

(12) PATENT
(19) AUSTRALIAN PATENT OFFICE

(11) Application No. **AU 199947770 B2**
(10) Patent No. **756763**

(54) Title
Thermal grafts of polyamides

(51)⁶ International Patent Classification(s)
C08G 081/00

(21) Application No: 199947770 (22) Application Date: 1999 . 06 . 23

(87) WIPO No: WO00/00536

(30) Priority Data

(31) Number	(32) Date	(33) Country
09/106452	1998 . 06 . 30	US

(43) Publication Date : 2000 . 01 . 17
(43) Publication Journal Date : 2000 . 03 . 30
(44) Accepted Journal Date : 2003 . 01 . 23

(71) Applicant(s)
MBT Holding AG; University of South Alabama

(72) Inventor(s)
Steven C. Sikes; Thomas M. Vickers

(74) Agent/Attorney
CULLEN and CO,GPO Box 1074,BRISBANE QLD 4001

(56) Related Art
US 5925728
US 5408028
US 5061750

47770/99



PCT

WORLD INTELLECTUAL PROPERTY ORGANIZATION
International Bureau

INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

(51) International Patent Classification ⁷ : C08G 81/00		A1	(11) International Publication Number: WO 00/00536
			(43) International Publication Date: 6 January 2000 (06.01.00)
(21) International Application Number: PCT/EP99/04378		(81) Designated States: AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, CA, CH, CN, CU, CZ, DE, DK, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MD, MG, MK, MN, MW, MX, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT, UA, UG, UZ, VN, YU, ZW, ARIPO patent (GH, GM, KE, LS, MW, SD, SL, SZ, UG, ZW), Eurasian patent (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European patent (AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE), OAPI patent (BF, BJ, CF, CG, CI, CM, GA, GN, GW, ML, MR, NE, SN, TD, TG).	
(22) International Filing Date: 23 June 1999 (23.06.99)		<p>Published</p> <p><i>With international search report.</i></p> <p><i>Before the expiration of the time limit for amending the claims and to be republished in the event of the receipt of amendments.</i></p>	
(30) Priority Data: 09/106,452 30 June 1998 (30.06.98) US			
(71) Applicants: MBT HOLDING AG [CH/CH]; Vulkanstrasse 110, CH-8048 Zürich (CH). UNIVERSITY OF SOUTH ALABAMA [US/US]; Mobile, AL 36688 (US).			
(72) Inventors: SIKES, Steven, C.; University of South Alabama, LSB 124, Mobile, AZ 36688 (US). VICKERS, Thomas, M.; 10140 Weathersfield Drive, Concord Township, OH 44060 (US).			
(74) Agent: McSTEa, John, Anthony; MBT (Schweiz) AG, Patents and Trade Marks, Vulkanstrasse 110, CH-8048 Zürich (CH).			
(54) Title: THERMAL GRAFTS OF POLYAMIDES			
(57) Abstract			
<p>Heating an intimate mixture of: (a) a polyamide which contains pendant carboxylic acid groups; and (b) a compound which contains an amino group efficiently affords graft polymers with a high graft density. Such graft polymers are particularly useful as additives for cementitious compositions.</p>			

TITLE OF THE INVENTION

THERMAL GRAFTS OF POLYAMIDES

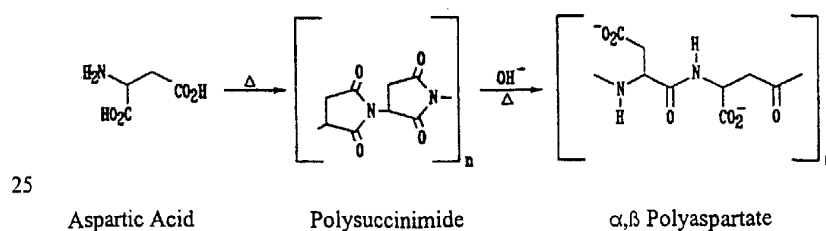
5 BACKGROUND OF THE INVENTION

Field of the Invention:

The present invention relates an efficient method of grafting compounds which contain an amino group, such as polyoxyalkylene amines, to polyamides, such as polyaspartic acid, polyglutamic acid, and copolymers of aspartic and glutamic acids, with pendant carboxylic acid groups. The present method also relates to the graft polymers produced by such a process. The present invention further relates to compositions which contain such graft polymers and methods of using such graft polymers.

15 Discussion of the Background:

Polyaspartate is economically produced by the thermal polymerization of aspartic acid or precursors of aspartic acid such as maleic acid and ammonia. The initial product is polysuccinimide (synonyms: polyanhydroaspartic acid, polyaspartimide), which is composed of residues of succinimide, the cyclic imide of aspartic acid. A review of patents and other literature dating to 1850 on the synthesis of polyaspartate and polysuccinimide is provided in U.S. Patent No. 5,594,077 (Groth et al), which is incorporated herein by reference. The production of polyaspartic acid by the thermal polymerization of aspartic acid and hydrolysis of the thus-obtained polysuccinimide is shown schematically below:



Synthesis of potentially useful derivatives of polysuccinimide and polyaspartate also has a long history. A common approach has been the nucleophilic addition of

primary amines such as ethanolamine to form an amide linkage upon opening of the imide ring of a succinimide residue, producing a polysuccinimide with pendant groups (see: U.S. Patent No. 3,846,380 (Fujimoto et al)). Because polysuccinimide is insoluble in water, the reactions are normally accomplished in organic solvents such as dimethyl formamide or N-methyl pyrrolidone. The derivatized polysuccinimide can be converted to a derivatized polyaspartate by mild alkaline hydrolysis of the remaining imide rings.

Thermal copolymerization of aspartic acid and other monomers also has frequently been reported beginning with U.S. Patent No. 3,052,655 (Fox et al) and extending to U.S. Patent No. 5,658,464 (Hann et al). U.S. Patent Nos. 5,357,004 (Wood et al); 5,442,038 (Wood et al); 5,502,117 (Wood et al); 5,510,426 (Wood et al); 5,510,427 (Wood et al); 5,519,110 (Wood et al); 5,521,279 (Wood et al); 5,371,179 (Paik et al); 5,371,177 (Paik et al); 5,410,017 (Bortnick et al); and 5,478,919 (Koskan et al) disclose copolymers produced by synthetic routes that included monoamines of polyoxyalkylenes. Applications such as detergent additives, dispersants, corrosion inhibitors, antiscalants, additives to fertilizers, additives to oral health care products and cosmetics, and others were disclosed.

Japanese Patent Application 7-172888 (Nagatomo et al) discloses derivatives of polysuccinimide and polyaspartate prepared by nucleophilic addition of monoamines such as ethanolamine and taurine to polysuccinimide in dimethyl formamide or water. Cement admixtures containing polyaspartic acid and these derivatives are disclosed, and some effects on the flow of Portland cement are disclosed.

Derivatives of polyaspartate have also been formed by thermal addition of coreactants to a preformed polyaspartate. U.S. Patent Nos. 5,247,068 (Donachy et al); 5,260,272 (Donachy et al); and 5,284,936 (Donachy et al) disclose that the use of a preformed polyaspartate as a reactant lead to improvements as compared to the copolymerization of monomeric reactants.

U.S. Patent Nos. 5,552,516 (Ross et al); 5,612,384 (Ross et al); 5,461,085 (Nagatomo et al); 5,525,682 (Nagatomo et al); 5,525,703 (Kalota) and WO 95/35337 (Ross et al); WO 96/08523 (Ross et al); and EP 0 658 586 A1 (Nagatomo et al) each disclose aqueous reactions of diamines with polysuccinimide to effect nucleophilic addition at elevated pH with the object of producing water-soluble, high MW, crosslinked polyaspartates upon alkaline treatment of the crosslinked polysuccinimides.

Alternatively, if the degree of crosslinking is sufficient, water-absorbent, insoluble materials were produced.

A problem with this approach is that the OH⁻ groups in water at elevated pH compete with the NH₂ groups of the diamines to donate electrons to imide rings of the polymer, thus ring-opening the polysuccinimide residues by hydrolysis. Another
5 problem is that the polysuccinimide is hydrophobic and water-insoluble such that much of the polysuccinimide is not readily available for reactions in the aqueous phase. Consequently, elevated amounts of diamines are required and the reactions are inefficient and incomplete, leaving partially reacted as well as unincorporated diamines.

10 U.S. Patent No. 5,639,832 (Kroner et al) discloses the aqueous reaction of monoamino compounds such as amino ethyl sulfonate (taurine), other amino acids, and monoamino poly(propylene-ethylene)glycol with polysuccinimide at elevated pH. Graft efficiency no higher than 60% was reported. The materials produced by the reactions, which include the entire mixture of reactants and products at the end of the
15 reactions, are disclosed as being useful as detergent additives, dispersants, and antiscalants.

In addition to polyaspartates, other polyanionic polymers such as polyacrylates have been derivatized with polyoxyalkylene side chains via thermal reaction of monofunctional groups of the derivatizing compound and the carboxyl groups of the
20 polymer (see: U.S. Patent Nos. 5,393,343 (Darwin et al); 5,614,017 (Shaw); and 5,670,578 (Shaw)). The derivatized polyacrylates are disclosed as being useful for improving the rheological properties of cement, acting as water-reducing and plasticizing additives.

Dispersants are used to improve the flow characteristics of cement slurries by
25 breaking up cement agglomerates and freeing the water, thus giving the slurries of lower viscosity and allowing desirable flow conditions to be obtained at lower pump pressures. V.S. Ramachandran, Concrete Admixtures Handbook: Properties, Science, and Technology, Second Edition, Noyes Publications, 1995, which is incorporated herein by reference. Superplasticizers such as sulfonated melamine formaldehyde
30 condensate (SMF) and sulfonated naphthalene formaldehyde condensate (BNS) are commonly used as dispersants. However, these compounds require more than the desired amount of material to achieve a desired level of concrete workability or water

reduction. In addition, these materials do not achieve the full range of water reducing capability for the various types of concrete admixtures (e.g., Type A and Type F as defined in ASTM C494 etc.).

It is important that dispersants be used in concrete situations where strength and durability are involved, as dispersants are a necessary component in high strength and high durability concretes. Due to the use of low water amounts in the high performance concretes, sometimes high dispersant amounts are necessary to achieve workable concretes. High BNS levels can lead to undesirable retardation of set and may not provide the required workability retention over time.

It is desirable to provide a material that is several times more efficient as a cement or concrete dispersant than the traditional materials such as BNS and SMF. Improving efficiency reduces the amount of material required to achieve a desired level of concrete workability or water reduction. With respect to BNS and SMF, it is also desirable to improve slump retention while maintaining normal setting characteristics. Providing a dispersant with full range (Type A and F) water-reducing capability is also a desirable characteristic.

Thus, there remains a need for a method of preparing polyamides, such as polyaspartic acid and polyglutamic acid, which contain pendant carboxylic acid groups grafted with amino compounds, such as methoxylated polyoxyalkylene amines, in high yield without the use and expense of excess coreactants or organic solvents. Such a method would involve no waste disposal and produce no harmful byproducts. There also remains a need for improved surfactants and dispersants for use as additives to cementitious materials.

25

SUMMARY OF THE INVENTION

Advantageously, the present invention may provide a novel method of grafting compounds which contain amino groups to polyamides, such as polyaspartic acid and polyglutamic acid, which contain pendant carboxylic acid groups.

It is another advantage of the present invention that there may be provided a method of grafting compounds which contain amino groups to polyamides, such as polyaspartic acid and polyglutamic acid, which contain pendant carboxylic acid groups, which method achieves a high graft efficiency and which therefore does not require the use of excess grafting agent.



The present invention may also advantageously provide a method of grafting compounds which contain amino groups to polyamides, such as polyaspartic acid and polyglutamic acid, which contain pendant carboxylic acid groups which method does not require the use of organic solvents.

5 The present invention may also advantageously provide a method of grafting compounds which contain amino groups to polyamides, such as polyaspartic acid and polyglutamic acid, which contain pendant carboxylic acid groups which method affords graft polymers with a high graft density.

The present invention may also advantageously provide novel graft polymers in
10 which compounds which contain amino groups are grafted to polyamides, such as polyaspartic acid and polyglutamic acid, which contain pendant carboxylic acid groups.

The present invention may also advantageously provide novel graft polymers, having a high graft density, in which compounds which contain amino groups are grafted to polyamides, such as polyaspartic acid and polyglutamic acid, which contain
15 pendant carboxylic acid groups.

The present invention may also advantageously provide novel compositions which contain such graft polymers in which compounds which contain amino groups are grafted to polyamides, such as polyaspartic acid and polyglutamic acid, which contain pendant carboxylic acid groups.

20 The present invention may also advantageously provide novel methods of improving the flow properties of cement.

These and other advantages that may be provided by the present invention, will become apparent during the following detailed description, have been achieved by the inventors' discovery that heating an intimate mixture of polyamide, such as polyaspartic
25 acid and polyglutamic acid, which contains pendant carboxylic acid groups, which mixture is substantially free of water or organic solvent, with a compound which contains an amino group affords a graft polymer in which the compound which contains an amino group is grafted to the polyamides which contain pendant carboxylic acid groups. The inventors have also discovered that the graft polymers so produced are
30 useful as additives to cementitious materials.



BRIEF DESCRIPTION OF THE DRAWINGS

A more complete appreciation of the invention and many of the attendant advantages thereof will be readily obtained as the same becomes better understood by reference to the following detailed description when considered in connection with the accompanying drawings, wherein:

Figure 1 is the infrared spectrum of the polysuccinimide prepared in Preparative Example 1, thermal polymer of aspartic acid, 220°C, 8 hours. Notice the prominent imide peak at $\sim 1716\text{ cm}^{-1}$. There also is some indication of amide linkages as shown by a small peak at $\sim 1630\text{ cm}^{-1}$ that is thought to be associated with a small number of branch points of this type of polysuccinimide. The carboxyl signal at 1400 cm^{-1} is associated with the C-termini of the main polymer chains and the branches;

Figure 2 is the infrared spectrum of the polyaspartate prepared by mild alkaline hydrolysis of the polysuccinimide prepared in Preparative Example 1. Notice the characteristic amide doublet at ~ 1600 and 1650 cm^{-1} , as well as the carboxyl signal at 1400 cm^{-1} ;

Figure 3 is the infrared spectrum of the derivatized polyaspartate prepared in Example 2, a graft of low MW polyoxyalkylene amine and high MW polyaspartate. Notice some residual imide signal at $\sim 1710\text{ cm}^{-1}$, the amide signals around 1600 and 1650 cm^{-1} , the carboxyl signal at $\sim 1400\text{ cm}^{-1}$ and 1540 cm^{-1} , and a prominent peak associated with the polyoxyalkylene side chains at 1103 cm^{-1} ;

Figure 4 is the gel permeation chromatography profile of the product of Example 2, an efficient thermal synthesis of a graft of low MW polyoxyalkylene amine and high MW polyaspartate. The derivatized polyaspartate eluted at about 28 minutes (peak 1), with the residual, unreacted polyoxyalkylene amine at 34 to 35 minutes (peak 2). The salt eluted at 43 minutes (peak 3). Compare to Figures 5 and 6 for examples of peaks of poorly grafted polyoxyalkylene amine under conditions of inefficient reaction with polysuccinimide or sodium aspartate;

Figure 5 is the gel permeation chromatography profile of the product of Comparative Example 1, grafting of a medium MW polyoxyalkylene amine on to a low MW polysuccinimide by nucleophilic addition in dimethyl formamide. The derivatized polyaspartate of this reaction eluted as a doublet between 22 and 28 minutes (peak 1),

SUBSTITUTE SHEET (RULE 26)

with the majority of the polyoxyalkylene amine remaining unreacted and eluting at about 34 minutes (peak 2): peak 3, salt;

Figure 6 is the infrared spectrum of the purified, derivatized polyaspartate of Comparative Example 1, a graft of medium MW polyoxyalkylene amine and low MW polysuccinimide by nucleophilic addition in dimethyl formamide. The amide signals are seen at 1600 and 1650 cm^{-1} , the polyoxyalkylene signal at 1104 cm^{-1} . There was essentially no imide signal, normally at 1710 cm^{-1} ;

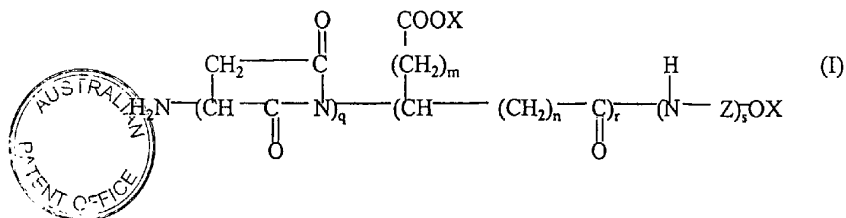
Figure 7 is the gel permeation chromatography profile of products of the reaction of Comparative Example 2, an attempt to graft a low MW polyoxyalkylene amine onto a low MW polysuccinimide by nucleophilic addition in the absence of solvent. Peak 1, grafted polyaspartate: peak 2, unreacted polyoxyalkylene amine: peak 3, salt;

Figures 8a and 8b are the gel permeation chromatography profiles of products of the reaction of Comparative Example 4, an attempt to copolymerize sodium aspartate and polyoxyalkylene amine by thermal treatment at 180°C. Figure 8a: sample after 2 hours of thermal treatment. Figure 8b: sample after 20.5 hours of thermal treatment. Peak 1, polyoxyalkylene amine: peak 2, salt.

DETAILED DESCRIPTION OF THE PREFERRED EMBODIMENTS

Thus, in a first embodiment, the present invention provides a method for producing graft polymers in which compounds which contain amino groups are grafted to polyaspartic acid. The present method involves heating an intimate mixture of a polyamide which contains pendant carboxylic acid groups, such as polyaspartic acid, polyglutamic acid, and copolymers of aspartic and glutamic acids, with a compound which contains an amino group selected from the group consisting of simple amines, C_{2-4} alkylene amines, polyoxyalkylene monoamines, aminosulfonates, aminophosphonates and hydrophobic amino acids.

The polyamides which contain pendant carboxylic acid groups will in many cases have the general formula (I):



MBC-0344

8

wherein:

each X independent of the other is hydrogen, an alkali metal ion, an alkaline earth metal ion, ammonium ion, or substituted ammonium, with the proviso that at least 50 mole %, preferably at least 75 mole %, more preferably at least 100 mole %, of X is hydrogen;

each -NH-Z- is a monomer unit derived from a compound which contains (i) a -NH₂ group and (ii) a -CO₂H or a -SO₃H group;

each n, independently of the other, is 0 or an integer of 1 to 20, preferably 0 to 2;

each m, independently of the other, is 0 or an integer of 1 to 20, preferably 0 to 2;

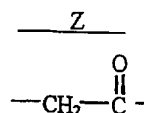
q, r, and s reflect the mole fraction of each type of monomer unit, wherein the sum of q + r + s equals one, wherein q is a decimal value from 0 to 0.95, preferably 0 to 0.50, more preferably 0 to 0.25; r is a decimal value of 0.05 to 1, preferably 0.50 to 1, more preferably 0.75 to 1; and s is a decimal value of 0 to 0.50, preferably 0 to 0.25, more preferably 0.

The total number of monomer units in the polyamide is 5 to 1000, preferably 10 to 250.

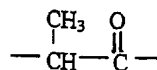
Examples of substituted ammonium include mono-, di-, tri-, and tetra-C₁₄-alkylammonium and mono-, di-, tri-, and tetra-C_{2,4}-alkoxyammonium.

Examples of the compound which contains: (i) a -NH₂ group and (ii) a -CO₂H or a -SO₃H group; include glycine, alanine, valine, leucine, isoleucine, asparagine, glutamine, *p*-aminobenzoic acid, sulfanilic acid, H₂N(CH₂)_n-CO₂H (where n is an integer of 2-8, preferably 6), etc. Thus, Z may be, e.g.:

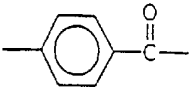
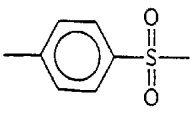
Compound
glycine

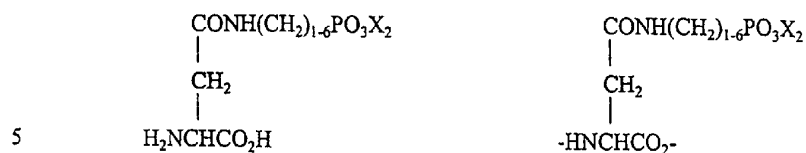


alanine



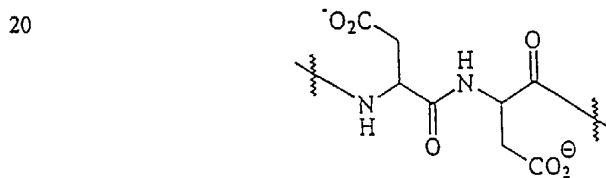
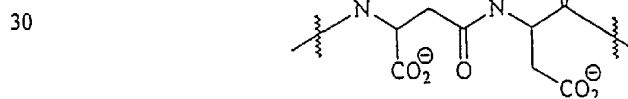
AMENDED SHEET

	leucine	$\begin{array}{c} (\text{CH}_3)_2\text{CHCH}_2 \\ \\ -\text{CH}-\text{C}(=\text{O})- \end{array}$
5	isoleucine	$\begin{array}{c} \text{CH}_3 \\ \\ \text{CH}_3\text{CH}_2\text{CH} \\ \\ -\text{CH}-\text{C}(=\text{O})- \end{array}$
10	valine	$\begin{array}{c} (\text{CH}_3)_2\text{CH} \\ \\ -\text{CH}-\text{C}(=\text{O})- \end{array}$
15	asparagine	$\begin{array}{c} \text{O} \\ \\ \text{H}_2\text{N}-\text{C}-\text{CH}_2 \\ \\ -\text{CH}-\text{C}(=\text{O})- \end{array}$
	glutamine	$\begin{array}{c} \text{O} \\ \\ \text{H}_2\text{N}-\text{C}-\text{CH}_2\text{CH}_2 \\ \\ -\text{CH}-\text{C}(=\text{O})- \end{array}$
20	p-aminobenzoic acid	
25	sulfanilic acid	
30	$\begin{array}{c} \text{H}_2\text{N}(\text{CH}_2)_n\text{CO}_2\text{H} \\ \\ \text{CONH}(\text{CH}_2)_{1-6}\text{SO}_3\text{X} \\ \\ \text{CH}_2 \\ \\ \text{H}_2\text{NCHCO}_2\text{H} \end{array}$	$\begin{array}{c} \text{O} \\ \\ -(\text{CH}_2)_n-\text{C}- \\ \\ \text{CONH}(\text{CH}_2)_{1-6}\text{SO}_3\text{X} \\ \\ \text{CH}_2 \\ \\ -\text{HNCHCO}_2- \end{array}$

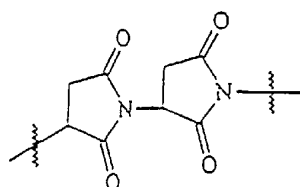


In a preferred embodiment, the polyamide which contains pendant carboxylic acid groups is polyaspartic acid (n is 0 or 1 and m is 0 or 1) or polyglutamic acid (n is 0 or 2 and m is 0 or 2). In a particularly preferred embodiment, the polyamide which contains pendant carboxylic acid groups is polyaspartic acid. Although the present process will be discussed below with particular reference to polyaspartic acid, it is to be understood that when applicable the considerations discussed in the context of polyaspartic acid, e.g., molecular weight, degree of neutralization, residue substitution, etc., also apply to the other polyamides which contain pendant carboxylic acid groups.

There is no limitation on the type of polyaspartic acid which can be used in the present process. In particular, it should be understood that the polyaspartic acid used in the present process may contain either α or β amide type residues or any combination of α and β amide type residues, which are shown below:

 α -amide linkage β -amide linkage

It should also be understood that the polyaspartic acid used in the present invention may be in a form in which a certain number of the aspartic acid residues are in the imide or succinimide form:



succinimide linkage

Thus, up to 95 %, preferably up to 50 %, more preferably up to 10 %, even more preferably essentially none, of the aspartic acid residues may be in the imide or succinimide form in the polyaspartic acid used in the present process.

The polyaspartic acid will typically have a gel permeation molecular weight of 1000 to 30,000, but it is possible to use polyaspartic acids which have a molecular weight outside this range. Good results have been achieved using polyaspartic acid having a gel permeation molecular weight of 1000 to 30,000, and even more typically 2000 to 10,000.

The polyaspartic acid may be prepared by any conventional process for producing polyaspartic acid. For example, the polyaspartic acid may be produced by solid state synthesis or the thermal polymerization of aspartic acid or precursors of aspartic acid such as maleic acid and ammonia followed by mild, alkaline hydrolysis of the polysuccinimide to produce polyaspartate and followed by acidification to produce polyaspartic acid, accompanied by the removal of cations. A review of the synthesis of polyaspartic acid is provided in U.S. Patent No. 5,594,077, which is incorporated herein by reference. Other methods of producing polyaspartic acid are disclosed in PCT Application No. WO 97/18249, Great Britain Patent No. 2307689, and Japanese Patent Application No. 5-321720, which are incorporated herein by reference.

The present process utilizes polyaspartic acid as opposed to polyaspartate. In the context of the present invention, the term polyaspartic acid includes polymers in which a small number of the pendant carboxy groups are neutralized. Thus, in the

present method, at least 50 %, preferably at least 75 %, more preferably at least 100 %, of the pendant carboxy groups in the polyaspartic acid are acidified as opposed to being salified.

In a preferred embodiment, the polyaspartic acid is essentially free of extraneous ions. As noted above, polyaspartic acid may be produced by the acidification of polyaspartate. In polyaspartate the pendent carboxylate groups are associated with a cation, typically sodium. Thus, the acidification of the sodium polyaspartate with an acid, e.g., hydrochloric acid, will yield, in addition to the desired polyaspartic acid, a salt such as sodium chloride. It is preferred that the polyaspartic acid used in the present process be essentially free of extraneous ions from such salts. A preferred way to avoid such salts is to convert the polyaspartate (e.g., sodium polyaspartate) to polyaspartic acid by dialysis against aqueous HCl (0.01 N HCl) followed by lyophilization (HCl is essentially completely removed by lyophilization). Alternatively, it is possible to use such a dialysate as is without lyophilization. It is also possible to convert the polyaspartate to polyaspartic acid by means of an ion exchange resin such as amberlyst. In the context of the present invention, the term essentially free of extraneous ions means that the polyaspartic acid contains less than 20 wt.%, preferably less than 5 wt.%, more preferably less than 1 wt.%, of salt (e.g., NaCl), based on the total weight of the polyaspartic acid.

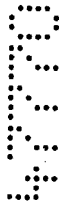
The compound containing the amino group may be selected from the group consisting of simple amines, C₂₋₄ alkylene amines, polyoxyalkylene monoamines, aminosulfonates, aminophosphonates and hydrophobic amino acids. Preferably the amino group is any medium to low molecular weight (MW from 600 to 25,000, preferably less than 10,000) amine which does not contain any other functional group which would interfere with the grafting reaction. Specific examples of such compounds containing an amino group include simple amines, such as include aliphatic monoamines (H₂NR, where R is C₁₋₂₂ alkyl), arylaliphatic monoamines, (H₂NR' where R' is C₆₋₁₀aryl, C₇₋₁₅ alkylaryl, or C₇₋₁₅ arylalkyl), C₂₋₄ alkanolamines, polyoxyalkylene monoamines, aminosulfonates (such as H₂N(CH₂)₁₋₆SO₃X), aminophosphonates (such as H₂N(CH₂)₁₋₆PO₃X), and hydrophobic amino acids. Preferred monoamines include ethanolamine, propylamine, leucine, isoleucine, valine, monoamino copolyethylene:propylene, laurylamine, and stearylamine. Particularly preferred



12a

monoamines include ethanolamine, leucine, isoleucine, valine, and laurylamine, and monamino copolyethylene:propylene. A preferred class of compounds containing an amino group include the polyoxyalkylene

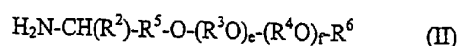
5



MBC-0344

13

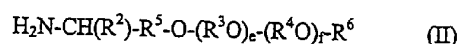
amines of the general the formula (II):



- 5 wherein R^2 and R^6 are, each independently, H, phenyl, substituted phenyl, or C_1 to C_4 linear or branched alkyl, such as methyl, ethyl, propyl, or butyl; R^5 is phenylene or C_1 to C_4 linear or branched alkylene, such as methylene, ethylene, propylene, or butylene; R^3 is ethylene, and R^3O is present randomly or in block form; e is 1 to 300, preferably 11 to 300; R^4 is propylene, and R^4O is present randomly or in block form; f is 0 to 180, preferably with a mole ratio of $\text{R}^3:\text{R}^4$ of 100:0 to 40:60.

Thus, in a preferred embodiment, the invention provides a graft polymer, produced by a process comprising heating an intimate mixture of: (a) a polyamide which contains pendant carboxylic acid groups; and (b) a compound which contains an amino group, to obtain said graft polymer,

- 15 wherein said compound which contains an amino group has the formula (II):



- 20 wherein R^2 and R^6 are, each independently, H, phenyl, substituted phenyl, or C_1 to C_4 linear or branched alkyl; R^5 is phenylene or C_1 to C_4 linear or branched alkylene; R^3 is ethylene; R^3O is present randomly or in block form; e is 1 to 300; R^4 is propylene; R^4O is present randomly or in block form; and f is 0 to 180; and

- wherein said polyamide which contains pendant carboxylic acid groups and said compound which contains an amino group are present in relative amounts such that said graft polymer has a graft density of at least 25%, preferably at least 30%.

25 In the graft polymer as hereinabove defined, said heating is preferably carried out without condensing and returning solvent to said intimate mixture.

- The polyamide which contains pendant carboxylic acid groups for use in the preparation of the graft polymers as hereinabove defined is preferably selected from the group consisting of polyaspartic acid, polyglutamic acid, and copolymers of aspartic acid and glutamic acid, and is most preferably polyaspartic acid.



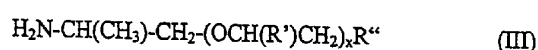
AMENDED SHEET

MBC-0344

14

Examples of substituted phenyl include mono-, di-, tri-, and tetrahalophenyl; mono-, di-, tri-, and tetra-C₁₋₄-alkoxy-phenyl; mono-, di-, tri-, and tetra-C₁₋₄-alkyl-phenyl; mono- and dinitrophenyl; etc. Specific examples of substituted phenyl include o-, p-, and m-chlorophenyl; o-, p-, and m-methylphenyl; o-, p-, and m-nitrophenyl; o-, p-, and m-cyanophenyl; and o-, p-, and m-methoxyphenyl.

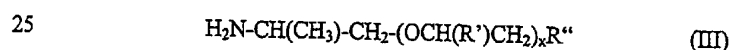
Another particularly preferred class of compounds containing an amino group includes monoamines, having the general formula (III):



in which each R' independently of the other is H or methyl, R'' is a residue derived from a monohydric alcohol (e.g., methoxy, ethoxy, propoxy, etc.), and x has a value such that the molecular weight of the compound of formula (III) is from 600 to 25,000. The ratio of R' = methyl to R' = H typically ranges from 3:19 to 9:1. Commercial examples of such monoamines include the JEFFAMINE[®] monoamines of Huntsman. Specific examples of such compounds include JEFFAMINE[®] XTJ-505 (M-600), JEFFAMINE[®] XTJ-506 (M-1000), JEFFAMINE[®] XTJ-507 (M-2005), JEFFAMINE[®] XTJ-508 (M-2070), and JEFFAMINE[®] 234.

Thus, in a further embodiment, the invention provides a graft polymer, produced by the process of heating an intimate mixture of: (a) a polyamide which contains pendant carboxylic acid groups; and (b) a compound which contains an amino group, to obtain said graft polymer,

wherein said compound which contains an amino group has the formula (III)



in which each R' independently of the other is H or methyl, R'' is a residue derived from a monohydric alcohol, and x has a value such that the molecular weight of said compound of formula (III) is from 600 to 25,000.

In this particular embodiment, said heating is preferably carried out in the solid state.

It is preferred that the compound containing an amino group exhibit a water



AMENDED SHEET

solubility of at least 5 g/l, preferably at least 500 g/l, at room temperature.

The ratio of the polyaspartic acid to the compound containing the amino group will depend on the graft density (GD) desired. In the context of the present invention, the term graft density (GD) refers to the average, numerical percentage of pendant
5 carboxyl groups of the polyaspartic acid which are grafted with the compound containing an amino group:

$$GD = CA_g / (CA_g + CA_u) \times 100$$

10 where CA_g is the number of pendant carboxyl groups of the polyaspartic acid which are grafted and CA_u is the number of pendant carboxyl groups of the polyaspartic acid which are ungrafted. For example, if the polyaspartic acid contains x equivalents of pendant carboxyl groups/gram and 0.25x equivalents/gram of these pendant carboxyl
15 groups are grafted with the compound containing an amino group and 0.75x of these pendant carboxyl groups are ungrafted, then the GD is 25 %:

$$GD = 0.25x / (0.25x + 0.75x) \times 100 = 25 \%$$

The graft efficiency of the reaction is such that the compound containing an
20 amino group is typically added in amount which is equivalent to or only slightly greater than that theoretically required to achieve the desired graft density. Thus, when it is desired to produce a graft polymer with a graft density of a certain percentage starting with a polyaspartic acid which contains Y equivalents of pendant carboxyl groups per gram, the compound which contains an amino group is suitably added in an amount of
25 A, according to the following formula:

$$A = ((GD \times Y) + Z) / 100$$

where A is the number of equivalents of amino groups added per gram of polyaspartic acid, GD is the numerical percentage of desired graft density, Y is the equivalents of
30 pendant carboxyl groups per gram of polyaspartic acid, and Z is 0 to 50 %, preferably 0 to 25 % of GD. Using the present method it is possible to achieve graft densities

SUBSTITUTE SHEET (RULE 26)

approaching 100 %.

Preferably, the polyaspartic acid and the compound containing an amino group are combined to form an intimate mixture prior to heating. A preferred method for forming an intimate mixture of the polyaspartic acid and the compound containing an amino group is to form an aqueous solution or dispersion of the polyaspartic acid and the compound containing an amino group. Typically, the aqueous dispersion is formed by mixing the polyaspartic acid and the compound containing an amino group in the desired ratio with sufficient water to result in a solution or dispersion with a total solids content of 700 g/l, preferably 500 g/l. It is also possible to include some low boiling organic solvent which is miscible with water, such as methanol, ethanol, or acetone, to assist in solubilizing or dispersing the compound containing an amino group.

The water (and possible organic solvent) in the aqueous dispersion or solution may be removed prior to the heating step to afford a dry, intimate mixture of the polyaspartic acid and the compound containing an amino group. The water (and possible organic solvent) may be removed by any conventional process, such as rotary evaporation, forced air at 50°C to 80°C, or lyophilization. It is preferred that the water be removed under conditions in which the hydrolysis of the polyaspartic acid is minimized.

Alternatively, the heating step may be commenced prior to the removal of the water (and possible organic solvent). In this case, the heating is carried out in either an open vessel or with some provision for removing vapors, and the water (and possible organic solvent) is driven off during the initial stages of the heating.

The mixture of the polyaspartic acid and the compound containing an amino group is suitably heated at a temperature of greater than 100°C, preferably 120° to 220°C, more preferably 150° to 200°C, for a time of 0.1 to 24 hours, preferably 0.25 to 12 hours, more preferably 0.5 to 8 hours. The heating step may be carried out without any stirring or the use of any special apparatus or equipment. Good results have been achieved using an inert atmosphere such as N₂.

After the heating step is complete, the product is typically obtained in the form of an oil or a solid. Then product may be used as is. In this state, ungrafted aspartic acid residues typically have been converted to succinimide residues, but upon addition with water to cementitious materials, the succinimide residues are ring-opened to the

MBC-0344

17

aspartate form due to the alkaline nature of the hydrating cement mixture. In the period of ring-opening, the presence of the succinimide residues may provide some increased dispersancy to the polymer. Alternatively, the product may be subjected to mild, alkaline hydrolysis to open the imide rings in the succinimide residues prior to use.

- 5 Such hydrolysis may be accomplished by treating the product (in water, up to 60 % by weight solids) with NaOH, KOH, or any such inorganic base, or simple organic base, while maintaining the pH in the range of 8 to 12, preferably 8 to 10, and the temperature at 40°C to 90°C, preferably 60°C to 80°C, for a time of from about 10 minutes to about 8 hours, preferably 1 to 2 hours, or just until the product is solubilized and the pH is
- 10 stable, indicating that no more COOH groups are being generated as a result of the ring opening of the succinimide residues.

The product solution may be used as is at that point, but is preferably neutralized to a pH of about 7 to 9 to avoid the possibility of alkaline hydrolysis of the polymer itself. Alternatively, the product may be spray-dried or otherwise rendered as a solid.

- 15 Typically the grafting reaction proceeds with a high graft efficiency of 50 to about 100%, more typically 80 to about 100%. In the context of the present invention the term graft efficiency is defined as:

$$GE = A'/A \times 100$$

20

where GE is the graft efficiency, A is the number of equivalents of amino groups per gram of polyaspartic acid added to the reaction, and A' is the number of equivalents of amino groups per gram of polyaspartic acid grafted to the polyaspartic acid.

- The present method also affords the desired products in yields of, typically, 50
- 25 to 100%, more typically 80 to 100%, based on polyaspartic acid, and 50 to 100%, more typically 80 to 100%, based on the compound containing an amino group.

- It is to be understood that the graft polymers in which compounds which contain amino groups are grafted to polyaspartic acid produced by the present process may then be partially or wholly neutralized. Thus, the product produced by the present process
- 30 may ultimately take a form in which the pendant carboxyl groups which are not grafted to the compound containing an amine group remain in the salt form or all or some of



AMENDED SHEET

these carboxyl groups are in the acid form. This complete or partial neutralization may be carried out by conventional treatment with a mineral acid or organic acid such as acetic acid. When the final product is to be used in a cementitious mixture which will be in contact with steel reinforcing bars, it may be preferred to use a mineral acid which
5 does not contain chloride, such as sulfuric acid.

The method of the present invention affords graft polymers in which compounds which contain amino groups are grafted to polyaspartic acid, without requiring the use of organic solvents. The method of the present invention also avoids the production of hazardous byproducts or wastes.

10 In another embodiment, the present invention provides novel graft polymers in which a compound containing an amino group is grafted to a polyamide, such as polyaspartic acid or polyglutamic acid, containing pendant carboxylic acid groups in a graft density of at least 5 %, preferably at least 10 %, more preferably at least 20 %. In the context of this embodiment, the polyamide containing pendant carboxylic acid
15 groups is the same as that described and defined above in the context of the method for producing graft polymers. In this embodiment, it is also preferred that the polyamide containing pendant carboxylic acid groups be polyaspartic acid or polyglutamic acid.

In contrast, the compound which contains an amino group is defined as including those polyoxyalkylene amines of the general formulae (II) and (III), as
20 defined above in the context of the method for producing graft polymers.

Although some graft polymers in which compounds which contain amino groups are grafted to polyaspartic acid are known, prior to the present invention, no method existed for grafting complex amines, such as those of formulae (II) or (III) to polyaspartic acid in a graft efficiency as high as 60 % or a graft density as high as 20 %.

25 As explained above in the context of the present method for producing graft polymers, the novel graft polymers of the present invention may exist in a form in which the pendant carboxyl groups which are not grafted to the compound containing an amine group are converted to succinimide residues and subsequently may be converted to the salt form or the acid form of aspartic acid residues.

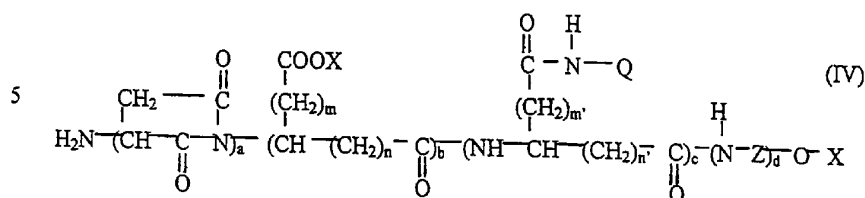
30 The novel graft polymers of the present invention may be prepared by the method described above, with the use of a polyoxyalkylene amines of the general formulae (II) or (III).

SUBSTITUTE SHEET (RULE 26)

MBC-0344

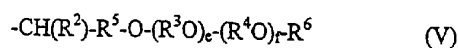
19

The novel graft polymers of the present invention have the formula (IV):



wherein:

- 10 the polymer has from 5 to 1000 monomer units;
 each X independent of the other is hydrogen, an alkali metal ion, an alkaline earth metal ion, ammonium ion, or substituted ammonium;
 each -NH-Z- is a monomer unit derived from a compound which contains (i) a -NH₂ group and (ii) a -CO₂H or a -SO₃H group;
 15 m, m', n, and n' are each independently 0 or an integer between 1 and 20, preferably 0 to 2;
 a, b, c, d each reflect the mole fraction of each unit wherein the sum of a + b + c + d equals one, wherein a is a decimal value of 0 to 0.75, preferably 0 to 0.70, more preferably 0 to 0.60; b is a decimal value of 0 to 0.75, preferably 0 to 0.70, more preferably 0 to 0.60; c is a decimal value of 0.25 to 1, preferably 0.3 to 1, more preferably 0.4 to 1; and d is a decimal value of 0 to 0.5, preferably 0 to 0.25; and
 20 Q has a molecular weight of from 600-25,000 and has the formulae (V) or (VI):



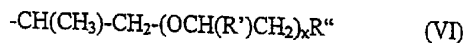
- 25 wherein R² and R⁶ are, each independently, H, phenyl, substituted phenyl, or C₁ to C₄ linear or branched alkyl, such as methyl, ethyl, propyl, or butyl; R⁵ is phenylene or C₁ to C₄ linear or branched alkylene, such as methylene, ethylene, propylene, or butylene; R³ is ethylene, and R³O is present randomly or in block form; e is 1 to 300, preferably 11 to 300; R⁴ is propylene, and R⁴O is present randomly or in block form; f is 0 to 180, preferably with a mole ratio of R³:R⁴ of 100:0 to 40:60.



AMENDED SHEET

MBC-0344

20



in which each R' independently of the other is H or methyl, R'' is a residue derived from a monohydric alcohol (e.g., methoxy, ethoxy, propoxy, etc.), and x has a value such that the molecular weight of the group of formula (VI) is from 600 to 25,000. The ratio of R' = methyl to R' = H typically ranges from 3:19 to 9:1.

It is noted that the higher the proportion of propylene oxide present in the side chain, the less hydrophilic the side chain will be.

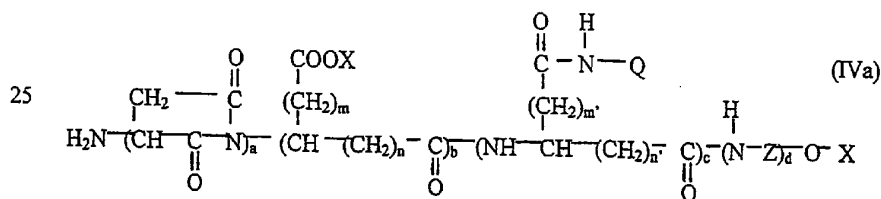
In formula (IV), preferably m, m', n, and n' are each independently an integer between 0 and 2. Most preferably, backbone is polyaspartic acid (n is 0 or 1 and m is 0 or 1) or polyglutamic acid (n is 0 or 2 and m is 0 or 2).

Examples of substituted ammonium include mono-, di-, tri-, and tetra-C₁₋₄-alkylammonium and mono-, di-, tri-, and tetra-C₂₋₄-alkoxyammonium.

The novel graft polymers of the present invention are useful in many applications including but not limited to dispersants, detergent additives, cement additives, water treatment additives, and others. The novel graft polymers of the present invention are particularly useful for modifying the flow properties of cement.

In another embodiment, the present invention provides novel compositions, which comprise:

- (a) a cementitious binder; and
- (b) the graft polymer of formula (IVa):



wherein:

- the polymer has from 5 to 1000 monomer units;
- each X independent of the other is hydrogen, an alkali metal ion, an alkaline earth metal ion, ammonium ion, or substituted ammonium;
- each -NH-Z- is a monomer unit derived from a compound which contains (i) a



AMENDED SHEET

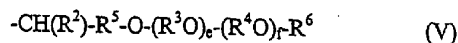
MBC-0344

21

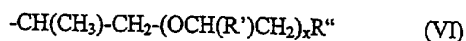
-NH₂ group and (ii) a -CO₂H or a -SO₃H group;

m, m', n, and n' are each independently 0 or an integer between 1 and 0, preferably 0 to 2;

a, b, c, d each reflect the mole fraction of each unit wherein the sum of a + b + c + d equals one, wherein a is a decimal value of 0 to 0.95, preferably 0 to 0.85, more preferably 0 to 0.75; b is a decimal value of 0 to 0.95, preferably 0 to 0.85, more preferably 0 to 0.75; c is a decimal value of 0.05 to 1, preferably 0.15 to 1, more preferably 0.25 to 1; and d is a decimal value of 0 to 0.5, preferably 0 to 0.25; and Q has a molecular weight of from 600-25,000 and has the formulae (V) or (VI):



wherein R² and R⁶ are, each independently, H, phenyl, substituted phenylene, or C₁ to C₄ linear or branched alkyl, such as methyl, ethyl, propyl, or butyl; R⁵ is phenyl or C₁ to C₄ linear or branched alkylene, such as methylene, ethylene, propylene, or butylene; R³ is ethylene, and R³O is present randomly or in block form; e is 1 to 300, preferably 11 to 300; R⁴ is propylene, and R⁴O is present randomly or in block form; f is 0 to 180, preferably with a mole ratio of R³:R⁴ of 100:0 to 40:60.



in which each R' independently of the other is H or methyl, R'' is a residue derived from a monohydric alcohol (e.g., methoxy, ethoxy, propoxy, etc.), and x has a value such that the molecular weight of the group of formula (VI) is from 600 to 25,000. The ratio of R' = methyl to R' = H typically ranges from 3:19 to 9:1.

Suitable cementitious binders include Portland cement, alumina cement, or blended cement, for example pozzolanic cement, slag cement, or other types, Portland cement being the preferred. The graft polymer of formula (IVa) is added to the present compositions in an amount from 0.001 to 5.0%, preferably from 0.01 to 3.0%, more preferably 0.05 to 1.0% by weight, based on the total weight of the cementitious binder.



AMENDED SHEET

MBC-0344

22

In a preferred embodiment, the graft polymer of formula (IVa) is the graft polymer of formula (IV).

When actually used, water is typically added to the present compositions. The water is typically added in an amount of 20 to 70 % by weight, preferably 25 to 50 % by weight, based on the total weight of the cementitious binder.

The present composition may further comprise an aggregate such sand, stone, slag, and mixtures thereof. Thus, the present composition may take the form of cement, mortar, concrete, etc.

Suitably, for example, the dispersant is preferably added prior to, during, or after initial mixing, to a cementitious composition mixture of a hydraulic cement, water, and aggregate, in which a reduced amount of water can be used compared to mixtures not containing the dispersant. The proportion of cement, water, and aggregate (if any) used in the cementitious composition mixture depends upon the end use desired, and is within the knowledge of the skilled artisan for a selected use.

The inventive polymer dispersant may be added to many different types of cementitious mixtures, and is not limited to the mixtures described above. Cementitious mixes in which the inventive polymer dispersant can be used as additives are mortars and concretes. The hydraulic binder used in the cementitious composition may be Portland cement, alumina cement, or blended cement, for example pozzolanic cement, slag cement, or other types, Portland cement being the preferred. Conventional admixtures in cement technology, such as set accelerators or retarders, frost protection agents, pigments and the like, may also be present in the cementitious mixes of the invention. The inventive polymer dispersant is capable of providing a Type A and Type F water-reducing effect, as defined in ASTM C494, Standard Specification for Chemical Admixtures for Concrete.

In another embodiment, the present invention provides a method for improving the flow properties of a cementitious composition, comprising adding an effective amount of the graft polymer of formula (IVa) to a cementitious composition which comprises a cementitious binder. The cementitious binder may be Portland cement, alumina cement, or blended cement, for example pozzolanic cement, slag cement, or other types, Portland cement being the preferred. The graft polymer of formula (IVa) is added to the cementitious composition in an amount from 0.001 to 5.0%,



AMENDED SHEET

preferably from 0.01 to 3.0%, preferably 0.05 to 1.0% by weight, based on the total weight of the cementitious binder. In a preferred embodiment, the graft polymer of formula (IV) is added to the cementitious composition.

Water may be added to the cementitious composition in an amount of 20 to 70 % by weight, preferably 25 to 50 % by weight, based on the total weight of the cementitious binder. However, it is to be understood that the water need not be added to the cementitious composition until just prior to pouring. Thus, the graft polymer may be mixed with the cementitious binder and the resulting mixture may be stored or sold as is with the addition of any water occurring later. Alternatively, the graft polymer may be added to the mixing water, prior to mixing with the cementitious binder, in an amount of from 0.001 to 5.0%, preferably from 0.01 to 3.0%, more preferably 0.05 to 1.0%, by weight, based on the total weight of the cementitious binder.

The cementitious composition to which the graft polymer of formula (IVa) is added may further comprise an aggregate such sand, stone, slag, and mixtures thereof. Thus, the cementitious composition may be a cement, mortar, concrete, etc. It should be understood that the aggregate need not be added to the cementitious composition until just prior to pouring, if any is to be added at all. Thus, the graft polymer may be mixed with the cementitious binder and the resulting mixture may be stored or sold as is with the addition of any aggregate occurring later.

In another embodiment, the present invention provides a method for reducing the amount of water which is needed to prepare a cementitious composition with suitable flow properties, comprising adding an effective amount of a graft polymer of formula (IVa) to a cementitious composition which comprises a cementitious binder. The cementitious binder may be Portland cement, alumina cement, or blended cement, for example pozzolanic cement, slag cement, or other types, Portland cement being the preferred. The graft polymer of formula (IVa) is added to the cementitious composition in an amount from 0.001 to 5.0%, preferably from 0.01 to 3.0%, preferably 0.05 to 1.0% by weight, based on the total weight of the cementitious binder. In a preferred embodiment, the graft polymer of formula (IV) is added to the cementitious composition.



AMENDED SHEET

MBC-0344

24

By means of this method it is possible to reduce the water of addition needed to achieve good flow properties. Thus, water may be added to the cementitious composition in an amount of 20 to 70 % by weight, preferably 25 to 50 % by weight, based on the total weight of the cementitious binder. However, it is to be understood that the water need not be added to the cementitious composition until just prior to pouring. Thus, the graft polymer may be mixed with the cementitious binder and the resulting mixture may be stored or sold as is with the addition of any water occurring later. Alternatively, the graft polymer may be added to the mixing water, prior to mixing with the cementitious binder, in an amount of from 0.001 to 5.0%, preferably from 0.01 to 3.0%, more preferably 0.05 to 1.0%, by weight, based on the total weight of the cementitious binder.

The cementitious composition may further comprise an aggregate such sand, stone, slag, and mixtures thereof. Thus, the cementitious composition may be a cement, mortar, concrete, etc. It should be understood that the aggregate need not be added to the cementitious composition until just prior to pouring, if any is to be added at all. Thus, the graft polymer may be mixed with the cementitious binder and the resulting mixture may be stored or sold as is with the addition of any aggregate occurring later.

Other features of the invention will become apparent in the course of the following descriptions of exemplary embodiments which are given for illustration of the invention and are not intended to be limiting thereof.

EXAMPLES

I. Methods:

A. Infrared Spectroscopy:

Infrared spectra of polysuccinimides and polyaspartates are measured by use of an FTIR spectrophotometer (Perkin Elmer, model 1600). Samples are mixed in KBr and disc pellets (13 mm) are made at 9000 lbs. for 3 minutes by use of a die (Spectratech) and press (Carver, Inc.).

Infrared spectra of grafted polyaspartates are measured by use of a Bio Rad FTIR (model FTS 175) with a single-bounce ATR (attenuated total reflectance) apparatus containing a silicon window. The spectra are ATR corrected.



AMENDED SHEET

B. Molecular Weight:

Molecular weights of polymers are determined by gel permeation chromatography. For polysuccinimides and polyaspartates, the standards are
5 polyaspartates made in-house by solid-phase methods (Asp₅ through Asp₆₀) and commercial polyaspartates (up to 32,000 MW; low-angle, laser light scattering, Sigma Chemical). A liquid chromatograph (Varian, model 5500) with a 7.5 mm x 30 cm column (G 4000 PW, Phenomenex) is used. The mobile phase is 0.01 M Tris, 0.1 M NaCl, pH 8.00, flow of 1 ml/min, with UV detection at 235 nm.

10 For grafted polyaspartates, the standards are polyethylene glycols (Polysciences). A Waters 600 liquid chromatograph, Alltech Varex MKIII detector, and Shodex OH Pak columns are used.

C. Phosphate Analysis:

Polymers that are synthesized in the presence of phosphoric catalysts/solvents
15 are cleansed by washing and dialysis. The removal of the catalyst/solvent is confirmed by use of the spectrophotometric assay for the phosphomolybdate complex which is formed in the presence of soluble inorganic phosphate (Eisenreich et al., Environmental Letters, vol. 9, pp. 43-53 (1975)). The polymers themselves also test negatively for phosphorus, following acid hydrolysis to release bound phosphorus, if any, as inorganic
20 phosphoric acid.

D. Cement Dispersancy:

Cement paste samples are prepared by mixing 500 g of Ashgrove I cement with water to obtain a final 0.35 water to cement ratio (w/w), including the water added with
25 the dispersant. The additive solution is then mixed with the cement paste by use of a mechanical mixer at 700 revolutions per minute for one minute. Flow is measured by transferring 300 g of paste to a 7.5 cm, 60-degree glass funnel. The amount of paste that flowed from the funnel versus time is recorded. Spread is measured by filling a cylindrical cup (4 cm diameter, 7 cm height) with paste, covering the open end with a
30 23 cm by 23 cm plate, inverting the cup, then placing it on a level surface. The cup is lifted and the average spread diameter is recorded.

II. Examples:

Preparation Example 1. Preparation of low molecular weight (MW) polysuccinimide and sodium polyaspartate by dry, thermal polymerization of aspartic acid.

A sample of 100 g of aspartic acid (Sigma Chemical Company) in a 600 ml beaker is placed in an oven equipped with a J-Kem temperature controller, model 250, and the temperature is maintained at $220^{\circ}\text{C} \pm 1^{\circ}\text{C}$. After 8 hours of reaction, the sample is removed, allowed to cool, and placed in a sample bottle. The product is collected in nearly quantitative yield (96% of theoretical) and is shown to be polysuccinimide by infrared spectroscopy (see Figure 1).

A sample of the polysuccinimide is converted to polyaspartate in aqueous solution by mild alkaline hydrolysis at pH 10 at 60°C for 2 hours via dropwise addition of 10 N NaOH. The solution is dialyzed (Spectrapor tubing, MW cutoff of 1000) against distilled water, then lyophilized to yield sodium polyaspartate, as also shown by infrared spectra (see Figure 2). The gel permeation MW is 2700. Polyaspartates made in-house by solid-phase peptide synthesis are used as standards.

Preparation Example 2. Preparation of high MW polysuccinimide and polyaspartate by thermal polymerization of aspartic acid in the presence of polyphosphoric acid.

A sample of 250 g of aspartic acid is placed in a pyrex baking dish (10.5" x 15" x 2") and preheated to 100°C . Next, 37.5 ml of polyphosphoric acid (Sigma Chemical, density = 2.06) is added and mixed manually with the aspartic acid on a hotplate at 80°C to form a paste. The elevated temperature rendered the polyphosphoric acid less viscous and pipettable, and permitted homogeneous mixing with the aspartic powder. The baking dish is then placed in a vacuum oven (VWR) at 190°C for 4.5 hours at a pressure of 38 mm of Hg. The product rises in the dish to form a very light crusty material that is easily scraped on to a cellulose filter (Whatman #3) and washed under gentle vacuum with distilled water. When the pH of the filtrate becomes neutral, the product polysuccinimide is dried and obtained in nearly quantitative yield. The polysuccinimide is a light, cream color. It is free of phosphoric acid as measured spectrophotometrically.

A sample of this polysuccinimide is hydrolyzed at 80°C and pH 10 for 2 hours by dropwise addition of 10 N NaOH. The resulting solution is dialyzed against distilled

water then lyophilized to yield sodium polyaspartate, as shown by the IR spectrum. The gel permeation MW is 30,000. Commercial polyaspartates (Sigma Chemical) are used as standards.

5 Example 1. Grafting of medium MW methoxy polyoxyalkylene amine on to low MW polyaspartic acid by thermal treatment at 180°C.

The polysuccinimide prepared in Preparation Example 1 is converted to sodium polyaspartate by mild alkaline hydrolysis, then to the polyaspartic acid by treatment with an insoluble acidic resin (Amberlyst 15). The solution of polyaspartic acid is
10 separated from the resin by filtration. The concentration of polyaspartic acid is determined from the weight of solids on drying of known volumes of the solution.

A sample of 0.54 g of polyaspartic acid in 6.55 ml of water is placed in a 50 ml round-bottom flask. Next, 1.66 g of methoxy polyoxyalkylene amine (JEFFAMINE M-2070, MW 2000, mole ratio of propyleneoxide:ethyleneoxide 10:32, 0.47 meq NH₂/g,
15 Huntsman Chemical) is added, along with 0.005 g of 4-methoxyphenol as an antioxidant. The reactant molar ratio of aspartic acid residues to polyoxyalkylene amine is 6:1. All reactants are soluble.

The flask is purged with dry N₂ and heated at 180°C for 1.5 hours, producing a viscous amber oil. The oil is dissolved in aqueous NaOH at room temperature, yielding
20 an aqueous solution at neutral pH of 20% solids by weight.

Infrared analysis of the product reveals a spectrum with peaks indicative of both amide and imide residues, and a large peak associated with derivatized side chains. Similarly, gel permeation chromatography of the product shows the presence of the derivatized polymer with only residual amounts of unreacted polymeric monoamine and
25 with an estimated density of underivatized:derivatized residues of 5:1.

Example 2. Grafting of low MW methoxy polyoxyalkylene amine on to high MW polyaspartic acid by thermal treatment at 180°C.

The polysuccinimide prepared in Preparation Example 2 is converted to
30 polyaspartic acid by the method of Example 1.

A sample of 1.13 g of polyaspartic acid in 6.21 ml of water is placed in a 50 ml round-bottom flask. Next, 2.01 g of methoxy polyoxyalkylene amine (JEFFAMINE M-

1000, MW 1000, mole ratio of propyleneoxide:ethyleneoxide 3:19, 0.98 meq NH_2/g , Huntsman Chemical) is added, along with 0.02 g of 4-methoxyphenol. The reactant molar ratio of aspartic acid residues to polyoxyalkylene amine is 5:1.

The clear solution in the flask is purged with N_2 and heated at 180°C for 2
5 hours, producing an homogeneous, viscous oil. The oil is dissolved in aqueous NaOH at room temperature, yielding an aqueous solution at pH 7.6 of 20% solids by weight.

Infrared and gel permeation chromatography analysis shows a mixed amide/imide polymer having derivatized side chains (see Figure 3), with only residual amounts of unreacted polymeric monoamine and with an estimated graft density of
10 underivatized:derivatized residues of 4:1 (see Figure 4).

Example 3. Grafting of medium MW methoxy polyoxyalkylene amine on to high MW polyaspartic acid by thermal treatment at 180°C .

The procedures and reaction conditions of Example 2 are repeated except for use of 4.18 g of methoxy polyoxyalkylene amine (JEFFAMINE M-2070). The reactant
15 molar ratio of aspartic acid residues to polyoxyalkylene amine is 5:1.

Again, an amber, homogeneous oil is produced and solubilized in NaOH at room temperature to give an aqueous solution at pH 7.0 of 40% solids by weight. Infrared analysis shows a mixed amide/imide polymer having derivatized side chains, with an estimated graft density of underivatized:derivatized residues of 4:1.
20 Accordingly, the graft efficiency is high, as shown by gel permeation chromatography profiles that reveal only residual amounts of unreacted polymeric monoamine.

Comparative Example 1. Grafting of medium MW methoxy polyoxyalkylene amine on to low MW polysuccinimide by nucleophilic addition in dimethyl formamide.

25 A sample of 4.0 g of the polysuccinimide prepared in Preparation Example 1 is dissolved in 20 g of dimethyl formamide (density at 20°C , 0.9487, Burdick and Jackson). Next, an amount of 21.9 g of methoxy polyoxyalkylene amine (JEFFAMINE M-2070) is added with magnetic stirring. The reaction is allowed to proceed for 23 hours at 80°C . The reactant molar ratio of succinimide residues to polyoxyalkylene
30 amine is 4:1.

The product is separated as a reddish oil in ethyl ether. The ether is decanted and the oil dissolved in acetone, followed by further washing with ethyl ether. The

product oil is then solidified by cooling on dry ice, followed by decanting of the ether and drying in vacuo. The product is dissolved in aqueous NaOH at 75°C for 1.5 hours to give a solution at pH 7 of 40 % solids by weight.

Based on the analysis of the peak sizes in the gel permeation profile (see Figure 5), the graft efficiency is about 40%, with about 60% of the polyoxyalkylene amine ungrafted. The purified product is estimated to have a graft density with a residue ratio of aspartate:grafted aspartate of 9:1. The infrared spectrum of the purified product reveals a polymer having an amide structure with no measurable imide residues and a prominent peak at 1104 cm^{-1} associated with the derivatized side chains (see Figure 6).

10

Example 4. Utility of grafts of polyaspartic acid and polyoxyalkylene amine as a dispersant in cement.

The performance of the grafted polyaspartic acids of the present invention as additives to cement is assessed. The paste samples are mixed as described in the Cement Dispersancy section above for 1 minute prior to testing, with the exception of the sample treated with material of example 2, which is mixed for 10 minutes, then tested. The products of examples 1, 2, and 3 are tested without purification, the product of Comparative Example 1 is tested following purification and mild alkaline hydrolysis to convert succinimide residues to the aspartate form. All cement paste mixtures are tested at a water to cement ratio of 0.35 (w/w). Representative results of the effect on the flow and spread of cement paste mixtures are shown in the following table.

20

TABLE

Material	% by weight cement powder	Initial Flow (g/sec)	Initial Spread (cm)
graft of polyaspartic acid: polyoxyalkylene amine of Example 1	0.1 %	76	12
graft of polyaspartic acid: polyoxyalkylene amine of Example 2	0.1 %	50	10
graft of polyaspartic acid: polyoxyalkylene amine of Example 3	0.1 %	46	10
graft of polysuccinimide: polyoxyalkylene amine of Comparative Example 1	0.1 %	96	15
control, no additive	0 %	1	0

Comparative Example 2. Grafting of low MW methoxy polyoxyalkylene amine on to low MW polysuccinimide acid by nucleophilic addition in the absence of solvent.

5 The polyoxyalkylene amine is itself a colorless, low-viscosity liquid when warmed. In this example, an attempt is made to derivatize polysuccinimide by nucleophilic addition of a monoamino polyoxyalkylene by mixing of the reactants in the absence of water or an organic solvent.

10 One gram of the polysuccinimide of the type prepared in Preparative Example 1 is combined in a glass test tube with 1.74 g of polyoxyalkylene monoamine (JEFFAMINE M-1000, 0.98 meq NH_2/g) to form a slurry with a reactant ratio of 6:1 succinimide residues:polymeric amine. The test tube is purged with dry N_2 and heated for 24 hours at 80°C.

15 The product is next treated with 0.4 g of 50% NaOH plus 5 ml of water and warmed at 40°C to convert the succinimide residues to the aspartate form. The

resulting, two-phase solution has a pH of approximately 7 at about 36% by weight solids. The formation of two solution phases at higher solids content coincides with poor graft efficiencies.

Gel permeation chromatography of the final product shows poor grafting of the polymeric amine to the polyaspartate, with most of the polyoxyalkylene amine eluting separately from the polyaspartate (see Figure 7). The performance of the final product as an additive to cement is poor.

Comparative Example 3. Grafting of low MW methoxy polyoxyalkylene amine on to low MW polyaspartic acid by nucleophilic addition in water.

The reactants and procedures of Comparative Example 2 are repeated except that 4 ml of water is added to the mixture of polysuccinimide and polyoxyalkylene amine. The mixture is warmed to 50°C and maintained at that temperature for 24 hours. The product again is converted to the aspartate form by mild alkaline hydrolysis, yielding a two-phase solution of approximately 40 % solids by weight.

Gel permeation chromatography of the final product shows poor grafting of the polymeric amine to the polyaspartate, with most of the polyoxyalkylene amine eluting separately from the polyaspartate. The performance of the final product as an additive to cement is poor.

Comparative Example 4. Copolymerization of sodium aspartate and monoamino polyoxyalkylene by thermal treatment at 180°C.

A mixture of 10 g of aspartic acid (0.0751 mole) in 28 ml of water plus 26.6 g of monoamino polyoxyalkylene (JEFFAMINE M-2070, 0.47 meq NH_2/g , 0.0125 mole NH_2) is mixed with 4.7 g of 50% NaOH. The salt of sodium aspartate and polyoxyalkylene amine is formed by lyophilization, and a sample of 0.05 g of 4-methoxyphenol is added as an antioxidant. This mixture is heated in a flask in an oil bath at 180°C for 20.5 hours. Subsamples are taken at 2, 4, 7, and 13.5 hours.

The samples are water-soluble, and are not further treated by mild alkaline hydrolysis. Gel permeation of the products reveals little change in the elution profiles versus time of reaction, suggesting poor chain lengthening (see Figure 8). The performance of the final product as an additive to cement is poor.

Comparative Example 5. Grafting of medium MW methoxy polyoxyalkylene amine on to low MW polysuccinimide by combined nucleophilic and thermal treatment at 180°C.

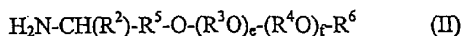
Two grams of the polysuccinimide prepared in Preparative Example 1 are mixed with 7.30 g of polyoxyalkylene monoamine (JEFFAMINE M-2070) plus 0.5 ml of
5 water, 0.1 g of 50% NaOH solution, and 0.02 g of 4-methoxyphenol in a 50 ml round-bottom flask. The mixture is heated at 180°C under a stream of N₂ for 12 hours. The reactant molar ratio of succinimide residues to polyoxyalkylene amine is 6:1.

The product is dissolved in aqueous NaOH at room temperature to yield a solution of 40% solids by weight. Gel permeation chromatography of the product
10 shows poor graft efficiency, with a dominant peak of unreacted polyoxyalkylene amine. The performance of the final product as an additive to cement is poor.

Obviously, numerous modifications and variations of the present invention are possible in light of the above teachings. It is therefore to be understood that, within the scope of the appended claims, the invention may be practiced otherwise than as
15 specifically described herein.

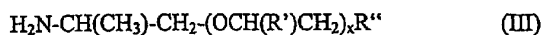
CLAIMS:

1. A method for preparing a graft polymer, said method comprising heating an intimate mixture which is substantially free of water or organic solvent of: (a) a polyamide which contains pendant carboxylic acid groups; and (b) a compound which contains an amino group selected from the group consisting of simple amines, C₂₋₄ alkylene amines, polyoxyalkylene monoamines, aminosulfonates, aminophosphonates and hydrophobic amino acids.
2. The method of Claim 1 wherein the amino group has a molecular weight in the range of from 600 to 25,000.
3. The method of either Claim 1 or Claim 2, wherein said polyamide which contains pendant carboxylic acid groups is selected from the group consisting of polyaspartic acid, polyglutamic acid, and copolymers of aspartic acid and glutamic acid.
4. The method of either Claim 1 or Claim 2, wherein said polyamide which contains pendant carboxylic acid groups is polyaspartic acid.
5. The method of any one of Claims 1 to 4, wherein said compound which contains an amino group has the formula (II):



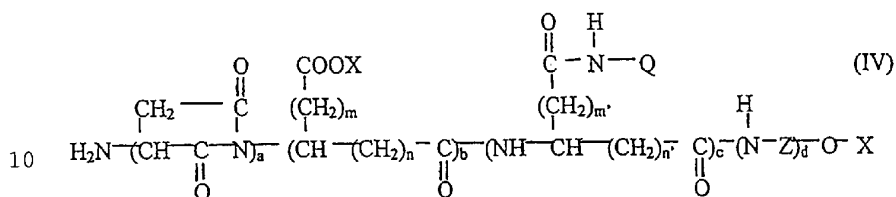
wherein R² and R⁶ are, each independently, H, phenyl, substituted phenyl, or C₁ to C₄ linear or branched alkyl; R⁵ is phenylene or C₁ to C₄ linear or branched alkylene; R³ is ethylene; R³O is present randomly or in block form; e is 1 to 300; R⁴ is propylene; R⁴O is present randomly or in block form; and f is 0 to 180; the compound of formula II having a molecular weight of from 600 to 25,000.

6. The method of any one of Claims 1 to 4, wherein said compound which contains an amino group has the formula (III):



in which each R' independently of the other is H or methyl, R'' is a residue derived from a monohydric alcohol, and x has a value such that the molecular weight of said compound of formula (III) is from 600 to 25,000.

7. A polymer of formula (IV):



wherein:

the polymer has from 5 to 1000 monomer units;

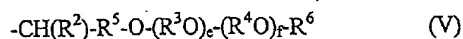
each X independent of the other is hydrogen, an alkali metal ion, an alkaline earth metal ion, ammonium ion, or substituted ammonium;

each -NH-Z- is a monomer unit derived from a compound which contains (i) a -NH₂ group and (ii) a -CO₂H or a -SO₃H group;

m, m', n, and n' are each independently 0 or an integer between 1 and 20;

a, b, c, d each reflect the mole fraction of each unit wherein the sum of a + b + c + d equals one, wherein a is a decimal value of 0 to 0.75; b is a decimal value of 0 to 0.75; c is a decimal value of 0.25 to 1; and d is a decimal value of 0 to 0.5; and

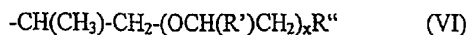
Q is selected from moieties of the formulae (V):



wherein R² and R⁶ are, each independently, H, phenyl, substituted phenyl, or C₁ to C₄ linear or branched alkyl; R⁵ is phenylene or C₁ to C₄ linear or branched alkylene; R³ is ethylene; R³O is present randomly or in block form; e is 1 to 300; R⁴ is propylene; R⁴O is present randomly or in block form; and f is 0 to 180;



and (VI):



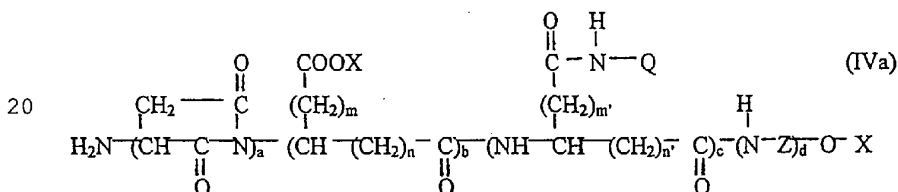
5 in which each R' independently of the other is H or methyl, R'' is a residue derived from a monohydric alcohol, and x has a value such that the molecular weight of the group of formula (VI) is from 600 to 25,000.

8. The polymer of Claim 7, wherein m , m' , n , and n' are each independently
10 0 to 2.

9. The polymer of either claim 7 or claim 8 wherein e is 11 to 300.

10. A composition, comprising:

- 15 (a) a cementitious binder; and
(b) a polymer of formula (IVa):



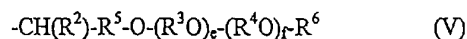
wherein:

- 25 the polymer has from 5 to 1000 monomer units;
each X independent of the other is hydrogen, an alkali metal ion, an alkaline earth metal ion, ammonium ion, or substituted ammonium;
each $-\text{NH}-\text{Z}-$ is a monomer unit derived from a compound which contains (i) a $-\text{NH}_2$ group and (ii) a $-\text{CO}_2\text{H}$ or a $-\text{SO}_3\text{H}$ group;
30 m , m' , n , and n' are each independently 0 or an integer between 1 and 20;



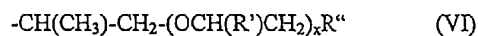
a, b, c, d each reflect the mole fraction of each unit wherein the sum of a + b + c + d equals one, wherein a is a decimal value of 0 to 0.95; b is a decimal value of 0 to 0.95; c is a decimal value of 0.05 to 1; and d is a decimal value of 0 to 0.5; and
Q is a moiety selected from the formulae (V):

5



wherein R^2 and R^6 are, each independently, H, phenyl, substituted phenyl, or C_1 to C_4 linear or branched alkyl; R^5 is phenylene or C_1 to C_4 linear or branched alkylene; R^3 is ethylene; R^3O is present randomly or in block form; e is 1 to 300;
10 R^4 is propylene; R^4O is present randomly or in block form; and f is 0 to 180;
and (VI):

15



in which each R' independently of the other is H or methyl, R'' is a residue derived from a monohydric alcohol, and x has a value such that the molecular weight of the group of formula (VI) is from 600 to 25,000.

20

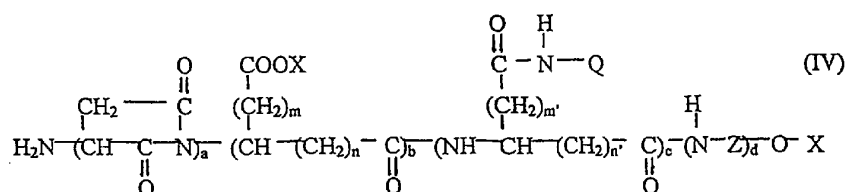
11. The composition of claim 10 wherein m, m', n, and n' are each independently 0 to 2.

12. The composition of either claim 10 or claim 11 wherein e is 71 to 300.

25

13. A method of improving the flow properties of a cementitious composition, comprising adding an effective amount of a polymer of formula (IV):

30



wherein:

the polymer has from 5 to 1000 monomer units;



each X independent of the other is hydrogen, an alkali metal ion, an alkaline earth metal ion, ammonium ion, or substituted ammonium;

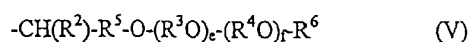
each -NH-Z- is a monomer unit derived from a compound which contains (i) a -NH₂ group and (ii) a -CO₂H or a -SO₃H group;

5 m, m', n, and n' are each independently 0 or an integer between 1 and 20;

a, b, c, d each reflect the mole fraction of each unit wherein the sum of a + b + c + d equals one, wherein a is a decimal value of 0 to 0.95; b is a decimal value of 0 to 0.95; c is a decimal value of 0.05 to 1; and d is a decimal value of 0 to 0.5; and

Q is a moiety selected from the formulae (V):

10



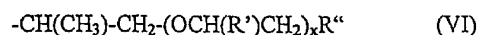
wherein R² and R⁶ are, each independently, H, phenyl, substituted phenyl, or C₁ to C₄ linear or branched alkyl; R⁵ is phenylene or C₁ to C₄ linear or branched alkylene; R³ is

15 ethylene; R³O is present randomly or in block form; e is 1 to 300;

R⁴ is propylene; R⁴O is present randomly or in block form; and f is 0 to 180;

and (VI):

20



in which each R' independently of the other is H or methyl, R'' is a residue derived from a monohydric alcohol, and x has a value such that the molecular weight of the group of formula (VI) is from 600 to 25,000;

to a cementitious composition which comprises a cementitious binder.

25

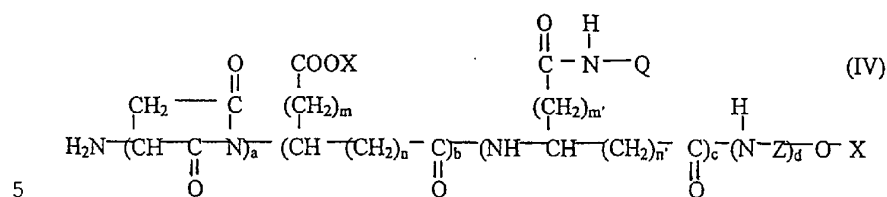
14. The method of Claim 13 wherein m, m', n, and n' are each independently 0 to 2.

15. The method of either Claim 13 or Claim 14 wherein e is 71 to 300.

30

16. A method for reducing the amount of water which is needed to prepare a cementitious composition with suitable flow properties, comprising adding an effective amount of polymer of formula (IV):





wherein:

the polymer has from 5 to 1000 monomer units;

each X independent of the other is hydrogen, an alkali metal ion, an alkaline

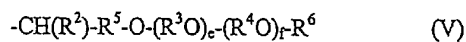
earth metal ion, ammonium ion, or substituted ammonium;

each -NH-Z- is a monomer unit derived from a compound which contains (i) a -NH₂ group and (ii) a -CO₂H or a -SO₃H group;

m, m', n, and n' are each independently 0 or an integer between 1 and 20;

a, b, c, d each reflect the mole fraction of each unit wherein the sum of a + b + c + d equals one, wherein a is a decimal value of 0 to 0.95; b is a decimal value of 0 to 0.95; c is a decimal value of 0.05 to 1; and d is a decimal value of 0 to 0.5; and

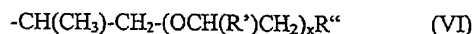
Q is a moiety selected from the formulae (V):



wherein R² and R⁶ are, each independently, H, phenyl, substituted phenyl, or C₁ to C₄ linear or branched alkyl; R⁵ is phenylene or C₁ to C₄ linear or branched alkylene; R³ is ethylene; R³O is present randomly or in block form; e is 1 to 300;

R⁴ is propylene; R⁴O is present randomly or in block form; and f is 0 to 180;

and (VI):



in which each R' independently of the other is H or methyl, R'' is a residue derived from a monohydric alcohol, and x has a value such that the molecular weight of the group of formula (VI) is from 600 to 25,000;

to a cementitious composition which comprises a cementitious binder.



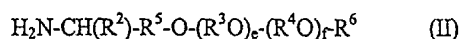
17. The method of Claim 16 wherein m, m', n, and n' are each independently 0 to 2.

18. The method of either Claim 16 or Claim 17 wherein e is 71 to 300.

5

19. A graft polymer, produced by a process comprising heating an intimate mixture of: (a) a polyamide which contains pendant carboxylic acid groups; and (b) a compound which contains an amino group, to obtain said graft polymer, wherein said compound which contains an amino group has a molecular weight of from 600-25,000 and corresponds to the formula (II):

10



15

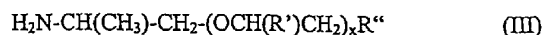
wherein R^2 and R^6 are, each independently, H, phenyl, substituted phenyl, or C_1 to C_4 linear or branched alkyl; R^5 is phenylene or C_1 to C_4 linear or branched alkylene; R^3 is ethylene; R^3O is present randomly or in block form; e is 1 to 300; R^4 is propylene; R^4O is present randomly or in block form; and f is 0 to 180; and

20

wherein said polyamide which contains pendant carboxylic acid groups and said compound which contains an amino group are present in relative amounts such that said graft polymer has a graft density of at least 25%.

25

20. A graft polymer, produced by a process comprising heating an intimate mixture of: (a) a polyamide which contains pendant carboxylic acid groups; and (b) a compound which contains an amino group, to obtain said graft polymer, wherein said compound which contains an amino group has the formula (III)



30

in which each R' independently of the other is H or methyl, R'' is a residue derived from a monohydric alcohol, and x has a value such that the molecular weight of said compound of formula (III) is from 600 to 25,000; and



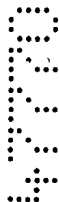
wherein said polyamide which contains pendant carboxylic acid groups and said compound which contains an amino group are present in relative amounts such that said graft polymer has a graft density of at least 25%.

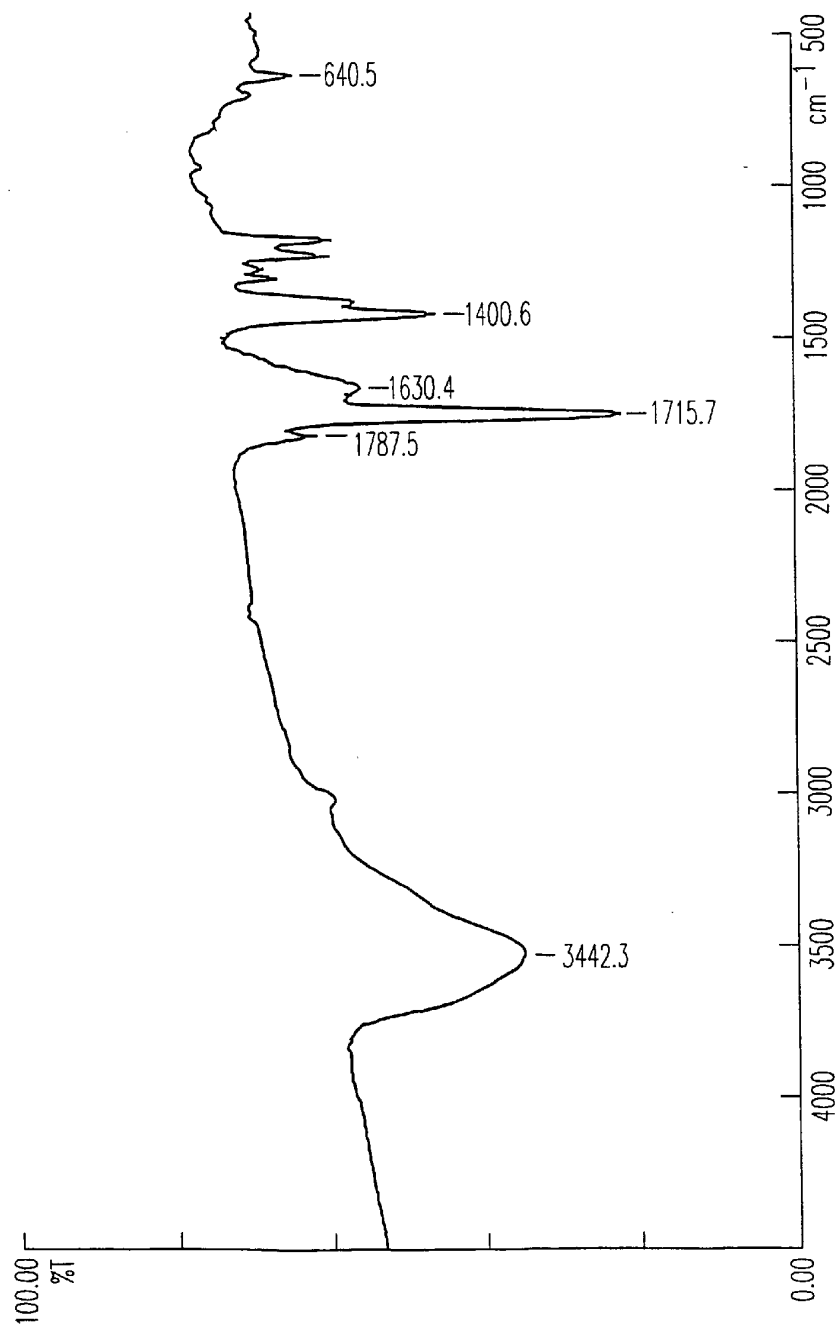
5 21. A method substantially as hereinabove defined with reference to the examples.

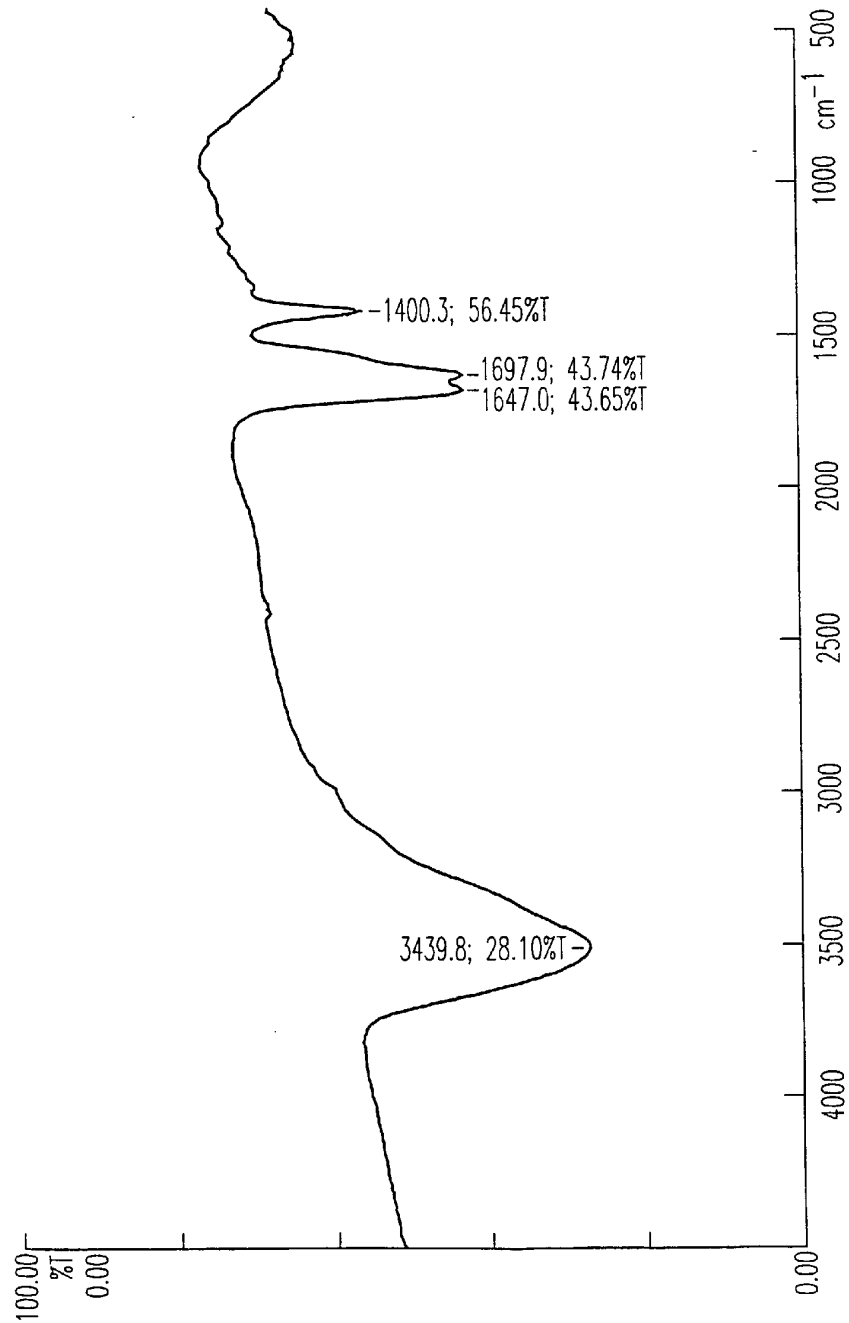
 22. A polymer substantially as hereinabove defined with reference to the examples.

10

 23. A composition substantially as hereinabove defined with reference to the examples.



**FIG. 1**

*FIG. 2*

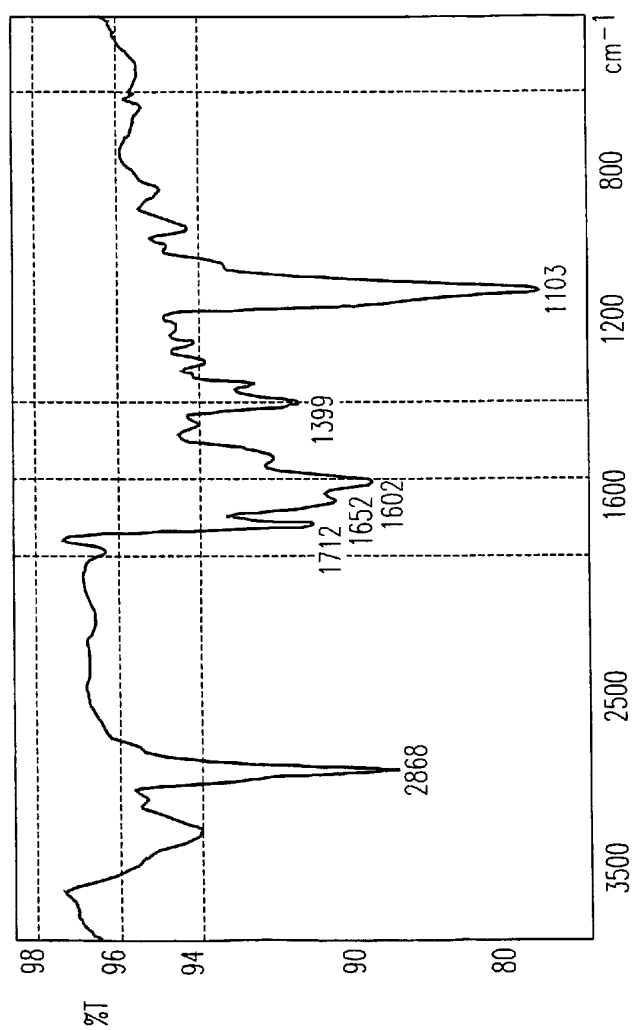
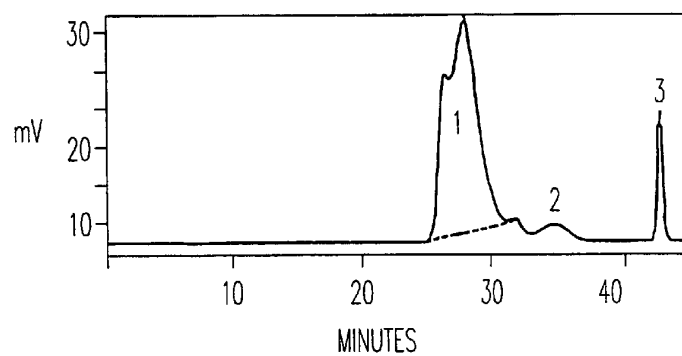
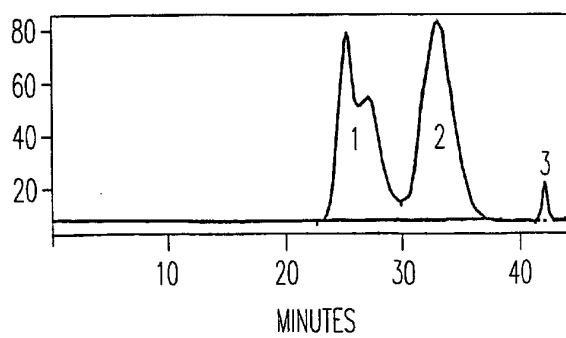
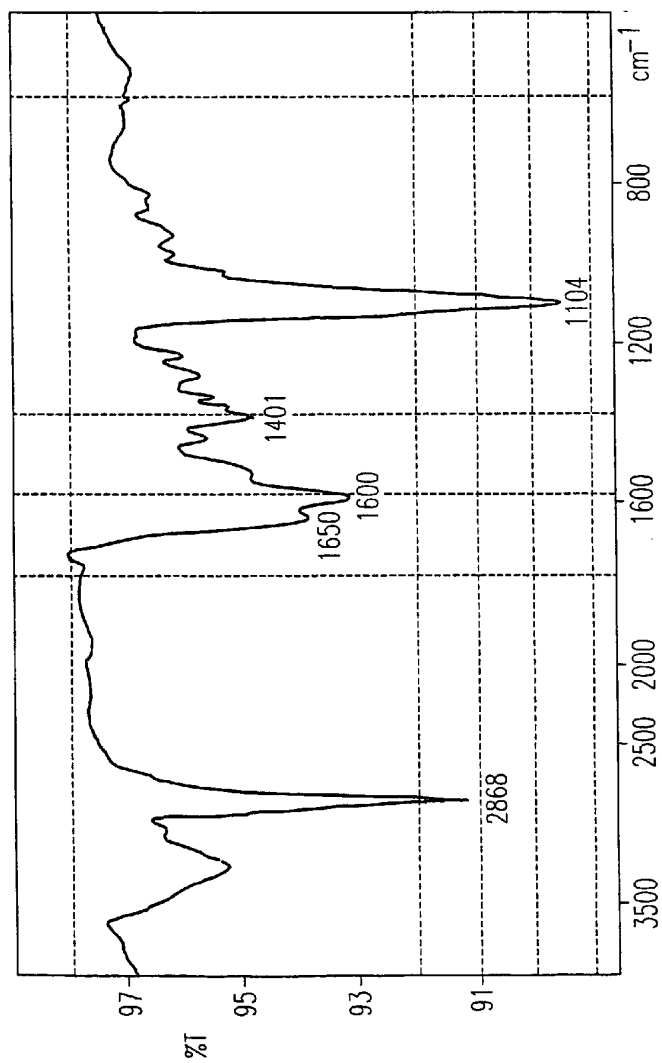
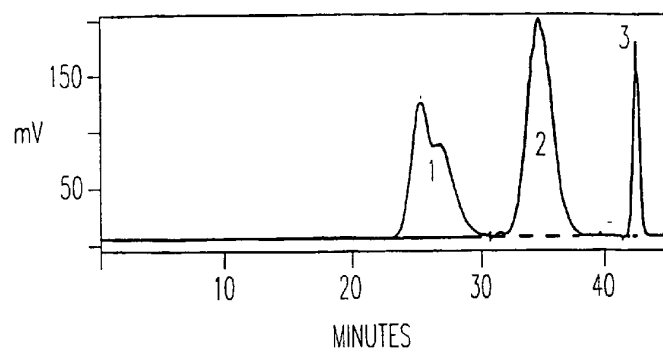
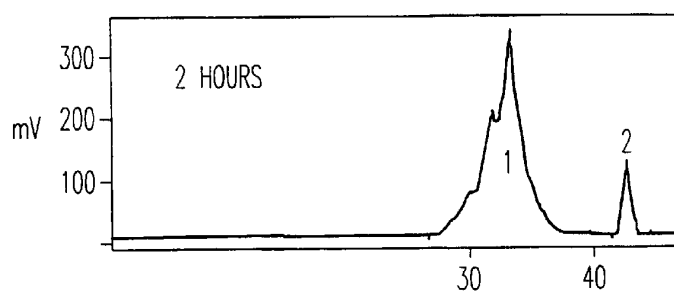
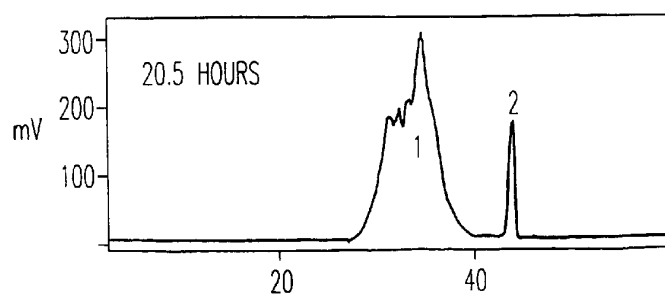
*FIG. 3*

FIG. 4*FIG. 5*

**FIG. 6**

6/6

*FIG. 7**FIG. 8A**FIG. 8B*

SUBSTITUTE SHEET (RULE 26)