

Office de la Propriété Intellectuelle du Canada

Un organisme d'Industrie Canada

Canadian Intellectual Property Office

Industry Canada

An agency of

CA 2068824 C 2002/12/24

(11)(21) 2 068 824

(12) BREVET CANADIEN CANADIAN PATENT

(13) **C**

(22) Date de dépôt/Filing Date: 1992/05/15

(41) Mise à la disp. pub./Open to Public Insp.: 1992/11/19

(45) Date de délivrance/Issue Date: 2002/12/24

(30) Priorité/Priority: 1991/05/18 (P 41 16 441.5) DE

(51) Cl.Int.⁵/Int.Cl.⁵ A01N 47/36, A01N 25/30

(72) Inventeurs/Inventors: FRISCH, GERHARD, DE; MAIER, THOMAS, DE

(73) Propriétaire/Owner: HOECHST AKTIENGESELLSCHAFT, DE

(74) Agent: FETHERSTONHAUGH & CO.

(54) Titre: DISPERSIONS AQUEUSES DE DERIVES DE SULFONYLUREE (54) Title: AQUEOUS DISPERSIONS OF SULFONYLUREA DERIVATIVES

(57) Abrégé/Abstract:

Herbicidal active substances from the sulfonylurea series were hitherto mainly formulated as WPs or WGs. The invention relates to novel, storage-stable aqueous suspensions of these active substances, comprising a polyacrylic acid derivative, to processes for their preparation, and to their use in crop protection.





HOE 91/F 151

Abstract of the disclosure

Aqueous dispersions of sulfonylurea derivatives

Herbicidal active substances from the sulfonylurea series were hitherto mainly formulated as WPs or WGs. The invention relates to novel, storage-stable aqueous suspensions of these active substances, comprising a polyacrylic acid derivative, to processes for their preparation, and to their use in crop protection.

HOECHST AKTIENGESELLSCHAFT 91/F 151

Dr. WS/PI

Description

Aqueous dispersions of sulfonylurea derivatives

5 The invention relates to aqueous preparations of sulfonylurea derivatives.

Active substances from the series of the sulfonylureas were hitherto formulated in particular as WPs (wettable powders) or WGs (water-dispersible granules).

The previous attempts to prepare aqueous suspensions of this substance class, where the active substance in a solid form is finely dispersed in the aqueous phase, have proven difficult. One reason for this may lie in the fact that some of these compounds are very highly sensitive to hydrolysis. EP-A-124,295 discloses the preparation of useful aqueous formulations with suitable and compatible inorganic acid salts or certain carboxylic acid salts, in a pH range of 6 - 10. However, in doing this, exact concentration conditions and temperatures must be adhered to, which makes the procedure complicated.

Surprisingly, it has now been found that aqueous dispersions of the abovementioned active substances can be prepared in a simple manner with the aid of surfactants from the class of the polyacrylic acid derivatives. These formulations are extraordinarily storage-stable. For example, the aqueous formulations of amidosulfuron (3-(4,6-dimethoxy-pyrimidin-2-yl)-1-(N-methylsulfonyl)-urea) are storage-stable over at least 3 months in a range from -10 °C to 40 °C and are not subject to chemical degradation. Storage at 50 °C over three months may result in a slight chemical degradation of the substance, without this endangering the stability of the suspension.

The surfactants mentioned protect the solid suspended active substance in such a manner that hydrolysis of the active substance is virtually completely prevented. They also exhibit their protective action when combined with one or more additional surfactants.

5

The invention therefore relates to aqueous herbicidal preparations which contain at least one herbicidal active substance from the series of the sulfonylureas and at least one polyacrylic acid derivative.

Suitable herbicides from the sulfonylurea series are 10 pyrimidine- or triazinylaminocarbonyl-[benzene-, pyridine-, pyrazole-, thiophene- and (alkylsulfonyl)alkylamino-]sulfamides. Preferred substituents on the pyrimidine ring or triazine ring are alkoxy, alkyl, haloalkoxy, haloalkyl, halogen or dimethylamino, it being 15 possible for all substituents to be combined independently of one another. Preferred substituents in the benzene, pyridine, pyrazole, thiophene or (alkylsulfonyl)alkylamino moiety are alkyl, alkoxy, halogen, nitro, 20 alkoxycarbonyl, aminocarbonyl, alkylaminocarbonyl, dialkylaminocarbonyl, alkoxyaminocarbonyl, alkyl, alkoxyaminocarbonyl, haloalkoxy, haloalkyl, alkylcarbonyl, alkoxyalkyl and (alkanesulfonyl)alkylamino.

Examples of suitable sulfonylureas are

- B1) phenyl- and benzylsulfonylureas and related compounds, for example 1-(2-chlorophenylsulfonyl)-3-(4-methoxy-6-methyl-1,3,5-triazin-2-yl)-urea (chlorsulfuron),
- 1-(2-ethoxycarbonylphenylsulfonyl)-3-(4-chloro-6-methoxypyrimidin-2-yl)-urea (chlorimuron-ethyl),
 - 1-(2-methoxyphenylsulfonyl)-3-(4-methoxy-6-methyl-1,3,5-triazin-2-yl)-urea (metsulfuron-methyl),
 - 1-(2-chloroethoxy-phenylsulfonyl)-3-(4-methoxy-6-methyl-
 - 1,3,5-triazin-2-yl)-urea (triasulfuron),

- 1-(2-methoxycarbonyl-phenylsulfonyl)-3-(4,6-dimethyl-pyrimidin-2-yl)-urea (sulfometuron-methyl),
- 1-(2-methoxycarbonylphenylsulfonyl)-3-(4-methoxy-6-methyl-1,3,5-triazin-2-yl)-3-methylurea
- 5 (tribenuron-methyl)
 - 1-(2-methoxycarbonylbenzylsulfonyl)-3-(4,6-dimethoxy-pyrimidin-2-yl)-urea (bensulfuron-methyl)
 - 1-(2-methoxycarbonylphenylsulfonyl)-3-(4,6-bis-(difluoro-methoxy)-pyrimidin-2-yl)-urea (pyrimisulfuron-methyl),
- 3-(4-ethyl-6-methoxy-1,3,5-triazin-2-yl)-1-(2,3-dihydrol,1-dioxo-2-methylbenzo[b]thiophene-7-sulfonyl)-urea (see EP-A-79,683) and
 - 3-(4-ethoxy-6-ethyl-1,3,5-triazin-2-yl)-1-(2,3-dihydro-
- 1,1-dioxo-2-methylbenzo[b]thiophene-7-sulfonyl)-urea (see
- 15 EP-A-79,683),
 - B2) thienylsulfonylureas, for example 1-(2-methoxycarbonylthiophen-3-yl)-3-(4-methoxy-6-methyl-1,3,5-triazin-2-yl)-urea (thifensulfuron-methyl),
 - B3) pyrazolylsulfonylureas, for example
- 1-(4-ethoxycarbonyl-1-methylpyrazol-5-yl-sulfonyl)-3(4,6-dimethoxypyrimidin-2-yl)-urea
 (pyrazosulfuron-methyl) and
 methyl 3-chloro-5-(4,6-dimethoxypyrimidin-2-ylcarbamoyl-sulfamoyl)-1-methyl-pyrazole-4-carboxylate (see EP-A-282,613),
- B4) sulfonediamide derivatives, for example

 3-(4,6-dimethoxypyrimidin-2-yl)-1-(N-methyl-N-methyl-sulfonylaminosulfonyl)-urea (amidosulfuron) and structural analogs (see EP-A-0,131,258 and Z. Pfl.

 30 Krankh. Pfl. Schutz, Special Edition XII, 489-497 (1990)),
 - B5) pyridylsulfonylureas, for example 1-(3-N,N-dimethylaminocarbonylpyridin-2-yl-sulfonyl)-3-(4,6-dimethoxypyrimidin-2-yl)-urea (nicosulfuron),

1-(3-ethylsulfonylpyridin-2-yl-sulfonyl)-3-(4,6-dimethoxy-pyrimidin-2-yl)-urea (DPX-E 9636, see Brighton Crop Prot. Conf. - Weeds - 1989, p. 23 et seq.), pyridylsulfonylureas as are described in European Patent Application EP-A-0510032, preferably those of the formula I or salts thereof,

in which

E is CH or N, preferably CH,

10 R⁸ is iodine or NR¹³R¹⁴,

is hydrogen, halogen, cyano, (C_1-C_3) -alkyl, (C_1-C_3) -alkoxy, (C_1-C_3) -haloalkyl, (C_1-C_3) -haloalkoxy, (C_1-C_3) -alkylmercapto, (C_1-C_3) -alkoxy- (C_1-C_3) -alkyl, (C_1-C_3) -alkoxy-carbonyl, mono- or di- (C_1-C_3) -alkyl-amino, (C_1-C_3) -alkyl-sulfinyl or -sulfonyl, $SO_2-NR^aR^b$ or $CO-NR^aR^b$, in particular H,

 R^a and R^b independently of one another are hydrogen, (C_1-C_3) -alkyl, (C_1-C_3) -alkyl, (C_1-C_3) -alkynyl, or together are $-(CH_2)_4-$, $-(CH_2)_5-$ or $(CH_2)_2-O-(CH_2)_2-$,

20 R^{10} is H or CH_3 ,

- R^{11} is halogen, (C_1-C_2) -alkyl, (C_1-C_2) -alkoxy, (C_1-C_2) -haloalkyl, preferably CF_3 , (C_1-C_2) -haloalkoxy, preferably OCH_2 or OCH_2CF_3 ,
- R^{12} is (C_1-C_2) -alkyl, (C_1-C_2) -haloalkoxy, preferably OCHF₂, or (C_1-C_2) -alkoxy, and
 - is (C_1-C_4) -alkyl and R^{14} is (C_1-C_4) -alkylsulfonyl, or R^{13} and R^{14} together are a chain of the formula $-(CH_2)_3SO_2-$ or $-(CH_2)_4SO_2-$, for example 3-(4,6-dimethoxypyrimidin-2-yl)-1-(3-N-methylsulfonyl-N-methylaminopyridin-2-yl)-sulfonylurea,
 - B6) alkoxyphenoxysulfonylureas as are described in EP-A-0,342,569, preferably those of the formula II or salts thereof,

$$\begin{pmatrix} R_{16} \end{pmatrix}_{n} \begin{pmatrix} R_{15} \\ 0 \\ 0 \end{pmatrix}_{N} \begin{pmatrix} 0 \\ N \\ N \end{pmatrix}_{R} \begin{pmatrix} 0 \\ 17 \end{pmatrix} \begin{pmatrix} 0 \\ 0 \\ 0 \end{pmatrix}_{R} \begin{pmatrix} 0 \\ 19 \\ 0 \end{pmatrix}_{R} \begin{pmatrix}$$

in which

10

15 E is CH or N, preferably CH,

R¹⁵ is ethoxy, propoxy or isopropoxy,

 R^{16} is hydrogen, halogen, nitro, CF_3 , CN, (C_1-C_4) -alkyl, (C_1-C_4) -alkoxy, (C_1-C_4) -alkylmercapto or (C_1-C_3) -alkoxy-carbonyl, preferably in the 6-position on the phenyl ring,

n is 1, 2 or 3, preferably 1,

 R^{17} is hydrogen, (C_1-C_4) -alkyl or (C_3-C_4) -alkenyl,

 R^{18} and R^{19} independently of one another are halogen, (C_1-C_2) -alkyl, (C_1-C_2) -alkoxy, (C_1-C_2) -haloalkyl, C_1-C_2 -haloalkoxy or (C_1-C_2) -alkoxy- (C_1-C_2) -alkyl, preferably OCH_3 or CH_3 , for example 3-(4,6-dimethoxy-pyrimidin-2-yl)-1-(2-ethoxyphenoxy)-sulfonylurea,

and other related sulfonylurea derivatives, and their mixtures.

The content of sulfonylurea derivatives, preferably sulfonediamide derivatives such as amidosulfuron, is preferably 0.1 - 60 % by weight, in particular 1 - 45 % by weight.

10

15

Suitable surfactants are in particular those polyacrylic acid derivatives as are available, for example, under the trade names *Sokalan CP10 (BASF), the *Geropon series (HB, DA, DG) (Rhone Poulenc) or *Dispersant series (Rhone Poulenc) or the *Degapas series (Degussa).

[®]Sokalan CP10 is a modified low molecular weight sodium polyacrylate which is prepared by a specific polymerization process (BASF Techn. Info TI/P 3039d, 1988.)

The [®]Geropon types HB, DA and DG and [®]Dispersant HB and FB are, according to Rhone-Poulenc Data Sheets of 1979 and 1989 respectively, alkali metal polyacrylates which are available both in liquid and in solid form.

The [®]Degapas series also consists of alkali metal salts, or ammonium salts, of polyacrylic acid derivatives.

These polyacrylic acid derivatives can preferably be employed in a range from 0.1 - 30 % by weight, but, in particular, from 0.5 - 20 % by weight.

As an additional surfactant which helps improve the dispersing properties of the solid suspended particles,

but which is not essential, the preparations according to the invention may contain a dodecyl- or tridecyl-benzenesulfonate in amounts of 0.01 - 12 % by weight. For example, [®]Maranil (sodium dodecylbenzenesulfonate), manufactured by Henkel, can be employed as a paste or a powder.

5

10

15

30

It is furthermore also possible to add up to 25 % by weight, preferably up to 15 % by weight, of commercially available adjuvants such as wetting agents, dispersants, defoamers, thickeners, preservatives and antifreeze agents.

Suitable additional wetting agents and dispersants are, for example, tributylphenol polyglycol ethers, such as the [®]Sapogenat T brands (Hoechst) or nonylphenol polyglycol ethers, such as the [®]Arkopal N brands (Hoechst).

Examples of suitable defoamers are those based on silicone, such as those from the **Silcolapse series (Rhone Poulenc) or Antischaummittel SH (Wacker).

Thickeners can be of inorganic or organic nature; they
can also be combined. Suitable examples are those based
on aluminum, xanthan, methylcellulose, polysaccharide,
alkaline earth metal silicate, gelatine and polyvinyl
alcohol such as, for example, Bentone EW, Veegum,
Rhodopol 23 or Kelzan S. If necessary, preservatives
can be used, for example those based on formaldehyde,
benzoic acid and triphenyltin such as, for example,
Robate C.

It is furthermore also possible to add antifreeze agents such as urea, salts, polyols (for example glycol, propylene glycol or glycerol) or sugars.

The invention furthermore relates to a method of controlling undesired vegetation, which comprises applying

a preparation according to the invention to plants, seeds of plants or the area under cultivation.

The Examples 1 - 22 which follow (Table 1) are intended to illustrate the herbicidal preparations according to the invention without restricting the invention thereto. The suspensions are prepared by wet grinding in a manner conventionally used by those skilled in the art (quantities are in percent by weight). The degradation rates are given in the lower part of the tables.

- Examples 18 and 19 are mixed formulations of isoproturon 10 and amidosulfuron which, at an initial value of 1.5 %, contain 1.47 and 1.46 %, respectively, of amidosulfuron after 5 months at room temperature (RT), and the content after 3 months at 40 °C is 1.40 and 1.42 %, respectively. Example 20 is a mixture of a commercially available 15 ®Arelon dispersion (DE-A-2,924,403) and an amidosulfuron dispersion content according to Example 2 in Table 1. 81.7% of [®]Arelon dispersion, 6% of amidosulfuron dispersion (Example 2) and 12.3 % of water were mixed thoroughly, and the mixture was stored. Degradation after 20 3 months at 40 °C is 1.11 %, and after 3 months at 50 °C content is < 0.1 %, while 0.4 and 0.49 %, the respectively, were still detectable in Examples 18 and 19 after 3 months at 50 °C.
- Besides the possibility of preparing a stable aqueous 25 formulation of sulfonylureas in the manner described herein, it is also possible to prepare mixed formulations of sulfonylureas with other biologically active substances in the same manner. This makes sense in particular if an addition of one or more further active 30 substances can result in a broadened biological spectrum of action, or even lead to synergism. Mainly suitable are herbicidally active substances such as here phenylurea derivatives known under the common names (Pesticides Manual) isoproturon, linuron, monolinuron, 35

monuron, diuron, neburon, chlortoluron, fluometuron, chlorbromuron, chloroxuron, fenuron, siduron, terbuthiuron, tetrafluron and metoxuron.

Further substances which are to be mentioned are those
from the class of the triazines such as tributylazin,
terbutryn, simetryn, simazine, secbumeton, propazine,
prometryn, procyazine, metribuzin, cyprazine, cyanazin,
atrazine, prometon, hexazinone, amitrol, dipropetryn,
desmetryn and ametryn. The following are furthermore
suitable: pendimethalin, trifluralin, CMPP, MCPA, 2, 4,
D, glufosinate, glyphosate, nitrofen, bifenox, diclofop,
ioxynil, bromoxynil, paraquat, ethalfluralin,
ethofumesate, diflufenican, metolachlor, alachlor,
acifluorfen, methalpropalin, nitralin, nitrofluorfen,
fluoroglycofen, oxyfluorfen, profluralin, propachlor,
2,4-DB, 2,4-DP and bentazone.

These formulations can be formulated in the form of pure mixed dispersions and in the form of suspoemulations as are described, for example, in EP-A-117,999 and EP-A-261,492.

 ∞ 4 25 3 ∞ 25 0.1 0 IJ 0 $\boldsymbol{\omega}$ 15 25 $\boldsymbol{\omega}$ 0.1 $\boldsymbol{\omega}$ 15 25 ന $\boldsymbol{\omega}$ o 0,1 25 8 9 4 $\boldsymbol{\omega}$ 8 0,1 5 2 $\boldsymbol{\omega}$ 0.1 2 $\boldsymbol{\omega}$ 3 5 2 $\boldsymbol{\omega}$ 2 2 $\boldsymbol{\omega}$ 25 0, 4 ∞ Antifreeze Rhodopol Degapas Geropon O Sokalan Maranii Kobate mido- \mathbf{m}

able 1

Table 1, continuation

	12	13	4	<u>t</u>	9	17	18	19	20	21	22
	25	25	25	25	45	30	1.55	1.55	*	25	25
	·)	
calan		6.0			10	10	10	10		, T	-
										2	2
pou		7.5	6.0	9							
2	8										
Maranil		(S)	2	-	2	2	2.0	4.0		4.0	4.0
freeze	8	8	8	8	8	8	80	8		8	00
dopol	0.2	0.2	0.2	0.2	0.1	0.1	0.1	0.1		0.2	
											<u> </u>
*Kobate C	0.1	0.1	0.1	0.1	0.1	0.1	0.1	0.1		0.1	0.1

Table 1, continuation

	-	2	3	4	5	9	7	8	6	10	1.1
Defoamer	~~	-	•	~	~	—					
•Darvan Nr. 3											
isoproturon											
Water to 100 %											
Analysis											
Initial	23,4	26,9			25,5						
3 months at	23,6	25,6			24,0	24.0	24.5				
3 months at 50 °C	2				23,7	20,5	22.0				

Table 1, continuation

22	-					22.7	23	
21						24.3	23.8	
28	*					1.60	# . # . # . # . # . # . # . # . # . # .	<0.1
19	2.0	1.0	38			1.51	1.42	0.49
18	2.0	1.0	38.0			1.49	1.40	0.4
17								
16	—	-						
15	—							
14		7						
13		-				23.6		
12							24.3	21.5
	Defoamer	•Darvan Nr. 3	Isoproturon	Water to 100 %	Analysis	Initial	3 months at	3 months at 50 °C

 $\boldsymbol{\omega}$ 9 S 3 continuation months 3 months
at RT
at months
at RT 5 months at RT

able 1, continuation

	12	13	14	15	16	17	18	19	20	21	22
3 months at RT											
14 months											23.2
at RT	•										
5 months at RT							1.47	1.46			

) see note above

The examples of Table 2 which follows are comparative examples in which surfactants, which are decisive as far as stability is concerned, from the class of the polyacrylates are exchanged for other, commercially available surfactants.

These examples demonstrate that the dispersion in Example 1a was virtually stable from the physical point of view, but chemical degradation by 21.4 % occurred after 3 months' storage at 50 °C resulting in a content of 27.5 %. Example 5a was not suitable for grinding. Example 3a was unstable from the physical point of view after one month and degradation resulted in a content of 39.9 %; similarly Example 6a, where degradation resulted in 37.2 %. Example 4a was unstable after 2 months, and degradation resulted in 37.7 %. Equally, Example 2a is unstable after one month at 50 °C and degradation resulted in a content of 38.8 %.

In contrast, the examples according to the invention listed in Table 1 show good stability values even after 3 months at 50 °C. With an initial value of 22.7 %, a value of 23.2 % - within the analytical error margin - for Example 22 after 14 months at RT means no significant degradation.

	12	2a	33	4 3	53	еа
Amidosulfuron	35	45	45	45	45	
Orotan 850		6.0				
Beroi 733				10	15	5
Soprophor 3033		3				
Genapol X 060		2				
Genapol X 080						
Vanisperse CB	5			30		S.
Darvan No. 3						
Antifreeze	8	8	8	8	8	
Thickener	0.1		0.1	0.1	0,1	0.1
Preservative						
Defoamer						

rable :

Table 2, continuation

	1	2a	33	43	5a	ба
Borresperse 3A			2			
Vanicell P					5	
H ₂ O to 100						
Degradation						
1 month at 50 °C	- 5.5 %	- 6.2 %	-5.1%		golid	. 7.8 %
2 months at 50 °C	. 19.3 %			. 7.3 %	when	
3 months at 50 °C	- 21.4 %				ground	
Stability	< 3 months	< 1 month	< 1 month	< 2 months		< 1 month

CLAIMS:

- 1. An aqueous herbicidal preparation comprising 0.1 60 % by weight of at least one herbicidal active substance from the sulfonylurea series and 0.1 30 % by weight of at least one polyacrylic acid derivative.
- A preparation as claimed in claim 1, comprising 1 45 % by weight of at least one herbicidal active substance from the sulfonylurea series and 0.5 20 % by weight of at least one polyacrylic acid derivative.
 - 3. A preparation as claimed in claim 1 or 2, addition-ally comprising
- 0.01 12 % by weight of a dodecyl- or tridecylbenzenesulfonate and 0 25 % by weight of conventional adjuvants from the series of the wetting
 agents, dispersants, defoamers, thickeners, preservatives and antifreeze agents.
- 4. A preparation as claimed in any one of claims 1 to 20 3, additionally comprising
 - 0.1 9 % by weight of a dodecyl- or tridecyl-benzenesulfonate and 0 15 % by weight of conventional adjuvants from the series of the wetting agents, dispersants, defoamers, thickeners, preservatives and antifreeze agents.
 - 5. A preparation as claimed in any one of claims 1 to 4, comprising at least one active substance from the series of the sulfonediamide derivatives.

- A preparation as claimed in claim 5 wherein the substance from the series of sulfonediamide derivatives is amidosulfuron.
- 7. A preparation as claimed in any one of claims 1 to 5, which exists in the form of a suspension.
 - 8. A preparation as claimed in any one of claims 1 to 5, which exists in the form of a suspoemulsion.
- 9. A process for preparing a preparation as claimed in any one of claims 1 to 8, which comprises combining the components and wet grinding the mixture.
 - 10. The use of a preparation as claimed in any one of claims 1 to 8 for controlling undesired vegetation.
- 11. A method of controlling undesired vegetation, which comprises applying a preparation as claimed in any one of claims 1 to 8 to plants, seeds of plants or the area under cultivation.

FETHERSTONHAUGH & CO.
OTTAWA, CANADA

PATENT AGENTS