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(54) Title: POLYMER THICK FILM SILVER CONDUCTOR WITH INVERTED CURE PROFILE BEHAVIOR

(57) Abstract: This invention is directed to a polymer thick film conductor composition that provides a better conductor when dried at 80°C than when dried at 130°C, in contrast to typical PTF conductors. More specifically, the polymer thick film conductor may be used in applications where low temperature curing is required.

TITLEPOLYMER THICK FILM SILVER CONDUCTOR  
WITH INVERTED CURE PROFILE BEHAVIOR

5

FIELD OF THE INVENTION

This invention is directed to a polymer thick film (PTF) conductor composition that performs better when dried at 80°C than when dried at 130°C, in contrast to typical PTF conductors. More specifically, the polymer thick film conductor composition may be used in applications 10 where low temperature curing is required.

BACKGROUND OF THE INVENTION

Conductive PTF circuits have long been used as electrical elements in low voltage circuitry. Although they have been used for years in these types of applications, the use of PTF silver conductors in applications 15 involving curing at temperatures less than 90°C is not common. This is particularly important in circuits where a highly conductive silver composition is needed on base substrates that cannot withstand temperatures greater than 90°C. The typical substrates used are polyester and polycarbonate which can withstand 130°C drying cycles. 20 When a typical PTF conductor is dried at 130°C, it exhibits optimum properties such as low (15 mohm/sq/mil) resistivity and good adhesion to the substrate. When a typical PTF conductor is dried at 80°C, the resulting properties are inferior and not acceptable for functional circuitry. However, more and more applications are being developed which require 25 the use of polyvinylidene difluoride (PVDF), polyvinyl chloride (PVC) or other such substrates which must not be exposed to temperatures greater than 90°C. One of the purposes of this invention is to deal with this issue and thus formulate a PTF conductor with excellent performance when dried at 80°C, such that the properties match a typical PTF conductor 30 dried at 130°C.

## SUMMARY OF THE INVENTION

This invention relates to a polymer thick film conductor composition comprising:

- (a) 30-90 wt% silver powder; and
- 5 (b) 10-70 wt% organic medium comprising 10-60 wt% thermoplastic vinylidene difluoride/hexafluoro propylene copolymer resin dissolved in 40-90 wt% organic solvent consisting of triethyl phosphate, wherein the weight percent of the thermoplastic vinylidene difluoride/hexafluoro propylene copolymer resin and 10 the triethyl phosphate organic solvent are based on the total weight of the organic medium;
- wherein said silver powder is dispersed in said organic medium and wherein the weight percent of the silver powder and the organic medium are based on the total weight of the polymer thick film conductor 15 composition.

## DETAILED DESCRIPTION OF INVENTION

The invention relates to a polymer thick film conductor composition for use in low-temperature substrate electrical circuits. A layer of the 20 polymer thick film conductor composition is printed and dried at 80°C on the substrate so as to produce a functioning circuit. This polymer thick film conductor composition exhibits inverted cure behavior, i.e., it results in a lower resistivity when dried at 80°C as compared to when dried at 130°C.

The substrates commonly used in polymer thick film circuits are 25 polycarbonate (PC), polyester (PET) among others. PET is generally preferred since it can be processed at higher temperatures such as 130°C. However, for many applications, substrates such as polyvinyl chloride (PVC) and polyvinylidene difluoride (PVDF) are used and they can only be exposed to approximately 80°C maximum temperature before they 30 deform.

The polymer thick film (PTF) conductor composition is comprised of (i) silver powder dispersed in (ii) an organic medium comprising a polymer

resin dissolved in an organic solvent consisting of triethyl phosphate. Additionally, powders and printing aids may be added to improve the composition. Herein, weight percent will be written as wt%.

A. Conductive Powder

5 In an embodiment, the conductive powders in the instant polymer thick film conductor composition are Ag conductor powders and may comprise Ag metal powder, alloys of Ag metal powder, Ag-coated copper powder or mixtures thereof. As used herein, "silver powder" and "metal powder" include all of the powders enumerated in the previous sentence.

10 Various particle diameters and shapes of the metal powder are contemplated. In an embodiment, the conductive Ag powder may include silver particles of any shape, including particles that are spherical, in the form of flakes (rods, cones, plates), and mixtures thereof. In one embodiment, the conductive silver powder comprises particles selected

15 from the group consisting of silver flakes, silver-coated copper particles and mixtures thereof. In an embodiment, the conductive silver powder comprises particles in the form of silver flakes. In another embodiment, the conductive silver powder comprises a mixture of particles of silver and silver-coated copper. The use of silver-coated copper particles results in

20 increased resistivity but a lower cost than the use of only silver particles and this is useful in certain applications.

In an embodiment, the particle size distribution of the conductive powders may be 1 to 100 microns; in a further embodiment, 2-10 microns.

25 In an embodiment, the surface area/weight ratio of the silver particles may be in the range of 0.1-1.0 m<sup>2</sup>/g.

Furthermore, it is known that small amounts of other metals may be added to silver conductor compositions to improve the properties of the conductor. Some examples of such metals include: gold, nickel, aluminum, platinum, palladium, molybdenum, tungsten, tantalum, tin, 30 indium, lanthanum, gadolinium, boron, ruthenium, cobalt, titanium, yttrium, europium, gallium, sulfur, zinc, silicon, magnesium, barium, cerium, strontium, lead, antimony, conductive carbon, and combinations thereof

and others common in the art of thick film compositions. The additional metal(s) may comprise up to about 1.0 percent by weight of the total composition. As used herein, silver powder encompasses such additional metals.

5 In various embodiments, the silver powder is present at 30 to 90 wt%, 40 to 80 wt%, or 58 to 70 wt%, based on the total weight of the composition.

B. Organic Medium

10 The organic medium is comprised of a thermoplastic vinylidene difluoride/hexafluoro propylene (VF2/HFP) co-polymer resin dissolved in an organic solvent consisting of triethyl phosphate. Triethyl phosphate is a critical component of the instant polymer thick film conductor composition. Changing the solvent to something other than triethyl phosphate will render the composition ineffective in providing a suitable thick film 15 conductor after an 80°C drying. As always, the solvent used must solubilize the resin.

In various embodiments, the organic medium is present at 10 to 70 wt%, 20 to 60 wt%, or 30 to 42 wt%, based on the total weight of the composition.

20 In one embodiment, the thermoplastic VF2/HFP resin is 10-60 wt% and the triethyl phosphate is 40-90 wt% of the total weight of the organic medium. In another embodiment the thermoplastic VF2/HFP resin is 20-45 wt% and the triethyl phosphate is 55-80 wt% of the total weight of the organic medium and in still another embodiment the thermoplastic 25 VF2/HFP resin is 25-35 wt% and the triethyl phosphate is 65-75 wt% of the total weight of the organic medium.

The polymer resin is typically added to the organic solvent by mechanical mixing to form the organic medium.

30 After the silver powder is dispersed in the organic medium, additional triethyl phosphate may be added to the polymer thick film conductor composition to adjust the viscosity of the composition. The

amount of triethyl phosphate added for this purpose is included in the amounts of triethyl phosphate solvent indicated above.

#### ADDITIONAL POWDERS

Various powders may be added to the PTF conductor composition

5 to improve adhesion, modify the rheology and increase the low shear viscosity thereby improving the printability.

#### APPLICATION OF THE PTF CONDUCTOR COMPOSITION

The PTF conductor composition, also referred to as a "paste", is typically deposited on a substrate, such as PVDF or PVC, that is

10 impermeable to gases and moisture. The substrate can also be a sheet of a composite material made up of a combination of plastic sheet with optional metallic or dielectric layers deposited thereupon.

The deposition of the PTF conductor composition is performed typically by screen printing, but other deposition techniques such as stencil

15 printing, syringe dispensing or coating techniques can be utilized. In the case of screen-printing, the screen mesh size controls the thickness of the deposited thick film.

Generally, a thick film composition comprises a functional phase that imparts appropriate electrically functional properties to

20 the composition. The functional phase comprises electrically functional powders dispersed in an organic medium that acts as a carrier for the functional phase. Generally, the composition is fired to burn out both the polymer and the solvent of the organic medium and to impart the electrically functional properties. However, in the

25 case of a polymer thick film, the polymer portion of the organic medium remains as an integral part of the composition after drying.

The PTF conductor composition is processed for a time and at a temperature necessary to remove all solvent. For example, the deposited thick film is dried by exposure to heat at 80°C for typically 5 min.

30 CIRCUIT CONSTRUCTION

Substrates such as polyvinyl chloride and polyvinylidene difluoride which can only be exposed to approximately 80°C maximum temperature

are suitable for use with the instant PTF conductor composition. The PTF conductor composition is printed and dried as per the conditions described above. Several layers can be printed and dried.

The base substrate used in the Examples and Comparative Experiments was PET so that drying could be carried out at 80°C and 130°C to show the advantage of the instant PTF conductor composition.

## EXAMPLES AND COMPARATIVE EXPERIMENTS

### EXAMPLE 1

10 The PTF conductor composition was prepared in the following manner. 34.31 wt% of the organic medium was used and was prepared by mixing 32.5 wt% vinylidene difluoride/hexafluoro propylene copolymer resin, ADS2 (Arkema, Inc. King of Prussia, PA), with 67.5 wt% triethyl phosphate (Eastman Chemicals, Kingsport, TN.) organic solvent. The 15 wt% of the resin and the triethyl phosphate are based on the total weight of the organic medium. The molecular weight of the resin was approximately 20,000. This mixture was heated at 90°C for 1-2 hours to dissolve all the resin. 63.72 wt% of a flake silver powder with an average particle size of approximately 5 microns was added. Finally, 1.97 wt% of 20 triethyl phosphate was added to reduce the viscosity of the paste. The wt% of the vinylidene difluoride/hexafluoro propylene copolymer resin, the organic medium and the added triethyl phosphate were based upon the total weight of the PTF conductor composition.

This composition was mixed for 30 minutes on a planetary mixer, 25 and then subjected to several passes on the three roll-mill to provide the PTF conductor composition.

A circuit was then fabricated as follows. On a 5 mil thick PET substrate, the PTF conductor composition was used to print a pattern of a series of serpentine silver lines using a 280 mesh stainless steel screen. 30 The patterned lines were dried at 80°C for 5 min. in a forced air box oven. The circuit was inspected and tested for resistivity and adhesion. Excellent resistivity (15 mohm/sq/mil) and adhesion (no removal from

substrate using ASTM Tape test) was found. Results are shown in Table I.

#### COMPARATIVE EXPERIMENT A

5 A circuit was produced exactly as described in Example 1. The only difference was that the conductive pattern was dried at 130°C for 10 min. As can be seen in Table 1, the properties of the PTF silver composition were worse when dried at 130°C as compared to those when dried at 80°C. Results are shown in Table I.

10 COMPARATIVE EXPERIMENT B

A circuit was produced exactly as described in Example 1. The only difference was that dibasic esters were used as the organic solvent instead of triethyl phosphate. Results of drying at 130°C vs 80°C were typical of a PTF conductor (better performance at higher temperatures).

15 The excellent performance at 80°C was not seen. Results are shown in Table I.

#### COMPARATIVE EXPERIMENT C

A circuit was produced exactly as described in Example 1. The only difference was that a commercial silver conductor paste, DuPont

20 5025 (DuPont Co., Wilmington, DE) was used to form the circuit. Results of drying at 130°C vs 80°C were typical of a PTF conductor (better performance at higher temperatures). The excellent performance at 80°C was not seen. Results are shown in Table I.

#### EXAMPLE 2

25 A circuit was produced exactly as described in Example 1 with the exception that the composition used here had an approximate 50/50 blend of flake silver and silver-coated copper (Ames Corporation; 3 µm average particle size). As can be seen in Table 1, the resistivity of the PTF silver composition was higher when dried at 130°C as compared to that when 30 dried at 80°C. Excellent adhesion was found. Results are shown in Table I.

**Table I**

Composition	Resistivity (Drying Temp. 80°C)	Resistivity (Drying Temp. 130°C)	Adhesion PET 0=Worst; 5 = Best
Example 1	15.5 mohm/sq/mil		4
Comparative Experiment A		40.2 mohm/sq/mil	3
Comparative Experiment B	50 mohm/sq/mil	16 mohm/sq/mil	4
Comparative Experiment C	45 mohm/sq/mil	13 mohm/sq/mil	5
Example 2	66 mohm/sq/mil	75 mohm/sq/mil	5

5

CLAIMS

What is claimed is:

- 5 1. A polymer thick film conductor composition, comprising:
  - (a) 30-90 wt% silver powder; and
  - (b) 10-70 wt% organic medium comprising 10-60 wt% thermoplastic vinylidene difluoride/hexafluoro propylene copolymer resin dissolved in 40-90 wt% organic solvent
- 10 consisting of triethyl phosphate, wherein the weight percent of said thermoplastic vinylidene difluoride/hexafluoro propylene copolymer resin and said triethyl phosphate organic solvent are based on the total weight of said organic medium;
- 15 wherein said silver powder is dispersed in said organic medium and wherein the weight percent of said silver powder and said organic medium are based on the total weight of said polymer thick film conductor composition.
- 20 2. The polymer thick film conductor composition of claim 1, said silver powder comprising particles selected from the group consisting of silver flakes, silver-coated copper particles and mixtures thereof.
- 25 3. An electrical circuit comprising a polymer thick film conductor formed from the polymer thick film conductor composition of claim 1.
4. The electrical circuit of claim 3, wherein said polymer thick film conductor composition has been dried at a temperature less than 90°C.
- 30 5. The electrical circuit of claim 4, said electrical circuit further comprising a substrate of polyvinylidene difluoride or polyvinyl chloride.

# INTERNATIONAL SEARCH REPORT

International application No  
PCT/US2015/039012

**A. CLASSIFICATION OF SUBJECT MATTER**  
INV. H01B1/22 H05K1/09 B05D5/12 C08K3/08 C09D127/16  
ADD. C08L27/16

According to International Patent Classification (IPC) or to both national classification and IPC

**B. FIELDS SEARCHED**

Minimum documentation searched (classification system followed by classification symbols)  
H01B H05K B05D C08K C09D C08L

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practicable, search terms used)

EPO-Internal, CHEM ABS Data, WPI Data

**C. DOCUMENTS CONSIDERED TO BE RELEVANT**

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	EP 1 538 637 A1 (DU PONT [US]) 8 June 2005 (2005-06-08) ----- A paragraphs [0002], [0003], [0011], [0014]; example 1	3,4
A	----- EP 1 538 180 A2 (DU PONT [US]) 8 June 2005 (2005-06-08) A paragraphs [0003], [0004], [0013], [0014]; example 1 -----	1,2,5
X	EP 1 538 180 A2 (DU PONT [US]) 8 June 2005 (2005-06-08) A paragraphs [0003], [0004], [0013], [0014]; example 1 -----	3,4
A	-----	1,2,5



Further documents are listed in the continuation of Box C.



See patent family annex.

\* Special categories of cited documents :

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**INTERNATIONAL SEARCH REPORT**

Information on patent family members

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