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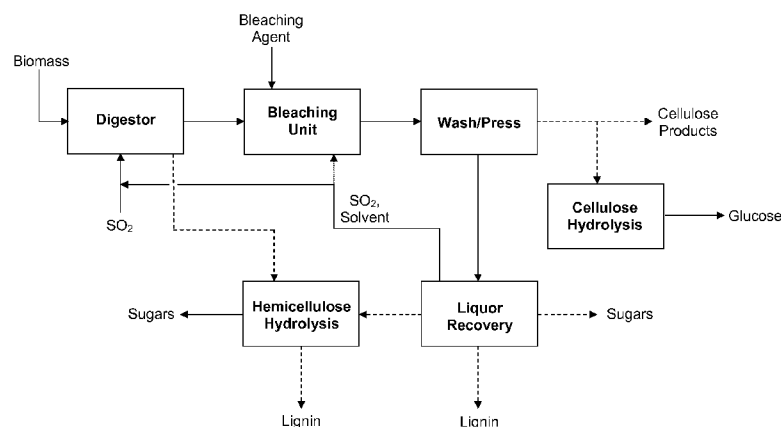
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- as to applicant's entitlement to apply for and be granted a patent (Rule 4.17(ii))
- as to the applicant's entitlement to claim the priority of the earlier application (Rule 4.17(iii))

[Continued on next page]

(54) Title: PROCESS FOR PRODUCING PURIFIED CELLULOSE

FIG. 1



(57) Abstract: In some variations, the invention provides a process for producing purified cellulose, comprising: providing a feedstock comprising lignocellulosic biomass; contacting the feedstock with sulfur dioxide, water, and a solvent for lignin, to produce intermediate solids and a liquid phase comprising hemicelluloses and lignin; mildly bleaching the intermediate solids to further delignify the intermediate solids, thereby generating cellulose-rich solids; and washing the cellulose-rich solids to generate purified cellulose with less than 2 weight percent lignin. The bleaching may employ bleaching agents including lignin-modifying enzymes. The bleaching and washing steps may be combined. It is also possible to carry out bleaching prior to, or simultaneously with, biomass fractionation in the digester, which may help reduce downstream lignin precipitation. The purified cellulose may be utilized for making cellulose materials or cellulose derivatives, or for hydrolysis to produce glucose.



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PROCESS FOR PRODUCING PURIFIED CELLULOSE

PRIORITY DATA

[0001] This international PCT patent application claims priority to U.S. Patent App. No. 14/141,415, filed December 26, 2013, and to U.S. Provisional Patent App. No. 61/747,408, filed December 31, 2012, each of which is hereby incorporated by reference herein.

FIELD

[0002] The present invention generally relates to fractionation processes for converting lignocellulosic biomass into high-purity cellulose materials.

BACKGROUND

[0003] Biomass refining (or biorefining) is becoming more prevalent in industry. Cellulose fibers and sugars, hemicellulose sugars, lignin, syngas, and derivatives of these intermediates are being used by many companies for chemical and fuel production. Indeed, we now are observing the commercialization of integrated biorefineries that are capable of processing incoming biomass much the same as petroleum refineries now process crude oil. Underutilized lignocellulosic biomass feedstocks have the potential to be much cheaper than petroleum, on a carbon basis, as well as much better from an environmental life-cycle standpoint.

[0004] Lignocellulosic biomass is the most abundant renewable material on the planet and has long been recognized as a potential feedstock for producing chemicals, fuels, and materials. Lignocellulosic biomass normally comprises

primarily cellulose, hemicellulose, and lignin. Cellulose and hemicellulose are natural polymers of sugars, and lignin is an aromatic/aliphatic hydrocarbon polymer reinforcing the entire biomass network. Some forms of biomass (e.g., recycled materials) do not contain hemicellulose.

[0005] There are many reasons why it would be beneficial to process biomass in a way that effectively separates the major fractions (cellulose, hemicellulose, and lignin) from each other. Cellulose from biomass can be used in industrial cellulose applications directly, such as to make paper or other pulp-derived products. The cellulose can also be subjected to further processing to either modify the cellulose in some way or convert it into glucose. Hemicellulose sugars can be fermented to a variety of products, such as ethanol, or converted to other chemicals. Lignin from biomass has value as a solid fuel and also as an energy feedstock to produce liquid fuels, synthesis gas, or hydrogen; and as an intermediate to make a variety of polymeric compounds. Additionally, minor components such as proteins or rare sugars can be extracted and purified for specialty applications.

[0006] In light of this objective, a major shortcoming of previous process technologies is that one or two of the major components can be economically recovered in high yields, but not all three. Either the third component is sacrificially degraded in an effort to produce the other two components, or incomplete fractionation is accomplished. An important example is traditional biomass pulping (to produce paper and related goods). Cellulose is recovered in high yields, but lignin is primarily consumed by oxidation and hemicellulose sugars are mostly degraded. Approximately half of the starting biomass is essentially wasted in this manufacturing process. State-of-the-art biomass-pretreatment approaches typically can produce high yields of hemicellulose sugars but suffer from moderate cellulose and lignin yields.

[0007] There are several possible pathways to convert biomass into intermediates. One thermochemical pathway converts the feedstock into syngas (CO and H₂) through gasification or partial oxidation. Another thermochemical pathway converts biomass into liquid bio-oils through pyrolysis and separation. These are both high-temperature processes that intentionally destroy sugars in biomass.

[0008] Sugars (e.g., glucose and xylose) are desirable platform molecules because they can be fermented to a wide variety of fuels and chemicals, used to grow

organisms or produce enzymes, converted catalytically to chemicals, or recovered and sold to the market. To recover sugars from biomass, the cellulose and/or the hemicellulose in the biomass must be hydrolyzed into sugars. This is a difficult task because lignin and hemicelluloses are bound to each other by covalent bonds, and the three components are arranged inside the fiber wall in a complex manner. This recalcitrance explains the natural resistance of woody biomass to decomposition, and explains the difficulty to convert biomass to sugars at high yields.

[0009] Fractionation of biomass into its principle components (cellulose, hemicellulose, and lignin) has several advantages. Fractionation of lignocellulosics leads to release of cellulosic fibers and opens the cell wall structure by dissolution of lignin and hemicellulose between the cellulose microfibrils. The fibers become more accessible for hydrolysis by enzymes. When the sugars in lignocellulosics are used as feedstock for fermentation, the process to open up the cell wall structure is often called “pretreatment.” Pretreatment can significantly impact the production cost of lignocellulosic ethanol.

[0010] One of the most challenging technical obstacles for cellulose has been its recalcitrance towards hydrolysis for glucose production. Because of the high quantity of enzymes typically required, the enzyme cost can be a tremendous burden on the overall cost to turn cellulose into glucose for fermentation. Cellulose can be made to be reactive by subjecting biomass to severe chemistry, but that would jeopardize not only its integrity for other potential uses but also the yields of hemicellulose and lignin.

[0011] Many types of pretreatment have been studied. A common chemical pretreatment process employs a dilute acid, usually sulfuric acid, to hydrolyze and extract hemicellulose sugars and some lignin. A common physical pretreatment process employs steam explosion to mechanically disrupt the cellulose fibers and promote some separation of hemicellulose and lignin. Combinations of chemical and physical pretreatments are possible, such as acid pretreatment coupled with mechanical refining. It is difficult to avoid degradation of sugars. In some cases, severe pretreatments (i.e., high temperature and/or low pH) intentionally dehydrate sugars to furfural, levulinic acid, and related chemicals. Also, in common acidic

pretreatment approaches, lignin handling is very problematic because acid-condensed lignin precipitates and forms deposits on surfaces throughout the process.

[0012] One type of pretreatment that can overcome many of these disadvantages is called “organosolv” pretreatment. Organosolv refers to the presence of an organic solvent for lignin, which allows the lignin to remain soluble for better lignin handling. Traditionally, organosolv pretreatment or pulping has employed ethanol-water solutions to extract most of the lignin but leave much of the hemicellulose attached to the cellulose. For some market pulps, it is acceptable or desirable to have high hemicellulose content in the pulp. When high sugar yields are desired, however, there is a problem. Traditional ethanol/water pulping cannot give high yields of hemicellulose sugars because the timescale for sufficient hydrolysis of hemicellulose to monomers causes soluble-lignin polymerization and then precipitation back onto cellulose, which negatively impacts both pulp quality as well as cellulose enzymatic digestibility.

[0013] An acid catalyst can be introduced into organosolv pretreatment to attempt to hydrolyze hemicellulose into monomers while still obtaining the solvent benefit. Conventional organosolv wisdom dictates that high delignification can be achieved, but that a substantial fraction of hemicellulose must be left in the solids because any catalyst added to hydrolyze the hemicellulose will necessarily degrade the sugars (e.g., to furfural) during extraction of residual lignin.

[0014] Contrary to the conventional wisdom, it has been found that fractionation with a solution of ethanol (or another solvent for lignin), water, and sulfur dioxide (SO₂) can simultaneously achieve several important objectives. The fractionation can be achieved at modest temperatures (e.g., 120-160°C). The SO₂ can be easily recovered and reused. This process is able to effectively fractionate many biomass species, including softwoods, hardwoods, agricultural residues, and waste biomass. The SO₂ hydrolyzes the hemicelluloses and reduces or eliminates troublesome lignin-based precipitates. The presence of ethanol leads to rapid impregnation of the biomass, so that neither a separate impregnation stage nor size reduction smaller than wood chips are needed, thereby avoiding electricity-consuming sizing operations. The dissolved hemicelluloses are neither dehydrated nor oxidized (Iakovlev, “SO₂-ethanol-water fractionation of lignocellulosics,” Ph.D. Thesis, Aalto

Univ., Espoo, Finland, 2011). Cellulose is fully retained in the solid phase and can subsequently be hydrolyzed to glucose. The mixture of hemicellulose monomer sugars and cellulose-derived glucose may be used for production of biofuels and chemicals.

[0015] Commercial sulfite pulping has been practiced since 1874. The focus of sulfite pulping is the preservation of cellulose. In an effort to do that, industrial variants of sulfite pulping take 6-10 hours to dissolve hemicelluloses and lignin, producing a low yield of fermentable sugars. Stronger acidic cooking conditions that hydrolyze the hemicellulose to produce a high yield of fermentable sugars also hydrolyze the cellulose, and therefore the cellulose is not preserved.

[0016] The dominant pulping process today is the Kraft process. Kraft pulping does not fractionate lignocellulosic material into its primary components. Instead, hemicellulose is degraded in a strong solution of sodium hydroxide with or without sodium sulfide. The cellulose pulp produced by the Kraft process is high quality, essentially at the expense of both hemicellulose and lignin.

[0017] Sulfite pulping produces spent cooking liquor termed sulfite liquor. Fermentation of sulfite liquor to hemicellulosic ethanol has been practiced primarily to reduce the environmental impact of the discharges from sulfite mills since 1909. However, ethanol yields do not exceed one-third of the original hemicellulose component. Ethanol yield is low due to the incomplete hydrolysis of the hemicelluloses to fermentable sugars and further compounded by sulfite pulping side products, such as furfural, methanol, acetic acid, and others fermentation inhibitors.

[0018] Solvent cooking chemicals have been attempted as an alternative to Kraft or sulfite pulping. The original solvent process is described in U.S. Patent No. 1,856,567 by Kleinert et al. Groombridge et al. in U.S. Patent No. 2,060,068 showed that an aqueous solvent with sulfur dioxide is a potent delignifying system to produce cellulose from lignocellulosic material. Three demonstration facilities for ethanol-water (Alcell), alkaline sulfite with anthraquinone and methanol (ASAM), and ethanol-water-sodium hydroxide (Organocell) were operated briefly in the 1990s.

[0019] In view of the state of the art, what is desired is to efficiently fractionate any lignocellulosic-based biomass (including, in particular, softwoods) into its primary components so that each can be used in potentially distinct processes.

While not all commercial products require pure forms of cellulose, hemicellulose, or lignin, a platform biorefinery technology that enables processing flexibility in downstream optimization of product mix, is particularly desirable. An especially flexible fractionation technique would not only separate most of the hemicellulose and lignin from the cellulose, but also render the cellulose highly reactive to cellulase enzymes for the manufacture of fermentable glucose.

[0020] The AVAP® fractionation process developed by American Process, Inc. and its affiliates is able to economically accomplish these objectives. Improvements are still desired to further reduce enzyme requirements and costs for cellulose hydrolysis, and for generating cellulose with very low lignin content.

SUMMARY

[0021] The present invention addresses the aforementioned needs in the art.

[0022] In some variations, the invention provides a process for producing purified cellulose, the process comprising:

- (a) providing a feedstock comprising lignocellulosic biomass;
- (b) contacting the feedstock with sulfur dioxide or sulfurous acid, water, and a solvent for lignin, to produce intermediate solids and a liquid phase comprising hemicelluloses and lignin;
- (c) bleaching the intermediate solids to further delignify the intermediate solids, thereby generating cellulose-rich solids;
- (d) washing the cellulose-rich solids to generate purified cellulose; and
- (e) recovering the purified cellulose.

[0023] In some embodiments, the bleaching is conducted with one or more chemicals selected from the group consisting of sodium hypochlorite, calcium hypochlorite, chlorine dioxide, ozone, hydrogen peroxide, sodium percarbonate, sodium perborate, organic peroxides, peracetic acid, sodium dithionite, and combinations thereof. In some embodiments, the bleaching utilizes lignin-modifying enzymes. Various combinations or sequences may be employed, as is known in the

bleaching art. Also, the washing in step (d) may be combined with the bleaching in step (c).

[0024] In some embodiments, the process further comprises hydrolyzing the purified cellulose to glucose. In these or other embodiments, the process may include converting the purified cellulose to a cellulose derivative.

[0025] In some embodiments, the process further comprises hydrolyzing the hemicelluloses to produce monomeric sugars. The lignin produced in step (b) may be recovered, in some embodiments.

[0026] In other variations, bleaching is conducted prior to the digester. In these embodiments, a process for producing purified cellulose comprises:

- (a) providing a feedstock comprising lignocellulosic biomass;
- (b) bleaching the feedstock to generate intermediate solids;
- (c) contacting the intermediate solids with sulfur dioxide or sulfurous acid, water, and a solvent for lignin, to produce cellulose-rich solids and a liquid phase comprising hemicelluloses and lignin;
- (d) washing the cellulose-rich solids to generate purified cellulose; and
- (e) recovering the purified cellulose.

[0027] The bleaching may be conducted with one or more chemicals selected from the group consisting of sodium hypochlorite, calcium hypochlorite, chlorine dioxide, ozone, hydrogen peroxide, sodium percarbonate, sodium perborate, organic peroxides, peracetic acid, sodium dithionite, and combinations thereof. In some embodiments, including when bleaching agents are used, the bleaching utilizes lignin-modifying enzymes.

[0028] In some embodiments, the process further comprises hydrolyzing the purified cellulose to glucose. In these or other embodiments, the process may include converting the purified cellulose to a cellulose derivative.

[0029] In some embodiments, the process further comprises hydrolyzing the hemicelluloses to produce monomeric sugars. The lignin produced in step (c) may be recovered, in some embodiments.

[0030] In certain embodiments, a bleaching additive is introduced in step (b) and is not removed. The bleaching agent therefore passes to step (c). The bleaching additive may be effective to reduce lignin precipitation during or downstream of step

(c). In some embodiments, a bleaching additive is introduced directly into a digester, so that steps (b) and (c) are combined.

BRIEF DESCRIPTION OF THE FIGURES

[0031] FIG. 1 is an exemplary block-flow diagram according to some variations of the invention, incorporating bleaching after the initial fractionation.

[0032] FIG. 2 is an exemplary block-flow diagram according to some variations of the invention, incorporating a bleaching agent prior to fractionation.

DETAILED DESCRIPTION OF SOME EMBODIMENTS

[0033] This description will enable one skilled in the art to make and use the invention, and it describes several embodiments, adaptations, variations, alternatives, and uses of the invention. These and other embodiments, features, and advantages of the present invention will become more apparent to those skilled in the art when taken with reference to the following detailed description of the invention in conjunction with any accompanying drawings.

[0034] As used in this specification and the appended claims, the singular forms “a,” “an,” and “the” include plural referents unless the context clearly indicates otherwise. Unless defined otherwise, all technical and scientific terms used herein have the same meaning as is commonly understood by one of ordinary skill in the art to which this invention belongs. All composition numbers and ranges based on percentages are weight percentages, unless indicated otherwise. All ranges of numbers or conditions are meant to encompass any specific value contained within the range, rounded to any suitable decimal point.

[0035] Unless otherwise indicated, all numbers expressing parameters, reaction conditions, concentrations of components, and so forth used in the specification and claims are to be understood as being modified in all instances by the

term “about.” Accordingly, unless indicated to the contrary, the numerical parameters set forth in the following specification and attached claims are approximations that may vary depending at least upon a specific analytical technique.

[0036] The term “comprising,” which is synonymous with “including,” “containing,” or “characterized by” is inclusive or open-ended and does not exclude additional, unrecited elements or method steps. “Comprising” is a term of art used in claim language which means that the named claim elements are essential, but other claim elements may be added and still form a construct within the scope of the claim.

[0037] As used herein, the phrase “consisting of” excludes any element, step, or ingredient not specified in the claim. When the phrase “consists of” (or variations thereof) appears in a clause of the body of a claim, rather than immediately following the preamble, it limits only the element set forth in that clause; other elements are not excluded from the claim as a whole. As used herein, the phrase “consisting essentially of” limits the scope of a claim to the specified elements or method steps, plus those that do not materially affect the basis and novel characteristic(s) of the claimed subject matter.

[0038] With respect to the terms “comprising,” “consisting of,” and “consisting essentially of,” where one of these three terms is used herein, the presently disclosed and claimed subject matter may include the use of either of the other two terms. Thus in some embodiments not otherwise explicitly recited, any instance of “comprising” may be replaced by “consisting of” or, alternatively, by “consisting essentially of.”

[0039] The present invention, in some variations, is premised on the realization that mild bleaching may be incorporated into the AVAP® process to further reduce the lignin content of cellulose-rich solids.

[0040] This disclosure describes processes and apparatus to efficiently fractionate any lignocellulosic-based biomass into its primary major components (cellulose, lignin, and if present, hemicellulose) so that each can be used in potentially distinct processes. An advantage of the process is that it produces cellulose-rich solids while concurrently producing a liquid phase containing a high yield of both hemicellulose sugars and lignin, and low quantities of lignin and hemicellulose degradation products. The flexible fractionation technique enables multiple uses for

the products. The cellulose is highly reactive to cellulase enzymes for the manufacture of glucose. Other uses for celluloses can be adjusted based on market conditions.

[0041] Certain exemplary embodiments of the invention will now be described. These embodiments are not intended to limit the scope of the invention as claimed. The order of steps may be varied, some steps may be omitted, and/or other steps may be added. Reference herein to first step, second step, etc. is for illustration purposes only.

[0042] In some variations, such as depicted in FIG. 1, the invention provides a process for producing purified cellulose, the process comprising:

- (a) providing a feedstock comprising lignocellulosic biomass;
- (b) contacting the feedstock with sulfur dioxide, water, and a solvent for lignin, to produce intermediate solids and a liquid phase comprising hemicelluloses and lignin;
- (c) bleaching the intermediate solids to further delignify the intermediate solids, thereby generating cellulose-rich solids;
- (d) washing the cellulose-rich solids to generate purified cellulose; and
- (e) recovering the purified cellulose.

[0043] In some embodiments, the bleaching is conducted with one or more chemicals selected from the group consisting of sodium hypochlorite, calcium hypochlorite, chlorine dioxide, ozone, hydrogen peroxide, sodium percarbonate, sodium perborate, organic peroxides, peracetic acid, sodium dithionite, and combinations thereof. In some embodiments, the bleaching utilizes lignin-modifying enzymes. Various combinations or sequences may be employed, as is known in the bleaching art. Bleaching technology is well-known; see, for example, Biermann, *Essentials of Pulping and Papermaking*, San Diego: Academic Press, Inc., 1993, which is hereby incorporated by reference herein.

[0044] Preferably, the bleaching conditions are relatively mild. Severe bleaching is not necessary because most of the lignin has typically been removed in the digester, in step (b). Optimization may be performed to determine the best combination of digester severity (e.g., time, temperature, or SO₂ concentration) and bleach-unit severity (e.g., time, temperature, and bleaching-agent concentration).

Thus, if desired, relatively mild fractionation conditions may be selected for the digester, coupled with relatively severe bleaching conditions, to achieve a desired level of delignification (i.e., lignin removal from the starting feedstock).

[0045] In various embodiments, a total delignification of at least about 80%, 81%, 82%, 83%, 84%, 85%, 86%, 87%, 88%, 89%, 90%, 91%, 92%, 93%, 94%, 95%, 96%, 97%, 98%, or 99% is achieved. In these embodiments, the digesting (fractionating) in step (b) may achieve at least about 50%, 55%, 60%, 65%, 70%, 75%, 80%, 85%, 90%, or 95% of the total delignification, while bleaching may achieve at least about 5%, 10%, 15%, 20%, 25%, 30%, 35%, 40%, 45%, or 50% of the total delignification. The purified cellulose contains less than about 5 wt%, 4 wt%, 3 wt%, 2 wt%, or 1 wt% lignin, in certain embodiments.

[0046] In various embodiments, bleaching increases the total delignification of solids by at least about 1%, 2%, 5%, 10%, 15%, 20%, 25%, or more, calculated as percentage points above the digesting delignification level. As an example, digesting may achieve 85% delignification and bleaching may result in 95% total delignification, which is an increase of 10% above the digesting delignification level.

[0047] The ratio of the mass of lignin removed during bleaching, divided by the mass of lignin removed during digesting, may be varied, such as about 0.01, 0.02, 0.03, 0.04, 0.05, 0.1, 0.15, 0.2, 0.25, 0.3, 0.35, 0.4, 0.45, 0.5, 0.55, 0.6, 0.65, 0.7, 0.75, 0.8, 0.85, 0.9, 0.95, or 1.0. Typically, this ratio will be about 0.2 or less, preferably about 0.1 or less.

[0048] The washing in step (d) may be combined with the bleaching in step (c). The washing and bleaching may be carried out in the same unit, or simultaneously or sequentially. In some embodiments, a countercurrent configuration is employed, where a dilute bleaching solution contacts delignified solids and the solution flows countercurrent to the direction of flow of solids. Bleaching chemicals are consumed in chemical reactions, so the concentration of the bleaching solution is reduced. At the feed of the unit, the concentration is very dilute, in some embodiments. Optionally, water is introduced to further aid washing.

[0049] In some embodiments, the process further comprises hydrolyzing the purified cellulose to glucose. In these or other embodiments, the process may include converting the purified cellulose to a cellulose derivative.

[0050] In some embodiments, the process further comprises hydrolyzing the hemicelluloses to produce monomeric sugars. The lignin produced in step (b) may be recovered, in some embodiments.

[0051] In other variations, such as depicted in FIG. 2, bleaching is conducted prior to the digester. In these embodiments, a process for producing purified cellulose comprises:

- (a) providing a feedstock comprising lignocellulosic biomass;
- (b) bleaching the feedstock to generate intermediate solids;
- (c) contacting the intermediate solids with sulfur dioxide, water, and a solvent for lignin, to produce cellulose-rich solids and a liquid phase comprising hemicelluloses and lignin;
- (d) washing the cellulose-rich solids to generate purified cellulose; and
- (e) recovering the purified cellulose.

[0052] The bleaching may be conducted with one or more chemicals selected from the group consisting of sodium hypochlorite, calcium hypochlorite, chlorine dioxide, ozone, hydrogen peroxide, sodium percarbonate, sodium perborate, organic peroxides, peracetic acid, sodium dithionite, and combinations thereof. In some embodiments, including when bleaching agents are used, the bleaching utilizes lignin-modifying enzymes.

[0053] The bleaching conditions may be selected along with the digester conditions, step (c), for economic optimization or other reasons. Mild bleaching in step (b) may be conducted to remove some lignin, so that in step (c), less solvent is necessary, for example. More-severe bleaching (e.g., higher temperature or more bleaching agent) in step (b) may be conducted to remove most of the lignin, so that in step (c), little or no solvent is necessary.

[0054] In various embodiments, a total delignification of at least about 80%, 81%, 82%, 83%, 84%, 85%, 86%, 87%, 88%, 89%, 90%, 91%, 92%, 93%, 94%, 95%, 96%, 97%, 98%, or 99% is achieved. In these embodiments, the bleaching in step (b) may achieve at least about 5%, 10%, 15%, 20%, 25%, 30%, 35%, 40%, 45%, 50%, 55%, 60%, 65%, 70%, 75%, 80%, 85%, 90%, or 95% of the total delignification, while digesting (fractionating) in step (c) may achieve at least about 95%, 90%, 85%, 80%, 75%, 70%, 65%, 60%, 55%, 50%, 45%, 40%, 35%, 30%,

25%, 20%, 15%, 10%, or 5% of the total delignification. The purified cellulose contains less than about 5 wt%, 4 wt%, 3 wt%, 2 wt%, or 1 wt% lignin, in certain embodiments

[0055] In various embodiments, bleaching increases the total delignification of solids by at least about 1%, 2%, 5%, 10%, 15%, 20%, 25%, or more, calculated as percentage points above the digesting delignification level. As an example, bleaching may result in 10% delignification and digesting may achieve 95% total delignification, which means the bleaching increased the total delignification by 10%.

[0056] The ratio of the mass of lignin removed during bleaching, divided by the mass of lignin removed during digesting, may be varied, such as about 0.01, 0.02, 0.03, 0.04, 0.05, 0.1, 0.15, 0.2, 0.25, 0.3, 0.35, 0.4, 0.45, 0.5, 0.55, 0.6, 0.65, 0.7, 0.75, 0.8, 0.85, 0.9, 0.95, or 1.0. Typically, this ratio will be about 0.2 or less, preferably about 0.1 or less.

[0057] In some embodiments, the process further comprises hydrolyzing the purified cellulose to glucose. In these or other embodiments, the process may include converting the purified cellulose to a cellulose derivative.

[0058] In some embodiments, the process further comprises hydrolyzing the hemicelluloses to produce monomeric sugars. The lignin produced in step (c) may be recovered, in some embodiments.

[0059] In certain embodiments, a bleaching additive is introduced in step (b) and is not removed. The bleaching agent therefore passes to step (c). The bleaching additive may be effective to increase delignification and/or reduce lignin precipitation during or downstream of step (c).

[0060] In other embodiments (different than FIGS. 1 or 2), a bleaching agent is added directly to the digester, without a separate bleaching unit or step. The bleaching agent for direct addition may be selected from the group consisting sodium hypochlorite, calcium hypochlorite, chlorine dioxide, ozone, hydrogen peroxide, sodium percarbonate, sodium perborate, organic peroxides, peracetic acid, sodium dithionite, lignin-modifying enzymes, and combinations thereof. Other additives may be desired, such as for pH adjustment.

[0061] The biomass feedstock may be selected from hardwoods, softwoods, forest residues, industrial wastes, pulp and paper wastes, consumer wastes, or

combinations thereof. Some embodiments utilize agricultural residues, which include lignocellulosic biomass associated with food crops, annual grasses, energy crops, or other annually renewable feedstocks. Exemplary agricultural residues include, but are not limited to, corn stover, corn fiber, wheat straw, sugarcane bagasse, sugarcane straw, rice straw, oat straw, barley straw, miscanthus, energy cane straw/residue, or combinations thereof.

[0062] As used herein, “lignocellulosic biomass” means any material containing cellulose and lignin. Lignocellulosic biomass may also contain hemicellulose. Mixtures of one or more types of biomass can be used. In some embodiments, the biomass feedstock comprises both a lignocellulosic component (such as one described above) in addition to a sucrose-containing component (e.g., sugarcane or energy cane) and/or a starch component (e.g., corn, wheat, rice, etc.).

[0063] Various moisture levels may be associated with the starting biomass. The biomass feedstock need not be, but may be, relatively dry. In general, the biomass is in the form of a particulate or chip, but particle size is not critical in this invention.

[0064] Reaction conditions and operation sequences may vary widely. Some embodiments employ conditions described in U.S. Patent No. 8,030,039, issued Oct. 4, 2011; U.S. Patent No. 8,038,842, issued Oct. 11, 2011; U.S. Patent No. 8,268,125, issued Sept. 18, 2012; and U.S. Patent App. Nos. 13/004,431; 12/234,286; 13/585,710; 12/250,734; 12/397,284; 12/304,046; 13/500,916; 13/626,220; 12/854,869; 61/732,047; 61/735,738; 61/739,343; 61/747,010; 61/747,105; 61/747,376; 61/747,379; and 61/747,382 including the prosecution histories thereof. Each of these commonly owned patent applications is hereby incorporated by reference herein in its entirety. In some embodiments, the process is a variation of the AVAP® process technology which is commonly owned with the assignee of this patent application.

[0065] Some variations may be understood with reference to FIGS. 1 and 2. Dotted lines denote optional streams. Various embodiments will now be further described, without limitation as to the scope of the invention. These embodiments are exemplary in nature.

[0066] In some embodiments, a first process step is “cooking” (equivalently, “digesting”) which fractionates the three lignocellulosic material components (cellulose, hemicellulose, and lignin) to allow easy downstream removal. Specifically, hemicelluloses are dissolved and over 50% are completely hydrolyzed; cellulose is separated but remains resistant to hydrolysis; and part of the lignin is sulfonated into water-soluble lignosulfonates.

[0067] The lignocellulosic material is processed in a solution (cooking liquor) of aliphatic alcohol, water, and sulfur dioxide. The cooking liquor preferably contains at least 10 wt%, such as at least 20 wt%, 30 wt%, 40 wt%, or 50 wt% of a solvent for lignin. For example, the cooking liquor may contain about 30-70 wt% solvent, such as about 50 wt% solvent. The solvent for lignin may be an aliphatic alcohol, such as methanol, ethanol, 1-propanol, 2-propanol, 1-butanol, 2-butanol, isobutanol, 1-pentanol, 1-hexanol, or cyclohexanol. The solvent for lignin may be an aromatic alcohol, such as phenol or cresol. Other lignin solvents are possible, such as (but not limited to) glycerol, methyl ethyl ketone, or diethyl ether. Combinations of more than one solvent may be employed.

[0068] Preferably, enough solvent is included in the extractant mixture to dissolve the lignin present in the starting material. The solvent for lignin may be completely miscible, partially miscible, or immiscible with water, so that there may be more than one liquid phase. Potential process advantages arise when the solvent is miscible with water, and also when the solvent is immiscible with water. When the solvent is water-miscible, a single liquid phase forms, so mass transfer of lignin and hemicellulose extraction is enhanced, and the downstream process must only deal with one liquid stream. When the solvent is immiscible in water, the extractant mixture readily separates to form liquid phases, so a distinct separation step can be avoided or simplified. This can be advantageous if one liquid phase contains most of the lignin and the other contains most of the hemicellulose sugars, as this facilitates recovering the lignin from the hemicellulose sugars.

[0069] The cooking liquor preferably contains sulfur dioxide and/or sulfurous acid (H_2SO_3). The cooking liquor preferably contains SO_2 , in dissolved or reacted form, in a concentration of at least 3 wt%, preferably at least 6 wt%, more preferably at least 8 wt%, such as about 9 wt%, 10 wt%, 11 wt%, 12 wt%, 13 wt%, 14 wt%, 15

wt%, 20 wt%, 25 wt%, 30 wt% or higher. The cooking liquor may also contain one or more species, separately from SO₂, to adjust the pH. The pH of the cooking liquor is typically about 4 or less.

[0070] Sulfur dioxide is a preferred acid catalyst, because it can be recovered easily from solution after hydrolysis. The majority of the SO₂ from the hydrolysate may be stripped and recycled back to the reactor. Recovery and recycling translates to less lime required compared to neutralization of comparable sulfuric acid, less solids to dispose of, and less separation equipment. The increased efficiency owing to the inherent properties of sulfur dioxide mean that less total acid or other catalysts may be required. This has cost advantages, since sulfuric acid can be expensive. Additionally, and quite significantly, less acid usage also will translate into lower costs for a base (e.g., lime) to increase the pH following hydrolysis, for downstream operations. Furthermore, less acid and less base will also mean substantially less generation of waste salts (e.g., gypsum) that may otherwise require disposal.

[0071] In some embodiments, an additive may be included in amounts of about 0.1 wt% to 10 wt% or more to increase cellulose viscosity. Exemplary additives include ammonia, ammonia hydroxide, urea, anthraquinone, magnesium oxide, magnesium hydroxide, sodium hydroxide, and their derivatives.

[0072] The cooking is performed in one or more stages using batch or continuous digestors. Solid and liquid may flow cocurrently or countercurrently, or in any other flow pattern that achieves the desired fractionation. The cooking reactor may be internally agitated, if desired.

[0073] Depending on the lignocellulosic material to be processed, the cooking conditions are varied, with temperatures from about 65°C to 175°C, for example 75°C, 85°C, 95°C, 105°C, 115°C, 125°C, 130°C, 135°C, 140°C, 145°C, 150°C, 155°C, 165°C or 170°C, and corresponding pressures from about 1 atmosphere to about 15 atmospheres in the liquid or vapor phase. The cooking time of one or more stages may be selected from about 15 minutes to about 720 minutes, such as about 30, 45, 60, 90, 120, 140, 160, 180, 250, 300, 360, 450, 550, 600, or 700 minutes. Generally, there is an inverse relationship between the temperature used during the digestion step and the time needed to obtain good fractionation of the biomass into its constituent parts.

[0074] The cooking liquor to lignocellulosic material ratio may be selected from about 1 to about 10, such as about 2, 3, 4, 5, or 6. In some embodiments, biomass is digested in a pressurized vessel with low liquor volume (low ratio of cooking liquor to lignocellulosic material), so that the cooking space is filled with ethanol and sulfur dioxide vapor in equilibrium with moisture. The cooked biomass is washed in alcohol-rich solution to recover lignin and dissolved hemicelluloses, while the remaining pulp is further processed. In some embodiments, the process of fractionating lignocellulosic material comprises vapor-phase cooking of lignocellulosic material with aliphatic alcohol (or other solvent for lignin), water, and sulfur dioxide. See, for example, U.S. Patent Nos. 8,038,842 and 8,268,125 which are incorporated by reference herein.

[0075] A portion or all of the sulfur dioxide may be present as sulfurous acid in the extract liquor. In certain embodiments, sulfur dioxide is generated *in situ* by introducing sulfurous acid, sulfite ions, bisulfite ions, combinations thereof, or a salt of any of the foregoing. Excess sulfur dioxide, following hydrolysis, may be recovered and reused.

[0076] In some embodiments, sulfur dioxide is saturated in water (or aqueous solution, optionally with an alcohol) at a first temperature, and the hydrolysis is then carried out at a second, generally higher, temperature. In some embodiments, sulfur dioxide is sub-saturated. In some embodiments, sulfur dioxide is super-saturated. In some embodiments, sulfur dioxide concentration is selected to achieve a certain degree of lignin sulfonation, such as 1%, 2%, 3%, 4%, 5%, 6%, 7%, 8%, 9%, or 10% sulfur content. SO₂ reacts chemically with lignin to form stable lignosulfonic acids which may be present both in the solid and liquid phases.

[0077] The concentration of sulfur dioxide, additives, and aliphatic alcohol (or other solvent) in the solution and the time of cook may be varied to control the yield of cellulose and hemicellulose in the pulp. The concentration of sulfur dioxide and the time of cook may be varied to control the yield of lignin versus lignosulfonates in the hydrolysate. In some embodiments, the concentration of sulfur dioxide, temperature, and the time of cook may be varied to control the yield of fermentable sugars.

[0078] Once the desired amount of fractionation of both hemicellulose and lignin from the solid phase is achieved, the liquid and solid phases are separated. Conditions for the separation may be selected to minimize the reprecipitation of the extracted lignin on the solid phase. This is favored by conducting separation or washing at a temperature of at least the glass-transition temperature of lignin (about 120°C).

[0079] The physical separation can be accomplished either by transferring the entire mixture to a device that can carry out the separation and washing, or by removing only one of the phases from the reactor while keeping the other phase in place. The solid phase can be physically retained by appropriately sized screens through which liquid can pass. The solid is retained on the screens and can be kept there for successive solid-wash cycles. Alternately, the liquid may be retained and solid phase forced out of the reaction zone, with centrifugal or other forces that can effectively transfer the solids out of the slurry. In a continuous system, countercurrent flow of solids and liquid can accomplish the physical separation.

[0080] The recovered solids normally will contain a quantity of lignin and sugars, some of which can be removed easily by washing. The washing-liquid composition can be the same as or different than the liquor composition used during fractionation. Multiple washes may be performed to increase effectiveness. Preferably, one or more washes are performed with a composition including a solvent for lignin, to remove additional lignin from the solids, followed by one or more washes with water to displace residual solvent and sugars from the solids. Recycle streams, such as from solvent-recovery operations, may be used to wash the solids.

[0081] After separation and washing as described, a solid phase and at least one liquid phase are obtained. The solid phase contains substantially undigested cellulose. A single liquid phase is usually obtained when the solvent and the water are miscible in the relative proportions that are present. In that case, the liquid phase contains, in dissolved form, most of the lignin originally in the starting lignocellulosic material, as well as soluble monomeric and oligomeric sugars formed in the hydrolysis of any hemicellulose that may have been present. Multiple liquid phases tend to form when the solvent and water are wholly or partially immiscible. The lignin tends to be contained in the liquid phase that contains most of the solvent.

Hemicellulose hydrolysis products tend to be present in the liquid phase that contains most of the water.

[0082] In some embodiments, hydrolysate from the cooking step is subjected to pressure reduction. Pressure reduction may be done at the end of a cook in a batch digester, or in an external flash tank after extraction from a continuous digester, for example. The flash vapor from the pressure reduction may be collected into a cooking liquor make-up vessel. The flash vapor contains substantially all the unreacted sulfur dioxide which may be directly dissolved into new cooking liquor. The cellulose is then removed to be washed and further treated as desired.

[0083] A process washing step recovers the hydrolysate from the cellulose. The washed cellulose is pulp that may be used for various purposes (e.g., paper or nanocellulose production). The weak hydrolysate from the washer continues to the final reaction step; in a continuous digester this weak hydrolysate may be combined with the extracted hydrolysate from the external flash tank. In some embodiments, washing and/or separation of hydrolysate and cellulose-rich solids is conducted at a temperature of at least about 100°C, 110°C, or 120°C. The washed cellulose may also be used for glucose production via cellulose hydrolysis with enzymes or acids.

[0084] In another reaction step, the hydrolysate may be further treated in one or multiple steps to hydrolyze the oligomers into monomers. This step may be conducted before, during, or after the removal of solvent and sulfur dioxide. The solution may or may not contain residual solvent (e.g. alcohol). In some embodiments, sulfur dioxide is added or allowed to pass through to this step, to assist hydrolysis. In these or other embodiments, an acid such as sulfurous acid or sulfuric acid is introduced to assist with hydrolysis. In some embodiments, the hydrolysate is autohydrolyzed by heating under pressure. In some embodiments, no additional acid is introduced, but lignosulfonic acids produced during the initial cooking are effective to catalyze hydrolysis of hemicellulose oligomers to monomers. In various embodiments, this step utilizes sulfur dioxide, sulfurous acid, sulfuric acid at a concentration of about 0.01 wt% to 30 wt%, such as about 0.05 wt%, 0.1 wt%, 0.2 wt%, 0.5 wt%, 1 wt%, 2 wt%, 5 wt%, 10 wt%, or 20 wt%. This step may be carried out at a temperature from about 100°C to 220°C, such as about 110°C, 120°C, 130°C,

140°C, 150°C, 160°C, 170°C, 180°C, 190°C, 200°C, or 210°C. Heating may be direct or indirect to reach the selected temperature.

[0085] The reaction step produces fermentable sugars which can then be concentrated by evaporation to a fermentation feedstock. Concentration by evaporation may be accomplished before, during, or after the treatment to hydrolyze oligomers. The final reaction step may optionally be followed by steam stripping of the resulting hydrolysate to remove and recover sulfur dioxide and alcohol, and for removal of potential fermentation-inhibiting side products. The evaporation process may be under vacuum or pressure, from about -0.1 atmospheres to about 10 atmospheres, such as about 0.1 atm, 0.3 atm, 0.5 atm, 1.0 atm, 1.5 atm, 2 atm, 4 atm, 6 atm, or 8 atm.

[0086] Recovering and recycling the sulfur dioxide may utilize separations such as, but not limited to, vapor-liquid disengagement (e.g. flashing), steam stripping, extraction, or combinations or multiple stages thereof. Various recycle ratios may be practiced, such as about 0.1, 0.2, 0.3, 0.4, 0.5, 0.6, 0.7, 0.8, 0.9, 0.95, or more. In some embodiments, about 90-99% of initially charged SO₂ is readily recovered by distillation from the liquid phase, with the remaining 1-10% (e.g., about 3-5%) of the SO₂ primarily bound to dissolved lignin in the form of lignosulfonates.

[0087] In a preferred embodiment, the evaporation step utilizes an integrated alcohol stripper and evaporator. Evaporated vapor streams may be segregated so as to have different concentrations of organic compounds in different streams. Evaporator condensate streams may be segregated so as to have different concentrations of organic compounds in different streams. Alcohol may be recovered from the evaporation process by condensing the exhaust vapor and returning to the cooking liquor make-up vessel in the cooking step. Clean condensate from the evaporation process may be used in the washing step.

[0088] In some embodiments, an integrated alcohol stripper and evaporator system is employed, wherein aliphatic alcohol is removed by vapor stripping, the resulting stripper product stream is concentrated by evaporating water from the stream, and evaporated vapor is compressed using vapor compression and is reused to provide thermal energy.

[0089] The hydrolysate from the evaporation and final reaction step contains mainly fermentable sugars but may also contain lignin depending on the location of lignin separation in the overall process configuration. The hydrolysate may be concentrated to a concentration of about 5 wt% to about 60 wt% solids, such as about 10 wt%, 15 wt%, 20 wt%, 25 wt%, 30 wt%, 35 wt%, 40 wt%, 45 wt%, 50 wt% or 55 wt% solids. The hydrolysate contains fermentable sugars.

[0090] Fermentable sugars are defined as hydrolysis products of cellulose, galactoglucomannan, glucomannan, arabinoglucuronoxylans, arabinogalactan, and glucuronoxylans into their respective short-chained oligomers and monomer products, i.e., glucose, mannose, galactose, xylose, and arabinose. The fermentable sugars may be recovered in purified form, as a sugar slurry or dry sugar solids, for example. Any known technique may be employed to recover a slurry of sugars or to dry the solution to produce dry sugar solids.

[0091] In some embodiments, the fermentable sugars are fermented to produce biochemicals or biofuels such as (but by no means limited to) ethanol, isopropanol, acetone, 1-butanol, isobutanol, lactic acid, succinic acid, or any other fermentation products. Some amount of the fermentation product may be a microorganism or enzymes, which may be recovered if desired.

[0092] When the fermentation will employ bacteria, such as *Clostridia* bacteria, it is preferable to further process and condition the hydrolysate to raise pH and remove residual SO₂ and other fermentation inhibitors. The residual SO₂ (i.e., following removal of most of it by stripping) may be catalytically oxidized to convert residual sulfite ions to sulfate ions by oxidation. This oxidation may be accomplished by adding an oxidation catalyst, such as FeSO₄·7H₂O, that oxidizes sulfite ions to sulfate ions. Preferably, the residual SO₂ is reduced to less than about 100 ppm, 50 ppm, 25 ppm, 10 ppm, 5 ppm, or 1 ppm.

[0093] In some embodiments, the process further comprises recovering the lignin as a co-product. The sulfonated lignin may also be recovered as a co-product. In certain embodiments, the process further comprises combusting or gasifying the sulfonated lignin, recovering sulfur contained in the sulfonated lignin in a gas stream comprising reclaimed sulfur dioxide, and then recycling the reclaimed sulfur dioxide for reuse.

[0094] The process lignin separation step is for the separation of lignin from the hydrolysate and can be located before or after the final reaction step and evaporation. If located after, then lignin will precipitate from the hydrolysate since alcohol has been removed in the evaporation step. The remaining water-soluble lignosulfonates may be precipitated by converting the hydrolysate to an alkaline condition (pH higher than 7) using, for example, an alkaline earth oxide, preferably calcium oxide (lime). The combined lignin and lignosulfonate precipitate may be filtered. The lignin and lignosulfonate filter cake may be dried as a co-product or burned or gasified for energy production. The hydrolysate from filtering may be recovered and sold as a concentrated sugar solution product or further processed in a subsequent fermentation or other reaction step.

[0095] Native (non-sulfonated) lignin is hydrophobic, while lignosulfonates are hydrophilic. Hydrophilic lignosulfonates may have less propensity to clump, agglomerate, and stick to surfaces. Even lignosulfonates that do undergo some condensation and increase of molecular weight, will still have an HSO_3 group that will contribute some solubility (hydrophilic).

[0096] In some embodiments, the soluble lignin precipitates from the hydrolysate after solvent has been removed in the evaporation step. In some embodiments, reactive lignosulfonates are selectively precipitated from hydrolysate using excess lime (or other base, such as ammonia) in the presence of aliphatic alcohol. In some embodiments, hydrated lime is used to precipitate lignosulfonates. In some embodiments, part of the lignin is precipitated in reactive form and the remaining lignin is sulfonated in water-soluble form.

[0097] The process fermentation and distillation steps are intended for the production of fermentation products, such as alcohols or organic acids. After removal of cooking chemicals and lignin, and further treatment (oligomer hydrolysis), the hydrolysate contains mainly fermentable sugars in water solution from which any fermentation inhibitors have been preferably removed or neutralized. The hydrolysate is fermented to produce dilute alcohol or organic acids, from 1 wt% to 20 wt% concentration. The dilute product is distilled or otherwise purified as is known in the art.

[0098] When alcohol is produced, such as ethanol, some of it may be used for cooking liquor makeup in the process cooking step. Also, in some embodiments, a distillation column stream, such as the bottoms, with or without evaporator condensate, may be reused to wash cellulose. In some embodiments, lime may be used to dehydrate product alcohol. Side products may be removed and recovered from the hydrolysate. These side products may be isolated by processing the vent from the final reaction step and/or the condensate from the evaporation step. Side products include furfural, hydroxymethyl furfural (HMF), methanol, acetic acid, and lignin-derived compounds, for example.

[0099] The cellulose-rich material is highly reactive in the presence of industrial cellulase enzymes that efficiently break the cellulose down to glucose monomers. It has been found experimentally that the cellulose-rich material, which generally speaking is highly delignified, rapidly hydrolyzes to glucose with relatively low quantities of enzymes. For example, the cellulose-rich solids may be converted to glucose with at least 80% yield within 24 hours at 50°C and 2 wt% solids, in the presence of a cellulase enzyme mixture in an amount of no more than 15 filter paper units (FPU) per g of the solids. In some embodiments, this same conversion requires no more than 5 FPU per g of the solids.

[00100] The glucose may be fermented to an alcohol, an organic acid, or another fermentation product. The glucose may be used as a sweetener or isomerized to enrich its fructose content. The glucose may be used to produce baker's yeast. The glucose may be catalytically or thermally converted to various organic acids and other materials.

[00101] In some embodiments, the cellulose-rich material is further processed into one more cellulose products. Cellulose products include market pulp, dissolving pulp (also known as α -cellulose), fluff pulp, purified cellulose, paper, paper products, and so on. Further processing may include bleaching, if desired. Further processing may include modification of fiber length or particle size, such as when producing nanocellulose or nanofibrillated or microfibrillated cellulose. It is believed that the cellulose produced by this process is highly amenable to derivatization chemistry for cellulose derivatives and cellulose-based materials such as polymers.

[00102] When hemicellulose is present in the starting biomass, all or a portion of the liquid phase contains hemicellulose sugars and soluble oligomers. It is preferred to remove most of the lignin from the liquid, as described above, to produce a fermentation broth which will contain water, possibly some of the solvent for lignin, hemicellulose sugars, and various minor components from the digestion process. This fermentation broth can be used directly, combined with one or more other fermentation streams, or further treated. Further treatment can include sugar concentration by evaporation; addition of glucose or other sugars (optionally as obtained from cellulose saccharification); addition of various nutrients such as salts, vitamins, or trace elements; pH adjustment; and removal of fermentation inhibitors such as acetic acid and phenolic compounds. The choice of conditioning steps should be specific to the target product(s) and microorganism(s) employed.

[00103] In some embodiments, hemicellulose sugars are not fermented but rather are recovered and purified, stored, sold, or converted to a specialty product. Xylose, for example, can be converted into xylitol.

[00104] A lignin product can be readily obtained from a liquid phase using one or more of several methods. One simple technique is to evaporate off all liquid, resulting in a solid lignin-rich residue. This technique would be especially advantageous if the solvent for lignin is water-immiscible. Another method is to cause the lignin to precipitate out of solution. Some of the ways to precipitate the lignin include (1) removing the solvent for lignin from the liquid phase, but not the water, such as by selectively evaporating the solvent from the liquid phase until the lignin is no longer soluble; (2) diluting the liquid phase with water until the lignin is no longer soluble; and (3) adjusting the temperature and/or pH of the liquid phase. Methods such as centrifugation can then be utilized to capture the lignin. Yet another technique for removing the lignin is continuous liquid-liquid extraction to selectively remove the lignin from the liquid phase, followed by removal of the extraction solvent to recover relatively pure lignin.

[00105] Lignin produced in accordance with the invention can be used as a fuel. As a solid fuel, lignin is similar in energy content to coal. Lignin can act as an oxygenated component in liquid fuels, to enhance octane while meeting standards as a renewable fuel. The lignin produced herein can also be used as polymeric material,

and as a chemical precursor for producing lignin derivatives. The sulfonated lignin may be sold as a lignosulfonate product, or burned for fuel value.

[00106] The present invention also provides systems configured for carrying out the disclosed processes, and compositions produced therefrom. Any stream generated by the disclosed processes may be partially or completely recovered, purified or further treated, and/or marketed or sold.

[00107] In this detailed description, reference has been made to multiple embodiments of the invention and non-limiting examples relating to how the invention can be understood and practiced. Other embodiments that do not provide all of the features and advantages set forth herein may be utilized, without departing from the spirit and scope of the present invention. This invention incorporates routine experimentation and optimization of the methods and systems described herein. Such modifications and variations are considered to be within the scope of the invention defined by the claims.

[00108] All publications, patents, and patent applications cited in this specification are herein incorporated by reference in their entirety as if each publication, patent, or patent application were specifically and individually put forth herein.

[00109] Where methods and steps described above indicate certain events occurring in certain order, those of ordinary skill in the art will recognize that the ordering of certain steps may be modified and that such modifications are in accordance with the variations of the invention. Additionally, certain of the steps may be performed concurrently in a parallel process when possible, as well as performed sequentially.

[00110] Therefore, to the extent there are variations of the invention, which are within the spirit of the disclosure or equivalent to the inventions found in the appended claims, it is the intent that this patent will cover those variations as well. The present invention shall only be limited by what is claimed.

CLAIMS

What is claimed is:

1. A process for producing purified cellulose, said process comprising:
 - (a) providing a feedstock comprising lignocellulosic biomass;
 - (b) contacting said feedstock with sulfur dioxide or sulfurous acid, water, and a solvent for lignin, to produce delignified intermediate solids and a liquid phase comprising hemicelluloses and lignin;
 - (c) bleaching said intermediate solids to further delignify said intermediate solids, thereby generating cellulose-rich solids;
 - (d) washing said cellulose-rich solids to generate purified cellulose; and
 - (e) recovering said purified cellulose.
2. The process of claim 1, wherein said bleaching is conducted with one or more chemicals selected from the group consisting of sodium hypochlorite, calcium hypochlorite, chlorine dioxide, ozone, hydrogen peroxide, sodium percarbonate, sodium perborate, organic peroxides, peracetic acid, sodium dithionite, and combinations thereof.
3. The process of either one of claim 1, wherein said bleaching utilizes lignin-modifying enzymes.
4. The process of claim 1, wherein steps (c) and (d) are combined in a single unit operation.
5. The process of claim 1, wherein said purified cellulose contains less than about 2 wt% lignin.
6. The process of claim 1, said process further comprising converting said purified cellulose to a cellulose derivative.

7. The process of claim 1, said process further comprising converting said purified cellulose to nanocellulose.
8. The process of claim 1, said process further comprising hydrolyzing said purified cellulose to glucose.
9. The process of claim 1, said process further comprising hydrolyzing said hemicelluloses to monomeric sugars.
10. The process of claim 1, said process further comprising recovering said lignin.
11. A process for producing purified cellulose, said process comprising:
 - (a) providing a feedstock comprising lignocellulosic biomass;
 - (b) bleaching said feedstock to generate delignified intermediate solids;
 - (c) contacting said intermediate solids with sulfur dioxide or sulfurous acid, water, and a solvent for lignin, to produce cellulose-rich solids and a liquid phase comprising hemicelluloses and lignin;
 - (d) washing said cellulose-rich solids to generate purified cellulose; and
 - (e) recovering said purified cellulose.
12. The process of claim 11, wherein said bleaching is conducted with one or more chemicals selected from the group consisting of sodium hypochlorite, calcium hypochlorite, chlorine dioxide, ozone, hydrogen peroxide, sodium percarbonate, sodium perborate, organic peroxides, peracetic acid, sodium dithionite, and combinations thereof.
13. The process of claim 11, wherein said bleaching utilizes lignin-modifying enzymes.
14. The process of claim 11, wherein steps (c) and (d) are combined in a single unit operation.

15. The process of claim 11, wherein said purified cellulose contains less than about 2 wt% lignin.
16. The process of claim 11, said process further comprising converting said purified cellulose to a cellulose derivative.
17. The process of claim 11, said process further comprising converting said purified cellulose to nanocellulose.
18. The process of claim 11, said process further comprising hydrolyzing said purified cellulose to glucose.
19. The process of claim 11, said process further comprising hydrolyzing said hemicelluloses to produce monomeric sugars, and optionally recovering said lignin.
20. The process of claim 11, wherein a bleaching additive is introduced in step (b) and is not removed prior to step (c); and wherein said bleaching additive reduces lignin precipitation during or downstream of step (c).

FIG. 1

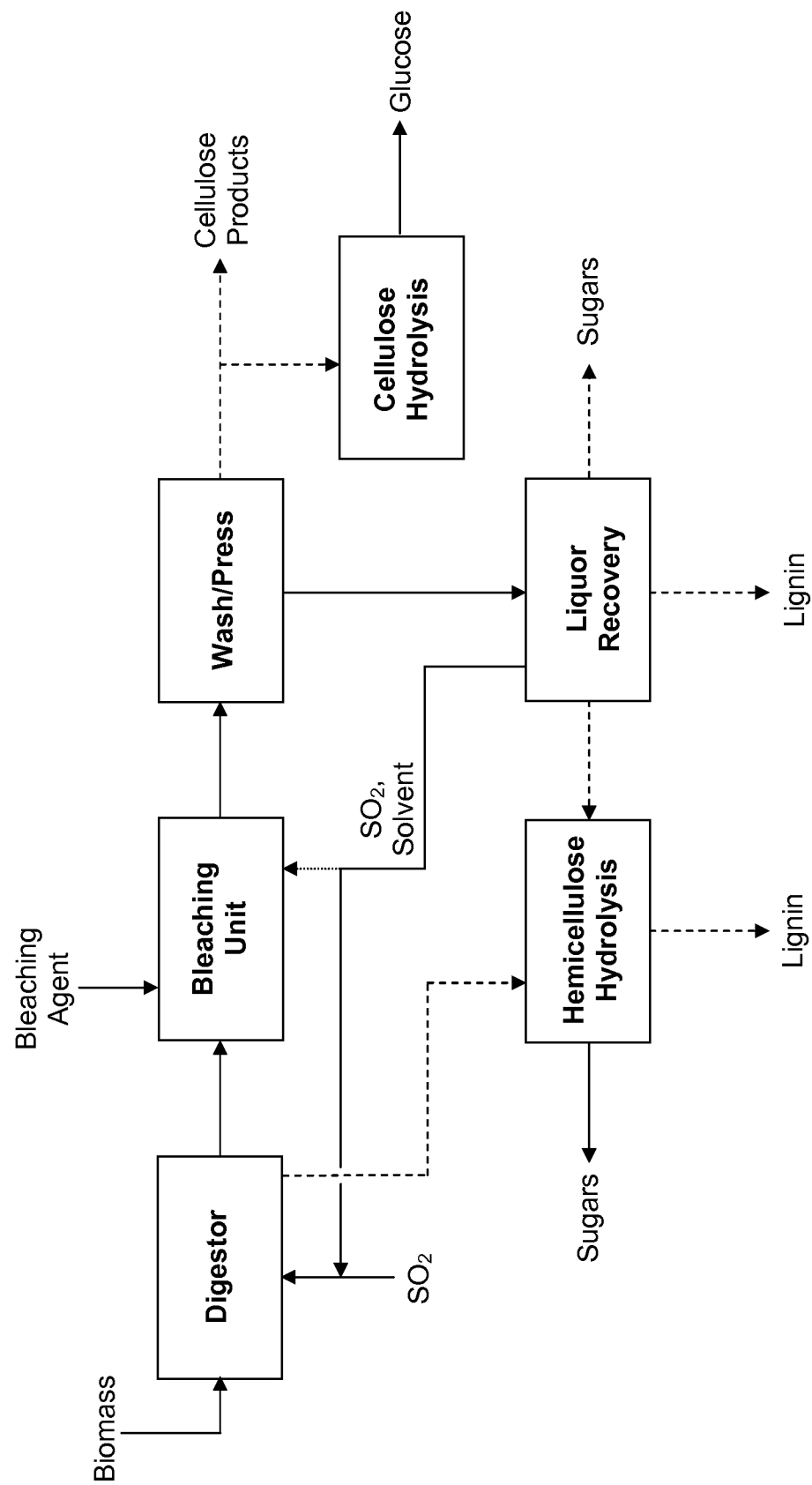
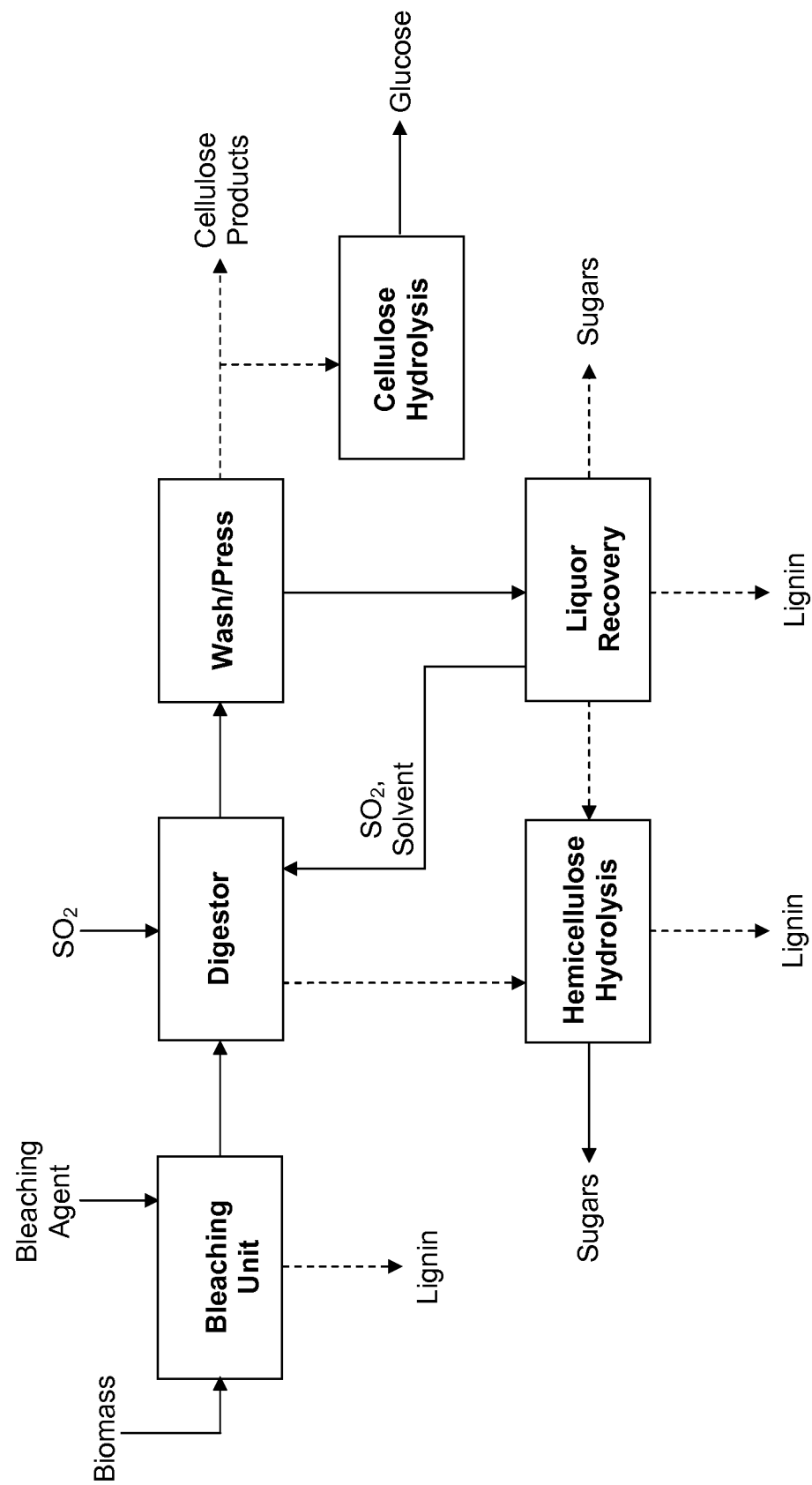


FIG. 2



INTERNATIONAL SEARCH REPORT

International application No.

PCT/US 2013/077897

A. CLASSIFICATION OF SUBJECT MATTER																						
(see extra sheet)																						
According to International Patent Classification (IPC) or to both national classification and IPC																						
B. FIELDS SEARCHED																						
Minimum documentation searched (classification system followed by classification symbols)																						
C08B 37/14, D21C 3/00, 3/04, 3/06, 3/20, 3/22, 5/00, C08B 1/00, C08H 8/00, D21C 9/10, 9/14, 9/153, B82Y 30/00																						
Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched																						
Electronic data base consulted during the international search (name of data base and, where practicable, search terms used)																						
PAJ, Espacenet, EAPATIS, USPTO, PatSearch (RUPTO internal), Information Retrieval System of FIPS (http://www1.fips.ru)																						
C. DOCUMENTS CONSIDERED TO BE RELEVANT																						
Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.																				
Y	RU 2456394 C1 (FEDERALNOE GOSUDARSTVENNOE BYUDZHETNOE UCHREZHDENIE NAUKI INSTITUT PROBLEM KHIMIKO-ENERGETICHESKIKH TEKHNOLGY SIBIRSKOGO OTDELENIYA ROSSYSKOI AKADEMII NAUK) 20.07.2012, abstract, claims	1-20																				
D,Y	US 2060068 A (CELANESE CORPORATION OF AMERICA) 10.11.1936, p. 1, left col., paragraphs 3-5, right col., lines 4-7	1-20																				
Y	US 5203964 A (CALL HANS PETER) 20.04.1993, abstract	3, 13																				
Y	WO 2009/0546024 A1 (SUNOPTA BIOPROCESS INC. et al.) 16.04.2009, claims	6, 8, 9, 16, 18, 19																				
Y	NZ 588539 A (INVENTIA AB) 27.04.2012, abstract	7, 17																				
☐ Further documents are listed in the continuation of Box C. ☐ See patent family annex.																						
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