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(54) Title: PROCESS FOR THE PREPARATION OF POLYSACCHARIDE NANOPARTICLES

(57) **Abstract:** The process for the preparation of nanoparticles from polysaccharides and derivatives thereof, by their specific partial oxidation to produce aldehyde groups and attach compounds with amino or other group with the R-NH₂ bond which react with aldehyde groups, comprising oxidizing the polysaccharide or derivative thereof by a known method to obtain aldehyde groups until the oxidation degree of 0,1% to 80% of the sugar rings is obtained, then adding at least one nanoparticle-forming agent which after attachment of the aldehyde group exhibits hydrophobic properties, selected from the group comprising: aliphatic or aromatic organic amines containing from 4 to 20 carbon atoms, amides and hydrazides of aliphatic and aromatic organic acids containing from 4 to 20 carbon atoms, hydrophobic amino acids, phosphatidylethanolamine, and at least one active substance containing at least one amino, amido or hydrazide group, to the solution of the oxidized polysaccharide in water or a mixture of water and an organic solvent, carrying the reaction out at pH of the solution from 1 to 9, at the temperature of from 10 to 100°C, wherein the total molar ratio of amino groups to aldehyde groups is from 20 to 0,5.

Process for the preparation of polysaccharide nanoparticles

The invention provides a process for the preparation of nanoparticles from polysaccharides and derivatives thereof by specific oxidation of polysaccharide chains and attaching hydrophobic compounds, including medicaments.

Conjugates of compounds from the group of polysaccharides and chemical substances which exhibit therapeutic activity and include amino groups are known. For example, WO 2012/004007 discloses conjugates of hydroxyethylated derivative of starch and various medicaments. WO 03/000738 discloses antibiotic/starch conjugates, where the antibiotic is attached to the reducing end of a polysaccharide through a peptide bond. Attachment is obtained by oxidation of the starch derivative with I_2 on its reducing end, in an alkaline aqueous solution, followed by coupling of the oxidized derivative with the antibiotic in an organic solution. Further, from the international application WO03/15826 a pharmaceutical composition is known for inhibiting metastases or preventing recurrence of a malignant cancer, which comprises as an active principle a polysaccharide with a carboxy group linked to an active substance with anticancer activity through amino acid or peptide comprised of 2 to 8 amino acids. The application WO03/074087 pertains to coupling proteins with a starch-derived modified polysaccharide, with the binding interaction between the modified polysaccharide and the protein comprising a covalent bond resulting from a coupling reaction between a terminal aldehyde group of the modified polysaccharide and a functional group of the protein able to react with said aldehyde group. The invention provides also pharmaceutical formulations which include coupled compounds prepared by coupling and use of said compounds for preventive or therapeutic treatment of humans or animals.

From the international application No. WO 2011/069475 a process is known for the preparation of oxidized hyaluronic acid derivatives and a process for modifying such derivatives. According to the process of that application, hyaluronic acid is oxidized with a specific oxidizing agent TEMPO, to obtain an acid derivative with aldehyde groups. The derivative is then used for linking with amines, diamines, amino acids, peptides and other amino-containing compounds. Such linking is implemented by reductive amination with $NaBH_3CN$, in water or a mixture of water and an organic solvent.

The above-mentioned inventions provide couplings (conjugates) of polysaccharides and various kinds of therapeutic substances, however, any of the solutions was aimed at obtaining polysaccharide nanoparticles. Meanwhile, nanoparticles are at present intensively studied as potential carriers for medicaments, due to a number of novel desired properties [Biodegradable nanoparticles are excellent vehicle for site directed in-vivo delivery of drugs and vaccines, Mahaparto A., Singh K., Journal of Nanobiotechnology, 9, 2011]. Nanoparticles with diameters of about 50 nm to about 200 nm, having suitable surface properties, could circulate for a long time in the blood avoiding elimination by kidneys, liver or spleen filtration (*long circulating particles, stealth particles*). The surface of such nanoparticles should induce neither a response of the immune system nor aggregation of small plasma proteins— opsonins. In that instance, a surface of hydrogel properties, as created by highly hydrophilic polymers such as polyethylene glycol, polysaccharides, polyvinyl alcohol, is particularly desirable. Polysaccharides are especially desirable due to their frequently natural origin, biodegradability and similarity to substances occurring in the body. Such long-circulating nanoparticles tend to accumulate in areas of tumors or inflammations (*passive targeting*) [Therapeutic Nanoparticles for Drug Delivery in Cancer, Kwangjae Cho, Xu Wang, Shuming Nie, Zhuo Chen, Dong M. Shin, Clin. Cancer Res., 2008 14; 1310]. The effect is due to the fact that cell membranes of endothelial cells padding the circulatory system are tightly sealed by appropriate proteins and a gap between them is several nanometers wide. In the tumor or inflammation area the gaps are much wider and reach several hundred nanometers. This makes the nanoparticles accumulate in the gaps and „leak” from the circulation into the surrounding afflicted tissue, including the tumor. Such passive accumulation of nanoparticles in the disease afflicted area allows to increase drug concentration in the areas, enhances efficacy of the treatment and reduces side effects. Additional feature of the nanoparticles is their ability to be surface-modified with suitable proteins, metabolites or antibodies to exhibit active affinity to specific cell types, including tumor cells. This allows to deliver medicaments primarily to the afflicted cells. It is desirable to prepare the surface of nanoparticles from polysaccharides due to the fact that tumor cells exhibit significantly increased demand for glucose (the Warburg effect) which in turn allows to obtain increased affinity of polysaccharide nanoparticles to tumor cells. Such polysaccharide nanoparticles with a drug would penetrate more efficiently the cancer cells and kill them, and when labeled with fluorescent marker, they become an efficient diagnostic tool. Another important application of nanoparticles is gene therapy. A nanoparticle

containing a RNA or DNA fragment is able to penetrate a cell and influence the gene-reading processes occurring in the cell. Hope arises to heal genetic diseases.

There are numerous methods for preparation of nanoparticles, but unfortunately most of them are very complex and require application of drastic conditions (ultrasounds, high temperatures), aggressive chemical compounds, toxic organic solvents or surface-active compounds [Biodegradable nanoparticles are excellent vehicle for site directed in-vivo delivery of drugs and vaccines, Mahaparto A., Singh K., Journal of Nanobiotechnology, 9, 2011]. Nanoparticles for therapeutic use should be non-toxic and most preferably biodegradable. Polysaccharides make a very good material for the preparation of such nanoparticles, due to their biocompatibility and biodegradability [Lemarchand C., R. Gref, P. Couvreur, Polysaccharide-decorated nanoparticles, European Journal of Pharmaceutics and Biopharmaceutics 58, 2004]. However, known methods for the preparation of nanoparticles from polysaccharides by attaching hydrophobic groups are complex and require use of surface-active materials or aggressive chemicals [Nanoparticles of hydrophobically modified dextrans as potential drug carrier system, Aumelas A., Serrero A., Durand, E., Dellacherie E., Leonard M., Colloids and Surfaces B, 59, 2007]. Nanoparticles such prepared must be further purified for a long period of time as a consequence of toxic properties of the compounds.

The invention was directed to a process for the preparation of polysaccharide nanoparticles in mild conditions, to allow for covalent bonding of therapeutic compounds which are sensitive to aggressive environment.

The process for the preparation of nanoparticles from polysaccharides and derivatives thereof, by their specific partial oxidation to produce aldehyde groups and attach compounds with amino or other group with the R-NH₂ bond which react with aldehyde groups of the invention is characterized in that the polysaccharide or derivative thereof is oxidized by a known method to give aldehyde groups until the oxidation degree of 0,1% to 80% of the sugar rings is obtained, then at least one nanoparticle-forming agent, which is an organic chemical compound with a R-NH₂ bond that after attachment of the aldehyde group exhibits hydrophobic properties, and at least one active substance comprising at least one R-NH₂ or N-H bond is added to the solution of the oxidized polysaccharide in water or a mixture of water and an organic solvent, the reaction being conducted at the pH of the solution of 1 to 9, at the temperature of 10 to 100°C, most preferably 20-60°C, where the total molar ratio of amine groups to aldehyde groups is from 20 to 0,5. The nanoparticle-forming agent is selected from a group comprising: aliphatic or aromatic organic amines comprising from 4 to 20 carbon atoms, amides and hydrazides of aliphatic and aromatic organic acids comprising from 4 to 20

carbon atoms, hydrophobic amino acids, phosphatidylethanolamine. The active substance contains an amino, amido or hydrazide group.

The nanoparticle-forming agent can be added simultaneously with the active substance or after the active substance is added. It is preferable to add the active substance first, since the reaction occurs slower after the nanoparticles are folded resulting possibly in incomplete addition of the active substance and loss thereof in the purification of the product. Sometimes, the active substance could be accumulated in the nanoparticles in the nanocrystal forms precipitated in hydrophobic areas created by the folding agent, and then it is not covalently, but physically bound and could be added in greater amounts. This is the case when it is poorly water-soluble around neutral pH, and well soluble at pH lower than 7.

Both forming agent and the active substance is preferably introduced in the form of salts that are readily water-soluble, e.g. hydrochlorides. If amine is introduced as the more readily soluble salt (e.g. hydrochloride), pH lowers in the course of the reaction, and the solution is then slowly neutralized with the aqueous base solution. The optimal pH is dependent on alkalinity of the amine employed. Generally, the increase in the hydrogen ion concentration activates the aldehyde group, but also results in the drop in concentration of the free non-protonated amine; the optimal pH of the first order reaction lies between 4 and 6, further increase in pH allows to complete the process by decreasing the amine cation concentration.

The process for the preparation of nanoparticles could also be conducted in organic or mixed solvents, such as water/DMSO, water/acetonitrile, water/ether. To increase resistance to hydrolysis of the obtained nanoparticles, reduction of the formed bonds with NaBH₄ or NaBH₃CN is utilized, also in the aqueous solution and under mild conditions.

Preferably, as a nanoparticle-forming agent the following are employed: butylamine, pentylamine, hexylamine, octylamine, decylamine, dodecylamine, tetradecylamine, hexadecylamine, cyclohexylamine, benzylamine, ethylphenylamine, sphingosines, oleic acid amide, palmitic acid amide, stearic acid hydrazide, palmitic acid hydrazide, oleic acid hydrazide, leucine, isoleucine, valine, methionine, alanine, phenylalanine or cephalin (phosphatidylethanolamine). It is preferable to add the forming agent as an aqueous solution of an amine salt, for example hydrochloride, nitrate or sulfate, since salts of the instant amines are usually better soluble in water.

Preferably, as an active substance comprising amino, amido or hydrazide group drugs are used which contain such groups, such as daunorubicin, doxorubicin, aminoacridins and derivatives thereof (like amsacrine), cisplatin and derivatives thereof, methotrexate,

cytarabine, gemcitabine, dapsone, acyclovir, azidothymidine, 5-fluorouracil, mercaptopurine, imatinib, sunitinib, bleomycin, actinomycin, mitomycin, dactinomycin, melphalan, temozolomide, celecoxib, nelarabine, cladribine, isoniazid or derived medicaments, where a carboxy group was converted into amide or hydrazide; RNA or DNA fragments suitable for gene therapy, or derivatives thereof; plain and fluorescent dyes such as 9-aminoacridine and other acridine dyes, DAPI, rhodamine and derivatives thereof, neutral red, trypan blue. It is preferable to add the active substance into the reaction medium as an aqueous solution of an amine salt, for example hydrochloride, nitrate or sulfate, salts of the instant amines are usually better water-soluble. If such an active substance dissolves in water in the acidic environment, but it is poorly soluble in the neutral environment, said substance could be bound within nanoparticles solely by physical interactions. On increase of pH it will precipitate within hydrophobic areas of nanoparticles as nanocrystals.

Preferably, as a polysaccharide, the polysaccharide is used which is soluble in water or other solvents, of the molecular weight up to 1000 kDa, most preferably dextran, starch and derivatives thereof (hydroxyethylstarch), amylose and derivatives thereof, cellulose derivatives (hydroxyethylcellulose, hydroxypropylcellulose, carboxymethylcellulose), glycogen, hyaluronic acid, heparin, alginic acid, carrageen.

Preferably oxidation is carried out up to the degree of oxidation of 0,5% to 80% of the oxidized sugar rings in the polysaccharide.

Preferably oxidation is carried out with the participation of the oxidation agent, which comprises periodate ions (e.g. sodium or potassium periodate), salts of lead with the oxidation number of 4, compounds of copper with the oxidation number of 2 or water oxidized in the presence of suitable catalysts such as e.g. vanadium oxides.

Moreover, the obtained nanoparticles could be modified with antibodies or peptides, or proteins, also by reaction of aldehyde groups of the oxidized polysaccharide with amine groups of peptides.

The aqueous suspension of nanoparticles thus obtained is purified by dialysis, precipitation, centrifugation, or employed directly. The obtained solution of nanoparticles could be lyophilized. For lyophilization, a substance could be added which plays a role of a protecting agent (cryoprotectant), such as a non-modified polysaccharide – e.g. dextran. The obtained nanoparticles display self-organizing capabilities, if lyophilized to a dry powder, after suspending in water or saline, form nanoparticles in several minutes.

According to the invention, a polysaccharide is pre-oxidized in an aqueous solution by adding a predetermined amount of an oxidizing agent. In the course of the specific oxidation,

monosaccharide, e.g. glucose rings are cleaved rather than the polysaccharide chain, and oxidized rings form aldehyde groups. This is one of the typical specific oxidation reaction employed in organic chemistry, as exemplified by a process of mono or polysaccharide oxidation with periodates. [Jeanes, Allene and Wilham, C.A. Periodate Oxidation of Dextran, Journal of American Chemical Society 72.6, (1950): 2655 - 2657]. In the oxidation process, a carbon-carbon bond is cleaved which has –OH groups on the adjacent carbons and aldehyde groups form on the both ends formed. Other methods of partial oxidation could be also employed which lead to formation of aldehyde groups along the polysaccharide chain by cleavage of monosaccharide (glucose) rings and proceeding without cleaving the polysaccharide chain. The degree of oxidation – a number of aldehyde groups can be determined in a known manner; for example by reacting aldehydes with hydroxylamine hydrochloride and titrating the liberated hydrochloric acid [Zhao, Huiru, Heindel, Ned D., Determination of Degree of Substitution of Formyl groups in Polyaldehyde Dextran by the Hydroxylamine Hydrochloride Method, Pharmaceutical Research, 8.3(1991):400–402]. To the prepared polysaccharide molecules, forming compounds are attached by reacting aldehyde groups with amino groups, which leads to spontaneous formation of nanoparticles.

According to the invention, the oxidized polysaccharide is simultaneously modified by at least two kinds of substances with the nature of amine: nanoparticle forming agent(s) hydrophobic by nature, with remaining substances being therapeutic agents or colorants, while it is possible to use in one nanoparticle at the same time several various active substances. This allows to obtain synergistic effect of combined activities of several medicaments. The simultaneous use of several medicaments significantly decreases the possibility of developing drug resistance by the tumor and enables active destruction of a cell regardless of the cell cycle phase. In the process of the invention, due to hydrophobic-hydrophilic interactions the polysaccharide nanoparticle is formed that contains the active substance and the hydrophobic folding agent inside, and its outer layer comprises hydrophilic components, mainly the polysaccharide. It is also possible to simultaneously modify the polysaccharide with the forming agent, the drug(s) and aldehyde group-reactive component which enhances affinity to specific cell types, such as antibodies, nucleotide bases or metabolites, for example folic acid. On formation of nanoparticles, the hydrophobic agents get inside, and the hydrophilic ones are positioned outside the nanoparticles.

The reaction of the invention proceeds in the aqueous environment, under the mild temperature conditions and without organic solvents or surfactants. The obtained nanoparticles are non-toxic as such (if prepared without a toxic drug) and could be used as

carriers for medicaments and color or fluorescent indicators, in the therapy and diagnostics of tumors in particular. The nanoparticles may contain one or more forming agents or several drugs in various combinations, and they may also contain adjuvants (diindolylmethane), that advance their efficiency as medicaments.

The stability of the prepared macromolecules is usually sufficient and reaches from several to over a dozen weeks in the aqueous environment. The dry stability after lyophilization is markedly higher and exceeds a year with proper storage. The stability of the bond formed between amino and aldehyde groups could be additionally enhanced by reducing aldehyde – amine bond.

The process of the invention was illustrated in more detail in the working examples.

Example 1.

Dextran of the molecular weight 70 kDa was oxidized with sodium periodate to oxidize about 5% of glucose rings, and purified. To conduct it, an aqueous dextran solution was prepared and sodium periodate added thereto. Stoichiometry of the reaction depends on oxidation conditions, a molecular weight, and, frequently, the origin of dextran, and equals to from 1 to 2 moles of periodate per a mole of oxidized glucose (two aldehyde groups formed), and has to be verified experimentally. The process of dextran oxidation was conducted at the room temperature in a vessel made of dark glass for one hour. Then, the solution was neutralized and purified by dialysis against distilled water, followed by stripping water *in vacuo*. The number of aldehyde groups was determined by the known hydroxylamine titration method. 5% solution of said dextran in distilled water was prepared. Then, daunorubicin hydrochloride at 15 mol % based on the number of moles of aldehyde groups in the used amount of oxidized dextran was added. The solution was stirred for 20 minutes at 30°C. Then, the 5% aqueous dodecylamine hydrochloride solution was added at 85 mol % based on the initial number of moles of aldehyde groups in the used amount of oxidized dextran and the temperature was raised to 35°C, and the reaction was continued for 60 minutes. The running reaction causes lowering pH of the reaction environment. Then, raising of pH was started by adding 5% aqueous NaOH solution. Adding was conducted in a manner to raise pH to pH 9 in 30 minutes. After pH=9 was reached, the reaction was continued for additional 30 minutes. Alanine was then added at 15 mol % based on the initial number of moles of aldehyde groups in the used amount of oxidized dextran, to bind all unreacted aldehyde groups. After 15 minutes of stirring, the solution was neutralized with 5% hydrochloric acid to pH=7 and purified by dialysis for 24 hours. Then, 20 weight % (based on the initial weight of oxidized dextran) of pure non-oxidized dextran was added as a cryoprotectant and the solution was

lyophilized. The powder was resuspended in water to give a suspension of nanoparticles. The distribution of diameters of the obtained nanoparticles was measured with the Malvern Zeta Sizer apparatus shown on Fig.1. Measurements made with the NanoSight apparatus with a 405 nm laser revealed the slightly lower mean particle diameter and the narrower diameter distribution.

Example 2.

Dextran of the molecular weight 40 kDa was oxidized with sodium periodate to oxidize about 20% of glucose rings, and purified. A 10% solution of such dextran in distilled water was prepared. Then, doxorubicin hydrochloride at 20 mol % based on the number of moles of aldehyde groups in the used amount of oxidized dextran was added. The solution was stirred for 20 minutes at 30°C. Then, the 5% aqueous octylamine hydrochloride solution was added at 80 mol % based on the initial number of moles of aldehyde groups in the used amount of oxidized dextran, and the temperature was raised to 40°C, and the reaction was continued for 60 minutes. The running reaction causes pH lowering. Then, raising of pH was started by adding 5% aqueous NaOH solution. Adding was conducted in a manner to raise pH to pH 8 in 30 minutes. After pH=8 was reached, the reaction was continued for additional 30 minutes. Alanine was then added at 10 mol % based on the initial number of moles of aldehyde groups in the used amount of oxidized dextran. After 15 minutes of stirring, the solution was neutralized with 5% hydrochloric acid to pH=7 and NaBH₃CN was added in 10 mol % excess based on the initial amount of aldehyde groups. Then, the reaction was carried out for 12 hours. The solution was neutralized and purified by extensive dialysis for 48 hours, then dextran was added at 50 weight % based on the initial weight of dextran and the solution lyophilized. After resuspending in water, the distribution of diameters of the obtained nanoparticles was measured with the NanoSight apparatus with a 405 nm laser and shown on Fig. 2.

Example 3.

A 4% aqueous solution of carboxymethylcellulose of the molecular weight about 100 kDa and the oxidation number of 5% was prepared, and pH was adjusted to pH 5. Then, 9-aminoacridine was added as its aqueous hydrochloride solution at 50 mol % based on the initial amount of aldehyde groups of the cellulose derivative used. Then, aqueous octylamine was added at 55 mol % based on the initial number of moles of aldehyde groups. The reaction was carried out at 40°C for one hour. Then the solution was neutralized by raising pH to pH 9 in 15 minutes, left for 30 minutes and dialyzed. The fluorescent nanoparticles were obtained with the 150 nm mean diameter.

Example 4.

Dextran of the molecular weight of 70 kDa was oxidized with sodium periodate to oxidize about 15% of glucose rings, and purified. A 10% solution of such dextran in distilled water was prepared. Then, doxorubicin hydrochloride at 25 mol % based on the number of moles of aldehyde groups in the used amount of oxidized dextran was added. The solution was stirred for 20 minutes at 35°C. Then, folic acid was added at 5 mol % based on the initial amount of aldehyde groups to enhance affinity of nanoparticles to tumor cells. After 15 minutes, the 5% aqueous isoleucine hydrochloride solution was added at 80 mol % based on the initial number of moles of aldehyde groups in the used amount of oxidized dextran, and the temperature was raised to 40°C, and the reaction conducted for 60 minutes. Then, raising of pH was started by adding the 5% aqueous NaOH solution. Adding was conducted in a manner to raise pH to pH 9,5 in 30 minutes. The reaction was continued for additional 30 minutes. Then the solution was neutralized and purified by dialysis for 24 hours. The mean diameter of the obtained nanoparticles was 140 nm.

Example 5

The carboxymethylcellulose sodium salt was oxidized in the aqueous solution with hydrogen peroxide in the presence of tetra-sulfo iron-phthalocyanine catalyst [Weber, J.H. et al, Complexes derived from strong field ligands..., Inorganic Chemistry, 1965, 4, 469-471]. The process was conducted for 12 hours at 40°C, then the product was purified by filtration, followed by dialysis. The amount of aldehyde groups in the obtained aldehyde carboxymethylcellulose derivative was determined by the known hydroxylamine titration method. The 5% solution of the obtained derivative in distilled water was prepared. Then, doxorubicin hydrochloride at 10 mol % based on the number of moles of aldehyde groups in the used amount of oxidized dextran was added. The solution was stirred for 20 minutes at 30°C. Then, the 5% aqueous dodecylamine hydrochloride solution was added at 90 mol % based on the initial number of moles of aldehyde groups in the used amount of oxidized dextran and the temperature was raised to 35°C, and the reaction was continued for 60 minutes. Then, raising of pH was started by adding 5% aqueous NaOH solution. Adding was conducted in a manner to raise pH to pH 9 in 30 minutes. After pH=9 was reached, the reaction was continued for additional 30 minutes. Alanine was then added at 30 mol % based on the initial number of moles of aldehyde groups in the used amount of oxidized dextran, to bind all unreacted aldehyde groups. After 15 minutes of stirring, the solution was neutralized with 5% hydrochloric acid to pH=7 and purified by dialysis for 24 hours. The mean diameter of the obtained nanoparticles as measured with the Malvern Zeta Sizer apparatus was 110 nm.

Example 6

The 1% aqueous hyaluronic acid sodium salt solution was prepared and oxidized with sodium periodate to the degree of oxidation of 5% as in Example 1. The pH was adjusted to pH 5, daunorubicin hydrochloride and cytarabine hydrochloride were added for the each drug comprising 10 mol % of all aldehyde groups of oxidized hyaluronic acid, and the reaction was conducted for 15 minutes at 30 degrees. Then, aqueous decylamine hydrochloride solution was added at 85 mol % based on the initial number of moles of aldehyde groups. The reaction was conducted at 40°C for one hour. Then, pH was raised to pH do 9 within 20 minutes, the solution was neutralized and dialyzed. An aqueous suspension of polysaccharide nanoparticles was obtained comprising two drugs with distinct mechanisms of action.

The discussion of documents, acts, materials, devices, articles and the like is included in this specification solely for the purpose of providing a context for the present invention. It is not suggested or represented that any or all of these matters formed part of the prior art base or were common general knowledge in the field relevant to the present invention as it existed before the priority date of each claim of this application.

Where the terms “comprise”, “comprises”, “comprised” or “comprising” are used in this specification (including the claims) they are to be interpreted as specifying the presence of the stated features, integers, steps or components, but not precluding the presence of one or more other features, integers, steps or components, or group thereof.

The claims defining the invention are as follows:

1. A process for the preparation of nanoparticles from polysaccharides and derivatives thereof, by their specific partial oxidation to produce aldehyde groups and attach compounds with amino or other group with the R-NH₂ bond which react with aldehyde groups, wherein the polysaccharide or derivative thereof is oxidized to obtain aldehyde groups until the oxidation degree of 0,1% to 80% of the sugar rings is obtained, then at least one nanoparticle-forming agent which after attachment of the aldehyde group exhibits hydrophobic properties, selected from the group comprising: aliphatic or aromatic organic amines containing from 4 to 20 carbon atoms, amides and hydrazides of aliphatic and aromatic organic acids containing from 4 to 20 carbon atoms, hydrophobic amino acids, phosphatidylethanolamine, and at least one active substance containing at least one amino, amido or hydrazide group, is added to the solution of the oxidized polysaccharide in water or a mixture of water and an organic solvent, the reaction is conducted at pH of the solution from 1 to 9, at the temperature of from 10 to 100°C, wherein the total molar ratio of amino groups to aldehyde groups is from 20 to 0,5.
2. The process according to claim 1, wherein the nanoparticle-forming agent is added simultaneously with the active substance.
3. The process according to claim 1, wherein the nanoparticle-forming agent is added after adding the active substance.
4. The process according to any one of claims 1 to 3, wherein the nanoparticle-forming agent is selected from the group comprising: butylamine, pentylamine, hexylamine, octylamine, decylamine, dodecylamine, tetradecylamine, hexadecylamine, cyclohexylamine, benzylamine, ethylphenylamine, sphingosines, oleic acid amide, palmitic acid amide, stearic acid hydrazide, palmitic acid hydrazide, oleic acid hydrazide, leucine, isoleucine, valine, methionine, alanine, phenylalanine, cephalin.
5. The process according to any one of claims 1 to 4, wherein as the active substance, the drugs comprising amino, amido or hydrazide groups or derived drugs in which a carboxy group was modified to amide or hydrazide, RNA or DNA fragments suitable for gene therapy, or derivatives thereof, plain and fluorescent dyes, are used.
6. The process according to any one of claims 1 to 5, wherein the active substance is

selected from the group comprising: daunorubicin, doxorubicin, aminoacridins and derivatives thereof, cisplatin and derivatives thereof, methotrexate, cytarabine, gemcitabine, dapsone, acyclovir, azidothymidine, 5-fluorouracil, mercaptopurine, imatinib, sunitinib, bleomycin, actinomycin, mitomycin, dactinomycin, melphalan, temozolomide, celecoxib, nelarabine, cladribine, isoniazid, 9-aminoacridine and other acridine dyes, DAPI, rhodamine and derivatives thereof, neutral red, trypan blue.

7. The process according to any one of claims 1 to 6, wherein both the forming agent and the active substance are added to the reaction in the form of readily soluble salts.
8. The process according to claim 7, wherein hydrochlorides, nitrates or sulfates are used as salts.
9. The process according to any one of claims 1 to 8, wherein dextran, starch and derivatives thereof, amylose and derivatives thereof, cellulose derivatives, glycogen, hyaluronic acid, heparin, alginic acid, carrageen, are used as polysaccharides.
10. The process according to any one of claims 1 to 9, wherein the process is carried out in the water/DMSO, water/acetonitrile, water/ether solvent mixtures.
11. The process according to any one of claims 1 to 9, wherein the oxidation is carried out until the oxidation degree of 0,1% to 80% of the sugar rings in the polysaccharide is obtained.
12. The process according to any one of claims 1 to 11, wherein the oxidation is carried out in the presence of an oxidizing agent comprising periodate ions, salts of lead with the oxidation number of 4, compounds of copper with the oxidation number of 2 or water oxidized in the presence of suitable catalysts.
13. The process according to any one of claims 1 to 12, wherein after the reaction is complete, the reduction of the bonds formed is carried out.
14. The process according to claim 13, wherein the reduction is carried out with NaBH_4 or NaBH_3CN in the aqueous solution.
15. The process according to any one of claims 1 to 14, wherein the aqueous nanoparticle suspension is purified by dialysis, precipitation, centrifugation, or it is used directly.
16. The process according to any one of claims 1 to 15, wherein the obtained nanoparticles are modified with antibodies, peptides or proteins, also by reaction of

aldehyde groups of the oxidized polysaccharide with peptide amino groups.

17. The process according to any one of claims 1 to 16, wherein the solution of the obtained nanoparticles is lyophilized.
18. The process according to claim 17, wherein lyophilization is carried out with a cryoprotective substance added.
19. The process according to claim 18, wherein the cryoprotectant is an unmodified polysaccharide.
20. Nanoparticles prepared according to the process of any one of claims 1 to 19.

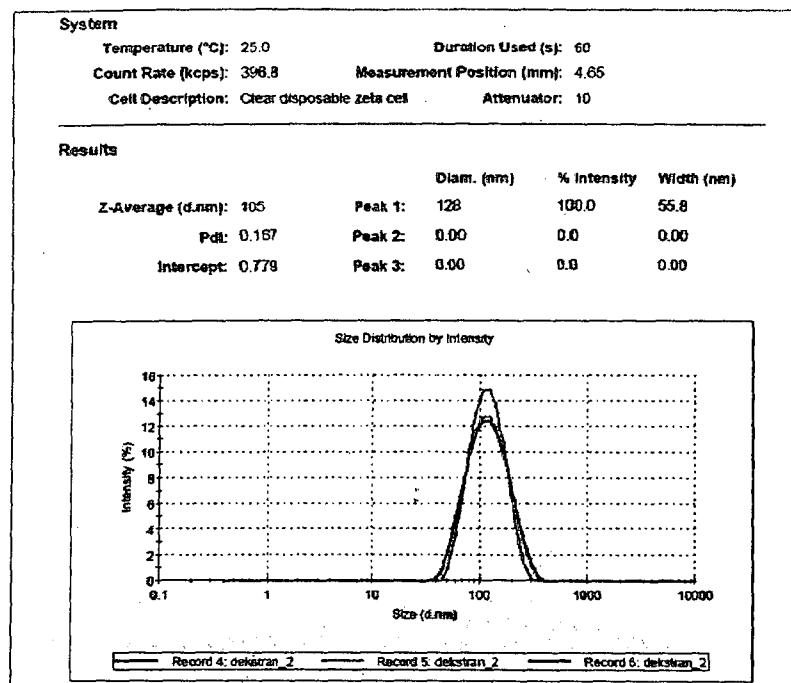


Fig. 1

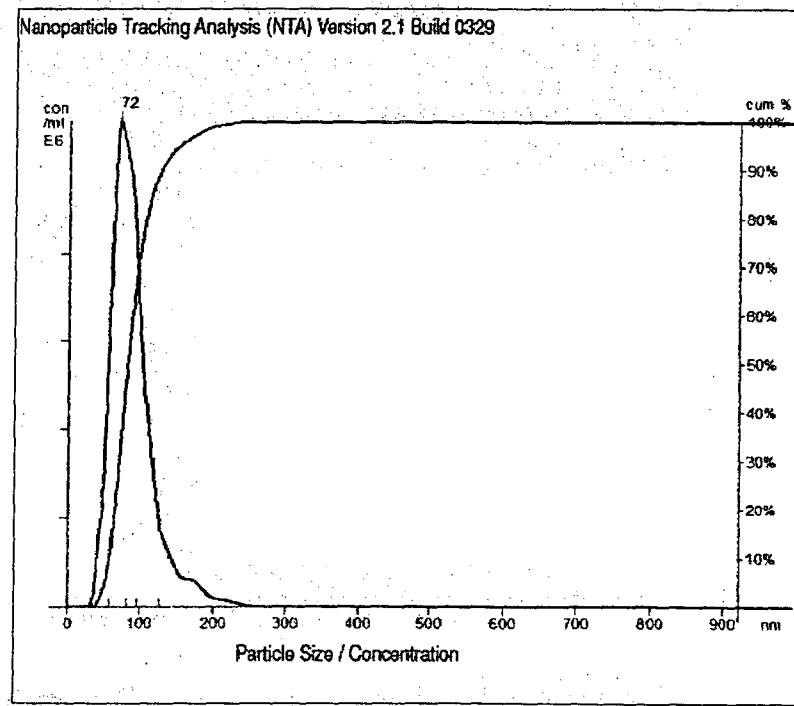


Fig. 2