about 51/49, depending on the thermal quality of the cryogen distribution system and the working pressure of the cryogen supply tank. Flow rates above the preferred range tend to increase process costs, as well as lead to the "popping" of the metal **300** out of the container **100** due to volumetric and mechanical expansion of the cryogen **500C** as it transitions from a liquid to a vapor. This creates a hazardous situation for users in the area around the container **100**.

In operation, the hood **230** directs the liquid cryogen **500A** into the container **100**, causing the liquid cryogen to fall from the lance **210** adjacent to the side wall **110** and form the small volume (pool **500C**) of liquid cryogen on the surface of the metal **300**, adjacent the side wall of the container **100**. The liquid volume **500C** vaporizes, creating an expanding gas **600** that expands across the entire surface of the metal **300**. At the same time, the hood **230** directs the vaporous gas **500C** downward, toward the metal surface, inhibiting the expansion of the expanding gas **600**, maintaining the reinforced vapor near the surface of the metal **300**.

Conventional processes use either already expanded inert gas or an inert cryogenic liquid as a protective barrier for the molten metal and/or charge material in the container. The vapor reinforced expanding gas approach to inert blanketing is distinguished from such conventional processes in that it offers a higher level of safety for the furnace operator, an increased consistency and effect of the inert blanket, and an increase in inert gas efficiency or lower application cost. It delivers the entire inert product from the source **400** through the delivery system **200** to the internal atmosphere of the container **100** at a point above the melt interface.

This above-describe system is effective to guide the vaporous cryogen **500B** into the container **100**, providing for the complete utilization of the vaporous

cryogen, using it to reinforce the expanding gas **600**. In conventional systems, a 3 – 15% of the inert cryogen is wasted of the tip of a lance due to flash losses. The present system avoids these losses by completely utilizing the vaporous cryogen **500B**, directing it into the container **100** in a manner (at a speed and in an amount) effective to minimize and/or avoid flash losses.

While the invention has been described in detail and with reference to specific embodiments thereof, it will be apparent to one skilled in the art that various changes and modifications can be made therein without departing from the spirit and scope thereof. For example, the hood **230** may possess any dimensions and shape suitable for its described purpose (directing a biphasic flow into the container), and may be modified based on factors such as manufacturing cost, manufacturing method, and application site parameters. In addition, while the flow rate is dependent primarily upon the surface area of the metal **300** in the container **100** requiring protection by the expanding gas **600**, secondary factors may be used to determine the flow rate of the liquid cryogen, such as the reactivity of the alloy or metal being protected, the existence and strength of the ventilation system, and the quality requirements of the end user for the metal being produced. Furthermore, while a single source **400** of inert cryogen is illustrated, it is understood that multiple sources **400** may be connected to lance **210** to provide multiple types of inert cryogen to the container, including mixtures.

In addition, the systems and methods described can include any one or more suitable controllers and/or sensors to facilitate monitoring and control of various operational parameters during heating of the load in the furnace. One or more suitable sensors and related equipment can also be provided to measure and monitor the concentration of the gaseous species within the furnace, preferably at locations in the immediate vicinity of the load surface. Also, when the container **100** is an induction furnace, the induction furnace can include any suitable number and different types of sensors to monitor one or more of the temperature, pressure, flow rate and concentration of nitrogen and/or any other gaseous species within the furnace.

It is to be understood that terms such as “top”, “bottom”, “front”, “rear”, “side”, “height”, “length”, “width”, “upper”, “lower”, “interior”, “exterior”, and the like as may be used herein, merely describe points of reference and do not limit the present invention to any particular orientation or configuration. Thus, it is intended that the present invention covers the modifications and variations of this invention provided they come within the scope of the appended claims and their equivalents.

Claims:

1. A method for reducing the oxidation of molten metal, the method comprising:

(a) forming molten metal within a container, the molten metal having an exposed surface defining a surface area;

5 (b) generating a biphasic inert cryogen comprising a liquid flow component and a vaporous flow component;

(c) directing the liquid flow component into contact with the molten metal to generate an expanding gaseous volume having a rate of expansion; and

10 (d) directing the vaporous flow component into the container to inhibit the rate of expansion of the gaseous volume.

2. The method of claim 1, wherein (b) comprises (b.1) directing a flow of biphasic inert cryogen at a flow rate effective to generate the expanding
15 gaseous volume that is substantially coextensive with the exposed surface of the molten metal.

3. The method of claim 2, wherein the flow rate is dependent upon the surface area of the molten metal.

20

4. The method of claim 2, wherein the flow rate is in the range of about 0.002 lb/in² to about 0.005 lb/in², based upon the surface area of the molten metal.

5. The method of claim 1, wherein the molten metal possesses a generally
25 meniscoid shape with a raised center meniscus portion and a lower edge meniscus portion, and (c) comprises (c.1) directing the liquid flow component into contact with the lower meniscoid portion.

6. The method of claim 1, further comprising (e) maintaining the flow rate to
30 localize the liquid flow component within a portion of the molten metal exposed surface.

7. The method of claim 1, wherein:

the container comprises:

- a bottom wall,
- a side wall, and
- an opening; and

(c) further comprises directing the liquid flow component proximate the side wall such that the liquid flow component contacts the molten metal at a point proximate the side wall.

8. The method of claim 7, wherein the flow rate of the inert cryogen is maintained such that liquid flow is localized within an area smaller than the total surface area of the molten metal exposed surface.

9. The method of claim 8, wherein the flow rate is in the range of about 0.002 lb/in² to about 0.005 lb/in², based upon the surface area of the molten metal.

10. The method of claim 1, wherein:

the container comprises a side wall;

the molten metal possesses a generally meniscoid shape with a raised center meniscus portion and a lower edge meniscus portion located proximate the side wall;

(b) generating the biphasic inert cryogen comprises (b.1) directing a liquid inert cryogen from a source through a diffuser to separate the liquid flow component from the vaporous flow component; and

(c) directing the liquid flow component comprises (c.1) directing the liquid flow component along the side wall such that it contacts the lower meniscoid portion to form a volume of vaporizing liquid cryogen localized within the lower meniscus edge portion.

11. The method of claim 1, wherein (b) generating the biphasic inert cryogen comprises (b.1) directing a liquid inert cryogen from a source through a diffuser to separate the liquid flow component from the vaporous flow component.

12. A heating system comprising:

a open vessel for containing molten metal, the vessel including a bottom wall, a side wall, and an opening;

5 an inert cryogen source, the inert cryogen including a liquid flow component and a vaporous flow component;

a delivery system disposed proximate the opening, the delivery system comprising:

10 a lance including an inlet and a outlet, wherein the inlet is connected to the inert cryogen source;

a hood coupled to the outlet of the lance, wherein the hood directs the components of the inert cryogen toward the molten metal,

15 wherein the hood is configured to direct the liquid component of the inert cryogen toward the bottom wall of the vessel such that the liquid component contacts the molten metal to form an expanding volume of gas having a rate of expansion, and

20 wherein the hood is further configured to direct the vaporous component toward the molten metal to inhibit the rate of expansion of the expanding volume of gas.

13. The heating system of claim 12, wherein the hood comprises a curved housing including an inlet and an outlet located downstream from the inlet.

14. The heating system of claim 13, wherein the hood possesses a degree of curvature of about 0° to about 90°.

15. The heating system of claim 12, wherein the hood comprises an outlet oriented such that it is generally coplanar with the opening of the vessel.

16. The heating system of claim 12, wherein the hood comprises outlet oriented within the vessel at a point slightly below the opening of the vessel.

17. The heating system of claim 12, wherein the delivery system is operable to generate a flow rate of inert cryogen the range of about 0.002 lb/in² to about 0.005 lb/in², based upon the surface area of the molten metal.

18. The heating system of claim 12, wherein the hood is oriented proximate the side wall of the vessel.

19. The heating system of claim 12, wherein the delivery system further comprises a diffuser disposed at the outlet of the lance and housed within the hood, the diffuser operable to separate the liquid flow component from the vaporous flow component.

20. The heating system of claim 12, wherein:

the hood comprises a curved housing including an inlet and an outlet located downstream from the inlet;

the outlet of the hood is either generally coplanar with the opening of the vessel or disposed below the opening of the vessel; and

the delivery system is operable to generate a flow rate of inert cryogen in a range of about 0.002 lb/in² to about 0.005 lb/in², based upon the total surface area of the molten metal.

21. The heating system of claim 20, wherein the outlet of the hood is oriented proximate the side wall of the vessel.

22. A heating system comprising:

an open vessel for containing molten metal, the vessel including a bottom wall, a side wall, and an opening;

a source of inert cryogen, the inert cryogen including a liquid flow component and a vaporous flow component;

a delivery system disposed proximate the opening, the delivery system comprising:

a lance including an inlet and a outlet, the inlet being connected to the inert cryogen source;

a means for receiving the inert cryogen from the lance and for directing the liquid component of the inert cryogen toward the bottom wall of the vessel such that the liquid component contacts the molten metal to form an expanding volume of gas having a rate of expansion, wherein the means
5 for receiving inert cryogen is further configured to direct the vaporous component toward the molten metal to inhibit the rate of expansion of the expanding volume of gas.

23. The heating system of claim 22, wherein the delivery system is operable to generate a flow rate of inert cryogen the range of about 0.002 lb/in² to about 0.005 lb/in², based upon the surface area of the molten metal.

24. The heating system of claim 22, wherein the means for receiving the inert cryogen is oriented proximate the side wall of the vessel.

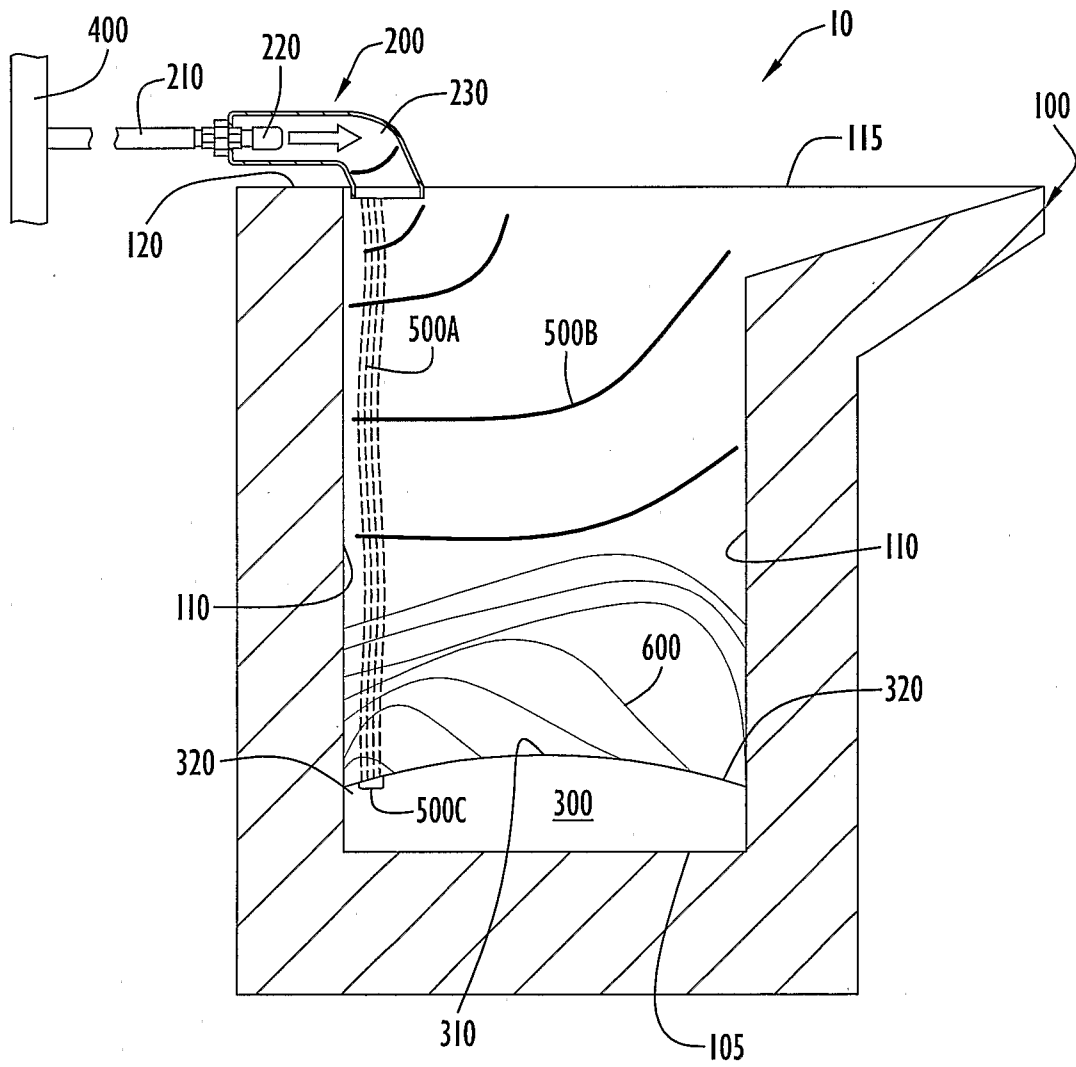


FIG. 1

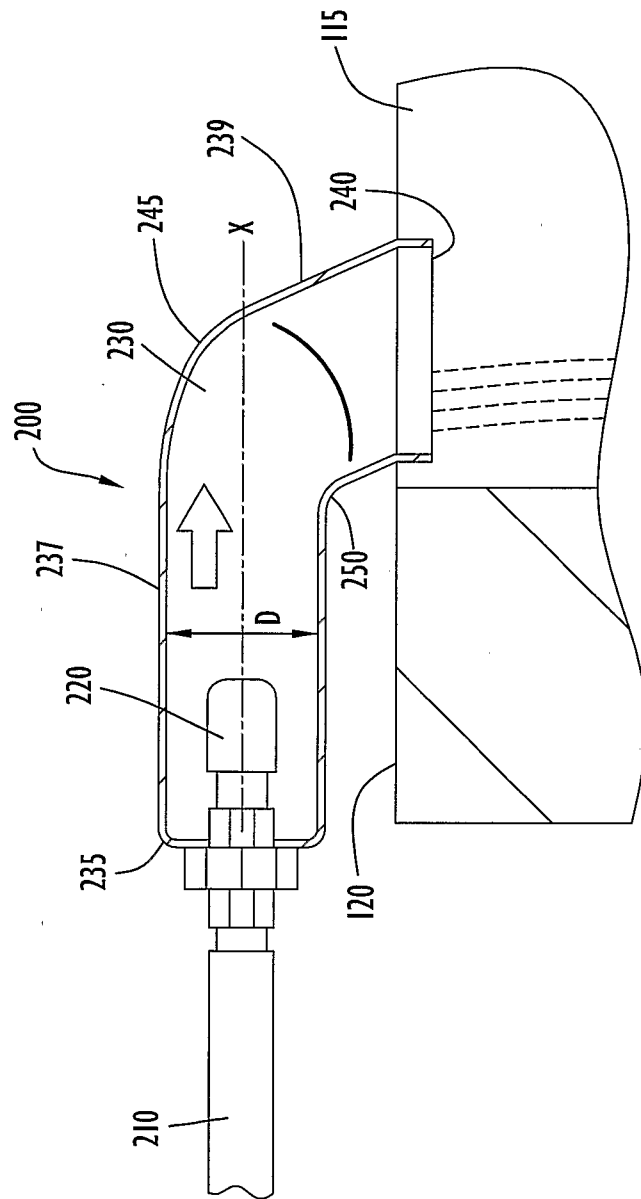


FIG.2

INTERNATIONAL SEARCH REPORT

International application No
PCT/IB2007/002353

A. CLASSIFICATION OF SUBJECT MATTER
 INV. B22D21/02 C21D1/74

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)
 B22D C21D

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)
 EPO-Internal, WPI Data, COMPENDEX

C. DOCUMENTS CONSIDERED TO BE RELEVANT		
Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	US 2002/070488 A1 (JEPSON STEWART C [US]) 13 June 2002 (2002-06-13) paragraphs [0017] - [0030]; figure 2 -----	1-4, 6-9, 22-24
X	US 4 990 183 A (ANDERSON SARA H [US] ET AL) 5 February 1991 (1991-02-05) column 2, lines 26-43; figure 1 ----- -/--	1-4, 6-9, 22-24

Further documents are listed in the continuation of Box C. See patent family annex.

<p>* Special categories of cited documents :</p> <p>*A* document defining the general state of the art which is not considered to be of particular relevance</p> <p>*E* earlier document but published on or after the international filing date</p> <p>*L* document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)</p> <p>*O* document referring to an oral disclosure, use, exhibition or other means</p> <p>*P* document published prior to the international filing date but later than the priority date claimed</p>	<p>*T* later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention</p> <p>*X* document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone</p> <p>*Y* document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.</p> <p>*&* document member of the same patent family</p>
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Date of the actual completion of the international search 28 November 2007	Date of mailing of the international search report 12/12/2007
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Name and mailing address of the ISA/ European Patent Office, P.B. 5818 Patentlaan 2 NL - 2280 HV Rijswijk Tel. (+31-70) 340-2040, Tx. 31 651 epo nl, Fax: (+31-70) 340-3016	Authorized officer Badcock, Gordon
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INTERNATIONAL SEARCH REPORT

International application No

PCT/IB2007/002353

C(Continuation). DOCUMENTS CONSIDERED TO BE RELEVANT		
Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	<p>TILL, K. (AIR LIQUIDE) ET AL: "The induction melting of stainless steel under the protection of liquid argon for powder metal manufacture." PUBLISHED BY: METAL POWDER INDUSTRIES FEDERATION. 105 COLLEGE ROAD EAST, PRINCETON, NJ 08540-6692, CONFERENCE: ADVANCES IN POWDER METALLURGY AND PARTICULATE MATERIALS-1994. VOL. 1. POWDER MANUFACTURING AND INDUSTRY TRENDS, TORONTO, ONT., CANADA, 8-1, 1994, XP009092752 Figure 2 and Discussion -----</p>	1-24
A	<p>BARBER, R.E. (FRANKLIN BRONZE AND ALLOY) ET AL: "Franklin bronze achieves dramatic results in SPAL application tests." INCAST, vol. 15, no. 5, 2002, pages 16-17, XP009092767 page 16 -----</p>	1-24
A	<p>EP 0 300 907 A (AIR LIQUIDE AMERICAN [US]) 25 January 1989 (1989-01-25) column 9, lines 47-65; figures 1,2 -----</p>	1-24

INTERNATIONAL SEARCH REPORT

Information on patent family members

International application No

PCT/IB2007/002353

Patent document cited in search report	Publication date	Publication date	Patent family member(s)	Publication date
US 2002070488	A1	13-06-2002	NONE	
US 4990183	A	05-02-1991	FR 2635789 A1	02-03-1990
EP 0300907	A	25-01-1989	AU 611462 B2	13-06-1991
			AU 1975888 A	27-01-1989
			AU 616126 B2	17-10-1991
			AU 6212390 A	29-11-1990
			CA 1276471 C	20-11-1990
			DE 3866988 D1	30-01-1992
			JP 1208426 A	22-08-1989
			US 4848751 A	18-07-1989