

Device for UV-spectrometric analysis of gaseous compounds

TECHNICAL FIELD

This invention relates to a device for UV-spectrometric analysis of gaseous
5 compounds according to the preamble of claim 1. In particular, the invention
relates to measurements of UV radiation with wavelengths in the far UV
region.

BACKGROUND OF THE INVENTION

10 Spectrophotometric measurement of gas and vapour absorption is well
known in various applications. An example is analytical chemistry where
ultraviolet (UV) spectrometers are combined with gas chromatography (GC)
for determining gaseous compounds. Examples of equipment for such use
are disclosed in US 4668091 and US 6305213. Various UV spectrometers or
15 spectrographs as well as gas chromatographs are commercially available.

Ultraviolet radiation has a wavelength shorter than that of visible light but
longer than X-rays, in the range 10-400 nm, and its spectrum can be
subdivided in a number of ways, for instance, near UV 400-300 nm, middle
20 UV 300-200 nm, far UV 200-122 and vacuum UV 200-100 nm. Ordinary
glass is opaque to shorter wavelengths. Windows made of quartz glass are
normally used when analyzing wavelengths shorter than around 350 nm.
Wavelengths shorter than around 190 nm, i.e. radiation in the far UV or
vacuum region, are normally absorbed too strongly also in quartz. In this
25 wavelength region it is possible to use windows and other optical elements
made of alkaline earth metal fluorides, such as MgF_2 . However, such
materials are not as resistant to chemicals as quartz and there are a number
of applications where these materials are not suitable. This is in particular a
problem at elevated temperatures.

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UV wavelengths in the far UV or vacuum region is of great interest in
chemical analysis since most chemical compounds absorb light in that

region. Many compounds that cannot be detected at longer wavelengths may be analyzed using radiation in the far UV region. However, also air, or rather oxygen (O₂) and water vapour, absorbs light strongly in this region (in the range below around 190 nm) which means that the analytic equipment must
5 be specially adapted to avoid interference from air. For this purpose there are, for instance, UV spectrographs commercially available that are adapted to be evacuated or to be filled with an inert gas, such as nitrogen (N₂) (which absorbs UV only at very short wavelengths).

10 Equipment for analyzing gases and/or vapours with UV spectroscopy typically includes an elongated heated channel (cell) that accommodates a sample gas during measurement, a UV-source (e.g. a deuterium lamp) positioned at one end the channel and a UV detector (a UV spectrograph) located at the opposite end of the channel. Gas tight and UV-transparent
15 windows, typically made of quartz, are provided at each end of the channel. The channel may be provided with inlets and outlets for leading sample gas, e.g. from a GC, and carrier gas to and from the channel in a continuous manner. UV light emitted by the UV-source passes through the window of the UV-source, through the entrance window of the channel and into and through
20 the channel where part of the light is absorbed in the sample gas. Remaining radiation passes the exit window of the channel and enters the UV spectrograph through a window or slit. The UV spectrograph measures the intensity at different wavelengths of the UV radiation that has passed through the channel and the absorption spectra obtained is used to identify and
25 quantify the compounds present in the sample gas.

To adapt such equipment to measurements of radiation in the far UV region a UV transparent (e.g. evacuated) UV detector can be used and the quartz windows can be replaced by windows made of an alkaline earth metal
30 fluoride. However, such an adaptation works well only in situations where the sample gas does not contain compounds that may decompose the sensitive material of the channel windows. Such degradation of the channel windows

is difficult to prevent in many cases, in particular at elevated temperatures, such as when the sample gas is supplied from a heated column of a gas chromatograph,

- 5 Accordingly, there is a need for improvements in the field of equipment for use in UV spectroscopy analysis of a gas, in particular for analyses in the far UV region.

SUMMARY OF THE INVENTION

- 10 An object of this invention is to provide a device for UV-spectrometric analysis of gaseous compounds that exhibit improved capabilities of performing analysis of radiation in the far UV region compared to conventional devices. A further object is to provide a less costly window arrangement. This object is achieved by the device defined by the technical
15 features contained in independent claim 1. The dependent claims contain advantageous embodiments, further developments and variants of the invention.

- The invention concerns a device for UV-spectrometric analysis of gaseous
20 compounds, said device comprising: a measurement channel intended to accommodate a flow of sample gas to be analyzed, a window member transparent for ultraviolet radiation arranged at a first end of the measurement channel, a radiation source capable of generating ultraviolet radiation arranged to emit radiation through the window member and into the
25 measurement channel, and a spectrographic member for measuring of ultraviolet radiation emitted by the radiation source arranged at a second, opposite, end of the measurement channel, wherein the device is arranged such that ultraviolet radiation entering the measurement channel at the first end can propagate through the measurement channel, interact with the
30 accommodated gas and be measured by the spectrographic member at the second end of the measurement channel.

The invention is characterized in that the spectrographic member is provided with an opening through which ultraviolet radiation passes into the spectrographic member during operation of the device, wherein the second end of the measurement channel is open towards the spectrographic member such that an inside of the spectrographic member and the measurement channel are in communication via said opening, wherein the spectrographic member is arranged to be filled with a protection gas, and wherein the device is arranged such that protection gas fed to the spectrographic member is allowed to flow through said opening and into the measurement channel.

By feeding protection gas through the spectrographic member via the opening (typically a slit) and further into the open-ended measurement channel, the protection gas forms a "gas window" that dispenses with the need for arranging a normal, solid exit window at this side of the measurement channel. Using a proper protection gas, such as N₂, the gas window formed can be made to be transparent to UV radiation of sufficiently short wavelength without the use of e.g. MgF₂ which is not only a sensitive material but also costly.

The protection gas entering the measurement channel via the opening in the spectrographic member can work as a carrier gas for the sample gas in the measurement channel before it is lead out. By properly adjusting the gas pressures the sample gas will not flow towards, or at least not through, the opening of the spectrographic member.

The protection gas supplied to the spectrographic member thus has at least two functions: to remove Interfering gases, such as O₂ and H₂O, from the spectrographic member (and at least a part of the measurement channel), and to form a "window" that allows ultraviolet radiation to pass but that prevents sample gas to escape. Depending on the design of e.g. inlets and

outlets of the measurement channel the protection gas supplied to the spectrographic member can also function as carrier gas.

5 The invention thus provides a window that is cost-effective and well-functioning and it also solves the problem of arranging the UV exit window such that UV radiation also in the far UV region can exit the channel and enter the spectrographic member without risking degradation of a sensitive window material.

10 To be able to perform a proper analysis in the far UV region of the sample gas also the UV entrance window at the first end of the measurement channel must be properly arranged so that UV radiation in the far UV region can enter the measurement channel. A conventionally arranged window made of an alkaline earth metal fluoride can be used for this purpose
15 provided that the sample gas does not degrade this sensitive material.

However, the inventive gas-window arrangement is useful also for analysis of longer wavelengths and can be used together with a conventional quartz window, or even a glass window, arranged as UV entrance window at the
20 UV-source side of the measurement channel.

In an embodiment of the invention the first end of the measurement channel is open towards the window member and in that a channel for guiding a protection gas is arranged in connection to the window member such that
25 protection gas fed through the protection gas channel is allowed to flow over and cover the side of the window member facing the measurement channel and to flow further into the measurement channel.

This design has the advantageous effect that it makes it possible to prevent
30 sample gas (and its content of potentially corrosive compounds) in the measurement channel to come into contact with the window member by feeding protection (inert) gas through the protection gas channel. Since the

protection gas passes the window member and then flows into the measurement channel via its open end it stops the sample gas in the channel and protects the window member. This allows the use of a channel entrance window made of an alkaline earth metal fluoride, such as MgF_2 , which is likely to be sensitive to compounds in the sample gas but at the same time is transparent to shorter UV wavelengths.

Nitrogen (N_2) is a suitable protection gas in that it protects the window well and is transparent to UV radiation at wavelengths above around 150 nm. The flow of N_2 or other suitable protection gas is useful also for use as carrier gas and for removing oxygen water vapour from the measurement channel.

This embodiment thus solves the problem of arranging the UV entrance window such that UV radiation in the far UV region can enter the measurement channel without risking degradation of the sensitive window material.

Together with the inventive gas-window arranged at the second end of the device, this allows for UV spectrometric analysis of UV radiation with wavelengths shorter than 190 nm down to around 150 nm depending on the protection gas used.

In an embodiment of the invention a void is arranged adjacent the window member at the side of the window member facing the measurement channel, wherein the void is in communication with the open end of the measurement channel and wherein the void is provided with a void inlet such that when a protection gas is fed to said void inlet, the protection gas is allowed to fill the void and flow further into the measurement channel.

In an embodiment of the invention the measurement channel is provided with at least one inlet for feeding sample gas into the measurement channel and

at least one outlet for leading sample gas and protection gas out from the measurement channel.

5 In an embodiment of the invention the device comprises seal members arranged such as to prevent air that surrounds the device from entering the measurement channel.

10 In an embodiment of the invention the device comprises a gas chromatography column for connection to the sample gas inlet.

In an embodiment of the invention the device comprises a gas flow regulator arranged to regulate the flow of protection gas fed to the protection gas channel.

15 In an embodiment of the invention the gas flow regulator is arranged to regulate the flow of sample gas.

20 In an embodiment of the invention the gas flow regulator is arranged to regulate the flow of protection gas fed to the spectrographic member.

In an embodiment of the invention the measurement channel has an elongated shape, wherein the window member and the spectrographic member are positioned at opposite ends of the measurement channel.

25 In an embodiment of the invention the measurement channel is a silica tube.

In an embodiment of the invention the measurement channel is arranged in a casing adapted to be heated.

30 In an embodiment of the invention the radiation source is a deuterium lamp. Preferably, the window member forms an integrated part of the deuterium lamp.

In an embodiment of the invention the window member is made of an alkaline earth metal fluoride. Preferably, the window member is made of MgF_2 .

- 5 In an embodiment of the invention the spectrographic member comprises reflectors and registration element adapted to short wavelength ultraviolet radiation.

In an embodiment of the invention the window member is made of quartz.

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BRIEF DESCRIPTION OF DRAWINGS

In the description of the invention given below reference is made to the following figure, in which:

- Figure 1 shows, in a schematic view, a first embodiment of the invention,
15 and
Figure 2 shows, in a schematic view, a second embodiment of the invention.

DESCRIPTION OF EXAMPLE EMBODIMENTS OF THE INVENTION

- 20 Figure 1 shows a first embodiment of a device 20 for UV-spectrometric analysis of gaseous compounds. The device 20 comprises a gas flow regulator 1, a spectrographic member 3, a connecting tube 4, a measurement channel 5, a sample injector 7, a gas chromatography packed column 8 and a UV-source 11 in the form of a deuterium lamp.

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- The spectrographic member 3 is in this example a grating spectrograph containing exclusively reflective optical elements 3b with a long registration element capable of registering UV light of very short wavelength. The spectrographic member 3 is provided with an inlet 2 for supplying protection
30 gas into the member 3 and an opening 12 in the form a slit for admitting UV radiation (dashed line 15) that has passed through the measurement channel 5 to enter the spectrographic member 3. The connecting tube 4 is arranged

in a sealed (gas tight) manner around the slit 12 and around a second open end 5b of the measurement channel 5 facing the spectrographic member 3. The tube 4 is arranged as an extension of the measurement channel 5. The spectrographic member 3 is gas tight which means that when the spectrographic member 3 has been filled with protection gas through the inlet 2 the gas will flow out through the slit 12 through the tube 4 and further into the measurement channel 5 via its open end 5b.

The deuterium lamp 11 is positioned at an opposite end of the measurement channel 5 in relation to the spectrographic member 3. A window member 16 that forms an integrated part of the deuterium lamp 11 is facing a first open end 5a of the measurement channel 5. The window member 16 is made of MgF_2 and it is arranged in a sealed (gas tight) manner in association with the first end 5a of the measurement channel 5.

The sample injector 7 is arranged to inject sample gas to the gas chromatography column 8.

The measurement channel 5 is a tube made of pure silica and it is arranged in a housing 17 adapted to be heated such as to keep the measurement channel 5 at a controlled, elevated temperature during operation of the device 20. An inlet 9 for introducing the sample gas into the measurement channel 5 is arranged in the measurement channel 5 in a position relatively close to the second end 5b of the channel 5 facing the spectrographic member 3. An outlet 10 for leading sample gas and protection gas out from the measurement channel 5 is positioned relatively close to the first end 5a of the channel 5 facing the UV source 11. An inlet 6 for protection/inert gas is arranged in association with the first end 5a of the channel 5 facing the UV source 11 so that gas fed to the this inlet 6 enters a void 18 between the window member 16 and the channel 5 so that this gas flushes over, covers and protects the side of the window member 16 facing the channel 5 before the gas continues to flow into the channel 5 through the open end 5a thereof.

The gas flow regulator 1 is arranged such as to feed a carrier gas through the column 8 and to feed protection gas (which in this case is the same gas as the carrier gas: N₂) to the inlet 2 of the spectrographic member 3 and to the inlet 6 at the window member 16.

During operation of the device 20, the gas flows are adjusted such that gas supplied to the spectrographic member 3 via the inlet 2 fills the spectrographic member 3 and flows further into the measurement channel 5 via the slit 12, the tube 4 and the open end 5b of the channel 5. The sample gas entering the channel 5 via the inlet 9 mixes with the flow (of nitrogen) from the spectrographic member 3 and flows through the channel 5 towards the UV source 11.

Simultaneously, protection or inert gas is fed to the inlet 6 at the window member 16. This gas flow fills the void 18 and contacts and covers the window member 16 and flows further into the channel 5 towards the spectrographic member 3. At the position of the outlet 10 the two oppositely directed gas flows in the channel 5 meet and mix, and flow out from the measurement channel 5 via the outlet 10. Preferably, protection/inert gas is allowed to enter the open ends 5a, 5b of the channel 5 before sample gas is introduced through the inlet 9.

Measurement of the sample gas can be carried out while it flows along the measurement channel 5. Because a gas chromatography column 8 that separates compounds is arranged upstream of the measurement channel 5, it is likely that only one or at least only a few sample gas compounds will be present in the measurement channel 5 at the same time. The result of the measurement carried out by the spectrographic member 3 is typically a spectrum showing the absorption of UV-radiation as a function of both wavelength and time.

The device 20 also comprises a control unit (not shown) for controlling the various parts of the device 20, such as the gas flow regulator 1 (for controlling the gas flows), the UV-source 11, the spectrographic member 3, the heating of the casing 17 and the column 8, and the sample injector 7.

5

Example 1:

A device according to the invention was assembled containing a deuterium lamp with an exit window made of MgF_2 (Hamamatsu, Japan), a cell/measurement channel made from a silica tube (I.D. 1 mm, 200 mm long),
10 inserted and fused into a heated Al bar of square cross-section provided with input fitting for gases and silicon seals both at the lamp and the spectrograph (Photon Control, Canada) with a grating 1,200 grooves/mm, optimised for 300 nm, and an optical element CCD Toshiba TCD1304 CCD with the silica glass cover removed (such as to create an opening). All components were
15 mounted on an optical bench made of an aluminium angle bar (100 mm x 100 mm x 10 mm). A stream of inert gas (nitrogen, purity 5.0) entered an opening made in the central part of the cell from a loop injector, used to add the compounds whose spectra were to be measured into the inert gas stream. Additional inert gas entered the inlet close to the lamp window and
20 also to the spectrograph which was properly sealed to ensure that the gas could exit only through the slit by which light from the flow cell enters. The gas flows were controlled in a manner ensuring that the analysed compounds did not contact or even got close to the lamp window. The instrument was used to measure spectra of aromatic hydrocarbons and ketones.

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Example 2

The spectrometer described in Example 1 was connected to a gas chromatograph (Labio, Czech Republic) with a chromatographic column of 2 mm diameter, 4 m long, packed with Supelcoport OV1 sorbent (Sigma
30 Aldrich). Mixtures of organic compounds dissolved in an aliphatic solvent were injected and spectra were measured and recorded as a function of time.

Figure 2 shows a second embodiment of a device 200 for UV-spectrometric analysis of gaseous compounds. A main difference from the embodiment described above is that no protection gas is in this case fed towards the window member 16 to protect it from the sample gas. The inlet 6 now forms an outlet 26 and the previous outlet 10 is removed. In the embodiment shown in figure 2 the sample gas and the nitrogen fed to the spectrographic member 3 flow through the measurement channel 5, passes the window member 16 and leaves the system via the outlet 26. The window member 16 is in this case made of quartz which normally is not affected by the sample gas.

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The embodiment shown in figure 1 can easily be converted to the embodiment of figure 2 by plugging the outlet 10, disconnecting the inlet 6 from the gas flow regulator 1 (as to form the outlet 26), and changing window member 16 (e.g. by changing deuterium lamp from one having an integrated MgF₂-window to one having an integrated quartz window. Conversion back to the embodiment shown in figure 1 is, of course, also easily carried out.

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The “window” formed at the second end 5b of the measurement channel 5 by the gas flowing from the spectrographic member 3 into the measurement channel 5 is useful also in applications where a conventional window is used as the window member 16.

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The invention is not limited by the embodiments described above but can be modified in various ways within the scope of the claims.

25

For instance, the gas inlets and outlets to and from the measurement channel 5 can be arranged in different ways. Generally it is advantageous if the sample gas is allowed to flow along most of the length of the channel 5 to let the UV light (dashed line 15) pass as much sample gas as possible before entering the spectrographic member 3. This increases normally the sensitivity of the analysis. With reference to figure 1, sample gas may enter the measurement channel 5 at a position close to the first end 5a of the channel

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- facing the UV source 11 (with an outlet arranged close to the second end 5b of the channel 5 facing the spectrographic member 3) or sample gas may be fed to a mid section of the measurement channel 5 (preferably with one outlet arranged at each end of the channel 5 so that the flow of sample gas can divide and flow in different directions towards each end of the channel 5). Of course, the gas flows must be regulated so that sample gas is prevented from coming into contact with the window member 16 and from entering the spectrographic member 3.
- 10 Moreover, the UV source 11 does not necessarily have to be a deuterium lamp.

CLAIMS

1. Device (20) for UV-spectrometric analysis of gaseous compounds, said
5 device (20) comprising:
- a measurement channel (5) intended to accommodate a flow of sample gas to be analyzed,
 - a window member (16) transparent for ultraviolet radiation arranged at a first end (5a) of the measurement channel (5),
 - 10 - a radiation source (11) capable of generating ultraviolet radiation arranged to emit radiation through the window member (16) and into the measurement channel (5),
 - a spectrographic member (3) for measuring of ultraviolet radiation emitted by the radiation source (11) arranged at a second, opposite, end (5b) of the
15 measurement channel (5),
- wherein the device (20) is arranged such that ultraviolet radiation entering the measurement channel (5) at the first end (5a) can propagate through the measurement channel (5), interact with the accommodated gas and be measured by the spectrographic member (3) at the second end (5b) of the
20 measurement channel (5),
- characterized in
- that the spectrographic member (3) is provided with an opening (12) through which ultraviolet radiation passes into the spectrographic member (3) during operation of the device (20), wherein the second end (5b) of the
25 measurement channel (5) is open towards the spectrographic member (3) such that an inside of the spectrographic member (3) and the measurement channel (5) are in communication via said opening (12), wherein the spectrographic member (3) is arranged to be filled with a protection gas, and wherein the device (20) is arranged such that protection gas fed to the
30 spectrographic member (3) is allowed to flow through said opening (12) and into the measurement channel (5).

2. Device (20) according to claim 1

characterized in

that the first end (5a) of the measurement channel (5) is open towards the window member (16) and in that a channel (6, 18) for guiding a protection gas is arranged in connection to the window member (16) such that protection gas fed through the protection gas channel (6, 18) is allowed to flow over and cover the side of the window member (16) facing the measurement channel (5) and to flow further into the measurement channel (5).

10

3. Device (20) according to claim 2,

characterized in

that a void (18) is arranged adjacent the window member (16) at the side of the window member (16) facing the measurement channel (5), wherein the void (18) is in communication with the open end (5a) of the measurement channel (5) and wherein the void (18) is provided with a void inlet (6) such that when a protection gas is fed to said void inlet (6), the protection gas is allowed to fill the void (18) and flow further into the measurement channel (5).

20 4. Device (20) according to claim 1 or 2,

characterized in

that the measurement channel (5) is provided with at least one inlet (9) for feeding sample gas into the measurement channel (5) and at least one outlet (10) for leading sample gas and protection gas out from the measurement channel (5).

25 5. Device (20) according to any of the above claims,

characterized in

that it comprises seal members arranged such as to prevent air that surrounds the device from entering the measurement channel (5).

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6. Device (20) according to claim 4,

characterized in

that it comprises a gas chromatography column (8) for connection to the sample gas inlet (9).

5 7. Device (20) according to claim 2 or 3,

characterized in

that it comprises a gas flow regulator (1) arranged to regulate the flow of protection gas fed to the protection gas channel (6, 18).

10 8. Device (20) according to any of the above claims,

characterized in

that the gas flow regulator (1) is arranged to regulate the flow of sample gas.

9. Device (20) according to any of the above claims,

15 characterized in

that the gas flow regulator (1) is arranged to regulate the flow of protection gas fed to the spectrographic member (3).

10. Device (20) according to any of the above claims,

20 characterized in

that the measurement channel (5) has an elongated shape, wherein the window member (16) and the spectrographic member (3) are positioned at opposite ends (5a, 5b) of the measurement channel (5).

25 11. Device (20) according to claim 10,

characterized in

that the measurement channel (5) is a silica tube.

12. Device (20) according to any of the above claims,

30 characterized in

that the measurement channel (5) is arranged in a casing (17) adapted to be heated.

13. Device (20) according to any of the above claims,
c h a r a c t e r i z e d i n
that the radiation source (11) is a deuterium lamp.

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14. Device (20) according to claim 13,
c h a r a c t e r i z e d i n
that the window member (16) forms an integrated part of the deuterium lamp.

10 15. Device (20) according to any of the above claims,
c h a r a c t e r i z e d i n
that the window member (16) is made of an alkaline earth metal fluoride.

15 16. Device (20) according to claim 15,
c h a r a c t e r i z e d i n
that the window member (16) is made of MgF_2 .

17. Device (20) according to any of the above claims,
c h a r a c t e r i z e d i n
20 that the spectrographic member (3) comprises reflectors and registration
element adapted to short wavelength ultraviolet radiation.

18. Device (20) according to any of claims 1-14,
c h a r a c t e r i z e d i n
25 that the window member (16) is made of quartz.

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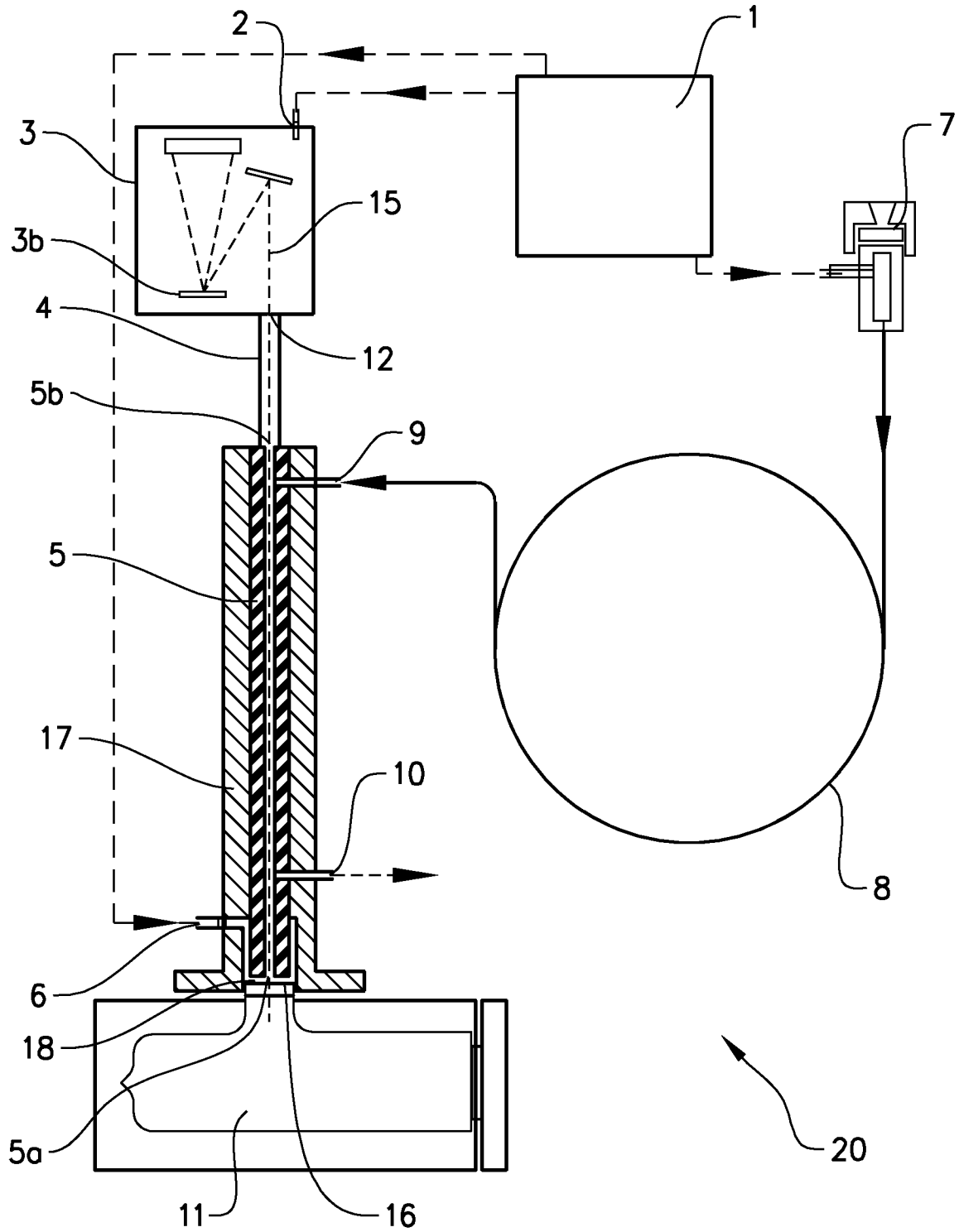


FIG. 1

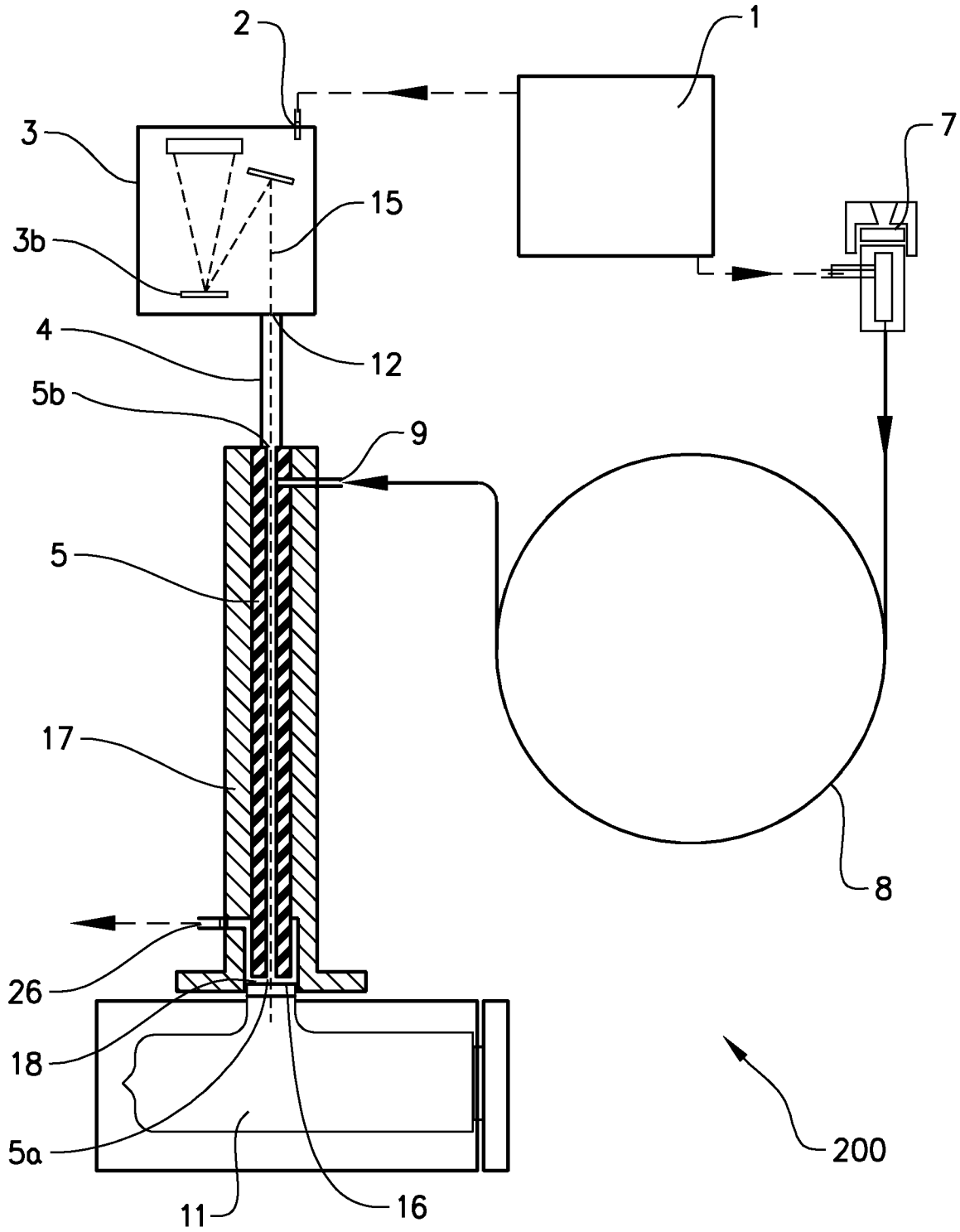


FIG. 2

INTERNATIONAL SEARCH REPORT

International application No
PCT/EP2011/053990

A. CLASSIFICATION OF SUBJECT MATTER
INV. G01N21/15 G01N21/33 G01J3/02 G01N21/09
ADD.
According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED
Minimum documentation searched (classification system followed by classification symbols)
G01N G01J
Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)
EPO-Internal, WPI Data

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
Y	US 6 305 213 B1 (LAGESSON VERNER [SE] ET AL) 23 October 2001 (2001-10-23) cited in the application column 2, line 20 - line 58 figure 1	1-18
Y	US 5 225 681 A (FALK HEINZ [DE] ET AL) 6 July 1993 (1993-07-06) column 2, line 48 - column 3, line 2 figure	1-18
A	DE 103 18 786 A1 (NATTKEMPER ANDREAS [DE]) 25 November 2004 (2004-11-25) paragraph [0001] - paragraph [0002] paragraph [0005] figure 1	1-18
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Further documents are listed in the continuation of Box C.

See patent family annex.

* Special categories of cited documents :

<p>"A" document defining the general state of the art which is not considered to be of particular relevance</p> <p>"E" earlier document but published on or after the international filing date</p> <p>"L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)</p> <p>"O" document referring to an oral disclosure, use, exhibition or other means</p> <p>"P" document published prior to the international filing date but later than the priority date claimed</p>	<p>"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention</p> <p>"X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone</p> <p>"Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.</p> <p>"&" document member of the same patent family</p>
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Date of the actual completion of the international search 30 May 2011	Date of mailing of the international search report 14/06/2011
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Name and mailing address of the ISA/ European Patent Office, P.B. 5818 Patentlaan 2 NL - 2280 HV Rijswijk Tel. (+31-70) 340-2040, Fax: (+31-70) 340-3016	Authorized officer Krametz, Edeltraud
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INTERNATIONAL SEARCH REPORT

International application No
PCT/EP2011/053990

C(Continuation). DOCUMENTS CONSIDERED TO BE RELEVANT		
Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	EP 0 591 758 A1 (KERNFORSCHUNGSZ KARLSRUHE [DE] KARLSRUHE FORSCHZENT [DE]) 13 April 1994 (1994-04-13) column 1, line 1 - line 15 column 2, line 27 - line 29 column 2, line 39 - column 3, line 6 figure 1 -----	1-18
A	US 2007/273882 A1 (SMITH STEVIE HORTON [US]) 29 November 2007 (2007-11-29) paragraph [0034] paragraph [0036] - paragraph [0039] figure 4 -----	1-18

INTERNATIONAL SEARCH REPORT

Information on patent family members

International application No

PCT/EP2011/053990

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
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