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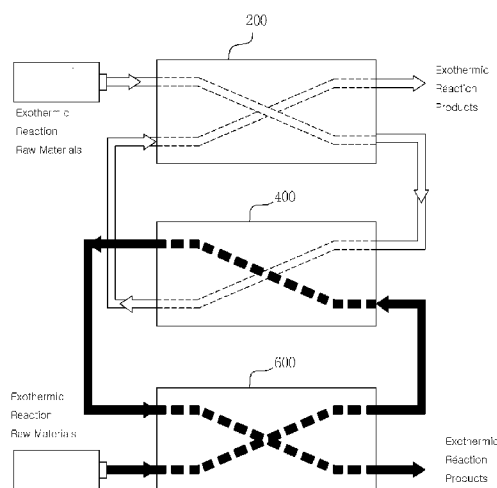
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(54) Title: A CHEMICAL REACTOR INCLUDING HEAT EXCHANGERS



(57) Abstract: The present invention discloses a chemical reactor with high heat efficiency and small volume, the chemical reaction of the present invention comprises a first unit for performing heat exchange of exothermic reaction products with exothermic reaction raw material fed for exothermic reaction; a second unit including a plate assembly for exothermic reactions and a plate assembly for endothermic reactions; and a third unit for performing heat exchange of endothermic reaction products with endothermic reaction raw material fed for endothermic reactions, the exothermic reaction raw material discharged from the first unit is fed into the plate assembly for exothermic reactions in the second unit and used in the exothermic reactions, the exothermic reaction products are fed into the first unit to pre-heat the exothermic reaction raw material, the endothermic reaction raw material discharged from the third unit is fed into the plate assembly for endothermic reactions in the second unit and used in the endothermic reactions, the endothermic reaction products are fed into the third unit to pre-heat the endothermic reaction raw material, and the first unit, the second unit and the third unit are laminately disposed.

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Description

A CHEMICAL REACTOR INCLUDING HEAT EXCHANGERS

Technical Field

- [1] The present invention relates to a chemical reactor with high heat efficiency and small volume. More particularly, the present invention relates to a chemical reactor which may be miniaturized and can heat-exchange effectively to improve heat efficiency, and thus effectively used in a reactor module for reforming hydrocarbons, and the like

[2]

Background Art

- [3] Recently, interest of environment problems is increased and thus researches for fuel cells and the like are vigorously conducted as alternatives to gasoline engines and fossil-fuel generation. In order to increase availability of fuel cells, every condition should be provided, which can smoothly produce and/or feed hydrogen used in fuels.
- [4] Since hydrogen is the lightest among gases and can be easily exploded in air, its storage and handling are very difficult. Therefore, it is in the currently technical level to feed hydrogen simply using large volumes of hydrogen storage tanks. Such hydrogen storage tanks have disadvantages which are difficult to completely solve the previously mentioned problems.
- [5] In addition, there is a disadvantage that initial equipment investments such as developing large volumes of hydrogen storage tanks and preparing larger volumes of hydrogen reservoirs should be vastly required. It cannot help being such a vast project as would be discussed as infrastructures in a national level.
- [6] Therefore, a plan of miniaturizing a chemical reactor such as a reforming reactor producing hydrogen to reduce weight and volume is very desirable in that hydrogen may be used in clean fuels without requiring vast equipment investments.
- [7] Production apparatuses of hydrogen for feeding it into fuel cells roughly consist of three units. That is, they may be divided by a reformer for reforming fuels to produce hydrogen, carbon dioxide and carbon monoxide, a water gas shift reactor for reacting carbon monoxide with a steam to lower a concentration of carbon monoxide to 1% and increase a concentration of hydrogen, and a selective CO oxidation reactor for lowering a concentration of the remaining carbon monoxide to 10 ppm or less.
- [8] Especially, in case of a steam reforming reaction, the reaction is an endothermic reaction performed at a high temperature of 650 °C or more, and thus needs sustained heat supply. As heat supply sources, burners are used or catalytic combustion is utilized. In addition, the water gas shift reaction occurs at a temperature of 250 to 450

°C, so that heat exchange of reformed gases discharged via the reformer is necessary.

[9] Furthermore, when the catalytic combustion is utilized, it is advantageous to inject the pre-heated combustion gases, and also advantageous to pre-heat and inject the reforming fuels. For such pre-heating, separate pre-heaters should be used or the combustion gases or the reforming fuels should be heat-exchanged with burned stack gases or reforming gases exited from the reformer.

[10] In KR Unexamined Patent Publication No. 1999-014655, a reformer, which could heat the reformer using combustion catalysts and simultaneously perform a function of developing steam and a function of pre-heating a mixture of air and fuels using heat of stack gases, was prepared. However, a heat exchanger is separately disposed so that heat loss occurred in the middle cannot be prevented. Therefore, it has disadvantages that heat efficiency is lowered and that the apparatus is complicated.

[11] In US Patent No. 6,998,096 B2, a burner, a reformer, a heat exchanger and an evaporator were integrally constituted to increase heat efficiency. However, said technique has disadvantages that the reactor has so large volume that heat loss through wall surfaces cannot help being large, and the heat exchanger is positioned on top of the reformer and thus heat loss becomes large through the side wall surfaces of the relatively wide reformer.

[12]

Disclosure of Invention

Technical Problem

[13] The present invention is conceived to solve the problems of heat loss of the chemical reactors such as reforming reaction apparatuses described above, an object of the present invention to provide an integral type reactor module having a more simple structure.

[14] To achieve the above object, the present invention uses plates with fluid passages through which fluids could pass and has a flow of fluids designed for achieving the effective heat exchange to minimize fluid lines exposed to an outside. In addition, the chemical reactor of the present invention utilizes micro channels to increase heat transfer efficiency.

[15]

Technical Solution

[16] The chemical reactor according to the present invention comprises

[17] a first unit including one or more heat exchange plate assemblies for exothermic reaction raw material and one or more heat exchange plate assemblies for exothermic reaction products, a heat exchange between exothermic reaction products and exothermic reaction raw material being performed in the first unit;

- [18] a second unit including one or more plate assemblies for exothermic reaction and one or more plate assemblies for endothermic reaction, the second unit providing heat generated through an exothermic reaction of said exothermic reaction raw material to endothermic reaction raw material to proceed the endothermic reaction; and
- [19] a third unit including one or more heat exchange plate assemblies for endothermic reaction raw material and one or more heat exchange plate assemblies for endothermic reaction products, a heat exchange between endothermic reaction products and endothermic reaction raw material being performed in the third unit.
- [20] The term "exothermic reaction raw material" used herein means raw material (for example, combustion gases) to be used in exothermic reactions such as combustion reactions, and the term "exothermic reaction products" means a state (for example, burned combustion gases [stack gases]) after subjecting the "exothermic reaction raw material" to exothermic reactions. In addition, the term "endothermic reaction raw material" means raw material (for example, hydrocarbon or alcohol) to be used in endothermic reactions such as reforming reactions of hydrocarbons, and the term "endothermic reaction products" means a state (for example, reforming gases) after subjecting the "endothermic reaction raw material" to endothermic reactions.
- [21]
- [22] *In particular, the chemical reactor of the present invention as mentioned above may be effectively used as a reactor module for reforming hydrocarbons which reforms hydrocarbon raw material to produce hydrogen. At this time, in the chemical reaction of the present invention, the exothermic reaction is a combustion reaction of combustion gases, and endothermic reaction is a reforming reaction for reforming hydrocarbon raw material to produce hydrogen.
- [23] In the chemical reactor of the present invention as described above, it is preferably to minimize a flow of fluids to be exposed to an outside, by means of the first and fourth plates placed on the first unit and below the third unit, respectively and the second and third plates having through holes and placed between two units, respectively.
- [24] Specifically, the chemical reactor according to the present invention further comprises, preferably,
- [25] a first plate for supplying exothermic reaction raw material to the heat exchange plate assembly for exothermic reaction raw material and discharging exothermic reaction products introduced from the heat exchange plate assembly for exothermic reaction products;
- [26] a second plate for supplying exothermic reaction raw material introduced from the heat exchange plate assembly for exothermic reaction raw material to the plate assembly for exothermic reaction, and discharging endothermic reaction raw material

and exothermic reaction products introduced from a third plate to the plate assembly for endothermic reaction and the heat exchange plate assembly for exothermic reaction products, respectively;

- [27] a third plate for supplying endothermic reaction raw material introduced from the heat exchange plate assembly for exothermic reaction raw material and exothermic reaction products introduced from the plate assemblies for exothermic reaction to the second plate, and supplying endothermic reaction products introduced from the plate assembly for endothermic reaction to the heat exchange plate assembly for endothermic reaction products; and
- [28] a fourth plate for supplying endothermic reaction raw material to the heat exchange plate assembly for endothermic reaction raw material, and discharging endothermic reaction products introduced from the heat exchange plate assembly endothermic reaction products.
- [29] Considering the volume of reactor, in addition, it is preferably that the first plate, the first unit, the second plate, the second unit, the third plate, the third unit and the fourth plate are laminately disposed order in the chemical reactor of the present invention. Also, it is preferably that the heat exchange plate assemblies for exothermic reaction raw material and the heat exchange plate assemblies for exothermic reaction products constituting the first unit are alternately laminated, the plate assemblies for exothermic reaction and the plate assemblies for endothermic reaction constituting the second unit are alternately laminated, the heat exchange plate assemblies for endothermic reaction raw material and the heat exchange plate assemblies for endothermic reaction products constituting the third unit are alternately laminated, respectively.
- [30] Further, considering heat exchange ability, in each plate assembly constituting said first to third units, it is preferably that the first channel plate having one or more micro channels formed on a lower surface thereof and the second channel plate having one or more formed on an upper surface thereof and corresponding micro channels of the first plate are joined to form one or more micro fluid passages.

[31]

Brief Description of the Drawings

- [32] Fig. 1 is a view schematically depicting an entire structure of the chemical reactor according to the present invention;
- [33] Fig. 2 is an exploded perspective view of the chemical reactor according to one aspect of the present invention;
- [34] Fig. 3 is an exploded perspective view of plate assemblies constituting the chemical reactor of Fig. 2;

- [35] Fig. 4 is a plane view of a channel plate constituting the plate assemblies of Fig. 3;.
- [36] Fig. 5 is a plane view of a gasket used in a plate assembly according to the present invention; and
- [37] Fig. 6 is a side cross-sectional view of a plate assembly constituting the chemical reactor of Fig. 2.

[38]

Best Mode for Carrying Out the Invention

- [39] The present invention is explained in detail below, with reference to the accompanying drawings.

[40] Fig. 1 is a view schematically depicting an entire structure of a chemical reactor according to the present invention. As shown in Fig. 1, the chemical reactor of the present invention comprises a first unit **200** for performing heat exchange of exothermic reaction products with exothermic reaction raw material fed for exothermic reaction; a second unit **400** including a plate assembly for exothermic reaction and a plate assembly for endothermic reaction; and a third unit **600** for performing heat exchange of endothermic reaction products with endothermic reaction raw material fed for endothermic reaction. At this time, the exothermic reaction raw material discharged from the first unit are fed into the plate assembly for exothermic reactions in the second unit and used in the exothermic reactions, the exothermic reaction products is fed into the first unit to pre-heat the exothermic reaction raw material, the endothermic reaction raw material discharged from the third unit are fed into the plate assembly for endothermic reactions in the second unit and used in the endothermic reactions, and the endothermic reaction products is fed into the third unit to pre-heat the endothermic reaction raw material.

- [41] The chemical reactor of the present invention as described above may be used, for example, as a reactor module for reforming hydrocarbons. The exothermic reaction is a combustion reaction of combustion gases, and the endothermic reaction is a reaction for reforming hydrocarbon raw material to produce hydrogen. In this reforming reaction, the exothermic reaction raw material, exothermic reaction products, endothermic reaction raw material and endothermic reaction products correspond to combustion gases, burned stack gases, fuels and reforming gases, respectively.

[42] Fig. 2 is an exploded perspective view of a chemical reactor according to the present invention, the chemical reactor comprises;

- [43] a first plate **100** which exothermic reaction raw material is fed, exothermic reaction products being discharged from the first plate;

[44] a first unit **200** including a heat exchange plate assembly **210** for exothermic reaction raw material to which exothermic reaction raw material discharged from the

first plate **100** is in-flowed and a heat exchange plate assembly **220** for exothermic reaction products to which exothermic reaction products discharged from the second plate **300** is in-flowed;

[45] a second plate **300** for feeding exothermic reaction raw material in-flowed from the first unit **200** into a plate assembly **410** for exothermic reactions and feeding endothermic reaction raw material in-flowed from a third plate **500** into a plate assembly **420** for endothermic reactions, exothermic reaction products discharged from the third plate **500** being in-flowed to the second plate;

[46] a second unit **400** including a plate assembly **410** for exothermic reactions to which exothermic reaction raw material discharged from the second plate **300** is in-flowed and a plate assembly **420** for endothermic reactions to which endothermic reaction raw material are in-flowed;

[47] the third plate **500** for discharging exothermic reaction products exited from the second unit **400** into the second plate **300**, feeding endothermic reaction products exited from the second unit **400** into a third unit **600**, and discharging endothermic reaction raw material in-flowed from the third unit **600** into the second plate **300**;

[48] a third unit **600** including a heat exchange plate assembly **620** for endothermic reaction products to which endothermic reaction products discharged from the third plate **500** is in-flowed and a heat exchange plate assembly **610** for endothermic reaction raw material to which endothermic reaction raw material discharged from the fourth plate **700**; is in-flowed; and

[49] the fourth plate **700** for discharging endothermic reaction products, endothermic reaction raw material being fed into the fourth plate.

[50] The configuration of each member and the interconnection among the members constituting the chemical reactor according to the present invention are explained in detail, via the attached drawings below.

[51]

[52] The First Plate **100**

[53] The first plate **100** acting as a top cover of the chemical reactor and feeding exothermic reaction raw material and discharging exothermic reaction products is a flat-shaped member, and is equipped with an inflow port **101** to which exothermic reaction raw material is introduced and an exhaust port **105** through which exothermic reaction products is discharged are formed at both side ends thereof. In addition, lower ends of the inflow port **101** and the exhaust port **105** are connected to the first unit **200** described below.

[54]

[55] The First Unit **200**

[56] The first unit **200** is a part in which exothermic reaction raw material is heat

exchanged with exothermic reaction products and positioned below the first plate **100**. The first unit **200** includes the heat exchange plate assembly **210** for exothermic raw material to which the exothermic reaction raw material is in-flowed, and the heat exchange plate assembly **220** for exothermic reaction products to which exothermic reaction products is in-flowed. Each heat exchange plate assembly is constituted by coupling the first channel plate with the second channel plate, and the first and second channel plate have the same structure and are present in a mirror image. When the first channel plate and the second channel plate are overlapped, concave portions formed on an inner surface of the first channel plate and concave portions formed on an inner surface of the second channel plate correspond to each other to form one or more micro fluid passages.

[57] The inflow port **101** of the first plate **100** is connected to an inlet **211** formed on the heat exchange plate assembly **210** for exothermic reaction raw material, and an outlet **212** formed on the heat exchange plate assembly **210** for exothermic raw material is connected to a through-hole **225** formed on the heat exchange plate assembly **220** for exothermic reaction products. Therefore, the exothermic reaction raw material in-flowed from the first plate **100** flows along the fluid passages formed in the heat exchange plate assembly **210** for exothermic raw material, and then in-flowed into the through-hole **305** of the second plate **300**.

[58] Meanwhile, an inlet **221** formed on the heat exchange plate assembly **220** for exothermic reaction products is connected to a first inflow port **301** formed on the second plate **300**, and an outlet **222** formed on the heat exchange plate assembly **220** for exothermic reaction products is connected to a through-hole **215** formed on the heat exchange plate assembly **210** for exothermic reaction raw material. Therefore, exothermic reaction products introduced from the second plate **300** flows along the fluid passages formed in the heat exchange plate assembly **220** for exothermic reaction products and are then discharged to an exhaust port **105** of the first plate **100** via the through-hole **215** formed on the heat exchange plate assembly **210** for exothermic reaction raw material. At this time, the exothermic reaction products with a high temperature of 500 °C or more that passes through the heat exchange plate assembly **220** for exothermic reaction products provides the exothermic reaction raw material that passes through the heat exchange plate assembly **210** for exothermic reaction raw material with heat to pre-heat exothermic reaction raw material by a temperature of about 500 °C.

[59] Of course, considering the heat transfer efficiency and the like, one or more heat exchange plate assemblies for exothermic reaction raw material to which exothermic reaction raw material such as combustion gases is introduced and one or more heat exchange plate assemblies for exothermic reaction products to which exothermic

reaction products such as stack gases is introduced may be alternately laminated in such first unit **200**.

[60]

[61] The second plate **300**

[62] On the second plate **300** positioned below the first unit **200**, a first inflow port **301** to which exothermic reaction products is introduced and a second inflow port **302** to which heat-exchanged endothermic reaction raw material is introduced are formed, and the through-hole **305** is formed for passing through heat-exchanged exothermic reaction raw material.

[63] The through-hole **305** formed on the second plate **300** is connected to the through-hole **225** of the heat exchange plate assembly **220** for exothermic reaction products constituting the upper first unit **200** and an inlet **411** of the plate assembly **410** for exothermic reactions constituting the lower second unit **400** described below.

[64] In addition, the first inflow port **301** is connected to an inlet **221** of the heat exchange plate assembly **220** for exothermic reaction products constituting the upper first unit **200**, and the second inflow port **302** is connected to a through-hole **415** of the plate assembly **410** for exothermic reactions constituting the lower second unit **400**.

[65]

[66] The Second Unit **400**

[67] The second unit **400** is a part in which exothermic reaction raw material pre-heated in the first unit **200** is exothermically reacted to provide heat to be used in the endothermic reactions and to proceed an endothermic reactions of endothermic reaction raw material pre-heated in the third unit **600**.

[68] The second unit **400** positioned below the second plate **300** comprises the plate assembly **410** for an exothermic reactions and the plate assembly **420** for an endothermic reactions. Here, the configurations of the plate assembly **410** for exothermic reactions and the plate assembly **420** for endothermic reactions, which constitute the second unit **400**, are the same as those of the heat exchange plate assembly **210** for exothermic reaction raw material and the heat exchange plate assembly **220** for exothermic reaction products, which constitute the first unit **200**.

[69] The through-hole **305** of the second plate **300** is connected to the inlet **411** formed on the plate assembly **410** for exothermic reactions, and an outlet **412** formed on the plate assembly **410** for exothermic reactions is connected to a through-hole **425** formed on the plate assembly **420** for endothermic reactions. Therefore, exothermic reaction raw material such as combustion gases in-flowed from the second plate **300** is subjected to exothermic reaction such as combustion reaction, with flowing along fluid passages formed in the plate assembly **410** for exothermic reactions, and then in-flowed into the first exhaust port **501** formed on the third plate **500** through which

exothermic reaction products is discharged. Accordingly, exothermic reaction raw material passing through the plate assembly **410** for exothermic reactions continuously provide heat to endothermic reaction raw material passing through the plate assembly **420** for endothermic reactions to proceed the endothermic reactions.

[70] On the other hand, an inlet **421** formed on the plate assembly **420** for endothermic reactions is connected to the second inflow port **302** formed on the second plate **300** via the through-hole **415** formed on the plate assembly **410** for exothermic reactions, and an outlet **422** formed on the plate assembly **420** for endothermic reactions is connected to a through-hole **505** formed on the third plate **500**. Therefore, endothermic reaction raw material in-flowed from the second plate **300** is subjected to endothermic reaction, with flowing along fluid passages formed within the plate assembly **420** for endothermic reactions, passes through the through-hole **505** formed on the third plate **500**, and are then in-flowed into a heat exchange plate assembly **620** for endothermic reaction products of the third unit **600**.

[71] Of course, considering the reaction capacity, one or more plate assemblies for exothermic reactions and one or more plate assemblies for endothermic reactions may be alternately laminated in the second unit **400** as described above. In the present invention, in addition, the fluid passages formed within the plate assembly for exothermic reactions may be coated with a catalyst for exothermic reactions, and the fluid passages formed within the plate assembly for endothermic reactions may be coated with a catalyst for endothermic reactions.

[72] Particular kinds of catalysts for exothermic reactions and endothermic reactions are not specifically limited. Depending on kinds of reactions being proceeding, usual catalysts in chemical fields may be used without limitation. For example, if the chemical reactor of the present invention is used in a reforming reaction of hydrocarbons, the catalyst for exothermic reactions may be a catalyst for combustion, and the catalyst for endothermic reactions may be a reforming catalyst. At this time, the catalyst for combustion is not specifically limited, as long as it can be used in catalyzing oxidation of hydrocarbons or alcohols, and the like. Examples of such catalyst for combustion may be one or more selected from the group consisting of a platinoid element such as platinum (Pt), rhodium (Rh), ruthenium (Ru), osmium (Os), iridium (Ir), and palladium (Pd); a precious metal element such as gold (Au), silver (Ag), and copper (Cu); or a mixture thereof. In addition, the catalyst for combustion may be used by supporting it on a catalyst support such as aluminum oxide, α , γ , θ -aluminum oxide, cerium oxide (CeO_2), zirconium oxide (ZrO_2) and/or silica (SiO_2) and the like.

[73] In addition, examples of the reforming catalyst may be one or more selected from platinum (Pt), rhodium (Rh), ruthenium (Ru), nickel (Ni), alkali metal oxides, alkali

earth metal oxides (for example, CaO , K_2O , or MgO) and a mixture thereof, or one or more selected from the group consisting of a composite of $\text{Cu/CeO}_2/\text{ZrO}_2$, a composite of $\text{Cu/ZnO}_2/\text{Al}_2\text{O}_3$, a composite of $\text{Cu/CeO}_2/\text{Al}_2\text{O}_3$, a composite of $\text{Cu/ZrO}_2/\text{Al}_2\text{O}_3$, or a solid-solution of copper-zinc-aluminum (CuZnAl) oxide. In addition, the reforming catalyst may also be used by supporting it on a catalyst support such as aluminum oxide, α,γ , θ -aluminum oxide, cerium oxide (CeO_2), zirconium oxide (ZrO_2) and/or silica (SiO_2) and the like.

[74]

[75] The Third Plate 500

[76] On the third plate **500** positioned below the second unit **400**, a first exhaust port **501** for discharging exothermic reaction products and a second exhaust port **502** for discharging heat-exchanged endothermic reaction raw material are formed, and a through-hole **615** for passing the heat-exchanged endothermic reaction products is formed.

[77] The through-hole **505** formed on the third plate **500** is connected to an outlet **422** formed on the plate assembly **420** for endothermic reactions constituting the upper second unit **400** and the through-hole **615** of the heat exchange plate assembly **610** for endothermic reaction raw material constituting the lower third unit **600**.

[78] In addition, the first exhaust port **501** is connected to the first inflow port **301** of the second plate **300**, and the second exhaust port **502** is connected to the second inflow port **302** of the second plate **300**.

[79]

[80] The Third Unit 600

[81] The third unit **600** is a part that endothermic reaction raw material and endothermic reaction products is heat-exchanged, and comprises the heat exchange plate assembly **610** for endothermic reaction raw material located below the third plate **500** and the heat exchange plate assembly **620** for endothermic reaction products, endothermic reaction raw material are in-flowed to the heat exchange plate assembly for endothermic reaction raw material and endothermic reaction products is in-flowed to the heat exchange plate assembly for endothermic reaction products. Here, the configurations of the heat exchange plate assembly **610** for endothermic reaction raw material and the heat exchange plate assembly **620** for endothermic reaction products, constituting the third unit **600**, are the same as those of the heat exchange plate assembly **210** for exothermic reaction raw material and the heat exchange plate assembly **220** for exothermic reaction products, constituting the first unit **200**.

[82] An inflow port **701** of the fourth plate **700** is connected to an inlet **611** formed on the heat exchange plate assembly **610** for endothermic reaction raw material, and an outlet **612** formed on the heat exchange plate assembly **610** for endothermic reaction

raw material is connected to the second exhaust port **502** formed on the third plate **500**. Therefore, endothermic reaction raw material in-flowed from the fourth plate **700** flows along fluid passages formed within the heat exchange plate assembly **610** for endothermic reaction raw material, and then are in-flowed into the second exhaust port **502** of the third plate **500**.

[83] On the other hand, an inlet **621** formed on the heat exchange plate assembly **620** for endothermic reaction products is connected to the through-hole **615** formed on the heat exchange plate assembly **610** for endothermic reaction raw material, and an outlet **222** formed on the heat exchange plate assembly **620** for endothermic reaction products is connected to an exhaust port **702** formed on the fourth plate **700**. Therefore, endothermic reaction products generated by an endothermic reaction and in-flowed from the third plate **500** flow along fluid passages formed within the heat exchange plate assembly **620** for endothermic reaction products and are then discharged via the exhaust port **702** formed on the fourth plate **700**. Accordingly, the endothermic reaction products at a high temperature of 650°C or more that passes through the heat exchange plate assembly **620** for endothermic reaction products provides heat to endothermic reaction raw material passing through the heat exchange plate assembly **610** for endothermic reaction raw material to pre-heat endothermic reaction raw material by a temperature of about 500°C or more.

[84] Of course, considering heat transfer efficiency, one or more heat exchange plate assemblies for endothermic reaction raw material and one or more heat exchange plate assemblies for endothermic reaction products may be alternately laminated in the third unit **600**.

[85]

[86] The Fourth Plate **700**

[87] The fourth plate **700** acts as a cove of the bottom of reactor. On the fourth plate **700**, an inflow port **701** for supplying endothermic reaction raw material and an exhaust port **702** for discharging endothermic reaction products are formed. Endothermic reaction raw material is introduced into the inflow port and endothermic reaction products is discharged through the exhaust port.

[88] The inflow port **701** is connected to the inlet **611** of the heat exchange plate assembly **610** for endothermic reaction raw material via a through-hole **625** of the heat exchange plate assembly **620** for endothermic reaction products of the third unit **600**, and the exhaust port **702** is connected to the outlet **622** of the heat exchange plate assembly **620** for endothermic reaction products of the third unit **600**.

[89] The plate assemblies constituting the first to third units are explained in more detail below, with reference to Figs. 3 to 6.

[90] Fig. 3 is an exploded perspective view depicting the disassembled heat exchange

plate assemblies **210** and **220** of the first unit **200**. As shown in Fig. 3, in the plate assembly according to the present invention, one or more micro fluid passages are formed by joining the first channel plate **210'** or **220'** having one or more micro channels formed on a lower surface thereof and the second channel plate **210''** or **220''** having one or more micro channels formed on an upper surface thereof and corresponding to micro channels of the first channel plate. The second channel plates **210''** and **220''** of the micro channel plates are the same as the first channel plates **210'** and **220'**, except for having a symmetrical structure in a mirror image. Below, accordingly, the same reference numerals are given to the same members, but they are distinguished and represented by mark ('').

[91] As shown in Fig. 5, a gasket **61** may be interposed between the micro channel plates **210'** and **220'**, **210''** and **220''**. The gasket **61** may preferably be made of a copper plate. Openings **62** and through-holes **63** are formed on the gasket for forming certain space between the micro channel plates **210'** and **220'**, **210''** and **220''**.

[92] As shown in Figs. 3 and 4, concave portions with a certain depth are formed at insides of the micro channel plates **210'**, **220'**, **210''** and **220''**. Inlets **211'** and **211''** and outlets **212'** and **212''** diagonally positioned against each other are formed in the concave portion, and the through-holes **214'** and **215'** diagonally positioned against each other are also formed on a periphery of the outer concave portions. In addition, a wall **53'** of concave portion serves as a barrier for controlling flow direction of raw material or reaction products passing through micro channels. The micro channels **51'** in the inside of concave portion are divided by one or more partitions **52'** and **52** to form plural channels.

[93] As shown in Fig. 6, a number of spaces **71** [micro fluid passages] are formed in the inside of each plate assembly for forming channels. In addition, in the plate assemblies for endothermic reaction and exothermic reaction, coating parts may be formed by applying catalyst for exothermic reaction or catalyst for endothermic reaction on the inside walls.

[94] In addition, as shown in Fig. 4, a number of converging grooves **54'** and converging projections **55'** are formed near the inlet **211'** and the outlet **212'** for restricting flow of raw material or reaction products so as to diffuse or collect raw material or reaction products from the inlet **211'** or the outlet **212'** to plural micro channels **51'**.

[95]

Industrial Applicability

[96] The above chemical reactor including heat exchangers, according to the present invention, minimizes an exposure of flows of exothermic reaction raw material and endothermic reaction raw material to an outside to reduce the volume of reactor and si-

multaneously minimize heat loss.

- [97] In addition, the heat efficiency is improved by placing the heat exchangers on both wide surfaces of the chemical reactor and using heat transferred via the surface of reactor in the heat exchangers. Exothermic reaction raw material and endothermic reaction raw material are pre-heated by heat-exchanging exothermic reaction products and endothermic reaction products with exothermic reaction raw material and endothermic reaction raw material and then introduced into the plate assemblies for exothermic reaction and endothermic reactions, and so the exothermic reactions and the endothermic reactions may be effectively carried out. Especially, the heat exchange plate assemblies and the plate assemblies for reactions are disposed in a contact state so that the heat exchange can be effectively occurred.
- [98] Moreover, there is the advantage that the volume of the reactor can be remarkably reduce by disposing in order the first plate, the first unit, the second plate, the second unit, the third plate, the third unit and the fourth plate in stack state.

Claims

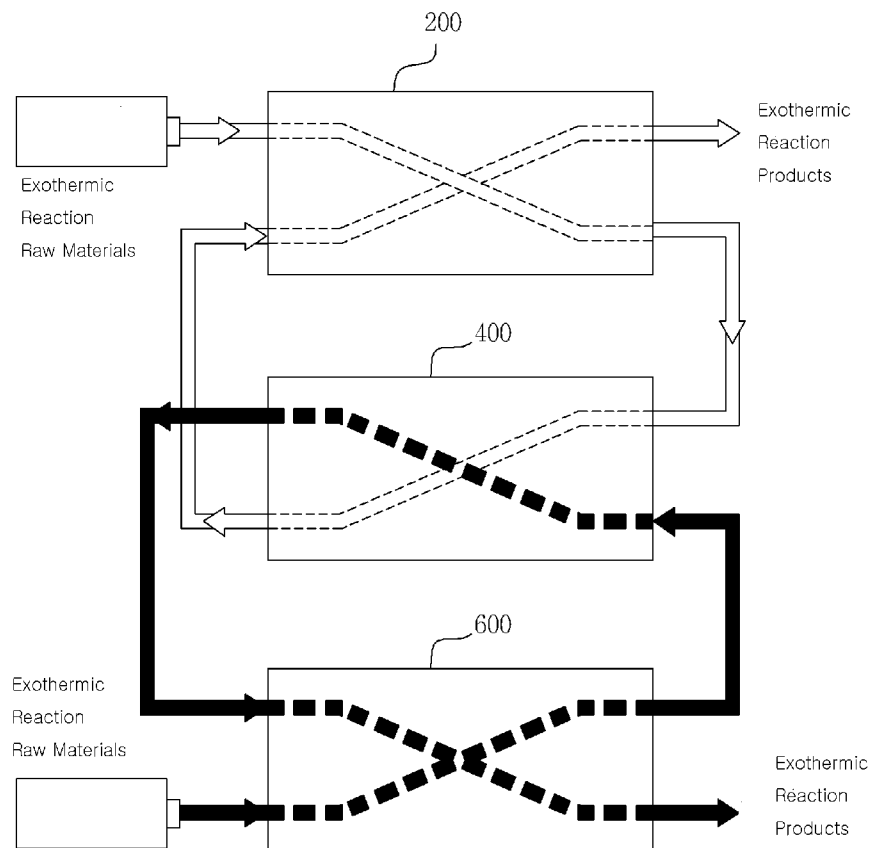
- [1] A chemical reactor, comprising;
a first unit including one or more heat exchange plate assemblies for exothermic reaction raw material and one or more heat exchange plate assemblies for exothermic reaction products, a heat exchange between exothermic reaction products and exothermic reaction raw material being performed in the first unit;
a second unit including one or more plate assemblies for exothermic reaction and one or more plate assemblies for endothermic reaction, the second unit providing heat generated through an exothermic reaction of said exothermic reaction raw material to endothermic reaction raw material to proceed the endothermic reaction; and
a third unit including one or more heat exchange plate assemblies for endothermic reaction raw material and one or more heat exchange plate assemblies for endothermic reaction products, a heat exchange between endothermic reaction products and endothermic reaction raw material being performed in the third unit.
- [2] The chemical reactor of claim 1, wherein the exothermic reactions are combustion reactions of combustion gases, and the endothermic reactions are reforming reactions for reforming hydrocarbon raw material to produce hydrogen.
- [3] The chemical reactor of claim 1, further comprising
a first plate for supplying exothermic reaction raw material to the heat exchange plate assembly for exothermic reaction raw material and discharging exothermic reaction products introduced from the heat exchange plate assembly for exothermic reaction products;
a second plate for supplying exothermic reaction raw material introduced from the heat exchange plate assembly for exothermic reaction raw material to the plate assembly for exothermic reaction, and discharging endothermic reaction raw material and exothermic reaction products introduced from a third plate to the plate assembly for endothermic reaction and the heat exchange plate assembly for exothermic reaction products, respectively;
a third plate for supplying endothermic reaction raw material introduced from the heat exchange plate assembly for exothermic reaction raw material and exothermic reaction products introduced from the plate assemblies for exothermic reaction to the second plate, and supplying endothermic reaction products introduced from the plate assembly for endothermic reaction to the heat exchange plate assembly for endothermic reaction products; and

a fourth plate for supplying endothermic reaction raw material to the heat exchange plate assembly for endothermic reaction raw material, and discharging endothermic reaction products introduced from the heat exchange plate assembly endothermic reaction products.

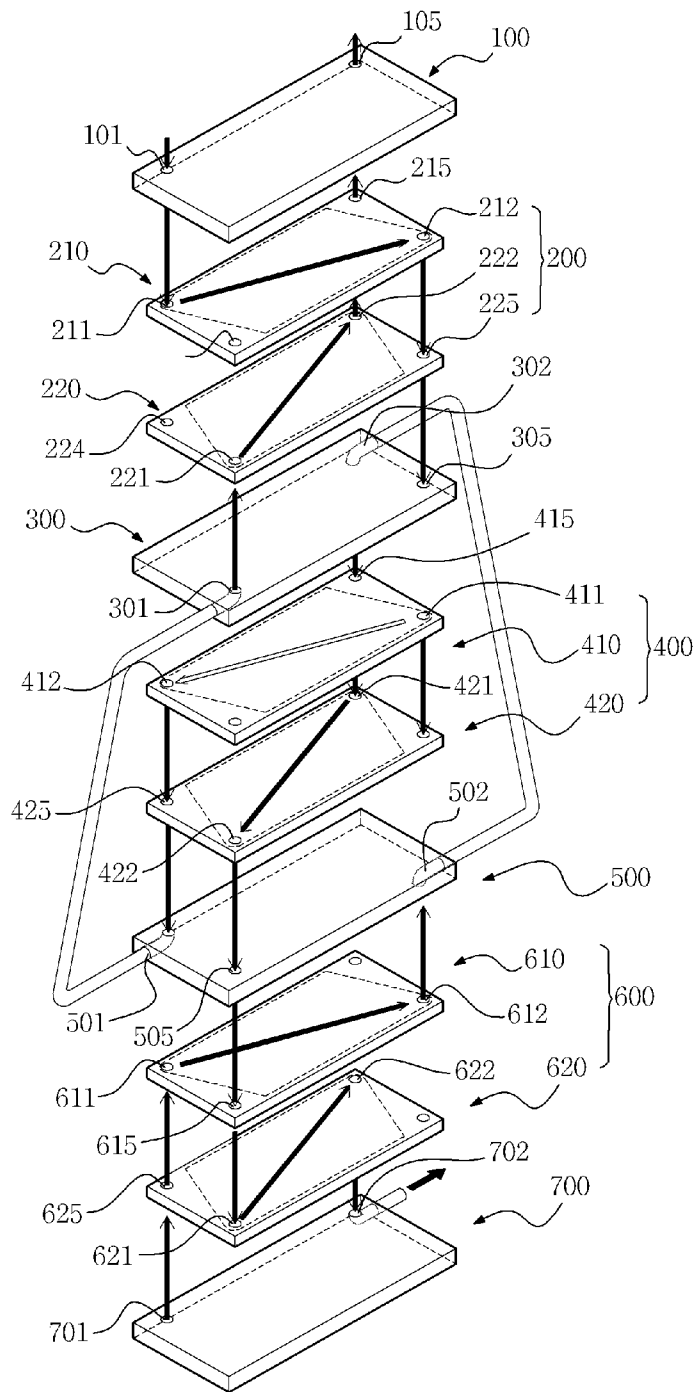
- [4] The chemical reactor of claim 3, wherein the first plate, the first unit, the second plate, the second unit, the third plate, the third unit and the fourth plate are laminated in order.
- [5] The chemical reactor of claim 1, wherein the first unit is formed such that the heat exchange plate assemblies for exothermic reaction raw material and the heat exchange plate assemblies for exothermic reaction products are alternatively laminated .
- [6] The chemical reactor of claim 1, wherein the second unit is formed such that the plate assemblies for exothermic reaction and the plate assemblies for endothermic reaction are alternatively laminated.
- [7] The chemical reactor of claim 1, wherein the third unit is formed such that the heat exchange plate assemblies for endothermic reaction raw material and the heat exchange plate assemblies for endothermic reaction products are alternately laminated.
- [8] The chemical reactor of claim 1, wherein the plate assembly comprises a first channel plate and a second channel plate joined each other, and one or more micro fluid passages are formed therein.
- [9] The chemical reactor of claim 8, wherein, the plate assembly has an inlet and outlet communicated in fluid to an inside thereof and through-holes perforating thereof.
- [10] The chemical reactor of claim 9, wherein, the plate assembly has control walls formed therein and allowing to diffuse in-flowed raw material or reaction products from the inlets or converge it to the outlets.
- [11] The chemical reactor of claim 1, wherein the plate assembly for exothermic reaction comprises a catalyst for exothermic reaction in which active ingredient is supported on one or more supports selected from the group consisting of aluminum oxide, α , γ , θ -aluminum oxide, CeO_2 , ZrO_2 and SiO_2 , and the active ingredient is one or more selected from the group consisting of Pt, Rh, Ru, Os, Ir, Pd, Au, Ag and Cu.
- [12] The chemical reactor of claim 1, wherein the plate assembly for endothermic reaction comprises a catalyst for endothermic reaction in which active ingredient is supported on one or more supports selected from the group consisting of aluminum oxide, α , γ , θ -aluminum oxide, CeO_2 , ZrO_2 and SiO_2 , and

the active ingredient is one or more selected from the group consisting of Pt, Rh, Ru, Ni, alkali metal oxides, and alkali earth metal oxides, or one or more selected from the group consisting of a composite of $\text{Cu/CeO}_2/\text{ZrO}_2$, a composite of $\text{Cu/ZnO}/\text{Al}_2\text{O}_3$, a composite of $\text{Cu/CeO}_2/\text{Al}_2\text{O}_3$, a composite of $\text{Cu/ZrO}_2/\text{Al}_2\text{O}_3$, and a solid-solution of Cu-Zn-Al oxide.

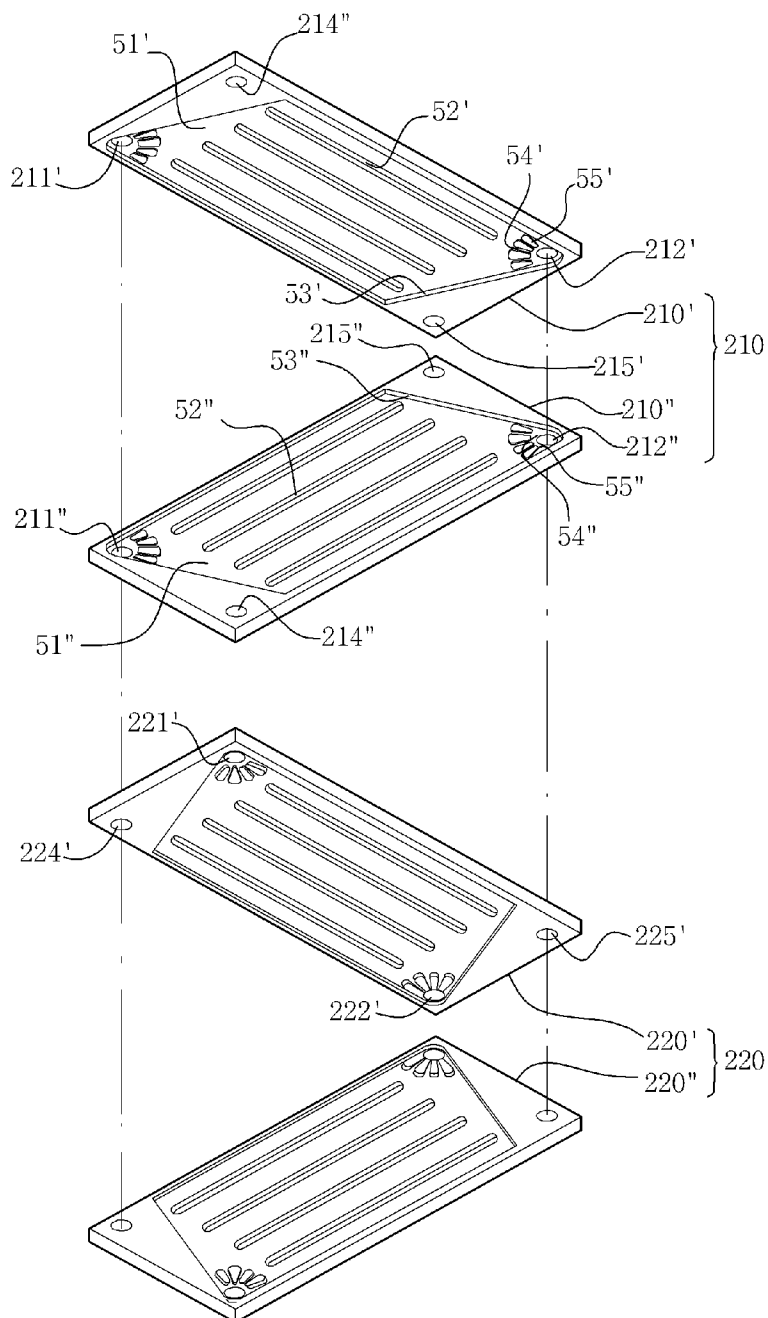
[Fig. 1]



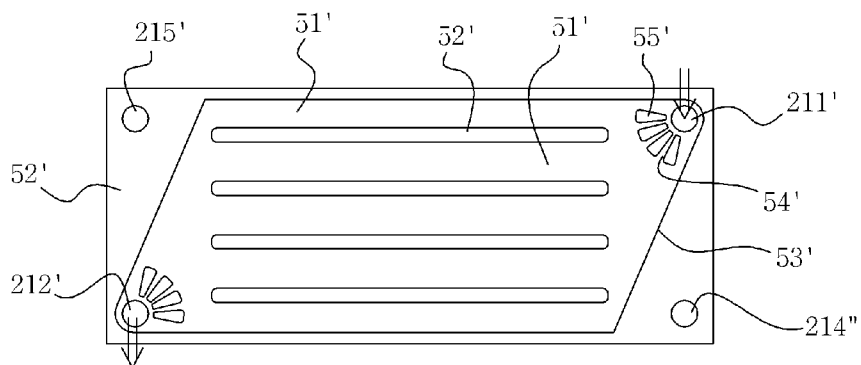
[Fig. 2]



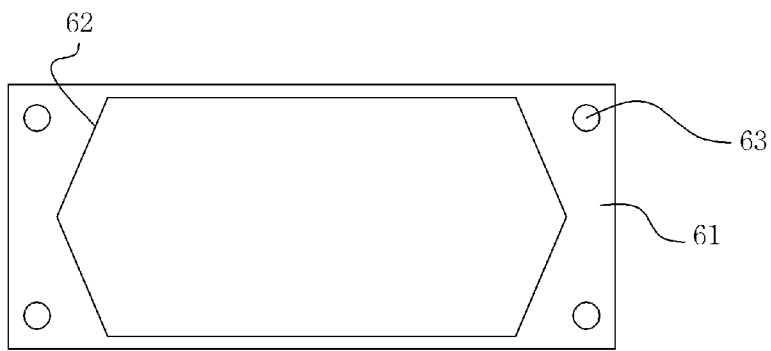
[Fig. 3]



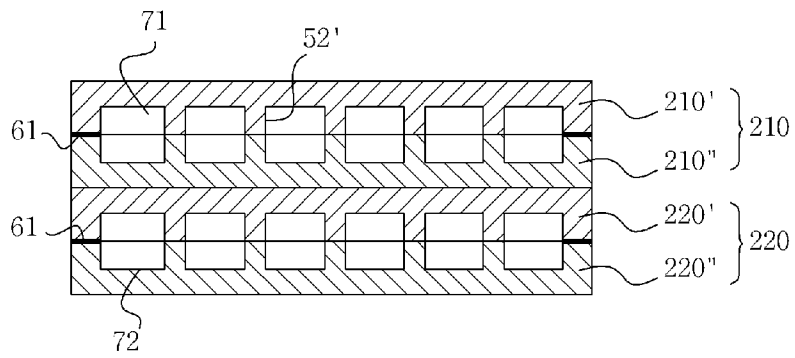
[Fig. 4]



[Fig. 5]



[Fig. 6]



INTERNATIONAL SEARCH REPORT

International application No.
PCT/KR2007/003483**A. CLASSIFICATION OF SUBJECT MATTER****C01B 3/38(2006.01)i**

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC 8 C01B 3/38

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Korean Utility models and applications for Utility models since 1975

Japanese Utility models and applications for Utility models since 1975

Electronic data base consulted during the international search (name of data base and, where practicable, search terms used)

eKIPASS (KIPO internal) " channel reactor, reform, hydrocarbon"

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	US 2002/0071797 A1 (DANIEL G. LOFFLER. et al.) 13 JUNE 2002 see the abstract; claims 1-9, 15-21; paragraphs [0011]-[0013]; figures 1-7	1, 2, 8-10
A	US 2005/0193628 A1 (JU-YONG KIM. et al.) 8 SEPTEMBER 2005 see claims 1-4, 9; paragraphs [0006]-[0008]; figures 2-7	1, 2, 4, 9, 10
A	US 7 048 897 B1 (CHOWDARY RAMESH KORIPPELLA. et al.) 23 MAY 2006 see claims 1-4; figures 1-4, 7 -----	1, 2

☐ Further documents are listed in the continuation of Box C.☒ See patent family annex.

* Special categories of cited documents:

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Date of the actual completion of the international search

29 OCTOBER 2007 (29.10.2007)

Date of mailing of the international search report

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INTERNATIONAL SEARCH REPORT

Information on patent family members

International application No.

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