

(19) World Intellectual Property Organization
International Bureau



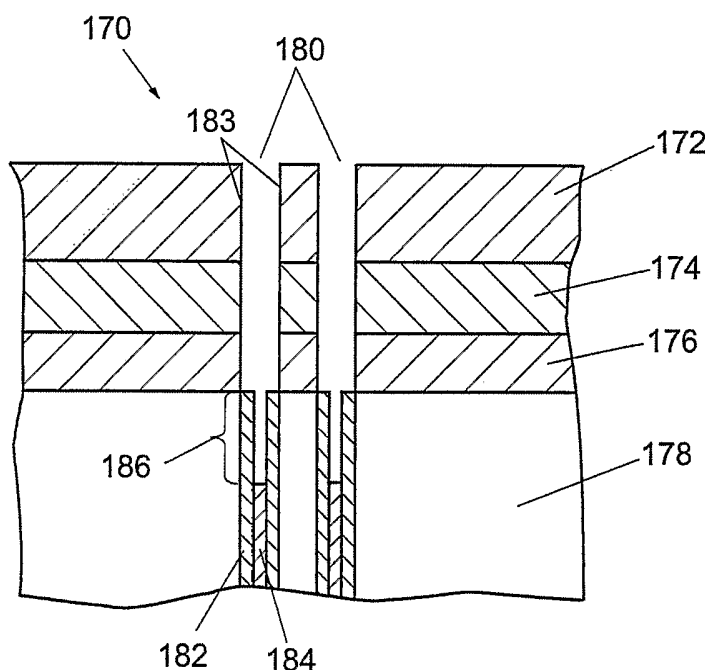
(43) International Publication Date
19 January 2006 (19.01.2006)

PCT

(10) International Publication Number
WO 2006/007005 A1

- (51) International Patent Classification⁷: **H01L 21/311** (74) Agents: **FREI, Donald, F.** et al.; Wood, Herron & Evans, L.L.P., 2700 Carew Tower, Cincinnati, OH 45202 (US).
- (21) International Application Number: PCT/US2005/013885 (81) Designated States (*unless otherwise indicated, for every kind of national protection available*): AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KM, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NA, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, SM, SY, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW.
- (22) International Filing Date: 22 April 2005 (22.04.2005)
- (25) Filing Language: English
- (26) Publication Language: English
- (30) Priority Data:
10/881,456 30 June 2004 (30.06.2004) US
- (71) Applicant (*for all designated States except US*): **TOKYO ELECTRON LIMITED** [JP/JP]; TBS Broadcast Center, 3-6, Akasaka 5-chome, Minato-ku, Tokyo 107 (JP).
- (71) Applicant (*for JP only*): **TOKYO ELECTRON AMERICA, INC.** [US/US]; 2400 Grove Blvd., Austin, TX 78741-6500 (US).
- (72) Inventors; and
- (75) Inventors/Applicants (*for US only*): **GALE, Glenn, W.** [US/JP]; Yonbancho Plaza #903, 9-6 Yonbancho, Chiyoda-ku, Tokyo 102-0081 (JP). **HILLMAN, Joseph, T.** [US/US]; 7777 E. Main Street #331, Scottsdale, AZ 85251 (US). **JACOBSON, Gunilla** [SE/US]; 156 Mosher Way, Palo Alto, CA 94304 (US). **PALMER, Bentley** [CA/US]; 14410 S. 40th Street, Phoenix, AZ 85044 (US).
- (84) Designated States (*unless otherwise indicated, for every kind of regional protection available*): ARIPO (BW, GH, GM, KE, LS, MW, MZ, NA, SD, SL, SZ, TZ, UG, ZM, ZW), Eurasian (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European (AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IS, IT, LT, LU, MC, NL, PL, PT, RO, SE, SI, SK, TR), OAPI (BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG).
- Published:
— with international search report
- For two-letter codes and other abbreviations, refer to the "Guidance Notes on Codes and Abbreviations" appearing at the beginning of each regular issue of the PCT Gazette.*

(54) Title: A SYSTEM AND METHOD FOR PROCESSING A SUBSTRATE USING SUPERCRITICAL CARBON DIOXIDE PROCESSING



(57) Abstract: A method and system for processing a substrate (178, 265) in a film removal system (200, 201). The method includes providing the substrate (178, 265) in a substrate chamber (250) of a film removal system (200, 201), where the substrate (178, 265) has a micro-feature containing a dielectric film (182) on a sidewall (183) of the micro-feature (170) and a photoresist film (184) covering a portion the dielectric film (182), and performing a first film removal process using supercritical CO₂ processing to remove the portion (186) of the dielectric film (182) not covered by the photoresist film (184). Following the first film removal process, a second film removal process using supercritical CO₂ processing can be performed to remove the photoresist film (184). Alternately, wet processing can be used to perform one of the first film removal process or the second film removal process.

TITLE OF THE INVENTION

A SYSTEM AND METHOD FOR PROCESSING A SUBSTRATE USING
SUPERCRITICAL CARBON DIOXIDE PROCESSING

FIELD OF THE INVENTION

[0001] The invention relates to semiconductor manufacturing, and more particularly, to utilizing supercritical CO₂ processing to remove a film from a micro-feature on a substrate.

BACKGROUND OF THE INVENTION

[0002] In the semiconductor industry, the minimum feature sizes of microelectronic devices are approaching the deep sub-micron regime to meet the demand for faster, lower power microprocessors and digital circuits. In the manufacturing of a trench capacitor for a dynamic random access memory (DRAM) device, a deep trench (DT) is etched several microns (μm) into a silicon substrate. During manufacturing of a deep trench capacitor, a dielectric film, such as doped silicon dioxide film (e.g., arsenic-doped silicon dioxide, also referred to as arsenosilicate glass (ASG)), is deposited on the sidewalls of the trench, in order to provide out-diffusion of the dopant (e.g., arsenic, As) from the doped dielectric film into the sidewalls of the silicon trench to form one plate of the capacitor.

[0003] In current trench capacitor technology, the width of the trench can be about 0.2 microns, or less, and the trench depth to diameter aspect ratio can be as great as about 50:1, or even greater. Due to these aggressive trench dimensions, it can be difficult to process films located in the trench. In addition to utilizing a trench with straight vertical sidewalls, current trench capacitor technology may use a "bottle-shaped" trench, in which the bottom portion of the trench is etched to be wider than the top portion of the trench, in order to increase the capacitor surface area. This presents further difficulties for processing films located in the trench.

SUMMARY OF THE INVENTION

[0004] A system and method are provided for removing a dielectric film and a photoresist film from a micro-feature on a substrate. A method is provided for processing a substrate having a micro-feature containing a dielectric film on the sidewalls of the micro-feature and a photoresist film covering a portion of the dielectric film, and performing a first film removal process on the substrate using supercritical CO₂ processing, wherein the portion of the dielectric film not covered by the photoresist film is removed.

[0005] In one embodiment of the invention, the method further contains a second film removal process that can be performed on the substrate using supercritical CO₂ processing to remove the photoresist film, wherein the second film removal process is performed following the first film removal process.

[0006] In another embodiment of the invention, the second film removal process can be performed on the substrate using wet processing to remove the photoresist film, wherein the second film removal process is performed following the first film removal process.

[0007] In yet another embodiment of the invention, a method is provided for processing a substrate having a micro-feature containing a dielectric film on the sidewalls of the micro-feature and a photoresist film covering a portion of the dielectric film, performing a first film removal process on the substrate using wet processing, wherein the portion of the dielectric film not covered by the photoresist film is removed, and performing a second film removal process on the substrate using supercritical CO₂ processing to remove the photoresist film, wherein the second film removal process is performed following the first film removal process.

[0008] A film removal system is provided for processing a substrate, the system including a substrate transfer system configured for transferring the substrate within the film removal system, a substrate chamber configured for performing a supercritical CO₂ film removal process on a substrate having a micro-feature containing a dielectric film and a photoresist film covering a portion of the dielectric film, wherein the supercritical CO₂ film removal

process includes at least one of a first film removal process to remove the dielectric film not covered by the photoresist film and a second film removal process to remove the photoresist film following the first film removal process, a CO₂ supply system configured for pressurizing the substrate chamber with supercritical CO₂ fluid, a solvent supply system configured for delivering a solvent to the substrate chamber, and a controller configured for controlling the film removal system.

BRIEF DESCRIPTION OF THE DRAWINGS

[0009] In the drawings:

[0010] FIG. 1A schematically shows a cross-sectional view of a micro-feature containing a trench having a dielectric film and a photoresist film covering a portion of the dielectric film according to an embodiment of the invention;

[0011] FIG. 1B schematically shows a cross-sectional view of the micro-feature in FIG. 1A following removal of the portion of the dielectric film not covered by the photoresist film according to an embodiment of the invention;

[0012] FIG. 1C schematically shows a cross-sectional view of the micro-feature in FIG. 1B following removal of the photoresist film according to an embodiment of the invention;

[0013] FIG. 2A is a schematic diagram showing a film removal system according to an embodiment of the invention;

[0014] FIG. 2B is a schematic diagram showing a film removal system according to another embodiment of the invention;

[0015] FIG. 3A is a flow diagram showing a simplified sequence of a dielectric film removal process according to an embodiment of the invention;

[0016] FIG. 3B is a flow diagram showing a simplified sequence of a photoresist film removal process according to another embodiment of the invention; and

[0017] FIG. 4 shows a general-purpose computer that may be used to implement embodiments of the invention.

DETAILED DESCRIPTION OF SEVERAL EMBODIMENTS OF THE INVENTION

[0018] The term micro-feature, as used herein, refers to a feature formed in a substrate and/or in a layer or layers formed on a substrate that has dimensions on the micrometer scale, and typically the sub-micron scale, i.e., less than 1 μm . FIG. 1A schematically shows a cross-sectional view of a micro-feature containing a trench having a dielectric film on the sidewalls of the trench and a photoresist film covering a portion of the dielectric film according to an embodiment of the invention. The micro-feature 170 contains a hard mask film 172 (e.g., borosilicate glass, BSG), a pad nitride film 174, a pad oxide film 176, and a silicon substrate 178. The micro-feature 170 further contains a trench 180 that is formed by etching through the films 172 -176 and into the silicon substrate 178.

[0019] The trench 180 can be formed using a photolithographic process and dry etching techniques that are well known to persons skilled in the art of lithography and plasma etching. The exemplary trench 180 may have a width of about 0.2 micron, or less, and the trench aspect ratio may be 50:1, or even greater. In FIG. 1A, the part of the trench 180 etched in the silicon substrate 178 contains a dielectric film 182 on the sidewalls 183 and a photoresist film 184 covering the dielectric film 182 except for portion 186 of the dielectric film 182. The dielectric film 182 can, for example, be an arsenic-doped silicon dioxide film.

[0020] Methods for depositing the dielectric film 182 and the photoresist film 184 are well known to persons skilled in the art. For example, the photoresist film 184 can be formed by spin-coating a photoresist solution into the trench 180 and subsequently baking the photoresist solution. Then, an oxygen plasma may be used to recess the photoresist film 184 to the level where the dielectric film should terminate in the trench. Next, the portion 186 of the dielectric film 182 is removed from the trench 180.

[0021] Following removal of the exposed dielectric film 182 from the trench 180, the photoresist film 184 is removed from the trench 180. Removal of the photoresist film 184 must be performed without significantly affecting other

materials in the micro-feature 170, including the pad oxide 176, the pad nitride 174, the hard mask 172, the remaining dielectric film 182, and silicon on the trench sidewalls 183.

[0022] An embodiment of the invention provides a method for selectively removing films from a micro-feature using supercritical carbon dioxide (CO₂) processing. CO₂ fluid is in a supercritical state when above the critical temperature T_c of about 31°C and above the critical pressure P_c of about 1,070 pounds per square inch gauge (psig). Supercritical CO₂ fluid has virtually no viscosity or surface tension and has therefore no difficulty in penetrating all the way to the bottom of a deep trench or a hole and removing a film from the trench or the hole. Furthermore, supercritical CO₂ processing can avoid or reduce the use of hazardous and environmentally damaging wet chemicals that are frequently used at very high temperatures.

[0023] An additional benefit of supercritical CO₂ processing is the elimination of large volumes of water associated with the traditional wet processing, and the absence of sulfur residues that can result from inadequate rinsing during wet processing. Yet another benefit of supercritical CO₂ processing is the absence of watermarks that are frequently associated with wet processing in a trench, since no water or materials dissolved in the water need to be removed from the trenches. Furthermore, when processing a micro-feature, performing at least one film removal process using supercritical CO₂ processing can reduce overall processing time when compared to conventional wet processing.

[0024] FIG. 1B schematically shows a cross-sectional view of the micro-feature 170 in FIG. 1A following removal of the portion 186 of the dielectric film 182 not covered by the photoresist film 184 according to an embodiment of the invention. According to one embodiment of the invention, the portion 186 of the dielectric film 182 may be selectively removed from the trench 180 by exposing the micro-feature 170 to supercritical CO₂ fluid and a first solvent dissolved in the supercritical CO₂ fluid. The first solvent dissolved in the supercritical CO₂ is capable of breaking down/dissolving the exposed dielectric film 182 without significantly affecting the photoresist film 184 and other materials in the micro-feature 170. The first solvent can, for example,

contain HF_(aq) or HF:pyridine. The micro-feature 170 is kept in contact with the supercritical CO₂ and the first solvent until the portion 186 of the dielectric film 180 has been broken down/dissolved and removed from the micro-feature 170 with the supercritical CO₂ fluid.

[0025] FIG. 1C schematically shows a cross-sectional view of the micro-feature in FIG. 1B following removal of the photoresist film 184 according to an embodiment of the invention. In one embodiment of the invention, the photoresist film 184 may be exposed to supercritical CO₂ fluid and a second solvent dissolved in the supercritical CO₂ fluid to selectively remove the photoresist film 184 from the trench 180. The second solvent dissolved in the supercritical CO₂ is capable of breaking down/dissolving the photoresist film 184 without significantly affecting other materials in the micro-feature 170. The second solvent can, for example, include N-methyl pyrrolidone, diisopropyl amine, triisopropyl amine, or diglycol amine, or a combination of two or more thereof.

[0026] In addition, the second solvent can, for example, further contain one of the following chemicals: methanol, ethanol, isopropyl alcohol, benzyl alcohol, acetone, butylene carbonate, propylene carbonate, dimethylsulfoxide, γ -butyrolactone, dimethyl formamide, dimethyl acetamide, ethyl lactate, hydrogen peroxide, benzoyl peroxide, oxygen, ozone, nitric acid, acetic acid, or formic acid, or a combination of two or more thereof. As persons skilled in the art will appreciate, the invention is not limited to these solvents and chemicals, as many other solvents and chemicals may be used to carry out an embodiment of the invention for removing the photoresist film 184 from the trench 180. The micro-feature 170 is kept in contact with the mixture of supercritical CO₂ and the second solvent, until the photoresist film 184 has been broken down/dissolved and removed from the micro-feature 170 with the supercritical CO₂ fluid.

[0027] According to an embodiment of the invention, both (a) the portion 186 of the dielectric film 182, and subsequently, (b) the photoresist film 184 may be removed from the trench 180 by supercritical CO₂ processing.

[0028] According to another embodiment of the invention, supercritical CO₂ processing can be performed in series with wet processing to remove the portion 186 of the dielectric film 182 and the photoresist film 184. In other words, the portion 186 of the dielectric film 182 can be removed by supercritical CO₂ processing, and the photoresist film 184 can be removed by wet processing, or alternately, the portion 186 of the dielectric film 182 can be removed by wet processing, and the photoresist film 184 may be removed by supercritical CO₂ processing. Wet processing of the dielectric film 182 can, for example, utilize an acid bath such as aqueous hydrofluoric acid (HF_(aq)), HF:pyridine, or HF/NH₄F. Wet processing of the photoresist film 184 may, for example, utilize an aqueous mixture of sulfuric acid and hydrogen peroxide (SPM) that is heated to approximately 120°C, or higher, HF/HNO₃, or H₂O/O₃.

[0029] Obviously, embodiments of the invention are not limited to micro-features having a trench, as micro-features having other geometries, for examples holes or other complex geometries, can be processed according to embodiments of the invention.

[0030] FIG. 2A is a schematic diagram showing a film removal system according to an embodiment of the invention. The film removal system 200 in FIG. 2A includes a substrate chamber 250, a chamber heater 204 coupled to the substrate chamber 250, a CO₂ supply system 206, a circulation loop 208, a circulation pump 210, a solvent supply system 212, a separating vessel 214, a liquid/solid waste collection vessel 217, and a liquifying/purifying system 219. Substrate 265 is transferred within the film removal system 200 using a (robotic) substrate transfer system 269. The substrate 265 can be of any size, for example a 200 mm substrate, a 300 mm substrate, or an even larger substrate.

[0031] The substrate chamber 250 includes chamber housing 270, substrate holder 260, and processing zone 267 for processing the substrate 265. The substrate chamber 250 further includes injection nozzles 275 for introducing the supercritical CO₂ fluid into the processing zone 267. During the supercritical CO₂ processing, the substrate 265 is present in the processing zone 267, where a supercritical CO₂ fluid is used in conjunction with a solvent to remove a film from a micro-feature on the substrate 265.

The process chamber heater 204 heats the substrate chamber 250 and may be a heating blanket.

[0032] The CO₂ supply system 206 includes a CO₂ supply vessel 216, a particle filter 225, a CO₂ pump 218, and a CO₂ heater 220. The solvent supply system 212 includes chemical solvent vessels 222 and 224, and first and second high pressure injection pumps 226 and 228.

[0033] The CO₂ supply vessel 216 is coupled to the circulation loop 208 via the CO₂ piping 230. The CO₂ piping 230 includes the heater 220 located between the CO₂ pump 218 and the circulation loop 208. The circulation pump 210 is located on the circulation loop 208, and the circulation loop 208 couples to the substrate chamber 250 at a circulation inlet 232 and at a circulation outlet 234. The solvent supply vessels 222 and 224 are coupled to the circulation loop 208 via solvent supply lines 236 and 238, respectively.

[0034] The separating vessel 214 is coupled to the process chamber 250 via exhaust gas piping 240. The liquid/solid waste collection vessel 217 is coupled to the separating vessel 214. The separating vessel 214 is also coupled to the liquifying/purifying system 219 via return gas piping 241. The liquifying/purifying system 219 is coupled to the CO₂ supply vessel 216 via liquid CO₂ piping 243. Alternatively, an off-site location houses the liquifying/purifying system 219, which receives exhaust gas in gas collection vessels and returns liquid CO₂ in liquid CO₂ vessels.

[0035] The first and second filters, 221 and 223, are coupled to the circulation loop 208. The first filter 221 can be a fine filter that is, for example, configured to filter 0.05 μm particles and larger particles from the cleaning fluid prior to its reaching the circulation inlet 232 to the process chamber 250. The second filter 223 can be a coarse filter that is, for example, configured to filter 2-3 μm particles and larger particles from the cleaning fluid after it leaves the process chamber 250 via circulation outlet 234. The third filter 225 couples the CO₂ supply vessel 216 to the CO₂ pump 218. The third filter 225 can, for example, be configured to filter 0.05 μm particles and larger particles from the CO₂ liquid prior to it reaching the circulation loop 208.

[0036] A controller 290 is coupled to and exchanges information with multiple components of the film removal system 200, including the substrate chamber 250, the CO₂ supply system 206, the solvent supply system 212, and the substrate transfer system 269. In addition, controller 290 is coupled to and exchanges information with valves, pumps, pressure gauges, heaters, and temperature gauges of the film removal system 200. The controller 290 is capable of generating control signals sufficient to communicate and control the inputs of the film removal system 200 as well as monitor the outputs from the film removal system 200.

[0037] It will be readily apparent to one skilled in the art that the film removal system 200 further includes valves, control electronics, and utility hookups which are typical of supercritical fluid processing systems. Further, it will be readily apparent to one skilled in the art that the injection nozzles 275 can be configured as part of the substrate holder 260 rather than as part of the chamber housing 270.

[0038] The film removal system 200 in FIG. 2A further contains a wet processing system 280 that is disposed in the film removal system 200. The wet processing system 280 is coupled to and exchanges information with the controller 290 and the substrate transfer system 269. According to an embodiment of the invention, supercritical CO₂ processing can be performed in series with wet processing to remove films from a micro-feature on the substrate 265. For example, as described in FIGS. 1A – 1C, the portion 186 of the dielectric film 182 can be removed by supercritical CO₂ processing, and the photoresist film 184 can be removed by wet processing, or alternately, the portion 186 of the dielectric film 182 can be removed by wet processing, and the photoresist film 184 can be removed by supercritical CO₂ processing. The wet processing system 280 can be a conventional wet processing system that is well known to artisans skilled in the art of wet processing.

[0039] FIG. 2B is a schematic diagram showing a film removal system according to another embodiment of the invention. In FIG. 2B, a wet processing system 281 is separate from, but operatively coupled to, a supercritical CO₂ film removal system 201. The film removal system 201 can be the same or similar to the supercritical CO₂ portion of the film removal

system 200 of FIG. 2A, i.e., it can include all components shown in FIG 2A except for wet processing system 280. The separate wet processing system 281 may operate with its own controller (not shown).

[0040] Operations of the film removal systems illustrated in FIGS. 2A and 2B will now be described. Parts of the film removal systems 200 and 201 configured to contain a supercritical CO₂ fluid are heated to a temperature above the critical temperature of 31°C, for example by heaters 204 and 220. In one embodiment of the invention, the temperature can be between about 31°C and about 200°C. Alternately, the temperature can be between about 40°C and about 120°C. Still alternately, the temperature can be between about 60°C and about 80°C.

[0041] A substrate 265 is provided in the substrate chamber 250 using substrate transfer system 269. Supercritical CO₂ fluid is introduced into the circulation loop 208 using CO₂ supply vessel 216 and CO₂ pump 218. A solvent is introduced into the circulation loop 208 from the solvent supply vessels 222 or 224 via the solvent supply lines 236 or 238 utilizing the first injection pump 226 or the second injection pump 228. The ratio of the solvent to the combination of the supercritical CO₂ and the solvent can, for example, be between about 0.1% and about 33% by volume. Next, the system is pressurized to the operating pressure. The supercritical CO₂ pressure in the film removal system 200 or 201 can, for example, be between about 1,070 psig and about 6,000 psig. In one embodiment of the invention, the supercritical CO₂ pressure can be between about 2,000 psig and about 2,500 psig. The supercritical CO₂ fluid containing the solvent is circulated by pump 210 through the processing zone 267 and the circulation loop 208 until the desired film is removed from the substrate 265.

[0042] Next, the solvent supply to the circulation loop 208 is discontinued and the processing zone 267 and the circulation loop 208 are continuously flushed for a predetermined time by flowing fresh supercritical CO₂ from the CO₂ supply vessel 216 through the processing zone 267 and exhausting the fluid to the separating vessel 214, while maintaining pressure above a critical pressure. The predetermined time can, for example, be between about 10

sec and about 1200 sec, and alternately, can be between about 20 sec and about 600 sec, and further alternately, can be between about 30 sec and about 180 sec. The flushing may further include a series of predetermined decompression steps that include sequentially pressurizing the processing zone 267 with supercritical CO₂ fluid, and subsequently exhausting supercritical CO₂ fluid to the separating vessel 216, while maintaining the pressure in the processing zone 267 and in the circulation loop 208 above the critical CO₂ pressure.

[0043] Next, the processing zone 267 is depressurized and the substrate 265 removed from the substrate holder 260 by the substrate transfer system 269.

[0044] The substrate 265 can be transferred from the substrate chamber 265 to the wet processing system 280 or 281, and vice versa, for further processing. In a wet process for removing a film from a micro-feature on the substrate 265, the substrate 265 is exposed to a wet fluid in the wet processing system 280 or 281. In the case of a dielectric film, the wet fluid is capable of removing the dielectric film from the substrate 265 and can, for example, be a HF_(aq) fluid. In the case of a photoresist film, the wet fluid is capable of removing the photoresist film from the substrate 265 and can, for example, be an aqueous mixture of sulfuric acid and hydrogen peroxide. When the wet processing has been performed for a desired amount of time to remove the dielectric film or the photoresist film, the substrate 265 is rinsed with deionized water and dried.

[0045] In an exemplary embodiment, wet processing system 280 or 281 is not needed, and supercritical CO₂ processing is used to remove both the dielectric film and the photoresist film. In this embodiment, after the dielectric film is removed from substrate 265 by the supercritical CO₂ fluid containing a first solvent, for example from solvent supply vessel 222, the processing zone 267 and circulation loop 208 are flushed, as described above, with fresh supercritical CO₂ until the first solvent is removed therefrom. Then, without transferring the substrate 265, the second solvent, for example from solvent supply vessel 224, is introduced to the circulation loop 208 for combining with the supercritical CO₂ fluid. The supercritical CO₂ fluid containing the second

solvent is then circulated by pump 210 through the processing zone 267 and the circulation loop 208 until the photoresist film is removed from the substrate 265. The flushing process is then repeated until the second solvent is flushed from the processing zone 267 and circulation loop 208.

[0046] FIG. 3A is a flow diagram showing a simplified sequence of a dielectric film removal process according to an embodiment of the invention. In the dielectric film removal process 300, the film removal system 200 in FIG. 2A or the supercritical CO₂ film removal system 201 or wet processing system 281 in FIG. 2B can be used for removing the dielectric film from a micro-feature on the substrate 265. At 302, the process is started. At 312, a substrate is provided in a substrate chamber. The substrate has a micro-feature containing a dielectric film on the sidewalls of the micro-feature and a photoresist film covering a portion of the dielectric film. At 320, a dielectric film removal process is performed on the substrate to remove the portion of the dielectric film not covered by the photoresist film. At 332, the process ends.

[0047] FIG. 3B is a flow diagram showing a simplified sequence of a photoresist film removal process according to another embodiment of the invention. In the film removal process 350, the film removal system 200 in FIG. 2A or the supercritical CO₂ film removal system 201 or wet processing system 281 in FIG. 2B can be used for removing a dielectric film from a micro-feature on the substrate 265. At 352, the process is started. At 362, a substrate is provided in a substrate chamber. The substrate can contain the micro-feature processed by the dielectric film removal process 300 in FIG. 3A. At 372, a film removal process is performed on the substrate to remove the photoresist film from the micro-feature. At 382, the process ends. At least one of dielectric film removal process 300 or photoresist film removal process 350 is performed using supercritical CO₂ processing.

[0048] In one embodiment of the invention, both the dielectric film removal process 300 and the photoresist film removal process 350 can be performed using supercritical CO₂ processing. In another embodiment of the invention, the dielectric film removal process 300 can be performed using supercritical CO₂ processing and the photoresist film removal process 350 can be

performed using wet processing. In yet another embodiment of the invention, the dielectric film removal process 300 can be performed using wet processing, and the photoresist film removal process 350 can be performed using supercritical CO₂ processing.

[0049] FIG. 4 illustrates a computer system 1201 with which an embodiment of the invention may be implemented. The computer system 1201 may be used as the controller 290 in the systems 200 and 201 of FIGS. 2A and 2B, respectively, to perform any or all of the functions described above. Computer system 1201 may also be used as a controller (not shown) for wet processing system 281 in FIG. 2B. The computer system 1201 includes a bus 1202 or other communication mechanism for communicating information, and a processor 1203 coupled with the bus 1202 for processing the information. The computer system 1201 also includes a main memory 1204, such as a random access memory (RAM) or other dynamic storage device (e.g., dynamic RAM (DRAM), static RAM (SRAM), and synchronous DRAM (SDRAM)), coupled to the bus 1202 for storing information and instructions to be executed by processor 1203. In addition, the main memory 1204 may be used for storing temporary variables or other intermediate information during the execution of instructions by the processor 1203. The computer system 1201 further includes a read only memory (ROM) 1205 or other static storage device (e.g., programmable ROM (PROM), erasable PROM (EPROM), and electrically erasable PROM (EEPROM)) coupled to the bus 1202 for storing static information and instructions for the processor 1203.

[0050] The computer system 1201 also includes a disk controller 1206 coupled to the bus 1202 to control one or more storage devices for storing information and instructions, such as a magnetic hard disk 1207, and a removable media drive 1208 (e.g., floppy disk drive, read-only compact disc drive, read/write compact disc drive, tape drive, and removable magneto-optical drive). The storage devices may be added to the computer system 1201 using an appropriate device interface (e.g., small computer system interface (SCSI), integrated device electronics (IDE), enhanced-IDE (E-IDE), direct memory access (DMA), or ultra-DMA).

[0051] The computer system 1201 may also include special purpose logic devices (e.g., application specific integrated circuits (ASICs)) or configurable logic devices (e.g., simple programmable logic devices (SPLDs), complex programmable logic devices (CPLDs), and field programmable gate arrays (FPGAs), (not shown). The computer system may also include one or more digital signal processors (DSPs) (not shown), such as the TMS320 series of chips from Texas Instruments, the DSP56000, DSP56100, DSP56300, DSP56600, and DSP96000 series of chips from Motorola, the DSP1600 and DSP3200 series from Lucent Technologies or the ADSP2100 and ADSP21000 series from Analog Devices. Other processors especially designed to process analog signals that have been converted to the digital domain may also be used.

[0052] The computer system 1201 may also include a display controller 1209 coupled to the bus 1202 to control a display 1210 for displaying information to a computer user. The computer system includes input devices, such as a keyboard 1211 and a pointing device 1212, for interacting with a computer user and providing information to the processor 1203. The pointing device 1212, for example, may be a mouse, a trackball, or a pointing stick for communicating direction information and command selections to the processor 1203 and for controlling cursor movement on the display 1210. In addition, a printer (not shown) may provide printed listings of data stored and/or generated by the computer system 1201.

[0053] The computer system 1201 performs a portion or all of the processing steps of the invention in response to the processor 1203 executing one or more sequences of one or more instructions contained in a memory, such as the main memory 1204. Such instructions may be read into the main memory 1204 from another computer readable medium, such as a hard disk 1207 or a removable media drive 1208. One or more processors in a multi-processing arrangement may also be employed to execute the sequences of instructions contained in main memory 1204. In alternative embodiments, hard-wired circuitry may be used in place of or in combination with software instructions. Thus, embodiments are not limited to any specific combination of hardware circuitry and software.

[0054] As stated above, the computer system 1201 includes at least one computer readable medium or memory for holding instructions programmed according to the teachings of the invention and for containing data structures, tables, records, or other data described herein. Examples of computer readable media are compact discs, hard disks, floppy disks, tape, magneto-optical disks, PROMs (EPROM, EEPROM, flash EPROM), DRAM, SRAM, SDRAM, or any other magnetic medium, compact discs (e.g., CD-ROM), or any other optical medium, punch cards, paper tape, or other physical medium with patterns of holes, a carrier wave (described below), or any other medium from which a computer can read.

[0055] Stored on any one or on a combination of computer readable media, the invention includes software for controlling the computer system 1201, for driving a device or devices for implementing the invention, and for enabling the computer system 1201 to interact with a human user (e.g., processing system personnel). Such software may include, but is not limited to, device drivers, operating systems, development tools, and applications software. Such computer readable media further includes the computer program product of the invention for performing all or a portion (if processing is distributed) of the processing performed in implementing the invention.

[0056] The computer code devices of the invention may be any interpretable or executable code mechanism, including but not limited to scripts, interpretable programs, dynamic link libraries (DLLs), Java classes, and complete executable programs. Moreover, parts of the processing of the invention may be distributed for better performance, reliability, and/or cost.

[0057] The term "computer readable medium" as used herein refers to any medium that participates in providing instructions to the processor 1203 for execution. A computer readable medium may take many forms, including but not limited to, non-volatile media, volatile media, and transmission media. Non-volatile media includes, for example, optical, magnetic disks, and magneto-optical disks, such as the hard disk 1207 or the removable media drive 1208. Volatile media includes dynamic memory, such as the main memory 1204. Transmission media includes coaxial cables, copper wire and fiber optics, including the wires that make up the bus 1202. Transmission

media also may also take the form of acoustic or light waves, such as those generated during radio wave and infrared data communications.

[0058] Various forms of computer readable media may be involved in carrying out one or more sequences of one or more instructions to processor 1203 for execution. For example, the instructions may initially be carried on a magnetic disk of a remote computer. The remote computer can load the instructions for implementing all or a portion of the invention remotely into a dynamic memory and send the instructions over a telephone line using a modem. A modem local to the computer system 1201 may receive the data on the telephone line and use an infrared transmitter to convert the data to an infrared signal. An infrared detector coupled to the bus 1202 can receive the data carried in the infrared signal and place the data on the bus 1202. The bus 1202 carries the data to the main memory 1204, from which the processor 1203 retrieves and executes the instructions. The instructions received by the main memory 1204 may optionally be stored on storage device 1207 or 1208 either before or after execution by processor 1203.

[0059] The computer system 1201 also includes a communication interface 1213 coupled to the bus 1202. The communication interface 1213 provides a two-way data communication coupling to a network link 1214 that is connected to, for example, a local area network (LAN) 1215, or to another communications network 1216 such as the Internet. For example, the communication interface 1213 may be a network interface card to attach to any packet switched LAN. As another example, the communication interface 1213 may be an asymmetrical digital subscriber line (ADSL) card, an integrated services digital network (ISDN) card or a modem to provide a data communication connection to a corresponding type of communications line. Wireless links may also be implemented. In any such implementation, the communication interface 1213 sends and receives electrical, electromagnetic or optical signals that carry digital data streams representing various types of information.

[0060] The network link 1214 typically provides data communication through one or more networks to other data devices. For example, the network link 1214 may provide a connection to another computer through a

local network 1215 (e.g., a LAN) or through equipment operated by a service provider, which provides communication services through a communications network 1216. The local network 1214 and the communications network 1216 use, for example, electrical, electromagnetic, or optical signals that carry digital data streams, and the associated physical film (e.g., CAT 5 cable, coaxial cable, optical fiber, etc). The signals through the various networks and the signals on the network link 1214 and through the communication interface 1213, which carry the digital data to and from the computer system 1201 maybe implemented in baseband signals, or carrier wave based signals. The baseband signals convey the digital data as unmodulated electrical pulses that are descriptive of a stream of digital data bits, where the term "bits" is to be construed broadly to mean symbol, where each symbol conveys at least one or more information bits. The digital data may also be used to modulate a carrier wave, such as with amplitude, phase and/or frequency shift keyed signals that are propagated over a conductive media, or transmitted as electromagnetic waves through a propagation medium. Thus, the digital data may be sent as unmodulated baseband data through a "wired" communication channel and/or sent within a predetermined frequency band, different than baseband, by modulating a carrier wave. The computer system 1201 can transmit and receive data, including program code, through the network(s) 1215 and 1216, the network link 1214, and the communication interface 1213. Moreover, the network link 1214 may provide a connection through a LAN 1215 to a mobile device 1217 such as a personal digital assistant (PDA) laptop computer, or cellular telephone.

[0061] The computer system 1201 may be configured to perform the method of the invention to process a substrate in a film removal system. The computer system 1201 may be further configured to control a supercritical CO₂ system for removing films from a micro-feature. The computer system 1201 may also be configured to control a wet processing system disposed in or operatively coupled to the supercritical CO₂ film removal system.

[0062] Although only certain embodiments of this invention have been described in detail above, those skilled in the art will readily appreciate that many modifications are possible in the exemplary embodiment without

materially departing from the novel teachings and advantages of this invention. Accordingly, all such modifications are intended to be included within the scope of this invention.

WHAT IS CLAIMED IS:

1. A method of processing a substrate in a film removal system, the method comprising:
 - providing the substrate in a substrate chamber of the film removal system, the substrate having a micro-feature containing a dielectric film on a sidewall of the micro-feature and a photoresist film covering a first portion of the dielectric film and not covering a second portion of the dielectric film;
 - performing a first film removal process on the substrate to remove the second portion of the dielectric film not covered by the photoresist film;
 - and
 - performing a second film removal process on the substrate to remove the photoresist film, wherein the second film removal process is performed following the first film removal process, and
 - wherein at least one of the performing a first film removal process and the performing a second film removal process comprises using supercritical CO₂ processing.
2. The method according to claim 1, wherein the performing a first film removal process comprises exposing the substrate to a supercritical CO₂ fluid and a first solvent capable of removing the dielectric film.
3. The method according to claim 2, wherein the first solvent comprises aqueous HF or HF:pyridine.
4. The method according to claim 2, wherein the performing a first film removal process further comprises:
 - pressurizing the substrate chamber with the supercritical CO₂ fluid;
 - delivering the first solvent in the supercritical CO₂ fluid;
 - exposing the substrate to the supercritical CO₂ fluid and the first solvent until the second portion of the dielectric film is removed from the sidewall; and
 - flushing the substrate chamber with fresh supercritical CO₂ fluid.

5. The method according to claim 2, wherein the performing a first film removal process comprises heating the supercritical CO₂ fluid to between about 31°C and about 200°C.
6. The method according to claim 2, wherein the performing a first film removal process comprises heating the supercritical CO₂ fluid to between about 40°C and about 120°C.
7. The method according to claim 2, wherein the performing a first film removal process comprises heating the supercritical CO₂ fluid to between about 60°C and about 80°C.
8. The method according to claim 2, wherein the performing a first film removal process comprises providing a CO₂ pressure between about 1,070 psig and about 6,000 psig.
9. The method according to claim 2, wherein the performing a first film removal process comprises providing a CO₂ pressure between about 2,000 psig and about 3,000 psig.
10. The method according to claim 2, wherein a ratio of the first solvent to the combination of the supercritical CO₂ fluid and the first solvent is between about 0.1% and about 33% by volume.
11. The method according to claim 2, wherein the performing a first film removal process comprises exposing the substrate to the supercritical CO₂ fluid and the first solvent for a time period between about 10 sec and about 1200 sec.
12. The method according to claim 2, wherein the performing a first film removal process comprises exposing the substrate to the supercritical CO₂ fluid and the first solvent for a time period between about 20 sec and about 600 sec.

13. The method according to claim 2, wherein the performing a first film removal process comprises exposing the substrate to the supercritical CO₂ fluid and the first solvent for a time period between about 30 sec and about 180 sec.
14. The method according to claim 2, wherein the performing a second film removal process comprises exposing the substrate to the supercritical CO₂ fluid and a second solvent capable of removing the photoresist film.
15. The method according to claim 14, wherein the second solvent comprises N-methyl pyrrolidone, diisopropyl amine, triisopropyl amine, or diglycol amine, or a combination of two or more thereof.
16. The method according to claim 15, wherein the second solvent further comprises methanol, ethanol, isopropyl alcohol, benzyl alcohol, acetone, butylene carbonate, propylene carbonate, dimethylsulfoxide, γ -butyrolactone, dimethyl formamide, dimethyl acetamide, ethyl lactate, hydrogen peroxide, benzoyl peroxide, oxygen, ozone, nitric acid, acetic acid, or formic acid, or a combination of two or more thereof.
17. The method according to claim 14, wherein the performing a second film removal process further comprises:
- pressurizing the substrate chamber with the supercritical CO₂ fluid;
 - delivering the second solvent in the supercritical CO₂ fluid;
 - exposing the substrate to the supercritical CO₂ fluid and the second solvent until the photoresist film is removed from the first portion of the dielectric film; and
 - flushing the substrate chamber with fresh supercritical CO₂ fluid.
18. The method according to claim 14, wherein the performing a second film removal process comprises heating the supercritical CO₂ fluid to between about 31°C and about 200°C.

19. The method according to claim 14, wherein the performing a second film removal process comprises heating the supercritical CO₂ fluid to between about 40°C and about 120°C.
20. The method according to claim 14, wherein the performing a second film removal process comprises heating the supercritical CO₂ fluid to between about 60°C and about 80°C.
21. The method according to claim 14, wherein the performing a second film removal process comprises providing a CO₂ pressure between about 1,070 psig and about 6,000 psig.
22. The method according to claim 14, wherein the performing a second film removal process comprises providing a CO₂ pressure between about 2,000 psig and about 3,000 psig.
23. The method according to claim 14, wherein a ratio of the second solvent to the combination of the supercritical CO₂ fluid and the second solvent is between about 0.1% and about 33% by volume.
24. The method according to claim 14, wherein the performing a second film removal process comprises exposing the substrate to the supercritical CO₂ fluid and the second solvent for a time period between about 10 sec and about 1200 sec.
25. The method according to claim 14, wherein the performing a second film removal process comprises exposing the substrate to the supercritical CO₂ fluid and the second solvent for a time period between about 20 sec and about 600 sec.
26. The method according to claim 14, wherein the performing a second film removal process comprises exposing the substrate to the supercritical

CO₂ fluid and the second solvent for a time period between about 30 sec and about 180 sec.

27. The method according to claim 1, wherein the micro-feature comprises a trench having a diameter less than about 0.2 microns.
28. The method according to claim 1, wherein the micro-feature comprises a trench having a depth to diameter aspect ratio greater than about 10:1.
29. The method according to claim 1, wherein the micro-feature comprises a trench having a depth to diameter aspect ratio greater than about 50:1.
30. The method according to claim 1, wherein the dielectric film comprises an arsenic-doped silicon dioxide film.
31. The method according to claim 2, wherein the performing a second film removal process comprises using wet processing to remove the photoresist film.
32. The method according to claim 31, wherein the wet processing utilizes an aqueous mixture of sulfuric acid and hydrogen peroxide, HF/HNO₃, or H₂O/O₃.
33. The method according to claim 1, wherein the performing a second film removal process comprises exposing the substrate to a supercritical CO₂ fluid and a second solvent capable of removing the photoresist film.
34. The method according to claim 33, wherein the second solvent comprises N-methyl pyrrolidone, diisopropyl amine, triisopropyl amine, or diglycol amine, or a combination of two or more thereof.

35. The method according to claim 34, wherein the second solvent further comprises methanol, ethanol, isopropyl alcohol, benzyl alcohol, acetone, butylene carbonate, propylene carbonate, dimethylsulfoxide, γ -butyrolactone, dimethyl formamide, dimethyl acetamide, ethyl lactate, hydrogen peroxide, benzoyl peroxide, oxygen, ozone, nitric acid, acetic acid, or formic acid, or a combination of two or more thereof.
36. The method according to claim 33, wherein the performing a second film removal process further comprises:
- pressurizing the substrate chamber with the supercritical CO₂ fluid;
 - delivering the second solvent in the supercritical CO₂ fluid;
 - exposing the substrate to the supercritical CO₂ fluid and the second solvent until the photoresist film is removed from the dielectric film; and
 - flushing the substrate chamber with fresh supercritical CO₂ fluid.
37. The method according to claim 33, wherein the performing a second film removal process comprises heating the supercritical CO₂ fluid to between about 31°C and about 200°C.
38. The method according to claim 33, wherein the performing a second film removal process comprises heating the supercritical CO₂ fluid to between about 40°C and about 120°C.
39. The method according to claim 33, wherein the performing a second film removal process comprises heating the supercritical CO₂ fluid to between about 60°C and about 80°C.
40. The method according to claim 33, wherein the performing a second film removal process comprises providing a CO₂ pressure between about 1,070 psig and about 6,000 psig.

41. The method according to claim 33, wherein the performing a second film removal process comprises providing a CO₂ pressure between about 2,000 psig and about 3,000 psig.

42. The method according to claim 33, wherein a ratio of the second solvent to the combination of the supercritical CO₂ fluid and the second solvent is between about 0.1% and about 33% by volume.

43. The method according to claim 33, wherein the performing a second film removal process comprises exposing the substrate to the supercritical CO₂ fluid and the second solvent for a time period between about 10 sec and about 1200 sec.

44. The method according to claim 33, wherein the performing a second film removal process comprises exposing the substrate to the supercritical CO₂ fluid and the second solvent for a time period between about 20 sec and about 600 sec.

45. The method according to claim 33, wherein the performing a second film removal process comprises exposing the substrate to the supercritical CO₂ fluid and the second solvent for a time period between about 30 sec and about 180 sec.

46. The method according to claim 33, wherein the performing a first film removal process comprises using wet processing to remove the second portion of the dielectric film.

47. The method according to claim 46, wherein the wet processing utilizes aqueous HF, HF:pyridine, or HF/NH₄F.

48. A method of processing a substrate in a film removal system, the method comprising:

providing the substrate in a substrate chamber of the film removal system, the substrate having a micro-feature containing a dielectric film on a

sidewall of the micro-feature and a photoresist film covering a first portion of the dielectric film and not covering a second portion of the dielectric film;

performing a first film removal process on the substrate by exposing the substrate to a supercritical CO₂ fluid and a first solvent capable of removing the second portion of the dielectric film not covered by the photoresist film;

performing a second film removal process on the substrate after the first film removal process by exposing the substrate to the supercritical CO₂ fluid and a second solvent capable of removing the photoresist film; and

maintaining the supercritical CO₂ fluid at a temperature between about 31°C and about 200°C and a pressure between about 1,070 psig and about 6,000 psig during the first and second film removal processes.

49. The method according to claim 48, wherein the micro-feature comprises a trench having a diameter less than about 0.2 microns.

50. The method according to claim 48, wherein the micro-feature comprises a trench having a depth to diameter aspect ratio greater than about 10:1.

51. The method according to claim 48, wherein the micro-feature comprises a trench having a depth to diameter aspect ratio greater than about 50:1.

52. The method according to claim 48, wherein the dielectric film comprises an arsenic-doped silicon dioxide film.

53. The method according to claim 48, wherein the first solvent comprises aqueous HF or HF:pyridine, and wherein the second solvent comprises N-methyl pyrrolidone, diisopropyl amine, triisopropyl amine, or diglycol amine, or a combination of two or more thereof.

54. The method according to claim 48, wherein the maintaining the supercritical CO₂ fluid is at a temperature between about 40°C and about 120°C.
55. The method according to claim 48, wherein the maintaining the supercritical CO₂ fluid is at a temperature between about 60°C and about 80°C.
56. The method according to claim 48, wherein the maintaining the supercritical CO₂ fluid is at a CO₂ pressure between about 2,000 psig and about 3,000 psig.
57. The method according to claim 48, wherein a ratio of each of the first and second solvents to the combination of the supercritical CO₂ fluid and the respective first and second solvent is between about 0.1% and about 33% by volume.
58. A computer readable medium containing program instructions for execution on a processor, which when executed by the processor, cause a film removal system to perform the steps in the method recited in claim 1.
59. A computer readable medium containing program instructions for execution on a processor, which when executed by the processor, cause a film removal system to perform the steps in the method recited in claim 48.
60. A film removal system for processing a substrate, comprising:
a substrate transfer system configured for transferring the substrate within the film removal system;
a substrate chamber configured for performing a supercritical CO₂ film removal process on a substrate having a micro-feature containing a dielectric film and a photoresist film covering a portion of the dielectric film, wherein the supercritical CO₂ film removal process includes at least one of a first film removal process to remove the dielectric film not covered by the

photoresist film and a second film removal process to remove the photoresist film following the first film removal process;

a CO₂ supply system configured for delivering supercritical CO₂ fluid having a temperature of at least about 31°C and a pressure of at least about 1070 psig to the substrate chamber;

a solvent supply system configured for delivering a solvent to the substrate chamber with the supercritical CO₂ fluid; and

a controller configured for controlling the film removal system.

61. The film removal system according to claim 60, wherein the solvent supply system is configured for providing a ratio of the solvent to the combination of the supercritical CO₂ fluid and the solvent that is between about 0.1% and about 33% by volume.

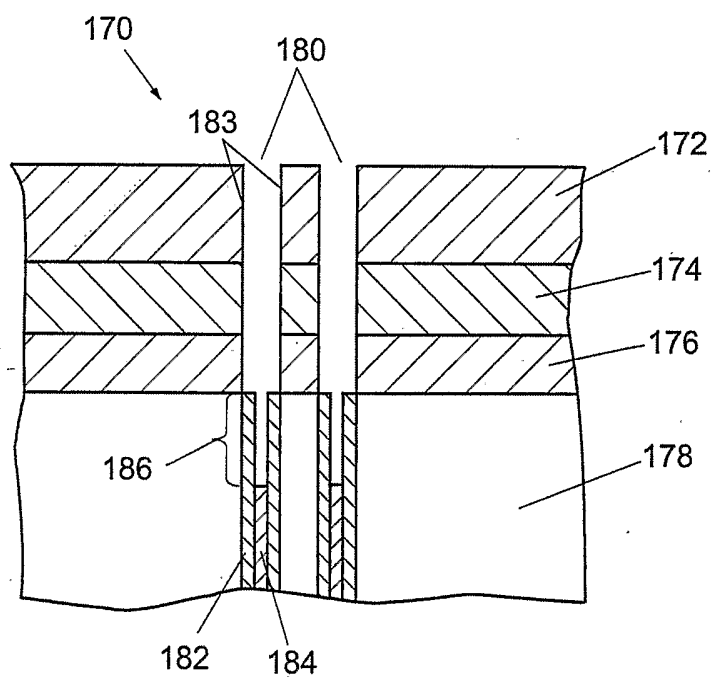
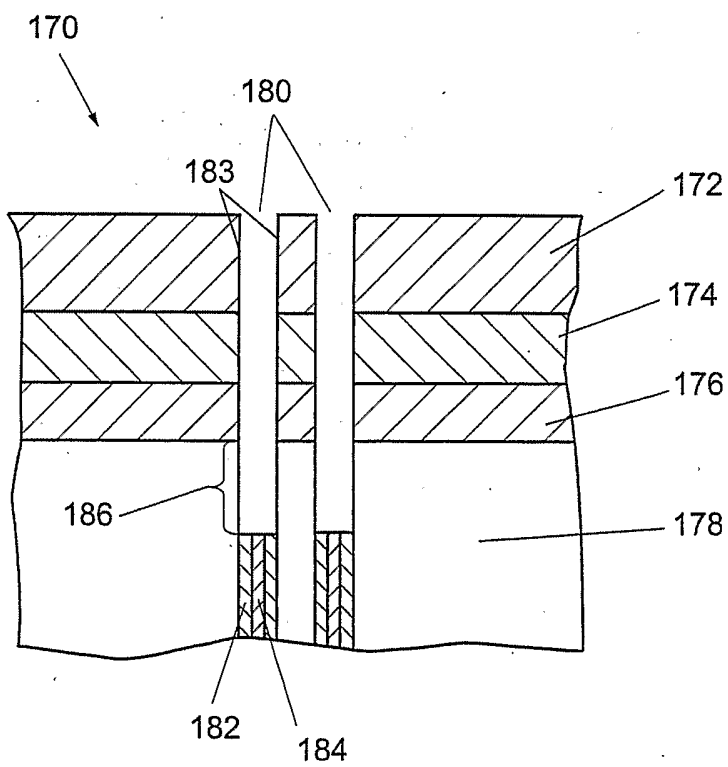
62. The film removal system according to claim 60, wherein the controller is configured for maintaining the supercritical CO₂ fluid and the solvent in contact with the substrate for a time period between about 10 sec and about 1200 sec.

63. The film removal system according to claim 60, further comprising a wet processing system disposed in or operatively coupled to the film removal system, wherein the wet processing system is configured to perform the first film removal process or the second film removal process.

64. The film removal system according to claim 63, wherein the wet processing system is configured for performing the first film removal process utilizing aqueous HF, HF:pyridine, or HF/NH₄F.

65. The film removal system according to claim 63, wherein the wet processing system is configured for performing the second film removal process utilizing an aqueous mixture of sulfuric acid and hydrogen peroxide, HF/HNO₃, or H₂O/O₃.

1/6

**FIG. 1A****FIG. 1B**

2/6

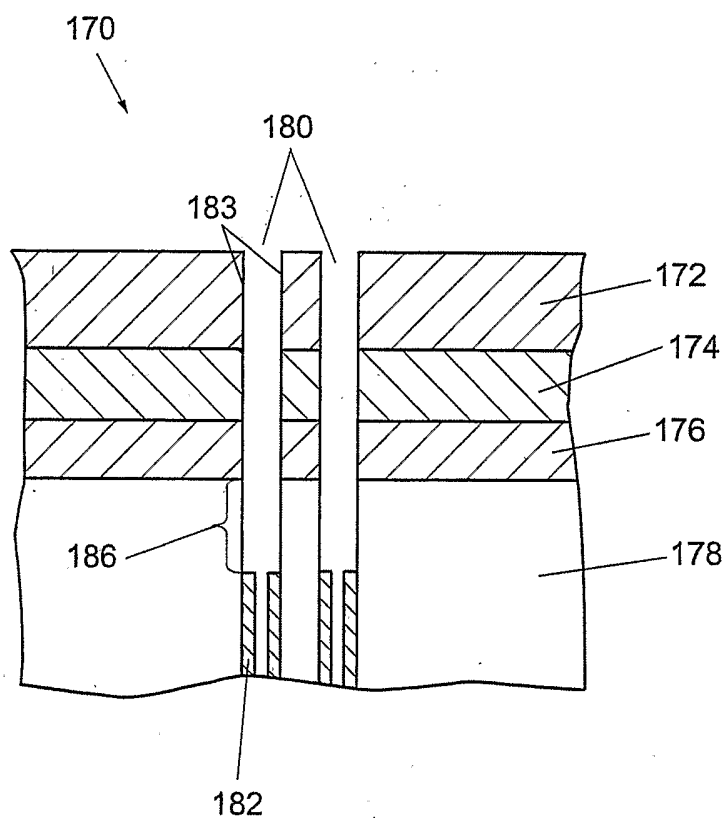


FIG. 1C

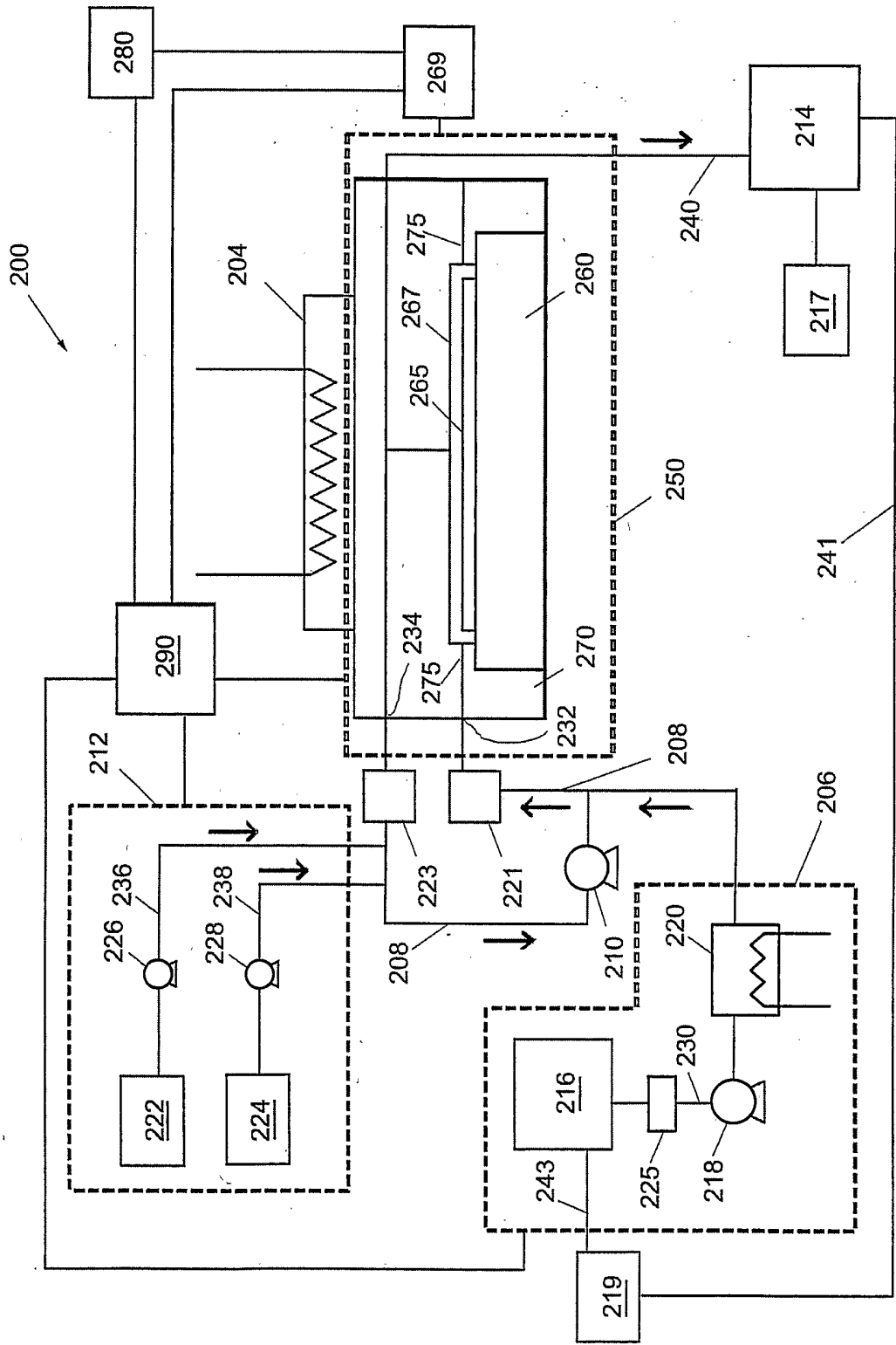


FIG. 2A

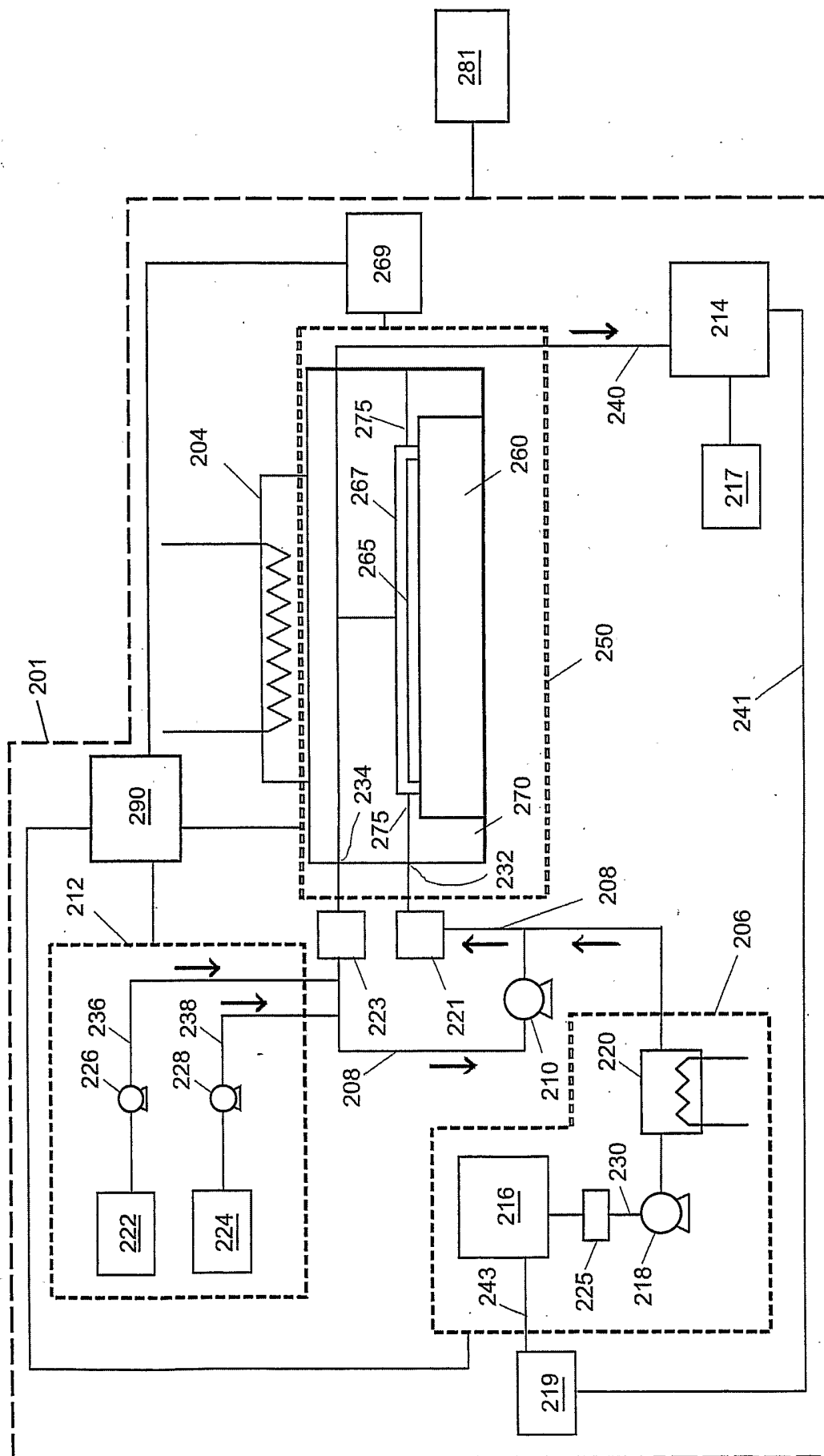


FIG. 2B

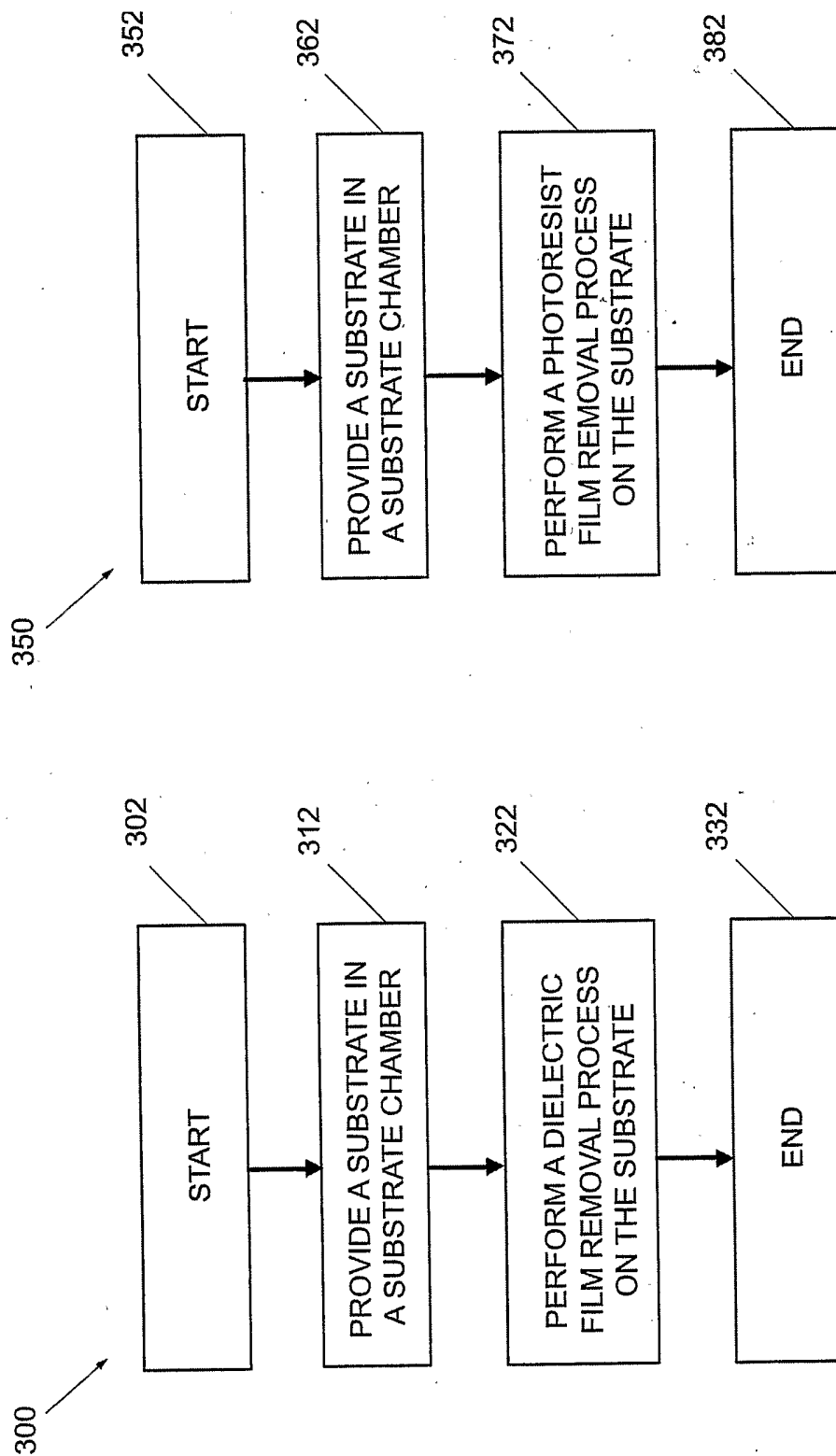


FIG. 3A

FIG. 3B

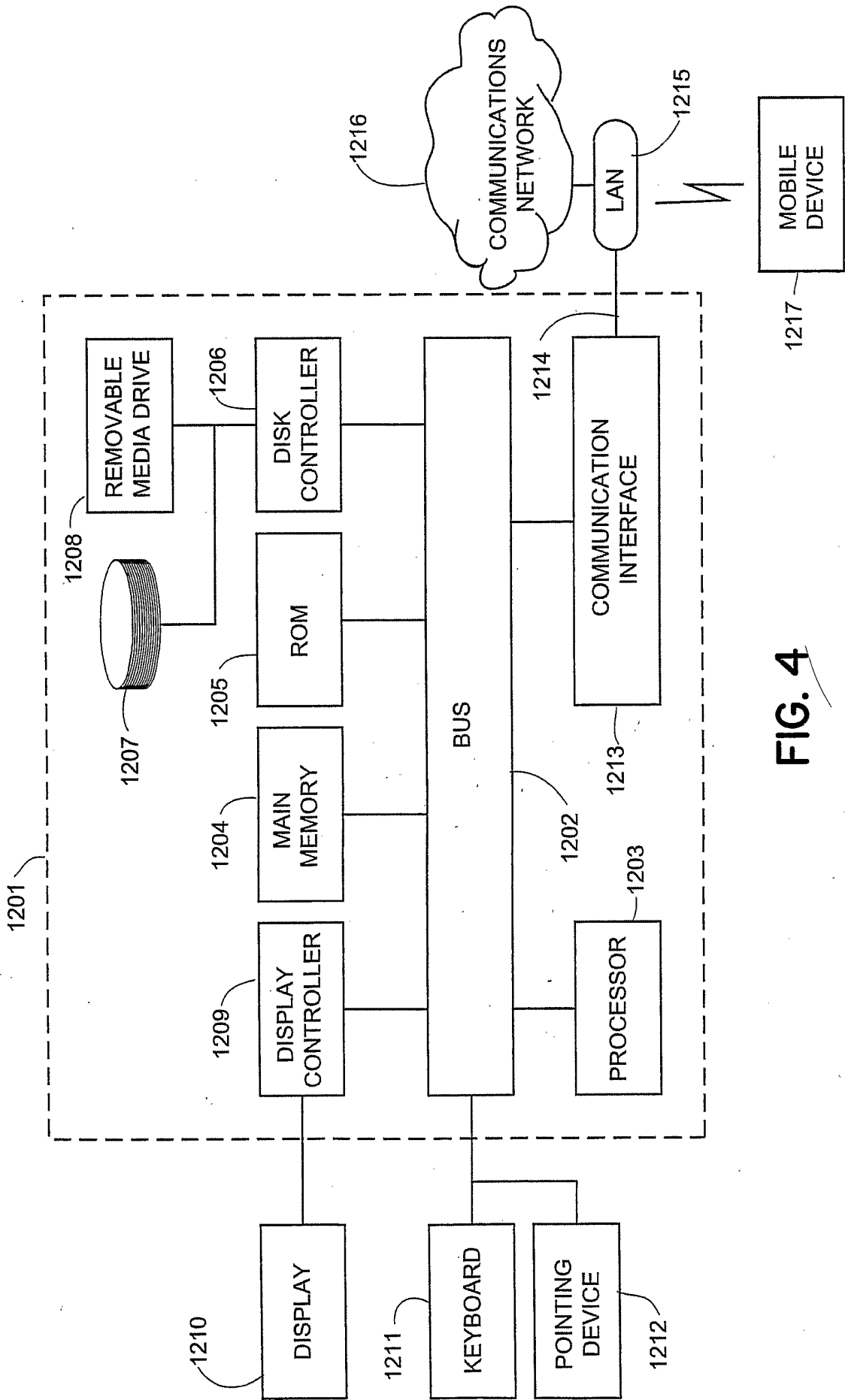


FIG. 4

INTERNATIONAL SEARCH REPORT

International Application No
PCT/US2005/013885

A. CLASSIFICATION OF SUBJECT MATTER

IPC 7 H01L21/311

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC 7 H01L

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

EPO-Internal, WPI Data, INSPEC

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	WO 02/15251 A (TOKYO ELECTRON LIMITED) 21 February 2002 (2002-02-21) pages 2-7; figures 2,3	1,27-30, 33-47, 58,60-65
X	WO 01/33613 A (SUPERCritical SYSTEMS, INC; TOKYO ELECTRON LIMITED) 10 May 2001 (2001-05-10) pages 7-10; claims 22,27,28; figure 7; examples 1,5,6	1,27-30, 33-47, 58,60-65
X	US 5 618 751 A (GOLDEN ET AL) 8 April 1997 (1997-04-08) column 4, lines 16-31; figures 5,6 column 6, lines 20-31	1-59
	----- -/--	

☒ Further documents are listed in the continuation of box C.

☒ Patent family members are listed in annex.

* Special categories of cited documents:

A document defining the general state of the art which is not considered to be of particular relevance

E earlier document but published on or after the international filing date

L document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)

O document referring to an oral disclosure, use, exhibition or other means

P document published prior to the international filing date but later than the priority date claimed

T later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention

X document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone

Y document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.

* & * document member of the same patent family

Date of the actual completion of the international search

12 October 2005

Date of mailing of the international search report

24/10/2005

Name and mailing address of the ISA

European Patent Office, P.B. 5618 Patentlaan 2
NL - 2280 HV Rijswijk
Tel. (+31-70) 340-2040, Tx. 31 651 epo nl,
Fax: (+31-70) 340-3016

Authorized officer

Szarowski, A

INTERNATIONAL SEARCH REPORT

International Application No
PCT/US2005/013885

C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT		
Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	JONES C A III ET AL: "HF etchant solutions in supercritical carbon dioxide for "dry" etch processing of microelectronic devices" CHEMISTRY OF MATERIALS AMERICAN CHEM. SOC USA, vol. 15, no. 15, 29 July 2003 (2003-07-29), pages 2867-2869, XP002348298 the whole document	1-26,31, 32, 48-57,59
A	GANGOPADHYAY S ET AL: "Supercritical CO2 treatments for semiconductor applications" MATERIALS, TECHNOLOGY AND RELIABILITY FOR ADVANCED INTERCONNECTS AND LOW-K DIELECTRICS-2004 13-15 APRIL 2004 SAN FRANCISCO, CA, USA, 2004, pages F4.6.1-F4.6.6, XP002348299 Materials, Technology and Reliability for Advanced Interconnects and Low-k Dielectrics-2004 (Materials Research Society Symposium Vol.812) Materials Research Soc Warrendale, PA, USA page F4.6.1.	1-57
A	US 2004/045588 A1 (DEYOUNG JAMES P ET AL) 11 March 2004 (2004-03-11) the whole document	1-57
A	US 2003/150559 A1 (BIBERGER MAXIMILIAN ALBERT ET AL) 14 August 2003 (2003-08-14) paragraphs '0035! - '0037!, '0047! - '0051!; figures 4,5	60-65

INTERNATIONAL SEARCH REPORT

Information on patent family members

International Application No
PCT/US2005/013885

Patent document cited in search report		Publication date	Patent family member(s)	Publication date
WO 0215251	A	21-02-2002	AU 6644200 A EP 1309990 A1 JP 2004507087 T	25-02-2002 14-05-2003 04-03-2004
WO 0133613	A	10-05-2001	AU 1455001 A CA 2387334 A1 CN 1384972 A EP 1226603 A2 HK 1050957 A1 JP 2003513342 T MX PA02004039 A	14-05-2001 10-05-2001 11-12-2002 31-07-2002 08-07-2005 08-04-2003 20-08-2003
US 5618751	A	08-04-1997	JP 3251878 B2 JP 10050943 A	28-01-2002 20-02-1998
US 2004045588	A1	11-03-2004	AU 2003237814 A1 CN 1653012 A EP 1503968 A1 JP 2005525706 T WO 03097550 A1 US 2003216269 A1	02-12-2003 10-08-2005 09-02-2005 25-08-2005 27-11-2003 20-11-2003
US 2003150559	A1	14-08-2003	US 2003136514 A1	24-07-2003