Braid

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[54]	LUBRICA ANTIOXI	NTS CONTAINING AMINE DANTS	3,347,791 3,660,290 3,378,491	11/1967 5/1972 4/1968	Thompson et al	
[75]	Inventor:	Milton Braid, Westmont, N.J.	3,247,111	4/1966	Oberright et al 252/50	
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[22]	Filed:	Nov. 17, 1971	Attorney—Andrew L. Gaboriault et al.			
[21]	Appl. No.	199,732				
[52] [51]			[57]		ABSTRACT	
[58]		earch	phenylami	ne and su	ing a mixture of p,p'-dioctyldi- bstituted naphthylamines have im-	
[56]		References Cited	proved an	tioxidant _J	properties.	
	UNI	TED STATES PATENTS				
3,282	,840 11/19	66 Foster et al		10 C	laims, No Drawings	

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LUBRICANTS CONTAINING AMINE ANTIOXIDANTS

BACKGROUND OF THE INVENTION

1. Field of the Invention

This invention relates to lubricant compositions having improved antioxidant characteristics. More particularly, the invention relates to lubricants to which have been added a combination of additives which imparts antioxidant properties thereto.

2. Discussion of the Prior Art

Lubricating oils and greases are subject to oxidative deterioration at elevated temperatures or upon prolonged exposure in an ordinary atmosphere. Such deterioration is in many instances evidenced by an increase 15 in acidity and in viscosity, and when the deterioration is severe enough it can cause metal parts to corrode. Additionally, there is a loss of lubricating properties which may lead to complete failure of the part or parts in contact with the oil. Numerous additives have been 20 tried as antioxidants in lubricants, but many of them are only marginally effective except at the higher concentrations.

The use of amines as lubricant antioxidants is, of course, well known. As an example, U.S. Pat. No. 25 3,126,344 discloses the use of such amines as phenyl-α-naphthylamine, p-amino diphenylamine and p,p'-dioctyldiphenylamine as antioxidants in synthetic ester lubricating oils. Representative of other patents disclosing the use of single amines are U.S. Pat. No. 3,231,499 30 and U.S. Pat. No. 3,309,314.

In addition, it is known to use mixtures of amines as antioxidants for lubricants. U.S. Pat. No. 3,148,147 discloses for instance, the use of a mixture of phenyl-α-naphthylamine and diphenylamine. In U.S. Pat. No. 353,247,111 it is taught that a particularly outstanding combination involves a mixture of phemyl-α-naphthylamine and dioctyl diphenylamine. U.S. Pat. No. 3,309,318 teaches the same combination.

As will appear hereinafter, the combinations of the ⁴⁰ present invention are superior to the mixtures taught by U.S. Pat. Nos. 3,247,111 and 3,309,318.

SUMMARY OF THE INVENTION

The invention provides a lubricant composition comprising a major proportion of a lubricant and an antioxidant amount of a mixture comprising p,p'-dioctyl diphenylamine and a substituted naphthylamine.

DESCRIPTION OF SPECIFIC EMBODIMENTS

The substituted naphthylamine useful as one component of the amine mixture has the formula

where R is an alkyl group having from about 4 to about 20 carbon atoms or a nitro group, Ar is phenyl or naphthyl, R' is hydrogen or a phenylazo group, x is 0 to 2 and y is 0 to 1, their sum being at least 1.

Effective antioxidant mixtures of this invention will contain from about 0.001% to about 5% by weight of the p,p'-dioctyl diphenylamine and of a second amine of the above formula. Thus, the mixed amines will comprise from about 0.002% to about 10%, preferably from about 0.1% to about 5% by weight, of the lubri-

cant. In selecting appropriate concentrations within these ranges, it will generally be found that the best antioxidant activity will be realized from mixtures wherein the ratios of p,p'-dioctyl diphenylamine to second amine are from about 4:1 to about 1:4.

The lubricants which may be improved by the addition thereto of the above-mentioned amine mixtures are mineral oils, both paraffinic and naphthenic, synthetic oils, and greases made from these oils. The syn-10 thetic oils include polyolefin fluids, polyglycols, polyacetals, the siloxanes and the like. Especially useful are those synthetic esters which are becoming more and more popular, particularly in aviation. These include esters made from polycarboxylic acids and monohydric alcohols, such as those obtained by reacting methanol, ethanol, propanol, butanol, neopentyl alcohol and the like with pimelic acid, azelaic acid, adipic acid and the like, or from polyhydric alcohols and monocarboxylic acids. Of these preferred members are those made from trimethylol alkanes, where alkane includes ethane, propane and butane, pentaerythritol, di- and tripentaerythritol, or mixtures thereof and an aliphatic monocarboxylic acid containing from 1 to about 20 carbon atoms or a mixture of such acids. The most preferred are esters from the pentaerythritols and monocarboxylic acids having from 5 to 9 carbon atoms, or mixtures of these acids. The polyolefin fluids contemplated are those prepared from monoolefins having from about 4 to about 22 carbon atoms. Preferred are the trimers and tetramers of decene-1.

The following examples and evaluations will serve to specifically illustrate the method of making the substituted naphthylamines used in the practice of this invention and the use of mixtures of amines as lubricant antioxidants. It will be understood that they are illustrative only and are not intended to limit the scope of the invention in any way.

EXAMPLE 1

N-2,4-Dinitrophenyl-1-naphthylamine

A mixture of 20.3 g. of 1-chloro-2,4-dinitrobenzene, 14.3 g of 1-naphthylamine, 10 g. of sodium acetate and 100 ml of ethanol was heated at reflux for several hours. The hot reaction mixture was filtered and the N-2,4-dinitrophenyl-1-naphthylamine crystallized on cooling. Recrystallization from ethanol afforded orange crystalline solids melting at 191°-193° C.

EXAMPLE 2

1,1'-Dinaphthylamine

A mixture of 90 g. of 1-naphtylamine and 72.4 g. of 1-naphtylamine hydrochloride was heated at 225° C. for 22 hours. Unreacted naphthylamine was removed by distillation. The residue was extracted with benzene and 1,1'-dinaphthylamine was obtained by crystallization from the alumina column chromatographed benzene solution as white crystals melting at 115°-116° C.

EXAMPLE 3

N-(4-tert-Octylphenyl)-4-phenylazo-1-naphthylamine

To a solution of 82.5 g. of N-4-tert-octylphenyl-1-naphthylamine in 500 ml of 1,2-dimethoxyethane cooled to 0.5° there was added while stirring during 0.5 hour an aqueous solution of benzenediazonium chloride prepared from 23.3 g. of aniline, 17.3 g. of sodium nitrite and dilute hydrochloric acid.

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The reaction mixture was allowed to warm to room temperature after the addition was completed and was further diluted with water. The gray solids which separated were collected, treated with ammonia in methanol suspension, and recrystallized from ethanol. The 5 N-(4-tert-octylphenyl)-4-phenylazo-1-naphthylamine was obtained as a red crystalline solid melting at 135°-137° C.

EVALUATION OF PRODUCTS OXIDATION TEST

The compounds produced in accordance with this invention were blended into a synthetic ester oil lubricant (made by reacting pentaerythritol with an equimolar mixture of C₅ and C₉ monocarboxylic acids) and 15 tested in an oxidation test in accordance with the following procedure.

A sample of the test composition was heated and air at the rate of about 5 liters per hour was passed through for a period of about 24 hours at 450° F. Present in the test sample were specimens of iron, copper, aluminum, and lead. It should be noted that the metals are typical metals of engine or machine construction, and they also provide some catalysis for the oxidation of organic materials. The results are tabulated in the following ta- 25 ble. The change in kinematic viscosity, measured in centistokes at 100° F. (% KV change) and the change in the neutralization number (NN change) are measures of the effectiveness of an antioxidant in lubricants.

TABLE

Additive	Weight percent	ΔΝΝ	ΔKV, percent	Pb loss, mg.	Sludge
0.4.31.41.70.4374		8. 25	585	13.7	Trace.
2,4-dinitro PAN ¹ VAN 81 ²	$\{1.0\}$	2.10	33	14.3	Nil.
Do	1.0 }	3. 55	85	10.5	Nil.
t-Octyl PAN` VAN 81	1.0 }	2.09	64	2	Nil.
1,1'-DNA 3 VAN 81	1.0 }	2, 25	43	32.3	Nil.
Do	0.5	2.57	43	15	Nil.
Phenylazo-t-octyl:	. 1.0)				
PAN VAN 81	$\{1,0\}$	2.0	53. 0	0	Nil.
Do{	2.0 } 1.0 }	1.9	55	0. 5	Nil.
Phenylazo-t-octyl:	2.0,				
PAN (Exam-	1.0	3.9	117	1.7	Nil.
ple 3). VAN 81	1,0	6, 64	390	2, 7	Trace.
,	1.0	3. 15	51	44.0	Heavy.
1,1'-DNA	0.5	3. 17	67	41.3	Do.
PAN	1	3. 54	92	4.1	Light.
PAN VAN 81	$\{1.0, 0, 1.0, 1.0, 1.0, 1.0, 1.0, 1.0, 1.$	5.35	98	13.6	Trace.

1 PAN=phenyl-1-naphthylamine. 2 VAN 81=p,p'-dioctyl diphenylamine. 3 1,1'-DNA=1,1'-dinaphthylamine.

Other substituted naphthylamines useful in the mixtures of this invention are the following.

R-Ar-NH

phenyl naphthyl phenyl naphthyl phenyl naphthyl phenyl naphthyl phenyl naphthyl

Although the present invention has been described with certain specific embodiments, it is to be understood that modefications and variations may be resorted to without departing from the spirit and scope of this invention as those skilled in the art will readily understand. Such modifications and variations are considered to be within the purview and scope of the appended claims.

I claim:

1. A lubricant composition comprising a major proportion of a lubricant and an antioxidant amount of a mixture comprising p, p'-dioctyldiphenylamine and a substituted naphthylamine of the formula

wherein R is NO2 or an alkyl of from 4 to 20 carbon 35 atoms, Ar is phenyl or naphthyl, R' is hydrogen or phenylazo, R' being hydrogen when R is nitro and x is 1 or 2.

- 2. The composition of claim 1 wherein the p,p'dioctyl diphenylamine is present therein to the extent 40 of from about 0.001% to about 5% by weight.
 - 3. The composition of claim 1 wherein the substituted naphthylamine is present therein to the extent of from about 0.001% to about 5% by weight.
- 4. The composition of claim 1 wherein the mixture 45 comprises from about 0.002% to about 10% by weight thereof.
 - 5. The composition of claim 1 wherein the lubricant is a lubricating oil or grease.
- 6. The composition of claim 5 wherein the lubricat-50 ing oil is a synthetic ester lubricating oil.
 - 7. The composition of claim 1 wherein the substituted naphthylamine is N-2,4-dinitrophenyl-1naphthylamine.
- 8. The composition of claim 1 wherein the substi-55 tuted naphthylamine N-tert-octylphenyl-1is naphthylamine.
 - 9. The composition of claim 1 wherein the substituted naphthylamine is N-(4-tert-octylphenyl)-4phenylazo-1-naphthylamine.
 - 10. The composition of claim 1 wherein the substituted naphthylamine is 1,1'-dinaphthylamine.