



(51) International Patent Classification:

B28B 11/04 (2006.01) E04B 2/26 (2006.01)
C04B 16/06 (2006.01)

(21) International Application Number:

PCT/US2019/057071

(22) International Filing Date:

18 October 2019 (18.10.2019)

(25) Filing Language:

English

(26) Publication Language:

English

(30) Priority Data:

62/747,595 18 October 2018 (18.10.2018) US

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(81) Designated States (unless otherwise indicated, for every kind of national protection available): AE, AG, AL, AM, AO, AT, AU, AZ, BA, BB, BG, BH, BN, BR, BW, BY, BZ, CA, CH, CL, CN, CO, CR, CU, CZ, DE, DJ, DK, DM, DO, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, GT, HN, HR, HU, ID, IL, IN, IR, IS, JO, JP, KE, KG, KH, KN, KP, KR, KW, KZ, LA, LC, LK, LR, LS, LU, LY, MA, MD, ME, MG, MK, MN, MW, MX, MY, MZ, NA, NG, NI, NO, NZ, OM, PA, PE, PG, PH, PL, PT, QA, RO, RS, RU, RW, SA, SC, SD, SE, SG, SK, SL, SM, ST, SV, SY, TH, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, ZA, ZM, ZW.

(84) Designated States (unless otherwise indicated, for every kind of regional protection available): ARIPO (BW, GH, GM, KE, LR, LS, MW, MZ, NA, RW, SD, SL, ST, SZ, TZ, UG, ZM, ZW), Eurasian (AM, AZ, BY, KG, KZ, RU, TJ,

TM), European (AL, AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HR, HU, IE, IS, IT, LT, LU, LV, MC, MK, MT, NL, NO, PL, PT, RO, RS, SE, SI, SK, SM, TR), OAPI (BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, KM, ML, MR, NE, SN, TD, TG).

Published:

— with international search report (Art. 21(3))

(54) Title: SYSTEMS AND METHODS FOR A POWER-GENERATING THERMOGALVANIC BRICK

(57) Abstract: Various embodiments of systems and methods for a thermogalvanic brick are disclosed.

SYSTEMS AND METHODS FOR A POWER-GENERATING THERMOGALVANIC BRICK

CROSS REFERENCE TO RELATED APPLICATIONS

[0001] This is a PCT application that claims benefit to U.S. provisional patent application serial number 62/747,595 filed on October 18, 2018, which is herein incorporated by reference in its entirety.

FIELD

[0002] The present disclosure generally relates to a thermogalvanic brick, and in particular to systems and methods for a thermogalvanic brick having the ability to convert temperature differences across the thermogalvanic brick into electricity using thermogalvanic electrochemistry.

BACKGROUND

[0003] To date, building components, such as bricks and concrete blocks, are passive devices that just provide structural support and thermal insulation, resulting in a significant temperature difference between the inside and outside of buildings. This temperature difference can be potentially exploited to generate useful electrical power, but should not compromise the thermal insulation properties of such building materials.

[0004] It is with these observations in mind, among others, that various aspects of the present disclosure were conceived and developed.

BRIEF DESCRIPTION OF THE DRAWINGS

[0005] FIG. 1 is an exploded view showing the various components of a thermogalvanic power-generating brick.

[0006] FIG. 2 is a perspective view of the assembled thermogalvanic power-generating brick of FIG. 1.

[0007] FIG. 3 is an image of a nonfunctional prototype embodiment having a clear acrylic plastic frame and an ABS plastic three dimensional Schwarz D geometry for structural strength, absent of thermogalvanic cell hardware or electrolyte.

[0008] FIG. 4 is an image of the outer casing for a nonfunctioning prototype embodiment of a thermogalvanic brick having electrodes, absent of electrolyte or an internal structure.

[0009] FIG. 5 is an image of a nonfunctional prototype embodiment of a thermogalvanic brick filled with electrolyte gel absent of the internal structure.

[0010] FIG. 6 shows a diagram of a thermogalvanic cell including demonstration of a redox reaction to generate electricity from heat differentials.

[0011] FIG. 7 shows 3D representations of eight Triply Periodic Minimal Surface (TPMS) structure configurations.

[0012] FIGS. 8A-8H are photographs of eight printed brick interiors with each brick interior configured using one of the eight TPMS structures (Schwarz D, Gyroid, Schwarz P, Neovius, Octo, Lidinoid, IWP, and Split P, respectively).

[0013] FIG. 9 is a CAD representation of the experimental setup for testing of each of the eight printed brick interiors of FIGS. 8A-8H.

[0014] FIG. 10 is a graphical representation of a measured change in temperature (K) over time for each of the eight printed TPMS brick interiors.

[0015] FIG. 11 is a graphical scatter plot representation of measured power output and measured current density of the eight printed TPMS brick interiors.

[0016] FIG. 12 is a graphical representation of measured power density for each of the eight printed TPMS brick interiors.

[0017] FIG. 13A is a graphical representation of a load-deformation curve associated with the Schwarz P TPMS structure.

[0018] FIG. 13B is a graphical representation of a load-deformation curve associated with the Gyroid TPMS structure.

[0019] FIG. **13C** is a graphical representation of a load-deformation curve associated with the Schwarz D TPMS structure.

[0020] FIG. **14** shows graphical representations of temperatures over time using Perspex Brick and Wish Graphite based on cold air, hot air, cold brick, and hot brick.

[0021] FIG. **15** shows graphical representations of temperatures over time using Perspex Brick and Wish Graphite based on cold air, hot air, cold brick, and hot brick.

[0022] FIG. **16** shows graphical representations of temperatures over time using Perspex Brick and Wish Graphite based on cold thermistor, hot thermistor, hot air, and cold air.

[0023] FIG. **17** is a graphical representation of a background measurement of a thermogalvanic brick.

[0024] FIG. **18** is a graphical representation of test results of a thermogalvanic brick filled with air.

[0025] FIG. **19** is a graphical representation of test results of a thermogalvanic brick filled with water.

[0026] FIG. **20** is a graphical representation of test results of a thermogalvanic brick filled with gel.

[0027] Corresponding reference characters indicate corresponding elements among the view of the drawings. The headings used in the figures do not limit the scope of the claims.

DETAILED DESCRIPTION

[0028] The interiors of buildings are typically held at a nominally constant, comfortable temperature relative to the exterior temperature. In cooler climates, the interior should be warmer; in hotter climates, shading and ventilation should keep the interior cooler. Buildings account for approximately 40% of all energy consumed, and are characterized by an ever-increasing number of embedded, wireless electronic devices that require periodic charging. Thermogalvanic cells convert temperature differences directly into electricity, using basic chemistry and no mechanical parts to do so. The present systems and methods disclosed herein incorporate a thermogalvanic cell into a building element, such as a brick, while at the same time improving (or at least not compromising) the thermal resistance of the element. These thermogalvanic bricks will thus transform “smart” building structures by enabling them to generate electric power.

[0029] Temperature differences between the inside and outside of buildings occur because of both natural and man-made causes. Most buildings in the USA and other developed countries rely on air conditioning and space heating to maintain a comfortable environment. The thermogalvanic power-generating bricks as disclosed herein convert that artificially maintained temperature difference into useful power, while maintaining or improving the thermal resistance of the building envelope. These thermogalvanic bricks, in fact, exploit the natural thermal energy storage capability of buildings, in that the outside/inside temperature difference can linger even after the air conditioning and/or heating system has been turned off. In summertime, attic temperatures in houses can reach extreme levels; the thermogalvanic power-generating bricks can be placed to capitalize on the resulting large temperature difference between the occupied space and the attic. In developing countries that don’t rely on artificial heating or cooling of buildings, massive building components such as masonry walls still yield appreciable inside/outside temperature differences that can be exploited to generate much-needed electric power at all times of the day or night. Referring to the drawings, embodiments of a thermogalvanic brick are illustrated and generally indicated as **100** in FIGS. **1-20C**.

[0030] The thermogalvanic brick **100** as shown in FIGS. **1-2** comprises a thermogalvanic cell **107** disposed within an outer casing **101**. As disclosed herein,

in order to generate electric power, the thermogalvanic cell **107** requires an electrolyte **109** (embodied in some embodiments of the thermogalvanic brick **100** as a gelled water electrolyte **109**) in contact with two conducting electrodes **106** that are maintained at different temperatures. Therefore, the thermogalvanic brick **100** needs to be in thermal contact with both the outer and inner surface temperatures of a building. The thermogalvanic brick **100** further comprises a three-dimensional interior structure **108** defining a lattice configuration and in some embodiments a Triply Minimal Periodic Surface (TPMS) configuration (shown in FIGS. **15-16**) to improve physical strength, power conversion performance, and thermal insulation of the thermogalvanic brick **100**. In addition, electricity generated by the thermogalvanic brick is exported through an outlet **112**.

[0031] An initial nonfunctioning brick **100** prototype with a clear acrylic outer casing **101** for visibility, and a yellow Schwarz D interior structure **108** made from inexpensive ABS (Acrylonitrile Butadiene Styrene) plastic, is shown in FIG. **3**. FIG. **4** shows an outer casing **101** of one embodiment of the thermogalvanic brick **100** and FIG. **5** shows a prototype of the thermogalvanic brick **100** having the gelled electrolyte **109** disposed within the outer casing **101** shown in FIG. **4**. Note that in FIG. **4** and FIG. **5** only graphite electrodes **106** are employed, within an all-plastic assembly, and thus metal has not been employed in this embodiment.

[0032] The thermogalvanic brick **100**, or more probably a plurality of thermogalvanic bricks **100**, can be positioned on any wall of a building, on the floor, the roof, or on the ceiling--wherever there is a temperature difference between the inside and outside. At the same time, the thermal insulation value (i.e., the "R" value) of the thermogalvanic brick **100** must not be less than that of the building component that it is replacing. Finally, the thermogalvanic brick **100** as disclosed herein must have at least the same structural integrity as an ordinary brick, that is, it must meet applicable standard strength guidelines for bricks used in construction.

[0033] In some embodiments, the thermogalvanic brick **100** has thermal insulation and mechanical properties like that of a conventional brick, but also has the added ability of converting temperature differences across the thermogalvanic brick **100** into electricity using thermogalvanic electrochemistry. As discussed above, much of the interior of the thermogalvanic brick **100** is filled with the gelled water electrolyte **109**. The gelled water electrolyte **109** includes a redox active compound dissolved within: ferricyanide and ferrocyanide are key examples.

The gelled electrolyte **109** fills the primary roles of (i) providing insulation, and (ii) allowing thermogalvanic electrochemistry to occur. In addition, the gelled electrolyte **109** also fills the minor role of (iii) providing structural integrity to the thermogalvanic brick **100**.

[0034] In some embodiments as discussed above, the interior structure **108** of the thermogalvanic brick **100** comprises a lattice or Triply Periodic Minimal Surface (TPMS) structure. A key example of this is a Schwarz-D plastic structure. The interior structure **108** provides most of the structural integrity and thermal insulation of the thermogalvanic brick **100**. In other embodiments, the interior structure **108** of the thermogalvanic brick **100** may be comprised of a general plastic frame.

[0035] In some embodiments, two sides **102** of the housing **101** of the thermogalvanic brick **100** may both act as metal electrodes **106** such that one embodiment of the design is a thin sheet of aluminum with the interior coated with a layer of graphite paint. This serves the primary roles of (i) acting as two electrodes **106** allowing thermogalvanic electrochemistry to occur, and (ii) providing a thermal interface such that their temperature reflects the interior and exterior temperature. In some embodiments, the other four faces **103**, **104**, and **105** of the thermogalvanic brick **100** form the remainder of the housing **101**. The housing **101** may form a water-tight seal such that no electrolyte **109** evaporates or leaks, nor can foreign substances enter the housing **101**.

[0036] One aspect of the thermogalvanic brick **100** is the control of the solar irradiation upon the exterior electrode **106** to exert some control over the temperature. When the exterior is hotter than the interior of the thermogalvanic brick **100**, additional solar irradiation will generate a temperature difference of up to 30°C greater than a shaded thermogalvanic brick **100**, and therefore more electricity. However, the thermogalvanic brick **100** must be installed carefully to ensure the thermal envelope of the building is maintained. Since the power directly expresses the temperature difference, and the temperature of each electrode **106** can also be individually determined, this effect can be intelligently tailored as needed (black for additional heating, shading used to reduce it, etc.).

[0037] Finally, the nature of the thermogalvanic brick **100** also results in new potential methods of construction. For example, a rapidly prefabricated and temporary accommodation in a humanitarian disaster could be constructed using

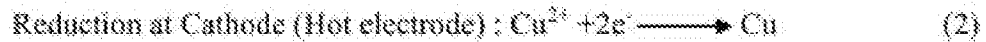
magnets (not shown) incorporated in the thermogalvanic bricks **100**. Conversely, solvent welding is a common process used to combine plastic sheets that takes seconds to achieve; it is how the prototypes of the thermogalvanic brick **100** have been constructed. If multiple thermogalvanic bricks **100** are solvent-welded together, such bricks will form an extremely robust, waterproof, insulated wall.

[0038] The initial thermogalvanic brick **100** prototypes are constructed largely from plastic, in part through additive manufacturing (“3D printing”). FIG. **1** shows an exploded view of one embodiment of the thermogalvanic brick **100**. Structural strength is provided by an interior lattice or TPMS structure **108**, made from plastic that is lightweight and uses relatively little solid material. The interior structure **108** is immersed in the gelled electrolyte **109**, and two conducting electrodes **106** are placed on either side of the interior structure **108** and in contact with the electrolyte **109**. The interior structure **108**, electrolyte **109**, and electrodes **106** are contained within the outer casing **101**. Because the interior structure **108** provides structural support, the outer casing **101** does not need to be load-bearing. The electrodes **106**, which in some embodiments are made partially of aluminum and carbon, are in good thermal contact with the outer and inner temperatures, respectively. In some embodiments, the outlet **112** for the generated electricity may be in the form of a USB port or some other kind of socket.

[0039] The complete, closed conceptual thermogalvanic brick **100** is shown in FIG. **2**. The dimensions of the thermogalvanic brick **100** are chosen to be compatible with the UK standard, but are not limited to those dimensions. It is anticipated that each thermogalvanic brick **100** will yield as much as 1 mW of power under ordinary temperature conditions, thereby making it necessary in most cases to couple the electric output of a number of bricks **100** to provide useful amounts of power.

[0040] A thermogalvanic cell **107** is defined as a galvanic cell with a non-uniform temperature, where energy is generated using temperature differentials. In practice, such cell will consist of two metallic electrodes **106**, not necessarily reversible or chemically identical, immersed in electrolyte **109**, which may or may not be homogenous in composition, and in which selectively permeable membranes in the form of an interior structure **108** might be interposed. As shown in FIG. **6**, the non-uniform temperature generates a potential difference which is directly proportional to the entropy change of the redox reaction. This potential difference

causes ions to move from the cold electrode **106** which serves as the anode to the hot electrode **106** which is the cathode when a load is connected.



[0041] The electrochemical performance of the thermogalvanic cell greatly depends on a property known as the Seebeck coefficient. This property relates the temperature difference to the electrode potential of thermogalvanic cell **107** and has a quadratic effect on the cell's performance. It is said that in thermogalvanic cells **107**, this property is three to four times greater than that of solid thermoelectrics. However, the internal resistance of the thermogalvanic **107** cell is high; a major hindrance to its overall performance.

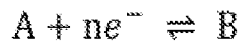
[0042] Another feature that affects the performance of the thermogalvanic cell **107** is the conduction of heat across its terminals or electrodes. An ideal thermogalvanic cell **107** has a high electrical conductivity (σ) but low thermal conductivity (K). However, both properties are directly related and hence, there is a significant tradeoff between the two parameters since a typical material of high electrical conductivity will also have high thermal conductivity. An ideal separator is porous and hence, allow ionic transport, yet highly thermal resistive in order to maintain a large temperature gradient between the electrodes.

[0043] The thermogalvanic brick **100** is a building material that converts the temperature difference across building surfaces into useful energy using the principle of Seebeck effect; recovering waste heat and converting the waste heat into electricity. The present disclosure describes and tests the effect of eight different Triply Periodic Minimal Surface Lattice Structures as interior structures **108** made from additive manufacturing on the power output and mechanical properties of the thermogalvanic brick with aqueous copper sulphate electrolyte so as to meet ASTM standards for compression strength (ASTM C67).

[0044] Thermogalvanic cells **107** are mainly characterized by their Seebeck coefficients (α) and the maximum power output (P_{max}) of the cells, as further discussed below.

[0045] As discussed earlier, the Seebeck coefficient (α) is a property which relates the temperature difference to the electrode potential of thermogalvanic

cell **107** and has a quadratic effect on the cell's performance. For a hypothetical redox reaction:



[0046] The Seebeck coefficient is expressed as:

$$(\alpha) = \frac{\partial E}{\partial T} \cong \frac{S_B - S_A}{nF}$$

[0047] where E is representative of electric potential difference, T is representative of temperature (Kelvin), S_A is representative of partial molar entropies of A, S_B is representative of partial molar entropies of B, n is representative of the number of electrons in the redox reaction, and F is representative of the Faraday constant.

[0048] The maximum power output is the highest power measured. It is obtained by connecting the external resistance to the thermogalvanic cell **107**. At maximum power output, the internal resistance of the thermogalvanic cell **107** is equal to that of the external resistance applied and the electric potential is half the open circuit voltage. Hence, from Ohm's law:

$$\text{Power output} = E \cdot I = \frac{E^2}{R_{ext}}$$

$$P_{max} = \frac{E_{oc}^2}{4R_{int}}$$

[0049] One of the critical properties of thermogalvanic cells **107** are their resistances. Thermogalvanic cells **107** are made up of three resistances. These are; activation (R_a), ohmic (R_Ω), and mass transport (R_m) resistances and together, they make up the internal resistance of the thermogalvanic cell **107**:

$$R_{int} = R_a + R_\Omega + R_m$$

[0050] The ohmic resistance is mainly affected by the properties of the electrolyte **109** or ionic resistance:

$$R_\Omega = \frac{L}{A\sigma}$$

[0051] where L is representative of the inter-electrode spacing, σ is representative of the ionic conductivity of the electrolyte **109**, and A is the surface area of the electrode **106**.

[0052] The activation resistance depends on the electron transfer reactions:

$$R_a = \frac{RT}{AnFj_o}$$

[0053] where R is representative of the universal gas constant, T is representative of the absolute temperature in Kelvin, A is representative of the surface area of the electrode **109**, n is representative of the number of electrons involved in the reaction, F is representative of the Faraday constant, and j_o is the exchange current density.

[0054] For mass transport (R_m) resistance, it depends on the mass transport process of the electrolyte **109** and its physical state. It can be derived by subtracting the other two resistances from the internal resistance:

$$R_m = R_{int} - R_a - R_\Omega$$

[0055] The internal resistance of the cell affects its power conversion efficiency by the relation expressed below:

$$\eta = \frac{\alpha^2 R_{th} \Delta T}{4R_{int}}$$

[0056] where R_{th} is the thermal resistance. The relation above shows that for higher cell efficiency, the thermal resistance must be increased and the internal resistance must be reduced.

[0057] A selective permeable membrane in the form of an interior structure **108** may be interposed into a thermogalvanic cell **107**. This permeable membrane interior structure **108** allows the movement of ions but restricts thermal conductivity. This is to say that with the right material and structure, the performance of the thermogalvanic cell **107** can be increased.

[0058] The endless design possibilities of additive manufacturing have made way for the implementation of cellular structures in modern designs. Their unique lightweight and energy absorbing capabilities make their applications limitless. Notable among these is a family of cellular structures known as Triply Periodic Minimal Surface Structures (TPMS), eight of which are shown in FIG. 7. They are believed to be the ideal geometries which mimic biomorphic geometries of human bones and other natural and pathological geometries but due to their high

complexities, can only be manufactured by 3D printing. However, their smooth surfaces and uniform radius of curvature give them their self-supporting capabilities resulting in their high manufacturability. TPMS structures are lightweight and tend to have excellent impact absorption, strength to weight ratio, acoustic insulation and heat transfer properties which enable them to be employed in a wide range of applications.

[0059] Mathematically, triply periodic minimal surfaces are solutions to scalar-valued functions of three independent variables which are mostly thought of as the x, y, and z coordinates of a point in three-dimensional Euclidean space. Below are the equations of all eight minimal surfaces Gyroid, Schwarz D, Neovius, Schwarz P, Lidinoid, IWP, Split P, and Octo. A set of trigonometric functions which define each structure is given below:

$$\textbf{Gyroid: } \cos x \sin y + \cos y \sin z + \cos z \sin x = t$$

$$\begin{aligned} \textbf{Schwarz D: } & \sin x \sin y \sin z \\ & + \sin x \cos y \cos z \\ & + \cos x \sin y \cos z \\ & + \cos x \cos y \sin z = t \end{aligned}$$

$$\begin{aligned} \textbf{Neovius: } & 3(\cos x + \cos y + \cos z) \\ & + 4(\cos x \cos y \cos z) = t \end{aligned}$$

$$\textbf{Schwarz P: } \cos x + \cos y + \cos z = t$$

$$\begin{aligned} \textbf{Lidinoid: } & 0.5(\sin 2x \cos y \sin z \\ & + \sin 2y \cos z \sin x \\ & + \sin 2z \cos x \sin y) \\ & - 0.5(\cos 2x \cos 2y \\ & + \cos 2y \cos 2z \\ & + \cos 2z \cos 2x) = t \end{aligned}$$

$$\begin{aligned} \textbf{IWP: } & \cos x \cos y + \cos y \cos z \\ & + \cos z \cos x \\ & - \cos x \cos y \cos z = t \end{aligned}$$

$$\begin{aligned}
 \text{Split P: } & 1.1(\sin 2x \cos y \sin z \\
 & + \sin 2y \cos z \sin x \\
 & + \sin 2z \cos x \sin y) \\
 & - 0.2(\cos 2x \cos 2y \\
 & + \cos 2y \cos 2z \\
 & + \cos 2z \cos 2x) - 0.4(\cos 2y \\
 & + \cos 2z + \cos 2x) = t
 \end{aligned}$$

$$\begin{aligned}
 \text{Octo: } & 4(\cos x \cos y + \cos y \cos z \\
 & + \cos z \cos x) \\
 & - 2.8(\cos x \cos y \cos z) \\
 & + (\cos x + \cos y + \cos z) + 1.5 \\
 & = t
 \end{aligned}$$

[0060] As mathematical structures, TPMS structures are typically designed using mathematical modeling tools like MathMod or Wolfram Mathematica with their respective mathematical equations expressed above. However, it is difficult to create high quality structures and smooth surfaces from these softwares due to a lack of proper discretization as a result from inadequate number of nodes during modeling.

[0061] To overcome these limitations, a software known as Flatt Pack was used to model these structures. The software was used to generate stereolithography files which were later 3D printed. With Flatt Pack, blocks of dimension 199mm x 88.8mm x 52.3mm blocks of all eight different internal structures **108** were generated at a volume fraction of 0.3 and respective number of cells across each axis as shown below in Table 1. The printed internal structures **108** are shown in FIGS. **8A-8H**. While Flatt Pack was used to model some embodiments, any suitable modeling tool may be used.

	X	Y	Z
Dimension (mm)	199mm	88.8mm	52.3mm
No. Cell	9.95	4.44	2.615

Table 1: Dimension and cells of 3D printed TPMS structures

[0062] The general class of cellular structures can be fabricated through polymer foaming processes using either physical or chemical blowing agents. However, these processes tend to generate random structures. The

emergence of additive manufacturing has created the opportunity to manufacture these TPMS internal structures **108**, offering outstanding control over the structures' architecture and periodic characteristics. For this research, the internal structures **108** were 3D printed by Fused Deposition Modeling (FDM) process using a Zortrax M200 3D printer and Z-ULTRAT material. Each internal structure **108** was printed at a layer height of 0.29, 100% infill and without support.

[0063] For testing purposes, the experimental setup **10** is shown in FIG. **9** and is comprised of a thermogalvanic brick **100**, PID controllers **15**, a power supply **16**, a Fluke multimeter **18**, a variable resistor box **13** and an NI DAQ **17**. This setup mimics the internal and external temperature conditions of a building, using a hot chamber **11** to model external conditions and a cold chamber **12** to model internal conditions.

[0064] For this experiment, 0.7 M CuSO_4 + 0.1 M H_2SO_4 was used for the electrolyte **109**. The electrolyte **109** was formulated in batches of 500mL each using almost a liter for each round of experimentation. This was done with Copper sulfate pentahydrate (blue powder / crystals), sulfuric acid, de-ionized water, beakers, measuring cylinders and a measuring scale. Below shows the steps in preparing the electrolyte **109**:

1. Measure out the correct amount of copper sulfate pentahydrate on the mass scale.
 - a. For 500 mL final volume, the total is 87.39 g. I usually distribute this evenly into *two beakers*, eg 44.00 g + 43.39 g.
2. Pour the copper sulfate pentahydrate into the large beakers and fill to roughly 80% of final volume, each with DI water.
 - a. *Each beaker: fill to 200 mL.*
3. Place one beaker at a time on the magnetic stir plate and stir at least medium speed for 5 minutes or until the crystals dissolve.
4. Place the small beaker on the scale, then tare the scale.
5. Carefully measure out the correct amount of sulfuric acid into the small beaker
 - a. *For 500 mL final volume, the total is 4.90 g.*
6. Carefully pour the acid into one of the large beakers having aqueous copper sulfate.

7. Rinse the small beaker with ~10 mL of DI water and drain that into the large beaker as well

8. Stir the beaker having acid for ~1 minute.

9. One beaker at a time, pour the aqueous copper sulfate into the graduated cylinder. Add water to top off at the correct volume. Measure from the bottom of the meniscus.

a. *For 500 mL final volume, each beaker should contribute 250 mL.*

10. Pour the solution from the graduated cylinder into one large beaker.

11. Label the beaker and cover with parafilm, or transfer to jar for longer storage.

12. At this point, the solution is often cloudy, possibly due to incompletely dissolved crystals if the solution is near the solubility limit. Stir the beaker until the solution shows no more opacity. This may take several hours.

[0065] The TPMS interior structure **108** was placed in between the electrodes **106** and filled with the electrolyte **109** to create the thermogalvanic brick **100**. The thermogalvanic brick **100** is then placed in the enclosure between the hot chamber **11** and the cold chamber **12**. A plurality of thermocouples (not shown) are then attached to the hot and cold sides of the thermogalvanic brick **100**- six thermocouples in all; three for each side. The thermogalvanic brick **100** was then covered and the remaining sides **103**, **104** and **105** were insulated. The enclosure was then closed and the edges insulated with tape. A set of TECs **14** were then turned on and after 2 undisturbed hours the system reached steady-state. Data was recorded with the aid of a LabVIEW VI program which recorded a sample of cell potential per second and sampled the temperature measurements at 100 samples a second. The recording was started as the external resistances were varied from 1-2000Ω using the variable resistor box **13**. Temperatures and voltages were recorded. The interior structure **108** and electrolyte **109** were changed, and the electrodes **106** sanded and cleaned with isopropyl alcohol. This process was repeated for each of the different Triply Periodic Minimal Surface interior structures **108**. A graphical representation of a measured change in temperature over time associated with each of the eight TPMS interior structures **108** is shown in FIG. **10**, and a scatter plot showing average power output and current density for thermogalvanic bricks **100** associated with each of the eight TPMS interior structures **108** is shown in FIG. **11**.

[0066] It can be observed from the results of FIG. 12 that, first, different triply periodic minimal surface interior structures **108** have effect on power output of the thermogalvanic brick **100**. Secondly, it is seen that the Split P TPMS structure gave the highest nominal power density at about $5 \times 10^{-4} \mu\text{W}/\text{cm}^2\text{K}^2$ followed by Lidinoid at about $3.255 \times 10^{-4} \mu\text{W}/\text{cm}^2\text{K}^2$.

[0067] To study the mechanical properties of the TPMS interior structures **108**, samples of 40mm x 40mm x 40mm were printed and tested. The material used was Z-ULTRAT from Zortrax. It has a Young modulus of 857 MPa and Poisson's ratio of 0.35. Samples were tested for compressive strength on Instron Compression Load frame and at 0.25 mm/s loading rate as per other literature. The loading was perpendicular to the plates. Results are shown in FIGS. 13A-13C for Schwarz P, Gyroid, and Schwarz D, respectively.

[0068] It is observed that Schwarz D and Gyroid showed a similar stress-strain behavior characterized by a short linear region. The two types of curves obtained shows that TPMS internal structures **108** could be used as both restorable and non-restorable energy absorbers and protective systems to replace most of current designs and might be more effective than conventional honeycombs and other cellular structures. It is recommended to use Schwarz D and Gyroid for applications requiring high stiffness and strain before failure and Schwarz P for low strain. Hence, it would be advisable to use Schwarz D or Gyroid for thermogalvanic brick **100** applications among the TPMS internal structure **108**; however, other embodiments may include any other TPMS or lattice internal structure **108**.

[0069] FIGS. 14-16 are graphical representations showing potential measurements and current measurements related to the development of the thermogalvanic brick **100**. FIG. 17 is a graphical representation showing background measurement of the thermogalvanic brick **100** during testing, FIG. 18 is a graphical representation showing test results of the thermogalvanic brick **100** filled with air, FIG. 19 is a graphical representation of test results showing the thermogalvanic brick **100** filled with water, and FIG. 20 is a graphical representation of test results showing the thermogalvanic brick **100** filled with gel. Initial testing shows that the thermogalvanic brick **100** with a black painted outer surface is capable of reaching 30°C higher exterior temperatures than a shaded 'brick' **100** (e.g. ~50°C in a 20°C ambient), even when the interior temperature is held at a constant ~0°C.

[0070] This disclosure is extended to what is conventionally accepted as a 'brick', as in a regular feature which can be easily combined in order to construct features such as walls, and entire habitations. Furthermore, bricks are also now employed as purely cosmetic features, in order to improve the appearance of a construction. The term "brick" is used herein to cover such applications. Furthermore, the thermogalvanic brick **100** describes a regular feature consisting of the described exterior and interior components; if dimensions are altered these thermogalvanic bricks **100** could be employed in some embodiments as ceiling tiles; in other embodiments as flooring tiles. In other embodiments, brick dimensions may be chosen which are compatible with standard cinder block building materials. Therefore, the term "thermogalvanic brick" is employed as a generic term, but dimensions and applications in certain embodiments expand to cover a multitude of conventional construction applications. In addition, in some embodiments the internal structure **108** may be fabricated using many different types of plastics, including recycled plastic.

[0071] It is envisioned that such thermogalvanic bricks **100** may be useful in both developed markets (e.g., USA, UK, Australia, Europe, etc.) and in developing markets (e.g., sub-Saharan Africa, Southeast Asia, etc.). In developed economies where HVAC is the norm for buildings then the thermogalvanic bricks **100** will be converting artificially maintained temperature differences (between the inside and outside of buildings) to electric power, but importantly without increasing the HVAC load. In developing economies where HVAC systems may not be present, the thermogalvanic bricks **100** will exploit naturally occurring temperature differences in buildings, such as across thick masonry walls, and across the ceiling between the occupied space and the attic. The economic break-even points for the thermogalvanic brick **100** design will likely depend on the target markets, such that more expensive and efficient thermogalvanic bricks **100** may be suitable for developed markets, and less expensive and less thermogalvanic efficient bricks **100** more suitable for developing markets.

[0072] It should be understood from the foregoing that, while particular embodiments have been illustrated and described, various modifications can be made thereto without departing from the spirit and scope of the invention as will be apparent to those skilled in the art. Such changes and modifications are within the scope and teachings of this invention as defined in the claims appended hereto.

CLAIMS

What is claimed is:

1. A thermogalvanic brick comprising:
 - an outer casing defining an interior surface and exterior surface forming an enclosure;
 - a three-dimensional interior structure disposed within the outer casing and immersed in an electrolyte, wherein the three dimensional interior structure defines a lattice configuration; and
 - a first electrode and a second electrode disposed on opposite sides of the three-dimensional interior structure;wherein a potential difference is induced between the first electrode and the second electrode when a temperature difference is present between the first electrode and the second electrode.
2. The thermogalvanic brick of claim 1, further comprising:
 - a USB port in communication with the outer casing, wherein the USB port is in operative communication with the three-dimensional interior structure for providing an output of power generated by the three-dimensional interior structure.
3. The thermogalvanic brick of claim 1, wherein the lattice configuration comprises a triply periodic minimal surface configuration.
4. The thermogalvanic brick of claim 1, wherein the outer casing of the three-dimensional interior structure comprises a clear acrylic plastic and a acrylonitrile butadiene styrene interior structure.
5. The thermogalvanic brick of claim 1, wherein the three-dimensional interior structure is immersed in a gelled or liquid electrolyte.
6. The thermogalvanic brick of claim 1, wherein the outer casing comprises a plastic material.

7. The thermogalvanic brick of claim 1, wherein exterior solar heating of the thermogalvanic brick has a controlled solar absorptivity, orientation, and shading to maximize the temperature difference between the exterior and interior surfaces of the thermogalvanic brick.

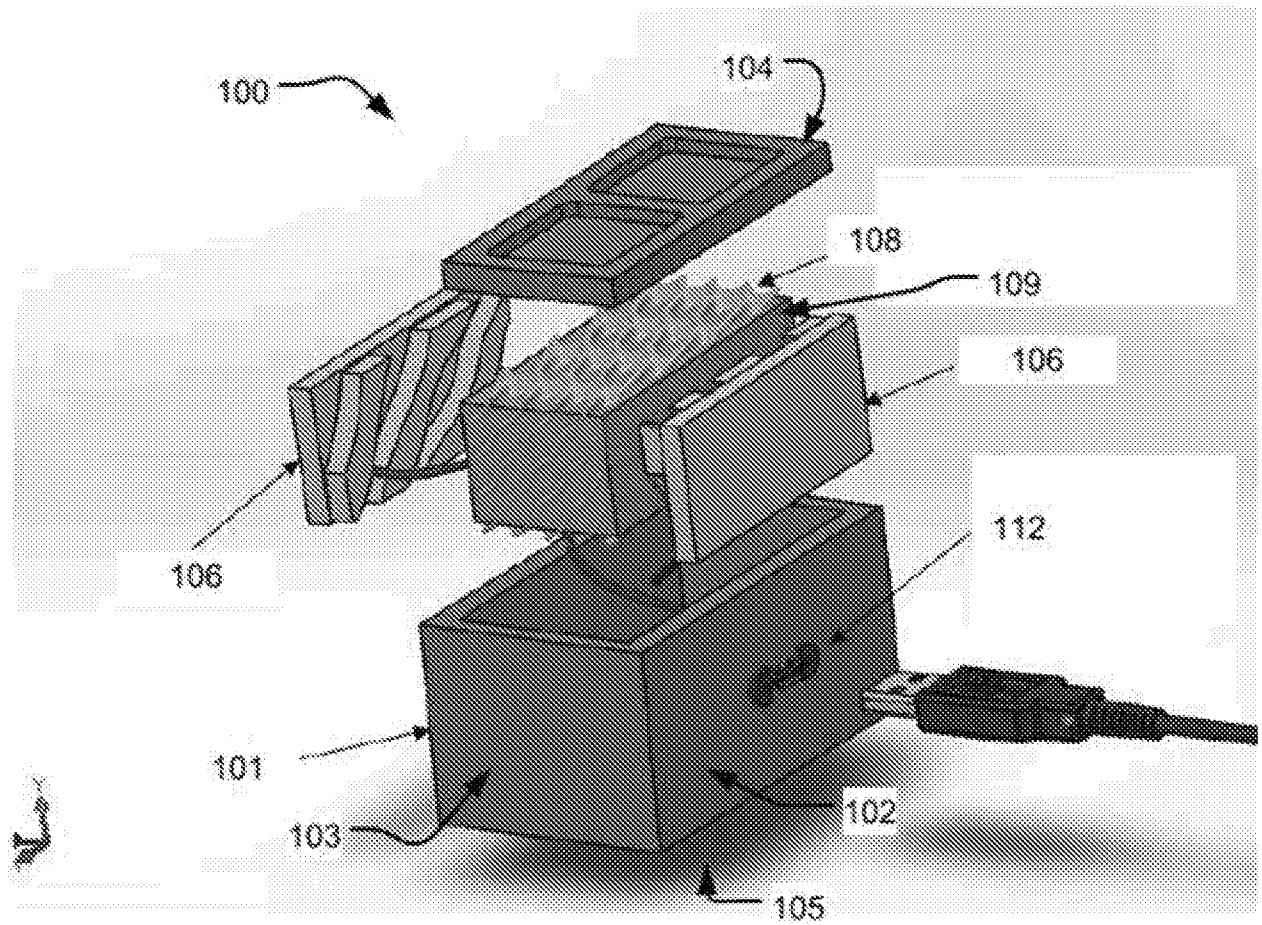


FIG. 1

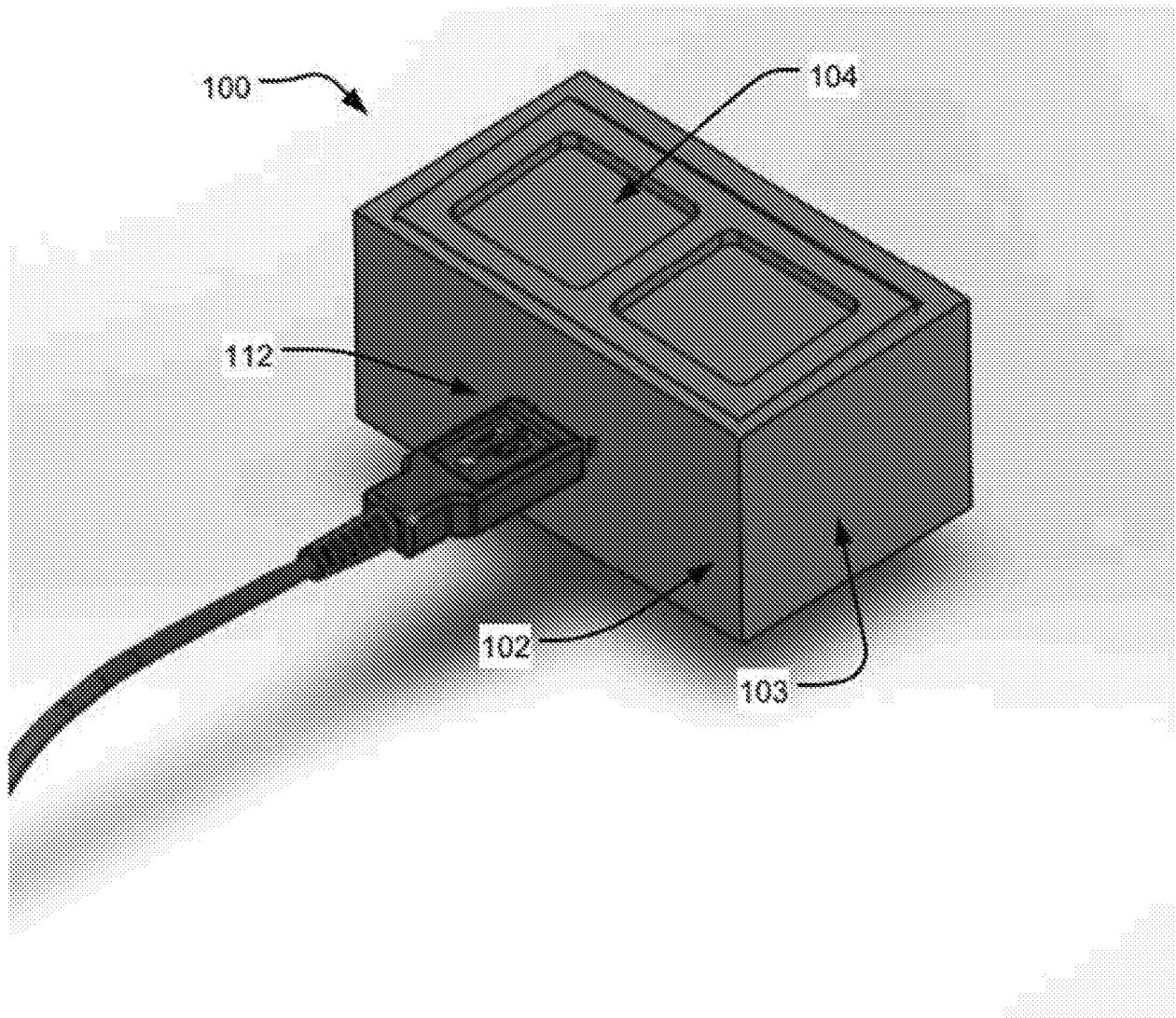


FIG. 2

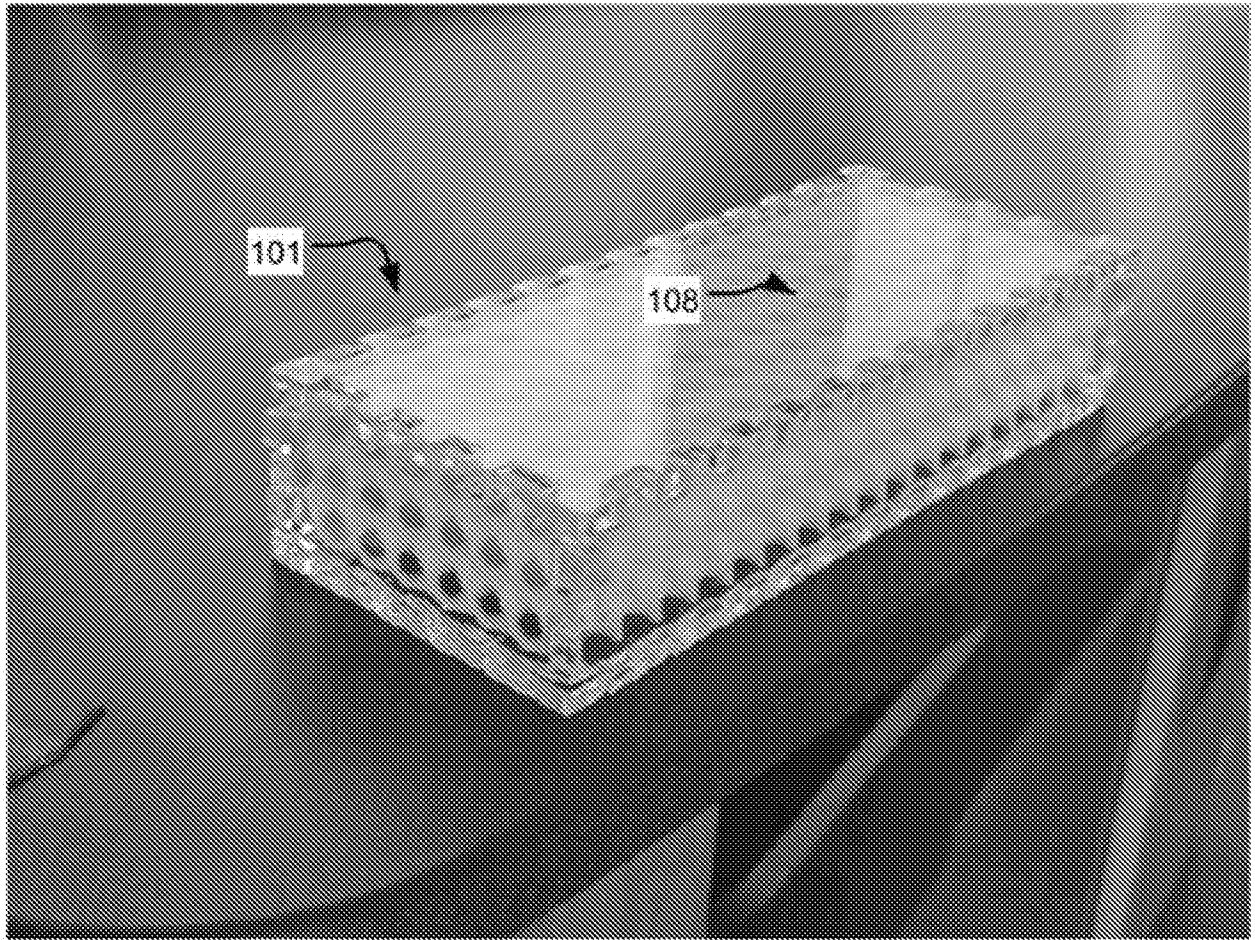


FIG. 3

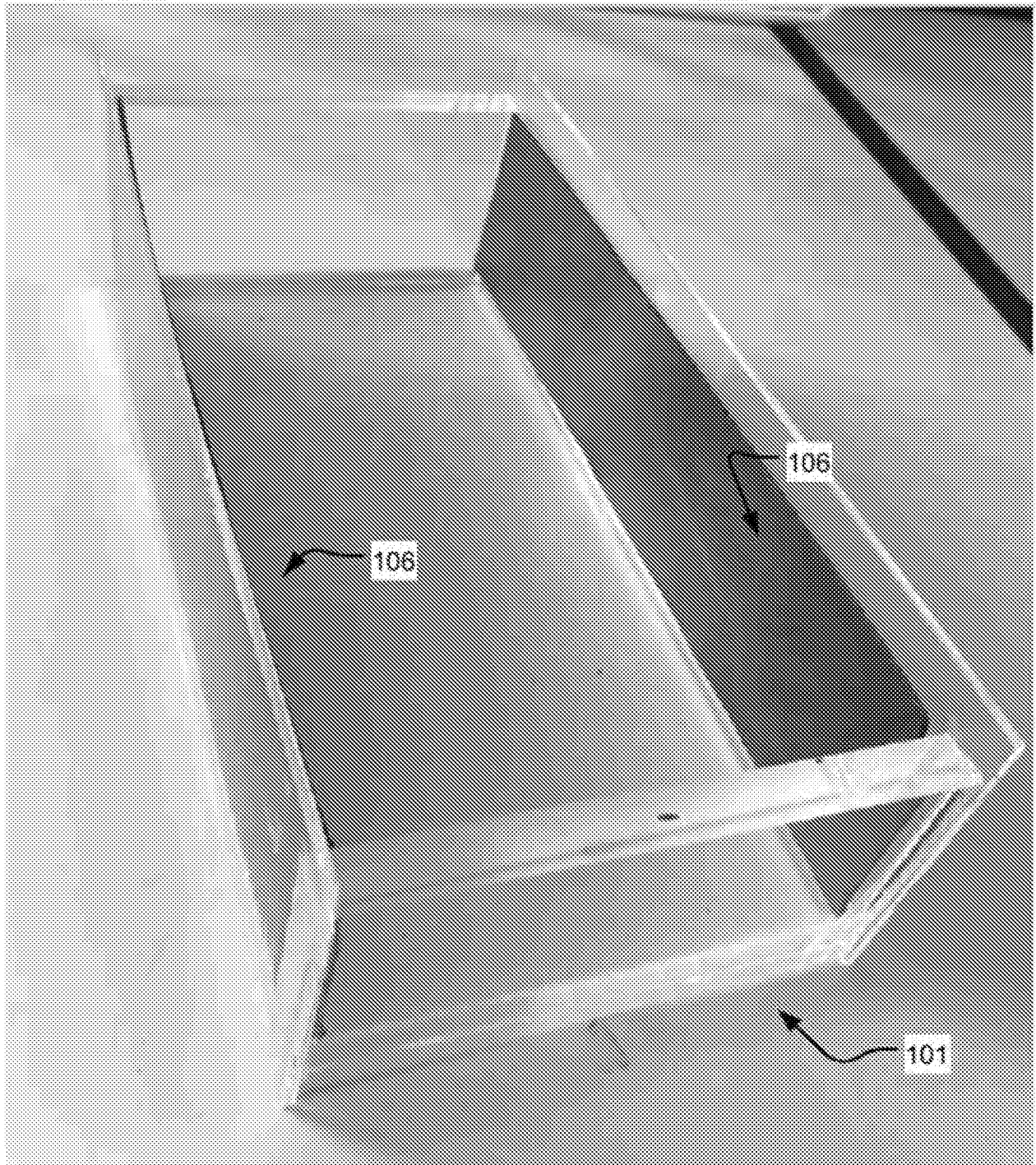


FIG. 4

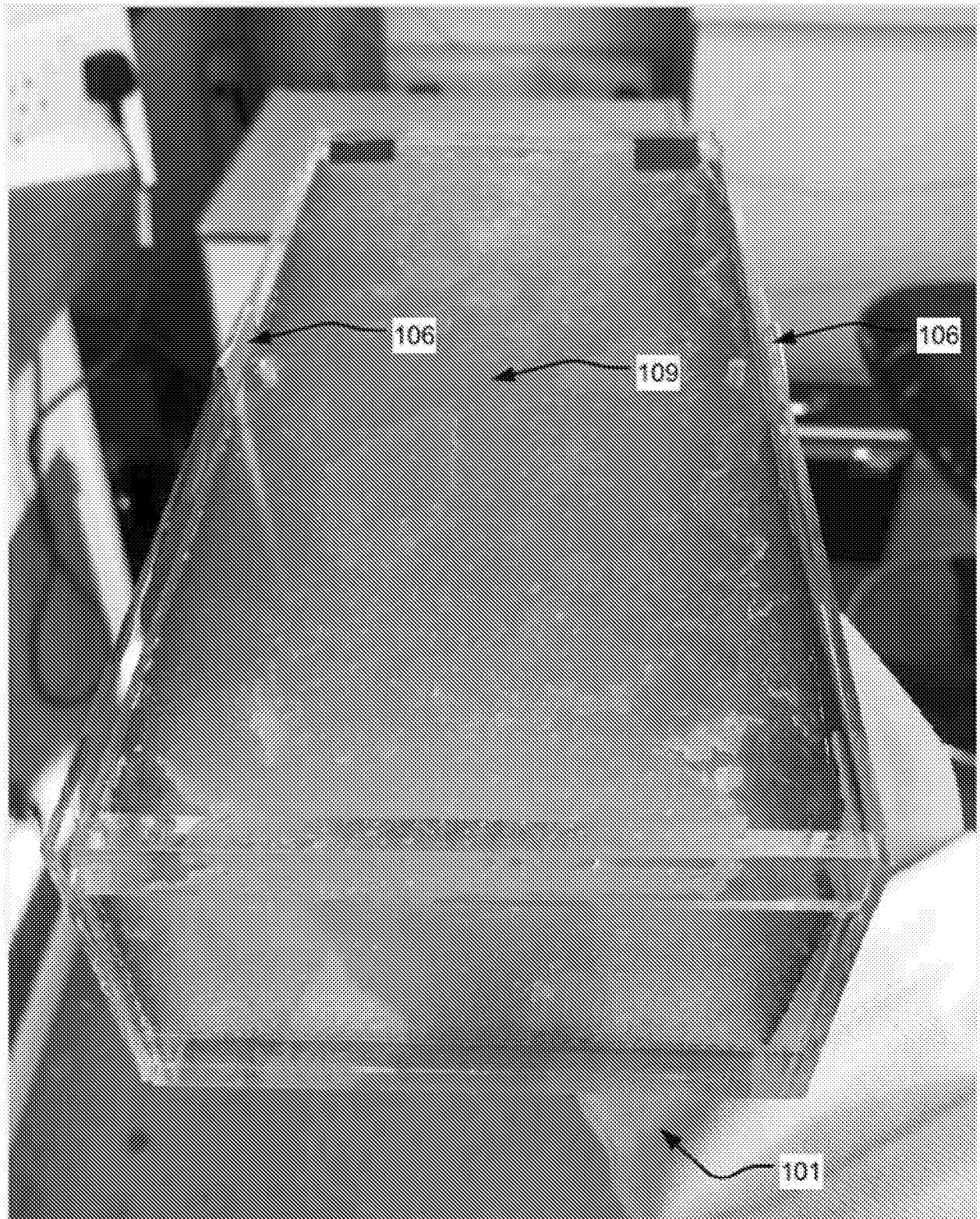


FIG. 5

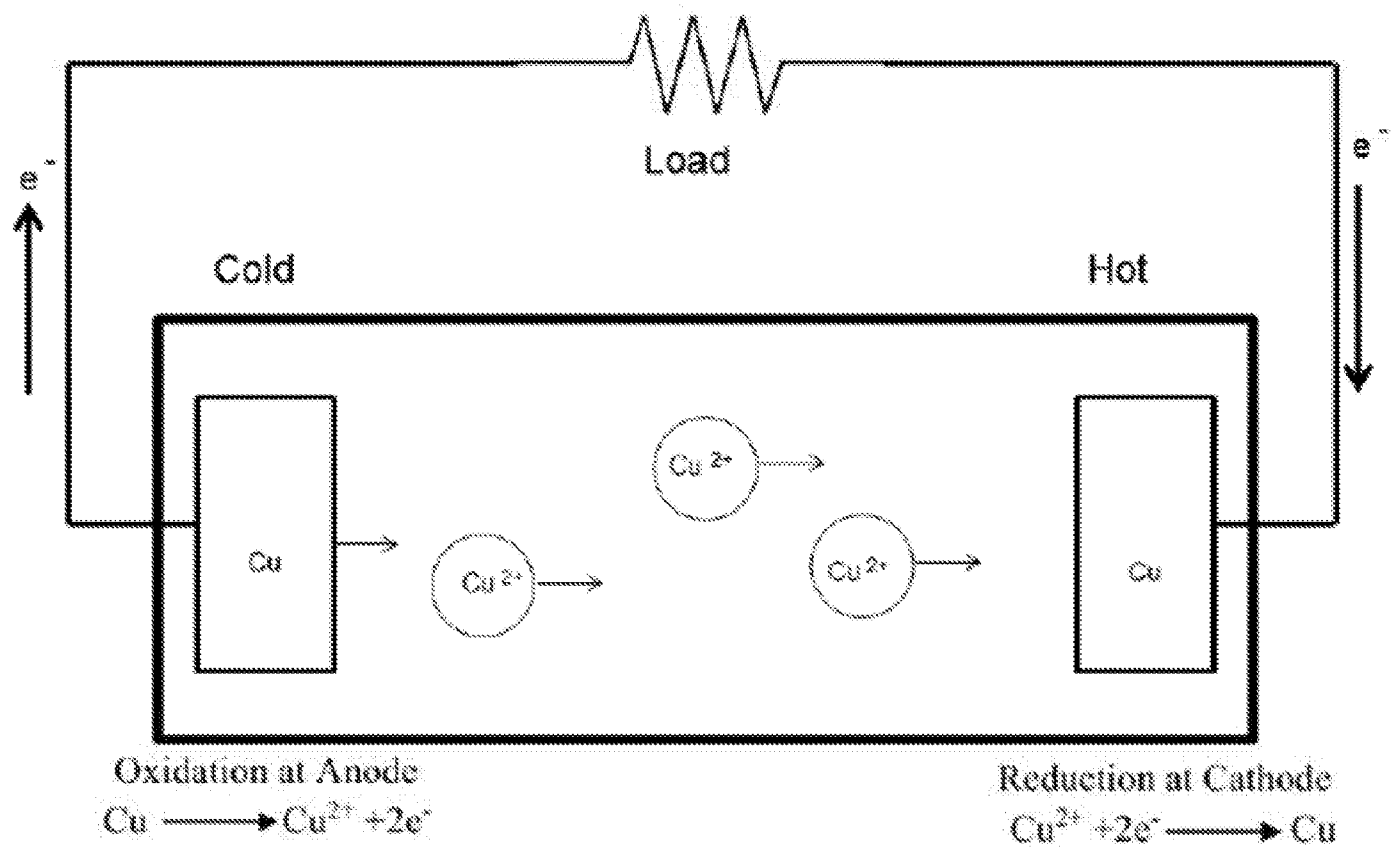


FIG. 6

107

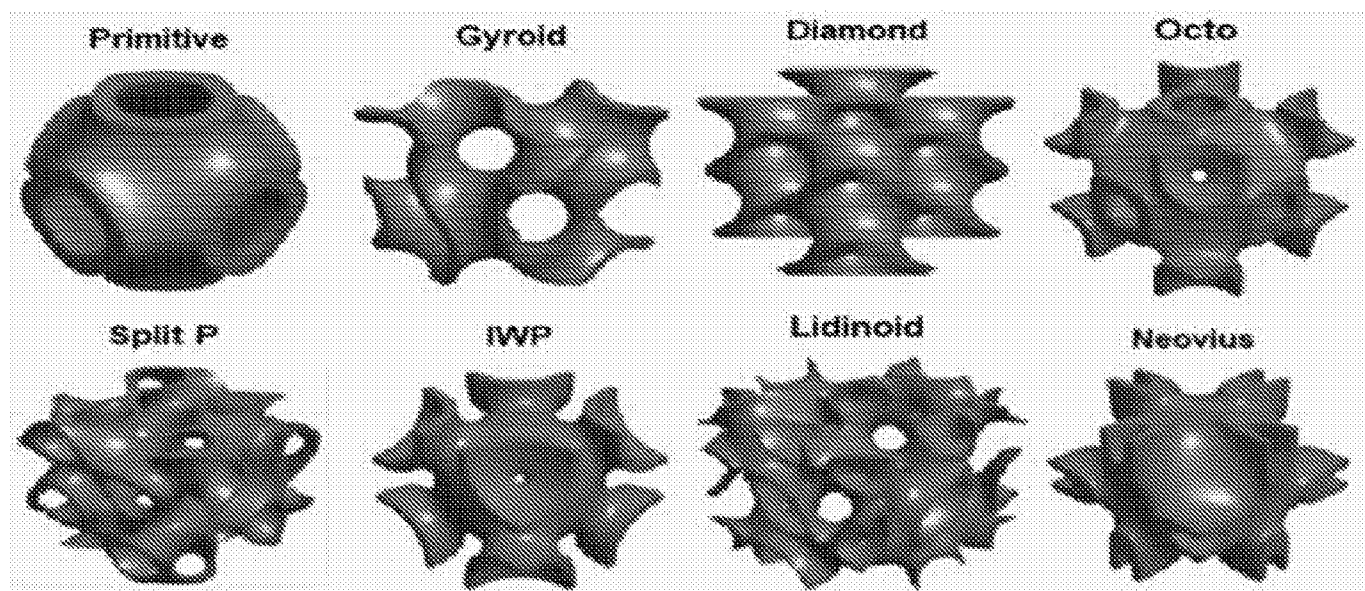
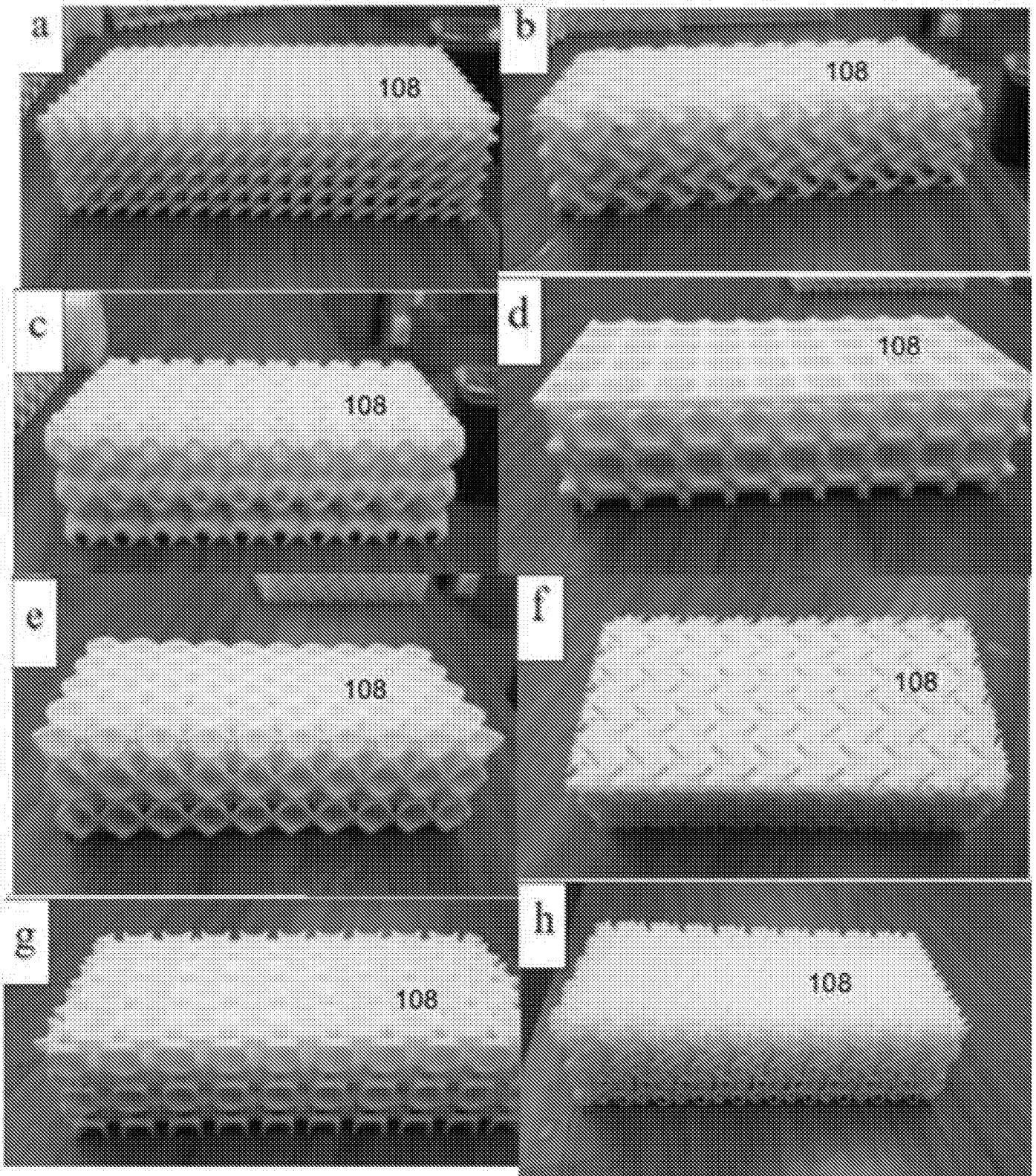


FIG. 7



FIGS. 8A-8H

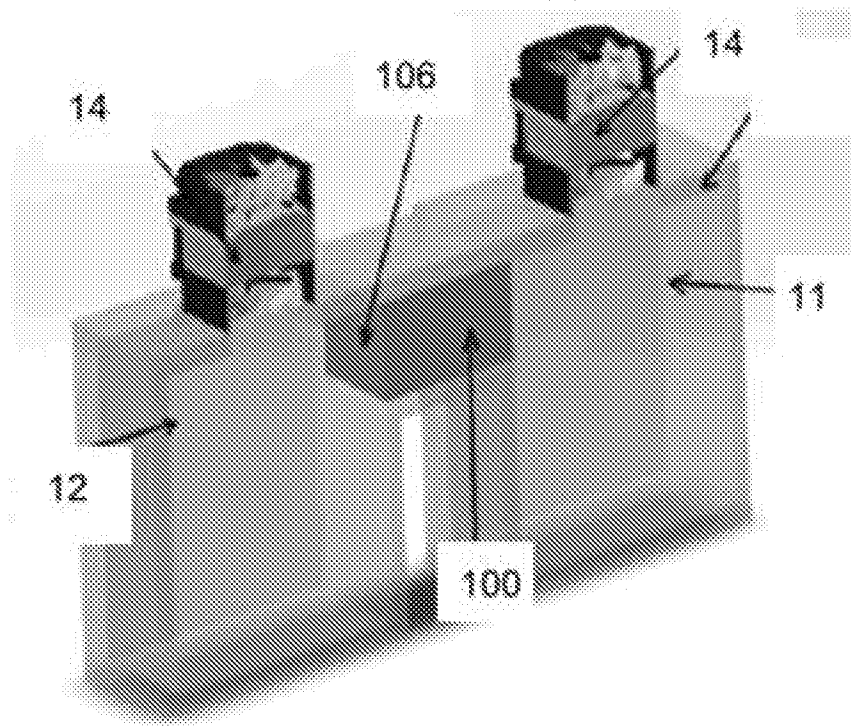


FIG. 9

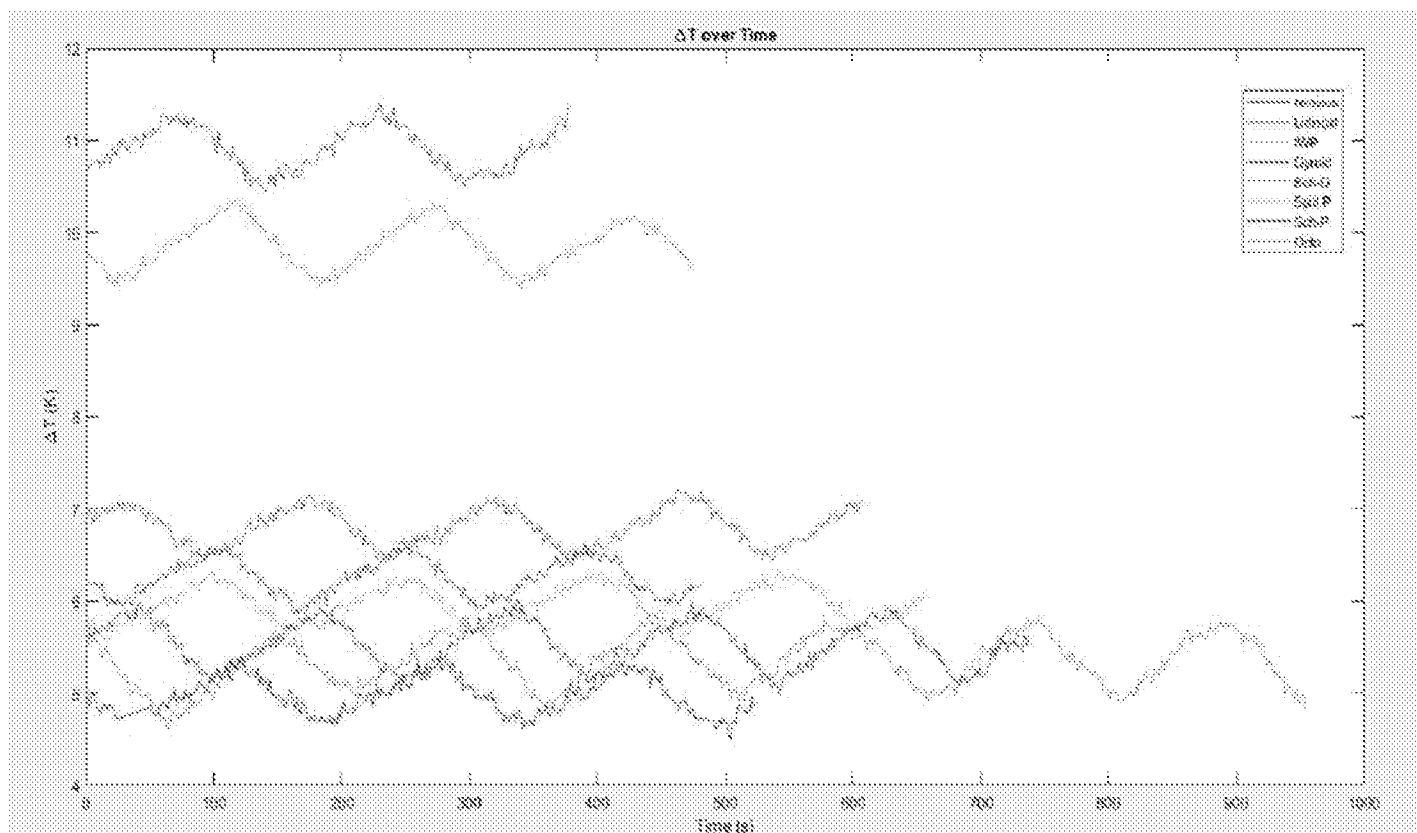
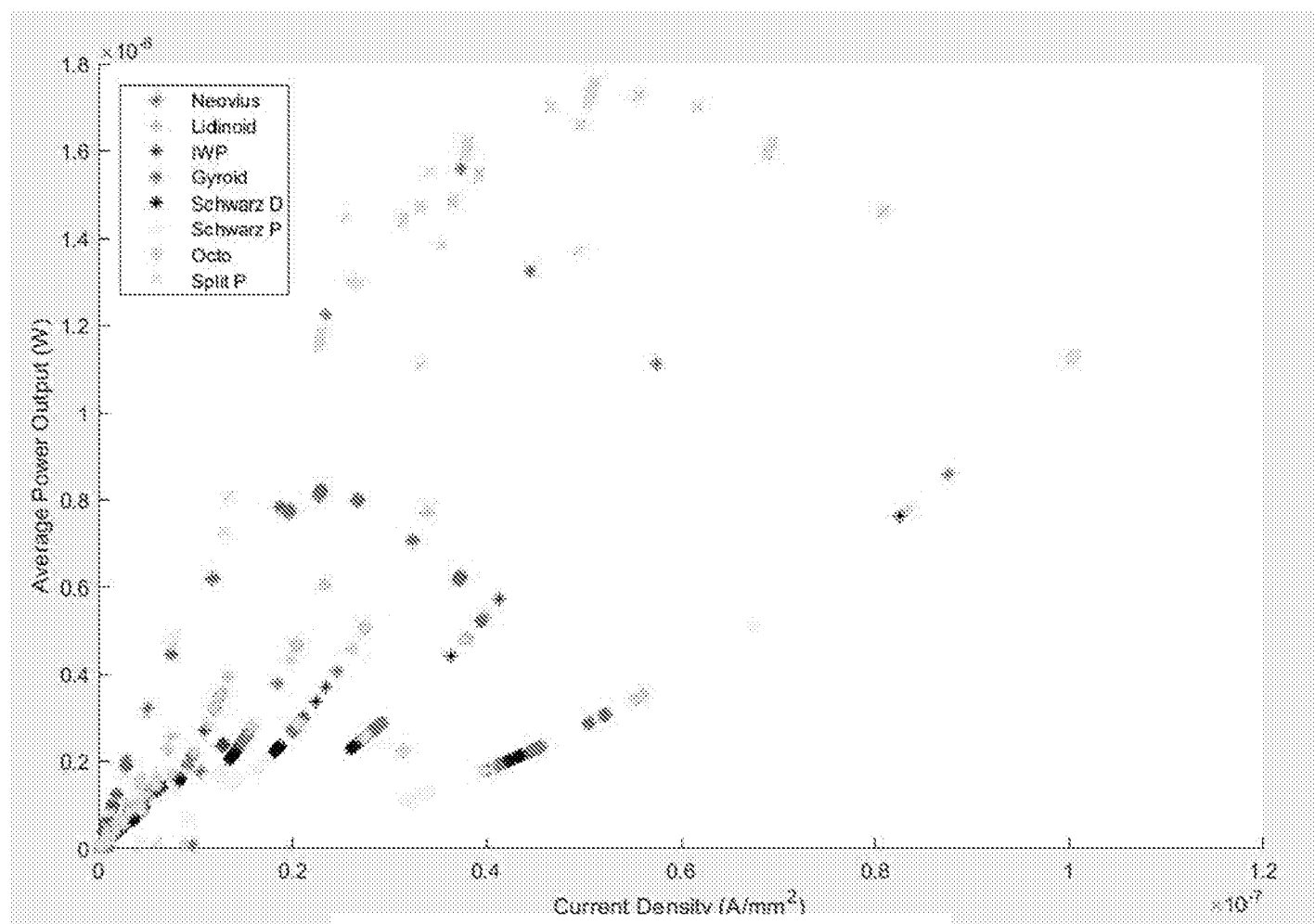


FIG. 10



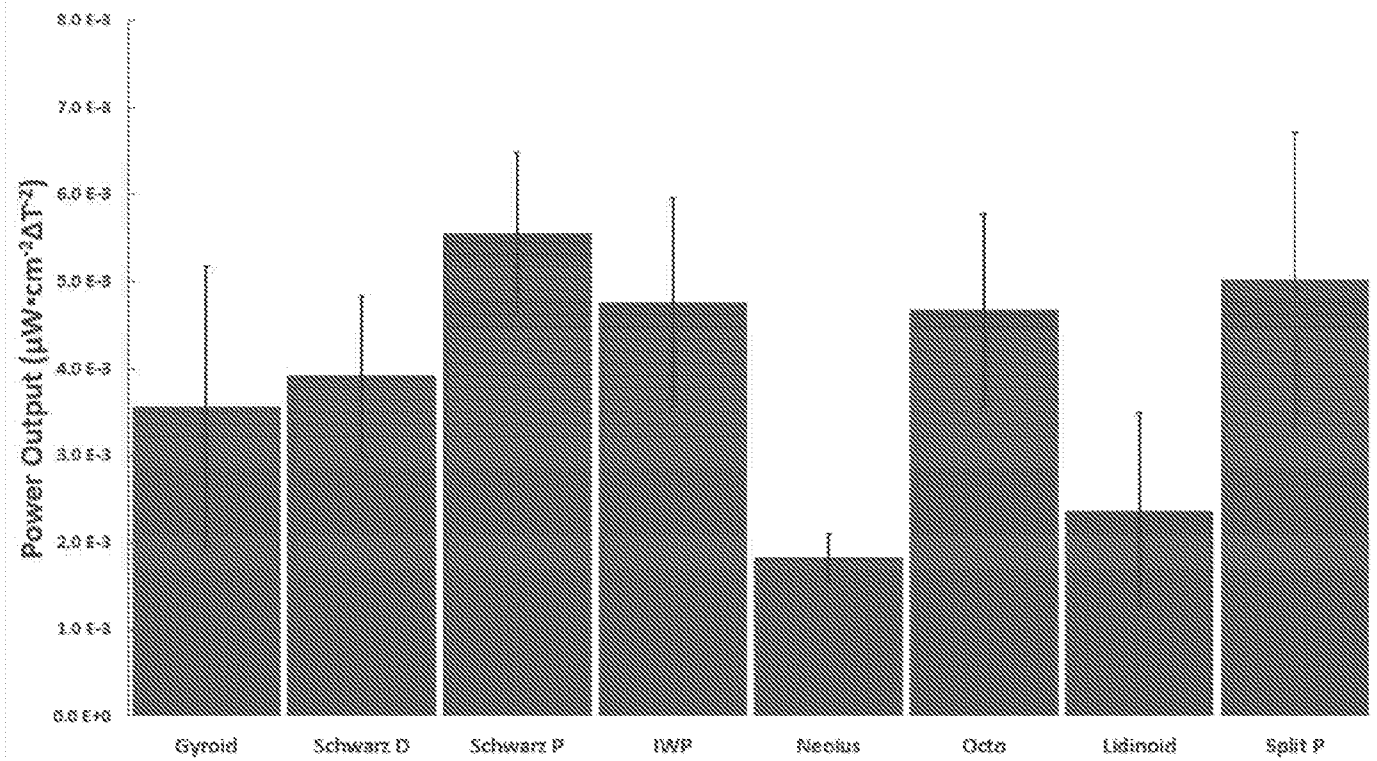


FIG. 12

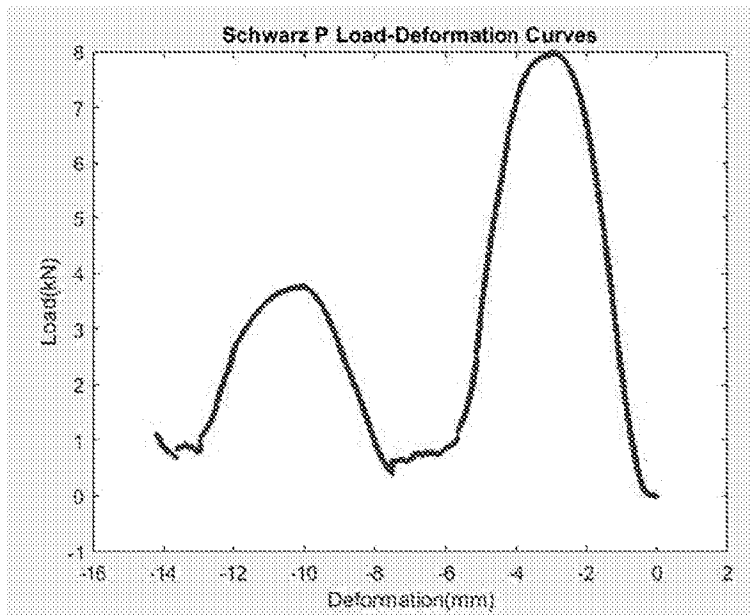


FIG. 13A

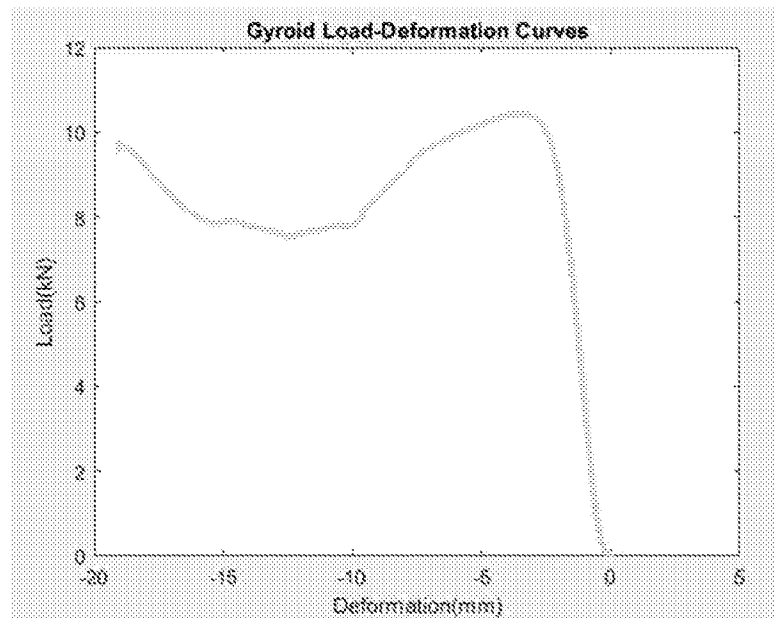


FIG. 13B

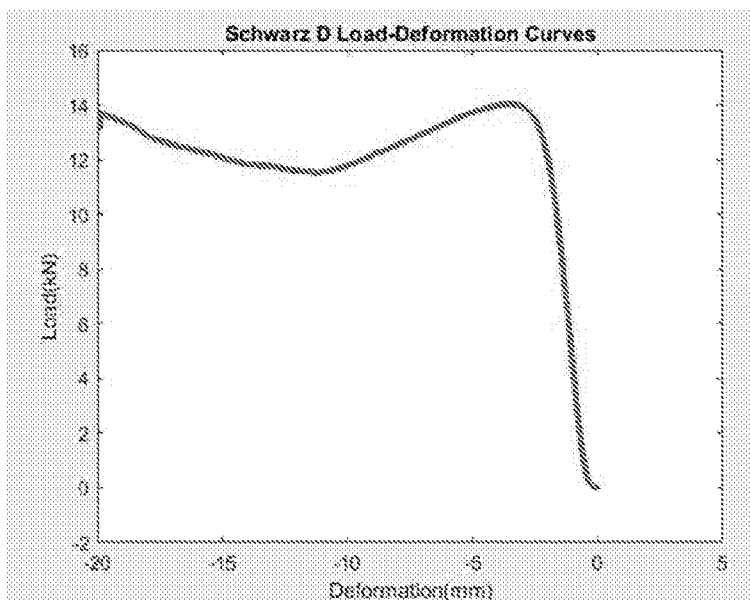


FIG. 13C

Potential Measurements 15.08.18

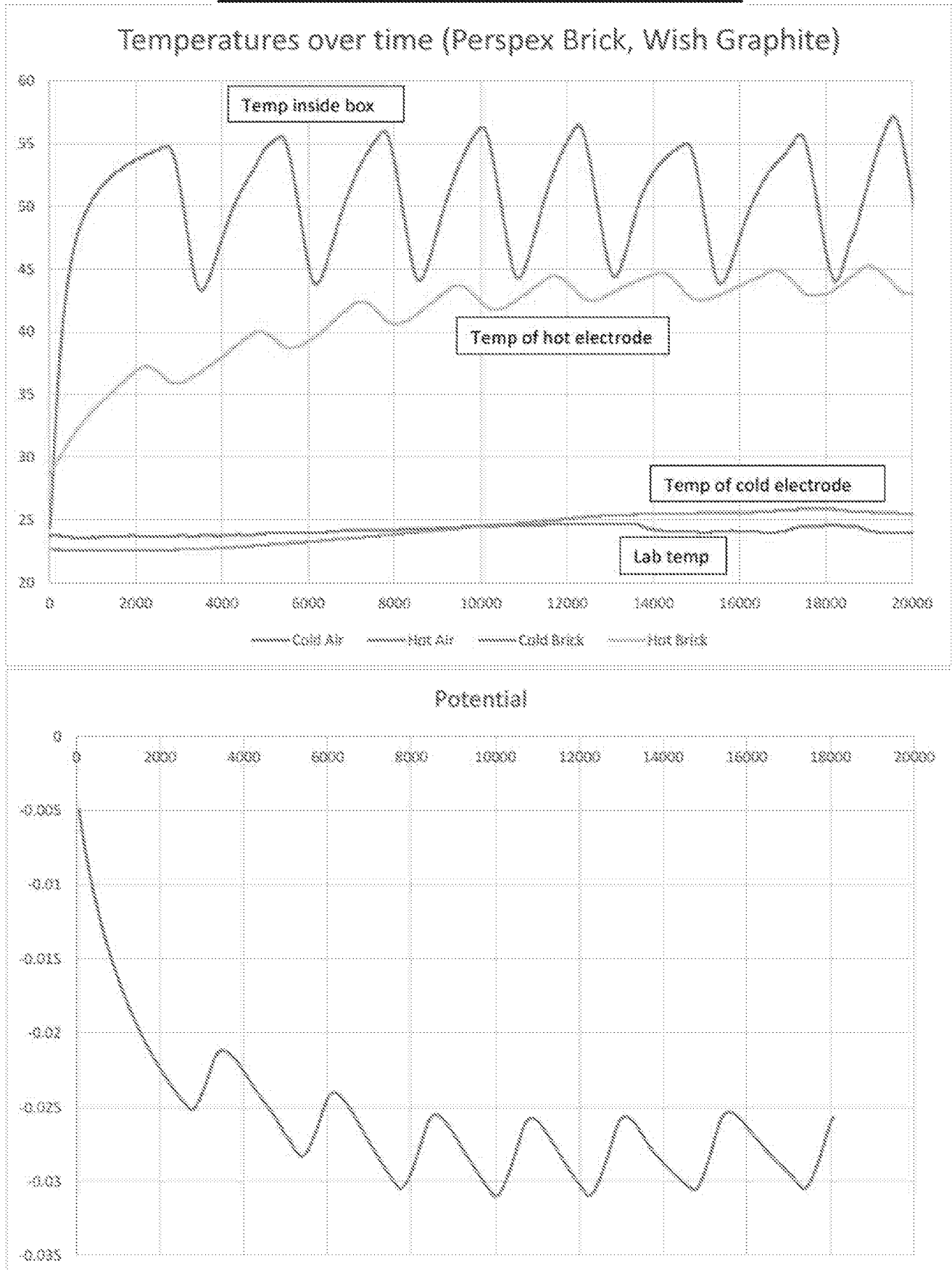
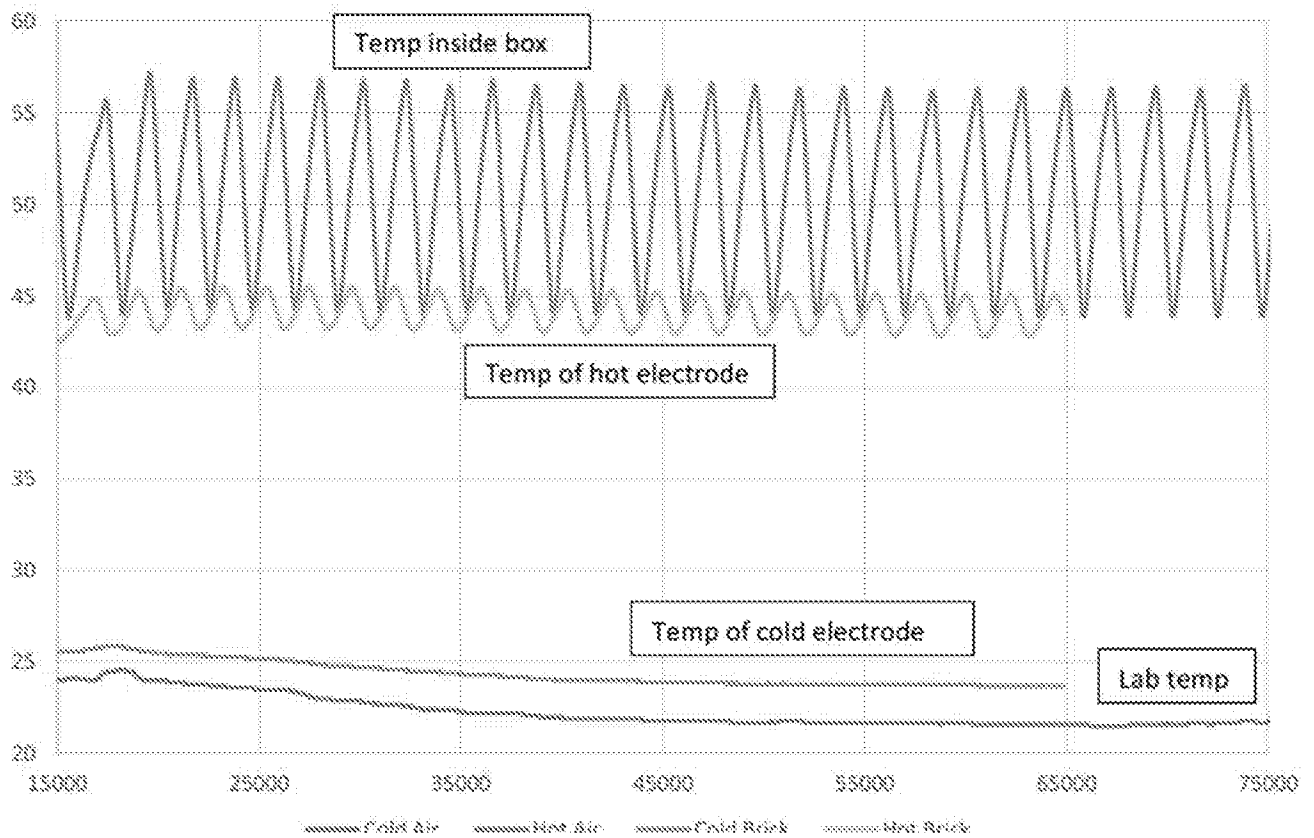


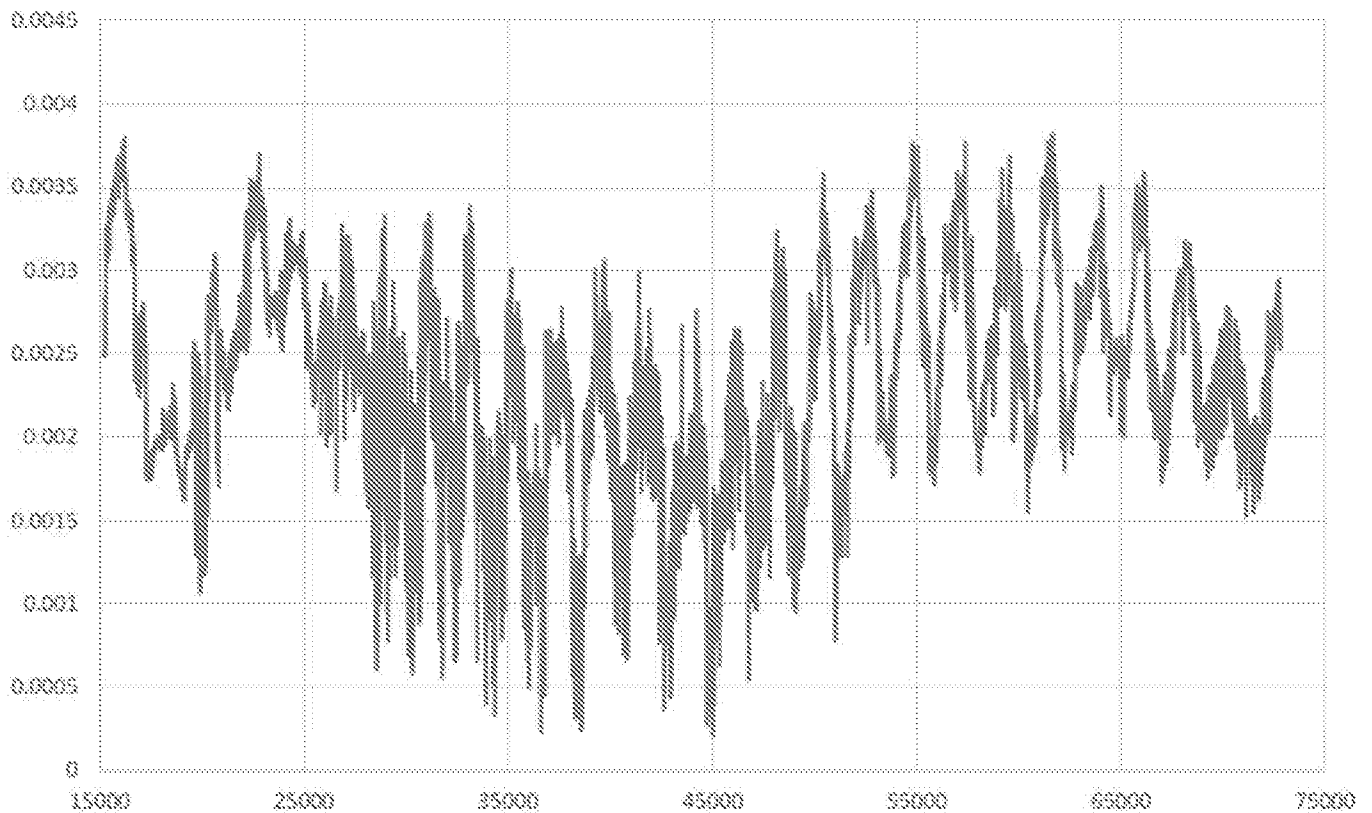
FIG. 14

Current Measurements 15.08.18

Temperatures over time (Perspex Brick, Wish Graphite)

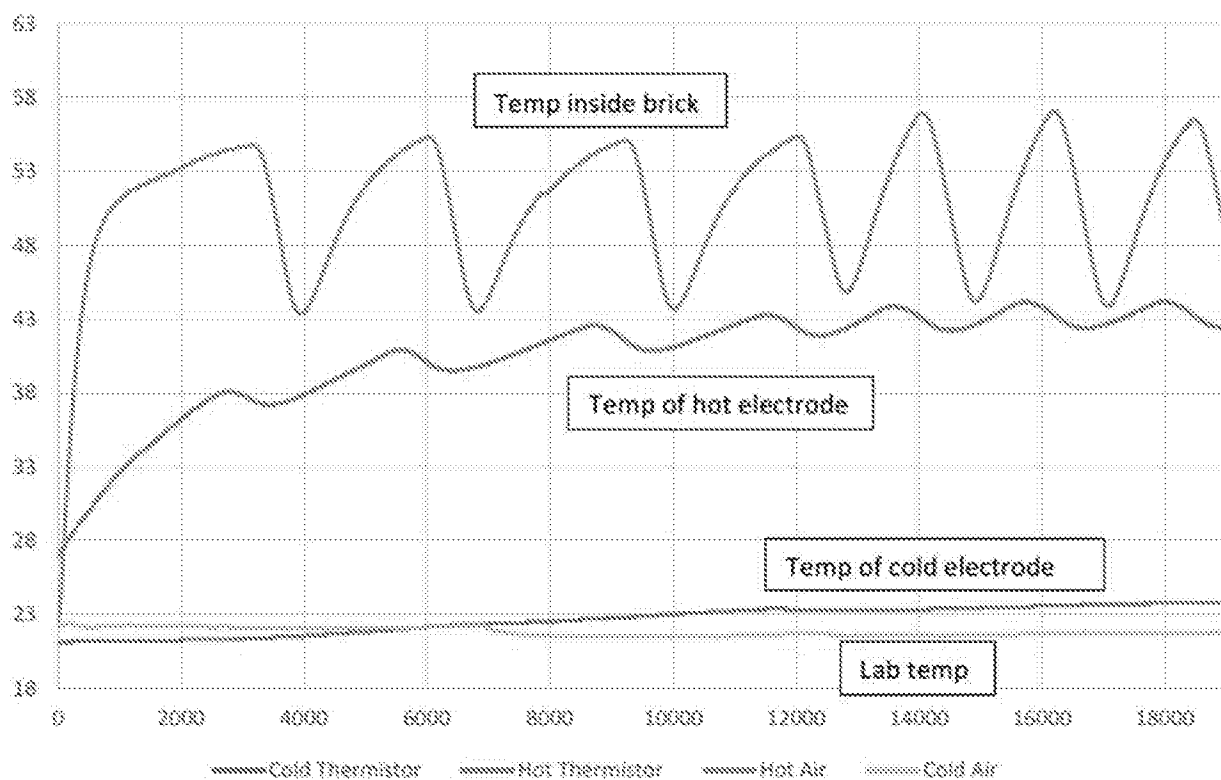


Current

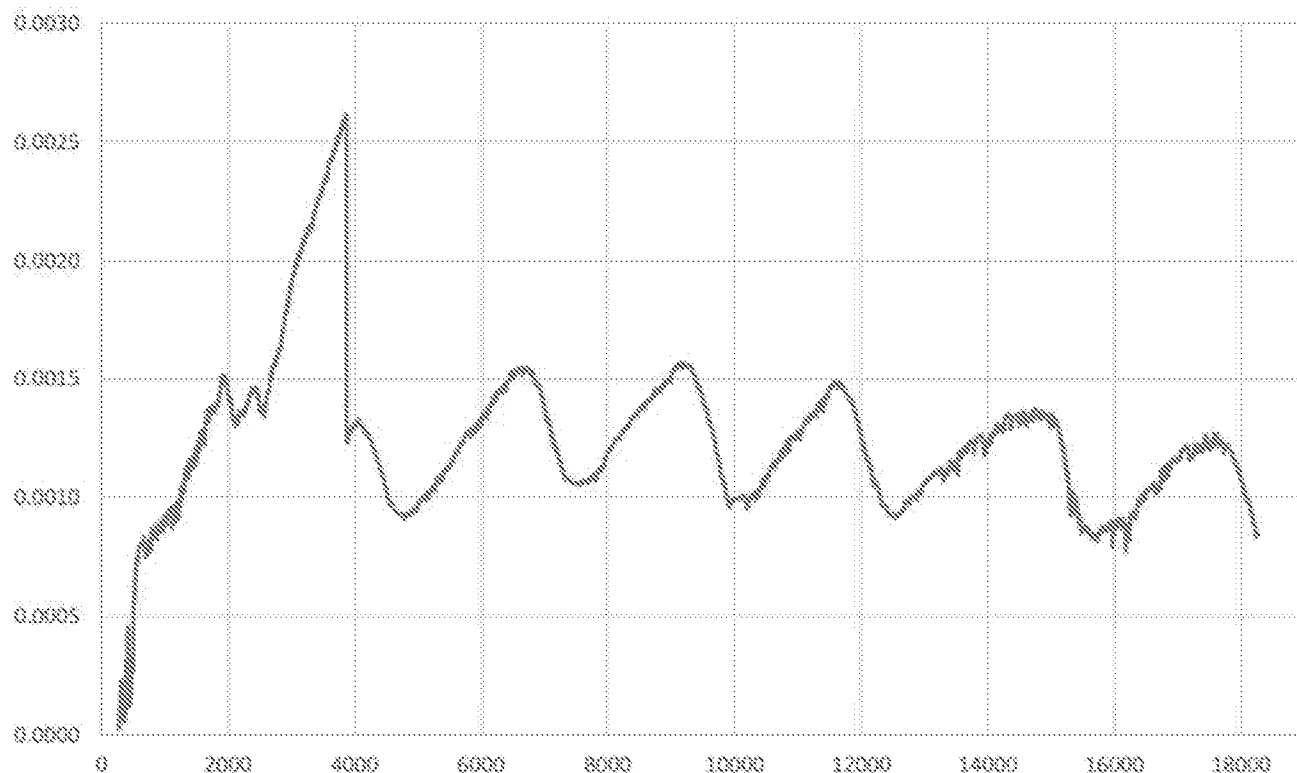


Current Measurements at Constant -0.015V 16.08.18

Temperatures over time (Perspex Brick, Wish Graphite)



Drawing current at a fixed potential (-15 mV)



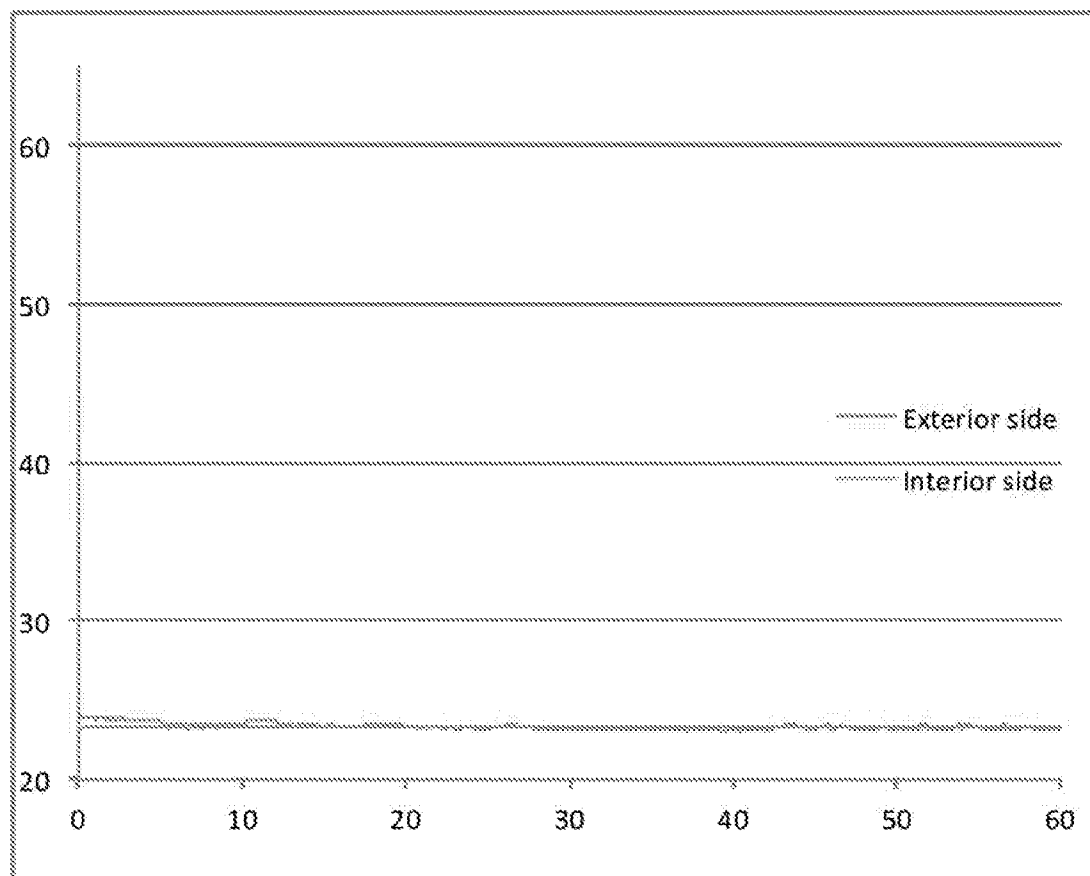


FIG. 17

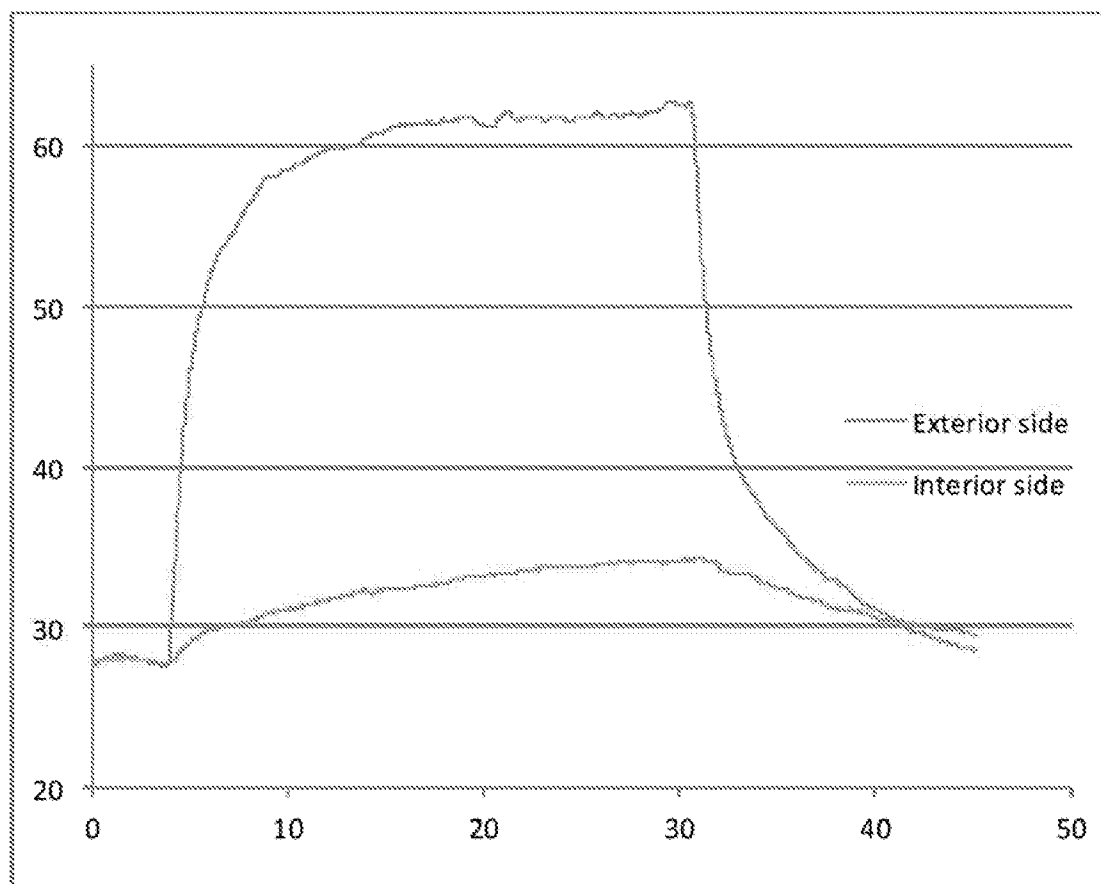


FIG. 18

Brick filled with air (5 min no heating; 25 min heating; 15 min no heating)

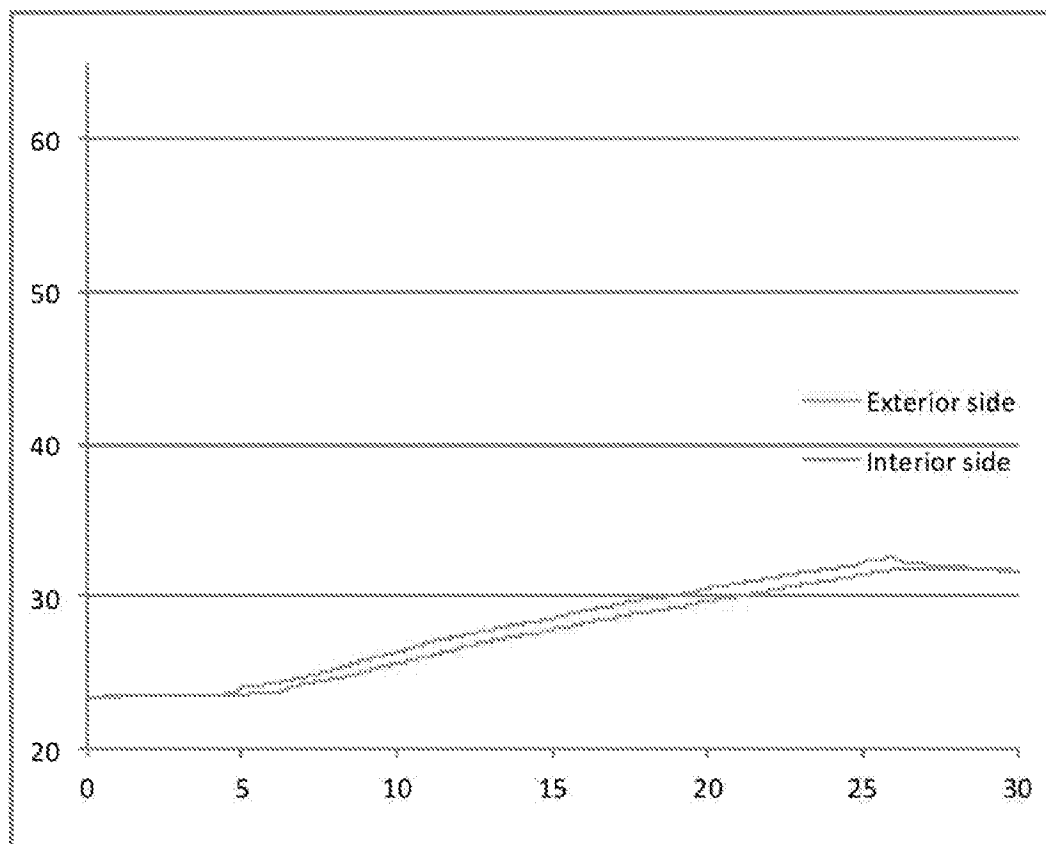


FIG. 19

Brick filled with water (5 min no heating; 20 min heating; 5 min no heating)

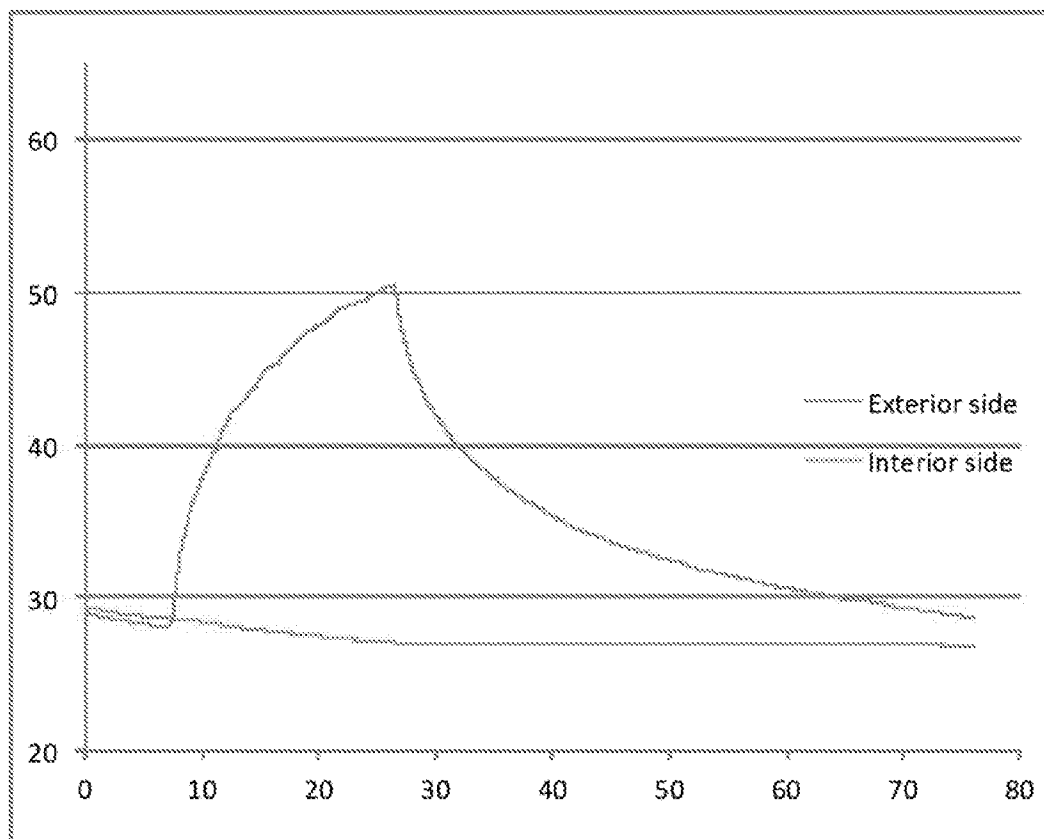


FIG. 20

Brick filled with gel (5 min no heating; 20 min heating; 50 min no heating)

INTERNATIONAL SEARCH REPORT

International application No.

PCT/US 19/57071

A. CLASSIFICATION OF SUBJECT MATTER

IPC - B28B 11/04; C04B 16/06; E04B 2/26 (2019.01)

CPC - B28B 11/043; C04B 28/18; C04B 2111/28; E04B 2/14

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

See Search History document

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

See Search History document

Electronic data base consulted during the international search (name of data base and, where practicable, search terms used)

See Search History document

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	WO 2017/092633 A1 (JIN), 8 June 2017 (08.06.2017), figures 1-4, pages 4-8 [via machine translation into English, retrieved 18 December 2019 (18.12.2019) from Google Patents].	1-7
A	US 4,376,155 A (PECK), 8 March 1983 (08.03.1983), figures 1-6, col 1 ln 12-14, col 5 ln 61-64, col 6 ln 8-10, col 6 ln 66-68, col 7 ln 1-12, col 9 ln 56-68, col 10 ln 1.	1-7
A	US 2015/0050543 A1 (ROBERT BOSCH GMBH), 19 February 2015 (19.02.2015), figures 1 and 8-11; para [0046]-[0050].	1-7
A	US 2,310,354 A (DEYSHER), 9 ebruary 1943 (09.02.1943), entire document.	1-7
P / X	LEE, 'Experimental Measurements of the Power Output of a Cu/Cu ²⁺ Thermogalvanic Brick', Arizona State University Digital Repository, December 2018 (12.2018), entire document, [retrieved 18 December 2019 (18.12.2019) via <http://hdl.handle.net/2286/R.A.211299>].	1-7

☐ Further documents are listed in the continuation of Box C.☐ See patent family annex.

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"Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art

"&" document member of the same patent family

Date of the actual completion of the international search

18 December 2019

Date of mailing of the international search report

15 JAN 2020

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