



(43) International Publication Date
5 July 2012 (05.07.2012)

(10) International Publication Number
WO 2012/090102 A2

- (51) **International Patent Classification:**
A01N 33/02 (2006.01) *A01P 1/00* (2006.01)
A01N 33/12 (2006.01)
- (21) **International Application Number:**
PCT/IB2011/055528
- (22) **International Filing Date:**
7 December 2011 (07.12.2011)
- (25) **Filing Language:** English
- (26) **Publication Language:** English
- (30) **Priority Data:**
12/982,169 30 December 2010 (30.12.2010) US
13/269,931 10 October 2011 (10.10.2011) US
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- (81) **Designated States (unless otherwise indicated, for every kind of national protection available):** AE, AG, AL, AM, AO, AT, AU, AZ, BA, BB, BG, BH, BR, BW, BY, BZ, CA, CH, CL, CN, CO, CR, CU, CZ, DE, DK, DM, DO, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, GT, HN, HR, HU, ID, IL, IN, IS, JP, KE, KG, KM, KN, KP, KR, KZ, LA, LC, LK, LR, LS, LT, LU, LY, MA, MD, ME, MG, MK, MN, MW, MX, MY, MZ, NA, NG, NI, NO, NZ, OM, PE, PG, PH, PL, PT, QA, RO, RS, RU, RW, SC, SD, SE, SG, SK, SL, SM, ST, SV, SY, TH, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, ZA, ZM, ZW.
- (84) **Designated States (unless otherwise indicated, for every kind of regional protection available):** ARIPO (BW, GH, GM, KE, LR, LS, MW, MZ, NA, RW, SD, SL, SZ, TZ, UG, ZM, ZW), Eurasian (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European (AL, AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HR, HU, IE, IS, IT, LT, LU, LV, MC, MK, MT, NL, NO, PL, PT, RO, RS, SE, SI, SK, SM, TR), OAPI (BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG).
- Published:**
— without international search report and to be republished upon receipt of that report (Rule 48.2(g))

(54) **Title:** DURABLE ANTIMICROBIAL COMPOSITION INCLUDING A SURFACTANT

(57) **Abstract:** Compositions having durable antimicrobial activity are disclosed herein. The durable antimicrobial compositions include a carbonate/bicarbonate salt of a quaternary ammonium cation, an organic acid, hydrogen peroxide, a surfactant, and a cationic polymer. The cationic polymer includes either a (3-acrylamidopropyl)trimethylammonium chloride monomer or a [2-(acryloyloxy)ethyl]trimethylammonium chloride monomer. The cationic polymer also includes another monomer selected from a polar, water-soluble monomer, a hydrophobic, silicone-containing monomer or mixtures of both types of monomers in combination with one of the trimethylammonium chloride monomers. The surfactant is selected from cationic surfactants, non-ionic surfactants, zwitterionic surfactants, and combinations thereof.



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DURABLE ANTIMICROBIAL COMPOSITION INCLUDING A SURFACTANT

RELATED APPLICATION DATA

This application is a continuation-in-part application claiming priority from presently copending U.S. Application No. 12/982,169 entitled "DURABLE ANTIMICROBIAL COMPOSITION" filed on 12/30/2010, in the names of Corey T. Cunningham et al.

FIELD

5 Compositions having durable antibacterial activity are described herein. The compositions include a carbonate/bicarbonate salt of a quaternary ammonium cation, an organic acid, hydrogen peroxide, a cationic polymer, and a surfactant. The cationic polymer includes a (3-acrylamidopropyl)trimethylammonium chloride monomer or a [2-(acryloyloxy)ethyl]trimethylammonium chloride monomer combined with another
10 monomer selected from a polar, water-soluble monomer, a hydrophobic, silicone-containing monomer, and mixtures of such monomers. The surfactant is selected from cationic surfactants, non-ionic surfactants, zwitterionic surfactants, and combinations thereof.

BACKGROUND

15 In order to protect health and maintain hygiene, a variety of environments require controlled and limited microbial growth. Such environments include temporary and permanent healthcare facilities, caregiver facilities (e.g., daycares, nursing homes, etc.) and households. When growth of potentially harmful microbes is not controlled/limited in these environments, the risk of infection and spread of disease increases. Infection and disease
20 may compromise the health and safety of humans and/or animals occupying these environments. While potentially not as sensitive as the above-identified environments, workplace and public environments may also be negatively impacted by uncontrolled/unlimited growth of disease-causing microbes.

Some types of microorganisms (bacteria, viruses, fungi, etc.) are capable of
25 negatively impacting the health and/or safety of living organisms. Such microorganisms

can be transmitted by contact with surfaces on which the microorganisms are present and/or multiplying and by contact between humans/animals already infected with particular microorganisms. When such microorganisms spread and infect new “hosts”, the “host” can either go from an otherwise healthy state to a state of illness or from a “compromised” state (i.e., a state of pre-existing illness or a weak immune system) to a more serious/severe state. The public health impact of the undesired spread of microorganisms is significant as reflected by time out of school, time away from work (either for self or to care for others not able to care for themselves), additional time for which professional health care is needed, etc. Therefore, it is desirable to be able to prevent or inhibit microbial presence/growth on targeted surfaces. The presence of microorganisms can be eliminated/controlled using surface treatments that may be applied directly (as from a spray bottle) and by using wipes or other carriers that include the surface treatment. Further, it is desirable that such surface treatments have durability and persistence so that they do not need to be re-applied on a frequent basis.

There are many detergent, disinfectant, cleaning and antimicrobial compositions known in the art for killing and preventing growth of microorganisms. Those compositions include components/ingredients that are well-known for antimicrobial functionality. For example, quaternary ammonium compounds are considered “broad spectrum” antimicrobial cationic compounds that are effective against both Gram positive (e.g., *Staphylococcus* species) and Gram negative (e.g., *Escherichia coli*) microorganisms. Other components/ingredients that may be incorporated into products for removing/reducing microorganisms on surfaces include alcohols, acids and bleaching agents, such as hydrogen peroxide. Not all of the antimicrobial components can be used at the same time because some of them form unstable combinations.

Disinfecting and cleaning compositions that provide antimicrobial activity over a period of time are also known in the art. For example, U.S. Patent No. 6,270,754 issued to Zhou et al. and entitled “Antimicrobial Cleaning Composition” (hereinafter “the ‘754 patent”) is directed to an antibacterial cleaning composition that exhibits germicidal activity for sustained periods of time. The ‘754 patent discloses an aqueous cleaning composition that includes a quaternary ammonium compound, an anionic polymer (where the anionic polymer has an acid number greater than 10 and the anionic polymer is partially or completely neutralized by the quaternary ammonium compound to form a polymer

complex), a dispersing agent and/or a water-miscible solvent. The aqueous cleaning composition of the '754 patent has antibacterial activity against both Gram positive and Gram negative bacteria. However, the components of the '754 patent may not be effective against a broader range of microorganisms, such as non-enveloped viruses.

5 In addition to the composition of the '754 patent, there are compositions known in the art that are effective against a broad spectrum of microorganisms and continue to have activity for a period of time. For example, U.S. Patent No. 7,598,214 issued to Cusack et al. and entitled "Disinfecting Compositions Containing A Polymer Complex Of an Organic Acid" (hereinafter "the '214 patent") is directed to compositions that include at least one
10 organic acid and at least one polymer capable of forming a complex with the at least one organic acid. The compositions of the '214 patent may also optionally include an anionic surfactant and an organic acid. The organic acid may be citric acid and examples of suitable polymers include vinylpyrrolidone/dimethylaminoethylmethacrylate copolymer, vinylpyrrolidone/vinylacetate copolymers, vinylpyrrolidone/vinylcaprolactum/ammonium
15 derivative terpolymers and polyvinylpyrrolidone. The compositions of the '214 patent need an organic acid and they are pH sensitive. Because of the acid-based reaction between the polymer having a tertiary amine functionality and the organic acid, the compositions are not effective in higher pH environments. In a higher pH environment, the reaction would reverse and the polymer would be rendered ineffective because it would be neutral. Though
20 the compositions of the '214 patent are effective against a broader spectrum of microorganisms, the compositions may not be effective against the spore-form of all microorganisms because the compositions cannot penetrate through the outer wall of the spores.

 While many antimicrobial compositions are known and while some of those
25 compositions maintain their antimicrobial activity over a period of time, there remains a need in the art for a durable antimicrobial composition that is effective against a broad range of microorganisms, including the spore-form of potentially harmful microorganisms. Additionally, there remains a need for a durable antimicrobial composition that is stable (i.e., is not reactive) so that it is not unnecessarily harsh (causing wear or corrosion) on the
30 surfaces on which it is used. Further, there remains a need for a durable antimicrobial composition that does not require a volatile solvent that may have an unpleasant smell.

Furthermore, there is a need to provide a durable antimicrobial composition that includes a stable surfactant such that cleaning is achieved and also provides a composition wherein a long-lasting biocide effect is maintained, rapid broad spectrum germicidal properties are maintained, and multiple fragrances can be employed for consumer acceptance.

SUMMARY

Durable antimicrobial compositions that are effective against a broad range of potentially harmful microorganisms and that do not have to be reapplied on a frequent basis to the surfaces on which controlled microbial growth is desired are described herein. The durable antimicrobial compositions are effective against a broad range of microorganisms, including the spore-form of microorganisms, because of the composition components, which are unexpectedly stable in combination with each other. Additionally, the durable antimicrobial compositions do not need to contain a volatile solvent that could make the compositions unpleasant to use.

In one aspect, the compositions described herein have durable antimicrobial activity and include a carbonate/bicarbonate salt of a quaternary ammonium cation, an organic acid, hydrogen peroxide, a surfactant and a cationic polymer. The cationic polymer includes a (3-acrylamidopropyl)trimethylammonium chloride monomer combined with another monomer selected from a polar, water-soluble monomer, a hydrophobic, a surfactant, silicone-containing monomer and mixtures of such monomers. The polar, water-soluble monomer may be selected from vinyl pyrrolidinone, hydroxyl ethyl acrylate, hydroxyl ethyl methacrylate, N,N'-dimethyl acrylamide, acrylamide and N-isopropyl acrylamide. The hydrophobic, silicone-containing monomer may be selected from unsubstituted or substituted vinyl or ethynyl group terminated siloxyl compounds, comprising monomethacryloxypropyl terminated polydimethylsiloxane, methacryloxypropyl tris(trimethylsiloxysilane) and methacryloxypropyl terminated T-structure siloxane. As described herein, the compositions have a durable or persistent activity to kill and prevent the growth of potentially-harmful microorganisms. The durability of the compositions is indicated by the compositions retaining antimicrobial activity after twenty-five insults of *E. coli* organisms as measured by a log 2 reduction in organisms upon the twenty-fifth insult of 10^6 total organisms. The durable antimicrobial composition are stable; the stability of the

compositions is reflected by the compositions maintaining their efficacy during shelf-life studies. For example, the compositions remain effective (meaning, they have the same level of durability to effect a log 2 reduction in organisms after twenty-five insults of 10^6 organisms) after storage for three months at 40°C; further, the compositions remain effective after storage for one month at 50°C, nine months at 25°C and after three freeze-thaw cycles. The compositions are liquid at room temperature and can be applied directly to a surface for which it is desired to prevent or inhibit microbial growth. The compositions may be applied using a spray bottle or other known structure for dispensing liquids. Alternatively, the compositions may be applied to a surface by transfer from a basesheet, such as a wiper, into which a representative composition has been incorporated. The basesheet may be made of a nonwoven material or of a cellulosic material. More particularly, the composition may include from 0.2 to 15.0 percent by weight of the carbonate/bicarbonate salt of a quaternary ammonium cation. The composition may include from 0.1 to 3.0 percent by weight of the organic acid, which may be selected from citric, malic, maleic, oxalic, glutaric, succinic, lactic, glycolic, fumaric, acetic, benzoic, propionic, sorbic, tartaric, formic and mixtures of such organic acids. The composition may include from 0.5 to 5.0 percent by weight of hydrogen peroxide and the composition may include from 0.5 to 10.0 percent by weight of cationic polymer. Even more particularly, the cationic polymer may include from 0.70 to 0.90 mole fraction of (3-acrylamidopropyl) trimethylammonium chloride monomer.

In another aspect, the compositions have durable antimicrobial activity and include a carbonate/bicarbonate salt of a quaternary ammonium cation, an organic acid, hydrogen peroxide and a cationic polymer. The cationic polymer includes a [2-acryloyloxy)ethyl]trimethylammonium chloride monomer combined with another monomer selected from a polar, water-soluble monomer, a hydrophobic, silicone-containing monomer and mixtures of such monomers. The polar, water-soluble monomer may be selected from vinyl pyrrolidinone, hydroxyl ethyl acrylate, hydroxyl ethyl methacrylate, N,N'-dimethyl acrylamide, acrylamide and N-isopropyl acrylamide. The hydrophobic, silicone-containing monomer may be selected from unsubstituted or substituted vinyl or ethynyl group terminated siloxyl compounds, comprising monomethacryloxypropyl terminated polydimethylsiloxane, methacryloxypropyl tris(trimethylsiloxysilane) and methacryloxypropyl terminated T-structure siloxane. As described herein, the compositions have a durable or persistent activity to kill and prevent

the growth of potentially-harmful microorganisms. The durability of the compositions is indicated by the compositions retaining antimicrobial activity after twenty-five insults of *E. coli* organisms as measured by a log 2 reduction in organisms upon the twenty-fifth insult of 10^6 total organisms. The durable antimicrobial composition are stable; the stability of the compositions is reflected by the compositions maintaining their efficacy during shelf-life studies. For example, the compositions remain effective (meaning, they have the same level of durability to effect a log 2 reduction in organisms after twenty-five insults of 10^6 organisms) after storage for three months at 40°C; further, the compositions remain effective after storage for one month at 50°C, nine months at 25°C and after three freeze-thaw cycles. The compositions are liquid at room temperature and can be applied directly to a surface for which it is desired to prevent or inhibit microbial growth. The compositions may be applied using a spray bottle or other known structure for dispensing liquids. Alternatively, the compositions may be applied to a surface by transfer from a basesheet, such as a wiper, into which a representative composition has been incorporated. The basesheet may be made of a nonwoven material or of a cellulosic material. More particularly, the composition may include from 0.2 to 15.0 percent by weight of the carbonate/bicarbonate salt of a quaternary ammonium cation. The composition may include from 0.1 to 3.0 percent by weight of the organic acid, which may be selected from citric, malic, maleic, oxalic, glutaric, succinic, lactic, glycolic, fumaric, acetic, benzoic, propionic, sorbic, tartaric, formic and mixtures of such organic acids. The composition may include from 0.5 to 5.0 percent by weight of hydrogen peroxide and the composition may include from 0.5 to 10.0 percent by weight of cationic polymer. Even more particularly, the cationic polymer may include from 0.70 to 0.90 mole fraction of [2-(acryloyloxy)ethyl] trimethylammonium chloride monomer.

These features will be described in greater detail herein. Further, it is to be understood that both the foregoing general description and the following detailed description are exemplary and are intended to provide further explanation of the invention claimed.

DETAILED DESCRIPTION

The present disclosure of the invention will be expressed in terms of its various components, elements, constructions, configurations, arrangements and other features that

may also be individually or collectively referenced by the term, “aspect(s)” of the invention, or other similar terms. It is contemplated that the various forms of the disclosed invention may incorporate one or more of its various features and aspects, and that such features and aspects may be employed in any desired, operative combination thereof.

5 It should also be noted that, when employed in the present disclosure, the terms “comprises”, “comprising” and other derivatives from the root term “comprise” are intended to be open-ended terms that specify the presence of any stated features, elements, integers, steps, or components, and are not intended to preclude the presence or addition of one or more other features, elements, integers, steps, components, or groups thereof.

10 Compositions having durable antimicrobial activity are disclosed herein. The compositions may be used to kill or to inhibit the growth of microorganisms that are potentially harmful or capable of causing disease. The durable antimicrobial compositions do not need to contain a volatile solvent and therefore, do not generate an unpleasant smell when used. The compositions are effective at killing and/or inhibiting growth of a broad
15 range of microorganisms. For example, the compositions are effective against both Gram positive and Gram negative bacteria. Additionally, the compositions are effective against viruses, fungi, mildew and mold. Further, the compositions are effective against bacteria that form spores, bacteria with waxy outer layers, fungi that form spores (fungal spores) and enveloped and non-enveloped viruses. Without wishing to be bound by theory, it is
20 believed that the composition is capable of breaking down the waxy outer layer of a bacteria or outer layer of a spore so that the composition can penetrate into the microorganism beyond the outer layer.

 The durable antimicrobial compositions may be used to control microbial growth on a variety of surfaces, including relatively durable objects having both hard and soft
25 surfaces; for example, appropriate surfaces may include door knobs, light switches, countertops, sinks, wash basins, telephones, keyboards, remote controls, medical instruments, upholstery, curtains, bedspreads, towels and shoes. The compositions may be applied to the targeted surface either directly, in liquid form, such as by a spray bottle or similar packaging capable of delivering a liquid composition in a relatively uniform
30 amount over the full surface to be covered. Alternatively, the compositions may be applied to the targeted surface by a carrier, such as a basesheet (i.e., a “wet” wipe or wiper). Because the compositions are liquid at room temperature, the composition may be applied

to a surface by wiping the surface with a basesheet that has been saturated with the composition; the composition will transfer from the basesheet to the surface. The basesheet may be formed from one or more woven materials, nonwoven materials, cellulosic materials, and combinations of such materials. More specifically, the basesheet may be
5 formed of nonwoven fibrous sheet materials that include meltblown, spunlace, coform, air-laid, bonded-carded web materials, hydroentangled materials, and combinations of such materials. Such materials can be made of synthetic or natural fibers, or a combination of such fibers. Typically, the basesheet will have a basis weight of from 25 to 120 grams per square meter and desirably from 40 to 90 grams per square meter.

10 The basesheet may be constructed of a coform material of polymer fibers and absorbent fibers having a basis weight of from 45 to 80 grams per square meter and desirably 60 grams per square meter. Typically, such coform basesheets are constructed of a gas-formed matrix of thermoplastic polymeric meltblown fibers and cellulosic fibers. Various suitable materials may be used to provide the polymeric meltblown fibers, such as,
15 for example, polypropylene microfibers. Alternatively, the polymeric meltblown fibers may be elastomeric polymer fibers, such as those provided by a polymer resin. For instance, VISTAMAXX elastic olefin copolymer resin designated PLTD-1810, available from ExxonMobil Corporation of Houston, TX, or KRATON G-2755, available from Kraton Polymers of Houston, TX, may be used to provide stretchable polymeric meltblown
20 fibers for the coform basesheets. Other suitable polymeric materials or combinations thereof may alternatively be utilized as known in the art.

The coform basesheet additionally may be constructed of various absorbent cellulosic fibers, such as, for example, wood pulp fibers. Suitable commercially available cellulosic fibers for use in the coform basesheets can include, for example, NF 405, which
25 is a chemically treated bleached southern softwood Kraft pulp, available from Weyerhaeuser Co. of Federal Way, WA; NB 416, which is a bleached southern softwood Kraft pulp, available from Weyerhaeuser Co.; CR-0056, which is a fully debonded softwood pulp, available from Bowater, Inc. of Greenville, SC; Golden Isles 4822 debonded softwood pulp, available from Koch Cellulose of Brunswick, GA; and
30 SULPHATATE HJ, which is a chemically modified hardwood pulp, available from Rayonier, Inc. of Jessup, GA. The relative percentages of the polymeric meltblown fibers and cellulosic fibers in the coform basesheet may vary over a wide range depending upon

the desired characteristics of the wipes. For example, the coform basesheet may have from 10 to 90 weight percent, desirably from 20 to 60 weight percent, and more desirably from 25 to 35 weight percent of polymeric meltblown fibers based on the dry weight of the coform basesheet.

5 The durable antimicrobial composition may be incorporated into the basesheet in an add-on amount of from 50 to 800 percent by weight of the basesheet. More specifically, the compositions may be incorporated into the basesheet in an add-on amount of from 200 to 600 percent by weight of the basesheet or from 400 to 600 percent by weight of the basesheet. The composition add-on amounts may vary depending on the composition of the
10 basesheet.

 The compositions disclosed herein having durable antimicrobial activity. The “durability” or “persistence” of antimicrobial activity is descriptive of a benefit provided by the durable antimicrobial composition. From a cost and efficiency standpoint, it is desirable to maintain antimicrobial activity on a surface over a period of time with one
15 application of a composition rather than having to frequently apply a composition because its antimicrobial activity rapidly dissipates. From a public health standpoint, a durable antimicrobial composition is desirable because such a composition is more likely to prevent microbial growth than a composition that is weaker to begin with and a durable antimicrobial composition introduces less liquid/material into the environment, thereby
20 decreasing the opportunity for microbes to develop resistance. The durability of the durable antimicrobial composition is measured by activity after twenty-five (25) insults with a representative Gram negative bacterium, *Escherichia coli* (*E. coli*). The durable antimicrobial compositions retain activity sufficient to cause a log 2 reduction upon the twenty-fifth insult of 10^6 total *E. coli* organisms. Additionally, the durability of the durable
25 antimicrobial composition is measured by ability to effect greater log 2 reduction against Gram positive bacteria, Gram negative bacteria, enveloped viruses, non-enveloped viruses, fungi, mildew and mold twenty-four (24) hours after application of the composition to a surface. Further, the durability of the durable antimicrobial composition is measured by ability to effect greater log 2 reduction in microorganisms in the presence of soil after
30 either of the first two assays described above (i.e., (1) twenty-fifth insult of 10^6 total organisms; or (2) twenty-four hours after application). From a practical standpoint, a standard surface, such as a countertop, table, telephone, etc., in a susceptible environment,

such as a hospital or daycare facility, is continuously exposed to potentially harmful microorganisms. Given the rate at which exposure to new microorganisms typically occurs, a durable antimicrobial composition may be applied to the surface in a timeframe of every 24 to 48 hours in order to kill and/or to prevent the growth of microorganisms.

- 5 Comparatively, an antimicrobial composition that is not durable would need to be applied continuously to a surface to maintain a comparable level of antimicrobial activity. In a less susceptible environment and with a less susceptible surface, such as draperies in a home, the durable antimicrobial composition may last up to seven days at full activity.

The durable antimicrobial compositions include a carbonate/bicarbonate salt of a quaternary ammonium cation. Quaternary ammonium compounds are generally considered “broad spectrum” antimicrobial cationic compounds that have efficacy against both Gram positive and Gram negative microorganisms. The carbonate/bicarbonate salts of quaternary ammonium cations may be selected from dioctyldimethylammonium carbonate, decyloctyldimethylammonium carbonate, didecyldimethylammonium carbonate, benzalkonium carbonate, benzethonium carbonate, stearylalkonium carbonate, cetrimonium carbonate, behentrimonium carbonate, dioctyldimethylammonium bicarbonate, decyloctyldimethylammonium bicarbonate, didecyldimethylammonium bicarbonate, benzalkonium bicarbonate, benzethonium bicarbonate, stearylalkonium bicarbonate, cetrimonium bicarbonate, behentrimonium bicarbonate and mixtures of one or more such carbonate salts. The durable antimicrobial composition may include from 0.2 to 15.0 percent by weight of one or more carbonate/bicarbonate salts of quaternary ammonium cations. Alternatively, the durable antimicrobial composition may include a chloride salt such as benzalkonium chloride, benzethonium chloride, cetrimonium chloride, stearylalkonium chloride and behentrimonium chloride.

25 The durable antimicrobial compositions also include an organic acid. Organic acids are also known to have efficacy against the growth of microorganisms. The organic acid may be selected from citric, malic, maleic, oxalic, glutaric, succinic, lactic, glycolic, fumaric, acetic, benzoic, propionic, sorbic, tartaric, formic and mixtures of one or more such organic acids. The durable antimicrobial composition may include from 0.1 to 3.0 percent by weight of one or more organic acids.

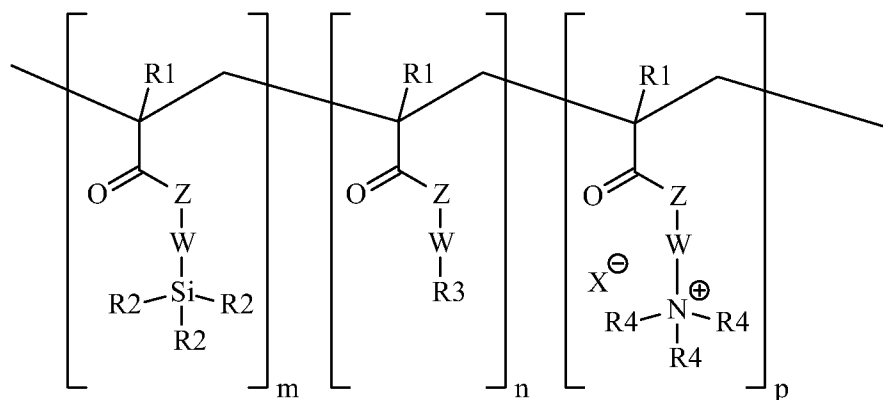
Additionally, the durable antimicrobial compositions also include hydrogen peroxide. The hydrogen peroxide is stable in the durable antimicrobial composition, despite

the presence of the carbonate/bicarbonate salt. Existing antimicrobial compositions do not contain stabilized hydrogen peroxide in combination with a carbonate/bicarbonate salt. The stability of the hydrogen peroxide is measured by the durable antimicrobial composition maintaining their initial concentration and efficacy during shelf-life studies. For example, the compositions remain effective (meaning, they have the same level of durability to effect a log 2 reduction in organisms after twenty-five insults of 10^6 organisms) after storage for three months at 40°C; further, the compositions remain effective after storage for one month at 50°C, nine months at 25°C and after three freeze-thaw cycles. While not wishing to be bound by theory, it is believed that the ability to provide compositions with stabilized hydrogen peroxide significantly expands the range of microorganisms that the durable antimicrobial compositions are effective against. Some microorganisms exist or are spread in spore form, where the spores have an outer layer; the outer layer presents a barrier to penetration by some conventional antimicrobial compositions. It is believed that the stabilized hydrogen peroxide in the durable antimicrobial composition is capable of penetrating the outer layer of spores, thereby facilitating exposure of the spore interior to the carbonate/bicarbonate salt of a quaternary ammonium cation. The carbonate/bicarbonate salt of a quaternary ammonium cation prevents future germination or development of the spore. The durable antimicrobial composition may include from 0.5 to 5.0 percent by weight of hydrogen peroxide.

The stability of the durable antimicrobial composition is also measurable by the ongoing/sustained detectable concentration of the carbonate/bicarbonate salt of a quaternary ammonium cation, organic acid and hydrogen peroxide components of the compositions. The carbonate/bicarbonate salt of a quaternary ammonium cation component may be detected using high pressure liquid chromatography (HPLC) with an evaporative light-scattering (ELS) detector. The mobile phase for the HPLC is an acidic mixture of acetonitrile and water. The organic acid component may be detected using HPLC with an ultra-violet (UV) absorption detector monitoring the 220 nanometer wavelength. The mobile phase for the HPLC for the detection of the organic acid is also an acidic mixture of acetonitrile and water. The hydrogen peroxide component may be detected by titrating the sample with a solution of ceric sulfate and ferroin indicator as described in the journal article, Frank P. Greenspan and Donald G. MacKellar entitled "Analysis of Aliphatic Per Acids" published in *Analytical Chemistry*, 1948, 20, 1061. The durable antimicrobial

compositions have a sustained and detectable presence of these components after experiencing the accelerated shelf-life conditions described herein.

The durability of the hydrogen peroxide in the presence of the carbonate/bicarbonate salt is provided by novel cationic polymers that are components of the durable antimicrobial composition. The cationic polymers have the following structure:



With respect to this structure, R1 may be independently selected from H (hydrogen) or methyl (CH₃); R2 may be independently selected from H (hydrogen), halide (fluoride, chloride, bromide, iodide), C₁₋₆ alkyl or alkoxy, aryl, linear or branched oligomeric or polymeric dimethyl siloxane; R3 may be independently selected from hydroxyl, alkyl amine, dialkyl amine or polyether; R4 may be independently selected from H (hydrogen), C₁₋₆ alkyl, or benzyl; Z may be independently selected from O (oxygen) or NH; W may be independently selected from C₁₋₆ alkyl; and X may be independently selected from fluoride, chloride, bromide, iodide, methosulfate or ethosulfate. Alternatively, the monomer represented by subscript “n” may be a vinyl pyrrolidinone. The values of m, n and p may be the same or they may be different. The values of m, n and p are integers and are selected to make the number average molecular weight in the range of 1000 to 100,000 g/mol.

One exemplary durable antimicrobial composition includes a cationic polymer that includes a (3-acrylamidopropyl)trimethylammonium chloride monomer combined with another monomer selected from a polar, water-soluble monomer, a hydrophobic, silicone-containing monomer, and mixtures of one or more polar, water-soluble monomers and hydrophobic, silicone-containing monomers. The polar, water-soluble monomer may be selected from vinyl pyrrolidinone, hydroxyl ethyl acrylate, hydroxyl ethyl methacrylate,

N,N'-dimethyl acrylamide, acrylamide, N-isopropyl acrylamide, and mixtures of one or more such polar water-soluble monomers. The hydrophobic, silicone-containing monomer may be selected from unsubstituted or substituted vinyl or ethynyl group terminated siloxyl compounds, comprising monomethacryloxypropyl terminated polydimethylsiloxane (commercially available from Gelest, Inc. as product "MCR-M11"), methacryloxypropyl tris(trimethylsiloxysilane), methacryloxypropyl terminated T-structure siloxane (commercially available from Gelest, Inc. as product "RTT-1011") and mixtures of one or more such hydrophobic, silicone-containing monomers. The durable antimicrobial composition may include from 0.5 to 10.0 percent by weight of cationic polymer. Additionally, the cationic polymer may include from 0.70 to 0.90 mole fraction of (3-acrylamidopropyl)trimethylammonium chloride monomer.

Another exemplary durable antimicrobial composition includes a cationic polymer that includes a [2-(acryloyloxy)ethyl]trimethylammonium chloride monomer combined with another monomer selected from a polar, water-soluble monomer, a hydrophobic, silicone-containing monomer, and mixtures of one or more polar, water-soluble monomers and hydrophobic, silicone-containing monomers. The polar, water-soluble monomer may be selected from vinyl pyrrolidinone, hydroxyl ethyl acrylate, hydroxyl ethyl methacrylate, N,N'-dimethyl acrylamide, acrylamide, N-isopropyl acrylamide, and mixtures of one or more such polar, water-soluble monomers. The hydrophobic, silicone-containing monomer may be selected from unsubstituted or substituted vinyl or ethynyl group terminated siloxyl compounds, comprising monomethacryloxypropyl terminated polydimethylsiloxane (commercially available from Gelest, Inc. as "MCR-M11"), methacryloxypropyl tris(trimethylsiloxysilane), methacryloxypropyl terminated T-structure siloxane (commercially available from Gelest, Inc. as product "RTT-1011") and mixtures of one or more such hydrophobic, silicone-containing monomers. The durable antimicrobial composition may include from 0.5 to 10.0 percent by weight of cationic polymer. Additionally, the cationic polymer may include from 0.70 to 0.90 mole fraction of [2-(acryloyloxy)ethyl]trimethylammonium chloride monomer.

The cationic polymers may be synthesized using a typical acrylate copolymer synthesis. For example, in a typical procedure, methacryloxypropyl tris(trimethylsiloxy) silane (0.2 g, 0.574 mmol), (3-acrylamidopropyl)trimethylammonium chloride solution (75 wt. % in water, 1.86 g, 6.77 mmol), hydroxyethyl acrylate (0.4 g, 3.44 mmol),

isopropanol (10 ml), and azobisisobutyronitrile (AIBN) solution (5 wt. % in THF, 0.015 mol/L, 0.98 ml) are added to a 20 ml vial inside a glove box under nitrogen atmosphere. The reagents are degassed by bubbling nitrogen through the solutions for 20 to 30 minutes at room temperature prior to entering the glove box. The reaction mixture may then be heated to 60°C for 18 hours under magnetic agitation. After the reaction is complete, the polymer solution is approximately 20 wt. %. The polymer may then be used without further purification. This exemplary procedure for synthesizing the cationic polymer is adapted from the synthetic procedures described in the journal article by Charles L. McCormick and Andrew B. Lowe entitled "Aqueous RAFT polymerization: recent developments in synthesis of functional water-soluble (co)polymers with controlled structure" *Accounts of Chemical Research*, **2004**, 37, 312-325 and the journal article by Yulia A. Vasilieva, David B. Thomas, Charles W. Scales, and Charles L. McCormick entitled "Direct controlled polymerization of a cationic methacrylamido monomer in aqueous media via the RAFT process" *Macromolecules*, **2004**, 37, 2728-2737.

Additionally, the durable antimicrobial compositions also include a compatible surfactant. The surfactant is selected from cationic surfactants, non-ionic surfactants, zwitterionic surfactants and combinations thereof. The durable antimicrobial composition may suitably include one or more compatible surfactants in an amount of from about 0.01 to about 10.0 percent by weight of the composition. Not all surfactants are compatible with durable antimicrobial composition. For example, an anionic surfactant will react with other components of the durable antimicrobial composition to form a coacervate. This leads to poor resistance to abrasion and poor efficacy.

As described above, the surfactant may be a nonionic surfactant. Nonionic surfactants typically have a hydrophobic base, such as a long chain alkyl group or an alkylated aryl group, and a hydrophilic chain comprising a certain number (e.g., 1 to about 30) of ethoxy and/or propoxy moieties. Examples of some classes of nonionic surfactants that can be used include, but are not limited to, ethoxylated alkylphenols, ethoxylated and propoxylated fatty alcohols, polyethylene glycol ethers of methyl glucose, polyethylene glycol ethers of sorbitol, ethylene oxide-propylene oxide block copolymers, ethoxylated esters of fatty (C₈₋₁₈) acids, condensation products of ethylene oxide with long chain amines or amides, condensation products of ethylene oxide with alcohols, and combinations thereof.

Various specific examples of suitable nonionic surfactants for use in the durable antimicrobial composition include, but are not limited to, methyl gluceth-10, PEG-20 methyl glucose distearate, PEG-20 methyl glucose sesquistearate, C₁₁₋₁₅ pareth-20, ceteth-8, ceteth-12, dodoxynol-12, laureth-15, PEG-20 castor oil, polysorbate-20, 5 steareth-20, polyoxyethylene-10 cetyl ether, polyoxyethylene-10 stearyl ether, polyoxyethylene-20 cetyl ether, polyoxyethylene-10 oleyl ether, polyoxyethylene-20 oleyl ether, an ethoxylated nonylphenol, ethoxylated octylphenol, ethoxylated dodecylphenol, ethoxylated fatty (C₁₋₂₂) alcohol, including 3 to 20 ethylene oxide moieties, polyoxyethylene-20 isohexadecyl ether, polyoxyethylene-23 glycerol laurate, PEG-80 10 sorbitan laurate, polyoxy-ethylene-20 glyceryl stearate, PPG-10 methyl glucose ether, PPG-20 methyl glucose ether, polyoxyethylene-20 sorbitan monoesters, polyoxyethylene-80 castor oil, polyoxyethylene-15 tridecyl ether, polyoxy-ethylene-6 tridecyl ether, laureth-2, laureth-3, laureth-4, PEG-3 castor oil, PEG-12 dioleate, PEG-8 dioleate, and combinations thereof.

15 Additional nonionic surfactants that can be used include water soluble alcohol ethylene oxide condensates, such as the condensation products of a secondary aliphatic alcohol containing between about 8 to about 18 carbon atoms in a straight or branched chain configuration condensed with between about 5 to about 30 moles of ethylene oxide. Such nonionic surfactants are commercially available under the trade name Tergitol from 20 The Dow Chemical Company (Midland, MI). Specific examples of such commercially available nonionic surfactants of the foregoing type are C₁₁₋₁₅ secondary alkanols condensed with either 9 moles of ethylene oxide (Tergitol 15-S-9) or 12 moles of ethylene oxide (Tergitol 15-S-12) marketed by The Dow Chemical Company (Midland, MI)..

Other suitable nonionic surfactants include the polyethylene oxide condensates of 25 one mole of alkyl phenol containing from about 8 to 18 carbon atoms in a straight or branched chain alkyl group with about 5 to 30 moles of ethylene oxide. Specific examples of alkyl phenol ethoxylates include nonyl condensed with about 9.5 moles of ethylene oxide per mole of nonyl phenol, dinonyl phenol condensed with about 12 moles of ethylene oxide per mole of phenol, dinonyl phenol condensed with about 15 moles of ethylene oxide 30 per mole of phenol and diisooctylphenol condensed with about 15 moles of ethylene oxide per mole of phenol. Commercially available nonionic surfactants of this type include Igepal CO-630 (a nonyl phenol ethoxylate) marketed by ISP Corp. (Wayne, NJ). Suitable non-

ionic ethoxylated octyl and nonyl phenols include those having from about 7 to about 13 ethoxy units. Such compounds are commercially available under the trade name Triton X by The Dow Chemical Company (Midland, MI).

Alkyl polyglycosides may also be used as a nonionic surfactant in the durable antimicrobial composition. Suitable alkyl polyglycosides are known nonionic surfactants that are alkaline and electrolyte stable. Alkyl mono and polyglycosides are prepared generally by reacting a monosaccharide, or a compound hydrolyzable to a monosaccharide with an alcohol such as a fatty alcohol in an acid medium. Commercially available nonionic surfactants of this type include Glucopon 425 marketed BASF (Ludwigschafen, Germany).

Suitable zwitterionic surfactants for use in the durable antimicrobial composition include, for example, alkyl amine oxides, silicone amine oxides, and combinations thereof. Various specific zwitterionic surfactants for use in the durable antimicrobial composition include, for example, Almondamidopropylamine Oxide, Babassuamidopropylamine Oxide, Behenamine Oxide, Cocamidopropylamine Oxide, Cocamine Oxide, Dihydroxyethyl Cocamine Oxide, Dihydroxyethyl Lauramine Oxide, Dihydroxyethyl Stearamine Oxide, Isostearamidopropylamine Oxide, Isostearamidopropyl Morpholine Oxide, Lauramidopropylamine Oxide, Decylamine Oxide, Lauramine Oxide, Methyl Morpholine Oxide, Myristamidopropylamine Oxide, Myristamine Oxide, Palmitamidopropylamine Oxide, Palmitamine Oxide, PEG-3 Lauramine Oxide, Soyamidopropylamine Oxide, Stearamidopropylamine Oxide, Stearamine Oxide, and combinations thereof. Commercially available alkyl amine oxide surfactants of this type include Mackamine CO (Cocamine Oxide) marketed by McIntyre Group, A Rhodia Company, (University Park, IL).

Suitable cationic surfactants for use in the durable antimicrobial composition include, for example, alkyl ammonium salts, polymeric ammonium salts, alkyl pyridinium salts, aryl ammonium salts, alkyl aryl ammonium salts, silicone quaternary ammonium compounds, and combinations thereof. Specific examples of cationic surfactants include behenyltrimonium chloride, stearylkonium chloride, distearalkonium chloride, chlorohexidine diglutamate, polyhexamethylene biguanide (PHMB), cetyl pyridinium chloride, benzammonium chloride, benzalkoniumchloride, and combinations thereof.

In addition to the components described herein, the durable antimicrobial composition may also include a polar carrier solvent, pH adjuster, fragrance, preservative, dye, corrosion inhibitor, builder, cleansing solvent and other components known to be useful in antimicrobial compositions. The durable antimicrobial composition may include
5 from 67 to 98 percent by weight of one or more of these other components.

While other blending methods may be used, an example of one method of blending the durable antimicrobial composition is as follows: (1) Add water to vessel for mixing of the components of the compositions; (2) Slowly add the carbonate/bicarbonate salt of the quaternary ammonium cation component to the vessel; (3) Slowly add the organic acid
10 component to the vessel and begin mixing at low revolutions per minute (RPM) (i.e., 150-250 RPM); (4) Continue mixing until any foam that is present dissipates (e.g., up to 10 minutes for a 1 liter batch); (5) If desired for additional stability when the final composition is applied to a surface, add a stabilizer such as urea and continue mixing at low RPM (e.g., add 0.4 percent by weight of urea if adding 3.0 percent by weight hydrogen peroxide); (6)
15 Slowly add hydrogen peroxide to the vessel and continue mixing at low RPM; (7) Slowly add the cationic polymer component to the vessel and continue mixing at low RPM; (8) If desired for solution clarity, an appropriate organic solvent (e.g., ethanol, isopropanol, ethylene glycol, propylene glycol, butylene glycol, ethylene glycol monobutyl ether, etc.) may slowly be added to the vessel; and (9) If necessary, adjust the pH of the final
20 composition in the vessel to pH 3.0 (+/- 0.25) with a dilute (10 to 25 percent by weight) solution of potassium hydroxide. Those of skill in the art will appreciate that there are other methods by which the components of the durable antimicrobial composition may be blended. However, the durable antimicrobial composition described herein is desirably made by the steps described above wherein the carbonate/bicarbonate salt of the quaternary
25 ammonium cation is neutralized by the addition of the organic acid in step 3 prior to the addition of the hydrogen peroxide.

Representative examples of the cationic polymers of the durable antimicrobial compositions are provided in Table 1 below. Each exemplary cationic polymer described in Table 1 was used in a durable antimicrobial composition that included the following
30 components: (1) 2 percent by weight of CARBOQUAT H solution as available from Lonza Group Limited Switzerland; (2) 0.85 percent by weight of citric acid; (3) 3 percent by weight hydrogen peroxide; and (4) 5 percent by weight of the cationic polymers described

in Table 1 below. The compositions also included 0.4 percent by weight urea and 20 percent ethanol; the remainder was water. Note, while ethanol was used for purposes of these examples, the ethanol is not needed for the durable antimicrobial composition to have the described efficacy and durability. In fact, before the treated surfaces were insulted as
5 described below, the treated surfaces were allowed to dry and the ethanol and water would have evaporated. Similarly, while not required, the urea is added to provide enhanced stability of the compositions after application to a surface. Each of these compositions produced at least a log 3 reduction of microorganisms within five minutes after twenty-five, individual insults of 10^6 *E. coli* organisms. The values in Table 1 represent the mole
10 fractions of the individual monomers forming the cationic polymer.

Table 1

Example	APTAC	AETAC	VP	HEA	DMA	AM	MAPDMS	TRIS	TPDMS
1		0.70-0.80	0.30-0.20						
2		0.70-0.80	0.15-0.25						0.05-0.10
3		0.70-0.80	0.10-0.20					0.05-0.10	
4		0.70-0.80	0.10-0.20				0.10		
5		0.90	0.05				0.05		
6		0.80-0.90		0.10-0.20					
7		0.80-0.90		0.05-0.15					0.05-0.10
8		0.70-0.90		0.05-0.20				0.05-0.10	
9		0.70-0.90		0.05-0.25			0.05-0.10		
10		0.90							0.10
11		0.90						0.10	
12	0.70-0.90			0.10-0.30					
13	0.70-0.90			0.05-0.20			0.05-0.10		
14	0.90				0.05		0.05		
15	0.80-0.90				0.05-0.10			0.05-0.10	
16	0.70-0.80				0.15-0.25				0.05-0.10
17	0.70-0.90		0.10-0.30						
18	0.70-0.90		0.05-0.25				0.05-0.10		
19	0.80-0.90		0.05-0.15					0.05-0.10	
20	0.70-0.90					0.10-0.30			
21	0.80-0.90					0.05-0.15	0.05-0.10		
22	0.70-0.90					0.05-0.25		0.05-0.10	
23	0.70-0.80					0.15-0.25			0.05-0.10
24	0.90						0.10		
25	0.90							0.10	
26	0.90								0.10

Legend: APTAC = (3-acrylamidopropyl)trimethylammonium chloride

AETAC = ([2-(Acryloyloxy)ethyl]trimethylammonium chloride

5 VP = vinyl pyrrolidinone

HEA = hydroxyl ethyl acrylate

DMA = dimethyl acrylamide

AM = acrylamide

MAPDMS = monomethacryloxypropyl terminated polydimethylsiloxane

10 (product MCR-M11 from Gelest, Inc.)

TRIS = methacryloxypropyl tris(trimethylsiloxysilane)

TPDMS = methacryloxypropyl T-Structure siloxane

(product RTT-1011 from Gelest, Inc.)

Unless indicated otherwise, the monomers were commercially-available and
 15 obtained from Sigma-Aldrich Co.

In addition to the combinations of monomers provided in Table 1 above, there are additional examples of monomer combinations forming the cationic polymers provided in Table 2 below. In addition to the mole fractions of the specific monomers forming the cationic polymer component, Table 2 also includes the log reduction in microorganisms after the 25th and (where appropriate) the 50th insults of 10⁶ *E. coli* organisms. The durable antimicrobial compositions represented in Table 2 have the same components described for the exemplary compositions in Table 1 and the same legend for the names of the monomers.

Table 2

Ex. #	APTAC	AETAC	VP	HEA	DMA	AM	MAPDMS	TRIS	TPDMS	Log Reduction after 25 insults	Log Reduction after 50 insults
1		0.90		0.05			0.05			4.6	1.5
2	0.90			0.05			0.05			4.2	1.0
3		0.90	0.05				0.05			5.6	1.9
4	0.90		0.05				0.05			4.4	0.4
5		0.80		0.15			0.05			4.7	1.5
6	0.80			0.15			0.05			3.3	not tested
7	0.80		0.15				0.05			2.5	not tested
8		0.70		0.25			0.05			4.9	0.5
9	0.70			0.25			0.05			4.2	1.7
10	0.70		0.25				0.05			3.0	not tested
11		0.90		0.05				0.05		5.6	4.1
12	0.90		0.05					0.05		3.2	not tested
13		0.80		0.15				0.05		4.1	4.8
14		0.80	0.15					0.05		3.8	not tested
15	0.80		0.15					0.05		2.7	not tested
16	0.70					0.25		0.05		2.1	not tested
17	0.70				0.25			0.05		2.1	not tested
18	0.90				0.05			0.05		6.8	2.8
19	0.90					0.05		0.05		3.3	not tested
20		0.90		0.10						5.3	2.5
21		0.90		0.10						3.3	not tested
22		0.90		0.10						3.1	not tested
23	0.90			0.10						4.2	0.4
24		0.90						0.10		4.0	5.3
25	0.90							0.10		4.7	1.8
26	0.90							0.10		3.3	not tested
27	0.90							0.10		2.0	not tested
28		0.80		0.10					0.10	3.4	not tested
29		0.70	0.20						0.10	2.6	not tested
30	0.70				0.20				0.10	2.5	not tested
31		0.90							0.10	3.1	not tested
32	0.90								0.10	3.9	1.5

Some of the examples in Table 2 were not tested at the 50th insult level. The reason is because the composition provided a log reduction in organisms of between 2 and 3 after twenty-five insults and therefore, was unlikely to achieve the same level of log reduction after an additional twenty-five insults.

5 Samples were also prepared to illustrate the stability of the composition with a surfactant included. To show stability, tests were run to determine the release profile of different biocides contained in exemplary durable antimicrobial formulations on hard surfaces and the durability of the films obtained from such formulations. To illustrate the release profile of the various compositions, sample compositions were prepared as
10 described below. To test durability, pre-weighed substrates were placed in a fume-hood and 80-microliters of durable antimicrobial formulation was dispensed on top of each substrate spreading the formulation across the entire surface. The substrate is a non-porous, pre-cleaned, passivated 1 inch by 1 inch piece of stainless steel (18 ga 304 sst with mirrored finish). Substrates were passivated by immersing substrates in the Carboquat-H 3 wt. %
15 solution for 2 hours; immersing them in the citric acid 3 wt. % solution for 2 hours and rinsing substrates twice by immersing them in clean DI-water for 30 minutes; and immersing them in the 3 percent hydrogen peroxide solution for 2 hours and rinsing the substrates twice immersing by them in clean DI-water for 30 minutes. After application of the formulation, the substrates were dried for 14 hours and then weighed to determine the
20 weight of the film.

 The substrates were extracted with 1.2 mL of DI water in a weighing dish placed on orbital-shaker (IKA Shuttler MTS4). The substrate is placed into the DI water with the polymer film facing the weighing dish, in the water, and the speed of the orbital shaker is set to 100. Aliquots of 300 microliters each of the extraction fluid are then transferred to
25 HPLC vials at 15 seconds, 30 seconds, 1 minute, 2 minutes and 7.5 minutes. An HPLC system with UV- & ELSD-detectors; and a Neptune Hilic Silica Column (5 μ 100A; 15 cm by 4.6 mm) from ES-Industries (Cat. # 135221-NPN-SI) was used to determine the amount of the biocide released from the system at each time. The UV-Detector was set at 195 nm and the ELSD-Detector was set at a gas flow of 0.6, with a Neb. Temperature of 100°C and
30 an Evap. Temperature of 80°C. The gradient was set at 0.08 percent TFA with an injection volume of 10 μ L.

 Representative examples of a surfactant of the durable antimicrobial composition are provided in Table 3 below. Each exemplary surfactant described in Table 3 was used in a

durable antimicrobial composition in a concentration of 1.0 percent active surfactant by weight that included the following components: (1) 2 percent by weight of CARBOQUAT H solution as available from Lonza Group Limited Switzerland; (2) 0.85 percent by weight of citric acid; (3) 3 percent by weight hydrogen peroxide; and (4) 2.5 percent by weight polyquaternium-22 polymer (MERQUAT 295 polymer available from Nalco Company).
 The compositions also included 0.4 percent by weight urea and 20 percent ethanol; the remainder was water. Table 3 also illustrates the biocide release of various actives of the durable antimicrobial compositions.

Table 3

Example	Surfactant	Trade Name	% Release of citric acid after 2 minutes	% Release of Carboquat H after 2 minutes
1	None	N/A	76%	82%
2	coco-glucoside	Glucopon 425N	67%	74%
3	decylamine oxide	Mackamine C-10	91%	82%
4	alcohol ethoxylate	Ecosurf EH-6	91%	68%
5	C11-15 pareth-9	Tergitol 15-S-9	83%	63%

10

The durable antimicrobial compositions including the exemplary surfactants represented in Table 3 have similar release profile of biocide and similar durability described for the exemplary durable antimicrobial composition not including a surfactant in Table 3. All of the durable antimicrobial compositions including the exemplary surfactants had quick release of the citric acid and CARBOQUAT H compound illustrating the compositions are stable. Similarly, the durable antimicrobial compositions provide similar durability required of the antimicrobial composition described herein.

While the durable antimicrobial compositions have been described in detail with respect to specific aspects thereof, it will be appreciated that those skilled in the art, upon attaining an understanding of the foregoing, may readily conceive of alterations to, variations of and equivalents to these compositions. Accordingly, the scope of the present invention should be assessed as that of the claims and any equivalents thereto.

20

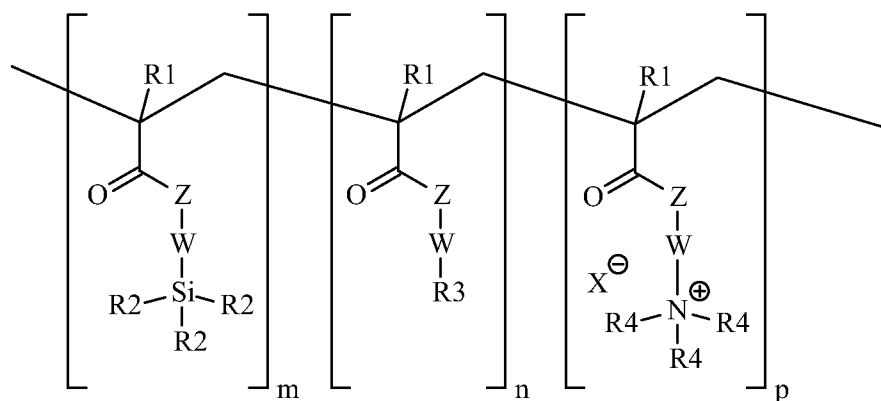
WE CLAIM:

1. A composition having durable antimicrobial activity comprising:
a carbonate/bicarbonate salt of a quaternary ammonium cation;
an organic acid;
hydrogen peroxide;
a surfactant selected from cationic surfactants, non-ionic surfactants, zwitterionic surfactants, and combinations thereof; and
a cationic polymer that includes a (3-acrylamidopropyl)trimethylammonium chloride monomer combined with another monomer selected from a polar, water-soluble monomer, a hydrophobic, silicone-containing monomer, and combinations thereof.
2. The composition of claim 1 wherein the polar, water-soluble monomer is selected from vinyl pyrrolidinone, hydroxyl ethyl acrylate, hydroxyl ethyl methacrylate, N,N'-dimethyl acrylamide, acrylamide and N-isopropyl acrylamide.
3. The composition of claim 1 wherein the hydrophobic, silicone-containing monomer is selected from unsubstituted or substituted vinyl or ethynyl group terminated siloxyl compounds, comprising monomethacryloxypropyl terminated polydimethylsiloxane, methacryloxypropyl tris(trimethylsiloxysilane) and methacryloxypropyl terminated T-structure siloxane.
4. The composition of claim 1 wherein the carbonate/bicarbonate salt of a quaternary ammonium cation is selected from dioctyldimethylammonium carbonate, decyloctyldimethylammonium carbonate, didecyldimethylammonium carbonate, benzalkonium carbonate, benzethonium carbonate, stearalkonium carbonate, cetrimonium carbonate, behentrimonium carbonate, dioctyldimethylammonium bicarbonate, decyloctyldimethylammonium bicarbonate, didecyldimethylammonium bicarbonate, benzalkonium bicarbonate, benzethonium bicarbonate, stearalkonium bicarbonate, cetrimonium bicarbonate, behentrimonium bicarbonate, and combinations thereof.
5. The composition of claim 1 wherein the organic acid is selected from citric, malic, maleic, oxalic, glutaric, succinic, lactic, glycolic, fumaric, acetic, benzoic, propionic, sorbic, tartaric, formic, and combinations thereof.

6. The composition of claim 1 wherein the composition further comprises urea.
7. The composition of claim 1 wherein the composition retains antimicrobial activity after twenty-five insults of *E. coli* as measured by a log 2 reduction in organisms upon the twenty-fifth insult of 10^6 total organisms.
8. The composition of claim 1 wherein the composition is effective against Gram positive bacteria, Gram negative bacteria, enveloped viruses, non-enveloped viruses, fungi, fungal spores, mildew and mold.
9. The composition of claim 1 wherein the composition includes from 0.2 to 15.0 percent by weight of the carbonate/bicarbonate salt of a quaternary ammonium cation.
10. The composition of claim 1 wherein the composition includes from 0.1 to 3.0 percent by weight of the organic acid.
11. The composition of claim 1 wherein the composition includes from 0.5 to 5.0 percent by weight of the hydrogen peroxide.
12. The composition of claim 1 wherein the composition includes from 0.05 to 3.0 percent by weight of the surfactant.
13. The composition of claim 1 wherein the surfactant is selected from ethoxylated alkylphenols, ethoxylated and propoxylated fatty alcohols, polyethylene glycol ethers of methyl glucose, polyethylene glycol ethers of sorbitol, ethylene oxide-propylene oxide block copolymers, ethoxylated esters of fatty (C_{8-18}) acids, condensation products of ethylene oxide with long chain amines or amides, condensation products of ethylene oxide with alcohols, alkyl ammonium salts, polymeric ammonium salts, alkyl pyridinium salts, aryl ammonium salts, alkyl aryl ammonium salts, silicone quaternary ammonium compounds, alkyl amine oxides, silicone amine oxides, and combinations thereof.
14. The composition of claim 1 wherein the composition includes from 0.5 to 10.0 percent by weight of the cationic polymer.
15. The composition of claim 14 wherein the cationic polymer includes from 0.70 to 0.90 mole fraction of the (3-acrylamidopropyl)trimethylammonium chloride monomer.

16. The composition of claim 1 wherein the composition is incorporated into a nonwoven basesheet.
17. A composition having durable antibacterial activity comprising:
- a carbonate/bicarbonate salt of a quaternary ammonium cation;
 - an organic acid;
 - hydrogen peroxide;
 - a surfactant selected from cationic surfactants, non-ionic surfactants, zwitterionic surfactants and combinations thereof; and
 - a cationic polymer that includes [2-(acryloyloxy)ethyl]trimethylammonium chloride monomer combined with another monomer selected from a polar, water-soluble monomer, a hydrophobic, silicone-containing monomer, and mixtures thereof.
18. The composition of claim 17 wherein the polar, water-soluble monomer is selected from vinyl pyrrolidinone, hydroxyl ethyl acrylate, hydroxyl ethyl methacrylate, N,N'-dimethyl acrylamide, acrylamide and N-isopropyl acrylamide.
19. The composition of claim 17 wherein the hydrophobic, silicone-containing monomer is selected from unsubstituted or substituted vinyl or ethynyl group terminated siloxyl compounds, comprising monomethacryloxypropyl terminated polydimethylsiloxane, methacryloxypropyl tris(trimethylsiloxysilane) and methacryloxypropyl terminated T-structure siloxane.
20. The composition of claim 17 wherein the surfactant is selected from ethoxylated alkylphenols, ethoxylated and propoxylated fatty alcohols, polyethylene glycol ethers of methyl glucose, polyethylene glycol ethers of sorbitol, ethylene oxide-propylene oxide block copolymers, ethoxylated esters of fatty (C₈₋₁₈) acids, condensation products of ethylene oxide with long chain amines or amides, condensation products of ethylene oxide with alcohols, alkyl ammonium salts, polymeric ammonium salts, alkyl pyridinium salts, aryl ammonium salts, alkyl aryl ammonium salts, silicone quaternary ammonium compounds, alkyl amine oxides, silicone amine oxides, and combinations thereof.
21. The composition of claim 17 wherein the composition includes from 0.05 to 3.0 percent by weight of the surfactant.

22. A composition having durable antimicrobial activity comprising:
- a carbonate/bicarbonate salt of a quaternary ammonium cation;
 - an organic acid;
 - hydrogen peroxide;
 - a surfactant selected from cationic surfactants, non-ionic surfactants, zwitterionic surfactants and combinations thereof; and
 - a cationic polymer that has the following structure:



wherein R1 is selected from hydrogen and methyl; R2 is selected from hydrogen, a halide and a C₁₋₆ alkyl or alkoxy, aryl, linear or branched oligomeric or polymeric dimethyl siloxane; R3 is selected from hydroxyl, alkyl amine, dialkyl amine and polyether; R4 is selected from hydrogen, C₁₋₆ alkyl and benzyl; Z is selected from oxygen and NH; W is selected from C₁₋₆ alkyl; X is selected from fluoride, chloride, bromide, iodide, methosulfate and ethosulfate; and the values of m, n and p are integers and are selected to make the number average molecular weight in the range of 1000 to 100,000 g/mol.

23. The composition of claim 22 wherein the surfactant is selected from ethoxylated alkylphenols, ethoxylated and propoxylated fatty alcohols, polyethylene glycol ethers of methyl glucose, polyethylene glycol ethers of sorbitol, ethylene oxide-propylene oxide block copolymers, ethoxylated esters of fatty (C₈₋₁₈) acids, condensation products of ethylene oxide with long chain amines or amides, condensation products of ethylene oxide with alcohols, alkyl ammonium salts, polymeric ammonium salts, alkyl pyridinium salts,

aryl ammonium salts, alkyl aryl ammonium salts, silicone quaternary ammonium compounds, alkyl amine oxides, silicone amine oxides, and combinations thereof.

24. The composition of claim 22 wherein the composition includes from 0.05 to 3.0 percent by weight of the surfactant.