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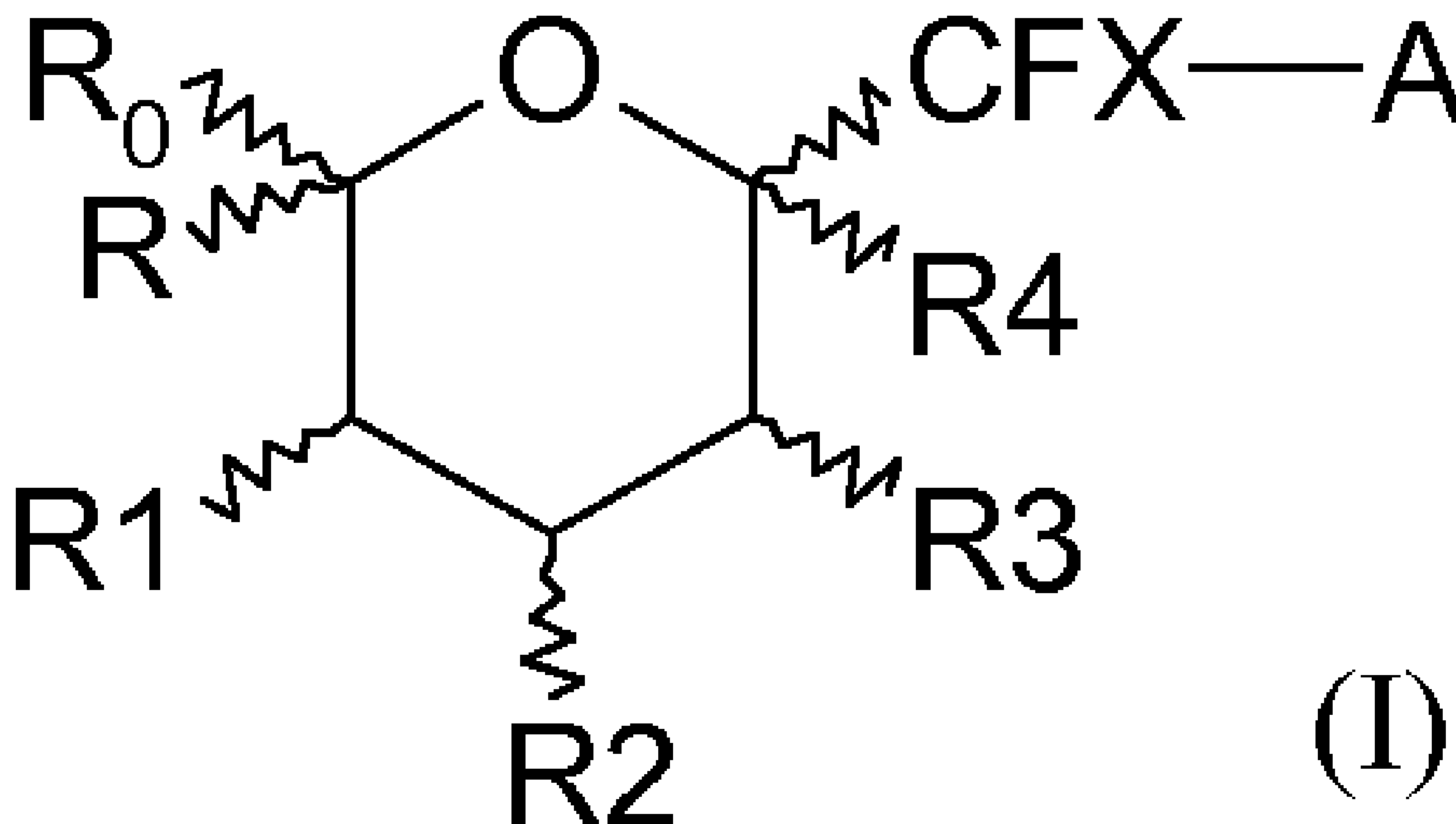
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(54) Titre : COMPOSES C-ARYL GLYCOSIDES POUR LE TRAITEMENT DU DIABETE ET DE L'OBESITE

(54) Title: C-ARYL GLYCOSIDE COMPOUNDS FOR THE TREATMENT OF DIABETES AND OBESITY



(57) Abrégé/Abstract:

This invention relates to a compound of generic formula (I): (I) as well as a pharmaceutically acceptable salt thereof, a tautomer, optical isomer or a mixture of optical isomers in any proportion, in particular a mixture of enantiomers, and particularly a racemate mixture, in particular for use thereof as a drug, notably in the treatment of diabetes.

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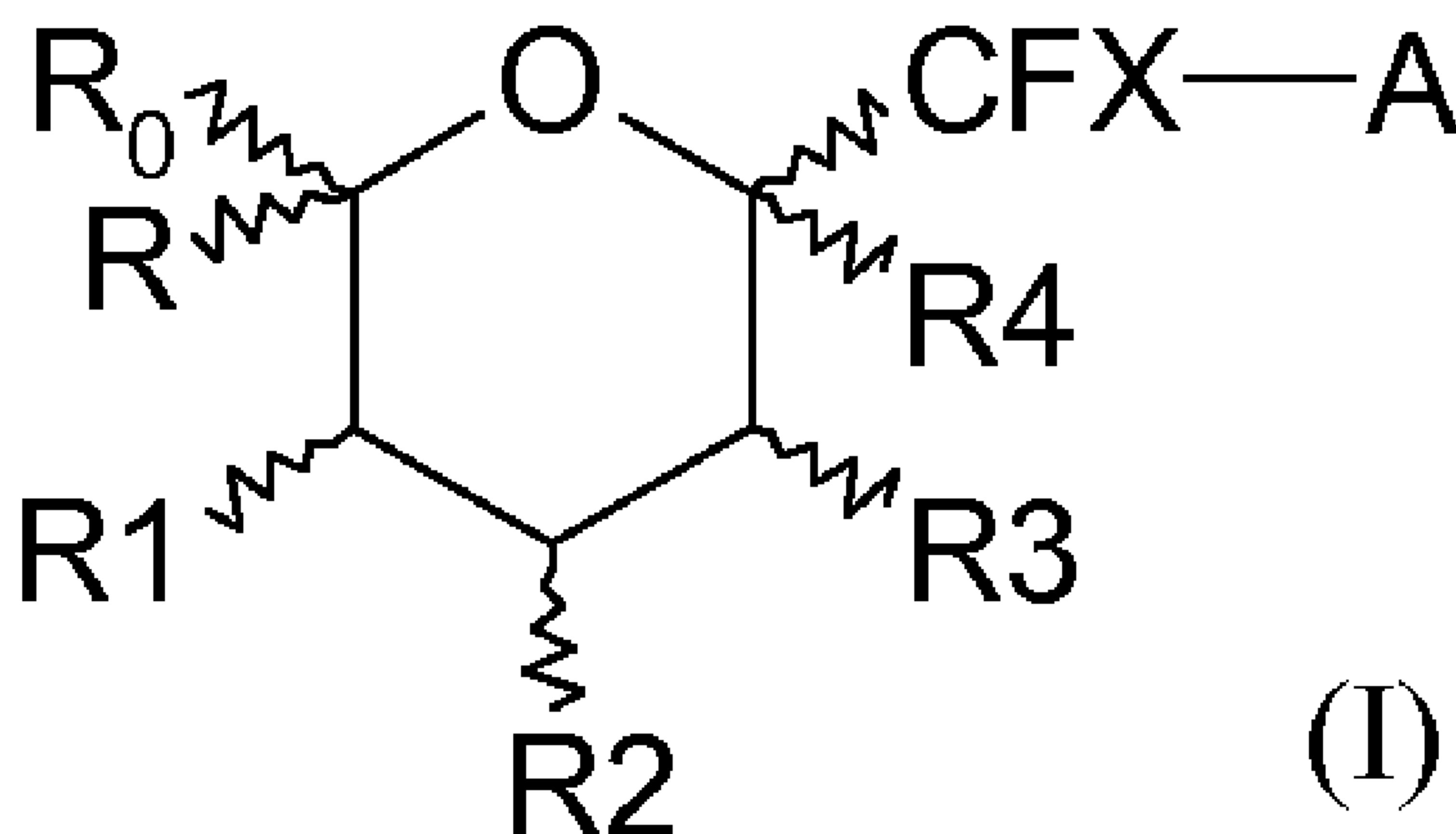
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(54) Title: C-ARYL GLYCOSIDE COMPOUNDS FOR THE TREATMENT OF DIABETES AND OBESITY



(57) Abstract: This invention relates to a compound of generic formula (I); (I) as well as a pharmaceutically acceptable salt thereof, a tautomer, optical isomer or a mixture of optical isomers in any proportion, in particular a mixture of enantiomers, and particularly a racemate mixture, in particular for use thereof as a drug, notably in the treatment of diabetes.

C-ARYL GLYCOSIDE COMPOUNDS FOR THE TREATMENT OF DIABETES AND OBESITY

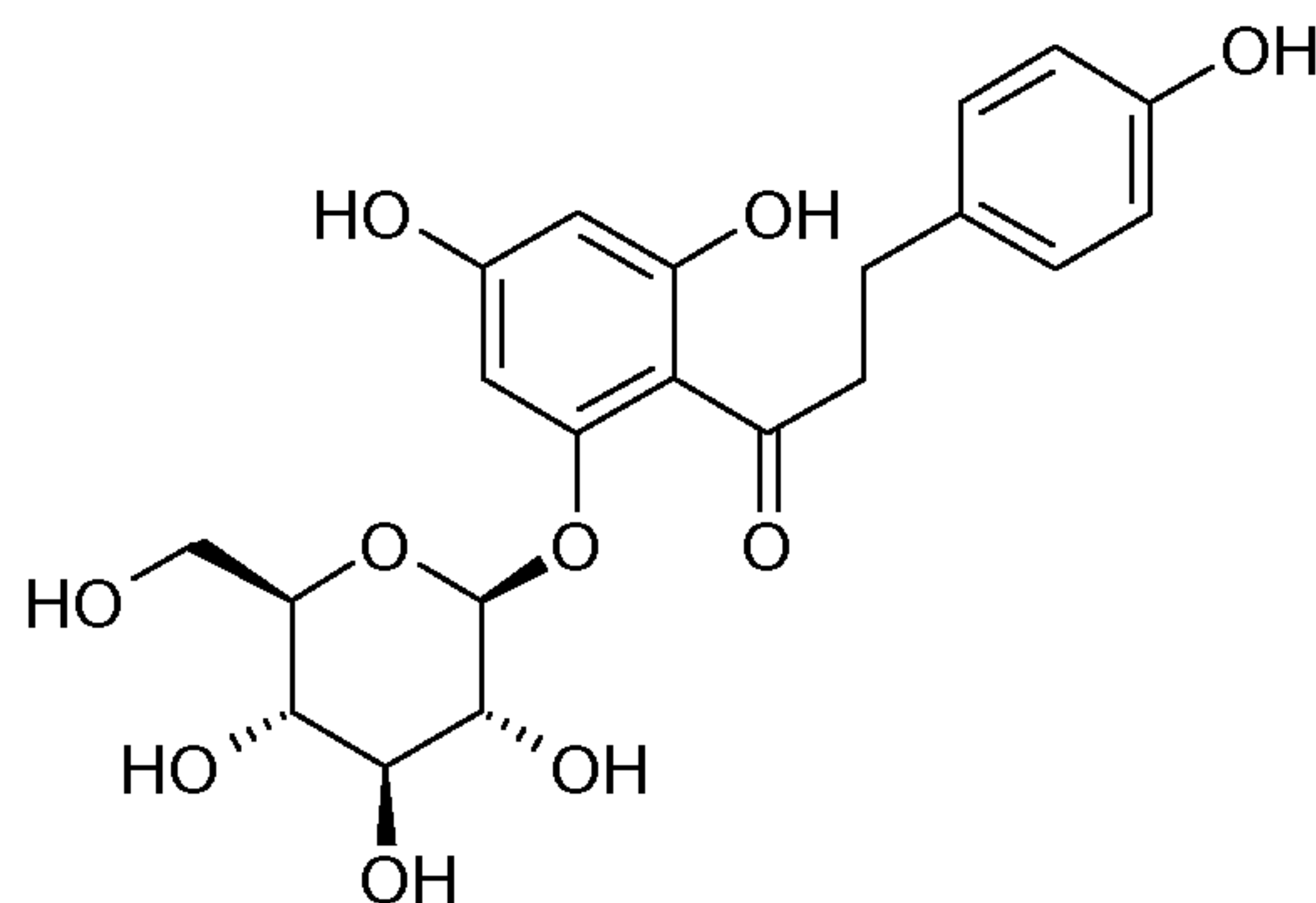
This invention relates to a family of fluorinated C-aryl glycoside compounds, the
5 process for their preparation, as well as the application of same in the pharmaceutical
and cosmetics fields, in particular for the treatment of diabetes and obesity.

Sugars and the derivatives thereof constitute one of the most common classes of
compounds in nature. Based on their chemical structures, they exhibit various physico-
chemical properties and can play a key role in a wide variety of biological processes.

10 In recent years, there has been a growing interest in discovering new glycosides
having advantageous properties in terms of improved efficacy, selectivity and stability.

Found among these compounds, in particular, are aryl glycosides or phenol
glycosides having applications in the field of cosmetics or in the treatment or prevention
of diseases such as diabetes, obesity, cancer, inflammatory diseases, auto-immune
15 diseases, infections, thromboses, and with regard to numerous other therapeutic fields.
By their biological properties and their structure, these compounds interest numerous
research teams.

Phlorizin may be cited in particular, as a molecule known for its inhibiting
activity with regard to sodium-dependent glucose co-transporters (SGLT) (Journal of
20 Clinical Investigation, vol. 79, p. 1510, (1987); *ibid.*, vol. 80, p. 1037 (1987); *ibid.*, vol.
87, p. 561 (1991); J. of Med. Chem., vol. 42, p. 5311 (1999); British Journal of
Pharmacology, vol. 132, p. 578, (2001)).



Phlorizin

Inhibitors of sodium-dependent glucose co-transporters (SGLT), found in particular in the intestines and kidney, are potentially usable for treating diabetes, and more specifically type-II diabetes, but also for hyperglycemia, hyperinsulinemia, obesity, hypertriglyceridemia, syndrome X (also known by the name of metabolic syndrome, J. of Clin. Endocrinol. Metabol., 82, 727-734 (1997)), diabetes-related complications or else atherosclerosis. As a matter of fact, it is known that hyperglycemia participates in the onset and evolution of diabetes and leads to a reduction in the secretion of insulin and a reduction in insulin sensitivity, which results in an increase in the glucose level, thereby exacerbating diabetes. The treatment of hyperglycemia can thus be considered as a mean to treat diabetes.

Such being the case, one of the methods for treating hyperglycemia is to promote the excretion of excess of glucose directly into the urine, e.g., by inhibiting the sodium-dependent glucose co-transporter in the proximal tubules of the kidneys, the effect of which is to inhibit the re-absorption of glucose and to thereby promote the excretion thereof into the urine, leading thus to a reduction in the blood-sugar level.

At present, a large number of drugs exist, which can be used for treating diabetes, such as biguanides, sulfonylureas, insulin resistance-improving agents, and inhibitors of α -glycosidases. However, these compounds have numerous side effects, thereby increasing the need for new drugs.

Therefore, the invention relates to C-aryl glycoside compounds, which are useful, in particular, for the treatment of diabetes.

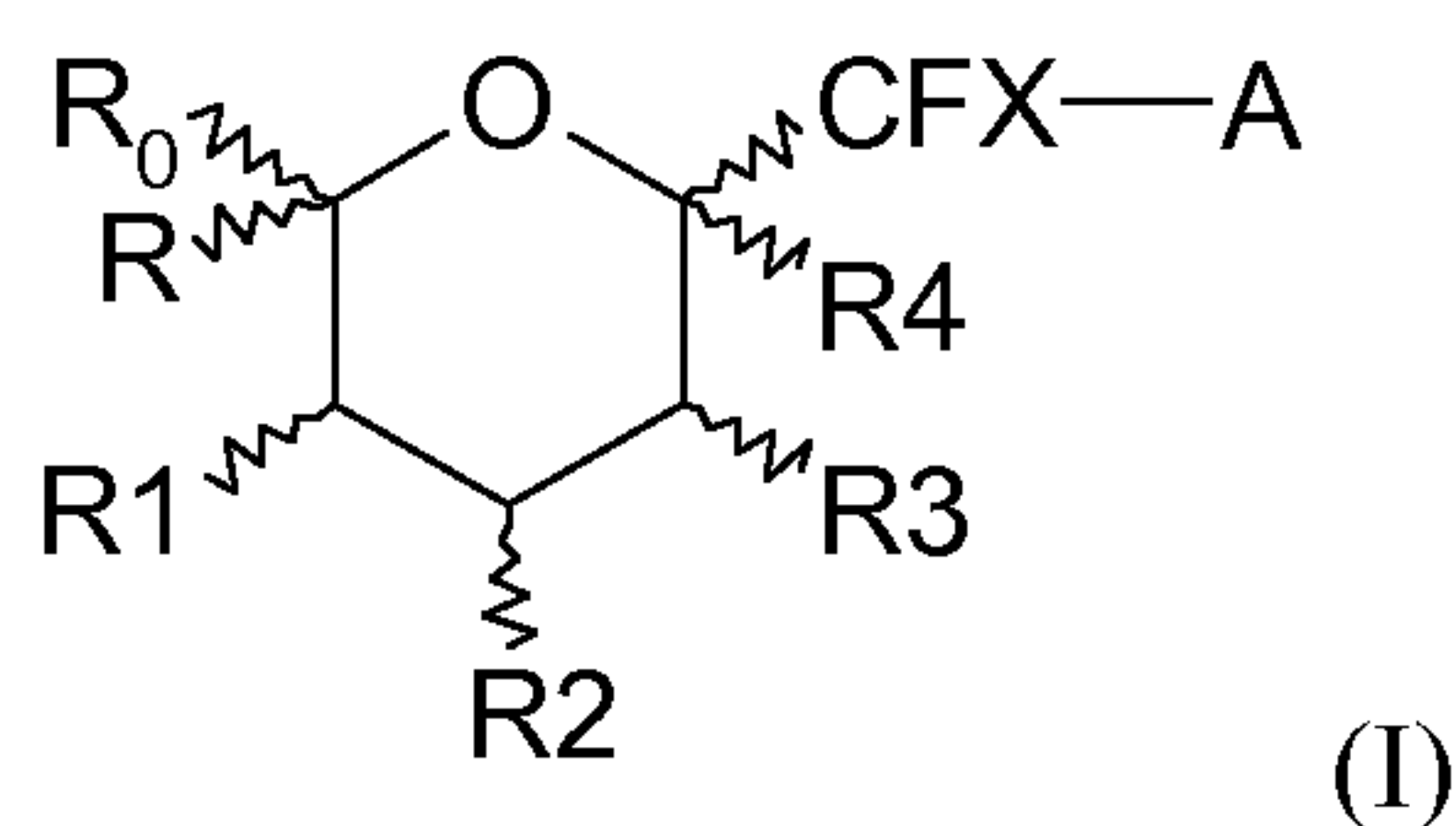
These compounds are analogues of O-aryl glycosides or phenol glycosides, wherein the anomeric oxygen is replaced by a carbon atom, carrying one or two fluorine atom(s), and have the distinctive feature of being stable analogues of O-aryl glycosides, which are stable when confronted with enzymatic degradation processes, in particular via glycosidase-type enzymes. Moreover, the mono or difluorinated carbon is a better mimic of oxygen than a CH_2 group.

Thus, contrary to the CH_2 -glycosides, the replacement of the anomeric oxygen by a CF_2 or a CFH group, in particular minimizes the electronic effects due to the substitution, while at the same time resulting in stable compounds, resistant when confronted with enzymatic degradations, and in particular via glycosidase-type enzymes, but also resistant to hydrolysis condition in acidic or basic media.

C-fluorinated-glycoside compounds substituted at the anomeric position by an alkyl chain possibly substituted are described in the patent applications WO 2004/014 928 and WO 2007/128 899 but no biological activity of these compounds with regard to inhibiting SGLT is demonstrated in these applications. Moreover, no C-aryl glycoside
 5 compound is described, such a compound being not obtainable by a process such as described in these patent applications.

The inventors have thus developed new synthetic approaches enabling access to C-aryl glycoside compounds, compounds useful as SGLT inhibitors, in particular for the treatment of diabetes and obesity.

10 Therefore, the object of the present invention is a compound having the generic formula (I):



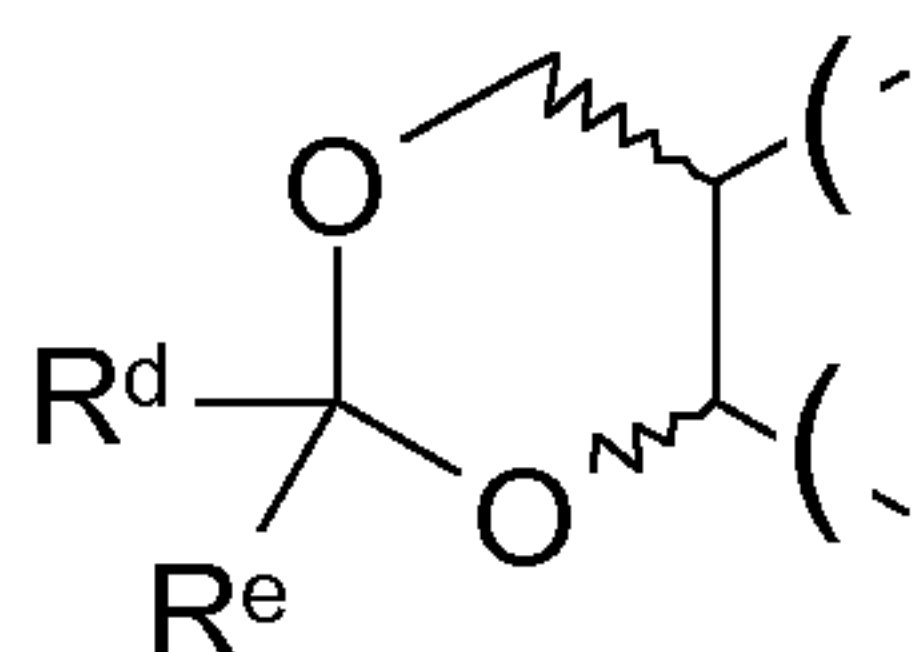
15 or a pharmaceutically acceptable salt thereof, a tautomer, an isomer or a mixture of isomers in any proportion, in particular a mixture of enantiomers, and particularly a racemate mixture,

wherein:

- X represents a hydrogen or a fluorine atom;
- 20 - R represents a hydrogen or a fluorine atom or a CH₃, CH₂F, CH₂OH, CH₂OSiR^aR^bR^c, CH₂OR¹¹, CH₂OCOR¹¹, CH₂OCO₂R¹¹, CH₂OCONR¹²R¹³, CH₂OP(O)(OR¹⁴)₂ or CH₂OSO₃R¹⁴ group;
- R1 and R2 represent, independently from one another, a fluorine atom or an OH, OSiR^aR^bR^c, OR¹¹, OCOR¹¹, OCO₂R¹¹ or OCONR¹²R¹³ group;
- 25 - R3 represents a hydrogen or fluorine atom or an OH, OSiR^aR^bR^c, OR¹¹, OCOR¹¹, OCO₂R¹¹, OCONR¹²R¹³, NR¹²R¹³ or NR¹²COR¹¹ group;
- R4 represents a hydrogen atom, an halogen atom or an OH, OSiR^aR^bR^c, OR¹¹, OCOR¹¹, OCO₂R¹¹, OCONR¹²R¹³, NR¹²R¹³, (C₁-C₆)-alkyl or (C₂-C₆)-alkenyl group;

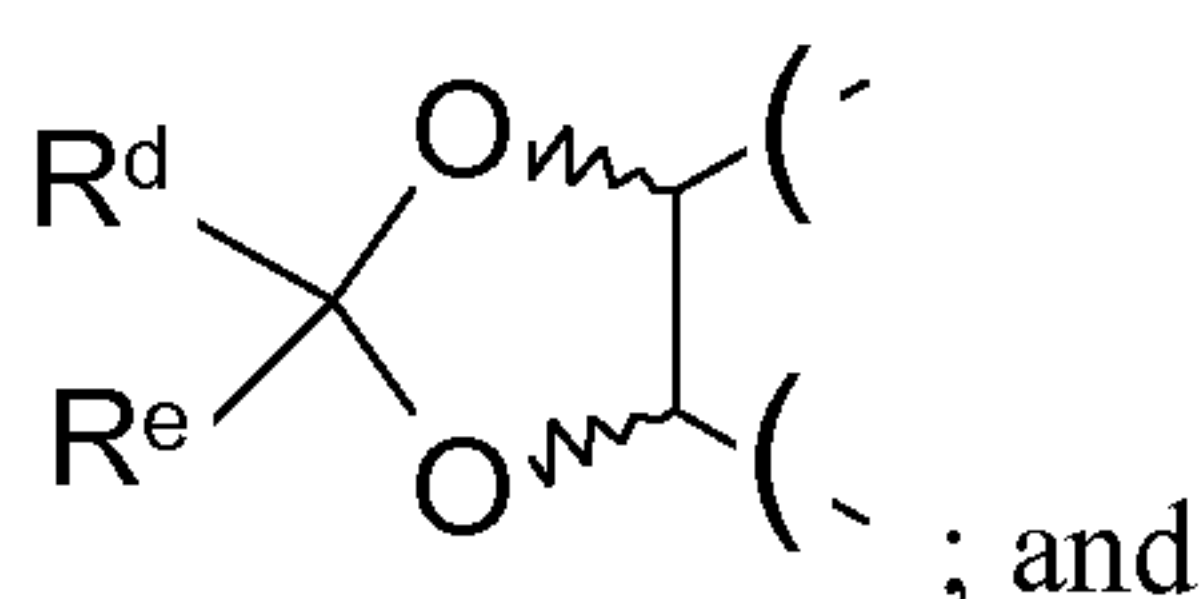
- R_0 represents a hydrogen or an halogen atom or an OH, $OSiR^aR^bR^c$, OR^{11} , $OCOR^{11}$, OCO_2R^{11} , $CONR^{12}R^{13}$, $OP(O)(OR^{14})_2$ or OSO_3R^{14} group;

or R and R1, together with the carbon atoms carrying them, form a cyclic acetal having the following formula:



5

and/or (R_0 and R1), (R1 and R2), (R2 and R3), and/or (R3 and R4), together with the carbon atoms carrying them, form a cyclic acetal having the following formula:



- A represents an aryl, heteroaryl or aryl-(C₁-C₆)-alkyl-aryl group, possibly substituted by one or more groups chosen among an halogen atom, a CN, SO₂, $SiR^aR^bR^c$, (C₁-C₆)-alkyl, (C₂-C₆)-alkenyl, (C₂-C₆)-alkynyl, (C₃-C₇)-cycloalkyl, 5 to 7 ring-membered heterocycloalkyl, aryl, heteroaryl, aryl-(C₁-C₆)-alkyl, heteroaryl-(C₁-C₆)-alkyl, (C₁-C₆)-alkyl-aryl, (C₁-C₆)-alkyl-heteroaryl, OR^{11} , COR^{11} , $OCOR^{11}$, CO_2R^{11} , $NR^{12}R^{13}$, $NR^{12}COR^{11}$, $CONR^{12}R^{13}$, SR^{11} , SO_2R^{11} , CSR^{11} and OSO_3R^{11} group,

the whole being possibly substituted by one or more groups chosen among an halogen atom, an OH, (C₁-C₆)-alkyl, (C₁-C₆)-alkoxy, COOH and CHO group;

with:

- R^{11} representing a (C₁-C₆)-alkyl, (C₂-C₆)-alkenyl, (C₂-C₆)-alkynyl, (C₃-C₇)-cycloalkyl, 5 to 7 ring-membered heterocycloalkyl, aryl, aryl-(C₁-C₆)-alkyl or (C₁-C₆)-alkyl-aryl group, this group being possibly substituted by one or more groups chosen among an halogen atom, an OH, COOH and CHO group;

- R^{12} and R^{13} representing, independently from one another, a hydrogen atom or a (C₁-C₆)-alkyl or aryl-(C₁-C₆)-alkyl group;

- R^{14} representing a hydrogen atom or a (C₁-C₆)-alkyl group;

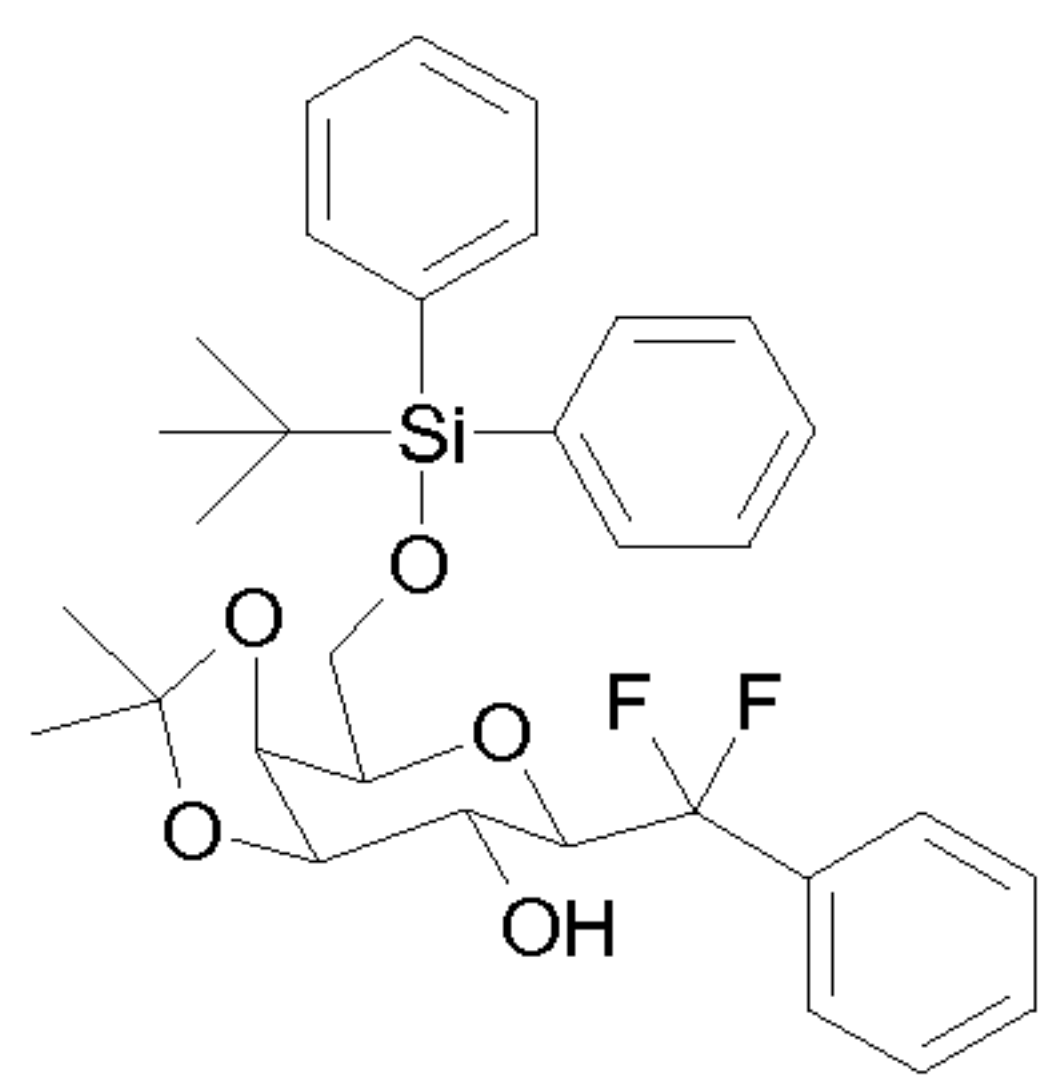
- R^a , R^b and R^c representing, independently from one another, a (C₁-C₆)-alkyl, aryl or aryl-(C₁-C₆)-alkyl group; and

- R^d and R^e representing, independently from one another, a hydrogen atom or a (C₁-C₆)-alkyl group;

25

with the proviso that when R_0 is different from a hydrogen atom, then R_4 represents a hydrogen atom, and

with the proviso that the compound of formula (I) is not the following compound:



5 The silylated compound cited above is described in Kurissery et al. (*Org. Lett.* 2007, 9, 8, 1441-1444) as synthesis intermediate. No biological activity of this compound is described or suggested in this publication.

In this invention, “pharmaceutically acceptable” is understood to mean what is useful in the preparation of a pharmaceutical composition which is generally safe, non-toxic and neither biologically nor otherwise undesirable and which is acceptable for
10 veterinary as well as human pharmaceutical use.

In this invention, “pharmaceutically acceptable salts” of a compound, is understood to designate salts which are pharmaceutically acceptable, as defined herein, and which possess the desired pharmacological activity of the parent compound. Such
15 salts include:

(1) hydrates and solvates,

(2) acid addition salts formed with inorganic acids such as hydrochloric acid, bromhydric acid, sulphuric acid, nitric acid, phosphoric acid or the like; or formed with organic acids such as acetic acid, benzenesulfonic acid, benzoic acid, camphorsulfonic
20 acid, citric acid, ethanesulfonic acid, fumaric acid, glucoheptonic acid, gluconic acid, glutamic acid, glycolic acid, hydroxynaphtoic acid, 2-hydroxyethanesulfonic acid, lactic acid, maleic acid, malic acid, mandelic acid, methanesulfonic acid, muconic acid, 2-naphtalenesulfonic acid, propionic acid, salicylic acid, succinic acid, dibenzoyl-L-tartaric acid, tartaric acid, p-toluenesulfonic acid, trimethylacetic acid, trifluoroacetic
25 acid and the like; and

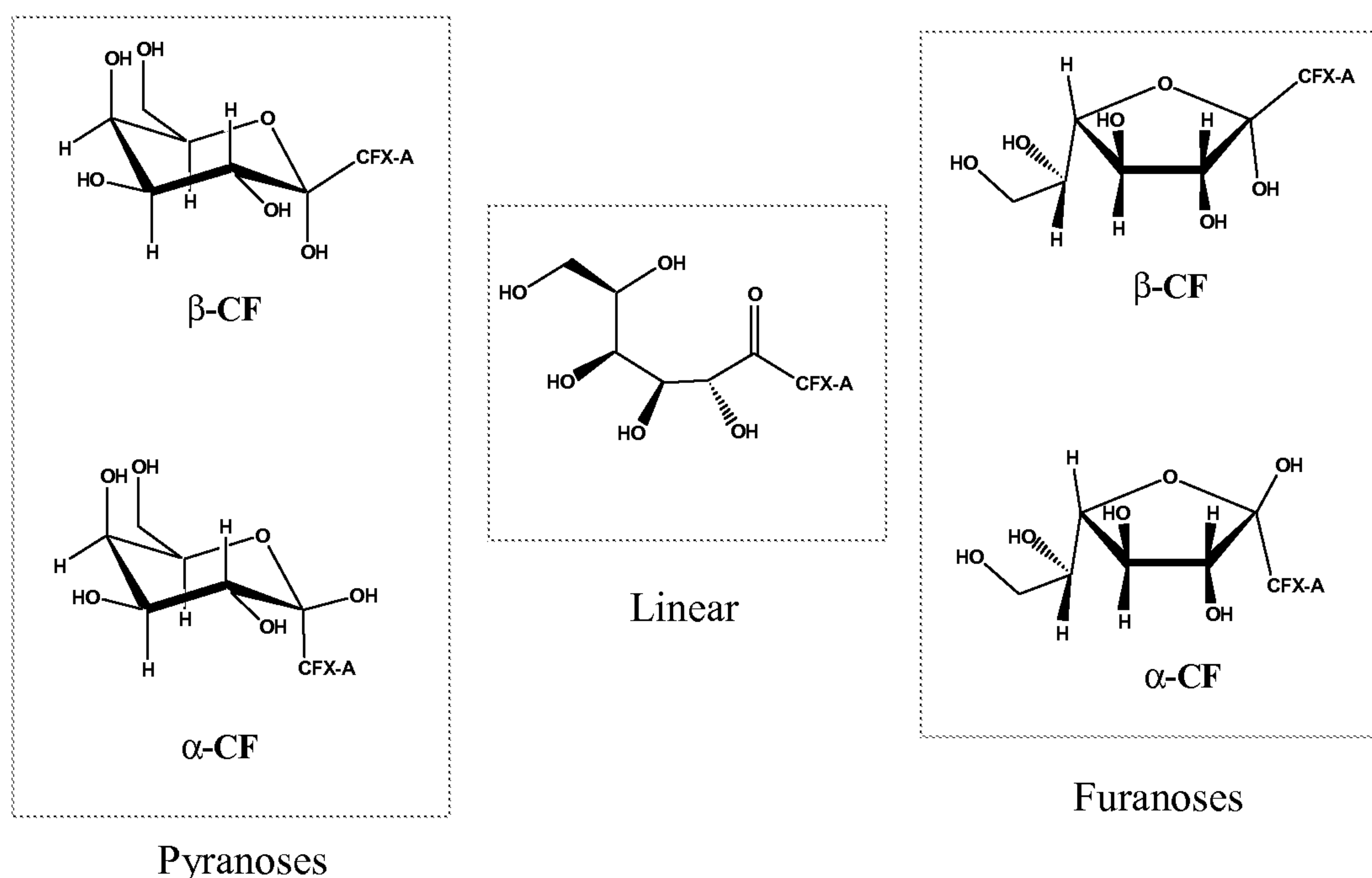
(3) salts formed when an acid proton present in the parent compound is either replaced by a metal ion, e.g., an alkali metal ion (e.g., Na^+ , K^+ or Li^+), an alkaline-earth metal ion (like Ca^{2+} or Mg^{2+}) or an aluminium ion; or coordinates with an organic or

inorganic base. Acceptable organic bases include diethanolamine, ethanolamine, N-methylglucamine, triethanolamine, tromethamine and the like. Acceptable inorganic bases include aluminium hydroxide, calcium hydroxide, potassium hydroxide, sodium carbonate and sodium hydroxide.

5 In this invention, "tautomer" is understood to designate the various tautomer forms that the sugar of compound (I) may assume, namely a pyranose (6-membered ring), furanose (5-membered ring) or linear (open form) form, and also the various tautomer forms that could be observed with a ketone moiety, when it is present on the molecule, such as a cyclisation between an hydroxyle group and the ketone moiety.

10 However, the compounds of the invention can assume various tautomer forms only when the radical R4 represents an OH group, R1 having also to represent an OH group in order that the compounds of the invention can be in the furanose form.

Thus, for example, in the galactose series, the compounds of the invention might appear under the following various forms:



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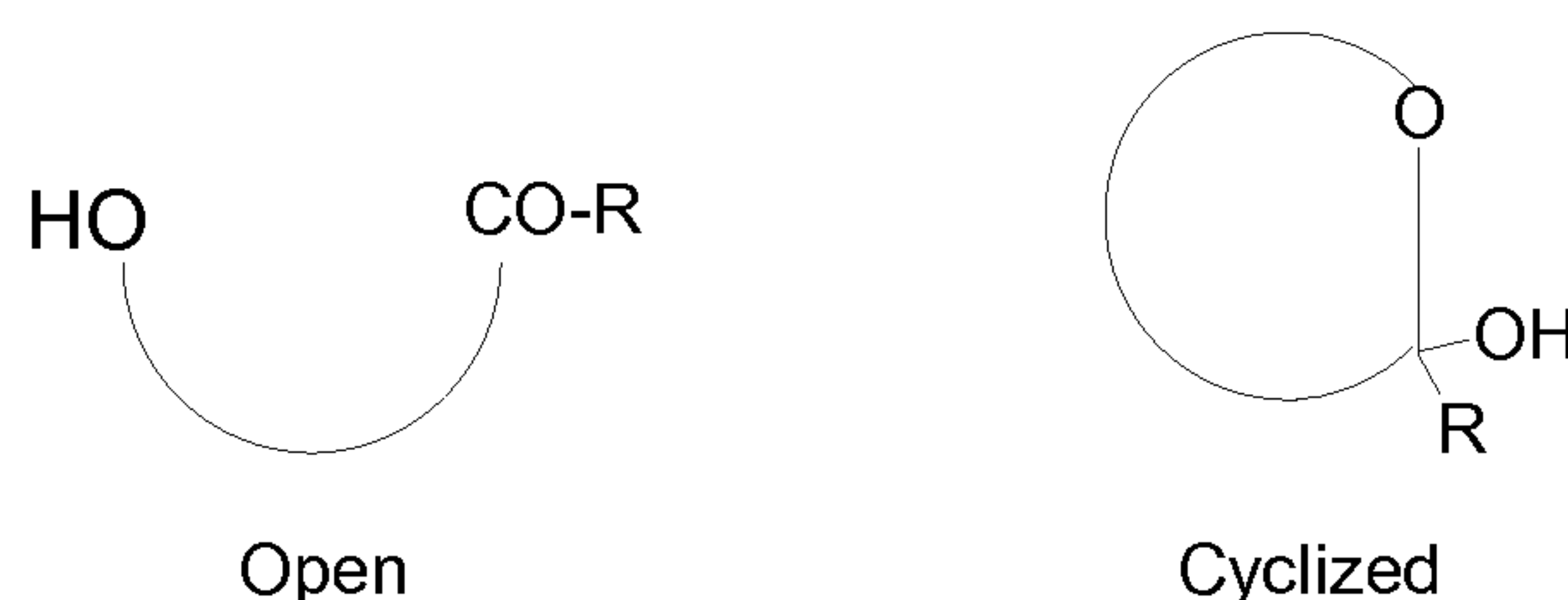
The anomeric carbon can thus appear in two different configurations in the closed pyranose and furanose forms.

20 The compounds of the invention can thus assume different tautomer forms which can be present in solution in equilibrium, with optionally a major tautomer form

relatively to the other(s) tautomer form(s), or the compounds of the invention can assume only one tautomer form, such as only a furanose form, in some cases.

In this last case where the sugar assumes only one tautomer form, it is thus possible to block the configuration of the sugar to this tautomer form when $R_4 = OH$ is transformed, notably by substitution of the OH group or conversion in a hydrogen or halogen atom.

In the case of the presence of OH and C=O functionalities in the same molecule, the following tautomer forms (open and cyclized) can be observed:



In this invention, “isomers,” within the meaning of this invention, is understood to designate diastereoisomers or enantiomers. These are therefore optical isomers also referred to as “stereoisomers”. Stereoisomers which are not mirror images of one another are thus designated as “diastereoisomers,” and stereoisomers which are non-superimposable mirror images are designated as “enantiomers”.

Notably, the sugar moiety of the compounds of the invention can belong to the D or L series, and preferably to the D series.

A carbon atom bound to four non-identical substituents is called a “chiral centre”.

An equimolar mixture of two enantiomers is called a racemate mixture.

Within the meaning of this invention, “halogen” is understood to mean an atom of fluorine, bromine, chlorine or iodine. Advantageously, this is an atom of fluorine, bromine or chlorine.

Within the meaning of this invention, “(C₁-C₆)-alkyl” group is understood to mean a saturated, linear or branched hydrocarbon chain comprising from 1 to 6 carbon atoms, in particular the methyl, ethyl, n-propyl, isopropyl, n-butyl, iso-butyl, sec-butyl, tert-butyl, n-pentyl, n-hexyl groups.

Within the meaning of this invention, “(C₁-C₆)-alkoxy” group is understood to mean a (C₁-C₆)-alkyl group as defined above, which is bound to the molecule by means

of an oxygen atom. It can be, in particular, a methoxy, ethoxy, n-propoxy, isopropoxy, n-butoxy, iso-butoxy, sec-butoxy, tert-butoxy, n-pentoxy or n-hexoxy group.

Within the meaning of this invention, “(C₂-C₆)-alkenyl” group is understood to mean a linear or branched hydrocarbon chain comprising at least one double bond and comprising from 2 to 6 carbon atoms, e.g., such as an ethenyl (vinyl) or propenyl group.

Within the meaning of the invention, “(C₂-C₆)-alkynyl” group is understood to mean a linear or branched hydrocarbon chain comprising at least one triple bond and comprising from 2 to 6 carbon atoms, e.g., such as an ethynyl or propynyl group.

Within the meaning of this invention, “(C₃-C₇)-cycloalkyl” group is understood to mean a saturated hydrocarbon ring comprising from 3 to 7, advantageously from 5 to 7, carbon atoms, in particular the cyclohexyl, cyclopentyl or cycloheptyl group.

Within the meaning of this invention, “heterocycloalkyl” group is understood to mean a saturated hydrocarbon ring having 5 to 7 members and containing one or more, advantageously one or two, heteroatoms, e.g., such as sulphur, nitrogen or oxygen atoms, e.g., such as the tetrahydrofuranyl, piperidinyl, pyrrolidinyl, tetrahydropyranyl, 1,3-dioxolanyl group.

Within the meaning of this invention, “aryl” group is understood to mean an aromatic group preferably comprising from 5 to 10 carbon atoms and including one or more fused rings, e.g., such as a phenyl or naphthyl group. This is advantageously phenyl.

Within the meaning of the invention, “heteroaryl” group is understood to mean any aryl group as defined above wherein one or more carbon atoms have been replaced by one or more heteroatoms, advantageously 1 to 4, and even more advantageously 1 to 2, e.g., such as sulphur, nitrogen or oxygen atoms. Examples of heteroaryl groups are the furyl, thiophenyl, pyrrolyl, pyridyl, pyrimidyl, pyrazolyl, imidazolyl, tetrazolyl or else indyl groups.

Within the meaning of this invention, “aryl-(C₁-C₆)-alkyl” group is understood to mean any aryl group as defined above, which is bound to the molecule by means of a (C₁-C₆)-alkyl group as defined above. In particular, a group such as this can be a benzyl group.

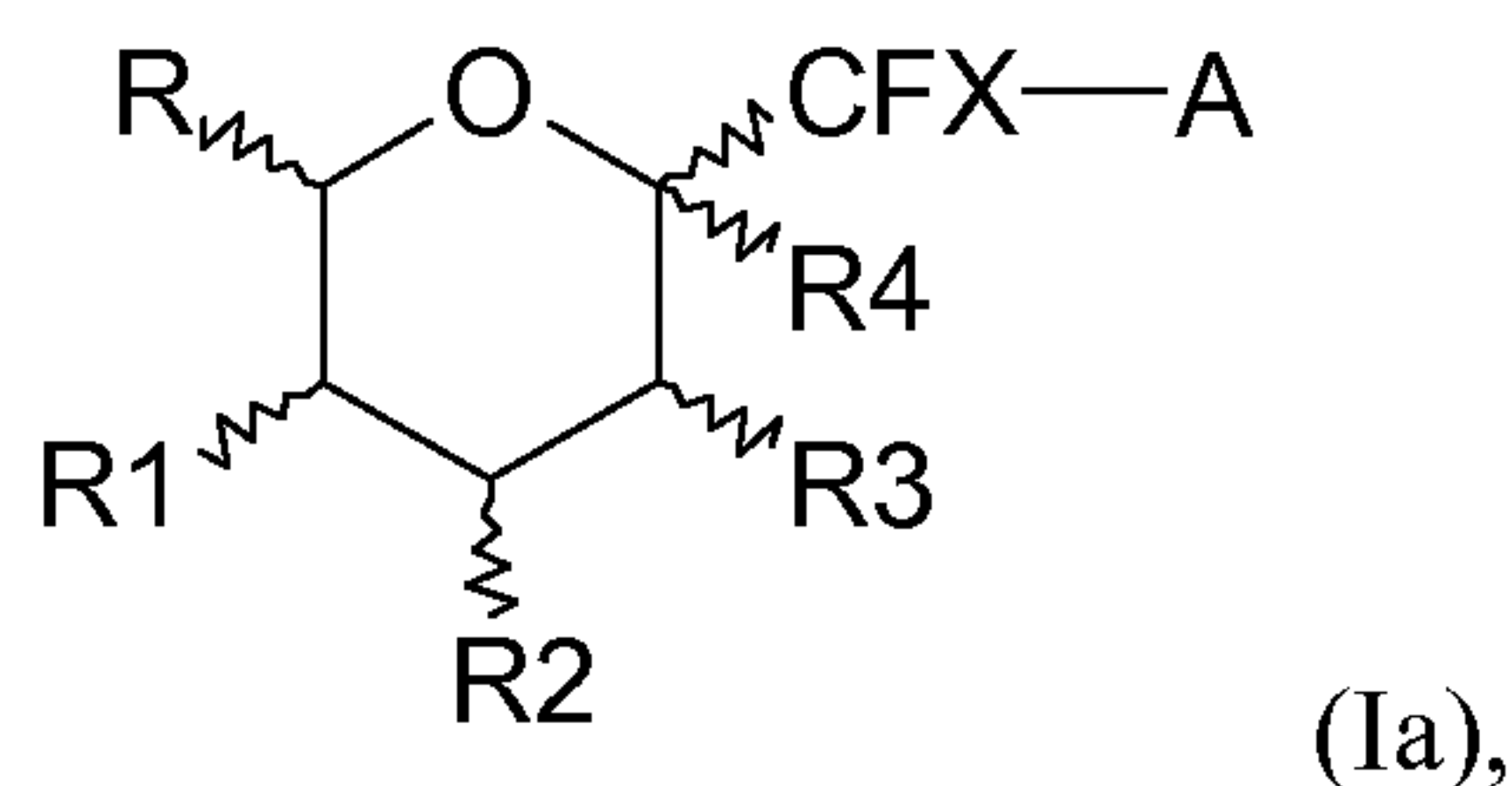
Within the meaning of this invention, “heteroaryl-(C₁-C₆)-alkyl” group is understood to mean a heteroaryl group as defined above, which is bound to the molecule by means of a (C₁-C₆)-alkyl group as defined above.

Within the meaning of this invention, “(C₁-C₆)-alkyl-aryl” group is understood to mean a (C₁-C₆)-alkyl group as defined above, which is bound to the molecule by means of an aryl group as defined above. In particular, a group such as this can be a methylphenyl group.

5 Within the meaning of this invention, “(C₁-C₆)-alkyl-heteroaryl” group is understood to mean a (C₁-C₆)-alkyl group as defined above, which is bound to the molecule by means of a heteroaryl group as defined above.

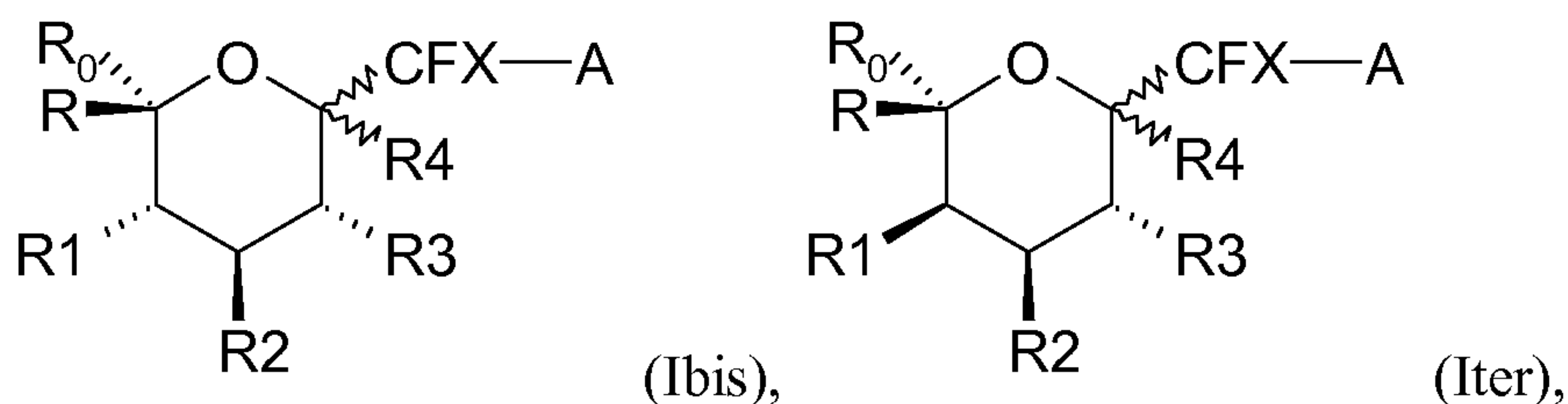
10 Within the meaning of this invention, “aryl-(C₁-C₆)-alkyl-aryl” group is understood to mean an aryl-(C₁-C₆)-alkyl group as defined above, which is bound to the molecule by means of an aryl group as defined above. In particular, such a group can be a benzyl-phenyl group.

According to a preferred embodiment, R₀ represents a hydrogen atom or an OH group and preferably a hydrogen atom. In this last case, when R₀ = H, the compounds of the invention respond to the following formula (Ia):



with R, R₁, R₂, R₃, R₄, X and A as defined above.

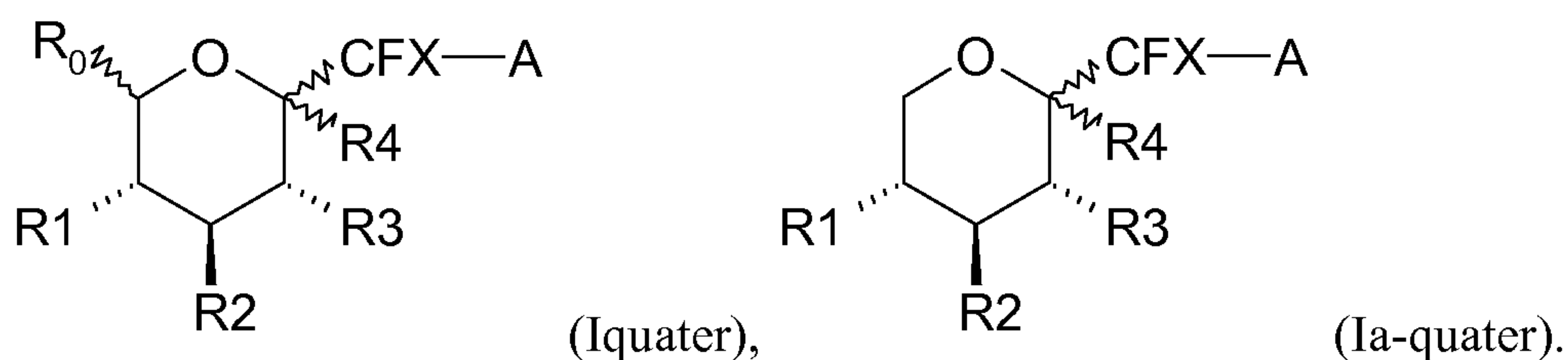
The compounds of the invention are advantageously based on the following formulas (Ibis) or (Iter):



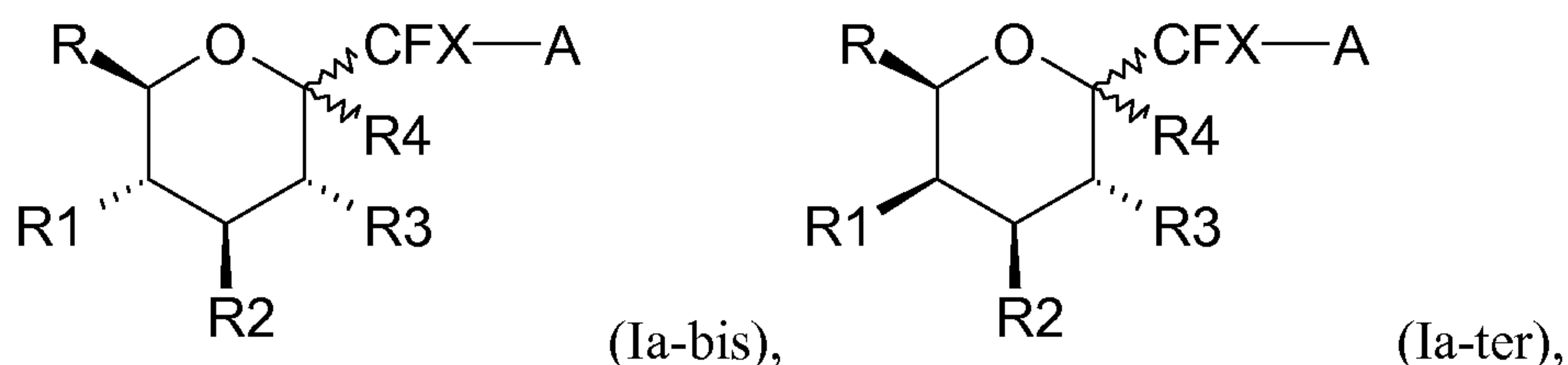
with R, R₁, R₂, R₃, R₄, R₀, X and A as defined above.

The compounds of the invention are advantageously based on the formula (Ibis).

Moreover, the compounds of the invention can also be based on the following formulas (Iquater) and (Ia-quarter), when R = H:



The compounds of the invention are more advantageously based on the following formulas (Ia-bis) or (Ia-ter):



5 with R, R1, R2, R3, R4, X and A as defined above.

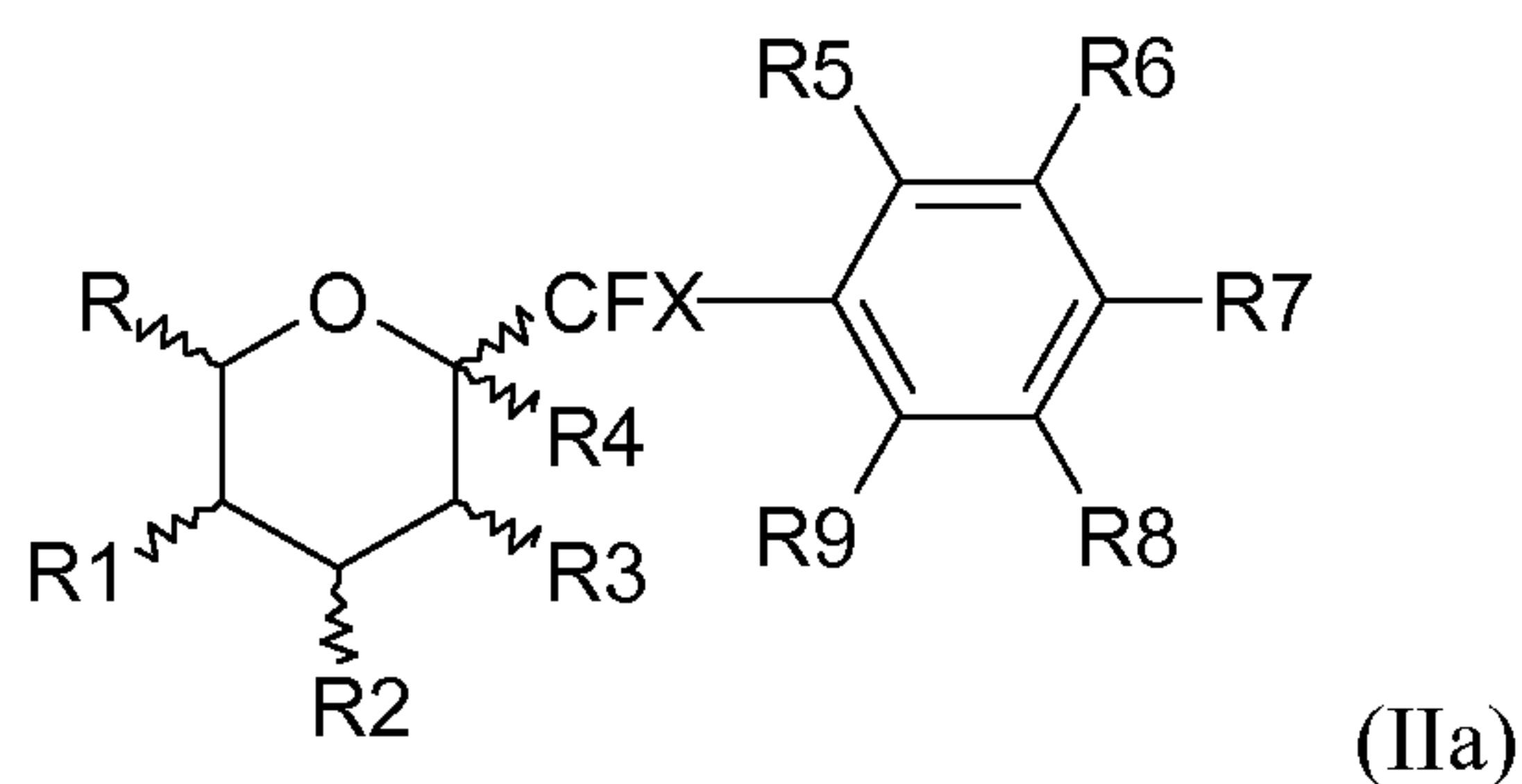
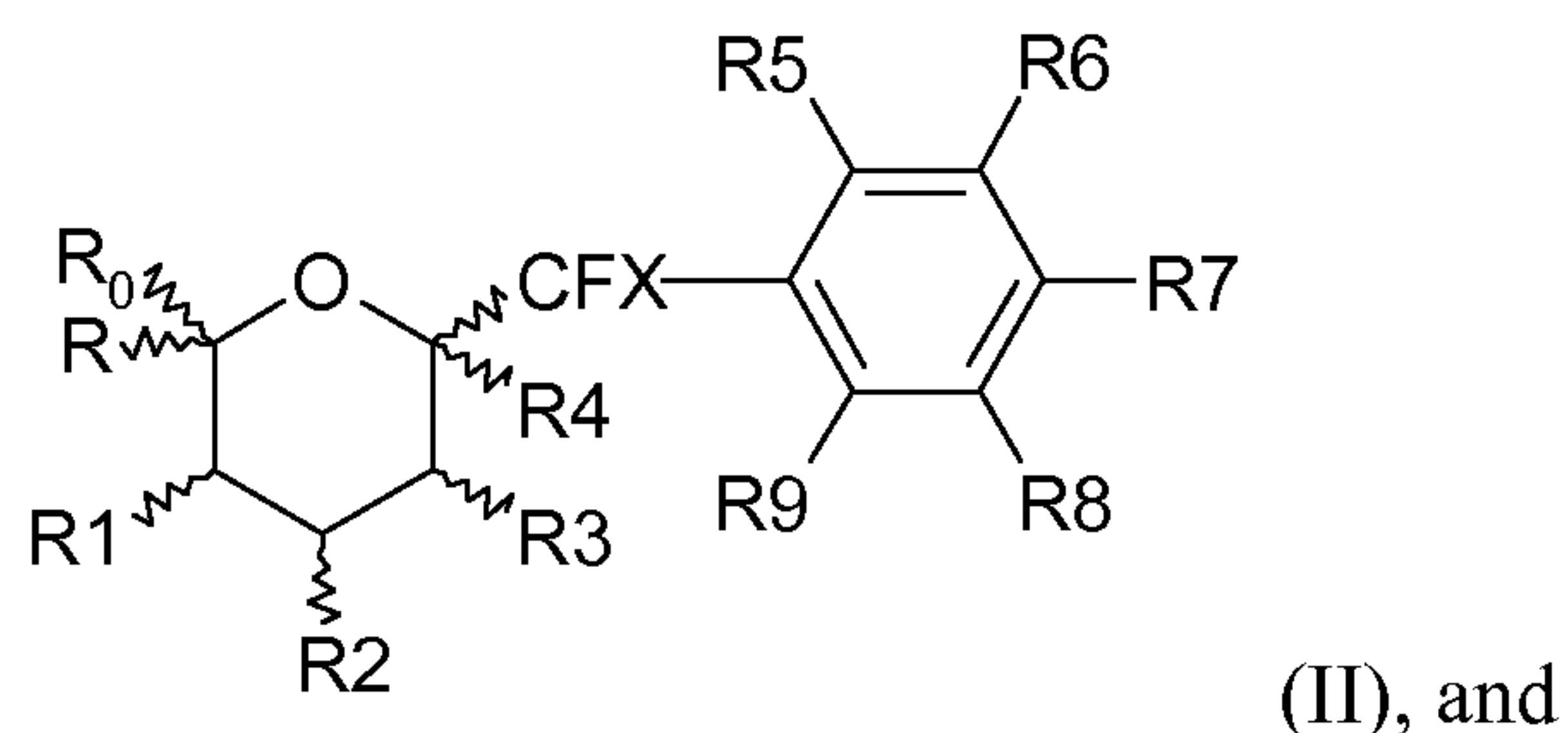
The compounds of the invention are more advantageously based on the formula (Ia-bis).

According to a particular embodiment of the invention, A represents an aryl or heteroaryl group, possibly substituted by one or more groups chosen among an halogen atom, a CN, SO₂, SiR^aR^bR^c, (C₁-C₆)-alkyl, (C₂-C₆)-alkenyl, (C₂-C₆)-alkynyl, (C₃-C₇)-cycloalkyl, 5 to 7 ring-membered heterocycloalkyl, aryl, heteroaryl, aryl-(C₁-C₆)-alkyl, heteroaryl-(C₁-C₆)-alkyl, (C₁-C₆)-alkyl-aryl, (C₁-C₆)-alkyl-heteroaryl, OR¹¹, COR¹¹, OCOR¹¹, CO₂R¹¹, NR¹²R¹³, CONR¹²R¹³, SR¹¹, SO₂R¹¹, CSR¹¹ and OSO₃R¹¹ group, the whole being possibly substituted by one or more groups chosen among an halogen atom, an OH, COOH and CHO group, R^a, R^b, R^c, R¹¹, R¹² and R¹³ being as defined above.

Advantageously, A represents a phenyl or benzylphenyl group, possibly substituted by one or more groups chosen among an halogen atom, a CN, SO₂, SiR^aR^bR^c, (C₁-C₆)-alkyl, (C₂-C₆)-alkenyl, (C₂-C₆)-alkynyl, (C₃-C₇)-cycloalkyl, 5 to 7 ring-membered heterocycloalkyl, aryl, heteroaryl, aryl-(C₁-C₆)-alkyl, heteroaryl-(C₁-C₆)-alkyl, (C₁-C₆)-alkyl-aryl, (C₁-C₆)-alkyl-heteroaryl, OR¹¹, COR¹¹, OCOR¹¹, CO₂R¹¹, NR¹²R¹³, NR¹²COR¹¹, CONR¹²R¹³, SR¹¹, SO₂R¹¹, CSR¹¹ and OSO₃R¹¹ group, the whole being possibly substituted by one or more groups chosen among an halogen atom, an OH, (C₁-C₆)-alkyl, (C₁-C₆)-alkoxy, COOH and CHO group, R^a, R^b, R^c, R¹¹, R¹² and R¹³ being as defined above.

In an equally advantageously manner, the radical A represents a phenyl group possibly substituted by one or more groups chosen among an halogen atom, a CN, SO₂, SiR^aR^bR^c, (C₁-C₆)-alkyl, (C₂-C₆)-alkenyl, (C₂-C₆)-alkynyl, (C₃-C₇)-cycloalkyl, 5 to 7 ring-membered heterocycloalkyl, aryl, heteroaryl, aryl-(C₁-C₆)-alkyl, heteroaryl-(C₁-C₆)-alkyl, (C₁-C₆)-alkyl-aryl, (C₁-C₆)-alkyl-heteroaryl, OR¹¹, COR¹¹, OCOR¹¹, CO₂R¹¹, NR¹²R¹³, CONR¹²R¹³, SR¹¹, SO₂R¹¹, CSR¹¹ and OSO₃R¹¹ group, the whole being possibly substituted by one or more groups chosen among an halogen atom, an OH, (C₁-C₆)-alkyl, (C₁-C₆)-alkoxy, COOH and CHO group, and notably among an halogen atom, an OH, COOH and CHO group, R^a, R^b, R^c, R¹¹, R¹² and R¹³ being as defined above.

Consequently, according to a first particular embodiment of the invention, a compound of the invention is advantageously based on the following generic formula (II), and more advantageously based on the following generic formula (IIa):



wherein:

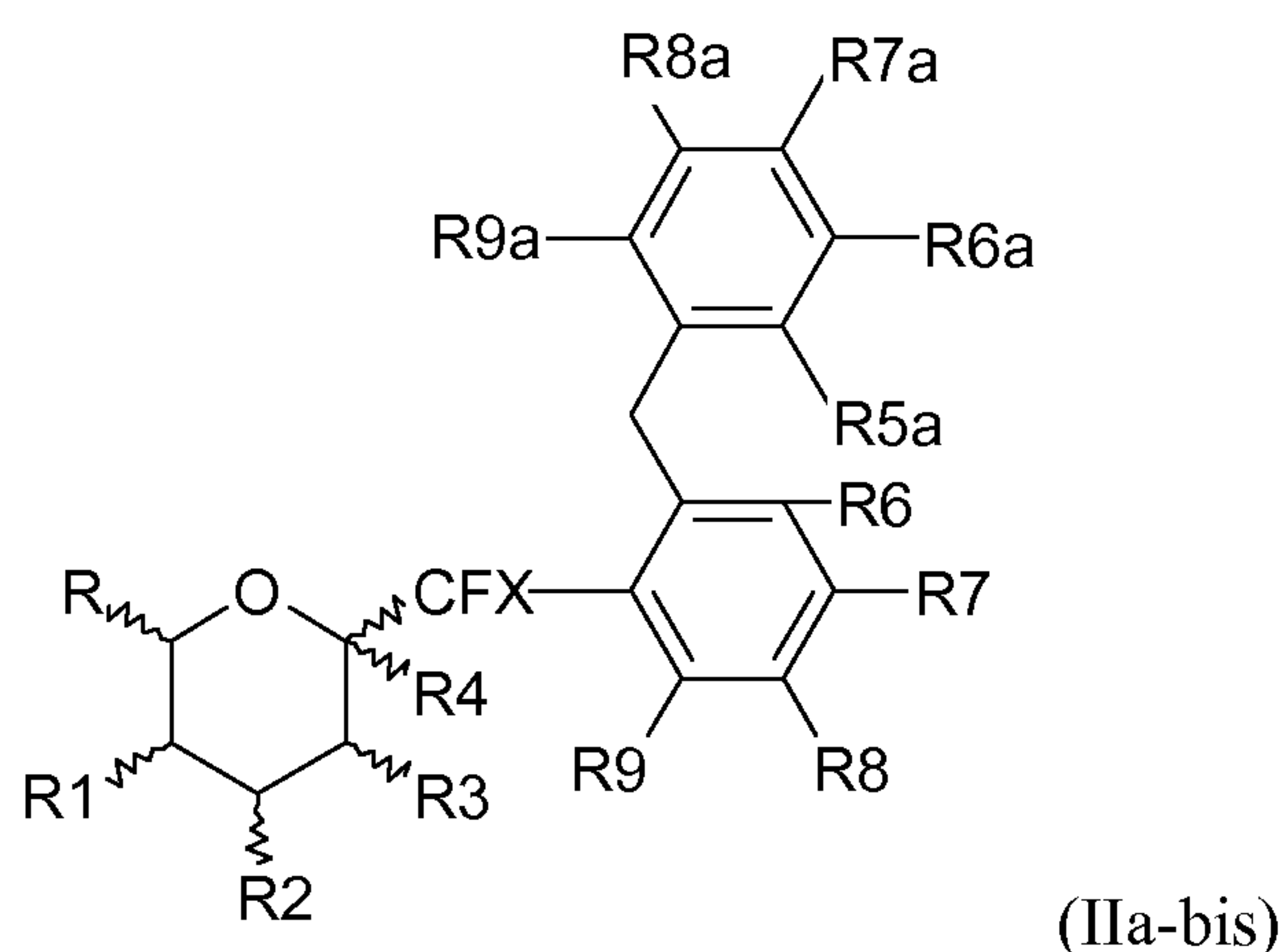
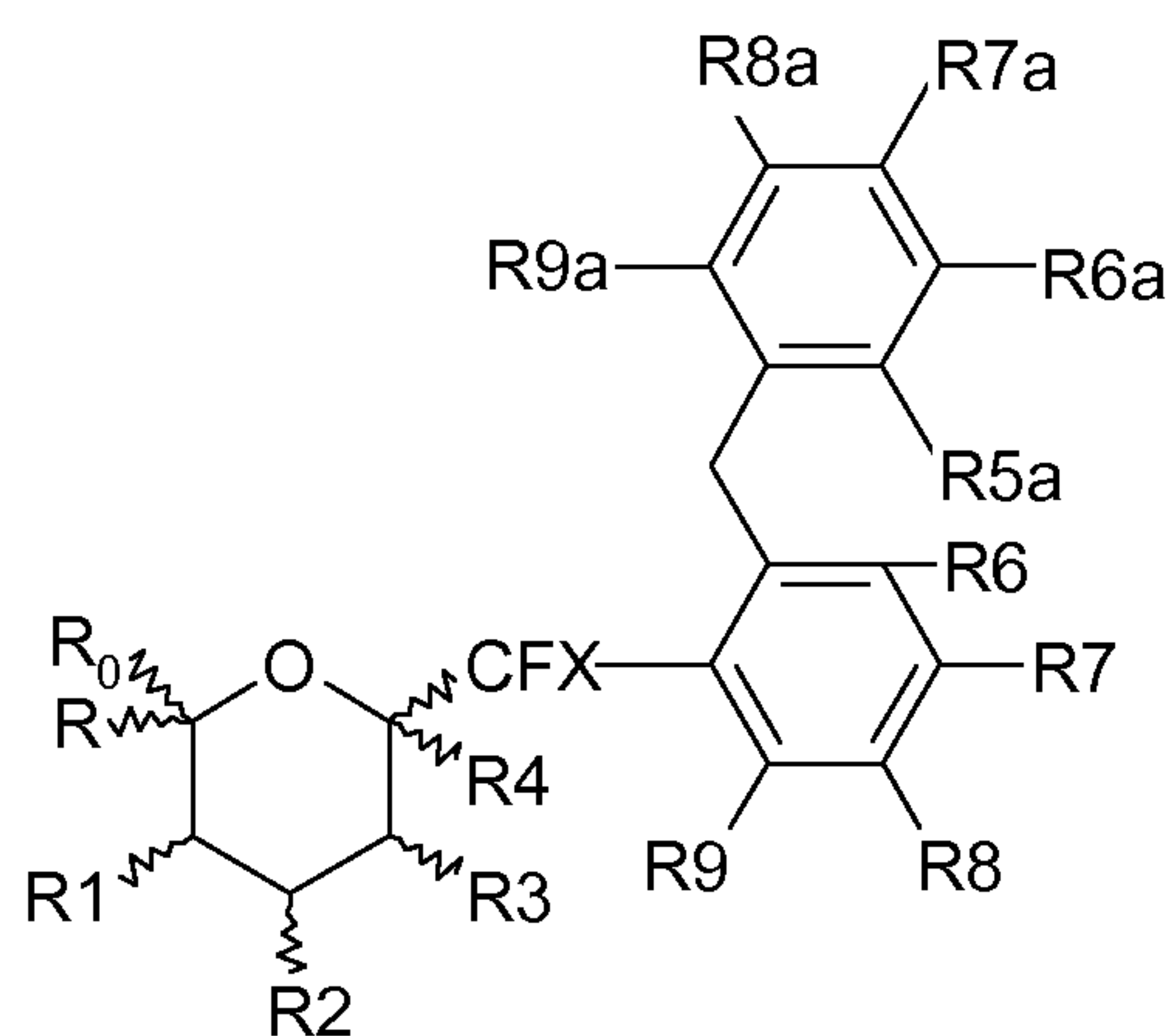
- R₅, R₆, R₇, R₈ and R₉ represent, independently from one another, a hydrogen atom, an halogen atom, a CN, SO₂, SiR^aR^bR^c, (C₁-C₆)-alkyl, (C₂-C₆)-alkenyl, (C₂-C₆)-alkynyl, (C₃-C₇)-cycloalkyl, 5 to 7 ring-membered heterocycloalkyl, aryl, heteroaryl, aryl-(C₁-C₆)-alkyl, heteroaryl-(C₁-C₆)-alkyl, (C₁-C₆)-alkyl-aryl, (C₁-C₆)-alkyl-heteroaryl, OR¹¹, COR¹¹, OCOR¹¹, CO₂R¹¹, NR¹²R¹³, NR¹²COR¹¹, CONR¹²R¹³, SR¹¹, SO₂R¹¹, CSR¹¹ or OSO₃R¹¹ group, the said group being possibly substituted by one or more

groups chosen among an halogen atom, an OH, (C₁-C₆)-alkyl, (C₁-C₆)-alkoxy, COOH and CHO group; and in particular by one or more groups chosen among an halogen atom, an OH, COOH and CHO group, and

- X, R, R₁, R₂, R₃, R₄, R₀, R¹¹, R¹² and R¹³ are as defined above.

Thus, compound of formula (IIa) corresponds to a compound of formula (II) wherein R₀ = H.

According to a second particular embodiment of the invention, a compound of the invention is advantageously based on the following generic formula (IIbis), and more advantageously based on the following generic formula (IIa-bis):



wherein:

- R₆, R₇, R₈, R₉, R_{5a}, R_{6a}, R_{7a}, R_{8a} and R_{9a} represent, independently from one another, a hydrogen atom, an halogen atom, a CN, SO₂, SiR^aR^bR^c, (C₁-C₆)-alkyl, (C₂-C₆)-alkenyl, (C₂-C₆)-alkynyl, (C₃-C₇)-cycloalkyl, 5 to 7 ring-membered heterocycloalkyl, aryl, heteroaryl, aryl-(C₁-C₆)-alkyl, heteroaryl-(C₁-C₆)-

alkyl-aryl, (C₁-C₆)-alkyl-heteroaryl, OR¹¹, COR¹¹, OCOR¹¹, CO₂R¹¹, NR¹²R¹³, NR¹²COR¹¹, CONR¹²R¹³, SR¹¹, SO₂R¹¹, CSR¹¹ or OSO₃R¹¹ group, the said group being possibly substituted by one or more groups chosen among an halogen atom, an OH, (C₁-C₆)-alkyl, (C₁-C₆)-alkoxy, COOH and CHO group; and in particular by one or more
 5 groups chosen among an halogen atom, an OH, COOH and CHO group, and
 - X, R, R₁, R₂, R₃, R₄, R₀, R¹¹, R¹² and R¹³ are as defined above.

Thus, compound of formula (IIa-bis) corresponds to a compound of formula (IIbis) wherein R₀ = H.

Preferably, R₁, R₂ and R₃ represent, independently from one another, a fluorine
 10 atom or an OH, OSiR^aR^bR^c, OR¹¹, OCOR¹¹, OCO₂R¹¹ or OCONR¹²R¹³ group

R₁, R₂ and R₃ may advantageously be chosen, independently from one another, among an OH, OR¹¹ and OCOR¹¹ group with R¹¹ as defined above.

Even more advantageously, R₁, R₂ and R₃ may be chosen, independently from one another, among an OH, -O-(C₁-C₆)-alkyl, -O-aryl, -O-(C₁-C₆)-alkyl-aryl and -
 15 OCO-(C₁-C₆)-alkyl group.

In particular, R₁, R₂ and R₃ may be chosen, independently from one another, among an OH, OSiMe₃ and benzyloxy (OBn) group, and preferably among OH and OBn.

According to a particular embodiment, R₁, R₂ and R₃ are identical.

20 According to another particular embodiment, R₁, R₂ and R₃ are identical and represent each an OH group and R represents a CH₂OH group.

R advantageously represents a hydrogen atom or a CH₃, CH₂OH, CH₂OR¹¹, CH₂OSiR^aR^bR^c, CH₂OCOR¹¹, CH₂OP(O)(OH)₂ or CH₂OSO₃H group, and in particular a hydrogen atom or a CH₃, CH₂OH, CH₂OR¹¹, CH₂OCOR¹¹, CH₂OP(O)(OH)₂ or
 25 CH₂OSO₃H group,

with R^a, R^b, R^c and R¹¹ as defined above, and with CH₂OR¹¹ advantageously representing a -CH₂O-(C₁-C₆)-alkyl, -CH₂O-aryl and -CH₂O-(C₁-C₆)-alkyl-aryl, and CH₂OCOR¹¹ group advantageously representing a -CH₂OCO-(C₁-C₆)-alkyl group.

Even more advantageously, R represents a CH₂OH, CH₂OSiR^aR^bR^c, CH₂OR¹¹
 30 or CH₂OCOR¹¹ group, and more advantageously a CH₂OH, CH₂OR¹¹ or CH₂OCOR¹¹ group, with R^a, R^b, R^c and R¹¹ as defined above.

Yet even more advantageously, R represents a CH₂OH, -CH₂O-(C₁-C₆)-alkyl, -CH₂O-aryl, -CH₂O-(C₁-C₆)-alkyl-aryl and -CH₂OCO-(C₁-C₆)-alkyl group.

In particular, R can represent a CH₂OH, CH₂OSiMe₃ or CH₂OBn group, and preferably a CH₂OH or CH₂OBn group.

5 In the same way, R₄ may advantageously represent a hydrogen or halogen atom or an OH or OR¹¹ group, and in particular a hydrogen atom or an OH or OR¹¹ group, with R¹¹ as defined above.

Yet even more advantageously, R₄ may represent a hydrogen or halogen atom or an OH, -O-(C₁-C₆)-alkyl, -O-aryl and -O-(C₁-C₆)-alkyl-aryl group, and in particular, a
10 hydrogen atom or an OH, -O-(C₁-C₆)-alkyl, -O-aryl and -O-(C₁-C₆)-alkyl-aryl group.

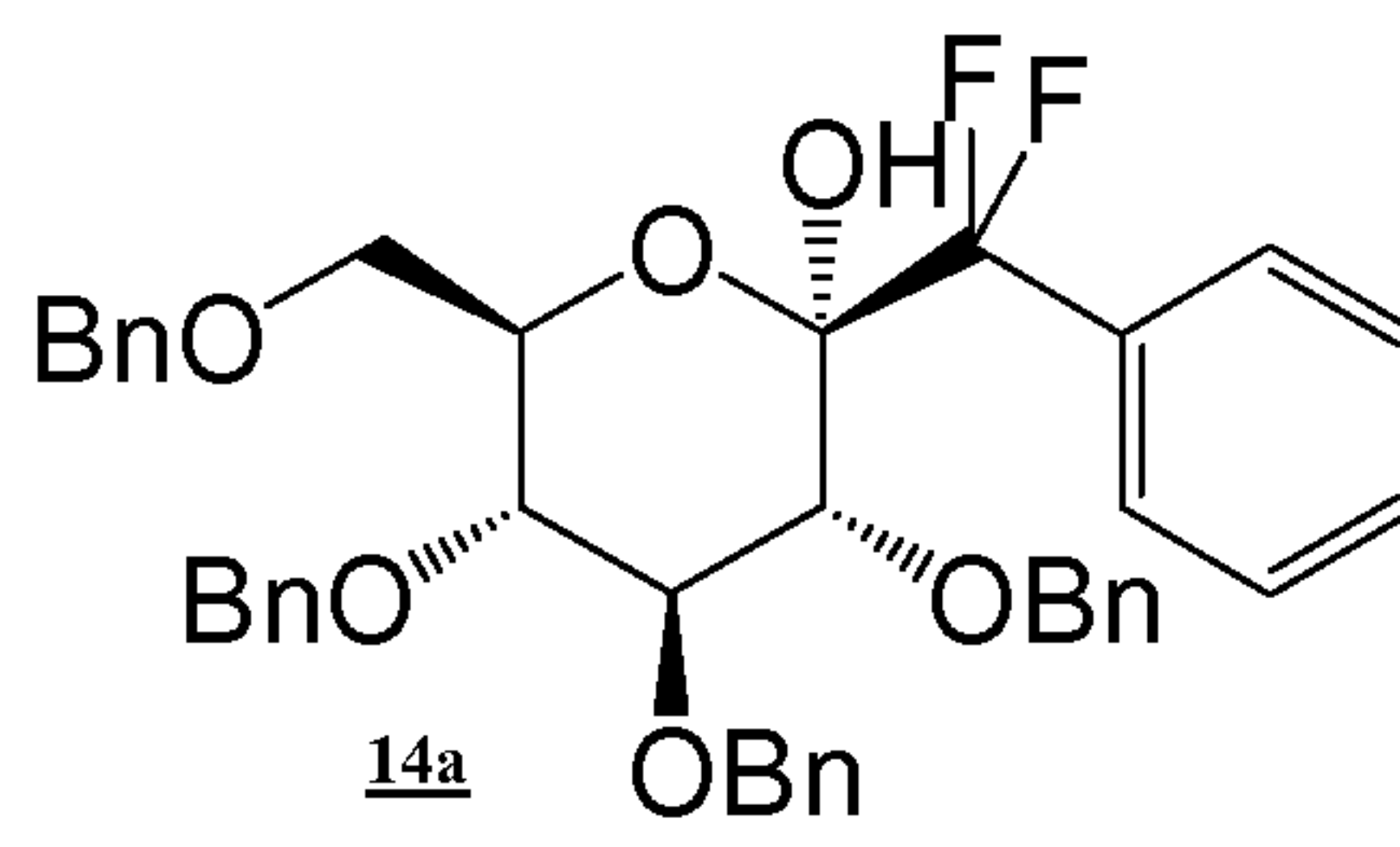
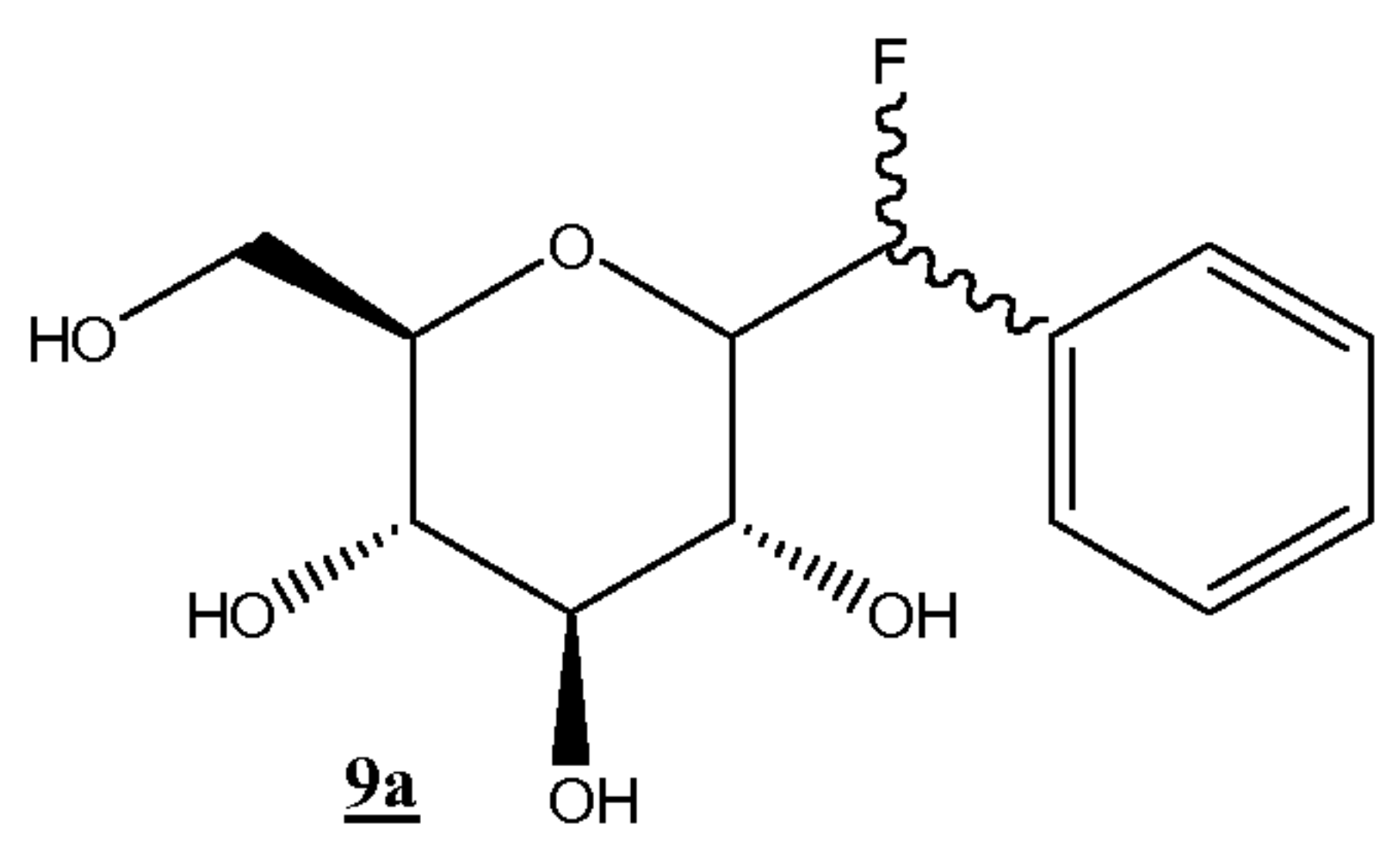
In particular, R₄ can represent a hydrogen or halogen (such as Br, Cl, F) atom or an OH group, and advantageously, a hydrogen atom or an OH group.

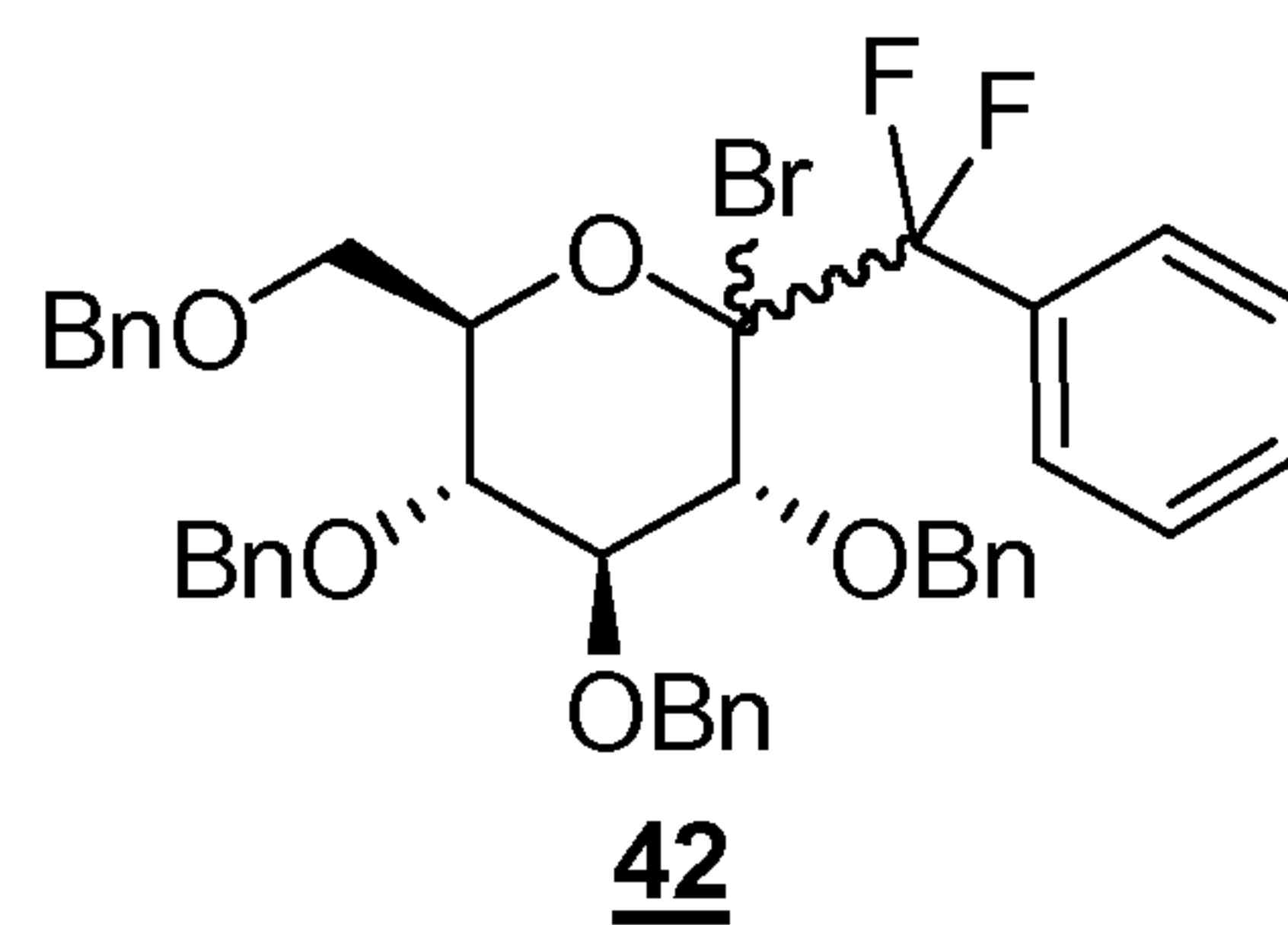
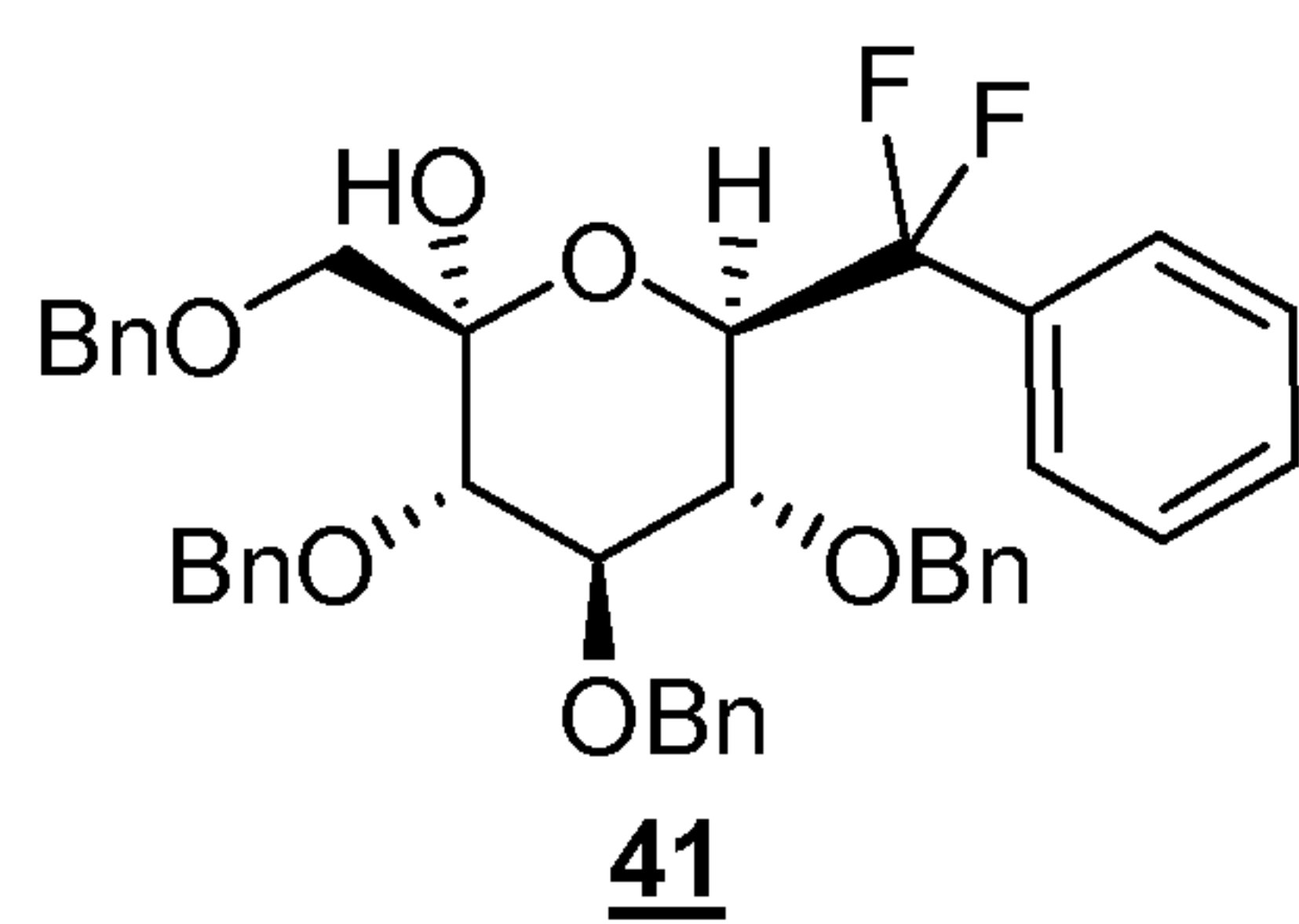
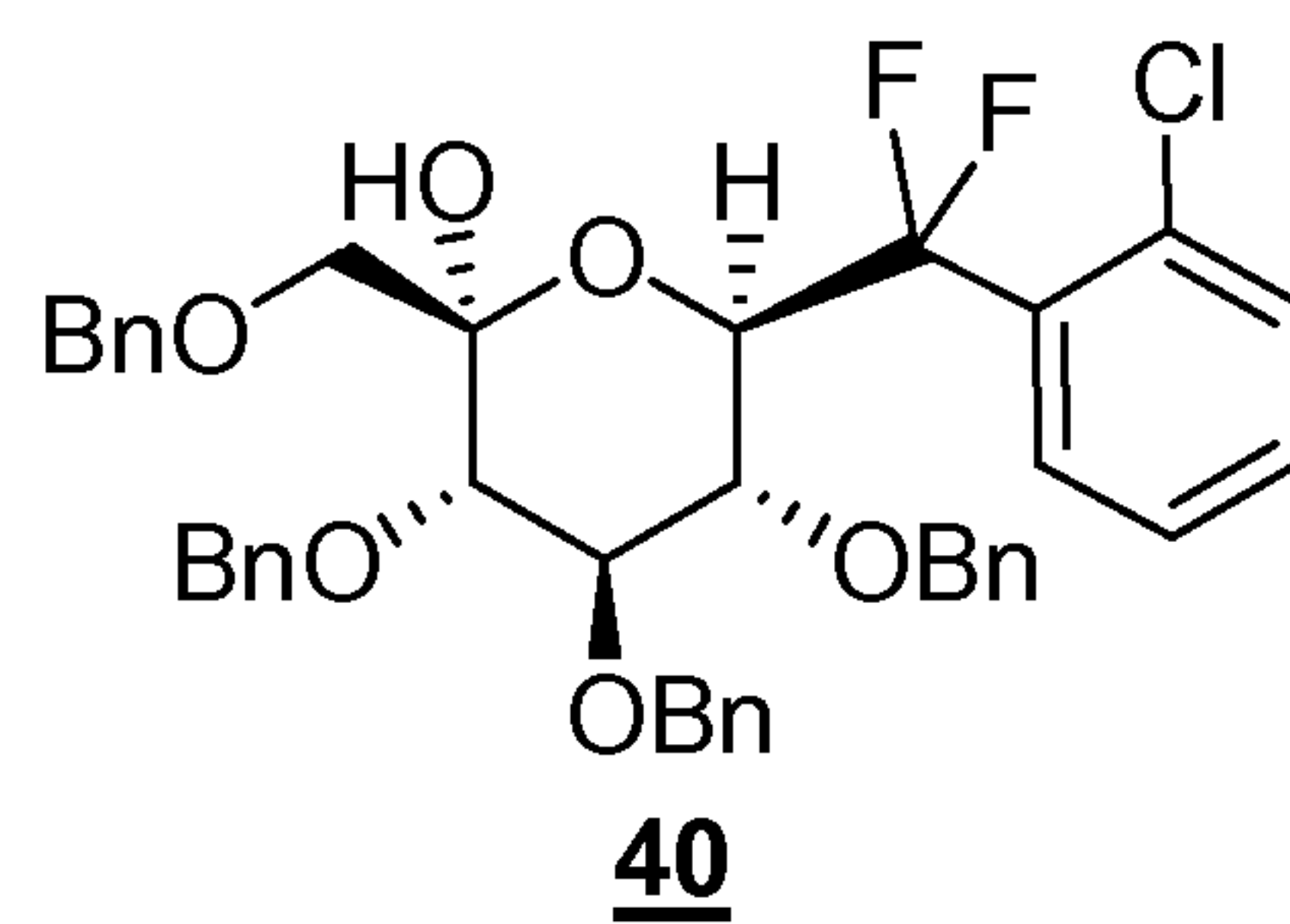
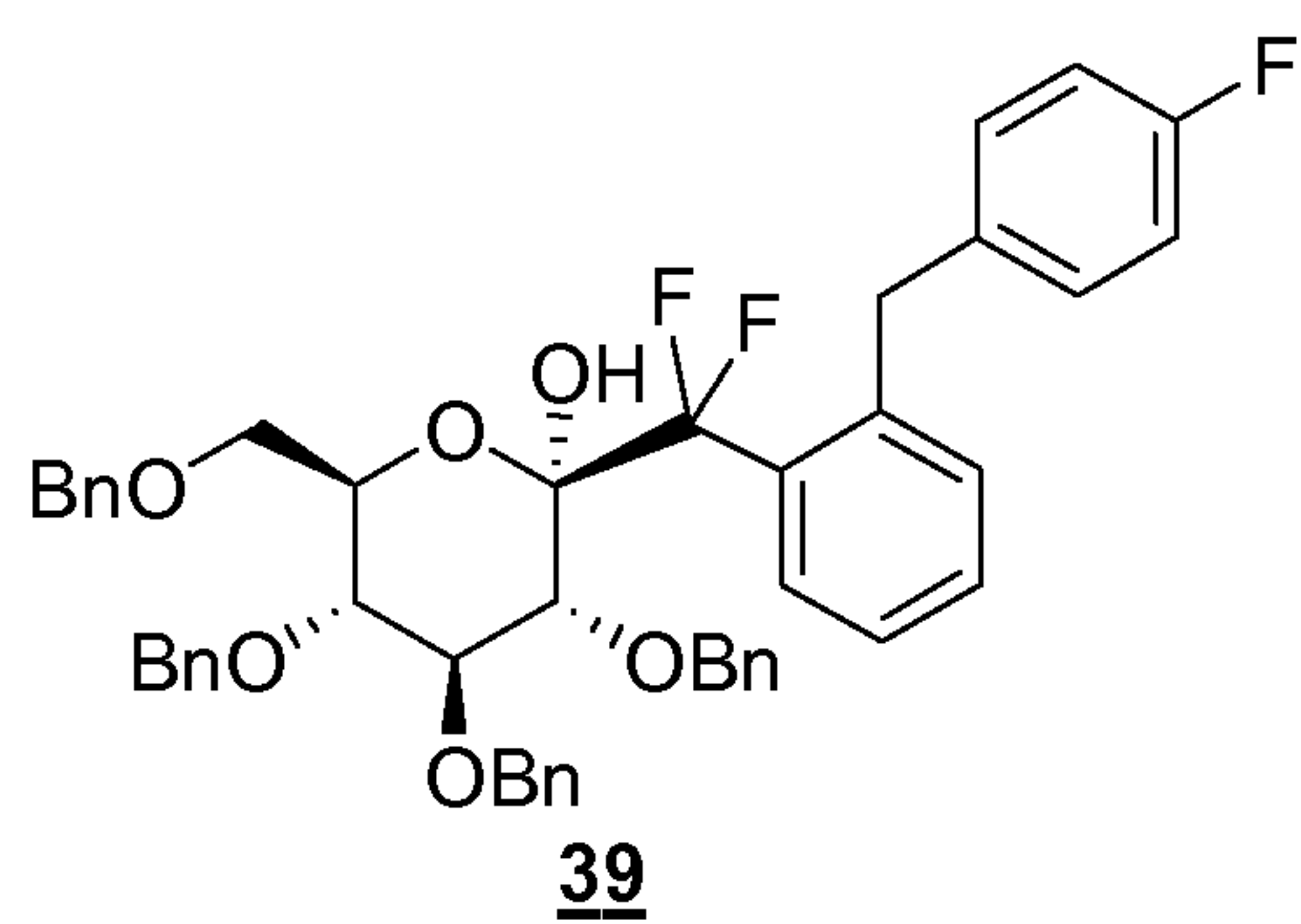
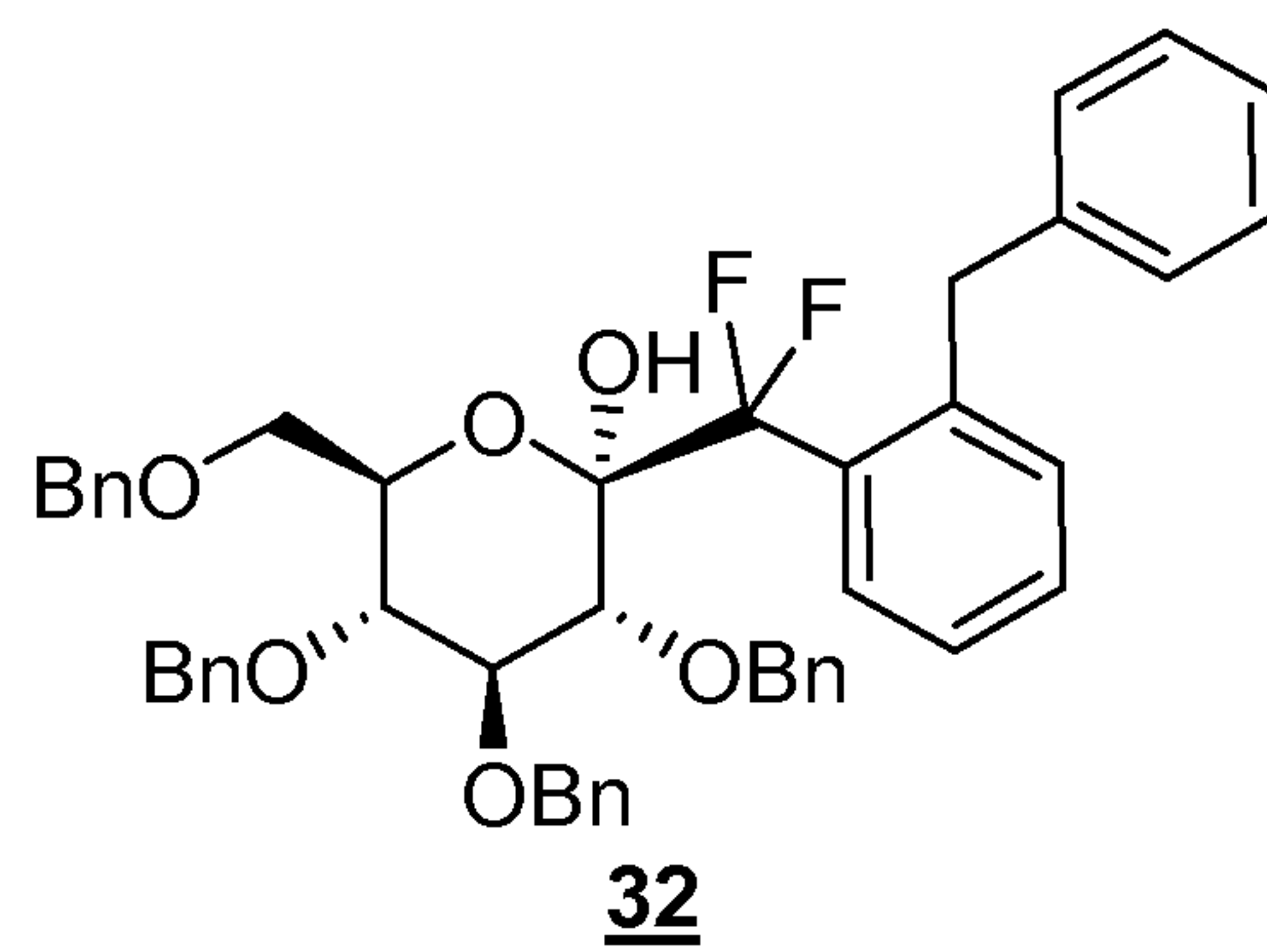
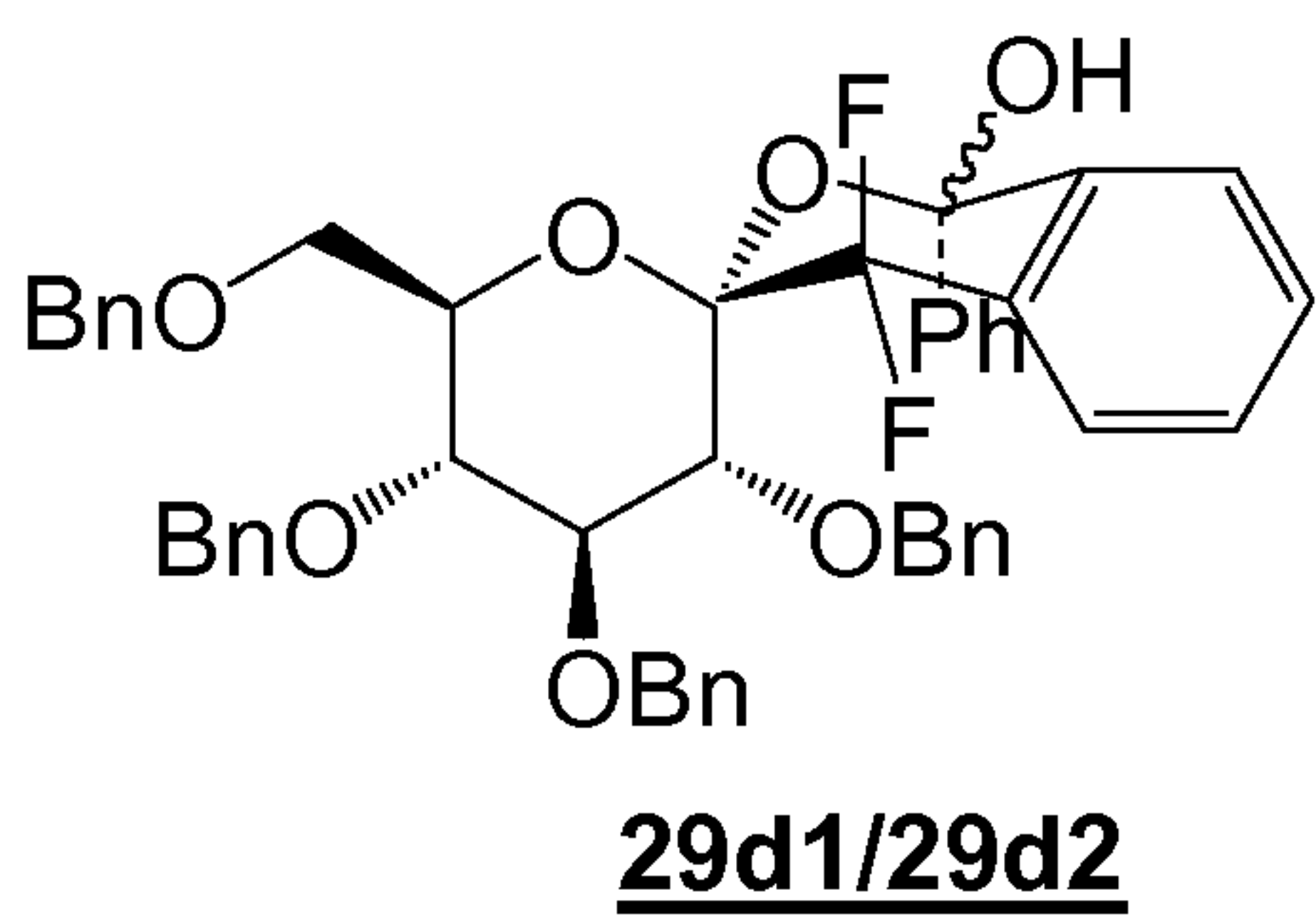
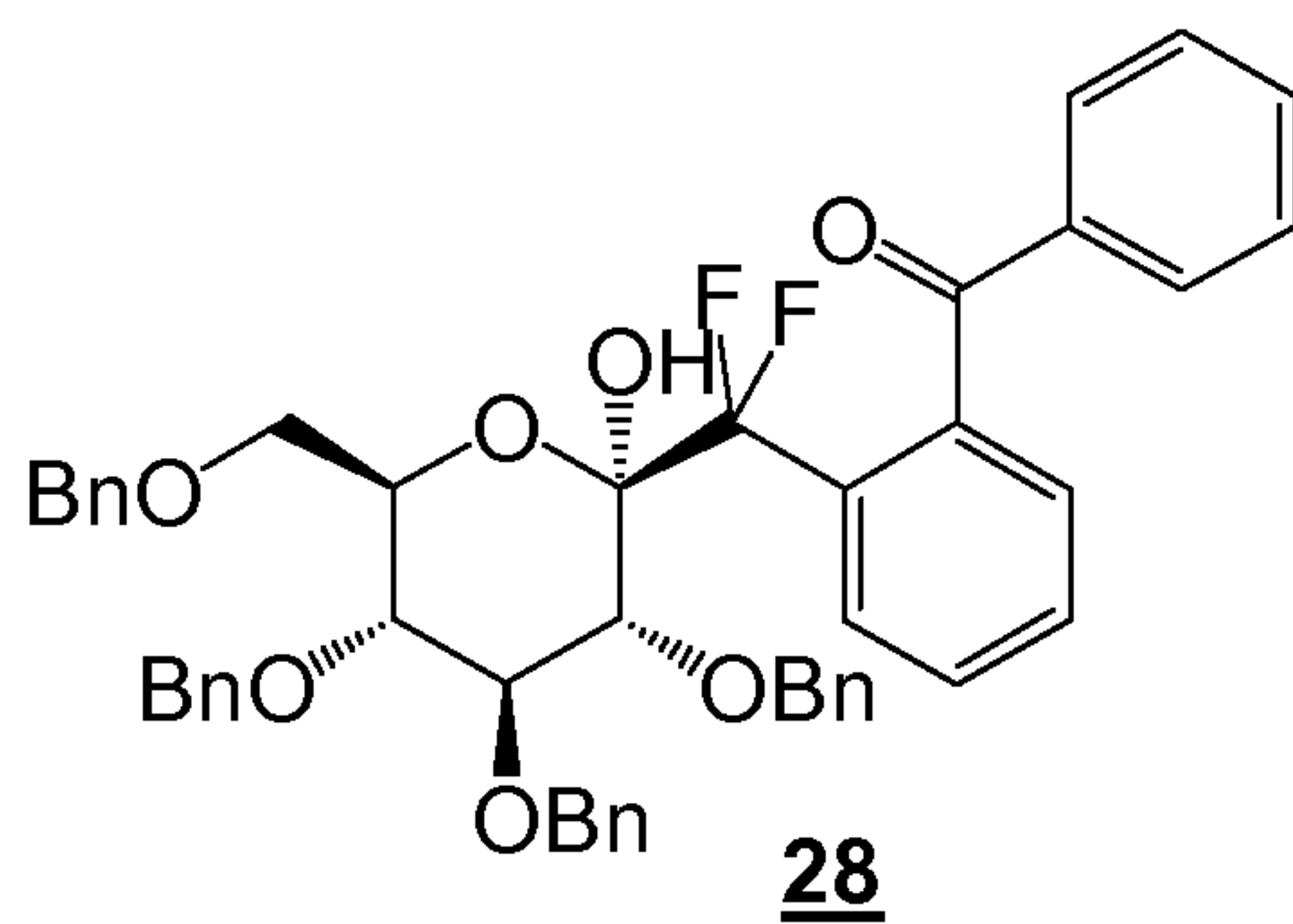
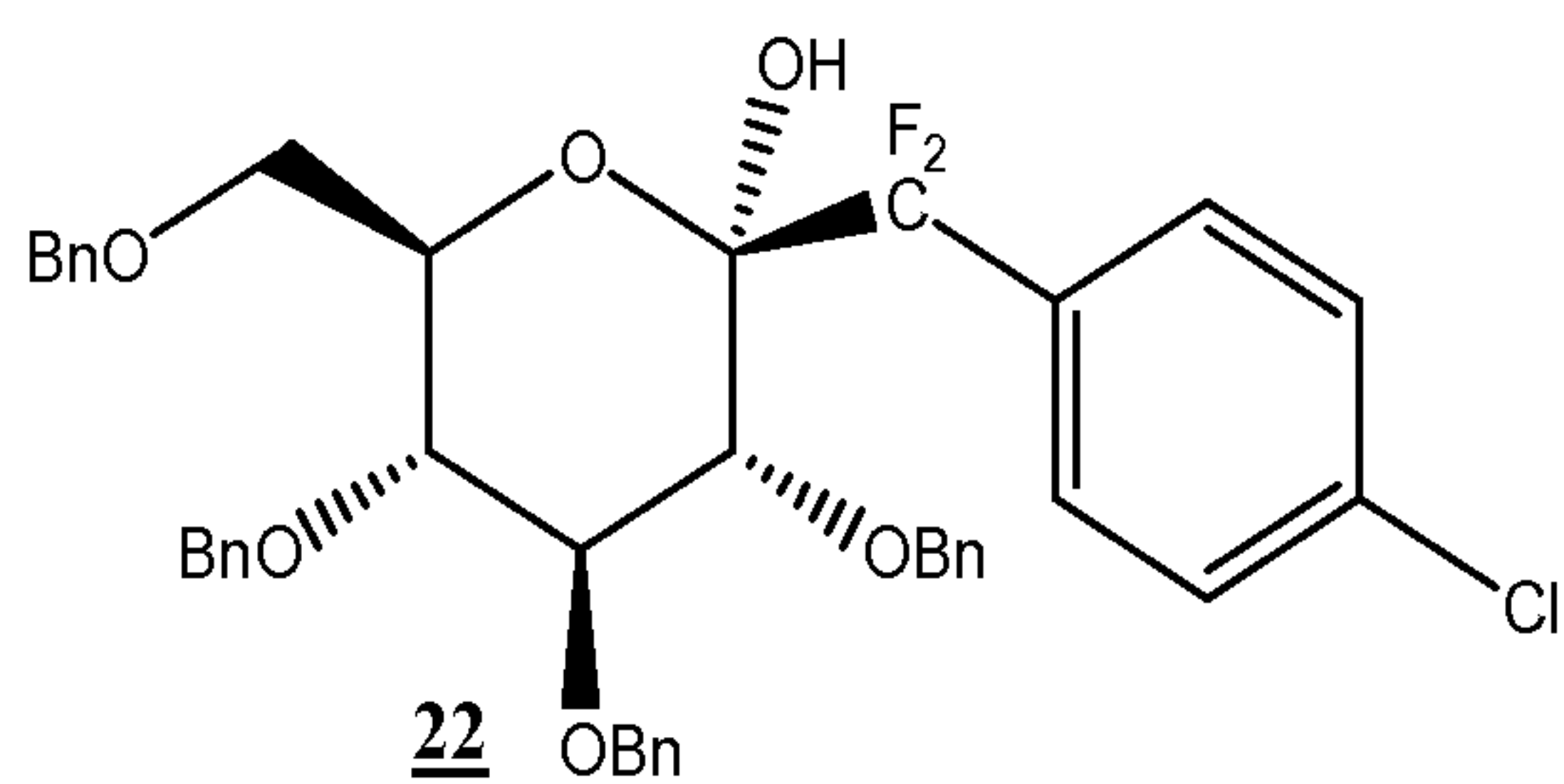
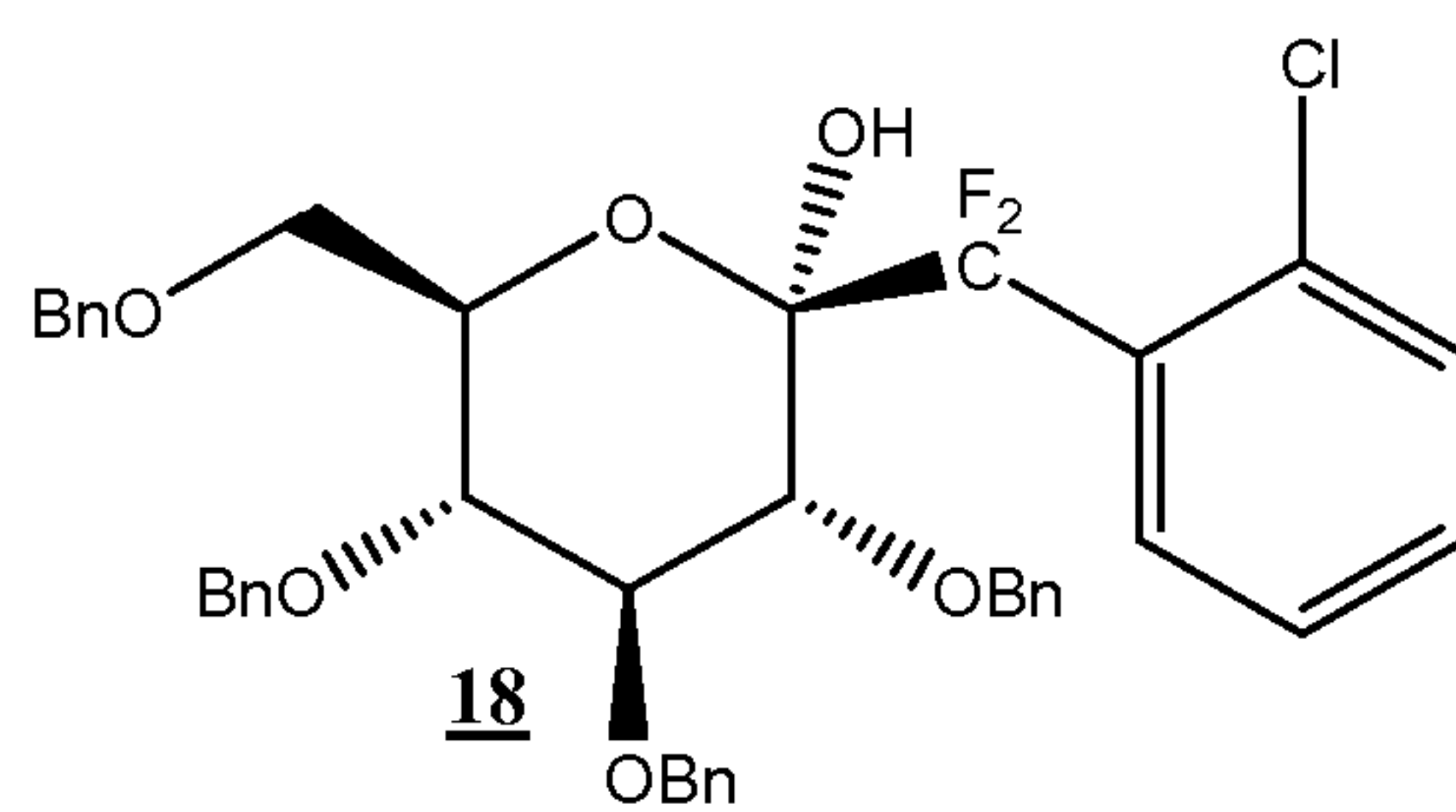
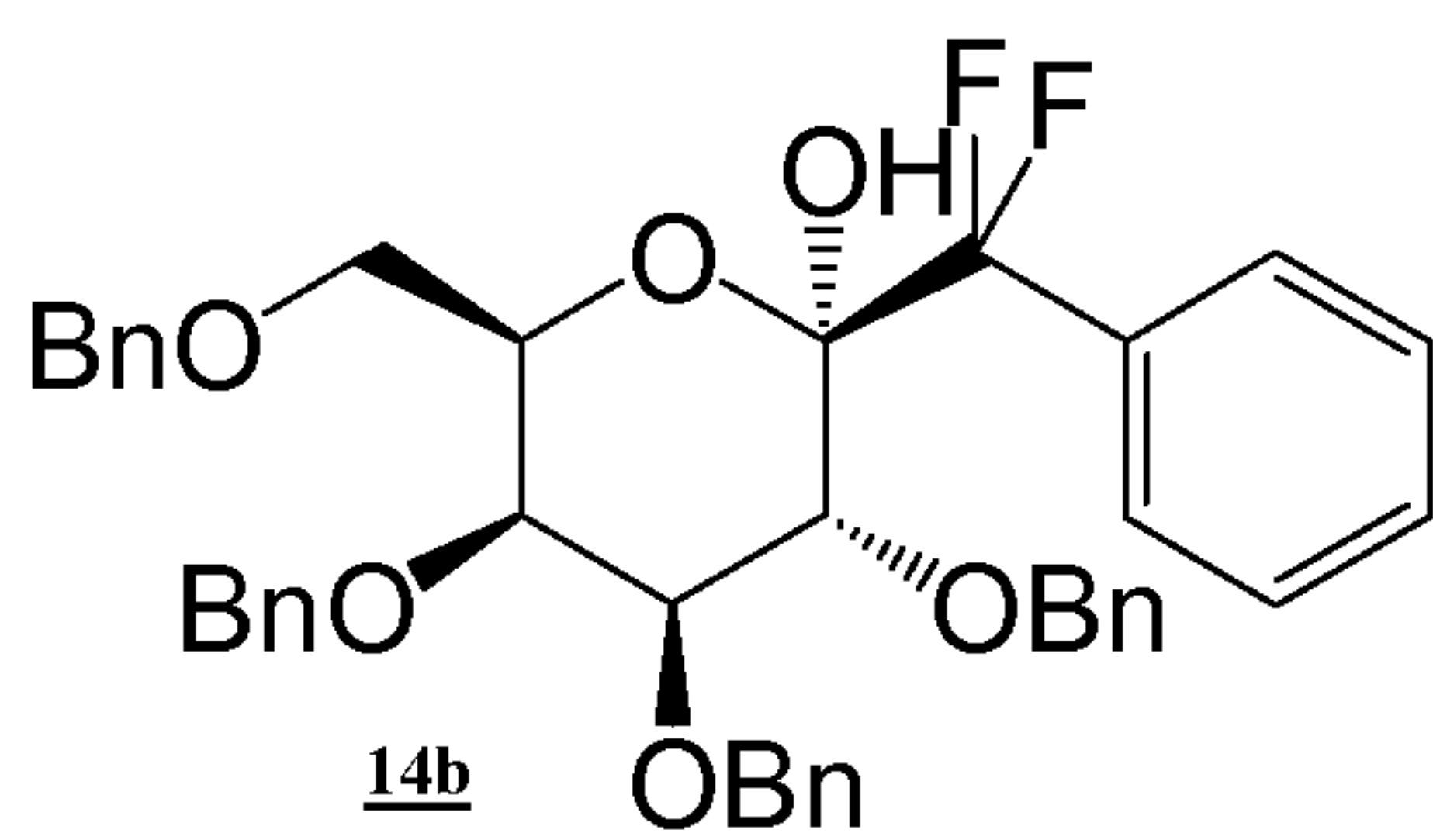
R₅, R₆, R₇, R₈, R₉, R_{5a}, R_{6a}, R_{7a}, R_{8a} and R_{9a} can be chosen among a hydrogen atom, a halogen atom, advantageously a chlorine atom, an aryl-(C₁-C₆)-alkyl
15 group, advantageously benzyl, the alkyl group being possibly substituted by an OH group.

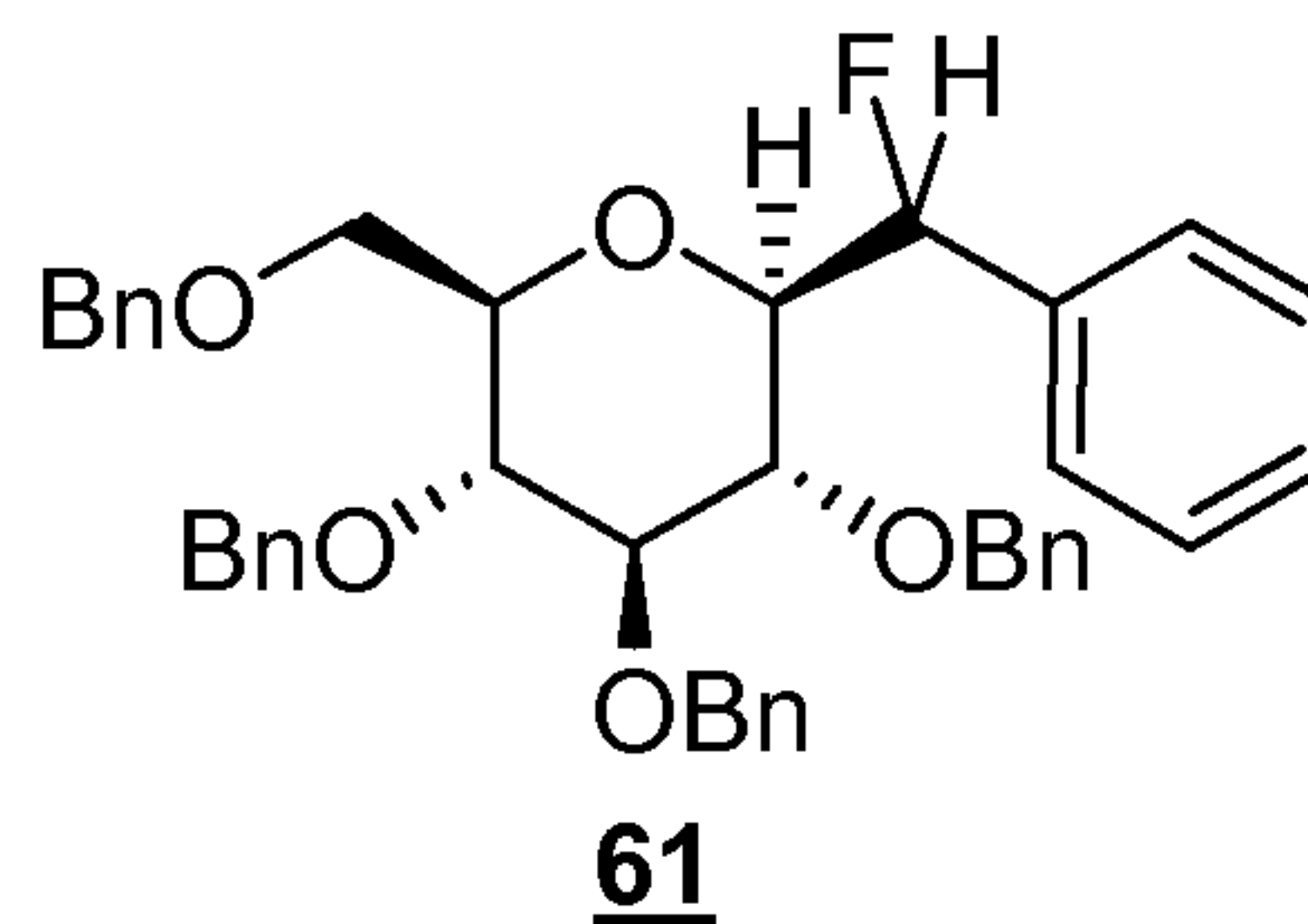
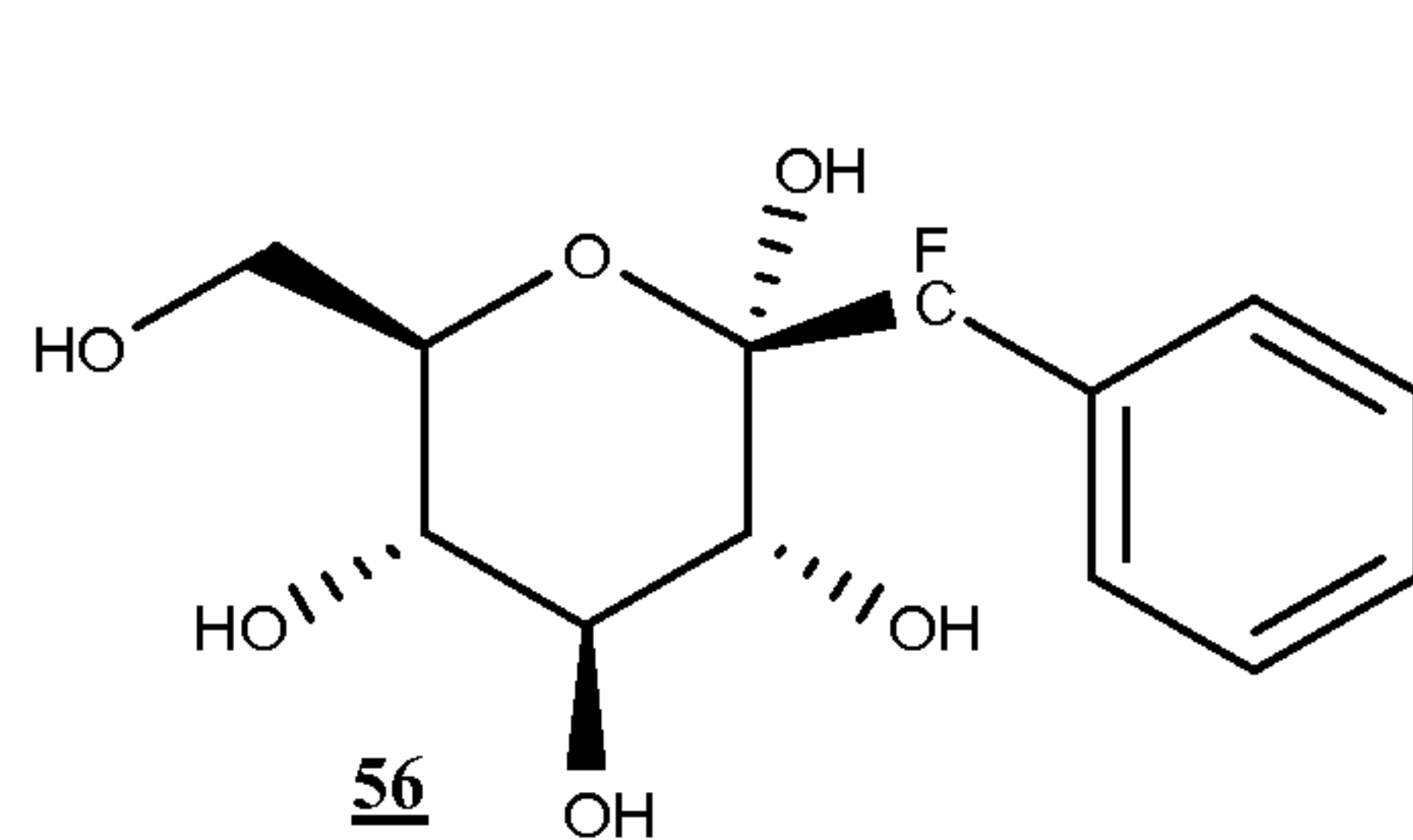
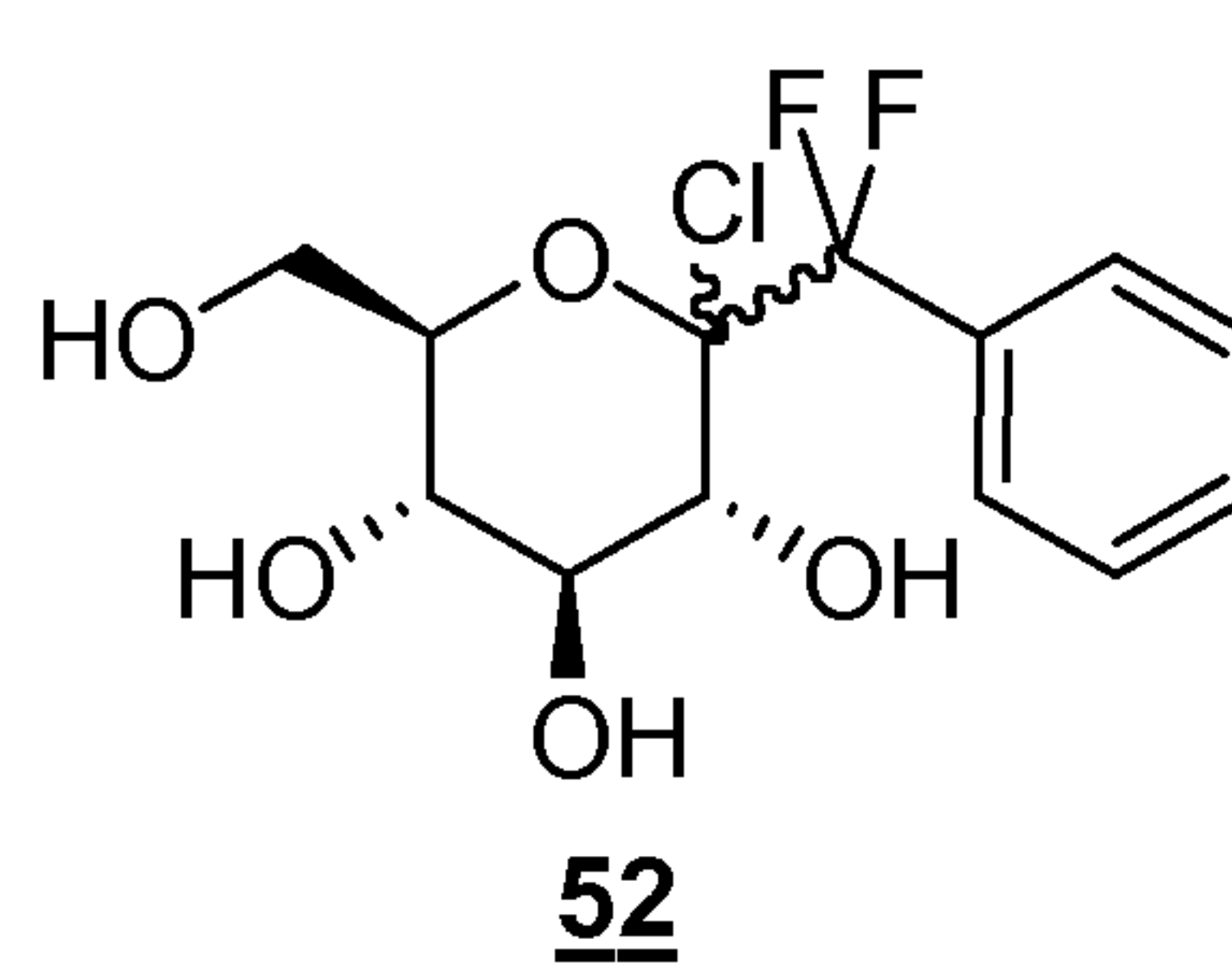
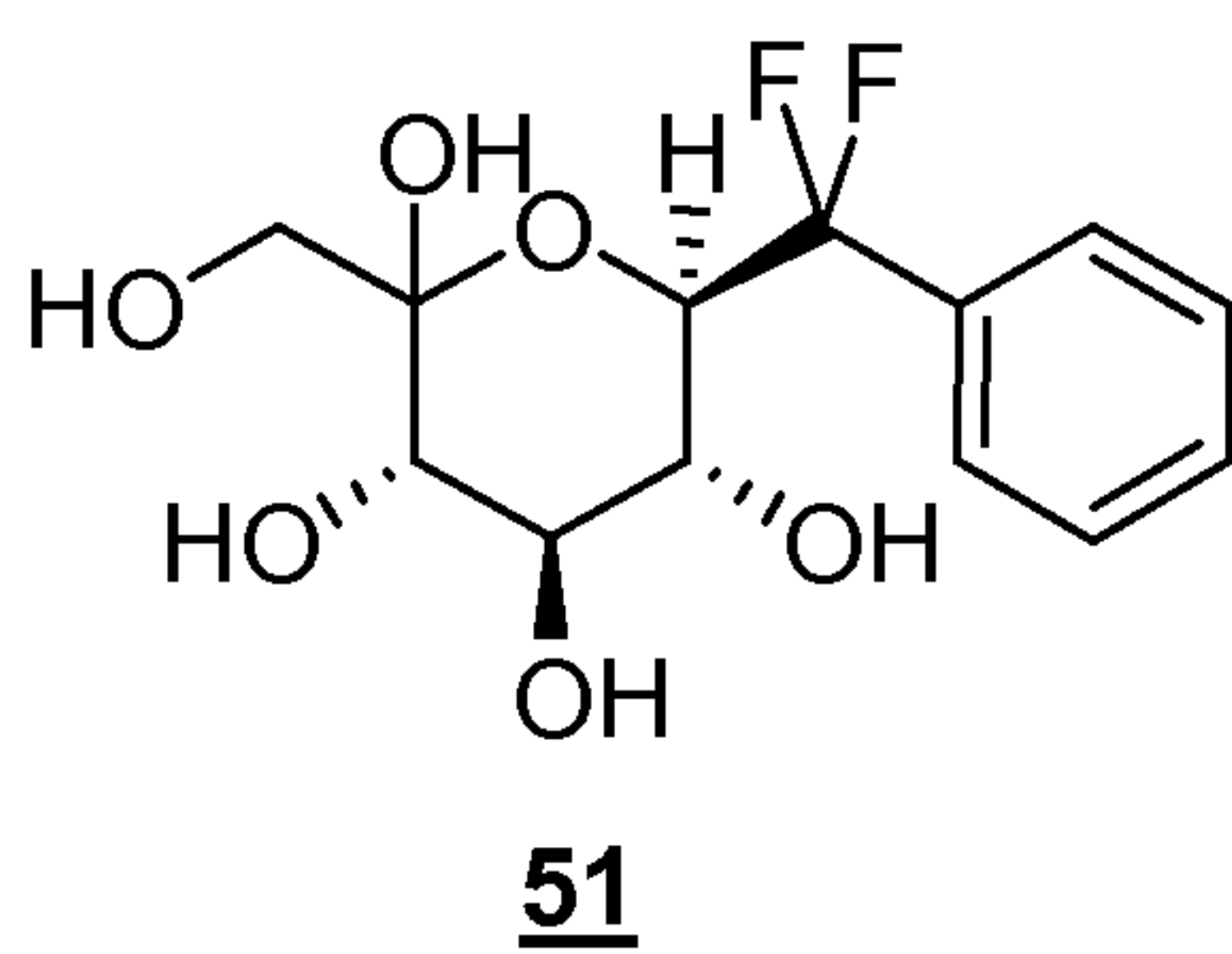
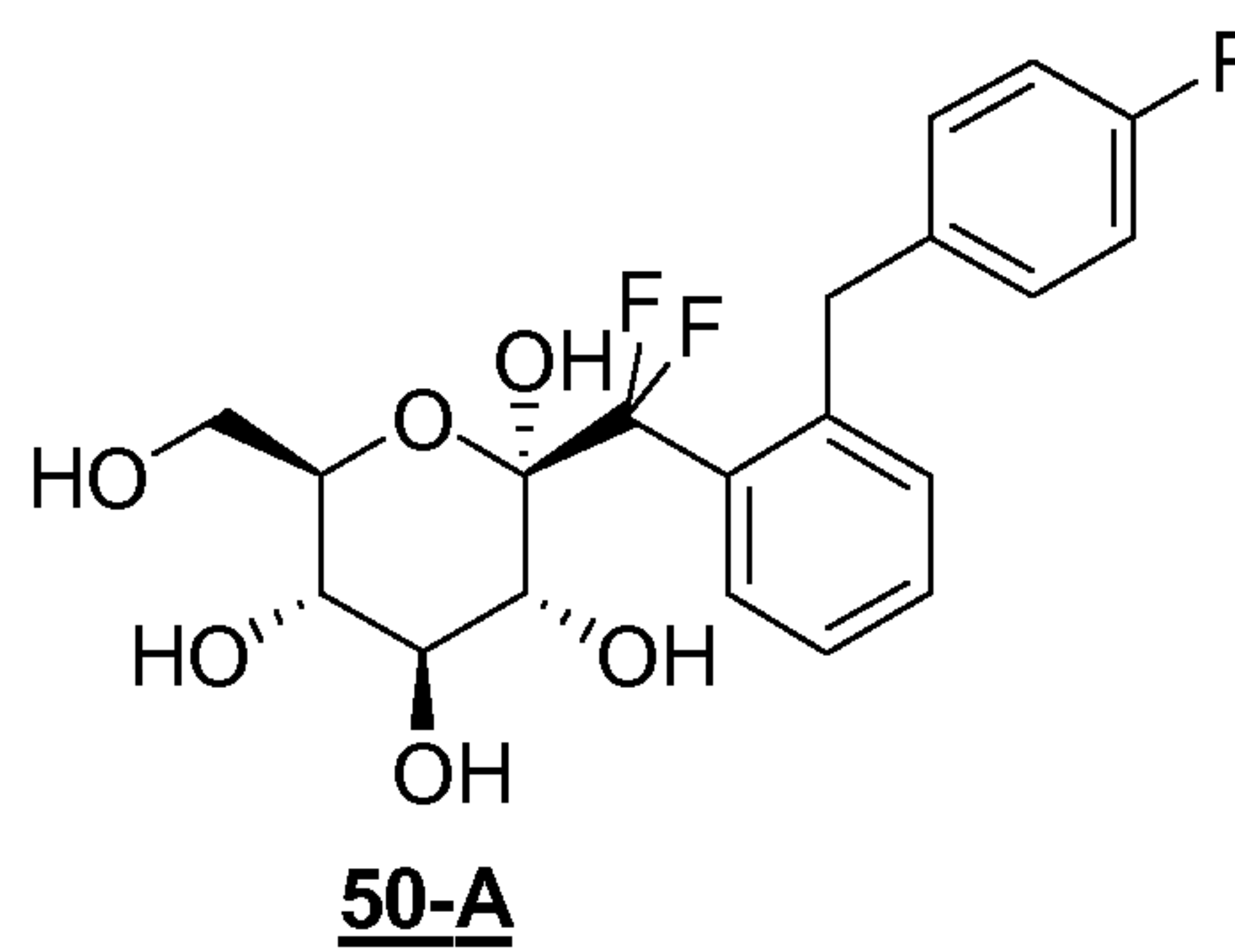
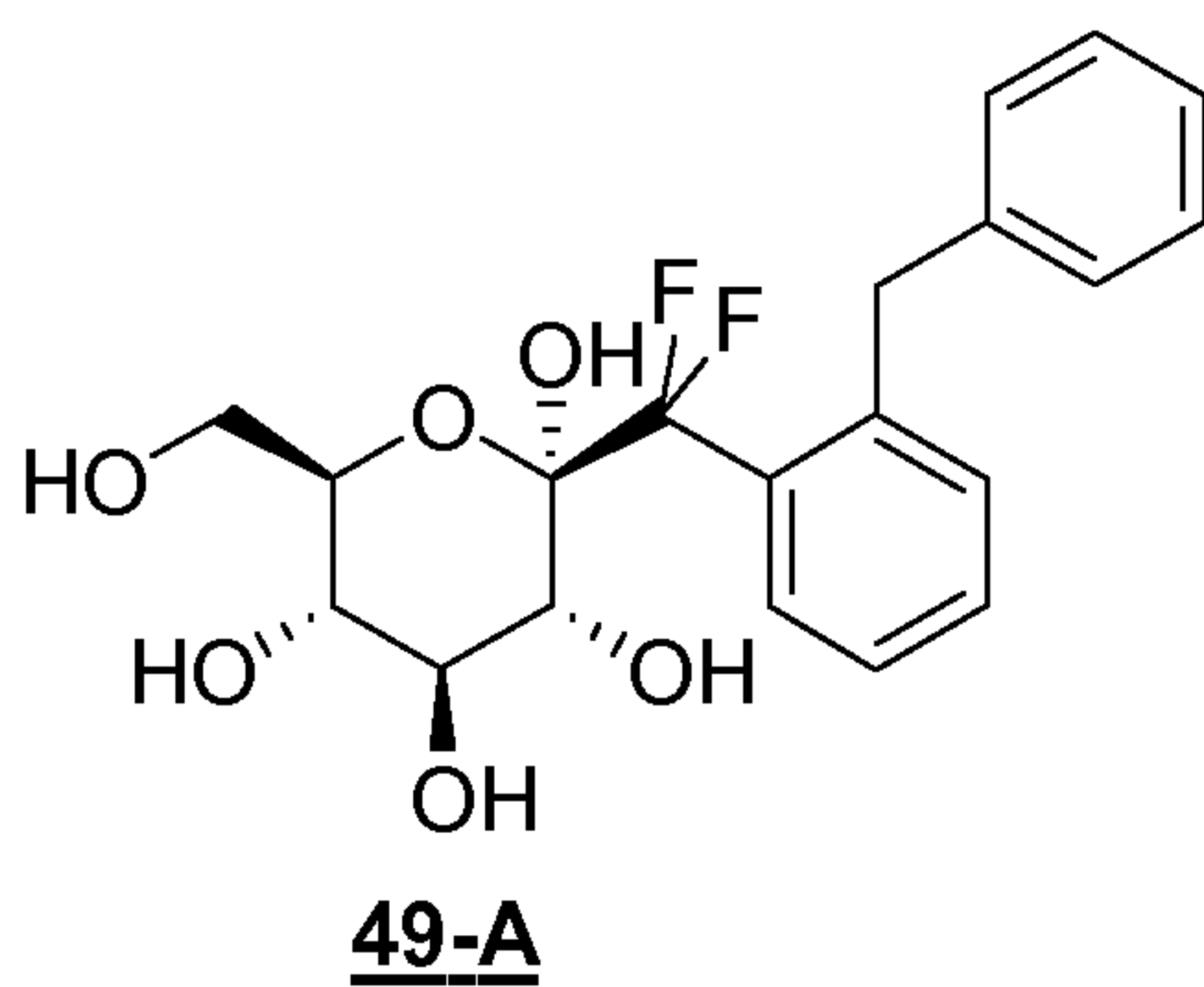
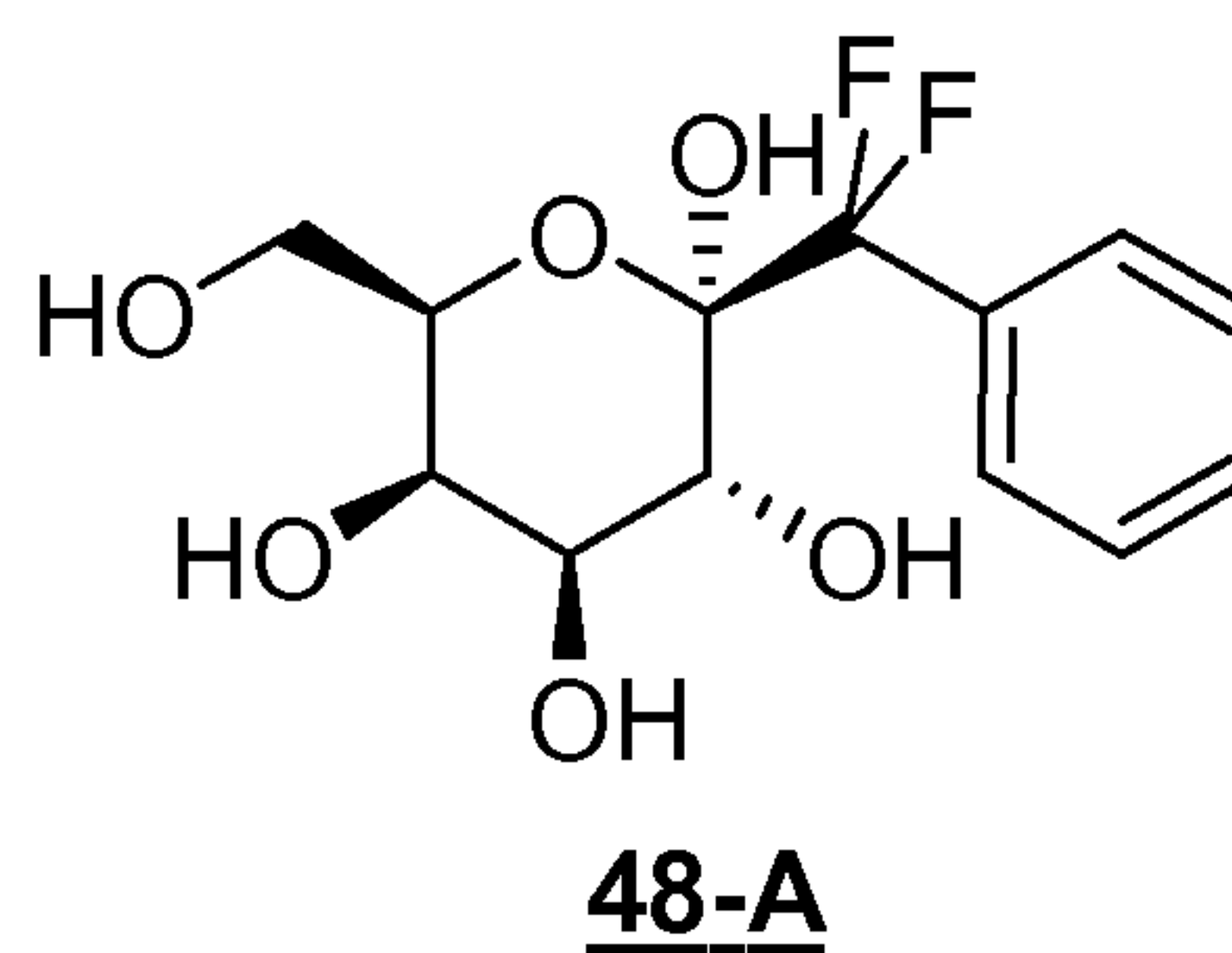
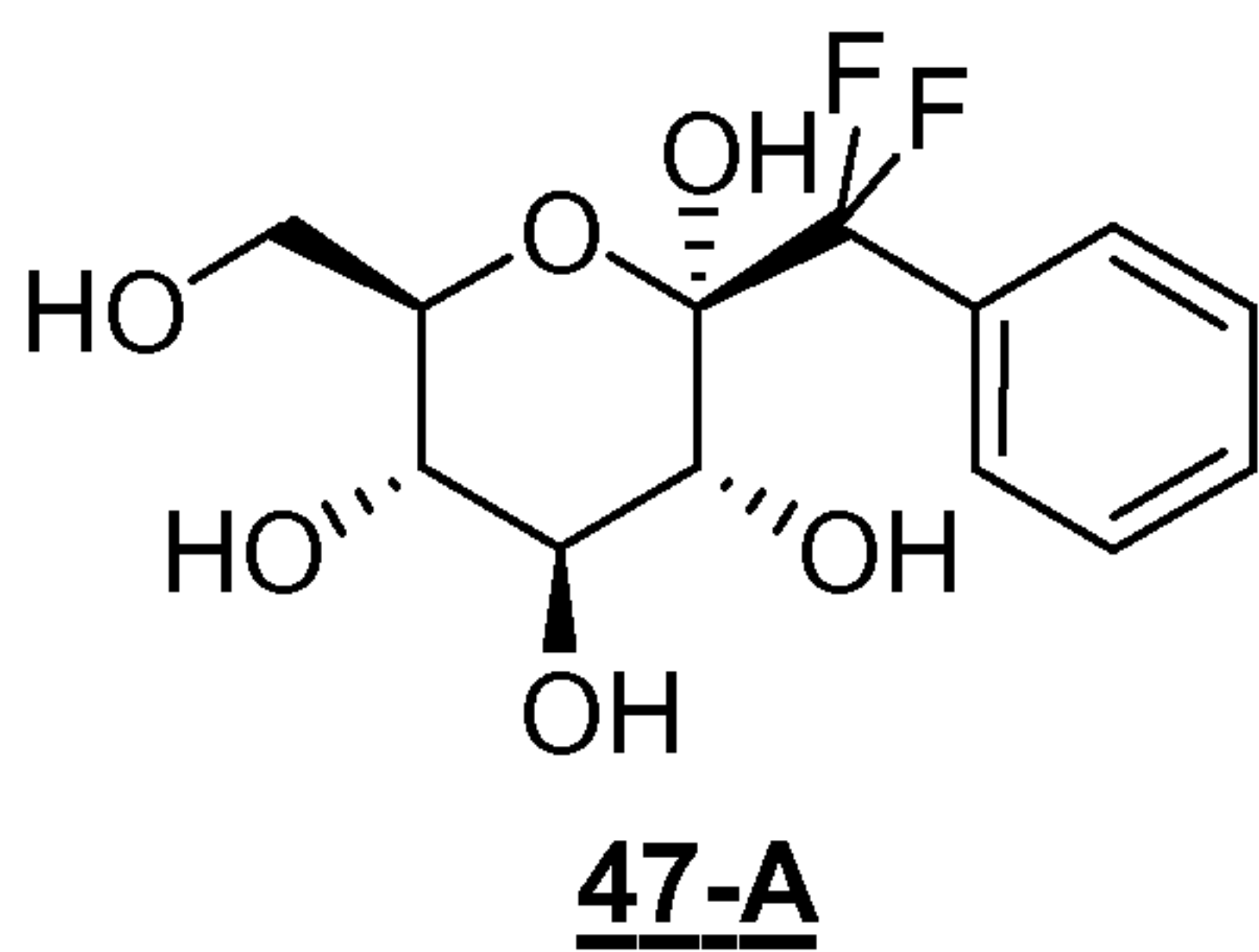
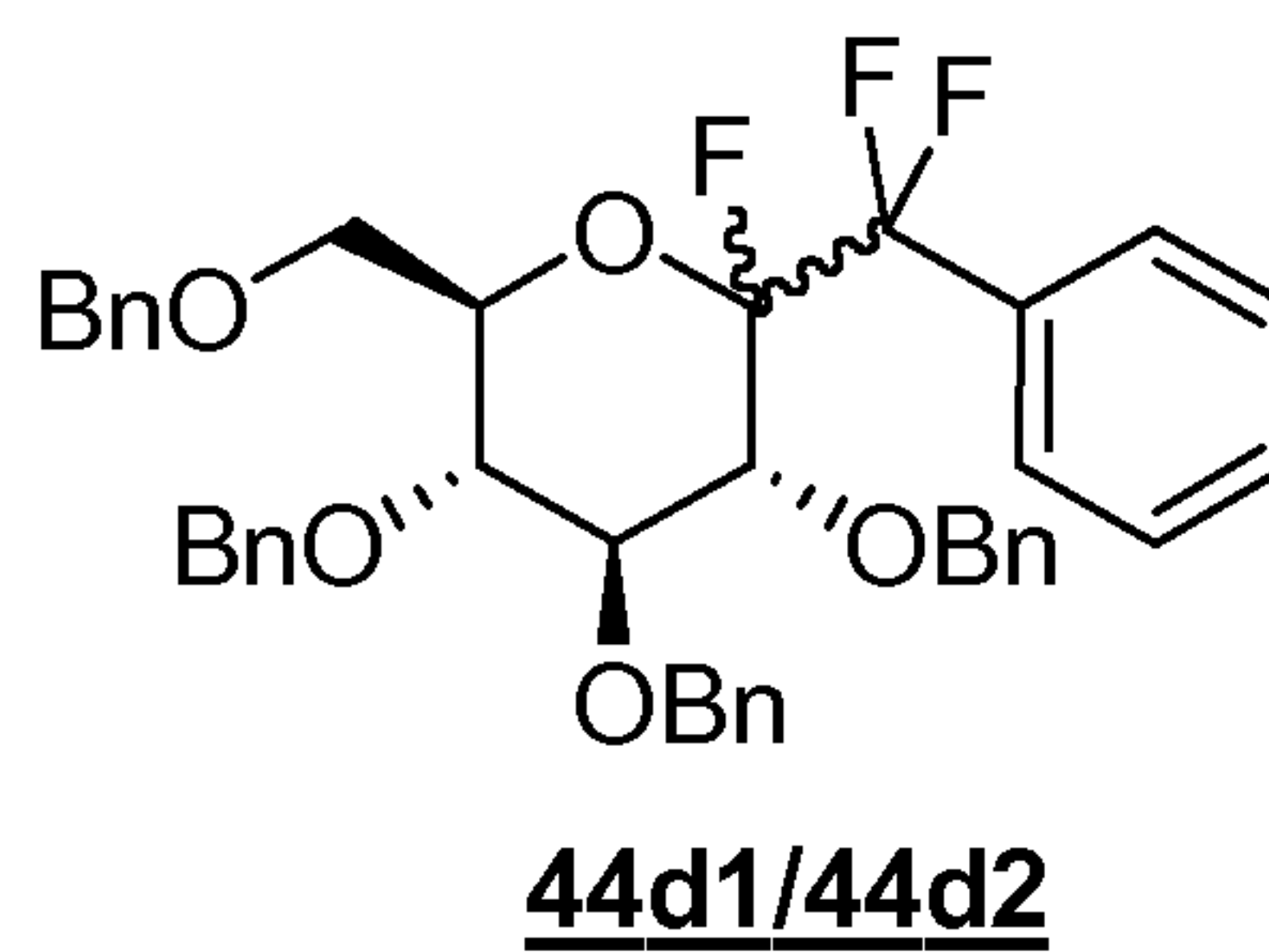
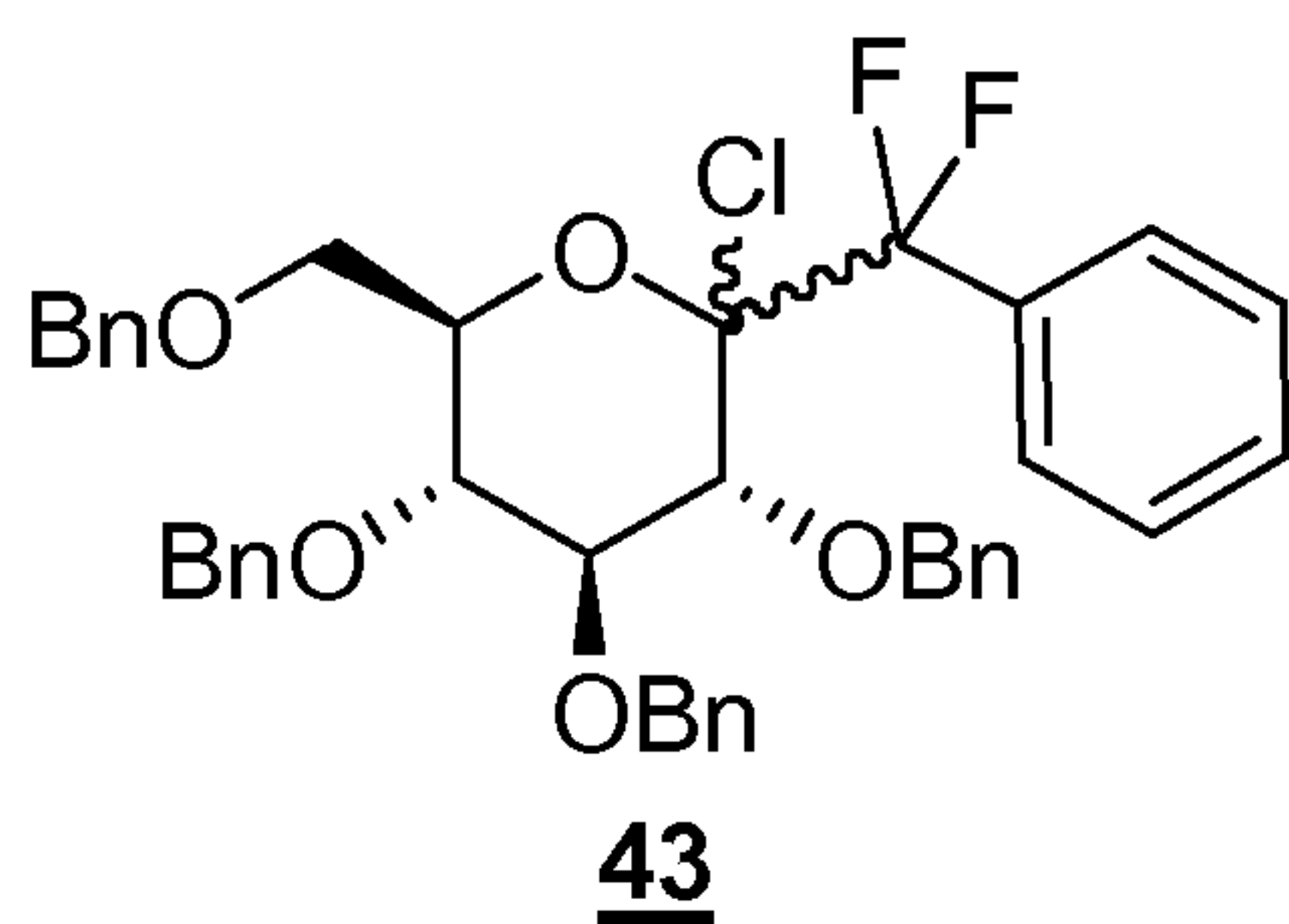
Advantageously, R₅, R₆, R₇, R₈, R₉, R_{5a}, R_{6a}, R_{7a}, R_{8a} and R_{9a} will be chosen, independently from one another, among a hydrogen atom, a halogen atom, advantageously a chlorine or fluorine atom, an aryl-(C₁-C₆)-alkyl, such as benzyl, aryl-
20 (C₁-C₆)-alkyl-O-, such as benzyloxy, or aryl-CO-, such as benzoyl, group, the alkyl moiety of said group being possibly substituted by an OH group and the aryl moiety of said group being possibly substituted by an halogen atom, such as fluorine, an OH, (C₁-C₆)-alkyl or (C₁-C₆)-alkoxy group.

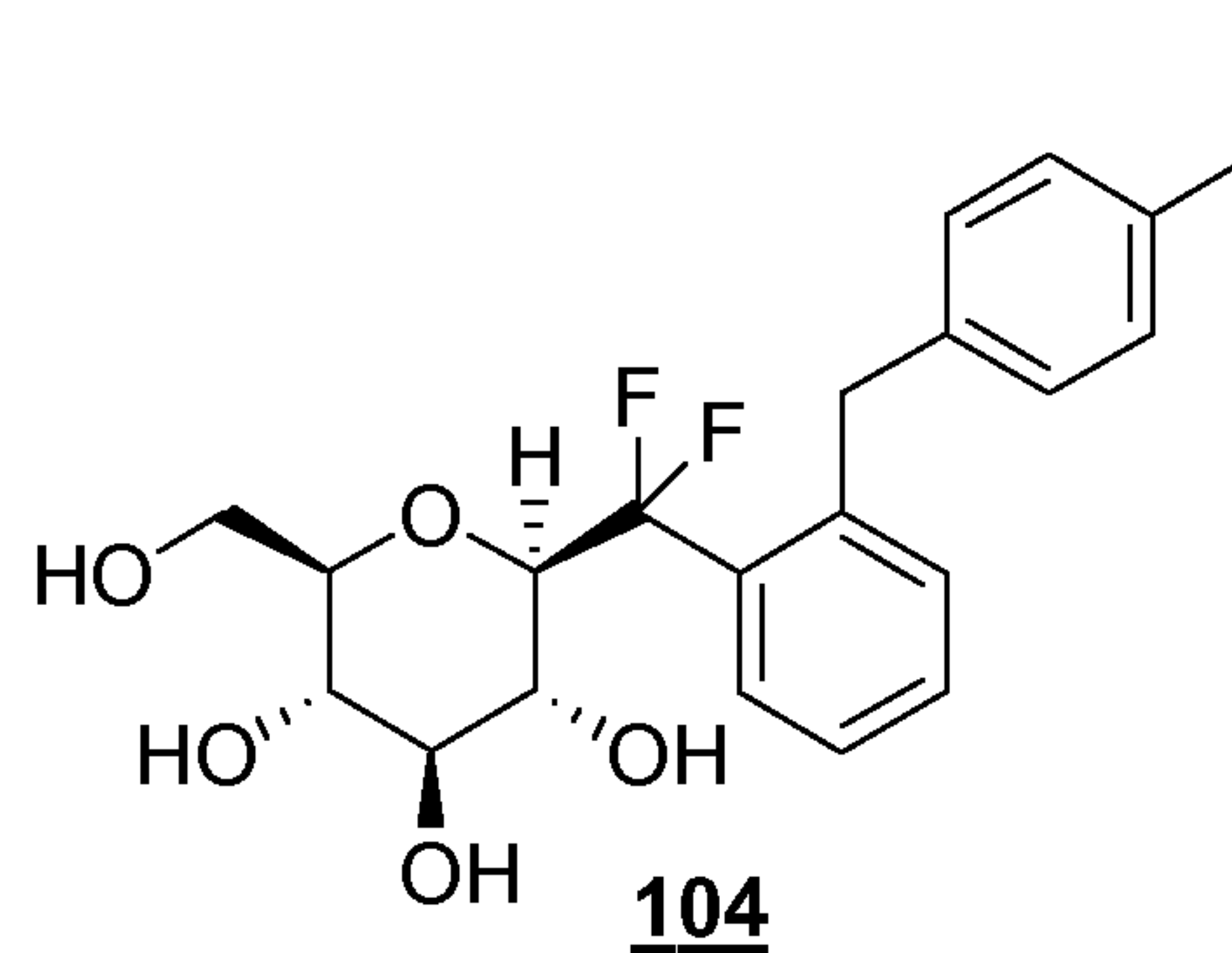
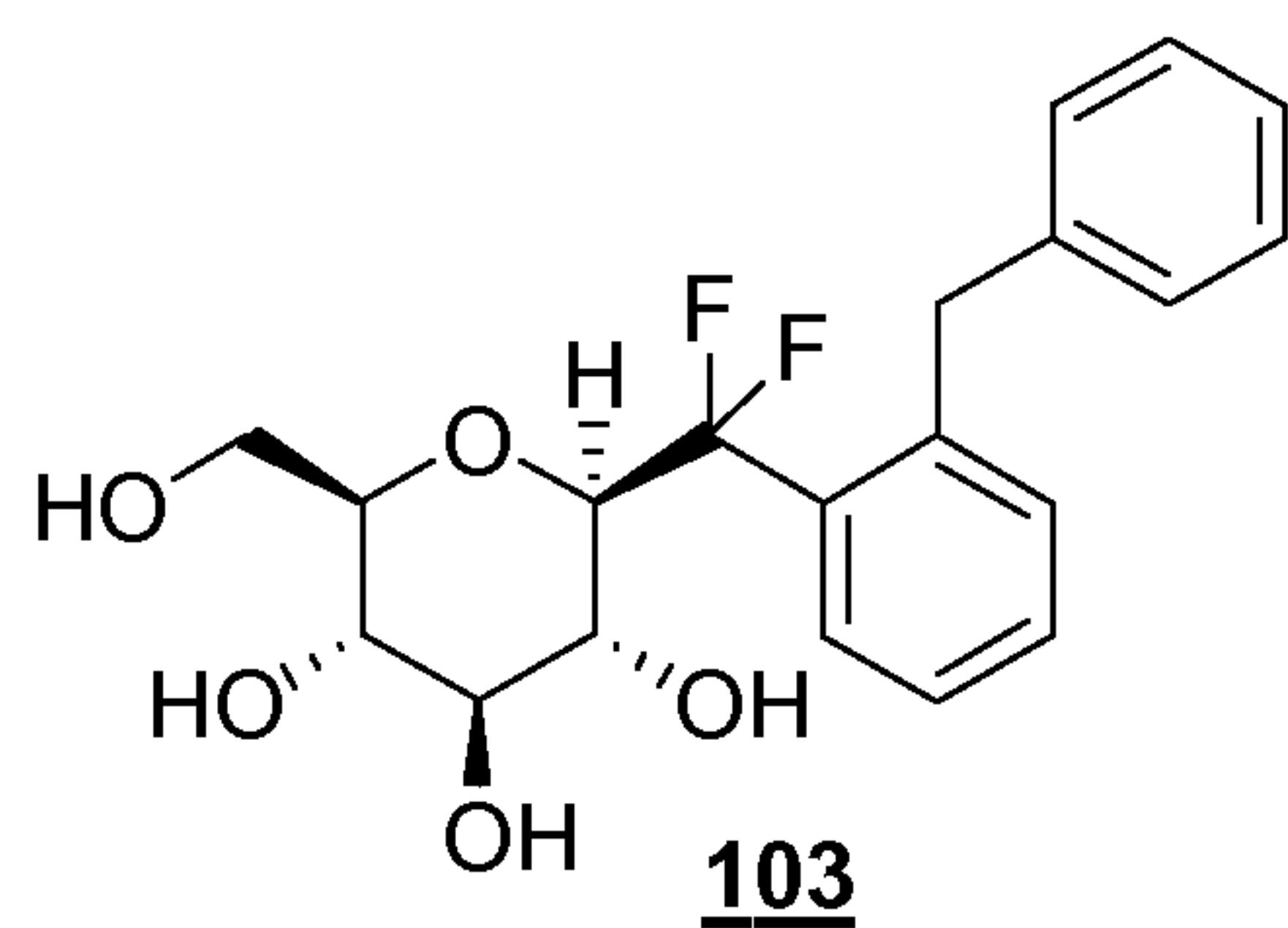
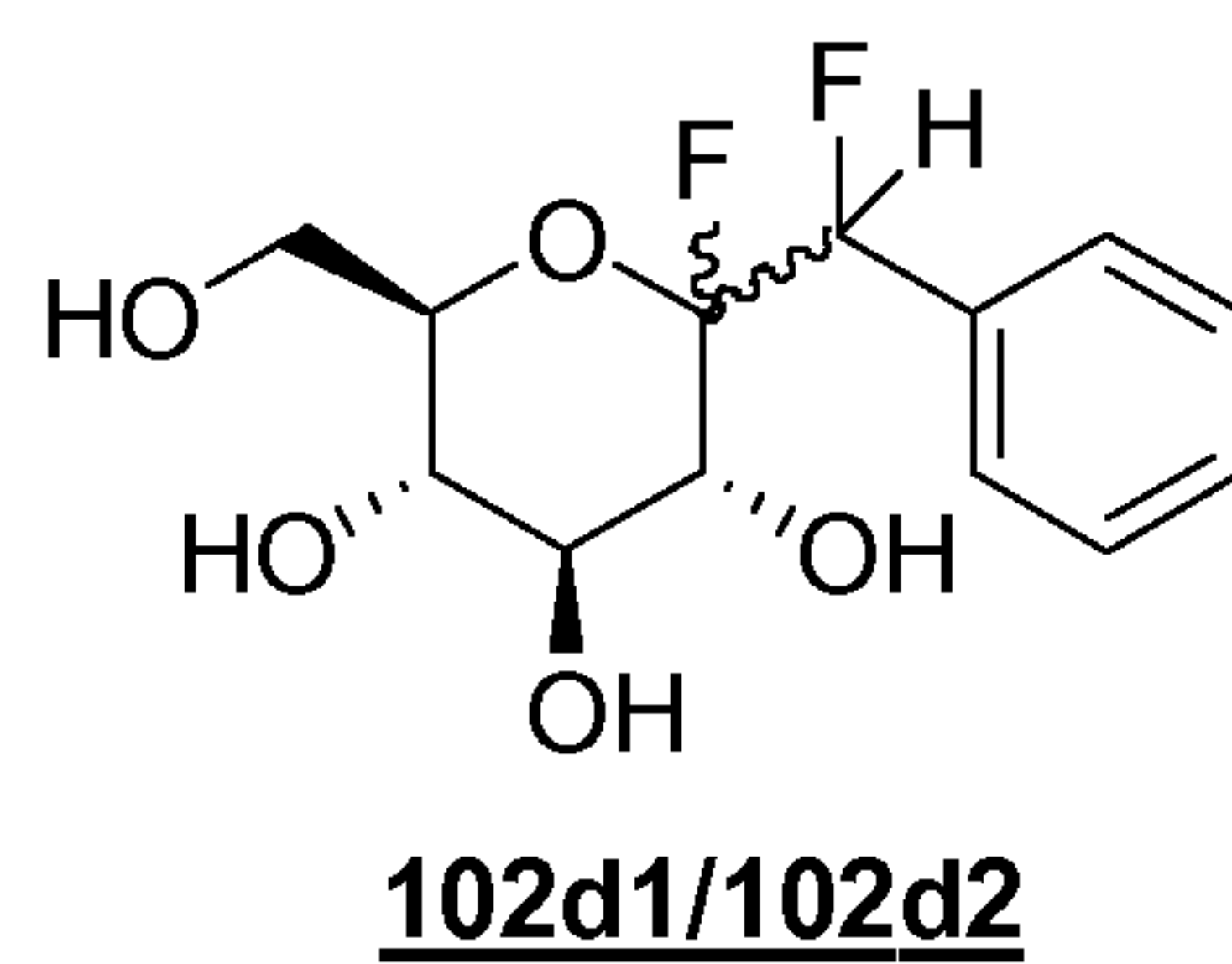
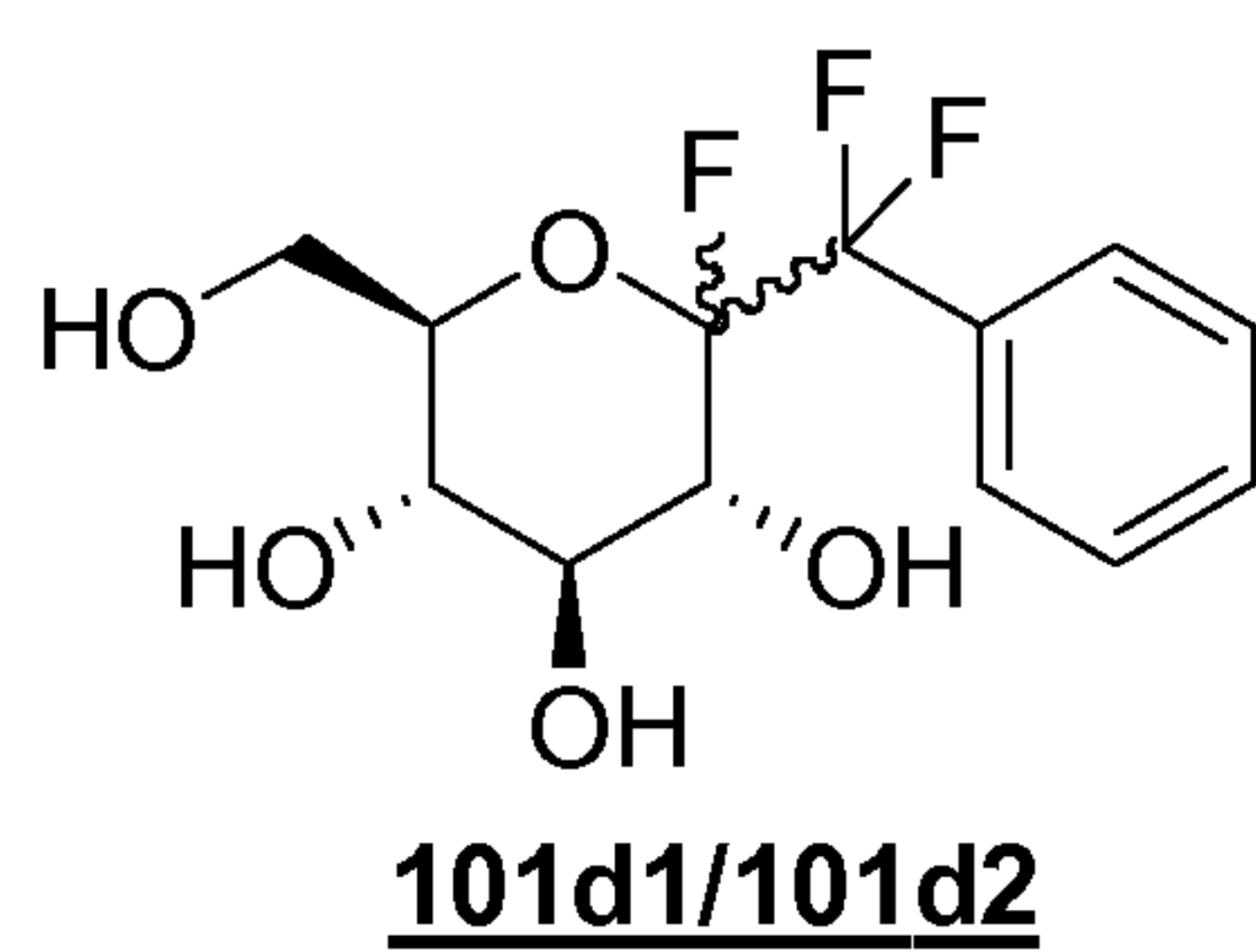
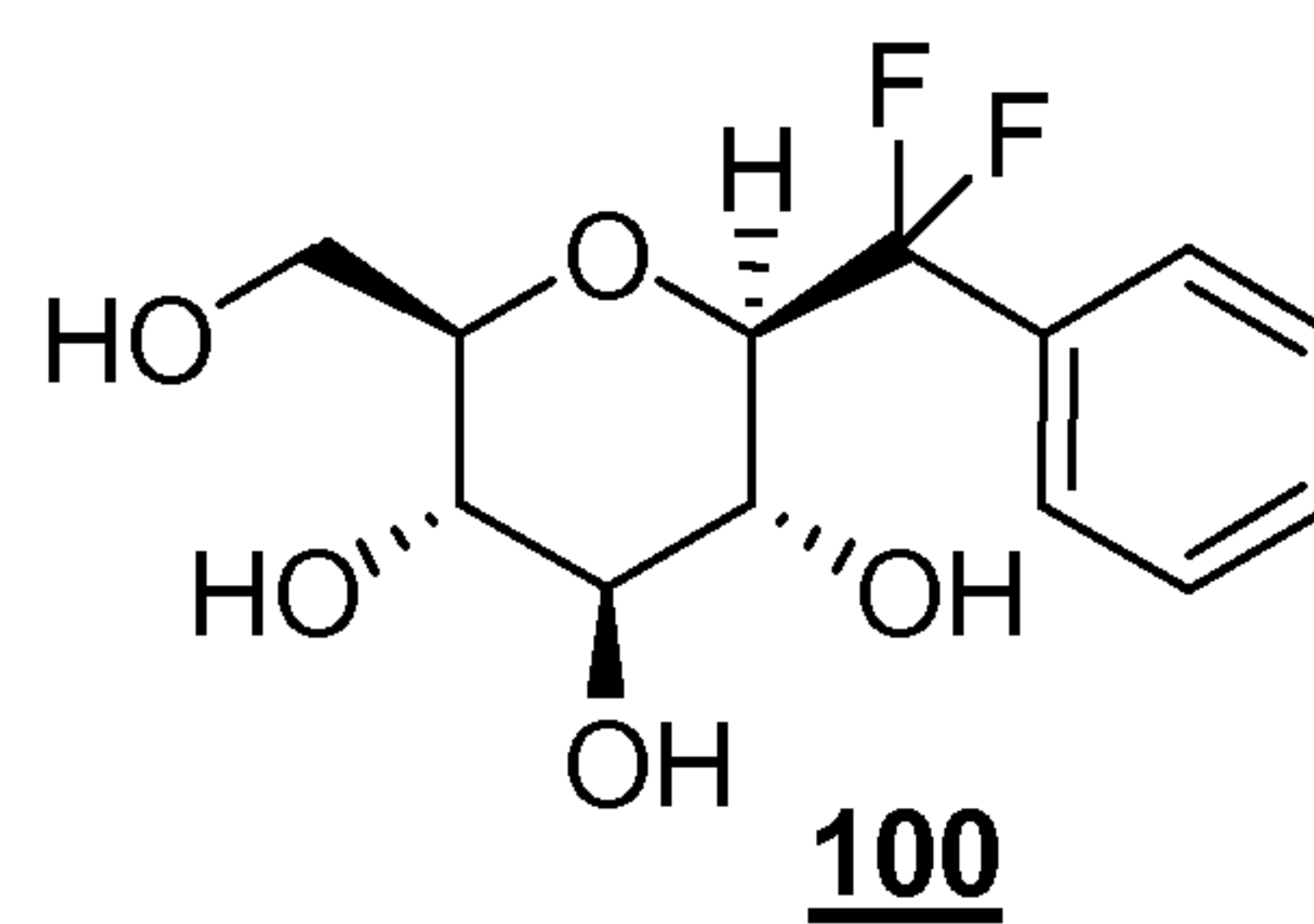
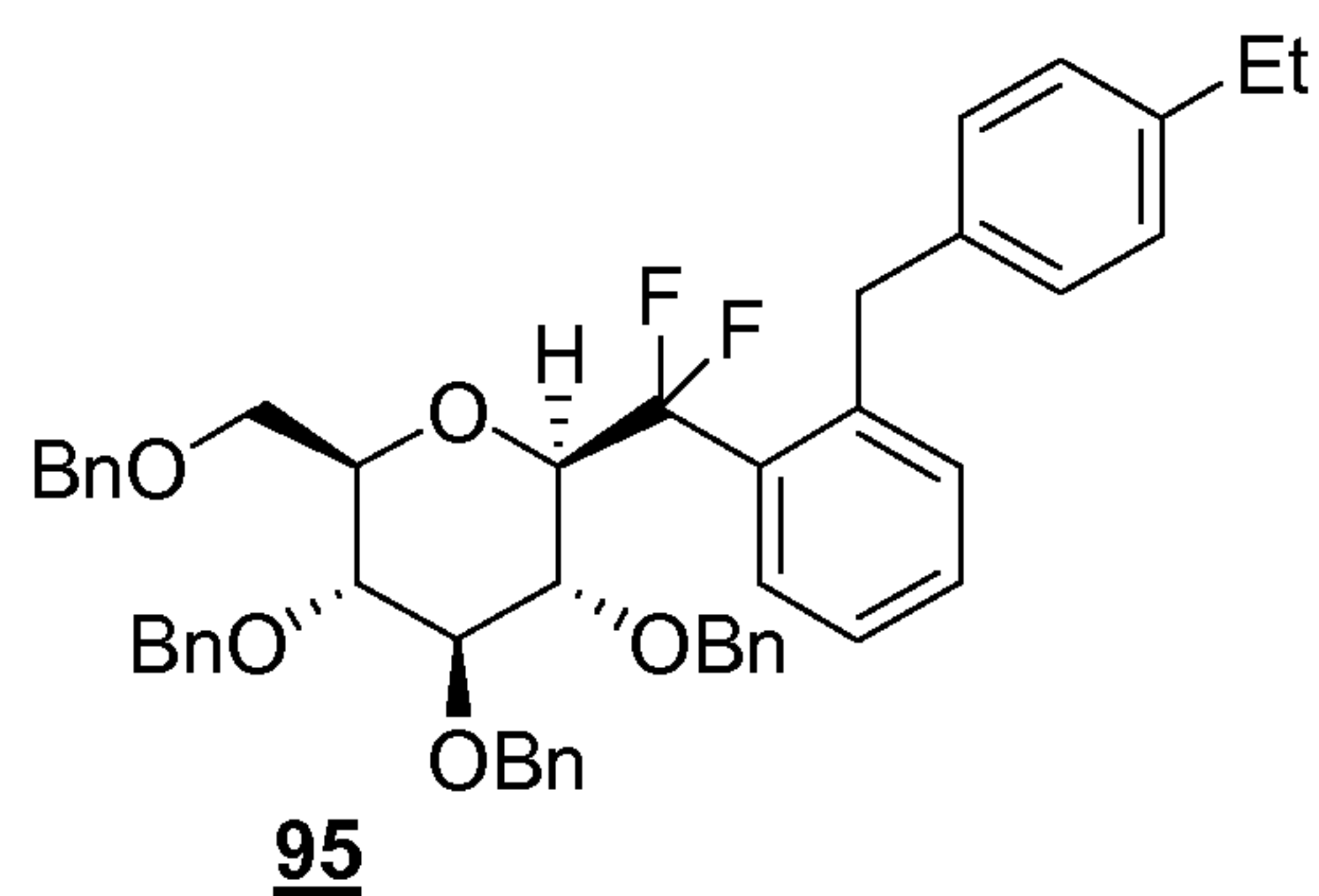
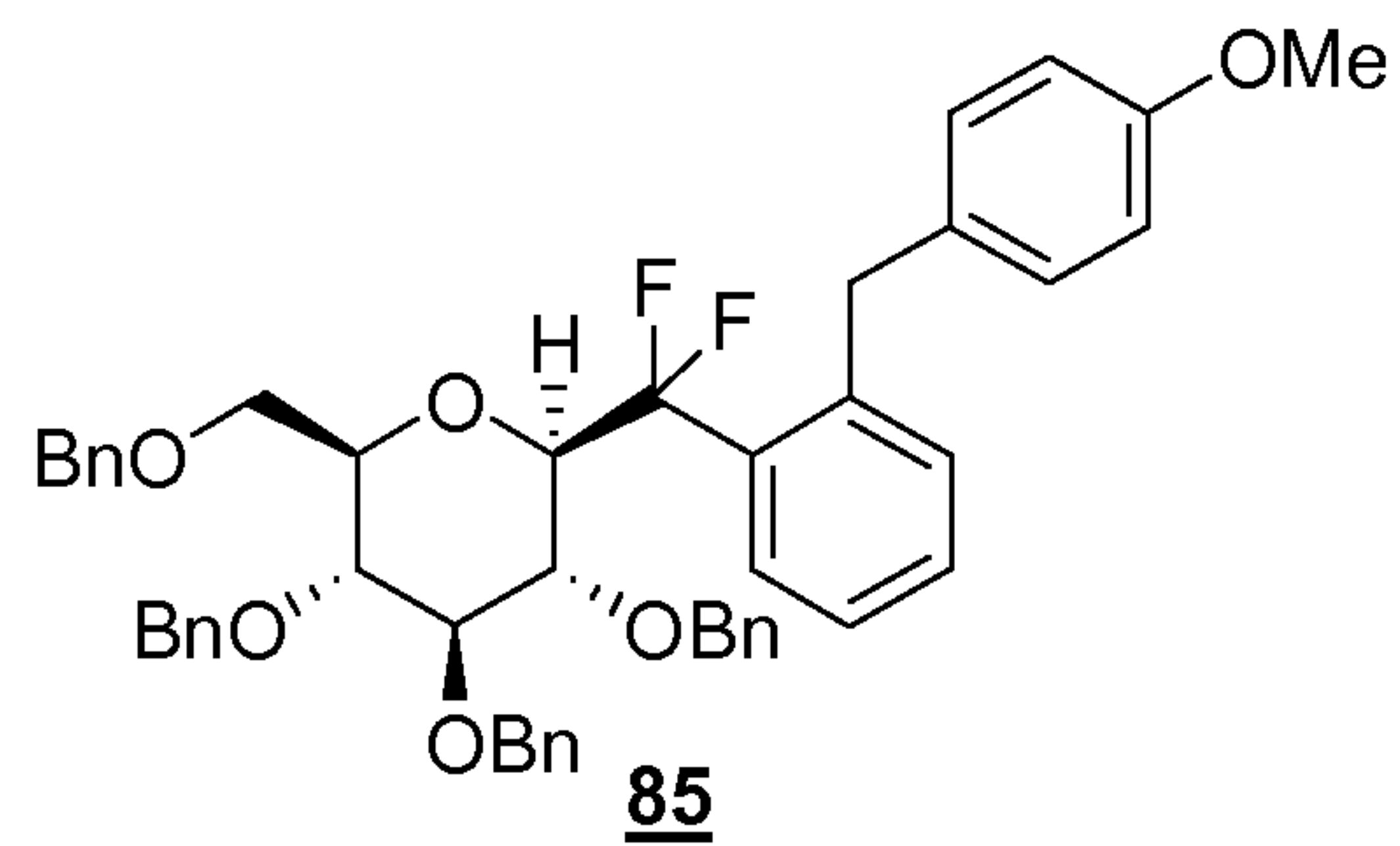
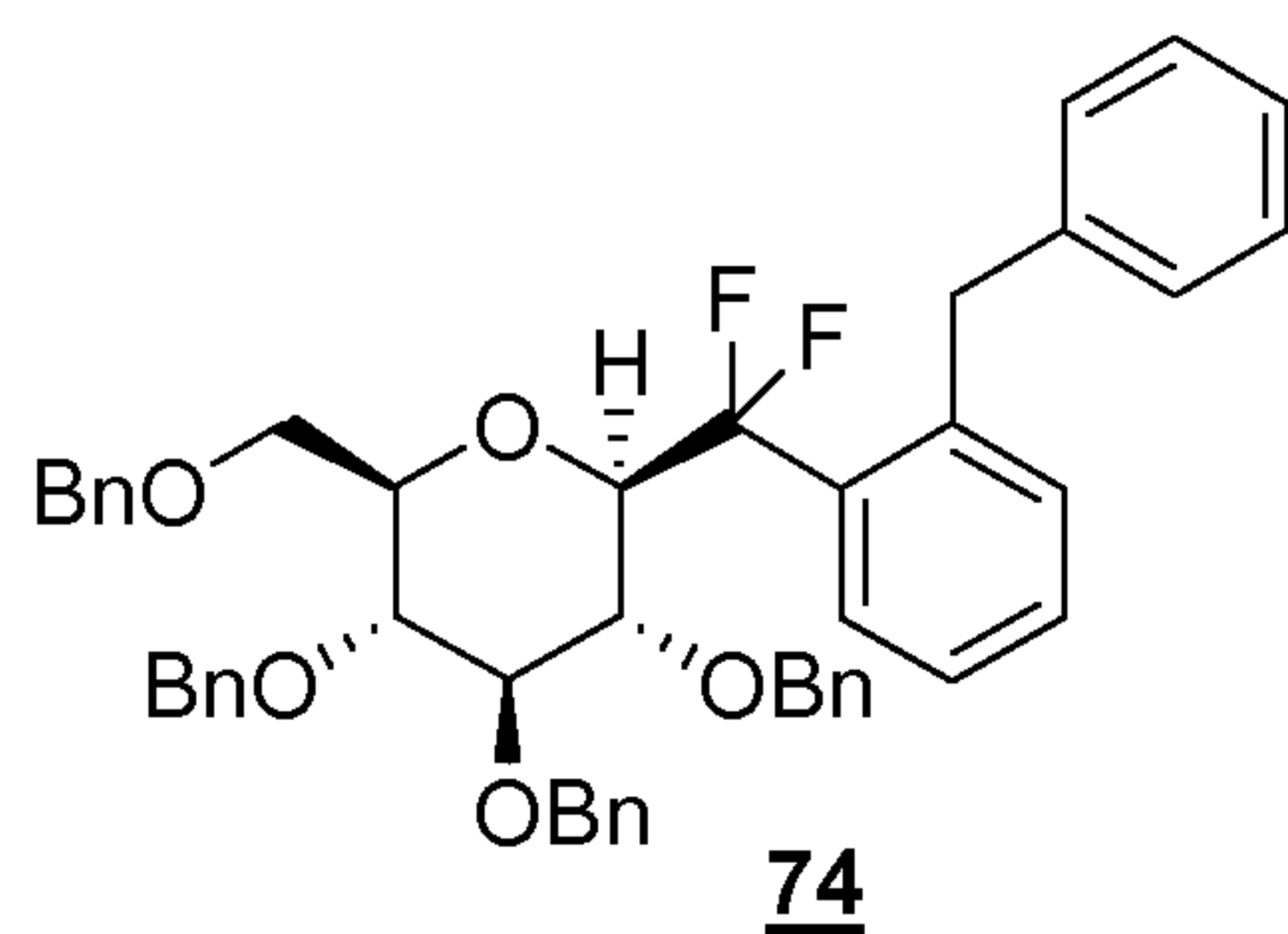
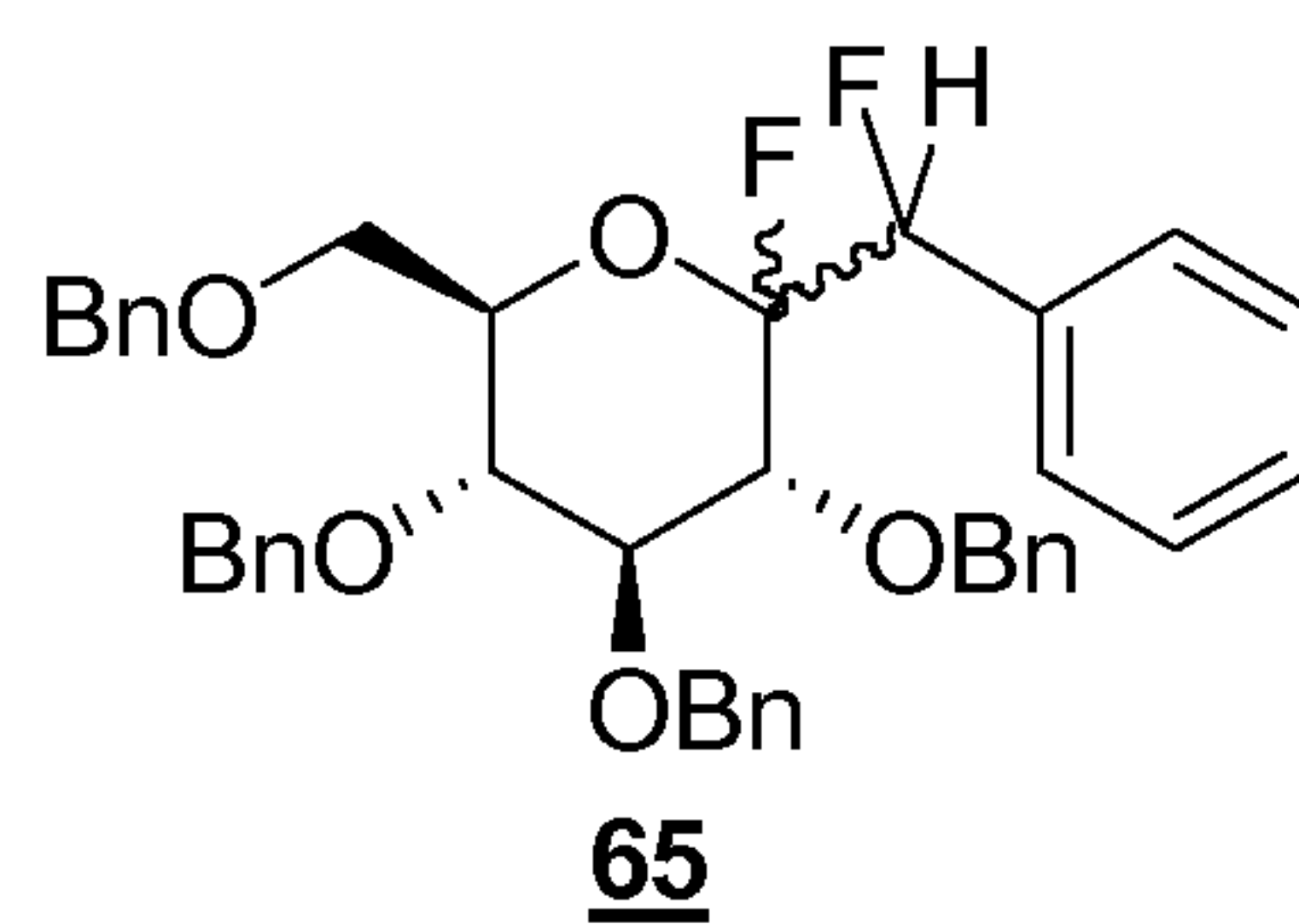
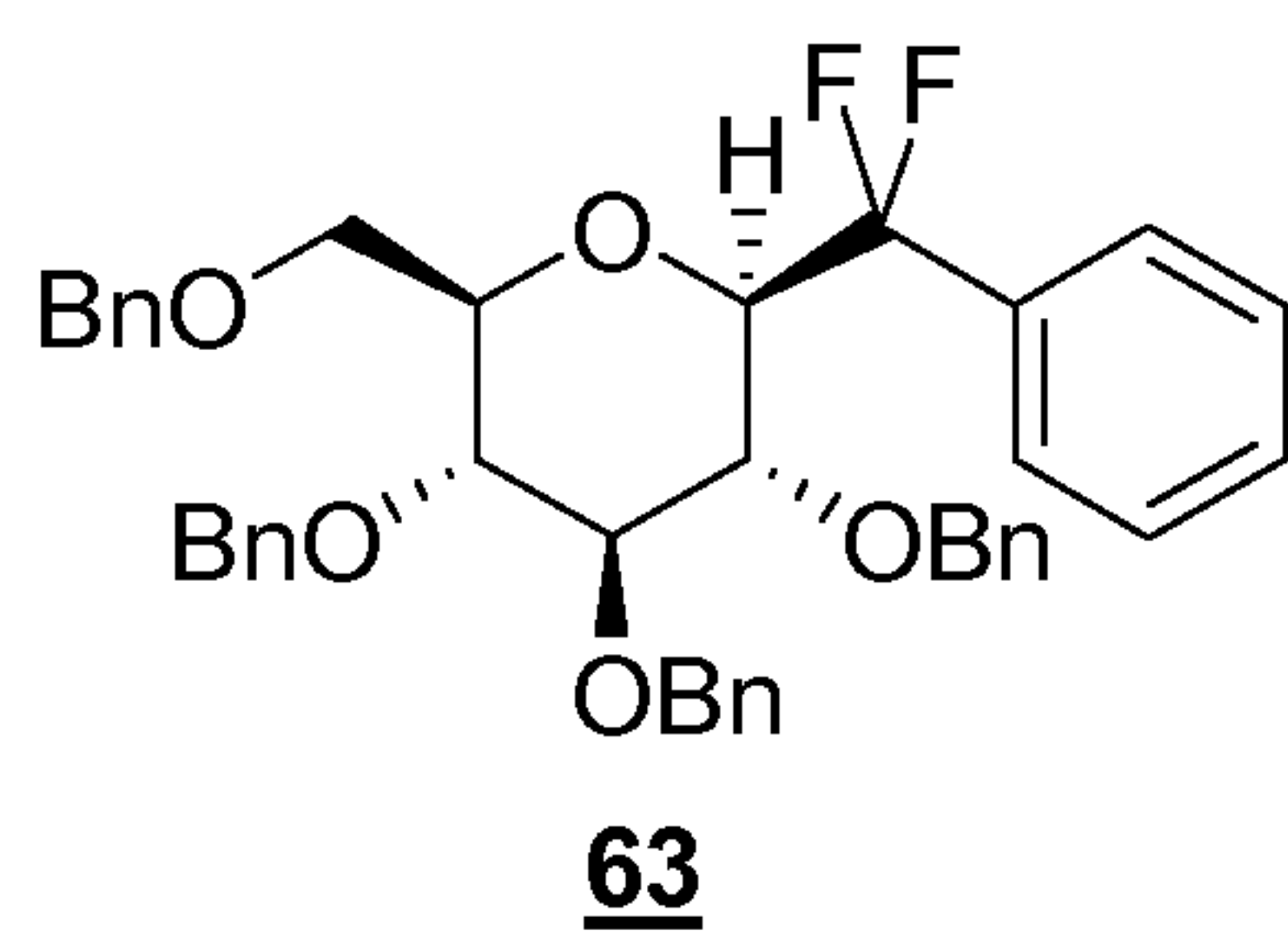
According to a particular embodiment, R₄ represents an NH₂ group.

25 In particular, the compounds of the invention can be chosen among the following molecules:









and

Another object of this invention is a compound as defined above, for use as a drug, in particular as an inhibitor of the sodium-dependent glucose co-transporter, such as SGLT1, SGLT2 and SGLT3.

Within the meaning of this invention, “inhibitor of the sodium-dependent glucose co-transporter” is understood to mean a compound capable of inhibiting partially or totally the sodium-dependent glucose co-transporter.

More particularly, the compounds of the invention may be used for treating diabetes, and more particularly type-II diabetes, diabetes-related complications, such as arteritis of the lower extremities, cardiac infarction, renal insufficiency, neuropathy or blindness, hyperglycemia, hyperinsulinemia, obesity, hypertriglyceridemia, X syndrome and arteriosclerosis.

The compounds of the invention may likewise be used as an anti-cancer, anti-infective, anti-viral, anti-thrombotic or anti-inflammatory drug.

The invention likewise relates to the use of a compound of the invention for the manufacture of a drug intended for the treatment of diabetes, and more particularly type-II diabetes, diabetes-related complications, such as arteritis of the lower extremities, cardiac infarction, renal insufficiency, neuropathy or blindness, hyperglycemia, hyperinsulinemia, obesity, hypertriglyceridemia, X syndrome and arteriosclerosis, as well as for the manufacture of an anti-cancer, anti-infective, anti-viral, anti-thrombotic or anti-inflammatory drug.

The invention likewise relates to a method for a treatment against diabetes, and more particularly type-II diabetes, diabetes-related complications, such as arteritis of the lower extremities, cardiac infarction, renal insufficiency, neuropathy or blindness, hyperglycemia, hyperinsulinemia, obesity, hypertriglyceridemia, X syndrome and arteriosclerosis, as well as for an anti-cancer, anti-infective, anti-viral, anti-thrombotic or anti-inflammatory treatment, including the administration of at least one compound of the invention to a patient in need thereof.

Silylated compounds of the present invention, as well as compounds with $R = CH_2OBn$, $R1 = OBn$, $R2 = OBn$ and/or $R3 = OBn$, will not be preferred for their use as medicament.

Another object of this invention is a pharmaceutical or cosmetic composition including at least one compound of the invention as defined above and at least one pharmaceutically or cosmetically acceptable vehicle.

In this invention, "cosmetically acceptable" is understood to mean what is useful
5 in the preparation of a cosmetic composition which is generally safe, non-toxic and neither biologically nor otherwise undesirable and which is acceptable for veterinary as well as human cosmetic use.

The compounds according to the invention can be administered orally, sublingually, parenterally, subcutaneously, intramuscularly, intravenously,
10 transdermally, locally or rectally.

In the pharmaceutical compounds of this invention, for oral, sublingual, parenteral, subcutaneous, intramuscular, intravenous, transdermal, local or rectal administration, the active ingredient can be administered in unit forms of administration, mixed together with conventional pharmaceutical carriers, for animals or human beings.
15 Suitable unit forms of administration include oral forms such as tablets, gel capsules, powders, granules and oral solutions or suspensions, sublingual or buccal forms of administration, parenteral, subcutaneous, intramuscular, intravenous, intranasal or intraocular forms of administration and rectal forms of administration.

When a solid composition is prepared in the form of tablets, the principal active
20 ingredient is mixed with a pharmaceutical vehicle such as gelatine, starch, lactose, magnesium stearate, talc, gum arabic or the like. The tablets can be coated with sucrose or other suitable materials or else treated in such a way that they have an extended or delayed activity and continuously release a predetermined amount of active principle.

A gel capsule preparation is obtained by mixing the active ingredient with a
25 diluent and by pouring the mixture obtained into soft or hard capsules.

A preparation in the form of a syrup or elixir can contain the active ingredient in conjunction with a sweetening agent, antiseptic, as well as a flavour-producing agent and appropriate colouring agent.

Powders or granules dispersible in water can contain the active ingredient mixed
30 together with dispersing agents, wetting agents, or suspending agents, as well as with taste correctors or sweetening agents.

For rectal administration, suppositories are used, which are prepared with binding agents melting at rectal temperature, e.g., cocoa butter or polyethylene glycols.

For parenteral, intranasal or intraocular administration, aqueous suspensions are used, isotonic saline solutions or sterile and injectable solutions, which contain
5 pharmacologically compatible dispersing agents and/or wetting agents.

The active principle can also be formulated as microcapsules, possibly with one or more additive carriers.

The compounds of the invention can be used at doses of between 0.01 mg and 1000 mg per day, given in a single dose once a day or administered in several doses
10 throughout the day, e.g., twice daily in equal doses. The daily dose administered is advantageously between 5 mg and 500 mg, even more advantageously between 10 mg and 200 mg. It may be necessary to use doses exceeding these ranges, of which those skilled in the art will themselves be aware.

In one particular embodiment of the invention, the pharmaceutical or cosmetic
15 composition can also be formulated for topical administration. It may be introduced in forms commonly known for this type of administration, i.e., in particular, lotions, foams, gels, dispersions, sprays, shampoos, serums, masks, body milks or creams, for example, with excipients enabling, in particular, penetration of the skin so as to improve the properties and accessibility of the active principle. Besides the composition according to
20 the invention, these compositions generally further contain a physiologically acceptable medium, which generally contains water or a solvent, e.g., alcohols, ethers or glycols. They can also contain surface-active agents, preservatives, stabilizers, emulsifiers, thickeners, other active principles producing a complementary or possibly synergic effect, trace elements, essential oils, perfumes, colouring agents, collagen, chemical or
25 mineral filters, hydrating agents or thermal waters.

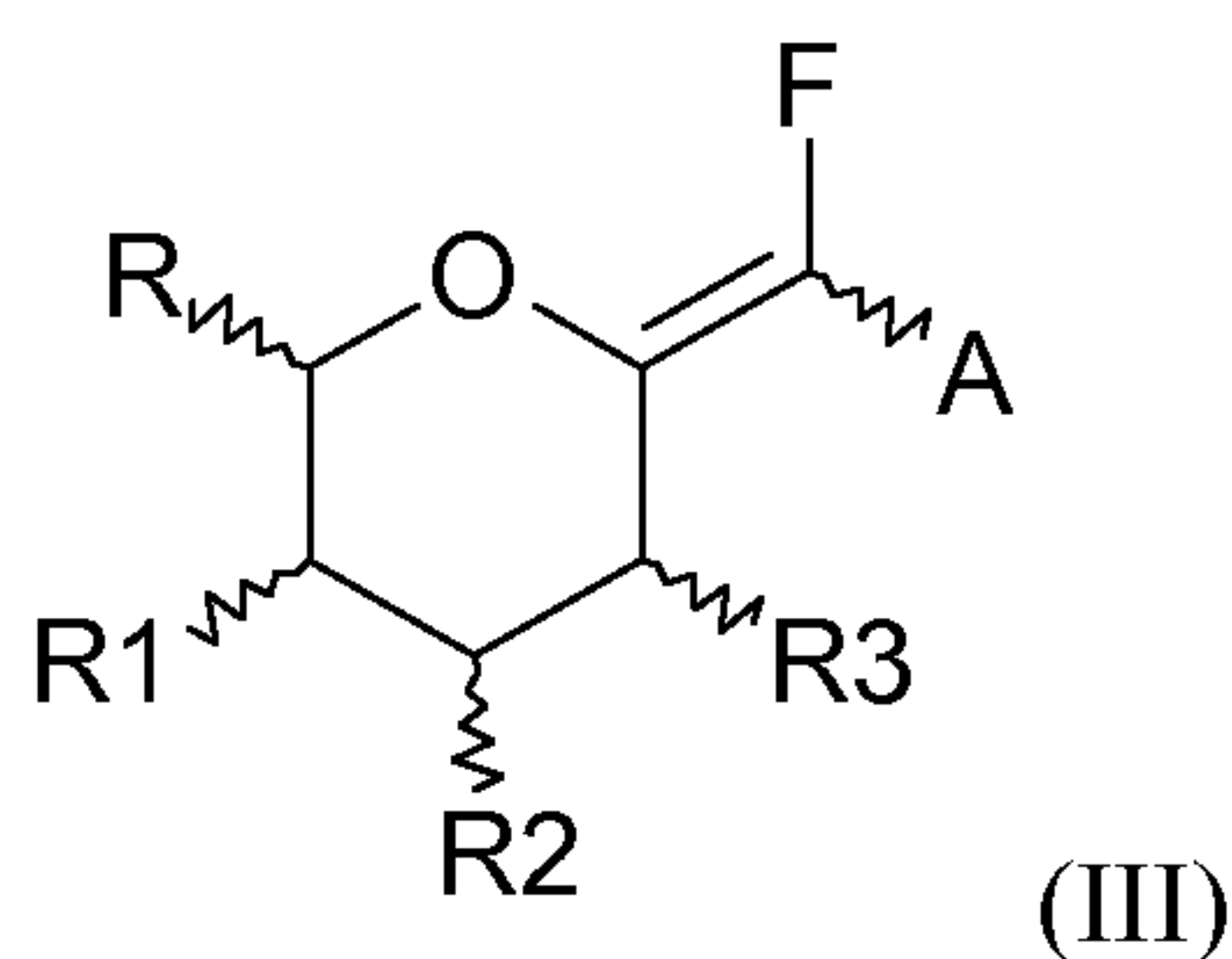
In one particular embodiment, the pharmaceutical composition of the invention may include at least one other active principle, in addition to the compound of the invention.

Examples of active principles that can be cited are antidiabetic agents, such as
30 sulfonylurea-type compounds which are hypoglycemic sulfamides which increase insulin secretion like, e.g., chlorpropamide, tolbutamide, tolazamide, glipizide, gliclazide, glibenclamide, gliquidone and glimepiride, biguanides which reduce the

hepatic glyconeogenesis and the insulin resistance like metformine, thiazolidinediones (also called glitazones) which increase the sensibility to insulin like rosiglitazone, pioglitazone and ciglitazone, alpha-glucosidases inhibitors which slow down the intestinal absorption of carbohydrates like acarbose, miglitol and voglibose, 5 meglitinides (also called glitinides) which increase insulin pancreatic secretion like repaglinide and nateglinide, incretin mimics like exenatide or dipeptidylpeptidase-4 (DPP4) inhibitors like sitagliptin, vildagliptin and insulin, or antilipidic agents, such as statins which reduce cholesterol by inhibiting the enzyme HMG-CoA reductase like atorvastatin and cerivastatin, fibrates like bezafibrate, gemfibrozil and fenofibrate, or 10 ezetimibe.

Another object of this invention is the cosmetic use of a compound of the invention as defined above, for lightening, bleaching, depigmenting the skin, removing blemishes from the skin, particularly age spots and freckles, or preventing pigmentation of the skin, via topical application in particular.

15 Another object of this invention is a process for preparing a compound of generic formula (Ia), as defined above, wherein X and R4 represent a hydrogen atom, characterized in that the compound of formula (Ia) is obtained by hydrogenation of the double bond of a compound of generic formula (III):



20 wherein A, R, R1, R2 and R3 are as defined above.

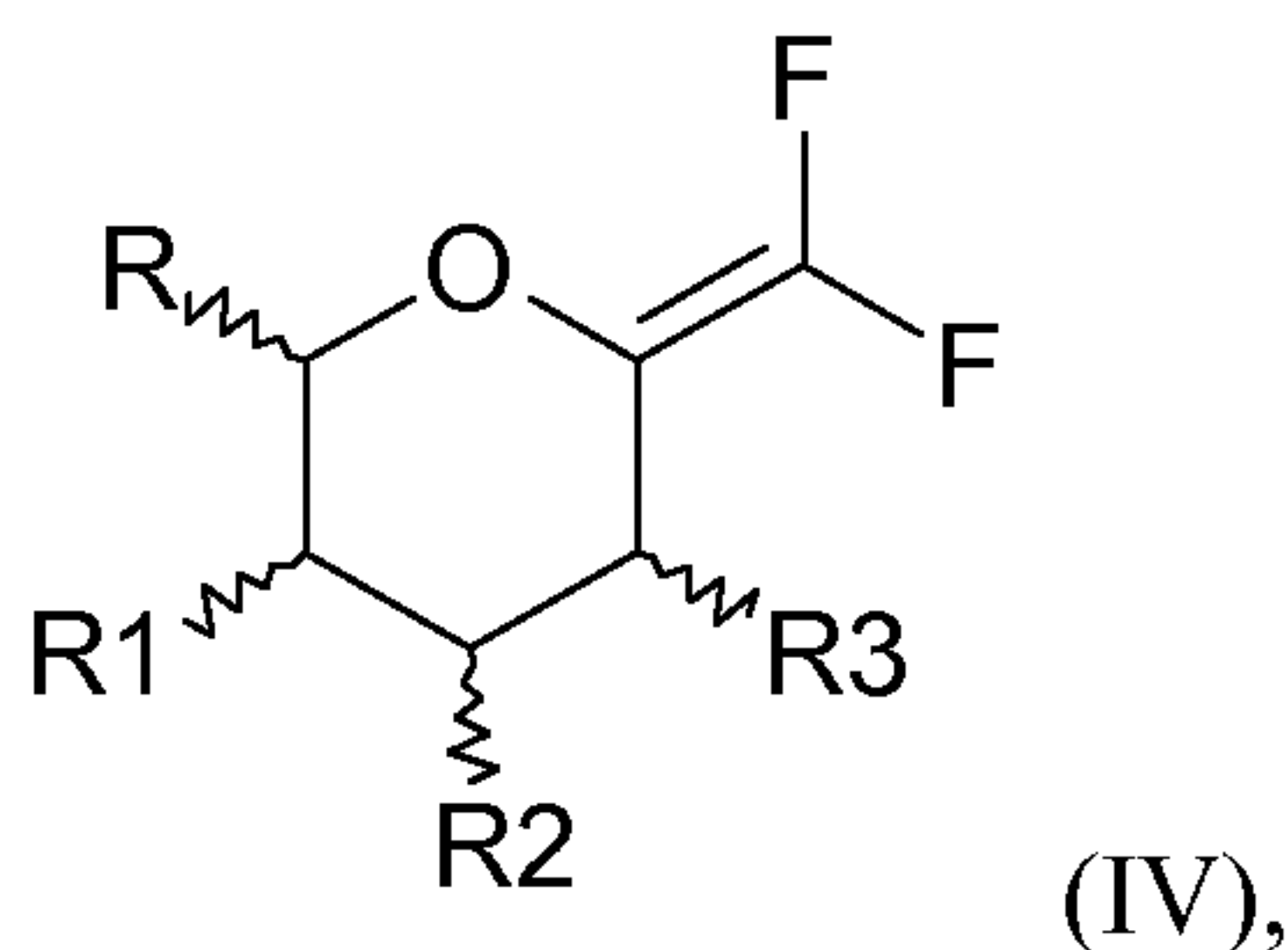
This hydrogenation occurs under a hydrogen atmosphere, in particular in the presence of palladium on carbon Pd/C.

25 According to a first alternative, the compound of generic formula (I) defined above can be obtained according to the following steps:

(a1) halogen-metal exchange between a compound of generic formula A-Hal, wherein A is as defined above and Hal represents an halogen atom, advantageously

bromine or chlorine, and a (C₁-C₆)-alkyl lithium, a (C₁-C₆)-alkyl magnesium halide or a di-(C₁-C₆)-alkyl magnesium, and

(b1) reaction of the compound obtained at the preceding step (a1) with a compound of generic formula (IV):



wherein R, R1, R2 and R3 are as defined above,
in order to obtain the compound of formula (III).

Preferably, the halogen-metal exchange of step (a1) is carried out with a (C₁-C₆)-alkyl lithium.

10 Advantageously, the (C₁-C₆)-alkyl lithium derivative will be n-butyllithium, sec-butyllithium or tert-butyllithium.

The halogen-metal exchange can be also carried out with a (C₁-C₆)-alkyl magnesium halide, preferably bromide or chloride, (Grignard reagent) or with a di-(C₁-C₆)-alkyl magnesium in place of the (C₁-C₆)-alkyl lithium, possibly in the presence of
15 lithium chloride LiCl, in order to accelerate the metalation process. The Grignard reagent is advantageously isopropylmagnesium or sec-butylmagnesium bromide or chloride, and the dialkyl magnesium is advantageously diisopropylmagnesium or di-sec-butylmagnesium.

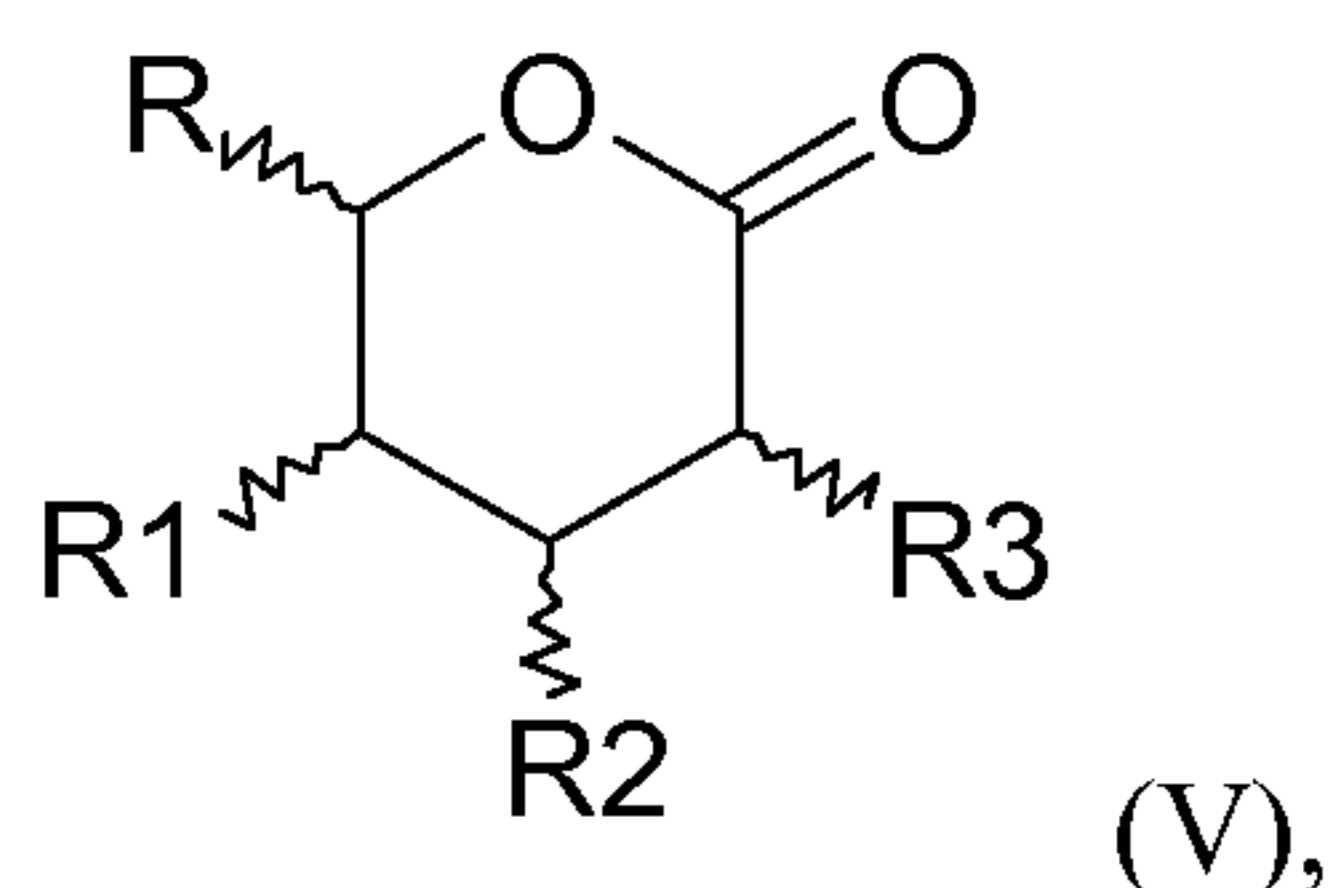
The halogen-metal exchange reactions are preferably conducted at temperatures
20 varying from -100°C to 40°C, advantageously in an inert solvent or solvent mixture, e.g., such as diethylether, dioxane, tetrahydrofurane, toluene, hexane, dimethylsulfoxide, dichloromethane.

The lithium or magnesium compounds obtained via halogen-metal exchange may be possibly transmetalated with metal salts such as cerium trichloride (CeCl₃), zinc
25 chloride or bromide (ZnCl₂, ZnBr₂), indium chloride or bromide (InCl₃, InBr₃) in order to form other organometallic compounds usable in the reaction of step (b1).

Alternatively, the halogen-metal exchange step (a1) could be replaced by a step for inserting a metal into the carbon-halogen bond of the halogen derivative A-Hal.

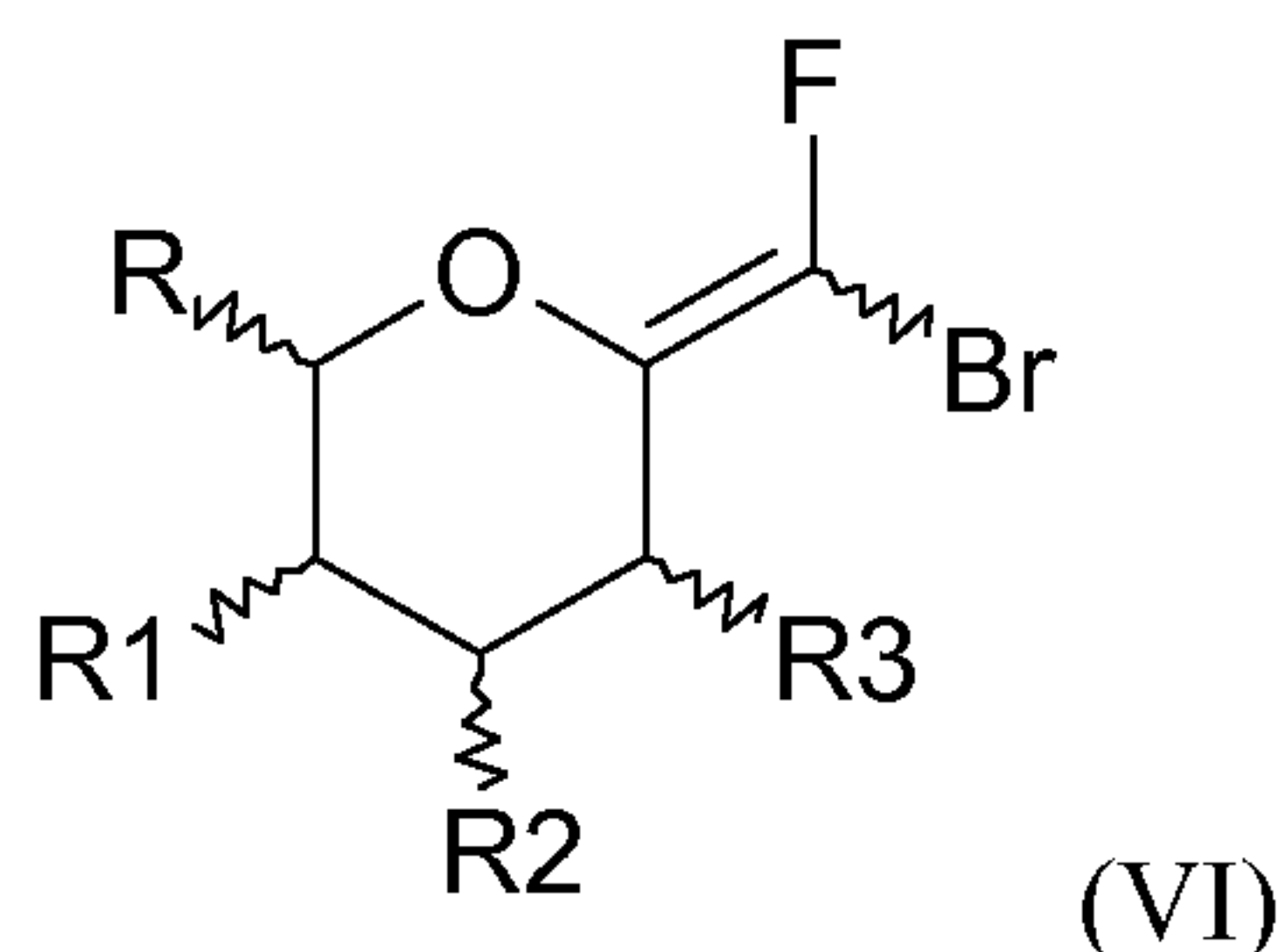
Lithium and magnesium are two metals that can be used for this type of reaction. The insertion can be performed in an inert solvent or solvent mixture, e.g., such as diethylether, dioxane, tetrahydrofurane, toluene, hexane, dimethylsulfoxide, advantageously at a temperature varying from -80°C to 100°C. In the case where no spontaneous reaction occurs, activation of the metal may be necessary, e.g., by treating with 1,2-dibromoethane, iodine, trimethylsilyl chloride, acetic acid, hydrochloric acid and/or via sonication. The addition of the organometallic compound thus obtained to the compound of formula (IV) (corresponding to step (b1)) is advantageously carried out at temperatures varying between -100°C to 60°C, advantageously in an inert solvent or solvent mixture, such as diethylether, dimethoxyethane, benzene, toluene, methylene chloride, hexane, tetrahydrofurane, dioxane, N-methylpyrrolidinone. These reactions can be conducted in air although an inert atmosphere is preferred, such as a nitrogen or argon atmosphere.

As far as the compound of generic formula (IV) is concerned, it can be obtained according to a process described in literature, in particular by a reaction on a lactone of formula (V) derived from a sugar as defined below:



wherein R, R1, R2 and R3 are as defined above, in the presence of dibromodifluoromethane CF_2Br_2 , hexamethylphosphotriamide HMPT, and possibly zinc, in a solvent such as tetrahydrofurane (Journal of the Chemical Society, Chemical Communications (1989), 19, 1437-1439; Tetrahedron (1993), 49 (36), 8087-8106; Angewandte Chemie, International Edition (2004), 43 (48), 6680-6683) or according to a procedure as described in J. of. Fluorine Chemistry (2006), 127 (4-5), 637-642).

According to a second alternative, the compound of generic formula (III) is obtained by reacting a compound of formula A-B(OH)_2 or $\text{A-SnR}'_3$, wherein A is as defined above and R' represents $(\text{C}_1\text{-C}_6)\text{-alkyl}$, with a compound having the following generic formula (VI):



wherein R, R1, R2, and R3 are as defined above, in the presence of a palladium catalyst and a base.

The above-described reaction thus consists of a coupling reaction (Suzuki
5 reaction or Stille reaction) between an organoboronic acid ($A-B(OH)_2$) or a stannylated derivative ($A-SnR'_3$) and a halogenated derivative (V) in the presence of a catalyst and a base.

Among the examples of bases used, in particular but not exclusively, are sodium or potassium carbonate and sodium or potassium hydroxide.

10 Among the examples of catalysts, any palladium catalyst that can be used for Suzuki coupling (in the case of an $A-B(OH)_2$ compound), or for Stille coupling (in the case of an $A-SnR'_3$ compound) can be used as tetrakis(triphenylphosphin)palladium $Pd(PPh_3)_4$, palladium (II) acetate $Pd(OAc)_2$, $Pd(PPh_3)_2Cl_2$, $Pd_2(dba)_3$, $PdCl_2(dppf)$ or $PdCl_2(dppe)$, with “dba” meaning dibenzylideneacetone, “dppf” meaning 1,1'-
15 bis(diphenylphosphino)ferrocene and “dppe” meaning 1,2-bis(diphenylphosphino)ethane. Advantageously, the palladium catalyst is tetrakis(triphenylphosphin)palladium $Pd(PPh_3)_4$, palladium (II) acetate $Pd(OAc)_2$ or $Pd(PPh_3)_2Cl_2$.

20 Among the reaction solvents that can be used, in particular but not exclusively, are tetrahydrofuran (THF), N,N-dimethylformamide (DMF), dimethylsulfoxide (DMSO), toluene, alcohol such as ethanol and water, as well as mixtures of solvents.

The coupling reaction can be carried out at a temperature varying from ambient temperature to 120°C.

By “ambient temperature,” it is understood to mean a temperature varying between 20°C and 35°C, and preferably of around 25°C.

25 The organoboronic acid ($A-B(OH)_2$) could be replaced by an organoborane, such as A-9-BBN (9-BBN corresponds to 9-borabicyclo[3.3.1]nonane) or a boronic ester.

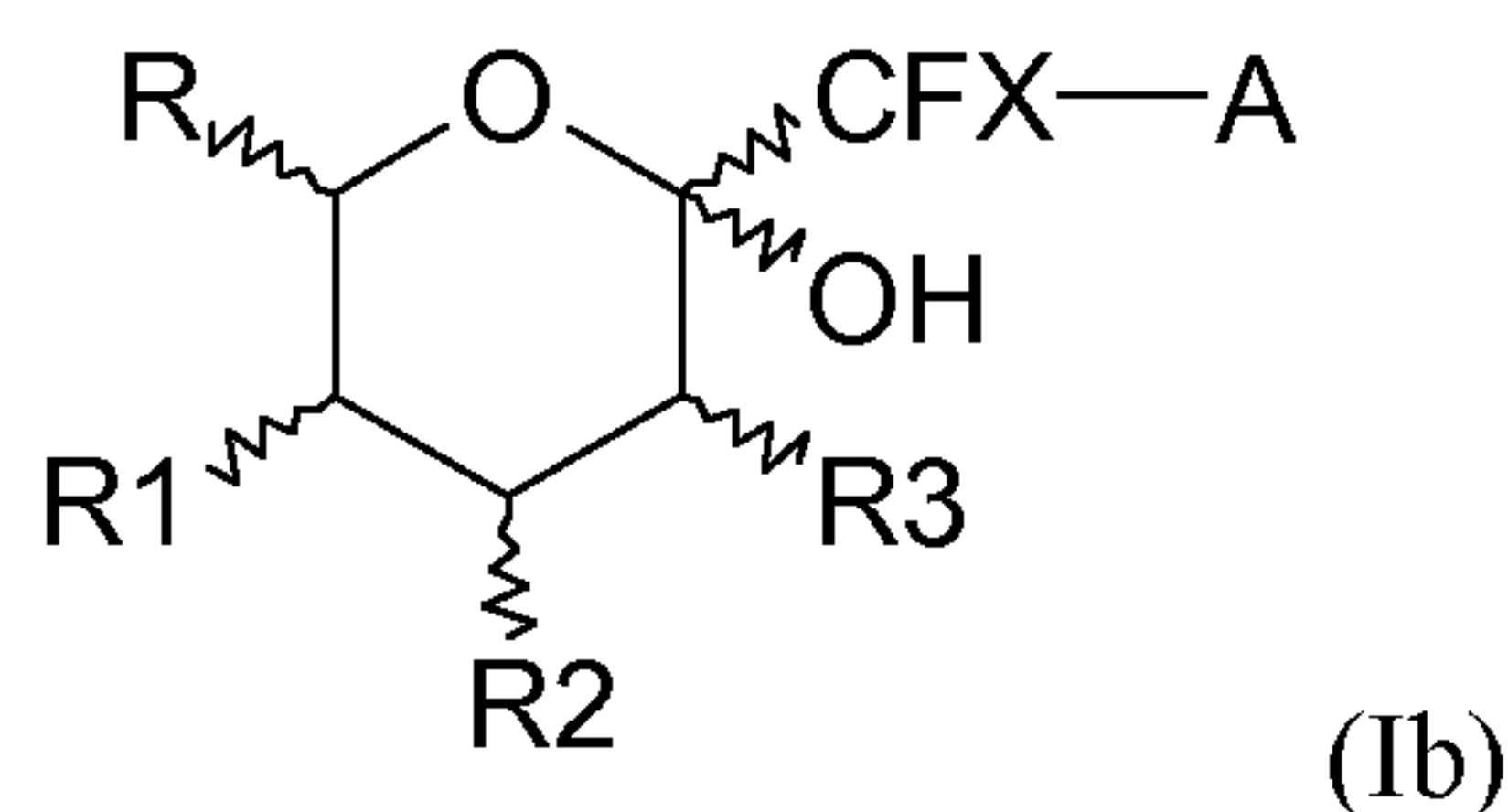
Furthermore, the compound of formula (III) could also be obtained via a coupling reaction between the halogenated derivative (VI) and an organometallic

derivative obtained from the halogenated compound A-Hal, where Hal represents a halogen, via halogen-metal exchange or insertion of a metal into the carbon-halogen bond, possibly followed by transmetallation as described above.

This coupling can be catalyzed by a palladium or nickel catalyst, such as
 5 $\text{Pd}_2(\text{dba})_3$, $\text{Pd}(\text{PPh}_3)_4$, $\text{PdCl}_2(\text{PPh}_3)_2$ or dmpeNiCl_2 (dmpe meaning (1,2-dimethylphosphino)ethane).

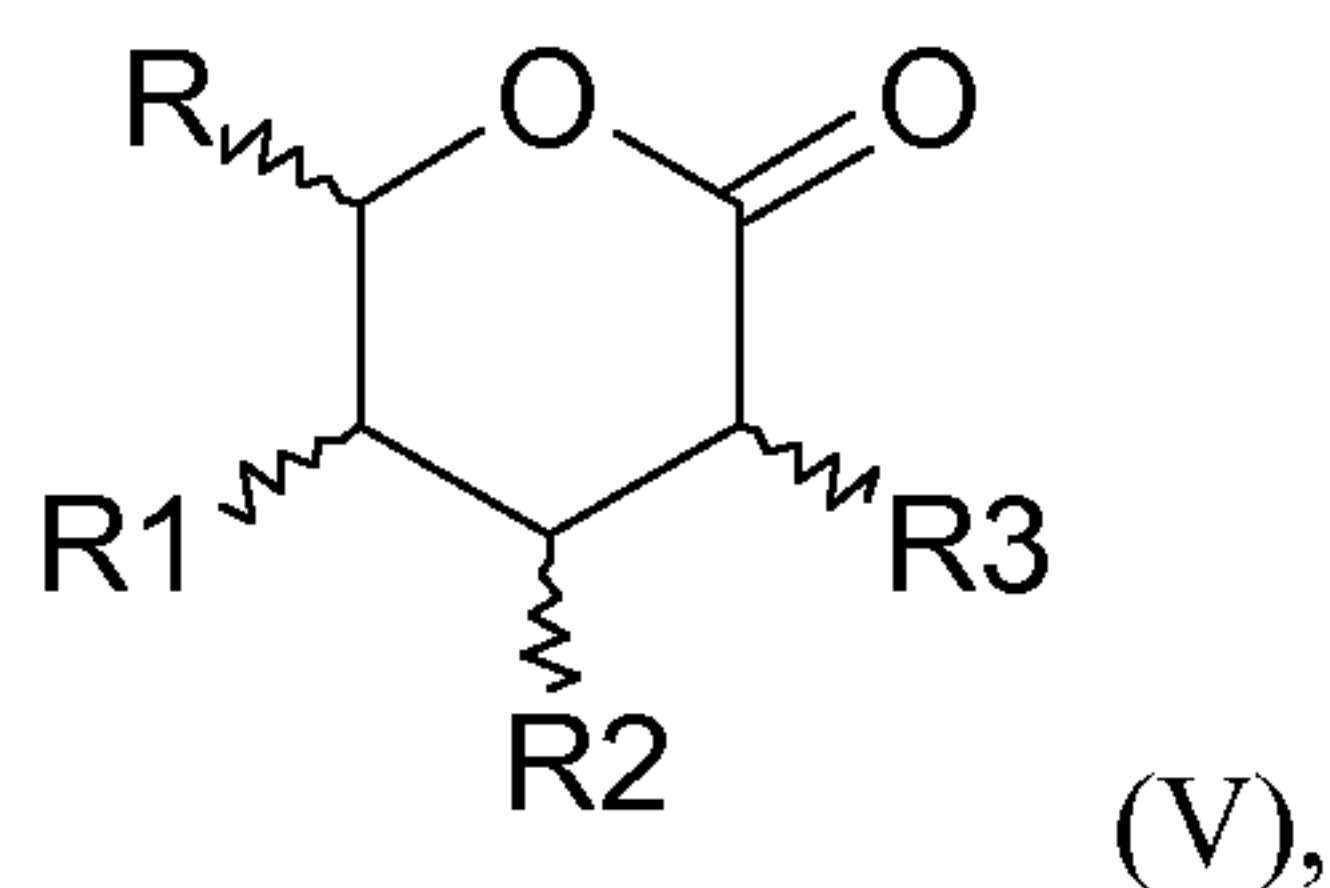
The compound of generic formula (V) can be obtained according to a process described in the patent application WO 2007/128 899, in particular via the reaction of a lactone derived from a sugar with CFBr_3 , Et_2Zn and PPh_3 , in a solvent such as THF.

10 Another object of this invention is a process for preparing a compound of generic formula (Ib) below:



corresponding to a compound of formula (I), as defined above, wherein X represents a hydrogen or a fluorine atom and R4 represents an OH group, according to the following
 15 steps:

(a3) placing a compound of formula $\text{A-CFXX}'$, wherein X is as defined above, A is as defined previously and X' represents a bromine or chlorine atom, in the presence of a compound of generic formula (V):



20 wherein R, R1, R2 and R3 are as defined previously, and

(b3) addition of a $(\text{C}_1\text{-C}_6)$ -alkyl lithium to the mixture of step (a3), in order to obtain a compound of formula (Ib).

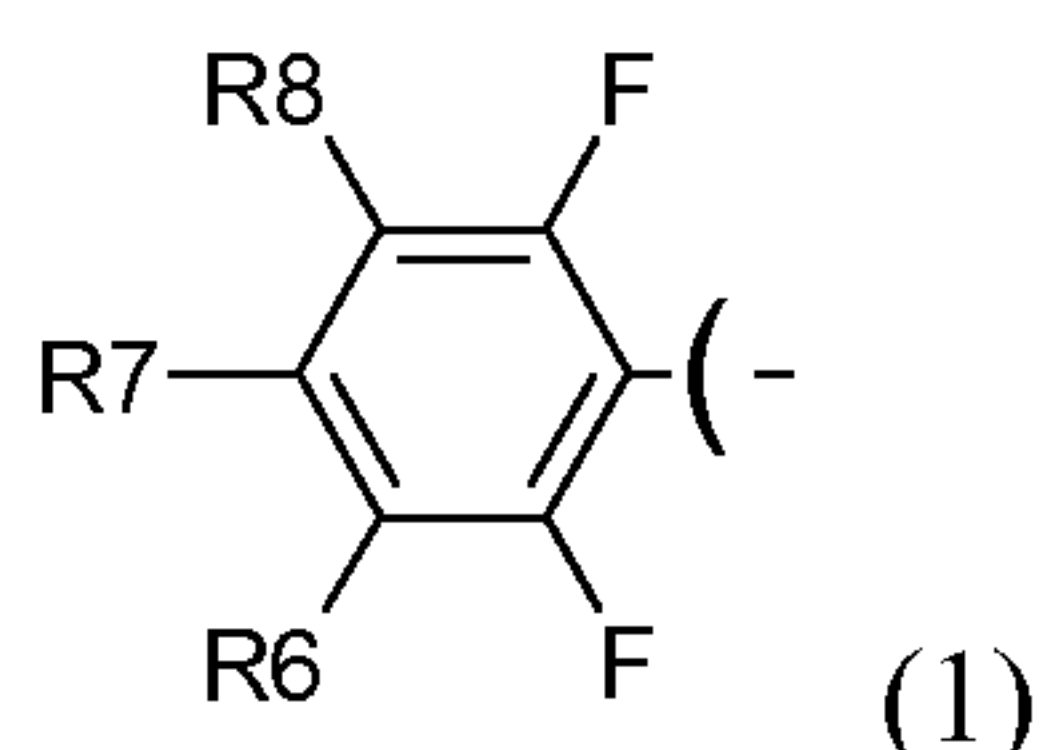
This reaction is thus carried out under Barbier conditions, the $(\text{C}_1\text{-C}_6)$ -alkyl lithium advantageously being n-butyllithium, sec-butyllithium or tert-butyllithium

25 However, indium could be used also in place of the $(\text{C}_1\text{-C}_6)$ -alkyl lithium.

According to a first alternative, the compound of formula A-CFXX' is obtained from a compound of formula A-CHO, when X represents a fluorine atom, or from a compound of formula A-CH₂OH or A-CH₂Br, when X represents a hydrogen atom, with A as defined above, via fluorination in the presence of diethylaminosulfur trifluoride (DAST) for A-CHO or A-CH₂OH or tetrabutylammonium fluoride (TBAF) for A-CH₂Br, followed by bromination or chlorination in the presence of *N*-bromosuccinimide (NBS), *N*-chlorosuccinimide (NCS) or Br₂, under ultraviolet radiation.

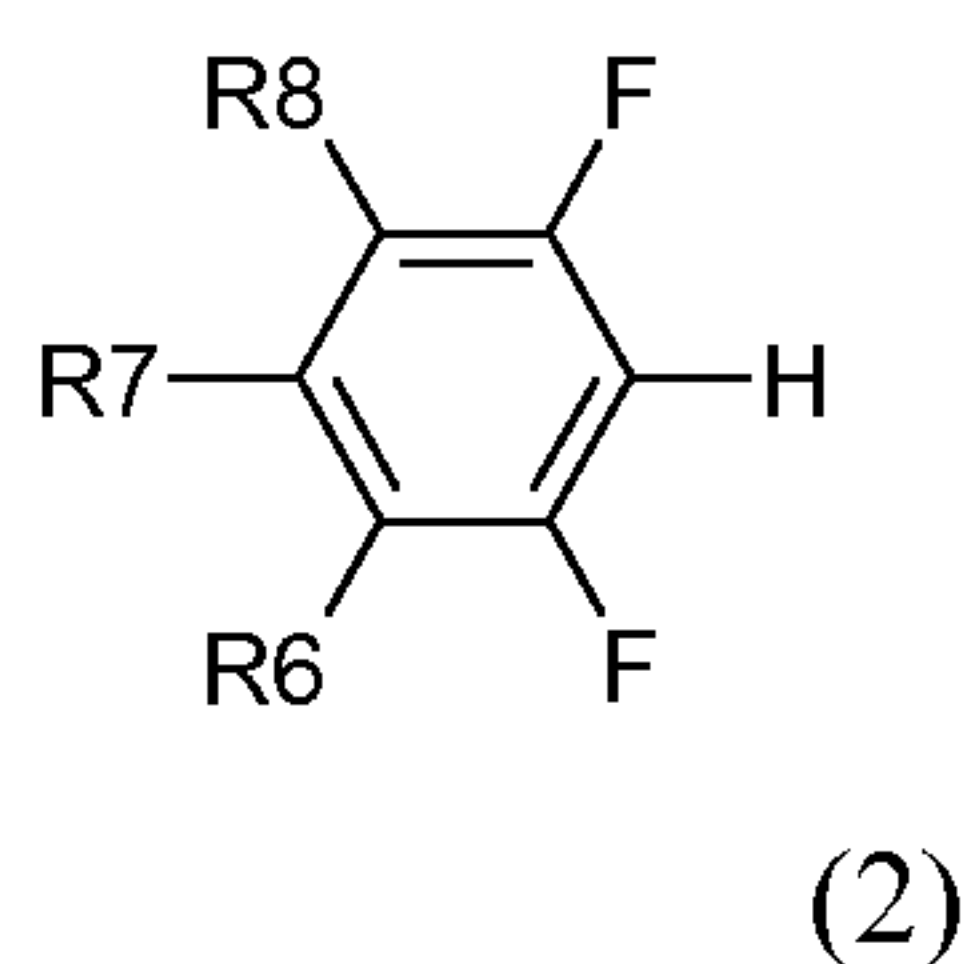
Such a process is described, in particular, in *Macromolecules* (2007), 40 (19), 6799-6809 and *Polymer* (2007), 48, 1541-1549.

According to a second alternative, when radical A corresponds to a phenyl ring of formula (1) below:



with R6, R7 and R8 as defined previously,

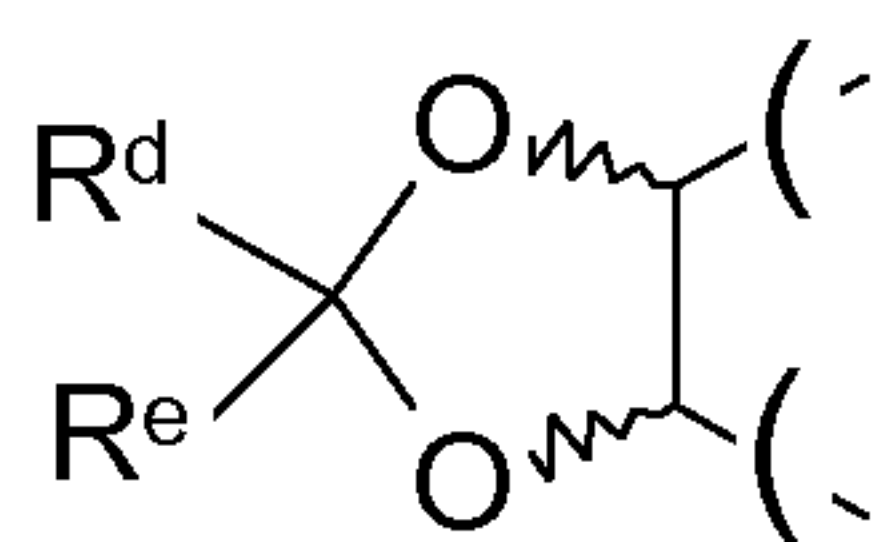
the compound of formula A-CFXX', wherein X = F and X' = Br, can be prepared from a compound of formula (2) below:



via deprotonation of compound (2) in the presence of a base such as *n*-butyllithium, the hydrogen atom torn away being the one situated between the two fluorine atoms, and then the anion thus obtained reacts with CF₂Br₂.

Another object of this invention is a process for preparing a compound of generic formula (Ia) as defined above, wherein:

- X represents a hydrogen or a fluorine atom and
- R4 represents a OSiR^aR^bR^c, OR¹¹, OCOR¹¹, OCO₂R¹¹, OCONR¹²R¹³ group or R3 and R4, together with the carbon atoms carrying them, form a cyclic acetal having the following formula:



with R^a , R^b , R^c , R^d , R^e , R^{11} , R^{12} and R^{13} as defined above, characterised in that the compound of formula (Ia) is obtained by substitution of the OH group of a compound of formula (Ib) as defined above.

5 Such a substitution reaction is well known from the person skilled in the art who will know to adapt the reaction conditions.

Compounds of formula (I) wherein $R_4 = OH$, obtained in particular by a previous process, can be further involved in one or more additional reaction steps for substituting the hydroxyl group in order to produce similar compounds of formula (I) wherein the OH group of radical R_4 has been replaced by an ether (OR^{11}), ester ($OCOR^{11}$), carbonate (OCO_2R^{11}), carbamate ($OCONR^{12}R^{13}$) or else a silyloxy ($OSiR^aR^bR^c$) group.

When R_3 and R_4 represent an OH group, a reaction with a ketone can give access to a compound of formula (I) wherein R_3 and R_4 , together with the carbon atoms carrying them, form a cyclic acetal as defined previously.

In the same way, it is possible to convert the preceding OH group in chlorine or bromine in the presence of $SOCl_2$ or $SOBr_2$ and pyridine, to give access thus to compounds of formula (I) in which R_4 represents a chlorine or bromine atom, or it is possible to convert this OH group in fluorine in the presence of a fluorinating agent such as DAST.

Starting with a compound of formula (I), wherein R_4 represents a halogen atom or a leaving group (e.g., in the form of a mesylate, tosylate or triflate), it is also possible to carry out a substitution reaction with, for example, a hydrogen, an amine ($HNR^{12}R^{13}$) or with an alkyl or alkenyl group in order to give access to compounds of formula (I) wherein R_4 represents a hydrogen atom or a $NR^{12}R^{13}$, (C_1-C_6) -alkyl or (C_2-C_6) -alkenyl group.

Compounds of formula (I) wherein $R_4 = OH$ and $R_0 = H$, obtained in particular by a previous process, can be also further involved in one or more additional reaction steps such as a concomitant magnesium derivative mediated C-1 reduction and C-5 oxydation using magnesium derivatives such as an alkoxide magnesium halide, a

benzylmagnesium halide or an alkylmagnesium halide to lead to compounds of formula (I) wherein $R_0 = OH$ and $R_4 = H$.

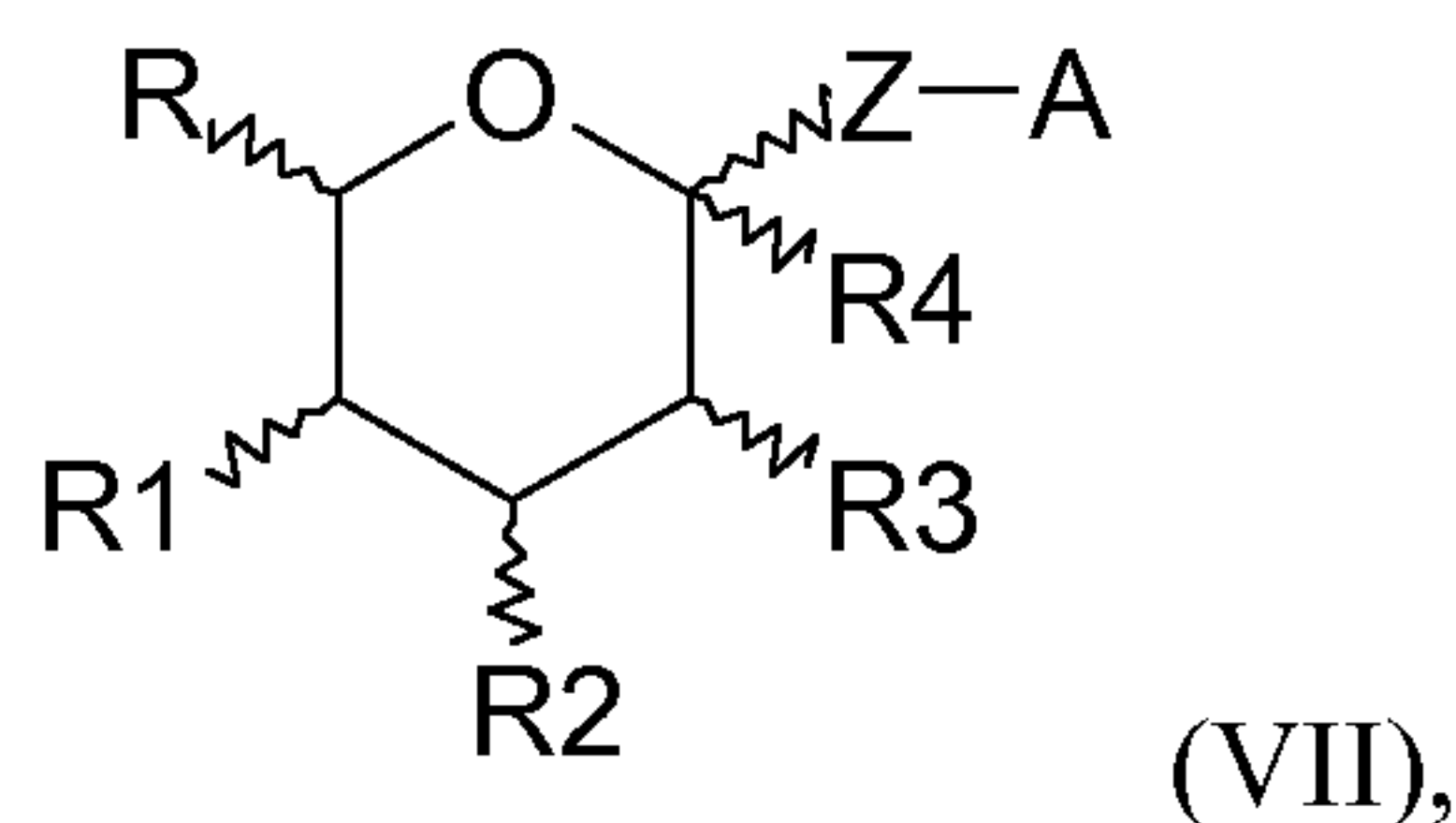
By “alkoxide magnesium halide”, is meant, in the sense of the present invention, a compound of formula $AlkO-Mg-Hal$, with Hal representing an halogen atom, such as a bromine atom, and Alk representing a (C_1-C_6) alkyl group as defined above. It can be in particular $tBuOMgBr$.

By “benzylmagnesium halide”, is meant, in the sense of the present invention, a compound of formula $Bn-Mg-Hal$, with Hal representing an halogen atom, such as a bromine atom. It is in particular a benzylmagnesium bromide.

By “alkylmagnesium halide”, is meant, in the sense of the present invention, a compound of formula $Alk-Mg-Hal$, with Hal representing an halogen atom, such as a bromine atom, and Alk representing a (C_1-C_6) alkyl group as defined above.

Furthermore, additional protection/deprotection and/or functionalization steps, well known from the person skilled in the art, can be anticipated in the preceding processes for preparing compounds of formula (I).

Another object of this invention is a process for preparing a compound of generic formula (IIa), characterized in that the compound of formula (IIa) is obtained by fluorination of a compound of the following formula (VII):



wherein A , R , R_1 , R_2 , R_3 and R_4 are as defined above and Z represents a $C=O$, $CHOH$ or $C(SR^{15})(SR^{16})$ group, with R^{15} and R^{16} representing, independently of each other, a (C_1-C_6) alkyl group or forming together an hydrocarbon chain of formula $-CH_2-(CH_2)_p-$, with $p = 1$ or 2 , between the two sulphur atoms.

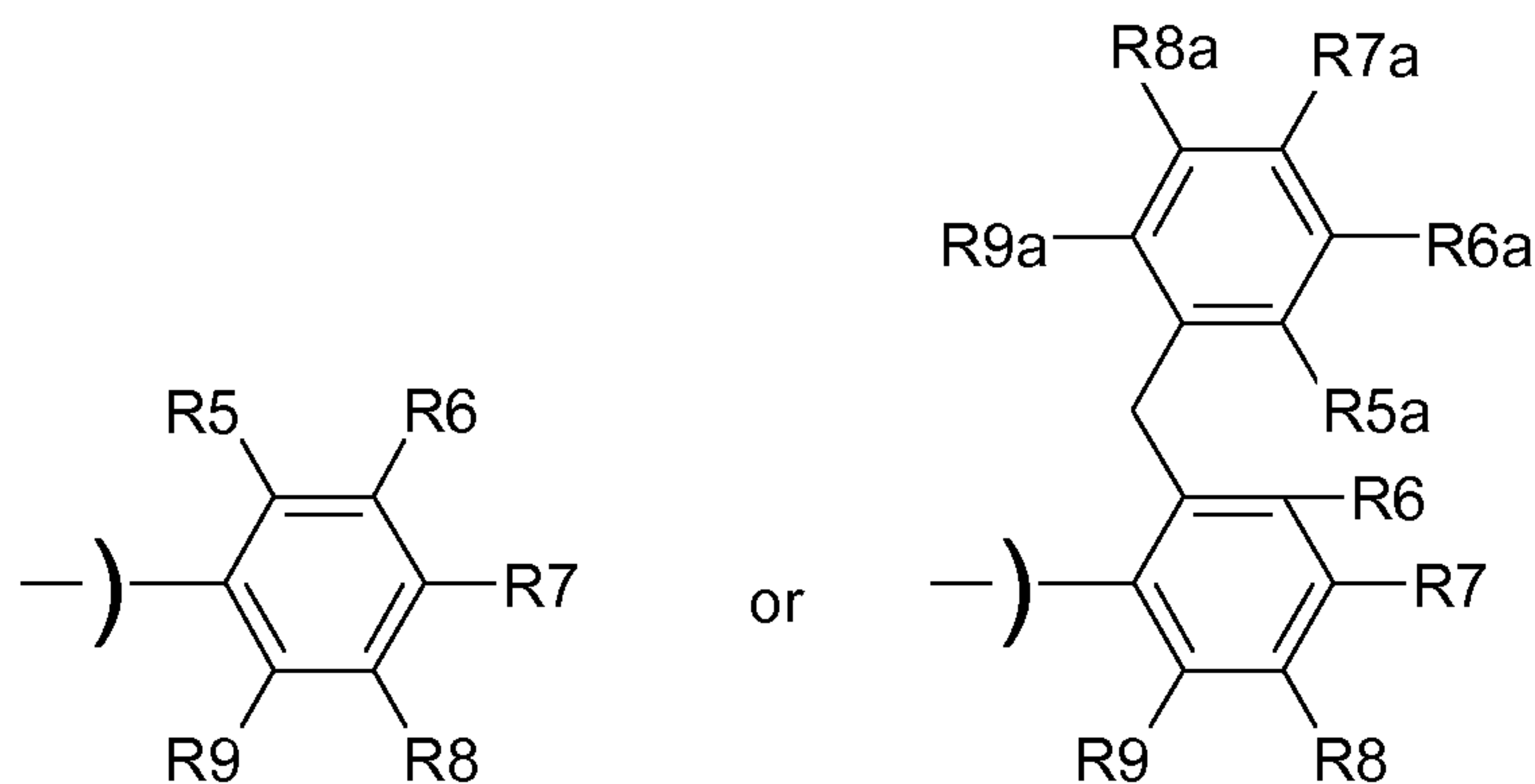
In the case of a radical $Z = C=O$ or $CHOH$, the fluorination can be carried out in the presence of a fluorinating compound such as DAST, preferably at a temperature comprised between ambient temperature and $45^\circ C$. A solvent such as dichloromethane can be used. The fluorination of a compound of formula (VII) wherein $Z = C=O$, respectively $CHOH$, gives access to a compound of formula (Ia) wherein $X = F$, respectively H .

In the case of a radical $Z = C(SR^{15})(SR^{16})$, the step of fluorination, which is accompanied of an oxidative desulfurization, can be carried out by using an oxidant such as NBS (N-bromosuccinimide), NIS (N-iodosuccinimide), $NO+BF_4^-$ or DBH (1,3dibromo-5,5-dimethylhydantoin), along with a fluorinating agent such as HF-pyridine, HF-triethylamine, $TBAH_2F_3$ (tetrabutylammonium dihydrogen trifluoride) or DAST, in solvent such as dichloromethane, notably at a temperature ranging from 0° to room temperature (Adv. Synth. Catal. 2001, 343, N°5, 235-250).

Preferably, R4 represents a hydrogen or halogen (e.g. F, Br, Cl) atom or an OH group. When R4 = OH, it is possible to modify this radical as previously described in the preceding processes.

When R4 = OH, it is preferable to protect it in order to avoid its fluorination, e.g. through the use of a base such as sodium hydride (NaH) and the addition of an electrophile, in particular an alkylhalide (such as methyl iodide) or a benzylhalide (such as benzyl bromide). All other classical protecting group known for a person skilled in the art can also be used to achieve protection of R4 = OH.

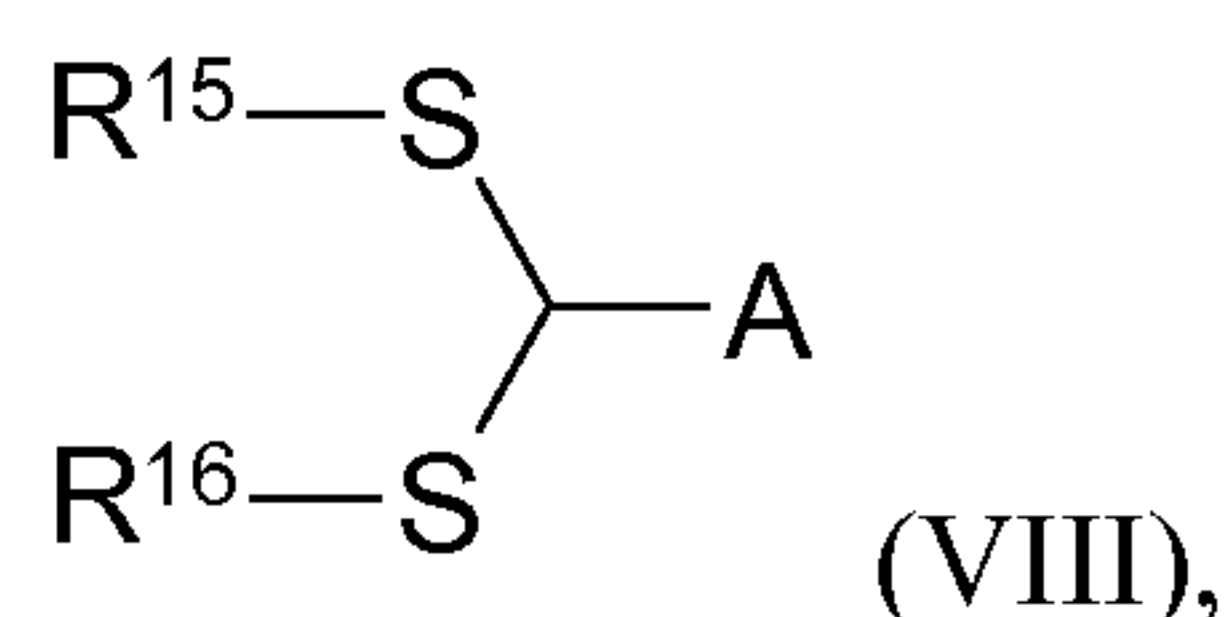
Preferably, A represents a radical:



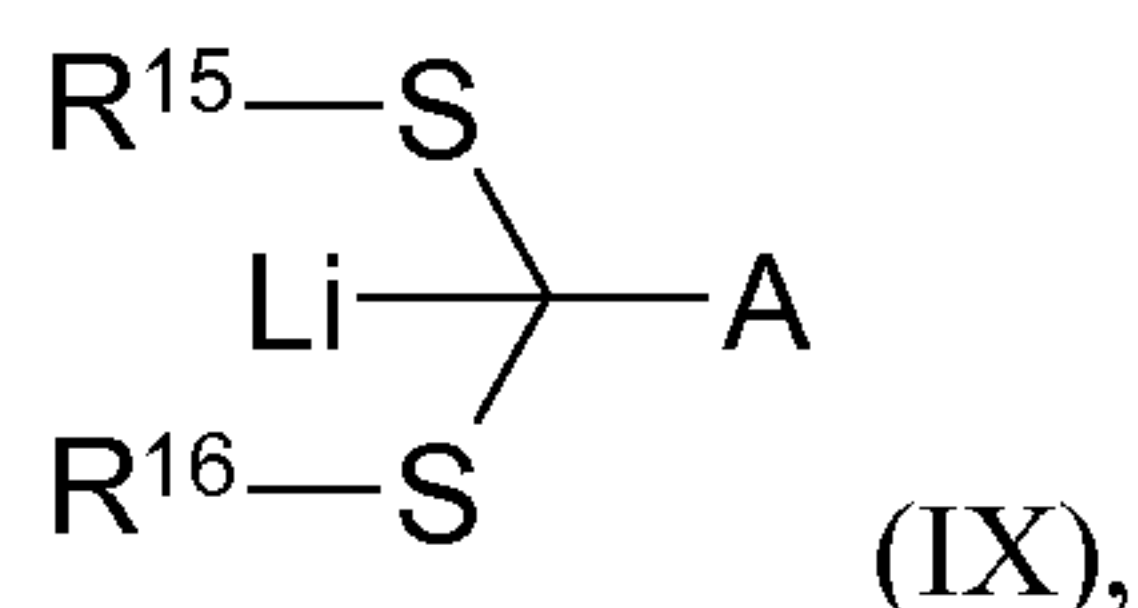
with, R5, R6, R7, R8, R9, R5a, R6a, R7a, R8a and R9a as defined above.

According to a first variant, the compound of formula (VII) can be prepared according to the following steps:

(a4) reaction between a lithio base and the dithiane compound of the following formula (VIII):

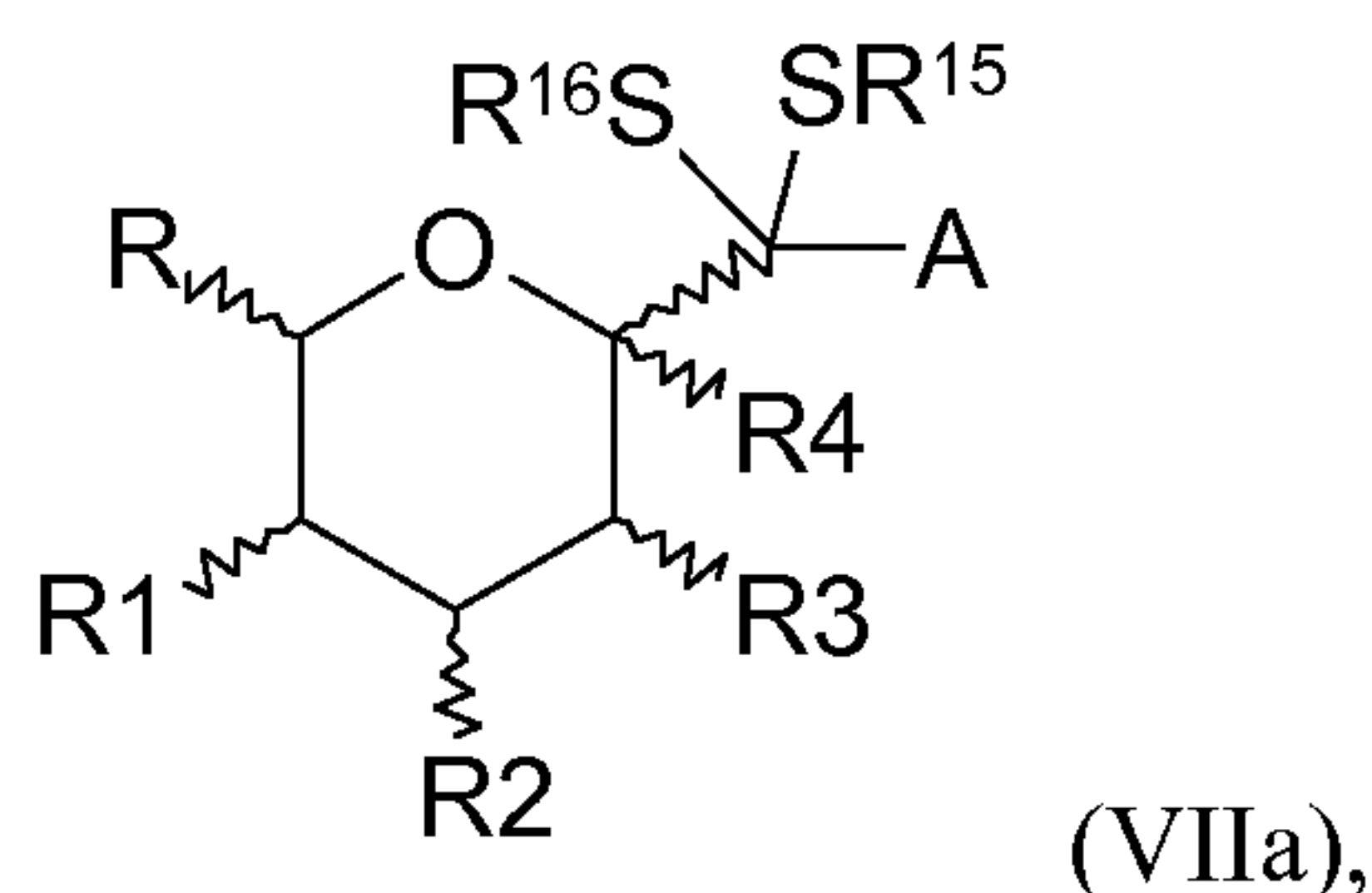


wherein A, R⁵, R⁶, R⁷, R⁸, R⁹, R¹⁵ and R¹⁶ are as defined above,
to give a lithio derivative of formula (IX):



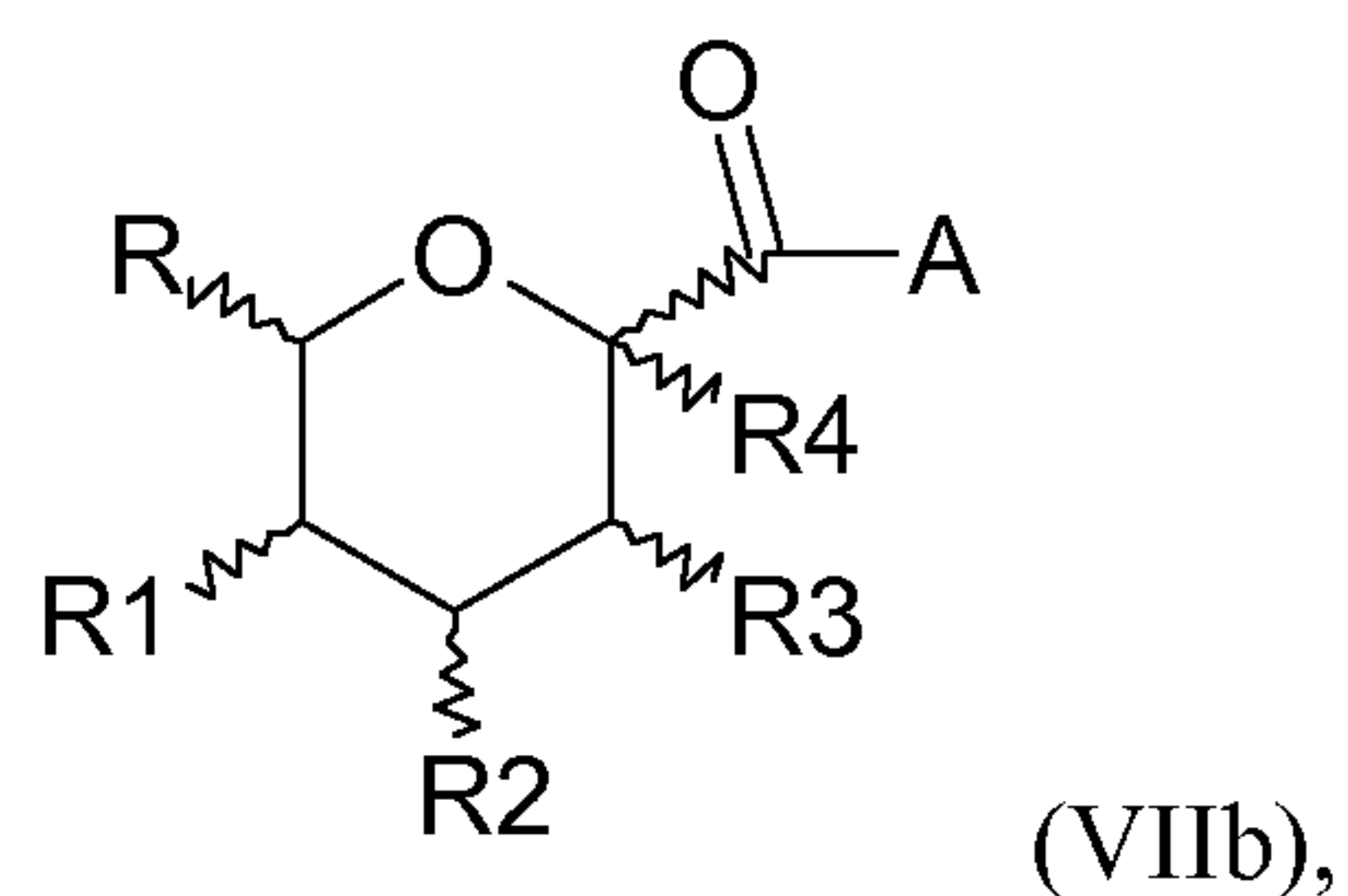
wherein A, R⁵, R⁶, R⁷, R⁸, R⁹, R¹⁵ and R¹⁶ are as defined above,

- 5 (b4) addition of the previous lithio derivative of formula (IX) obtained in the previous step (a4) onto a lactone of formula (V) as defined above to lead to a compound of formula (VIIa)



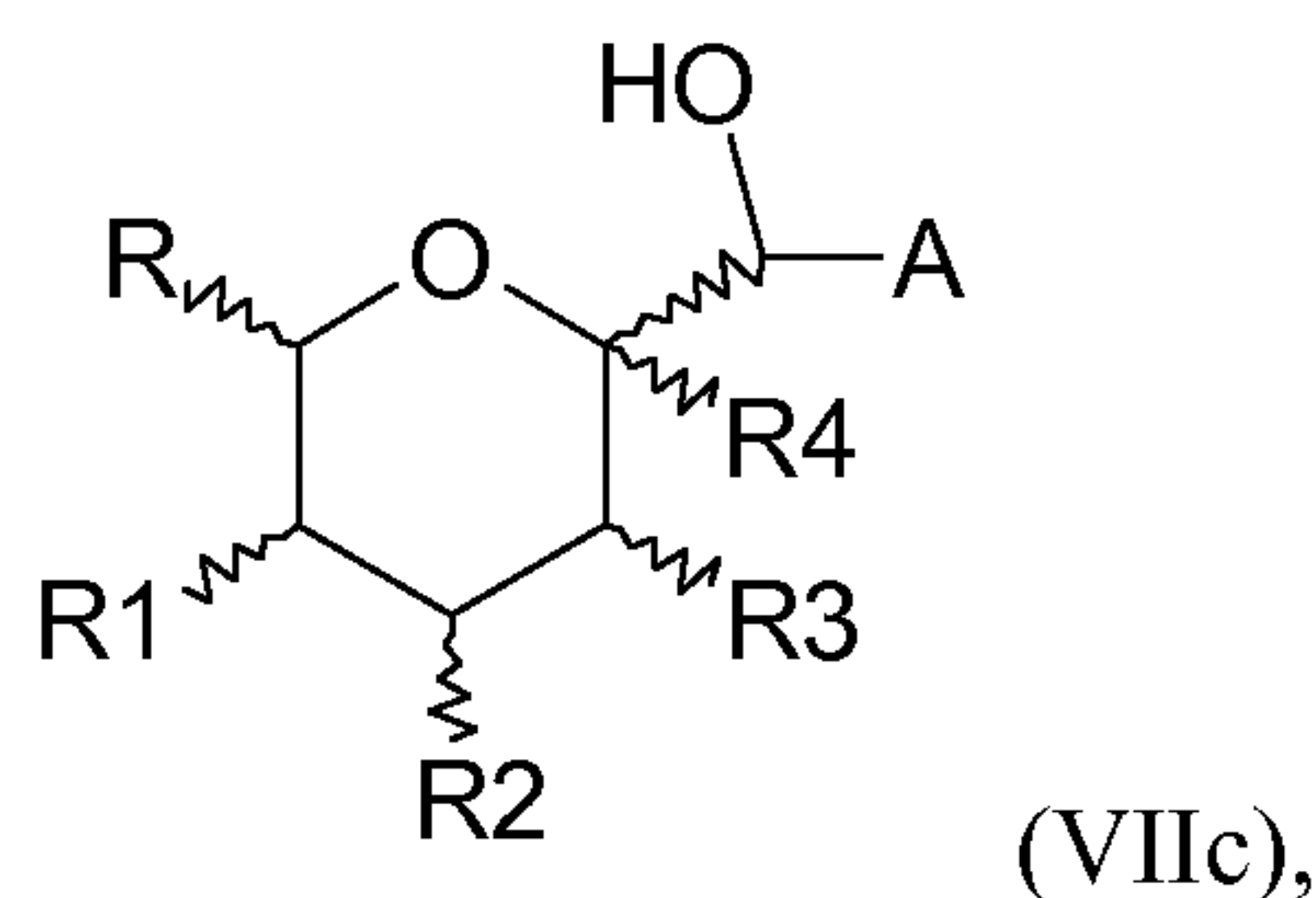
- wherein A, R, R¹, R², R³, R⁵, R⁶, R⁷, R⁸, R⁹, R¹⁵ and R¹⁶ are as defined above and R⁴
10 = OH,
which corresponds to a compound of formula (VII) wherein Z = C(SR¹⁵)(SR¹⁶) and R⁴ = OH,

(c4) hydrolysis of the dithiane moiety of the compound of formula (VIIa) obtained in the previous step (b4) to give a compound of formula (VIIb):



- 15 wherein A, R, R¹, R², R³, R⁵, R⁶, R⁷, R⁸ and R⁹ are as defined above and R⁴ = OH,
which corresponds to a compound of formula (VII) wherein Z = C = O and R⁴ = OH,

(d4) optionally reduction of the compound of formula (VIIb) obtained in the previous step (c4) in order to give the compound of the following formula (VIIc):



wherein A, R, R1, R2, R3, R5, R6, R7, R8 and R9 are as defined above and R4 = OH or H,

which corresponds to a compound of formula (VII) wherein Z = CHOH and R4 = OH or H, and

(e4) optionally oxidation of the compound of formula (VIIc) obtained at the previous step (d4) for which R4 = H to give a compound of formula (VIIb) for which R4 = H,

which corresponds to a compound of formula (VII) wherein Z = C = O and R4 = H.

Step (a4) can be carried out by using an appropriate lithio base which can undergo the deprotonation followed by the lithiation of the carbon atom bearing the two sulphur atoms. It can be notably a (C₁-C₆)-alkyl lithium, such as butyllithium, or lithium diisopropylamide (LDA). If necessary, the reaction can be carried out in the presence of hexamethylphosphoric triamide or tetramethylethylenediamine. The solvent used in this reaction can be advantageously chosen among the ethers, such as tetrahydrofuran.

Dithiane compound of formula (VIII), used in this step (a1), can be obtained easily by a classical condensation of a thiol or a dithiol on the corresponding aldehyde (*J. Org. Chem.* 1978, 43(21) 4172-4177; *J. Org. Chem.* 1979, 44(15), 2804-2805; *Org. Biomol. Chem.* 2003, 1, 306-317).

By “ether”, is meant, in the framework of the present invention, a compound of formula R¹⁷-O-R¹⁸, with R¹⁷ and R¹⁸ representing, independently of each other, a (C₁-C₆)alkyl group or form together an hydrocarbon chain of formula -CH₂-(CH₂)_p-, with p = 1 or 2, to give a cyclic ether.

Step (b4) will be carried out advantageously in the same solvent as for step (a4), preferably at a temperature of -90°C to 0°C.

The hydrolysis of step (c4) can be carried out in the presence of an oxidant such as NCS (N-chlorosuccinimide), NBS (N-bromosuccinimide), NIS (N-iodosuccinimide), MeI, Br₂ or I₂, with a base such as AgNO₃ or CaCO₃.

This step (c4) can be carried out in a mixture of solvent such as dichloromethane/H₂O, acetonitrile/ H₂O or HgCl₂ in H₂O.

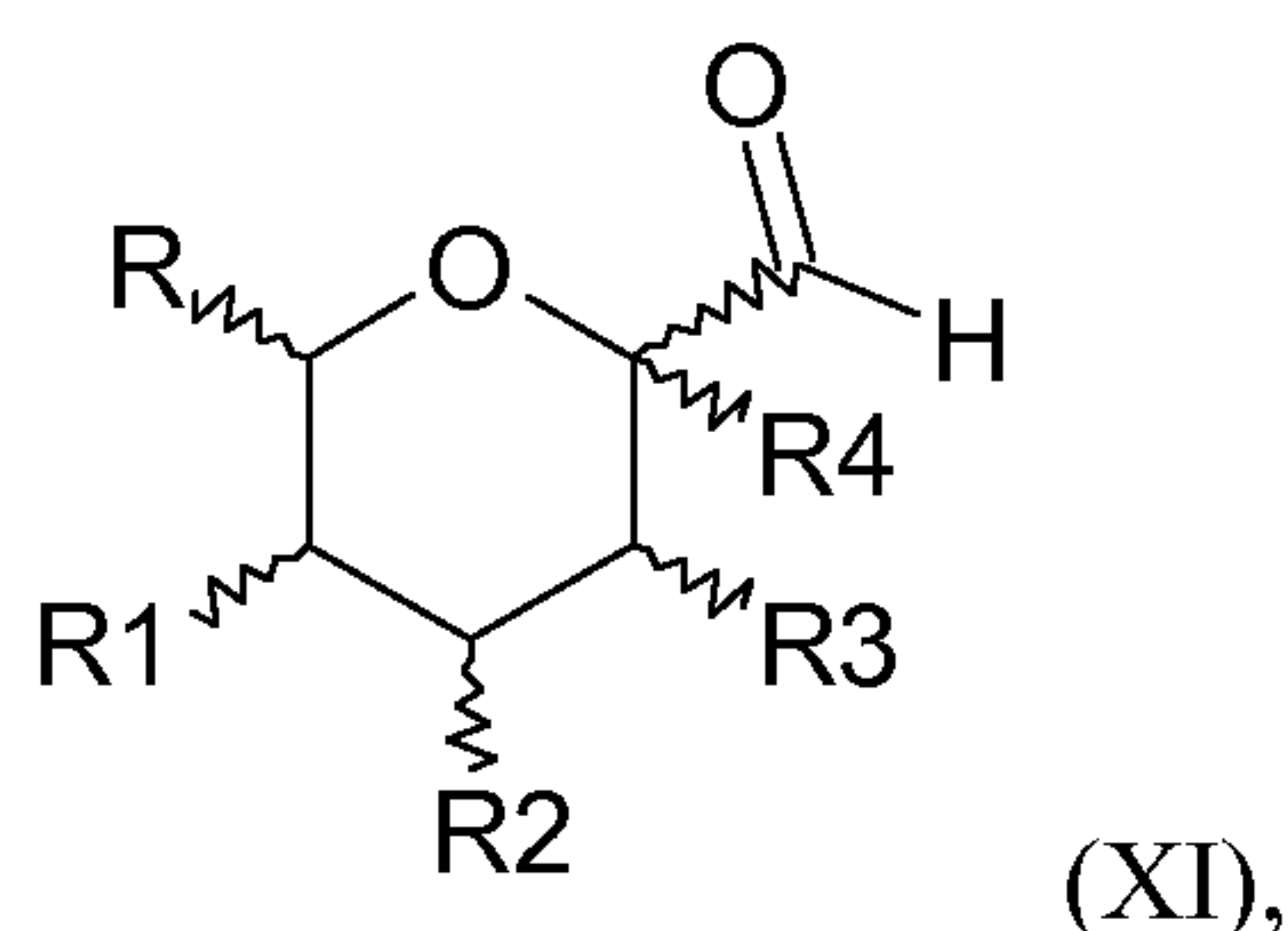
The step (d4) of reduction can be carried out by methods well known of the person skilled in the art. For example, a Lewis acid, such as BF₃-OEt or TMSOTf (trimethylsilyl trifluoromethanesulfonate), and a reducing agent, such as Et₃SiH, can be used, notably in a solvent such as dichloromethane, optionally in mixture with acetonitrile. Preferably the reaction will be carried out at a temperature comprised between -78°C and 0°C. In this reaction step it is possible to reduce the ketone moiety as well as the OH group of radical R₄, or to reduce selectively only the ketone moiety, according to the chosen reaction conditions. In particular, the selective reduction of the ketone moiety can be carried out at lower temperatures such as about -40°C, whereas the reduction of both moieties (ketone and hydroxyle) can be carried out at higher temperatures such as about -20°C.

The step (e4) allows to give access to compounds of formula (VII) in which Z = C = O and R₄ = H. This oxidation reaction can be carried out in the presence of classical oxidants well known of the person skilled in the art such as by using PCC (pyridinium chlorochromate). In this case, the reaction can be carried out in a solvent such as dichloromethane, advantageously at a temperature comprised between ambient temperature and 45°C.

Moreover, when R₄ = OH, it is possible to modify this radical as previously described in the preceding processes to give access to other substituents. Such a reaction of modification of the radical R₄ can be performed on the all the different intermediates.

According to a second variant, a compound of formula (VIIb) or (VIIc) as defined above can be prepared according to the following steps:

(a5) reaction between an aldehyde of generic formula (XI) :



wherein R, R₁, R₂, R₃ and R₄ are as defined above,

and a compound of formula A-M,
wherein A is as defined above and M represents lithium or magnesium halide, such as magnesium bromide,

to give a compound of formula (VIIc) as defined above, and

- 5 (b5) optionally, oxidation of the compound of formula (VIIc) obtained at the previous step (a5) to give a compound of formula (VIIb) as defined above.

In this second variant, R₄ represents preferably a hydrogen atom.

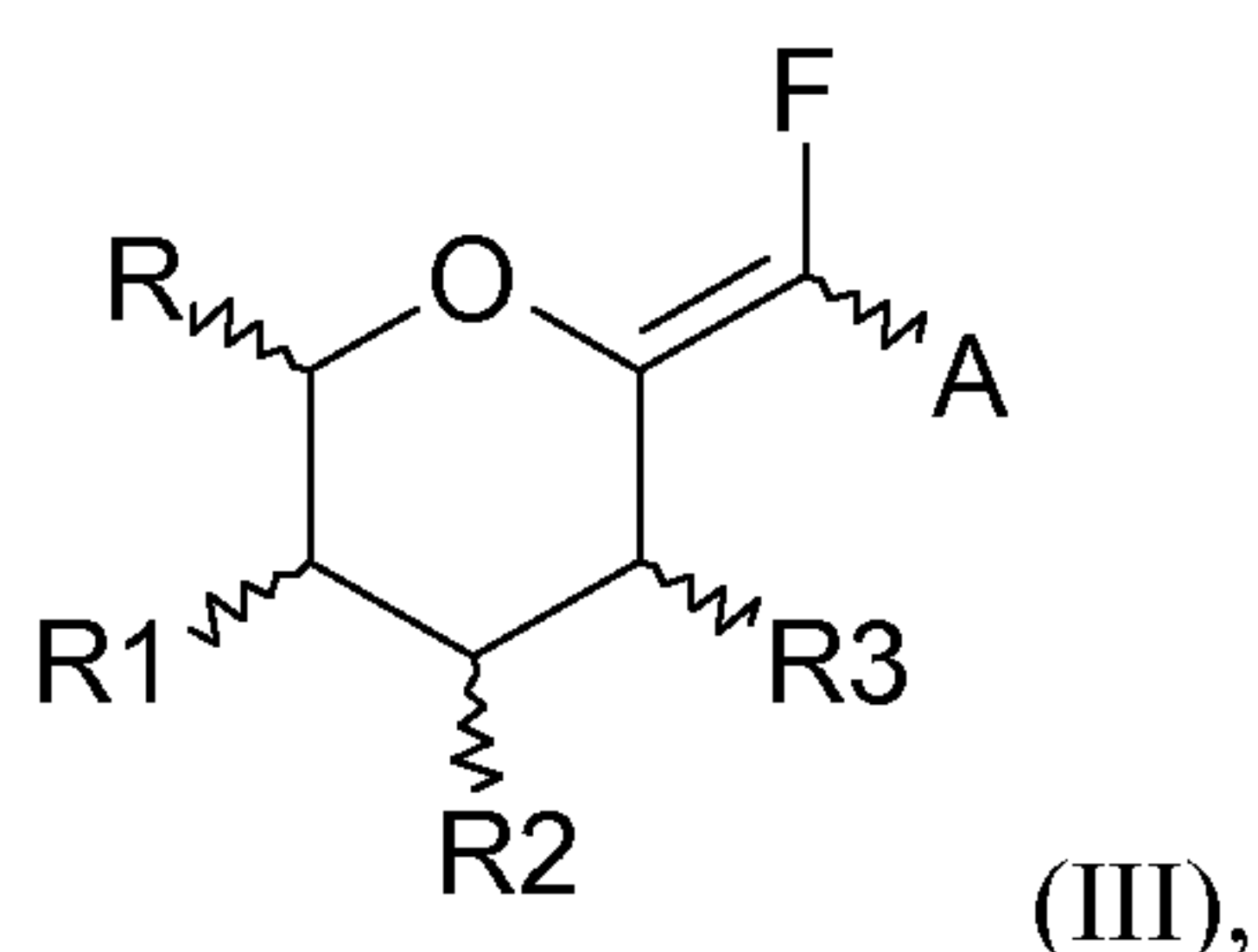
Step (a5) can be carried out by the addition of the compound of formula A-M
10 onto the aldehyde of formula (XI) according to the conditions described in the publication *Org. Biomol. Chem.* 2007, 5, 2311-2314, in particular in a solvent such as THF and preferably at about -78°C.

Compounds of formula A-M can be obtained from an halogen-metal exchange
between the corresponding halide derivative (A-Hal with Hal representing an halogen
15 atom) and a (C₁-C₆)-alkyl lithium, a (C₁-C₆)-alkyl magnesium halide or a di-(C₁-C₆)-alkyl magnesium as previously described, or from a reaction between the same halide derivative with magnesium or lithium.

Compounds of formula (XI) can be prepared according to the methods described
in the following publications: *Chem. Bio. Chem.* 2006, 7, 1017-1022; *Tetrahedron Lett.*
20 2004, 45, 7761-7763; *Tetrahedron Lett.* 2002, 43, 7271-7272; *Synlett* 2001, 1, 79-81;
and *Synlett* 1994, 9, 705-708.

The oxidation of step (b5) can be carried out by methods well known of the
person skilled in the art, such as by using PCC (pyridinium chlorochromate) as oxidant.
In this case, the reaction can be carried out in a solvent such as dichloromethane,
25 advantageously at a temperature comprised between ambient temperature and 45°C.

Another object of this invention is a compound of generic formula (III) as below:



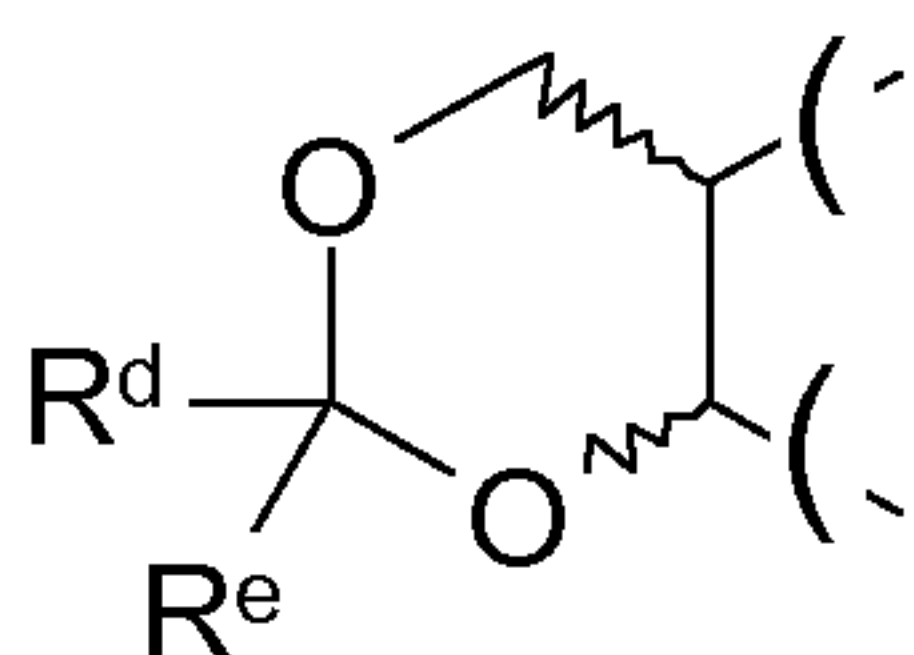
or a pharmaceutically acceptable salt thereof, a tautomer, an isomer or a mixture of isomers in any proportions, in particular a mixture of enantiomers, and particularly a racemate mixture, wherein:

- R represents a hydrogen or a fluorine atom or CH₃, CH₂F, CH₂OH, CH₂OSiR^aR^bR^c, CH₂OR¹¹, CH₂OCOR¹¹, CH₂OCO₂R¹¹, CH₂OCONR¹²R¹³, CH₂OP(O)(OR¹⁴)₂ or CH₂OSO₃R¹⁴ group;

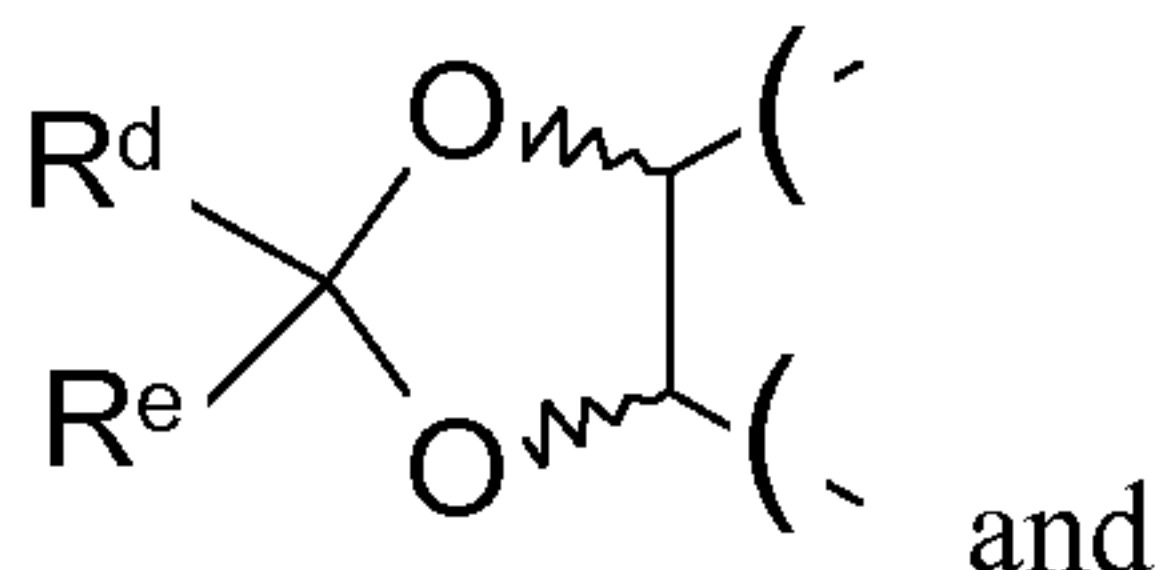
- R₁ and R₂ represent, independently from one another, a fluorine atom or an OH, OSiR^aR^bR^c, OR¹¹, OCOR¹¹, OCO₂R¹¹ or OCONR¹²R¹³ group;

- R₃ represents a hydrogen or fluorine atom or an OH, OSiR^aR^bR^c, OR¹¹, OCOR¹¹, OCO₂R¹¹, OCONR¹²R¹³, NR¹²R¹³ or NR¹²COR¹¹ group;

or R and R₁, together with the carbon atoms carrying them, form a cyclic acetal having the following formula:



and/or (R₂ and R₃) or (R₁ and R₂), together with the carbon atoms carrying them, form a cyclic acetal having the following formula:



- A represents an aryl or heteroaryl group, possibly substituted by one or more groups chosen among an halogen atom, a CN, SO₂, SiR^aR^bR^c, (C₁-C₆)-alkyl, (C₂-C₆)-alkenyl, (C₂-C₆)-alkynyl, (C₃-C₇)-cycloalkyl, 5 to 7 ring-membered heterocycloalkyl, aryl, heteroaryl, aryl-(C₁-C₆)-alkyl, heteroaryl-(C₁-C₆)-alkyl, (C₁-C₆)-alkyl-aryl, (C₁-C₆)-alkyl-heteroaryl, OR¹¹, COR¹¹, OCOR¹¹, CO₂R¹¹, NR¹²R¹³, NR¹²COR¹¹, CONR¹²R¹³, SR¹¹, SO₂R¹¹, CSR¹¹ and OSO₃R¹¹ group,

the whole being possibly substituted by one or more groups chosen among an halogen atom, an OH, (C₁-C₆)-alkyl, (C₁-C₆)-alkoxy, COOH and CHO group;

with:

- R¹¹ representing a (C₁-C₆)-alkyl, (C₂-C₆)-alkenyl, (C₂-C₆)-alkynyl, (C₃-C₇)-cycloalkyl, 5 to 7 ring-membered heterocycloalkyl, aryl, aryl-(C₁-C₆)-alkyl or (C₁-C₆)-

alkyl-aryl, this group being possibly substituted by one or more groups chosen among an halogen atom, an OH, COOH and CHO group;

- R^{12} and R^{13} representing, independently from one another, a hydrogen atom or a (C₁-C₆)-alkyl or aryl-(C₁-C₆)-alkyle group;

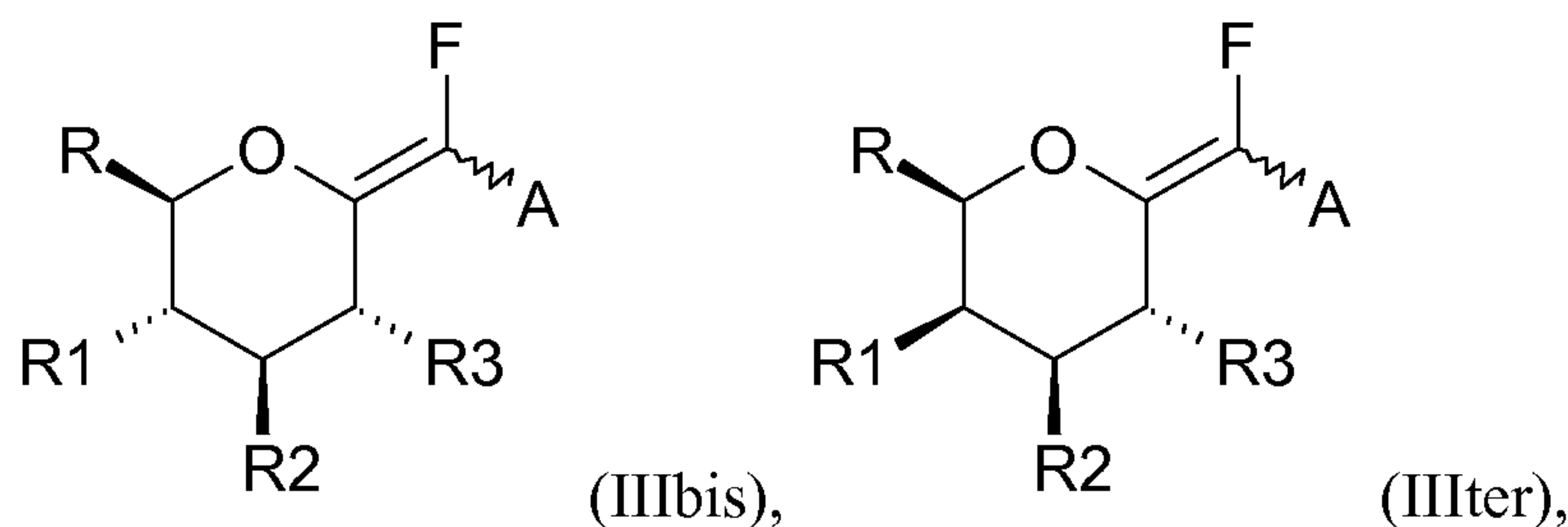
5 - R^{14} representing a hydrogen atom or a (C₁-C₆)-alkyl group;

- R^a , R^b and R^c representing, independently from one another, a (C₁-C₆)-alkyl, aryl or aryl-(C₁-C₆)-alkyl group; and

R^d and R^e representing, independently from one another, a hydrogen atom or a (C₁-C₆)-alkyl group.

10 The compounds of formula (III) are useful, in particular, as synthesis intermediates of the compounds of formula (I).

The preceding compounds respond advantageously to the formula (IIIbis) or (IIIter) below:



15 with R, R₁, R₂, R₃ and A as defined previously.

The preceding compounds respond advantageously to the formula (IIIbis).

According to a particular embodiment of the invention, A represents an aryl or heteroaryl group, possibly substituted by one or more groups chosen among an halogen atom, a CN, SO₂, SiR^aR^bR^c, (C₁-C₆)-alkyl, (C₂-C₆)-alkenyl, (C₂-C₆)-alkynyl, (C₃-C₇)-cycloalkyl, 5 to 7 ring-membered heterocycloalkyl, aryl, heteroaryl, aryl-(C₁-C₆)-alkyl, heteroaryl-(C₁-C₆)-alkyl, (C₁-C₆)-alkyl-aryl, (C₁-C₆)-alkyl-heteroaryl, OR¹¹, COR¹¹, OCOR¹¹, CO₂R¹¹, NR¹²R¹³, CONR¹²R¹³, SR¹¹, SO₂R¹¹, CSR¹¹ and OSO₃R¹¹ group, the whole being possibly substituted by one or more groups chosen among an halogen atom, an OH, COOH and CHO group,

25 R^a , R^b , R^c , R^{11} , R^{12} and R^{13} being as defined above.

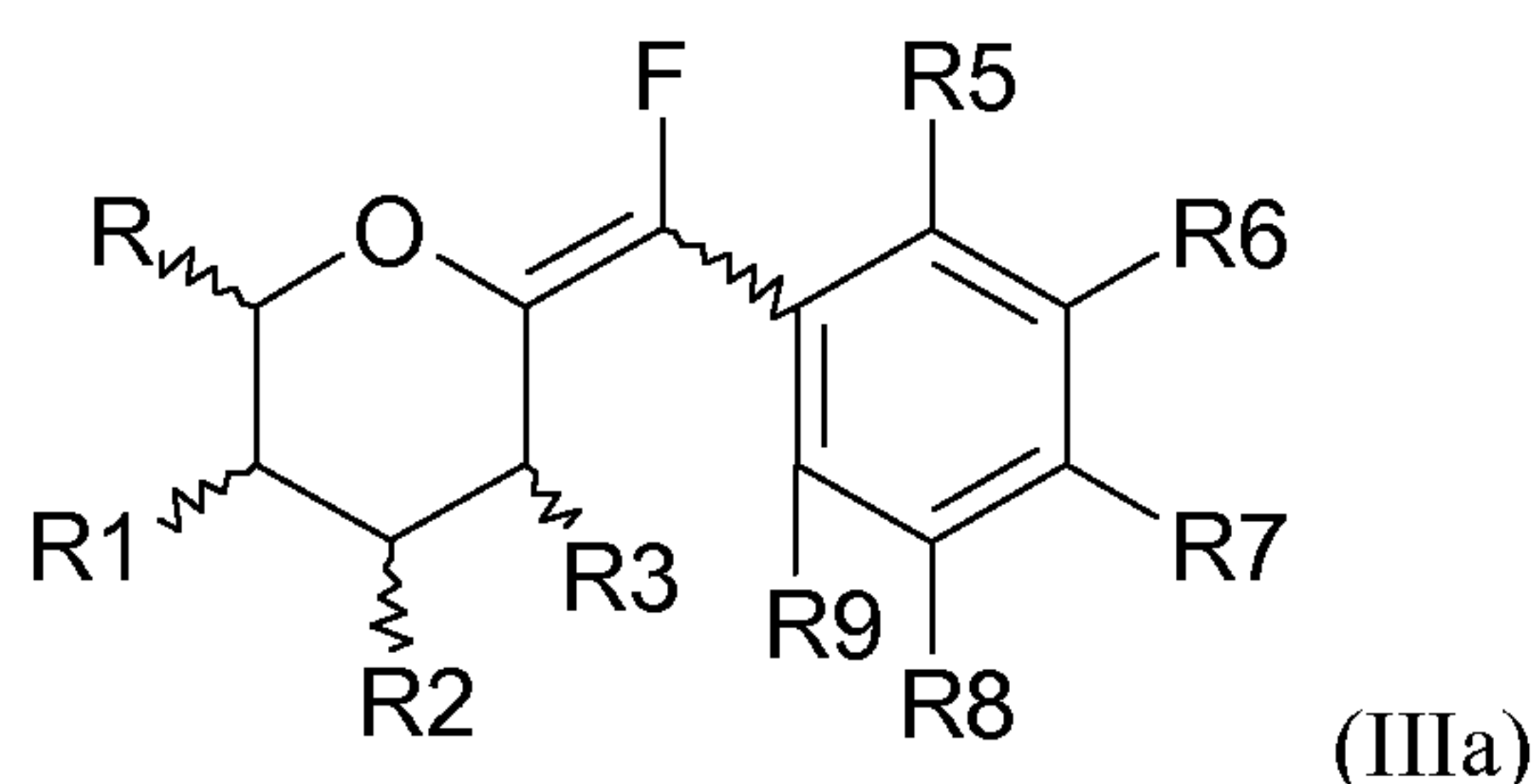
The radical A advantageously represents a phenyl group possibly substituted by one or more groups chosen among an halogen atom, a CN, SO₂, SiR^aR^bR^c, (C₁-C₆)-alkyl, (C₂-C₆)-alkenyl, (C₂-C₆)-alkynyl, (C₃-C₇)-cycloalkyl, 5 to 7 ring-membered

heterocycloalkyl, aryl, heteroaryl, aryl-(C₁-C₆)-alkyl, heteroaryl-(C₁-C₆)-alkyl, (C₁-C₆)-alkyl-aryl, (C₁-C₆)-alkyl-heteroaryl, OR¹¹, COR¹¹, OCOR¹¹, CO₂R¹¹, NR¹²R¹³, CONR¹²R¹³, SR¹¹, SO₂R¹¹, CSR¹¹ and OSO₃R¹¹ group,

the whole being possibly substituted by one or more groups chosen among an halogen atom, an OH, (C₁-C₆)-alkyl, (C₁-C₆)-alkoxy, COOH and CHO group, and notably among an halogen atom, an OH, COOH and CHO group,

R^a, R^b, R^c, R¹¹, R¹² and R¹³ being as defined above.

Consequently, according to a first particular embodiment of the invention, a compound of the invention advantageously responds to the following generic formula (IIIa):

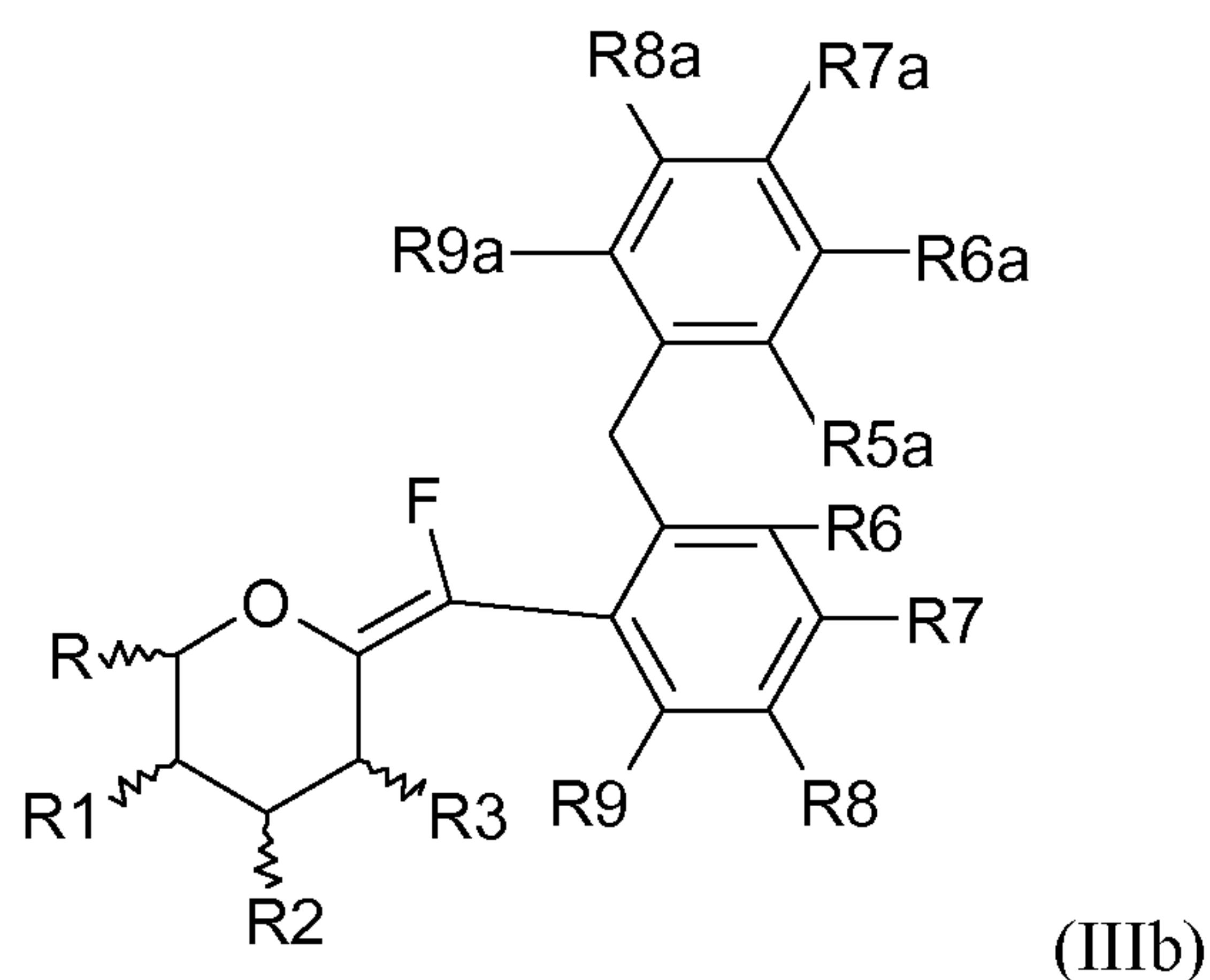


wherein:

- R₅, R₆, R₇, R₈ and R₉ represent, independently from one another, a hydrogen atom, an halogen atom, a CN, SO₂, SiR^aR^bR^c, (C₁-C₆)-alkyl, (C₂-C₆)-alkenyl, (C₂-C₆)-alkynyl, (C₃-C₇)-cycloalkyl, 5 to 7 ring-membered heterocycloalkyl, aryl, heteroaryl, aryl-(C₁-C₆)-alkyl, heteroaryl-(C₁-C₆)-alkyl, (C₁-C₆)-alkyl-aryl, (C₁-C₆)-alkyl-heteroaryl, OR¹¹, COR¹¹, OCOR¹¹, CO₂R¹¹, NR¹²R¹³, NR¹²COR¹¹, CONR¹²R¹³, SR¹¹, SO₂R¹¹, CSR¹¹ or OSO₃R¹¹ group, possibly substituted by one or more groups chosen among an halogen atom, an OH, (C₁-C₆)-alkyl, (C₁-C₆)-alkoxy, COOH and CHO group, and notably among an halogen atom, an OH, COOH and CHO group; and

- R, R₁, R₂, R₃, R¹¹, R¹² and R¹³ are as defined above.

Moreover, according to a second particular embodiment of the invention, a compound of the invention is advantageously based on the following generic formula (IIIb):



wherein:

- R6, R7, R8, R9, R5a, R6a, R7a, R8a and R9a represent, independently from one another, a hydrogen atom, an halogen atom, a CN, SO₂, SiR^aR^bR^c, (C₁-C₆)-alkyl, (C₂-C₆)-alkenyl, (C₂-C₆)-alkynyl, (C₃-C₇)-cycloalkyl, 5 to 7 ring-membered heterocycloalkyl, aryl, heteroaryl, aryl-(C₁-C₆)-alkyl, heteroaryl-(C₁-C₆)-alkyl, (C₁-C₆)-alkyl-aryl, (C₁-C₆)-alkyl-heteroaryl, OR¹¹, COR¹¹, OCOR¹¹, CO₂R¹¹, NR¹²R¹³, NR¹²COR¹¹, CONR¹²R¹³, SR¹¹, SO₂R¹¹, CSR¹¹ or OSO₃R¹¹ group, the said group being possibly substituted by one or more groups chosen among an halogen atom, an OH, (C₁-C₆)-alkyl, (C₁-C₆)-alkoxy, COOH and CHO group; and in particular by one or more groups chosen among an halogen atom, an OH, COOH and CHO group, and

- X, R, R1, R2, R3, R¹¹, R¹² and R¹³ are as defined above.

Preferably, R1, R2 and R3 represent, independently from one another, a fluorine atom or an OH, OSiR^aR^bR^c, OR¹¹, OCOR¹¹, OCO₂R¹¹ or OCONR¹²R¹³ group.

R1, R2 and R3 may advantageously be chosen, independently from one another, among an OH, OR¹¹ and OCOR¹¹ group with R¹¹ as defined above.

Even more advantageously, R1, R2 and R3 may be chosen, independently from one another, among an OH, -O-(C₁-C₆)-alkyl, -O-aryl, -O-(C₁-C₆)-alkyl-aryl and -OCO-(C₁-C₆)-alkyl group.

In particular, R1, R2 and R3 may be chosen, independently from one another, among an OH, OSiMe₃ and benzyloxy (OBn) group, and preferably among OH and OBn.

According to a particular embodiment, R1, R2 and R3 are identical.

Advantageously, R represents a hydrogen atom or a CH₃, CH₂OH, CH₂OSiR^aR^bR^c, CH₂OR¹¹, CH₂OCOR¹¹, CH₂OP(O)(OH)₂ or CH₂OSO₃H group, and

more advantageously a hydrogen atom or a CH_3 , CH_2OH , $\text{CH}_2\text{OR}^{11}$, $\text{CH}_2\text{OCOR}^{11}$, $\text{CH}_2\text{OP(O)(OH)}_2$ or $\text{CH}_2\text{OSO}_3\text{H}$ group,

with R^a , R^b , R^c and R^{11} as defined above, and with $\text{CH}_2\text{OR}^{11}$ advantageously representing a $-\text{CH}_2\text{O}-(\text{C}_1-\text{C}_6)\text{-alkyl}$, $-\text{CH}_2\text{O-aryl}$ and $-\text{CH}_2\text{O}-(\text{C}_1-\text{C}_6)\text{-alkyl-aryl}$ group,
 5 and $\text{CH}_2\text{OCOR}^{11}$ advantageously representing a $-\text{CH}_2\text{OCO}-(\text{C}_1-\text{C}_6)\text{-alkyl}$ group.

Even more advantageously, R represents a CH_2OH , $\text{CH}_2\text{OSiR}^a\text{R}^b\text{R}^c$, $\text{CH}_2\text{OR}^{11}$ or $\text{CH}_2\text{OCOR}^{11}$ group, and more advantageously a CH_2OH , $\text{CH}_2\text{OR}^{11}$ or $\text{CH}_2\text{OCOR}^{11}$ group, with R^a , R^b , R^c and R^{11} as defined above.

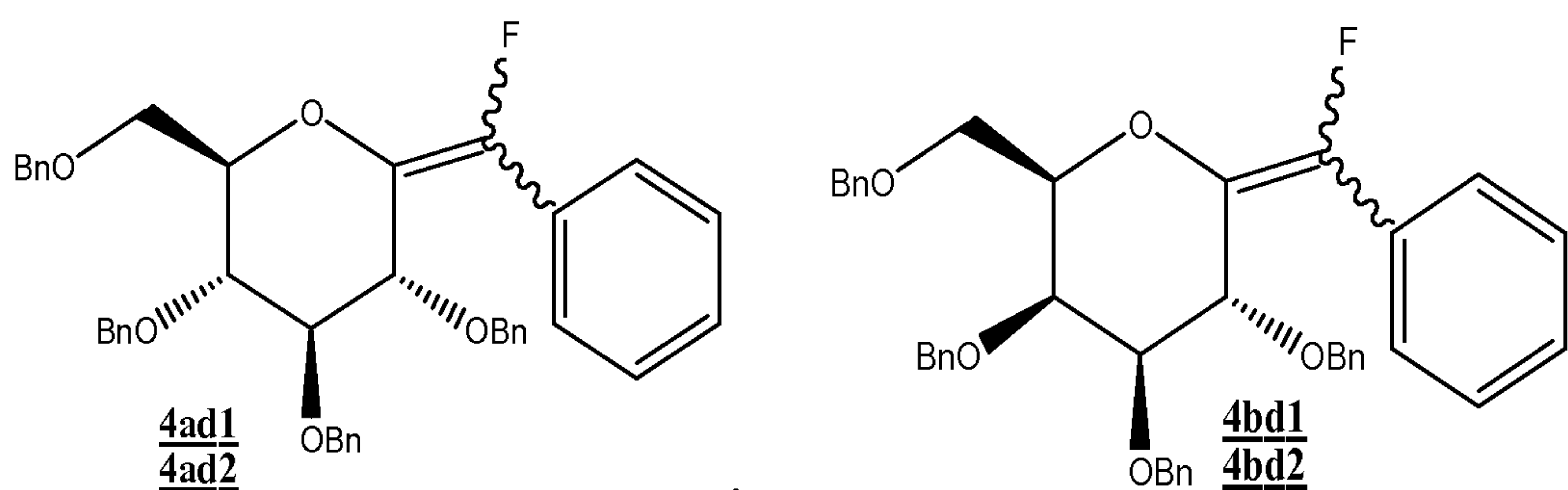
Even more advantageously, R represents a CH_2OH , $-\text{CH}_2\text{O}-(\text{C}_1-\text{C}_6)\text{-alkyl}$, $-\text{CH}_2\text{O-aryl}$, $-\text{CH}_2\text{O}-(\text{C}_1-\text{C}_6)\text{-alkyl-aryl}$ and $-\text{CH}_2\text{OCO}-(\text{C}_1-\text{C}_6)\text{-alkyl}$ group.
 10

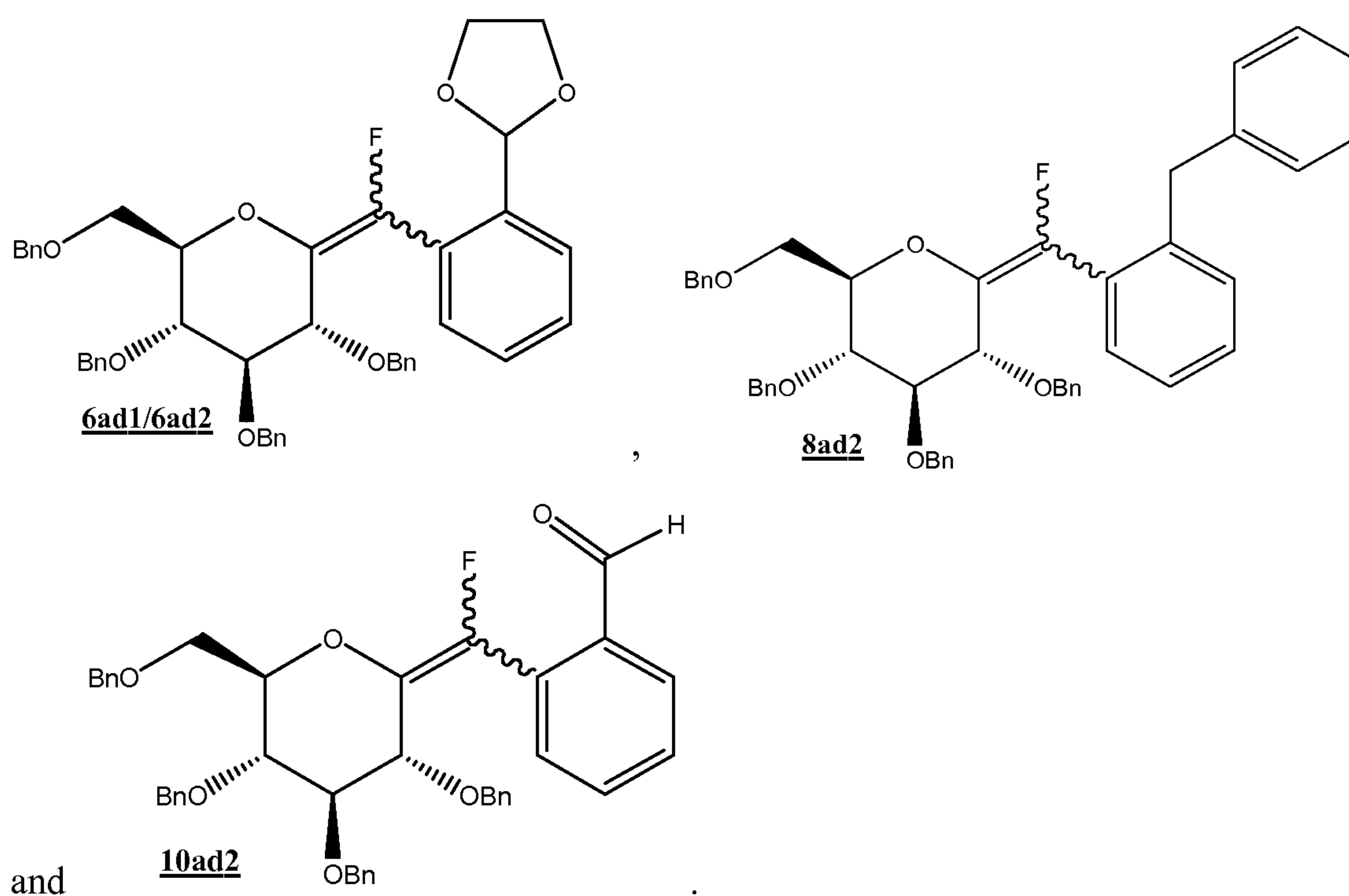
In particular, R can represent a CH_2OH , $\text{CH}_2\text{OSiMe}_3$ or CH_2OBn group and preferably a CH_2OH or CH_2OBn group.

R_5 , R_6 , R_7 , R_8 , R_9 , R_{5a} , R_{6a} , R_{7a} , R_{8a} and R_{9a} can be chosen among a hydrogen atom, an halogen atom, advantageously a chlorine atom, an aryl- $(\text{C}_1-\text{C}_6)\text{-alkyl}$ group, advantageously benzyl, the alkyl group being possibly substituted by an OH group.
 15

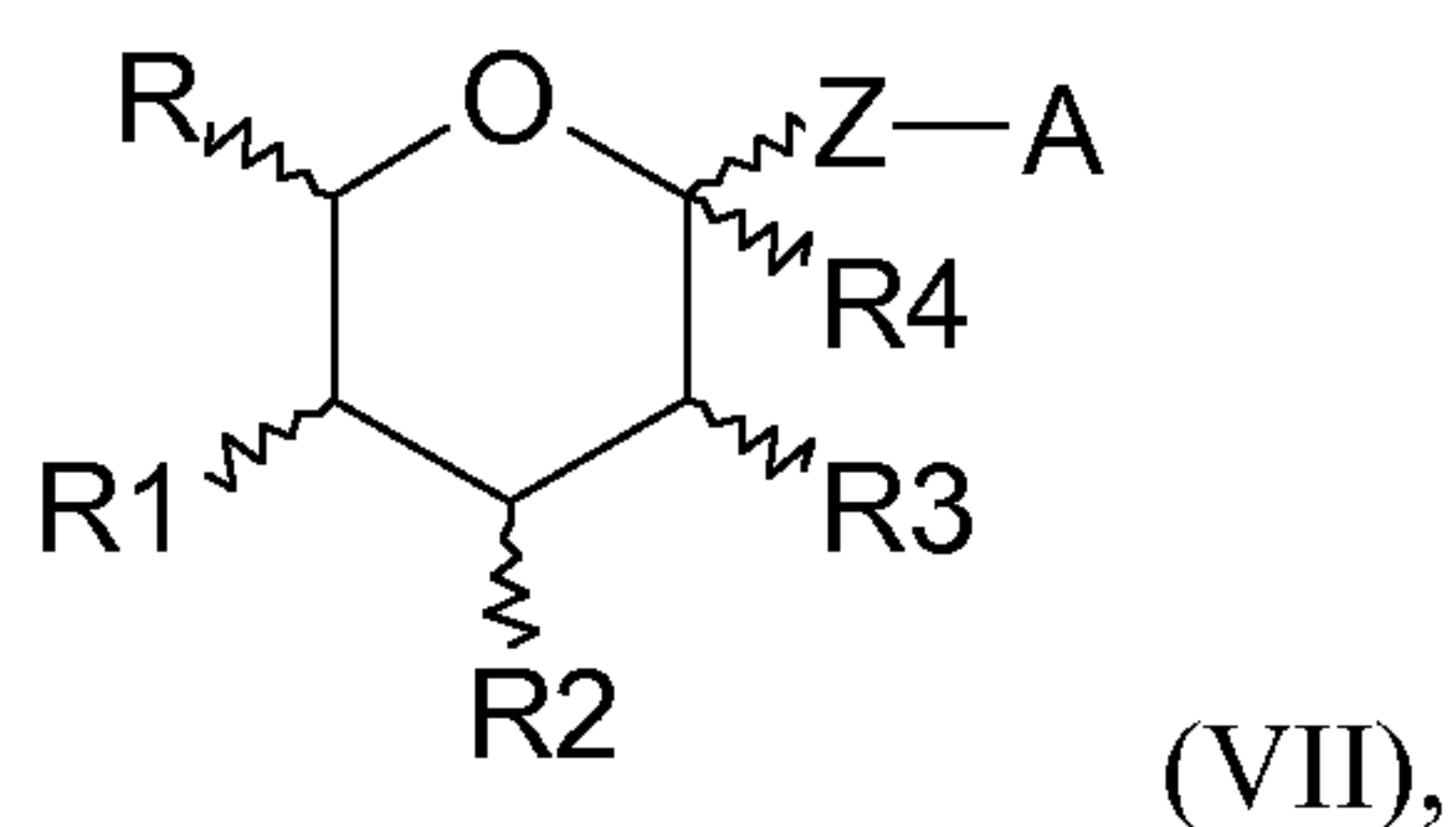
Advantageously, R_5 , R_6 , R_7 , R_8 , R_9 , R_{5a} , R_{6a} , R_{7a} , R_{8a} and R_{9a} will be chosen, independently from one another, among a hydrogen atom, a halogen atom, advantageously a chlorine or fluorine atom, a CHO group, a 5 to 7 ring-membered heterocycloalkyl group, an aryl- $(\text{C}_1-\text{C}_6)\text{-alkyl}$, such as benzyl, or aryl-CO-, such as benzoyl, group, the alkyl moiety of said group being possibly substituted by an OH group and the aryl moiety of said group being possibly substituted by an halogen atom, such as fluorine, a $(\text{C}_1-\text{C}_6)\text{-alkyl}$ or $(\text{C}_1-\text{C}_6)\text{-alkoxy}$ group.
 20

In particular, the compounds of the invention can be chosen among the
 25 following molecules:





Another object of the present invention is a compound of formula (VII):

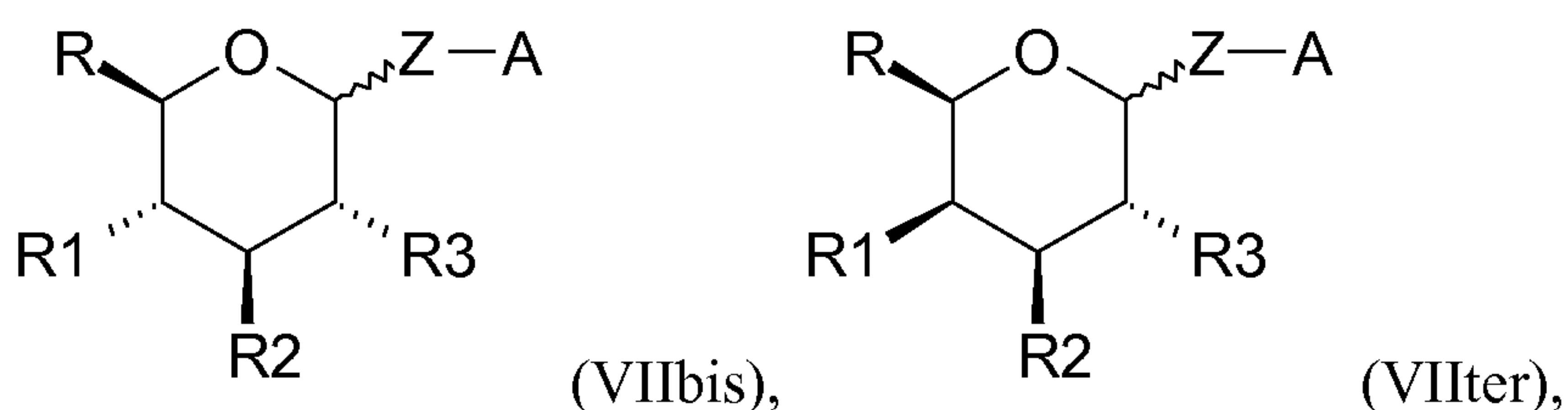


or a pharmaceutically acceptable salt thereof, a tautomer, an isomer or a mixture of isomers in any proportions, in particular a mixture of enantiomers, and particularly a racemate mixture

wherein A, Z, R, R1, R2, R3 and R4 are as defined above.

The compounds of formula (VII) are useful, in particular, as synthesis intermediates of the compounds of formula (I).

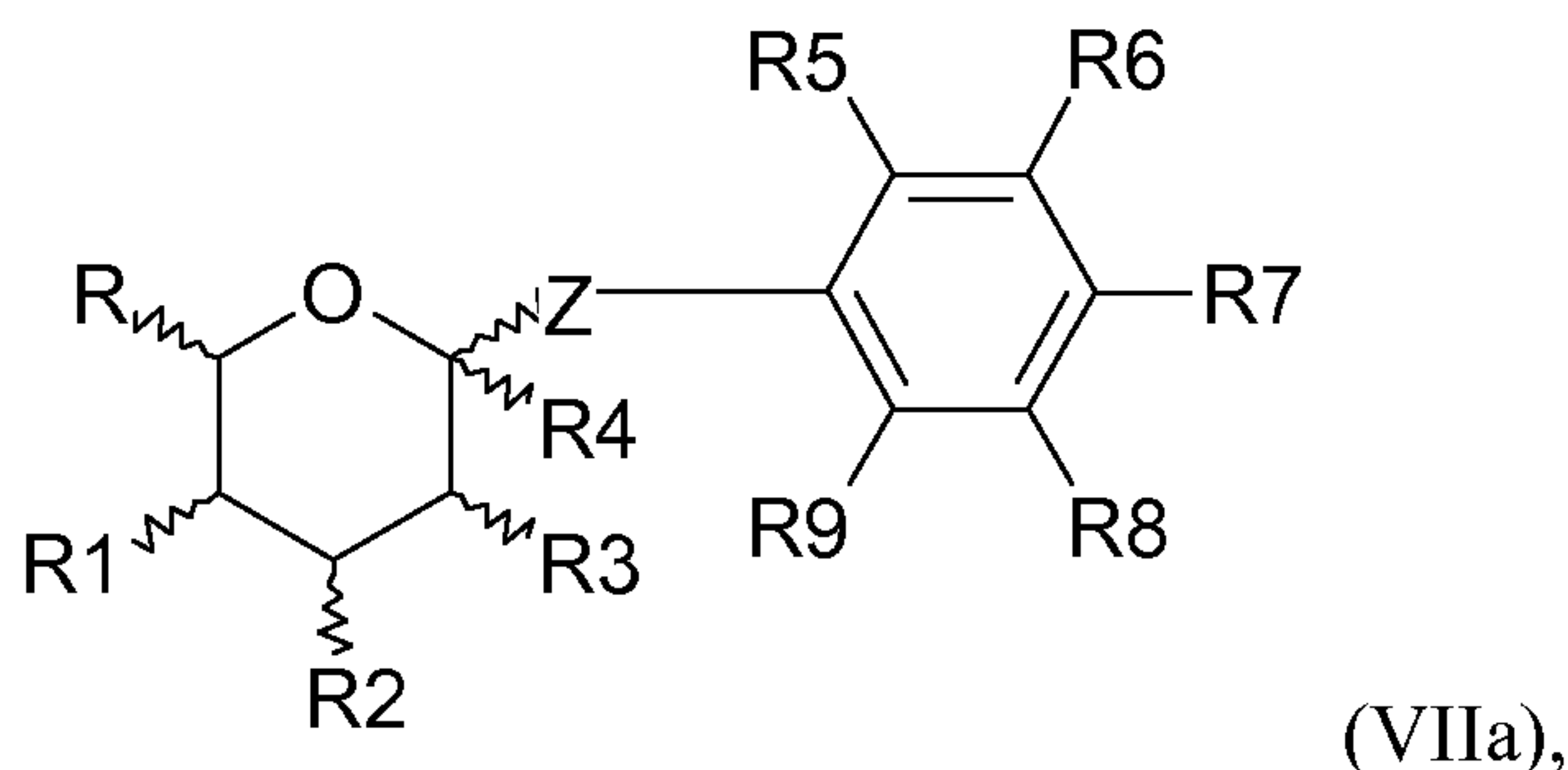
The preceding compounds respond advantageously to the formula (VIIbis) or (VIIter) below:



with R, R1, R2, R3, R4, Z and A as defined previously.

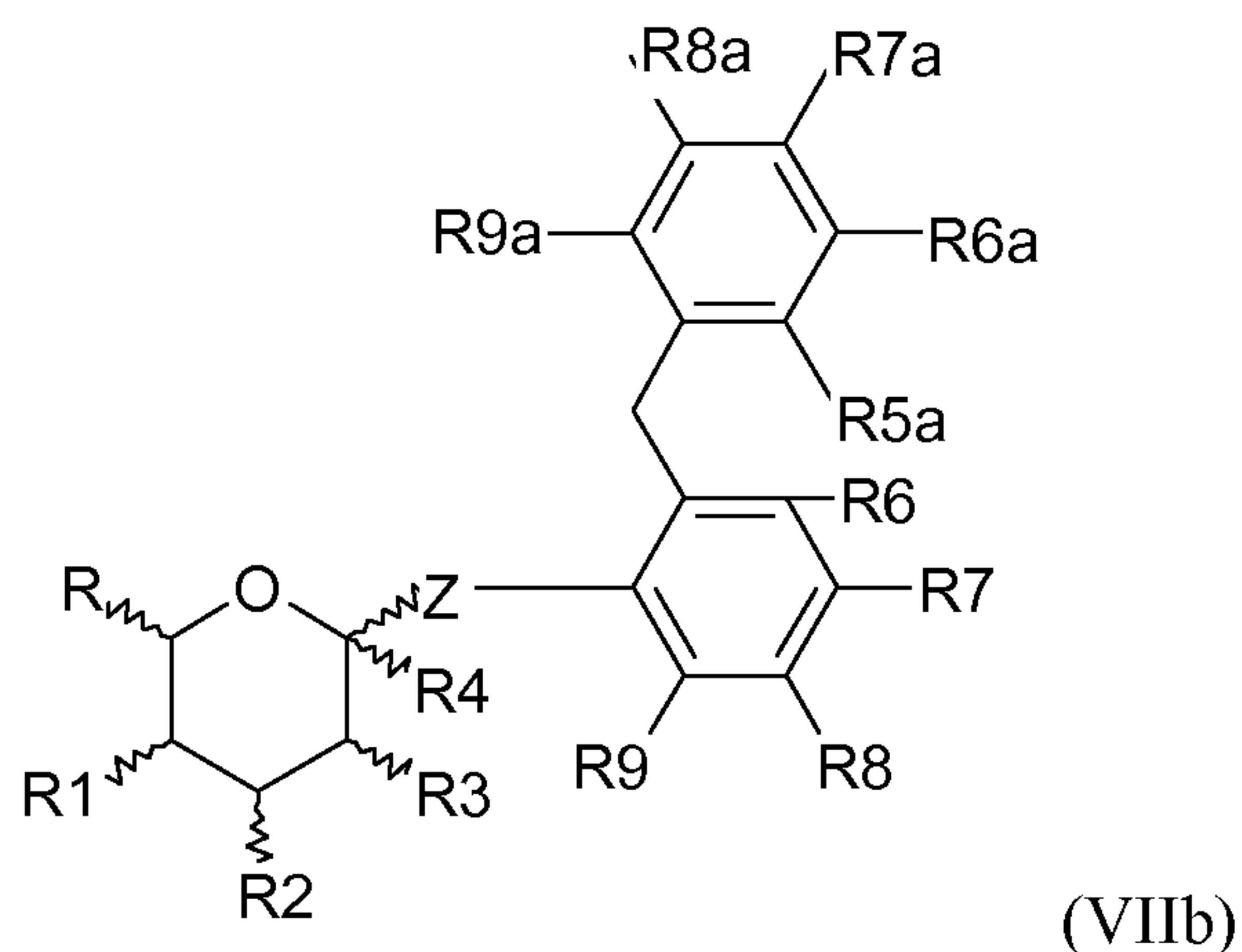
The preceding compounds respond advantageously to the formula (VIIbis).

According to a first particular embodiment of the invention, the compound is advantageously based on the following generic formula (VIIa):



with R, R1, R2, R3, R4, R5, R6, R7, R8, R9 and Z as defined above.

According to a second particular embodiment of the invention, the compound is advantageously based on the following generic formula (VIIb),:



with R, R1, R2, R3, R4, R6, R7, R8, R9, R5a, R6a, R7a, R8a, R9a and Z as defined above.

Preferably, R1, R2 and R3 represent, independently from one another, a fluorine atom or an OH, OSiR^aR^bR^c, OR¹¹, OCOR¹¹, OCO₂R¹¹ or OCONR¹²R¹³ group.

R1, R2 and R3 may advantageously be chosen, independently from one another, among an OH, OSiR^aR^bR^c, OR¹¹ and OCOR¹¹ group, and even more advantageously among an OH, OR¹¹ and OCOR¹¹ group, with R^a, R^b, R^c and R¹¹ as defined above.

Even more advantageously, R1, R2 and R3 may be chosen, independently from one another, among an OH, -O-(C₁-C₆)-alkyl, -O-aryl, -O-(C₁-C₆)-alkyl-aryl and -OCO-(C₁-C₆)-alkyl group.

In particular, R1, R2 and R3 may be chosen, independently from one another, among an OH, OSiMe₃ and benzyloxy (OBn) group, and preferably among OH and Obn.

According to a particular embodiment, R1, R2 and R3 are identical.

5 Advantageously, R represents a hydrogen atom or a CH₃, CH₂OH, CH₂OSiR^aR^bR^c, CH₂OR¹¹, CH₂OCOR¹¹, CH₂OP(O)(OH)₂ or CH₂OSO₃H group, with R^a, R^b, R^c and R¹¹ as defined above, and with CH₂OR¹¹ advantageously representing a -CH₂O-(C₁-C₆)-alkyl, -CH₂O-aryl and -CH₂O-(C₁-C₆)-alkyl-aryl group, and CH₂OCOR¹¹ advantageously representing a -CH₂OCO-(C₁-C₆)-alkyl group.

10 Even more advantageously, R represents a CH₂OH, CH₂OSiR^aR^bR^c, CH₂OR¹¹ or CH₂OCOR¹¹ group, with R^a, R^b, R^c and R¹¹ as defined above.

Even more advantageously, R represents a CH₂OH, -CH₂O-(C₁-C₆)-alkyl, -CH₂O-aryl, -CH₂O-(C₁-C₆)-alkyl-aryl and -CH₂OCO-(C₁-C₆)-alkyl group.

15 In particular, R can represent a CH₂OH, CH₂OSiMe₃ or CH₂OBn group, and preferably a CH₂OH or CH₂OBn group.

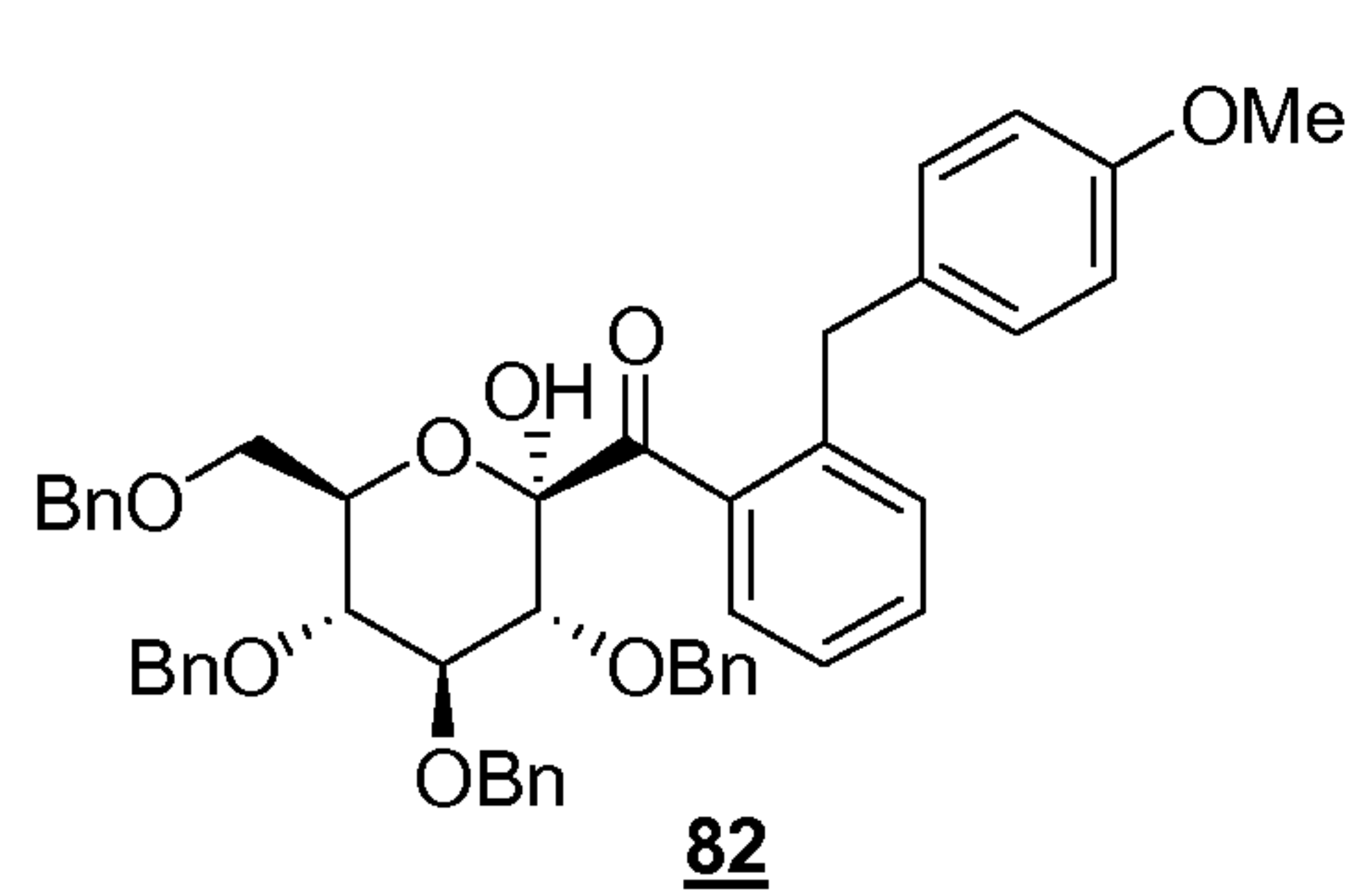
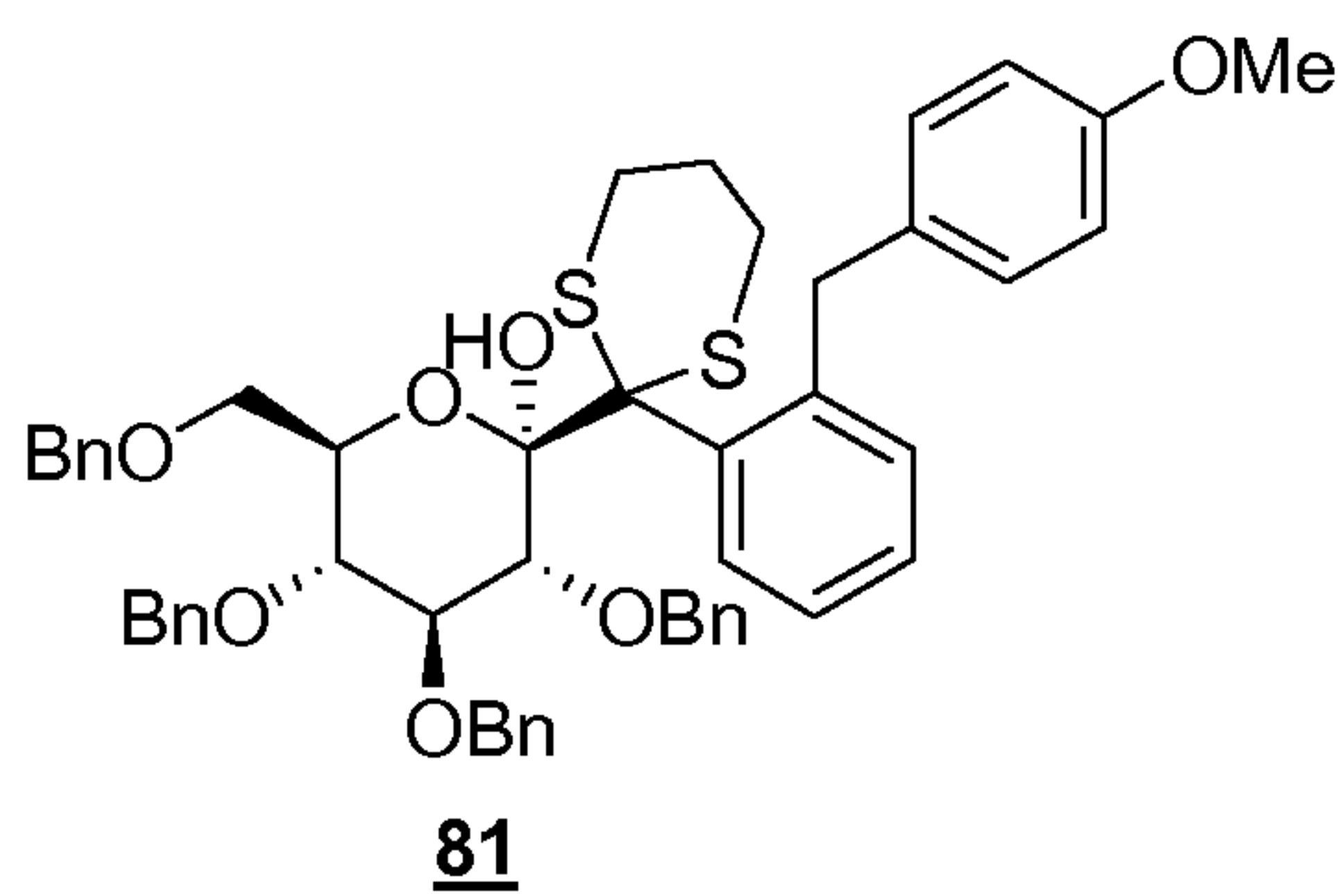
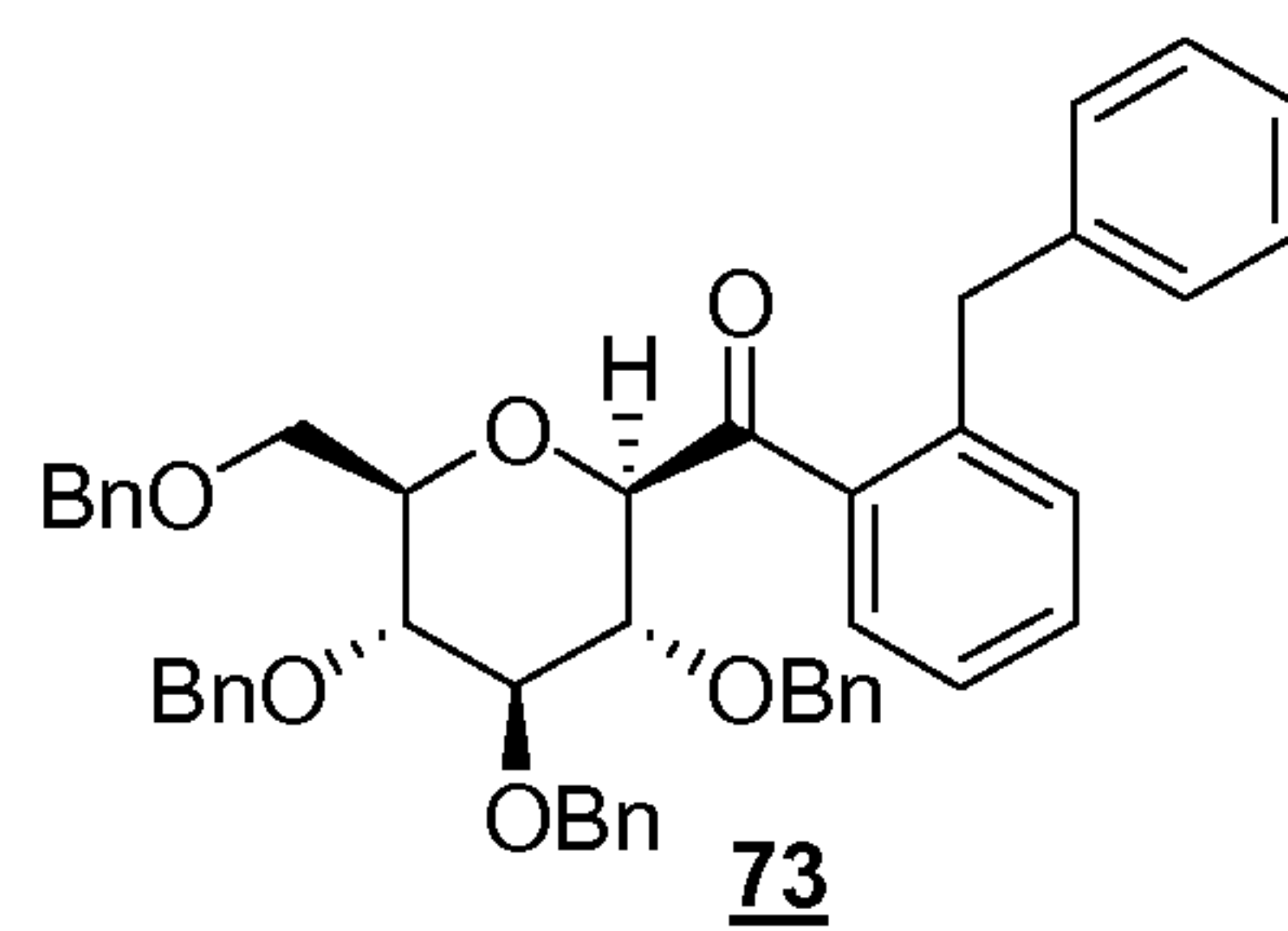
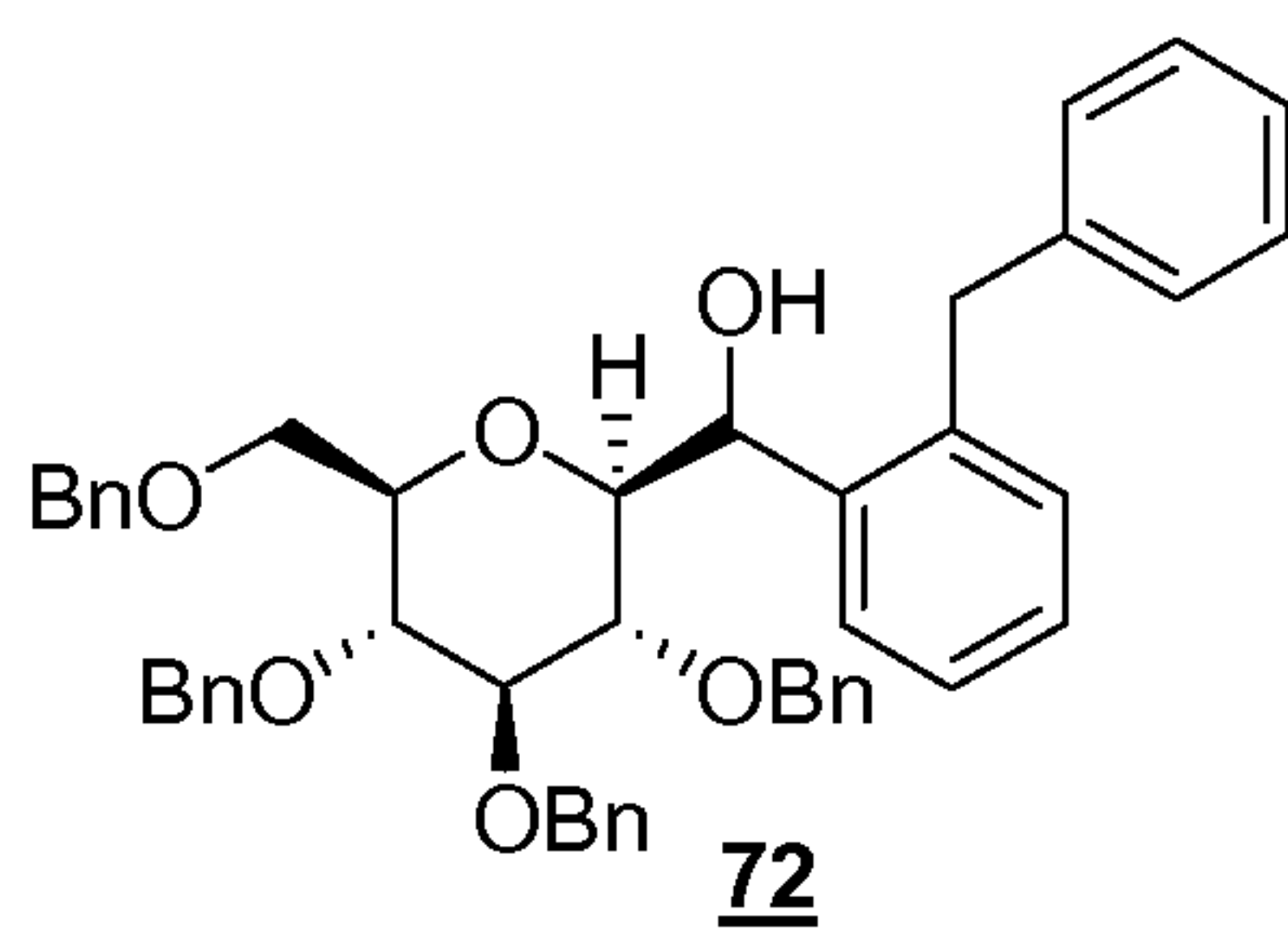
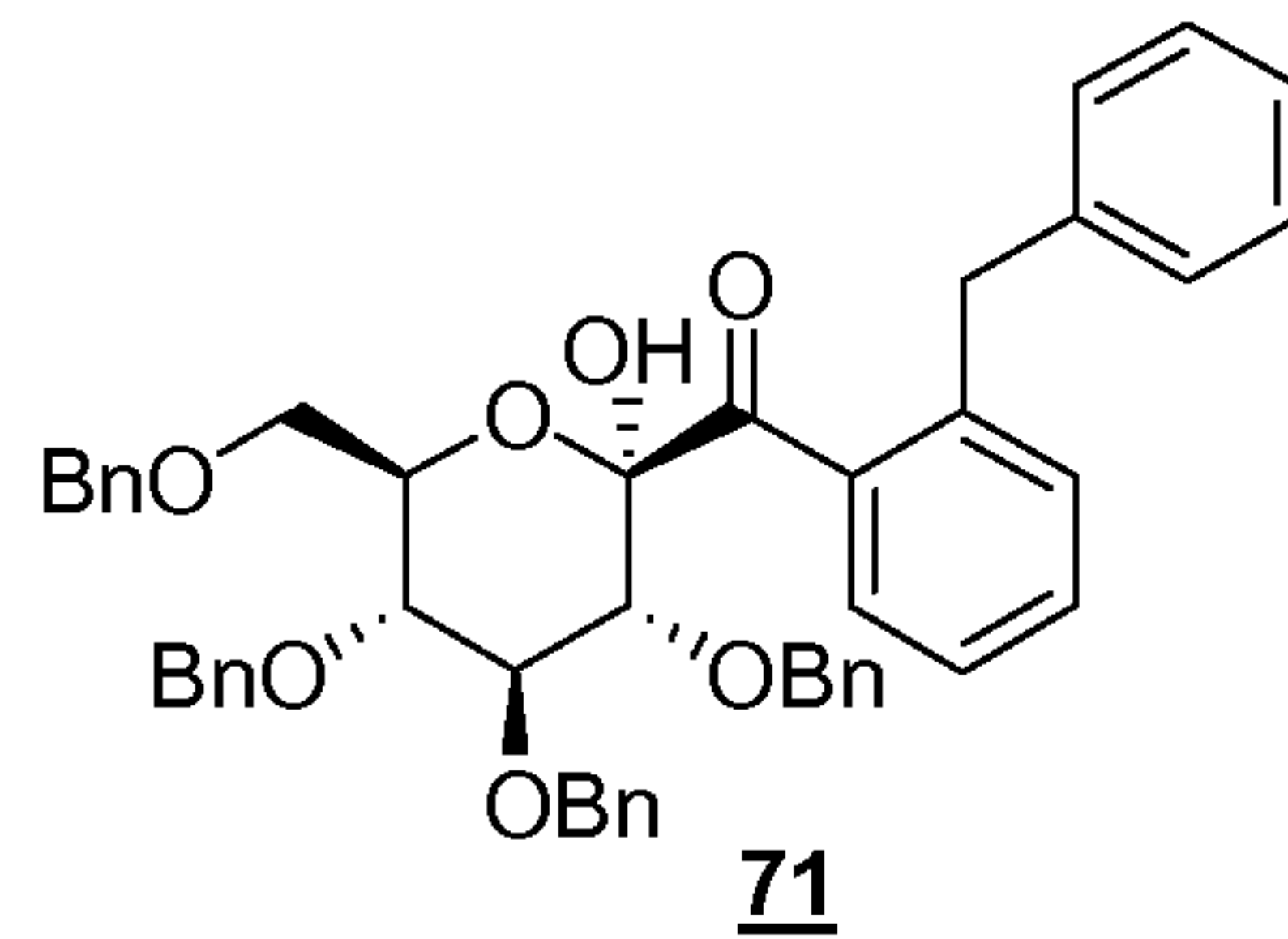
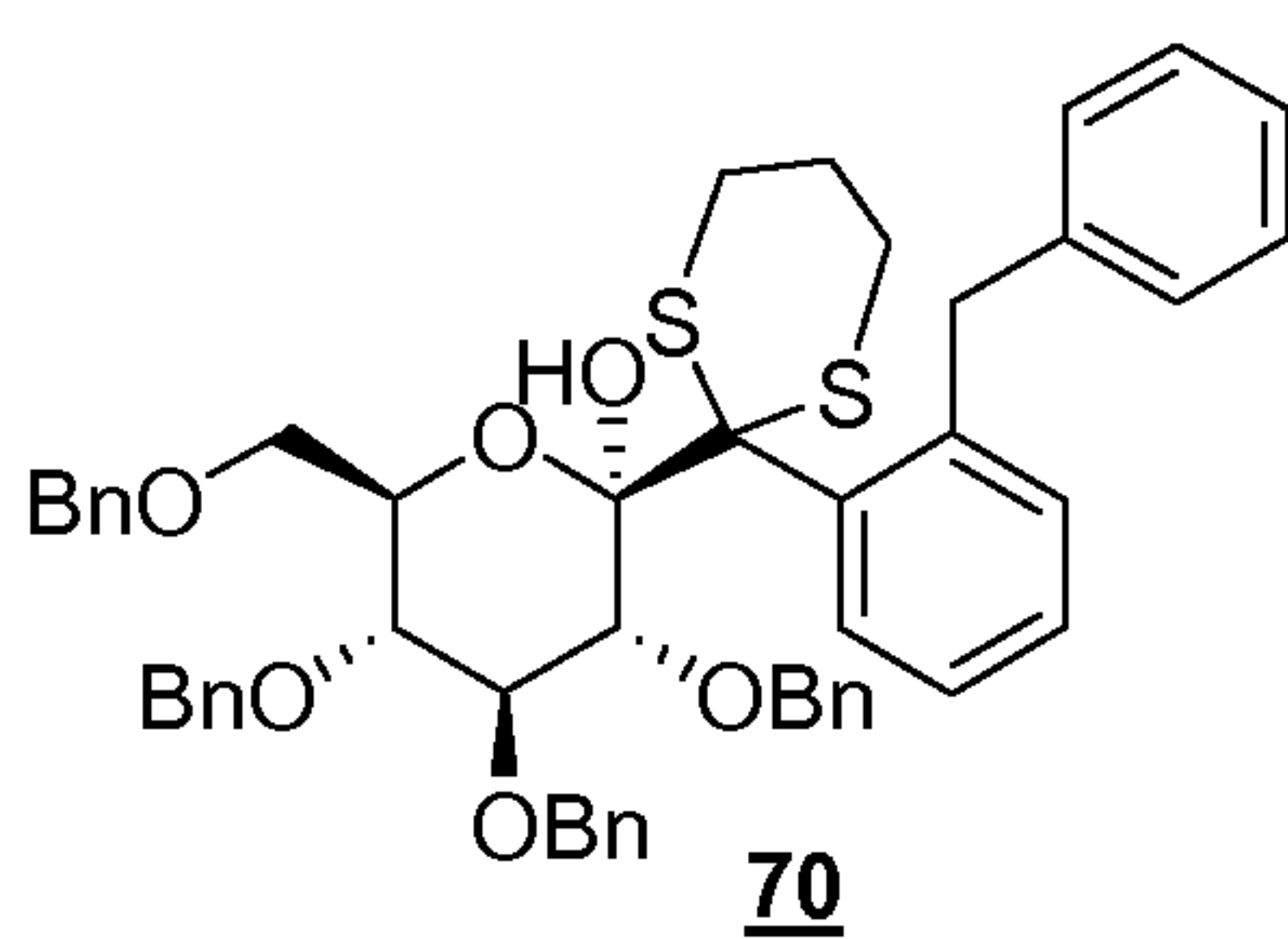
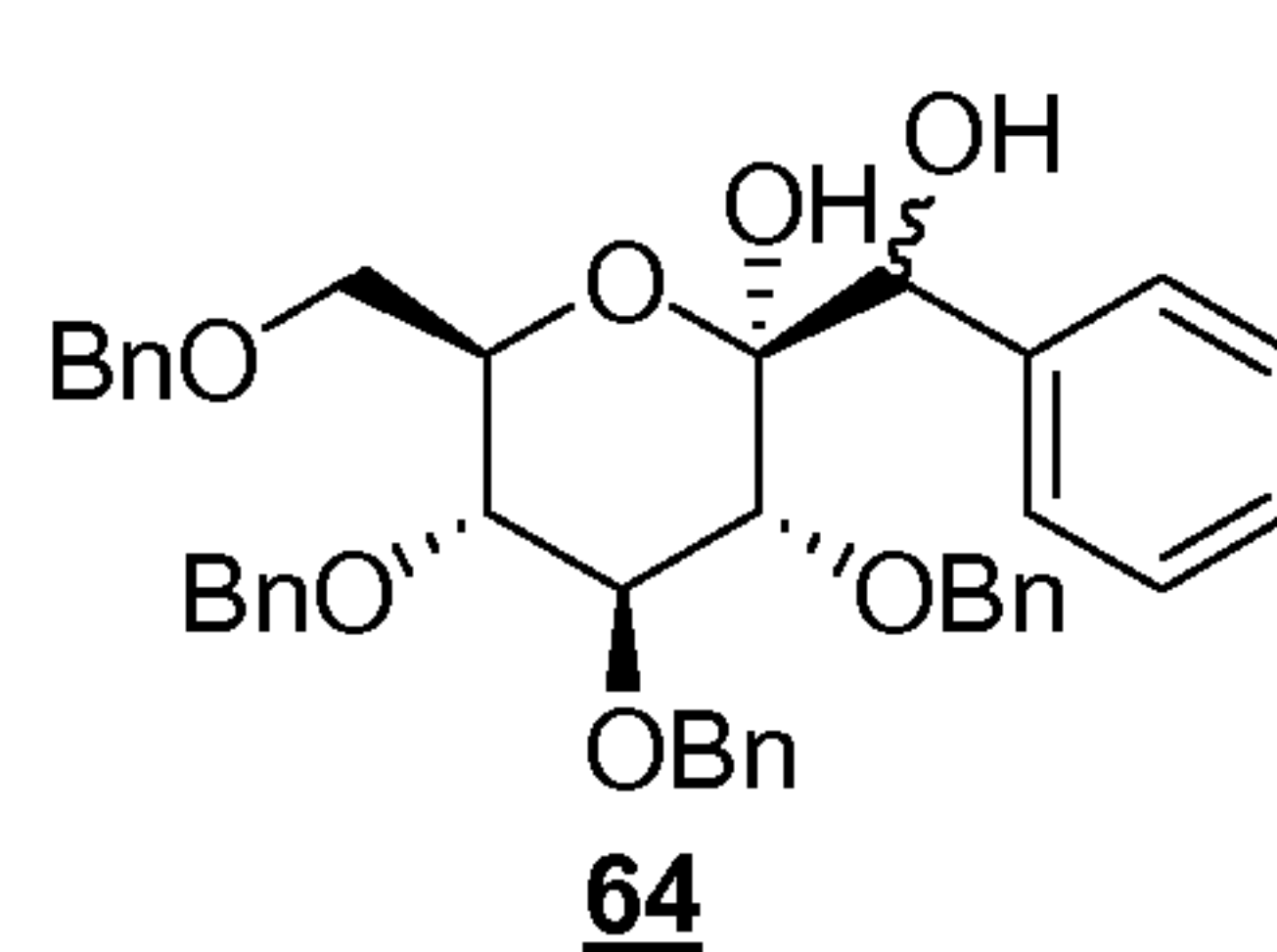
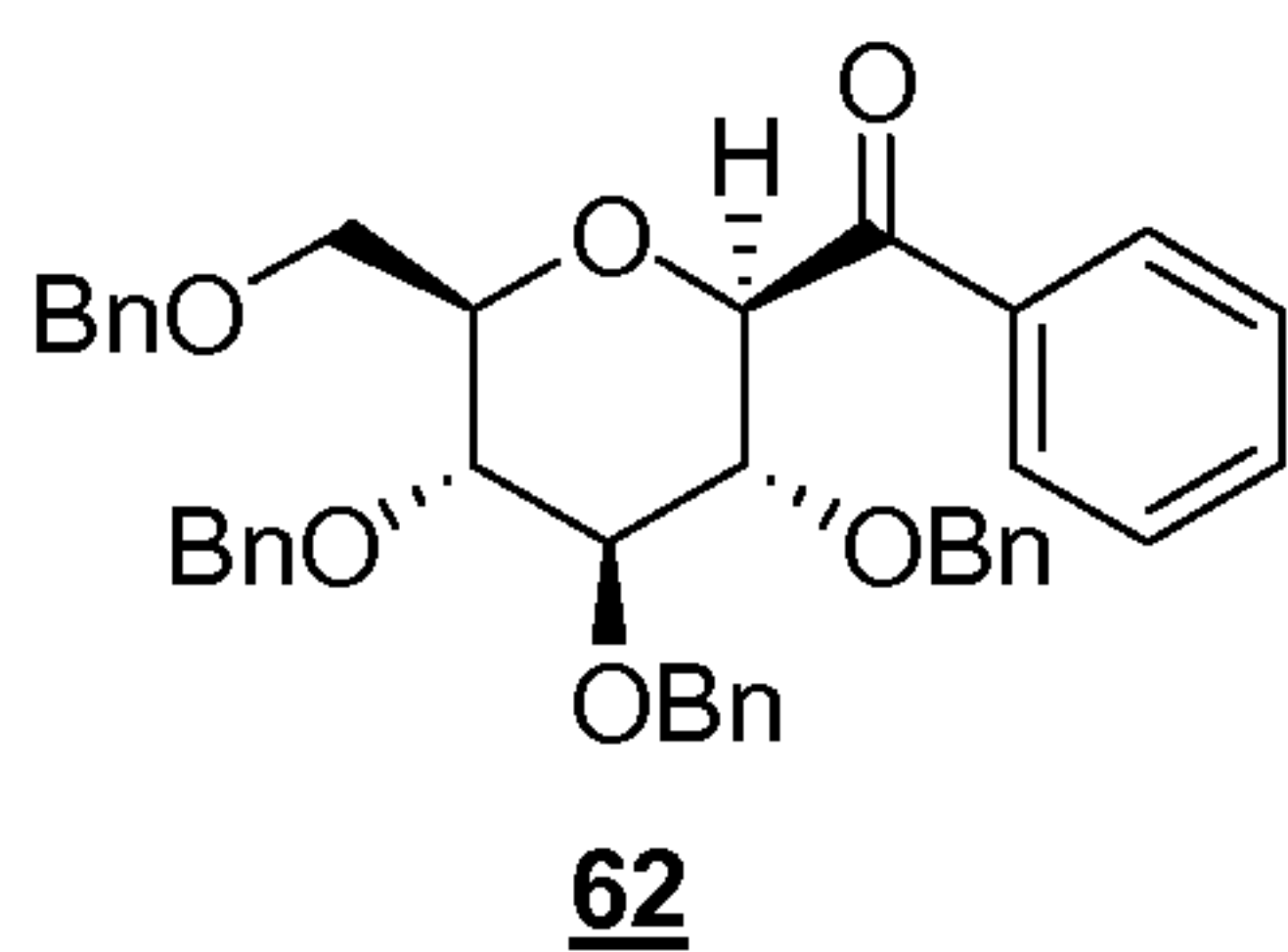
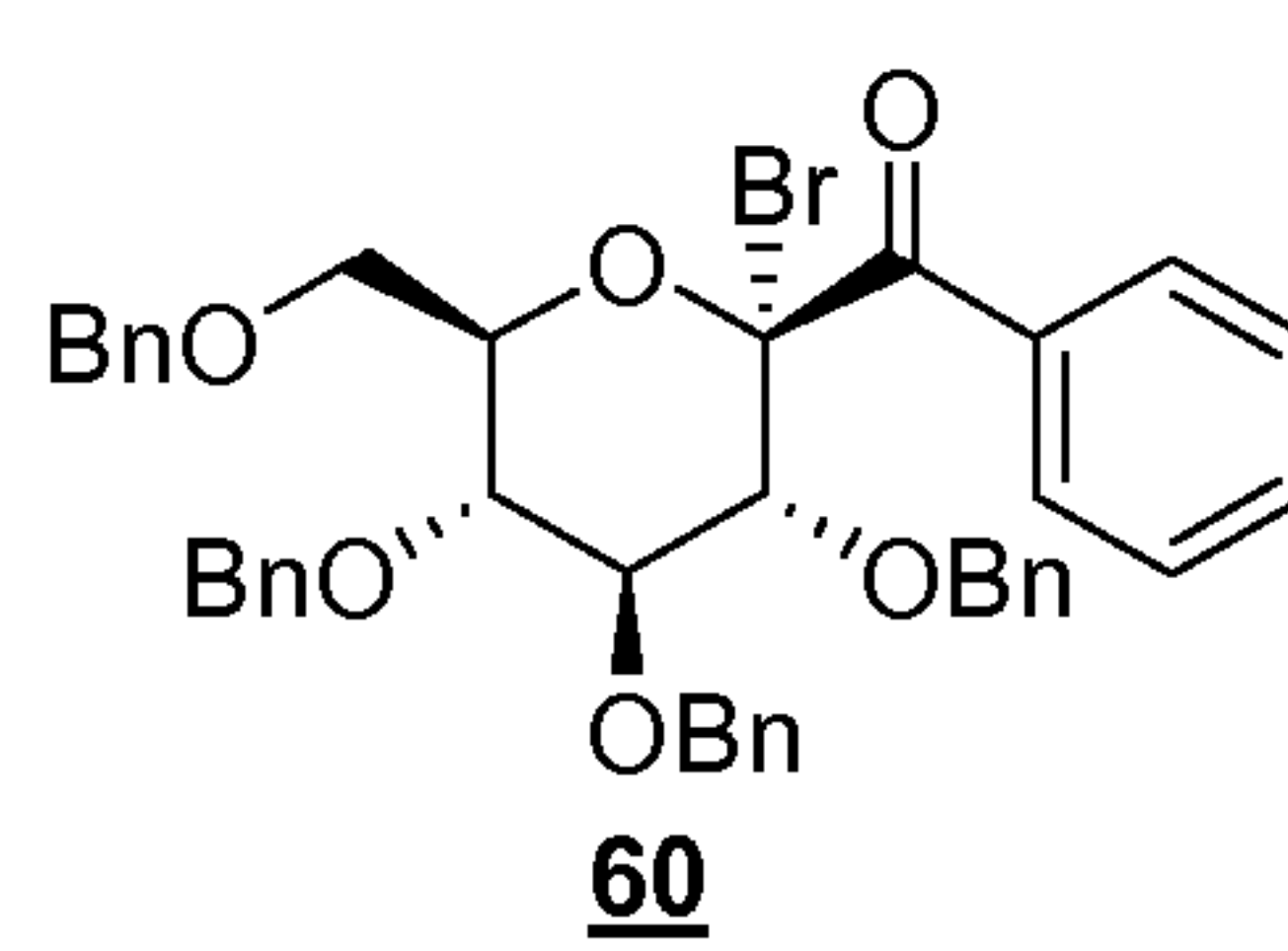
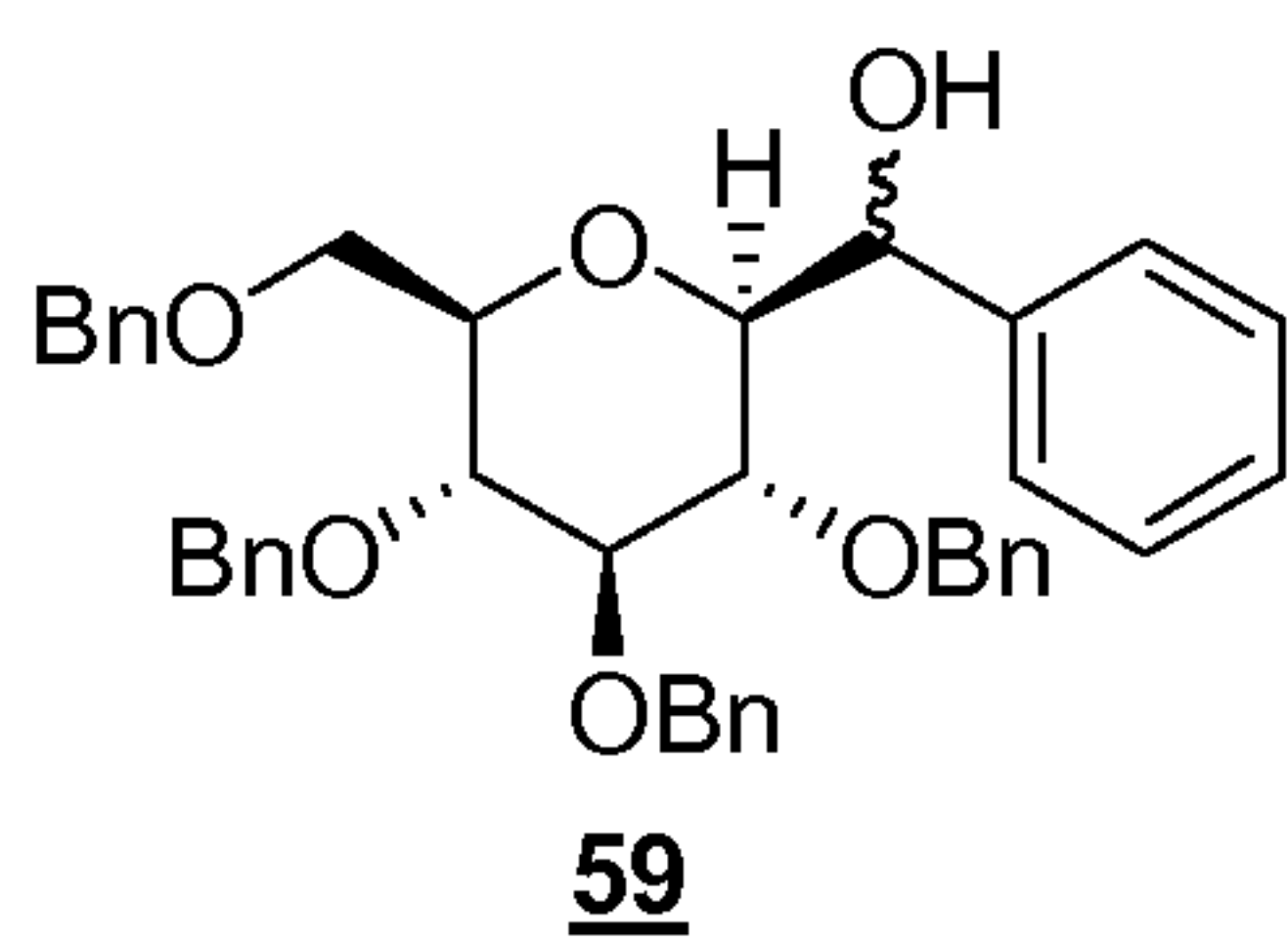
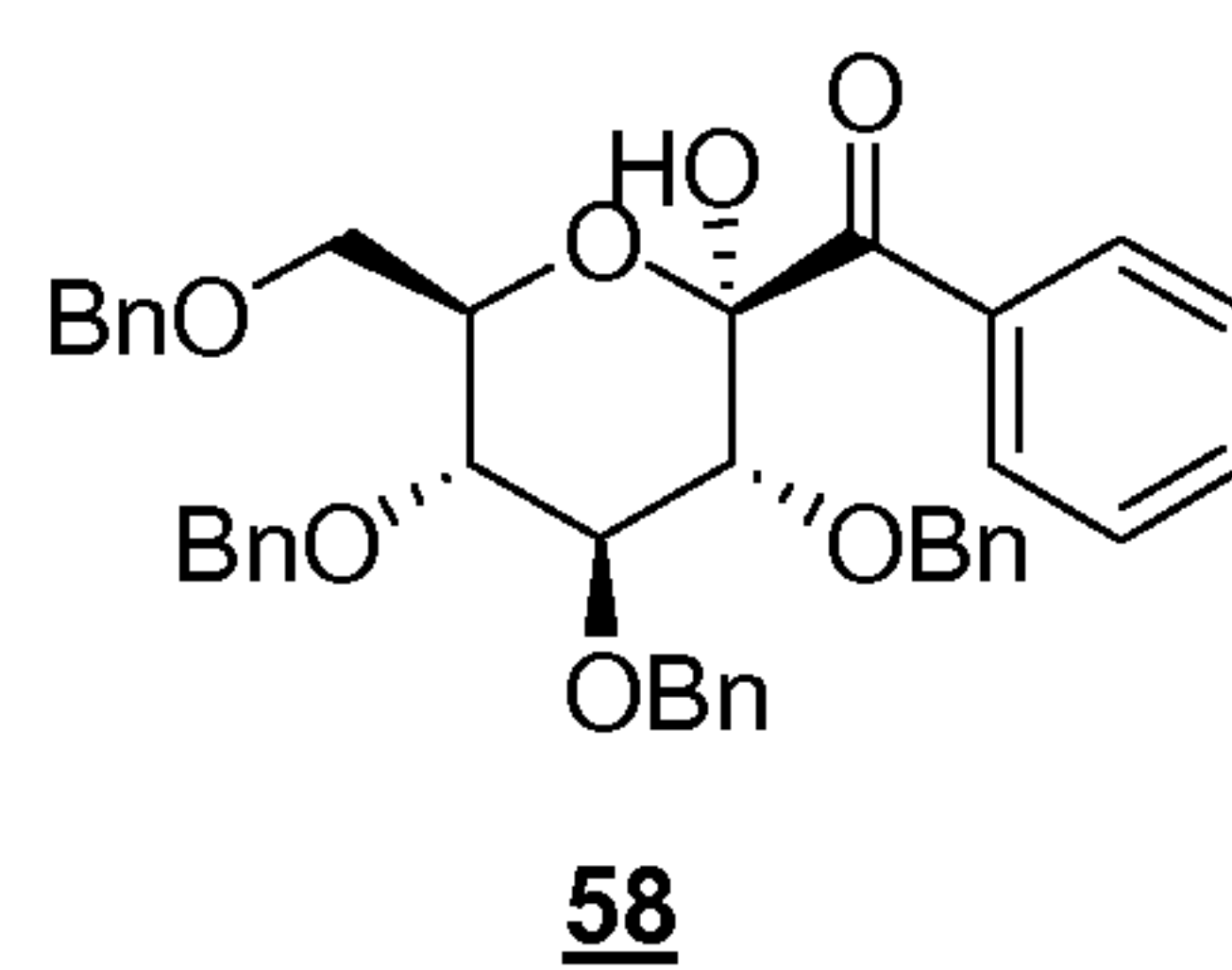
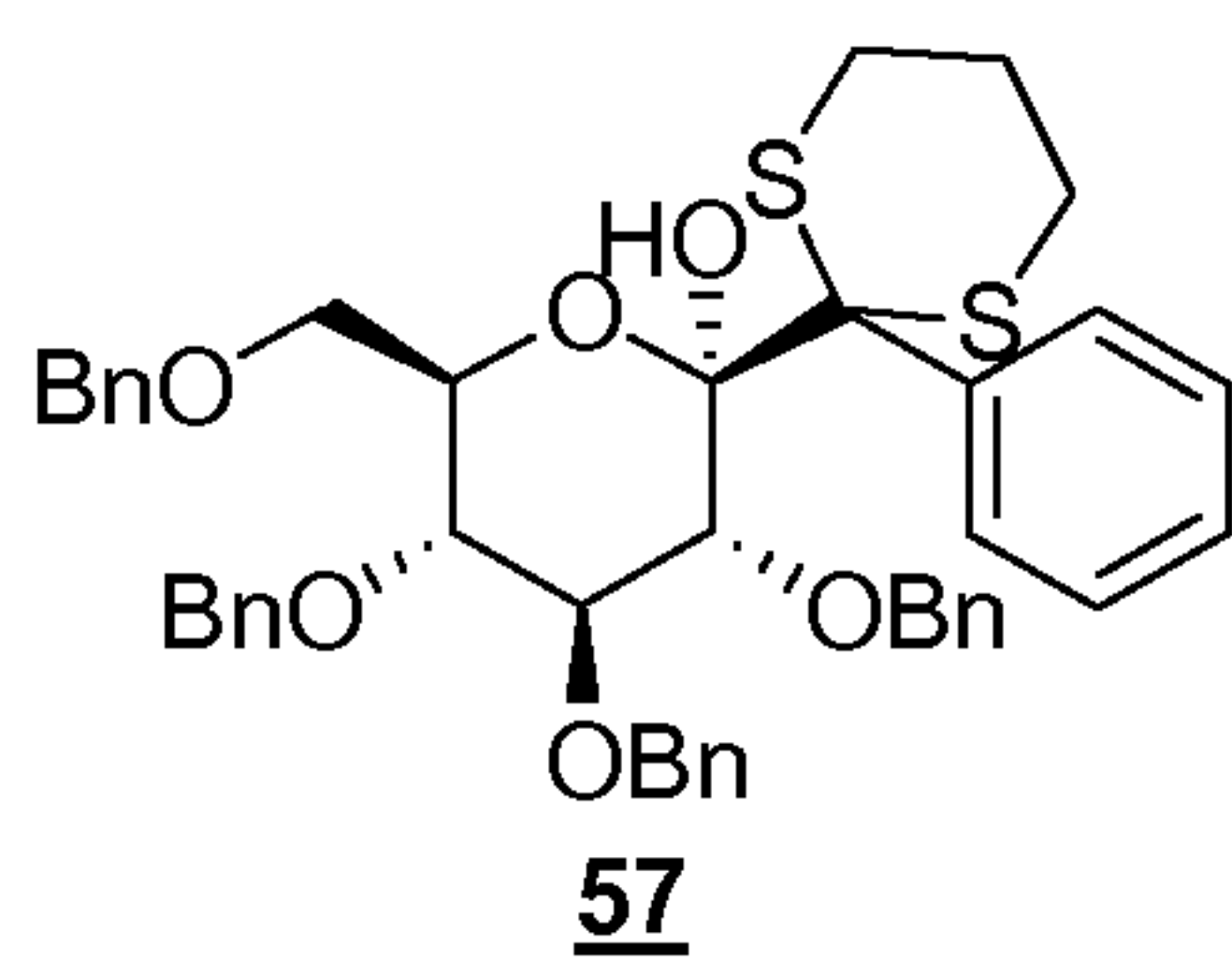
In the same way, R4 may advantageously represent a hydrogen or halogen atom or an OH or OR¹¹ group, and in particular a hydrogen atom or an OH or OR¹¹ group, with R¹¹ as defined above.

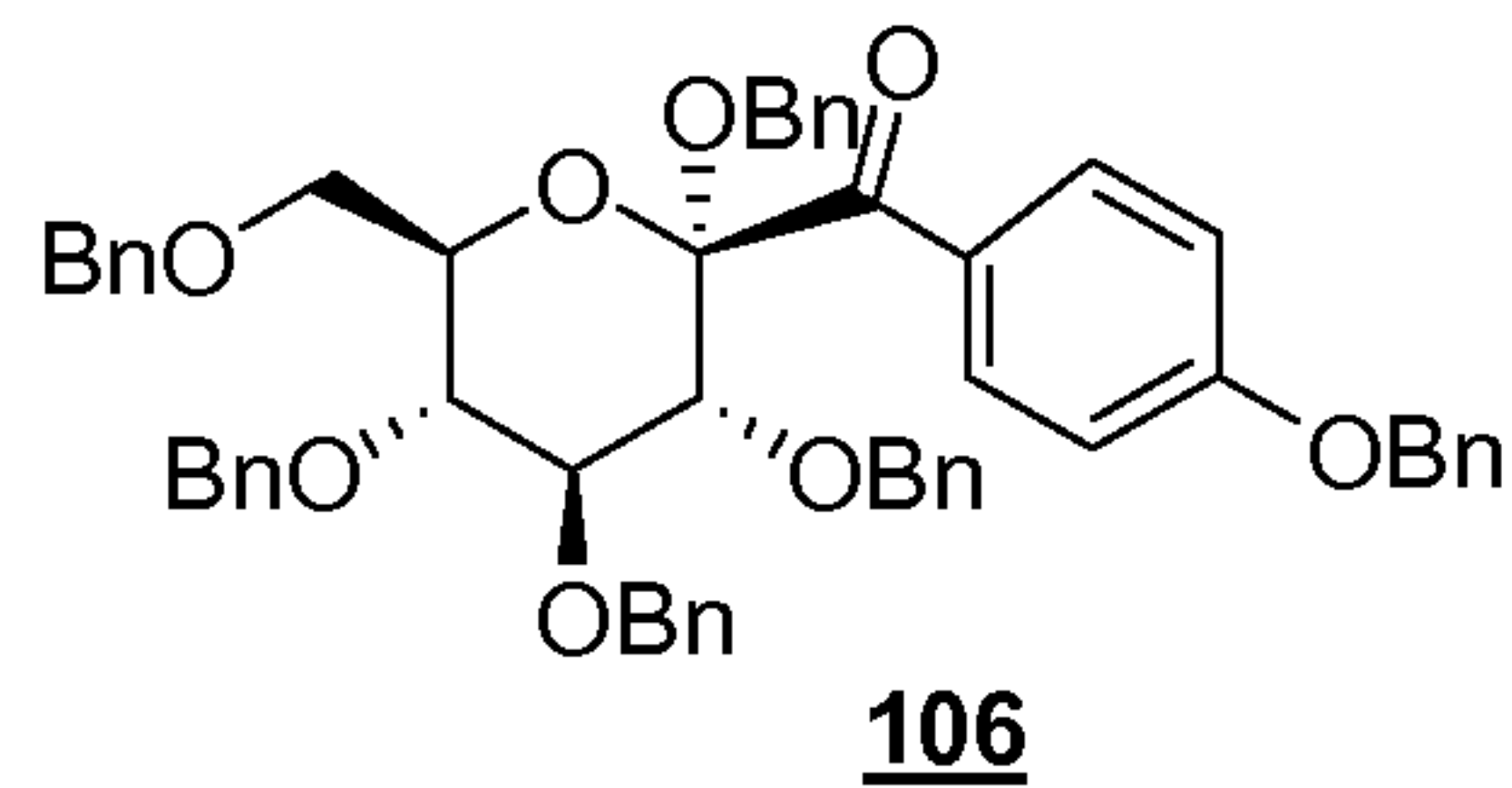
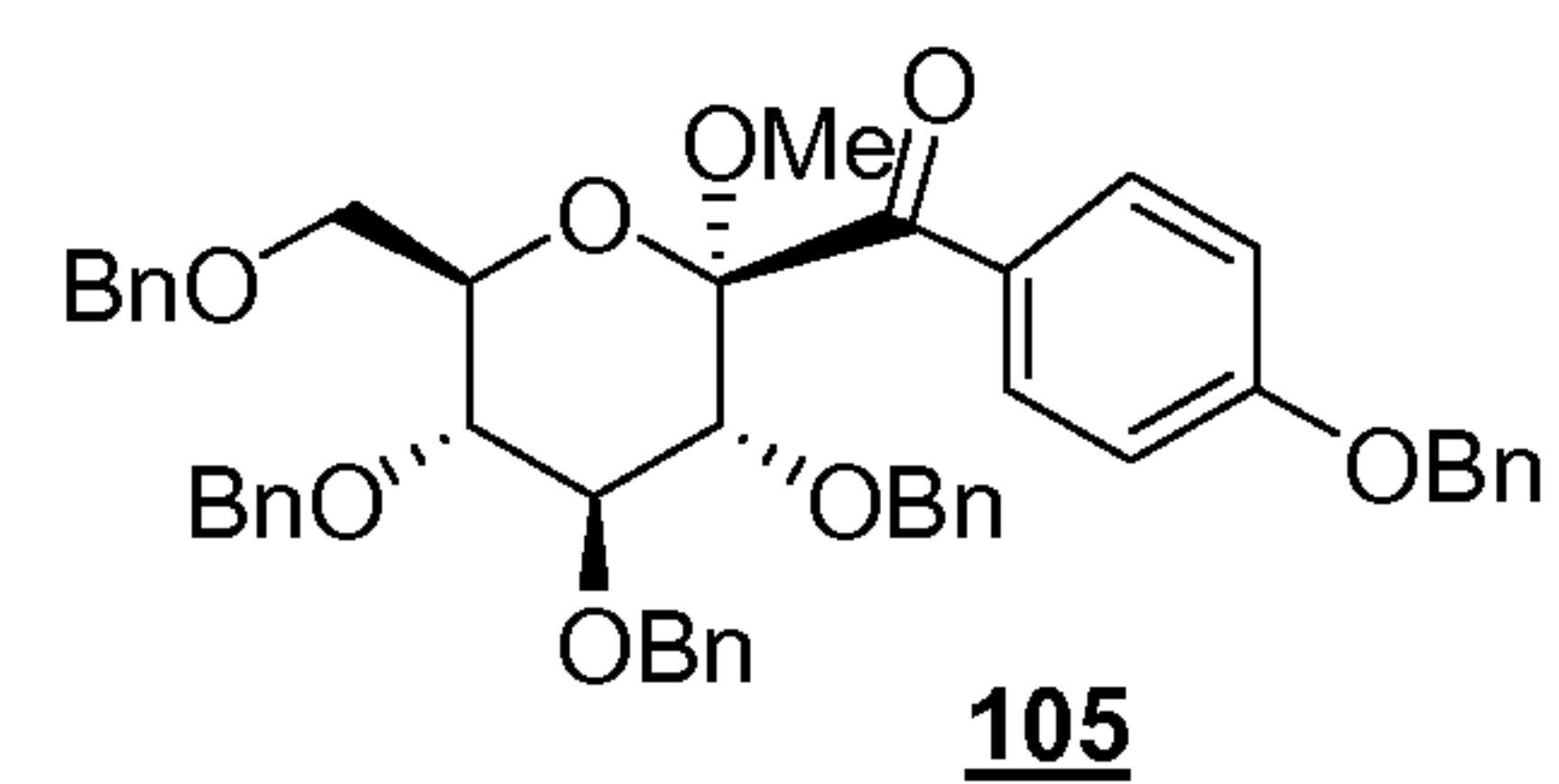
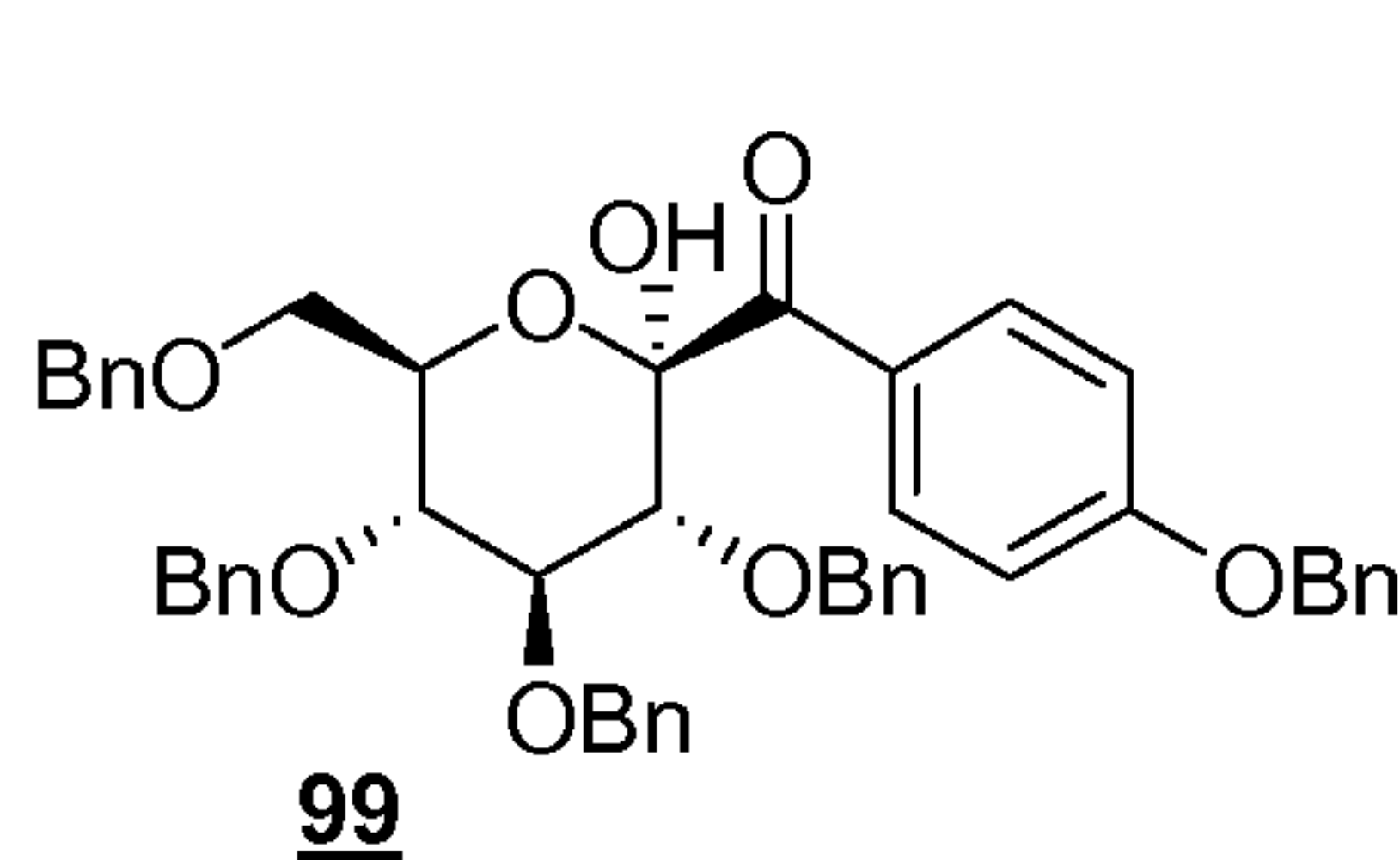
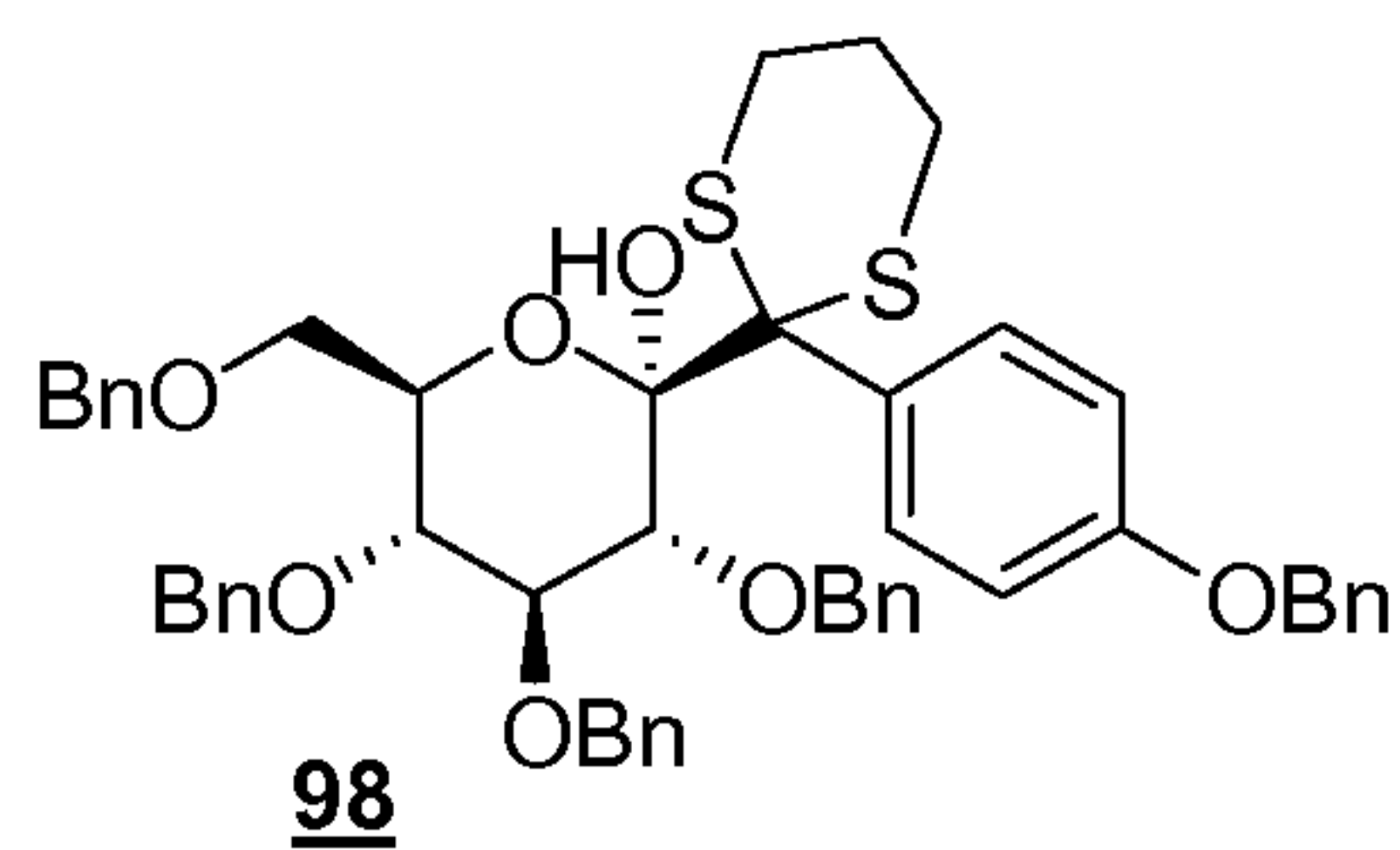
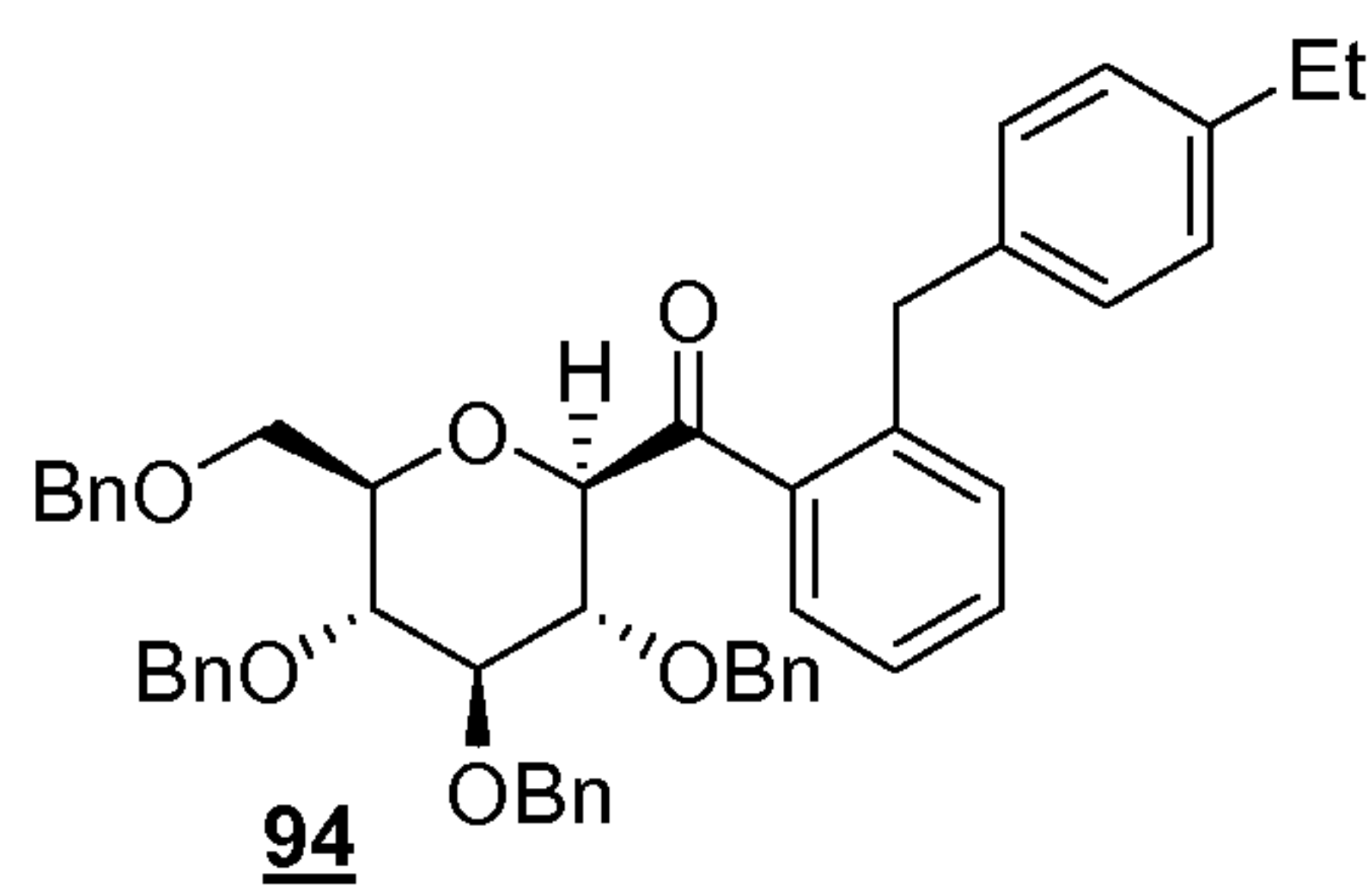
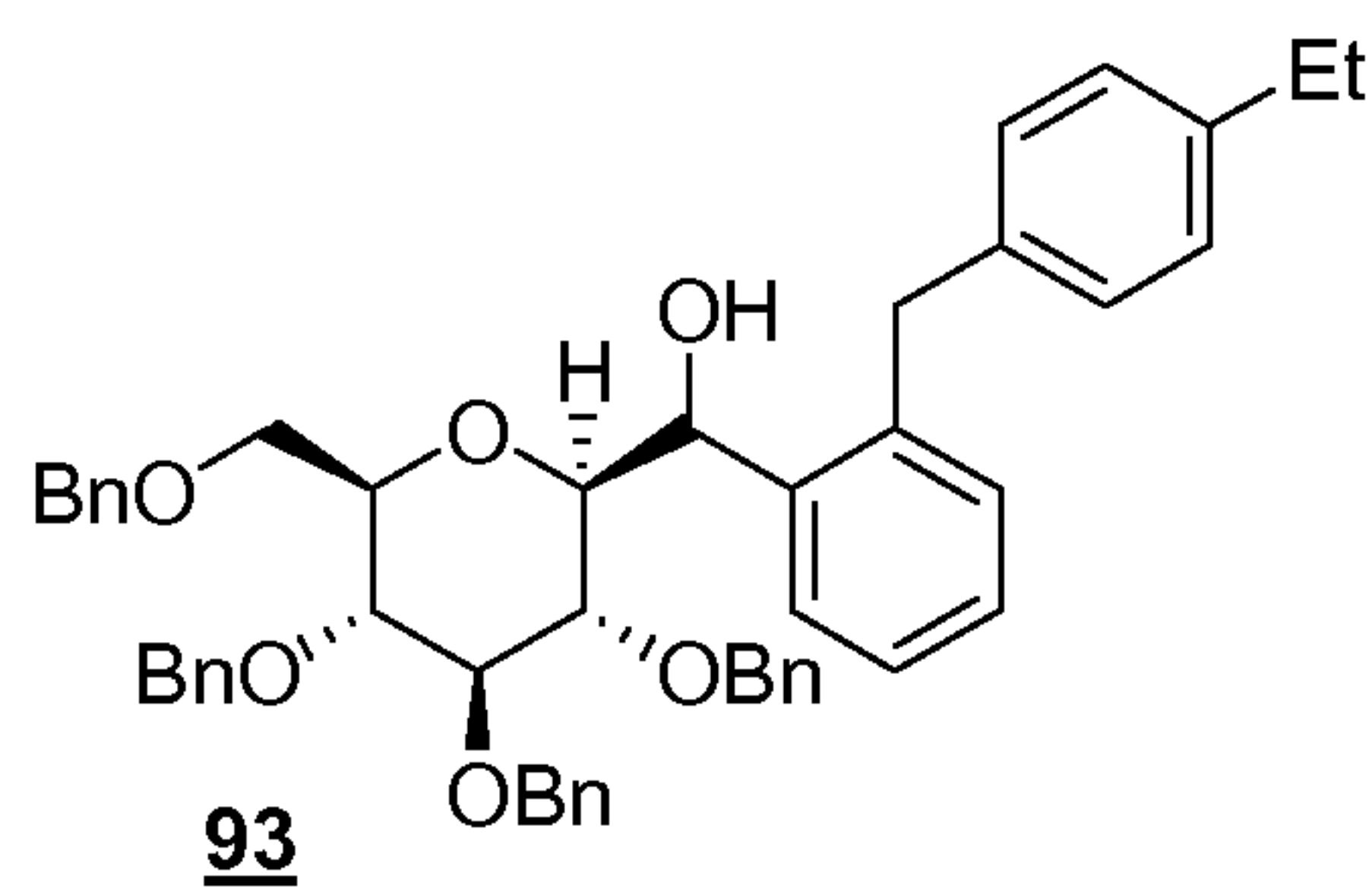
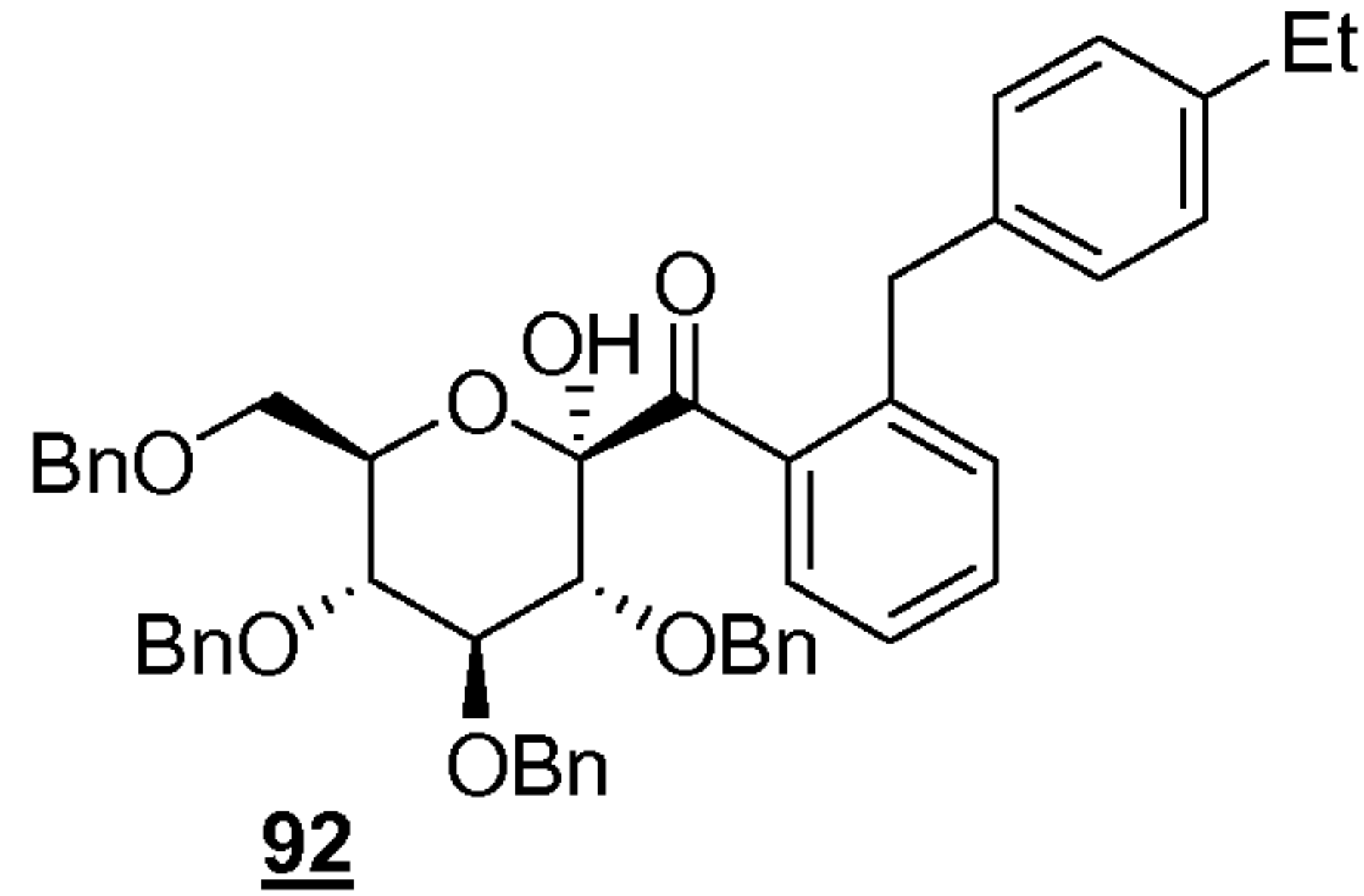
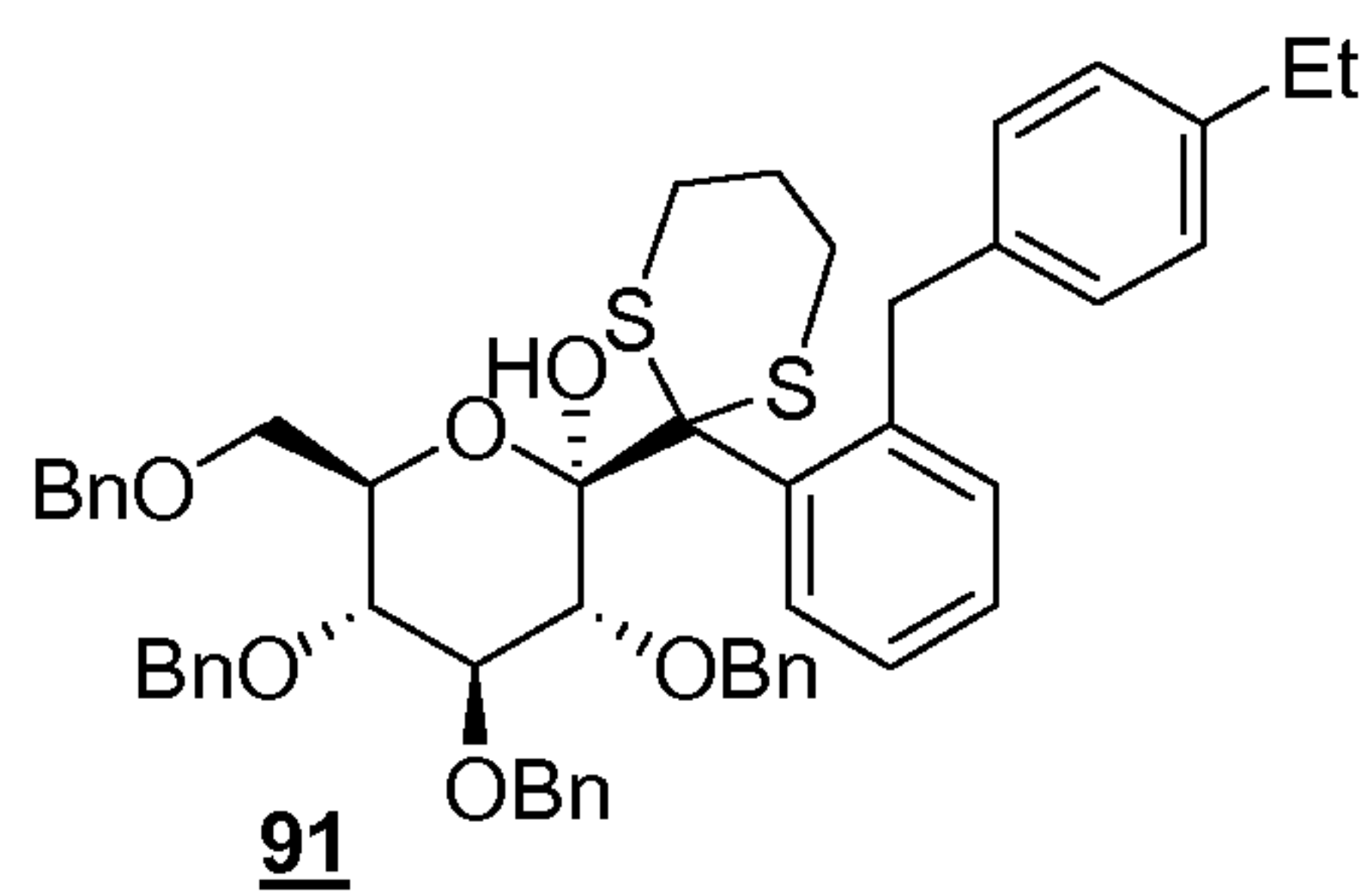
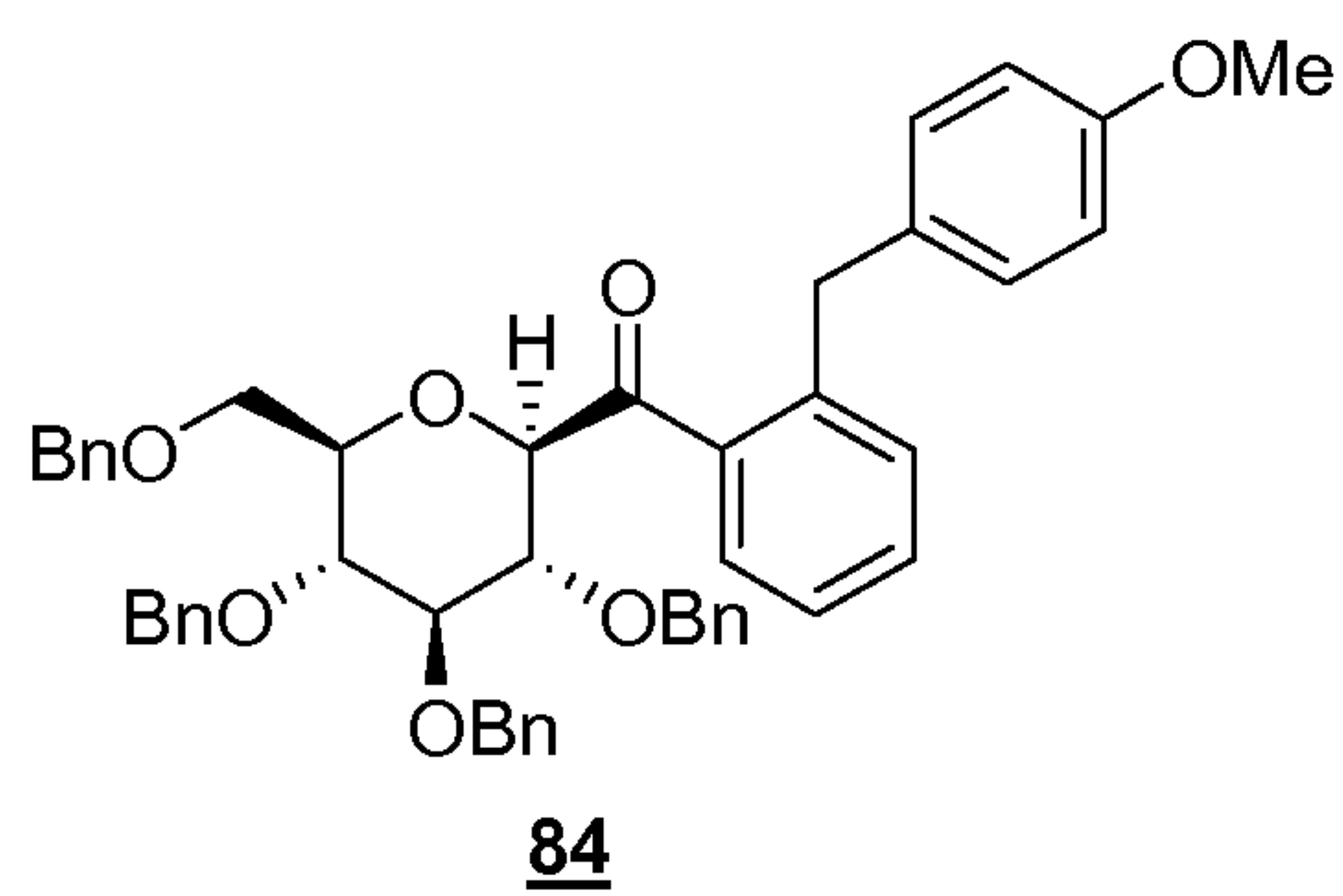
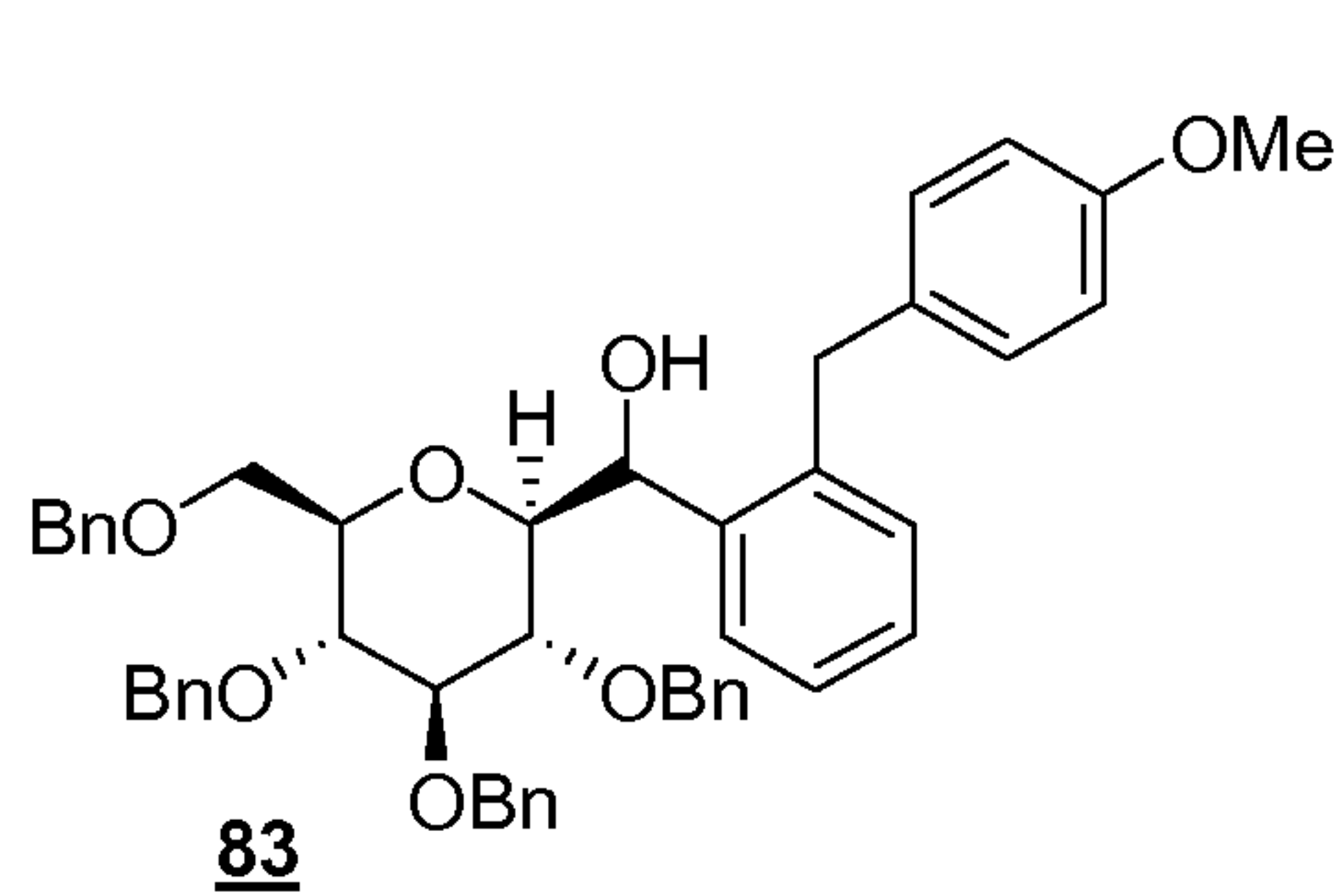
20 Yet even more advantageously, R4 may represent a hydrogen or halogen atom or an OH, -O-(C₁-C₆)-alkyl, -O-aryl and -O-(C₁-C₆)-alkyl-aryl group, and in particular, a hydrogen atom or an OH, -O-(C₁-C₆)-alkyl, -O-aryl and -O-(C₁-C₆)-alkyl-aryl group.

In particular, R4 can represent a hydrogen or halogen (such as Br, Cl, F) atom or an OH group, and advantageously, a hydrogen atom or an OH group.

25 Advantageously, R5, R6, R7, R8, R9, R5a, R6a, R7a, R8a and R9a will be chosen, independently from one another, among a hydrogen atom, a halogen atom, advantageously a chlorine or fluorine atom, an aryl-(C₁-C₆)-alkyl, such as benzyl, aryl-(C₁-C₆)-alkyl-O-, such as benzyloxy, or aryl-CO-, such as benzoyl, group, the alkyl moiety of said group being possibly substituted by an OH group and the aryl moiety of said group being possibly substituted by a halogen atom, such as fluorine, an OH, (C₁-C₆)-alkyl or (C₁-C₆)-alkoxy group.

30 In particular, the compounds of formula (II) can be chosen among the following molecules:





and

The invention will be better understood upon reading the following examples, these examples serving solely to illustrate the invention.

Examples

1. Preparation of the compounds of the invention

The abbreviations encountered are defined as follows:

eq.: equivalentg: gram Hz: Hertz
 5 mg: milligramme MHz: megahertz min.: minute
 mL: millilitre mmol: millimole μ mol: micromole
 nmol: nanomole de: diastereomeric excess

The features of the devices used to conduct analyses of all of the compounds described in this application are indicated hereinbelow:

10 The ^{19}F NMR spectra were recorded on BRUKER DPX 300 and DPX 600 spectrometers. The internal reference used is fluorotrichloromethane CFCl_3 . Chemical shifts (δ) are expressed in parts per million (ppm), and coupling constants (J) in Hertz (Hz).

The following abbreviations were used:

15 s for singlet, bs for broad singlet, d for doublet, t for triplet, qdt for quartet, m for multiplet or massive, dd for doublet of doublet, etc.

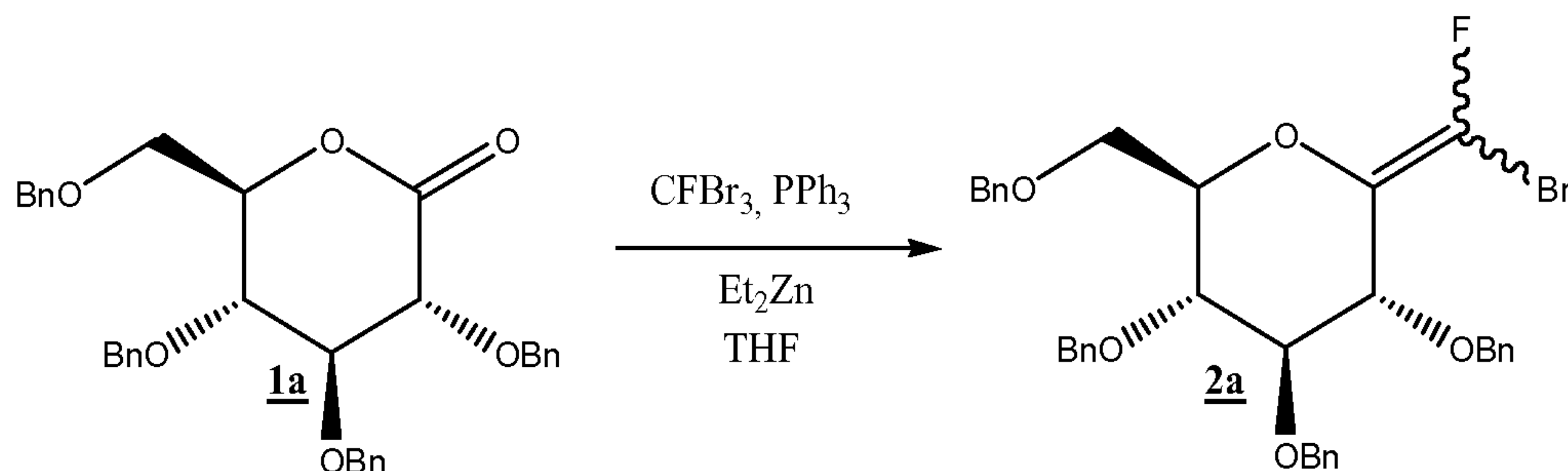
The mass spectra were obtained on a spectrophotometer of the Micromass TOF-SPEC E 20 kV, α -cyano type, for MALDI ionization and JEOL AX500, 3 kV, Canon FAB JEOL, Xe, 4 kV, 10 μA limiting current, Gly-NBA 50:50 for FAB ionization.

20 Separations via column chromatography are carried out under light pressure by following chromatography techniques on Kieselgel 60 silica (230-400 Mesh, Merck).

Follow-up is ensured via thin-layer chromatography (TLC) with Kieselgel 60F-254-0.25-mm plates. The ratio of the migration distance of a compound on a given support to the migration distance of an eluent is called the retardation factor (R_f).

25 Exemplary compound preparations according to the invention will be described hereinbelow, for non-limiting, illustrative purposes.

The compounds have been numbered by assigning the letter a to the glucose derivatives and b to the galactose derivatives.

Synthesis of compounds 2a

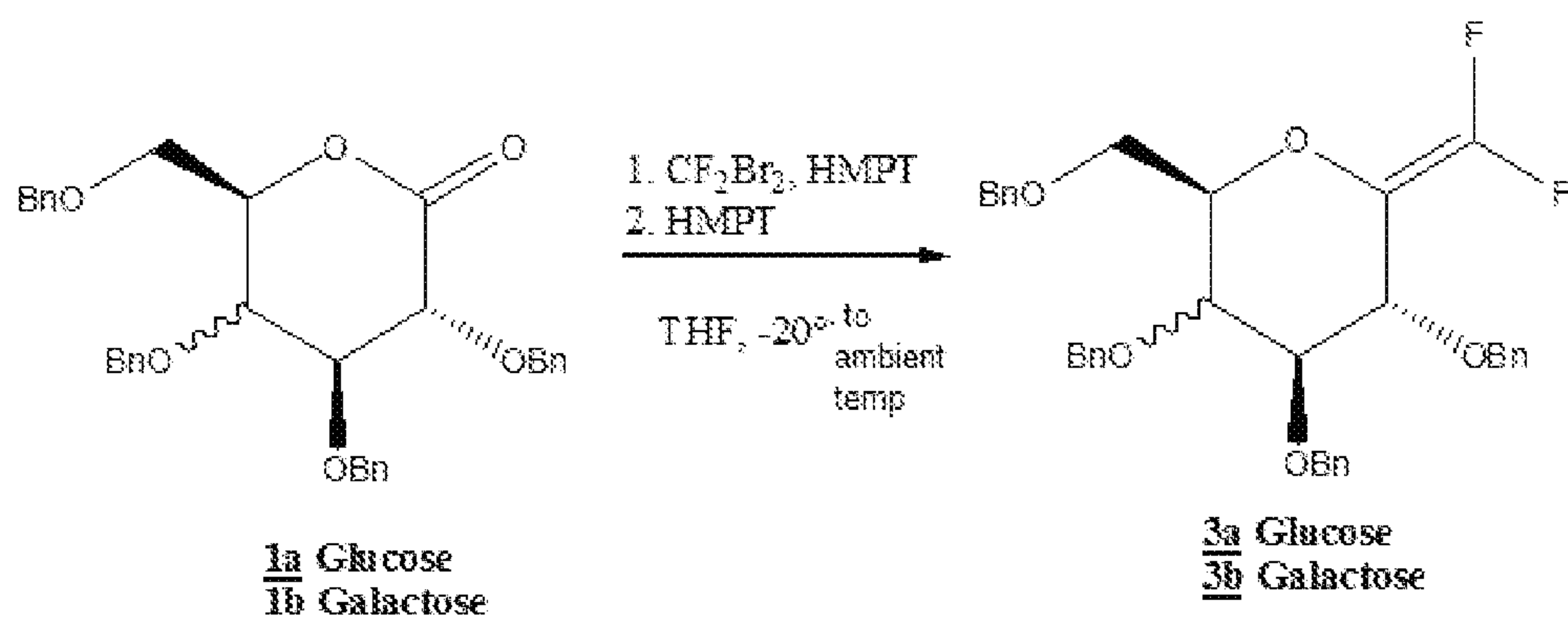
Into a round-bottom flask under an inert atmosphere containing the triphenylphosphine PPh₃ (849 mg; 3.2 mmol; 3.4 eq.), the tribromofluoromethane CFBr₃ (313 μ L; 3.2 mmol; 3.4 eq.) and the lactone 1a (synthesized according to *J. Org. Chem.* 1967, 32 (8) 2531-2534) (500 mg; 0.928 mmol; 1 eq.) in the anhydrous tetrahydrofuran (THF) (15 mL), a solution of diethylzinc Et₂Zn 1 M in hexane or toluene (3.2 mL; 3.2 mmol, 3.4 eq.) is added slowly dropwise over approximately three hours using a syringe driver. The mixture is stirred for 24 hours, and then MeOH is added and the reaction mixture is concentrated. The crude product is then purified on a chromatography column and the compound 2ad1/d2, in the form of a colourless oil containing the 2 diastereomers (d1 and d2), in a ratio of (33/67), is collected together with a 95/5 mixture of cyclohexane/ethyl acetate, and with a yield of 42%.

2ad1/d2: C₃₅H₃₄BrFO₅ M = 633.54 g.mol⁻¹

NMR ¹⁹F (CDCl₃, 282.5 MHz):

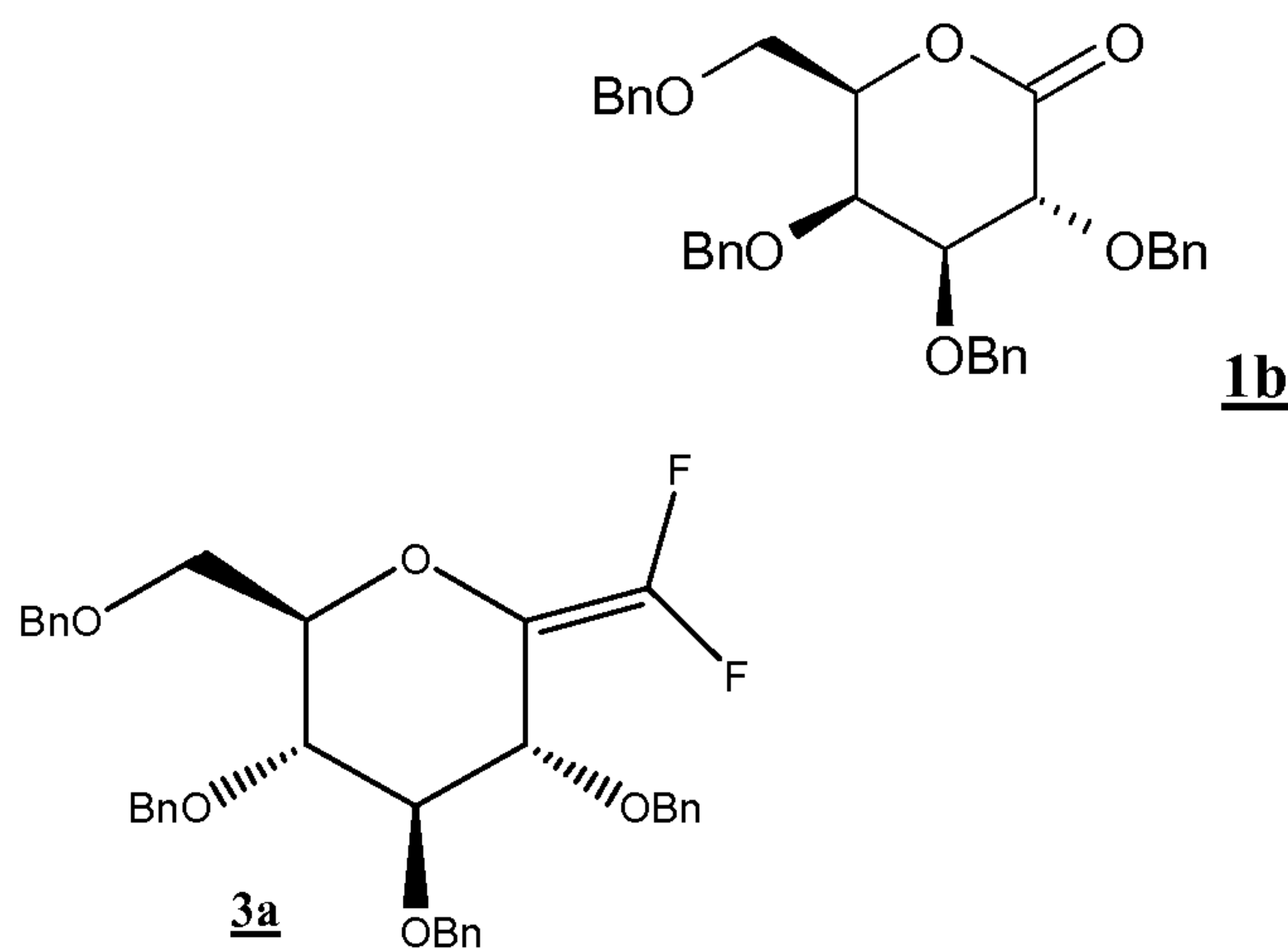
2ad1: -98.5 (dd, 2 Hz, 0.34F); 2ad2: -119.2 (d, 3 Hz, 0.66 F)

Mass: (ESI +) : 651 (M + H₂O)

Synthesis of compounds 3a and 3b

Into a round-bottom flask under an inert atmosphere at -20°C containing the lactone 1a (497 mg; 0.923 mmol; 1 eq.) in tetrahydrofurane THF, the dibromodifluoromethane CF₂Br₂ (422 µL; 4.62 mmol; 5 eq.), and hexamethylphosphoramide (HMPT) (847 µL; 4.62 mmol; 5 eq.) are then added. The temperature of the solution is brought back up to 10°C very slowly (in approximately 30 min) and then the hexamethylphosphoramide HMPT (2.5 mL; 13.8 mmol; 15 eq.) is added at this temperature. The solution is brought back to ambient temperature and stirred for 2 h 30 min. Diethyl ether is added and then the mixture is washed three times with a saturated aqueous copper sulphate solution. The organic phase is dried over magnesium sulphate, filtered, and then concentrated. The crude product thus obtained is chromatographed on silica gel with a (95/5) cyclohexane/ethyl acetate eluent mixture to produce the compound 3a, in the form of a yellow oil, with a yield of 58%.

Compound 3b was obtained in the form of a yellow oil with an isolated yield of 53%, by following the same procedure as above, replacing compound 1a with compound 1b (synthesized according to *J. Org. Chem.* 1967, 32 (8) 2531-2534) having the following formula:



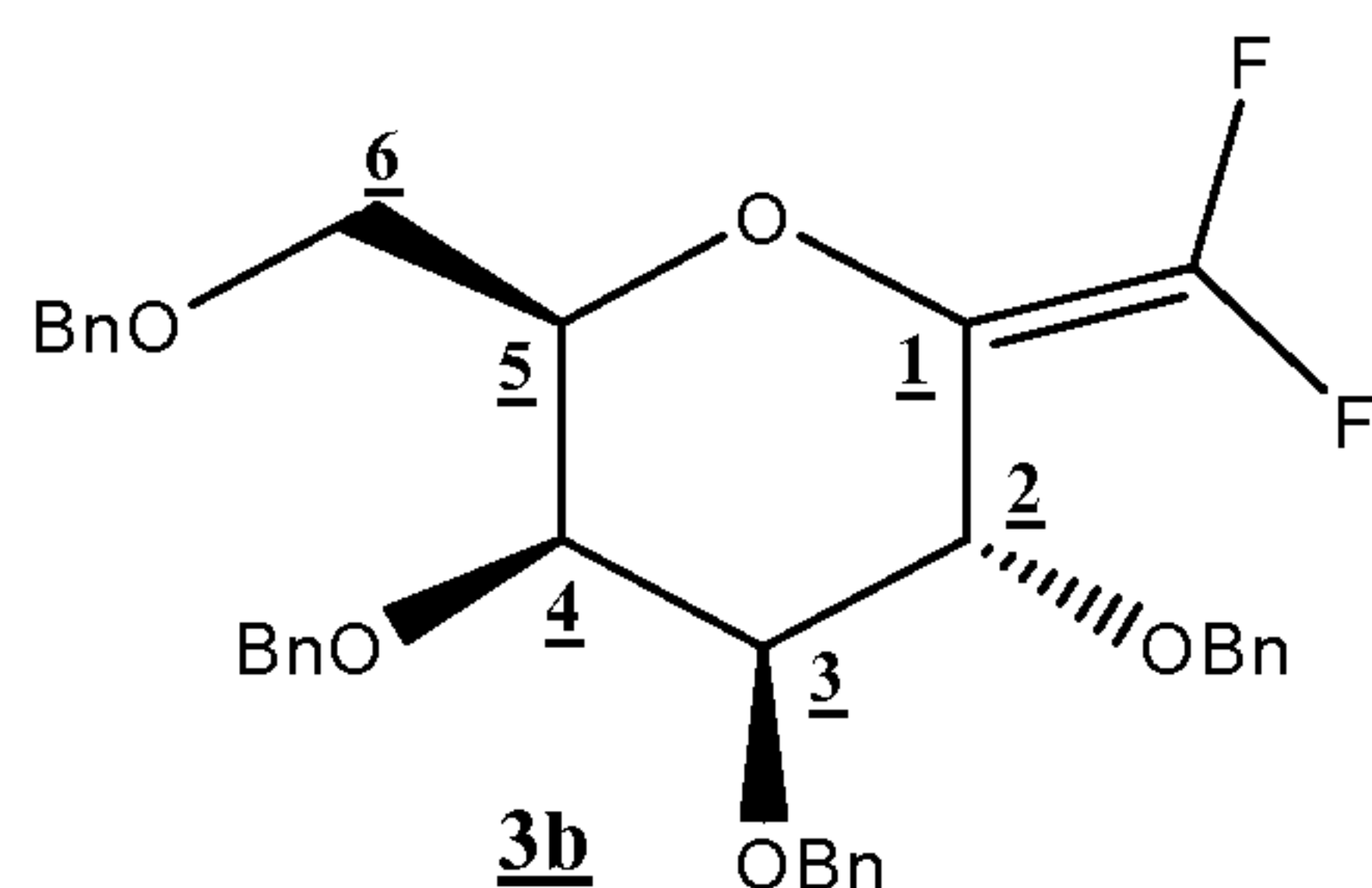
3a: C₃₅H₃₄F₂O₅ M = 572.64 g.mol⁻¹

R_f = 0.49 eluent: cyclohexane/ethyl acetate (9/1).

NMR ¹⁹F (CDCl₃, 282.5 MHz):

-99.3 (d, J_{F-F} = 74 Hz, 1F); -116.3 (d, J_{F-F} = 74 Hz, 1F)

Mass: (ESI +) : 590.40 (M + H₂O); 595.53 (M + Na); 612.27 (M + K)



3b: $C_{35}H_{34}F_2O_5$ $M = 572.64 \text{ g.mol}^{-1}$

NMR ^{19}F ($CDCl_3$, 282.5 MHz):

-97.0 (d, $J_{F-F} = 66 \text{ Hz}$, 1F); -110.9 (s, $J_{F-F} = 61 \text{ Hz}$, 1F).

5

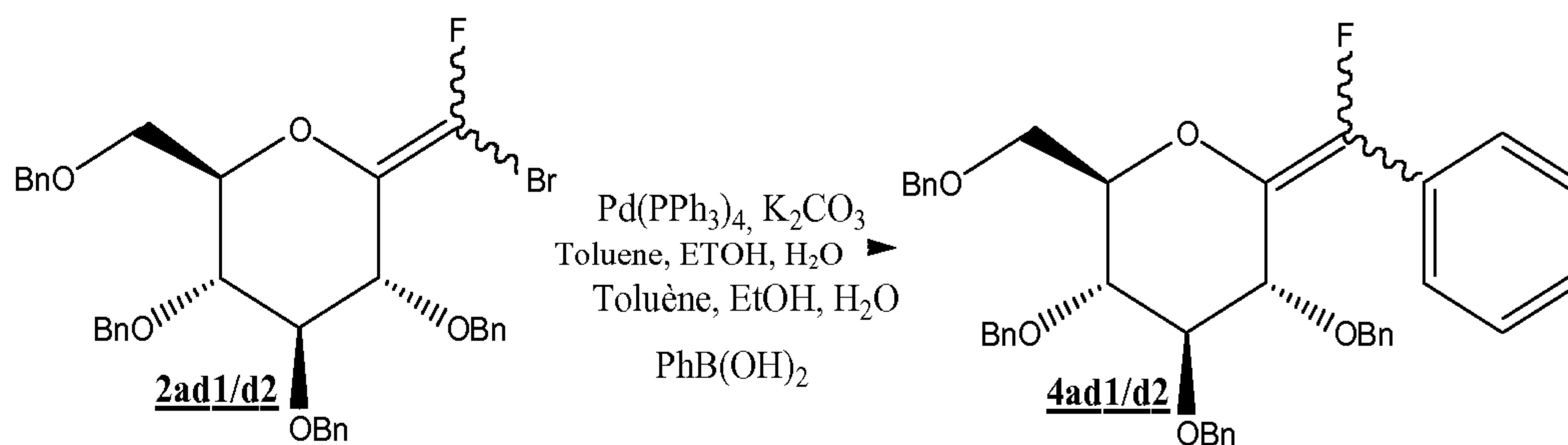
Mass: (ESI +) : 595 (M + Na); 611 (M + K)

Synthesis of compounds 4ad1/d2 and 4bd1/d2

Compound 4a was synthesized in the form of two isomers, according to two different processes. Compound 4b was synthesized from the second process.

10

First process:



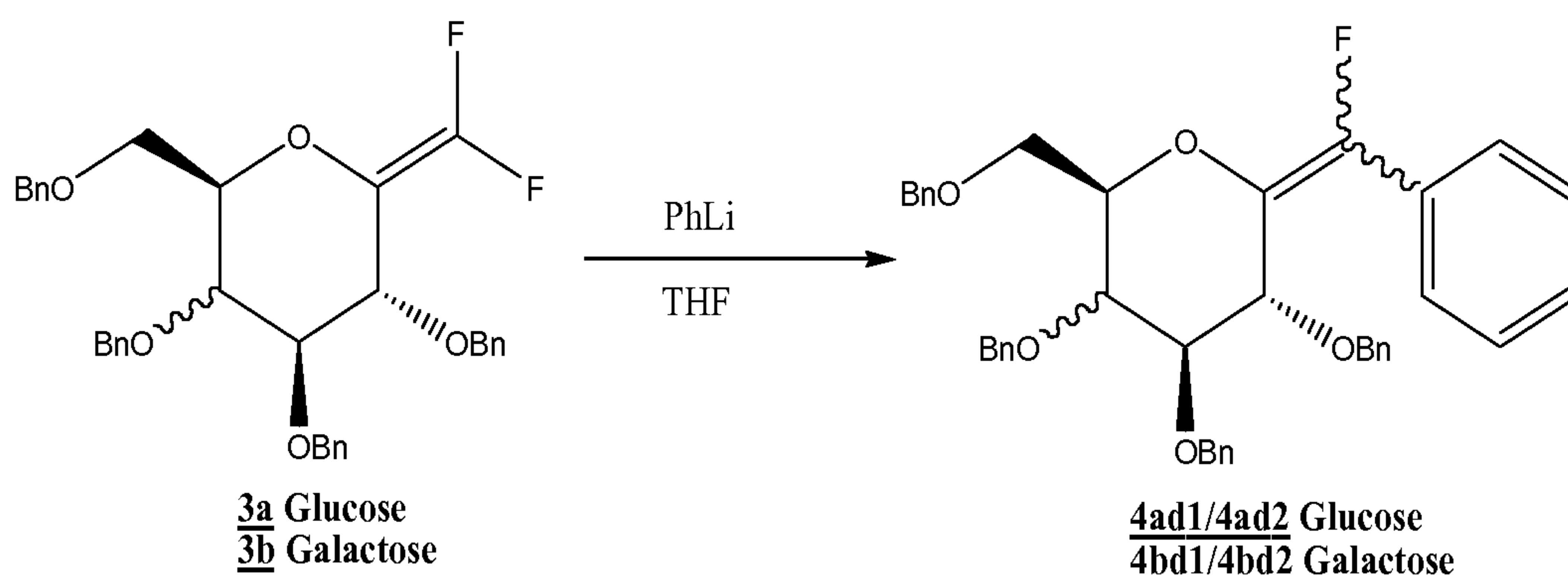
15

20

In a two-necked round-bottom flask containing the palladium tetrakis $Pd(PPh_3)_4$ (8 mg, 4% polarity), the potassium carbonate K_2CO_3 (75 mg, 0.54 mmol, 3 eq.) in a mixture of toluene (5.55 mL), ethanol EtOH (540 μ L) and water H_2O (540 μ L), compound 2ad1/d2 in the form of a mixture of the 2 isomers (in proportions of 33/67) is added and left under stirring at ambient temperature for 15 minutes. Then, the phenylboronic acid $PhB(OH)_2$ is added, and the reaction mixture is refluxed and thus kept under stirring for 3 hours. The reaction mixture is then brought back to ambient temperature, hydrolyzed and extracted three times with ether Et_2O . The organic phases are then collected, washed with a saturated sodium chloride solution (NaCl), then dried over magnesium sulphate $MgSO_4$, filtered and evaporated. The crude product containing the 2 isomers in a ratio of 66/34 is then purified on a silica column with a 99/1 cyclohexane/ethyl acetate mixture to produce a mixture of the major diastereomer 4ad2 and of the minor

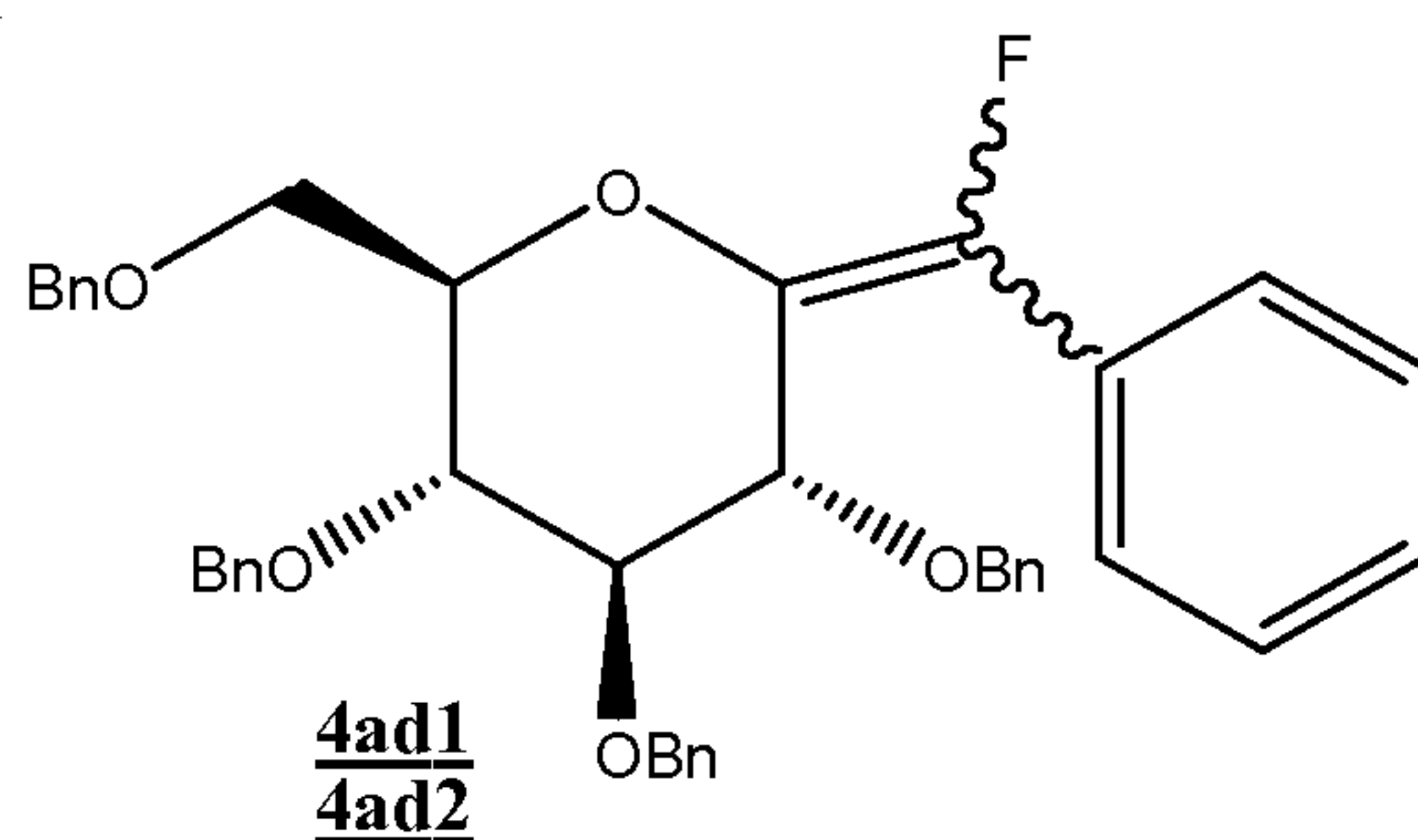
diastereomer 4ad1 with an overall yield of 90% in the form of a light yellow oil, each diastereomer being obtainable separately after purification on the silica column, if necessary.

5 Second process:



The phenyllithium 2 M (1.8 mL: 3.6 mmol; 3 eq.) is added to a round-bottom flask, under an inert atmosphere and containing compound 3a (689 mg; 1.2 mmol; 1 eq.), dissolved in anhydrous THF at 0°C. The mixture is left under stirring for 3 hours at 0°C, then gradually brought back to ambient temperature and left to stir for 12 hours. The mixture is hydrolyzed with a saturated sodium chloride solution, and dichloromethane is added. The two phases are separated, then the aqueous phase is extracted two more times with dichloromethane. The organic phases are collected, dried over magnesium sulphate, filtered and then evaporated. The crude mixture containing the two diastereomers in a ratio of 62/38 is then purified on a silica column with a 99/1 cyclohexane/ethyl acetate mixture to produce a mixture of the major diastereomer 4ad2 and of the minor diastereomer 4ad1 with an overall yield of 66%, each diastereomer being obtainable separately after purification on the silica column, if necessary.

The reaction is carried out in the same way as for compound 3b, but with 2 eq. of phenyllithium. The reaction is hydrolyzed after one hour at 0°C and, after purification, produces 2 diastereomers in a ratio of 87/13 (major 4bd2 and minor 4bd1) with an overall yield of 46%.



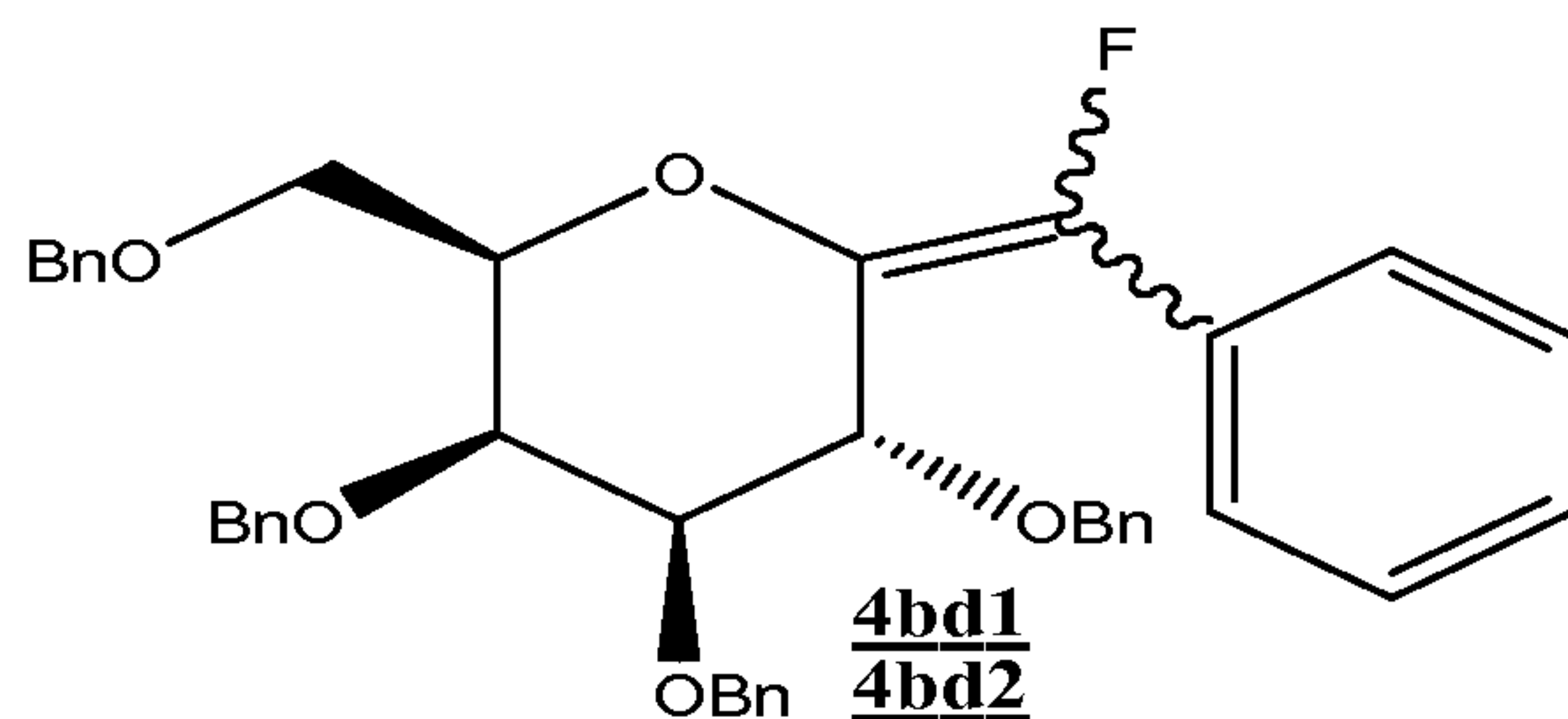
4ad1 and 4ad2: C₄₁H₃₉FO₅ M = 630.74 g.mol⁻¹

NMR ¹⁹F (CDCl₃, 282.5 MHz):

-122.2 (s) 4ad1; -154.3 (d, 2 Hz) 4ad2

Mass: (ESI +) : 648 (M + H₂O)

5



4bd1/4bd2: C₄₁H₃₉FO₅ M = 630.74 g.mol⁻¹

NMR ¹⁹F (CDCl₃, 282.5 MHz):

-114 (s) 4ad1; -145.1 (d, 3 Hz) 4ad2

Mass: (ESI +) : 648 (M + H₂O)

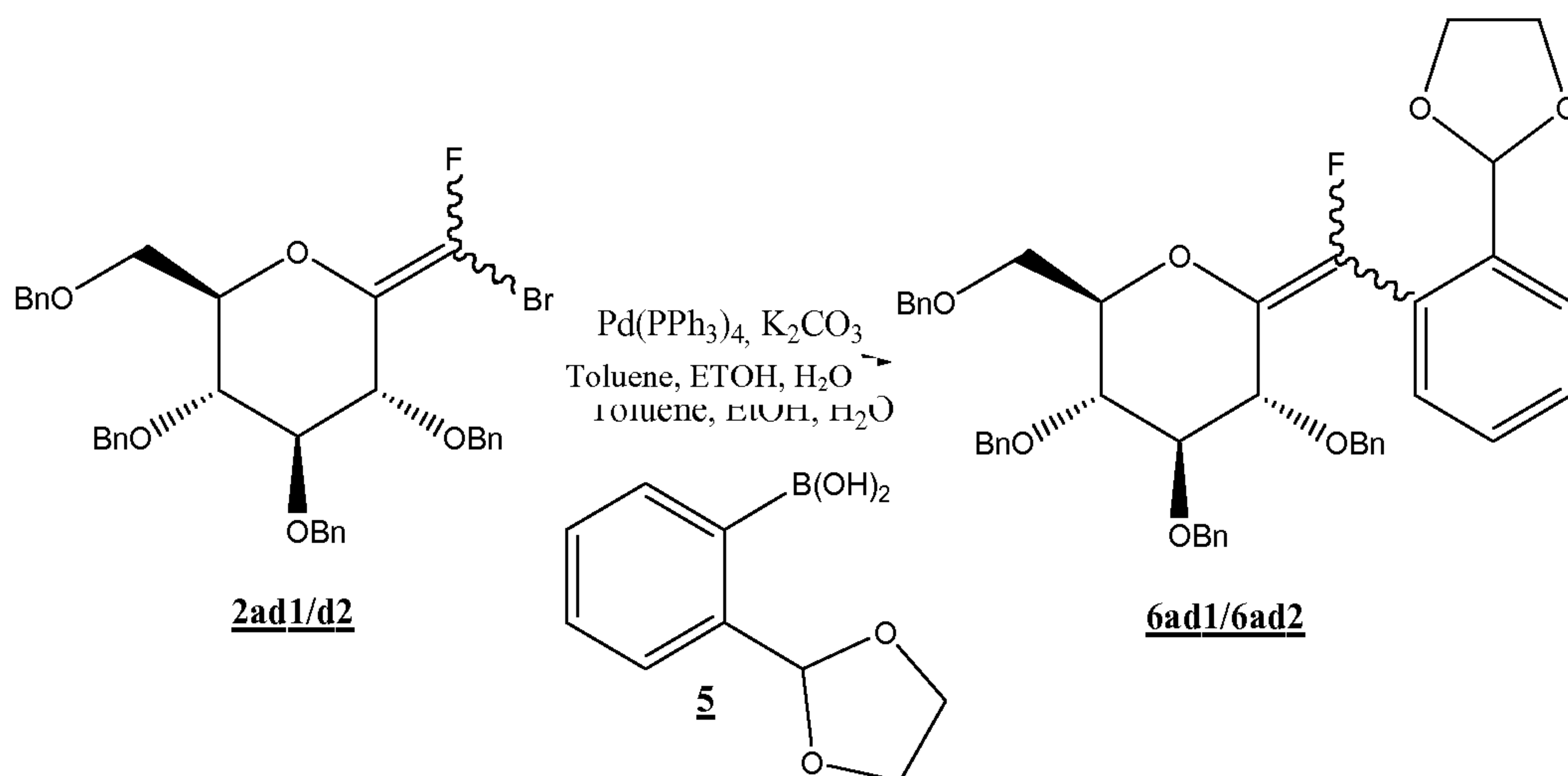
10

Synthesis of compounds 6ad1/6ad2

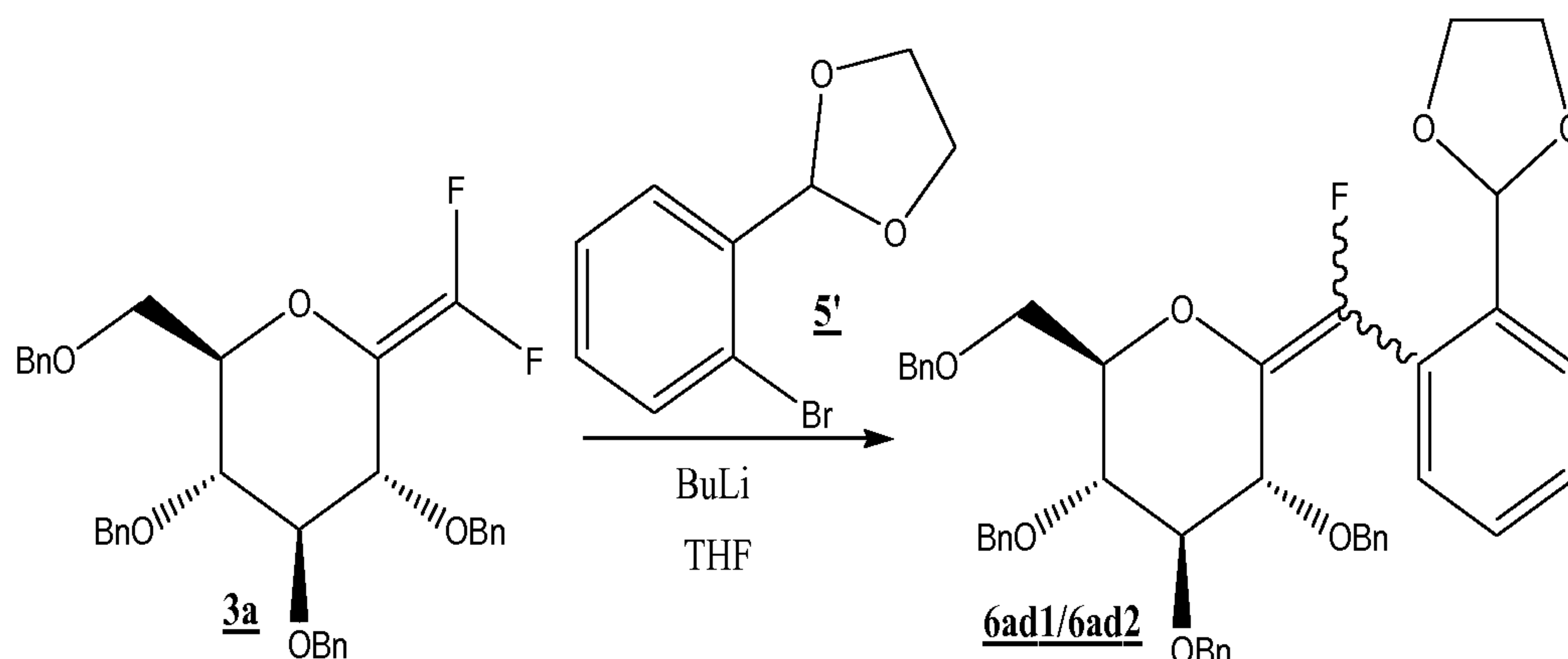
Compound 6a was synthesized in the form of two isomers 6ad1 and 6ad2, according to two different processes.

15

20

First process:

Into a round-bottom flask containing the palladium tetrakis $\text{Pd(PPh}_3)_4$ (8 mg, 4% polarity), the potassium carbonate K_2CO_3 (70 mg; 0.507 mmol, 3 eq.) in a mixture of
 5 toluene (5.55 mL), ethanol EtOH (540 μL) and water H_2O (540 μL), compound **2ad1/d2** in the form of a mixture of 2 isomers (33/67) is added and left under stirring at ambient temperature for 15 minutes. Then compound **5** is added (obtained in 2 steps according to procedures described in the Journal of Organic Chemistry (2006), 71 (20), 7840-7845 and Bull. Chem. Soc. Jpn (2002), 2267-2672), and the medium is refluxed
 10 and thus kept under stirring for 3 hours. The medium is then brought back to ambient temperature, hydrolyzed and extracted three times with ether Et_2O . The organic phases are then collected, washed with a saturated sodium chloride solution (NaCl), then dried over magnesium sulphate MgSO_4 , filtered and evaporated. The crude product containing the 2 isomers is then purified on a silica column with a 97/3
 15 cyclohexane/ethyl acetate mixture to produce a mixture of the major diastereomer **6ad2** and of the minor diastereomer **6ad1** with an overall yield of 55%, each diastereomer being obtainable separately after purification on the silica column, if necessary.

Second process:

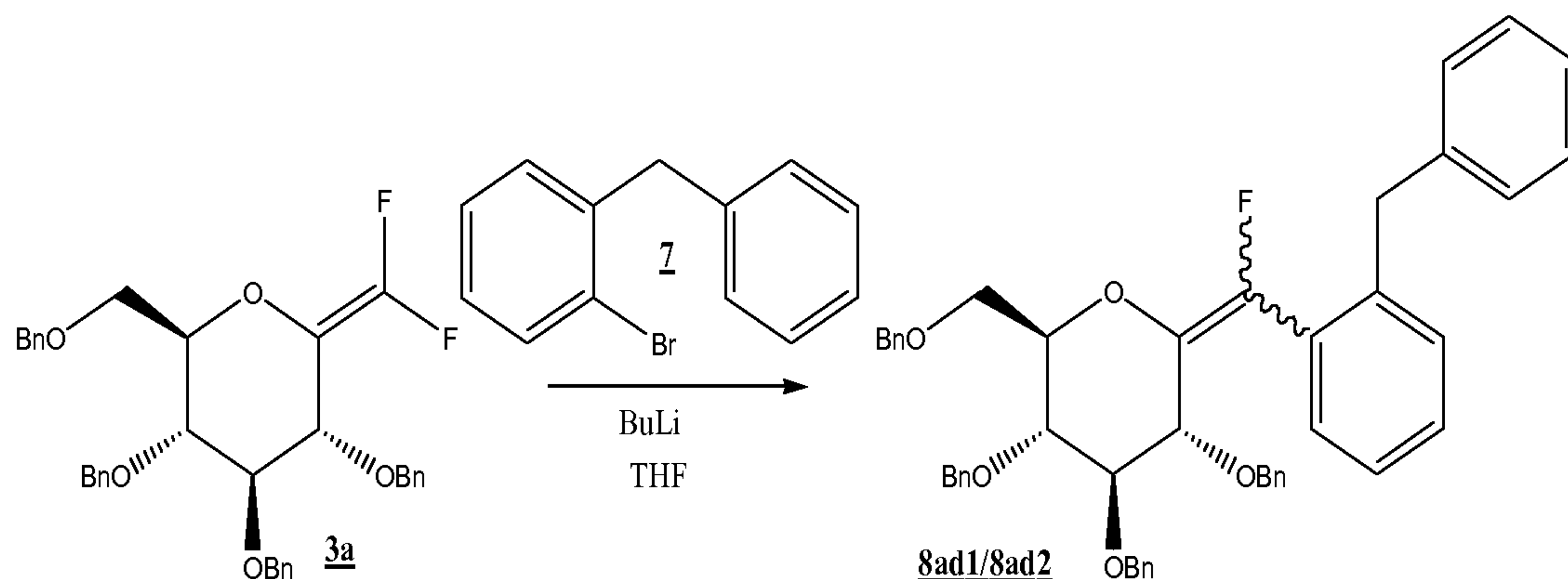
Into a round-bottom flask, under an inert atmosphere containing compound **3a** (900 mg; 1.57 mmol; 1 eq.) dissolved in anhydrous THF (20 mL) at -78°C, compound **5'** (synthesized according to *J. Org. Chem.* (2006), 71(20), 7840-7845) (1.07 g; 4.71 mmol; 3 eq.) is added, then the n-butyllithium (BuLi) 1.6 M (2.84 mL, 4.55 mmol, 2.9 eq.). The mixture is left under stirring for 3 hours at -78°C, then allowed to gradually rise back to ambient temperature and left to stir for 12 hours. The mixture is hydrolyzed with a saturated sodium chloride solution, and dichloromethane is added. The two phases are separated, and then the aqueous phase is extracted two more times with dichloromethane. The organic phases are collected, dried over magnesium sulphate, filtered and then evaporated. The crude mixture containing the two diastereomers is then purified on a silica column with a 97/3 cyclohexane/ethyl acetate mixture to produce a mixture of the major diastereomer **6ad2** and of the minor diastereomer **6ad1** with an overall yield of 55%, each diastereomer being obtainable separately after purification on the silica column, if necessary.

6ad1 and 6ad2: C₄₄H₄₃FO₇ M = 702.81 g.mol⁻¹

NRM ¹⁹F (CDCl₃, 282.5 MHz):

-136.4 (s) **6ad2**; -114.7 (s) **6ad1**

Mass: (ESI +) : 725 (M + H₂O)



to a round-bottom flask containing compound 3a (200 mg; 0.35 mmol; 1 eq.) dissolved in anhydrous THF (2 mL) at -78°C, compound 7 (172 mg; 0.698 mmol; 2 eq.) is added, followed by the n-butyllithium 1.6 M (414 µL, 0.66 mmol, 1.9 eq.). The mixture is left under stirring for 3 hours at -78°C, then allowed to gradually rise back up to ambient temperature and left to stir for 12 hours. The mixture is hydrolyzed with a saturated sodium chloride solution, and dichloromethane is added. The 2 phases are separated, and then the aqueous phase is extracted two more times with dichloromethane. The organic phases are collected, dried over magnesium sulphate, filtered and then evaporated. The crude mixture containing the two diastereomers is then purified on a silica column with a 97/3 cyclohexane/ethyl acetate mixture to produce the major diastereomer 8ad2 with an overall yield of 35%, only traces of compound 8ad1 being present, which do not allow the isolation of this compound.

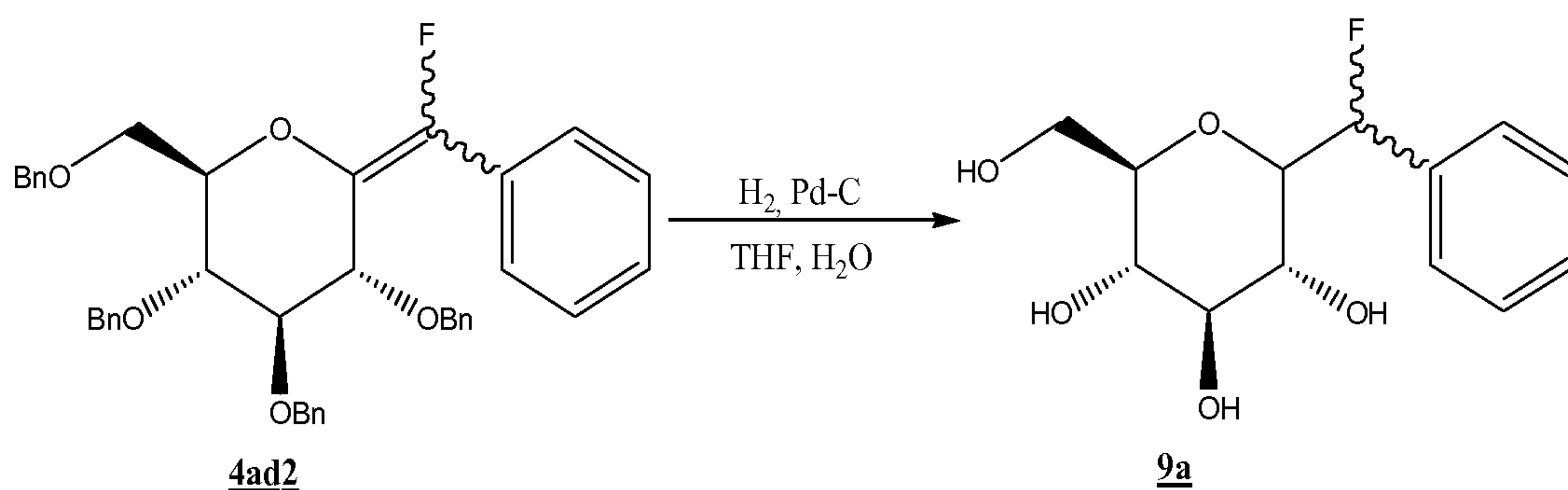
15 **8ad2: C₄₈H₄₅FO₅** M = 720.87 g.mol⁻¹

NMR ¹⁹F (CDCl₃, 282.5 MHz):

-136.9 (s) 8ad2

Mass: (ESI +) : 738 (M + H₂O)

25

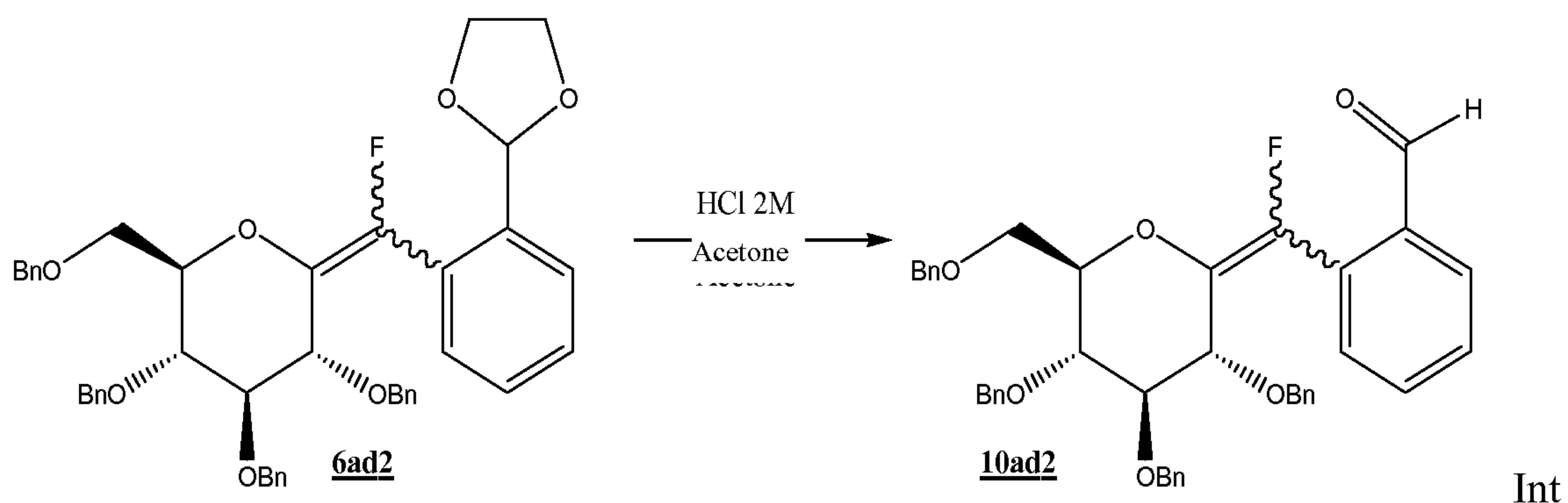
Synthesis of compound 9a

Compound **4ad2** (32.2 mg; mmol; 1 eq.) is placed inside a round-bottom flask and dissolved in a mixture of tetrahydrofurane (1 mL) and water (500 μ L), in the presence of a scoopula tip of Pd/C under a hydrogen atmosphere. The mixture is stirred for 24 h, then Millipore-filtered and evaporated to produce compound **9a** with a quantitative yield.

9a: $C_{13}H_{17}FO_5$ $M = 212.27 \text{ g.mol}^{-1}$

$NMR^{19}F$ ($CDCl_3$, 282.5 MHz):

Mass: (ESI-) : 211 (M-H); 246-248 (M+Cl)

Synthesis of compound 10ad2

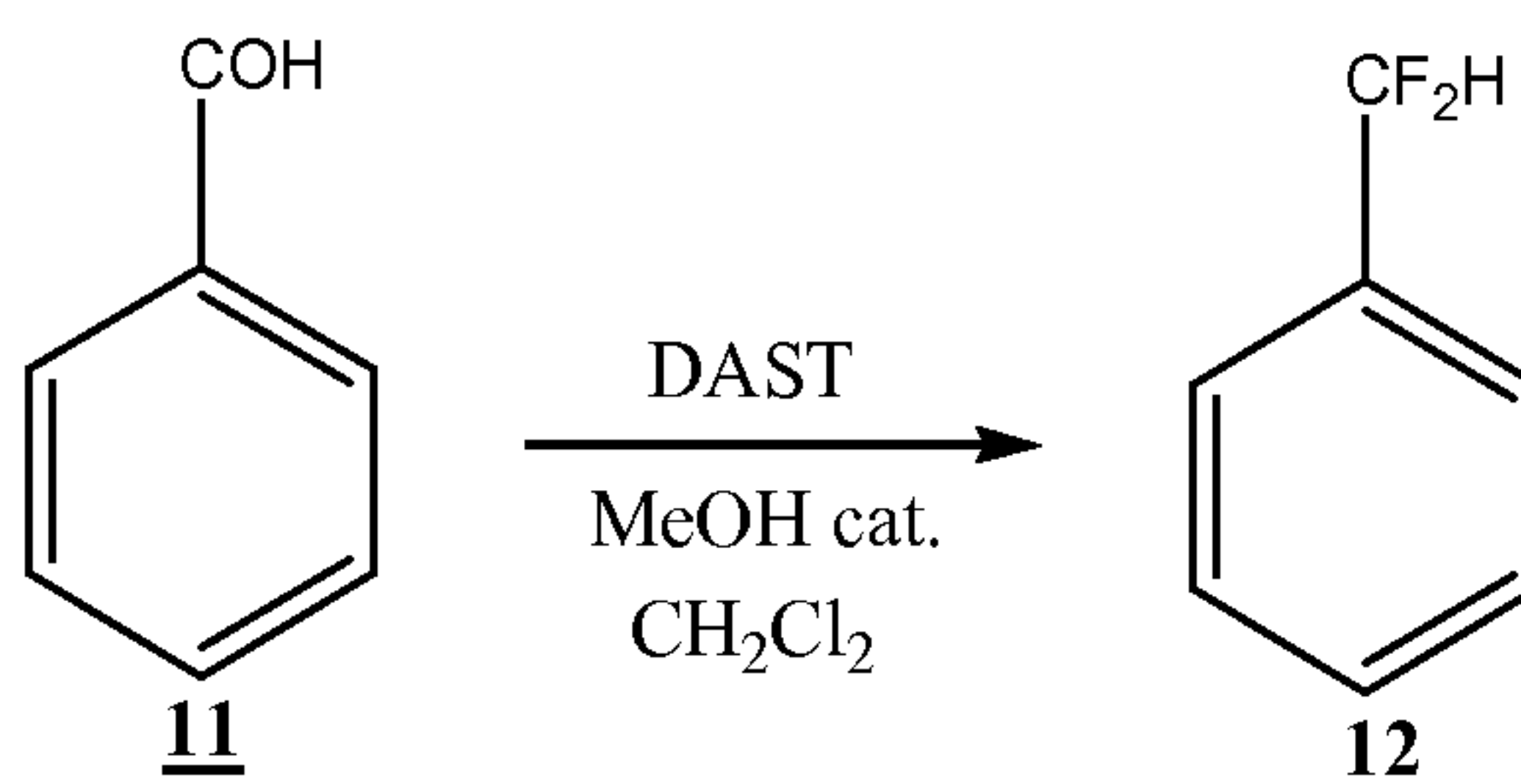
o a round-bottom flask containing compound **6ad2** (20 mg; 0.028 mmol; 1 eq.) in acetone (1 mL), an HCl 2M solution (200 μ L; 0.4 mmol; 14 eq.) is added, and then the mixture is kept under stirring for 48 hours. A saturated sodium hydrogencarbonate solution is added, then extracted three times with dichloromethane. The organic phases are collected, dried over magnesium sulphate, filtered and then evaporated to produce compound **10ad2**, in the form of a yellow oil, with a yield of 60%.

10ad2: $C_{42}H_{39}FO_6$ $M = 658.75 \text{ g.mol}^{-1}$

$NMR^{19}F$ ($CDCl_3$, 282.5 MHz): -139.2 (s)

Mass: (ESI +) : 677 (M + H₂O); 700 (M + H₂O + Na);

Synthesis of compound 12



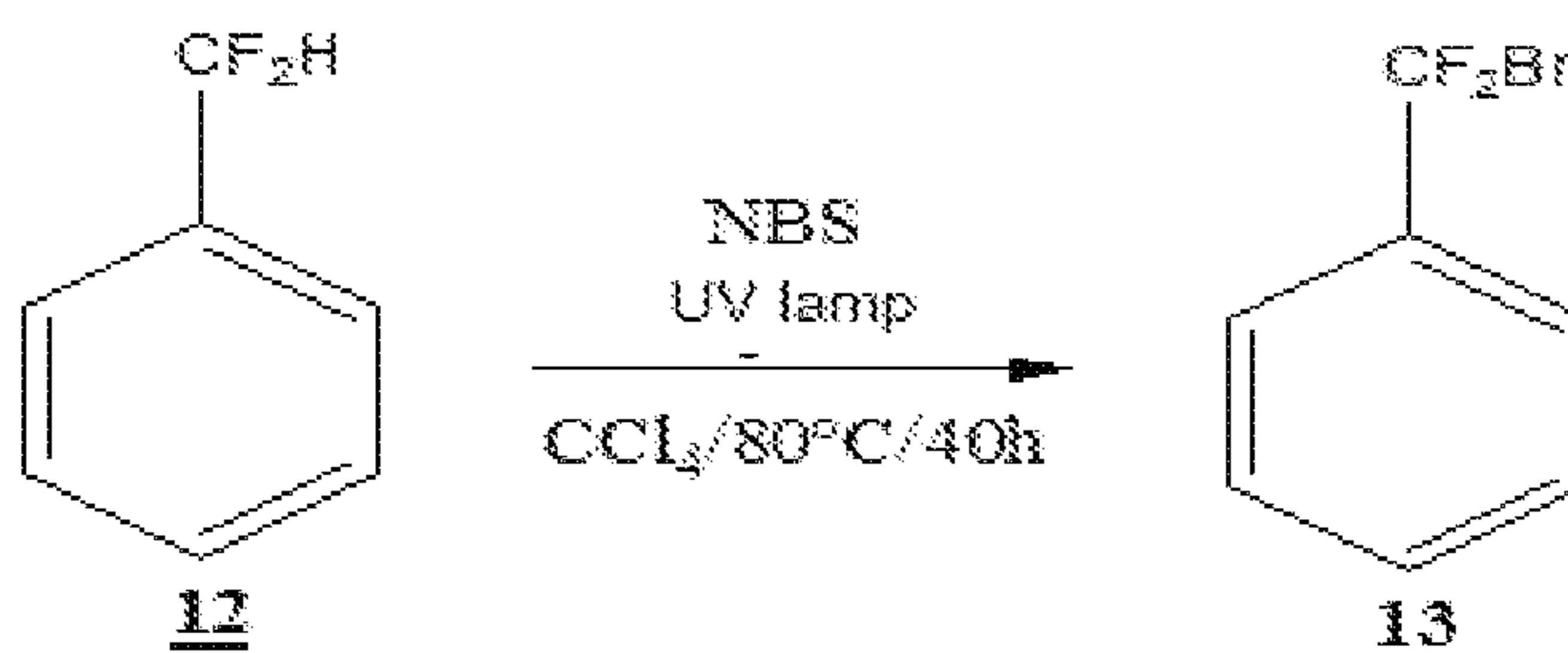
- 5 Diethylaminosulfur trifluoride (DAST) (11.8 mL; 95 mmol; 1.7 eq.) is added dropwise into a round-bottom flask under an inert atmosphere containing freshly distilled benzaldehyde 11 (5.67 mL; 56 mmol; 1 eq.) in dichloromethane (20 mL). A drop of anhydrous methanol is then added to the reaction medium in order to catalyze the reaction. The mixture is stirred for 16 h at ambient temperature and then cooled to 0°C,
- 10 before adding a saturated aqueous sodium bicarbonate solution until the neutral state is reached. The mixture is then extracted with dichloromethane. The organic phase is distilled under low pressure (bp T° = 35°C; P° = 61 mBar) to produce compound 12 in the form of a colourless liquid with a yield of 60%.

12: C₇H₆F₂ M = 128.12 g.mol⁻¹

- 15 NMR ¹⁹F (CDCl₃, 282.5 Mz): -111.0 (d, J = 56 Hz, 2F).

Mass: (IE): (M +)⁺127-128

Synthesis of compound 13



- 20 Into a round-bottom flask under an inert atmosphere containing a solution of (difluoromethyl) benzene 12 (1.74 g; 13 mmol; 1 eq.) in carbon tetrachloride (distilled over P₂O₅) is added N-bromosuccinimide (NBS) (5.07 g; 28 mmol; 2.1 eq.). The round-bottom flask is then provided with a cooler and the reaction medium is refluxed (80°C)

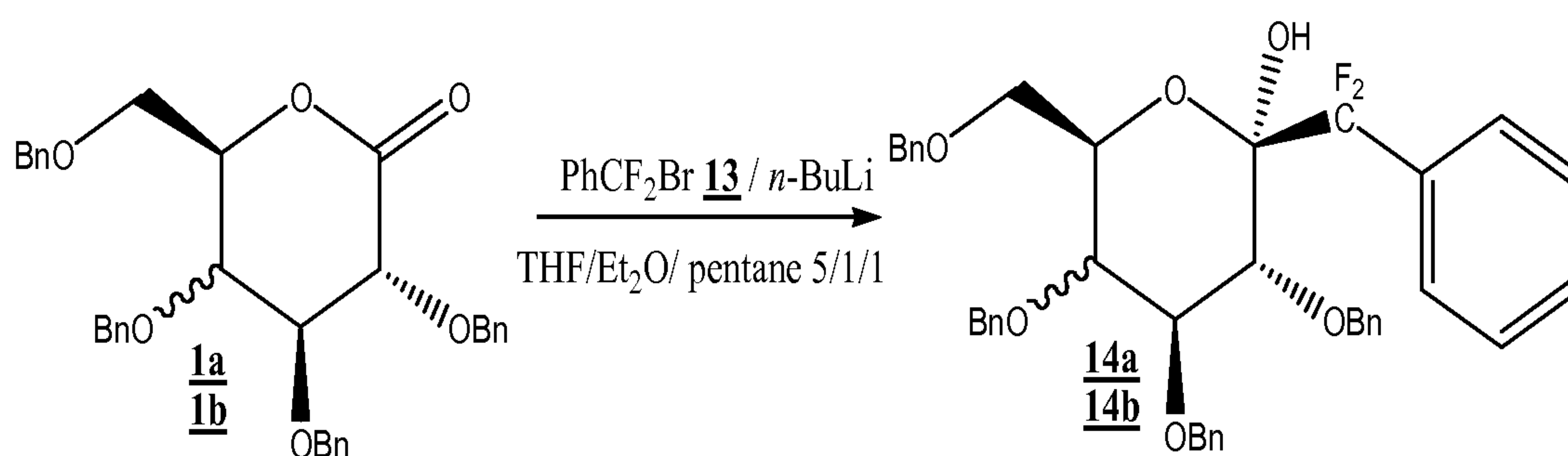
and irradiated by means of a mercury vapour UV lamp for 40 h. The mixture is then filtered, washed with water and extracted with dichloromethane. The organic phase is dried over magnesium sulphate, filtered, concentrated and then distilled under low pressure (bp $T^\circ = 47^\circ\text{C}$; $P^\circ = 61\text{ mBar}$) to produce compound 13 in the form of a colourless liquid, with a yield of 60%.

13: $\text{C}_7\text{H}_5\text{BrF}_2$ $M = 207.02\text{ g.mol}^{-1}$

NMR ^{19}F (CDCl_3 , 282.5 M_z): -43.5 (s, 2F)

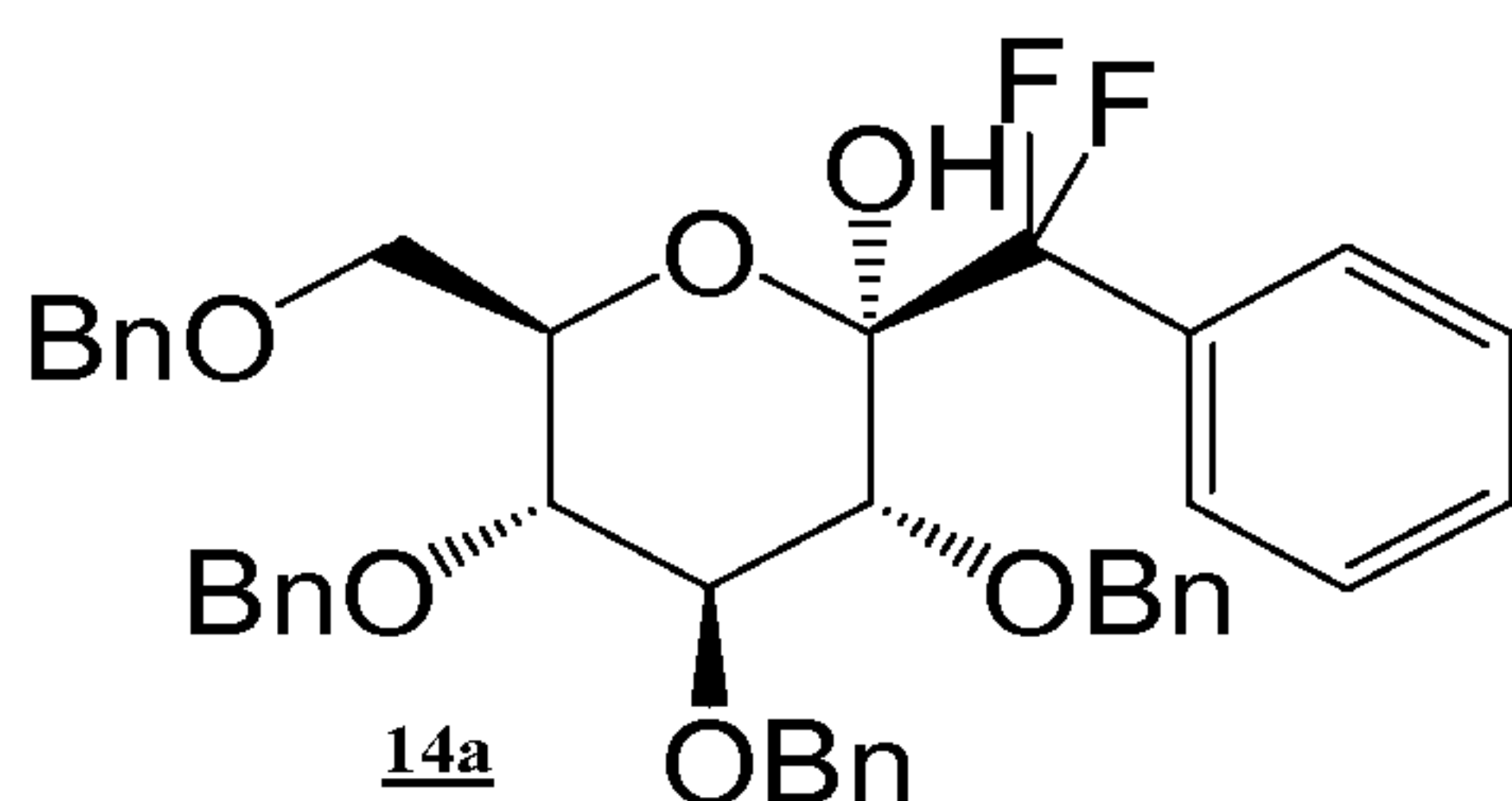
Mass: (IE): 206-208 (M^+).

10 Synthesis of compound 14a and 14b



A 1.6 M solution of *n*-butyllithium in hexane (3.19 mL, 5.10 mmol, 5.5 eq.) is added to a round-bottom flask under an inert atmosphere, which contains a solution of 13 (0.81 mg, 3.71 mmol, 4 eq.) and the lactone 1a (0.50 g, 0.93 mmol, 1 eq.) in THF (10 mL) at -78°C. The cooling bath was removed and the reaction mixture was stirred overnight at ambient temperature. A saturated aqueous ammonium chloride solution is then added. The reaction medium is extracted with ethyl acetate and then dried over magnesium sulphate, prior to being concentrated. The residue is then purified on a chromatography column (95/5 to 90/10 cyclohexane/ethyl acetate eluent) to produce compound 14a, in the form of a white solid, with a yield of 78%.

Compound 14b (0.616 g, 71% yield, white solid) is prepared according to the procedure described above from lactone 1b (0.70 g, 1.30 mmol, 1 eq.), 1-(bromodifluoromethyl)-2-chlorobenzene 13 (1.38 g, 5.20 mmol, 4 eq.), and 1.6 M *n*-butyllithium (4.06 mL, 6.50 mmol, 5 eq.).

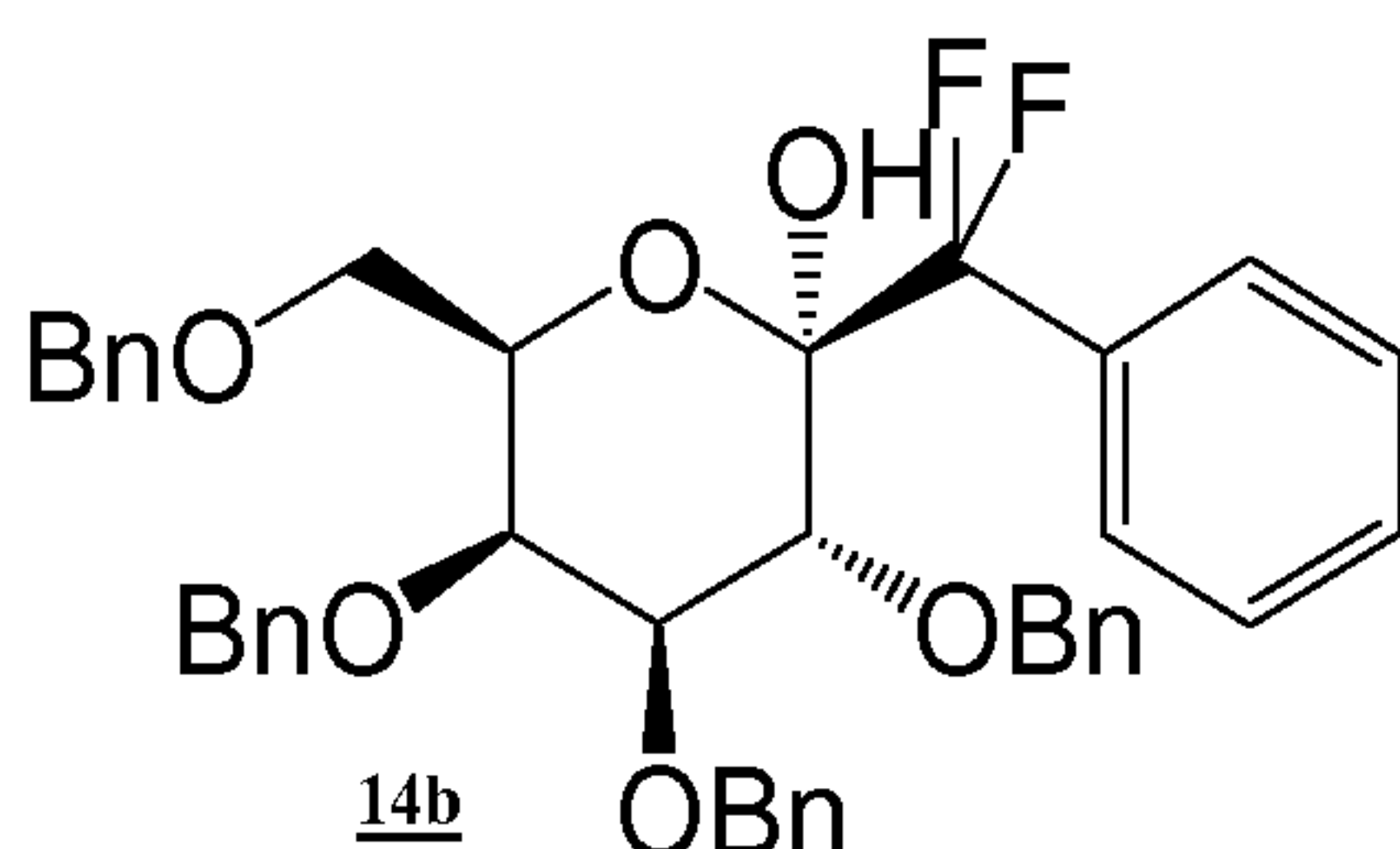


14a: C₄₁H₄₀F₂O₆ M = 666.75 g.mol⁻¹

Rf: 0.41 (cyclohexane/ethyl acetate) 8/2

NMR ¹⁹F (CDCl₃, 282.5 Mz): -108.2 (d, J = 250 Hz, 1F); -109.1 (d; J = 250 Hz, 1F).

Mass: (ESI +): 684.3 (M + H₂O); 689.3 (M + Na⁺) ; 705.3 (M + K⁺).



14b: C₄₁H₄₀F₂O₆ M = 666.75 g.mol⁻¹

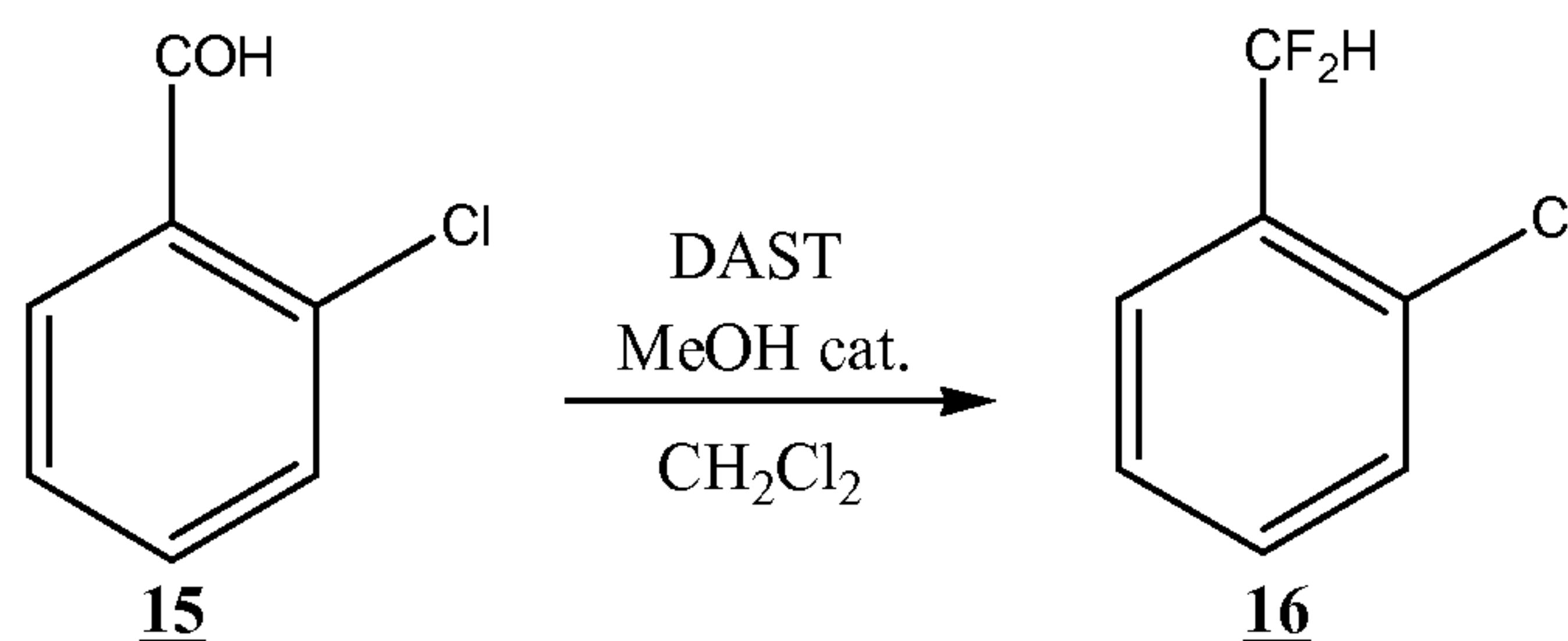
Rf: 0.41 (cyclohexane/ethyl acetate) 8/2

NMR ¹⁹F (CDCl₃, 282.5MHz) : -108.03 (1F, d, J=251Hz); -109.13 (1F, d, J=252Hz).

Mass (ESI⁺): 718.53 (M+H₂O).

Anal. Calcd : C, 73.86; H, 6.05 Found: C, 73.84; H, 5.99.

15 Synthesis of compound 16



Diethylaminosulfur trifluoride (DAST) (2.97 mL; 24 mmol; 1.7 eq.) is added dropwise into a round-bottom flask under an inert atmosphere containing ortho-chlorobenzaldehyde **15** (2 g; 14 mmol; 1 eq.) in dichloromethane (15 mL). A drop of anhydrous methanol is then added to the reaction medium in order to catalyze the

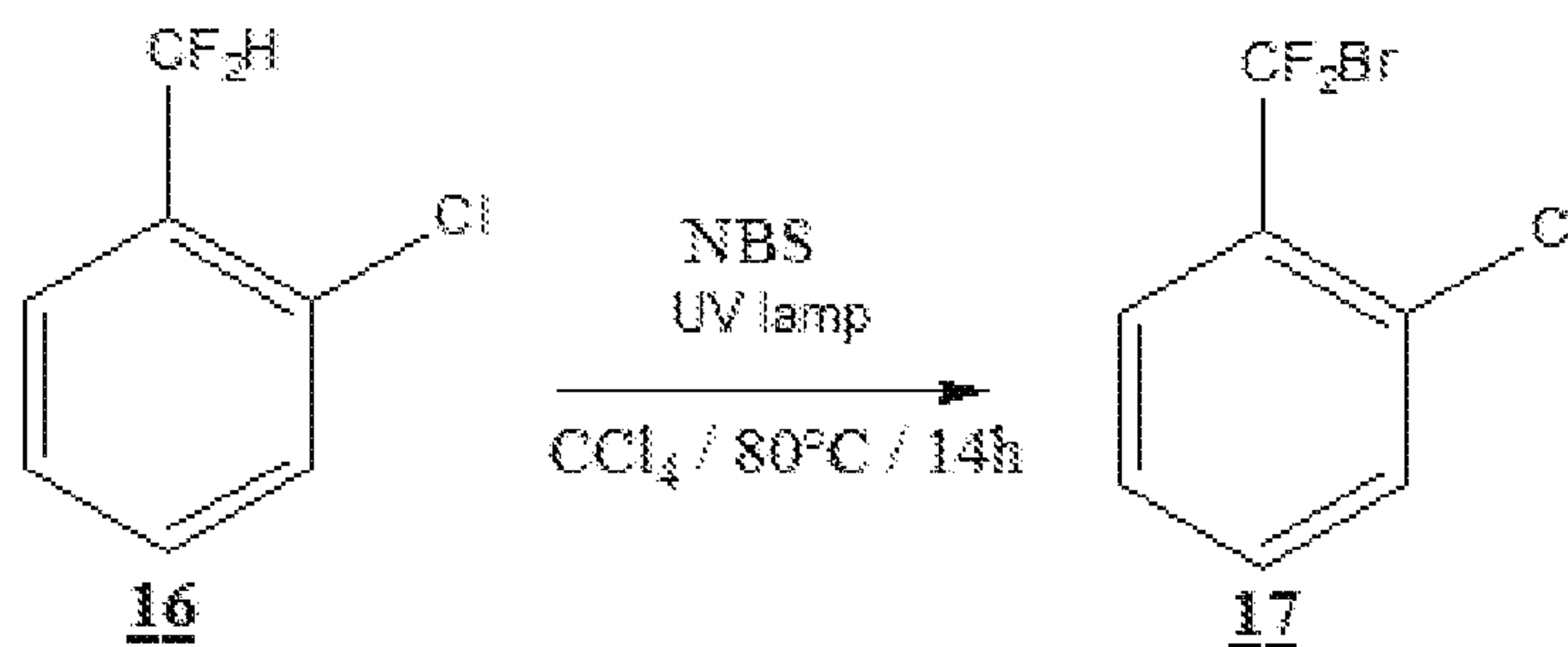
reaction. The mixture is stirred for 16 h at ambient temperature and is then cooled to 0°C before adding a saturated aqueous sodium bicarbonate solution until the neutral state is reached. The mixture is then extracted with dichloromethane and dried over magnesium sulphate. The organic phase is distilled under low pressure (bp $T^\circ = 40$ -48°C; $P^\circ = 61$ mBar) to produce compound 16 in the form of a colourless liquid, with a yield of 45%.

16: $C_7H_5ClF_2$ $M = 162.56 \text{ g.mol}^{-1}$

NMR ^{19}F ($CDCl_3$, 282.5 Mz): -115.2 (d, $J = 54$ Hz, 2F).

Mass: (IES): 161-162-163-164 (M)

Synthesis of compound 17

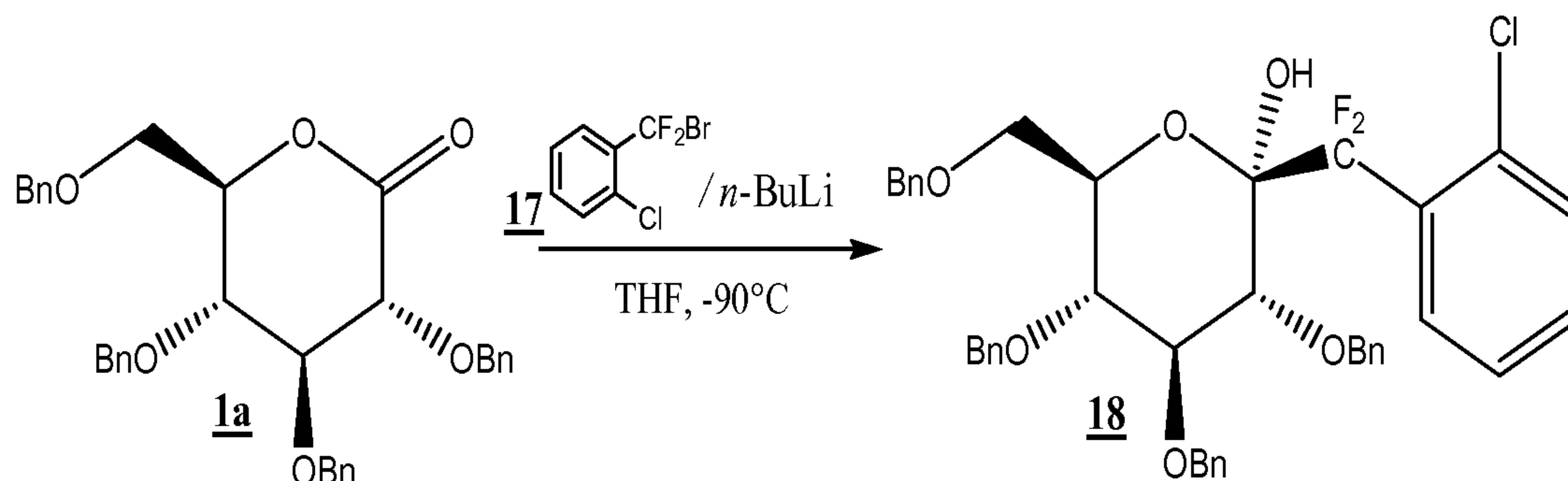


N-bromosuccinimide (0.115 g; 0.6 mmol; 2.1 eq.) is added to a quartz reactor under an inert atmosphere, which is surmounted by a mercury vapour UV lamp provided with a cooling system, and which contains a solution of ortho-chloro (difluoromethyl) benzene 16 (0.05 g; 0.3 mmol; 1 eq.) in carbon tetrachloride (distilled over P_2O_5). The reaction medium is refluxed and is irradiated for 14 h. The mixture is then filtered, washed with water and extracted with dichloromethane. The organic phase is dried over magnesium sulphate, filtered and then concentrated to produce compound 17 in the form of a yellow oil, with a conversion rate of 75% (^{19}F NMR).

17: $C_7H_4BrClF_2$ $M = 241.46 \text{ g.mol}^{-1}$

NMR ^{19}F ($CDCl_3$, 282.5 Mz): -45.5 (s, 2F)

Mass: (IE): (M⁺) 240-242-244.

Synthesis of compound 18

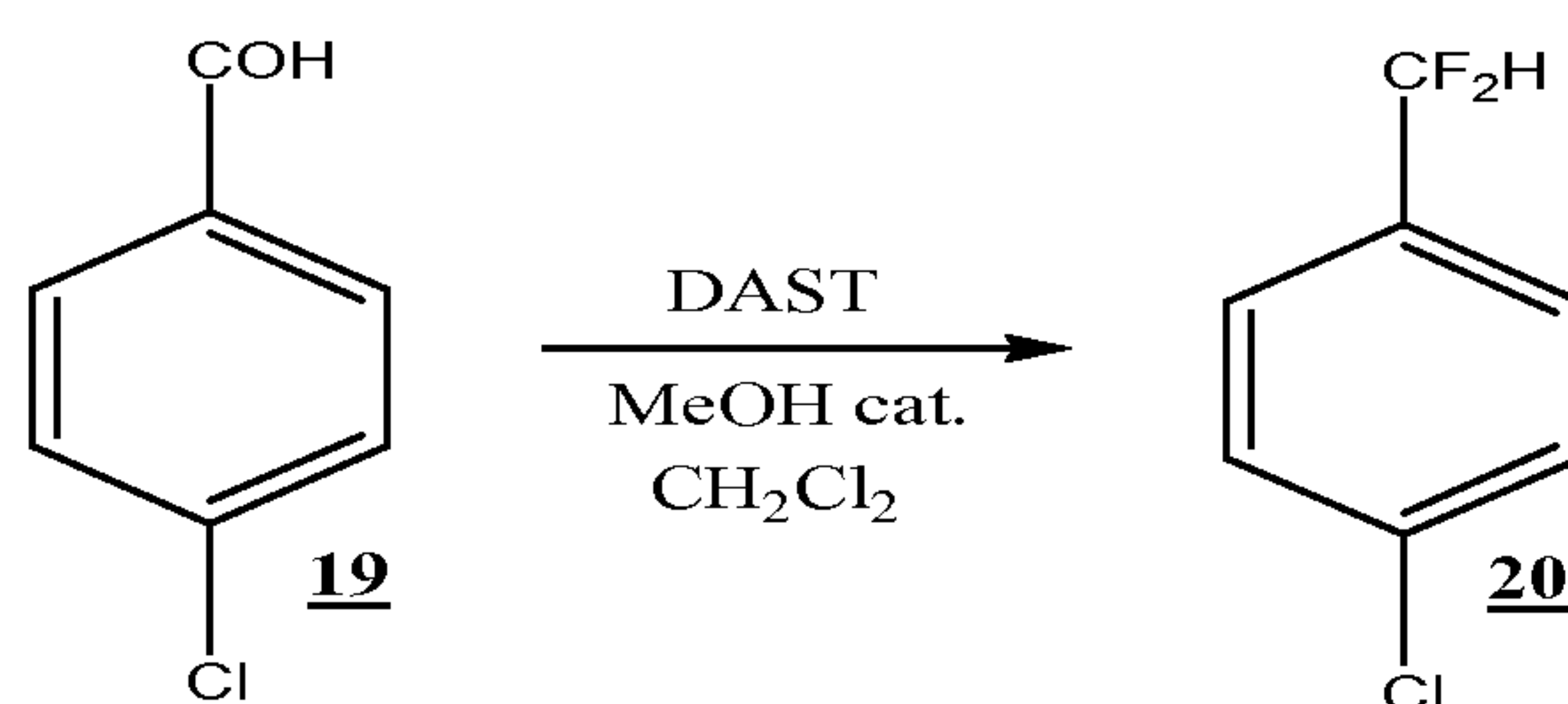
A 1.4 M solution of *n*-butyllithium in hexane (0.20 mL; 0.29 mmol; 1.5 eq.) is added to a round-bottom flask under an inert atmosphere, which contains the ortho-chloro (bromodifluoromethyl) benzene 17 (46 mg; 0.14 mmol; 0.7 eq.) and the lactone 1a (107 mg; 0.19 mmol; 1 eq.) at -90°C. The mixture is stirred for one hour at this temperature. A saturated aqueous ammonium chloride solution is then added at ambient temperature. The reaction medium is extracted with ethyl acetate and then dried over magnesium sulphate, prior to being concentrated. The residue is then purified on a chromatography column (95/5 cyclohexane/ethyl acetate eluent) to produce compound 18, in the form of a colourless oil, with a yield of 33%.

18: C₄₁H₃₉F₂O₆ M = 701.19 g.mol⁻¹

Rf: 0.35 (cyclohexane/ethyl acetate) 8/2

NMR ¹⁹F (CDCl₃, 282.5 Mz): -103.5 (d, *J* = 255 Hz, 1F), -106, 7 (d, *J* = 255 Hz, 1F)

Mass: (ESI +): 718.27 (M + H₂O); 723.33 (M + Na⁺)

Synthesis of compound 20

Diethylaminosulfur trifluoride (DAST) (1.48 mL; 12 mmol; 1.7 eq.) is added dropwise into a round-bottom flask under an inert atmosphere containing para-chlorobenzaldehyde 19 (1 g; 7.1 mmol; 1 eq.) in dichloromethane (15 mL). A drop of anhydrous methanol is then added to the reaction medium in order to catalyze the

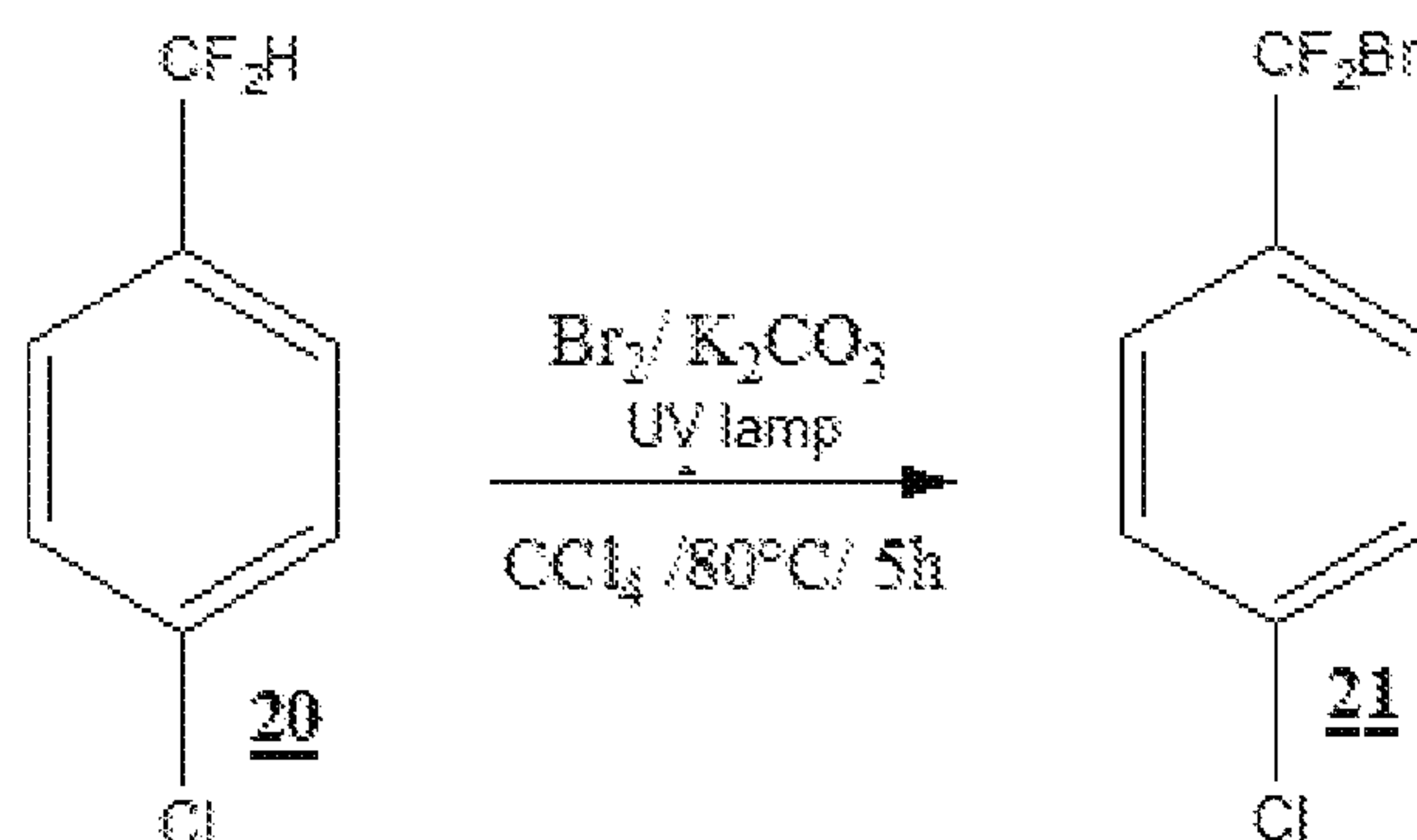
reaction. The mixture is stirred for 16 h at ambient temperature (88% conversion rate determined by gas chromatography (GC)), and then cooled to 0°C, before to add a saturated aqueous sodium bicarbonate solution until the neutral state is reached. The mixture is then extracted with dichloromethane, dried over magnesium sulphate, filtered and distilled under low pressure (bp T° = 40-48°C; P° = 61 mBar) to produce compound 20.

20: C₇H₅ClF₂ M = 162.56 g.mol⁻¹

NMR ¹⁹F (CDCl₃, 282.5 Mz): -110.79 (d, J = 56 Hz, 2F).

Mass: (IE): 161-162-163-164 (M +).

Synthesis of compound 21

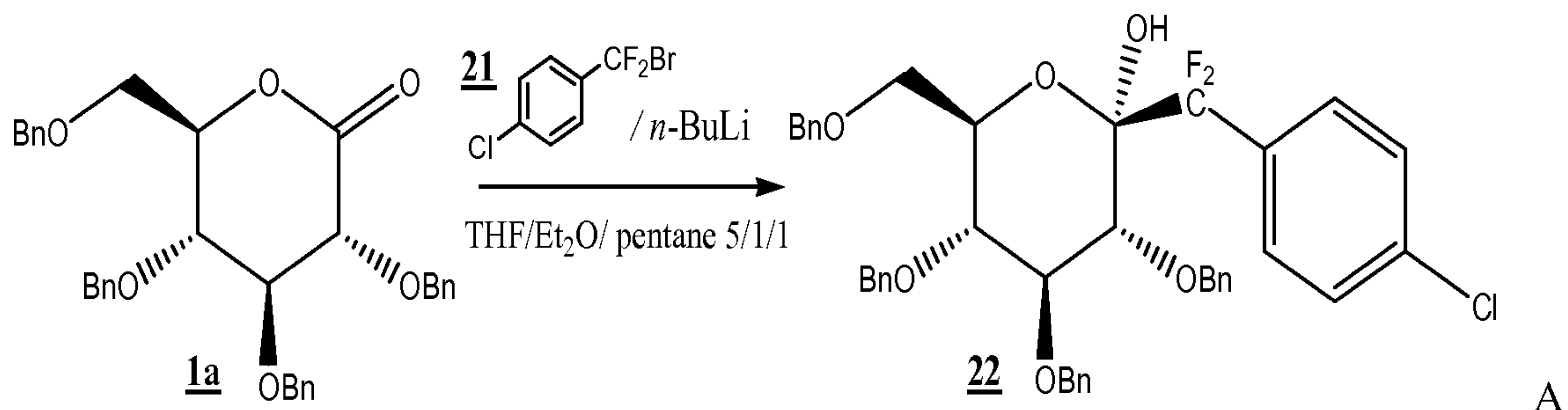


Bromine (17 µL; 0.34 mmol; 1.1 eq.) is added to a quartz reactor under an inert atmosphere, which is surmounted by a mercury vapour UV lamp equipped with a cooling system, and which contains a solution of para-chloro (difluoromethyl) benzene 20 (50 mg; 0.3 mmol; 1 eq.) in carbon tetrachloride (5 mL; distilled over P₂O₅) and potassium carbonate (0.212 g; 1.54 mmol; 5 eq.). The reaction medium is refluxed and is irradiated for 5 h (84% conversion rate determined by GC). The mixture is then filtered and concentrated. The product 21 is involved in the following step without purification.

21: C₇H₄BrClF₂ M = 241.46 g.mol⁻¹

NMR ¹⁹F (CDCl₃, 282.5 Mz): - 43.9 (s, 2F).

Mass: (IE): (M +) 240-242-244.

Synthesis of compound 22

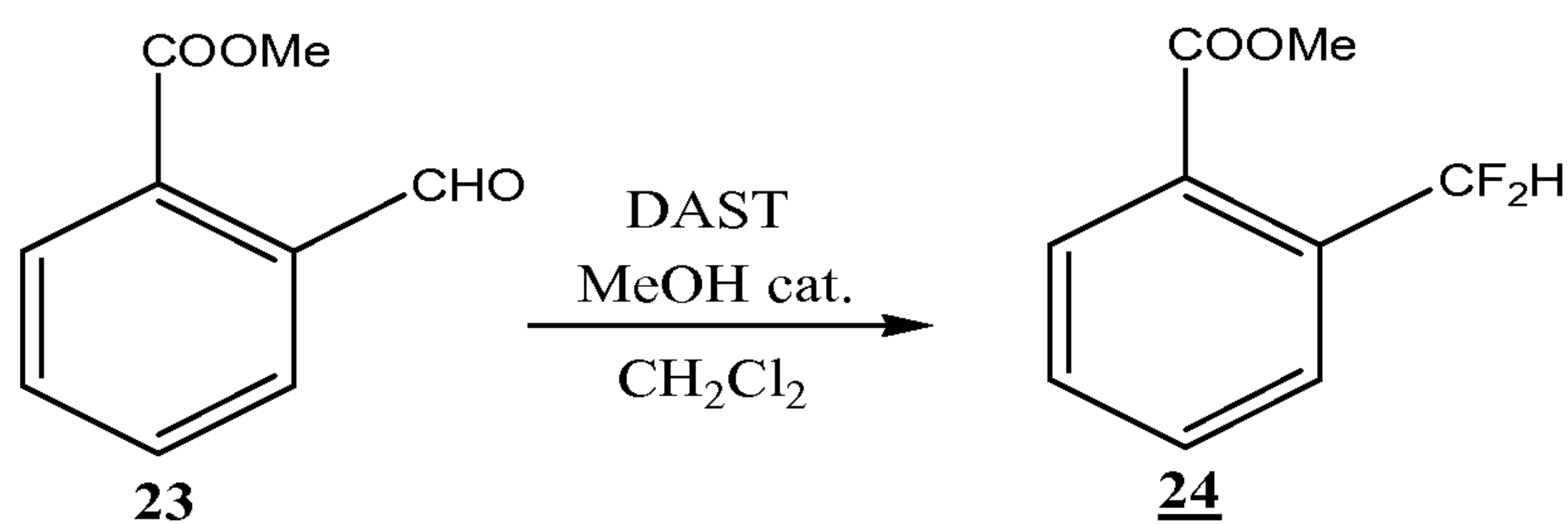
1.4 M solution of *n*-butyllithium in hexane (0.044 mL; 0.06 mmol; 1.5 eq.) is added, at -90°C, to a round-bottom flask under an inert atmosphere, which contains the para-chloro (bromodifluoromethyl) benzene 21 (15 mg; 0.06 mmol; 1.5 eq.) and the lactone 1a (22 mg; 0.04 mmol; 1 eq.) in a mixture of tetrahydrofurane, diethyl ether and pentane, in proportions of 5: 1: 1 (2.5 mL: 0.5 mL: 0.5 mL). The mixture is stirred for 1 h 30 min at this temperature. A saturated aqueous ammonium chloride solution is then added at ambient temperature. The reaction medium is then extracted with ethyl acetate and then dried over magnesium sulphate, prior to being concentrated, in order to produce compound 22.

22: C₄₁H₃₉F₂O₆ M = 701.19 g.mol⁻¹

R_f: 0.45 (cyclohexane/ethyl acetate 8/2).

NMR ¹⁹F (CDCl₃, 282.5 Mz): -108.8 (s, 1F), -108.8 (s, 1F).

Mass: (ESI +): 719.13 (M + H₂O).

Synthesis of compound 24

Diethylaminosulfur trifluoride (DAST) (3.8 mL; 31 mmol; 1.7 eq.) is added dropwise into a round-bottom flask under an inert atmosphere, which contains methyl 2-formylbenzoate 23 (3 g; 18 mmol; 1 eq.) in dichloromethane (20 mL). A drop of anhydrous methanol is then added to the reaction medium in order to catalyze the reaction. The mixture is stirred for 16 h at ambient temperature (82% conversion rate

determined by GC) and then cooled to 0°C prior to adding a saturated sodium bicarbonate solution until the neutral state is reached. The mixture is then extracted with dichloromethane, dried over magnesium sulphate and then concentrated. The residue is then purified on a chromatography column (95/5 cyclohexane/ethyl acetate eluent) to produce compound 24, in the form of a yellow oil, with a yield of 66%.

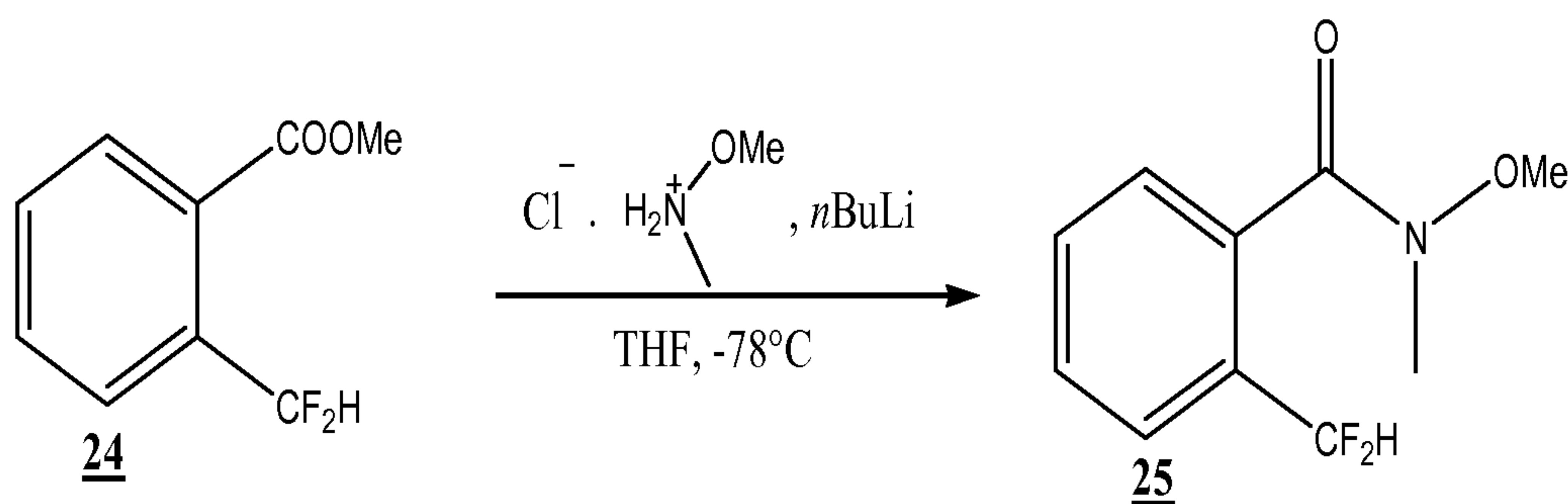
24: C₉H₈F₂O₂ M = 186.16 g.mol⁻¹

Rf: 0.44 (cyclohexane/ethyl acetate) 9/1.

NMR ¹⁹F (CDCl₃, 282.5 Mz): -114.2 (d, J = 56 Hz, 2F).

Mass: (ESI +) : 187.07 (M + H).

Synthesis of compound 25



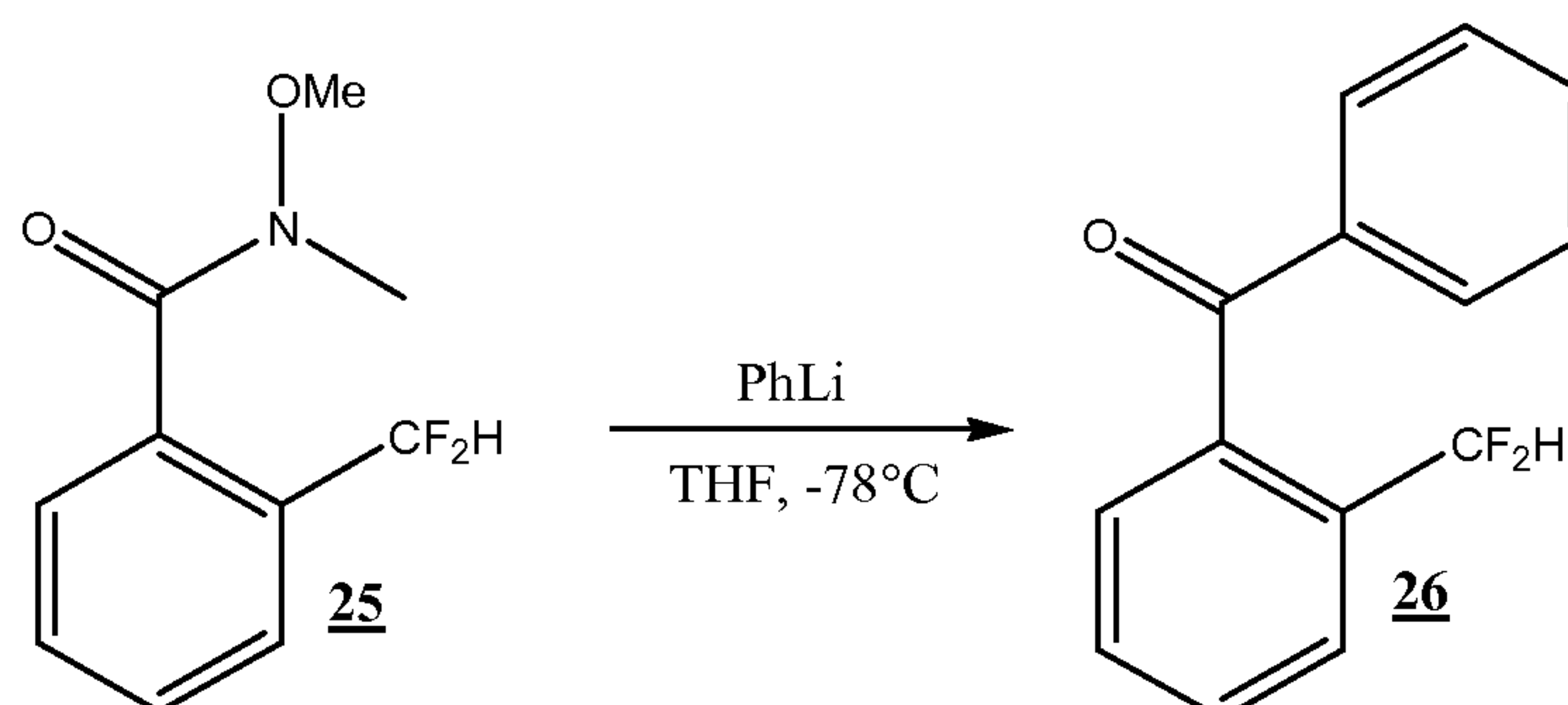
A solution of *n*-butyllithium 1.5 M (10.9 mL; 16 mmol; 6 eq.) is added, at -78°C, to a round-bottom flask under an inert atmosphere, which contains Weinreb amine (0.786 g; 8.0 mmol; 3 eq.) in anhydrous tetrahydrofuran (20 mL). The mixture is stirred at -78°C for 10 min. The methyl 2-difluoromethylbenzoate 24 (0.500 g; 2.69 mmol; 1 eq.) in tetrahydrofuran (5 mL) is then added at -78°C. After stirring for 20 min, the mixture can return to ambient temperature and saturated aqueous ammonium chloride solution is added. The reaction medium is then extracted with ethyl acetate, dried over magnesium sulphate and concentrated. The residue is then purified on a chromatography column (8/2 cyclohexane/ethyl acetate eluent) in order to produce compound 25, in the form of a yellowish oil, with a yield of 61%.

25: C₁₀H₁₁F₂NO₂ M = 215.20 g.mol⁻¹

Rf: 0.17 (cyclohexane/ethyl acetate) 8/2.

NMR ¹⁹F (CDCl₃, 282.5 Mz): -113.3 (brd, J = 50 Hz, 2F).

Mass: (IE): 215 (M +).

Synthesis of compound 26

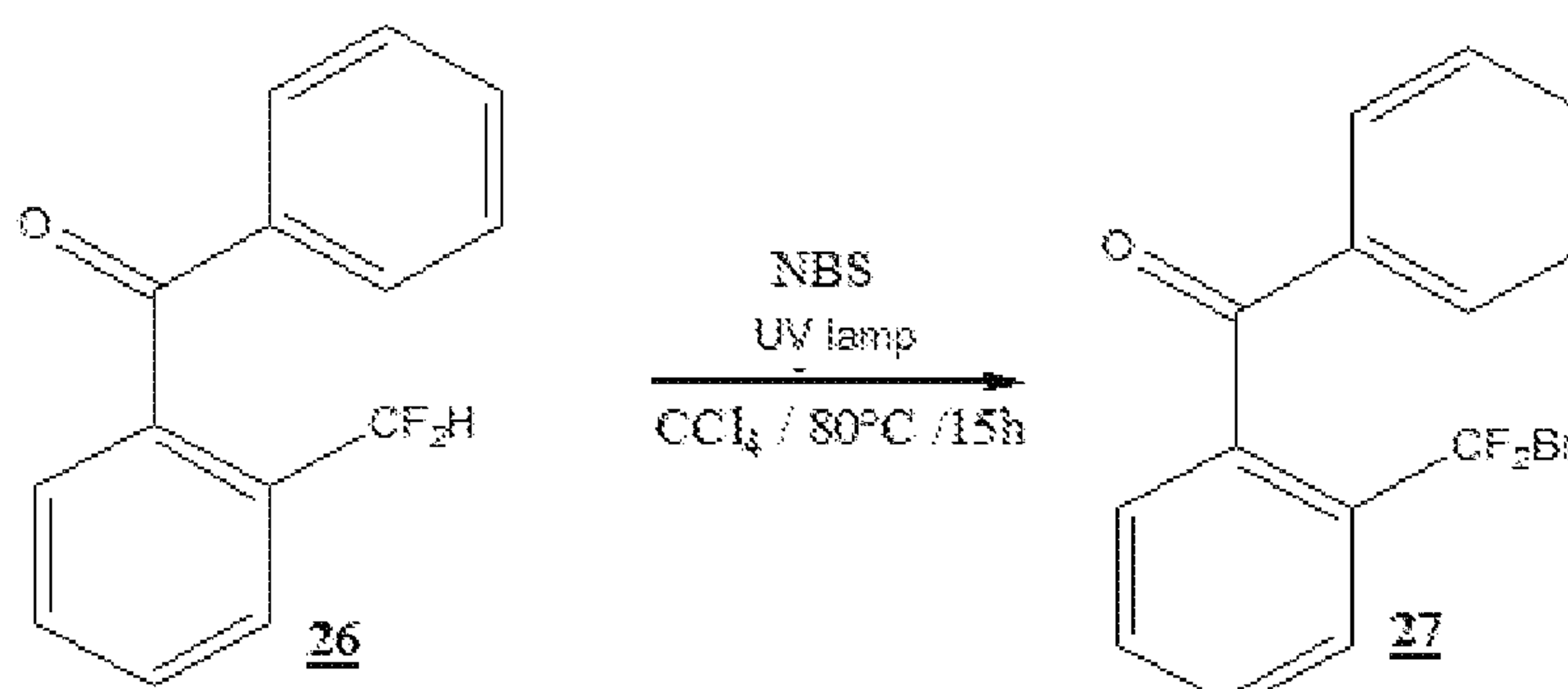
A solution of phenyllithium 1.8 M in diisobutylether (38.0 ml; 67.4 mmol; 2 eq.) is added to a round-bottom flask under an inert atmosphere which contains a solution of compound 25 (7.25 g, 33.7 mmol, 1 eq.) in dry tetrahydrofuran (75 mL) at -78°C . The reaction mixture is stirred for one hour at this temperature. A saturated aqueous ammonium chloride solution is then added at ambient temperature and the reaction medium is extracted with ethyl acetate. The organic phase is then dried over magnesium sulphate and concentrated. The residue is then purified on a chromatography column (9/1 cyclohexane/ethyl acetate eluent) in order to produce compound 26, in the form of a yellow oil, with a yield of 77%.

26: $\text{C}_{14}\text{H}_{10}\text{F}_2\text{OM} = 232.23 \text{ g.mol}^{-1}$

Rf: 0.52 (cyclohexane/ethyl acetate) 8/2.

NMR ^{19}F (CDCl_3 , 282.5 Mz): -112.2 (d, $J = 56 \text{ Hz}$, 2F).

Mass: (IE): 232 (M^+).

Synthesis of compound 27

N-bromosuccinimide (0.926 g; 5.2 mmol; 2.1 eq.) is added to a quartz reactor under an inert atmosphere, which is surmounted by a mercury vapour UV lamp provided with a cooling system, and which contains a solution of compound 26 (0.575 g; 2.48 mmol; 1

eq.) in carbon tetrachloride (15 mL; distilled over P₂O₅). The reaction medium is refluxed and is irradiated for 15 h, making a second addition of *N*-bromosuccinimide (0.926 g; 5.2 mmol; 2.1 eq.) after 7 h. The mixture is washed with water and extracted with dichloromethane. The organic phase is dried over magnesium sulphate and then concentrated. The residue is then purified via a chromatography column (9/1 cyclohexane/ethyl acetate eluent) in order to produce compound 27, in the form of a colourless oil, with a yield of 45%.

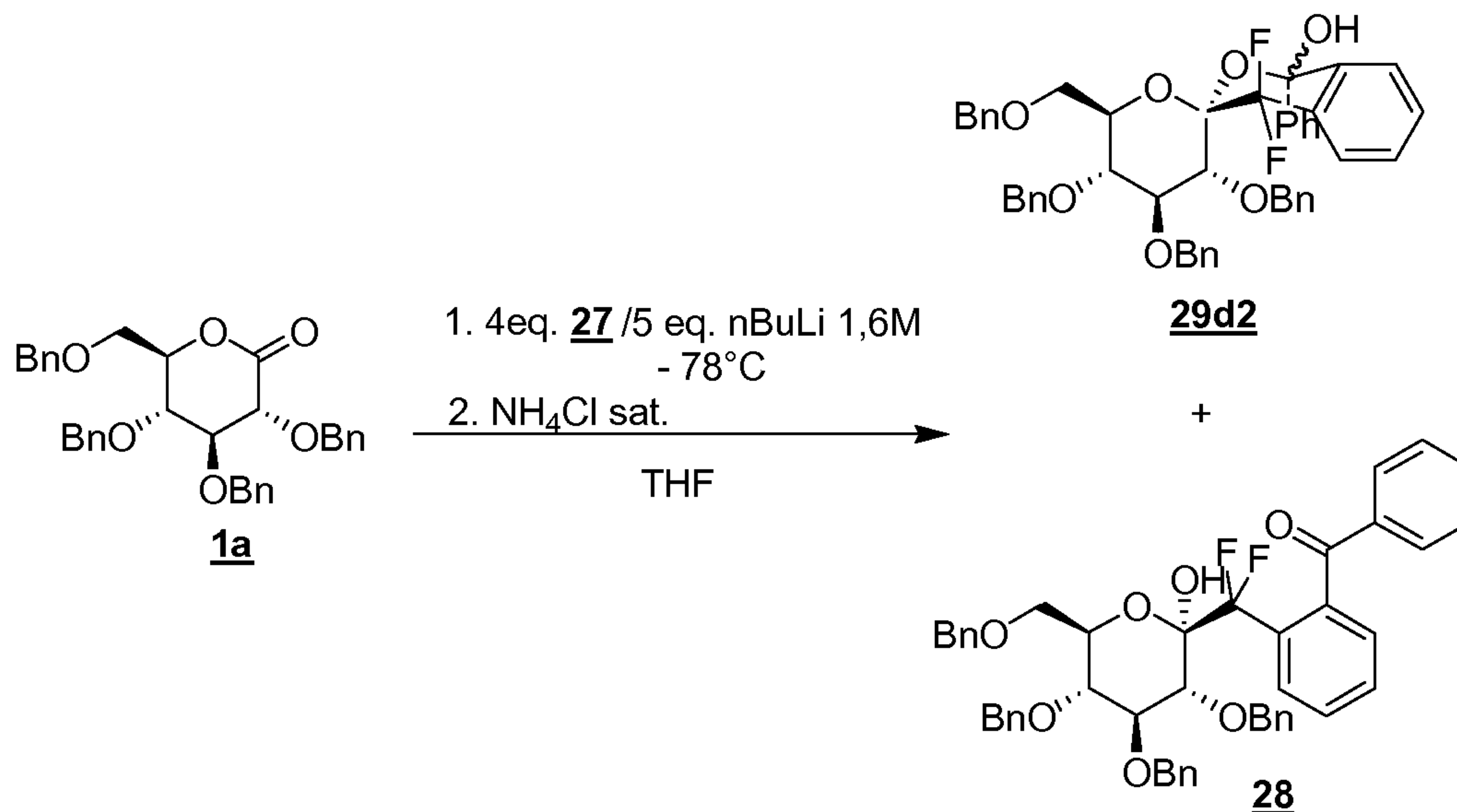
27: C₁₄H₉BrF₂O M = 331.12 g.mol⁻¹

Rf: 0.45 (cyclohexane/ethyl acetate) 8/2.

NMR ¹⁹F (CDCl₃, 282.5 Mz): -40.5 (s, 2F).

Mass: (IE): 331 (M⁺).

Synthesis of compound 28 and 29d1/d2



A 1.6 M solution of *n*-butyllithium in hexane (0.580 mL; 0.93 mmol; 5 eq.) is added to a round-bottom flask under an inert atmosphere, which contains a solution of 27 (0.304 g; 0.74 mmol; 4 eq.) and the lactone 1a (0.100 mg; 0.186 mmol; 1 eq.) in THF (10mL) at -78°C. The cooling bath was removed and the reaction mixture was stirred overnight. A saturated aqueous ammonium chloride solution is then added at ambient temperature. The reaction medium is extracted with ethyl acetate and then dried over magnesium sulphate, prior to being concentrated. The residue is then purified on a chromatography column (90/10 cyclohexane/ethyl acetate eluent) to produce compound 29d2 and 28 in the form of a colourless oil.

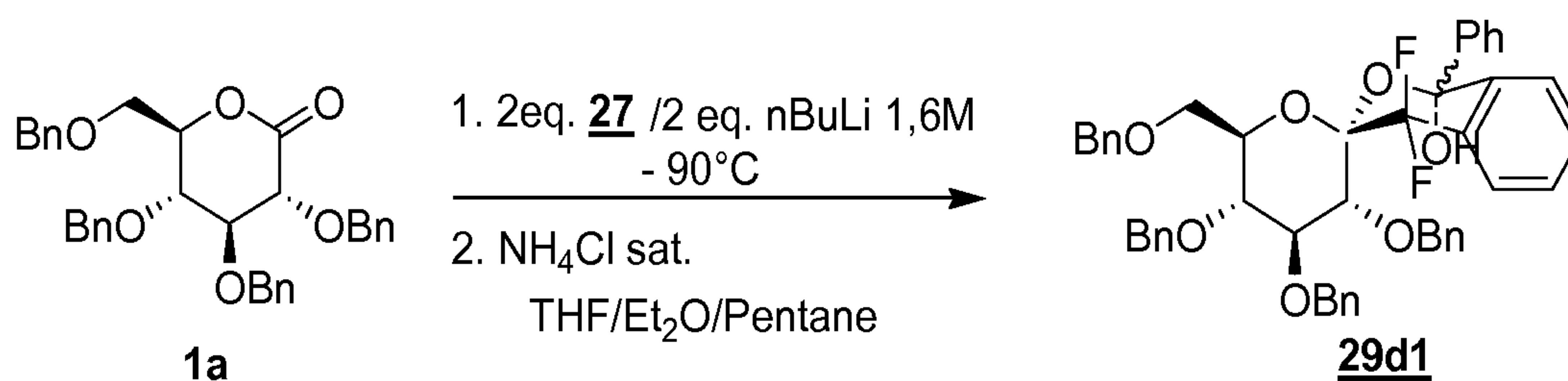
28 et 29d2 C₄₈H₄₄F₂O₇ M = 770.86 g.mol⁻¹

NMR ¹⁹F (CDCl₃, 282.5MHz) :

29d2: -93.6 (d, 263 Hz, 1F); -122.8 (d, 263 Hz, 1F)

28: -97.5 (d, 258Hz, 1F); -101.8 (d, 258Hz, 1F)

Mass (ESI⁺): 753.2 (M-H₂O+H); 788.2 (M+H₂O)



A

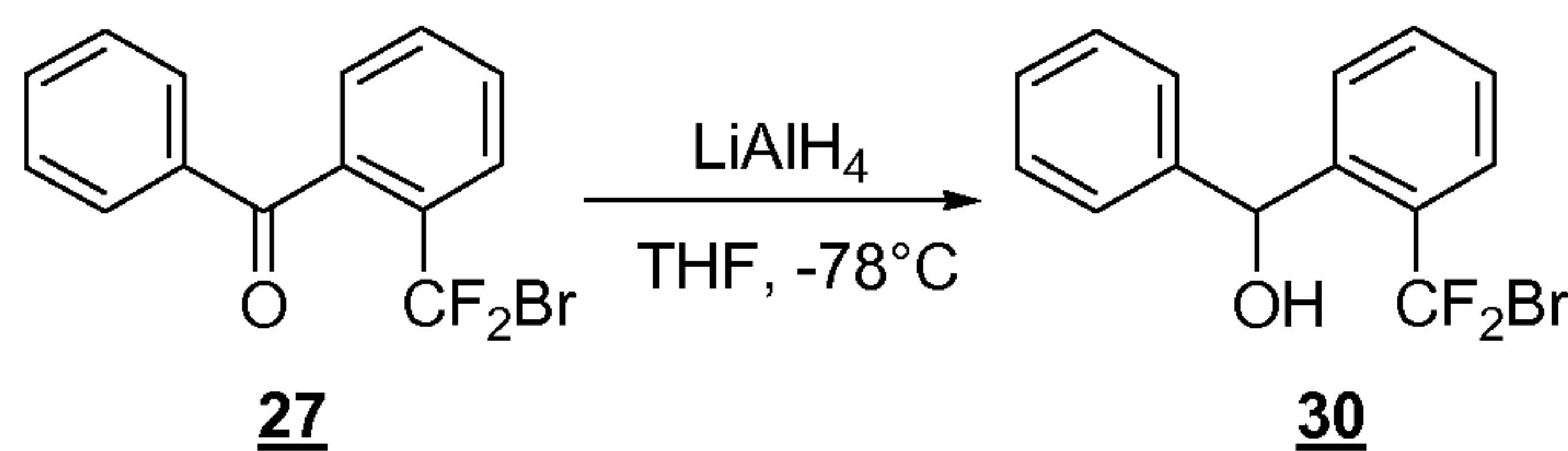
1.6 M solution of *n*-butyllithium in hexane (0.200 mL; 0.32 mmol; 2 eq.) is added to a round-bottom flask under an inert atmosphere, which contains a solution of 27 (100 mg; 0.32 mmol; 2 eq.) and the lactone 1a (0.86 mg; 0.16 mmol; 1 eq.) in a mixture of tetrahydrofuran, diethyl ether and pentane, in proportions of 5/1/1 (3.5 mL) at -90°C. The mixture is stirred for 1 h 30 min at this temperature. A saturated aqueous ammonium chloride solution is then added at ambient temperature. The reaction medium is extracted with ethyl acetate and then dried over magnesium sulphate, prior to being concentrated. The residue is then purified on a chromatography column (95/5 cyclohexane/ethyl acetate eluent) to produce compound 29d1, in the form of colourless oil, with a yield of 21%.

29d1: C₄₈H₄₄F₂O₇ M = 770.86 g.mol⁻¹

NMR ¹⁹F (CDCl₃, 282.5MHz) : -92.8 (1F, d, J = 259 Hz); -120.2 (1F, d, J = 259 Hz).

Mass (ESI⁺) : 753.2 (M-H₂O+H); 788.2 (M+H₂O).

Synthesis of compound 30



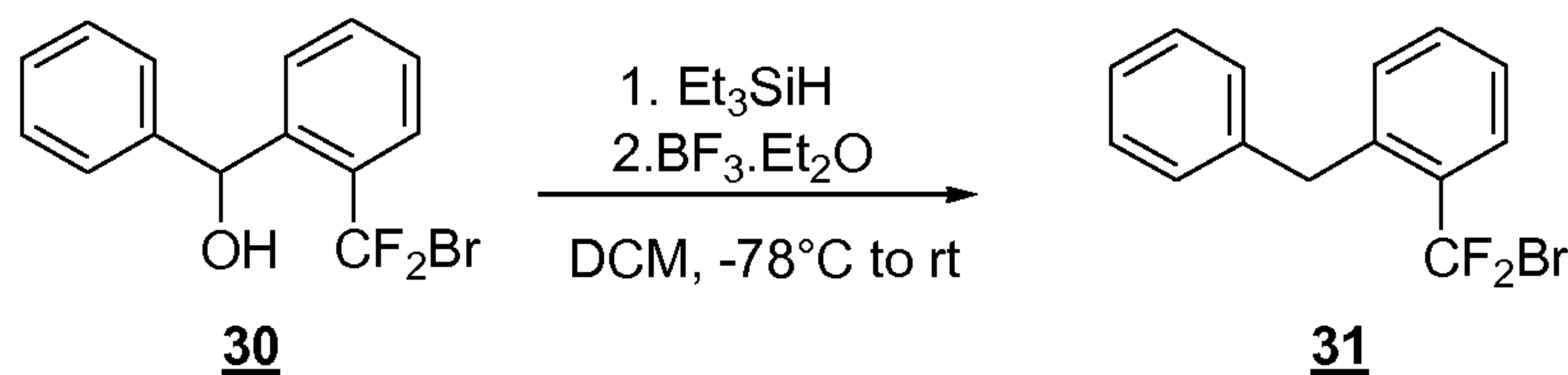
Lithium aluminium hydride (69.0 mg, 1.72 mmol, 1 eq.) is added in small portions, over a period of 15 min into a round-bottom flask under an inert atmosphere which contains a solution of 27 (0.54 g, 1.72 mmol; 1 eq.) in dry THF (17mL) at -78°C. The solution is stirred for 1h before a saturated ammonium chloride aqueous solution (a few drops) is added. The solution is filtered through celite and dried over magnesium sulphate, prior to being concentrated. The residue is then purified on a chromatography column (95/5 cyclohexane/ethyl acetate eluent) to produce compound 30, in the form of a light yellow liquid with a yield of 61%.

30: C₁₄H₁₁BrF₂O M = 313.14 g.mol⁻¹

NMR ¹⁹F (CDCl₃, 282.5 MHz): -42.5 (d, J = 160 Hz, 1F); -36.6 (d, J = 160 Hz, 1F).

Mass (EI): 231 (M-Br).

Synthesis of compound 31



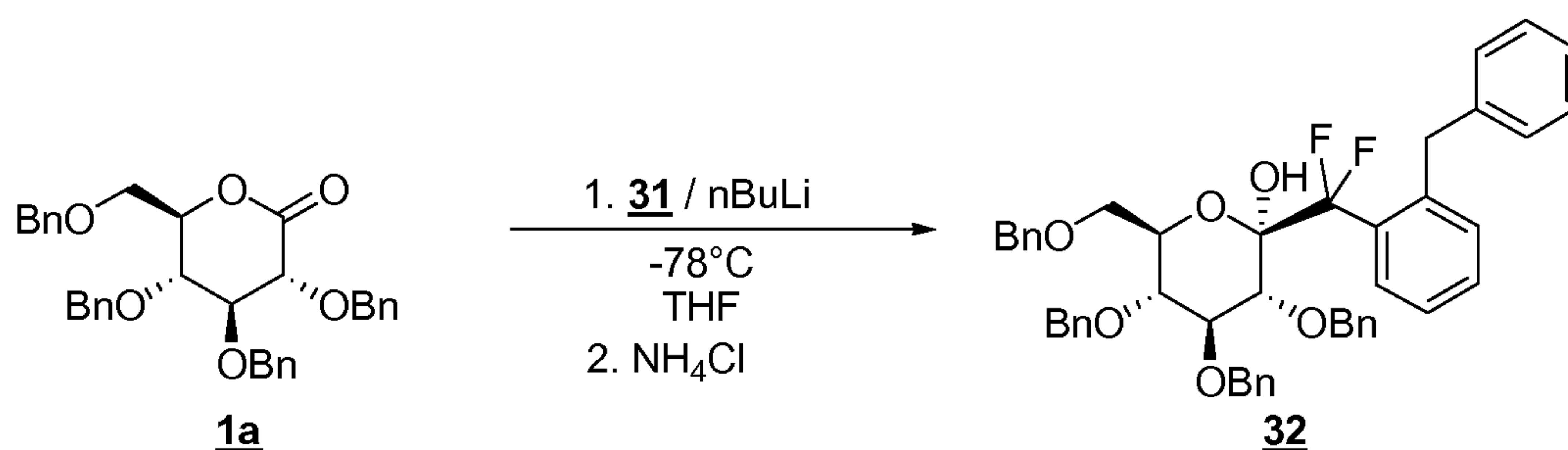
Triethylsilane (1.6 mL, 10.12 mmol, 10eq.) and boron trifluoride etherate (0.639 mL, 5.06 mmol, 5 eq.) are added successively into a round-bottom flask under an inert atmosphere which contains a solution of 30 (0.317 g, 1.01 mmol, 1 eq.) in dry dichloromethane (DCM) (15 mL) at -78°C. The cooling bath was removed and the reaction mixture was stirred overnight at ambient temperature. A saturated aqueous ammonium chloride solution is then added. The reaction medium is extracted with dichloromethane and then dried over magnesium sulphate, prior to being concentrated. The residue is then purified on a chromatography column (100 cyclohexane eluent) to produce compound 31, in the form of colourless oil, with a yield of 81%.

31: C₁₄H₁₁BrF₂ M = 297.11g.mol⁻¹

NMR ¹⁹F (CDCl₃, 282 MHz): -41.2 (s, 2F).

Mass (EI): 217 (M-Br).

Synthesis of compound 32



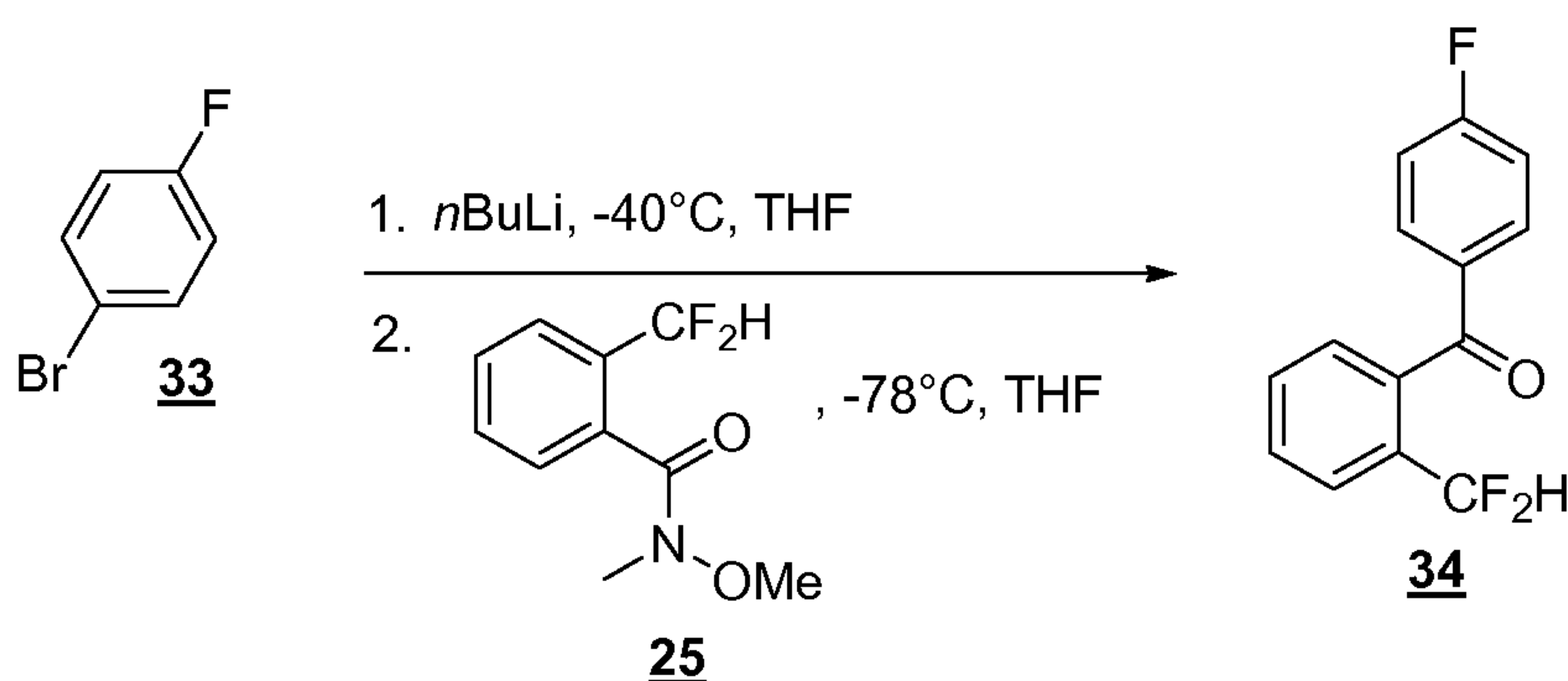
- 5 Compound 32 is prepared according to the procedure previously described (synthesis of compound 14a) from lactone 1a (0.100 g, 0.18 mmol, 1 eq.), compound 31 (0.166 g, 0.56 mmol, 3 eq.), and 1.5 M *n*-butyllithium (0.37 mL, 0.56 mmol, 3 eq.) to give a colorless oil, with a yield of 28%.

32: C₄₈H₄₆F₂O₆ M = 756.87 g.mol⁻¹

10 NMR ¹⁹F (CDCl₃; 282.5 MHz): -108.9 (d, J = 269 Hz, 1F); -101.2 (, d, J = 270 Hz, 1F).

Mass (ESI⁺): 777.33 (M+H₂O); 1529.53 (2M+H₂O).

Synthesis of compound 34



15

- A 1.5M solution of *n*-butyllithium in hexane (12.9 mL, 19.3 mmol, 2.8 eq.) is added to a round-bottom flask under an inert atmosphere, which contains a solution of 33 (2.3 mL, 20.7 mmol, 3 eq.) in anhydrous THF (50 mL) at -10°C. The mixture is stirred for 2h at -40°C. The temperature of the solution is brought down to -78°C and a solution of 25 (1.48 g, 6.90 mmol, 1 eq.) in THF (20 mL) is added at this temperature. The mixture is stirred for an additional 30 min and saturated ammonium chloride aqueous solution is added. The reaction medium is extracted with ethyl acetate and then dried over
- 20

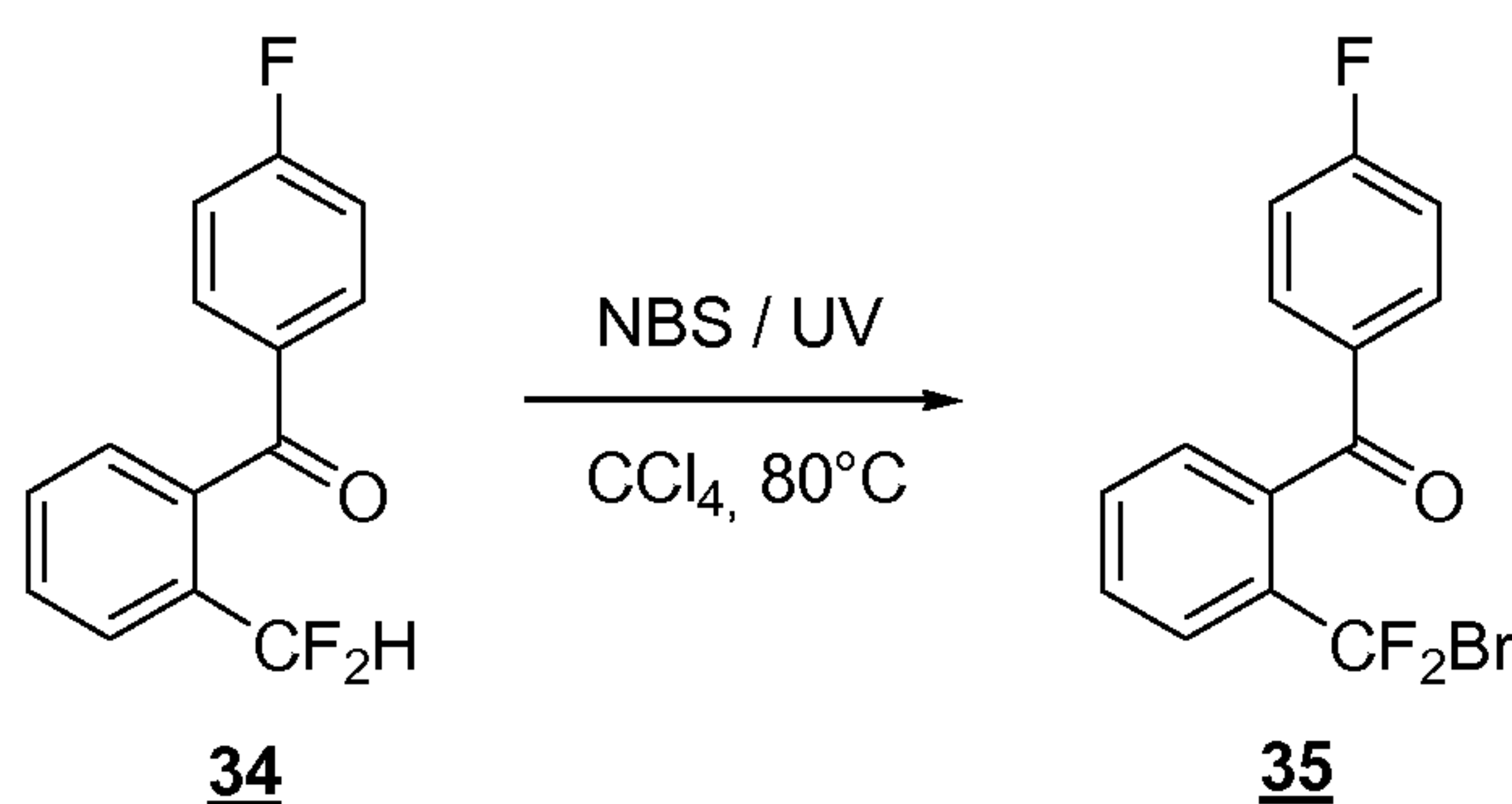
magnesium sulphate, prior to being concentrated. The residue is then purified on a chromatography column (100/0 to 97/03 cyclohexane/ethyl acetate eluent) to produce compound 34, in the form of a greenish oil, with a yield of 84%.

34: C₁₄H₉F₃O M = 250.22 g.mol⁻¹

5 NMR ¹⁹F (CDCl₃, 282.5 MHz): -124.4 (d, J = 56 Hz, 2F); 103.5 (m, 1F).

Mass (EI): 75-95-123-155-202-230-250.

Synthesis of compound 35



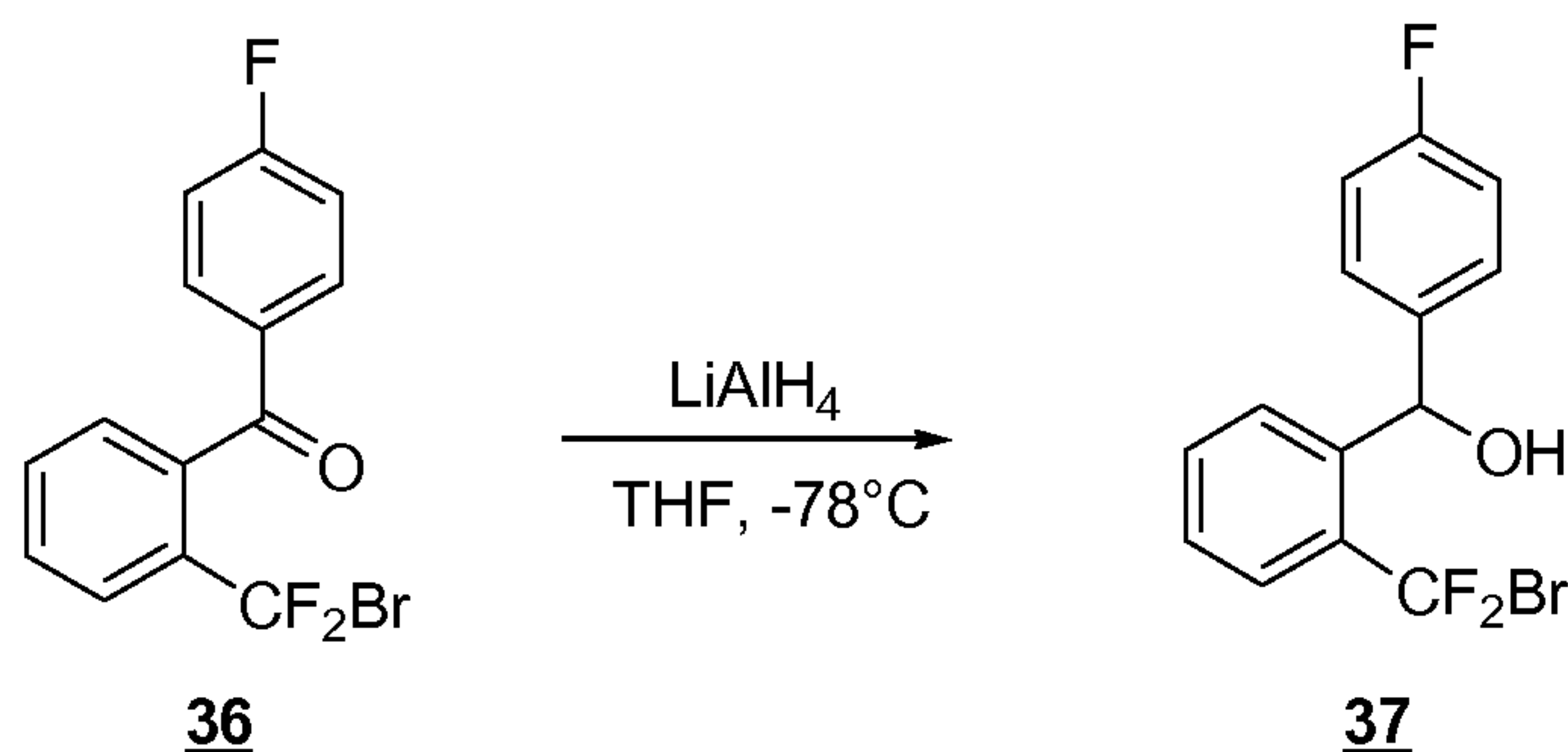
10 Compound 34 (1.45 g, 5.79 mmol, 1 eq.) was brominated with *N*-bromosuccinimide (4.33 g, 24.3 mmol, 4.1 eq.) according to the procedure previously described (synthesis of compound 27) to produce compound 35 in the form of a colourless oil, with a yield of 80%.

35: C₁₄H₈BrF₃O M = 329.11 g.mol⁻¹

15 NMR ¹⁹F (CDCl₃, 282.5 MHz): -103.6 (m, 1F); -40.6 (s).

Mass (EI): 75-95-123-201-229-249 (M-Br).

Synthesis of compound 37



20 A solution of 36 (0.95 mg, 2.90 mmol, 1 eq.) in dry THF (13 mL) is added to a round-bottom flask under an inert atmosphere which contains a suspension of lithium aluminium hydride (0.11 g, 2.90 mmol, 1 eq.) in dry THF (13 mL) at -78°C. The

solution is stirred for 1h30 before a saturated ammonium chloride aqueous solution (a few drops) is added. The solution is filtered through celite and dried over magnesium sulphate, prior to being concentrated. The residue is then purified on a chromatography column (98/02 to 80/20 cyclohexane/ethyl acetate eluent) to produce compound 37, in the form of an orange oil with a yield of 70%.

37: C₁₄H₁₀BrF₃O M = 331.13 g.mol⁻¹

NMR ¹⁹F (CDCl₃, 282.5 MHz): -36.7 (d, J = 159 Hz, 1F); -42.6 (d, J = 159 Hz, 1F); -115.1 (m, 1F).

Mass (EI): 77-97-125-127-183-201-211-231-249-330 (M).

Synthesis of compound 38



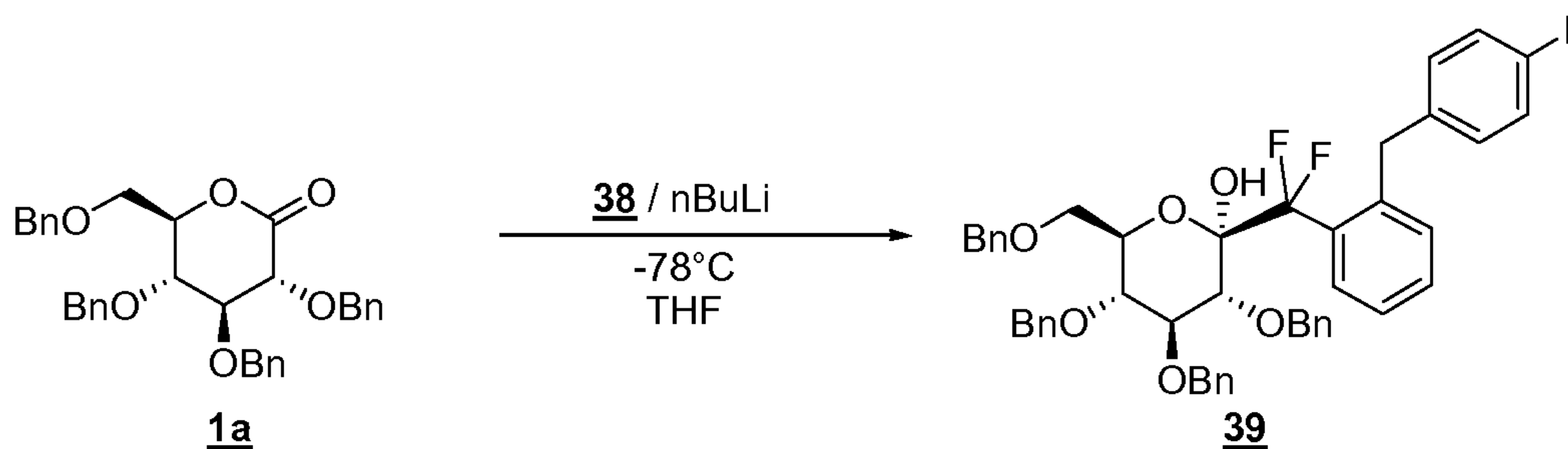
Compound 38 is prepared according to the procedure previously described (synthesis of 31) from compound 37 (0.477 g, 1.44 mmol, 1 eq.), triethylsilane (2.3 mL, 14.4 mmol, 10 eq.) and boron trifluoride etherate (0.91 mL, 7.20 mmol, 5 eq.), to give a yellowish liquid, with a yield of 100%.

38: C₁₄H₁₀BrF₃ M = 315.13 g.mol⁻¹

NMR ¹⁹F (CDCl₃, 282.5 MHz): -41.3 (s, 2F); -116.0 (m, 1F).

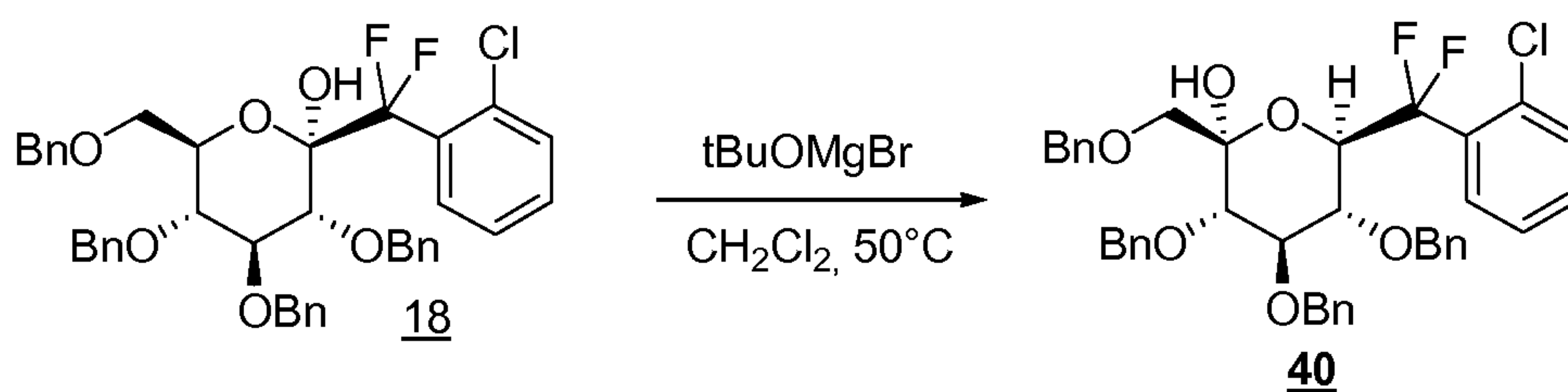
Mass (EI): 109-183-215-235-314-316 (M)

Synthesis of compound 39



Compound 39 is prepared according to the procedure previously described (synthesis of 14a) from lactone 1a (0.097 g, 0.18 mmol, 1 eq.), compound 38 (0.295 g, 0.72 mmol, 4 eq.), and *n*-buthyllithium 1.5 M (0.66 mL, 0.44 mmol, 5.5 eq.) to give a yellow oil, with a yield of 66%.

5 39: $C_{48}H_{45}F_3O_6$ $M = 774.86 \text{ g.mol}^{-1}$
 $NMR ^{19}F (CDCl_3; 282.5MHz)$: -102.9 (d, J = 254 Hz, 1F); -101.6 (d, J = 254 Hz, 1F); -118.1 (m, 1F).
 $Mass (ESI^+)$: 792.33 (M+H₂O).

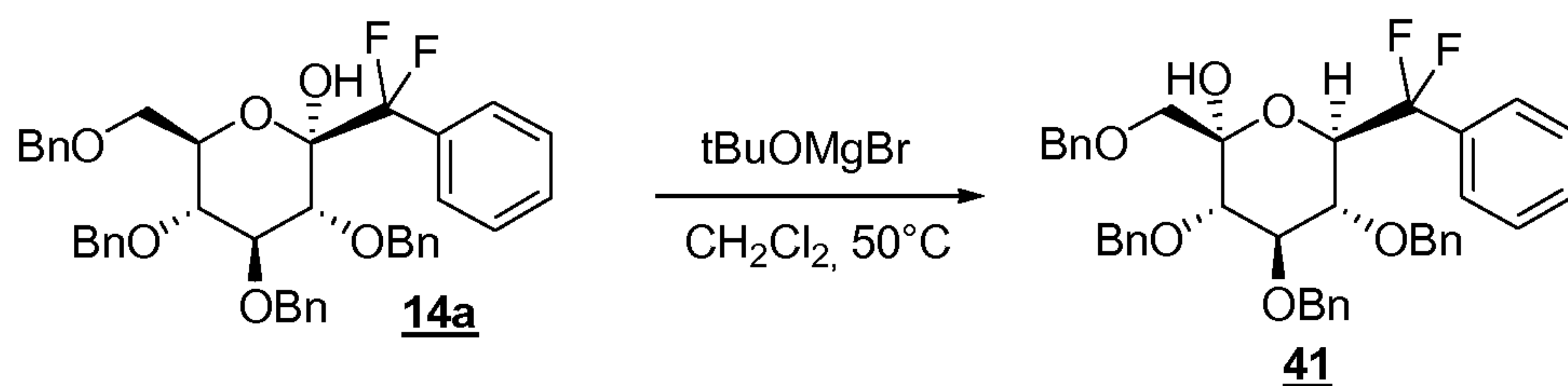
10 Synthesis of compound 40

A 3M solution of ethylmagnesium bromide in diethyl ether (0.133 mL, 0.40 mmol, 5 eq.) is slowly added to a solution of *tert*-butanol (0.038 mL, 0.40 mmol; 5 eq.) in diethyl ether (1 mL). The mixture is stirred for 15 min at ambient temperature. A solution of 18 (0.056 g, 0.079 mmol, 1 eq) in dichloromethane (0.5 mL) is then slowly added. The mixture is warmed to 50°C and stirred at this temperature for 3 days. A 1N aqueous solution of hydrochloric acid is then added at ambient temperature. The reaction medium is extracted with dichloromethane and then dried over magnesium sulphate, prior to being concentrated to produce compound 40 (no further purification) in the form of a colourless oil.

40: $C_{41}H_{39}ClF_2O_6$ $M = 700.24 \text{ g.mol}^{-1}$
NMR ^{19}F (CD_3OD), 282.5MHz : -98.4 (dd, J1 = 277 Hz, J2 = 7 Hz, 1F) ; -108.7
(dd, J1 = 272 Hz, J2 = 22Hz, 1F).
Mass (ESI $^{+}$) : 718.20 (M+H $_2$ O); 1417.73 (2M+H $_2$ O).

25

Synthesis of compound 41



Compound 41 is prepared according to the procedure previously described (synthesis of compound 40) from compound 14a (0.200 g, 0.3 mmol, 1 eq.), a 3M solution of ethylmagnesium bromide in diethyl ether (0.50 mL, 1.5 mmol, 5 eq.) and *tert*-butanol (0.142 mL, 1.5 mmol, 5 eq.). The residue is purified on a chromatography column (98/2 to 85/15 cyclohexane/ethyl acetate eluent) to produce compound 41 in the form of a colourless oil with a yield of 34%

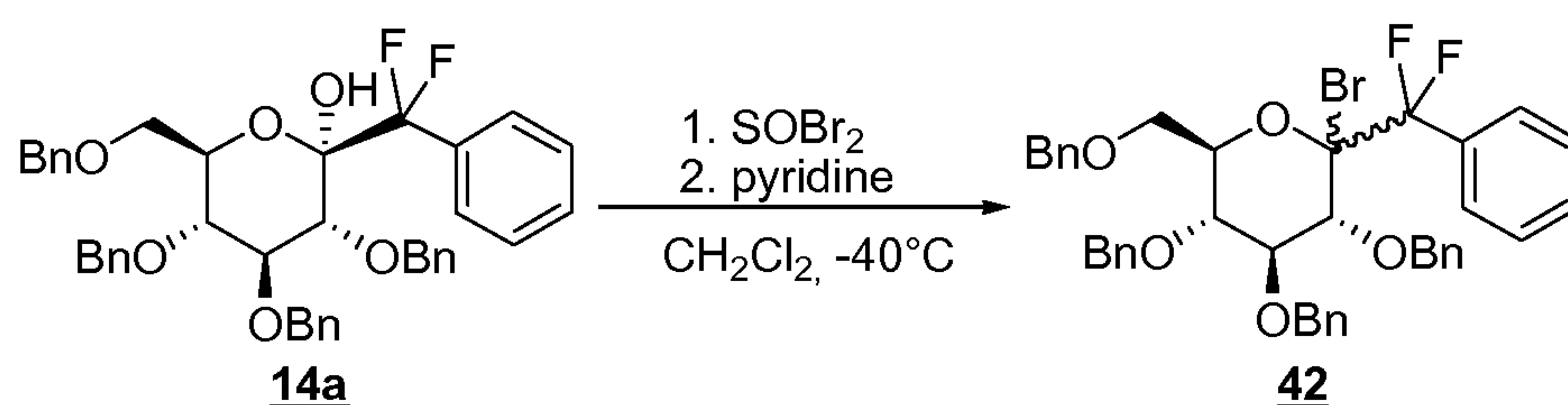
41: C₄₁H₄₀F₂O₆ M = 666.75 g.mol⁻¹

Rf: 0.28 (cyclohexane/ethyl acetate 8/2).

NMR ^{19}F (CDCl_3 , 282.5MHz) : -96.4 (dd, J1 = 255 Hz, J2 = 5Hz, 1F); -110.0 (dd, J1 = 254 Hz, J2 = 17 Hz, 1F).

Mass (ESI+): 684.13 (M+H₂O).

Synthesis of compound 42



Thionyl bromide (0.018 mL, 0.23 mmol, 1.5 eq.) is added to a round-bottom flask under inert atmosphere which contains a solution of 14a (0.101 g, 0.15 mmol, 1 eq.) in dichloromethane at -40°C. The mixture is stirred for 2h at this temperature before pyridine (0.018 g, 0.23 mmol, 1.5 eq.) is added. The solution is stirred for an additional period of 30min at this temperature. The solution is then brought back to ambient temperature and a 1N aqueous solution of hydrochloric acid is added. The reaction medium is extracted with dichloromethane and then dried over magnesium sulphate, prior to being concentrated. The residue is then purified on a preparative thin layer

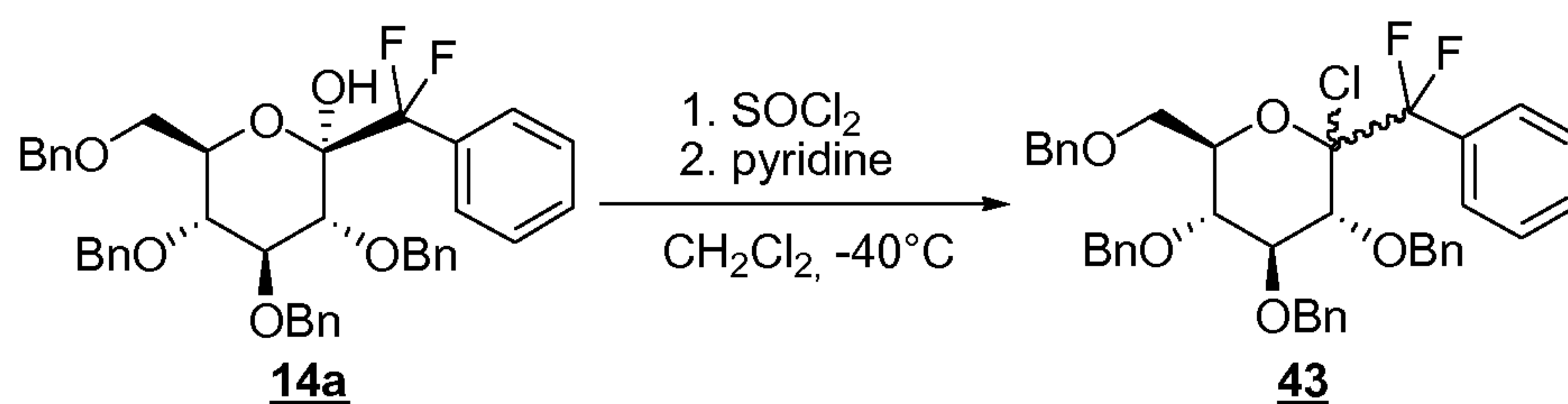
chromatography (85/15 cyclohexane/ethyl acetate eluent) to produce compound 42, in the form of white crystals, with a yield of 13%.

41: C₄₁H₃₉BrF₂O₅ M = 729.65 g.mol⁻¹

NMR ¹⁹F (CDCl₃), 282.5MHz : -100.3 (d, J = 247 Hz, 1F) ; -101.2 (d, J = 248 Hz, 1F).

Mass (ESI⁺): 746.07-747.93 (M+H₂O); 769.00 (M+K).

Synthesis of compound 43

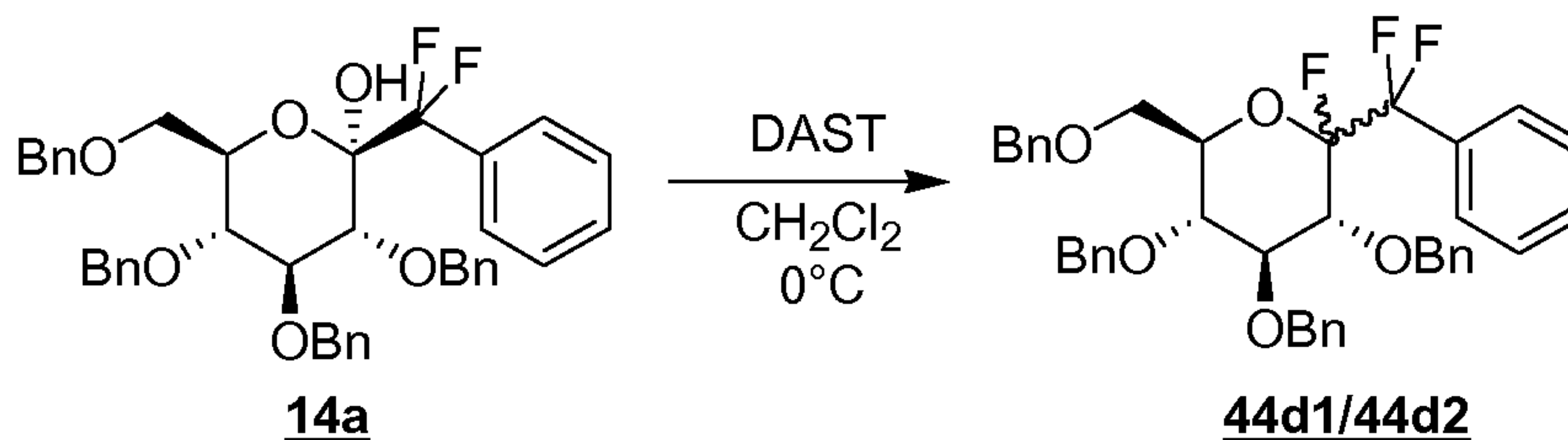


Thionyl chloride (0.037 mL, 0.51 mmol, 1.5 eq.) is added dropwise to a round-bottom flask under inert atmosphere which contains a solution of 14a (0.226 g, 0.34 mmol, 1 eq.) in dichloromethane (3.3 mL) at -30°C. The mixture is stirred for 30 min at this temperature before pyridine (0.041 mL, 0.51 mmol, 1.5 eq) is added. The solution is stirred for an additional period of 30 min at this temperature. The solution is then brought back to ambient temperature and a 2N aqueous solution of hydrochloric acid is added. The reaction medium is extracted with dichloromethane, washed with brine and then dried over magnesium sulphate, prior to being concentrated. The residue is then purified on a chromatography column (98/2 to 90/10 cyclohexane/ethyl acetate eluent) to produce compound 43 as a mixture of 2 anomers (60/40), in the form of an orange oil, with a yield of 77%.

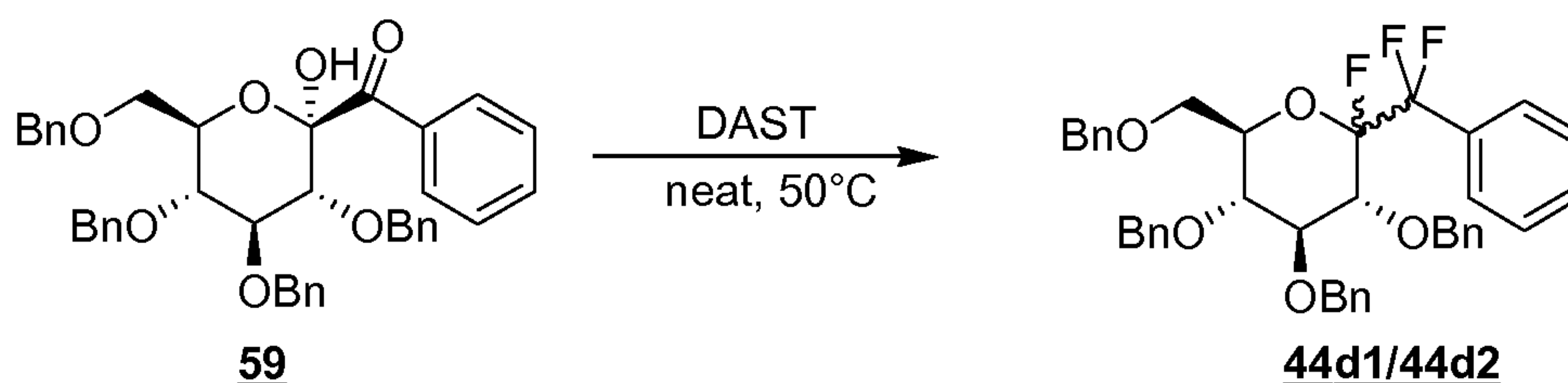
43: C₄₁H₃₉ClF₂O₅ M = 685.2g.mol⁻¹

NMR ¹⁹F (CDCl₃), 282.5MHz : -98.2 (d, J = 250 Hz, 1F); -101.5 (d, J = 250 Hz, 1F); -102.5 (d, J = 248 Hz, 1F); -104.4 (d, J = 249 Hz, 1F).

Mass (ESI⁺): 666.4 (M-HCl+H₂O); 671.47 (M-HCl+Na); 1314.13 ((2(M-HCl)+H₂O); 1318.80 (2(M-HCl)+Na).

Synthesis of compound 44d1/44d2*First process:*

Diethylaminosulfur trifluoride (DAST) (0.28 mL, 2.27 mmol, 2 eq.) is added into a round-bottom flask under an inert atmosphere which contains a solution of 14a (0.757 g, 1.14 mmol, 1 eq.) in dichloromethane (12 mL) at 0°C. The mixture is stirred for 1h at this temperature and overnight at room temperature. The reaction mixture is cooled to 0°C and methanol and solid sodium bicarbonate are carefully added at this temperature. Water is added and the reaction medium is extracted with dichloromethane, washed with water and brine and then dried over magnesium sulphate, prior to being concentrated. The residue is then purified on a preparative thin layer chromatography (80/20 cyclohexane/ethyl acetate eluent) to produce 44d1 and 44d2 as a mixture of two diastereomers in 40/60 proportion, in the form of a colourless oil, with a yield of 54%.

*Second process:*Synthesis of compound 44d1/44d2

A solution of compound 59 (see below for the preparation of compound 59) (0.032g, 0.049mmol, 1eq.) in diethylaminosulfur trifluoride (0.061mL, 0.49mmol, 10eq.) neat is stirred overnight at 50°C in a round-bottom flask under an inert atmosphere. Solid sodium bicarbonate and water are then carefully added at 0°C. The reaction medium is extracted with dichloromethane, washed with brine then dried over magnesium sulphate prior to being concentrated. The residue is then purified on a preparative thin layer chromatography (80/15 cyclohexane/ethyl acetate eluent) in order to produce compound 63 in the form of a colourless oil which slowly crystallizes, with a yield of 30%.

44d1/44d2: $C_{41}H_{40}F_3O_5$ $M = 668.74 \text{ g.mol}^{-1}$

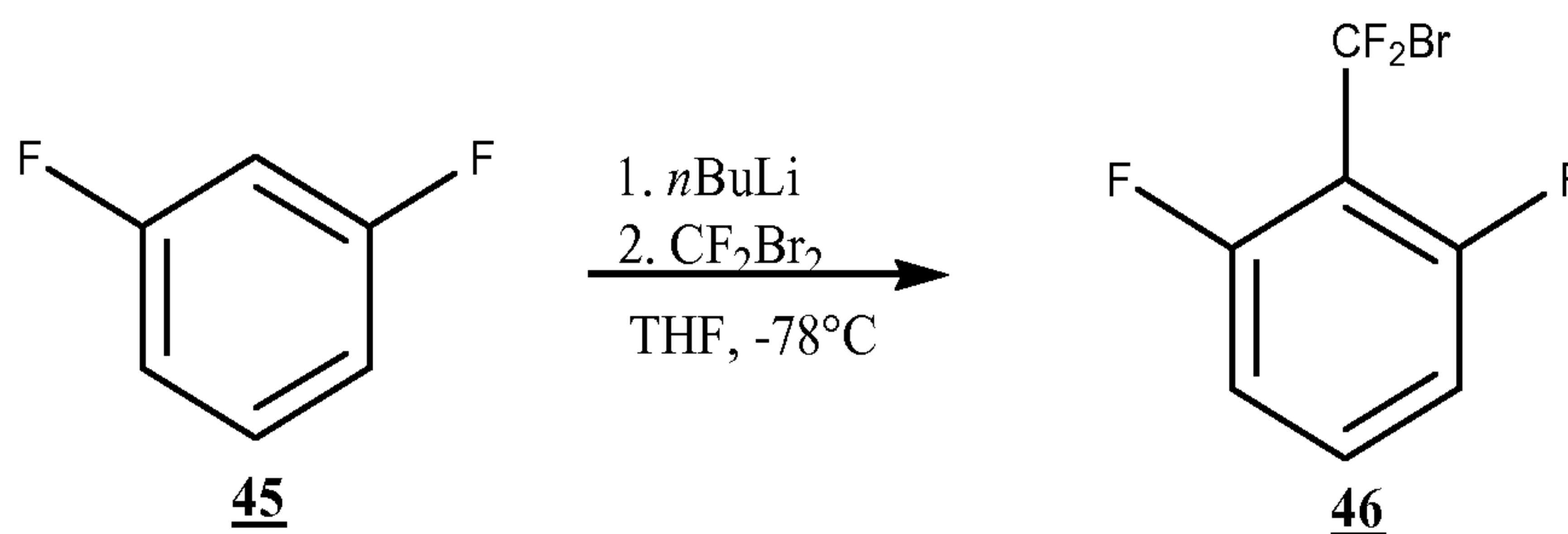
NMR ^{19}F ($CDCl_3$, 282.5 MHz) : 44d1: -106.2 (m, 1F); -107.2 (m, 2F).

44d2: -108.4 (m, 2F); -140.5 (m, 1F).

Mass (ESI^+) : 686.20 (M+H₂O).

5

Synthesis of compound 46



A solution of *n*-butyllithium 1.4 M in hexane (0.310 mL; 0.44 mmol; 1 eq.) is added dropwise into a round-bottom flask under an inert atmosphere, which contains the difluorobenzene 29 (0.05 g; 0.44 mmol; 1 eq.) in tetrahydrofurane (5 mL) at -78°C. After stirring for 1 h at this temperature, the dibromodifluoromethane (0.080 mL; 0.88 mmol; 2 eq.) is added. The reaction mixture is stirred for 1 additional h at -78°C and then a saturated aqueous ammonium chloride solution is added at ambient temperature. The mixture is extracted with ethyl acetate, dried over magnesium sulphate, filtered and concentrated in order to produce compound 30.

15

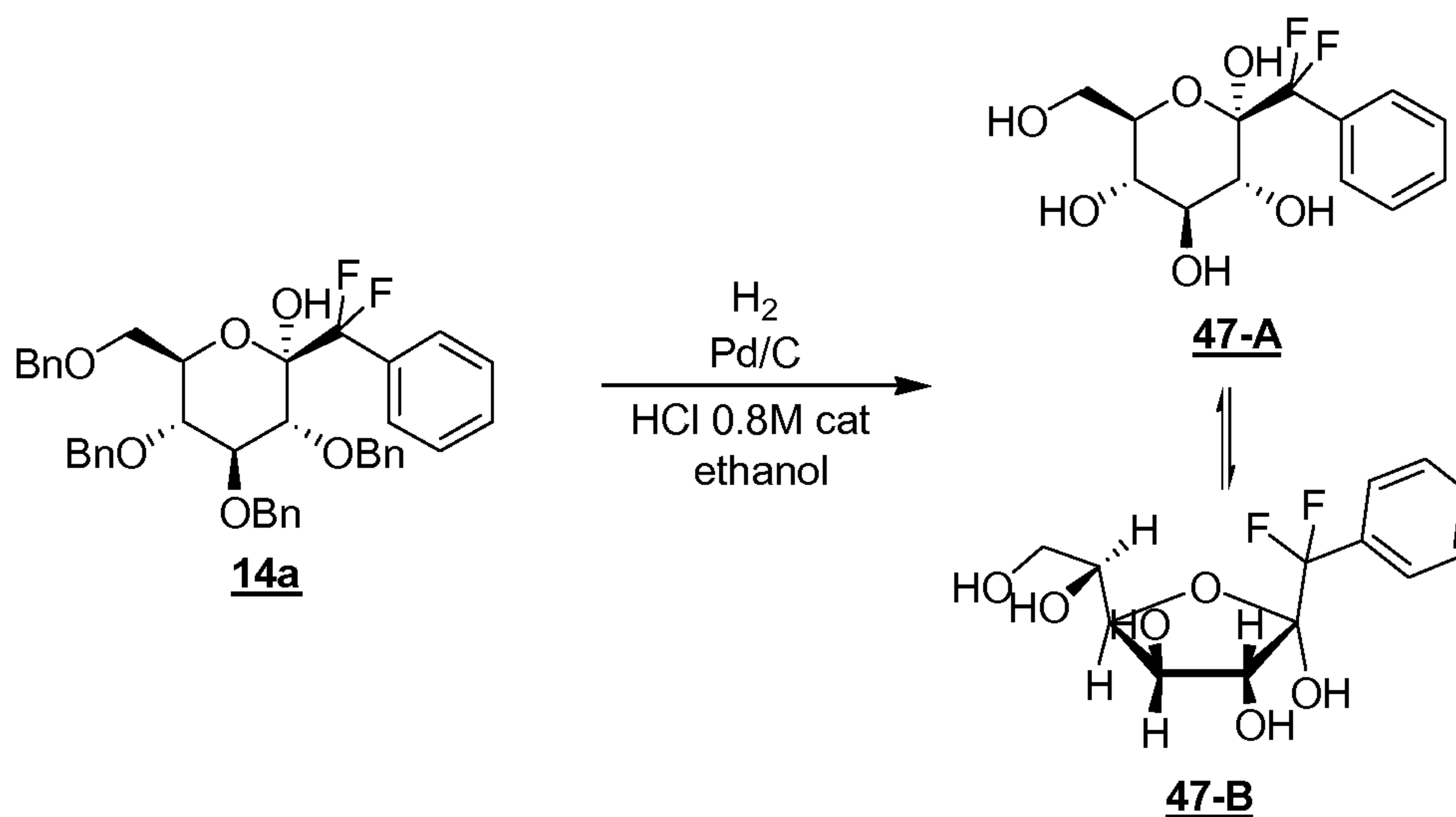
30: $C_7H_3BrF_4$ $M = 243.00 \text{ g.mol}^{-1}$

NMR ^{19}F ($CDCl_3$, 282.5 Mz): -40.70 (t, J = 30 Hz, 2F).

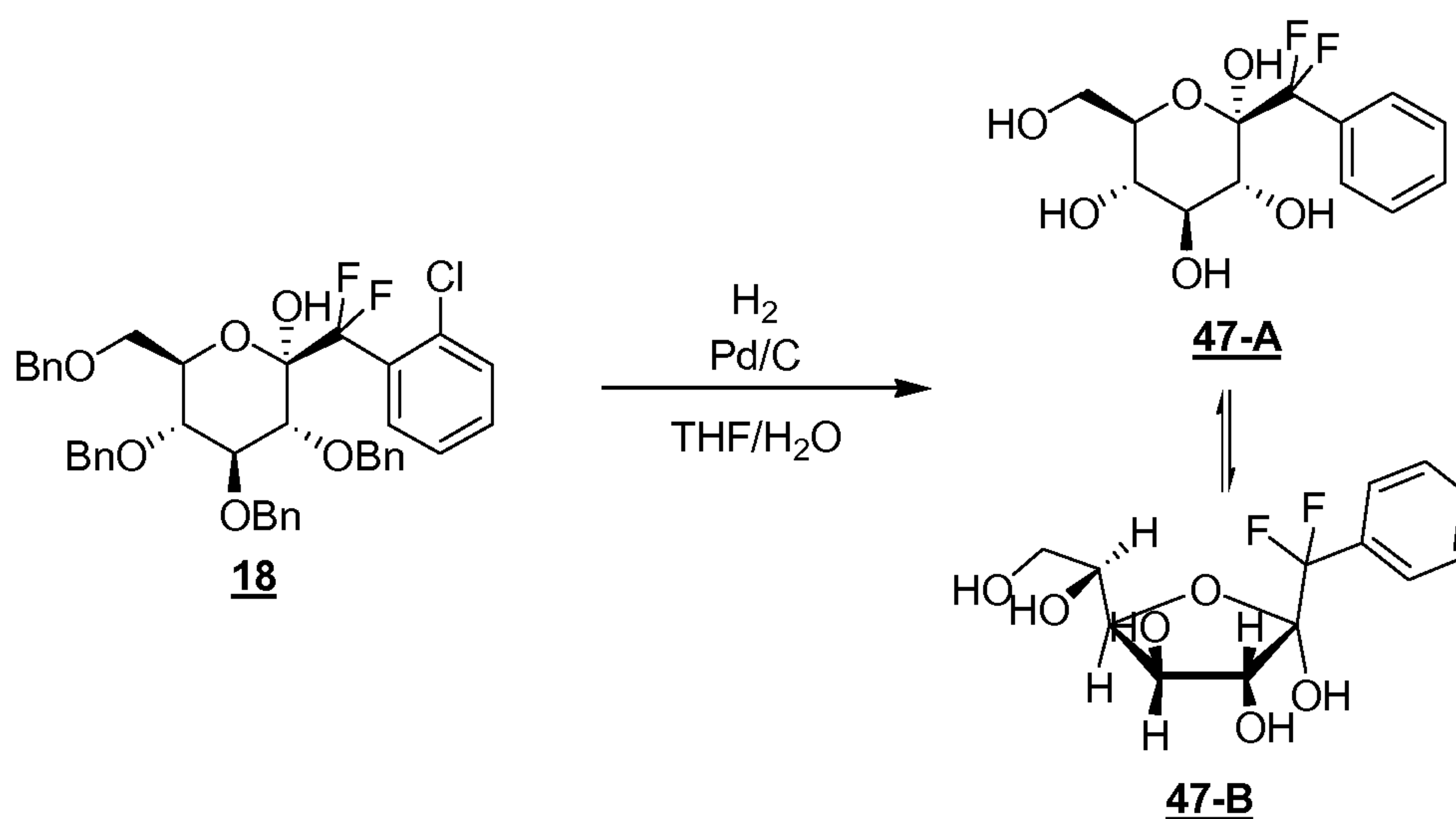
Mass: (ESI^+) : 163 (M-Br)

20

25

Synthesis of compound 47-A and 47-BFirst process:

Compound 14a (0.319 g, 0.48 mmol, 1 eq.) is placed inside a round-bottom flask and dissolved in a mixture of ethanol (4 mL) and 0.8M aqueous hydrochloric acid solution (two drops), in the presence of a spatula tip of Pd/C under a hydrogen atmosphere. The mixture is stirred for 48 h, then Millipore-filtered and evaporated to produce compound 47-A/47-B in the form of a white powder with a quantitative yield.

10 Second process:

Compound 18 (41.3 mg; 0.059 mmol; 1 eq.) is deprotected according to the procedure previously described (synthesis of compound 9a) to produce compound 47-A and 47-B in the form of a white powder with a quantitative yield.

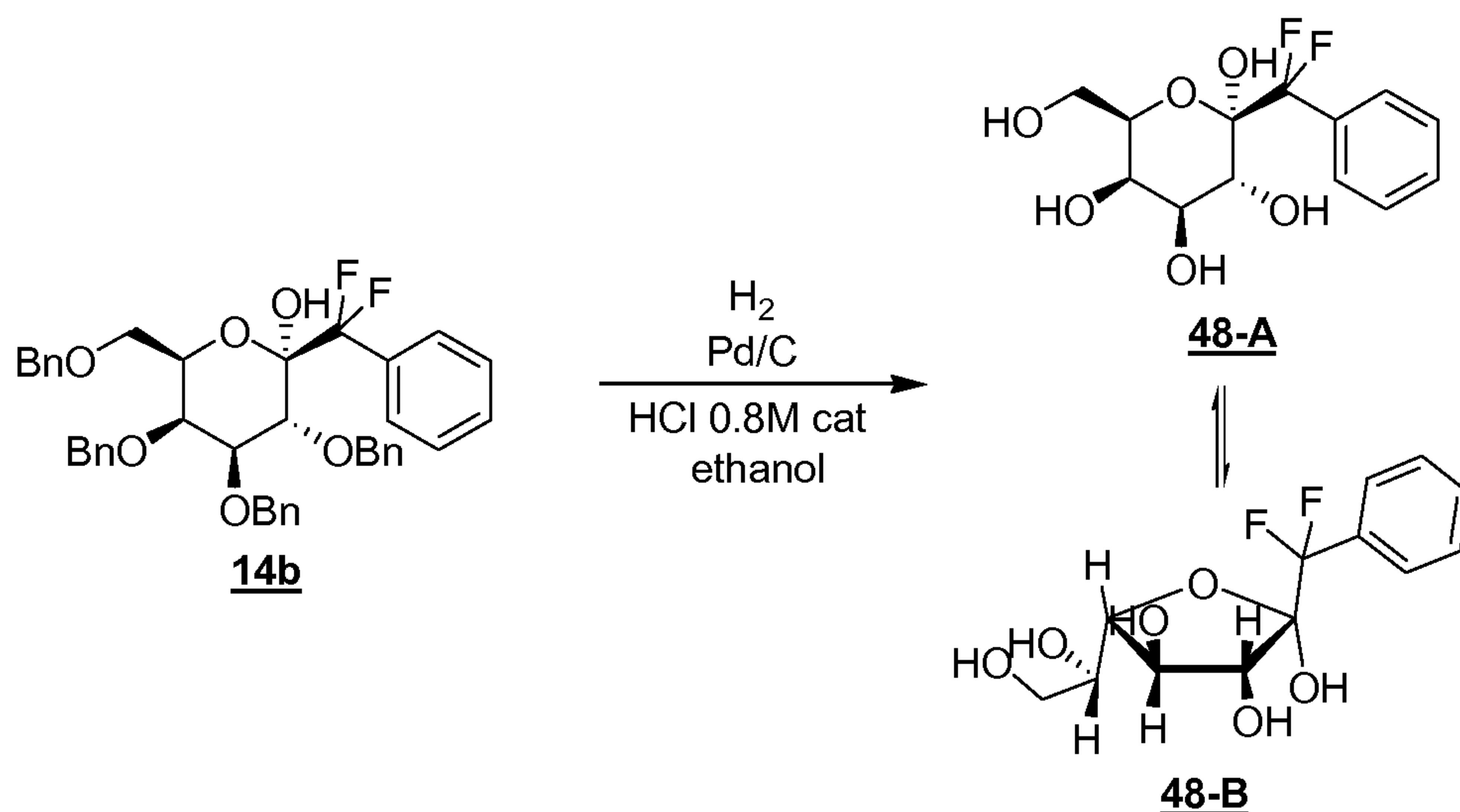
47-A/47-B: $C_{13}H_{16}F_2O_6$ $M = 306.26 \text{ g.mol}^{-1}$

NMR ^{19}F (D_2O , 282.5MHz) : 47-A: -109.7 (d, $J = 251 \text{ Hz}$, 1F); -107.2 (d, $J = 251 \text{ Hz}$, 1F); 47-B: -110.4 (d, $J = 250 \text{ MHz}$, 1F); -108.9 (d, $J = 253 \text{ Hz}$, 1F).

Mass: (ESI-): 305 (M-H) ; 341-343 (M+Cl).

5

Synthesis of compound 48-A and 48-B



Compound 14b (167 mg, 0.25 mmol, 1eq.) is deprotected according to the procedure described previously (synthesis of 47A-47-B, first process) to produce compound 48-A and 48-B in the form of a white powder with a quantitative yield.

10

48-A/48-B: $C_{13}H_{16}F_2O_6$ $M=306.26\text{g.mol}^{-1}$

^{19}F NMR (D_2O , 282,5MHz):

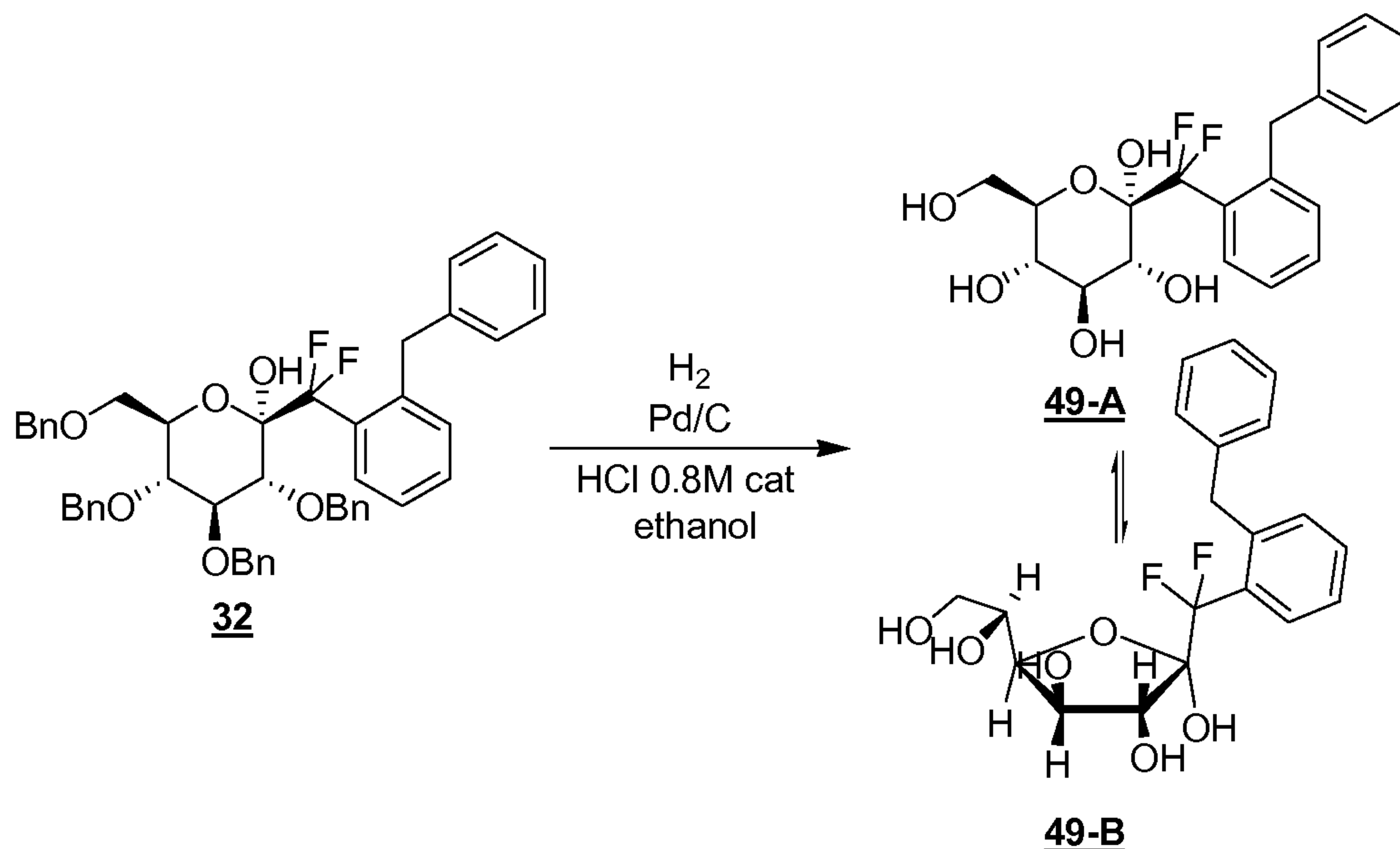
48-B: -112.4 (d, $J = 253 \text{ Hz}$, 1F); -114.2 (d, $J = 253 \text{ Hz}$, 1F).

48-A: -109.5 (d, $J = 252 \text{ Hz}$, 1F); -112.5 (d, $J = 250.6 \text{ Hz}$).

15

Mass (ESI) : 304.8 (M-H) ; 340.8 (M+Cl).

20

Synthesis of compound 49-A and 49-B

Compound 32 (39 mg, 0.05 mmol, 1 eq.) is deprotected according to the procedure described previously (synthesis of 47A-47-B, first process) to produce compound 49-A and 49-B in the form of a white powder with a quantitative yield.

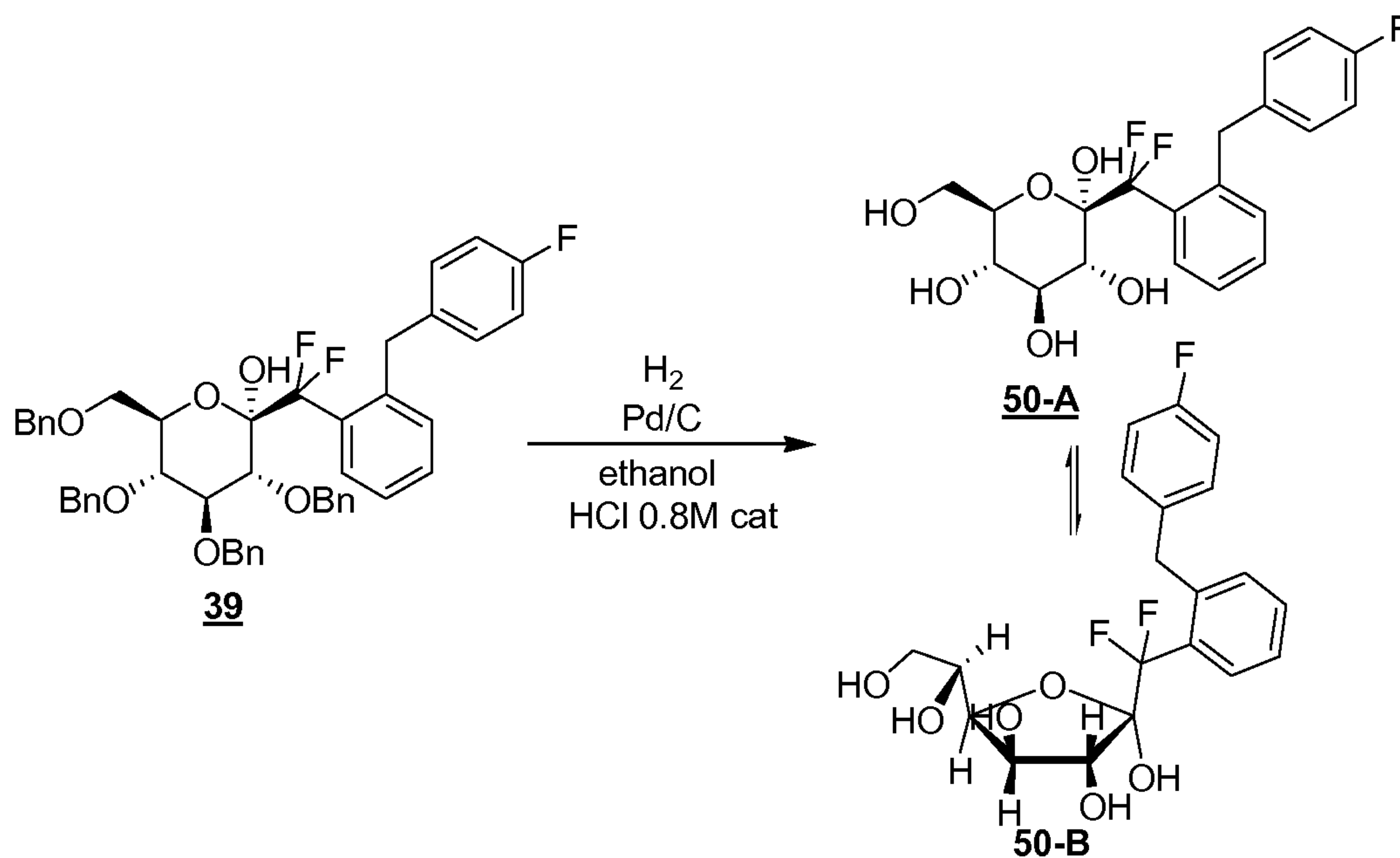
49-A/49-B: $\text{C}_{48}\text{H}_{22}\text{F}_2\text{O}_6$ $M = 396.38 \text{ g.mol}^{-1}$

NMR ^{19}F (CD_3OD , 282.5MHz) :

49-A: -100.7 (d, $J = 258 \text{ Hz}$, 1F); -104.9 (d, $J = 258 \text{ Hz}$, 1F)

49-B: -102.8 (d, $J = 258 \text{ Hz}$, 1F); -104.0 (d, $J = 259 \text{ Hz}$, 1F)

Mass (ESI) : 395.33 (M-H); 431.33 (M+Cl); 791.40 (2M-H).

Synthesis of compound 50-A and 50-B

Compound **39** (59 mg, 0.07 mmol, 1 eq.) is deprotected according to the procedure described previously (synthesis of **47A-47-B**, first process) to produce compound **50-A** and **50-B** in the form of a white powder with a 78% yield.

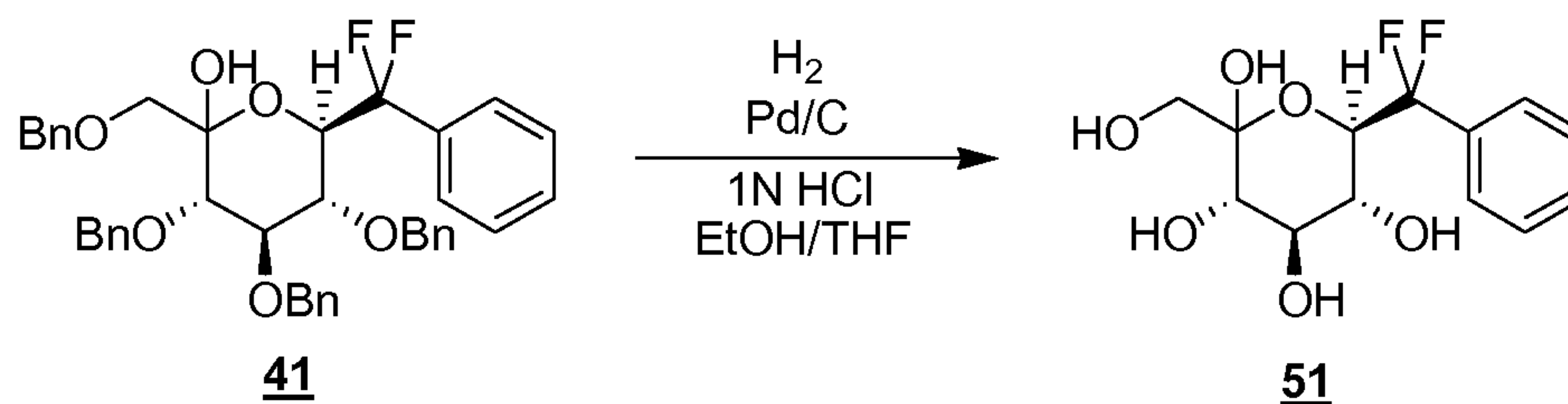
50-A/50-B: $C_{20}H_{21}F_3O_6$ $M = 414.37 \text{ g.mol}^{-1}$

$RMN^{19}F$ (CD_3OD , 282.5MHz):

Major form 50-A: -100.8 (d, $J = 258 \text{ Hz}$, 1F); -104.0 (d, $J = 259 \text{ Hz}$, 1F); -120.3 (dddd, 1F)

Minor form 50-B: -102.0 (d, $J = 258 \text{ Hz}$, 1F); -104.1 (d, $J = 259 \text{ Hz}$, 1F); -120.2 (dddd, 1F)

Mass (ESI) : 413.29 (M-H).

Synthesis of compound 51

Compound **41** (45.2 mg, 0.06 mmol, 1 eq.) is placed in a round-bottom flask and dissolved in a mixture of ethanol (1 mL), tetrahydrofuran (1 mL) and 1M hydrochloric acid solution (two drops) in the presence of a spatula tip of Pd/C , under a hydrogen

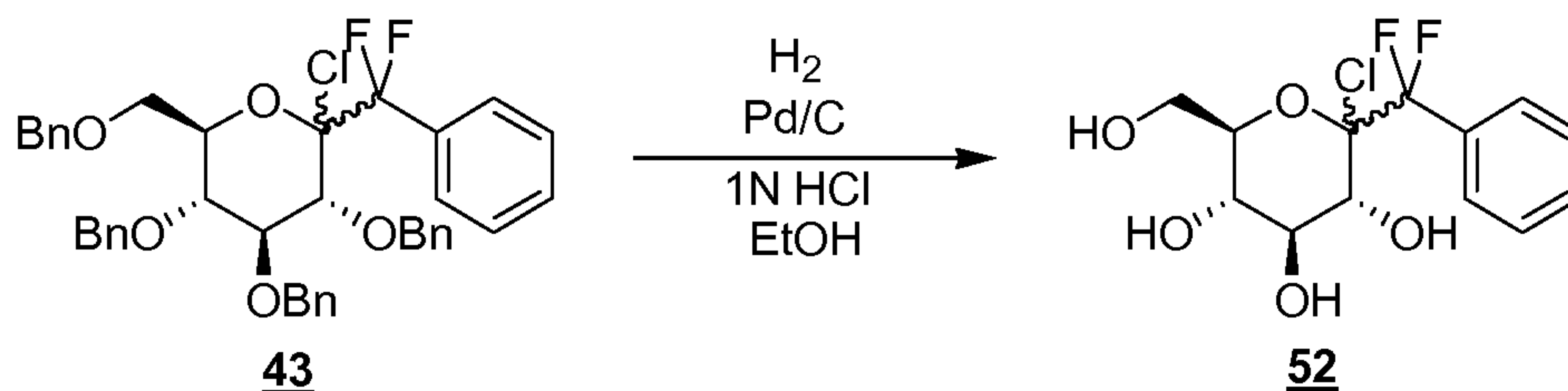
atmosphere. The mixture is stirred for 48 h, then Millipore-filtered and evaporated in order to produce compound 51, in the form of a white solid, with 96% yield.

51: C₁₄H₂₀F₂O₆ M = 322.30 g.mol⁻¹

NMR ¹⁹F (CDCl₃, 282.5MHz) : -98.7 (dd, J1 = 255 Hz, J2 = 7Hz, 1F); -107.7 (dd, J1 = 255 Hz; J2 = 13 Hz, 1F).

Mass (ESI-): 304.9 (M-H); 340.9 (M+Cl).

Synthesis of compound 52

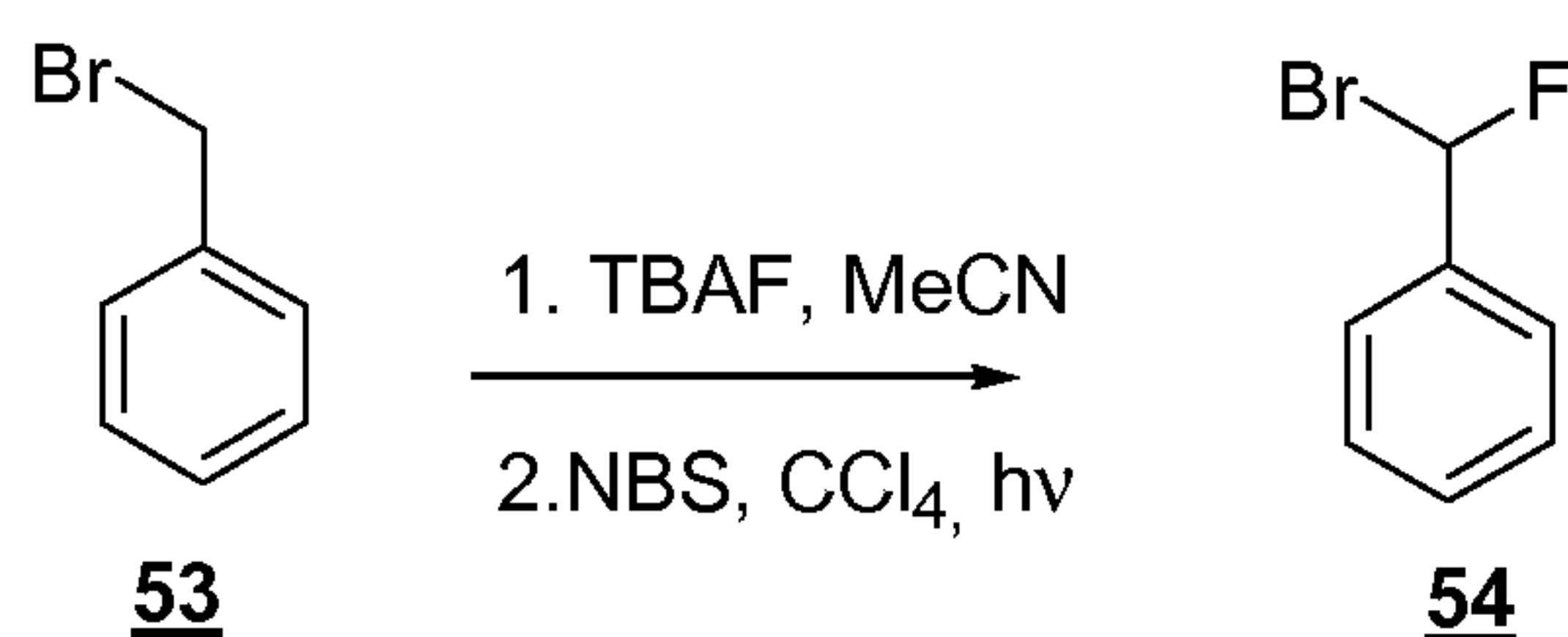


Compound 43 (44.3 mg, 0.07 mmol, 1 eq.) is deprotected according to the procedure described previously (synthesis of 47A-47-B, first process) to produce compound 52 in the form of a white solid, with 86% yield.

52: C₁₃H₁₅ClF₂O₅ M = 324.71 g.mol⁻¹

Mass (ESI-): 358.9 (M+Cl).

Synthesis of compound 54



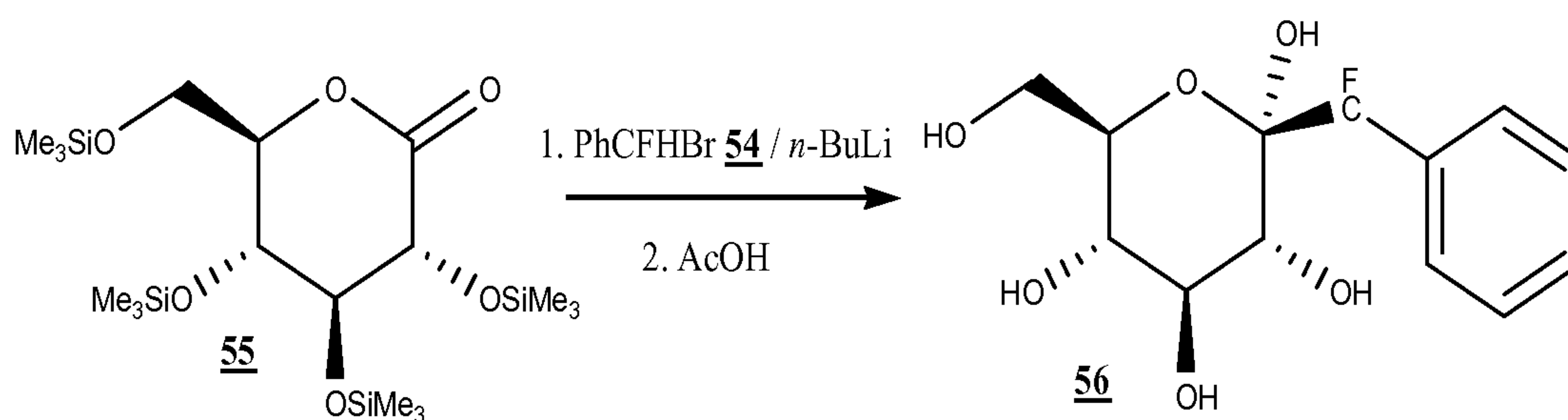
Benzyl bromide (2.4 mL, 20.0 mmol, 1 eq.) is added dropwise to a round-bottom flask under an inert atmosphere which contains a solution of tetra-n-butylammonium fluoride (12.62 g, 40.0 mmol, 2 eq.) in dry acetonitrile (40 mL) at ambient temperature. The reaction is stirred overnight at this temperature. Water is added (30 mL) and the reaction medium is extracted with pentane, and then dried over magnesium sulphate, prior to being concentrated to produce fluoro-methyl benzene with no further purification.

Fluoro-methyl benzene (1.37 g, 12.4 mmol, 1 eq.) is then added into a reactor under an inert atmosphere which contains a suspension of N-bromosuccinimide (2.21 g, 12.4 mmol, 1 eq.) in carbon tetrachloride (40 mL), surmounted by a mercury vapour UV lamp. The reaction mixture is irradiated overnight at ambient temperature. The mixture

is then filtered, extracted with dichloromethane, washed with water, dried over magnesium sulphate, filtered and then concentrated. The residue is then purified on a chromatography column (100% cyclohexane eluent) to produce compound 54 in the form of a colourless oil, with a 24% overall yield.

5 54: C₇H₆BrF M = 189,02 g.mol⁻¹
 NMR ¹⁹F (CDCl₃, 282.5MHz) : -130.1 (d, J = 49 Hz, 1F).
 Mass (CI+): 109 (M+H-Br)

Synthesis of compound 56

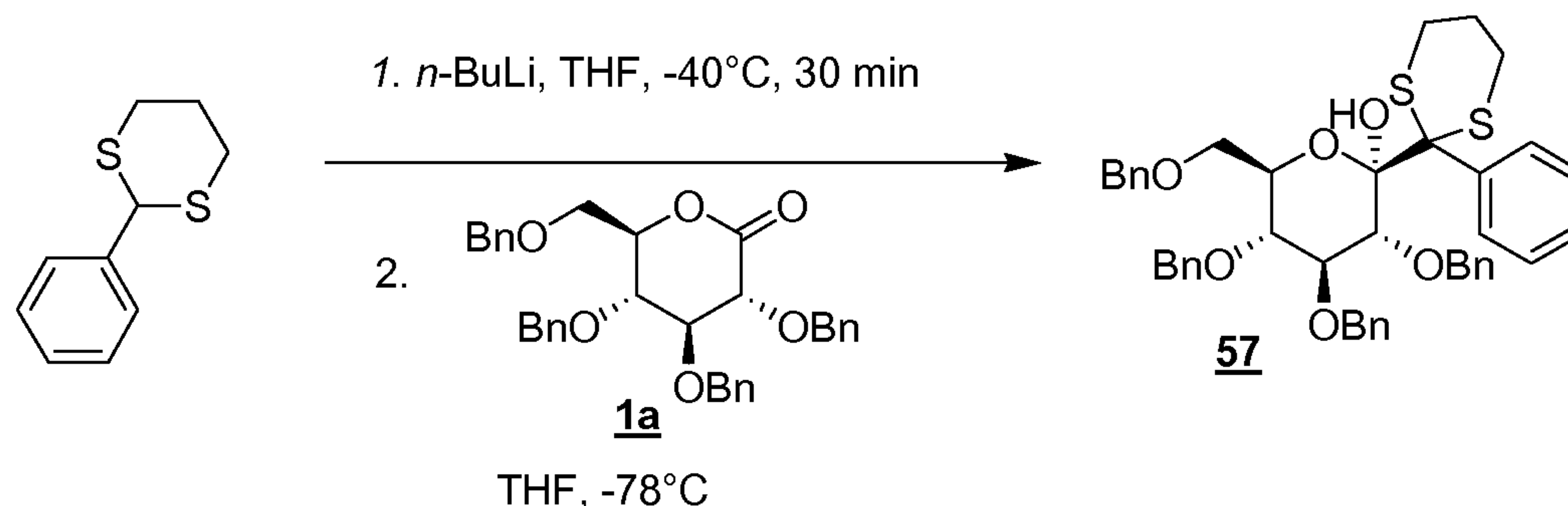


A 1.5 M solution of *n*-butyllithium in hexane (0.85 mL, 1.24 mmol, 5.5 eq.) is added to a round-bottom flask under an inert atmosphere, which contains a solution of compound 54 (0.170 mg, 0.40 mmol, 4 eq.) and lactone 55 (0.105 g, 0.22 mmol, 1 eq.) in dry tetrahydrofuran (3 mL) at -90°C. The mixture is stirred for 2 hours at this temperature.

15 A 1% aqueous acetic acid solution is added at this temperature and the mixture is brought back to ambient temperature. The reaction medium is extracted with diethyl ether, washed with brine and then dried over magnesium sulphate, prior to being concentrated. The residue is then diluted in methanol and a 1% aqueous solution of acetic acid (5 mL) is added. The mixture is stirred overnight at ambient temperature.

20 The solvent is removed and the reaction medium is extracted with ethyl acetate and then dried over magnesium sulphate, prior to being concentrated. The residue is then purified on a chromatography column (100/0 to 90/10 dichloromethane/methanol eluent) to produce compound 56 as a mixture of two diastereomers in 80/20 proportion, in the form of a colourless oil, with a yield of 4%.

25 **56:** C₁₃H₁₇FO₆ M = 288.27 g.mol⁻¹
 NMR ¹⁹F (MeOD, 282.5MHz) : -187.3 (d, J = 45 Hz, 1F);
 -200.0 (d, J = 45 Hz, 1F).
 Mass (ESI+): 306.1 (M+H₂O); 311.0 (M+Na); 327.1 (M+K).

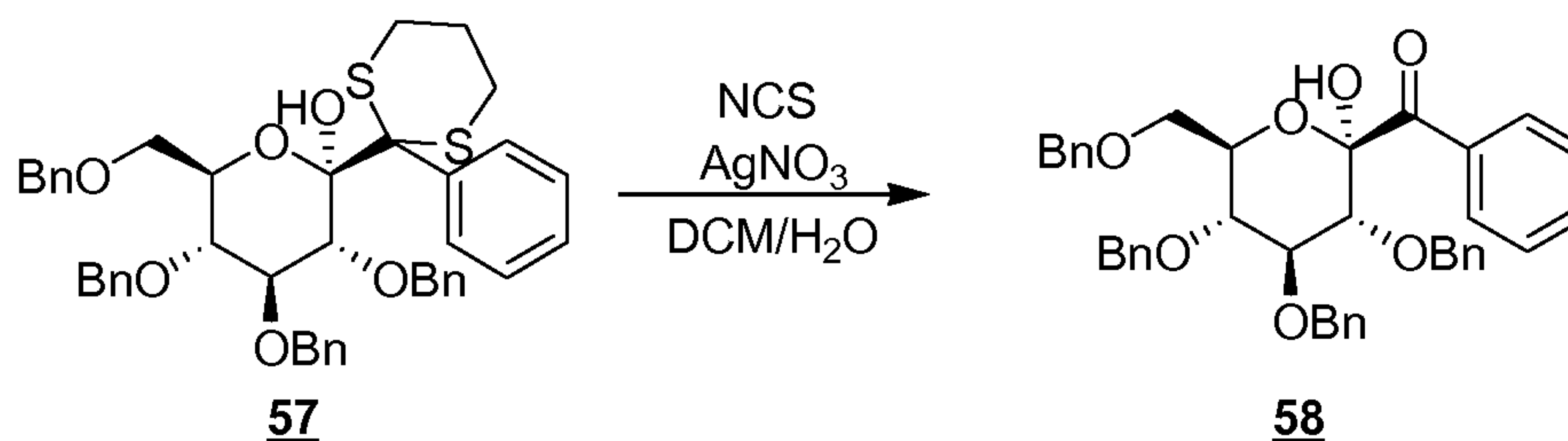
Synthesis of compound 57

A 1.6 M solution of *n*-butyllithium in hexane (45.6 mL, 73.0 mmol, 4.5 eq.) is added dropwise to a round-bottom flask under an inert atmosphere which contains a solution of 2-phenyl-1,3-dithiane (14.04 g, 71 mmol, 4.4 eq.) in dry tetrahydrofuran at -40°C. The mixture is stirred at -40°C for 30min before being cooled to -78°C. A solution of lactone **1a** (8.75 g, 16 mmol, 1 eq.) in tetrahydrofuran (10 mL) cooled at -78°C is added dropwise to the reaction mixture. At the end of the addition, the cooling bath is removed and saturated aqueous ammonium chloride solution (2 mL) is added. The reaction medium is extracted with diethyl ether, washed with brine and then dried over magnesium sulphate, prior to being concentrated. The residue is then purified on chromatography column (95/5 cyclohexane/ethyl acetate eluent) to produce **57** in the form of a white solid with a yield of 57%. The product can be recrystallised from acetonitrile to give colourless crystals.

57: C₄₄H₄₆O₆S₂ M = 734.96 g.mol⁻¹

Rf: 0.45 (cyclohexane/ethyl acetate 8/2).

Mass (ESI⁺) : 752.20 (M+H₂O); 1487.07 (2M+H₂O); 1507.87 (2M+K).

Synthesis of compound 58

20

A solution of **57** (1.75 g, 2.38 mmol; 1 eq.) in dichloromethane (4 mL) is quickly added to a round-bottom flask which contains *N*-chlorosuccinimide (1.27 g, 9.51 mmol, 4 eq.) and silver nitrate (1.82 g, 10.7 mmol, 4.5 eq.) in a mixture of dichloromethane and

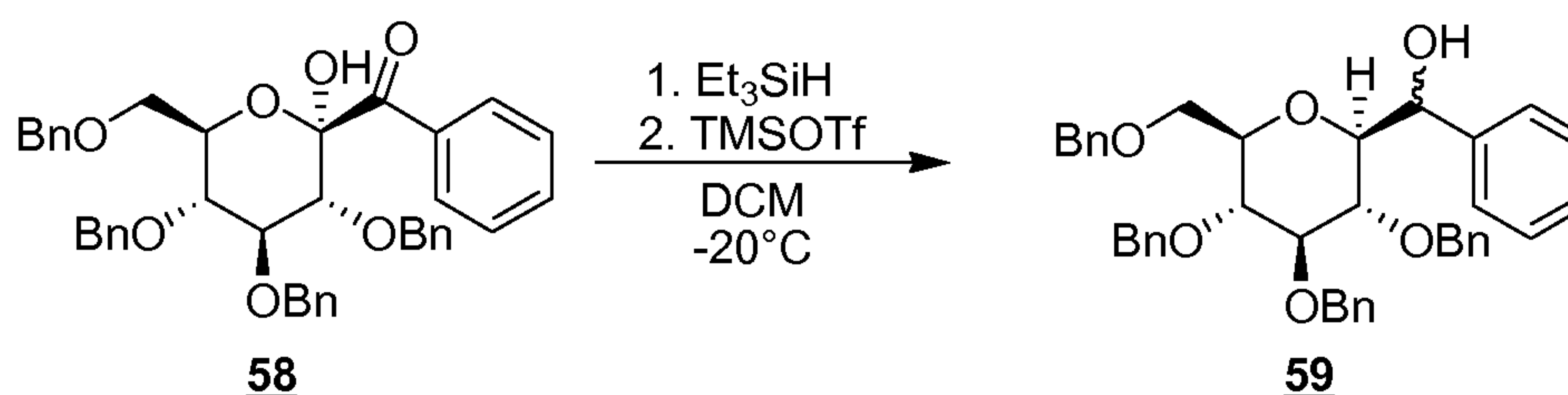
water in proportions of 8/2 (50 mL) at ambient temperature. The mixture is vigorously stirred for 15 min. The reaction medium is extracted with dichloromethane and then washed with a saturated aqueous solution of sodium sulfite (2 mL), sodium carbonate (2 mL) and brine (2 mL) then dried over magnesium sulphate, prior to being concentrated. The residue is then purified on chromatography column (90/10 cyclohexane/ethyl acetate eluent) to produce 58 in the form of a colourless oil with a yield of 79%.

58: C₄₁H₄₀O₇ M = 644.75 g.mol⁻¹

Rf: 0.42 (cyclohexane/ethyl acetate 8/2).

Mass (ESI⁺) : 662.33 (M+H₂O).

Synthesis of compound 59

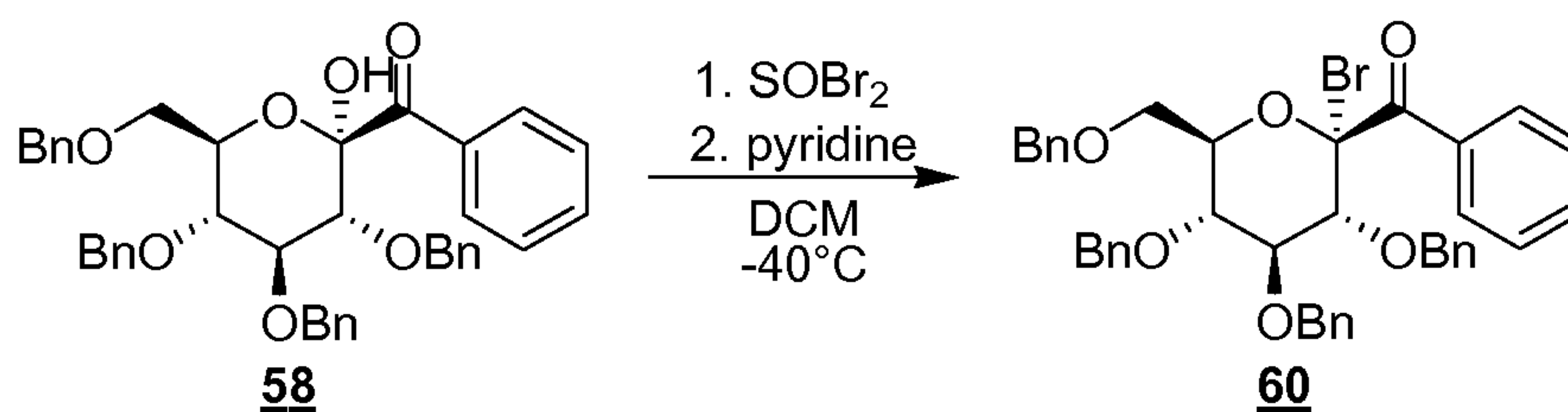


Triethylsilane (0.200 mL, 0.124 mmol, 8 eq.) and trimethylsilyl trifluoromethanesulfonate (0.028 mL, 0.15 mmol, 1 eq.) are successively added, to a round-bottom flask under an inert atmosphere which contains a solution of 58 (0.100 g, 0.15 mmol, 1 eq.) in dry dichloromethane (3mL) at -20°C. The mixture is stirred at this temperature for 7h. A saturated aqueous sodium carbonate solution is then added at ambient temperature and the reaction medium is extracted with dichloromethane, washed with brine then dried over magnesium sulphate prior to being concentrated. The residue is then purified on a chromatography column (10/0 to 8/2 cyclohexane/ethyl acetate eluent) in order to produce compound 59, in the form of a white solid, with a yield of 22%.

59: C₄₁H₄₀O₆ M = 628.76 g.mol⁻¹

Rf: 0.27(cyclohexane/ethyl acetate 8/2).

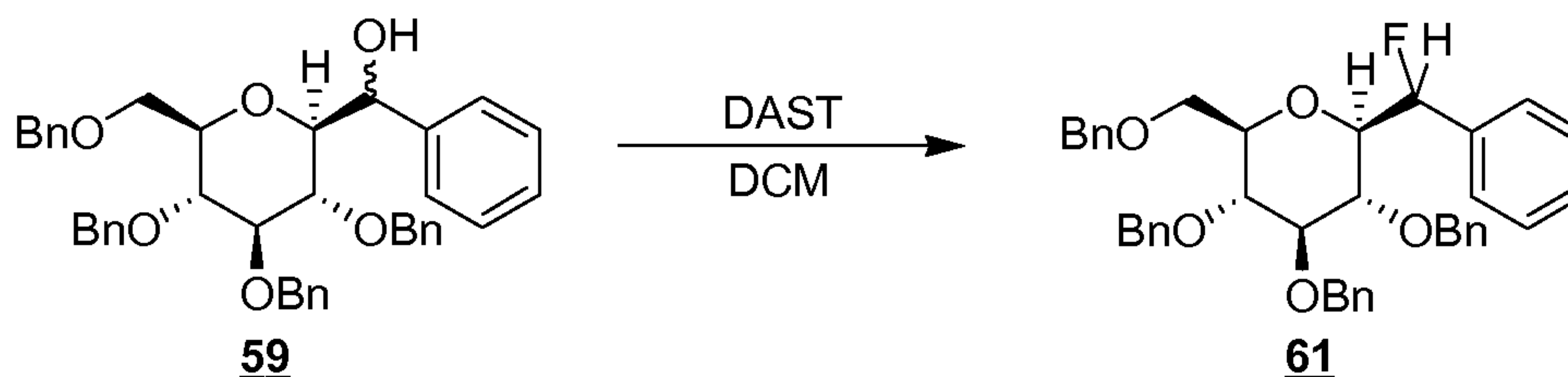
Mass (ESI⁺) : 629.27 (M+H); 646.20(M+H₂O); 1274.13 (2M+H₂O); 1278.93 (2M+Na).

Synthesis of compound 60

Compound 60 was prepared according to the procedure previously described (synthesis of compound 42) from compound 58 (0.100 g, 0.155 mmol, 1 eq.) , thionyl bromide (0.018mL, 0.132mmol, 1.5eq) and pyridine (0.019mL, 0.232 mmol, 1.5eq.) to give a colourless oil with a 51% yield.

60: C₄₁H₃₉BrO₆ M = 707.66 g.mol⁻¹

Mass (ESI⁺) : 729.27-731.27-732.27 (M+Na); 745.27-747.20-747.93-749.07 (M+K).

Synthesis of compound 61

Compound 61 was prepared according to the procedure previously described (synthesis of compound 44d1/44d2) from compound 59 (0.055g, 0.088mmol, 1eq.) and diethylaminosulfur trifluoride (0.018mL, 0.15mmol, 1.7eq.), as a mixture of two diastereomers in 58/42 proportion, in the form of colourless crystals.

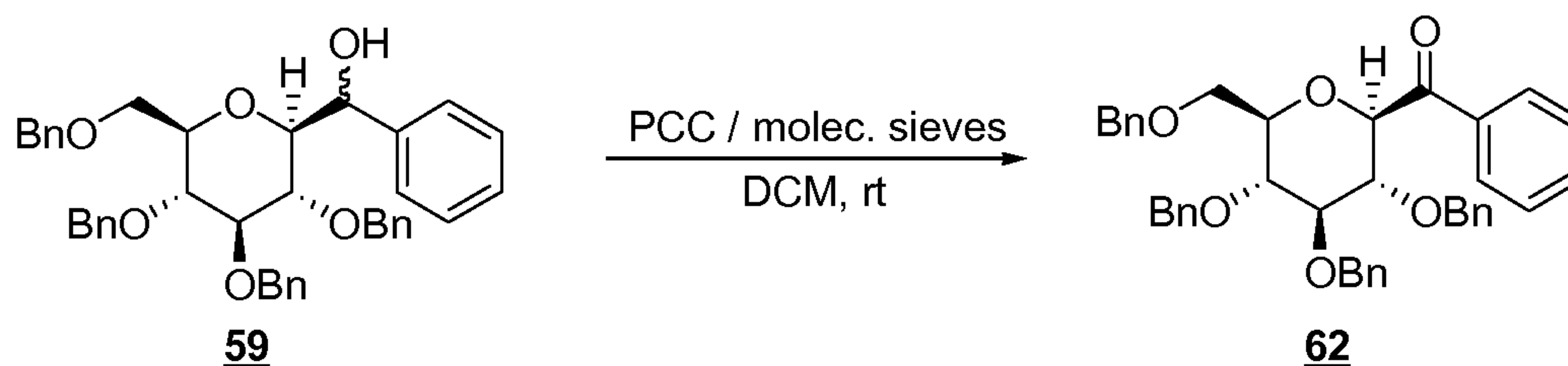
61: C₄₁H₄₀F₃O₅ M = 632.76 g.mol⁻¹

NMR ¹⁹F (CDCl₃, 282.5MHz) :

-183.4 (dd, J = 44.3 Hz, J2 = 14.4 Hz, 1F);

-197.2 (dd, J1 = 45.4 Hz, J2 = 27.8 Hz, 1F).

Mass (ESI⁺) : 650.20 (M+H₂O).

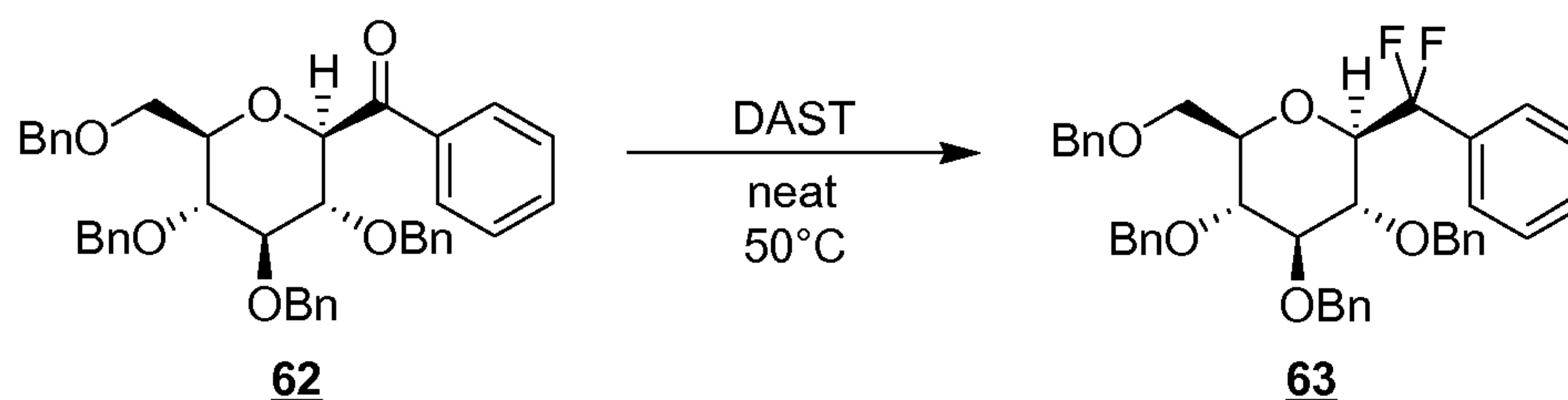
Synthesis of compound 62

Pyridinium chlorochromate (0.01 mg, 0.05 mmol, 1.7 eq) is added to a round-bottomed flask under inert atmosphere, which contains a solution of compound 59 (0.020 g, 0.03 mmol, 1 eq.) in dry dichloromethane (2mL) and molecular sieves. The mixture is stirred at ambient temperature overnight before another portion of PCC (1 eq.) is added. The mixture is stirred at ambient temperature for 5h and then filtered. Solvent is removed and the residue is purified on preparative thin layer chromatography (8/2 cyclohexane/ethyl acetate eluent) in order to produce compound 62, in the form of a white solid, with a yield of 58%.

62: C₄₁H₄₀O₆ M = 628.76 g.mol⁻¹

R_f: 0.39 (cyclohexane/ethyl acetate 8/2).

Mass (ESI⁺): 629.27 (M+H); 646.20 (M+H₂O); 1274.13 (2M+H₂O); 1278.9 (2M+Na).

Synthesis of compound 63

A solution of compound 62 (70.5 mg, 0.11 mmol, 1 eq.) in diethylaminosulfur trifluoride (0.300 mL) neat is stirred overnight at 50°C in a round-bottom flask under an inert atmosphere. Additional diethylaminosulfur trifluoride (0.100 mL) is then added at ambient temperature and the mixture is stirred at 50°C for an additional 24h. Solid sodium bicarbonate and water are then carefully added at 0°C. The reaction medium is extracted with dichloromethane, washed with brine then dried over magnesium sulphate prior to being concentrated. The residue is then purified on a chromatography column

(90/10 to 85/15 cyclohexane/ethyl acetate eluent) in order to produce compound 63 in the form of a colourless oil which slowly crystallizes, with a yield of 30%.

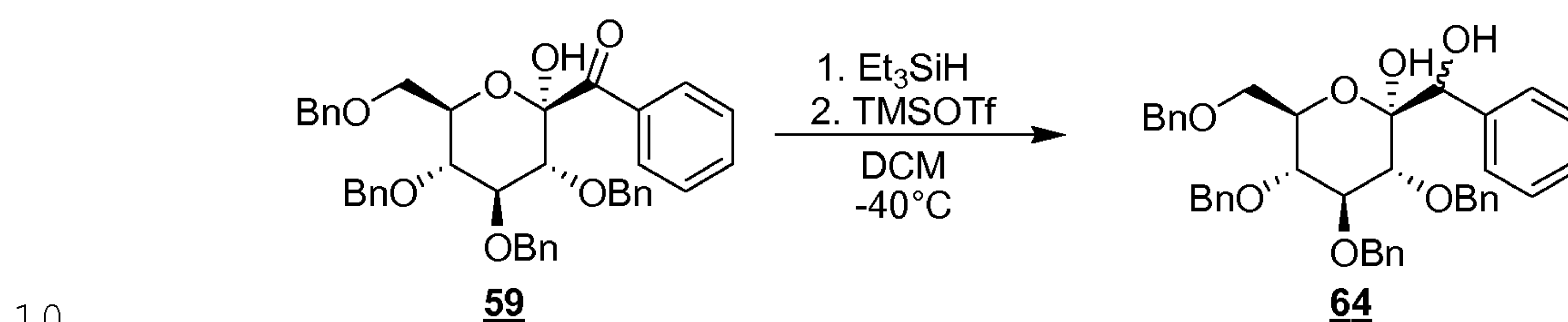
63: $C_{41}H_{40}F_2O_5$ $M = 650.75 \text{ g.mol}^{-1}$

Rf: 0.48 (cyclohexane/ethyl acetate 8/2).

5 NMR ^{19}F ($CDCl_3$, 282.5 MHz): -97.9 (dd, $J_1 = 4.12 \text{ Hz}$, $J_2 = 260.9 \text{ Hz}$, 1F); -109.4 (dd, $J_1 = 15 \text{ Hz}$, $J_2 = 257 \text{ Hz}$, 1F).

Mass (ESI^+): 668.20 ($M+H_2O$).

Synthesis of compound 64

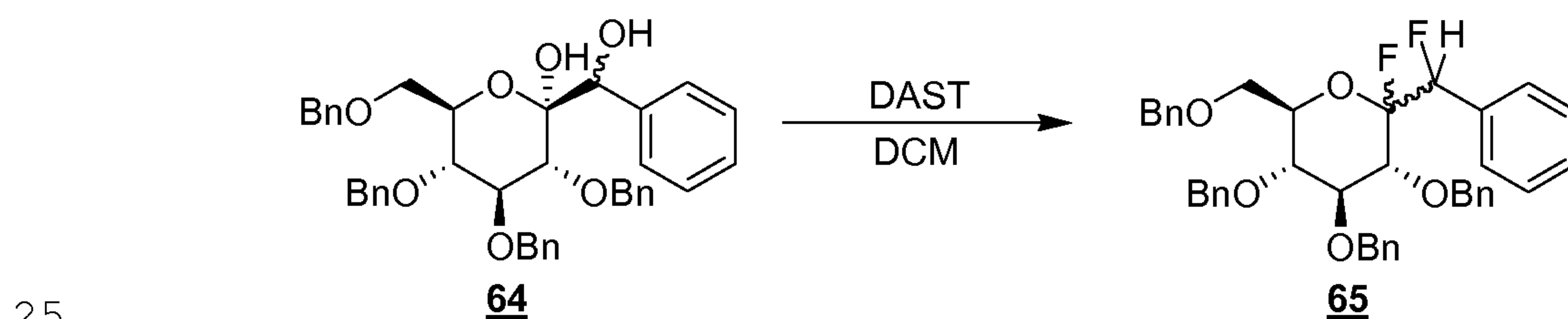


Triethylsilane (0.050mL, 0.31mmol, 4eq.) and trimethylsilyl trifluoromethanesulfonate (TMSOTf) (0.035mL, 0.19mmol, 2.5eq.) are successively added, to a round-bottom flask under an inert atmosphere which contains a solution of 59 (0.05g, 0.077mmol, 1eq.) in dry dichloromethane (1.5mL) at -40°C . The mixture is stirred at this temperature for 1h. A saturated aqueous sodium carbonate solution is then added at ambient temperature and the reaction medium is extracted with dichloromethane, washed with brine then dried over magnesium sulphate prior to being concentrated. The residue is then purified on a chromatography column (10/0 to 80/20 cyclohexane/ethyl acetate eluent) in order to produce compound 64, in the form of a white solid, with a yield of 45%.

64: $C_{41}H_{42}O_7$ $M = 646.78 \text{ g.mol}^{-1}$

Mass (ESI^+): 644.27 ($M+H_2O$); 1311.07 ($2M+H_2O$).

Synthesis of compound 65



Compound 65 was prepared according to the procedure previously described (synthesis of compound 44d1/44d2) from compound 64 (0.022 g, 0.035 mmol, 1eq.) and diethylaminosulfur trifluoride (0.017mL, 0.14mmol, 4eq.) as a mixture of 4 diastereomers in 33/33/25/5 proportion, in the form of a colourless oil, with a 41% yield.

5 65: C₄₁H₄₀F₂O₅ M = 650.75 g.mol⁻¹

NMR ¹⁹F (CDCl₃, 282.5MHz) :

-134.5 (ddd, J1 = 24 Hz, J2 = 18 Hz, J3 = 6.2 Hz, 1F); -192.0 (dd, J1 = 45 Hz, J2 = 17.5 Hz, 1F);

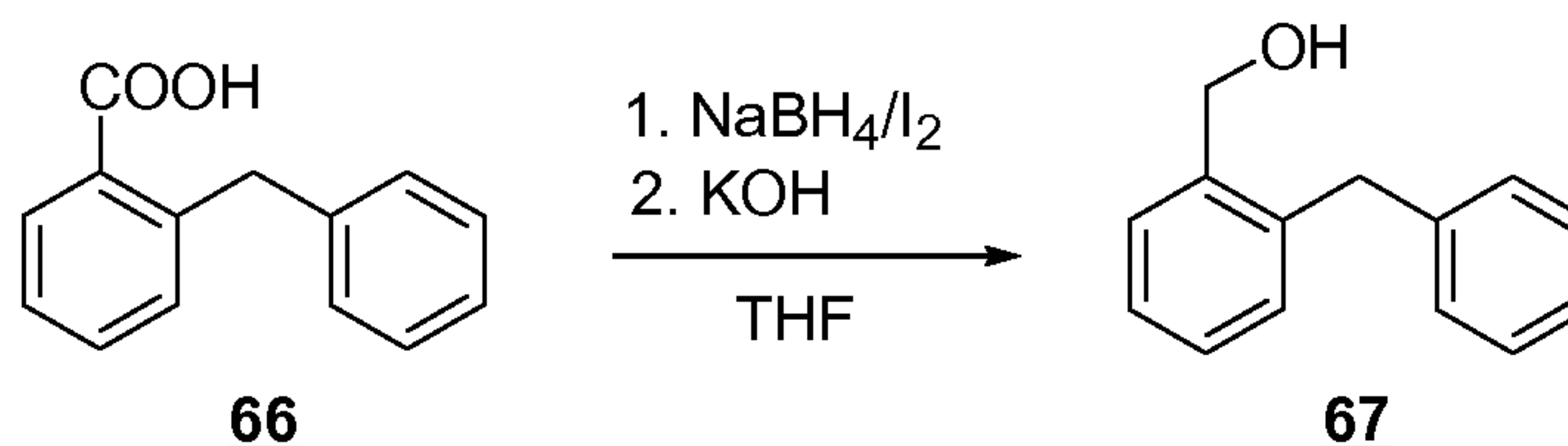
10 -135.9 (dd, J1 = 23 Hz, J2 = 5 Hz, 1F); -190.50 (dapp, J1 = 44 Hz, J2 = 5 Hz, 1F)

-115.3 (m, 1F); 189.3 (dd, J1 = 44 Hz, J2 = 17 Hz, 1F)

-111.9 (m, 1F); -189.8 (dd, J1 = 42 Hz, J2 = 9 Hz, 1F)

Mass (ESI⁺): 650.20 (M+H₂O).

15 Synthesis of compound 67



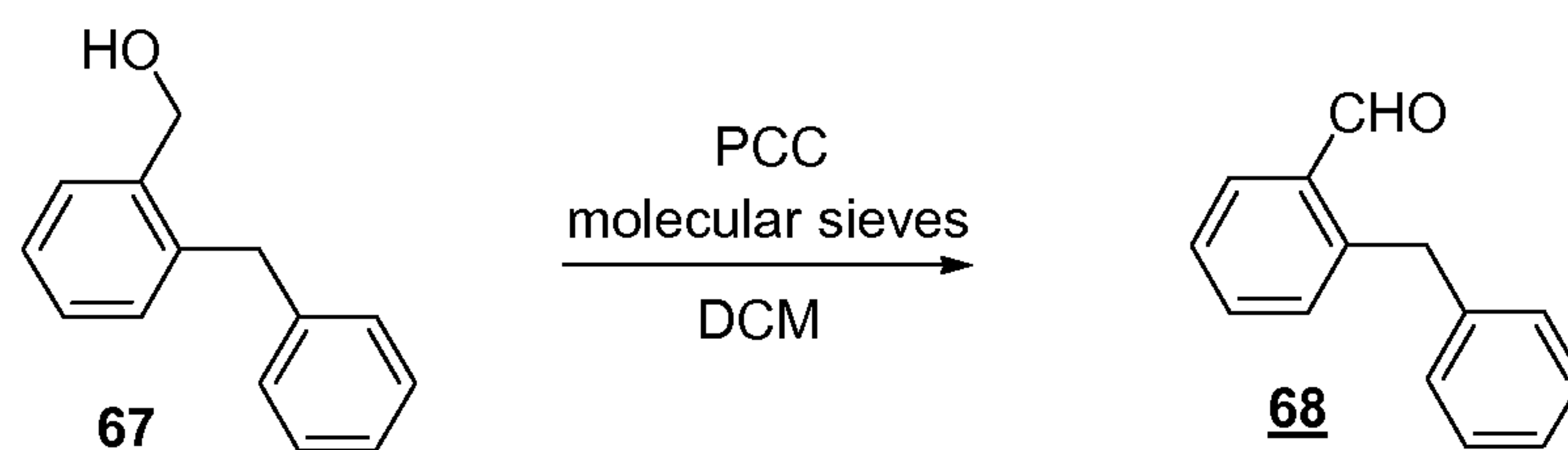
A solution of iodine (8.37 g, 33.0 mmol, 1 eq.) in dry tetrahydrofuran (60 mL) is added dropwise to a round-bottom flask under an inert atmosphere which contains a suspension of sodium borohydride (3.0 g, 79.0 mmol, 2.4 eq.) in dry tetrahydrofuran (60 mL) at 0°C. The mixture is stirred 5 min at this temperature and compound 66 is added. The mixture is refluxed overnight before being cooled to 0°C. Methanol (50 mL) is then added dropwise and the resulting mixture is stirred at ambient temperature for a further 30 min. Solvents are removed and a 20% potassium hydroxide aqueous solution (150 mL) is added to the residue. The solution is stirred for 4h at ambient temperature. The reaction medium is extracted with dichloromethane and dried over magnesium sulphate prior to being concentrated to produce compound 67, in the form of a yellow oil, with a yield of 92%. The compound can be involved in the next step without any further purification.

25 67: C₁₄H₁₄O M = 198.26 g.mol⁻¹

30 Rf: 0.23 (dichloromethane).

Mass (CI⁺): 181 (M-H₂O+H).

Synthesis of compound 68



- 5 Pyridinium chlorochromate (PCC) (4.56 g, 21 mmol, 1.4 eq.) is added to a round-bottom flask under inert atmosphere, which contains **67** (3.00 g, 15.0 mmol, 1 eq.) in dry dichloromethane (150 mL) and molecular sieves. The mixture is stirred overnight at ambient temperature and filtered through celite (dichloromethane eluent). Solvent is removed and the residue is purified on a chromatography column (90/10
10 cyclohexane/ethyl acetate eluent) in order to produce compound **68**, in the form of a white solid, with a yield of 67%.

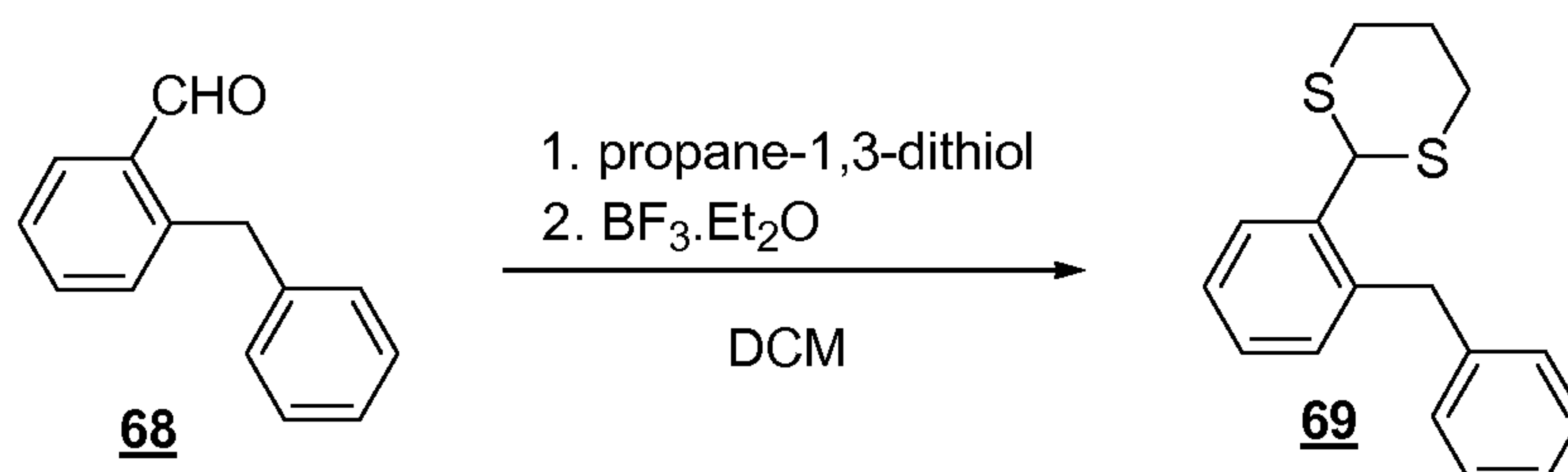
68: C₁₄H₁₂O M = 196.24 g.mol⁻¹

Rf: 0.87 (cyclohexane/ethyl acetate 7/3).

Mass (ESI⁺): 213.92 (M+H₂O).

15

Synthesis of compound 69



- Propane-1,3-dithiol (1.50 mL, 15.12 mmol, 1.5 eq.) is added in a round-bottom flask under an inert atmosphere which contains a solution of compound **68** (1.48 g, 10.1
20 mmol, 1 eq.) in dichloromethane (30 mL) at 0°C. Boron trifluoride etherate (0.25 mL, 1.98 mmol, 0.2 eq.) is added dropwise at this temperature. The mixture is stirred at 0°C for 15 min and overnight at room temperature. The reaction medium is extracted with dichloromethane, washed with a 5% sodium hydroxide aqueous solution, water and dried over magnesium sulphate prior to being concentrated. The residue is recrystallized

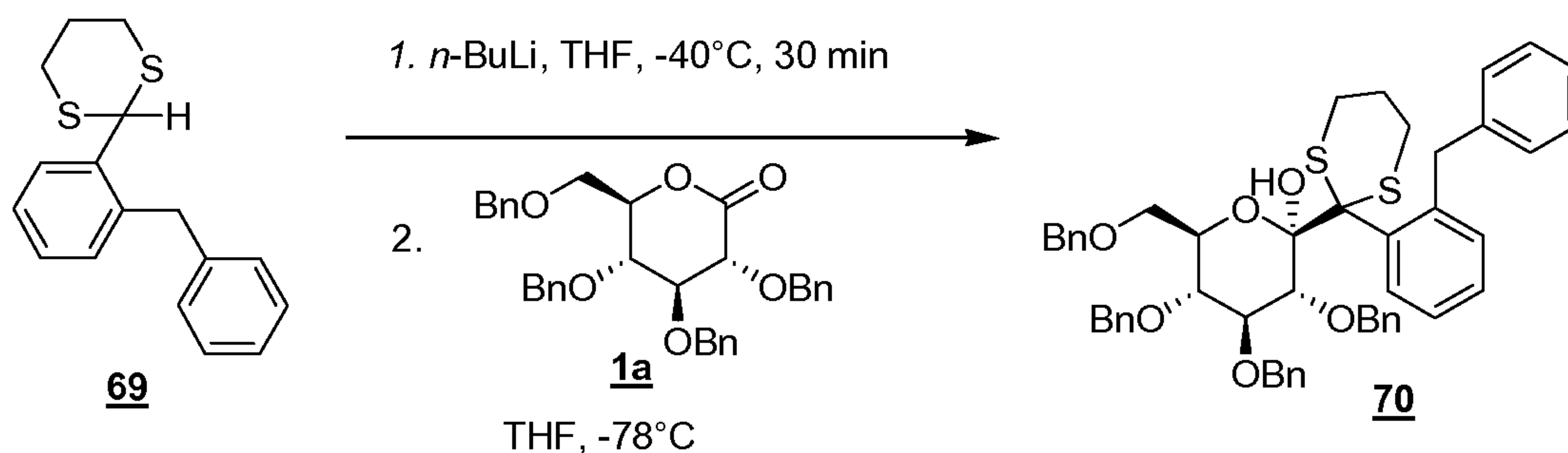
from acetonitrile to produce compound 69, in the form of a white solid, with a yield of 81%.

69: C₁₇H₁₈S₂ M = 286.45 g.mol⁻¹

Rf: 0.55 (cyclohexane/ethyl acetate 9/1).

5 Mass (ESI⁺): 287 (M+H).

Synthesis of compound 70



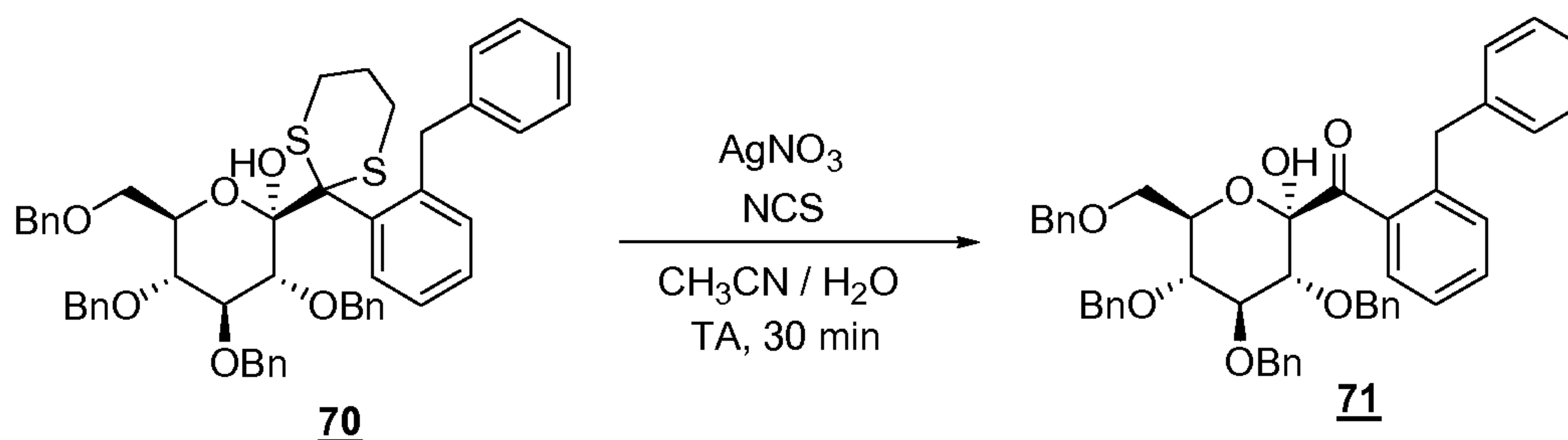
10 Compound 70 is prepared according to the procedure previously described (synthesis of compound 57) from compound 69 (2.23 g, 2.78 mmol, 2.1 eq.), 1.5M solution of *n*-butyllithium in hexane (5.4 mL, 8.15 mmol, 2.2 eq.) and lactone 1a (1.99 g, 3.70 mmol, 1 eq.) to give a white solid.

70: C₅₁H₅₂O₆S₂ M = 825.08 g.mol⁻¹

Rf: 0.51 (cyclohexane/ethyl acetate 75/25).

15 Mass (ESI⁺): 842.27 (M+H₂O).

Synthesis of compound 71



20 A solution of 70 (1.21 g, 1.57 mmol; 1 eq.) in acetonitrile (3 mL) is quickly added to a round-bottom flask which contains *N*-chlorosuccinimide (0.84 g, 6.28 mmol; 4 eq.) and silver nitrate (1.12 g, 6.60 mmol, 4.5 eq.) in a mixture of acetonitrile and water in proportions of 8:2 (30 mL) at room temperature. The mixture is vigorously stirred for 30 min. Saturated sodium sulfite aqueous solution (2 mL), saturated sodium carbonate

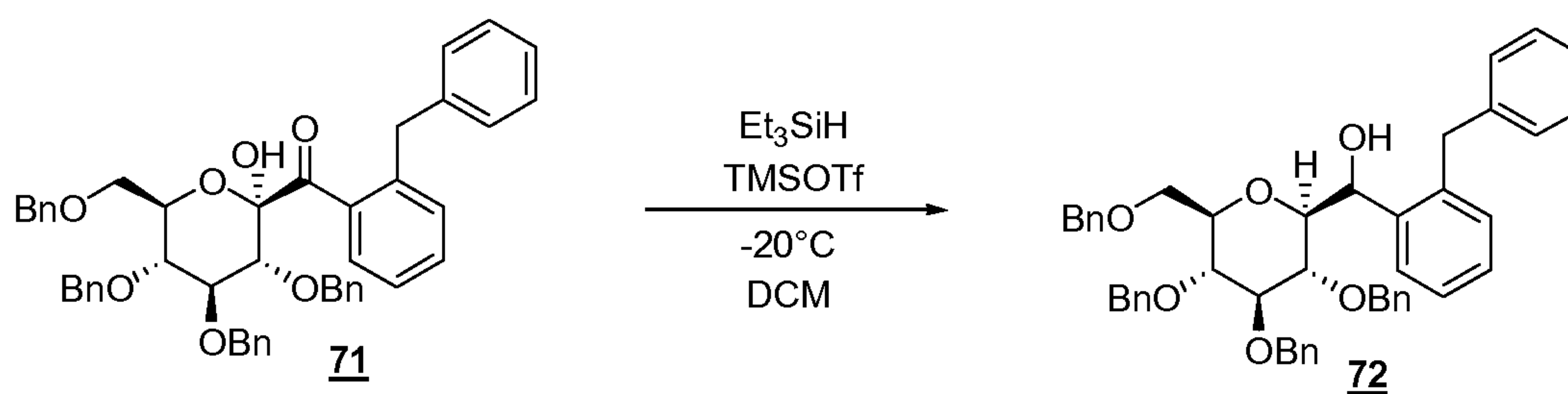
aqueous solution (2 mL), brine (2 mL) and cyclohexane (80 mL) are successively added to the reaction mixture. The reaction medium is filtered through celite, dried over magnesium sulfate prior to being concentrated. The residue is then purified on chromatography column (100/0 to 60/40 cyclohexane/ethyl acetate eluent) to produce 71 in the form of a white solid with a yield of 47%.

71: C₄₈H₄₆O₇ M = 734.87 g.mol⁻¹

Rf: 0.49 (cyclohexane/ethyl acetate 8/2).

Mass (ESI+): 752.27(M+H₂O); 1486.00 (2M+H₂O).

10 Synthesis of compound 72



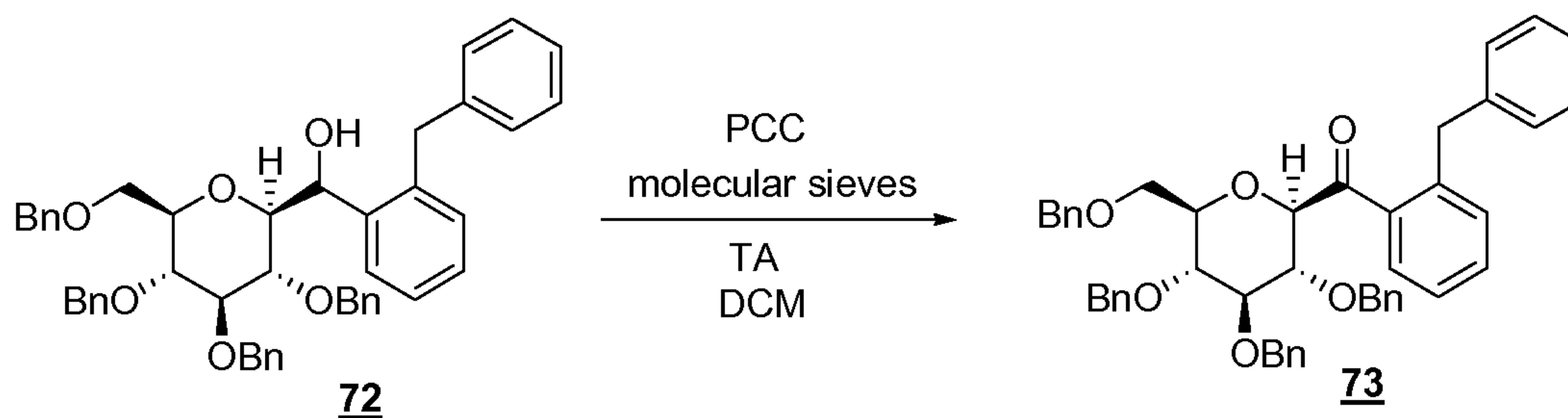
Compound 72 is prepared according to the procedure previously described (synthesis of compound 59) from triethylsilane (0.088 mL; 0.54 mmol, 4 eq.) and trimethylsilyl trifluoromethanesulfonate (0.025 mL, 0.14 mmol, 1eq.) in the form of a white solid, with a yield of 18 %.

72: C₄₈H₄₈O₆ M = 720.89 g.mol⁻¹

Rf: 0.24 (cyclohexane/ethyl acetate 85/15).

Mass (ESI+): 738.20 (M+H₂O); 1457.67 (2M+H₂O).

20 Synthesis of compound 73



Compound 73 is prepared according to the procedure previously described (synthesis of compound 62) at temperature ambient (TA) from compound 72 (0.016 g, 0.02 mmol)

and pyridinium chlorochromate (0.01 mg, 0.05 mmol, 2 eq.) to give a white solid, with a yield of 55 %.

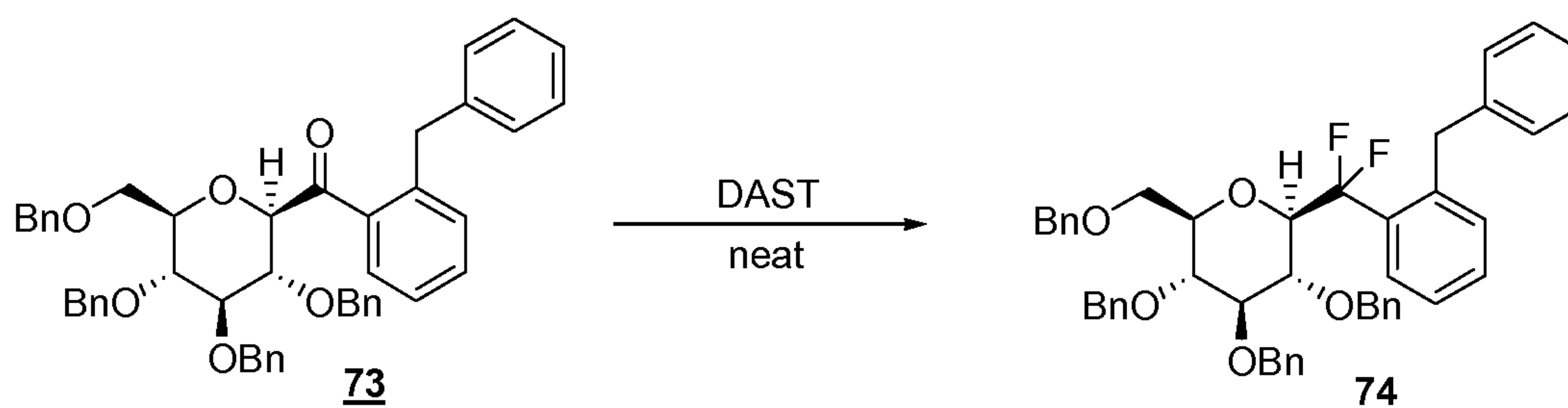
73: C₄₈H₄₆O₆ M = 718.88 g.mol⁻¹

Rf: 0.35 (cyclohexane/ethyl acetate 8/2).

5

Mass (ESI+): 736.27 (M+H₂O).

Synthesis of compound 74



Compound 73 is fluorinated 3 times with diethylaminosulfur trifluoride (0,3mL) neat by stirring overnight at 70°C in a round-bottom flask under an inert atmosphere according to the procedure previously described (synthesis of compound 63). Between each time, the residue needs to be purified on a chromatography column (80/20 cyclohexane/ethyl acetate eluent) to remove diethylaminosulfur trifluoride residues before being reintroduced in a fluorination reaction. The residue is purified on preparative HPLC (Kromasil 100-5C18, 15 cm*21.2 mm id, 100% acetonitrile, 254 nm).

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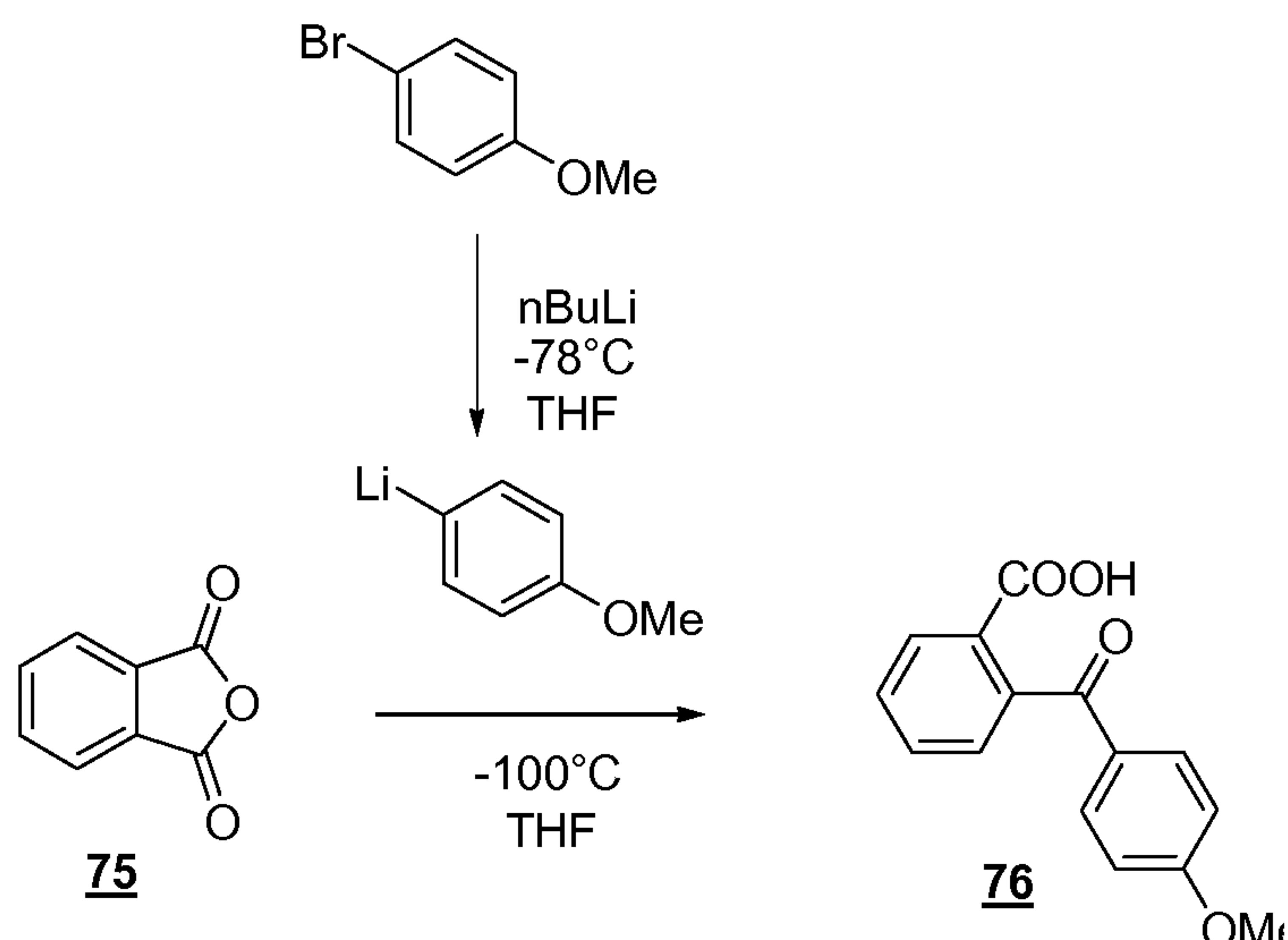
74: C₄₈H₄₆F₂O₅ M = 740.87 g.mol⁻¹

Rf: 0.5 (cyclohexane/ethyl acetate 8/2).

NMR ¹⁹F (CDCl₃, 282.5MHz) : -95.3 (d, J = 259 Hz, 1F); -105.2 (dd, J1 = 19 Hz, J2 = 259 Hz, 1F).

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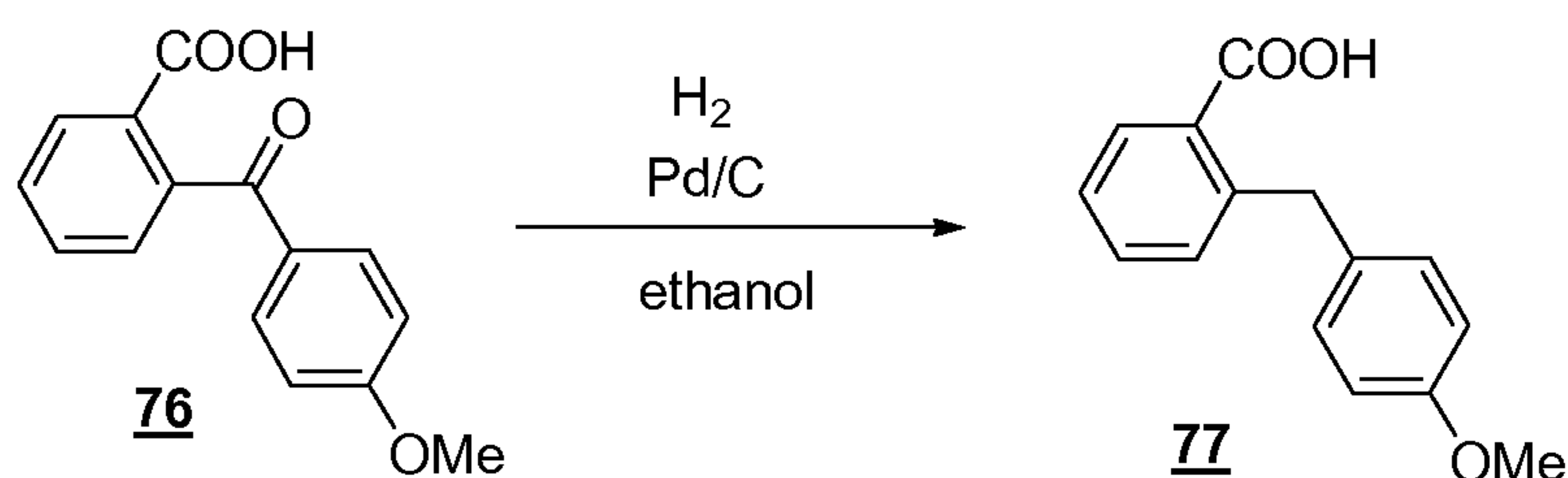
Synthesis of compound 76

A 2.5 M solution of *n*-butyllithium in hexane (13.5 mL, 33.8 mmol, 1 eq.) is added dropwise to a round-bottom flask under an inert atmosphere which contains 1-Bromo-4-methoxybenzene (4.7 mL, 37.1 mmol, 1.1 eq.) in dry tetrahydrofuran (100 mL) at -78°C . The mixture is stirred at this temperature for 1h before being quickly added to a solution of **75** (10.0 g, 67.5 mmol, 2 eq.) in tetrahydrofuran (10 mL) at -100°C . The mixture is stirred 1h at this temperature and 2h at ambient temperature. The mixture is concentrated and then diluted in diethyl ether. Water and then a 1N hydrochloric acid aqueous solution are added. The organic layer is washed with a saturated sodium carbonate solution and the aqueous layer is acidified with concentrated hydrochloric acid. The precipitate is filtered and dried to produce compound **76** in the form of a white solid with a yield of 35%.

76: $\text{C}_{15}\text{H}_{12}\text{O}_4$ $M = 256.25 \text{ g}\cdot\text{mol}^{-1}$

Rf: 0.27 (cyclohexane/ethyl acetate 3/7).

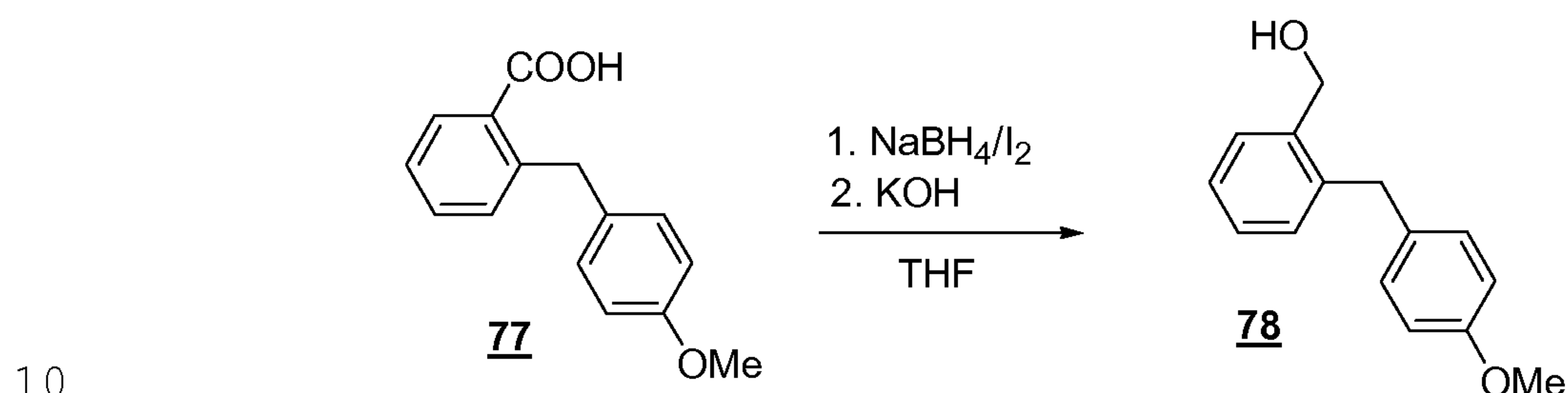
Mass (ESI⁺): 257.03 (M+H); 273.80 (M+H₂O).

Synthesis of compound 77

Compound 76 (2.98 g, 11.63 mmol, 1 eq.) is placed inside a round-bottom flask and dissolved in ethanol (115 mL) in the presence of a spatula tip of Pd/C under a hydrogen atmosphere. The mixture is stirred for 6 days, then Millipore-filtered and evaporated to produce compound 77 in the form of a white powder with a yield of 97%.

5 77: C₁₅H₁₄O₃ M = 242.27 g.mol⁻¹
 Rf: 0.3 (cyclohexane/ethyl acetate 3/7).
 Mass (ESI-): 241.38 (M-H); 482.94 (2M-H).

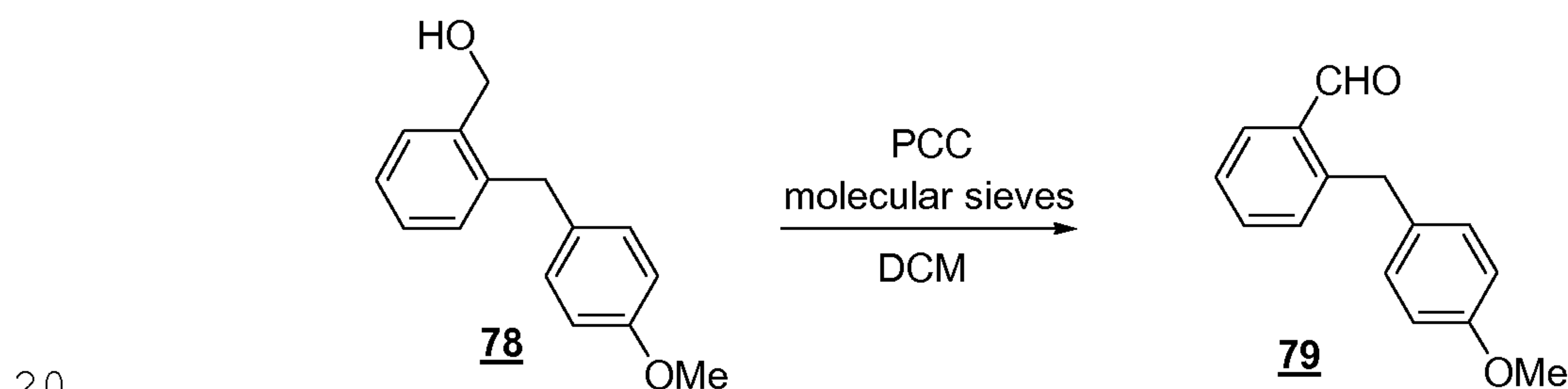
Synthesis of compound 78



Compound 78 was prepared according to the procedure previously described (synthesis of compound 67) from compound 77 (13.83 g, 57.1 mmol, 1 eq.) sodium borohydride (5.20 g, 137.0 mmol, 2.4 eq.) and iodine (14.5 g, 57.1 mmol, 1 eq.) in the form of a yellow oil with a quantitative yield.

15 78: C₁₅H₁₆O₂ M = 228.29 g.mol⁻¹
 Rf: 0.28 (cyclohexane/ethyl acetate 7/3).
 Mass (CI+): 228 (M)

Synthesis of compound 79



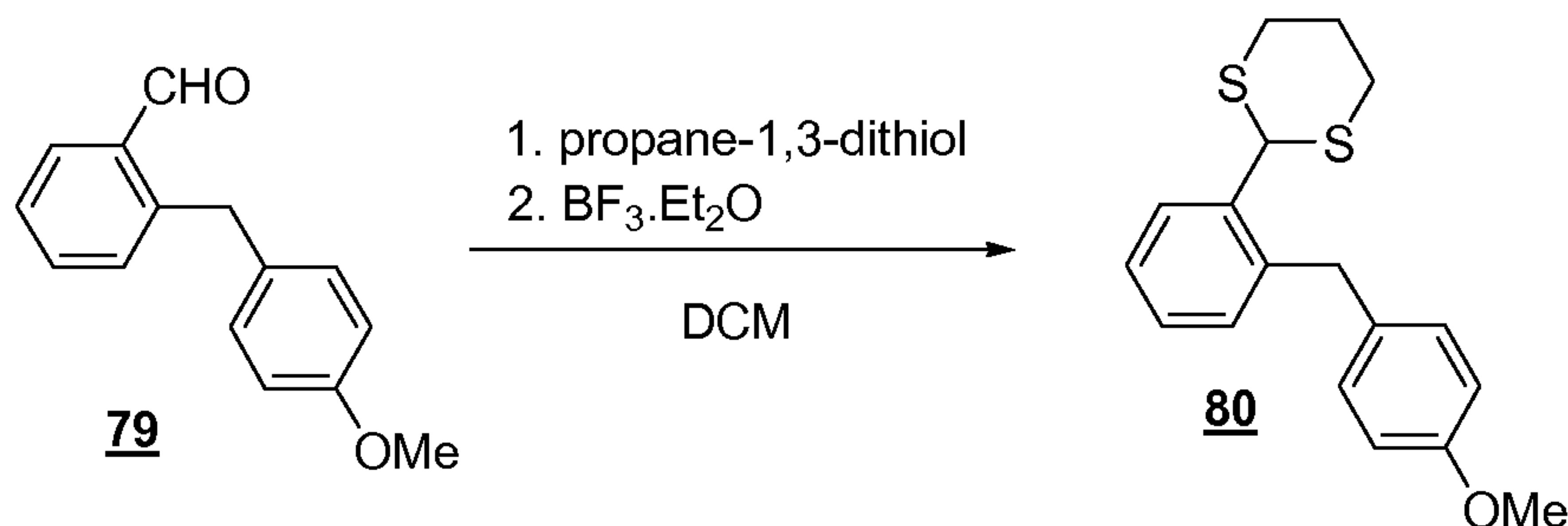
Compound 79 was prepared according to the procedure previously described (synthesis of compound 68) from compound 78 (13.0 g, 56.9 mmol, 1 eq.) and pyridinium chlorochromate (17.2 g, 79.7 mmol, 1.4 eq.) to give a yellow oil with a yield of 81%.

79: C₁₅H₁₄O₂ M = 226.27 g.mol⁻¹

Rf: 0.14 (cyclohexane/ethyl acetate 95/5).

Mass (CI+): 227(M+H).

Synthesis of compound 80



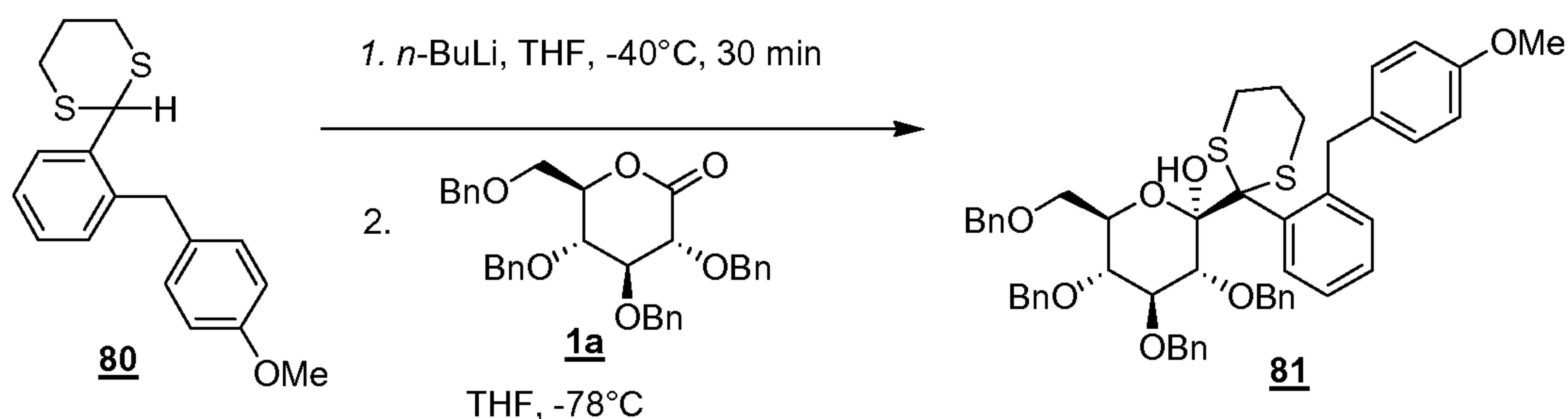
Compound 80 was prepared according to the procedure previously described (synthesis of compound 69) from compound 79 (1.76 g, 7.78 mmol, 1 eq.), propane-1,3-dithiol (1.20 mL, 11.7 mmol, 1.5 eq.) and boron trifluoride etherate (0.20 mL, 1.56 mmol, 0.2 eq.) to give a white solid with a yield of 94%.

80: $C_{18}H_{20}OS_2M = 316.48 \text{ g.mol}^{-1}$

Rf: 0.55 (cyclohexane/ethyl acetate 9/1).

Mass (CI+): 317 (M+H).

Synthesis of compound 81

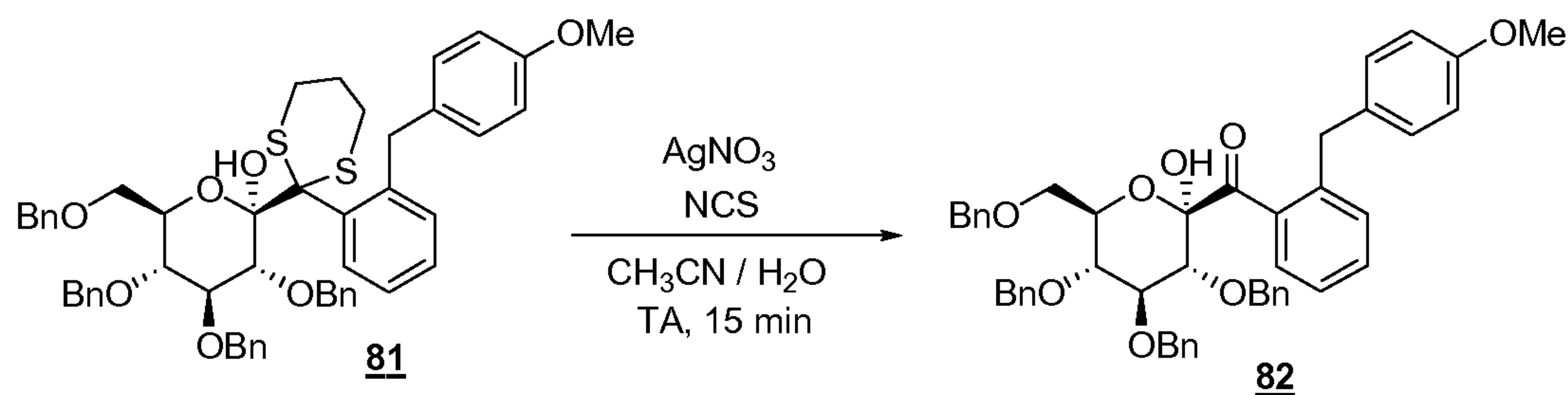


Compound 81 is prepared according to the procedure previously described (synthesis of compound 57) from compound 80 (2.19 g, 6.91 mmol, 2.1 eq.), 1.4M solution of *n*-butyllithium in hexane (5.17 mL, 7.24 mmol, 2.2 eq) and lactone 1a (1.77 g, 3.29 mmol, 1 eq.) to give a white solid.

81: $C_{52}H_{54}O_7S_2$ $M = 855.11 \text{ g.mol}^{-1}$

Rf: 0.22 (cyclohexane/ethyl acetate 8/2).

Mass (ESI+): 872.20 (M+H₂O)

Synthesis of compound 82

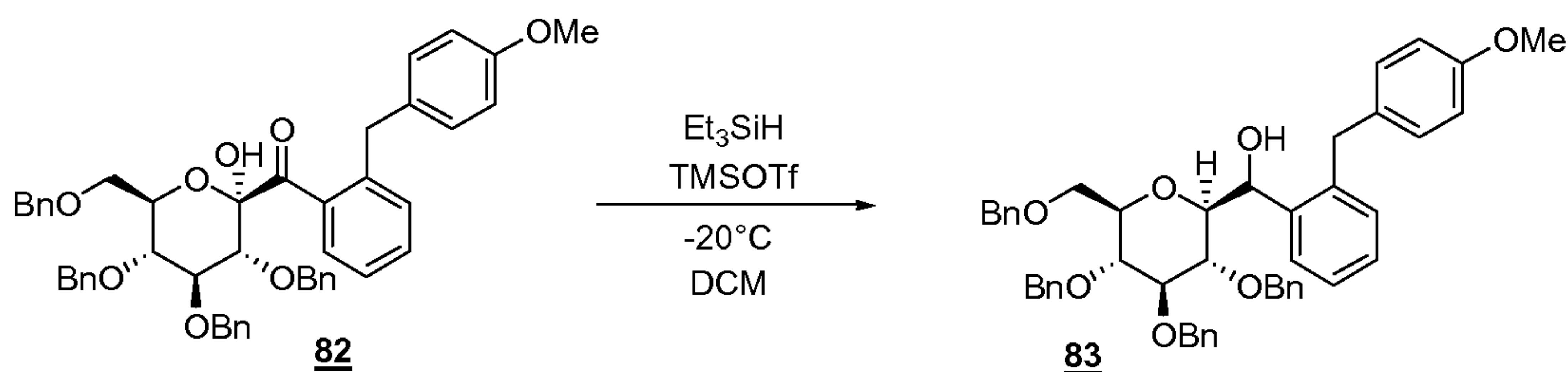
Compound **82** is prepared according to the procedure previously described (synthesis of compound **71**) from compound **81** (1.29 g, 1.50 mmol, 1 eq.), *N*-chlorosuccinimide (0.80 g, 6.00 mmol; 4 eq.) and silver nitrate (1.15 g, 6.76 mmol, 4.5 eq.) to give a white solid with a yield of 56%.

82: $\text{C}_{49}\text{H}_{48}\text{O}_8$ $M = 764.90 \text{ g.mol}^{-1}$

Rf: 0.46 (cyclohexane/ethyl acetate 75/25).

Mass (ESI+): 782.20 (M+H); 1546.13 (M+H₂O).

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Synthesis of compound 83

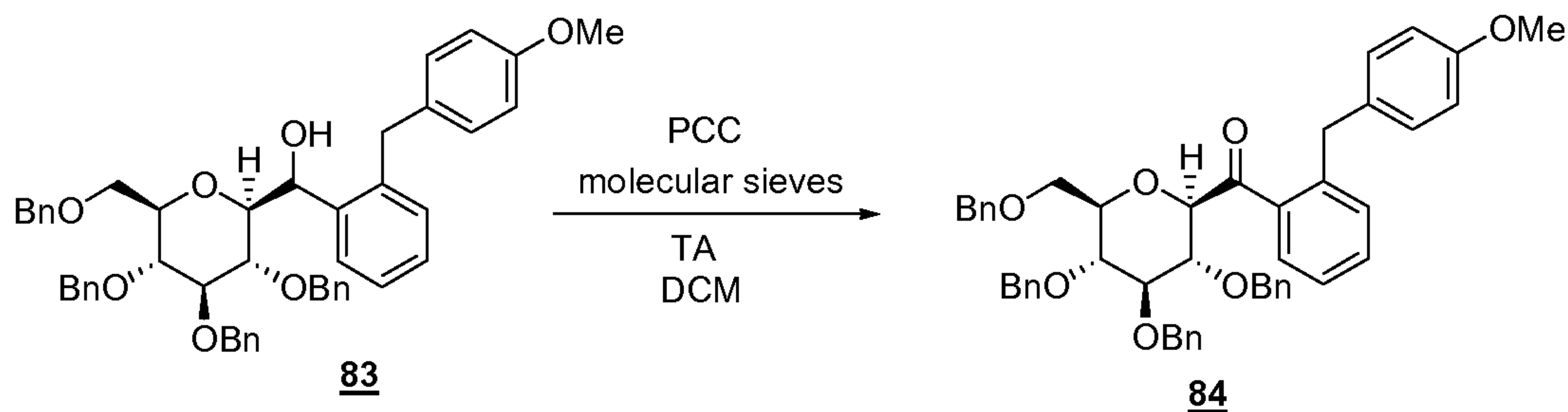
Compound **83** is prepared according to the procedure previously described (synthesis of compound **59**) from triethylsilane (0.087 mL; 0.54 mmol, 4 eq.) and trimethylsilyl trifluoromethanesulfonate (0.025 mL, 0.14 mmol, 1eq.) in the form of a white solid, with a yield of 17 %.

83: $\text{C}_{49}\text{H}_{50}\text{O}_7$ $M = 750.92 \text{ g.mol}^{-1}$

Rf: 0.25 (cyclohexane/ethyl acetate 8/2).

Mass (ESI+): 768.27 (M+H₂O); 1518.20 (2M+H₂O).

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Synthesis of compound 84

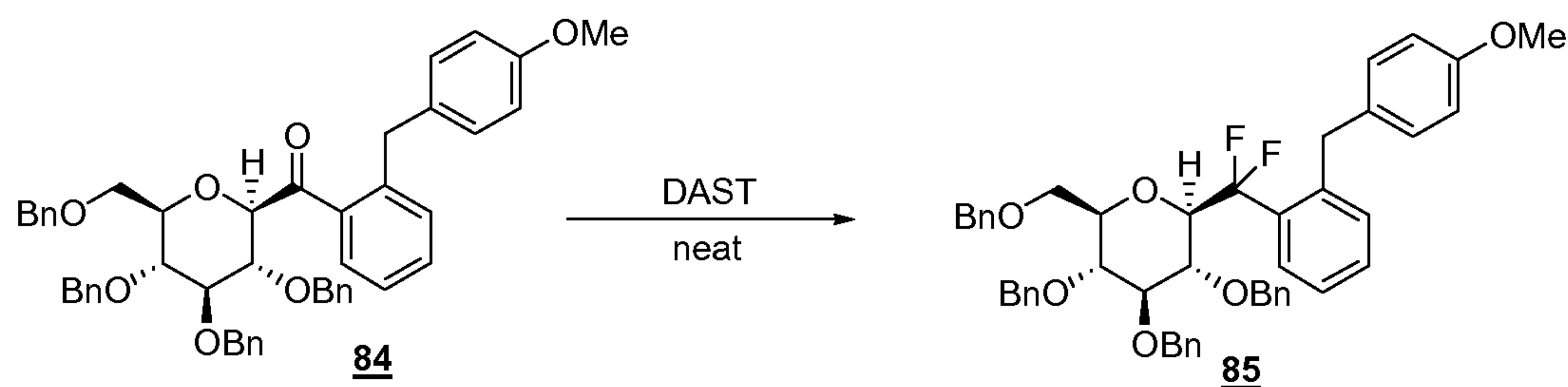
Compound **84** is prepared according to the procedure previously described (synthesis of compound **62**) from compound **83** (0.015 g, 0.02 mmol) and pyridinium chlorochromate (0.008 mg, 0.04 mmol, 2 eq) to give a white solid, with a yield of 56%.

84: C₄₉H₄₈O₇ M = 748.90 g.mol⁻¹

Rf: 0.43 (cyclohexane/ethyl acetate 8/2).

Mass (ESI+): 766.20 (M+H₂O); 1514.73 (2M+H₂O).

10 Synthesis of compound 85



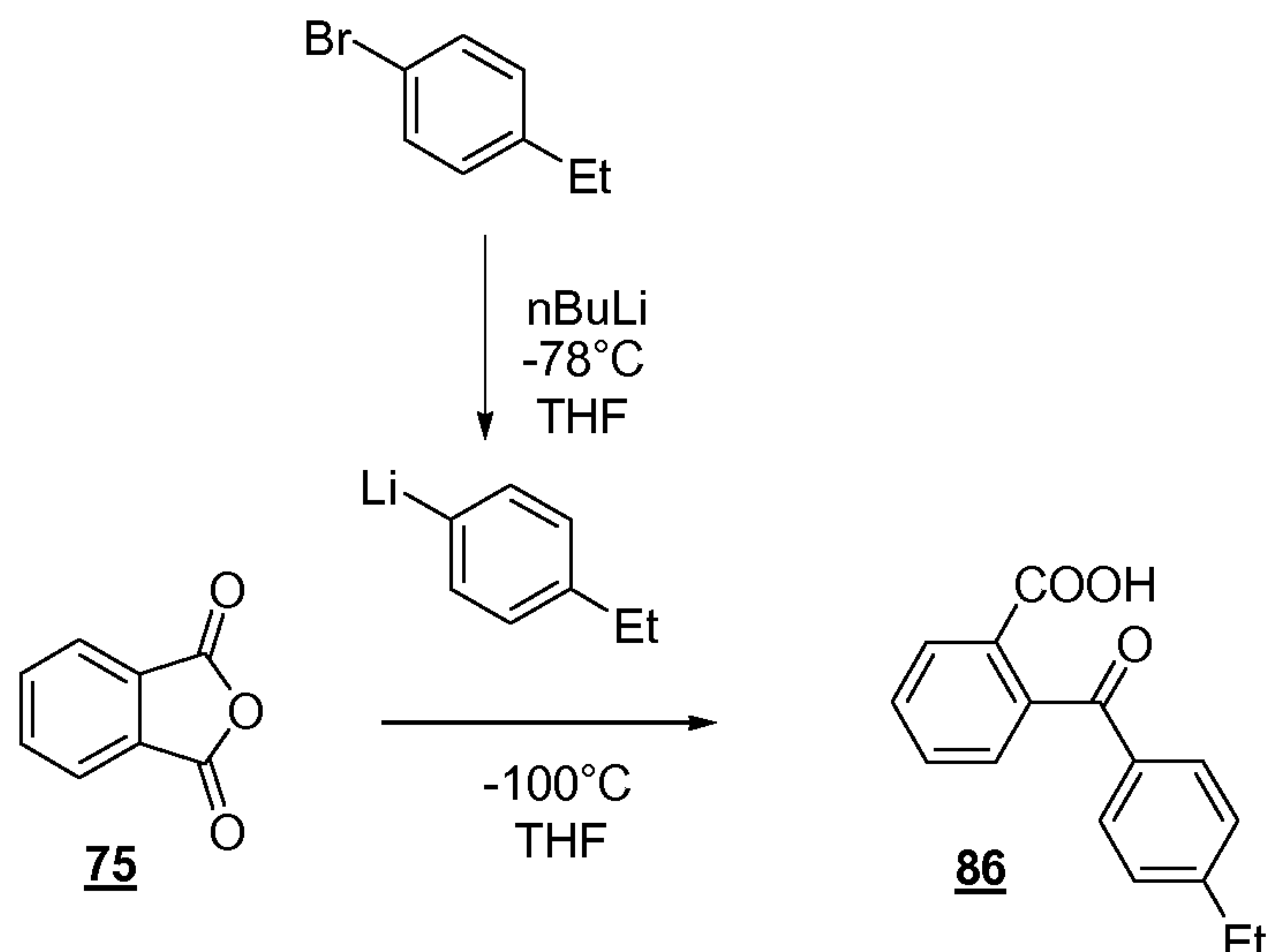
Compound **84** is fluorinated 3 times with diethylaminosulfur trifluoride (0,3 mL) neat by stirring overnight at 70°C in a round-bottom flask under an inert atmosphere according to the procedure previously described (synthesis of compound **63**). Between each time, the residue needs to be purified on a chromatography column (80/20 cyclohexane/ethyl acetate eluent) to remove diethylaminosulfur trifluoride residues before being reintroduced in a fluorination reaction. The residue is purified on preparative HPLC (Kromasil 100-5C18, 15 cm*21.2 mm id, 100% acetonitrile, 254 nm).

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85: C₄₉H₄₈F₂O₆ M = 770.90 g.mol⁻¹

Rf: 0.48 (cyclohexane/ethyl acetate 8/2).

NMR ¹⁹F (CDCl₃, 282.5MHz) : -95.2 (d, J = 259 Hz, 1F); -105.2 (dd, J1 = 19 Hz, J2 = 258 Hz).

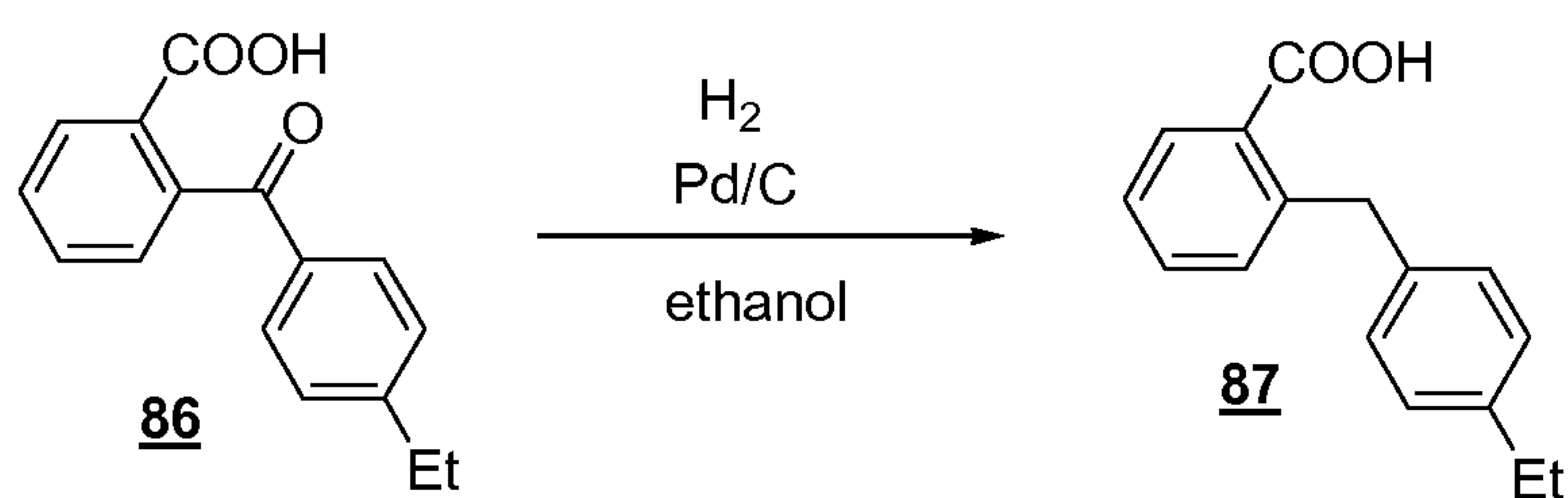
Synthesis of compound 86

Compound **86** was prepared according to the procedure previously described (synthesis of compound **76**) from compound **75** (10.0 g, 67.5 mmol, 2 eq.), 2.5M solution of *n*-butyllithium in hexane (13.5 mL, 33.8 mmol, 1 eq.) and 1-Bromo-4-ethylbenzene (5.1 mL, 37.1 mmol, 1.1 eq.) to give a white solid with a yield of 40%.

86: $\text{C}_{16}\text{H}_{14}\text{O}_3$ $M = 254.28 \text{ g.mol}^{-1}$

Rf: 0.24 (cyclohexane/ethyl acetate 5/5).

Mass (ESI+): 255.10 (M+H); 271.93 (M+H₂O).

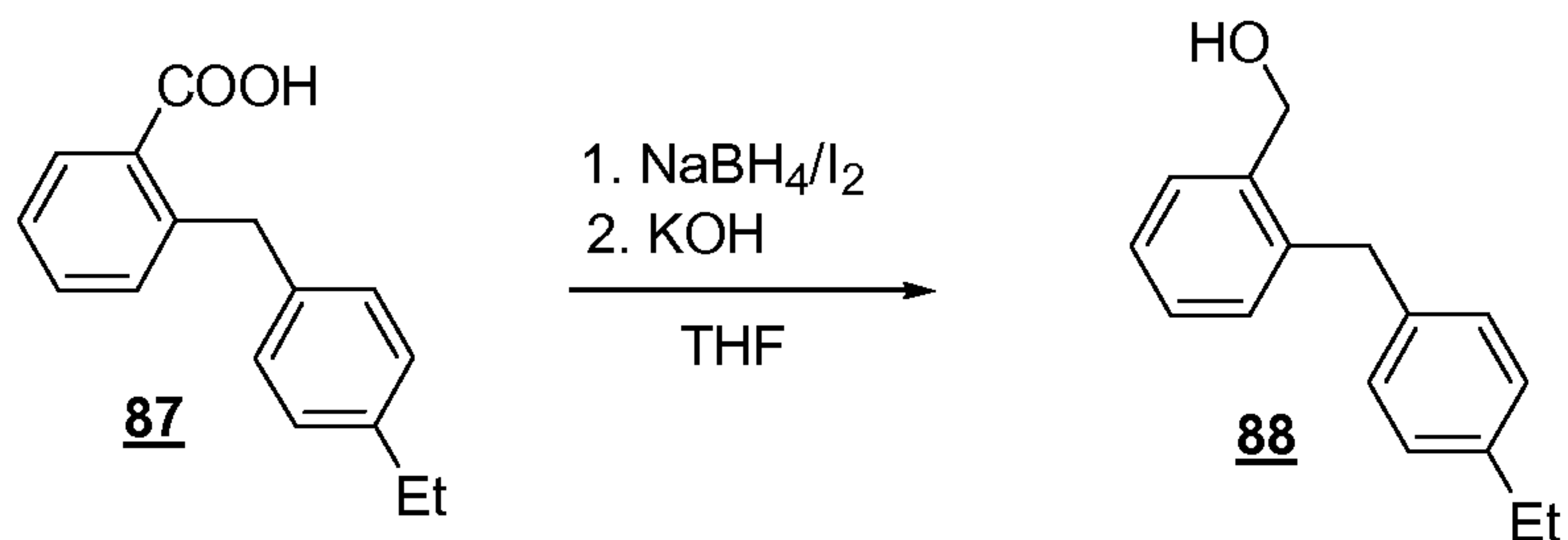
Synthesis of compound 87

Compound **86** was deprotected according to the procedure previously described (synthesis of compound **77**) in 48h, to give a white solid with a quantitative yield.

87: $\text{C}_{16}\text{H}_{16}\text{O}_2$ $M = 240.30 \text{ g.mol}^{-1}$

Rf: 0.61 (cyclohexane/ethyl acetate 5/5).

Mass (ESI-): 239.27 (M-H).

Synthesis of compound 88

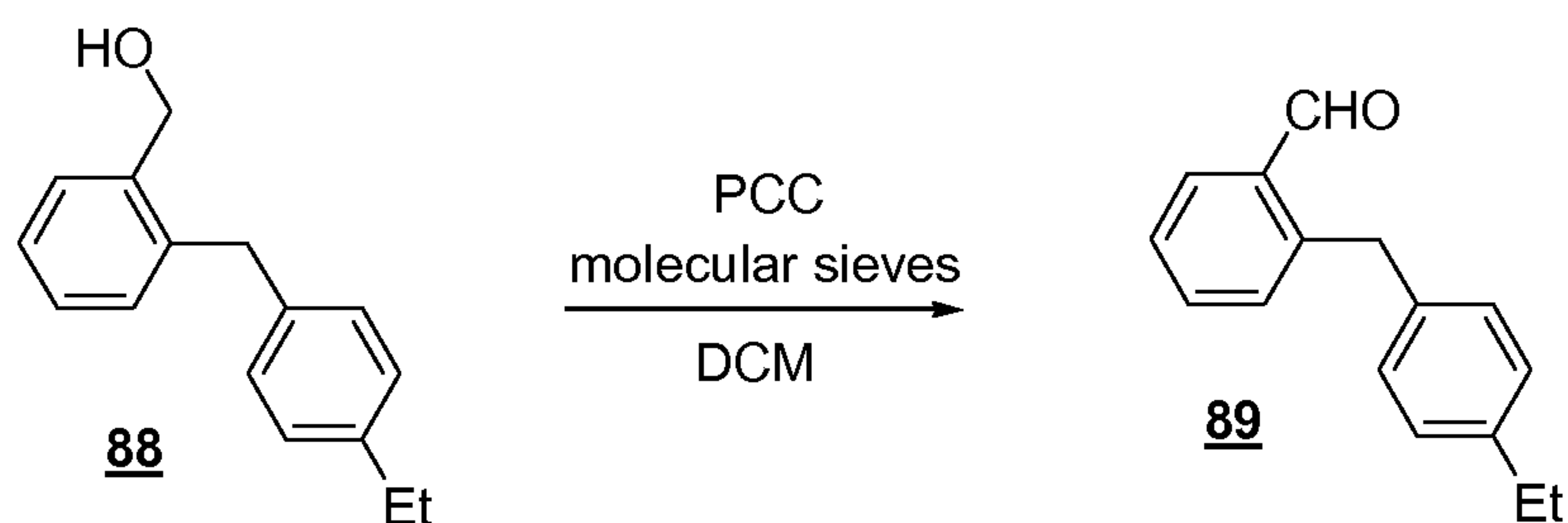
Compound 88 was prepared according to the procedure previously described (synthesis of compound 67) from compound 87 (22.7 g, 36.0 mmol, 1 eq.), sodium borohydride (8.51 g, 225 mmol, 2.4 eq.) and iodine (23.8 g, 93.6 mmol, 1 eq.) to give a colourless oil with a quantitative yield.

88: C₁₆H₁₈O M = 226.31 g.mol⁻¹

Rf: 0.53 (cyclohexane/ethyl acetate 7/3).

Mass (ESI+): 243.99 (M+H₂O).

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Synthesis of compound 89

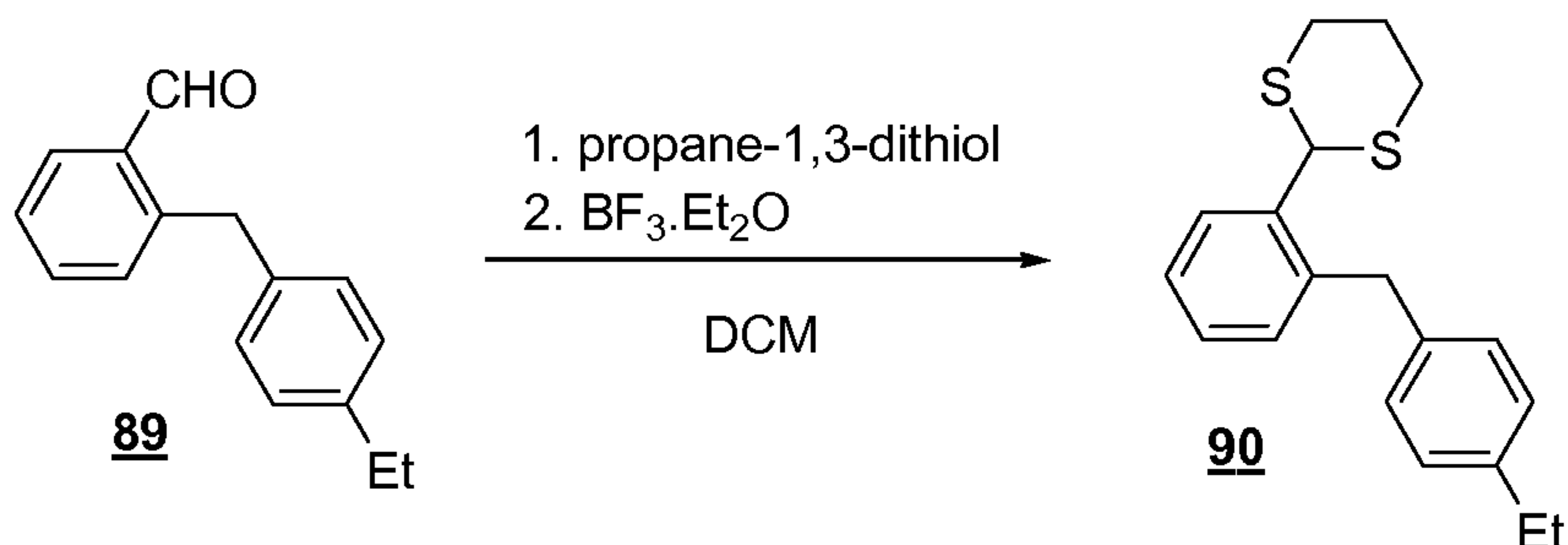
Compound 89 was prepared according to the procedure previously described (synthesis of compound 68) from compound 88 (21.2 g, 93.6 mmol, 1 eq.) and pyridinium chlorochromate (28.25 g, 131.0 mmol, 1.4 eq.) to give a yellow oil with a yield of 81%.

89: C₁₆H₁₆O M = 224.30 g.mol⁻¹

Rf: 0.39 (cyclohexane/ethyl acetate 95/5).

Mass (CI+): 225 (M+H).

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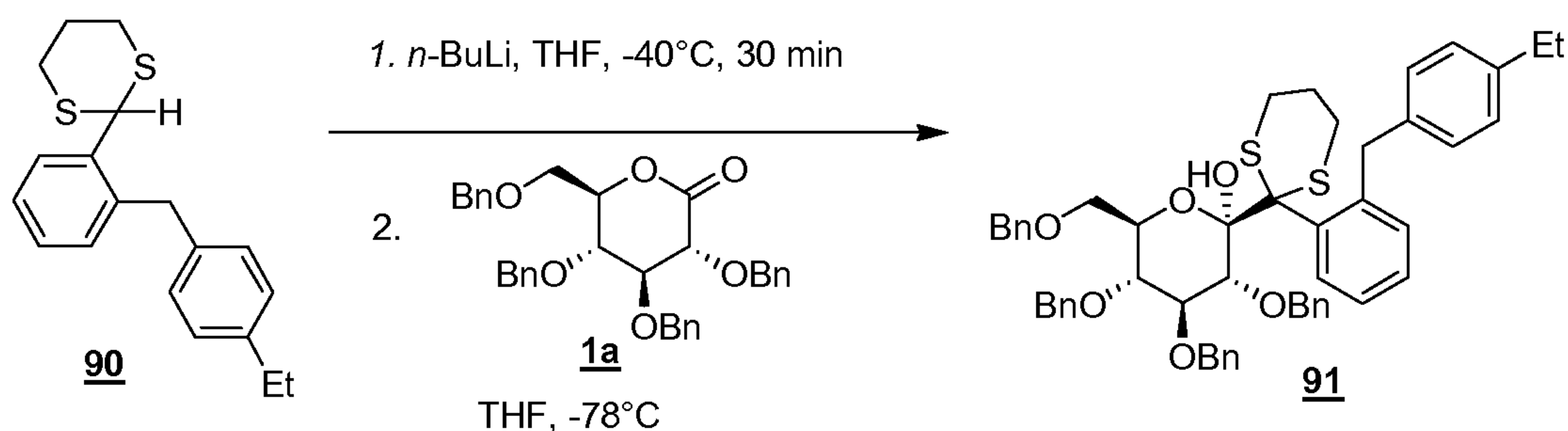
Synthesis of compound 90

Compound 90 was prepared according to the procedure previously described (synthesis of compound 69) from compound 89 (2.39 g, 10.7 mmol, 1 eq.), propane-1,3-dithiol (1.6 mL, 15.5 mmol, 1.5 eq.) and boron trifluoride etherate (0.27 mL, 2.13 mmol, 0.2 eq.) in the form of a white solid with a yield of 82%.

90: $\text{C}_{19}\text{H}_{22}\text{S}_2$ $M = 314.51 \text{ g.mol}^{-1}$

Rf: 0.63 (cyclohexane/ethyl acetate 9/1).

Mass (ESI⁺): 315 (M+H).

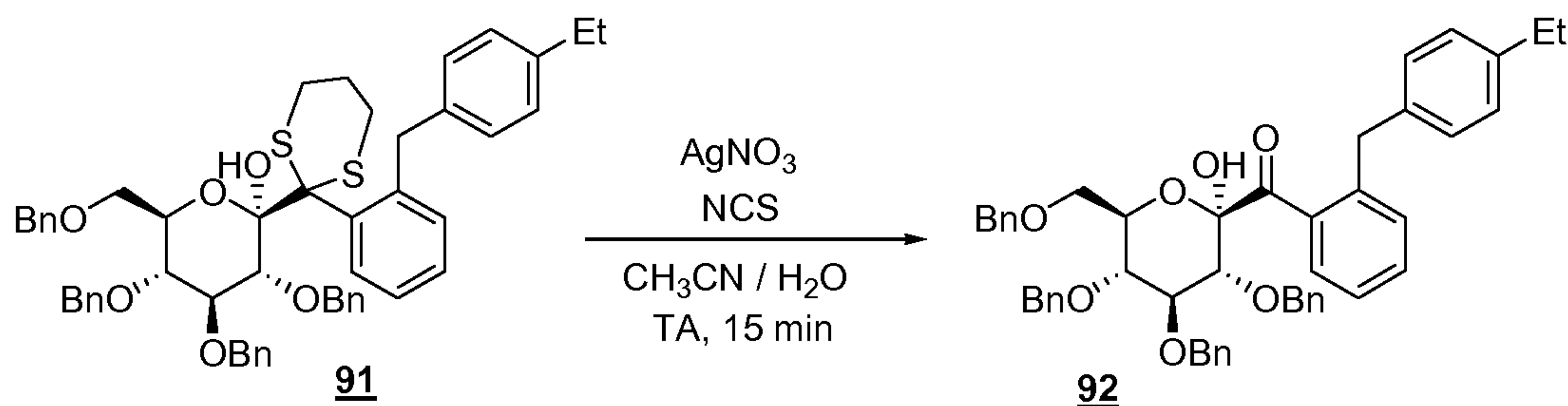
Synthesis of compound 91

Compound 91 is prepared according to the procedure previously described (synthesis of compound 57) from compound 90 (2.79 g, 8.87 mmol, 4.4 eq.), 1.4M solution of *n*-butyllithium in hexane (6.4 mL, 9.09 mmol, 4.5 eq) and lactone 1a (1.09 g, 2.02 mmol, 1 eq.) in the form of a yellow oil.

91: $\text{C}_{53}\text{H}_{56}\text{O}_6\text{S}_2$ $M = 853.14 \text{ g.mol}^{-1}$

Rf: 0.27 (cyclohexane/ethyl acetate 8/2).

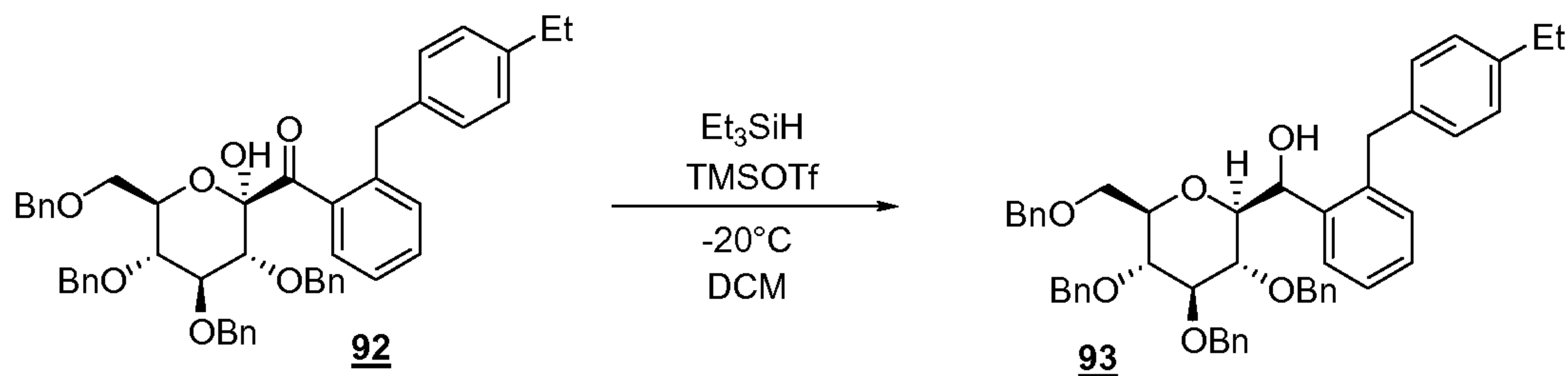
Mass (ESI⁺): 870.07 (M+H₂O).

Synthesis of compound 92

Compound **92** is prepared according to the procedure previously described (synthesis of compound **71**) from compound **91** (0.67 g, 0.79 mmol, 1 eq.), *N*-chlorosuccinimide (0.42 g, 3.14 mmol; 4 eq.) and silver nitrate (0.60 g, 3.53 mmol, 4.5 eq.) to give a white solid with a yield of 48%.

92: $\text{C}_{50}\text{H}_{50}\text{F}_2\text{O}_7$ $M = 762.93 \text{ g.mol}^{-1}$

Rf: 0.48 (cyclohexane/ethyl acetate 8/2).

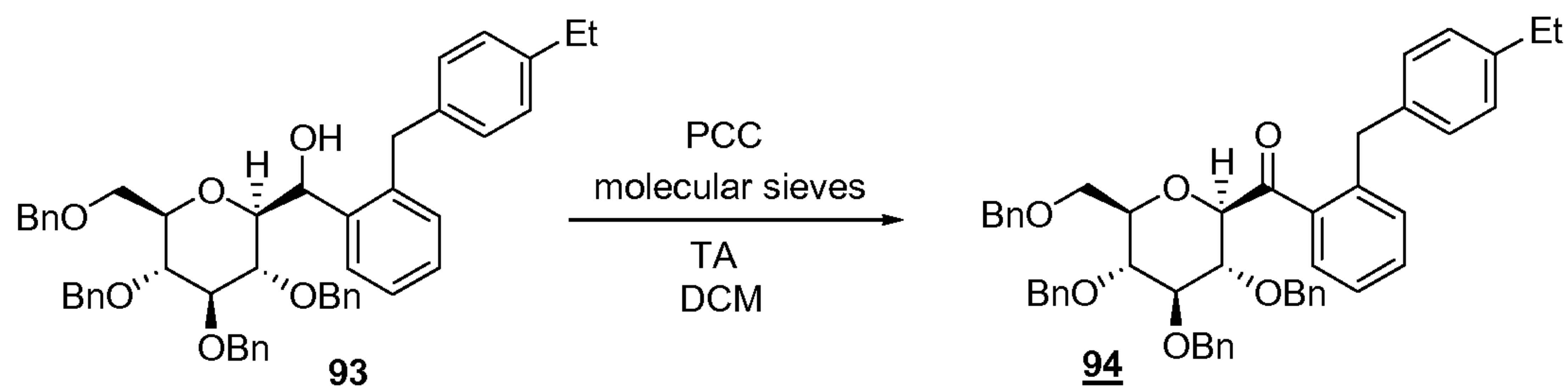
10 Synthesis of compound 93

Compound **93** is prepared according to the procedure previously described (synthesis of compound **59**) from triethylsilane (0.238 mL; 1.47 mmol, 4 eq.) and trimethylsilyl trifluoromethanesulfonate (0.067 mL, 0.37 mmol, 1eq.) to give a white solid, with a yield of 19 %.

93: $\text{C}_{50}\text{H}_{52}\text{O}_6$ $M = 748.94 \text{ g.mol}^{-1}$

Rf: 0.30 (cyclohexane/ethyl acetate 8/2).

Mass (ESI⁺): 766.20 (M+H₂O); 1514.93 (2M+H₂O)

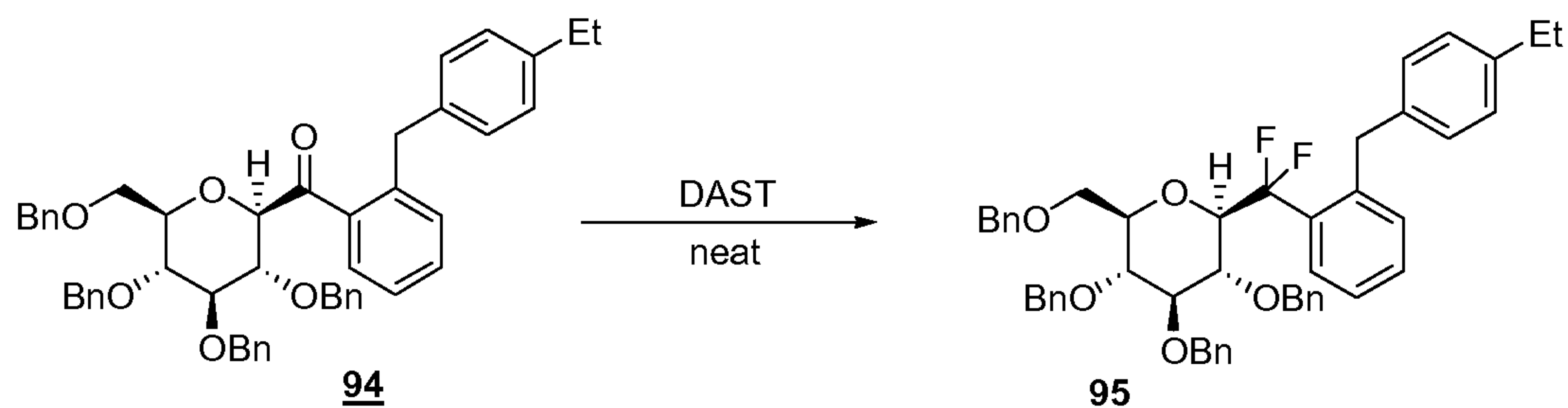
Synthesis of compound 94

Compound 94 is prepared according to the procedure previously described (synthesis of compound 62) from compound 93 (0.052 g, 0.07 mmol) and pyridinium chlorochromate (0.030 mg, 0.14 mmol, 2 eq.) to give a white solid, with a yield of 56%.

94: C₅₀H₅₀O₆ M = 746.93 g.mol⁻¹

Rf: 0.49 (cyclohexane/ethyl acetate 8/2).

Mass (ESI+): 764.40 (M+H₂O); 1510.93 (2M+H₂O).

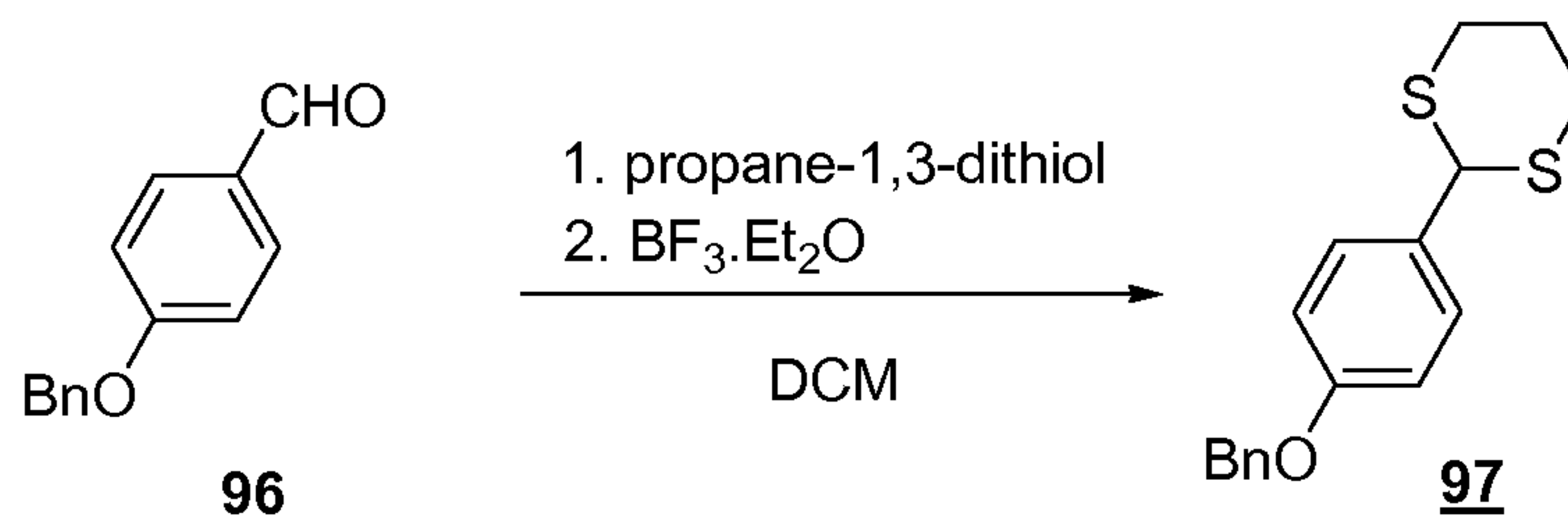
10 Synthesis of compound 95

Compound 94 (0.033 g, 0.04 mmol, 1 eq.) is fluorinated 3 times with diethylaminosulfur trifluoride (0.300 mL) neat by stirring overnight at 70°C in a round-bottom flask under an inert atmosphere according to the procedure previously described (synthesis of compound 63). Between each time, the residue needs to be purified on a chromatography column (80/20 cyclohexane/ethyl acetate eluent) to remove diethylaminosulfur trifluoride residues before being reintroduced in a fluorination reaction. The residue is purified on preparative HPLC (Kromasil 100-5C18, 15 cm*21.2 mm id, 100% acetonitrile, 254 nm) to produce compound 95 in the form of a yellow oil with a yield of 35%.

95: C₅₀H₅₀F₂O₅ M = 768.93 g.mol⁻¹

NMR ¹⁹F (CDCl₃, 282.5MHz) : -95.3 (d, J = 258 Hz, 1F); -105.3 (dd, J₁ = 19 Hz, J₂ = 258 Hz, 1F).

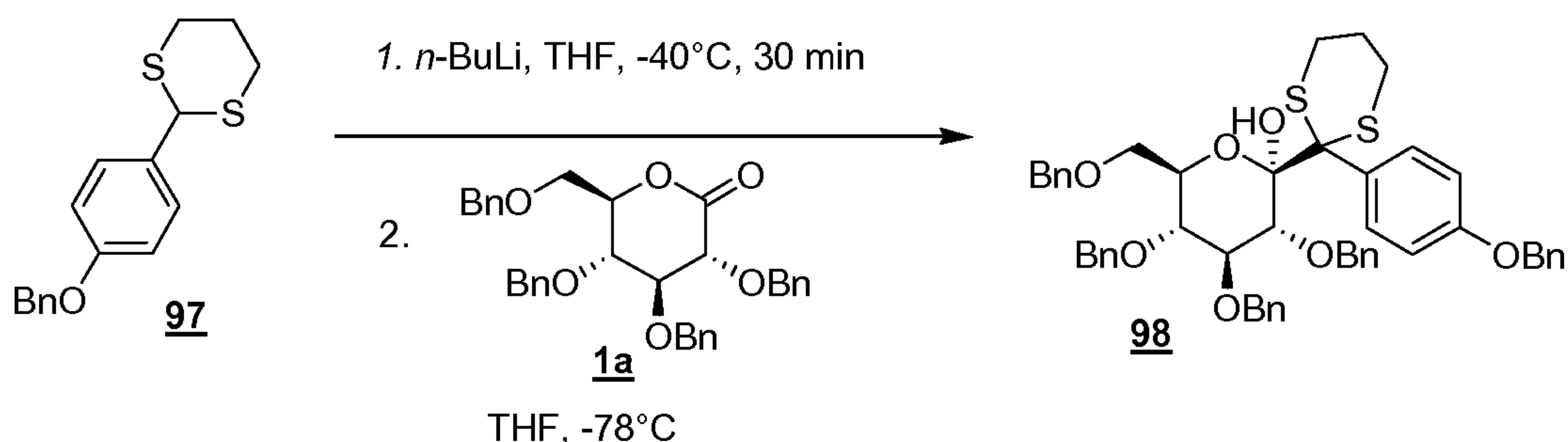
Rf: 0.55 (cyclohexane/ethyl acetate 8/2).

Synthesis of compound 97

Compound **97** is prepared according to the procedure previously described (synthesis of compound **69**) from compound **96** (10 g, 47.1 mmol, 1 eq.), 1,3-propanedithiol (7.15 mL ; 70.7 mmol ; 1.5 eq.) and boron trifluoride etherate (0.70 mL ; 5.54 mmol ; 0.1 eq.) to give white crystals with a yield of 89%.

97: C₁₇H₁₈OS₂ M = 302,45 g.mol⁻¹

Mass (EI): 302 (M)

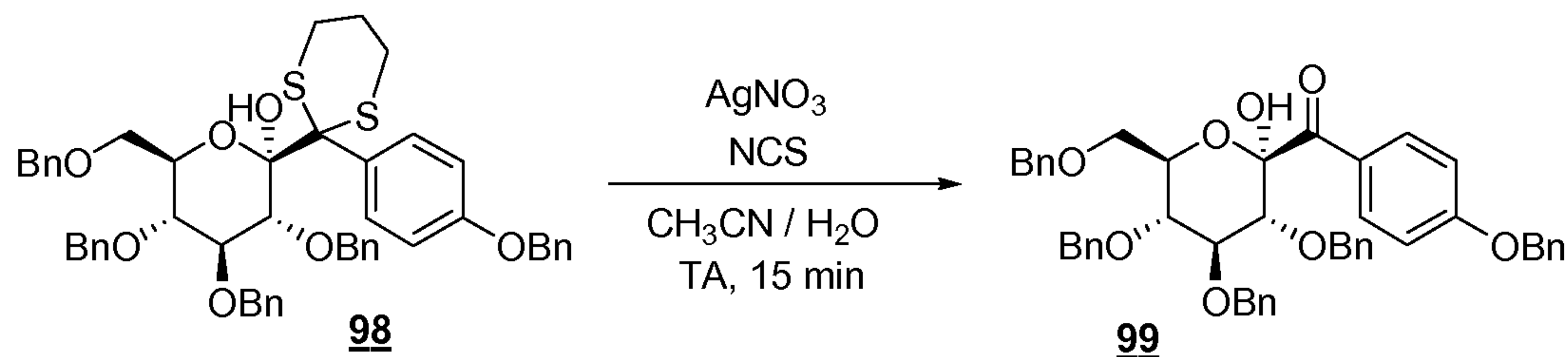
10 Synthesis of compound 98

Compound **98** is prepared according to the procedure previously described (synthesis of compound **57**) from compound **97** (3.54 g, 11.7 mmol, 2.1 eq.), 2,5M solution of *n*-Butyllithium in hexane (4.8 mL, 12.0 mmol, 2.2 eq.) and lactone **1a** (3.00 g, 5.57 mmol, 1 eq.) to give a white solid with a yield of 59%.

98: C₅₁H₅₂O₇S₂ M = 841.08 g.mol⁻¹

R_f: 0.33 (cyclohexane/ethyl acetate 8/2).

Mass (ESI+): 858.07 (M+H₂O)

Synthesis of compound 99

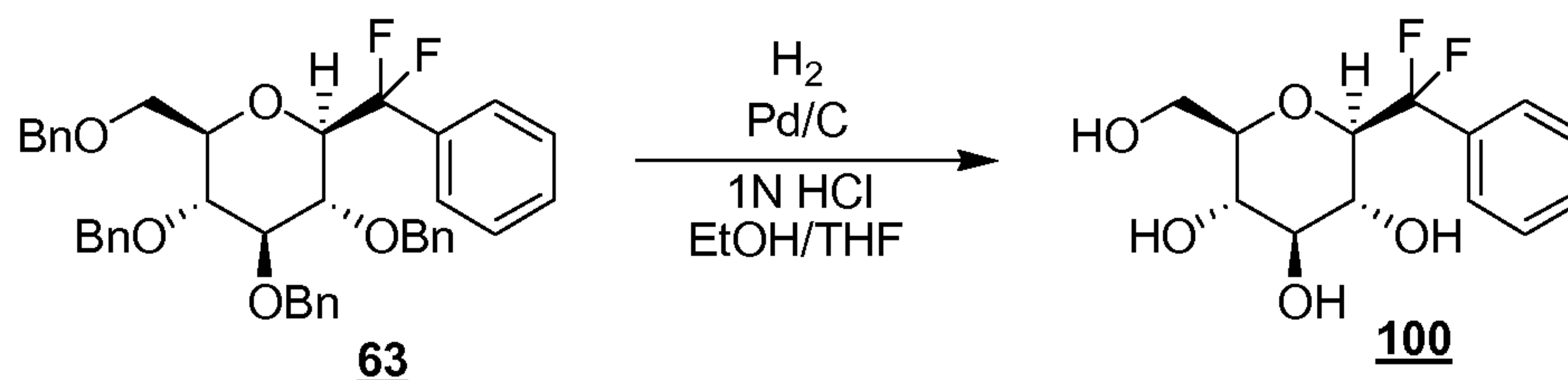
Compound **99** is prepared according to the procedure previously described (synthesis of compound **71**) from compound **98** (4.00 g, 4.76 mmol, 1 eq.), *N*-chlorosuccinimide (2.66 g, 19.0 mmol, 4 eq.) and silver nitrate (3.64 g, 21.0 mmol, 4.5 eq.) to give a colourless oil which slowly crystallizes with a yield of 84%.

99: $\text{C}_{48}\text{H}_{46}\text{O}_8$ $M = 750.87 \text{ g.mol}^{-1}$

Rf: 0.32 (cyclohexane/ethyl acetate 8/2).

Mass (ESI⁺): 768.13 (M+H₂O).

10

Synthesis of compound 100

Compound **63** (19.3 mg, 0.03 mmol, 1 eq.) is deprotected according to the procedure described previously (synthesis of **51**) to afford compound **100** in the form of a white solid, with a 87% yield.

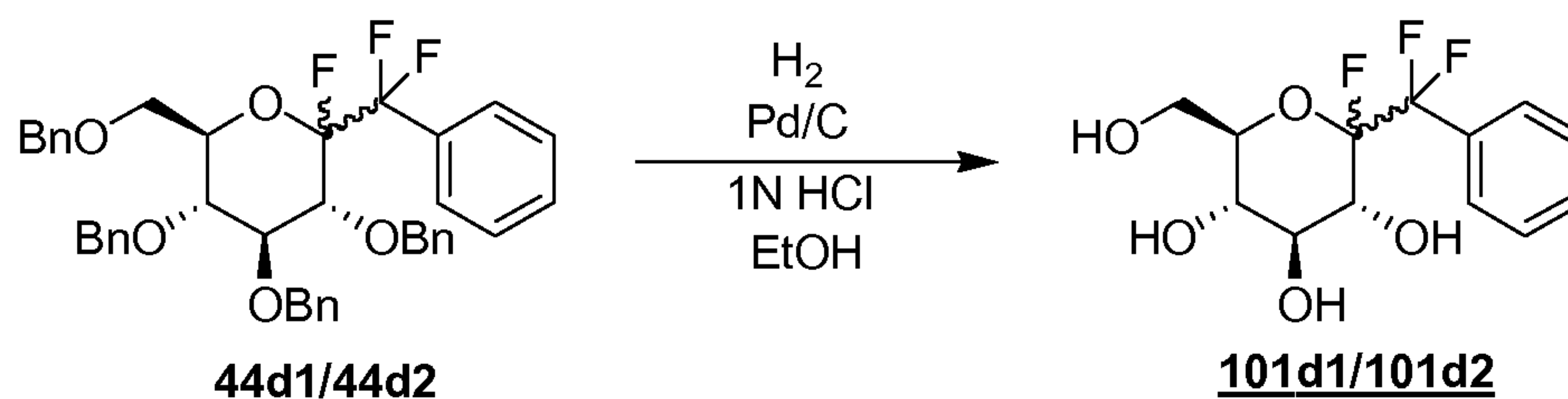
15

100: $\text{C}_{13}\text{H}_{16}\text{F}_2\text{O}_5$ $M = 290.26 \text{ g.mol}^{-1}$

NMR ^{19}F (*MeOD*, 282.5 MHz) : -98.4 (dd, J₁ = 260 Hz, J₂ = 6 Hz, 1F); -107.2 (dd, J₁ = 261 Hz, J₂ = 11 Hz, 1F).

Mass (ESI⁻): 325.0 (M+Cl).

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Synthesis of compound 101d1/101d2

Compound 44d1 (40.6 mg, 60.8 mmol, 1 eq.) is deprotected according to the procedure described previously (synthesis of 51) to afford compound 101d1 in the form of a white solid, with a quantitative yield.

Compound 44d2 (47.2 mg, 70.6 mmol, 1 eq.) is deprotected according to the procedure described previously (synthesis of 51) to afford compound 101d2 in the form of a white solid, with a quantitative yield.

101d1/101d2: C₁₃H₁₅F₃O₅ M = 308.25 g.mol⁻¹

NMR ¹⁹F (D₂O, 282.5 MHz) :

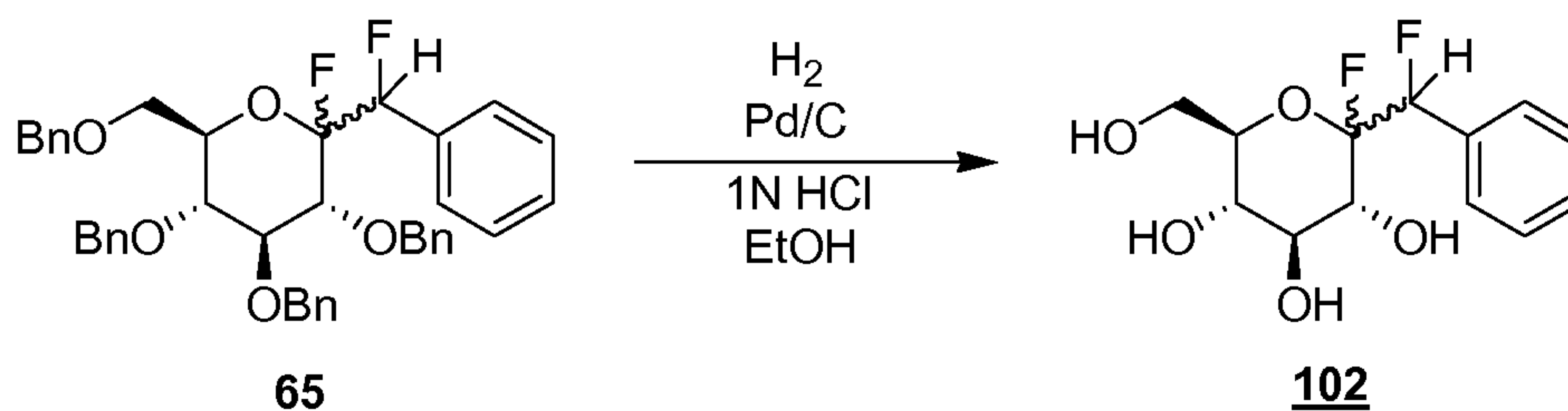
101d1: -108.4 (dd, J₁ = 5 Hz, J₂ = 259 Hz, 1F); -109.4 (d, J = 259 Hz, 1F); -142.9 (dd, J₁ = 5 Hz, J₂ = 23 Hz, 1F).

101d2: -102.0 (dd, J₁ = 8 Hz, J₂ = 264 Hz, 1F); -107.2 (dd, J₁ = 9 Hz, J₂ = 264 Hz, 1F) ; -113.0 (brd, J = 9Hz, 1F).

Mass (ESI+): 326.07 (M+H₂O); 331.13 (M+Na) 101d1.

326.07 (M+H₂O); 331.03 (M+Na) 101d2.

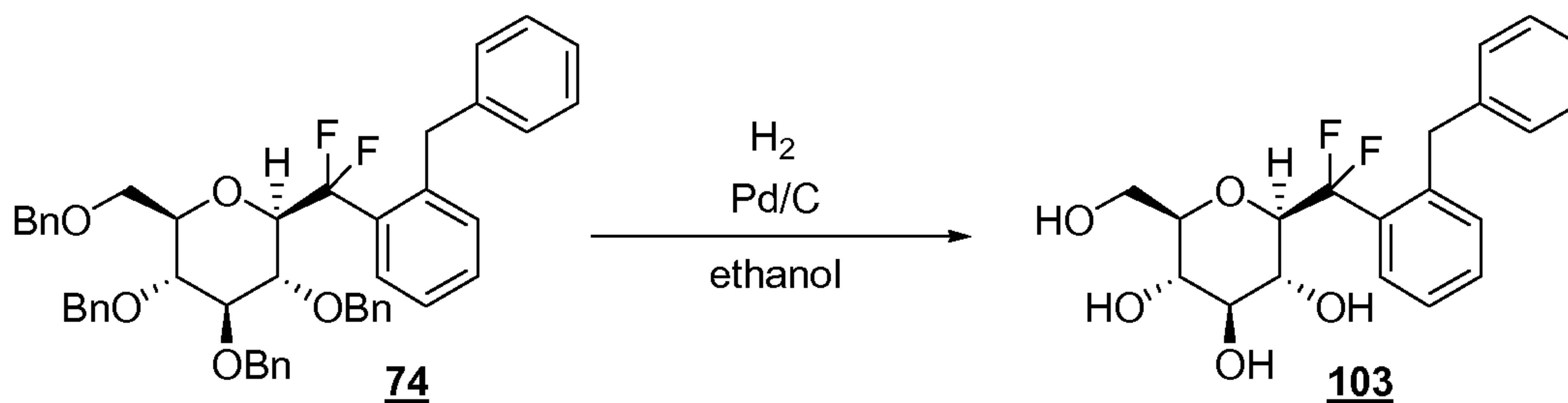
Synthesis of compound 102



Compound 65 (36.4 mg, 0.06 mmol, 1 eq.) is deprotected according to the procedure described previously (synthesis of 51) to afford compound 102 in the form of a yellow oil, with a quantitative yield.

102: C₁₃H₁₆F₂O₅ M = 290.26 g.mol⁻¹

Mass (ESI-): 287.0-289.0-291.0 (M-H); 323.0-325.0-327.0 (M+Cl).

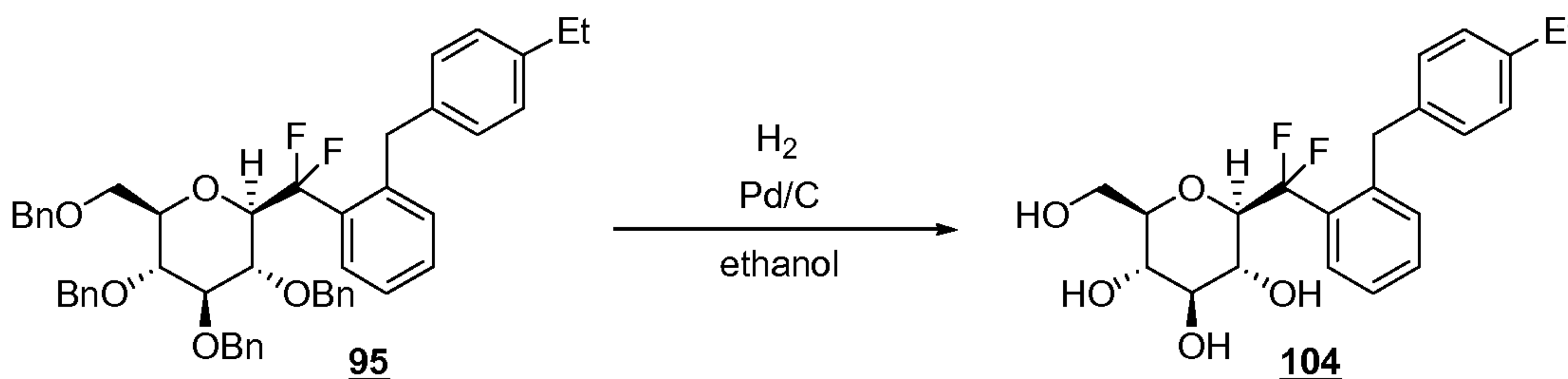
Synthesis of compound 103

Compound **74** (5.70 mg, 0.008 mmol, 1eq.) is deprotected according to the procedure described previously (synthesis of **47A-47-B**, first process) to afford compound **103** in the form of a colourless oil with a yield of 50%.

103: $\text{C}_{20}\text{H}_{22}\text{F}_2\text{O}_5$ $M = 380.38 \text{ g.mol}^{-1}$

$\text{NMR } ^{19}\text{F} (\text{MeOD}, 282.5\text{MHz})$: -95.6 (dd, $J_1 = 5 \text{ Hz}$, $J_2 = 262 \text{ Hz}$, 1F); -104.5 (dd, $J_1 = 14 \text{ Hz}$, $J_2 = 263 \text{ Hz}$, 1F).

$\text{Mass} (\text{ESI}^-)$: 379.0 (M-H); 415.1 (M+Cl).

Synthesis of compound 104

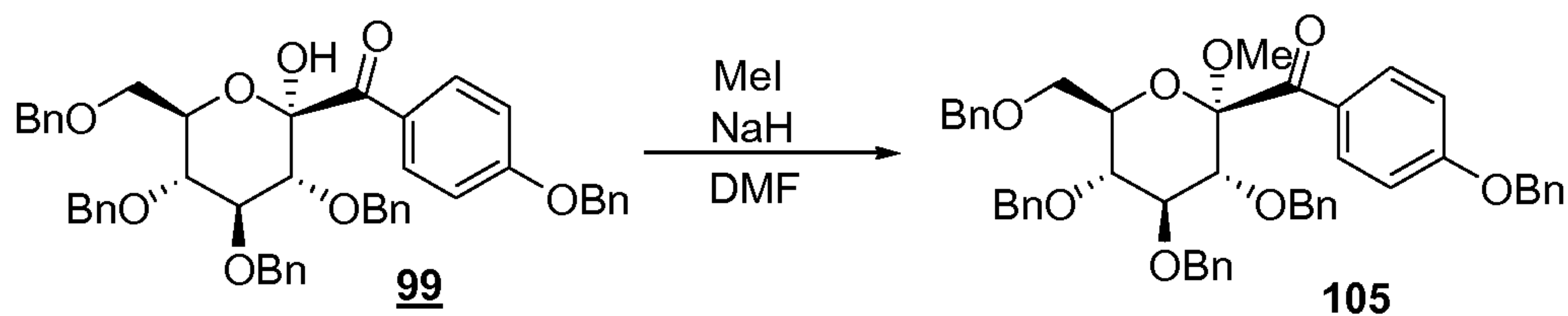
Compound **95** (0.011 mg, 0.01 mmol, 1eq.) is deprotected according to the procedure described previously (synthesis of **47A-47-B**, first process) to afford compound **104** in the form of a colourless oil with a yield of 30%.

105: $\text{C}_{22}\text{H}_{26}\text{F}_2\text{O}_5$ $M = 408.44 \text{ g.mol}^{-1}$

$\text{NMR } ^{19}\text{F} (\text{MeOD}, 282.5\text{MHz})$: -95.6 (d, $J = 262 \text{ Hz}$, 1F); -104.8 (dd, $J_1 = 14 \text{ Hz}$, $J_2 = 262 \text{ Hz}$, 1F).

$\text{Mass} (\text{ESI}^-)$: 407.1 (M-H); 443.1 (M+Cl).

Synthesis of compound 105

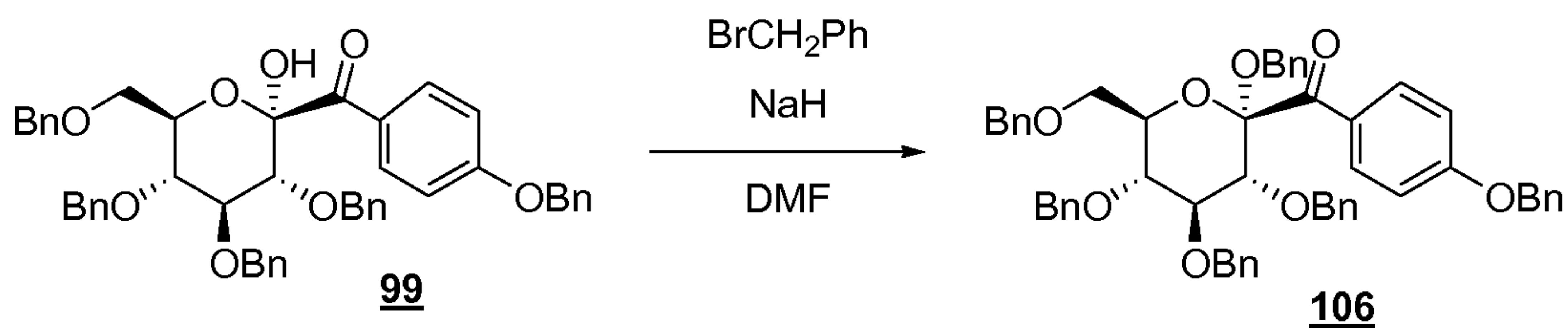


Methyliodide (0,028mL ; 0,45 mmol ; 1,5eq.) is added to a solution of compound **99** (225mg ; 0,30mmol ; 1eq.) in dimethylformamide (DMF) (2mL). NaH 95% (38,0mg ; 1,50mmol ; 5eq.) is added in one portion, and the media is stirred at room temperature during 30 minutes. A solution of chlorhydric acid 1M is then slowly added. Ethylacetate is then added and the organic phase is washed three times with water, then with brine. The organic layer is dried on MgSO₄, filtered and then concentrated. The residue is then purified on chromatography column (98/2 to 80/20 cylohexane/ethyl acetate eluent) to produce 105 in the form of a colourless oil with a yield of 89%.

105 : C₄₉H₄₈O₈ M=764,90g.mol⁻¹

Mass (ESI+) : 787.40 (M+Na) ; 1552.00 (2M+Na)

Synthesis of compound 106



15

Benzyl bromide (0.054 mL ; 0,45 mmol ; 1,5eq.) is added to a solution of compound **99** (228mg ; 0,30mmol ; 1eq.) in DMF (2mL). NaH 95% (36,4 mg ; 1,52 mmol ; 5eq.) is added in one portion, and the media is stirred at room temperature during 10 minutes. A solution of chlorhydric acid 1M is then slowly added. Ethylacetate is then added and the organic phase is washed three times with water, then with brine. The organic layer is dried on MgSO₄, filtered and then concentrated. The residue is then purified on chromatography column (98/2 to 80/20 cylohexane/ethyl acetate eluent) to produce **106** in the form of a colorless oil.

$$\underline{106} : \text{C}_{55}\text{H}_{52}\text{O}_8 \quad M=841,00\text{g.mol}^{-1}$$

Mass (ESI+) : 863.40 (M+Na) ; 1703.27 (2M+Na)

2. Biological activity

The compounds of the invention have been tested for their ability to inhibit **Human Sodium Glucose Co-Transporter 2 (Human SGLT2)** according to the following protocol:

5 1. *Preparation of human SGLT1 and human SGLT2 Expression Vectors:*

Human SGLT1 (Genbank M24847) cDNA was cloned from a pCMV6 vector containing the full length human SGLT1 gene (Origene NM_000343, Cat. #: RC221312) and Human SGLT2 (Genbank M95549) cDNA was cloned from a pCMV6 vector containing the full length human SGLT2 gene (Origene
10 NM_003041, Cat. #: RC224822). The full cDNAs were subcloned independently into mammalian cell expression plasmid pSPI1 and sequenced to verify the integrity of the construct.

2. *Preparation of CHO-K1 cells stably expressing human SGLT1 and human SGLT2:*

Transfection of CHO-K1 cells was performed using 2.5ug of pSPI1-SGLT1 or
15 pSPI1-SGLT2 plasmid with about 6ul of Lipofectamin 2000 (Invitrogen, Cat. #: 11668-019) in about 1.5×10^5 CHO-K1 cells using 12-well cell culture plate (Becton Dickinson, Cat. #: 353003) in the presence of DMEM medium (Dulbecco's Modified Eagle Medium) (Gibco, Cat. #: 11885-092) containing 10% FBS (Sigma, Cat. #: F1051-500ML). Transfectants were then selected in the presence of the
20 antibiotic G418 (GIBCO, Cat. #: 11811-031) at final concentration of 750 ug/ml. Individual clones for both SGLT1 and SGLT2 were then characterized using the functional cell-based assay described below.

3. *Cell-based assay for inhibition of uptake of methyl- α -D-glucopyranoside by human SGLT1 and human SGLT2:*

25 Selected cell lines stably expressing human SGLT1 or human SGLT2 were then used for functional analysis of sodium dependent glucose uptake. Sodium-dependent D-glucose transport was determined by measuring the uptake of ^{14}C -methyl- α -D-glucopyranoside (^{14}C -AMG) with a specific activity of 250-350mCi (9.25-13.0GBq)/mmol (PerkinElmer, Cat. #: NEC659250UC). The assay buffer
30 used to assess sodium-dependent D-glucose transport was Krebs-Ringer-Henseleit (KRH) solution containing 4.7mM KCl, 1.2mM MgCl_2 , 2.2mM CaCl_2 , 10mM Hepes pH 7.4 with Tris (Sigma). For sodium (Na^+) conditions the Assay Buffer

containing 120mM NaCl (Na^+) was used to assess sodium-dependent D-glucose transport (KRH- Na^+). For sodium free conditions, KRH solution containing 120mM N-methyl-glucamine (NMG) instead of NaCl (Na^+) was used to assess sodium-independent D-glucose transport (KRH-NMG). All buffer chemicals were purchased from Sigma.

In brief, the cells were plated at a density of 40,000 cells per well in a 96-well plate in DMEM media and allowed to grow for 24 hours. Cells were subsequently washed twice (2 x 100 μ L) with KRH buffer cells containing NMG. Cells in each well were incubated with KRH- Na^+ or KRH-NMG buffer containing 5 μ Ci ^{14}C -AMG, 50 μ M AMG and treated with compounds and then incubated for 1 hour at 37°C in a CO₂ incubator. After 1 hour the labeled cells were washed two times with KRH-Na or KRH-NMG containing 50 μ M AMG. After aspiration, cells in each well were solubilized with 50 μ L of lysis buffer by placing the 96-wellplate on a plate shaker for 5 min. Scintillation cocktail (100 μ L) was added and the ^{14}C -AMG radioisotope counted in a MicroBeta Trilux (PerkinElmer).

The results obtained are shown on the following tables:

SGLT1 Data

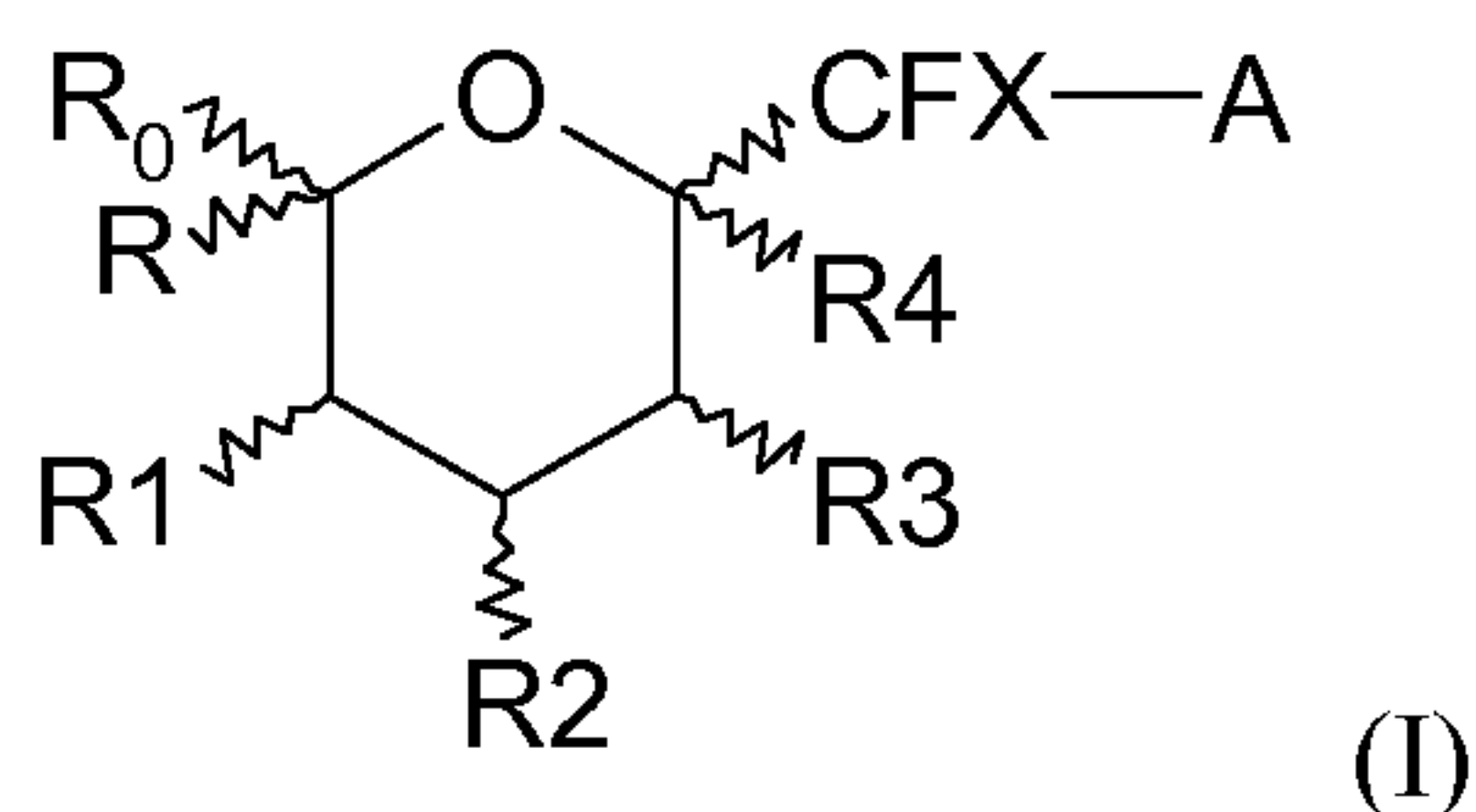
Compound	% Inhibition
Control	0
100 μ M <u>104</u>	48
100 μ M <u>100</u>	46
10 μ M <u>47</u>	11
100 μ M <u>47</u>	40

SGLT2 Data

Compound	% Inhibition
Control	0
10 μ M <u>104</u>	40
100 μ M <u>104</u>	56
10 μ M <u>100</u>	30
100 μ M <u>100</u>	79
10 μ M <u>47</u>	22
100 μ M <u>47</u>	58

CLAIMS

1. Compound of generic formula (I):

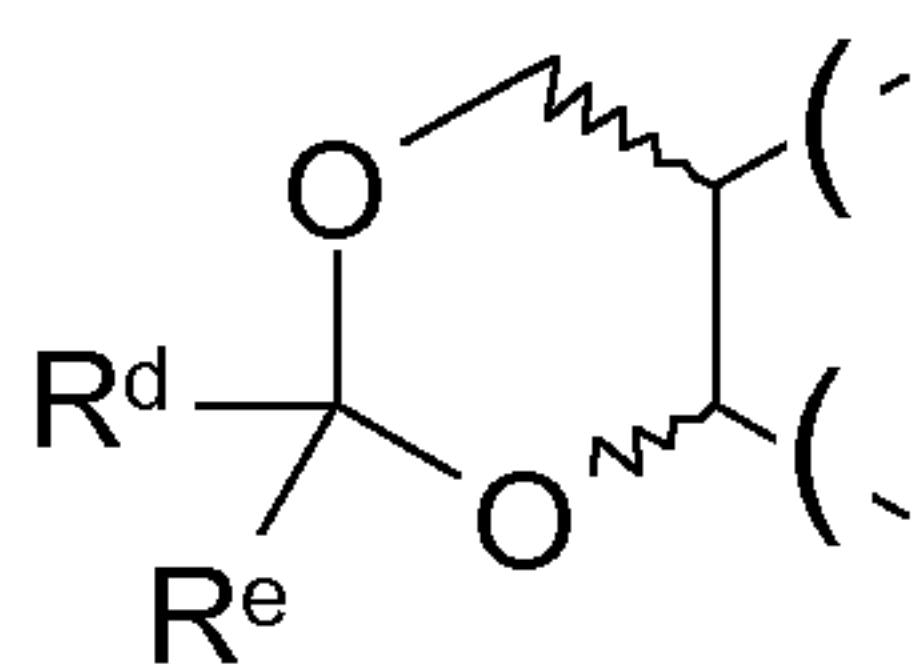


5 or a pharmaceutically acceptable salt thereof, a tautomer, an isomer or a mixture of isomers in any proportion, in particular a mixture of enantiomers, and particularly a racemate mixture,

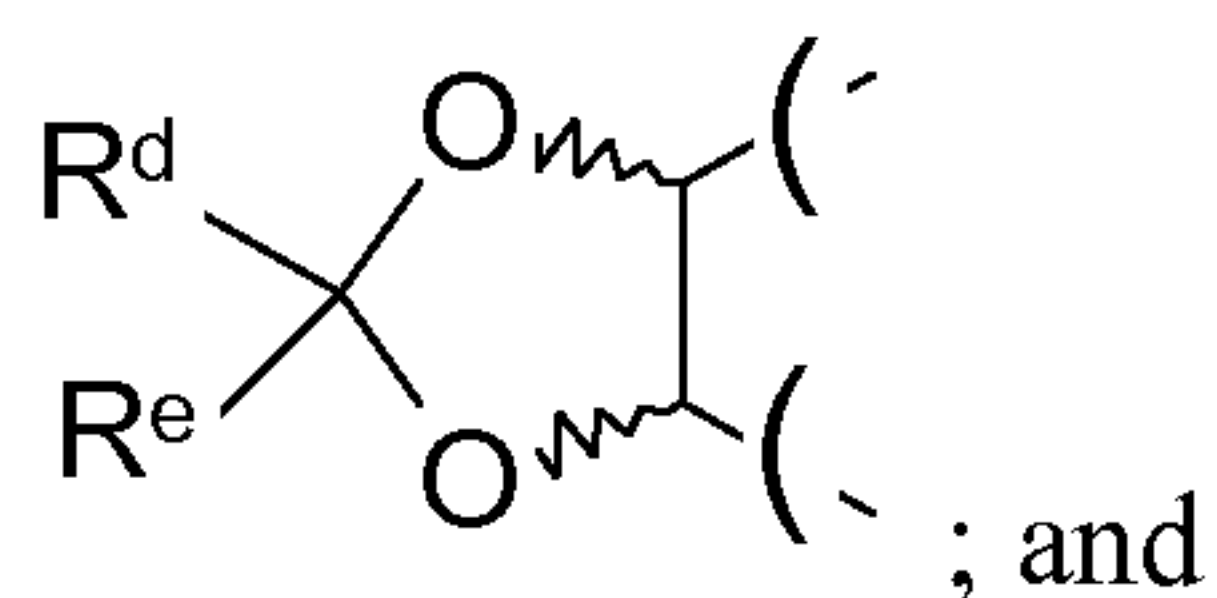
wherein:

- X represents a hydrogen or a fluorine atom;
- 10 - R represents a hydrogen or a fluorine atom or a CH₃, CH₂F, CH₂OH, CH₂OSiR^aR^bR^c, CH₂OR¹¹, CH₂OCOR¹¹, CH₂OCO₂R¹¹, CH₂OCONR¹²R¹³, CH₂OP(O)(OR¹⁴)₂ or CH₂OSO₃R¹⁴ group;
- R₁ and R₂ represent, independently from one another, a fluorine atom or an OH, OSiR^aR^bR^c, OR¹¹, OCOR¹¹, OCO₂R¹¹ or OCONR¹²R¹³ group;
- 15 - R₃ represents a hydrogen or fluorine atom or an OH, OSiR^aR^bR^c, OR¹¹, OCOR¹¹, OCO₂R¹¹, OCONR¹²R¹³, NR¹²R¹³ or NR¹²COR¹¹ group;
- R₄ represents a hydrogen atom, an halogen atom or an OH, OSiR^aR^bR^c, OR¹¹, OCOR¹¹, OCO₂R¹¹, OCONR¹²R¹³, NR¹²R¹³, (C₁-C₆)-alkyl or (C₂-C₆)-alkenyl group;
- R₀ represents a hydrogen or an halogen atom or an OH, OSiR^aR^bR^c, OR¹¹, OCOR¹¹, OCO₂R¹¹, OCONR¹²R¹³, OP(O)(OR¹⁴)₂ or OSO₃R¹⁴ group;
- 20

or R and R₁, together with the carbon atoms carrying them, form a cyclic acetal having the following formula:



and/or (R₀ and R₁), (R₁ and R₂), (R₂ and R₃), and/or (R₃ and R₄), together
 25 with the carbon atoms carrying them, form a cyclic acetal having the following formula:



- A represents an aryl, heteroaryl or aryl-(C₁-C₆)-alkyl-aryl group, possibly substituted by one or more groups chosen among an halogen atom, a CN, SO₂, SiR^aR^bR^c, (C₁-C₆)-alkyl, (C₂-C₆)-alkenyl, (C₂-C₆)-alkynyl, (C₃-C₇)-cycloalkyl, 5 to 7 ring-membered heterocycloalkyl, aryl, heteroaryl, aryl-(C₁-C₆)-alkyl, heteroaryl-(C₁-C₆)-alkyl, (C₁-C₆)-alkyl-aryl, (C₁-C₆)-alkyl-heteroaryl, OR¹¹, COR¹¹, OCOR¹¹, CO₂R¹¹, NR¹²R¹³, NR¹²COR¹¹, CONR¹²R¹³, SR¹¹, SO₂R¹¹, CSR¹¹ and OSO₃R¹¹ group,

the whole being possibly substituted by one or more groups chosen among an halogen atom, an OH, (C₁-C₆)-alkyl, (C₁-C₆)-alkoxy, COOH and CHO group;

with:

- R¹¹ representing a (C₁-C₆)-alkyl, (C₂-C₆)-alkenyl, (C₂-C₆)-alkynyl, (C₃-C₇)-cycloalkyl, 5 to 7 ring-membered heterocycloalkyl, aryl, aryl-(C₁-C₆)-alkyl or (C₁-C₆)-alkyl-aryl group, this group being possibly substituted by one or more groups chosen among an halogen atom, an OH, COOH and CHO group;

- R¹² and R¹³ representing, independently from one another, a hydrogen atom or a (C₁-C₆)-alkyl or aryl-(C₁-C₆)-alkyl group;

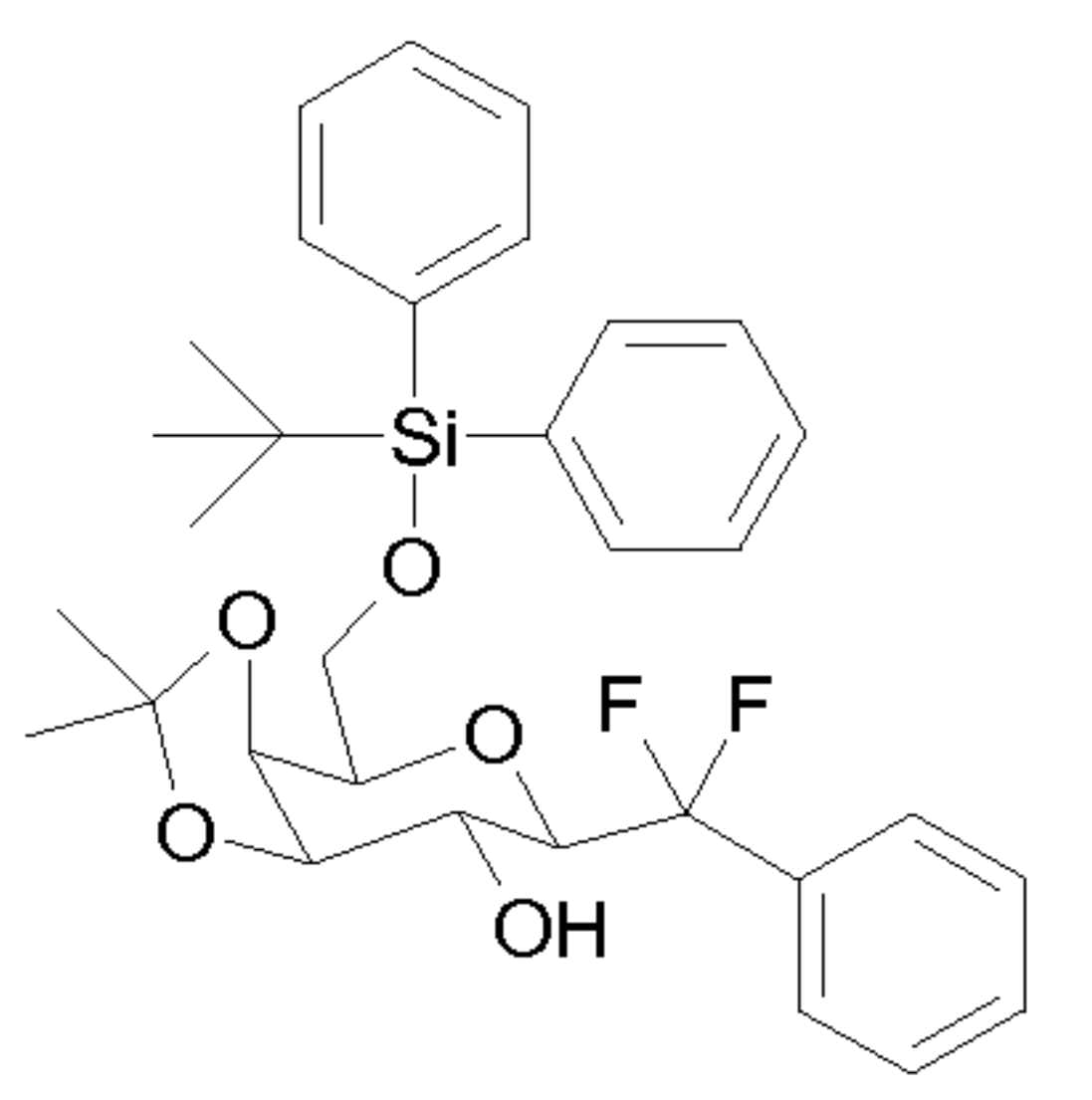
- R¹⁴ representing a hydrogen atom or a (C₁-C₆)-alkyl group;

- R^a, R^b and R^c representing, independently from one another, a (C₁-C₆)-alkyl, aryl or aryl-(C₁-C₆)-alkyl group; and

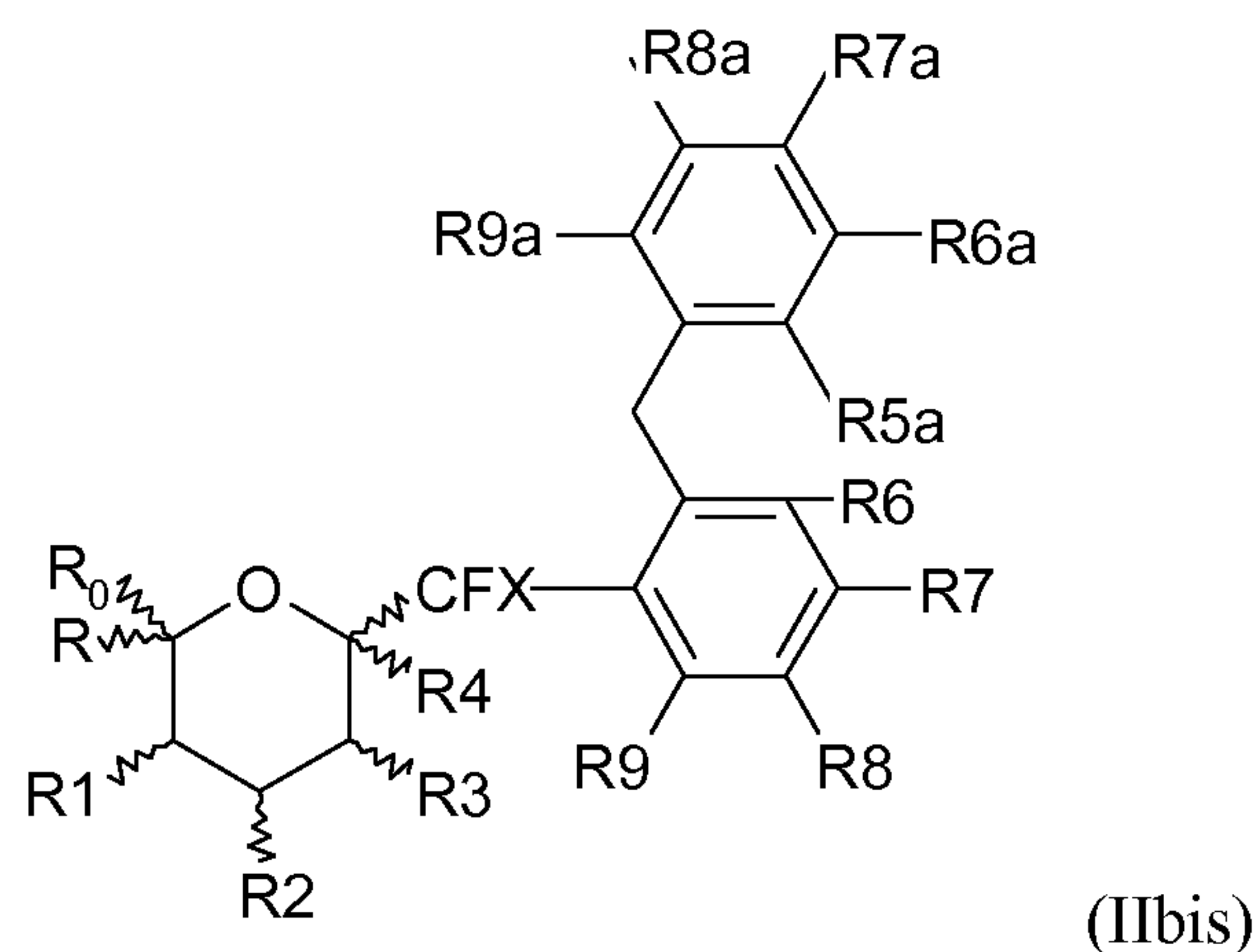
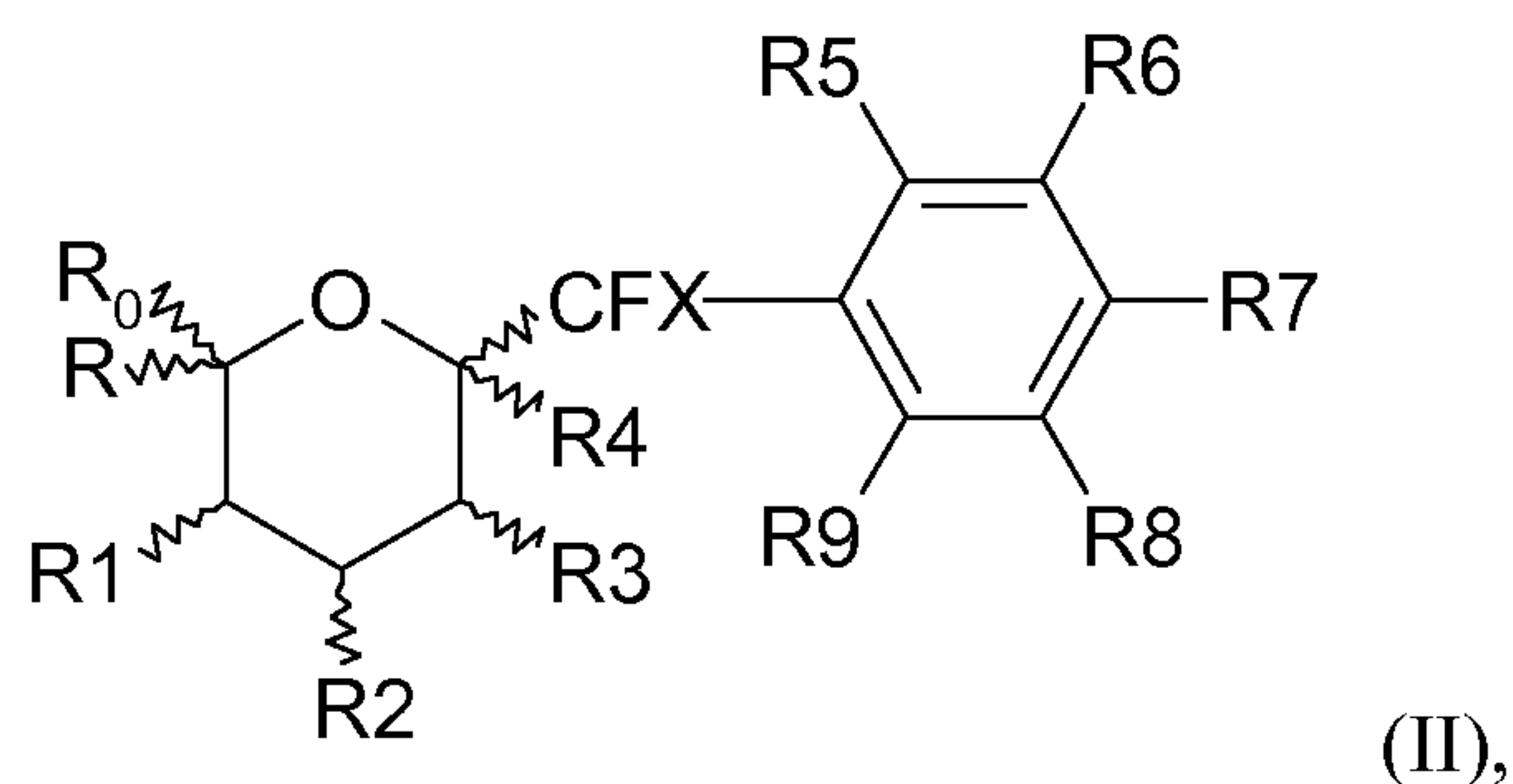
- R^d and R^e representing, independently from one another, a hydrogen atom or a (C₁-C₆)-alkyl group;

with the proviso that when R_0 is different from a hydrogen atom, then R_4 represents a hydrogen atom, and

with the proviso that the compound of formula (I) is not the following compound:



2. Compound according to claim 1, characterised in that it responds to the following generic formula (II) or (IIbis):

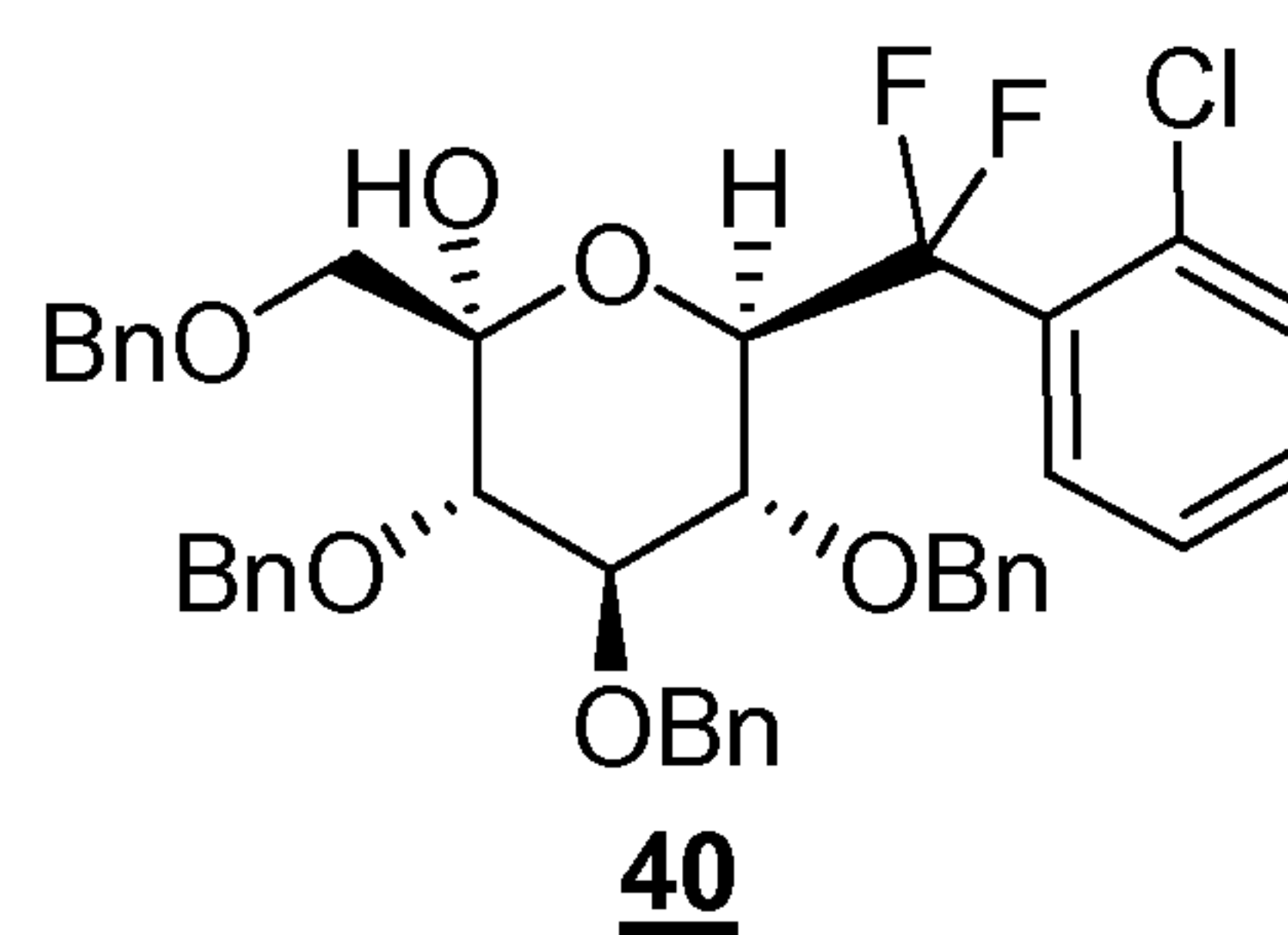
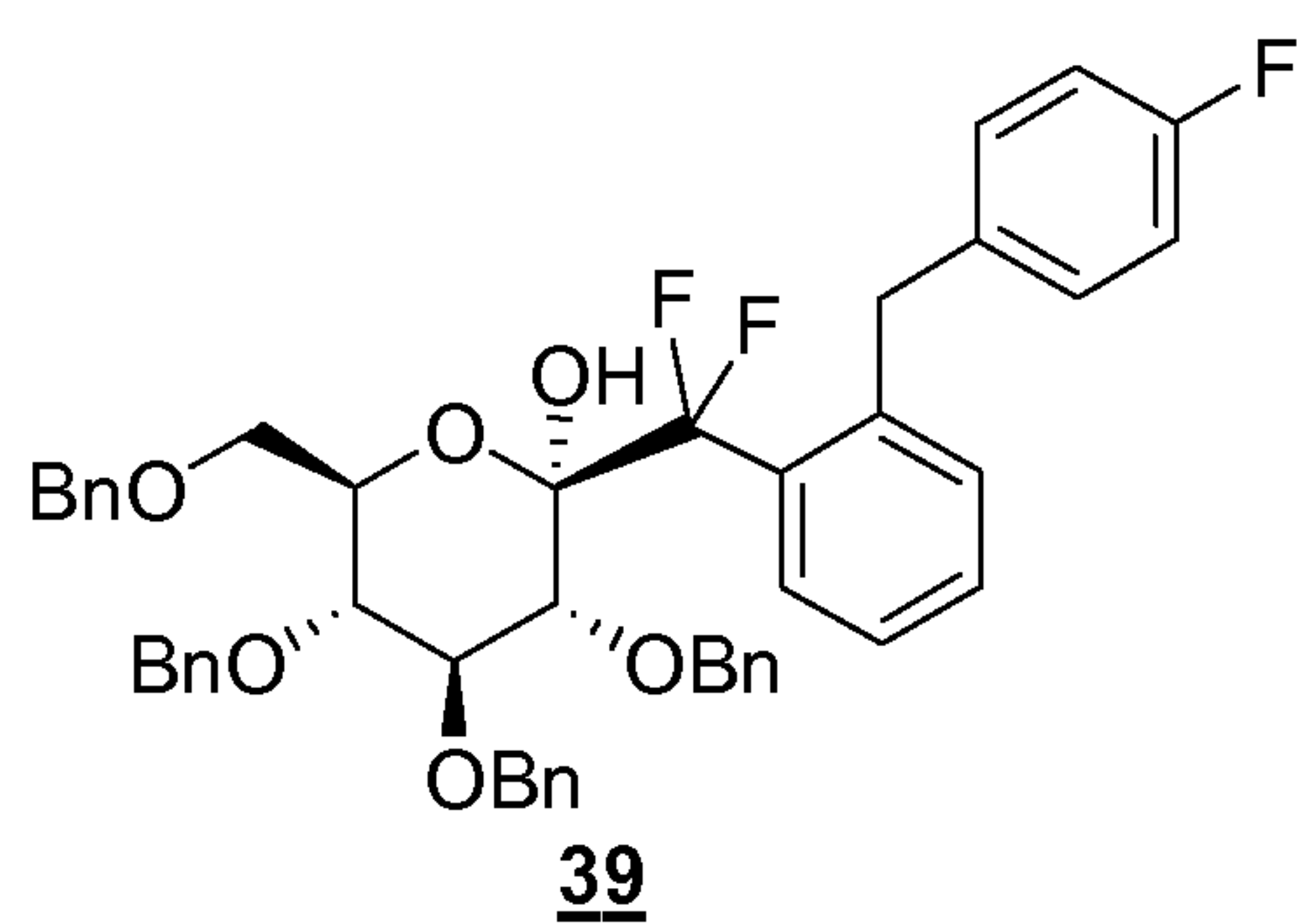
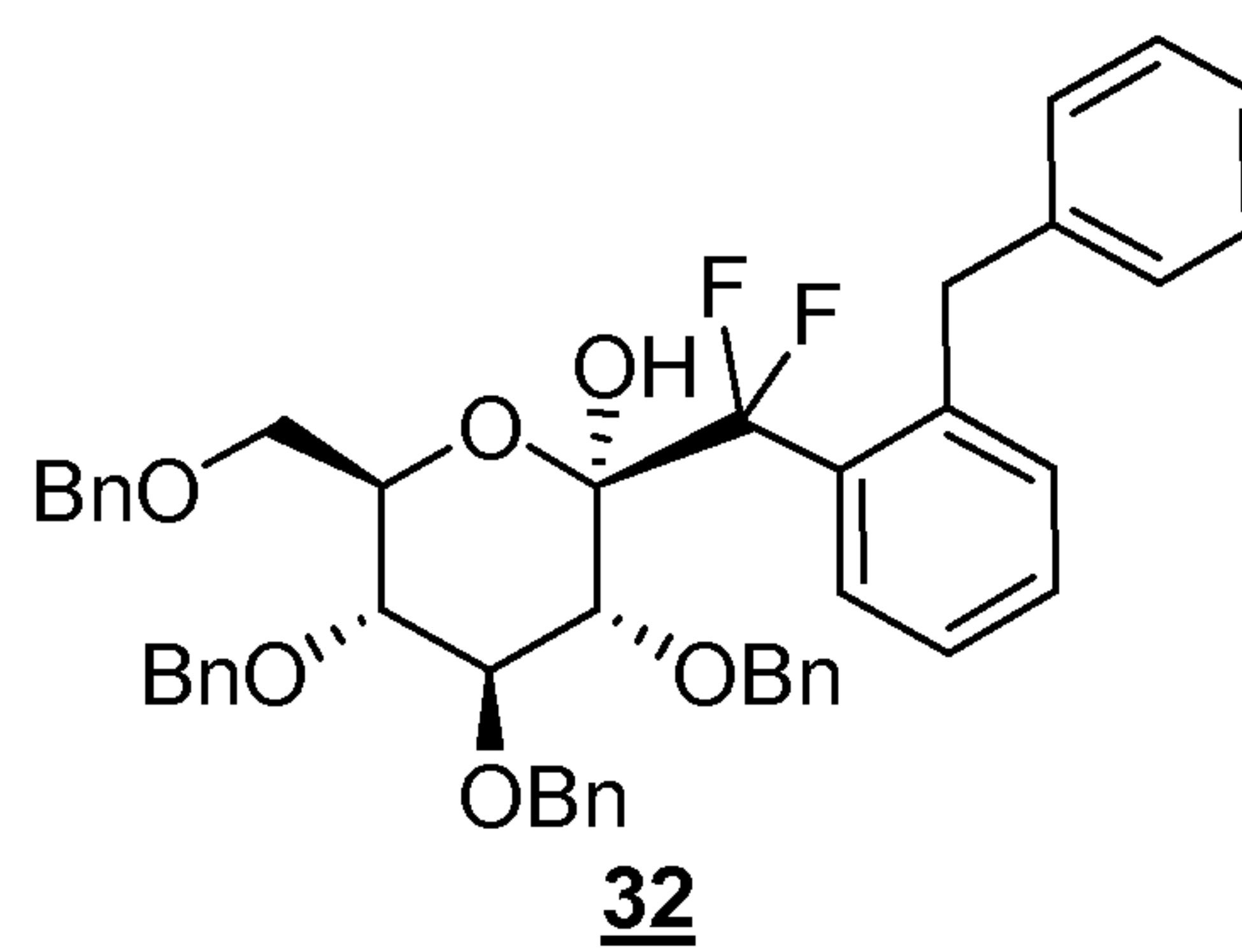
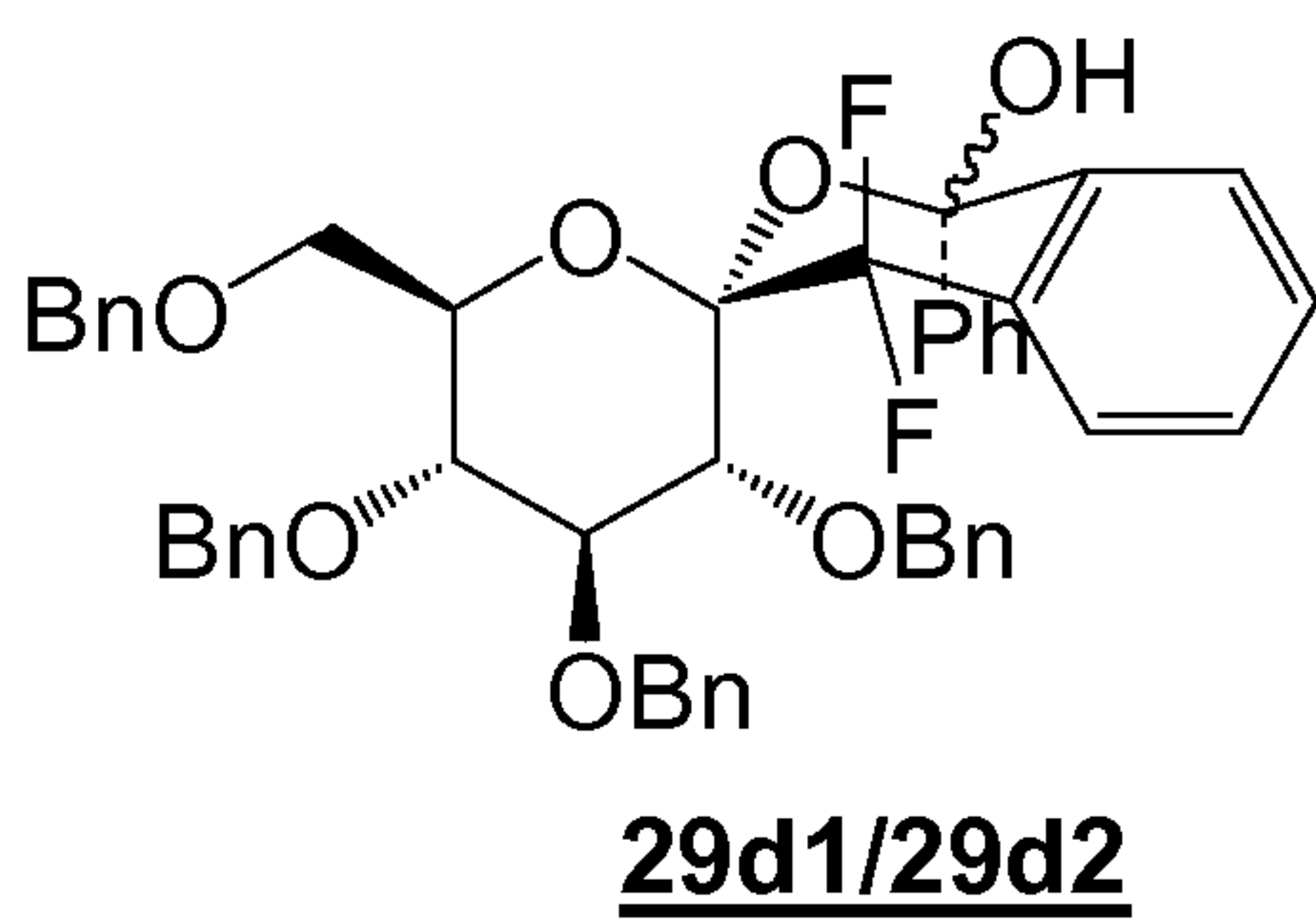
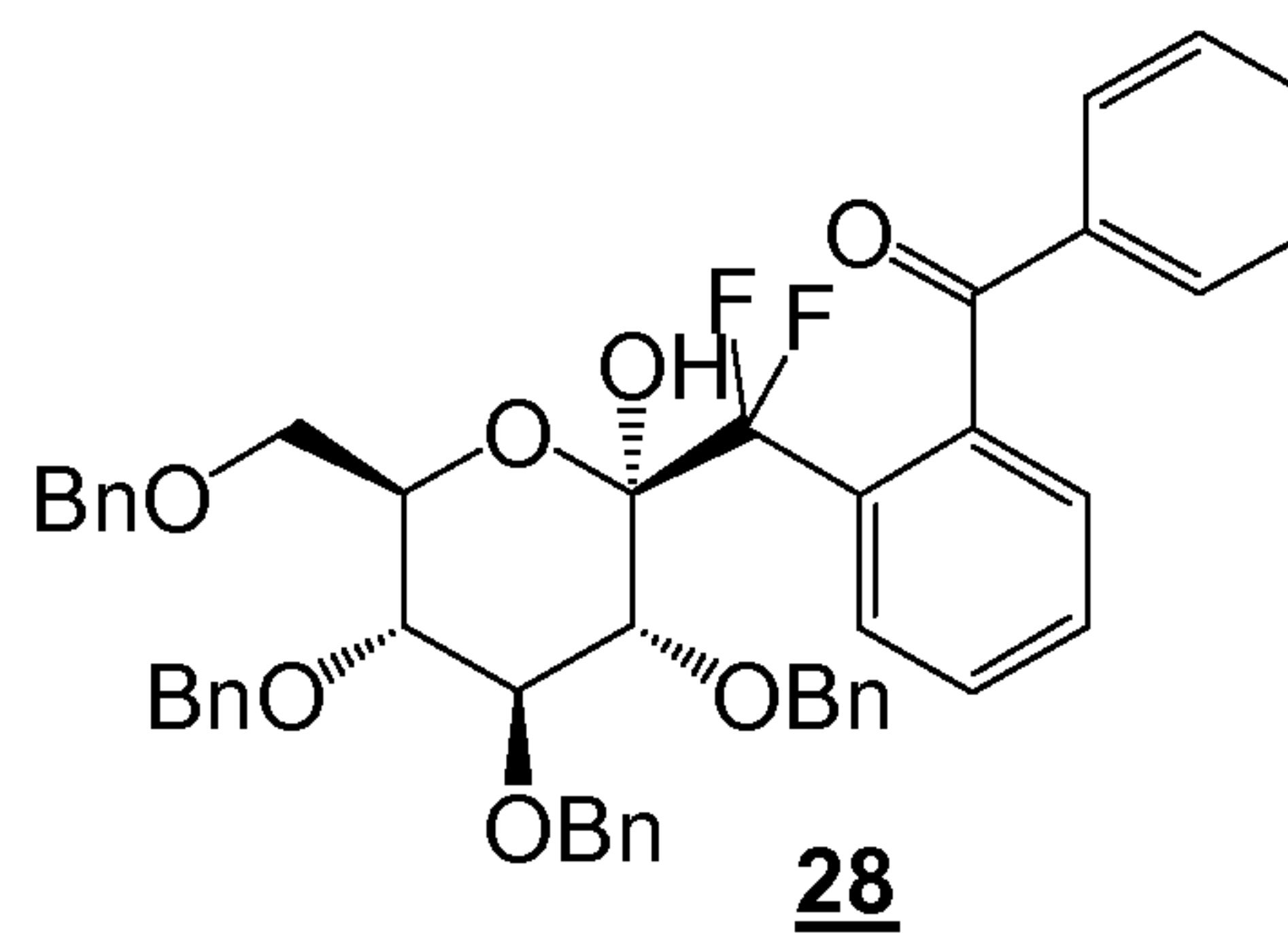
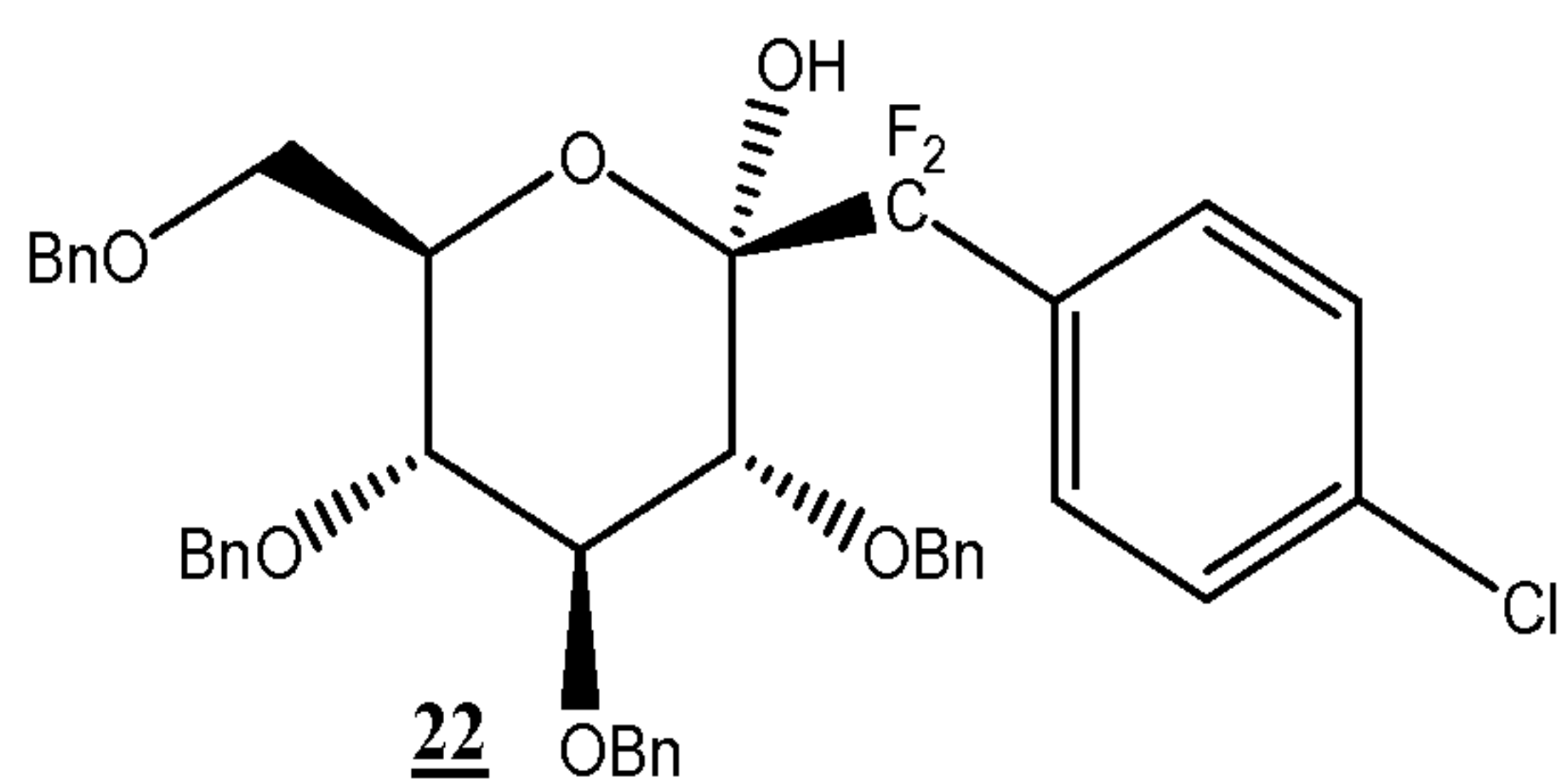
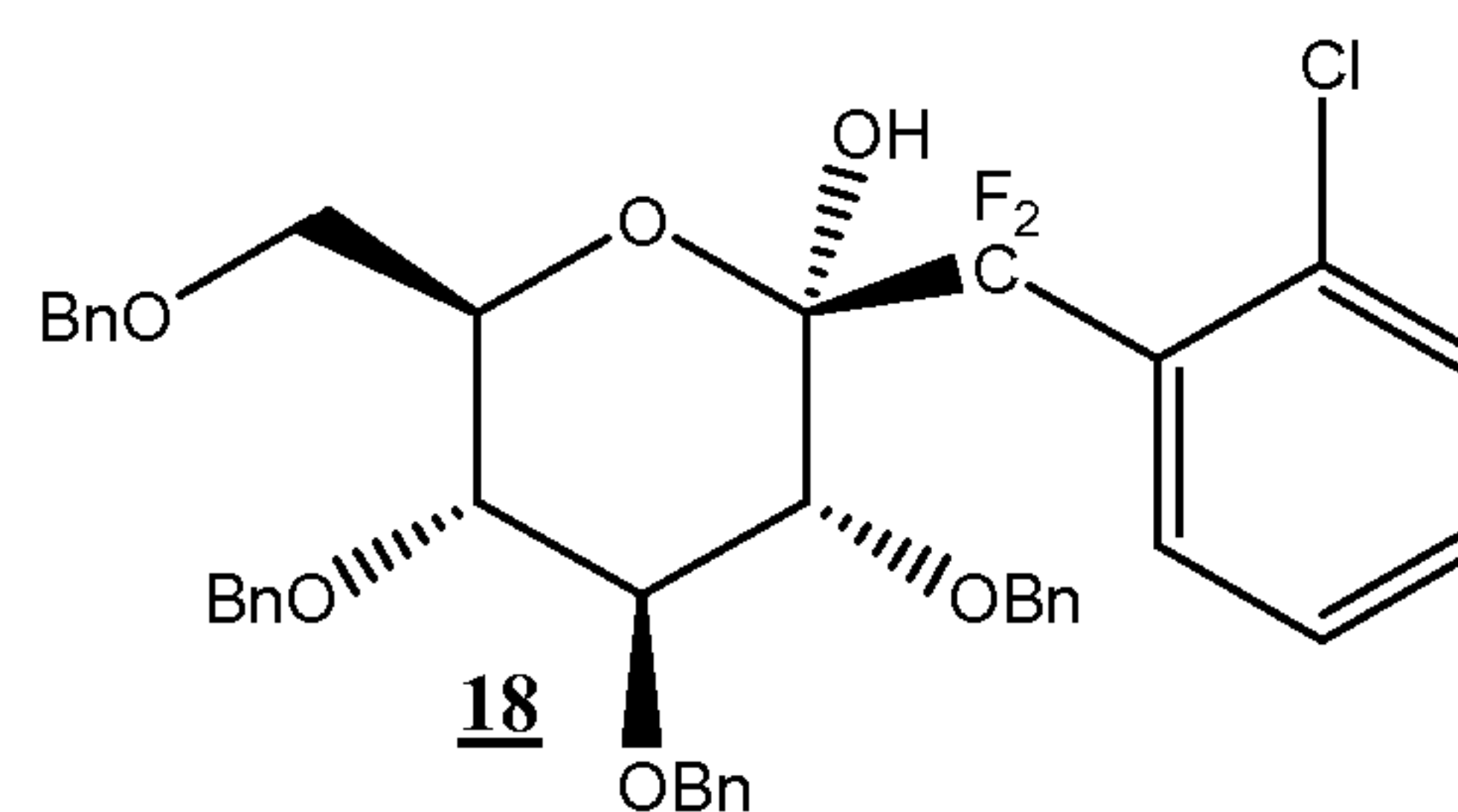
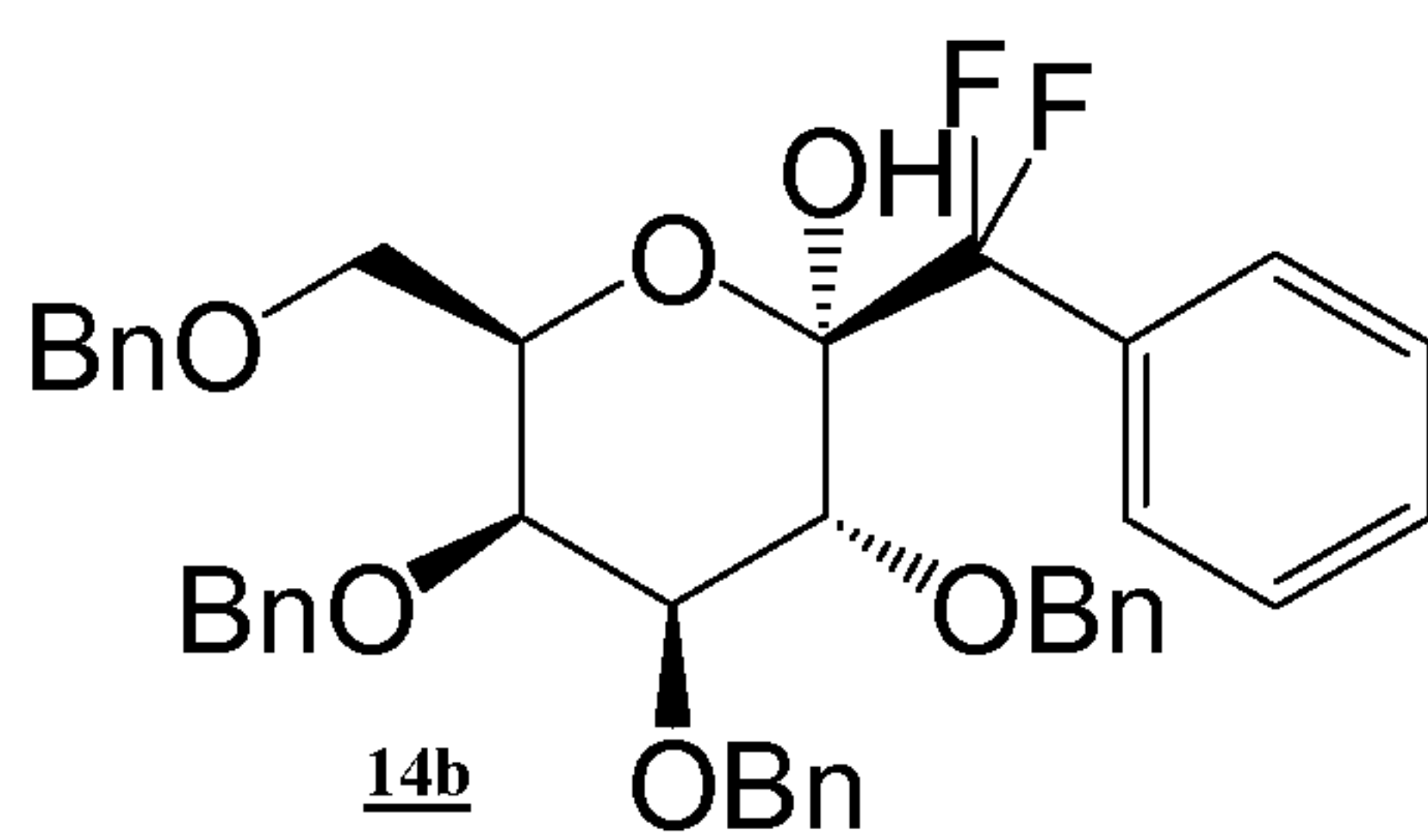
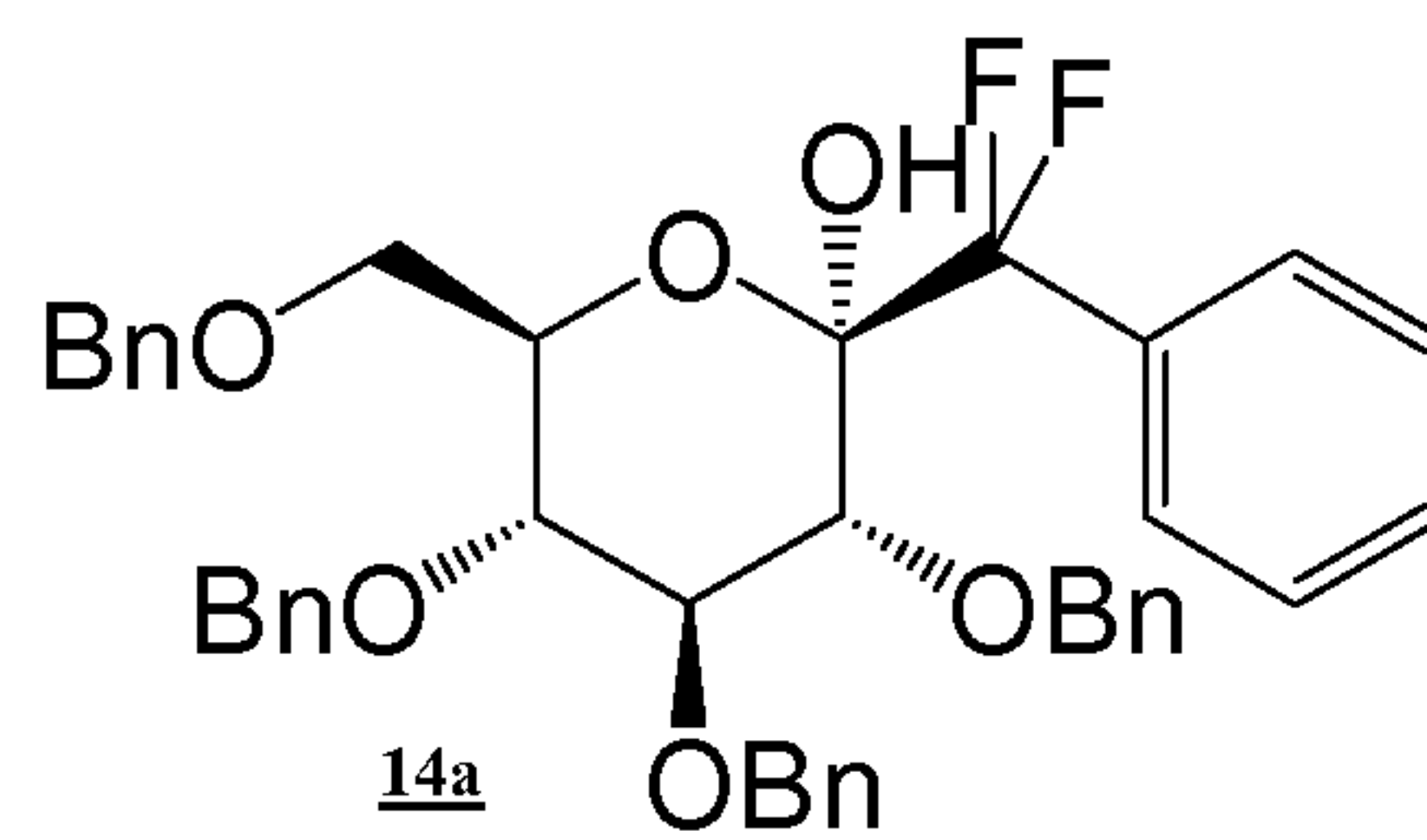
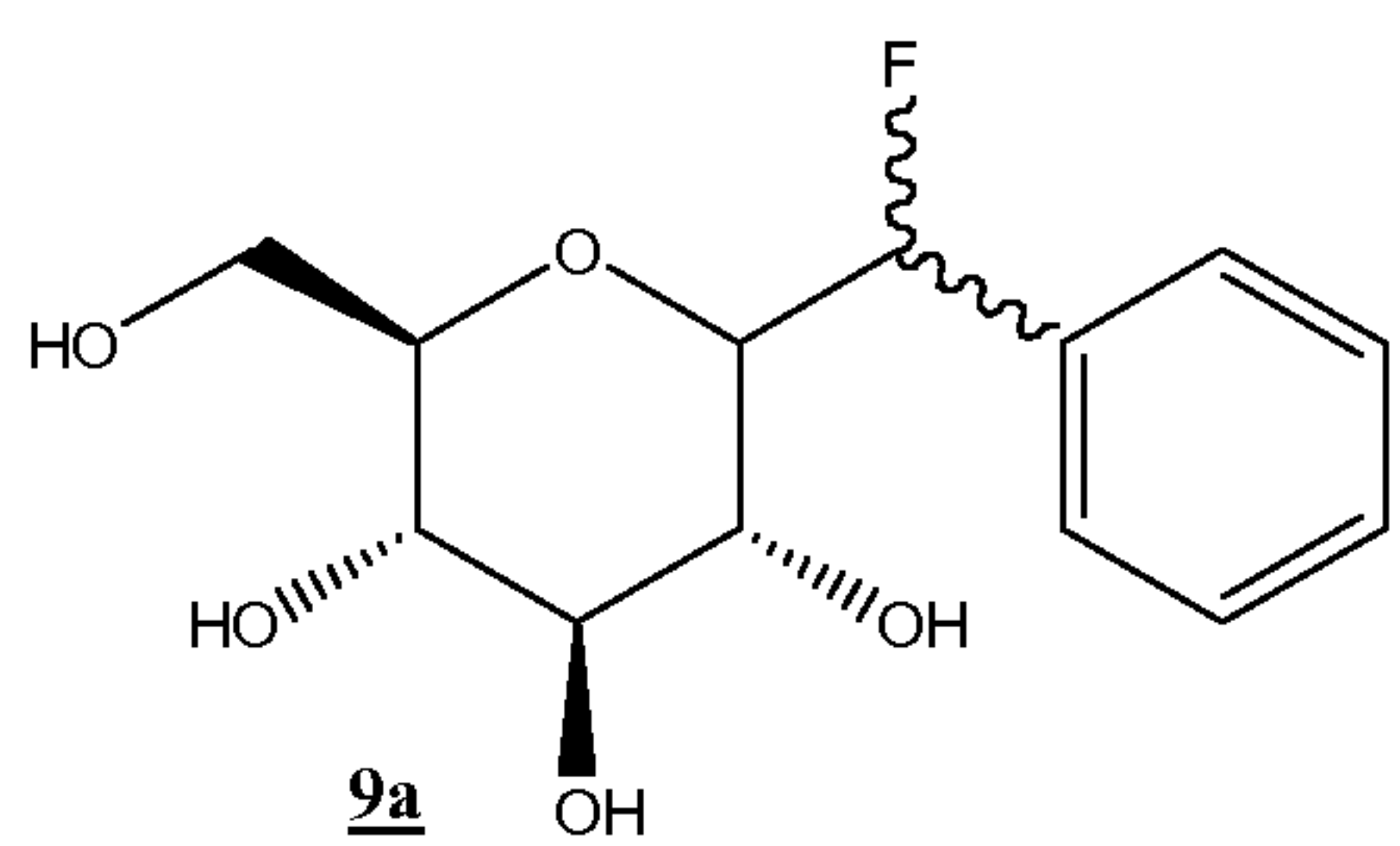


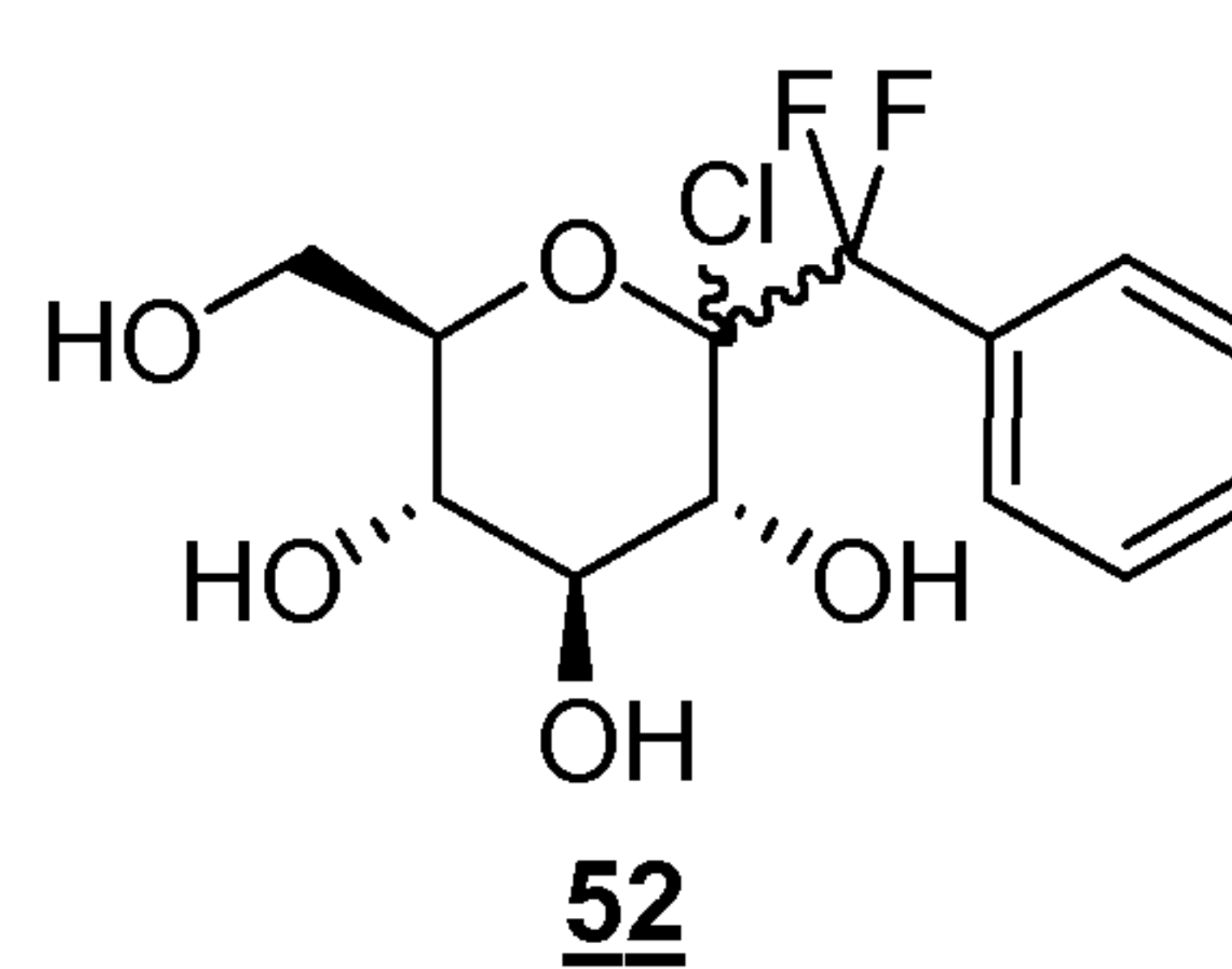
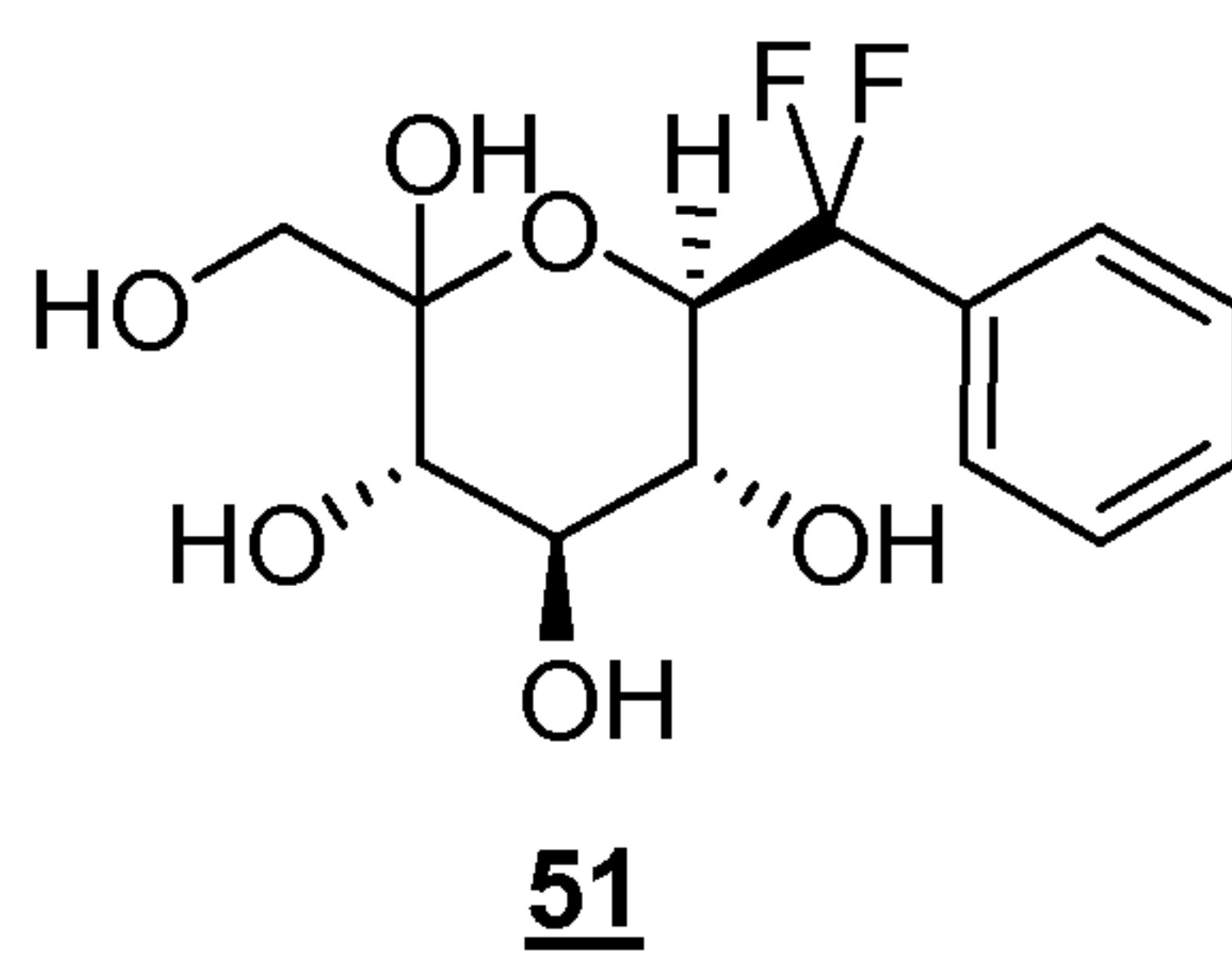
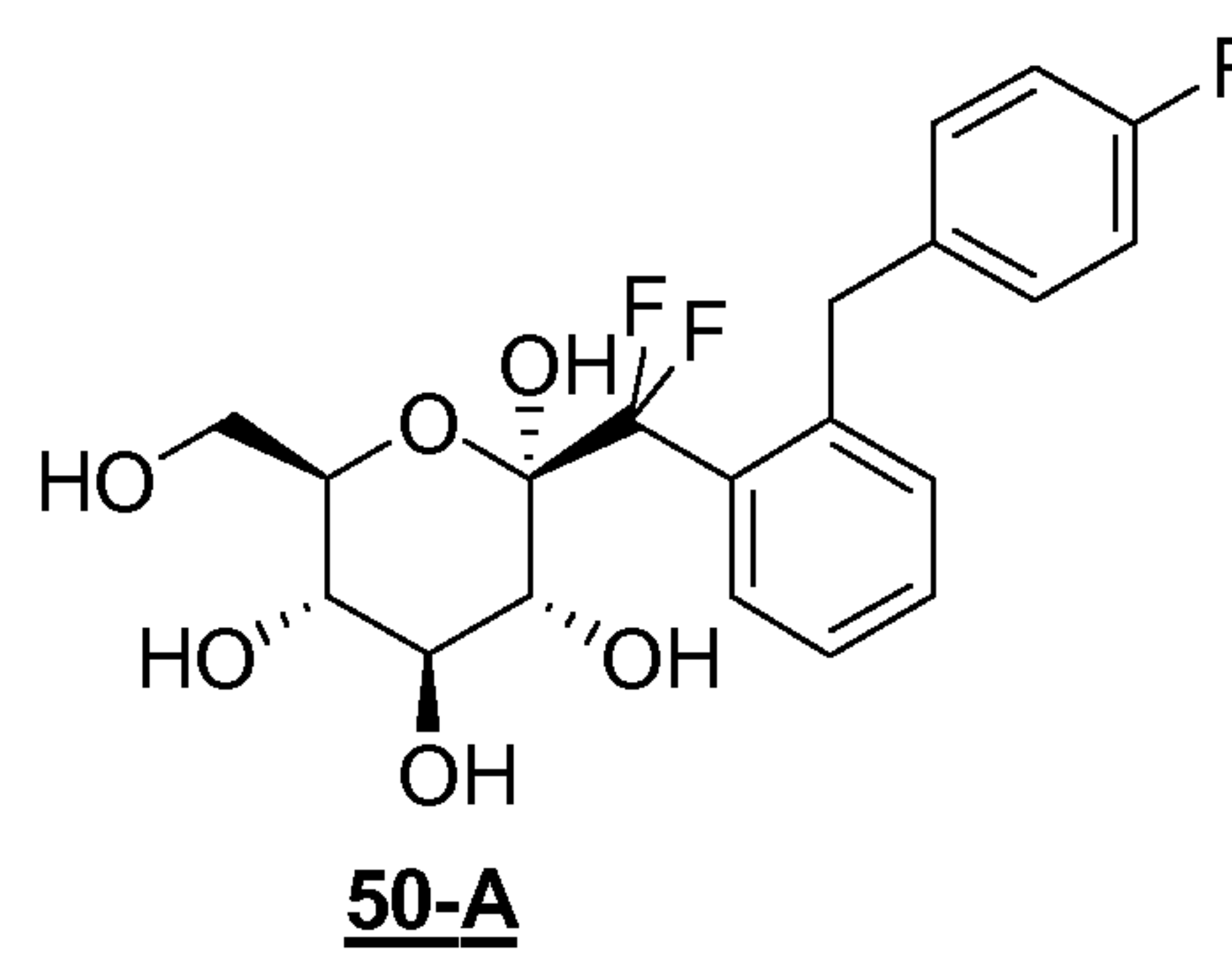
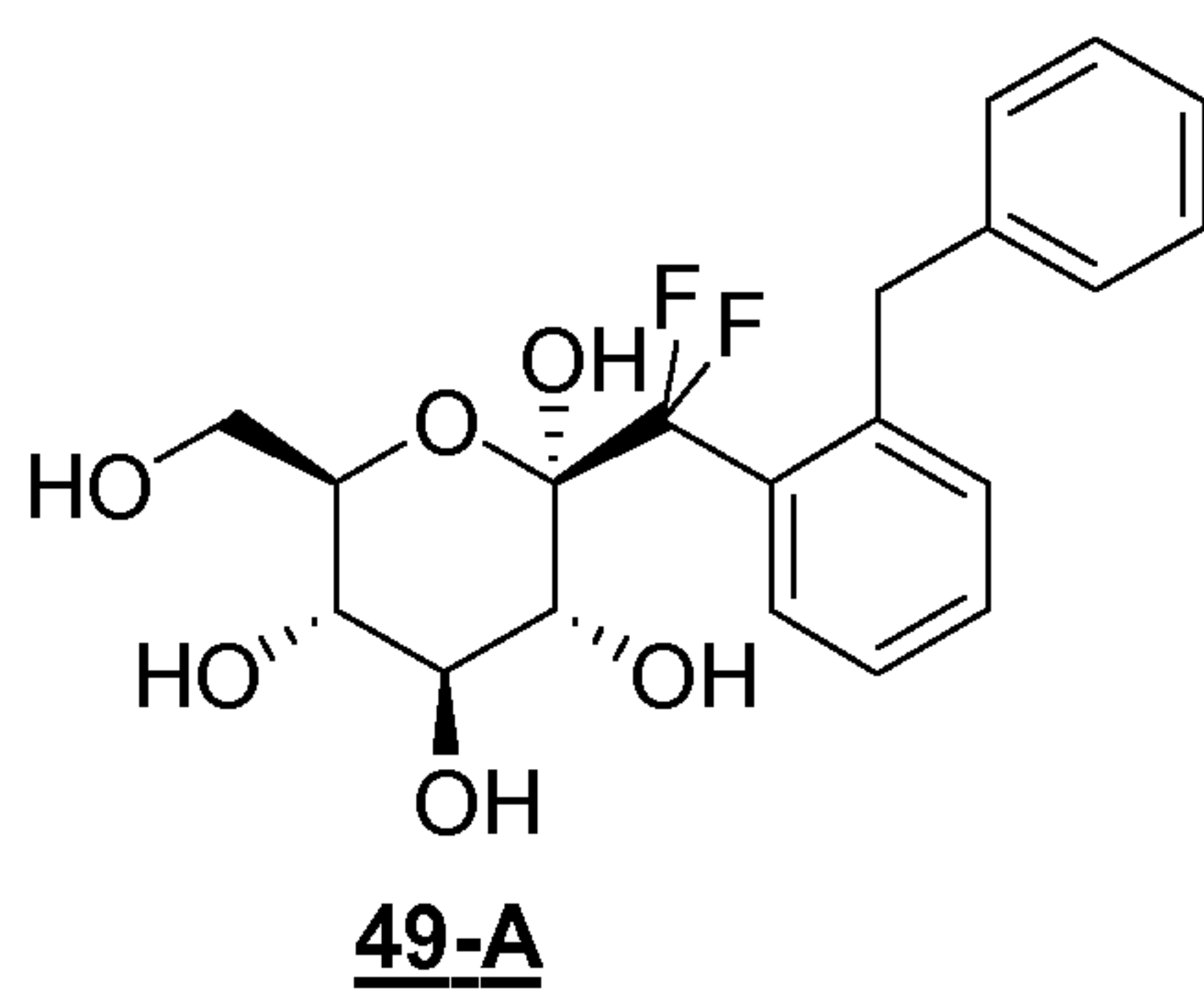
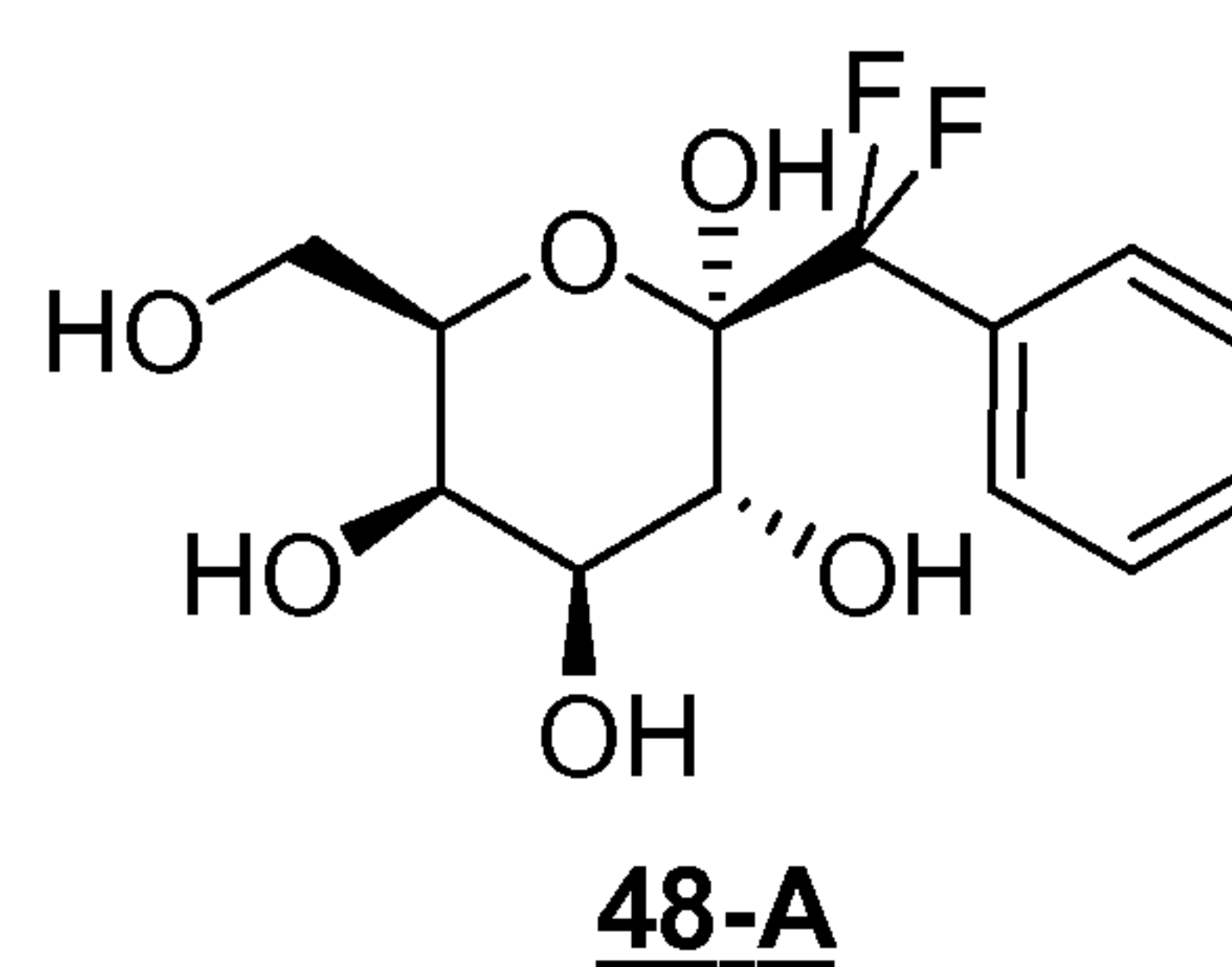
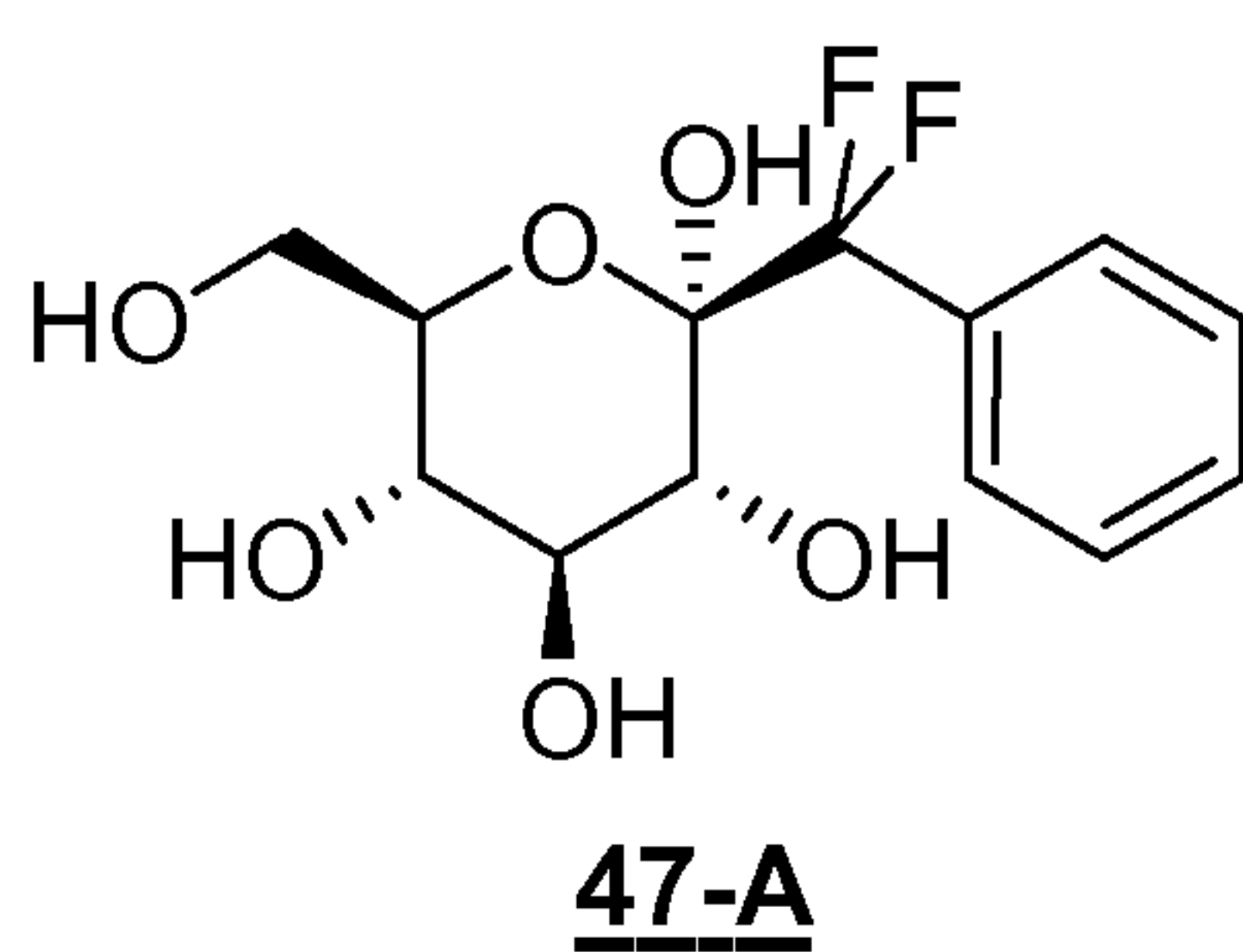
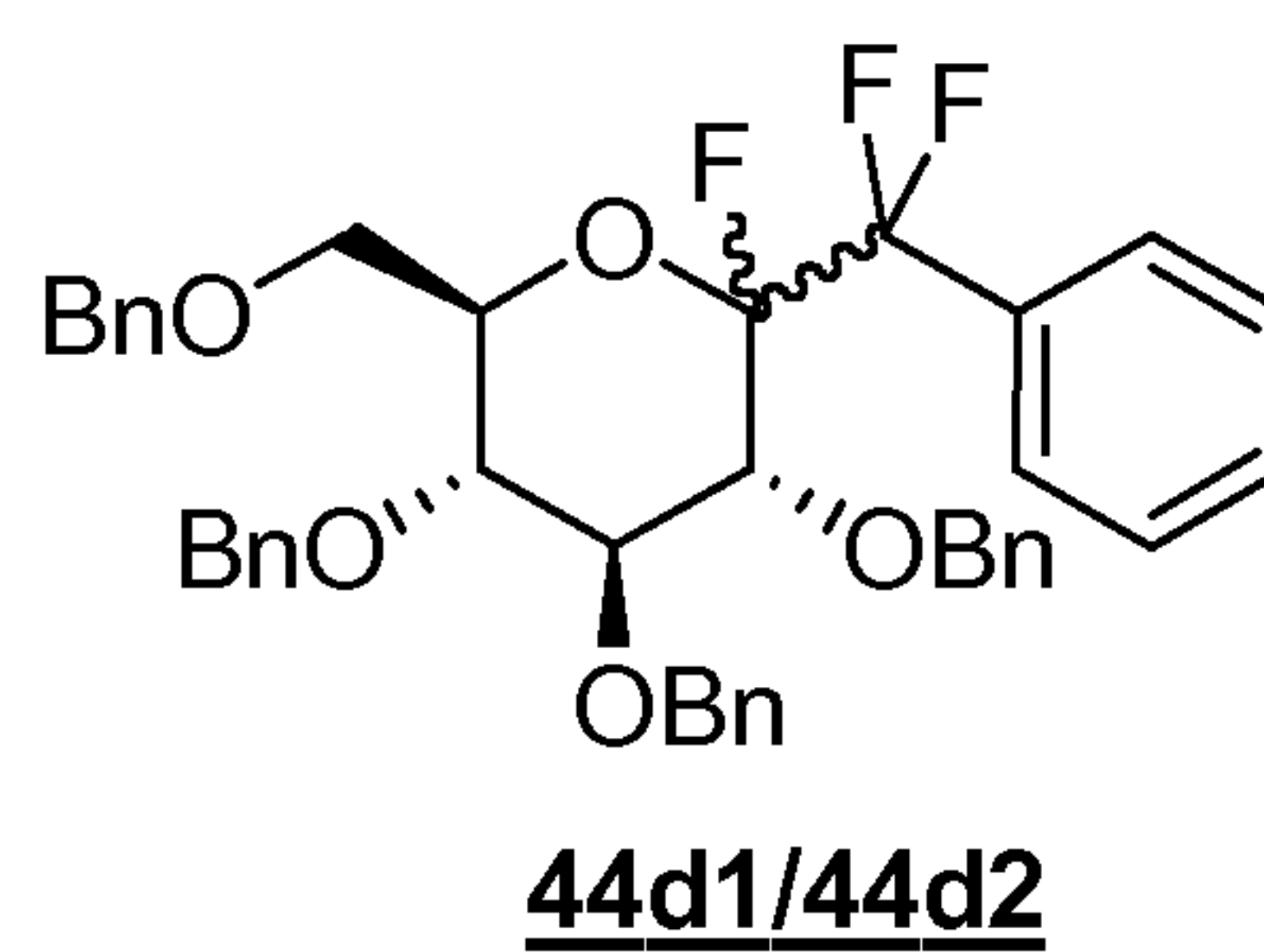
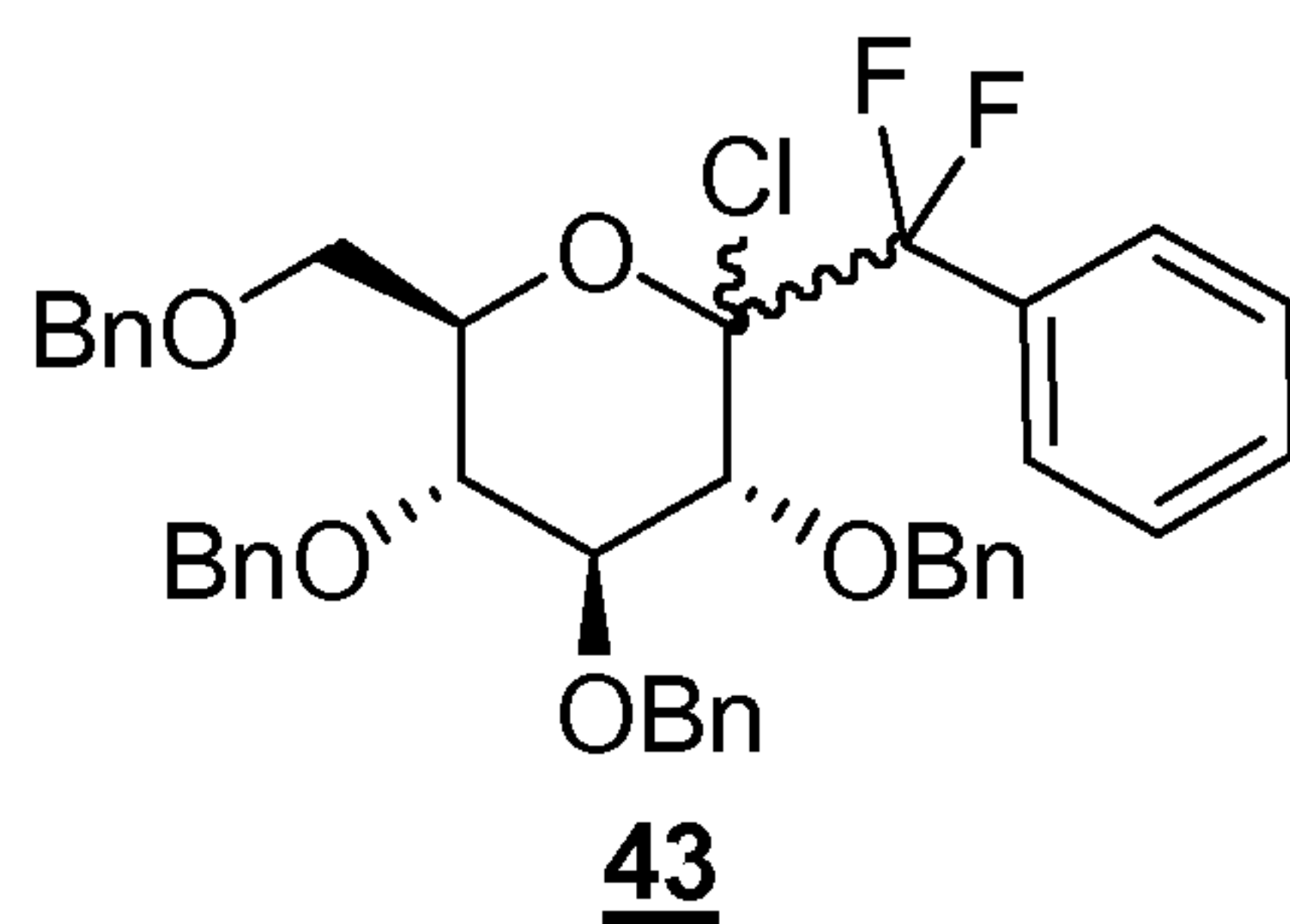
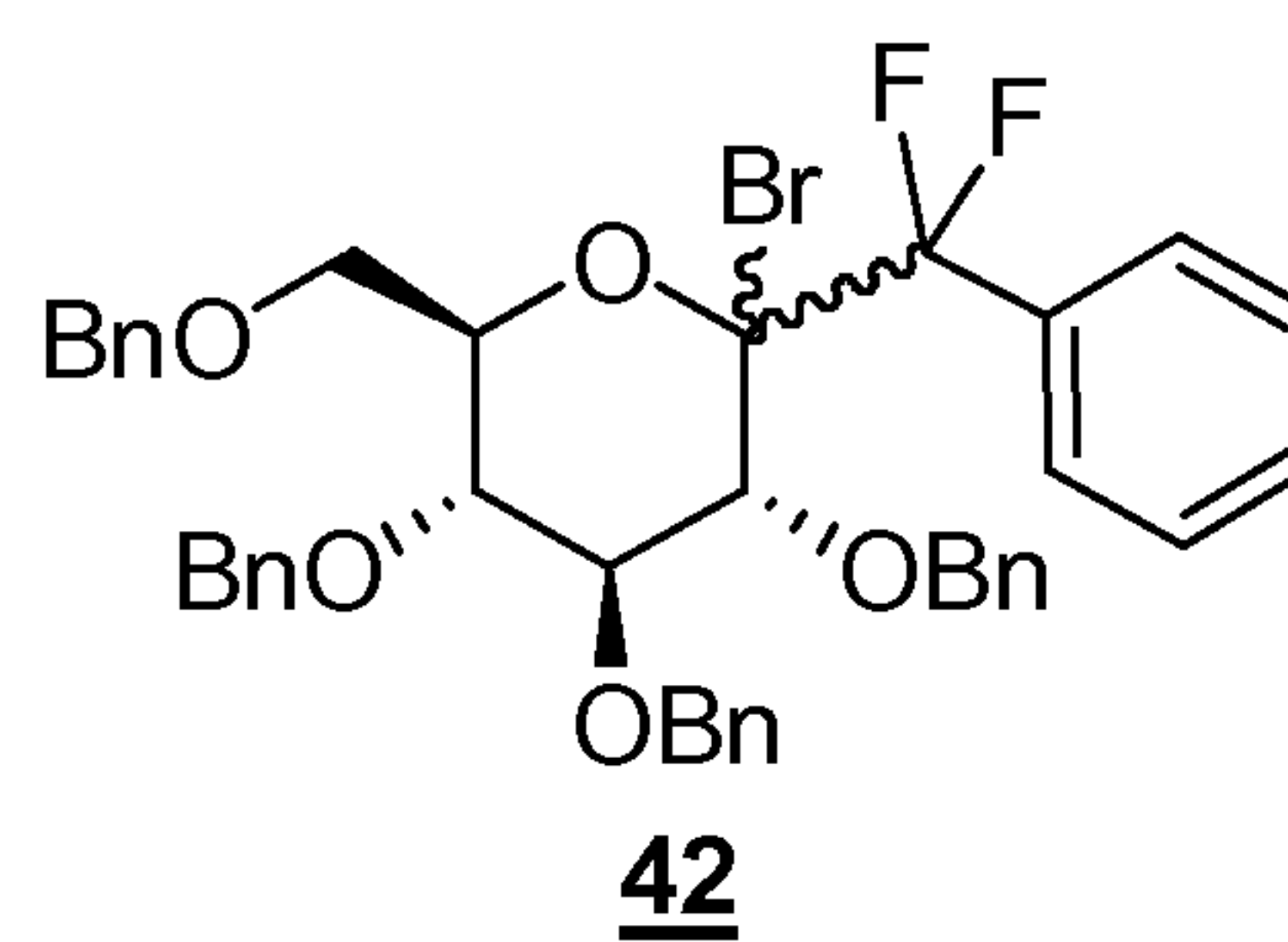
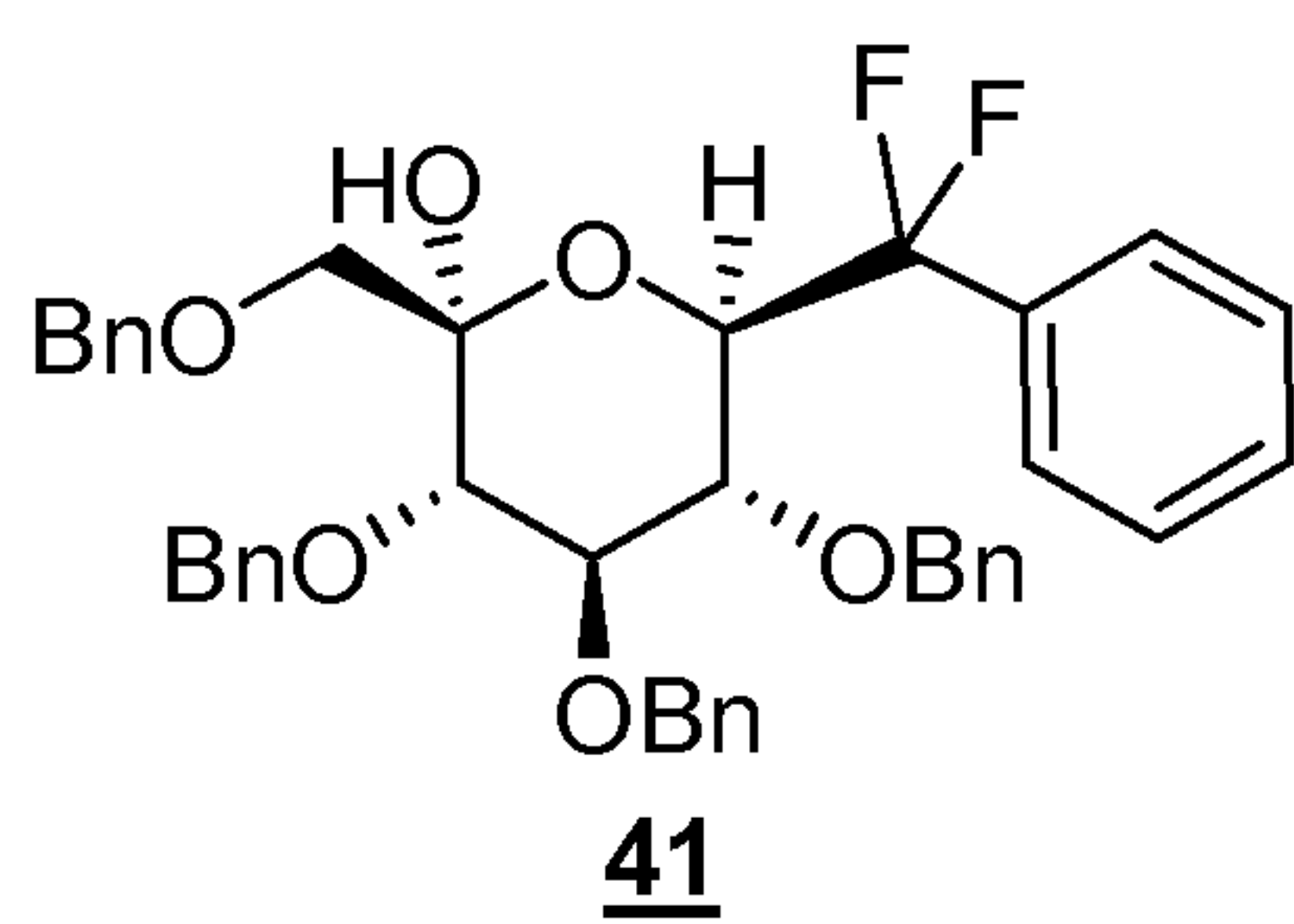
5 wherein:

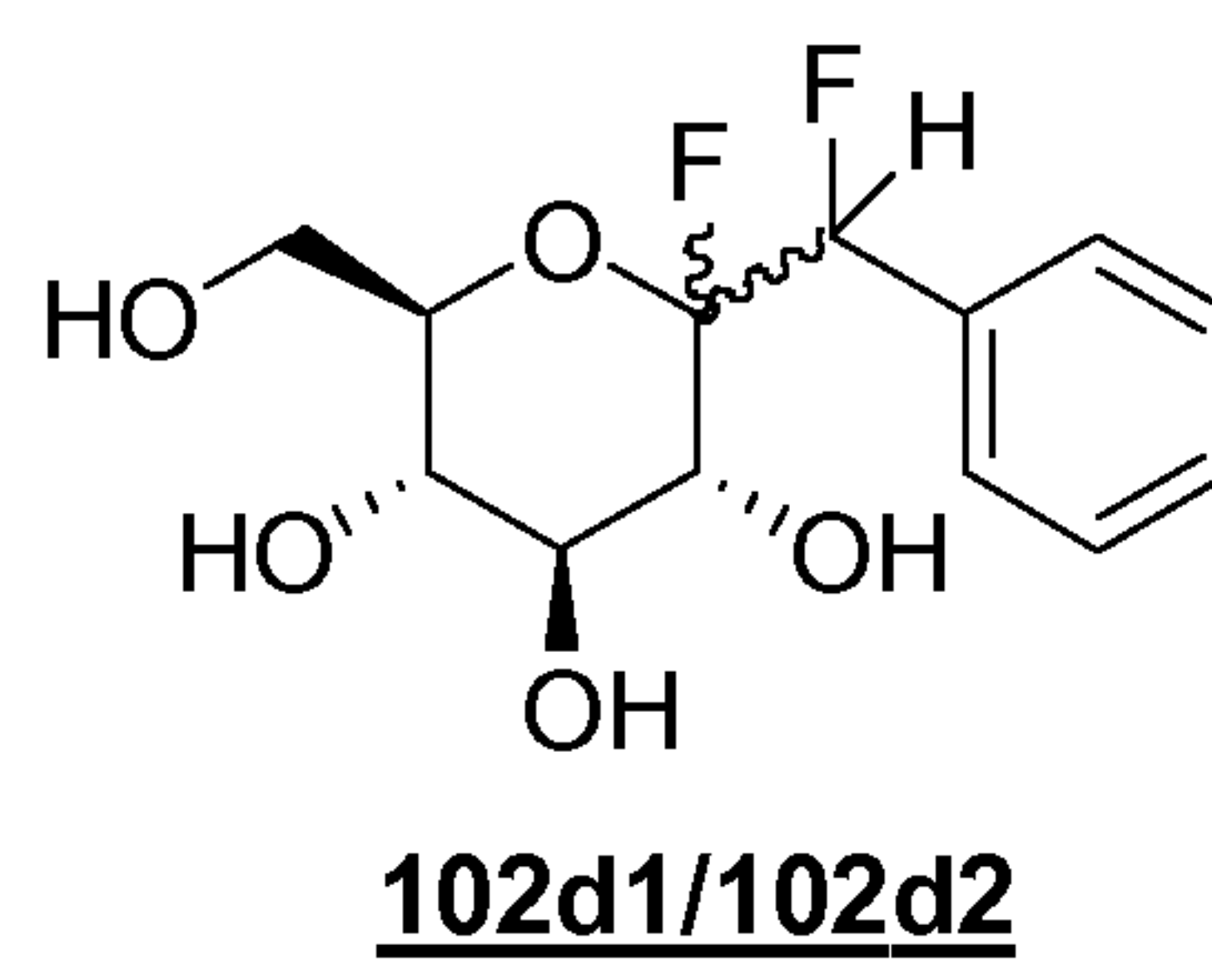
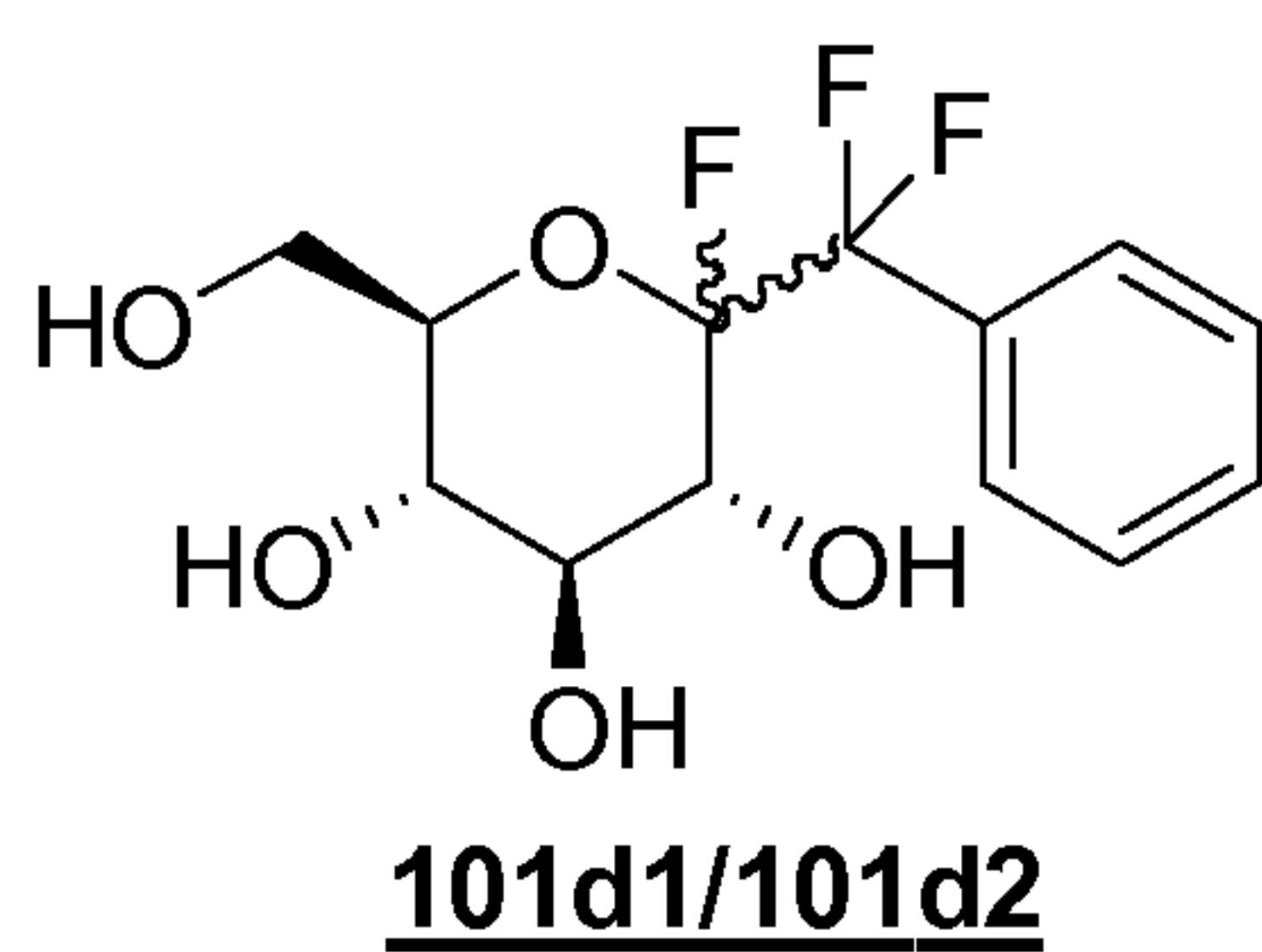
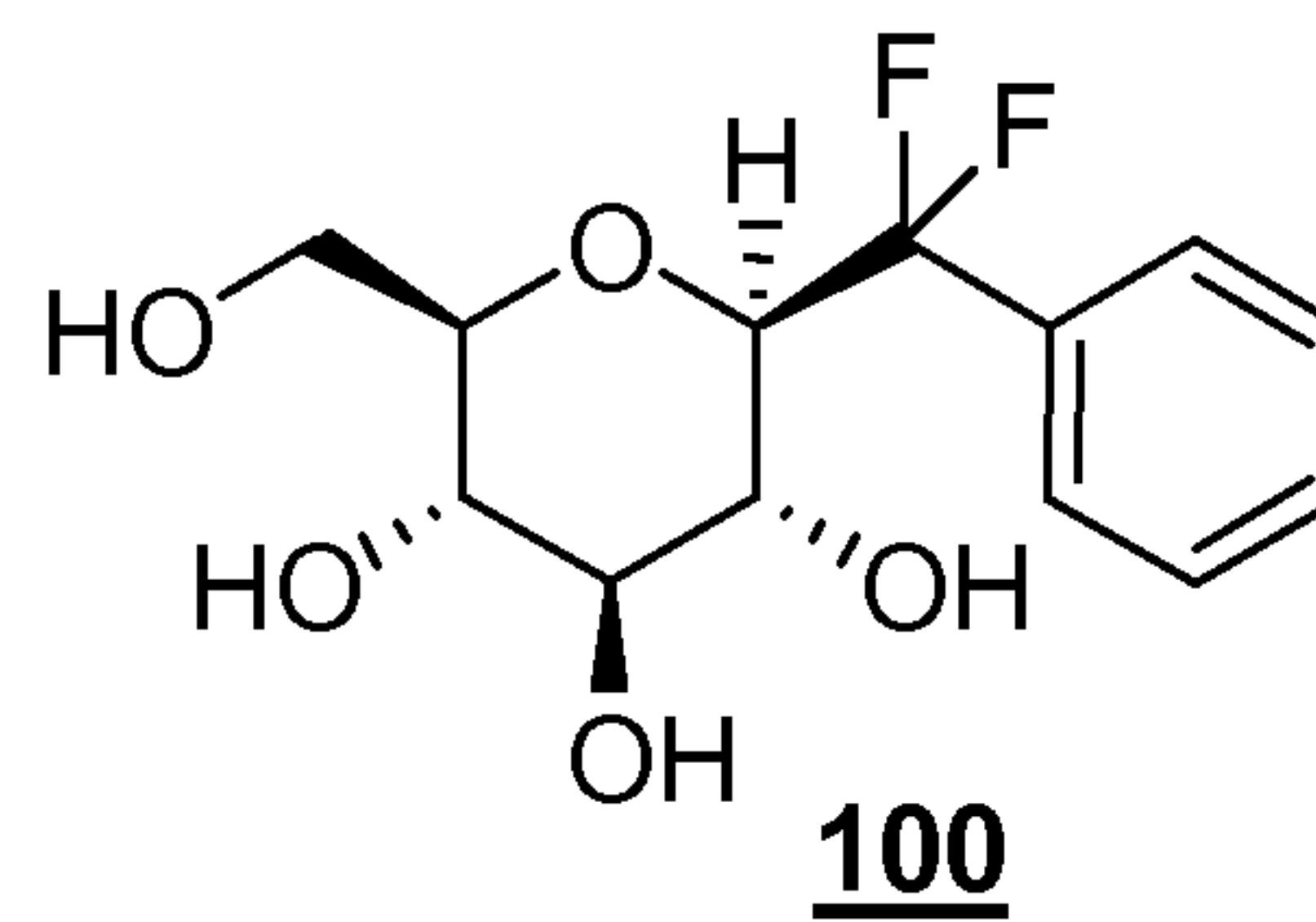
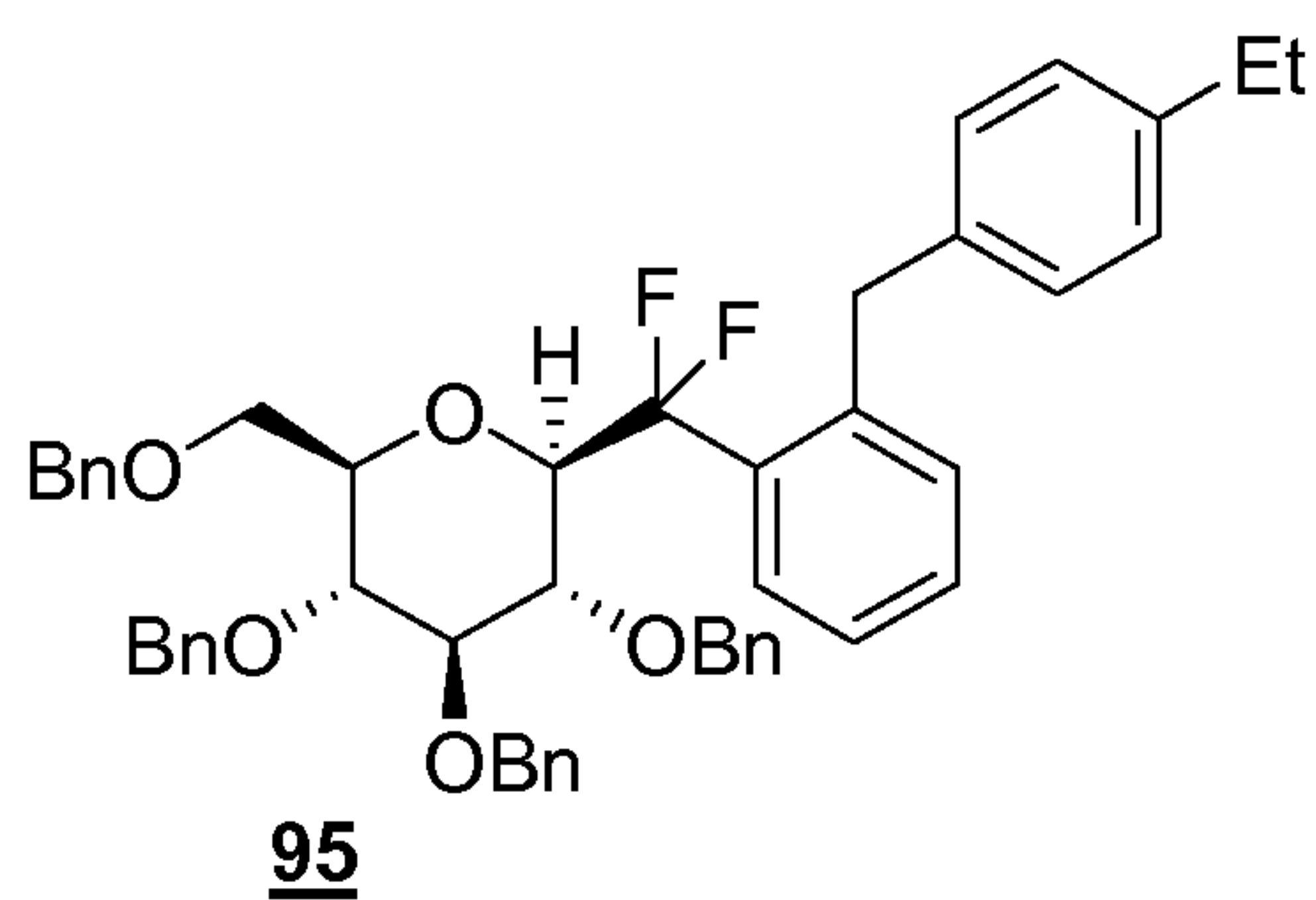
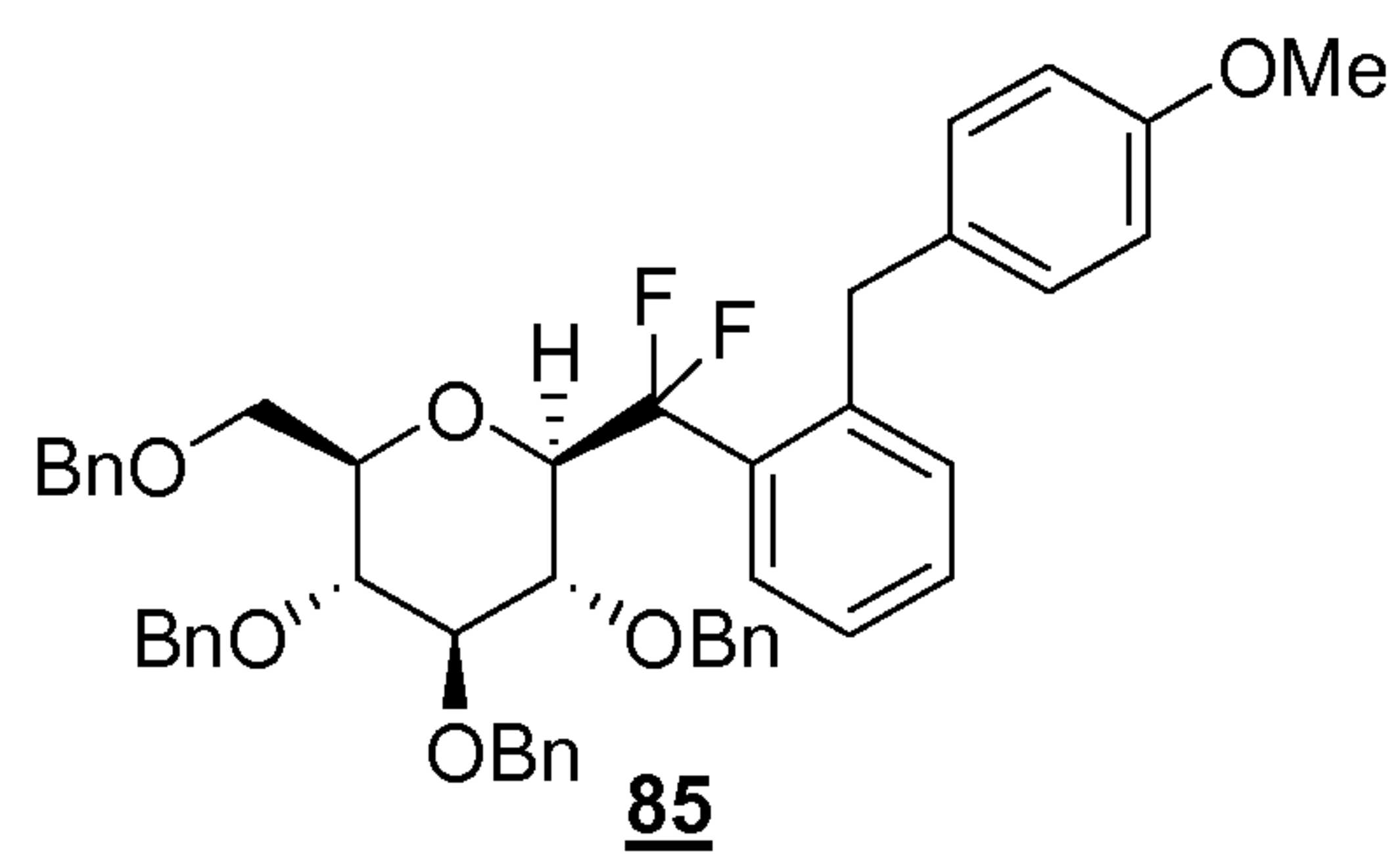
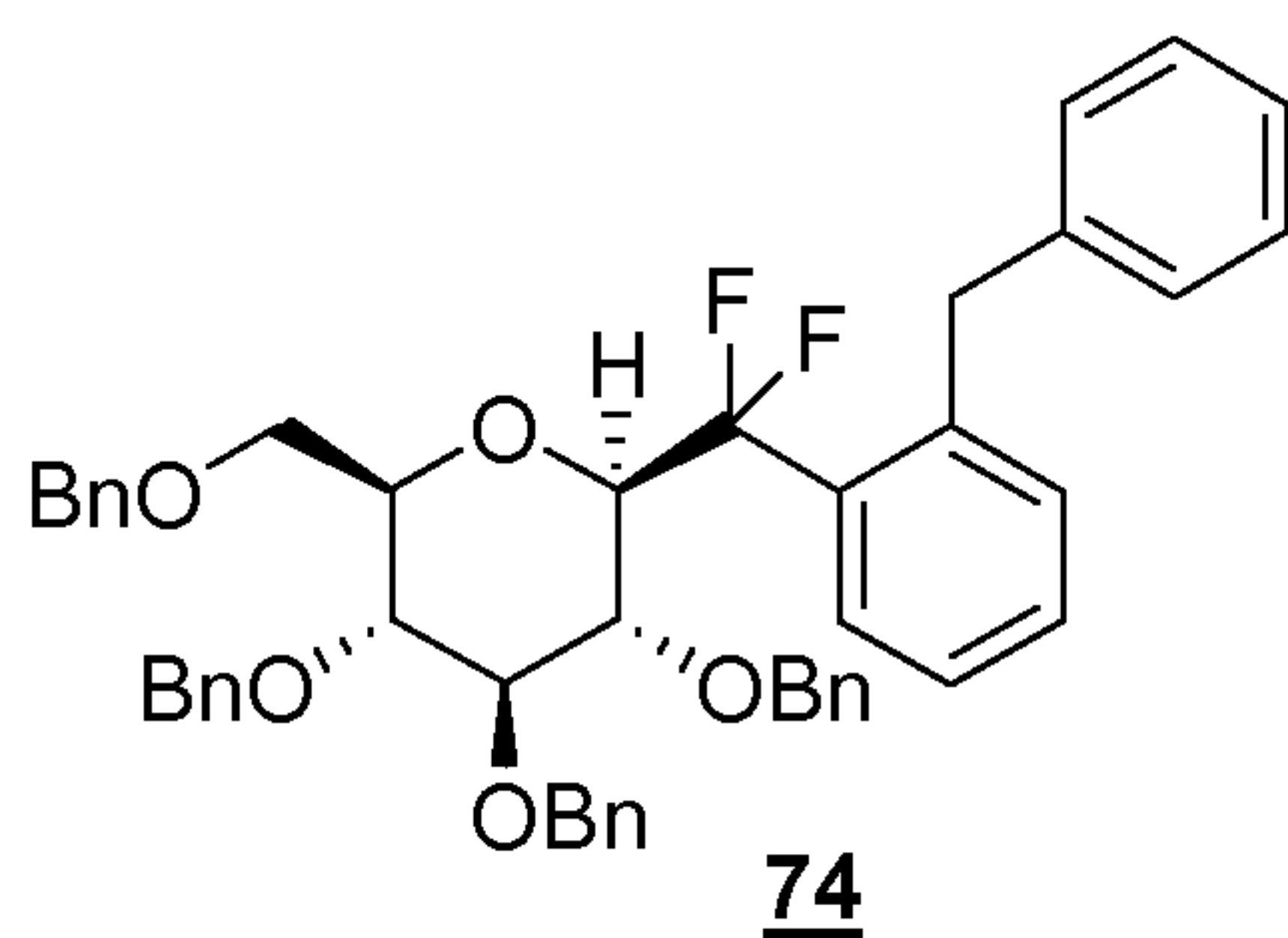
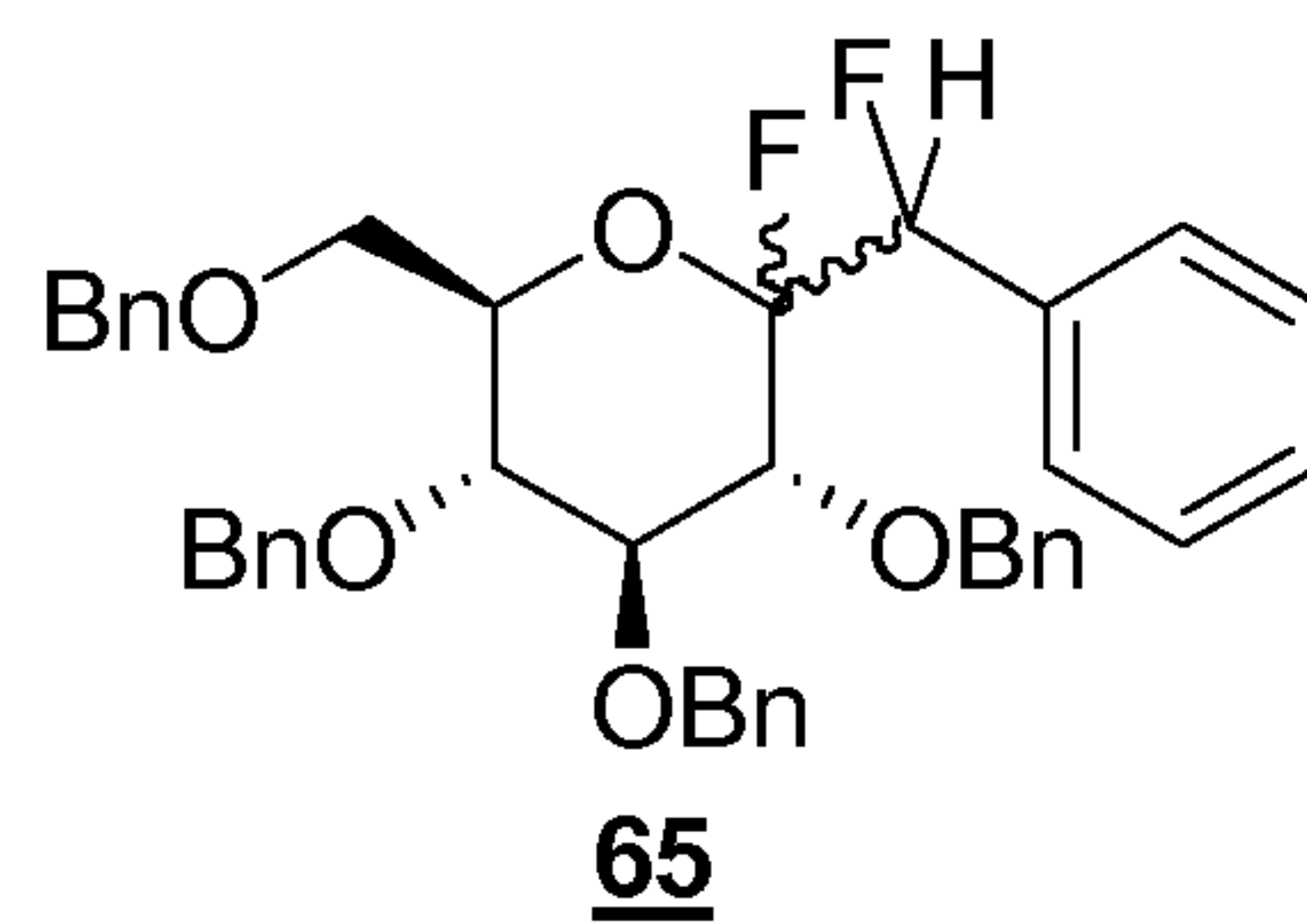
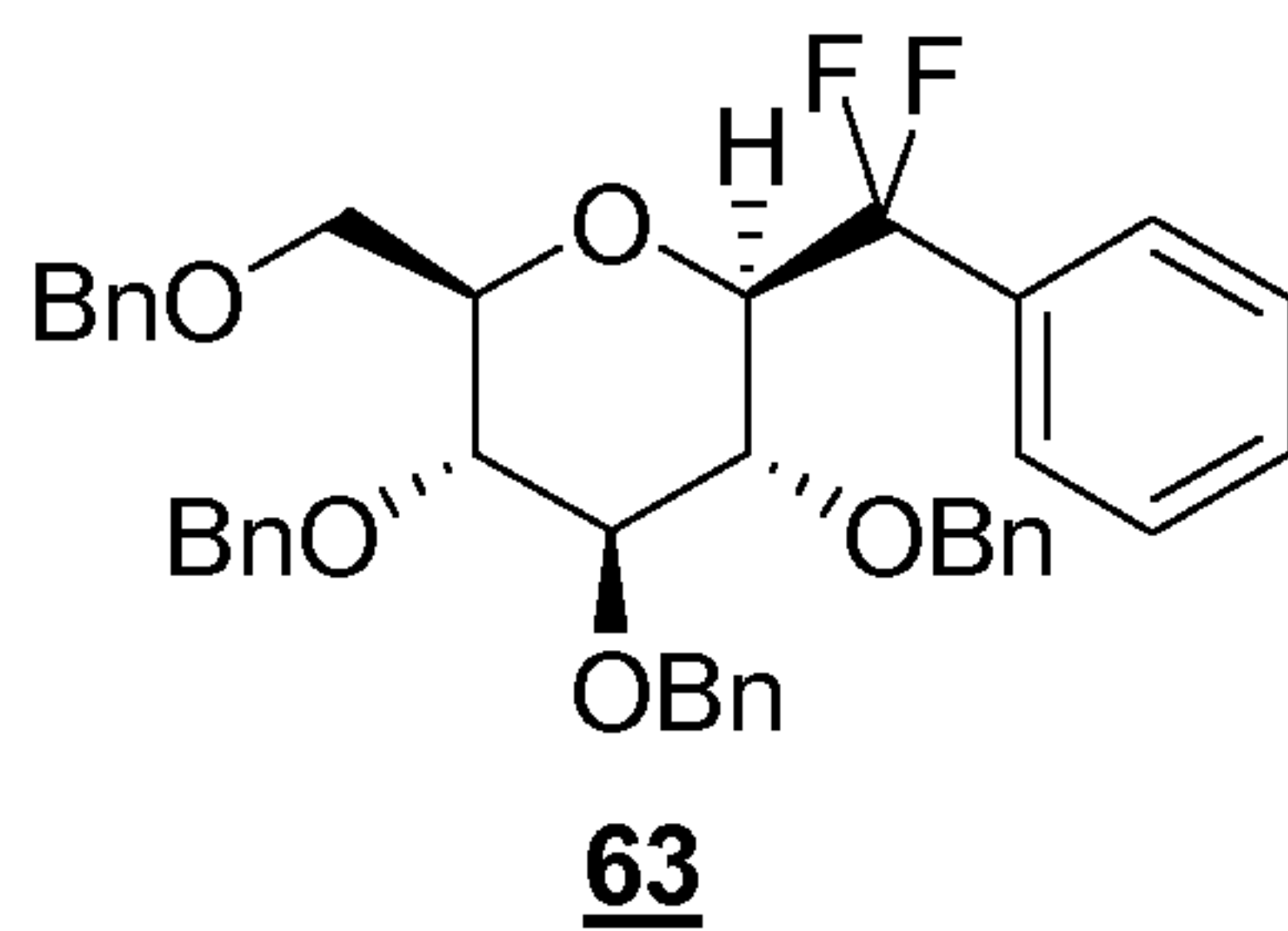
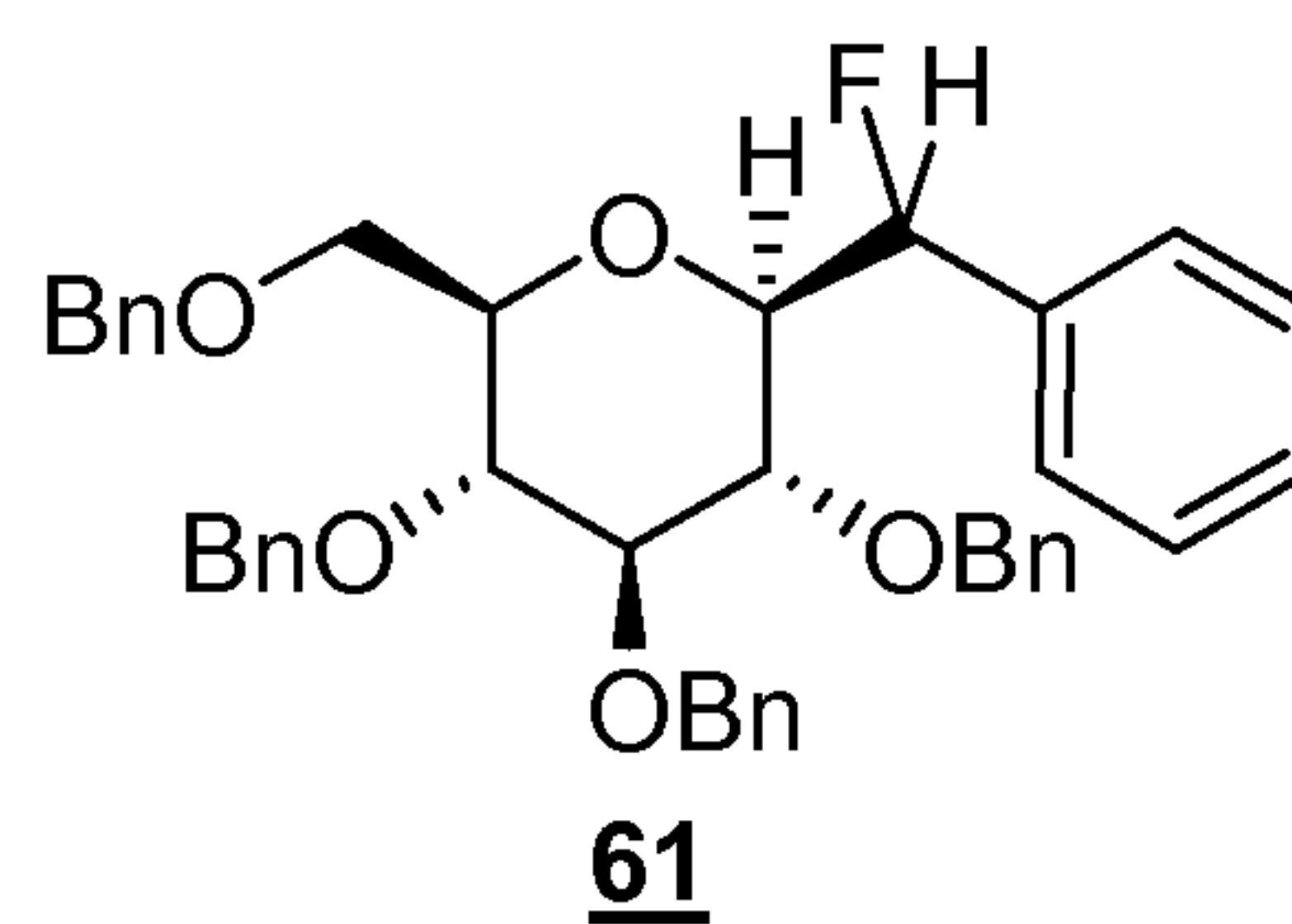
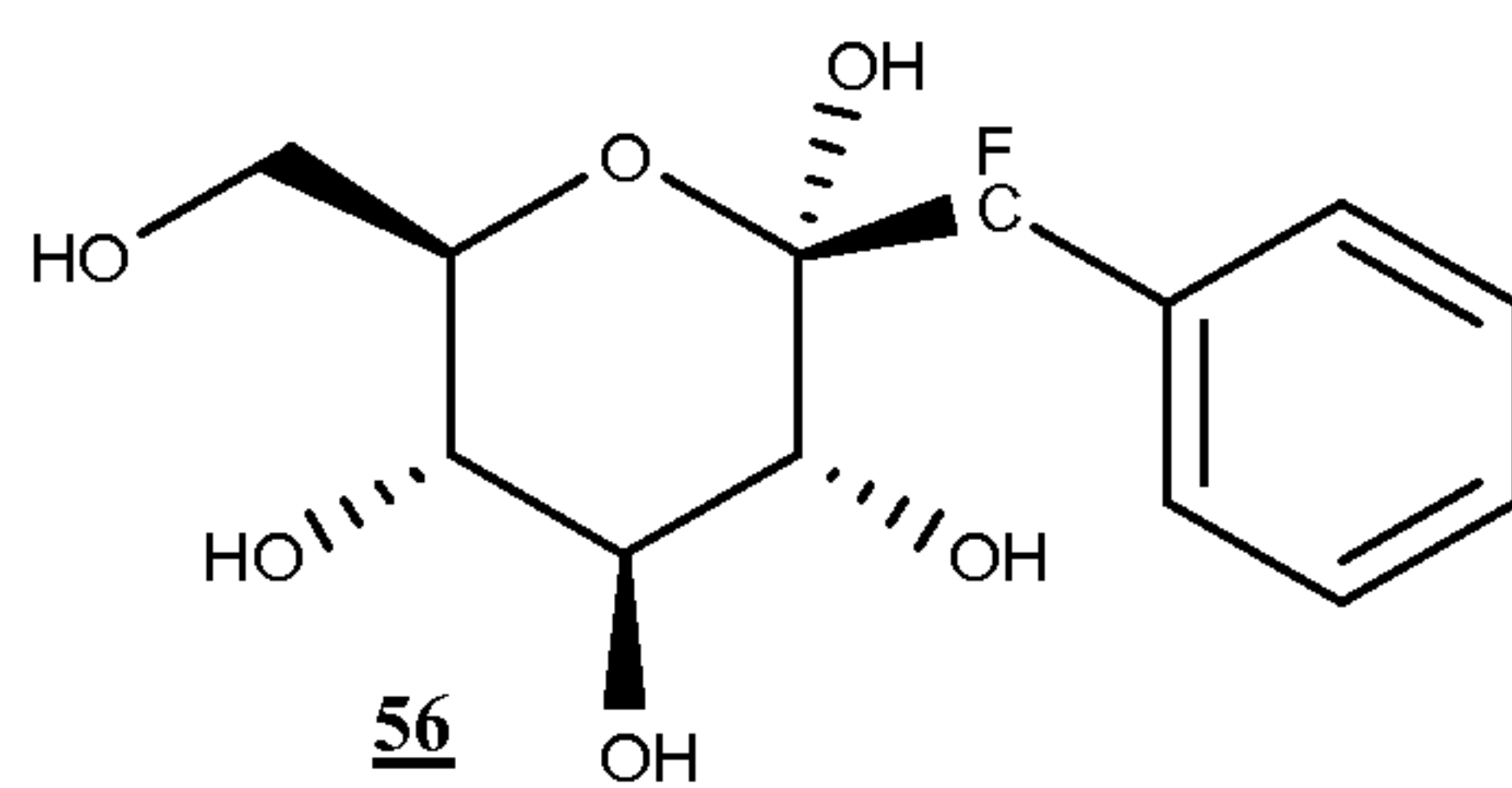
- R5, R6, R7, R8, R9, R5a, R6a, R7a, R8a and R9a represent, independently from one another, a hydrogen atom, an halogen atom, a CN, SO₂, SiR^aR^bR^c, (C₁-C₆)-alkyl, (C₂-C₆)-alkenyl, (C₂-C₆)-alkynyl, (C₃-C₇)-cycloalkyl, 5 to 7 ring-membered heterocycloalkyl, aryl, heteroaryl, aryl-(C₁-C₆)-alkyl, heteroaryl-(C₁-C₆)-alkyl, (C₁-C₆)-alkyl-aryl, (C₁-C₆)-alkyl-heteroaryl, OR¹¹, COR¹¹, OCOR¹¹, CO₂R¹¹, NR¹²R¹³, NR¹²COR¹¹, CONR¹²R¹³, SR¹¹, SO₂R¹¹, CSR¹¹ or OSO₃R¹¹ group, the said group being possibly substituted by one or more groups chosen among an halogen atom, an OH, (C₁-C₆)-alkyl, (C₁-C₆)-alkoxy, COOH and CHO group, and

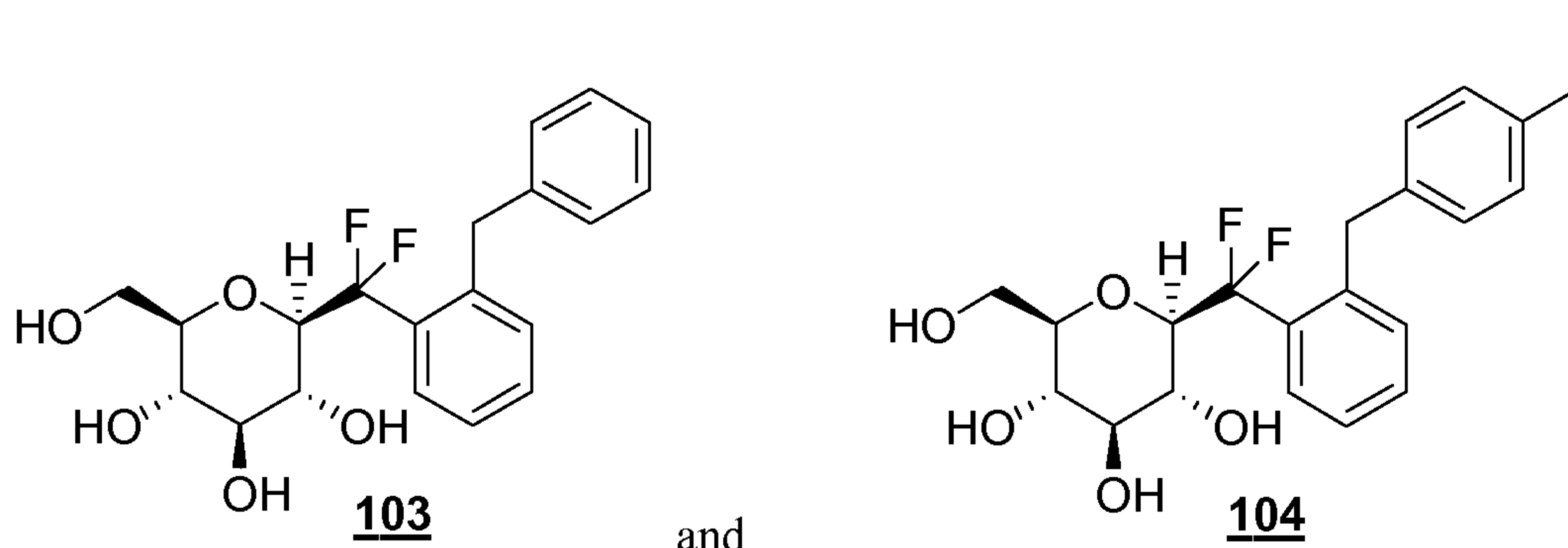
- X, R, R1, R2, R3, R4, R0, R¹¹, R¹² and R¹³ are as defined in claim 1.

3. Compound according to any of claims 1 and 2, characterised in that it is chosen among the following molecules:





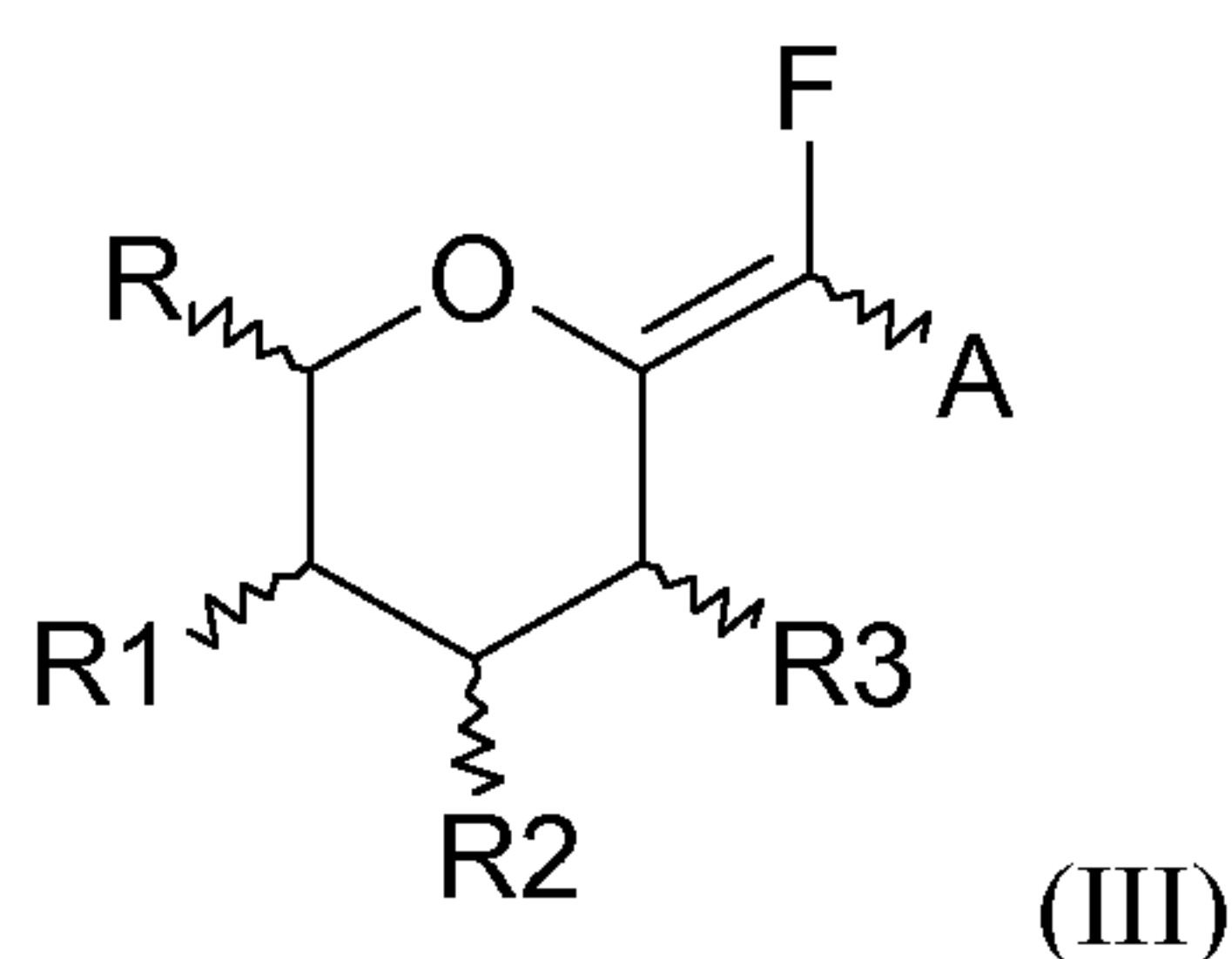




4. Compound according to any of claims 1 to 3, for use thereof as a drug, in particular as sodium-dependent glucose co-transporter inhibitor, notably for the treatment of diabetes, and more particularly type-II diabetes, diabetes-related complications, such as arteritis of the lower extremities, cardiac infarction, renal insufficiency, neuropathy or blindness, hyperglycemia, hyperinsulinemia, obesity, hypertriglyceridemia, X syndrome and arteriosclerosis, or as an anti-cancer, anti-infective, anti-viral, anti-thrombotic or anti-inflammatory drug.
5. Pharmaceutical or cosmetic composition including at least one compound as claimed in any of claims 1 to 3, and at least one pharmaceutically or cosmetically acceptable vehicle.
6. Pharmaceutical composition according to claim 5, characterised in that it includes at least one other active principle, advantageously chosen among antidiabetic agents, such as sulfonylurea-type compounds like chlorpropamide, tolbutamide, tolazamide, glipizide, gliclazide, glibenclamide, gliquidone and glimepiride; biguanides like metformine; thiazolidinediones like rosiglitazone, pioglitazone and ciglitazone; alpha-glucosidase inhibitors like acarbose, miglitol and voglibose; meglitinides like repaglinide and nateglinide; incretin mimics like exenatide; dipeptidylpeptidase-4 (DPP4) inhibitors like sitagliptin, vildagliptin and insulin; and antilipidic agents such as statins like atorvastatin and cerivastatin, fibrates like bezafibrate, gemfibrozil and fenofibrate, and ezetimibe.

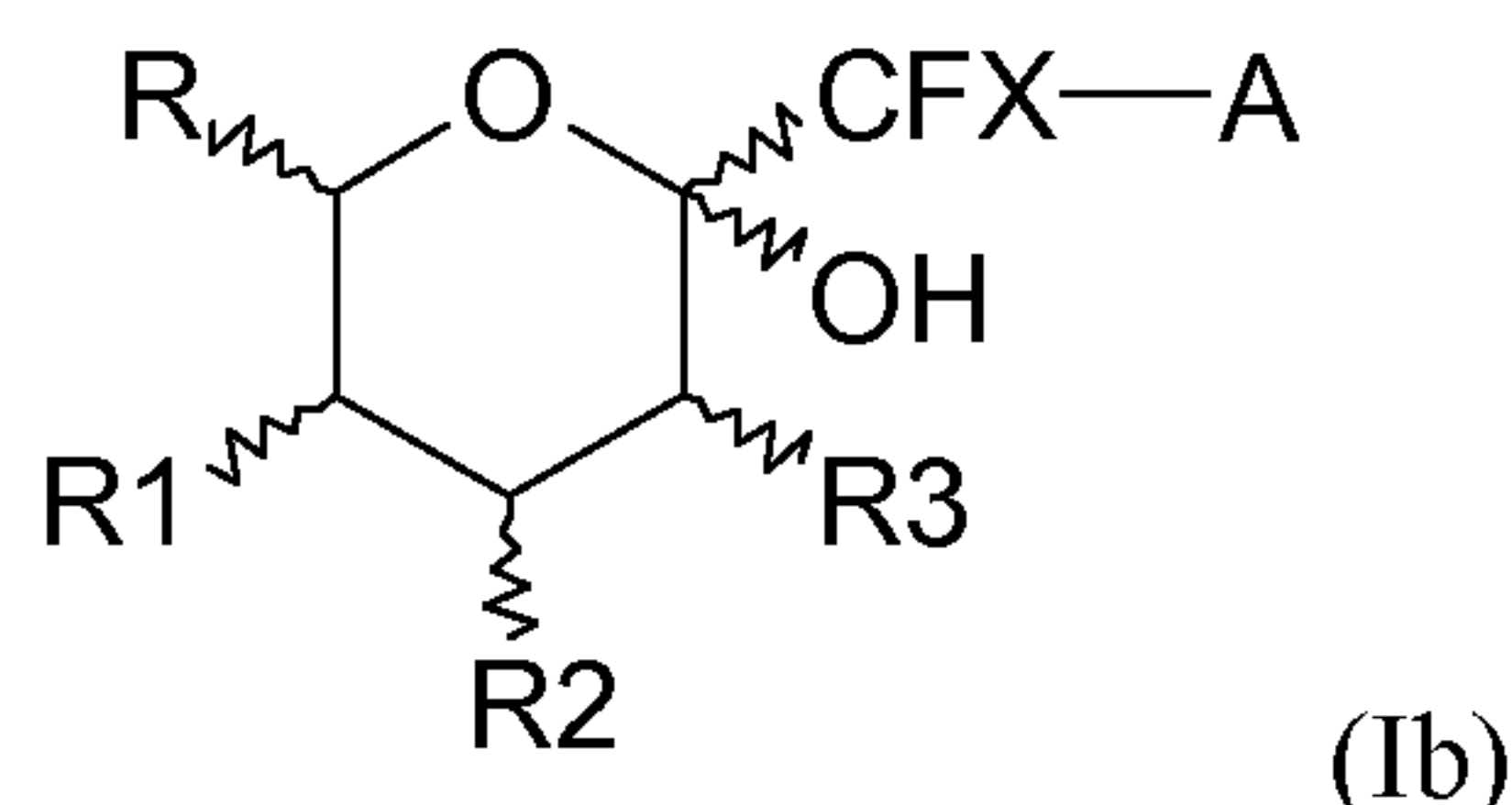
7. Cosmetic use of a compound according to any of claims 1 to 3, for lightening, bleaching, depigmenting the skin, removing blemishes from the skin, particularly age spots and freckles, or preventing pigmentation of the skin.

- 5 8. Process for preparing a compound of generic formula (I), according to any of claims 1 to 3, wherein X, R₀ and R₄ represent a hydrogen atom, characterized in that the compound of formula (I) is obtained by hydrogenation of the double bond of a compound of generic formula (III):



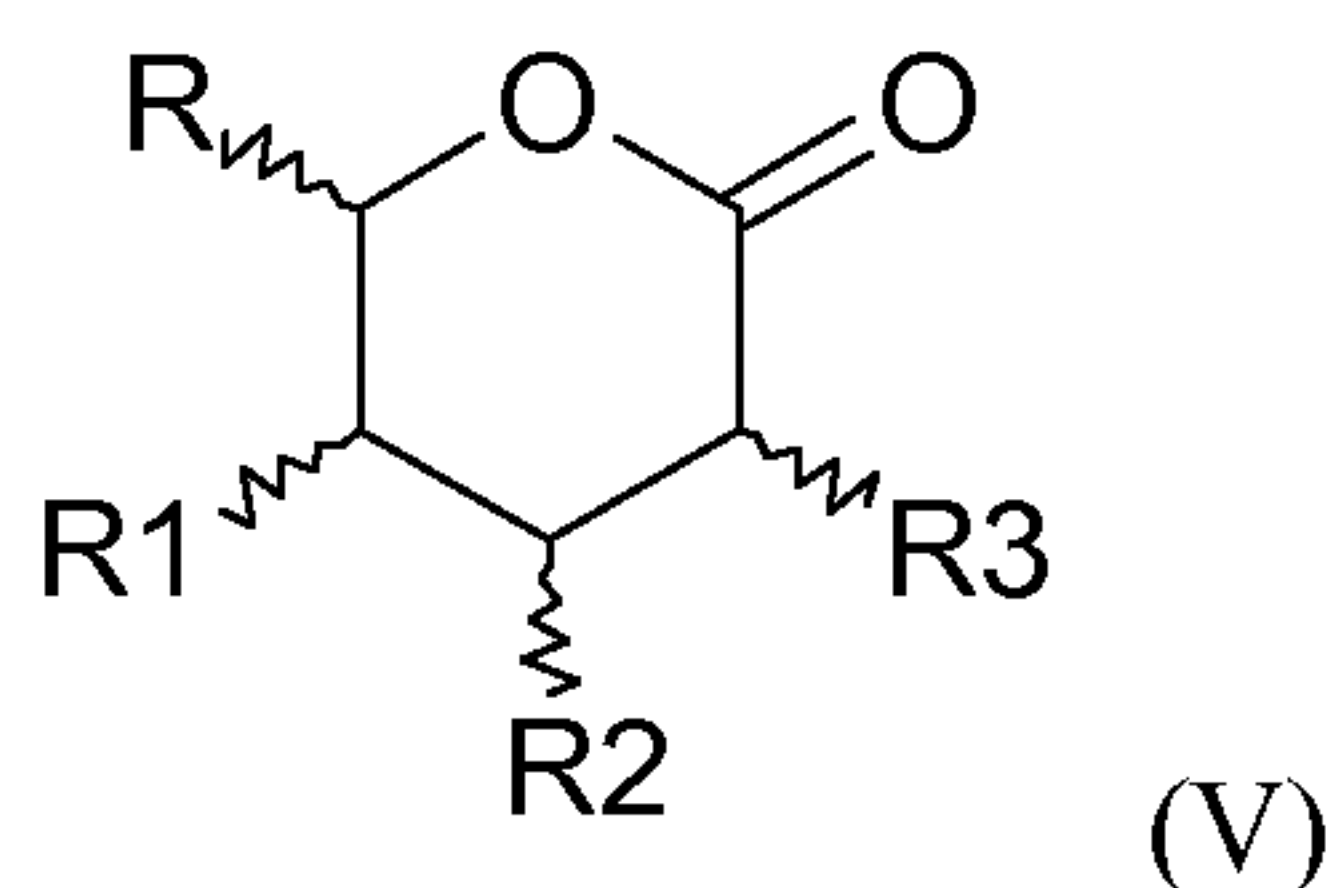
10 wherein A, R, R₁, R₂ and R₃ are as defined in claim 1.

9. Process for preparing a compound of generic formula (Ib) below:



15 corresponding to a compound of formula (I), as defined in claim 1, wherein X represents a hydrogen or a fluorine atom and R₄ represents an OH group, according to the following steps:

(a3) placing a compound of formula A-CFXX', wherein X is as defined above, A is as defined in claim 1 and X' represents a bromine or chlorine atom, in the presence of a compound of generic formula (V):



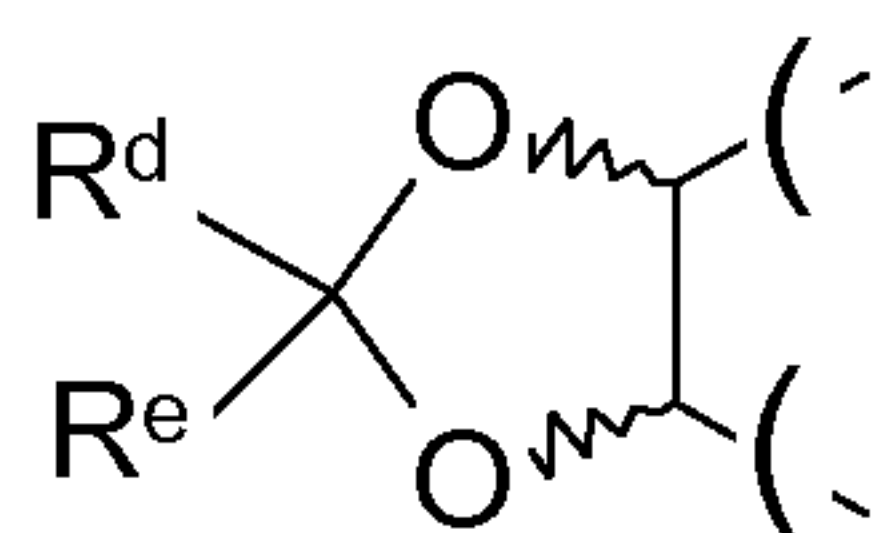
20

wherein R, R₁, R₂ and R₃ are as defined in claim 1, and

(b3) addition of a (C₁-C₆)-alkyl lithium to the mixture of step (a3), in order to obtain a compound of formula (Ib).

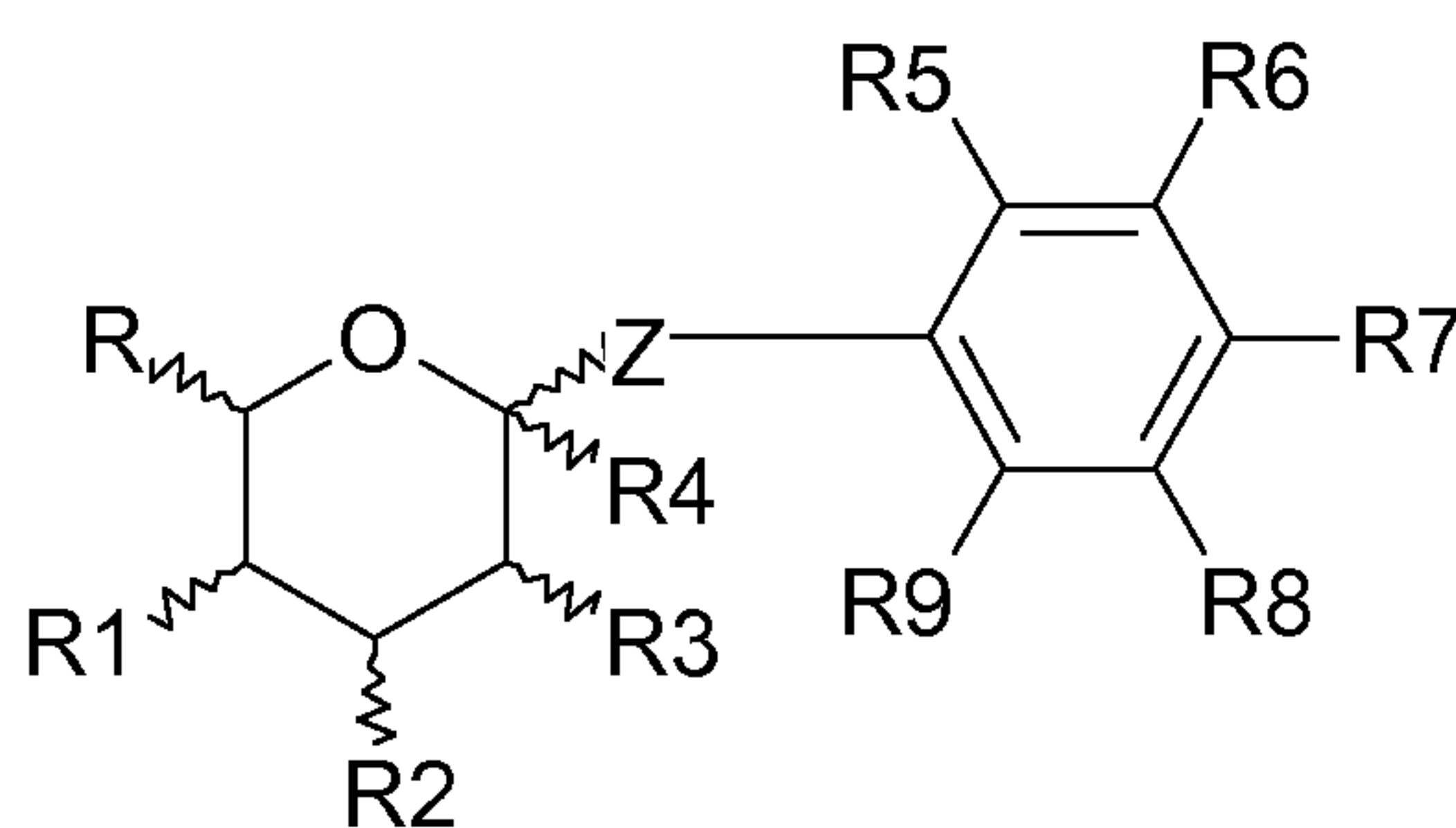
10. Process for preparing a compound of generic formula (I) as claimed in any of claims 1 to 3, wherein:

- X represents a hydrogen or a fluorine atom and
- R₄ represents a OSiR^aR^bR^c, OR¹¹, OCOR¹¹, OCO₂R¹¹, OCONR¹²R¹³ group or R₃ and R₄, together with the carbon atoms carrying them, form a cyclic acetal having the following formula:



with R^a, R^b, R^c, R^d, R^e, R¹¹, R¹² and R¹³ as defined in claim 1, characterised in that the compound of formula (I) is obtained by substitution of the OH group of a compound of formula (Ib) as defined in claim 9.

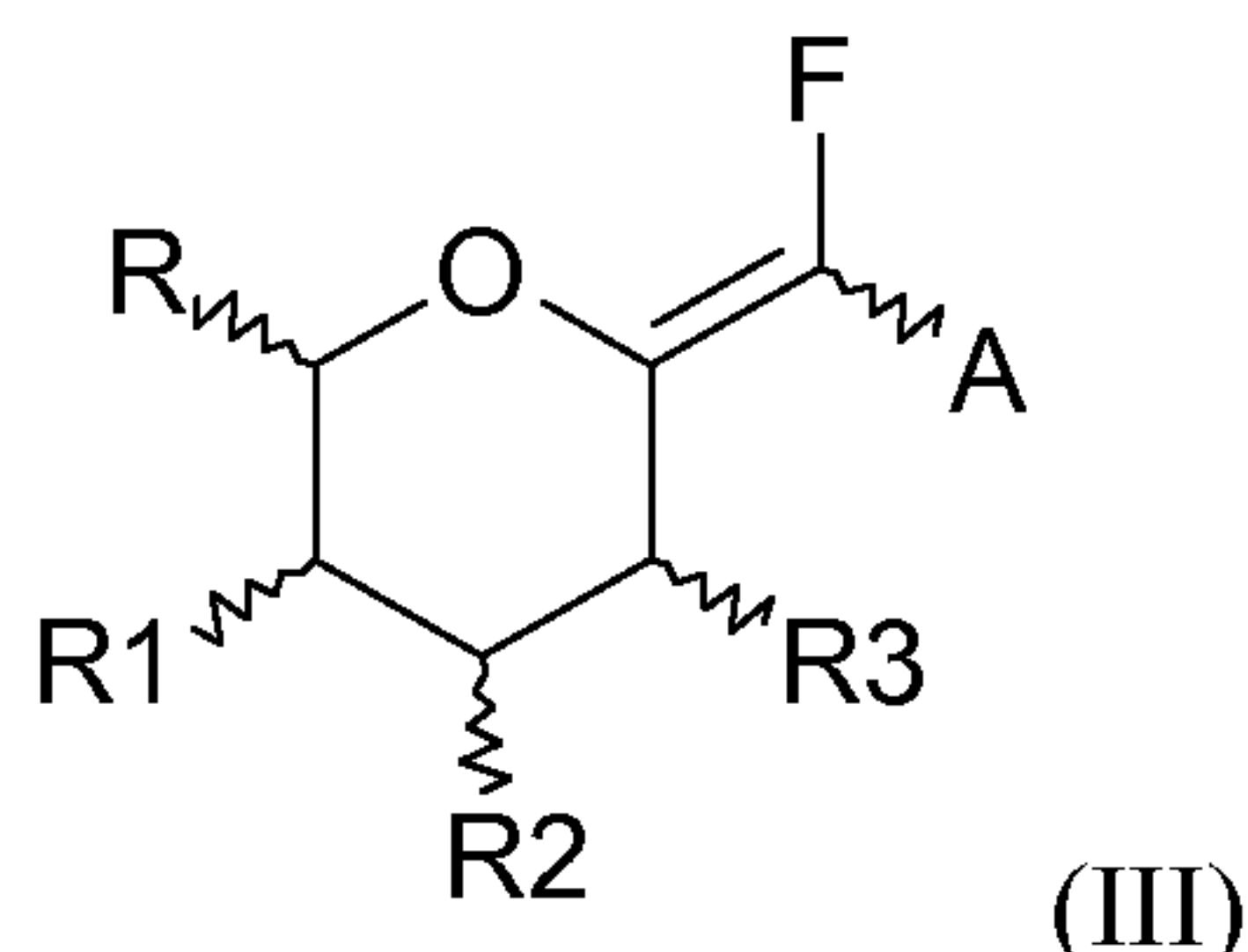
11. Process for preparing a compound of generic formula (II) as defined in claim 2, wherein R₀ = H, characterized in that the compound of formula (II) is obtained by fluorination of a compound of the following formula (VII):



(VII),

wherein R, R₁, R₂, R₃ and R₄ are as defined in claim 1, R₅, R₆, R₇, R₈ and R₉ are as defined in claim 2 and Z represents a C=O, CHOH or C(SR¹⁵)(SR¹⁶) group, with R¹⁵ and R¹⁶ representing, independently of each other, a (C₁-C₆)alkyl group or forming together an hydrocarbon chain of formula -CH₂-(CH₂)_p-, with p = 1 or 2, between the two sulphur atoms.

12. Compound of generic formula (III) below:



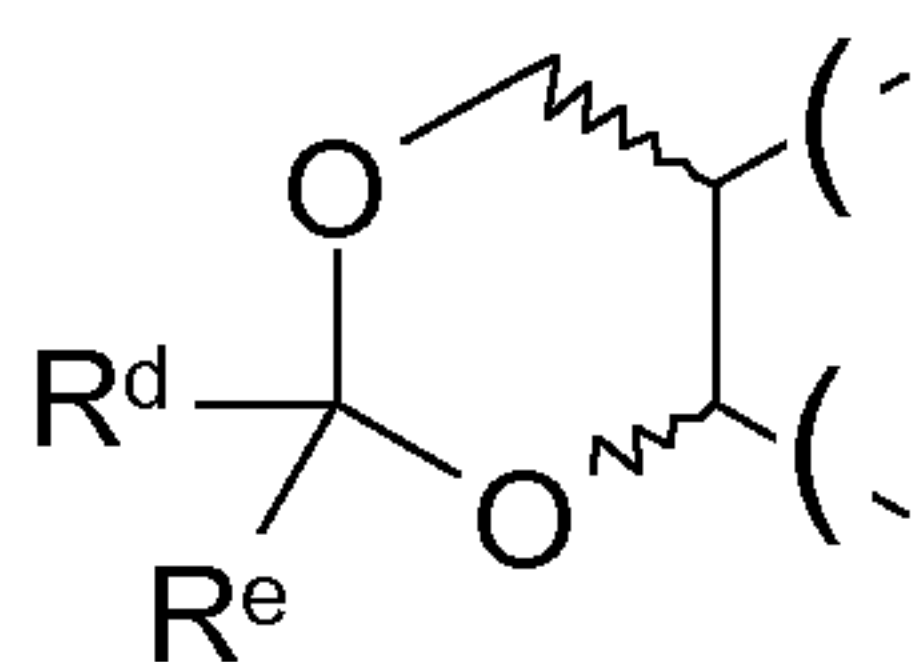
or a pharmaceutically acceptable salt thereof, a tautomer, an isomer or a mixture of isomers in any proportions, in particular a mixture of enantiomers, and particularly a racemate mixture, wherein:

- R represents a hydrogen or a fluorine atom or CH₃, CH₂F, CH₂OH, CH₂OSiR^aR^bR^c, CH₂OR¹¹, CH₂OCOR¹¹, CH₂OCO₂R¹¹, CH₂OCONR¹²R¹³, CH₂OP(O)(OR¹⁴)₂ or CH₂OSO₃R¹⁴ group;

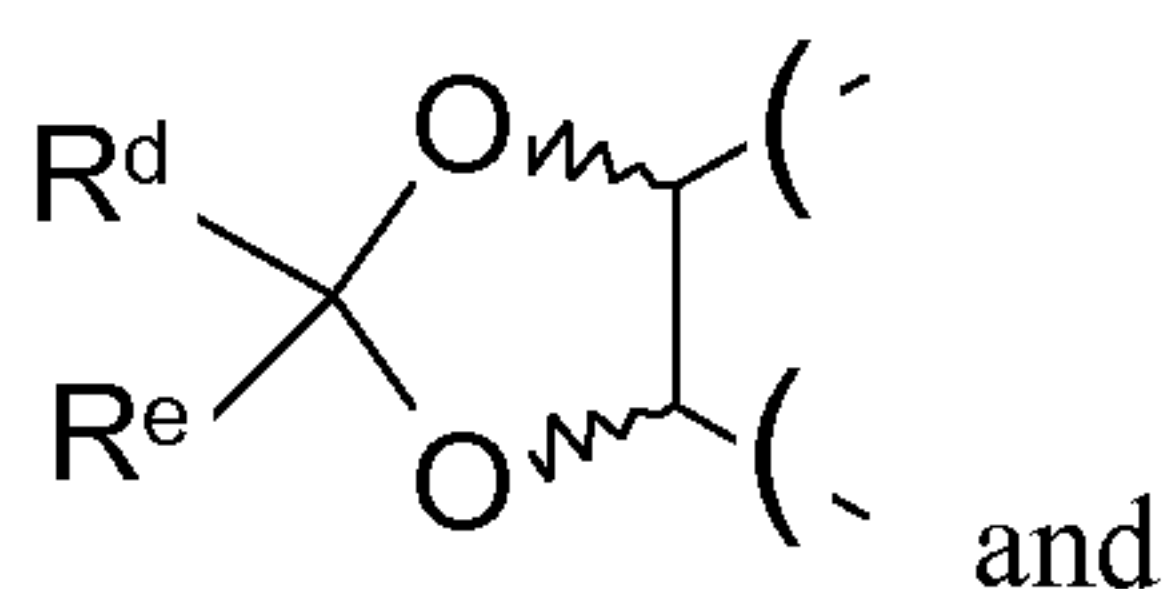
- R1 and R2 and represent, independently from one another, a fluorine atom or an OH, OSiR^aR^bR^c, OR¹¹, OCOR¹¹, OCO₂R¹¹ or OCONR¹²R¹³ group;

- R3 represents a hydrogen or fluorine atom or an OH, OSiR^aR^bR^c, OR¹¹, OCOR¹¹, OCO₂R¹¹, OCONR¹²R¹³, NR¹²R¹³ or NR¹²COR¹¹ group;

or R and R1, together with the carbon atoms carrying them, form a cyclic acetal having the following formula:



and/or (R2 and R3) or (R1 and R2), together with the carbon atoms carrying them, form a cyclic acetal having the following formula:



- A represents an aryl or heteroaryl group, possibly substituted by one or more groups chosen among an halogen atom, a CN, SO₂, SiR^aR^bR^c, (C₁-C₆)-alkyl, (C₂-C₆)-alkenyl, (C₂-C₆)-alkynyl, (C₃-C₇)-cycloalkyl, 5 to 7 ring-membered heterocycloalkyl, aryl, heteroaryl, aryl-(C₁-C₆)-alkyl, heteroaryl-(C₁-C₆)-alkyl, (C₁-C₆)-alkyl-aryl, (C₁-C₆)-alkyl-heteroaryl, OR¹¹, COR¹¹, OCOR¹¹, CO₂R¹¹, NR¹²R¹³, NR¹²COR¹¹, CONR¹²R¹³, SR¹¹, SO₂R¹¹, CSR¹¹ and OSO₃R¹¹ group,

the whole being possibly substituted by one or more groups chosen among an halogen atom, an OH, (C₁-C₆)-alkyl, (C₁-C₆)-alkoxy, COOH and CHO group;

with:

- R¹¹ representing a (C₁-C₆)-alkyl, (C₂-C₆)-alkenyl, (C₂-C₆)-alkynyl, (C₃-C₇)-cycloalkyl, 5 to 7 ring-membered heterocycloalkyl, aryl, aryl-(C₁-C₆)-alkyl or (C₁-C₆)-alkyl-aryl, this group being possibly substituted by one or more groups chosen among an halogen atom, an OH, COOH and CHO group;

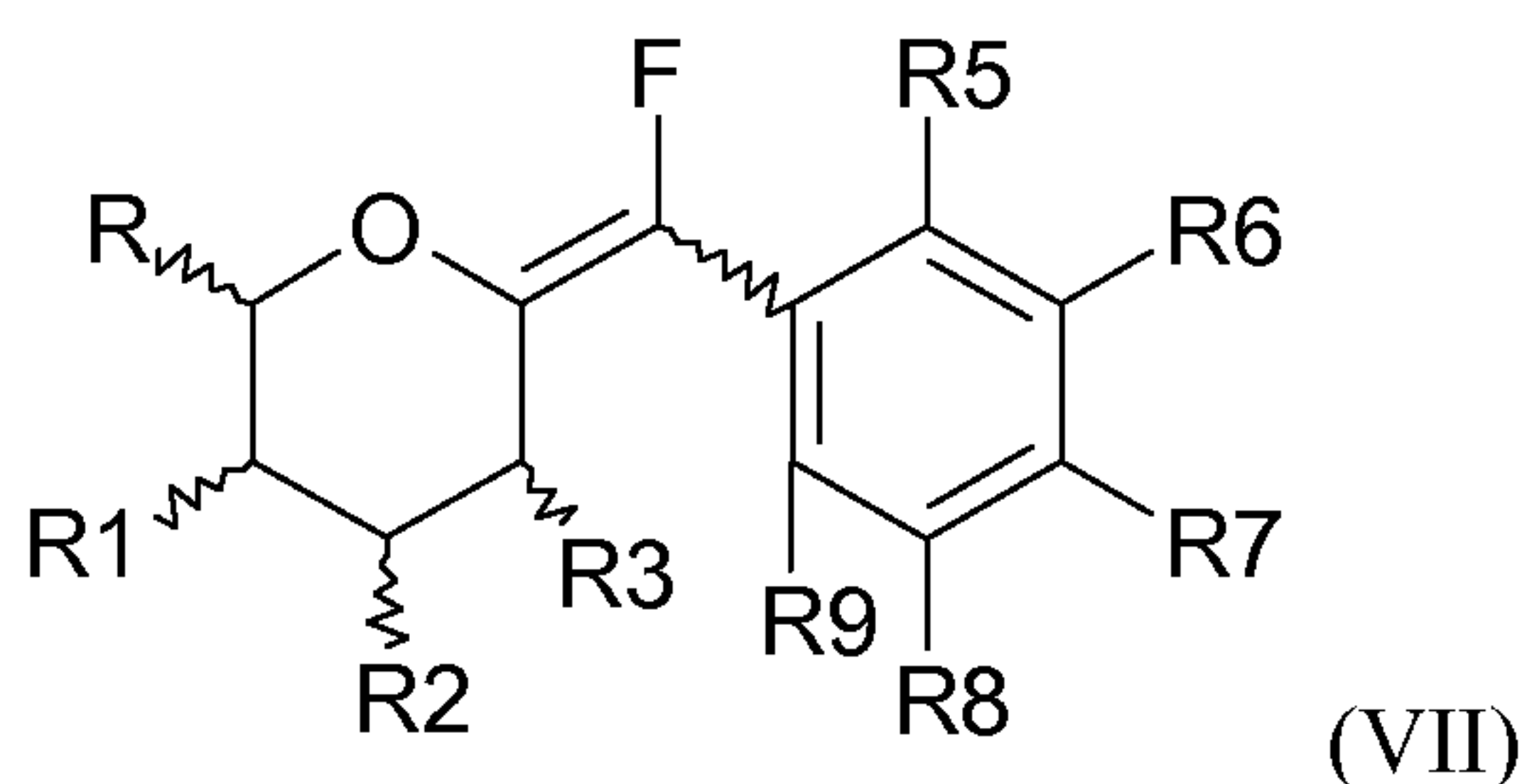
- R¹² and R¹³ representing, independently from one another, a hydrogen atom or a (C₁-C₆)-alkyl or aryl-(C₁-C₆)-alkyle group;

- R¹⁴ representing a hydrogen atom or a (C₁-C₆)-alkyl group;

- R^a, R^b and R^c representing, independently from one another, a (C₁-C₆)-alkyl, aryl or aryl-(C₁-C₆)-alkyl group; and

- R^d and R^e representing, independently from one another, a hydrogen atom or a (C₁-C₆)-alkyl group.

13. Compound according to claim 12, characterised in that it responds to the following generic formula (VII):

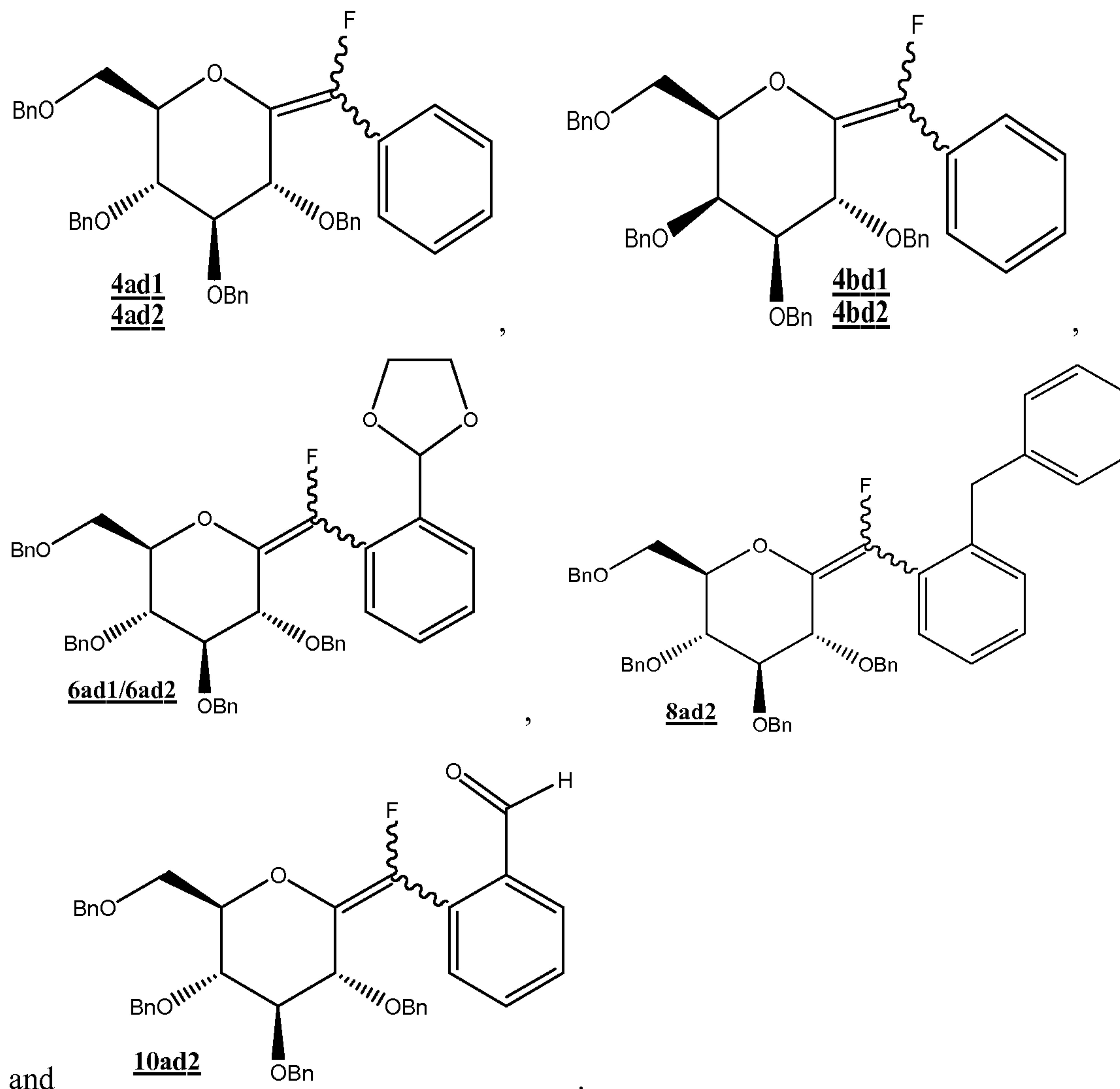


wherein:

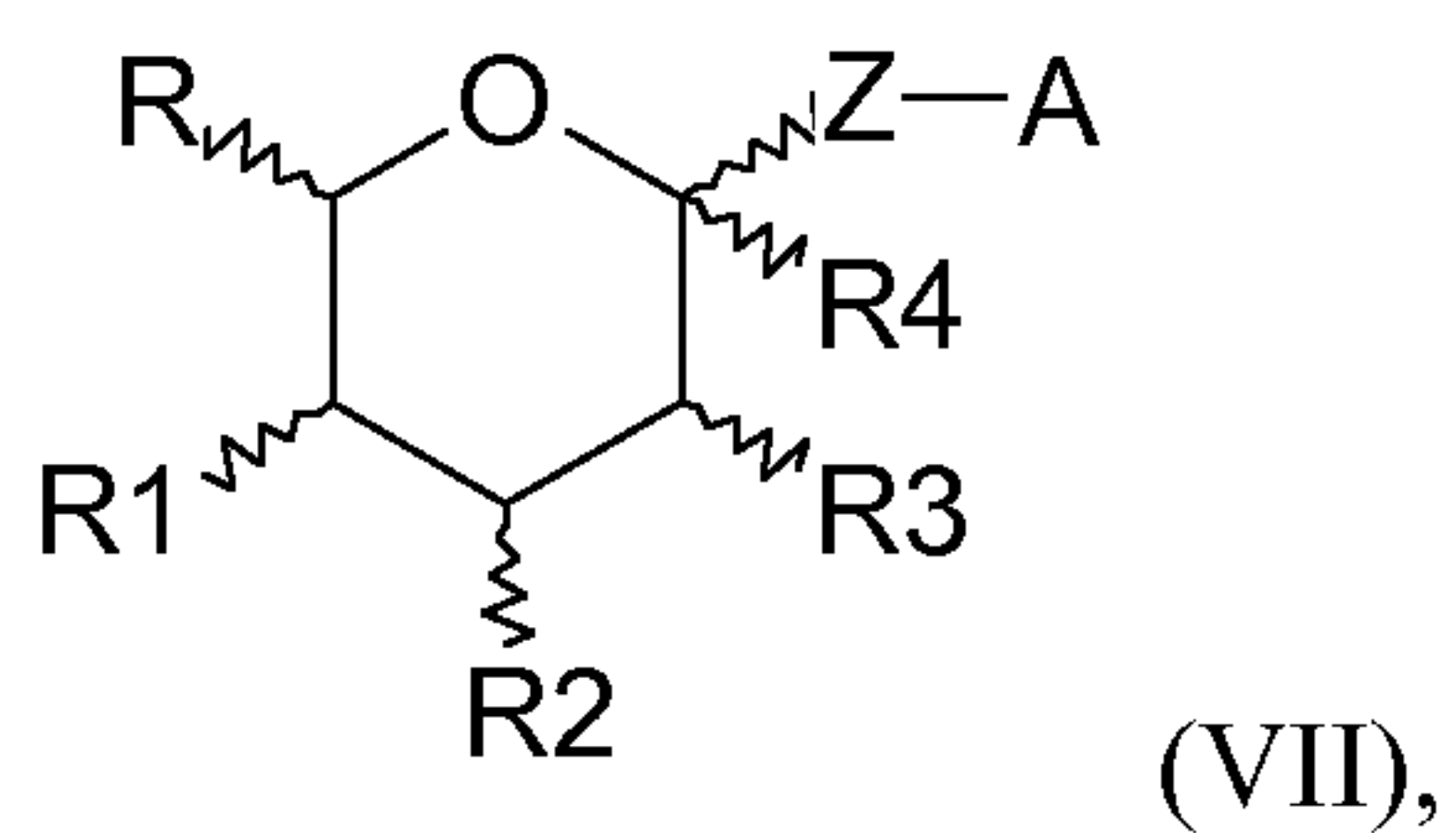
- R₅, R₆, R₇, R₈ and R₉ represent, independently from one another, a hydrogen atom, an halogen atom, a CN, SO₂, SiR^aR^bR^c, (C₁-C₆)-alkyl, (C₂-C₆)-alkenyl, (C₂-C₆)-alkynyl, (C₃-C₇)-cycloalkyl, 5 to 7 ring-membered heterocycloalkyl, aryl, heteroaryl, aryl-(C₁-C₆)-alkyl, heteroaryl-(C₁-C₆)-alkyl, (C₁-C₆)-alkyl-aryl, (C₁-C₆)-alkyl-heteroaryl, OR¹¹, COR¹¹, OCOR¹¹, CO₂R¹¹, NR¹²R¹³, NR¹²COR¹¹, CONR¹²R¹³, SR¹¹, SO₂R¹¹, CSR¹¹ or OSO₃R¹¹ group, possibly substituted by one or more groups chosen among an halogen atom, an OH, COOH CHO group; and

- R, R₁, R₂, R₃, R¹¹, R¹² and R¹³ are as defined in claim 1.

14. Compound according to claim 12 or 13, characterised in that it is chosen among:

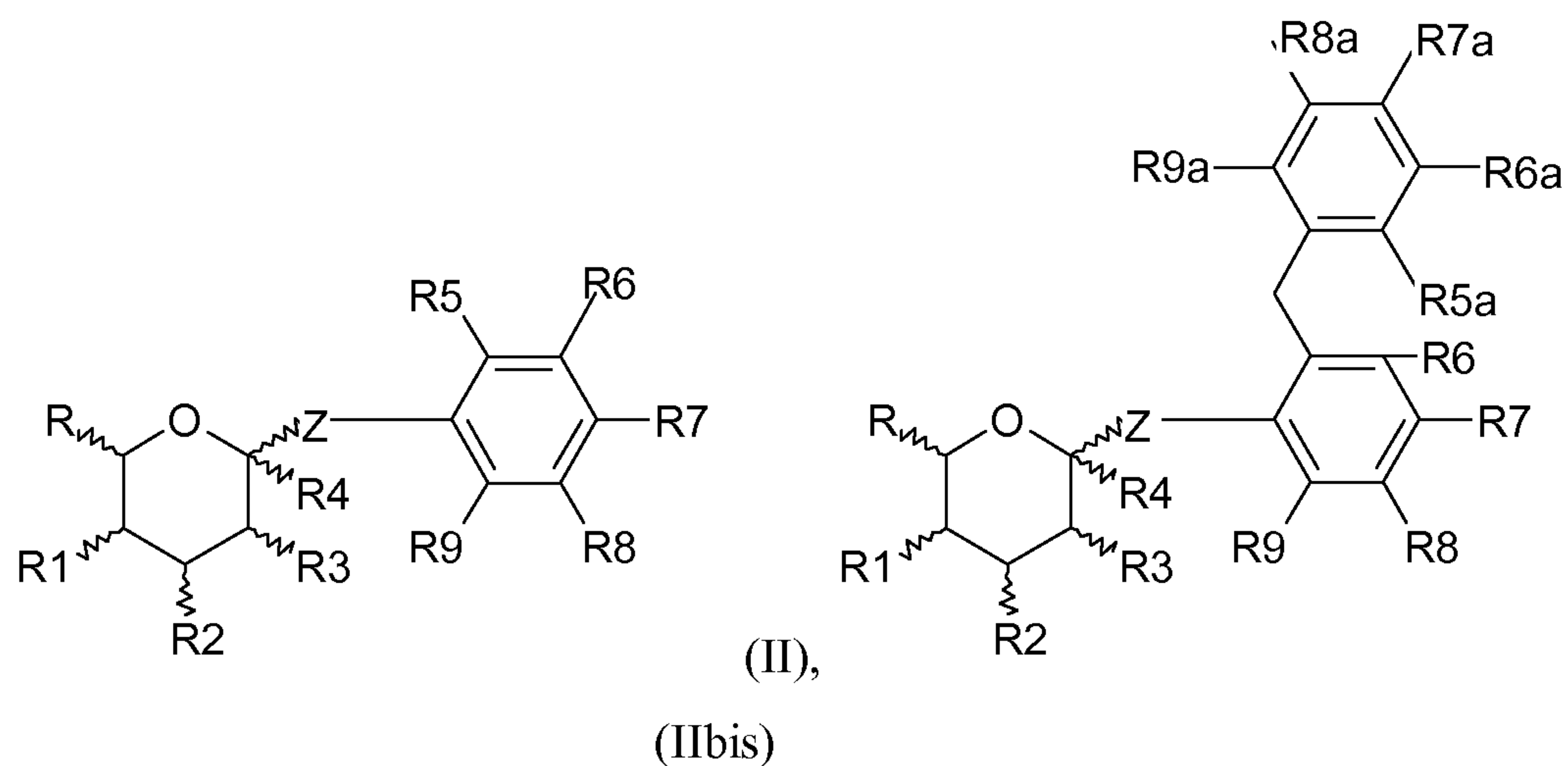


15. Compound of the following formula (VII):



- 5 or a pharmaceutically acceptable salt thereof, a tautomer, an isomer or a mixture of isomers in any proportions, in particular a mixture of enantiomers, and particularly a racemate mixture,
- wherein A, R, R1, R2, R3 and R4 are as defined in claim 1 and Z is as defined in claim 11.

16. Compound according to claim 15, characterised in that it responds to the following generic formula (VIIa) or (VIIb):



5

wherein R_0 , R, R1, R2, R3 and R4 are as defined in claim 1, R5, R6, R7, R8, R9, R5a, R6a, R7a, R8a and R9a are as defined in claim 2 and Z is as defined in claim 15.

17. Compound according to claim 15 or 16, characterised in that it is chosen among:

