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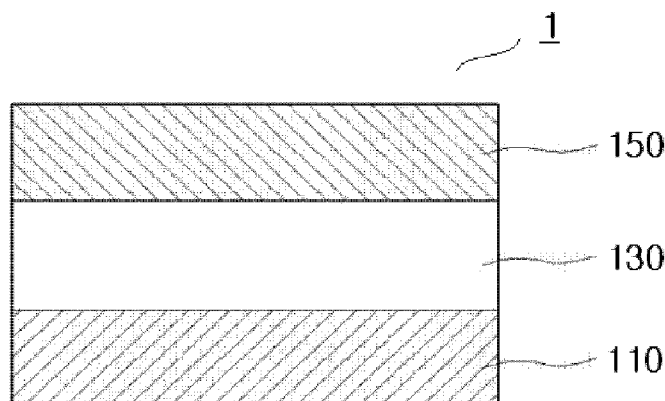
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(54) Title: COMPOUND FOR ORGANIC ELECTROLUMINESCENT DEVICE AND ORGANIC ELECTROLUMINESCENT DEVICE INCLUDING THE SAME



(57) Abstract: This invention relates to a compound for an organic electroluminescent device and to an organic electroluminescent device including the same. This compound for an organic electroluminescent device has high thermal stability, high triplet energy and hole transport capability, and the organic electroluminescent device including the same is improved in thermal stability and light emission efficiency. When this compound is used as a hole transport layer material, a triplet exciton of a phosphorescent light emitting material is confined, thus improving efficiency of the organic electroluminescent device.



Description

Title of Invention: COMPOUND FOR ORGANIC ELECTROLUMINESCENT DEVICE AND ORGANIC ELECTROLUMINESCENT DEVICE INCLUDING THE SAME

Technical Field

- [1] The present invention relates to a compound for an organic electroluminescent device and an organic electroluminescent device including the same, and more particularly, to an amine-based compound for an organic electroluminescent device, having a heterocyclic structure, and to an organic electroluminescent device including the same.

Background Art

- [2] Organic electroluminescent (EL) devices have a simpler structure, various processing advantages, higher brightness, superior viewing angle properties, quicker response rate, and a lower driving voltage compared to other flat panel displays such as liquid crystal displays (LCDs), plasma display panels (PDPs), field emission displays (FEDs), etc., and are thus being thoroughly developed so as to be utilized as light sources of flat panel displays such as wall-mountable TVs, etc. or backlight units of the displays, illuminators, advertisement boards and so on.
- [3] Typically, when a direct-current voltage is applied to an organic EL device, holes injected from an anode and electrons injected from a cathode recombine to form electron-hole pairs, namely, excitons. While the excitons return to a stable ground state, energy corresponding thereto is transferred to a light emitting material and is thereby converted into light.
- [4] In order to increase efficiency and stability of an organic EL device, since C. W. Tang et al. of Eastman Kodak Company made an organic EL device operating at low voltage by forming a tandem organic thin film between two opposite electrodes (C. W. Tang, S. A. Vanslyke, Applied Physics Letters, vol. 51, pp. 913, 1987), extensive and intensive research into organic materials for organic EL devices having a multilayered thin-film structure has been ongoing. The efficiency and lifetime of such a tandem organic EL device are closely related to the molecular structure of a material for the thin film. For example, quantum efficiency may greatly vary depending on the structure of the material for the thin film, particularly a host material, a hole transport layer material or an electron transport layer material. When thermal stability of the material decreases, the material may be crystallized at a high temperature or a driving temperature, undesirably shortening the lifetime of the device.
- [5] Hole transport materials for use in organic EL devices, which have been known to date, are problematic because thin films formed therefrom using vacuum deposition are

thermally and electrically unstable, and thus may rapidly crystallize due to heat generated upon device driving and also the film materials may change, undesirably deteriorating the light emission efficiency of the devices. Further, non-emission parts referred to as dark spots may increasingly occur, and the voltage may increase upon constant-current driving, undesirably damaging the devices.

- [6] Also, organic EL devices using a phosphorescent light emitting material do not confine a triplet exciton produced in the light emitting material of a light emitting layer due to low triplet energy, undesirably lowering the light emission efficiency of the devices.

Disclosure of Invention

Technical Problem

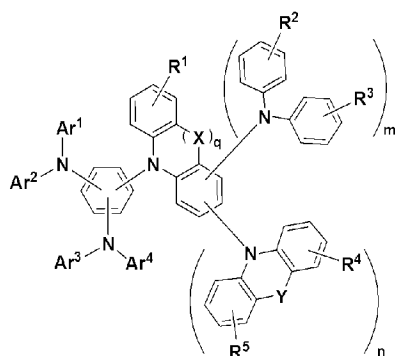
- [7] Accordingly, an object of the present invention is to provide a compound for an organic EL device, which may have high thermal stability, high triplet energy and hole transport capability.
- [8] Another object of the present invention is to provide an organic EL device, which includes the compound as above and is thus improved in thermal stability and light emission efficiency, and in which the above compound is used as a hole transport layer material, thereby confining a triplet exciton of a phosphorescent light emitting material, ultimately improving efficiency of the organic EL device.

Solution to Problem

- [9] In order to accomplish the above objects, an aspect of the present invention provides a compound for an organic EL device, as represented by Chemical Formula 1 below.

[10] [Chemical Formula 1]

[11]



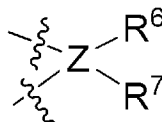
- [12] In Chemical Formula 1, Ar¹ to Ar⁴ are identical to or different from each other, and Ar¹ to Ar⁴ are each independently a substituted or unsubstituted C1 to C30 alkyl group, a substituted or unsubstituted C3 to C30 cycloalkyl group, a substituted or unsubstituted C1 to C30 heterocycloalkyl group, a substituted or unsubstituted C6 to C30 aryl group, or a substituted or unsubstituted C1 to C30 heteroaryl group, or Ar¹ and Ar², and Ar³ and Ar⁴, respectively, are linked to form a substituted or unsubstituted C1 to

C30 heterocycloalkyl group, or a substituted or unsubstituted C1 to C30 heteroaryl group, together with a nitrogen atom therebetween,

[13] R^1 is a hydrogen atom, a substituted or unsubstituted C1 to C30 alkyl group, a substituted or unsubstituted C3 to C30 cycloalkyl group, a substituted or unsubstituted C1 to C30 heterocycloalkyl group, a substituted or unsubstituted C6 to C30 aryl group, or a substituted or unsubstituted C1 to C30 heteroaryl group,

[14] R^2 to R^5 are identical to or different from each other, and R^2 to R^5 are each independently a hydrogen atom, a substituted or unsubstituted C1 to C30 alkyl group, a substituted or unsubstituted C3 to C30 cycloalkyl group, a substituted or unsubstituted C1 to C30 heterocycloalkyl group, a substituted or unsubstituted C6 to C30 aryl group, or a substituted or unsubstituted C1 to C30 heteroaryl group, or at least one of R^2 to R^5 is further coupled with a carbon atom adjacent to a carbon atom linked therewith to form a substituted or unsubstituted fused C3 to C30 cycloalkyl group, a substituted or unsubstituted fused C1 to C30 heterocycloalkyl group, a substituted or unsubstituted fused C6 to C30 aryl group, or a substituted or unsubstituted fused C1 to C30 heteroaryl group,

[15] X and Y are identical to or different from each other, and X and Y are each independently a valence bond or



atom, R^6 and R^7 are identical to or different from each other, and R^6 and R^7 are each independently a hydrogen atom, a substituted or unsubstituted C1 to C30 alkyl group, a substituted or unsubstituted C3 to C30 cycloalkyl group, a substituted or unsubstituted C1 to C30 heterocycloalkyl group, a substituted or unsubstituted C6 to C30 aryl group, or a substituted or unsubstituted C1 to C30 heteroaryl group,

[16] q is 0 or 1,

[17] m is 0 or 1,

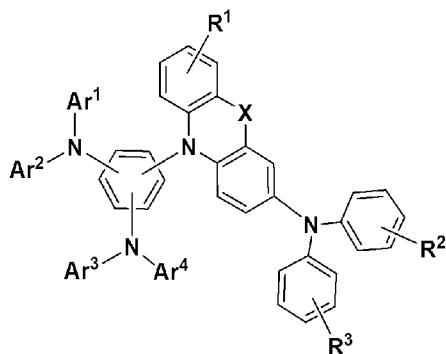
[18] n is 0 or 1, and

[19] $m+n \neq 0$.

[20] According to a preferred embodiment of the present invention, the compound for an organic EL device is represented by either of Chemical Formulas 2 and 3 below.

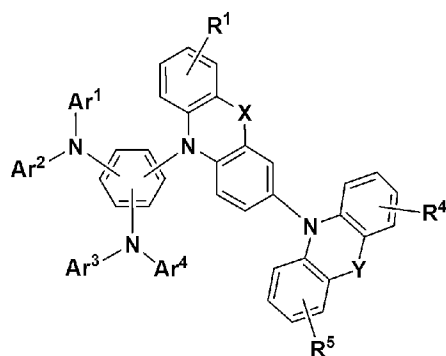
[21] [Chemical Formula 2]

[22]



[23] [Chemical Formula 3]

[24]



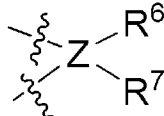
[25] In Chemical Formulas 2 and 3, Ar¹ to Ar⁴ are identical to or different from each other, and Ar¹ to Ar⁴ are each independently a substituted or unsubstituted C1 to C30 alkyl group, a substituted or unsubstituted C3 to C30 cycloalkyl group, a substituted or unsubstituted C1 to C30 heterocycloalkyl group, a substituted or unsubstituted C6 to C30 aryl group, or a substituted or unsubstituted C1 to C30 heteroaryl group, or Ar¹ and Ar², and Ar³ and Ar⁴, respectively, are linked to form a substituted or unsubstituted C1 to C30 heterocycloalkyl group, or a substituted or unsubstituted C1 to C30 heteroaryl group, together with a nitrogen atom therebetween,

[26] R¹ is a hydrogen atom, a substituted or unsubstituted C1 to C30 alkyl group, a substituted or unsubstituted C3 to C30 cycloalkyl group, a substituted or unsubstituted C1 to C30 heterocycloalkyl group, a substituted or unsubstituted C6 to C30 aryl group, or a substituted or unsubstituted C1 to C30 heteroaryl group,

[27] R² to R⁵ are identical to or different from each other, and R² to R⁵ are each independently a hydrogen atom, a substituted or unsubstituted C1 to C30 alkyl group, a substituted or unsubstituted C3 to C30 cycloalkyl group, a substituted or unsubstituted C1 to C30 heterocycloalkyl group, a substituted or unsubstituted C6 to C30 aryl group, or a substituted or unsubstituted C1 to C30 heteroaryl group, or at least one of R² to R⁵ is further coupled with a carbon atom adjacent to a carbon atom linked therewith to form a substituted or unsubstituted fused C3 to C30 cycloalkyl group, a substituted or unsubstituted fused C1 to C30 heterocycloalkyl group, a substituted or unsubstituted fused C6 to C30 aryl group, or a substituted or unsubstituted fused C1 to C30

heteroaryl group, and

- [28] X and Y are identical to or different from each other, and X and Y are each independently a valence bond or

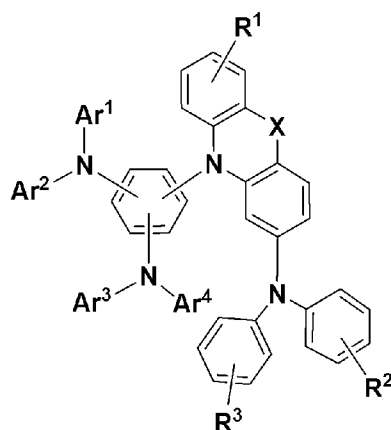


atom, and R^6 and R^7 are identical to or different from each other, and R^6 and R^7 are each independently a hydrogen atom, a substituted or unsubstituted C1 to C30 alkyl group, a substituted or unsubstituted C3 to C30 cycloalkyl group, a substituted or unsubstituted C1 to C30 heterocycloalkyl group, a substituted or unsubstituted C6 to C30 aryl group, or a substituted or unsubstituted C1 to C30 heteroaryl group.

- [29] According to a preferred embodiment of the present invention, the compound for an organic EL device is represented by either of Chemical Formulas 4 and 5 below.

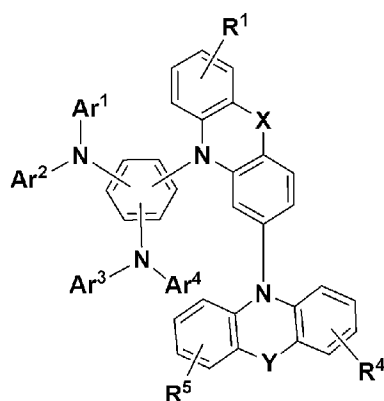
- [30] [Chemical Formula 4]

- [31]



- [32] [Chemical Formula 5]

- [33]



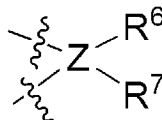
- [34] In Chemical Formulas 4 and 5, Ar^1 to Ar^4 are identical to or different from each other, and Ar^1 to Ar^4 are each independently a substituted or unsubstituted C1 to C30 alkyl group, a substituted or unsubstituted C3 to C30 cycloalkyl group, a substituted or unsubstituted C1 to C30 heterocycloalkyl group, a substituted or unsubstituted C6 to C30 aryl group, or a substituted or unsubstituted C1 to C30 heteroaryl group, or Ar^1

and Ar², and Ar³ and Ar⁴, respectively, are linked to form a substituted or unsubstituted C1 to C30 heterocycloalkyl group, or a substituted or unsubstituted C1 to C30 heteroaryl group, together with a nitrogen atom therebetween,

[35] R¹ is a hydrogen atom, a substituted or unsubstituted C1 to C30 alkyl group, a substituted or unsubstituted C3 to C30 cycloalkyl group, a substituted or unsubstituted C1 to C30 heterocycloalkyl group, a substituted or unsubstituted C6 to C30 aryl group, or a substituted or unsubstituted C1 to C30 heteroaryl group,

[36] R² to R⁵ are identical to or different from each other, and R² to R⁵ are each independently a hydrogen atom, a substituted or unsubstituted C1 to C30 alkyl group, a substituted or unsubstituted C3 to C30 cycloalkyl group, a substituted or unsubstituted C1 to C30 heterocycloalkyl group, a substituted or unsubstituted C6 to C30 aryl group, or a substituted or unsubstituted C1 to C30 heteroaryl group, or at least one of R² to R⁵ is further coupled with a carbon atom adjacent to a carbon atom linked therewith to form a substituted or unsubstituted fused C3 to C30 cycloalkyl group, a substituted or unsubstituted fused C1 to C30 heterocycloalkyl group, a substituted or unsubstituted fused C6 to C30 aryl group, or a substituted or unsubstituted fused C1 to C30 heteroaryl group, and

[37] X and Y are identical to or different from each other, and X and Y are each independently a valence bond or

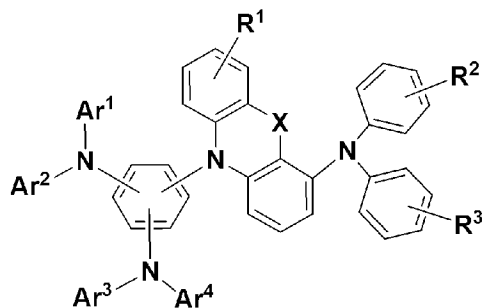


atom, and R⁶ and R⁷ are identical to or different from each other, and R⁶ and R⁷ are each independently a hydrogen atom, a substituted or unsubstituted C1 to C30 alkyl group, a substituted or unsubstituted C3 to C30 cycloalkyl group, a substituted or unsubstituted C1 to C30 heterocycloalkyl group, a substituted or unsubstituted C6 to C30 aryl group, or a substituted or unsubstituted C1 to C30 heteroaryl group.

[38] According to a preferred embodiment of the present invention, the compound for an organic EL device is represented by either of Chemical Formulas 6 and 7 below.

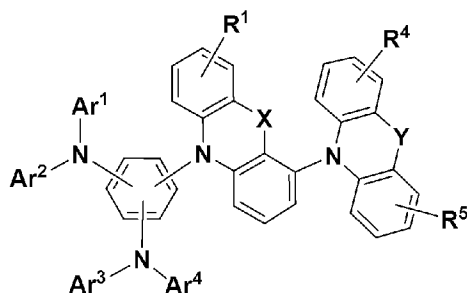
[39] [Chemical Formula 6]

[40]

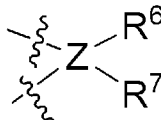


[41] [Chemical Formula 7]

[42]

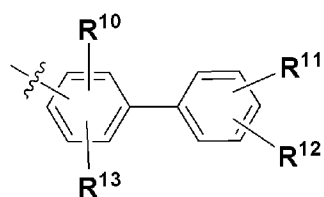
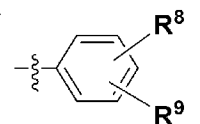


- [43] In Chemical Formulas 6 and 7, Ar¹ to Ar⁴ are identical to or different from each other, and Ar¹ to Ar⁴ are each independently a substituted or unsubstituted C1 to C30 alkyl group, a substituted or unsubstituted C3 to C30 cycloalkyl group, a substituted or unsubstituted C1 to C30 heterocycloalkyl group, a substituted or unsubstituted C6 to C30 aryl group, or a substituted or unsubstituted C1 to C30 heteroaryl group, or Ar¹ and Ar², and Ar³ and Ar⁴, respectively, are linked to form a substituted or unsubstituted C1 to C30 heterocycloalkyl group, or a substituted or unsubstituted C1 to C30 heteroaryl group, together with a nitrogen atom therebetween,
- [44] R¹ is a hydrogen atom, a substituted or unsubstituted C1 to C30 alkyl group, a substituted or unsubstituted C3 to C30 cycloalkyl group, a substituted or unsubstituted C1 to C30 heterocycloalkyl group, a substituted or unsubstituted C6 to C30 aryl group, or a substituted or unsubstituted C1 to C30 heteroaryl group,
- [45] R² to R⁵ are identical to or different from each other, and R² to R⁵ are each independently a hydrogen atom, a substituted or unsubstituted C1 to C30 alkyl group, a substituted or unsubstituted C3 to C30 cycloalkyl group, a substituted or unsubstituted C1 to C30 heterocycloalkyl group, a substituted or unsubstituted C6 to C30 aryl group, or a substituted or unsubstituted C1 to C30 heteroaryl group, or at least one of R² to R⁵ is further coupled with a carbon atom adjacent to a carbon atom linked therewith to form a substituted or unsubstituted fused C3 to C30 cycloalkyl group, a substituted or unsubstituted fused C1 to C30 heterocycloalkyl group, a substituted or unsubstituted fused C6 to C30 aryl group, or a substituted or unsubstituted fused C1 to C30 heteroaryl group, and
- [46] X and Y are identical to or different from each other, and X and Y are each independently a valence bond or



atom, and R⁶ and R⁷ are identical to or different from each other, and R⁶ and R⁷ are each independently a hydrogen atom, a substituted or unsubstituted C1 to C30 alkyl group, a substituted or unsubstituted C3 to C30 cycloalkyl group, a substituted or unsubstituted C1 to C30 heterocycloalkyl group, a substituted or unsubstituted C6 to C30 aryl group, or a substituted or unsubstituted C1 to C30 heteroaryl group.

- [47] In a preferred embodiment of the present invention, Ar¹ to Ar⁴ are identical to or different from each other, and Ar¹ to Ar⁴ are each independently



, a substituted or unsubstituted C1 to C30 alkyl group, a sub-

stituted or unsubstituted C3 to C30 cycloalkyl group, or a substituted or unsubstituted C1 to C30 heterocycloalkyl group, or Ar¹ and Ar², and Ar³ and Ar⁴, respectively, are linked to form a substituted or unsubstituted C1 to C30 heterocycloalkyl group, or a substituted or unsubstituted C1 to C30 heteroaryl group, together with a nitrogen atom therebetween, wherein R⁸ to R¹³ are identical to or different from each other, and R⁸ to R¹³ are each independently a hydrogen atom, a substituted or unsubstituted C1 to C30 alkyl group, a substituted or unsubstituted C3 to C30 cycloalkyl group, a substituted or unsubstituted C1 to C30 heterocycloalkyl group, a substituted or unsubstituted C6 to C30 aryl group, or a substituted or unsubstituted C1 to C30 heteroaryl group.

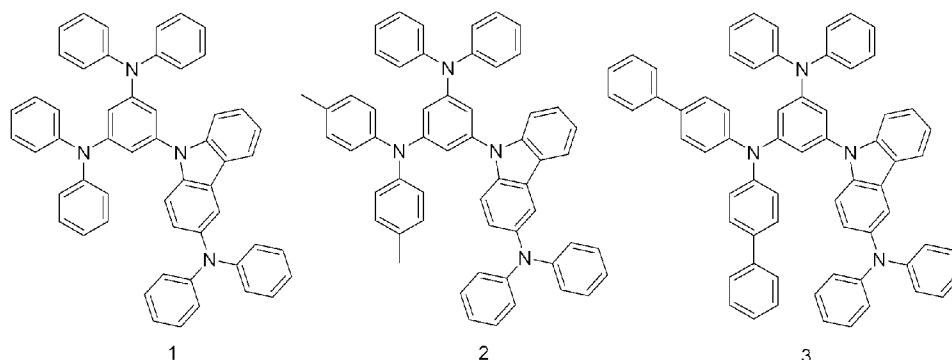
- [48] Examples of the substituted or unsubstituted C6 to C30 aryl group may include a substituted or unsubstituted phenyl group, a substituted or unsubstituted biphenyl group, a substituted or unsubstituted terphenyl group, a substituted or unsubstituted naphthalenyl group, a substituted or unsubstituted anthracenyl group, a substituted or unsubstituted phenanthrenyl group, a substituted or unsubstituted fluorenyl group, a substituted or unsubstituted spirofluorenyl group, a substituted or unsubstituted pyrenyl group, or a substituted or unsubstituted perylenyl group.

- [49] Examples of the substituted or unsubstituted C1 to C30 heteroaryl group may include a substituted or unsubstituted pyridinyl group, a substituted or unsubstituted pyrimidinyl group, a substituted or unsubstituted triazinyl group, a substituted or unsubstituted thiophenyl group, a substituted or unsubstituted pyrrolyl group, a substituted or unsubstituted benzothiophenyl group, a substituted or unsubstituted indolyl group, a substituted or unsubstituted imidazo[1,2-a]pyridinyl group, a substituted or unsubstituted benzimidazolyl group, a substituted or unsubstituted indazolyl group, a substituted or unsubstituted phenothiazinyl group, a substituted or unsubstituted phenazinyl group, a substituted or unsubstituted carbazolyl group, a substituted or unsubstituted dibenzothiophenyl group, a substituted or unsubstituted imidazolyl group, a substituted or unsubstituted triazolyl group, a substituted or unsubstituted tetrazolyl group, a substituted or unsubstituted oxadiazolyl group, a substituted or unsubstituted oxatriazolyl group, a substituted or unsubstituted thiatriazolyl group, a substituted or

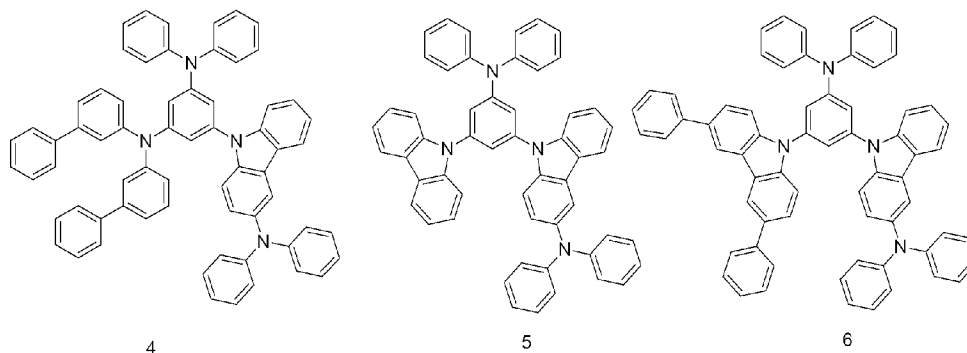
unsubstituted benzotriazolyl group, a substituted or unsubstituted pyrazinyl group, a substituted or unsubstituted pyridazinyl group, a substituted or unsubstituted purinyl group, a substituted or unsubstituted quinolinyl group, a substituted or unsubstituted isoquinolinyl group, a substituted or unsubstituted phthalazinyl group, a substituted or unsubstituted naphpyridinyl group, a substituted or unsubstituted quinoxalinyl group, a substituted or unsubstituted quinazolinyl group, a substituted or unsubstituted acridinyl group, or a substituted or unsubstituted phenanthrolinyl group. Preferably useful is a substituted or unsubstituted pyridinyl group, a substituted or unsubstituted pyrimidinyl group, a substituted or unsubstituted triazinyl group, a substituted or unsubstituted thiophenyl group, a substituted or unsubstituted pyrrolyl group, a substituted or unsubstituted benzothiophenyl group, a substituted or unsubstituted indolyl group, a substituted or unsubstituted imidazo[1,2-a]pyridinyl group, a substituted or unsubstituted benzimidazolyl group, a substituted or unsubstituted indazolyl group, a substituted or unsubstituted phenothiazinyl group, a substituted or unsubstituted phenazinyl group, a substituted or unsubstituted carbazolyl group, or a substituted or unsubstituted dibenzothiophenyl group.

[50] According to a preferred embodiment of the present invention, the compound for an organic EL device is any one elected from among Compounds 1 to 65 represented by the following chemical formulas.

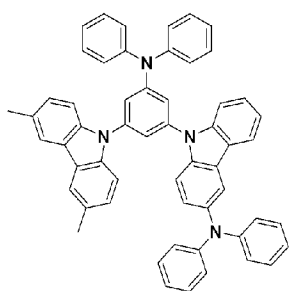
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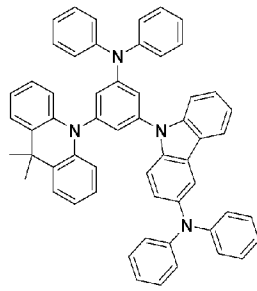
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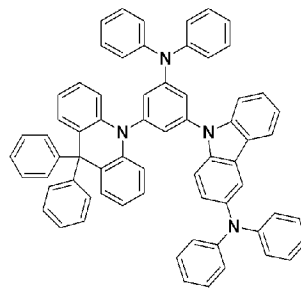
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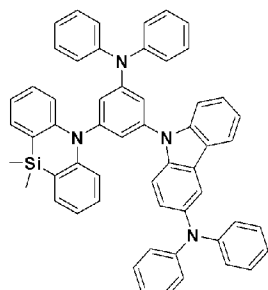


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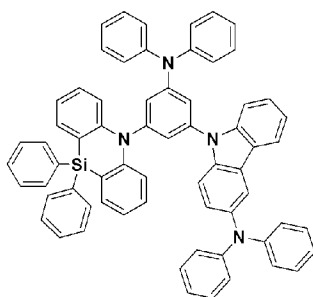


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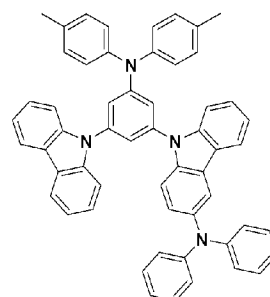
[54]



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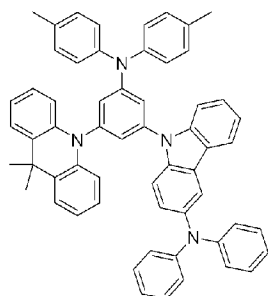


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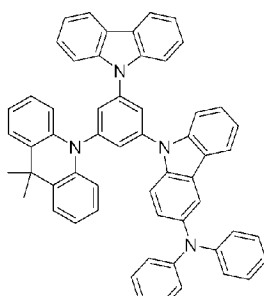


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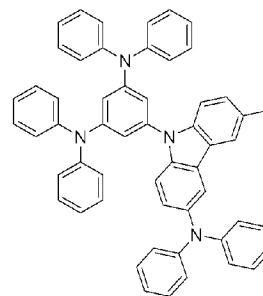
[55]



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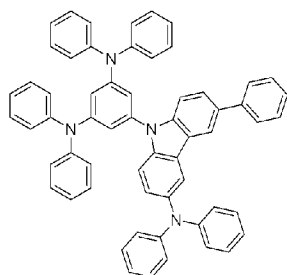


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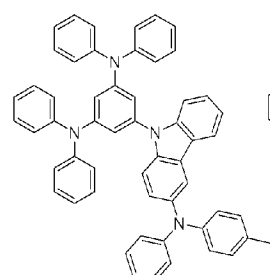


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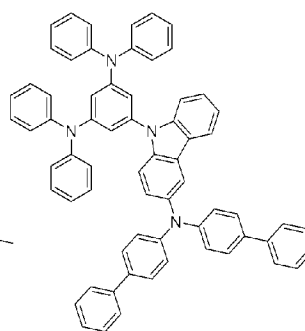
[56]



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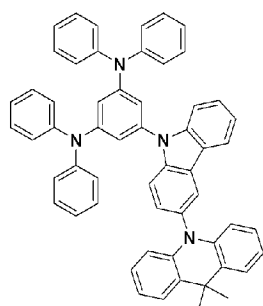


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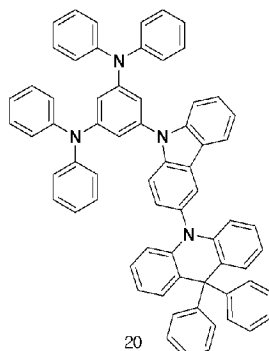


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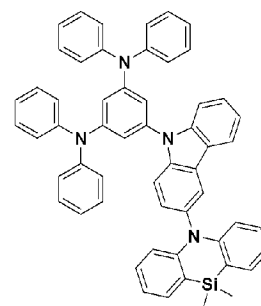
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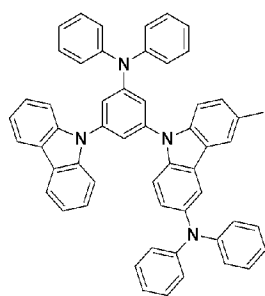


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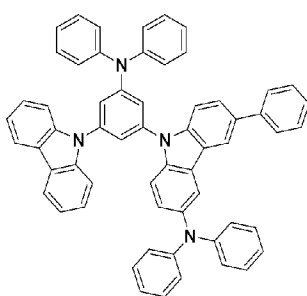


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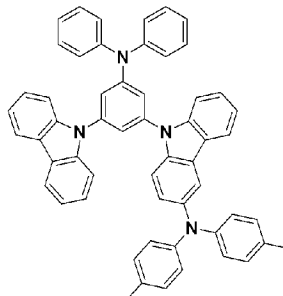
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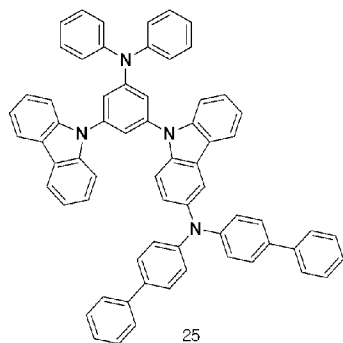


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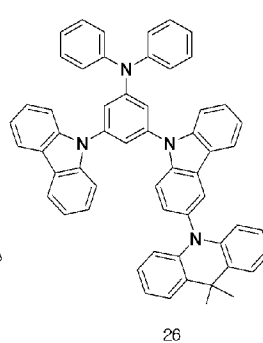


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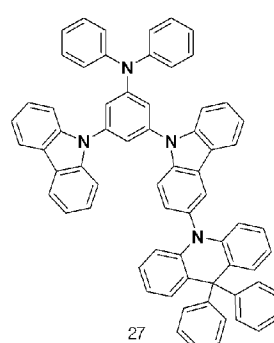
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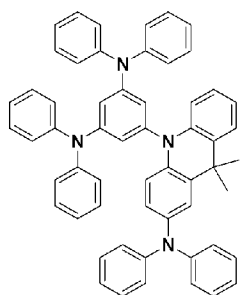


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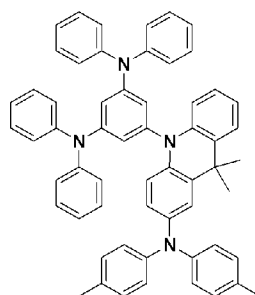


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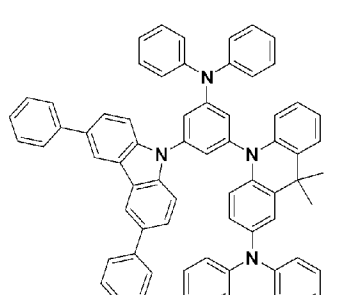
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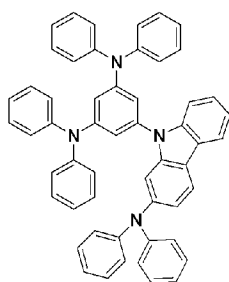


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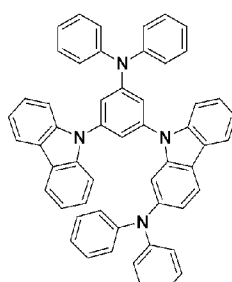


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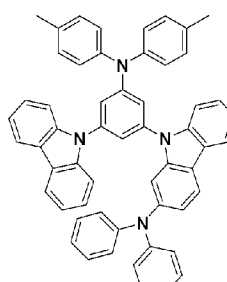
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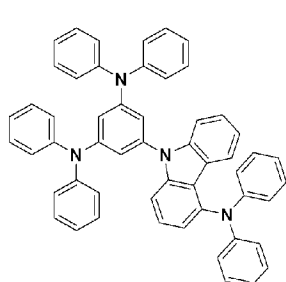


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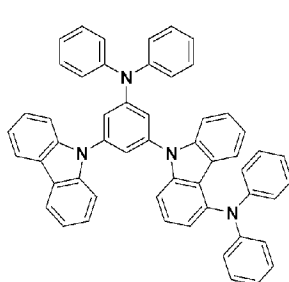


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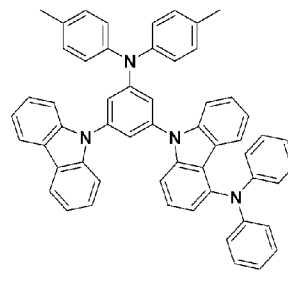
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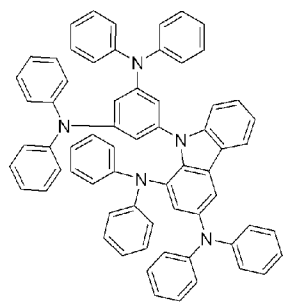


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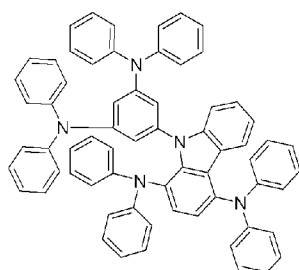


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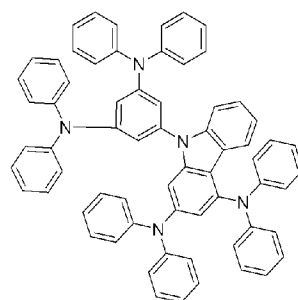
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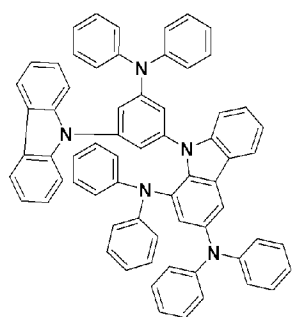


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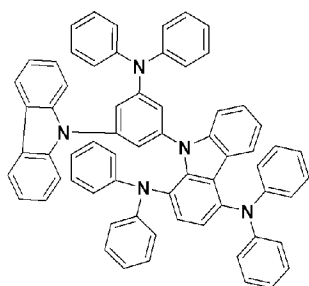


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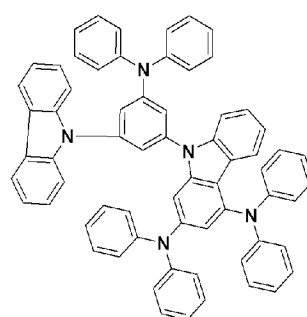
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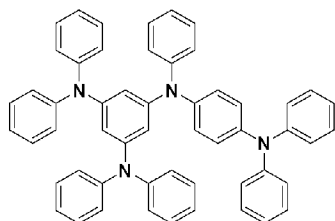


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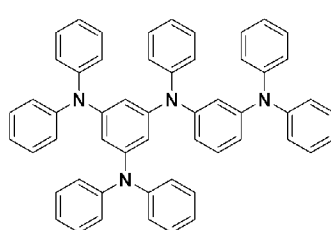


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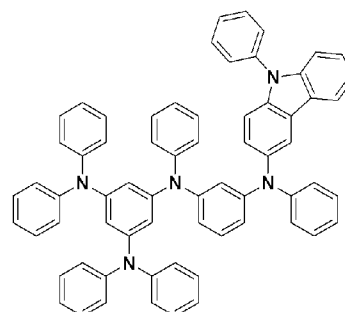
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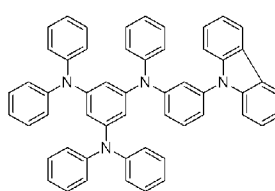


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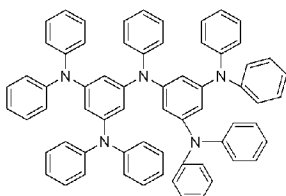


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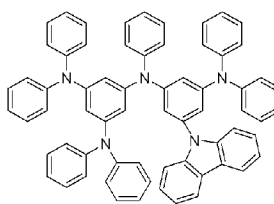
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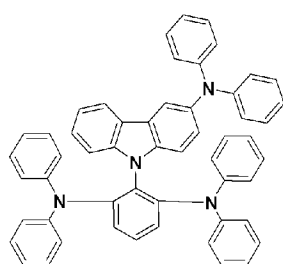


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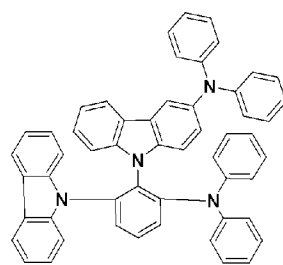


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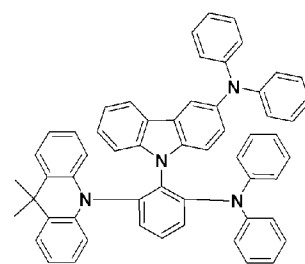
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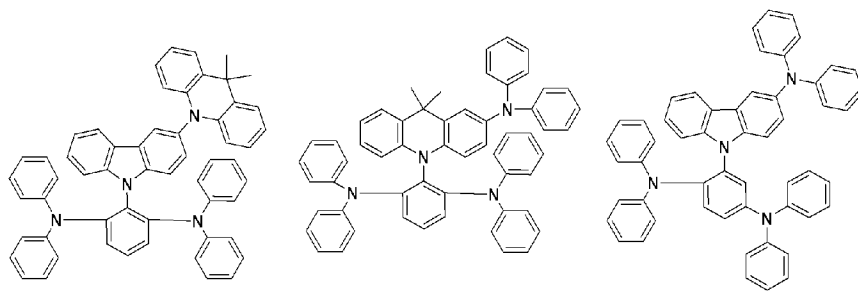


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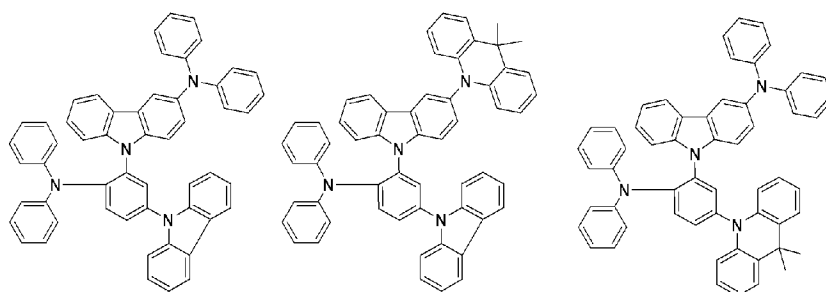


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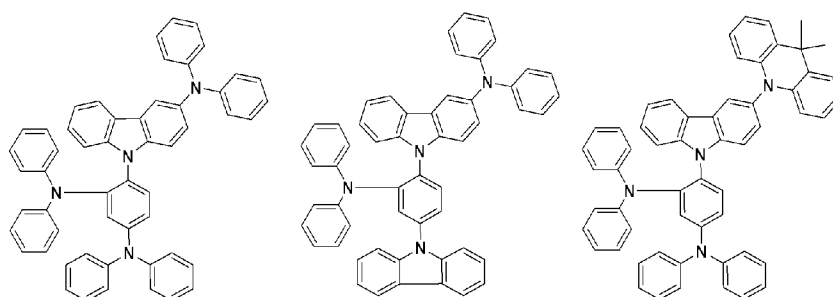


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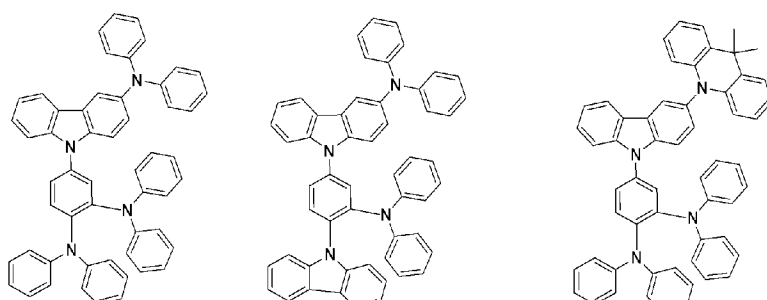


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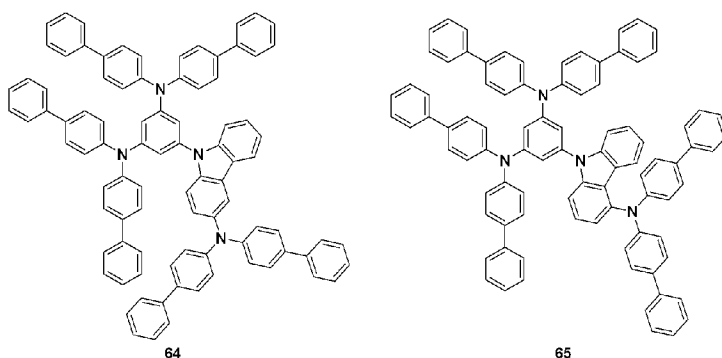


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- [73] According to an embodiment of the present invention, an organic electroluminescent (EL) device including the compound for an organic EL device according to the present invention may be provided.
- [74] According to an embodiment of the present invention, an organic EL device may include a first electrode, a second electrode, and a single organic layer or a plurality of organic layers between the first electrode and the second electrode, and one or more organic layers selected from among the single organic layer or the plurality of organic layers may include the compound for an organic EL device according to the present invention.
- [75] According to an embodiment of the present invention, the single organic layer or the plurality of organic layers may include a light emitting layer.
- [76] According to an embodiment of the present invention, the plurality of organic layers may include a light emitting layer, and the plurality of organic layers may further include one or more selected from among an electron injection layer, an electron transport layer, a hole blocking layer, an electron blocking layer, a hole transport layer and a hole injection layer.
- [77] According to an embodiment of the present invention, the light emitting layer may include a host and a dopant.

Advantageous Effects of Invention

- [78] According to an embodiment of the present invention, a compound including one or more derivatives having a substituted phenyl group is formed, making it possible to provide a variety of derivatives, whereby the properties of the compound can be adjusted. When this compound is contained in the organic layer of an organic EL device, various properties can be exhibited. Moreover, thermal stability and light emission efficiency of the organic EL device using the compound can be improved, and the use of the compound as a hole transport layer material enables a triplet exciton of a phosphorescent light emitting material to be confined, consequently improving the efficiency of the organic EL device.

Brief Description of Drawings

- [79] The above and other objects, features and advantages of the present invention will be more clearly understood from the following detailed description taken in conjunction with the accompanying drawings, in which:
- [80] FIG. 1 is a cross-sectional view illustrating an organic EL device according to an embodiment of the present invention; and
- [81] FIG. 2 is a cross-sectional view illustrating an organic EL device according to another embodiment of the present invention.

Mode for the Invention

- [82] The present invention may be variously modified, and may have a variety of embodiments, and is intended to illustrate specific embodiments. However, the following description does not limit the present invention to specific embodiments, and should be understood to include all variations, equivalents or substitutions within the spirit and scope of the present invention. Furthermore, in the description of the present invention, when it is determined that the detailed description of the related art would obscure the gist of the present invention, the description thereof will be omitted.
- [83] Also, in the following description, the terms “first,” “second” and the like are used to differentiate a certain component from other components, but the configuration of such components should not be construed to be limited by the terms. For example, a first component may be referred to as a second component, and a second component may be referred to as a first component, within the scope of the present invention.
- [84] Also, when any one component is mentioned to be “formed” or “stacked” on another component, it may be directly attached to the entire surface or one surface of another component, or a further component may be additionally interposed therebetween.
- [85] Unless otherwise stated, the singular expression includes a plural expression. In this application, the terms “include” and “have” are used to designate the presence of features, numbers, steps, operations, components, parts or combinations thereof described in the specification, not intending to exclude the presence or additional possibility of one or more different features, numbers, steps, operations, components, parts or combinations thereof are not excluded.
- [86] As used herein, unless otherwise defined, the term “valence bond” means a single bond, a double bond or a triple bond.
- [87] As used herein, unless otherwise defined, the term “substituted” means that at least one hydrogen on a substituent or a compound is substituted with deuterium, a halogen group, a hydroxyl group, an amino group, a C1 to C30 amine group, a nitro group, a C1 to C30 silyl group, a C1 to C30 alkyl group, a C1 to C30 alkylsilyl group, a C3 to C30 cycloalkyl group, a C1 to C30 heterocycloalkyl group, a C6 to C30 aryl group, a C1 to C30 heteroaryl group, a C1 to C20 alkoxy group, a C1 to C10 trifluoroalkyl group or a cyano group.
- [88] Further, among the halogen group, the hydroxyl group, the amino group, the C1 to C30 amine group, the C3 to C30 silyl group, the C1 to C30 alkyl group, the C1 to C30 alkylsilyl group, the C3 to C30 cycloalkyl group, the C6 to C30 aryl group, the C1 to C20 alkoxy group, the C1 to C10 trifluoroalkyl group or the cyano group, which is substituted, two adjacent substituents may be fused to form a ring.
- [89] As used herein, unless otherwise defined, the term “hetero” means a functional group containing 1 ~ 4 heteroatoms selected from the group consisting of N, O, S and P, the remainder being carbon.

- [90] As used herein, unless otherwise defined, the term “combination thereof” means that two or more substituents are coupled with each other by a linker or two or more substituents are condensed to each other.
- [91] As used herein, unless otherwise defined, the term “hydrogen” means hydrogen, deuterium or tritium.
- [92] As used herein, unless otherwise defined, the term “alkyl group” means an aliphatic hydrocarbon group.
- [93] The alkyl group may be a “saturated alkyl group” without any double bond or triple bond.
- [94] The alkyl group may be an “unsaturated alkyl group” with at least one double bond or triple bond.
- [95] The term “alkenylene group” means a functional group having at least one carbon-carbon double bond between at least two carbon atoms, and the term “alkynylene group” means a functional group having at least one carbon-carbon triple bond between at least two carbon atoms. The alkyl group may be branched, linear or cyclic, regardless of whether it is saturated or unsaturated.
- [96] The alkyl group may be a C1 to C30 alkyl group, preferably a C1 to C20 alkyl, more preferably a C1 to C10 alkyl group, and much more preferably a C1 to C6 alkyl group.
- [97] For example, a C1 to C4 alkyl group indicates an alkyl chain containing 1 ~ 4 carbon atoms, particularly an alkyl chain which is selected from the group consisting of methyl, ethyl, propyl, iso-propyl, n-butyl, iso-butyl, sec-butyl and t-butyl.
- [98] Specific examples of the alkyl group include a methyl group, an ethyl group, a propyl group, an isopropyl group, a butyl group, an isobutyl group, a t-butyl group, a pentyl group, a hexyl group, an ethenyl group, a propenyl group, a butenyl group, a cyclopropyl group, a cyclobutyl group, a cyclopentyl group, a cyclohexyl group, etc.
- [99] The “amine group” includes an arylamine group, an alkylamine group, an arylalkylamine group, or an alkylarylamine group.
- [100] The term “cycloalkyl group” refers to a monocyclic or fused-ring polycyclic (i.e., rings which share adjacent pairs of carbon atoms) functional group.
- [101] The term “heterocycloalkyl group” means a cycloalkyl group containing 1 ~ 4 heteroatoms selected from the group consisting of N, O, S and P, the remainder being carbon. In the case where the heterocycloalkyl group is a fused ring, each ring may contain 1 ~ 4 heteroatoms.
- [102] The term “aromatic group” means a cyclic functional group where all ring atoms have p-orbitals, and these p-orbitals form conjugation. Specific examples thereof include an aryl group and a heteroaryl group.
- [103] The term “aryl group” refers to a monocyclic or fused-ring polycyclic (i.e., rings which share adjacent pairs of carbon atoms) functional group.

- [104] The term “heteroaryl group” means an aryl group containing 1 ~ 4 heteroatoms selected from the group consisting of N, O, S and P, the remainder being carbon. In the case where the heteroalkyl group is a fused ring, each ring may contain 1 ~ 4 heteroatoms.
- [105] In the aryl group and the heteroaryl group, the number of ring atoms is the sum of the number of carbons and the number of non-carbon atoms.
- [106] When alkyl and aryl are used in combination as in “alkylaryl group” or “arylalkyl group,” “alkyl” and “aryl” respectively have the meanings as above.
- [107] The term “arylalkyl group” means an aryl substituted alkyl radical such as benzyl, and is incorporated in the alkyl group.
- [108] The term “alkylaryl group” means an alkyl substituted aryl radical, and is incorporated in the aryl group.
- [109] Below is a description of embodiments of the present invention with reference to the appended drawings, wherein the same or similar components are designated by the same reference numerals and the overlapping description thereof is omitted.
- [110] With reference to FIGS. 1 and 2, according to an embodiment of the present invention, an organic EL device 1 including the compound for an organic EL device according to the present invention may be provided.
- [111] According to another embodiment of the present invention, an organic EL device includes a first electrode 110, a second electrode 150, and a single organic layer or a plurality of organic layers 130 between the first electrode and the second electrode, and one or more organic layers selected from among the single organic layer or the plurality of organic layers 130 may include the compound for an organic EL device according to the present invention.
- [112] As such, the single organic layer or the plurality of organic layers 130 may include a light emitting layer 134.
- [113] The plurality of organic layers 130 include a light emitting layer 134, and the plurality of organic layers 130 may further include one or more selected from among an electron injection layer 131, an electron transport layer 132, a hole blocking layer 133, an electron blocking layer 135, a hole transport layer 136 and a hole injection layer 137.
- [114] Also, one or more selected from among the electron injection layer 131, the electron transport layer 132, the hole blocking layer 133, the light emitting layer 134, the electron blocking layer 135, the hole transport layer 136 and the hole injection layer 137 may be formed in a plural number.
- [115] The light emitting layer 134 may include a host and a dopant.
- [116] The organic EL device is preferably supported by a transparent substrate. The material for the transparent substrate is not particularly limited so long as it has good

mechanical strength, thermal stability and transparency. Specific examples thereof may include glass, a transparent plastic film, etc.

- [117] The anode material of the organic EL device according to the present invention may include a metal, an alloy, an electrically conductive compound or a mixture thereof, having a work function of 4 eV or more. Specific examples thereof may include Au metal or a transparent conductive material such as CuI, ITO (indium tin oxide), SnO₂ and ZnO. The thickness of the anode film is preferably set to 10 ~ 200 nm.
- [118] The cathode material of the organic EL device according to the present invention may include a metal, an alloy, an electrically conductive compound or a mixture thereof, having a work function of less than 4 eV. Specific examples thereof may include Na, a Na-K alloy, calcium, magnesium, lithium, a lithium alloy, indium, aluminum, a magnesium alloy, or an aluminum alloy. In addition, aluminum/AlO₂, aluminum/lithium, magnesium/silver or magnesium/indium may be used. The thickness of the cathode film is preferably set to 10 ~ 200 nm.
- [119] In order to increase light emission efficiency of the organic EL device, one or more electrodes preferably have a light transmittance of 10% or more. The sheet resistance of the electrodes is preferably hundreds of Ω/mm or less. The thickness of the electrodes falls in the range of 10 nm ~ 1 μm , and preferably 10 ~ 400 nm. Such electrodes may be manufactured in the form of a thin film using the above electrode material via vapor deposition such as chemical vapor deposition (CVD), physical vapor deposition (PVD) or the like, or sputtering.
- [120] When the compound for an organic EL device according to the present invention is used so as to be adapted for the purposes of the present invention, a hole transport material, a hole injection material, a light emitting layer material, a host material for a light emitting layer, an electron transport material, and an electron injection material, which are known, may be used alone in each organic layer, or may be used in selective combination with the compound for an organic EL device according to the present invention.
- [121] Examples of the hole transport material may include porphyrin compound derivatives including N,N-dicarbazoyl-3,5-benzene (mCP), poly(3,4-ethylenedioxythiophene):poly(styrenesulfonate) (PEDOT:PSS), N,N'-di(1-naphthyl)-N,N'-diphenylbenzidine (NPD), N,N'-diphenyl-N,N'-di(3-methylphenyl)-4,4'-diaminobiphenyl (TPD), N,N'-diphenyl-N,N'-dinaphthyl-4,4'-diaminobiphenyl, N,N,N',N'-tetra-p-tolyl-4,4'-diaminobiphenyl, N,N,N',N'-tetraphenyl-4,4'-diaminobiphenyl, 1,10,15,20-tetraphenyl-21H,23H-porphyrin copper(II), etc., triarylamine derivatives including polymers having an aromatic tertiary amine in the main chain or side chain

thereof, 1,1-bis(4-di-p-tolylaminophenyl)cyclohexane, N,N,N-tri(p-tolyl)amine and 4,4',4'-tris[N-(3-methylphenyl)-N-phenylamino]triphenylamine, carbazole derivatives including N-phenylcarbazole and polyvinylcarbazole, phthalocyanine derivatives including metal-free phthalocyanine and copper phthalocyanine, starburst amine derivatives, enaminestilbene-based derivatives, aromatic tertiary amine-containing styrylamine compound derivatives, polysilane, etc.

[122] Examples of the electron transport material may include diphenylphosphine oxide-4-(triphenylsilyl)phenyl (TSPO1), Alq₃, 2,5-diaryl sylo derivatives (PyPySPyPy), perfluorinated compounds (PF-6P), octasubstituted cyclooctatetraene compounds (COTs), etc.

[123] In the organic EL device according to the present invention, an electron injection layer, an electron transport layer, a hole transport layer and a hole injection layer may be provided in the form of a single layer containing one or more kinds of the above compound, or may be provided in the form of a plurality of stacked layers containing different kinds of compounds.

[124] The light emitting material may include, for example, photoluminescent fluorescent materials, fluorescent brighteners, laser dyes, organic scintillators and fluorescence analysis reagents. Specific examples thereof include carbazole-based compounds, phosphine oxide-based compounds, carbazole-based phosphine oxide compounds, polyaromatic compounds including bis((3,5-difluoro-4-cyanophenyl)pyridine)iridium picolinate (FCNIrpic), tris(8-hydroxyquinoline) aluminum (Alq₃), anthracene, phenanthrene, pyrene, chrysene, perylene, coronene, rubrene and quinacridone, oligophenylene compounds including quaterphenyl, scintillators for liquid scintillation including 1,4-bis(2-methylstyryl)benzene, 1,4-bis(4-methylstyryl)benzene, 1,4-bis(4-methyl-5-phenyl-2-oxazolyl)benzene, 1,4-bis(5-phenyl-2-oxazolyl)benzene, 2,5-bis(5-t-butyl-2-benzoxazolyl)thiophene, 1,4-diphenyl-1,3-butadiene, 1,6-diphenyl-1,3,5-hexatriene and 1,1,4,4-tetraphenyl-1,3-butadiene, metal complexes of oxine derivatives, coumarine dyes, dicyanomethylenepyran dyes, dicyanomethylenethiopyran dyes, polymethine dyes, oxobenzanthracene dyes, xanthene dyes, carbostyryl dyes, perylene dyes, oxazine compounds, stilbene derivatives, spiro compounds, oxadiazole compounds, etc.

[125] Each layer of the organic EL device according to the present invention may be provided in the form of a thin film using a known process such as vacuum deposition, spin coating or casting, or may be manufactured using each layer material. The thickness of each layer is not particularly limited, but may be appropriately set depending on the material properties, and may be typically determined in the range of 2 ~ 5,000 nm.

[126] Because the compound for an organic EL device according to the present invention

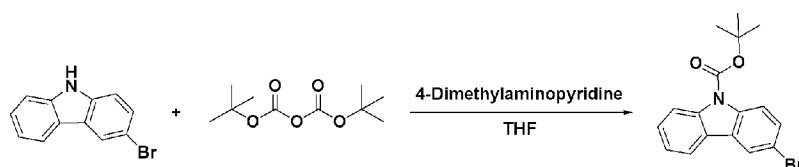
may be subjected to vacuum deposition, a thin film formation process is simple and a uniform thin film which does not substantially have pin holes may be easily obtained.

[127] A better understanding of the present invention regarding the synthesis of the compound for an organic EL device and the manufacture of the organic EL device including the same may be obtained through the following examples which are set forth to illustrate, but are not to be construed as limiting, the present invention.

[128] [Example]

[129] Preparation Example 1. Synthesis of Intermediate 1
(3-bromo-tert-butyl-9H-carbonylcarbazole)

[130]

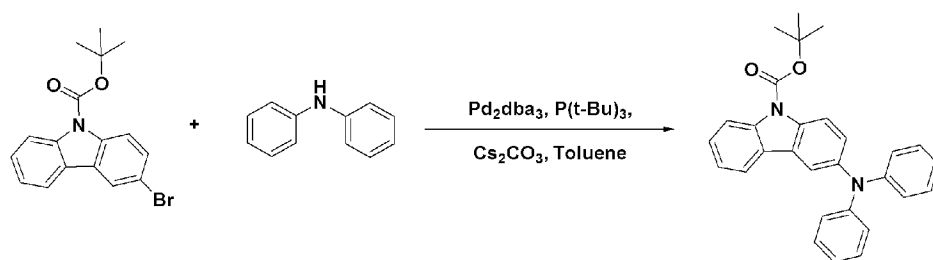


[131] In a 1 L round-bottom three-neck flask in a nitrogen atmosphere, 12.3 g of 3-bromocarbazole, 16.4 g of tert-butyldicarbonate, 0.9 g of 4-dimethylaminopyridine and 850 ml of THF were placed, and stirred at room temperature for 3 hr. The reaction solution was concentrated and then subjected to column chromatography using n-hexane, thus obtaining 16.1 g of Intermediate 1 (yield 93%).

[132] ¹H NMR (CDCl₃, 600 MHz) δ 8.27 (d, 1H), 8.19 (d, 1H), 8.08 (d, 1H), 7.92 (d, 1H), 7.54 (dd, 1H), 7.48 (td, 1H), 7.35 (t, 1H), 1.76 (s, 9H)

[133] Preparation Example 2. Synthesis of Intermediate 2
(3-diphenylamino-9H-tert-butylcarbonylcarbazole)

[134]

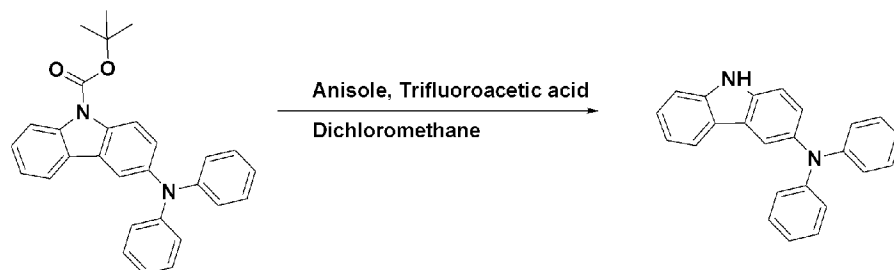


[135] In a 500 ml round-bottom three-neck flask in a nitrogen atmosphere, 16.1 g of Intermediate 1, 7.9 g of diphenylamine, 0.7 g of tris(dibenzylideneacetone)dipalladium(0), 1.9 g of 15% tris(tert-butyl)phosphine, 29.2 g of cesium carbonate and 300 ml of toluene were placed, and refluxed for 24 hr. The reaction solution was cooled, filtered with silica gel, concentrated and then subjected to column chromatography with a solvent mixture of dichloromethane and n-hexane, thus obtaining 13.1 g of Intermediate 2 (yield 64%).

[136] ¹H NMR (CDCl₃, 600 MHz) δ 8.29 (d, 1H), 8.19 (d, 1H), 7.82 (d, 1H), 7.73 (s, 1H), 7.44 (t, 1H), 7.28 (t, 1H), 7.25-7.23 (m, 5H), 7.11 (d, 4H), 6.99 (t, 2H), 1.74 (s, 9H)

[137] Preparation Example 3. Synthesis of Intermediate 3 (3-diphenylamino-9H-carbazole)

[138]

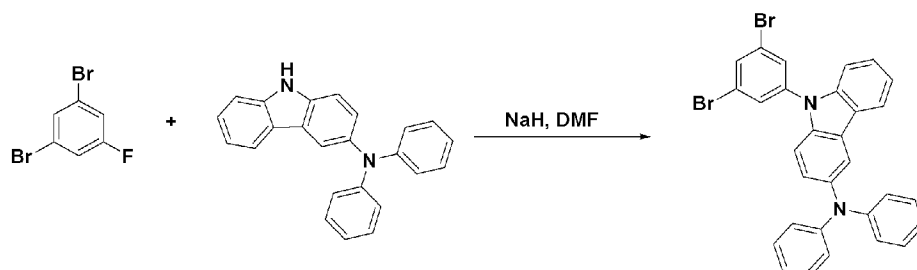


[139] In a 250 ml round-bottom three-neck flask in a nitrogen atmosphere, 9.0 g of Intermediate 2, 0.44 g of anisole, 50 ml of trifluoroacetic acid and 80 ml of dichloromethane were placed, and stirred at room temperature for 30 min. After completion of the reaction, the reaction solution was transferred into a separatory funnel and then neutralized with a sodium bicarbonate aqueous solution, after which the organic layer was dewatered with anhydrous magnesium sulfate, filtered and concentrated. The concentrated solution was recrystallized from dichloromethane and methanol, thus obtaining 4.7 g of Intermediate 3 (yield 68%).

[140] ^1H NMR (CDCl_3 , 600 MHz) δ 10.36 (s, 1H), 8.01 (d, 1H), 7.89 (d, 1H), 7.49 (t, 2H), 7.35 (t, 1H), 7.23-7.19 (m, 4H), 7.16 (dd, 1H), 7.11 (t, 1H), 7.02 (d, 4H), 6.91 (t, 2H)

[141] Preparation Example 4. Synthesis of Intermediate 4
(9-(3,5-dibromophenyl)-3-diphenylamino-9H-carbazole)

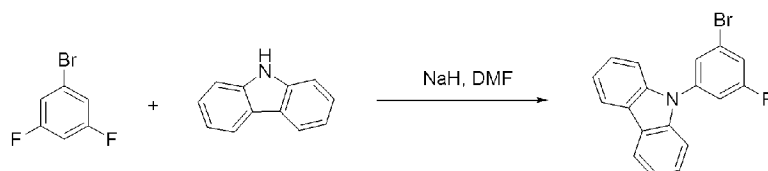
[142]



[143] In a 100 ml round-bottom three-neck flask in a nitrogen atmosphere, 2.5 g of 1,3-dibromo-5-fluorobenzene, 3.3 g of Intermediate 3, 0.8 g of 60% sodium hydride and 50 ml of DMF were placed, and stirred at 110°C for 4 hr. After completion of the reaction, the reaction solution was transferred into a separatory funnel, so that layers were separated with water and dichloromethane, after which the organic layer was dewatered with anhydrous magnesium sulfate, and filtered. The filtrate was concentrated and then subjected to column chromatography using a solvent mixture of dichloromethane and n-hexane, thus obtaining 4.3 g of Intermediate 4 (yield 75%).

[144] Preparation Example 5. Synthesis of Intermediate 5
(9-(3-bromo-5-fluorophenyl)-9H-carbazole)

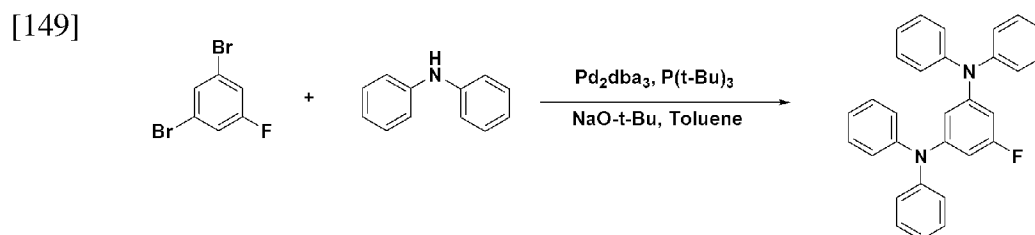
[145]



[146] In a 500 ml round-bottom three-neck flask in a nitrogen atmosphere, 29 g of 1-bromo-3,5-difluorobenzene, 8.5 g of carbazole, 4.0 g of 60% sodium hydride and 250 ml of DMF were placed, and stirred at 110°C for 4 hr. After completion of the reaction, the reaction solution was transferred into a separatory funnel, so that layers were separated with water and dichloromethane, after which the organic layer was dewatered with anhydrous magnesium sulfate, and filtered. The filtrate was concentrated and then subjected to column chromatography using a solvent mixture of dichloromethane and n-hexane, thus obtaining 8.6 g of Intermediate 5 (yield 50%).

[147] ¹H NMR (CDCl₃, 600 MHz) δ 8.13 (d, 2H), 7.57 (s, 1H), 7.43 (d, 4H), 7.36-7.30 (m, 4H)

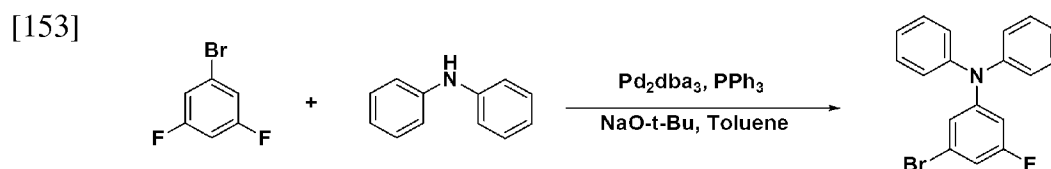
[148] Preparation Example 6. Synthesis of Intermediate 6
(1,3-bisdiphenylamino-5-fluorobenzene)



[150] In a 250 ml round-bottom three-neck flask in a nitrogen atmosphere, 5.0 g of 1,3-dibromo-5-fluorobenzene, 6.7 g of diphenylamine, 0.2 g of tris(dibenzylideneacetone)dipalladium(0), 0.3 g of 15% tris(tert-butyl)phosphine, 7.6 g of t-butoxy sodium and 100 ml of toluene were placed, and stirred at 80°C for 2 hr. The reaction solution was cooled, filtered with silica gel, concentrated, and then recrystallized from dichloromethane and methanol, thus obtaining 5.5 g of Intermediate 6 (yield 65%).

[151] ¹H NMR (CDCl₃, 600 MHz) δ 7.23 (t, 8H), 7.07 (d, 8H), 7.00 (t, 4H), 6.53 (s, 1H), 6.28 (dd, 2H)

[152] Preparation Example 7. Synthesis of Intermediate 7
(1-bromo-3-diphenylamino-5-fluorobenzene)



[154] In a 250 ml round-bottom three-neck flask in a nitrogen atmosphere, 10.2 g of 1-bromo-3,5-difluorobenzene, 3.4 g of diphenylamine, 0.4 g of tris(dibenzylideneacetone)dipalladium(0), 0.2 g of triphenylphosphine, 5.8 g of t-butoxy sodium and 100 ml of toluene were placed, and stirred at 100°C for 4 hr. The reaction solution was cooled, filtered with silica gel, concentrated and then subjected to column chromatography using a solvent mixture of dichloromethane and n-hexane,

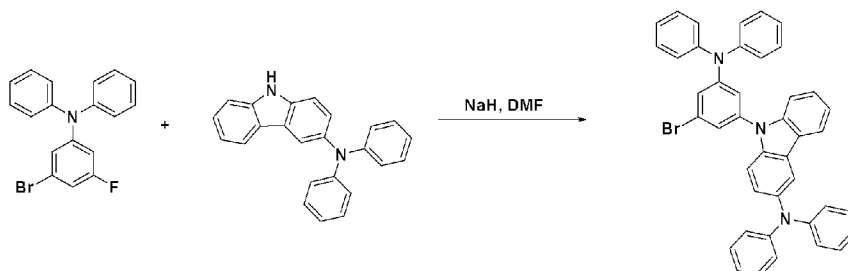
thus obtaining 4.8 g of Intermediate 7 (yield 70%).

[155] ^1H NMR (CDCl_3 , 600 MHz) δ 7.31 (t, 4H), 7.12-7.07 (m, 6H), 6.90 (s, 1H), 6.77 (d, 1H), 6.60 (d, 1H)

[156] Preparation Example 8. Synthesis of Intermediate 8

(9-(3-bromo-5-diphenylaminophenyl)-3-diphenylamino-9H-carbazole)

[157]

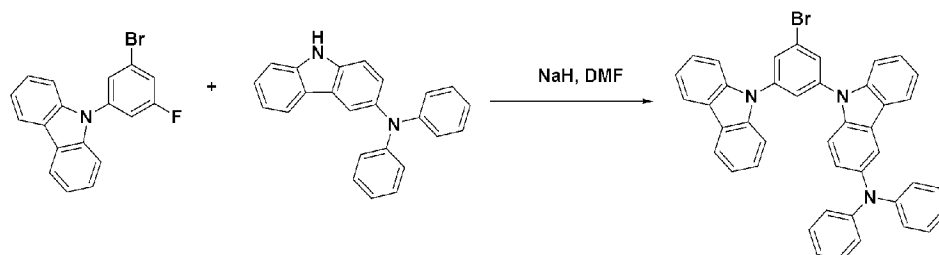


[158] In a 100 ml round-bottom three-neck flask in a nitrogen atmosphere, 3.4 g of Intermediate 7, 3.3 g of Intermediate 3, 0.8 g of 60% sodium hydride and 50 ml of DMF were placed, and stirred at 110°C for 2 hr. After completion of the reaction, the reaction solution was transferred into a separatory funnel, so that layers were separated with water and dichloromethane, after which the organic layer was dewatered with anhydrous magnesium sulfate, and filtered. The filtrate was concentrated and then subjected to column chromatography using a solvent mixture of dichloromethane and n-hexane, thus obtaining 4.9 g of Intermediate 8 (yield 75%).

[159] Preparation Example 9. Synthesis of Intermediate 9

(9-(3-bromo-5-(9H-carbazolyl)phenyl)-3-diphenylamino-9H-carbazole)

[160]

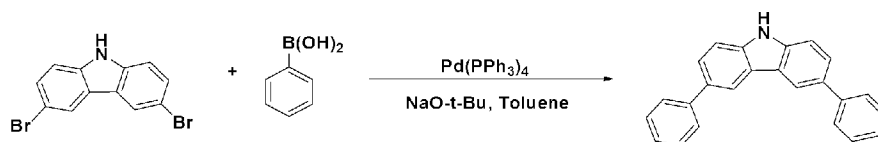


[161] In a 100 ml round-bottom three-neck flask in a nitrogen atmosphere, 4.1 g of Intermediate 5, 4.0 g of Intermediate 3, 1.0 g of 60% sodium hydride and 60 ml of DMF were placed, and stirred at 110°C for 4 hr. After completion of the reaction, the reaction solution was transferred into a separatory funnel, so that layers were separated with water and dichloromethane, after which the organic layer was dewatered with anhydrous magnesium sulfate, and filtered. The filtrate was concentrated and then subjected to column chromatography using a solvent mixture of dichloromethane and n-hexane, thus obtaining 5.5 g of Intermediate 9 (yield 70%).

[162] ^1H NMR (CDCl_3 , 600 MHz) δ 8.15 (d, 2H), 7.98 (d, 1H), 7.91 (s, 1H), 7.85 (d, 2H), 7.79 (s, 1H), 7.54-7.52 (m, 3H), 7.48-7.43 (m, 4H), 7.32 (t, 2H), 7.27-7.20 (m, 6H), 7.12 (d, 4H), 6.96 (t, 2H)

[163] Preparation Example 10. Synthesis of Intermediate 10 (3,6-diphenyl-9H-carbazole)

[164]

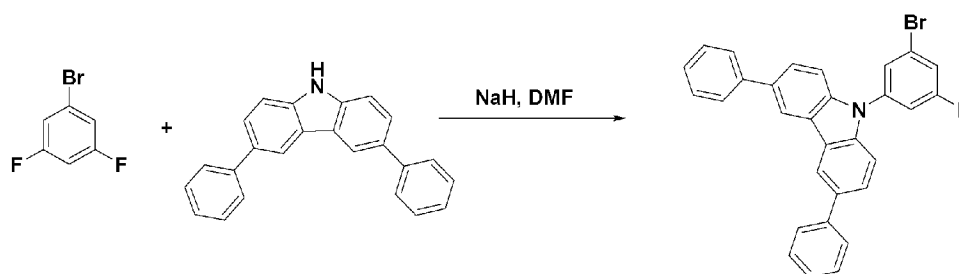


[165] In a 250 ml round-bottom three-neck flask in a nitrogen atmosphere, 5.0 g of 3,6-dibromocarbazole, 4.1 g of phenylboronic acid, 0.4 g of tetrakis(triphenylphosphine)palladium(0), 5.9 g of t-butoxy sodium and 150 ml of toluene were placed, and stirred under reflux for 24 hr. The reaction solution was cooled, filtered with silica gel, concentrated and then subjected to column chromatography with a solvent mixture of dichloromethane and n-hexane, thus obtaining 3.3 g of Intermediate 10 (yield 68%).

[166] ^1H NMR (CDCl_3 , 600 MHz) δ 8.33 (s, 2H), 8.11 (s, 1H), 7.72 (d, 4H), 7.69 (d, 2H), 7.51-7.46 (m, 6H), 7.34 (t, 2H)

[167] Preparation Example 11. Synthesis of Intermediate 11 (9-(3-bromo-5-fluorophenyl)-3,6-diphenyl-9H-carbazole)

[168]

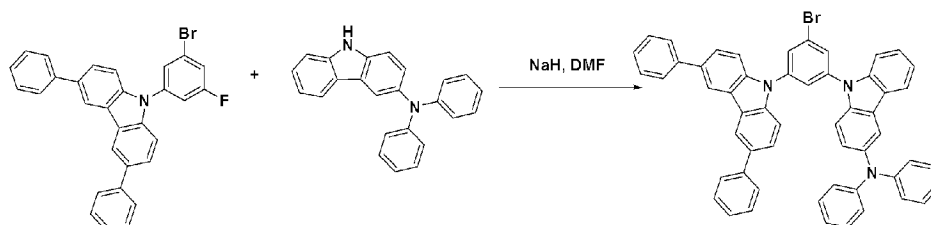


[169] In a 100 ml round-bottom three-neck flask in a nitrogen atmosphere, 5.7 g of 1-bromo-3,5-difluorobenzene, 3.3 g of Intermediate 10, 0.8 g of 60% sodium hydride and 50 ml of DMF were placed, and stirred at 110°C for 4 hr. After completion of the reaction, the reaction solution was transferred into a separatory funnel, so that layers were separated with water and dichloromethane, after which the organic layer was dewatered with anhydrous magnesium sulfate, and filtered. The filtrate was concentrated and then subjected to column chromatography using a solvent mixture of dichloromethane and n-hexane, thus obtaining 3.1 g of Intermediate 11 (yield 62%).

[170] ^1H NMR (CDCl_3 , 600 MHz) δ 8.39 (s, 2H), 7.75 (d, 4H), 7.70 (d, 2H), 7.60 (s, 1H), 7.55-7.48 (m, 6H), 7.41-7.36 (m, 3H), 7.32 (d, 1H)

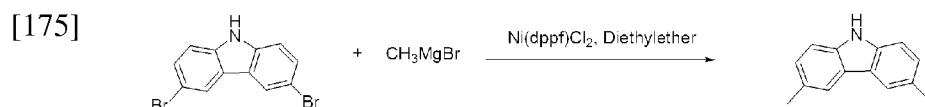
[171] Preparation Example 12. Synthesis of Intermediate 12 (9-(3-bromo-5-(3,6-diphenyl-9H-carbazolyl)phenyl)-3-diphenylamino-9H-carbazole)

[172]



[173] In a 100 ml round-bottom three-neck flask in a nitrogen atmosphere, 3.1 g of Intermediate 11, 2.2 g of Intermediate 3, 0.5 g of 60% sodium hydride and 35 ml of DMF were placed, and stirred at 110°C for 4 hr. After completion of the reaction, the reaction solution was transferred into a separatory funnel, so that layers were separated with water and dichloromethane, after which the organic layer was dewatered with anhydrous magnesium sulfate, and filtered. The filtrate was concentrated and then subjected to column chromatography using a solvent mixture of dichloromethane and n-hexane, thus obtaining 5.2 g of Intermediate 12 (yield 67%).

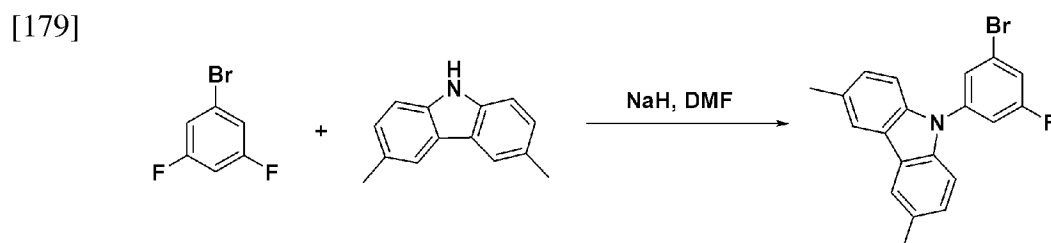
[174] Preparation Example 13. Synthesis of Intermediate 13 (3,6-dimethyl-9H-carbazole)



[176] In a 1 L round-bottom three-neck flask in a nitrogen atmosphere, 16 g of 3,6-dibromocarbazole, 1.4 g of nickel(diphenylphosphinoferrocene)chloride(2) and 600 ml of diethylether were placed, and 30 ml of 3M methylmagnesium bromide was slowly added dropwise at room temperature. The reaction mixture was refluxed for 8 hr, after which the reaction solution was cooled, washed with an ammonium chloride aqueous solution, and neutralized with a sodium carbonate aqueous solution. The organic layer was then dried with magnesium sulfate, filtered and concentrated. The concentrated solution was subjected to column chromatography using a solvent mixture of dichloromethane and n-hexane, thus obtaining 2.4 g of Intermediate 13 (yield 24%).

[177] ¹H NMR (Acetone, d₆, 600 MHz) δ 10.02 (s, 1H), 7.83 (s, 2H), 7.32 (d, 2H), 7.15 (d, 2H), 2.45 (s, 6H)

[178] Preparation Example 14. Synthesis of Intermediate 14 (9-(3-bromo-5-fluorophenyl)-3,6-dimethyl-9H-carbazole)



[180] In a 100 ml round-bottom three-neck flask in a nitrogen atmosphere, 5.7 g of 1-bromo-3,5-difluorobenzene, 2.0 g of Intermediate 13, 0.8 g of 60% sodium hydride and 50 ml of DMF were placed, and stirred at 110°C for 4 hr. After completion of the reaction, the reaction solution was transferred into a separatory funnel, so that layers were separated with water and dichloromethane, after which the organic layer was dewatered with anhydrous magnesium sulfate, and filtered. The filtrate was concentrated and then subjected to column chromatography using a solvent mixture of

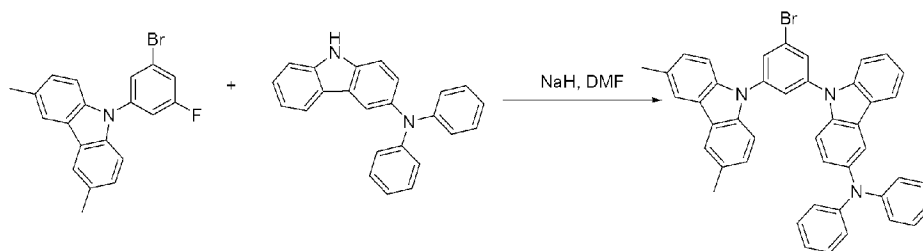
dichloromethane and n-hexane, thus obtaining 2.4 g of Intermediate 14 (yield 65%).

[181] ^1H NMR (CDCl_3 , 600 MHz) δ 7.87 (s, 2H), 7.54 (s, 1H), 7.33 (d, 2H), 7.30 (d, 1H), 7.26 (d, 1H), 7.22 (d, 2H), 2.53 (s, 6H)

[182] Preparation Example 15. Synthesis of Intermediate 15

(9-(3-bromo-5-(3,6-dimethyl-9H-carbazolyl)phenyl)-3-diphenylamino-9H-carbazole)

[183]



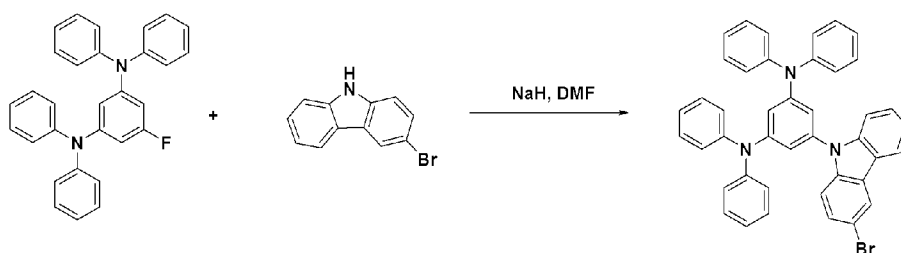
[184] In a 100 ml round-bottom three-neck flask in a nitrogen atmosphere, 2.4 g of Intermediate 14, 2.2 g of Intermediate 3, 0.5 g of 60% sodium hydride and 35 ml of DMF were placed, and stirred at 110°C for 4 hr. After completion of the reaction, the reaction solution was transferred into a separatory funnel, so that layers were separated with water and dichloromethane, after which the organic layer was dewatered with anhydrous magnesium sulfate, and filtered. The filtrate was concentrated and then subjected to column chromatography using a solvent mixture of dichloromethane and n-hexane, thus obtaining 3.7 g of Intermediate 15 (yield 83%).

[185] ^1H NMR (CDCl_3 , 600 MHz) δ 7.98 (d, 1H), 7.90 (s, 1H), 7.88 (s, 2H), 7.81 (d, 2H), 7.76 (s, 1H), 7.51 (d, 1H), 7.46 (d, 1H), 7.43-7.40 (m, 3H), 7.26-7.21 (m, 8H), 7.10 (d, 4H), 6.96 (t, 2H), 2.53 (s, 6H)

[186] Preparation Example 16. Synthesis of Intermediate 16

(9-(1,3-bis(diphenylamino)phenyl)-3-bromo-9H carbazole)

[187]



[188] In a 100 ml round-bottom three-neck flask in a nitrogen atmosphere, 5.5 g of Intermediate 6, 3.2 g of 3-bromocarbazole, 1.0 g of 60% sodium hydride and 65 ml of DMF were placed, and stirred under reflux for 24 hr. After completion of the reaction, the reaction solution was transferred into a separatory funnel, so that layers were separated with water and dichloromethane, after which the organic layer was dewatered with anhydrous magnesium sulfate, and filtered. The filtrate was concentrated and then subjected to column chromatography using a solvent mixture of dichloromethane and n-hexane, thus obtaining 4.0 g of Intermediate 16 (yield 47%).

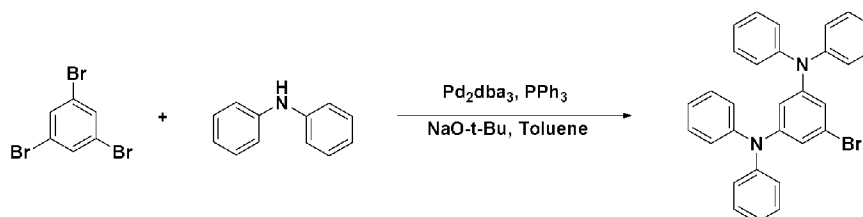
[189] ^1H NMR (CDCl_3 , 600 MHz) δ 8.14 (s, 1H), 8.98 (d, 1H), 7.42-7.37 (m, 3H),

7.28-7.22 (m, 10H), 7.15 (d, 8H), 6.99 (t, 4H), 6.87 (s, 1H), 6.70 (s, 2H)

[190] Preparation Example 17. Synthesis of Intermediate 17

(1-bromo-(3,5-bisdiphenylamino)benzene)

[191]



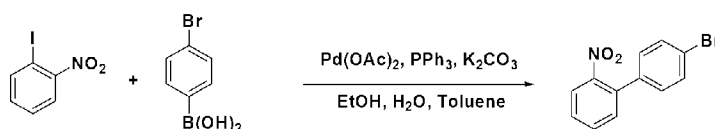
[192] In a 500 ml round-bottom three-neck flask in a nitrogen atmosphere, 9.0 g of 1,3,5-tribromobenzene, 8.0 g of diphenylamine, 0.3 g of tris(dibenzylideneacetone)dipalladium(0), 0.3 g of triphenylphosphine, 19.4 g of t-butoxy sodium and 200 ml of toluene were placed, and stirred at 80°C for 2 hr. The reaction solution was cooled, filtered with silica gel, concentrated and then subjected to column chromatography with a solvent mixture of dichloromethane and n-hexane, thus obtaining 6.3 g of Intermediate 17 (yield 85%).

[193] ^1H NMR (CDCl_3 , 600 MHz) δ 7.23 (t, 8H), 7.07 (d, 8H), 7.01 (t, 4H), 6.75 (d, 2H), 6.71 (d, 1H)

[194] Preparation Example 18. Synthesis of Intermediate 18

(4'-bromo-2-nitro-1,1'-biphenyl)

[195]



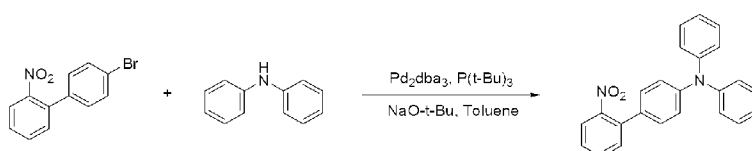
[196] In a 250 ml round-bottom three-neck flask in a nitrogen atmosphere, 4.9 g of 2-iodonitrobenzene, 4.3 g of 4-bromophenylboronic acid, 0.2 g of palladium acetate(II), 0.8 g of triphenylphosphine, 6.0 g of potassium carbonate, 20 ml of ethanol, 20 ml of water and 175 ml of toluene were placed, and stirred under reflux for 24 hr. The reaction solution was cooled and then extracted with dichloromethane and water, after which the organic layer was concentrated and then subjected to column chromatography using a solvent mixture of dichloromethane and n-hexane, thus obtaining 4.4 g of Intermediate 18 (yield 81%).

[197] ^1H NMR (CDCl_3 , 600 MHz) δ 7.88 (d, 1H), 7.62 (t, 1H), 7.56-7.51 (m, 2H), 7.50 (t, 1H), 7.39 (d, 1H), 7.19-7.17 (m, 2H)

[198] Preparation Example 19. Synthesis of Intermediate 19

(4'-diphenylamino-2-nitro-1,1'-biphenyl)

[199]

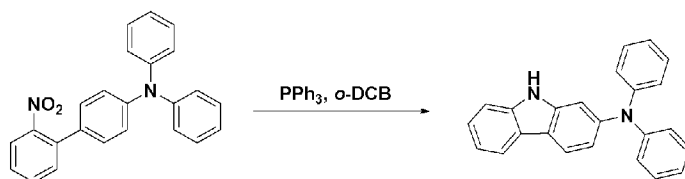


[200] In a 100 ml round-bottom three-neck flask in a nitrogen atmosphere, 4.4 g of Intermediate 18, 2.7 g of diphenylamine, 0.5 g of tris(dibenzylideneacetone)dipalladium(0), 0.6 g of 15% tris(tert-butyl)phosphine, 4.6 g of t-butoxy sodium and 80 ml of toluene were placed, and stirred at 80°C for 12 hr. The reaction solution was cooled, filtered with silica gel, concentrated and then subjected to column chromatography with a solvent mixture of dichloromethane and n-hexane, thus obtaining 4.9 g of Intermediate 19 (yield 84%).

[201] ¹H NMR (CDCl₃, 600 MHz) δ 7.80 (d, 1H), 7.57 (t, 1H), 7.46-7.42 (m, 2H), 7.29-7.26 (m, 4H), 7.17-7.14 (m, 6H), 7.08-7.04 (m, 4H)

[202] Preparation Example 20. Synthesis of Intermediate 20
(2-diphenylamino-9H-carbazole)

[203]

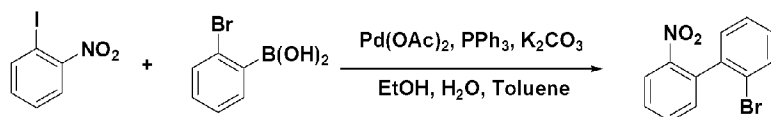


[204] In a 100 ml round-bottom three-neck flask in a nitrogen atmosphere, 4.9 g of Intermediate 19, 8.8 g of triphenylphosphine and 20 ml of o-dichlorobenzene were placed, and stirred under reflux for 24 hr. The reaction solution was cooled, after which the organic layer was concentrated and then subjected to column chromatography using a solvent mixture of dichloromethane and n-hexane, thus obtaining 2.4 g of Intermediate 20 (yield 53%).

[205] ¹H NMR (Acetone, d₆, 600 MHz) δ 10.15 (s, 1H), 8.02 (d, 1H), 7.99 (d, 1H), 7.43 (d, 1H), 7.30 (t, 1H), 7.28-7.25 (m, 4H), 7.13 (t, 2H), 7.07 (d, 4H), 7.00 (t, 2H), 6.91 (d, 1H)

[206] Preparation Example 21. Synthesis of Intermediate 21
(2'-diphenylamino-2-nitro-1,1'-biphenyl)

[207]

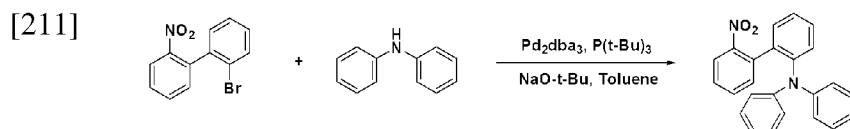


[208] In a 500 ml round-bottom three-neck flask in a nitrogen atmosphere, 6.2 g of 2-iodonitrobenzene, 5.5 g of 2-bromophenylboronic acid, 0.3 g of palladium acetate(II), 1.0 g of triphenylphosphine, 7.6 g of potassium carbonate, 25 ml of ethanol, 25 ml of water and 200 ml of toluene were placed, and stirred under reflux for 24 hr. The reaction solution was cooled and then extracted with dichloromethane and water, after which the organic layer was concentrated and then subjected to column chromatography using a solvent mixture of dichloromethane and n-hexane, thus obtaining 5.0 g of Intermediate 21 (yield 72%).

[209] ¹H NMR (CDCl₃, 600 MHz) δ 8.12 (d, 1H), 7.69 (t, 1H), 7.65 (d, 1H), 7.60 (d, 1H),

7.38 (t, 1H), 7.37 (d, 1H), 7.31-7.25 (m, 2H)

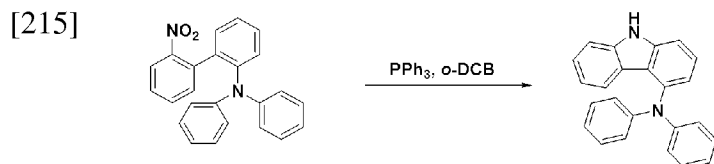
[210] Preparation Example 22. Synthesis of Intermediate 22
 (2'-diphenylamino-2-nitro-1,1'-biphenyl)



[212] In a 100 ml round-bottom three-neck flask in a nitrogen atmosphere, 5.0 g of Intermediate 21, 3.0 g of diphenylamine, 0.5 g of tris(dibenzylideneacetone)dipalladium(0), 0.6 g of 15% tris(tert-butyl)phosphine, 5.2 g of t-butoxy sodium and 90 ml of toluene were placed, and stirred at 80°C for 12 hr. The reaction solution was cooled, filtered with silica gel, concentrated and then subjected to column chromatography using a solvent mixture of dichloromethane and n-hexane, thus obtaining 5.1 g of Intermediate 22 (yield 77%).

[213] ¹H NMR (CDCl₃, 600 MHz) δ 7.69 (d, 1H), 7.34 (t, 1H), 7.27-7.22 (m, 4H), 7.18 (t, 1H), 7.02 (t, 4H), 6.91 (d, 1H), 6.83 (t, 2H), 6.76 (d, 4H)

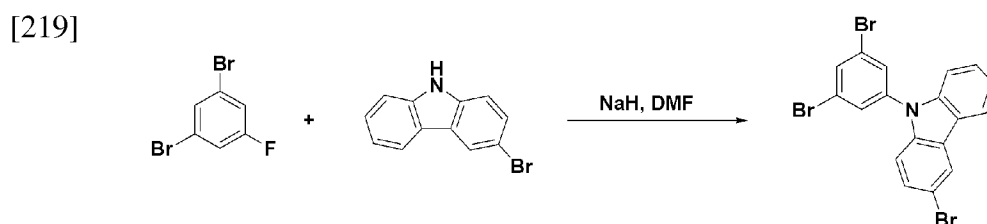
[214] Preparation Example 23. Synthesis of Intermediate 23
 (4-diphenylamino-9H-carbazole)



[216] In a 100 ml round-bottom three-neck flask in a nitrogen atmosphere, 5.1 g of Intermediate 22, 9.2 g of triphenylphosphine and 25 ml of o-dichlorobenzene were placed, and stirred under reflux for 24 hr. The reaction solution was cooled, after which the organic layer was concentrated and then subjected to column chromatography using a solvent mixture of dichloromethane and n-hexane, thus obtaining 2.9 g of Intermediate 23 (yield 62%).

[217] ¹H NMR (Acetone, d₆, 600 MHz) δ 10.53 (s, 1H), 7.69 (d, 1H), 7.99 (d, 1H), 7.46-7.43 (m, 2H), 7.40 (t, 1H), 7.27 (t, 1H), 7.21-7.18 (m, 4H), 7.03 (d, 4H), 6.90 (t, 2H), 6.86 (d, 1H)

[218] Preparation Example 24. Synthesis of Intermediate 24
 (9-(1,3-dibromophenyl)-3-bromo-9H-carbazole)



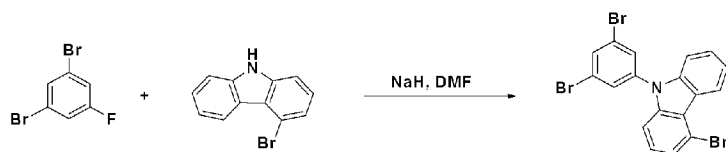
[220] In a 250 ml round-bottom three-neck flask in a nitrogen atmosphere, 6.0 g of

1,3-dibromo-5-fluorobenzene, 5.8 g of 3-bromocarbazole, 1.9 g of 60% sodium hydride and 120 ml of DMF were placed, and stirred at 160°C for 12 hr. After completion of the reaction, the reaction solution was transferred into a separatory funnel, so that layers were separated with water and dichloromethane, after which the organic layer was dewatered with anhydrous magnesium sulfate, and filtered. The filtrate was concentrated and then subjected to column chromatography using a solvent mixture of dichloromethane and n-hexane, thus obtaining 8.0 g of Intermediate 24 (yield 71%).

[221] ^1H NMR (CDCl_3 , 600 MHz) δ 8.23 (s, 1H), 8.06 (d, 1H), 7.78 (s, 1H), 7.65 (s, 1H), 7.64 (s, 1H), 7.51 (d, 1H), 7.46 (t, 1H), 7.39 (d, 1H), 7.32 (d, 1H), 7.27 (d, 1H)

[222] Preparation Example 25. Synthesis of Intermediate 25
(9-(1,3-dibromophenyl)-4-bromo-9H-carbazole)

[223]

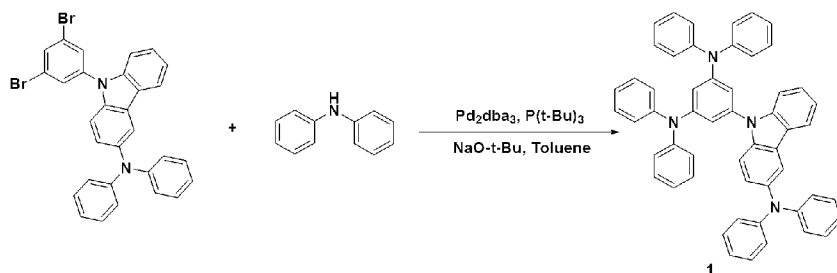


[224] In a 250 ml round-bottom three-neck flask in a nitrogen atmosphere, 6.0 g of 1,3-dibromo-3-fluorobenzene, 5.8 g of 4-bromocarbazole, 1.9 g of 60% sodium hydride and 120 ml of DMF were placed, and stirred at 160°C for 24 hr. After completion of the reaction, the reaction solution was transferred into a separatory funnel, so that layers were separated with water and dichloromethane, after which the organic layer was dewatered with anhydrous magnesium sulfate, and filtered. The filtrate was concentrated and then subjected to column chromatography using a solvent mixture of dichloromethane and n-hexane, thus obtaining 7.3 g of Intermediate 25 (yield 65%).

[225] ^1H NMR (CDCl_3 , 600 MHz) δ 8.83 (s, 1H), 7.80 (s, 1H), 7.65 (s, 1H), 7.64 (s, 1H), 7.50-7.47 (m, 2H), 7.38-7.35 (m, 2H), 7.31 (d, 1H), 7.27 (d, 1H)]

[226] Example 1. Synthesis of Compound 1

[227]



[228] In a 100 ml round-bottom three-neck flask in a nitrogen atmosphere, 3.3 g of Intermediate 4 synthesized in Preparation Example 4, 3.4 g of diphenylamine, 0.3 g of tris(dibenzylideneacetone)dipalladium(0), 0.4 g of 15% tris(t-butyl)phosphine, 5.8 g of t-butoxy sodium and 50 ml of toluene were placed, and stirred at 80°C for 2 hr. The

reaction solution was cooled, filtered with silica gel, concentrated and then subjected to column chromatography using a solvent mixture of dichloromethane and n-hexane, thus obtaining 6.3 g of Compound 1 (yield 85%).

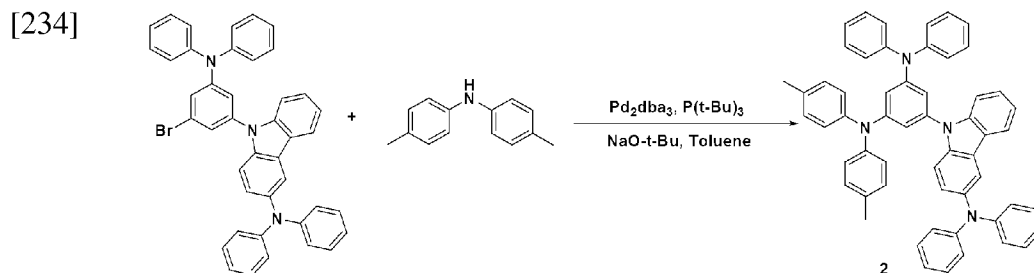
[229] ^1H NMR (CDCl_3 , 600 MHz) δ 7.88 (d, 1H), 7.81 (s, 1H), 7.35 (d, 1H), 7.34-7.32 (m, 2H), 7.25-7.14 (m, 22H), 7.07 (d, 4H), 6.98 (t, 4H), 6.93 (t, 2H), 6.86 (t, 1H), 6.77 (d, 2H)

[230] MS (ESI): $[\text{M}+\text{H}]^+$ 745

[231] Tg: 111°C

[232] Tm: 267°C

[233] Example 2. Synthesis of Compound 2

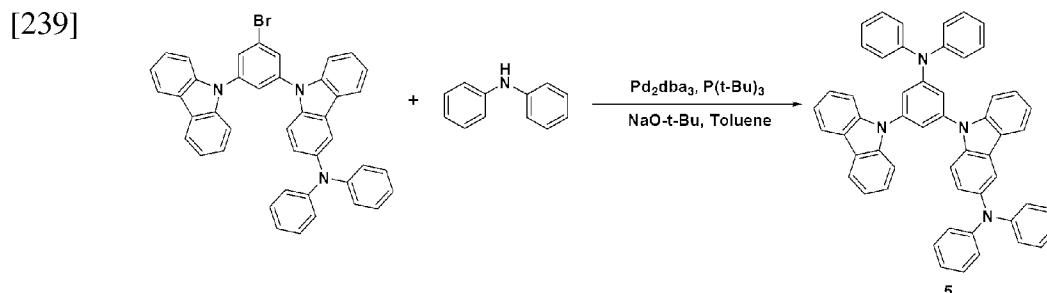


[235] In a 100 ml round-bottom three-neck flask in a nitrogen atmosphere, 3.3 g of Intermediate 8 synthesized in Preparation Example 8, 1.0 g of di-p-tolylamine, 0.1 g of tris(dibenzylideneacetone)dipalladium(0), 0.2 g of 15% tris(t-butyl)phosphine, 1.4 g of t-butoxy sodium and 50 ml of toluene were placed, and stirred at 80°C for 2 hr. The reaction solution was cooled, filtered with silica gel, concentrated and then subjected to column chromatography using a solvent mixture of dichloromethane and n-hexane, thus obtaining 3.3 g of Compound 2 (yield 85%).

[236] ^1H NMR (CDCl_3 , 600 MHz) δ 7.88 (d, 1H), 7.81 (s, 1H), 7.38 (d, 1H), 7.35-7.31 (m, 2H), 7.25-7.13 (m, 14H), 7.07 (d, 4H), 7.05-7.00 (m, 8H), 6.98 (t, 2H), 6.94 (t, 2H), 6.82 (s, 1H), 6.72 (d, 2H)

[237] MS (ESI): $[\text{M}+\text{H}]^+$ 773

[238] Example 3. Synthesis of Compound 5



[240] In a 100 ml round-bottom three-neck flask in a nitrogen atmosphere, 3.0 g of Intermediate 9 synthesized in Preparation Example 9 was placed, and the reaction solution was cooled, filtered with silica gel, concentrated and then subjected to column

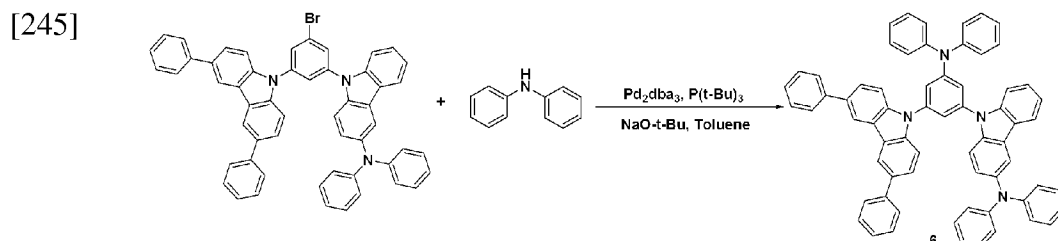
chromatography using a solvent mixture of dichloromethane and n-hexane, thus obtaining 3.1 g of Compound 5 (yield 92%).

[241] ^1H NMR (CDCl_3 , 600 MHz) δ 8.10 (d, 2H), 7.94 (d, 1H), 7.86 (s, 1H), 7.53-7.50 (m, 3H), 7.46 (d, 1H), 7.42-7.37 (m, 4H), 7.36-7.26 (m, 12H), 7.24-7.19 (m, 6H), 7.07-7.08 (m, 6H), 6.95 (t, 2H)

[242] MS (ESI): $[\text{M}+\text{H}]^+$ 743

[243] Tg: 133°C

[244] Example 4. Synthesis of Compound 6

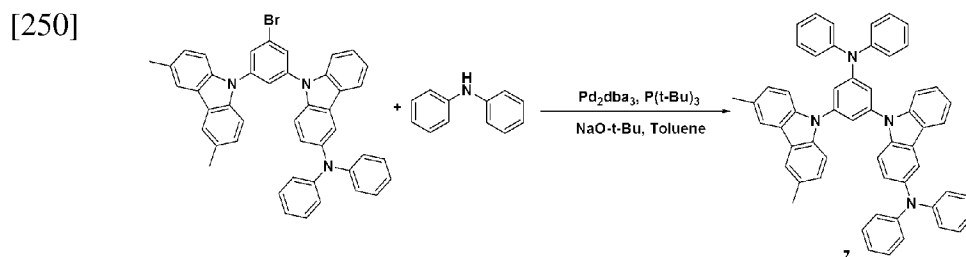


[246] In a 100 ml round-bottom three-neck flask in a nitrogen atmosphere, 4.0 g of Intermediate 12 synthesized in Preparation Example 12, 1.0 g of diphenylamine, 0.1 g of tris(dibenzylideneacetone)dipalladium(0), 0.2 g of 15% tris(t-butyl)phosphine, 1.4 g of t-butoxy sodium and 50 ml of toluene were placed, and stirred at 80°C for 2 hr. The reaction solution was cooled, filtered with silica gel, concentrated and then subjected to column chromatography using a solvent mixture of dichloromethane and n-hexane, thus obtaining 4.0 g of Compound 6 (yield 90%).

[247] ^1H NMR (CDCl_3 , 600 MHz) δ 8.36 (s, 2H), 7.95 (d, 1H), 7.87 (s, 1H), 7.71 (d, 4H), 7.67 (dd, 2H), 7.60 (d, 2H), 7.54 (d, 1H), 7.48-7.46 (m, 5H), 7.43-7.31 (m, 12H), 7.31 (s, 2H), 7.25-7.20 (m, 6H), 7.12-7.06 (m, 6H), 6.95 (t, 2H)

[248] MS (ESI): $[\text{M}+\text{H}]^+$ 895

[249] Example 5. Synthesis of Compound 7



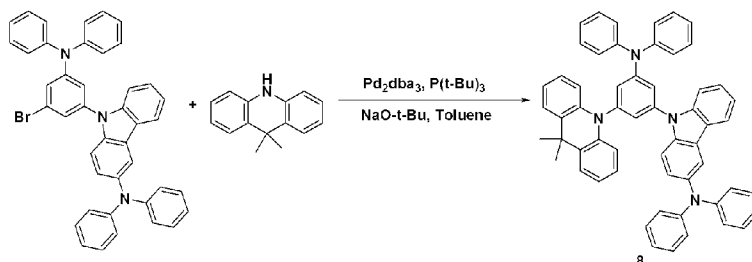
[251] In a 100 ml round-bottom three-neck flask in a nitrogen atmosphere, 3.4 g of Intermediate 15 synthesized in Preparation Example 15, 0.8 g of diphenylamine, 0.1 g of tris(dibenzylideneacetone)dipalladium(0), 0.2 g of 15% tris(t-butyl)phosphine, 1.4 g of t-butoxy sodium and 50 ml of toluene were placed, and stirred at 80°C for 2 hr. The reaction solution was cooled, filtered with silica gel, concentrated and then subjected to column chromatography using a solvent mixture of dichloromethane and n-hexane, thus obtaining 3.4 g of Compound 7 (yield 88%).

[252] ^1H NMR (CDCl_3 , 600 MHz) δ 7.94 (d, 1H), 7.87 (s, 1H), 7.85 (s, 2H), 7.51 (d, 1H), 7.46 (d, 1H), 7.41-7.30 (m, 13H), 7.23-7.19 (m, 9H), 7.10-7.08 (m, 6H), 6.95 (t, 2H), 2.51 (s, 6H)

[253] MS (ESI): $[\text{M}+\text{H}]^+$ 771

[254] Example 6. Synthesis of Compound 8

[255]



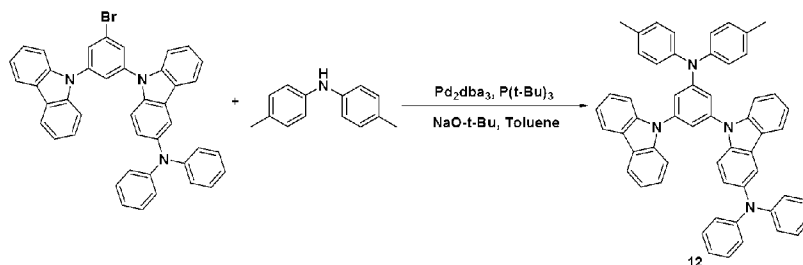
[256] In a 100 ml round-bottom three-neck flask in a nitrogen atmosphere, 3.1 g of Intermediate 8 synthesized in Preparation Example 8, 1.0 g of 9,9-dimethyl-9,10-dihydroacridine, 0.1 g of tris(dibenzylideneacetone)dipalladium(0), 0.2 g of 15% tris(t-butyl)phosphine, 1.4 g of t-butoxy sodium and 50 ml of toluene were placed, and stirred at 80°C for 2 hr. The reaction solution was cooled, filtered with silica gel, concentrated and then subjected to column chromatography using a solvent mixture of dichloromethane and n-hexane, thus obtaining 3.1 g of Compound 8 (yield 80%).

[257] ^1H NMR (CDCl_3 , 600 MHz) δ 7.92 (d, 1H), 7.85 (s, 1H), 7.46 (d, 1H), 7.43-7.40 (m, 3H), 7.38-7.29 (m, 6H), 7.26-7.24 (m, 4H), 7.22-7.18 (m, 6H), 7.12-7.04 (m, 10H), 6.94 (t, 4H), 6.56 (d, 2H)

[258] MS (ESI): $[\text{M}+\text{H}]^+$ 785

[259] Example 7. Synthesis of Compound 12

[260]



[261] In a 100 ml round-bottom three-neck flask in a nitrogen atmosphere, 2.5 g of Intermediate 9 synthesized in Preparation Example 9, 0.8 g of di-p-tolylamine, 0.1 g of tris(dibenzylideneacetone)dipalladium(0), 0.2 g of 15% tris(t-butyl)phosphine, 1.1 g of t-butoxy sodium and 50 ml of toluene were placed, and stirred at 80°C for 2 hr. The reaction solution was cooled, filtered with silica gel, concentrated and then subjected to column chromatography using a solvent mixture of dichloromethane and n-hexane, thus obtaining 2.6 g of Compound 12 (yield 90%).

[262] ^1H NMR (CDCl_3 , 600 MHz) δ 8.11 (d, 2H), 7.95 (d, 1H), 7.87 (s, 1H), 7.53-7.50 (m,

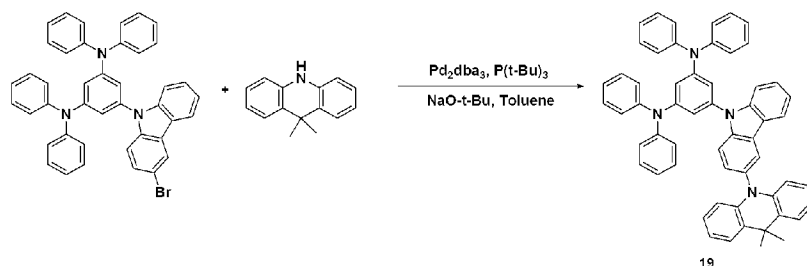
3H), 7.46 (d, 1H), 7.43-7.36 (m, 3H), 7.28-7.19 (m, 15H), 7.13 (d, 4H), 7.11 (d, 4H), 6.96 (t, 2H), 2.30 (s, 6H)

[263] MS (ESI): $[M+H]^+$ 771

[264] Tg: 134°C

[265] Example 8. Synthesis of Compound 19

[266]



[267] In a 100 ml round-bottom three-neck flask in a nitrogen atmosphere, 3.1 g of Intermediate 16 synthesized in Preparation Example 16, 1.0 g of 9,9-dimethyl-9,10-dihydroacridine, 0.1 g of tris(dibenzylideneacetone)dipalladium(0), 0.2 g of 15% tris(t-butyl)phosphine, 1.4 g of t-butoxy sodium and 50 ml of toluene were placed, and stirred at 80°C for 2 hr. The reaction solution was cooled, filtered with silica gel, concentrated and then subjected to column chromatography using a solvent mixture of dichloromethane and n-hexane, thus obtaining 3.1 g of Compound 19 (yield 85%).

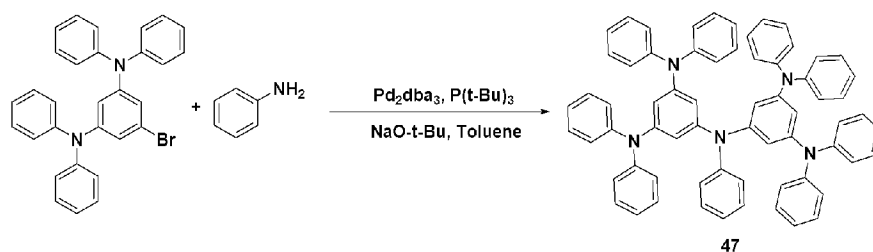
[268] ^1H NMR (CDCl_3 , 600 MHz) δ 8.00 (s, 1H), 7.97 (d, 1H), 7.60 (d, 1H), 7.47-7.44 (m, 3H), 7.41 (t, 1H), 7.27-7.21 (m, 10H), 7.18 (d, 8H), 7.00 (t, 4H), 6.91-6.88 (m, 5H), 6.84 (d, 2H), 6.28-6.27 (m, 2H), 1.72 (s, 6H)

[269] MS (ESI): $[M+H]^+$ 785

[270] Tg: 122°C

[271] Example 9. Synthesis of Compound 47

[272]



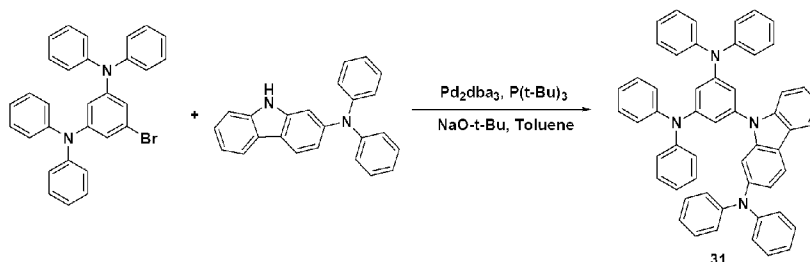
[273] In a 100 ml round-bottom three-neck flask in a nitrogen atmosphere, 4.9 g of Intermediate 17 synthesized in Preparation Example 17, 0.5 g of aniline, 0.3 g of tris(dibenzylideneacetone)dipalladium(0), 0.4 g of 15% tris(t-butyl)phosphine, 2.9 g of t-butoxy sodium and 50 ml of toluene were placed, and stirred at 80°C for 2 hr. The reaction solution was cooled, filtered with silica gel, concentrated and then subjected to column chromatography using a solvent mixture of dichloromethane and n-hexane, thus obtaining 4.3 g of Compound 47 (yield 93%).

[274] ^1H NMR (CDCl_3 , 600 MHz) δ 7.16 (t, 16H), 7.05 (t, 2H), 6.98 (d, 16H), 6.95-6.90 (m, 10H), 6.78 (t, 1H), 6.37-6.35 (m, 6H)

[275] MS (ESI): $[\text{M}+\text{H}]^+$ 914

[276] Example 10. Synthesis of Compound 31

[277]



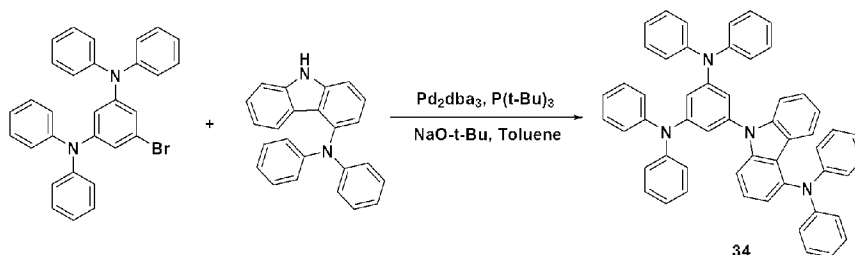
[278] In a 100 ml round-bottom three-neck flask in a nitrogen atmosphere, 1.8 g of Intermediate 17 synthesized in Preparation Example 17, 1.2 g of Intermediate 20 synthesized in Preparation Example 20, 0.1 g of tris(dibenzylideneacetone)dipalladium(0), 0.1 g of 15% tris(t-butyl)phosphine, 1.0 g of t-butoxy sodium and 20 ml of toluene were placed, and stirred at 80°C for 3 hr. The reaction solution was cooled, filtered with silica gel, concentrated and then subjected to column chromatography using a solvent mixture of dichloromethane and n-hexane, thus obtaining 2.1 g of Compound 31 (yield 79%).

[279] ^1H NMR (CDCl_3 , 600 MHz) δ 7.94 (d, 1H), 7.89 (d, 1H), 7.31-7.27 (m, 2H), 7.23-7.15 (m, 14H), 7.08-7.06 (m, 12H), 6.99-6.94 (m, 7H), 6.76 (s, 1H), 6.65 (s, 2H)

[280] MS (ESI): $[\text{M}+\text{H}]^+$ 745

[281] Example 11. Synthesis of Compound 34

[282]



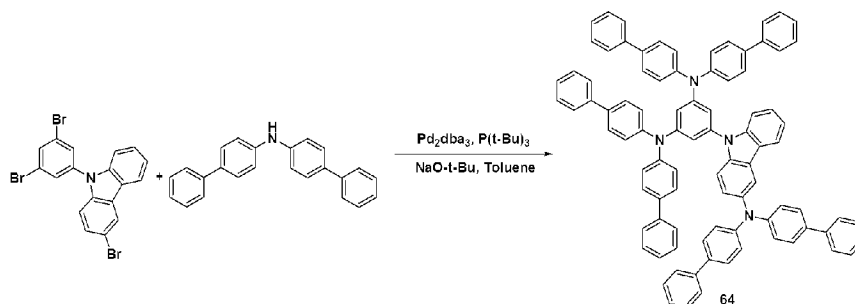
[283] In a 100 ml round-bottom three-neck flask in a nitrogen atmosphere, 2.4 g of Intermediate 17 synthesized in Preparation Example 17, 1.6 g of Intermediate 23 synthesized in Preparation Example 23, 0.1 g of tris(dibenzylideneacetone)dipalladium(0), 0.2 g of 15% tris(t-butyl)phosphine, 1.4 g of t-butoxy sodium and 25 ml of toluene were placed, and stirred at 80°C for 3 hr. The reaction solution was cooled, filtered with silica gel, concentrated and then subjected to column chromatography using a solvent mixture of dichloromethane and n-hexane, thus obtaining 2.9 g of Compound 34 (yield 81%).

[284] ^1H NMR (CDCl_3 , 600 MHz) δ 7.69 (d, 1H), 7.32 (t, 2H), 7.29 (d, 1H), 7.27-7.23 (m, 10H), 7.17-7.14 (m, 12H), 7.07 (d, 4H), 6.99 (t, 4H), 6.94-6.88 (m, 4H), 6.80 (d, 2H)

[285] MS (ESI): [M+H]⁺ 745

[286] Example 12. Synthesis of Compound 64

[287]



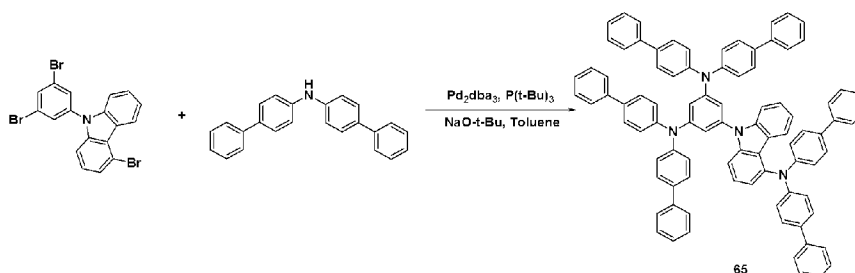
[288] In a 100 ml round-bottom three-neck flask in a nitrogen atmosphere, 1.5 g of Intermediate 24 synthesized in Preparation Example 24, 3.0 g of bisbiphenylamine, 0.2 g of tris(dibenzylideneacetone)dipalladium(0), 0.2 g of 15% tris(t-butyl)phosphine, 1.4 g of t-butoxy sodium and 20 ml of toluene were placed, and stirred at 80°C for 2 hr. The reaction solution was cooled, filtered with silica gel, concentrated and then subjected to column chromatography using a solvent mixture of dichloromethane and n-hexane, thus obtaining 3.0 g of Compound 64 (yield 79%).

[289] ¹H NMR (CDCl₃, 600 MHz) δ 7.95 (s, 1H), 7.92 (d, 1H), 7.56 (d, 4H), 7.51-7.45 (m, 22H), 7.42-7.39 (m, 5H), 7.36 (t, 8H), 7.31-7.25 (m, 15H), 7.19-7.18 (m, 5H), 6.98 (s, 1H), 6.96 (s, 2H)

[290] MS (ESI): [M+H]⁺ 1202

[291] Example 13. Synthesis of Compound 65

[292]



[293] In a 100 ml round-bottom three-neck flask in a nitrogen atmosphere, 1.5 g of Intermediate 25 synthesized in Preparation Example 25, 3.0 g of bisbiphenylamine, 0.2 g of tris(dibenzylideneacetone)dipalladium, 0.2 g of 15% tris(t-butyl)phosphine, 1.4 g of t-butoxy sodium and 20 ml of toluene were placed, and stirred at 80°C for 2 hr. The reaction solution was cooled, filtered with silica gel, concentrated and then subjected to column chromatography using a solvent mixture of dichloromethane and n-hexane, thus obtaining 3.2 g of Compound 65 (yield 85%).

[294] ¹H NMR (CDCl₃, 600 MHz) δ 7.80 (d, 1H), 7.53-7.50 (m, 20H), 7.45-7.41 (m, 7H), 7.38-7.35 (m, 12H), 7.32-7.29 (m, 13H), 7.27-7.24 (m, 3H), 7.19 (d, 4H), 7.05 (d, 1H), 7.00 (s, 1H), 6.98 (s, 2H)

[295] MS (ESI): [M+H]⁺ 1202

- [296] Device Example 1. Manufacture of organic EL device including Compound 1 as hole transport layer
- [297] A glass substrate coated with an ITO (indium tin oxide) thin film having a thickness of 100 nm was ultrasonically washed with an isopropyl alcohol solvent, dried, placed in a plasma cleaning system so that the substrate was cleaned using oxygen plasma for 5 min, and then transferred into a vacuum deposition system.
- [298] The ITO transparent electrode thus prepared was used as an anode, and DNTPD [N,N'-diphenyl-N,N'-bis-[4-(phenyl-m-tolylamino)-phenyl]-biphenyl-4,4'-diamine] was vacuum deposited on the ITO substrate, thus forming a hole injection layer having a thickness of 55 nm. Subsequently, Compound 1 was vacuum deposited to a thickness of 30 nm, thus forming a hole transport layer, and CBP [4,4-N,N-dicarbazolebiphenyl] as a host and 6 vol% of Ir(PPy)₃ [tris(2-phenylpyridine)iridium] as a dopant were vacuum deposited to a thickness of 25 nm on the hole transport layer, thus forming a light emitting layer.
- [299] Thereafter, an electron transport layer was formed to a thickness of 30 nm using Bphen [4,7-diphenyl-1,10-phenanthroline] on the light emitting layer. 2 nm thick Liq [lithium quinolate] and 100 nm thick Al were sequentially vacuum deposited on the electron transport layer to form a cathode, thereby manufacturing an organic EL device.
- [300] Device Example 2. Manufacture of organic EL device including Compound 5 as hole transport layer
- [301] An organic EL device was manufactured in the same manner as in Device Example 1, with the exception that Compound 5 was used instead of Compound 1.
- [302] Device Example 3. Manufacture of organic EL device including Compound 12 as hole transport layer
- [303] An organic EL device was manufactured in the same manner as in Device Example 1, with the exception that Compound 12 was used instead of Compound 1.
- [304] Device Example 4. Manufacture of organic EL device including Compound 19 as hole transport layer
- [305] An organic EL device was manufactured in the same manner as in Device Example 1, with the exception that Compound 19 was used instead of Compound 1.
- [306] Device Example 5. Manufacture of organic EL device including Compound 34 as hole transport layer
- [307] An organic EL device was manufactured in the same manner as in Device Example 1, with the exception that Compound 34 was used instead of Compound 1.
- [308] Device Example 6. Manufacture of organic EL device including Compound 47 as hole transport layer
- [309] An organic EL device was manufactured in the same manner as in Device Example

- 1, with the exception that Compound 47 was used instead of Compound 1.
- [310] Device Example 7. Manufacture of organic EL device including Compound 64 as hole transport layer
- [311] An organic EL device was manufactured in the same manner as in Device Example 1, with the exception that Compound 64 was used instead of Compound 1.
- [312] Device Example 8. Manufacture of organic EL device including Compound 65 as hole transport layer
- [313] An organic EL device was manufactured in the same manner as in Device Example 1, with the exception that Compound 65 was used instead of Compound 1.
- [314] Comparative Device Example 1. Manufacture of organic EL device including NPB as hole transport layer
- [315] An organic EL device was manufactured in the same manner as in Device Example 1, with the exception that NPB was used instead of Compound 1.
- [316] Device Example 9. Manufacture of organic EL device including Compound 1 as second hole transport layer
- [317] A glass substrate coated with an ITO thin film having a thickness of 100 nm was ultrasonically washed with an isopropyl alcohol solvent, dried, placed in a plasma cleaning system so that the substrate was cleaned using oxygen plasma for 5 min, and then transferred into a vacuum deposition system.
- [318] The ITO transparent electrode thus prepared was used as an anode, and DNTPD [N,N'-diphenyl-N,N'-bis-[4-(phenyl-m-tolylamino)-phenyl]-biphenyl-4,4'-diamine] was vacuum deposited on the ITO substrate, thus forming a hole injection layer having a thickness of 55 nm. Subsequently, NPB [N,N'-di(naphthalen-1-yl)-N,N'-diphenyl-benzidine] was vacuum deposited to a thickness of 20 nm, thus forming a first hole transport layer, and a second hole transport layer was formed to a thickness of 10 nm using Compound 1 on the first hole transport layer. CBP [4,4-N,N-dicarbazolebiphenyl] as a host and 6 vol% of Ir(PPy)₃ [tris(2-phenylpyridine)iridium] as a dopant were vacuum deposited to a thickness of 25 nm on the second hole transport layer, thus forming a light emitting layer.
- [319] Thereafter, an electron transport layer was formed to a thickness of 30 nm using Bphen [4,7-diphenyl-1,10-phenanthroline] on the light emitting layer. 2 nm thick Liq [lithium quinolate] and 100 nm thick Al were sequentially vacuum deposited on the electron transport layer to form a cathode, thereby manufacturing an organic EL device.
- [320] Device Example 10. Manufacture of organic EL device including Compound 5 as second hole transport layer
- [321] An organic EL device was manufactured in the same manner as in Device Example 9, with the exception that Compound 5 was used instead of Compound 1.

[322] Device Example 11. Manufacture of organic EL device including Compound 12 as second hole transport layer

[323] An organic EL device was manufactured in the same manner as in Device Example 9, with the exception that Compound 12 was used instead of Compound 1.

[324] Device Example 12. Manufacture of organic EL device including Compound 19 as second hole transport layer

[325] An organic EL device was manufactured in the same manner as in Device Example 9, with the exception that Compound 19 was used instead of Compound 1.

[326] Device Example 13. Manufacture of organic EL device including Compound 34 as second hole transport layer

[327] An organic EL device was manufactured in the same manner as in Device Example 9, with the exception that Compound 34 was used instead of Compound 1.

[328] Device Example 14. Manufacture of organic EL device including Compound 47 as second hole transport layer

[329] An organic EL device was manufactured in the same manner as in Device Example 9, with the exception that Compound 47 was used instead of Compound 1.

[330] Device Example 15. Manufacture of organic EL device including Compound 64 as second hole transport layer

[331] An organic EL device was manufactured in the same manner as in Device Example 9, with the exception that Compound 64 was used instead of Compound 1.

[332] Device Example 16. Manufacture of organic EL device including Compound 65 as second hole transport layer

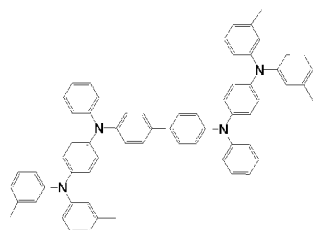
[333] An organic EL device was manufactured in the same manner as in Device Example 9, with the exception that Compound 65 was used instead of Compound 1.

[334] Comparative Device Example 2. Manufacture of organic EL device including NPB as second hole transport layer

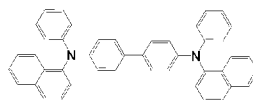
[335] An organic EL device was manufactured in the same manner as in Device Example 9, with the exception that NPB was used instead of Compound 1.

[336] The chemical formulas of DNTPD, NPB, CBP and Bphen used in the examples are represented below.

[337]

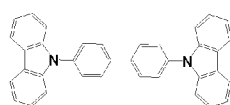


DNTPD

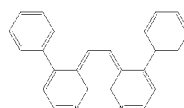


NPB

[338]



CBP



Bphen

[339] Test Example: Evaluation of properties of organic EL device

[340] The properties of the devices of Device Examples 1 to 14 and Comparative Device Examples 1 and 2 were evaluated at a brightness of 1000 cd/m². The results are shown in Tables 1 and 2 below.

[341] Current density

[342] In the manufactured organic EL devices, while a voltage was increased from 0 V to 10 V, current of each unit device was measured using a current-voltage meter (Keithley 2635A Source Meter), and the measured current value was divided by the area, thus obtaining current density.

[343] Brightness efficiency

[344] In the manufactured organic EL devices, while a voltage was increased from 0 V to 10 V, the brightness was measured using a brightness meter (Minolta CS-2000), and the measured brightness value was divided by the current value, thus obtaining brightness efficiency.

[345] Color coordinates

[346] The color coordinates were measured using a brightness meter (Minolta CS-2000).

[347] Table 1

[Table 1]

	Hole transport layer material	Current density(mA/cm ²)	Brightness efficiency (cd/A)	Color coordinates CIE (x,y)
Device Ex.1	Compound 1	6.8	31.3	0.32,0.62
Device Ex.2	Compound 5	7.5	29.0	0.32,0.61
Device Ex.3	Compound 12	7.5	29.3	0.32,0.61
Device Ex.4	Compound 19	7.5	29.4	0.32,0.61
Device Ex.5	Compound 34	6.9	31.0	0.32,0.61
Device Ex.6	Compound 47	7.2	29.9	0.32,0.61
Device Ex.7	Compound 64	7.7	27.8	0.32,0.61
Device Ex.8	Compound 65	7.6	28.2	0.32,0.62
Comp. Device Ex.1	NPB	7.8	27.5	0.31,0.61

[348] Table 2

[Table 2]

	2 nd Hole transport layer material	Current density(mA/cm ²)	Brightness efficiency(cd/A)	Color coordinates CIE (x,y)
Device Ex.9	Compound 1	7.5	28.6	0.31,0.62
Device Ex.10	Compound 5	7.8	27.2	0.31,0.62
Device Ex.11	Compound 12	8.3	25.8	0.32,0.61
Device Ex.12	Compound 19	7.8	27.6	0.32,0.61
Device Ex.13	Compound 34	7.6	28.4	0.32,0.61
Device Ex.14	Compound 47	8.0	26.9	0.32,0.61
Device Ex.15	Compound 64	8.3	25.7	0.32,0.61
Device Ex.16	Compound 65	8.1	26.5	0.32,0.62
Comp.Device Ex.2	NPB	8.4	25.5	0.31,0.61

[349] As is apparent from the results of manufacturing organic EL devices using the compounds according to the present invention as the material for the hole transport layer or the second hole transport layer, all of the devices exhibited superior properties compared to when using NPB as a conventional material.

[350] Although the preferred embodiments of the present invention have been disclosed for illustrative purposes, those skilled in the art will appreciate that various modifications, additions and substitutions are possible, without departing from the scope and spirit of the invention as disclosed in the accompanying claims.

Industrial Applicability

[351] According to an embodiment of the present invention, a compound including one or more derivatives having a substituted phenyl group is formed, making it possible to provide a variety of derivatives, whereby the properties of the compound can be adjusted. When this compound is contained in the organic layer of an organic EL device, various properties can be exhibited. Moreover, thermal stability and light emission efficiency of the organic EL device using the compound can be improved, and the use of the compound as a hole transport layer material enables a triplet exciton of a phosphorescent light emitting material to be confined, consequently improving the efficiency of the organic EL device.

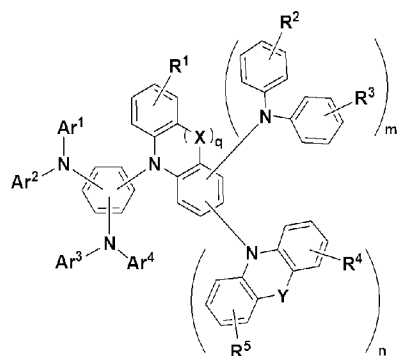
[352]

Claims

[Claim 1]

A compound for an organic electroluminescent device, represented by Chemical Formula 1 below:

[Chemical Formula 1]



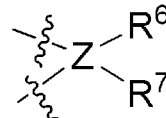
wherein Ar¹ to Ar⁴ are identical to or different from each other, and Ar¹ to Ar⁴ are each independently a substituted or unsubstituted C1 to C30 alkyl group, a substituted or unsubstituted C3 to C30 cycloalkyl group, a substituted or unsubstituted C1 to C30 heterocycloalkyl group, a substituted or unsubstituted C6 to C30 aryl group, or a substituted or unsubstituted C1 to C30 heteroaryl group, or Ar¹ and Ar², and Ar³ and Ar⁴, respectively, are linked to form a substituted or unsubstituted C1 to C30 heterocycloalkyl group, or a substituted or unsubstituted C1 to C30 heteroaryl group, together with a nitrogen atom therebetween,

R¹ is a hydrogen atom, a substituted or unsubstituted C1 to C30 alkyl group, a substituted or unsubstituted C3 to C30 cycloalkyl group, a substituted or unsubstituted C1 to C30 heterocycloalkyl group, a substituted or unsubstituted C6 to C30 aryl group, or a substituted or unsubstituted C1 to C30 heteroaryl group,

R² to R⁵ are identical to or different from each other, and R² to R⁵ are each independently a hydrogen atom, a substituted or unsubstituted C1 to C30 alkyl group, a substituted or unsubstituted C3 to C30 cycloalkyl group, a substituted or unsubstituted C1 to C30 heterocycloalkyl group, a substituted or unsubstituted C6 to C30 aryl group, or a substituted or unsubstituted C1 to C30 heteroaryl group, or at least one of R² to R⁵ is further coupled with a carbon atom adjacent to a carbon atom linked therewith to form a substituted or unsubstituted fused C3 to C30 cycloalkyl group, a substituted or unsubstituted fused C1 to C30 heterocycloalkyl group, a substituted or unsubstituted fused C6 to C30 aryl group, or a substituted or unsubstituted fused C1 to C30 heteroaryl group,

group,

X and Y are identical to or different from each other, and X and Y are each independently a valence bond or



carbon atom or a silicon atom, and R^6 and R^7 are identical to or different from each other, and R^6 and R^7 are each independently a hydrogen atom, a substituted or unsubstituted C1 to C30 alkyl group, a substituted or unsubstituted C3 to C30 cycloalkyl group, a substituted or unsubstituted C1 to C30 heterocycloalkyl group, a substituted or unsubstituted C6 to C30 aryl group, or a substituted or unsubstituted C1 to C30 heteroaryl group,

q is 0 or 1,

m is 0 or 1,

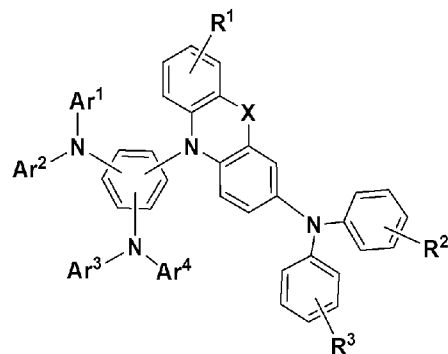
n is 0 or 1, and

$m+n \neq 0$.

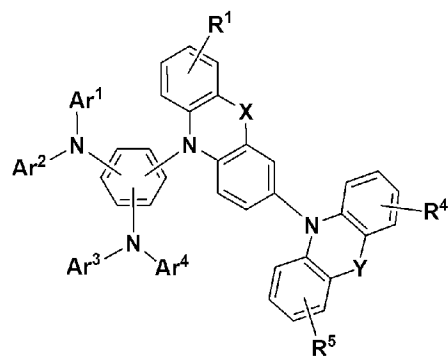
[Claim 2]

The compound of claim 1, which is represented by either of Chemical Formulas 2 and 3 below:

[Chemical Formula 2]



[Chemical Formula 3]



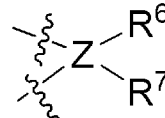
wherein Ar^1 to Ar^4 are identical to or different from each other, and Ar^1 to Ar^4 are each independently a substituted or unsubstituted C1 to C30

alkyl group, a substituted or unsubstituted C3 to C30 cycloalkyl group, a substituted or unsubstituted C1 to C30 heterocycloalkyl group, a substituted or unsubstituted C6 to C30 aryl group, or a substituted or unsubstituted C1 to C30 heteroaryl group, or Ar¹ and Ar², and Ar³ and Ar⁴, respectively, are linked to form a substituted or unsubstituted C1 to C30 heterocycloalkyl group, or a substituted or unsubstituted C1 to C30 heteroaryl group, together with a nitrogen atom therebetween,

R¹ is a hydrogen atom, a substituted or unsubstituted C1 to C30 alkyl group, a substituted or unsubstituted C3 to C30 cycloalkyl group, a substituted or unsubstituted C1 to C30 heterocycloalkyl group, a substituted or unsubstituted C6 to C30 aryl group, or a substituted or unsubstituted C1 to C30 heteroaryl group,

R² to R⁵ are identical to or different from each other, and R² to R⁵ are each independently a hydrogen atom, a substituted or unsubstituted C1 to C30 alkyl group, a substituted or unsubstituted C3 to C30 cycloalkyl group, a substituted or unsubstituted C1 to C30 heterocycloalkyl group, a substituted or unsubstituted C6 to C30 aryl group, or a substituted or unsubstituted C1 to C30 heteroaryl group, or at least one of R² to R⁵ is further coupled with a carbon atom adjacent to a carbon atom linked therewith to form a substituted or unsubstituted fused C3 to C30 cycloalkyl group, a substituted or unsubstituted fused C1 to C30 heterocycloalkyl group, a substituted or unsubstituted fused C6 to C30 aryl group, or a substituted or unsubstituted fused C1 to C30 heteroaryl group, and

X and Y are identical to or different from each other, and X and Y are each independently a valence bond or

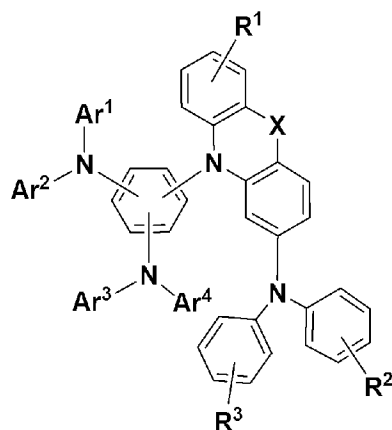


carbon atom or a silicon atom, and R⁶ and R⁷ are identical to or different from each other, and R⁶ and R⁷ are each independently a hydrogen atom, a substituted or unsubstituted C1 to C30 alkyl group, a substituted or unsubstituted C3 to C30 cycloalkyl group, a substituted or unsubstituted C1 to C30 heterocycloalkyl group, a substituted or unsubstituted C6 to C30 aryl group, or a substituted or unsubstituted C1 to C30 heteroaryl group.

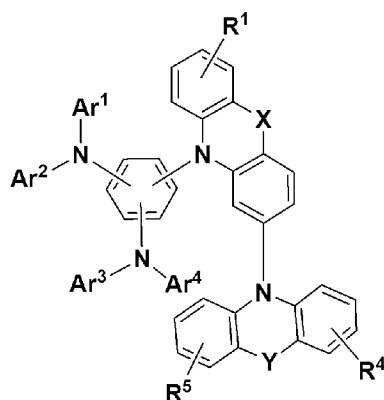
[Claim 3]

The compound of claim 1, which is represented by either of Chemical Formulas 4 and 5 below:

[Chemical Formula 4]



[Chemical Formula 5]



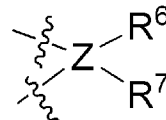
wherein Ar¹ to Ar⁴ are identical to or different from each other, and Ar¹ to Ar⁴ are each independently a substituted or unsubstituted C1 to C30 alkyl group, a substituted or unsubstituted C3 to C30 cycloalkyl group, a substituted or unsubstituted C1 to C30 heterocycloalkyl group, a substituted or unsubstituted C6 to C30 aryl group, or a substituted or unsubstituted C1 to C30 heteroaryl group, or Ar¹ and Ar², and Ar³ and Ar⁴, respectively, are linked to form a substituted or unsubstituted C1 to C30 heterocycloalkyl group, or a substituted or unsubstituted C1 to C30 heteroaryl group, together with a nitrogen atom therebetween,

R¹ is a hydrogen atom, a substituted or unsubstituted C1 to C30 alkyl group, a substituted or unsubstituted C3 to C30 cycloalkyl group, a substituted or unsubstituted C1 to C30 heterocycloalkyl group, a substituted or unsubstituted C6 to C30 aryl group, or a substituted or unsubstituted C1 to C30 heteroaryl group,

R² to R⁵ are identical to or different from each other, and R² to R⁵ are each independently a hydrogen atom, a substituted or unsubstituted C1 to C30 alkyl group, a substituted or unsubstituted C3 to C30 cycloalkyl group, a substituted or unsubstituted C1 to C30 heterocycloalkyl group,

a substituted or unsubstituted C6 to C30 aryl group, or a substituted or unsubstituted C1 to C30 heteroaryl group, or at least one of R² to R⁵ is further coupled with a carbon atom adjacent to a carbon atom linked therewith to form a substituted or unsubstituted fused C3 to C30 cycloalkyl group, a substituted or unsubstituted fused C1 to C30 heterocycloalkyl group, a substituted or unsubstituted fused C6 to C30 aryl group, or a substituted or unsubstituted fused C1 to C30 heteroaryl group, and

X and Y are identical to or different from each other, and X and Y are each independently a valence bond or

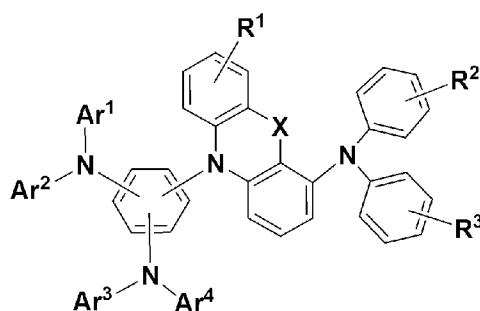


carbon atom or a silicon atom, and R⁶ and R⁷ are identical to or different from each other, and R⁶ and R⁷ are each independently a hydrogen atom, a substituted or unsubstituted C1 to C30 alkyl group, a substituted or unsubstituted C3 to C30 cycloalkyl group, a substituted or unsubstituted C1 to C30 heterocycloalkyl group, a substituted or unsubstituted C6 to C30 aryl group, or a substituted or unsubstituted C1 to C30 heteroaryl group.

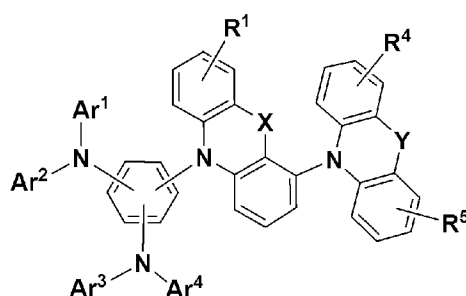
[Claim 4]

The compound of claim 1, which is represented by either of Chemical Formulas 6 and 7 below:

[Chemical Formula 6]



[Chemical Formula 7]



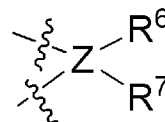
wherein Ar¹ to Ar⁴ are identical to or different from each other, and Ar¹

to Ar⁴ are each independently a substituted or unsubstituted C1 to C30 alkyl group, a substituted or unsubstituted C3 to C30 cycloalkyl group, a substituted or unsubstituted C1 to C30 heterocycloalkyl group, a substituted or unsubstituted C6 to C30 aryl group, or a substituted or unsubstituted C1 to C30 heteroaryl group, or Ar¹ and Ar², and Ar³ and Ar⁴, respectively, are linked to form a substituted or unsubstituted C1 to C30 heterocycloalkyl group, or a substituted or unsubstituted C1 to C30 heteroaryl group, together with a nitrogen atom therebetween,

R¹ is a hydrogen atom, a substituted or unsubstituted C1 to C30 alkyl group, a substituted or unsubstituted C3 to C30 cycloalkyl group, a substituted or unsubstituted C1 to C30 heterocycloalkyl group, a substituted or unsubstituted C6 to C30 aryl group, or a substituted or unsubstituted C1 to C30 heteroaryl group,

R² to R⁵ are identical to or different from each other, and R² to R⁵ are each independently a hydrogen atom, a substituted or unsubstituted C1 to C30 alkyl group, a substituted or unsubstituted C3 to C30 cycloalkyl group, a substituted or unsubstituted C1 to C30 heterocycloalkyl group, a substituted or unsubstituted C6 to C30 aryl group, or a substituted or unsubstituted C1 to C30 heteroaryl group, or at least one of R² to R⁵ is further coupled with a carbon atom adjacent to a carbon atom linked therewith to form a substituted or unsubstituted fused C3 to C30 cycloalkyl group, a substituted or unsubstituted fused C1 to C30 heterocycloalkyl group, a substituted or unsubstituted fused C6 to C30 aryl group, or a substituted or unsubstituted fused C1 to C30 heteroaryl group, and

X and Y are identical to or different from each other, and X and Y are each independently a valence bond or

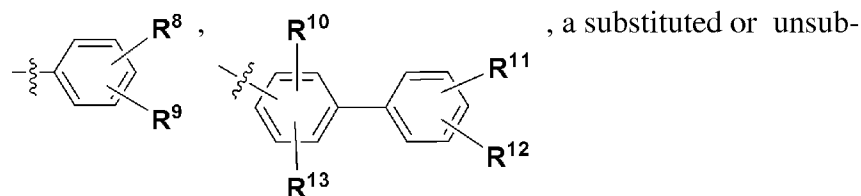


carbon atom or a silicon atom, and R⁶ and R⁷ are identical to or different from each other, and R⁶ and R⁷ are each independently a hydrogen atom, a substituted or unsubstituted C1 to C30 alkyl group, a substituted or unsubstituted C3 to C30 cycloalkyl group, a substituted or unsubstituted C1 to C30 heterocycloalkyl group, a substituted or unsubstituted C6 to C30 aryl group, or a substituted or unsubstituted C1 to C30 heteroaryl group.

[Claim 5]

The compound of claim 1, wherein Ar¹ to Ar⁴ are identical to or

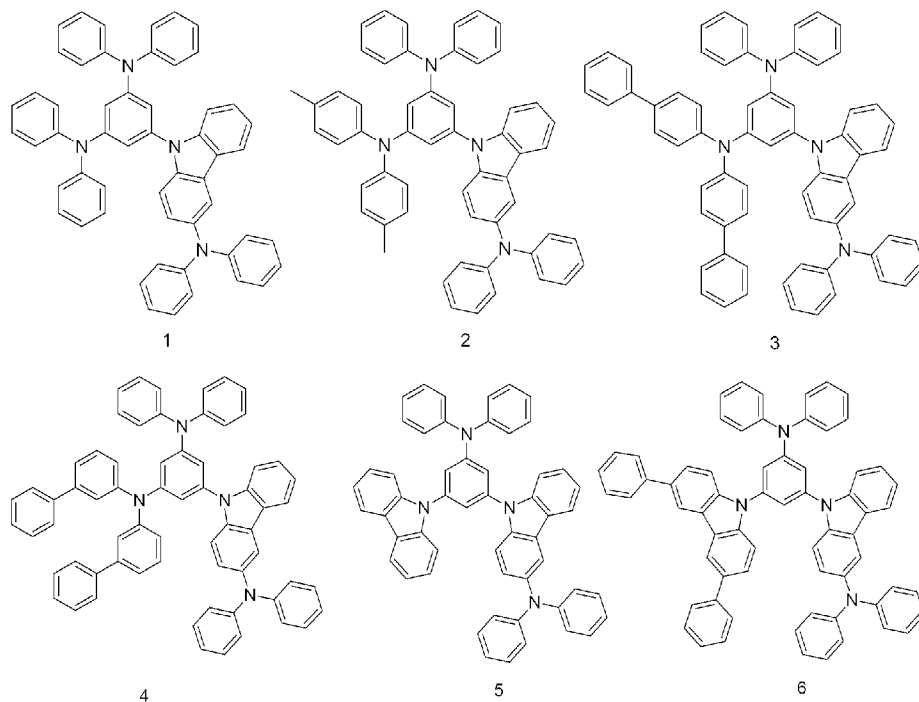
different from each other, and Ar¹ to Ar⁴ are each independently

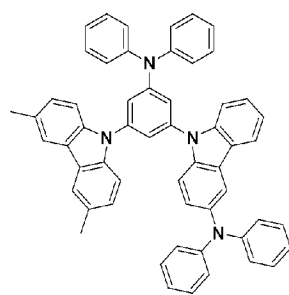


stituted C1 to C30 alkyl group, a substituted or unsubstituted C3 to C30 cycloalkyl group, or a substituted or unsubstituted C1 to C30 heterocycloalkyl group, or Ar¹ and Ar², and Ar³ and Ar⁴, respectively, are linked to form a substituted or unsubstituted C1 to C30 heterocycloalkyl group, or a substituted or unsubstituted C1 to C30 heteroaryl group, together with a nitrogen atom therebetween, wherein R⁸ to R¹³ are identical to or different from each other, and R⁸ to R¹³ are each independently a hydrogen atom, a substituted or unsubstituted C1 to C30 alkyl group, a substituted or unsubstituted C3 to C30 cycloalkyl group, a substituted or unsubstituted C1 to C30 heterocycloalkyl group, a substituted or unsubstituted C6 to C30 aryl group, or a substituted or unsubstituted C1 to C30 heteroaryl group.

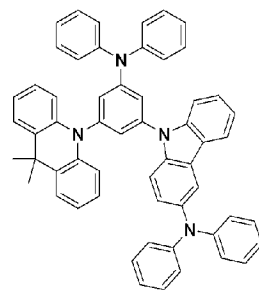
[Claim 6]

The compound of claim 1, which is any one selected from among Compounds 1 to 65 represented by the following chemical formulas:

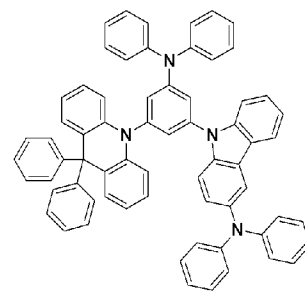




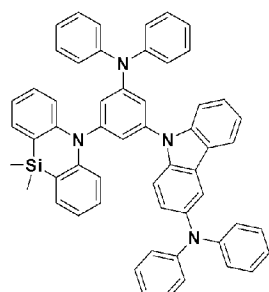
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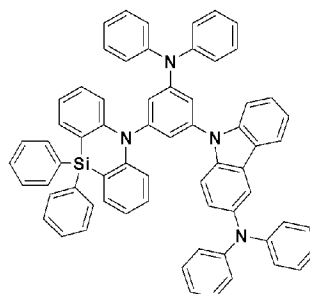
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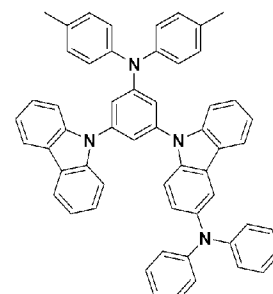
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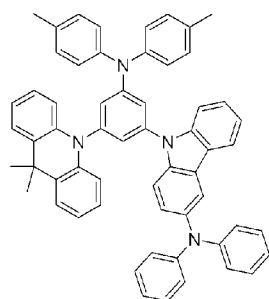
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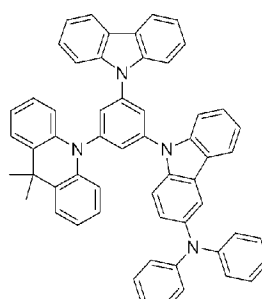
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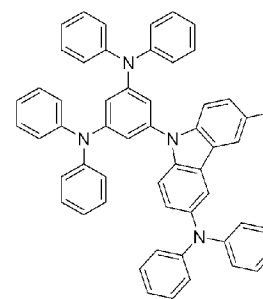
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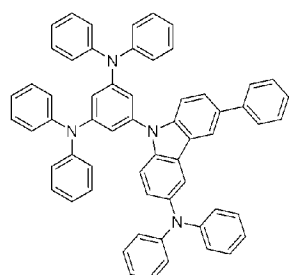
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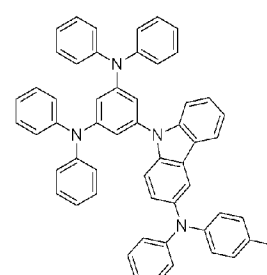
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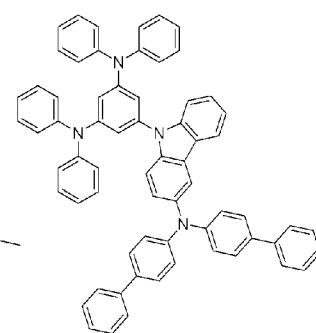
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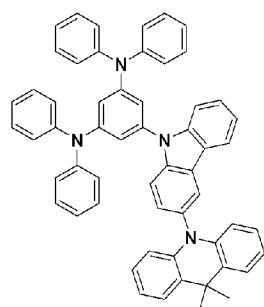
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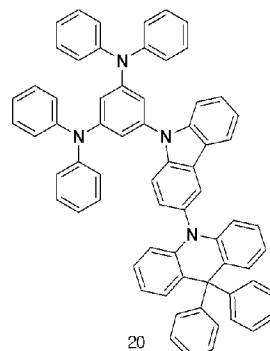
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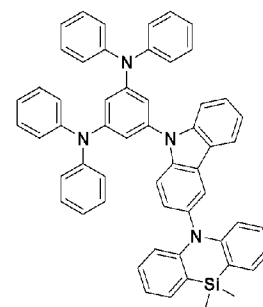
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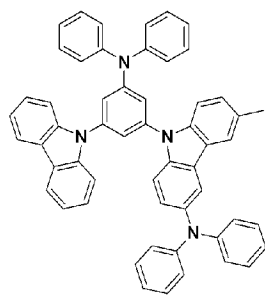
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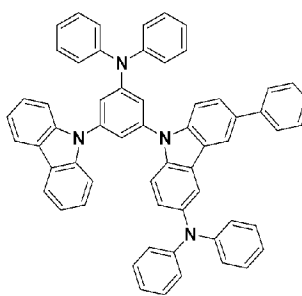
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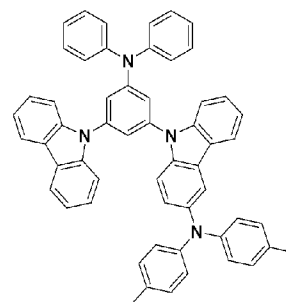
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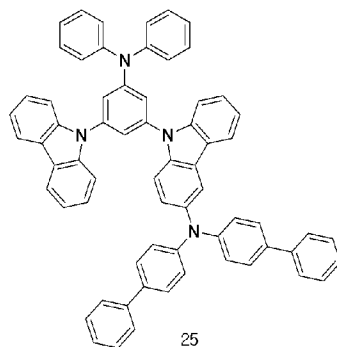
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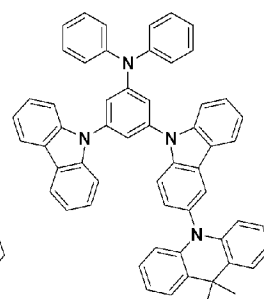
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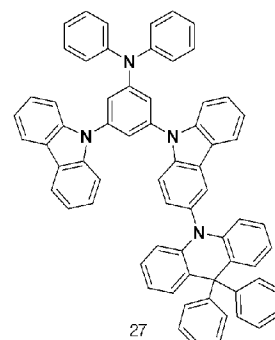
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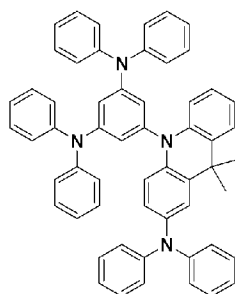
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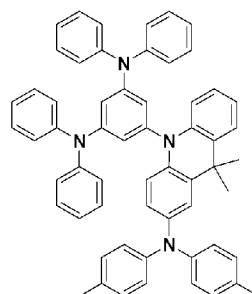
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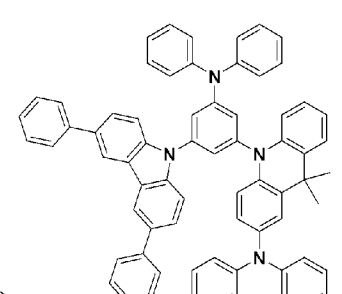
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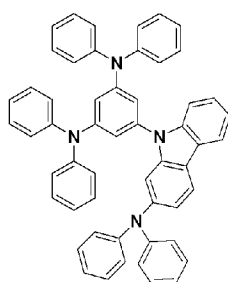
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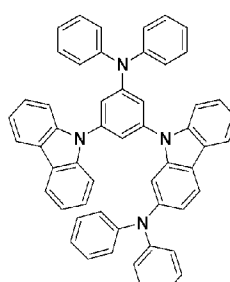
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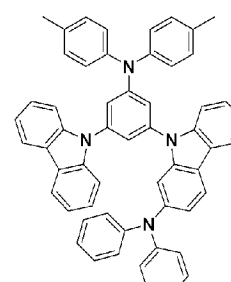
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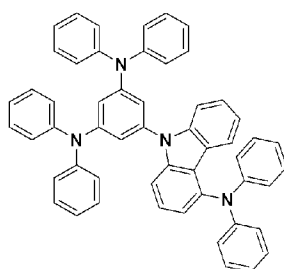
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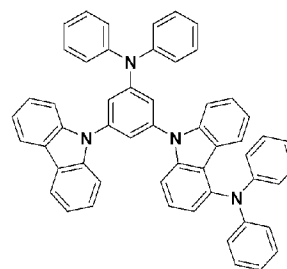
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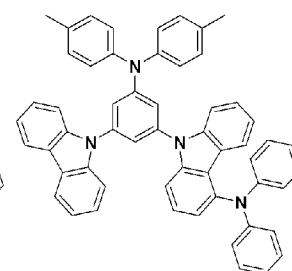
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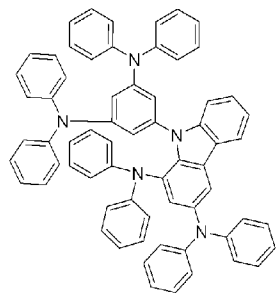
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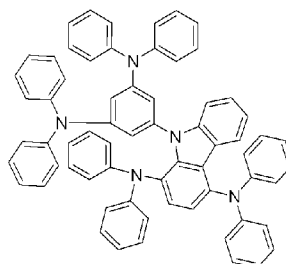
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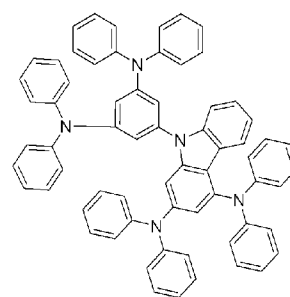
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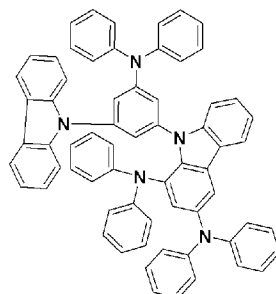
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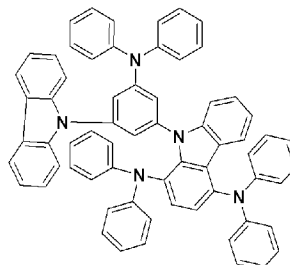
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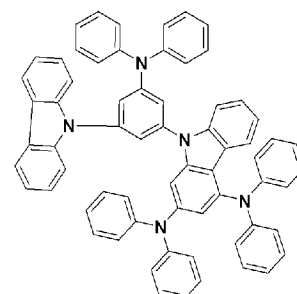
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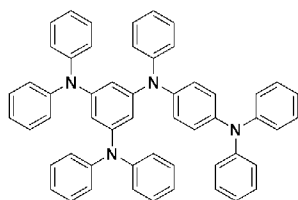
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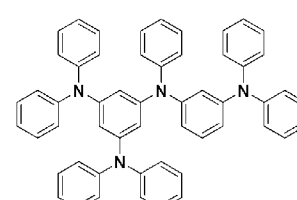
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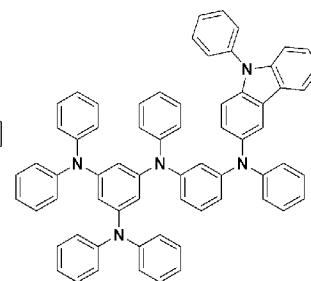
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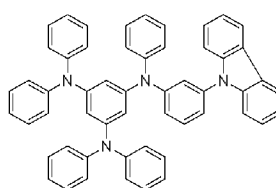
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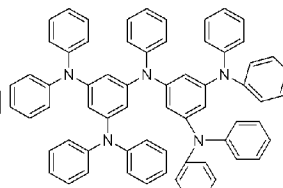
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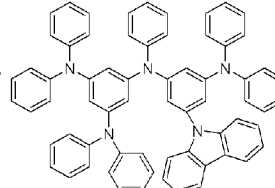
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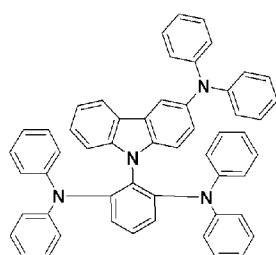
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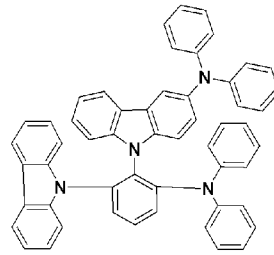
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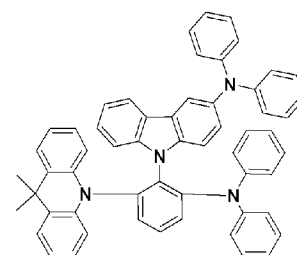
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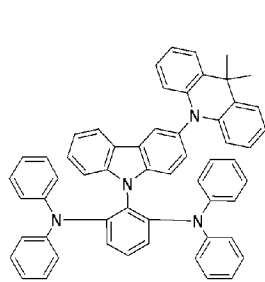
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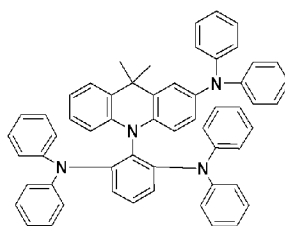
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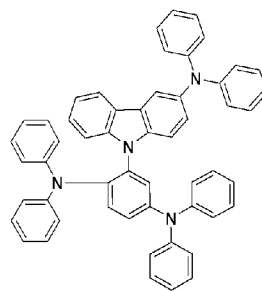
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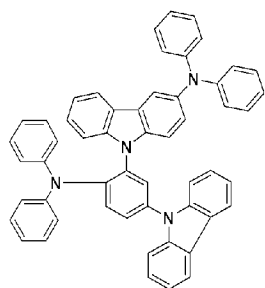
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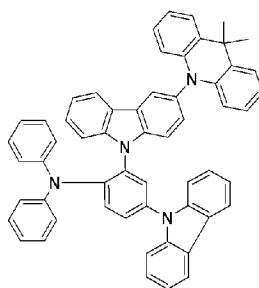
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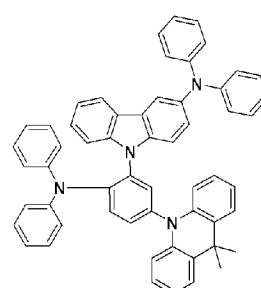
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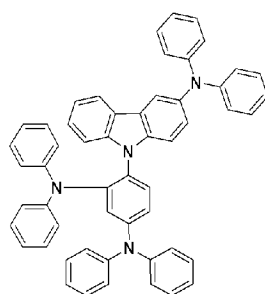
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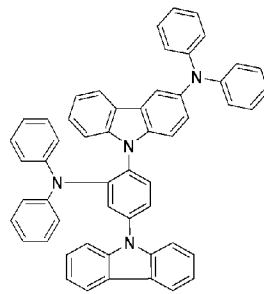
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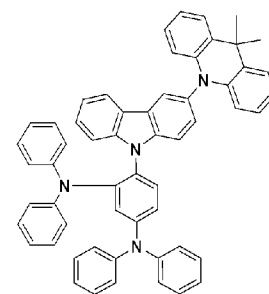
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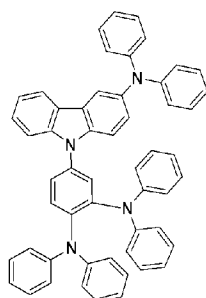
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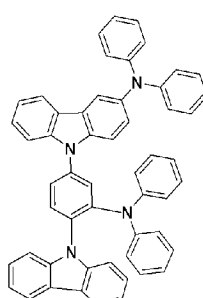
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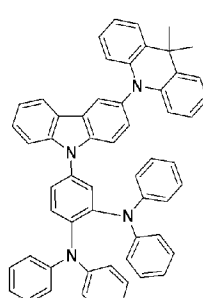
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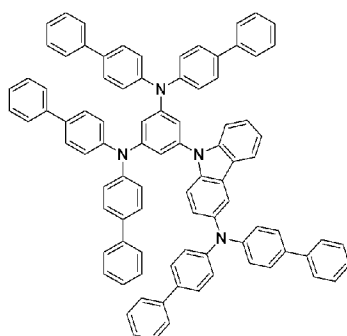
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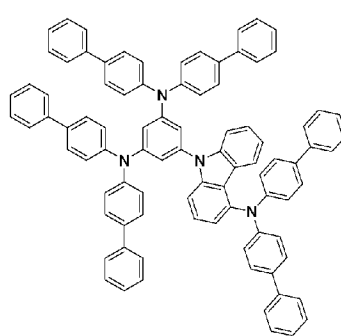
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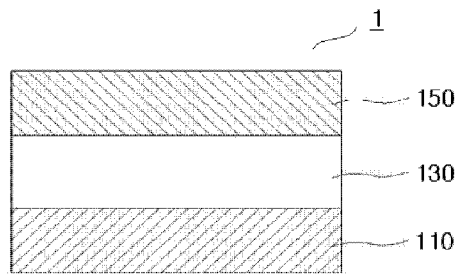
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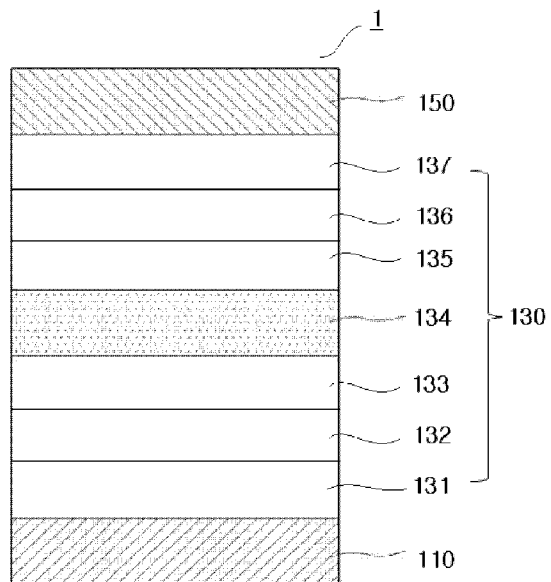
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- [Claim 7] An organic electroluminescent device, including the compound of claim 1.
- [Claim 8] An organic electroluminescent device, comprising a first electrode, a second electrode, and a single organic layer or a plurality of organic layers between the first electrode and the second electrode, wherein one or more organic layers selected from among the single organic layer or the plurality of organic layers include the compound of claim 1.
- [Claim 9] The organic electroluminescent device of claim 8, wherein the single organic layer or the plurality of organic layers include a light emitting layer.
- [Claim 10] The organic electroluminescent device of claim 8, wherein the plurality of organic layers include a light emitting layer, and the plurality of organic layers further include one or more selected from among an electron injection layer, an electron transport layer, a hole blocking layer, an electron blocking layer, a hole transport layer and a hole injection layer.
- [Claim 11] The organic electroluminescent device of claim 9, wherein the light emitting layer includes a host and a dopant.

[Fig. 1]



[Fig. 2]



INTERNATIONAL SEARCH REPORT

International application No.
PCT/KR2013/006439**A. CLASSIFICATION OF SUBJECT MATTER****C09K 11/06(2006.01)i, C07D 209/82(2006.01)i, H01L 51/50(2006.01)i**

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

C09K 11/06; C07D 403/14; H01L 51/54; C07D 209/82; H01L 51/50

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Korean utility models and applications for utility models

Japanese utility models and applications for utility models

Electronic data base consulted during the international search (name of data base and, where practicable, search terms used)

eKOMPASS(KIPO internal) & Keywords: organic electroluminescent device, OLED, carbazole.

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	KR 10-2006-0044424 A (LG CHEM.LTD. 16 May 2006 See claims 1-11.	1-11
A	WO 2011-049325 A2 (CHEIL INDUSTRIES INC.) 28 April 2011 See claims 1-15.	1-11
A	KR 10-2007-0114562 A (SAMSUNG SDI CO., LTD.) 04 December 2007 See claims 1-32.	1-11
A	KR 10-2011-0023090 A (SAMSUNG MOBILE DISPLAY CO., LTD.) 08 March 2011 See claims 1-19.	1-11



Further documents are listed in the continuation of Box C.



See patent family annex.

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"P" document published prior to the international filing date but later than the priority date claimed

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"Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art

"&" document member of the same patent family

Date of the actual completion of the international search

31 October 2013 (31.10.2013)

Date of mailing of the international search report

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INTERNATIONAL SEARCH REPORT

Information on patent family members

International application No.

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