

## COMMONWEALTH OF AUSTRALIA

PATENTS ACT 1952

597924

## APPLICATION FOR A STANDARD PATENT

☒ We, SECTION 34(4)(a) DIRECTION SEE FOLIO 5  
NAME DIRECTED BUNGE (AUSTRALIA) PTY. LTD.,  
of 140 William Street, Melbourne, Victoria,  
Australia.  
 of Melbourne, Vic. 3000  
AUSTRALIA



hereby apply for the grant of a standard patent for an invention  
 entitled "SOLUBILIZATION OF PROTEIN AGGREGATES"

which is described in the accompanying provisional/~~complete~~  
 specification.

~~Details of basic application(s) -~~

~~Number of basic application      Name of Convention country in which basic application was filed      Date of basic application~~

APPLICATION ACCEPTED AND AMENDMENTS

ALLOWED

28.3.90

~~My~~ our address for service is care of CLEMENT HACK & CO., Patent  
 Attorneys, 140 William Street, Melbourne, Victoria, 3000,  
 Australia.

DATED this 11th

day of

December,

1985

ADDRESS FOR SERVICE  
 ALTERED



BUNGE INDUSTRIAL LIMITED

Phillips, Ormonde and Fitzpatrick  
CLEMENT HACK & CO.

*C. B. Bunter*

TO: The Commissioner of Patents.

PF/App/9/83

# AUSTRALIA

Patents Act

## DECLARATION FOR A PATENT APPLICATION

### INSTRUCTIONS

(a) Insert "Convention" if applicable  
(b) Insert FULL name(s) of applicant(s)

In support of the (a) application made by  
(b) BUNGE (AUSTRALIA) PTY. LTD.

(c) Insert "of addition" if applicable  
(d) Insert TITLE of invention

(hereinafter called "applicant(s) for a patent (c) for an  
invention entitled (d)  
SOLUBILIZATION OF PROTEIN AGGREGATES

(e) Insert FULL name(s) AND address(es) of declarant(s)  
(See headnote\*)

I/We (e) Mr. L. Lucantonio, Company Secretary of  
616 St. Kilda Road, Melbourne, Victoria 3004

do solemnly and sincerely declare as follows:

~~XXXXXX We are the applicant(s)~~

(or, in the case of an application by a body corporate)

1. I am/We are authorized to make this declaration on behalf of the applicant(s).

~~XXXXXX We are the actual inventor(s) of the invention.~~

(or, where the applicant(s) is/are not the actual inventor(s))

2. (f)

JOSEPH JOHN PATRONI of 1 Eleanor Street, Preston West, Victoria  
Australia and MALCOLM ROY BRANDON of 14 Castella Street,  
Ivanhoe East, Victoria, Australia

is/are the actual inventor(s) of the invention and the facts upon which the applicant(s)  
is/are entitled to make the application are as follows:

(g)

Applicant is the assignee of Bunge Industrial Limited who is the  
assignee of the invention from the actual inventors

(Note: Paragraphs 3 and 4 apply only to Convention applications)

3. The basic application(s) for patent or similar protection on which the application is based  
is/are identified by country, filing date, and basic applicant(s) as follows:

(h)

4. The basic application(s) referred to in paragraph 3 hereof was/were the first application(s)  
~~made in a Convention country in respect of the invention the subject of the application.~~

(k) Insert PLACE of signing

Declared at (k)

MELBOURNE, VIC.

(l) Insert DATE of signing

Dated (l)

6th December, 1989

(m) Signature(s) of declarant(s)

(m)

MR. L. LUCANTONIO  
Company Secretary

Note: No legalization or other witness required

To: The Commissioner of Patents

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**(12) PATENT ABRIDGMENT      (11) Document No. AU-B-66874/86**  
**(19) AUSTRALIAN PATENT OFFICE      (10) Acceptance No. 597924**

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(54) Title  
**SOLUBILIZATION OF PROTEIN AGGREGATES**

International Patent Classification(s)  
(51)<sup>4</sup> **C07K 003/12**

(21) Application No. : **66874/86**                      (22) Application Date : **11.12.85**

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(43) Publication Date : **18.06.87**

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(60) Related to Provisional(s) : **PH3818**

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**BUNGE (AUSTRALIA) PTY. LTD.**

(72) Inventor(s)  
**JOSEPH JOHN PATRONI; MALCOLM ROY BRANDON**

(74) Attorney or Agent  
**PHILLIPS,ORMONDE & FITZPATRICK**

(56) Prior Art Documents  
**AU 77927/87**  
**AU 37973/78**

(57) Claim

1.            A method for the recovery of proteins which method includes providing a source of protein in an insoluble form, a source of at least one cationic surfactant; treating the insoluble protein with the at least one cationic surfactant in an amount sufficient to effect solubilisation without substantial modification to the structural backbone of the protein.

3.            A method according to claim 2 wherein the at least one cationic surfactant is present in an amount exceeding the critical micelle concentration.

6.            A method according to claim 5 wherein the cationic surfactant is selected from surfactants including a cation selected from

Cetyl trimethylammonium cations,  
Cetyl pyridinium cations,  
Tetradecyl trimethylammonium cations,

(11) AU-B-66874/86  
(10) 597924

-2-

Dodecyl trimethylammonium cations,  
Mixed n-alkyl dimethyl benzyl ammonium cations,  
N,N-dimethyl-N-[2-[2-[4-(1,1,3,3,-tetramethyl butyl)  
phenoxy]ethoxy]ethyl]benzenemethanaminium cations.

8. A method for the recovery of proteins which method includes providing a source of proteins in an insoluble form, a source of at least one cationic surfactant, and a source of at least one polar organic solvent; treating the insoluble protein with a mixture of from 5 to 70% volume/volume of the at least one polar organic solvent and at least one cationic surfactant in an amount and sufficient to effect solubilisation without substantial modification to the structural backbone of the protein; and separating the solubilised protein from the resulting solution.

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# COMPLETE SPECIFICATION

(ORIGINAL)

Application Number: 66874/86.

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Class

Int. Class

Complete Specification Lodged:

Accepted:

Published:

Priority

Related Art:

This document contains the amendments made under Section 49 and is correct for printing.

APPLICANT'S REF.: CAP of PH 3818

Name(s) of Applicant(s):

BUNGE (AUSTRALIA) PTY. LTD.

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JOSEPH JOHN PATRONI  
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PHILLIPS, ORMONDE AND FITZPATRICK  
Patent and Trade Mark Attorneys  
367 Collins Street  
Melbourne, Australia, 3000

Complete Specification for the invention entitled:

"SOLUBILIZATION OF PROTEIN AGGREGATES"

The following statement is a full description of this invention, including the best method of performing it known to applicant(s):

This invention relates to methods for recovery in soluble form of proteins initially present as insoluble aggregates.

Recombinant DNA technology provides potentially extremely valuable means of synthesizing large amounts of desirable eukaryotic (usually mammalian) proteins such as hormones, interferons, and enzymes. Although it has proved to be relatively easy to manipulate organisms such as bacteria in order to produce the desired protein, the host organism does not normally secrete the protein product into the culture medium. Thus lysis of the organisms (for example bacteria), followed by isolation of the desired protein is usually necessary.

With few exceptions, eukaryotic proteins expressed at high levels in Escherichia coli (E.coli) are present within the host cell in the form of insoluble protein aggregates (Marston, F.A.O., Lowe, P.A., Doel, M.T., Schoemaker, J.M., White, S., and Angal, S., Bio/Technology, 2 (9), 1984, 729). For example, human insulin produced by E. coli is present in the bacterial cells as morphologically characteristic inclusion bodies (Williams, D.C., Van Frank, R.M., Muth, W.L., and Burnett, J.P., Science 215, 1982, 687). These inclusion bodies may occupy a large proportion of the cell volume, and appear to be sites of localised accumulation of the eukaryotic proteins produced by precipitation and aggregation within the cell (ibid).

Whilst the insoluble aggregates can be usefully recovered in high yield by mechanical means, for example, by centrifugation or filtration, it is then usually necessary to render these proteins soluble prior to further purification.

According to the prior art, two principal methods have been used, both of which involve solubilisation with concomitant denaturation of the protein, followed by gradual removal of the denaturant agent; the protein can then slowly renature. In one of these methods solubilisation and denaturation is effected using very high concentrations (6-9M) of compounds such as urea or guanidine hydrochloride in water (International Patent Application WO 83/04418); in the other, the protein is solubilised and reversibly denatured by treatment with alkali at high pH and then

returned to a lower non-denaturing pH to allow the protein to renature. The two methods may be combined to give a further improvement in recovery (GB 8407570). However, even with the combined technique the recovery is still only of the order of 30% (GB 8407570).

These methods suffer from serious disadvantages. Those using urea or guanidine hydrochloride utilize reagents which may be very expensive, difficult and hazardous to handle, difficult to remove, ecologically harmful and are required in very large quantities. Solutions of these reagents must either be recycled or disposed of at additional expense. As a consequence these disadvantages tend to detract from their future large scale application (Emtage, J.S., Nature, 317, 1985, 185). Moreover, guanidine hydrochloride may be deleterious to the immunogenicity of the protein.

Alkaline solubilisation using for example sodium or potassium hydroxide is a cheaper alternative, but offers little real advantage in convenience; moreover, at high pH, alkaline aqueous solutions are reactive, causing irreparable damage to the dissolved protein molecule. In both methods, renaturation of the desired protein may be incomplete, resulting in poor recovery, and the processes are slow and tedious.

Accordingly it is an object of the present invention to overcome, or at least alleviate, one or more of the difficulties related to the prior art.

Accordingly, in a first aspect, there is provided a method for the recovery of proteins which method includes providing a source of protein in an insoluble form, a source of at least one cationic surfactant; treating the insoluble protein with the at least one cationic surfactant in an amount sufficient to effect solubilisation without substantial modification to the structural backbone of the protein.

Desirably the amount of the at least one cationic surfactant exceeds the critical micelle concentration.

The present invention is particularly applicable to the solubilization and recovery of biologically active proteins synthesized by microorganisms and eukaryotic cell

lines which have been modified by recombinant DNA technology. The protein aggregate may comprise an inclusion body isolated by disruption or lysis of a host cell which may have been transformed or transfected with a vector including a gene coding for the protein. However, it is not restricted thereto. In addition, the present invention is applicable to naturally occurring aggregate protein complexes common to numerous biological systems.

The protein aggregates which may be recovered according to the present invention may be selected from inclusion bodies and cytoplasmic aggregates. The inclusion bodies may be selected from biologically active polypeptides and peptides including growth hormones, interferons, immunogens and lymphokines. As used herein the term "structural backbone" refers to the primary and secondary structure of the protein.

Preferably the at least one cationic surfactant is present in amount of from approximately 2.5 to 50% weight/volume, more specifically 2.5 to 20% weight/volume. The upper limit of surfactant content may vary due to solubility limits of the selected surfactant.

We have now found that it is possible to solubilise aggregates of desired proteins, including inclusion bodies, by treatment of the insoluble form with a cationic surfactant in water, either in the presence or absence of a polar solvent or with the polar solvent alone. The process is rapid (5-60 min), and recovery of the solubilised protein can be optimally effected very easily. Only small quantities of inexpensive reagents, which are readily available and recyclable, are required. For example, the bulk of the solubilising agent may be water.

According to a preferred aspect of the present invention there is provided a method for the recovery of proteins which method includes providing a source of proteins in an insoluble form, a source of at least one cationic surfactant, and a source of at least one polar organic solvent; treating the insoluble protein with a mixture of from approximately 5 to 70% volume/volume of the at least one polar organic solvent and at least one cationic surfactant in an amount sufficient to effect solubilisation without



substantial modification to the structural backbone of the protein; and separating the solubilised protein from the resulting solution.

The protein may be maintained in an aqueous solution comprising a polar organic solvent and suitable buffering salts. The presence of a polar organic solvent, such as acetonitrile, preferably at a concentration of 5 to 70% alters the interaction between the insoluble protein and the aqueous solvent, thereby increasing the solubility of the hydrophobic regions of the protein. More preferably the concentration of the organic solvent is 10 to 20%.

Moreover, the incorporation of a cationic surfactant, such as a quaternary ammonium compound, at a level exceeding the critical micelle concentration and sufficient to overcome the associative forces within the aggregate, is highly advantageous and promotes the segregation, disruption and solubilization of the inclusion body constituents.

According to another embodiment of the invention, the addition of an aqueous solution of a suitable surfactant to a dried powder or an aqueous slurry of the protein aggregates is also a highly desirable and efficient means of solubilisation, particularly of whole inclusion bodies. The addition of the at least one cationic surfactant is at levels above the critical micelle concentration and within the limit of its solubility and economy. This procedure is highly desirable for its simplicity; moreover, in contrast to some prior art, the protein may be solubilised in a mild near neutral environment. Due to the nature of the solubilising agent, the recovery method is compatible with later processing steps, in contradistinction to the severe solubilisation treatments of the prior art. The solubilising agent has been found to be compatible with other ingredients utilising the encountered processing of protein aggregates. For example dithiothreitol, mercaptoethanol, reduced glutathione, dimethylsulfone, urea, sodium and potassium hydroxides.

The scope of the invention comprehends the use of all suitable single and multiple chain quaternary nitrogen or phosphorous compounds with various head groups, counter ions

and branched or derivatised carbon chains.

Preferably the at least one cationic surfactant is selected from

Cetyl trimethylammonium halide, e.g. bromide,  
Cetyl pyridinium halide, e.g. chloride,  
Tetradecyl trimethylammonium halide, e.g. bromide,  
Dodecyl trimethylammonium halide, e.g. bromide,  
Mixed n-alkyl dimethyl benzyl ammonium halide, e.g. chloride (50% C-14, 40% C-12, 10% C-16),  
N,N-dimethyl-N-[2-[2-[4-(1,1,3,3,-tetramethyl butyl) phenoxy]ethoxy]ethyl]benzenemethanaminium halide, e.g. chloride.

It should be understood, however, that the selection of the halide ions is illustrative only. The identity of the anion is unimportant. For example the halide may be substituted by other anions, e.g. sulfonates e.g. p-toluene sulfonates.

More preferably the cationic surfactant is cetyl trimethylammonium bromide or cetyl pyridinium chloride.

It is preferred that the cationic surfactant selected is one which does not absorb in the region of the ultraviolet spectrum where polypeptide absorbance is maximal, e.g. cetyl trimethylammonium bromide.

The invention provides significant economic advantages in large scale purification systems. As the cationic surfactants are considered to be active against bacteria, fungi and viruses (Goodman, L.S., and Gilman, S., "The Pharmacological Basis of Therapeutics", 5th Ed., Macmillan Publishing Co. Inc., N.Y., 1975, pl001), the risk of contamination, sample degradation, and the need for complex and expensive sterile containment facilities are all reduced. Furthermore, only low concentrations of the solubilising agents are required and these may be readily removed from solution by either chromatography, dialysis, Krafft point crystallisation or, additionally, in the case of the organic solvent, by distillation.

The method according to the present invention may include the further step of separating the solubilised protein from the resulting solution.

The separation step may include differential elution

of the solubilised protein through a chromatographic column, dialysis, ultrafiltration, differential precipitation, or ligand specific isolation. The chromatographic column may be a high performance liquid chromatography (HPLC) column, optionally a reversed phase high performance liquid chromatography (RP-HPLC) column. A column sold under the trade designation TSK-GEL (LC) and available from Toyo Soda Manufacturing Co. Ltd. (Japan) or Ultrapore RPSC and available from (Beckman Instruments Inc. (California, United States of America) have been found to be suitable. Due to the nature of the solubilising agent, the separation step may be conducted utilizing other known forms of chromatography including chromatography of the molecular sieve type, e.g. gel filtration chromatography, anion exchange chromatography, hydrophobic interaction chromatography, and ligand specific chromatography. Preferably the chromatography eluant is an aqueous solution of a cationic surfactant. A dilute solution may be used. The cationic surfactant may be present in amounts of from approximately 0.25% weight/volume to approximately 2.0% weight/volume, more preferably 0.4% weight/volume.

It will be understood that the chromatographic separation also functions to purify the protein product.

It will be understood that the method according to the present invention may be utilised in a method for the analysis of a polypeptide sample wherein the sample to be tested is subjected to the recovery process thereof. The results may provide a quantitative analysis of the composition of the polypeptide sample.

Embodiments of the present invention will now be illustrated by way of example only with reference to the following non-limiting examples, and the accompanying figures.

In the following examples the transmission electron micrographs were generated as follows:

A small portion of the wet pellet was fixed and embedded into L.R. White resin (London Resin Co., U.K.) and the block sectioned for inspection by electron microscopy using a Philips EM-300 transmission electron microscope.

#### EXAMPLE 1

#### Solubilisation of Natural Porcine Growth Hormone

Water-insoluble, lyophilised natural porcine growth hormone isolated in a manner similar to that previously reported<sup>(i)</sup> (50mg) was vigorously agitated (30 min, 25°) with an aqueous solution of cetyl trimethylammonium bromide (1.0ml of 10%) and B-mercaptoethanol (1%). The clarified solution was then centrifuged (13,000 r.p.m., 5 min) on a Beckman Microfuge<sup>TM</sup> 11 to give a clear supernatant. A 1.0ml sample of this solution gave a UV absorbance of 72 (A<sub>1cm</sub><sup>280nm</sup>), indicating that solubilisation had occurred.

#### EXAMPLE 2A

##### Solubilization of Synthetic Porcine Growth Hormone

An experiment was conducted with inclusion bodies obtained from transformed E.coli cells. The inclusion bodies, containing 1-190AA methionine-porcine growth hormone, derived from a plasmid, pMG935 were isolated, after cell disruption, by centrifugation. The insoluble protein pellet was washed twice with an aqueous solution of Triton X-100 (0.5%) and EDTA (10 mM) and twice with aqueous EDTA (5 mM); the final pellet was then lyophilised for storage.

The results are illustrated in Figure 1.

Figure 1 represents a transmission electron micrograph of methionine-porcine growth hormone inclusion bodies. Clearly visible inclusion bodies of some 0.2 to 0.5 micron in diameter were present along with some fibrous contaminant, most likely cell debris. Magnification is x 6,000.

A portion of the lyophilised inclusion bodies (50mg) was then treated with aqueous cetyl trimethylammonium bromide (1.0ml of 10%) in a test tube, and the mixture agitated for 1 hour at room temperature. The mixture was centrifuged (2000 r.p.m., 10 min) on a Beckman Microfuge II to give a clear supernatant and a small insoluble pellet. A small portion of the pellet was fixed and embedded into L.R. White resin and the block section sectioned for comparison by electron microscopy with the untreated material.

The results are shown in Figure 2. Figure 2 represents a transmission electron micrograph of the insoluble residue remaining following solubilization treatment. Magnification is x 4,460. In marked contrast to the untreated material, no inclusion bodies were to be seen after the solubilisation procedure.

#### EXAMPLE 2B

A similar experiment was conducted to that described in Example 1A above whereby lyophilised inclusion bodies containing the variant 4-190AA porcine growth hormone derived from a plasmid pMG936 (50mg) were treated with cetyl trimethylammonium bromide (1.0ml of 10%). The mixture was agitated (1h, room temperature) and centrifuged to give a clear supernatant and a small insoluble pellet. A 1.0ml sample of the supernatant gave an estimated UV absorbance of  $60(A_{280\text{nm}}^{1\text{cm}})$  indicating that solubilisation had occurred.

#### EXAMPLE 3

In another experiment, the lyophilised inclusion bodies containing either 1-190AA methionine or the variant 4-190AA porcine growth hormone (50 mg) were treated with a mixture of acetonitrile (0.2 ml), aqueous buffer (0.1 M glycine, pH 8.5; 0.8 ml) and aqueous cetyl trimethylammonium bromide (0.5 ml of 10%) in a test tube at 25°C. The mixtures were agitated for 1 hour. Again, as with the previous experiment, substantial solubilization of the inclusion bodies was observed.

#### EXAMPLE 4

Lyophilised inclusion bodies containing 1-190AA methionine-porcine growth hormone (50mg) were vigorously agitated with an aqueous solution of one of the cationic surfactants listed below (1.0ml of 20% in 0.1M TRIZMA, pH 10.0) in a test tube at 25°C. The mixtures were agitated for 30 minutes and then centrifuged on a Beckman Microfuge II for 30 minutes to give clear pale yellow supernatants with negligible pellets in all cases. Again, as with the previous experiments, substantial solubilisation of the inclusion bodies was achieved in each case.

- (a) Cetyl pyridinium chloride,
- (b) Tetradecyl trimethylammonium bromide,
- (c) Dodecyl trimethylammonium bromide,
- (d) Mixed n-alkyl dimethyl benzyl ammonium chloride (50% C-14, 40% C-12, 10% C-16),
- (e) N,N-dimethyl-N-[2-[2-[4-(1,1,3,3,-tetramethylbutyl)phenoxy]ethoxy]ethyl]benzenemethanaminium chloride.

#### EXAMPLE 5

##### Solubilisation of Human Malaria Parasite Antigen Fusion Protein

The B-galactosidase fusion protein (Ag 13) containing a surface antigen from the human malaria parasite Plasmodium falciparum was expressed in E.coli as unusual almost transparent inclusion bodies (Fig. 3) and was obtained as a gift from Dr. R.F. Anders<sup>(ii)</sup>.

The wet inclusion body paste (800 mg) containing Ag13, a polypeptide of molecular weight 156kDa, was vigorously agitated with an aqueous solution of cetyl trimethylammonium bromide (1.0ml of 19% in 50mM di-sodium ethylenediaminetetraacetic acid (EDTA) and 0.15M tris(hydroxymethyl)aminomethane (TRIZMA), pH10.0), a cetyl pyridinium bromide (0.25ml of 10% in 50mM EDTA and 0.15M TRIZMA, pH10.0) and dithiothreitol (5%) at room temperature.

After some 45 min the initially opaque mixture appeared transparent. The mixture was then centrifuged (13,000rpm, 10 min.) on a Beckman Microfuge II to give a clear, pale yellow supernatant and a small off-white pellet. The pellet contained no visible remnants of the inclusion bodies upon inspection by electron microscopy.

#### EXAMPLE 6

##### Solubilisation of Infectious Bursal Disease Virus Antigen Fusion Protein

Inclusion bodies (Fig. 4) containing the D1 fragment of the 32kDa structural protein from infectious bursal disease virus were obtained as a gift from Dr. K.J. Fahey<sup>(iii)</sup>. The inclusion bodies were isolated from transformed E.coli in a manner similar to that previously described<sup>(iv)</sup>.

The wet inclusion body paste (100 mg) containing the expressed viral fusion protein was vigorously agitated with an aqueous solution of cetyl trimethylammonium bromide (0.75ml of 19% in 50mM EDTA and 0.15M TRIZMA), cetyl pyridinium chloride (0.5ml of 10% in 50mM EDTA and 0.15M TRIZMA) and dithiothreitol (5%) at room temperature. After some 30 min the initially opaque mixture appeared completely transparent.

The mixture was then centrifuged (13,000rpm, 10min)

on a Beckman Microfuge II to give a clear, yellow supernatant and negligible pellet indicating near complete solubilisation. An immuno-dot blot analysis of the supernatant using nitro-cellulose paper and a monoclonal antibody to the D1 polypeptide confirmed the preservation of the active antigenic site on the fused polypeptide.

#### EXAMPLE 7

Reversed-phase H.P.L.C. on C3 or C1 Alkyl-bonded, Large Pore Silica. The chromatograms illustrated in Figures 5 and 6 were obtained by following the protocol outlined below.

Operating parameters are described as follows:

Elution was performed at a flow rate of 0.8 ml/min. at room temperature with water/acetonitrile mixtures containing 0.1% v/v trifluoroacetic acid as modifier: a stepwise linear gradient was constructed from 100% water (0.1 min) to 75% water/acetonitrile over 5 min., to 45% water/acetonitrile over 15 min., to 100% acetonitrile over 5 min. The 100% acetonitrile eluant was maintained for a further 10 min. before re-equilibration prior to the next injection. The composition of the solvent and flow rate may be varied slightly to achieve the desired resolution.

Solutions were all degassed and filtered (0.45um); injection volumes of 3ul were used for the chromatograms displayed in Figures 5 and 6. The injection volume can, however, be varied to suit, according to the detection limits and protein concentrations available.

H.P.L.C. was performed on a system comprising two Beckman 114M System Delivery Modules coupled to a 20 ul loop injector and a Beckman 421 Controller. Detection was by UV at 270nm fixed wavelength on a Beckman 165 Variable Wavelength Detector.

- (i) Reichert, L.E., JR., Methods in Enzymology, XLIII, 1975, 360.
- (ii) Coppel, R.L., Cowman, A.F., Anders, R.F., Bianco, A.E., Saint, R.B., Lingelbach, K.R., Kemp, D.J., and Brown, G.V., Nature, 310, 1984, 789.
- (iii) Azad, A.A., Fahey, K.J., Barrett, S.A., Erny, K.M., and Hudson, P.J., Virology, 149, 1986, 190.
- (iv) Stanley, K.K., and Luzio, J.P., The EMBO Journal 3,

1984, 1429.

The following words used hereinbefore are registered trade marks: TRITON, L.R. WHITE.

It will be clearly understood that the invention in its general aspects is not limited to the specific details referred to hereinabove.

1984, 1429.

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1984, 1429.



The claims defining the invention are as follows:

1. A method for the recovery of proteins which method includes providing a source of protein in an insoluble form, a source of at least one cationic surfactant; treating the insoluble protein with the at least one cationic surfactant in an amount sufficient to effect solubilisation without substantial modification to the structural backbone of the protein.

2. A method according to claim 1 wherein the solubilisation is conducted in aqueous solution.

3. A method according to claim 2 wherein the at least one cationic surfactant is present in an amount exceeding the critical micelle concentration.

4. A method according to claim 3 wherein the at least one cationic surfactant is present in an amount of from approximately 2.5% to 50% weight/volume.

5. A method according to claim 4 wherein the cationic surfactant is a quaternary ammonium compound and the insoluble protein is a protein aggregate.

6. A method according to claim 5 wherein the cationic surfactant is selected from surfactants including a cation selected from

Cetyl trimethylammonium cations,  
Cetyl pyridinium cations,  
Tetradecyl trimethylammonium cations,  
Dodecyl trimethylammonium cations,  
Mixed n-alkyl dimethyl benzyl ammonium cations,  
N,N-dimethyl-N-[2-[2-[4-(1,1,3,3,-tetramethyl butyl) phenoxy]ethoxy]ethyl]benzenemethanaminium cations.

7. A method according to claims 6 wherein the cationic surfactant is cetyl trimethylammonium bromide.

8. A method for the recovery of proteins which method includes providing a source of proteins in an insoluble form, a source of at least one cationic surfactant, and a source of at least one polar organic solvent; treating the insoluble protein with a mixture of from 5 to 70% volume/volume of the at least one polar organic solvent and at least one cationic surfactant in an amount and sufficient to effect solubilisation without substantial modification to the structural backbone of the protein; and separating the



solubilised protein from the resulting solution.

9. A method according to claim 8 wherein the at least one cationic surfactant is present in an amount exceeding the critical micelle concentration.

10. A method according to claim 9 wherein the at least one cationic surfactant is present in an amount of from 2.5% to 50% weight/volume.

11. A method according to claim 10 wherein the cationic surfactant is a quaternary ammonium compound and the insoluble protein is a protein aggregate.

12. A method according to claim 11 wherein the cationic surfactant is selected from surfactants including a cation selected from

Cetyl trimethylammonium cations,  
Cetyl pyridinium cations,  
Tetradecyl trimethylammonium cations,  
Dodecyl trimethylammonium cations,  
Mixed n-alkyl dimethyl benzyl ammonium cations,  
N,N-dimethyl-N-[2-[2-[4-(1,1,3,3,-tetramethyl butyl) phenoxy]ethoxy]ethyl]benzenemethanaminium cations.

13. A method according to claim 12 wherein the cationic surfactant is cetyl trimethylammonium bromide.

14. A method according to claim 11 wherein the separation step includes differential elution of the solubilised protein through a chromatographic column.

15. A method according to claim 14 wherein the eluant is a dilute aqueous solution of a cationic surfactant.

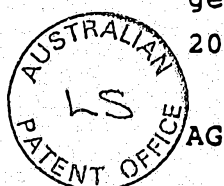
16. A method according to claim 11 wherein the separation step includes differential precipitation, dialysis, ultrafiltration and/or ligand specific isolation.

17. A method according to claim 16 wherein the at least one cationic surfactant is cetyl trimethylammonium bromide and the at least one polar organic solvent is acetonitrile.

18. A method according to claim 17 wherein the protein aggregate is an inclusion body isolated by lysis of a host cell.

19. A method according to claim 18 wherein the host cell has been transformed or translated with a vector including a gene coding for the protein.

20. A method according to claim 18 wherein the inclusion



body is selected from biologically active polypeptides and peptides including growth hormones, interferons, immunogens and lymphokines.

21. A protein whenever prepared by a method according to any one of the previous claims.

22. A method for the analysis of a polypeptide sample which method includes providing a sample to be tested and subjecting the sample to a method according to claim 8.

23. A method according to claim 22 wherein the separation step includes differential elution of the solubilised protein through a chromatographic column.

24. A method according to claim 1 substantially as hereinbefore described with reference to any one of the examples.

DATED: 23 March, 1990

PHILLIPS, ORMONDE AND FITZPATRICK

Attorneys for:

BUNGE (AUSTRALIA) PTY. LTD.

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