



US 20120123079A1

(19) **United States**

(12) **Patent Application Publication**  
**Ungerank et al.**

(10) **Pub. No.: US 2012/0123079 A1**

(43) **Pub. Date: May 17, 2012**

(54) **POLYIMIDE MEMBRANES MADE OF  
POLYMERIZATION SOLUTIONS**

(30) **Foreign Application Priority Data**

Jul. 23, 2009 (AT) ..... A1164/2009

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**Publication Classification**

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(51) **Int. Cl.**  
**B01D 71/64** (2006.01)  
**B29D 22/00** (2006.01)  
**C08G 18/76** (2006.01)  
**B05D 5/00** (2006.01)  
**B01D 53/22** (2006.01)  
**B01D 67/00** (2006.01)

(21) Appl. No.: **13/384,685**

(52) **U.S. Cl.** ..... **528/67**; 96/12; 96/10; 528/80;  
427/244; 264/103

(22) PCT Filed: **Jul. 22, 2010**

(86) PCT No.: **PCT/EP2010/060648**

(57) **ABSTRACT**

§ 371 (c)(1),  
(2), (4) Date: **Jan. 18, 2012**

The invention relates to polyimide membranes and to a phase inversion method for the production thereof. The polyimide membranes can be used to separate different gas mixtures.

Figure 1

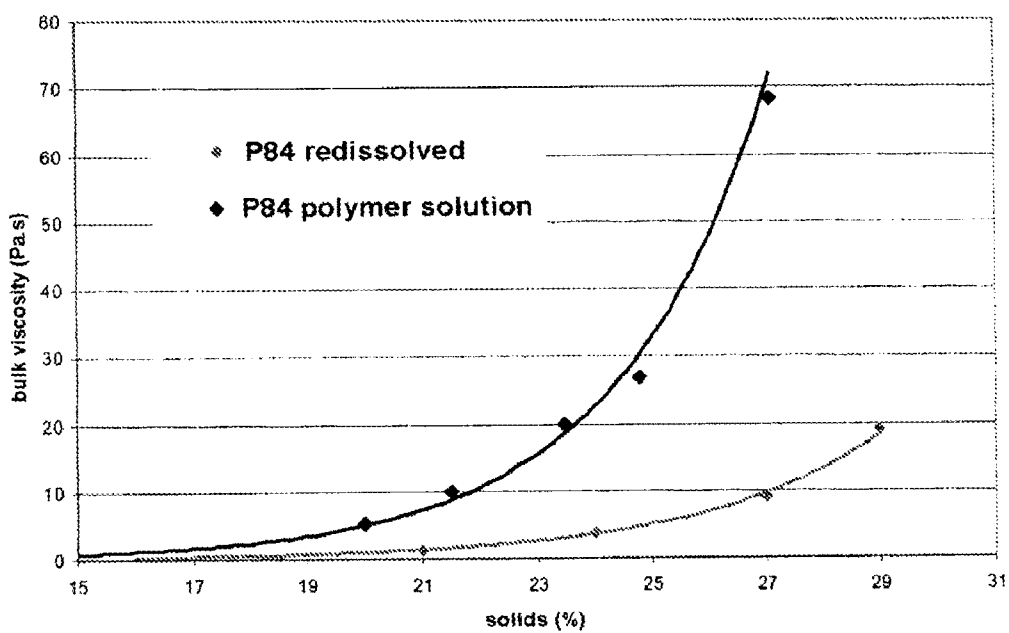


Figure 2

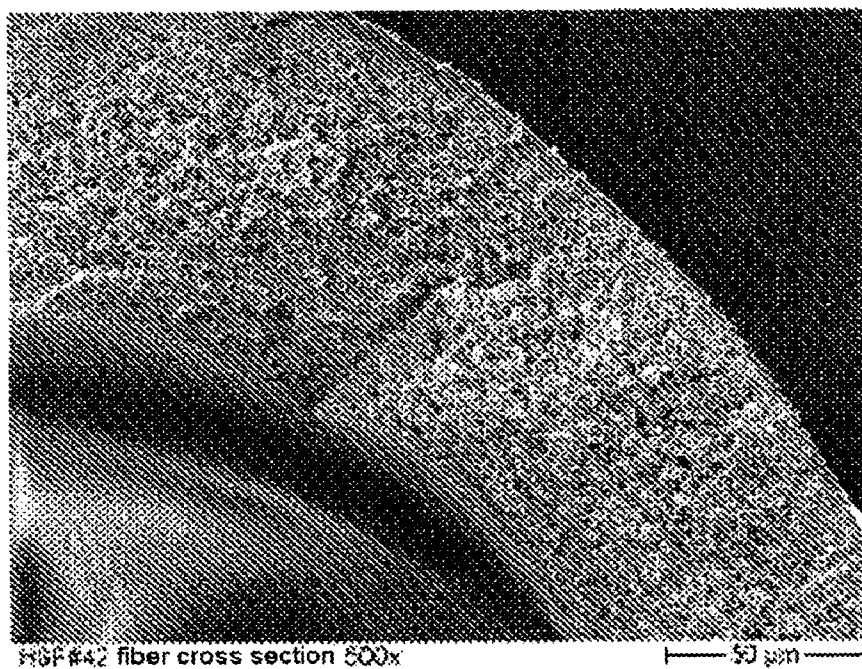
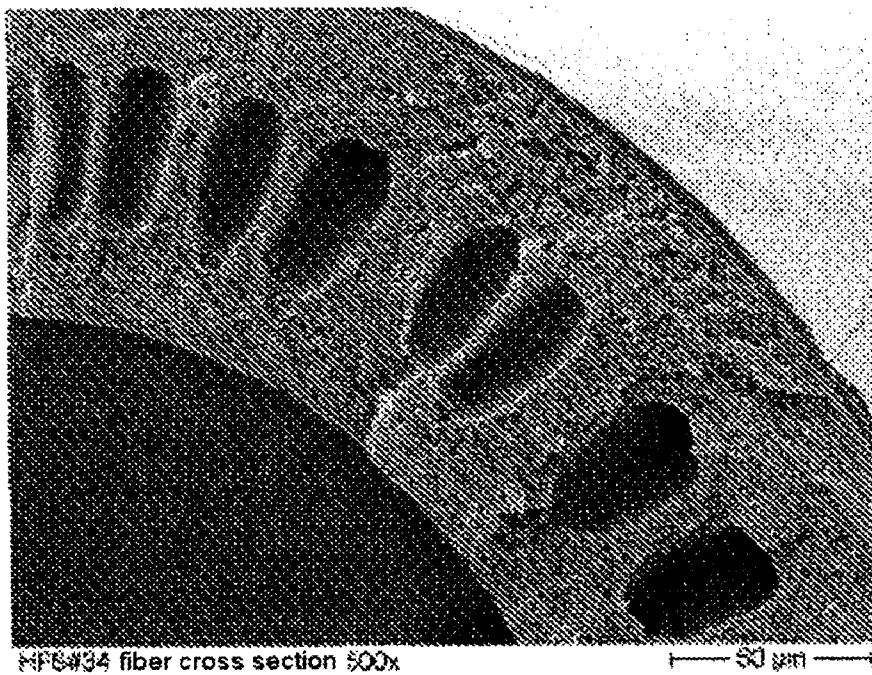
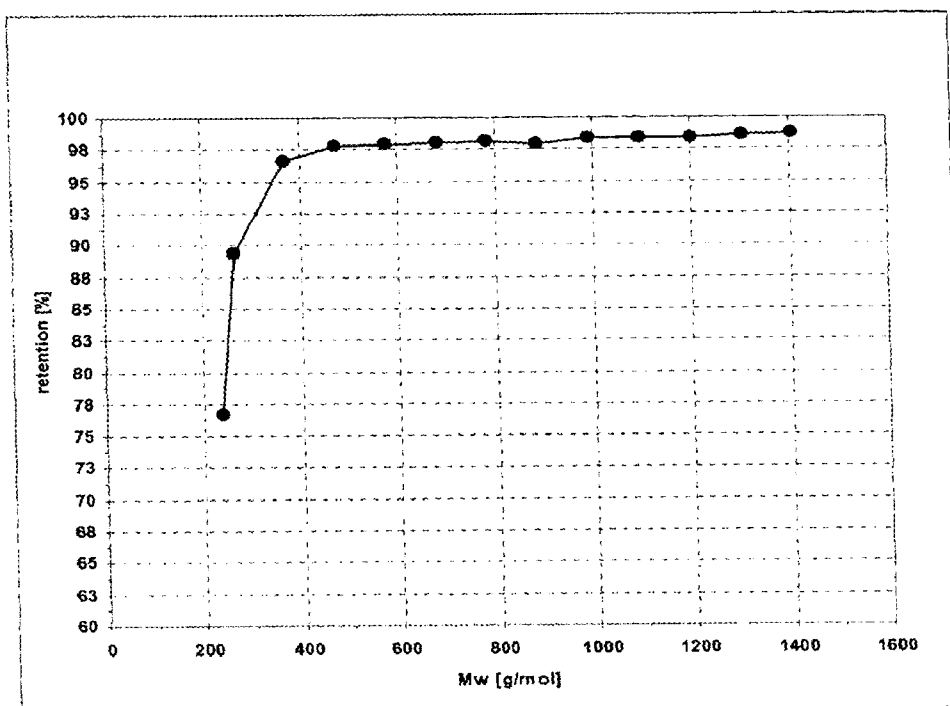


Figure 3:



## POLYIMIDE MEMBRANES MADE OF POLYMERIZATION SOLUTIONS

### FIELD OF INVENTION

[0001] The invention concerns polyimide membranes produced directly from a polyimide polymerization solution without the polyimide having been isolated in the form of a solid material, particularly not as dried solid material and more particularly not as dried powder, and then redissolved. The polyimide membranes concerning the invention can be either flat sheet membranes or hollow fiber membranes. The polyimide membranes can be not only porous membranes in the form of micro-, ultra- or nanofiltration membranes but also aporous membranes for separation of gases. All the membranes are integrally asymmetrical membranes and are produced by a phase inversion process.

### OBJECT

[0002] This invention has for its object to provide a production process for polyimides that does not use any substances which would be a disruptive influence in the subsequent membrane production process. It further has for its object that the process provided by the invention shall make it possible to produce membranes having sufficient mechanical properties.

[0003] Further objects, not explicitly mentioned, will become apparent from the overall context of the ensuing description, examples and claims.

### PRIOR ART

[0004] The production of phase inversion membranes generally requires polymers that are soluble in conventional water-miscible solvents. This process is currently being used to produce thousands of metric tons of polyether sulfone membranes. Possible solvents include inter alia but not exclusively dimethylformamide (DMF), dimethylacetamide or N-methylpyrrolidone. Many additives such as cosolvents, nonsolvents, pore-formers, hydrophilicizers etc are admixed in order to influence the properties of the membranes. The starting point for this is usually a pellet material, the casting solution being produced by pasting up with the solvents and the additives. Success in membrane production, as elsewhere, depends decisively on the molar mass and the distribution of the polymer used. In general, polymers with high molar masses and narrow distribution are required.

[0005] P84 is a polymer which is well known in the literature and is used for the production of flat sheet membranes and hollow fiber membranes (US 2006/0156920, WO 04050223, U.S. Pat. No. 7,018,445, U.S. Pat. No. 5,635,067, EP 1457253, U.S. Pat. No. 7,169,885, US 20040177753, U.S. Pat. No. 7,025,804). P84 is marketed in 2 modifications (P84 type 70 and P84 HT) in powder form by HP Polymer of Lenzing in Austria. The customers then redissolve this powder in aprotic dipolar solvents and admix it with additives. Membranes can then be produced therefrom. However, various customers (e.g., Air Liquide Medal, US 2006/156920) report that the films and membranes produced therefrom are very brittle and only blends with other polymers lead to stable films and hollow fiber membranes. The powder has to be subjected to a special treatment to have sufficiently high molar masses (Air Liquide WO 2006/092677). For this, the treatment time and the method are very important. The result is powders having slightly different properties but which

produce casting solutions having different viscosities. Therefore, uniform production of polymer membranes is only possible with great difficulty.

[0006] P84 is also made into a blend with other polymers (US 2006/156920), so that the membranes produced therefrom have sufficiently high stabilities. But the disadvantage here is that very good separation properties for gases, plasticization stabilities to CO<sub>2</sub> and chemical stabilities to many solvents are in part disruptively influenced, or even destroyed, by admixing other polymers.

[0007] The cause for the low molar mass resides in the production process of the P84 powder. It is at this stage that the polymer loses molar mass. The molar masses directly after polymerization and after production of the powder are depicted in Table 1.

TABLE 1

Molar masses after polymerization and after powder production of P84 type 70 and P84 HAT			
Type		Directly after polymerization in solution (g · mol <sup>-1</sup> )	After conversion into powder (g · mol <sup>-1</sup> )
P84 type 70	Mn	76 600 g	56 200
	Mw	146 200	101 500
	Mp	130 500	93 500
	PDI	1.91	1.81
P84 HT	Mn	75 500	52 200
	Mw	150 900	84 200
	Mp	122 200	84 100
	PDI	2.00	1.61

[0008] It is clearly apparent that the polymer loses molar mass in the course of conversion from the polymerization solution into the powder by a precipitation process.

[0009] P84 powder is also used for the production of flat sheet membranes (WO 2007/125367, WO 2000/06293). There are the same problems here as in the production of hollow fiber membranes.

### Measuring Technique

#### Viscosity Determination

[0010] Dynamic viscosity  $\eta$  is ascertained by shearing the polymer solution in a cylindrical gap at a constant 25° C. once by mandating various rotation rates  $\Omega$  (or shear gradients  $\gamma$ ) and then by mandating various shear stresses  $\tau$ .

[0011] The measuring instrument used is a HAAKE RS 600 with a liquid-heatable measuring cup receiver TEF/Z28, a cylindrical rotor Z25DIN53019/ISO3219 and a disposable aluminum measuring cup Z25E/D=28 mm.

[0012] The shear stress  $\tau$  is measured at a particular shear gradient. Dynamic viscosity  $\eta$  computes from ensuing formulae and is reported at a shear gradient of 10 s<sup>-1</sup> in Pa·s.

Viscosity function proper

$$\frac{\tau}{\gamma} = \eta * \gamma^2$$

Shear gradient  $\gamma = M * \Omega$

$\sigma$  . . . Shear stress

$\eta$  . . . Dynam. viscosity

M . . . Shear factor of rotor: 12350 rad/s

$\Omega$  . . . Angular viscosity

#### Molar Mass Determination

**[0013]** Molar mass is determined using a gel permeation chromatography system. The system is calibrated with polystyrene standards. The molar masses reported are therefore to be understood as relative molar masses.

**[0014]** Components and settings used were as follows:

HPLC	WATERS 600 pump, 717 autoinjector, 2487 UV detector
Precolumn	PSS SDV precolumn
Columns	PSS SDV 10 $\mu\text{m}$ 1000, 10 <sup>5</sup> and 10 <sup>6</sup> Å
Eluent	0.01M LiBr + 0.03M H <sub>3</sub> PO <sub>4</sub> in DMF (sterile-filtered, 0.45 $\mu\text{m}$ )
Flux	1.0 ml/min
Run time	45 min
Pressure	~1.550 psi
Wavelength	270 nm (with use of UV detector)
Injection volume	50 $\mu\text{l}$ or 20 $\mu\text{l}$ (for solutions c >1 g/l)
Standards	PS (polystyrene) standards (narrow distribution, 300-3 $\times$ 10 <sup>6</sup> , PSS)

#### Permeabilities

**[0015]** Gas permeabilities for films are reported in barrers (10<sup>-10</sup> cm<sup>3</sup>·cm<sup>-2</sup>·cm·s<sup>-1</sup>·cmHg<sup>-1</sup>). Permeances of hollow fibers or flat sheet membranes to gases are reported in GPU (Gas Permeation Unit, 10<sup>-6</sup> cm<sup>3</sup>·cm<sup>-2</sup>·s<sup>-1</sup>·cmHg<sup>-1</sup>). The fluxes of nano- and ultrafiltration membranes are reported in l·m<sup>-2</sup>·h<sup>-1</sup>·bar<sup>-1</sup>.

#### Gas Permeabilities

**[0016]** Permeabilities to gases are measured by the pressure rise method. A flat sheet film between 10 and 70 $\mu\text{m}$  in thickness has a gas or gas mixture applied to it from one side. On the other side, the permeate side, there is a vacuum (ca. 10<sup>-2</sup> mbar) at the start of the test. Then, pressure rise on the permeate side over time is reported.

**[0017]** The polymer's permeability can be computed by the following formula:

$$P = \frac{V_{dead} \cdot MW_{gas} \cdot l}{\rho \cdot R \cdot T \cdot A \cdot \Delta p} \cdot \frac{dp}{dt} \cdot 10^{10}$$

P . . . Permeability in barrers (10<sup>-10</sup> cm<sup>3</sup>·cm<sup>-2</sup>·cm·s<sup>-1</sup>·cmHg<sup>-1</sup>)

$V_{dead}$  . . . Volume of permeate side in cm<sup>3</sup>

$MW_{gas}$  . . . Molar mass of gas in g·mol<sup>-1</sup>

l . . . Layer thickness of film in cm

$\rho$  . . . Density of gas in g·cm<sup>-3</sup>

R . . . Gas constant in cm<sup>3</sup>·cmHg·K<sup>-1</sup>·mol<sup>-1</sup>

T . . . Temperature in kelvins

A . . . Area of film in cm<sup>2</sup>

$\Delta$  . . . A Pressure difference between feed and permeate side in cmHg dp/dt. Pressure rise per time on permeate side in cmHg·s<sup>-1</sup>

**[0018]** The permeability of hollow fibers is measured using the same pressure rise method.

**[0019]** Permeance is computed by the following formula:

$$p = \frac{V_{dead} \cdot MW_{gas}}{\rho \cdot R \cdot T \cdot A \cdot \Delta p} \cdot \frac{dp}{dt} \cdot 10^6$$

P . . . Permeance in GPU (gas permeation units.

10<sup>-6</sup> cm<sup>3</sup>·cm<sup>-2</sup>·cm·s<sup>-1</sup>·cmHg<sup>-1</sup>)

$V_{dead}$  . . . Volume of permeate side in cm<sup>3</sup>

$MW_{gas}$  . . . Molar mass of gas in g·mol<sup>-1</sup>

$\rho$  . . . Density of gas in g·cm<sup>-3</sup>

R . . . Gas constant in cm<sup>3</sup>·cmHg·K<sup>-1</sup>·mol<sup>-1</sup>

T . . . Temperature in kelvins

A . . . External area of hollow fiber in cm<sup>2</sup>

$\Delta p$  . . . Pressure difference between feed and permeate side in cmHg

dp/dt. Pressure rise per time on permeate side in cmHg·s<sup>-1</sup>

**[0020]** The selectivities of various pairs of gases are pure-gas selectivities. The selectivity between two gases computes from the ratio of permeabilities:

$$S = \frac{P_1}{P_2}$$

S . . . ideal gas selectivity

$P_1$  . . . permeability or permeance of gas 1

$P_2$  . . . permeability or permeance of gas 2

#### Liquid Permeances

**[0021]** Permeances of flat sheet membranes are determined using a Milipore stirred cell pressurized with 5 to 6 bar of nitrogen. What is measured is permeate flux per unit time at a defined pressure. Permeance is given by:

$$p = \frac{v}{\Delta p \cdot A}$$

P . . . Permeance in l·m<sup>-2</sup>·h<sup>-1</sup>·bar<sup>-1</sup>

v . . . Volume stream in l·h<sup>-1</sup>

$\Delta p$  . . . Pressure difference between feed and permeate side in bar

A . . . Filtration area in m<sup>2</sup>

**[0022]** Retention R is obtained from the following formula:

$$R = \left(1 - \frac{c_p}{c_f}\right) \cdot 100$$

R . . . Retention in %

**[0023]**  $C_p$  . . . Concentration of dissolved product in permeate

$C_f$  . . . Concentration of dissolved product in feed

**[0024]** When retention is 100%, the entire product is retained by the membrane. When retention is 0%, the membrane lets the entire dissolved product through.

#### SOLUTION TO PROBLEM

**[0025]** The problem of molar mass degradation in the production of P84 powder is circumvented by the polymer after

the polymerization in an aprotic dipolar solvent not being isolated in the form of a solid material, particularly not as dried solid material and more particularly not as dried powder, but instead the polymerization solution being used directly for producing the membranes.

**[0026]** The membrane production process involves the following subsidiary steps:

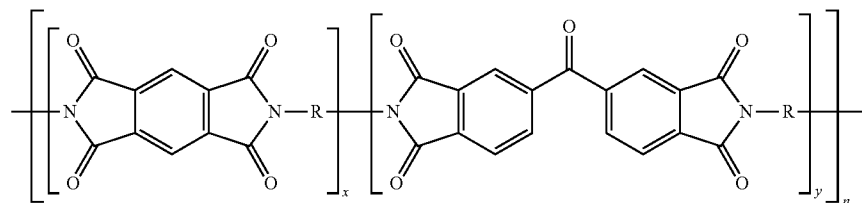
**[0027]** a) polymerization

**[0028]** b) producing the casting solution

**[0029]** c) membrane production

#### Polymerization

**[0030]** The polyimides are produced via a polycondensation of an aromatic tetracarboxylic anhydride with an aro-



matic diisocyanate by release of carbon dioxide. Preferably used substances and combinations thereof are described here-inbelow:

#### Dianhydrides:

**[0031]** 3,4,3',4'-benzophenonetetracarboxylic dianhydride, 1,2,4,5-benzenetetracarboxylic dianhydride, 3,4,3',4'-biphenyltetracarboxylic dianhydride, oxydiphthalic dianhydride, sulfonyldiphthalic dianhydride, 1,1,1,3,3,3-hexafluoro-2,2-propylidenediphthalic dianhydride

#### Diisocyanates:

**[0032]** 2,4-tolylene diisocyanate, 2,6-tolylene diisocyanate, 4,4'-methylenediphenyl diisocyanate, 2,4,6-trimethyl-1,3-phenylene diisocyanate, 2,3,4,5-tetramethyl-1,4-phenylene diisocyanate

**[0033]** The polymerization takes place in an aprotic dipolar solvent. Dimethylformamide, dimethylacetamide, N-methylpyrrolidone, N-ethylpyrrolidone and sulfolane are used preferably but not exclusively singly or in mixtures.

**[0034]** This involves the aromatic dianhydride or mixtures of aromatic dianhydrides being dissolved in concentrations of 10% by weight to 40% by weight, preferably between 18% by weight and 32% by weight and more preferably between 22% by weight and 28% by weight in an aprotic dipolar solvent and heated to from 50° C. to 150° C., preferably 70° C. to 120° C. and more preferably to from 80° C. to 100° C. This solution is admixed with 0.01% by weight to 5% by weight, preferably 0.05% by weight to 1% by weight and more preferably 0.1% by weight to 0.3% by weight of a basic catalyst. Useful catalysts include:

**[0035]** Alkali or alkaline earth metal hydroxides, methoxides, ethoxides, carbonates and phosphates such as for example but not exclusively sodium hydroxide, potassium hydroxide, sodium methoxide, potassium methox-

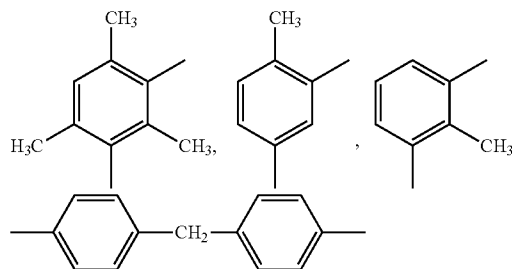
ide, sodium ethoxide, potassium ethoxide, sodium carbonate, sodium bicarbonate, potassium carbonate, potassium bicarbonate, potassium phosphate, potassium hydrogenphosphate, potassium dihydrogenphosphate

**[0036]** Tertiary amines such as for example but not exclusively: trimethylamine, triethylamine, tripropylamine, diazabicycloundecane, diaza-bicyclooctane, dimethylaminopyridine.

**[0037]** The diisocyanate is then added over a period of 1 to 25 hours, preferably 3 to 15 hours and more preferably 5 to 10 hours.

**[0038]** Particular preference is given to producing the following polyimides:

**[0039]** R is selected from the group consisting of



x, y: mole fraction with  $0 < x < 0.5$  and  $1 > y > 0.5$

**[0040]** The result is a clear golden yellow to dark brown polymer solution having a viscosity between 1 and 300 Pa·s, preferably 20 to 150 Pa·s and more preferably 40 to 90 Pa·s. The molar masses  $M_p$  are greater than  $100\,000\text{ g}\cdot\text{mol}^{-1}$  and therefore differ distinctly from the polyimide polymer powders, especially the P84 polymer powders.

**[0041]** The polyimide polymer of the present invention is obtained after the reaction as a solute in an aprotic dipolar solvent. There are no disruptive concomitants or by-products in the polymer solution. The viscosity is very high and suitable for production of membranes. For that reason, it is also economically advantageous for the polymer not to be precipitated and then redissolved in the same solvent. The solutions are therefore used directly—without isolating the polymer and preferably also without any other further treatment—for producing the casting solution.

#### Producing the Casting Solution

**[0042]** The polymer solutions obtained from the polymer condensation have a solids content between 22% by weight

and 28% by weight and can be used for producing the casting solution without further treatment.

**[0043]** The casting solution of the present invention is notable for the following properties:

**[0044]** It has a sufficiently high viscosity for production of flat sheet and hollow fiber membranes

**[0045]** It may contain additives to prevent the formation of large voids (macrovoids) in the membranes

**[0046]** It may contain volatile solvents to produce a surface having the desired pore size

**[0047]** Casting solution viscosity is ideal when it corresponds to the entanglement point in viscosity plotted as a function of solids content. This point is that point where the function of viscosity versus solids turns from linear to exponential. This point is also very highly dependent on molar mass. The higher the molar mass, the lower the solids content at which entanglement occurs.

**[0048]** In respect of viscosity, molar mass and molar mass distribution, the casting solutions obtainable via the process according to the present invention differ distinctly from the casting solutions of the prior art. It is only the process of the present invention that provides casting solutions combining a high viscosity with a high molar mass and a narrow molar mass distribution for the polyimide. The processes of the present invention thus make it possible to obtain membranes that have outstanding mechanical properties.

**[0049]** The processes of the prior art, i.e., dissolving pulverulent polyimides with subsequent aftertreatment to raise the molar mass, cannot be used to obtain any casting solutions having comparable combinations of properties.

**[0050]** The process of the present invention also makes it possible to add additives. Various amounts of additives result in different solids contents, which would then shift the entanglement point. Modulating the molar mass in the polymerization can be used to shift this entanglement point again.

**[0051]** As the casting solution composition moves very far away from the concentration at which phase separation takes place, the gradient between solvent and nonsolvent becomes very large in membrane production by phase inversion, and large voids are obtained in the membranes. These voids, which are also known as macrovoids, are responsible for lower stability of the membranes to pressure in use, and limit their usefulness for example in use in natural gas cleanup. The formation of macrovoids can be prevented by addition of nonsolvents. Suitable for this are the following water-miscible solvents or mixtures thereof.

**[0052]** This list is only illustrative, the practiced artisan will readily think of even more solvents.

**[0053]** alcohols such as for example methanol, ethanol, isopropanol, propanol, butanol, butanediol, glycerol,

**[0054]** water,

**[0055]** ketones such as for example acetone or butanone

**[0056]** To produce a defined surface on the membrane, several methods can be used in principle: As well as the delayed demixing method, evaporative removal of volatile cosolvents will also lead to very thin selective layers not only in the gas separation membrane sector but also in the nano- and ultrafiltration membrane sector. The degree of evaporative removal and hence the pore size is influenced by the species of volatile solvent, its concentration, the evaporation time, the casting solution temperature, the amount and temperature of ambient gas in the evaporative removal zone.

**[0057]** Useful volatile solvents include the following. They should be water miscible, for example acetone, tetrahydrofuran, methanol, ethanol, propanol, isopropanol, dioxane, diethyl ether.

**[0058]** Producing the casting solution is preferably effected by adding additives by metered addition of the mixture of additives or separately from each other in succession. The additives are gradually metered into the mixture under agitation. The metered addition takes between 10 min and 3 hours for preference and between 30 min and 2 hours for particular preference. Adding the cosolvents causes partial precipitation of polyimide at the drop entry point. But the solids dissolve again after a few minutes without leaving a residue. The clear solution is then additionally filtered through a 15 $\mu$  steel mesh sieve in order to remove destructive concomitants which would lead to imperfections in the membrane surface.

**[0059]** After filtration, the solution is left to stand in a sealed container at 50° C. for 2 days for removal of air bubbles and hence devolatilization.

#### Producing the Hollow Fibers

**[0060]** The devolatilized, filtered and additivized polyimide polymer solution is thermostated—preferably to from 20 to 100° C. and more preferably to from 30 to 70° C. The solution is gear pumped through the outer part of a two-material die. The external diameter of the two-material die is 600  $\mu$ m, the internal diameter is 160  $\mu$ m, pump rate is between 1.3 and 13.5 ml/min. A liquid mixture of water and one or more than one aprotic dipolar solvent in admixture is pumped in the inner part of the two-material die.

**[0061]** Useful solvents include inter alia but not exclusively dimethylformamide, dimethylacetamide, N-methyl-pyrrolidone, N-ethylpyrrolidone, sulfolane or dimethyl sulfoxide.

**[0062]** The composition is between solvent and water is between 10% by weight and 95% by weight of solvent and 90% by weight and 5% by weight of water, preferably between 30% by weight and 90% by weight of solvent and 70% by weight and 10% by weight of water and more preferably between 50% by weight and 80% by weight of solvent and 50% by weight and 20% by weight of water. Pump rate is between 0.2 ml/min and 10 ml/min.

**[0063]** The resulting hollow fiber then enters a tube flooded with a dry thermostated gas. Useful gases include: nitrogen, air, argon, helium, carbon dioxide, methane or other industrial inert gases. Gas temperature is adjusted via a heat exchanger and is preferably between 20 and 250° C., more preferably between 30 and 150° C. and even more preferably between 40 and 120° C.

**[0064]** Gas velocity in the tube is preferably between 0.1 and 10 m/min, more preferably between 0.5 and 5 m/min and even more preferably between 1 and 3 m/min. The distance and hence tube length is preferably between 5 cm and one meter and more preferably between 10 and 50 cm. The thread thus conditioned then dips into a water bath to coagulate the polymer mass and thus form the membrane. Water bath temperature is preferably between 1 and 60° C., more preferably between 5 and 30° C. and more preferably between 8 and 16° C.

**[0065]** The concentration of aprotic dipolar and other solvents such as for example but not exclusively dimethylformamide, dimethylacetamide, N-methyl-pyrrolidone, N-ethylpyrrolidone, sulfolane, dimethyl sulfoxide, tetrahydrofuran, dioxane, isopropanol, ethanol or glycerol in the coagulation bath is between 0.01% by weight and 20% by

weight, preferably between 0.1% by weight and 10% by weight and more preferably between 0.2% by weight and 1% by weight.

**[0066]** The hollow fibers are hauled off at between 2 and 100 m/min, preferably between 10 and 50 m/min and more preferably between 20 and 40 m/min. The fibers are wound up onto a bobbin and washed in water until the residual solvent content is below 1%. This is followed by treatment in ethanol and hexane. The fibers are then dried—preferably between room temperature and 150° C. and more preferably between 50 and 100° C. Fibers are obtained with external diameters of 100 to 1000 μm, preferably between 200 and 700 μm and more preferably between 250 and 400 μm.

**[0067]** The process of the present invention thus provides hollow fiber membranes of polyimides that exhibit high separation performances for various gases. An excerpt for various polymers and gases is summarized in Table 2.

TABLE 2

Permeances of various inventive polyimide hollow fibers in single gas measurements						
Polyimide polymer of hollow fiber	Internal diameter [μm]	External diameter [μm]	Gas	Trans-membrane pressure [bar]	Permeance [GPU]	Selectivity
P84 type 70	250	412	O <sub>2</sub>	5	1.45	8.8
	250	412	N <sub>2</sub>	5	0.165	
	250	412	CO <sub>2</sub>	5	6.0	72
	250	412	CH <sub>4</sub>	5	0.084	
	250	412	CO <sub>2</sub>	40	9.0	88
P84 T100	250	412	CH <sub>4</sub>	40	0.101	
	339	189	O <sub>2</sub>	10	0.56	7.8
	339	189	N <sub>2</sub>	10	0.072	
	339	189	CO <sub>2</sub>	10	1.68	72
P84 HT	339	189	CH <sub>4</sub>	10	0.023	
	174	300	O <sub>2</sub>	9	0.65	8.4
	174	300	N <sub>2</sub>	10	0.078	
	245	397	CO <sub>2</sub>	10	2.60	58
	245	397	CH <sub>4</sub>	9	0.044	

**[0068]** It is further notable that the membranes even under high CO<sub>2</sub> partial pressures scarcely exhibit any increase in methane permeance, retain their selectivity and therefore are scarcely plasticized. This property is necessary to process sour gases with high CO<sub>2</sub> contents and high pressures, as is the case for example with the workup of crude natural gas or crude biogas.

**[0069]** The hollow fiber membranes can also be crosslinked with amines. When the hollow fiber is crosslinked, this is done subsequent to the washing step. For this purpose, the hollow fiber is passed through a bath containing an amine with 2 or more amino groups per molecule such as, for example, a diamine, triamine, tetraamine or a polyamine. The amine may be primary or secondary or consist of mixtures of primary, secondary and tertiary amines in one molecule. Useful amines include aliphatic amines, aromatic amines and mixed aliphatic-aromatic amines. Silicone-based amines are also possible. Examples of aliphatic diamines include inter alia but not exclusively: diaminoethane, diaminopropane, diaminobutane, diaminopentane, diaminohexane, diaminoheptane, diaminoctane, diaminodecane or diamino compounds of branched or cyclic aliphatics (e.g. cis- and trans-1,4-cyclohexane) and of longer-chain compounds.

**[0070]** Useful aromatic compounds include inter alia but not exclusively: p-phenylenediamines, m-phenylenediamines, 2,4-tolylenediamines, 2,6-tolylenediamines, 4,4'-diaminodiphenyl ether.

**[0071]** Examples of mixed aliphatic-aromatic amines include inter alia but not exclusively: aminoalkyl-substituted aromatics such as, for example, p-bis(aminomethyl)-benzene.

**[0072]** Useful siliconic-based amines include inter alia but not exclusively: bis(aminoalkyl)siloxanes of differing chain length.

**[0073]** Useful representatives of polyfunctional amines include inter alia but not exclusively the following compounds: oligo- or polyethyleneimines having various molar masses (400 to 200 000 g/mol), N,N',N''-trimethylbis-(hexamethylene)triamine, bis(6-aminoethyl)amine

**[0074]** Crosslinking is effected by emplacement into or continuous pulling of the entire hollow fiber through a solution of the particular diamine in water or a mixture of water and water-miscible solvents or other solvents which do not influence the membrane structure and which dissolves the particular amines. Possibilities for this are for example but not exclusively:

**[0075]** Alcohols such as, for example, methanol, ethanol, isopropanol, propanol, butanol, butanediol, glycerol

**[0076]** Ethers such as, for example, diethyl ether, tetrahydrofuran, dioxane or polyethylene glycols or polyethylene glycol ethers

**[0077]** Aprotic dipolar solvents such as, for example, dimethylformamide, dimethylacetamide, N-methylpyrrolidone, N-ethylpyrrolidone, tetramethylurea, dimethyl sulfoxide or sulfolane

**[0078]** Ketones such as, for example, acetone or methyl ethyl ketone

**[0079]** Others such as ethyl acetate, dichloromethane, chloroform, toluene, xylene, aliphatics and cycloaliphatics such as hexane, heptane or cyclohexane.

**[0080]** The concentration of diamines is between 0.01% by weight and 10% by weight, but preferably between 0.05% by weight and 5% by weight and more preferably between 0.1% by weight and 1% by weight.

**[0081]** The crosslinking solution temperature is between 1 and 100° C., preferably between 10 and 70° C. and more preferably between 20 and 50° C.

**[0082]** Residence time is between 10 seconds and 10 hours, preferably between 1 minutes and 60 minutes and more preferably between 2 and 10 min.

**[0083]** To remove the residual amine, the membrane is washed with water. Wash bath temperature is between 10 and 90° C. and preferably between 20 and 60° C. Wash bath residence time is from 1 to 200 minutes, preferably between 2 and 50 minutes and more preferably between 3 and 10 minutes.

**[0084]** Hollow fibers are obtained that are no longer soluble in traditional organic solvents such as for example but not exclusively dimethylformamide, dimethylacetamide, N-methylpyrrolidone, N-ethylpyrrolidone, tetramethyl-urea, dimethyl sulfoxide or sulfolane, acetone, methyl ethyl ketone, diethyl ether, tetrahydrofuran, dioxane, ethyl acetate, dichloromethane, chloroform, toluene, xylene, hexane, heptane or cyclohexane. They can therefore be used in nano-, ultra- and microfiltration in organic solvents.

Producing the Flat Sheet Membranes

**[0085]** The additized and devolatilized solution is poured without bubbles into the applicator of a flat sheet membrane casting rig. Applicator width can be up to 1.2 m. A calendared

backing fleece of polymer fibers, preferably but not exclusively in polyimide, polypropylene, polyamide, polyester or polyphenylene sulfide, passes underneath the applicator at a speed of 0.1 to 10 m/min and preferably 1 to 5 m/min. Fleece thickness is between 30 and 300 $\mu$ m and preferably between 100 and 200 $\mu$ m. Basis weight is between 20 and 300 g/m<sup>2</sup> and preferably between 50 and 150 g/m<sup>2</sup>. Gap width between applicator and fleece is between 100 and 800 $\mu$ m and preferably between 200 and 400 $\mu$ m. The coated fleece enters a channel flooded with a countercurrent stream of gas. Useful gases include inter alia but not exclusively dry air, nitrogen, argon or helium. The gas flowing over the coated fleece moves at a speed in the range from 100 to 5000 m/h and preferably between 200 and 1000 m/h, gas temperatures can be between 10 and 150° C. and preferably between 15 and 90° C. The coated fleece then enters a coagulation bath, the polymer coagulates and forms the desired membrane. The coagulation bath consists of water or mixtures of water and one or more solvents that are miscible with water.

**[0086]** Possibilities for this include:

**[0087]** Alcohols such as, for example, methanol, ethanol, isopropanol, propanol, butanol, butanediol, glycerol

**[0088]** Ethers such as, for example, diethyl ether, tetrahydrofuran, dioxane or polyethylene glycols or polyethylene glycol ethers

**[0089]** Aprotic dipolar solvents such as, for example, dimethylformamide, dimethylacetamide, N-methylpyrrolidone, N-ethylpyrrolidone, tetramethylurea, dimethyl sulfoxide or sulfolane

**[0090]** Ketones such as, for example, acetone or methyl ethyl ketone.

**[0091]** Coagulation bath temperature is between 1 and 90° C. and preferably between 10 and 50° C. Following a short residence time of 10 s to 10 min and preferably 1 to 5 min the membrane is wound up in the wet state.

**[0092]** To remove residual solvent, the membrane is washed with water. Wash bath temperature is between 10 and 90° C. and preferably between 20 and 60° C. Wash bath residence time is from 1 to 200 minutes, preferably between 2 and minutes and more preferably between 3 and 10 minutes.

**[0093]** When the membrane is crosslinked, this is done subsequent to the washing step. For this purpose, the membrane is passed through a bath containing an amine with 2 or more amino groups per molecule such as, for example, a diamine, triamine, tetraamine or polyamine. The amine may be primary or secondary or consist of mixtures of primary, secondary and tertiary amines in one molecule. Useful amines include aliphatic amines, aromatic amines and mixed aliphatic-aromatic amines. Silicone-based amines are also possible.

**[0094]** Examples of aliphatic diamines include inter alia but not exclusively: diaminoethane, diaminopropane, diaminobutane, diaminopentane, diaminohexane, diaminoheptane, diaminoctane, diaminodecane or diamino compounds of branched or cyclic aliphatics (e.g. cis- and trans-1,4-cyclohexane) and of longer-chain compounds.

**[0095]** Useful aromatic compounds include inter alia but not exclusively: p-phenylenediamines, m-phenylenediamines, 2,4-tolylenediamines, 2,6-tolylenediamines, 4,4'-diaminodiphenyl ether.

**[0096]** Examples of mixed aliphatic-aromatic amines include inter alia but not exclusively: aminoalkyl-substituted aromatics such as, for example, p-bis(aminomethyl)-benzene.

**[0097]** Useful siliconic-based amines include inter alia but not exclusively: bis(aminoalkyl)siloxanes of differing chain length. Useful representatives of polyfunctional amines include inter alia but not exclusively the following compounds: oligo- or polyethyleneimines having various molar masses (400 to 200 000 g/mol), N,N',N''-trimethylbis(hexamethylene)triamine, bis(6-aminoethyl)amine

**[0098]** Crosslinking is effected by emplacing the entire membrane into a solution of the particular diamine in water or a mixture of water and water-miscible solvents or other solvents which do not influence the membrane structure and which dissolves the particular amines.

**[0099]** Possibilities for this are for example but not exclusively:

**[0100]** Alcohols such as, for example, methanol, ethanol, isopropanol, propanol, butanol, butanediol, glycerol

**[0101]** Ethers such as, for example, diethyl ether, tetrahydrofuran, dioxane or polyethylene glycols or polyethylene glycol ethers

**[0102]** Aprotic dipolar solvents such as, for example, dimethylformamide, dimethylacetamide, N-methylpyrrolidone, N-ethylpyrrolidone, tetramethylurea, dimethyl sulfoxide or sulfolane

**[0103]** Ketones such as, for example, acetone or methyl ethyl ketone

**[0104]** Others such as ethyl acetate, dichloromethane, chloroform, toluene, xylene, aliphatics and cycloaliphatics such as hexane, heptane or cyclohexane.

**[0105]** The concentration of diamines, the crosslinking solution temperature, the residence time and the way the washing step is carried out correspond to the values and procedures, respectively, indicated above for crosslinking the hollow fibers.

**[0106]** Following the washing/crosslinking operations, the membrane is impregnated to ensure pore preservation during subsequent drying. This is done by dipping the membrane into a mixture of water and a water-miscible high boiler.

**[0107]** Possibilities for this are for example but not exclusively: glycerol, polyethylene glycols of differing chain length in admixture or singly, polyethylene glycol dialkyl ethers of different chain lengths in admixture or singly as methyl or ethyl ethers, mono- or diols having a boiling point above 200° C. such as, for example, decanol, 1,4-butanediol, 1,6-hexanediol.

**[0108]** The concentration of high boiler in water is between 5% and 95%, but preferably between 25% by weight and 75% by weight. Impregnating solution temperature is between 1 and 100° C., preferably between 10 and 70° C. and more preferably between 20 and 50° C.

**[0109]** The residence time is between 10 seconds and 10 hours, preferably between 1 minutes and 60 minutes and more preferably between 2 and 10 min.

**[0110]** After impregnation, the membrane is dried. Drying can be done in the ambient air or continuously in a convective dryer. Drying temperature is in the range from 20 to 200° C. and preferably between 50 and 120° C. Drying time is between 10 seconds and 10 hours, preferably between 1 minutes and 60 minutes and more preferably between 2 and 10

min. After drying, the final membrane is wound up and can be further processed into spiral-wound elements or pocket modules.

**[0111]** The flat and hollow fiber membranes of the present invention thus comprise a polyimide having an  $M_p > 100\,000\text{ g}\cdot\text{mol}^{-1}$ , preferably  $110\,000$  to  $200\,000\text{ g}\cdot\text{mol}^{-1}$  and more preferably  $120\,000$  to  $170\,000\text{ g}\cdot\text{mol}^{-1}$  and a PDI in the range from 1.7 to 2.3 and preferably in the range from 1.8 to 2.1.  $M_p$  here corresponds to the peak maximum of the molar mass distribution on calibration against polystyrene standards in 0.01 mol/l of lithium bromide in dimethylformamide.

**[0112]** The high molar mass effectuates an improvement in mechanical properties regarding membrane strength and toughness. This is more particularly required at high pressures in the applications. Flat sheet membranes have to withstand at least 40 bar in operation and certain hollow fiber membranes above 100 bar in natural gas enrichment.

**[0113]** A high molar mass is also advantageous to achieve a sufficiently high viscosity even at moderate solids contents. Casting solutions require a certain viscosity for stable processing into membranes and hollow fibers and in order that dense and selective layers on the surface may be produced therewith.

#### PRODUCTION EXAMPLES

**[0114]** The examples which follow serve to provide more particular elucidation and better understanding of the present invention, but do not limit it in any way.

##### Producing the Polyimide Solutions

###### Example 1

###### Producing a P84 Type 70 Polyimide Solution in Dimethylacetamide

**[0115]** A 3 l glass reactor equipped with stirrer and reflux condenser is initially charged with 1622 g of anhydrous dimethylacetamide. A quantity of 456.4 g of 3,3',4,4'-benzophenonetetracarboxylic dianhydride are dissolved therein and the solution is heated to 90° C. To this solution is added 0.45 g of sodium hydroxide. Under nitrogen, 266.8 g of a mixture of 64% of 2,4-tolylene diisocyanate, 16% of 2,6-tolylene diisocyanate and 20% of 4,4'-diisocyanatodiphenylmethane are metered during several hours. In the process, CO<sub>2</sub> escapes as by-product and a polyimide results directly in solution.

**[0116]** The highly viscous solution obtained has a golden color, a solids content of 25% and a viscosity of 49 Pa·s.

**[0117]** The molar masses are determined by gel permeation chromatography as follows:  $M_n=80\,600\text{ g}\cdot\text{mol}^{-1}$ ,  $M_p=139\,600\text{ g}\cdot\text{mol}^{-1}$ ,  $M_w=170\,000\text{ g}\cdot\text{mol}^{-1}$  PDI=2.11

###### Example 2

###### Producing a P84 Type 70 Polyimide Solution in Dimethylformamide

**[0118]** A 3 l glass reactor equipped with stirrer and reflux condenser is initially charged with 1622 g of anhydrous dimethylformamide. A quantity of 456.4 g of 3,3',4,4'-benzophenonetetracarboxylic dianhydride are dissolved therein and the solution is heated to 90° C. To this solution is added 0.45 g of sodium hydroxide. Under nitrogen, 266.8 g of a mixture of 64% of 2,4-tolylene diisocyanate, 16% of 2,6-tolylene diisocyanate and 20% of 4,4'-diisocyanatodiphenylmethane

are metered during several hours. In the process, CO<sub>2</sub> escapes as by-product and a polyimide results directly in solution.

**[0119]** The highly viscous solution obtained has a golden color, a solids content of 27% and a viscosity of 48 Pa·s.

**[0120]** The molar masses are determined by gel permeation chromatography as follows:  $M_n=76\,600\text{ g}\cdot\text{mol}^{-1}$ ,  $M_p=130\,500\text{ g}\cdot\text{mol}^{-1}$ ,  $M_w=146\,200\text{ g}\cdot\text{mol}^{-1}$  PDI=1.91

###### Example 3

###### Producing a P84 Type 70 Polyimide Solution in N-methylpyrrolidone

**[0121]** A 3 l glass reactor equipped with stirrer and reflux condenser is initially charged with 1800 g of anhydrous N-methylpyrrolidone. A quantity of 456.4 g of 3,3',4,4'-benzophenonetetracarboxylic dianhydride are dissolved therein and the solution is heated to 90° C. To this solution is added 0.45 g of sodium hydroxide. Under nitrogen, 266.8 g of a mixture of 64% of 2,4-tolylene diisocyanate, 16% of 2,6-tolylene diisocyanate and 20% of 4,4'-diisocyanatodiphenylmethane are metered during several hours. In the process, CO<sub>2</sub> escapes as by-product and a polyimide results directly in solution.

**[0122]** The highly viscous solution obtained has a golden color, a solids content of 25% and a viscosity of 45 Pa·s.

**[0123]** The molar masses are determined by gel permeation chromatography as follows:  $M_n=65\,700\text{ g}\cdot\text{mol}^{-1}$ ,  $M_p=107\,200\text{ g}\cdot\text{mol}^{-1}$ ,  $M_w=147\,000\text{ g}\cdot\text{mol}^{-1}$  PDI=2.24

###### Example 4

###### Producing a P84 Type 70 Polyimide Solution in N-ethylpyrrolidone

**[0124]** A 3 l glass reactor equipped with stirrer and reflux condenser is initially charged with 1622 g of anhydrous N-ethylpyrrolidone. A quantity of 456.4 g of 3,3',4,4'-benzophenonetetracarboxylic dianhydride are dissolved therein and the solution is heated to 90° C. To this solution is added 0.45 g of sodium hydroxide. Under nitrogen, 266.8 g of a mixture of 64% of 2,4-tolylene diisocyanate, 16% of 2,6-tolylene diisocyanate and 20% of 4,4'-diisocyanatodiphenylmethane are metered during several hours. In the process, CO<sub>2</sub> escapes as by-product and a polyimide results directly in solution.

**[0125]** The highly viscous solution obtained has a golden color, a solids content of 27% and a viscosity of 87 Pa·s.

**[0126]** The molar masses are determined by gel permeation chromatography as follows:  $M_n=64\,600\text{ g}\cdot\text{mol}^{-1}$ ,  $M_p=105\,200\text{ g}\cdot\text{mol}^{-1}$ ,  $M_w=144\,700\text{ g}\cdot\text{mol}^{-1}$  PDI=2.24

###### Example 5

###### Producing a P84 T100 Polyimide Solution in Dimethylformamide

**[0127]** A 3 l glass reactor equipped with stirrer and reflux condenser is initially charged with 1800 g of anhydrous dimethylformamide. A quantity of 473.6 g of 3,3',4,4'-benzophenonetetracarboxylic dianhydride are dissolved therein and the solution is heated to 90° C. To this solution is added 1.8 g of diazabicyclooctane. Under nitrogen, 254.4 g of a mixture of 2,4-tolylene diisocyanate are metered during several hours. In the process, CO<sub>2</sub> escapes as by-product and a polyimide results directly in solution.

**[0128]** The highly viscous solution obtained has a golden color, a solids content of 25% and a viscosity of 59 Pa·s.

**[0129]** The molar masses are determined by gel permeation chromatography as follows: Mn=82 100 g·mol<sup>-1</sup>, Mp=151 500 g·mol<sup>-1</sup>, Mw=181 900 g·mol<sup>-1</sup> PDI=2.21

#### Example 6

##### Producing a P84 T80 Polyimide Solution in Dimethylformamide

**[0130]** A 3 l glass reactor equipped with stirrer and reflux condenser is initially charged with 1622 g of anhydrous dimethylformamide. A quantity of 473.6 g of 3,3',4,4'-benzophenonetetracarboxylic dianhydride are dissolved therein and the solution is heated to 90° C. To this solution is added 1.8 g of diazabicyclooctane. Under nitrogen, 254.4 g of a mixture of 80% of 2,4-tolylene diisocyanate and 20% of 2,6-tolylene diisocyanate are metered during several hours. In the process, CO<sub>2</sub> escapes as by-product and a polyimide results directly in solution.

**[0131]** The highly viscous solution obtained has a golden color, a solids content of 27% and a viscosity of 108 Pa·s.

**[0132]** The molar masses are determined by gel permeation chromatography as follows: Mn=83 800 g·mol<sup>-1</sup>, Mp=152 300 g·mol<sup>-1</sup>, Mw=173 800 g·mol<sup>-1</sup> PDI=2.07

#### Example 7

##### Producing a P84 HT Polyimide Solution in Dimethylformamide

**[0133]** A 3 l glass reactor equipped with stirrer and reflux condenser is initially charged with 1800 g of anhydrous dimethylformamide. A quantity of 316.4 g of 3,3',4,4'-benzophenonetetracarboxylic dianhydride and 142.8 g of pyromellitic dianhydride are dissolved therein and the solution is heated to 90° C. To this solution is added 1.8 g of diazabicyclooctane. Under nitrogen, 283.4 g of a mixture of 80% of 2,4-tolylene diisocyanate and 20% of 2,6-tolylene diisocyanate are metered during several hours. In the process, CO<sub>2</sub> escapes as by-product and a polyimide results directly in solution.

**[0134]** The highly viscous solution obtained has a golden color, a solids content of 27% and a viscosity of 70 Pa·s.

**[0135]** The molar masses are determined by gel permeation chromatography as follows: Mn=75 500 g·mol<sup>-1</sup>, Mp=122 200 g·mol<sup>-1</sup>, Mw=150 900 g·mol<sup>-1</sup> PDI=2.00

#### Example 8

##### Producing a P84 MDI Polyimide Solution in Dimethylformamide

**[0136]** A 3 l glass reactor equipped with stirrer and reflux condenser is initially charged with 1500 g of anhydrous dimethylformamide. A quantity of 369.2 g of 3,3',4,4'-benzophenonetetracarboxylic dianhydride are dissolved therein and the solution is heated to 90° C. To this solution is added 1.5 g of diazabicyclooctane. Under nitrogen, 222.3 g of 2,4,6-trimethyl-1,3-phenylene diisocyanate are metered during several hours. In the process, CO<sub>2</sub> escapes as by-product and a polyimide results directly in solution.

**[0137]** The highly viscous solution obtained has a pale yellow color, a solids content of 25% and a viscosity of 5 Pa·s.

**[0138]** The molar masses are determined by gel permeation chromatography as follows: Mn=55 200 g·mol<sup>-1</sup>, Mp=95 000 g·mol<sup>-1</sup>, Mw=112 000 g·mol<sup>-1</sup> PDI=2.03

#### Film Production and Intrinsic Gas Permeabilities

**[0139]** The polymerization solutions are filtered neat through a 15μ metal sieve. The films are produced using an instrument from Elcometer (Elcometer 4340) with an applicator. Glass plates are coated with the polymer solutions using an applicator and a gap size of 250μ. The solvent is subsequently evaporated off in a circulating air drying cabinet at 70° C. (0.5 h), 150° C. (2 h) and 250° C. (12 h). The films are then virtually free of solvents (content <0.1%) and are detached from the glass plates. The films obtained have a thickness of about 30 to 40 μm. None of the films was brittle and all exhibited good mechanical properties. These films were then examined under the microscope to find imperfection-free places and circularly round samples having a diameter of 46 mm are cut out. These samples are then emplaced into the self-built gas permeation apparatus and the permeability of various gases is determined by the vacuum method.

**[0140]** This involves pressurizing the films with a single gas (e.g. nitrogen, oxygen, methane or carbon dioxide) at various pressures and recording the increase in pressure on the permeate side. This is used to calculate the permeability in barrers (10<sup>-6</sup> cm<sup>3</sup>·cm<sup>-2</sup>·s<sup>-1</sup>·cmHg<sup>-1</sup>). Some examples are adduced in what follows.

#### Example 9

##### Gas Permeabilities of Various Polymers from Examples Above

**[0141]**

Polymer	Thickness (μm)	Gas	Trans-membrane pressure (bar)	Permeability (barrer)	Selectivity
Example 2 (P84 type 70)		O <sub>2</sub>		0.182	10.0
		N <sub>2</sub>		0.018	
		CO <sub>2</sub>		0.571	
		CH <sub>4</sub>		0.008	
Example 5 (P84 T100)	37.9	O <sub>2</sub>	10.9	0.250	30.9
		N <sub>2</sub>	11.0	0.008	
		CO <sub>2</sub>	25.2	0.622	
		CH <sub>4</sub>	17.5	0.005	
Example 6 (P84 T80)	36.0	O <sub>2</sub>	10.7	0.280	21.2
		N <sub>2</sub>	11.1	0.013	
		CO <sub>2</sub>	28.4	0.696	
		CH <sub>4</sub>	31.5	0.003	
Example 7 (P84 HT)	41.2	O <sub>2</sub>	12.0	0.53	13.2
		N <sub>2</sub>	12.3	0.04	
		CO <sub>2</sub>	36.8	1.69	
		CH <sub>4</sub>	36.4	0.01	

#### Additizing the Polymerization Solution

#### Example 10

##### Producing a Casting Solution from P84 Type 70 for Production of Polyimide Hollow Fibers

**[0142]** In a 3 l stirred tank of glass with powerful stirrer, 1168 g of P84 type 70 solution in dimethylformamide from Example 2 are admixed with a mixture of 94.1 g of tetrahy-

drofuran and 40.3 g of isopropanol added dropwise at room temperature. In the course of addition, the polymer briefly precipitates at the drop entry point, but quickly redissolves again. Stirring is continued until a homogeneous solution is produced. This homogeneous solution is then filtered through a sieve having a mesh size of 15 $\mu$  and left to stand for 2 days without stirring. The casting solution obtained has a solids content of 23.5%, a dimethylformamide content of 66.5%, a tetrahydrofuran content of 7% and an isopropanol content of 3%.

#### Example 11

##### Producing a Casting Solution from P84 Type 70 for Production of Polyimide Hollow Fibers

**[0143]** In a 3 l stirred tank of glass with powerful stirrer, 1034 g of P84 type 70 solution in dimethylformamide from Example 2 are admixed with a mixture of 58.6 g of tetrahydrofuran and 46.9 g of isopropanol added dropwise at room temperature. In the course of addition, the polymer briefly precipitates at the drop entry point, but quickly redissolves again. Stirring is continued until a homogeneous solution is produced. This homogeneous solution is then filtered through a sieve having a mesh size of 15 $\mu$  and left to stand for 2 days without stirring. The casting solution obtained has a solids content of 23.8%, a dimethylformamide content of 67.2%, a tetrahydrofuran content of 5% and an isopropanol content of 4%.

#### Example 12

##### Producing a Casting Solution from P84 Type HT for Production of Polyimide Hollow Fibers

**[0144]** In a 3 l stirred tank of glass with powerful stirrer, 1034 g of P84 type HT solution in dimethylformamide from Example 7 are admixed with a mixture of 47 g of tetrahydrofuran and 65 g of isopropanol added dropwise at room temperature. In the course of addition, the polymer briefly precipitates at the drop entry point, but quickly redissolves again. Stirring is continued until a homogeneous solution is produced. This homogeneous solution is then filtered through a sieve having a mesh size of 15 $\mu$  and left to stand for 2 days without stirring. The casting solution obtained has a solids content of 23.6%, a dimethylformamide content of 66.9%, a tetrahydrofuran content of 4% and an isopropanol content of 5.5%.

#### Example 13

##### Producing a Casting Solution from P84 T100 for Production of Polyimide Hollow Fibers

**[0145]** In a 3 l stirred tank of glass with powerful stirrer, 1034 g of P84 T100 solution in dimethylformamide from Example 5 are admixed with a mixture of 46.8 g of tetrahydrofuran and 58.5 g of isopropanol added dropwise at room temperature. In the course of addition, the polymer briefly precipitates at the drop entry point, but quickly redissolves again. Stirring is continued until a homogeneous solution is produced. This homogeneous solution is then filtered through a sieve having a mesh size of 15 $\mu$  and left to stand for 2 days without stirring. The casting solution obtained has a solids

content of 22.1%, a dimethylformamide content of 68.9%, a tetrahydrofuran content of 5% and an isopropanol content of 4%.

#### Example 14

##### Producing a Casting Solution from P84 Type for Production of Flat Sheet Membranes for Organophilic Nanofiltration

**[0146]** In a 3 l stirred tank of glass with powerful stirrer, 1034 g of P84 type 70 solution in dimethylformamide from Example 2 are admixed with 258.5 g of tetrahydrofuran added dropwise at room temperature. Stirring is continued until a homogeneous solution is produced. This homogeneous solution is then filtered through a sieve having a mesh size of 15 $\mu$  and left to stand for 2 days without stirring. The casting solution obtained has a solids content of 21.6%, a dimethylformamide content of 58.4% and a tetrahydrofuran content of 20%.

#### Hollow Fiber Production

#### Example 15

##### Hollow Fiber Production from a Casting Solution with P84 Type 70 in Dimethylformamide from Example 10

**[0147]** The devolatilized, filtered and additized solution of P84 type 70 in dimethylformamide from Example 10 is thermostated to 50° C. and gear pumped through a two-material die. Flux is 162 g/h. While the polymer solution is conveyed in the outer region of the two-material die, a mixture of 70% of dimethylformamide and 30% of water is conveyed in the inner region in order to produce the hole in the hollow fiber. Flux is 58 ml/h. After a distance of 40 cm, the hollow fiber enters cold water at 10° C. The hollow fiber is enveloped here with a tube. This tube is flooded with a 2 l/min stream of nitrogen, tube internal temperature being 41° C. The fiber is then pulled through a water wash bath and finally wound up at a speed of 15 m/min. After extraction with water for several hours, the hollow fibers are dipped first in ethanol and then in heptane and subsequently air dried to obtain hollow fibers having an outer diameter of 412 $\mu$ , a hole diameter of 250 $\mu$  and a wall thickness of 81 $\mu$ .

**[0148]** Single gas measurements gave the following permeances for the hollow fibers at a transmembrane pressure of 5 bar:

oxygen: 1.450 GPU  
 nitrogen: 0.165 GPU  
 carbon dioxide: 6.03 GPU  
 methane: 0.084 GPU

**[0149]** Single gas selectivities are thus 8.8 as between oxygen and nitrogen and 71.9 as between carbon dioxide and methane

**[0150]** Single gas measurements gave the following permeances for the hollow fibers at a transmembrane pressure of 40 bar:

carbon dioxide: 8.99 GPU  
 methane: 0.101 GPU

Single gas selectivities are 88.5 as between carbon dioxide and methane

## Example 16

Hollow Fiber Production from a Casting Solution  
with P84 Type 70 in Dimethylformamide from  
Example 11

**[0151]** The devolatilized, filtered and additized solution of P84 type 70 in dimethylformamide from Example 11 is thermostated to 50° C. and gear pumped through a two-material die. Flux is 162 g/h. While the polymer solution is conveyed in the outer region of the two-material die, a mixture of 70% of dimethylformamide and 30% of water is conveyed in the inner region in order to produce the hole in the hollow fiber. Flux is 58 ml/h. After a distance of 42 cm, the hollow fiber enters cold water at 10° C. The hollow fiber is enveloped here with a tube. This tube is flooded with a 2 l/min stream of nitrogen, tube internal temperature being 46° C. The fiber is then pulled through a water wash bath and finally wound up at a speed of 24 m/min. After extraction with water for several hours, the hollow fibers are dipped first in ethanol and then in heptane and subsequently air dried to obtain hollow fibers having an outer diameter of 310 $\mu$ , a hole diameter of 188 $\mu$  and a wall thickness of 61 $\mu$ .

**[0152]** Single gas measurements gave the following permeances for the hollow fibers at a transmembrane pressure of 9 bar:

oxygen: 1.463 GPU

nitrogen: 0.164 GPU

Single gas selectivities are thus 8.9 as between oxygen and nitrogen

## Example 17

Hollow Fiber Production from a Casting Solution  
with P84 T100 in Dimethylformamide from  
Example 13

**[0153]** The devolatilized, filtered and additized solution of P84 T100 in dimethylformamide from Example 13 is thermostated to 50° C. and gear pumped through a two-material die. Flux is 162 g/h. While the polymer solution is conveyed in the outer region of the two-material die, a mixture of 70% of dimethylformamide and 30% of water is conveyed in the inner region in order to produce the hole in the hollow fiber. Flux is 58 ml/h. After a distance of 42 cm, the hollow fiber enters cold water at 10° C. The hollow fiber is enveloped here with a tube. This tube is flooded with a 2 l/min stream of nitrogen, tube internal temperature being 46° C. The fiber is then pulled through a water wash bath and finally wound up at a speed of 20 m/min. After extraction with water for several hours, the hollow fibers are dipped first in ethanol and then in heptane and subsequently air dried to obtain hollow fibers having an outer diameter of 339 $\mu$ , a hole diameter of 189 $\mu$  and a wall thickness of 75 $\mu$ .

**[0154]** Single gas measurements gave the following permeances for the hollow fibers at a transmembrane pressure of 9 bar:

oxygen: 0.564 GPU

nitrogen: 0.072 GPU

carbon dioxide: 1.679

methane: 0.023

**[0155]** Single gas selectivities are thus 7.8 as between oxygen and nitrogen and 71.6 as between carbon dioxide and methane

## Example 18

Hollow Fiber Production from a Casting Solution  
with P84 HT in Dimethylformamide

**[0156]** The devolatilized, filtered and additized solution of P84 HT in dimethylformamide from Example 12 is thermostated to 50° C. and gear pumped through a two-material die. Flux is 162 g/h. While the polymer solution is conveyed in the outer region of the two-material die, a mixture of 70% of dimethylformamide and 30% of water is conveyed in the inner region in order to produce the hole in the hollow fiber. Flux is 58 ml/h. After a distance of 15 cm, the hollow fiber enters cold water at 10° C. The hollow fiber is enveloped here with a tube. This tube is flooded with a 1 l/min stream of nitrogen, tube internal temperature being 40° C. The fiber is then pulled through a water wash bath and finally wound up at a speed of 24 m/min. After extraction with water for several hours, the hollow fibers are dipped first in ethanol and then in heptane and subsequently air dried to obtain hollow fibers having an outer diameter of 306 $\mu$ , a hole diameter of 180 $\mu$  and a wall thickness of 63 $\mu$ .

**[0157]** Single gas measurements gave the following permeances for the hollow fibers at a transmembrane pressure of 10 bar:

carbon dioxide: 6.0 GPU

methane: 0.2 GPU

**[0158]** Single gas selectivities are thus 30 as between carbon dioxide and methane

## Example 19

Hollow Fiber Production from a Polymerization  
Solution with P84 Ht in Dimethylformamide from  
Example 7

**[0159]** The devolatilized filtered solution of P84 HT in dimethylformamide from Example 7 is thermostated to 50° C. and gear pumped through a two-material die. Flux is 162 g/h. While the polymer solution is conveyed in the outer region of the two-material die, a mixture of 70% of dimethylformamide and 30% of water is conveyed in the inner region in order to produce the hole in the hollow fiber. Flux is 58 ml/h. After a distance of 15 cm, the hollow fiber enters cold water at 10° C. The hollow fiber is enveloped here with a tube. This tube is flooded with a 1 l/min stream of nitrogen, tube internal temperature being 70° C. The fiber is then pulled through a water wash bath and finally wound up at a speed of 24 m/min. After extraction with water for several hours, the hollow fibers are dipped first in ethanol and then in heptane and subsequently air dried to obtain hollow fibers having an outer diameter of 307 $\mu$ , a hole diameter of 189 $\mu$  and a wall thickness of 59 $\mu$ .

**[0160]** Single gas measurements gave the following permeances for the hollow fibers at a transmembrane pressure of 10 bar:

carbon dioxide: 3.37 GPU

methane: 0.051 GPU

Single gas selectivities are thus 66 as between carbon dioxide and methane

**[0161]** The fiber was additionally measured at higher pressures in order to measure plasticization characteristics and pressure stability.

Pressure [bar]	CO2 permeance [GPU]	Methane permeance [GPU]	Selectivity
10	3.365	0.051	66
20	3.199	0.045	72
30	3.535	0.034	103
40	4.025	0.042	96
50	4.376	0.033	131
60	4.300	0.026	165
70	—	0.027	—
90	—	0.014	—

### Flat Sheet Membrane Production

#### Example 20

##### Producing a Flat Sheet Membrane from P84 Type 70

**[0162]** A flat sheet membrane rig is used to produce 35 cm wide membranes from a casting solution described in Example 14. For this, the casting solution is coated using an application and a casting gap of 200 $\mu$  onto a calendered polyester fleece having a basis weight of 100 g/m<sup>2</sup> and a speed of 5 m/min. The coated polyester fleece is then passed through a shaft through which nitrogen is flowed. The speed of flow is 339 m/h. The residence time thus achieved is 3 s. The coated fleece then dips into cold water at 10° C. The crude membrane is then wound up wet.

**[0163]** Subsequently, the membrane is at 70° C. extracted in water and impregnated with a conditioning agent (25% of polyethylene glycol dimethyl ether (PGDME 250 from Clariant) in water). It is dried in a festoon dryer at a temperature of 60° C.

**[0164]** The membrane is characterized in a Milipore stirred cell at a pressure of 5 bar. The solvent used is heptane in which hexaphenylbenzene is dissolved in a concentration of 12 mg/l. Measurement revealed a flux of 1.7 l·m<sup>-2</sup>·h<sup>-1</sup>·bar<sup>-1</sup> coupled with a retention of 94%

**[0165]** The membrane is subsequently also tested at a pressure of 30 bar and 30° C. in toluene. Oligostyrenes are used as test molecules. Flux in this test with toluene was 90 l·m<sup>-2</sup>·h<sup>-1</sup>. The membrane exhibits very high retention over the entire molar mass range and has a sharp cut-off in the region between 200 and 300 daltons (see FIG. 3).

### Crosslinking the Membranes with Diamines

#### Example 21

##### Crosslinking a Flat Sheet Membrane with Amines

**[0166]** The flat sheet membrane from Example 20 was placed for 16 h into a 0.1% ethanolic solution of an oligo-

ethyleneimine (#468533, Aldrich, typical molecular weight 423, contains 5-20% of tetraethylene-pentamine). The membrane crosslinks and exhibits no solubility in hexane, heptane, toluene, xylene, acetone, butanone, methanol, ethanol, isopropanol, tetrahydrofuran, dichloromethane, chloroform, dimethylformamide, dimethylacetamide, N-methyl-pyrrolidone, dimethyl sulfoxide and ethyl acetate.

**[0167]** The membrane is characterized in a Milipore stirred cell at a pressure of 5 bar. The solvent used is dimethylformamide in which hexaphenylbenzene is dissolved in a concentration of 2.2 mg/l. Measurement revealed a flux of 1.3 l·m<sup>-2</sup>·h<sup>-1</sup>·bar<sup>-1</sup> coupled with a retention of 89%.

### Example 22

#### Crosslinking a Hollow Fiber Membrane with Amines

**[0168]** The hollow fiber membrane from Example 19 was placed for 16 h into a 0.1% solution of hexamethylenediamine in ethanol. The membrane crosslinks and exhibits no solubility in hexane, heptane, toluene, xylene, acetone, butanone, methanol, ethanol, isopropanol, tetrahydrofuran, dichloromethane, chloroform, dimethylformamide, dimethylacetamide, N-methyl-pyrrolidone, dimethyl sulfoxide and ethyl acetate.

### FIGURE DESCRIPTION

**[0169]** FIG. 1: Influence of concentrations of P84 type 70 in DMF on viscosity of solution: comparing a P84 polymerization solution and a P84 solution prepared from a precipitated and redissolved polymer at 25° C.

**[0170]** FIG. 2: Cross sections of hollow fiber membranes with macrovoids (picture at left) and without macrovoids (picture at right)

**[0171]** FIG. 3:

**[0172]** Application test of membrane from Example 20.

1. A process for producing a polyimide membrane, the process comprising:

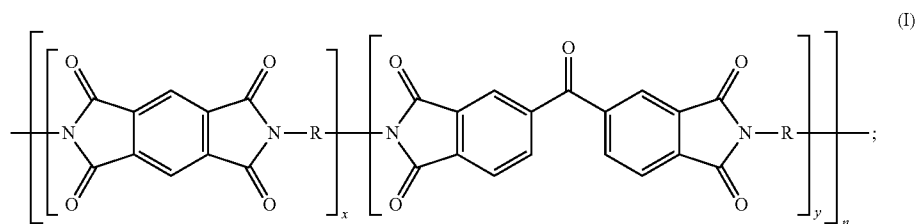
- forming a casting solution comprising a polyimide; and
- forming a polyimide membrane from the casting solution,

by a phase inversion process,

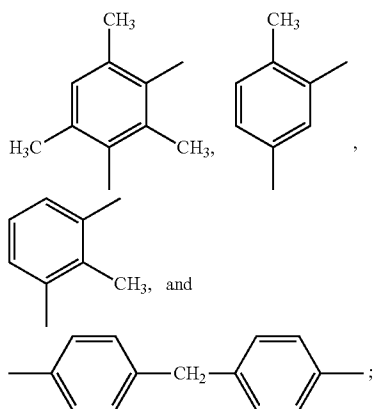
wherein the polyimide has not been isolated as a solid material and then redissolved prior to forming said casting solution.

2. The process of claim 1, further comprising forming the polyimide from an aromatic dianhydride, an aromatic diisocyanate, an aprotic dipolar solvent, or mixtures thereof.

3. The process of claim 1, wherein the polyimide is a polyimide of formula (I):



R is selected from the group consisting of



the polyimide is crosslinked with at least one aliphatic diamine or polyethyleneimine;

a crosslinking of the polyimide occurs by dipping the polyimide into a solution of diamine in water or at least one alcohol;

a crosslinking of the polyimide occurs at temperatures between 0 and 90° C.;

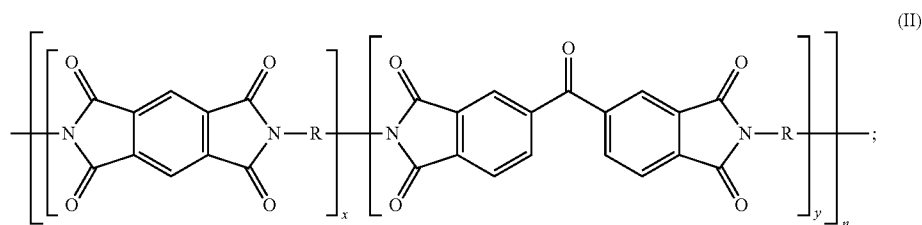
a crosslinking time for a crosslinking of the polyimide is between 10 seconds and 16 hours; and

a concentration of diamine in a crosslinking of the polyimide is between 0.01% by weight and 50% by weight.

9. A polyimide membrane obtained by the process of claim 1.

10. The polyimide membrane of claim 9, comprising a polyimide having an  $M_p > 100\,000\text{ g}\cdot\text{mol}^{-1}$  and a PDI in the range from 1.7 to 2.3.

11. The polyimide membrane of claim 9, wherein: the polyimide is a polyimide of formula (II):



$0 < x < 0.5$ ; and

$1 > y > 0.5$ .

4. The process of claim 1, wherein the casting solution further comprises a water-soluble additive.

5. The process of claim 1, wherein the forming (b) of the polyimide membrane comprises coating a backing fleece with the casting solution.

6. The process of claim 1, wherein the polyimide membrane is crosslinked with at least one aliphatic diamine or polyethyleneimine.

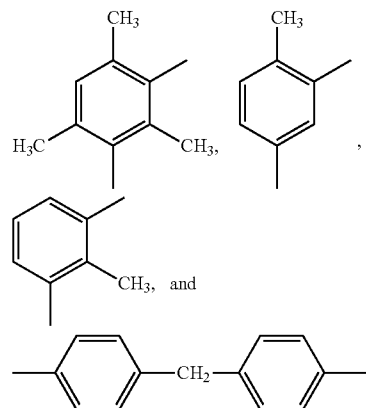
7. The process of claim 1, wherein the forming (b) of the polyimide membrane comprises forming an integrally asymmetrically hollow fiber membrane by spinning a hollow fiber with a two-material die from the casting solution and a bore solution in a continuous process.

8. The process of claim 7, wherein the forming (b) satisfies at least one condition selected from the group consisting of: the bore solution comprises a mixture of water and an alcohol with at least one selected from the group consisting of dimethylformamide, dimethylacetamide, N-methylpyrrolidinone, N-ethylpyrrolidinone, sulfolane, and dimethyl sulfoxide;

the two-material die is at a distance of 1 to 60 cm from a spin bath of water into which the hollow fiber is spun and an integrally asymmetrical hollow fiber membrane is formed by precipitating the polyimide;

during the spinning a hollow thread is subjected to a flow of a dry thermostated stream of nitrogen or air before entry into a spin bath in order to adjust separation properties of the polyimide membrane;

R is selected from the group consisting of



$0 < x < 0.5$ ; and

$1 > y > 0.5$ .

12. The polyimide membrane of claim 9, in the form of: a micro-, ultra- or nanofiltration membrane suitable for separating homogeneous dissolved or particulate products from organic solvents or from water; or a no-pore membrane suitable for separation of gases.

13. The polyimide membrane of claim 9, in the form of: an integrally asymmetrical flat sheet membrane on a backing fleece; or an integrally asymmetrical hollow fiber membrane.

**14.** The polyimide membrane of claim **13**, wherein the polyimide membrane is an integrally asymmetrical hollow fiber membrane suitable for separating a wide variety of gas mixtures.

**15.** A casting solution comprising a polyimide, wherein the polyimide is not isolated as a solid material and then redissolved.

**16.** The process of claim **4**, wherein the water-soluble additive is at least one selected from the group consisting of a water-miscible solvent, a non-solvent, and a pore-former.

**17.** The process of claim **5**, further comprising vaporizing some solvent of the casting solution with a dry thermostated stream of nitrogen or air, after the coating of the backing fleece with the casting solution, to adjust separation limits of the polyimide membrane.

**18.** The process of claim **6**, satisfying at least one condition selected from the group consisting of:

the polyimide membrane is crosslinked with at least one aliphatic diamine selected from the group consisting of diaminoethane, diaminopropane, diaminobutane, diaminopentane, diaminohexane, diaminooctane,

diaminodecane, diaminododecane, bis-4,4'-(aminomethyl)benzene, a oligoethyleneimine, and a polyethyleneimine;

the polyimide membrane is crosslinked by dipping into a solution of diamine in water or at least one alcohol;

the polyimide membrane is crosslinked at temperatures between 0 and 90° C.;

a crosslinking time is between 10 seconds and 16 hours; and

a concentration of diamine is between 0.01% by weight and 50% by weight.

**19.** The polyimide membrane of claim **13**, wherein the polyimide membrane is an integrally asymmetrical flat sheet membrane on a backing fleece of polyphenylene sulfide, polyethylene terephthalate or polypropylene.

**20.** The polyimide membrane of claim **14**, wherein the integrally asymmetrical hollow fiber membrane is suitable for separating at least one selected from the group consisting of: methane and carbon dioxide; oxygen and nitrogen; hydrogen and process gases; and water vapor and a gas or a mixture of gases.

\* \* \* \* \*