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## BERYLLIUM-COPPER ALLOYS

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## 4 Claims. (Cl. 75-159)

This invention relates to beryllium-copper al-  
loys, and especially to that general class of alloys  
in which the copper is the primary constituent,  
being present in amounts usually substantially  
greater than about 90% of the entire alloy.

This application is a divisional application of  
Serial No. 261,648 filed March 13, 1939.

Certain general properties of these alloys have  
been known for some time, including the high  
hardenability thereof, but various difficulties and  
disadvantages have been encountered with this  
general class of alloys. Some of these problems  
are discussed more in detail hereinafter, but it is  
here noted that, in general, the problems are met  
by the present invention by virtue of employing  
other metals in the alloy, in addition to the  
beryllium and copper. It should also be kept in  
mind that certain addition ingredients have al-  
ready been proposed for use in copper-beryllium  
alloys, but that the particular ingredients chosen  
in accordance with the present invention and the  
proportions thereof accomplish specially im-  
proved characteristics, as will appear more fully  
hereinafter.

One of the problems which has been en-  
countered in the development and use of the type  
of alloys here under consideration, is that it has  
been very difficult to secure uniformity of char-  
acteristics, especially hardenability, particularly  
when operating on any appreciable commercial  
scale.

Another difficulty which has been encountered  
in connection with the use of beryllium-copper  
alloys is that when such alloys have been em-  
ployed for sand castings there has been a con-  
siderable tendency toward warpage and a de-  
velopment of objectionable surface cracks during  
hardening, which has seriously limited this use.  
The alloys of the present invention aid materially  
in overcoming these difficulties.

As another important object of the invention  
may be mentioned reduction in beryllium con-  
tent, which is desirable because of the relatively  
high cost of beryllium, while at the same time  
obtaining physical and other characteristics  
which are not only equivalent to but, in some  
instances, superior to those of binary beryllium-  
copper alloys or ternary or quaternary alloys as  
heretofore proposed.

The present application is, in part, a continua-  
tion of my prior applications 219,999, filed July  
19, 1938; 107,948, filed October 28, 1936; and  
730,727, filed June 15, 1934, this last having is-  
sued February 18, 1936 as Patent No. 2,030,921.

Referring again to the problem of non-un-

iformity of characteristics, I have found that with  
beryllium-copper alloys in rolled form, a certain  
type of crystal structure is a very important  
factor in making possible uniformity of hardening  
by heat treatment. More particularly, as pointed  
out in application 107,948, the crystal structure  
should be homogeneous, and the addition in-  
gredients should be so chosen, and they and the  
beryllium content should be present in such  
amounts, as to avoid as far as possible a hetero-  
geneous crystal structure incorporating beta  
along with the alpha crystal forms.

As mentioned in said application 107,948, a  
homogeneous crystal structure and high physical  
properties are secured by the employment of  
beryllium from a trace up to about 2%, together  
with either or both of cobalt or silicon in an  
amount aggregating from about .1% to about  
1%.

Still further, as mentioned in said application  
107,948, especially desirable results are attained  
with an alloy containing the range of beryllium  
above indicated, together with both silicon and  
cobalt, the former being present in an amount  
about .2% and the latter in an amount about  
.4%. As indicated in the prior application, cer-  
tain other constituents may also be added, so long  
as they are not of such a nature or present in  
such amounts as to impair the desired homo-  
geneous crystal structure.

The general type of heat treatment which may  
effectively be used to harden alloys of the gen-  
eral class here involved is well known. Briefly,  
it includes quenching from a temperature between  
about 500° C. and 800° C. (for example, in the  
neighborhood of 750° C.) and subsequently re-  
heating at temperatures between 250° C. and 500°  
C. Between the quenching and reheating, the  
alloys can also be cold worked.

With regard to the foregoing type of hardening  
treatment, the present invention is further of  
considerable benefit in making possible the at-  
tainment of substantially uniform hardness with-  
out necessitating the fine or accurate control of  
temperature and time of treatment, such as are  
required with binary beryllium-copper alloys.

As noted in application 730,727 as originally  
filed, improved results can also be secured by  
adding to beryllium-copper (the beryllium be-  
ing present in an amount from a trace up to about  
3%, preferably not more than 2.5%) certain other  
metals, notably silicon, silver, and to a lesser ex-  
tent iron. In accordance with said application  
730,727 (as filed) especially good results are ob-  
tained with alloys containing even as low as 1%

beryllium, or thereabout, when there is added, for example, up to 2% silicon, or 5% silver. The preferred ranges given in said application for these metals are—silicon from about .5% to about 7%, silver from about .5% to about 10%, and iron from about .5% to 2%. The original disclosure of said prior application also points out that more than one of these metals may be present, in which event they may each be present in an amount upwards of a trace, and preferably of a total content of from about .5% to about 10%.

Thus, with the beryllium content kept below the preferred upper limit of 2.5%, the copper would constitute upwards of 87.5%.

Said original disclosure of application 730,727 also brings out that the use of the specified metals in addition to the beryllium and copper aids in raising the recrystallization temperature so that re-softening during heat treatment is not so likely to occur.

Furthermore, with the homogeneous character of many of the alloys of this invention it is possible to employ more cold working, and as a result the invention makes possible greater use of cold working between anneals, this being of advantage since it avoids the necessity for repeated and expensive anneals between cold working steps.

In addition to affording the improvements relating to crystal structure and hardenability, as is pointed out in my prior application 219,999, the specific addition metal cobalt is particularly desirable since it provides high tensile strength and also higher electrical conductivity than with certain other additions, the latter property being important when the alloy is to be used for electrical contact or other similar parts, and the combination of the two properties in a single alloy being unusually advantageous for certain purposes, such as wire required to carry a high tensile load in addition to an electric current.

For the purposes emphasized in my prior application 219,999, the following ranges are preferred:

	Per cent
Be-----	from .5 to 2
Co-----	from .1 to 1
Cu-----	Balance

In most instances the beryllium should be kept within from 1% to 2%, although for special purposes (notably castings) it may be carried up to as high as 2.5%.

The ranges of ingredients providing the best results for the general purposes mentioned in application 219,999 are as follows:

	Per cent
Be-----	from 1.25 to 2
Co-----	from .2 to .5
Cu-----	Balance

A desirable alloy is as follows:

	Per cent
Be-----	2
Co-----	.25
Cu-----	Balance

The cobalt addition is also particularly useful for sand castings which, with other alloys of this general type, have had a tendency to develop surface cracks during hardening treatment, the surface cracking being greatly reduced by the employment of cobalt. For these purposes, as is

mentioned in application 219,999, ranges as follows should preferably be used:

	Per cent
Be-----	from 1.75 to 2.25
Co-----	from .3 to 1
Cu-----	Balance

Castings from this alloy may be hardened to a very high degree.

As to conductivity, it is preferred to maintain the composition such that the electrical conductivity in the hardened condition equals at least 22% that of copper.

From the foregoing it will be seen that, as brought out in my prior applications, cobalt and silicon are of outstanding importance in improving various characteristics of beryllium-copper alloys, particularly hardenability, electrical conductivity and elongation, these ends being achieved with the use of a smaller amount of beryllium than would be required in a binary alloy for an equivalent degree of hardness.

The improvement in connection with uniformity of hardenability is especially noticeable when employing silicon and/or cobalt, the combination of silicon with cobalt, as mentioned in application 107,948, being especially advantageous with respect to the homogeneity of the crystal structure and uniformity of hardenability.

The addition of iron is also of importance, especially when employed in combination with silicon.

The silver addition is particularly suitable where the alloy is to be employed for special types of electrical parts, such as electrodes.

The cobalt addition is claimed in my copending application 219,999 above referred to.

To summarize the foregoing, it may be noted that ranges of metals given below may be employed in accordance with the invention.

Note first that the beryllium content, particularly for pieces to be rolled, should preferably be kept below about 2.5% and most desirably between an amount substantially greater than a trace and about 2%.

When silicon alone is used, the composition should be as follows:

Be-----	from a substantial amount above a trace up to about 2.5%
Si-----	from a substantial amount above a trace or about .1% to about 2%
Cu-----	balance

When cobalt is used in combination with silicon, the composition should be within the following ranges:

Be-----	from a substantial amount above a trace up to about 2.5%
Si-----	from a substantial amount above a trace or about .1% to about 2%
Co-----	from about .1% to about 1%
Cu-----	balance

When employing both cobalt and silicon, it is desirable that the cobalt should be present in an amount about twice that of the silicon content. For instance, the following composition is effective:

Be-----	from a substantial amount above a trace up to about 2.5%
Si-----	.2%
Co-----	.4%
Cu-----	balance

Silicon and cobalt together produce good results when the combined content of the two equals from about .1% to about 1%.

Where iron is used alone, the composition may be as follows:

Be\_\_\_\_\_from a substantial amount above a trace up to about 2.5%

Fe\_\_\_\_\_from a substantial amount above a trace, or preferably from about .5%, up to about 2%

Cu\_\_\_\_\_balance

If iron is used with certain other additions, for instance with silicon, the composition may be as follows:

Be\_\_\_\_\_from a substantial amount above a trace up to about 2.5%

Si\_\_\_\_\_from a substantial amount above a trace or about .1% to about 2%

Fe\_\_\_\_\_from a substantial amount above a trace, or preferably from about .5%, up to about 2%

Cu\_\_\_\_\_balance

In the last table above, it should be noted that the combined content of silicon and iron should preferably be within from .5% to about 10%. Still further, when some other metal, for instance silver, is used in combination with the iron and silicon, the total of the three should be kept within about .5% to about 10%.

Where silver is employed alone, the following composition may be used:

Be\_\_\_\_\_from a substantial amount above a trace up to about 2.5%

Ag\_\_\_\_\_from a substantial amount above a trace, and preferably from about .5%, to about 10%

Cu\_\_\_\_\_balance

In the event that silver is used in combination with other metals, such as silicon or iron, the ranges above indicated for these metals may be employed, and the total content of the group

should again be kept within about .5% to about 10%.

In all tables above and also in various of the claims, it will be understood that where the copper content is referred to (for instance— "Cu\_\_\_\_\_ balance") it is intended to include small amounts of other ingredients and/or characteristic impurities, so long as they do not materially alter the characteristic properties of the alloys.

What I claim is:

1. An alloy composed of the following ingredients in the proportions indicated:

Be\_\_\_\_\_from a substantial amount above a trace up to about 2.5%

Si\_\_\_\_\_from a substantial amount above a trace or about .1% to about 2%

Co\_\_\_\_\_from about .1% to about 1%

Cu\_\_\_\_\_balance

2. An alloy composed of the following ingredients in the proportions indicated:

Be\_\_\_\_\_from a substantial amount above a trace up to about 2.5%

Si\_\_\_\_\_2%

Co\_\_\_\_\_4%

Cu\_\_\_\_\_balance

3. An alloy composed of the following ingredients in the proportions indicated:

Be\_\_\_\_\_from a substantial amount above a trace to about 2%

Si } (combined content from about .1% to  
Co } about 1%

Cu\_\_\_\_\_balance

4. An alloy composed of the following ingredients in the proportions indicated:

Be\_\_\_\_\_from a substantial amount above a trace to about 2%

Si\_\_\_\_\_2%

Co\_\_\_\_\_4%

Cu\_\_\_\_\_balance

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