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(54) Title: A RUBBER COMPOSITION

(57) Abstract: A rubber composition, which can be used in particular as tread(s) for winter tire(s) capable of rolling over ground surfaces covered with ice or black ice, while at least maintaining or even improving a grip on wet, comprising a diene elastomer, between 50 and 150 phr of a reinforcing filler, between 20 and 160 phr of a plasticizer comprising more than 10 phr and less than 80 phr of a vegetable oil, between 2 and 40 phr of microparticles of a water soluble sulfate of an alkali metal or an alkaline earth metal.



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## **DESCRIPTION**

### **Title of Invention**

### **A RUBBER COMPOSITION**

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#### **1. FIELD OF THE INVENTION**

[0001] The invention relates to rubber compositions which can be used in particular as tread(s) for “winter tire(s)” capable of rolling over ground surfaces covered with ice or black ice without being provided with studs (also known as studless tires), while  
10 at least maintaining or even improving a grip on wet.

[0002] It relates more particularly to treads for winter tires specifically suited to rolling under “melting ice” conditions encountered within a temperature range typically of between -5°C and 0°C.

[0003] It should specifically be remembered that, within such a range, the pressure of the  
15 tires during the passage of a vehicle brings about surface melting of the ice, which is covered with a thin film of water harmful to a grip of these tires on the melting ice.

#### **2. BACKGROUND**

[0004] Patent applications WO 2010/0009850 and WO 2012/052331 filed by the applicants  
20 describe a specific rubber composition capable of generating an effective surface microroughness by virtues of specific microparticles and which makes it possible to improve the grip on ice of the treads and tires comprising them under melting ice conditions without being disadvantageous to the properties of reinforcement and hysteresis.

[0005] These microparticles, which protrude at the surface of the tread, perform a claw function well known without a disadvantage of being abrasive action on the surfacing of the ground surface itself and without a significantly deteriorated road behavior on a dry ground surface. Then, subsequently, after gradual expulsion from the rubber matrix, they release microcavities which act as storage volume and as channel for draining the film of water at the surface of the ice; under these conditions, the contact between the surface of the tread and the ice is no longer lubricated and the coefficient of friction is thus improved.

[0006] There is a constant objective of tire manufacturers to moreover improve the grip on ice, especially under melting ice, without however deteriorating a grip on wet which is one of essential tire performances.

### 3. BRIEF DESCRIPTION OF THE INVENTION

[0007] On continuing their research, the inventors have discovered a novel rubber composition that comprises a specific combination of plasticizer and water soluble sulfate particles which makes it possible to improve the grip on ice of the treads and tires comprising them under melting ice conditions, while at least maintaining or even improving the wet grip, that is the grip on wet roads.

[0008] Thus, a first subject matter of the invention is a rubber composition comprising at least, a diene elastomer, between 50 and 150 phr of a reinforcing filler, between 20 and 160 phr of a plasticizer comprising more than at 10 phr and less than 80 phr of a vegetable oil, and between 2 and 40 phr of microparticles of a water soluble sulfate of an alkali metal or an alkaline earth metal.

[0009] Moreover, aspects of the present invention can be as follows.

[1] A rubber composition comprising at least:

- a diene elastomer;
- between 50 and 150 phr of a reinforcing filler;
- between 20 and 160 phr of a plasticizer comprising more than 10 phr and  
5 less than 80 phr of a vegetable oil; and
- between 2 and 40 phr of microparticles of a water soluble sulfate of an  
alkali metal or an alkaline earth metal.

[2] The rubber composition according to [1], wherein the content of the vegetable oil is more than 10 phr and less than 70 phr.

10 [3] The rubber composition according to [1] or [2], wherein the vegetable oil is chosen from the group consisting of linseed oil, safflower oil, soybean oil, corn oil, cottonseed oil, turnip seed oil, castor oil, tung oil, pine oil, sunflower oil, palm oil, olive oil, coconut oil, groundnut oil, grapeseed oil, and mixtures thereof.

[4] The rubber composition according to [3], wherein the vegetable oil is sunflower  
15 oil.

[5] The rubber composition according to any one of [1] to [4], wherein the fatty acid from which the vegetable oil derives, comprises oleic acid in a weight fraction greater than or equal to 60%.

[6] The rubber composition according to any one of [1] to [5], wherein the  
20 microparticles have a median particle size by volume and a width of a volume particle size distribution, both of which are measured by laser diffraction methods in accordance with ISO standard 13320-1 and satisfy the following relations:

- $50\ \mu\text{m} < D_{50} < 150\ \mu\text{m}$ ;
- $0.50 < \text{Span} < 1.50$ ;

wherein:

- $D_{50}$  is the medium particle size by volume corresponding to 50 % of a cumulative distribution obtained from the volume particle size distribution;
- $\text{Span} = (D_{90} - D_{10}) / D_{50}$ ; and
- $D_{10}$  and  $D_{90}$  are the particle size corresponding to 10 volume % and to 90 volume %, respectively, of the cumulative particle distribution.

[7] The rubber composition according to any one of [1] to [6], wherein the content of the microparticles is between 2 and 30 phr.

[8] The rubber composition according to any one of [1] to [7], wherein the alkali metal or the alkaline earth metal is chosen from the group consisting of sodium, potassium, magnesium, calcium and mixtures thereof.

[9] The rubber composition according to any one of [1] to [8], wherein the water soluble sulfate is chosen from the group consisting of magnesium sulfate, potassium sulfate and mixtures thereof.

[10] The rubber composition according to any one of [1] to [9], wherein the water soluble sulfate is magnesium sulfate.

[11] The rubber composition according to any one of [1] to [10], wherein the diene elastomer is chosen from the group consisting of natural rubber, synthetic polyisoprenes, polybutadienes, butadiene copolymers, isoprene copolymers and the mixtures thereof.

[12] The rubber composition according to [11], wherein the rubber composition comprises more than 50 phr of natural rubber or of synthetic polyisoprene.

[13] The rubber composition according to [11], wherein the rubber composition comprises more than 50 phr of polybutadiene having a content of cis-1,4 bonds of greater than 90%.

[14] The rubber composition according to any one of [1] to [13], wherein the reinforcing filler comprises greater than 60 phr of carbon black.

[15] The rubber composition according to any one of [1] to [14], wherein the reinforcing filler comprises greater than 70 phr of a reinforcing inorganic filler.

[16] The rubber composition according to any one of [1] to [15], wherein the content of total reinforcing filler is between 60 and 120 phr.

[17] The rubber composition according to any one of [1] to [16], wherein the content of the plasticizer is between 30 and 120 phr.

[18] The rubber composition according to any one of [1] to [17], wherein the plasticizer further comprises a compound selected from the group consisting of hydrocarbon resins, liquid plasticizers other than the vegetable oil and the mixtures thereof.

[19] The rubber composition according to [18], wherein the hydrocarbon resin exhibits a T<sub>g</sub> greater than 20°C.

[20] The rubber composition according to [18] or [19], wherein the hydrocarbon resins are chosen from the group consisting of cyclopentadiene homopolymer or copolymer resins, dicyclopentadiene homopolymer or copolymer resins, terpene homopolymer or copolymer resins, C<sub>5</sub> fraction homopolymer or copolymer resins, C<sub>9</sub> fraction homopolymer or copolymer resins, alpha-methyl styrene homopolymer or copolymer resins, and the mixtures of these resins.

[21] The rubber composition according to any one of [18] to [20], wherein the content of hydrocarbon resin is between 3 and 60 phr.

[22] The rubber composition according to [18], wherein the liquid plasticizers other than the vegetable oils, are chosen from the group consisting of polyolefinic oils, naphthenic oils, paraffinic oils, Distillate Aromatic Extracts (DAE) oils, Medium  
5 Extracted Solvates (MES) oils, Treated Distillate Aromatic Extracts (TDAE) oils, Residual Aromatic Extracts (RAE) oils, Treated Residual Aromatic Extracts (TRAЕ) oils, Safety Residual Aromatic Extracts (SRAE) oils, mineral oils, ether plasticizers, ester plasticizers, phosphate plasticizers, sulphonate plasticizers and the mixtures  
10 thereof.

[23] The rubber composition according to [22], wherein the content of liquid plasticizer is less than 20 phr.

[24] Use of a composition in accordance with any one of [1] to [23] for the manufacture of treads for pneumatic tires.

15 [25] A tread for a pneumatic tire comprising a composition in accordance with any one of [1] to [23].

[26] A pneumatic tire comprising the tread according to [25].

[0010] Another subject matter of the invention is the use of such a rubber composition in the manufacture of treads for tires, whether the treads are intended for new tires or  
20 for the retreading of worn tires.

[0011] Another subject matter of the invention is these treads and these tires themselves when they comprise a rubber composition according to the invention.

[0012] The tires of the invention are particularly intended to equip passenger motor vehicles, including 4×4 (four-wheel drive) vehicles and SUV (Sport Utility Vehicles) vehicles,

two-wheel vehicles (in particular motorcycles), and also industrial vehicles in particular chosen from vans and heavy duty vehicles (i.e., underground, bus or heavy road transport vehicles (lorries, tractors, trailers), or off-road vehicles, such as agricultural vehicles or earthmoving equipment.

- 5   **[0013]** The invention and its advantages will be easily understood in the light of the description and implementation examples which follow.

#### 4. DETAILED DESCRIPTION OF THE INVENTION

- 10   **[0014]** In the present description, unless expressly indicated otherwise, all the percentages (%) shown are % by weight. Moreover, any interval of values denoted by the expression “between a and b” represents the range of values extending from greater than a to less than b (that is to say, limits a and b excluded) whereas any interval of values denoted by the expression “from a to b” means the range of values extending from a up to b (that is to say, including the strict limits a and b).

- 15   **[0015]** The rubber composition of the invention is based on at least a diene elastomer, a reinforcing filler, a plasticizer comprising a vegetable oil, and microparticles of a water soluble sulfate of an alkali metal or an alkaline earth metal components which are described in detail hereinafter.

- 20   **[0016]** The expression “based on” should be understood in the present application to mean a composition comprising the mixture and/or the product of the reaction of the various constituents used, some of the constituents being able or intended to react together, at least partly, during the various manufacturing phases of the composition, in particular during the vulcanization (curing).

**[0017]**



#### 4-1. Diene elastomer

A “diene” elastomer (or “rubber”, the two terms being considered to be synonymous) should be understood, in a known manner, to mean an (one or more is understood) elastomer resulting at least in part (i.e., a homopolymer or a copolymer) from diene monomers (monomers bearing two carbon-carbon double bonds which may or may not be conjugated).

[0018] These diene elastomers can be classified in a known way into two categories: those “essentially unsaturated” and those essentially saturated”. Butyl rubbers, such as, for example copolymers of dienes and of  $\alpha$ -olefins of EPDM type, come within the category of essentially saturated diene elastomers, having a content of units of diene origin which is low or very low, always less than 15% (mol %). In contrast, essentially unsaturated diene elastomer is understood to mean a diene elastomer resulting at least in part from conjugated diene monomers, having a content of units of diene origin (conjugated dienes) which is greater than 15% (mol %). In the category of “essentially unsaturated” diene elastomers, “highly unsaturated” diene elastomer is understood to mean in particular a diene elastomer having a content of units of diene origin (conjugated dienes) which is greater than 50%.

[0019] It is preferable to use at least one diene elastomer of the highly unsaturated type, in particular a diene elastomer chosen from the group consisting of polybutadienes (BR), synthetic polyisoprenes (IR), natural rubber (NR), butadiene copolymers, isoprene copolymers and mixtures of these elastomers. Such copolymers are more preferably chosen from the group consisting of butadiene/styrene copolymers (SBR), isoprene/butadiene copolymers (BIR), isoprene/styrene copolymers (SIR), isoprene/butadiene/styrene copolymers (SBIR) and mixtures of such copolymers.

[0020] The following are preferably suitable: polybutadienes, in particular those having a content of 1,2-units of between 4% and 80% or those having a content of cis-1,4 units of greater than 80%, polyisoprenes, butadiene/styrene copolymers in particular those having a styrene content of between 5% and 50% by weight and more particularly between 20% and 40%, a content of 1,2-bonds of the butadiene part of between 4% and 65% and a content of trans-1,4 bonds of between 20% and 80%, butadiene/isoprene copolymers, in particular those having an isoprene content of between 5% and 90% by weight and a glass transition temperature ("Tg"-measured according to ASTM D 3418 (1999)) of -40°C to -80°C, or isoprene/styrene copolymers, in particular those having a styrene content of between 5% and 50% by weight and a Tg of between -25°C and -50°C.

[0021] In the case of butadiene/styrene/isoprene copolymers, those having a styrene content of between 5% and 50% by weight and more particularly of between 10% and 40%, an isoprene content of between 15% and 60% by weight and more particularly between 20% and 50%, a butadiene content of between 5% and 50% by weight and more particularly of between 20% and 40%, a content of 1,2-units of the butadiene part of between 4% and 85%, a content of trans-1,4-units of the butadiene part of between 6% and 80%, a content of 1,2 plus 3,4-units of the isoprene part of between 5% and 70% and a content of trans-1,4-units of the isoprene part of between 10% and 50%, and more generally any butadiene/styrene/isoprene copolymer having a Tg of between -20°C and -70°C, are suitable in particular.

[0022] According to a particularly preferred embodiment of the invention, the diene elastomer is chosen from the group consisting of natural rubber, synthetic

polyisoprenes, polybutadienes having a content of cis-1,4 bonds of greater than 90%, butadiene/styrene copolymers and the mixtures thereof.

[0023] According to a more particular and preferred embodiment, the diene elastomer used is predominantly, that is to say for more than 50 phr (it should be remembered that  
5 “phr” means parts by weight per 100 parts of elastomer), of natural rubber (NR) or a synthetic polyisoprene (IR). More preferably, the said natural rubber or synthetic polyisoprene is then used as a blend with a polybutadiene (BR) having a content of cis-1,4 bonds which is preferably greater than 90%.

[0024] According to another particular and preferred embodiment, the diene elastomer used  
10 is predominantly, that is to say for more than 50 phr, a polybutadiene (BR) having a content of cis-1,4 bonds of greater than 90%. More preferably, said polybutadiene is then used as a blend with natural rubber or a synthetic polyisoprene.

[0025] According to another particular and preferred embodiment, the diene elastomer used is a binary blend (mixture) of NR (or IR) and of BR, or a ternary blend of NR (or IR),  
15 BR and SBR. Preferably in the case of such blends, the composition comprises between 25 and 75 phr of NR (or IR) and between 75 and 25 phr of BR, with which may or may not be associated a third elastomer (ternary blend) at a content of less than 30 phr, in particular of less than 20 phr. This third elastomer is preferably an SBR elastomer, in particular a solution SBR (“SSBR”). More preferably still, in the  
20 case of such a blend, the composition comprises from 35 to 65 phr of NR (or IR) and from 65 to 35 phr of BR. The BR used is preferably a BR having a content of cis-1,4 bonds of greater than 90%, more preferably of greater than 95%.

[0026] Synthetic elastomers other than diene elastomers, indeed even polymers other than elastomers, for example thermoplastic polymers, might be combined, in a minor amount, with the diene elastomers of the compositions of the invention.

[0027]

5 4-2. Reinforcing Filler

Use may be made of any type of reinforcing filler known for its capabilities of reinforcing a rubber composition which can be used for the manufacture of tires, for example an organic filler, such as carbon black, or a reinforcing inorganic filler, such as silica, with which a coupling agent is combined in a known way.

10 [0028] Such a reinforcing filler typically consists of nano particles, the mean size (by weight) of which is less than 500 nm, generally between 20 and 200 nm, in particular and preferably between 20 and 150 nm.

[0029] All carbon blacks, in particular blacks of the HAF, ISAF or SAF type, conventionally used in treads for tires ("tire-grade" blacks) are suitable as carbon blacks. Mention will more particularly be made, among the latter, of the reinforcing carbon blacks of the 100, 200 or 300 series (ASTM grades), such as, for example, the N115, N134, N234, N326, N330, N339, N347 or N375 blacks. The carbon blacks might, for example, be already incorporated in the isoprene elastomer in the form of a masterbatch (see, for example, Applications WO 97/36724 or W0 99/16600).

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[0030] Mention may be made, as examples of organic fillers other than carbon blacks, of the functionalized polyvinyl organic fillers as described in Applications WO 2006/069792, WO 2006/069793, WO2008/003434 and WO2008/003435.

[0031] The term “reinforcing inorganic filler” should be understood here as meaning any inorganic or mineral filler, whatever its colour and its origin (natural or synthetic), also known as “white filler” or sometimes “clear filler” in contrast to carbon black, capable of reinforcing by itself, without means other than an intermediate coupling agent, a rubber composition intended for the manufacture of tires, in other words capable of replacing, in its reinforcing role, a conventional tire-grade carbon black; such a filler is generally characterized, in a known way, by the presence of hydroxyl (-OH) groups at its surface.

[0032] Mineral fillers of the siliceous type, in particular silica ( $\text{SiO}_2$ ), or of the aluminous type, in particular alumina ( $\text{Al}_2\text{O}_3$ ), are suitable in particular as reinforcing inorganic fillers. The silica used can be any reinforcing silica known to a person skilled in the art, in particular any precipitated or pyrogenic silica exhibiting a BET surface and a CTAB specific surface both of less than  $450 \text{ m}^2/\text{g}$ , preferably from 30 to  $400 \text{ m}^2/\text{g}$ , in particular between 60 and  $300 \text{ m}^2/\text{g}$ . Mention will be made, as highly dispersible (“HD precipitated silicas”), for example, of the Ultrasil 7000 and Ultrasil 7005 silicas from Evonik, the Zeosil 1165 MP, 1135 MP and 1115 MP silicas from Rhodia, the Hi-Sil EZ150G silica from PPG, the Zeopol 8715, 8745 and 8755 silicas from Huber. Mention may be made, as examples of reinforcing aluminas, of the “Baikalox A125” or “Baikalox CR125” aluminas from Baikowski, the “APA-100RDX” alumina from Condea, the “Aluminoxid C” alumina from Degussa or the “AKP-G015” alumina from Sumitomo Chemicals.

[0033] Preferably, the content of total reinforcing filler (carbon black and/or reinforcing inorganic filler) is between 60 and 120 phr, in particular between 70 and 100 phr.

[0034] According to a specific embodiment, the reinforcing filler comprises predominantly carbon black; in such a case, the carbon black is present at a content preferably of greater than 60 phr, in combination or not with a reinforcing inorganic filler, such as silica, in a minor amount.

5 [0035] According to another specific embodiment, the reinforcing filler comprises predominantly an inorganic filler, in particular silica; in such a case, the inorganic filler, in particular silica, is present at a content preferably of greater than 70 phr, in combination or not with carbon black in a minor amount; the carbon black, when it is present, is preferably used at a content of less than 20 phr, more preferably less  
10 than 10 phr (for example between 0.1 and 10 phr).

[0036] Independently of the first aspect of the invention, namely the search for optimized grip on melting ice, the predominant use of a reinforcing inorganic filler, such as silica, is also advantageous from the viewpoint of the grip on a wet or snowy ground surface.

15 [0037] According to another possible embodiment of the invention, the reinforcing filler comprises a blend of carbon black and of reinforcing inorganic filler such as silica, in similar amounts; in such a case, the content of inorganic filler, in particular silica, and the content of carbon black are preferably each between 25 and 75 phr, more particularly each between 30 and 50 phr.

20 [0038] In order to couple the reinforcing inorganic filler to the diene elastomer, use is made, in a well-known way, of an at least bifunctional coupling agent (or bonding agent) intended to provide a satisfactory connection, of chemical and/or physical nature, between the inorganic filler (surface of its particles) and the diene elastomer. Use is made, in particular, of bifunctional organosilanes or polyorganosiloxanes.

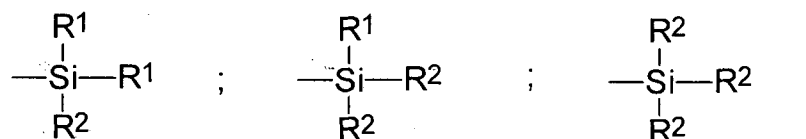
[0039] Use is made in particular of silane polysulphides, referred to as “symmetrical” or “asymmetrical” depending on their specific structure, as described, for example, in Applications WO 03/002648 (or US 2005/016651) and WO 03/002649 (or US 2005/016650). “Symmetrical” silane polysulphides corresponding to the following general formula (I):

(I)  $Z - A - S_x - A - Z$ , in which:

[0040] x is an integer from 2 to 8 (preferably from 2 to 5);

[0041] A is a divalent hydrocarbon radical (preferably,  $C_1$ - $C_{18}$  alkylene groups or  $C_6$ - $C_{12}$  arylene groups, more particularly  $C_1$ - $C_{10}$ , in particular  $C_1$ - $C_4$ , alkenes, especially propylene);

[0042] Z corresponds to one of the formulae below:



[0043] in which:

[0044] the  $R^1$  radicals, which are unsubstituted or substituted and identical to or different from one another, represent a  $C_1$ - $C_{18}$  alkyl,  $C_5$ - $C_{18}$  cycloalkyl or  $C_6$ - $C_{18}$  aryl group (preferably,  $C_1$ - $C_6$  alkyl, cyclohexyl or phenyl groups, in particular  $C_1$ - $C_4$  alkyl groups, more particularly methyl and/or ethyl),

[0045] the  $R^2$  radicals, which are unsubstituted or substituted and identical to or different from one another, represent a  $C_1$ - $C_{18}$  alkoxy or  $C_5$ - $C_{18}$  cycloalkoxy group (preferably a group chosen from  $C_1$ - $C_8$  alkoxy and  $C_5$ - $C_8$  cycloalkoxy, more preferably still a group chosen from  $C_1$ - $C_4$  alkoxy, in particular methoxy and ethoxy), are suitable in particular, without the above definition being limiting.

Mention will more particularly be made, as examples of silane polysulphides, of bis((C1-C4)alkoxyl(C1-C4)alkylsilyl(C1-C4)alkyl)polysulphides (in particular disulphides, trisulphides or tetrasulphides), such as, for example, bis(3-trimethoxysilylpropyl) or bis(3-triethoxysilylpropyl) polysulphides. Use is in particular made, among these compounds, of bis(3-triethoxysilylpropyl) tetrasulphide, abbreviated to TESPT, of formula  $[(C_2H_5O)_3Si(CH_2)_3S_2]_2$ , or bis(triethoxysilylpropyl)disulphide, abbreviated to TESPd, of formula  $[(C_2H_5O)_3Si(CH_2)_3S]_2$ . Mention will also be made, as preferred examples, of bis(mono(C1-C4)alkoxydi(C1-C4)alkylsilylpropyl) polysulphides (in particular disulphides, trisulphides or tetrasulphides), more particularly bis(monoethoxydimethylsilylpropyl) tetrasulphide, as described in Patent Application WO 02/083782 (or US 7 217 751).

[0046] Mention will in particular be made, as coupling agent other than alkoxysilane polysulphide, of bifunctional POSSs (polyorganosiloxanes) or of hydroxysilane polysulphides ( $R^2 = OH$  in the above formula (I)), such as described in Patent Applications WO 02/30939 (or US 6 774 255) and WO 02/31041 (or US 2004/051210), or of silanes or POSSs carrying azodicarbonyl functional groups, such as described, for example, in Patent Applications WO 2006/125532, WO 2006/125533 and WO 2006/125534.

[0047] In the rubber compositions in accordance with the invention, the content of coupling agent is preferably between 2 and 12 phr, more preferably between 3 and 8 phr.

[0048] A person skilled in the art will understand that a reinforcing filler of another nature, in particular organic nature, might be used as filler equivalent to the reinforcing inorganic filler described in the present section, provided that this reinforcing filler is



covered with an inorganic layer, such as silica, or else comprises, at its surface, functional sites, in particular hydroxyls, requiring the use of a coupling agent in order to form the connection between the filler and the elastomer.

[0049]

5 4-3. Plasticizer

The rubber composition according to the present invention has the other essential characteristic of comprising between 20 and 160 phr of a plasticizer comprising more than 10 phr and less than 80 phr of a vegetable oil. The vegetable oil is liquid at 20°C by definition.

10 [0050] The content of the essential plasticizer comprising the vegetable oil is preferably between 30 and 120 phr, more preferably between 40 and 100 phr, still more preferably between 50 and 80 phr, particularly from 55 to 70 phr. When the essential plasticizer is used in admixture with another essential plasticizer, the content should be understood to mean the total content of the two or more plasticizers.

15 [0051] Regarding the content of the vegetable (or plant) oil, below the indicated minimum, the targeted technical effect is insufficient. Wherein the indicated maximum, there is an issue of cost of the vegetable oil and a risk of worsening processability. For these reasons, the content of the vegetable oil is preferably between 10 and 70 phr, more preferably between 15 and 60 phr, still more preferably from 20 to 55 phr.

20 [0052] Mention may be made, as examples of the vegetable oil, of an oil selected from the group consisting of linseed oil, safflower oil, soybean oil, corn oil, cottonseed oil, turnip seed oil, castor oil, tung oil, pine oil, sunflower oil, palm oil, olive oil, coconut oil, groundnut oil and grapeseed oil, and mixtures of these oils. Sunflower oil is preferred.

[0053] The vegetable oil may be made up in large part of an unsaturated C<sub>18</sub> fatty acid, that is to say an unsaturated fatty acid chosen from the group consisting of oleic acid, linoleic acid, linolenic acid and the mixtures of these acids.

[0054] The vegetable oil is preferably rich in oleic acid, that is to say that the fatty acid (or all of the fatty acids if several are present) from which the vegetable oil derives, comprises oleic acid in a weight fraction at least equal to 60%, more preferably at least equal to 70%, still more preferably at least equal to 80% and particularly at least equal to 90%.

[0055] As the vegetable oil, use is advantageously made of a sunflower oil which is such that all of the fatty acids from which the vegetable oil derives comprise oleic acid in a weight fraction greater than or equal to 60%, preferably 70%, more preferably 80% and, according to one particularly advantageous embodiment of the invention, in a weight fraction greater than or equal to 90%.

[0056] T<sub>g</sub> of the vegetable oil may be by definition less than -20°C, preferably less than -40°C.

[0057] Preferably, the plasticizer further comprises a compound selected from the group consisting of liquid plasticizers other than the vegetable oil, hydrocarbon resins and the mixture thereof.

[0058] The liquid plasticizers other than the vegetable oil are liquid at 20°C by definition, their role is to soften the matrix by diluting the elastomer and the reinforcing filler; their T<sub>g</sub> is by definition less than -20°C, preferably less than -40°C.

[0059] Any extending oil, whether of aromatic or non-aromatic nature, any liquid plasticizing agent known for its plasticizing properties with regard to diene elastomers, can be used. At ambient temperature (20°C), these plasticizers or these

oils, which are more or less viscous, are liquids (that is to say, as a reminder, substances that have the ability to eventually take on the shape of their container), as opposed, in particular, to plasticizing hydrocarbon resins which are by nature solid at ambient temperature.

5   **[0060]** The liquid plasticizers other than the vegetable oil chosen from the group consisting of polyolefinic oils, naphthenic oils (low or high viscosity, in particular hydrogenated or otherwise), paraffinic oils, DAE (Distillate Aromatic Extracts) oils, MES (Medium Extracted Solvates) oils, TDAE oils (Treated Distillate Aromatic Extracts), RAE (Residual Aromatic Extracts) oils, TRAE (Treated Residual  
10   Aromatic Extracts) oils, SRAE (Safety Residual Aromatic Extracts) oils, mineral oils, ether plasticizers, ester plasticizers, phosphate plasticizers, sulphonate plasticizers and the mixtures of these compounds are particularly suitable.

**[0061]** Mention may be made, as phosphate plasticizers for example, of those that contain between 12 and 30 carbon atoms, for example trioctyl phosphate.

15   **[0062]** As examples of ester plasticizers, mention may especially be made of the compounds chosen from the group consisting of trimellitates, pyromellitates, phthalates, 1,2-cyclohexane dicarboxylates, adipates, azelates, sebacates, and mixtures thereof.

**[0063]** The content of liquid plasticizers other than the vegetable oil is preferably 0 phr or  
20   less than 20 phr, more preferably less than 15 phr, still more preferably less than 10 phr, particularly at most 5 phr.

**[0064]** The hydrocarbon resins, which are solid plasticizers (at 20°C), exhibit a T<sub>g</sub> of greater than +20°C, preferably greater than +30°C, such as is described, for example in Applications WO 2005/087859, WO 2006/061064 and WO 2007/017060.

[0065] The hydrocarbon resins are polymers well known to a person skilled in the art which are essentially based on carbon and hydrogen and thus miscible by nature in diene elastomer composition(s), when they are additionally described as being “plasticizing”. They have been described, for example, in the work entitled “Hydrocarbon Resins” by R. Mildenberg, M. Zander and G. Collin (New York, VCH, 1997, ISBN 3-527-28617-9), chapter 5 of which is devoted to their applications, in particular in the tire rubber field (5.5. “Rubber Tires and Mechanical Goods”). They can be aliphatic or aromatic or also of the aliphatic/aromatic type, that is to say based on aliphatic and/or aromatic monomers. They can be natural or synthetic and may or may not be petroleum-based (if such is the case, also known under the name of petroleum resins). They are preferably exclusively hydrocarbon, that is to say that they comprise only carbon and hydrogen atoms.

[0066] Preferably, the hydrocarbon resins as being “plasticizing” exhibit at least one, more preferably all, of the following characteristics:

[0067] a Tg of greater than 20°C (more preferably between 40 and 100°C;

[0068] a number-average molecular weight (Mn) of between 400 and 2000 g/mol (more preferably between 500 and 1500g/mol); and

[0069] a polydispersity index (PI) of less than 3, more preferably less than 2 (reminder:  $PI = M_w/M_n$  with  $M_w$  the weight-average molecular weight).

[0070] The Tg is measured in a known way by DSC (Differential Scanning Calorimetry) according to Standard ASTM D3418 (1999). The macrostructure ( $M_w$ ,  $M_n$  and PI) of the hydrocarbon resins are determined by steric exclusion chromatography (SEC): solvent tetrahydrofuran; temperature 35°C; concentration 1 g/l; flow rate 1 ml/min; solution filtered through a filter with a porosity of 0.45  $\mu m$  before injection; Moore

calibration with polystyrene standards; set of 3 "Waters" columns in series ("Styragel" HR4E, HR1 and HR0.5); detection by differential refractometer ("Waters 2410") and its associated operating software ("Waters Empower").

[0071] According to a particularly preferred embodiment, the hydrocarbon resins as being

5 "plasticizing" are chosen from the group consisting of cyclopentadiene (abbreviated to CPD) homopolymer or copolymer resins, dicyclopentadiene (abbreviated to DCPD) homopolymer or copolymer resins, terpene homopolymer or copolymer resins, C<sub>5</sub> fraction homopolymer or copolymer resins, C<sub>9</sub> fraction homopolymer or copolymer resins, alpha-methyl styrene homopolymer or copolymer resins and the  
10 mixtures thereof. Use is more preferably made, among the above copolymer resins, of those chosen from the group consisting of (D)CPD/vinylaromatic copolymer resins, (D)CPD/terpene copolymer resins, (D)CPD/C<sub>5</sub> fraction copolymer resins, (D)CPD/C<sub>9</sub> fraction copolymer resins, terpene/vinylaromatic copolymer resins, terpene/phenol copolymer resins, C<sub>5</sub> fraction/vinyl-aromatic copolymer resins, C<sub>9</sub>  
15 fraction/vinylaromatic copolymer resins, and the mixtures thereof.

[0072] The term "terpene" combines here, in a known way, the  $\alpha$ -pinene,  $\beta$ -pinene and limonene monomers; use is preferably made of a limonene monomer, which compound exists, in a known way, in the form of three possible isomers: L-limonene (laevorotatory enantiomer), D-limonene (dextrorotatory enantiomer) or else  
20 dipentene, the racemate of the dextrorotatory and laevorotatory enantiomers. Styrene,  $\alpha$ -methylstyrene, ortho-, meta- or para-methylstyrene, vinyltoluene, para-(tert-butyl)styrene, methoxystyrenes, chlorostyrenes, hydroxystyrenes vinylmesitylene, divinylbenzene, vinylnaphthalene, or any vinylaromatic monomer resulting from a C<sub>9</sub> fraction (or more generally from a C<sub>8</sub> to C<sub>10</sub> fraction) are suitable, for example, as

vinylaromatic monomer. Preferably, the vinylaromatic compound is styrene or a vinylaromatic monomer resulting from a C<sub>9</sub> fraction (or more generally from a C<sub>8</sub> to C<sub>10</sub> fraction). Preferably, the vinylaromatic compound is the minor monomer, expressed as molar fraction, in the copolymer under consideration.

- 5 [0073] The content of hydrocarbon resins is preferably between 3 and 60 phr, more preferably between 5 and 50 phr, still more preferably between 10 and 40 phr, particularly from 15 to 35 phr.

[0074]

#### 4-4. Water soluble Sulphate Microparticles

- 10 The rubber compositions of the invention have the other essential characteristic of comprising between 2 and 40 phr of microparticles of a water soluble sulphate of an alkali metal or an alkalie earth metal.

- [0075] Water solubility or solubility in water (it should be remembered that maximum weight of dissolved substance in water at given one temperature and one pressure) is one of basically well-known and constant physics (in particular, available in the "Handbooks of Chemistry and Physic") of the compounds which are organic or inorganic.

- [0076] The person skilled art in the art understands that the microparticles of the water soluble sulphate can dissolve at least partly in water. Preferably, the solubility of the microparticles in water at 0 °C under a pressure equal to 1 atm is higher than 1 g/100ml, more preferably higher than 2 g/100ml, still more preferably higher than 4 g/100ml.

[0077] According to a preferred embodiment, the alkali metal or the alkaline metal is chosen from the group consisting of sodium (Na), potassium (K), magnesium (Mg), calcium (Ca) and the mixture thereof.

[0078] According to a more preferred embodiment, the water soluble sulfate is chosen from the group consisting of magnesium sulfate, potassium sulfate and the mixture thereof. Still more preferably, the water soluble sulfate is magnesium sulfate.

[0079] According to a preferable embodiment, the microparticles have a specific median particle size by volume ( $D_{50}$ ) and a specific width of a volume particle size distribution (Span), both of which are measured by laser particle size methods in accordance with ISO standard 13320-1 and satisfy the following relations:

$$- 50 \mu\text{m} < D_{50} < 150 \mu\text{m};$$

$$- 0.50 < \text{Span} < 1.50;$$

[0080] The  $D_{50}$  is the median particle size (diameter) by volume corresponding to 50 (volume) % of a cumulative distribution obtained from the volume particle size distribution, that is to say, particle size by volume, 50% of the microparticles have smaller diameter(s) than  $D_{50}$ , and another 50% of the microparticles have larger diameter(s) than  $D_{50}$ . It is preferred that the  $D_{50}$  is between 75 and 125  $\mu\text{m}$  in order to optimally compromise between a desired surface roughness and a suitable contact between the rubber composition and the ice.

[0081] The Span which is the width of the volume particle size distribution is measured in a way known and is defined according to the following equation:

$$- \text{Span} = (D_{90} - D_{10}) / D_{50};$$

Preferably, the Span is between 0.75 and 1.25.

[0082] wherein:

$D_{10}$  and  $D_{90}$  are diameters (particle sizes) of the microparticles corresponding to 10 (volume) % and to 90 (volume)%, respectively, of the cumulative distribution.

[0083] According to another preferred embodiment, the  $D_{10}$  is greater than 30  $\mu\text{m}$ , more preferably greater than 40  $\mu\text{m}$ , in particular greater than 50  $\mu\text{m}$ .

5 [0084] According to another preferred embodiment combined or not with the previous embodiment, the  $D_{90}$  is less than 180  $\mu\text{m}$ , more preferably less than 160  $\mu\text{m}$ , in particular less than 150  $\mu\text{m}$ .

[0085] Moreover, in order to obtain an optimal compromise among the targeted technical effect (the creation of a suitable micro-roughness), the grip performance on melting  
10 ice and cost, the content of the microparticles in the rubber composition according to the present invention is preferably between 2 and 30 phr, more preferably between 5 and 20 phr.

[0086] The particle size distribution by volume (that is, volume particle distribution) is characterized in a known manner with a laser particle size analyzer in accordance  
15 with ISO 13320-1 (1999) unless expressly indicated otherwise.

[0087] A laser diffraction particle size analyzer, which is "Partica LA-950V2" equipped with its data processing software (P2000074001H version 4.11), provided by HORIBA Co., Ltd. is used. The programming parameters of the measurement are as follows: Use of the Fraunhofer model; no preliminary sample treatment with  
20 ultrasound; circulation velocity in the tank: 6 (corresponding to about 1400 revolutions / min); stirring speed in the cell: 6 (that is, about 2300 revolutions / min); red laser (650 nm); focal length from the object: 450 mm; preliminary alignment control of the laser and measurement of background noise (the tank with ethanol).



[0088] A suspension of the sample to be tested is prepared at first: for this purpose, about 500 mg of the sample is put in an empty beaker (volume about 100 ml), a few guttae of ethanol titrated at 95° is then utilized to moisturize the sample and to form a paste having the consistency of a toothpaste; finally, 40 ml of ethanol is added to this preparation which is then stirred with a bar magnet and a magnetic stirrer (magnetic stirrer IKA® “RET BASIC”, graduation 750) during about 15 seconds.

[0089] Then the suspension obtained is injected little by little, with a pipette, in the tank of the preparation of the particle size analyzer initially filled with 95° ethanol (low level, that is, about 180 ml), to an obscuration level of the measurement cell, displayed on the apparatus, between 15% and 20% (corresponding to the optimum concentration).

[0090] The measurement is performed just after stabilization of the obscuration level, with three successive measurements on the same preparation for a repeatability control. In order to validate this first measurement, a new preparation (suspension) is made and analyzed according to the same protocol: the two measurements must be identical (reproducibility in accordance with the recommendations of the standard). After each preparation and measurement, and the cell and the tank for the preparation in the laser particle size analyzer are rinsed twice with demineralized water.

[0091] The recorded data is ultimately used by the software that delivers the volume particle size distribution curve and the desired characteristics of the volume distribution ( $D_{10}$ ,  $D_{50}$ ,  $D_{90}$  and Span).

[0092]

#### 4-5. Various Additives

The rubber compositions of the invention also comprise all or a portion of the usual additives generally used in the elastomer compositions intended for the manufacture of treads for tires, in particular for winter tires, such as, for example, protection agents, such as antiozone waxes, chemical antiozonants, antioxidants, reinforcing resins, methylene acceptors (for example phenolic novolak resin) or methylene donors (for example HMT or H3M), a crosslinking system based either on sulphur or on donors of sulphur and/or per oxide and/or bismaleimides, vulcanization accelerators, or vulcanization activators.

[0093] These compositions can also comprise coupling activators when a coupling agent is used, agents for covering the inorganic filler or more generally processing aids capable, in a known way, by virtue of an improvement in the dispersion of the filler in the rubber matrix and of a lowering of the viscosity of the compositions, of improving their property of processing in the raw state; these agents are, for example, hydrolysable silanes, such as alkylalkoxysilanes, polyols, polyethers, amines, or hydroxylated or hydrolysable polyorganosiloxanes.

[0094]

#### 4-6. Manufactures of the Rubber Compositions and of the Treads

The rubber compositions of the invention are manufactured in appropriate mixers using two successive preparation phases according to a general procedure well known to a person skilled in the art: a first phase of thermomechanical working or kneading (sometimes described as “non-productive” phase) at high temperature, up to a maximum temperature of between 130°C and 200°C, preferably between 145°C and 185°C, followed by a second phase of mechanical working (sometimes

described as “productive” phase) at a lower temperature, typically of less than 120°C, for example between 60°C and 100°C, finishing phase during which the crosslinking or vulcanization system is incorporated.

[0095] A process which can be used for the manufacture of such compositions comprises,

5 for example and preferably, the following steps:

[0096] incorporating in the diene elastomer, in a mixer, the reinforcing filler, the plasticizer comprising the vegetable oil, the microparticles, during a first stage (“non productive” stage) everything being kneaded thermomechanically, in one or more goes, until a maximum temperature of between 130°C and 200°C is reached;

10 [0097] cooling the combined mixture to a temperature of less than 100°C;

[0098] subsequently incorporating, during a second stage (“productive” stage), a crosslinking system;

[0099] kneading everything up to a maximum temperature of less than 120°C;

[0100] extruding or calendering the rubber composition thus obtained, in particular in the  
15 form of a tire tread.

[0101] By way of example, the first (non-productive) phase is carried out in a single thermomechanical stage during which all the necessary constituents, the optional additional covering agents or processing aids, and various other additives, with the exception of the crosslinking system, are introduced into an appropriate mixer, such  
20 as a normal internal mixer. After cooling the mixture thus obtained during the first non-productive phase, the crosslinking system is then incorporated at low temperature, generally in an external mixer, such as an open mill; everything is then mixed (the second (productive) phase) for a few minutes, for example between 5 and 15 min.

[0102] The crosslinking system proper is preferably based on sulphur and on a primary vulcanization accelerator, in particular on an accelerator of sulphenamide type. Added to this vulcanization system are various known secondary accelerators or vulcanization activators, such as zinc oxide, stearic acid, guanidine derivatives (in particular diphenylguanidine), and the like, incorporated during the first non-productive phase and/or during the productive phase. The content of sulphur is preferably between 0.5 and 3.0 phr, and that of the primary accelerator is preferably between 0.5 and 5.0 phr.

[0103] Use may be made, as accelerator (primary or secondary) of any compound capable of acting as accelerator of the vulcanization of diene elastomers in the presence of sulphur, in particular accelerators of the thiazoles type and their derivatives, accelerators of thiurams types, or zinc dithiocarbamates. These accelerators are more preferably chosen from the group consisting of 2-mercaptobenzothiazyl disulphide (abbreviated to "MBTS"), N-cyclohexyl-2-benzothiazole sulphenamide (abbreviated to "CBS"), N,N-dicyclohexyl-2 benzothiazolesulphenamide ("DCBS"), N-ter't-butyl-2-benzothiazolesulphenamide ("TBBS"), N-tert-butyl-2 benzothiazolesulphenimide ("TBSI"), zinc dibenzylldithiocarbamate ("ZBEC"), Tetrabenzylthiuram disulfide ("TBZTD") and the mixtures thereof.

[0104] The final composition thus obtained is subsequently calendered, for example in the form of a sheet or of a plaque, in particular for laboratory characterization, or else extruded in the form of a rubber profiled element which can be used directly as winter tire tread.

[0105] The vulcanization (or curing) is carried out in a known way at a temperature generally of between 130°C and 200°C for a sufficient time which can vary, for

example, between 5 and 90 min depending in particular on the curing temperature, the vulcanization system adopted and the vulcanization kinetics of the composition under consideration.

[0106] The rubber compositions according to the invention can constitute all or a portion only of the tread in accordance with the invention, in the case of a tread of composite type formed from several rubber compositions of different formulations.

[0107] The invention relates to the rubber compositions and to the treads described above, both in the raw state (i.e., before curing) and in the cured state (i.e., after crosslinking or vulcanization).

[0108] The invention is further illustrated by the following non-limiting examples.

[0109]

## 5. EXAMPLES OF THE IMPLEMENTATION OF THE INVENTION

### 5.1 Preparation of the Rubber Compositions and of the Treads

[0110] The tests which follow were carried out in the following way: the reinforcing filler (for example a reinforcing inorganic filler, such as silica, and its associated coupling agent), the plasticizer, the microparticles, the diene elastomer (or blend of diene elastomers) and the various other ingredients, with the exception of the vulcanization system, were successively introduced into an internal mixer having an initial vessel temperature of approximately 60°C; the mixer was thus approximately 70% full (% by volume). Thermomechanical working (non-productive phase) was then carried out in one stage, which lasts in total approximately 3 to 4 minutes, until a maximum “dropping” temperature of 165°C was reached. The mixture thus obtained was recovered and cooled and then sulphur and an accelerator of sulphenamide type were

incorporated on an external mixer (homofinisher) at 30°C, everything being mixed (productive phase) for an appropriate time (for example between 5 and 12 min).

[0111] The compositions thus obtained were subsequently calendered, either in the form of sheets (thickness of 2 to 3 mm) or of fine sheets of rubber, for the measurement of their physical or mechanical properties, or in the form of profiled elements which could be used directly, after cutting and/or assembling to the desired dimensions, for example as tire semi-finished products, in particular as tire treads.

[0112]

## 5.2 Friction tests

In these tests, five compositions (identified as C-1, C-2, C-3, C-4 and C-5) based on diene elastomers (BR and NR blend comprising a content of cis-1,4 bonds of greater than 95%) are compared, the five compositions are reinforced with a blend of silica and carbon black, and comprise plasticizers with/without 55 or 20 phr an oleic sunflower oil (as a vegetable oil) which is or is not combined with a fraction (10 phr) of magnesium sulphate microparticles (as microparticles of a water soluble sulfate of an alkali metal or an alkaline earth metal), the microparticles obtained with a suitable filtering as shown in WO 2012/052331 A and having a specific volume distribution defined by D50, Span (also D10 and D90) measured by a laser diffraction method (with "Partica LA-950V2" equipped with its data processing software (P2000074001H version 4.11), provided by HORIBA Co., Ltd.) in accordance with ISO 13320-1:

Composition C-1: composition without the vegetable oil and without the microparticles (a reference);

Composition C-2: composition without the vegetable oil but with the microparticles  
(a comparative example);

Composition C-3: composition without the vegetable oil but with the microparticles  
(another comparative example);

5 Composition C-4: composition according to the present invention with 55 phr of the  
vegetable oil and with the microparticles;

Composition C-5: composition according to the present invention with 20 phr of the  
vegetable oil and with the microparticles;

[0113] The formulations of the five compositions (Table 1 - contents of the various products  
10 expressed in phr), and their ice friction results are given in Tables 1 and 2; the  
vulcanization system is composed of sulphur and sulphenamide.

[0114] These five compositions were subjected to a laboratory test consisting in measuring  
their friction coefficients on ice. The principle is based on a block of rubber  
composition that slides at a given speed (for example equal to 5 km/h) over an ice  
15 track with an imposed load (for example equal to 3 kg/cm<sup>2</sup>). The forces generated in  
the direction of travel (Fx) of the block and perpendicular to the travel (Fz) are  
measured. The Fx/Fz ratio determines the friction coefficient of the test specimen on  
the ice. The temperature during the measurement is set at -2°C. The surface of block  
of rubber composition in accordance with the invention, which contacted on the ice,  
20 had been cut a thin slice of the rubber by approximately 0.5mm before the friction  
test in order to expose the microparticles at the surface.

[0115] This test, the principle of which is well known to a person skilled in the art (see, for  
example, patent applications EP 1 052 270 and EP 1 505 112), makes it possible to  
evaluate, under representative conditions, the grip on the ice within the temperature

of range between 0°C and -5°C which would be obtained after a running test on a vehicle fitted with tires whose tread is composed of the same rubber compositions.

[0116] The results are expressed in Table 2. A value above that of the reference (composition C-1), arbitrarily set at 100, indicates an improved result, that is to say an aptitude for shorter braking distance, that is, the higher value, the better the ice grip. It is observed in this Table 2 that the compositions C-4 and C-5 in accordance with the present invention has a marked or certain increase in the friction coefficient at -2°C, relative to the reference composition C-1 and the comparative examples C-2 and C-3.

[0117]

### 5.3 Dynamic properties

Each of the above five compositions were tested in order to measure their dynamic properties on a viscosity (Metravib VA4000) according to Standard ASTM D5992-96. The response of a sample of vulcanized composition (cylindrical test specimen with a thickness of 4 mm and with a cross section of 400 m<sup>2</sup>), subjected to a simple alternating sinusoidal shear stress, at a frequency of 10 Hz, during a temperature sweep, under a stationary stress of 0.7MPa, was recorded; the  $\tan(\delta)$  value observed at 0°C was recorded. It should be remembered that, in a way well known to a person skilled in the art, a ratio of the  $\tan(\delta)_{0^\circ\text{C}}$  of the examples (or the comparative examples) to the reference, arbitrarily set at 100, is representative of the grip potential on wet roads: the higher ratio, the better wet grip than that of the reference.

[0118] The results of the ratios are expressed in Table 2. It is found that the composition C-5 in accordance with the present invention exhibits the value which is markedly greater than that of the reference composition C-1 and the comparative examples C-2



and C-3. Another example composition C-4 has the equivalent values to that of the reference C-1, the comparative examples C-2 and C-3 in the wet grip point of view.

[0119] In conclusion, the results of the tests commented above show that the rubber composition in accordance with the present invention having a plasticizer comprising a vegetable oil and microparticles of a water soluble sulfate of an alkali metal or an alkaline earth metal, gives the tires and their treads a better ice grip performance, especially melting ice grip performance, while maintaining or even improving a wet grip performance of the tire.

Table 1

Rubber Composition No.:	Ref.	Comparative examples		Examples of the invention	
	C-1	C-2	C-3	C-4	C-5
BR (1)	60	60	60	60	60
NR (2)	40	40	40	40	40
Silica (3)	80	80	80	80	80
Coupling agent (4)	5	5	5	5	5
Carbon black (5)	5	5	5	5	5
MES oil (6)	55	55			
TDAE oil (7)			55		
Vegetable oil (8)				55	20
Hydrocarbon resin (9)	15	15	15	15	35
Microparticles (10)		10	10	10	10
ZnO	1.2	1.2	1.2	1.2	1.2
Stearic acid	1	1	1	1	1
Antiozone wax	1.5	1.5	1.5	1.5	1.5
Antioxidant (11)	2	2	2	2	2
DPG (12)	1.5	1.5	1.5	1.5	1.5
Sulphur	2	2	2	2	2
Accelerator (13)	1.7	1.7	1.7	1.7	1.7

- (1) BR with 0.3% of 1,2; 2.7%-1,4 of trans; 97% of cis-1,4 ( $T_g = -104^\circ\text{C}$ );
- (2) Natural rubber (peptised);
- (3) Silica "Zeosill 165MP" from Rhodia, " (CTAB, BET: about  $160\text{ m}^2/\text{g}$ );
- (4) Coupling agent TESPT ("Si69" from Evonik);
- (5) Grade ASTM N234 (Cabot);
- (6) MES oil ("Catenex SNR" from Shell)
- (7) TDAE oil ("VivaTec 500" from H&R)
- (8) Oleic sunflower oil ("Agripure 80" from Cargil, Weight percent oleic acid: 100%);
- (9) Hydrocarbon resin C5/C9 type ("Escorez ECR-373" from Exxon);
- (10) Magnesium sulfate (filtering "MG-OK" from Ako Kasei, D10:  $52\text{ }\mu\text{m}$ , D50:  $84\text{ }\mu\text{m}$ , D90:  $140\text{ }\mu\text{m}$  and Span:  $1.05\text{ }\mu\text{m}$  measured by laser diffraction methods in accordance with ISO13320-1);
- (11) N-(1,3-dimethylbutyl)-N-phenyl-para-phenylenediamine ("Santoflex 6-PPD" from Flexsys);
- (12) Diphenylguanidine ("Perkacit DPG" from Flexsys);
- (13) N-dicyclohexyl-2-benzothiazolesulphenamide ("Santocure CBS" from Flexsys).

Table 2

Rubber Composition No.:	Ref.	Comparative examples		Examples of the invention	
	C-1	C-2	C-3	C-4	C-5
Melting ice grip <sup>(1)</sup>	100	111	113	131	114
Wet grip Index <sup>(2)</sup>	100	100	100	100	122

(1) Friction test on icy road (at -2 °C)

(2)  $\text{Tan}(\delta)[\text{at } 0^\circ\text{C, } 10\text{Hz, } 0.7\text{MPa, C-n}]/\text{Tan}(\delta)[\text{at } 0^\circ\text{C, } 10\text{Hz, } 0.7\text{MPa, C-1}]$   
 $\times 100$ ;

where: n=1, 2, 3, 4 and 5

### CLAIMS

1. A rubber composition comprising at least:

- a diene elastomer;
- between 50 and 150 phr of a reinforcing filler;
- between 20 and 160 phr of a plasticizer comprising more than 10 phr and less than 80 phr of a vegetable oil; and
- between 2 and 40 phr of microparticles of a water soluble sulfate of an alkali metal or an alkaline earth metal.

2. The rubber composition according to Claim 1, wherein the content of the vegetable oil is more than 10 phr and less than 70 phr.

3. The rubber composition according to Claim 1 or Claim 2, wherein the vegetable oil is chosen from the group consisting of linseed oil, safflower oil, soybean oil, corn oil, cottonseed oil, turnip seed oil, castor oil, tung oil, pine oil, sunflower oil, palm oil, olive oil, coconut oil, groundnut oil, grapeseed oil, and mixtures thereof.

4. The rubber composition according to Claim 3, wherein the vegetable oil is sunflower oil.

5. The rubber composition according to any one of Claims 1 to 4, wherein the fatty acid from which the vegetable oil derives, comprises oleic acid in a weight fraction greater than or equal to 60%.

6. The rubber composition according to any one of Claims 1 to 5, wherein the microparticles have a median particle size by volume and a width of a volume particle size distribution, both of which are measured by laser diffraction methods in accordance with ISO standard 13320-1 and satisfy the following relations:

- $50\text{ }\mu\text{m} < D_{50} < 150\text{ }\mu\text{m}$ ;
- $0.50 < \text{Span} < 1.50$ ;

wherein:

- $D_{50}$  is the medium particle size by volume corresponding to 50 % of a cumulative distribution obtained from the volume particle size distribution;
- $\text{Span} = (D_{90} - D_{10}) / D_{50}$ ; and
- $D_{10}$  and  $D_{90}$  are the particle size corresponding to 10 volume % and to 90 volume %, respectively, of the cumulative particle distribution.

7. The rubber composition according to any one of Claims 1 to 6, wherein the content of the microparticles is between 2 and 30 phr.

8. The rubber composition according to any one of Claims 1 to 7, wherein the alkali metal or the alkaline earth metal is chosen from the group consisting of sodium, potassium, magnesium, calcium and mixtures thereof.

9. The rubber composition according to any one of Claims 1 to 8, wherein the water soluble sulfate is chosen from the group consisting of magnesium sulfate, potassium sulfate and mixtures thereof.

10. The rubber composition according to any one of Claims 1 to 9, wherein the water soluble sulfate is magnesium sulfate.

5 11. The rubber composition according to any one of Claims 1 to 10, wherein the diene elastomer is chosen from the group consisting of natural rubber, synthetic polyisoprenes, polybutadienes, butadiene copolymers, isoprene copolymers and the mixtures thereof.

12. The rubber composition according to Claim 11, wherein the rubber composition  
10 comprises more than 50 phr of natural rubber or of synthetic polyisoprene.

13. The rubber composition according to Claim 11, wherein the rubber composition comprises more than 50 phr of polybutadiene having a content of cis-1,4 bonds of greater than 90%.

15 14. The rubber composition according to any one of Claims 1 to 13, wherein the reinforcing filler comprises greater than 60 phr of carbon black.

15. The rubber composition according to any one of Claims 1 to 14, wherein the  
20 reinforcing filler comprises greater than 70 phr of a reinforcing inorganic filler.

16. The rubber composition according to any one of Claims 1 to 15, wherein the content of total reinforcing filler is between 60 and 120 phr.

17. The rubber composition according to any one of Claims 1 to 16, wherein the content of the plasticizer is between 30 and 120 phr.

18. The rubber composition according to any one of Claims 1 to 17, wherein the plasticizer further comprises a compound selected from the group consisting of hydrocarbon resins, liquid plasticizers other than the vegetable oil and the mixtures thereof.

19. The rubber composition according to Claim 18, wherein the hydrocarbon resin exhibits a Tg greater than 20°C.

20. The rubber composition according to Claim 18 or 19, wherein the hydrocarbon resins are chosen from the group consisting of cyclopentadiene homopolymer or copolymer resins, dicyclopentadiene homopolymer or copolymer resins, terpene homopolymer or copolymer resins, C<sub>5</sub> fraction homopolymer or copolymer resins, C<sub>9</sub> fraction homopolymer or copolymer resins, alpha-methyl styrene homopolymer or copolymer resins, and the mixtures of these resins.

21. The rubber composition according to any one of claims 18 to 20, wherein the content of hydrocarbon resin is between 3 and 60 phr.

22. The rubber composition according to Claim 18, wherein the liquid plasticizers other than the vegetable oils, are chosen from the group consisting of polyolefinic oils, naphthenic oils, paraffinic oils, Distillate Aromatic Extracts (DAE) oils, Medium Extracted Solvates (MES) oils, Treated Distillate Aromatic Extracts (TDAE) oils, Residual Aromatic Extracts

(RAE) oils, Treated Residual Aromatic Extracts (TRAЕ) oils, Safety Residual Aromatic Extracts (SRAE) oils, mineral oils, ether plasticizers, ester plasticizers, phosphate plasticizers, sulphonate plasticizers and the mixtures thereof.

5   **23.** The rubber composition according to Claim 22, wherein the content of liquid plasticizer is less than 20 phr.

**24.** Use of a composition in accordance with any one of Claims 1 to 23 for the manufacture of treads for pneumatic tires.

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**25.** A tread for a pneumatic tire comprising a composition in accordance with any one of Claims 1 to 23.

**26.** A pneumatic tire comprising the tread according to Claim 25.

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## INTERNATIONAL SEARCH REPORT

International application No.

PCT/JP2015/084084

A. CLASSIFICATION OF SUBJECT MATTER			
Int.Cl. See extra sheet			
According to International Patent Classification (IPC) or to both national classification and IPC			
B. FIELDS SEARCHED			
Minimum documentation searched (classification system followed by classification symbols)			
Int.Cl. C08L9/00, B60C1/00, C08K3/00, C08K3/04, C08K3/24, C08L7/00, C08L91/00			
Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched			
Published examined utility model applications of Japan 1922-1996 Published unexamined utility model applications of Japan 1971-2015 Registered utility model specifications of Japan 1996-2015 Published registered utility model applications of Japan 1994-2015			
Electronic data base consulted during the international search (name of data base and, where practicable, search terms used)			
C. DOCUMENTS CONSIDERED TO BE RELEVANT			
Category*	Citation of document, with indication, where appropriate, of the relevant passages		Relevant to claim No.
Y	JP 2013-545835 A (COMPAGNIE GENERALE DES ETABLISSEMENTS MICHELIN, MICHELIN RECHERCHE ET TECHNIQUE S.A.) 2013.12.26, claims 1 to 20, paragraphs [0001], [0007], [0024], [0025], [0071]-[0074] & WO 2012/052331 A1 & EP 2629986 B1 & FR 2966157 A1 & CA 2813516 A1 & CN 103153647 A & EA 201390587 A		1-26
Y	US 5252649 A (SEMPERIT REIFEN AG) 1993.10.12, claim 1, column 2, lines 14 to 22, column 4, lines 3 to 6, Table1, Table2 & EP 524164 A1		1-26
<input type="checkbox"/> Further documents are listed in the continuation of Box C. <input type="checkbox"/> See patent family annex.			
* Special categories of cited documents: "A" document defining the general state of the art which is not considered to be of particular relevance "E" earlier application or patent but published on or after the international filing date "L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified) "O" document referring to an oral disclosure, use, exhibition or other means "P" document published prior to the international filing date but later than the priority date claimed "T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention "X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone "Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art "&" document member of the same patent family			
Date of the actual completion of the international search		Date of mailing of the international search report	
18.12.2015		28.12.2015	
Name and mailing address of the ISA/JP		Authorized officer	
Japan Patent Office		Isamu Komori	
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CLASSIFICATION OF SUBJECT MATTER

C08L9/00 (2006.01) i, B60C1/00 (2006.01) i, C08K3/00 (2006.01) i,  
C08K3/04 (2006.01) i, C08K3/24 (2006.01) i, C08L7/00 (2006.01) i,  
C08L91/00 (2006.01) i